Dissolved nitrous oxide (N_2O) dynamics in agricultural field drains and headwater streams in an intensive arable catchment

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9 ABSTRACT

Indirect nitrous oxide (N₂O) emissions produced by nitrogen (N) leaching into surface water and groundwater bodies are poorly understood in comparison to direct N₂O emissions from soils. In this study, dissolved N₂O concentrations were measured weekly in both lowland headwater streams and subsurface agricultural field drain discharges over a two-year period (2013–2015) in an intensive arable catchment, Norfolk, UK. All field drain and stream water samples were found to have dissolved N₂O concentrations higher than the water-air equilibrium concentration, illustrating that all sites were acting as a net source of N₂O emissions to the atmosphere. Soil texture was found to significantly influence field drain N₂O dynamics, with mean concentrations from drains in clay loam soils (5.3 μ g N L⁻¹) being greater than drains in sandy loam soils (4.0 μ g N L⁻¹). Soil texture also impacted upon the relationships between field drain N_2O concentrations and other water quality parameters (pH, flow rate, and nitrate (NO_3) and nitrite (NO_2) concentrations), highlighting possible differences in N₂O production mechanisms in different soil types. Catchment antecedent moisture conditions influenced the storm event mobilisation of N₂O in both field drains and streams, with the greatest concentration increases recorded during precipitation events preceded by prolonged wet conditions. N_2O concentrations also varied seasonally, with the lowest mean concentrations typically occurring during the summer months (JJA). Nitrogen fertiliser application rates and different soil inversion regimes were found to have no effect

on dissolved N₂O concentrations, whereas higher N₂O concentrations recorded in field drains under a winter cover crop compared to fallow fields revealed cover crops are an ineffective greenhouse gas emission mitigation strategy. Overall, this study highlights the complex interactions governing the dynamics of dissolved N₂O concentrations in field drains and headwater streams in a lowland intensive agricultural catchment.

33 Keywords: nitrous oxide; nitrate; nitrification; denitrification; river; tile drain.

1. Introduction

Nitrous oxide (N_2O) is a potent and persistent greenhouse gas with a present atmospheric concentration of 326.7 ppb (European Environment Agency, 2016). N₂O has 300 times greater global warming potential than CO_2 and accounts for ~5% of the total greenhouse effect (Omonode et al., 2011). N₂O participates in photochemical reactions in the stratosphere which lead to the destruction of the ozone (O_3) layer (Jacinthe and Dick, 1997) and is also linked to the release of nitric oxide and ammonia in the atmosphere which contribute to acid rain and the acidification of soils and drainage systems (Mosier and Kroeze, 1998). Atmospheric N₂O concentrations are increasing at an annual rate of ~0.26% (Forster et al., 2007) and agriculture is the largest source of anthropogenic N_2O emissions, accounting for ~60% of the total human-produced N₂O. Globally, agricultural N₂O emissions increased by nearly 17% from 1990 to 2005 (Smith et al., 2007).

Global agricultural N₂O emissions originate from three sources: direct emissions from nitrogen (N) fertilised soil (1.8 Tg N a^{-1}); direct emissions from animal production (2.3 Tg N a^{-1}); and indirect emissions from N used in agricultural systems (1.3 Tg N a^{-1}) (Syakila and Kroeze, 2011). Indirect emissions are in turn derived from three sources: atmospheric deposition (0.4 Tg N a^{-1}); human sewage (0.3 Tg N a^{-1}); and nitrogen leaching and runoff into water bodies (0.6 Tg N a^{-1}). Thus, indirect emissions from N leaching and runoff are

significant, equating to 46% of the indirect emissions from agriculture and 11% of total agricultural N₂O emissions. These indirect emissions from N leaching and runoff are equivalent to ~33% of the direct N₂O emissions from soils. Other studies (e.g. Mühlherr and Hiscock, 1997; Seitzinger and Kroeze, 1998; Naqvi et al., 2000) have previously highlighted the significance of N₂O emissions from agricultural N leaching and runoff to the overall N₂O budget. However, compared to direct N₂O emissions, these indirect emissions have been poorly studied to date (Outram and Hiscock, 2012).

Indirect emissions of N₂O occur when N fertiliser is lost from agricultural soils through leaching and runoff. This mobilised N enters groundwater, rivers, riparian areas, wetlands and eventually the ocean (Mosier et al., 1998). Nitrogen in groundwater and surface waters increases the biological production of N₂O as the N undergoes both nitrification and denitrification. Nitrification is an aerobic chemolithoautotrophic process in which bacteria oxidise ammonium (NH_1^+) to nitrate (NO_3) and N_2O . Denitrification is an anaerobic process in which bacteria sequentially reduce NO₃ to N₂O and dinitrogen (N₂) with a small amount of N_2O escaping in the process (Beaulieu et al., 2008). Therefore, the factors controlling nitrification and denitrification in the soil, such as moisture content, temperature, organic matter, availability of N and NH₄⁺, pH, redox conditions, texture and agricultural management practices may subsequently control both direct and indirect N₂O emissions (Bouwman et al., 1993; Panek et al., 2000). However, the majority of existing studies (e.g. Włodarczyk et al., 2005; Jarecki et al., 2009; Hénault et al., 2012) have only investigated the effect of these parameters on direct N_2O emissions, leaving the control of these parameters on indirect N_2O emissions poorly investigated in the literature. The aim of this study was to address these deficiencies through the following objectives:

 to investigate the spatial and temporal dynamics of dissolved N₂O concentrations in subsurface agricultural field drains and lowland headwater streams;

ii. to assess the impact of water quality parameters, soil texture and crop cultivation
regimes on dissolved N₂O concentrations.

2. Methods

80 2.1 Studylocation

The study area is located to the northwest of Norwich in the Blackwater sub-catchment of the River Wensum, Norfolk, UK (Figure 1). The Blackwater sub-catchment is intensively monitored as part of the Wensum Demonstration Test Catchment (DTC) project which aims to evaluate the extent to which on-farm mitigation measures can cost-effectively reduce the level of diffuse agricultural pollution (McGonigle et al., 2014). The field sites are in an area of intensive arable farming which includes cereals, oilseed rape, spring beans and sugar beet grown in a seven-year rotation. The average annual rainfall total is 674 mm and the mean annual temperature is 10.1°C. The soil parent material comprises glacial deposits over Cretaceous Chalk, with soil textures varying from sandy loam to sandy clay loam and clay loam. Part of the sub-catchment is extensively under-drained by a dense network of subsurface agricultural field ("tile") drains installed at a depth of 100-160 cm. Discharge from certain drains can be as high as 10 L s^{-1} , but varies greatly depending upon season, depth, catchment area and antecedent moisture conditions.

In 2013, nine fields covering 143 ha of arable land were identified for the trialling of a winter cover crop and reduced tillage practices aimed at reducing diffuse nutrient losses into the River Blackwater (Figure 1 and Table I). These nine fields were divided into three blocks of mitigation measures, with each block sown with the same crop and same fertiliser application rate during the 2013/14 (spring beans; 0 kg N ha⁻¹, 30 kg P ha⁻¹, 55 kg K ha⁻¹) and 2014/15 (winter wheat; 220 kg N ha⁻¹, 22 kg P ha⁻¹, 85 kg K ha⁻¹) farm years (September to August). Two fields were kept as a control (block J) and were cultivated by mouldboard ploughing to 101 25 cm depth prior to sowing. An oilseed radish (*Raphanus sativus*) cover crop (seed density = 102 18 kg ha⁻¹) was sown in treatment blocks L and P in late-August 2013. Block P then 103 underwent reduced tillage to a depth of 10 cm prior to sowing spring beans and block L 104 underwent direct drilling with no inversion. A winter crop (winter wheat) was grown in the 105 second year (2014/15) and so a cover crop was not sown, but direct drilling and reduced 106 tillage practices continued in block L and block P, respectively.

2.2 Sample collection

Water samples for N₂O and nutrient analysis were collected in 13 field drains and 4 stream locations (A, B, E and M) on a weekly basis between April 2013 and April 2015 (Figure 1). Over the duration of the study, 621 water samples were collected from field drains and 308 samples from stream sites, such that 929 samples were obtained in total. Samples for dissolved N₂O analysis were collected from the stream and drain outlet pipes in 20 mL glass syringes (SAMCO) with a three-way stopcock attached to each syringe by a Luer-Lock fitting. Syringes were flushed three times with water from the sampling point and any air bubbles contained in the syringes were expelled before the final sample was taken. No preservative was added to the sample. Samples were returned to cold storage at 4°C within 3 h and analysed for N₂O within 72 h of collection. Samples for nutrient analysis were collected in 1 L polypropylene bottles and also analysed within 72 h of collection. Rainfall was measured via a tipping bucket rain gauge installed in mini-catchment A. Soil samples for texture analysis were collected in May and September 2013 from 12 locations per field in a "W' layout at 0-30 cm depth using a Dutch auger and Hydrocare powered auger (Figure 1). Catchment areas of the field drains were divided into two dominant soil types (clay loam and sandy loam) based on the soil texture data collected in this study. Drains D3, D7, D8, D9,

124 D10 and D13 were predominantly within clay loam soils, whereas drains D1, D2, D4, D5,

2.3 Sample analysis

 N_2O was analysed by gas chromatography with an electron capture detector (GC-ECD). Samples were injected directly into a purge column of a helium flushed gas extraction line, which included traps for water vapour and CO₂ removal. A reverse-flow Nafion dryer was used in the extraction line with N₂O. N₂O was trapped at -190 °C above liquid nitrogen in a 3.2 mm stainless steel loop. The N₂O was remobilised by submerging the loop in a hot water bath at ~95 °C. The collected N₂O was analysed with a Shimadzu GC-8A at 300 °C using a 3.6 m by 3.2 mm diameter stainless steel Poropak Q column at 50 °C. The accuracy of N₂O measurements was within $\pm 3\%$ with a detection limit of ~0.0008 µg N L⁻¹. Further details of this method can be found in Mühlherr and Hiscock (1998). Nitrate was determined by ion chromatography using a Dionex ISC 2000 instrument with an accuracy of 0.19 mg N L⁻¹. Ammonium and nitrite were determined by a Continuous Flow Analyser - Skalar San++ with accuracies of 4.57 μ g N L⁻¹ and 1.52 μ g N L⁻¹, respectively. Soil texture was determined by laser diffraction.

For data analysis, the independent-sample *t*-test was used to examine the degree of significance (*p*-value < 0.05) between two groups, whilst multiple linear regression models for the prediction of N₂O concentration in stream and field drain waters were formulated in the *R* environment (R Core Team, 2016).

- **3. Results and Discussion**
- **3.1 Spatial variability of nitrous oxide concentrations**

147 N₂O concentrations in field drain (n = 621) and stream (n = 308) water samples collected 148 throughout this study are presented in Figure 2. All of the drain and stream water samples

were found to have dissolved N₂O concentrations higher than would be expected when atmospheric N₂O concentrations are at equilibrium with water (~0.36 μ g N L⁻¹; Weiss and Price, 1980), illustrating that all sites were acting as a net source of N₂O emissions to the atmosphere. Field drain N₂O concentrations ranged from 0.4 µg N L⁻¹, just above the atmospheric-water equilibrium, to 34.4 μ g N L⁻¹, 100 times greater than atmospheric-water equilibrium. Mean N₂O concentrations in drains within clay loam and sandy loam soils were 5.3 and 4.0 μ g N L⁻¹, respectively. Among the drains, D11 had the highest mean value (8.0 μ g N L⁻¹) and D2 the lowest (2.7 μ g N L⁻¹), with both draining sandy loam soils in block L.

There have been very few previously published studies of dissolved N₂O concentrations in field drains. Dowdell et al. (1979), who studied dissolved N₂O in agricultural drains for the first time, found a range of 1–132 μ g N L⁻¹ in three different locations across southern England. In a study of N₂O discharged from 28 drained agricultural areas in the upper Neckar region, Germany, Hack and Kaupenjohann (2002) observed a N₂O range of 0.4-60 µg N L⁻¹, whilst Reav et al. (2004) recorded a narrow range in N₂O concentration (2–4 μ g N L⁻¹) in one particular field drain under arable land planted with spring barely in Scotland over a 45 day period. Similar to the differences in soil texture reported here, differences in groundwater N₂O in the unsaturated zone between sites with contrasting geology was reported by Darling et al. (1998), who recorded mean N₂O concentrations of 2.8 and 1.5 µg N L⁻¹ in UK Chalk and Sandstone aquifers, respectively. Thus, the N₂O concentrations $(0.4-34.4 \ \mu g \ N \ L^{-1})$ measured in this study are within the range previously reported in the literature.

Across all stream samples, a mean N_2O concentration of 1.4 µg N L⁻¹ was measured with a range of 0.36–7.3 μ g N L⁻¹ (1–20 times greater than the atmospheric–water equilibrium).

Among the stream sampling sites, site M had the lowest mean concentration $(1.0 \ \mu g \ N \ L^{-1})$

and site B the highest $(1.8 \mu g N L^{-1})$. Compared to the field drains, stream samples contained

significantly (p < 0.05) lower N₂O concentrations, a consequence of N₂O being rapidly

degassed from field drain water once it has come into contact with the atmosphere upon discharge into the stream. This degassing of supersaturated N₂O from subsurface drainage and groundwater after discharge to surface water has also been reported in previous studies (e.g. Bowden and Bormann, 1986; Reay et al., 2003; Minamikawa et al., 2011; Li et al., 2013). For comparison, in a 13-month study at nine sites on the eutrophic San Joaquin River, California, Hinshaw and Dahlgren (2013) reported a mean dissolved N₂O concentration in surface waters of 0.91 µg N L⁻¹, whilst Outram and Hiscock (2012) recorded a mean N₂O concentration of $1.7 \mu g N L^{-1}$ in the lowland River Thurne, eastern England.

3.2 Temporal variability of nitrous oxide concentrations

183 3.2.1 Annual trends

The temporal variability in field drain and stream water N₂O concentration is presented in Figure 3. Gaps in the measurement of drain N₂O concentration are due to a lack of drain flow during the summer/autumn. As summer 2013 was approaching, N₂O concentrations gradually decreased in all drains, likely due to both drier antecedent conditions and a decline in potentially leachable nitrate due to crop uptake in this period. Drain samples contained lower N₂O concentrations in summer 2013 than summer 2014, possibly reflecting the lower rainfall totals in 2013 (106 mm) compared with 2014 (194 mm) reducing the amount of soil N flushing. Although high rainfall totals were recorded in autumn 2013 (244 mm), including the largest storm event in mid-October 2013 in which 68 mm fell in one week, N₂O concentrations in field drains remained low with no obvious peak corresponding to this storm event. In most drains, the low N₂O concentrations continued throughout winter 2013/14 and spring and summer 2014 with a slight gradual increase. This trend may relate to most of the drains being under fields planted with a spring bean crop which received either no N fertiliser or only 30 kg N ha⁻¹, thus limiting the availability of N for leaching into the subsurface

drainage network. A pronounced increase in N₂O concentration did, however, occur in autumn 2014 and winter 2014/15 when the highest values of the study period were recorded in drains D7 (32.9 μ g N L⁻¹) and D8 (34.4 μ g N L⁻¹) in clay loam soils. These higher N₂O concentrations under winter wheat continued throughout winter 2014/15, such that levels were considerably higher than they had been in the previous year (2013/14).

Temporal variability in the N₂O concentration of stream water was not as apparent as in the drain samples, most likely due to stream water being a composite of water originating from several different sources (e.g. groundwater, field drains and fresh rainwater) with differing N₂O concentrations. However, elevated stream N₂O concentrations (4.7–7.3 μ g N L⁻¹) were recorded in autumn 2014, corresponding with the higher concentrations observed in the field drains during this period.

No significant increase in N₂O concentration of either field drains or stream waters were observed during periods of N fertiliser application, indicating the absence of any direct linear relationship between N application and N loss as N₂O (Figure 3), Reav et al. (2004) noted some effect of N application, with a positive response recorded for several days after each application event, followed by an eventual decline in concentration around two weeks after application. However, Reay et al. (2004) concluded that both the spatial and temporal complexity of the processes responsible for N₂O production in agricultural drainage waters make a straightforward relationship between N₂O concentration and N application rate

217 unlikely, as was found to be the case in this study.

3.2.2 Storm events

N₂O concentrations in the field drains and streams responded differently to the three main
storm events that occurred during the study period (Figure 3). The largest rainfall event
(event 1), which yielded a weekly rainfall total of 68 mm, occurred in mid-October 2013

when the catchment had experienced dry antecedent conditions, with low stream flows (~0.005 m³ s⁻¹ at site A) and limited rainfall (3 mm) during the 14 days prior to the event. This event initiated no significant change in the N₂O concentration of either the flowing drains or stream water, although most of the dry drains did start flowing after event 1. The second event (event 2) during late May 2014, in which 62 mm of rainfall was recorded in one week, also produced no significant increase in the N₂O concentrations of most field drain and stream sites, with the exception of site M (0.5 to 1.6 μ g N L⁻¹), D1 (1.3 to 5.9 μ g N L⁻¹) and D8 (2.5 to 7.9 μ g N L⁻¹). Dry antecedent conditions had again preceded this event, with low stream flows (~ 0.014 m³ s⁻¹) and 0 mm of rainfall recorded in the 7 days prior to the event. In contrast, the storm event in mid-October 2014 (event 3), in which 54 mm of rainfall fell in one week, resulted in a pronounced rise in N₂O concentrations in all flowing field drains and stream sites. This event occurred during a period of wetter antecedent conditions in which 30 mm of rainfall had fallen in the 14 days prior to event 3 and the average stream flow was $0.021 \text{ m}^3 \text{ s}^{-1}$. The highest N₂O concentrations recorded throughout the monitoring period at all four stream locations occurred during storm event 3, which may also be associated with nitrification of residual soil nitrate post-harvest. N₂O concentrations at site M, for example, did not exceed 2.5 μ g N L⁻¹ in the previous 18 months of data collection, but during event 3, a concentration of 7.1 μ g N L⁻¹ was measured. N₂O concentrations in the field drain samples also peaked in mid-October, but this was less pronounced as most of the drains were not flowing prior to this rainfall event.

Overall, these results indicate that catchment antecedent moisture conditions influence the storm event mobilisation of N_2O into stream and field drain waters, with wetter conditions prior to an event typically resulting in elevated N_2O concentrations. There is some consistency here with the study by Reay et al. (2004) who found no clear relationship between field drain N_2O concentration and rainfall which they argued might be due to time 247 lags between rainfall and the resulting impact on dissolved N₂O concentrations. Such time
248 lags are themselves likely to be extremely variable depending upon antecedent moisture
249 conditions and due to the spatial heterogeneity of soil N processing.

²⁵⁰ **3.2.3 Seasonal trends**

To evaluate seasonal changes in field drain and stream water N₂O concentrations, all samples collected in a particular season were combined for spring (MAM), summer (JJA), autumn (SON) and winter (DJF) months (Figure 4). In all seasons, N₂O concentrations were significantly lower in stream samples than in field drains due to the rapid degassing of N₂O from the drain water once in contact with the atmosphere. In stream waters and field drains in sandy loam soils, N₂O concentrations were significantly (p < 0.05) lower during summer than any other season, with mean concentrations of 1.0 and 2.3 μ g N L⁻¹, respectively. Additionally, in field drains under sandy loam soils, N₂O concentrations were significantly lower in autumn (3.2 μ g N L⁻¹) than during winter (4.7 μ g N L⁻¹) or spring (4.8 μ g N L⁻¹). These low summer and autumn concentrations likely reflect a combination of drier antecedent conditions and increased nutrient uptake by crops during the growing season reducing the flushing of leachable soil NO_3 and thus reducing the pool of available N for conversion into N₂O. Lower summer N₂O concentrations in field drains and headwater streams has previously been reported for other arable catchments in southern Germany and Michigan, respectively, sites which have differing soil types and rainfall regimes to the study

266 presented here (Hack and Kaupenjohann, 2002; Beaulieu et al., 2008).

Lower summer N₂O levels were not apparent in field drains under clay loam soils where a high mean concentration (10.1 μ g N L⁻¹) was recorded. However, the number of samples for summer clay loam drains was small (n = 7) and the mean N₂O concentration was biased by very high concentrations discharging from just one drain (D10) at this time. Nevertheless,

further contrasts between field drains in clay loam and sandy loam soils were apparent during the autumn, when substantially higher mean concentrations under clay loam soils (5.9 µg N L^{-1}), particularly after the October 2014 storm event, indicate greater N₂O production and release from clay soils early in the hydrological year.

3.3 Potential factors controlling N₂O concentrations

3.3.1 Soil texture

Soil texture strongly influenced field drain N₂O concentrations, with a mean N₂O concentration under clay loam soils (5.3 μ g N L⁻¹) significantly (p < 0.05) higher than drains under sandy loam soils (4.0 μ g N L⁻¹) (Figure 5). This difference was largely driven by drains D7, D8, D9 and D10 in clay loam soils having high mean N_2O concentrations, whilst drains D2 and D16 in sandy loam soils had much lower concentrations. This was particularly the case during autumn 2014 and winter 2014/15 when field drain N_2O concentrations were substantially higher in clay loam soils (Figure 3). However, drains within sandy loam soils did not always have low N₂O concentrations, as was the case for D4 and D11 which both had high N₂O concentrations. Nevertheless, the data presented here suggests that drains within clay loam soils have the potential to yield higher N₂O concentrations than drains within sandy loam soils.

Very few of the published studies that investigated field drain N₂O concentrations considered

soil texture as a potential controlling factor, thus direct comparison with the results presented

here is difficult. One example is Jahangir et al. (2013), who observed that mean N_2O

concentrations in groundwater at agricultural sites with high permeability soils (sandy clay

loam and sandy loam) were significantly higher than low permeability soils (silty clay loam

and clay loam), in contrast to the findings presented here. In terms of direct N₂O emissions,

numerous studies have assessed the effects of soil texture. Rochette et al. (2008) stated that in

fine textured soils, higher N₂O emissions are often observed as a result of reduced oxygen levels within the soil matrix due to poor drainage. Włodarczyk et al. (2005) emphasised that soil texture and particle size distribution significantly affected the production of N_2O and concluded that heavier soils provided more favourable conditions for N₂O production than sandy soils. It has also been reported that the increased frequency of anaerobic conditions associated with higher water contents in heavier soils favours the production of N₂O by denitrification (Hénault et al., 2012). Therefore, assuming that dissolved N₂O concentrations are consistent with direct N₂O emissions from soils, the higher N₂O concentrations recorded here under heavier clay loam soils may be explained by increased anoxia caused by poorer soil drainage than in areas of sandy loam soils.

3.3.2 Drain flow rate

N₂O is highly soluble in water and so field drains with higher flow rates are expected to export higher loads of dissolved N₂O. However, the relationship between N₂O concentration and flow varied greatly among the drains (Figure 6), D2 (r = 0.77) and D1 (r = 0.75) had very strong positive correlations, whereas D10 (r = -0.35) and D8 (r = -0.05) had weak negative correlations. Figure 6 demonstrates that this variability is partially due to differences in soil texture across the study site, with drains in sandy loam soils having a stronger positive correlation (r = 0.24; p < 0.05) between the two parameters than drains in clay loam soils (r =0.06; p > 0.05). This is supported by the fact that the two drains with the strongest positive correlation (i.e. D1 and D2) were in sandy loam soils and the two drains with the strongest

negative correlation (i.e. D8 and D10) were in clay loam soils. However, there were exceptions to this, with drain D13 in a clay loam soil having a strong positive correlation (r= 0.60) and D4 in a sandy loam soil having a weak negative correlation (r = -0.04). Whilst overall there is no clear and dominant relationship between N₂O concentration and field drain

flow rate, the data presented here nevertheless demonstrates that soil texture does exerts some
 controlling influence upon this relationship.

³²¹ **3.3.3 pH**

As with flow rate, Figure 6 reveals that soil type affected the relationship between field drain N₂O concentration and pH. The pH values of the field drains ranged from 3.7 to 8.6, with a mean value of 7.7 and a 95% confidence interval for the mean of 7.67–7.76. A statistically significant, negative correlation (r = -0.25, p < 0.05) was established between pH and dissolved N₂O in clay loam soils, whereas a weaker negative correlation (r = -0.13; p < 0.05) was observed in sandy loam soils. Hénault et al. (2012) previously identified pH as one of the key soil parameters which significantly influences direct N_2O emissions, suggesting that N_2O emissions from acidic soils generally exceed those from alkaline soils due to higher N₂O emissions from nitrification and/or higher N₂O:N₂ ratios at lower pH levels. Weslien et al. (2009) also observed that soil N₂O emissions were significantly and negatively (r = -0.93) correlated with soil pH and suggested that this strong negative correlation is due to N_2O production being inhibited by alkaline pH. Whilst such strong correlations between N₂O concentration and pH were not observed in this study, the results presented here do support the hypothesis that N_2O production increases with decreasing pH, with the strength of this association partially linked to soil texture.

3.3.4 Other nitrogen species

The relationships between dissolved N₂O concentration and three other N species measured in field drains and stream water samples are presented in Figure 7. N₂O concentrations were generally three orders of magnitude smaller than dissolved NO₃ (Figure 7A and Figure S1), similar to the findings of previous studies (e.g. Ueda et al., 1993; Hack and Kaupenjohann, 2002; Vilain et al., 2011; Outram and Hiscock, 2012). The concentrations of N₂O and NO₃ were significantly and positively correlated in both field drains in sandy loam soils (r = 0.30; p < 0.05) and in stream water samples (r = 0.55; p < 0.05). However, a non-significant weak correlation was observed for field drains in clay loam soils (r = 0.06; p > 0.05). The individual drains with the strongest positive correlations between N₂O and NO₃ were D2 (r =0.80), D6 (r = 0.67) and D1 (r = 0.46), all of which were located within sandy loam soils. Conversely, drains D8 (r = -0.36), D13 (r = -0.29) and D7 (r = -0.15) located within clay loam soils had the strongest negative correlations.

Previous studies have suggested that a positive correlation between N₂O and NO₃ indicates that nitrification is the principle production mechanism for N_2O , whilst a negative correlation indicates denitrification is occurring (Ueda et al., 1993; Mühlherr and Hiscock, 1998; Hiscock et al., 2003). On this basis, the results presented here indicate that nitrification is likely to be the main production mechanism for N₂O in stream waters and field drains in sandy loam soils, whereas in clay loam soils the production mechanism is likely to be a combination of both nitrification and denitrification. This combination of nitrification and denitrification processes is supported by evidence from previous research in the River Wensum and neighbouring River Bure catchments, in which groundwater NO₃-N isotope $(\delta^{15}N)$ values in the range of -2.1 to +13.7‰ were measured at 36 locations (Feast et al., 1998). Isotopically light δ^{15} N values (+4 to +8‰) in these catchments are believed to be indicative of nitrification in areas covered by sand-rich glacial deposits in valley locations, whilst more enriched δ^{15} N values (+8 to +11‰) indicative of fractionation by denitrification

are associated with the presence of clay-rich till deposits at the valley margins.

Regarding other N species, N₂O was only weakly negatively correlated with NH₄ in stream water (r = -0.13; p < 0.05) and in field drains in clay loam (r = -0.09; p > 0.05) and sandy loam (r = 0.01; p > 0.05) soils, indicating the absence of any interconnected production mechanisms. Similarly, N₂O concentrations were not significantly correlated with NO₂ in

either the stream water samples (r = 0.04, p > 0.05) or the clay loam field drains (r = -0.01; p > 0.05), although a significant positive correlation (albeit weak) with sandy loam drains (r = 0.19; p < 0.05) was established, again highlighting potential differences in N₂O production mechanisms between different soil types.

To better assess the complexity and overall importance of these different factors in determining the observed variability in N₂O concentrations, Table II presents the results of three multiple linear regression models for the prediction of N_2O concentrations in stream water and field drains in sandy loam and clay loam soils. The stream model proved to be the best performing, being able to explain 33.1% of the variability in N₂O concentrations from three significant predictors (NO₃, NH₄ and NO₂), although NO₃ was by far the most dominant predictor ($R^2 = 0.31$). By contrast the field drain sandy loam and field drain clay loam models were only able to explain 16.6% and 6.2% of the variability in N_2O concentrations, respectively, with pH being the only significant predictor of N₂O in clay loam field drains. These model results highlight the complexity of N₂O production mechanisms in field drains and indicate that other drivers of N_2O variability exist which are not captured by the regression models.

3.3.5 Impact of a cover crop

³⁸⁵ During autumn and winter 2013/14, dissolved N₂O concentrations in field drains below the ³⁸⁶ winter oilseed radish cover crop ranged from 0.6–8.8 μ g N L⁻¹, whereas concentrations in ³⁸⁷ drains underlying fields without a cover crop ranged from 0.6–4.3 μ g N L⁻¹ (Figure 8). ³⁸⁸ Although the difference in the means was not statistically significant (p > 0.05), drains under ³⁸⁹ the cover crop did have a slightly higher mean N₂O concentration (2.6 μ g N L⁻¹) than drains ³⁹⁰ under fields without a cover crop (2.2 μ g N L⁻¹). This may be due to the accumulation of both ³⁹¹ carbon and N residues under the combined reduced tillage and cover crop management system and consequently higher substrate availability for nitrification and denitrification compared to conventional management (Abdalla et al., 2012). The primary goal of using a cover crop as a mitigation measure in agriculture is to improve soil fertility and decrease NO₃ leaching rather than to reduce greenhouse gas emissions; however the latter should not be neglected when assessing the overall effectiveness of such measures. The mean field drain NO₃ concentration under the cover crop (2.5 mg N L⁻¹) was significantly (p < 0.05) lower than drains beneath fallow fields (13.9 mg N L^{-1}), representing a ~82% reduction in NO₃ concentrations. Contradictory effects of cover crops on direct N₂O emissions from soil have been previously documented (e.g. Jarecki et al., 2009; Kallenbach et al., 2010; Dietzel et al., 2011; Abdalla et al., 2012; Sanz-Cobena et al., 2014), but to our knowledge the effects of a cover crop on indirect N₂O emissions from groundwater and surface waters has not been studied until now. Newell Price et al. (2011) did, however, summarise a list of mitigation measures to tackle environmental issues and stated that cover crops could reduce indirect N₂O emissions by a small amount. The results presented here contradict this and suggest that the use of cover crops (particularly oilseed radish) may actually increase indirect N_2O emissions. Thus, cover crops should not be recommended as a climate change mitigation strategy without further research.

3.3.6 Impact of reduced tillage

410 During the 2014/15 farm year, the different tillage options without a cover crop were

411 continued as mitigation measures and the impact upon field drain N₂O concentrations is 412 presented in Table III. Whilst the mean N₂O concentration in field drains under conventional 413 tillage (6.9 µg N L⁻¹) was not significantly (p > 0.05) different from that under direct drill 414 (6.2 µg N L⁻¹), the mean concentration under reduced tillage (4.8 µg N L⁻¹) was significantly 415 (p < 0.05) lower. Despite this finding, the lower N₂O concentrations in field drains under 416 reduced tillage are more likely to reflect that all these drains were within sandy loam soils

rather than truly representing differences in tillage practice. This is because of the four field drains under reduced tillage, only D16 had significantly lower N₂O concentrations, whereas D1, D3 and D5 showed no substantial decline in N₂O compared to the other drains. Moreover, if N₂O concentrations were truly lower under reduced tillage relative to conventional ploughing, then N₂O concentrations should have been even lower under direct drill systems where soil disturbance, and thus N mobilisation, is even lower.

To our knowledge, there have been no previous publications on the effects of different tillage methods on dissolved N_2O concentration until now, thus comparison with other studies is not possible. However, several studies have investigated the effects of soil management on direct N₂O emissions from soil and these have shown inconsistent results due to variability in environmental factors, such as soil water content, rates and types of fertiliser application, and depths of fertiliser placement (Baggs et al., 2003; Grant et al., 2004; Venterea et al., 2005; Omonode et al., 2011). Overall, the results presented here indicate that different soil inversion methods tended to have little impact on dissolved N₂O concentrations.

4. Conclusions

432 The research conducted here was undertaken to address the deficiency in the number of 433 existing studies investigating indirect N_2O emissions from agriculture. The key findings from 434 this work can be summarised as follows:

435 (i) All field drain and stream water samples collected, regardless of time or location of 436 sampling, contained a higher dissolved N_2O concentration than the water-air 437 equilibrium, demonstrating that all sites were acting as a source of N_2O emissions to 438 the atmosphere;

texture also impacted upon the relationships between field drain N_2O concentration and other water quality parameters, highlighting possible differences in N_2O production mechanisms between different soil types;

- $\begin{array}{ll} _{444} & (iii) & \text{Antecedent moisture conditions influenced the storm event mobilisation of N_2O in} \\ _{445} & \text{field drains and streams, with the greatest concentration increases occurring during} \\ _{446} & \text{events preceded by wet conditions. N_2O concentrations also varied seasonally, with} \\ _{447} & \text{the lowest concentrations typically occurring during the summer months;} \end{array}$
- (iv) Nitrogen fertiliser application and different soil inversion regimes were found to have no effect on dissolved N_2O concentrations either in field drains or stream waters;

450 (v) Higher N₂O concentrations recorded in field drains under a winter cover crop relative
451 to fallow fields indicate growing an oilseed radish cover crop is not an effective
452 greenhouse gas emission mitigation strategy.

Given the paucity of existing studies into the mechanisms involved in the production of indirect N₂O emissions from N leaching into surface water and groundwater bodies, further research conducted in a wider variety of agricultural catchments with a range of different soil types and rainfall regimes is highly recommended.

457 Acknowledgements

This research was funded by the Defra Agricultural Greenhouse Gas Platform (project
AC0116). ZQH acknowledges support from the Iraqi Kurdistan Regional Government. The

460 authors would like to thank: Gilla Suennenberg for GIS mapping; Jenny Stevenson,

- 461 Christopher Adams, Faye Outram, Simon Ellis, Nick Garrard, Steve Warnes and Steve
- 462 Dugdale for fieldwork support; and Liz Rix, Alina Mihailova, Kim Goodey, Tony Hinchliffe
- 463 and Andy Hind for laboratory analytical support. The authors would like to thank the Salle
- 464 Park Estate for their cooperation in providing access to the field sites.

465 Supporting Information

Figure S1: Time series of the mean N_2O/NO_3 ratio in stream water and field drains.

References

- Abdalla, M., Rueangritsarakul, K., Jones, M., Osborne, B., Helmy, M., Roth, B., Burke, J.,
 Nolan, P., Smith, P., & Williams, M. (2012). How Effective is Reduced Tillage–Cover
 Crop Management in Reducing N₂O Fluxes from Arable Crop Soils? Water, Air, & Soil *Pollution*, 223, 5155-5174.
- 472 Baggs, E., Stevenson, M., Pihlatie, M., Regar, A., Cook, H., & Cadisch, G. (2003). Nitrous
 473 oxide emissions following application of residues and fertiliser under zero and
 474 conventional tillage. *Plant and Soil*, **254**, 361-370.
- Beaulieu, J., Arango, C., Hamilton, S., & Tank, J. (2008). The production and emission of
 nitrous oxide from headwater streams in the Midwestern United States. *Global Change Biology*, 14, 878-894.
- Bouwman, A., Fung, I., Matthews, E., & John, J. (1993). Global analysis of the potential for

479 N2O production in natural soils. *Global Biogeochemical Cycles*, **7**, 557-597.

- Bowden, W.B., & Bormann, F. (1986). Transport and loss of nitrous oxide in soil water after
 forest clear-cutting. *Science*, 233, 867-869.
- ⁴⁸² Darling, W.G., Kinniburgh, D.G., & Gooddy, D.C. (1998). Gas compositions and processes
- in the unsaturated zone of the chalk and Triassic sandstone aquifers, England. In: Isotope
- 484 techniques in the study of environmental change. Proceedings of a symposium, Vienna,
- 485 14-18 April 1997, p. 265–274.
- 486 Dietzel, R., Wolfe, D., & Thies, J.E. (2011). The influence of winter soil cover on spring
 487 nitrous oxide emissions from an agricultural soil. *Soil Biology and Biochemistry*, 43, 1989488 1991.

Dowdell, R.J., Burford, J.R., & Crees, R. (1979). Losses of nitrous oxide dissolved in
drainage water from agricultural land. *Nature*, 278, 342-343.

- ⁴⁹¹ European Environment Agency. (2016). Atmospheric concentration of carbon dioxide,
 ⁴⁹² methane and nitrous oxide. Online: <u>http://www.eea.europa.eu/data-and-</u>
 ⁴⁹³ maps/daviz/atmospheric-concentration-of-carbon-dioxide#tab-chart 3.
- Feast, N.A., Hiscock, K.M., Dennis, P.F., & Andrews, J.N. (1998). Nitrogen isotope
 hydrochemistry and denitrification within the Chalk aquifer system of north Norfolk, UK. *Journal of Hydrology*, 20, 233-252.
 - 497 Forster, P., Ramaswamy, V., Artaxo, P., Berntsen, T., Betts, R., Fahey, D.W., Haywood, J.,
- Lean, J., Lowe, D.C., & Myhre, G. (2007). Changes in atmospheric constituents and in
 radiative forcing. *Climate Change*, 20, 129-234.
- Grant, B., Smith, W., Desjardins, R., Lemke, R., & Li, C. (2004). Estimated N₂O and CO₂
 emissions as influenced by agricultural practices in Canada. *Climatic Change*, 65, 315332.
- Hack, J., & Kaupenjohann, M. (2002). N₂O discharge with drain water from agricultural soils
 of the upper Neckar region in Southern Germany. In: Van Ham, J., Baede, A.P.M.R.,
 Guicherit, J.G., & Williams-Jacobse, F.M. (Eds.) Millpress Science Publishers, pp: 185190.
- 507 Hénault, C., Grossel, A., Mary, B., Roussel, M., & Léonard, J. (2012). Nitrous oxide
- 508 emission by agricultural soils: a review of spatial and temporal variability for mitigation.
- *Pedosphere*, **22**, 426-433.
- 510 Hinshaw, S.E., & Dahlgren, R.A. (2013). Dissolved nitrous oxide concentrations and fluxes
 511 from the eutrophic San Joaquin River, California. *Environmental Science & Technology*,
 - , 1313-1322.

513	Hiscock, K., Bateman, A., Mühlherr, I., Fukada, T., & Dennis, P. 2003. Indirect emissions of							
514	nitrous oxide from regional aquifers in the United Kingdom. Environmental Science &							
515	Technology, 37 , 3507-3512.							
516	Jacinthe, P-A., & Dick, W.A. (1997). Soil management and nitrous oxide emissions from							
517	cultivated fields in southern Ohio. Soil and Tillage Research, 41, 221-235.							
518	Jahangir, M.M., Johnston, P., Barrett, M., Khalil, M., Groffman, P., Boeckx, P., Fenton, O.,							
519	Murphy, J., & Richards, K.G. (2013). Denitrification and indirect N_2O emissions in							
520	groundwater: hydrologic and biogeochemical influences. Journal Of Contaminant							
521	<i>Hydrology</i> , 152 , 70-81.							
522	Jarecki, M.K., Parkin, T.B., Chan, A.S., Kaspar, T.C., Moorman, T.B., Singer, J.W., Kerr,							
523	B.J., Hatfield, J.L., & Jones, R. (2009). Cover crop effects on nitrous oxide emission from							
524	a manure-treated Mollisol. Agriculture, Ecosystems & Environment, 134, 29-35.							
525	Kallenbach, C.M., Rolston, D.E., & Horwath, W.R. (2010). Cover cropping affects soil N2O							
526	and CO2 emissions differently depending on type of irrigation. Agriculture, Ecosystems &							
527	Environment, 137, 251-260.							
528	Li, X., Tang, C., Han, Z., Jingqiu, P., Yingjie, C., & Chipeng, Z. (2013). Spatial and seasonal							
529	variation of dissolved nitrous oxide in wetland groundwater. Environment and Pollution,							
530	3 , 21-32.							
531	McGonigle, D., Burke, S., Collins, A., Gartner, R., Haft, M., Harris, R., Haygarth, P.,							
532	Hedges, M., Hiscock, K., & Lovett, A. (2014). Developing Demonstration Test							
533	Catchments as a platform for transdisciplinary land management research in England and							
534	Wales. Environmental Science: Processes & Impacts, 16, 1618-1628.							
535	Minamikawa, K., Hayakawa, A., Nishimura, S., Akiyama, H., & Yagi, K. (2011).							
536	Comparison of indirect nitrous oxide emission through lysimeter drainage between an							

Andosol upland field and a Fluvisol paddy field. Soil Science and Plant Nutrition, 57, 843-854. Mosier, A., & Kroeze, C. (1998). A new approach to estimate emissions of nitrous oxide from agriculture and its implications to the global N₂O budget. *IGBP newsletter*, **34**, 8-13. Mosier, A., Kroeze, C., Nevison, C., Oenema, O., Seitzinger, S., & Van Cleemput, O. (1998). Closing the global N₂O budget: nitrous oxide emissions through the agricultural nitrogen cycle. Nutrient Cycling in Agroecosystems, 52, 225-248. Mühlherr, I.H., & Hiscock, K.M. (1997). A preliminary assessment of nitrous oxide in chalk groundwater in Cambridgeshire, UK. Applied Geochemistry, 12, 797-802. Mühlherr, I.H., & Hiscock, K.M. (1998). Nitrous oxide production and consumption in British limestone aquifers. Journal of Hydrology, 211, 126-139. Naqvi, S., Jayakumar, D., Narvekar, P., Naik, H., Sarma, V., D'souza, W., Joseph, S., & George, M. (2000). Increased marine production of N₂O due to intensifying anoxia on the Indian continental shelf. Nature, 408, 346-349. Newell Price, J., Harris, D., Taylor, M., Williams, J., Anthony, S., Duethmann, D., Gooday, R., Lord, E., Chambers, B., & Chadwick, D. (2011). An inventory of mitigation methods and guide to their effects on diffuse water pollution, greenhouse gas emissions and ammonia emissions from agriculture. In: Report prepared as part of Defra Project WQ0106, ADAS and Rothamsted Research North Wyke. Omonode, R.A., Smith, D.R., Gál, A., & Vyn, T.J. (2011). Soil nitrous oxide emissions in

- corn following three decades of tillage and rotation treatments. *Soil Science Society of America Journal*, **75**, 152-163.
- Outram, F.N., & Hiscock, K.M. (2012). Indirect nitrous oxide emissions from surface water
 bodies in a lowland arable catchment: a significant contribution to agricultural greenhouse
 gas budgets? *Environmental Science & Technology*, 46, 8156-8163.

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Panek, J., Matson, P., Ortiz-Monasterio, I., & Brooks, P. (2000). Distinguishing nitrification 562 563 and denitrification sources of N₂O in a Mexican wheat system using ¹⁵N. Ecological 564 Applications, **10**, 506-514. R Core Team. (2016). R: A language and environment for statistical computing. R 565 Foundation for Statistical Computing, Vienna, Austria. https://www.R-project.org/. 566 567 Reay, D., Smith, K., & Edwards, A. (2004). Nitrous Oxide in Agricultural Drainage Waters 568 Following Field Fertilisation. Water, Air and Soil Pollution: Focus, 4, 437-451. Reay, D.S., Smith, K.A., & Edwards, A.C. (2003). Nitrous oxide emission from agricultural 569 drainage waters. Global Change Biology, 9, 195-203. 570 571 Rochette, P., Angers, D.A., Chantigny, M.H., & Bertrand, N. (2008). Nitrous oxide emissions 572 respond differently to no-till in a loam and a heavy clay soil. Soil Science Society of 573 America Journal, 72, 1363-1369. Sanz-Cobena, A., García-Marco, S., Quemada, M., Gabriel, J., Almendros, P., & Vallejo, A. 574 575 (2014). Do cover crops enhance N_2O , CO_2 or CH_4 emissions from soil in Mediterranean 576 arable systems? Science Of The Total Environment, 466, 164-174. 577 Seitzinger, S.P., & Kroeze, C. (1998). Global distribution of nitrous oxide production and N inputs in freshwater and coastal marine ecosystems. Global Biogeochemical Cycles, 12, 578 579 93-113. 580 Smith, P., Martino, D., Cai, Z., Gwary, D., Janzen, H., Kumar, P., McCarl, B., Ogle, S., 581 O'Mara, F., Rice, C., Scholes, B., & Sirotenko, O. (2007). Agriculture. Climate Change 582 2007: mitigation. In: Contribution of Working Group III to the Fourth Assessment Report 583 of the Intergovernmental Panel on Climate Change. 584 Syakila, A., & Kroeze, C. (2011). The global nitrous oxide budget revisited. Greenhouse Gas 585 Measurement and Management, 1, 17-26.

Page 25 of 34

Hydrological Processes

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586	Ueda, S., Ogura, N., & Yoshinari, T. (1993). Accumulation of nitrous oxide in aerobic
587	groundwaters. Water Research, 27, 1787-1792.

- Venterea, R.T., Burger, M., & Spokas, K.A. (2005). Nitrogen oxide and methane emissions
 under varying tillage and fertilizer management. *Journal of Environmental Quality*, 34,
 1467-1477.
- Vilain, G., Garnier, J., Tallec, G., & Tournebize, J. (2011). Indirect N₂O emissions from
 shallow groundwater in an agricultural catchment (Seine Basin, France). *Biogeochemistry*,
 111,253-271.
 - Weiss, R., & Price, B. (1980). Nitrous oxide solubility in water and seawater. *Marine Chemistry*, 8, 347-359.
- Weslien, P., Kasimir Klemedtsson, Å., Börjesson, G., & Klemedtsson, L. (2009). Strong pH
 influence on N₂O and CH₄ fluxes from forested organic soils. *European Journal of Soil Science*, 60, 311-320.
 - Włodarczyk, T., Stępniewski, W., & Brzezińska, M. (2005). Nitrous oxide production and
 consumption in Calcaric Regosols as related to soil redox and texture. *International Agrophysics*, 19, 263-271.

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Tables

Table I Description of the study area experimental treatments

Block	Field name	Area (ha)		2013/14				2014/15			
			Field drain	Cover crop	Tillage	Crop	Applied fertiliser (kg N ha ⁻¹)	Cover crop	Tillage	Сгор	Applied fertiliser (kg N ha ⁻¹)
J	Far Hempsky	13.8	-	No	Plough	SB	0	No	Plough	WW	226
	Potash	26.8	D8, D10	No	Plough	SB	7	No	Plough	WW	228
Р	Gatehouse Hyrne	17.3	D5	Yes	Reduced	SB	38	No	Reduced	WW	221
	Dunkirk	12.9	D1, D3	Yes	Reduced	SB	30	No	Reduced	WW	219
	Moor Hall Field	20.4	D16	Yes	Reduced	SB	0	No	Reduced	WW	229
L	Swanhills	10.4	D4, D6	Yes	DD	SB	26	No	DD	WW	219
	Sheds Field	14.9	-	Yes	DD	SB	28	No	DD	WW	227
	First Hempsky	14.1	D2	Yes	DD	SB	34	No	DD	WW	229
	Middle Hempsky	11.8	-	Yes	DD	SB	7	No	DD	WW	222

Note: DD: Direct drill, SB: spring beans, WW: winter wheat

Table II: Linear and multiple linear regression model results for the prediction of stream water and field drain N₂O concentrations. VIF is the variance inflation factor; VE is the variance explained. Only significant (p < 0.05) predictors were retained in the models.

Field drains: sandy loam	Predictor	Estimate	Std. Error	<i>t</i> -value	<i>p</i> -value	VIF	Proportion of VE (R ²)
	NO ₃	0.197	0.043	4.55	< 0.001	1.05	0.064
	Flow	3.180	0.795	3.99	< 0.001	1.04	0.045
	NO_2	0.105	0.030	3.52	< 0.001	1.02	0.034
	pH	-0.807	0.366	-2.20	0.03	1.07	0.023
						Fotal VE	0.166
Field drains:	Predictor	Estimate	Std Error	<i>t</i> -value	<i>p</i> -value	VIF	Proportion
clay loam							of VE (R^2)
	pH	-2.174	0.562	-3.87	< 0.001	1.00	0.062
						Total VE	0.062
Streams	Predictor	Estimate	Std Error	<i>t</i> -value	<i>p</i> -value	VIF	Proportion
							of VE (\mathbf{R}^2)
	NO ₃	0.146	0.013	11.16	< 0.001	1.04	0.310
	NH_4	-0.001	0.001	-1.45	0.015	1.30	0.012
	NO_2	0.005	0.002	2.21	0.028	1.26	0.009
						Total VE	0.331

Table III Field drain N₂O concentrations under different tillage practice during October 2014 to April 2015. Numbers followed by different superscripted letters are significantly different (p > 0.05).

Tillage type	п	$\frac{\text{Mean N}_2\text{O}}{(\text{ug N L}^{-1})}$	Standard Deviation
Conventional tillage	33	6.9 ^a	7.0
Reduced tillage	75	4.8 ^b	3.3
Direct drill	73	6.3 ^a	4.0

http://mc.manuscriptcentral.com/hyp





Figure 1: Location of the study area in the Blackwater sub-catchment of the River Wensum, Norfolk, UK. Map shows the locations of field drain, stream water and soil sampling sites.

99x70mm (600 x 600 DPI)



Figure 2: Boxplot of dissolved N2O concentrations in field drains in sandy loam and clay loam soils and in stream waters for samples collected during April 2013–April 2015. The central line is the median, the boxis the interquartile range and the whiskers are 1.5 times the interquartile range. The horizontal dashed line represents the atmospheric N2O concentration when in equilibrium with water (0.36 µg N L-1).

89x50mm (600 x 600 DPI)



Figure 4: Average N2O concentrations recorded in field drains and stream waters during different seasons in samples collected during April 2013–April 2015. Error bars represent one standard error. Significant differences (p < 0.05) are indicated by different letters for the same type of water sample.

99x62mm (600 x 600 DPI)







219x302mm (300 x 300 DPI)



Figure 5: Mean field drain N2O concentrations for drains underlying the two dominant soil texture types in the study area, clay loam and sandy loam, for samples collected during April 2013–April 2015. Error bars represent one standard error.

74x62mm (600 x 600 DPI)



Figure 6: Relationships between field drain N2O concentration and (A) flow rate and (B) pH, split by dominant soil type for samples collected during April 2013–April 2015. Dashed lines are linear regressions.

79x35mm (600 x 600 DPI)



Figure 7: Relationships between N2O concentration and (A) NO3, (B) NH4 and (C) NO2 concentrations in stream water and field drain samples from different soil types collected during April 2013–April 2015. Dashed lines are linear regressions.

64x22mm (600 x 600 DPI)



Figure 8: Relationship between dissolved N2O and NO3 concentrations in field drain samples collected during the growth of a winter oilseed radish cover crop (September 2013 to March 2014) from fields with (n = 114) and without (n = 29) the cover crop. Dashed lines are linear regressions.

79x71mm (600 x 600 DPI)