Journal of Applied Physics 117, 17E121 (2015). DOI: 10.1063/1.4916296

XMCD Study of Epitaxial Magnetite Ultrathin Film on MgO(100)

W. Q. Liu,^{1,2} M. Y. Song,³ N. J. Maltby,² S. P. Li,² J. G. Lin,³ M. G. Samant,⁴ S. S. P. Parkin,⁴ P. Bencok,⁵ Paul Steadman,⁵ Alexey Dobrynin,⁵ Y. B. Xu,^{1,2,*} R. Zhang^{1,*}

¹York-Nanjing International Center for Spintronics (YNICS), School of Electronics Science and Engineering, Nanjing University, Nanjing, 210093, China

² Spintronics and Nanodevice Laboratory, Department of Electronics, University of York, York YO10 5DD, UK

³Center for Condensed Matter Sciences, National Taiwan University, Taipei 106, Taiwan

⁴ IBM Research Division, Almaden Research Center, San Jose, CA 95120 USA

⁵Diamond Light Source, Didcot OX11 0DE, UK

Abstract

The spin and orbital magnetic moments of the Fe₃O₄ epitaxial ultrathin film synthesized by plasma assisted simultaneous oxidization on MgO(100) have been studied with X-ray magnetic circular dichroism (XMCD). The ultrathin film retains a rather large total magnetic moment, i.e. $(2.73\pm0.15)\mu_{\rm B}/f.u.$, which is ~ 70% of that for the bulk-like Fe₃O₄. A significant unquenched orbital moment up to $(0.54\pm0.05) \mu_{\rm B}/f.u.$ was observed, which could come from the symmetry breaking at the Fe₃O₄/MgO interface. Such sizable orbital moment will add capacities to the Fe₃O₄-based spintronics devices in the magnetization reversal by the electric field.

Keywords

magnetite, thin film, XMCD, orbital moment, spintronics

I. Background

Magnetite, or Fe₃O₄, is a competitive contender in the race to become the one of the key materials in the future spintronics computing, or the spin-operation-based data processing and sensing, due to its high spin polarization near the Fermi level (E_F). The experimentally true half-metallic state was reported by Dedkov *et al.*¹ by means of spin and angle-resolved photoemission spectroscopy, from which P = - (80 ± 5)% was obtained at E_F , consistent with the spin-split band energies from DFT calculations.² More desirably, the high Curie temperature (T_c) of Fe₃O₄ makes it a promising candidate for room temperature use. Fascinating properties of spin transport have also been presented in Fe₃O₄, i.e. spin Seebeck effect,³ spin filter effect,⁴ gate voltage-induced phase transition,⁵ and spin valve effect of Fe₃O₄/MgO/Fe₃O₄ junctions.^{6,7} Yet at the meantime, many fundamental properties of magnetite such as the half-metallicity, spin and orbital ordering, Verwey transition mechanism and the coupling mechanism between different sites have long been open issues, and with the thickness down to nanometer scale, these issues become even more sophisticated.

The rather complicated magnetic structure of Fe_3O_4 was partly proposed by Verwey and Hayman in 1941⁸ and the total structure was put forward by Neel in 1948,⁹ and then confirmed three years later by neutron scattering.¹⁰ Fe₃O₄ has cubic inverse spinel structure, where Fe³⁺ ions occupy tetrahedral sites (usually called A sites), whereas octahedral sites (B sites) are occupied by both Fe³⁺ and Fe²⁺ ions. The spin of Fe³⁺ ions at octahedral and tetrahedral sites are aligned antiparallel to each other leading to a net spin magnetic moment (m_{spin}) of 4 $\mu_{\rm B}/f.u.$, corresponding to a fully occupied local majority band (opposite for A and B sites). The presence of integer m_{spin} and vanishing orbital moment (m_{orb}) of magnetite are expected as an indication for a B-site minority electron conduction mechanism, and its accompanied full spin polarization at the $E_{\rm F}$. However, concerning with that, controversial results have been reported utilizing techniques such superconducting quantum interference device (SQUID) magnetometer,^{11,38} XMCD,^{12,27,30,39,38} and magnetic Compton scattering (MCS)^{33, 34} and calculations with local density approximation (LDA),³⁰ LDA+U,³⁰ local spin density approximation (LSDA)+U,² and moment analysis etc..¹² The fundamental magnetic properties of Fe₃O₄ show strong dependence on the sample preparation methodology. By simultaneous oxidation, Babu et al.³² observed $m_{spin} = (1.20\pm0.05) \mu_{B}/f.u.$ of Fe₃O₄ on BaTiO₃, and enhanced $m_{spin} = 7.7 \ \mu_{\rm B}/{\rm f.u.}$ was reported by Arora et al..¹¹ Whilst using pulsed laser deposition (PLD), Orna et al.³⁸ observed greatly reduced $m_{spin} = 1.83 \ \mu_B/f.u.$ of Fe₃O₄ thin film on MgO. And using floating zone methods, the reported $m_{\rm orb}$ of bulk Fe₃O₄ varies from $(0.67\pm0.09) \mu_{\rm B}/{\rm f.u.}$ by Huang et al., ³⁰ $(0.51\pm0.05) \mu_{\rm B}/{\rm f.u.}$ by Li et al., ³³ to $(0.06\pm0.14) \mu_{\rm B}/{\rm f.u.}$ by Duffy et al.³⁴ and all the way down to $-0.001\mu_{\rm B}/f.u.$ by Goering et al.²⁷ Theoretical

analyses were presented with even sharper contrast, varying from 0.43 $\mu_{\rm B}$ /f.u. by Huang *et al.*³⁰ to 0.02 $\mu_{\rm B}$ /f.u. by Antonov *et al.*² The well-known Verwey transition of magnetite is accompanied by a transition to a low symmetry structure, by which the $m_{\rm spin}$ and especially $m_{\rm orb}$ are expected to change significantly. The experimental work,^{11,34,38} however, has so far found no difference of them across the transition, which questions the picture of a fully A site Fe³⁺ and a mixed-valence B site configuration of magnetite.

Fe₃O₄/MgO heterostructure has attracted great interest within the spintronics community in the recent few years. Crystalline MgO-based magnetic tunnel junctions have achieved remarkable success in assisting efficient spin injections for various applications.^{13,14,15} An insulating layer of MgO can be used as a tunneling barrier, which not only relieves the conductivity mismatch problem but also works as a spin filter. Moreover, MgO forms an excellent diffusion barrier with thermal stability up to 800 °C, effectively preventing the intermixing at a given ferromagnet-semiconductor interface. Efforts have been made to explore the magnetic behavior of nanoscale epitaxial thin films on bulk MgO(100).¹¹, ¹⁶ Yet to fundamentally understand character of 3d electrons in Fe₃O₄, it is a prerequisite to determine the m_{soin} and m_{orb} , respectively and unambiguously. Recently, we have reported a unquenched $m_{\rm orb}$ and a well retained $m_{\rm spin}$ in Fe₃O₄ thin films grown by post-growth annealing method.¹⁷ In this paper, we furthermore present a XMCD study of Fe₃O₄ in a distinct configuration, prepared by plasma-assisted simultaneous oxidation, aiming to have an insight of the different magnetic properties of the Fe₃O₄ thin film caused by the substrate and preparation methodology, and to contribute to the open question of the magnetic moments of the Fe_3O_4 in nano scale regime.

II. Sample preparation and global characterization

The magnetite ultrathin film used in this study was grown by oxygen-plasmaassisted MBE at a substrate temperature of 623 K and in oxygen pressure of 2.5×10^{-7} torr. Prior to the growth, the MgO was annealed at 600 °C and after 30 minutes, a sharp MgO(100) was obtained as can be seen from the reflection high-energy electron diffraction (RHEED) pattern in Fig. 1. (a). The 12 nm Fe₃O₄ was then deposited by effusion cells at a rate of 0.04 Å per second at 350 °C. The comparatively large oxygen ions form a face-centered-cubic (fcc) lattice and the Fe atoms are located in interstitial sites as shown in Fig. 1. (b).

The resistance versus temperature (R-T) relationship of the 12 nm Fe₃O₄/MgO thin film was measured by patterning into a standard Hall bar geometry. As shown in Fig. 2, a discontinuous change of the resistivity was observed at ~ 97 K, corresponding to the Verwey transition. Ferromagnetic resonance (FMR) measurement was performed and published

elsewhere, by which different resonance features was observed below and above the transition.¹⁶

The macroscopic magnetization of the sample was characterized by means of SQUID magnetometer. Figure 3. (a) presents the magnetization versus temperature relationship (M-T), obtained by cooling the sample from 300 to 20 K in zero magnetic field, followed by an application of a static magnetic field of 100 Oe and recording the magnetization values during the warming cycle to 300 K. The magnetization drops with the decreasing temperature and the Verwey transition at ~ 97 K, consistent with the transport measurement, can be distinguished from the dM/dT (see the Fig. 3(a) inset). The magnetization versus field (M-H) loops in Fig. 3. (b) was obtained by applying the magnetic field in the film plane and with the subtraction of a small diamagnetic contribution from the sample holder. It can be seen that the sample saturates at ~ 500 Oe with a magnetization of ~ 270 emu/cm³, which is ~ 70% of the principle bulk-like magnetite saturation value, i.e., 480 emu/cm³ or 4.0 $\mu_{\rm B}/f.u.$

It is generally accepted that the presence of Verwey transition is very sensitive to the stoichiometry and homogeneity of magnetite thin films.¹⁸ The temperature (T_V) at which such a transition occurs has been commonly observed to decrease from the bulk value down to 85K and could even disappear with the decreasing Fe₃O₄ thicknesse.^{11,19,20,21,38} The observed transition temperature of the 12nm Fe₃O₄/MgO(100) thin film in this study agrees well with the reported T_V of the magnetite films on or near stoichiometry,^{11,22} suggesting the sample is not overly suffering from cation or anion vacancies.

III. XMCD measurement

X-ray absorption spectroscopy (XAS) and XMCD experiments at the Fe $L_{2,3}$ absorption edges were performed at the beamline I10 of Diamond Light Source, UK. Circularly polarized X-rays with 100% degree of polarization²³ were used in normal incidence with respect to the sample plane and parallel with the applied magnetic field, in order to minimize the nonmagnetic asymmetries, as shown in Fig. 4. The XAS spectra were obtained by total electron yield (TEY) detection. The XMCD was taken as the difference of the XAS spectra, i.e., σ - σ ⁺, obtained by flipping the X-ray helicity at a fixed magnetic field of 30 kOe, under which the sample is fully magnetized with little paramagnetic contribution. Typical XAS and XMCD spectra at 300 K of the 1 2nm Fe₃O₄/MgO(100) are presented in Fig. 5. The XMCD spectra well reproduce the features of those from the theoretical calculations and previous observations of the ferrimagnetic Fe₃O₄ with the three Fe alternative up and down peaks coming from the antiparallel spin orientations of the A and B sites. The relative

intensities of these L_3 peaks show lesser occupation of Fe²⁺ B sites, which may be associated with the slight formation of a secondary γ -Fe₂O₃ phase.^{24, 25} The complex form of the XMCD spectrum arises because of an overlap of different sets of multiplet structures. The B sites Fe³⁺ and Fe²⁺ spin-up states exhibit negative peaks at Fe L_3 edge and positive peaks at the Fe L_2 edge, while the A sites Fe³⁺ spin-down states behave the oppositely at the Fe L_3 and L_2 edges, respectively.

The $m_{\rm spin}$ and $m_{\rm orb}$ of the the 12 nm Fe₃O₄/MgO(100) were calculated by applying sum rules on the integrated XMCD and total XAS spectra of Fe $L_{2,3}$ edges based on equation (1),²⁶

$$< L_{z} > = -\frac{4}{3} n_{h} \frac{\int_{L_{2,3}} (\sigma^{+} - \sigma^{-}) dE}{\int_{L_{2,3}} (\sigma^{+} + \sigma^{-}) dE}$$

$$< S_{z} > = -n_{h} \frac{6 \int_{L_{3}} (\sigma^{+} - \sigma^{-}) dE - 4 \int_{L_{2,3}} (\sigma^{+} - \sigma^{-}) dE}{\int_{L_{2,3}} (\sigma^{+} + \sigma^{-}) dE} \times SC + < T_{z} >$$
(1)

where n_h - the effective number of 3*d*-band holes had been taken from literature.³⁰ In order to rule out non-magnetic parts of the XAS spectra an arctangent based step function is used to fit the threshold.²⁶ As can be seen from Fig. 2, unlike an infinite 'long tail' reported by Goering et al.,27 the integrated spectrum of both XMCD and total XAS of our data quickly saturate at ~ 726 eV. Therefore an integration range to 730 eV is sufficient, giving $m_{spin} = (2.19\pm0.1)$ $\mu_{\rm B}$ /f.u., $m_{\rm orb} = (0.54 \pm 0.05) \mu_{\rm B}$ /f.u. and the $m_{\rm orb}$ / $m_{\rm spin}$ ratio as large as 0.25. It should be noted that all sum rules-related values given here are the average information over the whole formula unit (f.u.) of the three cations contribution. Possible artefact of the experimental set up and data analysis of XMCD of magnetite were discussed in detail by Goering et al..27 In general, the nonmagenetic part of the raw data is smaller than 1/1000 of the total absorption. The saturation effect in our case is estimated to be about 3% in the normal incidence configuration. Besides, the magnetic dipole term $\langle T_z \rangle$ plays a rather insignificant role because of the predominantly cubic symmetry of magnetite, even under a scenario of additional surface symmetry breaking. Besides, the good agreement of the m_{s+l} obtained from SQUID measurement with that calculated from XMCD is an additional proof of the proper application of the sum rules in this study.

IV. Discussion

Unquenched m_{orb} , or strong spin-orbit coupling (*<LS>*), is a desired property in terms of the controllability by electric field in spintronics operation²⁸, however, which have been reported with controversy in magnetite. Early theoretical work based on the picture of bulk Fe₃O₄ possessing $m_{\rm spin}$ = 4.0 $\mu_{\rm B}/{\rm f.u.}$ and nearly vanishing $m_{\rm orb}$. McQueeney et al.²⁹ obtained a high $\langle LS \rangle$ of magnetite of the order of 10 meV, pointing to a large m_{orb} to expect. The XMCD performed by Huang et al.³⁰ suggests a large unquenched m_{orb} , typically 0.67 $\mu_{\rm B}/{\rm f.u.}$ along with a $m_{\rm spin}$ = 3.68 $\mu_{\rm B}/{\rm f.u.}$ at the temperature both above and below Verwey transition. These results are well resembled by calculations using the LDA + U scheme.³⁰ The large $m_{\rm orb}$ has been attributed to a strong on-site Coulomb interaction and corresponding 3dcorrelation effects. Similarly sizable m_{orb} was also observed by Kang et al.³¹ in Mn substitution at the A site, which changes the valence of the B-site Fe and by Babu et al.³² in ultrathin Fe₃O₄ on BaTiO₃(001). By sharply contrast, XMCD performed by Goering et al.²⁷ suggests that there is in fact a vanishingly small m_{orb} on the Fe sites. To avoid the systematic errors arises from the XMCD data analysis, MCS were performed, which still end in controversial results. Non-integral $m_{spin} = 3.54 \ \mu_{B}/f.u.$ and correspondingly $m_{orb} = 0.51 \ \mu_{B}/f.u.$ were observed by Li et al.,³³ while Duffy et al.³⁴ reported again nearly vanishing m_{orb} . Goering et al.¹² has recently tried to explain the large variety of published results by the independent analysis of the Fe $L_{2,3}$ edge XAS, moment analysis fit of the Fe $L_{2,3}$ edge XMCD, and by the comparison with O K edge XMCD. In consistent with Goering's, our data also exhibit an intensity ratio $r_{23} = 0.25$, strongly reduced from a pure statistical case where $r_{23} = 0.5$, which support the presence of large m_{orb} and respective $\langle LS \rangle$ expectation values. According to Goering's argument, these orbital moments are located at the A and B sites of magnetite and aligned antiparallel with each other, similar to the spin moments, though quantitatively, this scenario questions the picture of a fully A site Fe³⁺ and a mixed-valent B site configuration. Table 1 summaries some of the experimental and theoretical efforts within the thin film regime toward this issue. Regardless the controversial reports on various form of magnetite, our results (the first line of the table) suggest the existence of a large unquenched $m_{\rm orb}$ with $m_{\rm orb}$ / $m_{\rm spin}$ = 0.25 in the epitaxial Fe₃O₄ ultrathin film on MgO(100). While the unquenched $m_{\rm orb}$ might be a intrinsic property of the bulk-like Fe₃O₄, which is still a hotly debated topic, our result could also originates from modification of crystal lattice symmetry as ml in the low dimensional magnetic systems can be strongly enhanced by the reduced symmetry of the crystal field as found in the Fe/GaAs(100) system.³⁵

It is worth noting that not only the nanoscale full epitaxial Fe₃O₄/MgO(100) heterostructure exhibits considerably large m_{orb} , its total moment (m_{total}) also retains ~ 70% of the bulk value. The deviation of the magnetic moment of thin films from the bulk-like Fe₃O₄ is usually attributed to three forms of missing compensation or symmetry breaking. The first

one is the formation of antiphase boundaries (APBs) raised from the epitaxy growth process due to the fact that Fe₃O₄ has twice the unit-cell size of MgO.^{36, 37} In magnetite thin films, the magnetic interactions are altered at the APBs, across which the intrasublattice exchange interactions dominate, reversing the spin coupling. Therefore, the structural boundary separates oppositely magnetized regions and the resultant coupling between two domains turns out to be either frustrated or antiferromagnetic. Such antiferromagnetic exchange interactions usually lead to saturation fields as large as 70 kOe,³⁶ but can be neglected for our sample, whose magnetization saturate before 500 Oe (see Fig. 3(b)). The second mechanism of non-compensation occurs due to the less cubic symmetry of magnetite at the surface and the interface of Fe₃O₄/MgO. Among the very few work performed on Fe₃O₄ thin films, Orna et al.³⁸ reported significantly shrinking $m_{spin} = 1.83 \ \mu_B/f.u.in \ Fe_3O_4(8 \ nm)/MgO$, as well as the observation by Babu et al.³² of $m_{spin} = (1.20\pm0.05) \mu_B/f.u.$ in Fe₃O₄(2.5 nm)/BaTiO₃. Even in the bulk, strongly reduced m_{spin} down to (1.73±0.02) $\mu_{\rm B}$ /f.u. was observed by Goering et al..³⁹ By contrast again, large m_{spin} of 7.7 $\mu_B/f.u$ in Fe₃O₄(5 nm)/MgO was reported by Arora *et al.*,¹¹ who attributed the enhancement to the uncompensated spin between A and B sublattices at the surface and across the APBs. However, this enhancement may also come from the magnetic impurities as suggested by Orna et al.³⁸ The inter-diffusion of ions, which tends to substitute onto B-sites, is the third possibility. Although our sample were grown at a moderate growth temperature (350 °C), one may still predict an appreciable inter-diffusion given the ex situ measurement in the study were not carried out immediately after the growth. However, if any, such substitution would only happen at the first 1-2 atom layers at the Fe₃O₄/MgO interface. Therefore results presented in this paper are rather representative for a Fe_3O_4 thin film on MgO without the mixture of Mg²⁺ ions.

IV. Conclusion

To summarize, we have performed XMCD of a Fe₃O₄ epitaxial thin film on MgO(100) synthesized by plasma-assisted simultaneous oxidization MBE process. High quality XAS and XMCD spectra were obtained and carefully analysed with the sum rules. A significant unquenched m_{orb} was observed, which may be an intrinsic property of the Fe₃O₄ or come from the symmetry breaking at the Fe₃O₄/MgO(100) interface. Such sizable m_{orb} has strong implications for adding capacities to spintronics devices, since high *<LS>* coupling is mandatory for realizing the ultrafast switching of spin polarization by electric field and circularly polarized light. Moreover, our 12 nm Fe₃O₄/MgO(100) heterostructure retains a large m_{total} of ~ 70% of the bulk-like stoichiometric Fe₃O₄. Based on the presence of Verwey transition, such large magnetic moments can stay independent from the cation or anion

vacancies and the APBs, if any. Our results offer direct experimental evidence for addressing the open issue of the spin and orbital moments of magnetite, particularly, in its epitaxial ultrathin film form, which is significant for achieving high efficient spin injection and electrical spin manipulation in the full epitaxial spintronic heterostructures.

Acknowledgements

This work is supported by the State Key Programme for Basic Research of China (Grants No. 2014CB921101), NSFC (Grants No. 61274102) and PAPD project, UK EPSRC and STFC. We acknowledge Jill S. Weaver for sharing the computation code of the sum rule calculation.

Tables

Sample	Method	т _{огb} (µ _B / f.u.)	m _{spin} (µ _B /f.u.)	$m_{ ext{total}}$ ($\mu_{ ext{B}}$ /f.u.)	$m_{ m orb}/m_{ m spin}$	Ref.
12 nm Fe ₃ O ₄ /MgO(100)	XMCD	0.54±0.05	2.19±0.1	2.73±0.15	0.25	*
8 nm Fe ₃ O ₄ /MgO/GaAs(100)	XMCD	0.47±0.05	2.84±0.1	3.32±0.15	0.17	17
5 nm Fe ₃ O ₄ /MgO(001)	SQUID			7.7		11
8 nm Fe ₃ O ₄ /MgO(001)	XMCD			1.83	< 0.05	38
2.5 nm Fe ₃ O ₄ /BaTiO ₃ (001)	XMCD	0.44±0.05	1.20±0.05	1.64	0.37	32
Theory	LDA	0.06	4.0	4.06	0.015	30
Theory	LDA+U	0.43	4.0	4.43	0.108	30
Theory	LSDA+U	0.02	3.7	3.72	0.005	2

Table 1. The experimental and theoretical reports of the magnetic moments of magnetite thin
 films and the results of the present study (marked as *).

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Figure 1. Typical RHEED patterns of (a) MgO(100) after annealing, (b) $Fe_3O_4/MgO(100)$, taking during the MBE growth.



Figure 2. The resistance versus temperature relationship. Inset: the resistivity versus the inverse of temperature.



Figure 3. The global magnetization versus (a) temperature. Inset: the gradient of magnetization versus temperature. (b) applied magnetic field. Dash line indicates the value of bulk saturation.



Figure 4. A schematic diagram of the XAS and XMCD experimental configuration utilized in this study.

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Figure 5. A typical XAS pair and XMCD spectra of the magnetite thin film sample obtained at 30 kOe and room temperature.