

1       **Application of redox sensitive proxies and carbonate clumped isotopes to**  
2                   **Mesozoic and Palaeozoic radiaxial fibrous calcite cements**

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22 **ABSTRACT**

23 Ancient marine radiaxial calcite cements are commonly exploited as archives of marine porewater properties  
24 based on the argument that they lack metabolic effects often assigned to biogenic carbonates. Here we  
25 critically test the significance of conventional versus (with respect to these fabrics) less-conventional proxy  
26 data from Pennsylvanian, Triassic, and Cretaceous case examples. Conventional proxies include:  
27 cathodoluminescence, carbon and oxygen isotope ratios, main and trace elemental concentrations. Less  
28 conventionally applied proxies are: clumped isotope “ $\Delta_{47}$ ”, redox-sensitive, and rare earth elements sampled  
29 across a succession of Triassic radiaxial fibrous calcites. Radiaxial calcites are subdivided in three groups  
30 based on their luminescence characteristics: non-luminescent, patchy luminescent, and bright luminescent.  
31 Luminescence patterns are in fair agreement with isotope ratios, in particular with those of oxygen. The data  
32 fall into, or are close, to the range of reconstructed marine seawater values and often plot to the positive end  
33 member of the isotopic range. These results disagree with the commonly held view that isotope data from  
34 luminescent cements reflect a priori non-marine values. Further evidence for this comes from REE  
35 concentration patterns and cerium-anomalies suggesting normal marine porewater values for all except the  
36 very last generation of radiaxial calcites. This implies that luminescent radiaxial calcites must not  
37 necessarily represent significant diagenetic resetting. Kinetic effects during precipitation and different  
38 activator elements must be considered. Marine and earliest burial porewater temperatures of ~12-26°C are  
39 suggested by conventional calcite  $\delta^{18}\text{O}$  thermometry. Conversely, the application of the clumped isotope  
40 thermometer to the same radiaxial calcites suggests temperatures of 180-200°C, reflecting solid-state  
41 resetting of fully cemented limestones under a low water:rock ratio. Redox-sensitive elements, particularly  
42 Zn, Cd, U, and Cu are affected by kinetic processes overriding fluid Eh. Manganese concentrations and Ce-  
43 anomaly data point to gradually decreasing marine porewater oxygen levels from outer to inner cement  
44 fringes. Judging from REE patterns and Ce-anomalies, the cement layers in the central portions of the pore  
45 filling cement succession witnessed the end of marine precipitation and the onset of shallow marine  
46 diagenesis. Consequently, radiaxial calcite precipitation is suggested to continue in the early shallow  
47 (marine) burial domain. This study underscores the potential of radiaxial calcite successions as archives of

48 marine porewater to shallow burial diagenetic pathways. The combination of conventional and less  
49 conventional proxies is a clear strength of this study and documents that abiogenic carbonate archives are  
50 often underexplored.

51

## 52 **1. Introduction**

53 Radiaxial fibrous (RFC) and fascicular optical fibrous (FOFC) calcites (Bathurst, 1959; Kendall  
54 and Tucker, 1971; Kendall and Tucker, 1973; Kendall, 1976; Mazzullo et al., 1990; Swart, 2015) represent  
55 very common (up to 70 vol.-%) pore-filling cements in mainly Palaeozoic and Mesozoic marine carbonate  
56 buildups. Palaeogene and Neogene case examples of this marine carbonate fabric are remarkably scarce or  
57 perhaps even absent in the rock record (Aissaoui, 1988; van der Kooij et al., 2010). Although it is widely  
58 accepted that radiaxial fibrous calcite is an abiogenic precipitate (Richter et al., 2011), other authors, e.g.,  
59 Lees and Miller (1995) discuss a microbial origin of similar cements but did not find evidence for this  
60 themselves.

61 In many cases, the presence of several stratigraphically overlying fringes of radiaxial and fascicular  
62 optical fibrous calcites occluding pore space in reefal carbonates represents a time-resolved archive of fluid  
63 properties recording gradual changes from the marine to increasingly burial domains. The common view is  
64 that these fabrics represent early marine diagenetic stabilization products of a high-Mg radiaxial fibrous  
65 calcite precursor phase whereas the controls that lead to the formation of their converging or diverging  
66 crystal c-axes axes are as yet unclear (see review in Richter et al., 2011). Assuming that marine porewater  
67 properties represent reasonable analogues of coeval marine waters, numerous workers have exploited these  
68 archives (Saller, 1986; Carpenter and Lohmann, 1989; Gray and Adams, 1995; Tobin et al., 1996; Kaufmann  
69 and Wendt, 2000; Kim and Lee, 2003; Hasiuk and Lohmann, 2008).

70 Most previous studies dealing with Palaeozoic or Mesozoic radiaxial fibrous calcites apply standard  
71 cathodoluminescence microscopy and conventional stable isotope ( $\delta^{13}\text{C}$  and  $\delta^{18}\text{O}$ ) combined with major  
72 and trace elemental (Mg, Mn, Sr, Fe) geochemistry in order to assess the nature and degree of diagenetic

73 overprint (Carpenter and Lohmann, 1989; Mazzullo et al., 1990; Bruhn et al., 1995; van der Kooij et al.,  
74 2007; 2009; Hasiuk and Lohmann, 2008). The commonly held opinion is that non-luminescence (black to  
75 blue intrinsic) typifies well-preserved material, whilst patchy or bright luminescent (radial) calcites are  
76 considered as diagenetically altered and hence not applicable for the reconstruction of past seawater  
77 properties (see discussion in van der Kooij et al., 2007; 2009). Only a very limited number of workers  
78 exploited these fabrics in a more systematic manner as archives of their diagenetic pathways (e.g., Mazzullo  
79 et al., 1990; Kaufmann, 1997). Along these lines, Rare Earth Element (REE) data have the potential to shed  
80 light on the nature and properties of pore fluids and their evolution with time (e.g., German and Elderfield,  
81 1990; German et al., 1991; Bodin et al., 2013; Della Porta et al., 2015) but have, to the knowledge of the  
82 authors, not yet been applied to these fabrics. The same accounts for the clumped isotopes ( $\Delta_{47}$ ) proxy  
83 revealing important information on burial fluid temperatures (e.g., Bernasconi et al., 2011; Huntington et  
84 al., 2011; Budd et al., 2013; Huntington and Lechler, 2015; Kluge et al., 2015; Shenton et al., 2015).  
85 Acknowledging that the REE proxy is a widely applied tool in geochemistry, we argue that its application  
86 to the fabrics studied here is rarely seen. The same accounts for redox sensitive elements and carbonate  
87 clumped isotope data. Therefore, referring specifically to radial fibrous calcites, the label “less  
88 conventional” for the application of these proxies seems appropriate. The goal is to test if pore fluids from  
89 which some of these fabrics precipitated or stabilized might have properties that differed from that of coeval  
90 seawater.

91 The aims of this study are twofold: First, we document a comparison of geochemical and  
92 cathodoluminescence features of Pennsylvanian, Triassic and Cretaceous case examples of pore-filling  
93 successions of marine radial calcites. Second, a particularly extensive succession of Upper Triassic  
94 radial fibrous calcites was selected as a case example for an in-depth study of the diagenetic pathway  
95 recorded in these fabrics. In addition to conventional light stable isotopes and main and trace elements, we  
96 here also document clumped isotope, redox sensitive elements, and REE data from a transect across all  
97 paragenetic phases of these cements. This is performed with the intention to go beyond previously discussed

98 interpretations of these cement archives and to test the sensitivity of isotopic and elemental proxies for  
99 patterns and threshold limits in the diagenetic realm.

100

## 101 **2. Description of case settings, sample materials and rationale for the selection of study sites**

102

103 Below we briefly document the main characteristics of three Palaeozoic and Mesozoic case settings  
104 investigated here and refer to Table 1 for a summary of the main characteristics of the chosen time slices.  
105 The selection of sampling sites has been guided by the need for reasonably well dated, volumetrically  
106 significant, stratigraphically thick radial fibrous cement successions in a shallow marine environment. A  
107 detailed framework of environmental and diagenetic parameters for all sampling sites exists. Most of the  
108 data (about 60%) shown here were obtained from Upper Triassic (Carnian to Norian; *ca* 227 -209 Ma)  
109 fibrous cements of Hydra Island (Greece). These are particularly suitable for this study due to their  
110 uncommonly thick cement fringes (~ 5 to 10cm). Approximately 20% of the samples studied originate from  
111 Pennsylvanian (Bashkirian; *ca* 323 - 311 Ma) of Northern Spain and approximately 20% from the Late  
112 Cretaceous (Cenomanian; *ca* 99 - 93 Ma) of Dokos Island (Greece; Table 1). Combining these three sample  
113 sets allows for deducing conclusions that go beyond that of a single case study.

114

### 115 *2.1 Pennsylvanian of Asturias*

116 The carbonate platform sampled in the context of this study, the Sierra del Cuera mountain range, is  
117 situated in the province of Asturias, Northern Spain (Fig. 1). This area forms part of the Cantabrian Zone  
118 characterized by a wide marine foreland basin during early Pennsylvanian time (Serpukhovian, 328-318  
119 Ma). Nucleation of an extensive carbonate platform took place during Bashkirian (318-312 Ma) and  
120 Moscovian times (312-307 Ma; Colmenero et al., 2002; Bahamonde et al., 2007; van der Kooij et al., 2009).  
121 Due to tectonic rotation, outcrops of a high-rising margin of the platform are presently exposed in near-  
122 vertical position. The Bashkirian part of the margin, the Valdeteja Formation, has a progradational character

123 while the Moscovian portions (Picos de Europa Formation) are of aggradational nature (Bahamonde et al.,  
124 1997; Della Porta et al., 2004; van der Kooij et al., 2009). The Bashkirian upper-slope sequence, the  
125 sampling site of radiaxial calcites analysed in this study, is characterized by intercalated microbial  
126 boundstones, platform-shed grainstones and red-stained limestones with abundant successions of radiaxial  
127 fibrous calcites (Fig. 2A). These red intervals in the platform slope represent deepening pulses (van der  
128 Kooij et al., 2007).

129

## 130 *2.2 Triassic of Hydra*

131 The Greek island of Hydra is located offshore the Argolis Peninsula (Fig. 1). Hydra Island extends  
132 18 km in EW and 4 km in NS direction (64.4 km<sup>2</sup>) and forms part of the Subpelagonian Zone, a passive  
133 continental margin succession belonging to the Internal Hellenides and has a transitional position to the  
134 Pindos zone (Jacobshagen, 1986; Richter, 1999). The geology of Hydra is characterized by three Tethyan  
135 neritic carbonate to siliceous hemipelagic successions trusted tectonically on top of each other. The oldest  
136 unit (I) is Permian in age and comprises large mud mounds embedded in their surrounding host sedimentary  
137 rocks. The stratigraphically overlying unit (II) is mainly characterized by Anisian (Middle Triassic)  
138 Tubiphyte reefal facies. The third unit (III) comprises sedimentary rocks of Upper Triassic (Carnian to  
139 Rhaetian) age and holds the tectonically uppermost position. It is mainly built by sponge-, coral- and algae-  
140 reefal facies (Richter, 1999). Sample locations for radiaxial fibrous calcites are mainly in the porous reefal  
141 texture of Unit IIIs reefal limestone facies (Fig 2B).

142

## 143 *2.3 Cretaceous of Dokos*

144 The island of Dokos (~15 km<sup>2</sup>) is located between Hydra Island and the mainland of Greece (Fig. 1).  
145 Dokos is mainly built by grey to dark red Cenomanian (Upper Cretaceous) bedded, neritic Akros limestones  
146 yielding abundant orbitolinid foraminifera and rudist bivalve remains (Clift and Robertson, 1990; Clift,  
147 1996; Richter, 1999). Generally, the bedding of exposed units is thinning stratigraphically upwards and the

148 limestones represent an outer platform to slope setting. In northern Dokos, exposures of a stratigraphically  
149 thick, coarse breccia interval are exposed (Römermann, 1969). Pore spaces in these breccia bodies are often  
150 occluded by rims of radiaxial calcite cements sampled in the context of this study (Fig. 2C).

151

### 152 **3. Definitions and terminology**

153

154 Radiaxial fibrous (RFC) and fascicular optical fibrous (FOFC) carbonate fabrics (e.g., Kendall and  
155 Tucker, 1971; Kendall and Tucker, 1973; Kendall, 1976; Mazzullo et al., 1990) are composed of elongated  
156 (magnesian) calcite and rarely of dolomite (RFD; Hood and Wallace, 2012; Richter et al., 2014). These  
157 calcite cements show a characteristically distinct undulous extinction pattern when examined under crossed  
158 polarized light in a polarization microscope (Bathurst, 1959). Both fabrics reveal sub-crystals and a cloudy  
159 zonation and may include micro-inclusions (Davies and Nassichuk, 1990). Sub-crystals can either be  
160 identified by a displacement of the curvature of cleavage inside one crystal or by an unconformity in the  
161 movement of undulosity through the crystal. Radiaxial fibrous calcite fabrics, which are usually much more  
162 common relative to fascicular optical fibrous ones, are characterized by convergent c-axes while fascicular  
163 optical fibrous calcites reveal divergent c-axes (Fig. 3; Kendall, 1985). Furthermore, RFC fabrics display a  
164 characteristic upward concave curvature of cleavage, whilst the opposite is found for FOFC (Fig. 3; e.g.  
165 Kendall and Tucker, 1971; Kendall and Tucker, 1973; Mazzullo, 1980; Richter et al., 2011). Marine  
166 radiaxial calcites whereas such of shallow burial origin display increasingly planar cleavage. Following  
167 Richter et al. (2011), here the catch-it-all term “radiaxial calcite” is used for radiaxial fibrous and fascicular  
168 optical fibrous calcite unless specified differently. The finding of mainly radiaxial fibrous fabrics in  
169 Cretaceous belemnite guards (Richter et al., 2011) questions the strictly abiogenic nature of these particular  
170 cements. Radiaxial fibrous and fascicular optical fibrous calcite cements are not limited to marine  
171 environments but also common fabrics in vadose-meteoric cave depositional environments where they

172 include Holocene and present-day calcitic speleothems and flowstones (Richter et al., 2011), rimstones  
173 (Richter et al., 2015), or cryocalcites (Richter and Riechelmann, 2008).

174

## 175 **4. Methodology**

176

### 177 *4.1. Microscopy*

178 A total of ten thin sections from Pennsylvanian, Upper Triassic and Upper Cretaceous specimens  
179 (Fig. 2) have been analysed in detail by means of polarization and cathodoluminescence microscopy. Thin  
180 sections (2 x 3.6 cm) were used to characterize different fabrics under normal and polarized light.  
181 Cathodoluminescence (CL) microscopy has been applied in order to characterize zones of different CL  
182 characteristics using a HC1-LM hot cathode cathodoluminescence microscope developed at the Ruhr-  
183 University Bochum (Neuser et al., 1995). The electron beam is accelerated with 14 keV and the beam current  
184 is 0.2 mA. Rather than documenting spatially limited portions of thin sections by cathodoluminescence, a  
185 total of 120 to 150 individual CL images per thin section were combined to form luminescence maps with  
186 dimensions of approximately 2 x 3.6 cm, i.e. representing the full surface area of individual thin sections.  
187 The terminology applied here to differentiate between different luminescence zones is “non-  
188 luminescent/intrinsic”, “patchy luminescent” and “bright luminescent” (Fig. 4). This approach allows for  
189 very detailed assessment of stratigraphic and spatial changes within and across different cement rims (Fig.  
190 5).

191

### 192 *4.2. Geochemistry*

193 All studied rock samples have been analysed for their light stable isotope geochemistry ( $\delta^{13}\text{C}$  and  $\delta^{18}\text{O}$ ).  
194 Aliquots were used for the analysis of conventionally used main and trace elemental concentrations. Based  
195 on the features observed during cathodoluminescence microscopy, 152 powder sub-samples have been  
196 drilled from the corresponding rock slabs. A total of 34 samples are taken from Pennsylvanian red-stained



197 limestones, while 78 samples have been taken from Upper Triassic reefal limestone and 40 from Upper  
198 Cretaceous limestone breccia. A total of 78 samples were analysed from the Upper Triassic comprising 24  
199 general samples and 54 detailed samples from a transect through a 3-cm-thick RFC succession. The  
200 approach commonly applied when investigating different paragenetic stages in cement-filled pores is to  
201 place a transect across the centre of the pore aiming at a full coverage of all phases. Here we applied a  
202 different approach placing a transect across the portions of the pore, where the radial fibrous calcites are  
203 thickest and most complete (Fig. 5A).

204 Additionally, redox-sensitive and rare earth elements of the Triassic cement succession have been  
205 measured by means of Laser-Ablation-ICP-MS in order to gain independent information regarding the  
206 porewater evolution. Clumped isotope  $\Delta_{47}$  values (Fig. 5D) of four samples have been performed in order  
207 to shed light on the burial history of these samples.

208

#### 209 *4.2.1. Carbon and oxygen isotope analysis*

210 Based on cathodoluminescence maps, two different sampling approaches have been applied: (i)  
211 spatial sampling of carbonate powder from each of the typical CL zones observed using a Dremel hand  
212 drilling device with a 0.3 mm diamond drill head. (ii) In order to obtain a time-resolving data set across all  
213 paragenetic successions, a geochemical transect (54 data points) of 3 cm length, oriented perpendicular to  
214 the main orientation of cement seams, has been drilled from an Upper Triassic rock sample (Rk-1-H) with  
215 a computer-controlled CAM milling cutter device. This translates into sampling increments of 18  
216 samples/cm (Fig. 5).

217 Carbon and oxygen isotope ratios were analysed using a Gasbench coupled to a ThermoFinnigan  
218 MAT-253 mass spectrometer at the Ruhr-University Bochum. Carbon and oxygen isotopic ratios are given  
219 in ‰ relative to Vienna Pee-Dee Belemnite (VPDB) standard. Carbonate standards CO1, CO8, NBS19 as  
220 well as an internal standard were used for correction of the measured values. Prior to analysis, samples of  
221 0.35 to 0.40 mg have been weighed in into glass vials and then dried in an oven with a temperature of about  
222 105°C for 48 hours. Then the glass vials were closed gastight and put into a heated (70°C) autosampling

223 device and first flushed with He gas in order to remove the atmosphere inside the glass vials. After that a  
224 few drops of phosphoric acid are given on the sample and the emerging CO<sub>2</sub> is soaked in and measured for  
225 its carbon and oxygen isotopic composition in the coupled mass spectrometer.

226

#### 227 4.2.2. Clumped isotope ( $\Delta_{47}$ ) analysis

228 Four samples of pure RFC aliquots were selected for clumped isotope analysis in order to get  
229 information about temperature and burial history. The sample locations have been chosen close to the  
230 transect sampled for stable isotope ratios and trace element concentrations, one sample originating in both  
231 outermost cement layers (1b/t) as well as two samples in the central layers 4(b/t), where one sample is  
232 located on the left side of the transect in a bright luminescent zone and the other one on the right side of the  
233 transect in a non-luminescent zone (Fig. 5). Clumped isotope analysis were performed at the Qatar Stable  
234 Isotopes Laboratory at Imperial College, London, UK. Six to eight mg of powdered calcite were reacted for  
235 10 minutes in phosphoric acid held at 90°C. The CO<sub>2</sub> gas was purified by passage through a conventional  
236 vacuum line with multiple cryogenic traps and a Porapak-Q trap held at -35°C (Dennis and Schrag, 2010).  
237 CO<sub>2</sub> gas was analysed using a ThermoFinnigan MAT-253 gas source isotope ratio mass spectrometer  
238 configured to measure masses 44 through 49. Analytical protocols and corrections for non-linearity follow  
239 Huntington et al. (2009) and the data are reported in the “carbon dioxide equilibrium scale” (CDES) of  
240 Dennis et al. (2011). An acid correction factor of +0.069‰ was added to all measurements of  $\Delta_{47}$  CDES  
241 following Guo et al. (2009) and Wacker et al. (2013). Masses 48 and 49 were monitored to check for possible  
242 sample contamination following Affek and Eiler (2006) and Huntington et al. (2009; 2011). Each sample  
243 was measured four times to improve counting statistics. The  $\Delta_{47}$  CDES reproducibility is  $\pm 0.02$  permil based  
244 on regular analysis of Carrara marble and an intercalibration carbonate standard (‘ETH3’). The Carrara  
245 marble  $\Delta_{47}$  CDES mean value for the measurement period is  $0.385 \pm 0.016$ ‰ (1SD, n=20) and  $0.691 \pm 0.017$ ‰  
246 (1SD, n=16) for ETH3. The  $\Delta_{47}$  CDES values were converted to temperature solving numerically the  
247 calibration of Kluge et al. (2015):

248 
$$\Delta_{47}^{\text{eq}} = 0.98 * \left( \frac{-3.407 \times 10^9}{T^4} + \frac{2.365 \times 10^7}{T^3} - \frac{2.607 \times 10^3}{T^2} - \frac{5.880}{T} \right) + 0.293 \quad [1]$$

249 This calibration produced a good linear correlation of all data sets in the Kluge et al. (2015) study and  
250 thus confirms that the uncertainty of the acid fractionation correction has no impact on the temperature  
251 relationship. Additionally, as the samples of this study were processed, measured and evaluated in the same  
252 way as those of Kluge et al. (2015), this calibration is especially valid for this study.

253

#### 254 *4.2.3. ICP-AES and ICP-MS major and trace elemental analysis*

255 Major and trace elemental abundances (Ca, Mg, Sr, Fe and Mn) of all samples have been analysed  
256 making use of an inductively coupled plasma atomic emission spectrometer (ICP-AES) at Bochum. For the  
257 translation of the measured values into the unit ppm (parts per million), the certified reference materials  
258 BSC-CRM-512 (dolomite) and BSC-CRM-513 (limestone) were analysed. Analytical errors are given as  
259  $\pm\%$ RSD. The  $1\sigma$ -reproducibility for the major- and trace elements of the two standard materials is:  $\pm 0.18\%$   
260 for Ca,  $\pm 0.081\%$  for Mg,  $\pm 22$  ppm for Sr,  $\pm 17$  ppm for Fe and  $\pm 1$  ppm for Mn (CRM-512, n=111) and  
261  $\pm 0.36\%$  for Ca,  $\pm 0.002\%$  for Mg,  $\pm 1$  ppm for Sr,  $\pm 12$  ppm for Fe and  $\pm 1$  ppm for Mn (CRM-513, n=111;  
262 Geske et al., 2015).

263

#### 264 *4.2.4 Laser ablation-ICP-MS minor and rare earth element analyses*

265 Additional minor and rare earth elemental data have been obtained via laser ablation inductively-  
266 coupled plasma mass spectrometry (LA-ICP-MS) at the Institute for Mineralogy, Westfaelische Wilhelms-  
267 Universität, Münster. Within the limitation of sampling possibilities, this was done in order to collect  
268 comparable sample sets from sites that are situated as closely to each other as possible. For the LA-ICP-MS  
269 analysis, the sampled rock slice has been cut perpendicular to the drilled profile, turned by  $90^\circ$  and the laser  
270 track on the newly cut surface has been aligned as close as possible to the drilled transect (Fig. 5B). Sample  
271 ablation was performed with a 193nm ArF excimer laser (Analyte G2, Photon Machines). The repetition  
272 rate was set to 10 Hz using a fluence of  $4\text{J}/\text{cm}^2$ . The beam spot diameter was  $110\ \mu\text{m}$  for the entire session.

273 Ablated particles were transported by helium carrier gas and mixed with argon sample gas before entering  
274 the Element 2 (ThermoFisherScientific) single collector mass spectrometer. Forward power was 1300 W  
275 and reflected power < 2 W, gas flow rates were 1.1 l/min for total He and 0.9 l/m and 1.2 l/m for the Ar-  
276 auxiliary and sample gas, respectively. Cooling gas flow rate was set to 16 l/min. Prior to analyses, the  
277 system has been tuned on the NIST 612 reference material measuring  $^{139}\text{La}$ ,  $^{232}\text{Th}$  and  $^{232}\text{Th}^{16}\text{O}$  for high  
278 sensitivity and stability and low oxide rates ( $^{232}\text{Th}^{16}\text{O}/^{232}\text{Th} < 0.1\%$ ) during ablation. Overall measurement  
279 time for a single spot analysis was 60 s, with 20 s for background and 40 s for peak signal. The washout  
280 time between two spots was set to 15 s. A total of 47 elements were analysed using  $^{43}\text{Ca}$  as internal standard  
281 element and NIST 612 as external reference material (Jochum et al., 2011). Distance between single laser  
282 spots was 400 $\mu\text{m}$  in order to be close to resolution of the profile drilled with the computer-controlled mill  
283 along the same cement succession. This resolution yielded 95 data points over a line of 3.8cm. Groups of  
284 about 20 unknowns were bracketed with 3 NIST 612 calibration standards to keep track of instrumental drift  
285 over the course of the session. Raw data were further processed using the Glitter program (Griffin et al.,  
286 2008).

287 Due to known analytical problems in analysing carbonates with LA-ICP-MS (Jochum et al., 2012),  
288 the MACS-3 carbonate reference material provided by the USGS was measured along with the samples to  
289 monitor for accuracy and precision. Table A.1 (Appendix) shows that MACS-3 (n=13) trace element  
290 concentrations generally match the published range of values given in previous work (Chen et al., 2011;  
291 Jochum et al., 2012; Jochum et al., 2014; see GeoRem database version 18) proving that the analytical  
292 conditions chosen for this study give reasonably precise and accurate data with average 1-sigma standard  
293 deviation in the order of ~14% for concentrations <1ppm, 7% for element concentrations > 1 ppm and  
294 approximately 6% for concentrations higher than 100ppm.

295 Rare earth element values have been normalized to the North American Shale Composites (NASC)  
296 values given in McLennan (1989). Cerium-anomalies can be used as an individual paleo-redox proxy and  
297 have been calculated using the following equation (Nozaki, 2001), where “N” stands for shale-normalized  
298 concentration:

299 
$$\text{Ce} / \text{Ce}^* = 2\text{Ce}_N / (\text{La}_N + \text{Pr}_N) \quad [2]$$

300 In order to check Ce-anomalies for their reliability, also the Pr-anomaly has been calculated to be plotted  
301 against the Ce-anomaly to construct a La-anomaly diagram following work by Bau and Dulski (1996). For  
302 this the following equation has been used:

303 
$$\text{Pr} / \text{Pr}^* = 2\text{Pr}_N / (\text{Ce}_N + \text{Nd}_N) \quad [3]$$

304

## 305 **5. Results**

306

### 307 *5.1. Petrography and luminescence patterns*

308 All radiaxial fibrous cements studied here reveal spatially complex cathodoluminescence (CL)  
309 patterns (Fig. 4). Differential luminescence patterns are recognized stratigraphically, i.e. luminescence  
310 patterns change between successive layers, but also laterally within single cement layers (spatially).  
311 Luminescence colours range from dark blue and near non-luminescent (intrinsic luminescence; Fig. 4B),  
312 via patchy luminescence (intrinsic with yellowish patches or dots; Fig. 4D) to bright yellow or orange-brown  
313 (micro-patchy) luminescence colours (Fig. 4F). We here refer to non-luminescent, patchy luminescent, and  
314 bright luminescent patterns and have assigned geochemical data accordingly.

315 Bashkirian (Pennsylvanian) radiaxial fibrous calcite cements from Northern Spain nucleate on red  
316 micrite slope carbonates (Fig. 2A) and occlude inter-particle pore space between brachiopod shells,  
317 echinoderm fragments, and other bioclasts. They form part of paragenetic successions that, towards the  
318 centre of many pores, grade into scalenohedral and then blocky calcites (bright yellow/orange  
319 luminescence), locally with internal red sediments (crystal silt). Radiaxial fibrous calcites form fringes that  
320 are rather thin when compared to e.g., the Triassic ones. Specifically, individual layers reach some mm in  
321 thickness only with several superimposed layers amounting to 2 cm or less. Individual radiaxial fibrous  
322 crystals reach length:width ratios between 4:1 and 8:1. A more or less distinct cloudy zonation results from  
323 micro-inclusions and is also reflected in the corresponding luminescence patterns. All types of  
324 cathodoluminescence patterns used here are present.

325 From the sample material available, an Upper Triassic (Carnian to Norian) specimen (Rk-1-H; Figs  
326 2B, 5) from Hydra Island (Greece) was selected as a particularly representative case example of a  
327 stratigraphically thick succession of consecutive radiaxial calcite layers occluding primary pore space.  
328 Sample Rk-1-H is devoid of any mesoscopically visible late burial veining. A significant proportion of the  
329 data shown here were taken from this hand specimen. Nucleating on a reefal limestone substratum, sample  
330 Rk-1-H included a total of eight cement layers, representing four coeval couplets of cement precipitation  
331 (Figs 2B, 5C). All types of cathodoluminescence patterns used here are present. Thin-section microscopy  
332 of this rock sample revealed characteristic petrographic changes from the outer cement fringes towards the  
333 centre of the pore (Fig. 5C). In the outermost two cement fringes (cement layers 1b and 1t; Fig. 5C),  
334 nucleating on the limestone substratum, individual radiaxial calcites reach length:width ratios between 5:1  
335 and 8:1 similar to the Bashkirian ones. Individual crystals display a distinct undulosity and often reveal sub-  
336 crystals. In these outer fringes, radiaxial fibrous calcite (RFC) fabrics dominate (>95%) as based on thin  
337 section analysis. Single fascicular optical fibrous calcites (FOFC) are found in places but remain  
338 volumetrically insignificant (<5%). Cement rims 1b and 1t display well defined, large crystals lacking  
339 distinct sub-crystals and undulosity moves smoothly through the crystals when turning the microscope table.  
340 Initially, the orientation of individual crystals is at best sub-parallel. With increasing crystal growth, crystal  
341 c-axes are oriented increasingly parallel where crystals also show an increasing presence of sub-crystals.  
342 The most curved cleavage is present in earliest phases 1b/t and 2b/t but turns increasingly linear in later  
343 stages 3 to 4. All layers contain both radiaxial fibrous as well as fascicular optical fabrics. In cement rims  
344 3b/t and 4b/t, i.e., towards the centre of the pore, the fascicular optical fabrics increase in abundance (layer  
345 4b/t >90% FOFC). The Triassic samples display rather dull and dark luminescence features with localized  
346 layered as well as patchy areas with bright luminescence. Cement fringes 1t/b are mostly non-luminescent  
347 and are followed by two fringes of patchy luminescent radiaxial calcites (2b/t and 3b/t in Fig. 5D). The  
348 cement layers close to the suture (4b/t in Fig. 5D) are generally dull luminescent whilst both grade laterally  
349 into bright luminescent calcites.

350 Cenomanian (Upper Cretaceous) samples from Dokos Island (Greece) are characterized by  
351 radiaxial calcite cements occluding abundant pore space in outer platform limestone breccia (Fig. 2C).  
352 Successions of radiaxial calcite vary in thickness between 1 and 2 cm but paragenetic sequences may reach  
353 5 cm in width locally. In the latter case, radiaxial calcites are, towards the centre of the pore, overlain by  
354 blocky, bright luminescent calcite cement assigned to the burial diagenetic realm. Internal sediments are  
355 present in some cases. The radiaxial fibrous calcite crystals (length:width ratios of about 4:1) are rather  
356 inclusion-rich (cloudy) whilst they lack the distinct zonation than of their Bashkirian counterparts.  
357 Generally, they are smaller and thinner compared with crystals from the other samples. The Cenomanian  
358 fabrics from Dokos are characterized by two luminescence patterns: (i) patchy yellow luminescence that is  
359 well-correlated with individual cement seams, and (ii) bright yellow luminescence, locally superimposed  
360 by patches of dull luminescence.

361

## 362 *5.2 Geochemistry*

### 363 *5.2.1. Spatially distributed samples: Geochemical properties of luminescence zones*

364 The geochemical data set (Table 2, Figs 6 and 7) is described below in a condensed manner.  
365 Bashkirian radiaxial fibrous calcites are typified by rather invariant carbon-isotope values between +4.6 and  
366 +5.5‰ (mean = +5.0‰, standard deviation  $\sigma = 0.18$ ), whilst oxygen-isotope ratios show a broad range  
367 between -5.6 and +2.1‰ (mean = -2.4‰,  $\sigma = 2.08$ ; Fig. 6A). Non-luminescent (intrinsic) samples display  
368 more positive  $\delta^{18}\text{O}$  ratios (+2.1 to -3.7‰, mean = -0.7‰) compared to those with patchy (-5.6 to -1.4‰,  
369 mean = -3.5‰) or bright yellow luminescence (-5.2 to +1.1‰, mean = -2.7). Magnesium elemental  
370 concentrations are slightly enriched in non-luminescent samples (non-luminescent: 7482 to 11770ppm and  
371 bright luminescent: 6372 to 9473ppm), whilst Mn concentrations reveal no distinct trend in concentration  
372 from non-luminescent to bright luminescent patterns (non-luminescent: 5 to 571ppm, mean = 88ppm and  
373 bright luminescent: 14 to 219ppm, mean = 98ppm). Sr elemental abundances show a slightly opposite trend,  
374 as they are moderately depleted in bright luminescent samples compared with non-luminescent ones (non-  
375 luminescent: 302 to 430ppm, mean=382ppm and bright luminescent: 296 to 694ppm, mean=350ppm). Iron

376 concentrations show an increase from non- (32 to 366ppm, mean = 103ppm) to bright luminescent cements  
377 (25 to 485ppm, mean = 138ppm). In comparison to the two data sets described below, Bashkirian samples  
378 yield higher trace element concentrations in terms of Mg, Mn and Sr that show significant variability (Fig.  
379 7).

380 Upper Triassic radiaxial fibrous cements display a much smaller variability in oxygen (-1.6 to -  
381 0.9‰, standard deviation of all Triassic samples  $\sigma = 0.29$ ) and also in carbon isotope ratios (+2.8 to +3.7‰,  
382 mean = +3.3‰,  $\sigma = 0.16$ ; Fig. 6B). Bright luminescent samples coincide with lower  $\delta^{13}\text{C}$  values compared  
383 to those characterized by patchy or non-luminescence, while no distinct pattern between luminescence  
384 characteristics and  $\delta^{18}\text{O}$  ratios was observed. Bright luminescent samples yield lower Mg abundances  
385 compared to non- and patchy-luminescent calcites. Several studies revealed that luminescence in calcite is  
386 mainly caused by incorporation of Mn into the crystal lattice (ten Have and Heijnen, 1985; Habermann et  
387 al., 1998; (Richter et al., 2004). As expected, Mn concentrations increase from non- to bright luminescent  
388 calcites. Strontium concentrations of non- and patchy-luminescent cements are rather similar but show a  
389 decrease towards bright luminescent samples while Fe reveals a moderate increase from non-luminescent  
390 over patchy to bright luminescent calcites (Fig. 7).

391 Cenomanian fibrous cements yield  $\delta^{13}\text{C}$  ratios ranging between +3.0 and +3.4‰ and a  
392 corresponding mean value of +3.2‰ and the standard deviation for all Cenomanian samples is  $\sigma = 0.12$ .  
393 Carbon isotope ratios of bright luminescent samples are lower than those of non-luminescent calcites (Fig.  
394 6C). The corresponding  $\delta^{18}\text{O}$  values range between -3.2 and -1.2‰ and exhibit a mean value of -2.0‰ and  
395  $\sigma = 0.62$ . Radiaxial calcites characterized by dark blue luminescence plot in a narrow range ( $\delta^{13}\text{C}$ : +3.3 to  
396 +3.4‰, mean = +3.3‰;  $\delta^{18}\text{O}$ : -1.3 to -1.8‰, mean = -1.5‰). Bright luminescent calcites remain invariant  
397 in terms of their carbon isotope signatures but are more variable in their  $\delta^{18}\text{O}$  ratios ( $\delta^{13}\text{C}$ : +3.1 to +3.4‰,  
398 mean = +3.2‰;  $\delta^{18}\text{O}$ : -3.2 to -1.2‰, mean = -2.4‰). Oxygen isotope ratios tend to lower values from non-  
399 over patchy- to bright luminescent fibrous calcites, whereas some of the samples taken from bright  
400 luminescent calcites plot within the range of the non-luminescent samples. Magnesium abundances are  
401 rather homogenous but gradually increase from non- to patchy-luminescent cements. Bright luminescent



402 samples yield moderately depleted Mg concentrations. Manganese concentrations increase from non- to  
403 bright luminescent cements. Strontium concentrations display mean values between 178 and 193ppm, with  
404 minima found in non-luminescent and maxima in patchy-luminescent cements. Iron concentrations show  
405 an increase from non-, to patchy-, to bright luminescent cements (Fig. 7).

406

#### 407 *5.2.2 Triassic case example: Geochemical transect across different luminescence zones*

408 Magnesium, Mn, Sr, and Fe concentrations (ICP-OES) as well as carbon and oxygen isotope ratios  
409 are illustrated in Figure 8. Redox-sensitive element patterns (LA-ICP-MS) and rare earth element patterns  
410 are documented in Figures 9 through 12. Generally, the geochemical data acquired from the transect across  
411 the paragenetic cement succession display a very prominent symmetrical stratigraphic pattern from cement  
412 fringes 1b and 1t towards the central suture of the cement successions (Fig. 8). This geochemical pattern is  
413 near-identical in both, up- and downward growing cement successions, a feature considered evidence for a  
414 geochemical pattern that reflects pore water properties in a phreatic environment. Three out of four data sets  
415 (Mg, Fe, and Sr) and carbon and oxygen isotope values reveal a general trend to more depleted isotope  
416 values or decreasing elemental concentrations towards the central suture (Fig. 8). Conversely, Mn elemental  
417 abundances become gradually enriched towards the central suture and lack a comparable symmetrical  
418 pattern. Higher order patterns are superimposed upon this general trend. Specifically,  $\delta^{18}\text{O}$ , Sr, Mn, and Fe  
419 show a spatially limited shift to higher ratios/abundances at the boundary between non-luminescent/intrinsic  
420 (1b/t) and (patchy) luminescent (2b/t and 3b/t) cement fringes, albeit with rather variable amplitudes. The  
421 most pronounced excursion towards lower values is found in the case of Mg (amplitude: 1450 ppm), whilst  
422 oxygen isotope ratios display only a minor shift in the order of 0.5‰ ( $\sigma = 0.16$ ). Carbon isotope ratios and  
423 Mg concentrations co-vary. Moreover, geochemical patterns change between cement layers 3b/t and 4b/t.  
424 Here, a pronounced decrease in Sr, Mg, and Fe concentrations is observed, while a trend to increasing Mn  
425 concentrations is found. The main changes in geochemical patterns, and particularly so with reference to  
426 activators ( $\text{Mn}^{2+}$ ) and inhibitors ( $\text{Fe}^{2+}$ ), coincide largely with changes in luminescence characteristics but  
427 the issue is complicated by other activator elements such as REE (Richter et al., 2004).

428 Elements analysed by means of LA-ICP-MS (U, Cd, Zn, and Cu) display the same first order  
429 geochemical pattern as observed for Fe, Sr, and Mg concentration analysed by ICP-OES (Figs 8, 9).  
430 Specifically, Zn concentrations determined with LA-ICP-MS range between 4.2 and 87ppm ( $\sigma = 11.4$  ppm)  
431 with a mean value of 20.2ppm. Cadmium concentrations range between 0.1 and 1.7ppm ( $\sigma = 0.36$ ppm) with  
432 a corresponding mean value of 0.7ppm. Uranium concentrations range between 0.016 and 0.18ppm (mean  
433 = 0.06ppm;  $\sigma = 0.083$ ppm). Despite differences in total concentrations of these four elements and therefore  
434 also the varying ranges, all of them show a very similar symmetric pattern. Concentrations are generally  
435 highest in the outermost cement fringes (1t and 1b; Fig. 5) but decrease in layers 2t/b and increase again in  
436 layers 3t/b. Decreasing abundances are found in layers 4t/b.

437 Figure 10 displays the patterns of shale-normalized REE concentrations from all transect data. All  
438 samples are characterized by an overall slight increase in concentrations from light to heavier REE.  
439 Distinctly lower Ce concentrations compared to other REE are visible for all samples except one, this  
440 phenomenon is called cerium-anomaly (e.g., Alibo and Nozaki, 1999). The calculated Ce anomalies  
441 ( $Ce/Ce^*$ ) range between 0.03 and 1.2 with a mean value of 0.17 (Fig. 11). A distinct increase in Ce-anomaly  
442 is present in cement layers 4b and t, respectively. Values in the middle part vary between 0.08 and 1.2 and  
443 one maximum value of 1.2 while in the outer cement layers (1 t/b to 3 t/b) are mostly below 0.3.

444 Clumped isotope  $\Delta_{47}$  values of four radiaxial calcite samples from Hydra suggest temperatures  
445 between 179°C and 202°C. Specifically, the sample taken from the earliest cement rim 1b has a  $\Delta_{47}$  value  
446 of  $0.422 \pm 0.023\%$ , suggesting a calculated temperature of  $199 \pm 31^\circ\text{C}$ . The sample from the coeval cement  
447 rim 1t displays a  $\Delta_{47}$  value of  $0.420 \pm 0.006\%$  translating into a calculated temperature of  $202 \pm 7^\circ\text{C}$ . The two  
448 samples taken from layer 4b/t, i.e. the latest phase precipitated, yield  $\Delta_{47}$  values of  $0.431 \pm 0.019\%$  (bright  
449 luminescent) and  $0.439 \pm 0.007\%$  (n.l.) corresponding to temperatures of  $188 \pm 24^\circ\text{C}$  (bright luminescent) and  
450  $179 \pm 8^\circ\text{C}$  (non-luminescent). Within the limitations of 1SD, the calculated temperatures are identical (mean:  
451  $192 \pm 11$ ). In order to further evaluate the clumped isotope data, conventional  $\delta^{18}\text{O}$  and  $\delta^{13}\text{C}$  values, and such  
452 from the clumped analysis are compared. Carbon isotope values of the clumped isotopes method yield  
453 constant values with low uncertainties ( $\sim 0.06\%$ ), and are in very good agreement with the values analysed

454 using the MAT253 GasBench. Oxygen isotope ratios derived from clumped isotope measurements show  
455 higher uncertainties (~0.45‰) but are in agreement with those measured with the MAT253 GasBench  
456 approach.

457

## 458 **6. Interpretation and Discussion**

459

460 The main focus of this study is to re-evaluate the potential of time-resolved successions of radialial  
461 fibrous calcites as archives of their marine to burial diagenetic pathways. In order to achieve this, we  
462 compared spatially randomly distributed samples with those collected along a specific cement succession.  
463 A first important step is the subdivision of patterns assigned to different diagenetic stages. This is undertaken  
464 by assessing isotope data in the context of the cathodoluminescence patterns of the specific sample material  
465 and by comparing data shown here with previous work.

466

### 467 *6.1. Spatially distributed samples: Marine porewater diagenesis and comparison to published data sets*

468 The comparison of luminescence characteristics and isotope data from all samples reveals a  
469 significant pattern (Fig. 6). The most obvious relation between non-, patchy and bright luminescent calcites  
470 and isotope values is found in the Pennsylvanian data set, specifically with reference to oxygen isotopes  
471 (Fig. 6A). As commonly inferred (e.g., Bruhn et al., 1995; Kaufmann and Wendt, 2000), increasing  
472 luminescence patterns point to reducing conditions during burial favouring  $Mn^{2+}$  incorporation in the crystal  
473 lattice and agree with decreasing oxygen isotope values corresponding to increasingly warm pore fluids.  
474 The Upper Triassic data set is characterized by narrow oxygen and moderately variable carbon isotope ratios  
475 being clearly related to luminescence (Fig. 6B). Upper Cretaceous samples display moderately  $^{18}O$  enriched  
476 oxygen isotope values of non-luminescent, relative to luminescent cements, and near-invariant carbon  
477 isotope ratios. Two preliminary conclusions can be drawn: (i) Luminescence patterns are in rather good  
478 agreement with isotope data, mainly with reference to oxygen and less clearly to carbon isotope ratios. (ii)

479 It seems that the general assignment of bright luminescence as a priori diagenetic geochemical signature is  
480 not always true and has to be evaluated for every sample by a careful comparison of different data.

481 Following previous work of Tobin et al. (1996), the isotope patterns observed here for samples of  
482 different time periods are perhaps best explained by cementation and subsequent small-scale  
483 dissolution/reprecipitation during early diagenetic, shallow marine burial. Corresponding fluids are thought  
484 to be still marine to modified marine in origin except for bright luminescent samples from Pennsylvanian  
485 where burial fluids seem dominant. This is indicated by the considerable scatter in oxygen isotope values  
486 and the corresponding high calculated temperatures. The temperature equation used here in order to assess  
487  $\delta^{18}\text{O}$  derived fluid temperatures is that of Grossman (2012; derived from Kim and O'Neil, 1997). Patchy to  
488 bright luminescent Pennsylvanian cements translate into fluid temperatures between 15 and 37°C, pointing  
489 to shallow burial diagenesis. Pennsylvanian non-luminescent samples, on the other hand, reveal calculated  
490 fluid temperatures from  $\delta^{18}\text{O}$  values ranging from 1 to 26°C with a mean value of ~12°C.

491 Conversely, applying the same equation to the Upper Triassic samples, calculated fluid temperatures  
492 range between 13 and 16°C (a  $\delta^{18}\text{O}_{\text{water}}$  of -1 has been used, assuming an ice-free world), i.e., a range  
493 indicative of the normal marine domain. Calculated fluid temperatures from Cretaceous samples vary  
494 between ~14 and 24°C. Summing up, the data compiled here are, according to conventional interpretations,  
495 is best understood in the context of diagenetic stabilisation of a high-Mg radiaxial fibrous calcite precursor  
496 phase under very early marine burial conditions. Evidence for a significant later burial overprint is not found.  
497 An exception is found in the case of a bright luminescent zone (Fig. 5C; 4b/t, left) in portions of the centre  
498 of the cement-filled pore space in the Triassic case example. Specifically, the bright luminescent zone is  
499 related to an area with abundant fracturing and dense burial veining. Whilst the veins are filled with orange-  
500 brown luminescent cement, the radiaxial fibrous cement is altered and portions replaced by orange-brown  
501 and yellow luminescent phases.

502 When comparing the RFC carbon and oxygen isotope data presented in this study with estimates of  
503 coeval marine seawater values published in previous work, we obtain the following results. Hasiuk and  
504 Lohmann (2008) document, in comparison to data shown here,  $^{13}\text{C}$  depleted carbon isotope ratios for

505 Mississippian (Tournaisian–Visean) marine radiaxial fibrous cements. The meaning of this is difficult to  
506 assess and the problem of comparing data from different epochs of the Carboniferous, and from different  
507 settings, is acknowledged here.

508         Conversely, calcite  $\delta^{13}\text{C}$  and  $\delta^{18}\text{O}$  isotope ratios shown here are either in the same range or enriched  
509 relative to previous data (Fig. 6; e.g., Bruckschen et al., 1999; Veizer et al., 1999). Specifically, non-  
510 luminescent samples of Pennsylvanian and all samples of Upper Triassic radiaxial calcites plot towards the  
511  $^{13}\text{C}$  and  $^{18}\text{O}$  enriched range of these data sets (Fig. 6). Conversely, the Upper Cretaceous ones agree in terms  
512 of their  $\delta^{18}\text{O}$  range but are enriched in  $^{13}\text{C}$  relative to those shown in Bruckschen et al. (1999) and Veizer et  
513 al. (1999). Having said this, it must be noted that the data shown in Bruckschen et al. (1999) and Veizer et  
514 al. (1999) heavily rely on biogenic marine carbonate archives (skeletal carbonates; e.g., brachiopod shells)  
515 that should not be compared with inorganic precipitates in a non-critical manner (Immenhauser et al., 2015).

516         Summing up, the bulk of published data on radiaxial calcites from the time slices discussed here is  
517 remarkably small, too small to allow for a more systematic comparison. Moreover, only a limited number  
518 of the studies published compare geochemical data with luminescence properties of these cements.

519

520

## 521 *6.2. Relation of calcite elemental concentrations to luminescence properties*

522         Elemental data reveal increasingly depleted Sr and Mg concentrations from non- via patchy to bright  
523 luminescent cements. Both of these elements are commonly enriched in marine abiogenic precipitates and  
524 consequently, depletion of these elements is usually assigned to secondary diagenetic controls (e.g., Brand  
525 and Veizer, 1980; Satterley et al., 1994). Depletion in Sr and Mg coincides with an increase in Mn (activator)  
526 and Fe (quencher) concentrations of cements with increasing luminescence, the latter is usually assigned to  
527 burial diagenesis (~1 ppm in abiotic marine precipitates under oxic conditions; Major and Wilber, 1991  
528 versus >400ppm Mn in cements indicative of anoxic, non-marine fluids; Bruckschen et al., 1999). Satterley  
529 et al. (1994) suggest, that marine carbonates yield concentrations of less than 100 ppm Fe and less than 10  
530 ppm Mn. Considering the concentrations measured in the radiaxial cements studied here, exchange of

531 calcites with marine burial porewater seems likely. Tobin et al. (1996) describe radiaxial fibrous calcite  
532 cements from the Ordovician of Tennessee with Mn concentrations of  $\leq 10,000$  ppm and patchy  
533 luminescence patterns and associate these elemental values and luminescence patterns to overprint of a  
534 marine magnesian calcite precursor phase under reducing diagenetic conditions. For comparison, elemental  
535 data shown in this study have maximum values of 600 ppm Mn but more commonly range between some  
536 tens to a few hundred ppm (Fig. 7). Increasingly bright luminescence is assigned to increasing calcite Mn  
537 concentrations but values must not necessarily reach those considered indicative for diagenetic overprint  
538 under sub- or anoxic burial conditions. Specifically, the hot cathode device used in the context of this study  
539 is very sensitive to Mn and probably highlights patterns in Mn concentrations related to precipitation kinetics  
540 (Habermann et al., 1998) if not quenched by iron (ten Have and Heijnen, 1985) as discussed for the case  
541 example of Fe below.

542 A similar pattern is found in the case of Fe elemental concentrations. Specifically, Tobin et al. (1996)  
543 describe luminescent radiaxial fibrous calcites of Ordovician age with mean Fe concentrations of  $\leq 2000$   
544 ppm (400ppm and less for non-luminescent phases). Conversely, data presented in figure 7 show Fe  
545 concentrations of less than 500 ppm, with most values remaining below 250 ppm. Ten Have and Heijnen  
546 (1985) argue that where calcite iron concentrations remain below 200ppm, luminescence induced by Mn is  
547 not quenched. Based on precipitation experiments, ten Have and Heijnen (1985) documented that it is rather  
548 the absolute amount of calcite  $Mn^{2+}$ , as opposed to the  $Fe^{2+}/Mn^{2+}$  ratio, and equally important, changes in  
549 the rate of crystal growth that affect luminescence zonations. This implies that kinetic factors (e.g., calcite  
550 growth rates) interact in a complex manner with environmental ones (e.g., fluid chemistry). In the context  
551 of the paragenetic succession studied here, it seems more than likely that growth rates of successive  
552 generations of radiaxial calcites changed with time. The highest growth rates are expected during the early  
553 stages of radiaxial fibrous cement formation (i.e., phase 1b/t in Fig. 5) when the water circulation in the pore  
554 space is vigorous due to wave and current action. Under increasing cementation of pore space and related  
555 clogging of fluid circulation pathways, precipitation/recrystallization rates continuously decrease.

556 Generally the relation between Fe concentrations and degree of luminescence is present but weak in  
557 the data set shown here. Comparing different data sets from radiaxial fibrous calcite of different time  
558 intervals and basins, however, is not straightforward. Besides differences in the geochemistry of  
559 Pennsylvanian, Upper Triassic, and Upper Cretaceous seawater and related element distribution  
560 coefficients, perhaps even subtle patterns in diagenetic fluid chemistry, burial depth, and rate or non-  
561 equilibrium factors represent important factors (e.g., Reeder and Grams, 1987; Barnaby and Rimstidt, 1989;  
562 Rimstidt et al., 1998).

563 Two preliminary conclusions can be drawn after the previously described observations. (i) The  
564 general elemental patterns found in all three case examples agree with marine (porewater) radiaxial  
565 magnesian calcites that were subject to only very moderate diagenetic overprint under slightly reducing  
566 conditions. The main source of the fluids remained seawater except for patchy and bright luminescent  
567 samples of Pennsylvanian that have seen higher degrees of alteration under non-marine fluid influence. This  
568 notion agrees with conclusions previously drawn from the light stable isotope data documented here. (ii)  
569 Elemental concentrations are reflected, at least to some degree, in the luminescence patterns of the  
570 corresponding cements.

571  
572 *6.3. Radiaxial fibrous calcites as archives of diagenetic pathways: Application of less-conventional proxies*

573  
574 Having tentatively established, by means of conventional proxies for fluid properties, the low degree  
575 of diagenetic alteration of the calcite cements documented here, we now discuss the implications of the less-  
576 conventional proxy data compiled in the transect data across radiaxial calcites in the Upper Triassic  
577 specimen Rk-1-H (Figs 5 and 8 through 12). Additionally, we reconstruct a simple example for the  
578 diagenetic history of the cement succession and construct a model for cement paragenesis in the analysed  
579 Upper Triassic reefal limestone (Fig. 13).

580  $\Delta_{47}$  values of all four samples analysed are comparatively low and fall between 0.420 and 0.439‰.  
581 This is only marginally higher than the  $\Delta_{47}$  value of Carrara marble measured in the same analytical session

582 (~-0.395‰) and suggests a high-temperature origin of the sample clumped isotope signature (Kluge et al.,  
583 2015). Using Eq. 1, the fluid temperature estimates suggested by clumped isotope data from the Upper  
584 Triassic sample Rk-1-H stand in remarkable contrast to temperature estimates based on conventional  $\delta^{18}\text{O}$   
585 data using the temperature equation of Grossman (2012). Specifically, the  $\delta^{18}\text{O}$  ratios of the calcite layers  
586 4t/b, i.e. the last recorded precipitates suggest a maximum fluid temperature of  $<20^\circ\text{C}$  whilst those indicated  
587 by the clumped isotope thermometer applied to the same sample point to temperatures between 178 and  
588  $202^\circ\text{C}$ . If these temperatures are taken to represent a direct recrystallization of the calcite, and assuming a  
589 temperature/depth gradient of  $25^\circ\text{C}/\text{km}$  (Fridleifsson et al., 2008), a burial depth of  $\sim 7\text{-}8$  km is necessary to  
590 explain the data, whilst shallower depths of about 6-7 km or less are necessary when a gradient of  $30^\circ\text{C}/\text{km}$   
591 is taken. In any case, a burial depth of several kilometres in the context of the thrusting of tectonic naps  
592 (Römermann, 1969; Richter, 1999) that now build Hydra Island is likely. During recrystallization, the “true”  
593 temperature is recorded in  $\Delta_{47}$ , irrelevant how fast the uplift was. Uplift or cooling rates only become  
594 relevant when solid-state diffusion within crystals is assumed (Passey and Henkes, 2012; Stolper and Eiler,  
595 2015). Thus, a  $\Delta_{47}$  value between the closure temperature of clumped isotopes ( $\sim 100^\circ\text{C}$ ; Henkes et al., 2014)  
596 and the maximum burial temperature is expected. Assuming slow uplift rates and related slow cooling of  
597 the carbonate successions studied here, the clumped isotope temperatures are largely in agreement with  
598 earlier findings by Gillhaus et al. (1999) from Hydra Island reporting the outcome of different temperature  
599 proxies (e.g., illite crystallinity, authigenic albites, chert maturity) that suggest burial temperatures between  
600  $135$  and  $210^\circ\text{C}$ .

601         The discrepancy of two isotope-based palaeo-thermometers as observed here is remarkable but not  
602 unique as such. A prominent example is found in the case of Carrara marble, having seen peak temperatures  
603 in excess of  $200^\circ\text{C}$  (Vaselli et al., 2012) whilst being characterized by marine  $\delta^{13}\text{C}$  ratios of  $+1.5$  to  $+3\text{‰}$   
604 and  $\delta^{18}\text{O}$  ratios of  $-0.2$  to  $-3.5\text{‰}$  (Herz and Dean, 1986). The underlying reasons are not easily understood  
605 but perhaps best explained in the context of solid-state resetting (Passey and Henkes, 2012). Nevertheless,  
606 numerous workers have documented that marine calcites in the presence of a reducing burial pore fluid with  
607 significantly elevated temperatures undergo dissolution-reprecipitation (recrystallization) and exchange via



608 diffusion (Watson and Baxter, 2007) leading mainly to  $^{18}\text{O}$  depleted values. In the absence of a fluid phase,  
609 however, calcite  $\delta^{18}\text{O}$  ratios might be locked whilst the  $^{13}\text{C}$ - $^{18}\text{O}$  bond reordering responds to the elevated  
610 temperatures (e.g., Henkes et al., 2014).

611 The concept of burial in the absence of significant amounts of fluids, however, is not easily  
612 understood. On the level of a tentative working hypothesis, it is possible that marine radiaxial calcites  
613 occluded all of the available pore space and hence, fluid circulation in these tight limestones was reduced  
614 to a degree that fluid-carbonate interaction was a surficial rather than a volume phenomenon. This concept  
615 is in agreement with the conspicuous absence of genuine burial phases such as blocky calcites or saddle  
616 dolomites along the sampled transect that focussed on radiaxial fibrous calcites (Fig. 5). Essentially, some  
617 open pore space was present when entering the burial domain as documented in the central portions of the  
618 hand specimen shown in figure 5, but not in the sampled part of the specimen. The hypothesis that significant  
619 rock-fluid interaction was absent during most of the burial history of the studied sample is also supported  
620 by the spatially very limited alteration zone observed in luminescence zones 4b/t. Here, late burial fluids  
621 circulated along small veins and affected only a limited portion of the fibrous calcites (Fig. 5C).

622 A second remarkable observation is found in the patterns of redox-sensitive elements, particularly  
623 Zn, Cd, U and Cu (Fig. 9). Following previous workers (Thomson et al., 1995; Morford and Emerson, 1999;  
624 Tribovillard et al., 2006; Pattan and Pearce, 2009) enriched elemental concentrations of these elements point  
625 to increasingly reducing fluids. According to the data shown here, the fluid oxygen level reached highest  
626 values in cement phases 2b and 2t and in the late stage phases 4b and 4t (Fig. 9). This seems counter-intuitive  
627 given that pore fluids at a depth of a few mm to few cm in the sediment column reach the methanogenic  
628 diagenetic stage due to rapid respiration of organic matter (see discussion in Immenhauser et al., 2008).  
629 Additional evidence, that might shed light on the processes involved, comes from Mn elemental trends (Fig.  
630 8) and REE data (Fig. 10) compiled across the same transect.

631 Essentially, Mn concentrations are stable in cement phase 1t/b and then gradually increase towards the  
632 central suture (Fig. 8). This is the pattern expected from a gradually decreasing fluid  $E_h$  favouring  $\text{Mn}^{2+}$  as  
633 the dominant species (Bruckschen and Richter, 1994). A similar but less distinct gradual behaviour is only

634 observed for calculated Ce-anomalies which also points to gradually reducing conditions of radial fibrous  
635 calcite formation (i.e., the stabilisation of a precursor phase). Especially, the fully marine cement generations  
636 1b and 1t (Fig. 8) reflect well oxygenated seawater with a dominantly incompatible  $Mn^{4+}$  phase. At the onset  
637 of generation 2b/t, Mn incorporation in the crystal lattice increases due to increasingly reducing porewater  
638 chemistry characterized by  $Mn^{2+}$ . Elements like Zn, Cd, U and Cu are often considered to reflect fluid Eh.  
639 If true, this would imply oxidized pore fluids for the last cement generation precipitated. The controversial  
640 patterns found in Zn, Cd, U, and Cu could be related to non-equilibrium processes, especially with respect  
641 to variations in growth rates that strongly affect the incorporation of these elements in the calcite crystal  
642 lattice. In addition, it seems possible that some of the redox sensitive elements are less mobile (distribution  
643 coefficients) and more difficult to incorporate in the calcite lattice.

644 Rare earth element patterns (Fig. 10) display an upward enrichment trend towards the heavier REEs,  
645 a feature indicative for precipitation from marine fluids. Given that all cement phases display the typical  
646 REE pattern of seawater with a well-developed negative Ce-anomaly for most of them, except one (Fig. 10;  
647 Lécuyer et al., 2004; Olivier and Boyet, 2006; Bodin et al., 2013), it seems likely that the stabilization from  
648 an initially precipitated marine magnesium-rich radial fibrous calcite to a more Mg-depleted phase took  
649 place at an early burial stage in the presence of marine pore fluids.

650 Paleo-redox conditions can also be inferred from evolution of the Ce-anomaly ( $Ce/Ce^*$ , Fig. 11).  
651  $Ce/Ce^*$  values fluctuate around a value of 0.1 across cement layers 1b-3b and across cement layers 1t-3t. A  
652 value of 0.4 is representative for modern open-ocean shallow oxygenated seawater and probably also for  
653 Mesozoic seawater (Shields and Webb, 2004; Bodin et al., 2013). At the onset of cement rims 4b/t, Ce  
654 anomaly shifts distinctly towards values around 0.3, indicative for slightly more reducing conditions  
655 compared to the other cement phases, but still indicative for oxygenated waters.

656 In order to check whether the calculated Ce anomalies are due to a genuine depletion of Ce and not  
657 a positive La anomaly, the Pr-anomaly ( $Pr^*$ ) has been calculated and plotted against the Ce-anomaly which  
658 is then called La-anomaly plot, as suggested by Bau and Dulski (1996) (Fig. 12). This diagram helps to  
659 differ if a Ce-anomaly is of genuine nature or if it is caused by anomalous behaviour of other REEs. Data

660 that plots in field IIIb, as the vast majority of our data do, show real Ce-anomalies. Only a few data points  
661 do not plot in this field. The one data point that plots in field IIIa can be assigned to the maximum value of  
662 Ce-anomalies (a positive Ce-anomaly) and originates in the sampling of the burial vein that crosscuts the  
663 sampled cement transect. Plotting in the IIIa field however indicates that this is without a doubt a genuine  
664 positive Ce anomaly. Six out of the total of 94 data points plot in field IIa of the Bau and Dulski (1996) plot  
665 and may thus be related to anomalous enrichment/depletion of neighbouring elements. Those have been  
666 closely examined for their individual REE patterns and we tend to conclude from this examination that also  
667 for these samples a genuine negative Ce-anomaly can be assumed. Their slightly different behaviour can  
668 rather be linked to a slightly depleted Pr content compared to the other patterns than to any anomalous  
669 enrichment in La. We thus conclude that all Ce anomalies calculated here are genuinely related to  
670 depletion/enrichment of Ce content relative to the other REE. Even if we exclude the six samples plotting  
671 in the IIa field, no significant change in the Ce anomaly trend along the cement phases can be observed.

672         Assuming that these data and interpretations are valid, pore fluids did not reach fully anoxic  
673 conditions during late stage radiaxial calcite precipitation or diagenetic stabilization in the studied transect.  
674 Summing up, Mn and REE monitored paleo-redox conditions of the pore water suggesting initial radiaxial  
675 fibrous cement precipitation (1b/t) under highly oxidizing seawater conditions. These early cement seams  
676 are followed by radiaxial fibrous cement precipitation under increasingly reducing conditions but a still  
677 near-marine porewater. Other redox-sensitive proxies, and particularly Cd, Zn, Cu and U, appear to be  
678 dominated by kinetic factors rather than porewater Eh.

679

## 680 **7. Conclusions**

681

682 Marine radiaxial fibrous calcites in Pennsylvanian, Upper Triassic, and Upper Cretaceous reefal limestones  
683 are re-evaluated as archives of past seawater properties. Based on cathodoluminescence maps, fabrics are  
684 subdivided into non-, patchy- and bright luminescent types. There is a moderate agreement between isotope  
685 ratios and luminescence with all data plotting in the heavy side of reconstructed coeval seawater isotope

686 data. The commonly held notion that luminescent radiaxial calcites represent substantial diagenetic  
687 overprint represents an oversimplification. This as redox properties of (modified) marine porewater and  
688 kinetic effects (precipitation rates) affect the incorporation of activating elements in the crystal lattice and  
689 may result in the direct precipitation of luminescent radiaxial calcites in open pore space during initial burial.

690         Geochemical patterns along an Upper Triassic transect oriented along the growth directions of the  
691 cement layers display a symmetrical stratigraphic pattern from outer cement fringes towards the central  
692 suture. Three out of four data sets (Mg, Fe and Sr) and carbon and oxygen isotope values trend to more  
693 depleted isotope values or decreasing elemental concentrations towards the central suture. Conversely, Mn  
694 elemental abundances become gradually enriched. Redox-sensitive elements such as Zn, Cd, U, and Cu only  
695 in partly reflect porewater Eh whereas but more prominently kinetic factors. Conversely, Mn as well as Ce  
696 indicate oxic marine porewaters with a shift to moderately sub-oxic conditions and a decreasing influence  
697 of marine porewaters towards the latest stage radiaxial calcite precipitation.

698         Contrasting information is obtained from the comparison of two palaeothermometers:  $\delta^{18}\text{O}$  ratios  
699 ( $<20^\circ\text{C}$ ) and clumped isotope geochemistry ( $180$  to  $200^\circ\text{C}$ ). This controversy is perhaps best understood in  
700 the context of solid-state  $^{13}\text{C}$ - $^{18}\text{O}$  bond reordering in a burial setting under low fluid:rock ratios. REE patterns  
701 of all radiaxial calcite phases, suggest that these fabrics stabilized under the influence of increasingly oxygen  
702 depleted marine to altered marine porewaters at (shallow marine) burial depths but then ceased to record the  
703 subsequent burial history. Essentially, carbonate clumped isotope data represent the only proxy shown here  
704 that provides evidence of deeper burial conditions. This observation has general significance for  
705 palaeotemperature reconstructions of ancient marine carbonates.

706         The data presented here underline that the application of a multi-proxy data set including – with  
707 reference to radiaxial fibrous calcites - less conventional approaches reveals new and important insight into  
708 shallow (marine) to deep burial processes. The question whether radiaxial fibrous cement fabrics,  
709 precipitated from sub-oxic (altered) shallow marine burial fluids still qualify as marine cements *sensu stricto*  
710 merits consideration. This question seems adequate as petrographically, these fabrics are near-identical to  
711 fully marine radiaxial calcites.

712

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714

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724 **References**

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## 960 **Figure Captions**

961 **Fig. 1.** Overview map of Europe with indication of sampled locations (red crosses) in Spain and Greece.  
962 Lower inset shows position of Hydra and Dokos (Greece).

963 **Fig. 2.** Photographs of hand specimens studied in the context of this paper. (A) Red-stained Bashkirian  
964 (Pennsylvanian) slope facies from the Sierra de Cuera platform, northern Spain. Note gray areas  
965 represent predominantly pores filled by radiaxial fibrous calcites or shell fragments. (B) Radiaxial  
966 fibrous calcite (dark gray) occluding cavities in Carnian to Norian (Upper Triassic) reefal limestone  
967 (light gray) from Hydra Island, Greece. (C) Cenomanian (Upper Cretaceous) radiaxial fibrous calcite  
968 (dark gray) occluding former pores between submarine breccia clasts (light gray) from Dokos Island,  
969 Greece.

970 **Fig. 3.** Main characteristic features of radiaxial fibrous (A) and fascicular optical fibrous (B) calcites.  
971 Modified after Richter et al. (2011). (SC = subcrystal; UCC = upward concave curvature of cleavage;  
972 DCC = downward concave curvature of cleavage; CWE = clockwise extinction; CCWE =  
973 counterclockwise extinction; MD = microdolomites; Z = cloudy zonation)

974 **Fig. 4.** Thin section photomicrographs of Upper Triassic radiaxial calcites documenting three characteristic  
975 types of cathodoluminescence properties found in all analyzed samples. Arrows indicate growth  
976 directions of crystals. (A/B) Upper Triassic radiaxial fibrous calcite under normal polarized light (A) and  
977 under luminescence (B; non- to intrinsic blue luminescence = n.l.). (C/D) Upper Triassic radiaxial calcite  
978 under normal polarized light (C) and showing patchy luminescence (D, p.l.). Boundaries between patchy

979 luminescent (p.l.) and non-luminescent (n.l.) portions of cements are well defined. (E) Polarized light  
980 photomicrographs of Upper Triassic radiaxial fibrous calcite fringes in hand specimen from Hydra Island  
981 (Fig. 2B). The image depicts central suture (S) of converging cement seams. (F) Image of same area  
982 under cathodoluminescence showing bright luminescence (b.l.).

983 **Fig. 5.** (A) Triassic hand specimen Rk-1-H. Black lines indicate boundary between sedimentary substratum  
984 (reefal limestone) and the cavity occluded by four subsequent generations of radiaxial fibrous calcites  
985 (RFC). Red dashed rectangle indicates position of the thin section shown in (C/D). (B) Cartoon  
986 illustrating position of two sampling transects (computer-controlled milling device used for sampling for  
987 isotope and ICP-MS analysis and LA-ICP-MS laser track, slightly below milled transect). (C)  
988 Photomicrograph of thin section mapped under crossed polarized light. Different generations of radiaxial  
989 fibrous calcite are labelled 1b/t through 4b/t. Cements precipitating on floor of cavity are indicated b  
990 (base), those growing from the ceiling of the pore towards the center of the pore are labelled t (top).  
991 Central suture is denoted as S. (D) Cathodoluminescence map of same area as C. White rectangle  
992 indicates position of the transect sampled for geochemical analyses (5B). White circles (C1-C4) denote  
993 sampling positions for clumped isotope measurements. White arrows indicate growth directions of  
994 cement paragenesis.

995 **Fig. 6.** Cross-plots of carbon and oxygen isotope data from all measured carbonate samples compared with  
996 reconstructed coeval seawater values from Veizer et al. (1999; green rectangles) and Bruckschen et al.  
997 (1999; black rectangle) and with carbon isotope values of marine calcite cements reported in Carpenter  
998 and Lohmann (1997, red dashed line). Note key to color code for non-, patchy and bright luminescent  
999 cements in lower right. (A) Isotope data from Bashkirian non-luminescent radiaxial calcites are enriched  
1000 in  $^{18}\text{O}$  relative to reconstructed seawater values whilst patchy and bright luminescent data plot within  
1001 reconstructed marine range. (B) Data from Upper Triassic (Carnian/Norian) radiaxial calcites plot in  
1002 narrow oxygen isotope range with luminescent sampling material being moderately depleted in  $^{13}\text{C}$

1003 relative to non-luminescent ones. (C) Upper Cretaceous (Cenomanian) samples being enriched in  $^{13}\text{C}$   
1004 relative to published values.

1005 **Fig. 7.** Main and trace elemental concentrations of Carboniferous (Pennsylvanian; Bashkirian, Bashk.),  
1006 Upper Triassic (Norian, Nor) and Upper Cretaceous (Cenomanian, Cen) radial fibrous calcites plotted  
1007 according to luminescence properties. Note generally lower elemental concentrations of Triassic and  
1008 Cretaceous relative to Carboniferous cements. Decreasing  $\text{Fe}^{2+}$  and increasing  $\text{Mn}^{2+}$  concentrations as  
1009 generally assigned to increasingly luminescent calcites are present in some cases (Pennsylvanian) but  
1010 not obvious in Cretaceous and Triassic samples. Key for color coding is given in the lower right.

1011 **Fig. 8.** Carbon and oxygen isotope as well as elemental abundances (lines show five-point moving average)  
1012 plotted against the sampled transect in Upper Triassic radial fibrous calcites (refer to Fig. 5 for location of  
1013 sampled transect). Cathodoluminescence properties are shown to the left. Central suture is labelled S.  
1014 Growth directions are indicated with arrows. Note symmetrical patterns in isotope ratios and elemental  
1015 concentrations on both sides of central suture indicating that cements were not significantly overprinted.

1016 **Fig. 9.** LA-ICP-MS data of Zn, Cd, U and Cu concentrations across transect in Upper Triassic case example.  
1017 Lines show five-point moving averages. Symmetric patterns in Zn, Cd, Cu and U on both sides of the  
1018 central suture (S) correlated to cement rims suggest absence of significant late stage overprint. Labels  
1019 1b/t through 4b/t correspond to the different cement phases (Figs 5 and 8). Central suture is labelled with  
1020 S, growth direction indicated by arrows.

1021 **Fig. 10.** Characteristic seawater REE patterns of the radial fibrous calcite transect normalized to NASC  
1022 (North American Shale Composite) values. Data are summarized for the different cement layers (1t/b  
1023 through 4t/b) and labelled correspondingly. Ce concentrations are distinctly lower compared to the other  
1024 REE (negative Ce-anomaly). One data point (layer 4t) shows positive Ce-anomaly. Thick lines indicate  
1025 mean values of each cement layer.

1026 **Fig. 11.**  $\text{Ce}/\text{Ce}^*$  plot across sampled transect (cerium anomaly). Red line shows Lowess smoothing (10pts,  
1027  $\alpha=0.1$ ). Note significant  $\text{Ce}/\text{Ce}^*$  peak at central suture indicating trend towards decreasingly marine and

1028 increasingly suboxic conditions. This peak can be assigned to burial vein that crosscuts the transect.

1029 Central suture is indicated with S, growth directions are given as arrows.

1030 **Fig. 12.** Ce/Ce\* vs. Pr/Pr\* diagram (called La-anomaly) after Bau & Dulski (1996) to check for genuineness  
1031 of Ce-anomalies. Data points that plot in field I show no anomaly. Data in field IIa shows positive La-  
1032 anomaly which causes an apparent negative Ce-anomaly. IIb has positive La anomaly which leads to  
1033 apparent positive Ce-anomaly. Data in field IIIa shows genuine positive Ce-anomaly. All data points that  
1034 plot in field IIIb show a genuine negative Ce-anomaly. Data points are color coded regarding their  
1035 assignment to the different cement layers.

1036 **Fig. 13.** Paragenetic succession summarizing different depositional to diagenetic stages recorded in radiaxial  
1037 fibrous calcites in sample Rk-1-H.

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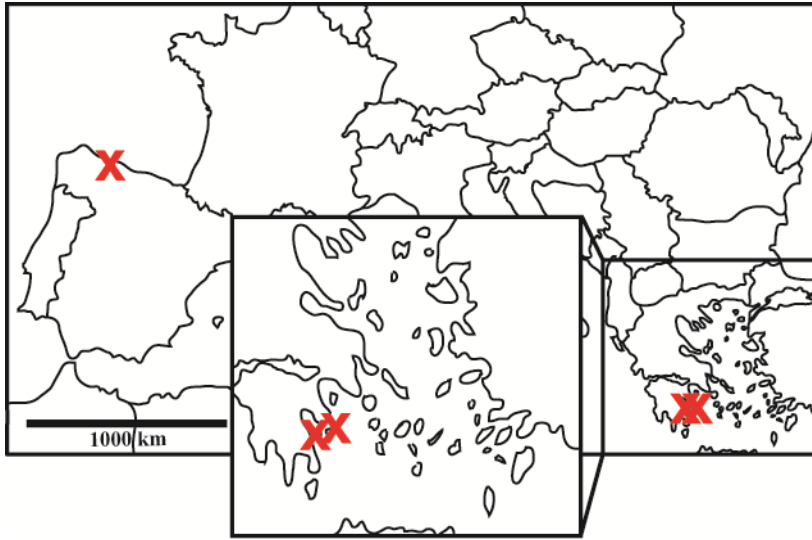
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1056 Fig. 1

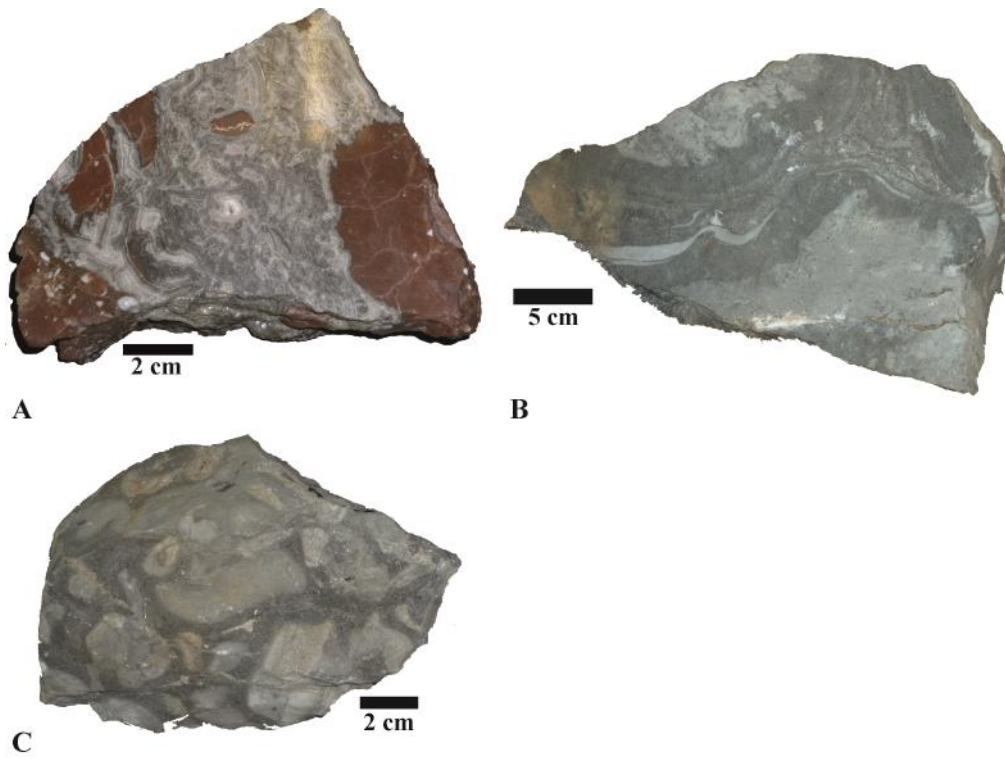


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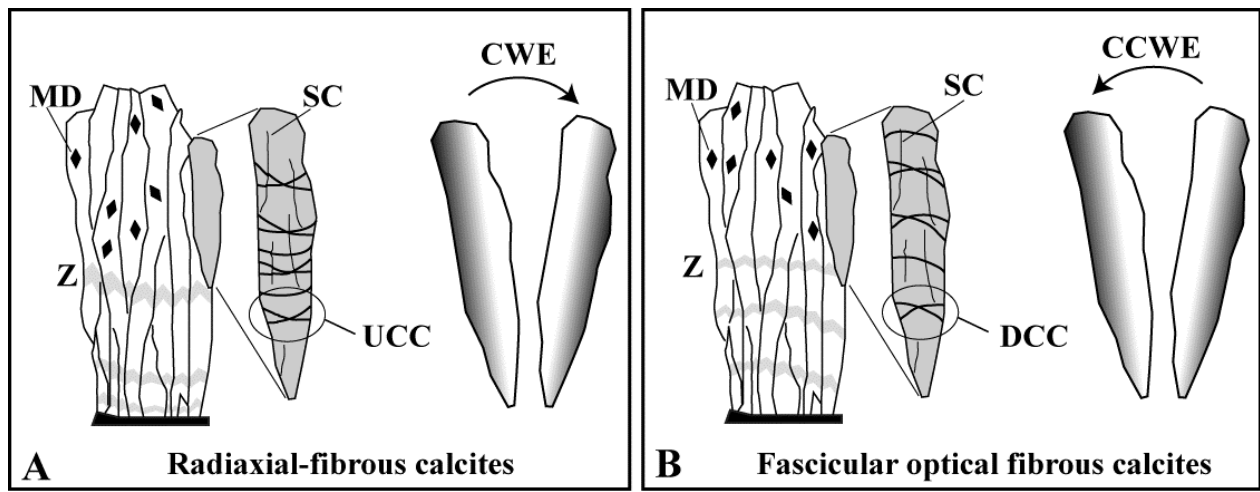


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1064 Fig. 3

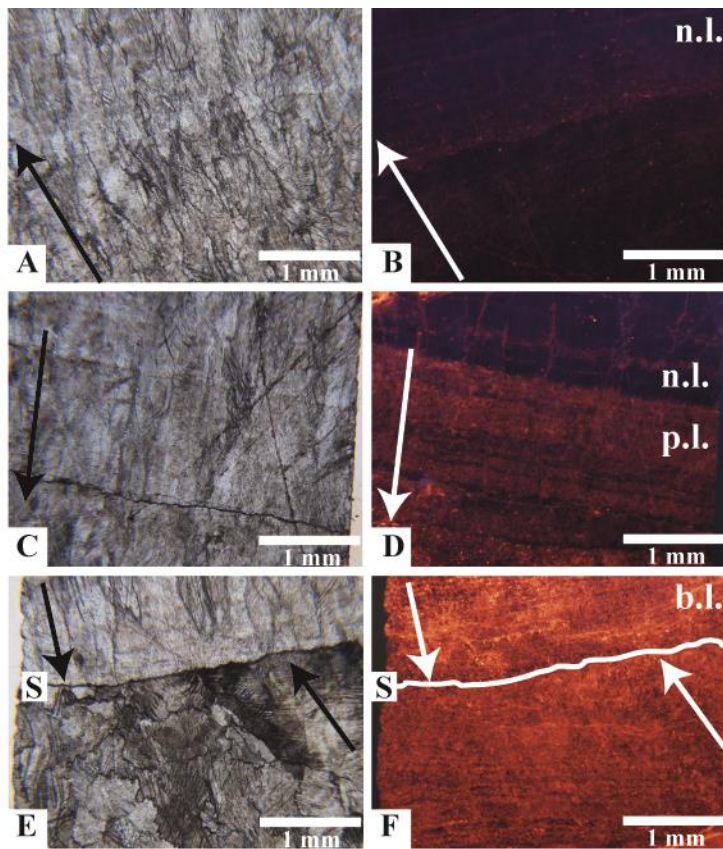


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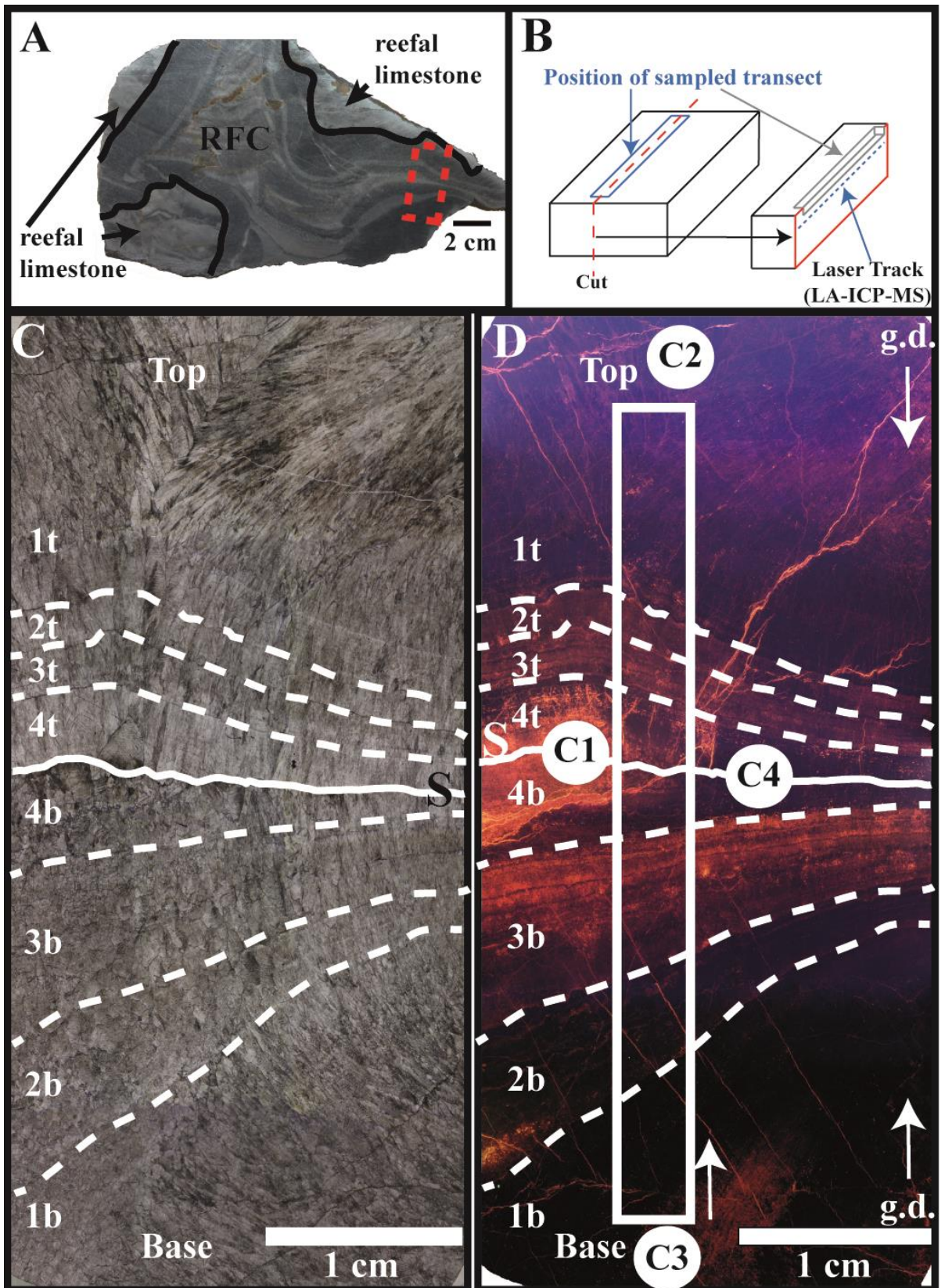
1068 Fig. 4



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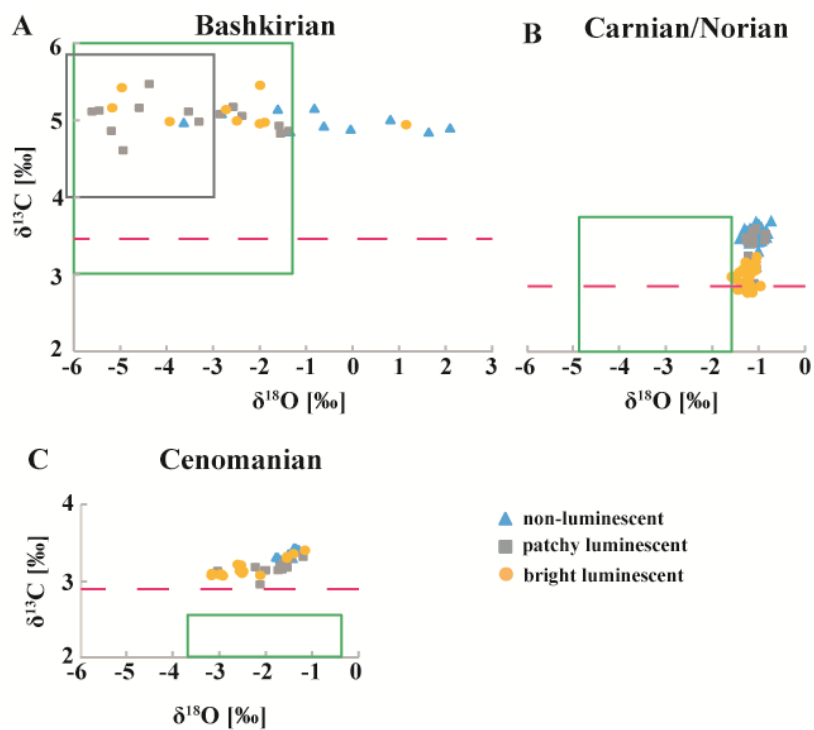
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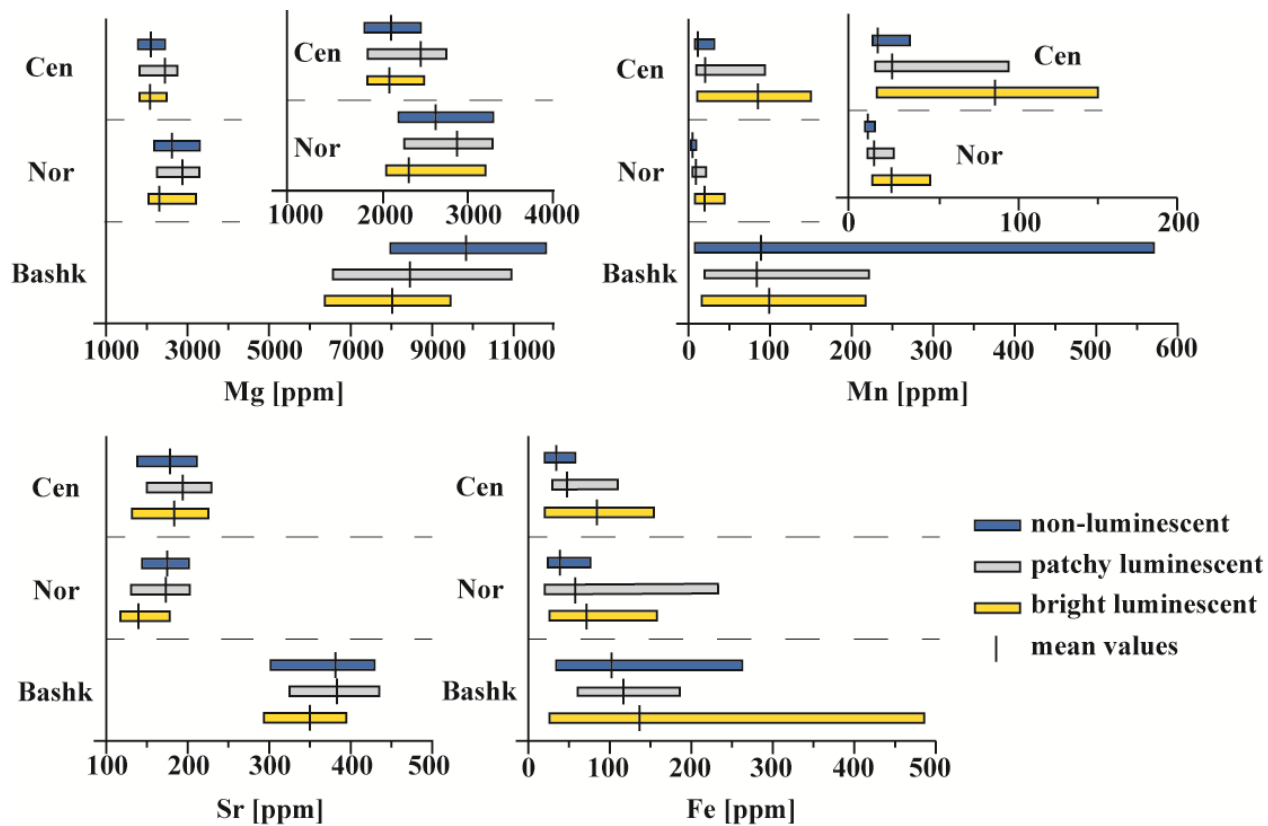


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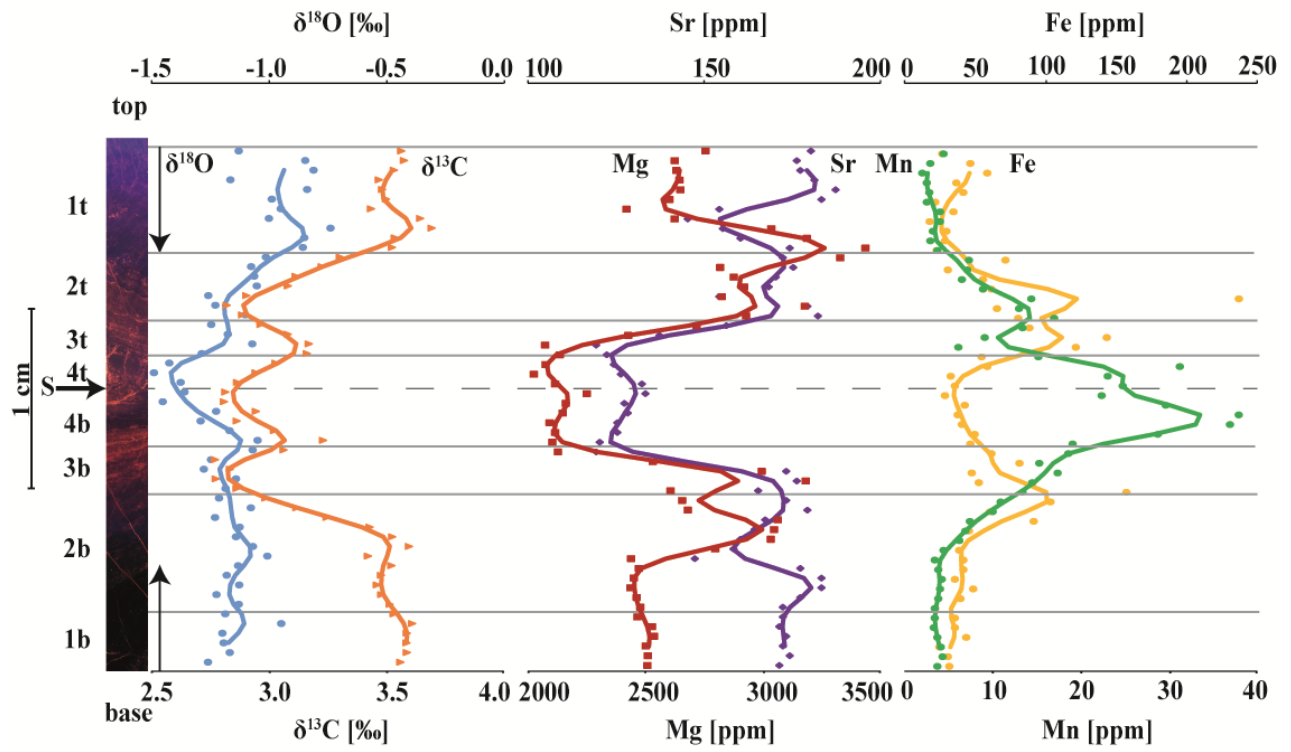
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1075 Fig. 7



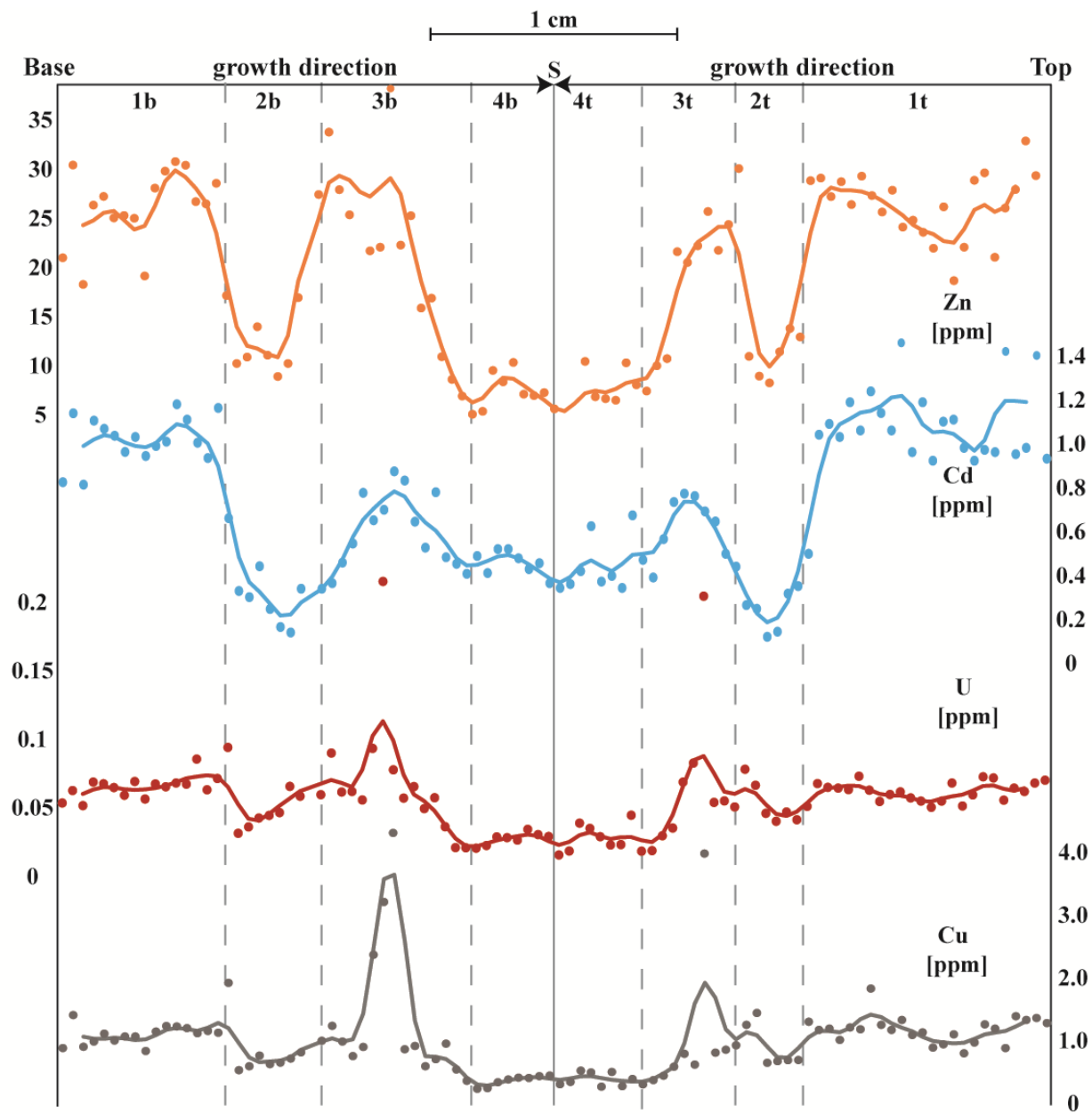
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1077 Fig. 8



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1092 Fig. 9



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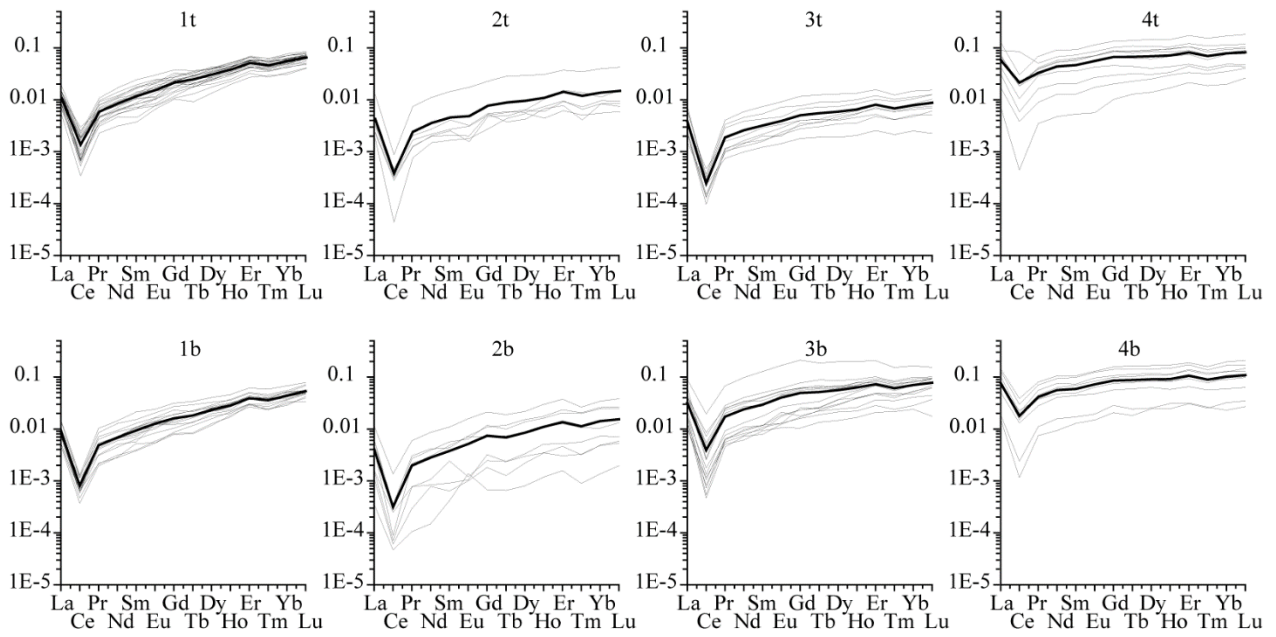
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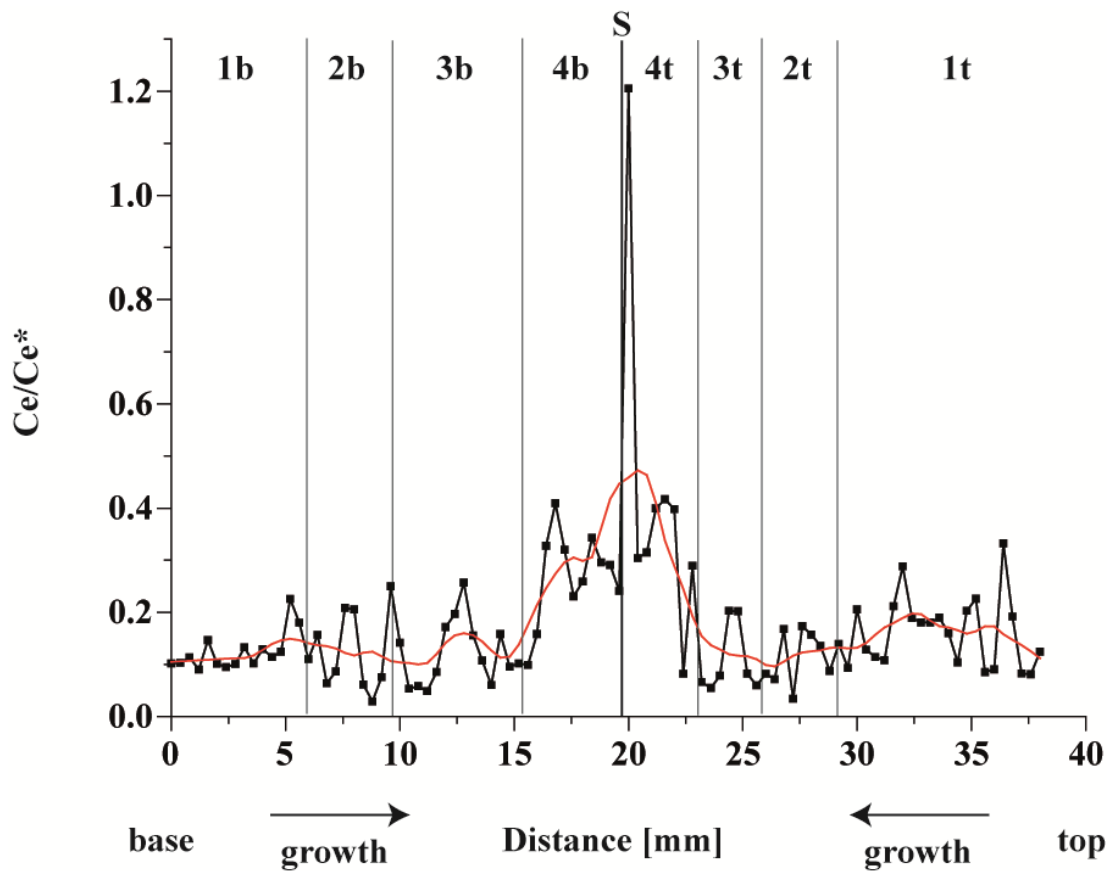
1100 Fig. 10



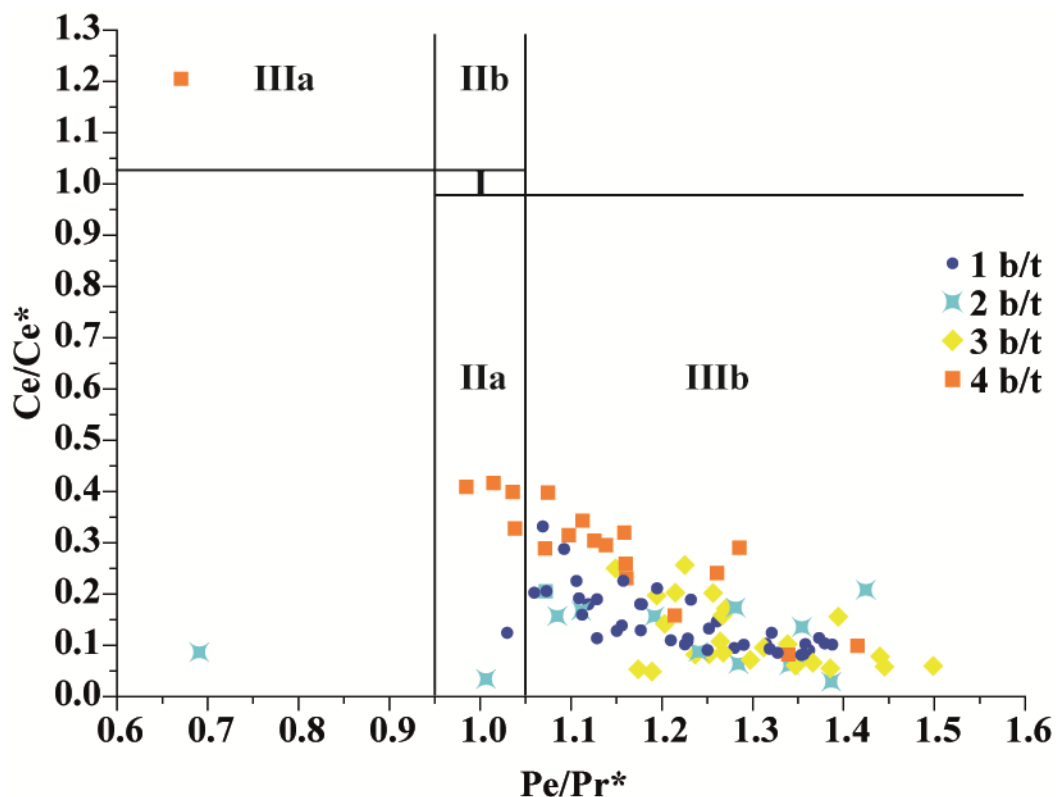
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1103 Fig. 11



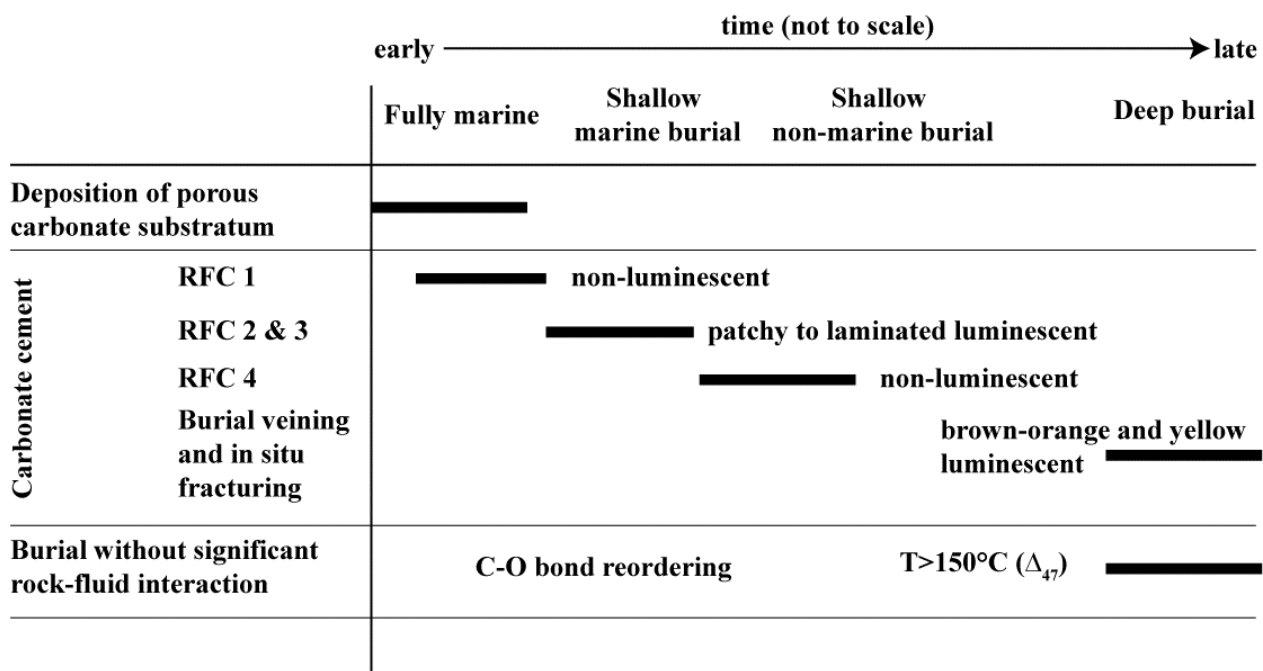
1117 Fig. 12



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Locality	Asturias (Spain)	Hydra (Greece)	Dokos (Greece)
Age	Late Carboniferous (Pennsylvanian (Bashkirian))	Late Triassic (Carman)	Late Cretaceous (Cenomanian)
Carbonate platform type	High-rising platform (up to 850m), progradational geometry	Reef complex, progradational geometry	
Location on platform	Upper slope (150 - 350m below platform top)	Platform top, neritic	Platform top, neritic
Host facies	Red-stained microbial boundstone	Reefal limestone	Polymict limestone breccia
Sea-level-mode	Transgressive	Transgressive	Transgressive
Climate-mode	Icelhouse	Greenhouse ("Hothouse")	Greenhouse
Global seawater chemistry	Aragonite-II sea	Aragonite-II sea	Calcite-II sea
	van der Kooij et al. (2007, 2009, 2010)	Richter (1999)	Bachmann and Riesch (1979), Clift & Robertson (1990)

**Table 1.** Description of the three different case settings analyzed in this study.

luminescence pattern	$\delta^{13}\text{C}$ [‰ V-PDB]		$\delta^{18}\text{O}$ [‰ V-PDB]		Ca [ppm]		Mg [ppm]		Sr [ppm]		Fe [ppm]		Mn [ppm]		Ba [ppm]									
	Min	Max	Mean	Min	Max	Mean	Min	Max	Mean	Min	Max	Mean	Min	Max	Mean	Min	Max	Mean						
<b>Carboniferous</b>	4.8	5.2	5.0	-3.6	2.1	-0.7	379280	387230	384144	7482	11770	9901	302	430	382.2	32	266	103	5	571	88.1	0.8	2.3	1.3
<b>Asturias</b>	4.6	5.2	5.0	-3.6	2.1	-3.5	372580	389650	384629	6553	10930	8468	324	436	381.5	62	186	119.4	22.9	223	85.2	1.3	2.8	1.9
	4.9	5.5	5.1	-5.2	1.1	-2.7	380960	390500	386582	6372	9473	8043	296	394	350.1	25	485	138.1	14.1	219	98.3	0.9	13.9	3.1
<b>Triassic</b>	3.1	3.7	3.5	-1.4	-0.7	-1.1	386590	400210	394995	2397	3325	2616	145	203	175	17	71	38	1.9	7.2	3.9	0.8	1.7	1.4
<b>Hydra</b>	2.8	3.6	3.3	-1.4	-0.9	-1.2	386470	402620	394839	2245	3432	2830	133	205	171	22	237	59	3.6	22.3	9.3	0.9	2.1	1.3
	2.8	3.2	3.0	-1.6	-1.0	-1.3	389270	403950	395305	2019	3178	2357	118	179	139	27	157	66	6	41.2	20.5	0.8	1.7	1.1
<b>Cretaceous</b>	3.3	3.4	3.3	-1.8	-1.3	-1.5	382050	397430	394041	1779	2426	2159	138	209	177.6	20	54	33.8	5.5	28.9	10.9	1.3	7.2	2.1
<b>Dokos</b>	3.0	3.3	3.2	-3.0	-1.2	-1.9	388540	395520	390775	1847	2754	2435	149	230	193	28	109	47.7	7.1	90.4	20.8	1	2.4	1.9
	3.1	3.4	3.2	-3.2	-1.2	-2.4	386190	396310	391763	1793	2508	2091	130	225	184.6	21	151	82.2	8.7	150	84.4	0.7	3.2	2.2

Table 2. Summary of isotope and trace elemental data.