

Optimization-based Methodology for the Development of Wastewater Facilities for Energy and Nutrient Recovery

C. Puchongkawarin^{a,b}, C. Gomez-Mont^c, D.C. Stuckey^a, B. Chachuat^{a,b,*}

^aDepartment of Chemical Engineering, Imperial College London, SW7 2AZ, UK

^bCentre for Process System Engineering, Imperial College London, London SW7 2AZ, UK

^cEnergy Futures Lab, Imperial College London, London SW7 2AZ, UK

Abstract

A paradigm shift is currently underway from an attitude that considers wastewater streams as a waste to be treated, to a proactive interest in recovering materials and energy from these streams. This paper is concerned with the development and application of a systematic, model-based methodology for the development of wastewater resource recovery systems that are both economically attractive and sustainable. With the array of available treatment and recovery options growing steadily, a superstructure modelling approach based on rigorous mathematical optimization appears to be a natural approach for tackling these problems. The development of reliable, yet simple, performance and cost models is a key issue with this approach in order to allow for a reliable solution based on global optimization. We argue that commercial wastewater simulators can be used to derive such models, and we illustrate this approach with a simple resource recovery system. The results show that the proposed methodology is computationally tractable, thereby supporting its application as a decision support system for selection of promising resource recovery systems whose development is worth pursuing.

Keywords: biological treatment, biorefining, energy recovery, nutrient recovery, superstructure optimization, wastewater treatment

1. Introduction

For a long time wastewater has been considered a human health concern and environmental hazard. For the most part, wastewater treatment design retains its

*Corresponding author. *Chemosphere*

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Email addresses: channarong.puchongkawarin10@imperial.ac.uk (C. Puchongkawarin), carlos.gomez-mont10@imperial.ac.uk (C. Gomez-Mont), d.stuckey@imperial.ac.uk (D.C. Stuckey), b.chachuat@imperial.ac.uk (B. Chachuat)

13 foundations in engineering traditions established back in the early 20th century
14 ([Daigger, 2009](#)). To produce an effluent that was of satisfactory quality for dis-
15 charge into the environment, processes were developed which used large amounts
16 of energy and land, and produced large amounts of sludge. However, a paradigm
17 shift is underway towards making wastewater treatment facilities more sustainable.
18 In this new paradigm, wastewater is regarded as a renewable resource from which
19 water, materials and energy can be recovered, thereby transitioning to *resource re-*
20 *covery facilities* ([Guest et al., 2009](#)). It has even been argued that the design of
21 wastewater facilities could have a significant impact on greenhouse gas emissions
22 ([Bufe, 2008](#)).

23 Until recently a majority of the activities related to resource recovery from
24 wastewater have focused on waste sludge streams, which are a by-product of bio-
25 logical treatment. Because these streams have relatively low flows in comparison
26 to the main wastewater stream, and are more concentrated, resources can be recov-
27 ered from them with minimal changes to the wastewater treatment infrastructure.
28 For instance, mesophilic anaerobic digestion of primary and waste-activated sludge
29 produces a methane-rich gas which is being used in most treatment facilities world-
30 wide to recover energy. It has been reported that a quarter to half of the energy re-
31 quirements for an activated sludge facility can be provided by such energy recovery
32 systems ([Crawford & Sandino, 2010](#); [McCarty et al., 2011](#)), and these figures keep
33 increasing as new practices and technologies are being deployed. A recent review
34 of the current opportunities for resource extraction from sludge streams—both en-
35 ergy and materials—was conducted by [Hydromantis Inc. \(2008\)](#).

36 The main focus of this paper is on direct resource recovery from municipal
37 and industrial wastewater—although it should be stressed that this methodology
38 can also be extended to encompass wastewater sludge as well. A diverse toolkit is
39 available and is increasingly being applied by practitioners to promote sustainabil-
40 ity. The recovery of water as a resource, known as water reclamation and reuse,
41 is an established practice ([Tchobanoglous et al., 2003](#)). Municipal water reuse is

42 practiced widely in water-short locations, to meet agricultural, industrial, and ur-
43 ban irrigation water needs; domestic water reuse, although less practiced, is also
44 increasing (Daigger, 2009). In this work, we are addressing applications where,
45 in combination with the production of water suitable for reuse, other resources are
46 recovered from the wastewater.

47 The increasing market value of wastewater components, such as ammonia
48 and phosphorus, are acting as key drivers for resource recovery from wastew-
49 ater. Because phosphorus is mined as a mineral, and thus is a limited re-
50 source, its commercial value will inevitably increase as it is depleted (Bufe, 2009;
51 Mihelcic et al., 2011). The production of ammonia, on the other hand, has a
52 substantial energy requirement and concomitant Greenhouse Gas (GHG) emis-
53 sion (Galloway & Cowling, 2002), and hence its price is increasing with energy
54 costs. Besides nutrients, organic carbon can be recovered from wastewater using
55 anaerobic digestion to produce a methane-rich biogas (Jeison & van Lier, 2008)
56 that can be combusted on-site for heat or electricity generation, or cleaned-up and
57 sold. Other examples of newer technologies for carbon recovery include microbial
58 fuel cells (Logan et al., 2006) as well as the production of bioplastics (Coats et al.,
59 2007). This potential for water, materials and energy recovery has also attracted a
60 lot of interest from industry (see, e.g., Veolia Environment, 2010).

61 There seems to be a general consensus that wastewater (and wastewater sludge)
62 is a potential source of valuable resources, and that the technology needed for such
63 resource recovery is maturing. Besides the technological and market penetration
64 barriers, the lack of decision-making tools and design methodologies remains a
65 major problem to identifying the most sustainable solutions in a given geographic
66 and cultural context (Guest et al., 2009; Parker, 2009). Such decisions need to be
67 made by considering whether a resource recovery facility will have a net positive
68 impact in terms of a number of economic, environmental and socio-cultural cri-
69 teria. These types of decisions are typically comparative in nature, and can take
70 advantage of life cycle analysis (LCA) or triple bottom line (3BL) techniques to

71 make quantitative assessments.

72 Nevertheless, as the array of technical options grows, a simple enumeration
73 of all possible alternatives quickly becomes unmanageable, let alone the fact that
74 each technology has its own parameters to specify or optimize. In this paper, we
75 advocate the use of systems engineering methods and tools to address this problem
76 in a systematic way. A superstructure modelling approach (Bielger et al., 1997) is
77 considered which can account for a large number of treatment and separation op-
78 tions (units) along with all feasible interconnections between them. Rigorous opti-
79 mization based on such a superstructure leads to mixed-integer nonlinear programs
80 (MINLPs) that can be implemented and solved using state-of-the-art mathematical
81 optimization software such as GAMS (<http://www.gams.com>). However, key
82 to the success of this methodology is the development/selection of mathematical
83 models for the units that are simple enough for the optimization problem to remain
84 tractable, yet provide reliable estimates of their performance and associated costs.

85 In the remainder of this paper we will review the main resource recovery op-
86 tions from wastewater in Sect. 2. We will then describe the superstructure opti-
87 mization approach in Sect. 3, with emphasis on performance models for the units
88 and assessment criteria. This approach is illustrated by the synthesis of a simple
89 resource recovery facility in Sect. 4. Finally, we conclude the paper with a general
90 discussion of the ensuing challenges and opportunities.

91 **2. Resource Recovery from Wastewater**

92 State-of-the-art wastewater treatment facilities are designed to remove sub-
93 stances that are considered as contaminants, rather than as resources. One striking
94 example of this is the characterization of organic carbon compounds in terms of
95 their (bio)chemical oxygen demand (BOD/COD), making it clear that such com-
96 pounds are undesirable. Nitrogenous and phosphorous compounds too are trans-
97 formed or removed due to their potential to trigger eutrophication (both N and P) or
98 their aquatic toxicity (ammonia). Moreover, the removal of carbon-, nitrogen- and

99 phosphorous-bearing compounds not only requires energy and/or chemicals, but it
100 often yields byproducts, typically of limited value and often requiring further pro-
101 cessing. The remainder of this section reviews a number of promising technologies
102 that are being developed for recovery of energy and materials from wastewater; the
103 emphasis is on proven technologies.

104 *2.1. Energy Recovery*

105 The organic compounds present in municipal (and many industrial) wastewa-
106 ters can be converted into a methane-rich biogas via anaerobic digestion. A quick
107 calculation suggests that about 30-60 L/day of methane per capita could be gen-
108 erated from typical municipal wastewater if all the biodegradable organic matter
109 were transformed into biogas, in addition to removing the extra costs for aeration
110 associated with conventional activated sludge processes (Owen, 1982). Moreover,
111 anaerobic digestion has little effect on ammonia or phosphates, and therefore it is
112 a perfect match for resource recovery where nutrients can be separated from the
113 effluent in downstream units.

114 In contrast to biogas generation from high-strength wastewater and wastewa-
115 ter sludge that has been employed for many years, direct anaerobic treatment of
116 low-strength wastewater has not been widely practiced so far, especially in tem-
117 perate climates where wastewater temperature is in the range of 15°C. Innova-
118 tions in reactor design to maintain elevated biomass inventories, such as upflow
119 anaerobic sludge blanket (UASB) and anaerobic membrane bioreactor (AnMBR),
120 have mitigated some of the limitations and extended the range of applications of
121 anaerobic treatment (Liao et al., 2006; Lew et al., 2009). Particularly promising
122 configurations include the submerged anaerobic membrane bioreactor (SAnMBR;
123 Hu & Stuckey, 2006; Lin et al., 2012) and, more recently, the anaerobic fluidized
124 membrane bioreactor (AFMBR; Kim et al., 2011). Research is underway to de-
125 velop improved membranes and reactor designs that reduce membrane fouling and
126 enhance dissolved methane recovery (Stuckey, 2012; Smith et al., 2012).

127 In those urban water systems where heat accumulates in the wastewater, ther-
128 mal energy can be recovered through the use of heat pumps or heat exchangers.
129 Although in the form of low-grade energy due to small temperature differences,
130 this energy can be suitable, e.g., for heating buildings (EPA, 2007). Besides biogas
131 and thermal energy, newer technologies that try to generate electricity or hydrogen
132 from wastewater are also worth mentioning (Logan et al., 2006; Kim & Logan,
133 2011).

134 2.2. Materials Recovery

135 Recent spikes in the price of mineral phosphorus (P) is making P recov-
136 ery an increasingly economically attractive option. In the absence of a precip-
137 itating or fixing agent, a majority of the phosphates (PO_4^{3-})—between 50-80%
138 of the total phosphorous compounds in municipal wastewater are phosphates
139 (Tchobanoglous et al., 2003)—will leave with the ‘treated’ effluent at a wastew-
140 ater facility, from where they can be recovered. Nitrogenous (N) compounds, on
141 the other hand, can be generated from atmospheric nitrogen and are not a limited
142 resource, albeit their production is an energy-intensive process. With increasing
143 energy costs, N recovery from wastewater effluents—between 50-80% of total ni-
144 trogen present in municipal wastewater is soluble and a majority is released in
145 inorganic form (Tchobanoglous et al., 2003)—is also becoming economically vi-
146 able.

147 Either ammonia, nitrate or phosphates can be recovered by passing the sec-
148 ondary effluent through adsorbent ion exchange columns (Liberti et al., 2001;
149 Johir et al., 2011). The columns need to be regenerated cyclically by desorbing
150 both N and P, usually with a low volume, concentrated brine solution. The resulting
151 N-and P-enriched solutions can then be mixed and further processed into a saleable
152 product (e.g. fertilizer, see below). Ion exchange presents the additional advantage
153 that N and P recovery can be achieved over a wide range of temperatures that are
154 challenging for biological removal processes. Nonetheless, some of the challenges
155 associated with the use of ion exchange include the potential for fouling with sus-

156 pended solids, the limited exchange capacity of certain adsorbents which require
157 regeneration every few hours, the limited selectivity due to ion competition for the
158 resin sites, and the large capital cost (Miladinovic & Weatherley, 2008). The use of
159 natural zeolites (e.g. clinoptilolite) to adsorb ammonium ions (NH_4^+) from wastew-
160 ater has received considerable attention (Aiyuk et al., 2004; Wang & Peng, 2010),
161 and more recently polymeric resins that have higher exchange capacities than zeo-
162 lites. Research is also underway to develop anion adsorbents with high phosphate
163 selectively and easy regeneration characteristics, including hydrotalcite (HTAL)
164 (Kuzawa et al., 2006) and polymeric resins with hydrated ferric oxide nanoparti-
165 cles (Martin et al., 2009).

166 An attractive alternative to ion exchange for P recovery is reactive filtra-
167 tion, which combines physical filtration of particulate P compounds with co-
168 precipitation and adsorption of soluble P compounds onto coated sand in a moving
169 bed filter. For instance, continuous backwash filters made of hydrous ferric oxide
170 (HFO) coated sand, where continuous regeneration is accomplished by adding 5-10
171 mg/L of ferric chloride (FeCl_3), have been reported to achieve up to 95% P removal
172 from secondary municipal effluents (Newcombe et al., 2008). In these processes,
173 P recovery can be simply achieved with a membrane separator receiving the back-
174 wash from the reactive filter (Sutton et al., 2011), although further processing is
175 then needed to convert P into a saleable product.

176 Another technology that has gained substantial interest to help recover phos-
177 phorus from concentrated wastewater streams is crystallization into reusable
178 compounds such as calcium phosphate ($\text{Ca}_3(\text{PO}_4)_2$) and struvite (MgNH_4PO_4)
179 (Le Corre et al., 2009). Although both can be used in agriculture as fertilizers,
180 struvite is preferred due to the fact that magnesium (Mg), N and P are released
181 simultaneously (1:1:1 molar ratio), and that the rate of nutrient release is slower
182 compared to other fertilizers. The recovery technology involves precipitation in
183 either stirred tank or fluidized bed reactors (Bhuiyan et al., 2008), the latter being
184 the most commonly used to crystallize struvite from wastewater. Most applica-

185 tions to date have used sludge liquors generated from anaerobic digesters as their
186 influent streams, with phosphate concentrations in the range 50-100 mg/L; removal
187 efficiencies of 80% or higher have been reported under well-controlled conditions
188 (Ueno & Fujii, 2001). For more dilute streams, such as secondary effluents with
189 phosphate concentrations in the range 4-12 mg/L, struvite crystallization can be
190 combined with adsorbent columns and fed with the enriched solutions from the ad-
191 sorbent regeneration; the RIM-NUT[®] process by Liberti et al. (2001) applies this
192 strategy.

193 Besides nutrients, organic carbon (OC) recovery can be achieved via the pro-
194 duction of polyhydroalkanoates (PHAs), a precursor of bioplastics (Coats et al.,
195 2007). PHAs are linear polyesters produced from the fermentation of sugars or
196 lipids by bacteria as energy storage (up to 50% weight). Their production is techno-
197 logically proven from synthetic wastewater and under well-controlled conditions.
198 However, full-scale applications are still at the embryonic stage, in part due to the
199 challenge associated with the separation of PHAs from the bacterial biomass. It
200 should also be clear that PHA production, when used in combination with anaer-
201 obic digestion, will result in a net reduction of biogas production by the digester
202 since it diverts part of the available OC.

203 Other materials recovery concern heavy metals, for which many separation
204 technologies have been proposed including, but not limited to, ion exchange, ad-
205 sorption, membrane filtration, and chemical precipitation (Fu & Wang, 2011).

206 **3. Development of Resource Recovery Facilities**

207 The combination of traditional wastewater treatment technology with energy
208 and materials recovery solutions offers considerable promise to improve the sus-
209 tainability and reduce the cost of wastewater facilities. Clearly, the ultimate goal
210 here is a closed-cycle, energy-sufficient process, where all waste streams are recy-
211 cled and the only output is saleable/valuable products.

212 Recently, Sutton et al. (2011) developed a wastewater process concept that in-

213 tegrates biological processing with physical and chemical separation units for the
214 conversion of OC to methane, the recovery of P and N, and the production of water
215 suitable for reuse. More generally, this raises the question as to how to select and
216 interconnect, from a wide variety of unit operations, those units which will lead
217 to the most sustainable wastewater treatment systems—this problem is known as
218 *synthesis* or *flowsheeting* in process engineering.

219 For the synthesis of sustainable resource recovery facilities from wastewater,
220 one must account for the trade-offs between capital and operating costs and sales
221 on the one hand, and water quality and other environmental considerations on the
222 other hand. While technical insights can significantly reduce the combinatorial
223 problem and often allow us to arrive at promising solutions, as in [Sutton et al.](#)
224 [\(2011\)](#), they do not always provide all the information that is required for an opti-
225 mal (or near optimal) system. In general, one may still be left with the problem of
226 having to search among a relatively large number of alternatives, thereby calling
227 for systematic optimization based on a superstructure representation.

228 *3.1. Superstructure Representation*

229 Systematic methods for the synthesis of complex chemical plants and biore-
230 fineries, based on superstructure modeling and optimization, are well developed
231 ([Bielger et al., 1997](#); [Kokossis & Yang, 2010](#); [Liu et al., 2011](#)). These approaches
232 are also increasingly applied to water network synthesis in process plants in order
233 to minimize fresh water consumption and wastewater generation through regener-
234 ation and recycle/reuse ([Faria & Bagajewicz, 2009](#); [Khor et al., 2012](#)). In contrast,
235 as far as municipal wastewater facilities are concerned, only limited work has been
236 reported to date ([Rigopoulos & Linke, 2002](#); [Alasino et al., 2007](#)), although the
237 need for systematic approaches was emphasized, e.g., in [Balkema et al. \(2001\)](#) and
238 [Hamouda et al. \(2009\)](#).

239 The synthesis problem statement starts with the specification of the following
240 data:

- 241 • a set of wastewater streams of given flow rates and compositions;

- 242 • a set of water sinks with known maximum concentration limits from local
243 regulations;
- 244 • a set of treatment and separation units with given performance for target
245 compounds.

246 These specifications can be represented by a generic superstructure, which consid-
247 ers every possible interconnection in a fixed network topology. One such super-
248 structure is illustrated in Fig. 1, for a simple network topology that consists of a
249 single wastewater stream, a single water sink, and six treatment/separation units.

250 [Figure 1 about here.]

251 The objective of the synthesis problem is to determine an optimal resource
252 recovery facility in terms of (i) its units, (ii) the piping interconnections between
253 the units, and (iii) the flowrates and compositions in the interconnections.

- 254 • The optimal system configuration is one that maximizes a certain sus-
255 tainability index of the facility, for instance a weighted sum of LCA im-
256 pacts (Corominas et al., 2013). Alternatively, one can choose to maximize
257 the economic efficiency of a facility, e.g. using life-cycle costing (LCC;
258 Rebitzer et al., 2003) or simply in terms of its net present value (NPV) over
259 the project lifetime:

$$\text{NPV} = -\text{CAPEX} + \sum_{\text{yr}=1}^{\text{lifetime}} \frac{\text{SALES} - \text{OPEX}}{(1 + \text{DISCOUNT_RATE})^{\text{yr}}}, \quad (1)$$

260 where SALES stands for revenues from energy/product sales.

- 261 • Regarding constraints, material balances on flows (F) and concentrations
262 (X) around the sources, the units, and the sinks are to be obeyed in addition
263 to the discharge limits and certain design and structural specifications. For
264 instance, the material balances around a given treatment unit k for a species
265 c can be written as:

$$\begin{aligned}
F_k^{\text{in}} &= \sum_{i=1}^{N_{\text{source}}} F_{i \rightarrow k} + \sum_{k'=1}^{N_{\text{unit}}} F_{k' \rightarrow k} \\
X_{k,c}^{\text{in}} F_k^{\text{in}} &= \sum_{i=1}^{N_{\text{source}}} F_{i \rightarrow k} X_{i,c} + \sum_{k'=1}^{N_{\text{unit}}} F_{k' \rightarrow k} X_{k',c}^{\text{out}} \\
X_{k,c}^{\text{in}} F_k^{\text{in}} (1 - \rho_{k,c}) &= \sum_{k'=1}^{N_{\text{unit}}} F_{k \rightarrow k'} X_{k,c}^{\text{out}} + \sum_{j=1}^{N_{\text{sink}}} F_{k \rightarrow j} X_{k,c}^{\text{out}} ,
\end{aligned}$$

266 where the superscripts ⁱⁿ and ^{out} refer to flows/concentrations entering or leaving
267 the unit, respectively; and ρ stands for the removal efficiency in the unit.
268 In turn, discharge limits (X^{max}) can be enforced on a given sink j for a species
269 c as:

$$\sum_{i=1}^{N_{\text{source}}} F_{i \rightarrow j} X_{i,c} + \sum_{k=1}^{N_{\text{unit}}} F_{k \rightarrow j} X_{k,c}^{\text{out}} \leq X_{j,c}^{\text{max}} .$$

270 Key to the accuracy of such a superstructure representation are the performance and
271 costing models used to populate the objective and constraint functions, as discussed
272 next.

273 3.2. Performance Models and Costing of the Units

274 The aim of a wastewater/resource recovery facility is to remove and recover
275 contaminants from the effluent and generate energy, and it is the role of the math-
276 ematical models of each treatment/separation unit to predict their performance.
277 To comply with the superstructure optimization approach and the current capa-
278 bilities of optimization technology, however, these models must remain as simple
279 as possible, typically linear, piecewise-linear or polynomial relationships between
280 the input/output variables. Needless to say, direct use of complex biodegradation
281 models such as ADM1 (Batstone et al., 2002) and ASM1-3 (Henze et al., 2007)
282 for the bioreactors or complex crystallization, adsorption and filtration models for
283 the separation units in the superstructure for performance prediction is currently
284 intractable from a computational standpoint.

285 [Figure 2 about here.]

286 In this context, one simple approach consists in assuming fixed conversion,
287 removal or split ratios in the treatment and separation units. This is the approach
288 typically used in water network synthesis problems (see, e.g., [Khor et al., 2012](#)).
289 In this work, we advocate an alternative, potentially more accurate, approach that
290 relies on detailed first principle models, as implemented in wastewater treatment
291 simulators such as GPS-XTM ([Hydromantis Inc., 2011b](#)). Based on the average
292 performance predicted by a simulator—either at steady state or averaged over a
293 cyclic steady state—for various influent compositions (COD, TSS, *etc*) and given
294 operation parameters (HRT, SRT, *etc*), simple regression models can be fitted to the
295 simulated data points. Fig. 2 shows an example of methane conversion efficiency
296 in a UASB reactor as predicted by the ManTIS2 model in GPS-XTM for various
297 HRTs, along with a corresponding piecewise-linear regression model for use in the
298 superstructure model.

299 Similarly, reliable costing information—both capital and operating costs—
300 can be obtained from preliminary costing software, such as CAPDETWORKSTM
301 ([Hydromantis Inc., 2011a](#)). Here again, data can be generated by such programs
302 for various unit volumes and/or flowrates and regressed to yield simple linear,
303 piecewise-linear or polynomial models. Using a common source and methodol-
304 ogy for costing various technologies is clearly advantageous with regards to con-
305 sistency.

306 3.3. Numerical Solution Strategy

307 In optimizing a superstructure for synthesizing a resource recovery system, the
308 optimization model will contain two types of decisions:

- 309 • the *discrete*, usually *binary*, decisions on the units that should be included in
310 the system along with their interconnections, here denoted by y ; and,
- 311 • the *continuous* decisions that define the flows and composition as well as
312 certain design and operating parameters, here denoted by x .

313 Such an optimization can be posed as the following mathematical programming
314 problem:

$$\begin{aligned} \min_{x,y} \quad & f(x, y) && \text{(P)} \\ \text{s.t.} \quad & h(x) = 0 \\ & g(x, y) \leq 0 \\ & x \geq 0, y \in \{0, 1\}. \end{aligned}$$

315 The objective function $f(x, y)$ in (P), which either expresses a sustainability or eco-
316 nomic index (see Sect. 3.1), is a function of both types of variables. The continuous
317 variables x , which for physical reasons are assumed to be non-negative, must obey
318 material balance equations of the form $h(x) = 0$ (see Sect. 3.2), where usually
319 $\dim(h) < \dim(x)$. Both continuous and binary variables must also satisfy the de-
320 sign specifications in terms of discharge allowance and physical operating limits,
321 as well as logical constraints such as the existence of piping interconnections for
322 nonzero flows or to enforce the sequencing of certain units. A detailed formulation
323 of the optimization model can be found in [Gomes-Mont \(2011\)](#).

324 The superstructure optimization model (P) yields a nonconvex MINLP prob-
325 lem due to the presence of bilinear terms that arise in the material balances
326 of the units as a result of contaminant mixing, in addition to other nonlinear-
327 ities in the performance and costing expressions. Such nonconvexity can result
328 in multiple local optimal solutions, thereby calling for the implementation of
329 global optimization techniques to guarantee a reliable solution. Fortunately, re-
330 cent work in water network synthesis ([Ahmetović & Grossmann, 2011](#); [Khor et al.,](#)
331 [2012](#)) demonstrates that deterministic global optimization solvers such as BARON
332 ([Tawarmalani & Sahinidis, 2004](#)) are now able to solve such problems with reason-
333 able run times to global optimality. Moreover, the situation is improving as new
334 solvers such as GloMIQO ([Misener & Floudas, 2013](#)) are becoming available.

335 3.4. Superstructure-based Optimization Methodology

336 Superstructure modelling and optimization is at the core of the synthesis of
337 sustainable resource recovery facilities. In an overall design methodology (Fig. 3,
338 the most promising alternatives would in turn be validated against the performance
339 and cost predicted by the wastewater treatment simulator. Typically, this would cre-
340 ate an iteration between the superstructure optimization and the simulator in order
341 to refine the regression models as appropriate. In particular, recent developments
342 in surrogate-based optimization can guarantee to the overall optimum with mini-
343 mum recourse to detailed models (Agarwal & Biegler, 2013; Biegler et al., 2014).
344 Finally, the selected process candidates would be considered for detailed perfor-
345 mance and cost analysis, including integration options and operability issues. Here
346 again, further iterations with the superstructure optimization block could prove
347 necessary in order to account for additional design and operational constraints.

348 [Figure 3 about here.]

349 The focus in this paper is more specifically on the components in the grey-
350 shaded area of Fig. 3. The main objective of the following case study is to provide
351 a proof-of-concept of this superstructure optimization approach based on simple
352 regression models, also showing that the underlying optimization problems are
353 indeed tractable to guaranteed global optimality.

354 4. Case Study

355 We consider the synthesis of a resource recovery facility for the treatment of
356 $100 \text{ m}^3/\text{h}$ of an industrial wine distillery effluent, whose average composition for
357 the main species is given in Table 1 (Melamane et al., 2007). The objective is to
358 maximize the NPV given in Eq. (1), while satisfying maximum discharge require-
359 ments as defined by the EU Directive 91/271/EEC on Urban Wastewater Treat-
360 ment; these requirements consist of meeting either minimum abatements of 75%
361 total COD, 80% total N and total P, and 90% TSS, or maximum concentrations of

362 10 g/L total COD, 0.4 g/L total N, 0.07 g/L total P, and 0.5 g/L TSS—but not
363 necessarily both.

364 [Table 1 about here.]

365 A small superstructure that consists of 2 biological treatment units (UASB,
366 SMBR), 2 filtration units (sand filter, membrane unit), and 2 nutrient recovery
367 units (struvite crystallizer, zeolite adsorber) is investigated. This problem is kept
368 relatively simple as the main objective of the case study is to assess the proposed
369 optimization methodology. More challenging problems that include a variety of
370 treatment/recovery options for carbon, nitrogen and phosphorus are beyond the
371 scope of this paper and will be investigated as part of future work.

372 • The performance of the UASB and the SMBR are approximated based on
373 the ManTIS2 and ASM1 models in GPS-XTM, respectively. The degradation
374 rates of total COD, total N, ammonia, total P, phosphate, and TSS, as well as
375 the production rate of methane in a UASB, are regressed using either linear
376 or piecewise-linear models within an HRT range of 2-20 day⁻¹. A subset of
377 such performance models are shown in Figure 4 (left and centre plots) for a
378 UASB unit.

379 [Figure 4 about here.]

380 • The performance of the filtration, membrane and struvite units is also pre-
381 dicted by using simple models in GPS-XTM, and then averaged to yield con-
382 stant removal/conversion efficiencies as a first approximation. The zeolite
383 (ion exchange) unit, currently unavailable in GPS-XTM, is also approximated
384 using constant efficiencies gathered from the literature (Gomes-Mont, 2011).

385
386 • The CAPEX and OPEX of all the units, with the exception of the membrane
387 unit, are estimated using CAPDETWORKSTM, and then regressed with simple
388 linear models as a function of the unit volume and/or processed flow rate

389 in the range of operation. This is illustrated in Figure 4 (right plots) for
390 the CAPEX and OPEX of a UASB unit. For the membrane unit, currently
391 unavailable in CAPDETWORKSTM, rough estimates of the CAPEX and OPEX
392 were used as recommended by membrane experts (Judd, 2012).

- 393 • Besides the treatment/separation units in the superstructure, an auxiliary
394 piece of equipment is the electricity generator from biogas. This technol-
395 ogy is well developed, with companies such as Alstom, Capston or General
396 Electric providing lines of engines specially adapted for biogas from anaer-
397 obic digestion. An average conversion efficiency of 40% is assumed here
398 for the generator, and the OPEX and CAPEX are estimated based on data
399 published by the United Nations Framework Convention on Climate Change
400 (UNFCCC) and the International Energy Agency (IEA); see Gomes-Mont
401 (2011) for details.

402 [Figure 5 about here.]

403 A constraint regarding the interconnections in the superstructure is that the
404 effluent from the UASB must pass either through the sand filter or through the
405 membrane unit before it can be sent to the zeolite and/or struvite units for nutri-
406 ent recovery; this is to prevent large concentrations of solids in the latter. The
407 computed optimal structure is displayed in the top part of Figure 5—Flowsheet
408 A. The wastewater stream is first split in such a way that about 60% is processed
409 in the UASB, before passing through the sand filter together with the remaining
410 40% of the raw wastewater stream. The sand filter outlet is then passed succes-
411 sively through the ion exchange column and the struvite crystallizer for nutrient
412 recovery. While the effluent satisfies all of the discharge requirements, it is the
413 minimum abatement of 80% in total P which turns out to be the most restrictive
414 here. Moreover, a majority of the residual COD concentration is comprised of
415 non-biodegradable soluble compounds, the other part being biodegradable soluble
416 compounds from the wastewater fraction not treated in the UASB.

417 An NPV of about -7.68 M\$ is found for this flowsheet over a period of 20
418 years, and the breakdown of these costs in terms of CAPEX, OPEX and SALES is
419 given in Table 2. This negative NPV is mainly due to the rather large CAPEX—
420 more than 40% incurred by the UASB unit, which cannot be offset by the net annual
421 profit of about 0.88 M\$, although the sales revenue, mainly electrical power from
422 biogas combustion, exceeds the OPEX. Therefore, while such a resource recov-
423 ery facility may not completely offset the infrastructure and operating costs solely
424 based on electricity and nutrient sales, it would greatly mitigate the cost of wastew-
425 ater treatment to comply with the discharge regulations.

426 [Table 2 about here.]

427 The decision to split the wastewater stream and not to process around 40% of
428 that stream in the UASB unit may first appear counter-intuitive given the fact that
429 most of the sales revenue is from the biogas produced in the UASB. However, in
430 terms of the NPV, producing more biogas is only justified when the added sales
431 revenue can offset the extra CAPEX and OPEX for a bigger UASB unit; this con-
432 dition is not met in the present case study, and it would be worth investigating the
433 sensitivity of this trade-off further to validate this finding.

434 To confirm this trend, we considered removing the possibility of bypassing
435 the UASB by sending part of the wastewater stream directly to the filtration units.
436 The resulting superstructure is depicted in the middle part of Figure 5—Flowsheet
437 B. In this second flowsheet, a small part of the wastewater stream is now mixed
438 directly with the treated effluent stream, without being processed. As expected
439 the estimated NPV value of Flowsheet B decreases compared to Flowsheet A—see
440 Table 2 for a break-down. In particular, this is a rather large decrease of about 3 M\$.
441 Finally, preventing part of the wastewater stream to be directly mixed with the
442 treated effluent leads to the superstructure depicted in the bottom part of Figure 5—
443 Flowsheet C. The estimated NPV of this third flowsheet turns out to be comparable
444 with the estimated NPV of Flowsheet B.

445 Future investigations around this case study will aim at extending the set of
446 treatment/recovery options in the superstructure, including anammox for ammonia
447 removal or adsorptive phosphorous removal. The possibility of having multiple
448 treatment stages, e.g., for the sand filtration units, will also be investigated in or-
449 der to yield more efficient wastewater facilities, as well as for coping with more
450 stringent effluent regulations.

451 **5. Conclusions and Future Directions**

452 In this paper, a systematic optimization-based methodology for the synthesis
453 of wastewater/resource recovery facilities has been discussed and illustrated with
454 a case study. By and large, this methodology should be regarded as a decision
455 support system for isolating, among hundreds or even thousands of alternatives,
456 those promising resource recovery systems whose development is worth pursuing.
457 Based on this preselection, further simulation and optimization studies can then be
458 undertaken to refine the performance and cost prediction by taking into account
459 detailed design and operation considerations, as well as process integration. Such
460 decomposition is indeed warranted as current computational capabilities and avail-
461 able algorithms do not allow for the optimal design and operation to be solved in
462 a single step due to complex unit configuration, multiple scales, time dependence,
463 and uncertainty.

464 A major hurdle in applying this methodology appears to be the availability
465 of reliable performance models for the treatment and separation units as well as
466 reliable (capital and operation) costing data. Here, we advocate the use of state-of-
467 the-art wastewater treatment simulators in order to derive simple response-surface
468 models, which are general enough to be independent of detailed design choices
469 and keep the superstructure optimization model computationally tractable—this
470 approach was demonstrated on a simple case study. Naturally, such simple models
471 carry significant uncertainty and usually only provide a rough approximation of the
472 actual performance of such complex units. A way to refine these models involves

473 performing an iteration between the detailed process simulators and the superstruc-
474 ture optimization problem. Moreover, for those treatment/recovery techniques that
475 are less well established or lack reliable performance models, a scenario-based
476 analysis can be applied, whereby multiple sets of resource recovery systems are de-
477 termined on account of the forecast performance and cost scenarios. In particular,
478 this analysis can be useful for resource allocation, for instance to help determine
479 which technologies are critical and focus further research and development effort.

480 As for future research directions, a key extension will be to integrate wastew-
481 ater and biosolids management within the same recovery system. Another impor-
482 tant research activity should be the development and regular update of information
483 databases as new advanced treatment and recovery technologies develop, or as the
484 economic, environmental and socio-cultural contexts evolve. Besides the avail-
485 ability of feasible technologies that can transform wastewater into a product, and
486 the downstream processing of this product into a saleable item, the circumstances
487 that are required to successfully establish a functioning and sustainable resource
488 recovery system also involves developing a distribution infrastructure and catching
489 investors' interest in developing such technologies.

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493 **References**

- 494 Agarwal, A., & Biegler, L. T. (2013). A trust-region framework for constrained
495 optimization using reduced order modeling. *Optimization & Engineering, 14*,
496 3–35.
- 497 Ahmetović, E., & Grossmann, I. E. (2011). Global superstructure optimization for
498 the design of integrated process water networks. *AIChE journal, 57*, 434–457.

- 499 Aiyuk, S., Amoaka, J., Raskin, L., van Haandel, A., & Verstraete, W. (2004). Re-
500 moval of carbon and nutrients from domestic wastewater using a low investment,
501 integrated treatment concept. *Water Research*, (pp. 3031–3042).
- 502 Alasino, N., Mussati, M. C., & Scenna, N. (2007). Wastewater treatment plant
503 synthesis and design. *Industrial & Engineering Chemistry Research*, *46*, 7497–
504 7512.
- 505 Balkema, A. J., Preisig, H. A., Otterpohl, R., Lambert, A. J., & Weijers, S. R.
506 (2001). Developing a model-based decision support tool for the identification of
507 sustainable treatment options for domestic wastewater. *Water Science & Tech-*
508 *nology*, *43*, 265–269.
- 509 Batstone, D., Keller, J., Angelidaki, I., Kalyuzhnyi, S., Pavlostathis, S., Rozzi, A.,
510 Sanders, W., Siegrist, H., & Vavilin, V. (2002). *Anaerobic Digestion Model No.*
511 *1 (ADM1)*. IWA Publishing, London.
- 512 Bhuiyan, M. I. H., Mavinic, D. S., & Koch, F. A. (2008). Phosphorus recovery from
513 wastewater through struvite formation in fluidized bed reactors: A sustainable
514 approach. *Water Science & Technology*, *57*, 175–181.
- 515 Biegler, L. T., Lang, Y., & Lin, W. (2014). Multi-scale optimization for process
516 systems engineering. *Computers & Chemical Engineering*, *60*, 17–30.
- 517 Bieler, L., Grossmann, I., & Westerberg, A. (1997). *Systematic Methods of Chem-*
518 *ical Process Design*. Prentice Hall.
- 519 Bufe, M. (2008). Getting warm? Climate change concerns prompt utilities to
520 rethink water resources, energy use. *Water Environment & Technology*, *20*, 29–
521 32.
- 522 Bufe, M. (2009). Phosphorous sources peaked out? *Water Environment & Tech-*
523 *nology*, *21*, 18–23.

- 524 Coats, E. R., Loge, F. J., Wolcott, M. P., Englund, K., & McDonald, A. G. (2007).
525 Synthesis of polyhydroxyalkanoates in municipal wastewater treatment. *Water*
526 *Environment Research*, 79, 2396–2403.
- 527 Corominas, L., Foley, F., Guest, J. S., Hospido, A., Larsen, H. F., Morera, S., &
528 Shaw, A. (2013). Life cycle assessment applied to wastewater treatment: State
529 of the art. *Water Research*, 47, 5480–5492.
- 530 Crawford, G., & Sandino, J. (2010). Energy efficiency in wastewater treatment
531 in North America: A compendium of best practices and case studies of novel
532 approaches. Report #OWSO4R07e, Water Environment Research Foundation
533 and IWA Publishing.
- 534 Daigger, G. T. (2009). Evolving urban water and residuals management paradigms:
535 water reclamation and reuse, decentralization, and resource recovery. *Water*
536 *Environment Research*, 81, 809–823.
- 537 EPA (2007). Opportunities for and benefits of combined heat and power at wastew-
538 ater treatment facilities. Report #EPA-430-R-07-003, U.S. Environmental Pro-
539 tection Agency.
- 540 Faria, D. C., & Bagajewicz, M. J. (2009). Profit-based grassroots design and retrofit
541 of water networks in process plants. *Computers & Chemical Engineering*, 33,
542 436–453.
- 543 Fu, F., & Wang, Q. (2011). Removal of heavy metal ions from wastewaters: A
544 review. *Journal of Environmental Management*, 92, 407–418.
- 545 Galloway, J. N., & Cowling, E. B. (2002). Reactive nitrogen and the world: 200
546 years of change. *Ambio*, 31, 64–71.
- 547 Gomes-Mont, C. (2011). *Modeling and Optimisation of a Biorefinery for Energy*
548 *and Nutrient Recovery*. Master's thesis Imperial College London, Energy Fu-
549 tures Laboratory.

550 Guest, J. S., Skerlos, S. J., Barnard, J. L., Beck, M. B., Daigger, G. T., Hilger, H.,
551 Jackson, S. J., Karvazy, K., Kelly, L., MacPherson, L., Mihelcic, J. R., Pramanik,
552 A., Raskin, L., Van Loosdrecht, M. C., Yeh, D., & Love, N. G. (2009). A new
553 planning and design paradigm to achieve sustainable resource recovery from
554 wastewater. *Environmental Science & Technology*, *43*, 6126–6130.

555 Hamouda, M. A., Anderson, W. B., & Huck, P. M. (2009). Decision support sys-
556 tems in water and wastewater treatment process selection and design: a review.
557 *Water Science & Technology*, *60*, 1757–1770.

558 Henze, M., Gujer, W., & Mino, T. (2007). *Activated Sludge Models ASM1, ASM2,*
559 *ASM2D and ASM3*. IWA Publishing, London.

560 Hu, A. Y., & Stuckey, D. C. (2006). Treatment of dilute wastewaters using a
561 novel submerged anaerobic membrane bioreactor. *Journal of Environmental*
562 *Engineering*, *132*, 190–198.

563 Hydromantis Inc. (2008). State of science report: Energy and resource recovery
564 from sludge. Report prepared for the Global Water Research Coalition.

565 Hydromantis Inc. (2011a). CapdetWorks (ver. 2.5). [http://www.hydromantis.](http://www.hydromantis.com/CapdetWorks.html)
566 [com/CapdetWorks.html](http://www.hydromantis.com/CapdetWorks.html).

567 Hydromantis Inc. (2011b). GPS-X (ver. 6.1). [http://www.hydromantis.com/](http://www.hydromantis.com/GPS-X.html)
568 [GPS-X.html](http://www.hydromantis.com/GPS-X.html).

569 Jeison, J., & van Lier, J. B. (2008). Anaerobic wastewater treatment and membrane
570 filtration: A one night stand or a sustainable relationship? *Water Science &*
571 *Technology*, *57*, 527–532.

572 Johir, M. A. H., George, J., Vignswaran, S., Kandasamy, J., & Grasmick, A. (2011).
573 Removal and recovery of nutrients by ion exchange from high rate membrane
574 bio-reactor (mbr) effluent. *Desalination*, *275*, 197–202.

575 Judd, S. J. (2012). Personal communication.

- 576 Khor, C. S., Chachuat, B., & Shah, N. (2012). A superstructure optimization ap-
577 proach for water network synthesis with membrane separation-based regenera-
578 tors. *Computers & Chemical Engineering*, *42*, 48–63.
- 579 Kim, J., Kim, K., Ye, H., Lee, E., Shin, C., McCarty, P. L., & Bae, J. (2011).
580 Anaerobic fluidized bed membrane bioreactor for wastewater treatment. *Envi-
581 ronmental Science & Technology*, *45*, 576–581.
- 582 Kim, Y., & Logan, B. E. (2011). Hydrogen production from inexhaustible supplies
583 of fresh and salt water using microbial reverse-electrodialysis electrolysis cells.
584 *Proceedings of the National Academy of Science of the USA*, *108*, 16176–16181.
- 585 Kokossis, A. C., & Yang, A. (2010). On the use of systems technologies and
586 a systematic approach for the synthesis and the design of future biorefineries.
587 *Computers & Chemical Engineering*, *34*, 1397–1405.
- 588 Kuzawa, K., Jung, Y. J., Kiso, Y., Yamada, M., T. Nagai, & Lee, T. G. (2006).
589 Phosphate removal and recovery with a synthetic hydrotalcite as an adsorbent.
590 *Chemosphere*, *62*, 45–52.
- 591 Le Corre, K. S., Valsami-Jones, E., Hobbs, P., & Parsons, S. A. (2009). Phosphorus
592 recovery from wastewater by struvite crystallization: A review. *Critical Reviews
593 in Environmental Science & Technology*, *39*, 433–477.
- 594 Lew, B., Tarre, S., Beliavski, M., Dosoretz, C., & Green, M. (2009). Anaerobic
595 membrane bioreactor (AnMBR) for domestic wastewater treatment. *Desalina-
596 tion*, *243*, 251–257.
- 597 Liao, B. Q., Kraemer, J. T., & Bagley, D. M. (2006). Anaerobic membrane biore-
598 actors: Applications and research directions. *Critical Reviews in Environmental
599 Science & Technology*, *36*, 489–530.
- 600 Liberti, L., Petruzzelli, D., & De Florio, L. (2001). REM NUT ion exchange plus
601 struvite precipitation process. *Environmental Technology*, *22*, 1313–1324.

- 602 Lin, H., Chen, J., Wang, F., Ding, L., & Hong, H. (2012). Feasibility evaluation of
603 submerged anaerobic membrane bioreactor for municipal secondary wastewater
604 treatment. *Desalination*, 281, 120–126.
- 605 Liu, P., Georgiadis, M. C., & Pistikopoulos, E. N. (2011). Advances in energy
606 systems engineering. *Industrial & Engineering Chemistry Research*, 50, 4915–
607 4926.
- 608 Logan, B. E., Hamelers, B., Rozendal, R., Schröder, U., Keller, J., Freguia, S., Ael-
609 terman, P., Verstraete, W., & Rabaey, K. (2006). Microbial fuel cells: Method-
610 ology and technology. *Environmental Science & Technology*, 40, 5181–5192.
- 611 Martin, B. D., Parsons, S. A., & Jefferson, B. (2009). Removal and recovery
612 of phosphate from municipal wastewaters using a polymeric anion exchanger
613 bound with hydrated ferric oxide nanoparticles. *Water Science & Technology*,
614 60, 2637–2645.
- 615 McCarty, P. L., Bae, J., & Kim, J. (2011). Domestic wastewater treatment as a net
616 energy producer—can this be achieved? *Environmental Science & Technology*,
617 45, 7100–7106.
- 618 Melamane, X. L., Strong, P. J., & Burgess, J. E. (2007). Treatment of wine distillery
619 wastewater: A review with emphasis on anaerobic membrane reactors. *South
620 African Journal of Enology & Viticulture*, 28, 25–36.
- 621 Mihelcic, J. R., Fry, L. M., & Shaw, R. (2011). Global potential of phosphorus
622 recovery from human urine and feces. *Chemosphere*, 84, 832–839.
- 623 Miladinovic, N., & Weatherley, L. R. (2008). Intensification of ammonia removal
624 in a combined ion-exchange and nitrification column. *Chemical Engineering
625 Journal*, 135, 15–24.
- 626 Misener, R., & Floudas, C. A. (2013). GloMIQO: Global mixed-integer quadratic
627 optimizer. *Journal of Global Optimization*, (pp. 3–50).

- 628 Newcombe, R. L., Strawn, D. G., Grant, T. M., Childers, S. E., & Möller, G.
629 (2008). Phosphorus removal from municipal wastewater by hydrous ferric oxide
630 reactive filtration and coupled chemically enhanced secondary treatment: Part i
631 – performance. *Water Environment Research*, 80, 238–247.
- 632 Owen, W. F. (1982). *Energy in Wastewater Treatment*. Englewood Cliffs: Prentice-
633 Hall.
- 634 Parker, W. (2009). Opportunities to exploit municipal and industrial wastewater as
635 a resource: A state of the science review. Report prepared for Ontario Centres
636 of Excellence and the Canadian Water Network.
- 637 Rebitzer, G., Hunkeler, D., & Jolliet, O. (2003). The economic pillar of sustain-
638 ability: methodology and application to wastewater treatment. *Environmental*
639 *Progress*, 22, 241–249.
- 640 Rigopoulos, S., & Linke, P. (2002). Systematic development of optimal activated
641 sludge process designs. *Computers & Chemical Engineering*, 26, 585–597.
- 642 Smith, A. L., Stadler, L. B., Love, N. G., Skerlos, S. J., & Raskin, L. (2012). Per-
643 spectives on anaerobic membrane bioreactor treatment of domestic wastewater:
644 A critical review. *Bioresource Technology*, 122, 149–159.
- 645 Stuckey, D. C. (2012). Recent developments in anaerobic membrane reactors.
646 *Bioresource Technology*, 122, 137–148.
- 647 Sutton, P. M., Melcer, H., Schraa, O. J., & Togna, A. P. (2011). Treating municipal
648 wastewater with the goal of resource recovery. *Water Science & Technology*, 63,
649 25–31.
- 650 Tawarmalani, M., & Sahinidis, N. V. (2004). Global optimization of mixed-integer
651 nonlinear programs: A theoretical and practical study. *Mathematical Program-*
652 *ming*, 99, 563–591.

- 653 Tchobanoglous, G., Burton, F. L., & Stensel, H. D. (2003). *Wastewater Engineer-*
654 *ing : Treatment and Reuse*. (4th ed.). New York: McGraw Hill.
- 655 Ueno, Y., & Fujii, M. (2001). Three years experience of operating and selling
656 recovered struvite from full-scale plant. *Environmental Technology*, 22, 1373–
657 1381.
- 658 Veolia Environment (2010). *The wastewater treatment plant of the future*. Scien-
659 tific Chronicles, Issue 17. [http://www.veolia.com/veolia/ressources/
660 documents/1/7772,Chroniques_scientifiques_n17-EN.pdf](http://www.veolia.com/veolia/ressources/documents/1/7772,Chroniques_scientifiques_n17-EN.pdf).
- 661 Wang, S., & Peng, Y. L. (2010). Natural zeolites as effective adsorbents in water
662 and wastewater treatment. *Chemical Engineering Journal*, 156, 11–24.

Table 1: Characteristics of the industrial wine distillery wastewater

Total COD	Soluble COD	VFA	TSS	VSS
40 g/L	16 g/L	4.8 g/L	5 g/L	2.8 g/L
Total N	Ammonia	Total P	Phosphates	Alkalinity
2 g/L	0.14 g/L	0.35 g/L	0.16 g/L	3100 meq/L

Table 2: Cost analysis for the case study. A: No further restrictions on the interconnections; B: Bypass from source to sand filter not allowed; C: Bypass from source to sand filter or source to sink not allowed.

Flowsheet	A	B	C
CAPEX [M\$]	18.70	27.85	28.25
UASB	41.2%	54.1%	54.2%
Electricity Generator	14.4%	16.4%	16.4%
Sand Filter	14.1%	9.4%	9.3%
Struvite Crystalizer	13.2%	8.8%	8.8%
Ion Exchange Column	17.1%	11.3%	11.3%
OPEX [M\$/year]	1.69	2.78	2.86
UASB	41.2%	54.1%	54.2%
Electricity Generator	14.4%	16.4%	16.4%
Sand Filter	14.1%	9.4%	9.3%
Struvite Crystalizer	13.2%	8.8%	8.8%
Ion Exchange Column	17.1%	11.3%	11.3%
SALES [M\$/year]	2.57	4.19	4.25
Electrical Power	80.4%	83.5%	83.5%
Struvite fertilizer	6.2%	4.0%	4.0%
Ammonia	13.4%	12.5%	12.5%
NPV [M\$]	-7.68	-10.75	-10.93

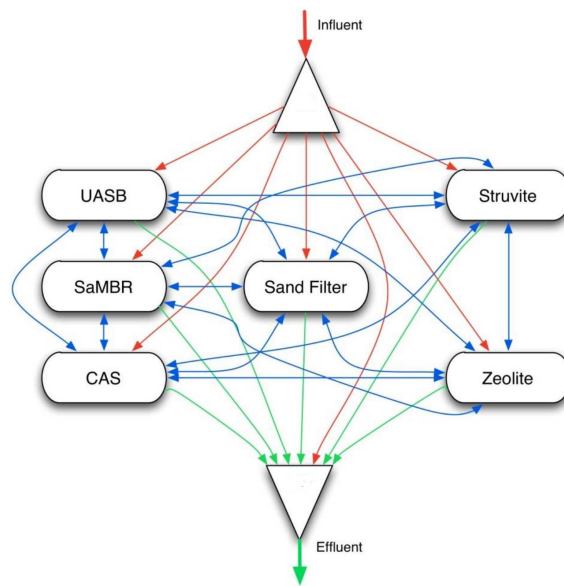


Figure 1: Illustration of a simple superstructure layout.

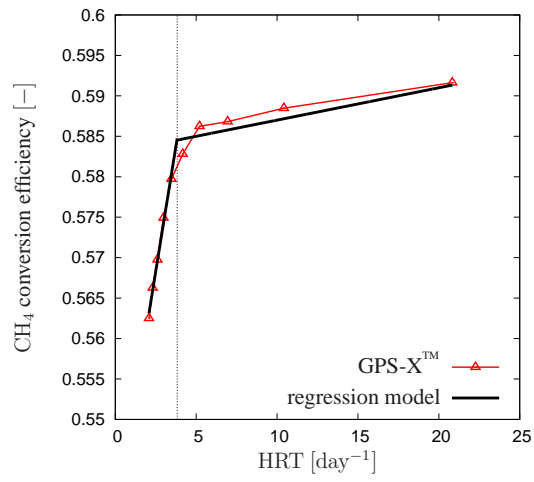


Figure 2: Illustration of a piecewise-linear performance model obtained from GPS-X™ simulated data.

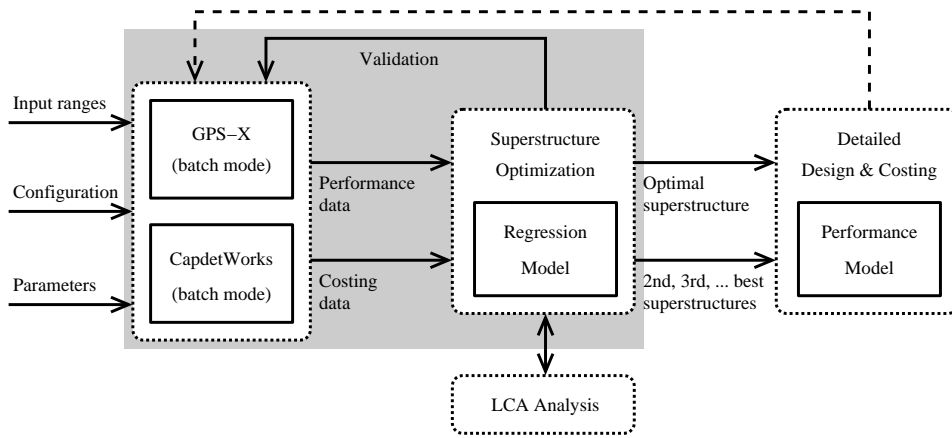


Figure 3: Illustration of the proposed methodology based on superstructure optimization and regression models.

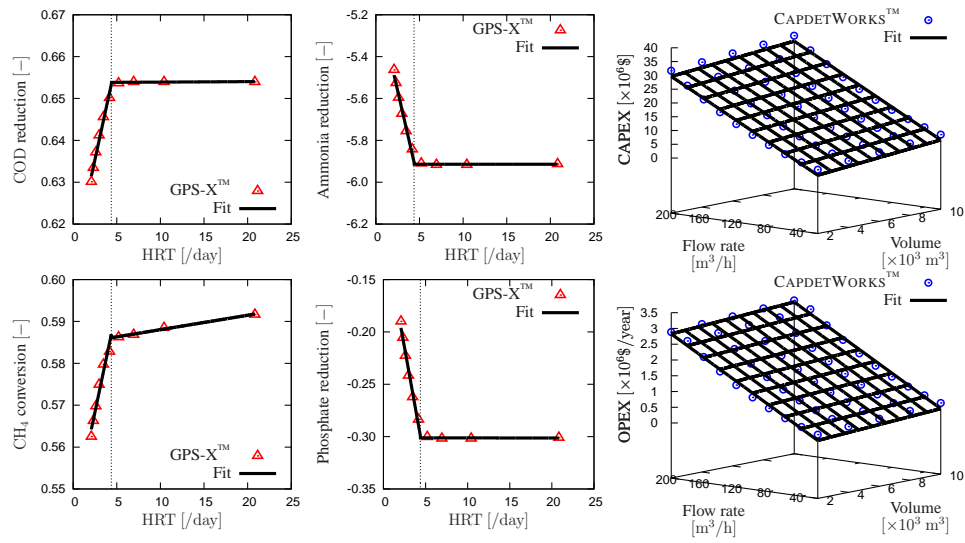


Figure 4: Subset of regression models for performance and cost prediction of a UASB unit. Top-left: COD reduction vs. HRT; Bottom-left: COD conversion into CH₄ (expressed as equivalent COD) vs. HRT; Top-center: Ammonia reduction vs. HRT;^a Bottom-center: Phosphate reduction vs. HRT;^a Top-right: CAPEX vs. unit volume and influent flow rate; Bottom-right: OPEX vs. unit volume and influent flow rate; Legend: Solid lines represent regression fits; Triangles denote GPS-XTM simulation results; Circles denote CAPDETWORKSTM simulation results.

^aNegative reduction ratios for ammonia and phosphates indicate a net increase due to the conversion of other forms of N and P inside the bioreactor.

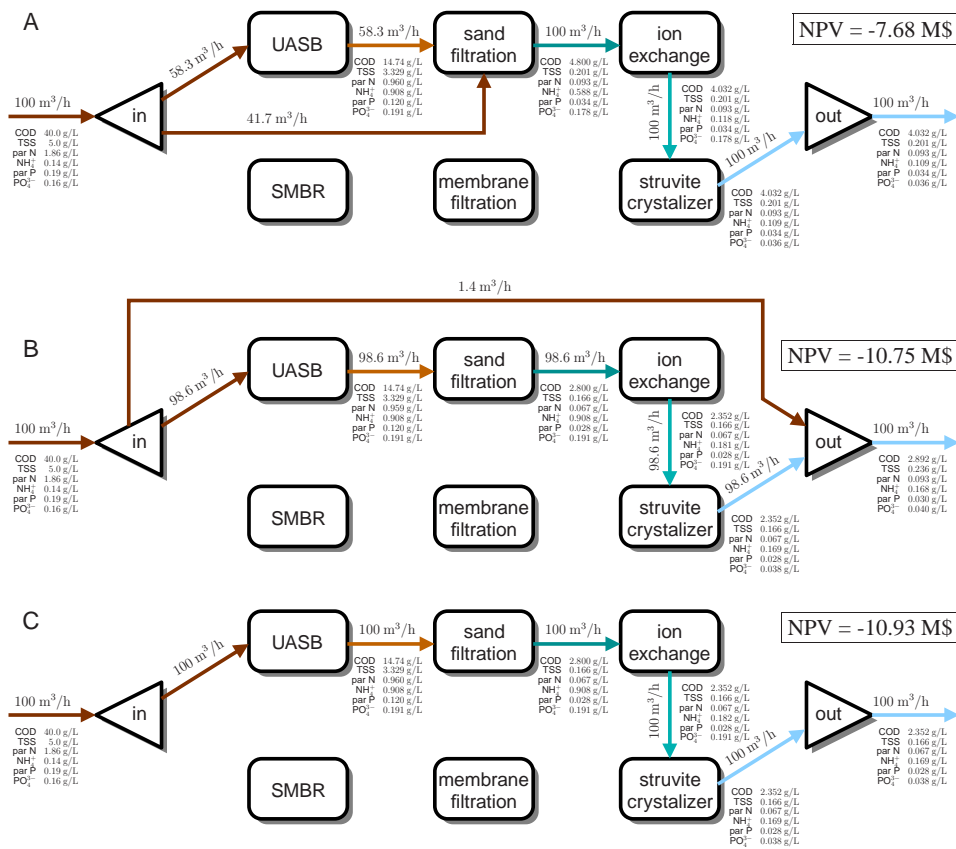


Figure 5: Optimal superstructures for the case study. A: No further restrictions on the interconnections; B: Bypass from source to sand filter not allowed; C: Bypass from source to sand filter or source to sink not allowed.