



Estimates of tropical bromoform emissions using an inversion method

M. J. Ashfold^{1,*}, N. R. P. Harris¹, A. J. Manning², A. D. Robinson¹, N. J. Warwick^{1,3}, and J. A. Pyle^{1,3}

¹Department of Chemistry, University of Cambridge, Cambridge, UK

²Met Office, Exeter, UK

³National Centre for Atmospheric Science, UK

* now at: School of Biosciences, University of Nottingham Malaysia Campus, Jalan Broga, 43500 Semenyih, Selangor Darul Ehsan, Malaysia

Correspondence to: M. J. Ashfold (matthew.ashfold@nottingham.edu.my)

Received: 20 July 2013 – Published in Atmos. Chem. Phys. Discuss.: 6 August 2013

Revised: 21 November 2013 – Accepted: 10 December 2013 – Published: 28 January 2014

Abstract. Bromine plays an important role in ozone chemistry in both the troposphere and stratosphere. When measured by mass, bromoform (CHBr_3) is thought to be the largest organic source of bromine to the atmosphere. While seaweed and phytoplankton are known to be dominant sources, the size and the geographical distribution of CHBr_3 emissions remains uncertain. Particularly little is known about emissions from the Maritime Continent, which have usually been assumed to be large, and which appear to be especially likely to reach the stratosphere. In this study we aim to reduce this uncertainty by combining the first multi-annual set of CHBr_3 measurements from this region, and an inversion process, to investigate systematically the distribution and magnitude of CHBr_3 emissions. The novelty of our approach lies in the application of the inversion method to CHBr_3 . We find that local measurements of a short-lived gas like CHBr_3 can be used to constrain emissions from only a relatively small, sub-regional domain. We then obtain detailed estimates of CHBr_3 emissions within this area, which appear to be relatively insensitive to the assumptions inherent in the inversion process. We extrapolate this information to produce estimated emissions for the entire tropics (defined as 20°S – 20°N) of $225\text{ Gg CHBr}_3\text{ yr}^{-1}$. The ocean in the area we base our extrapolations upon is typically somewhat shallower, and more biologically productive, than the tropical average. Despite this, our tropical estimate is lower than most other recent studies, and suggests that CHBr_3 emissions in the coastline-rich Maritime Continent may not be stronger than emissions in other parts of the tropics.

1 Introduction

Chemical cycles involving inorganic bromine species (Br_y) are known to destroy appreciable quantities of ozone (O_3) in both the stratosphere (e.g. Salawitch et al., 2005) and troposphere (Yang et al., 2005; Read et al., 2008; Parrella et al., 2012). The largest source of Br_y , by mass, appears to be release from sea-salt aerosols in the marine boundary layer (Ayers et al., 1999; Yang et al., 2005; Parrella et al., 2012). The lifetime of Br_y in the lower troposphere is short enough that this direct inorganic source does not appear to significantly influence chemistry at higher altitudes (e.g. Yang et al., 2005). In contrast, a class of man-made organic bromine compounds, the halons, provide a significant source of Br_y to the stratosphere (e.g. Wamsley et al., 1998). There are also naturally occurring organic bromine compounds, which are known to be emitted by seaweed (e.g. Gschwend et al., 1985; Manley et al., 1992; Itoh and Shinya, 1994) and phytoplankton (e.g. Sturges et al., 1992; Tokarczyk and Moore, 1994; Moore et al., 1996). These compounds have shorter atmospheric lifetimes than their man-made counterparts, typically of weeks to months, and are often referred to as very short-lived substances (VSLs). They may be the dominant source of Br_y in the upper troposphere and lowermost stratosphere (Dvortsov et al., 1999; Salawitch et al., 2005). Bromoform (CHBr_3) and dibromomethane (CH_2Br_2) have received most attention, and appear to account for the bulk of bromine bound within VSLs (Hos-saini et al., 2012). A further organic source of bromine is

methyl bromide (CH_3Br), which has an intermediate lifetime of ~ 1 yr and a more diverse range of both natural and man-made sources (Saltzman et al., 2004). Overall, these sources combine to give a present stratospheric Br_y loading of 22.5 (19.5–24.5) ppt (Montzka and Reimann, 2011).

Anthropogenic organic bromine emissions, and associated global average mixing ratios, are thought to be reasonably well quantified (Montzka and Reimann, 2011). Conversely, there remains much uncertainty in our understanding of the magnitude and distribution of the predominantly natural emissions of VSLS. Broadly, two main methods have been used to arrive at estimates of global CHBr_3 fluxes. In the “bottom-up” method local flux calculations are assumed to be representative of, and are extrapolated over, particular oceanic domains (e.g. Quack and Wallace, 2003; Carpenter and Liss, 2000; Butler et al., 2007; Ziska et al., 2013). In the “top-down” method an idealised pattern of emissions is typically assumed based on available observational evidence and the emission magnitude is then varied to match observations made in the “background” atmosphere (e.g. Warwick et al., 2006; Liang et al., 2010; Ordóñez et al., 2012). Inverse methods, in which both the magnitude and distribution of emissions are varied to provide an optimal match with observations, can be considered an extension of the top-down method, and have yet to be tried. Our use of an inversion method in this study therefore represents a novel approach to estimating CHBr_3 emissions. Finally, observations of compounds with a common source but different lifetimes can be used to derive emission estimates (e.g. Yokouchi et al., 2005), and parameterisations of emissions based on physical and biological variables have begun to be considered (Palmer and Reason, 2009).

Estimates of CHBr_3 emissions are made difficult by its short lifetime (as low as 2 weeks in the tropics, e.g. Warwick et al., 2006; Hossaini et al., 2010), which ensures that measured mixing ratios are often highly variable, and are likely to be influenced by only a small fraction of global sources. This difficulty is reflected in the following range of recent global emission estimates, all given in $\text{Gg CHBr}_3 \text{ yr}^{-1}$: ~ 450 (Liang et al., 2010); ~ 380 (Warwick et al., 2006, updated by Pyle et al. 2011); ~ 530 (Ordóñez et al., 2012); ~ 120 – 200 (Ziska et al., 2013). When defined consistently as 20°S – 20°N , the total tropical emission in these studies also varies substantially (again in $\text{Gg CHBr}_3 \text{ yr}^{-1}$): ~ 240 (Liang et al., 2010); ~ 310 (Warwick et al., 2006, updated by Pyle et al. 2011); ~ 300 (Ordóñez et al., 2012); ~ 70 – 90 (Ziska et al., 2013). In contrast, recent global emission estimates of the longer-lived CH_2Br_2 (which has a lifetime of months, e.g. Warwick et al., 2006; Hossaini et al., 2010) have converged on a value of $\sim 60 \text{ Gg yr}^{-1}$ (e.g. Liang et al., 2010; Ordóñez et al., 2012; Ziska et al., 2013). One part of the world where CHBr_3 emissions are particularly poorly constrained is the Maritime Continent, where long-term observations have only recently become available (e.g. Pyle et al., 2011; Robinson et al., 2014). The Maritime Continent is important because

a number of studies have suggested that emissions from this convectively active region may be much more likely to contribute to stratospheric Br_y (e.g. Levine et al., 2007; Aschmann et al., 2009; Hosking et al., 2010; Pisso et al., 2010).

In this study we attempt to improve our understanding of CHBr_3 emissions in the Maritime Continent. To do this we use the first long-term set of observations collected in this region (described by Robinson et al., 2014) and, for the first time, an inversion method (Manning et al., 2003, 2011). To date, this method has been used to estimate emissions of long-lived greenhouse gases and ozone-depleting substances (see also O’Doherty et al., 2004; Reimann et al., 2005; Polson et al., 2011). As such, this work, focussed on measurements of a short-lived compound in the deep tropics, represents a new application of this method. We therefore aim to consider the limitations and additional uncertainties associated with the choice of this type of modelling tool for our particular purpose. The paper is structured as follows: in Sect. 2 we introduce the observations we use to constrain our model; in Sect. 3 we provide details of the modelling tools we employ; in Sect. 4 we show that our method can successfully locate a “known” emission distribution; in Sect. 5 we present our estimates of CHBr_3 emissions in the Maritime Continent; and finally, in Sect. 6, we compare our estimate with previous work, we discuss the limitations of our method, and we recommend ways this type of study can be improved in the future.

2 Bromoform observations

We use CHBr_3 observations which are presented in detail by Robinson et al. (2014). The main features are summarised again here. The measurements were collected using University of Cambridge μ -Dirac instruments (Gostlow et al., 2010) at two locations in Sabah, Malaysia, and cover the period April 2008–May 2010. One instrument was located ~ 50 km from the nearest coast, at the Bukit Atur Global Atmospheric Watch site (elevation 426 m a.s.l.), within the Danum Valley Conservation Area. The second instrument was located at the coast, near the town of Tawau (elevation 15 m a.s.l.), and ~ 85 km south of Bukit Atur. The characteristics of the two measurement locations, and the quality of our data, are discussed by Robinson et al. (2014). Figure 1 shows the CHBr_3 measurements, which have been averaged over 3 h intervals to allow direct comparison with our model calculations (see Sect. 3). The sampling rate of μ -Dirac means there are typically ~ 6 measurements in these 3 h windows. We will refer to these averaged data simply as “observations”. Our CHBr_3 observations are a major enhancement to the quantity of information available in the Maritime Continent, where until recently measurements had been collected only during occasional cruises through the region (e.g. Quack and Suess, 1999; Yokouchi et al., 1999). Following the period we consider in this study, further CHBr_3 data have

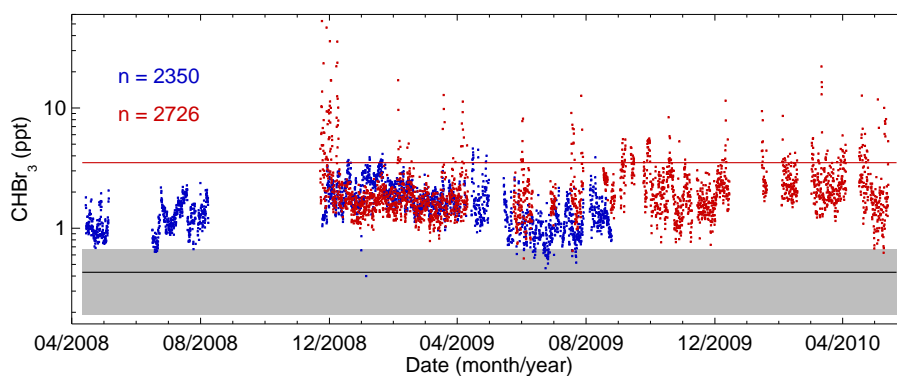


Fig. 1. Observations of CHBr_3 , averaged over 3 h periods, from two locations in northern Borneo for April 2008 to May 2010. The blue data are from Bukit Atur, a rainforest location, and the red data are from Tawau, a coastal site. The y axis is logarithmic. The solid black line marks the “baseline” mixing ratio of 0.43 ppt assumed in the majority of the inversion experiments (see Sect. 3.2 for details). The shaded grey region indicates one standard deviation (± 0.24 ppt) of the noise added to the observations (or equivalently the baseline, see Sect. 3.2.4). The solid red line, at 3.51 ppt, marks the 90th percentile of the coastal observations. The number of 3 h periods for which observations are available is noted to the top-left of the plot.

been collected during a cruise through the South China Sea (Mohd Nadzir et al., 2014) and near Borneo during the approximately month-long SHIVA campaign (e.g. the aircraft data presented by Hossaini et al., 2013).

The CHBr_3 mixing ratio at Tawau, on the coast, is occasionally many tens of ppt, with $\sim 10\%$ of the observations above ~ 3.5 ppt (a threshold marked by the red line in Fig. 1, and discussed with respect to our inversion calculations in Sect. 5.1). Additional quantitative information can be found in the probability density functions, PDFs, presented by Robinson et al. (2014). While Robinson et al. (2014) do not identify a specific local cause of these high mixing ratios, previous studies have typically found such levels of CHBr_3 when seaweed are nearby (e.g. Carpenter and Liss, 2000; Pyle et al., 2011). Aside from these periods, mixing ratios at Tawau are usually within the range 1–2 ppt. While there is often a small gradient in background mixing ratio between the coast and inland (again refer to the PDFs in Robinson et al., 2014), the majority of our inland data, and indeed many other observations apparently made away from strong CHBr_3 sources (see Table 1–7 of Montzka and Reimann, 2011), also lie within this range.

Our data are used here to inform an inversion process, and so it is important to consider how observational uncertainties might impact our estimates of CHBr_3 emissions. First, calibration errors might lead to biases in our observations, and in our subsequent emission estimates. A recent measurement inter-comparison (Jones et al., 2011) suggests that for CHBr_3 , discrepancies of ~ 10 – 20% are possible. As this uncertainty is likely to manifest itself as a bias, we attempt to assess its potential significance by repeating our inversion calculations with a range of “baseline” mixing ratios (see Sect. 3.2.2).

Another metric for measurement uncertainty is the precision, which we define as the standard deviation of calibration peak heights divided by the mean height of such peaks (see also Robinson et al., 2014). Though the measurement precision varied throughout the period covered, and was not identical at the two locations, we use 7% (i.e. 0.14 ppt, where the multi-site mean mixing ratio is 1.96 ppt) as a value representative of this type of uncertainty. The measurement precision uncertainty is likely to be smaller than the uncertainty attached to the model calculations (Sect. 3.1) and to the assumptions inherent in the inversion method (Sect. 3.2), and is therefore incorporated into an overall uncertainty which is implemented as “noise” within the inversion process (see Sect. 3.2.4).

3 Modelling tools

In this section we first describe the atmospheric transport model we have used to calculate the immediate history of the air masses measured in Borneo. Section 3.2 introduces the associated inversion method used to estimate CHBr_3 emissions, and discusses the assumptions and approximations this method requires.

3.1 NAME trajectory calculations

To interpret our measurements we have used the UK Met Office’s Lagrangian atmospheric dispersion model, NAME (Jones et al., 2007). NAME allows the calculation of trajectories using three-dimensional meteorological fields, which are produced by the UK Met Office’s Unified Model (UM) during operational weather forecasts, and are available at 3 hourly intervals. For most of our observational period they have a horizontal resolution of 0.5625° longitude by 0.375°

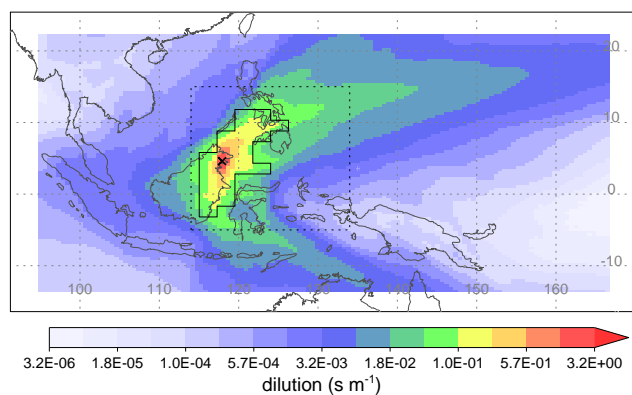


Fig. 2. The mean dilution matrix, averaged over backward trajectories released from both measurement locations, and over all 3 h periods for which observations are available at a particular measurement location. The black cross marks the average of the two proximate measurement locations. The grid covers $\sim 95\text{--}165^\circ\text{E}$, $13.5^\circ\text{S}\text{--}22^\circ\text{N}$ (128 by 96 grid cells). The units are s m^{-1} and the colour scale is logarithmic. The solid black contour encloses the “fine” cells in the solution grid for which emission estimates are subsequently given. The dotted black box highlights the region plotted in Fig. 5 and 6.

latitude, though this was improved to $\sim 0.35^\circ$ by $\sim 0.23^\circ$ in February 2010. Within NAME a parameterisation of turbulence is also used (see Webster et al., 2003; Morrison and Webster, 2005).

NAME was used to calculate batches of 33 000 backward trajectories, released at random throughout each 3 h period for which CHBr_3 measurements were available at a particular location. In our measurement window there were a total of 2350 such periods at Bukit Atur, and 2726 at Tawau (see also Fig. 1). The trajectories were started randomly within an altitude range of 0–100 m. They ran for 12 days, and every 15 min the location of all trajectories within the lowest 100 m of the model atmosphere was recorded on a grid with the same resolution as the driving meteorological fields (always 0.5625° by 0.375°). By counting only the “near-surface” trajectories a picture was built up of the areas from which surface emissions could have influenced the measured air masses during the previous 12 days.

In more detail, each trajectory is assigned an arbitrary mass, and the density of trajectories in each grid cell, integrated over the 12 day travel time (units are g s m^{-3}), is recorded. This density is then divided by the total mass of trajectories released (g), and multiplied by the area of the grid cell (m^2). Physically, the resulting quantity (which has units of s m^{-1}) can be considered the mean residence time in a particular grid cell, for all trajectories started in a particular 3 h period, divided by the cell depth (100 m). Importantly, this quantity is the multiplicative factor by which emissions in a particular grid cell are diluted by the time they have travelled to a measurement location (see Eq. (1), and Manning

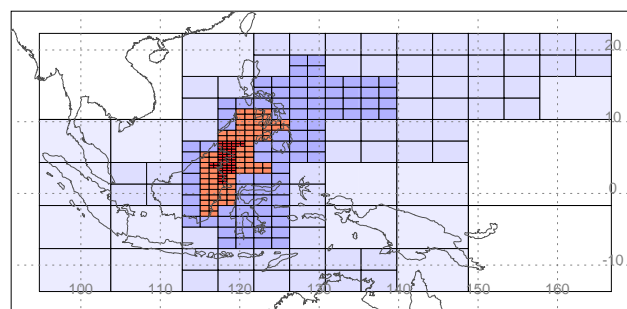


Fig. 3. The solution grid, which contains cells of various sizes, and covers the same area as the dilution grid (Fig. 2). The “fine” grid cells (of size 1 by 1 and 2 by 2 dilution grid cells) are coloured in red/orange.

et al., 2011). So, a low dilution value means that few trajectories reaching Bukit Atur or Tawau have travelled through the boundary layer of a particular grid cell, and therefore that the impact of emissions from that grid cell on a measurement site will be correspondingly small. A grid of dilution values exists for the batch of trajectories started in each 3 h period. We will refer to the full three-dimensional (longitude-latitude-time) grid of dilution values as the “dilution matrix”. We approximate the effect of photochemical loss of CHBr_3 by allowing the mass associated with each trajectory to decay with an e-folding time of 15 days (similar to the expected lifetime of CHBr_3 in the tropical lower troposphere, Warwick et al., 2006; Hossaini et al., 2010; Liang et al., 2010). This has the effect of decreasing the dilution value in grid cells further from the measurement locations.

Figure 2 shows the time-averaged dilution matrix, calculated over all 3 h periods for which observations are available at either location. It shows that the sites in Borneo are influenced by two prevailing wind directions, with northeasterly trade winds dominating during Northern Hemisphere winter, and southeasterlies during Northern Hemisphere summer. It also shows that large fractions of the plotted domain are crossed by relatively few trajectories (dilution values range over six orders of magnitude), and emissions in these regions are unlikely to appreciably affect our measurements. One consequence is that our emission estimates are made on a non-uniform grid, with larger grid cells used where the dilution values are smaller (see Fig. 3 and the discussion later in Sect. 3.2.1).

Our emission estimates will be affected by any errors in the meteorological data used to drive NAME, and in the dispersion calculations within NAME. The estimates of Manning et al. (2011) were relatively insensitive to the choice of meteorology, though their study was focussed on mid-latitudes. In the tropics, the representation of convection will be more important, and will be necessarily incomplete in any set of meteorological analyses. In addition, the local meteorology at both Bukit Atur, where changes in boundary layer

height appear to be critical to determining atmospheric composition (e.g. Pike et al., 2010), and Tawau, where a “sea breeze” circulation is common (Qian et al., 2013), may not be represented accurately in a global model such as the UM. Finally, the rate at which air masses (i.e. trajectories) are mixed between the boundary layer and free troposphere in NAME will have a significant impact on our estimated emissions. Pyle et al. (2011) showed, and we reiterate in Sect. 4, that in the Maritime Continent this type of mixing seems to be weaker in NAME than in another model, p-TOMCAT. We attempt to assess the possible impact of these model-related uncertainties both by adding “noise” to our inversion process, and by adjusting the “baseline” CHBr₃ mixing ratio (see Sect. 3.2.4).

3.2 Inversion method

The inversion method employed here has been described by Manning et al. (2003, 2011), and is now known as the Inversion Technique for Emission Modelling (InTEM). Using an optimisation process, the aim is to solve Eq. (1), and thereby locate the emission map, that can be multiplied by a NAME dilution matrix, to give modelled concentrations at a measurement site that most closely resemble the observed concentrations (converted from volume mixing ratios using UM meteorological data).

$$\begin{aligned} \text{emission (g m}^{-2} \text{ s}^{-1}) \times \text{dilution (s m}^{-1}) & \quad (1) \\ & = \text{concentration (g m}^{-3}) \end{aligned}$$

This calculation is conducted for each individual (source) grid cell: the emission in that grid cell, multiplied by the local value of the dilution matrix derived from the trajectory calculations, gives the contribution from those emissions to the simulated CHBr₃ concentration at a measurement location (receptor). The sum over all source grid cell contributions gives the total simulated CHBr₃ concentration at one particular time.

From Eq. (1) we obtain a spatially varying emission field. In solving this equation we assume the emissions are constant in time; clearly, when considering a compound with large natural emissions such as CHBr₃, this is likely to be a source of error in our estimates. In this section we discuss some of the other assumptions that are implicit in the method, which has typically been used to estimate emissions of compounds with lifetimes of years. So we also aim to highlight further uncertainties that are peculiar to our compound of interest, CHBr₃, with its much shorter lifetime.

3.2.1 The solution grid

The time-averaged dilution matrix in Fig. 2 shows that air is most likely to travel towards our measurement locations from either the southeast or northeast, and more likely to travel over grid cells near to a measurement location than those further away. It is difficult to assess how the observations will be

affected by emissions from a particular grid cell if air seldom passes over it. Such grid cells are therefore grouped together using the method of Manning et al. (2011), into progressively larger squares of 2 by 2 cells, 4 by 4 cells and so on. As a result, the contribution to the observations from emissions in the new, grouped grid cell is larger or more frequent. This solution grid, containing grouped cells, is shown in Fig. 3, and is used for each of the inversion experiments in this study. We tested changing the resolution of the solution grid but found that such changes had a small impact on the results we present in Sect. 5.

We can use the dilution matrix to assess the influence that emissions in solution grid cells of particular sizes might have on our measurements. If we assume that a tracer is emitted uniformly across our region of interest the dilution matrix can be used to derive a simulated concentration at both measurement sites (Eq. 1). Figure 4 shows the cumulative contribution to this modelled concentration when the individual dilution grid cells in Fig. 2 are added in turn (beginning with the largest dilution value). The red line (and label) shows that 44 % of this modelled concentration is due to emissions in solution grid cells of size 1 by 1, and that these cells cover an area of $1.65 \times 10^{11} \text{ m}^2$ ($\sim 0.1\%$ of the tropics). The orange (2 by 2), blue (4 by 4) and grey (8 by 8 to 32 by 32) lines indicate this relationship when solution grid cells of increasing size are also included in the calculation.

While the inversion process always returns estimated emissions for the entire solution domain, throughout the rest of the paper we will present emissions only in the smaller grid cells, of size 1 by 1 and 2 by 2 only (coloured in red and orange in Fig. 3). We refer to these as “fine” grid cells. We make this choice because they are the grid cells for which most dilution information is available and from which emissions are likely to have the greatest impact on the observations. Conversely, emissions from the coarser solution grid cells can vary significantly while having only a small impact on the observations, and will therefore be less reliably constrained by the inversion method. Figure 4 shows that the “fine” cells account for almost two thirds of the observed signal from a uniform emission across our solution grid, while accounting for only $\sim 3\%$ of the area. This demonstrates that the compromise between considering emissions for which there is reasonable dilution information, and estimating emissions over a usefully-sized area, is a delicate one. We have repeated our analyses using only the 1 by 1 cells, and using all cells up to size 4 by 4 (the areas covered by these different grids, along with estimates of emissions in one of the subsequent inversion experiments, are given in Table 1), and will occasionally refer to these results. Overall though, we find that our choice of “fine” grid does not significantly affect our conclusions.

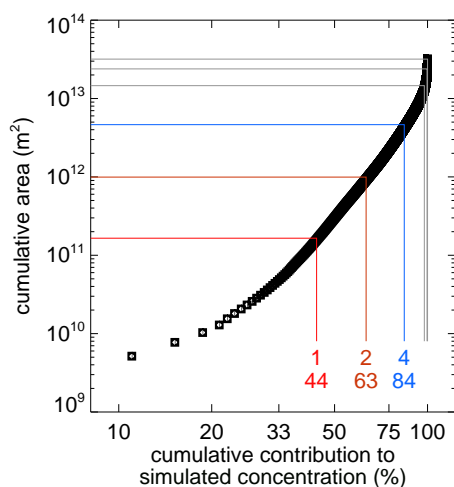


Fig. 4. Each black square represents an individual grid cell in the dilution matrix presented in Fig. 2 and, assuming a uniform emission throughout that area, shows the cumulative percentage of a simulated concentration at both measurement sites that is accounted for by adding emissions from that grid cell. The grid cells are counted successively, beginning with the grid cell containing the highest dilution value and finishing with the grid cell containing the lowest. This accumulated percentage is plotted against the surface area covered by the increasing number of grid cells considered. The lines relate to the different sized grid cells in the inversion solution grid. The red line (and label) shows that 44 % of the hypothetical modelled concentration is caused by emissions in grid cells of size 1 by 1, which cover an area of $\sim 1.6 \times 10^{11} \text{ m}^2$. The orange (2 by 2), blue (4 by 4) and grey (8 by 8 to 32 by 32) lines indicate this relationship when solution grid cells of increasing size are also included in the calculation. For reference, the surface area between 20° S – 20° N is $1.75 \times 10^{14} \text{ m}^2$.

Table 1. The area (in m^2) occupied when different maximum solution grid cell sizes (see also Figs. 3 and 4) are considered, and the percentage of the tropics (20° S – 20° N) this area covers. Then, for experiment D, the total emission from the selected grid (E_{grid}), and that emission extrapolated across the tropics (E_{trop}). Emission units are $\text{Gg CHBr}_3 \text{ yr}^{-1}$.

Max. size	Area	%trop	E_{grid}	E_{trop}
1 by 1	1.65×10^{11}	0.09 %	0.17	180
2 by 2	9.94×10^{11}	0.57 %	1.28	225
4 by 4	4.64×10^{12}	2.65 %	6.39	240

3.2.2 The “baseline”

Our backward trajectories can only be used to account for emissions occurring within the last 12 days. The contribution from earlier emissions therefore needs to be estimated, and subtracted from the observations. In the case of long-lived gases like CFCs, well-defined “baseline” mixing ratios can be observed at remote, unpolluted locations. Subtracting the

baseline mixing ratio from such observations should therefore leave only pollution episodes caused by recent emissions. Here, though, where the compound of interest is short-lived, the concept of a baseline is less applicable. In addition, and in contrast to gases such as CFCs which have geographically discrete sources, CHBr_3 is thought to be emitted over much of the ocean (e.g. Butler et al., 2007). Therefore, in the Maritime Continent, air masses travelling in any direction towards the measurement locations may be subject to CHBr_3 emissions. With a lifetime of ~ 15 days though, it is also likely there is some contribution to the CHBr_3 observations from emissions that occur more than 12 days previously, and that are therefore outside our solution grid.

In our experiments we use a constant baseline. This is a simplification, but there is no clear reason to select particular air masses that might enable the construction of a time-varying baseline. In addition, our results are similar if a seasonally varying baseline is employed (not shown). In the majority of the inversions we use a baseline mixing ratio of 0.43 ppt. This is the mean contribution from “rest of world” sources (i.e. those outside of a Southeast Asian domain of similar size and location to our solution grid) to the modelled CHBr_3 mixing ratio at Bukit Atur in a multi-year p-TOMCAT simulation containing Pyle et al. (2011) emissions. Our choice of baseline mixing ratio is also close to the minimum observed value (0.40 ppt, see Fig. 1) so is consistent with an alternative definition based on the CHBr_3 measurements. To examine how sensitive our emission estimates are to this choice, we conduct additional experiments in which the baseline is shifted up/down by one standard deviation of the p-TOMCAT “rest of world” contribution to Bukit Atur (± 0.19 ppt). These additional experiments also allow us to indirectly assess the possible impact of biases (e.g. in measurement calibration, or in the rate of modelled boundary layer to free troposphere mixing) on our estimated emissions.

3.2.3 Optimisation

The aim of the optimisation process, which is outlined in this section, is to find the emission distribution that provides the “best” match between the modelled time series, calculated using Eq. (1), and the observed time series. We use the normalised mean square error (NMSE, Eq. 2) as a cost function to judge the strength of the agreement between the simulated concentrations (m) and the three hourly average observations (o), and it is this error that the optimisation process will seek to minimise. In Eq. (2), n is the sum of the number of observations at both measurement locations ($2350 + 2726 = 5076$, see Fig. 1).

$$\text{NMSE} = \frac{1}{n} \sum_{i=1}^n \frac{(m_i - o_i)^2}{\bar{m} \bar{o}} \quad (2)$$

Our cost function could have many configurations, so we acknowledge that this choice may be a source of uncertainty.

However, we obtain similar results if we use either the root-mean-square error or a cost function similar to that employed by Manning et al. (2011) with a mix of statistical measures.

If something is thought to be known about an emission distribution (for example, Stohl et al. (2009) assume zero oceanic emissions of selected anthropogenic halocarbons) then this knowledge can be incorporated into the cost function. Again, the weighting (equivalently the confidence) assigned to this so-called a priori information is subjective. As Manning et al. (2011) aim to generate an emission estimate that is entirely independent of “bottom-up” inventories, they do not use any a priori information. We believe that the characteristics of CHBr_3 emissions are not known sufficiently well to warrant the use of a priori information, with the exception that there is little evidence for terrestrial sources. As such, rather than altering the cost function, our sole use of a priori information is to fix land emissions to be zero in some of our experiments.

The optimisation method used to minimise the cost function is “simulated annealing”, so-called because of its conceptual similarity to annealing in metallurgy. The method is described in detail by both Kirkpatrick et al. (1983) and Press et al. (1986), and its application here is described by Manning et al. (2011). This particular approach is useful when, as is the case here, there are so many possible solutions to a problem that an exhaustive search for the optimum solution (i.e. the minimum cost) is impractical. The method therefore aims to find a solution that can not be improved upon significantly while using only limited computing time. In brief, the method investigates a solution space of possible emission configurations. It iteratively searches the solution space to find the emission map that, after transformation to a simulated concentration using the dilution matrix, most closely matches (as measured by Eq. 2) the observed time-series. We repeat each inversion experiment independently 25 times to obtain a range of emission estimates, with different solutions emerging due to the different noise added to the observations (Sect. 3.2.4) in each of the 25 repetitions.

3.2.4 “Noise” and uncertainty

As suggested in Sect. 2, there is a range of sources of uncertainty in the emission estimates based on Eq. (1). While we are able to quantify the uncertainty related to measurement precision with reasonable confidence (Sect. 2), the uncertainty attached to the model calculations and to the assumptions inherent in the inversion method is more difficult to ascertain (see also Manning, 2011). In an attempt to account for these uncertainties a time series of “noise” (selected at random from a normal distribution with a mean of zero, and with a specified standard deviation) is added to the observations (or equivalently, to the baseline which is subtracted from the observations, see Fig. 1).

For the inversion experiments presented in Sect. 5 we use a standard deviation of ± 0.24 ppt for the normally distributed

random numbers we add to the observations. This is the sum of two normal distributions: the first, with a standard deviation of 0.14 ppt, represents measurement precision uncertainty (see Sect. 2), and the second, with a standard deviation of 0.19 ppt, represents baseline uncertainty (Sect. 3.2.2). We have tried other forms of noise, including uniform distributions, and different magnitudes, but find these changes have only a minor impact on our central emission estimates, and therefore on our conclusions. To reiterate a point made in Sect. 3.2.2, we also assess the importance of biases in both observations and modelling by adjusting the baseline mixing ratio.

3.2.5 Outline of inversion experiments

In Sect. 5 we will present results from six inversion experiments. The experiments are labelled A–F, and are summarised in Table 2. In experiments A–C we make use of different subsets of the observations shown in Fig. 1, which enables us to explore the usefulness of observations with different characteristics when used in this inversion method. In experiments D–F we consider the use of a priori information (i.e. fixing land emissions to zero), and vary the baseline to assess the possible impact of biases within our method.

In each experiment we use the dilution matrix described in Sect. 3.1, which is based on trajectories calculated over the entire measurement period. Recall that the mass associated with these trajectories decays with a CHBr_3 -like 15 day e-folding lifetime. We calculate emissions for the whole solution grid shown in Fig. 3, but will present results for only the “fine” cells, or size 1 by 1 and 2 by 2, in this grid. We use a constant baseline mixing ratio (typically 0.43 ppt, but shifted up and down in experiments E and F) and, in an attempt to account for our uncertainties, include normally distributed random noise with a standard deviation of 0.24 ppt. The cost function is given in Eq. (2), and each experiment is repeated with a different application of noise 25 times.

4 Pseudo observations

Before using our observations to estimate emissions, we test the ability of our selected inversion method to find a “known” emission distribution. We generate “pseudo observations” using the dilution matrix and a “known” emission distribution (dimensionally, $\text{s m}^{-1} \times \text{g m}^{-2} \text{s}^{-1} = \text{g m}^{-3}$). Then, given only these pseudo observations and the dilution matrix, the inversion system should be able to iterate towards emissions close to those originally prescribed (e.g. Fig. 9 of Manning et al., 2011).

For our “known” emission distribution we use a re-gridded version of the CHBr_3 emissions reported by Pyle et al. (2011) as an update to scenario 5 of Warwick et al. (2006) (see Fig. 5a). Pseudo observations for both Bukit Atur and Tawau, for each 3 h period in 2008, are then generated. The solution

Table 2. A summary of various parameters used in, and results from, inversion experiments A–F. Observational constraint is provided by data collected at the inland site, Bukit Atur (BA), and the coastal location, Tawau (T). The coastal data is sometimes capped at the 90th percentile (T_{90}). The baseline mixing ratio (units of ppt), and whether an a priori condition of zero emissions from land is employed, is also noted for each experiment. The table also summarises the CHBr_3 emission estimated by each experiment, which are discussed in Sect. 5. Emissions are listed for the “fine” solution grid cells (inside the solid black contour in Fig. 2, and in red/orange in Fig. 3). The grid cell-by-grid cell mean emission magnitude is given, with the range of the 25 separate solutions in parentheses. Emission units are $\text{Gg CHBr}_3 \text{ yr}^{-1}$. The percentage of the total emission from land (La), coast (Co) and ocean (Oc) grid cells is also noted, with surface definitions taken from the UM land/sea mask. Finally, the cost function score (Eq. 2) for each experiment is noted.

Exp.	Obs.	Base.	a priori	Emission	%La	%Co	%Oc	NMSE
A	BA	0.43	No	1.3 (1.2–1.4)	12	16	72	0.24
B	BA,T	0.43	No	2.2 (2.0–2.3)	23	26	51	1.05
C	BA, T_{90}	0.43	No	1.4 (1.3–1.5)	17	21	62	0.26
D	BA, T_{90}	0.43	Yes	1.3 (1.1–1.4)	8	25	67	0.26
E	BA, T_{90}	0.24	Yes	1.4 (1.3–1.6)	9	26	65	0.21
F	BA, T_{90}	0.62	Yes	1.1 (1.0–1.2)	8	25	67	0.34

grid was identical to that derived from the real observations (Fig. 3), and a version of our “known” emissions, degraded to this non-uniform grid, is presented in Fig. 5b. Emissions in only the “fine” grid cells are shown in Fig. 5c (i.e. all larger grid cells are removed).

The average of 25 separate inversion solutions is shown in Fig. 5d (where the mean is calculated grid cell-by-grid cell), with only the “fine” grid cells shown in Fig. 5e. The statistical agreement between the pseudo observations, which were not adjusted with noise, and the inversion-derived set of observations is excellent ($\text{NMSE} = 0.0002$). If we use the degraded emissions (Fig. 5b) to generate the pseudo observations, so that an exact solution is now possible, the statistical agreement is even stronger. The method is also able to quantify accurately the magnitude of regional emissions (the total emission from the plotted grid cells is given above each plot in Fig. 5).

Figure 5 also shows that our method is able to find nearby coastal emissions (that might be expected for CHBr_3) if they exist at our grid resolution, and if there is a clear signal in the observations. However, in more distant parts of the region, where the solution grid is coarser, coastal grid boxes (and emissions) cannot be separately resolved. We wish to retain information about the split between land, coast and ocean in our estimated emissions ahead of extrapolation to global scales in Sect. 5, so this problem provides additional motivation for including only smaller grid cells in our definition of “fine” (see Sect. 3.2.1).

Consistent with the comparison of Pyle et al. (2011), these emissions developed for use in p-TOMCAT lead NAME to overestimate CHBr_3 mixing ratios at both Bukit Atur (observed mean, 1.55 ppt, and modelled mean, 2.04 ppt) and Tawau (2.32 ppt and 2.75 ppt). Pyle et al. (2011) reported a larger difference, and highlighted a discrepancy in the rate at which the boundary layer is vented in the two models, with this process less efficient in NAME. We discuss this type of model-to-model difference further in Sect. 6, but it

is certainly a significant, and difficult to quantify, source of uncertainty in the emission estimates we present in the next section.

5 Emission estimates

5.1 Local emissions

In this section we present emission estimates from six different inversion experiments, which are labelled A–F, and are summarised in Table 2. In Sect. 3.2 we discussed choices and assumptions that need to be made in the inversion method we employ; in general the results we present here are relatively insensitive to these assumptions.

In experiment A only the observations from Bukit Atur, our inland measurement location, are used. As noted in Sect. 2, these data do not appear to have been influenced by nearby sources and therefore might be most suitable for estimating regional-scale emissions. As with the subsequent experiments, the mean emission distribution in only the “fine” solution grid cells is shown in Fig. 6. The estimated emission magnitude from these “fine” cells is $1.3 \text{ Gg CHBr}_3 \text{ yr}^{-1}$, with the 25 individual solutions ranging between $1.2\text{--}1.4 \text{ Gg CHBr}_3 \text{ yr}^{-1}$. The individual solutions to experiment A containing the lowest and highest emission are shown in Fig. 7a and 7b; while the grid cell by grid cell details of the distribution of emissions in these two solutions are somewhat different, the main features are consistent, with emissions found largely in the oceanic parts of the map, and much smaller over land.

Using the emissions from experiment A and the dilution matrix we now solve Eq. (1) to obtain simulated CHBr_3 concentrations at Bukit Atur. This simulated timeseries is presented in Fig. 8a, along with the measured timeseries. Recall that it is the difference between these timeseries that the inversion process tries to minimise, and here, they are in reasonable statistical agreement ($\text{NMSE} = 0.24$). The

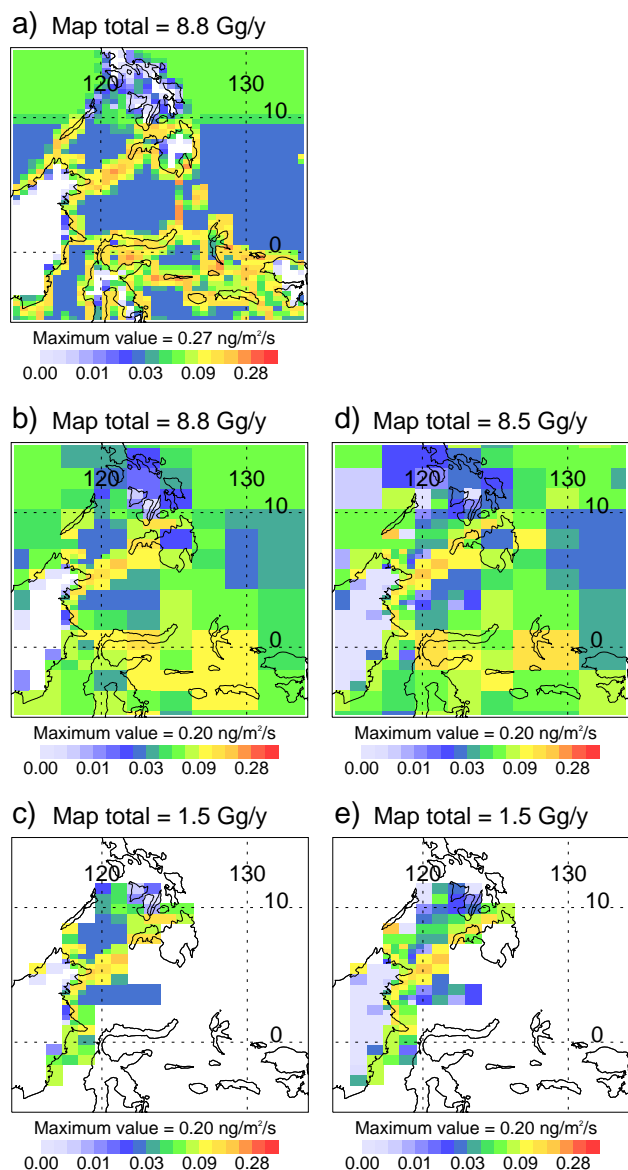


Fig. 5. (a) Shows CHBr_3 emissions according to the Pyle et al. (2011) update to scenario 5 in Warwick et al. (2006) after re-gridding to match NAME's grid. These emissions are used to create pseudo observations. (b) Shows these emissions degraded to the resolution of the solution grid (see Fig. 3), and (c) shows these degraded emissions when only the "fine" solution grid cells are included. For comparison, (d) shows the mean solution of 25 inversions which use pseudo observations derived from the emissions in (a), and (e) shows this mean solution when only the "fine" grid cells are included. The total emission from the plotted grid cells is noted above each plot (units are $\text{Gg CHBr}_3 \text{ yr}^{-1}$), and the colour scale is exponential.

figure shows that, while the inversion-derived model time-series does not capture all the observed high frequency variability, the monthly mean mixing ratios are in good agreement. Further, Fig. 8b demonstrates that the overall observed

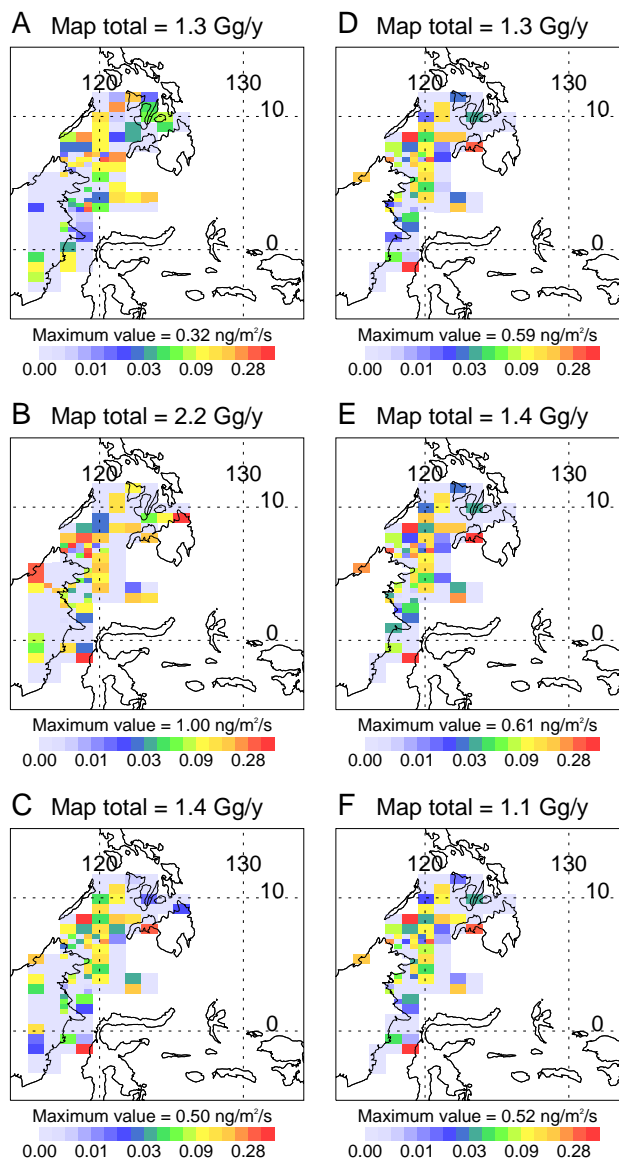


Fig. 6. Estimated emissions of CHBr_3 , for the "fine" grid cells (shades of red in Fig. 3). Experiment labels (see Table 2) are to the top-left of the plots, and in each case the mean of 25 separate solutions to the inversion process is shown. The total emission from the "fine" solution grid cells is noted above the plot (units are $\text{Gg CHBr}_3 \text{ yr}^{-1}$), and the colour scale is exponential.

mean mixing ratio, 1.55 ppt, is close to the modelled value, 1.64 ppt, and that the two time-series contain similar characteristics of variability.

We have used the UM land/sea mask, which has the same resolution as the meteorological analyses (0.5625° by 0.375°), to divide the "fine" grid cells into land, coast (ocean next to land) and ocean (excluding coast). These three categories cover, respectively, $\sim 32\%$, $\sim 22\%$ and $\sim 45\%$ of the selected area. Most emissions (72%, see Table 2) are placed in ocean grid cells, which is encouraging given the known

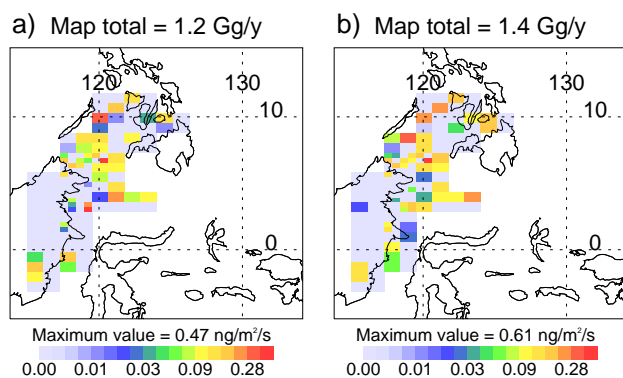


Fig. 7. For experiment A, considering only the “fine” grid cells: (a) shows the individual inversion solution (one of 25) with the lowest estimated CHBr₃ emission, and (b) shows the individual solution with the highest estimated CHBr₃ emission. As in Fig. 6, the total emission from the “fine” solution grid cells is noted above the plot (units are Gg CHBr₃ yr⁻¹).

oceanic sources of CHBr₃. However, emissions from coastal grid cells are relatively small (16%); as seaweed are thought to be a major CHBr₃ source in this region (Pyle et al., 2011) this finding is perhaps unexpected.

Of course, the observations collected at Bukit Atur and Tawau are not independent of one another; they are the result of the same distribution of CHBr₃ emissions. So in experiment B the observations from both sites are used in the same inversion. The total emission from the “fine” solution grid cells increases by ~70% to 2.2 (2.0–2.3 Gg CHBr₃ yr⁻¹) when we include these coastal data with, on average, higher mixing ratios. The distribution of emissions is also somewhat different, with a larger percentage of emissions found both along coastlines (26%, again see Table 2), which, given known CHBr₃ sources, we might expect, and inland areas (23%), which we might not. The statistical agreement between the observations and the modelled concentration is weaker (NMSE = 1.05) than in experiment A. This difference is largely caused by the occasional very high mixing ratios at the coast, which are presumably strongly influenced by emissions occurring very close to the Tawau site. These emissions will not be diluted according to the local value of the dilution matrix, and will therefore be difficult for the inversion method to account for.

In experiment C we attempt to remove the impact of these very local emissions. This is done by capping the coastal observations at the 90th percentile (3.51 ppt, marked with a red line in Fig. 1). All higher mixing ratios are set to this threshold. We accept that this is a crude approximation, but in support of our choice, we note that the 90th percentile of the coastal observations is very similar to the 99th percentile mixing ratio observed inland at Bukit Atur (3.55 ppt), where we assume there are no local sources. We also repeated this experiment using the 80th (2.65 ppt) and 95th

(4.75 ppt) percentiles as our threshold and found only small (~10%) differences in the total estimated emission magnitude. The total emission from the “fine” grid cells is now 1.4 (1.3–1.5) Gg CHBr₃ yr⁻¹, slightly higher than in experiment A, but lower than experiment B where the full range of coastal observations was included. Similarly, the distribution of emissions lies between experiments A and B, with 17% from land, 21% from coastal grid cells and 62% from the ocean (Table 2). The statistical agreement is almost as strong as in experiment A (NMSE = 0.26), confirming that the highest CHBr₃ mixing ratios at the coast lead to the poorer statistical agreement in experiment B.

In experiment D we repeat experiment C, but enforce an a priori condition of zero emissions from solution grid cells that contain only land. The inversion process remains free to attribute emissions to grid cells containing a mix of land and ocean. As CHBr₃ emissions are thought to be dominated by oceanic sources it is encouraging that, according to our cost function, the solutions to experiments C and D are of equal quality (NMSE = 0.26). Emissions from the “fine” solution grid cells total 1.3 (1.1–1.4) Gg CHBr₃ yr⁻¹, a little lower than without the “no land” constraint, and are now mostly distributed between the coast (25%) and ocean (67%). The remaining land emissions are placed in solution grid cells which contain both land and ocean, and so in reality are likely to be coastal emissions. In a similar way, some of the emissions attributed to land in experiments A, B and C are likely, in reality, to be due to coastal sources.

In the final pair of experiments, we repeat experiment D twice with a different baseline mixing ratio. In experiment E the baseline is lower than our best estimate, at 0.24 ppt. In F it is higher, at 0.62 ppt. As noted at the end of Sect. 3.2.2, these experiments allow us to assess the possible impact of biases, such as calibration errors, on our estimated emissions. As might be expected, the total emission from the “fine” grid cells in experiment D falls between the total with a lower baseline, 1.4 (1.3–1.6) Gg CHBr₃ yr⁻¹, and the total with a higher baseline, 1.1 (1.0–1.2) Gg CHBr₃ yr⁻¹. The split of emissions between land, coast and ocean is very similar in each of experiments D, E and F (Table 2, Fig. 6). The mean square error is also similar (within ~1%), with the differences in NMSE (Table 2) arising from the normalisation, which uses the mean modelled and mean observed mixing ratios after the baseline has been subtracted.

So, in all experiments but B, both the magnitude and the distribution of estimated emissions are similar. We feel that the decision to exclude the very high coastal mixing ratios included in experiment B is justified, and that our results are therefore robust to the range of reasonable sensitivities we have explored. In the following section we extrapolate these regional emissions to a global scale.

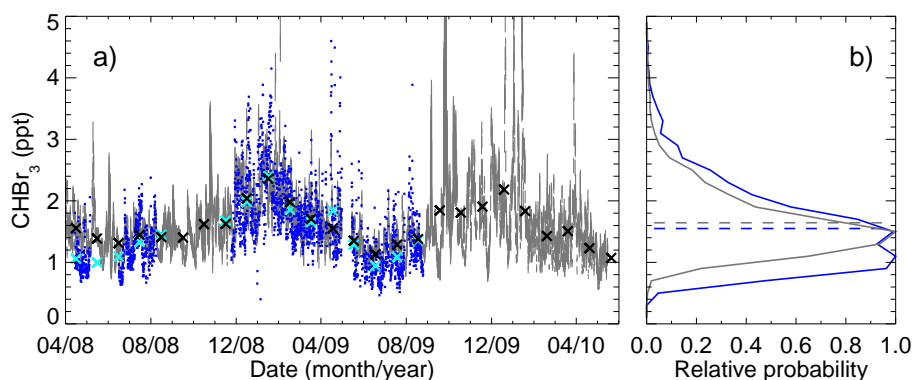


Fig. 8. (a) Shows both observed (blue) and modelled (grey) CHBr_3 timeseries at Bukit Atur. The model data are presented as a range, between the smallest and largest mixing ratio simulated in a particular 3 h window using emissions from each of the 25 individual solutions to experiment A. Monthly mean observed (light blue crosses) and modelled (black crosses) mixing ratios are also presented. (b) Shows the probability density function for each of these timeseries, normalised using the respective maximum probabilities, and with mean mixing ratios indicated by the dashed lines.

5.2 Extrapolated emissions

We have focussed so far on one part of the Maritime Continent, a region where transport from the troposphere to stratosphere appears to be particularly efficient. However, it is also useful to scale our estimate across a larger area to enable a comparison with other global-scale estimates of CHBr_3 emissions. Rather than extrapolating globally we wish to focus on the tropics (defined as 20°S – 20°N), from which emissions are most likely to have an impact on stratospheric ozone, and for which our measurements may be most representative.

In our first extrapolation, we simply assume that emissions are uniform across the entire tropics, and extrapolate the “fine” emissions from the mean solution in experiment D using the ratio of surface areas. We include land in this calculation, but would obtain a similar answer using only the oceanic areas because, according to our land/sea mask, land comprises a similar percentage of these two domains: 27 % of the entire surface area between 20°S – 20°N , and 32 % of our “fine” grid. This results in a best estimate of a source strength for the global tropics of $225\text{ Gg CHBr}_3\text{ yr}^{-1}$. We choose experiment D because we believe that experiments A and B do not make the best use of our observations, with A ignoring the coastal data altogether, and B being too strongly influenced by emissions very close to our coastal instrument. We also prefer the assumption of zero land emissions (i.e. experiment D rather than C) because it is supported by the majority of observational evidence, and because we are able to obtain equally strong statistical agreement between model and measurement with or without this condition.

Next, we can examine how certain choices in our method affect our extrapolated emission magnitude. To begin, we use the estimates from experiment E (with a lower baseline) and

from experiment F (with a higher baseline) to obtain a possible emission range related to the uncertainties in our inversion method. Extrapolating the maximum emission from the 25 individual solutions to experiment E and the minimum from experiment F yields a relatively narrow range of 180 – $280\text{ Gg CHBr}_3\text{ yr}^{-1}$. We can also consider the extent to which our definition of “fine” grid cells impacts upon our extrapolation. If “fine” were defined with a maximum grid size of 1 by 1 then extrapolation gives a total tropical emission of $180\text{ Gg CHBr}_3\text{ yr}^{-1}$. Conversely, if the maximum grid cell size were 4 by 4 the total tropical emission after extrapolation is $240\text{ Gg CHBr}_3\text{ yr}^{-1}$. These values are summarised in Table 1. They also fall within the range given by experiments E and F, and show that our choice of “fine” grid has had a relatively small impact on our extrapolated emission magnitude.

As well as emission magnitudes, we also try to quantify the split between oceanic and coastal emissions. We continue to use the UM land/sea mask to determine which grid cells fall in to a particular category, focus on our central estimate, experiment D, and re-assign all land emissions to the coast. Within our “fine” grid, an emission of $0.86\text{ Gg CHBr}_3\text{ yr}^{-1}$ from the 174 ocean grid cells leads to a mean oceanic flux of $0.06\text{ ng m}^{-2}\text{ s}^{-1}$. Following the reassignment of land emissions, $0.42\text{ Gg CHBr}_3\text{ yr}^{-1}$ is emitted from the 86 coastal grid cells, which leads to an identical mean coastal flux of $0.06\text{ ng m}^{-2}\text{ s}^{-1}$. Now, across the entire tropics, $\sim 73\%$ of the grid cells are ocean, and $\sim 3\%$ match our definition of coast. Using the above fluxes, the resulting tropical emission magnitudes are $247\text{ Gg CHBr}_3\text{ yr}^{-1}$ from the ocean, and $11\text{ Gg CHBr}_3\text{ yr}^{-1}$ from the coast. Their sum is broadly similar to, but due to slightly different assumptions not equal to, the total emission of $225\text{ Gg CHBr}_3\text{ yr}^{-1}$ found above. In contrast to our finding, many previous studies have suggested an approximately even split between emissions from

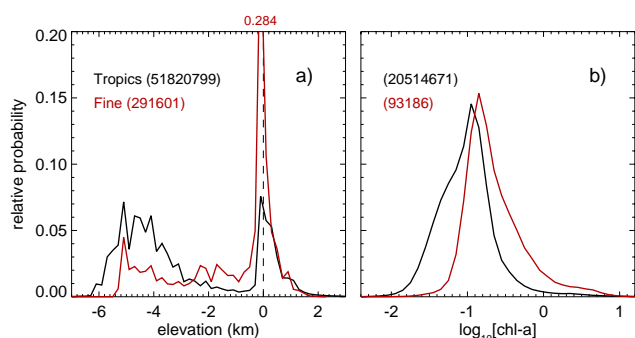


Fig. 9. (a) Shows a comparison of the distribution of surface elevation in the tropics (20° S– 20° N, in black) and in the “fine” grid cells (in red). The zero line, separating ocean from land, is marked with a dashed line. For the elevation data a bin size of 0.2 km is used, and the number of elevations in each classification is noted on the plot in parentheses. (b) Shows a similar plot for the chlorophyll *a* concentration measured by SeaWiFS. These data (original units are mg m^{-3}) are log-transformed, and a bin size of 0.1 is used. Again, the total number of pixels contributing to each distribution is noted in parentheses.

the global ocean and the global coast; we return to this subject in the next section.

For all these extrapolations it is reasonable to ask whether our region of focus around Borneo is representative of the tropics as a whole. To do this we briefly discuss two variables, ocean depth and surface chlorophyll *a* concentration, that have been related to CHBr_3 emissions in previous studies (e.g. Carpenter et al., 2009; Palmer and Reason, 2009; Ordóñez et al., 2012). Figure 9 shows PDFs for surface elevation (negative values give ocean depth) and chlorophyll *a* concentration for the whole of the tropics and for the “fine” grid cells. The elevation data have a horizontal resolution of ~ 2 km, and are available from http://www.gebco.net/data_and_products/gridded_bathymetry_data/. The chlorophyll *a* data have a horizontal resolution of ~ 9 km, and are from the satellite-borne Sea-viewing Wide Field-of-view Sensor (SeaWiFS, available from <http://daac.gsfc.nasa.gov/giovanni/>); the PDFs are constructed using all available monthly mean products within our analysis period (i.e. November 2008–May 2010). Plot (a) shows that the “fine” grid cells contain proportionately more shallow water (depths of 0–200 m) and less deep ocean (depths ~ 5 km) than the tropics as a whole. In addition, plot (b) of this figure shows that the distribution of chlorophyll *a* concentrations is shifted towards higher values in the “fine” region than in the tropics (the tropical mean chlorophyll *a* concentration is 0.20 mg m^{-3} , while the corresponding value for the “fine” grid cells, 0.45 mg m^{-3}).

While it has been suggested that, in broad terms, shallow water and high chlorophyll *a* concentrations may be associated with enhanced emissions of CHBr_3 (e.g. Carpenter et al., 2009; Palmer and Reason, 2009; Ordóñez et al., 2012), no robust relationships of this type have yet been demon-

strated. To illustrate, we have shown that coastal (similar to shallow water) emissions in our estimates are not stronger than non-coastal, or “open ocean” emissions. Further, our estimated emission distributions bear little resemblance to the distribution of either chlorophyll *a*, or of ocean depth ($r^2 < 0.05$ when the elevation and chlorophyll *a* data within the “fine” grid cells are degraded to the resolution of our emission estimates). As such we can only tentatively suggest that our extrapolations above, based upon estimates covering an area of relatively shallow and productive ocean, are more consistent with an upper limit for total tropical emissions.

We have also repeated the tropical emission extrapolation using an alternative definition of coast, based on ocean depth (elevation data as above). The distribution of land, coast (ocean shallower than 180 m, following Carpenter et al., 2009) and ocean within each grid cell was used to attribute emissions to each surface type. However, this new definition yielded similar results, and does not affect our conclusions. To illustrate, in the case of Experiment D, our original definition of coast, based on the UM land/sea mask, led to 25 % of emissions in coastal areas, and 67 % in open ocean areas (see Table 2). Using this new definition the equivalent percentages are 16 % and 72 %.

6 Summary and discussion

In this study we have described a new method to quantify CHBr_3 emissions. For the first time, an inversion process has been used to investigate systematically their distribution and magnitude. This process is informed by new long-term, high-frequency observations from Borneo, which are the first of their kind in the poorly sampled and convectively active Maritime Continent. We have examined how sensitive the estimated CHBr_3 emissions are to the characteristics of our measurements, and to assumptions in the inversion methodology.

We find that our local measurements of a short-lived compound like CHBr_3 , with presumed diffuse sources, are overwhelmingly influenced by emissions from within a relatively small “footprint”, covering < 1 % of the tropics (20° S– 20° N). This is also likely to be true of other CHBr_3 measurements, and has the important implication that observations from a very large number of sites would be required to obtain global coverage. Within the footprint region our results are robust, with the various sensitivities considered often leading to relatively minor changes, and emissions totalling 1.1 – $1.4 \text{ Gg CHBr}_3 \text{ yr}^{-1}$. Even without a priori constraint these emissions were largely distributed within oceanic grid cells, which is broadly consistent with expected sources due to the presence of either micro- or macroalgae.

An extrapolation of the flux calculated in experiment D results in a best estimate of tropical (20° S– 20° N) emissions of $225 \text{ Gg CHBr}_3 \text{ yr}^{-1}$, with possible lower and upper bounds (180 and $280 \text{ Gg CHBr}_3 \text{ yr}^{-1}$) provided by varying

the baseline mixing ratio in experiments E and F. We note that the ocean in the “fine” region on which our extrapolations are based appears to be typically shallower, and more biologically productive, than the tropical average. Despite this, our estimate is slightly lower than those arrived at in recent global modelling studies. For the tropics defined as 20° S–20° N, these emissions include ~ 240 Gg CHBr₃ yr⁻¹ (Liang et al., 2010), ~ 310 Gg CHBr₃ yr⁻¹ (Warwick et al., 2006, updated by Pyle et al. 2011) and ~ 300 Gg CHBr₃ yr⁻¹ (Ordóñez et al., 2012). Our estimate, however, is somewhat higher than the recent “bottom up” estimate of Ziska et al. (2013), who used two different statistical methods to arrive at a tropical flux of ~ 70 – 90 Gg CHBr₃ yr⁻¹. We also emphasise that the spatial distribution of CHBr₃ emissions in our “fine” grid cells is not similar to the distribution of either chlorophyll *a* or ocean depth. Our study therefore echoes previous findings: based on the two possible proxies we have examined, no simple relationship that would enable a straightforward physical parameterisation of CHBr₃ emissions appears to exist.

Hossaini et al. (2013) have recently compared the impact of including the four other estimates discussed above in the same global model, and found that no single inventory allows a satisfactory match with observations in all locations. For the tropics, they found that the relatively low emissions of Ziska et al. (2013) provided the best match with the few available datasets. The paucity of tropical data, added to the fact that the different studies used different subsets of those data to construct their inventories, is likely to be one cause of the differing tropical emission values. Overall, the study of Hossaini et al. (2013) supports our finding, that using CHBr₃ measurements from a relatively small number of locations makes constructing (or evaluating) a global emission inventory extremely difficult.

Our estimated emission distributions suggest that coastal CHBr₃ sources, presumably related to seaweed, contribute only a minor fraction of the total tropical emission. We certainly believe that our observations from Tawau were affected by strong, nearby coastal CHBr₃ emissions, but the typical width of the coastal band that is affected by such sources is still not clear (see also Carpenter et al., 2009). The average CHBr₃ mixing ratios we observed inland were lower than at the coast, and we also saw no elevated mixing ratios (i.e. > 5 ppt) inland. This observed gradient occurs over a distance of ~ 50 km, which is well within a typical global model grid cell. The recent study of Leedham et al. (2013), which also focussed on the Maritime Continent, also suggests that the band of coastline directly influenced by macroalgae is likely to be rather narrow. Further investigation of this type of gradient should be a priority because the global impact of coastal emissions is still poorly understood, and the best way to represent them in coarse global models is, therefore, still not clear. This is especially relevant for the Maritime Continent; it is conceivable that much of this region is influenced by coastal processes in some way.

Our examination of the “fine” grid cells near Borneo suggests a moderate ($\sim 20\%$) difference between the emissions that are most suitable for p-TOMCAT (see Fig. 5) and for NAME (i.e. our inversion solutions, Fig. 6). In comparison, the study of Pyle et al. (2011) reported differences of up to a factor of three in the CHBr₃ emissions inferred using these two models, and for a similar area of the Maritime Continent. This may in part be because here we utilise year-round measurements, whereas Pyle et al. (2011) considered only a short period in which only southeasterly flow was sampled. Another important difference is that this study uses coastal data, with higher average CHBr₃ mixing ratios, as a constraint. In contrast the p-TOMCAT emissions reported by Pyle et al. (2011), and used here to generate our pseudo observations (Sect. 4), were constrained only by data collected inland at Bukit Atur. Further, those CHBr₃ mixing ratios observed at Bukit Atur in mid-2008 have a mean of 1.21 ppt, which is substantially lower than the long-term (2008–2010) mean of 1.55 ppt for the dataset employed here.

We expect that underlying differences in the efficiency with which these models mix air (i.e. emissions) out of the boundary layer remain an important uncertainty in our emission estimates (see also Robinson et al., 2012). The size of such systematic uncertainties could be explored by using meteorological information from a different forecast centre to drive the NAME dispersion calculations (see Manning et al., 2011, who performed such a test and found little difference in mid-latitudes) and by using a different dispersion model, containing different, or additional parameterisations. Further studies assessing the characteristics of turbulence and related tracer distributions in and immediately above the tropical boundary layer, the influence of small-scale “sea breeze” type circulations, and the representation of these processes in dispersion models, are certainly required.

The limited constraint our new data provide on regional CHBr₃ emissions argues for an expanded network of measurement locations in the Maritime Continent. The spacing of such a network should be similar to the size of the footprint region described here. We have made progress in this direction by recently starting to make measurements of CHBr₃ and other halocarbons at sites near Darwin, in Northern Australia, and in Bachok, on the East coast of Peninsular Malaysia. Further, our two current measurement locations appear to be too close to distinguish between different regional scale influences on air mass composition. This is illustrated by the very similar patterns of variability in our observations of tetrachloroethene (C₂Cl₄), which appears to have few local sources (see Robinson et al., 2014). In addition, the coastal observations used in this study are probably too close to strong local CHBr₃ sources to be used to estimate regional-scale emissions. When we attempted to remove the data most influenced by very local emissions we obtained an improved emission estimate, as measured by the cost function. So ideally, measurement locations should be influenced, but not overwhelmed, by coastal emissions. Surveys made

across the coastal gradient would help to determine where such a level of influence occurs. We also emphasise that measurement locations should, as much as is possible in the Maritime Continent, avoid complicated terrain associated with prevailing smaller-scale circulations that are particularly difficult to represent accurately in global meteorological models.

Acknowledgements. M. Ashfold thanks the Natural Environment Research Council (NERC) for a research studentship, and is grateful for support through the ERC ACCI project (project number 267760). N. Harris is supported by a NERC Advanced Research Fellowship. This work was supported through the EU SHIVA project, through the NERC OP3 project, and NERC grants NE/F020341/1 and NE/J006246/1. We also acknowledge the Department of Energy and Climate Change for their support in the development of InTEM (contract GA0201). For field site support we thank S.-M. Phang, A. A. Samah and M. S. M. Nadzir of Universiti Malaya, S. Ong and H. E. Ung of Global Satria, Maznorizan Mohamad, L. K. Peng and S. E. Yong of the Malaysian Meteorological Department, the Sabah Foundation, the Danum Valley Field Centre and the Royal Society. This paper constitutes publication no. 613 of the Royal Society South East Asia Rainforest Research Programme.

Edited by: W. T. Sturges

References

- Aschmann, J., Sinnhuber, B.-M., Atlas, E. L., and Schauffler, S. M.: Modeling the transport of very short-lived substances into the tropical upper troposphere and lower stratosphere, *Atmos. Chem. Phys.*, 9, 9237–9247, doi:10.5194/acp-9-9237-2009, 2009.
- Ayers, G. P., Gillett, R. W., Caaney, J. M., and Dick, A. L.: Chloride and bromide loss from sea-salt particles in Southern Ocean air, *J. Atmos. Chem.*, 33, 299–319, doi:10.1023/A:1006120205159, 1999.
- Butler, J. H., King, D. B., Lobert, J. M., Montzka, S. A., Yvon-Lewis, S. A., Hall, B. D., Warwick, N. J., Mondeel, D. J., Aydin, M., and Elkins, J. W.: Oceanic distributions and emissions of short-lived halocarbons, *Global Biogeochem. Cy.*, 21, GB1023, doi:10.1029/2006GB002732, 2007.
- Carpenter, L. J. and Liss, P. S.: On temperate sources of bromoform and other reactive organic bromine gases, *J. Geophys. Res.-Atmos.*, 105, 20539–20547, doi:10.1029/2000JD900242, 2000.
- Carpenter, L. J., Jones, C. E., Dunk, R. M., Hornsby, K. E., and Woeltjen, J.: Air-sea fluxes of biogenic bromine from the tropical and North Atlantic Ocean, *Atmos. Chem. Phys.*, 9, 1805–1816, doi:10.5194/acp-9-1805-2009, 2009.
- Dvortsov, V. L., Geller, M. A., Solomon, S., Schauffler, S. M., Atlas, E. L., and Blake, D. R.: Rethinking reactive halogen budgets in the midlatitude lower stratosphere, *Geophys. Res. Lett.*, 26, 1699–1702, doi:10.1029/1999GL900309, 1999.
- Gostlow, B., Robinson, A. D., Harris, N. R. P., O'Brien, L. M., Oram, D. E., Mills, G. P., Newton, H. M., Yong, S. E., and Pyle, J.: μ Dirac: an autonomous instrument for halocarbon measurements, *Atmos. Meas. Tech.*, 3, 507–521, doi:10.5194/amt-3-507-2010, 2010.
- Gschwend, P. M., Macfarlane, J. K., and Newman, K. A.: Volatile Halogenated Organic-Compounds Released To Seawater From Temperate Marine Macroalgae, *Science*, 227, 1033–1035, doi:10.1126/science.227.4690.1033, 1985.
- Hosking, J. S., Russo, M. R., Braesicke, P., and Pyle, J. A.: Modelling deep convection and its impacts on the tropical tropopause layer, *Atmos. Chem. Phys.*, 10, 11175–11188, doi:10.5194/acp-10-11175-2010, 2010.
- Hossaini, R., Chipperfield, M. P., Monge-Sanz, B. M., Richards, N. A. D., Atlas, E., and Blake, D. R.: Bromoform and dibromomethane in the tropics: a 3-D model study of chemistry and transport, *Atmos. Chem. Phys.*, 10, 719–735, doi:10.5194/acp-10-719-2010, 2010.
- Hossaini, R., Chipperfield, M. P., Feng, W., Breider, T. J., Atlas, E., Montzka, S. A., Miller, B. R., Moore, F., and Elkins, J.: The contribution of natural and anthropogenic very short-lived species to stratospheric bromine, *Atmos. Chem. Phys.*, 12, 371–380, doi:10.5194/acp-12-371-2012, 2012.
- Hossaini, R., Mantle, H., Chipperfield, M. P., Montzka, S. A., Hamer, P., Ziska, F., Quack, B., Krüger, K., Tegtmeier, S., Atlas, E., Sala, S., Engel, A., Bönisch, H., Keber, T., Oram, D., Mills, G., Ordóñez, C., Saiz-Lopez, A., Warwick, N., Liang, Q., Feng, W., Moore, F., Miller, B. R., Maréchal, V., Richards, N. A. D., Dorf, M., and Pfeilsticker, K.: Evaluating global emission inventories of biogenic bromocarbons, *Atmos. Chem. Phys.*, 13, 11819–11838, doi:10.5194/acp-13-11819-2013, 2013.
- Itoh, N. and Shinya, M.: Seasonal Evolution Of Bromomethanes From Coralline Algae (Corallinaceae) And Its Effect On Atmospheric Ozone, *Mar. Chem.*, 45, 95–103, doi:10.1016/0304-4203(94)90094-9, 1994.
- Jones, A., Thomson, D., Hort, M., and Devenish, B.: The U.K. Met Office's Next-Generation Atmospheric Dispersion Model, NAME III, in: *Air Pollution Modeling and Its Application XVII*, edited by: Borrego, C. and Norman, A.-L., 580–589, Springer US, doi:10.1007/978-0-387-68854-1_62, 2007.
- Jones, C. E., Andrews, S. J., Carpenter, L. J., Hogan, C., Hopkins, F. E., Laube, J. C., Robinson, A. D., Spain, T. G., Archer, S. D., Harris, N. R. P., Nightingale, P. D., O'Doherty, S. J., Oram, D. E., Pyle, J. A., Butler, J. H., and Hall, B. D.: Results from the first national UK inter-laboratory calibration for very short-lived halocarbons, *Atmos. Meas. Tech.*, 4, 865–874, doi:10.5194/amt-4-865-2011, 2011.
- Kirkpatrick, S., Gelatt, C. D., and Vecchi, M. P.: Optimization by Simulated Annealing, *Science*, 220, 671–680, doi:10.1126/science.220.4598.671, 1983.
- Leedham, E. C., Hughes, C., Keng, F. S. L., Phang, S.-M., Malin, G., and Sturges, W. T.: Emission of atmospherically significant halocarbons by naturally occurring and farmed tropical macroalgae, *Biogeosciences*, 10, 3615–3633, doi:10.5194/bg-10-3615-2013, 2013.
- Levine, J. G., Braesicke, P., Harris, N. R. P., Savage, N. H., and Pyle, J. A.: Pathways and timescales for troposphere-to-stratosphere transport via the tropical tropopause layer and their relevance for very short lived substances, *J. Geophys. Res.-Atmos.*, 112, D04308, doi:10.1029/2005JD006940, 2007.
- Liang, Q., Stolarski, R. S., Kawa, S. R., Nielsen, J. E., Douglass, A. R., Rodriguez, J. M., Blake, D. R., Atlas, E. L., and Ott, L. E.: Finding the missing stratospheric Bry: a global modeling study

- of CHBr_3 and CH_2Br_2 , *Atmos. Chem. Phys.*, 10, 2269–2286, doi:10.5194/acp-10-2269-2010, 2010.
- Manley, S. L., Goodwin, K., and North, W. J.: Laboratory Production Of Bromoform, Methylene Bromide, And Methyl Iodide By Macroalgae And Distribution In Nearshore Southern California Waters, *Limnol. Oceanogr.*, 37, 1652–1659, 1992.
- Manning, A. J.: The challenge of estimating regional trace gas emissions from atmospheric observations, *Philos. T. R. Soc. A*, 369, 1943–1954, doi:10.1098/rsta.2010.0321, 2011.
- Manning, A. J., Ryall, D. B., Derwent, R. G., Simmonds, P. G., and O'Doherty, S.: Estimating European emissions of ozone-depleting and greenhouse gases using observations and a modeling back-attribution technique, *J. Geophys. Res.-Atmos.*, 108, 4405, doi:10.1029/2002JD002312, 2003.
- Manning, A. J., O'Doherty, S., Jones, A. R., Simmonds, P. G., and Derwent, R. G.: Estimating UK methane and nitrous oxide emissions from 1990 to 2007 using an inversion modeling approach, *J. Geophys. Res.*, 116, D02305, doi:10.1029/2010JD014763, 2011.
- Mohd Nadzir, M. S., Phang, S. M., Abas, M. R., Abdul Rahman, N., Abu Samah, A., Sturges, W. T., Oram, D. E., Mills, G. P., Leedham, E. C., Pyle, J. A., Harris, N. R. P., Robinson, A. D., Ashfold, M. J., Mead, M. I., Latif, M. T., Mohd Hanafiah, M., Khan, M. F., and Amiruddin, A. M.: Bromocarbons in the tropical coastal and open ocean atmosphere during the Prime Expedition Scientific Cruise 2009 (PESC 09), *Atmos. Chem. Phys. Discuss.*, 14, 953–984, doi:10.5194/acpd-14-953-2014, 2014.
- Montzka, S. A. and Reimann, S.: Ozone-Depleting Substances (ODSs) and Related Chemicals, vol. Scientific Assessment of Ozone Depletion: 2010, Global Ozone Research and Monitoring Project, Report No. 52, chap. 1, World Meteorological Organization (WMO), Geneva, 2011.
- Moore, R. M., Webb, M., Tokarczyk, R., and Wever, R.: Bromoperoxidase and iodoperoxidase enzymes and production of halogenated methanes in marine diatom cultures, *J. Geophys. Res.-Oceans*, 101, 20899–20908, doi:10.1029/96JC01248, 1996.
- Morrison, N. L. and Webster, H. N.: An Assessment of Turbulence Profiles in Rural and Urban Environments Using Local Measurements and Numerical Weather Prediction Results, *Bound.-Lay. Meteorol.*, 115, 223–239, doi:10.1007/s10546-004-4422-8, 2005.
- O'Doherty, S., Cunnold, D. M., Manning, A., Miller, B. R., Wang, R. H. J., Krummel, P. B., Fraser, P. J., Simmonds, P. G., McCulloch, A., Weiss, R. F., Salameh, P., Porter, L. W., Prinn, R. G., Huang, J., Sturrock, G., Ryall, D., Derwent, R. G., and Montzka, S. A.: Rapid growth of hydrofluorocarbon 134a and hydrochlorofluorocarbons 141b, 142b, and 22 from Advanced Global Atmospheric Gases Experiment (AGAGE) observations at Cape Grim, Tasmania, and Mace Head, Ireland, *J. Geophys. Res.*, 109, D06310, doi:10.1029/2003JD004277, 2004.
- Ordóñez, C., Lamarque, J.-F., Tilmes, S., Kinnison, D. E., Atlas, E. L., Blake, D. R., Sousa Santos, G., Brasseur, G., and Saiz-Lopez, A.: Bromine and iodine chemistry in a global chemistry-climate model: description and evaluation of very short-lived oceanic sources, *Atmos. Chem. Phys.*, 12, 1423–1447, doi:10.5194/acp-12-1423-2012, 2012.
- Palmer, C. J. and Reason, C. J.: Relationships of surface bromoform concentrations with mixed layer depth and salinity in the tropical oceans, *Global Biogeochem. Cy.*, 23, GB2014, doi:10.1029/2008GB003338, 2009.
- Parrella, J. P., Jacob, D. J., Liang, Q., Zhang, Y., Mickley, L. J., Miller, B., Evans, M. J., Yang, X., Pyle, J. A., Theys, N., and Van Roozendael, M.: Tropospheric bromine chemistry: implications for present and pre-industrial ozone and mercury, *Atmos. Chem. Phys.*, 12, 6723–6740, doi:10.5194/acp-12-6723-2012, 2012.
- Pike, R. C., Lee, J. D., Young, P. J., Carver, G. D., Yang, X., Warwick, N., Moller, S., Misztal, P., Langford, B., Stewart, D., Reeves, C. E., Hewitt, C. N., and Pyle, J. A.: NO_x and O_3 above a tropical rainforest: an analysis with a global and box model, *Atmos. Chem. Phys.*, 10, 10607–10620, doi:10.5194/acp-10-10607-2010, 2010.
- Pisso, I., Haynes, P. H., and Law, K. S.: Emission location dependent ozone depletion potentials for very short-lived halogenated species, *Atmos. Chem. Phys.*, 10, 12025–12036, doi:10.5194/acp-10-12025-2010, 2010.
- Polson, D., Fowler, D., Nemitz, E., Skiba, U., McDonald, A., Famulari, D., Marco, C. D., Simmons, I., Weston, K., Purvis, R., Coe, H., Manning, A., Webster, H., Harrison, M., O'Sullivan, D., Reeves, C., and Oram, D.: Estimation of spatial apportionment of greenhouse gas emissions for the UK using boundary layer measurements and inverse modelling technique, *Atmos. Environ.*, 45, 1042–1049, doi:10.1016/j.atmosenv.2010.10.011, 2011.
- Press, W. H., Teukolsky, S. A., Vetterling, W. T., and Flannery, B. P.: *Numerical Recipes: The Art of Scientific Computing*, Cambridge University Press, 1986.
- Pyle, J. A., Ashfold, M. J., Harris, N. R. P., Robinson, A. D., Warwick, N. J., Carver, G. D., Gostlow, B., O'Brien, L. M., Manning, A. J., Phang, S. M., Yong, S. E., Leong, K. P., Ung, E. H., and Ong, S.: Bromoform in the tropical boundary layer of the Maritime Continent during OP3, *Atmos. Chem. Phys.*, 11, 529–542, doi:10.5194/acp-11-529-2011, 2011.
- Qian, J.-H., Robertson, A. W., and Moron, V.: Diurnal Cycle in Different Weather Regimes and Rainfall Variability over Borneo Associated with ENSO, *J. Climate*, 26, 1772–1790, doi:10.1175/JCLI-D-12-00178.1, 2013.
- Quack, B. and Suess, E.: Volatile halogenated hydrocarbons over the western Pacific between 43 degrees and 4 degrees N, *J. Geophys. Res.-Atmos.*, 104, 1663–1678, doi:10.1029/98JD02730, 1999.
- Quack, B. and Wallace, D. W. R.: Air-sea flux of bromoform: Controls, rates, and implications, *Global Biogeochem. Cy.*, 17, 1023, doi:10.1029/2002GB001890, 2003.
- Read, K. A., Mahajan, A. S., Carpenter, L. J., Evans, M. J., Faria, B. V. E., Heard, D. E., Hopkins, J. R., Lee, J. D., Moller, S. J., Lewis, A. C., Mendes, L., McQuaid, J. B., Oetjen, H., Saiz-Lopez, A., Pilling, M. J., and Plane, J. M. C.: Extensive halogen-mediated ozone destruction over the tropical Atlantic Ocean, *Nature*, 453, 1232–1235, doi:10.1038/nature07035, 2008.
- Reimann, S., Manning, A. J., Simmonds, P. G., Cunnold, D. M., Wang, R. H. J., Li, J. L., McCulloch, A., Prinn, R. G., Huang, J., Weiss, R. F., Fraser, P. J., O'Doherty, S., Grealley, B. R., Stemmler, K., Hill, M., and Folini, D.: Low European methyl chloroform emissions inferred from long-term atmospheric measurements, *Nature*, 433, 506–508, doi:10.1038/nature03220, 2005.
- Robinson, N. H., Allan, J. D., Trembath, J. A., Rosenberg, P. D., Allen, G., and Coe, H.: The lofting of Western Pacific regional aerosol by island thermodynamics as observed around Borneo,

- Atmos. Chem. Phys., 12, 5963–5983, doi:10.5194/acp-12-5963-2012, 2012.
- Robinson, A. D., Harris, N. R. P., Ashfold, M. J., Gostlow, B., Warwick, N. J., O'Brien, L. M., Beardmore, E. J., Nadzir, M. S. M., Phang, S. M., Samah, A. A., Ong, S., Ung, H. E., Peng, L. K., Yong, S. E., Mohamad, M., and Pyle, J. A.: Long term halocarbon observations from a coastal and an inland site in Sabah, Malaysian Borneo, Atmos. Chem. Phys. Discuss., 14, 1919–1969, doi:10.5194/acpd-14-1919-2014, 2014.
- Salawitch, R. J., Weisenstein, D. K., Kovalenko, L. J., Sioris, C. E., Wennberg, P. O., Chance, K., Ko, M. K. W., and McLinden, C. A.: Sensitivity of ozone to bromine in the lower stratosphere, Geophys. Res. Lett., 32, L05811, doi:10.1029/2004GL021504, 2005.
- Saltzman, E. S., Aydin, M., De Bruyn, W. J., King, D. B., and Yvon-Lewis, S. A.: Methyl bromide in preindustrial air: Measurements from an Antarctic ice core, J. Geophys. Res.-Atmos., 109, D05301, doi:10.1029/2003JD004157, 2004.
- Stohl, A., Seibert, P., Arduini, J., Eckhardt, S., Fraser, P., Grealley, B. R., Lunder, C., Maione, M., Mühle, J., O'Doherty, S., Prinn, R. G., Reimann, S., Saito, T., Schmidbauer, N., Simmonds, P. G., Vollmer, M. K., Weiss, R. F., and Yokouchi, Y.: An analytical inversion method for determining regional and global emissions of greenhouse gases: Sensitivity studies and application to halocarbons, Atmos. Chem. Phys., 9, 1597–1620, doi:10.5194/acp-9-1597-2009, 2009.
- Sturges, W. T., Cota, G. F., and Buckley, P. T.: Bromoform Emission From Arctic Ice Algae, Nature, 358, 660–662, doi:10.1038/358660a0, 1992.
- Tokarczyk, R. and Moore, R. M.: Production Of Volatile Organohalogenes By Phytoplankton Cultures, Geophys. Res. Lett., 21, 285–288, doi:10.1029/94GL00009, 1994.
- Wamsley, P. R., Elkins, J. W., Fahey, D. W., Dutton, G. S., Volk, C. M., Myers, R. C., Montzka, S. A., Butler, J. H., Clarke, A. D., Fraser, P. J., Steele, L. P., Lucarelli, M. P., Atlas, E. L., Schauffler, S. M., Blake, D. R., Rowland, F. S., Sturges, W. T., Lee, J. M., Penkett, S. A., Engel, A., Stimpfle, R. M., Chan, K. R., Weisenstein, D. K., Ko, M. K. W., and Salawitch, R. J.: Distribution of halon-1211 in the upper troposphere and lower stratosphere and the 1994 total bromine budget, J. Geophys. Res.-Atmos., 103, 1513–1526, doi:10.1029/97JD02466, 1998.
- Warwick, N. J., Pyle, J. A., Carver, G. D., Yang, X., Savage, N. H., O'Connor, F. M., and Cox, R. A.: Global modeling of biogenic bromocarbons, J. Geophys. Res.-Atmos., 111, D24305, doi:10.1029/2006JD007264, 2006.
- Webster, H. N., Thomson, D. J., and Morrisson, N. L.: New turbulence profiles for NAME, Turbulence and Diffusion Note No. 288, 2003.
- Yang, X., Cox, R. A., Warwick, N. J., Pyle, J. A., Carver, G. D., O'Connor, F. M., and Savage, N. H.: Tropospheric bromine chemistry and its impacts on ozone: A model study, J. Geophys. Res.-Atmos., 110, D23311, doi:10.1029/2005JD006244, 2005.
- Yokouchi, Y., Li, H.-J., Machida, T., Aoki, S., and Akimoto, H.: Isoprene in the marine boundary layer (Southeast Asian Sea, eastern Indian Ocean, and Southern Ocean): Comparison with dimethyl sulfide and bromoform, J. Geophys. Res., 104, 8067–8076, doi:10.1029/1998JD100013, 1999.
- Yokouchi, Y., Hasebe, F., Fujiwara, M., Takashima, H., Shiotani, M., Nishi, N., Kanaya, Y., Hashimoto, S., Fraser, P., Toom-Saunty, D., Mukai, H., and Nojiri, Y.: Correlations and emission ratios among bromoform, dibromochloromethane, and dibromomethane in the atmosphere, J. Geophys. Res.-Atmos., 110, D23309, doi:10.1029/2005JD006303, 2005.
- Ziska, F., Quack, B., Abrahamsson, K., Archer, S. D., Atlas, E., Bell, T., Butler, J. H., Carpenter, L. J., Jones, C. E., Harris, N. R. P., Hepach, H., Heumann, K. G., Hughes, C., Kuss, J., Krüger, K., Liss, P., Moore, R. M., Orlikowska, A., Raimund, S., Reeves, C. E., Reifenhäuser, W., Robinson, A. D., Schall, C., Tanhua, T., Tegtmeier, S., Turner, S., Wang, L., Wallace, D., Williams, J., Yamamoto, H., Yvon-Lewis, S., and Yokouchi, Y.: Global sea-to-air flux climatology for bromoform, dibromomethane and methyl iodide, Atmos. Chem. Phys., 13, 8915–8934, doi:10.5194/acp-13-8915-2013, 2013.