1A mathematical model of microbial folate biosynthesis and 2utilisation: implications for antifolate development

7J. Enrique Salcedo-Sora and Mark T. Mc Auley

10The metabolic biochemistry of folate biosynthesis and utilisation has evolved into a complex 11network of reactions. Although this complexity represents challenges to the field of folate 12 research it has also provided a renewed source for antimetabolite targets. A range of improved 13 folate chemotherapy continues to be developed and applied particularly to cancer and chronic 14inflammatory diseases. However, new or better antifolates against infectious diseases remain 15much more elusive. In this paper we describe the assembly of a generic deterministic 16mathematical model of microbial folate metabolism. Our aim is to explore how a 17mathematical model could be used to explore the dynamics of this inherently complex set of 18biochemical reactions. Using the model it was found that: (1) a particular small set of folate 19 intermediates are overrepresented, (2) inhibitory profiles can be quantified by the level of key 20folate products, (3) using the model to scan for the most effective combinatorial inhibitions of 21 folate enzymes we identified specific targets which could complement current antifolates, and 22(4) the model substantiates the case for a substrate cycle in the folinic acid biosynthesis 23reaction. Our model is coded in the systems biology markup language and has been deposited 24in the BioModels Database (MODEL1511020000), this makes it accessible to the community 25as a whole.

55

56

57

581 Introduction

59Infectious diseases are still a major burden to human health and economic development. For 60example, in 2013 mortality due to tuberculosis 2013 was estimated at 1.4 million people.1 61Moreover, Malaria causes an astonishing 200 to 500 million of clinical episodes a year2–4 62with nearly 600 thousand deaths in 2013.4 Folate metabolism is a proven drug target with 63significant clinical efficacy and antifolates have been deployed for the treatment of a wide 64range of infectious diseases.5 However, due to increasing drug resistance their efficacy has 65been compromised forcing in some cases the withdrawal of formulations of antifolates unless 66they are combined with a further antimicrobial that works through a different mechanism of 67action.6 Furthermore, the available antibiotics are not extensive nor comprehensive. For 68instance, antimicrobials against Gram-negative bacteria are limited and the drugs to treat 69morbid parasitic infections are scarce and their treatment is clinically unsafe.7 Since the 70production of new antibiotics is lengthy and costly, it is imperative that there is a continued 71effort to identify pharmacological approaches to extend the life of this well known source of 72antimicrobial targets and counteract the detrimental consequences of antifolate drug 73 resistance. Due to the knowledge accumulated over eight decades on folate metabolism and 74the evidence on the efficacy of antifolates at killing sensitive cells, the folate biosynthesis and 75usage pathways continue to be a worthwhile avenue for antimicrobial developmental and 76 repurposing.8 Nonetheless, in order to identify key biochemical targets it is necessary to 77appreciate fully the dynamics of folate in microbial metabolism and cell growth. 78Fundamentally, cell proliferation requires folate for the biosynthesis of nucleic acids and the 79metabolism of amino acids. Animals can derive sufficient folate from their diet or by 80symbiotic relationships in their intestinal microflora. Thus, they have disposed of the 81endogenous folate biosynthesis pathway.9,10 However, unlike animals, most free living 82microorganisms (and plants) are capable of either salvaging folate from their environment, or 83producing de novo folate when there is a decline in folate availability. Therefore, if the 84immediate environment and diet do not offer this essential vitamin the biosynthetic pathway 85of folate becomes key for the viability of active proliferatingmicroorganisms.11 Crucially, 86this biochemical switch characterises the behaviour of most microbial pathogens including 87bacteria12 and parasitic single-cell eukaryotes such as the malaria parasites.8 To combat these 88pathogens antifolates of clinical use in medical and veterinary practice target primarily the 89following three enzymes: dihydropteroate synthase (DHPS), dihydrofolate reductase (DHFR) 90or thymidylate synthase (TS), while an enzyme of the Shikimate pathway is the target of the 91herbicidal glyphosate. Thus, the pharmacological utility of the extensive range of other 92enzymes involved in the folate biosynthesis and utilisation network are still to be fully 93exploited.

94

95Mainly due to the association its dysregulation has with cancer folate metabolism has been 96more intensively studied in mammals. This experimental work has been used to inform the 97assembly of a number of mathematical models of folate metabolism.13–19 In silico 98mammalian models have been used to represent in vivo purine biosynthesis,20 the kinetics of 99the folate cycle in human breast carcinoma cells,21 the impact of vitamin B12 deficiency on 100the folate cycle,13 the influence of genetic polymorphisms in methylene tetrahydrofolate 101reductase and thymidylate synthesis,22 the effect of epithelial folate concentrations on DNA 102methylation rate and purine and thymidylate synthesis,23 the high correlation between tissue 103and plasma folate and the low correlation between liver and plasma folate,16 and how vitamin 104B-6 restriction alters one-carbon metabolism in cultured HepG2 cells.19 These mathematical 105models have all worthwhile features and have deepened our understanding of the complex 106dynamics which underpins the folate cycle in mammals. However, to our knowledge at 107present, there is no mathematical model which has represented microbial de novo 108biosynthesis as well as the usage of folate (the folate cycle). Thus, it could be argued that the

109microbial biochemical folate system remains less well understood than its mammalian 110counterpart.

111

112

113In this paper we describe the assembly of a mathematical model of the microbial biosynthetic 114and usage pathways. This model is based on the biochemical architecture of a single celled 115microorganism, and is underpinned by known enzyme kinetics. The robustness of our model 116is based on its capacity to represent known folate inhibitory profiles as well as its capacity to 117predict effective new drug combinatorial profiles. Furthermore, this model includes folate 118metabolites recently identified as being involved in dormancy related persister bacteria and 119illustrates the likely metabolic folate profile of such a phenotype. Together these features of 120the emodel suggest that our model is a suitable template which could help to exploit novel 121aspects of this complex network for new antifolate chemotherapy.

122

123

1242 Methods

125

126The model proposed here comprises 31 reactions and 51 metabolites. The different reactions 127 are in Table 1 with extended annotation in Table S1 with the metabolites abbreviated as in 128Table S2 (ESI⁺). The components of our model are informed by the existing kinetic models 129briefly described above, and by the most recent reviews of microbial folate metabolism.8,24 130Moreover, a number of microbial metabolic representations that describe folate related 131 reactions were explored. These pathways are archived within the KEGG database (Kyoto 132encyclopedia of genes and genomes http://www.genome.jp/kegg/) (accessed July 2015)25 133which is based on the comparative genomics from the hundredsof microbial genomes 134sequenced to date.26 Kinetic parameters were compiled from the enzyme database BRENDA 135(accessed July 2015)27 (Tables S3 and S4, ESI⁺). Kinetic parameters for ADCS (reaction 136(8)), for E. coli, were extrapolated from ref. 28. Kinetic parameters for ADCL (reaction (9)), 137for P. falciparum, were extrapolated from ref. 29. The final curated model consists of 138 reactions reported from all three microbial models E. coli, S. cerevisiae, and P. falciparum 139and encompasses a biosynthesis component (the Shikimate pathway leading to the synthesis 140of pABA fromglycolytic intermediates, the pteridin biosynthesis pathway from GTP, and the 141reactions leading to the production of fully reduced and polyglutamated folate), and an 142interconversion cycle of reduced andpolyglutamated folate products (Fig. 1). All reactions are 143listed in Table S1 and all metabolites and their abbreviations are listed in Table S2 (ESI[†]).

144

145

146

147The vectorial assembly of this model was created with systems biology graphical notation 148(SBGN) (http://www.sbgn.org/Main Page)30 and implemented in VANTED (Version 2.2.1, 149http://vanted.ipkgatersleben.de/).31 We then converted this biochemical network into a series 150of reactions (Table 1 and Table S1, ESI†) and asembled them in Version 4.14.89 of the 151modelling and simulation software tool Copasi.32 The initial velocity of each reaction is 152underpinned by a rate law that depends on the concentrations of the reaction substrates, 153products, and co-factors. These rate laws are nonlinear and in general are described by 154Michaelis–Menten kinetics (list of ODEs in ESI†) for either one, two, or three substrates 155assuming a random-order mechanism.33 The following mathematical expressions 156exemplified the different Michaelis–Menten equations as used for this model for one 157substrate, two substrates, and three substrates:33

159

160The equations for reactions that had metabolite modifiers (inhibitors) included (see Table 1) 161are exemplified by eqn (4) where the concentration of the inhibitory metabolite is taken

162into account, together with its affinity constant (Ki), by its effect on the Km values in the 163denominator. Dihydrofolate (DHF) has been shown to act as inhibitor of a number of

164reactions in the folate cycle. Chiefly among them is folylpolyglutamate synthase (reaction 165(16)) which DHF inhibits with a constant of 3.1 mM.34 The model built here includes DHF 166as a modifier (inhibitor) of reactions (16), (19), (21) and (22). THF has also been involved in 167regulatory feedback by inhibiting reactions (10) and (17).8 This is reflected here by including 168THF as a modifier in such reactions (list of ODEs in ESI†). The model does not include 169membrane transport of folates. Folate membrane transporters have been found in folate 170heterotrophs24 and organisms with dual de novo folate biosynthesis and salvage 171capabilities.35–37 The former are obviously not covered by the model assembled here. Some 172single-celled eukaryotes (i.e. Apicomplexan such as malaria and toxoplasmosis parasites) 17337 and plants can perform both biosynthesis as well as salvage of folate from environment 174mainly via the FBT family of transporters.38 This extra layer of complexity is not included in 175our model.

177

178

179Substrate steady state concentrations available mainly from ref. 39 were included as initial 180concentrations. The initial concentration of boundary metabolites was fixed. These included 181PEP, EP, and GTP. These are the substrates for the initial reactions of the Shikimate pathway 182and the pterin biosynthesis, respectively. Also fixed were the initial concentrations of 183cofactors for which recycling reactions are not part of the system: ATP, NADH, NADPH, 184Gln, Gly, Ser and Lp. The reactions that generate the products using folates as cofactors in 185anabolic reactions (i.e. Met, dTMP and formyl-mtRNA) needed to have their substrates (Hcy, 186dUMP, and mtRNA) also fixed (Table S5, ESI⁺). Additionally, it was also considered that the 187average microorganism would have its folate pool most polyglutamated. For example, E. coli 188has approximately 50% of folates polyglutamated and S. cerevisiae nearly fully 189polyglutamated.12,40 To reflect this, all folate intermediates containing fully reduced 190tetrahydrofolate (THF) in this model are denoted as polyglutamated by using the suffix Glu in 191their abbreviations. The maximal rates (Vmax in micromoles (Litre)_1 (min)_1) of enzymatic 192reactions were calculated from the specific activities 41 of purified protein extracts as reported 193in BRENDA (micromoles (mg of protein) 1 (min) 1).27 In a bacterial cell such as E. coli 194proteins constitute about 55% of the dry cell weight and the cytoplasm has a density of 1.1 195 with 70% water.41 With these parameters the Vmax values were calculated by converting 196the mass (mg of protein) of the specific activity of a given enzyme to volume in litres to 197 represent Vmax values 41 as explained in Table S4 (ESI⁺). The model is encoded in the 198systems biology markup language (SBML)42 and was submitted to the BioModels Database, 199a repository for computational models of biological processes.43 This means that the model 200is accessible and can be updated as the biological knowledge of the system advances. 201

201

202

2043 Results

205

206

207**3.1 Initial examination of the model**

2080nce the initial set of parameters were added to the model we ran a number of simulations. It 209was found that the system reached steady state at approximately 300 minutes (Fig. 2 and Fig. 210S1 and Table S5, ESI†). Fig. 2 captures the steady state values for the folate cycle. Fig. 2A 211represents the intermediates of the cycle, while Fig. 2B represents the products of the cycle, 212namely methionine, dTMP and formyl-met-tRNA (fmtRNA). The concentrations of the 213metabolites and the fluxes related to the biosynthesis of folates range over several orders of 214magnitude as summarized in Table S5 (ESI†). Importantly, the folate pool seems to be stored 215mainly as two intermediates: the polyglutamated and fully reduced form THFGlu and its 216intermediate carrying the one-carbon unit as methenyl (meTHFGlu) (Fig. 2A and Table 2). 217This is an important finding of the model since neither THFGlu nor meTHFGlu are direct 218cofactors for the anabolic reactions where folates are involved. On the other hand, the

¹⁷⁶

219products derived directly from the folate cycle reactions are represented by methionine at a 220concentration of 172 mMand dTMP at a concentration of 45.7 mM. Themodified methionyl-221transfer RNA (fmtRNA) reaches a steady state at a much lower level (2.15 mM) than the 222other products (Fig. 2B). From these the only metabolite with a reported steady state 223concentration in microorganisms is methionine at a mean value of 142 mM39 which is close 224to the value derived from the simulation of this model (172 mM).

225

226

227

228

2293.2 Modelling the effect of known antifolates

230We modelled the effect of inhibiting enzymes by running parameter scans of the Vmax of a 231 given enzyme from the initial Vmax value entered for that enzyme down to decimal minimal 232values approaching zero (0.01 micromoles (Litre)_1 (min)_1) to simulate maximal inhibition 233(Fig. 3). The most commonly targeted enzyme by antifolates of clinical use is DHFR. Seven 234folate intermediates (THF, THFGlu, mvTHFGlu, meTHFGlu, fTHFGlu, ffTHFGlu and 235MTHFGlu) and the three products (Met, dTMP and fmtRNA) were all affected by the 236reduction of the Vmax of DHFR (Fig. 3). The effect on metabolites present at much lower 237levels such as fTHFGlu and fmtRNA is less visible. Importantly, the metabolite 238concentrations were at their lowest from the point where approximately a reduction of 90% of 239the Vmax had been reached. The methyl carrier MTHFGlu, and both products methionine and 240dTMP are at 4% and 10%, respectively, of their initial steady state concentrations when 241DHFR was inhibited. The inhibition of DHPS, another commonly targeted folate enzyme 242(currently by using sulfa drugs), affects the levels of THF and myTHFGlu, the two immediate 243products of de novo folate biosynthesis and one-carbon folate metabolism, respectively (Fig. 244S2, ESI⁺). Similarly, the effects of targeting the Shikimate pathway, simulated here by PSCVT (phosphoenolpyruvate: 3-phosphoshikimate 5-O-(1-carboxyvinyl)-245inhibiting 246transferase), target of glyphosate, presented a similar inhibition profile to that observed for 247DHPS. The inhibition of TS on the other hand, was limited to the decline of dTMP to 248negligible levels (Fig. S2, ESI⁺).

249

250

251 252**3.3 Modelling the effect of known antifolate combination therapies**

253

254Antifolate chemotherapy has been deployed using inhibitors that target at least two enzymes 255of folate biosynthesis and usage pathways and usually work due to a synergistic effect.44 The 256most common of such combinations is a DHFR inhibitor and a DHPS inhibitor for the 257treatment of infectious diseases. Targeting DHFR and TS has also been used to kill cancerous 258cells.45 The effects of the combined reduction of the Vmax for DHFR and DHPS (Fig. 4), 259and DHFR and TS (Fig. S3, ESI⁺) were simulated. The response of the two folate products 260(Met and dTMP) and the two metabolic intermediates (THFGlu and meTHFGlu) were used to 261illustrate the effects of these combined inhibitions. The inhibition of DHFR and DHPS has an 262overall effect on all of these metabolites while the inhibition of DHFR and TS has its main 263effect on dTMP, which was significantly reduced (Fig. S3, ESI⁺). An important aim of this 264model was to find new potential inhibitory combinations that could reduce the levels of folate 265metabolites and products which could work more effectively than the current antifolates. As 266the model successfully simulated the known effects of inhibiting DHFR46 (Fig. 3), we 267therefore decided that it would be logical to investigate the effects of inhibiting DHFR and a 268second target. Using the levels of dTMP as an indicator of cell survival, a scan of the Vmax of 269DHFR was performed while the Vmax of a second enzyme was set to negligible levels (0.01 270micromoles (Litre)_1 (min)_1). Firstly, we simulated the known synergism of inhibiting both 271DHFR and DHPS, which is the most common antifolate combinatorial chemotherapy against 272infectious microorganisms. The levels of dTMP when DHFR was inhibited alone reached the 273lowest point (5 mM) when the Vmax for DHFR was just below 1000 micromoles (Litre) 1 274(min)_1. When DHPS was also inhibited (Vmax = 0.01 micromoles (Litre)_1 (min)_1) the 275levels of dTMP were minimal even at high values of DHFR Vmax (2500 micromoles 276(Litre)_1 (min)_1) (Fig. 5A). Therefore, based on this output we reasoned that the model was 277suitable for the simulation of new potential combinations.

278 279

280It was decided that any other possible combination should be compared against the combined 281inhibition of DHFR and DPHS as illustrated above. Explicitly, the Vmax values at which 282DHFR render low levels of dTMP would be above, the same, or below the DHFR Vmax mark 283of 2500 micromoles (Litre)_1 (min)_1 as witnessed when DHPS is also inhibited. 284Unexpectedly, inhibitors of the Shikimate pathway did not render a change in the Vmax for 285DHFR that could be considered an improvement of the inhibition of DHFR alone. Namely, 286the inhibition of PSCVT (the targetof glyphosate) was no better than inhibiting DHFR on its 287own. On the other, and rather encouragingly the inhibition of other potential targets such as 288the enzyme that modifies folates by polyglutamation (FPGS) displayed a much more 289pronounced reduction in the levels of dTMP than the inhibition of DHPS (Fig. 5).

291

292**3.4 Predicting additive inhibitory effects of new combinations of antifolates** 293

294

295The above results prompted us to formulate a method that could facilitate a clear visualisation 296of additivity on cell toxicity by new combinations of a given candidate inhibitor and an anti-297DHFR compound. In experimental pharmacology cellular toxicity is measured as the 298concentrations of a molecule that affect cell survival: inhibitory concentrations IC50 or IC90. 299The type of screening we believe is worthwhile formulating is that which detects drugs or 300compounds that reduce the ICs of an anti-DHFR inhibitor at the same or lower levels than the 301known synergistic combinations with anti-DHPS drugs (Fig. 5). The difference between for 302instance, the IC90 of an anti-DHFR alone and in the presence of another molecule would be a 303coefficient. Such a coefficient can then be used as the exponential of a natural numeric base to 304render a positive scale where the point of no effect is one (anti-DHFR IC90 minus itself 305produces an exponential zero).

306

307Using 2 as the base this scale will show maximum possible effects (strong additivity or 308synergy) as an asymptote that approaches two, and minimal effects (antagonistic) as an 309asymptote that approaches zero (Fig. 5B). Simply stated the formula is: Danti-DHFR = 3102(A_B). Where A is the IC90 of an anti-DFHR acting alone and B the IC90 of such anti-311DHFR in the presence of another inhibitor at a set concentration. In this in silico model we 312simulated this type of assays by running Vmax scans for DHFR while reducing the Vmax of 313another of the enzymes of the model to negligible levels (i.e. 0.01 micromoles (Litre)_1 314(min)_1). The representation of the known synergistic effect of an anti-DHFR and an anti-315DHPS is observed under this method as a change of 1.5 in the inhibitory concentrations of an 316anti-DHFR (a reduction in its ICs (IC90 or IC50) of 50%) (Fig. 5B).

317

318When the same simulation was run with all other possible targets, significantly, inhibiting 319enzymes of the Shikimate pathway (e.g. PSCVT) did not seem to enhance the effect of 320inhibiting DHFR alone (observed as a change in the levels of dTMP). A similar trend was 321observed when lowering the Vmax values for SHMT, an enzyme directly involved in the 322one-carbon transfer to folates. On the other hand, an effect well above the reference (set by 323inhibiting DHPS) was observed when reducing the levels of the Vmax for FPGS. Inhibiting 324DHFR was significantly improved in the latter case with a score approaching 2. An increased 325efficacy of 100% for an anti-DHFR inhibitor when in the presence of an anti-FPGS 326compound (Fig. 5B).

327

328

3313.5 Sensitivity of the system to cell energy and redox status

332A question to address in folate metabolism relates to the effect that the energy status of a cell 333will have on the biosynthesis and usage of folate. As it takes four molecules of ATP to 334produce a new fully reduced monoglutamated folate (every additional condensation of a 335glutamate will cost an extra ATP), the full biosynthesis of folate ought to be sensitive to the 336energy status of the cell. Folate biosynthesis also requires reductive equivalents in the form of 337both NADH and NADPH. Recent findings from experimental work have confirmed that the 338folate metabolic network has a crucial role to play in maintaining the homoeostasis of cell 39biomass.47 Depending on the direction of the reactions of the folate cycle, the folate one-340carbon reactions can equally generate net energy and reductive equivalents (i.e. ATP and 341NADPH).48,49 Consequently, there is a need for a framework that integrates folate 342metabolism with cell growth and energy homoeostasis. In the model presented here reducing 343the levels of ATP to 1% reduced the concentration of most folate metabolites (Table 2 and 344Fig. 6). The reduction of other substrates such as glutamine, and NADPH had a similar effect 345on the folate pool (Table 2, ESI†). However, changes to ATP and NADPH were 346most significant (Fig. S5, ESI†).

350We were interested in metabolites whose concentration increased under restrictive energy 351conditions, as these could be feedback molecules for the folate biosynthesis and utilisation 352pathways. For instance under low ATP the monoglutamated THF accumulates (Table 2 and 353Fig. 6). Importantly, a similar trend is followed by 5-formyl-THFGlu (ffTHFGlu: folinic 354acid). Both molecules are known to be negative regulators of folate biosynthesis enzymes 355SHTM and GTPCH-I (ref. 8 and 50). Two other metabolites, SK and SAmDLp, increase to 356very higher levels (Fig. S5, ESI†). A synergistic effect by SK with other carbon sources in the 357promotion of cell growth has been observed in bacteria.51 However, the roles of SK, and 358SAmDLp, during limited nutrient availability and low ATP are unknown. Similarly, when the 359levels of NADPH decreased among the expected metabolites to become abundant, THF and

360DHF are again known regulators of the folate biosynthesis. However, the functions of 361DHSK,myTHFGlu, and again SAmDLp, whose concentrations are

362significantly higher in low NADPH (Fig. S5, ESI⁺), are unknown.

3684 Discussion

369Folate metabolismin microbes currently suffers from a paradox. Although, there is an 370abundance of experimental information derived mainly from microbial models such as E. coli 371and S. cerevisiae, on closer inspection there is still a lack of understanding of the regulatory 372mechanisms which underpin the dynamics of folate biosynthesis and utilisation. In mammals 373specific concentrations of intracellular and circulating folates are known to have predictable 374implications.52 In contrast, we lack a quantitative framework for understanding the 375relationship between intrinsic folate levels and microbial cell growth and multiplication.53 A 376basic initial challenge is to know the intracellular concentration of intracellular rmetabolites. 377Although, there have been efforts to quantify steady state metabolite content in microbes (e.g. 378E. coli),39,54 cofactors such as folates pose inherent difficulty for detection because they are 379in submicromolar concentrations and mostly protein-bound. Further complications arise from 380genomic driven automatic annotation of the myriad of microbial genomes. The genotypes of 381folate biosynthesis enzymes appear to have local gene variability that have compounding 382effects on gene annotation. Nonetheless the architecture of folate biosynthesis pathways 383seems evolutionary constrained.24,55 Consequently, the model presented here centres on the

384metabolic reactions that are widely regarded as fundamental to a fully biosynthetic 385microorganism and attempt to capture a broad set of parameters that allows us to integrate a 386systems level overview of microbial folate metabolism.

387

388

389The concentration of all folate metabolites and products represented here reach a steady state. 390THFGlu and meTHFGlu represent the main forms of folate in this model under steadystate 391conditions. This is a meaningful feature of the model since THFGlu is the product of the de 392novo biosynthesis of folate and meTHFGlu is the product of the condensation of the one 393carbon unit (from serine or glycine) on to THFGlu. Crucially, meTHFGlu is the substrate for 394the futile cycle with folinic acid (Fig. 1).56 Furthermore, THFGlu is a known regulatory 395(inhibitor) metabolite of folate biosynthesis enzymes such as GTPCHI which catalyses the 396 first reaction of the pteridin biosynthesis pathway.8 It is therefore noteworthy that the model 397assembled here demonstrates that these two folate intermediates, with such essential roles in 398the known biochemistry of folate utilisation are the main reservoir of the folate pool. The 399combined inhibition of DHFR and DHPS affects the levels of both of the main folate 400intermediates THFGlu and meTHFGlu while the levels of methionine and dTMP do not differ 401significantly from the levels observed when inhibiting DHFR alone (Fig. 3–5). The 402combinatorial inhibition of DHFR and TS on the other hand, has drastic effects on the levels 403of dTMP mainly (Fig. S3, ESI⁺). Importantly, the thymineless death is known as the 404mechanism mediating cell toxicity of antifolates.57 The profiles of these inhibitory trends of 405folate metabolites and products fit with the fact that anti-DHFR inhibitors are the most 406effective antifolate mono-therapy followed only by anti-TS compounds.5 However, it is clear 407that combinatorial approaches with an anti-DHFR and a second antifolate further improve the 408efficacy of anti-DHFR inhibitors to shut down folate usage reactions.58,59 Accordingly, it 409was decided to explore combinations of DHFR inhibitors and a second target. Particularly, 410targeting enzymes that are current candidates for antifolate chemotherapy such as SHMT and 411FPGS (Fig. S4, ESI⁺). The best known methods for evaluating drug-drug interactions are 412based on the Loewe additivity model, visualised by isobolograms and measured by the 413combination index analysis.60 These empirical implementations of representing drug-drug 414interactions serve the need for methods to study cell toxicity. Particularly given that usually, 415evidence on the mechanisms of action and interactions of drug-drug and drugs targets is 416lacking. None of these methods however, have found applicability in high-throughput (HTP) 417drug screening. The need to use a range of concentrations for each of the drugs increases the 418work load exponentially to levels that defeat the purpose of screening large chemical libraries. 419Therefore, drug additivity is not routinely an aim in HTP drug screening. This simple method 420that we decided to use here to represent antifolate combinatorial inhibition could find use in 421the search for chemical hits that complement synergistically the established effects of 422inhibitors such as anti-DHFR drugs. Consistently, this approach shows important trends such 423as the drastic synergistic effect of inhibiting polyglutamation of folates on top of the 424inhibition of DHFR. An effect that has been demonstrated experimentally in mammalian cell 425lines.61 Somewhat disappointingly when we used the model to simulate the combinatorial 426inhibition of other enzymes such as SHMT and enzymes involved in the Shikimate pathway, 427the effects of inhibiting DHFR alone was not enhanced (Fig. 5). Nonetheless, screening large 428chemical libraries is arguably a worthwhile strategy to look for drug additivity, and simple 429methods such as the one presented here to measure potential synergistic interactions in HTP 430projects are a necessity.

431

432

433The finding that folinic acid increases when the level of ATP is reduced has important 434implications when considered within the context of the regulation of the metabolism of 435microbial cell growth. Folinic acid is the most chemically stable form of reduced folates that 436seems to function as a metabolic sink for the folate cycle.56 Folinic acid itself is not a 437substrate for folate utilising enzymes, it has to be transformed back into meTHFGlu by the 438ATP-driven enzyme 5-formyl THF cyclo-ligase(5-FCL: reaction (29)) before re-entering the

439 folate cycle. Folinic acid has been given potential roles as a reservoir of cellular folate and as 440a regulatory metabolite through the inhibition of a number of folate biosynthesis 441enzymes.50,56,62 As a potential folate reservoir folinic acid is present in high levels (over 44270% of the folate pool) in dormant cellular forms such as plant seeds and fungi spores, 56, 63 443and the overexpression of 5-FCL has been associated with bacterial dormant phenotypes in 444liquid culture64 as well as in biofilms.65 Also, inhibiting 5-FCL has been shown to affect cell 445growth.56,66 Related to the latter, 5-FCL has been described as a pathogenic factor necessary 446for antifolate drug resistance in Mycobacterium.67 Thus, folinic acid seems to be part of a 447substrate cycle with invested value since it is seemingly used for both cell dormancy as well 448as actively cell growth. It is possible that this ATP-driven reaction is used by the folate cycle 449as an energy sensor whereby cellular stress and low ATP is sensed by the folinic acid 450substrate loop. When conditions are more favourable, activation of folinic acid restores the 451 flux downstream this futile cycle. Sensitivity in metabolic regulation is the relationship 452between the relative change in enzyme activity and the relative change in concentration of a 453 regulator.68 As an outlining feature this model of the folate biosynthesis pathway and the 454folate cycle substantiates the cited works that propose the 5-FCL reaction as a potential 455substrate cycle as part of the regulatory signals of the folate metabolism.

456 457

458A kinetic model to detect parameter dependencies can have limitations. The model outputs are 459influenced by the accuracy of the enzymatic kinetic parameters. However, these parameters 460have inherent variability due to differences in the experimental conditions in which they were 461quantified. Particularly, when as in this model, the objective was to build a generic 462construction of the relevant microbial pathways. We have mitigated against this limitation by 463compiling a metabolic network of consensus reactions for folate biosynthesis across species, 464and the distributions for a large number of values for the relevant kinetic from generic 465databases as well as the literature. Additionally, the model is well informed by the inclusion 466of the initial steady state concentrations for the majority of metabolites from studies on 467microbial model organisms which report absolute values using modern metabolomics 468techniques. The robustness and accuracy of this type of model then becomes apparent, as is 469the case in this work, by the steady-state values of metabolic products that agree with the 470literature data and the predictability of the effects of local parameter variations. The latter 471includes the agreement of the model with the known effects of existing inhibitors.

472

473 474

4755 Conclusions

476We have assembled a generic mathematical model of microbial folate biosynthesis and usage. 477This model is able to reproduce many of the key biochemical dynamics which underpin folate 478metabolism in microorganisms. We acknowledge that the model has limitations. For instance, 479model outputs are inexorably dictated by enzymatic kinetic parameters. These parameters 480have inherent variability due to differences in the experimental conditions in which they were 481quantified. Equally relevant to the validity of the model is that its foundations are based on 482the general consensus within the field that these are the accepted reactions of folate 483biosynthesis and utilisation. For example, for some reactions such as the initial steps of the 484pterin biosynthesis pathway alternative catalytic steps have been proposed.69Nonetheless, the 485model is consistent with the biology of folate metabolism and provides a number of useful 486biochemical insights as well as results which have meaningful implications. These include the 487presentation of two folate intermediates of the folate cycle, THFGlu and meTHFGlu, as the 488main components of the network of folate substrates. The simulation of the inhibition of 489certain folate enzymes seems to us particularly useful. DHFR stands out as the most 490efficacious target to inhibit and any combinatorial approach should consider including an anti-491DHFR. A combination that results with effects stronger than the benchmark of inhibiting 492DHPS and DHFR seems to e the inhibition of the polyglutamation (FPGS) of folates together 493 with inhibiting DHFR. These findings could be pertinent for the future development of

494antifolates. Lastly, and of significant interest this model supports that the folinic acid 495biosynthesis loop appears to act as a folate-mediated regulatory circuit in cell growth. In the 496 future we hope to use this model to explore this finding in greater depth.

497 498

499Acknowledgements

500The authors would like to acknowledge the support by the School of Health Science 501(Liverpool Hope University, UK) and the Department of Chemical Engineering (University of 502Chester, UK) in the research work that led to this paper

503

504

505

506

507**References**

508 5091 C. J. L. Murray, K. F. Ortblad, C. Guinovart, S. S. Lim, T. M. Wolock, D. A. Roberts, E. A. 510Dansereau, N. Graetz, R. M. Barber, J. C. Brown, H. Wang, H. C. Duber, M. Naghavi, 511D.Dicker, L. Dandona, J. A. Salomon, K. R. Heuton, K. Foreman, D. E. Phillips, T. D. 512Fleming, A. D. Flaxman, B. K. Phillips, E. K. Johnson, M. S. Coggeshall, F. Abd-Allah, S. F. 513Abera, J. P. Abraham, I. Abubakar, L. J. Abu-Raddad, N. M. Abu- Rmeileh, T. Achoki, A. O. 514Adeyemo, A. K. Adou, J. C. Adsuar, E. E. Agardh, D. Akena, M. J. Al Kahbouri, D. 515Alasfoor, M. I. Albittar, G. Alcala'-Cerra, M. A. Alegretti, Z. A. Alemu, R. Alfonso-516Cristancho, S. Alhabib, R. Ali, F. Alla, P. J. Allen, U. Alsharif, E. Alvarez, N. Alvis-Guzman, 517A. A. Amankwaa, A. T. Amare, H. Amini, W. Ammar, B. O. Anderson, 518C. A. T. Antonio, P. Anwari, J. A"rnlo"v, V. S. A. Arsenijevic, 519A. Artaman, R. J. Asghar, R. Assadi, L. S. Atkins, A. Badawi, 520K. Balakrishnan, A. Banerjee, S. Basu, J. Beardsley, T. Bekele, 521M. L. Bell, E. Bernabe, T. J. Beyene, N. Bhala, A. Bhalla, 522Z. A. Bhutta, A. B. Abdulhak, A. Binagwaho, J. D. Blore, 523D. Bose, M. Brainin, N. Breitborde, C. A. Castan^eda-Orjuela, 524F. Catala'-Lo'pez, V. K. Chadha, J.-C. Chang, P. P.-C. Chiang, 525T.-W. Chuang, M. Colomar, L. T. Cooper, C. Cooper, K. J. 526Courville, B. C. Cowie, M. H. Criqui, R. Dandona, A. Dayama, 527D. De Leo, L. Degenhardt, B. Del Pozo-Cruz, K. Deribe, D. C. Des 528Jarlais, M. Dessalegn, S. D. Dharmaratne, U. Dilmen, E. L. Ding, 529T. R. Driscoll, A. M. Durrani, R. G. Ellenbogen, S. P. Ermakov, 530A. Esteghamati, E. J. A. Faraon, F. Farzadfar, S.-M. 531Fereshtehnejad, D. O. Fijabi, M. H. Forouzanfar, U. Fra.Paleo, 532L. Gaffikin, A. Gamkrelidze, F. G. Gankpe', J. M. Geleijnse, 533B. D. Gessner, K. B. Gibney, I. A. M. Ginawi, E. L. Glaser, 534P. Gona, A. Goto, H. N. Gouda, H. C. Gugnani, R. Gupta, 535R. Gupta, N. Hafezi-Neiad, R. R. Hamadeh, M. Hammami, 536G. J. Hankey, H. L. Harb, J. M. Haro, R. Havmoeller, S. I. Hay, 537M. T. Hedavati, I. B. H. Pi, H. W. Hoek, J. C. Hornberger, 538H. D. Hosgood, P. J. Hotez, D. G. Hoy, J. J. Huang, K. M. Iburg, 539B. T. Idrisov, K. Innos, K. H. Jacobsen, P. Jeemon, P. N. Jensen, 540V. Jha, G. Jiang, J. B. Jonas, K. Juel, H. Kan, I. Kankindi, 541N. E. Karam, A. Karch, C. K. Karema, A. Kaul, N. Kawakami, 542D. S. Kazi, A. H. Kemp, A. P. Kengne, A. Keren, M. Kereselidze, 543Y. S. Khader, S. E. A. H. Khalifa, E. A. Khan, Y.-H. Khang, 544I. Khonelidze, Y. Kinfu, J. M. Kinge, L. Knibbs, Y. Kokubo, 545S. Kosen, B. K. Defo, V. S. Kulkarni, C. Kulkarni, K. Kumar, 546R. B. Kumar, G. A. Kumar, G. F. Kwan, T. Lai, A. L. Balaji, 547H. Lam, Q. Lan, V. C. Lansingh, H. J. Larson, A. Larsson, 548J.-T. Lee, J. Leigh, M. Leinsalu, R. Leung, Y. Li, Y. Li,

549G. M. F. De Lima, H.-H. Lin, S. E. Lipshultz, S. Liu, Y. Liu, 550B. K. Lloyd, P. A. Lotufo, V. M. P. Machado, J. H. Maclachlan, 551C. Magis-Rodriguez, M. Majdan, C. C. Mapoma, W. Marcenes, 552M. B. Marzan, J. R. Masci, M. T. Mashal, A. J. Mason-Jones, 553B. M. Mayosi, T. T. Mazorodze, A. C. Mckay, P. A. Meaney, 554M. M. Mehndiratta, F. Mejia-Rodriguez, Y. A. Melaku, Z. A. 555Memish, W.Mendoza, T. R.Miller, E. J. Mills, K. A. Mohammad, 556A. H.Mokdad, G. L. Mola, L. Monasta, M. Montico, A. R. Moore, 557R. Mori, W. N. Moturi, M. Mukaigawara, K. S. Murthy, 558A. Naheed, K. S. Naidoo, L. Naldi, V. Nangia, K. M. V. 559Narayan, D. Nash, C. Nejjari, R. G. Nelson, S. P. Neupane, 560C. R. Newton, M. Ng, M. I. Nisar, S. Nolte, O. F. Norheim, 561V. Nowaseb, L. Nyakarahuka, I.-H. Oh, T. Ohkubo, B. O. 562Olusanya, S. B. Omer, J. N. Opio, O. E. Orisakwe, J. D. 563Pandian, C. Papachristou, A. J. P. Caicedo, S. B. Patten, 564V. K. Paul, B. I. Pavlin, N. Pearce, D. M. Pereira, A. Pervaiz, 565K. Pesudovs, M. Petzold, F. Pourmalek, D. Qato, A. D. Quezada, 566D. A. Quistberg, A. Rafay, K. Rahimi, V. Rahimi-Movaghar, 567S. U. Rahman, M. Raju, S. M. Rana, H. Razavi, R. Q. Reilly, 568G. Remuzzi, J. H. Richardus, L. Ronfani, N. Roy, N. Sabin, 569M. Y. Saeedi, M. A. Sahraian, G. M. J. Samonte, M. Sawhney, 570I. J. C. Schneider, D. C. Schwebel, S. Seedat, S. G. Sepanlou, 571E. E. Servan-Mori, S. Sheikhbahaei, K. Shibuya, H. H. Shin, 572I. Shiue, R. Shivakoti, I. D. Sigfusdottir, D. H. Silberberg, 573A. P. Silva, E. P. Simard, J. A. Singh, V. Skirbekk, K. Sliwa, 574S. Soneji, S. S. Soshnikov, C. T. Sreeramareddy, V. K. 575Stathopoulou, K. Stroumpoulis, S. Swaminathan, B. L. Sykes, 576K. M. Tabb, R. T. Talongwa, E. Y. Tenkorang, A. S. Terkawi, 577A. J. Thomson, A. L. Thorne-Lyman, J. A. Towbin, J. Traebert, 578B. X. Tran, Z. T. Dimbuene, U. S. Tsilimbaris, Miltiadis and bibexport. 579sh o short.bib mwe.aux Uchendu, K. N. Ukwaja, 580A. J. Vallely, T. J. Vasankari, N. Venketasubramanian, 581F. S. Violante, V. V. Vlassov, S. Waller, M. T. Wallin, L. Wang, 582S. X. Wang, Y. Wang, S. Weichenthal, E. Weiderpass, 583R. G. Weintraub, R. Westerman, R. A. White, J. D. Wilkinson, 584T. N.Williams, S. M. Woldevohannes, J. O.Wong, G. Xu, Y. C. 585Yang, Y. Yano, P. Yip, N. Yonemoto, S.-J. Yoon, M. Younis, C. Yu, 586K. Y. Jin, M. El Saved Zaki, Y. Zhao, Y. Zheng, M. Zhou, J. Zhu, 587X. N. Zou, A. D. Lopez and T. Vos, Lancet, 2014, 384, 1005–1070. 5882 S. I. Hay, E. A. Okiro, P. W. Gething, A. P. Patil, A. J. Tatem, 589C. A. Guerra and R. W. Snow, PLoS Med., 2010, 7, e1000290. 5903 R. W. Snow, C. A. Guerra, A. M. Noor, H. Y. Mvint and 591S. I. Hay, Nature, 2005, 434, 214–217. 5924 WHO, World Malaria Report, WHO Press, 2014. 5935 J. Walling, Invest. New Drugs, 2006, 24, 37–77. 5946 WHO, Guidelines for the treatment of malaria, WHO Press, 5952006. 5967 D. Brown, Nat. Rev. Drug Discovery, 2015, 14, 821-832. 5978 J. E. Salcedo-Sora and S. A. Ward, Mol. Biochem. Parasitol., 5982013, 188, 51-62. 5999 A. S. Tibbetts and D. R. Appling, Annu. Rev. Nutr., 2010, 30, 60057-81. 60110 S. Blatch, K. W. Meyer and J. F. Harrison, Fly, 2010, 4, 602312-319. 60311 I. B. Mu["]ller and J. E. Hyde, Mol. Biochem. Parasitol., 2013,

604188, 63-77.

- 60512 J. M. Green and R. G. Matthews, EcoSal Plus, 2007, 2, DOI: 60610.1128/ecosalplus.3.6.3.6.
- 60713 H. F. Nijhout, M. C. Reed, P. Budu and C. M. Ulrich, J. Biol. 608Chem., 2004, 279, 55008–55016.
- 60914 C. M. Ulrich, M. C. Reed and H. F. Nijhout, Nutr. Rev., 2008, 61066(Suppl. 1), S27–S30.
- 61115 H. F. Nijhout, M. C. Reed and C. M. Ulrich, in Folic Acid
- 612Folates, Vitam. Horm, ed. G. Litwack, Academic Press, 2008,
- 613vol. 79, pp. 45-82.

61416 T. M. Duncan, M. C. Reed and H. F. Nijhout, Nutrients, 2013, 6155, 2457–2474.

61617 M. Scotti, L. Stella, E. J. Shearer and P. J. Stover, Wiley

617Interdiscip. Rev.: Syst. Biol. Med., 2013, 5, 343–365.

61818 M. T.Mc Auley, C. J. Proctor, B.M. Corfe, G. C. J. Cuskelly and

- 619K. M. Mooney, J. Comput. Sci. Syst. Biol., 2013, 6, 271–285.
- 62019 V. R. da Silva, M. A. Ralat, E. P. Quinlivan, B. N. DeRatt,
- 621T. J. Garrett, Y.-Y. Chi, H. F. Nijhout, M. C. Reed, J. F. Gregory,
- 622H. Frederik Nijhout, M. C. Reed and J. F. Gregory, Am.
- 623J. Physiol.: Endocrinol. Metab., 2014, 307, E93–E101.
- 62420 W. C. Werkheiser, Ann. N. Y. Acad. Sci., 1971, 186, 343–358.
- 62521 P. F. Morrison and C. J. Allegra, J. Biol. Chem., 1989, 264,

62610552–10566.

62722 C.M.Ulrich, M. Neuhouser, A. Y. Liu, A. Boynton, J. F. Gregory,

628B. Shane, S. J. James, M. C. Reed and H. F. Nijhout, Cancer

629Epidemiol., Biomarkers Prev., 2008, 17, 1822–1831.

63023 M. L. Neuhouser, H. F. Nijhout, J. F. Gregory, M. C. Reed,

- 631S. J. James, A. Liu, B. Shane and C. M. Ulrich, Cancer
- 632Epidemiol., Biomarkers Prev., 2011, 20, 1912–1917.

63324 V. de Cre´cy-Lagard, Comput. Struct. Biotechnol. J., 2014, 10, 63441–50.

- 63525 M. Kanehisa and S. Goto, Nucleic Acids Res., 2000, 28, 27–30.
- 63626 E. V. Koonin and Y. I.Wolf, Nucleic Acids Res., 2008, 36, 6688–6719.
- 63727 A. Chang, I. Schomburg, S. Placzek, L. Jeske, M. Ulbrich,
- 638M. Xiao, C. W. Sensen and D. Schomburg, Nucleic Acids Res.,

6392015, 43, D439–D446.

64028 V. K. Viswanathan, J. M. Green and B. P. Nichols,

641J. Bacteriol., 1995, 177, 5918–5923.

64229 G. Magnani, M. Lomazzi and A. Peracchi, Biochem. J., 2013, 643455, 149–155.

- 64430 N. Le Nove`re, M. Hucka, H. Mi, S. Moodie, F. Schreiber,
- 645A. Sorokin, E. Demir, K. Wegner, M. I. Aladjem, S. M.
- 646Wimalaratne, F. T. Bergman, R. Gauges, P. Ghazal, H. Kawaji,
- 647L. Li, Y.Matsuoka, A. Ville´ger, S. E.Boyd, L. Calzone, M.Courtot,
- 648U.Dogrusoz, T. C. Freeman, A. Funahashi, S. Ghosh, A. Jouraku,
- 649S. Kim, F. Kolpakov, A. Luna, S. Sahle, E. Schmidt, S. Watterson,
- 650G.Wu, I. Goryanin, D. B. Kell, C. Sander, H. Sauro, J. L. Snoep,
- 651K. Kohn and H. Kitano, Nat. Biotechnol., 2009, 27, 735–741.
- 65231 H. Rohn, A. Junker, A. Hartmann, E. Grafahrend-Belau,
- 653H. Treutler, M. Klapperstu[°]ck, T. Czauderna, C. Klukas and
- 654F. Schreiber, BMC Syst. Biol., 2012, 6, 139.
- 65532 S. Hoops, S. Sahle, R. Gauges, C. Lee, J. Pahle, N. Simus,
- 656M. Singhal, L. Xu, P. Mendes and U. Kummer, Bioinformatics,

6572006, 22, 3067–3074.

65833 A. Cornish-Bowden, Fundamentals of Enzyme Kinetics, Wiley-

659Blackwell, Berlin, 4th edn, 2012.

66034 Y. K. Kwon, W. Lu, E. Melamud, N. Khanam, A. Bognar and 661J. D. Rabinowitz, Nat. Chem. Biol., 2008, 4, 602–608.

66235 K. Lanthaler, E. Bilsland, P. Dobson, H. Moss, P. Pir, D. Kell 663and S. Oliver, BMC Biol., 2011, 9, 70.

66436 J. A. Delmar and E. W. Yu, Protein Sci., 2015, DOI: 10.1002/665pro.2820.

66637 J. E. Salcedo-Sora, E. Ochong, S. Beveridge, D. Johnson, A. Nzila,

667G. A. Biagini, P. A. Stocks, P. M. O'Neill, S. Krishna, P. G. Bray

668and S. A. Ward, J. Biol. Chem., 2011, 286, 44659–44668.

66938 S. S. Pao, I. T. Paulsen and M. J. Saier, Microbiol. Mol. Biol. 670Rev., 1998, 62, 1–34.

67139 B. D. Bennett, E. H. Kimball, M. Gao, R. Osterhout, S. J. Van

672Dien and J. D. Rabinowitz, Nat. Chem. Biol., 2009, 5, 593–599.

67340 H. Cherest, J. Biol. Chem., 2000, 275, 14056–14063.

67441 O. V. Demin, G. V. Lebedeva, A. G. Kolupaev, E. A. Zobova,

675T. Y. Plyusnina, A. I. Lavrova, A. Dubinsky, E. A. Goryacheva,

676F. Tobin and I. I. Goryanin, in Modelling in Molecular

677Biology, ed. G. Ciobanu and G. Rozenberg, Springer-

678Verlag, 2004, pp. 59–124.

67942 M. Hucka, A. Finney, B. J. Bornstein, S. M. Keating,

680B. E. Shapiro, J. Matthews, B. L. Kovitz, M. J. Schilstra,

681A. Funahashi, J. C. Doyle and H. Kitano, Evolving a lingua

682franca and associated software infrastructure for computational

683systems biology: the Systems Biology Markup Language

684(SBML) project, 2004, http://digital-library.theiet.org/content/685journals/10.1049/ sb_20045008.

68643 V. Chelliah, N. Juty, I. Ajmera, R. Ali, M. Dumousseau,

687M. Glont, M. Hucka, G. Jalowicki, S. Keating, V. Knight-

688Schrijver, A. Lloret-Villas, K. N. Natarajan, J.-B. Pettit,

689N. Rodriguez, M. Schubert, S. M. Wimalaratne, Y. Zhao,

690H. Hermjakob, N. Le Nove`re and C. Laibe, Nucleic Acids Res., 6912015, 43, D542–D548.

69244 R. L. Kisliuk, Pharmacol. Ther., 2000, 85, 183–190.

69345 H. M. Faessel, H. K. Slocum, Y. M. Rustum and W. R. Greco,

694Int. J. Oncol., 2003, 23, 401–409.

69546 B. I. Schweitzer, A. P. Dicker and J. R. Bertino, FASEB J.,

6961990, 4, 2441–2452

697

69847 J. W. Locasale, Nat. Rev. Cancer, 2013, 13, 572–583.

69948 A. Vazquez, E. K. Markert and Z. N. Oltvai, PLoS One, 2011, 7006, e25881.

70149 J. Fan, J. Ye, J. J. Kamphorst, T. Shlomi, C. B. Thompson and 702J. D. Rabinowitz, Nature, 2014, 510, 298–302.

70350 P. Stover and V. Schirch, J. Biol. Chem., 1991, 266, 1543–1550.

70451 H. Teramoto, M. Inui and H. Yukawa, Appl. Environ. Microbiol., 7052009, 75, 3461–3468.

70652 K. Pietrzik, Y. Lamers, S. Bramswig and R. Prinz-Langenohl, 707Am. J. Clin. Nutr., 2007, 86, 1414–1419.

70853 F. C. Neidhardt and R. Curtiss, Escherichia Coli and Salmonella: 709Cellular and Molecular Biology, American Society for Microbiology, 7102nd edn, 1996.

71154 N. Tepper, E. Noor, D. Amador-Noguez, H. S. Haraldsdo´ttir,

712R. Milo, J. Rabinowitz, W. Liebermeister and T. Shlomi,

713PLoS One, 2013, 8, e75370.

- 71455 X.-Y. Zhi, J.-C. Yao, H.-W. Li, Y. Huang and W.-J. Li, Mol.
- 715Phylogenet. Evol., 2014, 75, 154–164.
- 71656 P. Stover and V. Schirch, Trends Biochem. Sci., 1993, 18, 717102–106.
- 71857 Y. K. Kwon, M. B. Higgins and J. D. Rabinowitz, ACS Chem. 719Biol., 2010, 5, 787–795.
- 72058 R. W. Lacey, J. Antimicrob. Chemother., 1979, 5, 75–83.
- 72159 P. A. Masters, T. A. O'Bryan, J. Zurlo, D. Q. Miller and
- 722N. Joshi, Arch. Intern. Med., 2003, 163, 402–410.
- 72360 M. G. W. Liang Zhao, J. L.-S. Au, M. G. W. Liang Zhao and
- 724J. L.-S. Au, Front. Biosci., 2010, 2, 241–249.
- 72561 H. M. Faessel, H. K. Slocum, R. C. Jackson, T. J. Boritzki,
- 726Y. M. Rustum, M. G. Nair and W. R. Greco, Cancer Res.,
- 7271998, 58, 3036–3050.
- 72862 A. Goyer, E. Collakova, R. Daz de la Garza, E. P. Quinlivan,
- 729J. Williamson, J. F. Gregory, Y. Shachar-Hill and
- 730A. D. Hanson, J. Biol. Chem., 2005, 280, 26137–26142.
- 73163 V. Piironen, M. Edelmann, S. Kariluoto and Z. Bedo, J. Agric.
- 732Food Chem., 2008, 56, 9726–9731.
- 73364 S.Hansen, K. Lewis and M. Vulic´, Antimicrob. Agents Chemother., 7342008, 52, 2718–2726.
- 73565 D. Ren, L. A. Bedzyk, S. M. Thomas, R. W. Ye and
- 736T. K. Wood, Appl. Microbiol. Biotechnol., 2004, 64, 515–524.
- 73766 M. S. Field, D. M. E. Szebenyi, C. A. Perry and P. J. Stover,
- 738Arch. Biochem. Biophys., 2007, 458, 194–201.
- 73967 S. Ogwang, H. T. Nguyen, M. Sherman, S. Bajaksouzian,
- 740M. R. Jacobs, W. H. Boom, G.-F. Zhang and L. Nguyen,
- 741J. Biol. Chem., 2011, 286, 15377-15390.
- 74268 E. A. Newsholme, J. R. Arch, B. Brooks and B. Surholt,
- 743Biochem. Soc. Trans., 1983, 11, 52–56.
- 74469 S. Dittrich, S. L. Mitchell, A. M. Blagborough, Q. Wang,
- 745P. Wang, P. F. G. Sims and J. E. Hyde, Mol. Microbiol., 2008,
- 74667, 609–618.
- 747
- 748
- 749
- 750
- 751
- 752
- 753
- 754
- 755