

1 **Title: Reflectance: current state of research and future directions for archaeological**
2 **charcoal: results from a pilot study on Irish Bronze Age cremation charcoals**

3
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17 **Highlights**

- 18 • This study is a first attempt to use the reflectance method to measure absolute
19 burn temperatures from charcoal of archaeological cremation burials.
- 20 • Reflectance as a method for estimating charcoal burn temperature has been under
21 investigation for some time. To date studies have used modern pre-charred wood
22 samples (formed under a variety of conditions) to establish basic performance
23 parameters and calibration curves to apply to charcoals recovered from both
24 archaeological and modern contexts.
- 25 • The method shows promise in its application to archaeologically recovered
26 charred materials, especially wood, although to date, only a small number of
27 studies have been completed.
- 28 • Reflectance results of the charcoal did not demonstrate the range of expected
29 temperatures associated with cremation (ca. 650°C to as much as 1000°C). A
30 variety of explanations are considered.

- 31 • In particular, proof of this method's utility in archaeology will require better
32 rationalisation of the calibration curves used to date as these currently represent a
33 variability of typically 100-150°C (and up to as high as 180°C) for any one
34 reflectance value
- 35 • Un-sieved soil samples should be collected routinely for this method to gather the
36 smallest charcoal fractions

37

38 **Keywords**

39 Reflectance, charcoal absolute burn temperature, reflectance calibration curve, cremation
40

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101 **Illustrations**

102

103 *Figures*

104 Figure 1: Calibration graph of reflectance vs burn temperature

105

106 Figure 2: Map of site locations

107

108 Figure 3: Cremation pits and possible pyre pits, looking southwest. (Doody, 2008, p. 117)

109

110 Figure 4: Graph of comparison of calibration curves of McParland and Braadbart *et al.*

111

112 Figure 5: Graph of Hudspith calibration curve

113

114 *Tables*

115 Table 1: Context types, characteristics, and reflectance results of charcoals tested

116

117 Table 2: Absolute temperatures inferred from reflectance, by context type, showing
118 contexts containing bone vs those not containing bone

119

120 Table 3: Variations in reflectance readings from six different studies, from McParland
121 (2010, ch 10).

122 The maximum difference of 1.81 (at 500 °C) represents a potential variation of up to 180
123 °C (i.e. +/- 90 °C). This is the most extreme variation, and most fall within a range
124 representing a variation of ca. 100 °C (i.e. +/- 50 °C).

125 **Abstract**

126

127 *'Reflectance' is a method that estimates the absolute burn temperature of charcoal from*
128 *the 'shininess' of resin mounted samples. The method's usefulness for archaeological*
129 *charcoal is yet to be comprehensively studied. This report details first results from*
130 *reflectance testing of archaeological charcoals excavated from Irish Bronze Age*

131 *cremations, which included calcined bone. As calcination of bone commences at 650 °C,*
132 *it was expected that the charcoals would reflect at least this temperature. This was not*
133 *the case for taxonomically identified charcoals >2mm, nor for micro-charcoals of c.*
134 *250µm, although measured temperatures rose slightly with decreasing fraction size of*
135 *charcoal remains. Depositional practice, combustion completeness and taphonomic*
136 *influences may have all played a part in this result, and these will need careful*
137 *consideration in different archaeological circumstances. However, the greatest challenge*
138 *for reflectance of archaeological materials lies in obtaining full agreement on the*
139 *production and use of reflectance calibration curves. Current calibration curves differ*
140 *substantially, by 100-150 °C (+/- 50-75°C) and in one instance up to as much as 180 °C*
141 *(+/- 90°C). Without better agreement on calibration, the method's ultimate usefulness in*
142 *archaeological research will be limited. At the level of refinement currently possible, it*
143 *will still be useful for determining very high or very low temperature processes, and*
144 *possibly the difference between charcoal fuel and raw wood fuel fires. The latter has*
145 *distinct implications for estimating ancient forest wood consumption, since more wood is*
146 *consumed in processes employing charcoal fuel. Proving the utility of reflectance for*
147 *archaeological purposes may also require modification of normal practice for*
148 *archaeological field collection of charcoal, to include collection and laboratory*
149 *processing of un-sieved soil samples.*

150

151

152 **1 Introduction**

153

154 In the last forty years, the systematic study of archaeological charcoal has greatly
155 increased our knowledge of past environments as well as socio-economic activity relating
156 to fuel collection and consumption (Asouti and Austin, 2005; Chabal, 1992; Chabal *et al.*,
157 1999, Dufraisse, 2006, Théry-Parisot *et al.*, 2010, 143). In the laboratory, charcoal
158 analysis has been substantially limited to taxonomic identification, and more recently,
159 estimation of cropping indicators by examination of annual tree ring patterns, see for
160 example: Ludemann (2006), Veal (2012), Marguerie and Hunot (2007) . More recently,
161 attempts to further characterise modern charred materials in terms of their chemical and
162 physical characteristics have been made in experimental procedures designed to assist
163 archaeologists in their interpretation of ancient charred remains (Braadbaart *et al.*, 2009;
164 Braadbaart *et al.*, 2012; Braadbaart *et al.*, 2016; Chrzavzez *et al.*, 2011; Chrzavzez *et al.*,
165 2014, Lancelotti *et al.*, 2010).

166

167 Reflectance testing has been attested as a useful tool for demonstrating approximate burn
168 temperatures of modern, and some archaeological charcoal (Braadbaart and Poole, 2008;

169 McParland *et al.* 2009a,b; McParland *et al.*, 2010; Scott and Glasspool, 2005). A more
170 detailed explanation is provided in the supplementary material. (LINK HERE to
171 [Supplementary material 1](#)) McParland *et al.* (2010) have in particular, demonstrated the
172 almost linear relationship of reflectance of specially prepared samples of charcoal across
173 a range of wood species, temperatures, and charring times (as opposed to earlier work
174 which concentrated on one or two species, and sometimes limited charring times).
175 Extensive testing of the reflectance method over a range of archaeological depositional
176 types however has not yet been carried out. In this study, charcoals of a range of size
177 fractions from cremation burials were evaluated, since the expected temperature for
178 successful human cremation is inferred to reach *ca.* 650°C as a minimum (if calcined
179 bone is observed (Wahl, 1982:27.)). One domestic pit fill and two hearth contexts were
180 also evaluated as a comparison.

181

182 **2 Background**

183 *2.1 Reflectance of modern charcoals*

184 Wood charcoal is formed through the heating of wood in the absence of oxygen, and can
185 be formed intentionally (in the manufacture of charcoal fuel), or as a by-product of wood-
186 fire burning. Soil examined from archaeological sites normally contains a mixture of
187 charcoal (incomplete burning of raw wood, or ‘re-burning’ of charcoal fuel) as well as
188 ash (the remains of wood burned to completion in the presence of oxygen).

189 Morphological, physical and chemical properties of charcoal can differ depending on two
190 groups of variables associated with the heating process. The first group consists of heat
191 related variables, which include temperature, time of exposure and heating rate (°C/min).
192 The second group consists of wood property variables, which include taxon, size, thermal
193 conductivity and other variables that can change during the charcoalification process
194 itself, see for example: Braadbaart *et al.* (2007), Braadbaart and Poole (2008), and Asouti
195 (2007).

196

197 The connection between increasing temperature formation and increasing mean random
198 reflectance value (studied in polished blocks under oil) is relatively well established
199 (Braadbart and Poole, 2008; McParland *et al.*, 2007; Scott and Glasspool, 2005). The

200 reflectance (%Ro) of charcoaled organic material provides information regarding the
201 absolute temperature to which the material in question has been exposed. This reflectance
202 is quantified and measured by comparison with the reflectance of known standards,
203 achieved through experimental work. The most comprehensive work on reflectance of
204 modern woods (McParland *et al.*, 2009a) measured temperatures of modern wood
205 samples burned under a variety of controlled circumstances. Mean reflectance
206 measurements carried out on different species (*Quercus*, *Corylus*, *Acer*, *Fraxinus*, *Betula*,
207 *Pinus*, *Erica*, *Calluna* and *Ulex*) corresponded closely to the original temperatures at
208 which the modern charcoal was produced. This choice of woods is particularly pertinent
209 to ancient fuel studies as these taxa are very commonly observed in archaeological
210 assemblages. This method discriminated changes in reflectance levels at intervals of 50
211 °C ranging from 300 °C to 1100 °C. It proved (as did earlier studies) that a near linear
212 relationship exists between charcoal burn temperature, and average reflectance
213 measurements, regardless of time of heat exposure and at a variety of temperatures
214 (McParland *et al.*, 2009a, 2010) (figure 1). Reviewing published calibration curves
215 shows that while the linear trend is always demonstrated, for any individual reflectance
216 measurement, agreement among the curves as to the associated temperature can vary by
217 as much as 100-180 °C (Section 5.1).

218

219

220 Fig 1 Calibration graph of reflectance vs burn temperature developed from charring of
221 modern oak wood under a variety of temperatures and times (modified from McParland
222 2010)

223

224 Experimental work on modern charcoals produced under laboratory conditions, provides
225 the backdrop for the present study, with the reflectance curve for *Quercus* developed by
226 McParland (2010) (also discussed in McParland *et al.* (2009a)), acting as the calibration
227 curve. If reflectance is to be of use in an archaeological setting, then we must exclude the
228 possibility that taphonomic processes undergone by archaeological charcoal will dull or
229 obliterate the reflectance signal of the charcoal remains and/or recognise when such
230 limitations may be present. If this concern can be allayed, then the technique may help

231 establish actual burn temperatures under different technological conditions (for example,
232 metal smelting, ceramics and glass production), and thus improvements in technology (as
233 represented by higher heat processes). It should also be possible to discriminate between
234 charcoal and raw fuel fires.

235

236 2.2 *Irish Bronze Age Cremation*

237 During the Bronze Age in Ireland, cremation was the predominant rite of treatment for
238 human remains (Lynch and O'Donnell, 2007, 105). The most likely form of the structure
239 may have been similar to that more visibly attested in later historic periods, for example
240 on coins from the Roman period (Toynbee, 1971, 32), and those pyres used even today
241 on the Ganges in the Hindu rite of cremation. Pyres are typically built by alternating and
242 increasingly smaller levels of logs built in a roughly pyramidal shape. The cremation
243 process would have been challenging in prehistory, depending as it does on time,
244 temperature and oxygen (Mc Kinley, 2000, 404).

245

246 The maximum temperature achievable in the combustion process will be affected by the
247 size, shape and quality (i.e. calorific potential) of the fuel; the structure of the pyre, (or
248 hearth or kiln); the body weight and fat content, ambient weather conditions, and the
249 supply of oxygen. A rough (but imperfect) proxy for calorific potential of a wood is its
250 specific gravity (a measure of relative density at a particular moisture content (Veal 2012,
251 33-34). In general, denser woods such as *Quercus* and *Fraxinus* produce a longer-lasting
252 source of heat than less dense woods such as *Salix*, *Populus* and *Alnus* (Gale 2003, 36).
253 In the study area *Quercus* appears to have been selected for cremation pyres during the
254 Bronze Age, possibly because of its high calorific potential (O'Donnell 2011; 2016).

255

256 Heat is not a fixed value for any one wood type since it will vary with moisture
257 conditions, size and shape of logs burnt, and other ambient factors in combustion (Lyons
258 *et al.*, 1985). We speak of calorific potential in any fire process, as the actual calorific
259 return achieved will depend on the amount of heat value transferred to the object of
260 combustion. In open pyre cremation, a large amount of heat is lost to the atmosphere.
261 After cremation has completed, Irish research has shown that a sub-sample of bone,

262 charcoal or a mixture of the two was taken from the funeral pyre and buried, within urns
263 in grave pits, or directly into grave pits (Lynch and O'Donnell, 2007). Charcoal from this
264 study is derived both from pyres and from cremation graves. Taphonomic processes
265 differ between the two contexts, for example in the pyres, wood would have burnt *in situ*,
266 suggesting that samples could be taken from the centre and outskirts of the pyre. This
267 may result in differing reflectance values, charcoal at the periphery of the pyre may have
268 burnt at lower temperatures than the centre. In any single cremation event, however,
269 bone and wood from a cremation pyre may have origins from anywhere in the fire, from
270 the centre, to the periphery and have been exposed to varying fire temperatures. As
271 burning progresses, fuel, both burnt and unburnt, can potentially move around in the fire
272 due to a range of agents, for example, differing temperature patches in the fire will be
273 present due to the varying flammability of materials present; oils or perfumes thrown
274 onto the fire will momentarily increase temperature, and once temperature reaches body
275 fat ignition point, burning will progress more quickly. (LINK TO Supplementary
276 material 2). Fatter body parts (and their nearby fuel) may be more likely to burn and drop
277 through the pyre first. Observation of remains from some urns suggests a range of sizes
278 of charcoal are present at deposition. Upon excavation from urns or pits, the origins of
279 the recovered charcoal (in terms of its position in the cremation pyre) cannot be
280 determined, nor may its time spent in the pyre be estimated (McKinley 2008, 167-68).
281 While not the case at this study site, sometimes bones are found arranged vertically in
282 anatomic order within urns, however even in these (rare) cases, it is unlikely the charcoal
283 is also so organized. This would be extremely difficult to carry out, but also, due to
284 charcoal's lightness and fragility, settlement will continue as the urn is being filled,
285 carried and potentially even after deposition. McKinley (2008) also notes that in her
286 experiments, roughly 700-900 kgs of wood are required and the main process of
287 combustion occurs in about 2-3 hours, with the pyre being left overnight to cool. Even
288 after 8 hours, some of the body may still remain unburnt.

289

290 2.3 *Archaeological background*

291 Templenoe, Co. Tipperary, Ireland is the largest Bronze Age flat cemetery excavated in
292 Ireland (figure 2) where the remains of 89 grave pits, 57 of which contained cremated

293 bone were excavated. Charcoal results from six graves and one potential pyre are
294 presented here (figure 3). Four possible pyre ventilation pits were also identified in
295 association with the burials. *In situ* burning was not evident, but they are classified as
296 potential pyres due to their location in the site and their larger dimensions than the grave
297 pits. Other domestic features, potentially unrelated to the cemetery include pits and
298 postholes, indicative of settlement activity. The cemetery was in use from the Early to the
299 Middle Bronze Age (dated by AMS radiocarbon determinations) (Mc Quade *et al* 2009,
300 130-133).

301

302 Fig 2 Map of location

303

304 Fig 3 Cremation pits and possible pyre pits, looking southwest. (Doody 2008, p. 117)

305

306 Charcoal was analysed mainly from the grave and potential pyre pits, of which seven
307 samples are included in the present study. Seven native Irish wood species were
308 identified, dominated by *Quercus* (deciduous only, there being no evergreen *Quercus* in
309 Ireland), and Maloideae. Nomenclature follows Stace (1997) . The other taxa present
310 were *Fraxinus*, *Corylus avellana*, *Ulmus*, *Prunus avium/padus* and *Alnus*.

311

312 Human skeletal remains were identified from 31 pit features from Templenoe. The
313 majority of the bone deposits within the cemetery contained less than 10g of cremated
314 bone, with an overall range of between only 0.08g and 697g, suggesting that token
315 deposits of bone only were buried, based on the average weight of a cremated adult
316 individual (Mc Kinley 1993). Four non-adults and 20 adults were present while one male
317 and one female were identified (Geber 2009, 213-215). Modern studies have shown that
318 temperatures ranging from 650 to $\geq 800^{\circ}\text{C}$ are required to successfully cremate human
319 bone (i.e. until the bones are whitish to white in colour) (Wahl 1982, 27). In
320 archaeological samples however, the reliance upon colour as an indicator of exposure
321 temperature essentially an imprecise criterion both because of individual differences in
322 the ability to perceive fine colour distinctions and because burnt bones may change
323 colour if they are buried (Shipman et al 1984). A variety of colours are often observable

324 in the remaining bones, and a complex interaction of many factors can influence colour,
325 and thus it cannot always be a reliable indicator of temperature (Devlin and Herrmann
326 2008). Some scholars have even noted that cremation may have occurred in a lower
327 temperature range in pre-history, from about 500-600 °C (Barber 1990; van Andringa et
328 al. 2013, Vol 1:8-9). Ignition starts from about 350 °C, with sufficient oxygen and
329 flammable material. In the process of cremation, temperatures of up to 1000°C can be
330 reached. Starter materials may have included brushwood, oils, perfumes, and of course,
331 textiles. These would all burn to completion and thus may be lost to the archaeological
332 record.

333

334 The people at Templenoe were good at cremating their dead. Taking various caveats into
335 account (Devlin and Hermann 2008), the grey-white colour of the majority of the
336 collected bone fragments in the samples indicates successful cremation. The bones
337 exhibit the fourth and fifth category of degree of burning, according to Wahl (1982, 21),
338 and Geber (2008), suggesting burning temperatures of 650-800 °C. This corresponds
339 most closely with burn colour codes 5 and 6 as described by Steiner and Kuhn (1995).
340 Therefore, the samples should provide a useful control with which to compare the
341 burning temperatures as measured through reflectance. As a further comparison,
342 reflectance values were also measured from charcoal from one domestic pit at Templenoe
343 and from charcoal from two nearby Bronze Age domestic hearths, at Lissava and
344 Ballylegan (figure 2) (Mc Quade *et al* 2009).

345

346 **3 METHODS**

347

348 *3.1 Processing and identification of charcoal*

349 Soil samples were processed by flotation (O'Donnell 2007, 28). Charcoal was identified
350 following known standards (Marguerie and Hunot 2007; Schweingruber, 1978) and a
351 modern reference collection.

352

353 *3.2 Subsampling of charcoal from available material*

354 Small (*c.* 2cm longitudinally) charcoal fragments, as well as fine charcoal dust were sub-
355 sampled and reflectance measurements taken as follows:

356

- 357 • Charcoal was examined to measure absolute burn temperatures achieved in
358 cremation pyres with successful (Reflectance Sample (RS) numbers 1, 3, 4, 5, 7,9,
359 19, 22 and 24), and less successful cremations, (RS 2, 6, 25, 26 and 27), as
360 denoted by bone colour.
- 361 • Charcoal from domestic fires was tested as a comparison with the cremation pyre
362 charcoal to examine differences in temperature ranges (RS 8, 11, 12).

363

364 Some differences are notable within the sub-soils from the three sites, although these are
365 not thought to have affected charcoal preservation. At Templenoe, the sub-soil was dark
366 boulder clay (Doody, 2008). At Lissava, the sub-soil was an orange-yellow gravelly,
367 sandy clays (Molloy, 2007). At Ballylegan, the sub-soil was compact, yellow brown
368 sandy clay (Mc Quade, 2007). Soil pH was not recorded at the sites. Modern
369 experimentation has shown that highly alkaline environments (such as may exist from
370 high concentrations of ash from combustion of wood) can weaken charcoal structure,
371 suggesting lower reflectance values may be observed (Braadbaart *et al.*, 2009). Ash not
372 noted from Templenoe, and the context descriptions of the cremation deposits are very
373 cohesive, indicating they were filled with loose, black silty clay (Doody, 2008). Even if
374 high alkalinity were present at burial it is difficult to know the rate/range of pH changes
375 that may have occurred due to percolation of rainwater, and/or groundwater over time.

376

377 3.3 *Reflectance testing in two stages*

378 Taxonomically identified charcoals >2mm identified were roughly crushed and mounted
379 in one of two methods (cold set, or hot set epoxy), highly polished, and inspected at
380 x1,000 magnification using a reflecting Nikon microphot microscope. Fifty
381 measurements were taken and averaged for each individual sample. The hot and cold set
382 epoxy methods have different utility depending on sample size and other factors, but a
383 control test revealed results were not affected by setting method. The samples here were
384 prepared both by cold set (>2mm samples), and hot set (250µm-2mm samples). In the hot

385 set method, careful selection of the ratio of charcoal dust to epoxy powder is needed. Too
386 much epoxy powder results in insufficient charcoal at the sample interface; too little
387 epoxy results in a gritty sample that is difficult to polish. A ratio of about 1/3 charcoal
388 dust to 2/3 epoxy mix was found to be suitable for these samples. (Further method details
389 can be found in Supplementary materials 1)

390

391 Testing of samples of individual and mixed taxa was carried out. In a second round of
392 measurements, smaller charcoals of various fraction sizes were tested (from the same
393 contexts). Table 1 details the characteristics of the charcoals tested and the reflectance
394 results. Sample numbers are not contiguous as some failed to be 'readable' for reflectance
395 after preparation due to the challenges of sample preparation, in particular, the need to
396 provide at the end of the process good exposure of the charcoal to the reflecting laser, i.e.
397 at the very uppermost surface of the resin.

398

399

400 3.4 Reporting reflectance results: average and maximum temperatures

401 Standard procedure in past reflectance testing has reported ranges of average
402 temperatures (calculated from 50-100 measurements of each individual sample). Here
403 we follow convention, but also consider more critically, the maximum temperatures
404 observed for each sample. Temperatures were inferred using a 1, 6, 12 and 24 hour
405 calibration curve developed by MacParland for *Quercus* (figure 1). Results are expressed
406 as a range of temperature e.g. a sample with a reflectance value of 5%Ro would need to
407 be charred at 880 °C for one hour or 800 °C for 24 hours giving a range of 800-880 °C.
408 The *Quercus* (deciduous) curve, was adopted as this was the most common wood
409 observed within the Irish archaeological samples. No experimentation has been carried
410 out on evergreen oak (a factor of relevance for Mediterranean data) and a wood that is
411 usually harder and of higher specific gravity.

412

413 Table 1: Context types, characteristics, and reflectance results of tested charcoals (using
414 McParland reflectance curve in fig. 1)

415

416

417 4 RESULTS

418

419 4.1 Summary

420 In the first round of readings from >2mm sized charcoals, the average temperature varied
421 from a low of 360-410°C (Samples 7 and 10) to a high of 390-450°C (Sample 12) (table
422 1). The highest temperature observed was 525°C (Sample 3). Little or no variation could
423 be correlated with wood species. As these readings were well below those expected (650-
424 800°C), a decision was made to seek smaller fraction charcoals (by way of dry sieving
425 the extant archaeological material). McParland *et al.* (2009a and b), Braadbart and Poole
426 (2008) and others suggest that due to the increasingly brittle nature of charcoals with
427 increased charring temperature, higher reflecting charcoals may be limited to the smallest
428 size fraction. Careful sieving was made of the sub 2mm charcoals into 2-1mm, 1mm-500
429 µm, and 500-250 µm fractions (however, subsequently these sub-divisions provided no
430 extra information). As the flotation material kept was processed over a 250 µm mesh, it
431 was possible to examine material as small as this, but no smaller, from the same contexts
432 from which the identified charcoals arose. It cannot be proven that the charcoals in the
433 range 250 µm - 2mm are the same wood types, but it is highly probable (in any event,
434 wood type was determined to have little bearing on the process). It should also be borne
435 in mind that the size of a particular charcoal fragment may not only be due to fire
436 process, but also to depositional and taphonomic phenomena, and excavation and post-
437 excavation handling. All result in further fragmentation of archaeological charcoal.

438

439 In the second round of readings from the >250 µm – 2mm fractions, average
440 temperatures ranged from a low of 360-410°C (Sample 26, correlating with Samples 2
441 and 27) to an average temperature high of 390-450°C (Sample 19, correlating with
442 Sample 9). The highest temperature recorded was 515°C (Sample 27, correlating with
443 Samples 2 and 26).

444

445 Thus the lowest and highest average temperatures are the same from the >2mm, and
446 >250µm – 2mm fractions (360-410°C/390-450°C). In some cases, the >250µm – 2mm

447 fraction actually provided a lower average temperature than the >2mm fraction (for
448 example Sample 26, correlating with Sample 2). A slight temperature increase can be
449 noted however, within the average temperatures of the smaller fraction samples than the
450 larger ones as shown in **table 1**. Observation showed that the difference between
451 ‘successful’ and ‘less successful’ cremations as determined by bone colour could not be
452 explained by a difference in the calorific potential of the woods.

453

454 One of the questions considered was whether changes in temperature would be noted
455 from different archaeological contexts. Would a cremation pyre of mixed wood taxa (a
456 specialised construction with increased body fats) burn at a higher temperature than
457 domestic fires of mixed wood taxa? To test this, controls were taken measuring
458 reflectance values from Templenoe (Grave pit F179, Domestic pit F33), Lissava (Trough
459 pit F12) and and Ballylegan (Hearth F446). Average reflectance results from the mixed
460 taxa group of the grave pit from Templenoe (Sample 5) were 370-415°C, and these do
461 not differ to those obtained from the domestic hearth at Ballylegan (Sample 11). Results
462 from the domestic pit at Templenoe F33 (Sample 8) are slightly higher, at 375-425°C.
463 The highest temperature from this comparative group is 390-450°C (Sample 12) from the
464 trough fill at Lissava.

465

466 Table 2 summarises the maximum temperatures observed for each deposit type,
467 comparing contexts containing bone, with those that did not, or were the controls of
468 hearth and trough. Inspecting this table shows that generally the contexts with bone have
469 slightly higher temperatures, the maximum reading for contexts with bone was 525°C,
470 and for those of the controls/ no bone: 490°C, but the difference is marginal. More data,
471 and/or employing a reflectance reading strategy that uses Rmax proper (i.e. with the
472 transverse section aligned perpendicularly for the ‘best’ possible reading), as opposed to
473 %Ro (average random readings, on unaligned sections) may provide higher temperature
474 readings.

475

476 **Table 2: Comparison of maximum temperatures observed in contexts with bone, and**
477 **without bone**

478

479

480 **5 DISCUSSION**

481

482 5.1 *Reconsidering calibration curves*

483 The temperatures measured using reflectance on the cremation deposits are lower than
484 expected. This may be linked to the current available calibration curves.

485 A number of studies have published calibration graphs of the temperature / random
486 reflectance relationship with a positive correlation between increasing temperature of
487 formation and increasing mean random reflectance (%Ro) (Ascough *et al.*, 2010;
488 Braadbaart and Poole, 2008; Bustin and Guo, 1999; Guo and Bustin, 1998; Jones *et al.*,
489 .1991; McParland *et al.* 2007; Scott and Jones 1991; Scott and Glasspool 2005).

490 The curves are in broad agreement, but they were each constructed using different woods
491 and under a variety of conditions. Woods studied included: (conifers) *Sequoia*
492 *sempervirens*, *Pseudotsuga menziesii*, *Picea abies* and *Pinus sylvestris*; (deciduous
493 broadleaves); *Quercus robur*, *Fagus sylvatica*, *Corylus avellana* and *Alnus glutinosa*.
494 The two curves developed by Bustin and Guo (1999) and Guo and Bustin (1998) aimed
495 to measure Rmax (i.e. by carefully orienting the samples to provide a transverse surface
496 exactly perpendicular to the incident light), while the rest of the studies examine %Ro
497 (average random reflectance – without special orientation of the sample).

498

499 McParland (2010, table 10.1) provides a detailed table comparing these studies, however,
500 among all of these studies, it is only her own that attempts measurements across five
501 taxa, while all the others create curves using just one taxon. Comparison of the studies is
502 further complicated by the fact that different researchers formalise calibration
503 measurement points at different temperatures. **Table 3** summarises the maximum
504 variation observed in %Ro where four or more comparanda readings are available, except
505 for those measurements from 700-1100°C, where the low reading is always from
506 McParland (2010), while the higher reading is from Braadbaart and Poole (2008). **Figure**
507 **4** illustrates the comparison of these latter two curves, and exemplifies how at one %Ro,
508 quite a range of temperatures are theoretically possible. These are the only two
509 calibration curves which to date examine oak. Differences in the reported mean random
510 reflectance at a given formation temperature and duration may be accounted for by

511 variable factors in different experiment designs, such as length of seasoning time of the
512 calibration wood, dimensions of the sample material charred, or differences in charring
513 protocol (for example, charring under nitrogen atmosphere, which results in complete
514 exclusion of oxygen; or charring by wrapping in aluminium foil, a ‘low’ oxygen method).
515 It may be that differences in polishing level, and/or calibration of differing laboratory
516 reflectance systems may also account for these variations, as well as the fact that all
517 measure %Ro (i.e. the average of the maxima and minima observed – without perfect
518 alignment of the transverse plane). In figure 4, Braadbart and Poole’s 2008 curve, the
519 reflectance values as observed in this experiment result in temperature readings that can
520 be as much as 150° C higher for the same reflectance value. McParland *et al.*’s (2010)
521 preparatory methods however more closely mimic actual ancient fire conditions: low
522 oxygen charring, over a variety of time periods, and, importantly, across a range of taxa
523 most often found in European charcoal assemblages, thus this curve has been used in
524 calibrating the reflectance results for this study. Calibration curves built on, for example,
525 sequoia, or other conifers (summarised well in Scott *et al.* 2014:section 2.4) may have
526 little relevance to European conditions (although sequoia of course may be useful in
527 American studies). A further new curve is found in Hudspith (2015:3) who also measures
528 reflectance of multiple woods types: *Betula nana*, *Picea mariana*, *Picea glauca*, *Betula*
529 *papyrifera* and *Populus tremuloides* (figure 5). These taxa were chosen for their
530 relevance in the modern wildfire study to which she applies her curve.–The resinous
531 nature of conifers (as opposed to hardwoods) may affect calibration due to the possible
532 recondensation of resins during charcoalification, however no recognisable pattern of
533 reflectance calibration curves could be discerned that distinguished conifers from
534 broadleafed species, although wood specific gravity did seem to have some influence
535 (higher SGs give higher reflectance at a given temperature).

536

537 Table 3: Variation in reflectance measurements from six different studies (from
538 McParland 2010, table 10.1). Studies of: Ascough *et al.*, 2010; Braadbaart and Poole,
539 2008; Bustin and Guo, 1999; Guo and Bustin, 1998; Jones *et al.*, .1991; McParland *et al.*
540 2007; Scott and Jones 1991; Scott and Glasspool 2005. Table shows %Ro where four or
541 more comparanda readings were available, except for those measurements from 700-

542 1100°C, where the low reading is always from McParland (2010), while the higher
543 reading is from Braadbart and Poole (2008).

544

545 As noted, Braadbart and Poole (2008) is the only other study which derives a curve from
546 *Quercus*, but they used very small cubes of wood which were charred only for an hour
547 (conditions unlikely to mimic the long process of cremation), however, their remarks on
548 this curve bear some examination. Firstly, they note that a higher rate of increase of
549 reflectance (and therefore temperature) lies in the range 600-850° C (so their curve is not
550 completely 'flat'). They also note a tailing off of reflectance values after 850° C, and at
551 this temperature we may expect (as they note) that if full combustion has occurred, then
552 in all likelihood there will be very little carbonised material to observe in the imperfect
553 and variable conditions of reality (as opposed to laboratory) burning. Testing of small
554 microcharcoals from inside iron slag (Veal *et al.*, in preparation), has had some success
555 however, with reflectance values translating into temperatures over 1000°C. The fact
556 that the 'best' calibration curve may not in fact be linear, is also demonstrated by the
557 most recently published work by Hudspith *et al.* (2015, Fig. 5) whose fitted curve is S
558 shaped., but whose raw data points closely follow the patterns of McParland (2010) and
559 Braadbaart & Poole (2008)

560

561 Figure 4 : Braadbart and Poole's (2008) calibration curve and that of McParland (2010).
562 Both derived from *Quercus*, although with very different preparation strategies.

563

564 Figure 5 : Hudspith *et al.*'s (2015) calibration curve for five experimentally charred
565 (American) boreal woods. Mean random reflectance under oil (Roman) and standard
566 deviations represent all species. Derived from charring small pieces of wood for 1 hour
567 each at varying temperatures.

568

569

570

571 According to the calibration curve we believe to be most appropriate to the burning
572 conditions of cremation (Mc Parland , 2010), the results (average temperature high of

573 390-450°C) show that charcoals collected above 250 µm from known pyre contexts have
574 not demonstrated the temperatures associated with cremation (above 650°C as previously
575 discussed). A range of variable results were obtained, with, in some cases a small upward
576 trend for the smaller fraction charcoals. Reading from alternatives curves, such as that of
577 Braadbaart and Poole (2008), however, the temperatures expected may be inferred in a
578 range around and above 650° C. This demonstrates how the calibration curve used can
579 greatly influence results and calls for greater agreement in calibration curves.

580

581 A number of explanations may be considered:

- 582 • Is the reflectance method capable of measuring temperatures in archaeological
583 charcoals as high as 650-800°C? Experimental work creating charcoal under
584 laboratory conditions indicates that temperatures up to 1,100°C can be measured
585 through reflectance (Mc Parland *et al.*, 2009a, 253). When applied to archaeological
586 materials, however, average measured temperature values reported to date are 330-
587 410°C (Mc Parland *et al.*, 2009b, 182) and 375-530°C (Mc Parland *et al.*, 2009a, 6).
588 Further work must be conducted on measuring reflectance values from archaeological
589 charcoal to demonstrate the maximum temperatures the method can record.
- 590 • The collected remains may reflect fuel waste located at the periphery of the fire, (i.e.
591 the cooler part of it), and/or which had been thrown on late in the cremation (and thus
592 not exposed to the high central pyre temperature.) The cremated remains, ash and
593 charcoal are however, culturally sorted before deposition in an urn or grave, which
594 ultimately contains selected bones and some charcoal, a process which will have
595 likely mixed the remains, so selective collection by the archaeologist of lower
596 temperature (only) exposed charcoals seems improbable.
- 597 • Higher temperature charcoals tend to be more brittle and vulnerable to breakage, as
598 noted, and therefore there is likely to be a bias towards lower temperature materials in
599 any identified charcoal assemblage (the smaller the material size, the more probable it
600 will be lost in surrounding soils, and/or not collected).
- 601 • The collected remains may have reached the temperature of cremation, but the
602 taphonomic processes over *c.* 3500 years (such as percolation or mineralisation), have
603 altered the reflectance of the charcoal. Fresh breaks are revealed before reflectance

604 measurements are taken making this possibility less probable in the case of external
605 mineralisation, however, highly acid or alkaline waterlogged sites (not indicated
606 here), may be problematic. Generally speaking, archaeological charcoals absorb
607 chemicals in soils around them, and in particular water, diagenesis of charcoal does
608 occur over long time periods, and more testing of this phenomena through chemical
609 and physical proxies, as well as reflectance, may assist.

610

611

612 • While the smaller, (i.e. 250 μ m) charcoals mostly did reveal a slight increase, the
613 change in average temperature was not statistically significant. Higher temperatures
614 may be measureable from even smaller material, therefore this needs collection by
615 way of un-sieved ash/charcoal/soil samples, although if combustion is virtually
616 complete, no remains may be archaeologically detectable, as also noted by Braadbaart
617 and Poole (2008) and others. This will depend on the origins of the charcoal, and the
618 manner of collection

619

620 **6 CONCLUSIONS AND FURTHER WORK**

621 The results suggest that reflectance will need to be applied in a selective manner and
622 results interpreted carefully in terms of cultural practice, combustion processes and
623 taphonomy. Reflectance results outlined here (according to our chosen calibration curve)
624 do not match those expected from cremation deposits, however, they reach the minimum
625 temperature expected using the curve of Braadbaart and Poole (2008). We have not used
626 this curve because its development involved preparation methods less close to those of
627 prehistoric cremation, as already noted.

628

629 The calibration curve used here, and others published elsewhere need to be further
630 refined so we can obtain well-tested curves for major archaeological fuel woods
631 (predominantly hardwoods). The calibration line of best fit may well be a curve, rather
632 than a straight line relationship. The apparent steeper rise in reflectance from about 550-
633 600 °C , needs to be investigated, and we suggest may relate to a particular state of
634 rearrangement of the carbon atoms and aromatic compounds within the charcoal.

635

636 Exchange of samples for testing in different laboratories will assist, together with
637 checking of standards used. To this end it is suggested that a group of researchers be
638 formed to test and agree a way forward.

639

640 Ultimately the science of reflectance may require nuanced calibration of curves for
641 different taxonomic groups, and perhaps according to anatomical structure to see if
642 resolution of estimated temperature may be improved beyond the 100-150°C range
643 currently observed. Testing of other charred materials (such as seeds) is underway as
644 olive and grape pressings are also found as fuels, see for example, Braadbaart *et al.*
645 (2016), Rowan (forthcoming), Coubray *et al.* (forthcoming). The diagenesis of charcoal
646 increases over time, and this needs further examination, perhaps by way of examining
647 other chemical and physical properties of differently aged archaeological charcoal.

648

649 Combustion, once temperatures reach the high levels expected, may be wholly
650 destructive under some fire regimes, and it may not be possible to collect any remnant
651 charcoals that have been exposed to the high temperatures, although testing of reflectance
652 on charcoals from an iron smelting site in Medieval Angkor (South East Asia) has shown
653 more encouraging results, with some results ca. 1000°C (Veal *et al* in preparation).
654 Testing of small materials would be desirable. Normal field collection of charcoal is by
655 dry sieving over 4 or 5mm mesh, and/or flotation over 250µm or larger mesh. The
656 absolute size of charcoal fragments for taxonomic identification purposes is usually
657 >2mm. It will be useful in the future to collect un-sieved soil samples of ash and micro-
658 charcoals (where present), and separate the charcoal by gravimetric or other method for
659 reflectance testing to further resolve this issue. Small bulk samples collected by hand (we
660 suggest approximately 500g would be adequate. These should be stored in a cool place,
661 and not subjected to changing ranges of temperature or moisture after excavation. As this
662 strategy also accords with that required for a range of archaeometric studies, its burden to
663 the archaeologist would be minimal.

664

665 The method does have utility even at this level of refinement as it appears to differentiate
666 high and low level temperature processes. Further work using the reflectance method in

667 other archaeological context types is also required, for instance, the testing of ‘charcoal
668 only’ fires, vs ‘raw wood’ fires, should provide contrasting temperature profiles.

669

670 **ACKNOWLEDGEMENTS**

671

672 The authors acknowledge the support of the CSIRO, Sydney; the Department of
673 Archaeology, and the Australian Centre for Microscopy and Microanalysis, University of
674 Sydney (especially materials preparation specialist, Mr Adam Sikorski); and the
675 McDonald Institute for Archaeological Research, University of Cambridge. Excavations
676 were carried out by Margaret Gowen & Co. Ltd, funded by the National Roads Authority,
677 Ireland. This research was partly funded by the Government of Ireland, Irish Research
678 Council (Project id GOIPD/2013/387) supported by the School of Archaeology,
679 University College Dublin. Charcoal results are stored in the WODAN database at
680 www.wodan.ie.

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684 **References**

685

686 Ascough, P.L., Bird, M.I., Scott, A.C., Collinson, M.E., Cohen-Ofri, I., Snape, C.E. and
687 Le Manquais, K., 2010. Charcoal reflectance measurements: implications for
688 structural characterization and assessment of diagenetic alteration. *Journal of*
689 *Archaeological Science* **37**, 1590- 1599.

690

691 Asouti, E, Austin, P., 2005. Reconstructing Woodland Vegetation and its Exploitation by
692 Past Societies, based on the Analysis and Interpretation of Archaeological Wood
693 Charcoal Macro-Remains. *Environmental Archaeology* **10**, 1-18.

694 Barber, B. 1990. Cremation. *Journal of Indo-European Studies* **18** (3-4), 379-388.

695 Braadbaart,, F.and Poole, I., 2008. Morphological, chemical and physical changes during
696 charcoalfication of wood and its relevance to archaeological contexts. *Journal of*
697 *Archaeological Science* **35**, 2434-2445.

698

699 Asouti E. , 2007. *Charcoal Analysis Web*. University of Liverpool. Available from
700 <http://pcwww.liv.ac.uk/~easouti/>.

701

702 Barber, B. 1990. Cremation. *Journal of Indo-European Studies* no. 18 (3-4):379-388.

703

704 Belcher, C. and Hudspith, V.A. (2016). The formation of charcoal reflectance and its
705 potential use in post-fire assessments. *International Journal of Wildland Fire* (in press)
706 <http://dx.doi.org/10.1071/WF15185>.

707

708 Braadbaart, F., Marinova, E., and Sarpaki, A., 2016. Charred olive stones: experimental
709 and archaeological evidence for recognizing olive processing residues used as fuel.
710 *Vegetation History and Archaeobotany* (in press), DOI: 10.1007/s00334-016-0562-2.

711

712 Braadbaart, F., Poole, I., 2008. Morphological, Chemical and Physical Changes During
713 Charoalification of Wood and its Relevance to Archaeological Contexts, *Journal of*
714 *Archaeological Science* **35**, 2434-2445.

715

716 Braadbaart, F., Poole, I., and van Brussel, A.A., 2009. Preservation potential of charcoal
717 in alkaline environments: an experimental approach and implications for the
718 archaeological record. *Journal of Archaeological Science* **36**, 1672-1679.
719

720 Braadbaart, F., Poole, I., Huisman, H.D.J. and van Os, B., 2012. Fuel, Fire and Heat: an
721 experimental approach to highlight the potential of studying ash and char remains from
722 archaeological contexts. *Journal of Archaeological Science* **39**, 836-847.
723

724 Braadbaart, F., Wright, P. J., Vander Horst, J. and Boon, J. J., 2007. A laboratory
725 simulation of the carbonization of sunflower achenes and seeds. *Journal of Analytical
726 and Applied Pyrolysis* **78**, 316-327.
727

728 Bustin RM. and Guo Y., 1999. Abrupt changes (jumps) in reflectance
729 values and chemical compositions of artificial charcoals and
730 inertinite in coals. *Int J Coal Geol* **38**:237–260.
731

732 Chabal, L., 1992 La représentativité paléo-écologique des charbons de bois archéologiques
733 issus du bois de feu. *Bulletin de la Société Botanique de France* **139**, 213-236.
734 Chabal, L., Fabre, L., Terral, J.-F., Théry, I., 1999. L'anthracologie. In Bourquin-
735 Mignot, C., Brochier, J.-E., Chabal, L., Crozat, S., Fabre, L., Guibal, F., Marnival, P.,
736 Richard, H., Terral, J.-F. & Théry, I. (Eds.) *La Botanique*. Paris: France.
737

738 Chrzavzez, J., Théry-Parisot, I., Terral, J.-F., Ducom, A. and Fiorucci, G., 2011.
739 Differential preservation of anthracological material and mechanical properties of wood
740 charcoal, an experimental approach of fragmentation, in: Badal, E., Carrión, Y., Grau, E.,
741 Macías, M., Ntinou, M. (Eds.), *5th International Meeting of Charcoal Analysis. The
742 charcoal as cultural and biological heritage*. Saguntum, Papeles del Laboratorio de
743 Arqueología de València, Department de Prehistòria i Arqueologia, València, 29-30.
744

745 Chrzavzez, J., Théry-Parisot, I. Fiorucci, G., Terral, J.F., and Thibaut, B., 2014. I'impact
746 of post-depositional processes on charcoal fragmentation and archaeobotanical

747 implications: experimental approach combining charcoal analysis and biomechanics.’
748 *Journal of Archaeological Science* **44**, 30-42.
749
750 Coubray, S., V. Zech-Matterne, V. and Monteix N., forthcoming. ‘Of Olives and Wood.
751 Baking Bread in Pompeii.’ In *Fuel and Fire in the Ancient Roman World: towards an*
752 *integrated economic understanding*, edited by R. Veal and V. Leitch. Cambridge:
753 McDonald Institute for Archaeological Research Monographs.
754
755 Devlin, J.B. and Herrmann, N.P. 2008. Bone colour as an interpretive tool of the
756 depositional history of archaeological cremains. In *The Analysis of Burned Human*
757 *Remains*, edited by C.W. Schmidt and S.A. Symes (eds.) London: Elsevier Academic
758 Press, 109-128.
759
760 Dufraisse, A., 2006. *Charcoal Analysis: New Analytical Tools and Methods for*
761 *Archaeology: Papers from the Table-Rhonde held in Basel 2004*. BAR International
762 Series 1483. Oxford: Archaeopress.
763
764 Doody, M., 2008. Final report, Templenoe, Co. Tipperary. N8 Cashel to Mitchelstown
765 road scheme. Ministerial Direction Scheme Reference number A035/000. Registration
766 number E2290. Unpublished report for Margaret Gowen & Co. Ltd.
767
768 Gale, R. 2003. Wood-based industrial fuels and their environmental impact. In Murphy,
769 P. and Wiltshire, P.E.J. (Eds.) *The Environmental Archaeology of Industry. Symposia of*
770 *the Association of Environmental Archaeology No. 20*. Oxford: Oxbow books.
771
772
773 Geber, J., 2008. The cremation burials from Templenoe (E2290). Unpublished report for
774 Margaret Gowen & Co. Ltd.
775

776 Geber, J. 2009. Chapter 7. The human remains. In M. McQuade, B. Molloy and C.
777 Moriarty (eds.) *In the shadow of the Galtees. Archaeological excavations along the N8*
778 *Cashel to Mitchelstown*. Dublin: The National Roads Authority, 209-240.
779

780 Guo, Y, and Bustin, R.M. 1998. FTIR spectroscopy and reflectance of modern charcoals
781 and fungal decayed woods: implications for studies of inertinite in coals. *Int J Coal Geol*
782 **37**:29–53.
783

784 Harrison, K ., 2013. The application of forensic fire investigation techniques in the
785 archaeological
786 record. *Journal of Archaeological Science* 40:955-59.
787

788 Hudspith A., Belcher, C., Kelly R., and Hu, F.S., 2015. Charcoal Reflectance Reveals
789 Early Holocene Boreal Deciduous Forests Burned at High Intensities. *PloS One*
790 DOI:10.1371/journal.pone.0120835.
791

792 Jones, T.P, Scott A.C. and Cope, M. 1991. Reflectance measurements and the temperature
793 of formation of modern charcoals and implications for studies of fusain. *Bull Soc Geol*
794 *France* **162**, 193–200.
795

796 Lancelotti, C., Madella, M., P., A. and Petrie, C.A., 2010. Temperature, compression and
797 fragmentation: an experimental analysis to assess the impact of taphonomic processes on
798 charcoal preservation. *Archaeological and Anthropological Sciences* **2**, 307-320.
799

800 Ludemann, T., 2006. Anthracological Analysis of Recent Charcoal-Burning in the Black
801 Forest, SW Germany, in: Dufraisse, A. (Ed.), *Charcoal Analysis: New Analytical Tools*
802 *and Methods for Archaeology*, Archaeopress, Oxford, pp. 61-70.
803

804 Lynch, L. and O'Donnell, L., 2007. Cremation in the Bronze Age: Practice, process and
805 belief. In Grogan, E., O'Donnell, L. and Johnston, P. (Eds.) *The Bronze Age Landscapes*
806 *of the Pipeline to the West: An integrated archaeological and environmental assessment*.
807 Bray: Wordwell.
808

809 Lyons, G., Lunny, F. and Pollock, H.P., 1985. A Procedure for Estimating the Value of
810 Forest Fuels. *Biomass*, **8**, 283-300.
811

812 Marguerie, D. and Hunot, J.Y. 2007. Charcoal analysis and dendrology: data from
813 archaeological sites in north-western France. *Journal of Archaeological Science* **34**,
814 1417-1433.
815

816 McKinley, J. , 1993. Bone fragment size and weight of bone from modern British
817 cremations and the implications for the interpretation of archaeological cremations.
818 *International Journal of Osteoarchaeology*, **3**, 287-283.
819

820 McKinley, J., 2000. The analysis of cremated bone. In Cox, M. and Mays, S. (Eds.)
821 *Human osteology in archaeology and forensic science*. London: Greenwich Medical
822 Media Ltd, 403-421.
823

824 Mc Kinley, J. 2008. In the heat of the pyre: efficiency of oxidation in Romano-British
825 cremations - did it really matter? In C.W. Schmidt and S.A. Symes (eds.) *The Analysis of*
826 *Burned Human Remains*. London: Elsevier Academic Press, 163-184.
827

828 McParland, L., Collinson, M. E., Scott, A. C., Steart, D. C., Grassineau, N. V. and
829 Gibbons, S.. 2007. Fern and fires: experimental charring of ferns compared to wood and
830 implications for paleobiology, paleoecology, coal petrology and isotope geochemistry
831 *PALAIOS* **22**, 528-538.
832

833 McParland, L., Collinson, E., Scott, A. and Campbell, G., 2009a. The use of reflectance
834 values for the interpretation of natural and anthropogenic charcoal assemblages.
835 *Archaeological Anthropological Science* **1**, 249-261.
836

837 McParland, L., Hazell, Z., Campbell, G., Collinson, M. E. and Scott, A. C., 2009b. How
838 the Romans got themselves into hot water: temperatures and fuel types used in firing a
839 hypocaust. *Environmental archaeology* **14**, 176-183.

840

841 McParland, L.C., 2010. Utilisation of quantified reflectance values to determine
842 temperature and processes of formation for human produced charcoals . Ph.D. Thesis
843 (unpublished). Royal Holloway, University of London.

844

845

846 McParland, L., Collinson, M., Scott, A., Campbell, G. and Veal, R. 2010. Is vitrification
847 in charcoal a result of high temperature burning of wood? *Journal of Archaeological*
848 *Science* **37** 1-9.

849

850 McQuade, M., 2007. Final report, Ballylegan, Co. Tipperary. N8 Cashel to Mitchelstown
851 road scheme. Ministerial Direction Scheme Reference number A035/000. Registration
852 number E2265. Unpublished report for Margaret Gowen & Co. Ltd.

853

854 McQuade, M., Molloy, B., Moriarty, C., 2009. *In the shadow of the Galtees*
855 *Archaeological excavations along the N8 Cashel to Mitchelstown*. Dublin: The National
856 Roads Authority.

857

858 Molloy, B., 2007. Final report, Lissava, Co. Tipperary. N8 Cashel to Mitchelstown road
859 scheme. Ministerial Direction Scheme Reference number A035/000. Registration number
860 E2296. Unpublished report for Margaret Gowen & Co. Ltd.

861

862

863 O'Donnell, L., 2007. Charcoal and wood. In: Grogan, E., O' Donnell, L. and Johnston, P.
864 (Eds.), *The Bronze Age Landscapes of the Pipeline to the West. An integrated*
865 *archaeological and environmental assessment*. Bray: Wordwell.

866

867 O'Donnell, L. 2011. People and woodlands, an investigation of charcoal remains as
868 indicators of cultural remains and environmental indicators in Bronze Age Ireland. PhD
869 thesis, University College Dublin.

870

871 O'Donnell, L. 2016. The power of the pyre – a holistic study of cremation focusing on
872 charcoal remains. *Journal of Archaeological Science* 65, 161-171.
873

874 Rowan, E. "The Energy Potential of Pomace Fuel in the Roman World." In *Fuel and Fire*
875 *in the Ancient Roman World: Toward an Integrated Economic Understanding*, edited by
876 R. Veal and V. Leitch. Cambridge: McDonald Institute for Archaeological Research
877 Monographs, forthcoming.
878

879 Schweingruber, F.H., 1978. *Microscopic wood anatomy*. Birmensdorf: Swiss Federal
880 Institute for Forest, Snow and Landscape Research.
881

882 Scott A.C. and Jones T.P. 1991. Microscopical observations of recent and
883 fossil charcoal. *Microsc anal* 24:13–15.
884

885 Scott A.C. and Glasspool I.J. 2005 Charcoal reflectance as a proxy for the
886 emplacement temperature of pyroclastic flow deposits. *Geology*. 33: 589-592.
887

888 Scott, A.C., Bowman, D.M.J.S., Bond, W.J., Pyne, S.J. and Alexander, M.E., 2014.
889 *Fire on Earth: an Introduction*. John Wiley & Sons Ltd:Chichester.
890

891 Shipman, P., Foster, G. and Schoeninger, M. 1984. Burnt bones and teeth: an experimental study
892 of color, morphology, crystal structure and shrinkage. *Journal of Archaeological Science* 11,
893 Issue 4, 307-325.
894

895 Stace, C. 1997. *New Flora of the British Isles*. Second edition. Bath: The Bath Press.
896

897 Steiner, M.C. and Kuhn, S.L. 1995. Differential Burning, Recrystallization, and Fragmentation of
898 Archaeological Bone. *Journal of Archaeological Science* 22, 223–237
899

900 Théry-Parisot, I., Chabal, L. and Chrzavzez, J., 2010. Anthracology and taphonomy,
901 from wood gathering to charcoal analysis. A review of the taphonomic processes

902 modifying charcoal assemblages, in archaeological contexts. *Palaeogeography,*
903 *Palaeoclimatology, Palaeoecology* **291**, 142-153.

904

905 Toynbee, J..M.C., 1971. *Death and burial in the Roman world*. Cornell: Cornell
906 University Press.

907

908 Van Andringa, W.H, Duday, S., Lepetz, S., Joly, D. and Lind, T. 2013. *Mourir à*
909 *Pompéi. Fouille d'un quartier funéraire de la nécropole romaine de Porta Nocera (2003-*
910 *2007)*. Rome: Collection de l'Ecole française de Rome, 2 vol.

911

912 Veal R., 2012. From Context to Economy: charcoal and its unique potential in
913 archaeological interpretation: a case study from Pompeii, I.E. Schrüfer-Kolb (ed.), More
914 than just numbers? The role of science in Roman archaeology, Vol.91 (*Journal of Roman*
915 *Archaeology Supplement*.) Portsmouth: *Journal of Roman Archaeology*. 19-52.

916

917 Veal, R., L. McParland, L. and Hendrickson, M. (forthcoming). Testing the Temperature
918 of Archaeological Charcoals Using the Reflectance Method: A Proxy for Estimating
919 Operating Temperatures and Fuel Type of Medieval Iron Smelting Furnaces at Preah
920 Khan of Kompong Svay, Cambodia.

921

922 Wahl, J., 1982. Leichenbranduntersuchungen. Ein Überblick über die Bearbeitungs—
923 und Aussagemöglichkeiten von Brandgräbern. *Prähistorische Zeitschrift* **57**, 2-125.

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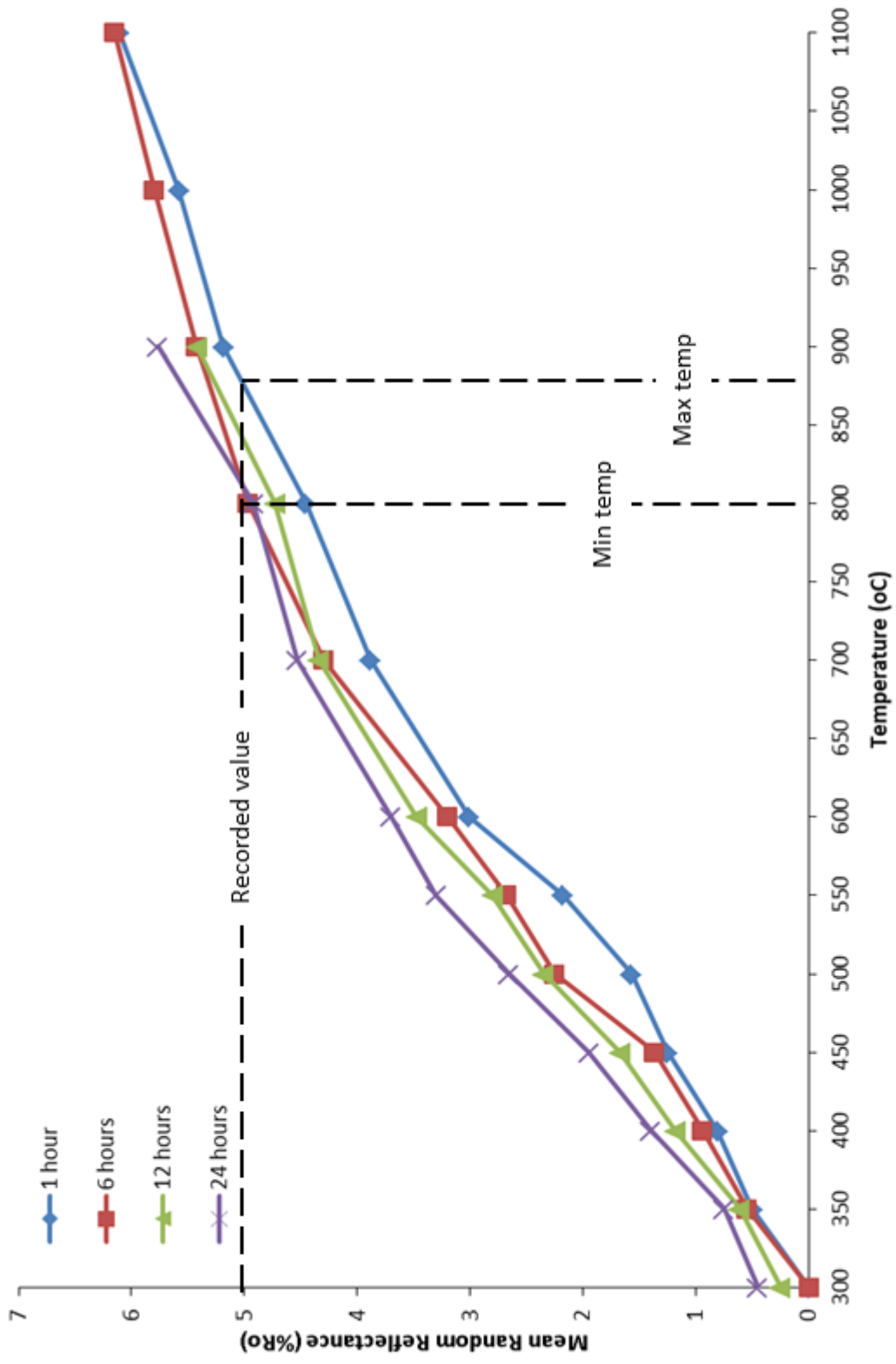
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Templenoe

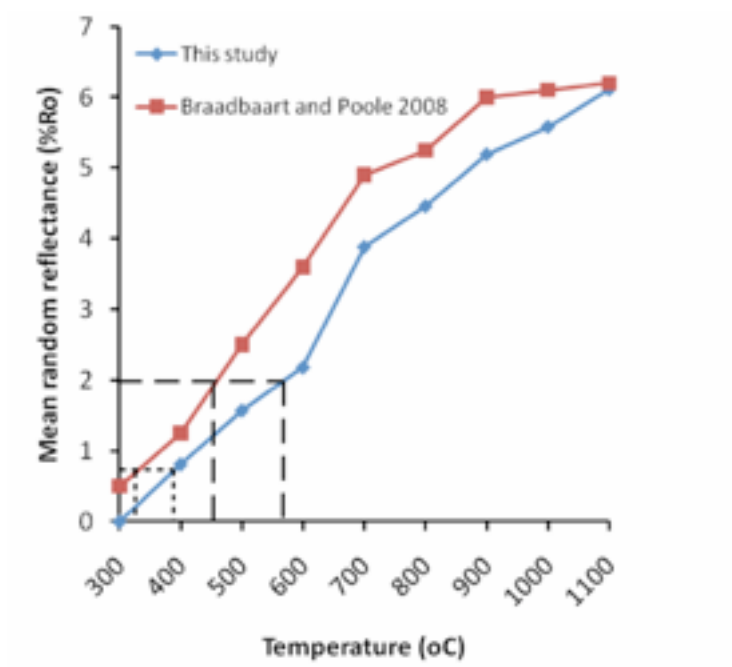
Lissava

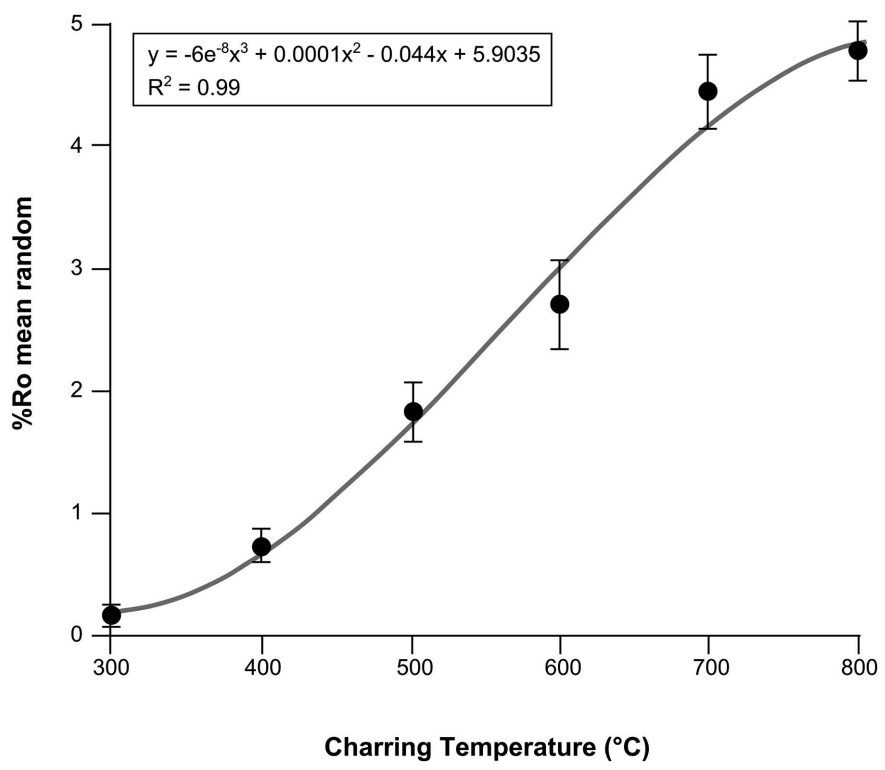
Ballylegan



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Site	Reflectance sample no	Feature No. and type	Charcoal	Bone	Bone colour	Mean reflect.	Min reflect.	Max reflect.	No. of readings	SD	Temp - average (°C)	Temp - max range (°C)	Absolute max per sample (°C)	Absolute max per feature (°C)
Templece	1	141 Grave pit	Maloideae C. Weber., <i>Quercus petraea</i> (Matt.) Liebl. <i>Quercus robur</i> L.	Human	White	1.00	0.67	1.60	50	0.22	375-425	325-510	510	510
	22		Fine dust >250 µm	Human	White	1.20	0.90	1.38	50	0.13	390-440	360-475	475	
	2	133 Grave pit	Maloideae C. Weber., <i>Fraxinus excelsior</i> L.	Human	Grey-white	1.07	0.67	1.41	50	0.18	380-430	325-475	475	515
	26		Fine dust >250 µm	Human	Grey-white	0.87	0.67	1.15	50	0.11	360-410	325-440	440	
	27		Fine dust >250 µm	Human	Grey-white	1.15	0.63	1.67	50	0.26	385-430	325-515	515	
	3	179 Grave pit	<i>Quercus petraea</i> (Matt.) Liebl. <i>Quercus robur</i> L.	Human	White	1.12	0.59	1.83	50	0.38	385-430	325-525	525	525
	4		Maloideae C. Weber sp.	Human	White	1.14	0.73	1.52	50	0.21	385-430	345-490	490	
	24		Fine dust >250 µm	Human	White	1.01	0.59	1.53	50	0.27	375-425	325-500	500	
	5		Maloideae C. Weber. & <i>Quercus petraea</i> (Matt.) Liebl. <i>Quercus robur</i> L.	Human	White	0.97	0.55	1.18	50	0.15	370-415	325-440	440	
	6	207 Grave pit	<i>Corylus avellana</i> L., Maloideae C. Weber., <i>Ulmus glabra</i> Huds.	Human	Grey-white	0.96	0.56	1.49	50	0.21	370-415	325-490	490	490
	25		Fine dust >250 µm	Human	Grey-white	1.10	0.67	1.47	50	0.22	385-430	325-490	490	
	7	265 Grave pit	<i>Quercus petraea</i> (Matt.) Liebl. <i>Quercus robur</i> L., Maloideae C. Weber.	Human	White	0.90	0.56	1.12	50	0.14	360-410	325-430	430	430
	8	33 Domestic pit	Maloideae C. Weber., <i>Corylus avellana</i> L., <i>Ilex aquifolium</i> L., <i>Prunus spinosa</i> L.	No bone		1.05	0.88	1.30	50	0.09	375-425	360-460	460	460
	9	193 Grave pit	<i>Quercus petraea</i> (Matt.) Liebl. <i>Quercus robur</i> L.	Human (older adult)	White	1.09	0.68	1.35	50	0.16	380-430	325-470	470	500
19	Fine dust >250 µm		Human (older adult)	White	1.26	0.93	1.56	50	0.16	390-450	360-500	500		
10	199 Possible pyre	<i>Quercus petraea</i> (Matt.) Liebl. <i>Quercus robur</i> L.	No bone		0.89	0.68	1.15	50	0.12	360-410	325-435	435	435	
Ballylegan	11	446 Hearth	<i>Prunus</i> <i>Prunus avium</i> L. (L.) <i>Prunus padus</i> L., <i>Salix</i> L., <i>Alnus glutinosa</i> (L.) Gaertn., <i>Euonymus</i> L., Maloideae C. Weber., <i>Quercus petraea</i> (Matt.) Liebl. <i>Quercus robur</i> L.			0.97	0.48	1.48	50	0.20	370-415	325-480	480	480
Lissava	12	12 Trough	Maloideae C. Weber., <i>Quercus petraea</i> (Matt.) Liebl. <i>Quercus robur</i> L., <i>Fraxinus excelsior</i> L., <i>Ulmus glabra</i> Huds., <i>Corylus avellana</i> L., <i>Salix</i> L.			1.25	0.85	1.50	50	0.15	390-450	355-490	490	490

	Feature No. and type	Absolute max per feature containing bone (°C)	Absolute max temp. per feature - no bone (°C)
Templenoce	141 Grave pit	510	
	133 Grave pit	515	
	179 Grave pit	525	
	207 Grave pit	490	
	265 Grave pit	430	
	33 Domestic pit		460
	193 Grave pit	500	
	199 Possible pyre		435
Ballylegan	446 Hearth		480
Lissava	12 Trough		490

Temperature (°C)	Low/High %Ro	Maximum %Ro variation
300	0 - 1.2	1.2
350	0.26 - 1	0.83
400	0.6 - 2	1.4
450	1.05 - 1.4	0.35
500	1.39 - 2.5	1.81
550	2.1 - 2.6	0.5
600	2.65 - 3.85	1.2
700 *	3.15 - 4.9	1.75
800 *	3.71 - 5.4	1.69
900 *	4.95 - 6	1.05
1000 *	5.23 - 6.1	0.29
1100 *	5.91 - 6.2	0.26

Overview of reflectance

Reflectance testing has been adopted in coal assaying where it is standardized and used to rank coal quality (see for example <https://www.astm.org/Standards/D7708.htm>).

Charcoal also has an exceptional ability to reflect light when viewed using reflectance microscopy. The amount of light reflected is variable depending on the differential ordering of graphite-like phases within the charcoal itself. The charcoal's carbon atoms organize into an increasingly regular matrix over time. An analogous process is that of the less regular structure of graphite, progressing to the highly organized structure of diamond gradually over time, and with heat and pressure. In charcoal it has been demonstrated that this relates to the temperature of formation, whereby higher (absolute) formation temperatures result in higher charcoal reflectance (due to the more formal ordering of the carbon atoms). Cell wall fusion or homogenization may also play a role. This phenomenon occurs above 325 °C (Scott and Glasspool, 2005; McParland *et al.* 2007, 2009a, 2009b). Testing the reflectance of charcoal, (as for coal), requires collection and mounting of samples in resin. Transverse sections of each of the fragments of charcoal are embedded in polyester resin and polished to a highly smooth surface. In this study, two methods were employed, cold and hot set. The cold set method requires careful manual placement of charcoals at the base of a mould, followed by mixing and pouring in of two part epoxy resin, (without disturbing the sample), and curation (about 24 hours), before careful polishing. The automated hot epoxy method raises the materials to no more than 80° C. Resin tablets can be prepared in about 15 minutes from powdered epoxy, (Citopress hot press) and polishing can be more automated so as to polish multiple samples at once, thus reducing preparation time. At the Materials Laboratory at the University of Sydney, six samples were polished simultaneously, (Struers equipment, using increasingly fine grinding surfaces) allowing complete preparation of six samples in about two hours. Reflectance is measured using a reflecting microscope (in this case, the Nikon microphot microscope attached to Leica QWin image analysis software (Leica Image systems Ltd., 1997). Specimens are measured under Cargill immersion oil (refractive index, R_o , of 1.518 at 23 °C), using the x40 objective lens. Calibration is required, and in this study, the performance of the laser was calibrated against five

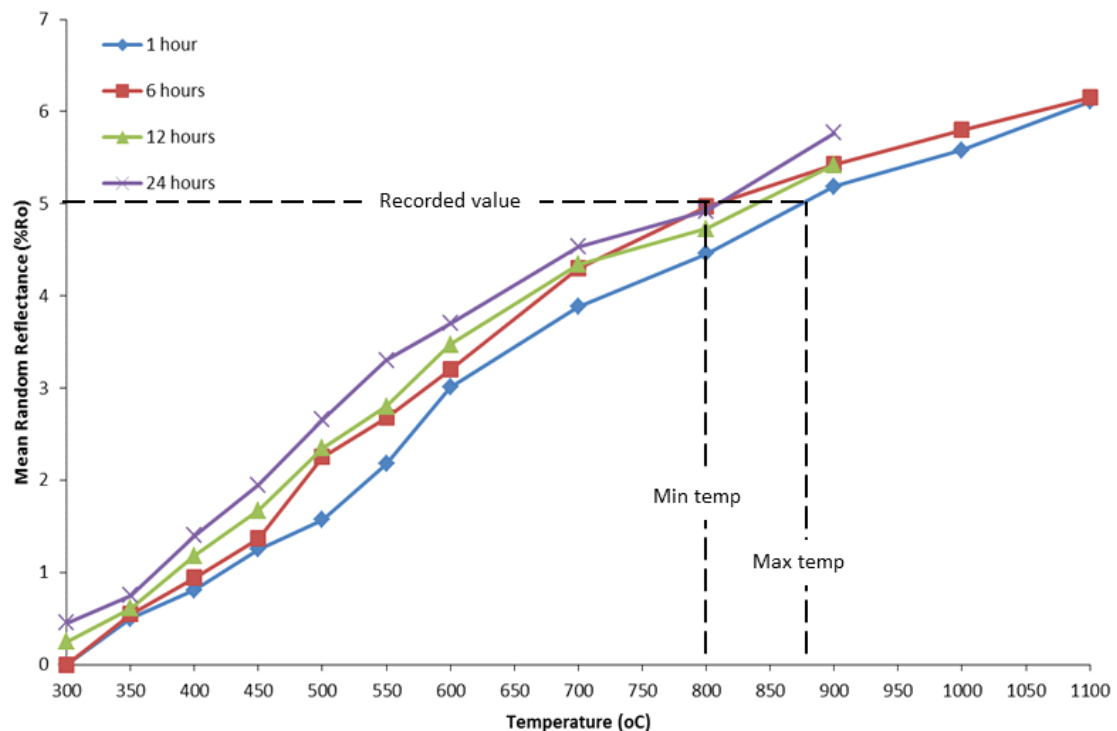
standards: Spinel (Ro 0.393), YAG (Ro 0.929), GGG (Ro 1.7486), cubic zirconium (Ro 3.188) and silicon carbide (Ro 7.506).

Once calibrated, the laser light is bounced off the charcoal surface and its reflected value (Ro) recorded. Usually a number of measurements are made on one sample (50, or even 100), thus providing a range of Ro readings for one fragment of charcoal. If the charcoal concerned is very small (i.e. micro, rather than macro-sized), measurements in the 'one' sample may come from several small fragments. From the range of measurements for one sample, an average is calculated for that sample (Ro avge), and of course there are Ro min and Ro max measurements for that sample (as measured, but these may or may not represent actual minimum and maximum of the sample).

Reflectance calibration curves

A number of researchers have charred woods of various types under varying (controlled) conditions, measuring the absolute temperature of charring, over various timeframes, and then subsequently measuring the reflectance of these modern samples, in order to produce a calibration curve that relates the temperatures observed with the measured reflectance values. McParland et al. (2009b) produced their calibration curve by plotting the reflectance results (y-axis) against the known temperature of formation (x-axis) (Fig. 1). The formation temperatures of samples with unknown charring temperature may then be obtained from the curve using the measured reflectance value (on the y axis), and reading the associated temperature from the x-axis. In this study, the temperature calibration curve for oak created by McParland et al (2009) has been used since oak was the most common taxon in the cremation assemblages. Other curves for softer, harder, and/or more resinous woods read slightly differently. Temperature results of the archaeological material examined here were inferred using a 1, 6, 12 and 24 h calibration curve, i.e. the modern wood samples were charred and their temperatures measured – after 1 hour, 6 hours, 12, and 24 hours to create the calibration curve. The results are expressed as a range of temperatures e.g. a reflectance of 5%Ro would need to be charred at 900° C for one hour or 800° C for 24 h giving a range of 800-900 ° C. A natural outworking of these observations suggests that quick charring to a particular temperature, (i.e. in one

hour), will produce a higher reflectance value than slower charring (e.g. 24 hours) to the same temperature. Since cremation of a full body takes about one day, we might infer that slower charring might have been occurring, however, flame temperatures fluctuate over the cremation process, for example, upon original ignition of clothes, and starter materials (brush woods, or potentially oils), and again when body fats reach ignition temperature. As stated in the main text, much of the body is burnt in a few hours, and even when cremation is ‘completed,’ some body parts may not have been burnt.

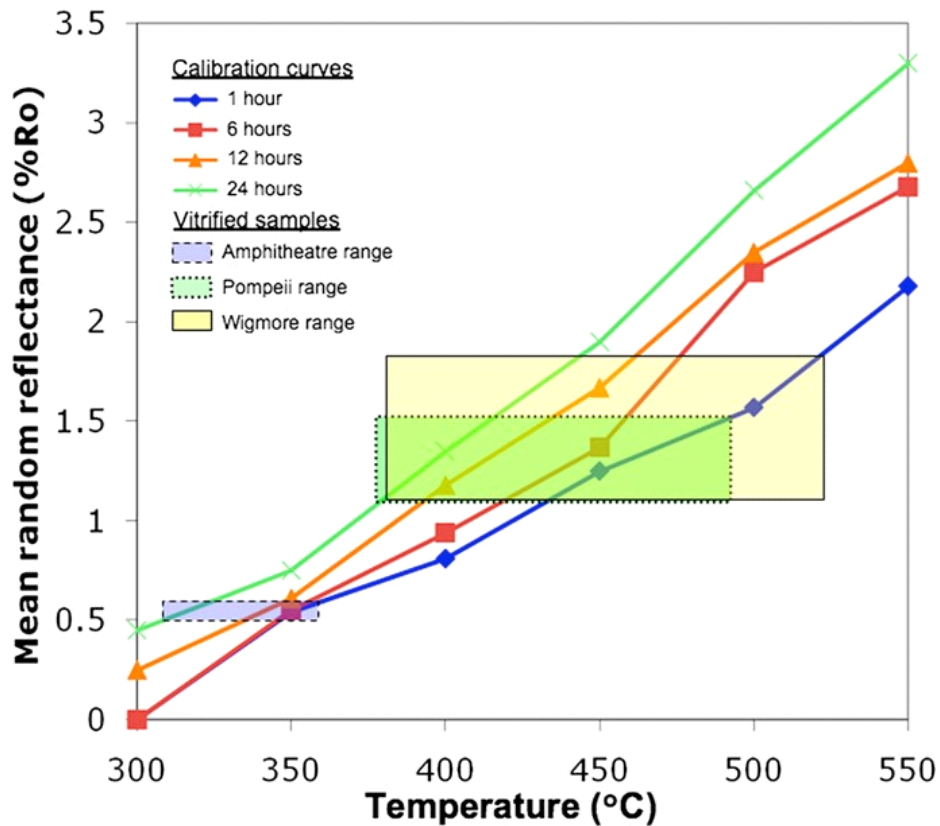


SUPPLEMENTARY FIGURE 1 Calibration graph of reflectance vs burn temperature (modified from McParland et al. 2009b) (Corresponds to figure 1 in main text).

The reflectance observed in charcoals is anisotropic in nature and therefore, the

measurements taken will depend upon the position of the sample relative to the plane of polarisation of the laser light. Reflectance is reported either as maximum reflectance (R max) or random reflectance (%Ro). R max is measured by orientating the sample or polariser to provide the reflectance maximum (which takes time to carry out). %Ro (mean random reflectance) simply takes a mean from the sample (from a number of measurements taken at random). Samples in this case have not been aligned to find the Ro max and therefore will often be lower. Mean random reflectance is reported throughout this study, in that the sample is not orientated before measuring. There is still a minimum random reflectance value and a maximum random reflectance value, however, these should not be confused with true maximum reflectance (R max).

Generally, in coal samples, and in some other charcoal reflectance studies, calculating the average measurement is sufficient, however, given we are interested in the highest observable temperature as randomly measured (i.e. %Ro max), we ultimately refer to the highest random measurement taken for each sample. By way of further explanation, in a study examining the phenomenon of 'vitrification' in archaeological charcoal, (McParland *et al.*, 2010), the reflectance ranges were measured for a variety of charcoals obtained from different charring/burning modalities (see Supplementary figure 2). Reflectance of wildfire charcoals is also now carried out to test differences in wildfire regimes, e.g. crown fires vs ground fires, the former being much hotter than the latter, (Hudspith *et al.*, 2015; Belcher and Hudspith, 2016).



SUPPLEMENTARY FIGURE 2. From McParland *et al.* 2010. Range of mean random reflectance (%Ro) of the vitrified areas of the archaeological fragments from three archaeological sites: Pompeii, Wigmore and Chester amphitheatre, plotted against the calibration curves of temperature vs. reflectance (for experimentally charred oak) in order to show the range in temperature of formation of the vitrified charcoals. Laboratory calibration charring data from McParland *et al.* (2009a).

Fire, temperature and fuel behavior (supplementary material 2)

A detailed understanding of how fire behaves may inform our understanding of fuel behavior during cremation, and its probable location after cremation is concluded. Much of our understanding about fire arises from studying wildfires, as well as forensic studies of modern fires where human are incinerated. This discussion draws on these sources, as well as limited works on ancient cremation.

The building of a cremation pyre will result in the provision of: the major fuel, timber logs possibly constructed in a pyramidal shape (with decreasing diameter); the secondary fuel (the body, especially the body fat); and the ignition fuels (which may be a range of products from kindling and shrubs to oils, we have little evidence for these from cremation sites). Bodies in modern and Roman cremations are normally placed on top of the pyre, in a shroud, or indeed in clothing (either of these provide more low ignition fuels). The major ignition fuels are placed around the base of, and potentially inside, (or in the case of oil, poured on to), the pyre, and ignition of these commences the burning process. The fire so started, must eventually transfer enough heat to the unburnt major fuel, the wood (Scott *et al.* 2014:302), and then also to the body; heat may be transferred by three different processes (Scott *et al.* 2014:303):

i) Convection: the transfer of heat by movement of a fluid (liquid or gas), including by actual flame contact. In a cremation, this is the natural movement of hot air and gaseous combustion products upwards, a process that will be further assisted by a draft. Convective heat is not visible to the naked eye.

ii) Radiation: transfer of heat in straight lines at the speed of light from hot particles of matter (solid, liquid, or gas) to cooler regions in the fuel layer, or its surroundings. In a modern open air cremation, this is felt from the visible flames, and will increase as the cremation proceeds. Uneven burning/spreading of flames will cause hot air turbulence, which may be significant and result in movement of burnt and partially burnt fuels.

iii) Conduction: the transfer of heat through matter from a region of high temperature to a region of lower temperature. Again, conduction may add to turbulence, and thus fuel movement within the pyre.

As the ignition fuels burn, white vapours are emitted around 100 °C (if the fuel is not completely dry), this is water vapour. Following this stage as the temperature rises, organic hydrocarbons vaporize and are emitted (blue smoke). A similar pattern occurs with the wood. More detailed information is provided in Braadbaart *et al.* (2012:844), and Harrison (2013). Once flaming combustion is established on a fuel surface, it reaches an equilibrium condition where its temperature stabilizes no matter what the temperatures of the flames above may be (Dehann, 2008: 5). For wood the surface temperature of the horizontal fuel is of the order of 350-400 °C. Radiant heat from flames above, and smouldering combustion in the surface are balanced against radiative and convective losses, and the vaporization of the organic volatiles. Wood fires produce a maximum flame temperature of 1027 °C (Dehann, 2008: 4), however, flame temperature varies with height from the fuel surface. A steady flame reaches a maximum temperature just above the fuel surface (800-900 °C) (Denhann, 2008:6). Fires are typically ‘turbulent’, with the surface of the burning fuel rising and falling, and the atmosphere ranging from reductive to occasionally oxidative. This living and changing nature of actual fire is very different to the conditions wrought by placing bone (or wood), in a laboratory oven at a specific temperature.

The best fuel in the body is the subcutaneous fat, which has an auto-ignition temperature of approximately 350 °C (Dehann 2008:9). Body fat does not smoulder and will only burn as a flame, and it requires a rigid porous ‘wick’ to maintain the flame. This wick can be charred wood, clothing, or even bone. Body fat produces an average temperature of about 800 °C, and turbulent flames (Dehann 2008:9). It contributes substantially to the burning process, evaporating body fluids, degrading, drying and finally burning skin, and muscle (which burns reluctantly).

Modern house fires, where hot gases and pyrolysis products collect in a room, eventually reach a 'flashpoint' where the fire suddenly ignites these products, and temperatures can reach 1000 °C, with post flashover temperatures as high as 1200 °C (Dehann 2008:11). Some Roman cremations were held 'inside' four walls (open to the sky), presumably to keep heat in, as substantial marble constructions have been excavated (e.g. the ustrina of Antininus Pius and that of Marcus Aurelius). In a prehistoric cremation, we have no knowledge of the positioning of regularly used pyre sites in this respect, but it cannot be discounted.