

1 **Ocean acidification does not impact shell growth or repair of the Antarctic**
2 **brachiopod *Liothyrella uva* (Broderip, 1833)**

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21 **Abstract**

22 Marine calcifiers are amongst the most vulnerable organisms to ocean acidification
23 due to reduction in the availability of carbonate ions for skeletal/shell deposition.
24 However, there are limited long-term studies on the possible impacts of increased
25 $p\text{CO}_2$ on these taxa. A 7 month CO_2 perturbation experiment was performed on one
26 of the most calcium carbonate dependent species, the Antarctic brachiopod
27 *Liothyrella uva*, which inhabits the Southern Ocean where carbonate ion saturation
28 levels are amongst the lowest on Earth. The effects of the predicted environmental
29 conditions in 2050 and 2100 on the growth rate and ability to repair shell in *L.uva*
30 were tested with four treatments; a low temperature control (0°C, pH7.98), a pH
31 control (2°C, pH8.05), mid-century scenario (2°C, pH7.75) and end-century scenario
32 (2°C, pH7.54). Environmental change impacts on shell repair are rarely studied, but
33 here repair was not affected by either acidified conditions or temperature. Growth
34 rate was also not impacted by low pH. Elevated temperature did, however, increase
35 growth rates. The ability of *L.uva* to continue, and even increase shell production in
36 warmer and acidified seawater suggests that this species can acclimate to these
37 combined stressors and generate suitable conditions for shell growth at the site of
38 calcification.

39

40 **1. Introduction**

41 Increasing CO_2 levels from anthropogenic activities over the past 250 years has
42 altered our oceans through warming and also acidification (Caldeira and Wickett,
43 2003, 2005; IPCC, 2013; Orr et al., 2005). This latter process has received much

44 attention recently with the chemical implications now being fairly well described,
45 although the biological and ecological consequences are less well described
46 (Gattuso et al., 2013; Wittmann and Pörtner, 2013). However, there is a consensus
47 that marine calcifying organisms are the most susceptible animal group to ocean
48 acidification because the predicted reduction in the availability of carbonate ions will
49 make it more difficult and more energetically expensive for shell production (Byrne,
50 2011; Byrne and Przeslawski, 2013; Doney et al., 2009; Watson et al., 2012). To
51 date studies have reported varied responses of calcifying organisms to future
52 predicted pH levels with an increasing number of studies indicating that some
53 species are tolerant (Havenhand and Schlegel, 2009; Parker et al., 2012; Ries et al.,
54 2009). However, it must be recognised that the majority of experiments have been
55 conducted on relatively short time scales so the ability of organisms to acclimate or
56 adapt is largely unknown (Byrne, 2011; Byrne and Przeslawski, 2013; Gattuso et al.,
57 2013; Wittmann and Pörtner, 2013). Longer term studies are increasing though,
58 which are providing insights into how organisms are coping with acidifying oceans
59 (Form and Riebesell, 2012; Kelly et al., 2013; Pandolfi et al., 2011; Pespeni et al.,
60 2013).

61 The fastest rates of change in carbonate chemistry are expected in the
62 Southern Ocean (Caldeira and Wickett, 2005; McNeil and Matear, 2008). CO₂ is
63 more soluble in cold water (Revelle and Fairbridge, 1957) resulting in naturally low
64 carbonate ion saturation levels compared to temperate and tropical regions. Acid-
65 base coefficients are also more sensitive in cold temperatures making this high
66 latitude region a forerunner of biological ocean acidification impacts for other oceans
67 (Fabry et al., 2009). Furthermore, the absence of shell-crushing predators, such as

68 crabs, lobsters and heavily jawed fish (Aronson et al., 2007) and the difficulty of
69 extracting Ca^{2+} from seawater at low temperature (Aronson et al., 2007; Harper,
70 2000) has resulted in Antarctic species generally having thin, weakly calcified shells
71 (Vermeij, 1978; Watson et al., 2012). This, added to the low physiological rates of
72 Antarctic marine species (Peck et al., 2007), especially low metabolic rates (Peck
73 and Conway, 2000), slow growth rates (Arntz et al., 1994), delayed reproduction
74 (Meidlinger et al., 1998) and high longevity (Pearse et al., 1991), indicates that these
75 organisms are likely to be amongst the most vulnerable species worldwide to
76 acidifying oceans. Although there are several studies on the potential impacts of this
77 aspect of climate change on the larval stage of Antarctic calcifying organisms (see
78 review by Byrne (2011)), there are limited studies on adults (Cummings et al., 2011;
79 McClintock et al., 2009), and the longest of these lasted 4 months.

80 Rhychonelliform brachiopods are potentially the most calcium carbonate
81 dependent group of marine animals because their calcareous skeleton and other
82 support structures makes up over 90% of their dry mass (Peck, 1993, 2008). They
83 have locally also been important organisms in shallow water communities, providing
84 a habitat for a diverse range of epifauna including encrusting sponges and algae
85 (Barnes and Peck, 1996), for the last 500 million years surviving several geological
86 periods where the pH has fluctuated. Ocean pH has declined in the past 250 years at
87 a rate of at least an order of magnitude faster than has occurred for millions of years
88 (Doney and Schimel, 2007; Doney et al., 2009). Despite this, only two studies have
89 addressed the potential climate change impacts on extant brachiopods (McClintock
90 et al., 2009; Peck, 2008) and only the former investigated ocean acidification effects
91 where significant dissolution in *Liothyrella uva* (our target taxon) valves was found

92 after only 14 days in pH 7.4 conditions. However, only empty valves were used so
93 the biological response and ability of *L.uva* to compensate for the impacts of future
94 pH conditions therefore remain to be investigated.

95 *Liothyrella uva* (Broderip, 1833) is a large (maximum recorded length is 55
96 mm), epifaunal, sessile, suspension-feeding terebratulide brachiopod with a
97 circumpolar distribution (Peck et al., 2001). It is found down to 300 metres and is
98 highly abundant in habitats protected from anchor ice and ice scour with reported
99 densities up to 3000 individuals per m² (Foster, 1974; Peck et al., 2001). *L.uva* is
100 typically found attached singly or in clumps to vertical and overhanging rocks around
101 the South Orkney Islands, the Antarctic Peninsula and Peter I Island (Foster, 1974).
102 Previous growth studies on *L.uva* have recorded slower growth rates than temperate
103 rhychonelliform brachiopods and it can live for over 55 years (Peck and Brey, 1996).
104 It also has a limited tolerance to elevated temperature, surviving up to 4.5°C (Peck et
105 al., 2001).

106 The shell is essential to the existence of *L.uva*, providing protection from
107 predators and preventing any encounters with harmful substances and the loss of
108 body fluids (Harper et al., 2012). Any environmental insult negatively impacting the
109 production, maintenance and/or repair of their shell could thus prove fatal. *L.uva* also
110 becomes naturally damaged and their shells require repairing in the environment
111 (Harper et al., 2009). Given this, and the current focus on ocean acidification, the
112 aims of this study were to determine how shell growth rates and the frequency of
113 shell repair following damage in *L.uva* were affected in a 7 month experiment using
114 predicted mid and end century pH levels.

115

116 2. Materials and methods

117 2.1. Sampling collection

118 Specimens of *Liothyrella uva* (Broderip, 1833) were hand collected by SCUBA
119 divers from Trolval Island, Ryder Bay, Antarctica (67° 35.44' S, 68° 12.44' W) at 15-
120 25 m depth in May 2012. Animals remained in their conspecific clumps with only the
121 pedicle of the central brachiopod attached to the cliff face being cut ensuring that the
122 majority of specimens were not damaged during collection. Environmental conditions
123 in Ryder Bay at 15-25m depth consist of seawater temperatures that range from -1.8
124 to +1.0°C, however, temperatures rarely exceed +0.5°C (Clarke et al., 2008), the pH
125 range is 8.04-8.10 (McNeil and Mearns, 2008) and salinity is 33.0-34.0 (Clarke et al.,
126 2008). Brachiopods were kept underwater during the short transportation from the
127 sampling site to the marine laboratory in Rothera and in recirculating aquaria (0.0 ±
128 0.5°C) whilst being transported back to the UK. Specimens remained in an ambient
129 recirculating seawater system in the UK in similar conditions for a further two weeks
130 to habituate to aquarium conditions before the experiment began.

131 2.2. Experimental Design

132 This study was conducted in a recirculating CO₂ microcosm adapted from
133 (Suckling et al., in press) at the British Antarctic Survey (BAS), UK. Four treatments
134 were used where two functioned as lowered pH treatments (pH7.75 and pH7.54)
135 based on the IPCC 'business-as-usual' scenario of the predicted reduction of 0.3-0.5
136 pH units from the present day average of pH8.05 in oceanic surface waters by 2100
137 (Table 1) (IPCC, 2013). The third was a pH control as the seawater remained at
138 ambient pH (pH8.05). As a concurrent 2°C increase in temperature is expected to

139 occur alongside this forecasted decrease in pH by the end of the century (Mitchell et
140 al., 1998), these three systems were maintained at 2°C throughout the experiment.
141 The fourth remaining system was a temperature control which was held at the
142 present-day average surface seawater temperature for Ryder Bay, 0°C (Clarke et al.,
143 2008). The average pH of this treatment was pH7.98 which was slightly lower than,
144 but close to, the pH of the pH control treatment probably as a result of the increased
145 solubility of CO₂ and carbonates (CO₃²⁻) in seawater at the lower temperatures.

146 The pH of the lowered pH treatments was altered by intermittently bubbling CO₂
147 gas through a ceramic diffuser to maintain the pH at the predetermined pH levels via
148 a solenoid valve connected to an Aqua Medic pH controlled computer and glass
149 electrode (with plastic shaft) system. The pH control had a similar set up but without
150 the pH control system. An Aqua Medic Ocean Runner power head 2000 circulated
151 the seawater in the mixing tank to ensure a constant pH. Seawater was then gravity
152 fed from each mixing tank at a rate of $0.65 \pm 0.03 \text{ L min}^{-1}$ into the experimental tank.
153 Seawater temperature was manipulated by the use of temperature-controlled
154 laboratories. Air temperature was maintained at -2.5°C in the laboratory with the pH
155 control and both lowered pH treatments but the lifting pumps (Aqua Medic Ocean
156 Runner 3500) and the mixing power heads (Aqua Medic Ocean Runner 2000) in
157 each treatment's mixing tank caused the seawater temperature to raise to the
158 desired ~2°C, with minimal variability (Table 1). The temperature control treatment
159 was situated in the main BAS aquarium where the air temperature was set at -1.5°C
160 and the absence of lifting and circulating pumps caused the seawater temperature to
161 be maintained at ~0°C.

162 Seawater temperatures ($^{\circ}\text{C}$, Digital Testo 106) and pH_{NIST} (Aquamedic pH
163 controlled computer and electrode system) were monitored and recorded daily.
164 pH_{NIST} was also more accurately measured once a week with a temperature
165 compensated HANNA bench top meter pH/ORP 115 v pH21-01). Salinity (Tropical
166 Marine Centre V2 Handheld refractometer), TCO_2 (mmol L^{-1} ; Ciba Corning TCO_2
167 Analyzer 965, Olympic Analytical. UK) and nutrient content (silicate and phosphate;
168 according to methods in Nickell et al. (2003)) of each treatment were also measured
169 weekly. Twice a week, the Aqua Medic pH probes were calibrated with NIST certified
170 pH buffers. Other carbonate system parameters, including the partial pressure of CO_2
171 (pCO_2) and the saturation values for calcite (Ω_{C}) and aragonite (Ω_{A}), were modelled
172 from applying TCO_2 and pH_{NIST} data to the program CO2SYS (Lewis and Wallace,
173 1998) with refitted constants (Dickson and Millero, 1987; Mehrbach et al., 1973).
174 Brachiopods in each treatment were fed weekly with microalgal concentrate of
175 approximately 331×10^4 cells L^{-1} which is within the natural range of phytoplankton
176 cell abundance along the west Antarctic Peninsula ($62 - 1150 \times 10^4$ cells L^{-1} ;
177 (Garibotti et al., 2005; Garibotti et al., 2003). Water quality was maintained and
178 alkalinity and other ions replenished through water changes twice a week and weekly
179 siphoning of the aquaria to remove any debris.

180 2.3. *Growth rates*

181 One hundred and eleven specimens of *L.uva* across a wide size range (2.6-40.9
182 mm in length) and varying number of individuals in each conspecific clump (30
183 conspecific clumps with 2-15 individuals and 20 single specimens) were used in this
184 experiment. Prior to the beginning of the experiment, shell lengths of each individual
185 in each conspecific clump were measured to the nearest 0.1 mm using Vernier

186 calipers. Conspecific clumps were placed into one of the four treatments ensuring
187 there was a similar size range of specimens and numbers of conspecific clumps and
188 single individuals in each treatment. The experiment began in mid June 2012 and
189 lasted for 7 months. At the mid-way and end of the experiment, the length of each
190 specimen was measured again, recorded and the shell edge photographed. Growth
191 rates were then calculated from the increase in length ($\mu\text{m day}^{-1}$).

192 2.4. *Shell repair frequencies*

193 Forty five specimens (8-12 individuals in each treatment) with a range of sizes
194 (5.0-37.0 mm in length) were selected at random and damaged by creating a 1-2 mm
195 deep notch at the valve edge using a metal file. Care was exercised to create
196 notches of equivalent size and severity and not to break shells or cause other
197 damage. After 3.5 and 7 months of the experiment, lengths were measured for each
198 specimen and the damaged region of the shell edge photographed (Figure 1).

199 2.5. *Statistical analyses*

200 All data were analysed using Minitab (Statistical Software™ Version 15). Growth
201 rate data for each treatment were all shown to be significantly different from normal
202 (Anderson-Darling Test; $P < 0.008$). These data were still not normally distributed after
203 square root, logarithmic and double logarithmic transformations because of the
204 presence of zeros in the dataset. Non-parametric Kruskal-Wallis tests were thus used
205 to determine whether damage affected growth rate in each treatment and whether
206 treatment affected growth. When a significant difference was found in the growth rate
207 of undamaged individuals, a further Kruskal-Wallis Multiple Comparisons test was
208 used to identify which treatments were statistically different from each other. Chi-

209 squared tests were used to determine if treatment affected the percentage of
210 individuals that had completed shell repair, the percentage of individuals that had
211 produced shell after fully repairing their notch and the percentage of individuals that
212 did not grow throughout the experiment.

213

214 **3. Results**

215 Throughout this 7 month experiment, the saturation states of aragonite and
216 calcite were just below the range usually reported for polar shallow coastal seawater
217 (Table 1; Barry et al. (2010)). In both controls, aragonite was slightly undersaturated
218 ($\Omega < 1$) whereas calcite was supersaturated ($\Omega > 1$). Saturation states with respect to
219 aragonite and calcite in every other treatment were undersaturated (Table 1). The
220 order of treatments with decreasing saturation state with respect to calcite was pH
221 control > temperature control > pH7.75 > pH7.54. Despite these potentially challenging
222 conditions, mortality across all treatments was low at 3.9% (total N=155) with no
223 differences between treatments (temperature control = 4% (n=3), pH control = 8%
224 (n=2), pH7.75 = 0% (n=0) and pH7.54 = 4% (n=1)). Also, the animals showed none
225 of the previously reported signs of stress (slow snapping responses, remaining
226 closed for extended periods, wide gaped when open; Peck (2001)) and responded
227 rapidly to physical stimulation when disturbed, throughout the study. A few females in
228 each treatment also managed to successfully complete brooding and release larvae
229 with large numbers (up to 100) settling on shells of older brachiopods and continuing
230 development to juveniles (E.C. unpubl. obs.). Survival of these newly settled
231 juveniles between 3 and 7 months was also high (over 90%).

232 3.1. *Shell repair frequencies*

233 After 3 months, all damaged individuals had at least started shell repair with >50% of
234 individuals fully repairing their notch across all treatments (Table 2). Following the
235 completion of shell repair, >33% of specimens had also produced new shell at the
236 anterior margin (Figure 1). After 7 months, >83% of individuals had completed shell
237 repair in every treatment (Table 2) with no significant difference between treatments
238 ($\chi^2 = 0.839$, $p = 0.840$). Punctae, shell perforations, which are part of the shell
239 structure in this species were present in the repaired notches and the new shell
240 growth. A high proportion (>64%) of specimens had also produced new shell at the
241 growth margin in three of the four treatments, including in both acidified treatments.
242 However, only a moderate proportion (38%) of specimens produced growth at the
243 shell margin in the pH control treatment. Treatment appeared to have an effect on
244 the proportion of individuals that had continued shell production after the completion
245 of repair ($\chi^2 = 9.670$, $p = 0.022$), however, a further Chi-squared test on the same
246 dataset but excluding the pH control indicated there was no significant difference (χ^2
247 = 0.815, $p = 0.665$). All individuals that had completed shell repair and then made
248 new shell growth after 3 months continued to produce new shell after 7 months
249 (Figure 1).

250 3.2. *Growth rates*

251 In the undamaged individuals, the majority, but not all individuals, grew (Figure 2 and
252 3). In those that did grow, growth rates ranged from 2.4 to 33.7 $\mu\text{m day}^{-1}$ in all three
253 treatments at the higher temperature (pH control, pH7.75 and pH7.54; Figure 2).
254 However, growth rates in the lower temperature trial (temperature control treatment)
255 were lower ranging from 1.2 to 13.2 $\mu\text{m day}^{-1}$ (Figure 3). The only clear ontogenetic

256 trend in every treatment was that individuals with initial lengths of >32.0 mm
257 produced no growth. Growth rates of undamaged individuals in both the acidified
258 treatments were not significantly different from the pH control (Kruskal-Wallis, H =
259 3.96, p = 0.138). A further Kruskal-Wallis test including the temperature control,
260 showed a significant effect of treatment on the growth rate (H = 13.06, p = 0.005). A
261 Kruskal-Wallis Multiple Comparisons test indicated that higher temperature increased
262 the growth rate over the temperature control in two treatments (pH control: Z = 2.20,
263 p = 0.028; pH7.75: Z = 3.21, p = 0.001), however, growth rate in the lowest pH
264 treatment (pH7.54) was not significantly different from the temperature control
265 treatment (Z = 0.93, p = 0.352).

266 In damaged individuals that produced new growth at the shell margin, growth
267 rates ranged from 1.2 to 23.1 $\mu\text{m day}^{-1}$ across all treatments with no ontogenetic
268 trend (Figure 2 and 3), and there was no effect of treatment on growth rate of
269 damaged individuals (Kruskal-Wallis, H = 1.24, P = 0.743).

270 Growth rates were not significantly different in damaged and undamaged
271 individuals in any treatment (Temperature control – H = 0.24, P = 0.623; pH control –
272 H = 2.82, P = 0.093; pH7.75 – H = 3.96, P = 0.046; pH7.54 – H = 0.06, P = 0.807).
273 Treatment also had no detectable effect on the proportion of damaged or undamaged
274 individuals that did not grow throughout the experiment ($\chi^2 = 2.590$, P = 0.459).

275

276 4. Discussion

277 Clearly *Liothyrella uva* are able to tolerate predicted 2050 and 2100 pH levels
278 as specimens showed no signs of prolonged stress and mortality was low. A similar

279 mortality rate was found in a study of growth rates of *Liothyrella uva* in their natural
280 habitat (2% yr⁻¹; Peck et al. (1997)). The ability of the damaged individuals to repair
281 their shells was not affected by acidified conditions or temperature and >90% of all
282 injured specimens had completed shell repair and >63% had continued shell
283 deposition after completion of repair after 7 months. This suggests that *Liothyrella*
284 *uva* will be able to repair damage to their shells under the predicted pH and warming
285 conditions in the next century. Similarly, shell repair rates in the gastropod *Subnivalia*
286 *undulata* were also not affected by ocean acidification, however, the gastropod
287 *Austrocochlea porcata* exhibited a decreased rate (Coleman et al., 2014). This study
288 concluded that different species have varying tolerances and therefore the response
289 is species-specific. Shell repair rates in living brachiopods have only been
290 investigated in purposely damaged specimens of the temperate brachiopod,
291 *Terebratulina retusa*, in ambient seawater where shell repair began after four weeks
292 and caeca developed after eight weeks (Alexander et al., 1992). All notches in the
293 Antarctic brachiopod in this study were made in the anterior margin where new shell
294 is laid down, and so relatively easily repaired. The chemical environments differed
295 between the treatments, but this did not affect the success rate of repairs in
296 *Liothyrella uva*, possibly reflecting an increase in the rate of shell regeneration in the
297 altered pH conditions as seen in some other species (e.g. Wood et al. (2008)).
298 Energetic costs of repairing damaged shells, and whether the repair material has the
299 same shell structure under altered pH conditions is under investigation.

300 Growth rates in both undamaged and damaged individuals were not affected
301 in the pH control or lowered pH, indicating that *L.uva* should be able to continue shell
302 production in the predicted future ocean acidification conditions even after the

303 disturbance of completing repair to shell damage. Other studies of shell growth in
304 altered pH conditions, which have all been conducted on non-Antarctic molluscs,
305 have demonstrated mixed responses. Early studies showed reduced growth in two
306 congeneric bivalves, *Mytilus galloprovincialis* (Michaelidis et al., 2005) and *M. edulis*
307 (Berge et al., 2006), after medium-term exposure of 90 and 44 days, respectively.
308 However, acidified conditions as low as pH7.3 (Michaelidis et al., 2005) and pH6.7
309 (Berge et al., 2006) were used and no growth occurred in the latter. The negative
310 effect in *M. edulis* started between pH7.4 and pH7.1 (Berge et al., 2006), a level
311 which is below the range of pH used in the current work but also below the predicted
312 end century acidified oceans. More recently, studies have predominantly used
313 predicted mid and end century pH levels (>pH7.5). However, short or medium-term
314 exposure (49-84 days, or less) is still generally the norm and this may not allow for
315 acclimation and trans-generational effects. Decreased growth rates in some species
316 of gastropods (*Urosalpinx cinerea*, *Littorina littorea*, *Nucella lamellosa*) are still seen
317 in these more environmentally relevant experiments (Nienhuis et al., 2010; Ries et
318 al., 2009). Although, lowered pH conditions have also had no impact on growth rates
319 in some molluscs (*Arctica islandica* and *Mytilus edulis*) (Hiebenthal et al., 2012; Ries
320 et al., 2009; Thomsen et al., 2010) and *Crepidula fornicata* has even demonstrated a
321 positive effect to intermediate $p\text{CO}_2$ levels (605 and 903 ppm) (Ries et al., 2009). An
322 extensive study on 18 benthic marine calcifiers concluded that the affect of ocean
323 acidification on the calcification process is species-specific (Ries et al., 2009). The
324 ability of *L. uva* to continue shell production in low pH and undersaturated with
325 respect to aragonite and calcite, suggests that this species can generate suitable
326 conditions at the site of calcification (Gazeau et al., 2013; Ries, 2011; Wittmann and
327 Pörtner, 2013). It is believed that marine calcifiers can elevate pH in calcifying

328 compartments to facilitate calcium carbonate precipitation, however, the mechanisms
329 are largely unknown, especially in the less-studied brachiopods. Potential methods
330 are via either proton channelling (McConnaughey and Falk, 1991), Ca²⁺-activated
331 proton-translocating ATPase (Cohen and McConnaughey, 2003; McConnaughey and
332 Falk, 1991; McConnaughey and Whelan, 1997), transcellular symporter and co-
333 transporter proton-solute shuttling (McConnaughey and Whelan, 1997) or cellular
334 extrusion of hydroxyl ions into the calcifying medium (Ries, 2011; Ries et al., 2009).
335 Further research is needed to determine which mechanism is used in the study
336 species, however, it is apparent that *Liothyrella uva* has a robust control on the
337 calcification process similar to molluscs.

338 Increasing the temperature to the predicted warming conditions alongside
339 ocean acidification positively affected the growth rate of *Liothyrella uva*. This
340 suggests shell deposition could occur at a faster rate in 50-100 years as a
341 consequence of elevated temperatures irrespective of the acidity of the oceans. The
342 long-term upper temperature tolerance limit of the Antarctic brachiopod is 4.5°C
343 (Peck, 1989) and higher shell growth rates have been shown with increasing
344 temperatures in other species (Almada-Vilela et al., 1982; Lewis and Cerrato, 1997;
345 Stor et al., 1982), therefore, warming to 2°C is unlikely to negatively impact shell
346 production in *Liothyrella uva*. Several ocean acidification studies have involved
347 combined effects with other environmental variables, particularly with temperature
348 (Courtney et al., 2013; Findlay et al., 2008; Hiebenthal et al., 2012; Martin and
349 Gattuso, 2009; Reynaud et al., 2003; Rodolfo-Metalpa et al., 2011). Organism
350 responses to the interaction of these variables have been varied where temperature
351 has both increased and decreased the effect size of ocean acidification, which has

352 been found in closely related molluscs, *M.edulis* and *M.galloprovincialis* (Hiebenthal
353 et al., 2012; Rodolfo-Metalpa et al., 2011). Rodolfo-Metalpa et al. (2011) found a
354 negative synergistic effect of low pH (pH_T 7.4) and elevated temperature (25°C) on
355 the gross calcification rate of *M.galloprovincialis* whereas Hiebenthal et al. (2012)
356 found that acidified conditions (pCO_2 $1093.3 \pm 123.6 \mu\text{atm}$ and $1654.5 \pm 82.7 \mu\text{atm}$)
357 did not have an impact on shell growth of closely related *M.edulis*, however, elevated
358 temperature did. Growth in *M.edulis* increased up to 20°C but then decreased at
359 25°C indicating that 25°C is above this species temperature threshold limit
360 (Hiebenthal et al., 2012; Pörtner, 2008). Growth rates in the temperature control
361 treatment here, which was used to compare the predicted scenarios to current
362 Antarctic conditions, were in the same range (0-15 μm per day) as individuals of the
363 same size (2.5-40.9 mm) in a study of *Liothyrella uva* growth rates in the wild (Peck
364 et al., 1997). No growth occurred in individuals >32.0 mm in the current study. This is
365 the approximate size that this species reaches sexual maturity (Meidlinger et al.,
366 1998; Peck and Holmes, 1989), and our data possibly results from a shift in energy
367 from somatic growth to reproduction.

368 Even though brachiopods arguably have the largest proportion of skeleton to
369 tissue mass of any group and *Liothyrella uva* inhabits the Southern Ocean which has
370 the naturally lowest carbonate ion saturation levels, our study shows that this species
371 can survive, repair shell damage and deposit new shell after 7 months exposure to
372 forecasted 2050 and 2100 pH conditions. However, there may be a cost to this
373 apparent resilience under future climate change scenarios. The ophiuroid brittlestar,
374 *Amphiura filiformis*, increased its metabolism and calcification to compensate for
375 increased seawater acidity, which came as a substantial cost as muscle wastage

376 was reported to coincide with this increase (Wood et al., 2008). *L.uva* is a low energy
377 species that uses low metabolic rates to survive the long period of low food
378 availability in the Antarctic winter, although it can opportunistically exploit
379 resuspended benthic microalgae when available (Peck et al., 2005). Increased
380 metabolic costs from elevated temperature may, however, compromise abilities to
381 survive periods of low food availability. The current study is a long-term CO₂
382 perturbation experiment measuring shell growth on a marine organism with a shell
383 permanently exposed to seawater, therefore, providing insights into how these highly
384 calcium carbonate dependent organisms can acclimate in future CO₂ scenarios. The
385 main mechanisms that organisms can use to respond to environmental change are
386 acclimation and genetic adaptation (Peck, 2011; Somero, 2010). Acclimation to
387 altered temperatures can take 6 months in Antarctic marine invertebrates (Peck et
388 al., 2014), and rates of acclimation to altered pH in negatively impacted species
389 remain to be quantified. Comparable long term studies on other species from polar to
390 tropical environments are essential to increase our knowledge of the capability of
391 these integral organisms to succeed under changing environmental conditions.

392

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- 629

630 **Figure legends**

631 **Figure 1** – Examples of shell repair and further shell growth in damaged individuals
632 after 3 months (top row) and 7 months (bottom row) in the following four treatments:
633 pH control (a, e), temperature control (b, f), pH7.75 (c, g) and pH7.54 (d, h). The
634 same individual is shown in each treatment at both growth periods where the arrow
635 indicates the notch created at the start of the experiment. Scale bar = 100µm. (Colour
636 on Web only)

637 **Figure 2** – Growth rates of specimens that were left undamaged (●) and were
638 damaged at the start of the experiment (○) after 7 months in the treatments kept at
639 the higher temperature; (a) pH control (pH8.05 ± 0.03, T=1.7°C ± 0.3), (b) pH7.75
640 (pH7.75 ± 0.03, T=1.9°C ± 0.4) and (c) pH7.54 (pH7.54 ± 0.03, T=2.2°C ± 0.4).

641 **Figure 3** - Growth rates of specimens that were left undamaged (●) and were
642 damaged at the start of the experiment (○) after 7 months in the low temperature
643 control (pH7.98 ± 0.02, T=-0.3 °C ± 0.1).