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Production of BTX from Ethane

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Production of BTX from Ethane

Abstract

This process will produce 0.5 MM tons of benzene, toluene, and xylene (BTX) per year from a feedstock of 0.82 MM tons of fractional-grade ethane. The conversion of ethane to BTX occurs in a steam reformer reactor operating at 1170°F, using a germanium-incorporated H-ZSM-5 zeolite catalyst. After the reaction occurs, the light components are separated from the product stream using a multistage compressor and flash followed by the PRISM separation system. The light hydrocarbons (ethylene, ethane, propylene, and propane) are recycled to the reactor and the hydrogen and methane are burned. The final BTX product is separated from the heavy stream using two distillation columns, and the remaining heavy components are sold for use in gasoline. The plant will be located on the Gulf Coast due to the abundance of fracking operations in the area, allowing for easy access to ethane feedstock and gases used in the regeneration process. The process is currently unprofitable with an IRR of 2.87% and a net present value of -\$166,765,000 at a discount rate of 18%. The financials for this venture are highly sensitive to the price of catalyst components and catalyst lifetime. A few major reasons that make this venture unprofitable are the high cost of catalyst, the high equipment cost, and the short catalyst lifespan. After in depth analysis of the financials, we recommend that this project only be executed if market prices for input components significantly decrease, prices for BTX significantly increase, or catalyst lifetime significantly increases.

Disciplines

Biochemical and Biomolecular Engineering | Chemical Engineering | Engineering

PRODUCTION OF BTX FROM ETHANE

Design Project By:

Arthur Chen Fiona Crowley Jonathan Lym Pablo Sanchez

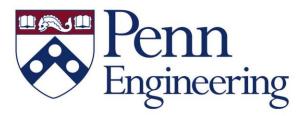
Presented To:

Professor Leonard Fabiano Dr. Daeyeon Lee

April 14, 2015

Department of Chemical and Biomolecular Engineering University of Pennsylvania School of Engineering and Applied Science Professor Leonard Fabiano Dr. Daeyeon Lee University of Pennsylvania School of Engineering and Applied Science Department of Chemical and Biomolecular Engineering 220 S 33rd Street Philadelphia, PA 19104 April 14, 2015

Dear Professor Fabiano and Dr. Lee,



We are pleased to present our process for the production of *BTX from Ethane* proposed by Dr. Richard Bockrath. Our plant, located along the Gulf Coast, is designed to produce 0.5 MM tons of BTX (62% Benzene, 30% Toluene, 8% Xylene) per year from 0.82 MM tons of fractional-grade ethane per year. This process utilizes the dehydroaromatization of ethane over a platinum zeolite catalyst using Group 14 elements such as germanium, tin and lead in its framework. This reaction takes place at 630°C and 1 atmosphere in a furnace with packed tubes similar to the reactor used for the steam reformation of methane. The products that are leaving the reactor are separated using flash vessels, distillation units, and PRISM membrane separators. Unreacted ethane, ethylene, propane, and propylene are recycled to minimize waste and maximize conversion. Hydrogen and methane generated in this process are burned in fired furnaces to generate heat for the process. Heavier hydrocarbons that are produced are sold as a gasoline additive to maximize profits. Detailed equipment designs and a preliminary economic analysis of the plant are enclosed within.

This method of BTX production would compete with two current methods of production, namely catalytic reforming of naphtha in petroleum refineries and steam cracking of hydrocarbons. Unlike its competitors, this process uses ethane, which has become much cheaper since the shale gas boom.

We have determined that at the current prices of raw materials, products and utilities, the process would be unprofitable. Our base case currently assumes a catalyst lifespan of four weeks; however it was found that in order to have a positive Net Present Value after 15 years, the catalyst would have to be repurchased no more frequently than once every 8.5 weeks. However, provided that the market outlook is positive in the future, this plant has potential to achieve a strong hold on the emerging market. Sincerely,

Auch de

- File

Pable St

Arthur Chen

Fiona Crowley

Jonathan Lym

Pablo Sanchez

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A-6.3. Dr. Richard Bockrath, DuPont	

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Abstract

Abstract

This process will produce 0.5 MM tons of benzene, toluene, and xylene (BTX) per year from a feedstock of 0.82 MM tons of fractional-grade ethane. The conversion of ethane to BTX occurs in a steam reformer reactor operating at 1170°F, using a germanium-incorporated H-ZSM-5 zeolite catalyst. After the reaction occurs, the light components are separated from the product stream using a multistage compressor and flash followed by the PRISM separation system. The light hydrocarbons (ethylene, ethane, propylene, and propane) are recycled to the reactor and the hydrogen and methane are burned. The final BTX product is separated from the heavy stream using two distillation columns, and the remaining heavy components are sold for use in gasoline. The plant will be located on the Gulf Coast due to the abundance of fracking operations in the area, allowing for easy access to ethane feedstock and gases used in the regeneration process. The process is currently unprofitable with an IRR of 2.87% and a net present value of -\$166,765,000 at a discount rate of 18%. The financials for this venture are highly sensitive to the price of catalyst components and catalyst lifetime. A few major reasons that make this venture unprofitable are the high cost of catalyst, the high equipment cost, and the short catalyst lifespan. After in depth analysis of the financials, we recommend that this project only be executed if market prices for input components significantly decrease, prices for BTX significantly increase, or catalyst lifetime significantly increases.

Introduction and Objective-Time Chart

Due to the recent increase in fracking in the United States, there is a high availability of light paraffinic hydrocarbons to be used as feedstocks, especially ethane. While research into alkanes-toaromatics processes was more popular in the 90s, with advances by companies such as Cyclar and Aromax, these processes fell out of favor due to the difficulty in finding a suitable catalyst. However, a recent patent by Shell showed promising conversions and selectivities to BTX for a catalyst consisting of a germanium incorporated into a zeolite.

This project is also motivated by the decrease in traditional sources of BTX, leading to a demand for new methods of production. BTX is typically made through catalytic naphtha reform or by extracting from naphtha-fed ethylene crackers, both of which require relatively expensive crude oil. As ethane is a common by-product of fracking, it is much cheaper, thus allowing for very high potential profits from this process.

Completion Date	Milestone
February 3	Submit preliminary material balance and computer-
	drawn block flow diagram.
February 24	Submit base case material balance and computer-
	drawn process flow diagram.
March 17	Submit detailed equipment design for a key process
	unit.
March 24	Major equipment designed.
March 31	Finances completed.
April 7	Written reports due.
April 14	Revised written reports due.
April 21	Design presentations.

Table 1. Objective Time Chart of Project Milestones

Chemistry Background

The general reaction mechanism is shown in Figure 1. Alkanes, such as ethane and propane, undergo dehydrogenation to form alkenes. This step is the rate-limiting step for the process. The alkenes then oligomerize to form longer chain hydrocarbons. The oligomers then cyclize to form naphthenes and then are dehydrogenated to form aromatics, the final product. Hydrogen is produced throughout this process and causes the reactions to be endothermic.

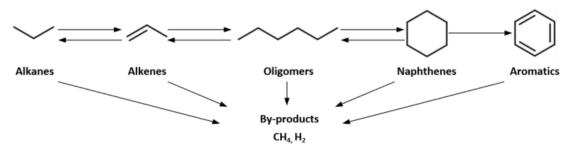


Figure 1. General mechanism for the aromatization of alkanes¹.

While the general mechanism of this process is well-known, only three stoichiometric equations of the conversion of ethane to benzene, toluene and xylene could be found². Otherwise, no information could be found on the exact stoichiometry or the kinetics of the steps. Table 2 shows the chemical equations that have been proposed for the project and the components of the reactor effluent were assumed to be the same as the results provided in US Patent 209,795³ and US Patent 324,778⁴.

Table 2. Reaction network and heat of reaction per mass of C2 or C3 hydrocarbon consumed. Note that the reactions	S
in which ethane and propane are consumed are endothermic.	

Reaction Number	Stoichiometric Reaction	ΔH _{rxn} BTU/lb of C2 or C3 consumed
1	$C_2H_6 + H_2 \rightarrow 2CH_4$	-3.81
2	$C_2H_6 \rightarrow C_2H_4 + H_2$	8.04
3	$3C_2H_6 \rightarrow 2C_3H_6 + 3H_2$	5.76
4	$3C_2H_6 \rightarrow 2C_3H_8 + H_2$	0.905
5	$3C_2H_6 \rightarrow C_6H_6 + 6H_2$	5.92
6	$4C_2H_6 \rightarrow C_7H_8 + CH_4 + 6H_2$	4.60
7	$4C_2H_6 \rightarrow C_8H_{10} + CH_4 + 7H_2$	4.61
8	$5C_2H_6 \rightarrow C_9H_{12} + CH_4 + 7H_2$	7.56
9	$C_2H_4 + 2H_2 \rightarrow 2CH_4$	-13.3
10	$3C_2H_4 \rightarrow 2C_3H_6$	-2.56
11	$3C_2H_4 + 2H_2 \rightarrow 2C_3H_8$	-8.03
12	$3C_2H_4 \rightarrow C_6H_6 + 3H_2$	-2.38
13	$4C_2H_4 \to C_7H_8 + 2H_2 + CH_4$	-3.87
14	$4C_2H_4 \to C_8H_{10} + 3H_2$	-3.86
15	$4C_2H_4 \to C_9H_{12} + 2H_2 + CH_4$	-0.543
16	$2C_3H_8 \rightarrow C_6H_6 + 5H_2$	3.08
17	$3C_3H_8 \rightarrow C_7H_8 + C_2H_6 + 5H_2$	2.21
18	$3C_3H_8 \rightarrow C_8H_{10} + CH_4 + 5H_2$	1.70
19	$3C_3H_8 \rightarrow C_9H_{12} + 3H_2$	4.86
20	$C_3H_6 + H_2 \rightarrow CH_4 + C_2H_4$	-1.07
21	$2C_3H_6 \rightarrow C_6H_6 + 3H_2$	0.104
22	$2C_3H_6 \rightarrow C_7H_8 + C_2H_6 + 2H_2$	-0.803
23	$3C_3H_6 \rightarrow C_8H_{10} + CH_4 + 2H_2$	-1.34
24	$3C_3H_6 \rightarrow C_9H_{12} + 6H_2$	1.97

Many of the reactions shown in Table 2 above are endothermic. Therefore the option of supplying heat to the reactor was considered when designing the reactor so that the reactants would remain at the reaction conditions of 1170°F.

Process Charter

Project Name: BTX from Ethane

Project Champion: Richard Bockrath, Process Engineer - formerly DuPont

Daeyeon Lee, Associate Professor - University of Pennsylvania

Leonard Fabiano, Adjunct Professor - University of Pennsylvania

Project Leaders: Arthur Chen, Fiona Crowley, Jonathan Lym, Pablo Sanchez

Specific Goals: Design a preliminary competitive plant and process to create 0.5 million tons of BTX/year using fractional grade ethane.

Project Scope:

In Scope

- Fully specified equipment lists (sizing, materials, etc.) and subsequence process design
- Stoichiometric reactor model
- Catalyst specifications
- Methods of separation
- Size and configuration of recycle loops
- Cost analysis

Out of Scope

• Kinetic reactor model

Deliverables:

- Completed process model showing every piece of equipment and its specifications, including operating parameters.
- Capital cost and financial economic analyses demonstrating feasibility and profitability of the design.
- Written report describing our findings/design.
- PowerPoint presentation of our findings/design.

Timeline:

- Weekly progress reports until final report submitted.
- Written design report completed by Tuesday, April 14, 2015.
- Oral presentation given on Tuesday, April 21, 2015.

Innovation Map

Recent breakthroughs in technology have made this process attractive. The innovation map, shown in Figure 2, shows how certain technologies affect the process and benefits customers.

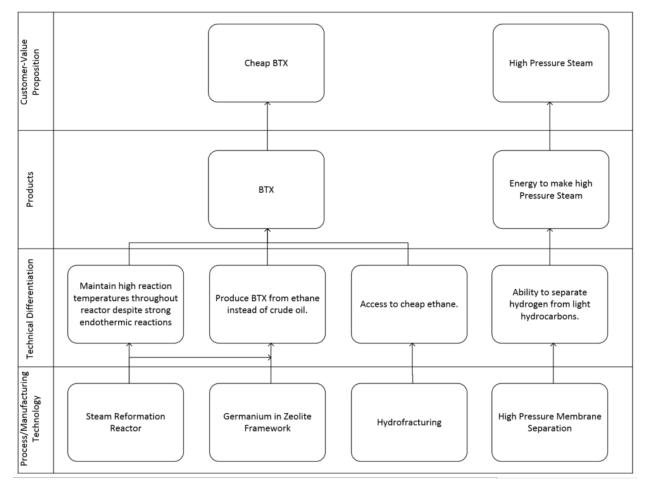


Figure 2. Innovation Map for the Proposed Process Design.

Market and Competitive Analyses

Market Analysis

The Use of Ethane over Oil

Currently, BTX is primarily produced in plants using the method of naphthalene reformation from crude oil. As demand for crude oil inevitably increases in the near future, alternatives to the use of crude oil have been a major area of research due to the possibility of permanent increases in price over time. Besides economic reasons, the use of crude oil in general has become increasingly unpopular in recent years due to stigma regarding its importation from foreign sources and negative environmental impact. Due to the emergence of the hydraulic fracturing process to economically retrieve previously unreachable natural gas in shale rocks, the price of United States natural gas has fallen by 80% in the past five years from a 2008 price of about \$10 per 1000 cubic feet to a current price of between \$2 and \$3 per 1000 cubic feet⁵. This low in the price of natural gas has opened an entirely new area of research into new uses of natural gas that could not have been considered earlier due to pricing issues. The study of the feasibility of producing BTX from natural gas components has been motivated by this price decrease, including a patent that converts propane to BTX. However, fractional grade ethane, which can be purchased as undesired byproduct from power plants, is substantially cheaper and readily available, and therefore worth studying as a feedstock for BTX production. For the purposes of this design project, the price of ethane is \$0.067 per pound and is tabulated in Table 5.

Natural gas has traditionally been used as a source of energy and is currently the second largest energy source in the world behind crude oil. The use of natural gas for energy is preferred in our increasingly environmentally conscious world for reasons besides its economic benefits; its combustion releases 30% less carbon dioxide than oil and up to 45% less carbon dioxide than coal per Btu of heat produced. It is also in abundance in the U.S. as a domestic energy source with proven resources at 354 trillion cubic meters as of 2014, making up 5% of the world's proved reserves. Figure 5 shows the distribution of natural gas reserves by state in the United States.

The recent drop in crude oil prices is also a consideration, as this makes the traditional naphthalene reformation process of producing BTX more economically feasible. However, both natural gas and natural gas liquid (e.g. ethane) prices have dropped just as much, as seen in Figures 3 and 4. While the drop in crude oil prices has made some of the least profitable fracking companies stop drilling,

the natural gas prices have not appeared to have been affected. If anything, natural gas prices have fallen alongside those of crude oil, making our process just as competitive as before the price drop.



Figure 3. Natural gas prices from April 2014 to April 2015⁵.



Figure 4. Ethane spot prices (per gallon) from December 2014 to April 2015⁵.

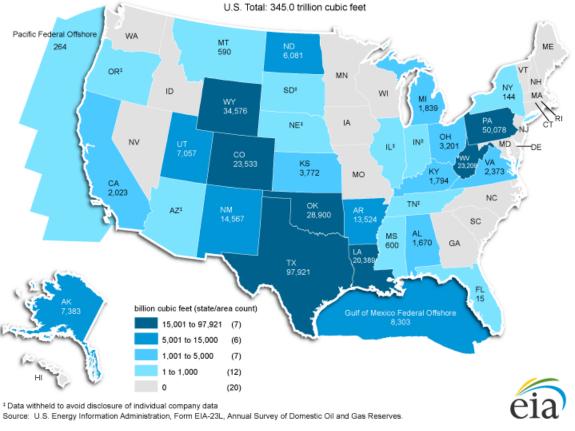


Figure 5. Natural gas proved reserves by sales/area, 2013⁵

Benzene

Benzene is the starting chemical reagent to create a wide array of aromatic chemical products. It is the most widely used aromatic petrochemical used in industrial processes. In the United States, about 50% of benzene is used to produce styrene, 20% is used to produce cumene, and 15% is used to produce nylon. Although the demand for benzene has been relatively stable in recent years, there are worries that weakening demand for polystyrene may negatively impact future prices of benzene. Despite this outlook, however, the price of benzene has been increasing in the past five years to a current high. For the purposes of this report, the price of our BTX mixture is taken to be \$0.41 per pound as tabulated in Table 6.

Toluene

Toluene is a common organic solvent and is commonly used as paint thinner. The largest industrial use of toluene is in the production of benzene and xylene – making up over 46% of annual consumption in the United States. It is also used to make dyes, photographic chemicals, and pharmaceuticals. Its use is limited in consumer applications due to its status as a known carcinogen. The price of toluene is

considered by the Independent Chemical Information Service as being volatile and is correlated to the price of gasoline. Because the majority of BTX produced is via crude oil cracking, as the price of crude oil has been increasing in the recent past, so has the price of toluene. It is predicted that production outages and political tensions are likely to keep the price of toluene volatile in the foreseeable future. For the purposes of this report, the price of our BTX mixture is taken to be \$0.41 per pound as tabulated in Table 6.

Xylene

Para-xylene is the most valuable product per pound that is created in the production of BTX. It is used in the large scale to create the polyester PET, which is used to produce clothing fibers, plastic bottles, film, and other synthetic plastic and polymer products. The demand in the market for p-xylene is therefore highly correlated with the demand for PET polyesters and their derived products. Tecnon OrbiChem, a UK-based consulting firm, estimates world consumption to increase by 7% per year. This represents an attractive marketplace for new entrants such as our company. O-xylene is used in industry as a necessary agent to create plasticizers. The main application however for mixed xylenes is to be converted to more valuable p-xylene which can be then used for a wide variety of applications listed earlier. For the purposes of this report, the price of our BTX mixture is taken to be \$0.41 per pound as tabulated in Table 6.

BTX Market

The market for BTX is currently valued at \$80.8 billion in sales per year. According to IBISWorld reports, the market for BTX production is mature and very heavily regulated²⁶. Although the industry is profitable and expected to grow at an average rate of 2.7% over the next four years to \$96.2 billion in 2018, performance is heavily influenced by high levels of volatility. This level of growth is expected to be fuelled by the growth of the chemical and plastic product manufacturing industry. For example, the demand for housing starts which increase the demand for insulation derived from para-xylene is expected to grow at an annualized rate of 11.1%. The market for BTX is considered at this point to be mature due to product saturation and market acceptance. It is also outlined by IBISWorld US in 2012. It is still possibly an attractive industry to enter, however, because of an expected shortage in supply to match the increase in demand expected through 2020. The primary markets are those in the domestic chemical manufacturing industry – only 1.8% of revenue of 2015 is predicted to be due to exports.

Competitive Analysis

Competition in this market is seen to be medium but increasing due to the growth rate of the BTX industry and the potential for innovative methods of its production. The two largest competitors in this industry are Exxon Mobil Corporation and The Dow Chemical Company. Anellotech Inc. is another smaller competitor that aims to produce BTX products from renewable sources.

- Exxon Mobil Chemical: Exxon Mobil Chemical is the petrochemical arm of Exxon Mobil Corporation. Exxon is one of the world's largest petrochemical companies. It is the largest North American producer of benzene and toluene and the second-largest producer of mixed xylene – holding an overall market share of 3.0%. Exxon, the country's largest oil refiner, is currently expanding their Texas facilities to take advantage of the close proximity to the gulf as well as further integrate the petroleum and petrochemical refining process. Its sales are estimated at \$2.45 billion per year and it is expected to grow at an annual rate of 15.8%.
- 2. The Dow Chemical Company: Dow produces more than 5000 products in 197 manufacturing facilities in 36 countries. It is one of the largest petrochemical companies in the world. Unlike Exxon, Dow has its materials transferred to its chemical plants at a net cost. Dow currently operates six business segments with feedstocks and energy being the only relevant business segment to the petrochemical industry. In the five years to 2015, IBISWorld estimates that Dow's industry-specific revenue has increased at an annualized rate of 10.3% to \$4.4 billion. It holds an overall BTX market share of 2.6%. Its sales are estimated at \$2.1 billion per year.
- 3. Anellotech Inc.: Anellotech has developed a clean technology platform to inexpensively produce BTX from renewable biomass. It is expected that their products will be inexpensive compared to petroleum derived counterparts while providing identical benefit. Anellotech will own and operate its own plants and sell proprietary technology to licensees.

Porter's Five Forces

The Porter's Five Forces framework is widely used in industry to identify and evaluate the competitive forces that must be considered when entering a prospective market. It was formed by Michael Porter of the Harvard Business School in 1979 as a tool to derive five market forces that determine the competitive industry and therefore the potential for profitability in a target market. The five forces are the (1)

competitive rivalry within an industry, (2) bargaining power of suppliers, (3) bargaining power of customers, (4) threat of new entrants, and (5) threat of substitute products. By carefully considering how each of these forces affects our position in our target market, a more informed decision to enter the market can be made.

1. Competitive Rivalry Within the Industry

Currently, BTX is produced mostly via plants that incorporate naphthalene reforming processes from crude oil. The potential for new innovative processes to maintain a competitive advantage is inherently low as there are relatively few sources that hydrocarbons can be economically derived from sources other than fossil fuels. According to ICIS, some markets may be saturated in terms of the amount of BTX that is floating. In Asia, for example, there is an excess of benzene that is causing a cut in production to avoid a drop in price. Because existing BTX producers are dependent on the price of crude oil in order to justify the profitability of their operations, there is clearly a market opportunity in presenting an alternative and possibly more economical reagent such as natural gas. Because there are so many large competitors in the BTX production industry, however, competitive rivalry is expected to be high.

2. Bargaining Power of Suppliers

Because the natural gas that is needed for our process is plentiful in the Gulf Coast region, it is unlikely that suppliers will be able to pose a significant threat to our industry. This is because the infrastructure needed in the transportation of natural gas offsite is very expensive and the price of natural gas is determined largely by market forces. It would be relatively easy to switch to another supplier since the costs of changing suppliers will be low due to the high concentration of refineries in the region. Other suppliers, such as that of electricity and catalyst components, may have more bargaining power based on their higher flexibility to cater to other customers.

Suppliers of proprietary equipment needed to carry through with this process design have significant bargaining power that must be accounted for in making our final economic decision. Proprietary equipment such as the PRISM separation unit that must be purchased from outside vendors incurs a significant cost due to their necessity in our systems.

3. Bargaining Power of Consumers

The main threat that players in the BTX production industry face is the possibility of backward integration by consumers which would cut demand from producers like our company. Just as Coca-Cola recently invested in renewable para-xylene for their bottles, other customers may be more inclined to backwards integrate if the price of their supplies reaches a critical threshold. Many of the products that are derived from BTX incorporate a large portion of their overall costs from the price of BTX and so customers will be very price sensitive. Our BTX is undifferentiated from that of other companies, which causes a threat to us in terms of switching to a competitor. However, since we are treating ourselves as our own customer for BTX, these concerns are alleviated.

4. Threat of New Entrants

As the price of crude oil continues to decrease, there will continue to be an increasing economic interest in creating a method of producing BTX from alternative sources. In 2011, the Coca-Cola Company has invested in Geno, Virent, and Avantium partnerships in an aim to produce para-xylene for its bottles in a completely renewable fashion. This investment represents the kind of vertical integration that is very threatening for competitive players in this industry especially considering the increasing price of all BTX products.

In addition, barriers to entry into the industry are high, and are expected to remain that way. This industry faces a number of barriers to entry, a major one being the level of capital required. The costs to construct a plant can range anywhere from a couple hundred million to billions of dollars. For example, Dow Chemical Co. is in the works of securing permits to construct a \$1.7 billion plant in Freeport, TX. Aside from the costs of building a plant, regulatory costs are high as well. Because the industry deals with chemical waste and emissions, companies are subject to rules set by the Environmental Protection Agency. Moreover, given the heavy reliance on various feedstocks, such as natural gas liquids and petroleum, the ability to access a steady supply of competitively priced raw materials is essential. Many of the established players are part of integrated oil companies that operate in integrated oil refining and petrochemical complexes, a position that gives them a significant competitive edge over potential stand-alone newcomers. Significant competition also comes from imported products. Infrastructure in oil rich nations, such as Saudi Arabia, continues to satisfy a large portion of domestic demand.

5. Threat of Substitute Products

The use of BTX in the products outlined in the Market Analysis section largely does not face a threat by substitute products. Benzene in particular cannot easily be replaced in industrial processes because it is the simplest six membered aromatic substance and many chemical processes rely on it to form product. The demand for Toluene is more price-sensitive because its use as a solvent can be replaced with other products. Para-xylene yields the most threat of substitute products because its use in hard plastic containers (PET) is threatened by a push to produce more renewable hard plastics. It should be considered, however, that with the overall trend toward higher prices for BTX that there will be an economic incentive to find substitutes and so there will be an increased effort in this area.

Preliminary Process Synthesis

Location of Plant

The two major concerns when determining the location of the plant were availability of ethane feedstock and cost of transporting the final BTX product. Our company is both a producer and a consumer of BTX, with customers in the Gulf Coast, Rotterdam, and Shanghai. The final decision to place the plant on the Gulf Coast was made due to both the ease of access to fractional-grade ethane from the numerous fracking operations in the area and to the proximity of BTX consumers. The cost of shipping liquid BTX is far lower than that of shipping ethane gas; therefore, the product can also be shipped to customers farther afield more economically. Additionally, the Gulf Coast has conveniently located pipelines for some of the reagents used for the catalyst regeneration process (i.e. nitrogen gas, oxygen gas, hydrogen gas).

Reactor Model

An initial choice between fixed or fluidized bed models was made by referring to US Patent $209,795^3$, which described the catalyst used in this process. In their experiments, the authors used a fixed bed packed with extrudate pellets; this and the high price of designing and scaling up a fluidized bed reactor motivated the choice of a fixed bed.

Due to the high temperature and endothermic nature of the reaction, a series of packed bed reactors would not be an efficient model, as seven reactors would be required to limit the temperature drop in each to less than 100°F. Therefore, the steam reformer model was selected, in which the feed is passed through tubes packed with catalyst, with direct heating on the outside of the tubes to maintain the necessary heat.

Catalyst Choice

The problem statement recommended two catalysts developed in US Patent 324,778⁴, each composed of a zeolite with platinum deposited on its surface, one with germanium incorporated into the zeolite structure and the other with titanium. The titanium catalyst had a higher overall conversion of about 80%, but a lower selectivity to BTX of about 40%. The germanium catalyst, on the other hand, had an overall conversion of about 40% but selectivity to BTX of about 52%. The patent gave conversions and selectivities for the conversion of propane to BTX, but these values were assumed to be the same for ethane to BTX. However, another patent, US Patent 209,795³, was found, with a germanium catalyst specifically for the conversion of an ethane feedstock to BTX. This patent provided a more detailed breakdown of product selectivities and had a higher overall conversion (46.6%) and selectivity for

aromatics (67.68%). The higher selectivity for BTX and the more knowledge available made the germanium catalyst a more attractive choice for this process.

US Patent 209,795³ gave selectivity and conversion information of many catalysts. The tables found in the patent have been reproduced below. Table 3 shows catalysts A through I, which have varying levels of platinum and tin. Table 4 shows catalysts J through N, which have varying levels of platinum and germanium. Catalyst L was chosen as it offered the best combination of high conversion, high selectivity of BTX and low selectivity of inert byproducts like methane and C9+ hydrocarbons. Note that the conversion to C4 and C5 hydrocarbons was so low that it was assumed to be negligible. A catalyst with germanium present also benefitted the project as US Patent 324,778⁴ could be used to provide conversion and selectivity data for any recycled propane to the reactor.

Catalyst	Α	В	С	D	Ε	F	G	Н	Ι
Analyzed Pt Level, % wt.	0.006	0.011	0.025	0.0437	0.04	0.1	0.103	0.123	0.233
Analyzed Sn Level, % wt.	0.005	0.01	0.012	0.0395	0.093	0.076	0.0601	0.11	0.217
Ethane Conversion, %	44.4	44.72	48.02	50.39	45.42	55.44	61.62	55.73	56.6
		Select	ivities %	% (C basi	s)				
Methane	15.68	18.84	24.22	24.62	21.1	30.16	38.39	29.85	29.67
Ethylene	13.86	14.18	12.6	11.17	9.84	10.37	8.88	9.46	9.3
Propylene	2.23	2.13	1.63	1.36	1.27	1.1	0.85	0.97	0.86
Propane	1.67	1.69	1.4	1.23	1.56	0.91	0.64	0.84	0.76
C4 Hydrocarbons	0.46	0.43	0.33	0.28	0.28	0.24	0.18	0.21	0.19
C5 Hydrocarbons	0.04	0.04	0.01	0	0.01	0	0.01	0.01	0.04
Benzene	35.19	37.18	36.61	34.32	36.54	31.67	30.39	31.59	31.69
Toluene	18.48	19.28	18.27	18.05	19.17	16.34	14.85	16.53	15.94
C8 Aromatics	3.73	3.83	3.27	3.7	3.82	2.9	2.57	3.3	3
C9+ Aromatics	8.68	2.4	1.65	5.24	6.41	6.31	3.24	7.24	8.56
Total Aromatics	66.07	62.69	59.81	61.31	65.94	57.22	51.05	58.66	59.16
Total Desired Aromatics	57.4	60.29	58.15	56.07	59.53	50.91	47.81	51.42	50.63

 Table 3. US Patent 290,795 Selectivity and Conversion Data of Catalysts A through I, which all have different levels of platinum and tin.

Table 4. US Patent 290,795 Selectivity and Conversion Data of Catalysts J through N, which all have different levels of platinum and germanium. Note that the selected choice has been highlighted in gray.

Catalyst	J	K	L	Μ	Ν
Analyzed Pt Level, % wt.	0.046	0.0436	0.0441	0.0436	0.122
Analyzed Ge Level, % wt.	0.0216	0.0442	0.0844	0.121	0.1235
Ethane Conversion, %	46.94	46.39	46.6	45.07	50.16
Sele	ctivities 9	% (C bas	is)		
Methane	22.65	18.24	16.27	15.36	20.81
Ethylene	9.51	11.97	12.67	12.96	11.33
Propylene	1.19	1.5	1.54	1.65	1.2
Propane	1.44	1.48	1.47	1.62	1.15
C4 Hydrocarbons	0.27	0.32	0.34	0.36	0.29
C5 Hydrocarbons	0.03	0	0.04	0.02	0.03
Benzene	35.52	37.4	37.28	36.68	35.43
Toluene	19.25	19.8	19.85	19.73	18.35
C8 Aromatics	4.07	3.68	4.15	3.95	3.7
C9+ Aromatics	6.06	5.6	6.4	7.67	7.71
Total Aromatics	64.91	66.48	67.68	68.03	65.2
Total Desired Aromatics	58.84	60.88	61.28	60.36	57.48

Catalyst Regeneration

The regeneration process was modeled after that given in US Patent 154,079⁶, which laid out a five-step regeneration process for catalyst similar to that used in this process. It suggested a possible sixth step, in which the catalyst was sulfided. Sulfidation is used in regeneration processes to minimize coking, as sulfides break down groups of platinum on the surface of the catalyst into smaller ensembles, making it more difficult for coke to form. However, the sulfidation step was rejected due to the already lengthy nature of the regeneration process (the estimate used in this analysis was five and a half days) and to the potential poisoning of the catalyst.

Separation of Hydrogen and Methane

Hydrogen and methane are produced in the reactors (see Figure 7) and have a detrimental effect on the equipment. The equipment that process hydrogen and methane had to be scaled larger to accommodate these species. Also, the separation of these components was necessary before using distillation columns as these species reduced the saturation temperature and increased the saturation pressure greatly. As a result, these components had to be removed.

The PRISM unit (see Figure 8) in the process is used to separate hydrogen and methane. However, a pressure-swing adsorption system was considered for the process as well. The PRISM system was preferred as it was difficult to find a suitable surface that would be able to bind all the other components of the stream. Additional costs would arise in order to refrigerate the system to temperatures to desorb its contents. Also, a pressure-swing adsorption system would complicate the process as the batch nature of the adsorption and desorption process will have to be managed.

Purge

The most expensive pieces of equipment in the process are the compressors (see Figure 8) valued at \$18,221,847 each; the reactors (see Figure 7) valued at \$55,000,000; and the PRISM Separation Unit (see Figure 8) valued at \$17,000,000. These units are so expensive due to the large flow rates around the process. The flows could be reduced by implementing a purge stream. Stream S-211 (see Figure 8) is the optimal candidate to purge as it contains no BTX product. If this stream were purged and its contents burned, more high-pressure steam could be generated and sold and the size and, as a result, the cost of several pieces of equipment could be reduced due to the lower demand on the units (heat exchangers H-201, H-202, and H-203, PRISM unit PR-101, turbines T-101, T-102, and T-103, reactor R-101, R-102, R-103, compressors C-201 and C-202). On the other hand, more fractional grade ethane would have to be obtained as some of the purged stream contains valuable reactants that could have been recycled into the

reactor; the cost of the steam boilers would have to be adjusted to accommodate the larger heat duty; and the turbines in Section 1 (see Figure 7) would produce less electricity. Unfortunately, the purge option could not be investigated in full detail as estimates for the costs of the PRISM unit and the reactors with the adjusted flow rates could not be obtained. However, it remains an option for potential future study if this project is taken into further development.

Lifetime of Catalyst

As testing of this catalyst has only occurred at a lab scale and information on its lifespan was not provided in the patent, an assumption of catalyst lifetime had to be made. Of catalytic processes commonly used, this is most similar to naphtha reform, which uses platinum alloys supported on alumina⁷. This process occurs at harsh operating conditions, with temperatures ranging from 860 to 980°F and pressures of 120 to 740 psi. Catalysts can last up to six months; therefore, a lifespan of one month was used as a conservative assumption for the germanium-incorporated zeolites used in this process.

Assembly of Database

Thermodynamic and Transport Data

All transport and thermodynamic data were taken from the ASPEN Plus Program and used directly in models. The Redlich-Kwong equation of state was the thermodynamic model used for gaseous species and the NRTL equation of state was used for liquid species.

Pricing Data

The prices for the inputs and outputs of the process were compiled from various sources.

Feed Materials

Table 5: Mass flow rate and	price for feed materials ⁶⁻¹⁵
Tuble 5. Muss now rule und	price for feed materials.

Material Name	Mass Flow Rate (lb/year)	Price (\$/lb)
Fractional Grade Ethane	1.47E+09	0.067
Chlorine	1.07E+07	0.68
Oxygen	4.12E+07	0.09
Hydrogen	6.78E+06	0.64
Nitrogen	3.39E+08	0.18

Products

Table 6: Mass flow rate, density and price for products of the process.^{14,15}

Product Name	Mass Flow Rate (lb/year)	Density (lb/gal)	Price (\$/gal)	Price (\$/1000 lb)
BTX	9.04E+08	7.26	3.0	-
Gasoline Blend	9.26E+07	7.31	3.0	-
High Pressure Steam (450 psig)	8.06E+09	-	-	8.0

Catalyst

Table 7: Mass required per year and the price of the catalyst components. Note that this mass assumes that catalyst has to be purchased every month.¹⁶⁻¹⁸

Catalyst Component	Mass Required (lb/year)	Price (\$/lb)		
Germanium	59840	862		
Platinum	31210	18240		
Zeolite	5.66E+07	5.67		
Alumina	1.42E+07	0.159		

Utilities

Table 8: Amount of each utility used and the corresponding price.¹⁴

Utility	Amount Required (Unit/year)	Price (\$/Unit)		
Cooling Water	3.20E+10 lb	0.0120/1000 lb		
Process Water	8.06E+09 lb	0.0962/1000 lb		
Electricity	2.59E+08 kWh	0.07/kWh		

Safety and MSDS

The Material Safety Data Sheets for the chemicals used in this process are provided in Appendix A-4: Thermophysical Data & Material Safety Data Sheets for reference.

Process Flowsheet and Material Balances

For the ease of reading, the following process is divided into four connected sections. An overview has been provided in Figure 6. A description of the process may be found under the Process Description Section. The process flow diagram will be presented on the left and mass flow of each material, temperature, pressure, vapor/liquid fraction, and enthalpy is shown on the following pages.

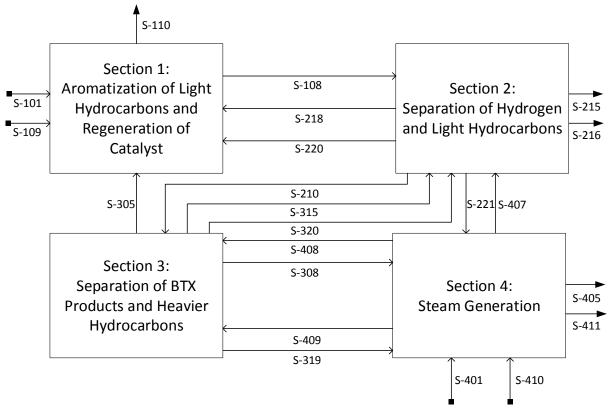


Figure 6. Overview of the interactions among the four different sections of the process.

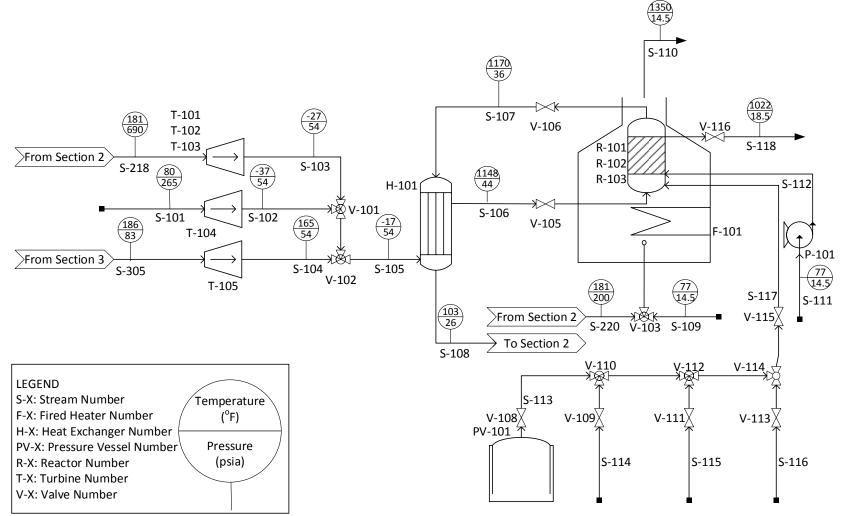


Figure 7. Section 1: Aromatization of Light Hydrocarbons and Regeneration of Catalyst

	Mass Flow (lb/hr)									
Species	S-101	S-102	S-103	S-104	S-105	S-106	S-107	S-108	S-109	S-110
Hydrogen	0	0	2758	Trace	2758	2758	18388	18388	0	0
Methane	989	989	115104	2717	118810	118810	162584	162584	0	0
Ethane	179794	179794	202499	7038	389331	389331	209537	209537	0	0
Ethylene	0	0	21482	608	22090	22090	22090	22090	0	0
Propane	5436	5436	8316	711	14463	14463	9027	9027	0	0
Propylene	0	0	2630	143	2774	2774	2774	2774	0	0
Benzene	0	0	9641	8373	18014	18014	88283	88283	0	0
Toluene	0	0	397	835	1232	1232	35905	35905	0	0
P-Xylene	0	0	3.73	23.7	27.5	27.5	9178	9178	0	0
1,3,5-Trimethyl										
Benzene	0	0	Trace	Trace	Trace	Trace	11739	11739	0	0
Water	0	0	0	0	0	0	0	0	0	42201
Oxygen	0	0	0	0	0	0	0	0	63797	10633
Nitrogen	0	0	0	0	0	0	0	0	210109	210109
Carbon Dioxide	0	0	0	0	0	0	0	0	0	21573
Total Flow (lb/hr)	186219	186219	362832	20449	569500	569500	569504	569504	273906	284516
Total Flow (cuft/hr)	114604	490675	1371780	67664	1960370	9149310	13895200	6551820	3768190	14251200
Temperature (°F)	80.0	-37	-27	165	-17	1148	1166	103	77.0	1346
Pressure (psia)	265	53.7	53.7	53.7	53.7	43.7	36.4	26.4	14.5	14.5
Vapor Frac	1.00	1.00	0.99	0.99	0.99	1.00	1.00	0.99	1.00	1.00
Liquid Frac	0.00	0.00	0.01	0.01	0.01	0.00	0.00	0.01	0.00	0.00
Enthalpy (Btu/lb)	-1.21E+3	-1.25E+3	-1.32E+3	-4.65E+2	-1.26E+3	-4.55E+2	-9.35E+1	-9.03E+2	-1.00E-1	-7.74E+2
Enthalpy (Btu/hr)	-2.26E+8	-2.32E+8	-4.79E+8	-9.50E+6	-7.20E+8	-2.59E+8	-5.33E+7	-5.14E+8	-2.74E+4	-2.20E+8

Table 9. Stream Report of Streams S-101 to S-110 showing the mass flow of each material, the temperature, pressure, vapor/liquid fraction and enthalpy.

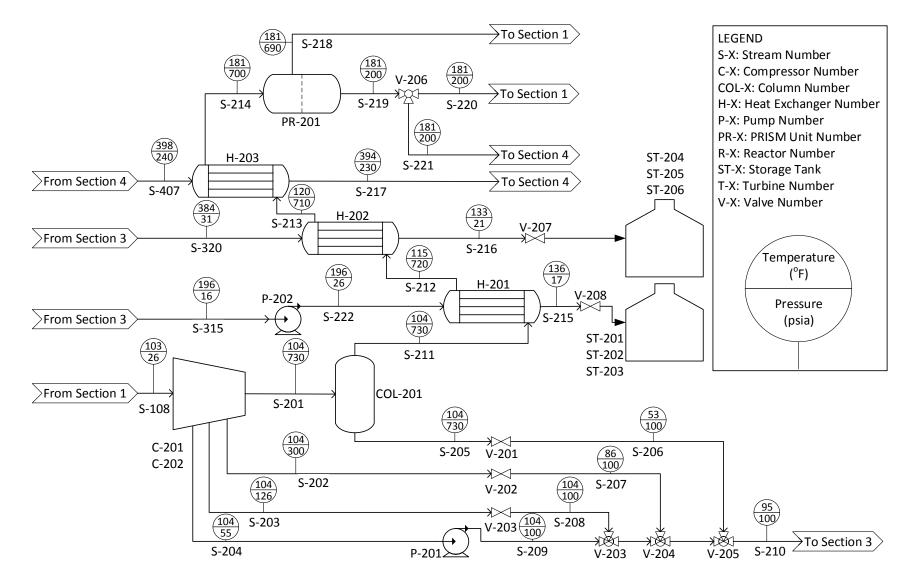


Figure 8. Section 2: Separation of Hydrogen and Light Hydrocarbons

	Mass Flow (lb/hr)									
Species	S-201	S-202	S-203	S-204	S-205	S-206	S-207	S-208	S-209	S-210
Hydrogen	18388	Trace	Trace	Trace	Trace	Trace	Trace	Trace	Trace	Trace
Methane	161167	783	391	243	1301	1301	783	391	243	2717
Ethane	205548	2131	1135	723	3049	3049	2131	1135	723	7038
Ethylene	21752	182	95.5	60.5	270	270	182	95.5	60.5	608
Propane	8615	217	119	76.5	298	298	217	119	76.5	711
Propylene	2695	40.9	22.2	15.7	64.2	64.2	40.9	22.2	15.7	143
Benzene	20942	20551	23052	23738	11301	11301	20551	23052	23738	78642
Toluene	1654	4717	10593	18941	1257	1257	4717	10593	18941	35508
P-Xylene	35.4	300	1726	7117	31.6	31.6	300	1726	7117	9175
1,3,5-Trimethyl										
Benzene	2.56	61.2	967	10708	2.45	2.45	61.2	967	10708	11739
Total Flow (lb/hr)	440798	28984	38100	61622	17574	17574	28984	38100	61622	146280
Total Flow (cuft/hr)	210921	617	755	1191	448	8454	4019	755	1191	9756
Temperature (°F)	104	104	104	104	104	53.2	86.3	104	104	96.0
Pressure (psia)	730	300	126	55.1	730	100	100	100	100	100
Vapor Frac	0.99	0.00	0.00	0.00	0.00	0.43	0.14	0.00	0.00	0.06
Liquid Frac	0.01	1.00	1.00	1.00	1.00	0.57	0.86	1.00	1.00	0.94
Enthalpy (Btu/lb)	-1.25E+3	5.26E+1	1.17E+2	5.66E+1	-2.04E+2	-2.04E+2	5.26E+1	1.17E+2	5.68E+1	4.03E+1
Enthalpy (Btu/hr)	-5.53E+8	1.53E+6	4.45E+6	3.49E+6	-3.59E+6	-3.59E+6	1.53E+6	4.45E+6	3.50E+6	5.89E+6

Table 10. Stream Report of Streams S-201 to S-210 showing the mass flow of each material, the temperature, pressure, vapor/liquid fraction and enthalpy.

	Mass Flow	v (lb/hr)								
Species	S-211	S-212	S-213	S-214	S-215	S-216	S-217	S-218	S-219	S-220
Hydrogen	18388	18388	18388	18388	0	0	0	2758	15630	2746
Methane	159866	159866	159866	159866	0	0	0	115104	44762	7864
Ethane	202499	202499	202499	202499	0	0	0	202499	0	0
Ethylene	21482	21482	21482	21482	0	0	0	21482	0	0
Propane	8316	8316	8316	8316	0	0	0	8316	0	0
Propylene	2630	2630	2630	2630	0	0	0	2630	0	0
Benzene	9641	9641	9641	9641	70269	Trace	0	9641	0	0
Toluene	397	397	397	397	34673	Trace	0	397	0	0
P-Xylene	3.73	3.73	3.73	3.73	9136	14.7	0	3.73	0	0
1,3,5-Trimethyl										
Benzene	Trace	Trace	Trace	Trace	40.5	11698	0	Trace	0	0
Water	0	0	0	0	0	0	22826	0	0	0
Total Flow (lb/hr)	423224	423224	423224	423224	114118	11713	22826	362832	60392	10610
Total Flow (cuft/hr)	210472	218944	224528	256757	2192	224	3033	147963	364193	63983
Temperature (°F)	104	115	120	181	136	133	394	181	181	181
Pressure (psia)	730	720	710	700	16.0	22.0	230	690	200	200
Vapor Frac	1.00	1.00	1.00	1.00	0.00	0.00	0.05	1.00	1.00	1.00
Liquid Frac	0.00	0.00	0.00	0.00	1.00	1.00	0.95	0.00	0.00	0.00
Enthalpy (Btu/lb)	-1.30E+3	-1.29E+3	-1.29E+3	-1.24E+3	2.01E+2	-2.03E+2	-6.43E+3	-1.23E+3	-1.35E+3	-1.35E+3
Enthalpy (Btu/hr)	-5.49E+8	-5.46E+8	-5.44E+8	-5.27E+8	2.29E+7	-2.37E+6	-1.47E+8	-4.47E+8	-8.13E+7	-1.43E+7

Table 11. Stream Report of Streams S-211 to S-220 showing the mass flow of each material, the temperature, pressure, vapor/liquid fraction and enthalpy.

	Mass Flow (lb/hr)			
Species	S-221	S-222		
Hydrogen	12884	0		
Methane	36898	0		
Benzene	0	70269		
Toluene	0	34673		
P-Xylene	0	9136		
1,3,5-Trimethyl				
Benzene	0	40.5		
Total Flow (lb/hr)	49782	114118		
Total Flow (cuft/hr)	300210	2287		
Temperature (°F)	181	196		
Pressure (psia)	200	26.0		
Vapor Frac	1.00	0.00		
Liquid Frac	0.00	1.00		
Enthalpy (Btu/lb)	-1.35E+3	2.29E+2		
Enthalpy (Btu/hr)	-6.70E+7	2.62E+7		

Table 12. Stream Report of Streams S-221 to S-222 showing the mass flow of each material, the temperature, pressure, vapor/liquid fraction and enthalpy.

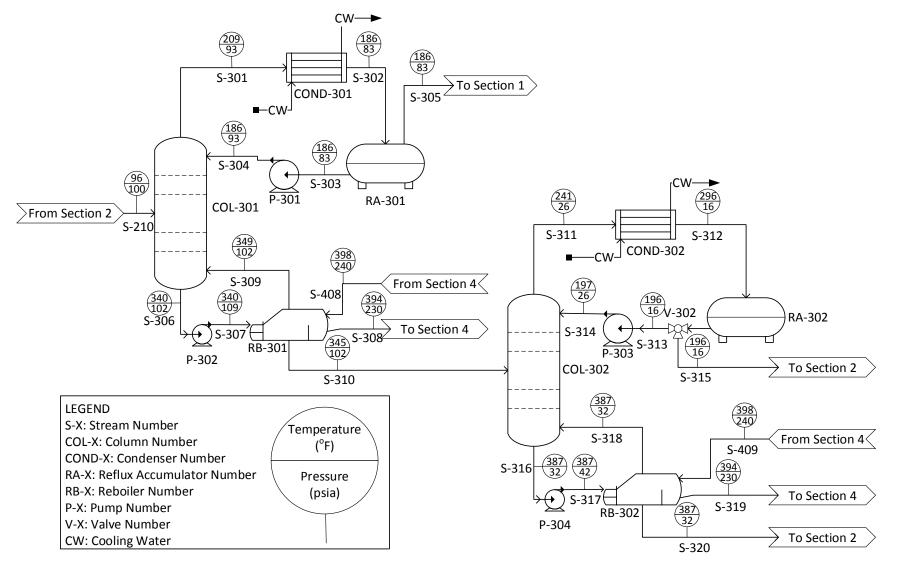


Figure 9. Section 3: Separation of BTX Products and Heavier Hydrocarbons

	Mass Flow	v (lb/hr)								
Species	S-301	S-302	S-303	S-304	S-305	S-306	S-307	S-308	S-309	S-310
Hydrogen	Trace	Trace	Trace	Trace	Trace	Trace	Trace	0	Trace	Trace
Methane	2723	2723	5.39	5.39	2717	Trace	Trace	0	Trace	Trace
Ethane	7089	7089	50.7	50.7	7038	Trace	Trace	0	Trace	Trace
Ethylene	612	612	3.61	3.61	608	Trace	Trace	0	Trace	Trace
Propane	720	720	9.08	9.08	711	Trace	Trace	0	Trace	Trace
Propylene	144	144	1.07	1.07	143	Trace	Trace	0	Trace	Trace
Benzene	10037	10037	1664	1664	8373	174298	174298	0	104030	70269
Toluene	1186	1186	351	351	835	61609	61609	0	26936	34673
P-Xylene	42.8	42.8	19.1	19.1	23.7	13244	13244	0	4094	9151
1,3,5-Trimethyl										
Benzene	Trace	Trace	Trace	Trace	Trace	14893	14893	0	3154	11739
Water	0	0	0	0	0	0	0	25705	0	0
Total Flow (lb/hr)	22553	22553	2104	2104	20449	264044	264044	25705	138214	125831
Total Flow (cuft/hr)	43772	36943	43.2	43.2	45287	6005	6006	3187	3177	2883
Temperature (°F)	209	186	186	186	186	340	340	394	349	349
Pressure (psia)	93.0	83.0	83.0	93.0	83.0	102	109	230	102	102
Vapor Frac	1.00	0.90	0.00	0.00	1.00	0.00	0.00	0.05	0.00	0.00
Liquid Frac	0.00	0.10	1.00	1.00	0.00	1.00	1.00	0.95	1.00	1.00
Enthalpy (Btu/lb)	-3.65E+2	-4.07E+2	2.30E+2	2.30E+2	-4.53E+2	3.03E+2	3.03E+2	-6.43E+3	3.40E+2	2.72E+2
Enthalpy (Btu/hr)	-8.23E+6	-9.18E+6	4.83E+5	4.83E+5	-9.27E+6	7.99E+7	7.99E+7	-1.65E+8	4.70E+7	3.42E+7

Table 13. Stream Report of Streams S-301 to S-310 showing the mass flow of each material, the temperature, pressure, vapor/liquid fraction and enthalpy.

	Mass Flow	v (lb/hr)								
Species	S-311	S-312	S-313	S-314	S-315	S-316	S-317	S-318	S-319	S-320
Benzene	115357	115357	70269	70269	70269	Trace	Trace	Trace	0	Trace
Toluene	56920	56920	34673	34673	34673	Trace	Trace	Trace	0	Trace
P-Xylene	14998	14998	9136	9136	9136	404	404	389	0	14.7
1,3,5-Trimethyl										
Benzene	66.5	66.5	40.5	40.5	40.5	194050	194050	182351	0	11698
Total Flow (lb/hr)	187342	187342	114118	114118	114118	194453	194453	182741	31936	11713
Total Flow (cuft/hr)	620515	3755	2287	2287	2287	4449	4449	406501	4501	268
Temperature (°F)	241	196	196	197	196	387	387	387	394	387
Pressure (psia)	26.0	16.0	16.0	45.0	16.0	31.9	41.9	32.0	230	32.0
Vapor Frac	1.00	0.00	0.00	0.00	0.00	0.00	0.00	1.00	0.06	0.00
Liquid Frac	0.00	1.00	1.00	1.00	1.00	1.00	1.00	0.00	0.94	1.00
Enthalpy (Btu/lb)	4.06E+2	2.29E+2	2.29E+2	2.29E+2	2.29E+2	-7.64E+1	-7.64E+1	5.56E+1	-6.43E+3	-7.63E+1
Enthalpy (Btu/hr)	7.60E+7	4.30E+7	2.62E+7	2.62E+7	2.62E+7	-1.49E+7	-1.49E+7	1.02E+7	-2.05E+8	-8.94E+5

Table 14. Stream Report of Streams S-311 to S-320 showing the mass flow of each material, the temperature, pressure, vapor/liquid fraction and enthalpy.

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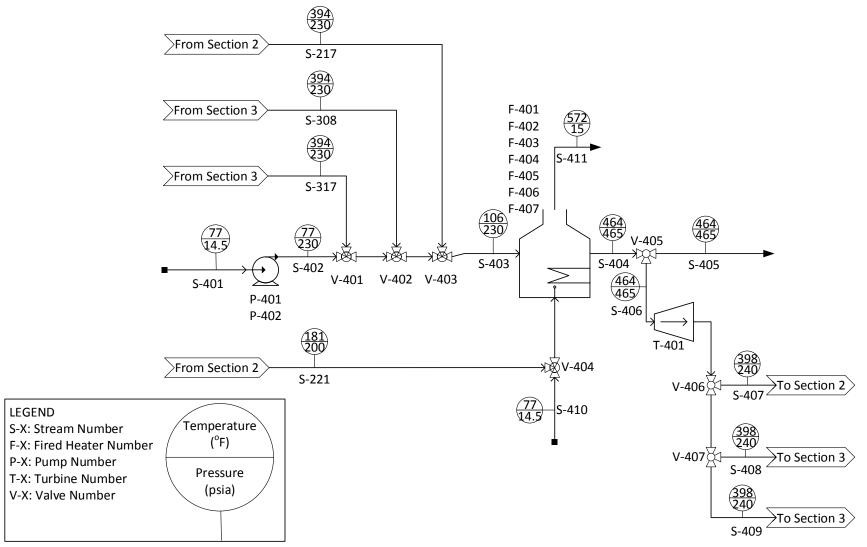


Figure 10. Section 4: Steam Generation

	Mass Flow	v (lb/hr)								
Species	S-401	S-402	S-403	S-404	S-405	S-406	S-407	S-408	S-409	S-410
Water	1015890	1015890	1096360	1096360	1015890	80467	22826	25705	31936	0
Oxygen	0	0	0	0	0	0	0	0	0	299340
Nitrogen	0	0	0	0	0	0	0	0	0	985838
Total Flow (lb/hr)	1015890	1015890	1096360	1096360	1015890	80467	22826	25705	31936	1285180
Total Flow (cuft/hr)	16372	16374	17955	1165440	1079910	85537	43974	49520	61524	17680500
Temperature (°F)	77.0	77.2	106	465	465	465	398	398	398	77.0
Pressure (psia)	14.5	230	230	465	465	465	240	240	240	14.5
Vapor Frac	0.00	0.00	0.00	1.00	1.00	1.00	0.97	0.97	0.97	1.00
Liquid Frac	1.00	1.00	1.00	0.00	0.00	0.00	0.03	0.03	0.03	0.00
Enthalpy (Btu/lb)	-6.82E+3	-6.82E+3	-6.79E+3	-5.62E+3	-5.62E+3	-5.62E+3	-5.67E+3	-5.67E+3	-5.67E+3	-1.00E-1
Enthalpy (Btu/hr)	-6.93E+9	-6.93E+9	-7.45E+9	-6.17E+9	-5.71E+9	-4.52E+8	-1.29E+8	-1.46E+8	-1.81E+8	-1.29E+5

Table 15. Stream Report of Streams S-401 to S-410 showing the mass flow of each material, the temperature, pressure, vapour/liquid fraction and enthalpy.

Table 16. Stream Report of Stream S-411 showing the mass flow of each material, the temperature, pressure, vapour/liquid fraction and enthalpy.

	Mass Flow (lb/hr)
Species	S-411
Water	198009
Oxygen	49890
Nitrogen	985838
Carbon Dioxide	101223
Total Flow (lb/hr)	1334960
Total Flow (cuft/hr)	38194900
Temperature (°F)	572
Pressure (psia)	14.5
Vapor Frac	1.00
Liquid Frac	0.00
Enthalpy (Btu/lb)	-1.01E+3
Enthalpy (Btu/hr)	-1.35E+9

Process Description

Process Description

Section 1 - Aromatization of Light Hydrocarbons and Regeneration of Catalyst

Light hydrocarbons, such as ethane, ethylene, propane and propylene, in this section are reacted to form benzene, toluene and xylene.

Pretreatment

Firstly, stream S-101, which contains fractional-grade ethane with a mole composition of 1% methane, 97% ethane and 2% propane¹⁸, enters the process through available pipelines along the Gulf Coast at a temperature of 88°F and a pressure of 265 psia. This feed stream will mix with S-218, the retentate of the PRISM system from Section 2, and S-305, the distillate of the Distillation Column COL-301 from Section 3. These streams contain mostly C2 and C3. The streams must be expanded to a suitable pressure to undergo aromatization. Each stream passes through a turbine that will reduce its pressure to 54 psia and produce some electricity. The streams mix and then are fed to the pretreatment section of the reactor, a heat exchanger that transfers heat between the inlet of the reactor and the outlet, H-101. The stream is preheated from -17°F to 1148°F.

Aromatization

The preheated stream, S-106, then enters the furnace where it is heated to 1170°F and then passes into one of the three reactors, R-101, R-102, and R-103. These reactors are steam-reforming type reactors, provided by the Jacobs Engineering Group, in which tubes packed with catalysts are enclosed in a furnace, are maintained at a temperature of 1170°F by using fired heaters on the outsides of the tubes. This temperature is maintained by continuously burning hydrogen and methane recovered from the permeate of the PRISM separation unit in Section 2 with 20% excess air. The feed entered the reactor at a pressure of 43.7 psia and left at a pressure of 36.4 psia; these pressures were selected to permit flow through the packed tubes. It was assumed in this analysis that the conversions and selectivities given in the patents would remain the same at these higher pressures. The reactor effluent, S-107 contains a wide array of products, including methane, C2 and C3 alkanes and alkenes, benzene, toluene, xylene, and heavier hydrocarbons.

The selectivity and conversion of ethane was provided by US Patent 209,795² and the selectivity was assumed to be the same as ethane. However, since the conversion of ethane to ethylene was the ratelimiting step, it was assumed that all the ethylene entering the reactor would be consumed. The specific catalyst chosen and the corresponding conversion and selectivity are shown under the 'Catalyst Choice' subsection under Preliminary Process Synthesis. The selectivity and conversion of propane and propylene were taken from US Patent 324,778³. This patent did not give a specific breakdown for the selectivity of benzene, toluene and xylene so a split of 35% benzene, 45% toluene, and 20% xylene was used.¹⁹

Posttreatment

The reactor effluent will move to Section 200 to separate the light components but its temperature must be reduced in order to prevent damage to the multistage compressors, C-201 and C-202. The stream entering the multistage compressors must be less than 110°F. The reactor effluent, S-107, passes through the tube side of the shell and tube heat exchanger, H-101, where it is cooled to a safe temperature of 103°F.

Regeneration of Catalyst

Coke builds up on the catalyst through its operation. This coke reduces the conversion of the reactant and the catalyst must be regenerated to maintain continuous and consistent production of BTX.

The procedure to regenerate the platinum-germanium zeolite catalyst has been provided in Patent US20080154079 A1.⁶ There are five stages to the regeneration process. The first stage and second stage remove coke and sulfur from the catalyst via oxidation. The third stage redisperses platinum on the surface of the catalyst by using chlorine gas. The fourth stage removes chlorine and binds platinum to the surface of the zeolite by steaming. The fifth stage reduces the catalyst in hydrogen. Chlorine, water vapor, oxygen, hydrogen and the nitrogen are fed in different ratios to accomplish each step. The conditions of the reactor and the feed are shown in Table 17. Valves V-108, V-109, V-111, and V-113, control the quantity of reagent fed to the reactor that requires catalyst regeneration. Chlorine is stored on-site in pressure vessel, PV-101; water is vaporized in the column after being pumped by pump P-101; and all other gases are supplied by local pipelines.

Stage	Time Duration (hr)	Temp (°F)	Mass Flow (lb/hr)				
			Chlorine	Water Vapor	Oxygen	Hydrogen	Nitrogen
1	24	1022	797	1613	3594	0	69300
2	24	1022	335	678	3594	0	27300
3	48	1022	3140	1610	8570	0	64000
4	12	1022	0	1610	8570	0	65300
5	24	1022	0	0	0	4710	9370

Table 17. Time duration, temperature and mass flow rate of each reagent used in each stage of the regeneration process.

One regeneration cycle takes five and a half days to complete and so a Gantt chart was created to determine the minimum number of reactors necessary for continuous operation. The Gantt chart is shown in Figure 11. From this chart, it was determined that three reactors would be necessary for continuous operation.

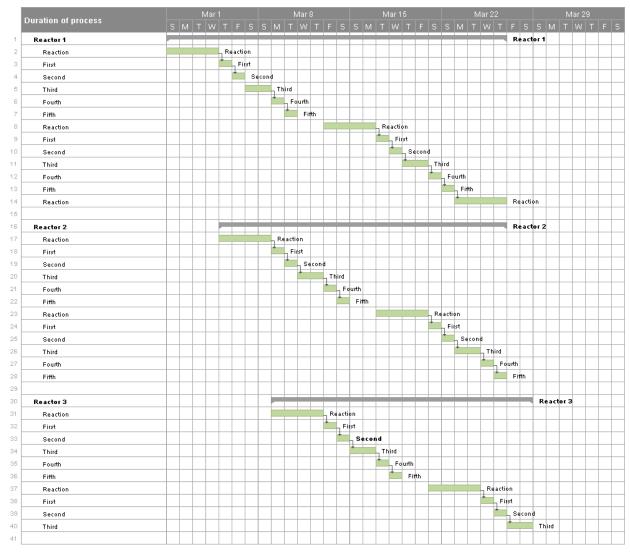


Figure 11. Gantt chart for the regeneration process. 'Reaction' represents the period in which the reactor is in operation. "First", "Second", "Third", "Fourth", and "Fifth" represent the different stages of the regeneration process.²⁰

Section 2 - Separation of Hydrogen and Light Hydrocarbons

This section separates the reaction products from Section 1 in stream S-108 into stream S-219, comprised of hydrogen and methane, that will be used to satisfy heating requirements around the plant; stream S-218, which is mainly C2 and C3 hydrocarbons that will be recycled to the reactor; and stream S-

210, which has a mix of all components except hydrogen and methane that will require further separation in Section 3.

Compression

In order to utilize the PRISM Separation system, PR-201, the feed must be at a pressure of 700 psia. Stream S-108 first enters two parallel multistage compressors, C-201 and C-202. Each compressor is comprised of four stages. Each compression stage's output pressure is 2.47 times the input pressure. Between each stage, the contents are cooled to 104 °F using cooling water. As a result, each stage becomes saturated and liquid is recovered as streams S-202, S-203 and S-204. The output stream, S-201, is also saturated and so must be fed to a flash vessel, COL-201, to separate the liquid phase from the vapor phase. The liquid streams leaving the multistage compressors and the liquid stream leaving the flash vessel (S-202, S-203, S-204 and S-205) will be fed to a distillation column that is operating at a pressure of 100 psia. Before mixing, all the streams must be at the same pressure. Stream S-204 is lower than the required pressure so it is fed through a pump, P-201. On the other hand, the other streams are higher than the required pressure so they are fed through valves to expand the stream to the desired pressure. The streams are then mixed to form Stream S-210 and sent to stage 3.

Pretreatment

Before feeding the light stream S-211, to the PRISM Unit PR-201, the temperature must be increased to 181°F. This was accomplished through a network of heat exchangers. The vapor stream of COL-201, stream S-211, is fed into the shell side of the heat exchanger H-201. The hot stream entering H-201, Stream S-315, is the distillate of the distillation column, COL-302. The hot stream cools to 136°F and is sent to the three one-day storage tanks, ST-201, ST-202 and ST-203. The cold stream, S-212, exits at 115°F and goes into the next heat exchanger, H-202. This heat exchanger uses the bottoms product of the distillation column, COL-302, in the tube side at a temperature of 384°F. The hot stream exits at 113°F and is then fed to three one-day storage tanks, ST-204, ST-205, and ST-206. The vapor stream exits the shell side of H-202 at 120°F to be fed into the final shell and tube heat exchanger, H-203. The steam condenses and leaves in Stream S-217 where the water is recycled to the steam boiler. The light hydrocarbon stream, S-214, is now at a temperature of 181°F and a pressure of 700 psia, which are suitable conditions for the PRISM system, PR-201.

PRISM Unit

The PRISM Unit is responsible for separating hydrogen from the vapor stream. It accomplishes this using membrane technology. At the conditions of Stream S-214, the permeate of the PRISM system, S-219, will

recover 85% of the hydrogen in Stream S-214 and will have a purity of 72% hydrogen with the balance being methane. The hydrogen recovered is at a pressure of 200 psia. The permeate is split by the valve V-206 and the gas is sent to each furnace to provide the required heat to operate the unit.

Section 3 - Separation of BTX Product and Heavier Hydrocarbons

The primary purpose of this section is to separate the liquid feed, S-210, into light components that will be recycled to the reactor; the BTX product; and heavier hydrocarbons that will be sold as a gasoline blend.

COL-301 is a 65 stage separation column that treats propane as the light key and benzene as the heavy key. First the stream S-210 enters COL-301 above stage 22. Each stage was assumed to have a Murphree tray efficiency of 65%. The vapor stream, S-301 exits the column at a temperature of 209°F and a pressure of 93 psia. It then passes through a partial condenser COND-301 that uses cooling water to condense the stream. The saturated mixture then enters a reflux accumulator, RA-301. The mass reflux ratio is set to 0.103 and the mass distillate to feed ratio is set to 0.140. The distillate leaves the reflux accumulator as stream S-305 to be recycled to the reactors while the liquid is pumped by pump P-301 back into the column. Liquid leaves the bottom of the column at a temperature of 340°F and a pressure of 102 psia as stream S-306. The liquid is pumped by pump P-302 and is fed to a kettle reboiler, RB-301. The boil-up ratio of the reboiler is 1.08 and the bottoms to feed ratio is 0.858. The contents of the partial reboiler are vaporized by steam obtained from S-408. The steam condenses and returns to the steam boilers, F-401 through F-407, in Stream S-308. The vapor stream S-309 is recycled back to the column and the liquid is taken from S-310.

The bottoms product of the first column then enters the second distillation column, COL-302. This column has 42 stages and it accepts the feed above tray 23. This distillation column was had a Murphree tray efficiency of 65%. Stream S-311, the vapor recovered from the top of the column, is at a temperature of 241°F and a pressure of 26 psia. Stream S-311 passes through a total condenser where it is cooled to a temperature of 200°F and the liquid is fed to a reflux accumulator, RA-302. The mass reflux ratio of the condenser is 0.642 and the distillate to feed ratio is set to 0.907. The distillate is recovered as S-315 and is sent to the heat exchanger S-202 and the reflux is pumped by pump P-303 and returned to the column. The liquid from the bottom of the column leaves as a temperature of 387°F and a pressure of 32 psia. The liquid is pumped by P-304 to a reboiler, RB-302. Steam from the steam boilers, F-401 through F-407, satisfies the heat requirements of the reboiler. After it condenses, it returns to the steam boilers to be evaporated again. The reboiler operates at a mass boil up ratio of 15.9 and a bottoms to feed ratio of 0.093. The vapor returns to the column as stream S-318 and the gasoline-blend product goes to heat exchanger H-202 to be cooled.

Process Description

Section 4 - Steam Generation

This section of the process satisfies all the heating requirements of the plant aside from those of the reactor. Water at ambient conditions is pumped to the pressure of the recycled stream using pump P-401. Steam from heat exchanger H-203 and reboilers RB-301 and RB-302 is mixed with the liquid water and fed to seven steam boilers, F-401 through F-407. The furnaces are heated by combusting hydrogen and methane from the permeate of the PRISM unit, PR-201. The hydrogen and methane are combusted with 20% excess oxygen and the combustion products leave the boilers at a temperature of 572°F. The heat generated produces high pressure steam, S-404. However, not all the steam is needed to satisfy the plant's utility requirements. Steam S-405 represents steam that is sold. The temperature of the steam is too high for the plant's needs and so it is passed through a turbine, T-401 to adjust the pressure and temperature to more suitable conditions of 398°F and 240 psia. Some electrical energy is recovered in the process that is used to drive the compressors, C-201 and C-202.

Energy Balance and Utility Requirements

The proposed utility requirements have been designed to use all recovered heat and electricity to reduce the total utility cost. Leftover requirements were fulfilled with either cooling water or steam.

Steam Integration

Stream S-401 is cooling water pressurized with P-401/P-402 and mixed with leftover steam from heat exchanger H-203, and reboilers RB-301 and RB-302. The mixed stream, S-403, is in the liquid phase with a temperature of 106°F and 230 psia. The fuel for the fired heaters is a mixture of S-221, containing methane and hydrogen, and S-410, containing oxygen and nitrogen for the combustion of S-221's gases. The heat produced from F-401 through F-407 is 1,280 MMBtu/hr and will be used to transfer heat to S-403 to produce steam. The outlet steam in S-404 is split at V-405 to sell approximately 1,015,660 lb/hr of the steam at 464°F and 465 psia. The leftover steam, containing 452 MMBtu/hr, is sent through turbine T-401 to generate electricity and then split again to fulfill utility requirements to RB-301, RB-302, and H-203.

Reactor Duty

The heat recovery section of the reactor, represented by heat exchanger H-101 and fired heater F-101, meets the utility requirements of reactors R-101, R-102, and R-103. H-101 will transfer 461 MMBtu/hr of heat to inlet stream S-105, allowing it to reach a temperature of 1148°F. F-101 will allow S-106 to reach the operating temperature of 1350°F and keep the reactors at an isothermal temperature of 1350°F. The heat duty required by F-101 is 205.9 MMBtu/hr.

Cooling Water

The process uses cooling water as the only external cooling source. Cooling water is used in condensers COND-301 and COND-302 to condense the vapors from the top of the columns. It is also used in the multistage condensers to cool the pressurized gas to 104°F so that the heavier components may be separated as a liquid. Table 18 summarizes the amount of energy removed from the hot stream by the cooling water.

Equipment	Source	Energy Requirement (Btu/hr)
COND-301	Cooling Water	548155
COND-302	Cooling Water	33077700
C-201, C-202 (Combined)	Cooling Water	181261000

Table 18. Energy requirements for condensers and compressors.

Electricity

This process uses several pieces of equipment that consume and generate electricity. Pumps and compressors are required throughout the plant to account for the pressure drop along pipes and through equipment. To mitigate the electricity costs for these units, turbines are installed when a vapor could be expanded. Table 19 shows a summary of the electricity required and supplied by each unit.

Table 19. Electricity requirement for pumps, turbines, and compressors. Note that a negative electricity requirement indicates that electricity is being generated.

Equipment	Number of	Electricity Requirement per unit	Total Electricity Requirement
-	Units	(hp)	(hp)
C-201	2	29985	59970
C-202			
P-101	1	0.0237	0.0237
P-201	1	4.86	4.86
P-202	1	2.08	2.08
P-301	1	0.0393	0.0393
P-302	1	4.85	4.85
P-303	1	6.03	6.03
P-304	1	4.04	4.04
P-401	1	321	321
T-101	3	-4090	-12270
T-102			
T-103			
T-104	1	-2564	-2564
T-105	1	-93.5	-93.5
T-401	1	-1470	-1470
Net Require	ement		43915

Heat Exchanger Network

This process consumes a tremendous amount of energy. A heat exchanger network was drawn to determine streams that could transfer energy in order to economically minimize external heating and cooling utilities. Figure 12 shows the process' heat exchanger network. Tables 19 through 30 show the design specifications of the heat exchangers and temperatures of the streams interacting with the entering and leaving the heat exchangers. Appropriate overall heat transfer coefficients, U, were obtained from Product and Process Design Principles¹⁴.

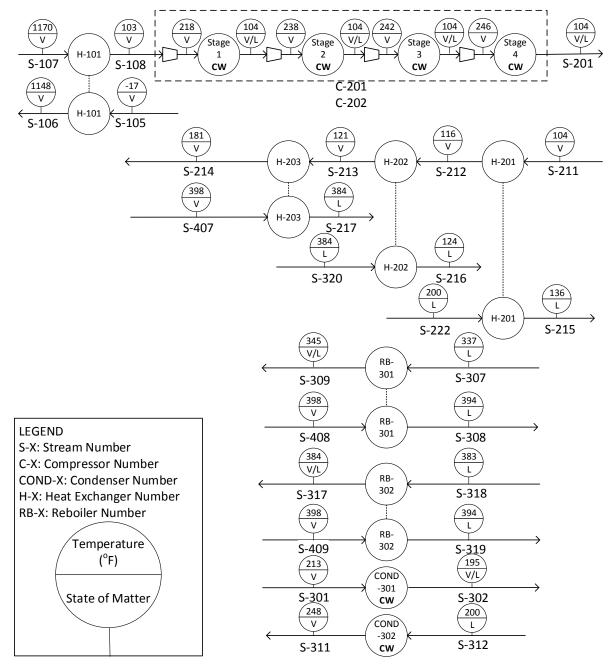


Figure 12. Heat Exchanger Network for the Process.

Unit		H-101						
	Stream	Temperature (°F)	U (Btu/hr-ft ² -°F)	Heat duty (MMBTU/hr)				
Hot in	S-107	1170	149.7	461.3				
Hot out	S-108	103						
Cold in	S-105	-17	Area (ft ²)	LMTD (°F)				
Cold out	S-106	1148	53340	57.8				

Tables 19-30. Heat Exchanger Network Tables.

Unit	C-101 Stage 1						
	Stream	Temperature (°F)	U (Btu/hr-ft ² - ^o F)	Heat duty (MMBTU/hr)			
Hot in	S-108	218	149.7	446.3			
Hot out	S-108	104					
Cold in	CW	90	Area (ft ²)	LMTD (°F)			
Cold out	CW	120	7166	43.2			

Unit		C-101 Stage 2						
	Stream	Temperature (°F)	U (Btu/hr-ft ² - ^o F)	Heat duty (MMBTU/hr)				
Hot in	S-108	238	149.7	46.32				
Hot out	S-108	104						
Cold in	CW	90	Area (ft ²)	LMTD (°F)				
Cold out	CW	120	6342	48.8				

Unit	C-101 Stage 3				
	Stream	Temperature (°F)	U (Btu/hr-ft ² - ^o F)	Heat duty (MMBTU/hr)	
Hot in	S-108	242	149.7	44.78	
Hot out	S-108	104			
Cold in	CW	90	Area (ft ²)	LMTD (°F)	
Cold out	CW	120	5996	49.9	

Unit		C-101 Stage 4				
	Stream	Temperature (°F)	U (Btu/hr-ft ² - ^o F)	Heat duty (MMBTU/hr)		
Hot in	S-108	246	149.7	44.07		
Hot out	S-108	140				
Cold in	CW	90	Area (ft ²)	LMTD (°F)		
Cold out	CW	120	3580	82.2		

Unit	H-201				
	Stream	Temperature (°F)	U (Btu/hr-ft ² - ^o F)	Heat duty (MMBTU/hr)	
Hot in	S-222	196	149.7	3.492	
Hot out	S-215	136			
Cold in	S-211	104	Area (ft ²)	LMTD (°F)	
Cold out	S-212	115	433.6	53.9	

Unit	H-202				
	Stream	Temperature (°F)	U (Btu/hr-ft ² -°F)	Heat duty (MMBTU/hr)	
Hot in	S-320	384	149.7	1.453	
Hot out	S-216	133			
Cold in	S-212	115	Area (ft ²)	LMTD (°F)	
Cold out	S-213	120	133.0	73.0	

Unit	H-203				
	Stream	Temperature (°F)	U (Btu/hr-ft ² - ^o F)	Heat duty (MMBTU/hr)	
Hot in	S-407	398	149.7	17.14	
Hot out	S-217	384			
Cold in	S-213	120	Area (ft ²)	LMTD (°F)	
Cold out	S-214	181	478.5	239	

Unit		RB-301				
	Stream	Temperature (°F)	U (Btu/hr-ft ² - ^o F)	Heat duty (MMBTU/hr)		
Hot in	S-408	398	149.7	19.51		
Hot out	S-308	394				
Cold in	S-307	340	Area (ft ²)	LMTD (°F)		
Cold out	S-309	349	2371	55.0		

Unit	RB-302				
	Stream	Temperature (°F)	U (Btu/hr-ft ² -°F)	Heat duty (MMBTU/hr)	
Hot in	S-409	398	149.7	24.69	
Hot out	S-319	394			
Cold in	S-317	387	Area (ft ²)	LMTD (°F)	
Cold out	S-318	387	13260	12.4	

Unit	COND-301				
	Stream	Temperature (°F)	U (Btu/hr-ft ² - ^o F)	Heat duty (MMBTU/hr)	
Hot in	S-301	209	149.7	0.507	
Hot out	S-302	186			
Cold in	CW	90	Area (ft ²)	LMTD (°F)	
Cold out	CW	120	150	98.9	

Unit	COND-302			
	Stream	Temperature (°F)	U (Btu/hr-ft ² - ^o F)	Heat duty (MMBTU/hr)
Hot in	S-311	241	149.7	33.19
Hot out	S-312	196		
Cold in	CW	90	Area (ft ²)	LMTD (°F)
Cold out	CW	120	1867	119

Equipment List and Unit Descriptions

Compressors

C-201 and C-202 are two cast iron/carbon-steel centrifugal compressors in parallel. These compressors raise the pressure of stream S-108, the cooled product stream leaving the reactor, from 26 psia to 730 psia. The compressors will run at an isentropic efficiency of 80% and will each require 29,990.65 hp of electricity. The exit streams from the compressor will experience a small temperature drop from 106° F to 104° F. The bare module costs for C-201 and C-202 are each \$18,221,846.

Condensers

COND-301 is a condenser for column COL-301. COND-301 is designed as a shell-and-tube, floating-head heat exchanger composed of carbon steel. The tube-side fluid is stream S-301, the vapor stream exiting the top of COL-301, and it is cooled from 209°F to 186°F at an outlet pressure of 83 psia. The shell-side fluid is cooling water that is supplied externally from the process. The tube length is 20 ft and has a surface area of 150 ft². The shell-side pressure is kept at 83 psig. COND-301 has an overall heat transfer coefficient of 250 Btu/hr-ft²-°F and has a heat duty of 0.548 MMBtu/hr. The bare-module cost of COND-301 is \$57,297.

COND-302 is a condenser for column COL-302. COND-302 is designed as a shell-and-tube, floating-head heat exchanger composed of carbon steel. The tube-side fluid is stream S-311, the vapor stream exiting COL-302, and it is cooled from 241° F to 196° F at an outlet pressure of 16 psia. The shell-side fluid is cooling water that is supplied externally from the process. The tube length is 20 ft and has a surface area of 4265 ft². The shell-side pressure is kept at 2.3 psig. COND-302 has an overall heat transfer coefficient of 250 Btu/hr-ft²- $^{\circ}$ F and has a heat duty of 33.1 MMBtu/hr. The bare-module cost of COND-302 is \$174,664.

Distillation Columns and Flash Vessels

COL-201 is a single sieve tray, carbon steel flash vessel that separates S-201 into vapor and liquid streams. S-201 is the stream exiting the multistage compressor. The vapor product will be sent to the PRISM separation unit, PR-201, to separate hydrogen and methane from the C2 and C3 hydrocarbons. The liquid product proceeds to Section 3 of the process for further separation. The column has a height of 15.5 ft, a diameter of 4.92 ft, a wall thickness of 1.93 inch, and a weight of 24,464 lb. It operates at 715.3 psig. The bare module cost of COL-301 is \$496,260.

COL-301 is a sieve tray, carbon steel distillation tower into which stream S-210 is fed. S-210 consists of the heavier streams that left the multistage compressors (C-201 and C-202) between each

compression stage as well as the liquid stream that left the flash vessel (COL-201). The bottoms product of COL-301 consists primarily of C6+ hydrocarbons that are sent to COL-302 for further separation. The distillate consists of lighter hydrocarbons that are recycled back to the reactor. There are 65 carbon steel trays, with the feed location at tray 22. The column is operated at a mass reflux ratio of 0.103. The column has a height of 112 ft, a diameter of 2.28 ft, a wall thickness of 1.13 in, and a weight of 38,899 lb. It operates at 87.8 psig. The bare module cost of COL-301 is \$946,960.

COL-302 is a sieve tray, carbon steel distillation tower into which the bottoms product of COL-301 is fed in order to separate the C9+ hydrocarbons from the BTX product. The distillate product and the bottoms product are fed through heat exchangers H-201 and H-202 respectively to reduce their temperature to appropriate levels for storage. COL-302 contains 42 carbon steel trays, with the feed location at tray 25. The column is operated at a mass reflux ratio of 0.642. The column has a height of 77 ft, a diameter of 9.14 ft, a wall thickness of 0.563 in, and a weight of 55,923 lb. It operates at 18 psig. The bare module cost of COL-302 is \$1,422,607.

Fired Heaters

F-101 represents the section of the heat recovery unit that will raise the temperature of the feed to the reactor to the appropriate reaction conditions after heat exchange with the reactor outlet occurs. The bare module cost and utility requirements of the entire heat recovery section are included in the cost and requirements for the reactor units R-101, R-102, and R-103.

F-401 through F-407 are steam boilers that will take S-403 and produce steam to fulfill heating requirements for multiple units in the process. S-403 comes from V-403 and will have a pressure increase of 230 psia to 465 psia and a temperature change of 106°F to 1350°F. The fired heater has a heat duty of 1,280 MMBtu/hr. The bare module cost of each unit is \$1,526,030.

Heat Exchangers

H-101 represents the section of the heat recovery unit that will exchange heat between the feed and the effluent streams of the reactor unit. The bare module cost and utility requirements of the entire heat recovery section are included in the cost and requirements for the reactor units R-101, R-102, and R-103.

H-201 is a floating-head shell-and- tube heat exchanger that will partially pre-heat stream S-211 to prepare for the PRISM separation unit, PR-201. S-211 is the vapor stream leaving the flash vessel (COL-201). This stream is on the shell side and will be heated from 104°F to 116°F with a pressure drop of 10 psia. The tube-side fluid is the distillate from COL-301, S-315, which is cooled from 200°F to 136°F with a pressure drop of 10 psia. The heat exchanger is composed of carbon steel for both the tube

and shell side. The tube length of H-201 is 20 ft with a surface area of 430.9 ft². The shell-side pressure will be kept at 705.3 psig. H-201 has an overall heat transfer coefficient of 149.7 Btu/hr-ft²- $^{\circ}$ F and will transfer 3.28 MMBtu/hr of heat. The bare-module cost of the heat exchanger is \$78,762. No external utilities will be required for H-201.

H-202 is a floating-head shell-and-tube heat exchanger that will continue to pre-heat S-211 to reach the PRISM unit's operating temperature. The shell-side fluid is the outlet stream, S-212, from heat exchanger H-201 that is heated from 116° F to 121° F with a pressure drop of 10 psia. The tube-side fluid is S-320, the bottoms product of COL-302, which is cooled from 384°F to 124°F. The heat exchanger is composed of carbon steel for both the tube and shell side. The tube length of H-202 is 20 ft with a surface area of 150 ft². The shell-side pressure will be kept at 695.3 psig. H-202 has an overall heat transfer coefficient of 149.7 Btu/hr-ft²-°F and will transfer 1.49 MMBtu/hr of heat. The bare-module cost of the heat exchanger is \$68,821. No external utilities will be required for H-202.

H-203 is a shell-and-tube heat exchanger that will raise S-213's temperature to the temperature of the PRISM unit. The shell-side fluid is the outlet shell-side stream from H-202, S-213, and is heated from 121°F to 181°F with a pressure drop of 10 psia. The tube-side fluid is S-407, which is a fraction of the steam from boilers F-401 through F-407. It is condensed with a temperature drop of 398° F to 384° F and has a pressure drop of 10 psia. The heat exchanger is composed of carbon steel for both the tube and shell side. The tube length of H-203 is 20 ft and has a surface area of 478.6 ft². The shell-side pressure will be kept at 685.3 psig. H-203 has an overall heat transfer coefficient of 149.7 Btu/hr-ft²-°F and will transfer 17.3 MMBtu/hr of heat. The bare-module cost of the heat exchanger is \$80,482. No external utilities will be required for H-203.

Pressure Vessel

PV-101 is a spherical pressure vessel that stores the chlorine gas used in the regeneration process. The vessel is modeled as a carbon steel storage tank due to its low vapor pressure, as suggested by Dr. Richard Bockrath. The vessel stores sufficient chlorine for one regeneration cycle, or 480,839 gallons of chlorine. The bare module cost of PV-101 is \$1,270,743.

PRISM Unit

PR-201 is a commercial membrane separation system that will separate methane and hydrogen from the lighter hydrocarbons products in stream S-214. The operating temperature and pressure of PR-201 are 181°F and 200 psia, respectively. At these conditions, the system will recover 85% of the hydrogen from the feed and will have a purity of 72% hydrogen, with the balance being methane. The cost of PR-201 is \$17,000,000, as suggested by Karin Fair from Air Products and Chemicals, Inc.

Pumps

P-201 is a cast-iron centrifugal pump that will raise the pressure of stream S-204, the first stream to leave the multistage compressors, from 55 psia to 100 psia. This pump is required as the streams coming from the compressors (C-201 and C-202) and the flash vessel (COL-201) must be at the same pressure when mixed to form the feed to COL-301. The pump has a volumetric flow rate of 148.5 gpm and generates a head of 125 ft. The electricity requirement for P-201 is 4.859 hp at an efficiency of 80%. The bare module cost of P-201 is \$12,308. An additional pump with the same specifications and cost will be purchased as a backup for the primary pump.

P-301 is a cast iron centrifugal pump that will raise the pressure of stream S-303 from 83 psia to 93 psia. This pump is required for raising the stream coming from reflux accumulator RA-301 to COL-301's operating pressure. The pump has a volumetric flow rate of 5.39 gpm and generates a head of 29.6 ft. The electricity requirement for P-301 is 0.0392 hp at an efficiency of 80%. The bare module cost of P-301 is \$11,053. An additional pump with the same specifications and cost will be purchased as a backup for the primary pump.

P-302 is a cast iron centrifugal pump that will raise the pressure of stream S-306, the bottoms product of COL-301, from 102 psia to 109 psia. This pump is required to raise S-306 to the reboiler RB-301's operating pressure. The pump has a volumetric flow rate of 749 gpm and generates a head of 21.9 ft. The electricity requirement for P-302 is 3.658 hp at an efficiency of 80%. The bare module cost of P-302 is \$14,602. An additional pump with the same specifications and cost will be purchased as a backup for the primary pump.

P-303 is a cast iron centrifugal pump that will raise the pressure of stream S-313, the reflux stream returning to COL-302 from RA-302, from 16 psia to 26 psia. This pump is required to raise S-313 to column COL-302's operating pressure. The pump has a volumetric flow rate of 285 gpm and generates a head of 83.7 ft. The electricity requirement for P-303 is 6.03 hp at an efficiency of 80%. The bare module cost of P-303 is \$11,692. An additional pump with the same specifications and cost will be purchased as a backup for the primary pump.

P-304 is a cast iron centrifugal pump that will raise the pressure of stream S-316, the bottoms product of COL-302, from 32 psia to 37 psia. This pump is responsible for raising S-316 to the reboiler RB-302's operating pressure. The pump has a volumetric flow rate of 555 gpm and generates a head of 32.9 ft. The electricity requirement for P-304 is 4.04 hp at an efficiency of 80%. The bare module cost of P-304 is \$14,731. An additional pump with the same specifications and cost will be purchased as a backup for the primary pump.

P-401 is a cast iron centrifugal pump. These pumps will raise the pressure of stream S-401 from 14.5 psia to 230 psia. This pump is required to allow the streams S-217, S-308, S-317 to be at the same

pressure when mixing occurs for fired heaters F-401 and F-402. The pump has a volumetric flow rate of 2040 gpm and generates a head of 500 ft. The electricity requirement for P-401 is 321 hp at an efficiency of 80%. The bare module cost of P-401 is \$49,893. An additional pump with the same specifications and cost will be purchased as a backup for each primary pump.

Reactors

R-101, R-102, and R-103 compose a group of three steam reforming reactors for the conversion of ethane to BTX, in which the feed is passed through tubes packed with catalyst. Direct fired heating is used on the outsides of the tubes to maintain a constant reaction temperature of 1170°F. The model will be purchased from Jacobs Engineering Group at a total installed cost of \$55 million per unit, including a heat recovery section in which the outlet stream is heat exchanged with the inlet stream. The dimensions of a single reactor, including the heat recovery section and all civil and structural works, are a length of 95 ft, width of 59 ft, and a height of 105 ft. Three reactors are used to ensure that at least one reactor will be operating at all times, while the other two are either regenerating or on standby. Each reactor will operate in production mode for 5 days, regeneration mode for 5.5 days, and standby mode for 1.5 days. Standby time is required to ensure that only one reactor is producing BTX at a time, preventing fluctuation in flow rate at the outlet of the reactor section.

The highly endothermic reaction converts ethane to BTX as well as smaller hydrocarbons (e.g., methane, propane, propylene, and ethylene), trimethylbenzene, and hydrogen. The C2 and C3 hydrocarbons are recycled to produce a higher yield of BTX. The catalyst used is Pt/Ge/ZSM-5 (604,223 lb). The inlet to the reactor, S-106, has a mass flow rate of 570,023 lb/hr at 1166° F and 43.7 psia. The outlet from the reactor, S-107, has a mass flow rate of 570,023 lb/hr at 1166° F and 36.4 psia. The adiabatic temperature drop across the reactor is 627°F and the heat duty is 103 MMBtu/hr.

Reboilers

RB-301 is a partial reboiler for column COL-301. This heat exchanger is designed as a shell-andtube kettle vaporizer composed of carbon steel for both sides. The tube-side fluid is stream S-408, which is a fraction of the steam from the steam boilers F-401 through F-407. It condenses with a temperature drop of 398°F to 394° F and an outlet pressure of 230 psia. The shell-side fluid is stream S-307, the liquid leaving the bottom of COL-301, which will be partially vaporized at an outlet pressure of 99 psia. The outlet is split into S-309, which is sent back to COL-301, and S-310, which is sent to H-202. The tube length is 20 ft and has a surface area of 1420 ft². The shell-side pressure is kept at 84.3 psig. RB-301 has an overall heat transfer coefficient of 250 Btu/hr-ft²-°F and has a heat duty of 19.5 MMBtu/hr. The baremodule cost of RB-301 is \$148,379. All utilities are provided internally by the system, so no external utilities are required for RB-301.

RB-302 is a partial reboiler for column COL-302. This heat exchanger is designed as a shell-andtube kettle vaporizer composed of carbon steel for both sides. The tube side is stream S-409, which is a fraction of the steam from the steam boilers F-401 through F-407. It condenses with a temperature drop of 398°F to 394° F and an outlet pressure of 230 psia. The shell-side fluid is stream S-317, the liquid leaving the bottom of COL-302, which will be partially vaporized at an outlet pressure of 31 psia. The outlet is split into S-318, which is sent back to COL-302, and S-320, which is sent to H-202. The tube length is 20 ft and has a surface area of 571.7 ft². The shell-side pressure is kept at 16.3 psig. RB-301 has an overall heat transfer coefficient of 250 Btu/hr-ft²-°F and has a heat duty of 1.78 MMBtu/hr. The baremodule cost of RB-302 is \$107,486. All utilities are provided internally by the system, so no external utilities are required for RB-302.

Reflux Accumulators

RA-301 is a reflux accumulator for column COL-301. This piece of equipment is a horizontal vessel composed of carbon steel with a diameter of 18.3 ft, a length of 36.6 ft, a wall thickness of 0.978 inch, and a weight of 118,143 lb. The operating conditions for RA-301 are 83 psi and 195° F. The bare module cost of RA-301 is \$663,053.

RA-302 is a reflux accumulator for column COL-302. This piece of equipment is a horizontal vessel composed of carbon steel with a diameter of 7.41 ft, a length of 14.8 ft, a wall thickness of 0.500 inch, and a weight of 9,907 lb. The operating conditions for RA-302 are 17 psi and 250 $^{\circ}$ F. The bare module cost of RA-302 is \$133,482.

Storage Tanks

ST-201 through ST-203 are storage tanks for the BTX product. The BTX product will enter the storage at 136°F and 17 psia. Stream S-215 is split between three tanks and the mixture contains 70,269 lb/hr of benzene, 34,673 lb/hr of toluene, and 9,099 lb/hr of xylene. The storage tank has a floating roof and is composed of carbon steel. Each tank will contain 590,366 gal of BTX product. The bare module cost of each storage tank is \$472,692.

ST-204 through ST-206 are storage tanks for the heavy hydrocarbon products. Stream S-216 goes into each storage tank and will enter at 136°F and 17 psia. Stream S-216 contains approximately 11,661 lb/hr of heavy hydrocarbons. The storage tank has a floating roof and is composed of carbon steel. Each tank will contain 40,429 gal of BTX product. The bare module cost of each storage tank is \$82,619.

Turbines

T-101, T-102, and T-103 are power-recovery turbines that take the outlet stream, S-218, from the PRISM separation unit and convert the pressure discharge to electricity, with an isentropic efficiency of 80%. The turbines cause a pressure drop of 690 psia to 54 psia and a temperature drop of 181° F to -27° F. Each turbine produces approximately 4089.8 hp. The bare module cost of each turbine is \$506,252.

T-104 is a power-recovery turbine that takes the ethane feedstock stream, S-101, and converts the pressure discharge to electricity, with an isentropic efficiency of 80%. The turbine causes a pressure drop of 265 psia to 54 psia and a temperature drop of 80° F to -37° F. These operating conditions produce approximately 2563.9 hp. The bare module cost of T-104 is \$346,820.

T-105 is a power-recovery turbine that takes the outlet stream, S-305, from RA-301, which is the distillate product of COL-301. This turbine converts the pressure discharge to electricity, with an isentropic efficiency of 80%. The turbine causes a pressure drop of 83 psia to 54 psia and a temperature drop of 186° F to 165° F. These operating conditions produce approximately 95.9 hp. The bare module cost of T-104 is \$24,222.

T-401 is a power-recovery turbine that takes the outlet stream, S-406, from V-405, which is a portion of the steam boilers' product stream. This turbine converts the pressure discharge to electricity with an isentropic efficiency of 80%. The turbine causes a pressure drop of 465 psia to 240 psia and a temperature drop of 464° F to 398° F. These operating conditions produce approximately 1466 hp. The bare module cost of T-401 is \$220,567.

Specification Sheets Compressors

compressors	COMPRESSOR					
Identification: C-201,	Identification: C-201, C-202 (Run in parallel)					
Function: Compresse	s the reactor outlet t	o 730 psia and co	ondenses heavi	er species.		
Operation: Continuou	IS					
	Mass Flow (lb/hr)	Mass Flow (lb/hr)				
	Feed	Exit 1	Exit 2	Exit 3	Exit 4	
Materials Handled	S-108	S-201	S-202	S-203	S-204	
Hydrogen	18388	18388	Trace	Trace	Trace	
Methane	162584	161167	783	391	243	
Ethane	209537	205548	2131	1135	723	
Ethylene	22090	21752	182	95.5	60.5	
Propane	9027	8615	217	119	76.5	
Propylene	2774	2695	40.9	22.2	15.7	
Benzene	88283	20942	20551	23052	23738	
Toluene	35905	1654	4717	10593	18941	
P-Xylene	9178	35.4	300	1726	7117	
1,3,5-						
Trimethylbenzene	11739	2.56	61.2	967	10708	
Total Flow (lb/hr)	569504	440798	28984	38100	61622	
Design Data						
Construction Material		Cast Iron/Carbon-Steel				
Drive		Electric Motor Drive				
Consumed Power (hp)		71442				
Purchase Cost (for a	ll units)		\$32,137,296.88			
Costing Information	Source	Product and Process Design Principles ¹⁴				

Condensers

CONDENSER				
Identification: COND-301				
Function: Partially condenses the vapor from the top of	of COL-301 from 213°F to 195°F			
Operation: Continuous				
Mass Flow (lb/hr)				
	Feed	Exit		
Materials Handled:	S-301	S-302		
Hydrogen	Trace	Trace		
Methane	2723	2723		
Ethane	7089	7089		
Ethylene	612	612		
Propane	720	720		
Propylene	144	144		
Benzene	10037	10037		
Toluene	1186	1186		
P-Xylene	42.8	42.8		
1,3,5-Trimethylbenzene	Trace	Trace		
Total Flow (lb/hr)	22553	22553		
Design Data	-			
Shell/Tube Material	Carbon Steel			
Tube Length (ft)	20			
Surface Area (ft ²)	150			
Purchase Cost	\$57,297.17			
Costing Information Source Product and Process Design Principles ¹⁴				

CONDENSER				
Identification: COND-302				
Function: Condenses the vapor from the top of COL-302 from	om 248°F to 200°F			
Operation: Continuous				
	Mass Flow (lb/h	r)		
	Feed	Exit		
Materials Handled:	S-311	S-312		
Benzene	115357	115357		
Toluene	56920	56920		
P-Xylene	14998	14998		
1,3,5-Trimethylbenzene	66.5	66.5		
Total Flow (lb/hr)	187342	187342		
Design Data				
Shell/Tube Material	Carbon Steel			
Tube Length (ft)	20			
Surface Area (ft ²)	4265			
Purchase Cost \$174,664.49				
Costing Information Source	Product and Pro Principles ¹⁴	ocess Design		

Distillation Columns and Flash Vessel

	FLASH VESSEL			
Identification: COL-201				
Function: Separates the saturated stream fro	om the multistage cor	npressors (C-201, C-	202) into vapor and	
liquid streams.				
Operation: Continuous				
	Mass Flow (lb/hr)			
	Feed	Exit 1	Exit 2	
Materials Handled:	S-201	S-211	S-205	
Hydrogen	18388	18388	Trace	
Methane	161167	159866	1301	
Ethane	205548	202499	3049	
Ethylene	21752	21482	270	
Propane	8615	8316	298	
Propylene	2695	2630	64.2	
Benzene	20942	9641	11301	
Toluene	1654	397	1257	
P-Xylene	35.4	3.73	31.6	
1,3,5-Trimethylbenzene	2.56	Trace	2.45	
Total Flow (lb/hr)	440798	423224	17574	
Design Data				
Height (ft)	15.5			
Inner Diameter (ft)	4.92			
Wall Thickness (inches)	1.93			
Design Pressure (psig)	715.3			
Vessel Material	Carbon Steel			
Tray/Packing	Tray			
Tray Type	Sieve			
Tray Material	Carbon Steel			
Number of Trays	1			
Purchase Cost \$496,260.10				
Costing Information Source	Product and Produc	ocess Design Princip	oles ¹⁴	

DISTILLATION COLUMN

Identification: COL-301

Function: Takes the liquid streams from compressors (C-201, C-202) and separates C1-C3 from BTX compounds.

Operation: Continuous

	Mass Flow (lb/hr)					
	Feed 1	Feed 2 (Reflux)	Feed 3 (Boil up)	Exit 1	Exit 2	
Materials Handled:	S-210	S-304	S-309	S-301	S-306	
Hydrogen	Trace	Trace	Trace	Trace	Trace	
Methane	2717	5.39	Trace	2723	Trace	
Ethane	7038	50.7	Trace	7089	Trace	
Ethylene	608	3.61	Trace	612	Trace	
Propane	711	9.08	Trace	720	Trace	
Propylene	143	1.07	Trace	144	Trace	
Benzene	78642	1664	104030	10037	174298	
Toluene	35508	351	26936	1186	61609	
P-Xylene	9175	19.1	4094	42.8	13244	
1,3,5-Trimethylbenzene	11739	Trace	3154	Trace	14893	
Total Flow (lb/hr)	146280	2104	138214	22553	264044	
Design Data						
Height (ft)	112					
Inner Diameter (ft)	2.28					
Wall Thickness (inches)	1.13					
Design Pressure (psig)	114					
Vessel Material	Carbon Stee	el				
Tray/Packing	Tray	Tray				
Tray Type	Sieve					
Tray Material	Carbon Steel					
Number of Trays	65					
Mass Reflux Ratio	0.103					
Mass Distillate to Feed Ratio	0.142					
Purchase Cost	\$835,061.3					
Costing Information Source	Product an	d Process Desi	ign Principles ¹⁴	4		

Identification: COL-302

DISTILLATION COLUMN

Function: Takes the bottoms product of COL-301 and separates BTX compounds from heavier hydrocarbons.

Operation: Continuous

Operation. Continuous						
	Mass Flow (lb/hr)					
		Feed	Feed	Exit 1	Exit 2	
	Feed	(Reflux)	(Boil up)	(Distillate)	(Bottoms)	
Materials Handled:	S-310	S-314	S-318	S-311	S-316	
Benzene	70269	70269	Trace	115357	Trace	
Toluene	34673	34673	Trace	56920	Trace	
P-Xylene	9151	9136	389	14998	404	
1,3,5-Trimethylbenzene	11739	40.5	182351	66.5	194050	
Total Flow (lb/hr)	125831	114118	182741	187342	194453	
Design Data						
Height (ft)	77					
Inner Diameter (ft)	9.14					
Wall Thickness (inches)	0.56					
Design Pressure (psig)	26					
Vessel Material	Carbon Stee	el				
Tray/Packing	Tray					
Tray Type	Sieve					
Tray Material	Carbon Stee	Carbon Steel				
Number of Trays	42					
Mass Reflux Ratio	0.642					
Mass Distillate to Feed Ratio	0.907					
Purchase Cost	\$1,422,606.78					
Costing Information Source	Costing Information Source Product and Process Design Principles ¹⁴					

Fired Heaters

FIRED HEATER								
Identification: F-401, F-402, F-403, F-404, F-405, F-406, F-407								
	Function: Burns hydrogen and methane from the PRISM Unit to boil water to form high pressure steam							
for heating requireme				0 1				
Operation: Continuous								
	Mass Flow (lb/h	r)						
	Combustion Side			Water Sid	e			
	Feed 1	Feed 2	Exit 1	Feed 3	Exit 2			
Materials Handled:	S-221	S-410	S-411	S-403	S-404			
Hydrogen	18388	0	0	0	0			
Methane	159866	0	0	0	0			
Water	0	0 0 198009 1096360 1096360						
Oxygen	0	0 299340 49890 0 0						
Nitrogen	0	0 985838 985838 0						
Carbon Dioxide	0	0 0 101223 0 0						
Total Flow (lb/hr)	423224 1285180 1334960 1096360 1096360							
Design Data								
Subtype	Steam Boiler							
Heat Duty (BTU/hr)	191217482							
Purchase Cost (for all units) \$10,682,210								
Costing Information Source Dr. Bockrath								

Heat Exchangers

HEAT EXCHANGER						
Identification: H-101						
Function: Exchange heat between the feed stream and the reactor outlet.						
Operation: Continuous						
	Mass Flow (lb/hr)					
	Feed 1 (Hot)	Feed 1 (Hot)Exit 1 (Hot)Feed 2 (Cold)Exit 2 (Cold)				
Materials Handled:	S-107 S-108 S-105 S-106					
Hydrogen	18388	18388	2758	2758		
Methane	162584	162584	118810	118810		
Ethane	209537	209537	389331	389331		
Ethylene	22090	22090	22090	22090		
Propane	9027	9027	14463	14463		
Propylene	2774	2774	2774	2774		
Benzene	88283	88283	18014	18014		
Toluene	35905	35905	1232	1232		
P-Xylene	9178	9178	27.5	27.5		
1,3,5-Trimethylbenzene	11739	11739	Trace	Trace		
Total Flow (lb/hr)	569504	569504	569500	569500		
Purchase Cost	Included in R-101's cost.					
Costing Information Source	Jacobs UK					

	HEAT	EXCHANGER					
Identification: H-201							
Function: Exchange heat betw	ween COL-201's vap	or product and COL	. 302's vapor produ	ct.			
Operation: Continuous	*	*	* *				
	Mass Flow (lb/hr	Mass Flow (lb/hr)					
	Feed 1 (Hot)	Feed 1 (Hot)Exit 1 (Hot)Feed 2 (Cold)Exit 2					
Materials Handled:	S-315	S-215	S-211	S-212			
Hydrogen	0	0	18388	18388			
Methane	0	0	159866	159866			
Ethane	0	0	202499	202499			
Ethylene	0	0	21482	21482			
Propane	0	0	8316	8316			
Propylene	0	0	2630	2630			
Benzene	70269	70269	9641	9641			
Toluene	34673	34673	397	397			
P-Xylene	9136	9136	3.73	3.73			
1,3,5-Trimethylbenzene	40.5	40.5	Trace	Trace			
Total Flow (lb/hr)	114118	114118	423224	423224			
Design Data							
Subtype	Floating Head, Shell and Tube						
Shell/Tube Material	Carbon Steel/Carbon Steel						
Tube Length	20 ft						
Surface Area (ft ²)	430.917						
Shell-Side Pressure (psig)	705.3						
Purchase Cost	\$78,762.43						
Costing Information Source	Product and Pro	cess Design Princip	les ¹⁴				

	HEAT	EXCHANGER			
Identification: H-202					
Function: Exchange heat betw	veen H-201's shell s	ide stream and COL	302's liquid produ	ct.	
Operation: Continuous			* *		
	Mass Flow (lb/hr	Mass Flow (lb/hr)			
	Feed 1 (Hot)	Exit 1 (Hot)	Feed 2 (Cold)	Exit 2 (Cold)	
Materials Handled:	S-320	S-216	S-212	S-213	
Hydrogen	0	0	18388	18388	
Methane	0	0	159866	159866	
Ethane	0	0	202499	202499	
Ethylene	0	0	21482	21482	
Propane	0	0	8316	8316	
Propylene	0	0	2630	2630	
Benzene	Trace	Trace	9641	9641	
Toluene	Trace	Trace	397	397	
P-Xylene	14.7	14.7	3.73	3.73	
1,3,5-Trimethylbenzene	11698	11698	Trace	Trace	
Total Flow (lb/hr)	11713	11713	423224	423224	
Design Data					
Subtype	Floating Head, Sh	Floating Head, Shell and Tube			
Shell/Tube Material	Carbon Steel/Carbon Steel				
Tube Length	20 ft	20 ft			
Surface Area (ft ²)	150				
Shell-Side Pressure (psig)	695.3	695.3			
Purchase Cost	\$68,821.01				
Costing Information Source	Product and Pro	cess Design Princip	les ¹⁴		

	HEAT EXCHANGER					
Identification: H-203						
Function: Exchange heat	between steam produc	ed from F-401 through	F-407 and H-20	02's shell side		
stream.						
Operation: Continuous						
	Mass Flow (lb/hr)					
	Feed 1 (Hot)	Exit 1 (Hot)	Feed 2 (Cold)	Exit 2 (Cold)		
Materials Handled:	S-407	S-217	S-213	S-214		
Hydrogen	0	0	18388	18388		
Methane	0	0	159866	159866		
Ethane	0	0	202499	202499		
Ethylene	0	0 0 21482 21482				
Propane	0	0 0 8316 8316				
Propylene	0	0 0 2630 2630				
Benzene	0 0 9641 9641					
Toluene	0 0 397 397					
P-Xylene	0	0	3.73	3.73		
1,3,5-Trimethylbenzene	0	0	Trace	Trace		
Water	22826	22826	0	0		
Total Flow (lb/hr)	22826	22826	423224	423224		
Design Data						
Subtype	Floating Head, Shell a	nd Tube				
Shell/Tube Material	Carbon Steel/Carbon Steel					
Tube Length	20 ft					
Surface Area (ft ²)	478.5673					
Shell-Side Pressure (psig)	685.3					
Purchase Cost	\$80,482.61					
Costing Information Source Product and Process Design Principles ¹⁴						

Pressure Vessel

PRESSURE VESSEL				
Identification: PV-101				
Function: Stores a regeneration cycle's worth of chlorine.				
Operation: Batch				
	Mass Flow (lb/hr)			
Exit				
Materials Handled:	S-113			
Chlorine	Varies based on regeneration stage.			
Total Flow lb/hr	Varies based on regeneration stage.			
Design Data				
Storage Capacity (gal)	480839			
Purchase Cost	\$1,270,734.45			
Costing Information Source	Product and Process Design Principles ¹⁴			

PRISM Unit

PRISM				
Identification: PR-201				
Function: Separates hydrogen and methane from its feed stream, S-214, using membrane separation.				
Operation: Continuous				
	Mass Flow (lb/hr)			
	Feed	Exit 1	Exit 2	
Materials Handled:	S-214	S-218	S-219	
Hydrogen	18388	2758	15630	
Methane	159866	115104	44762	
Ethane	202499	202499	0	
Ethylene	21482	21482	0	
Propane	8316	8316	0	
Propylene	2630	2630	0	
Benzene	9641	9641	0	
Toluene	397	397	0	
P-Xylene	3.73	3.73	0	
1,3,5-Trimethylbenzene	Trace	Trace	0	
Total Flow (lb/hr)	423224	362832	60392	
Purchase Cost	urchase Cost \$17,000,000.00			
Costing Information Source	Air Products an	nd Chemicals, Inc.		

Pumps

PUMP			
Identification: P-101			
Function: Pumps water to R-101, R-102, or R-103, depending on which reactor is being regenerated.			
Operation: Semi-continuous			
Mass Flow (lb/hr)			
	Feed	Exit	
Materials Handled:	S-111	S-112	
Water	Varies b	Varies based on regeneration stage.	
Total Flow lb/hr	Varies b	Varies based on regeneration stage.	
Design Data			
Subtype	Centrifugal		
Construction Material	Cast Iron	Cast Iron	
Flow Rate (gpm)	3.24		
Head (ft)	23.2		
Pump Type		SC, 40 - 500 ft, 50 - 900 gpm,	
Tump Type	75 hp		
Purchase Cost	\$15,500.06		
Costing Information Source	Product and	Process Design Principles ¹⁴	

	PUMP		
Identification: P-202			
Function: Pumps the distillate product from COL-302 to H-201.			
Operation: Continuous			
	Mass Flow (lb/hr)		
	Feed	Exit	
Materials Handled:	S-315	S-222	
Benzene	70269	70269	
Toluene	34673	34673	
P-Xylene	9136	9136	
1,3,5-Trimethylbenzene	40.5	40.5	
Total Flow (lb/hr)	114118	114118	
Design Data			
Subtype	Centrifugal		
Construction Material	Cast Iron		
Flow Rate (gpm)	289	289	
Head (ft)	28.9		
Pump Type	3600 RPM, VSC, 40 75 hp	3600 RPM, VSC, 40 - 500 ft, 50 - 900 gpm, 75 hp	
Purchase Cost	\$10,818.20		
Costing Information Source	Product and Proces	s Design Principles ¹⁴	

	PUMP		
Identification: P-301			
Function: Pumps the liquid reflux from the r	reflux accumulator RA-301 to the	distillation column, COL-	
301.			
Operation: Continuous			
	Mass Flow (lb/hr)		
	Feed	Exit	
Materials Handled:	S-303	S-304	
Hydrogen	Trace	Trace	
Methane	5.39	5.39	
Ethane	50.7	50.7	
Ethylene	3.61	3.61	
Propane	9.08	9.08	
Propylene	1.07	1.07	
Benzene	1664	1664	
Toluene	351	351	
P-Xylene	19.1	. 19.1	
1,3,5-Trimethylbenzene	Trace	e Trace	
Total Flow (lb/hr)	2104	2104	
Design Data	·		
Subtype	Centrifugal		
Construction Material	Cast Iron		
Flow Rate (gpm)	18.2		
Head (ft)	98.4	98.4	
Pump Type	3600 RPM, VSC, 4 75 hp	3600 RPM, VSC, 40 - 500 ft, 50 - 900 gpm, 75 hp	
Purchase Cost	\$11,324.71		
Costing Information Source	Product and Proce	ss Design Principles ¹⁴	

	PUMP		
Identification: P-302			
Function: Pumps the liquid from the bottom of	f the column COL-301 to the rebo	iler RB-301.	
Operation: Continuous			
	Mass Flow (lb/hr)		
	Feed	Exit	
Materials Handled:	S-306	S-307	
Benzene	174298	174298	
Toluene	61609	61609	
P-Xylene	13244	13244	
1,3,5-Trimethylbenzene	14893	14893	
Total Flow (lb/hr)	264044	264044	
Design Data			
Subtype	Centrifugal Pump		
Construction Material	Cast Iron		
Flow Rate (gpm)	5.57	5.57	
Head (ft)	29.3	29.3	
Pump Type	3600 RPM, VSC, 40 75 hp	3600 RPM, VSC, 40 - 500 ft, 50 - 900 gpm, 75 hp	
Purchase Cost	\$14,602.61		
Costing Information Source	Product and Proce	ss Design Principles ¹⁴	

	PUMP			
Identification: P-303				
Function: Pumps the liquid reflux from the 302.	reflux accumulator RA-302 to the d	listillation column, COL-		
Operation: Continuous				
	Mass Flow (lb/hr)			
	Feed	Exit		
Materials Handled:	S-313	S-314		
Benzene	70269	70269		
Toluene	34673	34673		
P-Xylene	9136	9136		
1,3,5-Trimethylbenzene	40.5	40.5		
Total Flow (lb/hr)	114118	114118		
Design Data	· · · · · · · · · · · · · · · · · · ·			
Subtype	Centrifugal Pump			
Construction Material	Cast Iron			
Flow Rate (gpm)	285	285		
Head (ft)	83.7			
Pump Type	3600 RPM, VSC, 40 75 hp	3600 RPM, VSC, 40 - 500 ft, 50 - 900 gpm, 75 hp		
Purchase Cost	\$13542.95	\$13542.95		
Costing Information Source	Product and Proces	s Design Principles ¹⁴		

	PUMP		
Identification: P-304			
Function: Pumps the liquid from the bottom of the column COL-302 to the reboiler RB-302.			
Operation: Continuous			
Mass Flow (lb/hr)			
	Feed	Exit	
Materials Handled:	S-316	S-317	
Benzene	Trace	Trace	
Toluene	Trace	Trace	
P-Xylene	404	404	
1,3,5-Trimethylbenzene	194050	194050	
Total Flow (lb/hr)	194453	194453	
Design Data			
Subtype	Centrifugal Pump		
Construction Material	Cast Iron	Cast Iron	
Flow Rate (gpm)	628	628	
Head (ft)	444	444	
Pump Type	3600 RPM, VSC, 4 75 hp	3600 RPM, VSC, 40 - 500 ft, 50 - 900 gpm, 75 hp	
Purchase Cost	\$22,006.77		
Costing Information Source	Product and Proce	ss Design Principles ¹⁴	

	PUMP	
Identification: P-401		
Function: Pumps water to the steam boiler.		
Operation: Continuous		
Mass Flow (lb/hr)		
	Feed	Exit
Materials Handled:	S-401	S-402
Water	1015890	1015890
Total Flow (lb/hr)	1015890	1015890
Design Data		
Subtype	Centrifugal Pump	
Construction Material	Cast Iron	
Flow Rate (gpm)	2040	
Head (ft)	500	
Pump Type	1800 RPM, HSC, 50 gpm, 250 Hp) - 500 ft, 250 - 5000
Purchase Cost	\$49,893.17	
Costing Information Source	Product and Proce	ss Design Principles ¹⁴

Reactor

REA	ACTOR	
Identification: R-101, R-102, R-103		
Function: Converts ethane feedstock to desired BTX	K product.	
Operation: Continuous		
Mass Flow (lb/hr)		
	Feed	Exit
Materials Handled:	S-106	S-107
Hydrogen	2758	18388
Methane	118810	162584
Ethane	389331	209537
Ethylene	22090	22090
Propane	14463	9027
Propylene	2774	2774
Benzene	18014	88283
Toluene	1232	35905
P-Xylene	27.5	9178
1,3,5-Trimethylbenzene	Trace	11739
Total Flow (lb/hr)	569500	569504
Purchase Cost (for all units)	\$165,000,000	
Costing Information Source	Jacobs UK Inc.	

Reboilers

REBOILER							
Identification: RB-301							
Function: Uses high pressure stea 301.	am to vaporize	e the liquid l	eaving the bottor	n of distillation	column COL-		
Operation: Continuous							
	Mass Flow	(lb/hr)					
	Hot Side		Cold Side				
	Feed 1	Exit 1	Feed 2	Exit 2	Exit 3		
Materials Handled:	S-408	S-308	S-307	S-309	S-310		
Benzene	0	0 0 174298 104030 702					
Toluene	0	0 0 61609 26936 34673					
P-Xylene	0	0	13244	4094	9151		
1,3,5-Trimethylbenzene	0	0	14893	3154	11739		
Water	25705	25705	0	0	0		
Total Flow (lb/hr)	25705 25705 264044 138214 125831						
Design Data							
Subtype	Kettle Vaporizer						
Shell/Tube Material	Carbon Steel/Carbon Steel						
Tube Length (ft)	20						
Surface Area (ft ²)	971						
Shell-Side Pressure (psig)	84.5						
Purchase Cost\$112,298.87							
Costing Information Source Product and Process Design Principles ¹⁴							

REBOILER						
Identification: RB-302						
Function: Uses high pressure steam to vaporize the liquid leaving the bottom of distillation column COL-						
302.						
Operation: Continuous						
	Mass Flow (lb/hr)					
	Hot Side		Cold Side			
	Feed 1	Exit 1	Feed 2	Exit 2	Exit 3	
Materials Handled:	S-409	S-319	S-317	S-318	S-320	
Benzene	0	0	Trace	Trace	Trace	
Toluene	0 0 Trace Trace Trace					
P-Xylene	0 0 404 389 14.					
1,3,5-Trimethylbenzene	0	0	194050	182351	11698	
Water	31936	31936	0	0	0	
Total Flow (lb/hr)	31936 31936 194453 182741 11713					
Design Data						
Subtype	Kettle Vapo	orizer				
Shell/Tube Material	Carbon Stee	Carbon Steel/Carbon Steel				
Tube Length (ft)	20					
Surface Area (ft ²)	488					
Shell-Side Pressure (psig)	44.5					
Purchase Cost	\$90,969.10					
Costing Information Source Product and Process Design Principles ¹⁴						

Reflux Accumulators

REFLUX ACC	CUMULATOR			
Identification: RA-301				
Function: Stores reflux from top of column to mainta	ain constant reflux and	l distillate flow r	ates.	
Operation: Continuous				
	Mass Flow (lb	Mass Flow (lb/hr)		
	Feed Exit 1 Exit 2			
Materials Handled:	S-302	S-303	S-305	
Hydrogen	Trace	Trace	Trace	
Methane	2723	5.39	2717	
Ethane	7089	50.7	7038	
Ethylene	612	3.61	608	
Propane	720	9.08	711	
Propylene	144	1.07	143	
Benzene	10037	1664	8373	
Toluene	1186	351	835	
P-Xylene	42.8	19.1	23.7	
1,3,5-Trimethylbenzene	Trace	Trace	Trace	
Total Flow (lb/hr)	22553	2104	20449	
Design Data				
Length (ft)	36.6			
Inner Diameter (ft)	18.3	18.3		
Wall Thickness (in)	0.625	0.625		
Operating Pressure (psig)	10	10		
Vessel Material	Carbon Steel			
Purchase Cost	\$418,852.67			
Costing Information Source Product and Process Design Principles ¹⁴				

REFLUX ACCUMULATOR Identification: RA-302 Function: Stores reflux from top of column to maintain constant reflux and distillate flow rates. **Operation: Continuous** Mass Flow (lb/hr) Exit 1 Feed Exit 2 S-313 S-312 S-315 Materials Handled: Benzene 115357 70269 70269 Toluene 56920 34673 34673 P-Xylene 14998 9136 9136 1,3,5-Trimethylbenzene 40.5 40.5 66.5 Total Flow (lb/hr) 187342 114118 114118 **Design Data** Length (ft) 14.8 Inner Diameter (ft) 7.41 Wall Thickness (in) 0.500 10 **Operating Pressure (psig)** Vessel Material Carbon Steel \$38,593.29 **Purchase Cost Product and Process Design Principles**¹⁴ **Costing Information Source**

Storage Tanks

STORAGE T	ANK			
Identification: ST-201, ST-202, ST-203				
Function: Stores the BTX product. Each tank has a capacity of one day.				
Operation: Batch				
	Mass Flow (lb/hr)			
	Feed			
Materials Handled:	S-215			
Benzene	70269			
Toluene	34673			
P-Xylene	9136			
1,3,5-Trimethylbenzene	40.5			
Total Flow (lb/hr)	114118			
Design Data				
Storage Capacity (gal)	590,367			
Purchase Cost	\$1,418,078.05			
Costing Information Source	Product and Process Design Principles ¹⁴			

ST	ORAGE TANK	
Identification: ST-204, ST-205, ST-206		
Function: Stores the gasoline-blend product. E	ach tank has a capacity of one day.	
Operation: Batch		
	Mass Flow (lb/hr)	
Feed		
Materials Handled:	S-216	
Benzene	Trace	
Toluene	Trace	
P-Xylene	14.7	
1,3,5-Trimethylbenzene	11698	
Total Flow (lb/hr)	11713	
Design Data		
Storage Capacity (gal)	40429	
Purchase Cost	\$59,246.38	
Costing Information Source	Product and Process Design Principles ¹⁴	

Turbines

	TURBINE		
Identification: T-101, T-102, T-103			
Function: To recover work from permeate co	oming out of PRISM		
Operation: Continuous			
	Mass Flow (lb/hr)		
	Feed	Exit	
Materials Handled:	S-218	S-103	
Hydrogen	2758	2758	
Methane	115104	115104	
Ethane	202499	202499	
Ethylene	21482	21482	
Propane	8316	8316	
Propylene	2630	2630	
Benzene	9641	9641	
Toluene	397	397	
P-Xylene	3.73	3.73	
1,3,5-Trimethylbenzene	Trace	Trace	
Total Flow (lb/hr)	362832	362832	
Design Data	-		
Power Extracted (hp)	4089.8		
Purchase Cost	\$446,431.21		
Costing Information Source	Product and Process De	sign Principles ¹⁴	

TU	IRBINE			
Identification: T-104				
Function: To recover work from feedstock ethane				
Operation: Continuous				
Mass Flow (lb/hr)				
Feed Exit				
Materials Handled:	S-101	S-102		
Methane	989	989		
Ethane	179794	179794		
Propane	5436	5436		
Total Flow (lb/hr)	186219	186219		
Design Data				
Power Extracted (hp) 2563.94				
Purchase Cost \$305,837.85				
Costing Information SourceProduct and Process Design Principles14				

	TURBINE	
Identification: T-105		
Function: To recover work from distillate f	from the first distillation column's reflu	x accumulator
Operation: Continuous		
	Mass Flow (lb/hr)	
	Feed	Exit
Materials Handled:	S-305	S-104
Methane	2717	2717
Ethane	7038	7038
Ethylene	608	608
Propane	711	711
Propylene	143	143
Benzene	8373	8373
Toluene	835	835
P-Xylene	23.7	23.7
1,3,5-Trimethylbenzene	Trace	Trace
Total Flow (lb/hr)	20449	20449
Design Data		
Power Extracted (hp)	95.91	
Purchase Cost	\$21,359.83	
Costing Information Source	Product and Process De	sign Principles ¹⁴

	TUF	RBINE				
Identification: T-401						
Function: Extracts energy from requirements of the reboilers, RE				for the heating		
Operation: Continuous						
	Mass Flow (lb/h	Mass Flow (lb/hr)				
	Feed	Feed Exit 1 Exit 2 Exit 3				
Materials Handled:	S-406	S-407	S-408	S-409		
Water	80467	22826	25705	31936		
Total Flow (lb/hr)	80467	22826	25705	31936		
Design Data						
Power Extracted (hp)	1691					
Purchase Cost	\$218,254.90					
Costing Information Source	Product and Pr	ocess Design Princ	ciples ¹⁴			

Equipment Cost Summary

Table 31. Cost and source of each unit of equipment. Pricing sources for equipment that was not priced with Product and Process Design Principles may be found in Appendix A-6.

ID	Description	Price Source	Cost
COND-301	Condenser	Product and Process Design Principles ¹⁴	\$64,974.99
COND-302	Condenser	Product and Process Design Principles ¹⁴	\$198,069.53
C-201	Multistage Compressor	Product and Process Design Principles ¹⁴	\$16,068,647.94
C-202	Multistage Compressor	Product and Process Design Principles ¹⁴	\$16,068,647.94
COL-201	Flash Drum	Product and Process Design Principles ¹⁴	\$496,260.10
COL-301	Distillation Column	Product and Process Design Principles ¹⁴	\$946,959.53
COL-302	Distillation Column	Product and Process Design Principles ¹⁴	\$1,422,606.78
F-401	Fired Heater	Consultant Provided	\$10,682,210.00
H-201	Heat Exchanger	Product and Process Design Principles ¹⁴	\$78,762.43
H-202	Heat Exchanger	Product and Process Design Principles ¹⁴	\$68,821.01
H-203	Heat Exchanger	Product and Process Design Principles ¹⁴	\$80,482.61
P-101	Centrifugal Pump	Product and Process Design Principles ¹⁴	\$15,500.06
P-201	Centrifugal Pump	Product and Process Design Principles ¹⁴	\$10,854.43
P-202	Centrifugal Pump	Product and Process Design Principles ¹⁴	\$10,818.20
P-301	Centrifugal Pump	Product and Process Design Principles ¹⁴	\$9,747.16
P-302	Centrifugal Pump	Product and Process Design Principles ¹⁴	\$12,877.08
P-303	Centrifugal Pump	Product and Process Design Principles ¹⁴	\$13,542.95
P-304	Centrifugal Pump	Product and Process Design Principles ¹⁴	\$12,990.48
P-401	Centrifugal Pump	Product and Process Design Principles ¹⁴	\$49,893.17
PR-201	PRISM	Air Products and Chemicals, Inc.	\$17,000,000.00
R-101	Steam Reformer Reactor	Jacobs UK	\$55,000,000.00

ID	Description	Price Source	Cost
R-102	Steam Reformer Reactor	Jacobs UK	\$55,000,000.00
R-103	Steam Reformer Reactor	Jacobs UK	\$55,000,000.00
RB-301	Reboiler	Product and Process Design Principles ¹⁴	\$148,379.15
RB-302	Reboiler	Product and Process Design Principles ¹⁴	\$107,486.22
RA-301	Reflux Accumulator	Product and Process Design Principles ¹⁴	\$663,053.24
RA-302	Reflux Accumulator	Product and Process Design Principles ¹⁴	\$133,482.62
T-101	Power-Recovery Turbine	Product and Process Design Principles ¹⁴	\$506,252.99
T-102	Power-Recovery Turbine	Product and Process Design Principles ¹⁴	\$506,252.99
T-103	Power-Recovery Turbine	Product and Process Design Principles ¹⁴	\$506,252.99
T-104	Power-Recovery Turbine	Product and Process Design Principles ¹⁴	\$346,820.12
T-105	Power-Recovery Turbine	Product and Process Design Principles ¹⁴	\$24,222.05
T-401	Power-Recovery Turbine	Product and Process Design Principles ¹⁴	\$220,567.45
ST-201	Storage Tank	Product and Process Design Principles ¹⁴	\$472,692.68
ST-202	Storage Tank	Product and Process Design Principles ¹⁴	\$472,692.68
ST-203	Storage Tank	Product and Process Design Principles ¹⁴	\$472,692.68
ST-204	Storage Tank	Product and Process Design Principles ¹⁴	\$82,619.28
ST-205	Storage Tank	Product and Process Design Principles ¹⁴	\$82,619.28
ST-206	Storage Tank	Product and Process Design Principles ¹⁴	\$82,619.28
PV-101	Pressure Vessel	Product and Process Design Principles ¹⁴	\$1,441,023.07

Fixed-Capital Investment Summary

The total fixed costs were calculated from the total cost of operations, maintenance, operating overhead, property insurance, and taxes. The total fixed cost was calculated as \$12.7MM. Table 32 summaries the fixed-capital investment breakdown.

Operations	Annual Cost		
Direct Wages and Benefits	\$3,328,000		
Direct Salaries and Benefits	\$499,200		
Operating Supplies and Services	\$199,680		
Technical Assistance to Manufacturing	\$2,400,000		
Control Laboratory	\$2,600,000		
Total Operations	\$9,026,880		
<u>Maintenance</u>	Annual Cost		
Wages and Benefits	\$10,616,126		
Salaries and Benefits	\$2,654,032		
Materials and Services	\$10,616,126		
Maintenance Overhead	\$530,806		
Total Maintenance	\$24,417,091		
Operating Overhead	Annual Cost		
General Plant Overhead:	\$1,213,912		
Mechanical Department Services:	\$410,337		
Employee Relations Department:	\$1,008,744		
Business Services:	\$1,265,204		
Total Operating Overhead	\$3,898,198		
Property Taxes and Insurance	Annual Cost		

Property Taxes and Insurance:	\$6,066,358
Total Fixed Costs	\$43,408,526

Guidelines provided in Chapter 23 of Product and Process Design Principles¹⁴ were used to arrive at these numbers. Eight operators will operate this plant per shift. This number was calculated by multiplying four operators (one for each process section) by a factor of two since this plant is a continuous-flow process that produces over 1,000 tons of products per day. Direct wages and benefits (DW&B) are assumed at \$40 per operator-hour. In addition, costs for direct salaries and benefits for supervisory and engineering personnel is set at 15% of DW&B and operating supplies and services is set at 6% of DW&B. Finally, \$60,000 per operator-shift-year is added for technical assistance and \$65,000 per operator-shift-year for the control laboratory. Wages and benefits for maintenance staff (MW&B) is set at 3.5% of total depreciable capital (TDC), since this process deals with fluids only. Costs of maintenance staff wages and benefits, materials and services, and maintenance overhead are estimated at 25%, 100%, and 5% of MW&B costs respectively. Furthermore, general plant overhead was set at 7.1% of maintenance and operations wages and benefits (MOW&B). Costs of mechanical department services, employee relations, and business services are set at 2.4%, 5.9%, and 7.4%, respectively. Finally, property taxes and property insurance costs were estimated at 2% of TDC. Since this process is located on a fairly large plant, this analysis did not take into account any additional costs such as rental fees and licensing fees.

Operating Cost - Cost of Manufacture

Table 33. Variable cost breakdown.				
Variable Costs at 100% Capacity:				
General Expenses				
Selling / Transfer Expenses:			\$11,198,277	
Direct Research:			\$17,917,244	
Allocated Research:			\$1,866,380	
Administrative Expense:			\$7,465,518	
Management Incentive Compensation:			\$4,665,949	
Total General Expenses			\$43,113,367	
Raw Materials	\$0.341	per lb of BTX	\$318,535,578	
Byproducts	\$0.113	per lb of BTX	(\$102,397,305)	
<u>Utilities</u>	\$0.030	per lb of BTX	\$27,762,245	
Total Variable Costs			\$287,013,885	

Profitability Analysis - Business Case

The return on investment is calculated as the ratio of net annual earnings, including income tax and a fiveyear MACRS depreciation schedule, and the total capital investment, adjusted for working capital effects. This process has an ROI of -0.94% based on the third production year.

Annual Sales	\$288,418,629
Annual Costs	\$(266,487,209)
Depreciation	\$(27,177,283)
Income Tax	\$1,940,969
Net Earnings	\$(3,304,894)
Total Capital Investment	\$351,308,449
ROI	-0.94%

Table 34. ROI Analysis (Third Production Year)

The Internal Rate of Return (IRR) for this project is 2.87% and the Net Present Value (NPV) for the project is -\$166,765,000 at a discount rate of 18%. These values were calculated using the "Profitability Analysis" Excel worksheet introduced as part of the Product and Process Design Principles¹⁴. The full spreadsheet is attached in Appendix A-3: Profitability Analysis Reports. Looking at these numbers, it is evident that the plant is currently an economically unfavorable business venture. The most important variables that contribute to profitability are primary input (ethane) costs, catalyst component costs (primarily platinum and zeolite), and catalyst lifespan. Other important factors that lead to this high internal rate of return have been addressed in the following sections. These factors include the price of nitrogen, price of BTX as the main product, and fixed and variable costs.

Overall Revenue and Cost Analysis

Revenue Analysis

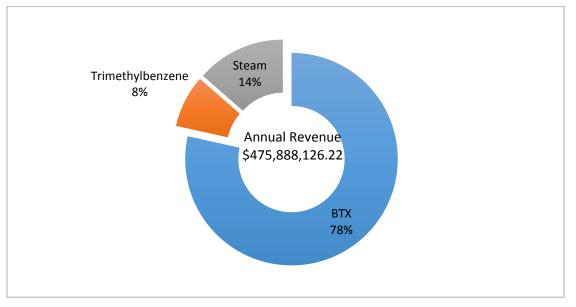
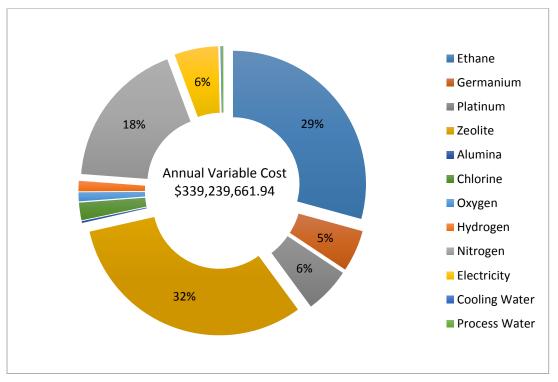


Figure 10. Revenue Stream Analysis

Analysis on the revenue stream was conducted to examine the source of revenue of the project and is presented in Figure 10. As expected, BTX accounts for 78% of the entire revenue stream and trimethylbenzene accounts for 8%. Surprisingly, the revenue generated from selling excess steam created during the process brings in a greater proportion of revenue than trimethylbenzene.



Cost Analysis

Figure 11. Variable Cost Analysis

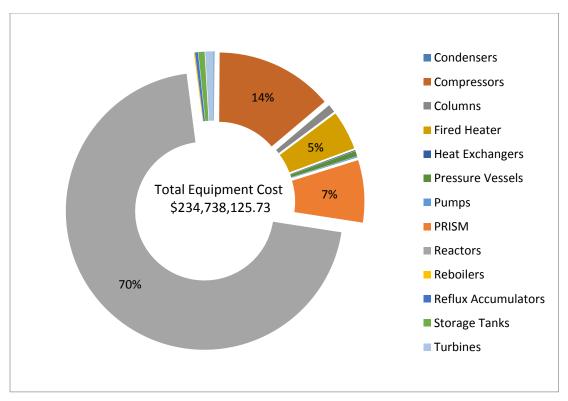
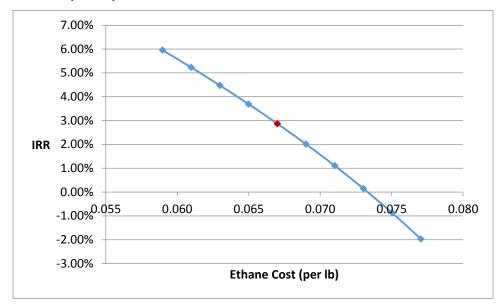


Figure 12. Equipment Cost Analysis

Analysis on the cost streams was conducted on variable and equipment costs. As shown in Figure 11 above, 42% of the variable cost consisted of catalyst costs. An additional 29% was contributed by the cost of the main feedstock ethane. Regeneration chemicals nitrogen, chlorine, hydrogen, and oxygen added 22%. Conversely, utilities contributed only 5% of total variable costs. Thus, the IRR and NPV for the project depend very much on catalyst and ethane prices, which will be shown in the Sensitivity Analyses section below. Meanwhile, the three reactors required for the plant constitute 70% of equipment costs. The PRISM separation membrane, multistage compressor, and fired heater contribute an additional 26% of the total costs, such that these six pieces of equipment make up almost the entirety of the cost of manufacture.

Sensitivity Analyses

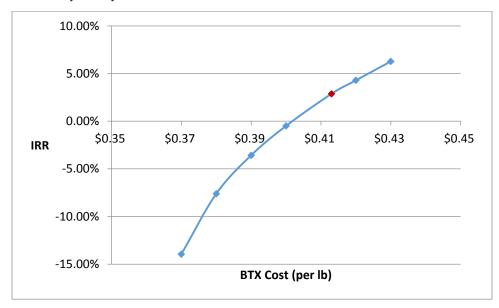


Sensitivity Analysis on Feed Price

Figure 13. Effect of Ethane Price on IRR

With an estimated price of ethane of \$0.067 per lb, the IRR is 2.87%. The IRR of the project is fairly sensitive to the price of ethane, as it is the only feedstock into the process. Sensitivity analysis was conducted on ethane price on an increase/decrease of up to 15%. Historical prices for ethane have been fairly stable, as exhibited in Figure 4, thus a limited range of volatility was chosen. As shown in Figure 13, the IRR remains positive within most of this range of prices. Further, it is unlikely that ethane's cost would be higher than this range, since the estimate comes from current market spot prices. In reality, the feed of ethane is actually fractional grade ethane coming as a waste product from other hydrocarbon

processing plants in the Gulf Coast, which is not pure ethane and would likely be cheaper than our estimates. Thus, in the event overall natural gas prices fall significantly, the price of ethane would likely follow suit, and greatly increase the IRR.



Sensitivity Analysis on Product Price

As seen in Figure 10 in the revenue stream analysis, 78% of the revenue comes from BTX, which is the main product from the process. A sensitivity analysis was conducted on the selling price of BTX, as the profitability of the project is heavily dependent on it. As stated in the problem statement for this project, BTX prices have been fairly stable at \$3.00 per gallon, or \$0.413 per lb. As expected, the IRR increases as the price of BTX increases. However, the project can maintain a positive IRR only in market conditions where the price of BTX stays above \$0.40 per lb, a level that can be tested by a drop in crude oil prices.

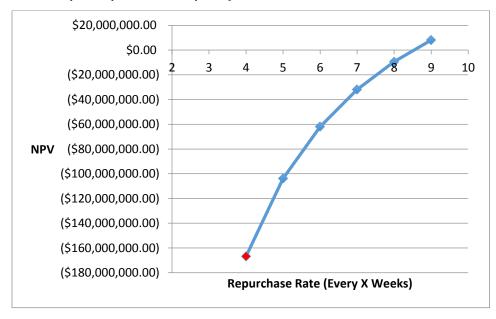
Figure 14. Effect of BTX price on IRR



Sensitivity Analysis on Catalyst Cost

Figure 15. Effect of zeolite price on IRR

As 77% of the variable cost, the catalyst and its components greatly influence the profitability of the project. At 57% of total variable cost, zeolite is the costliest of the components, and thus a sensitivity analysis was conducted on the price of zeolite. At a price of \$12.50 per kg, the IRR is 2.87%. Due to disagreeing estimates of the cost of \$5/kg and \$15/kg, a range of prices between the two were considered. In order to retain a positive IRR, the zeolite price can be no greater than \$13.50 per kg. This calculates to an 8% premium on the current estimated price, a small but not insignificant. The low estimate of \$5/kg would meet an IRR of 16.10%.



Sensitivity Analysis on Catalyst Lifetime

Figure 16. Effect of catalyst lifetime on NPV

For the same reasons stated under Sensitivity Analysis on Catalyst Cost, a sensitivity analysis was conducted on catalyst lifetime. An increase in the expected lifetime of the catalyst would lead to less frequent repurchases of the catalyst components, and hence lower costs. Figure 16 shows the effect of the catalyst lifetime on NPV, and one can see that an increase of just a few weeks in lifetime puts the NPV into positive territory. The current estimate of a lifetime of four weeks or one month is already a conservative estimate for catalyst lifetimes in industrial processes, so we see potential for the project will to become profitable. However, due to the sensitivity of NPV to changes in the catalyst lifetime, every step should be taken to maximize the lifetime of the catalyst. A one-month increase in the catalyst lifetime from one to two months increases the NPV by over \$150 million.

Assumptions for Economic Analysis

Tax rate: 37%

Cost of capital: 18%. Due to the presumed mature nature of the Company, a discount rate in the range of 10-25% was chosen²⁵.

Tax Credit: In economic analysis, the Company was assumed to be profitable during the lifetime of plant operation, and therefore the Company holds no tax credits or carryover losses.

Operating Times: The plant was assumed to operate 330 days a year for 24 hours a day. First year production was also assumed to be 50% of full production capacity, with a linear ramp-up over three years to 90% of full capacity in the third year and onwards.

Salvage Value: Given the highly specific nature of the equipment and a long operating period of 15 years, there was assumed to be no salvage value to the plant.

Other Important Considerations

Safety

Almost all streams and equipment in this process operate at high pressures and temperatures. Also, the hydrocarbons in this process are volatile and flammable. All operators and employees on the plant should be equipped with safety equipment, well trained on safety procedures if a disaster should arise and be cognizant of the normal operating conditions of all equipment on the plant. Leaks should be reported immediately and all units should be inspected regularly for preventative maintenance. All chemicals should be handled with care and copies of the Material Safety Data Sheets (MSDS) that are located in Appendix A-4 should be made available to all employees.

Maintenance

The process design allows the plant to be continuously operational for 330 days in a year, with the remainder reserved for maintenance. For the reactors (R-101, R-102, and R-103), the spent catalyst will be replaced monthly. The heat exchangers will have to be cleaned and inspected for fouling for improved heat transfer. In addition, condensers and reboilers should be inspected for fouling and corrosion.

The turbines will need to have motors and pumps replaced in case of failure.²² The storage tanks and pressure vessel should be cleaned of leftover product, sludge, and solids built up throughout the 330 day cycle. Whenever a primary pump in the process fails, the safety pump should be operational and workers should identify whether the primary pump can be fixed or if a new pump will have to be purchased. Distillation columns²³ and flash vessels are associated with failure modes such as overheating, impurities, and leaks⁶. This could potentially lead to weekly maintenance checks on these pieces of equipment due to the large amount of material passing through them. For the steam boilers, failures in the fan should be avoided to ensure that enough steam utility is supplied for the process⁸. Lastly, the PRISM unit will need to be checked regularly due to its high operating conditions and large feed flow rates.

Startup

The cost of plant startup will be close to 10% of the total depreciable capital cost¹³. Due to the large amounts of chemicals being processed and the size of the equipment being considered, enough time must be taken to achieve steady state in the major pieces of equipment (e.g., reactor, distillation columns) before production can begin. The startup plan must account for additional pieces of equipment, such as heat exchangers, which will assist in achieving appropriate temperatures for safe operation. Safe methods for the transport of feedstock, product, and other key chemicals will also be required; the costs of setting

up pipelines to ferry ethane feedstock and hydrogen to the site must be included, as well as the costs of transporting the final BTX product to an adjacent plant for use.

Environmental

This process consumes a significant amount of energy that is produced through the burning of fuel gas (a mixture of hydrogen, methane, C2, and C3-hydrocarbons). Therefore the process will produce significant NO_x and CO_2 emissions, which will require government permits in order to follow federal and state regulations. In addition, the gaseous chemicals involved in the regeneration cost may need to be scrubbed before release into the atmosphere.

Control

The implementation of flow-control valves is necessary to reduce disturbances and keep the process under a strict, continuous schedule. The flow control valves should operate either with proportional-integral or proportional-integral-derivative control. All flow control valves contacted with streams below 900°F can be constructed with carbon steel²⁴; otherwise, stainless steel AISI 304L can be used for temperatures below 1400°F. A type of control valve to consider is the single-port valve body due to its frequent usage in industrial plants⁸.

Flow control valves should be used for the multistage compressors C-201 and C-202 to keep the compressors below an electricity requirement of 30,000 hp. For storage tanks ST-201 through ST-206, the BTX and heavy hydrocarbon products will have a valve corresponding to each storage tank to ensure that products are evenly distributed. For condensers COND-301 and COND-302, valves should be put for the cooling water utilities to adjust for disturbances in the tube side inlet temperatures. Similarly, for reboilers RB-301 and RB-302, valves should be placed on the steam utilities to adjust for disturbances in the shell side inlet temperatures. For the reactor inlet stream, S-106, a valve should be implemented to adjust for a higher or lower expected amount of material from the recycle streams. In addition, there should be three valves to ensure that flow does not go through a reactor undergoing regeneration. Lastly, the pipelines containing the materials used in the regeneration process should each have a valve to adjust the amount of material going into a reactor with respect to regeneration stage.

Pilot Plant

Several assumptions have been made in this preliminary analysis that will need to be tested in a pilot plant before proceeding to full industrial scale. As the catalyst is novel and has not yet been used industrially, it will need to be tested to determine whether there are significant changes to conversion and

selectivity data when scaling up from laboratory experiments. Another key assumption is the lifetime of the catalyst: the patent authors performed short single pass conversion tests that did not account for deactivation. At the current harsh operating conditions of the reactor, coke formation will certainly foul the catalyst surface, and the costs of the catalyst itself and of the regeneration process depend on the number of times the catalyst can be regenerated and replaced. Significant deactivation of catalyst would lead to higher operating costs, increasing the length of time to achieve profitability.

Acknowledgements

Conclusions and Recommendations

After a meticulous analysis of materials, utilities, equipment, and profitability of the process, we have concluded that the process will be unprofitable under the internal rate of return criterion of 15%. The total capital cost of the process is \$347,191,577 and the operating cost is \$287,013,885 per year. Overall, the IRR for this project is estimated as 2.87% and the NPV is estimated at -\$166,765,000 at a discount rate of 18%.

Due to the plentiful supply of raw materials and to the availability of consumers of BTX and other side products in the area, the Gulf Coast is the optimal location for this plant. The main factors that affect the IRR are the cost of ethane, cost of catalyst material, and the catalyst lifespan. Under sensitivity analysis of our inputs, we find that in order for this project to be profitable, the repurchase rate/lifetime of our catalyst needs to be 8.5 weeks or longer. Alternatively, a zeolite price of under \$5.50 per kg would return an IRR of over 15%. In addition, a rise in the price of BTX to \$0.47 per lb will also yield an IRR value of over 15%. Further, there are concerns surrounding price stability due to the recent drop in crude oil and natural gas prices. Therefore, this process should only be carried out under particular advantageous market conditions.

Multiple assumptions made through this process design will need to be investigated to improve the profitability of this project. The transition from experimental- to industrial-scale production must be carefully observed over a suggested span of one month to evaluate any failures in the process equipment and to determine effects of scale-up on conversions. In particular, the effects of scale-up on the lifetime of the catalyst must be examined in detail, as no information was provided in the patent. In addition, the regeneration process should undergo further research to reduce the time and amount of material required per stage. Methods to reduce or eliminate hydrogen and methane should be developed to reduce the cost of separation and compression utilities. Process equipment with large flow rates such as the PRISM unit will need careful maintenance to prevent buildup of excess methane and hydrogen.

Based on the economic analyses and recommendations stated above, construction of the proposed plant to produce 0.5MM tons of BTX per year is not recommended. The IRR of 2.87% and the negative NPV of \$167 million, along with a trend of decreasing price of natural gas and of BTX, are the important considerations supporting this recommendation.

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Appendices

A-1. Problem Statement

BTX from Ethane (recommended by Richard Bockrath, Consultant – formerly DuPont)

Background. The recent explosion of fracking in the USA is dramatically increasing the availability of light paraffinic hydrocarbons for use as feedstocks. This is especially true for ethane. Numerous olefin cracker projects have recently been announced along the Gulf Coast using ethane as the feedstock. Your company is a major consumer of BTX for plastics and other downstream products and your management has asked your team to evaluate the feasibility of using ethane as a feedstock to make BTX. You are to view this as a preliminary analysis. If your analysis is promising, more resources will be added to address the uncertainties you will likely uncover to flesh out a detailed proposal.

Current Situation. Alkanes-to-aromatic processes were a significant research area in the early 90s (Cyclar, Aromax, etc), but research has been less active since then – possibly due to the inability to find suitable catalysts. Your team has uncovered a recent patent by Sabic, which has a potentially novel catalyst for this application. It shows several propane-to-BTX examples. While ethane may behave differently than propane, it is about 1/6 the price of propane and more stable in pricing. Your management would like you to evaluate the patent by assuming ethane and propane will give the same conversions and selectivities. Obviously this is a key uncertainty for a future team to study if the economics look promising.

The best candidate catalyst contains germanium, which you have never seen used in a petrochemical catalyst. You will need to understand the germanium market to determine whether this catalyst is commercially viable.

When locating your plant, consider feedstock availability and the cost of transporting a liquid product(s). You are a global BTX producer with customer plants along the Gulf Coast, in the Rotterdam area, and in Shanghai. Your largest downstream customers are in the USA and Mainland China, where much of the TPA from para-xylene is consumed.

Since your team will likely lead the next phase of this analysis, clearly document your key assumptions for management. You don't want them to be surprised by your omissions!

The US Government maintains pricing information for major hydrocarbon feedstocks. A recent price for ethane appears to be stable-to-declining at \$0.25/gal. (http://www.eia.gov/todayinenergy/detail.cfm?id=12291).

Your supply-chain experts predict that BTX will sell for an average of \$3.00/gal in the foreseeable future.

A-2. Sample Calculations

A-2.1. Area of a Heat Exchanger

The methods for calculating the area of a heat exchanger is based on methods in Chapter 18 in <u>Process</u> and <u>Product Design Principles</u>, 4th ed. Heat exchanger H-201 will be the basis of our sample calculation, which involves transferring heat for the vapor product of COL-201 and the distillate product of COL-302. An overall heat exchanger coefficient, U, is assumed to be 149.7 BTU/hr-ft²-°F due to the transferring of heat between vapor and liquid streams. Temperatures of each stream and the heat duty were calculated using ASPEN PLUS:

For counter-current flow:

 $T_{hot,in} = 200.2^{\circ}F$ $T_{hot,out} = 136.4^{\circ}F$ $T_{cold,in} = 104.0^{\circ}F$ $T_{cold,out} = 116.1^{\circ}F$ Q = 3497690 BTU/hr

Calculation of Log Mean Temperature Difference:

$$\Delta T_1 = T_{hot,in} - T_{cold,out} = 200.2^{\circ}F - 116.1^{\circ}F = 84.1^{\circ}F$$
$$\Delta T_2 = T_{hot,out} - T_{cold,in} = 136.4^{\circ}F - 104.0^{\circ}F = 32.4^{\circ}F$$
$$\Delta T_{lm} = \frac{(\Delta T_1 - \Delta T_2)}{\ln\left(\frac{\Delta T_1}{\Delta T_2}\right)} = \frac{(84.1^{\circ}F - 32.4^{\circ}F)}{\ln\left(\frac{84.1^{\circ}F}{32.4^{\circ}F}\right)} = 54.2^{\circ}F$$

Calculation of area:

$$A = \frac{Q}{U * \Delta T_{lm}} = \frac{3497690 \text{ BTU/hr}}{(149.7 \frac{\text{BTU}}{\text{hr} - \text{ft}^2 - ^{\text{o}}\text{F}} * 54.2^{\circ}\text{F})} = 431.1 \text{ ft}^2$$

A-2.2. Reactor Volume

The weight of catalyst required depends on the volume in which reaction takes place. The gas hourly space velocity (GHSV) was specified in US Patent 209,795 as 1000 h^{-1} . The flow rate of the inlet to the reactor was obtained from the ASPEN stream report as $9.15*10^6 \text{ ft}^3/\text{h}$. Reactor volume was then calculated as follows:

Reactor volume
$$[ft^3] = \frac{Inlet \, flow \, rate}{GHSV} = \frac{9.15 * 10^6 \frac{ft^3}{h}}{1000 \, h^{-1}} = 9150 \, ft^3$$

A-2.3. Catalyst Weight

The total weight of catalyst required was calculated using a bulk density of H-ZSM-5 zeolite catalyst of 1.06 g/cm^3 , or 66.0 lb/ft^3 . The catalyst weight was calculated as follows:

Catalyst weight
$$[lb] = (Bulk \ density) * (Reactor \ volume) = 66.0 \ \frac{lb}{ft^3} * 9150 \ ft^3 = 604,000 \ lb$$

A-2.4. Dimensions of a Distillation Column

The diameter of the flash vessel, COL-201, distillation columns COL-301 and COL-302 were calculated by using the correlations found in Seider et al^{13} .

Notation

Symbol	Description
A _a	Active area on tray
$\mathbf{A}_{\mathbf{h}}$	Total hole area
C_{SB}	Parameter related to a correlation based on commercial-size towers covering tray spacing from
	6 to 36 inches.
D_t	Diameter of column
$F_{\rm F}$	Foaming factor
F_{HA}	Hole-area factor
F_{LG}	Flow ratio parameter
F_{ST}	Surface tension factor
Н	Height of the column
H _{head}	Height of the head Space
space	
H_{sump}	Height of the Sump Space
H _{tray}	Height of a single tray
L	Liquid mass flow rate
N _{trays}	Number of trays
T _c	Critical Temperature
T_r	Reduced Temperature
U	Velocity of a droplet in the column
U_{f}	Flooding Velocity
V	Vapor mass flow rate

x_L Mass fraction of liquid

 ρ_L Density of the liquid

 ρ_V Density of the vapor

 σ Surface Tension

Using ASPEN stream report and block report, located in Appendix A-5, for data on COL-302.

$$H = H_{head space} + (N_{trays} - 1) \times H_{tray} + H_{sump}$$
(1)

Using $H_{head space} = 4$ ft, $N_{trays} = 42$, $H_{tray} = 1.5$ ft, $H_{sump} = 10$ ft: H = 77 ft

Assuming

$$U = 0.85U_f \tag{2}$$

 U_f may be calculated by using equation 3.

$$U_f = C_{SB} F_{ST} F_F F_{HA} \left(\frac{\rho_L - \rho_V}{\rho_V}\right)^{\frac{1}{2}}$$
(3)

C_{SB}, F_{ST}, F_F and F_{HA} must be found before the flooding velocity can be calculated.

 F_{ST} can be calculated using Equation 4.

$$F_{ST} = \left(\frac{\sigma}{20}\right)^{0.5} \tag{4}$$

 σ may be found by using Equation 5.

$$\sigma = 71.815 \times (1 - T_r)^{1.2362} \tag{5}$$

The average critical temperature was calculated for the liquid on stage 2 by using equation 6:

$$T_{c,avg} = \sum_{i} x_{L,i} T_{c,i} \tag{6}$$

 $T_{c,avg}$ was determined to be 1013 R.

The temperature on the second stage was found to be 617 R. The reduced temperature was calculated using Equation 7:

$$T_{r,avg} = \frac{T}{T_{c,avg}} \tag{7}$$

 $T_{r,avg}$ was found to be 0.609.

 σ was then found to be 22.5 dynes/cm by substituting T_{r,avg} in equation 5.

 F_{ST} was then found to be 1.02 by substituting σ into equation 4.

C_{SB} may be found by finding F_{LG}, and the plate spacing and then using Figure 19.4 in Seider et al.

 F_{LG} may be found by using equation 8.

$$F_{LG} = \left(\frac{L}{V}\right) \times \left(\frac{\rho_V}{\rho_L}\right)^{0.5} \tag{8}$$

From the ASPEN Block Report, L was found to be 0.637 lb/s and V was found to be 6.74 lb/s. A stream was created of the same composition as the liquid and vapor on the second stage of the column and the densities were found to be 48.8 lb/ft³ and 0.932 lb/ft³ respectively.

 F_{LG} was then found to be 0.0131. The corresponding C_{SB} value for 1.5 ft spacing was found to be 0.3 ft/s.

 F_F was assumed to be 1 since the contents of the column were a non-foaming system.

 F_{HA} was also 1 as it was assumed that $\frac{A_h}{A_a} = 0.1$.

Substituting these values into equation 3 gives a U_f of 2.2 ft/s and a U of 1.87 ft/s.

Finally the diameter may be calculated using equation 9:

$$D_t = \frac{4}{\pi} \times \frac{V}{\left(1 - \frac{A_h}{A_a}\right) U \rho_V} \tag{9}$$

The diameter of the column was found to be 2.33 ft.

A-2.5. Flow Rates for Regeneration Process

Notation

Symbol	Description
m _{coke}	Mass of coke built up on the catalyst surface
m _{catalyst}	Mass of the catalyst
n _{coke}	Number of moles of coke built up on the catalyst suface
n _{O2}	Number of moles of oxygen required to react with all the coke on the surface of the catalyst
'n _{j,i}	Molar flow rate of component i on stage j
$y_{j,i}$	Mole fraction of compound i on stage j

Assuming that the catalyst must be regenerated when the coke deposited on the surface is 5% of its weight and all of the coke must be burned off to be regenerated.

$$m_{coke} = 0.05 m_{catalyst} \tag{10}$$

Using a value of 1.29E+06 lb, the mass of coke, m_c, on the surface was calculated to be 6.47E+05 lb.

The number of moles of coke, n_{coke} , was then calculated by using its molecule weight and was found to be 5390 lb-mol.

By using Equation 11, the number of moles of oxygen required may be calculated.

$$C + O_2 \to CO_2 \tag{11}$$

The number of moles of oxygen, n_{02} , was found to be 5390 lb-mol. Since all of the carbon was burned off after stage 2, it was assumed that an equal amount of coke was burned off in each stage. Therefore, each stage required 2695 lb-mol of oxygen was required for each stage. The flow rate of oxygen was then found by dividing by the time duration of the stage. Since stage 1 is conducted for 24 hours, the flow rate of oxygen required was found to be 112 lb-mol/hr.

The partial pressures listed for stage 1 from the patent were then used to determine the mole fraction of each gas by using Dalton's Law of Partial Pressures. The flow rate of each gas could then be calculated by using Equation 12.

$$\dot{n}_{1,i} = \dot{n}_{1,0_2} \times \frac{y_{1,i}}{y_{1,0_2}} \tag{12}$$

The flow rates of the other reagents are listed in Table 35.

Species	Mole Fraction	Flow Rate Required for Stage 1
		(lb-mol/hr)
Chlorine	0.004	11.2
Water Vapor	0.033	89.6
Oxygen	0.042	112
Hydrogen	0.000	0
Nitrogen	0.921	2475

Table 35. Mole fractions and flow rates required for stage 1 for each species in the regeneration process.

The total flow rate of stage 1 can be found by setting $y_{1,i}$ to 1. The total flow was found to be 2688 lb-mol/hr.

Since the pressure and temperature of each stage were the same, it was assumed that the flow rate of each stage was the same. The mole flow rate of each component could then be calculated by using the mole fraction and the total mole flow rate as shown in Equation 13.

$$\dot{n}_{j,i} = \dot{n}_1 \times y_{j,i} \tag{13}$$

A-3. Profitability Analysis Spreadsheet

INTRODUCTION

Prepared by Brian K. Downey, Equity Research - US Royal Gas Exploration / Production, Sanford C. Bernstein & Co., LLC

This Introduction provides brief guidlines for using the Profitability Analysis 4.0. Further and extensive details about each tab are provided in Section 17.8 of the textbook (4th Edition)

You cannot enter values in the green tabs. However, you can enter values in the the other tabs, although not in every cell. This is to prevent undesired edits from the user.

Different parameters can be entered for the evaluation of the project. One can also use parameters mentioned in Table 17.1. Equipment costs can be generated from APEA or the equations in Chapter 16 of the textbook. These two methods are mentioned in Chapter 17, Section 17.8.

A-3.1. Input Summary

neral Information		
Process Title:	BTX from Ethane	
Product:	BTX	
Plant Site Location:	Gulf Coast	
Site Factor:	1.00	
Operating Hours per		
Year:	7919	
Operating Days Per		
Year:	330	
Operating Factor:	0.9040	

This Process will Yield

114,000	lb of BTX per hour
2,736,000	lb of BTX per day
902,770,560	lb of BTX per year

Price

\$0.41 /lb

Chronology

CIIIOIO	уу				
		Distribution of	Production	Depreciation	Product Price
Year	<u>Action</u>	Permanent Investment	<u>Capacity</u>	5 year MACRS	
2015	Design		0.0%		
2016	Construction	100%	0.0%		
2017	Production	0%	45.0%	20.00%	\$0.41
2018	Production	0%	60.0%	32.00%	\$0.42
2019	Production	0%	75.0%	19.20%	\$0.43
2020	Production		90.0%	11.52%	\$0.43
2021	Production		90.0%	11.52%	\$0.44
2022	Production		90.0%	5.76%	\$0.45
2023	Production		90.0%		\$0.45
2024	Production		90.0%		\$0.46
2025	Production		90.0%		\$0.47
2026	Production		90.0%		\$0.47
2027	Production		90.0%		\$0.48
2028	Production		90.0%		\$0.49
2029	Production		90.0%		\$0.49
2030	Production		90.0%		\$0.50
2031	Production		90.0%		\$0.51

Equipment Costs		
Equipment Description		Bare Module Cost
COL-		
201	Process Machinery	\$437,618
COL- 301	Process Machiner	¢010 500
COL-	Process Machinery	\$818,588
302	Process Machinery	\$1,380,192
C-201	Process Machinery	\$16,068,648
C-202	Process Machinery	\$16,068,648
PRISM	Fabricated Equipment	\$17,000,000
COND-		
301	Process Machinery	\$57,297
COND-	_	
302	Process Machinery	\$174,664
H-201	Process Machinery	\$69,445
H-202	Process Machinery	\$60,689
H-203	Process Machinery	\$70,972
P-201	Process Machinery	\$10,854
P-301	Process Machinery	\$9,747
P-302	Process Machinery	\$12,877
P-303	Process Machinery	\$11,943
P-304	Process Machinery	\$12,990
P-401	Process Machinery	\$43,998
PV-101	Process Machinery	\$1,270,743
RA-301	Process Machinery	\$584,703
RA-302	Process Machinery	\$117,710
RB-301	Process Machinery	\$130,846
RB-302	Process Machinery	\$94,785
ST-204	Storage	\$72,750

ST-205	Storage	\$72,750
ST-206	Storage	\$72,750
ST-201	Storage	\$418,837
ST-202	Storage	\$418,837
ST-203	Storage	\$418,837
T-101	Process Machinery	\$446,431
T-102	Process Machinery	\$446,431
T-103	Process Machinery	\$446,431
T-104	Process Machinery	\$305,838
T-105	Process Machinery	\$21,360
T-401	Process Machinery	\$194,504
P-101	Process Machinery	\$15,500
P-202	Process Machinery	\$10,818
#REF!	#REF!	#REF!

Additional Equipment

<u>Total</u>

\$175,810,937 <u>#REF!</u>

Raw Materials							
	Raw Material:	<u>Unit:</u>	Required Ratio:			Cost of Raw M	aterial:
1	Ethane	lb		1.6335088	lb per lb of BTX	\$0.067	per lb
2	Germanium	kg		1.001E-05	kg per lb of BTX	\$1,900.00	per kg
3	Platinum	kg		5.22E-06	kg per lb of BTX	\$4,014.20	per kg
4	Zeolite	kg		0.0094668	kg per lb of BTX	\$12.50	per kg
5	Alumina	kg		0.0023667	kg per lb of BTX	\$0.30	per kg
6	Chlorine	lb		0.0118041	lb per lb of BTX	\$0.68	per lb
7	Oxygen	lb		0.0455605	lb per lb of BTX	\$0.09	per lb
8	Hydrogen	lb		0.0074995	lb per lb of BTX	\$0.64	per lb
9	Nitrogen	lb		0.3747448	lb per lb of	\$0.18	per lb

	Total Weighted Average:					\$0.353	per lb of BTX
oroduc	cts						
	Byproduct:	<u>Unit:</u>	Ratio to Product			Byproduct Sell	ing Price
	T	п		0.4000040	lb per lb of	\$0.444	11
1	Trimethylbenzene	lb		0.1026316	BTX lb per lb of	\$0.411	per lb
2	Steam (450 psig)	lb		8.9113158	BTX	\$8.000E-03	per lb
	Total Weighted Average:					\$0.113	per lb of BTX
ities						\$0.113	per lb of BTX
ities		<u>Unit:</u>	Required Ratio			\$0.113 <u>Utility Cost</u>	per lb of BTX
ities 1	Average:	<u>Unit:</u> kWh	Required Ratio	0	kWh per lb of BTX		per lb of BTX
ities 1 2	Average: Utility:		Required Ratio	0	kWh per lb of BTX	Utility Cost	
1	Average: <u>Utility:</u> High Pressure Steam	kWh	Required Ratio		kWh per lb of BTX lb per lb of BTX	<u>Utility Cost</u> \$0.000E+00	per kWh
1 2	Average: <u>Utility:</u> High Pressure Steam Low Pressure Steam	kWh kWh	Required Ratio	0	kWh per lb of BTX lb per lb of	<u>Utility Cost</u> \$0.000E+00 \$0.000E+00	per kWh per kWh
1 2 3	Average: <u>Utility:</u> High Pressure Steam Low Pressure Steam Process Water	kWh kWh lb	Required Ratio	0 8.9022692	kWh per lb of BTX lb per lb of BTX lb per lb of	<u>Utility Cost</u> \$0.000E+00 \$0.000E+00 \$8.000E-04	per kWh per kWh per lb

BTX

Total Weighted

\$0.031 per lb of BTX

Ave	erage:				
	#REF!	#REF!	#REF!		#REF!
Variable Costs	-				
Ge	neral Expenses:		Colling / Transfer Evenences	3.00%	of Sales
			Selling / Transfer Expenses:		
			Direct Research: Allocated Research:	4.80% 0.50%	of Sales of Sales
			Administrative Expense:	2.00%	of Sales
			Management Incentive Compensation:	1.25%	of Sales
Working Capit	tal				
Acc	counts Receivable		⇔	30	Days
Cas	sh Reserves (exclud	ing Raw			-
	terials)		c >	30	Days
	counts Payable		⇔	30	Days
	X Inventory		⇔	3	Days
Rav	w Materials		₽	2	Days
Total Permane	ent Investment				
			Cost of Site Preparations:	5.00%	of Total Bare Module Costs
			Cost of Service Facilities:	5.00%	of Total Bare Module Costs
			Allocated Costs for utility plants and related		
			facilities:	\$0	
			Cost of Contingencies and Contractor Fees:	18.00%	of Direct Permanent Investment
			Cost of Land:	2.00%	of Total Depreciable Capital
			Cost of Royalties:	\$0	
			Cost of Plant Start-Up:	10.00%	of Total Depreciable Capital
Fixed Costs					
Ор	erations				
				_	

Operators per Shift: Direct Wages and Benefits:

8 (assuming 5 shifts) \$40 /operator hour

		Direct Salaries and Benefits: Operating Supplies and Services: Technical Assistance to Manufacturing: Control Laboratory:	15% 6% \$60,000.00 \$65,000.00	of Direct Wages and Benefits of Direct Wages and Benefits per year, for each Operator per Shift per year, for each Operator per Shift
<u>Maintenance</u>		Wages and Benefits: Salaries and Benefits: Materials and Services: Maintenance Overhead:	3.50% 25% 100% 5%	of Total Depreciable Capital of Maintenance Wages and Benefits of Maintenance Wages and Benefits of Maintenance Wages and Benefits
Operating Overhead				
		General Plant Overhead:	7.10%	of Maintenance and Operations Wages and Benefits of Maintenance and Operations Wages and
		Mechanical Department Services:	2.40%	Benefits
		Employee Relations Department:	5.90%	of Maintenance and Operations Wages and Benefits
		Business Services:	7.40%	of Maintenance and Operations Wages and Benefits
Property Taxes and Insurance		Property Taxes and Insurance:	2%	of Total Depreciable Capital
			270	
<u>Straight Line Depreciation</u> Direct Plant:	8.00%	of Total Depreciable Capital, less 1.18 times the A	llocated Costs	for Utility Plants and Related Facilities
Allocated Plant:	6.00%	of 1.18 times the Allocated Costs for Utility Plants	and Related Fa	-
Other Annual Expenses				
<u> </u>		Rental Fees (Office and Laboratory Space):	\$0	
		Licensing Fees:	\$0	

Depletion Allowance

	,	Annual Depletion A	lowance	:	\$0	
3.2. Cost Summary						
ariable Cost Summary						-
Variable Costs at 100% Capacity:						
<u>General Expenses</u>						
Sellin	ng / Transfer Expenses:		\$	11,198,277		
Direc	ct Research:		\$	17,917,244		

4,665,949

287,013,885

<u>\$</u>

Allocated Research: \$ 1,866,380

Administrative Expense:	\$ 7,465,518

Management Incentive Compensation:	\$

Total General Expenses			\$ 43,113,367
Raw Materials	\$0.352842	per lb of BTX	\$318,535,578
Byproducts	\$0.113426	per lb of BTX	(\$102,397,305)
<u>Utilities</u>	\$0.030752	per lb of BTX	\$27,762,245

Total Variable Costs

A-3.2. Cost Summary

Variable Cost Summary

Fixed Cost Summary

Operations

	Direct Wages and Benefits	\$ 3,328,000
	Direct Salaries and Benefits	\$ 499,200
	Operating Supplies and Services	\$ 199,680
	Technical Assistance to Manufacturing	\$ 2,400,000
	Control Laboratory	\$ 2,600,000
	Total Operations	\$ 9,026,880
<u>Maintenance</u>		
	Wages and Benefits	\$ 10,616,126
	Salaries and Benefits	\$ 2,654,032
	Materials and Services	\$ 10,616,126
	Maintenance Overhead	\$ 530,806
	Total Maintenance	\$ 24,417,091
Operating Overhead		
	General Plant Overhead:	\$ 1,213,912
	Mechanical Department Services:	\$ 410,337
	Employee Relations Department:	\$ 1,008,744
	Business Services:	\$ 1,265,204

	Total Operating Overhead	\$	3,898,198
Property Taxes and Ins	surance		
	Property Taxes and Insurance:	\$	6,066,358
Other Annual Expenses	<u>s</u>		
	Rental Fees (Office and Laboratory Space): Licensing	\$	-
	Fees:	\$	-
	Miscellaneous:	\$	-
	Total Other Annual Expenses	\$	<u> </u>
Total Fixed Costs		<u>\$</u>	43,408,526

Investment Summary	
Total Bare Module Costs:	
Fabricated Equipment	\$ 192,682,210
Process Machinery	\$ 39,395,272
Spares Storage	\$ 128,727

Other Equipment Catalysts Computers, Software, Etc.	\$ \$ \$	1,474,758 - - -	
Total Bare Module Costs:			<u>\$ 233,680,967</u>
Direct Permanent Investment			
Cost of Site Preparations:	\$	11,684,048	
Cost of Service Facilities: Allocated Costs for utility plants and related facilities:	\$ \$	11,684,048 -	
Direct Permanent Investment			<u>\$257,049,064</u>
Total Depreciable Capital			
Cost of Contingencies & Contractor Fees	\$	46,268,832	
Total Depreciable Capital			<u>\$ </u>
Total Permanent Investment			
Cost of Land: Cost of Royalties:	\$ \$	6,066,358 -	
Cost of Plant Start-Up:	\$	30,331,790	
Total Permanent Investment - Unadjusted			

Site Factor	\$	339,716,043 1.00
Total Permanent Investment	<u>\$</u>	339,716,043

Working Capital

	<u>2016</u>	<u>2017</u>		<u>2018</u>
Accounts Receivable	\$ 13,806,095	\$	4,602,032	\$ 4,602,032
Cash Reserves	\$ 2,632,344	\$	877,448	\$ 877,448
Accounts Payable	\$ (12,808,276)	\$	(4,269,425)	\$ (4,269,425)
BTX Inventory	\$ 1,380,610	\$	460,203	\$ 460,203
Raw Materials	\$ 785,430	\$	261,810	\$ 261,810
Total	\$ 5,796,203	\$	1,932,068	\$ 1,932,068
Present Value at 18%	\$ 4,912,036	\$	1,387,581	\$ 1,175,916

Total Capital Investment

<u>\$ 347,191,577</u>

A-3.3. Cash Flows

Cash Flow Summary

	Deveryters of								anning y		
<u>Year</u>	<u>Percentage of</u> <u>Design</u> <u>Capacity</u>	Product Unit Price	<u>Sales</u>	<u>C</u>	apital Costs	Working Capital	Var Costs	Fixed Costs	<u>Depreciation</u>	<u>Depletion</u> Allowance	<u>Taxil</u>
2015	0%			-	-	-	-	-	-	-	
2016	0%			- (339,	716,000)	(5,796,200)	-	-	-	-	
2017	45%	\$0.41	167,974,200		-	(1,932,100)	(129,156,200)	(43,408,500)	(60,663,600)	-	(65,2
2018	60%	\$0.42	227,325,000		-	(1,932,100)	(174,791,500)	(44,059,700)	(97,061,700)	-	(88,5
2019	75%	\$0.43	288,418,600		-	(1,932,100)	(221,766,700)	(44,720,500)	(58,237,000)	-	(36,3
2020	90%	\$0.43	351,293,900		-	-	(270,111,800)	(45,391,400)	(34,942,200)	-	848,5
2021	90%	\$0.44	356,563,300		-	-	(274,163,500)	(46,072,200)	(34,942,200)	-	1,385
2022	90%	\$0.45	361,911,700		-	-	(278,275,900)	(46,763,300)	(17,471,100)	-	19,40
2023	90%	\$0.45	367,340,400		-	-	(282,450,100)	(47,464,800)	-	-	37,42
2024	90%	\$0.46	372,850,500		-	-	(286,686,800)	(48,176,700)	-	-	37,98
2025	90%	\$0.47	378,443,300		-	-	(290,987,100)	(48,899,400)	-	-	38,55
2026	90%	\$0.47	384,119,900		-	-	(295,351,900)	(49,632,900)	-	-	39,13
2027	90%	\$0.48	389,881,700		-	-	(299,782,200)	(50,377,400)	-	-	39,72
2028	90%	\$0.49	395,730,000		-	-	(304,278,900)	(51,133,000)	-	-	40,31
2029	90%	\$0.49	401,665,900		-	-	(308,843,100)	(51,900,000)	-	-	40,92
2030 2031	90% 90%	\$0.50 \$0.51	407,690,900		-	- 11,592,400	(313,475,800)	(52,678,500)	-	-	41,53

Appendix A-3. Profitability Analysis Spreadsheet

	413,806,300	(318,177,900)	(53,468,700)	- 42,1
A-3.4. Profitability Measu	ıres			
Profitability Measures	S			
The Internal Rate of Retur	n (IRR) for this project is	2.87%		
The Net Present Value (NF	PV) of this project in 2015 is	\$ (166,765,000)		
ROI Analysis (Third Produ	uction Year)			
Annual Sales Annual	288,418,629			
Costs	(266,487,209)			
Depreciation	(27,177,283)			
Income Tax	1,940,969			
Net Earnings	(3,304,894)			
Total Capital Investment ROI	<u>351,308,449</u> -0.94%			

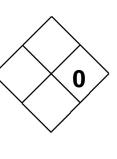
Sensitivity Analyses

Note: The Sensitivity Analyses section below takes quite a bit of memory to update each time a cell is changed; therefore, automatic calculations are turned off. After making your axis selections, prettie IRR values. (These two lines may be deleted before printing.)

	Vary Initial Value by +/-
x-axis	15%
y-axis	15%

						Variable Costs				
_	\$243,961,803	\$252,572,219	\$261,182,636	\$269,793,052	\$278,403,469	\$287,013,885	\$295,624,302	\$304,234,718	\$312,845,135	\$321,455,552
\$0.35	-1.29%	-4.92%	-10.03%	-20.58%	Negative IRR	Negative IRR	Negative IRR	Negative IRR	Negative IRR	Negative IRR
\$0.36	2.40%	-0.36%	-3.71%	-8.18%	-15.73%	Negative IRR	Negative IRR	Negative IRR	Negative IRR	Negative IRR
\$0.38	5.44%	3.14%	0.52%	-2.60%	-6.61%	-12.67%	Negative IRR	Negative IRR	Negative IRR	Negative IRR
\$0.39	8.07%	6.06%	3.85%	1.35%	-1.59%	-5.24%	-10.37%	-20.75%	Negative IRR	Negative IRR
\$0.40	10.42%	8.61%	6.66%	4.52%	2.13%	-0.64%	-4.01%	-8.51%	-16.06%	Negative IRR
\$0.41	12.57%	10.91%	9.15%	7.25%	5.17%	2.87%	0.24%	-2.90%	-6.93%	-13.01%
\$0.43	14.57%	13.02%	11.39%	9.66%	7.81%	5.80%	3.58%	1.07%	-1.88%	-5.55%
\$0.44	16.44%	14.99%	13.46%	11.86%	10.17%	8.36%	6.40%	4.26%	1.85%	-0.93%
\$0.45	18.22%	16.83%	15.39%	13.89%	12.32%	10.66%	8.89%	6.99%	4.91%	2.60%
\$0.46	19.91%	18.59%	17.22%	15.80%	14.32%	12.77%	11.14%	9.41%	7.55%	5.54%
\$0.48	21.54%	20.26%	18.95%	17.60%	16.19%	14.73%	13.21%	11.61%	9.91%	8.10%







Material Safety Data Sheet

Benzene MSDS

Section 1: Chemical Produc	t and Company Identification			
Product Name: Benzene	Contact Information:			
Catalog Codes: SLB1564, SLB3055,	Sciencelab.com, Inc.			
SLB2881	14025 Smith Rd. Houston, Texas 77396			
CAS#: 71-43-2	US Sales: 1-800-901-7247			
RTECS: CY1400000	International Sales: 1-281-441-4400			
TSCA: TSCA 8(b) inventory: Benzene	Order Online: ScienceLab.com CHEMTREC (24HR Emergency Telephone),			
Cl#: Not available.	call:			
Synonym: Benzol; Benzine	1-800-424-9300			
Chemical Name: Benzene	International CHEMTREC, call: 1-703-527-3887			
Chemical Formula: C6-H6	For non-emergency assistance, call: 1-281-441- 4400			

Section 2: Composition and Information on Ingredients			
position:			
Name	CAS #	% by Weight	

Toxicological Data on Ingredients: Benzene: ORAL (LD50): Acute: 930 mg/kg [Rat]. 4700 mg/kg [Mouse].

DERMAL (LD50): Acute: >9400 mg/kg [Rabbit]. VAPOR (LC50): Acute: 10000 ppm 7 hours [Rat].

Section 3: Hazards Identification

Potential Acute Health Effects:

Very hazardous in case of eye contact (irritant), of inhalation. Hazardous in case of skin contact (irritant, permeator), of ingestion. Inflammation of the eye is characterized by redness, watering, and itching.

Potential Chronic Health Effects:

CARCINOGENIC EFFECTS: Classified A1 (Confirmed for human.) by ACGIH, 1 (Proven for human.) by IARC. MUTAGENIC EFFECTS: Classified POSSIBLE for human. Mutagenic for mammalian somatic cells. Mutagenic for bacteria and/or yeast. TERATOGENIC EFFECTS: Not available. DEVELOPMENTAL TOXICITY: Classified Reproductive system/toxin/female [POSSIBLE]. The substance is toxic to blood, bone marrow, central nervous system (CNS). The substance may be toxic to liver, Urinary System. Repeated or prolonged exposure to the substance can produce target organs damage.

Section 4: First Aid Measures

Eye Contact:

Check for and remove any contact lenses. In case of contact, immediately flush eyes with plenty of water for at least 15 minutes. Cold water may be used. WARM water MUST be used. Get medical attention immediately.

Skin Contact:

In case of contact, immediately flush skin with plenty of water. Cover the irritated skin with an emollient. Remove contaminated clothing and shoes. Wash clothing before reuse. Thoroughly clean shoes before reuse. Get medical attention.

Serious Skin Contact:

Wash with a disinfectant soap and cover the contaminated skin with an anti-bacterial cream. Seek immediate medical attention.

Inhalation:

If inhaled, remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention if symptoms appear.

Serious Inhalation:

Evacuate the victim to a safe area as soon as possible. Loosen tight clothing such as a collar, tie, belt or waistband. If breathing is difficult, administer oxygen. If the victim is not breathing, perform mouth-to-mouth resuscitation. Seek medical attention.

Ingestion:

Do NOT induce vomiting unless directed to do so by medical personnel. Never give anything by mouth to an unconscious person. If large quantities of this material are swallowed, call a physician immediately. Loosen tight clothing such as a collar, tie, belt or waistband.

Serious Ingestion: Not available.

Section 5: Fire and Explosion Data

Flammability of the Product: Flammable.

Auto-Ignition Temperature: 497.78°C (928°F)

Flash Points: CLOSED CUP: -11.1°C (12°F). (Setaflash)

Flammable Limits: LOWER: 1.2% UPPER: 7.8%

Products of Combustion: These products are carbon oxides (CO, CO2).

Fire Hazards in Presence of Various Substances:

Highly flammable in presence of open flames and sparks, of heat. Slightly flammable to flammable in presence of oxidizing materials. Non-flammable in presence of shocks.

Explosion Hazards in Presence of Various Substances:

Risks of explosion of the product in presence of mechanical impact: Not available. Risks of explosion of the product in presence of static discharge: Not available. Explosive in presence of oxidizing materials, of

Fire Fighting Media and Instructions:

Flammable liquid, soluble or dispersed in water. SMALL FIRE: Use DRY chemical powder. LARGE FIRE: Use alcohol foam, water spray or fog.

Special Remarks on Fire Hazards:

Extremely flammable liquid and vapor. Vapor may cause flash fire. Reacts on contact with iodine heptafluoride gas. Dioxygenyl tetrafluoroborate is as very powferful oxidant. The addition of a small particle to small samples of benzene, at ambient temperature, causes ignition. Contact with sodium peroxide with benzene causes ignition. Benzene ignites in contact with powdered chromic anhydride. Virgorous or incandescent reaction with hydrogen + Raney nickel (above 210 C) and bromine trifluoride.

Special Remarks on Explosion Hazards:

Benzene vapors + chlorine and light causes explosion. Reacts explosively with bromine pentafluoride, chlorine, chlorine trifluoride, diborane, nitric acid, nitryl perchlorate, liquid oxygen, ozone, silver perchlorate. Benzene + pentafluoride and methoxide (from arsenic pentafluoride and potassium methoxide) in trichlorotrifluoroethane causes explosion. Interaction

Appendix A-4. MSDS Information

of nitryl perchlorate with benzene gave a slight explosion and flash. The solution of permanganic acid (or its explosive anhydride, dimaganese heptoxide) produced by interaction of permanganates and sulfuric acid will explode on contact with benzene. Peroxodisulfuric acid is a very powferful oxidant. Uncontrolled contact with benzene may cause explosion. Mixtures of peroxomonsulfuric acid with benzene explodes.

Section 6: Accidental Release Measures

Small Spill: Absorb with an inert material and put the spilled material in an appropriate waste disposal.

Large Spill:

Flammable liquid. Keep away from heat. Keep away from sources of ignition. Stop leak if without risk. Absorb with DRY earth, sand or other non-combustible material. Do not touch spilled material. Prevent entry into sewers, basements or confined areas; dike if needed. Be careful that the product is not present at a concentration level above TLV. Check TLV on the MSDS and with local authorities.

Section 7: Handling and Storage

Precautions:

Keep locked up.. Keep away from heat. Keep away from sources of ignition. Ground all equipment containing material. Do not ingest. Do not breathe gas/fumes/ vapor/spray. In case of insufficient ventilation, wear suitable respiratory equipment. If ingested, seek medical advice immediately and show the container or the label. Avoid contact with skin and eyes. Keep away from incompatibles such as oxidizing agents, acids.

Storage:

Store in a segregated and approved area. Keep container in a cool, well-ventilated area. Keep container tightly closed and sealed until ready for use. Avoid all possible sources of ignition (spark or flame).

Section 8: Exposure Controls/Personal Protection

Engineering Controls:

Provide exhaust ventilation or other engineering controls to keep the airborne concentrations of vapors below their respective threshold limit value. Ensure that eyewash stations and safety showers are proximal to the work-station location.

Personal Protection:

Splash goggles. Lab coat. Vapor respirator. Be sure to use an approved/certified respirator or equivalent. Gloves.

Personal Protection in Case of a Large Spill:

Splash goggles. Full suit. Vapor respirator. Boots. Gloves. A self contained breathing apparatus should be used to avoid inhalation of the product. Suggested protective clothing might not be sufficient; consult a specialist BEFORE handling this product.

Exposure Limits:

TWA: 0.5 STEL: 2.5 (ppm) from ACGIH (TLV) [United States] TWA: 1.6 STEL: 8 (mg/m3) from ACGIH (TLV) [United States] TWA: 0.1 STEL: 1 from NIOSH TWA: 1 STEL: 5 (ppm) from OSHA (PEL) [United States] TWA: 10 (ppm) from OSHA (PEL) [United States] TWA: 3 (ppm) [United Kingdom (UK)] TWA: 1.6 (mg/m3) [United Kingdom

(UK)] TWA: 1 (ppm) [Canada] TWA: 3.2 (mg/m3) [Canada] TWA: 0.5 (ppm) [Canada]Consult local authorities for acceptable exposure limits.

Section 9: Physical and Chemical Properties

Physical state and appearance: Liquid.

Odor:

Aromatic. Gasoline-like, rather pleasant. (Strong.)

Taste: Not available.

Molecular Weight: 78.11 g/mole

Color: Clear Colorless. Colorless to light

yellow. pH (1% soln/water): Not available.

Boiling Point: 80.1 (176.2°F)

Melting Point: 5.5°C (41.9°F)

Critical Temperature: 288.9°C (552°F)

Specific Gravity: 0.8787 @ 15 C (Water = 1)

Vapor Pressure: 10 kPa (@ 20°C)

Vapor Density: 2.8 (Air = 1)

Volatility: Not available.

Odor Threshold: 4.68 ppm

Water/Oil Dist. Coeff.: The product is more soluble in oil; log(oil/water) = 2.1

lonicity (in Water): Not available.

Dispersion Properties: See solubility in water, diethyl ether, acetone.

Solubility:

Miscible in alcohol, chloroform, carbon disulfide oils, carbon tetrachloride, glacial acetic acid, diethyl ether, acetone. Very slightly soluble in cold water.

Section 10: Stability and Reactivity Data

Stability: The product is stable.

Instability Temperature: Not available.

Conditions of Instability: Heat, ignition sources, incompatibles.

Incompatibility with various substances: Highly reactive with oxidizing agents, acids.

Corrosivity: Non-corrosive in presence of glass.

Special Remarks on Reactivity:

Benzene vapors + chlorine and light causes explosion. Reacts explosively with bromine pentafluoride, chlorine, chlorine trifluoride, diborane, nitric acid, nitryl perchlorate, liquid oxygen, ozone, silver perchlorate. Benzene + pentafluoride and methoxide (from arsenic pentafluoride and potassium methoxide) in trichlorotrifluoroethane causes explosion. Interaction of nitryl perchlorate with benzene gave a slight explosion and flash. The solution of permanganic acid (or its explosive anhydride, dimaganese heptoxide) produced by interaction of permanganates and sulfuric acid will explode on contact with benzene may cause explosion. Mixtures of peroxomonsulfuric acid with benzene explodes.

Special Remarks on Corrosivity: Not available.

Polymerization: Will not occur.

Section 11: Toxicological Information

Routes of Entry: Absorbed through skin. Dermal contact. Eye contact. Inhalation.

Toxicity to Animals:

WARNING: THE LC50 VALUES HEREUNDER ARE ESTIMATED ON THE BASIS OF A 4-HOUR EXPOSURE. Acute oral toxicity (LD50): 930 mg/kg [Rat]. Acute dermal toxicity (LD50): >9400 mg/kg [Rabbit]. Acute toxicity of the vapor (LC50): 10000 7 hours [Rat].

Chronic Effects on Humans:

CARCINOGENIC EFFECTS: Classified A1 (Confirmed for human.) by ACGIH, 1 (Proven for human.) by IARC. MUTAGENIC EFFECTS: Classified POSSIBLE for human. Mutagenic for mammalian somatic cells. Mutagenic for bacteria and/or yeast. DEVELOPMENTAL TOXICITY: Classified Reproductive system/toxin/female [POSSIBLE]. Causes damage to the following organs: blood, bone marrow, central nervous system (CNS). May cause damage to the following organs: liver, Urinary System.

Other Toxic Effects on Humans:

Very hazardous in case of inhalation. Hazardous in case of skin contact (irritant, permeator), of ingestion.

Special Remarks on Toxicity to Animals: Not available.

Special Remarks on Chronic Effects on Humans:

May cause adverse reproductive effects (female fertility, Embryotoxic and/or foetotoxic in animal) and birth defects. May affect genetic material (mutagenic). May cause cancer (tumorigenic, leukemia)) Human: passes the placental barrier, detected in maternal milk.

Special Remarks on other Toxic Effects on Humans:

Acute Potential Health Effects: Skin: Causes skin irritation. It can be absorbed through intact skin and affect the liver, blood, metabolism, and urinary system. Eyes: Causes eye irritation. Inhalation: Causes respiratory tract and mucous membrane irritation. Can be absorbed through the lungs. May affect behavior/Central and Peripheral nervous systems (somnolence, muscle weakness, general anesthetic, and other symptoms similar to ingestion), gastrointestinal tract (nausea), blood metabolism, urinary system. Ingestion: May be harmful if swallowed. May cause gastrointestinal tract irritation including vomiting. May affect behavior/Central and Peripheral nervous systems (convulsions, seizures, tremor, irritability, initial CNS stimulation followed by depression, loss of coordination, dizziness, headache, weakness, pallor, flushing), respiration (breathlessness and chest constriction), cardiovascular system, (shallow/rapid pulse), and blood.

Section 12: Ecological Information

Ecotoxicity: Not available.

BOD5 and COD: Not available.

Products of Biodegradation:

Possibly hazardous short term degradation products are not likely. However, long term degradation products may arise.

Toxicity of the Products of Biodegradation: The products of degradation are less toxic than the product itself.

Special Remarks on the Products of Biodegradation: Not available.

Section 13: Disposal Considerations

Waste Disposal:

Waste must be disposed of in accordance with federal, state and local environmental control regulations.

Section 14: Transport Information

DOT Classification: CLASS 3: Flammable liquid.

Identification: : Benzene UNNA: 1114 PG: II

Special Provisions for Transport: Not available.

Section 15: Other Regulatory Information

Federal and State Regulations:

California prop. 65: This product contains the following ingredients for which the State of California has found to cause cancer, birth defects or other reproductive harm, which would require a warning under the statute: Benzene California prop. 65 (no significant risk level): Benzene: 0.007 mg/day (value) California prop. 65: This product contains the following ingredients

Appendix A-4. MSDS Information

for which the State of California has found to cause cancer which would require a warning under the statute: Benzene Connecticut carcinogen reporting list.: Benzene Connecticut hazardous material survey.: Benzene Illinois toxic substances disclosure to employee act: Benzene Illinois chemical safety act: Benzene New York release reporting list: Benzene Rhode Island RTK hazardous substances: Benzene Pennsylvania RTK: Benzene Minnesota: Benzene Michigan critical material: Benzene Massachusetts RTK: Benzene Massachusetts spill list: Benzene New Jersey: Benzene New Jersey spill list: Benzene Louisiana spill reporting: Benzene California Director's list of Hazardous Substances: Benzene TSCA 8(b) inventory: Benzene SARA 313 toxic chemical notification and release reporting: Benzene CERCLA: Hazardous substances.: Benzene: 10 lbs. (4.536 kg)

Other Regulations:

OSHA: Hazardous by definition of Hazard Communication Standard (29 CFR 1910.1200). EINECS: This product is on the European Inventory of Existing Commercial Chemical Substances.

Other Classifications:

WHMIS (Canada):

CLASS B-2: Flammable liquid with a flash point lower than 37.8°C (100°F). CLASS D-2A: Material causing other toxic effects (VERY TOXIC).

DSCL (EEC):

R11- Highly flammable. R22- Harmful if swallowed. R38- Irritating to skin. R41- Risk of serious damage to eyes. R45- May cause cancer. R62- Possible risk of impaired fertility. S2- Keep out of the reach of children. S26- In case of contact with eyes, rinse immediately with plenty of water and seek medical advice. S39- Wear eye/face protection. S46- If swallowed, seek medical advice immediately and show this container or label. S53- Avoid exposure - obtain special instructions before use.

HMIS (U.S.A.):

Health Hazard: 2

Fire Hazard: 3

Reactivity: 0

Personal Protection: h

National Fire Protection Association (U.S.A.):

Health: 2

Flammability: 3

Reactivity: 0

Specific hazard:

Protective Equipment:

Gloves. Lab coat. Vapor respirator. Be sure to use an approved/certified respirator or equivalent. Wear appropriate respirator when ventilation is inadequate. Splash goggles.

Section 16: Other Information

Other Special Considerations: Not available.

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MATERIAL SAFETY DATA SHEET

	SECTION	1.	PRODUCT		
IDENTIFICATION PRODU	JCT NAME:	Carbon	Dioxide		
CHEMICAL NAME:	Carbon Dioxi	de		FORMULA:	CO_2
SYNONYMS:	Carbonic Anh	nydride, Carl	ponic Acid Gas,	Carbon Anhyd	ride
MANUFACTURER:	Air Products	and Chemi	cals, Inc.		
	7201 Hamilton	Boulevard			
	Allentown, PA	18195-150	1		
PRODUCT INFORMATIO	N: 1 - 800 - 752	- 1597			
MSDS NUMBER: 1005				REVISION: 5	
REVISION DATE: March	า 1993				REVIEW
DATE: March					
1994					

COMPOSITION / INFORMATION ON SECTION 2.

INGREDIENTS CONCENTRATION: Carbon dioxide is sold as pure product

> 99%.

BROD

CAS NUMBER: 124-38-9

EXPOSURE LIMITS:

OSHA: PEL-TWA = 5000 ppm **ACGIH:** TLV-TWA = 5000 ppm **NIOSH:** None established

SECTION 3. HAZARDS IDENTIFICATION

EMERGENCY OVERVIEW

Carbon dioxide is a nonflammable liquefied compressed gas packaged in cylinders under its own vapor pressure of 838 psig at 70 F (21.1 C). High concentrations can cause rapid suffocation and can also increase respiration and heart rate. Contact with liquid may cause frostbite. Avoid breathing gas. Self contained breathing apparatus (SCBA) may be required by rescue workers.

EMERGENCY TELEPHONE

NUMBERS

800 - 523 - 9374 Continental U.S., Canada, or Puerto Rico

610 - 481 - 7711 other locations

POTENTIAL HEALTH EFFECTS:

INHALATION: Carbon dioxide is an asphyxiant. Concentrations of 10% or more can produce unconsciousness or death.

EYE CONTACT: Contact with liquid or cold vapor can cause freezing of tissue.

SKIN CONTACT: Contact with liquid or cold vapor can cause frostbite.

EXPOSURE INFORMATION:

ROUTE OF ENTRY:

Inhalation

TARGET ORGANS: Central nervous system

EFFECT: Asphyxiation (suffocation). Overexposure may cause damage to retinal ganglion cells and central nervous system.

MSDS # 1005 Pub Carbon Dioxide

Page 1 of 6 310-408 **SYMPTOMS:** Headache, sweating, rapid breathing, increased heartbeat, shortness of breath, dizziness, mental depression, visual disturbances, and shaking. **CHRONIC EFFECTS:** None established.

MEDICAL CONDITIONS AGGRAVATED BY OVEREXPOSURE: None

CARCINOGENICITY: Carbon dioxide is not listed by NTP, OSHA or IARC.

SECTION 4. FIRST

AID

INHALATION: Persons suffering from overexposure should be moved to fresh air. If victim is not breathing, administer artificial respiration. If breathing is difficult, administer oxygen. Obtain prompt medical attention.

EYE CONTACT: Contact with liquid or cold vapor can cause freezing of tissue. Gently flush eyes with lukewarm water. Obtain medical attention immediately.

SKIN CONTACT: Contact with liquid or cold vapor can cause frostbite. Immediately warm affected area with lukewarm water not to exceed 107 F.

NOTES TO PHYSICIAN: There is no specific antidote. Treatment for overexposure should be directed at the control of symptoms and the clinical condition.

SECTION 5. FIRE AND EXPLOSION

FLASH POINT: Not Applicable Nonflammable

AUTOIGNITION: Nonflammable FLAMMABLE LIMITS:

EXTINGUISHING MEDIA: Carbon dioxide is nonflammable and does not support combustion. Carbon dioxide is an extinguishing agent for class B and C fires. Use extinguishing media appropriate for the surrounding fire.

HAZARDOUS COMBUSTION PRODUCTS: None known.

FIRE FIGHTING PROCEDURES: Evacuate personnel from danger area. Carbon dioxide is nonflammable. If possible, without risk, remove cylinders from fire area or cool with water. Self contained breathing apparatus (SCBA) may be required for rescue workers.

UNUSUAL HAZARDS: Upon exposure to intense heat or flame, cylinder will vent rapidly and or rupture violently. Most cylinders are designed to vent contents when exposed to elevated temperatures. Pressure in a container can build up due to heat and it may rupture if pressure relief devices should fail to function.

SECTION 6. ACCIDENTAL RELEASE MEASURES

Evacuate all personnel from affected area. Increase ventilation to release area and monitor oxygen level. Use appropriate protective equipment (SCBA). If leak is from cylinder or cylinder valve call the Air Products emergency telephone number. If leak is in user's system close cylinder valve and vent pressure before attempting repairs.

SECTION 7. HANDLING AND STORAGE

STORAGE: Cylinders should be stored upright in a well-ventilated, secure area, protected from the weather. Storage area temperatures should not exceed 125 F (52 C). Storage should be away from heavily traveled areas and emergency exits. Avoid areas where salt or other corrosive materials are present. Valve protection caps and valve outlet seals should remain on cylinders not connected for use. Separate full from empty cylinders. Avoid excessive inventory and storage

time. Use a first-in first-out system. Keep good inventory records.

HANDLING:Use a suitable hand truck for cylinder movement.Never attempt to lift acylinder by its valve protection valve cap.Never apply flame or localized heat directly toanypartofthecylinder.Donot

allow any part of the cylinder to exceed 125 °F (52 °C). High temperature may cause damage to cylinder and/or premature failure of pressure relief device which will result in venting of cylinder contents. If user experiences any difficulty operating cylinder valve discontinue use and contact supplier. Never insert an object (e.g., wrench, screwdriver, pry bar, etc.) into valve cap openings. Doing so may damage valve causing a leak to occur. Use an adjustable strap wrench to remove over-tight or rusted caps.

Only the proper CGA connections should be used, never use adapters. Use piping and equipment adequately designed to withstand pressures to be encountered. If liquid product is being used ensure steps have been taken to prevent entrapment of liquid in closed systems. The use of pressure relief devices may be necessary. Use a check valve or other protective apparatus in any line or piping from the cylinder to prevent reverse flow.

Carbon dioxide is compatible with all common materials of construction. Pressure requirements should be considered when selecting materials and designing systems.

Use a "FULL", "IN USE", and "EMPTY" tag system on cylinders. This will reduce the chances of inadvertently connecting or operating the wrong cylinder.

SPECIAL REQUIREMENTS: Always store and handle compressed gases in accordance with Compressed Gas Association, Inc. (ph. 703-979-0900) pamphlet CGA P-1, *Safe Handling of Compressed Gases in Containers.* Local regulations may require specific equipment for storage or use.

-CAUTION: Compressed gas cylinders shall not be refilled except by qualified producers of compressed gases. Shipment of a compressed gas cylinder which has not been filled by the owner or with the owner's written consent is a violation of federal law.

SECTION 8. PERSONAL PROTECTION / EXPOSURE CONTROL

ENGINEERING CONTROLS: Provide ventilation and/or local exhaust to prevent accumulation of carbon dioxide concentrations above 5000 ppm.

RESPIRATORY PROTECTION:

Emergency Use: Self contained breathing apparatus (SCBA) or positive pressure airline with mask and escape pack are to be used in oxygen deficient atmosphere. Air purifying respirators will not provide protection.

EYE PROTECTION: Safety glasses are recommended when handling, connecting, or disconnecting cylinders, and when pressurizing systems

OTHER PROTECTIVE EQUIPMENT: Safety shoes and leather work gloves when handling cylinders.

SECTION 9. PHYSICAL AND CHEMICAL PROPERTIES

APPEARANCE, ODOR AND STATE: Colorless and odorless. A slightly acid gas. It is felt by some to have a slight pungent odor and biting taste.

MOLECULAR WEIGHT: 44.01

GAS DENSITY (at 70 F (21.1 C) and 1 atm): 0.1144 lb/ft³ (1.832 kg/m³)

VAPOR PRESSURE (at 70 F (21.1 C)): 838 psig

 SPECIFIC GRAVITY (Air =1): 1.522

 SPECIFIC VOLUME (at 70 F (21.1 C) and 1 atm): 8.74 ft³/lb (0.5457 m³/kg)

 BOILING POINT: -109.3 F (-78.5 C)

 TRIPLE POINT (At 60.4 psig): -69.9 F (-56.6 C)

 SOLUBILITY IN WATER (Vol./Vol. at 68 F (20 C)): 0.90

SECTION10.STABILITYANDREACTIVITYSTABILITY:
StableStableNoneIncompatibilityIncompatibilit

SECTION 11. TOXICOLOGICAL

INFORMATION

Carbon dioxide is an asphyxiant. It initially stimulates respiration and then causes respiratory depression. High concentrations result in narcosis. Symptoms in humans are as follows:

CONCENTRATION	EFFECT
1%	Slight increase in breathing rate
2%	Breathing rate increases to 50% above normal. Prolonged exposure can cause headache and tiredness.
3%	Breathing increases to twice the normal rate and becomes
labored.	
	Weak narcotic effect. Impaired hearing, headache, increase in blood pressure and pulse rate.
4-5%	Breathing increases to approximately four times the normal
	rate, symptoms of intoxication become evident and slight choking may be felt.
5-10%	Characteristic sharp odor noticeable. Very labored breathing,
	headache, visual impairment and ringing in the ears. Judgment <u>may be impaired, followed within minutes by loss of</u>
	consciousness.
50-100%	Unconsciousness occurs more rapidly above 10% level. Prolonged exposure to high concentrations may eventually result in death from asphyxiation.

SECTION 12. ECOLOGICAL

INFORMATION

No adverse ecological effects are expected. No adverse ecological effects are expected. Carbon dioxide does not contain any Class I or Class II ozone depleting chemicals. Carbon dioxide is not listed as a marine pollutant by DOT (49 CFR 171).

SECTION 13.

DISPOSAL

UNUSED PRODUCT / EMPTY CYLINDER: Return cylinder and unused product to supplier. Do not attempt to dispose of unused product. Ensure cylinder valve is properly closed, valve outlet cap has been reinstalled, and valve protection cap is secured before shipping cylinder.

WASTE DISPOSAL METHODS: For emergency disposal, secure the cylinder and slowly discharge gas to the atmosphere in a well ventilated area or outdoors. Small amounts may be disposed of by reacting with a mild base.

SECTION 14. TRANSPORT INFORMATION

DOTSHIPPINGNAME:CarbondioxideHAZARDCLASS:2.2(Nonflammable Gas)IDENTIFICATIONNUMBER:UN1013

PRODUCT RQ: None

SHIPPING LABEL(s): Nonflammable gas

PLACARD (when required): Nonflammable gas

SPECIAL SHIPPING INFORMATION: Cylinders should be transported in a secure upright position in a well ventilated truck. Never transport in passenger compartment of a vehicle.

SECTION 15. REGULATORY

INFORMATION U.S. FEDERAL REGULATIONS: ENVIRONMENTAL PROTECTION AGENCY (EPA):

CERCLA: Comprehensive Environmental Response, Compensation, and Liability Act of 1980 requires notification to the National Response Center of a release of quantities of hazardous substances equal to or greater than the reportable quantities (RQ's) in 40 CFR 302.4.

CERCLA Reportable Quantity: None.

SARA TITLE III: Superfund Amendment and Reauthorization Act of 1986

SECTION 302/304: Requires emergency planning on threshold planning quantities (TPQ) and release reporting based on reportable quantities (RQ) of EPA's extremely hazardous substances (40 CFR 355).

Extremely Hazardous Substances: None

Threshold Planning Quantity (TPQ): None

SECTIONS 311/312: Require submission of material safety data sheets (MSDSs) and chemical inventory reporting with identification of EPA defined hazard classes. The hazard classes for this product are:

IMMEDIATE HEALTH: Yes PRESSURE: Yes DELAYED HEALTH: No

REACTIVITY: No FLAMMABLE: No

SECTION 313: Requires submission of annual reports of release of toxic chemicals that appear in 40 CFR 372.

Carbon dioxide does not require reporting under Section 313

40 CFR Part 68 - Risk Management for Chemical Accident Release Prevention: Requires the development and implementation of risk management programs at facilities that manufacture, use, store, or otherwise handle regulated substances in quantities that exceed specified thresholds.

Carbon dioxide is not listed as a regulated substance.

TSCA - TOXIC SUBSTANCES CONTROL ACT: Carbon dioxide is listed on the TSCA inventory.

OSHA - OCCUPATIONAL SAFETY AND HEALTH ADMINISTRATION:

29 CFR 1910.119: Process Safety Management of Highly Hazardous Chemicals. Requires facilities to develop a process safety management program based on Threshold Quantities (TQ) of highly hazardous chemicals.

Carbon dioxide is not listed in Appendix A as a highly hazardous chemical.

STATE REGULATIONS:

CALIFORNIA:

Proposition 65: This product does NOT contain any listed substances which the

State of California requires warning under this statute. SCAQMD Rule: VOC = Not applicable

	SECTION	16. 0	OTHER	INFORMATION
HAZARD RATINGS: NFPA RATINGS:				
		HMIS RATINGS:		
HEALTH:	1	HEALTH:	0	
FLAMMABILITY	: 0	FLAMMABILIT	-Y: 0	
REACTIVIT	0	REACTIVITY:	0	
Y: SPECIAL: *Compressed G to designate sim		recommendation		
Revision information: New	format with a	dditional sections		

added. Reformatted September 1998.



Safety Data Sheet **CHLORINE**

	OHLO	: 03-A Page 1 of 5
I. PRODUCT	AND COMPANY IDENTIFICATION	Precautionary statements:
Product Name	Liquid Chlorine	Prevention P220: Keep/Store away from clothing/combustible materials.
Recommended use	Used in water treatment and disinfection;	P244: Keep reduction valves free from grease and oil.
of the chemical and	as bleaching agent, particularly for	P261: Avoid breathing dust/fume/gas/mist/vapours/spray.
restrictions on use	paper and textiles; in the manufacture	P264: Wash thoroughly after handling.
	of bleaching powder and chemicals such as PVC, chlorinated hydrocarbons, ethylene glycol, glycerine and tetraethyl lead.	P280: Wear protective gloves/protective clothing/eye protection/face protection.P284: Wear respiratory protection.P271: Use only outdoors or in a well- ventilated area
-	Chlorine is 2.5 times heavier than air. It	P273: Avoid release to the environment.
	is not an explosive or flammable gas, but reacts violently with oils, solvents,	Response: P370+P376: In case of fire: Stop leak if safe to do so
	grease, ammonia, acetone, etc.	P304+P340: IF INHALED: Remove victim to fresh air and keep
Manufacturer	Mabuhay Vinyl Corporation 3 rd Floor Philamlife, 126 L.P. Leviste St. Salcedo Village, Makati City For Assistance Call : (02) 817-8971 to 76 loc 214; Direct line (02) 817-1830	at rest in a position comfortable for breathing. P301 + P330 + P331: IF SWALLOWED: Rinse mouth. Do NOT induce vomiting P305+P351+P338: IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.
lligan Plant	Assumption Heights, Iligan City Tel: (063) 221-9466, 221-1190	P337+P313: If eye irritation persists: Get medical advice/ attention.
Mabuhay Premium Bleach Plant	LTAI, Brgy. Biñan, Biñan, Laguna Tel: (049) 541-1923	P303 + P361 + P353: IF ON SKIN (or hair): Remove/Take off immediately all contaminated clothing. Rinse skin
Batangas Depot	BBTI Compound, Bauan, Batangas Tel: (043) 980-5869	with water/shower P363: Wash contaminated clothing before reuse P310: Immediately a POISON CENTER or dector/physician
Cebu Depot	Ceniza St., Ouano, Mandaue City Tel: (032) 344-5259, 345-0639	P310: Immediately call a POISON CENTER or doctor/physician. P312: Call a POISON CENTER or doctor/physician if you feel unwell.
Davao Depot	Bunawan, Davao City Tel: (082) 236-0015	P320: Specific treatment is urgent (see MSDS). P391 Collect spillage.
		Storage:

II. HAZARDS IDENTIFICATION

Symbols



P405: Store locked up.

Disposal: P501: Dispose of contents/container in accordance with applicable local, regional, national, and/or international regulations

III. COMPOSITION / INFORMATION ON INGREDIENTS

P403: Store in a well ventilated place.

P233: Keep container tightly closed.

P410: Protect from sunlight



Safety Data Sheet CHLORINE

IV. FIRST AID MEASURES

Description of first-aid measures: In all instances, seek immediate medical attention. Show this safety data sheet to the physician in attendance.

In case of frostbite place the frostbitten part in warm water. Do not use hot water! If warm water is not available wrap the affected parts gently in blankets. Encourage victim to gently exercise the affected part while being warmed. Seek immediate medical attention.

Inhalation: Remove to fresh air. Give artificial respiration if not breathing, preferably mouth-to-mouth . If breathing is difficult, administer oxygen. Keep the affected person warm at rest. In mild cases, give milk to relieve throat irritation.

Ingestion: Not a likely route of exposure.

- Skin contact: Wash with plenty of soap and water while removing contaminated clothing and shoes. Get medical attention. Wash clothing before reuse. Destroy contaminated shoes. Do not apply greases unless ordered by a physician.
- Eye contact: Immediately flush eyes with a directed stream of water for at least 15 minutes, forcibly holding eyelids apart to ensure complete irrigation of all eye and lid tissues. Washing eyes within several seconds is essential to achieve maximum effectiveness. Do not attempt chemical neutralization of any kind.

Most important symptoms/effects, both acute and delayed Toxic and irritating, with Inhalation as the major potential route of exposure. May cause severe irritation of mucous membranes of the nose, throat and respiratory tract followed by severe coughing, burning, chest pain, vomiting, headache, anxiety and feeling of suffocation. Severe breathing difficulties may occur which may be delayed in onset. Severe exposure may lead to chemical pneumonitis and pulmonary edema and may be fatal. Repeated or prolonged exposure may result in reduced pulmonary capacity and dental erosion.

Skin contact with liquid chlorine may cause serious burns, blistering and tissue destruction. Chlorine vapors can cause

V. FIRE FIGHTING MEASURES

irritation, burning and blisters. Contact with rapidly expanding gas poses a frostbite hazard.

Indication of any immediate medical attention and special treatment needed: No known antidote. Treatment for inhalation is symptomatic and supportive. Keep patient at rest until respiratory symptoms subside. Sedation for apprehension or restlessness may be considered as well as diuretics and antibiotics to alleviate edema and protect against secondary infection. Administer oxygen under exhalation pressure not exceeding 4 cm water for 15 minutes each hour until symptoms subside (except in presence of impending or existing cardiovascular failure). Steroid therapy, if given early, has been reported effective in preventing pulmonary edema. It is recommended that anyone exposed to chlorine gas by inhalation obtain a chest x-ray to check for pulmonary edema.

First Aid Facilities: Eye wash station, safety shower and normal washroom facilities.

Appendix A-4. MSDS Information

Extinguishing media

- Suitable extinguishing media: Water spray, fog or foam. For large fires, flood with fine water spray. Use water to keep fire - exposed containers cool and continue until well after fire is out.
- Unsuitable extinguishing media: Do not use carbon dioxide or halogenated extinguishing agents.
- Special hazards arising from the substance or mixture: Although non-flammable, chlorine is a strong oxidizer and may react to cause fire and/or explosion upon contact with turpentine, ether, ammonia, hydrocarbons, certain metal hydrides, carbides, nitrides, oxides, sulfides, easily oxidized phosphides, materials, organic materials or other flammables. It forms corrosive Hydrogen Chloride on contact with water. Chlorine gas is heavier than air and will collect in low-lying areas.
- Special protective actions for firefighters: Selfcontained breathing equipment, eye protection and full protective clothing is required. Move container from fire area if it can be done without risk. Stay away from the ends of tanks. Keep unnecessary people away, isolate hazard area and deny entry. Cool containers with water spray until well after the fire is out. Apply water from a protected location or from a safe distance. Avoid inhalation of

material or combustion by-products. Stay upwind and keep out of low areas. Evacuation radius: 800 meters (1/2 mile). Do not allow contaminated extinguishing water to enter the soil, groundwater or surface waters.

VI. ACCIDENTAL RELEASE MEASURES

- Personal precautions, protective equipment and emergency procedures: Evacuate all unprotected personnel. Put on protective equipment (see Section 8). Avoid direct contact with skin, eyes and clothing. Ensure adequate ventilation/exhaust extraction. Avoid low-lying areas. Work upwind if possible.
- Environmental precautions: Avoid entry of product into drains, sewers, surface/ground water system or soil. Drains for storage or use areas should have retention basins for pH adjustments and dilution of spills before discharge or disposal of material.
- Methods and material for containment and cleaning up: When possible, move leaking or damaged cylinders outdoors or to an isolated location. If source of leak is a cylinder and the leak cannot be stopped in place, remove the leaking cylinder to a safe place in the open air and repair the leak or allow the cylinder to empty through a reducing agent such as caustic soda, soda ash, or hydrated lime solutions. Turn leaking cylinder with the leak up to prevent escape of gas in liquid state. Chlorine vapors are heavier than air, and pockets of chlorine are likely to be trapped in low lying areas. Use water fog to dampen a chlorine cloud and reduce vapours. Do not spray water directly on the leak or chlorine container. Liquid or solid residues must be disposed of in accordance with all applicable regulations.



Safety Data Sheet **CHLORINE**

Appendix A-4. MSDS Information MVC-MSDS-C-001P Issue Date : 11/15/13 Rev. Code :03

Page 3 of 5

VII. HANDLING AND STORAGE

- Precautions for safe handling: Use protective equipment (see Section 8). Provide adequate ventilation. Avoid inhalation of vapors and skin and eye contact. Change contaminated or soaked clothing immediately. Wash hands after handling.
- Provide special training to workers handling chlorine. Regularly test and inspect piping and containment used for chlorine service. Liquid levels should be less than 85% of tank or cylinder capacity.
- Do not drag, slide or roll cylinders. Use a suitable hand truck for cylinders movement. Protect cylinders and containers from physical damage. Keep containers tightly closed when not in use. Chlorine emergency equipment should be available near the point of use.
- Keep away from foodstuffs, drinks and tobacco.
- Keep away from incompatible products.
- Conditions for safe storage, including any incompatibilities Store chlorine containers and cylinders below 45°C in cool, dry, well ventilated areas of non-combustible construction away from sunlight, precipitation, heavily trafficked areas and emergency exit. Cylinders should be stored upright and firmly secured to prevent falling or being knocked over. Full and empty cylinders should be segregated. Full cylinders should not be stored for more than six months. Liquid levels should be less than 85 % of container or cylinder capacity.
 - Non suitable packaging material: Acetal, aluminum, brass,
 - Bronze, carbon steel, cast iron, chrome, CPVC, epoxy, LDPE, natural rubber, neoprene, nitrile, nylon, polyether-
 - ether ketone(PEEK) , polypropylene, polyurethane, PPS,

Keep away from heat, sparks, open flames and incompatible substances(see Section X).

VIII. EXPOSURE CONTROLS AND PROTECTION

Control parameters

ACGIH	0.5 ppm TWA 1 ppm STEL
Europe	0.5 ppm STEL; 1.5 mg/m3 STEL
OSHA (Final)	1 ppm Ceiling; 3 mg/m3 Ceiling
OSHA (Vacated)	0.5 ppm TWA; 1.5 mg/m3 TWA

	т ррп
NIOSH	0.5 pp
	Ceiling
Philippines	3 mg/r

1 ppm STEL; 3 mg/m3 STEL om Ceiling (15 min); 1.45 mg/m3 g (15 min) m3 (TWA) OEL 1 ppm (TWA) OEL

Appropriate engineering controls: A system of local and / or general exhaust is recommended to keep employee exposure as low as possible. Use enclosed, isolated processing and handling whenever possible.

PVC, silicone, titanium

Personal protective equipment

Maintain eye wash fountain and guick-drench facilities in

work area. Final choice of appropriate protection will vary

according to methods handling, engineering of controls and risk assessments undertaken.

- Respiratory protection NIOSH-approved full- or half face piece (with goggles) respiratory protective equipment Up to 5 ppm:
 - cartridge chemical respirator Any with cartridge(s) providing protection against the compound of concern

Any supplied-air

respirator Up to 10 ppm:

supplied-air Any respirator operated in а continuous-flow mode

- Any powered, air-purifying respirator with cartridge(s) providing protection against the compound of concern
- Any chemical cartridge respirator with а full facepiece and cartridge(s) providing protection against the compound of concern
- Any air-purifying, full-facepiece respirator (gas mask) with a chin-style, front- or back-mounted canister providing protection against the compound of concern
- Any self-contained breathing apparatus with а full facepiece
- Any supplied-air respirator with a full facepiece

Appendix A-4. MSDS Information

Emergency or planned entry into unknown concentrations or IDLH conditions:

Any self-contained breathing apparatus that has a full

facepiece and is operated in a pressure-demand or other positive-pressure mode

Any supplied-air respirator that has a full facepiece and is operated in a pressure-demand or other positivepressure mode in combination with an auxiliary selfcontained positive-pressure breathing apparatus

Escape:

- Any air-purifying, full-facepiece respirator (gas mask)
 - with a chin-style, front- or back-mounted canister
 - providing protection against the compound of concern
- Any appropriate escape-type, self-contained breathing apparatus

Hand protection: PVC, rubber or neoprene gloves

- Eye / face protection: splash-proof safety goggles with side shields or face shield
- Skin protection: Appropriate impermeable protective clothing(made of Viton, butyl rubber, Teflon, chlorinated polyethylene material) to protect against possible skin contact. When responding to accidental release of unknown concentrations, wear one-piece, total encapsulating suit of Butyl coated nylon or equivalent.



Safety Data Sheet CHLORINE

X. PHYSICAL AND CH	EMICAL PROPERTIES
Appearance	Greenish-yellow to amber gas
Odor	Irritating, pungent odor
Odor threshold	0.2ppm
oH	Not applicable
Freezing point	-101°C
Boiling point /range	34°C
Flash point	Not applicable
Evaporation rate	Not applicable
Flammability	Non_flammable
Flammability/explosive limits	Not applicable
Vapor pressure	5168 mm Hg at 21°C
Vapor density	2.49 (air = 1.0)
Relative density(water = 1) Solubility(in water) Partition coefficient: n-octanol/water Auto-ignition temperature Decomposition temperature Viscosity	1.41 at 20°C(liquid) 14.6gpl at 0°C; 6.9gpl at 20°C Not available Not applicable Not available 14 Pa.s at 20°C

X. STABILITY AND REACTIVITY

Reactivity: May react violently with combustible materials. Reacts with water to form corrosive acids. May react violently with alkalis. With water causes rapid corrosion of some metals. May react violently with reducing agents. Violently oxidises organic material.

Stability: Stable under recommended storage conditions.

- Possibility of hazardous reactions or polymerization: Will not polymerize. Reacts violently with a variety of substances over a broad range of conditions including reducing agents and combustible materials.
- Conditions to avoid: Heat, sparks, sunlight, moisture and incompatible substances.

XI. TOXICOLOGICAL INFORMATION

Routes of exposure: inhalation, ingestion, skin and eye contact

- Symptoms related to the physical, chemical and toxicological characteristics: causes severe irritation of the eyes and respiratory tract with eye injury, restlessness, shortness of breath, cough, choking sensation, sneezing, running nose, chest pain, dizziness, headache, nausea, cyanosis (lack of oxygen in the blood) and respiratory failure. Following respiratory tract injury, onset of severe breathing difficulties, including bronchitis, lung edema (accumulation of fluid in the lungs) and pneumonia, may be delayed and life threatening.
- Delayed and immediate effects and also chronic effects from short and long term exposure: High concentrations of chlorine over a short period of time may aggravate preexisting heart conditions, and cause congestive heart failure. At high concentrations, chlorine gas irritates the skin and can produce sensations of burning and pricking of the skin, with inflammation and blister formation. Exposure to concentrations as low as 5-10 ppm is reported to cause severe irritation of the eyes, nose and respiratory tract which is intolerable after a few minutes.
- Numerical measures of toxicity The hazards via inhalation at different concentrations is reported to be as follows:

0.2-0.5 ppm	No toxic, long term effect
1-3 ppm 5-8 ppm 30 ppm 34-51 ppm 40-60 ppm	Definite odor: irritation of eyes and nose Throat, eye, and mucous membrane irritation Intense coughing fits Lethal in 1 to 1.5 hours exposure
100 ppm	Exposure for 30-60 minutes without effective respiration may cause bronchitis, pulmonary edema or bronchopneumonia
	May be lethal after 50 minutes of exposure (estimated)
430 ppm	Lowest concentration known to cause lethality after 30 minutes of exposure
1000 ppm	May be fatal with a few deep breaths

- Incompatible materials: Hydrocarbons, combustible materials, bases, acids, metals, metal salts, carbides, oxides, phosphides, nitrides, sulfides, reducing agents, oxidizing materials, halogens, halo carbons, amines, ammonia, arsenic, calcium, iodine, ethers, fluorine
- Hazardous decomposition products: Does not decompose but reacts violently to form hydrochloric acid and other potentially toxic and/ or corrosive substances.

XII. ECOLOGICAL INFORMATION

Ecotoxicity

- FISH TOXICITY: 390 ug/L 96 hour(s) LC50 (Mortality) Orangethroat darter (Etheostoma spectabile)
- INVERTEBRATE TOXICITY: 637.5 ug/L 1 hour(s) LC50 (Mortality) Pacific oyster (Crassostrea gigas)
- ALGAL TOXICITY: 50-1000 ug/L 23 hour(s) (Population) Algae,phytoplankton,algal mat (Algae)

PHYTOTOXICITY: 20 ug/L 96 day(s) (Growth) Water-milfoil (Myriophyllum spicatum)



Safety Data Sheet CHLORINE

Appendix A-4. MSDS Information

MVC-MSDS-C-001P Issue Date : 11/15/13 Rev. Code : 03

Page 5 of 5

Persistence and degradability:

Biodegradation (In water): Half-life: 1.3 to 5 hours.

Photodegradation (In air) : Half-life: 10 minutes, lifetime: 14 minutes.

Bioaccumulative potential : An accumulation in aquatic organisms is not to be expected

Mobility

- In water, chlorine is transformed to free available chlorine (gaseous chlorine), hypochlorous acid and hypochlorite ions, whose relative amounts depend on the pH and other physicochemical properties of the water. At environmental pH, only hypochlorous acid and hypochlorite will be present.
- In the atmosphere, Cl_2 will degrade during daylight, with halflives ranging from minutes to several hours, depending on latitude, season, and time of day.
- In soil, the high water solubility of chlorine can lead to a high mobility in soil, although chlorine as vapour or as aqueous solution is normally irreversibly bound to soil organics within the first few millimeters or centimeters of the soil surface.

XIII. DISPOSAL CONSIDERATIONS

Dispose of in accordance with all Government and Local regulations. Container is returnable and must be properly identified with return tag and returned as promptly as possible to supplier, in accordance with all applicable DOT regulations. All valves must be closed tight and closures or caps secured. It is illegal to ship a leaking Chlorine container.

Chlorine gas will disperse to the atmosphere leaving no residue. When possible, move leaking container to an isolated area. Position to release gas, not liquid. Absorb in alkaline solution of caustic soda, soda ash, or hydrated lime.

XIV. TRANSPORT INFORMATIC

UN Number UN Proper Shipping Name Transport hazard class

1017 CHLORINE 2.3(Toxic Gas, Corrosive, Oxidizing) Not applicable Environmental Hazards Marine Pollutant : Yes

Environmentally Hazardous : Yes No

IMDG ADR/RID IATA

Label

Packing group

Special precautions

Transport in open ventilated vehicle, cylinders upright and secured, drum placed lengthwise in the truck tray, with the valve end facing away from the vehicle. Do not transport in confined spaces like refrigerated compartments of vehicles, truck cabs or in passenger compartments. Ensure vehicle driver is aware of the potential hazards of the load and knows what to do in the event of an accident or an emergency.

XV. REGULATORY INFORMATION

U.S. Regulations:

This material contains one or more of the following chemicals required to be identified under SARA Section 302/304 (40 CFR 355 Appendix A), SARA Section 313 (40 CFR 372.65), CERCLA (40 CFR 302.4), TSCA 12(b), and/or require an OSHA process safety plan.

(7782-50)-5)		
100	lb		
10 lb E	PCF	RA RQ	
1.0	%	de	minimis
CERCLA:		10 lb	final RQ;
Q OSHA:		1500 lb	TQ
	` 100 10 lb E 1.0 CERCLA:	10 lb EPCF 1.0 % CERCLA:	100 lb 10 lb EPCRA RQ 1.0 % de CERCLA: 10 lb

SARA Title III Sections 311/312 Hazardous Categories (40

CFR 370.21):

Acute: Yes	Chronic: No	Fire:
No Reactive: No	Sudden release: No)

National Inventories Australian Chemical Inventory(AICS) : Listed Canadian Chemical Inventory(DSL) : Listed China Chemical Inventory(IECS) : Listed European Union Inventory(EINECS) : 231-959-5 Japan Chemical Inventory(ENCS): Not listed Korean Chemical Inventory(KECL) : KE-05486 New Zealand Chemical Inventory(NZIOC) : Listed

Philippines - Priority Chemical List(PICCS) : Listed U.S. Inventory (TSCA) : Listed

XVI. OTHER INFORMATION

The information herein is presented in good faith and believed to be correct as of the date of issue. However, no warranty, expressed or implied, is made by Mabuhay Vinyl Corporation regarding the product's merchantability, fitness for a particular purpose, performance, safety or stability. This information is not intended to be all-inclusive as to the manner and conditions of use, handling, storage, disposal and other factors that may involve other or additional legal, environmental, safety or performance considerations, and Mabuhay Vinyl Corporation assumes no liability whatsoever for the use of or reliance upon this information. No suggestions for use are intended as, and nothing herein shall be construed as, a recommendation to infringe any existing patents or to violate any existing laws or regulations.

THE LINDE GR	ROUP			Linde
	Saf	ety data she Ethane	et	
Creation date : Revision date :	28.01.2005 04.01.2011	Version : 2.0	DE / E	SDS No. : 051a page 1 / 3
1 IDENTIFICATION OF THE COMPANY	OF THE SUBSTANCE/PREPARATIO		onary Statement Disposal	
Product name			None.	
Ethane EC No (from EINECS)) [,] 200-814-8	3 COM	POSITION/INFORMATION C	N INGREDIENTS
CAS No: 74-84-0	. 200 011 0	Substan	ce/Preparation: Substance.	
Index-Nr. 601-002-00-			ents/Impurities	
Chemical formula C2 REACH Registration		Ethane CAS No:	. 74-84-0	
Not available.	indition.		r.: 601-002-00-X	
Known uses		EC No (f	rom EINECS): 200-814-8	
Not known. Company identificati	ion	REACH Not avail	Registration number: able	
• •	Division, Seitnerstraße 70, D-82049 Pull		no other components or impl	urities which will influence th
E-Mail Address Info@ Emergency phone n	ĝde.linde-gas.com umbers (24h): 089-7446-0		ation of the product.	
2 HAZARDS IDENT	TIFICATION	Inhalatio		
Classification of the	substance or mixture		concentrations may cause a	
Classification acc	to Regulation (EC) No 1272/20		oss of mobility/consciousness tion. In low concentrations	
(CLP/GHS)			ns may include dizziness, head	
	gas) - Contains gas under pressure; ma	ay ordinatio	n. Remove victim to uncont	taminated area wearing se
explode if heated. Flam. Gas 1 - Extreme	oly flammable das		d breathing apparatus. Keep v opply artificial respiration if brea	
riam. Gas i - Extreme	ery naminable gas.		contact	atiling stopped.
	Directive 67/548/EEC & 1999/45/EC		d spillage - flush with water fo	or at least 15 minutes Obtai
F+; R12 Extremely flammable.		medical a	assistance.	
	and the environment		is not considered a potential	route of exposure.
Liquefied gas.				
Contact with liquid ma Label Elements	y cause cold burns/frost bite.		FIGHTING MEASURES	
			hazards e to fire may cause containers	to rupture/explode
- Labelling Pictogran	ns		us combustion products	to rupturo oxprodo.
			ete combustion may form carbo	on monoxide.
< 3% > < -			extinguishing media n extinguishants can be used.	
			methods	
	×		le, stop flow of product. Move om a protected position. Do	
- Signal word		flame u	nless absolutely necessary.	Spontaneous/explosive re
I	Danger		nay occur. Extinguish any othe protective equipment for fire	
- Hazard Statements	3		ed space use self-contained b	
H280	Contains gas under pressure; may			BF0
H220	explode if heated. Extremely flammable gas.		IDENTAL RELEASE MEASU	KED
11220	Extremely numinable gas.		I precautions If-contained breathing apparat	us when entering area unles
- Precautionary State	ements	atmosph	ere is proved to be safe. Eva	cuate area. Ensure adequate
Precautionary Stater	ment Prevention		ation. Eliminate ignition source mental precautions	es.
P210	Keep away from heat/sparks/open	Try to st	op release. Prevent from ente	ering sewers, basements an
	flames/hot surfaces No smoking		or any place where its accum	ulation can be dangerous.
Precautionary Stater	nent Reaction		o methods area. Keep area evacuated a	and free from ignition source
P377	Leaking gas fire: Do not extinguish	n, until any	spilled liquid has evaporated.	
P381	unless leakcan be stopped safely. Eliminate all ignition sources if saf			
F 301	So.		DLING AND STORAGE	
		Handling		
Precautionary Stater		Ensure 4	equipment is adequately early	hed. Suck back of water into

THE LINDE GROUP



Safety data sheet Ethane

Creation date : Revision date :	28.01.2005 04.01.2011	Versio	n : 2.0	DE / E	SDS No. : 05′ page 2 / 3	1a
properly specified equip supply pressure and te doubt. Keep away discharges). Refer to su Storage Secure cylinders to pre gases and other oxidan	allow backfeed into the co oment which is suitable fo imperature. Contact your from ignition sources pplier's handling instruction event them falling. Segreg ts in store. Keep container Observe "Technische Re	r this product, its gas supplier if in (including static is. gate from oxidant below 50°C in a	14 TRANSPORT INFO ADR/RID Class UN number and prope UN 1035 Ethane UN 1035 Ethane Labels Packing Instruction	2 (rshipping nar	Classification Code ne Hazard number	2F 23
(TRG) 280 Ziffer 5" 8 EXPOSURE CONTI Personal protection Ensure adequate ventila	ROLS/PERSONAL PROTE ation. Do not smoke while and protection shoes wh	ECTION	IMDG Class UN number and prope UN 1035 Ethane Labels Packing Instruction EmS	2.1 er shipping nar 2.1 P200 FD,SU	me	
-,			IATA Class UN number and prope UN 1035 Ethane Labels	2.1	me	
Important information Molecular weight: 30 g Melting point: -183 °C Boiling point: -88,6 °C Critical temperature: 3 Autoignition temperatu Flammability range: 2, Relative density, gas: - Relative density, liquid Vapour Pressure 20 °C Solubility mg/l water: 6 Other data	on environment, health a /mol 2 °C ure: 515 °C 7 %(V) - 16 %(V) 1 1: 0,54 :: 37,8 bar		Packing Instruction Other transport inform Avoid transport on vehi from the driver's compa potential hazards of the an accident or an containers ensure that cylinder valve is closed cap nut or plug (where valve protection device adequate ventilation. Er 15 REGULATORY INF	icles where the rtment. Ensure a load and kno emergency. I they are firm and not leaking provided) is co (where provid nsure complian	e vehicle driver is awa ws what to do in the Before transporting nly secured. Ensure g. Ensure that the va orrectly fitted. Ensure led) is correctly fitted	are of the event of product that the live outlet e that the d. Ensure
particularly at or below g 10 STABILITY AND RE Stability and reactivity Can form explosive m oxidants. 11 TOXICOLOGICAL I	EACTIVITY nixture with air. May rea	act violently with	Further national regula Pressure Vessel Regula Gefahrstoffverordnung (Technische Regeln für (Regulations for the prev Water pollution class Not polluting to waters a	ation (GefStoff∨) Gefahrstoffe (T ⁄ention of indus	strial accidents	
Acute toxicity No known toxicological e	effects from this product.		16 OTHER INFORMAT Ensure all national/local understand the flamma often overlooked and	l regulations ar bility hazard. 1 must be stres	The hazard of asphy sed during operator	viation is training.
13 DISPOSAL CONSIL General Do not discharge into explosive mixture with suitable burner with flas	mage caused by this produ	sk of forming an flared through a ischarge into any	Before using this prod thorough material comp out. Advice Whilst proper care ha document, no liability fo be accepted. Details <u>c</u> correct at the time of go Further information Hommel: Handbook of of Kühn-Birett: Merkblätter Linde safety advice	patibility and si as been taker or injury or dam given in this d ing to press. dangerous goo	afefy study should b n in the preparation age resulting from its locument are believ ds	e carried n of this s use can

THE LINDE GROUP					
Safety data sheet Ethylene					
Creation date : Revision date :	28.01.2005 04.01.2011	Versio	n : 6.0	DE / E	SDS No. : 055a page 1 / 3
1 IDENTIFICATION OF THE COMPANY	OF THE SUBSTANCE/PREPARATIO	N AND	P377	unless leak	s fire: Do not extinguish, can be stopped safely.
Product name Ethylene			P381	Eliminate a so.	Il ignition sources if safe to do
EC No (from EINECS CAS No: 74-85-1 Index-Nr. 601-010-00-	3		Precautionary States P403		vell-ventilated place.
Chemical formula C2 REACH Registration Not available.			Precautionary State	ment Disposal None.	
Known uses Not known. Company identification Linde AG, Linde Gas Division, Seitnerstraße 70, D-82049 Pullach E-Mail Address Info@de.linde-gas.com Emergency phone numbers (24h): 089-7446-0 2 HAZARDS IDENTIFICATION Classification of the substance or mixture Classification acc. to Regulation (EC) No 1272/2008/EC (CLP/GHS) Press. Gas (Liquefied gas) - Contains gas under pressure; may explode if heated. Flam. Gas 1 - Extremely flammable gas. STOT SE 3 - May cause drowsiness or dizziness. Classification acc. to Directive 67/548/EEC & 1999/45/EC F+; R12 R67 Extremely flammable. Vapours may cause drowsiness and dizziness. Risk advice to man and the environment Compressed gas. Label Elements		 3 COMPOSITION/INFORMATION ON INGREDIENTS Substance/Preparation: Substance. Components/Impurities Ethylene CAS No: 74-85-1 Index-Nr.: 601-010-00-3 EC No (from EINECS): 200-815-3 REACH Registration number: Not available. Contains no other components or impurities which will influence the classification of the product. 4 FIRST AID MEASURES Inhalation In high concentrations may cause asphyxiation. Symptoms may include loss of mobility/consciousness. Victim may not be aware of asphyxiation. In low concentrations may cause anarcotic effects. Symptoms may include dizziness, headache, nausea and loss of coordination. Remove victim to uncontaminated area wearing self contained breathing apparatus. Keep victim warm and rested. Call a doctor. Apply artificial respiration if breathing stopped. Ingestion is not considered a potential route of exposure. 			
- Hazard Statements H280 H220 H336 - Precautionary State Precautionary State	Danger Contains gas under pressure; may explode if heated. Extremely flammable gas. May cause drowsiness or dizzines ements ment Prevention	SS.	water from a protect flame unless absolution may occur. Ex- Special protective et In confined space use 6 ACCIDENTAL RE Personal precaution Wear self-contained that atmosphere is proved	cause containers ion products on may form carbd ng media nts can be used. of product. Move ed position. Do utely necessary. ttinguish any othe quipment for fire self-contained b ELEASE MEASU s preathing apparat t to be safe. Eva	on monoxide. e container away or cool with not extinguish a leaking gas Spontaneous/explosive re- er fire. e fighters reathing apparatus. RES us when entering area unless cuate area. Ensure adequate
P210 P260	Keep away from heat/sparks/oper flames/hot surfaces No smoking Do not breathe gas, vapours.		air ventilation. Eliminate ignition sources. Environmental precautions Try to stop release. Prevent from entering sewers, basements		es. ering sewers, basements and
Precautionary Stater P304+P340+P315	nent Reaction IF INHALED: Remove victim to fre and keep at rest in a position comfortable for breathing. Get immediate medical advise/attentic				ulation can be dangerous.

Linde THE LINDE GROUP Safety data sheet Ethylene 28 01 2005 Version : 6.0 DE / E SDS No. : 055a Creation date : Revision date : 04.01.2011 page 2/3 7 HANDLING AND STORAGE place where its accumulation could be dangerous. Contact supplier if guidance is required. Handling EWC Nr. 16 05 04* Ensure equipment is adequately earthed. Suck back of water into the container must be prevented. Purge air from system before 14 TRANSPORT INFORMATION introducing gas. Do not allow backfeed into the container. Use only properly specified equipment which is suitable for this product, its ADR/RID supply pressure and temperature. Contact your gas supplier if in Class 2 Classification Code 2F doubt. Keep away from ignition sources (including static UN number and proper shipping name discharges). Refer to supplier's handling instructions UN 1962 Ethylene Storage UN 1962 Ethylene Secure cylinders to prevent them falling. Segregate from oxidant Labels 21 Hazard number 23 gases and other oxidants in store. Keep container below 50°C in a well ventilated place. Observe "Technische Regeln Druckgase IMDG (TRG) 280 Ziffer 5' Class 21 UN number and proper shipping name EXPOSURE CONTROLS/PERSONAL PROTECTION UN 1962 Ethylene 8 Labels 2.1 P200 Packing Instruction Personal protection EmS FD,SU Ensure adequate ventilation. Do not smoke while handling product. Carry working gloves and protection shoes while handling gas IATA cylinders Class 2.1 UN number and proper shipping name 9 PHYSICAL AND CHEMICAL PROPERTIES UN 1962 Ethylene General information 2.1 Labels Appearance/Colour: Colourless gas P200 Packing Instruction Odour: Poor warning properties at low concentrations. Sweetish. Other transport information Important information on environment, health and safety Avoid transport on vehicles where the load space is not separated Molecular weight: 28 g/mol from the driver's compartment. Ensure vehicle driver is aware of the Melting point: -169 °C potential hazards of the load and knows what to do in the event of an accident or an emergency. Before transporting product containers ensure that they are firmly secured. Ensure that the Boiling point: -103 °C Critical temperature: 9,5 °C Autoignition temperature: 425 °C cylinder valve is closed and not leaking. Ensure that the valve outlet Flammability range: 2,3 %(\lor) - 34 %(\lor) cap nut or plug (where provided) is correctly fitted. Ensure that the Relative density, gas: 1 valve protection device (where provided) is correctly fitted. Ensure Relative density, liquid: 0,57 adequate ventilation. Ensure compliance with applicable regulations. Vapour Pressure 20 °C: Not applicable. Solubility mg/l water: No reliable data available. 15 REGULATORY INFORMATION Other data Gas/vapour heavier than air. May accumulate in confined spaces, Further national regulations particularly at or below ground level. Pressure Vessel Regulation Gefahrstoffverordnung (GefStoffV) 10 STABILITY AND REACTIVITY Technische Regeln für Gefahrstoffe (TRGS) Stability and reactivity Regulations for the prevention of industrial accidents May decompose violently at high temperature and/or pressure or in Water pollution class the presence of a catalyst Can form explosive mixture with air. May Not polluting to waters according to VwVwS from 17.05.99. react violently with oxidants TA-Luft Not classified according to TA-Luft. 11 TOXICOLOGICAL INFORMATION **16 OTHER INFORMATION** Acute toxicity Ensure all national/local regulations are observed. Ensure operators No known toxicological effects from this product. understand the flammability hazard. The hazard of asphyxiation is often overlooked and must be stressed during operator training. Before using this product in any new process or experiment, a 12 ECOLOGICAL INFORMATION thorough material compatibility and safety study should be carried General out No known ecological damage caused by this product. Advice Whilst proper care has been taken in the preparation of this 13 DISPOSAL CONSIDERATIONS document, no liability for injury or damage resulting from its use can General be accepted. Details given in this document are believed to be Do not discharge into areas where there is a risk of forming an correct at the time of going to press. explosive mixture with air. Waste gas should be flared through a

Further information suitable burner with flash back arrestor. Do not discharge into any Hommel: Handbook of dangerous goods

055a / EDV / 29.12.2010



Material Safety Data Sheet: Hydrogen

Product Name: Hydrogen	CAS : 1333-74-0	
Hydrogen, Compressed (D.O.T); Water Gas	DOT I.D No.: UN 1049	
Chemical Name and Synonyms: Hydrogen, Normal Hydrogen	DOT Hazard Class: Division 2.1	
Formula: H ₂	Chemical Family: Inorganic Flammable Gas	

Health Hazard Data

Time Weighted Average Exposure Limit:

Hydrogen is defined as a simple asphyxiant (ACGIH 1994-1995); OSHA 1993 PEL (8 Hr. TWA) = No Listing

Symptoms of Exposure:

<u>Inhalation:</u> High concentrations of hydrogen so as to exclude an adequate supply of oxygen to the lungs causes dizziness, deeper breathing due to air hunger, possible nausea and eventual unconsciousness.

Toxicological Properties:

Hydrogen is inactive biologically and essentially nontoxic; therefore, the major property is the exclusion of an adequate supply of oxygen to the lungs.

Hydrogen is not listed in the IARC, NTP or by OSHA as a carcinogen or potential carcinogen.

Persons in ill health where such illness would be aggravated by exposure to hydrogen should not be allowed to work with or handle this product.

Hazardous Mixtures of other Liquids, Solids or Gases: Hydrogen is flammable over a very wide range in air.			
PHYSICAL DATA			
Boiling Point: -423°F (-252.8°C)	Liquid Density at Boiling Point: 4.43 lb/ft3 (70.96 kg/m3)		
Vapor Pressure @ 70°F (21.1 °C) = Above the critical temperature of 399.8°F (-239.9°C)	Gas Density at 70°F. 1 atm .0052		
Solubility in Water: Very slightly	Freezing Point: -434.6°F (-259.2°C)		
Evaporation Rate: N/A (Gas)	Specific Gravity (AIR=1) @ ^{70F} (21.1°C)=.069		
Appearance and Odor: Colorless, odorless gas			

Fire and Explosion Hazard Data

Flash Point (Method used): N/A Gas	Auto Ignition Temperature: 1058°F (570°C)	Flammable Limits % by Volume: LEL 4 UEL 74.5
Extinguishing Media: Water, carbon dioxide, dry chemical Group B		
Special Fire fighting Procedures: If possible, stop the flow of hydrogen. Cool surrounding containers with water spray. Hydrogen burns with an almost invisible flame of relatively low thermal radiation.		
Unusual Fire and Explosion Hazards: Hydrogen is very light and rises very rapidly in air. Should a hydrogen fire be extinguished and the flow of gas continue, increase ventilation to prevent an explosion hazard, particularly in the upper portions.		

<u>Reactivity Data</u>

Stability: Stable

Incompatibility (Materials to Avoid): Oxidizers

Hazardous Decomposition Products: None

Hazardous Polymerization: Will not occur

Conditions to Avoid None

Spill or Leak Procedures

Steps to be taken in case material is released or spilled:

Evacuate all personnel from affected area. Use appropriate protective equipment. If leak is in user's equipment, be certain to purge piping with an inert gas prior to attempting repairs. If leak is in container or container valve, contact your closest supplier location or call the emergency telephone number listed herein.

Waste disposal methods:

Do not attempt to dispose of waste or unused quantities. Return in the shipping container properly labeled, with any valve outlet plugs or caps secured and valve protection cap in place to your supplier. For emergency disposal assistance, contact your closest supplier location or call the emergency telephone number listed herein.

Special Protection Information

Respiratory Protection (Specify type): Positive pressure air line with mask or selfcontained breathing apparatus should be available for emergency use.

Ventilation: Hood with forced ventilation

Local Exhaust: To prevent accumulation above the LEL

Mechanical (Gen.): In accordance with electrical codes

Protective Gloves: Plastic or rubber

Eye Protection: Safety goggles or glasses

Other Protective Equipment: Safety shoes, safety shower

Special Precautions

Special Labeling Information:

DOT Shipping Name: Hydrogen, Compressed

DOT Hazard Class: Division 2.1

DOT Shipping Label: Flammable Gas

I.D. No.: UN 1049

Special Handling Recommendation:

Use only in well-ventilated areas. Valve protection caps must remain in place unless container is secured with valve outlet piped to use point. Do not drag, slide or roll cylinders. Use a suitable hand truck for cylinder movement. Use a pressure reducing regulator when connecting cylinder to lower pressure (<3,000 psig) piping or systems. Do not heat cylinder by any means to increase the discharge rate of product from the cylinder. Use a check valve or trap in the discharge line to prevent hazardous back flow into the cylinder. For additional handling recommendations, consult Compressed Gas Association's Pamphlets G-5, P-1, P-14, and Safety Bulletin SB-2.

Special Storage Recommendations:

Protect cylinders from physical damage. Store in cool, dry, well-ventilated area of I noncombustible construction away from heavily trafficked areas and emergency exits. Do not allow the temperature where cylinders are stored to exceed 125F (52C). Cylinders should be stored upright and firmly secured to prevent falling or being knocked over. Full and empty cylinders should be segregated. Use a "first in -first out" inventory system to prevent full cylinders being stored for excessive periods of time. Post "No Smoking or Open Flames" signs in the storage or use area. There should be no sources of ignition in the storage or use area. For additional storage recommendations, consult Compressed Gas Association's Pamphlets G-5, P-1, P-14, and Safety Bulletin SB-2.

Other Recommendations or Precautions:

Earth-ground and bond all lines and equipment associated with the hydrogen system. Electrical equipment should be non-sparking or explosion proof. Compressed gas cylinders should not be refilled except by qualified producers of compressed gases. Shipment of a compressed gas cylinder which has not been filled by the owner or with his (written) consent is a violation of Federal Law (49CFR).

Special Packaging Recommendations:

Hydrogen is non-corrosive and may be used with any common structural material.

Appendix A-4. MSDS Information

MSDS

MATERIAL SAFETY DATA SHEET

CBV2314 ONAL

ZEOLITE AMMONIUM ZSM-5 POWDER

Date

Zeolite

Prepared: 03/14/06 Page: 1 of 4

1. CHEMICAL PRODUCT AND COMPANY IDENTIFICATION

Product name:	CBV2314
	ZEOLITE AMMONIUM ZSM-5
Product description:	POWDER
	Zeolyst
Manufacturer:	International
	P. O. Box 830
	Valley Forge, PA 19482 USA
	610-651-
Telephone:	4200
In case of emergency	610-651-
call:	4200
For transportation emergency	
	800-424-
Call CHEMTREC:	9300
2. COMPOSITION/INFORM	MATION ON
INGREDIENTS	
Chemical and Common	
Name CAS K	Pegistry Wt. % OSHA PEL ACGIH TLV
Numb	er 3

15mg/m³ total dust 10 mg/m³ 100%

5mg/m³ respirable 3 mg/m³ respirable

1318-02-1

<u>3. HAZARDS IDENTIFICATION</u>

	White, odorless, powder. Causes respiratory irritation. Causes mild
Emergency Overview:	eye
	irritation. May cause skin irritation. Ammonia released on contact
	with
	strong bases. Noncombustible. Ammonia or nitrogen oxides may be
	released
	at high temperatures.
Eye contact:	Causes mild eye irritation.
Skin contact:	Prolonged or repeated contact may dry skin and cause irritation.
Inhalation:	Causes irritation.
Ingestion:	No known hazards. Inedible.
Chronic hazards:	No known hazards.
	Absorbs water from air and fluids. Generates heat when it absorbs
Physical hazards:	water.

4. FIRST AID MEASURES

	In case of contact, immediately flush eyes with plenty of water for at
Eye:	least 15
	minutes. Get medical attention if irritation persists.
	In case of contact, immediately flush skin with plenty of water.
Skin:	Remove
	contaminated clothing and shoes. Get medical attention if irritation
	develops
	aevelops

Trade Name: CBV2314

	ZEOLITE AMMONIUM ZSM5 POWDER			
Date				
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 Appendix A-4. MSDS Information

 and persists. Wash clothing before reuse. Thoroughly clean shoes

 before

 reuse.

 Remove to fresh air. If not breathing, give artificial respiration. If

 Inhalation:
 breathing

 is difficult, give oxygen. Get medical attention.

 Ingestion:
 Not applicable.

<u>5. FIRE FIGHTING MEASURES</u>

Flammable limits: This material is noncombustible.

Extinguishing Media: This material is compatible with all extinguishing media. *Hazards to fire-fighters:* Ammonia or nitrogen oxides may be released at high

temperatures.

Fire-fighting equipment: The following protective equipment for fire fighters is recommended when this material is present in the area of a fire: self-contained breathing apparatus (SCBA), chemical goggles, body-covering protective clothing, chemical resistant gloves, and rubber boots.

6. ACCIDENTAL RELEASE MEASURES

resistant gloves	es, body-covering protective clothing, chemical , and rubber boots, NIOSH-approved dust dust occurs. See section 8.			
Environmental Hazards: Sinks in water. No know environmental hazards.				
Small spill cleanup: Carefully shovel or	sweep up spilled material and place in suitable container.			
Equipment (PPE Large spill cleanup: Keep unnecessary p Do not touch or sweep up spilled generating dust. (PPE). See section CERCLA RQ: There is no CERCLA Rep	people away; isolate hazard area and deny entry. walk through spilled material. Carefully shovel or material and place in suitable container. Avoid Use appropriate Personal Protective Equipment			

7. HANDLING AND

STORAGE

	Avoid contact with eyes, skin and clothing. Avoid breathing dust.
Handling:	Keep
	container closed. Promptly clean up spills. Wash thoroughly after
	handling.
Storage:	Keep containers closed. Store separated from strong bases in original
	containers or clean metal, plastic, or fiber containers.

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Trade Name:	CBV2314
	ZEOLITE AMMONIUM ZSM5 POWDER
Date	
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8. EXPOSURE CONTROLS/PERSONAL PROTECTION

	Use with adequate ventilation. Safety shower and eyewash fountain
Engineering controls:	should be
	within direct access.
	Use a NIOSH-approved dust respirator where dust occurs. Observe
Respiratory protection:	OSHA
	regulations for respirator use (29 C.F.R. §1910.134)
Skin protection:	Wear body-covering protective clothing and gloves.
Eye protection:	Wear safety goggles.

9. PHYSICAL AND CHEMICAL PROPERTIES

Appearance:	White powder.
Odor:	Odorless.
pH:	Not applicable.
Specific Gravity:	>1.
Solubility in water:	Negligible.

10. STABILITY AND REACTIVITY

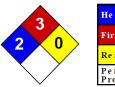
Stability:	This material is stable.
Conditions to avoid:	None.
Materials to avoid:	Ammonia may be released on contact with strong bases.
Hazardous	
decomposition	
products:	Ammonia, nitrogen oxides.

11. TOXICOLOGICAL INFORMATION

Acute Data: When tested for primary irritation potential, similar materials caused mild
eye irritation and were slightly irritating or non-irritating to the
skin. Human experience with similar materials indicates that
prolonged or repeated contact may dry skin and cause irritation.
This material has not been tested for acute inhalation toxicity. It
contains fine particles which can cause respiratory irritation.
The acute oral toxicity of this material has not been tested.
There are currently no reports of human toxicity for
ingested zeolite.Subchronic Data:This material has not been tested for subchronic toxicity potential.Special Studies:This material has not been tested for chronic inhalation toxicity. Zeolite

Beta is not listed by NTP, IARC, or OSHA as a carcinogen.





Health2Fire3Reactivity0Personal
ProtectionJ

Material Safety Data Sheet m-Xylene MSDS

Section 1: Chemical Product and Company Identification				
Product Name: m-Xylene	Contact Information:			
Catalog Codes: SLX1066	Sciencelab.com, Inc.			
CAS#: 108-38-3	14025 Smith Rd. Houston, Texas 77396			
RTECS: ZE2275000	US Sales: 1-800-901-7247			
TSCA: TSCA 8(b) inventory: m-Xylene	International Sales: 1-281-441-4400 Order Online: <u>ScienceLab.com</u>			
CI#: Not applicable.	CHEMTREC (24HR Emergency Telephone), call:			
Synonym: m-Methyltoluene	1-800-424-9300			
Chemical Name: 1,3-Dimethylbenzene	International CHEMTREC, call: 1-703-527-3887			
Chemical Formula: C6H4(CH3)2	For non-emergency assistance, call: 1-281-441-4400			

Section 2: Composition and Information on Ingredients				
Composition:				
Name	CAS #	% by Weight		
{m-}Xylene	108-38-3	100		

Toxicological Data on Ingredients: m-Xylene: ORAL (LD50): Acute: 5000 mg/kg [Rat.]. DERMAL (LD50): Acute: 14100 mg/kg [Rabbit.].

Section 3: Hazards Identification

Potential Acute Health Effects:

Very hazardous in case of skin contact (irritant), of eye contact (irritant). Slightly hazardous in case of skin contact (permeator), of ingestion, of inhalation. Inflammation of the eye is characterized by redness, watering, and itching. Skin inflammation is characterized by itching, scaling, reddening, or, occasionally, blistering.

Potential Chronic Health Effects:

Hazardous in case of skin contact (irritant), of eye contact (irritant). Slightly hazardous in case of skin contact (permeator), of ingestion, of inhalation. CARCINOGENIC EFFECTS: Not available. MUTAGENIC EFFECTS: Not available. TERATOGENIC EFFECTS: Not available. DEVELOPMENTAL TOXICITY: Not available. The substance is toxic to blood, kidneys, the nervous system, liver. Repeated or prolonged exposure to the substance can produce target organs damage.

Section 4: First Aid Measures

p. 1

Eye Contact: Check for and remove any contact lenses. Do not use an eye ointment. Seek medical attention.

Skin Contact:

After contact with skin, wash immediately with plenty of water. Gently and thoroughly wash the contaminated skin with running water and non-abrasive soap. Be particularly careful to clean folds, crevices, creases and groin. Cover the irritated skin with an emollient. If irritation persists, seek medical attention. Wash contaminated clothing before reusing.

Serious Skin Contact:

Wash with a disinfectant soap and cover the contaminated skin with an anti-bacterial cream. Seek medical attention.

Inhalation: Allow the victim to rest in a well ventilated area. Seek immediate medical attention.

Serious Inhalation: Not available.

Ingestion:

Do not induce vomiting. Loosen tight clothing such as a collar, tie, belt or waistband. If the victim is not breathing, perform mouth-to-mouth resuscitation. Seek immediate medical attention.

Serious Ingestion: Not available.

Section 5: Fire and Explosion Data

Flammability of the Product: Flammable.

Auto-Ignition Temperature: 527°C (980.6°F)

Flash Points: CLOSED CUP: 25°C (77°F). OPEN CUP: 28.9°C (84°F) (Cleveland).

Flammable Limits: LOWER: 1.1% UPPER: 7%

Products of Combustion: These products are carbon oxides (CO, CO2).

Fire Hazards in Presence of Various Substances: Highly flammable in presence of open flames and sparks, of heat.

Explosion Hazards in Presence of Various Substances:

Risks of explosion of the product in presence of mechanical impact: Not available. Risks of explosion of the product in presence of static discharge: Not available.

Fire Fighting Media and Instructions:

Flammable liquid, insoluble in water. SMALL FIRE: Use DRY chemical powder. LARGE FIRE: Use water spray or fog. Cool containing vessels with water jet in order to prevent pressure build-up, autoignition or explosion.

Special Remarks on Fire Hazards:

Explosive in the form of vapor when exposed to heat or flame. Vapor may travel considerable distance to source of ignition and flash back. When heated to decomposition it emits acrid smoke and irritating fumes.

Special Remarks on Explosion Hazards: Not available.

Section 6: Accidental Release Measures

Small Spill: Absorb with an inert material and put the spilled material in an appropriate waste disposal.

Large Spill:

Flammable liquid, insoluble in water. Keep away from heat. Keep away from sources of ignition. Stop leak if without risk. Absorb with DRY earth, sand or other non-combustible material. Do not get water inside container. Do not touch spilled material. Prevent entry into sewers, basements or confined areas; dike if needed. Eliminate all ignition sources. Call for assistance on disposal. Be careful that the product is not present at a concentration level above TLV. Check TLV on the MSDS and with local authorities.

Section 7: Handling and Storage

Precautions:

Keep away from heat. Keep away from sources of ignition. Ground all equipment containing material. Do not ingest. Do not breathe gas/fumes/ vapour/spray. If ingested, seek medical advice immediately and show the container or the label. Avoid contact with skin and eyes Keep away from incompatibles such as oxidizing agents.

Storage:

Flammable materials should be stored in a separate safety storage cabinet or room. Keep away from heat. Keep away from sources of ignition. Keep container tightly closed. Keep in a cool, well-ventilated place. Ground all equipment containing material. A refrigerated room would be preferable for materials with a flash point lower than 37.8°C (100°F).

Section 8: Exposure Controls/Personal Protection

Engineering Controls:

Provide exhaust ventilation or other engineering controls to keep the airborne concentrations of vapors below their respective threshold limit value. Ensure that eyewash stations and safety showers are proximal to the work-station location.

Personal Protection: Splash goggles. Lab coat. Gloves.

Personal Protection in Case of a Large Spill:

Splash goggles. Full suit. Boots. Gloves. Suggested protective clothing might not be sufficient; consult a specialist BEFORE handling this product.

Exposure Limits:

TWA: 100 STEL: 150 (ppm) from ACGIH (TLV) TWA: 434 STEL: 651 (mg/m3) from ACGIHConsult local authorities for acceptable exposure limits.

Section 9: Physical and Chemical Properties
Physical state and appearance: Liquid. (Liquid.)
Odor: Not available.
Taste: Not available.
Molecular Weight: 106.17 g/mole
Color: Colorless.
pH (1% soln/water): Not applicable.
Boiling Point: 139.3°C (282.7°F)
Melting Point: -47.87°C (-54.2°F)
Critical Temperature: Not available.
Specific Gravity: 0.86 (Water = 1)
Vapor Pressure: 6 mm of Hg (@ 20°C)
Vapor Density: 3.7 (Air = 1)
Volatility: Not available.
Odor Threshold: 0.62 ppm
Water/Oil Dist. Coeff.: Not available.
Ionicity (in Water): Not available.
Dispersion Properties: See solubility in water, methanol, diethyl ether.
Solubility: Easily soluble in methanol, diethyl ether. Insoluble in cold water, hot water.

Section 10: Stability and Reactivity Data

Stability: The product is stable.

Instability Temperature: Not available.

Conditions of Instability: Not available.

Incompatibility with various substances: Reactive with oxidizing agents.

Corrosivity: Non-corrosive in presence of glass.

Special Remarks on Reactivity: Not available.

Special Remarks on Corrosivity: Not available.

Polymerization: No.

Section 11: Toxicological Information

Routes of Entry: Eye contact.

Toxicity to Animals:

Acute oral toxicity (LD50): 5000 mg/kg [Rat.]. Acute dermal toxicity (LD50): 14100 mg/kg [Rabbit.].

Chronic Effects on Humans: The substance is toxic to blood, kidneys, the nervous system, liver.

Other Toxic Effects on Humans:

Very hazardous in case of skin contact (irritant). Slightly hazardous in case of skin contact (permeator), of ingestion, of inhalation.

Special Remarks on Toxicity to Animals: Not available.

Special Remarks on Chronic Effects on Humans:

0347 Animal: embryotoxic, foetotoxic, passes through the placental barrier. 0900 Detected in maternal milk in human. Narcotic effect; may cause nervous system disturbances.

Special Remarks on other Toxic Effects on Humans: Material is irritating to mucous membranes and upper respiratory tract.

Section 12: Ecological Information

Ecotoxicity: Not available.

BOD5 and COD: Not available.

Products of Biodegradation:

Possibly hazardous short term degradation products are not likely. However, long term degradation products may arise.

Toxicity of the Products of Biodegradation: The products of degradation are more toxic.

Special Remarks on the Products of Biodegradation: Not available.

Section 13: Disposal Considerations

Waste Disposal:

Section 14: Transport Information

DOT Classification: Class 3: Flammable liquid.

Identification: : Xylene : UN1307 PG: III

Special Provisions for Transport: Not available.

Section 15: Other Regulatory Information

Federal and State Regulations:

Pennsylvania RTK: m-Xylene Massachusetts RTK: m-Xylene TSCA 8(b) inventory: m-Xylene SARA 313 toxic chemical notification and release reporting: m-Xylene CERCLA: Hazardous substances.: m-Xylene

Other Regulations: OSHA: Hazardous by definition of Hazard Communication Standard (29 CFR 1910.1200).

Other Classifications:

WHMIS (Canada):

CLASS B-2: Flammable liquid with a flash point lower than 37.8°C (100°F). CLASS D-2B: Material causing other toxic effects (TOXIC).

DSCL (EEC):

R10- Flammable. R38- Irritating to skin. R41- Risk of serious damage to eyes.

HMIS (U.S.A.):

Health Hazard: 2

Fire Hazard: 3

Reactivity: 0

Personal Protection: j

National Fire Protection Association (U.S.A.):

Health: 2

Flammability: 3

Reactivity: 0

Specific hazard:

Protective Equipment:

Gloves. Lab coat. Wear appropriate respirator when ventilation is inadequate. Splash goggles.

Section 16: Other Information

References:

-Hawley, G.G.. The Condensed Chemical Dictionary, 11e ed., New York N.Y., Van Nostrand Reinold, 1987. -Material safety data sheet emitted by: la Commission de la Santé et de la Sécurité du Travail du Québec. -SAX, N.I. Dangerous Properties of Indutrial Materials. Toronto, Van Nostrand Reinold, 6e ed. 1984. -The Sigma-Aldrich Library of Chemical Safety Data, Edition II. -Guide de la loi et du rÃ"glement sur le transport des marchandises dangeureuses au canada. Centre de conformité internatinal Ltée. 1986.

Other Special Considerations: Not available.

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MATERIAL SAFETY DATA SHEET

Prepared to U.S. OSHA, CMA, ANSI and Canadian WHMIS Standards

1. PRODUCT IDENTIFICATION

CHEMICAL NAME: CLASS: METHANE

SYNONYMS:	Methyl Hydride; Marsh Ga	s
CHEMICAL EA	MILY Alkane (hydrocarb	on)

FORMULA: CH4

Document Number: 50006 (Replaces ISC MSDS No.1810-2312, 1810-4778)

Note: This Material Safety Data Sheet is for Methane supplied in cylinders with 33 cubic feet (935 liters) or less gas capacity (DOT 39 cylinders). For Methane in large cylinders refer to Document Number 10060

CALGAZ

PRODUCT USE:

ADDRESS:

1

SUPPLIER/MANUFACTURER'S NAME:

EMERGENCY PHONE: BUSINESS PHONE:

General MSDS Information: Fax on Demand: 821 Chesapeake Drive Cambridge, MD 21613 CHEMTREC: 1-800-424-9300 1-410-228-6400 1-713/868-0440 1-800/231-1366

Calibration of Monitoring and Research Equipment

2. COMPOSITION and INFORMATION ON INGREDIENTS

CHEMICAL NAME	CAS #	mole %	EXPOSURE LIMITS IN AIR					
			ACGI	H-TLV	OSHA-TLV		NIOSH	OTHER
			TWA	STEL	TWA	STEL	IDLH	
			ppm	ppm	ppm	ppm	ppm	ppm
Methane	74-82-8	> 98%	There are no specific exposure limits for Methane. Methane is a simple asphyxiant (SA). Oxygen levels should be maintained above 19.5%.					
Maximum Impurities < 2.0%		associated w provided in th	rith the product his Material Sat	. All hazard i fety Data She	nformation pert	inent to this pr irements of the	to the hazards oduct has been e OSHA Hazard tandards.	

NOTE (1): ALLI WHINS required information is included in appropriate sections based on the ANSI Z400.1-1998 format. This product has been classified in accordance with the hazard criteria of the CPR and the MSDS contains all the information required by the CPR.

3. HAZARD IDENTIFICATION

EMERGENCY OVERVIEW: This product is a colorless, flammable gas. The main health hazard associated with overexposure to this gas is asphyxiation, by displacement of oxygen. The gas poses a serious fire hazard when accidentally released. Flame or high temperature impinging on a localized area of the cylinder of this product can cause the cylinder to burst or rupture without activating the cylinder's relief devices. The gas is lighter than air, and may spread long distances. Distant ignition and flashback are possible. Provide adequate fire protection during emergency response situations.

SYMPTOMS OF OVER-EXPOSURE BY ROUTE OF EXPOSURE: The most significant route of over-exposure for this product is by inhalation. INHALATION: Due to the small size of an individual cylinder of this product, no unusual health effects from exposure to the product are anticipated under routine circumstances of use. If this product is released in a small, poorly ventilated area (i.e. an enclosed or confined space), an oxygendeficient environment may occur. It should be noted that before suffication could occur, the lower flammability limit of Methane in air would be exceeded; possibly causing an oxygen-deficient and explosive atmosphere. Individuals breathing an oxygen deficient atmosphere may experience

symptoms which include headaches, ringing in ears, dizziness, drowsiness, unconsciousness, nausea, vomiting, and depression of all the senses. Under some circumstances of over-exposure, death may occur. The following effects associated with various levels of oxygen are as follows

OXYGEN CONCENTRATION	SYMPTOM OF EXPOSURE	HEALTH HAZARD	(BLUE)
12-16% Oxygen:	Breathing and pulse rate increased, muscular coordination slightly disturbed.		
10-14% Oxygen: 6-10% Oxygen: Below 6%	Emotional upset, abnormal fatigue, disturbed respiration. Nausea and vomiting, collapse or loss of consciousness. Convulsive movements, possible respiratory collapse, and	FLAMMABILITY HAZARD) (RED)
	death.		
	R RISKS FROM EXPOSURE: An Explanation in Lay Terms. Over- ture may cause the following health effects:	PHYSICAL HAZARD	(YELLOW)

exposure to this gas mixture may cause the following health effects:

ACUTE: Due to the small size of the individual cylinder of this product, no unusual health effects from exposure to the product are anticipated under routine circumstances of use. The most significant hazard associated with this product is inhalation of oxygen-deficient atmospheres. Symptoms of oxygen deficiency include respiratory difficulty, ringing in ears, headaches, shortness of breath, wheezing, headache, dizziness, indigestion, nausea, and, at high concentrations, unconsciousness or death may occur. The skin of a victim of over-exposure may have a blue color. chronic data and a start of a sta

PROTECTIVE EQUIPMENT EYES RESPIRATORY HANDS See Section 8 effect the heart and nervous system. TARGET ORGANS: ACUTE: Respiratory system. CHRONIC: Heart, central nervous system. For Routine Industrial Use and Handling Applications

METHANE- CH4 MSDS - 50006 PN 3471

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HAZARDOUS MATERIAL IDENTIFICATION SYSTEM

(YELLOW) 0

4

BODY

NFPA RATING

4 0

OTHER

4. FIRST-AID MEASURES

RESCUERS SHOULD NOT ATTEMPT TO RETRIEVE VICTIMS OF OVER-EXPOSURE TO THIS PRODUCT WITHOUT ADEQUATE PERSONAL PROTECTIVE EQUIPMENT. At a minimum, Self-Contained Breathing Apparatus and Fire-Retardant Personal Protective equipment should be worn. Adequate fire protection must be provided during rescue situations.

No unusual health effects are anticipated after exposure to this product, due to the small cylinder size. If any adverse symptom develops after overexposure to this product, remove victim(s) to fresh air, as quickly as possible. Only trained personnel should administer supplemental oxygen and/or cardio-pulmonary resuscitation, if necessary. Victim(s) who experience any adverse effect after over-exposure to this product must be taken for medical attention. Rescuers should be taken for medical attention, if necessary. Take copy of label and MSDS to physician or other health professional with victim(s).

professional with victim(s). THERMAL BURNS: In the event personnel are burned as a result of a Hydrogen release, if burns are first degree or second degree with closed blisters, flush area with cold water until pain subsides. Apply loose, moist, sterile dressings, and bandage. Treat for shock. If burns are second degree with open blisters or third degree, apply loose, dry, sterile dressings and bandage. Treat for shock. Transport victim immediately to hospital or emergency center. Burns over an area of 20% or more of body are life-threatening, medical attention should be immediately sought. MEDICAL CONDITIONS AGGRAVATED BY EXPOSURE: None known.

RECOMMENDATIONS TO PHYSICIANS: Treat symptoms and reduce over-exposure

5. FIRE-FIGHTING MEASURES

FLASH POINT: -187.7°C (-306°F)

AUTOIGNITION TEMPERATURE: 650°C (1202°F) FLAMMABLE LIMITS (in air by volume, %):

FLAMMABLE LIMITS (in air by volume, %): Lower (LEL): 5.0% Upper (UEL): 15.0% FIRE EXTINGUISHING MATERIALS: Extinguish Methane fires by shutting-off the source of the gas. Use water spray to cool fire-exposed containers, structures, and equipment. UNUSUAL FIRE AND EXPLOSION HAZARDS: When involved in a fire, this material may decompose and produce toxic gases including carbon monoxide and carbon dioxide. DANGER! Fires impinging (direct fineme) on the outside surface of unprotected cylinders of this product can be very dangerous. Exposure to fire could cause a catastrophic failure of the outside surfacements into a fineful and ourdering of relaxed new. The resulting fire

cylinder releasing the contents into a fireball and explosion of released gas. The resulting fire and explosion can result in severe equipment damage and personnel injury or death over a large area around the cylinder. For massive fires in large areas, use unmanned hose holder

large area around the cylinder. For massive fires in large areas, use unmanned nose holder or monitor nozzles; if this is not possible, withdraw from area and allow fire to burn.
 <u>Explosion Sensitivity to Mechanical Impact</u>: Not sensitive.
 <u>Explosion Sensitivity to Static Discharge</u>: Static discharge may cause this product to ignite explosively, if released.
 SPECIAL FIRE-FIGHTING PROCEDURES: Structural fire-fighters must wear Self-Contained Breathing Apparatus and full protective equipment.
 Because of the potential for a fire, evacuation of non-emergency personnel is essential. If water is not available for cooling or protection of cylinder exposures, evacuate the area. The North American Emergency Response Guidebook (Guide #115) recommends 0.5 miles.

6. ACCIDENTAL RELEASE MEASURES

LEAK RESPONSE: LEAK RESPONSE: Due to the small size and content of the cylinder, an accidental release of this product presents significantly less risk of a flammability and Oxygen deficient environment and other safety hazards than a similar release from a larger cylinder. However, as with any chemical release, extreme caution must be used during emergency response procedures. In the event of a release in which the atmosphere is unknown, and in which other chemicals are potentially involved, evacuate immediate area. Uncontrolled releases should be affected area, protect people, and respond with trained personnel.

Eliminate any possible sources of ignition, and provide maximum explosion-proof ventilation. If the gas is leaking from cylinder contact the supplier. Adequate fire protection must be provided. Use only non-sparking tools and equipment during the response. Allow the gas, which is lighter than air to dissipate. If necessary, monitor the surrounding area (and the original area of the release) for oxygen. Oxygen levels must be above 19.5% before non-emergency personnel are allowed to re-enter area. Combustible gas concentration must be below 10% of the LEL (5.0%) prior to entry.

If leaking incidentally from the cylinder contact your supplier

THIS IS AN EXTREMELY FLAMMABLE GAS. Protection of all personnel and the area must be maintained.

7. HANDLING and USE WORK PRACTICES AND HYGIENE PRACTICES: Be aware of any signs of dizziness or fatigue; exposures to fatal concentrations of this product could occur without any significant warning symptoms. Use non-sparking tools when working with Methane.

STORAGE AND HANDLING PRACTICES. Cylinders should be firmly secured to prevent falling or being knocked-over. Cylinders must be protected from the environment, and preferably kept at room temperature approximately 21°C, 70°F. Cylinders should be stored in dry, wellventilated areas away from sources of heat, ignition and direct sunlight. Keep storage area clear of materials which can burn. Protect cylinders

Vehilated areas away non-sources of the gradient of the source of the so

Storage areas must meet national electrical codes for Class 1 Hazardous Areas. Post "No Smoking or Open Flames" signs in storage or use equipment in the storage area (i.e. sprinkler system, portable fire extinguishers). Full and empty cylinders should be segregated. Use a first-in, first-out inventory system to prevent full containers from being stored for long periods

of time. These cylinders are not refillable. WARNING! Do not refill DOT 39 cylinders. To do so may cause personal injury or property damage

SPECIAL PRECAUTIONS FOR HANDLING GAS CYLINDERS: WARNING! Compressed gases can present significantly safety hazards. During

cylinder use, use equipment designed for these specific cylinders. Ensure all lines and equipment are rated for proper service pressure. PROTECTIVE PRACTICES DURING MAINTENANCE OF CONTAMINATED EQUIPMENT: Follow practices indicated in Section 6 (Accidental Release Measures). Make certain application equipment is locked and tagged-out safely. Purge gas handling equipment with inert gas (i.e. nitrogen) before attempting repairs. Always use product in areas where adequate ventilation is provided

8. EXPOSURE CONTROLS - PERSONAL PROTECTION

VENTILATION AND ENGINEERING CONTROLS: No special ventilation systems or engineering controls are needed under normal incrumstances of use. As with all chemicals, use this product in well-ventilated areas. Monitoring may be appropriate to detect if the level of Methane exceeds 10% of the LEL (5%).

RESPIRATORY PROTECTION: No special respiratory protection is required under normal circumstances of use. Use supplied air respiratory protection if Oxygen levels are below 19.5%, or unknown, during emergency response to a release of this product. If respiratory protection is needed, use only protection authorized in the U.S. Federal OSHA Standard (29 CFR 1910.134), applicable U.S. State regulations, or the Canadian CSA Standard Z94.4-93 and applicable standards of Canadian Provinces. Oxygen levels below 19.16.33% are considered IDLH by OSHA. In such atmospheres, use of a full-facepiece pressure/demand SCBA or a full facepiece, supplied air respirator with auxiliary self-contained air supply is required under OSHA's Respiratory Protection Standard (1910.134-1998).

METHANE- CH ₄ MSDS - 50006	
PN 3471	

8. EXPOSURE CONTROLS - PERSONAL PROTECTION (CONTINUED)

EYE PROTECTION: Safety classes, If necessary, refer to U.S. OSHA 29 CFR 1910,133 or appropriate Canadian Standard HAND PROTECTION: No special protection is needed under normal circumstances of use. If necessary, refer to U.S. OSHA 29 CFR 1910.138 or appropriate Standards of Canada.

BODY PROTECTION: No special protection is needed under normal circumstances of use. If a hazard of injury to the feet exists due to falling objects, rolling objects, where objects may pierce the soles of the feet or where employee's feet may be exposed to electrical hazards, use foot protection, as described in U.S. OSHA 29 CFR 1910.136.

9. PHYSICAL and CHEMICAL PROPERTIES

GAS DENSITY @ 60°F (15.6°C) and 1 atm: 0.042 35 lb/ft

 GAS DENSITY @ 60°F (15.6°C) and 1 atm: 0.042 35 lbm"

 BOILING POINT: -161°C (-258.7°F)

 FREEZING/MELTING POINT (@ 10 psig): -182°C (-296.5°F)

 SPECIFIC GRAVITY (air = 1) @ 70°F (21.1°C): 0.555

 SOLUBILITY IN WATER vol/vol at 100°F (37.8°C): Very slight. MOLECULAR WEIGHT: 16.042

 EVAPORATION RATE (nBuAc = 1): Not applicable.

 EVAPORATION RATE (nBuAc = 1): Not applicable.

 SPECIFIC SOLUBILITY IN WATER vol/vol at 100°F (37.8°C): Very slight. MOLECULAR WEIGHT: 16.042

 EVAPORATION RATE (nBuAc = 1): Not applicable.

 SPECIFIC VOLUME (#
 EXPANSION RATIO: Not applicable SPECIFIC VOLUME (ft³/lb): 23.6

VAPOR PRESSURE @ 70°F (21.1°C) (psig): Not applicable. COEFFICIENT WATER/OIL DISTRIBUTION: Not applicable.

APPEARANCE, ODOR AND COLOR: Colorless, godrelss gas. HOW TO DETECT THIS SUBSTANCE (warning properties): There are no distinct warning properties of this gas. In terms of leak detection, fittings and joints can be painted with a soap solution to detect leaks, which will be indicated by a bubble formation. NOTE: This gas is lighter than air and must not be allowed to accumulate in elevated locations

10. STABILITY and REACTIVITY

STABILITY: Stable under conditions of normal temperature and pressure.
DECOMPOSITION PRODUCTS: When ignited in the presence of air, this gas will burn to produce carbon monoxide, carbon dioxide.
MATERIALS WITH WHICH SUBSTANCE IS INCOMPATIBLE: Strong oxidizers (i.e. chlorine, bromine pentafluoride, oxygen, oxygen difluoride, and nitrogen trifluoride). HAZARDOUS POLYMERIZATION: Will not occur.

CONDITIONS TO AVOID: Contact with incompatible materials and exposure to heat, sparks and other sources of ignition. Cylinders exposed to high temperatures or direct flame can rupture or burst.

11. TOXICOLOGICAL INFORMATION

TOXICITY DATA: There is no specific toxicology data for Methane. Methane is a simple asphyxiant, which acts to displace oxygen in the

SUSPECTED CANCER AGENT: Methane is not found on the following lists: FEDERAL OSHA Z LIST, NTP, IARC, CAL/OSHA; therefore is not considered to be, nor superted to be cancer-causing agent by these agencies. IRRITANCY OF PRODUCT: Methane is not irritating, however, contact with rapidly expanding gases can cause frostbite to exposed tissue. SENSITIZATION OF PRODUCT: Methane is not a sensitizer.

REPRODUCTIVE TOXICITY INFORMATION: Listed below is information concerning the effects of Methane on the human reproductive system. <u>Mutagenicity</u>: No mutagenicity effects have been described for Methane

<u>Tendpyotoxicity</u>: No empryotoxic effects have been described for Methane. <u>Teratogenicity</u>: No teratogenicity effects have been described for Methane. <u>Reproductive Toxicity</u>: No reproductive toxicity effects have been described for Methane.

<u>Reproductive Toxicity</u>. No reproductive toxicity effects inave been described for Mentana A <u>mutagen</u> is a chemical which causes permanent changes to genetic material (DNA) such that the changes will propagate through generation lines. An <u>embryotoxin</u> is a chemical which causes damage to a developing embryo (i.e. within the first eight weeks of pregnancy in humans), but the damage does not propagate across generational lines. A <u>teratogen</u> is a chemical which causes damage to a developing fetus, but the damage does not propagate across generational lines. A <u>teratogen</u> is a chemical which interferes in any way with the reproductive process. **BIOLOGICAL EXPOSURE INDICES (BEIs)**: Currently, Biological Exposure Indices (BEIs) have not been determined for Methane.

12. ECOLOGICAL INFORMATION

ENVIRONMENTAL STABILITY: This gas will be dissipated rapidly in well-ventilated areas. EFFECT OF MATERIAL ON PLANTS or ANIMALS: Due to the small cylinder size no adverse effect on animals or animals would be anticipated if one cylinder of this product is release

EFFECT OF CHEMICAL ON AQUATIC LIFE: No evidence is currently available on this product's effects on aquatic life.

13. DISPOSAL CONSIDERATIONS

PREPARING WASTES FOR DISPOSAL: Waste disposal must be in accordance with appropriate Federal, State, and local regulations. Cylinders with undesired residual product may be safely vented outdoors with the proper regulator. For further information, refer to Section 16 (Other Information).

14. TRANSPORTATION INFORMATION

THIS GAS IS HAZARDOUS AS DEFINED BY 49 CFR 172.101 BY THE U.S. DEPARTMENT OF TRANSPORTATION.

ROPER SHIPPING NAME: M	Methane, compressed
------------------------	---------------------

HAZARD CLASS NUMBER and DESCRIPTION: 2.1 (Flammable Gas) UN IDENTIFICATION NUMBER: UN 1971

PACKING GROUP: DOT LABEL(S) REQUIRED: Not applicable Class 21. (Flammable Gas)

NORTH AMERICAN EMERGENCY RESPONSE GUIDEBOOK NUMBER (2000): 115 MARINE POLLUTANT: Methane is not classified by the DOT as Marine Pollutants (as defined by 49 CFR 172.101, Appendix B).

SPECIAL SHIPPING INFORMATION: Cylinders should be transported in a secure position, in a well-ventilated vehicle. The transportation of compressed gas cylinders in automobiles or in closed-body vehicles can present serious safety hazards. If transporting these cylinders in vehicles, ensure these cylinders are not exposed to extremely high temperatures (as may occur in an enclosed vehicle on a hot day).

Additionally, the vehicle should be well-ventilated during transportation. Note: DOT 39 Cylinders ship in a storig outer carton (overpack). Pertinent shipping information goes on the outside of the overpack. DOT 39 Cylinders do not have transportation information on the cylinder itself.

TRANSPORT CANADA TRANSPORTATION OF DANGEROUS GOODS REGULATIONS: This gas is considered as Dangerous Goods, per regulations of Transport Canada

PROPER SHIPPING NAME:	Methane, compressed
HAZARD CLASS NUMBER and DESCRIPTION:	2.1 (Flammable Gas)
UN IDENTIFICATION NUMBER:	UN 1971
PACKING GROUP:	Not Applicable
HAZARD LABEL:	Class 2.1 (Flammable Gas)
SPECIAL PROVISIONS:	None
EXPLOSIVE LIMIT AND LIMITED QUANTITY INDEX:	0.12
ERAP INDEX:	3000
PASSENGER CARRYING SHIP INDEX:	Forbidden
PASSENGER CARRYING ROAD VEHICLE OR PASSENG	ER CARRYING RAILWAY VEHICLE INDEX: Forbidden
NORTH AMERICAN EMERGENCY RESPONSE GUIDEBO	OOK NUMBER (2000): 115

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15. REGULATORY INFORMATION

NOTE: Shipment of compressed gas cylinders via Public Passenger Road Vehicle is a violation of Canadian law (Transport Canada Transportation of Dangerous Goods Act, 1992)

ADDITIONAL U.S. REGULATIONS:

U.S. SARA REPORTING REQUIREMENTS: This gas is not subject to the reporting requirements of Sections 302, 304 and 313 of Title III of the Superfund Amendments and Reauthorization Act., as follows:

U.S. SARA THRESHOLD PLANNING QUANTITY: There are no specific Threshold Planning Quantities for this gas. The default Federal MSDS submission and inventory requirement filing threshold of 10,000 lb (4,540 kg) may apply, per 40 CFR 370.20. U.S. TSCA INVENTORY STATUS: Methane is listed on the TSCA Inventory.

U.S. CERCLA REPORTABLE QUANTITIES (RQ): Not applicable. U.S. STATE REGULATORY INFORMATION: Methano is accurated

U.S. STATE REGULATORY INFORMATION.	vietnane is covered under specific State regulations,	as denoted below.
Alaska - Designated Toxic and	Michigan - Critical Materials Register:	Pennsylvania - Hazardous Substance
Hazardous Substances: Methane.	No.	List: Methane.
California - Permissible Exposure	Minnesota - List of Hazardous	Rhode Island - Hazardous Substance
Limits for Chemical Contaminants:	Substances: Methane.	List: Methane.
Methane.	Missouri - Employer Information/Toxic	Texas - Hazardous Substance List:
Florida - Substance List: No.	Substance List: Methane.	No.
Illinois - Toxic Substance List:	New Jersey - Right to Know Hazardous	West Virginia - Hazardous Substance
Methane.	Substance List: Methane.	List: No.
Kansas - Section 302/313 List: No.	North Dakota - List of Hazardous	Wisconsin - Toxic and Hazardous
Massachusetts - Substance List: Methane.	Chemicals, Reportable Quantities: No.	Substances: No.

CALIFORNIA SAFE DRINKING WATER AND TOXIC ENFORCEMENT ACT (PROPOSITION 65): Methane is not on the California Proposition 65 list

OTHER U.S. FEDERAL REGULATIONS:

- Methane does not contain any Class I or Class II ozone depleting chemicals (40 CFR part 82). Methane is subject to the reporting requirements of Section 112(r) of the Clean Air Act. The Threshold Quantity for of this gas is 10,000 pounds.
- Depending on specific operations involving the use of this product, the regulations of the Process Safety Management of Highly Hazardous Chemicals may be applicable (29 CFR 1910.119). Under this regulation Methane is not listed in Appendix A, however, any process that involves a flammable gas on-site, in one location, in quantities of 10,000 lbs (4,553 kg) or greater is covered under this regulation unless it is used as a fuel.

A Debug a loss of the second secon

CANADIAN DSL/NDSL INVENTORY STATUS: Methane on the Canadian DSL Inventory.

CANADIAN ENVIRONMENTAL PROTECTION ACT (CEPA) PRIORITIES SUBSTANCES LISTS: Methane is not on the CEPA Priorities Substances Lis

CANADIAN WHMIS CLASSIFICATION: Methane is categorized as a Controlled Product, Hazard Classes A and B1, as per the Controlled Product Regulations

16. OTHER INFORMATION

INFORMATION ABOUT DOT-39 NRC (Non-Refillable Cylinder) PRODUCTS

DOT 39 cylinders ship as hazardous materials when full. Once the cylinders are relieved of pressure (empty) they are not considered hazardous material or waste. Residual gas in this type of cylinder is not an issue because toxic gas mixtures are prohibited. Calibration gas mixtures typically packaged in these cylinders are Nonflammable n.o.s., UN 1956. A small percentage of calibration gases packaged in DOT 39 cylinders are flammable gas mixtures.

For disposal of used DOT-39 cylinders, it is acceptable to place them in a landfill if local laws permit. Their disposal is no different than that employed with other DOT containers such as spray paint cars, household arensols, or disposable cylinders of propane (for camping, torch etc.). When feasible, we recommended recycling for scrap metal content. CALGAZ will do this for any customer that wishes to return cylinders to us prepaid. All that is required is a phone call to make arrangements so we may anticipate arrival. Scrapping cylinders involves some preparation before the metal dealer may accept them. We perform this operation as a service to valued customers who want to participate

MIXTURES: When two or more gases or liquefied gases are mixed, their hazardous properties may combine to create additional, unexpected hazards. Obtain and evaluate the safety information for each component before you produce the mixture. Consult an Industrial Hygienist or other trained person when you make your safety evaluation of the end product. Remember, gases and liquids have properties which can cause serious injury or death

Further information can be found in the following pamphlets published by: Compressed Gas Association Inc. (CGA), 1725 Jefferson Davis Highway, Suite 1004, Arlington, VA 22202-4102. Telephone: (703) 412-0900.

- P-1 "Safe Handling of Compressed Gases in Containers
- P-14 "Accident Prevention in Oxvaen-Rich and Oxvaen Deficient Atmospheres"
- "Use of Oxy-fuel Gas Welding and Cutting Apparatus" SB-8
- SB-2 "Oxygen Deficient Atmospheres"
 - "Handbook of Compressed Gases"

PREPARED BY:

CHEMICAL SAFETY ASSOCIATES, Inc. PO Box 3519, La Mesa, CA 91944-3519 619/670-0609



This Material Safety Data Sheet is offered pursuant to OSHA's Hazard Communication Standard, 29 CFR, 1910.1200. Other government regulations must be reviewed for applicability to this product. To the best of CALGAZ knowledge, the information contained herein is reliable and accurate as of this date; however, accuracy, suitability or completeness are not guaranteed and no warranties of any type, either express or implied, are provided. The information contained herein relates only to this specific product. If this product is combined with other materials, all component properties must be considered. Data may be changed from time to time. Be sure to consult the latest edition.

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AB MATERIAL SAFETY DATA SHEET (MSDS) NITROGEN

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Ref. No.: MS095		
1 PRODUCT AND COM		Extinguishing Media
Product Name	Nitrogen	As Nitrogen is an inert gas, it does not contribute to a fire, but coul
Chemical Formula	N ₂	help with the extinguishing by reducing the oxygen content of the a by dilution to below the level to support combustion.
Trade Names	Nitrogen, Compressed (Tec)	Specific Hazards
	Nitrogen, Instrument Grade	Nitrogen does not support life. It can act as a simple asphyxiant b
	Nitrogen, Pharmaceutical Grade	diluting the concentration of oxygen in the air below the levels t
	Nitrogen, ELCAP	support life.
Colour coding	Compressed, Instrument, ultra high purity	Emergency Actions
	& Pharmaceutical Grades have French	If possible, shut off the source of excess Nitrogen. Evacuate area
	Grey (H.30) bodies with black shoulders. Relevant decals/stencilling shall be on	All cylinders should be removed from the vicinity of the fire
	bodies of cylinders. ELCAP shall have a	Cylinders that cannot be removed should be cooled with water from
	Protea Pink (A.58) body, with "ELCAP"	safe distance. Cylinders which have been exposed to excessive here should be clearly identified and returned to supplier. CONTACT TH
	stencilled on body of the cylinder.	NEAREST AFROX BRANCH.
Valve	ELCAP No. 2 type-Brass 5/8inch BSP	Protective Clothing
	right hand female. All the other grades	Self-contained breathing apparatus. Safety gloves and shoes, o
	shall be fitted with 3 SN - Brass, 3/4 inch	boots, should be worn when handling cylinders.
	BSP right hand female valves.	Environmental Precautions
Company Identification	African Oxygen Limited 23 Webber Street	Nitrogen is lighter than air and disperses rapidly in the atmosphere
	Johannesburg, 2001	Care should be taken when entering a potentially oxygen-deficie
	Tel No: (011) 490-0400	environment. If possible, ventilate the affected area.
	Fax No: (011) 490-0506	6 ACCIDENTAL RELEASE MEASURES
EMERGENCY NUMBER	0860 020202 or (011) 873 4382	Personal Precautions
	(24 hours)	Do not enter any area where nitrogen has been spilled unless test
2 COMPOSITION/INFO	RMATION ON INGREDIENTS	have shown that it is safe to do so.
2 COMPOSITION/INFO	RMATION ON INGREDIENTS	Environmental Precautions
Chemical Name	Nitrogen	Nitrogen does not pose a hazard to the environment.
Chemical Family	Inert gas	Small Spills
CAS No.	7727-37-9	Shut off the source of escaping nitrogen. Ventilate the area.
UN No.	1066	Large Spills
ERG No.	121	Evacuate the area. Shut off the source of the spill if this can be don without risk. Restrict access to the area until completion of the clear
Hazchem Warning	2 C Non-flammable Gas	up procedure. Ventilate the area using forced-draught if necessary.
3 HAZARDS IDENTIFIC	ATION	7 HANDLING AND STORAGE
Main Hazards		Do not allow cylinders to slide or come into contact with sharp edge
	gas containers, and must be regarded as	Nitrogen cylinders may be stacked horizontally provided that they are
	nes. Nitrogen does not support life. It can it by diluting the concentration of oxygen in	firmly secured at each end to prevent rolling. Use a "first in - first ou
air below the levels neces	sarv to support life.	inventory system to prevent full cylinders from being stored for
Adverse Health Effects		excessive periods of time. Keep out of reach of children.
Inhalation of nitrogen in	excessive concentrations can result in	8 EXPOSURE CONTROLS/PERSONAL PROTECTION
	g, loss of consciousness and death.	
Chemical Hazards		Occupational Exposure Hazards
	nert to most materials under ordinary	As nitrogen is a simple asphyxiant, avoid any areas where spillag has taken place. Only enter once testing has proved the atmosphere
	nore reactive at elevated temperatures, and	to be safe.
combines with hydrogen, o Biological Hazards	No known effect.	Engineering Control Measures
Vapour Inhalation		Engineering control measures are preferred to reduce exposure
	ple asphyxiant death may result from errors	Oxygen-depleted atmospheres. General methods include forced
in judgement, confusion,	or loss of consciousness which prevents	draught ventilation, separate from other exhaust ventilation system
	gen concentrations, unconsciousness and	Ensure that sufficient fresh air enters at, or near floor level. Personal Protection
death may occur in secon	as without warning.	Self-contained breathing apparatus should always be worn whe
4 FIRST AID MEASURE	IS	entering area where oxygen depletion may have occurred. Safe
Eye/Skin Contact	No known effect.	goggles, gloves and shoes or boots should be worn when handlin
Ingestion	(See Section 3 above)	cylinders.
Inhalation		Skin No known effect.
Prompt medical attention	is mandatory in all cases of overexposure	9 PHYSICAL AND CHEMICAL PROPERTIES
to Nitrogen. Rescue pe	ersonnel should be equipped with self-	
	paratus. Conscious persons should be	PHYSICAL DATA
	ninated area and inhale fresh air. Quick	Chemical Symbol N2 Molecular Weight 28,013
	nated area is most important. Unconscious	Specific Volume @ 20°C & 101,325 kPa 861,5ml/g
	ved to an uncontaminated area, and given tion and supplemental oxygen.	Density, gas @ 101,325 kPa and 20°C 1,25 kg/m ³
		Relative density (Air = 1) @ 101,325 kPa 0,967
5 FIRE FIGHTING MEA	SURES	Colour None
		Taste None

DATE: April 2013	Version 2				Page 2 of 2
Ref. No.: MS095 Odour	None	•			
10 STABILITY AND REACT					
Conditions to avoid		NON-FLAMMA	ABLE GAS		
The dilution of the oxygen ca which cannot support life. No or for any other purpose than cylinders to excessive heat, pressure to rupture the cylind Incompatible Materials As Nitrogen is inert it may I any of the common metals withstand the pressures invol Hazardous Decomposition	be contained in systems constructed of which have been designed to safely ved.	ICAO/IA Class Packagir Packagir - Carg - Pass	ng group ng instructions	1066 2.2 200 200	
None		- Carg	0	150kg	
11 TOXICOLOGICAL INFO	RMATION		senger	75kg	
Acute Toxicity Skin & eye contact Chronic Toxicity	No known effect No known effect No known effect		ULATORY INFO		ammable
Carcinogenicity Mutagenicity	No known effect No known effect	Risk Phrase	Description	Safety Phrase	Description
Reproductive Hazards (For further information see	No known effect e Section 3. Adverse Health effects)	R20	Harmful by inhalation	S2	Keep out of reach of Child
	ATION d can cause pockets of oxygen depleted as. It does not pose a hazard to the	R44	Risk of explosion if heated under confinement	S9	Keep container in a ventilated place
13 DISPOSAL CONSIDERA	ATIONS			S15 S37	Keep way from heat Wear suitable gloves
Disposal Methods Small amounts may be bloc conditions. Large amounts	wn to the atmosphere under controlled should only be handled by the gas			S38	In case of insuffi ventilation, wear sui respiratory equipment
supplier. Disposal of Packaging				S51	Use only in well-ventil areas
	st only be handled by the gas supplier.		legislation Nor SABS 0265 for e		of the above
14 TRANSPORT INFORMA ROAD TRANSPORTATION	lion			·	
UN No ERG No Hazchem warning SEA TRANSPORTATION IMDG Class	1066 121 2C Non-flammable Gas 1066	Bibliogra Compres Handboo Matheso		ation, Arling d Gases – s Data Boo	3 ^{ra} Edition % – 6 th Edition
Packaging group	Non flowmobile goo	17 EXC	LUSION OF LIA	BILITY	
laper	Non-flammable gas	publication use of th	on. The compar	the use, a	cation is accurate at the da t accept liability arising from application, adaptation or pr

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MATERIAL SAFETY DATA SHEET

SECTION 1. PRODUCT IDENTIFICATION

PRODUCT NAME: CHEMICAL NAME: SYNONYMS: MANUFACTURER:

Oxygen, CompressedOxygenFORMULA: O2Oxygen gas, Gaseous Oxygen, GOXAir Products and Chemicals, Inc.7201HamiltonBoulevard

Allentown, PA 18195 - 1501

PRODUCT INFORMATION:

1-800-752-1597

MSDS NUMBER: 1012

REVISION DATE: January 1995

REVISION: 5 REVIEW DATE: August 1997**

SECTION 2. COMPOSITION / INFORMATION ON INGREDIENTS

Oxygen is sold as pure product > 99%. CAS NUMBER: 7782-44-7 EXPOSURE LIMITS: OSHA: Not established ACGIH: Not established NIOSH: Not established

SECTION 3. HAZARD IDENTIFICATION

EMERGENCY OVERVIEW

Oxygen is an odorless, colorless, nonflammable gas stored in cylinders at high pressure. It is an oxidizing gas and vigorously accelerates combustion. Keep away from oils or grease. Rescue personnel should be aware of the extreme fire hazards associated with oxygen-enriched (greater than 23%) atmospheres, and that self contained breathing apparatus (SCBA) may be required.

EMERGENCY TELEPHONE NUMBERS

(800) 523-9374 Continental U.S., Canada and Puerto Rico (610) 481-7711 other locations

POTENTIAL HEALTH EFFECTS INFORMATION:

INHALATION: Breathing 80% or more oxygen at atmospheric pressure for more than a few hours may cause nasal stuffiness, cough, sore throat, chest pain and breathing difficulty. Breathing oxygen at higher pressure increases the likelihood of adverse effects within a shorter time period. Breathing pure oxygen under pressure may cause lung damage and also central nervous system effects resulting in dizziness, poor coordination, tingling sensation, visual and hearing disturbances, muscular twitching, unconsciousness and convulsions. Breathing oxygen under pressure may cause prolongation of adaptation to darkness and reduced peripheral vision.

EYE / SKIN CONTACT: No adverse effect.

EXPOSURE INFORMATION:

ROUTE OF ENTRY:

Inhalation

TARGET ORGANS: Eyes, central nervous system

MEDICAL CONDITIONS AGGRAVATED BY OVEREXPOSURE: Patients with chronic obstructive pulmonary disease retain carbon dioxide abnormally. If oxygen is administered to them, raising the oxygen concentration in the blood depresses their breathing and raises their retained carbon dioxide to a dangerous level.

CARCINOGENIC POTENTIAL: Oxygen is not listed as a carcinogen or potential carcinogen by NTP, IARC, or OSHA Subpart Z.

SECTION 4. FIRST AID

INHALATION: Move victim to fresh air or if in elevated pressures reduce oxygen

pressures to one atmosphere. Call a physician. The physician should be advised that the victim has been

victim has been

exposed to a high concentration of oxygen. No treatment is required in the absence of symptoms or high pressure exposure.

EYE / SKIN CONTACT: Not applicable

NOTES TO PHYSICIAN: Animal studies suggest that the administration of certain drugs, including phenothiazine drugs and chloroquine, increase the susceptibility to toxicity from oxygen at high pressures. Animal studies also indicate that vitamin "E" deficiency may increase susceptibility to oxygen toxicity.

Airway obstruction during high oxygen tension may cause alveolar collapse following absorption of the oxygen. Similarly, occlusion of the Eustachian tubes may cause retraction of the eardrum and obstruction of the paranasal sinuses may produce "vacuum-type" headache.

All individuals exposed for long periods to oxygen at high pressure and who exhibit overt oxygen toxicity should have ophthalmologic examinations.

SECTION 5. FIRE AND EXPLOSION

FLASH POINT: Not applicable AUTOIGNITION: Nonflammable FLAMMABLE LIMITS: Nonflammable

EXTINGUISHING MEDIA: Oxygen is nonflammable but will support combustion. Use extinguishing media appropriate for surrounding fire.

HAZARDOUS COMBUSTION PRODUCTS: None

SPECIAL FIRE FIGHTING INSTRUCTIONS: Evacuate all personnel from the danger area. If possible, shut off flow of oxygen which is supporting the fire. Immediately cool containers with water spray from maximum distance. When cool move cylinders from fire area, if possible without risk. Self contained breathing apparatus may be required for rescue workers.

UNUSUAL FIRE AND EXPLOSION HAZARDS: Oxygen vigorously accelerates combustion. Some materials which are noncombustible in air will burn in the presence of an oxygen enriched atmosphere (greater than 23%). Fire resistant clothing may burn and offer no protection in oxygen rich atmospheres. Oxygen may form explosive compounds when exposed to combustible materials or oil, grease, and other hydrocarbon materials. Pressure in a

container can build up due to heat and it may rupture if pressure relief devices should fail to function. Upon exposure to intense heat or flame cylinder will vent rapidly and/or rupture violently. Most cylinders are designed to vent contents when exposed to elevated temperatures. Pressure in a container can build up due to heat and it may rupture if pressure relief devices should fail to function.

SECTION 6. ACCIDENTAL RELEASE MEASURES

Evacuate all personnel from affected area. Shut off source of oxygen if possible. Increase ventilation to release area. Personnel who have been exposed to high concentrations of oxygen should stay in a well-ventilated or open area for 30 minutes before going into a confined space or near an ignition source.

If leak is from container or its valve, call the Air Products emergency telephone number. If leak is in user's system close cylinder valve and vent pressure before attempting repairs.

SECTION 7. STORAGE AND HANDLING

STORAGE: Cylinders should be stored upright in a well-ventilated, secure area, protected from the weather. Storage area temperatures should not exceed 125 °F (52 °C) and area should be free of combustible materials. Storage should be away from heavily traveled areas and emergency exits. Avoid areas where salt or other corrosive materials are present. Cylinders should be separated from flammables by a minimum distance of 20 ft. or by a barricade of non-combustible material at least five ft. high having a fire resistance rating of at least 1/2 hour. Valve protection caps and valve outlet seals should remain on cylinders not connected for use. Separate full from empty cylinders. Avoid excessive inventory and storage time. Use a first-in first-out system. Keep good inventory records.

HANDLING: Do not drag, roll, or slide cylinder. Use a suitable handtruck designed for cylinder movement. Never attempt to lift a cylinder by its cap. Secure cylinders at all times while in use. Use a pressure reducing regulator or separate control valve to safely discharge gas from cylinder. Use a check valve to prevent reverse flow into cylinder. Do not overheat cylinder to increase pressure or discharge rate. Always open cylinder valve slowly. Do not use rapid opening valves (i.e., ball valves). If user experiences any difficulty operating cylinder valve, discontinue use and contact supplier. Never insert an object (e.g., wrench, screwdriver, pry bar, etc.) into valve cap openings. Doing so may damage valve causing a leak to occur. Use an adjustable strap-wrench to remove over-tight or rusted caps.

All gauges, valves, regulators, piping and equipment to be used in oxygen service must be cleaned for oxygen service in accordance with Compressed Gas Association pamphlet G-4.1.

Carbon steel, stainless steel, copper, brass, nickel and their alloys are materials of construction that can be used in oxygen service. Use piping and equipment adequately designed to withstand pressures to be encountered. Oxygen is not to be used as a substitute for compressed air. Never use an oxygen jet for cleaning purposes of any sort, especially clothing, as it increases the likelihood of an engulfing fire. Use a check valve or other protective apparatus in any line or piping from the cylinder to prevent reverse flow.

When used in welding and cutting read and understand the manufacturer's instructions and the precautionary label on the products. Never strike an arc on a compressed gas cylinder or make a cylinder a part of an electrical circuit.

SPECIAL REQUIREMENTS: Always store and handle compressed gases in accordance with Compressed Gas Association, Inc. (ph. 703-412-0900) pamphlet CGA P-1, *Safe Handling of*

Compressed Gases in Containers. Local regulations may require specific equipment for

storage or use.

CAUTION: Compressed gas cylinders shall not be refilled except by qualified producers of compressed gases. Shipment of a compressed gas cylinder which has not been filled by the owner or with the owner's written consent is a violation of federal law.

SECTION 8. PERSONAL PROTECTION / EXPOSURE CONTROL

ENGINEERING CONTROLS: Provide ventilation and/or local exhaust to prevent accumulation of high concentrations of gas (greater than 23%).

RESPIRATORY PROTECTION:

GENERAL USE: None required

EMERGENCY: Use SCBA do to possibility of fire when concentrations exceed 23%.

OTHER PROTECTIVE EQUIPMENT: Safety shoes and work gloves are recommended when handling cylinders. Clothing exposed to high concentrations may retain oxygen 30 minutes or longer and become a potential fire hazard. Stay away from ignition sources.

ECTION 9. PHYSICAL AND CHEMICAL PROPERTIES

APPEARANCE: Colorless gas **ODOR:** Odorless MOLECULAR WEIGHT: 32.0 **BOILING POINT (1 atm):** -297.3 °F (-183.0 °C) SPECIFIC GRAVITY (Air =1): 1.10 SPECIFIC VOLUME (at 70 °F 21.1 °C) and 1 atm): 12.08 ft³/lb (0.754 m³/kg)

FREEZING / MELTING POINT: -361.9 °F (-218.8 °C) VAPOR PRESSURE: Not applicable at 70°F GAS DENSITY (At 70°F (21.1°C) and 1 Atm): 0.083 lb /ft³ (1.326 kg/m³)

SOLUBILITY IN WATER (Vol./Vol. at 32°F (0°C)): 0.049

SECTION 10. REACTIVITY / STABILITY

CHEMICAL STABILITY: Stable

CONDITIONS TO AVOID: None

INCOMPATIBILITY: Oils, grease, hydrocarbons and flammable materials.

HAZARDOUS DECOMPOSITION PRODUCTS: None

HAZARDOUS POLYMERIZATION: Will not occur

SECTION 11. TOXICOLOGICAL INFORMATION

At atmospheric concentration and pressure, oxygen poses no toxicity hazards.

Premature infants exposed to high oxygen concentrations may suffer delayed retinal damage which can progress to retinal detachment and blindness. Retinal damage may also occur in adults exposed to 100% oxygen for extended periods (24 to 48 hr).

At two or more atmospheres central nervous system (CNS) toxicity occurs. Symptoms include nausea, vomiting, dizziness or vertigo, muscle twitching, vision changes, and loss

Appendix A-4. MSDS Information

of consciousness and generalized seizures. At three atmospheres, CNS toxicity occurs in less than two hours, and at six atmospheres in only a few minutes.

SECTION 12. ECOLOGICAL INFORMATION

The atmosphere contains 21% oxygen. No adverse ecological effects are expected. Oxygen does not contain any Class I or Class II ozone depleting chemicals. Oxygen is not listed as a marine pollutant by DOT (49 CFR 171).

SECTION 13. DISPOSAL

UNUSED PRODUCT / EMPTY CONTAINER: Return container and unused product to supplier. Do not attempt to dispose of residual or unused quantities.

DISPOSAL: For emergency disposal, secure cylinder and slowly discharge gas to the

atmosphere in a well ventilated area or outdoors.

SECTION 14. TRANSPORTATION

DOT HAZARD CLASS: 2.2 (Nonflammable Gas) **DOT SHIPPING LABEL:** Nonflammable Gas,

DOT SHIPPING NAME: Oxygen, compressed REPORTABLE QUANTITY (RQ): None Oxidizer IDENTIFICATION NUMBER: UN 1072 PLACARD: Nonflammable Gas or Oxygen

SPECIAL SHIPPING INFORMATION: Cylinders should be transported in a secure upright position in a well ventilated truck. Never transport in passenger compartment of a vehicle. An oxygen label may be used for domestic shipment in the United States and Canada in place of the Non-flammable and Oxidizer labels (49CFR Part 172).

SECTION 15. REGULATORY INFORMATION

U.S. FEDERAL REGULATIONS:

EPA - ENVIRONMENTAL PROTECTION AGENCY:

CERCLA: Comprehensive Environmental Response, Compensation, and Liability Act of 1980 requires notification to the National Response Center of releases of quantities of hazardous substances equal to or greater than the reportable quantities (RQ) in 40 CFR 302.4.

CERCLA Reportable Quantity: None

SARA TITLE III: Superfund Amendments and Reauthorization Act of 1986

SECTION 302: Requires emergency planning based on threshold planning quantities (TPQ) and release reporting based on reportable quantities (RQ) of EPA's extremely hazardous substances (40 CFR 355).

Oxygen is not listed as an Extremely Hazardous Substance.

SECTIONS 311/312: Require submission of material safety data sheets (MSDSs) and chemical inventory reporting with identification of EPA defined hazard classes. The hazard classes for this product are:



Appendix A-4. MSDS Information Ye PRESSURE: s REACTIVITY: No Ye FIRE: s

SECTION 313: Requires submission of annual reports of releases of toxic chemicals that appear in

40 CFR 372.

Oxygen is not listed as a toxic chemical.

40 CFR PART 68: Risk Management for Chemical Accident Release Prevention. Requires the development and implementation of risk management programs at facilities that manufacture, use, store, or otherwise handle regulated substances in quantities that exceed specified thresholds.

Oxygen is not listed as a regulated substance.

TOXIC SUBSTANCE CONTROL ACT (TSCA): Oxygen is listed on the TSCA inventory.

OSHA - OCCUPATIONAL SAFETY AND HEALTH ADMINISTRATION

29 CFR 1910.119: Process Safety Management of Highly Hazardous Chemicals. Requires facilities to develop a process safety management program based on Threshold Quantities (TQ) of highly hazardous chemicals.

Oxygen is not listed as a Highly Hazardous Chemical.

STATE REGULATIONS

CALIFORNIA:

Proposition 65:This product does NOT contain any listed substances for which the State of California requires warning under this statute.

SCAQMD Rule: VOC = Not applicable

Material Safety Data

Sheet	Appendix A-4. MSDS Information He a lt h
o-Xy	lene MSDS Fire 3
	Reactivit ₀ y
	Personal _H Protectio
	n
Section 1: Chemical Pro	duct and Company Identification
Product Name: o-Xylene	Contact Information:
Catalog Codes: SLX1012	Sciencelab.com, Inc.
CAS#: 95-47-6	14025 Smith Rd. Houston, Texas 77396
RTECS: ZE2450000	US Sales: 1-800-901-7247
TSCA: TSCA 8(b) inventory: o-	International Sales: 1-281-441-4400
Xylene	Order Online: ScienceLab.com
	CHEMTREC (24HR Emergency Telephone),
CI#: Not applicable.	call:
Synonym: 1,2-Dimethylbenzene	1-800-424-9300
Chemical Name: o-Xylene	International CHEMTREC, call: 1-703-527-3887
Chemical Formula: C6H4(CH3)2	For non-emergency assistance, call: 1-281-441- 4400

Section 2: Composition and Information on Ingredients				
Composition:				
Name	CAS #	% by Weight		
{o-}Xylene	95-47-6	100		

Toxicological Data on Ingredients: o-Xylene LD50: Not available. LC50: Not available.

Section 3: Hazards Identification
Potential Acute Health Effects: Hazardous in case of skin contact (irritant, permeator), of eye contact (irritant), of ingestion, of inhalation.

Potential Chronic Health Effects:

CARCINOGENIC EFFECTS: A4 (Not classifiable for human or animal.) by ACGIH, 3 (Not classifiable for human.) by IARC. MUTAGENIC EFFECTS: Not available. TERATOGENIC EFFECTS: Classified POSSIBLE for human. DEVELOPMENTAL TOXICITY: Classified Reproductive system/toxin/male [POSSIBLE]. The substance may be toxic to kidneys, liver, upper respiratory tract, skin, eyes, central nervous system (CNS). Repeated or prolonged exposure to the substance can produce target organs damage.

Section 4: First Aid Measures

Check for and remove any contact lenses. Immediately flush eyes with running water for at least 15 minutes, keeping eyelids open. Get medical attention.

Skin Contact:

In case of contact, immediately flush skin with plenty of water. Cover the irritated skin with an emollient. Remove contaminated clothing and shoes. Wash clothing before reuse. Thoroughly clean shoes before reuse. Get medical attention.

Serious Skin Contact:

Wash with a disinfectant soap and cover the contaminated skin with an anti-bacterial cream. Seek immediate medical attention.

Inhalation:

If inhaled, remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention.

Serious Inhalation:

Evacuate the victim to a safe area as soon as possible. Loosen tight clothing such as a collar, tie, belt or waistband. If breathing is difficult, administer oxygen. If the victim is not breathing, perform mouth-to-mouth resuscitation. WARNING: It may be hazardous to the person providing aid to give mouth-to-mouth resuscitation when the inhaled material is toxic, infectious or corrosive. Seek medical attention.

Ingestion:

Do NOT induce vomiting unless directed to do so by medical personnel. Never give anything by mouth to an unconscious person. If large quantities of this material are swallowed, call a physician immediately. Loosen tight clothing such as a collar, tie, belt or waistband.

Serious Ingestion: Not available.

Section 5: Fire and Explosion Data

Flammability of the Product: Flammable.

Auto-Ignition Temperature: 463°C (865.4°F)

Flash Points: CLOSED CUP: 17°C (62.6°F).

Flammable Limits: LOWER: 0.9% UPPER: 6.7%

Products of Combustion: These products are carbon oxides (CO, CO2).

Fire Hazards in Presence of Various Substances: Highly flammable in presence of open flames and sparks, of heat.

Explosion Hazards in Presence of Various Substances:

Risks of explosion of the product in presence of mechanical impact: Not available. Slightly explosive in presence of open flames and sparks, of heat.

Fire Fighting Media and Instructions:

Flammable liquid, insoluble in water. SMALL FIRE: Use DRY chemical powder. LARGE FIRE: Use water spray or fog.

Special Remarks on Fire Hazards:

Vapors are heavier than air and may travel considerable distance to source of ignition and flash back. When heated to decomposition it emits acrid smoke and irritating fumes.

Special Remarks on Explosion Hazards:

Explosive in the form of vapor when exposed to heat or flame. Vapors may form explosive mixtures with air. Containers may explode when heated. Runoff to sewer may create fire or explosion hazard

Section 6: Accidental Release Measures

Small Spill: Absorb with an inert material and put the spilled material in an appropriate waste disposal.

Large Spill:

Toxic flammable liquid, insoluble or very slightly soluble in water. Keep away from heat. Keep away from sources of ignition. Stop leak if without risk. Absorb with DRY earth, sand or other non-combustible material. Do not get water inside container. Do not touch spilled material. Prevent entry into sewers, basements or confined areas; dike if needed. Call for assistance on disposal. Be careful that the product is not present at a concentration level above TLV. Check TLV on the MSDS and with local authorities.

Section 7: Handling and Storage

Precautions:

Keep locked up.. Keep away from heat. Keep away from sources of ignition. Ground all equipment containing material. Do not ingest. Do not breathe gas/fumes/ vapor/spray. Wear suitable protective clothing. In case of insufficient ventilation, wear suitable respiratory equipment. If ingested, seek medical advice immediately and show the container or the label. Avoid contact with skin and eyes. Keep away from incompatibles such as oxidizing agents, acids.

Storage:

Store in a segregated and approved area. Keep container in a cool, well-ventilated area. Keep container tightly closed and sealed until ready for use. Avoid all possible sources of ignition (spark or flame).

Section 8: Exposure Controls/Personal Protection

Engineering Controls:

Provide exhaust ventilation or other engineering controls to keep the airborne concentrations of vapors below their respective threshold limit value. Ensure that eyewash stations and safety showers are proximal to the work-station location.

Personal Protection:

Splash goggles. Lab coat. Vapor respirator. Be sure to use an approved/certified respirator or equivalent. Gloves.

Personal Protection in Case of a Large Spill:

Splash goggles. Full suit. Vapor respirator. Boots. Gloves. A self contained breathing apparatus should be used to avoid inhalation of the product. Suggested protective clothing might not be sufficient; consult a specialist BEFORE handling this product.

Exposure Limits:

TWA: 434 STEL: 651 (mg/m3) from ACGIH (TLV) [United States] TWA: 100 STEL: 150 (ppm) from ACGIH (TLV) [United States] STEL: 150 (ppm) from NIOSH STEL: 655 (mg/m3) from NIOSHConsult local authorities for acceptable exposure limits.

Section 9: Physical and Chemical Properties

Physical state and appearance: Liquid. (Mobile, nonpolar liquid.)

Odor: Aromatic. Sweetish.

Taste: Not available.

Molecular Weight: 106.17 g/mole

Color: Colorless.

- pH (1% soln/water): Not applicable.
- Boiling Point: 144.4°C (291.9°F)

Melting Point: -25°C (-13°F)

Critical Temperature: 359°C (678.2°F)

Specific Gravity: 0.88 (Water = 1)

Vapor Pressure: 0.9 kPa (@ 20°C)

Vapor	Density:	3.7	(Air	=	1)
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Appendix A-4. MSDS Information

Volatility: Not available.

Odor Threshold: 0.05 ppm

Water/Oil Dist. Coeff.: The product is more soluble in oil; log(oil/water) = 3.1

lonicity (in Water): Not available.

Dispersion Properties:

Dispersed in diethyl ether. Is not dispersed in cold water, hot water. See solubility in diethyl ether, acetone.

Solubility:

Soluble in diethyl ether, acetone. Insoluble in cold water, hot water.

Section 10: Stability and Reactivity Data

Stability: The product is stable.

Instability Temperature: Not available.

Conditions of Instability: Heat, ignition sources, flames, incompatible materials.

Incompatibility with various substances: Reactive with oxidizing agents, acids.

Corrosivity: Non-corrosive in presence of glass.

Special Remarks on Reactivity:

Photochemically reactive. Incompatible with strong oxidizers(e.g. chlorine, bromine, fluorine), and strong acids (e.g. nitric acid, acetic acid).

Special Remarks on Corrosivity: Not available.

Polymerization: Will not occur.

Section 11: Toxicological Information

Routes of Entry: Absorbed through skin. Dermal contact. Eye contact. Inhalation.

Toxicity to Animals:

Lowest Published Lethal Dose - Inhalation (LCL): 6125 ppm 12 hours [Rat]; 6125 ppm 12 hours [Human] Lowest Published Lethal Dose - Oral: 5000 mg/kg [Rat]

Chronic Effects on Humans:

CARCINOGENIC EFFECTS: A4 (Not classifiable for human or animal.) by ACGIH, 3 (Not classifiable for human.) by IARC. TERATOGENIC EFFECTS: Classified POSSIBLE for human. DEVELOPMENTAL TOXICITY: Classified Reproductive system/ toxin/male [POSSIBLE]. May cause damage to the following organs: kidneys, liver, upper respiratory tract, skin, eyes, central nervous system (CNS).

Other Toxic Effects on Humans: Hazardous in case of skin contact (irritant, permeator), of ingestion, of

inhalation.

Special Remarks on Toxicity to Animals: Not available.

Special Remarks on Chronic Effects on Humans:

May cause adverse reproductive effects (male) and birth defects based on animal data. 0347 Animal: embryotoxic, foetotoxic, passes through the placental barrier. 0900 Detected in maternal milk in human. Narcotic effect; may cause nervous system disturbances.

Special Remarks on other Toxic Effects on Humans:

Acute Potential Health Effects Skin: May cause skin irritation. May be absorbed through skin i harmful amounts. Eves: Causes severe eve irritation. Inhalation: Causes respiratory tract and mucous membranes irritation. May affect sense organs, behavior (Central Nervous system) which may result in dizziness, general weakness, central nervous system depression, confusion, ataxia, disorientation, lethargy, drowsiness, headaches. May also affect respiration, cardiovascular system, liver, blood, and digestive system (nausea, vomiting) Ingestion: Harmful if swallowed. Causes digestive tract irritation with nausea, vomiting

and diarrhea. May also affect metabolism, liver, and urinary system, and central nervous system (excitement followed by headache, dizziness, drowsiness and nausea). Chronic Potential Health Effects: Skin: Prolonged or repeated contact may cause defatting of skin and dermatitis. Eyes: Prolonged or repeated exposure may cause conjunctivitis or permanent eye damage. Inhalation: Chronic inhalation may cause effects similar to those of acute inhalation.

Section 12: Ecological Information

Ecotoxicity: Not available.

BOD5 and COD: Not available.

Products of Biodegradation:

Possibly hazardous short term degradation products are not likely. However, long term degradation products may arise.

Toxicity of the Products of Biodegradation: The products of degradation are less toxic than the product itself.

Special Remarks on the Products of Biodegradation: Not available.

Section 13: Disposal Considerations

Waste Disposal:

Waste must be disposed of in accordance with federal, state and local environmental control regulations.

Section 14: Transport Information

DOT Classification: CLASS 3: Flammable liquid.

Identification: : Xylene UNNA: 1307 PG: III

Special Provisions for Transport: Not available.

Section 15: Other Regulatory Information

Federal and State Regulations:

Connecticut hazardous material survey.: o-Xylene Illinois chemical safety act: o-Xylene New York release reporting list: o-Xylene Pennsylvania RTK: o-Xylene Florida: o-Xylene Massachusetts RTK: o-Xylene Massachusetts spill list: o-Xylene New Jersey: o-Xylene New Jersey spill list: o-Xylene Louisiana spill reporting: o-Xylene California Director's List of Hazardous Substances: o-Xylene TSCA 8(b) inventory: o-Xylene TSCA 8(d) H and S data reporting: o-Xylene: Effective: 10/4/82; Sunset: 10/4/92 SARA 313 toxic chemical notification and

release reporting: o-Xylene CERCLA: Hazardous substances.: o-Xylene: 1000 lbs. (453.6 kg)

Other Regulations:

OSHA: Hazardous by definition of Hazard Communication Standard (29 CFR 1910.1200). EINECS: This product is on the European Inventory of Existing Commercial Chemical Substances.

Other Classifications:

WHMIS (Canada):

CLASS B-2: Flammable liquid with a flash point lower than 37.8°C (100°F). CLASS D-2A: Material causing other toxic effects (VERY TOXIC).

DSCL (EEC):

HMIS (U.S.A.):

Health Hazard: 2

Fire Hazard: 3

Reactivity: 0

Personal Protection: h

National Fire Protection Association (U.S.A.):

Health: 2

Flammability: 3

Reactivity: 0

Specific hazard:

Protective Equipment:

Gloves. Lab coat. Vapor respirator. Be sure to use an approved/certified respirator or equivalent. Wear appropriate respirator when ventilation is inadequate. Splash goggles.

Section 16: Other Information

References:

-Hawley, G.G.. The Condensed Chemical Dictionary, 11e ed., New York N.Y., Van Nostrand Reinold, 1987. -Material safety data sheet emitted by: la Commission de la Santé et de la Sécurité du Travail du Québec. -The Sigma-Aldrich Library of Chemical Safety Data, Edition II. -Guide de la loi et du rÃ[°]glement sur le transport des marchandises dangeureuses au canada. Centre de conformité internatinal Ltée. 1986.

Other Special Considerations: Not available.

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MATERIAL SAFETY DATA SHEET FOR ODORIZED PROPANE

1. Chemical Product and Company Identification				
Product Name: Odorized Commercial Propar Chemical Name: Propane Chemical Family: Paraffinic Hydrocarbon Formula: C ₃ H ₈ Synonyms: Dimethylmethane, LP-Gas, Lique		ppane, Propyl Hydride	<u>Name & Address:</u> AmeriGas Propane, L.P. P. O. Box 965 Valley Forge, PA. 19482	
Transportation Emergency Number	r:		For General Information, Call:	
CHEMTREC 1-800-424-9300	_		1-610-337-1000, Safety Dept.	
2. Composition / Information on Ingredients				
INGREDIENT NAME /CAS NUMBER Propane / 74-98-6 Ethane / 74-84-0 Propylene / 115-07-1 Butanes / 106-97-8 Ethyl Mercaptan / 75-08-1.	0 - 7.0 0 - 5.0 0 - 2.5	OSHA PEL 1,000 ppm 0.5 ppm	ACGIH TLV Simple asphyxiant Simple asphyxiant Simple asphyxiant 0.5 ppm	
WARNING: The intensity of the chemical of				

adsorption or absorption. Individuals with nasal perception problems may not be able to smell the odorant. Leaking propane from underground gas lines may lose its odor as it passes through certain soils. No odorant is effective 100% of the time. Therefore, circumstances can exist when individuals are in the presence of leaking propane and not be alerted by the smell. Contact AmeriGas for more information about odor, propane gas detectors and other safety considerations associated with the handling, storage and use of propane.

3. Hazards Identification

EMERGENCY OVERVIEW

HEALTH HAZARD

(Blue)

Minimal 0

Slight 1

FIRE HAZARD

SPECIAL

Moderate 2

Serious 3

(Red)

REACTIVITY

*(Ref. NFPA 704)

(Yellow)

HAZARDS*

Severe 4

DANGER! Flammable liquefied gas under pressure. Keep away from heat, sparks, flame, and all other ignition sources. Vapor replaces oxygen available for breathing and may cause suffocation in confined spaces. Use only with adequate ventilation. Reliance upon detection of odor may not provide adequate warning of potentially hazardous concentrations. Vapor is heavier than air; may collect at low levels. Liquid can cause freeze burn similar to frostbite. Do not get liquid in eyes, on skin, or on clothing. Avoid breathing vapor. Keep service valve closed when not in use.

POTENTIAL HEALTH EFFECTS INFORMATION

ROUTES OF EXPOSURE:

Inhalation: Asphyxiation. Before suffocation could occur, the lower flammability limit of propane in air would be exceeded, possibly causing both an oxygen-deficient and explosive atmosphere. Exposure to concentrations >10% may cause dizziness. Exposure to atmospheres containing 19% or less oxygen will bring about unconsciousness without warning. Lack of sufficient oxygen may cause serious injury or death.

Eye Contact: Contact with liquid can cause freezing of tissue.

Skin Contact: Contact with liquid can cause frostbite.

Skin Absorption: None.

Ingestion: Ingestion is not expected to occur in normal use. However, liquid can cause freeze burn similar to frostbite.

CHRONIC EFFECTS: None.

CARCINOGENICITY: Propane is not listed by NTP, OSHA or IARC.

4	ŀ.	First	Aid	Meas	ures

INHALATION: Individuals suffering from lack of oxygen should be removed to fresh air. If victim is not breathing, administer artificial respiration. If breathing is difficult, administer oxygen. Obtain immediate medical assistance.

EYE CONTACT: Gently flush eyes with lukewarm water. Obtain immediate medical assistance.

SKIN CONTACT: Remove saturated clothes, shoes and jewelry. Immerse affected area in lukewarm water not exceeding 105° F. Keep immersed. Obtain immediate medical assistance.

1

INGESTION: If swallowed, obtain immediate medical assistance.

5. Fire Fighting Measures

IGNITION TEMPERATURE IN AIR: 920°F to 1120°F (493°C to 549°C) FLAMMABLE LIMITS IN AIR (% by

volume): Lower: 2.15% Upper: 9.6%

EXTINGUISHING MEDIA: Dry chemical, CO₂, water spray or fog for surrounding area. Do not attempt to extinguish fire until propane source is isolated.

SPECIAL FIRE-FIGHTING INSTRUCTIONS: Evacuate all unnecessary personnel from the area. Allow only properly trained and protected emergency response personnel in area. A NIOSH approved self-contained breathing apparatus may be required. If gas flow cannot be shut off, <u>do not attempt to extinguish fire</u>. Allow fire to burn itself out. Use high volume water supply to cool exposed pressure containers and nearby equipment. Approach a flame-enveloped container from the sides, never from the ends. Use extreme caution when applying water to a container that has been exposed to heat or flame for more than a short time. For uncontrollable fires and/or when flame is impinging on container, withdraw all personnel and evacuate vicinity immediately.

UNUSUAL FIRE AND EXPLOSION HAZARDS: Propane is heavier than air and can collect in low areas. Flash back along a vapor trail is possible. Pressure in a container can build up due to heat; and, container may rupture suddenly and violently without warning if pressure relief devices fail to function properly. If flames are against the container, withdraw immediately on hearing a rising sound, if venting increases in volume or intensity or if there is discoloration of the container due to fire. Propane released from a properly functioning relief valve on an overheated container can also become ignited.

HAZARDOUS COMBUSTION PRODUCTS: None.

6. Accidental Release Measures

IF MATERIAL IS RELEASED OR SPILLED: Evacuate the immediate area. Eliminate any possible sources of ignition and provide maximum ventilation. Shut off source of propane, if possible. If leaking from container or valve, contact your supplier or AmeriGas immediately.

7. Handling and Storage

HANDLING PRECAUTIONS: Propane vapor is heavier than air and can collect in low areas that are without sufficient ventilation. Conduct system checks for leaks with a leak detector or solution, never with flame. Make certain the container service valve is shut off prior to connecting or disconnecting. If container valve does not operate properly, discontinue use and contact AmeriGas. Never insert an object (e.g., wrench, screwdriver, pry bar, etc.) into pressure relief valve or cylinder valve cap openings. Do not drop or abuse cylinders. Never strike an arc on a gas container or make a container part of an electrical circuit. See Section 16, "OTHER INFORMATION", for additional precautions.

STORAGE PRECAUTIONS: Store in a safe, authorized location (outside, detached storage is preferred) with adequate ventilation. Specific requirements are listed in NFPA 58, LP-GAS CODE. Isolate from heat and ignition sources. Containers should never be allowed to reach temperature exceeding 125°F (52°C). Isolate from combustible materials. Provide separate storage locations for other compressed and flammable gases. Propane containers should be separated from oxygen cylinders or other oxidizers by a minimum distance of 20 feet, or by a barrier of non-combustible material at least 5 feet high having a fire rating of at least 1/2 hour. Full and empty cylinders should be segregated. Keep cylinders in an upright position at all times so that each pressure relief valve communicates with the vapor space. Keep container valve closed and plugged or capped when not in use. Install protective caps when cylinders are not connected for use. Empty containers retain some residue and should be treated as if they were full.

8. Exposure Control / Personal Protection

ENGINEERING CONTROLS

Ventilation: Provide ventilation adequate to ensure propane does not reach a flammable mixture.

RESPIRATORY

PROTECTION

General Use: None.

Emergency Use: If concentrations are high enough to warrant supplied-air or NIOSH self-contained breathing apparatus, then the atmosphere may be flammable (See Section 5). Appropriate precautions must be taken regarding flammability.

PROTECTIVE CLOTHING: Avoid skin contact with liquid propane because of possibility of freeze burn. Wear gloves and protective clothing that are impervious to the product for the duration of the anticipated exposure.

EYE PROTECTION: Safety glasses, goggles or face shields are recommended when handling cylinders.

9. Physical and Chemical Properties

BOILING POINT: @ 14.7 psia = -44° F (@ 1.00 atm.pressure = -42° C) **SPECIFIC GRAVITY OF VAPOR** (Air = 1) at 60° F (15.56°C): 1.50 **SPECIFIC GRAVITY OF LIQUID** (Water = 1) at 60° F: 0.504

VAPOR PRESSURE: @ 70° F (20°C) = 127 psig; @ 105° F (45°C) = 210 psig; @ 130°F (55°C) = 287 psig

EXPANSION RATIO (From liquid to gas @ 14.7 psia): 1

to 270 SOLUBILITY IN WATER: Slight, 0.1 to 1.0%

APPEARANCE AND ODOR: A colorless and tasteless gas at normal temperature and pressure. An odorant (ethyl mercaptan) is added to provide a strong unpleasant odor. Should a propane-air mixture reach the lower limits of flammability, the ethyl mercaptan concentration will be approximately 0.5 ppm in air.

ODORANT WARNING: Odorant is added to aid in the detection of leaks. One common odorant is ethyl mercaptan, CAS No. 75-08-1. Odorant has a foul smell. The ability of people to detect odors varies widely. Also, the odor level can be reduced by certain chemical reactions with material in the propane system or when fugitive propane gas from underground leaks passes through certain soils. No odorant will be 100% effective in all circumstances. If the presence of the odorant is not obvious, notify AmeriGas immediately.

10. Stability and Reactivity

STABILITY: Stable.

Conditions to Avoid: Keep away from high heat, strong oxidizing agents and sources of ignition.

REACTIVITY:

Hazardous Decomposition Products: Under fire conditions, fumes, smoke, carbon monoxide, aldehydes and other decomposition products. In most applications where there is inadequate venting to the outside air, incomplete combustion will produce carbon monoxide (a toxic gas) and potentially develop concentrations that can create a serious health hazard. **Hazardous Polymerization:** Will not occur.

11. Toxicological Information

Propane is non-toxic and is a simple asphyxiant. It has slight anesthetic properties. Higher concentrations

may cause dizziness.

IRRITANCYOFMATERIAL:None.REPRODUCTIVEEFFECTS:NoneTERATOGENICITY: None

SENSITIZATION TO MATERIAL: None MUTAGENICITY: None SYNERGISTIC MATERIALS: None

12. Ecological Information

No adverse ecological effects are expected. Propane does not contain any Class I or Class II ozone-depleting chemicals (40 CFR Part 82). Propane is not listed as a marine pollutant by DOT (49 CFR Part 171).

13. Disposal Considerations

WASTE DISPOSAL METHOD: Do not attempt to dispose of residual or unused product in the container; return it to your supplier or contact AmeriGas for safe disposal. Residual product within a process system may be burned at a controlled rate if a suitable burning unit is available on site, and is done in accordance with federal, state and local regulations.

14. Transport Information

·	SHIPPING LABEL (S): Flammable Gas PLACARD (WHEN REQUIRED): Flammable Gas
IMO IDENTIFICATION NUMBER: UN 1978 HAZARD CLASS: 2.1 (Flammable Gas) PRODUCT RQ: None	SPECIAL SHIPPING INFORMATION: Container must be transported in a well-ventilated vehicle, secured, and in a position such that the pressure relief device is in communication with the vapor space.

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15. Regulatory Information

The following information concerns U.S. Federal regulatory requirements potentially applicable to this product. Not all such requirements are identified. Users of this product are responsible for their own regulatory compliance on a federal, state [provincial] and local level.

U.S. FEDERAL REGULATIONS

Environmental Protection Agency (EPA)

Comprehensive Environmental Response, Compensation and Liability Act of 1980 (CERCLA) - 40 CFR Parts 117 and 302

Reportable Quantity (RQ): None

Superfund Amendment and Reauthorization Act (SARA)

Sections 302/304: Relates to emergency planning on threshold planning quantities (TPQ) and release reporting based on reportable quantities (RQ) of EPA's extremely hazardous substances (40 CFR Part 355).

Extremely Hazardous Substances: None Threshold Planning Quantity (TPQ): None

- Sections 311/312: Relates to submission of material safety data sheets (MSDSs) and chemical inventory reporting with identification of EPA-defined hazard classes (40 CFR Part 370). The hazard classes for this product are:
 IMMEDIATE: No PRESSURE: Yes DELAYED: No REACTIVITY: No FLAMMABLE: Yes
- Section 313: Relates to submission of annual reports of release of toxic chemicals that appear in 40 CFR Part 372. Propane does not require reporting under Section 313.

Toxic Substance Control Act

(TSCA) Propane is listed on the TSCA inventory.

<u>Occupational Safety and Health Administration</u> (OSHA) The following 29 CFR Parts may apply to propane:

29 CFR 1910.110: Storage and Handling of Liquefied Petroleum Gases

29 CFR 1910.119: *Process Safety Management of Highly Hazardous Chemicals* **29 CFR 1910.1200:** *Hazardous Communications*

Food and Drug Administration (FDA)

21 CFR 184.1655: <u>Generally recognized as safe (GRAS)</u> as a direct human food ingredient when used as a propellant, aerating agent and gas.

16. Other Information

SPECIAL PRECAUTIONS: Use piping and equipment adequately designed to withstand pressure to be encountered. NFPA 58, LP-GAS CODE and OSHA 29 CFR 1910.10 require that all persons employed in handling LP-gases be trained in proper handling and operating procedures, which the employer shall

document. Contact your propane supplier or AmeriGas to arrange for the required training. Allow only trained and qualified persons to install and service propane containers and systems.

ISSUE INFORMATION

Issue Date: December 2002 Issued By: Director of Safety

Supersedes Date: April 2002 Phone Number: 1-610-337-7000

This material safety data sheet and the information it contains is offered to you in good faith as accurate. This Supplier does not manufacture this product, but is a supplier of the product that is independently produced by others. Much of the information contained in this data sheet was received from sources outside our Company. To the best of our knowledge this information is accurate, but this Supplier does not guarantee its accuracy or completeness. Health and safety precautions in this data sheet may not be adequate for all individuals and/or situations. It is the user's obligation to evaluate and use this product safely, comply with all applicable laws and regulations and to assume the risks involved in the use of this product.

GAS INNOVATIONS Appendix A-4. MSDS Information

PROPYLENE

CHEMICAL PRODUCT AND COMPANY IDENTIFICATION

GAS INNOVATIONS	Emergency (Emergency Contact: 3 E Company 866-303-2640				
18005 E. Hwy 225	3 E Company					
La Porte, TX 77571 Information: 281-471-2	200 Calls Origina	ating Outside the US:				
	281-471-2200) (Collect Calls Accepted)				
SUBSTANCE: PROPYL						
		Date prepared: September 7, 2007				
PRODUCT	D.O.T. SHIPPING NAME	Propylene				
IDENTIFICATION	■SYNONYM (S)	Propene				
	D.O.T. I.D. NUMBER	UN-1077				
	D.O.T. HAZZARD CLASS	2.1 Flammable Gas				
	D.O.T. LABEL (S)	Flammable Gas				
	C.A.S. NUMBER	115-07-01				
	CHEMICAL FORMULA	C_3H_6 or CH_3CH : CH_2				
PHYSICAL DATA	MOLECULAR WEIGHT	42.081				
		-185.2°C, -301.4°F				
		-47.7°C, -53.9 °F				
	 VAPOR PRESSURE SPECIFIC VOLUME 	942 kPa (gauge), 136.5 psig				
	•RELATIVE DENSITY,					
	(air=1)	0.567 m³/kg, 9.06ft₃/lb @ 1 atm, 21.1°C				
	SOLUBILITY IN WATER	1.48 @ 1 atm, 20°C				
	DESCRIPTION	22.05 cm ³ /100 ml @ 1 atm, 20°C				
		At room temperature and atmospheric pressure propylen is a colorless, flammable, relatively nontoxic gas, with characteristic natural gas odor. It is shipped as a liquefie				
		gas under its own vapor pressure.				
FIRE AND EXPLOSION	•FLAMMABLE LIMITS					
HAZARD DATA	IN AIR	2.0 – 11.1 % by volume				
	■AUTO-IGNITION					
	TEMPERATURE	480.0°C, 896.0°F				
	■FIRE FIGHTING	The only safe way to extinguish a propylene fire is to				
	PROCEDURES	stop the flow of gas. IF the flow cannot be stopped, let th fire burn out while cooling the cylinder and th surroundings using a water spray. Personnel may have t wear approach-type protective suits and positive pressur self-contained breathing apparatus. Firefighters' turnou gear may be inadequate.				
	UNUSUAL HAZARDS	1. Cylinders exposed to fire may rupture with viole				
	- UNUSUAL HAZARDS 213					

force. Extinguish surrounding fire and keep cylinders cool

by applying water from a maximum possible distance with a water

spray.

 Flammable gases may spread from a spill after the fire is extinguished and be subject to re-ignition.

GAS INNOVATIONS

Appendix A-4. MSDS Information **MSDS-PROPYLENE**

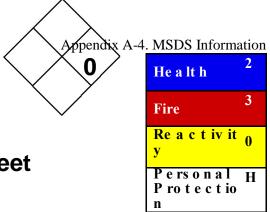
PAGE 2 OF3

HEALTH HAZARD DATA	 PERMISSIBLE EXPOSURE LIMITS ACCUTE EFFECTS OVEREXPOSURE 	OSHA TWA None established. ASGIH TWA None established. Propylene is nontoxic but can act as a simple asphyxiant by displacing air. Symptoms of asphyxia include rapid respirations, dizziness and fatigue. Contact with the liquid phase or with the cold gas escaping from cylinder may cause frostbite.
	CHRONIC EFFECTS OF OVEREXPOSURE	None known.
FIRST AID INFORMATION	 INHALATION 	Move victim to fresh air. If not breathing, give artificial respiration, preferably mouth-to-mouth. If breathing is difficult, give
	 CONTACT 	oxygen. Call a physician. Treat for frostbite.
REACTIVITY DATE	 STABILITY INCOMPATIBILITY HAZARDOUS DECOMPOSITION/ 	(X) Stable. () Unstable. Oxidizing material. Carbon monoxide, carbon dioxide
	OXIDATION PRODUCTS	
	POLYMERIZATION	(X) Will not occur () May Occur
SPILL OR LEAKAGE PROCEDURE	C C	entilate the area. For controlling large flows, personnel may ctive suits and self-contained breathing apparatus.
PRECAUTIONS	 STORAGE RECOMMENDATIONS 	Cylinders should be stored and used in dry, well- ventilated areas away from sources of heat or ignition. Do not store with oxidizers.
	 PERSONAL PROTECTIVE EQUIPMENT 	 Eye protection – Safety glasses should be worn. Respiratory protection – Approved respiratory equipment must bet be worn when airborne concentrations exceed safe levels. Skin protection – No special equipment is required. Gloves are recommended for cylinder handling.
	BEFORE USING THE GAS	 Secure the cylinder to prevent it from falling or being knocked over. Install check valves or traps to prevent suckback to the
		cylinder.3. Ground all lines and equipment.4. Leak check the lines and equipment.
	compiled form reference mate	 Have an emergency plan covering steps to be taken in the even of an accidental release. guaranteed by Gas Innovations or its affiliates, nor is any responsibility assumed or implied for any loss
		S's accuracy or completeness is not or damage resulting

Appendix A-4. MSDS Information

from inaccuracies or omissions. Since conditions of use are beyond our control, no warranties of merchantability of fitness for a particular purpose are expressed or implied. This MSDS is not intended as a license to operate under, or recommendation to infringe on, any patents. Appropriate warnings and safe handling procedures should be provided to handlers and users





Material Safety Data Sheet

p-Xylene MSDS

Section 1: Chemical	Product and Company Identification		
Product Name: p-Xylene	Contact Information:		
Catalog Codes: SLX1120	Sciencelab.com, Inc.		
CAS#: 106-42-3	14025 Smith Rd. Houston, Texas 77396		
RTECS: ZE2625000	US Sales: 1-800-901-7247		
TSCA: TSCA 8(b) inventory: p- Xylene Cl#: Not applicable.	International Sales: 1-281-441-4400 Order Online: ScienceLab.com CHEMTREC (24HR Emergency Telephone), call:		
Synonym: p-Methyltoluene	1-800-424-9300		
Chemical Name: 1,4-Dimethylbenzene	International CHEMTREC, call: 1-703-527-3887		
Chemical Formula: C6H4(CH3)2	For non-emergency assistance, call: 1-281-441- 4400		

Section 2: Composition and Information on Ingredients			
Composition:			
Name	CAS #	% by Weight	
{p-}Xylene	106-42-3	100	
Toxicological Data on Ingredients: p-Xylene: ORAL (LD50): Acute: 5000 mg/kg [Rat] DERMAL (LD50):			

Toxicological Data on Ingredients: p-Xylene: ORAL (LD50): Acute: 5000 mg/kg [Rat.]. DERMAL (LD50) Acute: 12400 mg/ kg [Rabbit.]. VAPOR (LC50): Acute: 4550 ppm 4 hour(s) [Rat].

Section 3: Hazards Identification	

Potential Acute Health Effects:

Very hazardous in case of skin contact (irritant), of eye contact (irritant). Slightly hazardous in case of skin contact (permeator), of ingestion, of inhalation. Inflammation of the eye is characterized by redness, watering, and itching. Skin inflammation is characterized by itching, scaling, reddening, or, occasionally, blistering.

Potential Chronic Health Effects:

Hazardous in case of skin contact (irritant), of eye contact (irritant). Slightly hazardous in case of skin contact (permeator), of ingestion, of inhalation. CARCINOGENIC EFFECTS: Not available. MUTAGENIC EFFECTS: Not available. TERATOGENIC EFFECTS: Not available. DEVELOPMENTAL TOXICITY: Not available. The substance is toxic to blood, kidneys, the nervous system, liver. Repeated or prolonged exposure to the substance can produce target organs damage.

Section 4: First Aid Meas	ures
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Appendix A-4. MSDS Information

Eye Contact: Check for and remove any contact lenses. Do not use an eye ointment. Seek medical attention.

Skin Contact:

After contact with skin, wash immediately with plenty of water. Gently and thoroughly wash the contaminated skin with running water and non-abrasive soap. Be particularly careful to clean folds, crevices, creases and groin. Cover the irritated skin with an emollient. If irritation persists, seek medical attention. Wash contaminated clothing before reusing.

Serious Skin Contact:

Wash with a disinfectant soap and cover the contaminated skin with an anti-bacterial cream. Seek immediate medical attention.

Inhalation: Allow the victim to rest in a well ventilated area. Seek immediate medical attention.

Serious Inhalation: Not available.

Ingestion:

Do not induce vomiting. Examine the lips and mouth to ascertain whether the tissues are damaged, a possible indication that the toxic material was ingested; the absence of such signs, however, is not conclusive. Loosen tight clothing such as a collar, tie, belt or waistband. If the victim is not breathing, perform mouth-to-mouth resuscitation. Seek immediate medical attention.

Serious Ingestion: Not available.

Section 5: Fire and Explosion Data

Flammability of the Product: Flammable.

Auto-Ignition Temperature: 527°C (980.6°F)

Flash Points: CLOSED CUP: 25°C (77°F). OPEN CUP: 28.9°C (84°F) (Cleveland).

Flammable Limits: LOWER: 1.1% UPPER: 7%

Products of Combustion: These products are carbon oxides (CO, CO2).

Fire Hazards in Presence of Various Substances: Highly flammable in presence of open flames and sparks, of heat.

Explosion Hazards in Presence of Various Substances:

Risks of explosion of the product in presence of mechanical impact: Not available. Risks of explosion of the product in presence of static discharge: Not available.

Fire Fighting Media and Instructions:

Flammable liquid, insoluble in water. SMALL FIRE: Use DRY chemical powder. LARGE FIRE: Use water spray or fog. Cool containing vessels with water jet in order to prevent pressure build-up, autoignition or explosion.

Special Remarks on Fire Hazards:

Explosive in the form of vapor when exposed to heat or flame. Vapor may travel considerable distance to source of ignition and flash back. When heated to decomposition it emits acrid smoke and irritating fumes.

Section 6: Accidental Release Measures

Small Spill: Absorb with an inert material and put the spilled material in an appropriate waste disposal.

Large Spill:

Toxic flammable liquid, insoluble or very slightly soluble in water. Keep away from heat. Keep away from sources of ignition. Stop leak if without risk. Absorb with DRY earth, sand or other non-combustible material. Do not get water inside container. Do not touch spilled material. Prevent entry into sewers, basements or confined areas; dike if needed. Eliminate all ignition sources. Call for assistance on disposal. Be careful that the product is not present at a concentration level above TLV. Check TLV on the MSDS and with local authorities.

Section 7: Handling and Storage

Precautions:

Keep away from heat. Keep away from sources of ignition. Ground all equipment containing material. Do not ingest. Do not breathe gas/fumes/ vapour/spray. If ingested, seek medical advice immediately and show the container or the label. Avoid contact with skin and eyes Keep away from incompatibles such as oxidizing agents.

Storage:

Flammable materials should be stored in a separate safety storage cabinet or room. Keep away from heat. Keep away from sources of ignition. Keep container tightly closed. Keep in a cool, well-ventilated place. Ground all equipment containing material. A refrigerated room would be preferable for materials with a flash point lower than 37.8°C (100°F).

Section 8: Exposure Controls/Personal Protection

Engineering Controls:

Provide exhaust ventilation or other engineering controls to keep the airborne concentrations of vapors below their respective threshold limit value. Ensure that eyewash stations and safety showers are proximal to the work-station location.

Personal Protection:

Splash goggles. Lab coat. Vapor respirator. Be sure to use an approved/certified respirator or equivalent. Gloves.

Personal Protection in Case of a Large Spill:

Splash goggles. Full suit. Vapor respirator. Boots. Gloves. A self contained breathing apparatus should be used to avoid inhalation of the product. Suggested protective clothing might not be sufficient; consult a specialist BEFORE handling this product.

Exposure Limits:

TWA: 100 STEL: 150 (ppm) from ACGIH (TLV) TWA: 434 STEL: 651 (mg/m3) from ACGIHConsult local authorities for acceptable exposure limits.

Section 9: Physical and Chemical Properties

Physical state and appearance: Liquid. (Liquid.)

Odor: Not available.

Taste: Not available.

Molecular Weight: 106.17 g/mole

Color: Colorless.

pH (1% soln/water): Not applicable.

Boiling Point: 138°C (280.4°F)

Critical Temperature: Not available.

Specific Gravity: 0.86 (Water = 1)

Vapor Pressure: 9 mm of Hg (@ 20°C)

Vapor Density: 3.7 (Air = 1)

Volatility: Not available.

Odor Threshold: 0.62 ppm

Water/Oil Dist. Coeff.: Not available.

lonicity (in Water): Not available.

Solubility:

Easily soluble in methanol, diethyl ether. Insoluble in cold water, hot water.

Section 10: Stability and Reactivity Data

Stability: The product is stable.

Instability Temperature: Not available.

Conditions of Instability: Not available.

Incompatibility with various substances: Reactive with oxidizing agents.

Corrosivity: Non-corrosive in presence of glass.

Special Remarks on Reactivity: Not available.

Special Remarks on Corrosivity: Not available.

Polymerization: No.

Section 11: Toxicological Information

Routes of Entry: Eye contact.

Toxicity to Animals:

WARNING: THE LC50 VALUES HEREUNDER ARE ESTIMATED ON THE BASIS OF A 4-HOUR EXPOSURE. Acute oral toxicity (LD50): 5000 mg/kg [Rat.]. Acute dermal toxicity (LD50): 12400 mg/kg

[Rabbit.]. Acute toxicity of the vapor (LC50): 4550 ppm 4 hour(s) [Rat].

Chronic Effects on Humans: The substance is toxic to blood, kidneys, the nervous system, liver.

Other Toxic Effects on Humans:

Very hazardous in case of skin contact (irritant). Slightly hazardous in case of skin contact (permeator), of ingestion, of inhalation.

Special Remarks on Toxicity to Animals: Not available.

Special Remarks on Chronic Effects on Humans:

0347 Animal: embryotoxic, foetotoxic, passes through the placental barrier. 0900 Detected in maternal milk in human. Narcotic effect; may cause nervous system disturbances.

Special Remarks on other Toxic Effects on Humans: Material is irritating to mucous membranes and upper respiratory tract.

Section 12: Ecological Information

Ecotoxicity: Not available.

BOD5 and COD: Not available.

Products of Biodegradation:

Possibly hazardous short term degradation products are not likely. However, long term degradation products may arise.

Toxicity of the Products of Biodegradation: The products of degradation are more toxic.

Special Remarks on the Products of Biodegradation: Not available.

Section 13: Disposal Considerations

Waste Disposal:

Section 14: Transport Information

DOT Classification: Class 3: Flammable liquid.

Identification: : Xylene : UN1307 PG: III

Special Provisions for Transport: Not available.

Section 15: Other Regulatory Information

Federal and State Regulations:

Pennsylvania RTK: p-Xylene Florida: p-Xylene Massachusetts RTK: p-Xylene New Jersey: p-Xylene TSCA 8(b) inventory: p-Xylene SARA 313 toxic chemical notification and release reporting: p-Xylene CERCLA: Hazardous substances.: p-Xylene

Other Regulations: OSHA: Hazardous by definition of Hazard Communication Standard (29 CFR 1910.1200).

Other Classifications:

WHMIS (Canada):

CLASS B-2: Flammable liquid with a flash point lower than 37.8°C (100°F). CLASS D-2B: Material causing other toxic effects (TOXIC).

DSCL (EEC):

R10- Flammable. R38- Irritating to skin. R41- Risk of serious damage to eyes. R48/20- Harmful: danger of serious damage to health by prolonged exposure through inhalation.

HMIS (U.S.A.):

Health Hazard: 2

Fire Hazard: 3

Reactivity: 0

Personal Protection: h

National Fire Protection Association (U.S.A.):

Health: 2

Flammability: 3

Effective Date: 9/07/2004

Reactivity: 0

Specific hazard:

Protective Equipment:

Gloves. Lab coat. Vapor respirator. Be sure to use an approved/certified respirator or equivalent. Wear appropriate respirator when ventilation is inadequate. Splash goggles.

Section 16: Other Information

References:

-Hawley, G.G.. The Condensed Chemical Dictionary, 11e ed., New York N.Y., Van Nostrand Reinold, 1987. -Material safety data sheet emitted by: la Commission de la Santé et de la Sécurité du Travail du Québec. -SAX, N.I. Dangerous Properties of Indutrial Materials. Toronto, Van Nostrand Reinold, 6e ed. 1984. -The Sigma-Aldrich Library of Chemical Safety Data, Edition II. -Guide de la loi et du rÃ" glement sur le transport des conformité marchandises dangeureuses au canada. Centre de internatinal Ltée. 1986. Toluene MSDS Number: M1003

Material Safety Data Sheet Toluene

Section 1 -

Chemical Product and Company Identification

MSDS Name: Toluene

Synonyms: Methacide; Methylbenzene; Methylbenzol; Phenylmethane;

Toluol **Company Identification:** VEE GEE Scientific, Inc. 13600 NE 126th PI Ste A Kirkland, WA 98034 **For information in North America, call:** 425-823-4518

Section 2 -

Composition, Information on Ingredients

CAS# 108-88-3 Chemical Name Toluene

Percent

EINECS/ELINCS 203-625-9

Section 3 -

Hazards Identification

Emergency Overview

Appearance: Colorless. Flash Point: 40°F. Warning! Flammable liquid and vapor. May cause central nervous system depression. May cause liver and kidney damage. This substance has caused adverse reproductive and fetal effects in animals. Causes digestive and respiratory tract irritation. May cause skin irritation. Aspiration hazard if swallowed. Can enter lungs and cause damage. **Danger!** Harmful or fatal if swallowed. Causes eye irritation and possible transient injury. **Poison!** May be absorbed through intact skin. Vapor harmful. Call physician immediately.

Target Organs: Kidneys, central nervous system, liver.

Potential Health Effects

Eye Contact: Causes eye irritation. May result in corneal injury. Vapors may cause eye irritation.

Skin Contact: Causes moderate skin irritation. May cause cyanosis of the extremities.

Ingestion: Aspiration hazard. May cause irritation of the digestive tract. May cause effects similar to those for inhalation exposure. Aspiration of material into the lungs may cause chemical pneumonitis, which may be fatal.

Inhalation: Inhalation of high concentrations may cause central nervous system effects characterized by nausea, headache, dizziness, unconsciousness and coma. Inhalation of vapor may cause respiratory tract irritation. May cause liver and kidney damage. Vapors may cause dizziness or suffocation. Overexposure may cause dizziness, tremors, restlessness, rapid heart beat, increased blood pressure, hallucinations, acidosis, kidney failure.

Chronic Exposure: Prolonged or repeated skin contact may cause dermatitis. May cause cardiac sensitization and severe heart abnormalities. May cause liver and kidney damage.

Section 4 -

First Aid Measures

Eye Contact: Flush eyes with plenty of water for at least 15 minutes, occasionally lifting the upper and lower eyelids. Get medical aid immediately. **Skin Contact:** Flush skin with plenty of soap and water for at least 15 minutes while removing contaminated clothing and shoes. Get medical aid if irritation develops or persists.

Ingestion: Do NOT induce vomiting. If victim is conscious and alert, give 2-4 cupfuls of milk or water. Never give anything by mouth to an unconscious person. Possible aspiration hazard. Get medical aid immediately.

Inhalation: Get medical aid immediately. Remove from exposure to fresh air immediately. If not breathing, give artificial respiration. If breathing is difficult, give oxygen.

Notes to Physician: Causes cardiac sensitization to endogenous catelcholamines which may lead to cardiac arrhythmias. Do NOT use adrenergic agents such as epinephrine or pseudoepinephrine.

Section 5 -

Fire Fighting Measures

General Information: Containers can build up pressure if exposed to heat and/or fire. As in any fire, wear a self-contained breathing apparatus in pressure-demand, MSHA/NIOSH (approved or equivalent), and full protective gear. Water runoff can cause environmental damage. Dike and collect water used to fight fire. Vapors may form an explosive mixture with air. Vapors can travel to a source of ignition and flash back. Flammable Liquid. Can release vapors that form explosive mixtures at temperatures above the flashpoint. Use water spray to keep fire-exposed containers cool. Water may be ineffective. Material is lighter than water and a fire may be spread by the use of water. Vapors may be heavier than air. They can spread along the ground and collect in low or confined areas. Containers may explode when heated.

Fire Extinguishing Media: Use water spray to cool fire-exposed containers. Water may be ineffective. Do NOT use straight streams of water. For small fires, use dry chemical, carbon dioxide, water spray or regular foam. Cool containers with flooding quantities of water until well after fire is out. For

Section 5 -

Fire Fighting Measures

Autoignition Temperature: 422°C (792°F) Flash Point: 7°C (45°F) Explosion Limits, lower: 1.2 vol%. Explosion Limits, upper: 7.1 vol% NFPA Rating: (estimated) Health: 2; Flammability: 3; Instability: 0

Section 6 -

Accidental Release Measures

General Information: Use proper personal protective equipment as indicated in Section 8.

Spills/Leaks: Avoid runoff into storm sewers and ditches which lead to waterways. Remove all sources of ignition. Absorb spill using an absorbent, non-combustible material such as earth, sand, or vermiculite. Do not use combustible materials such as saw dust. A vapor suppressing foam may be used to reduce vapors. Water spray may reduce vapor but may not prevent ignition in closed spaces.

Section 7 -

Handling and Storage

Handling: Wash thoroughly after handling. Use with adequate ventilation. Ground and bond containers when transferring material. Avoid contact with eyes, skin, and clothing. Empty containers retain product residue, (liquid and/or vapor), and can be dangerous. Keep container tightly closed. Avoid contact with heat, sparks and flame. Avoid ingestion and inhalation. Do not pressurize, cut, weld, braze, solder, drill, grind, or expose empty containers to heat, sparks or open flames.

Storage: Keep away from heat, sparks, and flame. Keep away from sources of ignition. Store in a tightly closed container. Store in a cool, dry, well-ventilated area away from incompatible substances.

Section 8 - Exposure Controls, Personal Protection				
Chemical Name Toluene	ACGIH 50 ppm TWA	NIOSH 100 ppm TWA 375 mg/m3 TWA 500 ppm IDLH	OSHA - Final PELs 200 ppm TWA C 300 ppm	OSHA - Vacated Pels 100 ppm TWA 375 mg/m3 TWA 150 ppm STEL 560 mg/m3 STEL

Engineering Controls: Facilities storing or utilizing this material should be equipped with an eyewash facility and a safety shower. Use adequate general or local exhaust ventilation to keep airborne concentrations below the permissible exposure limits. **Personal Protective Equipment**

Eyes: Wear appropriate protective eyeglasses or chemical safety goggles as described by OSHA's eye and face protection regulations in 29 CFR 1910.133 or European Standard EN166.

Skin: Wear appropriate protective gloves to prevent skin exposure.

Clothing: Wear appropriate protective clothing to prevent skin exposure.

Respirators: Follow the OSHA respirator regulations found in 29CFR 1910.134 or European Standard EN 149. Always use a NIOSH or European Standard EN 149 approved respirator when necessary.

Section 9 -

Physical and Chemical Properties

Physical State:Clear liquidAppearance:ColorlessOdor:Sweet, pleasantpH:Not available

 Vapor Pressure: 36.7 mm Hg @ 30° C

 Vapor Density: 3.1

 Evaporation
 Rate:
 2.4

 Viscosity:
 0.59 cP @ 20° C

Boiling Point: 232° F Freezing/Melting Point: -139° F Decomposition Temperature: Not available Solubility: Insoluble Specific Gravity/Density: 0.9 Molecular Formula: C6H5CH3 Molecular Weight: 92.056

Section 10 -

Stability and Reactivity

Chemical Stability: Stable under normal temperatures and pressures. Conditions to Avoid: Incompatible materials, ignition sources, excess heat. Incompatibilities with Other Materials: Ntrogen tetroxide, nitric acid plus sulfuric acid, silver perchlorate, strong oxidizers, sodium difluoride. Hazardous Decomposition Products: Carbon monoxide, carbon dioxide. Hazardous Polymerization: Has not been reported.

Section 11 -

Toxilogical Information

Carcinogenicity: CAS# 108-88-3: ACGIH: A4 - Not Classifiable as a Human Carcinogen

Section 11 -

Toxilogical Information (continued)

Epidemiology: No information available.

Teratogenicity: Specific developmental abnormalities included craniofacial effects involving the nose and tongue, musculoskeletal effects, urogenital and metabolic effects in studies on mice and rats by the inhalation and oral routes of exposure. Some evidence of fetotoxicity with reduced fetal weight and retarded skeletal development has been reported in mice and rats.

Reproductive Effects: Effects on fertility such as abortion were reported in rabbits by inhalation. Paternal effects were noted in rats by inhalation. These effects involved the testes, sperm duct and epididymis.

Neurotoxicity: No information available.

Mutagenicity: No information available.

mutagenicity: No information available.

Section 12 -

Ecological Information

Ecotoxicity: No data available. Bluegill LC50=17 mg/L/24H Shrimp LC50=4.3 ppm/96H Fathead minnow LC50=36.2 mg/L/96HSunfish (fresh water) TLm=1180 mg/L/96H

Environmental: From soil, substance evaporates and is microbially biodegraded. In water, substance volatilizes and biodegrades. **Physical:** Photochemically produced hydroxyl radicals degrade substance. **Other:** None.

Section 13 -

Disposal Considerations

Chemical waste generators must determine whether a discarded chemical is classified as a hazardous waste. US EPA guidelines for the classification determination are listed in 40 CFR Parts 261.3. Additionally, waste generators must consult state and local hazardous waste regulations to ensure complete and accurate classification.

RCRA P-Series: None listed.

RCRA U-Series: CAS# 108-88-3: waste number U220.

Section 14 -	Transport Information		
Shipping Name Hazard Class UN Number Packing Group Other	US DOT Toluene 3 UN1294 II	Canada TDG Toluene 3 (9.2) UN1294 II FP 4C	

Section 15 -

Regulatory Information

US Federal

TSCA: CAS# 108-88-3 is listed on the TSCA inventory.

Health & Safety Reporting List: None of the chemicals are on the Health & Safety Reporting List.

CAS# 108-88-3: Effective Date: October 4, 1982; Sunset Date: October 4 , 1992

Chemical Test Rules: None of the chemicals in this product are under a Chemical Test Rule.

Section 12b: None of the chemicals are listed under TSCA Section 12b.

TSCA Significant New Use Rule: None of the chemicals in this material have a SNUR under TSCA.

SARA: Section 302 (RQ): CAS# 108-88-3: final RQ = 1000 pounds (454 kg)

Section 302 (TPQ): None of the chemicals in this product have a TPQ.

SARA Codes: CAS # 108-88-3: acute, flammable.

Section 313: This material contains Toluene (CAS# 108-88-3, 99%), which is subject to the reporting requirements of Section 313 of SARA Title III and 40 CFR Part 373.

Clean Air Act: CAS# 108-88-3 is listed as a hazardous air pollutant (HAP). This material does not contain any Class 1 Ozone depletors. This material does not contain any Class 2 Ozone depletors.

Clean Water Act: CAS# 108-88-3 is listed as a Hazardous Substance under the CWA. CAS# 108-88-3 is listed as a Priority Pollutant under the Clean Water Act. CAS# 108-88-3 is listed as a Toxic Pollutant under the Clean Water Act.

OSHA: None of the chemicals in this product are considered highly hazardous by OSHA.

STATE: CAS# 108-88-3 can be found on the following state right to know lists: California, New Jersey, Florida, Pennsylvania, Minnesota, Massachusetts.

WARNING: This product contains Toluene, a chemical known to the state of California to cause birth defects or other reproductive harm. California No

Significant Risk Level: CAS# 108-88-3: NOEL = 7000 ug/day

European/International Regulations

European Labeling in Accordance with EC Directives Hazard Symbols: XN F Risk Phrases:

R 11 Highly flammable.

R 20 Harmful by inhalation

M1003

Page 3/4

Effective Date: 9/07/2004

Regulatory Information (continued)

Safety Phrases:

Section 15 -

S 16 Keep away from sources of ignition - No smoking. S 25 Avoid contact with eyes. S 29 Do not empty into drains.

S 33 Take precautionary measures against static discharges.

WGK (Water Danger/Protection): CAS# 108-88-3: 2

Canada - DSL/NDSL: CAS# 108-88-3 is listed on Canada's DSL List. Canada - WHMIS: This product has a WHMIS classification of B2, D2B

Canadian Ingredient Disclosure List: CAS# 108-88-3 is listed on Canada's Ingredient Disclosure List.

Exposure Limits: CAS# 108-88-3: OEL-AUSTRALIA:TWA 100 ppm (375 mg/m3);STEL 150 ppm (560 mg/m3) OEL-BELGIUM:TWA 100 ppm (377 g/ m3);STEL 150 ppm (565 mg/m3) OEL-CZECHOSLOVAKIA:TWA 200 mg/m3;STEL 1000 mg/m3 OEL-DENMARK:TWA 50 ppm (190 mg/m3);Skin OEL-FINLAND:TWA 100 ppm (375 mg/m3);STEL 150 ppm;Skin OEL-FRANCE:TWA 100 ppm (375 mg/m3);STEL 150 ppm (560 mg/m3) OEL-GERMANY:TWA 100 ppm (380 mg/m3) OEL-HUNGARY:TWA 100 mg/m3;STEL 300 mg/m3;SKin OEL-JAPAN:TWA 100 ppm (380 mg/m3) OEL-THE NETHERLANDS:TWA 100 ppm (375 mg/m3);Skin OEL-THE PHILIPPINES:TWA 100 ppm (375 mg/m3) OEL-POLAND:TWA 100 mg/m3 OEL-USSIA:TWA 100 ppm;STEL 50 mg/m3 OEL-SWEDEN:TWA 50 ppm (200 mg/m3);STEL 100 ppm (400 mg/m3);Skin OEL-SWITZERLAND:TWA 100 ppm (380 mg/m3);STEL 500 ppm OEL-THAILAND:TWA 200 ppm;STEL 300 ppm OEL-TURKEY:TWA 200 ppm (750 mg/m3) OEL-UNITED KINGDOM :TWA 100 ppm (375 mg/m3);STEL 150 ppm;Skin OEL IN BULGARIA, COLOMBIA,JORDAN, KOREA check ACGIH TLV OEL IN NEW ZEALAND, SINGAPORE, VIETNAM check ACGI TLV

Section 16 -

Additional Information

MSDS Creation Date: 09/07/2004

The information above is believed to be accurate and represents the best information currently available to us. However, we make no warranty of merchantability or any other warranty, express or implied, with respect to such information, and we assume no liability resulting from its use. Users should make their own investigations to determine the suitability of the information for their particular purposes. In no event shall VEE GEE Scientific be liable for any claims, losses, or damages of any third party or for lost profits or any special, indirect, incidental, consequential or exemplary damages, howsoever arising, even if VEE GEE Scientific has been advised of the possibility of such damages.



Material Safety Data Sheet

HAZARD WARNINGS	RISK PHRASES		PROTECTIVE CLOTHING
	Combustible material; avoid heat and sources of ignition. Harmful compound, minimize exposure. Irritating to skin, eyes, and the respiratory system.		
Section I. Chem	ical Product and Company Identification	n	
Section 1. Chem	ical Froduct and Company identification	лт —	
Chemical 1	,3,5-Trimethylbenzene		
Name	s	upplier	TCI America

	T0470		9211 N. Harborgate St. Portland OR
Catalog Number	Mesitylene		1-800-423-8616
Synonym	(CH ₃) ₃ C ₆ H ₃	 In case of	Chemtrec®
Chemical Formula	108-67-8	Emergency Call	(800) 424-9300 (U.S.) (703) 527-3887 (International)
CAS Number			

CAS Number

Section II. Composition and Information on Ingredients						
Chemical Nan	ne	CAS Number	Percent (%)	TLV/PEL		Toxicology Data
1,3,5-Trimethylber	nzene	108-67-8	Min. 97.0 (GC)	Not available.		Rat LC ₅₀ (inhalation) 24gm/m ³ /4H
Section III. Ha	zards Identi	fication				
Acute Health Effects	Irritating to eyes ar eye is characterize or, occasionally, bli	Harmful if ingested or inhaled. Minimize exposure to this material. Severe overexposure can result in injury or death. Irritating to eyes and skin on contact. Inhalation causes irritation of the lungs and respiratory system. Inflammation of the eye is characterized by redness, watering, and itching. Skin inflammation is characterized by itching, scaling, reddening, or, occasionally, blistering. Follow safe industrial hygiene practices and always wear proper protective equipment when handling this compound.				
Chronic Health Effects	CARCINOGENIC EFFECTS : Not available. MUTAGENIC EFFECTS : Not available. TERATOGENIC EFFECTS : Not available. DEVELOPMENTAL TOXICITYNot available. Repeated or prolonged exposure to this compound is not known to aggravate existing medical conditions.					
Section IV	/. Aid Measu	ires				
Eye Contact	Check for and remove any contact lenses. In case of contact, immediately flush eyes with plenty of water for at least 15 minutes. Get medical attention.					
Skin Contact		In case of contact, immediately flush skin with plenty of water. Remove contaminated clothing and shoes. Wash clothing before reuse. Thoroughly clean shoes before reuse. Get medical attention.				
Inhalation		If the victim is not breathing, perform mouth-to-mouth resuscitation. Loosen tight clothing such as a collar, tie, belt or waistband. If breathing is difficult, oxygen can be administered. Seek medical attention if respiration problems do not improve.				
Ingestion INDUCE VOMITING by sticking finger in throat. Lower the head so that the vomit will not reenter the mouth and throat. Loosen tight clothing such as a collar, tie, belt or waistband. If the victim is not breathing, perform mouth-to-mouth resuscitation. Examine the lips and mouth to ascertain whether the tissues are damaged, a possible indication that the toxic material was ingested; the absence of such signs, however, is not conclusive. SEEK IMMEDIATE MEDICAL ATTENTION in case of ingestion of a radioactive material.						
Section V. Fire	e and Explos	ion Data				
Flammability	Combustible.		Α	uto-Ignition	Not availa	able.
					LOWER:	0.88%

Appendix A-4. MSDS Information

Flash Points	44.44°C (112°F).	Flammable Limits
Combustion Products	These products are toxic carbon oxides (CO, CO 2).	
Fire Hazards	Risks of explosion of the product in presence of mechanical impact: Not available. Risks of explosion of the product in presence of static discharge: Not available.	
Explosion Hazards	Combustible liquid.	
	SMALL FIRE: Use DRY chemical powder.	
Fine Fielding Mardia		am, water spray or fog. Cool containing vessels with water jet in order to prevent pressure

3/11/2005.

Acute TOXIC Effects

Irritating to eyes and skin on contact. Inhalation causes irritation of the lungs and respiratory system. Inflammation of the eye is characterized by redness, watering, and itching. Skin inflammation is characterized by itching, scaling, reddening, or executionally bitching.

Appendix A-4. MSDS Information

0470

1,3,5-Trimethylbenzene

Page 3

Section XII.	Ecological Information		
Ecotoxicity	Toxic to aquatic organisms. May cause long-term adverse effects in the aquatic environment.		
Environmental Fate	Mesitylene's production and use as a dyestuff intermediate, solvent, paint thinner, and as a UV oxidation stabilizer for plastics may result in its release to the environment through various waste streams. Mesitylene is released directly to the environment as a component of gasoline and as an emission from gasoline-powered vehicles, municipal waste-treatment plants, and coal-fired power stations. If released to the atmosphere, mesitylene will exist solely in the vapor phase in the ambient atmosphere, based on a measured vapor pressure of 2.48 mm Hg at 25 deg C. Vapor-phase mesitylene is degraded in the atmosphere by reaction with photochemically-produced hydroxyl radicals and nitrate radicals with half-lives of about 7 hours and 10-67 days, respectively. A measured Koc value of 660 suggests that mesitylene will have low mobility in soil. Volatilization from moist and dry soil surfaces should occur based on a measured Henry's Law constant of 8.77X10-3 atm-cu m/mole and the vapor pressure of this compound, respectively. Mesitylene should aerobically biodegrade in both soil and water. Mesitylene was not degraded in methanogenic aquifer microcosms. In water, mesitylene may adsorb to sediment or particulate matter based on its Koc value. This compound should volatilize from water surfaces given its Henry's Law constant. Estimated half-lives for a model river and model lake are 3 hours and 4 days, respectively. Bioconcentration in aquatic organisms may occur based on BCF values of 23-342, measured in carp. The general population will be exposed to mesitylene via inhalation of ambient air, ingestion of food and drinking water, and dermal contact with vapors, food and other products containing mesitylene. Occupational exposure may occur through inhalation and dermal contact with this compound at workplaces where it is produced or used.		
Section XIII.	Disposal Considerations		
Waste Disposal	Recycle to process, if possible. Consult your local regional authorities. You may be able to dissove or mix material with a combustible solvent and burn in a chemical incinerator equipped with an afterburner and scrubber system. Observe all federal, state and locl regulations when disposing of the substance.		
Section	XIV. sport Information		
DOT Classification	CLASS 3: Flammable liquid.		
PIN Number	UN2325		
Proper Shipping Name	1,3,5-Trimethylbenzene		
	III		
Packing Group (PG)			
DOT Pictograms			
Section XV. Other Regulatory Information and Pictograms			
TSCA Chemical Invento (EPA)	This compound is ON the EPA Toxic Substances Control Act (TSCA) inventory list.		
WHMIS Classification (Canada)	CLASS B-3: Combustible liquid with a flash point between 37.8°C (100°F) and 93.3°C (200°F). This product is subject to SARA section 313 reporting requirements (8a PAIR).		

EINECS Number (EEC)	203-604-4
EEC Risk Statements	R20/21/22- Harmful by inhalation, in contact with skin and if swallowed. R36/37/38- Irritating to eyes, respiratory system and skin. R51- Toxic to aquatic organisms.
Japanese Regulatory Data	R53- May cause long-term adverse effects in the aquatic environment. Not available.

Section XVI. Other Information

Version 1.0

Validated on 6/7/2002.

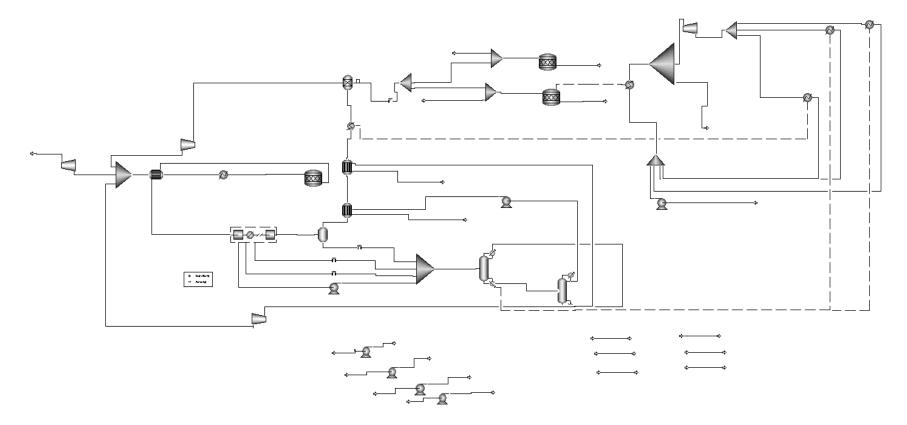
Printed 3/11/2005.

Notice to Reader

TCI laboratory chemicals are for research purposes only and are NOT intended for use as drugs, food additives, households, or pesticides. The information herein is believed to be correct, but does not claim to be all inclusive and should be used only as a guide. Neither the above named supplier nor any of its subsidiaries assumes any liability whatsoever for the accuracy or completeness of the information contained herein. Final determination of suitability of any material is the sole responsibility of the user. All chemical reagents must be handled with the recognition that their chemical, physiological, toxicological, and hazardous properties have not been fully investigated or determined. All chemical reagents should be handled only by individuals who are familiar with their potential hazards and who have been fully trained in proper safety, laboratory, and chemical handling procedures. Although certain hazards are described herein, we can not guarantee that these are the only hazards which exist. Our MSDS sheets are based only on data available at the time of shipping and are subject to change without notice as new information is obtained. Avoid long storage periods since the product is subject to degradation with age and may become more dangerous on hazardous. It is the responsibility of the user to request updated MSDS sheets for products that are stored for extended periods. Disposal of unused product must be undertaken by qualified personnel who are knowledgeable in all applicable regulations and follow all pertinent safety precautions including the use of anorrorize protective equinment (e.g. protective could be asseed conducting breathing equinment facial mask furme hood). For proper handling and follow all pertinent safety precautions including the use of anorrorize protective equinment (e.g. protective equinment (

Printed 3/11/2005.

A-5. ASPEN Data



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A-5.1. Input File
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```
;Input Summary created by Aspen Plus Rel. 32.0 at 17:36:27 Sat Apr 11, 2015
;Directory S:\Documents\CBE 459\New Catalyst L ver2-6-1 1
                                                                      Filename
C:\Users\jlym\AppData\Local\Temp\~ap972.txt
DYNAMICS
    DYNAMICS RESULTS=ON
IN-UNITS ENG
DEF-STREAMS CONVEN ALL
MODEL-OPTION
DATABANKS 'APV86 PURE32' / 'APV86 AQUEOUS' / 'APV86 SOLIDS' / &
        'APV86 INORGANIC' / NOASPENPCD
PROP-SOURCES 'APV86 PURE32' / 'APV86 AQUEOUS' / 'APV86 SOLIDS' &
         / 'APV86 INORGANIC'
COMPONENTS
   HYDROGEN H2 /
    METHANE CH4 /
    ETHANE C2H6 /
    ETHENE C2H4 /
    PROPANE C3H8 /
    PROPENE C3H6-2 /
    BENZENE C6H6 /
    TOLUENE C7H8 /
    P-XYL-01 C8H10-3 /
    1:3:5-01 C9H12-8 /
    WATER H2O /
    OXYGEN O2 /
    NITROGEN N2 /
    CARBO-01 CO2
SOLVE
    RUN-MODE MODE=SIM
FLOWSHEET
    BLOCK R-101 IN=S-106, OUT=S-107
    BLOCK F-101, IN=S-106 OUT=S-106,
    BLOCK F-101 IN=S-220, OUT=S-110
    BLOCK V-103 IN=S-109 S-220 OUT=S-220,
    BLOCK V-203-5 IN=S-209 S-208 S-207 S-206 OUT=S-210
    BLOCK P-201 IN=S-204 OUT=S-209
    BLOCK PR-201 IN=S-214 OUT=S-219, S-218
    BLOCK V-101-2 IN=S-102 S-104 S-103 OUT=S-105
    BLOCK V-206 IN=S-219 OUT=S-221 S-220
    BLOCK T-105 IN=S-305 OUT=S-104
    BLOCK T-104 IN=S-101 OUT=S-102
    BLOCK COL-301 IN=S-210 OUT=S-305 S-310 HEAT3
```

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BLOCK F-401-2 IN=S-221, OUT=S-411 HEAT1
    BLOCK V-404 IN=S-221 S-410 OUT=S-221,
    BLOCK H-101 IN=S-107 S-105 OUT=S-108 S-106
    BLOCK COL-302 IN=S-310 OUT=S-315 S-320 HEAT4
    BLOCK V-201 IN=S-205 OUT=S-206
    BLOCK V-203 IN=S-203 OUT=S-208
    BLOCK V-202 IN=S-202 OUT=S-207
    BLOCK C-201-3 IN=S-108 OUT=S-201 S-204 S-203 S-202
    BLOCK F-401-2, IN=S-403 HEAT1 OUT=S-404
    BLOCK V-405 IN=S-404 OUT=S-405 S-406
    BLOCK RB-301 IN=S-408 HEAT3 OUT=S-308
    BLOCK RB-302 IN=S-409 HEAT4 OUT=S-319
    BLOCK H-201 IN=S-222 S-211 OUT=S-215 S-212
    BLOCK H-203 IN=S-213 OUT=S-214 HEAT2
    BLOCK H-202 IN=S-320 S-212 OUT=S-216 S-213
    BLOCK T-101-3 IN=S-218 OUT=S-103
    BLOCK COL-201 IN=S-201 OUT=S-211 S-205
    BLOCK H-203, IN=S-407 HEAT2 OUT=S-217
    BLOCK PR-201, IN=S-219, OUT=S-219
    BLOCK V-401-3 IN=S-402 S-319 S-217 S-308 OUT=S-403
   BLOCK P-401 IN=S-401 OUT=S-402
    BLOCK T-401 IN=S-406 OUT=S-406,
    BLOCK V-406-7 IN=S-406, OUT=S-408 S-409 S-407
    BLOCK P-301 IN=S-303 OUT=S-304
    BLOCK P-302 IN=S-306 OUT=S-307
   BLOCK P-303 IN=S-313 OUT=S-314
   BLOCK P-304 IN=S-316 OUT=S-317
    BLOCK P-202 IN=S-315 OUT=S-222
PROPERTIES NRTL-RK
PROP-DATA NRTL-1
    IN-UNITS MET PRESSURE=bar TEMPERATURE=C DELTA-T=C PDROP=bar &
        INVERSE-PRES='1/bar'
    PROP-LIST NRTL
    BPVAL PROPENE BENZENE 0.0 178.8998000 .300000000 0.0 0.0
                                                              æ
        0.0 25.0000000 25.0000000
    BPVAL BENZENE PROPENE 0.0 46.73830000 .3000000000 0.0 0.0
                                                               £
       0.0 25.0000000 25.0000000
    BPVAL BENZENE TOLUENE -2.719200000 1083.397500 .300000000
                                                                δ
       0.0 0.0 0.0 40.0000000 110.7500000
    BPVAL TOLUENE BENZENE 2.067100000 -831.3335000 .300000000
                                                                δ
        0.0 0.0 0.0 40.0000000 110.7500000
    BPVAL BENZENE P-XYL-01 0.0 222.4249000 .3000000000 0.0 0.0
                                                               æ
        0.0 83.6000000 129.000000
    BPVAL P-XYL-01 BENZENE 0.0 -197.7876000 .3000000000 0.0 &
        0.0 0.0 83.6000000 129.000000
    BPVAL TOLUENE P-XYL-01 0.0 -114.6844000 .300000000 0.0
                                                            &
        0.0 0.0 111.2300000 137.1000000
    BPVAL P-XYL-01 TOLUENE 0.0 107.5213000 .3000000000 0.0 0.0
                                                               ß
        0.0 111.2300000 137.1000000
    BPVAL BENZENE WATER 45.19050000 591.3676000 .2000000000 0.0
                                                                 æ
        -7.562900000 0.0 .800000000 77.00000000
    BPVAL WATER BENZENE 140.0874000 -5954.307100 .200000000
        0.0 -20.02540000 0.0 .800000000 77.00000000
    BPVAL TOLUENE WATER -247.8792000 14759.75980 .200000000 &
        0.0 35.58200000 0.0 -9.00000000 93.0000000
```

BPVAL WATER TOLUENE 627.0528000 -27269.35550 .200000000 æ 0.0 -92.71820000 0.0 -9.00000000 93.0000000 BPVAL P-XYL-01 WATER 2.773400000 296.6645000 .200000000 £ 0.0 .1174000000 0.0 0.0 294.9000000 BPVAL WATER P-XYL-01 162.4774000 -6045.999500 .2000000000 & 0.0 -23.46720000 0.0 0.0 294.9000000 BPVAL 1:3:5-01 WATER -3.726400000 2542.779300 .2000000000 æ 0.0 0.0 0.0 20.0000000 40.0000000 BPVAL WATER 1:3:5-01 10.25210000 -66.03660000 .200000000 8 0.0 0.0 0.0 20.0000000 40.0000000 BPVAL WATER CARBO-01 10.06400000 -3268.135000 .200000000 æ 0.0 0.0 0.0 0.0 200.000000 BPVAL CARBO-01 WATER 10.06400000 -3268.135000 .200000000 & 0.0 0.0 0.0 0.0 200.000000 STREAM S-101 SUBSTREAM MIXED TEMP=80.0000000 PRES=264.6959487 & MASS-FLOW=5707.767968 MOLE-FRAC METHANE 0.01 / ETHANE 0.97 / PROPANE 0.02 STREAM S-109 SUBSTREAM MIXED TEMP=77.00000000 PRES=14.50377377 & MOLE-FLOW=22046.22622 MOLE-FRAC OXYGEN 0.21 / NITROGEN 0.79 STREAM S-301 SUBSTREAM MIXED TEMP=209.392 PRES=93. MASS-FLOW=22552.844 & FREE-WATER=NO NPHASE=1 PHASE=V MASS-FRAC HYDROGEN 4.57E-009 / METHANE 0.120733261 / & ETHANE 0.31430703 / ETHENE 0.027124142 / PROPANE & 0.031919275 / PROPENE 0.006390641 / BENZENE æ 0.445045079 / TOLUENE 0.052578284 / P-XYL-01 & 0.001899723 / 1:3:5-01 2.56E-006 / WATER 0. / OXYGEN & 0. / NITROGEN 0. / CARBO-01 0. STREAM S-302 SUBSTREAM MIXED TEMP=186.1 VFRAC=0.897120543 & MASS-FLOW=22552.844 MASS-FRAC HYDROGEN 4.57E-009 / METHANE 0.120733261 / & ETHANE 0.31430703 / ETHENE 0.027124142 / PROPANE & 0.031919275 / PROPENE 0.006390641 / BENZENE & 0.445045079 / TOLUENE 0.052578284 / P-XYL-01 & 0.001899723 / 1:3:5-01 2.56E-006 STREAM S-303 SUBSTREAM MIXED TEMP=186.1 PRES=83. MASS-FLOW=2103.78883 & FREE-WATER=NO NPHASE=1 PHASE=L MASS-FRAC HYDROGEN 2.21E-016 / METHANE 0.002560832 / & ETHANE 0.024119202 / ETHENE 0.001717492 / PROPANE æ 0.004315422 / PROPENE 0.000507373 / BENZENE & 0.790953211 / TOLUENE 0.166730683 / P-XYL-01 & 0.009080698 / 1:3:5-01 1.51E-005 STREAM S-306 SUBSTREAM MIXED TEMP=339.938 PRES=102.3 MASS-FLOW=264044.408 FREE-WATER=NO NPHASE=1 PHASE=L MASS-FRAC HYDROGEN 2.33E-274 / METHANE 3.23E-067 / ETHANE &

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2.39E-024 / ETHENE 9.62E-028 / PROPANE 2.2E-017 / &
       PROPENE 3.29E-026 / BENZENE 0.660110215 / TOLUENE &
        0.233327963 / P-XYL-01 0.050159506 / 1:3:5-01 &
        0.056402315
STREAM S-309
    SUBSTREAM MIXED TEMP=348.53 PRES=102.45 MASS-FLOW=138213.865 &
        FREE-WATER=NO NPHASE=1 PHASE=L
   MASS-FRAC HYDROGEN 4.45E-274 / METHANE 6.1E-067 / ETHANE &
        4.14E-024 / ETHENE 1.68E-027 / PROPANE 3.63E-017 / &
       PROPENE 5.67E-026 / BENZENE 0.752673327 / TOLUENE &
       0.194888485 / P-XYL-01 0.029617377 / 1:3:5-01 &
        0.022820812
STREAM S-311
    SUBSTREAM MIXED TEMP=241.229 PRES=26. MASS-FLOW=187342.041 &
        FREE-WATER=NO NPHASE=1 PHASE=V
    MASS-FRAC BENZENE 0.615754339 / TOLUENE 0.303831852 / &
       P-XYL-01 0.080058695 / 1:3:5-01 0.000355114
STREAM S-312
    SUBSTREAM MIXED TEMP=196.384 PRES=16. MASS-FLOW=187342.041 &
       FREE-WATER=NO NPHASE=1 PHASE=L
   MASS-FRAC BENZENE 0.615754339 / TOLUENE 0.303831852 / &
       P-XYL-01 0.080058695 / 1:3:5-01 0.000355114
STREAM S-313
    SUBSTREAM MIXED TEMP=196.384 PRES=16. MASS-FLOW=114117.806 &
        FREE-WATER=NO NPHASE=1 PHASE=L
    MASS-FRAC BENZENE 0.615754339 / TOLUENE 0.303831852 / &
       P-XYL-01 0.080058695 / 1:3:5-01 0.000355114
STREAM S-316
    SUBSTREAM MIXED TEMP=386.837 PRES=31.85 MASS-FLOW=194453.214 &
       FREE-WATER=NO NPHASE=1 PHASE=L
   MASS-FRAC BENZENE 8.08E-011 / TOLUENE 9.14E-007 / &
       P-XYL-01 0.002075488 / 1:3:5-01 0.997923598
STREAM S-318
    SUBSTREAM MIXED TEMP=387.281 PRES=32. MASS-FLOW=182740.516 &
       FREE-WATER=NO NPHASE=1 PHASE=V
   MASS-FRAC HYDROGEN 0. / METHANE 0. / ETHANE 0. / &
       ETHENE 0. / PROPANE 0. / PROPENE 0. / BENZENE &
        8.49E-011 / TOLUENE 9.52E-007 / P-XYL-01 0.002128228 / &
       1:3:5-01 0.99787082 / WATER 0. / OXYGEN 0. / &
       NITROGEN 0. / CARBO-01 0.
STREAM S-401
    SUBSTREAM MIXED TEMP=77.00000000 PRES=14.50377377 &
       MASS-FLOW=1.03970664E+6
    MASS-FRAC WATER 1.
STREAM S-410
    SUBSTREAM MIXED TEMP=77.00000000 PRES=14.50377377 &
       MOLE-FLOW=2.20462262E+5
   MOLE-FRAC OXYGEN 0.21 / NITROGEN 0.79
```

DEF-STREAMS HEAT HEAT1 DEF-STREAMS HEAT HEAT2 DEF-STREAMS HEAT HEAT3 DEF-STREAMS HEAT HEAT4 BLOCK V-101-2 MIXER PARAM BLOCK V-103 MIXER PARAM BLOCK V-203-5 MIXER PARAM BLOCK V-401-3 MIXER PARAM BLOCK V-404 MIXER PARAM BLOCK V-206 FSPLIT FRAC S-221 0. ;0.014605878 ; 0.016515892 BLOCK V-405 FSPLIT FRAC S-406 0.071210559 BLOCK V-406-7 FSPLIT FRAC S-408 0.329935017 / S-409 0.431538716 BLOCK PR-201 SEP PARAM PRES=-10.0000000 FRAC STREAM=S-219, SUBSTREAM=MIXED COMPS=HYDROGEN METHANE & ETHANE ETHENE PROPANE PROPENE BENZENE TOLUENE P-XYL-01 & 1:3:5-01 WATER OXYGEN NITROGEN CARBO-01 FRACS=0.85 & 0.28 0. 0. 0. 0. 0. 0. 0. 0. 0. 0. 0. 0. BLOCK F-101, HEATER PARAM TEMP=1166.000000 PRES=0.0 BLOCK F-401-2, HEATER PARAM PRES=450. psig> NPHASE=2 TOL=0.0001 T-EST=384.2600000 BLOCK-OPTION FREE-WATER=NO BLOCK H-203 HEATER PARAM TEMP=181.4000000 PRES=-10.00000000 BLOCK H-203, HEATER PARAM PRES=-10.0000000 BLOCK RB-301 HEATER PARAM PRES=-10.0000000 NPHASE=2

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BLOCK-OPTION FREE-WATER=NO
BLOCK RB-302 HEATER
    PARAM PRES=-10.0000000 NPHASE=2
    BLOCK-OPTION FREE-WATER=NO
BLOCK COL-201 FLASH2
    PARAM TEMP=104.0000000 PRES=0.0 TOL=0.0001
BLOCK H-101 HEATX
    PARAM DELT-COLD=18. CALC-TYPE=DESIGN PRES-HOT=-10.00000000 &
       PRES-COLD=-10.00000000 MIN-TAPP=18.00000000 U-OPTION=PHASE &
       F-OPTION=CONSTANT CALC-METHOD=SHORTCUT
    FEEDS HOT=S-107 COLD=S-105
    OUTLETS-HOT S-108
    OUTLETS-COLD S-106
    HOT-SIDE DP-OPTION=CONSTANT DPPARMOPT=NO
    COLD-SIDE DP-OPTION=CONSTANT
    TQ-PARAM CURVE=YES
BLOCK H-201 HEATX
    PARAM DELT-HOT=18. <C> CALC-TYPE=DESIGN &
       PRES-HOT=-10.0000000 PRES-COLD=-10.0000000 &
       MIN-TAPP=18.0000000 U-OPTION=PHASE F-OPTION=CONSTANT &
       CALC-METHOD=SHORTCUT
    FEEDS HOT=S-222 COLD=S-211
    OUTLETS-HOT S-215
   OUTLETS-COLD S-212
    HOT-SIDE DP-OPTION=CONSTANT
    COLD-SIDE DP-OPTION=CONSTANT
    TO-PARAM CURVE=YES
BLOCK H-202 HEATX
    PARAM DELT-HOT=10. <C> CALC-TYPE=DESIGN &
        PRES-HOT=-10.0000000 PRES-COLD=-10.0000000 &
       U-OPTION=PHASE F-OPTION=CONSTANT CALC-METHOD=SHORTCUT
    FEEDS HOT=S-320 COLD=S-212
    OUTLETS-HOT S-216
    OUTLETS-COLD S-213
   HOT-SIDE DP-OPTION=CONSTANT
    COLD-SIDE DP-OPTION=CONSTANT
   TO-PARAM CURVE=YES
BLOCK COL-301 RADFRAC
    PARAM NSTAGE=65 ALGORITHM=STANDARD EFF=MURPHREE MAXOL=25 &
        DAMPING=NONE
    COL-CONFIG CONDENSER=PARTIAL-V
    RATESEP-ENAB CALC-MODE=EQUILIBRIUM
    FEEDS S-210 22
    PRODUCTS S-305 1 V / S-310 65 L
    PRODUCTS HEAT3 65
    P-SPEC 1 83.
    COL-SPECS MASS-D:F=0.089716729 DP-STAGE=0.15 &
       MASS-RR=0.10287944 DP-COND=10.
    STEFF-SEC SECNO=1 1 65 0.65
```

BLOCK COL-302 RADFRAC

```
PARAM NSTAGE=42 ALGORITHM=STANDARD EFF=MURPHREE MAXOL=25 &
       DAMPING=NONE
    COL-CONFIG CONDENSER=TOTAL
    FEEDS S-310 23
    PRODUCTS S-315 1 L / S-320 42 L
    PRODUCTS HEAT4 42
    P-SPEC 1 16.
    COL-SPECS MASS-D:F=0.906916873 DP-STAGE=0.15 &
       MASS-RR=0.641654779 DP-COND=10.
    STEFF-SEC SECNO=1 1 42 0.65
BLOCK F-101 RSTOIC
    PARAM TEMP=1346.000000 PRES=14.50377377 HEAT-OF-REAC=YES
    STOIC 1 MIXED METHANE -1. / OXYGEN -2. / CARBO-01 1. / &
       WATER 2.
    STOIC 2 MIXED ETHANE -2. / OXYGEN -7. / CARBO-01 4. / &
       WATER 6.
    STOIC 3 MIXED PROPANE -1. / OXYGEN -5. / CARBO-01 3. / &
       WATER 4.
    STOIC 4 MIXED HYDROGEN -2. / OXYGEN -1. / WATER 2.
    CONV 1 MIXED METHANE 1.
    CONV 2 MIXED ETHANE 1.
    CONV 3 MIXED PROPANE 1.
    CONV 4 MIXED HYDROGEN 1.
    HEAT-RXN REACNO=1 CID=METHANE / REACNO=2 CID=ETHANE / &
       REACNO=3 CID=PROPANE / REACNO=4 CID=HYDROGEN
BLOCK F-401-2 RSTOIC
    PARAM TEMP=572.0000000 PRES=14.50377377 HEAT-OF-REAC=YES
    STOIC 1 MIXED METHANE -1. / OXYGEN -2. / CARBO-01 1. / &
        WATER 2.
    STOIC 2 MIXED HYDROGEN -2. / OXYGEN -1. / WATER 2.
    CONV 1 MIXED METHANE 1.
    CONV 2 MIXED HYDROGEN 1.
    HEAT-RXN REACNO=1 CID=METHANE / REACNO=2 CID=HYDROGEN
BLOCK R-101 RSTOIC
    PARAM TEMP=1166.000000 PRES=-7.251886887 SERIES=NO
                                                        æ
       HEAT-OF-REAC=YES
    STOIC 1 MIXED ETHANE -1. / HYDROGEN -1. / METHANE 2.
    STOIC 2 MIXED ETHANE -1. / ETHENE 1. / HYDROGEN 1.
    STOIC 3 MIXED ETHANE -3. / PROPENE 2. / HYDROGEN 3.
    STOIC 4 MIXED ETHANE -3. / PROPANE 2. / HYDROGEN 1.
    STOIC 5 MIXED ETHANE -3. / BENZENE 1. / HYDROGEN 6.
    STOIC 6 MIXED ETHANE -4. / TOLUENE 1. / METHANE 1. / &
       HYDROGEN 6.
    STOIC 7 MIXED ETHANE -4. / P-XYL-01 1. / HYDROGEN 7.
    STOIC 8 MIXED ETHANE -5. / 1:3:5-01 1. / METHANE 1. /
       HYDROGEN 7.
    STOIC 9 MIXED ETHENE -1. / HYDROGEN -2. / METHANE 2.
    STOIC 10 MIXED ETHENE -3. / PROPENE 2.
    STOIC 11 MIXED ETHENE -3. / HYDROGEN -2. / PROPANE 2.
    STOIC 12 MIXED ETHENE -3. / BENZENE 1. / HYDROGEN 3.
    STOIC 13 MIXED ETHENE -4. / TOLUENE 1. / HYDROGEN 2. / &
       METHANE 1.
    STOIC 14 MIXED ETHENE -4. / P-XYL-01 1. / HYDROGEN 3.
    STOIC 15 MIXED ETHENE -5. / 1:3:5-01 1. / HYDROGEN 2. / &
```

```
METHANE 1.
STOIC 16 MIXED PROPANE -2. / BENZENE 1. / HYDROGEN 5.
STOIC 17 MIXED PROPANE -3. / TOLUENE 1. / ETHANE 1. / &
   HYDROGEN 5.
STOIC 18 MIXED PROPANE -3. / P-XYL-01 1. / METHANE 1. / &
   HYDROGEN 5.
STOIC 19 MIXED PROPENE -1. / HYDROGEN -1. / METHANE 1. / &
   ETHENE 1.
STOIC 20 MIXED PROPENE -2. / BENZENE 1. / HYDROGEN 3.
STOIC 21 MIXED PROPENE -3. / TOLUENE 1. / ETHANE 1. /
   HYDROGEN 2.
STOIC 22 MIXED PROPENE -3. / P-XYL-01 1. / METHANE 1. / &
   HYDROGEN 2.
STOIC 23 MIXED PROPANE -3. / 1:3:5-01 1. / HYDROGEN 6.
STOIC 24 MIXED PROPENE -3. / 1:3:5-01 1. / HYDROGEN 3.
CONV 1 MIXED ETHANE 0.07582
CONV 2 MIXED ETHANE 0.05904
CONV 3 MIXED ETHANE 0.00718
CONV 4 MIXED ETHANE 0.00685
CONV 5 MIXED ETHANE 0.17372
CONV 6 MIXED ETHANE 0.0925
CONV 7 MIXED ETHANE 0.01934
CONV 8 MIXED ETHANE 0.02982
CONV 9 MIXED ETHENE 0.1871
CONV 10 MIXED ETHENE 0.0177
CONV 11 MIXED ETHENE 0.0169
CONV 12 MIXED ETHENE 0.4287
CONV 13 MIXED ETHENE 0.2283
CONV 14 MIXED ETHENE 0.0477
CONV 15 MIXED ETHENE 0.0736
CONV 16 MIXED PROPANE 0.18153
CONV 17 MIXED PROPANE 0.2333
CONV 18 MIXED PROPANE 0.1037
CONV 19 MIXED PROPENE 0.3487
CONV 20 MIXED PROPENE 0.228
CONV 21 MIXED PROPENE 0.2931
CONV 22 MIXED PROPENE 0.1303
CONV 23 MIXED PROPANE 0.064656
CONV 24 MIXED PROPENE 0.08119
HEAT-RXN REACNO=1 CID=ETHANE TEMP=1166.000000 / REACNO=2 &
    CID=ETHANE TEMP=1166.000000 / REACNO=3 CID=ETHANE &
   TEMP=1166.000000 / REACNO=4 CID=ETHANE TEMP=1166.000000 / &
   REACNO=5 CID=ETHANE TEMP=1166.000000 / REACNO=6 &
   CID=ETHANE TEMP=1166.000000 / REACNO=7 CID=ETHANE &
    TEMP=1166.000000 / REACNO=8 CID=ETHANE TEMP=1166.000000 /
   REACNO=9 CID=ETHENE TEMP=1166.000000 / REACNO=10 &
   CID=ETHENE TEMP=1166.000000 / REACNO=11 CID=ETHENE &
   TEMP=1166.000000 / REACNO=12 CID=ETHENE TEMP=1166.000000 / &
   REACNO=13 CID=ETHENE TEMP=1166.000000 / REACNO=14 &
   CID=ETHENE TEMP=1166.000000 / REACNO=15 CID=ETHENE &
    TEMP=1166.000000 / REACNO=16 CID=PROPANE
                                             &
   TEMP=1166.000000 / REACNO=17 CID=PROPANE
                                              δ
   TEMP=1166.000000 / REACNO=18 CID=PROPANE
                                             æ
   TEMP=1166.000000 / REACNO=19 CID=PROPENE
                                              æ
   TEMP=1166.000000 / REACNO=20 CID=PROPENE &
   TEMP=1166.000000 / REACNO=21 CID=PROPENE &
   TEMP=1166.000000 / REACNO=22 CID=PROPENE &
```

```
TEMP=1166.000000 / REACNO=23 CID=PROPANE &
        TEMP=1166.000000 / REACNO=24 CID=PROPENE &
       TEMP=1166.000000
BLOCK P-201 PUMP
    PARAM PRES=100. EFF=0.8
BLOCK P-202 PUMP
    PARAM DELP=10. EFF=0.8
BLOCK P-301 PUMP
   PARAM PRES=93. EFF=0.8
BLOCK P-302 PUMP
    PARAM PRES=109. EFF=0.8
BLOCK P-303 PUMP
    PARAM PRES=45. EFF=0.8
BLOCK P-304 PUMP
    PARAM DELP=10. EFF=0.8
BLOCK P-401 PUMP
    PARAM PRES=230. EFF=0.8
BLOCK T-101-3 COMPR
    PARAM TYPE=ISENTROPIC PRES=53.66396296 SEFF=0.8 NPHASE=2 &
        SB-MAXIT=30 SB-TOL=0.0001 MODEL-TYPE=TURBINE
    BLOCK-OPTION FREE-WATER=NO
BLOCK T-104 COMPR
    PARAM TYPE=ISENTROPIC PRES=53.66396296 SEFF=0.8 NPHASE=2 &
        SB-MAXIT=30 SB-TOL=0.0001 MODEL-TYPE=TURBINE
    BLOCK-OPTION FREE-WATER=NO
BLOCK T-105 COMPR
    PARAM TYPE=ISENTROPIC PRES=53.66396296 SEFF=0.8 NPHASE=2 &
        SB-MAXIT=30 SB-TOL=0.0001 MODEL-TYPE=TURBINE
    BLOCK-OPTION FREE-WATER=NO
BLOCK T-401 COMPR
    PARAM TYPE=ISENTROPIC PRES=240. SEFF=0.8 NPHASE=2 &
        SB-MAXIT=30 SB-TOL=0.0001 MODEL-TYPE=TURBINE
    BLOCK-OPTION FREE-WATER=NO
BLOCK C-201-3 MCOMPR
    PARAM NSTAGE=4 TYPE=ISENTROPIC SB-MAXIT=30 SB-TOL=0.0001
    FEEDS S-108 1
    PRODUCTS S-204 1 L / S-203 2 L / S-202 3 L / S-201 &
       4
    COMPR-SPECS 1 PRATIO=2.465032536 SEFF=0.8 / 2 &
       PRATIO=2.465032536 SEFF=0.8 / 3 PRATIO=2.465032536 &
        SEFF=0.8 / 4 PRATIO=2.465032536 SEFF=0.8
    COOLER-SPECS 1 TEMP=104.0000000 PDROP=10.00000000 / 2 &
       TEMP=104.0000000 PDROP=10.00000000 / 3 TEMP=104.0000000 &
       PDROP=10.00000000 / 4 TEMP=104.0000000 PDROP=10.00000000
```

```
BLOCK PR-201, VALVE
    PARAM P-OUT=200.000000
BLOCK V-201 VALVE
    PARAM P-OUT=100.
BLOCK V-202 VALVE
    PARAM P-OUT=100.
BLOCK V-203 VALVE
    PARAM P-OUT=100.
DESIGN-SPEC BTXFLOW
    DEFINE BTXOUT STREAM-VAR STREAM=S-315 SUBSTREAM=MIXED &
        VARIABLE=MASS-FLOW UOM="MMtons/year"
    SPEC "BTXOUT" TO "0.5"
    TOL-SPEC "0.01"
    VARY STREAM-VAR STREAM=S-101 SUBSTREAM=MIXED &
        VARIABLE=MASS-FLOW UOM="MMtons/year"
    LIMITS "0" "100000"
DESIGN-SPEC CH4SPLIT
    DEFINE FURNQ BLOCK-VAR BLOCK=F-101 VARIABLE=NET-DUTY &
        SENTENCE=RESULTS UOM="cal/sec"
    DEFINE HEATO BLOCK-VAR BLOCK=F-101, VARIABLE=NET-DUTY &
        SENTENCE=RESULTS UOM="cal/sec"
    DEFINE REACTO BLOCK-VAR BLOCK=R-101 VARIABLE=NET-DUTY &
        SENTENCE=RESULTS UOM="cal/sec"
    SPEC "FURNQ" TO "- (HEATQ+REACTQ)"
    TOL-SPEC "0.1"
    VARY BLOCK-VAR BLOCK=V-206 SENTENCE=FRAC VARIABLE=FRAC &
        ID1=S-221
    LIMITS "0" "1"
DESIGN-SPEC SEP1PUR
    DEFINE BENBOT MASS-FLOW STREAM=S-310 SUBSTREAM=MIXED &
        COMPONENT=BENZENE UOM="kg/hr"
    DEFINE BENFEED MASS-FLOW STREAM=S-210 SUBSTREAM=MIXED &
       COMPONENT=BENZENE UOM="kg/hr"
    SPEC "BENBOT/BENFEED" TO "0.9"
    TOL-SPEC "0.01"
    VARY BLOCK-VAR BLOCK=COL-301 VARIABLE=MASS-D:F &
        SENTENCE=COL-SPECS
    LIMITS "0" "1"
DESIGN-SPEC STEAMAMT
    DEFINE VAPFRAC STREAM-VAR STREAM=S-403 SUBSTREAM=MIXED &
        VARIABLE=VFRAC
    SPEC "VAPFRAC" TO "1"
    TOL-SPEC "0.1"
    VARY STREAM-VAR STREAM=S-401 SUBSTREAM=MIXED &
        VARIABLE=MASS-FLOW UOM="kg/hr"
    LIMITS "0" "6000000"
DESIGN-SPEC STEAMT
    DEFINE STEAMT STREAM-VAR STREAM=S-404 SUBSTREAM=MIXED &
        VARIABLE=TEMP UOM="F"
```

```
SPEC "STEAMT" TO "465"
    TOL-SPEC "1"
    VARY STREAM-VAR STREAM=S-401 SUBSTREAM=MIXED &
        VARIABLE=MASS-FLOW UOM="kg/hr"
    LIMITS "400000" "500000"
DESIGN-SPEC STM1SPL
    DEFINE STM1TOUT STREAM-VAR STREAM=S-308 SUBSTREAM=MIXED &
        VARIABLE=VFRAC
    SPEC "STM1TOUT" TO "0.05"
    TOL-SPEC "0.01"
   VARY BLOCK-VAR BLOCK=V-406-7 SENTENCE=FRAC VARIABLE=FRAC &
       ID1=S-408
    LIMITS "0.30" "0.40"
DESIGN-SPEC STM2SPL
    DEFINE STM2TOUT STREAM-VAR STREAM=S-319 SUBSTREAM=MIXED &
        VARIABLE=VFRAC
    SPEC "STM2TOUT" TO "0.05"
    TOL-SPEC "0.01"
    VARY BLOCK-VAR BLOCK=V-406-7 SENTENCE=FRAC VARIABLE=FRAC &
       ID1=S-409
    LIMITS "0.35" "0.5"
DESIGN-SPEC STM3SPL
    DEFINE STM3TOUT STREAM-VAR STREAM=S-217 SUBSTREAM=MIXED &
        VARIABLE=VFRAC
    SPEC "STM3TOUT" TO "0.05"
    TOL-SPEC "0.01"
    VARY BLOCK-VAR BLOCK=V-405 SENTENCE=FRAC VARIABLE=FRAC &
        ID1=S-406
    LIMITS "0.05" "0.1"
EO-CONV-OPTI
CALCULATOR FURNIAIR
    IN-UNITS MET PRESSURE=bar TEMPERATURE=C DELTA-T=C PDROP=bar &
        INVERSE-PRES='1/bar'
    DEFINE XCH4 MOLE-FRAC STREAM=S-220 SUBSTREAM=MIXED &
       COMPONENT=METHANE
    DEFINE XC2H6 MOLE-FRAC STREAM=S-220 SUBSTREAM=MIXED &
        COMPONENT=ETHANE
    DEFINE XC3H8 MOLE-FRAC STREAM=S-220 SUBSTREAM=MIXED &
        COMPONENT=PROPANE
    DEFINE XH2 MOLE-FRAC STREAM=S-220 SUBSTREAM=MIXED &
        COMPONENT=HYDROGEN
    DEFINE AIRREQ STREAM-VAR STREAM=S-109 SUBSTREAM=MIXED &
        VARIABLE=MOLE-FLOW UOM="kmol/hr"
    DEFINE XO2 MOLE-FRAC STREAM=S-109 SUBSTREAM=MIXED &
        COMPONENT=OXYGEN
    DEFINE FUELFLO STREAM-VAR STREAM=S-220 SUBSTREAM=MIXED &
       VARIABLE=MOLE-FLOW UOM="kmol/hr"
     AIRREQ = 1.2 \times FUELFLO \times (2 \times XCH4 + XH2/2) / XO2
ਜ
    READ-VARS XCH4 XC2H6 XC3H8 XO2 FUELFLO XH2
   WRITE-VARS AIRREQ
   EXECUTE BEFORE BLOCK V-103
```

```
CALCULATOR FURN2AIR
    IN-UNITS MET PRESSURE=bar TEMPERATURE=C DELTA-T=C PDROP=bar &
        INVERSE-PRES='1/bar'
    DEFINE AIRREQ STREAM-VAR STREAM=S-410 SUBSTREAM=MIXED &
       VARIABLE=MOLE-FLOW UOM="kmol/hr"
    DEFINE H2FLOW MOLE-FLOW STREAM=S-221 SUBSTREAM=MIXED &
       COMPONENT=HYDROGEN UOM="kmol/hr"
    DEFINE CH4FLOW MOLE-FLOW STREAM=S-221 SUBSTREAM=MIXED &
       COMPONENT=METHANE UOM="kmol/hr"
F
           AIRREO = 1.2*(2*CH4FLOW+H2FLOW/2)/0.21
   READ-VARS H2FLOW CH4FLOW
   WRITE-VARS AIRREQ
   EXECUTE AFTER BLOCK V-206
CALCULATOR PUMPS
    IN-UNITS MET PRESSURE=bar TEMPERATURE=C DELTA-T=C PDROP=bar &
        INVERSE-PRES='1/bar'
    DEFINE PUMP1P BLOCK-VAR BLOCK=P-201 VARIABLE=PRES &
       SENTENCE=PARAM UOM="bar"
    DEFINE VALVE2P BLOCK-VAR BLOCK=V-203 VARIABLE=P-OUT &
       SENTENCE=PARAM UOM="bar"
    DEFINE VALVE3P BLOCK-VAR BLOCK=V-202 VARIABLE=P-OUT &
       SENTENCE=PARAM UOM="bar"
    DEFINE VALVE4P BLOCK-VAR BLOCK=V-201 VARIABLE=P-OUT
                                                        æ
       SENTENCE=PARAM UOM="bar"
    DEFINE COLP BLOCK-VAR BLOCK=COL-301 VARIABLE=PRES &
       SENTENCE=PROFILE ID1=14 UOM="bar"
     PUMP1P = COLP
F
     VALVE2P = COLP
F
F
     VALVE3P = COLP
F
     VALVE4P = COLP
   READ-VARS COLP
   WRITE-VARS PUMP1P VALVE2P VALVE3P VALVE4P
    EXECUTE AFTER BLOCK COL-301
CONV-OPTIONS
    PARAM TEAR-METHOD=WEGSTEIN TOL=0.0001 SPEC-METHOD=SECANT
STREAM-REPOR MOLEFLOW MASSFLOW MOLEFRAC MASSFRAC
DISABLE
   CALCULATOR PUMPS
    DESIGN-SPEC STEAMAMT
```

A-5.2. Stream Report

HEAT1 HEAT2 HEAT3 HEAT4							
TO :	F-401-2 F-401-2,	HEAT2 H-203 H-203, HEAT	COL-301 RB-301	COL-302 RB-302			
STREAM ATTRIBUTES: HEAT	1 2050-00	1 5005.05		0.0000.07			
Q BTU/HR TBEG F TEND F	94.7830	-1.5007+07 129.6288 181.4000	195.7211	323.0151			
S-101 S-102 S-103 S-10	04 S-105						
STREAM ID FROM : TO :		S-102 T-102 V-101-2	T-101	T-103	V-101-2		
SUBSTREAM: MIXED PHASE:		VAPOR	MIXED	MIXED	MIXED		
ETHANE	0.0 62.8392 6095.4040 0.0 125.6784 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0	0.0 125.6784 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0	7314.0134 6865.2255 780.6162 192.2615 63.7247 125.7796 4.4111 3.5762-02 9.2656-04 0.0 0.0	172.6175 238.5242 22.0910 16.4274 3.4654 108.5439 10.4562 0.1014 $1.8392-07$ 0.0 0.0 0.0	7549.4701 1.3199+04 802.7072 334.3674 67.1900 234.3235 14.8673 0.1372 9.2675-04 0.0 0.0		
HYDROGEN METHANE ETHANE ETHENE PROPANE PROPENE BENZENE TOLUENE P-XYL-01 1:3:5-01 WATER OXYGEN NITROGEN CARBO-01	$\begin{array}{c} 0.0\\ 1.0000-02\\ 0.9700\\ 0.0\\ 2.0000-02\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0$	$\begin{array}{c} 0.0\\ 1.0000-02\\ 0.9700\\ 0.0\\ 2.0000-02\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0$	$\begin{array}{r} 8.3318-02\\ 0.4369\\ 0.4101\\ 4.6629-02\\ 1.1485-02\\ 3.8065-03\\ 7.5133-03\\ 2.6350-04\\ 2.1362-06\\ 5.5347-08\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0$	$\begin{array}{c} 9.1005-08\\ 0.3017\\ 0.4168\\ 3.8605-02\\ 2.8708-02\\ 6.0559-03\\ 0.1897\\ 1.8273-02\\ 1.7728-04\\ 3.2141-10\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0$	$5.9110-02 \\ 0.3199 \\ 0.5594 \\ 3.4017-02 \\ 1.4170-02 \\ 2.8474-03 \\ 9.9302-03 \\ 6.3005-04 \\ 5.8146-06 \\ 3.9274-08 \\ 0.0 $		

COMPONENTS: LB/HR					
HYDROGEN	0.0	0.0	2811.7695	1.0498-04	
METHANE	1008.1145			2769.2615	1.2111+05
ETHANE	1.8329+05	1.8329+05	2.0643+05	7172.3375	3.9689+05
ETHENE	0.0	0.0	2.1899+04	619.7359	2.2519+04
PROPANE	5541.9815	5541.9815	8478.0651	724.3933	1.4744+04
PROPENE	0.0	0.0	2681.5743	145.8246	2827.3989
BENZENE	0.0	0.0	9825.0998		1.8304+04
TOLUENE	0.0	0.0	406.4449	963.4368	1369.8817
P-XYL-01	0.0	0.0	3.7968	10.7700	14.5668
1:3:5-01	0.0	0.0	0.1114		0.1114
WATER	0.0	0.0	0.0	0.0	0.0
OXYGEN	0.0	0.0	0.0	0.0	0.0
NITROGEN	0.0	0.0	0.0	0.0	0.0
CARBO-01	0.0	0.0	0.0	0.0	0.0
COMPONENTS: MASS FRAC					
HYDROGEN	0.0	0.0	7.6019-03		
METHANE	5.3104-03		0.3172	0.1326	0.2086
ETHANE	0.9655	0.9655	0.5581	0.3434	
ETHENE	0.0	0.0	5.9207-02		
PROPANE	2.9193-02	2.9193-02	2.2921-02		
PROPENE	0.0	0.0	7.2499-03	6.9824-03	
BENZENE	0.0	0.0	2.6563-02		
TOLUENE	0.0	0.0	1.0989-03		
P-XYL-01	0.0	0.0	1.0265-05		
1:3:5-01	0.0	0.0	3.0109-07	1.0585-09	1.9185-07
WATER	0.0	0.0	0.0	0.0	0.0
OXYGEN	0.0	0.0	0.0	0.0	0.0
NITROGEN	0.0	0.0	0.0	0.0	0.0
CARBO-01	0.0	0.0	0.0	0.0	0.0
TOTAL FLOW:	COOD 001 C	COO2 001 C	1 (741)04		0 0 0 0 7 1 0 4
LBMOL/HR	6283.9216				
LB/HR	1.8984+05	1.8984+05		2.0885+04	
CUFT/HR	1.1683+05	5.0021+05	1.3984+06	6.9614+04	1.9998+06
STATE VARIABLES: TEMP F	00 0000	-37.4871	26 0520	166 6616	17 0020
PRES PSIA	80.0000 264.6959	53.6640		166.6616 53.6640	-17.0838 53.6640
VFRAC	1.0000	1.0000	53.6640	0.9943	0.9874
LFRAC	0.0	0.0	9.1642-03	5.6769-03	1.2647-02
SFRAC				0.0	
BTIL/I.BMOI.	-3 6611+04	-3 7670+04	-2 9149+04	-1 6819+04	-3 1119+04
BTU/LBMOL BTU/LB	-1211.8994	-1246.9321	-1319.2949	-460.8216	-1264.7549
				-9.6240+06	
ENTROPY:	2.0000.00	2.00/1/00	1.0,00,00	2.0210.00	
BTU/LBMOL-R	-47.7691	-47.1251	-29.9197	-31,7680	-34.1442
BTU/LB-R				-0.8704	
DENSITY:					
	5.3787-02	1.2563-02	1.1971-02	8.2200-03	1.1800-02
LB/CUFT					
AVG MW	30.2099	30.2099	22.0943	0.3000 36.4969	24.6048
S-106 S-106, S-107 S-					
STREAM ID	S-106	S-106,	S-107	S-108	
FROM :	H-IOI	F-101,	K-IOI	H-IOI	

MAX CONV. ERROR					
SUBSTREAM: MIXE		0.0	0.0	0.0	0.0
PHASE:	VAPOR	VAPOR	VAPOR	MIXED	VAPOR
COMPONENTS: LBMG HYDROGEN METHANE ETHANE ETHENE PROPANE PROPENE BENZENE TOLUENE P-XYL-01 1:3:5-01 WATER	1394.8100 7549.4701 1.3199+04 802.7072 334.3674 67.1900 234.3235 14.8673 0.1372 9.2675-04 0.0	7549.5749 1.3199+04 802.7072 334.3674 67.1900 234.3241 14.8673 0.1372 9.2676-04 0.0	1.0331+04 7103.7498 802.7072 208.6890 67.1900 1151.3585 398.4794 88.0037 99.5612 0.0	9298.7333 1.0331+04 7103.7498 802.7072 208.6890 67.1900 1151.3585 398.4794 88.0037 99.5612 0.0	0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0
OXYGEN NITROGEN	0.0 0.0	0.0 0.0	0.0	0.0 0.0	2032.4726 7645.9682
CARBO-01	0.0	0.0	0.0	0.0	0.0
COMPONENTS: MOLI HYDROGEN METHANE ETHANE ETHENE PROPANE PROPENE BENZENE TOLUENE P-XYL-01 1:3:5-01 WATER OXYGEN NITROGEN CARBO-01	$\begin{array}{c} 5.9110-02\\ 0.3199\\ 0.5594\\ 3.4017-02\\ 1.4170-02\\ 2.8474-03\\ 9.9302-03\\ 6.3005-04\\ 5.8146-06\\ 3.9274-08\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0$	0.5594 3.4017-02 1.4170-02 2.8474-03 9.9302-03 6.3005-04 5.8145-06	0.3496 0.2404 2.7165-02 7.0624-03 2.2738-03 3.8964-02 1.3485-02 2.9782-03	$\begin{array}{c} 0.3147\\ 0.3496\\ 0.2404\\ 2.7165-02\\ 7.0624-03\\ 2.2738-03\\ 3.8964-02\\ 1.3485-02\\ 2.9782-03\\ 3.3693-03\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ \end{array}$	0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0
COMPONENTS: LB/I HYDROGEN METHANE ETHANE ETHENE PROPANE PROPENE BENZENE TOLUENE P-XYL-01 1:3:5-01 WATER OXYGEN NITROGEN CARBO-01 COMPONENTS: MASS HYDROGEN METHANE ETHANE ETHENE	2811.7696 1.2111+05 3.9689+05 2.2519+04 1.4744+04 2827.3989 1.8304+04 1369.8817 14.5668 0.1114 0.0 0.0 0.0 0.0 0.0	0.2086 0.6836	1.6574+05 2.1361+05 2.2519+04 9202.4586 2827.3990 8.9937+04 3.6716+04 9343.1188 1.1967+04 0.0 0.0 0.0 0.0 3.2286-02 0.2855 0.3679	1.8745+04 1.6574+05 2.1361+05 2.2519+04 9202.4586 2827.3990 8.9937+04 3.6716+04 9343.1188 1.1967+04 0.0 0.0 0.0 3.2286-02 0.2855 0.3679 3.8786-02	$\begin{array}{c} 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0$

PROPENE BENZENE TOLUENE P-XYL-01 1:3:5-01 WATER OXYGEN NITROGEN CARBO-01 TOTAL FLOW:	4.8698-03 3.1526-02 2.3594-03 2.5089-05 1.9185-07 0.0 0.0 0.0 0.0	3.1526-02 2.3594-03 2.5089-05	6.3238-02 1.6092-02	0.1549 6.3238-02 1.6092-02	0.0 0.0 0.0 0.0 0.0 0.0 0.2329 0.7671 0.0
	2.3597+04 5.8060+05 9.3271+06	5.8060+05	5.8060+05	5.8060+05	2.7923+05
TEMP F PRES PSIA VFRAC LFRAC SFRAC		43.6640	36.4121	102.9178 26.4121 0.9928 7.2463-03 0.0	14.5038
	-1.1199+04 -455.1462 -2.6426+08	-438.3481	-93.4799	-903.0861	-0.1002
BTU/LBMOL-R BTU/LB-R DENSITY:				-18.5047 -0.9418	
	2.5299-03 6.2249-02 24.6048	6.1558-02	4.0988-02		7.2689-02
S-110 S-201 S-202 S-	203 S-204				
STREAM ID FROM : TO :	S-110 F-101	S-201 C-201-3 COL-201	S-202 C-201-3 V-202	S-203 C-201-3 V-203	
SUBSTREAM: MIXED PHASE: COMPONENTS: LBMOL/HR	VAPOR	MIXED	LIQUID	LIQUID	LIQUID
HYDROGEN METHANE ETHANE ETHENE PROPANE PROPENE BENZENE TOLUENE P-XYL-01 1:3:5-01 WATER OXYGEN NITROGEN CARBO-01 COMPONENTS: MOLE FRA	0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0	$\begin{array}{c} 9298.7333\\ 1.0241+04\\ 6968.5278\\ 790.4191\\ 199.1530\\ 65.2797\\ 273.1375\\ 18.3614\\ 0.3389\\ 2.1683-02\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ \end{array}$	49.7312 72.2086 6.6169 5.0230 0.9903 267.8718 52.3147 2.8720 0.5189 0.0 0.0 0.0 0.0	300.5154 117.4845 16.5359 8.1958 0.0 0.0 0.0 0.0 0.0 0.0	3.9946-06 15.4312 24.5482 2.2003 1.7713 0.3815 309.8338 210.3187 68.2569 90.8248 0.0 0.0 0.0 0.0
HYDROGEN	0.0	0.3338	3.0803-08	1.2889-08	5.5207-09

METHANE ETHANE ETHENE PROPANE PROPENE BENZENE TOLUENE P-XYL-01 1:3:5-01 WATER OXYGEN NITROGEN CARBO-01	0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.2196 3.1156-02 0.7032 4.5961-02	$\begin{array}{c} 0.3677\\ 0.2502\\ 2.8376-02\\ 7.1497-03\\ 2.3436-03\\ 9.8057-03\\ 6.5918-04\\ 1.2166-05\\ 7.7842-07\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0$	$\begin{array}{c} 0.1085\\ 0.1576\\ 1.4443-02\\ 1.0964-02\\ 2.1616-03\\ 0.5847\\ 0.1142\\ 6.2686-03\\ 1.1325-03\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0$	$\begin{array}{c} 4.8456-02\\ 7.5011-02\\ 6.7685-03\\ 5.3464-03\\ 1.0501-03\\ 0.5860\\ 0.2291\\ 3.2247-02\\ 1.5983-02\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0$	3.3927-02 3.0409-03 2.4481-03 5.2728-04 0.4282 0.2907
COMPONENTS: LB/HR					
HYDROGEN METHANE ETHANE ETHENE PROPANE PROPENE BENZENE TOLUENE P-XYL-01 1:3:5-01 WATER OXYGEN NITROGEN CARBO-01 COMPONENTS: MASS FRAC	$\begin{array}{c} 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0$	$\begin{array}{c} 1.8745+04\\ 1.6429+05\\ 2.0954+05\\ 2.2174+04\\ 8781.9546\\ 2747.0108\\ 2.1336+04\\ 1691.8323\\ 35.9784\\ 2.6062\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\end{array}$	$\begin{array}{c} 2.8448-05\\797.8258\\2171.2867\\185.6288\\221.4973\\41.6738\\2.0924+04\\4820.3074\\304.9079\\62.3656\\0.0\\0.0\\0.0\\0.0\\0.0\\0.0\end{array}$	$\begin{array}{c} 1.3323-05\\ 398.6320\\ 1156.6322\\ 97.3707\\ 120.8966\\ 22.6597\\ 2.3474+04\\ 1.0825+04\\ 1755.5779\\ 985.0911\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\end{array}$	247.5598 738.1570 61.7273 78.1101 16.0546 2.4202+04 1.9379+04
HYDROGEN METHANE ETHANE ETHENE PROPANE PROPENE BENZENE TOLUENE P-XYL-01 1:3:5-01 WATER OXYGEN NITROGEN CARBO-01 TOTAL FLOW:	$\begin{array}{c} 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0$	$\begin{array}{r} 4.1716-02\\ 0.3656\\ 0.4663\\ 4.9347-02\\ 1.9544-02\\ 6.1133-03\\ 4.7482-02\\ 3.7651-03\\ 8.0068-05\\ 5.7999-06\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ \end{array}$	$\begin{array}{c} 9.6337-10\\ 2.7018-02\\ 7.3528-02\\ 6.2861-03\\ 7.5008-03\\ 1.4112-03\\ 0.7086\\ 0.1632\\ 1.0325-02\\ 2.1119-03\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ \end{array}$	$\begin{array}{c} 3.4307-10\\ 1.0264-02\\ 2.9782-02\\ 2.5072-03\\ 3.1130-03\\ 5.8347-04\\ 0.6044\\ 0.2787\\ 4.5205-02\\ 2.5365-02\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ \end{array}$	3.9366-03 1.1738-02 9.8157-04
LBMOL/HR LB/HR CUFT/HR	1.0872+04 2.9004+05 1.4528+07	2.7855+04 4.4935+05 2.1502+05	458.1474 2.9530+04 628.3836	512.7958 3.8836+04 769.1817	6.2886+04
STATE VARIABLES: TEMP F PRES PSIA VFRAC LFRAC SFRAC ENTHALPY:	1346.0000 14.5038 1.0000 0.0 0.0	104.0000 729.9998 0.9869 1.3132-02 0.0	104.0000 300.1988 0.0 1.0000 0.0	104.0000 125.8396 0.0 1.0000 0.0	104.0000 55.1066 0.0 1.0000 0.0
BTU/LB	-774.2822	-2.0222+04 -1253.5260 -5.6327+08	3386.3202 52.5377 1.5514+06	8841.0617 116.7377 4.5337+06	56.5828

ENTROPY:					
BTU/LBMOL-R					
BTU/LB-R	0.3277	-1.4927	-0.8536	-0.8195	-0.8618
DENSITY:					
LBMOL/CUFT	7.4838-04 1.9964-02	0.1295	0.7291	0.6667	0.5954
LB/CUFT	1.9964-02	2.0898	46.9935	50.4904	51.7502
AVG MW	26.6768	16.1318	64.4551	75.7344	86.9111
S-205 S-206 S-207 S-2	208 S-209				
STREAM ID	S-205	S-206	S-207	S-208	S-209
FROM :	COL-201	V-201	V-202 V-203-5	V-203	P-201
то :	V-201	V-203-5	V-203-5	V-203-5	V-203-5
SUBSTREAM: MIXED					
PHASE:	LIQUID	MIXED	MIXED	MIXED	LIQUID
COMPONENTS: LBMOL/HR			4 4440 05		
HYDROGEN	2.7359-05			6.6093-06	
METHANE	82.6084	82.6084	49.7312	24.8481	
ETHANE	103.3017	103.3017	72.2086	38.4651	24.5482
ETHENE	9.8029	9.8029	6.6169	3.4709	2.2003
PROPANE	6.8914	6 8914	5.0230	2.7416	
PROPENE	1.5550	1.5550	0.9903	0.5385	
BENZENE		147.3575		300.5154	
TOLUENE			52.3147		
P-XYL-01	0.3031			16.5359	
1:3:5-01	2.0756-02			8.1958	
WATER	0.0	0.0	0.0	0.0	0.0
OXYGEN	0.0	0.0	0.0	0.0	0.0
NITROGEN	0.0	0.0	0.0	0.0	0.0
CARBO-01	0.0	0.0	0.0	0.0	0.0
COMPONENTS: MOLE FRAC					
HYDROGEN	7.4795-08	7.4795-08	3.0803-08	1.2889-08	5.5207-09
METHANE	0.2258	0.2258	0.1085	4.8456-02	
ETHANE	0.2824	0.2824	0.1576	7.5011-02	3.3927-02
ETHENE	2.6799-02	2.6799-02	1.4443-02	6.7685-03	3.0409-03
PROPANE	1.8840-02	1.8840-02	1.0964-02	5.3464-03	
PROPENE	4.2511-03	4.2511-03	2.1616-03	1.0501-03	5.2728-04
BENZENE	0.4028	0.4028	0.5847	0.5860	0.4282
TOLUENE	3.8137-02	3.8137-02	0.1142	0.2291	0.2907
P-XYL-01	8.2867-04	8.2867-04	6.2686-03	3.2247-02	9.4334-02
1:3:5-01	5.6744-05	5.6744-05	1.1325-03	1.5983-02	0.1255
WATER	0.0	0.0	0.0	0.0	0.0
OXYGEN	0.0	0.0	0.0	0.0	0.0
NITROGEN	0.0	0.0	0.0	0.0	0.0
CARBO-01	0.0	0.0	0.0	0.0	0.0
COMPONENTS: LB/HR					
HYDROGEN	5.5153-05	5.5153-05	2.8448-05	1.3323-05	8.0526-06
METHANE	1325.2672	1325.2672	797.8258	398.6320	247.5598
ETHANE	3106.2458	3106.2458	2171.2867	1156.6322	738.1570
ETHENE	275.0077	275.0077	185.6288	97.3707	61.7273
PROPANE	303.8877	303.8877	221.4973	120.8966	78.1101
PROPENE	65.4362	65.4362	41.6738	22.6597	16.0546
BENZENE	1.1511+04	1.1511+04	2.0924+04	2.3474+04	2.4202+04
TOLUENE	1285.3839	1285.3839	4820.3074	1.0825+04	1.9379+04
P-XYL-01	32.1816	32.1816	304.9079	1755.5779	

1:3:5-01 WATER OXYGEN NITROGEN CARBO-01 COMPONENTS: MASS FRA	2.4948 0.0 0.0 0.0 0.0	2.4948 0.0 0.0 0.0 0.0	62.3656 0.0 0.0 0.0 0.0 0.0	985.0911 0.0 0.0 0.0 0.0	1.0917+04 0.0 0.0 0.0 0.0 0.0
HYDROGEN METHANE ETHANE ETHENE PROPANE PROPENE BENZENE TOLUENE	3.0801-09 7.4010-02 0.1735 1.5358-02 1.6971-02 3.6543-03 0.6428 7.1783-02	3.0801-09 7.4010-02 0.1735 1.5358-02 1.6971-02 3.6543-03 0.6428 7.1783-02	9.6337-10 2.7018-02 7.3528-02 6.2861-03 7.5008-03 1.4112-03 0.7086 0.1632	3.4307-10 1.0264-02 2.9782-02 2.5072-03 3.1130-03 5.8347-04 0.6044 0.2787	1.2805-10 3.9366-03 1.1738-02 9.8157-04 1.2421-03 2.5530-04 0.3849 0.3082
P-XYL-01 1:3:5-01 WATER OXYGEN NITROGEN CARBO-01	1.7972-03 1.3932-04 0.0 0.0 0.0 0.0	1.7972-03 1.3932-04 0.0 0.0 0.0 0.0 0.0	1.0325-02 2.1119-03 0.0 0.0 0.0 0.0	4.5205-02 2.5365-02 0.0 0.0 0.0 0.0	0.1152 0.1736 0.0 0.0 0.0 0.0 0.0
TOTAL FLOW: LBMOL/HR LB/HR CUFT/HR STATE VARIABLES:	365.7911 1.7907+04 456.9599	365.7911 1.7907+04 8.8319+04		512.7958 3.8836+04 3.5235+04	
TEMP F PRES PSIA VFRAC LFRAC SFRAC ENTHALPY:	104.0000 729.9998 0.0 1.0000 0.0	20.7594 11.8000 0.5545 0.4455 0.0	63.1154 11.8000 0.3023 0.6977 0.0	85.5026 11.8000 0.1368 0.8632 0.0	103.9130 11.8000 0.0 1.0000 0.0
BTU/LBMOL BTU/LB BTU/HR ENTROPY:	-204.1864	-9995.5132 -204.1864 -3.6563+06	3386.3202 52.5377 1.5514+06	8841.0617 116.7377 4.5337+06	4906.9032 56.4589 3.5505+06
BTU/LBMOL-R BTU/LB-R DENSITY: LBMOL/CUFT	-48.8363 -0.9976 0.8005	-45.6916 -0.9334 4.1417-03	-0.8346	-61.6982 -0.8147 1.4554-02	-74.8959 -0.8618 0.5955
LB/CUFT AVG MW S-210 S-211 S-212 S-	39.1862 48.9529	0.2027 48.9529	0.4480	1.1022	
STREAM ID FROM : TO :	V-203-5	S-211 COL-201 H-201	H-201	H-202	
MAX CONV. ERROR: SUBSTREAM: MIXED	-1.2179-05	1.0193-05	0.0	0.0	0.0
PHASE: COMPONENTS: LBMOL/HR		VAPOR			
HYDROGEN METHANE ETHANE	172.6175	9298.7332 1.0158+04 6865.2255	1.0158+04		1.0158+04

ETHENE PROPANE PROPENE BENZENE TOLUENE P-XYL-01 1:3:5-01 WATER OXYGEN NITROGEN CARBO-01 COMPONENTS:	MOLE EDIC	22.0910 16.4274 3.4654 1025.5849 394.0676 87.9679 99.5603 0.0 0.0 0.0 0.0	780.6162 192.2615 63.7247 125.7796 4.4111 3.5762-02 9.2656-04 0.0 0.0 0.0 0.0	780.6162 192.2615 63.7247 125.7796 4.4111 3.5762-02 9.2656-04 0.0 0.0 0.0 0.0	780.6162 192.2615 63.7247 125.7796 4.4111 3.5762-02 9.2656-04 0.0 0.0 0.0 0.0	
	MOLL FRAC		0 2202	0 2202	0 2202	0 0000
HYDROGEN		2.5276-08	0.3383	0.3383	0.3383	0.3383
METHANE		8.3782-02	0.3695	0.3695	0.3695	0.3695
ETHANE		0.1158	0.2497	0.2497	0.2497	0.2497
ETHENE		1.0722-02	2.8397-02	2.8397-02	2.8397-02	
PROPANE		7.9733-03	6.9941-03	6.9941-03	6.9941-03	
PROPENE		1.6820-03	2.3182-03	2.3182-03	2.3182-03	
BENZENE		0.4978	4.5756-03	4.5756-03	4.5756-03	
TOLUENE		0.1913	1.6047-04	1.6047-04	1.6047-04	
P-XYL-01		4.2697-02		1.3010-06	1.3010-06	
1:3:5-01		4.8323-02	3.3706-08	3.3706-08	3.3706-08	3.3706-08
WATER		0.0	0.0	0.0	0.0	0.0
OXYGEN		0.0	0.0	0.0	0.0	0.0
NITROGEN		0.0	0.0	0.0	0.0	0.0
CARBO-01		0.0	0.0	0.0	0.0	0.0
COMPONENTS:	LB/HR	1 0 4 0 0 0 4	1 0745.04	1 0745.04	1 0745.04	1 0745-04
HYDROGEN		1.0498-04	1.8745+04	1.8745+04	1.8745+04	
METHANE		2769.2619	1.6297+05	1.6297+05	1.6297+05	
ETHANE		7172.3384 619.7360	2.0643+05	2.0643+05	2.0643+05	2.0643+05
ETHENE PROPANE		724.3933	2.1899+04 8478.0651	2.1899+04 8478.0651	2.1899+04 8478.0651	2.1899+04 8478.0651
PROPANE PROPENE		145.8246	2681.5743	2681.5743	2681.5743	2681.5743
BENZENE		8.0112+04	9825.0998	9825.0998	9825.0998	9825.0998
TOLUENE		3.6310+04	406.4449	406.4449	406.4449	406.4449
P-XYL-01		9339.3203	3.7968	3.7968	3.7968	3.7968
1:3:5-01		1.1967+04	0.1114	0.1114	0.1114	0.1114
WATER		0.0	0.0	0.0	0.0	0.0
OXYGEN		0.0	0.0	0.0	0.0	0.0
NITROGEN		0.0	0.0	0.0	0.0	0.0
CARBO-01		0.0	0.0	0.0	0.0	0.0
COMPONENTS:	MASS FRAC					
HYDROGEN		7.0380-10	4.3448-02	4.3448-02	4.3448-02	4.3448-02
METHANE		1.8566-02	0.3777	0.3777	0.3777	0.3777
ETHANE		4.8085-02	0.4785	0.4785	0.4785	0.4785
ETHENE		4.1549-03	5.0758-02	5.0758-02	5.0758-02	5.0758-02
PROPANE		4.8565-03	1.9651-02	1.9651-02	1.9651-02	1.9651-02
PROPENE		9.7764-04	6.2154-03	6.2154-03	6.2154-03	6.2154-03
BENZENE		0.5371	2.2773-02	2.2773-02	2.2773-02	2.2773-02
TOLUENE		0.2434	9.4206-04	9.4206-04	9.4206-04	9.4206-04
P-XYL-01		6.2613-02	8.8002-06	8.8002-06	8.8002-06	
1:3:5-01		8.0227-02	2.5813-07	2.5813-07	2.5813-07	2.5813-07
WATER		0.0	0.0	0.0	0.0	0.0
OXYGEN		0.0	0.0	0.0	0.0	0.0
NITROGEN		0.0	0.0	0.0	0.0	0.0
CARBO-01		0.0	0.0	0.0	0.0	0.0
TOTAL FLOW:						

LBMOL/HR LB/HR	1.4916+05	4.3144+05	2.7489+04 4.3144+05	4.3144+05	4.3144+05
CUFT/HR STATE VARIABLES:	2.2662+05	2.1456+05	2.2845+05	2.3339+05	2.6174+05
TEMP F PRES PSIA VFRAC LFRAC SFRAC	11.8000 0.2247 0.7753	729.9998 1.0000 0.0	126.3420 719.9998 1.0000 0.0 0.0	709.9998 1.0000 0.0	699.9998 1.0000 0.0
	40.0873	-1297.0781	-2.0122+04 -1282.0629 -5.5314+08	-1279.6658	-1244.8834
ENTROPY: BTU/LBMOL-R BTU/LB-R DENSITY:	-0.8499	-1.5132	-1.4854	-1.4797	-1.4214
LBMOL/CUFT LB/CUFT AVG MW	0.6582	2.0108	0.1203 1.8886 15.6950	1.8486	0.1050 1.6483 15.6950
S-215 S-216 S-217 S-2	18 S-219 				
STREAM ID FROM : TO :			S-217 H-203, V-401-3		PR-201,
SUBSTREAM: MIXED PHASE: COMPONENTS: LBMOL/HR	LIQUID	LIQUID	MIXED	VAPOR	VAPOR
HYDROGEN METHANE ETHANE ETHENE PROPANE PROPENE BENZENE TOLUENE P-XYL-01 1:3:5-01 WATER OXYGEN NITROGEN CARBO-01 COMPONENTS: MOLE FRAC	0.0 0.0 0.0 917.0410 383.6114 87.8297 0.2518 0.0 0.0 0.0 0.0	$\begin{array}{c} 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 4.8566-09\\ 1.8879-05\\ 3.6709-02\\ 99.3085\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.$	0.0	1394.8100 7314.0134 6865.2255 780.6162 192.2615 63.7247 125.7796 4.4111 3.5762-02 9.2656-04 0.0 0.0 0.0 0.0	2844.3385 0.0 0.0 0.0 0.0
HYDROGEN METHANE ETHANE ETHENE PROPANE PROPENE BENZENE TOLUENE P-XYL-01 1:3:5-01 WATER	0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.6603	1.9003-07	0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0	8.3318-02 0.4369 0.4101 4.6629-02 1.1485-02 3.8065-03 7.5133-03 2.6350-04 2.1362-06 5.5347-08 0.0	0.7354 0.2646 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.

OXYGEN NITROGEN CARBO-01	0.0 0.0 0.0	0.0 0.0 0.0	0.0 0.0 0.0	0.0 0.0 0.0	0.0 0.0 0.0
COMPONENTS: LB/HR HYDROGEN METHANE ETHANE ETHENE PROPANE PROPENE BENZENE TOLUENE P-XYL-01 1:3:5-01 WATER OXYGEN NITROGEN CARBO-01	$\begin{array}{c} 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 7.1633+04\\ 3.5346+04\\ 9324.6534\\ 30.2652\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ \end{array}$	$\begin{array}{c} 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 3.7936-07\\ 1.7395-03\\ 3.8973\\ 1.1936+04\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ \end{array}$	$\begin{array}{c} 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0$	2811.7695 1.1734+05 2.0643+05 2.1899+04 8478.0651 2681.5743 9825.0998 406.4449 3.7968 0.1114 0.0 0.0 0.0 0.0	1.5933+04 4.5631+04 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0
COMPONENTS: MASS FRAC HYDROGEN METHANE ETHANE ETHENE PROPANE PROPENE BENZENE TOLUENE P-XYL-01 1:3:5-01 WATER OXYGEN NITROGEN CARBO-01 TOTAL FLOW:	$\begin{array}{c} 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.6158\\ 0.3038\\ 8.0154-02\\ 2.6016-04\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ \end{array}$	$\begin{array}{c} 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 3.1772-11\\ 1.4568-07\\ 3.2640-04\\ 0.9997\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ \end{array}$	$\begin{array}{c} 0 \ . \ 0 \\ 0 \ . \ 0 \\ 0 \ . \ 0 \\ 0 \ . \ 0 \\ 0 \ . \ 0 \\ 0 \ . \ 0 \\ 0 \ . \ 0 \\ 0 \ . \ 0 \\ 0 \ . \ 0 \\ 1 \ . \ 0 \\ 0 \ . \ 0 \\ 0 \ . \ 0 \\ 0 \ . \ 0 \\ 0 \ . \ 0 \\ 0 \ . \ 0 \\ 0 \ . \ 0 \end{array}$	$\begin{array}{c} 7.6019-03\\ 0.3172\\ 0.5581\\ 5.9207-02\\ 2.2921-02\\ 7.2499-03\\ 2.6563-02\\ 1.0989-03\\ 1.0265-05\\ 3.0109-07\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ \end{array}$	0.2588 0.7412 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.
LBMOL/HR LB/HR CUFT/HR	1388.7340 1.1633+05 2234.7351	99.3452 1.1940+04 230.2501	1073.4269 1.9338+04 2327.6736		
STATE VARIABLES: TEMP F PRES PSIA VFRAC LFRAC SFRAC	136.3993 25.0000 0.0 1.0000 0.0	144.3420 3.7500 0.0 1.0000 0.0	380.2914 196.0000 4.2023-02 0.9580 0.0	181.1316 689.9998 1.0000 0.0 0.0	180.6861 200.0000 1.0000 0.0 0.0
ENTHALPY: BTU/LBMOL BTU/LB BTU/HR ENTROPY:	200.5672	-197.9592	-6455.7857	-2.7248+04 -1233.2512 -4.5615+08	-1346.3470
BTU/LBMOL-R BTU/LB-R DENSITY:	-63.9918 -0.7639	-127.9778 -1.0648	-29.6377 -1.6451		
LBMOL/CUFT LB/CUFT AVG MW	0.6214 52.0574 83.7702	0.4315 51.8576 120.1891	8.3079	2.4522	0.1658

S-219, S-220 S-220, S-221 S-221,

STREAM ID	S-219.	S-220	s-220,	S-221	S-221,
FROM :	PR-201	V-206	V-103	V-206	V-404
TO :	PR-201,	V-103	F-101	V-404	F-401-2
	,				
SUBSTREAM: MIXED					
PHASE:	VAPOR	VAPOR	VAPOR	VAPOR	VAPOR
COMPONENTS: LBMOL/HR					
HYDROGEN		1388.6102			
METHANE	2844.3385		499.7110	2344.6275	
ETHANE	0.0	0.0	0.0	0.0	0.0
ETHENE	0.0	0.0	0.0	0.0	0.0
PROPANE	0.0	0.0	0.0	0.0	0.0
PROPENE	0.0	0.0	0.0	0.0	0.0
BENZENE	0.0	0.0	0.0	0.0	0.0
TOLUENE	0.0	0.0	0.0	0.0	0.0
P-XYL-01	0.0	0.0	0.0	0.0	0.0
1:3:5-01	0.0	0.0	0.0	0.0	0.0
WATER	0.0	0.0	0.0	0.0	0.0
OXYGEN NITROGEN	0.0	0.0 0.0	2032.4726 7645.9682	0.0 0.0	9536.2938 3.5875+04
CARBO-01	0.0 0.0	0.0	0.0	0.0	0.0
COMPONENTS: MOLE FRAC	0.0	0.0	0.0	0.0	0.0
HYDROGEN	0.7354	0.7354	0.1201	0.7354	0.1201
METHANE	0.2646	0.2646		0.2646	
ETHANE	0.0	0.0	0.0	0.0	0.0
ETHENE	0.0	0.0	0.0	0.0	0.0
PROPANE	0.0	0.0	0.0	0.0	0.0
PROPENE	0.0	0.0	0.0	0.0	0.0
BENZENE	0.0	0.0	0.0	0.0	0.0
TOLUENE	0.0	0.0	0.0	0.0	0.0
P-XYL-01	0.0	0.0	0.0	0.0	0.0
1:3:5-01	0.0	0.0	0.0	0.0	0.0
WATER	0.0	0.0	0.0	0.0	0.0
OXYGEN	0.0	0.0	0.1757	0.0	0.1757
NITROGEN	0.0	0.0	0.6610	0.0	0.6610
CARBO-01	0.0	0.0	0.0	0.0	0.0
COMPONENTS: LB/HR					
HYDROGEN			2799.2715		
METHANE	4.5631+04	8016.7439			3.7614+04
ETHANE	0.0	0.0	0.0	0.0	0.0
ETHENE	0.0	0.0	0.0	0.0	0.0
PROPANE	0.0	0.0	0.0	0.0	0.0
PROPENE	0.0	0.0	0.0	0.0	0.0
BENZENE	0.0	0.0	0.0	0.0	0.0
TOLUENE P-XYL-01	0.0	0.0 0.0	0.0 0.0	0.0 0.0	0.0 0.0
1:3:5-01	0.0	0.0	0.0	0.0	0.0
WATER	0.0	0.0	0.0	0.0	0.0
OXYGEN	0.0	0.0	6.5037+04	0.0	3.0515+05
NITROGEN	0.0	0.0	2.1419+05	0.0	1.0050+06
CARBO-01	0.0	0.0	0.0	0.0	0.0
COMPONENTS: MASS FRAC					
HYDROGEN	0.2588	0.2588	9.6512-03	0.2588	9.6512-03
METHANE	0.7412	0.7412	2.7640-02	0.7412	2.7640-02
ETHANE	0.0	0.0	0.0	0.0	0.0
ETHENE	0.0	0.0	0.0	0.0	0.0

PROPANE	0.0	0.0	0.0	0.0	0.0
PROPENE	0.0	0.0	0.0	0.0	0.0
BENZENE	0.0	0.0	0.0	0.0	0.0
TOLUENE	0.0	0.0	0.0	0.0	0.0
P-XYL-01	0.0	0.0	0.0	0.0	0.0
1:3:5-01	0.0	0.0	0.0	0.0	0.0
WATER	0.0	0.0	0.0	0.0	0.0
OXYGEN	0.0	0.0	0.2242	0.0	0.2242
NITROGEN	0.0	0.0	0.7385	0.0	0.7385
CARBO-01	0.0	0.0	0.0	0.0	0.0
TOTAL FLOW:					
LBMOL/HR	1.0748+04	1888.3212	1.1567+04	8859.9405	5.4271+04
LB/HR		1.0816+04			1.3609+06
CUFT/HR	1.0918+05				
STATE VARIABLES:	1.0010.00	0.0220.01	1.,111.00	0.0001000	2.2203.07
TEMP F	181 1316	180.6861	94 7830	180.6861	94.7830
PRES PSIA	689.9998			200.0000	
VFRAC	1.0000	1.0000	1.0000	1.0000	1.0000
LFRAC	0.0	0.0	0.0	0.0	0.0
SFRAC	0.0	0.0	0.0	0.0	0.0
ENTHALPY:	0.0	0.0	0.0	0.0	0.0
	-7711.6697	7711 ((07	1001 2001	7711 ((07	1001 2001
BTU/LB				-1346.3470	
BTU/HR	-8.288/+0/	-1.4562+07	-1.4590+07	-6.8325+07	-6.8456+0/
ENTROPY:	10 0015		4 0 4 5 4		1 0 1 5 1
BTU/LBMOL-R				-7.8360	
BTU/LB-R	-1.8020	-1.3680	5.3641-02	-1.3680	5.3641-02
DENSITY:					
	9.8443-02				
LB/CUFT	0.5639				
AVG MW	5.7278	5.7278	25.0755	5.7278	25.0755
S-305 S-308 S-309 S-	31/ 9-317				
STREAM ID	S-305	S-308	S-309	S-314	S-317
FROM :	COL-301	RB-301	COL-301	COL-302	RB-302
то :	T-103	V-401-3	COL-302	H-201	V-401-3
SUBSTREAM: MIXED PHASE:	VAPOR	MIYED	LIQUID	LIQUID	MIVED
COMPONENTS: LBMOL/HR		MIXED	TIÕOID	TIÕOID	MIXED
		0 0	0 0	0 0	0 0
HYDROGEN	5.2075-05	0.0	0.0	0.0	0.0
METHANE	172.6175	0.0	0.0	0.0	0.0
ETHANE	238.5242	0.0	1.2669-29	0.0	0.0
ETHENE	22.0910	0.0	2.9603-33		0.0
PROPANE	16.4274	0.0	5.5478-23		0.0
PROPENE	3.4654	0.0	5.8969-30	0.0	0.0
BENZENE	108.5439	0.0	917.0410		0.0
TOLUENE	10.4562	0.0	383.6115		0.0
P-XYL-01	0.1014	0.0	87.8664		0.0
1:3:5-01	1.8392-07	0.0	99.5603	0.2518	0.0
WATER	0.0	646.0829	0.0	0.0	2074.3381
OXYGEN	0.0	0.0	0.0	0.0	0.0
NITROGEN	0.0	0.0	0.0	0.0	0.0
CARBO-01	0.0	0.0	0.0	0.0	0.0
COMPONENTS: MOLE FRA	С				

HYDROGEN METHANE	9.1005-08 0.3017	0.0	0.0	0.0	0.0
ETHANE	0.4168	0.0	8.5135-33	0.0	0.0
ETHENE	3.8605-02	0.0	0.0	0.0	0.0
PROPANE	2.8708-02	0.0	3.7282-26	0.0	0.0
PROPENE	6.0559-03	0.0	3.9628-33	0.0	0.0
BENZENE	0.1897	0.0	0.6163	0.6603	0.0
TOLUENE	1.8273-02	0.0	0.2578	0.2762	0.0
P-XYL-01	1.7728-04	0.0	5.9047-02		0.0
1:3:5-01	3.2141-10	0.0	6.6905-02	1.8132-04	0.0
WATER	0.0	1.0000	0.0	0.0	1.0000
OXYGEN	0.0	0.0	0.0	0.0	0.0
NITROGEN	0.0	0.0	0.0	0.0	0.0
CARBO-01	0.0	0.0	0.0	0.0	0.0
COMPONENTS: LB/HR	1 0 4 0 0 0 4	0 0	0.0	0 0	0.0
HYDROGEN	1.0498-04	0.0	0.0	0.0	0.0
METHANE	2769.2615	0.0	0.0	0.0	0.0
ETHANE ETHENE	7172.3375 619.7359	0.0	3.8095-28 8.3046-32	0.0	0.0
PROPANE	724.3933	0.0	2.4464-21	0.0	0.0
PROPENE	145.8246	0.0	2.4815-28	0.0	0.0
BENZENE	8478.7610	0.0	7.1633+04	7.1633+04	0.0
TOLUENE	963.4368	0.0	3.5346+04	3.5346+04	0.0
P-XYL-01	10.7700	0.0	9328.5507	9324.6534	0.0
1:3:5-01	2.2106-05	0.0	1.1967+04	30.2652	0.0
WATER	0.0	1.1639+04	0.0	0.0	3.7370+04
OXYGEN	0.0	0.0	0.0	0.0	0.0
NITROGEN	0.0	0.0	0.0	0.0	0.0
CARBO-01	0.0	0.0	0.0	0.0	0.0
COMPONENTS: MASS FRAC					
HYDROGEN	5.0266-09	0.0	0.0	0.0	0.0
METHANE	0.1326	0.0	0.0	0.0	0.0
ETHANE	0.3434	0.0	2.9698-33	0.0	0.0
ETHENE	2.9674-02	0.0	0.0	0.0	0.0
PROPANE	3.4686-02	0.0	1.9071-26	0.0	0.0
PROPENE	6.9824-03	0.0	1.9345-33	0.0	0.0
BENZENE	0.4060	0.0	0.5584	0.6158	0.0
TOLUENE	4.6132-02	0.0	0.2756	0.3038	0.0
P-XYL-01	5.1570-04	0.0	7.2723-02	8.0154-02	0.0
1:3:5-01 WATER	1.0585-09 0.0	0.0 1.0000	9.3289-02 0.0	2.6016-04	0.0 1.0000
OXYGEN	0.0	0.0	0.0	0.0	0.0
NITROGEN	0.0	0.0	0.0	0.0	0.0
CARBO-01	0.0	0.0	0.0	0.0	0.0
TOTAL FLOW:	0.0	0.0	0.0	0.0	0.0
LBMOL/HR	572.2272	646.0829	1488.0792	1388.7340	2074.3381
LB/HR		1.1639+04			3.7370+04
CUFT/HR	4.6730+04	1400.9990	2583.3950	2431.5464	4498.1002
STATE VARIABLES:					
TEMP F	193.0605	380.2914	200.4235	249.7144	380.2914
PRES PSIA	83.0000	196.0000	16.0000	35.0000	196.0000
VFRAC	1.0000	4.2023-02	0.0	0.0	4.2023-02
LFRAC	0.0	0.9580	1.0000	1.0000	0.9580
SFRAC	0.0	0.0	0.0	0.0	0.0
ENTHALPY:					
		-1.1630+05		2.1466+04	
BTU/LB	-449.1031	-6455.7857	193.5382	256.2534	-6455.7857

	-9.3793+06	-7.5141+07	2.4826+07	2.9811+07	-2.4125+08
ENTROPY:					
	-31.9389	-29.6377	-63.6657	-56.8798	-29.6377
BTU/LB-R	-0.8751	-1.6451	-0.7386	-0.6790	-1.6451
DENSITY:					
LBMOL/CUFT	1.2245-02	0.4612	0.5760	0.5711	0.4612
LB/CUFT	0.4469	8.3079			
AVG MW	36.4969	18.0153	86.2015	83.7702	18.0153
a 010 a 401 a					
5-318 5-401 5-	-402 S-403 S-404				
STREAM ID	S-318	S-401	S-402	S-403	S-404
FROM :	COL-302		P-401	V-401-3	F-401-2,
TO :	S-318 COL-302 H-202	P-401	V-401-3	F-401-2,	V-405
			0 0	0 0	
	DR: 0.0	0.0	0.0	0.0	1.3/55-08
SUBSTREAM: MIX		TTOUTD	TTOUTD	TTOUTD	
PHASE: COMPONENTS: LI		LIQUID	LIQUID	LIQUID	VAPOR
HYDROGEN	3MOL/HR	0 0	0 0	0 0	0 0
METHANE	0.0	0.0 0.0 0.0	0.0	0.0	0.0
ETHANE	0.0	0.0	0.0	0.0	0.0
ETHENE	0.0	0.0	0.0	0.0	0.0
PROPANE	0.0			0.0	
PROPENE				0.0	
	4.8566-09			0.0	
TOLUENE	1.8879-05			0.0	
P-XYL-01	3 6709-02	0.0		0.0	
1:3:5-01	3.6709-02 99.3085	0.0	0.0	0.0	0.0
WATER	0.0	5.7897+04			
OXYGEN	0.0			0.0	
NITROGEN		0.0	0.0	0.0	0.0
CARBO-01	0.0	0.0		0.0	
COMPONENTS: MO		0.0	0.0	0.0	0.0
HYDROGEN		0.0	0.0	0.0	0.0
METHANE		0.0			
ETHANE	0.0	0.0	0.0	0.0	0.0
ETHENE	0.0	0.0	0.0	0.0	0.0
PROPANE	0.0	0.0	0.0	0.0	0.0
PROPENE	0.0	0.0	0.0	0.0	0.0
BENZENE	4.8886-11	0.0	0.0	0.0	0.0
TOLUENE	1.9003-07	0.0	0.0	0.0	0.0
P-XYL-01	3.6951-04	0.0	0.0	0.0	0.0
1:3:5-01	0.9996	0.0	0.0	0.0	0.0
WATER	0.0	1.0000	1.0000	1.0000	1.0000
OXYGEN	0.0	0.0	0.0	0.0	0.0
NITROGEN	0.0	0.0	0.0	0.0	0.0
CARBO-01	0.0	0.0	0.0	0.0	0.0
COMPONENTS: LE	B/HR				
HYDROGEN	0.0	0.0	0.0	0.0	0.0
METHANE	0.0	0.0	0.0	0.0	0.0
ETHANE	0.0	0.0	0.0	0.0	0.0
ETHENE	0.0	0.0	0.0	0.0	0.0
PROPANE	0.0	0.0	0.0	0.0	0.0
PROPENE	0.0	0.0	0.0	0.0	0.0

BENZENE TOLUENE P-XYL-01 1:3:5-01 WATER OXYGEN NITROGEN CARBO-01 COMPONENTS: MASS FRAC	3.7936-07 1.7395-03 3.8973 1.1936+04 0.0 0.0 0.0 0.0 0.0	$\begin{array}{c} 0.0\\ 0.0\\ 0.0\\ 0.0\\ 1.0430+06\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ \end{array}$	$\begin{array}{c} 0.0\\ 0.0\\ 0.0\\ 0.0\\ 1.0430+06\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ \end{array}$	0.0 0.0 0.0 0.0 1.1114+06 0.0 0.0 0.0	$\begin{array}{c} 0.0\\ 0.0\\ 0.0\\ 0.0\\ 1.1114+06\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ \end{array}$		
HYDROGEN METHANE ETHANE ETHENE PROPANE PROPENE BENZENE TOLUENE P-XYL-01 1:3:5-01 WATER OXYGEN NITROGEN CARBO-01	$\begin{array}{c} 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 3.1772-11\\ 1.4568-07\\ 3.2640-04\\ 0.9997\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.$	0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 1.0000 0.0 0.	0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 1.0000 0.0 0.	0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 1.0000 0.0 0.	0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 1.0000 0.0 0.		
TOTAL FLOW: LBMOL/HR LB/HR CUFT/HR STATE VARIABLES:	99.3452 1.1940+04 259.3541	1.0430+06 1.6809+04	1.0430+06 1.6811+04	1.1114+06 1.8136+04	1.1114+06 1.1820+06		
TEMP F PRES PSIA VFRAC LFRAC SFRAC ENTHALPY:	323.7883 13.7500 0.0 1.0000 0.0	77.0000 14.5038 0.0 1.0000 0.0	77.2090 205.6316 0.0 1.0000 0.0	99.9141 196.0000 0.0 1.0000 0.0	465.2781 464.6959 1.0000 0.0 0.0		
BTU/LBMOL BTU/LB	-111.3439 -1.3295+06	-6820.4540 -7.1140+09	-6819.7415 -7.1132+09	-1.2246+05 -6797.3592 -7.5545+09	-5623.1311 -6.2494+09		
BTU/LB-R DENSITY: LBMOL/CUFT LB/CUFT	0.3830 46.0383	62.0507	3.4439 62.0437	3.4015 61.2788	-0.7430 5.2194-02 0.9403		
AVG MW 120.1891 18.0153 18.0153 18.0153 18.0153 S-404, S-405 S-406 S-407 S-408							
STREAM ID FROM : TO :	V-405	V-405	T-401	S-407 V-406-7 H-203,	V-406-7		
SUBSTREAM: MIXED PHASE: COMPONENTS: LBMOL/HR HYDROGEN METHANE		VAPOR 0.0 0.0			MIXED 0.0 0.0		

ETHANE		0.0	0.0	0.0	0.0	0.0
ETHENE		0.0	0.0	0.0	0.0	0.0
PROPANE		0.0	0.0	0.0	0.0	0.0
PROPENE		0.0	0.0	0.0	0.0	0.0
BENZENE		0.0	0.0	0.0	0.0	0.0
TOLUENE		0.0	0.0	0.0	0.0	0.0
P-XYL-01		0.0	0.0	0.0	0.0	0.0
1:3:5-01		0.0	0.0	0.0	0.0	0.0
WATER		3793.8480	5.7897+04	3793.8480	1073.4269	646.0829
OXYGEN		0.0	0.0	0.0	0.0	0.0
NITROGEN		0.0	0.0	0.0	0.0	0.0
CARBO-01		0.0	0.0	0.0	0.0	0.0
COMPONENTS:	MOTE EDAC	0.0	0.0	0.0	0.0	0.0
HYDROGEN	MOLE FRAC	0.0	0.0	0 0	0.0	0.0
				0.0		
METHANE		0.0	0.0	0.0	0.0	0.0
ETHANE		0.0	0.0	0.0	0.0	0.0
ETHENE		0.0	0.0	0.0	0.0	0.0
PROPANE		0.0	0.0	0.0	0.0	0.0
PROPENE		0.0	0.0	0.0	0.0	0.0
BENZENE		0.0	0.0	0.0	0.0	0.0
TOLUENE		0.0	0.0	0.0	0.0	0.0
P-XYL-01		0.0	0.0	0.0	0.0	0.0
1:3:5-01		0.0	0.0	0.0	0.0	0.0
WATER		1.0000	1.0000	1.0000	1.0000	1.0000
OXYGEN		0.0	0.0	0.0	0.0	0.0
NITROGEN		0.0	0.0	0.0	0.0	0.0
CARBO-01		0.0	0.0	0.0	0.0	0.0
COMPONENTS:	LB/HR					
HYDROGEN		0.0	0.0	0.0	0.0	0.0
METHANE		0.0	0.0	0.0	0.0	0.0
ETHANE		0.0	0.0	0.0	0.0	0.0
ETHENE		0.0	0.0	0.0	0.0	0.0
PROPANE		0.0	0.0	0.0	0.0	0.0
PROPENE		0.0	0.0	0.0	0.0	0.0
BENZENE		0.0	0.0	0.0	0.0	0.0
TOLUENE		0.0	0.0	0.0	0.0	0.0
P-XYL-01		0.0	0.0	0.0	0.0	0.0
1:3:5-01		0.0	0.0	0.0	0.0	0.0
WATER		6.8347+04	1.0430+06	6.8347+04	1.9338+04	1.1639+04
OXYGEN		0.0	0.0	0.0	0.0	0.0
NITROGEN		0.0	0.0	0.0	0.0	0.0
CARBO-01		0.0	0.0	0.0	0.0	0.0
COMPONENTS:	MASS FRAC					
HYDROGEN		0.0	0.0	0.0	0.0	0.0
METHANE		0.0	0.0	0.0	0.0	0.0
ETHANE		0.0	0.0	0.0	0.0	0.0
ETHENE		0.0	0.0	0.0	0.0	0.0
PROPANE		0.0	0.0	0.0	0.0	0.0
PROPENE		0.0	0.0	0.0	0.0	0.0
BENZENE		0.0	0.0	0.0	0.0	0.0
TOLUENE		0.0	0.0	0.0	0.0	0.0
P-XYL-01		0.0	0.0	0.0	0.0	0.0
1:3:5-01		0.0	0.0	0.0	0.0	0.0
WATER		1.0000	1.0000	1.0000	1.0000	1.0000
OXYGEN		0.0	0.0	0.0	0.0	0.0
NITROGEN		0.0	0.0	0.0	0.0	0.0
CARBO-01		0.0	0.0	0.0	0.0	0.0

TOTAL FLOW:			2702 0400	1072 4000	
LBMOL/HR	3793.8480			1073.4269	
LB/HR	6.8347+04			1.9338+04	
CUFT/HR	7.2687+04	1.1093+06	1.5112+05	4.2759+04	2.5736+04
STATE VARIABLES:					
TEMP F				384.4670	
PRES PSIA					
VFRAC		1.0000			
LFRAC				4.0290-02	
SFRAC	0.0	0.0	0.0	0.0	0.0
ENTHALPY:					
				-1.0232+05	
				-5679.7743	
	-3.8433+08	-5.8651+09	-3.8820+08	-1.0984+08	-6.6109+07
ENTROPY:					
BTU/LBMOL-R					
BTU/LB-R	-0.7430	-0.7430	-0.7263	-0.7263	-0.7263
DENSITY:					
	5.2194-02	5.2194-02	2.5104-02	2.5104-02	2.5104-02
LB/CUFT	0.9403	0.9403	0.4523	0.4523	0.4523
AVG MW	18.0153	18.0153	18.0153	18.0153	18.0153
S-409 S-410 S-411 S1	S2				
STREAM ID	S-409		S-411	S1	S2
FROM :	V-406-7		F-401-2		BOCKRATH
	PB-302	V-404		BOCKRATH	·
TO :	IND JUZ	V-404		DUCKRAID	L
10 :	ND 502	V-404		BUCKKAIR	L
TO : SUBSTREAM: MIXED	ICD 502	V-404		DUCKKAIT	L
	MIXED	VAPOR		LIQUID	-
SUBSTREAM: MIXED	MIXED				-
SUBSTREAM: MIXED PHASE:	MIXED				-
SUBSTREAM: MIXED PHASE: COMPONENTS: LBMOL/HR	MIXED	VAPOR	VAPOR	LIQUID	VAPOR
SUBSTREAM: MIXED PHASE: COMPONENTS: LBMOL/HR HYDROGEN	MIXED 0.0	VAPOR 0.0	VAPOR 0.0	LIQUID 0.0	VAPOR 0.0
SUBSTREAM: MIXED PHASE: COMPONENTS: LBMOL/HR HYDROGEN METHANE	MIXED 0.0 0.0	VAPOR 0.0 0.0	VAPOR 0.0 0.0	LIQUID 0.0 0.0	VAPOR 0.0 0.0
SUBSTREAM: MIXED PHASE: COMPONENTS: LBMOL/HR HYDROGEN METHANE ETHANE ETHENE	MIXED 0.0 0.0 0.0 0.0 0.0	VAPOR 0.0 0.0 0.0 0.0 0.0	VAPOR 0.0 0.0 0.0 0.0 0.0	LIQUID 0.0 0.0 0.0 0.0 0.0	VAPOR 0.0 0.0 0.0 0.0 0.0
SUBSTREAM: MIXED PHASE: COMPONENTS: LBMOL/HR HYDROGEN METHANE ETHANE	MIXED 0.0 0.0 0.0 0.0 0.0 0.0	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0	LIQUID 0.0 0.0 0.0 0.0 0.0 0.0	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0
SUBSTREAM: MIXED PHASE: COMPONENTS: LBMOL/HR HYDROGEN METHANE ETHANE ETHENE PROPANE	MIXED 0.0 0.0 0.0 0.0 0.0 0.0 0.0	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0	LIQUID 0.0 0.0 0.0 0.0 0.0 0.0 0.0	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0
SUBSTREAM: MIXED PHASE: COMPONENTS: LBMOL/HR HYDROGEN METHANE ETHANE ETHENE PROPANE PROPENE BENZENE	MIXED 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0	LIQUID 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0
SUBSTREAM: MIXED PHASE: COMPONENTS: LBMOL/HR HYDROGEN METHANE ETHANE ETHENE PROPANE PROPENE BENZENE TOLUENE	MIXED 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0	LIQUID 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.
SUBSTREAM: MIXED PHASE: COMPONENTS: LBMOL/HR HYDROGEN METHANE ETHANE ETHENE PROPANE PROPENE BENZENE TOLUENE P-XYL-01	MIXED 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	LIQUID 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.
SUBSTREAM: MIXED PHASE: COMPONENTS: LBMOL/HR HYDROGEN METHANE ETHANE ETHENE PROPANE PROPENE BENZENE TOLUENE P-XYL-01 1:3:5-01	MIXED 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	LIQUID 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.
SUBSTREAM: MIXED PHASE: COMPONENTS: LBMOL/HR HYDROGEN METHANE ETHANE ETHENE PROPANE PROPENE BENZENE TOLUENE P-XYL-01 1:3:5-01 WATER	MIXED 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	LIQUID 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.
SUBSTREAM: MIXED PHASE: COMPONENTS: LBMOL/HR HYDROGEN METHANE ETHANE ETHENE PROPANE PROPENE BENZENE TOLUENE P-XYL-01 1:3:5-01 WATER OXYGEN	MIXED 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	LIQUID 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.
SUBSTREAM: MIXED PHASE: COMPONENTS: LBMOL/HR HYDROGEN METHANE ETHANE ETHENE PROPANE PROPENE BENZENE TOLUENE P-XYL-01 1:3:5-01 WATER OXYGEN NITROGEN	MIXED 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	LIQUID 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.
SUBSTREAM: MIXED PHASE: COMPONENTS: LBMOL/HR HYDROGEN METHANE ETHANE ETHENE PROPANE PROPENE BENZENE TOLUENE P-XYL-01 1:3:5-01 WATER OXYGEN NITROGEN CARBO-01	MIXED 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	LIQUID 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.
SUBSTREAM: MIXED PHASE: COMPONENTS: LBMOL/HR HYDROGEN METHANE ETHANE ETHENE PROPANE PROPENE BENZENE TOLUENE P-XYL-01 1:3:5-01 WATER OXYGEN NITROGEN CARBO-01 COMPONENTS: MOLE FRA	MIXED 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	LIQUID 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.
SUBSTREAM: MIXED PHASE: COMPONENTS: LBMOL/HR HYDROGEN METHANE ETHANE ETHENE PROPANE PROPENE BENZENE TOLUENE P-XYL-01 1:3:5-01 WATER OXYGEN NITROGEN CARBO-01 COMPONENTS: MOLE FRA HYDROGEN	MIXED 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	LIQUID 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.
SUBSTREAM: MIXED PHASE: COMPONENTS: LBMOL/HR HYDROGEN METHANE ETHANE ETHENE PROPANE PROPENE BENZENE TOLUENE P-XYL-01 1:3:5-01 WATER OXYGEN NITROGEN CARBO-01 COMPONENTS: MOLE FRA HYDROGEN METHANE	MIXED 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	LIQUID 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.
SUBSTREAM: MIXED PHASE: COMPONENTS: LBMOL/HR HYDROGEN METHANE ETHANE ETHENE PROPANE PROPENE BENZENE TOLUENE P-XYL-01 1:3:5-01 WATER OXYGEN NITROGEN CARBO-01 COMPONENTS: MOLE FRA HYDROGEN METHANE ETHANE	MIXED 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	LIQUID 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.
SUBSTREAM: MIXED PHASE: COMPONENTS: LBMOL/HR HYDROGEN METHANE ETHANE ETHENE PROPANE PROPENE BENZENE TOLUENE P-XYL-01 1:3:5-01 WATER OXYGEN NITROGEN CARBO-01 COMPONENTS: MOLE FRA HYDROGEN METHANE ETHANE ETHENE	MIXED 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	LIQUID 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.
SUBSTREAM: MIXED PHASE: COMPONENTS: LBMOL/HR HYDROGEN METHANE ETHANE ETHENE PROPANE PROPENE BENZENE TOLUENE P-XYL-01 1:3:5-01 WATER OXYGEN NITROGEN CARBO-01 COMPONENTS: MOLE FRA HYDROGEN METHANE ETHANE ETHENE PROPANE	MIXED 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	LIQUID 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.
SUBSTREAM: MIXED PHASE: COMPONENTS: LBMOL/HR HYDROGEN METHANE ETHANE ETHENE PROPANE PROPENE BENZENE TOLUENE P-XYL-01 1:3:5-01 WATER OXYGEN NITROGEN CARBO-01 COMPONENTS: MOLE FRA HYDROGEN METHANE ETHANE ETHENE PROPANE PROPENE	MIXED 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	LIQUID 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.
SUBSTREAM: MIXED PHASE: COMPONENTS: LBMOL/HR HYDROGEN METHANE ETHANE ETHENE PROPANE PROPENE BENZENE TOLUENE P-XYL-01 1:3:5-01 WATER OXYGEN NITROGEN CARBO-01 COMPONENTS: MOLE FRA HYDROGEN METHANE ETHANE ETHENE PROPANE PROPENE BENZENE	MIXED 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	LIQUID 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.
SUBSTREAM: MIXED PHASE: COMPONENTS: LBMOL/HR HYDROGEN METHANE ETHANE ETHENE PROPANE PROPENE BENZENE TOLUENE P-XYL-01 1:3:5-01 WATER OXYGEN NITROGEN CARBO-01 COMPONENTS: MOLE FRA HYDROGEN METHANE ETHENE PROPANE PROPENE BENZENE TOLUENE	MIXED 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	LIQUID 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.
SUBSTREAM: MIXED PHASE: COMPONENTS: LBMOL/HR HYDROGEN METHANE ETHANE ETHENE PROPANE PROPENE BENZENE TOLUENE P-XYL-01 1:3:5-01 WATER OXYGEN NITROGEN CARBO-01 COMPONENTS: MOLE FRA HYDROGEN METHANE ETHANE ETHENE PROPANE PROPENE BENZENE	MIXED 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	LIQUID 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.	VAPOR 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.

WATER	1.0000	0.0	0.2196	1.0000	1.0000
OXYGEN	0.0	0.2100	3.1156-02	0.0	0.0
NITROGEN	0.0	0.7900	0.7032	0.0	0.0
CARBO-01	0.0	0.0	4.5961-02	0.0	0.0
COMPONENTS: LB/HR					
HYDROGEN	0.0	0.0	0.0	0.0	0.0
METHANE	0.0	0.0	0.0	0.0	0.0
ETHANE	0.0	0.0	0.0	0.0	0.0
ETHENE	0.0	0.0	0.0	0.0	0.0
PROPANE	0.0	0.0	0.0	0.0	0.0
PROPENE	0.0	0.0	0.0	0.0	0.0
BENZENE	0.0	0.0	0.0	0.0	0.0
TOLUENE	0.0	0.0	0.0	0.0	0.0
P-XYL-01	0.0	0.0	0.0	0.0	0.0
1:3:5-01	0.0	0.0	0.0	0.0	0.0
WATER	3.7370+04	0.0	2.0185+05	1.6500+05	1.6500+05
OXYGEN	0.0	3.0515+05			0.0
NITROGEN	0.0	1.0050+06			0.0
CARBO-01	0.0	0.0	1.0319+05		0.0
COMPONENTS: MASS FRAC		0.0	1.0019.00	0.0	0.0
HYDROGEN	0.0	0.0	0.0	0.0	0.0
METHANE	0.0	0.0	0.0	0.0	0.0
ETHANE	0.0	0.0	0.0	0.0	0.0
ETHENE	0.0	0.0	0.0	0.0	0.0
PROPANE	0.0	0.0	0.0	0.0	0.0
PROPENE	0.0	0.0	0.0	0.0	0.0
BENZENE	0.0	0.0	0.0	0.0	0.0
TOLUENE	0.0	0.0	0.0	0.0	0.0
P-XYL-01	0.0	0.0	0.0	0.0	0.0
1:3:5-01	0.0	0.0	0.0	0.0	0.0
WATER	1.0000	0.0	0.1483	1.0000	1.0000
OXYGEN	0.0	0.2329	3.7372-02	0.0	0.0
NITROGEN	0.0	0.7671	0.7385	0.0	0.0
CARBO-01	0.0	0.0	7.5824-02	0.0	0.0
TOTAL FLOW:					
LBMOL/HR	2074.3381	4.5411+04	5.1013+04	9158.8918	9158.8918
LB/HR	3.7370+04				
CUFT/HR	8.2629+04	1.8024+07		2659.1155	6.5896+05
STATE VARIABLES:	0.2029104	1.0024107	5.0550107	2009.1100	0.0000100
TEMP F	384.4670	77.0000	572.0000	77.0000	338.0404
PRES PSIA	206.0000	14.5038			114.6959
VFRAC	0.9597	1.0000	1.0000		1.0000
LFRAC	4.0290-02	0.0	0.0	1.0000	0.0
SFRAC	0.0	0.0	0.0	0.0	0.0
ENTHALPY:					
				-1.2287+05	
				-6820.4540	
	-2.1225+08	-1.3125+05	-1.3735+09	-1.1254+09	-9.3416+08
ENTROPY:					
BTU/LBMOL-R	-13.0838	1.0423	4.2301	-38.9666	-11.5948
BTU/LB-R	-0.7263	3.6128-02	0.1586	-2.1630	-0.6436
DENSITY:					
LBMOL/CUFT	2.5104-02	2.5195-03	1.3102-03	3.4443	1.3899-02
LB/CUFT	0.4523	7.2689-02	3.4951-02		0.2504
AVG MW	18.0153	28.8504		18.0153	18.0153
	-				-

A-5.3. Block Report

INLET ST				
	REAMS: S-108 TREAMS: S-201 S-204	TO STAGE	1	
OUTLET S	TREAMS: S-201	FROM STAGE	4	
	S-204	FROM STAGE	1	
	S-203	FROM STAGE	2	
		FROM STAGE		
PROPERTY	OPTION SET: N	RTL-RK RENON	(NRTL) / REDLICH-KW	IONG
	* * *	MASS AND ENERGY IN	BALANCE *** OUT	ספואייזער סינייע
TOTAL B	ALANCE			
MOLE	(LBMOL/HR)	29549.4	29549.4	0.00000
MASS	(LB/HR)	580601.	580601.	-0.200508E-15
ENTH	ALPY(BTU/HR)	-0.524333E+0	29549.4 580601. 09 -0.553627E+09	0.529134E-01
	* * *	CO2 EQUIVALENT S	SUMMARY ***	
	REAMS CO2E			
	STREAMS CO2E			
NET STR	EAMS CO2E PRODUC	TION 0.00000	LB/HR	
UTILITI	ES CO2E PRODUCTI	ON 0.00000	LB/HR	
TOTAL C	02E PRODUCTION	0.00000	LB/HR	
	*	** INPUT DATA	* * *	
ISENTROP	IC CENTRIFUGAL C	OMPRESSOR		
NUMBER O	OF STAGES			4
	COMPRES	SOR SPECIFICATIO	ONS PER STAGE	
STAGE	PRESSURE			
		MECHANICAL	ISENTROPIC	
NUMBER	RATIO	MECHANICAL EFFICIENCY	ISENTROPIC EFFICIENCY	
NUMBER	RATIO 2.465	EFFICIENCY	EFFICIENCY	
NUMBER 1	RATIO	EFFICIENCY 1.000	EFFICIENCY 0.8000	
NUMBER 1 2	RATIO 2.465	EFFICIENCY 1.000 1.000	EFFICIENCY 0.8000 0.8000	
NUMBER 1 2 3	RATIO 2.465 2.465	EFFICIENCY 1.000 1.000 1.000	EFFICIENCY 0.8000 0.8000 0.8000	
NUMBER 1 2 3	RATIO 2.465 2.465 2.465 2.465	EFFICIENCY 1.000 1.000 1.000	EFFICIENCY 0.8000 0.8000 0.8000 0.8000	
NUMBER 1 2 3	RATIO 2.465 2.465 2.465 2.465	EFFICIENCY 1.000 1.000 1.000 1.000 SPECIFICATIONS 1	EFFICIENCY 0.8000 0.8000 0.8000 0.8000	
NUMBER 1 2 3 4	RATIO 2.465 2.465 2.465 2.465 COOLER	EFFICIENCY 1.000 1.000 1.000 1.000 SPECIFICATIONS 1	EFFICIENCY 0.8000 0.8000 0.8000 0.8000	
NUMBER 1 2 3 4 STAGE	RATIO 2.465 2.465 2.465 2.465 COOLER PRESSURE DROP	EFFICIENCY 1.000 1.000 1.000 SPECIFICATIONS I TEMPERATURE	EFFICIENCY 0.8000 0.8000 0.8000 0.8000	
NUMBER 1 2 3 4 STAGE NUMBER	RATIO 2.465 2.465 2.465 2.465 COOLER PRESSURE DROP PSI	EFFICIENCY 1.000 1.000 1.000 SPECIFICATIONS I TEMPERATURE F	EFFICIENCY 0.8000 0.8000 0.8000 0.8000	
NUMBER 1 2 3 4 STAGE NUMBER 1	RATIO 2.465 2.465 2.465 2.465 COOLER PRESSURE DROP PSI 10.00	EFFICIENCY 1.000 1.000 1.000 SPECIFICATIONS I TEMPERATURE F 104.0	EFFICIENCY 0.8000 0.8000 0.8000 0.8000	
NUMBER 1 2 3 4 STAGE NUMBER 1 2	RATIO 2.465 2.465 2.465 2.465 COOLER PRESSURE DROP PSI 10.00 10.00	EFFICIENCY 1.000 1.000 1.000 SPECIFICATIONS I TEMPERATURE F 104.0 104.0	EFFICIENCY 0.8000 0.8000 0.8000 0.8000	
NUMBER 1 2 3 4 STAGE NUMBER 1 2 3	RATIO 2.465 2.465 2.465 2.465 COOLER PRESSURE DROP PSI 10.00 10.00 10.00 10.00 10.00	EFFICIENCY 1.000 1.000 1.000 SPECIFICATIONS I TEMPERATURE F 104.0 104.0 104.0 104.0	EFFICIENCY 0.8000 0.8000 0.8000 0.8000 PER STAGE	
NUMBER 1 2 3 4 STAGE NUMBER 1 2 3 4	RATIO 2.465 2.465 2.465 2.465 COOLER PRESSURE DROP PSI 10.00 10.00 10.00 10.00 10.00	EFFICIENCY 1.000 1.000 1.000 SPECIFICATIONS P TEMPERATURE F 104.0 104.0 104.0 104.0 104.0 104.0	EFFICIENCY 0.8000 0.8000 0.8000 PER STAGE	30.000
NUMBER 1 2 3 4 STAGE NUMBER 1 2 3 4 FINAL PR	RATIO 2.465 2.465 2.465 2.465 COOLER PRESSURE DROP PSI 10.00 10.00 10.00 10.00 10.00	EFFICIENCY 1.000 1.000 1.000 SPECIFICATIONS P TEMPERATURE F 104.0 104.0 104.0 104.0 104.0 104.0	EFFICIENCY 0.8000 0.8000 0.8000 PER STAGE ** 73 61,13	

*** PROFILE ***

COMPRESSOR PROFILE

STAGE NUMBER	OUTLET PRESSURE PSIA	PRESSURE RATIO	OUTLET TEMPERATURE F		
1 2 3 4	65.11 135.8 310.2 740.0	2.465 2.465 2.465 2.465	218.2 238.1 242.2 246.2		
STAGE NUMBER 1 2	INDICATED HORSEPOWER HP 0.1557E+05 0.1555E+05	BRAKE HORSEPOWER HP 0.1557E+05 0.1555E+05			
3 4	0.1521E+05 0.1481E+05	0.1521E+05 0.1481E+05			
STAGE NUMBER	HEAD DEVELOPED FT-LBF/LB	VOLUMETRIC FLOW CUFT/HR	ISENTROPIC EFFICIENCY		
1 2 3 4	0.4248E+05 0.4756E+05 0.5031E+05 0.5221E+05	0.6585E+07 0.3143E+07 0.1343E+07 0.5456E+06	0.8000 0.8000 0.8000 0.8000 0.8000		
		COOLER PROFII	LE		
STAGE NUMBER	OUTLET TEMPERATURE F	OUTLET PRESSURE PSIA	COOLING LOAD BTU/HR	VAPOR FRACTION	
1 2 3 4	104.0 104.0 104.0 104.0	55.11 125.8 300.2 730.0	4709E+08 4720E+08 4564E+08 4492E+08	0.9822 0.9838	
BLOCK: COI	-201 MODEL: F	LASH2			
OUTLET LI	APOR STREAM: QUID STREAM:		DN (NRTL) / RED	LICH-KWON	G
	* * *	MASS AND ENER IN	RGY BALANCE ** OU		RELATIVE DIFF.
MASS	LBMOL/HR) LB/HR)	27854.9 449349 -0.563270		9.	-0.576129E-06 -0.485841E-06 0.825102E-06
	*** REAMS CO2E STREAMS CO2E	0.41073	NT SUMMARY *** 32E+07 LB/HR 33E+07 LB/HR		

NET STREAMS CO2E UTILITIES CO2E P TOTAL CO2E PRODU	RODUCTION 0.	/	R	
TWO PHASE TP SPECIFIED TEMPERA PRESSURE DROP MAXIMUM NO. ITERA CONVERGENCE TOLER	FLASH TURE F PSI TIONS	' DATA ***	104.000 0.0 30 0.000	100000
OUTLET TEMPERATUR OUTLET PRESSURE HEAT DUTY VAPOR FRACTION		ILTS ***	104. 730. 0.605 0.986	00 78E-01
V-L PHASE EQUILIB	RIUM :			
COMP HYDROGEN	F(I) 0.333	X(I)	Y(I) 0.74795E-07	K(I) 0.33827
0.45226E+07	0.333	0.5	0.747958-07	0.33627
METHANE	0.36765	0.22583	0.36954	1.6363
ETHANE	0.25017	0.28241	0.24974	0.88434
ETHENE	0.28376E-01	0.26799E-01	0.28397E-01	1.0596
PROPANE	0.71497E-02	0.18840E-01	0.69941E-02	0.37124
PROPENE	0.23436E-02	0.42511E-02	0.23182E-02	0.54531
BENZENE	0.98057E-02	0.40285	0.45756E-02	0.11358E-
01 TOLUENE 02	0.65918E-03	0.38137E-01	0.16047E-03	0.42077E-
P-XYL-01	0.12166E-04	0.82867E-03	0.13010E-05	0.15700E-
02 1:3:5-01	0.77842E-06	0.56744E-04	0.33707E-07	0.59402E-
03				
BLOCK: COL-301 MO	DEL: RADFRAC			
	STAGE 1 STAGE 42 STAGE 42	RENON (NRTL) / I	REDLICH-KWONG	
	*** MASS AND	ENERGY BALANCE	* * *	
				ATIVE DIFF.
TOTAL BALANCE MOLE (LBMOL/HR MASS (LB/HR ENTHALPY (BTU/		0.31 20 159. 14 7939E+07 0.6	91590.	
	*** ^^^ דיזיז	ALENT SUMMARY *	* *	
FEED STREAMS CO2 PRODUCT STREAMS NET STREAMS CO2E	E 69 CO2E 69	231.5 LB/H 231.5 LB/H	R R	

		DN 0.00000 -0.854772E-0		
	* * *	**************************************	* * * *	
**** INPUI	PARAMETERS	* * * *		
INSIDE LOOF DESIGN SPEC MAXIMUM NO. MAXIMUM NO. MAXIMUM NUM FLASH TOLEF	PTION TION VION OPTION PARAMETER CALC CONVERGENCE IFICATION MET OF OUTSIDE I OF INSIDE LO IBER OF FLASH	METHOD THOD LOOP ITERATIONS DOP ITERATIONS ITERATIONS		42 STANDARD NO STANDARD NO BROYDEN NESTED 25 10 30 0.000100000 0.000100000
**** COL-S	PECS ****			
MASS REFLUX	C DIST / TOTAI C RATIO LATE TO FEED	-		1.00000 0.10288 0.14001
**** PROF	'ILES ****			
P-SPEC	STAGE	1 PRES, PSIA 2		83.0000 10.0000
	* *	**************************************	* *	
*** COMPON	ENT SPLIT FRA	ACTIONS ***		
		OUTLET STREAMS		
	S-305	S-309		
COMPONENT: HYDROGEN METHANE ETHANE ETHENE PROPANE PROPENE BENZENE TOLUENE P-XYL-01 1:3:5-01	1.0000 1.0000 1.0000 1.0000 1.0000 .10584 .26534E-01 .11532E-02 .18473E-08	0.0000 0.0000 0.0000 0.0000 0.0000 .89416 .97347 .99885 1.0000		

*** SUMMARY OF KEY RESULTS	* * *	
TOP STAGE TEMPERATURE	F	193.060
BOTTOM STAGE TEMPERATURE	F	200.424
TOP STAGE LIQUID FLOW	LBMOL/HR	28.0315
BOTTOM STAGE LIQUID FLOW	LBMOL/HR	1,488.08
TOP STAGE VAPOR FLOW	LBMOL/HR	572.227
BOILUP VAPOR FLOW	LBMOL/HR	653.807
MOLAR REFLUX RATIO		0.048987
MOLAR BOILUP RATIO		0.43936
CONDENSER DUTY (W/O SUBCOOL)	BTU/HR	435,077.
REBOILER DUTY	BTU/HR	9,032,280.

**** MAXIMUM FINAL RELATIVE ERRORS ****

DEW POINT	0.21263E-05	STAGE=	L
BUBBLE POINT	0.59084E-05	STAGE=	3
COMPONENT MASS BALANCE	0.16503E-05	STAGE= 1	5 COMP=METHANE
ENERGY BALANCE	0.25582E-06	STAGE= 1	1

**** PROFILES ****

NOTE REPORTED VALUES FOR STAGE LIQUID AND VAPOR RATES ARE THE FLOWS FROM THE STAGE INCLUDING ANY SIDE PRODUCT.

				ENTHA	ALPY		
STAGE	TEMPERATUR	E PRESSUE	RE	BTU/I	BMOL	HEAT DU	ГҮ
	F	PSIA	LI	QUID	VAPOR	BTU/HR	
1	193.06	83.000	175	70.	-16391.	.43508+0	06
2	96.784	10.000	161	66.	-15530.		
12	102.99	11.500	145	77.	-16255.		
13	103.91	11.650	140	53.	-16287.		
14	157.25	11.800	160	54.	14238.		
15	176.73	11.950	174	80.	33700.		
39	193.92	15.550	184	27.	36227.		
40	194.59	15.700	184	29.	36226.		
41	195.72	15.850	182	75.	36163.		
42	200.42	16.000	166	83.	35714.	.90323+0	7
STAGE	-			FEED RATE		PRODUC	
	LBMOL	/HR		LBMOL/HF	ર	LBMOI	L/HR
	LIQUID	VAPOR	LIQUID	VAPOR	MIXED	LIQUID	VAPOR
1		28.03					572.2
572.22	271						
2	19.69	600.3					
12	16.01	588.7					
13	17.78	588.2		462.8550)		
14	1947.	127.2	1597.4512				
15	2049.	458.8					
39	2141.	650.1					
40	2144.	653.2					
41	2142.	655.8					
42	1488.	653.8				1488.0792	

**** MASS FLOW PROFILES ****

STAGE FLOW RATE LB/HR			FEED RATE LB/HR				PRODUCT RATE LB/HR	
	LIQUID	VAPOR	LIQUID	VAPOR	MIXED	LIQUID	VAPOR	
1		2149.					0.2088E+05	
.2088	35+05							
2	1600.	0.2303E+05						
12	1324.	0.2225E+05						
13	1478.	0.2221E+05		.14236+05				
14	0.1643E+06	8127.	.13492+06					
15	0.1731E+06	0.3606E+05						
39	0.1806E+06	0.5208E+05						
40	0.1809E+06	0.5233E+05						
41	0.1811E+06	0.5260E+05						
42	0.1283E+06	0.5283E+05				.12827+00	5	

			**** MOLE-X	-PROFILE **	* * *	
	STAGE	HYDROGEN	METHANE	ETHANE	ETHENE	PROPANE
	1	0.88549E-14	0.11043E-01	0.58174E-01	0.44557E-02	0.70493E-
02						
	2	0.24465E-15	0.32870E-02	0.10857E-01	0.80200E-03	0.19598E-
02						
0.0	12	0.31759E-15	0.37860E-02	0.12317E-01	0.90227E-03	0.21132E-
02	1 0		0 202400 02		0 000000 00	0 011700
02	13	0.32645E-15	0.38248E-02	0.12417E-01	0.90883E-03	0.21170E-
UΖ	14	0.27575E-22	0.48442E-03	0.47760E-02	0.30069E-03	0.16300E-
02	14	0.2/3/36-22	0.40442E-05	0.4//006-02	0.30009E-03	0.10300E-
02	15	0.13229E-29	0.13120E-04	0.45753E-03	0.23410E-04	0.28654E-
03	10	0.102291 29	0.101201 01	0.10,001 00	0.201101 01	0.200011
	39	0.14121-204	0.54073E-44	0.12277E-28	0.50725E-32	0.10119E-
22						
	40	0.86934-212	0.12405E-45	0.11112E-29	0.37763E-33	0.16260E-
23						
	41	0.54660-219	0.28322E-47	0.10034E-30	0.28110E-34	0.25825E-
24						
	42	0.36824-226	0.61458E-49	0.85135E-32	0.19893E-35	0.37282E-
25						

0.5	STAGE 1	PROPENE 0.89414E-03	**** MOLE-: BENZENE 0.74353	X-PROFILE TOLUENE 0.17072	**** P-XYL-01 0.41364E-02	1:3:5-01 0.17122E-
07	2	0.22921E-03	0.71023	0.26044	0.12196E-01	0.11530E-
06	12	0.25717E-03	0.66520	0.24836	0.57667E-01	0.93962E-
02	13	0.25863E-03	0.65819	0.24296	0.54913E-01	0.24408E-
01	14	0.13475E-03	0.67133	0.22218	0.47124E-01	0.52040E-
01	15	0.14337E-04	0.68700	0.21726	0.45271E-01	0.49676E-
01 01	39	0.39239E-29	0.69517	0.21322	0.43806E-01	0.47801E-

Appendix A-5. ASPEN Data

01	40	0.39454E-30	0.69351	0.21482	0.43881E-01	0.47790E-
	41	0.39919E-31	0.68295	0.22339	0.45130E-01	0.48526E-
01	42	0.39628E-32	0.61626	0.25779	0.59047E-01	0.66905E-
υT						

			**** MOLE-Y-	-PROFILE	* * * *	
	STAGE	HYDROGEN	METHANE	ETHANE	ETHENE	PROPANE
	1	0.91005E-07	0.30166	0.41683	0.38605E-01	0.28708E-
01 01	2	0.86755E-07	0.28809	0.40009	0.37011E-01	0.27696E-
01	12	0.88452E-07	0.29330	0.40548	0.37548E-01	0.27962E-
01						
	13	0.88527E-07	0.29355	0.40582	0.37579E-01	0.27984E-
01						
0.1	14	0.32385E-14	0.62693E-01	0.19776	0.15503E-01	0.34794E-
01	15	0.11701E-21	0.20556E-02	0.20266E-01	L 0.12759E-02	0.69167E-
02	ТЭ	0.11/016-21	0.20550E-02	0.202006-01	L 0.12/J9E-02	0.0910/E-
02	39	0.76572-197	0.77489E-42	0.44758E-27	7 0.22496E-30	0.20723E-
21						
	40	0.46289-204	0.17725E-43	0.40226E-28	0.16623E-31	0.33085E-
22						
0.0	41	0.28420-211	0.40539E-45	0.36133E-29	0.12300E-32	0.52310E-
23	42	0.17907-218	0.91386E-47	0.30932E-30	0.87560E-34	0.76119E-
24	42	0.1/90/-218	0.91300E-4/	0.309328-30	U.0/300E-34	0./01198-
27						

			**** MOLE-Y	Y-PROFILE **	**	
	STAGE	PROPENE	BENZENE	TOLUENE	P-XYL-01	1:3:5-01
	1	0.60559E-02	0.18969	0.18273E-01	0.17728E-03	0.32141E-
09	2	0.58149E-02	0.21555	0.25392E-01	0.36217E-03	0.11060E-
08	12	0.58932E-02	0.20310	0.24813E-01	0.18042E-02	0.96641E-
04 03	13	0.58980E-02	0.20263	0.24537E-01	0.17424E-02	0.25581E-
03	14	0.47962E-02	0.60069	0.75464E-01	0.57916E-02	0.25025E-
02	15	0.57180E-03	0.84995	0.10668	0.84543E-02	0.38282E-
02	39	0.12916E-27	0.87611	0.11060	0.90366E-02	0.42565E-
02	40	0.12853E-28	0.87494	0.11169	0.90883E-02	0.42804E-
02	41	0.12808E-29	0.86880	0.11732	0.94668E-02	0.44165E-
02	42	0.12176E-30	0.83475	0.14510	0.13455E-01	0.66935E-

		**** K-VAI	JUES	* * * *	
STAGE	HYDROGEN	METHANE	ETHANE	ETHENE	PROPANE
1	0.10277E+08	27.316	7.1652	8.6641	4.0724
2	0.35461E+09	87.643	36.852	46.148	14.132

	12 13 14 15 39 40 41 42	0.27851E+09 0.27118E+09 0.11744E+09 0.88450E+08 0.54224E+08 0.53247E+08 0.51993E+08 0.48628E+08	77.470 76.748 129.42 156.68 143.30 142.88 143.14 148.70	32.921 32.683 41.407 44.296 36.455 36.200 36.012 36.333	41.614 41.348 51.557 54.504 44.349 44.019 43.758 44.015	13.232 13.219 21.346 24.138 20.479 20.347 20.255 20.417
			**** K-VALU	ES **	* * *	
	STAGE	PROPENE	BENZENE	TOLUENE	P-XYL-01	1:3:5-01
01	1	6.7728	0.25512	0.10703	0.42859E-01	0.18773E-
	2	25.369	0.30349	0.97498E-01	0.29696E-01	0.95922E-
02	12	22.916	0.30532	0.99907E-01	0.31287E-01	0.10285E-
01 01	13	22.805	0.30786	0.10099	0.31731E-01	0.10481E-
	14	35.594	0.89478	0.33965	0.12290	0.48088E-
01	15	39.882	1.2372	0.49103	0.18675	0.77064E-
01	39	32.917	1.2603	0.51870	0.20629	0.89046E-
01 01	40	32.578	1.2616	0.51991	0.20711	0.89566E-
01	41	32.086	1.2721	0.52517	0.20977	0.91013E-
ΟI	42	30.725	1.3546	0.56285	0.22786	0.10005
			**** MASS-X	-PROFILE **	* * *	
	STAGE	HYDROGEN	METHANE	ETHANE	ETHENE	PROPANE
02	1		0.23114E-02	0.22822E-01	0.16308E-02	0.40555E-
02	-	0.23289E-15	0.231146-02			
02	2	0.23289E-15 0.60685E-17	0.64887E-03	0.40170E-02	0.27685E-03	0.10634E-
02				0.40170E-02 0.44807E-02	0.27685E-03 0.30623E-03	0.10634E- 0.11274E-
02	2	0.60685E-17	0.64887E-03			
02 02	2 12	0.60685E-17 0.77455E-17	0.64887E-03 0.73481E-03	0.44807E-02	0.30623E-03	0.11274E-
02 02 03	2 12 13	0.60685E-17 0.77455E-17 0.79165E-17	0.64887E-03 0.73481E-03 0.73814E-03	0.44807E-02 0.44915E-02	0.30623E-03 0.30671E-03	0.11274E- 0.11230E-
02 02 03 03	2 12 13 14	0.60685E-17 0.77455E-17 0.79165E-17 0.65858E-24	0.64887E-03 0.73481E-03 0.73814E-03 0.92071E-04	0.44807E-02 0.44915E-02 0.17014E-02	0.30623E-03 0.30671E-03 0.99938E-04	0.11274E- 0.11230E- 0.85155E-
02 02 03 03 23	2 12 13 14 15	0.60685E-17 0.77455E-17 0.79165E-17 0.65858E-24 0.31565E-31	0.64887E-03 0.73481E-03 0.73814E-03 0.92071E-04 0.24912E-05	0.44807E-02 0.44915E-02 0.17014E-02 0.16284E-03	0.30623E-03 0.30671E-03 0.99938E-04 0.77733E-05	0.11274E- 0.11230E- 0.85155E- 0.14956E-
02 02 03 03 23 24	2 12 13 14 15 39	0.60685E-17 0.77455E-17 0.79165E-17 0.65858E-24 0.31565E-31 0.33751-206	0.64887E-03 0.73481E-03 0.73814E-03 0.92071E-04 0.24912E-05 0.10285E-44	0.44807E-02 0.44915E-02 0.17014E-02 0.16284E-03 0.43770E-29	0.30623E-03 0.30671E-03 0.99938E-04 0.77733E-05 0.16872E-32	0.11274E- 0.11230E- 0.85155E- 0.14956E- 0.52903E-
02 02 03 03 23 24 24	2 12 13 14 15 39 40	0.60685E-17 0.77455E-17 0.79165E-17 0.65858E-24 0.31565E-31 0.33751-206 0.20772-213	0.64887E-03 0.73481E-03 0.73814E-03 0.92071E-04 0.24912E-05 0.10285E-44 0.23588E-46	0.44807E-02 0.44915E-02 0.17014E-02 0.16284E-03 0.43770E-29 0.39604E-30	0.30623E-03 0.30671E-03 0.99938E-04 0.77733E-05 0.16872E-32 0.12557E-33	0.11274E- 0.11230E- 0.85155E- 0.14956E- 0.52903E- 0.84985E-
02 02 03 03 23 24	2 12 13 14 15 39 40 41	0.60685E-17 0.77455E-17 0.79165E-17 0.65858E-24 0.31565E-31 0.33751-206 0.20772-213 0.13032-220	0.64887E-03 0.73481E-03 0.73814E-03 0.92071E-04 0.24912E-05 0.10285E-44 0.23588E-46 0.53736E-48	0.44807E-02 0.44915E-02 0.17014E-02 0.16284E-03 0.43770E-29 0.39604E-30 0.35681E-31	0.30623E-03 0.30671E-03 0.99938E-04 0.77733E-05 0.16872E-32 0.12557E-33 0.93262E-35	0.11274E- 0.11230E- 0.85155E- 0.14956E- 0.52903E- 0.84985E- 0.13468E-

07	1	0.49089E-03	0.75774	0.20522	0.57294E-02	0.26849E-
	2	0.11868E-03	0.68266	0.29528	0.15932E-01	0.17053E-
06	12	0.13092E-03	0.62863	0.27686	0.74068E-01	0.13663E-
01	13	0.13092E-03	0.61849	0.26930	0.70132E-01	0.35291E-
01	14	0.67179E-04	0.62127	0.24254	0.59273E-01	0.74105E-
01	15	0.71409E-05	0.63517	0.23694	0.56888E-01	0.70670E-
01	39	0.19577E-29	0.64382	0.23293	0.55140E-01	0.68117E-
01	40	0.19679E-30	0.64209	0.23461	0.55218E-01	0.68083E-
01	41	0.19866E-31	0.63093	0.24343	0.56665E-01	0.68979E-
01	42	0.19345E-32	0.55844	0.27555	0.72723E-01	0.93289E-
01						
			**** MASS-Y-	PROFILE **	* *	
	STAGE 1	HYDROGEN 0.50266E-08	METHANE 0.13260	ETHANE 0.34343	ETHENE 0.29674E-01	PROPANE 0.34686E-
01	1					
01 01	1 2	0.50266E-08 0.45577E-08	0.13260 0.12045	0.34343 0.31352	0.29674E-01 0.27058E-01	0.34686E- 0.31828E-
	1 2 12	0.50266E-08 0.45577E-08 0.47189E-08	0.13260 0.12045 0.12453	0.34343 0.31352 0.32268	0.29674E-01 0.27058E-01 0.27876E-01	0.34686E- 0.31828E- 0.32631E-
01 01	1 2	0.50266E-08 0.45577E-08	0.13260 0.12045	0.34343 0.31352	0.29674E-01 0.27058E-01	0.34686E- 0.31828E-
01 01 01	1 2 12	0.50266E-08 0.45577E-08 0.47189E-08	0.13260 0.12045 0.12453	0.34343 0.31352 0.32268	0.29674E-01 0.27058E-01 0.27876E-01	0.34686E- 0.31828E- 0.32631E-
01 01 01 01	1 2 12 13	0.50266E-08 0.45577E-08 0.47189E-08 0.47270E-08	0.13260 0.12045 0.12453 0.12474	0.34343 0.31352 0.32268 0.32322	0.29674E-01 0.27058E-01 0.27876E-01 0.27924E-01	0.34686E- 0.31828E- 0.32631E- 0.32685E-
01 01 01 01 02	1 2 12 13 14	0.50266E-08 0.45577E-08 0.47189E-08 0.47270E-08 0.10214E-15	0.13260 0.12045 0.12453 0.12474 0.15736E-01	0.34343 0.31352 0.32268 0.32322 0.93039E-01	0.29674E-01 0.27058E-01 0.27876E-01 0.27924E-01 0.68045E-02	0.34686E- 0.31828E- 0.32631E- 0.32685E- 0.24005E-
01 01 01 01	1 2 12 13 14 15	0.50266E-08 0.45577E-08 0.47189E-08 0.47270E-08 0.10214E-15 0.30016E-23	0.13260 0.12045 0.12453 0.12474 0.15736E-01 0.41963E-03	0.34343 0.31352 0.32268 0.32322 0.93039E-01 0.77545E-02 0.16803E-27	0.29674E-01 0.27058E-01 0.27876E-01 0.27924E-01 0.68045E-02 0.45548E-03 0.78792E-31	0.34686E- 0.31828E- 0.32631E- 0.32685E- 0.24005E- 0.38811E- 0.11409E-
01 01 01 01 02	1 2 12 13 14 15 39 40	0.50266E-08 0.45577E-08 0.47189E-08 0.47270E-08 0.10214E-15 0.30016E-23 0.19271-198 0.11647-205	0.13260 0.12045 0.12453 0.12474 0.15736E-01 0.41963E-03 0.15520E-42 0.35493E-44	0.34343 0.31352 0.32268 0.32322 0.93039E-01 0.77545E-02 0.16803E-27 0.15098E-28	0.29674E-01 0.27058E-01 0.27876E-01 0.27924E-01 0.68045E-02 0.45548E-03 0.78792E-31 0.58208E-32	0.34686E- 0.31828E- 0.32631E- 0.32685E- 0.24005E- 0.38811E- 0.11409E- 0.18210E-
01 01 01 01 02 21	1 2 12 13 14 15 39	0.50266E-08 0.45577E-08 0.47189E-08 0.47270E-08 0.10214E-15 0.30016E-23 0.19271-198	0.13260 0.12045 0.12453 0.12474 0.15736E-01 0.41963E-03 0.15520E-42	0.34343 0.31352 0.32268 0.32322 0.93039E-01 0.77545E-02 0.16803E-27	0.29674E-01 0.27058E-01 0.27876E-01 0.27924E-01 0.68045E-02 0.45548E-03 0.78792E-31	0.34686E- 0.31828E- 0.32631E- 0.32685E- 0.24005E- 0.38811E- 0.11409E-
01 01 01 01 02 21 22	1 2 12 13 14 15 39 40	0.50266E-08 0.45577E-08 0.47189E-08 0.47270E-08 0.10214E-15 0.30016E-23 0.19271-198 0.11647-205	0.13260 0.12045 0.12453 0.12474 0.15736E-01 0.41963E-03 0.15520E-42 0.35493E-44	0.34343 0.31352 0.32268 0.32322 0.93039E-01 0.77545E-02 0.16803E-27 0.15098E-28	0.29674E-01 0.27058E-01 0.27876E-01 0.27924E-01 0.68045E-02 0.45548E-03 0.78792E-31 0.58208E-32	0.34686E- 0.31828E- 0.32631E- 0.32685E- 0.24005E- 0.38811E- 0.11409E- 0.18210E-

			**** MASS-	Y-PROFILE **	**	
	STAGE	PROPENE	BENZENE	TOLUENE	P-XYL-01	1:3:5-01
08	1	0.69824E-02	0.40598	0.46132E-01	0.51570E-03	0.10585E-
08	2	0.63769E-02	0.43880	0.60972E-01	0.10020E-02	0.34643E-
03	12	0.65629E-02	0.41985	0.60506E-01	0.50692E-02	0.30740E-
03	13	0.65740E-02	0.41925	0.59884E-01	0.48999E-02	0.81442E-
02	14	0.31578E-02	0.73414	0.10879	0.96203E-02	0.47061E-

	15	0.30618E-03	0.84483	0.1250	0 80	.11421E-01	0.58550E-	
02	39	0.67857E-28	0.85441	0.1272	22 0	.11978E-01	0.63872E-	
02	40	0.67512E-29	0.85308	0.1284	45 0	.12044E-01	0.64217E-	
02	41	0.67195E-30	0.84609	0.1347	77 0	.12530E-01	0.66180E-	
02	42	0.63404E-31	0.80692	0.1654	45 0	.17677E-01	0.99560E-	
02								
BL(оск: 	COL-302 MODEL:	RADFRAC					
	OUTLE	TS - S-314 S-318	STAGE 27	RENON (NRI	IL) / REDLI	CH-KWONG		
		* *	* MASS AND H	ENERGY BAI	LANCE ***			
	ΤΟΤΑΙ	BALANCE		IN	OUT	RELAT	FIVE DIFF.	
	TOTAL BALANCE MOLE (LBMOL/HR) 1488.08 -0.152797E-15 MASS(LB/HR) 128275. 128275. -0.159352E-09 ENTHALPY(BTU/HR) 0.248261E+08 -517737. 1.02085							
		* *	* CO2 EQUIV		1ARY ***			
		STREAMS CO2E CT STREAMS CO2E	0.0		LB/HR LB/HR			
		TREAMS CO2E PRO			LB/HR LB/HR			
	UTILI	TIES CO2E PRODU	CTION 0.0	00000	LB/HR			
	TOTAL	CO2E PRODUCTIO	N 0.0	00000	LB/HR			
			* * * * * * * * * * * * *	* * * * * * * * * *	· *			
			**** INPUT					
			* * * * * * * * * * * *	* * * * * * * * * *	* *			
	* * * *	INPUT PARAMETE	RS ****					
	NUMBE	R OF STAGES				27		
		ITHM OPTION				STANDARD		
		BER OPTION ALIZATION OPTIO	NT			NO STANDARD		
		ULIC PARAMETER				NO		
		E LOOP CONVERGE				BROYDEN		
	DESIGN SPECIFICATION METHOD					NESTED		
	MAXIMUM NO. OF OUTSIDE LOOP ITERATIONS MAXIMUM NO. OF INSIDE LOOP ITERATIONS					25		
		UM NO. OF INSID UM NUMBER OF FL				10 30		
		TOLERANCE				0.00010	00000	
	OUTSI	DE LOOP CONVERG	ENCE TOLERAN	CE		0.00010	0000	
	* * * *	COL-SPECS **	* *					
	MOLAR	VAPOR DIST / T	OTAL DIST			0.0		

	REFLUX RATIO) IO FEED RATIO			0.64165 0.90692
* * * *	PROFILES	* * * *			
P-SP	EC S	STAGE 1 PRE 2	S, PSIA		35.0000 10.0000
		**** RE	*********** SULTS **** *****		
* * *	COMPONENT SI	PLIT FRACTIONS	T STREAMS		
COMP	ONENT:	314 S-3	18		
TOLU P-XY	ENE 1.000 ENE 1.000 L-01 .9993 5-01 .2520	.4921	3E-07 8E-03		
***		KEY RESULTS	***		040 714
BOTT TOP BOTT TOP BOIL MOLA	STAGE TEMPERA OM STAGE TEM STAGE LIQUID OM STAGE LIQU STAGE VAPOR I UP VAPOR FLOU R REFLUX RAT R BOILUP RAT	PERATURE FLOW UID FLOW FLOW N IO	F F LBMOL/HR LBMOL/HR LBMOL/HR LBMOL/HR	1	249.714 323.788 891.088 99.3452 0.0 ,709.75 0.64165 17.2102
COND		W/O SUBCOOL)	BTU/HR BTU/HR		-0.253438+08 0.289994+08
* * * *	MAXIMUM FI	NAL RELATIVE E	RRORS ****		
BUBB COMP	POINT DLE POINT ONENT MASS BA GY BALANCE	ALANCE	0.12342E-04 0.79140E-05 0.20754E-06 0.93082E-05	STAGE= 12 STAGE= 6	COMP=1:3:5-01
* * * *	PROFILES	* * * *			
**NOT		VALUES FOR SI STAGE INCLUDI			ES ARE THE FLOWS
STAGE T F	EMPERATURE	PRESSURE PSIA	ENTHA BTU/L LIQUID	LPY BMOL VAPOR	HEAT DUTY BTU/HR
	49.71 87.19	35.000 10.000	21466. 12043.	36973. 32583.	25344+08

3	197.61	10.150	9384.1	30990.	
13	227.44	11.650	319.31	29091.	
14	230.05	11.800	-420.05	28954.	
15	233.51	11.950	-1188.0	28557.	
16	277.59	12.100	-8901.8	17347.	
25	322.22	13.450	-13469.	3477.7	
26	323.02	13.600	-13427.	3502.1	
27	323.79	13.750	-13382.	3531.6	.28999+08

STAGE FLOW RATE LBMOL/HR		FEED RATE LBMOL/HR			PRODUCT RATE LBMOL/HR		
	LIQUID	VAPOR	LIQUID	VAPOR	MIXED	LIQUID	VAPOR
1	2280.	0.000				1388.7339	
2	639.6	2280.					
3	622.9	2028.					
13	521.9	1920.					
14	516.3	1911.		79.3889			
15	1729.	1826.	1408.6902				
16	1733.	1630.					
25	1808.	1707.					
26	1809.	1708.					
27	99.35	1710.				99.3452	

**** MASS FLOW PROFILES ****

STAGE	FLOW RATE LB/HR		FEED RATE LB/HR		PRODUCT LB/HR	
1 0.19 2 0.57 3 0.57	QUID VAPOR 10E+06 0.000 59E+05 0.1910E+ 87E+05 0.1739E+ 43E+05 0.1702E+	-06 -06	VAPOR 1	MIXED	LIQUID .11633+06	VAPOR
15 0.17 16 0.19 25 0.21 26 0.21	26E+05 0.1698E+ 99E+06 0.1632E+ 64E+06 0.1679E+ 72E+06 0.2051E+ 74E+06 0.2053E+ 94E+05 0.2055E+	.12186+06 06 06	6414.8241		.11940+05	
27 0.11	94E105 0.2055E1	00			.11940105	
		**** MOLE-2	K-PROFILE	* * * *		
STAGE	BENZENE	TOLUENE	P-XYL-01	1:3	:5-01	
1	0.66034	0.27623	0.63244E-01	1 0.1813	32E-03	
2	0.37831	0.39510	0.22505	0.154	56E-02	
3	0.28423	0.38064	0.33096	0.417	50E-02	
13	0.19921	0.22658	0.21974	0.354	47	
14	0.19461	0.21768	0.19540	0.3923		
15	0.18345	0.21111	0.18076	0.424	68	
16	0.40730E-01			0.693		
25	0.21096E-08	0.21952E-05				
26	0.32174E-09	0.64889E-06	0.67845E-03	3 0.9993	32	
27	0.48886E-10	0.19003E-06	0.36951E-03	3 0.999	63	
		**** MOLE-1	-PROFILE	* * * *		
STAGE	BENZENE	TOLUENE	P-XYL-01	1:3	:5-01	
1	0.82629	0.15722	0.16461E-01	1 0.2334	47E-04	
2	0.66034	0.27623	0.63244E-01	1 0.1813	32E-03	
3	0.57141	0.31371	0.11427	0.611	52E-03	

DIF		BALANCE	IN	OUT	GENERATION	KELA'I'IVE
		* *	* MASS AND EI			D
	OUTLET	STREAM: STREAM: TY OPTION SET:		ENON (NRTL) /	REDLICH-KWONG	
		F-101 MODEL:				
	27	0.21942E-09				
		0.14491E-08				
		0.95341E-08				
	16	0.14756	0 20030	0.19759		
	14	0.45537	0.27351	0.122005	0.14901	
	13 14		0.27239	0.13829 0.12665	0.11438 0.13116	
	3	0.52055 0.47103	0.33711 0.27630	0.14148	0.85720E-03	
	2	0.61575	0.30383	0.80154E-01	0.26016E-03	
	1	0.79900	0.17933	0.21634E-01		
	STAGE	BENZENE	TOLUENE	P-XYL-01	1:3:5-01	
			**** MASS-1	Y-PROFILE	* * * *	
	27	0.31772E-10	0.14568E-06	0.32640E-03	0.99967	
	26	0.20911E-09	0.49748E-06	0.59933E-03	0.99940	
	25	0.13712E-08	0.16831E-05	0.10866E-02	0.99891	
	16	0.28059E-01	0.79179E-01	0.15745	0.73531	
	15	0.13777	0.18700	0.18450	0.49073	
	14	0.14736	0.19443	0.20111	0.45709	
	13	0.15200	0.20393	0.22789	0.41618	
	3	0.23896	0.37747	0.37817	0.54008E-02	
	2	0.32822	0.40434	0.26537	0.20633E-02	
	STAGE 1	BENZENE 0.61575	TOLUENE 0.30383	P-XYL-01 0.80154E-01	1:3:5-01 0.26016E-03	
				X-PROFILE	**** 1.2.5 01	
	20	6.9057	3.5550	1.8847	0.99967	
	26	6.9294	3.5628	1.8867	0.99940	
	16 25	4.7788 6.9522	2.2987 3.5700	1.1403 1.8884	0.56169 0.99890	
	15	2.8403	1.2568	0.56869	0.26093	
	14	2.7459	1.2067	0.54243	0.24715	
	13	2.6829	1.1730	0.52535	0.23794	
	3	2.0104	0.82418	0.34526	0.14647	
	2	1.7455	0.69915	0.28103	0.11732	
	1	1.2513	0.56917	0.26027	0.12876	
	STAGE	BENZENE	TOLUENE	P-XYL-01	1:3:5-01	
			**** K-VALU	TES	****	
	27	0.33759E-09	0.67555E-06	0.69641E-03	0.99930	
	26	0.22294E-08	0.23118E-05	0.12800E-02		
	25	0.14666E-07	0.78369E-05	0.23227E-02	0.99767	
	15 16	0.52106 0.19463	0.26533 0.22397	0.10280 0.19175	0.11081 0.38964	
	14	0.53438	0.26267	0.10599	0.96960E-01	
	13	0.53444	0.26577	0.11544	0.84346E-01	

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MOLE (LBMOL/HR)11566.810872.5MASS (LB/HR)290043.290043.ENTHALPY (BTU/HR)-0.145901E+08-0.224575E+09			-694.305	94.305 0.00000 0.00000 0.935032		
FEED STREAMS CO2E PRODUCT STREAMS CO2 NET STREAMS CO2E PI UTILITIES CO2E PRODUCT	2E RODUCTION DUCTION	200419. 21992.2 -178426. 0.00000	LB/HR LB/HR LB/HR			
STOICHIOMETRY MATRIX		PUT DATA	* * *			
REACTION # 1: SUBSTREAM MIXED METHANE -1.00		2.00	OXYGEN	-2.00	CARBO-01	1.00
REACTION # 2: SUBSTREAM MIXED ETHANE -2.00		6.00	OXYGEN	-7.00	CARBO-01	4.00
REACTION # 3: SUBSTREAM MIXED PROPANE -1.00		4.00	OXYGEN	-5.00	CARBO-01	3.00
REACTION # 4: SUBSTREAM MIXED HYDROGEN -2.00		2.00	OXYGEN	-1.00		
REACTION CONVERSION REACTION # 1: SUBSTREAM:MIXED REACTION # 2: SUBSTREAM:MIXED REACTION # 3: SUBSTREAM:MIXED REACTION # 4: SUBSTREAM:MIXED	KEY COMP: KEY COMP: KEY COMP:	METHANE ETHANE PROPANE	CONV FRAC: CONV FRAC:	1.000		
TWO PHASE TP FLASH SPECIFIED TEMPERATURE F SPECIFIED PRESSURE PSIA MAXIMUM NO. ITERATIONS CONVERGENCE TOLERANCE SIMULTANEOUS REACTIONS GENERATE COMBUSTION REACTIONS FOR FEED SPECIES				3	4.5038	
OUTLET TEMPERATURE OUTLET PRESSURE HEAT DUTY VAPOR FRACTION	*** R F PSIA BTU/HR	ESULTS	* * *	_	1346.0 14.504 0.20998E+09 1.0000	

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HEAT OF REACTIONS:

HEAT OF
REACTION
BTU/LBMOL
-0.34514E+06
-0.61431E+06
-0.87852E+06
-0.10400E+06

REACTION EXTENTS:

REACTION	REACTION
NUMBER	EXTENT
	LBMOL/HR
1	499.71
2	0.0000
3	0.0000
4	694.31

V-L PHASE EQUILIBRIUM :

MAXIMUM NO. ITERATIONS

COMP	F(I)	X(I)	Y(I)	K(I)
WATER	0.21964	0.92206	0.21964	532.39
OXYGEN	0.31156E-01	0.37057E-02	0.31156E-01	18791.
NITROGEN	0.70324	0.74236E-01	0.70324	21172.
CARBO-01	0.4596	61E-01 0.	26327E-05	0.45961E-01
00100.00				

0.39019E+08

BLOCK: F-101, MODEL: HEATER INLET STREAM: S-106 OUTLET STREAM: S-106, PROPERTY OPTION SET: NRTL-RK RENON (NRTL) / REDLICH-KWONG

*** MAS	S AND ENERGY BAL	ANCE ***	
	IN	OUT	RELATIVE DIFF.
TOTAL BALANCE			
MOLE(LBMOL/HR)	23597.0	23597.1	-0.447184E-05
MASS(LB/HR)	580599.	580601.	-0.298171E-05
ENTHALPY (BTU/HR)	-0.264258E+09	-0.254505E+09	-0.369042E-01
*** CO2	EQUIVALENT SUMM	IARY ***	
FEED STREAMS CO2E	0.302786E+07	LB/HR	
PRODUCT STREAMS CO2E	0.302790E+07	LB/HR	
NET STREAMS CO2E PRODUCTIO	N 42.0148	LB/HR	
UTILITIES CO2E PRODUCTION	0.0000	LB/HR	
TOTAL CO2E PRODUCTION	42.0148	LB/HR	
* * *	INPUT DATA ***		
TWO PHASE TP FLASH			
SPECIFIED TEMPERATURE	F		1,166.00
PRESSURE DROP	PSI		0.0

0.000100000

	*** RESU	LTS ***		
OUTLET TEMPERATURE OUTLET PRESSURE HEAT DUTY OUTLET VAPOR FRACT	F PSIA BTU/HR		116 43. 0.97 1.0	664 541E+07
V-L PHASE EQUILIBRI	IUM :			
COMP HYDROGEN	F(I) 0.5910	X(I) 9E-01	Y(I) 0.43786E-03	K(I) 0.59109E-01
ETHANE ETHENE PROPANE PROPENE BENZENE TOLUENE P-XYL-01 1:3:5-01 BLOCK: F-401-2 MODE	0.55935 0.34017E-01 0.14170E-01 0.28474E-02 0.99302E-02 0.63005E-03 0.58145E-05 0.39274E-07 EL: RSTOIC	0.44805 0.18588E-01 0.30878E-01 0.41948E-02 0.35301 0.50190E-01	1 0.31994 0.55935 1 0.34017E-01 1 0.14170E-01 2 0.28474E-02 0.99302E-02 1 0.63005E-03 3 0.58145E-05 5 0.39274E-07	1302.1 1908.7 478.63 707.98 29.340 13.093
INLET STREAM: OUTLET STREAM: OUTLET HEAT STREAM: PROPERTY OPTION SET	S-221, S-411 HEAT1	RENON (NRTL)	/ REDLICH-KWONG	
DIFF.	*** MASS AND IN		CE *** GENERATIC	N RELATIVE
TOTAL BALANCE MOLE(LBMOL/HR) MASS(LB/HR) ENTHALPY(BTU/HR) 14	0.136087E+0	7 0.136087E-	+07	0.00000 0.00000 -0.152372E-
FEED STREAMS CO2E PRODUCT STREAMS CO NET STREAMS CO2E H UTILITIES CO2E PRO TOTAL CO2E PRODUCT	D2E10PRODUCTION-83DDUCTION0.	0357. LH 3187. LH 7171. LH 00000 LH	3/HR 3/HR 3/HR 3/HR	
STOICHIOMETRY MATR:	*** INPUT IX:	DATA ***		
REACTION # 1: SUBSTREAM MIXED METHANE -1.00		.00 OXYGE1	N -2.00 CAR	BO-01 1.00

REACTION # 2: SUBSTREAM MIXED HYDROGEN -2.00	: WATER	2.00	OXYGEN	-1.00	
REACTION CONVERSION REACTION # 1:	SPECS: NUMB	ER= 2			
SUBSTREAM:MIXED REACTION # 2:	KEY COMP:M	ETHANE	CONV FRAC:	1.000	
SUBSTREAM:MIXED	KEY COMP:H	YDROGEN	CONV FRAC:	1.000	
TWO PHASE TP F SPECIFIED TEMPERATU SPECIFIED PRESSURE MAXIMUM NO. ITERATI CONVERGENCE TOLERAN SIMULTANEOUS REACTI	RE F PSIA ONS CE			5	572.000 14.5038 30 0.000100000
GENERATE COMBUSTION	REACTIONS F	OR FEED	SPECIES	1	10
OUTLET TEMPERATURE OUTLET PRESSURE HEAT DUTY VAPOR FRACTION	*** RE F PSIA BTU/HR	SULTS *	* *		572.00 14.504 -0.13050E+10 1.0000
HEAT OF REACTIONS:					
REACTION NUMBER	REFERENCE COMPONENT		HEAT OF REACTION BTU/LBMOL		
1 2	METHANE HYDROGEN		-0.34514E -0.10400E		
REACTION EXTENTS:					
REACTION NUMBER 1 2	REACTION EXTENT LBMOL/HR 2344.6 3257.7				

V-L PHASE EQUILIBRIUM :

COMP	F(I)	X(I)	Y(I)	K(I)
WATER	0.21964	0.97237	0.21964	67.110
OXYGEN	0.31156E-01	0.18842E-02	0.31156E-01	4912.9
NITROGEN	0.70324	0.25672E-01	0.70324	8138.6
CARBO-01	0.4596	61E-01 0.	72309E-04	0.45961E-01
0.18885E+06				

BLOCK: F-401-2, MODEL: HEATER

INLET STREAM: INLET HEAT STREAM: OUTLET STREAM: PROPERTY OPTION SET:	S-404	.) / REDLICH-KW	ONG
* * *	MASS AND ENERGY BALA IN	NCE *** OUT	RELATIVE DIFF.
TOTAL BALANCE MOLE(LBMOL/HR) MASS(LB/HR) ENTHALPY(BTU/HR)	61691.1 0.111138E+07 -0.624944E+10	61691.1 0.111138E+07	-0.353826E-15 -0.209497E-15
*** FEED STREAMS CO2E PRODUCT STREAMS CO2E NET STREAMS CO2E PRODUCT UTILITIES CO2E PRODUCTION	0.00000 0.00000 JCTION 0.00000 FION 0.00000	LB/HR LB/HR	
TWO PHASE PQ FLASH SPECIFIED PRESSURE DUTY FROM INLET HEAT ST MAXIMUM NO. ITERATIONS CONVERGENCE TOLERANCE	PSIA		464.696 0.130502+10 30 0.000100000
OUTLET TEMPERATURE F OUTLET PRESSURE F OUTLET VAPOR FRACTION			465.28 464.70 1.0000
V-L PHASE EQUILIBRIUM :	:		
	(I) X(I) .0000 1.0000	Y(I) 1.0000	K(I) 1.0492
BLOCK: H-101 MODEL: H	HEATX		
HOT SIDE:			
INLET STREAM: OUTLET STREAM: PROPERTY OPTION SET: COLD SIDE:	S-107 S-108 NRTL-RK RENON (NRTI	.) / REDLICH-KW	ONG
INLET STREAM: OUTLET STREAM: PROPERTY OPTION SET:	S-105 S-106 NRTL-RK RENON (NRTI	.) / REDLICH-KW	ONG
***	MASS AND ENERGY BALA IN		RELATIVE DIFF.
TOTAL BALANCE MOLE (LBMOL/HR)	53146.5	53146.5	0.00000

MASS(LB/HR) 0. ENTHALPY(BTU/HR) -0.			
FEED STREAMS CO2E 0	0.00000	LB/HR LB/HR LB/HR LB/HR	
*** INP	UT DATA ***		
FLASH SPECS FOR HOT SIDE: TWO PHASE FLASH MAXIMUM NO. ITERATIONS CONVERGENCE TOLERANCE			30 0.000100000
FLASH SPECS FOR COLD SIDE: TWO PHASE FLASH MAXIMUM NO. ITERATIONS CONVERGENCE TOLERANCE			30 0.000100000
FLOW DIRECTION AND SPECIFICATIO COUNTERCURRENT HEAT EXCHANG SPECIFIED COLD APPROACH TEMP SPECIFIED VALUE TEMPERATURE TOLERANCE LMTD CORRECTION FACTOR			18.0000 0.01800 1.00000
PRESSURE SPECIFICATION: HOT SIDE PRESSURE DROP COLD SIDE PRESSURE DROP	PSI PSI		10.0000 10.0000
HEAT TRANSFER COEFFICIENT SPECI HOT LIQUID COLD LIQUID HOT 2-PHASE COLD LIQUID HOT VAPOR COLD LIQUID HOT LIQUID COLD 2-PHASE HOT 2-PHASE COLD 2-PHASE HOT VAPOR COLD 2-PHASE HOT LIQUID COLD VAPOR HOT 2-PHASE COLD VAPOR HOT 2-PHASE COLD VAPOR	BTU/HR-SQF" BTU/HR-SQF" BTU/HR-SQF" BTU/HR-SQF"	I-R 1 I-R 1	L49.6937 L49.6937 L49.6937 L49.6937 L49.6937 L49.6937 L49.6937 L49.6937
*** OVERA	LL RESULTS	* * *	
STREAMS:			
S-107> T= 1.1660D+03 1.0292D+02 P= 3.6412D+01 2.6412D+01 V= 1.0000D+00	НОТ	 	> S-108 T= P= V= 9.9275D-
01		I	

S-106 < T= 1.1480D+03 1.7084D+01	COLD	< S-105 	T= -
P= 4.3664D+01		I	P=
5.3664D+01			0 0 0 0 5 5
V= 1.0000D+00 01		V=	9.8735D-
DUTY AND AREA: CALCULATED HEAT DUTY	BTU/HR	470058043.9075	
CALCULATED (REQUIRED) AREA		60388.0036	
ACTUAL EXCHANGER AREA	SQFT	60388.0036	
PER CENT OVER-DESIGN		0.0000	
HEAT TRANSFER COEFFICIENT:			
AVERAGE COEFFICIENT (DIRTY)	BTU/HR-SQFT-R	149.6937	
UA (DIRTY)	BTU/HR-R	9039701.0442	
LOG-MEAN TEMPERATURE DIFFERENCE:			
LMTD CORRECTION FACTOR		1.0000	
LMTD (CORRECTED)	F	51.9993	
NUMBER OF SHELLS IN SERIES		1	
PRESSURE DROP:			
	PSI	10.0000	
COLDSIDE, TOTAL	PSI	10.0000	

*** ZONE RESULTS ***

TEMPERATURE LEAVING EACH ZONE:

			НОТ			
HOT IN OUT	 VAP		VAP		COND	 HOT
> 1166.0	 	 152.1 	119.9	 9 		> 102.9
COLDOUT <	VAP		BOIL		BOIL	COLDIN <
1148.0		41.9 	12.3	3		-17.1

COLD

ZONE HEAT TRANSFER AND AREA:

ZONE	HEAT DUTY	AREA	LMTD	AVERAGE U	UA
	BTU/HR	SQFT	F	BTU/HR-SQFT-F	R BTU/HR-R
1	451409557.974	59266.5658	50.8812	149.6937	8871828.9247
2	9967205.951	611.3995	108.9043	149.6937	91522.6296
3	8681279.983	510.0382	113.7045	149.6937	76349.4899
HEATX (COLD-TQCU H-101	TQCURV INLET			

PRESSURE PROFILE:	CONSTANT2				
PRESSURE DROP:	-10.0000	PSI			
PROPERTY OPTION SET:	NRTL-RK	RENON	(NRTL)	/	REDLICH-KWONG

 ! DUTY !	 ! pres	! TEMP !	· · · · · · · · · · · · · · · · · · ·				
!	! !	! !	! ! ! !				
! BTU/HR	PSIA	! F !	· · · · · · · · · · · · · · · · · · · ·				
! 0.0 ! 2.2384+07 ! 4.4767+07 ! 6.7151+07 ! 8.9535+07	53.6640 53.6640	! 1106.3022 ! ! 1063.9048 ! ! 1020.8060 !	! 1.0000 ! ! 1.0000 ! ! 1.0000 !				
!	53.6640 53.6640 53.6640 53.6640 53.6640	! 932.2784 ! ! 886.7175 ! ! 840.1877 ! ! 792.5954 !	1.0000 ! 1.0000 ! 1.0000 ! 1.0000 ! 1.0000 ! 1.0000 ! 1.0000 !				
! 2.2384+08 ! ! 2.4622+08 ! ! 2.6860+08 ! ! 2.9099+08 ! ! 3.1337+08 !	53.6640 53.6640 53.6640	! 642.2347 ! ! 589.0530 !	! 1.0000 ! ! 1.0000 ! ! 1.0000 !				
! 3.3576+08 ! ! 3.5814+08 ! ! 3.8052+08 ! ! 4.0291+08 ! ! 4.2529+08 !	53.6640 53.6640 53.6640	! 353.9281 ! ! 287.2685 ! ! 216.0633 !	! 1.0000 ! ! 1.0000 ! ! 1.0000 !				
! 4.4767+08 ! ! 4.5136+08 ! ! 4.5783+08 ! ! 4.7006+08 !	53.6640 53.6640	! 42.1066 ! 23.3187	! DEW>1.0000 ! ! 0.9947 !				
HEATX HOT-TQCUR H-101 TQCURV INLET							
PRESSURE PRO PRESSURE DRO PROPERTY OPT	DP: 0	CONSTANT2 0.0 PSI IRTL-RK RENON		DLICH-KWON			

!	DUTY	!	PRES	!	TEMP	!	VFRAC	!
!		!		!		!		!
!		!		!		!		!
!		!		!		!		!
!	BTU/HR	!	PSIA	!	F	!		!
!		!		!		!		!
! =	===========	:!=:	===========	:!=:	===========	=!=		=!
!	0.0	!	36.4121	!	1166.0000	!	1.0000	!

! 2.2384+07 ! 36.4121 ! 4.4767+07 ! 36.4121 ! 6.7151+07 ! 36.4121 ! 8.9535+07 ! 36.4121 !	! 1084.0343 ! ! 1042.1961 ! ! 999.7354 !	1.0000 ! 1.0000 ! 1.0000 !	
! 1.1192+08 ! 36.4121 ! 1.3430+08 ! 36.4121 ! 1.5669+08 ! 36.4121 ! 1.7907+08 ! 36.4121 ! 2.0145+08 ! 36.4121 !	956.6074	1.0000 !	
! 2.2384+08 ! 36.4121 ! 2.4622+08 ! 36.4121 ! 2.6860+08 ! 36.4121 ! 2.9099+08 ! 36.4121	! 728.9590 ! ! 680.5035 ! ! 630.8374 ! ! 579.8209 !	1.0000 ! 1.0000 ! 1.0000 ! 1.0000 !	
! 3.1337+08 ! 36.4121 !	! 416.7732 ! ! 358.2202 ! ! 296.9623 !	1.0000 ! 1.0000 ! 1.0000 !	
! 4.4767+08 ! 36.4121 ! 4.5136+08 ! 36.4121 ! 4.5783+08 ! 36.4121 ! 4.7006+08 ! 36.4121	! 164.2910 !	1.0000 !	
BLOCK: H-201 MODEL: HE HOT SIDE:			
INLET STREAM: OUTLET STREAM: PROPERTY OPTION SET: N COLD SIDE:	S-314 S-215 RTL-RK RENON	(NRTL) / REDLICH-	KWONG
INLET STREAM: OUTLET STREAM: PROPERTY OPTION SET: N	S-212	(NRTL) / REDLICH-	KWONG
***	MASS AND ENERGY IN	BALANCE *** OUT	RELATIVE DIFF.
TOTAL BALANCE MOLE (LBMOL/HR) MASS (LB/HR) ENTHALPY (BTU/HR)	28877.9 547777. -0.529803E+	547777.	0.00000 0.00000
FEED STREAMS CO2E PRODUCT STREAMS CO2E NET STREAMS CO2E PRODUC UTILITIES CO2E PRODUCTI	0.407420E TION 0.00000	+07 LB/HR +07 LB/HR LB/HR LB/HR	
*	** INPUT DATA	* * *	

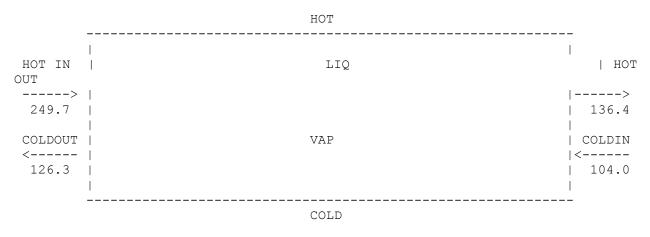
FLASH SPECS FOR HOT SIDE: TWO PHASE FLASH		
MAXIMUM NO. ITERATIONS		30
CONVERGENCE TOLERANCE		0.000100000
		0.000100000
FLASH SPECS FOR COLD SIDE:		
TWO PHASE FLASH		
MAXIMUM NO. ITERATIONS		30
CONVERGENCE TOLERANCE		0.000100000
FLOW DIRECTION AND SPECIFICATI	ION:	
COUNTERCURRENT HEAT EXCHAN	IGER	
SPECIFIED HOT APPROACH TEMP		
SPECIFIED VALUE	F	32.4000
TEMPERATURE TOLERANCE	F	0.01800
LMTD CORRECTION FACTOR		1.00000
PRESSURE SPECIFICATION:		
HOT SIDE PRESSURE DROP	PSI	10.0000
COLD SIDE PRESSURE DROP	PSI	10.0000
HEAT TRANSFER COEFFICIENT SPEC		
HOT LIQUID COLD LIQUID		149.6937
HOT 2-PHASE COLD LIQUID	-	
	BTU/HR-SQFT-R	
HOT LIQUID COLD 2-PHASE		
HOT 2-PHASE COLD 2-PHASE	- / - ~	
HOT VAPOR COLD 2-PHASE	BTU/HR-SQFT-R	
HOT LIQUID COLD VAPOR	BTU/HR-SQFT-R	149.6937
HOT 2-PHASE COLD VAPOR	BTU/HR-SQFT-R	149.6937
HOT VAPOR COLD VAPOR	BTU/HR-SQFT-R	149.6937
*** OVEF	RALL RESULTS ***	
STREAMS:		

S-314>		НОТ	 > S	-215	
T= 2.4971D+02					T=
1.3640D+02 P= 3.5000D+01 2.5000D+01	1				P=
V= 0.0000D+00 0.0000D+00	1				V=
a. 010		0.01 5		011	
S-212 < T= 1.2634D+02		COLD	< S	-211	T=
1.0400D+02	I			I	1-
P= 7.2000D+02					P=
7.3000D+02					
V= 1.0000D+00					V=
1.0000D+00					
DUTY AND AREA:					
CALCULATED HEAD	F DUTY	BTU/HR	6478222.00	23	

CALCULATED (REQUIRED) AREA ACTUAL EXCHANGER AREA PER CENT OVER-DESIGN	SQFT SQFT	636.0483 636.0483 0.0000
HEAT TRANSFER COEFFICIENT:		
AVERAGE COEFFICIENT (DIRTY)	BTU/HR-SQFT-R	149.6937
UA (DIRTY)	BTU/HR-R	95212.3982
LOG-MEAN TEMPERATURE DIFFERENCE:		
LMTD CORRECTION FACTOR		1.0000
LMTD (CORRECTED)	F	68.0397
NUMBER OF SHELLS IN SERIES		1
PRESSURE DROP:		
HOTSIDE, TOTAL	PSI	10.0000
COLDSIDE, TOTAL	PSI	10.0000

*** ZONE RESULTS ***

TEMPERATURE LEAVING EACH ZONE:



ZONE HEAT TRANSFER AND AREA:

zone 1	HEAT DUTY BTU/HR 6478222.002	SQFT	LMTD F 68.0397	AVERAGE U BTU/HR-SQFT-R 149.6937	
НЕАТХ СО	LD-TQCU H-201	TQCURV INLE	ΣT		
PRESSU			PSI RENON (NRTL) /	REDLICH-KWONG	
! DUTY ! !	 ! PRES ! !	! TEMP ! ! !	! VFRAC ! !	 ! ! !	
! BTU/H !	R ! PSIA !	! F !	- ! !	! !	
! 0.	0 ! 729.99	98 ! 126.67	736 ! 1.00	: 00 !	

! !	3.0849+05 6.1697+05 9.2546+05 1.2339+06	! ! !	729.9998 729.9998 729.9998 729.9998 729.9998	! ! !	125.5962 124.5186 123.4407 122.3625	! ! !	1.0000 ! 1.0000 ! 1.0000 ! 1.0000 !
! ! !	1.5424+06 1.8509+06 2.1594+06 2.4679+06 2.7764+06	! ! ! !	729.9998 729.9998 729.9998 729.9998 729.9998 729.9998	! ! ! !	121.2841 120.2054 119.1265 118.0474 116.9680	! ! ! !	1.0000 ! 1.0000 ! 1.0000 ! 1.0000 ! 1.0000 !
! ! !	3.0849+06 3.3934+06 3.7018+06 4.0103+06 4.3188+06	! ! ! !	729.9998 729.9998 729.9998 729.9998 729.9998 729.9998	! ! ! !	115.8884 114.8086 113.7286 112.6484 111.5680	! ! ! !	1.0000 ! 1.0000 ! 1.0000 ! 1.0000 ! 1.0000 !
! ! ! !	4.6273+06 4.9358+06 5.2443+06 5.5528+06 5.8612+06 6.1697+06 6.4782+06	! ! ! ! ! !	729.9998 729.9998 729.9998 729.9998 729.9998 729.9998 729.9998 729.9998	! ! ! ! ! !	110.4874 109.4066 108.3257 107.2445 106.1633 105.0819 103.9998	! ! ! !	1.0000 ! 1.0000 ! 1.0000 ! 1.0000 ! 1.0000 ! 1.0000 ! 1.0000 !

HEATX HOT-TQCUR H-201 TQCURV INLET

PRESSURE	PROFILE:	CONSTANT2				
PRESSURE	DROP:	0.0	PSI			
PROPERTY	OPTION SET:	NRTL-RK	RENON	(NRTL)	/	REDLICH-KWONG

!	DUTY	!	PRES	!	TEMP	!	VFRAC	!
1		1		!		1		1
1		1		!		1		1
1		1		!		1		1
1	BTU/HR	1	PSIA	!	F	1		1
!		!		!		!		!
! =		= ! =		! =		=!=		=!
!	0.0	!	35.0000	!	249.7144	1	0.0	!
!	3.0849+05	!	35.0000	!	244.5803	!	0.0	!
!	6.1697+05	!	35.0000	!	239.4215	!	0.0	!
!	9.2546+05	!	35.0000	!	234.2379	!	0.0	!
!	1.2339+06	!	35.0000	!	229.0292	!	0.0	!
! -		-+-		+ -		-+-		-!
!	1.5424+06	!	35.0000	!	223.7951	!	0.0	!
!	1.8509+06	!	35.0000	!	218.5355	!	0.0	!
!	2.1594+06	!	35.0000	!	213.2499	!	0.0	!
!	2.4679+06	!	35.0000	!	207.9382	!	0.0	!
!	2.7764+06	!	35.0000	!	202.6000	!	0.0	!
! -		-+-		+ -		-+-		-!
!	3.0849+06	!	35.0000	!	197.2351	!	0.0	!
!	3.3934+06	!	35.0000	!	191.8430	!	0.0	!
!	3.7018+06	!	35.0000	!	186.4235	!	0.0	!
!	4.0103+06	!	35.0000	!	180.9762	!	0.0	!

! 4.3188+06 ! 35.0000 ! 175.5008 ! 0.0 ! 4.6273+06 !35.0000 !169.9968 !0.0 !4.9358+06 !35.0000 !164.4640 !0.0 ! ! 1 ! 5.2443+06 ! 35.0000 ! 158.9019 ! 0.0 ! ! 5.5528+06 ! 35.0000 ! 153.3102 ! ! 5.8612+06 ! 35.0000 ! 147.6886 ! 0.0 ! 0.0 1 !-----+-----! ! 6.1697+06 ! 35.0000 ! 142.0365 ! 0.0 ! ! 6.4782+06 ! 35.0000 ! 136.3537 ! 0.0 ! BLOCK: H-202 MODEL: HEATX _____ HOT SIDE: _____ INLET STREAM: S-318 OUTLET STREAM: S-216 PROPERTY OPTION SET: NRTL-RK RENON (NRTL) / REDLICH-KWONG COLD SIDE: _____ S-212 INLET STREAM: OUTLET STREAM: S-213 PROPERTY OPTION SET: NRTL-RK RENON (NRTL) / REDLICH-KWONG *** MASS AND ENERGY BALANCE *** OUT IN RELATIVE DIFF. TOTAL BALANCE MOLE (LBMOL/HR)27588.527588.50.00000MASS (LB/HR)443383.443383.0.00000 ENTHALPY(BTU/HR) -0.554466E+09 -0.554466E+09 0.00000 *** CO2 EQUIVALENT SUMMARY *** FEED STREAMS CO2E0.407420E+07LB/HRPRODUCT STREAMS CO2E0.407420E+07LB/HR NET STREAMS CO2E PRODUCTION 0.00000 LB/HR 0.00000 0.00000 UTILITIES CO2E PRODUCTION LB/HR TOTAL CO2E PRODUCTION LB/HR *** INPUT DATA *** FLASH SPECS FOR HOT SIDE: TWO PHASE FLASH MAXIMUM NO. ITERATIONS 30 CONVERGENCE TOLERANCE 0.000100000 FLASH SPECS FOR COLD SIDE: TWO PHASE FLASH MAXIMUM NO. ITERATIONS 30 CONVERGENCE TOLERANCE 0.000100000 FLOW DIRECTION AND SPECIFICATION: COUNTERCURRENT HEAT EXCHANGER SPECIFIED HOT APPROACH TEMP SPECIFIED VALUE F 18.0000 TEMPERATURE TOLERANCE F 0.01800 LMTD CORRECTION FACTOR 1.00000

PRESSURE SPECIFICAT	ION:		
HOT SIDE PRESSUR	E DROP	PSI	10.0000
COLD SIDE PRESSUR	E DROP	PSI	10.0000
HEAT TRANSFER COEFF	ICIENT SPECIE	FICATION:	
HOT LIQUID COL	D LIQUID	BTU/HR-SQFT-R	149.6937
HOT 2-PHASE COL	D LIQUID	BTU/HR-SQFT-R	149.6937
HOT VAPOR COL	D LIQUID	BTU/HR-SQFT-R	149.6937
HOT LIQUID COL	D 2-PHASE	BTU/HR-SQFT-R	149.6937
HOT 2-PHASE COL	D 2-PHASE	BTU/HR-SQFT-R	149.6937
HOT VAPOR COL	D 2-PHASE	BTU/HR-SQFT-R	149.6937
HOT LIQUID COL	D VAPOR	BTU/HR-SQFT-R	149.6937
HOT 2-PHASE COL	D VAPOR	BTU/HR-SQFT-R	149.6937
HOT VAPOR COL	D VAPOR	BTU/HR-SQFT-R	149.6937

*** OVERALL RESULTS ***

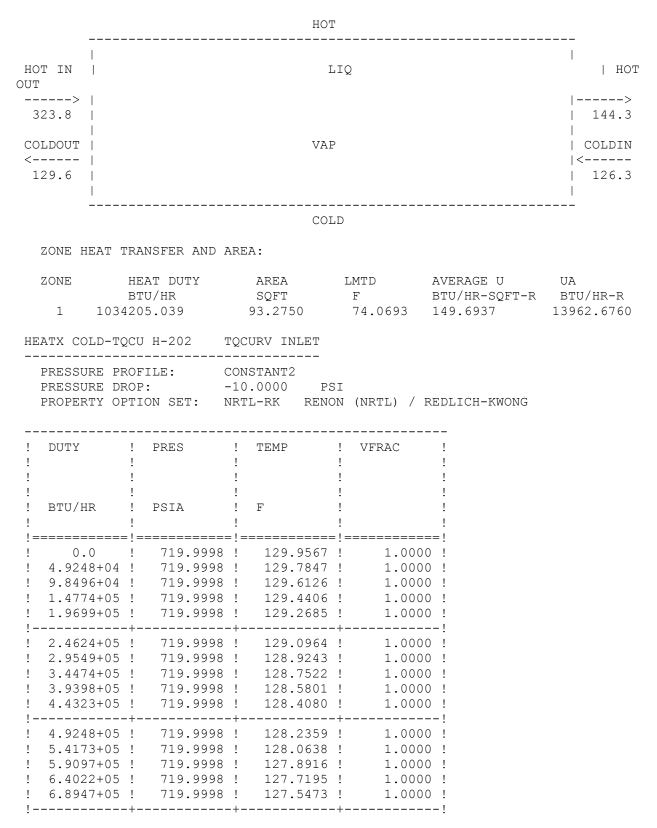
STREAMS:

	I		1		
S-318	>	НОТ	>	S-216	
T= 3.2379D+02					T=
1.4434D+02					
P= 1.3750D+01					P=
3.7500D+00					
V= 0.0000D+00					V=
0.0000D+00					
S-213 <	-	COLD	<	S-212	
T= 1.2963D+02					T=
1.2634D+02					
P= 7.1000D+02					P=
7.2000D+02					
V= 1.0000D+00					V=
1.0000D+00					

DUTY AND AREA: CALCULATED HEAT DUTY CALCULATED (REQUIRED) AREA ACTUAL EXCHANGER AREA PER CENT OVER-DESIGN	BTU/HR SQFT SQFT	1034205.0395 93.2750 93.2750 0.0000
HEAT TRANSFER COEFFICIENT: AVERAGE COEFFICIENT (DIRTY) UA (DIRTY)	BTU/HR-SQFT-R BTU/HR-R	149.6937 13962.6760
LOG-MEAN TEMPERATURE DIFFERENCE: LMTD CORRECTION FACTOR LMTD (CORRECTED) NUMBER OF SHELLS IN SERIES	F	1.0000 74.0693 1
PRESSURE DROP: HOTSIDE, TOTAL COLDSIDE, TOTAL	PSI PSI	10.0000 10.0000

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*** ZONE RESULTS ***
```

TEMPERATURE LEAVING EACH ZONE:



! 7.8797+05 ! 8.3721+05 ! 8.8646+05 ! 9.3571+05	! 719.9998 ! ! 719.9998 !	127.2030 ! 127.0308 ! 126.8587 ! 126.6865 !	1.0000 1.0000 1.0000 1.0000	
PRESSURE DRO	OFILE: CC	DNSTANT2 0 PSI		DLICH-KWONG
! ! !	! PRES ! ! ! ! ! ! ! ! PSIA ! ! !	TEMP ! ! ! F !	VFRAC	-
! 4.9248+04 ! 9.8496+04 ! 1.4774+05	!========= ! 13.7500 ! ! 13.7500 ! ! 13.7500 ! ! 13.7500 ! ! 13.7500 !	315.9888 ! 308.1228 ! 300.1892 !	0.0 0.0 0.0	
	! 13.7500 ! ! 13.7500 ! ! 13.7500 ! ! 13.7500 ! ! 13.7500 ! ! 13.7500 !	275.9711 !	0.0	
	! 13.7500 !	234.1361 ! 225.5354 ! 216.8533 !	0.0 0.0 0.0	
! 7.3872+05 ! 7.8797+05 ! 8.3721+05 ! 8.8646+05 ! 9.3571+05	! 13.7500 ! ! 13.7500 ! ! 13.7500 !	181.2802 ! 172.1684 !	0.0	
! 9.8496+05 ! 1.0342+06			0.0	! ! -

BLOCK: H-203 MODEL: HEATER

INLET STREAM:	S-213
OUTLET STREAM:	S-214
OUTLET HEAT STREAM:	HEAT2

PROPERTY OPTION	SET: NRTL-RK	RENON (NRTL) /	REDLICH-KWC	DNG
	*** MASS AND	ENERGY BALANCE IN	*** OUT	RELATIVE DIFF.
TOTAL BALANCE MOLE (LBMOL/H MASS (LB/HR ENTHALPY (BTU) 431		7489.1 31442. 552102E+09	0.00000 0.00000 0.00000
NET STREAMS CO2 UTILITIES CO2E	*** CO2 EQUIV 2E 0.4 CO2E 0.4 E PRODUCTION 0. PRODUCTION 0. UCTION 0.	07420E+07 LB/ 07420E+07 LB/ 00000 LB/ 00000 LB/	HR HR HR HR	
		DATA ***		
TWO PHASE TP SPECIFIED TEMPER PRESSURE DROP MAXIMUM NO. ITER CONVERGENCE TOLE	ATURE ATIONS	F PSI		181.400 10.0000 30 0.000100000
OUTLET PRESSURE HEAT DUTY OUTLET VAPOR FRA V-L PHASE EQUILI	BTU/HR CTION			700.00 0.15007E+08 1.0000
COMP	F(I)	(=)	Y(I)	
HYDROGEN 14976E+07	0.3382	27	0.47059E-	06 0.33827
METHANE ETHANE ETHENE PROPANE PROPENE BENZENE	0.36954 0.24974 0.28397E-01 0.69941E-02 0.23182E-02 0.45756E-02	0.21974 0.45435 0.44096E-01 0.22279E-01 0.58043E-02 0.23448	0.36954 0.24974 0.28397E- 0.69941E- 0.23182E- 0.45756E	020.65404020.83209
TOLUENE	0.16047E-03	0.18875E-01	0.16047E	-03 0.17712E-
P-XYL-01	0.13010E-05	0.34907E-03	0.13010E	-05 0.77647E-
1:3:5-01	0.33706E-07	0.20497E-04	0.33706E	-07 0.34261E-
1:3:5-01				

BLOCK: H-203, MODEL: HEATER INLET STREAM: S-407 INLET HEAT STREAM: HEAT2 OUTLET STREAM: S-217

PROPERTY OPTION SET:	NRTL-RK	RENON (NR	.TL) / RE	EDLICH-KW	IONG
**	* MASS AN	ID ENERGY BA IN			RELATIVE DIFF.
TOTAL BALANCE MOLE (LBMOL/HR) MASS (LB/HR) ENTHALPY (BTU/HR	1 1 0 - 0 .				
*** FEED STREAMS CO2E PRODUCT STREAMS CO2E NET STREAMS CO2E PROD UTILITIES CO2E PRODUCTION	DUCTION	0.00000 0.00000 0.00000	LB/HR LB/HR LB/HR LB/HR		
TWO PHASE PQ FLA:		PUT DATA **	*		
PRESSURE DROP DUTY FROM INLET HEAT MAXIMUM NO. ITERATION CONVERGENCE TOLERANCE	STREAM(S)	PSI BTU/H			10.0000 -0.150066+08 30 0.000100000
OUTLET TEMPERATURE OUTLET PRESSURE OUTLET VAPOR FRACTION	F PSIA	SULTS ***			380.29 196.00 0.42023E-01
V-L PHASE EQUILIBRIUM	:				
COMP : WATER :	F(I) 1.0000	X(I) 1.0000		Y(I) 1.0000	K(I) 1.0000
BLOCK: P-201 MODEL:	PUMP				
INLET STREAM: OUTLET STREAM: PROPERTY OPTION SET:	S-204 S-209 NRTL-RK	RENON (NR	TL) / RE	EDLICH-KW	JONG
**	* MASS AN	ND ENERGY BA IN		***)UT	RELATIVE DIFF.
TOTAL BALANCE MOLE (LBMOL/HR) MASS (LB/HR) ENTHALPY (BTU/HR	6		6288		0.00000 0.115701E-15 0.218946E-02
*** FEED STREAMS CO2E PRODUCT STREAMS CO2E NET STREAMS CO2E PRODUC UTILITIES CO2E PRODUCTION	DUCTION CTION	JIVALENT SUM 6188.99 6188.99 0.00000 0.00000 0.00000	MARY *** LB/HR LB/HR LB/HR LB/HR LB/HR		

*** INPUT DATA *** OUTLET PRESSURE PSIA 11.8000 PUMP EFFICIENCY 0.80000 1.00000 DRIVER EFFICIENCY FLASH SPECIFICATIONS: LIOUID PHASE CALCULATION NO FLASH PERFORMED MAXIMUM NUMBER OF ITERATIONS 30 TOLERANCE 0.000100000 *** RESULTS *** VOLUMETRIC FLOW RATE CUFT/HR 1,215.18 PRESSURE CHANGE PSI -43.3066 NPSH AVAILABLE FT-LBF/LB 0.0 FLUID POWER HP -3.82731 BRAKE POWER HP -3.06185 ELECTRICITY KW -2.28322 PUMP EFFICIENCY USED 0.80000 NET WORK REQUIRED HP -3.06185 HEAD DEVELOPED FT-LBF/LB -120.505 BLOCK: P-401 MODEL: PUMP _____ INLET STREAM: S-401 OUTLET STREAM: S-402 S-402 PROPERTY OPTION SET: NRTL-RK RENON (NRTL) / REDLICH-KWONG *** MASS AND ENERGY BALANCE *** OUT RELATIVE DIFF. IN TOTAL BALANCE 0.00000 57897.2 57897.2 0.104303E+07 0.104303E+07 MOLE (LBMOL/HR) MASS(LB/HR) 0.00000 ENTHALPY(BTU/HR) -0.711397E+10 -0.711322E+10 -0.104463E-03 *** CO2 EQUIVALENT SUMMARY *** FEED STREAMS CO2E0.00000LB/HRPRODUCT STREAMS CO2E0.00000I.R/HR PRODUCT STREAMS CO2E0.00000LB/HRNET STREAMS CO2E PRODUCTION0.00000LB/HRUTILITIES CO2E PRODUCTION0.00000LB/HRTOTAL CO2E PRODUCTION0.00000LB/HR TOTAL CO2E PRODUCTION 0.00000 LB/HR *** INPUT DATA *** OUTLET PRESSURE PSIA 205.632 PUMP EFFICIENCY 0.80000 DRIVER EFFICIENCY 1.00000 FLASH SPECIFICATIONS: LIQUID PHASE CALCULATION NO FLASH PERFORMED MAXIMUM NUMBER OF ITERATIONS 30 0.000100000 TOLERANCE *** RESULTS *** VOLUMETRIC FLOW RATE CUFT/HR 16,809.4 PRESSURE CHANGE PSI 191.128 NPSH AVAILABLE FT-LBF/LB 32.5915

FLUID POWER HP 233.654 BRAKE POWER HP 292.067 ELECTRICITY KW 217.795 0.80000 PUMP EFFICIENCY USED NET WORK REQUIRED HP 292.067 HEAD DEVELOPED FT-LBF/LB 443.547 BLOCK: PR-201 MODEL: SEP _____ INLET STREAM: S-214 OUTLET STREAMS: S-219, S-218 PROPERTY OPTION SET: NRTL-RK RENON (NRTL) / REDLICH-KWONG *** MASS AND ENERGY BALANCE *** IN OUT RELATIVE DIFF. TOTAL BALANCE MOLE (LBMOL/HR)27489.127489.10.00000MASS (LB/HR)431442.431442.-0.269828E-15 ENTHALPY(BTU/HR) -0.537095E+09 -0.539039E+09 0.360651E-02 *** CO2 EOUIVALENT SUMMARY *** FEED STREAMS CO2E0.407420E+07LB/HRPRODUCT STREAMS CO2E0.407420E+07LB/HR NET STREAMS CO2E PRODUCTION 0.00000 LB/HR UTILITIES CO2E PRODUCTION0.00000LB/HRTOTAL CO2E PRODUCTION0.00000LB/HR *** INPUT DATA *** INLET PRESSURE DROP PSI 10.0000 FLASH SPECS FOR STREAM S-219, TWO PHASE TP FLASH PRESSURE DROP PSI 0.0 MAXIMUM NO. ITERATIONS 30 CONVERGENCE TOLERANCE 0.000100000 FLASH SPECS FOR STREAM S-218 TWO PHASE TP FLASH PRESSURE DROP PSI 0.0 MAXIMUM NO. ITERATIONS 30 CONVERGENCE TOLERANCE 0.000100000 FRACTION OF FEED SUBSTREAM= MIXED STREAM= S-219, CPT= HYDROGEN FRACTION= 0.85000 0.28000 METHANE 0.0 ETHANE ETHENE 0.0 PROPANE 0.0 PROPENE 0.0 BENZENE 0.0 TOLUENE 0.0 P-XYL-01 0.0 1:3:5-01 0.0 WATER 0.0

OXYGEN	0.0
NITROGEN	0.0
CARBO-01	0.0

*** RESULTS ***

HEAT DUTY		BTU/HR		-0.19441E+07
COMPONENT = STREAM S-219, S-218	SUBSTREAM MIXED	SPLIT	FRACTION 0.85000 0.15000	
COMPONENT = STREAM S-219, S-218	METHANE SUBSTREAM MIXED MIXED	SPLIT	FRACTION 0.28000 0.72000	
COMPONENT = STREAM S-218	SUBSTREAM	SPLIT	FRACTION 1.00000	
COMPONENT = STREAM S-218	ETHENE SUBSTREAM MIXED	SPLIT	FRACTION 1.00000	
COMPONENT = STREAM S-218	SUBSTREAM	SPLIT	FRACTION 1.00000	
COMPONENT = STREAM S-218	PROPENE SUBSTREAM MIXED	SPLIT	FRACTION 1.00000	
COMPONENT = STREAM S-218	SUBSTREAM	SPLIT	FRACTION 1.00000	
COMPONENT = STREAM S-218	SUBSTREAM	SPLIT	FRACTION 1.00000	
COMPONENT = STREAM S-218	SUBSTREAM	SPLIT	FRACTION 1.00000	
COMPONENT = STREAM S-218	SUBSTREAM	SPLIT	FRACTION 1.00000	
BLOCK: PR-20	01, MODEL:			
INLET STREA OUTLET STRE	AM: EAM:	S-219, S-219	RENON (NRTL) / REDLICH-KWONG

*** MASS AND ENERGY BALANCE *** OUT RELATIVE DIFF. IN TOTAL BALANCE MOLE (LBMOL/HR)10748.310748.30.00000MASS (LB/HR)61564.461564.40.00000 ENTHALPY(BTU/HR) -0.828870E+08 -0.828870E+08 0.00000 *** CO2 EQUIVALENT SUMMARY *** FEED STREAMS CO2E0.114078E+07LB/HRPRODUCT STREAMS CO2E0.114078E+07LB/HR NET STREAMS CO2E PRODUCTION 0.00000 LB/HR 0.00000 LB/HR 0.00000 LR/UD UTILITIES CO2E PRODUCTION 0.00000 TOTAL CO2E PRODUCTION *** INPUT DATA *** VALVE OUTLET PRESSURE PSIA 200.000 VALVE FLOW COEF CALC. NO FLASH SPECIFICATIONS: NPHASE 2 MAX NUMBER OF ITERATIONS 30 0.000100000 CONVERGENCE TOLERANCE *** RESULTS *** VALVE PRESSURE DROP PSI 490.000 BLOCK: R-101 MODEL: RSTOIC _____ OUTLET STREAM: S-106, S-107 PROPERTY OPTION SET: NRTL-RK RENON (NRTL) / REDLICH-KWONG *** MASS AND ENERGY BALANCE *** OUT GENERATION RELATIVE ΤN DIFF. TOTAL BALANCE 23597.1 29549.4 5952.29 0.123115E-MOLE(LBMOL/HR) 15 MASS(LB/HR) 580601. 580601. 0.200508E-15 ENTHALPY(BTU/HR) -0.254505E+09 -0.542745E+08 -0.786745 *** CO2 EQUIVALENT SUMMARY *** FEED STREAMS CO2E0.302790E+07LB/HRPRODUCT STREAMS CO2E0.414343E+07LB/HR NET STREAMS COZE FRODUCTION 0.00000 DUTILITIES CO2E PRODUCTION 0.111552E+07 LB/HR *** INPUT DATA *** STOICHIOMETRY MATRIX: REACTION # 1: SUBSTREAM MIXED : HYDROGEN -1.00 METHANE 2.00 ETHANE -1.00

REACTION # 2: SUBSTREAM MIXED : HYDROGEN 1.00 ETHANE -1.00 ETHENE 1.00 REACTION # 3: SUBSTREAM MIXED : HYDROGEN 3.00 ETHANE -3.00 PROPENE 2.00 REACTION # 4: SUBSTREAM MIXED : HYDROGEN 1.00 ETHANE -3.00 PROPANE 2.00 REACTION # 5: SUBSTREAM MIXED : HYDROGEN 6.00 ETHANE -3.00 BENZENE 1.00 REACTION # 6: SUBSTREAM MIXED : HYDROGEN 6.00 METHANE 1.00 ETHANE -4.00 TOLUENE 1.00 REACTION # 7: SUBSTREAM MIXED : HYDROGEN 7.00 ETHANE -4.00 P-XYL-01 1.00 REACTION # 8: SUBSTREAM MIXED : HYDROGEN 7.00 METHANE 1.00 ETHANE -5.00 1:3:5-01 1.00 REACTION # 9: SUBSTREAM MIXED : HYDROGEN -2.00 METHANE 2.00 ETHENE -1.00 REACTION # 10: SUBSTREAM MIXED : ETHENE -3.00 PROPENE 2.00 REACTION # 11: SUBSTREAM MIXED : HYDROGEN -2.00 ETHENE -3.00 PROPANE 2.00 REACTION # 12: SUBSTREAM MIXED : HYDROGEN 3.00 ETHENE -3.00 BENZENE 1.00 REACTION # 13: SUBSTREAM MIXED : HYDROGEN 2.00 METHANE 1.00 ETHENE -4.00 TOLUENE 1.00 REACTION # 14: SUBSTREAM MIXED : HYDROGEN 3.00 ETHENE -4.00 P-XYL-01 1.00 REACTION # 15: SUBSTREAM MIXED : HYDROGEN 2.00 METHANE 1.00 ETHENE -5.00 1:3:5-01 1.00

REACTION # 16: SUBSTREAM MIXED HYDROGEN 5.00	: PROPANE	-2.00	BENZENE	1.00		
REACTION # 17: SUBSTREAM MIXED HYDROGEN 5.00	: ETHANE	1.00	PROPANE	-3.00	TOLUENE	1.00
REACTION # 18: SUBSTREAM MIXED HYDROGEN 5.00	: METHANE	1.00	PROPANE	-3.00	P-XYL-01	1.00
REACTION # 19: SUBSTREAM MIXED HYDROGEN -1.00	: METHANE	1.00	ETHENE	1.00	PROPENE	-1.00
REACTION # 20: SUBSTREAM MIXED HYDROGEN 3.00	: PROPENE	-2.00	BENZENE	1.00		
REACTION # 21: SUBSTREAM MIXED HYDROGEN 2.00	: ETHANE	1.00	PROPENE	-3.00	TOLUENE	1.00
REACTION # 22: SUBSTREAM MIXED HYDROGEN 2.00	: METHANE	1.00	PROPENE	-3.00	P-XYL-01	1.00
REACTION # 23: SUBSTREAM MIXED HYDROGEN 6.00	: PROPANE	-3.00	1:3:5-01	1.00		
REACTION # 24: SUBSTREAM MIXED HYDROGEN 3.00	: PROPENE	-3.00	1:3:5-01	1.00		
REACTION CONVERSION REACTION # 1:	SPECS: NUM	IBER= 2	2.4			
SUBSTREAM:MIXED	KEY COMP:	ETHANE	CONV FRAC:	0.7582E-	01	
REACTION # 2: SUBSTREAM:MIXED	KEY COMP:	ETHANE	CONV FRAC:	0.5904E-	01	
REACTION # 3: SUBSTREAM:MIXED	KEY COMP:	ETHANE	CONV FRAC:	0.7180E-	02	
REACTION # 4: SUBSTREAM:MIXED	KEY COMP:	ETHANE	CONV FRAC:	0.6850E-	02	
REACTION # 5: SUBSTREAM:MIXED	KEY COMP:	ETHANE	CONV FRAC:	0.1737		
REACTION # 6: SUBSTREAM:MIXED	KEY COMP:	ETHANE	CONV FRAC:	0.9250E-	01	
REACTION # 7: SUBSTREAM:MIXED			CONV FRAC:			
REACTION # 8:						
SUBSTREAM:MIXED REACTION # 9:			CONV FRAC:		υT	
SUBSTREAM:MIXED REACTION # 10:	KEY COMP:	ETHENE	CONV FRAC:	0.1871		

SUBSTREAM:MIXED	KEY	COMP:ETHENE	CONV FRAC:	0.1770E-01
REACTION # 11:				0 1 0 0 0 0 1
SUBSTREAM:MIXED	KEY	COMP: ETHENE	CONV FRAC:	0.1690E-01
REACTION # 12:				0 4007
SUBSTREAM:MIXED	KEY	COMP:ETHENE	CONV FRAC:	0.4287
REACTION # 13:	777737			0 0000
SUBSTREAM:MIXED	KEY	COMP:ETHENE	CONV FRAC:	0.2283
REACTION # 14:	777737			0 47700 01
SUBSTREAM:MIXED	KEY	COMP:ETHENE	CONV FRAC:	0.4770E-01
REACTION # 15:				0 7000 01
SUBSTREAM:MIXED	KEY	COMP: ETHENE	CONV FRAC:	0.7360E-01
REACTION # 16:	777737			0 1015
SUBSTREAM:MIXED	KEY	COMP: PROPANE	CONV FRAC:	0.1815
REACTION # 17:				0 0000
SUBSTREAM:MIXED	KEY	COMP: PROPANE	CONV FRAC:	0.2333
REACTION # 18:	777737			0 1007
SUBSTREAM:MIXED	KEY	COMP: PROPANE	CONV FRAC:	0.1037
REACTION # 19:				0 0407
SUBSTREAM:MIXED	KEY	COMP: PROPENE	CONV FRAC:	0.348/
REACTION # 20:	777737			0 0000
SUBSTREAM:MIXED	KEY	COMP: PROPENE	CONV FRAC:	0.2280
REACTION # 21:			CONV FRAC:	0 2021
SUBSTREAM:MIXED	KEI	COMP: PROPENE	CONV FRAC:	0.2931
REACTION # 22:	777737			0 1 2 0 2
SUBSTREAM:MIXED	KEY	COMP: PROPENE	CONV FRAC:	0.1303
REACTION # 23:			CONTREPAC-	
SUBSTREAM:MIXED	KLI	COMP: PROPANE	CONV FRAC:	0.6466E-01
REACTION # 24:				0 01100 01
SUBSTREAM:MIXED	КĿТ	COMP: PROPENE	CONV FRAC:	0.8119E-01

TWOPHASETPFLASHSPECIFIED TEMPERATURE F1,166.00PRESSURE DROPPSIMAXIMUM NO. ITERATIONS30CONVERGENCE TOLERANCE0.000100000SIMULTANEOUS REACTIONS0.000100000GENERATE COMBUSTION REACTIONS FOR FEED SPECIESNO

	*** RESULTS ***	
OUTLET TEMPERATURE	F	1166.0
OUTLET PRESSURE	PSIA	36.412
HEAT DUTY	BTU/HR	0.20023E+09
VAPOR FRACTION		1.0000

HEAT OF REACTIONS:

REFERENCE	HEAT OF
COMPONENT	REACTION
	BTU/LBMOL
ETHANE	-31236.
ETHANE	61556.
ETHANE	44994.
	COMPONENT ETHANE ETHANE

4	ETHANE	7607.0
5	ETHANE	53804.
6	ETHANE	37863.
7	ETHANE	43254.
8	ETHANE	32513.
9	ETHENE	-92791.
10	ETHENE	-16563.
11	ETHENE	-53951.
12	ETHENE	-7751.3
13	ETHENE	-23693.
14	ETHENE	-18302.
15	ETHENE	-29043.
16	PROPANE	69298.
17	PROPANE	44281.
18	PROPANE	41058.
19	PROPENE	-21553.
20	PROPENE	13218.
21	PROPENE	-11800.
22	PROPENE	-15023.
23	PROPANE	47987.
24	PROPENE	-8094.4

REACTION EXTENTS:

REACTION	REACTION
NUMBER	EXTENT
	LBMOL/HR
1	1000.8
2	779.28
3	31.590
4	30.138
5	764.32
6	305.23
7	63.818
8	78.720
9	150.19
10	4.7360
11	4.5219
12	114.71
13	45.815
14	9.5723
15	11.816
16	30.349
17	26.003
18	11.558
19	23.429
20	7.6597
21	6.5645
22	2.9183
23	7.2063
24	1.8184

V-L PHASE EQUILIBRIUM :

COMP	F(I)	X(I)	Y(I)	K(I)
HYDROGEN	0.	31468	0.64800E-03	0.31468
0.17991E+06				

METHANE ETHANE ETHENE PROPANE PROPENE BENZENE TOLUENE P-XYL-01 1:3:5-01	0.34962 0.24040 0.27165E-01 0.70624E-02 0.22738E-02 0.38964E-01 0.13485E-01 0.29782E-02 0.33693E-02		01 0.24040 02 0.27165E-0 02 0.70624E-0 03 0.22738E-0 0.38964E-0	02 610.46 02 936.21 01 32.113 01 16.383 02 16.498
BLOCK: RB-301 M	ODEL: HEATER			
INLET STREAM: INLET HEAT STREA OUTLET STREAM: PROPERTY OPTION	S-308	RENON (NRTL)	/ REDLICH-KWON	IG
	*** MASS ANI) ENERGY BALAN IN	ICE *** OUT	RELATIVE DIFF.
TOTAL BALANCE MOLE (LBMOL/H MASS (LB/HR ENTHALPY (BTU) 11	639.4	646.083 11639.4 -0.751412E+08	0.00000 0.00000 0.198309E-15
FEED STREAMS CO PRODUCT STREAMS NET STREAMS CO2 UTILITIES CO2E TOTAL CO2E PROD	2E (CO2E (E PRODUCTION (PRODUCTION ().00000 I).00000 I).00000 I	RY *** LB/HR LB/HR LB/HR LB/HR LB/HR	
TWO PHASE PO		JT DATA ***		
PRESSURE DROP DUTY FROM INLET MAXIMUM NO. ITER CONVERGENCE TOLE	HEAT STREAM(S) ATIONS	PSI BTU/HR	-9,032	10.0000 2,280. 30 0.000100000
	*** 000	SULTS ***		
OUTLET TEMPERATU OUTLET PRESSURE OUTLET VAPOR FRA	RE F PSIA		-	380.29 196.00 42023E-01
V-L PHASE EQUILI	BRIUM :			
COMP WATER	F(I) 1.0000	X(I) 1.0000	Y(I) 1.0000	K(I) 1.0000
BLOCK: RB-302 M INLET STREAM: INLET HEAT STREA OUTLET STREAM:	S-409			

PROPERTY OPTION SE	ET: NRTL-RE	K RENON (NRI	CL) / REDLICH-KW	IONG
	*** MASS A	AND ENERGY BAI IN		RELATIVE DIFF.
TOTAL BALANCE MOLE (LBMOL/HR) MASS (LB/HR) ENTHALPY (BTU/H	IR) –(2074.34 37369.8).241251E+09	2074.34 37369.8 -0.241251E+09	0.00000 0.00000 0.00000
FEED STREAMS CO2P PRODUCT STREAMS (NET STREAMS CO2E UTILITIES CO2E PP TOTAL CO2E PRODUC	E CO2E PRODUCTION RODUCTION CTION	0.00000 0.00000 0.00000 0.00000	LB/HR LB/HR LB/HR LB/HR LB/HR	
TWO PHASE PQ		NPUT DATA ***	-	
PRESSURE DROP DUTY FROM INLET HE MAXIMUM NO. ITERAT CONVERGENCE TOLERA	EAT STREAM(S) FIONS	PSI BTU/HF		10.0000 -0.289994+08 30 0.000100000
OUTLET TEMPERATURI OUTLET PRESSURE OUTLET VAPOR FRACT	E F PSIA	RESULTS ***		380.29 196.00 0.42023E-01
V-L PHASE EQUILIBE	RIUM :			
COMP WATER	F(I) 1.0000	X(I) 1.0000	Y(I) 1.0000	K(I) 1.0000
BLOCK: T-101 MOI	DEL: COMPR			
INLET STREAM: OUTLET STREAM: PROPERTY OPTION SP		K RENON (NRI	L) / REDLICH-KW	IONG
	*** MASS A	AND ENERGY BAI IN	ANCE *** OUT	RELATIVE DIFF.
TOTAL BALANCE MOLE (LBMOL/HR) MASS (LB/HR) ENTHALPY (BTU/H		16740.9 369878.).456152E+09	369878.	0.00000 0.00000 0.652194E-01
FEED STREAMS CO2H PRODUCT STREAMS (NET STREAMS CO2E UTILITIES CO2E PH TOTAL CO2E PRODUC	E CO2E PRODUCTION RODUCTION	QUIVALENT SUMM 0.293342E+07 0.293342E+07 0.00000 0.00000 0.00000	LB/HR LB/HR	

*** INPUT DATA *** ISENTROPIC TURBINE OUTLET PRESSURE PSIA 53.6640 ISENTROPIC EFFICIENCY 0.80000 MECHANICAL EFFICIENCY 1.00000 *** RESULTS *** INDICATED HORSEPOWER REQUIREMENT HP -12,508.0 BRAKE HORSEPOWER REQUIREMENT HP -12,508.0 NET WORK REQUIRED HP -12,508.0 POWER LOSSES HP 0.0 ISENTROPIC HORSEPOWER REQUIREMENT HP -15,634.9 CALCULATED OUTLET TEMP F -26.9538 ISENTROPIC TEMPERATURE F -70.4389 EFFICIENCY (POLYTR/ISENTR) USED 0.80000 OUTLET VAPOR FRACTION 0.99084 HEAD DEVELOPED, FT-LBF/LB -83,695.7 MECHANICAL EFFICIENCY USED 1.00000 INLET HEAT CAPACITY RATIO 1.33025 INLET VOLUMETRIC FLOW RATE , CUFT/HR OUTLET VOLUMETRIC FLOW RATE, CUFT/HR 150,837. 1,398,420. INLET COMPRESSIBILITY FACTOR 0.90406 OUTLET COMPRESSIBILITY FACTOR 0.96534 AV. ISENT. VOL. EXPONENT 1.21336 AV. ISENT. TEMP EXPONENT 1.24255 AV. ACTUAL VOL. EXPONENT 1.14686 AV. ACTUAL TEMP EXPONENT 1.18167 BLOCK: T-102 MODEL: COMPR _____ INLET STREAM: OUTLET STREAM: S-101 S-102 PROPERTY OPTION SET: NRTL-RK RENON (NRTL) / REDLICH-KWONG *** MASS AND ENERGY BALANCE *** OUT IN RELATIVE DIFF. TOTAL BALANCE MOLE (LBMOL/HR)6283.926283.92MASS (LB/HR)189837.189837. 0.00000 0.00000 ENTHALPY(BTU/HR) -0.230063E+09 -0.236713E+09 0.280951E-01 *** CO2 EQUIVALENT SUMMARY *** FEED STREAMS CO2E25202.9LB/HRPRODUCT STREAMS CO2E25202.9LB/HR LB/HR 25202.9 NET STREAMS CO2E PRODUCTION 0.00000 LB/HR UTILITIES CO2E PRODUCTION 0.00000 LB/HR TOTAL CO2E PRODUCTION 0.00000 LB/HR *** INPUT DATA *** ISENTROPIC TURBINE OUTLET PRESSURE PSIA 53.6640 ISENTROPIC EFFICIENCY 0.80000 MECHANICAL EFFICIENCY 1.00000

*** RESULTS *** INDICATED HORSEPOWER REQUIREMENT HP -2,613.74 BRAKE HORSEPOWER REQUIREMENT HP -2,613.74 NET WORK REQUIRED HP -2,613.74 POWER LOSSES HP 0.0 ISENTROPIC HORSEPOWER REQUIREMENT HP CALCULATED OUTLET TEMP F -3,267.17 -37.4871 ISENTROPIC TEMPERATURE F -59.9360 EFFICIENCY (POLYTR/ISENTR) USED 0.80000 OUTLET VAPOR FRACTION 1.00000 HEAD DEVELOPED, FT-LBF/LB -34,076.7 MECHANICAL EFFICIENCY USED 1.00000 INLET HEAT CAPACITY RATIO 1.32990 INLET VOLUMETRIC FLOW RATE , CUFT/HR116,830.OUTLET VOLUMETRIC FLOW RATE, CUFT/HR500,206. INLET COMPRESSIBILITY FACTOR 0.84974 OUTLET COMPRESSIBILITY FACTOR 0.94285 AV. ISENT. VOL. EXPONENT 1.14878 AV. ISENT. TEMP EXPONENT 1.23166 AV. ACTUAL VOL. EXPONENT 1.09733 AV. ACTUAL TEMP EXPONENT 1.18182 BLOCK: T-103 MODEL: COMPR _____ INLET STREAM: S-305 OUTLET STREAM: S-104 PROPERTY OPTION SET: NRTL-RK RENON (NRTL) / REDLICH-KWONG *** MASS AND ENERGY BALANCE *** OUT RELATIVE DIFF. IN TOTAL BALANCE
 MOLE (LBMOL/HR)
 572.227
 572.227
 0.00000

 MASS (LB/HR)
 20884.5
 20884.5
 0.00000

 ENTHALPY (BTU/HR)
 -0.937930E+07
 -0.962404E+07
 0.254296E-01
 *** CO2 EQUIVALENT SUMMARY *** FEED STREAMS CO2E69231.5LB/HRPRODUCT STREAMS CO2E69231.5LB/HR PRODUCT STREAMS CO2E69231.5LB/HRNET STREAMS CO2E PRODUCTION0.00000LB/HR UTILITIES CO2E PRODUCTION0.00000LB/HRTOTAL CO2E PRODUCTION0.00000LB/HR *** INPUT DATA *** ISENTROPIC TURBINE OUTLET PRESSURE PSIA 53.6640 ISENTROPIC EFFICIENCY 0.80000 MECHANICAL EFFICIENCY 1.00000 *** RESULTS *** INDICATED HORSEPOWER REQUIREMENT HP -96.1846 BRAKE HORSEPOWER REQUIREMENT HP -96.1846 NET WORK REQUIRED HP -96.1846 HP 0.0 -120.231 ISENTROPIC HORSEPOWER REQUIREMENT HP

CALCULATED OUTLET TEMP F 166.662 ISENTROPIC TEMPERATURE F 165.112 EFFICIENCY (POLYTR/ISENTR) USED 0.80000 OUTLET VAPOR FRACTION 0.99432 HEAD DEVELOPED, FT-LBF/LB -11,398.7 MECHANICAL EFFICIENCY USED 1.00000 INLET HEAT CAPACITY RATIO 1.17062 INLET VOLUMETRIC FLOW RATE , CUFT/HR OUTLET VOLUMETRIC FLOW RATE, CUFT/HR 46,729.6 69,613.8 INLET COMPRESSIBILITY FACTOR 0.96763 OUTLET COMPRESSIBILITY FACTOR 0.97129 AV. ISENT. VOL. EXPONENT 1.11882 AV. ISENT. TEMP EXPONENT 1.11154 AV. ACTUAL VOL. EXPONENT 1.09411 AV. ACTUAL TEMP EXPONENT 1.10457 BLOCK: T-401 MODEL: COMPR _____ INLET STREAM: S-404, OUTLET STREAM: S-406 PROPERTY OPTION SET: NRTL-RK RENON (NRTL) / REDLICH-KWONG *** MASS AND ENERGY BALANCE *** OUT RELATIVE DIFF. IN TOTAL BALANCE MOLE (LBMOL/HR)3793.853793.850.00000MASS (LB/HR)68347.268347.20.00000 ENTHALPY(BTU/HR) -0.384325E+09 -0.388197E+09 0.997281E-02 *** CO2 EQUIVALENT SUMMARY *** FEED STREAMS CO2E0.00000LB/HRPRODUCT STREAMS CO2E0.00000LB/HR NET STREAMS CO2E PRODUCTION0.00000LB/HRUTILITIES CO2E PRODUCTION0.00000LB/HRTOTAL CO2E PRODUCTION0.00000LB/HR *** INPUT DATA *** ISENTROPIC TURBINE 206.000 OUTLET PRESSURE PSIA ISENTROPIC EFFICIENCY 0.80000 MECHANICAL EFFICIENCY 1.00000 *** RESULTS *** INDICATED HORSEPOWER REQUIREMENT HP -1,521.52 BRAKE HORSEPOWER REQUIREMENT HP -1,521.52 NET WORK REQUIRED -1,521.52 HP POWER LOSSES HP 0.0 POWER LOSSES HP ISENTROPIC HORSEPOWER REQUIREMENT HP CALCULATED OUTLET TEMP F -1,901.90 CALCULATED OUTLET TEMP F 384.467 ISENTROPIC TEMPERATURE F 384.467 EFFICIENCY (POLYTR/ISENTR) USED 0.80000 OUTLET VAPOR FRACTION 0.95971 -55,097.6 HEAD DEVELOPED, FT-LBF/LB MECHANICAL EFFICIENCY USED 1.00000 INLET HEAT CAPACITY RATIO 1.43709

INLET VOLUMETRIC FLOW RATE , CUFT/HR 72,687.4 151,123. OUTLET VOLUMETRIC FLOW RATE, CUFT/HR INLET COMPRESSIBILITY FACTOR 0.89696 OUTLET COMPRESSIBILITY FACTOR 0.90584 AV. ISENT. VOL. EXPONENT 1.13878 AV. ISENT. TEMP EXPONENT 1.12661 AV. ACTUAL VOL. EXPONENT 1.11146 1.12661 AV. ACTUAL TEMP EXPONENT BLOCK: V-101-2 MODEL: MIXER _____ INLET STREAMS: S-102 OUTLET STREAM: S-105 S-104 S-103 PROPERTY OPTION SET: NRTL-RK RENON (NRTL) / REDLICH-KWONG *** MASS AND ENERGY BALANCE *** OUT RELATIVE DIFF. IN TOTAL BALANCE 23597.023597.00.00000580599.580599.-0.200509E-15 MOLE(LBMOL/HR) MASS(LB/HR) ENTHALPY (BTU/HR) -0.734316E+09 -0.734316E+09 0.162341E-15 *** CO2 EOUIVALENT SUMMARY *** FEED STREAMS CO2E0.302786E+07LB/HRPRODUCT STREAMS CO2E0.302786E+07LB/HR NET STREAMS CO2E PRODUCTION 0.00000 LB/HR UTILITIES CO2E PRODUCTION 0.00000 LB/HR TOTAL CO2E PRODUCTION 0.00000 LB/HR *** INPUT DATA *** TWO PHASE FLASH 30 MAXIMUM NO. ITERATIONS CONVERGENCE TOLERANCE 0.000100000 OUTLET PRESSURE: MINIMUM OF INLET STREAM PRESSURES BLOCK: V-103 MODEL: MIXER _____ INLET STREAMS: S-109 OUTLET STREAM: S-220, S-220 PROPERTY OPTION SET: NRTL-RK RENON (NRTL) / REDLICH-KWONG *** MASS AND ENERGY BALANCE *** IN OUT RELATIVE DIFF. TOTAL BALANCE MOLE (LBMOL/HR)11566.811566.80.00000MASS (LB/HR)290043.290043.0.00000 ENTHALPY(BTU/HR) -0.145901E+08 -0.145901E+08 -0.255330E-15 *** CO2 EQUIVALENT SUMMARY *** FEED STREAMS CO2E200419.LB/HRPRODUCT STREAMS CO2E200419.LB/HR PRODUCT STREAMS CO2E200419.NET STREAMS CO2E PRODUCTION0.000000.000000.00000 LB/HR UTILITIES CO2E PRODUCTION 0.00000 LB/HR TOTAL CO2E PRODUCTION 0.00000 LB/HR *** INPUT DATA *** TWO PHASE FLASH

MAXIMUM NO. ITERATIONS 30 0.000100000 CONVERGENCE TOLERANCE OUTLET PRESSURE: MINIMUM OF INLET STREAM PRESSURES BLOCK: V-201 MODEL: VALVE _____ INLET STREAM: S-205 OUTLET STREAM: S-205 PROPERTY OPTION SET: NRTL-RK RENON (NRTL) / REDLICH-KWONG *** MASS AND ENERGY BALANCE *** OUT IN RELATIVE DIFF. TOTAL BALANCE MOLE (LBMOL/HR) 365.791 365.791 365.791 17906.5 17906.5 0.00000 MASS(LB/HR) 0.00000 ENTHALPY(BTU/HR) -0.365627E+07 -0.365627E+07 0.00000 PRODUCT STREAMS CO2E 33131.7 LB/HR NET STREAMS CO2E 33121 7 *** CO2 EQUIVALENT SUMMARY *** NET STREAMS CO2E PRODUCTION 0.00000 LB/HR UTILITIES CO2E PRODUCTION 0.00000 LB/HR 0.00000 TOTAL CO2E PRODUCTION LB/HR *** INPUT DATA *** VALVE OUTLET PRESSURE PSIA 11.8000 VALVE FLOW COEF CALC. NO FLASH SPECIFICATIONS: 2 NPHASE MAX NUMBER OF ITERATIONS 30 0.000100000 CONVERGENCE TOLERANCE *** RESULTS *** VALVE PRESSURE DROP PSI 718.200 BLOCK: V-202 MODEL: VALVE _____ INDET STREAM:S-202OUTLET STREAM:S-207 S-207 PROPERTY OPTION SET: NRTL-RK RENON (NRTL) / REDLICH-KWONG *** MASS AND ENERGY BALANCE *** OUT RELATIVE DIFF. ΤN TOTAL BALANCE 458.147458.14729529.929529.9 MOLE(LBMOL/HR) MASS(LB/HR) 0.00000 MASS(LB/HR) 0.00000 0.155143E+07 0.155143E+07 ENTHALPY (BTU/HR) 0.00000 *** CO2 EQUIVALENT SUMMARY *** FEED STREAMS CO2E19945.6LB/HRPRODUCT STREAMS CO2E19945.6L.R/HP NET STREAMS CO2E PRODUCTION 0.00000 LB/HR 0.00000 LB/HR 0.00000 LB/HR UTILITIES CO2E PRODUCTION 0.00000 TOTAL CO2E PRODUCTION

*** INPUT DATA *** 11.8000 VALVE OUTLET PRESSURE PSIA NO VALVE FLOW COEF CALC. FLASH SPECIFICATIONS: NPHASE 2 MAX NUMBER OF ITERATIONS 30 CONVERGENCE TOLERANCE 0.000100000 *** RESULTS *** VALVE PRESSURE DROP PSI 288.399 BLOCK: V-203 MODEL: VALVE _____ INLET STREAM: OUTLET STREAM: S-203 S-208 PROPERTY OPTION SET: NRTL-RK RENON (NRTL) / REDLICH-KWONG *** MASS AND ENERGY BALANCE *** OUT RELATIVE DIFF. IN TOTAL BALANCE MOLE (LBMOL/HR)512.796512.7960.00000MASS (LB/HR)38836.338836.30.00000ENTHALPY (BTU/HR)0.453366E+070.453366E+070.00000 MOLE (LBMOL/HR) *** CO2 EQUIVALENT SUMMARY *** rtel STREAMS CO2E9965.80LB/HRPRODUCT STREAMS CO2E9965.80TB/UD LB/HR NET STREAMS CO2E PRODUCTION 0.00000 UTILITIES CO2E PRODUCTION 0.00000 LB/HR TOTAL CO2E PRODUCTION 0.00000 LB/HR *** INPUT DATA *** VALVE OUTLET PRESSURE PSIA 11.8000 VALVE FLOW COEF CALC. NO FLASH SPECIFICATIONS: 2 NPHASE MAX NUMBER OF ITERATIONS 30 CONVERGENCE TOLERANCE 0.000100000 *** RESULTS *** VALVE PRESSURE DROP PSI 114.040 BLOCK: V-203-5 MODEL: MIXER -----INLET STREAMS: S-209 OUTLET STREAM: S-210 S-208 S-207 S-206 PROPERTY OPTION SET: NRTL-RK RENON (NRTL) / REDLICH-KWONG *** MASS AND ENERGY BALANCE *** OUT RELATIVE DIFF. ΤN

TOTAL BALANCE MOLE (LBMOL/HR)2060.302060.31-0.247041E-05MASS (LB/HR)149159.149159.-0.299691E-05 ENTHALPY (BTU/HR) 0.597930E+07 0.597939E+07 -0.151762E-04 *** CO2 EQUIVALENT SUMMARY *** PRODUCT STREAMS CO2E 69232.1 LB/HR NET CODUCT LB/HR NET STREAMS CO2E PRODUCTION -0.574376 0.00000 -0.574376 LB/HR UTILITIES CO2E PRODUCTION TOTAL CO2E PRODUCTION LB/HR *** INPUT DATA *** TWO PHASE FLASH MAXIMUM NO. ITERATIONS 30 CONVERGENCE TOLERANCE 0.000100000 OUTLET PRESSURE: MINIMUM OF INLET STREAM PRESSURES BLOCK: V-206 MODEL: FSPLIT _____ INLET STREAM: S-219 OUTLET STREAMS: S-221 S-220 PROPERTY OPTION SET: NRTL-RK RENON (NRTL) / REDLICH-KWONG *** MASS AND ENERGY BALANCE *** OUT IN RELATIVE DIFF. TOTAL BALANCE 0.00000 MOLE (LBMOL/HR)10748.310748.3MASS (LB/HR)61564.461564.4 0.00000 ENTHALPY(BTU/HR) -0.828870E+08 -0.828870E+08 0.00000 *** CO2 EQUIVALENT SUMMARY *** FEED STREAMS CO2E0.114078E+07LB/HRPRODUCT STREAMS CO2E0.114078E+07LB/HR NET STREAMS CO2E PRODUCTION 0.00000 LB/HR UTILITIES CO2E PRODUCTION0.00000TOTAL CO2E PRODUCTION0.00000 LB/HR LB/HR *** INPUT DATA *** FRACTION OF FLOW STRM=S-221 FRAC= 0.82431 *** RESULTS *** STREAM= S-221 SPLIT= 0.82431 KEY= 0 STREAM-ORDER= 1 0.17569 Ο S-220 2 BLOCK: V-401-3 MODEL: MIXER _____ INLET STREAMS: S-402 OUTLET STREAM: S-403 S-317 S-217 S-308 PROPERTY OPTION SET: NRTL-RK RENON (NRTL) / REDLICH-KWONG *** MASS AND ENERGY BALANCE *** OUT RELATIVE DIFF. ΤN

TOTAL BALANCE MOLE (LBMOL/HR)61691.161691.10.00000MASS (LB/HR)0.111138E+070.111138E+07-0.209497E-15ENTHALPY (BTU/HR)-0.755446E+10-0.755446E+100.252480E-15 *** CO2 EOUIVALENT SUMMARY ***
 reluid STREAMS CO2E
 0.00000
 LB/HR

 PRODUCT STREAMS CO2E
 0.00000
 LD/HR
 NET STREAMS CO2E PRODUCTION 0.00000 LB/HR UTILITIES CO2E PRODUCTION0.00000TOTAL CO2E PRODUCTION0.00000 LB/HR LB/HR *** INPUT DATA *** TWO PHASE FLASH MAXIMUM NO. ITERATIONS 30 CONVERGENCE TOLERANCE 0.000100000 OUTLET PRESSURE: MINIMUM OF INLET STREAM PRESSURES BLOCK: V-404 MODEL: MIXER _____ INLET STREAMS: S-221 OUTLET STREAM: S-221, S-221 S-410 PROPERTY OPTION SET: NRTL-RK RENON (NRTL) / REDLICH-KWONG *** MASS AND ENERGY BALANCE *** OUT IN RELATIVE DIFF. TOTAL BALANCE MOLE (LBMOL/HR)54270.954270.90.00000MASS (LB/HR)0.136087E+070.136087E+070.00000 54270.9 ENTHALPY (BTU/HR) -0.684562E+08 -0.684562E+08 -0.217674E-15 *** CO2 EQUIVALENT SUMMARY *** FEED STREAMS CO2E940357.LB/HRPRODUCT STREAMS CO2E940357.LB/HR NET STREAMS CO2E PRODUCTION 0.00000 LB/HR UTILITIES CO2E PRODUCTION0.00000TOTAL CO2E PRODUCTION0.00000 LB/HR LB/HR *** INPUT DATA *** TWO PHASE FLASH MAXIMUM NO. ITERATIONS 30 CONVERGENCE TOLERANCE 0.000100000 OUTLET PRESSURE: MINIMUM OF INLET STREAM PRESSURES BLOCK: V-405 MODEL: FSPLIT _____ INLET STREAM: S-404 OUTLET STREAMS: S-405 S-404, PROPERTY OPTION SET: NRTL-RK RENON (NRTL) / REDLICH-KWONG *** MASS AND ENERGY BALANCE *** OUT RELATIVE DIFF. IN TOTAL BALANCE
 MOLE (LBMOL/HR)
 61691.1
 61691.1
 0.00000

 MASS (LB/HR)
 0.111138E+07
 0.111138E+07
 -0.209497E-15

 ENTHALPY (BTU/HR)
 -0.624944E+10
 -0.624944E+10
 0.152601E-15

*** CO2 EQUIVALENT SUMMARY *** FEED STREAMS CO2E0.00000LB/HRPRODUCT STREAMS CO2E0.00000LB/HR 0.00000 NET STREAMS CO2E PRODUCTION 0.00000 LB/HR UTILITIES CO2E PRODUCTION 0.00000 LB/HR TOTAL CO2E PRODUCTION 0.00000 LB/HR *** TNPUT DATA *** STRM=S-404, FRAC= FRACTION OF FLOW 0.061498 *** RESULTS *** STREAM= S-405 SPLIT= 0.93850 KEY= 0 STREAM-ORDER= 2 S-404, 0.061498 0 1 BLOCK: V-406-7 MODEL: FSPLIT _____ INLET STREAM: S-406 OUTLET STREAM: S-400 S-409 S-407 PROPERTY OPTION SET: NRTL-RK RENON (NRTL) / REDLICH-KWONG *** MASS AND ENERGY BALANCE *** OUT IN RELATIVE DIFF. TOTAL BALANCE MOLE (LBMOL/HR)3793.853793.850.119864E-15MASS (LB/HR)68347.268347.20.00000 ENTHALPY (BTU/HR) -0.388197E+09 -0.388197E+09 0.00000 *** CO2 EQUIVALENT SUMMARY *** FEED STREAMS CO2E0.00000LB/HRPRODUCT STREAMS CO2E0.00000LB/HR NET STREAMS CO2E PRODUCTION 0.00000 LB/HR UTILITIES CO2E PRODUCTION0.00000TOTAL CO2E PRODUCTION0.00000 LB/HR LB/HR *** INPUT DATA *** STRM=S-408 FRAC= STRM=S-409 FRAC= FRACTION OF FLOW 0.17030 0.54676 *** RESULTS *** STREAM= S-408 SPLIT= 0.17030 KEY= 0 STREAM-ORDER= 1 0.54676 S-409 0 2 S-407 0.28294 0 3

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A-5.4.1. TPFQ Profile for COL-301

Stag	Temperatu re	Pressu re	Heat duty	Liquid from (Mass)	Vapor from (Mass)	Liquid enthalpy	Vapor enthalpy	Liquid flow (Mass)	Vapor flow (Mass)
0	F	psia	Btu/hr	lb/hr	lb/hr	Btu/lbmol	Btu/lbmol	lb/hr	lb/hr
1	1 186	83.0	-548155	2104	20449	17461	-16513	2104	20449
2	209	93.0	0	2179	22553	18086	-13986	2179	22553
3	20)	93.2	0	2163	22628	17872	-13918	2163	22628
4	211	93.3	0	2105	22612	17719	-13945	2150	22612
5	212	93.5	0	2140	22599	17619	-13964	2140	22599
6	212	93.6	0	2131	22589	17545	-13978	2131	22589
7	212	93.8	0	2123	22580	17487	-13989	2123	22580
8	212	93.9	0	2115	22572	17438	-13999	2115	22572
9	212	94.1	0	2108	22564	17395	-14007	2108	22564
10	212	94.2	0	2102	22558	17358	-14014	2102	22558
11	213	94.4	0	2095	22551	17322	-14021	2095	22551
12	213	94.5	0	2089	22544	17288	-14028	2089	22544
13	213	94.7	0	2082	22538	17252	-14034	2082	22538
14	213	94.8	0	2075	22531	17213	-14041	2075	22531
15	213	95.0	0	2067	22524	17166	-14048	2067	22524
16	213	95.1	0	2057	22516	17108	-14057	2057	22516
17	213	95.3	0	2047	22507	17033	-14067	2047	22507
18	214	95.4	0	2038	22496	16935	-14078	2038	22496
19	214	95.6	0	2053	22487	16807	-14091	2053	22487
20	214	95.7	0	2296	22502	16661	-14088	2296	22502
21	210	95.9	0	4541	22745	16592	-13934	4541	22745
22	190	96.0	0	174113	21844	12445	-7877	174113	21844
23	283	96.2	0	221917	48283	20266	17208	221917	48283
24	317	96.3	0	247393	96087	23942	31976	247393	96087
25	326	96.5	0	255826	121562	25200	37085	255826	121562
26	329	96.6	0	258622	129996	25624	38784	258622	129996
27	330	96.8	0	259600	132791	25771	39353	259600	132791

Stag	Temperatu	Pressu	Heat	Liquid	from	Vapor	from		Vapor	Liquid	flow	Vapor	flow
e	re	re	duty	(Mass)		(Mass)		enthalpy	enthalpy	(Mass)		(Mass)	
28	331	96.9	0		259992		133769	25827	39549		259992		133769
29	331	97.1	0		260195		134162	25852	39621		260195		134162
30	331	97.2	0		260335		134365	25867	39649		260335		134365
31	331	97.4	0		260454		134505	25877	39662		260454		134505
32	332	97.5	0		260565		134623	25886	39669		260565		134623
33	332	97.7	0		260673		134734	25894	39675		260673		134734
34	332	97.8	0		260780		134843	25902	39679		260780		134843
35	332	98.0	0		260887		134950	25910	39683		260887		134950
36	332	98.1	0		260993		135056	25918	39687		260993		135056
37	332	98.3	0		261100		135163	25925	39691		261100		135163
38	333	98.4	0		261206		135269	25933	39694		261206		135269
39	333	98.6	0		261312		135375	25941	39698		261312		135375
40	333	98.7	0		261418		135481	25949	39702		261418		135481
41	333	98.9	0		261524		135587	25957	39706		261524		135587
42	333	99.0	0		261630		135693	25964	39709		261630		135693
43	333	99.2	0		261736		135799	25972	39713		261736		135799
44	333	99.3	0		261842		135905	25980	39717		261842		135905
45	334	99.5	0		261948		136011	25988	39721		261948		136011
46	334	99.6	0		262054		136117	25995	39724		262054		136117
47	334	99.8	0		262159		136223	26003	39728		262159		136223
48	334	99.9	0		262265		136329	26011	39732		262265		136329
49	334	100.1	0		262371		136435	26018	39735		262371		136435
50	334	100.2	0		262477		136541	26026	39739		262477		136541
51	334	100.4	0		262584		136647	26033	39742		262584		136647
52	335	100.5	0		262691		136753	26041	39746		262691		136753
53	335	100.7	0		262799		136860	26048	39749		262799		136860
54	335	100.8	0		262909		136969	26055	39751		262909		136969
55	335	101.0	0		263022		137079	26060	39753		263022		137079
56	335	101.1	0		263141		137192	26065	39753		263141		137192

Stag	Temperatu	Pressu	Heat	Liquid	from	Vapor	from	Liquid	Vapor	Liquid	flow	Vapor	flow
e	re	re	duty	(Mass)		(Mass)		enthalpy	enthalpy	(Mass)		(Mass)	
57	335	101.3	0	26	63269		137310	26067	39751		263269		137310
58	335	101.4	0	26	63413		137438	26065	39745		263413		137438
59	336	101.6	0	26	63586		137583	26056	39731		263586		137583
60	336	101.7	0	26	63806		137755	26034	39704		263806		137755
61	336	101.9	0	26	64108		137975	25987	39653		264108		137975
62	337	102.0	0	26	64551		138278	25897	39559		264551		138278
63	338	102.2	0	26	65214		138720	25724	39386		265214		138720
64	340	102.3	0	26	64044		139384	25387	39071		264044		139384
			196044										
65	349	102.5	03	12	25831		138214	23436	38680		125831		138214

A-5.4.2. Liquid Mass Composition Profile for COL-301

Stage	Hydrogen	Methane	Ethane	Ethylene	Propane	Propylene	Benzene	Toluene	Xylene	Trimethylbenzene
1.00E+00	2.21E-16	2.56E-03	2.41E-02	1.72E-03	4.32E-03	5.07E-04	7.91E-01	1.67E-01	9.08E-03	1.51E-05
2.00E+00	3.05E-16	2.11E-03	2.30E-02	1.66E-03	4.16E-03	5.12E-04	7.55E-01	1.98E-01	1.48E-02	3.30E-05
3.00E+00	3.09E-16	2.07E-03	2.28E-02	1.65E-03	4.13E-03	5.13E-04	7.37E-01	2.12E-01	2.01E-02	5.93E-05
4.00E+00	3.11E-16	2.06E-03	2.28E-02	1.65E-03	4.12E-03	5.15E-04	7.27E-01	2.17E-01	2.47E-02	9.69E-05
5.00E+00	3.12E-16	2.05E-03	2.28E-02	1.65E-03	4.12E-03	5.16E-04	7.22E-01	2.18E-01	2.88E-02	1.50E-04
6.00E+00	3.13E-16	2.05E-03	2.28E-02	1.65E-03	4.12E-03	5.17E-04	7.19E-01	2.17E-01	3.22E-02	2.26E-04
7.00E+00	3.14E-16	2.05E-03	2.28E-02	1.65E-03	4.13E-03	5.18E-04	7.17E-01	2.17E-01	3.52E-02	3.33E-04
8.00E+00	3.15E-16	2.05E-03	2.28E-02	1.65E-03	4.13E-03	5.19E-04	7.15E-01	2.16E-01	3.77E-02	4.83E-04
9.00E+00	3.16E-16	2.05E-03	2.28E-02	1.65E-03	4.13E-03	5.20E-04	7.13E-01	2.15E-01	3.97E-02	6.95E-04
1.00E+01	3.17E-16	2.05E-03	2.28E-02	1.65E-03	4.14E-03	5.21E-04	7.12E-01	2.14E-01	4.14E-02	9.91E-04
1.10E+01	3.18E-16	2.05E-03	2.28E-02	1.65E-03	4.14E-03	5.21E-04	7.11E-01	2.14E-01	4.28E-02	1.41E-03
1.20E+01	3.19E-16	2.05E-03	2.29E-02	1.65E-03	4.14E-03	5.22E-04	7.10E-01	2.13E-01	4.39E-02	1.98E-03
1.30E+01	3.20E-16	2.05E-03	2.29E-02	1.66E-03	4.15E-03	5.23E-04	7.09E-01	2.12E-01	4.48E-02	2.79E-03
1.40E+01	3.21E-16	2.05E-03	2.29E-02	1.66E-03	4.15E-03	5.23E-04	7.08E-01	2.11E-01	4.54E-02	3.91E-03
1.50E+01	3.22E-16	2.05E-03	2.29E-02	1.66E-03	4.15E-03	5.24E-04	7.07E-01	2.11E-01	4.57E-02	5.45E-03
1.60E+01	3.23E-16	2.05E-03	2.29E-02	1.66E-03	4.15E-03	5.24E-04	7.05E-01	2.10E-01	4.59E-02	7.58E-03

Stage	Hydrogen	Methane	Ethane	Ethylene	Propane	Propylene	Benzene	Toluene	Xylene	Trimethylbenzene
1.70E+01	3.24E-16	2.04E-03	2.29E-02	1.66E-03	4.16E-03	5.25E-04	7.03E-01	2.09E-01	4.59E-02	1.05E-02
1.80E+01	3.25E-16	2.04E-03	2.29E-02	1.66E-03	4.16E-03	5.25E-04	7.01E-01	2.08E-01	4.56E-02	1.45E-02
1.90E+01	3.26E-16	2.04E-03	2.29E-02	1.66E-03	4.16E-03	5.26E-04	6.98E-01	2.06E-01	4.50E-02	1.97E-02
2.00E+01	3.26E-16	2.05E-03	2.30E-02	1.67E-03	4.17E-03	5.27E-04	6.96E-01	2.03E-01	4.34E-02	2.58E-02
2.10E+01	3.19E-16	2.18E-03	2.38E-02	1.72E-03	4.28E-03	5.40E-04	7.08E-01	1.95E-01	3.74E-02	2.65E-02
2.20E+01	1.77E-22	2.69E-03	2.87E-02	2.01E-03	5.15E-03	7.00E-04	6.07E-01	2.29E-01	5.61E-02	6.97E-02
2.30E+01	1.63E-28	2.26E-04	1.39E-02	8.72E-04	3.83E-03	3.63E-04	6.71E-01	2.06E-01	4.73E-02	5.70E-02
2.40E+01	1.54E-34	1.09E-05	4.74E-03	2.72E-04	1.96E-03	1.28E-04	6.99E-01	1.98E-01	4.41E-02	5.23E-02
2.50E+01	1.47E-40	4.52E-07	1.47E-03	7.75E-05	9.05E-04	4.03E-05	7.08E-01	1.95E-01	4.32E-02	5.09E-02
2.60E+01	1.41E-46	1.80E-08	4.48E-04	2.16E-05	4.06E-04	1.24E-05	7.11E-01	1.95E-01	4.29E-02	5.05E-02
2.70E+01	1.36E-52	7.06E-10	1.35E-04	6.00E-06	1.81E-04	3.77E-06	7.12E-01	1.94E-01	4.28E-02	5.04E-02
2.80E+01	1.31E-58	2.76E-11	4.08E-05	1.66E-06	8.06E-05	1.15E-06	7.12E-01	1.94E-01	4.28E-02	5.03E-02
2.90E+01	1.27E-64	1.08E-12	1.23E-05	4.61E-07	3.59E-05	3.49E-07	7.13E-01	1.94E-01	4.28E-02	5.03E-02
3.00E+01	1.24E-70	4.22E-14	3.72E-06	1.28E-07	1.60E-05	1.06E-07	7.13E-01	1.94E-01	4.28E-02	5.03E-02
3.10E+01	1.21E-76	1.65E-15	1.12E-06	3.56E-08	7.10E-06	3.23E-08	7.13E-01	1.94E-01	4.27E-02	5.03E-02
3.20E+01	1.18E-82	6.45E-17	3.39E-07	9.89E-09	3.16E-06	9.84E-09	7.13E-01	1.94E-01	4.27E-02	5.02E-02
3.30E+01	1.16E-88	2.52E-18	1.03E-07	2.75E-09	1.41E-06	3.00E-09	7.13E-01	1.94E-01	4.27E-02	5.02E-02
3.40E+01	1.14E-94	9.87E-20	3.11E-08	7.67E-10	6.29E-07	9.15E-10	7.13E-01	1.94E-01	4.27E-02	5.02E-02
3.50E+01	1.12E-100	3.86E-21	9.41E-09	2.14E-10	2.81E-07	2.79E-10	7.13E-01	1.94E-01	4.27E-02	5.02E-02
3.60E+01	1.11E-106	9.95E-23	2.85E-09	5.97E-11	1.25E-07	8.52E-11	7.13E-01	1.94E-01	4.27E-02	5.02E-02
3.70E+01	1.10E-112	2.56E-24	8.66E-10	1.67E-11	5.59E-08	2.60E-11	7.13E-01	1.94E-01	4.27E-02	5.02E-02
3.80E+01	1.09E-118	6.61E-26	2.63E-10	4.66E-12	2.50E-08	7.96E-12	7.13E-01	1.94E-01	4.27E-02	5.02E-02
3.90E+01	1.09E-124	1.71E-27	8.00E-11	1.30E-12	1.12E-08	2.44E-12	7.13E-01	1.94E-01	4.27E-02	5.01E-02
4.00E+01	1.09E-130	4.40E-29	2.43E-11	3.65E-13	5.00E-09	7.46E-13	7.13E-01	1.94E-01	4.27E-02	5.01E-02
4.10E+01	1.09E-136	1.14E-30	7.41E-12	1.02E-13	2.24E-09	2.29E-13	7.13E-01	1.94E-01	4.26E-02	5.01E-02
4.20E+01	1.09E-142	2.93E-32	2.26E-12	2.87E-14	1.00E-09	7.01E-14	7.14E-01	1.94E-01	4.26E-02	5.01E-02
4.30E+01	1.10E-148	7.56E-34	6.89E-13	8.07E-15	4.49E-10	2.15E-14	7.14E-01	1.94E-01	4.26E-02	5.01E-02
4.40E+01	1.11E-154	1.95E-35	2.10E-13	2.27E-15	2.02E-10	6.60E-15	7.14E-01	1.94E-01	4.26E-02	5.01E-02
4.50E+01	1.12E-160	5.04E-37	6.42E-14	6.38E-16	9.04E-11	2.03E-15	7.14E-01	1.94E-01	4.26E-02	5.01E-02
4.60E+01	1.13E-166	1.30E-38	1.96E-14	1.80E-16	4.06E-11	6.23E-16	7.14E-01	1.94E-01	4.26E-02	5.00E-02

Stage	Hydrogen	Methane	Ethane	Ethylene	Propane	Propylene	Benzene	Toluene	Xylene	Trimethylbenzene
4.70E+01	1.15E-172	3.36E-40	6.01E-15	5.07E-17	1.82E-11	1.92E-16	7.14E-01	1.94E-01	4.26E-02	5.00E-02
4.80E+01	1.17E-178	8.68E-42	1.84E-15	1.43E-17	8.19E-12	5.89E-17	7.14E-01	1.94E-01	4.26E-02	5.00E-02
4.90E+01	1.20E-184	2.24E-43	5.64E-16	4.04E-18	3.68E-12	1.82E-17	7.14E-01	1.94E-01	4.26E-02	5.00E-02
5.00E+01	1.22E-190	5.79E-45	1.73E-16	1.14E-18	1.66E-12	5.59E-18	7.14E-01	1.93E-01	4.26E-02	5.00E-02
5.10E+01	1.26E-196	1.50E-46	5.31E-17	3.23E-19	7.46E-13	1.72E-18	7.14E-01	1.93E-01	4.26E-02	5.00E-02
5.20E+01	1.29E-202	3.87E-48	1.63E-17	9.15E-20	3.36E-13	5.32E-19	7.14E-01	1.93E-01	4.25E-02	5.00E-02
5.30E+01	1.33E-208	1.00E-49	5.01E-18	2.59E-20	1.51E-13	1.64E-19	7.14E-01	1.93E-01	4.25E-02	4.99E-02
5.40E+01	1.38E-214	2.58E-51	1.54E-18	7.35E-21	6.82E-14	5.08E-20	7.14E-01	1.94E-01	4.25E-02	4.99E-02
5.50E+01	1.43E-220	6.68E-53	4.75E-19	1.51E-21	3.08E-14	1.57E-20	7.14E-01	1.94E-01	4.25E-02	4.99E-02
5.60E+01	1.48E-226	1.73E-54	1.46E-19	3.09E-22	1.39E-14	4.85E-21	7.14E-01	1.94E-01	4.25E-02	4.99E-02
5.70E+01	1.54E-232	4.47E-56	4.51E-20	6.35E-23	6.27E-15	1.10E-21	7.13E-01	1.94E-01	4.26E-02	4.99E-02
5.80E+01	1.61E-238	1.15E-57	1.39E-20	1.30E-23	2.83E-15	2.47E-22	7.12E-01	1.95E-01	4.26E-02	4.99E-02
5.90E+01	1.69E-244	2.98E-59	4.30E-21	2.68E-24	1.28E-15	5.59E-23	7.11E-01	1.96E-01	4.27E-02	5.00E-02
6.00E+01	1.78E-250	7.69E-61	9.68E-22	5.52E-25	5.76E-16	1.27E-23	7.09E-01	1.98E-01	4.29E-02	5.01E-02
6.10E+01	1.88E-256	1.98E-62	2.18E-22	1.14E-25	2.59E-16	2.87E-24	7.04E-01	2.02E-01	4.34E-02	5.04E-02
6.20E+01	2.00E-262	5.07E-64	4.89E-23	2.33E-26	1.16E-16	6.50E-25	6.97E-01	2.08E-01	4.43E-02	5.11E-02
6.30E+01	2.14E-268	1.29E-65	1.09E-23	4.78E-27	5.12E-17	1.47E-25	6.84E-01	2.18E-01	4.62E-02	5.26E-02
6.40E+01	2.33E-274	3.23E-67	2.39E-24	9.62E-28	2.20E-17	3.29E-26	6.60E-01	2.33E-01	5.02E-02	5.64E-02
6.50E+01	2.67E-280	7.40E-69	4.71E-25	1.76E-28	6.28E-18	6.83E-27	5.58E-01	2.76E-01	7.27E-02	9.33E-02

A-5.4.3. Vapor Mass Composition Profile for COL-301

Stage	Hydrogen	Methane	Ethane	Ethylene	Propane	Propylene	Benzene	Toluene	Xylene	Trimethylbenzene
1.00E+00	5.04E-09	1.33E-01	3.44E-01	2.97E-02	3.48E-02	7.00E-03	4.09E-01	4.08E-02	1.16E-03	1.27E-06
2.00E+00	4.57E-09	1.21E-01	3.14E-01	2.71E-02	3.19E-02	6.39E-03	4.45E-01	5.26E-02	1.90E-03	2.56E-06
3.00E+00	4.55E-09	1.20E-01	3.13E-01	2.70E-02	3.18E-02	6.37E-03	4.43E-01	5.60E-02	2.48E-03	4.32E-06
4.00E+00	4.56E-09	1.20E-01	3.13E-01	2.71E-02	3.18E-02	6.38E-03	4.41E-01	5.72E-02	2.97E-03	6.82E-06
5.00E+00	4.56E-09	1.20E-01	3.14E-01	2.71E-02	3.18E-02	6.38E-03	4.40E-01	5.76E-02	3.41E-03	1.04E-05
6.00E+00	4.56E-09	1.20E-01	3.14E-01	2.71E-02	3.19E-02	6.38E-03	4.39E-01	5.76E-02	3.78E-03	1.54E-05
7.00E+00	4.56E-09	1.21E-01	3.14E-01	2.71E-02	3.19E-02	6.38E-03	4.39E-01	5.75E-02	4.09E-03	2.25E-05

Stage	Hydrogen	Methane	Ethane	Ethylene	Propane	Propylene	Benzene	Toluene	Xylene	Trimethylbenzene
8.00E+00	4.56E-09	1.21E-01	3.14E-01	2.71E-02	3.19E-02	6.39E-03	4.38E-01	5.74E-02	4.36E-03	3.25E-05
9.00E+00	4.57E-09	1.21E-01	3.14E-01	2.71E-02	3.19E-02	6.39E-03	4.38E-01	5.72E-02	4.58E-03	4.65E-05
1.00E+01	4.57E-09	1.21E-01	3.14E-01	2.71E-02	3.19E-02	6.39E-03	4.38E-01	5.71E-02	4.77E-03	6.61E-05
1.10E+01	4.57E-09	1.21E-01	3.14E-01	2.71E-02	3.19E-02	6.39E-03	4.38E-01	5.70E-02	4.92E-03	9.36E-05
1.20E+01	4.57E-09	1.21E-01	3.14E-01	2.71E-02	3.19E-02	6.39E-03	4.37E-01	5.69E-02	5.03E-03	1.32E-04
1.30E+01	4.57E-09	1.21E-01	3.14E-01	2.71E-02	3.19E-02	6.40E-03	4.37E-01	5.68E-02	5.12E-03	1.85E-04
1.40E+01	4.57E-09	1.21E-01	3.14E-01	2.71E-02	3.19E-02	6.40E-03	4.37E-01	5.67E-02	5.19E-03	2.59E-04
1.50E+01	4.57E-09	1.21E-01	3.15E-01	2.72E-02	3.19E-02	6.40E-03	4.37E-01	5.66E-02	5.23E-03	3.61E-04
1.60E+01	4.58E-09	1.21E-01	3.15E-01	2.72E-02	3.19E-02	6.40E-03	4.37E-01	5.64E-02	5.25E-03	5.01E-04
1.70E+01	4.58E-09	1.21E-01	3.15E-01	2.72E-02	3.20E-02	6.40E-03	4.37E-01	5.63E-02	5.25E-03	6.94E-04
1.80E+01	4.58E-09	1.21E-01	3.15E-01	2.72E-02	3.20E-02	6.41E-03	4.36E-01	5.61E-02	5.23E-03	9.56E-04
1.90E+01	4.58E-09	1.21E-01	3.15E-01	2.72E-02	3.20E-02	6.41E-03	4.36E-01	5.60E-02	5.19E-03	1.31E-03
2.00E+01	4.58E-09	1.21E-01	3.15E-01	2.72E-02	3.20E-02	6.41E-03	4.36E-01	5.59E-02	5.16E-03	1.80E-03
2.10E+01	4.53E-09	1.20E-01	3.12E-01	2.69E-02	3.17E-02	6.34E-03	4.38E-01	5.72E-02	5.42E-03	2.60E-03
2.20E+01	1.83E-15	7.51E-02	2.55E-01	2.08E-02	3.02E-02	5.66E-03	5.22E-01	7.75E-02	8.76E-03	5.47E-03
2.30E+01	6.37E-22	9.72E-03	1.03E-01	7.26E-03	1.86E-02	2.53E-03	7.32E-01	1.06E-01	1.26E-02	8.26E-03
2.40E+01	3.77E-28	5.21E-04	3.20E-02	2.01E-03	8.85E-03	8.38E-04	8.17E-01	1.15E-01	1.41E-02	9.47E-03
2.50E+01	3.14E-34	2.22E-05	9.65E-03	5.53E-04	3.99E-03	2.60E-04	8.44E-01	1.17E-01	1.45E-02	9.83E-03
2.60E+01	2.89E-40	8.89E-07	2.90E-03	1.52E-04	1.78E-03	7.93E-05	8.53E-01	1.18E-01	1.46E-02	9.95E-03
2.70E+01	2.74E-46	3.50E-08	8.73E-04	4.21E-05	7.91E-04	2.41E-05	8.56E-01	1.18E-01	1.47E-02	9.99E-03
2.80E+01	2.63E-52	1.37E-09	2.63E-04	1.16E-05	3.52E-04	7.32E-06	8.57E-01	1.18E-01	1.47E-02	1.00E-02
2.90E+01	2.55E-58	5.36E-11	7.91E-05	3.22E-06	1.56E-04	2.22E-06	8.57E-01	1.18E-01	1.47E-02	1.00E-02
3.00E+01	2.47E-64	2.09E-12	2.38E-05	8.94E-07	6.94E-05	6.76E-07	8.57E-01	1.18E-01	1.47E-02	1.00E-02
3.10E+01	2.40E-70	8.18E-14	7.19E-06	2.48E-07	3.09E-05	2.05E-07	8.57E-01	1.18E-01	1.47E-02	1.00E-02
3.20E+01	2.34E-76	3.19E-15	2.17E-06	6.88E-08	1.37E-05	6.25E-08	8.57E-01	1.18E-01	1.47E-02	1.00E-02
3.30E+01	2.29E-82	1.25E-16	6.56E-07	1.91E-08	6.12E-06	1.90E-08	8.57E-01	1.18E-01	1.47E-02	1.00E-02
3.40E+01	2.24E-88	4.88E-18	1.98E-07	5.32E-09	2.73E-06	5.80E-09	8.57E-01	1.18E-01	1.47E-02	1.00E-02
3.50E+01	2.21E-94	1.91E-19	6.00E-08	1.48E-09	1.22E-06	1.77E-09	8.57E-01	1.18E-01	1.47E-02	1.00E-02
3.60E+01	2.17E-100	7.45E-21	1.82E-08	4.13E-10	5.42E-07	5.39E-10	8.57E-01	1.18E-01	1.47E-02	1.00E-02
3.70E+01	2.15E-106	1.92E-22	5.51E-09	1.15E-10	2.42E-07	1.65E-10	8.57E-01	1.18E-01	1.47E-02	1.01E-02

Stage	Hydrogen	Methane	Ethane	Ethylene	Propane	Propylene	Benzene	Toluene	Xylene	Trimethylbenzene
3.80E+01	2.13E-112	4.95E-24	1.67E-09	3.22E-11	1.08E-07	5.03E-11	8.57E-01	1.18E-01	1.47E-02	1.01E-02
3.90E+01	2.11E-118	1.28E-25	5.08E-10	8.99E-12	4.82E-08	1.54E-11	8.57E-01	1.18E-01	1.47E-02	1.01E-02
4.00E+01	2.10E-124	3.29E-27	1.54E-10	2.51E-12	2.16E-08	4.70E-12	8.57E-01	1.18E-01	1.47E-02	1.01E-02
4.10E+01	2.10E-130	8.48E-29	4.69E-11	7.04E-13	9.64E-09	1.44E-12	8.57E-01	1.18E-01	1.48E-02	1.01E-02
4.20E+01	2.10E-136	2.19E-30	1.43E-11	1.97E-13	4.31E-09	4.40E-13	8.57E-01	1.18E-01	1.48E-02	1.01E-02
4.30E+01	2.10E-142	5.64E-32	4.35E-12	5.53E-14	1.93E-09	1.35E-13	8.57E-01	1.18E-01	1.48E-02	1.01E-02
4.40E+01	2.11E-148	1.46E-33	1.33E-12	1.55E-14	8.65E-10	4.14E-14	8.57E-01	1.18E-01	1.48E-02	1.01E-02
4.50E+01	2.13E-154	3.76E-35	4.05E-13	4.37E-15	3.88E-10	1.27E-14	8.57E-01	1.18E-01	1.48E-02	1.01E-02
4.60E+01	2.15E-160	9.69E-37	1.24E-13	1.23E-15	1.74E-10	3.90E-15	8.57E-01	1.18E-01	1.48E-02	1.01E-02
4.70E+01	2.18E-166	2.50E-38	3.78E-14	3.46E-16	7.81E-11	1.20E-15	8.57E-01	1.18E-01	1.48E-02	1.01E-02
4.80E+01	2.21E-172	6.46E-40	1.16E-14	9.74E-17	3.51E-11	3.68E-16	8.57E-01	1.18E-01	1.48E-02	1.01E-02
4.90E+01	2.25E-178	1.67E-41	3.54E-15	2.75E-17	1.57E-11	1.13E-16	8.57E-01	1.18E-01	1.48E-02	1.01E-02
5.00E+01	2.30E-184	4.31E-43	1.08E-15	7.76E-18	7.08E-12	3.49E-17	8.57E-01	1.18E-01	1.48E-02	1.01E-02
5.10E+01	2.35E-190	1.11E-44	3.32E-16	2.19E-18	3.18E-12	1.07E-17	8.57E-01	1.18E-01	1.48E-02	1.01E-02
5.20E+01	2.41E-196	2.87E-46	1.02E-16	6.20E-19	1.43E-12	3.31E-18	8.57E-01	1.18E-01	1.48E-02	1.01E-02
5.30E+01	2.48E-202	7.42E-48	3.13E-17	1.76E-19	6.45E-13	1.02E-18	8.57E-01	1.18E-01	1.48E-02	1.01E-02
5.40E+01	2.56E-208	1.92E-49	9.62E-18	4.97E-20	2.90E-13	3.15E-19	8.57E-01	1.18E-01	1.48E-02	1.01E-02
5.50E+01	2.64E-214	4.96E-51	2.96E-18	1.41E-20	1.31E-13	9.73E-20	8.57E-01	1.18E-01	1.48E-02	1.01E-02
5.60E+01	2.74E-220	1.28E-52	9.11E-19	2.89E-21	5.90E-14	3.01E-20	8.56E-01	1.19E-01	1.48E-02	1.02E-02
5.70E+01	2.84E-226	3.31E-54	2.81E-19	5.92E-22	2.66E-14	9.30E-21	8.56E-01	1.19E-01	1.49E-02	1.02E-02
5.80E+01	2.96E-232	8.56E-56	8.65E-20	1.22E-22	1.20E-14	2.10E-21	8.55E-01	1.20E-01	1.49E-02	1.02E-02
5.90E+01	3.09E-238	2.21E-57	2.67E-20	2.50E-23	5.41E-15	4.74E-22	8.53E-01	1.21E-01	1.51E-02	1.03E-02
6.00E+01	3.24E-244	5.70E-59	8.23E-21	5.13E-24	2.44E-15	1.07E-22	8.51E-01	1.24E-01	1.53E-02	1.04E-02
6.10E+01	3.40E-250	1.47E-60	1.85E-21	1.05E-24	1.10E-15	2.42E-23	8.46E-01	1.28E-01	1.58E-02	1.07E-02
6.20E+01	3.59E-256	3.78E-62	4.15E-22	2.17E-25	4.89E-16	5.47E-24	8.37E-01	1.35E-01	1.67E-02	1.14E-02
6.30E+01	3.81E-262	9.66E-64	9.29E-23	4.44E-26	2.15E-16	1.23E-24	8.22E-01	1.46E-01	1.86E-02	1.28E-02
6.40E+01	4.07E-268	2.45E-65	2.04E-23	8.94E-27	9.17E-17	2.74E-25	7.96E-01	1.65E-01	2.23E-02	1.59E-02
6.50E+01	4.45E-274	6.10E-67	4.14E-24	1.68E-27	3.63E-17	5.67E-26	7.53E-01	1.95E-01	2.96E-02	2.28E-02

A-5.4.4. TPFQ Profile for COL-302

Stag	Temperat	Pressu		Liquid from	Vapor from	1	Vapor	Liquid flow	Vapor flow
e	ure F	re	Heat duty	(Mass)	(Mass)	enthalpy	enthalpy	(Mass)	(Mass)
1	_	psia	Btu/hr	lb/hr	lb/hr	Btu/lbmol	Btu/lbmol	lb/hr	lb/hr
1	196	16.0	-33077660	187342) 19206	34800	73224	0
2	241	26.0	0	80177	18734		33997	80177	187342
3	251	26.2	0	80163	19429		32850	80163	194294
4	258	26.3	0	80114	19428		32115	80114	194281
5	262	26.5	0	80008	19423		31673	80008	194232
6	265	26.6	0	79878	19412		31406	79878	194126
7	267	26.8	0	79739	19399		31239	79739	193996
8	268	26.9	0	79595	19385	7 10938	31130	79595	193857
9	270	27.1	0	79440	19371	3 10698	31057	79440	193713
10	271	27.2	0	79269	19355	8 10483	31003	79269	193558
11	272	27.4	0	79073	19338	7 10268	30958	79073	193387
12	273	27.5	0	78846	19319	1 10028	30916	78846	193191
13	274	27.7	0	78578	19296	3 9744	30869	78578	192963
14	275	27.8	0	78266	19269	5 9400	30815	78266	192696
15	276	28.0	0	77905	19238	4 8984	30748	77905	192384
16	278	28.1	0	77499	19202	3 8490	30667	77499	192023
17	280	28.3	0	77053	19161	7 7919	30570	77053	191617
18	281	28.4	0	76584	19117	1 7282	30458	76584	191171
19	284	28.6	0	76109	19070	1 6601	30334	76109	190701
20	286	28.7	0	75663	19022	7 5906	30202	75663	190227
21	288	28.9	0	75506	18978	1 5237	30066	75506	189781
22	289	29.0	0	80889	18962	3 4741	29908	80889	189623
23	288	29.2	0	163448	15040	3 5160	26970	163448	150403
24	315	29.3	0	169477	15173		21912	169477	151735
25	337	29.5	0	176292	15776		17144	176292	157765
26	352	29.6	0	181972	16457		13515	181972	164579
27	363	29.8	0	185912	17025		11098	185912	170259

Stag	Temperat	Pressu		Liquid from	Vapor from	Liquid	Vapor	Liquid flow	Vapor flow
e	ure	re	Heat duty	(Mass)	(Mass)	enthalpy	enthalpy	(Mass)	(Mass)
28	369	29.9	0	188453	174199	-7543	9561	188453	174199
29	374	30.1	0	190093	176740	-8118	8581	190093	176740
30	377	30.2	0	191189	178381	-8503	7941	191189	178381
31	379	30.4	0	191953	179476	-8764	7513	191953	179476
32	380	30.5	0	192507	180240	-8940	7223	192507	180240
33	382	30.7	0	192923	180794	-9058	7025	192923	180794
34	383	30.8	0	193248	181211	-9134	6891	193248	181211
35	384	31.0	0	193508	181535	-9182	6800	193508	181535
36	384	31.1	0	193722	181795	-9208	6741	193722	181795
37	385	31.3	0	193905	182010	-9219	6704	193905	182010
38	385	31.4	0	194065	182193	-9220	6683	194065	182193
39	386	31.6	0	194208	182352	-9213	6673	194208	182352
40	386	31.7	0	194340	182496	-9201	6672	194340	182496
41	387	31.9	0	194453	182627	-9185	6676	194453	182627
42	387	32.0	24137701	11713	182741	-9168	6686	11713	182741

<u>A-5.4.5.1</u> Stage	Benzene	Toluene	Profile for COL- Xylene	<u>Trimethylbenzene</u>
1	0.616	0.304	0.080	0.000
2	0.420	0.399	0.180	0.001
3	0.314	0.416	0.267	0.002
4	0.263	0.398	0.335	0.002
5	0.238	0.372	0.384	0.004
6	0.236	0.348	0.418	0.009
7	0.229	0.329	0.440	0.012
8	0.210	0.316	0.453	0.012
9	0.211	0.306	0.460	0.024
10	0.208	0.298	0.461	0.032
11	0.206	0.290	0.459	0.042
12	0.200	0.287	0.452	0.056
13	0.204	0.282	0.443	0.073
13	0.202	0.276	0.430	0.095
15	0.197	0.270	0.413	0.120
16	0.193	0.262	0.393	0.151
17	0.189	0.254	0.370	0.186
18	0.185	0.245	0.345	0.224
19	0.180	0.236	0.318	0.265
20	0.176	0.227	0.291	0.306
21	0.171	0.218	0.265	0.346
22	0.169	0.211	0.241	0.378
23	0.170	0.234	0.225	0.371
24	0.075	0.173	0.252	0.500
25	0.028	0.108	0.243	0.620
26	0.010	0.061	0.211	0.718
27	0.003	0.032	0.172	0.793
28	0.001	0.016	0.134	0.849
29	0.000	0.008	0.102	0.890
30	0.000	0.004	0.076	0.920
31	0.000	0.002	0.056	0.942
32	0.000	0.001	0.041	0.958
33	0.000	0.000	0.030	0.970
34	0.000	0.000	0.022	0.978
35	0.000	0.000	0.016	0.984
36	0.000	0.000	0.011	0.989
37	0.000	0.000	0.008	0.992
38	0.000	0.000	0.006	0.994
39	0.000	0.000	0.004	0.996
40	0.000	0.000	0.003	0.997
41	0.000	0.000	0.002	0.998

A-5.4.5. Liquid Mass Composition Profile for COL-302

S	tage	Benzene	Toluene	Xylene	Trimethylbenzene
	42	0.000	0.000	0.001	0.999

A-5.4.6. Vapor Mass Composition Profile for COL-302

Stage	Benzene	Toluene	Xylene	Trimethylbenzene
1	0.744	0.216	0.040	0.000
2	0.616	0.304	0.080	0.000
3	0.535	0.343	0.121	0.001
4	0.491	0.350	0.157	0.001
5	0.470	0.343	0.185	0.002
6	0.460	0.332	0.205	0.003
7	0.455	0.322	0.219	0.004
8	0.452	0.314	0.228	0.005
9	0.451	0.309	0.233	0.007
10	0.449	0.305	0.236	0.010
11	0.449	0.302	0.236	0.013
12	0.448	0.299	0.235	0.018
13	0.448	0.297	0.232	0.023
14	0.447	0.295	0.228	0.030
15	0.447	0.293	0.222	0.039
16	0.446	0.290	0.215	0.049
17	0.445	0.287	0.207	0.061
18	0.444	0.284	0.197	0.075
19	0.443	0.280	0.187	0.090
20	0.442	0.277	0.175	0.106
21	0.440	0.273	0.164	0.122
22	0.439	0.270	0.154	0.138
23	0.342	0.279	0.181	0.198
24	0.183	0.252	0.243	0.322
25	0.080	0.185	0.271	0.463
26	0.030	0.116	0.260	0.593
27	0.010	0.065	0.226	0.699
28	0.003	0.034	0.183	0.779
29	0.001	0.017	0.143	0.839
30	0.000	0.008	0.108	0.883
31	0.000	0.004	0.081	0.915
32	0.000	0.002	0.059	0.939
33	0.000	0.001	0.043	0.956
34	0.000	0.000	0.032	0.968
35	0.000	0.000	0.023	0.977
36	0.000	0.000	0.016	0.983
37	0.000	0.000	0.012	0.988

Stage	Benzene	Toluene	Xylene	Trimethylbenzene
38	0.000	0.000	0.008	0.992
39	0.000	0.000	0.006	0.994
40	0.000	0.000	0.004	0.996
41	0.000	0.000	0.003	0.997
42	0.000	0.000	0.002	0.998

A-6. Consultant E-mail Correspondence

A-6.1. Ms. Karin Fair, Air Products and Chemicals, Inc.

------ Forwarded message ------From: **Fair,Karin Aurora** <<u>FAIRKA@airproducts.com</u>> Date: Mon, Mar 23, 2015 at 10:41 PM Subject: RE: PRISM Separation To: Arthur Chen <<u>arthurchen24@gmail.com</u>>

Hello, Arthur.

Answers are in your e-mail below.

Regards, Karin Fair

From: Arthur Chen [mailto:arthurchen24@gmail.com]
Sent: Sunday, March 22, 2015 1:08 PM
To: Fair,Karin Aurora
Subject: Re: PRISM Separation

Hi Karin,

Thank you so much for the help. We just had a couple of followup questions as we complete our project.

1) Are there any temperature considerations for the separation at 1200 psig?

With operating at either feed pressure of 700 psig or 1200 psi, typical operating pressure for feed gas containing H2 and hydrocarbons is 83°C. Typical pretreatment is cooler (if feed gas temperature is more than about 45°C) to cool to 40°C, mist eliminator to remove condensed liquids, and feed heater to superheat the feed gas to 83°C.

2) What are the outlet pressures for the permeate and diffusate in the same 1200 psig case?

The permeate (H2-rich / fast gas) stream is 200 psig. The non-permeate (HC-rich / slower gas) stream would be at about 1150 psig and then can be let down to a lower pressure if necessary. 3) Is PRISM suitable for a second further separation of methane? Such a separation would be looking to achieve ethane purity.

Are you asking if the membrane can be used to separate methane from ethane and the other HC's? If so, the answer is no. There is not enough difference between the C1 and C2 component permeabilities to make that an effective separation.

4) Lastly, about how much would such a piece of equipment cost?

I think the MW of the feed gas is 16.21 kg/kgmole. So for 135,000 kg/hr feed flow rate, that means 8333 kmol/hr or 167 MMSCFD. This is a large feed flow, and given the low purity feed gas (only 42 mol% H2), I'm not so sure that a membrane is the most economic solution. For 8333 kmol/hr, 700 psig feed, 200 psig permeate, a rough cost for the membrane system is \$17MM (17 million dollars).

Also some added information: we are using a feed flow rate of 1.35 x 10⁵ kg/hr and a pressure of 700 psig.

Thank you in advance for your kind advice.

Best, Arthur

On Mon, Mar 2, 2015 at 11:33 PM, Fair,Karin Aurora <<u>FAIRKA@airproducts.com</u>> wrote: Hello,

The higher level of aromatics in the right-hand composition could be problematic, so would likely be better for the membrane to have heavy HC's removed upstream of the membranes. Here are some general comments:

- Membranes use partial pressure as the driving force for the separation. Feed is at higher pressure. Permeate (H2-rich) is at lower pressure.

- If feed is at 700 psig and permeate is at 200 psig, with targeting 85% recovery, expected permeate purity is about 72 mol% H2.

- If feed is at 1200 psig and permeate is at 200 psig, with targeting 85% recovery, expected permeate purity is about 81 mol% H2.

- Higher H2 recovery results in lower H2 purity permeate.

You didn't specify either a H2 recovery or a H2 purity that you're looking to achieve, so not sure if the recovery and/or purity that I report above is close to what you're expecting.

-Karin

From: Arthur Chen [mailto:arthurchen24@gmail.com]
Sent: Monday, March 02, 2015 3:57 PM
To: Fair,Karin Aurora
Subject: Re: PRISM Separation

Thank you for your quick reply.

Depending on the placement of the membrane separation in our process, there are two possible feed compositions (on left: after separation of heavy hydrocarbons; on right: before separation).

Mole Frac		
HYDROGEN	0.421367	0.3868498
METHANE	0.1507092	0.139271
ETHANE	0.3714968	0.3478525
ETHYLENE	0.0422207	0.0393063

PROPANE	0.00977494	0.010147
PROPYLENE	0.0034205	0.00329065
BENZENE	0.000996064	0.0453434
TOLUENE	1.46E-05	0.0187711
P-XYLENE	9.53E-08	0.00429653
TRIMETHYLBENZENE	2.20E-09	0.0048715

The feed gas pressure and temperature are flexible, as we have yet to set those conditions in stone. The same goes for the permeate stream. As for the performance we wish to achieve, it is H2 recovery.

Again, all your help is deeply appreciated.

Best, Arthur Chen

On Mon, Mar 2, 2015 at 11:04 AM, Fair,Karin Aurora <<u>FAIRKA@airproducts.com</u>> wrote: Hello,

I have some questions in order to assess if a membrane is a viable technology for you.

• Can you please provide a more complete feed composition? The composition you provided (H2, C1, other products) only adds up to 0.882.

- What is the feed flow rate?
- What is the feed gas pressure?
- What pressure do you need the permeate (H2-rich) stream?
- What performance are you wanting to achieve either H2 recovery and/or permeate H2 purity?

Regards, Karin Fair

From: Arthur Chen [mailto:arthurchen24@gmail.com]
Sent: Monday, March 02, 2015 10:35 AM
To: Fair,Karin Aurora
Subject: PRISM Separation

Dear Karin Fair,

Good morning. I am writing to you on behalf of my chemical engineering senior design team. We received your contact information from our classmates on another team. Our project has some similarities to theirs, namely in that we have a stream with a mixture of hydrogen and light hydrocarbons from which we wish to separate out the hydrogen for sale or other use. Specifically, our stream had a composition of 0.389 hydrogen, 0.130 methane, and 0.363 plus small quantities of other products.

We are hoping that you could provide some insight as to what specific PRISM conditions we would theoretically require to separate out our hydrogen, and how much that would cost. Furthermore, we are curious if PRISM can separate methane as well. Any suggestions or insight you can provide would be greatly appreciated.

Best regards, Arthur Chen

A-6.2. Ms. Sasha Maitala, Jacobs UK

------ Forwarded message ------From: "Maitala, Sasha" <<u>Sasha.Maitala@jacobs.com</u>> Date: Mon, 30 Mar 2015 16:40:37 +0100 Subject: Steam Reformer Reactor To: "<u>crowleyf@seas.upenn.edu</u>" <<u>crowleyf@seas.upenn.edu</u>>

Hi Fiona,

I have been asked to provide you with some information for your senior design project.

If you go with a steam reformer type reactor unit you should know that it has a significant amount of heat that can and should be recovered from the flue gas. The temperature you have specified is a little low for typical steam reformer duties but a solution could be easily found.

The total installed cost of a steam reformer with the absorbed heat duty you have quoted is in the region of 55 million USD. The price includes the heat recovery section of the reactor.

The footprint of the reactor with the heat recovery section and all civil/structural works is: 29 m Length 18 m Width 32 m Height to top of penthouse roof (not including stack height)

Attached is a picture of an existing reformer. Your unit will be roughly twice the twice of that one.

Hope you find this information helpful,

Regards, Sasha

Sasha Maitala | Jacobs | Process Engineer | <u>+44 (0) 20 7378 2841</u> | sasha.maitala@jacobs.com<mailto:sasha.maitala@jacobs.com> | www.jacobs.com <http://www.jacobs.com/ NOTICE - This communication may contain confidential and privileged information that is for the sole use of the intended recipient. Any viewing, copying or distribution of, or reliance on this message by unintended recipients is strictly prohibited. If you have received this message in error, please notify us immediately by replying to the message and deleting it from your computer.

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A-6.3. Dr. Richard Bockrath, DuPont

------ Forwarded message ------From: **Richard Bockrath** <<u>richardbockrath@gmail.com</u>> Date: Wed, Apr 1, 2015 at 6:36 AM Subject: Re: BTX from Ethane: Steam Boiler To: Arthur Chen <<u>arthurchen24@gmail.com</u>>

Attached is the boiler info I have. The boiler is sized for about 50% of your steam generation rates so you would need two of these boilers. We had talked about the value of having two boilers anyways and so this is probably a good size for your work. Our engineering firm said this was "a small boiler for a refinery or petrochemical plant". The price was for January, 2014 and so you would need to adjust it for inflation from then to when you would buy it for your plant. The pressure rating is lower than you desire. I am not sure if the database you are currently using shows the price impact of different output pressures in its tables. If it doesn't then I would just add 10% to the price in the attachment for higher output pressure.

Hope this helps. Let me know if you need more.

Rick

On Tue, Mar 31, 2015 at 10:25 PM, Arthur Chen <<u>arthurchen24@gmail.com</u>> wrote: Hi Dr. Bockrath,

Just wanted to follow up on our meeting this afternoon. We mentioned that we needed a steam boiler that could handle a heat requirement of 254 million BTU/hr. You mentioned that you were working with on a project with a similar large steam boiler and may have some information on the sizing and cost of such a unit.

Thanks, Arthur

15. Steam Boiler

The steam demand requirements were based on the Processium energy balance and the design basis is listed in Table 22. The assumed steam pressure is 100 psig to provide sufficient temperature differential to meet maximum reboiler temperatures. The temperatures in the Processium material balance are subject to change based on results during pilot plant operation.

One vendor was contacted for a budgetary estimate for the steam boiler and a summary of the proposal is provided in Table 23.

Parameter	Value	Comments
Steam Supply Pressure	100 psig	Based on maximum estimated
		reboiler temperatures in
		Processium energy balance
Steam Demand	165,000 lb/hr	Based on Processium steady
		state energy balance and GBE
		fermentation CIP process

Table 22. Steam Boiler Design Basis

Vendor	Babcock and Wilcox
Boiler Capacity	165,000 lb/hr
Produced Steam Pressure	100 psig
Produced Steam Temperature	Saturated
Boiler Package	FM120-112 Water Tube
Pricing FOB point of fabrication	\$1,387,300

Fisher Scientific sells at lab scale and so their price will be far too high. Alibaba is an industrial eBay for China. They always low ball their prices. I would pick something between the two. I would guess it is more like \$10-15/kg. Say \$12.5/kg.

Rick

On Apr 4, 2015 5:05 PM, "Jonathan Lym" <<u>ilym@seas.upenn.edu</u>> wrote: Hi Dr. Bockrath,

That's good news to hear! Thank you!

Sorry to barrage you with more questions but we were talking to another group about the price of ZSM-5 and found that we were using two completely different values.

Our estimate was \$92.2 per kg from Fisher Scientific but their group's estimate was \$4.9 per kg from Alibaba. Which cost do you think is more reasonable?

Thank you again for all the help you've provided.

Sincerely, Jon

Jonathan Lym

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On Fri, Apr 3, 2015 at 11:43 AM, Richard Bockrath <<u>richardbockrath@gmail.com</u>> wrote:

Praxair and Linde both have numerous pipelines in the Texas petrochemical hub. If you dig a hole in that area you will probably break a pipe! So I think it is quite reasonable to assume you can get hydrogen, N2 and O2 from them. Chlorine is stored in carbon steel pressurized tanks. Its vapor pressure is low enough that the pressure rating on the tank is very reasonable. This is Texas and so assume the tank will be exposed to the heat of the summer ie the pressure rating must be able to handle Cl2 at 50oC. So on site storage is the norm. People try to store as little Cl2 as possible due to permitting issues. So I would size the tank to either a) hold one smaller rail car of Cl2 (This would be about 80,000 lb of Cl2) or b) one week of usage whichever is larger.

Good luck and it has been a pleasure working with you all.

Rick

Hi Dr. Bockrath,

Firstly I want to say thank you so much for your help. We have made significant progress with your help.

We had found the flowrates required for the regeneration process but now our group was discussing the sources of the gases (i.e. chlorine, nitrogen, hydrogen and oxygen). We found <u>this website</u> on Praxair's website that says it has pipelines available on the Gulf Coast for nitrogen, hydrogen and oxygen.

Do you see any issues with using these pipelines for our process? Also we were planning on storing chlorine on site in a pressure vessel. Is this standard for an industrial plant?

Thank you again for the guidance!

Best, Jon --

Yes you should assume fractionation grade. I would just guess it is 1% methane, 97% ethane and 2% propane.

Rick

Hi Dr. Bockrath,

Sorry about the late response. I had fallen asleep. I had initially got that information from the <u>US Department of</u> <u>Transportation</u>

and some pages in a book called the Fundamentals of Natural Gas Processing. Upon further examination of the chapter, I found this is a figure that gives a more accurate composition of liquid products (see page 141 of the book available here

):

Could we then assume we're using Fractionation grade ethane?

Best, Jon