

Understanding SOEC Degradation Processes by means of a Systematic Parameter Study

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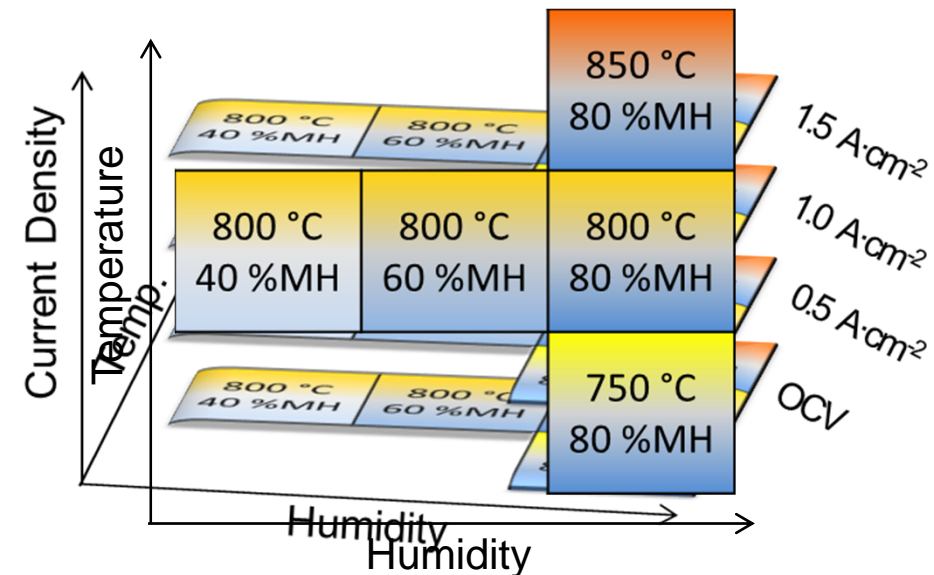


Knowledge for Tomorrow

Experimental concept and data interpretation

Systematic study: operating parameters → degradation

- Temperature (T)
- Fuel gas humidity (MH)
- Current density (i)

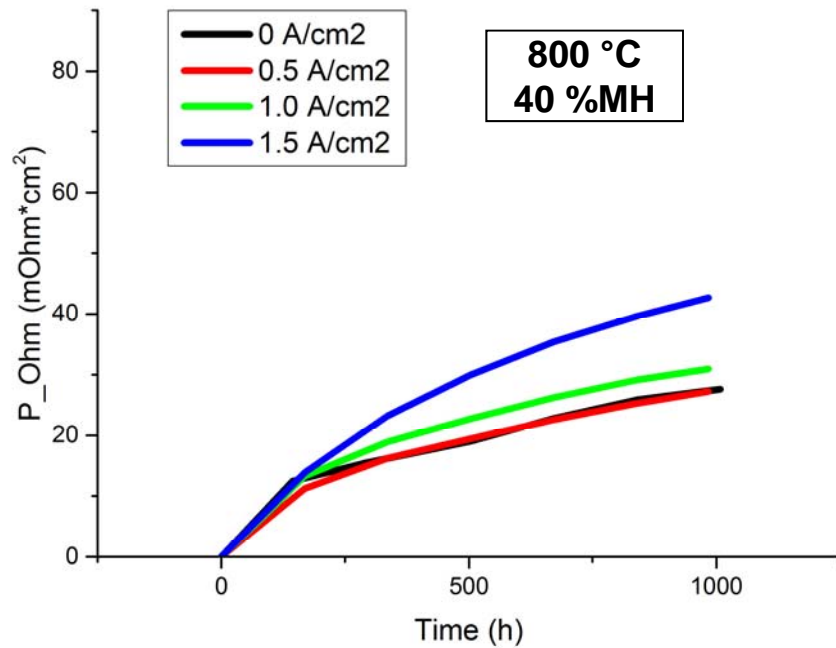


Experimental concept:

- 20 comparable degradation experiments over 1000 h
- Identical measurements except for investigated parameter
- Fuel electrode supported cells from CeramTec (16 cm²)
→ Ni-8YSZ support | Ni-8YSZ | 8YSZ | CGO | LSCF



Degradation results: Ohmic resistance



- Ohmic degradation at all current densities
- Influence of current density only at high current densities
→ effect rather limited

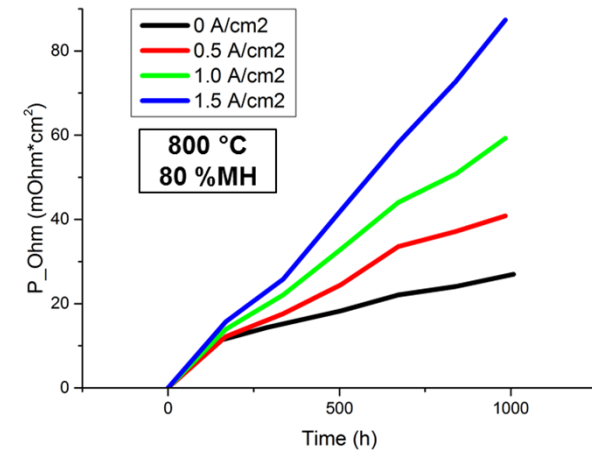
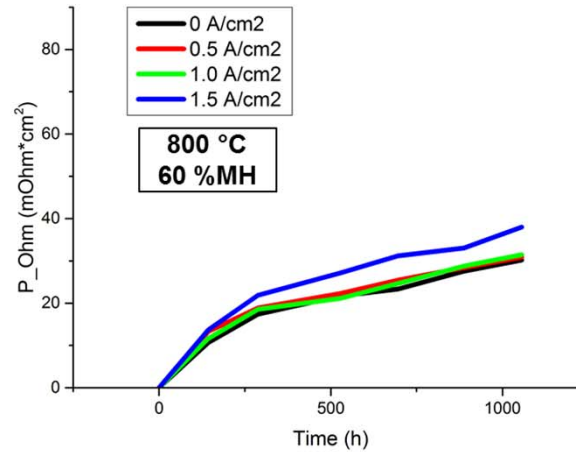
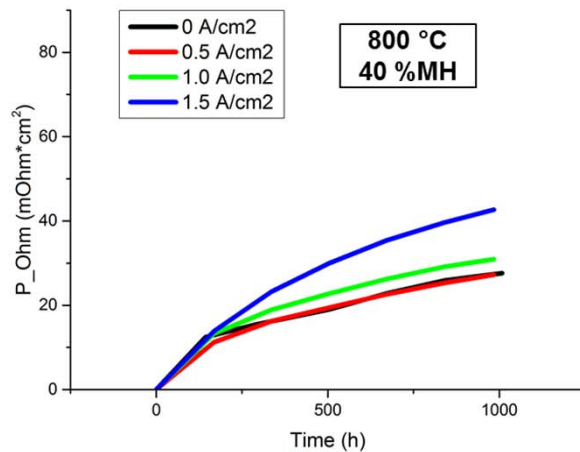


Degradation results: Ohmic resistance

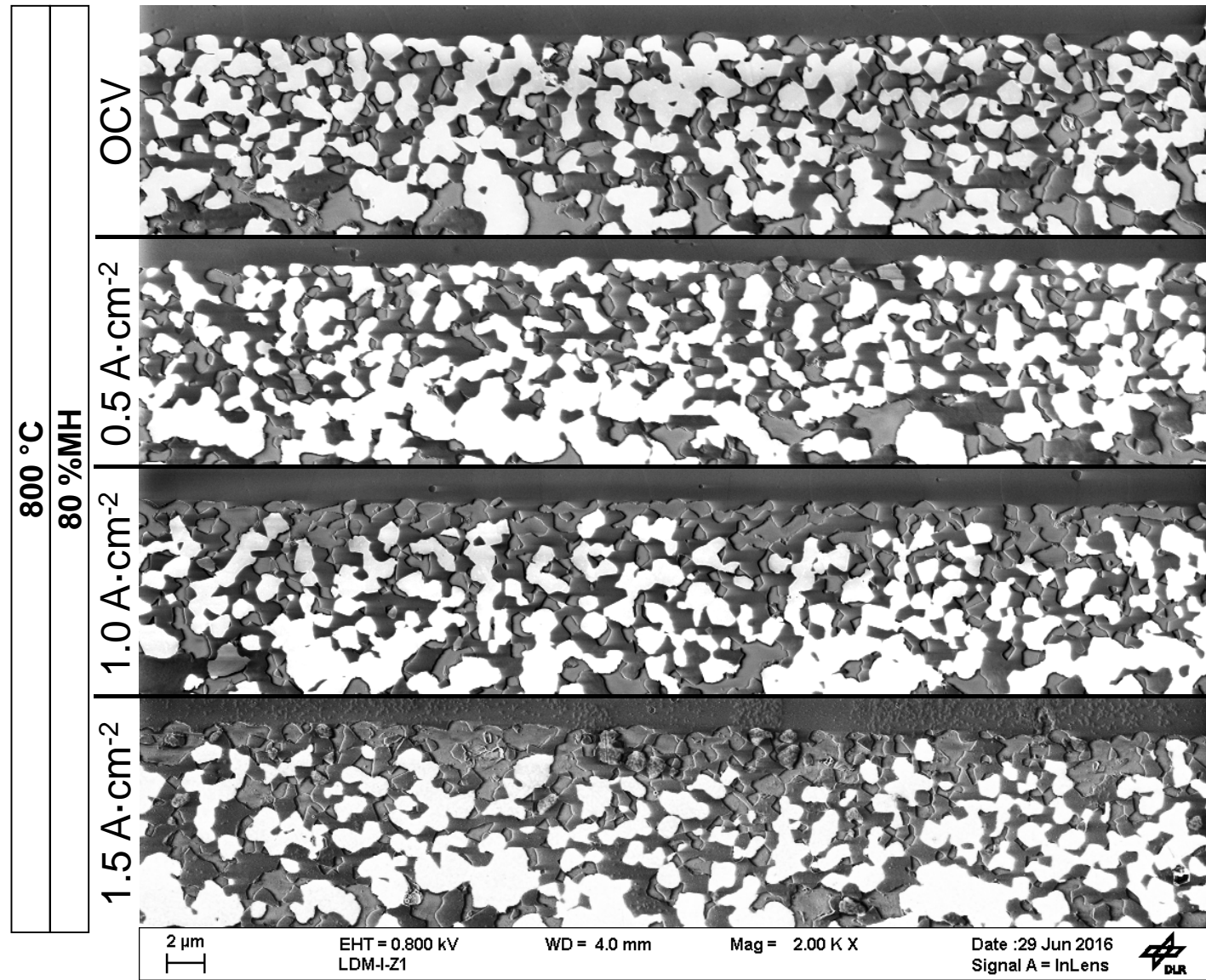
40 %MH and 60 %MH

80 %MH

- Degradation of ohmic resistance similar
- Influence of current density much stronger
- Current density has effect even at low current densities



Degradation results: Ohmic resistance



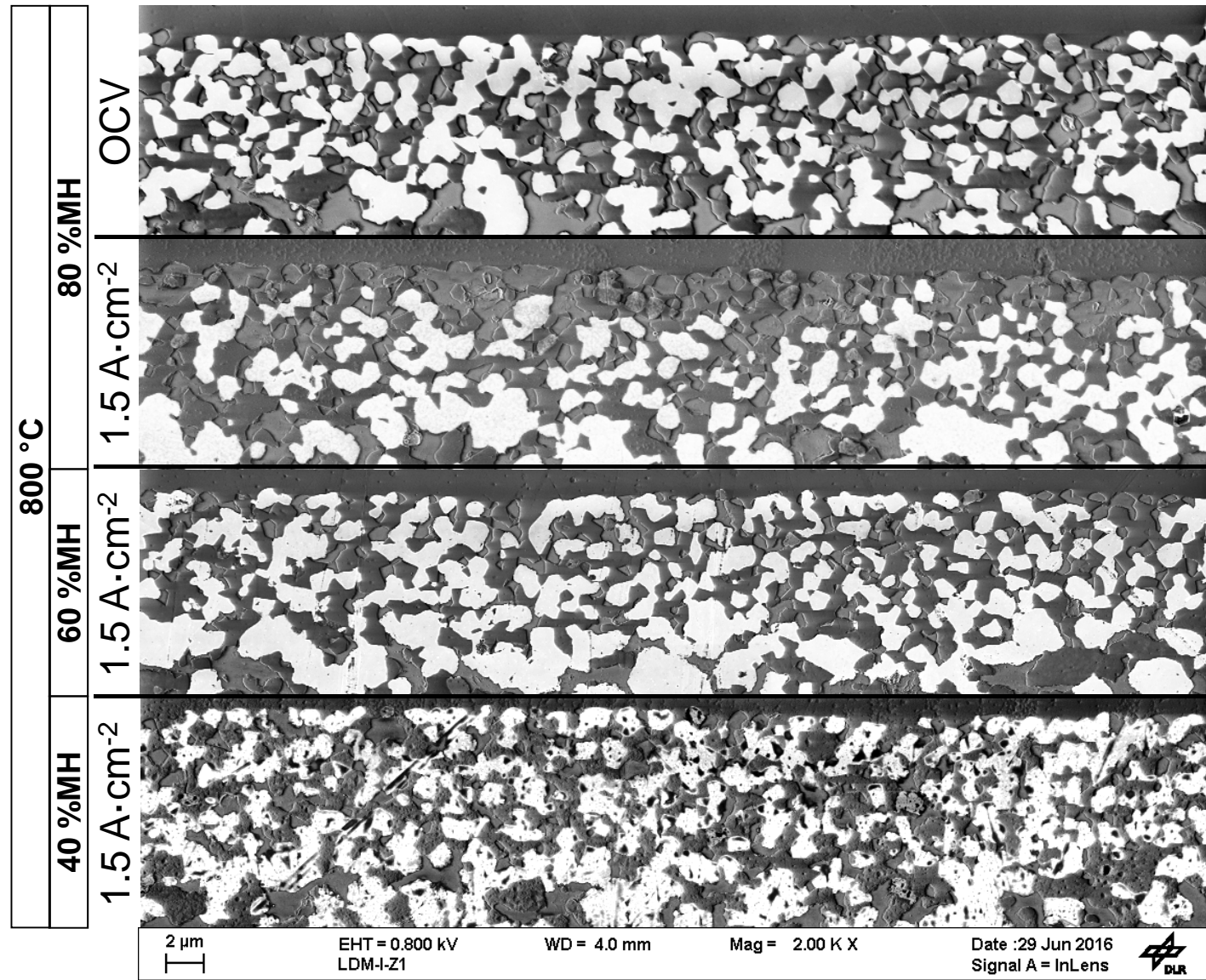
Ni-depletion at Electrode|Electrolyte interface

→ Effectively increasing electrolyte thickness

Direct correlation between Ni-depletion and current density



Degradation results: Ohmic resistance



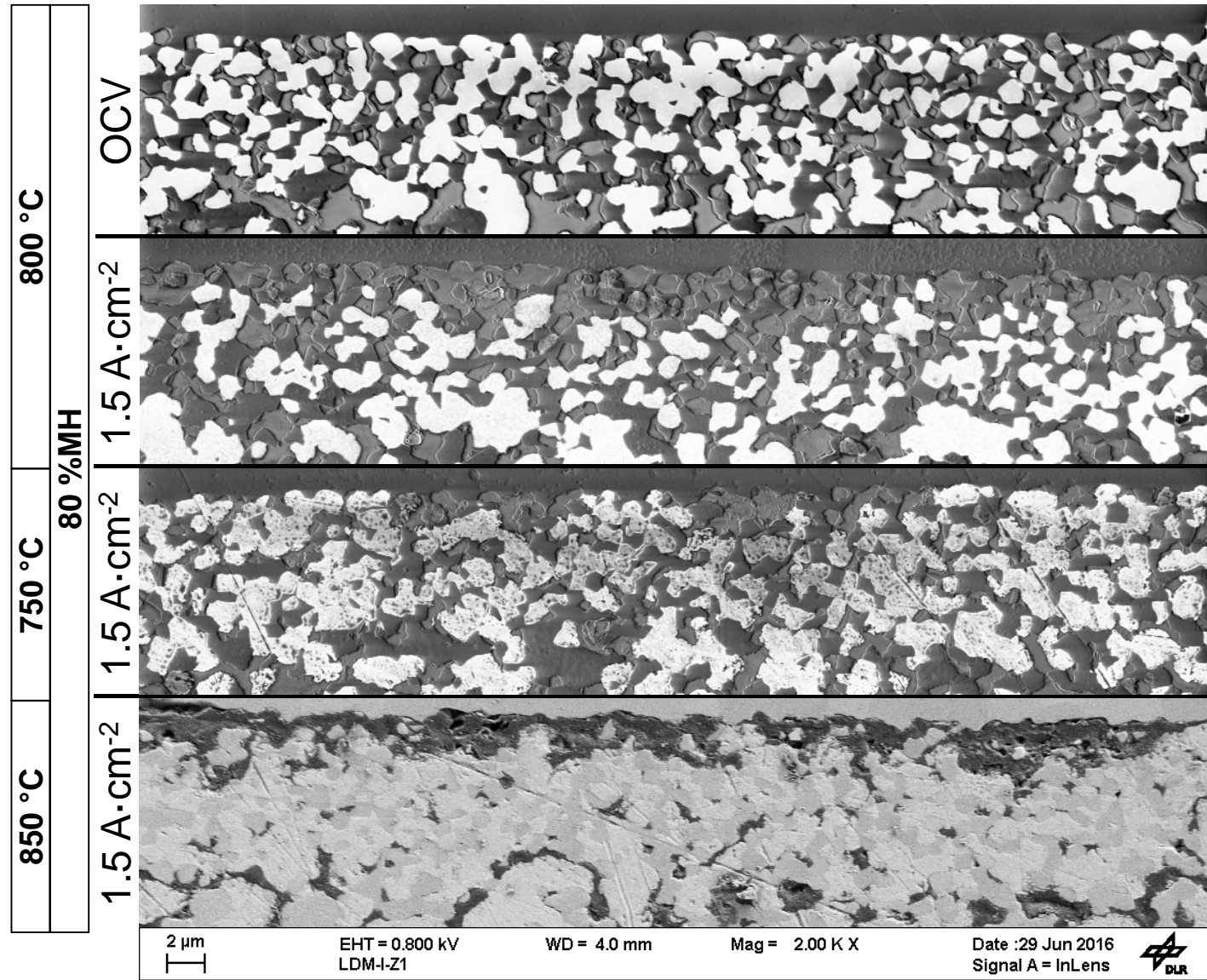
Ni-depletion at Electrode|Electrolyte interface

→ Effectively increasing electrolyte thickness

No Ni-depletion at lower humidity



Degradation results: Ohmic resistance



Ni-depletion at Electrode|Electrolyte interface

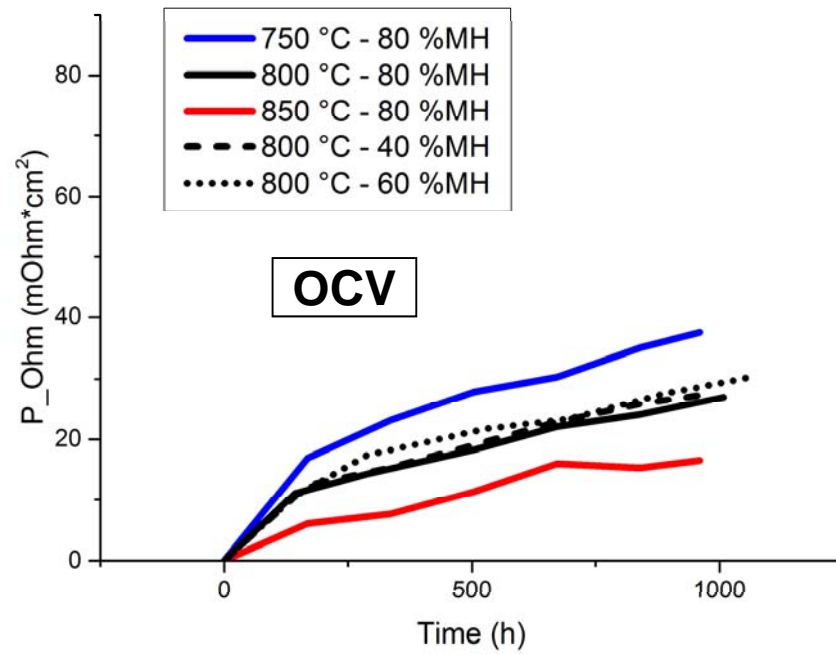
→ Effectively increasing electrolyte thickness

Less Ni-depletion at lower temperature

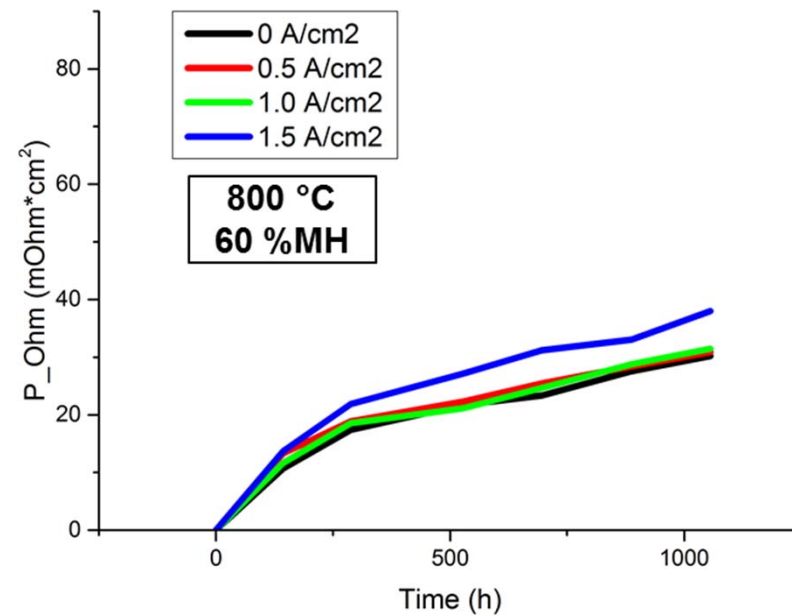


Degradation results: Ohmic resistance

“Underlying” Degradation:



- Higher temperature leads to lower degradation
- Independent of humidity
- Little influence of current density



Degradation results: Ohmic resistance - Summary

Two major degradation processes

Ni-Depletion:

- Direct correlation between current density and Ni-Depletion
- Minimum temperature and humidity threshold

“Underlying” Degradation:

- Temperature dependence: higher temperature → lower degradation
- No influence of humidity or current density

Deterioration of YSZ integrity

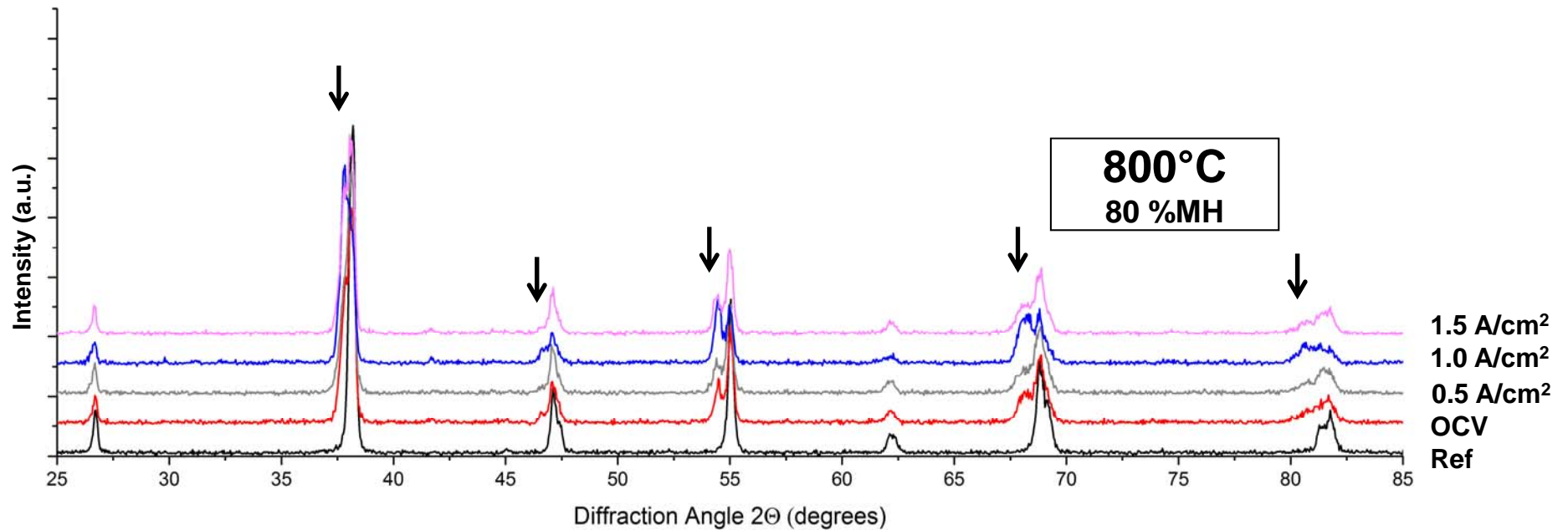
- Weakening of YSZ structure at high current densities

Not observed

- Correlation between YSZ deterioration and significant increase in ohmic resistance
- SrZrO₃ formation



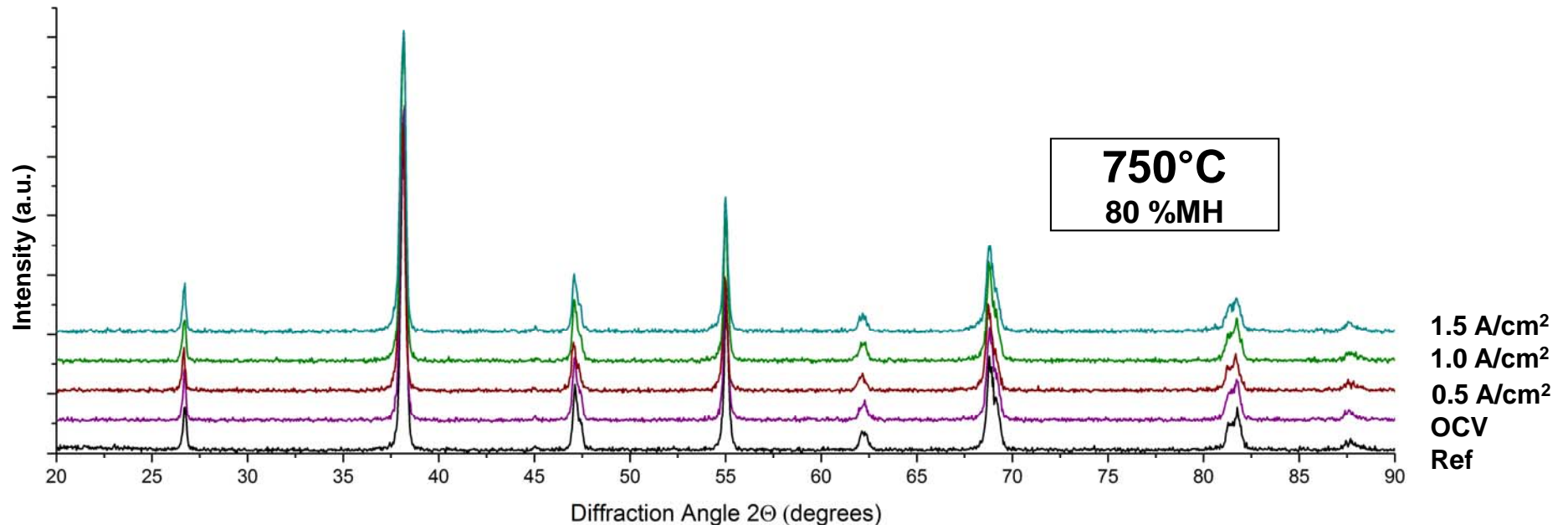
Degradation results: Oxygen Electrode



- New peaks in XRD pattern → suggest formation of new crystalline phase
- Observable at all current densities, but no clear trend
- Similar at 800 °C and 850 °C



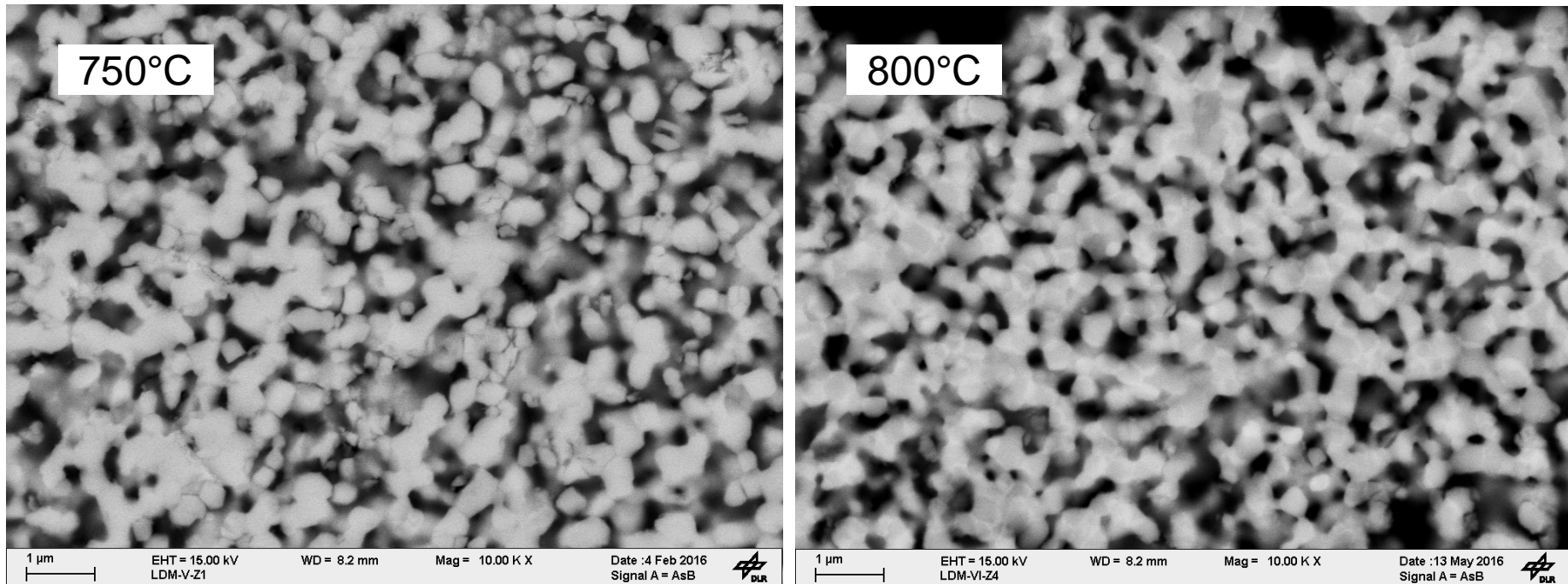
Degradation results: Oxygen Electrode



- New peaks in XRD pattern → suggest formation of new crystalline phase
- Observable at all current densities, but no clear trend
- Similar at 850 °C and 800 °C
- Not detectable at 750 °C



Degradation results: Oxygen Electrode



- Change in phase composition observable in BSE-SEM
- Correlates with new Peaks on XRD-Pattern
- Correlate with degradation of electrochemical activity



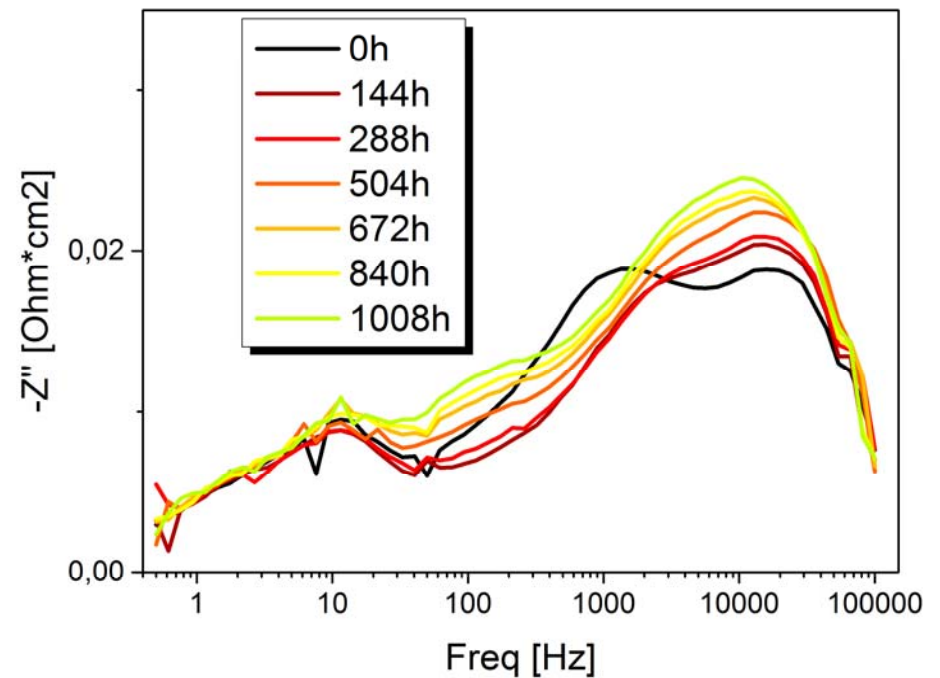
Degradation results: Fuel electrode

- Generally less significant
- Generally stronger at higher current density
- Ni-agglomeration at 850°C

Depending on conditions even activation!

Two fuel electrode processes
→ Behave differently

Discussion wanted!



Summary

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- Ohmic degradation dominates overall degradation
- Two major ohmic degradation processes:
 - Ni-depletion: $f(i)$ above T and humidity threshold
 - “Underlying” Degradation: lower at higher temperatures, $f(i, \text{humidity})$
- *Changes in the oxygen electrode:*
 - *XRD and BSE-SEM images show change of phase composition*
 - *Correlates with degradation of electrochemical activity*
- Fuel electrode degradation:
 - Stronger at higher current densities
 - Ni-agglomeration at high T

Thank you for your attention



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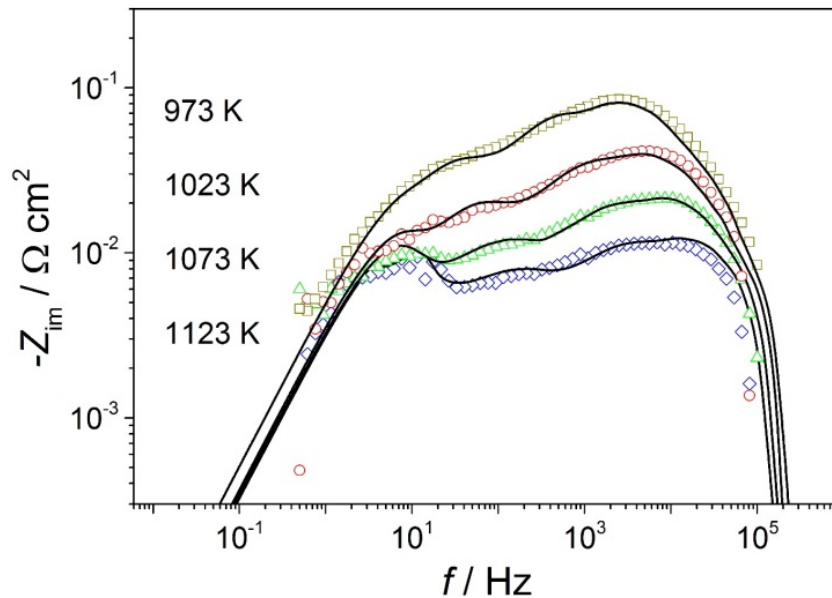
Thank you for your attention



Experimental concept and data interpretation

Understanding and interpreting impedance data:

- 1) DRT-Analysis
- 2) Use of a physico-chemical model (B0811)



Measured and simulated impedance spectra at various temperatures, at OCV and 60 %MH

→ Better identification and understanding of individual rate limiting processes

