'X-Ray Studies of the Structures of Some Biologically Significant Molecules'

> A Thesis Presented for the Degree of Doctor of Philosophy of the University of London

> > by

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#### Abstract

The crystal structures of three natural-product derivatives have been examined by three-dimensional X-ray diffraction techniques.

6-(N-Benzylformamido)penicillanic acid  $(C_{16}H_{18}O_4N_2S)$ is an isomer of the antibiotic, benzylpenicillin, but possesses only one ten-thousandth part of its antibacterial activity. The crystals are monoclinic with a = 19.58 Å, b = 6.427 Å, c = 13.98 Å,  $\beta$  = 108° and spacegroup C2, with four molecules in the unit cell. The structure, which is shown below was determined with the aid of a graphical Patterson



superposition technique. It is similar to those found by previous X-ray studies on the penicillins except for the position of one of the possible hydrogen-bonding groups (0'). This difference is consistent with current explanations for the antibiotic properties of the penicillins.

The crystal structure of a spiro-cyclic derivative of another antibiotic series, the cephalosporins, has been determined. This spiro compound  $({}^{C}_{19}{}^{H}_{22}{}^{N}_{4}{}^{0}_{4}{}^{S}_{2}$ .  ${}^{3H}_{2}{}^{0}_{3H})$ crystallizes in the monoclinic system with a = 8.063 Å, b = 7.184 Å, c = 19.72 Å and  $\beta$  = 100.3°. The spacegroup is  ${}^{P2}_{1}$  with two molecules per cell. The molecule, represented by



the diagram above, forms a novel tricyclic system and presents many interesting aspects of structure and conformation, which are discussed in some detail.

Attempts have also been made to determine the crystal structure of zeorin acetate  $(C_{32}H_{54}O_{3})$ , a member of the triterpene series. The crystals are orthorhombic with a = 8.78 Å, b = 11.59 Å, c = 29.15 Å and belong to the spacegroup  $P2_{1}2_{1}2_{1}^{2}$ , with four molecules in the unit cell. This compound presents substantial crystallographic problems and efforts have been made to interpret its Patterson function mainly with the aid of a relatively modern Patterson searching technique.

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#### CHAPTER 1

The Phase Problem and an

Introduction to the Patterson Function

One of the first stages in the X-ray analysis of a crystal structure is the measurement of the intensities of the spectra that arise from the diffraction of X-radiation by the threedimensional crystal lattice. Each diffracted beam can be regarded as the 'reflexion' of X-rays by a particular set of parallel crystal planes. Directionally, a diffracted beam will depend only upon the orientation and separation of its reflecting planes with respect to the 'unit cell' of the crystal lattice. The intensity, however, depends upon the type and arrangement of the atoms with regard to these planes.

If there are N atoms in the unit cell, the diffraction pattern of the crystal lattice can be regarded as made up from the patterns of N atomic lattices and the total wave diffracted from the unit cell can be described by the equation

$$F(hkl) = \sum_{r}^{N} f_{r} exp^{i\phi_{t}} \qquad \dots \qquad (1)$$

where F(hkl) is the 'Structure Factor' for the particular reflexion and  $f_r$  is the 'form factor' of the rth. atom. The

form factor describes the power with which atoms of type r 'reflect' X-rays and it depends upon the atomic number and decreases with increasing Bragg angle,  $\Theta$ .  $\phi_r$  is the phase angle of the diffracted beam from the rth. atomic lattice.

Since a crystal is periodic in three dimensions, Bragg (1929) has shown that the electron density ( $\varrho$ ) at any point x, y, z can be represented by a three-dimensional Fourier series of the form

$$\left( (xyz) = \frac{1}{V} \sum_{h,k} \sum_{h=0}^{\infty} \sum_{h=0}^{\infty} F(hkl) \exp(-2\pi i(hx + ky + lz)) \dots (2) \right)$$

where V is the volume of the unit cell and the function is everywhere continuous. A good and convenient approximation to the structure factor is obtained assuming that all the electron density in the crystal occurs in spherically symmetrical atoms. The structure factor (equation 1) can then be written as

$$F(hkl) = \sum_{r=1}^{N} f_{r} \exp 2\pi i(hx_{r} + ky_{r} + lz_{r}) \quad ..... \quad (3)$$

where  $x_r$ ,  $y_r$ ,  $z_r$  are the coordinates of the rth. atom expressed as fractions of unit-cell edges. Thus the expression (2) for the electron density can be written as

$$\rho(xyz) = \frac{1}{V} \sum_{k,k,\ell}^{A} \sum_{r} \left\{ \sum_{i=r}^{A} f_{r} \exp -2\pi i \left( h(x-x_{r}) + k(y-y_{r}) + l(z-z_{r}) \right) \right\} \dots (4)$$

The Fourier series (4) will consist of N resolved, spherically symmetrical peaks at the positions of the atoms  $x_r$ ,  $y_r$ ,  $z_r$  and

the number of electrons in each peak will be equal to  $Z_r$ , the atomic number. (The Fourier in the form of equation (4) is pertinent to the discussion of the Patterson in later sections).

Crystal structure analysis involves substituting the values of the structure factors, F(hkl), into equation (2) in order to obtain a representation of the electron density distribution in the unit cell. However, the structure factors are, in general, complex quantities and the X-ray crystallographer observes intensities, which are proportional to the square of the modulus of the structure factors,  $|F(hkl)|^2$ . It is the determination of the phases that constitutes the 'phase problem' in X-ray crystallography.

One of the most successful methods proposed to overcome this problem is due to Patterson (1935). He derived a Fourier series with the coefficients, F(hkl), replaced by  $|F(hkl)|^2$ and related the series to the inter-atomic vectors in the unit cell. The rest of this chapter is devoted to a description of this 'Patterson series' as it has been used extensively throughout the thesis. Some of the simpler applications are also described here.

Using for simplicity the one-dimensional series of period 'a' the equation

$$\rho(x) = \frac{1}{a} \sum_{h=-ab}^{ab} F(h) \exp -2\pi i(hx)$$
 ..... (5)

gives the electron density at any point x. The density distribution about x can also be expressed as a function of a parameter u, i.e.  $\rho(x + u)$ . The total amount of material between x and x + dx is  $\rho(x)dx$ , which gives the weighted distribution about x as  $\rho(x) \rho(x + u)dx$  and the weighted average distribution as

$$A(u) = \int_{0}^{1} \rho(x) \rho(x + u) dx \qquad \dots \qquad (6)$$

Thus from equation (5) and utilising the property that

$$\int_{0}^{1} \exp -2\pi i(m + n)x \, dx = 1 \text{ if } n = -m \text{ and is otherwise}$$

zero, equation (6) gives

$$A(u) = a^{-2} \sum_{h=-\infty}^{\infty} |F(h)|^2 \exp 2\pi ihu$$
 ..... (7)

This is the one-dimensional Patterson series and the derivation can be extended to three dimensions to give

$$A(uvw) = V^{-2} \sum_{h,k,e_{2} \to 0} \sum_{h,k,e_{2} \to 0} |F(hkl)|^{2} \exp(2\pi i(hu + kv + lw)) \dots (8)$$

where the distribution is now a function of the three parameters u, v, w about the point x, y, z. The general form of the three-dimensional function is usually written as

$$P(uvw) = \frac{1}{V} \sum_{h,k,\ell} \sum_{n=0}^{\infty} \left| F(hkl) \right|^2 \cos 2\pi (hu + kv + lw) \dots (9)$$

Equation (9) differs from equation (8) for A(uvw) by a factor of 1/V and in the use of the cosine. The former factor is merely a result of defining the function as an average value of the electron density product, and the latter because in the absence of fluorescence,  $F(hkl) = F^*(\bar{h}k\bar{l})$ , so that their imaginary components cancel. As the cosine function is even, it follows that the Patterson function is centrosymmetric.

The physical meaning of the Patterson can be readily seen by considering it in the form of equation (6). Thus A(u) will only be large when both Q(x) and Q(x + u) are large, so that peaks appear in the Patterson corresponding to peaks in the electron density distribution separated by a vector 'u'. The Patterson function, then, will consist of a large, multiple peak at the origin (ie. the sum of the product Q(x) Q(x) for all x), and peaks whose vectors from the origin represent inter-atomic vectors in the crystal. Although a Fourier series is a continuous function it is sometimes useful to describe the Patterson as the vector set of the electron density.

#### Properties of the Patterson

Writing the one-dimensional structure factor in the form of equation (3), it follows that

$$|F(h)|^2 = \sum_{\mathbf{f},\mathbf{s}>1}^{n} f_{\mathbf{f},\mathbf{s}} \exp 2\pi ih(\mathbf{x}_{\mathbf{r}} - \mathbf{x}_{\mathbf{s}}) \dots (10)$$

Substituting equation (10) in (7) gives

$$A(u) = a^{-2} \sum_{h_{2}-\delta}^{\infty} \left\{ \sum_{\substack{t,s>1\\ t,s>1}}^{N} f_{r} f_{g} \exp 2\pi ih(u - (x_{g} - x_{r})) \right\}$$
  
=  $a^{-2} \sum_{\substack{h_{2}-\delta}}^{\infty} \sum_{\substack{t=1\\ t_{2}-\delta}}^{N} f_{r}^{2} \exp 2\pi ihu + a^{-2} \sum_{\substack{h_{2}-\delta}}^{\infty} \sum_{\substack{t=1\\ t_{2}-\delta}}^{N} f_{r} f_{g} \exp 2\pi ih(u - (x_{g} - x_{r})) \dots (11)$ 

The Patterson peaks thus correspond to the Fourier representation of artificial atoms with specific 'form factors'. N of these, represented by the first term on the right hand side of equation (11) coincide at the origin with 'form factors' of  $f_r^2$ ; the remaining N(N-1) are situated at points  $\overline{+}(x_s - x_r)$  with 'form factors' of  $f_r f_s$ .

Thus the total number of electrons in the origin peak is  $\sum_{r=1}^{N} Z_{r}^{2}$ , and peaks corresponding to a vector between two atoms at  $x_{r}$  and  $x_{s}$  respectively will contain  $Z_{r}Z_{s}$  'electrons' and the peak height will be roughly proportional to this product.in the three-dimensional case. It can also be noted that the volume of a three-dimensional Patterson peak is roughly eight times that of a corresponding peak in O(xyz). Coupled with the fact that the Patterson contains  $N^2$  peaks in the same volume as the electron density contains N, this means that the resolution of the former is poor compared to that of the latter.

Also, the origin peak can be removed by using as coefficients in the summation,  $|F_0(h)|^2$ , where

$$\left|F_{o}(h)\right|^{2} = \left|F(h)\right|^{2} - \sum_{r>1}^{N} f_{r}^{2}$$

The new series will retain all the information regarding interatomic vectors and may also show peaks that were previously obscured by the large origin peak.

# Sharpening

Patterson (1935) suggested a method of improving the resolution of the Patterson series, which in effect attempts to transform the observed  $|F(h)|^{\nu}$  terms into what they would be if the crystal were composed of point atoms. As the different form-factor curves are, in fact, very similar when scaled to unity, he suggested setting up an average f-factor per electron of

$$\hat{f}(h) = \sum_{r=1}^{N} f_r(h) / \sum_{r=1}^{N} Z_r$$

and the sharpened synthesis is then computed by dividing each  $|F(h)|^2$  term by the appropriate  $\hat{f}(h)$ .

Another factor to be taken into account is the thermal

motion of the atoms. This, if isotropic and uniform for all atoms, tends to reduce the observed values of |F(h)| by a factor of

exp ( -B 
$$\sin^2 \Theta / \lambda^2)$$

where  $\lambda$  is the wavelength of the X-radiation and B a constant for the particular crystal, which can be estimated from the intensity data by the method of Wilson (1942). Thus, division of each  $|F(h)|^2$  by

$$\hat{T}^{2}(h) \exp(-2B \sin^{2} \theta / \lambda^{2})$$

willgive a Patterson function sharpened to represent point atoms at rest.

An undesirable result of sharpening the Patterson is the appearance of false peaks. Equation (2) indicates that a Fourier series is ideally the sum of an infinite number of terms. In practice the number of terms is limited but the unsharpened series converges rapidly because of the thermal attenuation of the terms. However, if sharpening is applied, the importance of these high-order terms will be greatly increased. The truncation of the data becomes obvious and produces series\_ termination ripples in the Fourier map.

In recent years, more empirical sharpening functions have been used which attempt to compromise between resolving power and freedom from non-convergence effects. For example, Shoemaker et al. (1953) have used the function

$$(2\sin\theta/\lambda)^4 \exp - \alpha^2 (2\sin\theta/\lambda)^2$$
 with  $\alpha = 2.2$ 

in the structure analysis of DL-serine. The function had a maximum value at  $\sin \theta = 0.5 \sin \theta_{max}$  and tended to reduce the Fourier ripple.

## The Use of Symmetry

Harker (1936) suggested that the use of the symmetry of the spacegroup could lead to a simplification in the interpretation of the Patterson.

For a diad, coincident with the b axis (for example), the coordinates of a pair of atoms related by the symmetry are

The vectors between them will have components

+(2x 0 2z)

and the 'Harker section' in the Patterson at  $\mathbf{Y} = 0$  will contain peaks corresponding to vectors between all atoms related by this symmetry element. Equation (9) takes the form

$$P(uOw) = \frac{1}{V} \sum_{\substack{k,k,\ell=-\infty \\ k_{1} \neq k_{2} \neq \infty}} \sum_{\substack{k \neq k_{2} \neq \infty \\ k_{2} \neq \infty}} \left| F(hkl) \right|^{2} \cos 2\pi (hu + lw)$$

$$= \frac{1}{V} \sum_{\substack{k,\ell=-\infty \\ k_{2} \neq \infty}} \sum_{\substack{k \neq k_{2} \neq \infty \\ k_{1} \neq \infty}} \sum_{\substack{k \neq k_{2} \neq \infty \\ k_{1} \neq \infty}} \sum_{\substack{k \neq k_{2} \neq \infty \\ k_{1} \neq \infty}} \sum_{\substack{k \neq k_{2} \neq \infty \\ k_{1} \neq \infty}} \sum_{\substack{k \neq k_{2} \neq \infty \\ k_{1} \neq \infty}} \sum_{\substack{k \neq k_{2} \neq \infty \\ k_{1} \neq \infty}} \sum_{\substack{k \neq k_{2} \neq \infty \\ k_{1} \neq \infty}} \sum_{\substack{k \neq k_{2} \neq \infty \\ k_{1} \neq \infty}} \sum_{\substack{k \neq k_{2} \neq \infty \\ k_{1} \neq \infty}} \sum_{\substack{k \neq k_{2} \neq \infty \\ k_{1} \neq \infty}} \sum_{\substack{k \neq k_{2} \neq \infty \\ k_{2} \neq \infty}} \sum_{\substack{k \neq k_{1} \neq \infty \\ k_{2} \neq \infty}} \sum_{\substack{k \neq k_{2} \neq \infty \\ k_{1} \neq \infty}} \sum_{\substack{k \neq k_{2} \neq \infty \\ k_{2} \neq \infty}} \sum_{\substack{k \neq k_{2} \neq \infty \\ k_{2} \neq \infty}} \sum_{\substack{k \neq k_{2} \neq \infty \\ k_{2} \neq \infty}} \sum_{\substack{k \neq k_{2} \neq \infty \\ k_{2} \neq \infty}} \sum_{\substack{k \neq k_{2} \neq \infty \\ k_{2} \neq \infty}} \sum_{\substack{k \neq k_{2} \neq \infty \\ k_{2} \neq \infty}} \sum_{\substack{k \neq k_{2} \neq \infty \\ k_{2} \neq \infty}} \sum_{\substack{k \neq k_{2} \neq \infty \\ k_{2} \neq \infty}} \sum_{\substack{k \neq k_{2} \neq \infty \\ k_{2} \neq \infty}} \sum_{\substack{k \neq k_{2} \neq \infty \\ k_{2} \neq \infty}} \sum_{\substack{k \neq k_{2} \neq \infty \\ k_{2} \neq \infty}} \sum_{\substack{k \neq k_{2} \neq \infty \\ k_{2} \neq \infty}} \sum_{\substack{k \neq k_{2} \neq \infty \\ k_{2} \neq \infty}} \sum_{\substack{k \neq k_{2} \neq \infty}} \sum_{\substack{k \neq k_{2} \neq \infty}} \sum_{\substack{k \neq \infty \\ k_{2} \neq \infty}} \sum_{\substack{k \neq k_{2} \neq \infty}} \sum_{\substack{k \neq k_{2} \neq \infty}} \sum_{\substack{k \neq \infty \\ k_{2} \neq$$

where  $C_{h,l} = \sum_{k=-\infty}^{\infty} |F(hkl)|^2$ . Use of equation (12) reduces computation to that of the two-dimensional case but as threedimensional data are used the resolution of the Harker section will be far better than the corresponding two-dimensional projection. Analogous results can be derived for screw axes but the relevant Harker plane is then displaced along the screw. For mirror and glide planes, the summation reduces to onedimensional Harker lines normal to the plane.

In any Harker region there are always some extra peaks which arise accidentally and arise from non-equivalent atoms. Such peaks confuse the impression and are not readily distinguishable from true Harker peaks.

### Interpretation of the Patterson

The complexity of this problem depends upon the nature of the substance being examined and consequently, the methods of solution provide one of the most diverse and interesting aspects of crystal structure analysis. This thesis is concerned with organic molecules which for convenience can be divided into two groups :

(a) Molecules containing a relatively large number of atoms of low atomic number with only a few heavy scatterers.

(b) Molecules containing only light-atom scatterers.

The interpretation of the Patterson of crystals containing type (a) molecules is reasonably straightforward. The initial step is the determination of the coordinates of the heavy atoms.

## The Heavy Atom Method

It has been shown (page 12) that the height of a Patterson peak representing a vector between two atoms of atomic number  $Z_r$  and  $Z_s$  respectively is roughly proportional to  $Z_r Z_s$ . If the crystal contains relatively few atoms of high scattering power, the Patterson peaks of these 'heavy' atoms will stand out against a background of smaller peaks. It is then usually a simple matter to determine the coordinates of the heavy atoms, especially by making use of Harker regions. If partial structure factors,  $F^H(hkl)$ , are then calculated from these positions, the total structure factor is given by

 $F(hkl) = F^{H}(hkl) + F^{L}(hkl)$ 

where  $F^{L}(hkl)$  is the structure factor contribution from the light atoms. In general,  $F^{H}(hkl)$  will be much greater than  $F^{L}(hkl)$  and the computation of a Fourier series using the phases of  $F^{H}(hkl)$  applied to the observed structure amplitudes should enable the location of light atoms. In the computer program available here, a rejection test can be applied where F(hkl) is excluded from the Fourier synthesis if

 $|F^{H}(hkl)| \leq x |F_{o}(hkl)|$ 

where x is normally 0.25 - 0.33 and  $|F_o(hkl)|$  is the observed structure amplitude. In this way, structure factors to which the heavy atoms contribute very little can be excluded from the summation because of uncertainties in phase. In most cases, the first Fourier map is not clear enough to locate all the light atom peaks . The above process must be repeated several times, using at each stage the atomic positions located in the preceding map to phase the structure factors for the next. Lipson and Cochran (1953) have suggested that the method works best when the sum of the squares of the atomic numbers of the heavy and light atoms are approximately equal, i.e. each group contributes approximately equally to the average intensity.

#### Image Seeking

Clastre and Gay (1950), Garrido (1950), McLachlan (1950) and Beevers and Robertson (1950) simultaneously published methods for the interpretation of the Patterson which are classed as 'superposition' methods since they involve the superposition of one Patterson map on another. These methods can be applied when the proportion of heavy atoms in the crystal is insufficient to determine the phases of a majority of the

structure factors. Since then, superposition methods have been formalised and given a rigorous theoretical treatment as imageseeking methods by M.J.Buerger, whose work from 1950 onwards is presented in his book, Vector Space (1959).

Considering the Patterson as the vector set of the electron density (page 11), solution of the Patterson can be looked upon as identifying its fundamental set. That this can be done for discrete points was first shown by Wrinch (1939), who used the term 'image' to describe, say, the point ab at the end of the vector  $\overrightarrow{ab}$ , as the way point b looks from a. Thus the Patterson can be imagined as the superposition of the images of the structure as seen from each atom in turn, each image being translated to the origin of the Patterson. Searching for these images can be carried out graphically or by using a digital computer.

A single peak in the Patterson that represents a vector between two symmetry-related atoms is first selected, (it may be useful in this respect to predict the expected peak heights based upon the height of the origin peak). For instance, this may be a heavy-atom peak from which the heavy-atom positions can be found (at points  $r_1, \ldots, r_j$ ). The values of the Patterson are then compared for the set of points  $(r_1+r, r_2+r, \ldots, r_j+r)$  and  $\vec{r}$  is regarded as a vector of the fundamental set only if the Patterson at all these points is large. Buerger

(1959) describes three types of functions for testing the 'vector-end' Patterson values; the product, sum and minimum functions. The product function is defined as the product of the weighted values of the Patterson at the ends of the vectors, and the other two functions are respectively the sum and the minimum of these Patterson values. This multiple comparison is carried out right through the cell by varying the common vector r. In the above heavy-atom example, the weight of each Patterson value will be the same, but if other atomic species . are included as 'image points', the vector-end Patterson values must be weighted according to their scattering powers. Buerger discusses the merits of each image-seeking function and their relation to the electron density. He recommends the minimum function as it is simple to use and the other two tend to give false maxima. Thus a contoured map of, say, the minimum function can be obtained for the unique part of the unit cell, and the peaks will represent possible atomic positions.

If two image points are used in the search, the resulting function is said to be of rank two. Normally, this is not enough to reveal the complete structure and a stepwise procedure has to be employed, whereby functions of higher rank are computed by using as image points, atomic positions that have been determined from lower-ranking functions.

Graphically, the procedure can be carried out by preparing

a duplicate contoured map of the Patterson on tracing paper and laying it on the original map with its origin on the selected single peak of the lower map. The minimum function can then be contoured on a perspex sheet superimposed on both maps. In practice it is difficult to prepare, by this method, minimum functions of rank higher than two.

For structures of type (b) (page 16), containing no heavy atoms, the solution of the Patterson is generally quite complex. The function normally consists of clusters of unresolved peaks, some of which may not be distinguishable from the background. Recognition of single peaks becomes extremely difficult and consequently, normal image-seeking methods are unsuitable. However, in recent years, modified image-seeking methods have been developed for specific use with light-atom structures. A description of these methods is deferred until later (Chapter 7), prior to the account of attempts to solve the crystal structure of zeorin acetate  $(C_{32}H_{54}O_3)$  by a Patterson searching method.

#### CHAPTER 2

#### Data Treatment and

General Crystallographic Methods Used

All intensity data were collected photographically by the Weissenberg method using filtered CuKA radiation. Two fourfilm exposures on Ilford Industrial 'G' were carried out for each layer, the time ratio being approximately 60 : 1. Using the equi-inclination setting for non-zero layers, the bulk of the data was collected about the shortest crystallographic axis, and between one and three layers about a second axis for data Intensities were estimated visually using calicorrelation. brated'wedges', and inter-film scales were calculated by hand. No corrections were applied for absorption. Unless the mass absorption coefficient of the crystal is very high, it is expedient to try to avoid making absorption corrections by choosing a specimen of small enough dimensions. This was done in each case and the pertinent data are given in the relevant chapters. Primary extinction is due to the interference effects of out-of-phase rays that have been 'back-reflected' into the primary and reflected beams by the stack of planes in the particular reflecting position. A large reduction in intensity can occur for strong, low-order reflexions and consequently

any suspected primary extinction was dealt with by omitting the reflexions from the structure refinement.

A series of computer programs were used to obtain a complete set of data on one relative scale. All the programs mentioned form part of a crystallographic system for the Atlas computer. Formulated mainly by M.G.B.Drew (1966), and named ATSYS, this system facilitates the handling of data and enhances continuity by the option of storing both programs and data on magnetic tape. Further details can be obtained from the reference cited and the list of references therein. In each case, the programmer's name is given, and the author, who fully appreciates the exasperation sometimes involved, is grateful to them all.

LSCD.....R.D.Diamand.

The unit cell dimensions are calculated by a leastsquares method outlined by Alexander and Klug (1954). The method assumes that any systematic errors are a function of the Bragg angle,  $\Theta$ , so that an error term is added to the observational equations. In this work the data were derived from Weissenberg films.

FIFI..... R.D.Diamand.

Lorentz and polarisation corrections of the form

$$(Lp)^{-1} = 2 \frac{cos \theta}{1 + cos^2} 2\theta$$

are applied to a set of intensity data, where 5 is the radial cylindrical coordinate of the reciprocal lattice point.

POLO..... M.G.B.Drew.

Procedures can be carried out as follows: (a) Pick out common reflexions from separate lists of data and print out their ratios. Common reflexions will be used to calculate the inter-layer scales (b), and this list usually indicates mis-indexing and other trivial mistakes. (b) Calculate the inter-layer scales by the least-squares method of Hamilton, Rollett and Sparks (1965), and apply them to the data.

(c) Sort a list into any order and modify parts of the list.

LOLA..... R.A.Sparks, modified by the author.

Inter-layer scales can be calculated by the method of Rollett and Sparks (1960), now superseded by the method in POLO.

Other ATSYS programs used for calculations in this thesis are :-

MATT..... the author.

Wilson's (1942) method is applied to estimate the absolute scale and overall temperature factor, from two or

three-dimensional data. The procedure used is analogous to that outlined by Rogers (1965). The data are output to facilitate the plotting of the graph of  $\log_e(\langle I \rangle / \sigma_2)$ against  $\sin^2 \theta / \lambda^2$ , where  $\langle I \rangle$  and  $\sigma_2$  are the average values of  $|F(hkl)|^2$  (on a relative scale) and  $\sum_{r=1}^{n} f_r^2$  for shells of the reciprocal lattice extending over equal intervals of  $\sin^2 \theta$ . The values of B, the overall isotropic temperature factor and K, the scale to be applied to |F(hkl)| are also calculated by a simple least-squares routine.

BOSS..... N.M.Harding, modified by M.G.B.Drew.

The program is basically for the calculation of Fourier series but includes a routine to calculate structure factors from a set of atomic positions. An agreement analysis can also be prepared, i.e. a list of  $\sum F_0(hkl)$  and  $\sum F_c(hkl)$ and R for various groups of reflexions where **k** 

$$R = \sum k \left| |F_{o}| - |F_{c}| \right|$$

 $\sum k |F_{o}|$ 

is the scale factor and  $F_c$  the calculated structure factor, (adopting the more convenient notation), and the summations are over the appropriate groups of reflexions.

DIDO..... R.A.Sparks, modified by the author.

Using the eigenvalue method outlined by Rollett (1965)

the best plane through a set of specified atoms is calculated together with the distances of other atoms from this plane.

BABA..... R.D.Diamand, modified by M.G.B.Drew.

The program calculates structure factors from input atomic coordinates and refines, by a block diagonal, leastsquares method, positions, temperature factors and an overall scale. The anisotropic temperatures factors are of the form  $\beta_{ij}$  where the contribution to the structure factor is

$$\exp - (\beta_{11}h^{2} + \beta_{22}k^{2} + \beta_{33}l^{2} + \beta_{12}hk + \beta_{13}hl + \beta_{23}kl)$$

The method is essentially that outlined by Cruickshank (1961). The scale, applied to  $F_0$ , is coupled with an overall temperature factor in a 2 x 2 matrix, and 3 x 3 and 6 x 6 matrices are computed for positions and anisotropic vibrations respectively. The function minimised is

$$\sum w (k |F_0| - |F_c|)^2$$

where  $\sqrt{w}$  is the weight applied to each observation. The weighting scheme used is that described by Rollett (1961) where

$$\int w = F^{\dagger} / F_{o} \text{ if } F_{o} > F^{\dagger}$$

$$\int w = 1$$
 if  $F_0 < F^{\dagger}$ 

The value of F can be chosen from the agreement analysis of

BOSS to give roughly constant values of  $\sum w (|F_0| - |F_c|)^2$  (i.e.w $\Delta^2$ ) over ranges of  $|F_0|$ .

The program also optionally applies a 'fudge factor' for modification of parameter shifts. This is particularly useful to dampen the oscillation of non-centrosymmetric structures whose origin is not defined.

ELSI..... R.D.Diamand and M.G.B.Drew.

Bond distances and angles are calculated and their standard deviations. The variances of the normal equations obtained from BABA are used to calculate the standard deviations, which are thus lower than their true value as most of the offdiagonal terms are ignored. The formula used for the estimation of the standard deviation,  $\sigma$ , of a variable  $x_r$  is that of Cruickshank (1959) where

 $\sigma^{2}(\mathbf{x}_{r}) = \mathbf{V}_{rr} \sum w \Delta^{2}/(m - n)$ 

 $V_{rr}$  is the variance of the parameter  $x_r$ , m the number of observations and n the number of parameters refined.

A limited number of programs have been used from the 'X-Ray '63' system for the I.B.M. 7090. Thanks are extended to the authors (mentioned in the text) and especially J.Stewart who was chiefly responsible for making the programs available at Imperial College.

### CHAPTER 3

# Chemistry of the Penicillins

and Cephalosporins.

#### The Penicillins.

That the antibiotic penicillin is produced from a mould, later identified as a strain of Penicillium notatum, was first observed by Fleming (1929). Penicillin is active against a large number of organisms such as Diplococcus pneumoniae (pneumonia organism), Streptococcus pyrogenes (associated with scarlet and rheumatic fevers) and Staphylococcus aureus (skin infections). but early clinical use was limited by its instability. This also complicated the chemical work on the structure of penicillin and a further difficulty was that certain bacteria secrete an enzyme. penicillinase, which inactivates the antibiotic. Nevertheless, the potentialities of penicillin for full-scale use against many infectious diseases led to some very intense work from 1940-45. During this period of the Second World War, chemical information was restricted and consequently, an historical account of the structure determination is complex. The following outline of the fundamental chemical points, for which no references are given, was taken from 'The Chemistry of Penicillin', edited by Clarke,

Johnson and Robinson (1949).

The first solid forms of penicillin , though stable when dry, were soon found to be inhomogeneous, and it was not until 1943 that a chemically pure compound possessing the properties of penicillin was isolated. Further studies showed that more than one such compound existed, and the chemical differences between them soon became evident. Hydrolytic degradation of the individual penicillins gave, in each case, the same sulphur-containing amino-acid, penicillamine (I), carbon dioxide and a penillo-



aldehyde (R.CO.NH.CH<sub>2</sub>.CH:O). The structure of (I) was confirmed by synthesis, its absolute configuration being D. The chemical differences between the penicillins were attributed to the identity of the group R in the penilloaldehydes. This provided a basis for nomenclature, so that with  $R = C_6H_5$ .CH<sub>2</sub> the parent penicillin was given the name benzylpenicillin.

Although the penicillins are strong monobasic acids, the action of methanol on benzylpenicillin under mild conditions, gave a monomethyl ester of a dibasic acid, benzylpenicilloic acid. This indicated that the penicillins also contain a labile, but 'masked' carboxyl group, since esterification of free carboxyl groups normally takes place under more vigorous conditions. The monomethyl ester of benzylpenicilloic acid was further degraded to penicillamine (I) and the benzylpenaldic ester (II). Further hydrolysis of the latter compound gave benzylpenaldic acid which

 $Ph.CH_2.CO.NH.CH.CH=0$ 

## (II)

loses carbon dioxide to give benzylpenilloaldehyde.

These reactions are indicated schematically in Figure 1, and showed the free carboxyl group to be associated with the penicillamine residue, and the masked group with the penilloaldehyde. It had also been demonstrated that penicillamine yielded thiazolidines on condensation with carbonyl compounds and consequently, the monomethyl ester of benzylpenicilloic acid was formulated as (III), which was supported by synthesis.



(III)



Thus, for the organic chemists, the problem was reduced to postulating a linkage, working from (III), that would explain the masked carboxyl group of the penicillins. However, any postulated structure also had to satisfy other criteria that had arisen during the very exhaustive chemical work. Three main structural formulae were proposed, namely the oxazolone (IV), tricyclic (V) and  $\beta$ -lactam (VI).



## (IV) Oxazolone



(V) Tricyclic



The  $\beta$ -lactam structure gradually became the most popular, and was finally confirmed by an X-ray analysis by Crowfoot, Bunn, Rogers-Low and Turner-Jones (1949).

Two-dimensional X-ray investigations were carried out on the sodium, potassium and rubidium salts of benzylpenicillin. The latterstwo salts are isomorphous and the Patterson series was used to locate approximately, the metal ion positions. Unfortunately, it was found that the heavy atoms contributed to only a limited number of reflexions, so that the Fourier maps were difficult to interpret except that the sulphur atom was located. The sodium salt was examined by optical methods and the derived Fourier maps were little better than those from the isomorphous salts. However, by comparison of the two sets of Fouriers it became clear that the molecule existed as a 'curled' configuration. not extended as had previously been thought, and when the results were viewed without bias towards any of the proposed structures (the oxazolone structure was originally taken as model), the adoption of trial positions and subsequent refinement, led to the  $\beta$ -lactam structure (VI) for benzylpenicillin. The general configuration of the molecule is represented by (VII). The thiazolidine ring is not planar, C2 lies out of the plane of the other atoms on the opposite side to  $C_7$  and similarly,  $O_8$  projects out of the plane of the  $\beta$ -lactam ring system towards the sulphur atom. The amide group in the phenylacetyl side-chain

is planar and possesses dimensiona of a normal amide.



## (VII)

The crystal structure of phenoxymethylpenicillin has been determined by Abrahamsson, Crowfoot Hodgkin and Maslen (1963). The molecule shows marked similarities in the general configurations of the thiazolidine and  $\beta$ -lactam rings to those in benzylpenicillin. The main difference is that in benzyle penicillin, the plane of the benzene ring is approximately parallel to the  $\beta$ -lactam-thiazolidine system, whereas in phenoxymethylpenicillin, the benzene ring is considerably inclined to this system. The differences may be largely due to packing considerations in the crystals.

In 6-aminopenicillanic acid (6-APA), as determined by Diamand (1963), the out-of-plane atom in the thiazolidine ring is  $N_1$ , which is projected on the same side of the plane as  $N_{14}$ . This is in contrast to the situation in the other two penicillins where **C**<sub>9</sub> deviates most from the best plane. The mechanism by which penicillin kills bacterial cells is still not clear. Evidence has been presented by Park and Strominger (1957) that the antibiotic interferes with the synthesis of the bacterial cell wall, by reacting preferentially with an enzyme binding site that normally takes part in cell-wall synthesis. Collins and Richmond (1962) suggested that penicillin may bind to the enzyme in preference to some derivative of N-acetylauramic acid (VIII). They point out that



in one of the permissible conformations of N-acetylmuramic acid, the carboxyl group can occupy the same position as that in benzylpenicillin and similarly  $0_3'$  and  $0_{10}'$  in N-acetylmuramic acid can occupy identical positions to  $N_1$  and  $0_{16}$  in benzylpenicillin. They further suggest that the binding of the Nacetylmuramic acid residue to an 'active centre' will depend primarily on ionic and hydrogen-bonding forces, and that the high affinity of penicillin for the active centre could be explained by the fact that the possible hydrogen-bonding group  $0_8$ , has no counterpart in N-acetylmuramic acid. Similar considerations can be applied to the cephalosporins.

Tipper and Strominger (1965) have suggested that penicillin interferes with cell-wall synthesis by binding to an active centre in preference to an acyl-D-alanyl-D-alanine (IX), and



# (IX)

not N-acetylmuramic acid. They show that similar spatial considerations apply as in Collins and Richmond's hypothesis described previously.

6-(N-Benzylformamido)penicillanic acid (6-NBF-PA) represented by (X), is an isomer of benzylpenicillin but Housley



(X)
and Spooner (1965) have measured its antibiotic activity as less than one ten-thousandth of that of benzylpenicillin. A threedimensional X-ray study (described in Chapter 4) has been undertaken mainly to determine the positions of the hydrogen-bonding groups. These studies have, in fact, shown that the carbonyl group  $C_{15} - O_{16}$  is on the wrong' side of the molecule, in terms of Collins and Richmond's explanation, to participate in possible hydrogen bonding to an active centre, and hence its low antibiotic activity is consistent with their theory.

### The Cephalosporins

The isolation of cephalosporin C (ceph. C) has been described by Newton and Abraham (1955). A species of the mould 'cephalosporium' produced a mixture of penicillin N (XI), and



(XI)

another antibiotic, ceph. C. Chemical evidence given by Abraham and Newton (1961) for the structure of ceph. C indicated a monoamino dicarboxylic acid present as a residue of  $\alpha$ -aminoadipic acid linked to the rest of the molecule by its S carboxyl group. The infra-red absorption spectrum of the sodium salt showed a maximum at 5.62  $\mu$ , which is close to the carbonyl stretching frequency of the  $\beta$ -lactam ring in the penicillins. This fact and the results of hydrogenolysis experiments indicated the presence of the part structure (XII).



(XII)

These initial lines of enquiry were based on the assumption that ceph. C contained the characteristic skeleton of the penicillins, although the ultra-violet absorption spectrum of ceph. C showed a maximum at 260 m $\mu$ , which could not be explained in terms of the fused four and five-membered ring system. Further chemical work and nuclear magnetic resonance spectroscopy threw more doubts on this hypothesis, which was thus rejected.

Abraham and Newton then proposed the structure (XIII),



#### (XIII)

which was consistent with all the chemical reactions, while the 3-4 double bond would account for the U.V. maximum at 260 m $\mu$ . However, the possibility of other structures, especially those with a 2-3 double bond were not excluded.

Before any further chemical evidence became available regarding this structure, the X-ray analysis by Crowfoot Hodgkin and Maslen (1961) indicated the presence of a six-membered ring containing sulphur. Thenceforward, the chemical and X-ray crystallographic progress was rapid, and both finally substantiated the structure (XIII) for ceph. C.

The full X-ray analysis was complicated by the low

reflecting power and rapid deterioration of the crystals. Consequently, the accuracy achieved was limited but the analysis showed that the atoms  $C_2$ ,  $C_3$ ,  $C_4$ ,  $N_5$ ,  $C_{10}$  and  $C_{15}$  form a plane with  $S_1$  about 0.6 Å above the plane, and  $C_6$  about 0.6 Å below it. The length of the bond  $C_3 - C_4$  (1.31 Å) is in agreement with its formulation as a double bond. The carbonyl group of the amide is nearly normal to the  $\beta$ -lactam ring and the protons at  $C_6$ ,  $C_7$  are cis and in the same configuration as in the penicillins.

Lactonisation of ceph. C by hydrochloric acid at room temperature produces another antibiotic, ceph. Cc, represented by (XIV). The X-ray analysis of ceph. Cc (carried out by





Diamand (1963)) has confirmed the general configuration of the molecule and provided quite accurate bond lengths, the true o standard deviations probably being 0.02 - 0.03 A.

The biological activity of the cephalosporins has been studied by Newton and Abraham (1956). Ceph. C shows only 0.1 %

of the activity of benzylpenicillin against Staph. aureus, but is relatively stable to penicillinase. By analogy with the penicillins Loader, Newton and Abraham (1961) suggested that the substitution of, say, a phenylacetyl group for the *d*-aminoadipic acid residue in ceph. C might produce more active compounds that would still be stable to penicillinase. Treatment of ceph. C with dilute acid at room temperature produced the ceph. C nucleus, 7-aminocephalosporanic acid (XV), which although



(XV)

biologically inactive itself, was acylated to give 7-phenylacetamidocephalosporanic acid (XVI). This possesses approximately



(XVI)

one hundred times the activity of ceph. C against Staph. aureus, i.e. similar to benzyl penicillin but more stable to penicillinase.

Another interesting reaction of ceph. C was observed by Hale, Newton and Abraham (1961). The acetoxy group in ceph. C can be displaced by nucleophiles, especially hetero tertiary bases such as pyridine. The resulting compounds, known as the ceph.  $C_{A}$  family, show increased antibacterial activity and are represented by (XVII), X = pyridine etc.





Muggleton, O'Callaghan and Stevens (1964) have prepared many analogues of ceph. C, in which both the  $\alpha$ -aminoadipic acid sidechain and the acetoxy group were replaced. Their work led to the production of cephaloridine (XVIII) (Trade name 'Ceporin'



(XVIII)

Glaxo Ltd.), a highly active antibiotic of low toxicity.

Cocker et al. (1965) have replaced the acetoxy group in ceph. C with sulphur nucleophiles such as thiourea, and Fazakerly et al. (1965) produced spirocyclic compounds by the reaction of certain bidentate nucleophiles with 7-phenylacetamido cephalosporanic acid. Nucleophiles such as pyrid-2-thione and thiouracil give products that lack the U.V. absorption maximum at 260 m $\mu$ , and it was suggested that the reaction, which is illustrated below for pyrid-2-thione, is a two-stage process.



First, the acetoxy group is displaced from the cephalosporanic acid derivative by the sulphur end of the nucleophile, followed by internal Michael addition by the nitrogen atom giving compounds of type (XIX), with a spiro atom at the 3 position. The spiro compounds are dextrorotatory and possess weak antibacterial activity. A substance thought to be the thiazolinium derivative (XX),  $R_1$ ,  $R_2$  = Et, H or H, Et was formed when



7-phenylacetamidocephalosporanic acid and N-ethylthiourea were mixed in aqueous solution at  $37^{\circ}$  for several days. Electrophoresis experiments, together with an I.R. absorption maximum at 1610 - 1630 cm<sup>-1</sup> due to an ionized carboxyl group, indicated that it occurred as a zwitter-ion. The compound possesses only one fifteenth of the anti-bacterial activity of 7-phenyl acetamidocephalosporanic acid against Staph. aureus (Long (1966)). If (XX) is correct it contains two new centres of asymmetry.

The prospects of determining the detailed stereochemistry of this molecule by any means other than X-rays seemed poor. Thus can three dimensional X-ray determination has been carried out to determine the stereochemistry at the asymmetric centres, and to study the conformations of this new tricyclic system. The results, which are reported in Chapter 5, indicate that (XX) is essentially correct, but should be written more precisely as (XXa).



(XXa)

# CHAPTER 4

The Crystal Structure of

6-(N-Benzylformamido)penicillanic Acid

## Preliminary Data

The crystals were prepared and the compound characterized by J.R.Housley and D.F.Spooner of the Boots Pure Drug Co. Ltd., Nottingham and supplied to the author by J.F.Collins of the Medical Research Council, Mill Hill, London. After treating 6-benzylaminopenicillanic acid at room temperature with an ethereal solution of acetic formic anhydride, 6-NBF-PA was crystallized from ether/light petroleum ( $40^{\circ} - 60^{\circ}$ ) at  $0^{\circ}$  C mainly as clusters of ill-formed,monoclinic needles, elongated along [010]. The solution in dioxan is dextrorotatory with  $\left[ \mathcal{A} \right]_{D}^{25} = +351^{\circ}$ . The structural formula and numbering scheme are shown in the pull-out diagram inside the back cover.

Formula  $C_{16}H_{18}O_4N_2S$  Molecular Weight (formula) = 335 Mass absorption coefficient for  $CuK\sigma_{\mu}=1.8.5 \text{ cm}^{-1}$ 

Unit-cell dimensions  
a = 19.58 (0.01 Å), b = 6.427 (0.003) Å, c = 13.98 (0.006) Å  

$$\beta = 108.0^{\circ}$$
 (0.1), V = 1673 Å<sup>3</sup>, Z = 4 F(000) = 704 electrons.

 $D_{obs}$  (by flotation) = 1.33 g.cm<sup>-3</sup>  $D_{calc}$  = 1.33 g.cm<sup>-3</sup>

Absent Spectra:

only among hkl for h + k = 2n + 1

The compound is optically active, so that the spacegroup is uniquely determined as C2. 1580 independent reflexions were estimated visually and correlated from six layers collected about the b axis and reflexions from the zero layer about the c axis. The crystal used for the collection of the main axis data had the approximate dimensions  $0.50 \times 0.15 \times 0.10$  mm and in view of the value of the absorption coefficient(18.5 cm<sup>-1</sup>) absorption corrections were considered unnecessary.

A Wilson plot based on three-dimensional data was calculated by MATT and is shown in Figure 1. The data were divided into 17 shells, each containing 90 - 110 reflexions. Unobserved reflexions were included as  $|F_{unobs}| = 2/3 |F_{min}|$ , where  $|F_{min}|$  is the minimum observable value of |F| at the appropriate point in reciprocal space. The least-squares method gave B = 2.98, K = 145.9 and the results from the graph are B = 2.64, K = 159.2. (After structure refinement the final scale was found to be 131.4.)

### Structure Determination

The four sulphur atoms in the unitcell give rise to ± six S - S inter-vectors which overlap to give ± three



Figure 1. 6-NBF-PA Wilson plot based on three-dimensional data.

distinct peaks in the three-dimensional Patterson. In terms of the coordinates x, y, z of one of the sulphur atoms their positions are given by

$$\frac{1}{2} \begin{cases} 2x, 0, 2z \\ \frac{1}{2}, \frac{1}{2}, 0 \\ 2x + \frac{1}{2}, \frac{1}{2}, 2z \end{cases}$$

Only the first peak need be considered as the second and third are merely related to the origin and first peaks respectively by the C-face centering of the spacegroup.

The [010] Patterson projection was computed with each  $|F|^2$  term modified by its value of  $(Lp)^{-1}$  as a crude sharpening function. The position of the S - S peak could not be unambiguously assigned, and of the four most likely peaks, the highest was chosen to give the tentative sulphur position of x = 0.205, y = 0.0, z = 0.208 (Coordinates of the peak were estimated by Booth's (1948) method.).

A solution of the crystal structure directly from the sulphur position was attempted using the FATAL program written by J.Rollett for the Mercury computer. The program was used to calculate structure factors from the above sulphur position, then compute a three-dimensional Fourier map and finally to search the Fourier map for peaks above a certain height. Six additional peaks were recognised possessing geometry reasonably consistent with that of a fused  $\beta$  -lactam-thiazolidine ring system. However, when the additional positions were input to a second FATAL run, the R value increased from 0.58 to 0.62 and the resulting peaks showed no indication of any further structural details. The method was therefore abandoned (Only two of the additional peaks were near final atomic positions).

In order to check the position of the sulphur atom, a three-dimensional Patterson function was computed using as sharpening function

 $\frac{1}{f^2} \exp(3\sin^2\theta/\lambda^2)$ 

where f is the form factor for sulphur and the resulting Patterson is specifically sharpened for sulphur peaks. The S - S peak was easily located in the Harker section P(u, 0, w) and gave the position of the sulphur atom as x = 0.206, y = 0.0, z = 0.214, which is close to that derived from the [010] Patterson projection. It was thought that the FATAL method may have failed because the sulphur atom was not heavy enough to determine unambiguously enough of the phases, the ratio  $\sum_{H}^{2} Z_{H}^{2}$  to

 $\sum Z_L^2$  being only 0.27. Even though Fourier methods, applied carefully, may have led to the solution of the structure, it was felt that superposition methods would provide a simpler route.

Duplicate copies of the (u,w) sections of the sharpened, three-dimensional Patterson were prepared on tracing paper and contoured at arbitrary but uniform intervals. The origin of the zero section was then laid on the S - S peak of the duplicate and the minimum function plotted on a perspex sheet by drawing the minimum contours of coincident peaks in the superposed Pattersons. The other sections were paired off with their duplicates using an identical u,w displacement. The resulting minimum function was of rank two as the procedure was equivalent to selecting as image points the two sulphur atoms at x = 0.206, y = 0, z = 0.214 and x = -0.206, y = 0, z = -0.214, i.e. related by a diad. Thus the origin of the minimum function was situated half-way along the u,w displacement.

Figure 2 is a composite diagram derived from all the sections of the minimum function. The mirror plane of the Patterson spacegroup, C2/m, is not eliminated in this function, which contains a pseudo mirror plane through y = 0. Although this rendered the y coordinates ambiguous, the main structural features of the molecule were recognisable, as shown in Figure 2. However, the structure near N<sub>14</sub> was ambiguous, though the cluster of peaks to the left of this strongly suggested the location of the benzene ring, giving the molecule a curled conformation as in benzylpenicillin. (This was in fact later vindicated).

Three-dimensional structure factor and Fourier calculations were carried out including the fourteen positions marked in Figure 2. The Fourier map, in which the pseudo mirror plane was now eliminated, gave unambiguous peaks for all 23 atoms, and



indicated that  $C_9$  and  $C_{10}$  had been misplaced. The details near N<sub>14</sub> were clarified and showed that  $C_{15}$  and  $O_{16}$  (diagram inside back cover) were projecting 'down' as seen in Figure 2. The suggested position of the benzene ring was confirmed. Before starting least-squares refinement, a Fourier map phased on all 23 atomic positions was calculated and when deriving the revised atomic positions, double shifts were applied to the observed peak positions as recommended by Donohue (1950).

#### Refinement

Rollett's Mercury program for the least-squares refinement of parameters was now used to refine the atomic coordinates and isotropic temperature factors for all 23 atoms. The method is essentially that used in BABA and described briefly in Chapter 2. Rollett's weighting scheme 2 was used and checks were made at various stages of refinement with an agreement analysis program written by G.A.Mair and the bond distances and angles program of R.A.Sparks.

The course of the refinement is shown in Table 1. At cycle A6, all the thermal parameters were converted into their anisotropic form but only the sulphur coordinates and thermal parameters refined. The anisotropic refinement was then continued including all atoms and switched to the Atlas program Table 1. The refinement process for 6-NBF-PA

	I	
Cycle	🤃 R 🐭	$\sum w \Delta^2 \times 10^{-4}$
Il	0.308	16419
12	0.225	10393
13	0.188	8184
14	0.179	7134
15	0.167	5838
A6	0.154	4622
Α7	0.140	35 4 4
8 <b>A</b>	0.105	1314
A9	0.099	990
A10	0.127	3177
A11	0.102	2154
Al2	0.096	1951
A13	0.095	1912
A14	0.095	1894
A15	0.109	2398
A16	0.106	2197
A17	0.106	2179
<b>A18</b>	0.106	2169
A19	0.103	2038
A20	0.101	1989
¥51	0.100	1984
¥55	0.100	1982
A23	0.100	1981
A24	0.100	1981
A25	0.100	1980
A26	0.100	1980
A27	0.100	1980

I = isotropic refinement

A = anisotropic refinement

BABA at A8. (The F+ value in the weighting scheme was inadvertently decreased at A8 but corrected again at A10). During cycles A12, A13 and A14 it was noted that little convergence was taking place and that the refinement was oscillatory. Thus 164 unobserved reflexions were included at  $\frac{2}{3}F_{min}$  and a fudge factor of 0.75 applied. The process was again disturbed at A18 when the calculated positions of the 18 hydrogen atoms were included in the structure factor calculations but not refined. Hydrogen atom positions were calculated on an X-Ray '63 program written by D.F.High, assuming staggered conformations for the Co and C10 methyl groups. An earlier difference Fourier showed peaks at or near the calculated hydrogen atom positions varying in height from 0.1 to 0.6  $e/A^3$  but the difference map exhibited ripple as high as 0.5 e/  $A^3$  and cannot be taken as conclusive evidence for the hydrogen atom positions - the positions of the methyl hydrogens were especially ill-defined.

Subsequent refinement was extremely slow, and at A27 the coordinate shifts were less than one tenth of their standard deviations, so that refinement was ceased.

### Results and Discussion

The final atomic coordinates are shown in Table 2, together with their standard deviations, and the anisotropic

<u>Table 2</u>. Atomic coordinates and standard deviations for 6-(N-benzylformamido) penicillanic acid (fractional values).

	X	Y	Z	$\sigma_{x}$	$\sigma_{y}$	$\sigma_{z}$
Nl	0.19398	0.19816	0.36500	0.00027	0.00135	0.00042
<b>C</b> 2	0.12891	0.06769	0.34228	0.00035	0.00140	0.00056
C3	0.11559	-0.01477	0.23186	0.00035	0.00200	0.00056
S4	0.20502	-0.00849	0.21308	0.00009	0.00058	0.00014
C5	0.23075	0.21859	0.28823	0.00036	0.00160	0,00054
C6	0.30337	0.22048	0.37611	0.00034	0.00161	0.00054
C7	0.25648	0.16794	0.41425	0.00033	0.00163	0.00055
08	0.26642	0.12197	0.52951	0.00027	0.00123	0.00038
C9	0.08592	-0.22774	0.22020	0.00057	0.00203	0.00087
<b>C1</b> 0	0.06584	0.13802	0.15406	0.00051	0.00232	0.00066
C11	0.06760	0.19723	0.35791	0.00034	0.00166	0.00057
012	0.01415	0.07078	0.36209	0.00030	0.00118	0.00058
013	0.06669	0.38004	0.36511	0.00031	0.00127	0.00056
N14	0.35902	0.07255	0.376 <b>9</b> 9	0.00029	0.00116	0.00047
C15	0 <b>.</b> 3550 <b>6</b>	-0.12767	0.39730	· 0.00041	0.00166	0.00064
016	0.39630	-0.26675	0.38916	0.00030	0.00122	0.00050
C17	0.41454	0.14853	0• 33 <b>3</b> 27	0.00036	0.00157	0.00060
<b>C</b> 18	0.38742	0.15183	0.21828	0.00043	0.00177	0.00070
C19	0•38852	-0.02767	0.16578	0.00066	0.00256	0 <u>.00079</u>
<b>C2</b> 0	0.36315	-0.03048	0.06097	0.00089	0.00299	0.00085

Table 2 continued

	X	Y	Z	σ	$\sigma_y$	σz
<b>C</b> 21	0.33591	0.15432	0.01082	0 <b>.</b> 0 <b>0090</b>	0.00327	0.00096
C22	0.33318	0.33570	0.06531	0.00077	0.00299	0.00100
C23	0.36263	0.33517	0.16991	0.00064	0.00219	0.00090

Table 3. Anisotropic temperature factor coefficients for 6-NBF-PA.

	β"	Brz	β33	B23	ß13	BIZ
Nl	0.00076	0.02892	0.00363	-0100141	0.00102	0.00000
<b>C</b> 2	0.00126	0.01857	0.00476	-0.00145	0.00197	-0.00075
C3	0.00141	0.02953	0.00482	-0.00397	0.00060	0.00063
<b>S</b> 4	0.00172	0.03483	0.00469	-0.00725	0.00226	-0.00134
C5	0.00152	0.02347	0.00402	0.00054	0.00170	-0100120
<u></u> C6	0.00120	0.02536	0.00431	-0.00098	0.00205	-0.00169
C7	0.00098	0.02762	0.00436	-0.00276	0.00128	-0100011
<b>08</b>	0.00184	0.03298	0.00399	-0.00110	0.00168	0.00392
C9	0.00359	0.02473	0.00924	-0.01337	0.00471	-0.00704
<b>C1</b> 0	0.00269	0.04859	0.00430	-0.00059	-0.00052	0.00300
C11	0.00106	0.02594	0.00489.	-0.00133	0.00132	-0.00227
012	0.00162	0.02382	0.01028	-0.00654	0.00458	-0.00344
013	0.00176	0.02437	0.00946	-0.00154	0.00332	0.00069
N14	0.00124	0.01738	0 <b>.004</b> 75	-0.00050	0.00186	-0.00105
C15	0.00172	0.02048	0.00591	0.00442	0.00555	-0.00121
016	0.00218	0.02204	0.00804	0.00320	0.00501	0.00384
C17	0.00130	0.02296	0.00553	-0.00036	0.00255	-0.00156
C18	0.00217	0.02514	0.00696	0.00902	0.00442	0.00236
C19	0.00529	0.03373	0.00658	0.00307	0.00461	0.00438
<b>C</b> 20	0.00867	0.04386	0 <b>.00</b> 607	0.00664	0.007 <u>3</u> 1	-0.00130

Table 3 continued

	β"	\$m	B33	\$23	B13	B12
<b>C</b> 21	0.00713	0.06227	0.00677	0.00839	0.00559	-0.00320
C22	0.00495	0.06718	0.00829	0.02674	0.00268	0.00124
C2 3	0.00408	0.03213	0.00831	0.01258	0.00387	0.00236

All temperature factors in this Table are positivedefinite. Table 4. 6-NBF-PA observed and calculated structure factors

( x 100) in the order

\* h k

 $|\mathbf{F}_0| |\mathbf{F}_c| \ll (radians)$ 

Unobserved reflexions are marked + , and the reflexions

not inluded are :- .

as they were all obscured by the backstop.

•	0	0		•				10	. 1981	1809	3.142
3	3318	3263	0.000					11	131	62	3.142 🕇
4	4295	4473	3.142	•	1	3		12	1088	653	0,000
5	6885	7072	C.000	-16	66	192	1.980 T	13	348	. 303	0.000
6	1524	1411	0.000	-15	335	430	-0.282	14	322	297	3.142
7	4924	4669	3.142	-14	348	337	-3.012	15	834	843	3.14Z
8	1602	1455	0.000	-13	147	126	2.536 1	16	360	376	0.000
9	114	130	. 0.000 +	-12	869	920	2.686	•	2	2	
10	696	557	3.142	-11	329	323	+2.119	-16	534	. 681 .	-0.396
11	809	484	0.000	-10	689	585	+0.506	-15	218	248	2.865
12	1548	1446	3.142	- 9	726	729	-0.382	-14	354	317	-2.740
13	1172	1077	0.000	- 6	70.	662	-2.740				1.172 +
14	1113	909	0.000	- 7	505	601	-1.609	-12	<b>517</b>	\$27	1 854
15	114	114	3.142 +	-6	1350	1348	1.141	-14	1761	1010	-0 703
16	492	475	3.142	- 5	2001	1744	+0 701	-10	748	782	-1 344
17	66	40	3.142 1		445	344	-1 607	- 14	1403	1741	-1,270
	ñ	2			441	4043	-2.478		1442	1303	-2.92/
	3403	4340	1.717	-2	100	1040	-0.141		010		1.004
í	3168	3777	-0.074		1241	1//0	1 544		1630	1470	0.519
-	1422	1443	3.417	Š	1107	7004	-7 047		1037	14/9 .	-0.740
÷.	2457	2687	2.013	<b>4</b>	1449	3004	-3.007	-7	680	/01	-3.089
4	4850	1404	2.900		1000	1/16	-0.029		2280	2/83	-2.030
	1004	1051	-0.301	2	1320	2037	-0.274	- 3	2304	2202	V.851
. 7	100	894	-0.496	. 7	961	896	1.871	-2	617	587	-0.208
2	1020	1483	1.611	•	1308	1390	-3.039	-1	1790	2139	1.694
1	1507	1862	-2.912	?	915	921	3,118	D	588	667	-1.789
	995	1067	2.404	8	1153	1238	0.045	1	2402	3052	2.611
	1347	1412	0.668	9	1323	1417	0.217	2	4246	4184	-0.913
10	823	7.1	-1.129	10	735	700	-1.111	3	2509	2789	•0.288
11	1020	1070	2.319	11	775	683	2.903	4	1056	1054	-2.027
12	410	411	-1.637	12	400	469	-0,255	5	1606	1992	2.817
13	93	74	1.298 🕇	13	301	290	-1.206	6	975	1019	2.475
14	341	299	0.607	141	114	308	0.647 +	7	1682	1789	-0.576
15	271	328	0.161	• 15	93	170	2.330 1		1123	1248	•0.635
16	66	333	-2.913 4	•	1	5			813	823	3.064
•	ā	4		-13	348	456	-2 807	10	1240	1455	2 802
C	1136	1019	-1.347	-12	354	439	2.229	11	274	362	-0 215
1	092	989	0.096	-11	540	575	0 704	12	474	724	0.041
2	1841	1734	2.044	-10	E 1 2	404	-1 203		255	214	-1 510
3 -	623	575	-1 104	- 10	313		1 178 4	13	277	214	-1,934
Ă	1027	004	-6 544		500	574	-2 704	1.	340	300	2.040
	1022	0.77	-0.174	- 0	399	248	-2.790	19	414		-2.07/
	1955	927	-0.134	-/	410	342	-2.720	10		133	-V.203 T
•	1910	981	2.708	-0	469	342	0.425	•	2	4	
	7/5	280	-2-913	-7	1221	120/	-0.432	-17	308	372	-0.520
	308	195	-1.782		864	/09	2.068	-14	451	541	-3.053
	400	320	-1.032	-3	554	684	2.205	13	378	365	1.648
10	573	528	0.300	- 2	714	707	-1.977	-12	474	433	-0.794
11	513	446	2.964	· <b>-1</b>	955	906	-0.569	-11	-542	509	0.170
12	609	615	2.855	0	879	1014	0.546	-10	699	673	0.715
13	237	243	-1.469	1	1600	1605	2.636	-9	769	711	-2.929
14	465	518	-0.497	2	708	756	2.066	÷.	810	789	5.113
•	0	6		3	752	636	-0.113	- 7	732	727	0.424
5	613	583	0.996	4	839	697	-1.037	- 6	617	514	-0.695
	0	8		5	335	360	1.068	-5	606	496	1.281
0	501	324	-1.227	6	995	856	2.271	14	263	291	2.665
	1	1		. 7	754	621	-3.055	- 3	1855	1954	3.003
-16	174	153	-1.683	, i	1115	1121	-0.496	- 2	1055	2053	0.001
-15	383	315	-0.203	ě	114		-0.253 +			628	+0.ALL
-14	758	A1 5	0.651	10	444	377	1 284	- 1	1.1.4	1174	2 313
-13	1314	1340	-2.411	11	402	443	-2 504	¥	14-4	1 1 J / D	E. 636
-12	869	787	1.047	40	340	260	-2.297	1	1911	1000	
-11	284	261	1.241	12	440	644	-0.411	ć	+207	1076	-0.916
-1.	1084	114.	-0.740	10	207		-0.433	· 3	1140	10//	U.OUZ
-0	622	441	-0.700		, e			. 1	1049	802	-0.221
- 4	1257	1204		-17	00	24	U.VUUT		1547	1540	3.046
- 7	1271	11 4		-18	93	140	3.142 T	6	905	810	-3.018
	1500	11:4 74	2.012	-15	041	613	0.000	7	1010	920	-0.632
- 6	1741	14 4	0.386	-14	037	205	3,142	•	918	913	0.404
	1144	19.4	-1.035	•13	1163	1193	3.142	9	546	502	1.075
	2907	ee28	-2-170	-12	1867	1797	0.000	10	970	1003	-2.756
- 3	4177	3475	2.167	-11	279	199	3.142	11	114	245	5.0214
-2	2957	2707	-2.716	-10	186	194	0.000	12	526	592	-1.654
5	1313	1407	1.30	-9	301	313	. 0.000	13	228	505	•0.296
5	1037	997	1.151	-8	2186	1856	3.142	14	451	599	2.300
4	4591	5538	-1-255	-7	197	230	0.000		2	6	
5	2355	2364	1.415	-6	620	556	0.000	0	647	569	3.067
6	2938	2973	-3.163	-5	732	712	3.142		2		
7	522	553	-2. 26	-4	5065	4696	3.142	۵	378	198	2.014
8	1426	1235	C.63C	-3	36+3	3089	3.142	•	- 'i		
9	1569	1644		ŏ	4330	3833	3.142	- 1 7	141		
1:	1136	1170	3. 40	X	6347	6822	0.000	-14	104	104	0.114
11	920	873	2.2.8		3504	3604	3.142	-14	241	304	2 864
12	854	848	-1.4.	ě	286	241	3.142	-17	203	804	-3 714
13	961	1926	1.654	2	2140	1941	0 004	-14	- 978	1 1 2 4	-2./14
14	3 8 3	335	-1.202	7	944	78	0.0004	-13	1600	441	-3 474 4
15	474	40	-2.800	<u> </u>	3880	2144	0.000 F	-14		144	-6.0/4 ]
10	394	421	3.114		1660	4 143	3.142	•11	492		-2.007

. .

-9	1306	1271	2.006	11	383	469	-0.073	-1	1135	1394	2.212
-8	1526	1585	-0.934	12	389	479 .	-0.944	i ū	478	220	-1.506
-7	2228	19.2	0.343	•	4	0		1	1514	1556	-0,220
-0	689	054	2.026	-17	93	233	0.000 +	2	791	781	2.911
-4	1728	1407	-2.160	-10	335	320	3,142		660	527	0.307
-3	3203	3245	-0.077	-14	950	865	0.000		0/2	1 27	-0.167 +
-2	3012	3653	0.392	-13	131	51	3,142 +		964	951	-0.396
-1	1194	1837	1.184	-12	1052	935	3.142	7	271	274	-2.988
C	2370	3016	1.388	-11	197	129	3,142		431 -	374	1.938
1 2	2277	2849	2+615	-10	1172	973	3.142		372	307	3.021
3	712	3093 R)1	-0.831		2010	2323	-0.000	10	509	. 505	-0.046
4	2343	2311	-3.036	-7	561	745	- U. UUU G. OAA	12	200	416	1 005
5	2261	2346	1.431	-6	904	746	0.000	13	400	488	2.849
6	1078	1051	0.586	-5	5537	4891	3,142		4	6	
7	1782	18:6	-1.303	-4	4632	4339	0,000	8	147	86	-0.484
8	1449	1308	1.184	-3	2826	2813	0.000	•	5	1	
10	3222	2154	-2+797	-1	2665	2690	3,142	-17	329	401	-2.974
11	1230	1231	*1+612	4	7494	20/2	3,142	-10	271	216	-2.007
12	562	598	0.139	2	1989	1973	0.000	-12	1052	200	0 428
13	426	455	-3.037	3	5490	4831	3.142	-13	802	836	-1.839
14	383	475	2.534		1276	1085	0,000	-12	657	629	2.507
15	301	329	-1.553	5.	1298	1133	3.142	-11	813	778	-2.487
16	378	455	0.236	6	306	223	3,142	-10	1086	1167	-0.020
	3	3	A 444 4	7	1581	1331	0.000	-9	644	928	2.032
-16	93 500	120	U.190 T		2079	1928	3,142	- 1	294	236	-2.904
-14	416	436	4.004	10	1084	. 207	3.142	-/	2164	2003	-2.819
-13	147	119	2.264 +	11	1366	1364	n 000	-6	2440	2705	-0 481
-12	904	953	-0.571	12	986	943	3.142	-4	2616	2377	2.074
-11	904	937	1.541	13	513	470	3,142	-3	1674	1650	2.888
-10	1555	1592	2.996	14	114	109	3.142 +	- 2	5458	5832	+1.601
-9	1274	1277	2.679	15	322	339	0.000	-1	1924	1861	1.803
-8	1003	977	0.155	16	66	13	0,000 T	0	3602	3818	-0.105
	877	844	-1.079		4			1	1222	1284	-0.421
-5	1960	1727	-3.123	-14	405	434	-3 110	2	3/9/	4190	-2.993
-4	1422	1262	-2.553	-15	228	206	2.481	4	821	#16	1.011
- 3	2478	1841	0.101	-14	93	197	1.376 +	Ś	1987	1604	0.235
-2	2408	1996	-1.428	-13	577	566	0,222	6	1080	972	0.549
-1	1872	1738	1.738	-12	93	204	-2.124 +	7	1963	1985	3.099
0	1843	1728	2.357	-11	613	964	2.768	6	1007	1002	-2.171
1	3610	2397	-1.188	-10	197	182	2.634	9	1197	1123	0.603
2 1	544	26/3	0.004	- 9	1041	107/	-0.035	10	341	276	-0.08/
Ă	1987	2042	-3.100	-0	1347	1309	2 010	14	294	200	-2 847
5	1165	1240	2.399	-6	1410	1523	-2.625	13	147	168	0.086
6	988	976	-0.835	-5	970	949	-0.405	14	304	415	0.126
7	823	916	-0.313	-4	1872	1949	-0.431	15	66	152	-1.510 +
8	1504	1585	2.397	-3	2547	2681	-1.125	•	5	3	
10	1170	1401	-2.851	-2	3291	3992	2.161	-16	366	343	1.854
14	329	389	1.401	-1	2469	2987	-2,242	-15	315	271	-0.140
12	147	91/ 05	-1.061.4	1	1891	2047	1.020	-13	360	337	-1.088
13	400	484	1.817	2	1566	1697	0.006	-12	905	743	-2.453
14	372	536	3.112	3	1714	1853	-2.817	-11	1094	989	-0.188
15	301	401	-0.929	4	1570	1390	2.747	-10	961	958	0.670
•	. 3	5		5	1533	1653	-0.136	-9	746	661	-1.129
-14	66	179	-2.985 +	6	1123	1159	0.167	-8	1261	1193	2.128
-13	131	123	0.488	7	647	611	1.890	-7	1062	1015	-2.688
-11	400	462	-0.98	8	1078	1155	-2,127		772	758	0.499
-10	421	325	1.001	10	704	1114	-1.731	- 2	1773	1591	0.405
-ġ	114	88	-8.964 <del>+</del>	11	405	470	1.424	-3	2130	2284	-1.249
-8	546	5.6	1.777	12	366	361	-2.790	-0	1334	. 1486	2.264
-7.	546	478	-0.812	13	410	483	2.706	-1	1047	998	-1.996
-6	641	649	-1.314	14	426	473	-1.576	0	1131	1231	0.036
-5	1217	1046	2.580	15	410	485	-0.114	1	1623	1484	-0.514
-4	517 110	379	-0-515		4	4		2	823	675	-0.162
-3	834	920	1.285	-15	206	262	2,566	3	1615	1617	2.728
-1	1443	1429	-1.754	-14	203	21/	-V.104 .0 772	4	1140	1138	-1.091
Ē	973	1064	-3.055	-12	63	144	-2.505 4	7	1204	1203 524	U.004
1	1010	739	C.428	-11	542	476	-2.838	7	22Å-	247	0.345
ş	218	168	-0.427	-10	821	684	0.029	É	228	151	-2.158
3	538	374	0-519	-9	660	508	-0.177	9	329	310	-1.659
4	573	515	-2.377	-8	218	216	-3.138	10	405	370	-0.141
2	370 312	771	2.810	-7	474	417	-1.229	11	630	773	2.612
7	623	200 541	-1.766	- 6	928	828	2.749	12	206	307	2.978
é	651	588	1.740	-7	013 461	8 V V 8 7 1	0.014	13	114	103	-2.117 1
9	174	188	2.993	-3	492	482	-0.399	17,		370	-0.770
10	360	296	-2.954	-2	769	870	-3,021	-14	218	200	-9.374

- 17	047			_				· · _			
-19	203	201	3.108	9	925	872	-0.497	-3	641	579	0.455
+12	218	207	-2.329	10	558	603	1.727	2	1072	1048	0.963
-11	465	378	1.307	11	542	552	-3.013	-1	1074	1009	-1.488
-10	301	224	-1.560	12	93	114	-1,548 🕈	0	1519	1360	2.626
-9	627	564	-0.651	13	348	365	0.033	i	402	437	-2.038
~8	752	639	2.754	14	03	20.0	-0.0A1 🕈		728	444	-1 344
• 7	526	540	1.701		,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,		***** *		127	990	-1.044
- 6	380	300	-0 784	-18			-0.047	· •	1992	1934	0,452
		340	-0.701	-15	217	000	-0.94/	4	620	587	2.946
	<b>y</b> 3	239	2.626 +	-14	322	285	2,286	5	1233	1334	-2.378
-4	671	591	0.475	-13	554	609	2.802	6	1076	1113	1.507
-3	538	5u4	-2.198	-12	114	133	2.818 +	7	724	492	-0 680
-2	1561	1602	2.411	-11	414	537	0 035	<u> </u>	729	1 4 4 1 1	-0.007
-1	1086	1204	-0 207	-14	0.34		-1 700		700	1019	-0.338
5	550	1290	-0.207	-10	A5A	DK4	-1./99	9	329	336	1.130
ž	370	JOE	-1-128	-9	394	393	-2.538	10	637	688	-3.118
1	422	412	2.485	-8	657	658	2.305	11	131	162	-0.580 <b>+</b>
2	580	579	-0.843	-7	389	421	-1,944	12	-114	230	0.490 +
. 3	1022	986	2.861	-6	943	933	0.040	13	344	170	-0 743
4	802	812	0.160	- 6	318	244	3 101				
5	651	637	-1.321		1 8 7 2	1 24 4	-4 550				
Å	. 244	4 74	2 204		1072	1210	-1.997	-1-	222	678	2.423
	240	130	2.201	-3	1252	1276	2.341	-13	93	79	0.195 🕈
	301	185	-3.102	-2	961	902	-0.237	-12	627	533	0.039
8	294	292	-3.054	-1	1217	1328	-1.646	-11	114	98	+0.707 +
9	114	13	-0.069 🔶	0	1248	1412	1.542	-10	244	281	-1 807
10	335	356	0.357	i	720	633	-2 874	- 0	744	4.34	-1.00/
11	186	245	2.450		403		-2 477		/	051	2.010
12	66	2.14	-1 004 4		78/		-2.133	- 5	114	37	1.256 +
**_		200	-3.001 T	3	<b>400</b>	391	0,266	-7	1005	955	-0.761
<b>*</b>		U		<b>4</b> -	405	289	-1,150	-6	735	736	1.074
-1/	2/9	325	3.142	. 5	586	568	3.138	-5	789	775	2.825
-16	591	567	0.000	. 6	271	360	0.820	-4	588	505	-1.810
-15	859	746	0.000		448	410	2 887	- 7	200	48	
-14	354	250	0.000		405	40.0	-1 400	- 3	43		U. 974 T
-13	1434	1244	7 140		202	700	-1,420	-2	884	810	+1.119
-10	1404	4614	-3-1-2	9	/38	/69	. 0.003	-1	1113	1108	1.287
-12	021	523	3.142	10	573	627	-3,059	Ç	1153	1157	2.779
-11	1701	1538	0.000	11	93	110	-2,775 🕈	1	93	178	-1.410 +
-10	530	412	3.142	12	263	221	-3.084	2	1074	0.90	-0.159
-9	952	9.5	3.142	13	271	316	-0.370	i	448	102	-2 524
- 8	279	<b>a</b> n	3.149						.705	372	
- 7	180	7.4	3 140						238	409	2.156
	007	3.0	3.1.72	U	40.5	382	2,192		550	445	0.725
-0	2800	2906	0.000	+	7	1		6	925	858	-2.905
-2	218	276	0.000	-17	474	513	0.387	7	891	071	-0.215
-4	3922	3683	3.142	-16	360	307	-0.520		237	260	0.483
-3	3744	3657	0.000	-15	349	333	1 375	Š	304	470	9 414
-1	8206	9714	0.000	-14	0.40	000	-2 775		394		C.000
, F	3016	2575	3 140	·	999	922	-<.//2	10	294	315	+1.525
	2700	2535	3.1.72	-13	041	633	2.930	11	228	240	-3.004
1	4209	3948	3.142	-12	975	996	-0.204		8	· 0	
2	2217	1910	0.000	-11	1694	1763	-0.073	-17	469	449	0.600
· 4	1369	1085	0.000	-10	1852	1774	-2.590	-16	114		0 000 🕈
5	1784	1622	0.000		2141	50.84	1 167		222		7 4 4 2
Ä	2882	2714	7 140		100	2004	-1 8-4	-15	000	204	3.142
š	2002	¢/30	0.172	-8	329	423	-1.220	-14	1001	907	3.142
	000	07/	3.142	-7	1212	1344	-1,291	-13	1183	1062	0.000
	1027	808	0.000	-6	2276	1568	0.720	-12	1226	1147	U.000
9	577	577	3.142	-5	2924	2760	-2.633	-11	1497	1255	3.142
10	1106	1110	0.000	-4	2006	2045	2.545	-10	120	230	1 142
11	995	894	3.142	- 3	251.	2400	0 404		430	EAE	1 4 4 2
12	341	314	0.000	- 2	505	6.78	-1 212	-7	0/0	545	3.142
13	645	5.1	0.000		2993	020	-1.616	*0.	101	19	3.142
	202	941	0.000	-1	2364	2022	3.099	-7	1256	1214	0.000
	75	112	3-142 4	Ū	1519	1501	-2.380	-6	1984	2837	3.142
15	3/2	413	3.142	1	2391	2235	-2,205		32A1	3222	3.142
. +	6	2		ź	2493	2203	0.827	-4	2245	2404	0.000
-17	66	226	2.248 +	Ĵ	2276	1486	-0.940	-3	204	304	8 1 4 2
-16	348	397	0.813	Ă	01.8	1024	1.847		6-1E	200	1 449
-15	271	244	-0.850	2	710	1064	2.07/	- 2	. 472		3.142
-14	444	448	0 044		1339	1391	E.049	-1	3/65	3723	3,142
	440	717	0.904	Ó	1912	1588	-1.300	0	800	552	0.000
-13	077	6 ; C	-2.555	7	562	515	0.638	1	1319	1092	0.000
+12	641	749	-2.869	8	372	372	-0.728	2	2414	1741	0.000
-11	1219	1143	1.)24		237	134	-2.782	1	1010	1741	1 449
-10	501	470	-0.200	10	044	1027	2 114	3	**10	/71	0.19E
- 9	816	711	2.474	. 44	700		E. 1 4 4	1	114		0.000 T
- 0	1144	11.1	E	11	/02	/37	-1.021	_ <b>5</b>	834	753	U.QOQ '
	1100	TTOU	2.120	12	595	683	0,405	6	366	288 -	0,000
-7	629	832	-1.806	13	93	110	-3,080 🕈	ž	123A	1022	0.000
-6	1084	965	0.288	14	474 -	537	-2.244	Å	1165	1050	3.142
-5	1547	1567	0.72Å		7	3		с С	140	478	8 142
-4	2064	235A	-2.711	-14	<b>F</b> 4 4	414	-0 0=0		300	1/0	3.176
•3	1967	21 97	- 3=-	-15	270	030	-0.780	10	206	130	3.142
• 2	1246	1330	1.377	-15	416	319	2.210	11	818	690	0.000
	1677	1270	-0.932	-14	366	352	2.887	12	114	68	3.142 🕈
-1	1138	1096	2.555	-13	711	681	-2.949	13	348	361	3.142
r.	2726	2714	-0.504	-12	660	462	0.430		104	441	1 442
1	1824	1811	3.330	_11	447	E 74	-D #43	14	344		3.T45
2	2495	2100	1.107		31/	277	-0.002	•			
ā	1740	17	4 6 14	-10	/91	01/	-0.901	-17	441	535	-0.520
	1, 77	-/00	1.31	-9	1435	1370	2,915	-16	66	39	-2.521 +
2	A09	975	-0.525	-8	465	394	-0.794	-15	378	322	3.039
2	789	8 8	1.846	-7	1533	1481	0.439	-14	0.3	40	2.770 +
6	1222	1340	2.905	-6	QR4	1010	-1.526	_ 17	£74	443	0 005
7	664	662	-2.314		Ros	402	+0.554	-10	739	107	0,007
8	1106	1107	1.084		447	712	2 7 1 2	-12	1403	+24/	-0./73

-10	1471	1				_						
- 0	344	1022	2.643		12	383	418	-3,059	6	1024	1006	0.000
- 4	2020	2044	2.035		13	246	260	-2.784	9	948	996	0.000
.7	2029	2039	-0.103		14	11 0	118	-0.270-4	10	197	123	0.000
- 4	1409	1912	-1,740		•	9	. 3		11	673	494	3 142
- 6	1009	1093.	2.396		-16	534	60 <b>B</b>	3.022	12	348	250	3.142
- 4	000	/50	-2.018		-15	114	135	-2.618 🕇	13	66	127	3 142 +
- 1	977	979	2.539		-14	620	614	0.846		10	;	0.144
-0	102/	2228	0.089		-13	834	905	-0.867	•17	255	258	-2 744
-2	766	525	0.206		-12	1066	956	1.954	-14	403	2/0	-2,/44
	1533	1550	3.094		-11	1261	1265	2.958	-18	772	202	2.9/8
Q	2117	1697	0.563		-10	1368	1230	-2.582	-19	110	3/4	0.489
1	2939	2836	-1.539		- 0	2120	2218	- 0 4 4 4	-14	778	844	-0.579
2	565	493	1.531		- 6	746	440	-1 742	-13	550	485	2,666
3	1420	1483	0.252		-7	1217	1288	-1./76	-12	271	297	2.216
4	2352	2516	3.041		- 4	1418	1270	- 2.012	•11	509	459	3.121
5	474	410	0.117			1010	1200	1,87/	-10	975	915	-0.070
6	1699	1942	+0.570		-4	2110	1/3/	-2.198	- 9	1631	1745	-0,43
7	1104	1125	-2.802		_	2139	2139	0.075	-1	2425	2565	-3.072
ė	93	147	0 700 -		- 4	172	731	1.682	-7	378	425	3.118
ē	1003	1008	2.203		- <b>X</b>	1793	1773	-2,567	-6	1473	1499	-0.111
10	478	520	2.203		-1	955	786	1,336	-5	301	221	-2.034
11	754	227	-2.233		0	1659	1580	-0.508	-4	1276	1438	-0.718
12	170	633	0.495		1	263	180	3.123	+3	2316	2705	2 811
12	1/	208	-1.974		2	939	840	1.683	-2	2218	2800	2,300
13	00	142	-2.961 +		3	1779	1607	3.088	-1	2654	2405	-2.010
*	8	4			4	605	825	-1.764	- 1	6 7 7 4	2493	0.143
-15	513	543	2.894		5	1631	1697	0 438		/26	647	1.099
-14	93	116	1.454 🕈	•	6	400	443		1	749	712	~1.489
-13	761	733	-0.164		7	0.05	4 1 4 8	7	2	1037	956	2.561
-12	588	527	0.084			405	1010	3.900	3	1413	1371	2.589
-11	588	448	1.532			405	394	2.138	4	2231	- 2442 .	-0.553
-10	134A	1386	-2.904			17/	200	*1.734 T	51	831	885	0.977
-9	279	238	2.277		10	301	318	-0.090	<b>6</b>	821	815	2.873
-8	1276	1428	E·2//		11	114	230	1,016 🜪	7	943	1027	-1.519
-7	1000	1044	4 344		12	161	256	-3.073	6	. 346	418	1.287
- 4	1070	1044	-1-345		•	9	5		9	560	564	-0.403
-6	960	842	1.926	-	14	66	26	-1.509 🕈	10	341	144	3 047
	11-8	1255	2.731	-	13	667	577	0.092	11	246	240	-2 438
	10/0	1213	-2.134	-	12	623	586	-2.273	12	44	126	-2.407
- 3	2338	2545	0.378	-	11	735	647	2.643			163	-2.993 T
-2	906	629	-2.791	-	10	654	513	0 102		10		
-1	864	892	-1.885		- 0	114	7.8	7 489 4	-15	280	250	1.074
ŋ	1111	1079	2.964		- 6	800	410	2.4021	-14	372.	332	-0.277
1	1266	1217	-0.640		-7	470	434	-0./0/	-13	569	637	-1.481
2	854	858	-0.152	•	-4	1447	120	-0.561	-12	752	- 637	2.549
3	1564	1615	2.164			1017	1942	2.667	-11	509	438	-0.933
Ā	1475	1812	-4 434		~?	966	920	-0.378	-10	677	641	-0.139
5	270	107	-1.027			1144	1109	-0.287	-9	699	666	1.901
	600	183	6.125		-3	896	768	2.777	-8	1226	1236	-1.032
?	877	858	1.192		-2	1121	1001	-1.265	- 7	1463	1423	2 6 16
	1027	989	-0.721		-1	1031	966	2.890	- 6	487	416	51210
	849	1 i Q	2.158	•	0	920	855	1.448		1401	1010	1.001
. 9	627	640	-2.886		1	1772	1866	-0 780		1081	1914	-0.615
10	538	535	-0.763	•	2	1102	40.85	0.700		975	945	2.285
11	206	271	1.618		ā	671	. 471	-7 470	- 3	1076	1100	2.690
12	308	410	-0.250		Ă	910	750	-3.139	-2	383	339	-2.740
	8	6			2	744	750	-1.470	-1	602	474	1.446
0	341	443	-2.621			/17	/1/	£.438	0	2064	2184	0.374
	õ	1	21013		2	/94	810	-0.198	1	961	1030	-3.102
-17	246	980	-4 000	·	1	394	341	2.183	2	1612	988	2.086
-16	320	3.44	-1.545		6	165	506	2.703	3	877	772	-1.742
-15	654	341	2.3/4		9	360	413	-1.071	i i	329	254	0.565
-14	229	0.00	2.077		10	66	265	-0.350 🕈	, i	1048	1137	0 816
_17	000	02/	0.195		٠	10	O			464	441	-2 744
-12	240	154	-1.033	-	17	599	591	3.142	, i	501	481	-2.704
-15	341	264	2.674	-	16	329	208	3.142	<u> </u>	201	. 451	E.23/
-11	984	970	-3.124	•	15	263	244	0.000	, in the second s	<00	210	1.529
-10	1360	1399	1.804	-	14	131	123	0 000 +		228	212	-0.634
-9	2657	2485	-0.864		13	5.1	461	0.000 T	10	66	175	0,494 T
-8	2107	1960	B.162	_	12	1144	495	0.000	11	322	392	2.098
-7	2705	26:14	2.714			401	1010	0.142	•	10	6	
-6	1348	1118	-1.615				434	3.142	Û	526	315	-0.745
~ 5	1799	16. 2	4.043	•		1/76	1590	0.000	•	11	1	
-4	2020	2447	-0.770		- 9	1281	1177	0.000	-17	279	262	2.811
- 3	3784	3004	-0.070		- 5	562	538	3.142	-16	204	222	0.362
-2	4864	4766	0.094	•	•7	1868	1642	3,142	-15	600	724	-0.217
	11.40	4349	2.687	•	- 6	1720	1505	3,142	-14	304	414	-1 444
-1	1142	1071	2.653		- 5	4202	4194	0.000	-4.8	594	~ 4 ~	1.000
L'	1306	914	-0.289		-4	1648	2182	0.000	-10	334	770	6./72
1	1212	899	1.597		- 3	1762	2522	3 142	-12	1108	1057	-2.818
5	1394	1293	-1.348		-2	4484	3854	3 4 4 3	-11	1288	1381	0.804
3	2312	2211	1.865		-1	444	434	J.176	-10	546	516	-0.620
4	1360	13.6	2.674			1462	720	0.000	-9	741	766	-2.576
5	1548	2024	-0.4R			1002	1493	0.000	- 6	1475	1588	-3.048
6	389	321	2.21		-	114	242	0.000 9	- 7	766	673	-2.625
7	372	304	C. CT ]		4	2786	2424	3.142	-6	1566	1765	0.247
ė	565	4.0	C· 775		3	711	627	0.000	-5	662	672	0.044
ē	204	107	-2.289		4	410	• 375	3.142	-4	1773	1842	-2.424
1		222	0.619		5	660	399	0.008		2176	2221	-2 411
11	7.75	/	-0.001		6.	341	304	0.000	-9	1044	1443	- 2123
41	£10	128	0.577		7	2352	2138	3.142		5 y 4 4	2024	0./03
										E J V V	CZ JI	e II . 001

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	044		· · · ·	-							
	7472	. 857	0.068	2	1895	1599	0.000	- 3	255	337	0.219
2	4116	3295	-3.048	3	957	901	0.000	-2	509	620	-2.359
ж х	1428	992	0 • 764	4	197	124	3,142	-1	680	453	-1.911
Å	467	1191	-0.086	5	1759	1593	3.142	0	1552	1603	-3.039
š	286	284	-0.4/1	6	657	555	3.142	1	2371	2054	0.501
,	1972	200	1.017	7	1734	1609	0.000	2	1391	1303	-1.059
7	114	1041	-3.063		131	54	3,142 +	3	613	588	2.382
Á	1554	1414	2.503 +	9	131	27	3.142 +	4	966	893	-2.906
ŏ	228	263	0.2//	10	764	770	3,142	5	950	806	-1.930
10	301	274	-3.113	11	263	280	0.000	6	1052	994	0.252
11	263	247	-2.000	12	228	305	0.000	7	580	558	-0.009
12	64	247	-2.704	•	_ 12	2		8	634	624	2.780
· · · .	11	11/	-2-239 T	-17	360	397	1.320	9	546	598	3.096
-16	115	170	0 430	-18	421	510	-0.222	10	93	176	-0.991 +
-15	431	420	0.470	-12	329	334	-2.181	11	354	430	+0.0A3
-14	402	440	-0.229	-14	558	520	2.822	•	13	3	
.13	417		-1.23/	-13	664	613	-2,621	-16	66	37	-2.394 +
-12	4047	284	2.280	-12	1066	979	1,551	-15	03	176	-2.000 +
- 1 4	417	740	-1.213	-11	1513	1540	-0.142	-14	591	592	2.761
-15	01/ 813	249	1.248	-10	752	698	-2,497	-13	465	415	-0.287
-0	480	021	0+102	-9	1257	1391	2.603	-12	496	441	-0.638
- 9	780	061	-2.643	-8	714	734	-0.619	-11	400	371	-0.610
- 7	464	874	2.705	-7	315	266	0.635	-10	1119	1107	3.009
	871	279	0.099	-6	1642	1893	Q.697	-9	161	71	2 572 +
-0	171	450	-0.214	-5	1425	1532	-2.404	- 8	1183	1183	0.220
	131	78	0 - 259 🕈	-4	1368	1542	2.267	-7	1022	962	-1.486
	1237	1481	2.072	- 3	941	1037	0.060	-6	147	98	1.586 +
- 3	744	427	2.864	-2	692	668	-1,279	-5	1403	1370	2.378
	740	/58	-1.705	-1	1016	991	-0.397	-4	580	489	-1.920
- <b>1</b>	210/	2085	-0.196	0	1981	2019	2,917	- 3	1276	1326	0.051
	200	499	1.963	1	1651	1684	3.107	-2	1201	1114	-0.424
	007	682	-3.070	2	2166	2231	0.360	-1	2077	1833	2 847
-	973	050	2.887	3	896	849	-0.684	ō	469	345	+3.080
3	1049	1025	-0.468	4	884	860	-3.025	Í	228	178	-0.003
	1001	988	0.359	5	927	855	2.852	2	1464	1314	0.003
2	262	479	-3.018	6	538	495	-2.526	3	161	121	-2 00o +
2	23/	234	2.633	7	1305	1334	0,508	4	1104	1135	-3.122
	329	328	-1.521	8	416	424	-0.069	5	320	300	1 441
	303	290	0.366	9	692	784	2,858	6	80.8	787	-0 712
4 0	131	82	0+294 🕈	10	174	178	-1.898	7	637	649	0 243
14	- 00	433	2.860	11	66	53	-0.309 4	à	431	539	2.840
- <b></b> -	90	151	2.379	•	12	4	-		372	460	2 775
-13	770	, , ,		-15	147	185	-0.838	10	301	246	-0.575
-12	575	369	3.038	-14	431	561	2.471		13	5	••••
.11	497	20/	1.981	-13	426	418	2,602	-13	66	453	-0.217 +
-15	407	428	-1.093	-12	360	402	-1.080	-12	66 -	11	0.542 +
-0	114	1/3	-0.001 T	-11	93	93	2.706 🕇	-11	93	73	2.760 +
	440	42	2.201 T	-10	487	458	-0.983	-10	255	204	-2.357
	500	772	2-405	-9	114	236	2.710 🕈	-9	460	388	2.532
- 6	506	399	2.549	-8	1027	1039	2,660	- 8	602	562	-0.219
- 5	140	0.30	-0.528	-7	1417	1444	-0,565	-7	421	408	0.589
-4	644	5:0	-0.304	-6	114	136	2.509 🕈	-6	546	478	3.134
-3	1100	239	-2./74	-5	451	430	-1.762	-5	577	465	-2.043
-2	974	744	2.142	-4	573	511	2.261	-4	492	360	2.167
-1	844	/44 40 P	-0.921	-3	522	463	-3,049	- 3	673	604	-0.475
ā	440	302	-0.381	-2	1337	· 1322	-0.701	-2	639	721	1.308
1	349	392	2.922	-1	1228	1092	1.305	-1	816	719	-2.667
5	744	440	-2.292	, O	1240	1195	3,115	0	114	187	-0.061 +
3	627	444	2.309	1	262	486	-1.480	1	114	118	1.090 +
Ă	764	10	-0.070	2	366	326	1,993	2	478	412	-1.124
5	505	644	1 000	3	623	684	-0.731	3	400	349	1.561
6	436	4.8	1.802	4	279	264	1,144	- <b>4</b>	383	357	2.182
2	348	334		2	906	935	2.252	5	383	408	-2.208
Å	66	140	-132	6	634	633	-1.676	6	538	540	0.088
	12	107	1.00/ 4	7	246	273	-0.488	7	66	122	0.408 +
•17	444	301	A	. 8	501	562	0.123	•	14	Ū Ū	
-16	505	440	0.000		513	557	-2.812	-17	66	106	0.000 +
-15	186	2.8	0.000	10	308	422	2.074	-16	620	626	3.142
-14	55Å	405	3.142	. •	12	6		-15	114	193	3.142 +
-13	505	444	3.145	Ū,	114	237	-2.587	-14	651	730	0.000
-12	913	1612	0.10-	*	13	_ 1		-13	131	46	0.000 +
-11	465	365	0.000	•17	335	209	-0.174	-12	686	693	3.142
-10	131	171	0.000 30.445 ±	-16	66	121	2.680 +	-11	591	573	0.000
- 9	831	7 4	3+1+2 T	-15	720	763	-2.759	-10	1231	1288	3.142
-8	534	4 4 4	3+1+2	-14	348	273	0.812	- 9	286	295	0.000
-7	1300	12 4	3+1+2	-13	372	340	0.052	-8	1313	1413	0.000
- 6	1819	1610	0.000 0.000	-12	1076	1153	-0.189	-7	1503	1497	3.142
-5	1587	1401	U.UUQ	-11	591	590	-2.428	- 6	717	474	3.142
-4	580	44E	3.1.72	-10	1094	1187	2.050	-5	1305	1208	0.000
- 3	831	880	3.142	-9	977	953	-1.873	-4	1467	1270	3.142
-2	2746	27 . 6	C+UUQ	· • •	483	400	0.280	- 3	1131	966	0.000
-1	1000	000	V+290	-7	1306	1455	0.506	-2	1106	1004	3.142
7	1117	040	0.142	6	1443	1511	-2.225	-1	1523	1300	3.142
i	114	149	3.142	-5	1697	1841	2.386	0	1744	1599	0.000
-		40J	3.142 🍟	-4	294	321	0.615	4	474	726	

•.

2	1064	924	3.142		8.87	802	-0 010				- 1 4 10
ī	754	3.1	0.000		007		- 0.717		513	203	-3.132
	324	2/1	0.000	2	/86	776	0.450	 - 8	492	648	.2.379
	1142	960	3.142	6	558	575	1.648	-7	775	. 809	-0.496
5	421	339	0.000	7	623	653	-2.231	- 6	436	396	1.183
6	1648	1466	0.000	8	237	223	-2.755	- 5	436	417	-2.938
7	657	640	3.142	à	782	787	0 465	·	5 7 7	510	-1 464
Å	456	417	1 440	10	1.1	147	0.733		2/3	539	-1.474
Ň		71/	0.142	10	315	902	-1,7/0	- 3.	839	873	•3.110 ·
	114	148	3.142 T	•	· 15	3		2	308	280	-0.114
10	613	722	0.000	-15	416	356	0.564	-1	831	865	-0.023
- 11	246	341	3.142	-14	416	472	0.177	0	538	524	1 473
	14	2		-13	640	442	-1 048		408		- 3 407
-16			-9 477		700	1006	-3,988	+	620	204	-2.497
-15	101	190	-2.033	12	. 301	372	2.849	2	580	503	3.102
-19	14/	<u>5.00</u>	2.851	-11	147	188	2.401 T	3	664	644	-0.292
-14	348	3.7	0.418	10	823	821	-0.212	4	534	534	+D.064
-13	670	690	-0.544	-9	147	139	0.973 1	5	400	415	2 043
-12	421	4.1	-7.403	- 9	44.0	402	-2 754			435	2.003
-11		1344	~2.003	-0	000	OUZ	-2.754	6 1	294	- 318	-2.441
	1010	1291	2.347	•7	469	468	3.097	7	66	96	2.665 T
-10	879	901	-1.662	-6	660	639	1.981		66	125	-1.020 +
-9	831	870	0.061	-5	98.8	1110	-0.495	· ·			
-8	647	645	-0.170	- 4	530	417	-2 384	- 4 7	10		
• 7	824	871	- 2 41 7		266	407	-2.050	-10	00	105	*U.397 T
		10/1	-2.01/	- 3	222	235	1.075	-12	496	446	1.750
-0	11/2	1551	2.549	- 2	657	649	-2.756	-11	569	576	·0.702
- 5	294	288	-0.256	-1	522	480	-0.540	-10	244	206	1 602
-4	913	9.16	-0.808	ñ	804	040	1 110		240		2.076
-3	1642	1710	0 265		077	755			203	142	-2.947
~ 7	074	*/37	0.205	1	013	/22	-1.356	-8	378	365	-2.599
-2	700	751	-2.878	· 2	87,2	865	2.628	-7	421	. 394	1.039
-1	1305	1388	2.870	3	147	. 95	2.664 +	-6	534	528	+0.185
5	1041	1078	-0.492	· · · · ·	70.0	748	-0 400		455	250	
1	766	695	0.442	Ĕ	100	. 70	-0.400		422	372	-2.144
-	550	400	0.442		301	170	0.376	-4	1027	992	2.262
	220	489	0.016	6	441	431	-3.034	- 3	366	293	-0.964
3	1188	1160	2.926	7	354	407	2.667	-2	122	101	-0 543
- 4	599	500	-2.872	Á.			1 448 4		526		-0.940
5	1416	1474	-0 474	· · ·	10		1.400 1	· •	244	514	-0,011
-	440		-0.078		19	2		Q	436	363	-2.637
	442	348	1.497	-12	66	393	2.104 +	1	580	501	2.527
/	487	517	2.666	-11	301	277	0.960	2	263	218	-0.980
8	341	424	2.789	-10	404	520	-1 142				0.505
	114	226	-2.104 +		256	100	1 407		240	524	0.022
1 1	414.	523	0 041		299	300	1.403	4	441	402	+0.287
- C	710	225	0.003	-8	228	207	-1.745	5	237	268	-2.057
•	14	- 4		-7	483	420	-2.757	6	378	429	2 455
-14	580	645	0.176	-6	400	386	1 5.13				
-13	66	03	1.269 1		840	0.00			10		
-12	3.00	2.4	1		049	067	-0.342	U	366	292	1.352
	000	201	3.140	-4	237	191	-2.768		17	. 1	
-11	224	578	2.862	-3	580	493	2.816	-16	465	488	-0.166
-10	93	141	1.694 🕇	-2	410.	308	-1.503	-15	470	471	-1 807
-9	906	830	0.234	-1	426	577	1 400		4/0	-/3	-1.04/
-8	366	2. 9	-1 470		020	277	1,007	-14	778	832	2.362
- 7	400	7.4	-1.4/9	v	952	812	-0.991	-13	505	495	-2.325
	400	394	-2.819	. 1	354	291	1.210	-12	986	09B	0.322
-6	680	620	2.607	2	644	662	3.016	-11	114	110	1 4 1 4 +
-5	738	721	2.727	3	378	374	-0.970		117	147	1.000 1
-4	1644	19-0	-8.280	7	744	2070		-10	1243	1121	-2.236
-1	370	170	-0.207		341	30/	1.06/	- 9	929	667	2.112
ž	372	3/4	2.427	•	542	471	-9.603	- 8	114	109	-0.915 +
	/38	/45	-2.562	•	16	0	•	- 7	0.4	021	-1 497
1	400	318	-0.035	-16	434	410	0.01		1404	1602	-0.007
5	630	611	2.140	-16	0.1	4 80	0.000	-0.	1480	1246	1.205
1	604	743	-0.747	- 4 4		130	0.000 1	- 5	1901	1872	-2.300
-	400	100	- 5.717	- 74	220	451	0.000	- 4	1140	1141	-2.354
4 \	-00	3.7	0.979	-13	1226	1212	3,142	- 3	2767	2809	6 A07
3	844	791	-2.968	-12	131	193	0.000 🕈	- 2	2314	2412	-1 760
4	93	120	-0.246 +	+11	0.4 3	000	0 000		1.010		-1.702
5	436	435	0.915		140	770	0.000	-1	1170	11/4	1.196
	301	3=7	-0 050	- A V		214	0,142	Q	738	719	2.3/2
ž	174		-0.422	- 9	294	125	3.142	1 -	1242	1257	-2.260
~			e.332	-8	706	860.	3,142	2	1206	1123	0.470
8	229	/17	2.767	-7	874	810	3.142	3	673	620	-0 Kno
•	14	6		-6	2503	2709	0.000		570		-0.309
0	93	181	1.985 4	- 6	114		T 440 4	4	6/4	0-0	- 2. / 48
	15				131	20	3,192 T	-5	657	710	1.620
- 4 4				*4	2157	2123	3.142	6	237	306	-2.525
- 10	770	. 712 .	-2.925	-3	131	38	0.000 +	7	271	332	+0.112
-15	410	354	0.191	- 2	573	359	0.000		6 0 0	410	0.434
-14	329	28(-	P.321	-1	877	401	0 000	. <b>a</b>	309	010	0.034
-13	354	347	2 860	:			0.000	•	17	· 3	
-12	440	4.07	2.000	Ų	1012	10/0	0.000	-14	400	429	-3.053
	-04		2 512	1	1885	1665	3.142	-13	161	2.53	-0.300
-11	-71	371	6.347	2	513	472	3,142	- 1 2	456	454	1 214
-1	1208	1247	-1.576	3	934	771	0 000				
-9	1569	16 3	0.655	¥		24.7	3 4 4 9 4	-11	575	511	-9.589
- 14	0.29	803	0.075	1	191	213	3.142 T	-10	478	387	-2.635
		0.02	21/08	7	279	329	0.000	- 9	441	483	2.343
-/	¥66	12.1	~2.627	6	623	573	3.142		100	744	-0 431
-0	1499	1653	J.280	ĩ	343	417	3 142	- 0	247	300	-0.021
- 5	758	821	+1.444	<b>.</b>	880	~ 4 /	0.146	•7	400	429	3.346
- 4	1286	1244	1.500	<b>_</b> .	034	<b>y</b> =U	0.000	- 6	147	236	2.129 7
.1	1054		0.009	9	174	209	0.000	-5	465	364	-2.723
	1024	072	2.994		16	2		- 4	445	. 347	+0.Ath
-2	1692	1633	2.927	-15	714	766	-0.118		427	410	1 6 7 4
-1	647	8.2	-0.661	-14	414	4 8 0	2 444	-3	07/	839	3.020
7	1710	16 0	0.674	_ 4 7	491	400	E.010	- 2	1037	. 1120	-0.207
•	373	3.47	0.9/4	-13	378	414	-2.901	-1	613	576	2.331
5	·		-2.011	-12	451	469	0.391	0	406	539	+2.0en
4	209	443	2.066	-11	93	213	+1.185 t		1 4 7	1	-1 70V
3	j 1197	1139	3.01A	•1n	378	303	0.040	1	1.1	103	<b>.</b>
						474	41946	2	368	<b>510</b>	v.023

								•			
3	465	446	0.861	3	66	114	0.371 🕈	-6	869	899	•0.307
	414	487	-4 474		104	387	3 030		488	407	-0.538
-	430	42/	-1.4/1	•	394	377	0.000		422		-0,300
,	440	409	2.588	•	_19	1			229	204	0.271
6	161	199	2.005	-15	714	673	1.220	-3	1041	1158	2.910
7	66	348	-0.931 #	-14	761	687	-1.236	-2.	530	486	-3.068
	17	5	-	-13	657	660	0.226	-1	1197	1215	-0.051
-10	66	412	2.057 #	-12	538	454	-2.902	ŏ	03	92	-0.845 #
	21.6	104	2 414	-11	044		3 078		218	917	3 800
	610	141	2.411		700				210	20/	3,900
-8	93	110	1.038 4	-10	492	442	V.227	<u> </u>	470	204	2.172
-7	588	543	-1.415	-9	723	785	-0.356	- 3	186	231	-1.654
-6	569	480	0.599	-8	114	144	0,434 🕈	4	630	722	-0.268
- 5	667	588	3.098	-7	5.01	448	-2.856		20		
- 4	335	207	2.084	- 4	044	037	2 841	-10	474	440	-0.440
	505	627	2.004	-0	700	907	2.001		7/7		
-3	200	211	-0.008	- ?	478	411	0.283	- 4	174	102	-1.0/3
-2	421	366	-1.636	-4	877	942	-0.056	~ 8	478	442	2.469
-1	93	73	1.250 🕂	-3	1027	1083	-2.566	-7	627	557	-3.008
ñ	577	484	2.068	-2	602	618	1.293	-4	441	392	0.845
-		4.4	-9 127		BOA		-2 824		4078	1074	-0 477
	320		-2.12/	-1	220	207	-6.760	- 2	**/*		
ć	400	-03	0.494	ų	040	/30	V. 204		437	490	6./4/
3	. 66	164	-1-511 👕	1	677	616	-1.149	- 3	509	530	3.097
•	18	0		2	554	563	2.816	+2	294	242	-0.540
-15	93	116	3.142 +	3	208	229	3.123	-1	255	248	1.937
-14	03		3.142 +		364	218	-0 845	ñ	402	816	+6.517
	70		3.172 4		501	200	-0.005		492	124	
-13	702	0 ij 2	0.000	2	36 U -	341	0.242	1		169	1,400 T
-12	664	555	0.000	6	66	131	~0,175 <b>f</b>	2	363	534	2.953
-11	699	531	3.142	•	19	3		•	21	1 .	
-10	1174	1170	3.142	-13	344	351	-0.457	-13	1070	043	-2.670
	1144	1144	0.000	-12	441	424	-3 044	-12	228	21.8	1.507
	-222	+100	0.000		274		2 740	-16	620		-0 424
- 8	470	284	0.000	-11	220	294	6./00	-11	027	/00	-4.461
-7	513	466	0.000	-10	. 114	151	-0.670 T	-19	542	388	0.935
-6	360	272	3.142	-9	595	691	0.067	- 9	378	462	2.369
- 5	2184	2270	3.142	- A	284	350	-1.269	- 8	934	884	-2.506
- 4	21.84	2102	0.000	.,	761	733	2 972	-7	400	371	0.354
	5104	E106	0.000		771		0.075	- 4			- 0 042
- 3	807	062	0.000	-0	/11	001	2.035		A0 8	877	
-2	1895	1895	3.142	-5	894	932	-1.100	- 7	93	170	V.301 T
-1	1604	1478	0.000	-4	720	716	0.267	-4	975	1021	-3.016
Ď	964	944	3.142	- 3	483	453	0.555	-3	460	478	2.339
1	131	1 7 1	n.oon 🔶	-2	1630	1.617	-3.040	-2	602	612	-0.484
-	46.00		0.000	-1	444	447	6 203		1074	1001	0 141
2	1966	1482	0.000	-1	707	406	0.293		10/0	1071	0.000
3	1484	1410	3.142	Ç	469	329	-0.247	U U	813	82/	-2.0UV
4	114	17	0.000 🕈	1	436	424	-0,558	1	329	340	2.643
5	341	383	0.000	2	366	403	1.541	2	348	286	-2.928
Å	93	64	3.142 +	ā	711	822	-2.719	3	366	465	-0.413
-	474	4.7.7	0.000		11	4 7 7	0.407.4		705	741	
	-30	-/3	0.000		40	1/3	0.797 1	•	/05	101	. 0.000
8	00	399	3.142 🕈		66	378	0.155 <b>T</b>	· · · ·	21	3	
٠	18	2.		•	19	5		-11	389	408	-0.165
-15	802	815	2.573	-8	255	146	0.008	-10	329	316	-0.769
-14	66	114	-2.450 4	- 7	341	487	2 328		KTA	877	2.822
	E 20	***	-61920 T		1041		2.440		500		2
-13	722	292	0.273	-0	324	007	-2.100		460		2.040
-12	542	536	-0.100	-5	66	455	0.843 🕂	-7	436	368	0.182
-11	711	729	-2-496	-4	66	205	-0.103 🕈	-6	606	697	-0,710
-10	744	712	2.630	-3	66	136	-2.041 🛨	-5	315	245	2.547
- 0	741	681	0.581	-2	66	216	2.977		416	431	2.402
- 6	= 2 2	450	4 070					_ 4	407	403	-2 867
-0	700	079	-1.038	-1			1.003 1	-0	021	743	-6.007
-/	702	/60	1.254	Q	301	233	-1.340	-2	000	/ 4/	-0.113
- 6	595	614	-2-652	•	20	0		-1	237	271	0.746
-5	271	268	-2.468	- 14	441	321	0.000	0	279	304	-1.911
- 4	872	0.4	1.269	-13	03	38	0.000 +		641	702	2.547
-3	818	874	-1 046	-12	4 8 0	417	3 142	<b>3</b> -	44	290	+0.781 +
_ ~	200	<b>1</b> /7	- 4 + 4 - 7	- 16	000	017	7 4 4 5	6		2.70	
- 2	. 308	303	-2.183	-11	225	400	J.142	*	.22	~ <sup>0</sup>	7 440
-1	1082	1211	2,288	-19	1066	913	0.000	-13	378	246	3.142
Ç	322	353	-1.631	-9	271	203	0.000	-12	990	975	0.000
1	591	SAD	-0.015	- A	335	292	3.142	-11	03	62	3.142 +
5	244	245	0.201	-7	740	744	3.142		214	238	3.142
	600	272	-3.0/-				0 000	- 10	210	490	1 4 4 3
3	244	025	-2.202	-0	400	348	0.000	- 9	460	44 9	3.196
4	446	484	2.602	-5	755	719	0.000	-8	651	560	3.142
5	286	3:1	1.267	-4	451	452	3,142	-7	1632	1638	0.000
6	301	349	-0.241	-3	107	1 52	3.142	-Å	114	59	3.142 +
ž	44	40	2.324	ě	770	74.2	3.142	_	1400	1440	3.142
· .	4.6	0,	21930 T		1/2		V, 476	• 2	140v		0 888
	18	. 1		-1	AS0 -	811	v.uuv	-4	501	403	0.000
-12	492	457	-0.053	0	580	593	0.000	- 3	301	272	0.000
-11	400	299	-3.001	1	1121	1105	3.142	-2	831	790	0.000
-10	246	241	-2.987	2	301	314	0.000	-1	400	241	3.142
- 0	372	A 3	2.314		4	44 0	3 1 4 2	-	763		3.142
			2.004	0	70/	717	0.276	y y	176	0,0	3 440 4
-6	123	754	-0.756	4	294	200	0.000	1	93	38	3.142 T
-7	322	266	0.355	5	1196	1084	0.000	· 2	341	429	0.000
-6	569	495	2.766	6	546	889	3.142	3	606	531	0.000
-5	301	244	-2.470		20	• 2			22	2	
-4	335	207	-0.04m	-4 2	470	A 4 0	8.141	-11	7 Å Å	484	5,102
	474	47/		-13	3/0		* ***	-14	97/		-2 -4 -
- 3	4/4	42%	0.512	-12	627	013	0.081	-10	399	02/	-2.003
-¥	508	125	1.493	-11	354	220	-0.401	- <b>- 9</b>	446 .	250	2.B/U
-1	764	648	3.039	-10	492	602	-0.046	-8	441	509	0,107
0	237	200	-1.500	- 9	378	410	-0.331	-7	66	192	-1.007 🕇
1	329	203	+0.647	- i	6.80	651	2.944	- 4	383	340	0,720
5	344	172		_	222	194	2 641	-	204	474	-2 614
	9-6	994	1+101	-7	Q# /	040	E . 790	- 7	241	7/4	- * * 8 3 8

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-4	1024	1122	3,120
-3	872	897	0.343
-2	651	723	-0.242
-1	329	422	3.078
Ō	416	457	2.904
1	394	390	-2.165
2	558	607	0.031
	22	4	
-5	686	712	1.923
-4	565	501	-1.801
•	23	1	-
-11	186	176	-2.190
-10	831	820	-3.102
-9	421	397	0.654
-8	723	741	-0.266
-7	405	403	2.254
-6	394	420	-2.070
-5	797	789	-2.756
-4	1007	1103	0.964
-3	1058	1188	-0.897
-5	469	530	2.202
-1	595	676	2 911
Ç	410	419	-1.575
1	197	249	0.270
•	23	3	
-8	308	254	-0.495
-7	496	526	-1.024
-6	492	553	1.698
-5	758	578	-2.559
-4	308	202	2.788
-3	869	899	0-192
-2	335	329	-2.470
	24	0	
-10	66	147	0.000 🕇
-9	1339	1201	0.000
-8	908	725	3.142
-7	93	163	3.142 🕈
-6	400	352	3.142
-5	93	59	0.000 T
	1166	996	0.000
-3	400	4:5	3.142
-2	496	431	3.142
0	208	256	3.142
.*	24	2	
-8	522	495	-0.995
-7	228	309	2.804
-6	394	376	2.354
-5	186	150	2.937
-4	602	652	-0.479
-3	308	318	0.691

Table 5.	6-NBF-PA	intra-molecu	ılar	bond d	listances	and	their
estimated	l standard	l deviations	in .	Angstro	om units.		

Bond

Nl

Nl

N1

Bond

<b>-</b> C2	1.476 (0.009)	N14 - C17	1.485 (0.008)
- 05	1.471 (0.008)	C15 - 016	1.233 (0.011)
- C7	1.387 (0.008)	C17 - C18	1.529 (0.012)
- 03	1.576 (0.011)	C18 - C19	1.371 (0.018)
- 011	1.529 (0.009)	C18 - C23	1.371 (0.017)
<b>-</b> \$4	1.850 (0.005)	C19 - C20	1.394 (0.015)
- 09	1.476 (0.017)	C20 - C21	1.398 (0.026)
- C10	1.563 (0.014)	C21 - C22	1.403 (0.026)
- 05	1.778 (0.010)	C22 - C23	1.396 (0.018)
<b>-</b> C6	1.566 (0.009)	, •	
- C7	1.551 (0.008)		
- N14	1.443 (0.009)		

C2 Ċ2 C3 C3 03 **S4** C5 **C**6 C6 c7 - 98 1.184 (0.009) 1.342 (0.009) C11 - 012 C11 - 013 1.180 (0.013) N14 - C15 1.325 (0.013)

C2 - N1 - C5	118.6 (0.6)	C6 - N14 - C15	123.2 (1.1)
C2 - N1 - C7	125.6 (0.7)	C6 - N14 - C17	115.3 (0.5)
C5 - N1 - C7	94.9 (0.5)	C15 - N14 - C17	120.4 (0.9)
N1 - C2 - C3	105.8 (0.7)	N14 - C15 - 016	126.9 (1.5)
N1 - C2 - C11	108.9 (0.5)	C2 - <sup>C</sup> 11 - 012	109.5 (0.5)
C3 - C2 - C11	114.9 (0.8)	C2 - C11 - 013	126.1 (1.3)
C2 - C3 - S4	104.7 (0.5)	012 - 011 - 013	124.4 (1.4)
c2 - c3 - c9	111.1 (1.1)	N14 - C17 - C18	112.1 (0.8)
C2 - C3 - C10	110.5 (0.8)	C17 - C18 - C19	119.6 (1.3)
s4 - c3 - c9	111.2 (0.7)	C17 - C18 - C23	119.1 (1.3)
S4 - C3 - C10	107.5 (0.6)	C19 - C18 - C23	121.4 (1.7)
C9 - C3 - C10	111.5 (1.4)	C18 - C19 - C20	120.9 (1.7)
C3 - S4 - C5	92.4 (0.4)	C19 - C20 - C21	118.2 (1.7)
N1 - C5 - S4	104.2 (0.6)	C20 - C21 - C22	120.4 (2.6)
N1 - C5 - C6	87.6 (0.4)	C21 - C22 - C23	119.6 (1.9)
S4 - C5 - C6	119.5 (0.8)	C18 - C23 - C22	119.2 (1.7)
C5 - C6 - C7	85.1 (0.5)	N1 - C7 - C6	91.3 (0.4)
C5 - C6 - N14	120.1 (0.7)	N1 - C7 - O8	132.0 (1.1)
C7 - C6 - N14	116.1 (0.6)	c6 - c7 - o8	136.7 (1.1)



Figure 3. 6-NBF-PA bond distances and angles.

temperature factor coefficients are shown in Table 3. The observed and calculated structure factors are listed in Table 4, the final R value being 0.100. Bond distances and angles and their estimated standard deviations were calculated by ELSI and are shown in Tables 5 and 6, and in Figure 3.

### Configuration of the Molecule

Figure 4 is a projection of the 6-(N-benzylformamido)penicillanic acid molecule down the b axis, as determined by the analysis. <sup>T</sup>he projection down the c axis is shown in Figure 5. With the exception of the N-benzylformamido sidechain, the structure of the molecule is similar to those found for the other penicillins mentioned in Chapter 3.

The benzyl group in the side-chain is curled round towards the fused ring system as in benzyl penicillin, but its position in 6-NBF-PA is slightly different. It can be expressed as an imaginary rotation , of approximately 90, of the benzyl group about the bond N14 - C17 in the direction of the sulphur atom. The least-squares planes through the groups of atoms in the molecule have been calculated with DIDO and are shown in Table 7. The amide group in the side-chain is planar, although the deviations are not as low as would be expected (Table 7a), and orientated approximately normal to the plane of the **p**-lactam ring. The dimensions of the group are consistent with those


Figure 4. 6-NBF-PA projected down the b axis



<u>Table 7</u>. Deviations (A) of the atoms in 6-NBF-PA from the calculated least-squares planes. Atoms marked with \$ were not included in the calculation of the plane.  $\sum d^2$  is the sum of the squares of the deviations from the plane(in A<sup>2</sup>)

a. Ami $\sum_{d=1}^{d} d^2$	.de = 0.00943	b. Benzene Ring $\sum d^2 = 0.00297$					
Atom	<b><i><b>Jevi</b>ation</i></b>	Atom	Deviation				
C6	-0.043	C17	-0.020				
C15	0.017	C18	0.007				
016	-0.033	C19	0.002				
N14	0.075	C20	0.005				
C17	-0.015	C21	0.000				
		C22	-0.032				
c. Thi	azolidine Ring	C23	0.040				

C. TOT	azorrarue ving
$\sum a$	<sup>2</sup> = 0.00014
Atom	Deviation
C3	-0.004
C2	0.007
Nl	-0.008
C <u>5</u>	0.004
S4 \$	0.729

a. β - Σ a <sup>2</sup>	lactam ≈ 0.01858	e. $\beta$ -lactam $\sum d^2 = 0.00002$					
Atom	Deviation	Atom	Deviation				
Nl	0.072	Nl	-0,001				
C5	-0.089	C6	-0.001				
C6	0.057	C7	0.004				
C7	0.008	08	-0.002				
08	-0.047	005.\$	0.209				

of a normal resonating amide. 016 is cis to the benzyl group and C15 is trans to the proton at C6, which is opposite to the analogous position in the other penicillins.

In terms of Collins and Richmond's explanation of the biological activity of the penicillins, 016 is on the 'wrong' side of the molecule with respect to the other polar groups (N1, 013, 08) to form a possible hydrogen-bonding system with a protein substrate similar to that postulated for N-acetylmuramic acid and the penicillins. As pointed out by Hunt and Rogers (1964), 016 can only be brought into the required position by rotation about C6 - N14 followed by cis-trans isomerization of the amide group. This combination seems improbable even in physiological solutions; the rotation would be hindered by the size of the benzyl group and the isomerization would require energy equal to the resonance energy of the amide. (Wheland (1955) estimates this value as 17 - 60 K cal./mole but with a large degree of uncertainty). Hence, the low antibacterial activity of 6-NBF-PA is consistent with Collins and Richmond's theory, and, it should be noted, with Tipper and Strominger's view that penicillin binds to the active centre in preference to a substituted D-alanyl-D-alanine.

#### Packing

The unitsell contents and the adjacent molecules in the 6-NBF-PA crystal are shown projected down the b axis in

Figure 6 and down the c axis in Figure 7. Some inter-molecular distances are shown in both diagrams and distances less o than 3.25 A are shown in Table 8. Atoms related to those in Table 2 by the symmetry of the spacegroup are referred to by terms in parentheses following the atom name.

The distances between the atoms 016 and 012  $(x + \frac{1}{2}, 0)$ y -  $\frac{1}{2}$ , z) of 2.665 Å strongly suggests a hydrogen bond between these two atoms. This is supported by the bond lengths in the carboxyl group, C11 - 012 (1.342 Å) and C11 - 013 (1.180 Å) which agree quite well with the values of 1.358 Å and 1.233 Å given by Sutton (1965) for the unionized form. The result is that the molecules related by centering form an approximately linear hydrogen-bonding system roughly along the ab diagonals. That this is a 'criss-cross' arrangement can be seen from Figure 7.

The general packing arrangement is similar to that described for potassium benzylpenicillin by Crowfoot et al. (1949) and for phenoxymethylpenicillin by Abrahamsson et al. (1963). The polar groups are on one side of the molecule and the hydrocarbon parts on the other, so that the molecules make contact with one another in alternate polar and nonpolar layers, which in 6-NBF-PA are parallel to the a axis.



Figure 6. Crystal structure of 6-NBF-PA projected down

the b axis

Figure 7.



<u>Table 8</u>. Inter-molecular distances in 6-NBF-PA in A. The atomic positions are related to those in Table 2 by the terms in parentheses which are :-

(1)	=	ł	-	x,	у	+	1/2,	1-z
(2)	Ξ	x	+	1 29	у,	-	<del>1</del> ,	z
(3)	u	x	+	1 2,	у	+	1 2,	z

N1 - 08 (1)	3,080	C11 - 016 (1)	<b>3.</b> 398
C6 - O8 (1)	3.374	013 - C15 (1)	3.200
C7 - 08 (1)	2.992	013 - 016 (1)	3.417
08 - 08 (1)	3.331	016 - 012 (2)	2.665
08 - C15(1)	3.286	C17 - 012 (3)	3.293
C11- C15(1)	3.479	C17 - 013 (2)	3.351

# The Thiazolidine and B-Lactam Rings

Diamand (1963) has tabulated the bond lengths and angles of the penicillin nucleus from the determination of the crystal structures of 6-APA, potassium benzylpenicillin refined by Rollett and Vaciago, phenoxymethylpenicillin and p-bromo-phenoxymethylpenicillin by Watson. Most of the values are in good agreement with those listed in Tables 5 and 6 for 6-NBF-PA. The shortening of the bond C3 - C9 (1.476 (0.017 Å)) is probably not significant in terms of the real standard deviations, and there is no apparent reason for the distortion of the angle C3 - C2 - C11 (114.9°) from the tetrahedral value of  $109^{\circ}$  28'.

The thiazolidine ring has four atoms, N1, C2, C3 and C5 in a plane (Table 7c) with S4 0.729 Å below this plane viewing the molecule down the b axis as in Figure 4. This conformation is different to that of the thiazolidine rings in the other penicillins. As mentioned in Chapter 3, in benzylpenicillin and phenoxymethylpenicillin C2 is the out-of-plane atom, in the latter case by 0.51 Å on the opposite side of the plane to C7. In 6-APA, N1 deviates from the best plane through the other four atoms by 0.42 Å, on the same side of the plane as N14. The smaller value for the angle C3 - S4 - C5 (92.4°) compared with 96°- 97° in the other penicillins probably results from this conformational difference. Although the bond 54 - C5 (1.778 A) is only 4 e.s.d.'s lower than the average value of 1.817 Å, given by Sutton (1965) for a single C - S bond, 54 - C3 (1.850 A) is 7 e.s.d.'s higher. This agrees with the values for the other penicillins except potassium benzylpenicillin for which Rollett and Vaciago give 54 - C5 (1.85 A) and 54 - C3 (1.82 A). The lengthening of the bond 54 - C3 has been discussed by Diamand (1963). By considering the mean value for this bond length and its true standard deviation from previous determinations, he tentatively suggests 54 - C3 has a bond length in the range 1.84 A to 1.87 A. The value in 6-NBF-PA lies in this range.

The non-planarity of the  $\beta$  -lactam ring can be seen from Table 7d, the deviation of C5 from the calculated plane being as high as 0.089 Å. Table 7e shows that by excluding C5 from the least-squares calculation, the cyclic amide! atoms N1, C6, C7 and 08 lie in a plane well within the limits of experimental accuracy, with C5 0.204 Å out of the plane on the same side as S4. This is similar to the arrangement in 6-APA, phenoxymethylpenicillin, ceph. Cc and the 3-spiro-thiazolinium derivative (Chapter 5) where the deviations of the put-ofplane atom vary from approximately 0.14 Å to 0.25 Å. In each case it is the atom bonded to the sulphur atom that lies out of the plane.

As Diamand (1963) has mentioned, the tendency of the group



to be planar, so that resonance can occur in the cyclic amide, is probably opposed to some extent by the tendency of the five-(or six) membered ring to adopt its most stable conformation. <sup>Th</sup>is will result in C5 being pulled out of the plane of the  $\beta$ -lactam ring. That resonance occurs to some extent in 6-NBFo PA is indicated by the length of the bond N1 - C7 (1.387 Å), which is nearer to the value of the C - N bond length in amides (1.333 Å) than the accepted value of 1.472 Å for a C - N single bond. The angle C2 - N1 - C5 (118.6°) suggests that N1 is tending to planarity.

83.

#### CHAPTER 5

The Crystal Structure of

(3S,4S,6R,7R)-2'-Amino-3'-ethyl-2'-thiazolinium-4'-spiro-3-(7-phenylacetamidocepham-4-carboxylate) trihydrate

#### Preliminary Data

The crystals, which were supplied by Dr.A.G.Long of Glaxo Ltd., and prepared as indicated on page 44, were colourless monoclinic needles. The crystallographically unique axis, b, was parallel to the needle axis and the crystal selected for data collection had the approximate dimensions  $0.6 \ge 0.2 \ge 0.13$  mm. The structural formula and numbering scheme are shown in the diagram inside the back cover.

Formula  $C_{19}^{H}_{28}N_{4}O_{7}S_{2}$  Molecular Weight (formula) = 488 Mass absorption coefficient for  $CuK\alpha$ ,  $\mu = 25.5 \text{ cm}^{-1}$ 

Unit-cell dimensions a = 8.063 (0.008) Å, b = 7.184 (0.008) Å, c = 19.72 (0.02) Å  $\beta$  = 100.3° (0.2) V = 1123 Å<sup>3</sup>, Z = 2, F(000) = 516 electrons D<sub>obs</sub>(by flotation) = 1.45 g.cm<sup>-3</sup> D<sub>calc</sub> = 1.44 g.cm<sup>-3</sup> Absent Spectra: only among OkO for k = 2n + 1 As the compound is dextrorotatory (page 43), the spacegroup is uniquely determined as  $P2_1$ . 2269 independent reflexions were estimated visually using three intensity wedges to accommodate spot-shape changes from layer to layer. The data were correlated from six layers collected about the b axis and three about the a axis, using LOLA for the determination of the inter-layer scale factors. It was considered that the value of  $\mu$  and the crystal dimensions rendered absorption corrections unnecessary ( for data collected about the a axis, a crystal was sliced normal to the needle axis to give a rough cube of side 0.3 mm).

Figure 1 shows the Wilson plot calculated for threedimensional data including the unobserved reflexions at  $|F_{unobs}| = \frac{2}{3}|F_{min}|$ . The data were divided into 22 shells each containing 80 - 120 reflexions. The least-squares method gave B = 2.95, K = 100.4. From the graph, the best line through the points in the middle of the  $\sin^2 \theta/\lambda^2$  range gave B = 2.91, and the intercept gave K = 104.6 (The final scale after structure refinement was 101.2)

## Structure Determination

## 1. Location of the Sulphur Atoms

The four sulphur atoms in the unit cell give rise to





on three-dimensional data.

six centrosymmetrically related pairs of S - S inter vectors. In each six, two correspond to peaks in the three-dimensional Patterson, occuring in the Harker section at  $v = \frac{1}{2}$ , and the remaining four correspond to two peaks of double weight at

> $x_1 - x_2, y_1 - y_2, z_1 - z_2$  $x_1 + x_2, y_1 - y_2 + \frac{1}{2}, z_1 + z_2$

where  $x_1$ ,  $y_1$ ,  $z_1$  and  $x_2$ ,  $y_2$ ,  $z_2$  are the positions of the two independent sulphur atoms. A three-dimensional Patterson was computed with coefficients sharpened by the function

$$\frac{1}{f^2} \exp \left( 3 \sin^2 \theta / \lambda^2 \right)$$

where f is the formfactor for sulphur. Four peaks were recognised which complied with the conditions described above and gave the following fractional coordinates for the sulphur atoms :

x	У	z
0.155	0	0.069
0.395	0.182	0.293

The distance between these two positions, 4.7 A, seemed reasonable assuming the approximate geometry of ceph. Cc as determined by Diamand (1963)

# 2. Location of the Light Atoms

The ratio of the sum of the squares of the scattering powers of the sulphur atoms to that of the light atoms is 0.39 イ

and Fourier methods were used to locate the positions of the light atoms. The first Fourier map, phased on the coordinates of the sulphur atoms revealed five additional peaks, one of which was subsequently found to have been misplaced. <sup>Th</sup>ree further Fouriers enabled the assignment of all thirty-two positions and the calculated and observed structure factors showed an overall agreement factor, R, of 0.29. During the course of the analysis, reflexions for which

$$|F_{o}| < 0.25 \text{ k} |F_{c}|$$
 where  $k = \sum_{i=1}^{n} |F_{o}|$ 

were omitted from the Fourier calculations, and double shifts were applied to the estimated coordinates before input to the next stage, as recommended by Donohue (1950)

## Refinement

Four cycles of isotropic, block-diagonal, least-squares refinement reduced R to 0.114 and a maximum coordinate shift of 0.03 A. At this stage a bond-distance calculation indicated sensible bond lengths and a difference Fourier confirmed that the postulated structure was substantially correct. An agreement analysis was used to check the validity of the interlayer scaling and that the choice of  $F^{\dagger} = 1500$  in the weighting scheme produced fairly constant values of  $\sum w \Delta^2$  over ranges of |F<sub>o</sub>|.

The progress of the structure refinement is shown in Table 1. It should be noted that the R value associated with a particular cycle was calculated from atomic coordinates before refinement. Agreement analyses and molecular geometry calculations were used to check the course of the refinement after cycles 18, A10 and A16.

Four further cycles of isotropic least squares (15 to 18) produced very little convergence and so for input to A9, the individual isotropic temperature factors were converted to the equivalent anisotropic form and refinement continued on all 32 atoms, applying a fudge factor of 0.8 to all parameter shifts. At cycle A11, two reflexions  $(1 \ 0 \ \overline{2} \text{ and } 1 \ 0 \ \overline{1})$  were omitted for suspected extinction and 214 unobserved reflexions included at  $\frac{2}{3}F_{\min}$ , where F is the minimum observed value of F at the relevant  $\sin^2 \theta$ . After cycle A16, a difference Fourier was calculated and suggested the positions of all hydrogen atoms except one bonded to C6, two bonded to C15, the three methylgroup hydrogens at C23 and the six of the three water molecules. The location of two peaks near N29 in approximately the correct orientation suggested the presence of the group  $=NH_2^+$ . However, the peak heights in the difference map varied from 0.13 to  $0.3 \text{ eA}^{-3}$ , and as the difference Fourier still exhibited ripple of the order 0.2  $eA^{-3}$  these positions were not taken as

Cycle	Agreement	$\sum w \Delta^2_{x10}^{-4}$	Maximum coord-
Number	Factor, R	с. Х	inate shift (Å)
Il	.0.289	35503	0.15
IS	0.175	14427	0.20
13	0.133	7340	0.05
14	0.114	5991	0.03
15	0.113	5485	0.01
16	0.112	5370	0.01
17	0.112	5357	0.005
18	0.112	5353	0.002
A9	0.112	5366	
A10	0.089	3487	0.014
A11	0.082	3034	0.008
A12	0.083	2634	0.014
A13	0.081	2441	0.007
A14	0.081	2420	0.004
A15	0.081	2411	0.002
A16	0.081	2408	0.002
A17	0.077	2073	0.009
A18	0.075	1963	0.003
A19	0.075	1955	0.0015
A20	0.075	1950	0.0008

<u>Table 1</u>. The refinement process for the 3-spiro-thiazolinium-7-PAC derivative. I = isotropic, A = anisotropic refinement. conclusive proof of hydrogen atom positions.

Using a program written by D.F.High for the I.B.M. 7090, hydrogen positions were calculated assuming the staggered conformation for the methyl group at C23. The observed and calculated positions are shown in Table 2, excluding those of the water molecules. They show good agreement except for the atom bonded to C20. All the calculated hydrogen positions listed in Table 2 were included in subsequent structure-factor calculations with isotropic temperature factors 1.5 times those to which the hydrogen atoms are bonded. The hydrogens of the water molecules were not included as it was considered that their positions could not be well enough defined by stereochemical considerations.

Refinement was considered complete after cycle A20 when the position shifts were about one tenth of their standard deviations. Structure factors were calculated from the positions output from A20 to give a final R value of 0.075. The calculated and observed structure factors are listed in Table 3a.

A difference Fourier was calculated from the structure factors listed in Table  $3\sigma_{i}$  in an attempt to locate the hydrogen atoms of the water molecules. The map exhibited a ripple up to about 0.17 eA<sup>-3</sup> and the areas around the oxygen atom positions gave no indication of the hydrogen atoms.

Table 2. Observed and calculated hydrogen positions(in fractional

coordinates) for the 3-spiro-thiazolinium-7-PAC derivative.

Atom.	Bonded Calcul		alculate	d	Ob	Observed		
	t0-	x	У	Z	x	У	z	
H1	C2	0.475	0.632	0.201	0.491	0.625	0.200	
H2	C2	0.453	0.872	0.178	0.438	0.883	0.167	
H3	C4	0.052	0.674	0.235	0.054	0.675	0.233	
H4	C6	0.271	0.608	0.363		-* -		
H5	C7	0.460	0.331	0.374	0.417	0.350	0.383	
нб	<b>C</b> 9	0.167	1.012	0.186	0.167	1.000	0.183	
H7	C9	-0.017	0.882	0.152	0.000	0.875	0.150	
н8	N13	0.674	0.586	0.311	0.646	0.600	0.308	
Н9	C15	0.979	0.488	0.317				
H10	C15	1.050	0.368	0.397				
H11	C17	1.090	0.540	0.485	1.083	0.500	0.500	
H12	C18	1.329	0.387	0.567	1.375	0.450	0.550	
H13	<b>C1</b> 9	1.444	0.090	0.539	1.438	0.000	0.542	
H14	C20	1.319	-0.073	0.428	1.542	-0.150	0.400	
H15	C21	1.089	0.080	0.346	1.125	0.133	0.367	
H16	C22	0.199	0.407	0.033	0.167	0.350	0.067	
H17	C22	0.248	0.364	0.124	0.250	0.350	0.117	
H18	C23	0.495	0.284	0.067			·	
H19	C23	0.484	0.449	0.038				
H20	C23	0.544	0.537	0 <b>.12</b> 3				
H21	N29	0.248	0.612	-0.041	0.229	0.600	-0.027	
H22	N29	0.191	0.849	-0.054	0.188	<b>0.850</b>	-0.050	

<u>Table 3</u>**a**. The 3-spiro-thiazolinium-7-PAC derivative observed and calculated structure factors ( x 100) in the order

\* h k  
1 
$$|F_{r}| < F_{r}| < (radians)$$

Unobserved reflexions are marked  $\uparrow$ , and 0 0 1 and 0 0 2 were obscured by the backstop. The two reflexions omitted for suspected extinction were

h	k	1	F	Fc	d
1	0	2	6003	8419	0.000
1	0	1	5789	8266	0.000

3	0	0 753	6.560	10	2038	1861	-2.714	5	.910	918	•
4	1998	2057	3.142	12	506	387	2.045	8	8#2 770	802	
2	3643	3929	3,142 0.906 ♥	14 15	1056.	1049 311	-2.029	. 9	1405	1375	-
7	564.	5453	3.142	16	757	708	2.134	ii	1547	1649	
ě	2269	2307	3.142	17	858 175	759 116	1.738 -2.685 <b>t</b>	12	664 679	948	•
1 11	3419	3258	0.000 0.000	19	365	379	2.766	14	192	171	-
12	910	854	0.900	21	496	502	-1.435	15	192	260 975	
14	1271	1203	0.000 0.000	22 23	392	302	2.824		0	. 8	
15	615	450 369	0.000		0	4		1	1327	1334	
17	615	558	3.142	1	3515	1574 3566	-0.992 0.323	2	1365	1300	-
10	554 175	440 177	3.142 3.142 <b>†</b>	2	2480	2688	1.404	4.	764	740	
2,	, 175	9 <b>9</b>	3.142 +	•	2186	2274	-1.073	.6	922	991	-
22	1222	1190	3,142	6	1350	1444 1013	0.492	7	. 938 508	1040	
23 24	143	51 769	3.142 T 0.000	7	2223	2126	-1.824	9	392	334	-
•	0	1	-1 540		441	456	-2.456	11	143	813 58	-
2	4261	4724	2.672	10 . 11	2381 572	2386	-1.954 2.689	12	701	627	(
3	2615 5343	2926 5082	0.455 -2.876	12	2457	2544	1.556	1	632	450	(
5	3762	3672	1.710	14	632	1303	1.434	3	1380 1567	1269 1569	
7	2565	2568	2,761	15	970 777	89 <b>3</b> 760	0.580	4	679	630	•
8	3:03	2982	1,862	17	1334	1330	-1.164		1	. 54/	1
1	2170	1999	-1.102	18	143	913 144	2.469 +	-25	554 441	453	1
11 12	1214 2626	1247 2515	2,918	20	1017	1136	-2.187	-23	598	715	3
13	3401	3414	-1.162	22	708	771	2.724	-22	402	412 476	
15	1201	1045	2,045		0 1637	1765	0.686	-20	350	328	
16 17	1717	1660	-0.094	2	1574	1429	-1.268	-16	365	476	1
18	1705	1581	1.241	4	3993	664 3927	2.096	-17	350 784	182	1
19 2-	764 905	826 941	-0.031 -0.810	5	1941	2022	1.192	-15	581	477	i
21	1167	1178	2.429	7	1380	1297	0.778	-14	790 1153	706	2
23	671	643	2.340 T 0.097	· 8	991 1930	961 1895	-2.160	-12	1357	1243	
24	101	181	-1.016 4	10	656	611	-2.178	-10	2953	2770	2
¢.	5124	5676	1.563	12	1650	1777	1.49/	-9	4661 2311	4552	3
2	2232 2467	2581	0.064	13	1541	1535	-1.151	-7	374	161	Ċ
3	3229	3489	1.690	15	1295	1328	0.437	-5	1247 8392	1293 10458	3
5	6445	0675	-2,365	10	336 598	247	-0.058	-4	4258	4573	3
6 7	2754 3011 -	2719	2.948	18	694	691	0.915	0	1288	1195	0
8	2223	2197	-2.109	20	864	417	0.961 -3.050	1 2	2340	3145	3
r i	3, 35	2890 3429	-1.042 -2.645	•	0. 1740	1625	n 897	3	3639	4243	3
11. 12	1387	1298	0.814	i	248	235	2.985	5	4479	4906 1865	3
3	1140	1086	1.296	23	581 1702	518 1810	-0.994 -0.427	. 0.	1874	1749	q
19	1342 2445	1241	1,485 8,056	Å.	590	379	2.656	8	101	174	0 3
16	1376	1421	-D.417	5	2214	2040	-2.871	9 10	<b>516</b> 101	- <b>536</b> 46	, j
14	1162	1221	2.176 -1.212	7 A	226 1729	395	-0.591	11	3058	2977	0
9	563	547	-1.081	9	2285	2226	-2.731	12 13	2316 516	2146	2
21	554	607	2.677	10 11	422 485	759 789	1.267	14	1311	1224	3
18 13	563 545	573 524	-2.966 '	12	202	341	0.566	16	516	402	1 
24	429	490	1.728	14	722	698	0.053	17	1452	1494	ņ
1	3070	3537	-1,199	15	202	239	2.745	19	1126	1147	÷
23	2911 2144	3251 2343	2.259	17	1071	1176	0.263	20 21	874	497 817	3
-	25.0	2153	-1.198	18 19	409 501	323 678	-2.172 -2.724	22	4#5 248	431	2
7 6	1 42 2899	2970	1.360 1.743		) 224	7		24	842	<u>914</u>	9 3
7. F	2 85	2001	-0.011	2	422	849	1.224	-25	1 101	1 271	-
9	305	291	-0.865	3	1427 1357	1491 1345	1.160	-24	101	145	2
	2038	3861	-2.714	5	-10	e () 9	-2,945	-23	960 640	17.*\$ 671	- 6
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-21	743	738	2.937 2.183	14 15	1469 1061	1455 918	0.294 -2.968	5	2883	3269	-2.878
-19	1431	1444	-1.268	16	1534	1597	2.669	6	2675	2647	3.044
+18	1494	1509	2.226	17	1193	1184	-3.011	7	2071	1443	1.440
-16	1963	1686	-0./56	19	816	-812	1.972	9	1554	1541	-1.703
-15	922	814	-0.290	20	441	131	-2.000	10	1625	1638	0.433 6.397
-14	1871 526	1718	-1.437	21	581	695	2.591	12	1735	1777	-0.481
-12	1628	1456	4.837	23	365	390	2.120	13	1413	1349	0.286
-11	3273	3185	2,662	-23	101	79	1.383 +	14	1222	1272	-1.629
-9	2615	2406	0.039	-22	598	529	-1.686	16	679	708	-0.213
-8	722	754	-2.987	-21	392	322	-0.843	17	624	189	-1.729 1
-6	2872	2601	3.026	-19	876	913	2,751	19	441	448	-2.910
-5	3595	2973	-2,830	-18	1090	1098	-1,654	20	736	749	2.838
-3	3684	3869	-3,049	-16	1180	962	0.696	· · ·	1		
-2	5301	5691	0.601	-15	663	618	-2.526	-21	379	351	0.253
-1	5107	5857 3537	-3,130	-14	1251	1240	-2./6/	-20	598	551	-1.431
i	2907	2790	-2.739	-12	2098	1952	1.077	-18	350	343	-0.077
2	2:33	1997	1.611	-11	1534	1394	0.425	-17	143	140	0./00 T
4	2160	2268	0.975		3737	3823	2.331	-15	722	636	2.245
5	3804	3683	3.013	-8	922	816	-1.206	-14	1376	1346	1.471
<b>9</b>	3529	3925	2.724	-7	3364	2432	-0.932	-13	1108	1057	-2.642
8	2902	3030	-0.416	-5	3131	3048	-2.043	-11	1726	1600	0.404
	1653	1567	-2.637	-4	3773	3909	-1.363	-10	1445	1494	1.235 -1.401 T
11	3523	3818	-1.400	-2	5167	5449	0-494	-8	1099	1073	-1.620
12	2805	2894	-0.821	-1	2911	2900	0.030	-7	1868	1784	-1.712
13	960	836	1.276	0	3762	3715	1.090	-5	2432	2502	-2.203
15	1455	1376	-0.273	2	269,0	2986	0.872	-4	303	315	-1.503
10	2193	2344	1.776	3	944 730	. 891	0.105	-3	1136	1015	-0.965
18	1612	1720	0.693	5	598	583	-2.970	-1	3431	3260	1.366
19	790	793	-2.704	6	1037	1137	-2.482	8	2011	2191	1.789
20	91U 526	973	-3.002	/ 8	1717	1682	-1.286	1	648	527	2.094
22	464	565	-0.708	9	1838	1874	-1.378	. 3	1761	1669	2.007
23	417	468		10	1222	1188	-2.064	4	1995	2187	2.852
	1	2	-E,200 P	12	701	717	1.752	6	1216	1123	2.601
-24	417	396	-1.790	13	175	165	-0.539 +	7	2242	2290	-1.457
-23	1292	14.3	• 2.468	19	1749	1717	1.852	9	1775	1693	0.556
-21	722	686	2,915	16	750	765	2.990	10	474	344	2.397
-2.	632	613	-0.149	17	4/4.	393	2.371	11	1268	1212	-0.807
-18	708	697	0.630	19	441	416	2.741	13	750	710	0.187
-17	2472	2511	0.128	20	1239	1338	-1.435	14	840	760	0.644
-15	464	400	-1.507	22	563	591	-3.091	16	828	887	1.732
-14	2207	2077	-0.472	23	722	890	-0.754	17	828	980	0.819
-13	1829	1/52	-0.014 2.997	-22	965	984	-2.393	19	464	600 386	-2.769
-11	1668	1687	2.908	-21	452	515	-2.827	20	379	424	0.384
-1;	2402	2324	-2.409	-20	671	716	1.784	- 16	1066	6 1076	6.128
-8	2842	2780	1,489	-18	464	427	-0.635	-17	417	323	0.783
-7	1158	967	-3.001	-17	1214	1125	0.566	-16	816	767	0.252
-0	3:20	3911	1.123	-10	708	635	-0.970	-15	175	1061	-3.121 1
-4	3608	37.5	-0.452	-14	581	493	2.451	-13	175	376	-0.335 1
-3	4197	4839 2803	-1.270	-13	2155	2186	2.235	-12	1272	1145	9.362 2.354
-1	1898	1871	-0.905	-11	970	1017	-3.078	-10	1222	1175	-2.623
•	2655	2856	-0.146 #8.470	-10	2273	2374	-2.579	-9	1380	1423	-1.747
2	526	4,9	-3,133	-9	2033	2013	-2.497	-7	2579	2518	-2.10/
3	4154	4824	-2.968	-7	1574	1458	1.301	- 6	1012	908	-2.93/
4	1827 3851	1634	2.823	-6	1749	1468	0.284	-5	2515 1941	2657	0.894
6	2879	2922	2.155	-4	840	750	0.306	- 3	615	705	1.237
7	648	499	3,695	-3	1511	1459	1.050	-2	1409	1411	0.307
9	1732	18.9	-5.500	-2 -1	2834	2915	-0.034		701	595	9./13 9.061
1	2761	2787	-0.227	ō	1066	1219	•0.163	i	607	575	-1.406
11	2.23 1988	1087 13.4	+U.023 2.857	1	933	931 3397	-1.301 -1.74A	2	496 2033	410	-2.164
13	722	749	0.825	23	302	131	-2.032		2416	2404	-1.071

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				- 6 4	147	440	-a naa 🕈	15	1113	1044	1.804
5	1427	1418	2.722	-21	149	100			1040	1004	2.072
6	1.12	862	-2,400	-50	175	97	0.000 1	10	777	779	+0.360
5	715	680	2.538	-19	175	68	3.142 T	17			-7 400
é	4 8 1 4	1512	-3.095	-18	175	205	0.000 T	18	1099	.1114	-2.070
	3914	1216	A 132	-17	1452	1524	0.000	19	715	678	-2.101
. *	1424	1447	0.132		1383	1398	0.000	20	648	615	-1.050
1.	922	888	0.016	-10	1300	474		21	701	774	-1.596
11	202	336	-1.668	-17	0/9.	0/0	0.000	10	440	403	-2.985
12	656	737	0.691	-14	1705	1629	3.142	~~~			-2.053
	202	70	0.691	-13	1222	1024	0.000	23	101	71	-21730
10	202	404	-1 911	-12	2965	2743	3.142		2	8	
1.	41/	400	- 340		4995	4457	3.142	-24	464	324	-0.071
15	671	768	0.310		34.04	2488	1 142	- 23	640	620	2.963
16	202	89	-0.478 T	-10	2174	E100	0.1-2	- 22 '	379	322	1.585
17	743	644	-2.677	-9	1100	TIOT	0.000	- 04			-0.637
18	545	636	-3.078	-8	2246	2141	3.142	-21			4 455
		7		-7	2579	2624	0.000	-20	200	5.10	1.022
- • •		405	1 349	-6	6076	6398	0.000	-19	663	628	0.027
-10	41/	402	1 0 7 0		3037	2860	0.000	-18	1423	1385	-0.067
-17	240	022	1.7/7		4474	4079	3.142	-17	960	829	-0.503
-14	899	985	2.5/2		4470 6770	7272	0.000	-16	671	320	1.812
-13	834	862	1.627	-3	23/2	2429	0.000	- 45	7 76	5.8.0	+1.377
-12	694	683	0.636	-2	2003	2309	3.142	- 19	130	011	-2.039
-11	417	470	-2.561	-1	3703	3996	3.142	-14	424	414	- 873
	744	B4 6	1.260	Ó	6470	7279	3.142	-13	1954	1605	2.3/0
-12	/04	170	-0.054		1090	1854	0.000	- 12	1380	·1380	-2.001
-9	040	0/7	-0.074	· .	5770	6670	1 142	-11	1445	1356	1.874
-8	1.99	1065	-2.318	<u> </u>	2/19	0070		+10	2813	2632	2.084
-7	1112	970	~1.685	3	0923	6//2	3.142		7003	2404	1.620
-6	764	687	-0,713	. 4	1303	1125	0.000	- 7	3020	2308	_4 141
	844	628	-0.282	5	648	542	0.000	-8	2/59	2720	-1.101
	840	5.0	0 014	Ā	1022	1072	3.142	-7	2974	2915	0.212
	701		4 4 7 4	7	1784	1780	0.000	-6	4117	4083	0.182
-3	222	190	1,1/4	<u>′</u>	1/07	4/07	7 149 4		3312	3145	0.600
-2	496	428	0.923	6	101	100	3.175		2814	2484	-1-604
-1	496	472	-1.885	. 9	101	133	0.000 1		2440	1400	2 1 4 4
-	864	829	1.232	10	2353	2448	-0.000	-3	3414	3429	2.100
	4005	48.9	6.802	11	1843	1868	0.000	-2	3507	3379	0.111
-	1072	10.0	1 519	42	1131	1195	0.000	-1	5285	5769	-0.663
2	1.51	1142	1,210	14	1634	1440	3 142		3484	3776	1.491
3	1307	1381	-3.088	10	1201	1407	0.1-2		4288	44.09	2.486
4	1:42	1010	-1.772	14	1363	1309	0.000	· •	7448	ISER	-1.103
5	910	¯8υ7	1.223	15	496	449	3.142	· · · · · ·	3060	3350	-11200
	054	011	1.878	16	1587	1563	3,142	· 3	4827	2182	-0.705
2		040	-0 876	17	1239	1213	0.000	4	2953	3013	-2.918
	440		-0.070		270	757	0.000	5	2151	2358	2.429
8	1295	1352	-0.939	10	,,,,	544	7 142		2248	2145	n.913
9	1085	1125	-1,668	19	679	290	3.172	, ş	1808	1477	3.0VA
10	516	473	-2.034	20	175	58 .	0.000 1	<u>'</u>	1040	10//	
11	175	147	6.270 +	21	526	514	0.000	· 8	2512	2/21	0.033
12	176		=0.657 4	22	303	271	3.142	9	2073	2175	-0.08-
. 16	1/2	4 75	7 840	23	554	513	3.142	10	417	311	-0.837
15	800	0/3	2.049	£*,		- 1		11	1938	1902	2.005
14	526	503	1.301			4 - 4 -	-4 380	19	1209	1254	0.812
15	729	814	0.685	-24	100/	1040	-1.200		24E1	24.26	-0.531
·   •	. 1	8		-23	722	734	-2.135	13	2021	2120	-0.200
-11	404.	717	-2.483	-22	516	440	-2.955	- 14	922	880	-3.024
	60/	473	-2 519	- 21	1090	1219	-0.564	15	1987	2241	-3.019
-12	340	255	-3 540	- 20	0.76	1076	-0.784	16	1500	1543	-2.902
-9	485	372	-2.980	-20	0.04	046	0 220	17	607	592	-2.828
-8	770	834	-2.782	-19	900	404	4 774	14	840	870	1.832
- 7	535	494	-1.885	-16	5/2	481	1.33	10	707	374	- 146
-6	485	418	-0.275	-17	1755	1792	0.024	14	/ 4/	730	-1.140
-5	952	914	2.650	-16	1521	1508	1.063	20	367	252	-5.243
	475	145	1.111	-15	1197	1170	1.484	21	876	986	2.140
	1/2		-0.941		405	243	0.736	22	350	321	-1.992
• 3	1383	1004			1030	1 4 4 7	2.053	23	736	769	-1.016
-2	1117	10,3	0.770	-13	1237	111/	21520			3	
-1	79L	732	0.525	-12	1527	1513	2.723		E - 1		A. 484
	797	780	-0,601	-11	1631	1389	0.422	-23	263	262	-170
1	303	314	-2.390	-10	3869	4145	-0.380	+22	261	001	-1-299
	A4 7	310	2.936	20	2883	2803	-1.199	-21	143	A7	-0.319
ć		B 4	6 700	- 9	1327	1349	-2.376	-20	927	915	2.912
3	110	001	U . 776	-0	2024	2217	-0.110	-19	572	561	-1.756
4	175	2.5	-2./00	-/	2003	1601	-0 404	- 1 8	R45	AAR	0.551
5	1551	1563	-2,998	-6	2003	1949	- 0.070	-19	1051		
6	893	915	2.613	-5	1772	1660	-2.2/4	-17	1021	· • • • •	1.30
7	972	390	-0.118	-4	7133	6204	-1.823	-16	803	760	-2.485
	417	384	-1.907	-3	3554	3824	1.000	-15	876	790	-2.52
0	41/		2 974	- 2	5247	5019	2.703	-14	3315	3517	2.270
, <b>y</b>	472	440	E. 7/0		7087	8000	2.151	-13	1004	1032	2. A01
· 1.	496	510	0.241	-1	1872	1	4 4 4 4 2		1042	1031	. 105
11	485	511	05407	0	349/	3913	1,303	-16	1046	1001	-1.32
•	. 1	9		1	3863	4280	2.263	-11	1004	1124	-2.31
	152	5.3	-1.074	ź	1007	1038	0.972	-10	2291	2211	1.87
		077	-0.860	-	4731	5289	-0.550	-9	1887	1878	-0.81
-1	1.61		-2 847	3	24AP	2629	-1.094	- 6	2630	2518	-1.25
-3	452	461	-6.00/	4	2000	4 74 8	-2 610	_,	001		
-2	1131	10.9	2.528	5	1627	1/17	-2.017	-/	776		
-1	663	573	1.471	6	3902	4286	-1.0/9	-0	2001	246/	-1.04
-	581	512	0.694	7	4980	5644	-0.730	-5	4385	4336	-1.25
	354	4017	2.055		2300	2469	3.027	-4	2543	2576	2.53
1 ·	424	1037	2.271.	0 P	1832	1019	3.051	-3	1319	1236	2.19
8	429	774	<b>E</b> • <b>E</b> • <b>I</b> ·		4 140	1 . 1 1	-D. 47A		2443	230A	A.47
•	5	9	A	10	1/47	1001	- 0/0		1210	134+	
-52	161	435	3.142 Ŧ	11	1001	1739	2.016	-1	1210	1211	1.47
-24	161	46	3,142 T	12	2521	2792	2.504	Q	3440	3515	1.63
- 23	163	262	3.142 🕈	13	2000	2110	0.610	1	1871	1711	1.92
_ 9.9	143	24	0.388 <b>+</b>	14	257	693	2.870	2	1793	1673	-2.22
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	3	965	793	-0.402	-2	3011	2936	1.254	•	590	502	-1.033	
	5 6 7	2778 1956 757 679	3179 2161 /37 521	-0,094 -1.664 0.123 2.976	-1 0 1 2	1145 1243 1729 1956	969 1087 1701 2002	1.818 -2.307 1.447 1.751	10 11 12	175 656 175 922	107 666 314 927	-0.256 T 0.215 1.817 T 2.216	
	8 9	1483 1162 1867	1463 1219 1737	-1.409 -0.372 1.408	3 4 5	2424 736 1665	2608 668 1762	-2.114	13	893 2 822	961 8 847	3.139	
	11 12 13	777	677 957 1660	3.090 -1.900 1.462	6.7.1	2056 1914	2065	-1.144	-11 -10	607 960	579 940	-2.765	
	14 15	1451 3214	1018 1150	2.271	9	1593 1075	1559 1034	-0.104	- <b>6</b> -7	581 392	621 327	-8.435 1.130	
	17	1690 840	1762 864	1.286	11 12 13	143 797 1319	207 781 1481 .	2.614 T' 1.171 -2.889	-6 -5 -4	157 910 1368	383 884 1416	2.748 T -0.376 -0.047	
	2	777 598 615	743 617 615	-1.009 -2.782 -1.328	14 15 16	816 1158 671	819 1259 734	3.015 1.482 1.868	-3 -2 -1	474 . 822 169	452 864 251	0.383 1.200 -1.251 T	
	22 -22	846 2 474	934 4 480	-0.905 Q.651	17 18 19	949 101 535	917 130 482	-2.930 1.165 t	0 1 2	1099 1188 157	1192 1237	-2.408 3.057 -2.995 T	
	-21 -25 -19	581 516 432	586 525	1.291	-19	2 464 286	6 647	0.266	3	991 1584	1070	-2.725 3.080	
	-18	933 598	856 523	1.064	-16 -17 -16	694 679	652 671	-0.180 0.410	5 6 7	157 1842 143	10 <b>53</b> 323	3.000 T -0.757 3.080 T	
	-15 -14	1113 981	1160 945	-2.975	-15 -14 -13	1027 624	793 1040 614	-2.001 -2.001 1.246	9 10	663 1046 986	724 1203 860	1.574 0.308 -0.038	
	-13 -12 -11	563 97( 1868	471 967 1795	-2.148 -2.766 2.963	-12 -11 -10	1243 1504 1108	1201 1802 957	2.975 -3.117 1.214	-3 -2	2 1197 581	9 1198 544	2.234	
	-1 -9 -8	1737 694 1175	1708 589 1046	-1.511 0.790 0.901	-9 -8 -7	202 2068 899	207 1972 881	-2.721 -2.618	-1	392 797	430 812	1.471 2.102	
	-7 -6 -5	143 2031	207 1952 842	2.526 T -1.588	-6	2787	2773 2012	0.536	-24	-3 101	0 180	0.000 +	
	-4 -3	3459 1:32	3384 890	0.877	-3 -2	1434 828	1104 1408 704	-0.187 0.083 -0.613	-23 -22 -21	101 682 960	269 930 986	· 3.142 T 0.000 0.000	
	-1	1849 1943 598	1913 1877 424	-2.543 2.026 -1.357	-1 0 1	202 954 474	329 1016 504	-2.490 2.377 -0.223	-20 -19 -18	143 441 864	108 413 932	0.000 <b>†</b> 3.142 0.000	
	1 2 3	2807 1338 1807	2721 1167 - 1845	-1.664 1.194 2.837	2 3 4	828 2174 202	701 2318 275	-2.345 -2.740 -1.152	-17 -16 -15	143 679 1452	121 592	0.000 +	
	4 5 6.	4249 1136 3347	4484 1663 3565	-2.729 2.614	5	764	633 1018	1.454	-14	175	156 2894	3.142 T 3.142	
	7	172.	1745	-2.365	. 8	1113	861 1314	0.081	-12 -11 -16	1600 3226 876	1597 3420 788	3.142 0.000 0.000	•
	1 1	2329	2426 2597	-0.271 -0.475	10 1 <b>1</b> 12	790 938 192	736 92 <b>3</b> 19 <b>1</b>	-0.860 0.607 -1.747	-9 -8 -7	632 976- 1307	676 854 1243	3.142 0.000 0.000	
	12 13 14	441 143 392	377 139 360	1.487 0.964 † -1.012	13 14 15	563 743 336	533 778 338	3.030 1.583 3.073	-6 -5	4777 2743	5054 2688	-0.000	
	15 16 17	694 656 474	608 748 429	1.693 1.684 -2.744	16 17	949 485 2	1011 579 7	-2.990 1.991	-3 -2	4181 6003	4150 7205	3.142	
•.	18 19 2	640 506	663 430	-2.684 2.145	-16 -15	516 1061	536 1035	3.056	-1	101 6261	144 269 7299	0.000 T 0.000 T 3.142	
	21	h32 2	694	-0.940	-14 -13 -12	961 160 954	992 246 988	1.024 2.059 T -2.242	2 3 4	1167 101 1580	1054 112 1440	3.142 3.142 T 0.000	
	-21 -2. -19	392 516 417	4,7 534 429	-1.461 -1.915 1.270	-11 -10. -9	485 506 743	453 452 664	-2.447 -0.248 -1.394	567	101 464 1483	186 295	3.142 T 3.142 7.142	
•	-18 -1/ -16	35 101 1251	337 271 1329	2.046 2.220.9 0.557	-8 -7	624 1099	568 917	-1.963	6 9	506 2264	334 2958	3.142 0.000	
	-17 -14 -13	121. N;3 474	1239	0.719	-5	797	631 1006	-3,094	11 12	2053	1297 1998 2410	0.000 3.142 0.000	
	-12 -11	1365 1235	14.3	-0.223	-3 -2 -1	876 1517 202	829 1466 300	0.218 2.936 0.286	13 14 15	840 581 2669	701 479 2683	0.000 3.142 3.142	
	-1 -9 -8.	1 46 32 35.	961 255 331	-1.414 -1.225 1./85	0 1 2	1784 1280 202	1784 1321 324	0.816 1.964 -2.980 +	17 18 19	1231 828	1135	3.142	
	-7 -6 -5	2217 729 152	2235 173 826	-2.161 -2.189 0.517	3	656 1122	527 1060	-0.296	20 21	143 516	149 378	0.000 T 3.142	
	-4 -3	572	495	-1.944 2.498	. <del>6</del> . 7.	764 1524	07/ 812 1611	3.130 -1.049 -0.484	-24	226 3 -	206 1 423	0.000	

1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 -1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 -

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-23	161	170	2.083 4	14	1042	1106	-2.470	7	607	543	0.160
•22	922	9 J B	-0.467	15	864	857	-2.198		1665	1863	0.722
-21	506	445	-3.015	18	622	936	2,749	. 9	1660	1884	-0.447
-20	1446	1035	2.400	1/	846	674	2.077 1	11	922	925	2.745
-18	1.17	1019	0.301	19	175	240	0.259 +	12	392	377	-2.791
-17	572	475	1.426	20	545	509	0.825	13	496	406	-1-610
-16	350	336	1.562	21	905	619	-0.563 .	14	1149	1120	2.0//
-12	1061	1046	3.079	۴.	3	. 3	-100	16	1145	1222	-2.424
-13	607	499	1.082 .	-23	563	· 565	-1.530	17	111	220	2.500
-12	1380	1480	-1.305	- 22	- 545	471	2.189	18	508	459	1.762
-11	1383	1357	-1.784	-21	417 -	433	*1.774	20 1	248	302	.n.116
	1879	1709	-0.967	-19	1368	1412	1,526	•	3	. 5	
- 6	2729	2732	0.983	-18	607	575	2.137	-20	365	325	-1.650
-7	1995	1850 .	2.763	-17	392	341	0.071	-19	738	70/	3.127
-0	4711	4673	-0.648	-15	2366	2447	1.563	-17	1272	1418	0.569
-4	905	929	1.223	-14	474	407	2.981	-16	770	900	1.497
-3	3090	2925	0.132	-13	1795	1722	-1.607	-15	485	460	2.955
-2	2228	1770	-1.313	-12	1737	1578	3.047	-13	492 581	470	0.462
-	4458	5172	1.229	-10	3626	3631	-1.277	-12	770	745	-1.752
Í,	3194	3259	0.368	-9	452	361	-1.866	-11	701.	740	-2.611
2	3453	3966	-0.287	-8	927	813	-2.577	-10	893 -	638	-0.317
3	944	1049	-2.438	-7	303	10/	-1.091	- 4	1583	1377	-1.674
- 5	2144	2318	-0.576	-5	1840	1503	2.384	-7	1455	1277	1.030
6.	1051	1016	-1.616	-4	1784	1732	0.995	- 6	1099	1006	1.375
7	1117	1217	-2.304	-3	2359	2204	1.488	-5	465	404	2.058
8	1951	2279	-0,102	-2	2451	2406	2.210	- 4	2061	1742	0.943
10	429	1735	2.169	-1	1770	1575	-2.315	-2	1145	976	2.699
11	2565	2791	1,880	ĩ	722	520	-2.964	-1	991	895	1.634
12	1717	2014	0.677	2	858	896	-1.581	0	2262	2234	1.642
13	607	664	-2,883	3	2081	2138	-1.917	1	1068	1709	-0.508
14	1193	1280	-3.069		1801	1768	-1.930	3	1231	1145	-0.982
16	905	920	-2.514	6	1815	1713	-1.476	4	516	469	0.394
17	976	959	-3.067	. 7	1201	1114	-1.957	5	2093	2236	-1.818
18	405	314	-1.474	8	2083	2183	1.614	<u>0</u>	1612	1550	-0.032
20	268	342	-2.371	10	1323	134R	-0.310	á	405	353	1.696
21	615	532	-3,134	11	1184	1093	1.024	9	797	775	2.145
22	624	580	-0,402	12	1821	1966	1.444	10	1473	1670	1.588
- 7.4	3	2	1 676	13	927	1023	-2.695	11	1007	1100	2.279
-23	200 545	488	0.554	19	671	659	-1.765	13	615	607	2.005
-22	701	723	-0.193	16	743	794	0.601	14	590	575	1.396
-21	143	183	1.905 1	17	828	827	-0.653	15	303	- 316	-2.178
-20	175	114	-0.514 7	18	648	562	-2.019	16	834	769	-2.639
-17	175	147	2.174 1	19	405	615	-0.702	18	632	593	-0.519
-17	864	833	3.032	21	1117	1158	-0.624	•	4	. 0	
-16	1772	1673	-1.971	•	3	4	-	-23	1247	1238	0.000
-15	2158	1917	2.702	-22	417	481	0.661	-22	1012	997	0.000
-14	1/84	1/59	-2.506 1	-21	350	360	-0.500	-20	143	124	3-142 T
-12	572	532	2.408	-19	976	1035	1.254	-19	143	104	3.142 +
-11	2355	2462	2.631	-18	545	557	0.299	-18	816	736	3.142
-10	1494	1327	-2.876	•17	143	238	-1.854 T	-17	864	806	3-142
-9	1210	1020	-1.270	-10	1483	1442	2.204	-10	175	245	0.000 T
•7	4203	4223	0.043	•14	496	279	2.844	-14	1251	1187	3.142
-6	5934	6561	-1.047	-13	545	419	2.619	-13	175	105	3.142 +
-5	1821	1627	1.775	-12	870	899	-2.642	-12	2451	2238	n.00ú
-4	1459	1318	2.370	-11	1099	1047	-1.251	-11	2763	2658	0.000
-2	2.78	1936	-0.190	-10	1705	1656	1.290	-9.	1149	1070	0.000
-1	4966	5175	2.606	-8	686	572	2.007	-8	2794	2733	0.000
ŝ	1772	1640	3.075	-7	1180	1070	-0.583	- 7	1402	1363	3-142
1	2267	2281	0.028	-6	2543	2430	0.265	-6	1108	944	3.142
3	2293	2111	-0,992	-5	2110	1357	-1.955	-4	2002	2453	0.000
4	2375	2450	-1.560	-3	2038	1951	2.061	-3	2276	2174	3+1=6
. 5.	1531	1565	0.009	-2	1948	1765	2.678	-2	405	538	3.142
6	2073	2.054	8,917	-1	3253	3447	-2.612	-1	3220	3375	3.142
7	/9/ 432	/98 674	U.012	0	1651	1619	*2.010	0	1402	1399	3.142
ÿ	870	758	8.085	2	1634	1666	-2.765	2	47 n4	. 2074	0.000
<b>1</b> .0	1113,	1062	1.545	j	1547	1420	-2.299	. 3	965	750	-0.000.
11	1761	1754 .	0.715	4	764	693	2.193	<b>4</b> -	143	124	3.142 +
12	686	/53	· U.716	. 5	1890	965	0.166	5	2134	2226	0.000
**			~~.~~		300	700	-1.0//	0	7267	1755	0.000

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,	701	642	0.088	-2	3065	3212	2.811	-3	143	234	2.609 +
í.	143	248	1.142 +	-1	1511	1398	+2.949	+ Ž	1153	1040	-3.064
ŏ	1911	1740	3.142	- i	1117	1013	-2.145	-1	1674	1722	2.831
16	3444	3548	3.142	-1	1717	1701	2.645	. 0	1632	1544	1.604
11	2430	2276	8.908	2	2328	2245.	3.010	. 1	764	770	1.729
12	1:32	1040		3	1235	1229	-0.475	2	2255	2233	-1.577
13	1427	1237	3,142		2888	3206	0.342	3	1987	2048	1.673
14	1514	1432	3.142	5	1398	1412	1.282	4	1365	1263	-0.195
15	671	671	3.142	6	554	538	-2.878	5	1987	1971	-1-189
16	465	427	3.142	7	722	756	-0.720	6	1001	938	-0.201
17	694	681	3.142	8	535	532	-2.656	. 7	2177	2504	0.310
18	175	42	0.000 🕈	9	392	267	1.049		1149	1170	-0.594
19	379	283	9.008	10	743	707	-0.206	9	1243	1342	3.141
2	834	741	3.142	11	1108	1100	-2.527	10	1327	1342	1.895
21	663	529	0.000	12	496	430	2.637	11	803	815	+2.200
	÷. 4	1		13	864	, 840.	-0.843	12	686	624	-2.171
-24	679	715	9.106	14	202	129	0.109 🕇	13	1153	1160	2.134
-23	535	516	-0,181	15	729	685	2.139	14	899	925	-3.020
-22	516	6u3	-2.889	16	452	466	0.671	15	910	923	-2.304
-21	646	7u1	2.831	17	175	210	-0.345 T	10	336	324	2.540
-20	615	657	0.205	10	663	567	1.089	17	367	. 373	•2./30
-19	143	142	8.983 T	19	572	560	0.520	10	60/	205	-5+190
-18	1483	1587	2.225	20	1323	1159	-0.61/			7.5	0 334
•17	722	644	2.871	•	4	3		-20	330	317	2.224
-16	671	693	-2.837	-23	417	412	0.204	-14	200	279	1.044
-15	870	912	2.898	-22	764	/30	-0.//4	10	050	040	-9 898 -
-14	179	278	1.082 T	-21	101	202	-0.111 T	-1/	. 248	788	1 124
-13	1534	1573	-8.715	-20	996	1011	1.740	-10		/72	- 0.081 <b>+</b>
-12	1012	994	-2.236	-19	927	903	2.357	-19	101	171	4 552
-11	2346	2321	-2.1/3	-18	0/0	¥10	-0.55/	-14	1104	4 7 4 8	-1.572
*10	3221	3234	-1.303	-17	1413	1300	1.195	-10	1344	2719	8-130
-9	224	428	0.030	-10	204	801	1.916	-12	447	160	4.115
-8	3167	3152	1.991	-17	. 000	011	-4 686	-11	870	874	
-/	933	1000	2.383	-14	111/	1077	-1.033	-10	8.8	203	-0.950
- 0	2013	5550	1.931	-19	1760	1744	-1 471		404		0.486
-7	000	72/	1.844	-12	1/26	2740	-0.045	- 0	1584	1619	0.914
	1919	1030	.1.700	-11	1331	1100	-1.931	-6	1136	1072	1.608
	2027	2847	0.514	- 10	1459	1247	4 409	-5	2103	2053	1.603
	674	1140	2 898		1778	1668	0.526	-4	757	736	0.926
-	4702	1819	-0.427	-8	701	486	1.723	- 3	2226	2325	-1.978
	3448	4177	-8.669	-6	2353	2433	2.650	-2	1201	1099	-2.987
5	2044	2082	-1.091		2396	2283	1.818	-1	2141	1918	1.562
• 3	1046	1058	-1.970	-4	143	271	· -0.546 +	õ	1117	1057	-1.897
Ă	2840	3126	-0.870	+3	2278	2069	2.171	i	1887	1896	-0.952
ŝ	2163	2274	-2.895	-2	1122	975	3.086	2.	2160	2350	-0.586
	1372	1429.	2.076	-1	2935	2906	-2.044	3	1126	1063	-0.054
7	2865	2558	0.492	ō	2181	2173	2.011	4	526	519	-2.394
8	852	831	0.059	i	1416	1279	-0.062	5	1214	1240	-0.237
, j	2507	2652	2.386	ž	2624	2634	-0.680	. 6	764	. 739	1.299
10	1376	1474	1.431	3	770	642	3.118	- 7	545.	439	-2.359
- <b>ii</b> '	1514	1678	0,965	4	822	700	-2.907	. 8	615	630	1.972
12	949	949	1,002	5	1865	1844	-1.910	9.1	1625	1774	1.764
13	1507	1401	-2.719	6	1956	1883	-2,343	1.0	986	1089	2.025
14	1681	1883	-1.034	7	1603	1700	1,581	11	441	438	2.752
15	1311	1348	-2.058	8	656	623	1.204	12	379	411	1.400
16	581	565	-2.611	9	905	903	-1.626	13	750	784	g.235
17	1027	1016	-9,361	10	1445	1592	1.316	1,4	1099	976	-2.076
18 -	350	382	-0.160	11	1402	1449	1.786	15	365	282	-2.346
19	607	541	2,435	12	485	420	-2.569	16	876	803	0.142
20	336	268	0.432	13	870	647	-2.016	•	5	. 0	
21	350	362	0.243	14	598	601	0.810	-22	320	343	3.142
•	4	2	• _	15	. 175	248	-1.072 T	-21	286	243	3.142
-24	656	723	9.176	16	1413 -	1346	-1.354	-20	392	347	3.142
-23	694	657	0.394	17	572	561	-2.183	-19	632	554	3.142
-22	365	352	1.624	18	632	552	2.460	-10	143	80	3-142
-21	464	395	-1.591	19	320	293	-1,188	-17	143	182	0.000
-2.	143	140	-1.810 T	•	- 4			-16	650	700	3.142
-19	656	658	-2.670	-20	365	381	-2.142	-15	1473	150/	3,142
-18	840	041	-3.103	-19	365	450	*1./71	-14	241	571	7.004
-17	1,71	1115	-2.071	*18	101	109	1,/00 T	-13	1274	1229	7.000
-16	1323	1922	2.760 2.761	-17	370	346	2.783	-12	1/7		3.144
-17	929	450	2./90	-10	6/7	/ 10	-1.07/	-11	770	70/	3.144
-15	1.01	1170	- 4.300	- 12	301	677	-U.UJO 2 403	-10	1340	1274	-0.000
-13	202	412	T 9044	-14	470	100	3.103		22/3	#311 1020	3.142
-12	1202	1107	2.V93	-13	378	377	-0.417		1173	1020	3.142
-11	9/0	676		-12	1730	1100	4 897	-/	1535	1420	7.000
-10	017	297	4.470 1.654		1707	1112	1.840	-0	1283	404	3.142
-7	1903	2740	1.051	-10	1414	1444	1.007	-9	1717	3442	9.000 1
-0	3084	1100	1 N74	-7	1934	7444	-2.400		3/3/		7.000
	4041	1920	2 534		3130	2148	8 440	-0	2724	2840	Q • UUU W 4 4 %
-0	2714	1497 And	1.944	-/	1140	1285	-0.921		E/0/	27-9	3.246
-7	. 9487	DALL	-1.107		444 A44			-1	10.68	375	4.004
-3	1887	1810	-2.521	•7	1056	044	-0.381	1	1010	1873	9.000

2	2:53	1840	3.142	-1	1383	1277	-2.920	0	1808	2005	-0.229
3	1801	1643	0.000	0	3420	3612	2.763	i	526	356	+0.680
4	1.71	000	0.000	i	2298	2275	-2.609	2	1184	1135	-0.054
5	143	247	8.000 +	2	2531	2590	-0.417	3	1272	1194	-0.930
6	3799	3095	0.000	3	1264	1305	0.434	4	1527	1468	-0.597
7	1810	1731	0.000	4.	175	200	-2.958 +	5	2223	2301	0.648
8	1259	1118	3,142	5	464	373	0.683	6	1303	1308	-0.009
9	516	466	3.142	6	392	386	2.755	7	1001	1071	-2.069
1.0	1131	952	3.142	- 7	624	620	1.031	8	1714	1861	2.318
11	905	750	0.000	8	1175	1270	-0.527	9	1541	1721	2.310
12	1531	1485	3.142	9	1438	1568	2.595	10	485	501	-2.477
13	572	593	3.142	10	1531	1688	-3.116	11	500	574	2.739
14	175	42	0.000 +	11	405	339	-2.618	12	485	419	2:384
15	1942	996	3,142	12	632	664	-1.075	13	986	1010	-1.614
10	429	365	3.142	13	777	712	-2.302	14	590	540	-0.219
17	1387	1255	0.000	14	202	192	1.657 🕈	15	474	495	0.528
18	379	360	0.000	15	816	718	-0.210	16	615	577	-1.793
19	143	30	3.142 🕈	16	452	347	-1.977		5	5	
	5	1	•	. 17	526	491	1.471	-19	320	395	2.842
-23	864	. 965	2,515	18	405	343	0.131	-18	607	622	-2.418
-22	101	326	1.134 T	19	350	367	-1.282	-17	202	249	3.123
-21	590	643	0.277	•	5	3		-16	757	706	0.100
-20	679	773	2.723	-22	417	549	1.636	-15	1022	1166	-1.351
-19	764	847	2,551	-21	858	938	2.018	-14	671	739	-2.177
-18	722	773	-1,231	-50	797	674	2.840	-13	392	423	1.559
*17	663	633	2,871	-19	101	206	-1.080 T	-12	485	492	-0.163
-16	803	799	2,916	-18	1149	1125	0.843	-11	809	772	-2.23/
-15	1080	1330	-1.093	-17	581	556	-1.925	-10	506	528	2.830
-14	441	503	-0.823	-16	1544	1557	-1.217	- 9	1051	1079	1.458
-13.	938	929	3.043	-15	954	982	-0.956	-8	852	757	-2.576
-12	1012	1269	-1,440	-14	392	397	-1.030	-7	1251	1250	2.149
-11	834	83/	-1.022	-13	1153	1189	-0.875	-0	1434	1508	1.492
-10	1307	1351	-2.038	-12	1476	1499	-0.719	-5	916	793	-2.35/
• •	1077	1581	2.001	-11	4/4	38/	2,984	-4	1050	1149	-2.047
- 8	. 10/2	106/	1.110	-10	1/46	1759	1.426	3	1131	1198	-0.364
- 14	2103.	2050	1.201	- 9	598	562	-0.1//	-2	933	850	-1.4//
	2277	2/53	2 201	- 0	1077	3260	2.200	-1	1237	1140	*2.270
	1482	1416	-8 701	-/	1085	1016	1.034	U A	1024	1962	-0./2/
-3	777	748	-1 570	-0	1054	1010	-0.200	1	444	774	0.000
-9	094	059	-2 425	- 9	1920	277	-7 104	2 1	444	3/4	-2.220
_1	9477	2464	-0.922		3048	3012	-1 474	3	449	309	- 0.407
-	1102	1115	-7 380	-0	1202	1242	-1.4/4	-	1040	016	1.707
ň	1818	1841	-2 157	-2	790	1205	-2.301	2	1633	1030	2.210
5	9174	2164	-0 584	-1	2085	1040	-0.710		11026	10/0	2.413
3	624	539	-1.747	1	1008	1993	~U.717 +0 840	4	1104	1000	2 1
- i	283	2288	2,025		2300	2307	4442		143	448	1.202
5	1468	1757	1.851	2	1046	003	0 720	10	462	145	-1.020 1
6	976	973	0.913		091	1046	-1 250	11	001	1040	1.007
7	970	902	1.480	5	1080	1064	2.251	12	770	878	-0.62/
8	1662	1851	3.089	í.	1342	1380	1.704	13	1022	639	-11.02/
9	1531	17.3	0.030	7	175	169	-0.901 🛨	14	392	111	0.406
19	572	6.6	1.266	Á	175	337	-2.322 +	15	572	634	-0.266
11	1551	1733	2.431	ŏ	743	734	1.191		6		
12	1361	1415	-1.288	10	485	476	1,930	- 20	736	743	3.142
13	1184	1219	-0.829	11	1284	1258	-1.940	-19	615	652	3.142
14	474	514	-1.710	12	169	137	+2.413	-18	143	107	3,142 +
15	954	1025	-0.873	13	722	680	1.556	-17	803	A13	3.142
16	750	821	-0.329	14	. 590	652	-1.561	-16	429	455	3.142
17	464	429	0.840	15	1303	1251	-1.449	-15	679	602	0.000
18	286	273	2.365	16	336	328	-2.879	-14	581	485	3.142
19	893	841	0,599	17	944	822	2,986	-13	175	146	9.000 +
٠	. 5	2		18	143	141	-3.124 🕇	-12	1644	1719	-9.000
-23	379	363	-1.425	•	5	4		-11	816	500	0.000
- 22	535	461	-1.089	-21	648	718	-1.805	-10	722	694	3.142
-21	365	348	-1.808	-20	405	559	-1.486	-9	1007	A77	3.142
-2v	809	821	2.686	-19	572	735	1.931	-8	1056	1003	3.142
+19	350	239	0,917	- 18	535	693	2.999	-7	2577	2407	3.142
-18	535	427	2.123	-17	743	864	-2.836	-6	175	44	0.000 <b>†</b>
-17	554	534	2.068	-16	572	562	2.199	-5	485	356	n.000
-16	1637	1717	1.107	-15	417	424	1.903	- 4	175	100	3.142 4
-15	648	525	-0.867	-14	922	1049	-1.773	- 3	1821	1712	3.142
-14	648	680	2.997	-13	. 590	529	0.326	-2	1587	1417	0.000
-13	1284	1157	-0.454	-12	1264	1307	1.003	-1	1517	1430	9.005
-12	3604	3712	-0,810	-11	893	938	-1.605	0	995	775	3.142
-11	485	361	0.992	-10	143	319	-1.186 🕈	1	485	243	9.000
-1:	5003	20.9	1.787	-9	1487	1384	0.469	2	656	629	9.000
-9	2151	2022	-1.388	-8	1007	850	-1.932	3	965	1060	9.009
-8	828	824	-0.577	-7	1017	915	-3.086	4	175	252	0.000 🕈
-7	991	942	2.073	-6	526	417	0.640	5	1251	1077	3.142
-0	1551	14.6	1.809	-5	1243	1169	1.111	6	1462	1300	3.000
- 5	496	455	2.375	-4	429	461	2.308	7	175	143	3.142 1
-4	2443	2274	2.236	-3	2667	2850	-2.297		1797	1690	3.142
- 3	944	856	2.966	-5	2278	2244	2.590	9	452	334	3.142
- 2	2756	5621	-1,244	-1	1353	1395	0.750	10	922	749	3.142

						470		- 0		437	- 7 844
12	1922	1004	3.172	19	286	303	3 491		445	444	4 152
13	852	786	0,000	. 17	441	403	+1.730	-0 +7	1046	1133	2.159
14	516	433	3.142	•	6	3		-6	809	728	3.003
15	1334	1111	0.000	- ? 1	365	394	-1.858	-5	954	968	-0.544
16	1887	1613	0.000	-20	535	551	-0.122	-4	840	829	-0.462
17	143	4	0.000 🕈	-19	101	235	-0.484 🕇	-3	736	710	-1,941
•	6	<u>1</u>		-18	379	431	-2.266	-2	916	946	-2.11/
-21	729	6u1	2.700	-17	1307	1382	-1.261	-1	679	632	-2.240
-2	718	741	-1.033	-16	/01	690	-1.314	. 0	495	481	-2.821
-14	412	/13	-4.444	*15	472	200	2.200	1	232	484	-2.343
•17	705	787	-1 753	-1-	1172	1371	2 402	2 1	770	202	2.430
-16	846	935	-0.842	-13	1596	1571	2 905	Å	516	517	2.205
-15	563	536	0.059	-11	392	320	-1.941	5	770	A91	2.654
-14	840	740	1,207	-10	686	. 564	2.035	6	464	457	1.758
-13	441	379	1.643	-9	1656	1559	2.742	7	345	417	1.003
-12	441	489	-2.535	- 8	624	527	2.964	8	303	302	-2.740
-11	1497	1562	2.531	-7	1210	1179	0.478	9	143	102	-n.850 <del>1</del>
-1	1544	1582	1.400	-6	1104	1045	0.392	30	701	201	0.043
-9	1386	1375	0.040	-5	1714	1608	-0.361	. 11	1108	1112	-1.723
-7	602	/22	-0.493	-4	2000	1945	-0.812	12	883	739	-2.194
	675	450	3.042	- 3	1094	1002	2.909	-10	100		
-5	1259	1376	-0 687		1506	1687	-2.504	-14	101	142	1.142 +
-4	944	979	=1.141	1-	454	1000	-1.523	-17	572	449	0.000
-3	1.27	1024	-1.338	· 1	1104	1081	1.056	-16	648	620	0.000
-2.	2496	26.9	-2.263	2	624	633	1.960	-15	143	224	3.142 +
-1	1708	1514	-2.599	3	686	806	-2.089	-14	429	367	-0.000
	2963	2253	2.576	4	888	884	0.951	-13	1749	1793	n.00V
1	336	311	-1,471	5	1795	1903	0.958	-12	1117	. 1096	- 3.142
2	417	412	1.493	6	662	859	-0.462	-11	365	354	3.142
3	2223	2348	2.336	7	485	334	1.035	-10	1473	1413	0.000
	965	1034	1.4/2	8	6/1	66/	1.755	-9	175	44	3.142 T
	1331	1363	2 354		1/5	193	-2.024 T	- 8	2117	2061	3.1.42
. 7	2146	2431	-1.489	. 10	350	361	1.933	- 6	175	44	3.142 #
В	1746	1/01	-0.870	12	634	807	2.411	-5	171	697	3.142
ÿ	175	195	2.672 +	13	563	549	-1.546	-4	933	913	0.000
1	965	649	-2,935	14	/08	607	-1.583	- 3	933	842	0.000
11	417	353	-0.922	15	143	70	1.138 +	-2	452	368	3-142
12	175	90	0.709 <b>T</b>		6	4		-1	1699	1519	3-142
13	175	192	-2.199 T	-19	496	459	-2.736	0	736	623	0.000
14	405	439	-0.037	-18	496	446	-2.979	1	216D	1948	0.000
15	485	536	0.310	-17	554	525	1.853	2	526	465	0.000
10		74	-2.069 🕈	-16	516	556	-2.255	. 3	2669	2689	3-142
17	703	219	2.004	-15	4/4	771	-2.000	1	1210	1151	3+142
- 22	345	4 7	-2 145	-14	141	241	1.040	2	1/2	· 1/ ·	1 142
-21	545	454	1.731	-12	143	245	2.981 +	7	350	281	3.142
-2	111	227	-1.908 +	-11	/22	733	-0.316	8	177	416	3,142
-19	101	282	2.994 +	-10	545	426	1.372	9	1483	1443	3.142
-18	545	473	0.951	-9	143	77	1.020 🕈	10	175	228	n.000 🕇
-17	433	912	+0.288	- 8	177	765	-2.593	11	1066	1041	0.000
-10	379	425	-2./81	-7	632	613	2.764	12	330	275	3.142
-15	554	489	2,767	-6	840	924	1.494	13	516	497	n.000
-14	044	953	-0.994	-5	1117	1028	-2.378	.14	1409	1181	ŋ.QOù
-13	15//	1598	-0.2/2	-4	5505	2227	-2.553	15	1051	950	g.DCU
-12	1387	1034	0.101	- 3	143	207	-1.092 T	- 7.4	504	1	
-1	1 76	1045	-0.143	-2	1857	1708	-U.VCJ -D.GRM	-21	520	552	-9.99U -0.049
-1 -1	2.2	3.6	8.809 4	-1	671	672	1.543	•19	302	105	- 1.034
- H	1323	125	2.289	ň	590	520	D.560	-18	101	152	-1 405
+7	148.	1435	-2.522	2	870	823	-0.260	-17	624	654	-1.820
-6	1.37	993	-1.839	3	708	685	0.770	-16	101	245	-2.644 +
-5	172.	1726	1.961	4	1315	1432	1.303	-15	697	549	n.674
••	1.61	911	3.120	5	496	418	-0.913	-14	715	686	9.435
-3	1596	1522	-1.392	6	615	645	-2.200	-13	143	174	-9.454 🕈
•2	474	371	-1,796	7	1046	1062	2.250	-12	615	' 542	2.12+
-1	1372	1473	2.924	6	506	571	-2.904	-11	1971	1115	1.991
	16(9	1543	-0.082	9	773	865	-2.737	-10	1346	1399	- 7 . 6 9 2
1	2422	24.54	U.009 1 46*	10	1 2 3	908	2.021	- 9	441	410	-2.345
ć	11.66	1247	1,733	11	<b>7</b> 5 1 5 1 5 1 5 1 5 1 5 1 5 1 5 1 5 1 5	231	-1-1-4	- 7	970 494	54/	3-941
4	5/2	400	-4.456	16	715	660	C. 22H	- /	6027	570	-1+1/4
5	946	NA4	-1.447	13	764	776	0.397	-5	6-)0 R≰û	517 A16	-1.4945 3.166
é	581	558	0.041		6	5		-4	624	. A10	- 377
7	202	133	-0.012 4	-17	365	290	-0.151	- 3	1162	1123	-1.514
ö	1625	1/42	-2.617	-16	581	610	-1.604	-2	1455	1315	2.553
Ŷ	1 99	1172	2,586	-15	303	317	-1.660	-1	1130	1.55	1.575
1 -	615	5.6	-1.739	-14	730	826	0.642	0	1046	955	-1.235
11	8¢3	843	-2.831	-13	303	294	-0.830	1	1597	1558	1.875
12	882	952	2.690	-12	581	661	-3.130	2	175	336	-1-634 🕂
13	632	261	U.729	-11	679	771	1.602	3	1323	133¢	1.055
		88.8		- 1 0			D 223	=	1 5 4 4	1 4 4 7	

5	1334	1445	2.596	-15	506	565	0.879	12	143	70	n.000 +
6	554	561	-0.062	-14	452	476	-0.312			1	•
7	694	839	-1.014	-13	111	199	-1.352 🕈	-18	286	309	-0.676
8	175	213	-3.097 🕇	-12	506 -	445	1.029	-17	101	149	-3.082
	828	884	-2.025	-11	764	570	0.923	-16	200	539	2.320
11	175	49	-2.365 +	-10	1158	1424	-1.914	-15	464	435	0.303 I -2.888
12	535	491	2.721	-8	729	739	1.896	-13	809	852	1.419
13	642	539	-0.176	-7	828	810	2.516	-12	949	1131	0.930
14	750	7 0 3	-0.275	-6	1259	1282	-2.484	-11	143	120	1.022
15	764	737	3.024	-5	722	682	2.575	-10	624	677	-2.633
	700	2	-7 604	-4	485	412	-0.487	-9	590	591	-1.540
-19	303	285	-0.113	-3	572	574	-0.43	- 8	1280	1302	-1.200
-18	572	641	-0.561	-1	750	693	1.479	-6	607	556	-1.755
-17	101	91	0.207 +	0	128	259	-1.437 +	- 5	175	129	-2.414 +
-16	417	380	\$,037	1	607	606	-1.111	-4	350	273	2.779
-15	615	635	-1.129	· 2	1175	1244	2.016	-3	175	114	1.903 4
-14	648	655	-0,042	3	976	1035	2.343	-2	10A0	1045	0.357
•12	470	1004	1 102		1037	1009	-2.431	-1	980	957	1.205
-11	694	711	-0.385	6	464	425	-2.167	1	729	684	-0.212
-10	701	789	2.010	7	516	583	-2.077	2	797	1029	-0.677
-9	1726	1768	-3.108	8	336	385	-2.940	3	784	774	1.756
-8	1846	1868	-2.234	9	474	473	- 1.645	4	175	187	2.009 t
-7	202	230	2,586 4	10	452	366	-0.922	5	590	639	-0.820
-0	1679	1981	2.132	11	701	649	-1.119	<b>4</b> .	220	562	-0.735
- 2	1323	1289	-1.190	- • 6	A 70	206	A 457		250 5e1	200	2.45/
-3	764	75ù	-2.936	-14	303	365	-1.816	9	648	740	-0.669
-2	1500	1624	0,335	-13	350	328	2.224	10	722	713	3.111
-1	2239	2432	-0.029	-12	1323	1506	1.450	11	417	371	2.731
	1117	1047	0.030	-11	286	293	0.677	12	336	290	-0.138
1	1387	1454	1,200	-10	286	231	-0.082	•	8	2	
ž	704	000	-2 002	-9	101	248	-0.289 T	-15	101	290	2.381 T
4	800	706	2.234	-7	545	543	-2.743	-14	336	390	2./21
5	1162	1295	-1.342	-6	516	546	0.217	-12	143	187	1.319 4
6	554	567	-1 987	-5	554	562	-0.385	-11	1099	1121	2.598
7	715	137	2.670	-4	834	846	-1.194	-10	1188	1265	-2.567
8	526	520	1.57/	-3	526	505	-1.634	-9	876	817	-1-917
,9	708	736	-0.651	-2	496	460	1.813	-8	464	342	-2.721
17	632	0.53	-3.132	-1	452	378	3.019	-/	663	679	1.2//
12	663	546	2.J70 8 140	1	67U 628	450	2.003	-0	474	404	1.240
13	656	677	0.539	2	392	495	0.922	-4	954	865	1.336
14	563	452	0,991	Ĵ	248	345	2.692	-3	1405	1424	-0.272
	7	3		4	101	81	1.739 🕇	-2	1066	1099	-0.629
-19	686	617	-1,639	5	/50	721	0.691	-1	175	278	0.964 🕈
-10	545	558	-1.938	6	208	251	-1.885	0	175	329	0.062 +
-16	708	/60	-2.383	, ,	202	182			590	264	-0.898
15	329	282	2.404		722	. 734	-1.256	د ۱	1108	1 2 4 1	2.007
-14 +	1251	13.7	1.616		ē	0		4	175	258	-2.35/
-13	1.37	1113	1.612	-19	535	534	0.000	5	485	416	-2.557
12	143	259	-0.106 🕇	-18	1153	1211	0.000	6	743	719	1.874
-11	1.94	1062	2.553	-17	106	69	0.000 1	7	590	630	0.571
-1	404	444	-2.000	-16	320	242	-3,142	8	172	220	0.569 4
-8	175	151	-0.733 +	-19	101	186	0.000	. 10	1/2	10/	2.020 1
-7	736	648	-2.973	-13	1051	1012	3.142	11	663	630	-1.000 T
- 6	1 01	822	-1.135	-12	143	259	3.142 +		8	3	- () • 2 • 1
-5	1247	1253	-1.843	-11	1094	925	0.000	-16	545	478	1.629
-4	175	212	0.507 🕈	-10	143	128	0.000 <b>t</b>	-15	656	567	1.164
-3	792	7.1	1.878	-9	429	268	3.142	-14	268	283	1-556
	1472	1254	-1.4/4	-8	143	58	0.000 T	-13	500	530	2.339
-1	022	/ 30 974	1 333	-/	143	103	3.142 T	-12	101	204	2.752 T
1	784	722	-1.062	-5	1051	933	0.000	-10	816	801	-0.203
2	175	289	-1.341 +	-4	1323	1258	0.000	-9	417	417	-1.336
3	1 22	960	1.815	- 3	441	406	3.142	-8	607	589	2.341
4	A93	839	1.623	2	1323	1187	3.142	- 7	1007	973	-1.45/
2	590	621	-0.274	<b>1</b>	175	60	3.142 T	- 6	441	450	-9.645
7	581 70.	44D 844	2.204	0	1405	1325	• 0.000 <b>•</b>	-5	441	359	3.109
Á	940	040		1	1/5	/U #34	U.UUU T 2.449		441 746	421	-2-617
ş	640	646	-2.003	۲ ۲	900 91n	173	3.147	- 2	679	272	-1.781
1	175	293	1.844 9	Ă	893	831	3.142	-1	563	509	1.031
11	379	326	-0.776	5	603	748	3.142	õ	350	347	-1.784
12	441	377	-1.070	6	226	227	0.000	1	991	997	-2.840
13	441	384	0.799	7	202	248	0.000	2	1188	1196	2.162
	984	4	1.171	8	572	491	0.000	3	175	113	0.919 +
-10	248	244	1.092	y	141	200	U.UUU 1.142 4		172	100	1.235 1
	=0·			10	173	20	3+472 T	2	3/7	394	1.972

,	864	740	-1 680	_							
à	143	102	-2 014 -	2	545	543	3.033	1	899	1010	2.648
9.	268	243	-0.941	3	143	175	-1.863 🕈	٠	10	· 2	
•	8	4	*****		705	1135	-0.633	-8	365	432	0.092
-15	392	421	-0.207	2	330	210	2.162	-7	506	578	1.185
-14	656	795	-1.487	2	870	44/ 501	2.2/2	-6	4 <u>1</u> 7	446	-0.642
-13	. 615	611	1.752	· ·	312	241	-0,795	-5	350	346	-0.667
-12	268	361	0.975	-14	320	270		-4	464	611	3.116
11	268	347	-1.554	-13	524	579	-3.009	-3	656	882	-3.107
-1	115	2:6	0.687 +	-12	514	5/7	2.738	-2	101	128	-2.224
-9	392	338	-2.570	-11	581	207	3.122	-1	392	477	-1.877
-8	1012	1096	-2.653	-10	764	841	-1 504				
-7	485	445	2.545		420	412	-1.304				
-0	615	633	1.319	-8	379	327	-0 071				
	526	563	0.843	7	496	531	*0 500			•	
	840	892	-0.587	-6	143	217	-0.461 +				
- 3	124	261	-0.045 🕂	-5.	916	947	1.042				
-2	417	465	0.900	-4	392	422	0.017				
-1	554	570	-0.431	-3	143	155	-1.544 🕈				
	111	215	0.873 🕈	-2	640	732.	2.783				
5	024	714	2.324	-1	392	356	-0.895				
1	804	/18	3.068	· 0	535	548	-1.185			•	
3	500	461	-2.793	1	1046	1256	2.848	• •			
5	220	579	2,810	2	922	1042	3.077				
	530	-03	-2.365	3	392	413	-2.247				
2	201	034	-1,888	· • •	365	409	-1.913				
Å	<b>654</b>	200	2.843 7	5	474	561	2.485	•			
	,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	2190	u. 452	6	143	172	1.641 +				
-1	268	2=1	4 348	· 7	143	202	0.875 +			1	
- 0	224	274	2,700	· · · · •	9	. 3					
-8	268	2/0	-1.902	-12	607	645	-0.275				
-7	200	105	-1,904	-11	1001	957	-0.737				
-6	616	193	0.900 T	-10	598	241	0.783				
- 5	284	470	-0.073	-9	248	237	2.617				
-4	800	200	-2.200	- 8	535	429	-1.985				
· -3	485	5.5	0.937	· -7	101	46	0.561 +		•		
- 2	320	202	-1.040	-6	392	358	2.319				
-1	514	515	-1.909	-5	405	360	-1.664				
-	624	202	1,993	-4	143	69	2.218 🕇				
i	379	796	1.095	-3	694	651	1.627			*	
2	248	344	1.009	-2	392	380	1.163				
3	452	240	-0.304	-1	143	144	-1.451 🕇		•		
Ă	441	405	-0.100	0	379	410	2.218			•	
•		-05	-0,178	1	797	816	2.565				
-16	764	608		2	554	466	-1.623				
-15	78	164	1 143 4	3	143	84	2.613 🕇				
-14	1113	194	3,142 T	1	320	278	2.057				
-13	101	45	3 440 4	5.	101	620	-1,534 t				
-12	708	7	0 000		. 9	4					
-11	474	472	4 149	-9	516	609	-1.569				
-1	777	761	3 142	-8	303	405	-0.512				•
-9	365	356	0.000	-/	648	797	1.027				
-8	365	323	3 142	-0	379	386	0.454				
-7	143	227	3 142 4	-5	392	372	-0.078				
-6	976	873	0.000		679	779	1.085				
-5	876	8.5	0.000	-3	268	271	0.248				
- 4	143	146	3 140 -	-2	429	368	-1.822				
-3	143	260	3.142 +	-1	563	594	2.608				
-2	784	679	0 000	0	405	376	2.221				
-1	715	615	0.000	1	330	331	-1.095				
2	143	116	3.142 #	· •	10	0					
1	624	519	0.000	-10	590	548	0.000				
2	452	247	3.142		10	91	0.000				
3	797	726	3.142	- 9	303	259	3.142				
4	640	488	3.142	-/	/20	687	0.000				
5	736	489	3.142	-0	101	284	0.000 🕇				
6	679	622	0.000	- 9	101	33	3.142 +		•		
1	1 99	1026	0.000		101	130	0.000 T				
8	1037 -	911	0.000	-0	415	703	3.142		•		
•	9	1		-2.	541	39¥ 674	3.172				
-12	496	567	2.855	- <b>.</b>	101	197			:		
-11	485	6.1	-3.139	4	370	16/	0.000 T				
-13	101	206	-1.226 +	. *	10	040 <u>_</u>	0.000				
- 9	632	677	-1.698		301	382	-0.040				
-8	679	740	-2.114	-0	420	444	-2.742				
-7	143	9 <b>8</b>	-3.076 4	-7	767 51A	471 674	-1./01		•		
-6	417	460	0.403	· _7	00	34	V.204	•			
-5	764	797	2.318	_4	368	212	0.20/				
-4	143	201	-0.842 +	_6	101	0K 010	2.710			•	
-3	790	8ú9	-0.218		545	484	2.630 T				
-2	554	574 .	1.396	_3	365	347	-1.107		•		
-1	927	977	1.368	-0	607	37/ 881	1.073				
1	496	507	-0.103		534 -	484	1.405				
1	598 ·	661	-1.779	-	A14	803	1 011				

## Results and Discussion

The atomic coordinates of the non-hydrogen atoms and their standard deviations are shown in Table 34, and Table 4 lists the anisotropic temperature factor coefficients. ELSI was used to calculate the intra-molecular bond distances and angles and their standard deviations shown in Tables 5 and 6. The intra-molecular bond distances and angles are also shown in Figure 2. That the estimated standard deviations are too low can be seen from the dimensions of the benzene ring, in which the C - C bond lengths range from 1.34 Å to 1.43. Å, indicating that the e.s.d.'s in Tables 5 and 6 have been underestimated by perhaps a factor of three. This is, no doubt due to the block-diagonal approximation of the least-squares program.

The Configuration of the Molecule

The view of the molecule as shown in Figure 3 was derived from the final atomic coordinates by redefining their values with respect to the plane of the atoms S1, C6 and C4. Figure 3 shows that in the crystal at least, the molecule exists in a rather extended configuration as opposed to the curled forms of benzylpenicillin and 6-NBF-PA. The asymmetric centres at C6 and C7 are both R, (using the terms for chirality proposed and defined by Cahn, Ingold and Prelog (1966)), which is the Table 36. Atomic coordinates and standard deviations output from cycle A20 for the 3-spiro-thiazolinium-7-PAC derivative.

(fr	actional X	values) Y	Z	$\sigma_{x}$	$\sigma_{\rm y}$	σz
<sup>s</sup> ı	0.40182	0.81764	0.29429	0.00019	0.00027	0.00007
°2	0.39849	0.75748	0.20363	0.00066	0.00092	0.00028
с <sub>3</sub> .	0.21655	0.71990	0.16468	0.00060	<b>Q.00088</b>	0.00026
с <sub>4</sub>	0.12234	0.58335	0.20692	0.00062	0.00084	0.00026
N 5	0.24289	0.48782	0.26052	0.00052	0.00077	0.00021
с <sub>6</sub>	0.33625	0.58921	0 <b>.</b> 32 <b>018</b>	0.00073	0 <b>.00098</b>	0.00026
с <sub>7</sub>	0.46553	0.42010	0.33023	0.00070	0.0009 <b>8</b>	0.00028
с <sub>8</sub>	0.36282	<b>0 • 35 3 30</b>	0.25995	0.00066	0.00092	0.00027
с <sub>9</sub>	0.11601	0.90107	0.15082	0.00071	0.00091	0.00030
<sup>S</sup> 10	0.15454	0.99421	0.06962	0.00021	0.00027	0.00008
с <sub>11</sub>	0.20262	0.77005	0.04416	0.00063	0.00096	0.00027
<sup>N</sup> 12	0.23123	0.64606	0.09555	0.00056	0.00070	0.00021
<sup>N</sup> i3	0.64054	0.46150	0.33197	0.00058	0.00090	0.00025
°14	0 <b>.</b> 76 <b>3</b> 34	0.34013	0.35927	0.00078	0.00111	0.00034
°15	0.93905	0.40895	0.35974	0.00083	0.00156	0.00048
с <sub>16</sub>	1.07545	0.31144	0.41139	0.00081	0.00131	0.00033
с <sub>17</sub>	1.14044	0.40453	0.47206	0.00099	0 <b>.001</b> 40	0.00040
с <sub>18</sub>	1.27654	0.31844	0.51807	0.00106	0.00179	0.00036
с <sub>19</sub> .	1.33923	0.15357	0.50272	0.00110	0.00164	0.00038
C <sub>20</sub> ,	1.27030	0.06257	0.44115	0.00107	0.00154	0.00042
<sup>C</sup> 21	1.14061	0.14745	0• 39573	.0.000093	0.00130	0.00037

Table 36 continued

•	X	Ŷ	Z	σ	Oy	$\sigma_{_{\mathbf{z}}}$
°22	0.27546	0.45226	0.08176	0.00067	0.00084	0.00028
°23	0.46137	0.42789	0.07739	0.00077	0.00116	0.00033
°24	-0.01397	0 <b>. 4569</b> 0	0.16453	0.00062	0.00088	0.00028
0 <sub>25</sub>	-0,10451	0.53246	0.11447	0.00050	0.00068	0.00020
<sup>0</sup> 26	-0.02332	0.29427	0.18508	0.00054	0;00072	0.00024
<sup>0</sup> 27	0.38446	0.23180	0.21984	0.00051	0.00068	0.00020
<sup>0</sup> 28	0.72824	0.18588	0.38048	0.00065	0.00099	0.00036
<sup>N</sup> 29	0.21564	0•73742	-0.01975	0.00059	0.00082	0.00022
<sup>0</sup> 30	0.70759	0.08559	0.19846	0.00062	0.00082	0.00029
<sup>0</sup> 31	0 <b>.</b> 69064	-0.12790	0.08104	0.00069	0.00087	0,00024
0 · 32	0.83998	-0.17257	0.30179	0.00080	0.00108	0.00036

Table 4. 3-Spiro-thiazolinium-7-PAC derivative anisotropic temperature factor coefficients

	B <sub>11</sub>	3	333	323	Bis	B12
s <sub>1</sub>	0.01324	0.01220	0.00187	-0.00206	0.00064	-0.00504
c <sup>2</sup>	0.00903	0.01125	0.00197	0.00026	0.00086	-0.00217
°3	0.00734	0.01075	0.00164	-0.00223	0.00090	-0.00045
°4	0.00786	0.00808	0.00158	-0.00056	0.00030	0.00101
N 5	0.00926	0.00950	0.00148	-0.00042	0.00068	-0.00296
с <sub>6</sub>	0.01284	0.01448	0.00127	-0.00205	0.00099	-0.00650
с <sub>7</sub>	0.01100	0,01362	0.00160	0.00103	0.00022	-0.00552
c <sub>8</sub>	0.00975	0.01222	0 <b>.00166</b>	0.00068	0.00067	-0.00319
с <sub>9</sub>	0.01071	0.00948	0.00217	-0.00085	0.00101	0.00188
<sup>S</sup> 10	0.01699	0.00784	0.00258	0.00162	0.00185	0.00242
°11	0.00803	0.01461	0.00178	0.00190	-0.00030	-0.00442
N <sub>12</sub>	0.01089	0.00809	0.00134	-0.00034	0.00134	0.00126
<sup>N</sup> 13	0.00968	0.01821	0.00186	0.00190	0.00021	-0.00072
c <sub>14</sub>	0.01317	0.01506	0.00264	0,00395	0.00071	0.00166
с <sub>15</sub>	0.00995	0.03357	0.00476	0.01460	0.00175	0.00231
°16	0.01310	0.02097	0.00245	0,00450	0.00145	-0.00224
°17	0.01874	0.02357	0.00293	-0.00231	0.00341	0.00332
c <sub>18</sub>	0.02237	0.03500	0.00215	-0.00224	0.00032	0.00438
с <sub>19</sub>	0.02146	0.03230	0.00236	0.00161	0.00001	0.00536
с <sub>20</sub>	0.02163	0.02569	0.00304	0.00097	0.00220	0.01332
с <sub>21</sub>	0.01679	0.02045	0.00264	-0.00189	0.00030	0.00571

# Table 4 continued

	β"	B 12	B33	B23	B13	Biz
с <sub>22</sub>	0.01081	0.00798	0.00185	-0.00108	0.00220	0.00277
°23	0.01271	0.01931	0.00236	-0.00047	0.00400	0.00858
c <sub>24</sub>	0.00816	0.01076	0.00194	-0.00154	0.00022	-0.00327
0 <sub>25</sub>	0.01222	0.01287	0.00195	-0.00138	-0.00012	<del>-0</del> .00492
<sup>0</sup> 26	0.01310	0.01081	0.00302	0.00097	0.00058	-0.00694
0 <sub>27</sub>	0.01206	0.01117	0.00211	-0.00107	0.00071	0.00150
0 <sub>28</sub>	0.01386	0.02159	0.00582	0.01169	0.00399	0.00008
<sup>N</sup> 29	0.01151	0.01258	0.00150	0.00179	0.00099	-0.00395
°30	0.01459	0.01452	0.00415	-0.00168	0.00483	-0.00280
°31	0.02132	0.01960	0.00238	-0.00145	0.00261	0.00092
0 <sub>32</sub>	0.02324	0.02141	0.00524	0.00452	.0.00544	-0.00097

All temperature factors in this table are positive definite.
<u>Table 5</u>. 3-Spiro-thiazolinium-7-PAC derivative intra-molecular bond distances and estimated standard deviations in Angstroms.

Bond

Bond

S1 - C2	1.835 (.005)
si -006	1.825 (.007)
C2 - C3	1.553 (.007)
C3 - C4	1.568 (.007)
C3 - C9	1.532 (.009)
C3 - N12	1.487 (.006)
C4 - N5	1.472 (.007)
C4 - C24	1.550 (.008)
N5 - C6	1.471 (.007)
N5 - C8	1.369 (.008)
C6 - C7	1.590 (.009)
C7 - C8	1.558 (.007)
C7 - N13	1.437 (.007)
C8 <b>-</b> 027	1.211 (.007)
C9 <b>-</b> S10	1.813 (.005)
S10 - C11	1.751 (.007)
C11 - N12	1.338 (.007)
C11 - N29	1.306 (.006)
N12 - C22	1.475 (.008)
N13 - C14	1.351 (.009)
C14 - C15	1.507 (.010)

C14 - 028	1.233 (.010)
C15 - C16	1.529 (.011)
C16 - C17	1.389 (.010)
C16 - C21	1.348 (.012)
C17 - C18	1.433 (.012)
C18 - <sup>C</sup> 19	1.343 (.016)
C19 <u>-</u> C20	1.403 (.011)
C20 - C21	1.390 (.011)
C22 - C23	1.527 (.008)
024 - 025	1.243 (.006)
C24 - O26	1.243 (.008)

C2 - S1 - C6	96.1 (0.3)	C9 - S10 - C11	89.7 (0.3)
S1 - C2 - C3	111.8 (0.4)	S10 - C11 - N12	114.3 (0.7)
C2 - C3 - C4	110.5 (0.5)	S10 - C11 - N29	120.5 (0.6)
C2 - C3 - C9	111.3 (0.6)	N12 - C11 - N29	125.1 (0.7)
C2 - C3 - N12	107.0 (0.5)	C3 - N12 - C22	124.6 (0.6)
04 - 03 - 09	109.6 (0.6)	C3 - N12 - C11	115.2 (0.5)
C4 - C3 - N12	112.9 (0.5)	C11 - N12 - C22	120.2 (0.7)
C9 - C3 - N12	105.4 (0.5)	C7 - N13 - C14	121.1 (0.7)
$C_3 - C_4 - N_5$	110.7 (0.52)	N13 - C14 - C15	114.2 (0.7)
$C_3 - C_4 - C_{24}$	116.4 (0.6)	N13 - C14 - 028	121.7 (1.1)
N5 - C4 - C24	115.3 (0.5)	C15 - C14 - 028	124.1 (1.1)
C4 - N5 - C6	121.4 (0.5)	C14 - C15 - C16	114.9 (0.7)
C4 - N5 - C8	134.5 (0.8)	C15 - C16 - C17	118.0 (0.8)
C6 - N5 - C8	95.5 (0.5)	C15 - C16 - C21	120.8 (1.1)
c6 - c7 - c8	83.8 (0.4)	C17 - C16 - C21	121.0((1.1)
C6 - C7 - N13	117.7 (0.7)	C16 - C17 - C18	117.8 (0.9)
C8 - C7 - N13	116.7 (0.6)	<b>C17 - C18 - C19</b>	120.9 (1.4)
N5 - C8 - C7	92.3 (0.4)	<b>C18 - C19 - C20</b>	119.9 (1.3)
N5 - C8 - 027	134.3 (1.0)	C19 - C20 - C21	119.3 (1.0)
C7 - C8 - 027	133.4 (0.8)	C16 - C21 - C20	121.0 (1.2)
S10 - C9 - C3	107.6 (0.5)	N12 - C22 - C23	113.0 (0.6)

# Table 6 continued

C4 - C24 - O25	115.6 (0.6)
C4 - C24 - O26	117.0 (0.7)
025 - 024 - 026	127.5 (0.9)

# $C_{10} = \frac{1.39}{C_{10}} = C_{11}$



$$C_2 - C_3 - N_{12} = 107^\circ$$
  
 $C_4 - C_3 - C_9 = 110^\circ$ 

lengths and angles.



same as in all previously determined cephalosporins and penicillins. The asymmetric centre at C4 is S, whereas the corresponding centres in benzylpenicillin, phenoxymethylpenicillin and 6-NBF-PA possess the opposite chirality, R. Using the same notation, the spiro atom at C3 is S. It was the difficulty of describing the configuration at C3 that led the author, with the assistance of Dr.A.G.Long (1966), to adopt the Cahn-Ingold-Prelog notation which gives the name (3S,4S,6R,7R)-2'-amino-3'-ethyl-2'-thiazolinium-4'-spiro-3-(7-phenylacetamidocepham-4-carboxylate) trihydrate, where the 'cepham' system, suggested by Morin et al. (1962) is shown in Figure 4.



Figure 4. The cepham system

The analysis has shown that the ethyl group (C22, C23) is bonded to the cyclic nitrogen atom (N12) of the thiazolinium group.

### The Thiazolinium Ring

DIDO was used to calculate the least-squares planes through sets of atoms in the mobecule, and the results are shown in Table 7.

The atoms C3, S10, C11, N12, C22 and N29 form a plane (Table 7a), the average deviation from the plane being 0.01 Å, and C9 lies 0.45 Å below the plane looking at the molecule as seen in Figure 3. The bond C9 - S10 (1.813 Å) is in good agreement with the average value of 1.817 Å for a single C - S bond given by Sutton (1965). However, S10 - C11 (1.751 Å) is approximately 9 e.s.d.'s shorter than this and nearer to the value of 1.718 Å in the conjugated, heterocyclic compound thiophen. Sutton gives the average C - N bond length in conjugated heterocyclics as 1.339 Å, with which the bond C11 - N12 (1.338 Å) is in good agreement, and C11 - N29 (1.306 Å) is approximately 5 e.s.d.'s shorter.

These results suggest that the thiazolinium system can be represented as in Figure 5, N29 being probably hydrogen



Figure 5. The thiazolinium ring.

bonded (see page 122). The bond lengths of the carboxyl group, o C24 - 025 (1.243 Å) and C24 - 026 (1.243 Å) indicate that it is present in the ionized form (C - 0 = 1.26 Å (Sutton)), so that the analysis substantiates the electrophoresis and infrared data that suggested the compound occurs as a zwitter-ion. The carboxylate ion is also planar (Table 7e).

The bond angle  $C9 - S10 - C11 (89.7^{\circ})$  agrees quite well with the C - S - C angle in thiophen ( $91.3^{\circ}$ ), although the comparison is questionable as thiophen is completely conjugated. The angle N12 - C22 - C23 ( $113.0^{\circ}$ ) differs significantly from the equilibrium value of  $109^{\circ}$  28'. This can be explained by repulsion between the C23 methyl group and the  $\Pi$  electrons at N12, the distance N12 - C23 being 2.503 Å.

The Tetrahydrothiazine Ring

The atoms S1, C2, C3, C4, N5 and C6 form a distorted boat conformation with C2 and N5 at the prow and stern. The boat is distorted in that the bond S1 - C6 is far from parallel to C3 - C4. This can be seen in Figure 3; for an undistorted boat, the atoms S1, C3, C4 and C6 would be in a plane, whereas C3 actually lies 0.656 Å above the plane through the other three atoms.

The reasons for the preferred boat conformation in the crystal structure are difficult to assess. One explanation

may be that the chair form would separate the charges of the thiazolinium system and the carboxylate ion. Whether this is the case was difficult to decide by constructing models, especially as it is not known where the charge is concentrated in the thiazolinium group. The shortest inter-atomic distances relevant to this point are given below in A

C11 <b>-</b> 025	3.492
C11 <mark>-</mark> C24	3.902
N12 - C24	2.927
N12 - 025	2.915

Changing the model from the boat to the chair form moved 025 away from N12 but did not noticeably affect the N12 - C24 distance, though it did bring the exocyclic ethyl group into crowding with the  $\beta$ -lactam-tetrahydrothiazine ring system. Even in the boat conformation, the ethyl group is quite close to C2 and 027 as shown by the distances C2 - C23 (3.537 Å), C2 - C22 (3.273 Å), C22 - O27 (3.137 Å) and C23 - O27 (3.299 Å).

Another close contact that may affect the conformation of the six-membered ring is that between the group  $\frac{H}{H} > C22$  and the carboxylate ion. The distance C22 - C24 is 3.080 A, and one of the C22 hydrogen atoms points towards the carboxylate ion. This probably accounts for the fact that the angles C3 - C4 - C24 (116.4°) and N5 - C4 - C24 (115.3°) are both approximately 10 e.s.d.'s from the tetrahedral value and the

distortion may be favoured in that it results in a partial staggering of the bonds C3 - N12 and C4 - C24. It is also possible that the role of the water molecules in the crystal could contribute to the stability of a particular conformer. As will be described later, the water molecules enter into a hydrogen-bonding scheme that includes N29, O25 and O26.

The high degree of overcrowding that has been described may result in restricted rotation of the carboxylate ion around the bond C4 - C24 and possibly account for the low antibacterial activity of the spiro compound, in that it would be difficult for either 025 or 026 to acquire a 'penicillin-like' position. Rotation would be especially hindered by the close contacts between 025 and C22 (3.293 Å), 026 and 027 (3.267 Å) and C9 and 025 (3.199 Å).

The rest of the dimensions in the tetrahydrothiazine o ring are unremarkable. The bonds S1 - C2 (1.835 Å) and S1 - C6 (1.825 Å) are not significantly different from the value given by Sutton (1.817 Å) for a C - S single bond and the value of 1.80 Å determined by Diamand (1963) for ceph. Cc. The dimensions of the ionized carboxyl group agree well with those listed by Sutton, for example in the tartaric acid ion and the zwitterion, L-threonine.

# The $\beta$ -Lactam Ring

The bond lengths and angles of the  $\beta$ -lactam ring in the 3-spiro-thiazolinium-7-PAC derivative are similar to those determined for 6-NBF-PA. The length of the bond C6 - C7 (1.590 Å) is possibly not significant in terms of the real standard deviations; previous crystallographic work has not indicated a bond significantly longer than the accepted value of 1.539 Å.

The least-squares plane through the atoms of the  $\beta$ -lactam ring including 027 shows deviations between 0.02 and 0.10 A (Table 7b). However, the atoms N5, C7, C8 and 027 form a plane wellwithin the experimental accuracy, with C6 projecting 0.245 A out of the plane on the same side as S1. This arrangement agrees with that found for 6-NBF-PA.

That resonance takes place to a certain extent in the ocyclic amide is shown by the bond length N5 - C8 (1.369 Å) and the angle C4 - N5 - C6 (121.4°) indicates the corresponding tendency for N5 to be planar. However, the nitrogen atom in the 3-spiro-thiazolinium-7-PAC derivative has achieved a greater degree of planarity than the corresponding atom in 6-NBF-PA, as shown by the sum of the angles round both atoms (351° and  $340^{\circ}$  respectively). This is probably because the six-membered ring in the spiro cephalosporin possesses more degrees of freedom than the thiazolidine ring, so that planarity at N5 can be achieved without exerting a great deal of strain at C6. Table 7. DEviations (A) of the atoms from the calculated least-squares planes. Atoms marked with \* were not included in the calculation of the plane.  $\sum d^2$  is the sum of the squares of the deviations (A) from the plane.

(a) Thi	azolinium Ring	(b) <mark>/3 —</mark> 1	(b) 3 -Lactam Ring			
$\sum d^2$	= 0.00089	$\sum d^2 = 0.02569$				
Atom	Deviation	Atom	Deviation			
S10	-0.001	N5	-0.080			
C11	-0.020	c6	0.104			
N 12	-0.008	C7	-0.063			
N29	0.017	<b>c</b> 8	-0.023			
C3	0.013	027	0.062			
C22	-0.002					
C9 *	-0.451					

(c) \$ -Lac	tam Ring
$\sum d^2$	= 0.00010
Atom	Deviation
N5	0.002
C7	0.002
C8	-0.009
027	0.004
C6 *	0.245

(d)	Benzene Ring			
	$\sum d^2 = 0.00040$			
Atom	Deviation			
C16	-0.008			
C17	0.000			
C18	0.003			
C19	0.003			
C20	-0.011			
C21	0.014			
C15	• 0.091			

# Table 7 continued

(e) Ca	arboxylate Ion	(f) Ami	de Group
Z	$d^2 = 0.00011$	Zd	2 = 0.00107
Atom	Deviation	Atom	Deviation
025	-0.003	N 13	0.025
C24	0.009	C14	-0.002
026	-0.003	028	0.006
C4	-0.002	C15	-0.011
•		C7 •	-0.017

### The Phenylacetamido Side Chain

The amide group is planar (Table 7f) and the bond lengths are consistent with those of a normal resonating amide. The angle N13 - C14 - C15  $(114.2^{\circ})$  is 8 e.s.d.'s from the expected value of  $120^{\circ}$ , and C14 - C15 - C16  $(114.9^{\circ})$  is 7 e.s.d.'s above the tetrahedral value. <sup>Th</sup>ese results can be explained by repulsive forces between the carbonyl electrons and the  $\Pi$  electrons of the benzene ring, the distance C16 - O28 being 2.900 Å. The extended form of the side chain which results in this close contact is probably stabilized by packing conditions in the crystal.

### Packing and Hydrogen Bonding

In the following discussion, atoms that are related to those of Table 3b by the symmetry of the spacegroup are referred to by terms in parentheses following the atom symbol.

Some close-contact distances are shown in Figure 8 which is a projection of adjacent unit cells down the b axis. Inter-molecular distances less than 3.5 Å are listed in Table 8, including those between the oxygen atoms (030, 031, 032) of the three water molecules.

The molecules of the spiro compound lie diagonally across the cell; the arrangements around the screw axes at



<u>Table 8.</u> Inter-molecular distances (A) less than 3.5 A in the crystal of the 3-spiro-thiazolinium-7-PAC derivative. The terms in parentheses following the atom name relate the position of the atom to the values listed in Table 3b. The first term describes the symmetry relation where

$$0/ = x, y, z$$
  
 $1/ = -x, y - \frac{1}{2}, -z$ 

and the other terms give the number of cell translations in the directions x, y and z respectively.

Vect	:01	?	Symmet	try	I	Distance	Vector	Symmet	try	Distance
<b>S1</b>	÷	027	(0/0	1	0)	3.310	C24 - N29	(1/0	0 0)	3.412
C2	-	027	(0 <u>/</u> 0	1	0)	3.426	025 - 031	(0/-1	1 00)	2.954
C2	-	030	(0/0	1	0)	3.445	025 <b>-</b> N29	(1/0	o o)	2.866
N5	-	C15	(0/-1	0	0)	3.445	026 - 030	(0/-1	0 0)	2.689
<b>C</b> 9	-	026	(0/0	1	0)	3.157	026 <b>-</b> N29	(1/0	0 0)	3.376
C9	-	027	(0/0	1	0)	3.336	027 <b>-</b> 030			2.909
C9	-	031	(0/-1	1	0)	3.466	028 - 032			3.217
s10	-	C22	(0/0	1	0)	3.428	030 - 031			2.760
S10	-	N29	(1/0	1	0)	3.444	030 - 032			2.820
N 13	_	032	(0/0	1	0)	3.193	0 <b>31 -</b> N29	(1/1	0 0)	3.035
C15	-	032	(0/0	1	0)	3.264	c <b>1</b> 8 - 028	(1/2	1 1)	3.317
<u>6-8</u>								•		•

z = 0, 1 etc. concern the polar groups, whereas the screw axes at z = 1/2, 3/2 etc. are surrounded mainly by the non-polar benzyl groups. The close contacts around the screw axis at  $x = \frac{1}{2}$ ,  $z = \frac{1}{2}$  are between the benzene rings and the  $\beta$  -lactam rings though the distances are quite large as shown in Figure 8.

The oxygen atoms of the three water molecules are located (Figure 8) such that the cephalosporinemolecules in adjacent unit cells are linked in all three crystallographic directions by hydrogen bonds. The network so formed can be seen in Figure 9, which is a projection down the a axis. The positions of the atoms shown in Figure 9 are related to those in Table 3b as follows :

> CB11 = C11 (1 = x, y -  $\frac{1}{2}$ , -z) CC11 = C11 (1 - x, y +  $\frac{1}{2}$ , -z) CA24 = C24 (1 + x, y, z) OA25 = O25 (1 + x, y, z) OA26 = O26 (1 + x, y, z) NB29 = N29 (1 - x, y -  $\frac{1}{2}$ , -z) NC29 = N29 (1 - x, y +  $\frac{1}{2}$ , -z) OA30 = O30 (x, y + 1, z) OA31 = O31 (x, y + 1, z) OA32 = O32 (x, y + 1, z)

030 appears to be tetrahedrally bonded to 032 (2.820 A), o o o 031 (2.760 A), 026 (1 +x, y, z) (2.689 A) and 027 (2.909 A).

NC<sub>29</sub> 0A30 OAal OA32 CA 24 0 112 N<sub>13</sub> C<sub>14</sub> OCB<sub>II</sub> 115 Pite 116 0<sub>28</sub> ( NĒ 99 030 112 101 5 129 031 11,2 () | 032 Scale L

bonding system projected down the a axis



2Å

The angles around 030 are shown in Figure 9. Pimental and McClellan (1960) have tabulated the known data for the  $0 - H \dots 0$  distances and the  $0 \dots 0 \dots 0$  angles for water in organic crystals. The values vary over a wide range; the distances from 2.65 Å to 2.99 Å, and the angles from  $83^{\circ}$ to  $120^{\circ}$ , and the values associated with 030 are in good agreement with this range. If the hydrogen bonding is as suggested, the hydrogen atoms bonded to 030 must lie approximately in the directions of 027 and 026 (1 + x, y, z).

The atoms surrounding 031 are 025 (1 + x, y - 1, z)(2.954 A), N29  $(1 - x, y - \frac{1}{2}, -z)$  (3.035 A) and 030 (2.760 A). N12 (1 + x, y, z) is toofar from 031 (4.101 A) to take part in any hydrogen bonding. The location of two possible hydrogen atoms bonded to N29 has been described on page 89. For the calculation of the position of H21 (1 - x, y -  $\frac{1}{2}$ , -z) (Table 2), it was assumed that a linear hydrogen bond existed between 031 and N29 (1 - x, y -  $\frac{1}{2}$ , -z). The assumption for the calculation of H22 was that the atoms 031, C11(1 - x, y -  $\frac{1}{2}$ , -z) and the hydrogen atom would form a trigonal arrangement around N29 (1 - x, y -  $\frac{1}{2}$ , -z). The observed and calculated positions agree well, and the observed position of H21 (1 - x, y -  $\frac{1}{2}$ , -z) lies 19° off the line N29 (1 - x, y -  $\frac{1}{2}$ , -z) - 031 and that of H22 (1 - x, y -  $\frac{1}{2}$ , -z) is situated 7<sup>°</sup> from the line N29 (1 - x, y -  $\frac{1}{2}$ , -z) - 025 (1 + x, y, z). Thus, the hydrogen atoms bonded to 031 probably lie in the directions of 030 and

025 (1 + x, y, z).

The distances 0.32 - 0.28 (3.217 A) and 0.32 - 0.13 (x, y - 1, z) (3.193 Å) suggest that weak hydrogen bonding may exist between these atoms. The angles around 0.32 are shown in Figure 9 and only 0.30 - 0.32 - 0.28 (74°) lies out of the range given by Pimentel and McClellan, and the two lone pair orbitals of 0.28 will not be pointing well towards 0.32, resulting in only a weak hydrogen bond. The calculated and observed positions of the hydrogen bonded to N13 (x, y - 1, z) are situated respectively 28° and 17° off the line 0.32 - 0.13 (x, y - 1, z), which indicates that the hydrogen atoms bonded to 0.32 are orientated towards 0.30 and 0.28.

, The hydrogen-bonding network that has been described must contribute substantially to the stability of the crystal, and as mentioned on page //8 possibly plays an important part in stabilising the observed boat conformation of the tetrahydrothiazine ring.

### CHAPTER 6

The Chemistry of Zeorin

As mentioned at the end of Chapter 1, attempts have been made to solve the crystal structure of zeorin acetate, a compound containing only carbon, hydrogen and oxygen  $(C_{32}H_{54}^{\circ}O_{3})$ and derived from the natural product zeorin, a member of the triterpenoid series. The chemical work, which is briefly described in this chapter has established the structure of zeorin fairly well and shown that it is derived from a unique, saturated hydrocarbon zeorinane. The proposed structure of zeorin fits in well with the theory of squalene as the biogenetic precursor of the triterpenoids.

That zeorin is a component of foliaceous lichens was recognised by Hesse (1906) and Zopf (1909) but its chemistry received little attention until Asahina and Akagi (1938) demonstrated that it belongs to the triterpenoid series and has the formula  $C_{30}H_{52}O_2$ . Mild acetylation produced the mono-acetate, zeorin acetate, and Asahina and Yosioka (1940) showed that more vigorous acetylation gave anhydrozeorin acetate, by elimination of the elements of water from zeorin acetate. This indicated that zeorin contained a tertiary - OH group.

Zeorin, when heated with alcoholic hydrochloric acid gave a dehydration product, zeorinin, which also formed a monoacetate and when oxidised with dichromate gave a ketone, zeorininone  $(C_{30}H_{48}O)$ , indicating the presence in zeorin of a secondary - OH group.

These conclusions were substantiated and extended by Barton and Bruun (1952). Infra-red data for zeorinone (produced by chromic acid oxidation of zeorin) indicated a cyclohexanone part-structure, but the oxygen atom could not be at the  $C_3$ position, which is most common in the friterpenoids and shown in (I), as zeorinone is unreactive to normal carbonyl reagents



and reduction. Dehydration of zeorin acetate gave zeorinin acetate or an isomer, isozeorinin acetate, depending upon the reaction conditions. The infra-red absorption spectrum of isozeorinin acetate (probably equivalent to the anhydrozeorin acetate of Asahina and Yosioka) indicated the presence of a methylene group as  $\sum CH_2$ , hence the tertiary - OH of zeorin as  $\sum CMe = OH$ . That zeorin is not a simple derivative of a known triterpene was demonstrated by the preparation of zeorinane, the unique and fully saturated hydrocarbon.

The above and other important reactions of zeorin and its derivatives are shown in Figure 1.

After further work, Barton, de Mayo and Orr (1958) proposed the structure (II) for zeorin.



Ozonolysis of crude isozeorinin acetate gave a methyl ketone with the loss of one carbon atom, indicating the part-structure (III) in isozeorinin acetate (that the E ring is five-membered





is shown in the following section). The ozonolysis also produced acetone which suggested the presence of the

# Figure 1. Some reactions of zeorin.



isopropylidene isomer (IV) in the crude starting material.



Recognition of the ketone (V) was made by further oxidation



of the product with trifluoro peracetic acid to give the  $\delta$ lactone (VI) by the following mechanism.



(VI)

(VI) was recognised as a  $\delta$ -lactone from its infra-red spectrum, and its formation indicated that the E ring must be fivemembered. Hence zeorinin was assigned the part-structure (VII)



# (VII)

where the double bond of isozeorinin,(III), has migrated into the E ring.

Other oxidative experiments on zeorinin acetate using osmium tetroxide and lead tetra-acetate, resulted in the 1:5 diketone (VIII). Barton et al. predicted that (VIII) should



### (VIII)

undergo a reverse Michael addition, to give after oxidation, the diketone (IX). The product from the action of alkali in



(IX)

ethylene glycol on (VIII) followed by oxidation showed a single band in the I.R. spectrum at 1703 cm<sup>-1</sup>, characteristic of a cyclohexanone structure and confirming that the D ring must be six-membered, as in (IX).

The probable identity of the C ring was determined by studies on neozeorininone, obtained from acid isomerisation of zeorininone. By analogy with other rearrangements in the triterpenoid series, neozeorininone was assigned the structure (X).



This was supported by oxidation of neozeorininone with chromic

acid to give three products whose spectral properties could be explained in terms of the part-structures (XI), (XII) and (XIII).



Selenium dioxide oxidation of neozeorininone afforded a conjugated, heteroannular diene (XIV), which was converted into an aldol



# condensation product via the dialdehyde (XV). The formation



of an aldol condensation product and its probable structure, deduced from spectral properties, excluded the partial structure (XVI) for the parent diol, and together with the reactions so



far, suggested the part-structure (XVII) for the C, D and E rings in zeorin.



The secondary - OH group is located at the C6 position in the B ring. This was demonstrated by dehydration of deoxyzeorin (obtained by hydrogenolysis of zeorin) to give, as main product, a compound represented by the part-structure (XVIII).



Chromic acid oxidation of this unsaturated hydrocarbon gave an  $\checkmark \beta$ -unsaturated ketone (XIX), which was stable to the action



of bromine and selenium dioxide, in agreement with the absence in (XIX) of a replaceable  $\mathcal{A}$ -hydrogen atom, and suggesting that the B ring possesses the part-structure (XX).



The structure of the A and B rings was further supported by the fact that Asahina and Yosioka obtained 1 : 2 : 5 trimethyl naphthalene (XXI) from zeorin. It is not likely that the C



and D rings would have given this product.

Barton, de Mayo and Orr (1958) then suggest that it is 'attractive' to place the part-structure (XVI) and the A and B rings together as implied in structure (II) so that the biogenesis of zeorin, with squalene as precursor,(XXII), can be represented by the cyclisation of squalene followed by rearrangement and hydroxylation, as follows :







hydroxylation

ZEORIN

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Structure (II) has also received support from the

elucidation of the structure of hydroxyhopanone (XXIII), a



triterpenoid obtained from dammar resin. This compound, studied by Dunstan et al. (1957), Schaffner et al. (1957) and Fazakerley, Halsall and Jones (1959) forms a series of compounds analogous to zeorinone. By consideration of the molecular rotation changes between the derivatives, it is suggested that seven of the nine asymmetric centres in zeorin can be assigned as in hydroxyhopanone, only the stereochemistry at C17 and C21 being ambiguous. Thus zeorin can be written as in (XXIV), the



- OH at C6 being  $\checkmark$ -equatorial, as suggested by the ease of acetylation. Similarities between the zeorin and hydroxyhopane series have also been pointed out by Huneck (1961). Wolff - Kishner reduction of zeorinone produced 6-deoxyzeorin, which was shown by physical methods to be probably identical to hydroxyhopane. Huneck (1963) used nuclear magnetic resonance spectroscopy on zeorin acetate to deduce that the proton at C6 is  $\beta$ -axial, and that the protons at C17 and C21 are both  $\beta$ -axial.

Thus, the chemical work that has been described establishes the structure of zeorin fairly well, but it was felt that a confirmation of the structural details was desirable. Two crystalline derivatives of zeorin were initially supplied by Professor D.H.R.Barton of this department, with a view to determining the crystal structure by the Heavy-Atom method. However, both zeorin iodoacetate and zeorin-m-iodobenzoate deteriorated very rapidly when exposed to X-rays.

Crystals of zeorin acetate, again supplied by Professor Barton,did not deteriorate noticeably when exposed to X-rays for long periods. Although lack of a heavy atom in the crystal would cause substantial crystallographic problems, it was felt that the postulated structure and stereochemistry of the molecule would provide a useful starting point in an X-ray

analysis by Patterson searching techniques. It was considered that if the method had proved successful, it might have been applicable to similar compounds in the triterpene series and in the steroids.

Another possible means of determining the crystal structure of zeorin acetate was by one of the 'Direct Methods' of phase determination (e.g. Harker and Kasper (1947), Sayre (1952), Hauptman and Karle (1953)) but these methods are difficult to apply to non-centrosymmetric crystals. Zeorin acetate is in fact noncentrosymmetric (spacegroup  $P2_12_12_1$ ) and of the two possible centrosymmetric projections, down <u>a</u> (8.78 Å) and <u>b</u> (11.59 Å) preliminary investigations (Chapter 8) indicated that the former was an inconvenient 'edge-on' view of the molecule and a considerable degree of overlap would occur in the latter.

### CHAPTER 7

Interpretation of the Patterson for Complex Structures

In recent years there have been basically two different approaches towards interpreting the Patterson of molecules containing a large number of similarly weighted atoms. All the methods are dependent upon the use of a high-speed digital computer.

If no reasonable assumptions concerning the structure of the molecule can be made, it may be possible to achieve a solution by the use of an image-seeking function to compare for each point in the unit cell (x, y, z), values of the Patterson corresponding to all vectors between x, y, z and its symmetry-related points. Mighell and Jacobson (1963) have used this method which they call Vector Verification; if all the vectors have corresponding peaks in the Patterson, then the point may define an approximate atomic location. The validity of all suggested positions can then be tested by computing vectors between them and searching the Patterson for the corresponding peaks.

Simpson et al. (1963) made use of the Symmetry Minimum Function (SMF) in the solution of the structure of iso -  $B_{18}H_{22}$ .

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The method is essentially the same as that of Vector Verification and was used first to locate trial atomic positions, and then, in conjunction with normal image-seeking procedures to locate the rest of the molecule. Methods along similar lines have been developed by other authors (Alden et al. (1964), Hamilton (1965)).

The second approach to the problem of Patterson interpretation is to presuppose that the molecule contains a fragment of approximately known internal geometry. Such a method has been described by Nordman and Nakatsu (1963), Nordman and Kumra (1965) and more recently by Nordman (1966).

The orientation of the molecular fragment in the unit cell can be expressed in terms of three Eulerian angles  $\phi$ ,  $\theta$ ,  $\psi$ with respect to a Cartesian system, fixed in relation to the crystallographic axes. The position  $(x_0, y_0, z_0)$  of one atom of the fragment is then required to define uniquely the location of the entire group in the unit cell. The vector set of the molecular fragment is computed over ranges of  $\phi$ ,  $\theta$ ,  $\psi$ , and at each angular combination the Patterson values at the ends of the vectors can be examined by an image-seeking function. Nordman and Nakatsu chose the minimum function

 $M_p(\phi, \theta, \psi) = Min.(P_1, P_2, \dots, P_p)$ 

where p is the number of vectors in the vector set and  $P_{p}$  is
the value of the Patterson at the end of the p'th. vector. The values of  $\phi$ ,  $\Theta$ ,  $\psi$  in this orientation search' that give peaks in the three-dimensional minimum function may then define possible orientations of the molecular fragment in the unit cell.

To find possible values of the positional parameter  $(x_0, y_0, z_0)$ , a second search (location search) of the Patterson is carried out, but this time computing vectors between atoms in the suitably orientated fragment and their symmetryrelated positions. Vectors within a particular fragment are excluded as they have already been considered in the orientation search, so the vector set used will comprise Harker and general vectors. Selecting any arbitrary values as starting point (unless there is evidence of a limited range) the search ranges are successive values of x, y, z along the three Cartesian axes. The search is less than three-dimensional whenever the origin of the unit cell is arbitrary in one or two dimensions. If tentative values of  $x_0, y_0, z_0$  can be found, the rest of the molecule can be identified by normal image-seeking techniques or by Fourier methods. Nordman (1966) has described improvements to the method that include the assignment of vector weights, and a vector refinement procedure to maximise and differentiate between different 'fits' that may be obtained in practice.

A similar method which presupposes a known molecular

fragment has been developed by Hoppe (1957) and is termed the Convolution Molecule Method. It is possible to calculate the 'partial Pattersons' corresponding to both vectors within a molecular fragment and those between symmetry-related fragments. The resulting 'convolution molecules' are termed 'even-indexed' and 'mixed-indexed' respectively as they are calculated by Fourier summations using as coefficients  $F_i F_i^*$  and  $F_i F_j^*$  where  $F_i$ ,  $F_i$  are 'molecular structure factors'.

The even-indexed convolution molecule may then be used in an orientation search and the mixed-indexed function in a translational search as described for the previous method. In theory, the degree of fit at each orientation and translation could be judged by subtracting the convolution molecule from the Patterson value (both on the same scale) at all points in Patterson space and then summing separately, the positive and negative differences. A large negative sum is unacceptable but a large positive value merely indicates that additional peaks are present in the Patterson due to vectors not accounted for in the convolution molecule. However, this is impracticable in terms of computing time (automated by Huber (1965)) and so the procedure is limited to the maxima of the convolution molecules. The weight of each vector is naturally taken into account by virtue of using the peak heights in this way.

#### The Computer Programs Used

In view of the structure proposed for zeorin as a result of chemical evidence (Chapter 6), it seemed reasonable to attempt an interpretation of the Patterson of zeorin acetate by an approach in which this information could be utilized. The A, B, C, D, and E rings most probably form a rather rigid system, parts of which would be suitable to define a molecular fragment. The automated version of the Convolution Molecule method was not available but Dr. Nordman of the University of Michigan kindly supplied the listings of computer programs similar to those used by Nordman and Nakatsu (1963). Slight modifications have been made and the programs are now available for use on the I.B.M. 7090 at Imperial College. The three distinct programs are written mainly in MAP (Macro Assembly Programming Language) with several FORTRAN IV subroutines for input and output.

#### Offentation Search

A flow diagram is shown in Figure 1.

1. The relative coordinates of the atoms in the molecular fragment are input to the program as unweighted Cartesian vector components (A), together with three Eulerian-angle search ranges (for convenience the angles  $\phi, \phi, \psi$  will be written



as A, B, C) and the precomputed Patterson (section 5, page 151), which is stored in the computer in a packed format.

2. The search ranges are input as nine angles :-

(i) The values through which the vectors are initially rotated to the starting point of the search (INA, INB, INC).

(ii) The values by which A, B, and C are to be successively incremented (DELA, DELB, DELC)

(iii) The final values of the three angles (FINA, FINB, FINC). The system is shown in Figure 2, where X, Y and Z are the Cartesian axes of Patterson space. The input vectors are first rotated about the X axis through INA. This rotation also moves the Y and Z axes to Y' and Z' in the YZ plane, making angles of INA with the original axes. Then follows a rotation of INB about Y', which also rotates the X axis through INB to give the X'' axis. The third rotation of INC is about X" to give the vectors their starting orientation for the search.

..... X"..... INC(DELC)FINC. Calculation REPEAT Reset C = INCREPEAT Reset B = INBREPEAT

# Figure 2. The Eulerian angles.



4. At each orientation, the appropriately transformed Cartesian vector components are converted to fractional crystal coordinates, which are adjusted to lie in the range  $-\frac{1}{2}$  to  $+\frac{1}{2}$  by successive addition or subtraction of unity, and then their signs are changed according to the Patterson symmetry, i.e. 2/m for monoclinic and mmm for orthorhombic spacegroups. The vectors are then expressed in sixtieths, as this, at the moment, is the only permissible grid interval. The Patterson values at the ends of the vectors are then evaluated by linear interpolation between four neighbouring grid points, and the two minimum values are stored in the print bands LOW and NEXT, which are subsequently printed out to give the rotational minimum function.

5. The routines for reading and packing the Patterson function in core storage require that the Patterson be initially input on punched cards in a strict format. This is rather unsatisfactory in that it limits each Patterson value to two digits (otherwise the number of cards becomes excessive). A better approach would have been to revise the input routines to accept off I.B.M. magnetic tape, a Patterson with a larger range of values. Unfortunately, no satisfactory Fourier program was available and so it was found expedient to calculate and store the Patterson on I.C.T. magnetic tape using BOSS, and then punch the required cards by a simple Atlas program, 'Patterson Search Punch'(**fsp**).

#### Location Search

A flow diagram is shown in Figure 3.

1. A set of Cartesian coordinates (A) of the molecular fragment are input to the program together with one or more possible sets of Eulerian angles A, B and C, obtained from the orientation search. Up to three symmetry-element matrices are input for the generation of the equivalent positions of the fragment, and also three search ranges as follows :-

(i) The initial values in fractional coordinates to be added to the input positions (INX, INY, INZ).

(ii) The values by which the positions are to be incremented along the crystallographic axes (DELX, DELY, DELZ).

(iii) The final values (FINX, FINY, FINZ).

After input of the Patterson and rotation of the fragment through the Eulerian angles, the Cartesian coordinates are converted to fractional values and the fragment then moved to some arbitrary starting point by addition of INX, INY, INZ to its atomic positions.

2. One or more symmetry-related fragments are then generated and the unique vectors between them and the original fragment are computed. The vectors are adjusted and vector-end Patterson values interpolated and printed as in the orientation search.

3. The fragment is then moved in the crystallographic



directions through the specified search ranges and (2) is repeated at every stage. The process can be described by three nested loops as follows :-

> INX(DELX)FINX INY(DELY)FINY INZ(DELZ)FINZ Calculation REPEAT Reset Z = INZ REPEAT Reset Y = INY REPEAT

Superposition

A flow diagram is shown in Figure 4.

 The positions of the known molecular fragment (plus the symmetry-related positions if required) are input as fractional coordinates together with the search ranges INX, Y, Z; DELX, Y, Z; FINX, Y, Z and the Patterson function.

2. Vectors from input positions to the point INX, INY, INZ are computed, then adjusted and the Patterson values interpolated and printed as in the previous two programs.

3. The process is repeated by computing vectors from the input positions to all the points generated by the search ranges



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Superposition program.

#### as follows :-

INX(DELX)FINX -INY(DELY)FINY -INZ(DELZ)FINZ-Calculation REPEAT Reset Z = INZREPEAT Reset Y = INY REPEAT

4. The minimum function is printed as parallel sections up an optional crystallographic axis, in a form suitable for contouring.

Further details of the programs are described in "Patterson Search Programs" by Hunt (1966). Many difficulties were initially experienced in attempting to generalise and make the programs available for any I.B.M. 7090/7094 installation. It soon became evident that the programs had been written mainly for one spacegroup, P2/m, and consequently many routines were, in the author's opinion, illogical and unnecessary when applied to other symmetries. Changes have been made in a way to make most use of what was already given, and the result (as the flow diagrams indicate), is a rather inefficient sequence of operations. Nevertheless, the computing times on the I.B.M. 7090 are quite realistic, for example an orientation search with 66 vectors (i.e. all the vectors within a molecular fragment of 12 atoms) over 7000 angular combinations took approximately 25 minutes, though increasing the number of vectors above this value leads to excessive computing times. The location search and the superposition programs are both approximately as fast as in the example given.

#### CHAPTER 8

An X-Ray Investigation of the

Crystal Structure of Zeorin Acetate.

## Preliminary Data

The crystals were colourless orthorhombic laths,, elongated along [100] with (001) prominent. The dimensions of the crystal selected for data collection were approximately 0.40 x 0.15 x 0.12 mm. The structural formula proposed by Barton, de Mayo and Orr (1958) is shown in Figure 1.



Figure 1. The proposed structure of zeorin acetate.

Formula  $C_{32}H_{54}O_{3}$  Molecular Weight (formula) = 486 Unitcell Dimensions :  $a = 8.78 (0.02) \stackrel{\circ}{A}, b = 11.59 (0.02) \stackrel{\circ}{A}, c = 29.15 (0.04) \stackrel{\circ}{A}, V = 2967 \stackrel{\circ}{A^3}, Z = 4, F(000) = 1080 e.$ 

Mass absorption coefficient for  $CuKa, \mu = 5.2 \text{ cm}^{-1}$ 

$$D_{obs}$$
 (flotation) = 1.09 g.cm<sup>-3</sup>  $D_{calc}$  = 1.09 g.cm<sup>-3</sup>

Absent spectra only among

h00 for 
$$h = 2n + 1$$
  
Ok0 for  $k = 2n + 1$   
O01 for  $l = 2n + 1$ 

so that the spacegroup is uniquely determined as  $P2_{1}2_{1}2_{1}$ . Intensity data for the layers Okl to 7kl were estimated visually using three graduated wedges to take account of spot-shape changes on the higher layers. In view of the dimensions of the crystal used and the low value of the absorption coefficient, corrections for absorption were considered unnecessary. The data were correlated by comparison with reflexions common to hOl, giving a total of 2490 independent reflexions.

The Wilson plot based on three-dimensional data is shown in Figure 2. The data were divided into 25 shells each containing 110 - 140 reflexions, and unobserved reflexions were included as zero. The abnormality in the curve is not surprising in view of the structure proposed for zeorin acetate. It is evident from the Patterson (Figure 3) that many inter-atomic vectors accidentally coincide so that at certain  $\sin^2 \Theta$  values the intensity averages will be abnormally high. The least-squares method gave B = 3.2, K = 23.7 and



 $\mathcal{A}$ 

Figure 2. Zeorin acetate Wilson plot based on three-

dimensional data.

the graph B = 3.8, K = 25.8.

The approximatellength of the triterpenoid nucleus (i.e. the A, B, C, D and E rings) is 12 Å and the lengths of the a and b axes (8.78 Å and 11.59 Å) are too short to accommodate this dimension including the side chain. This strongly suggests that the length of the molecule lies in the c direction. A brief examination of the crystals with a polarizing microscope and quartz wedge showed  $n_c > n_a > n_b$ , (where n is the refractive index along the axial direction denoted by the suffix) but the birefringencies were not large. This confirmed that the molecules have their longest dimension parallel to c and suggested that their planes are roughly parallel to (010). The latter conclusion was corroborated by the observation that the two strongest reflexions are from (020) and (021).

#### Visual Examination of the Patterson

The unsharpened, three-dimensional Patterson function is shown in Figure 3 on pages /62 to  $/7^{o}$ , computed as sections normal to the a axis (the axes have been labelled x, y and z). The function is on an arbitrary scale and contoured at levels of 50 starting at 0. The computation did not include the term  $F(000)^{2}$  and the negative parts of the Patterson have not been contoured, though to avoid ambiguity the areas within the

# Figure 3.

Zeorin acetate

unsharpened Patterson.



 $X = \frac{O}{12}$ 



 $X = \frac{1}{12}$ 



x=<mark>15</mark>



 $X = \frac{3}{12}$ 



 $x = \frac{4}{12}$ 



 $X = \frac{5}{12}$ 



 $X = \frac{6}{12}$ 



broken contours are negative.

The height of the origin peak on an arbitrary scale is 2000 and this is proportional to  $\sum_{i=1}^{N} f_{i}^{2} = 5592$ . Thus, the peak heights corresponding to single inter-atomic vectors should be approximately :-

Vector	Peak Height
C - C	14
C - 0	17
0 = 0	24

These peaks will be indistinguishable in the negative areas of the Patterson, and are dwarfed by multiple peaks.

The set of points corresponding approximately to the postulated structure of the A, B, C and D rings is shown in Figure 4a. The system is puckered and the points marked ( • ) are regarded as above the plane of the paper. Figure 4b is the vector set of the points in Figure 4a; it forms three parallel 'sheets' and in broadside view the points make up an equilateral triangular mesh. Those in the central sheet are 'strongest' and form the larger mesh of side  $\approx 2.4$  Å. The points in the other two sheets are mirror images, offset by roughly 0.5 Å and approximately 1.5 Å apart . Peaks in the Patterson corresponding to the outer sheets will probably not be as easily distinguished as those in the central layer.

The Patterson does, in fact, exhibit a broad, flattish region of high density around the origin . It is fairly well



Figure 4a. The postulated structure of the A, B, C and D rings. The system is puckered, atoms marked ( • ) are regarded as being above the plane of the paper.



Figure 4b.

Part of the vector set of the points in Figure 4a. Points marked ( • ) lie in the central sheet Points marked ( • ) represent superimposed points athwart the central sheet. confined to the  $(u, \theta, w)$  plane, so all the molecules in the cell must be roughly normal to <u>b</u>. The peak clusters around  $(u, \frac{1}{2}, w)$  and  $(\frac{1}{2}, v, w)$ , presumably between the <u>b</u> and <u>a</u> screwrelated molecules also confirm the general alignment.

In the u = 0 section of the Patterson (page 163), the strong peak labelled (1) was attributed to the accumulation of vectors of the type  $\vec{v_1}$  in Figure 4a and peak (2) to those of type  $\vec{v_2}$ . The length of the vectors from the origin to these peaks (2.4 Å and 4.9 Å respectively) agrees with the calculated distances across the cyclohexane rings. The strong peak (3) (pages 165 and 166 ) was attributed to vectors of the type  $\vec{v_3}$  in Figure 4a, its distance from the origin being 2.5 Å. Thus, the origin peak and peaks (1) and (3) form the equilateral triangular mesh associated with the points in the central layer in Figure 4b. Their positions suggest two possible orientations for a zeorin acetate molecule as shown in Figures 5a and 5b.

The projection of the vector from the origin to peak (3) on to (001) makes an angle with (O10) of approximately,  $\measuredangle = 30^{\circ}$ . With the aid of Dreiding models it was apparent that this peak could arise from two possible orientations of the molecule seen 'end-on' as in Figures 6a and 6b (the vectors marked are as in Figure 4a). If the orientation is similar to that shown in Figure 6b, then there should be a peak in the Patterson







approximately 1.5 A from the origin, on or close to the <u>a</u> axis and corresponding to  $\overrightarrow{v_4}$ . No such peak was discernible. However, assuming an orientation similar to that in Figure 6a, it was possible to locate Patterson peaks corresponding to the vectors  $\overrightarrow{v_4}$  and  $\overrightarrow{v_5}$ . These peaks, (4) and (5) are shown in Figure 3 on pages 165 and 164, their distances from the origin being approximately 1.6 Å and 1.5 Å respectively.

## The Use of Patterson-Searching Techniques

The method and computer programs used have been described in Chapter 7.

As shown in Figures 4a and 4b, the vector set of a single molecule (postulated) consists of three parallel sheets (considering the ring system only). The peaks in the Patterson corresponding to the inmost sheet are the strongest and there o should be several less than 3.4 A (i.e. they will be clear of inter-molecular vectors). Although the longer vectors are more sensitive to angular changes, they mayrum into the region of inter-molecular vectors and be difficult to disentangle. It therefore seemed advisable at first to deal only with the shorter and medium-length vectors.

Considering only the rings A, B, C and D, they form .

an approximately centrosymmetric group with vectors too numerous and too long to use in initial searches. Computing time becomes excessive with more than about 78 vectors (i.e. all the vectors between 13 atoms). A cyclohexane ring is too small a unit and occurs too often either as complete or incomplete rings. A decalin ring is possibly acceptable but it was decided to use the group



as the computing times would still be reasonable and it would possibly be more selective than the ten-atom group; it can occur four times in the molecule and possibly more considering the five-membered ring and other substituent groups.

As the unit cell contains four symmetry-related molecules, the region round the origin in the Patterson contains sets of peaks corresponding to four overlapping sets of intra-molecular vectors. However, this is implicit in the mmm symmetry of the Patterson so that only one quadrant of space need be searched. Any multiple fits are then due to repetition of the search group within one molecule.

The structure of methyl melaleucate iodoacetate (Figure 7)

determined by Hall and Maslen (1965), was used to define the



Figure 7. Structural formula of methyl melaleucate iodoacetate.

geometry of a molecular fragment. The bond distances of the A, B, C and D rings are shown in Figure 8a, which also gives the approximate orientation of the molecule projected down [010]The naming of the axes in Figure 8a is not the same as that given by Hall and Maslen. The a and c axes have been interchanged so that the orientation is similar to one postulated for zeorin acetate. Hall and Maslen noted that the degree of substitution at carbon atoms bears a direct relationship to the lengthening of the C - C bonds. Long range steric effects may also account for some of the abnormal bond lengths and angles. As the substitution in zeorin acetate (Figure 8b) is probably very similar to that in methyl melaleucate iodoacetate, except at C<sub>3</sub>, it is reasonable to assume that the geometry of the A, B, C and D rings of the two compounds will not differ to a great extent.



Figure 8. (a) Bond lengths in methyl melaleucate iodoacetate. (b) The A, B, C and D rings in the postulated structure of zeorin acetate.

# Work with the Unsharpened Patterson

Orientation Search

As the unsharpened Patterson had given an approximate orientation of the molecule, it was decided to proceed with the unsharpened function for the search. This was computed on a 31 x 31 x 31 grid and punched on cards with PSP. As mentioned in Chapter 7, input format requires not more than two digits for each Patterson grid point. Consequently, the Patterson as shown in Figure 3 was scaled down so that most values fell within the range -48 to +48. Each value was then adjusted by addition of 48 to give the range 0 to 96. Any negatives were punched as 0 and those above 99 (mainly in the origin peak) The origin peak was not removed, which would reduce the as 99. power of the search in that the Patterson values at the ends of the vectors between adjacent atoms would be obscured by the origin peak in some orientations. In Figure 3, only peak (5) (of those visible) was affected in this way and it was considered that this would have little effect on the position of the minimum, especially as it is the coordinates of the longer vectors that are more susceptible to angular changes and hence define the position of the minimum more accurately.

The 66 vectors between the atoms C1 - C11, C14 of methyl melaleusate iodoacetate (Figure **g**a) were input to the orient-
ation search program (although they form rather a deformed group, especially about C3 and C5). Increments of 10<sup>°</sup> were initially employed to make computing time practical.

The highest peak in the rotational minimum function had LOW = 41, (O in Figure 3 corresponds to 48 in the Patterson used for the search), NEXT = 43 with A =  $0^{\circ}$ , B =  $60^{\circ}$  and C =  $-10^{\circ}$ , and corresponds approximately to a clockwise rotation, by  $60^{\circ}$  about the b axis, of the A and B rings (as shown in Figure &a), to give the orientation shown in Figure 9. This



# Figure 9.

corresponds quite closely with a possible position suggested by the preliminary investigations of the Patterson. Other peaks in the minimum function with LOW  $\approx$  36 occurred in the Eulerangle ranges

$$A = -30^{\circ} \text{ to } + 20^{\circ} \begin{cases} B = -70^{\circ} \text{ to } -60^{\circ} & C = 10^{\circ} \\ B = -30^{\circ} \text{ to } -20^{\circ} & C = 20^{\circ} \\ B = 85^{\circ} \text{ to } 95^{\circ} & C = 30^{\circ} \end{cases}$$

The first peak may be accounted for by a fit with the B/C rings

as shown in Figure 9.

<sup>T</sup>he increment was reduced to  $5^{\circ}$  and the Patterson searched over ranges that included all four peaks. <sup>T</sup>he minimum values for the latter three still remained at approximately 36, but for the first peak were LOW = 42, NEXT = 43. Decreasing the increment to  $1^{\circ}$  showed that this peak was in fact triplet with

Ao	во	င္ရွိ	LOW/NEXT
0	<b>5</b> 8	-14	43/43
-1	61	-13	43/43
-1	52	-14	41/41

Two of these peaks could be accounted for as representing the fit of the molecular fragment with the A/B and C/D rings respectively, suggesting that the C and D rings are as shown in Figure 9. The third peak which is fractionally weaker may represent the partial fit with a single ring and its appendages. However, the boundaries between the three peaks are ill-defined and the differences between the assumed and actual geometry must be a factor that contributes to this effect.

The orientation of the molecular fragment with  $A = 0^{\circ}$ , B = 58<sup>°</sup> and C = -14<sup>°</sup> is shown projected down the b axis in Figure 10a, and down the c axis in Figure 10b.



### Location Search

For the spacegroup  $P2_12_1^2_1$ , this process can be carried out by two separate two-dimensional searches. With the molecular fragment in the orientation  $A = 0^\circ$ ,  $B = 58^\circ$  and  $C = -14^\circ$ , the Patterson function was searched along the **X** and Y directions, computing vectors between the fragments related by the c screw. The appropriate ranges were :-

INX, Y = 0.0 DELX, Y = 0.01 FINX, Y = 0.50 o The increments are approximately 0.1 A and the searching was carried out in the unique part of the unit cell.

The translational minimum function is shown in <sup>F</sup>igure 11. Only the values of LOW are contoured at the levels 38, 40, 42 and 44. The background level was generally between 25 and 32. The peak heights agree quite well with those of rotational minimum function but as can be seen, the function is poorly resolved especially in the X direction, although the highest values are confined to two narrow ranges in the Y direction, from 0.34 to 0.37 and 0.07 to 0.10.

The translational search computing vectors across the <u>a</u> screw tended to confirm the ranges for the Y values shown in Figure 11. Outside these ranges the minimum function was generally of the order 25 to 32 although there were, in places, peaks as high as 39. <sup>Th</sup>e function also suggested that



directions.

two different Z ranges, 0.40 to 0.44 and 0.36 to 0.38, are associated with the two peaks at Y = 0.07 to 0.10 and Y = 0.34to 0.37 respectively. A third search computing vectors across the <u>b</u> screw did little to define the minimum more clearly and gave a minimum function in which the peak heights were generally lower than in the other two searches.

In an attempt to find unambiguously, the translational parameters  $(x_0, y_0, z_0)$  (page 144), two three-dimensional searches were carried out. Both were over the complete X range (0.0 to 0.5) but with different Y and Z ranges corresponding to those suggested by the two-dimensional minimum functions. The resulting peak heights (background level = 30 to 34) were generally lower than those in the twodimensional maps. The four highest peaks are shown in Table 1.

<u>Table 1</u>. The highest peaks in the three-dimensional, translational minimum function.

Peal	2	X	Y	Z	LOW/NEXT
A	٠	0.14	0.34	0.37	39/39
В		0.14	0.07	0.43	39/40
C		0.08	0.33	0.39	39/39
D		0.23	0.08	0.43	38/39

Table 1 shows that the differences between the heights of the four peaks are very small and thus from the results so far it was felt that the original Patterson was not sharp enough to define, unambiguously, the translational parameters  $x_0$ ,  $y_0$ ,  $z_0$ . However, two three-dimensional Fourier maps were computed using the coordinates of the two highest peaks in Table 1 (A and B) as values of  $x_0$ ,  $y_0$ ,  $z_0$  for the structure factor calculations. The agreement between the observed and calculated structure factors (scale of 20.0 applied to Fo) was 0.55 in each case. It was difficult to interpret the Fourier map derived from the coordinates of peak A in a way consistent with the expected geometry of the zeorin acetate molecule. The Fourier derived from peak B was a little more encouraging and gave tentative positions for 18 atoms as shown schematically in Figure 12. The peaks corresponding to the atoms marked 12 and 16 in Figure 12 were extremely weak. An attempt was



Figure 12.

made to refine these coordinates by the Fourier Refinement program of the X-Ray '63 System. This program, written by J.R.Holden calculates the 27-point block Fourier around each

atomic positions, estimates the coordinates of the point of maximum electron density within the block and then repeats the process with these peak positions as new atomic coordinates. The method, when applied to the 18 positions in Figure 12, reduced the R value to only 0.52 and some of the interatomic bond lengths attained unacceptable values. The shifts of the atomic positions were unsystematic and did not indicate that the group as a whole ought to be moved in any particular direction. A difference Fourier derived from the original positions was similarly uninformative.

## 2. Work with the Sharpened Patterson

In view of the ambiguity in the determination of the translational parameters,  $x_0$ ,  $y_0$ ,  $z_0$  that has been described, the method was repeated using the Patterson function sharpened to point atoms at rest by the modification

 $1/f^2 \cdot \exp(6 \sin^2 \theta/\lambda^2)$ 

where f is the formfactor for carbon. The resulting Patterson did not exhibit any serious diffraction ripple except in the neighbourhood of the origin peak and the zeroth section is shown in Figure 13 which can be compared with the corresponding section of the unsharpened Patterson on page /63. A sharpening

Y ጉ 0 0 Figure 13. Zeorin acetate 0 sharpened Patt-0 erson.  $\subset$ Section  $\mathbf{x} = 0$ 0  $\sim$ C Ô C

function of the type

exp ( 
$$6\sin^2 \theta / \lambda^2$$
)

similar to that used by Nordman and Nakatsu (1963), did not produced any marked differences between the sharpened and unsharpened Pattersons.

Orientation Search

During the course of the work with the unsharpened Patterson, there were indications that the distortion of the molecular fragment may have contributed to the poor resolution of the minimum functions. Consequently, the ten atoms comprising the B and C rings of Hall and Maslens's (1965) determination of methyl melaleucate iodoacetate were selected as molecular fragment. In the orientation search, vectors between adjacent atoms were excluded to avoid the effects of the ripple near the origin peak. The approximate orientation of the molecule had been fairly well defined, as described in the preceeding sections and so the ranges of the rotational search were limited to  $A = -30^{\circ}$  to  $+30^{\circ}$ ,  $B = -90^{\circ}$  to  $+90^{\circ}$  and  $C = -50^{\circ}$  to  $+50^{\circ}$ .

The minimum function showed four strong peaks for which LOW and NEXT were approximately 36 and 37 respectively compared with the background level of 10 - 25 (0 in Figure 13 was adjusted

to 36 for the search). After decreasing the angular increments the values shown in Table 2 were obtained for the four peaks,

Table 2. Rotational parameters of the four highest peaks

in the minimum function.

Peak	A	в	C <sub>O</sub>	LOW/NEXT
1	-5	-70	10	36/37
2	+7	-59	23	37/39
3	-2	47	-35	39/40
4	1	57	_19	37/38

although the angles could not be fixed within  $\pm 2^{\circ}$ . The orientation of the molecular fragment defined by the parameters of peak 2 in Table 2 is shown projected down the b axis in Figure 14a and down the c axis in Figure 14b and roughly agrees with the positions deduced in the earlier work. The values of the angles  $\alpha_1 = 39^{\circ}$  and  $\alpha_2 = 28^{\circ}$  can be compared with the corresponding angle of  $30^{\circ}$  obtained from peak (3) in the unsharpened Patterson.

Table 2 shows that the peaks occur in two pairs 1,2 and 3,4 related to each other by a rotation of approximately  $120^{\circ}$  about the b axis. The differences between peaks 1 and 2 can be explained by the fits of the molecular fragment with the A/B and C/D rings respectively, similar to the situation described for the unsharpened search (there were also three peaks with LOW  $\approx 32$  in the ranges A =  $-5^{\circ}$  to  $+5^{\circ}$ , B =  $-10^{\circ}$  to

 $+10^{\circ}$ , C =  $-10^{\circ}$  to  $+25^{\circ}$  which may have been due to fits with the B/C rings and the A and D rings plus appendages, but they were not investigated further). However, the differences between peaks 1 and 2 are in this case quite marked and strongly suggest that the molecule is situated in the unit cell as shown in Figure 15. This is different from that suggested by the Fourier



### Figure 15.

maps derived from the unsharpened Patterson.

Using Dreiding models it was seen that with the molecular fragment in theorientation corresponding to peak 2 (Figures 14a and 14b), the position of peak 4 is equivalent to an approximate rotation of the group through  $180^{\circ}$  about the c axis. The relationship between peaks 2 and 4 occurs because the ten atoms of the molecular fragment approximately form a centrosymmetric set. Rotating the vectors through  $180^{\circ}$  about a



crystallographic axis and making all their signs positive to deal with the symmetry of the spacegroup, will give a set of vectors nearly, but not quite, equivalent to the original set. Peak 1 must be related to peak 3 in the same manner.

## Location Search

The translational minimum function derived from the search along the Y and Z directions with the molecular fragment in the orientation  $A = 7^{\circ}$ ,  $B = -59^{\circ}$  and  $C = 23^{\circ}$  contained a large number of peaks with LOW = 32 to 39 compared with the backgroundlevel of 15 - 25. However, the number of peaks was far too large to enable any unambiguous assignment of values to the translational parameters y, z. It was noticed that the peaks occurred at very regular intervals, 0.7 A in Y and 1.2 A in Z. The search along X and Z showed a similar phenomenon the intervals being 0.4 A in X and 1.2 A in Z. These repeat distances are approximately one half of the projected interatomic distances in the molecular fragment shown in Figures 14a and 14b. In order to try and reduce this effect, the positions were calculated for a molecular fragment comprising two trans decalin rings both in the orientation  $A = 7^{\circ}$ ,  $B = 59^{\circ}$  and  $\dot{}$  $C = 23^{\circ}$ , and joined at C8 - C9 as in Figure 15. In the resulting translational minimum functions the periodicity was not reduced to any marked extent.

Attempts were made to interpret these results by postulating partial fits between the molecule fragment and the postulated structure of zeorin acetate, as the former was translated along the crystallographic directions in the orientation shown in Figures 14a and 14b, and taking into account packing considerations. These methods proved abortive and as it was becoming almost as difficult to interpret the translational minimum function as it would be to visually interpret the Patterson, the method was abandoned.

# Conclusion

In retrospect, there are several factors that may have contributed to the failure of this Patterson search technique as a method for the determination of the crystal structure of zeorin acetate.

Certainly for structures that have been solved by this method, the geometry of the molecular fragment has been well defined. The two alkaloid structures analysed by Nordman and Nakatsu (1963) and Nordman and Kumra (1965) contained planar oxindole and indole groups respectively and the solution of a pyrethosin derivative by Nordman and Gabe (1965) was attained using an isoxazolone group and three other atoms as molecular fragment. In both the sharpened and unsharpened work on zeorin acetate the rotational parameters A, B and C were not well defined, which must have resulted in part from the differences between the geometry of the trans decalin rings in methyl melaleucate iodoacetate and zeorin acetate. The vector refinement program described by Nordman (1966) would probably be useful in this respect but was not available for this work.

The 'periodicity' that was noted in the translational minimum function during the work on the sharpened Patterson suggests that more serious difficulties are present. Although it was not recognised at the time, this effect was also present in the unsharpened work, but to a less marked extent. It was felt that the effect was due to the fact that the molecular fragment must consist of at least a trans decalin ring system, as this is the most well-defined feature of the molecule. The structure postulated from the chemical evidence contains three such systems of the A/B, B/C and C/D rings and a similar system in the part structure defined by the D/E rings. It was conceivable that the 'periodicity' could be reduced by using a molecular fragment comprising a larger number of atoms, though the results in this respect were disappointing. In the previous structure determinations to which this method has been applied, the compounds have all possessed only one, unique part structure that was used as molecular fragment and their unit cells con-

tained only two molecules (i.e. spacegroup P2, in each case).

The Convolution Molecule method that has been developed by Hoppe (1957) and described briefly on page 145 offers an alternative approach to the determination of the structure of zeorin acetate that would possibly not involve some of the difficulties encountered here. Attention is, in fact, now being turned towards this method.

It does seem that this method is not seriously dependent upon an accurate knowledge of the geometry of a molecular fragment. Huber and Hoppe (1965) have described the structure determination of the steroid, ecdysone, in which a ring junction initially assumed to be trans fused was later shown to be cis. However, it is not clear whether the 'periodicity' effect would be a problem. The Convolution Molecule method would also overcome a defect in the Patterson search technique used here, that of applying vector weights. However, it should be a simple modification to incorporate a scheme similar to that described by Nordman (1966) into the Patterson search programs.

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Structural formula for 6-NBF-PA.



Structural formula for the 3-spiro-thiazolinium-7-PAC derivative.