



Contents lists available at ScienceDirect

Journal of Luminescence

journal homepage: www.elsevier.com/locate/jlumin

Recent advances in lanthanide spectroscopy in Brazil

Q1 Fabiana R. Gonçalves e Silva^a, Renato G. Capelo^b, Rodrigo Q. Albuquerque^b^a Chemistry and Biology Academic Department, Federal technological University of Paraná, Curitiba, PR, Brazil^b São Carlos Institute of Chemistry (IQSC), University of São Paulo, 13560-044 São Carlos, SP, Brazil

ARTICLE INFO

Article history:

Received 12 January 2015

Received in revised form

13 May 2015

Accepted 29 June 2015

Keywords:

Lanthanide

Rare earths

Q4 Luminescence

Glasses

Sol-gel

Crystals

Light-emitting diodes

Nanoparticles

Metal-organic frameworks

Coordination polymers

Thin films

Energy transfer

Upconversion

Theoretical methods

ABSTRACT

This review discusses recent advances in lanthanide spectroscopy involving luminescence applications carried out in Brazil. The revised topics include glasses, sol-gel, light-emitting diodes, nanoparticles, metal-organic frameworks, coordination polymers, thin films, energy transfer processes, upconversion and development of new theoretical tools. The important role played by Prof. Oscar L. Malta on this subject is evidenced by his many contributions to the broad range of investigations reported here and this review is dedicated to him, on the occasion of his 60th birthday. Q2

© 2015 Published by Elsevier B.V.

1. Introduction

The great interest in trivalent lanthanide (Ln^{3+}) compounds is in part due to their strong luminescence observed in a broad spectral range and to their long-lived excited states. For this reason, lanthanide compounds have been used in a variety of applications involving sol-gel, glasses, light-emitting diodes (LEDs), metal-organic frameworks (MOFs), among others. In many investigations of lanthanide-based systems, experimental measurements are interpreted with the help of theoretical models, through which properties and processes of excited and ground states are also described [1]. The development of new theoretical models and methodologies in this field has been shown to be very important in the last decades, since they nicely complement and enrich experimental investigations, as well as guide new experiments. Theoretical models addressing energy transfer processes [2,3], ligand field interactions [4], electronic transitions [5], chemical bond properties [6], geometry optimization [7], among others [1], have been extensively used due to their large applicability to this research field. Many of these theoretical models and

investigations carry the name of Prof. Oscar L. Malta, who has done pioneering work in the field of lanthanide spectroscopy in Brazil for almost four decades, and whose contributions are also internationally recognized.

The aim of this review is to describe the recent advances in the field of lanthanide luminescence in Brazil and to highlight the important contributions of Prof. Malta in honor of his 60th birthday. The revised topics include glasses, sol-gel, LEDs, nanoparticles, crystals, metal-organic frameworks, coordination polymers, thin films, energy transfer, upconversion, and the development of theoretical methods. Spectroscopic treatments in general are discussed inside each topic.

2. Results and discussion

This review describes the work published in the last decade (2005–2014) on lanthanide luminescence by Brazilian corresponding authors. The bibliographic search under these conditions has given up to date 235 publications. This number would be 284 if one considered publications involving Brazilians without the restriction of Brazilian corresponding authorship. However, discussing these additional investigations is out of the scope of this

E-mail addresses: fabianah@utfpr.edu.br (F.R. Gonçalves e Silva), rodrigo_albuquerque@iqsc.usp.br (R.Q. Albuquerque).

<http://dx.doi.org/10.1016/j.jlumin.2015.06.047>
0022-2313/© 2015 Published by Elsevier B.V.

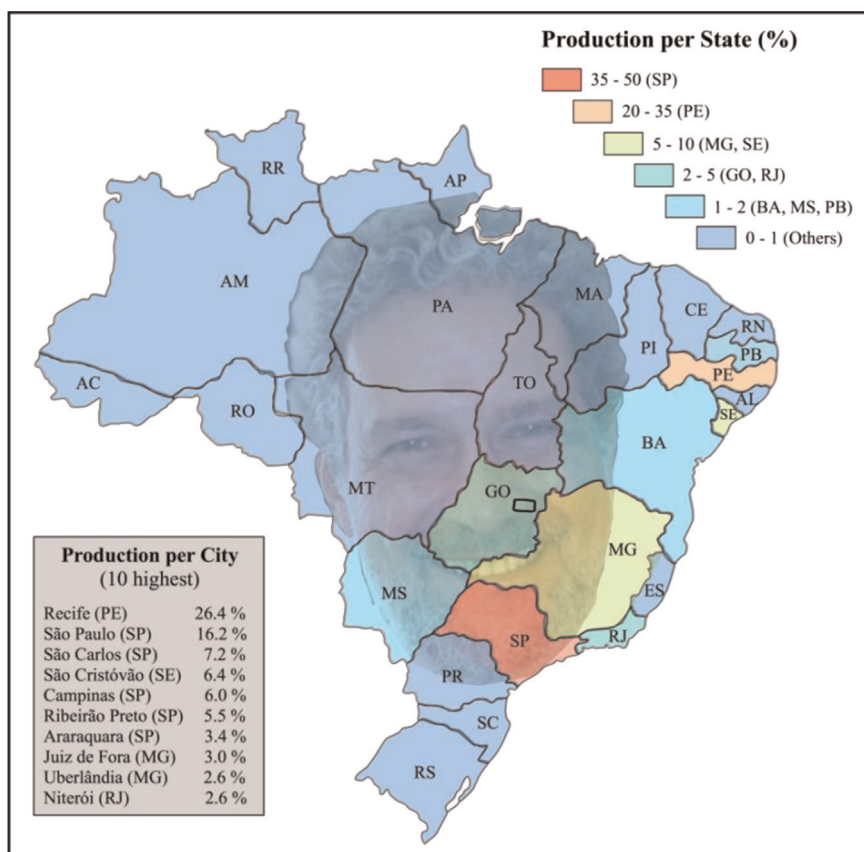


Fig. 1. Distribution of the scientific production (235 publications) in the field of lanthanide luminescence in the last decade (2005–2014) over different regions and cities of Brazil. The background shows a photo of Prof. Oscar L. Malta.

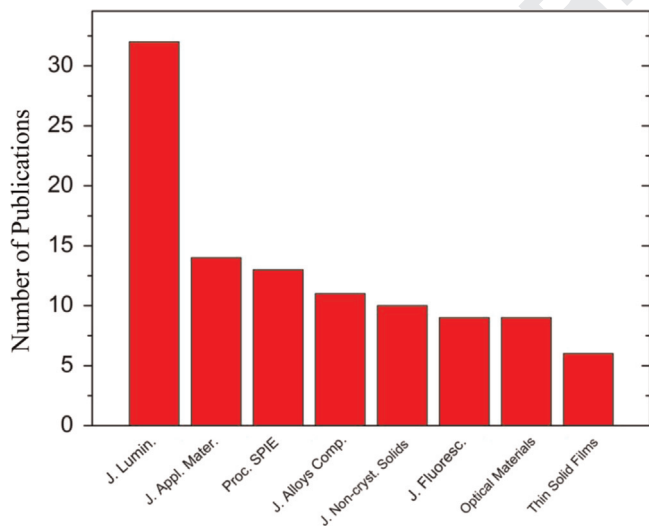


Fig. 2. Number of publications per journal of the scientific Brazilian production shown in Fig. 1. The top-eight journals are shown.

review. These international publications came from collaborations between Brazil and other countries, such as Germany, England, France, Canada, Finland, Switzerland, Spain, South Korea, Poland, Italy, Portugal, India, Turkey, Serbia, and China.

The published work selected here is described in the frame of the research topics shown in the keywords of this paper. Before going into detail about each topic, it might be instructive to get a general picture of the distribution of the scientific production of the last decade found in the field of lanthanide luminescence (235 papers) over different states and cities of Brazil (Fig. 1). Note that

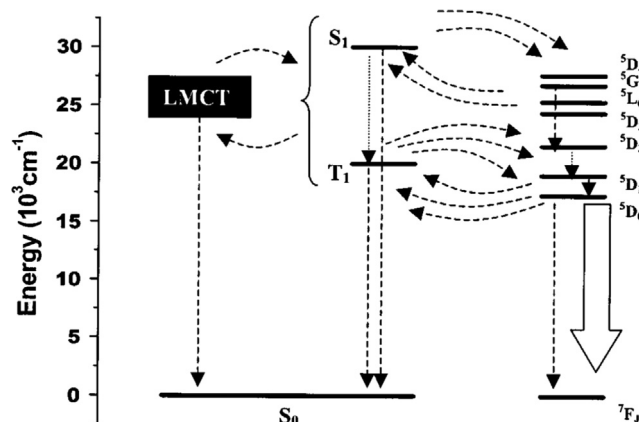


Fig. 3. ET processes taking place in a Eu^{3+} complex with the explicit participation of a LMCT state. (Reproduced with permission from Ref. [8].)

the background picture is a recent photo of Prof. Malta, which is justified by his expressive participation in about 11% of the whole Brazilian scientific production in that period, as well as by the fact that his theoretical models and findings are cited in almost all analyzed publications.

From the distribution of published papers per state, São Paulo (SP) has the highest production, followed by Pernambuco (PE). This may be due to a very large number of universities and much larger investment in science in the former. The scientific production per city delivers a much sharper picture of where science is mostly applied in this field. The capital of Pernambuco state, Recife, is highly active, even surpassing the production of the city of

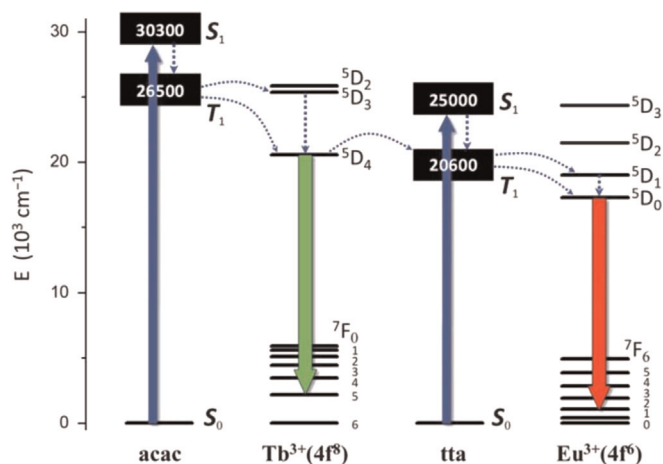


Fig. 4. Intermolecular and intramolecular energy transfer paths involving two lanthanide complexes. (Reproduced with permission from Ref. [11].)

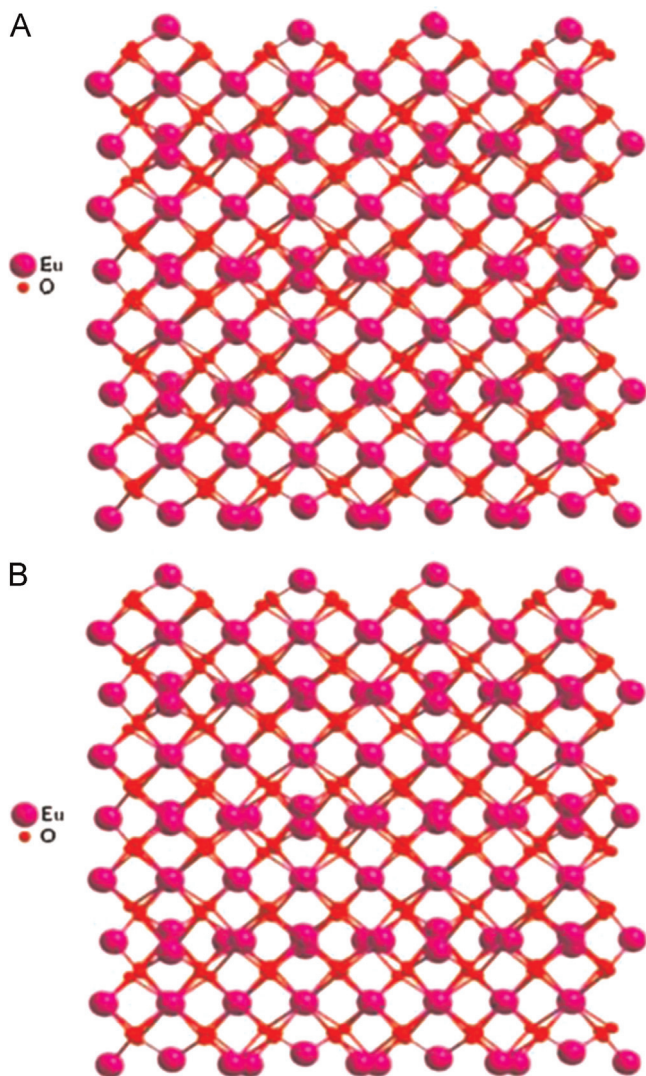


Fig. 5. Comparison between the (a) crystallographic and (b) Sparkle/PM3-optimized structure of Eu_2O_3 . (Reproduced with permission from Ref. [17].)

São Paulo. This is not only explained by the tradition of the scientific community of Recife in investigating lanthanide systems, but also by the presence of Prof. Malta there, which has been

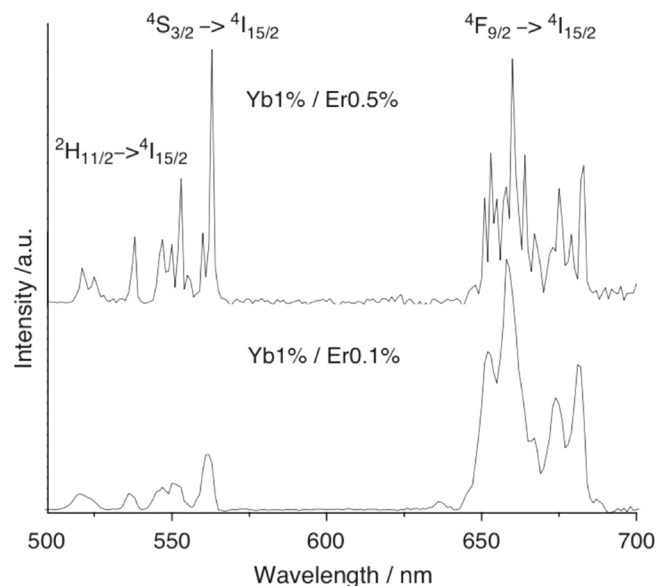


Fig. 6. Emission spectra of $\text{Y}_2\text{O}_3:\text{Er}^{3+}(0.1\%),\text{Yb}^{3+}(1\%)$ and $\text{Y}_2\text{O}_3:\text{Er}^{3+}(0.5\%),\text{Yb}^{3+}(1\%)$ prepared from a polymeric precursor, under excitation at 980 nm ($\text{Yb}^{3+}:^2\text{F}_{7/2} \rightarrow ^2\text{F}_{5/2}$ transition). (Reproduced with permission from Ref. [121].)

influencing different generations of scientists and directly contributing to different investigations on several topics as well.

The papers discussed in Fig. 1 can also be organized in terms of the journals where they were published, as shown in Fig. 2. Most investigations were published in the Journal of Luminescence, followed by the Journal of Applied Materials.

The scientific production shown in Fig. 1 is now divided into the topics shown below. Many publications are discussed in the frame of two or more topics.

2.1. Energy transfer

More than 40% of the publications shown in Fig. 1 describe experimental and/or theoretical investigations of lanthanide-based luminescent materials where energy transfer (ET) phenomena play a central role. ET processes described as ligand-to-metal, metal-to-ligand, ion-to-ion, ion-to-nanoparticle, and nanoparticle-to-ion occurring in several different systems have been often investigated, and their mechanism discussed. They can be classified into intramolecular and intermolecular ET. The calculated ET rates can be used to build a system of rate equations [1], from where the emission quantum yield (ϕ) is found. The latter parameter is one of the most important quantities used to design new luminescent materials exhibiting desired optical properties.

Theoretical and phenomenological treatments of ET processes involving lanthanide-based systems have been continuously developed by Prof. Malta, who has greatly enhanced the general understanding of such processes [8–14]. The references cited here are representative of the last decade only, which is the time range discussed in this review, although one should not forget that Malta's contributions already date from almost four decades.

The investigation of non-radiative ET involving lanthanide ions has been described in Ref. [10] and has attracted much attention since its publication. This work is better discussed in Section 2.2, since it involves ion-to-ion ET processes, which occur during up-conversion phenomena.

The role played by ligand-to-metal charge transfer (LMCT) states in intramolecular ET processes taking place in lanthanide complexes was discussed by Faustino et al. [8] and has been often quoted by the lanthanide community. Fig. 3 shows a schematic energy level diagram of one of the investigated systems, where ϕ

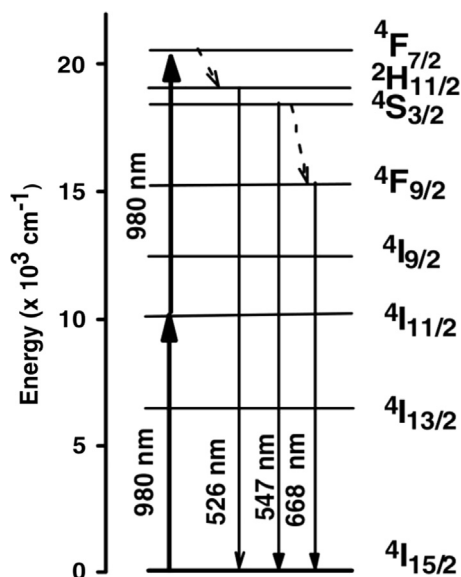


Fig. 7. Upconversion mechanism observed for the Er^{3+} -doped glasses investigated. Solid lines represent absorption or emission and dotted lines nonradiative decays by multiphonon emission. (Reproduced with permission from Ref. [122].)

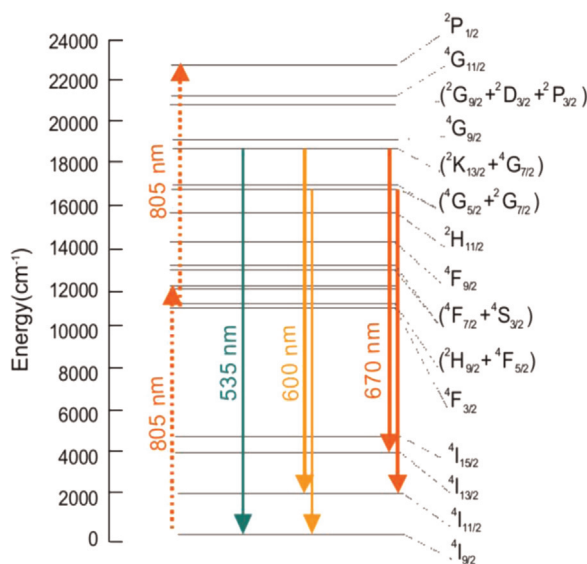


Fig. 8. Upconversion mechanism of the Nd^{3+} ions. (Reproduced with permission from Ref. [123].)

was analyzed as a function of the relative energy of the LMCT state in a typical Eu^{3+} compound.

Theoretical investigations calculate intensity and ligand field parameters, as well as ET rates of lanthanide compounds aiming at comparing the results with experimental data. This allows one to validate and optimize the theoretical parameters. In this context, Freire et al. [15] have done a theoretical investigation on the luminescence of Eu^{3+} cryptates with high ϕ values. This work used the Judd–Ofelt theory to calculate intensity parameters and the model proposed by Malta et al. [2,3] to calculate ET rates between metal and ligand, from where ϕ could be predicted and compared with the experimental one. Teotonio et al. [16] carried out a theoretical–experimental investigation of the ET process of a β -diketonate Eu^{3+} complex, also using the Judd–Ofelt theory and the calculation of ET rates to predict ϕ . In this work, some emphasis was given on the triboluminescence exhibited by the Eu^{3+} compound.

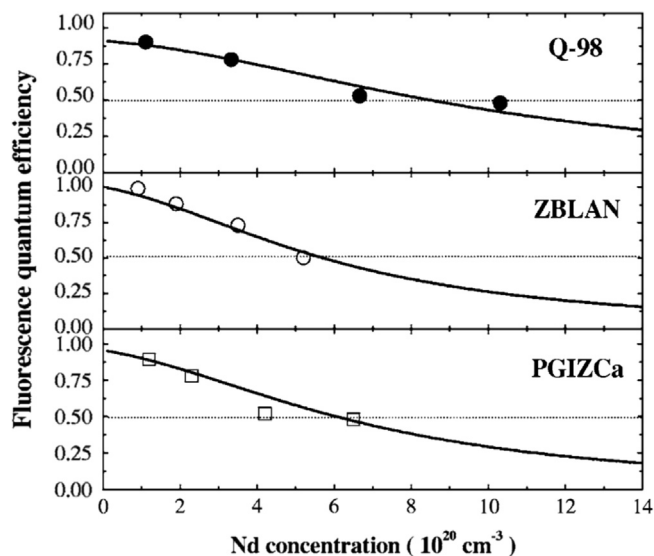


Fig. 9. η of Phosphate (Q-98), fluorozirconate (ZBLAN), and fluorindate (PGIZCa) glasses doped with Nd^{3+} as a function of $[\text{Nd}^{3+}]$ determined using the normalized-lifetime thermal lens technique. (Reproduced with permission from Ref. [148].)

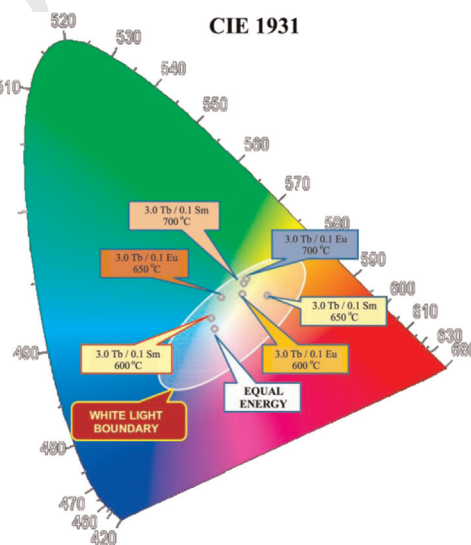


Fig. 10. CIE (Commission Internationale de l'Éclairage) chromaticity diagram of $75\text{SiO}_2:25\text{PbF}_2$ doped with Eu/Tb and Sm/Tb samples excited at 395 nm. (Reproduced with permission from Ref. [149].)

The work of Kai et al. [11] has received much attention in the last years and it describes ET processes and photostability of Eu^{3+} and Tb^{3+} β -diketonate complexes doped in poly(methylmethacrylate) (PMMA) films. In this study, besides the usually-observed intramolecular (ligand \rightarrow ion) ET, intermolecular ET was surprisingly observed, occurring from the emitting level of a near Tb^{3+} ion belonging to one complex to the excited triplet state of the β -diketonate ligand coordinated to the Eu^{3+} ion of another complex (see Fig. 4).

Ion-to-ion energy transfer in Eu_2O_3 microspheres was recently investigated by Alves et al. [17] using theoretical methods and experimental techniques. In this work semiempirical methods based on Sparkle/X ($X = \text{AM1}, \text{PM3}, \text{PM6}$) were used to treat part of the solid Eu_2O_3 system comprising 600 atoms (Fig. 5). The ET rates calculated via dipole–dipole or dipole–quadrupole mechanisms were one order of magnitude smaller than those found for the quadrupole–quadrupole mechanism, indicating that the latter

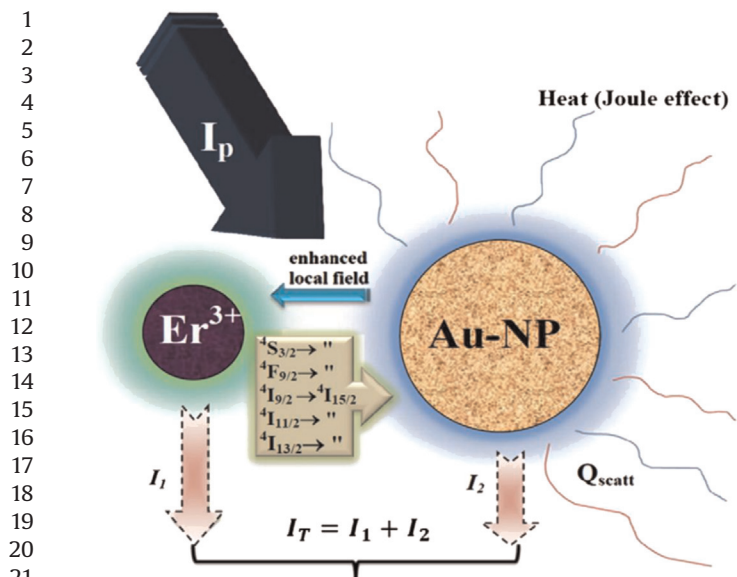


Fig. 11. Schematic representation of the $\text{Er}^{3+}:\text{Au-NP}$ system and photophysical processes occurring after irradiating the sample. (Reproduced with permission from Ref. [40].)

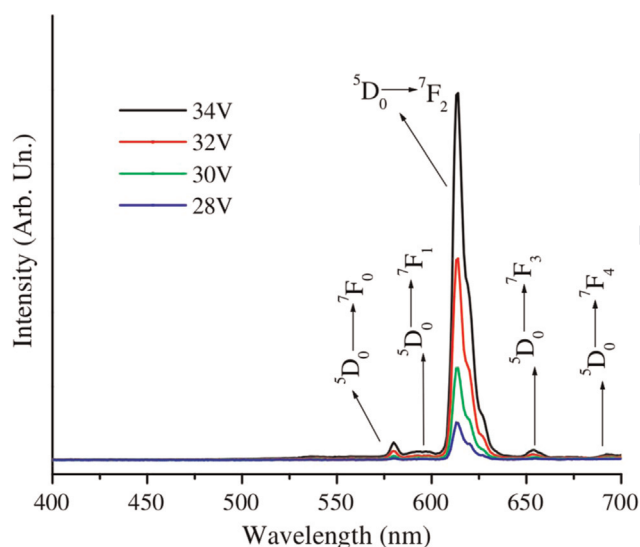


Fig. 12. Electroluminescence spectra of $\text{Eu}(\text{DBM})_3\text{phen}$ -based OLED recorded upon applying different voltages. (Reproduced with permission from Ref. [150].)

must be dominant in such materials.

Several other interesting investigations discussing to different extents ET processes involving lanthanide compounds have been reported in the last decade [18–119].

2.2. Upconversion

Upconversion is related to sequential energy transfer processes involving different lanthanide ions, from which low-energy photons can be converted into high-energy ones [120]. Upconversion techniques can be employed to understand several non-linear optical processes observed in lanthanide-doped solids. It can take place via ion-to-ion energy transfer, although other mechanisms are also possible (e.g. excited state absorption), as discussed below. Malta [10] has given an important contribution in this field by investigating ET processes taking place in inorganic systems with lanthanide ions. In his work, fundamentals of intramolecular and ion-to-ion energy transfer through the Coulomb and exchange

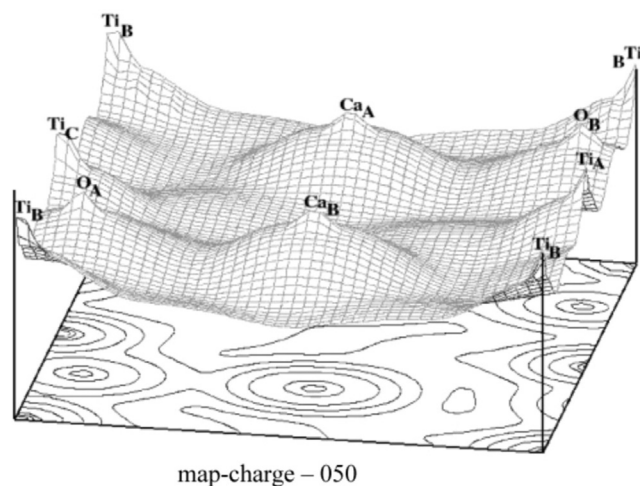


Fig. 13. Charge density contour and surface plots (100) for Ti-O displacements inside a perovskite doped with Sm^{3+} . (Reproduced with permission from Ref. [185].)

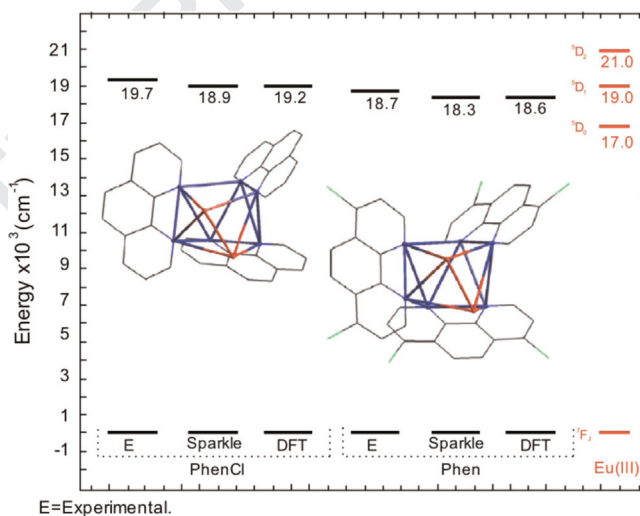


Fig. 14. Comparison between theoretical and experimental energies of the triplet excited states showing the influence of the theoretical method used in the geometry optimization step (Sparkle versus DFT) and of the presence of chlorine atoms in the Phen ligands to form the PhenCl ligand. (Reproduced with permission from Ref. [48].)

mechanisms have been discussed. The shielding effects produced by the filled 5s and 5p sub-shells of the lanthanide ion and the forced electric dipole intensity parameters of 4f–4f transitions were reported to play an essential role in the description of those processes [10]. In the case of ion-to-ion energy transfer, the exchange mechanism was shown to be irrelevant for ion–ion distances larger than 4 Å.

The work of Pires et al. [121] about luminescent and morphological studies with yttrium oxide compounds doped with ytterbium and erbium has received much attention since its publication in 2005. It has shown how to tune particle size and shape by appropriate choice of the oxide precursor. Energy transfer taking place from Yb^{3+} ($^2\text{F}_{5/2}$) to Er^{3+} ($^4\text{I}_{11/2}$) was responsible for the appearance of a red emission assigned to the $^4\text{F}_{9/2} \rightarrow ^4\text{I}_{15/2}$ transition of Er^{3+} . There is an improvement of intensity for samples having smaller crystallite sizes and a higher percentage of Yb^{3+} . On the other side, green emission from Er^{3+} ($^4\text{S}_{3/2} \rightarrow ^4\text{I}_{15/2}$) also generated after energy transfer from Yb^{3+} was observed to become more intense for lower percentages of Yb^{3+} (Fig. 6).

Upconversion studies involving germanate glasses in ternary

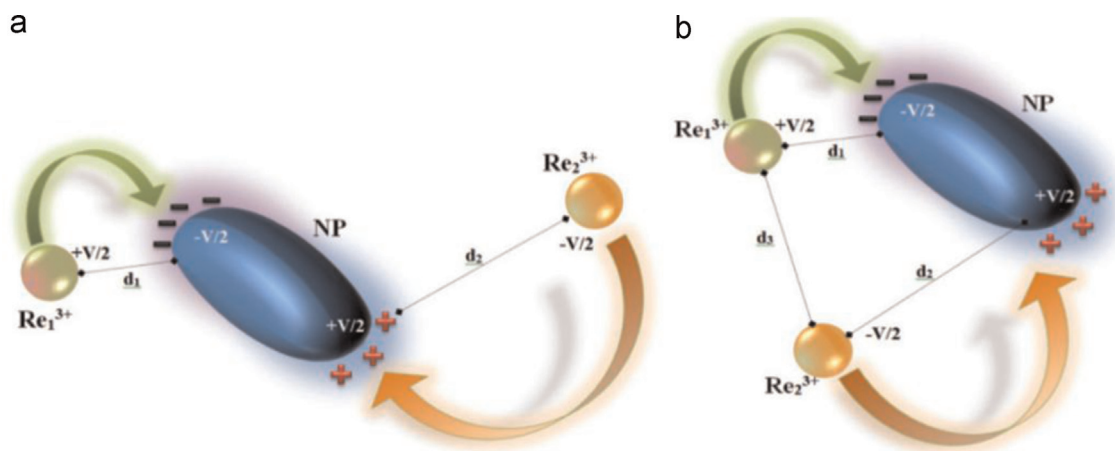


Fig. 15. Schematic representation of the lanthanide ion-NP interactions in two geometrically different scenarios. (Reproduced with permission from Ref. [82].) The label "Re" represents rare earth ions and not Rhenium.

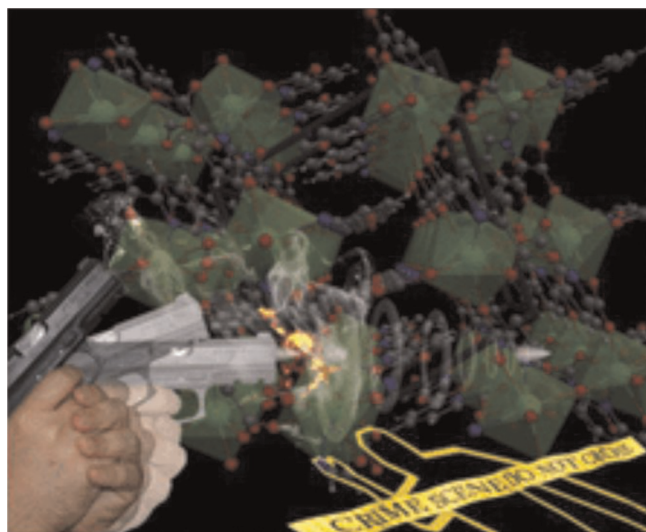


Fig. 16. Representation of the Ln-based MOF (Ln=Yb³⁺ and Tb³⁺) used as photoluminescence marker for gunshot residues. (Reproduced with permission from Ref. [138].)

GeO₂-PbO-Bi₂O₃ and binary GeO₂-Bi₂O₃ systems containing Er₂O₃ (0.1–1.0 wt%) were reported by Kassab et al. [122], where Judd-Ofelt parameters were also calculated. These materials exhibit upconverted green emissions at 530 and 550 nm, and red emission at 668 nm, under excitation at 980 nm. These upconverted emissions are due to excited state absorptions occurring in the Er³⁺ ion, where one low-energy photon (980 nm) is first absorbed (⁴I_{15/2} → ⁴I_{11/2}) and then a second one of the same energy is absorbed again (⁴I_{11/2} → ⁴F_{7/2}), as shown in Fig. 7. These systems were reported to be excellent candidates for applications in nonlinear optics, optical amplification and upconversion optical fiber devices [122].

Da Silva et al. [123] have carried out upconversion luminescence studies in Nd³⁺-doped PbO-GeO₂ glasses containing silver nanoparticles (NPs), whose mechanism of excited state absorptions is shown in Fig. 8. The presence of silver NPs in the lead-germanate glasses was shown to enhance the intensity of the Nd³⁺ emission due to the increase of the local field on the Nd³⁺ ions located in the vicinity of the NPs. The enhancement of the optical properties of Er³⁺-doped glasses by metallic NPs is also discussed in Ref. [124]; this effect is being attributed to energy

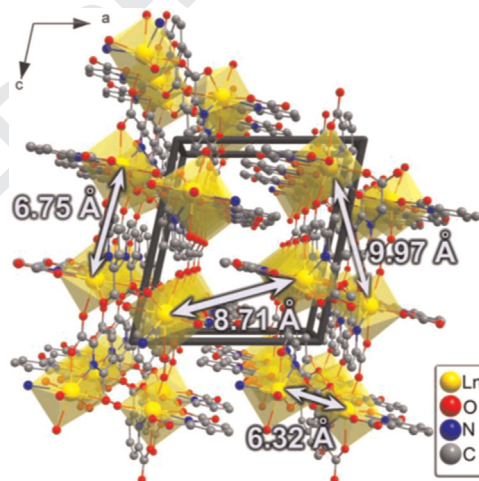


Fig. 17. Schematic representation of the Ln-based MOF material investigated (Ln=Eu³⁺ and Tb³⁺). (Reproduced with permission from Ref. [20].)

transfer from the NP to the Er³⁺ ion.

Several investigations on lanthanide-based upconverted luminescence discussed in the frame of different technological applications have been reported with Eu³⁺ and Tb³⁺ [90], Tm³⁺/Ho³⁺ [87], Er³⁺ and Er³⁺/Yb³⁺ [108], among others [19,20,23,24,26,27,29,33–35,41–44,50,52,60,65,69,70,82–84,95,97,99,100,102,103,125–147].

2.3. Glasses, LEDs, OLEDs and crystals

Great part of the analyzed papers deal with the investigation of glasses in the frame of several subtopics, such as materials science, optics, physics, chemistry, science technology, engineering, metallurgy and metallurgical engineering, spectroscopy, crystallography, biochemistry or molecular biology [148–195]. Several of them discuss upconversion or ET phenomena. For instance, studies involving upconversion in glasses were reported by Kassab et al. [123], Barbosa et al. [19] and Jakutis et al. [24].

Jacinto et al. [148] have reported a relatively simple method to determine emission quantum efficiency (η) and thermo-optical coefficients. This method does not require a direct comparison between the luminescence properties of reference and sample and is based on the normalized-lifetime thermal lens (TL) technique. This newly developed TL-based technique explores the linear

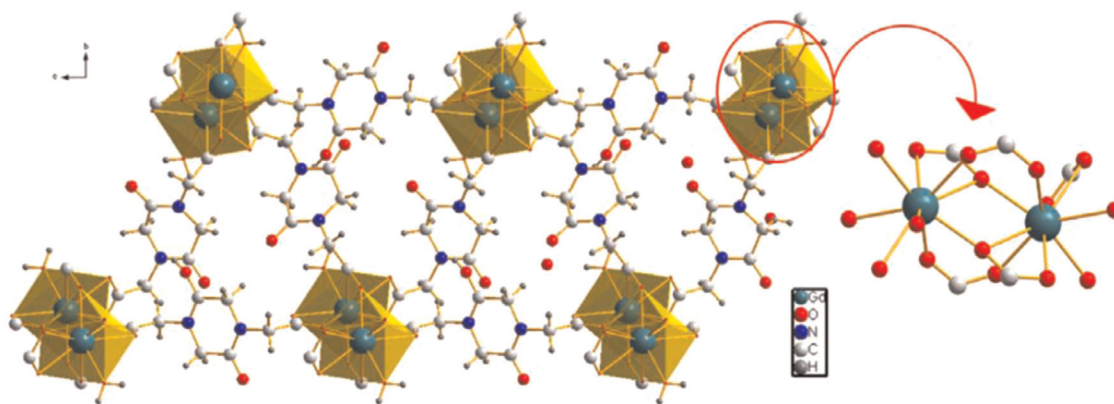


Fig. 18. 2D view of a coordination polymer based on the 2,5-piperazinedione-1,4-diacetic acid (H₂PDA) with lanthanide ions. (Reproduced with permission from Ref. [200].)

dependence of the TL signal with the experimental lifetimes for different samples exhibiting different lanthanide concentrations. η values of Phosphate (Q-98), fluorozirconate (ZBLAN), and fluorindate (PGIZCa) glasses doped with Nd³⁺ as a function of [Nd³⁺] were determined, as shown in Fig. 9. The authors reported a very good agreement between the η values found for the investigated glasses with those values calculated using standard techniques (e.g. multiwavelength or reference sample).

Gouveia-Neto et al. [149] have reported the synthesis of phosphors/vitroceraic hybrid materials composed by 75SiO₂:25PbF₂ doped with Eu³⁺, Tb³⁺, and Sm³⁺ using the sol-gel process. The lanthanide ions were doped into the low-phonon-energy PbF₂ nanocrystals, which were in turn dispersed in the aluminosilicate matrix. The single-doped phosphors presented typical emission spectra in the red, green and blue regions. The appropriate combination of the dopants inside the material gave rise to the emission colors depicted in Fig. 10, which shows the CIE (*Commission Internationale de l'Eclairage*) chromaticity diagram for different samples. One can see that the prepared glasses are all inside the white-light boundary represented by the white ellipse (Fig. 10), revealing that they are promising materials for use in white-light LED technological applications.

A recent investigation reported by Rivera et al. [40] discussed the best way to achieve efficient coupling between Er³⁺-doped germanium-tellurite TeO₂-Na₂O-ZnO-GeO₂-Er₂O₃ glasses and Au NPs. The upconversion process exhibited by the Er³⁺ ion inside the glass matrix was shown to be favored by the presence of NPs. The Au NPs, which were ca 1.6 nm in diameter and were grown inside the germanium-tellurite glass during thermal treatments, were responsible for enhancing the local field in the surroundings of the Er³⁺ ion. Energy transfer between the NP and the lanthanide ion was also observed. Fig. 11 shows a schematic representation of all photophysical processes (absorption, energy transfer, emission) occurring in the Er³⁺:Au-NP glass.

Few investigations have reported the use of lanthanides in LEDs or Organic Light-Emitting Diodes (OLEDs). Cremona et al. [150] have synthesized new quinoline-based lanthanide complexes (Ln=Y³⁺, La³⁺ and Lu³⁺) and determined their electroluminescent, electrical and optical properties in OLED devices. Emissions assigned to the organic ligand in the green and yellow spectral ranges were observed for the prepared complexes. When using a β -diketonate Eu³⁺ complex as the emitting layer of the OLED, the typical emission lines of Eu³⁺ were observed, as shown in Fig. 12. The Eu³⁺-based OLED exhibited a turn-on voltage of 15 V and the maximum luminance value of 20 cd/m² at 0.2 mA. Santos et al. [189] have reported the use of a β -diketonate Eu³⁺ complex with 1,10-phenanthroline to build OLEDs. Other investigations have reported the fabrication of LED and OLED devices with lanthanides [35,97,159,168,182,190–195].

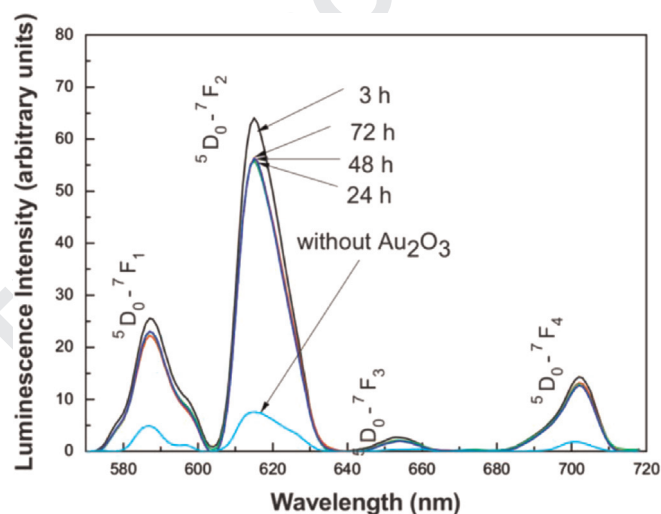


Fig. 19. Sensitization of the Eu³⁺ emission caused by its interaction with Au NPs. (Reproduced with permission from Ref. [177].)

The luminescent properties of lanthanide-doped amorphous silicon nanocrystals (Ln=Er³⁺, Nd³⁺, Eu³⁺ and Tb³⁺) have been reported by Tessler [151]. Other lanthanide systems based on crystals [152–155], silica [156], kaolinite [157], PMMA [158], alumina [159,160], and poly- β -hydroxybutyrate (PHB) [161] have been discussed. The preparation of glasses containing lanthanides and their luminescence properties have been discussed in the literature for several different systems and discussed in the frame of a wide range of applications [10,20,26,29,34,35,38–42,44,50–52,54,60,61,65,67,69,70,73,82–84,87,90,92,94,95,97–100,102,103,123–132,134,136,141–145,162–188].

2.4. Theoretical methods and methodologies

The theoretical methods used to treat lanthanide compounds are usually related to predictions of photophysical properties, description of the electronic structure, numerical calculation of emission quantum yields, estimates of ET rates, calculation of Judd-Ofelt intensity parameters, prediction of the ground state geometry using semiempirical models as well as first principles methods and description of the chemical environment around the lanthanide ion.

Freire et al. [37] have recently presented a detailed investigation of the photoluminescence of a Eu³⁺ complex, where different theoretical tools were introduced step-by-step to predict several properties of interest. In addition, the Sparkle Model and Density Functional Theory (DFT) methods were used to predict ground

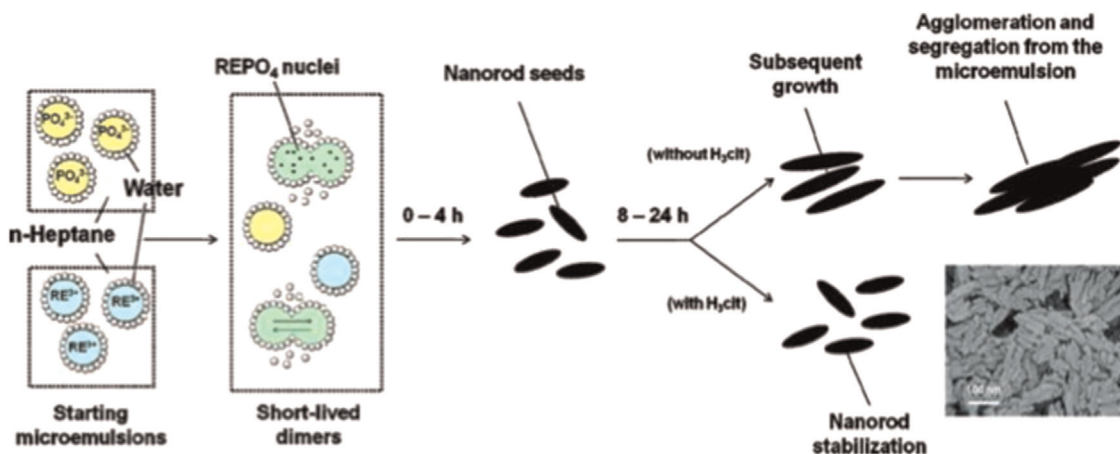


Fig. 20. The synthetic route used for preparing lanthanide-based nanorods. (Reproduced with permission from Ref. [139].)

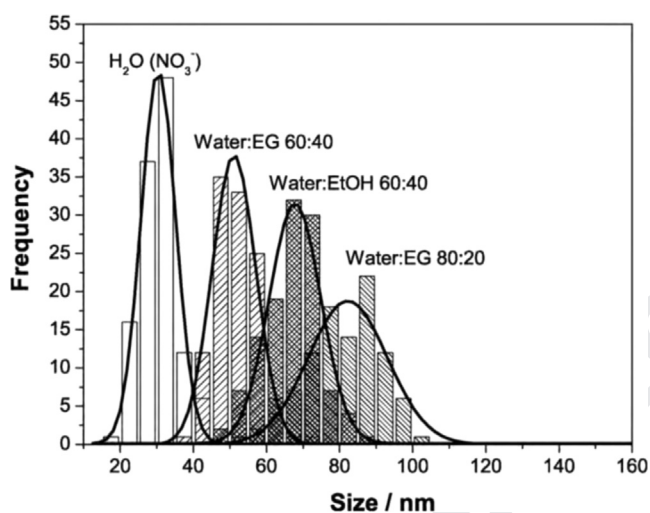


Fig. 21. Fine tuning of the diameter of $Gd_2O_3:Eu^{3+}$ NPs. (Reproduced with permission from Ref. [229].)

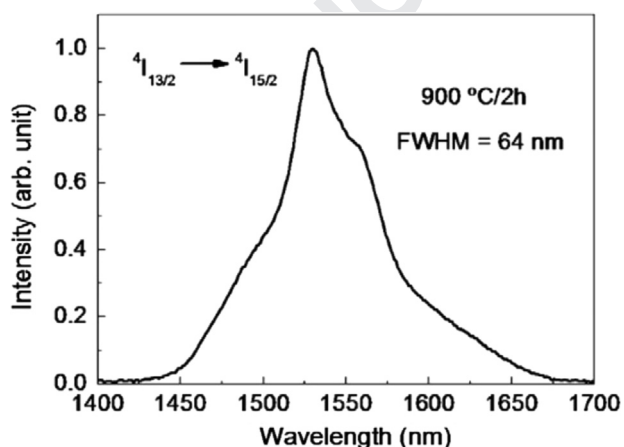


Fig. 22. Broadband emission of Er^{3+} -doped $SiO_2-Ta_2O_5$ nanocomposite films. (Reproduced with permission from Ref. [141].)

state geometries. Prediction of excited states (singlets and triplets) of the ligands is carried out by using the semiempirical INDO/S-CIS model and the Judd-Ofelt theory is used to calculate intensity parameters based on the model developed by Malta et al. [196] and ϕ 's were predicted. Malta et al. [36] have also used such

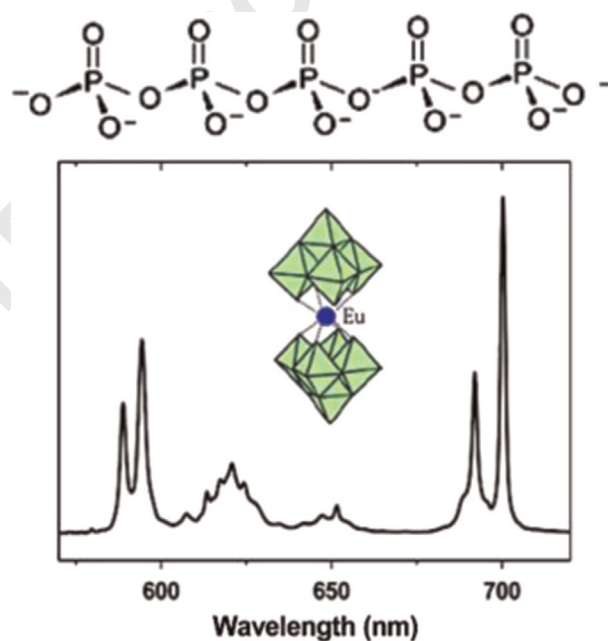


Fig. 23. Emission spectra of the luminescent Eu^{3+} -containing polyphosphate-tungstate aqueous colloidal systems. (Reproduced with permission from Ref. [53].)

theoretical tools to make a systematic investigation of the luminescence properties of several β -diketonate Eu^{3+} complexes and could observe a linear relationship between theoretical and experimental ϕ values.

The use of periodic DFT methods to predict ground state geometries, partial charges and density-of-states of perovskite doped with Sm^{3+} was reported by Figueiredo et al. [185]. In this study, the effect of the order-disorder transition of the simulated material over its photoluminescent properties was discussed. The authors have concluded that the luminescence exhibited by the Sm^{3+} -doped material in its disordered state is independent of the Sm^{3+} emission. The ordered and disordered structures were associated to the Ti sites in the TiO_6 and TiO_5 clusters, respectively. The Ti-O displacements of the investigated material were reported to influence the formation of TiO_6 clusters, as revealed by the charge density contour plots (see Fig. 13).

Belian et al. [48] have investigated the luminescent properties of Eu^{3+} complexes, in which the 1,10-phenanthroline (Phen) ligand was substituted at the 4 and 7 positions by chlorine atoms. Their intensity parameters were calculated using the model

developed by Malta et al. [196] and compared with experimental results. Molecular structures of the complexes and excited states were predicted using the Sparkle model and DFT method, and the INDO/S-CI model, respectively. Fig. 14 shows the theoretical (Sparkle/DFT) and experimental triplet state energies of the investigated Eu^{3+} systems. The energies predicted for the triplet states of the complexes followed the experimental data, where a blue shift was observed upon chlorinating the ligand.

Among the different theoretical methods developed to investigate luminescent lanthanide systems, many of them being developed by Prof. Malta, the prediction of ground state geometries plays a very important role, since the optimized atomic coordinates are always the starting point for other theoretical models. The Sparkle model [197] is often used for this purpose and it is implemented in the MOPAC program, which, until very recently, could not be run in parallel, therefore limiting its range of application. However, the Sparkle model can now be used to treat much larger lanthanide systems due to the pioneering work of Rocha et al. [198], which made possible running parallel computations to optimize lanthanide compounds using the MOPAC program.

The recent publication of Rivera et al. [82] reports the incorporation of either Ag or Au NPs into Er^{3+} - Tm^{3+} -doped tellurite glasses and makes a complete description on how the photon-plasmon interaction can help control the radiative properties of these materials. The Hamiltonian used to describe the oscillator strength or the local crystal field in the material expresses the dynamic coupling mechanism which is associated with the interaction between the NP and the lanthanide ion. The noble metal NPs enhance the emission intensity and expand the bandwidth of the luminescence spectrum of the doped glasses by both the combined emission of Er^{3+} and Tm^{3+} ions and efficient ET processes. The latter are thoroughly described in Ref. [82]. The schematic representation of these interactions is shown in Fig. 15.

Other investigations reporting the use and/or development of theoretical models or methodologies have been discussed in the literature [8,12,13,15,17,28,43,55,56,86,199–207].

2.5. Thin films, metal-organic frameworks and coordination polymers

The application of metal-organic frameworks (MOFs) with Yb^{3+} and Tb^{3+} as optical problems to detect gunshot residues has been reported [138]. These luminescent markers (Fig. 16) have allowed one to identify ammunition origins and to make caliber recognition by adding them directly to conventional gunpowder in a ratio of 10 wt% before reassembling the bullet. The luminescence associated to each marker arises from excitation of Yb^{3+} at 980 nm, where ground and excited state absorptions take place using two photons, followed by the upconverted Tb^{3+} emission in the visible spectral range.

Rodrigues et al. [20] have carried out an experimental and theoretical investigation of a Ln-based MOF ($\text{Ln}=\text{Eu}^{3+}$ and Tb^{3+}), where ion-to-ion ($\text{Tb}^{3+} \rightarrow \text{Eu}^{3+}$) ET takes place. In this system (Fig. 17), the Tb^{3+} emission (${}^5\text{D}_4 \rightarrow {}^7\text{F}_j$) is strongly quenched concomitantly with the appearance of an intense red emission from Eu^{3+} (${}^5\text{D}_0 \rightarrow {}^7\text{F}_j$). The model developed by Malta et al. [2,10] was used to predict that the dipole-dipole and dipole-quadrupole ET mechanisms were the dominant ones to describe the $\text{Tb}^{3+} \rightarrow \text{Eu}^{3+}$ ET process.

Alves et al. [200] have synthesized 3D coordination polymers based on the 2,5-piperazinedione-1,4-diacetic acid (H_2PDA) and Eu^{3+} and Gd^{3+} ions. These materials were shown to be crystalline with chemical formula $[\text{Ln}(\text{PDA})_{1.5}(\text{H}_2\text{O})]$ and presented the characteristic red emission from the transition ${}^5\text{D}_0 \rightarrow {}^7\text{F}_2$ in the case of the Eu^{3+} -based polymer. The energy transfer (ligand-to-metal) in

this material was also described using the model developed by Malta et al. [2,3], and Judd-Ofelt intensity parameters were obtained for the prepared material. The 2D view of this coordination polymer is shown in Fig. 18.

The work of Kai et al. [11], discussed on a previous section is an example of a lanthanide-based thin film material, where PMMA was used as a matrix and Eu^{3+} and Tb^{3+} complexes as luminescent species. The PMMA promotes the close interaction between those complexes, from where intermolecular ET could be observed, as well as it enhances the overall luminescence intensity since it acts as a co-sensitizer. The materials have exhibited emission with green, red, and intermediate colors, which were fine-tuned by optimizing composition and excitation wavelength.

Other investigations involving thin films, MOFs and coordination polymers have been reported in the literature [18,63,73,78,88,94,133,141,169,178,182,185,201,208–228].

2.6. Nanoparticles, nanorods and sol-gel process

The work of Pires et al. [121] is related to morphological and luminescent studies on nanosized yttrium oxide NPs doped with Yb^{3+} and Er^{3+} . Luminescent NPs as small as 12 nm have been reported, but depending on the synthetic protocol used, 90 nm-diameter NPs could also be made. The Ln-doped NPs reported in Ref. [121] were shown to be excellent candidates for use in immunoassays involving analytes of nanometer size, greatly contributing to the development of the upconversion phosphor technology.

Au and Ag NPs with 10 nm in diameter were used by Kassab et al. [177] to enhance the photoluminescence of germanate glasses doped with Eu^{3+} for wavelengths overlapping with the surface plasmon band of the NPs. The presence of noble-metal NPs has shown to influence the forced electric-dipole and magnetic-dipole transitions starting from the ${}^5\text{D}_0$ level of Eu^{3+} , strongly increasing the luminescence intensity from f-f transitions (Fig. 19). This effect was attributed to ET from the NPs to the Eu^{3+} ion and to the confined electromagnetic field near the NPs.

Nanorods of 50–100 nm in length with composition of MPO_4 : Ln^{3+} ($\text{M}=\text{La}$, Y , Gd , or Yb ; $\text{Ln}=\text{Eu}$, Tm , or Er) were prepared by Serra et al. [139] using a synthetic route based on the reversion microemulsion method (Fig. 20). The Eu^{3+} -doped nanorods have exhibited the characteristic red emission mainly due to the ${}^5\text{D}_0 \rightarrow {}^7\text{F}_2$ transition and the large number of Stark levels have revealed that the Eu^{3+} ion occupied low-symmetry sites. The LaPO_4 : Tm^{3+} and YPO_4 : Tm^{3+} nanorods have exhibited a blue emission with high color purity, while the one with composition YbPO_4 : Er^{3+} showed an upconverted red emission upon excitation at 980 nm. The authors have reported that these new materials can be designed to exhibit desired properties for a broad range of applications.

Sigoli et al. [229] have shown how to tune the size of luminescent Gd_2O_3 : Eu^{3+} NPs prepared using a modified homogeneous precipitation method. The NP size can be controlled to have diameters in the range 30–90 nm (Fig. 21) by means of appropriate selection of the counter anion types (nitrate versus chloride), chemical nature of the alcohol (high versus low viscosity) and dielectric constant of the reaction medium. The synthesized materials have exhibited a strong red luminescence due to the ${}^5\text{D}_0 \rightarrow {}^7\text{F}_2$ transition of Eu^{3+} . The results indicate that the strong interaction between nitrate anions and Gd^{3+} is a key to promote the stabilization of small NPs.

The article of Ferrari et al. [141] reports the preparation, structural analyses and optical properties of Er^{3+} -doped SiO_2 - Ta_2O_5 nanocomposite films obtained by the sol-gel process. The Er^{3+} -doped nanocomposite annealed at 900 °C consisted of Ta_2O_5 NPs with average diameter around 2.4 nm and a very small size

distribution, dispersed in the SiO₂ amorphous matrix. It was observed a broad band emission (Fig. 22) peaked at around 1532 nm and assigned to the ⁴I_{13/2} → ⁴I_{15/2} transition of Er³⁺. The photoluminescence lifetime measured for this band was 5.4 ms. The optimal structural and optical properties found for this material suggest it may be used in active planar waveguides.

Pereira et al. [230] have presented a non-hydrolytic sol-gel route to prepare Y₃Al₅O₁₂ (YAG) phosphors doped with Eu³⁺, Tb³⁺, and Tm³⁺ in order to generate white light. The typical luminescence bands of Eu³⁺ (⁵D₀ → ⁷F₂), Tb³⁺ (⁵D₄ → ⁷F₅) and Tm³⁺ (¹D₂ → ³H₄) corresponding to red, green and blue emissions, respectively, were observed in the prepared material. The green emission was more intense and narrow in relation to the red and blue emissions, which was reported to be due to differences in the size of the three incorporated ions. The results have reinforced the important role of the non-hydrolytic sol-gel route to efficiently prepare red-green-blue phosphors for applications in display technology and white light generation.

Ribeiro et al. [53] have reported the synthesis of europium sodium-phosphate-tungstate colloidal systems in which the Eu³⁺ ions occupy cage-like sites composed of phosphate groups from the metaphosphate chains. The tungstate addition has led to the formation of W-rich nanoparticles (5–10 nm in diameter) and removal of water molecules from the first coordination sphere of Eu³⁺. The authors also report an efficient energy transfer from the O–W charge-transfer states to Eu³⁺ at room temperature and in aqueous systems. The strong ligand field around the lanthanide ion, together with the absence of water in its first coordination sphere was shown to be responsible for a very high emission quantum efficiency value of 80%. A similar strategy was used by De Camargo et al. [74], where complete removal of water molecules from the first coordination sphere of a nine-coordinated Eu³⁺ complex gave rise to ϕ = 85% in solution. Fig. 23 shows the emission spectra of the luminescent Eu³⁺-containing polyphosphate-tungstate aqueous colloidal systems.

Investigations of luminescent lanthanide complexes in solution bearing different applications were also reported in the literature [231–234]. Other investigations on the luminescence properties of lanthanide-based systems with NPs, nanorods or systems assembled using sol-gel techniques have been reported in the literature [17,22,27,33,34,38,40,43,45,49,50,60,61,67,73,78,82,88,93,107,127–129,133,136,140,142,143,149,163,169–171,176,178,201,211,212,214,235–246].

3. Conclusions

The overview of the representative research carried out in Brazil in the last decade in the field of luminescent lanthanide systems was presented, being discussed in the frame of several subtopics. Recent publications involving glasses, upconversion systems, coordination polymers, modeling of ET processes, among others, were discussed here. The strong contribution of Prof. Oscar L. Malta to the Brazilian science and technology is notorious, not only because he was co-author of more than 10% of the papers analyzed in this review, but also because his many important theoretical models and findings were both used and cited in almost all publications. This makes evident the great importance of Prof. Malta for the research carried out in the field of lanthanide spectroscopy in Brazil, and it does not rule out his international leadership in this field, which was out of the scope of this review.

Acknowledgments

The authors are grateful to CNPq (Grant 305082/2013-2), CAPES

(Grant A061_2013) and FAPESP (Grants 2014/01174-5 and 2014/02071-5) for financial support.

References

- [1] G.F. de Sá, O.L. Malta, C.D. Donega, A.M. Simas, R.L. Longo, P.A. Santa-Cruz, E.F. da Silva, *Coord. Chem. Rev.* 196 (2000) 165.
- [2] O.L. Malta, F.R. Gonçalves e Silva, *Spectrochim. Acta A* 54 (1998) 1593.
- [3] O.L. Malta, F.R. Gonçalves e Silva, R. Longo, *Chem. Phys. Lett.* 307 (1999) 518.
- [4] a) O.L. Malta, *Chem. Phys. Lett.* 87 (1982) 27;
b) O.L. Malta, *Chem. Phys. Lett.* 88 (1982) 353.
- [5] O.L. Malta, S.J.L. Ribeiro, M. Faucher, P. Porcher, *J. Phys. Chem. Solids* 52 (1991) 587.
- [6] O.L. Malta, H.J. Batista, L.D. Carlos, *Chem. Phys.* 282 (2002) 21.
- [7] G.B. Rocha, R.O. Freire, A.M. Simas, J.J.P. Stewart, *J. Comput. Chem.* 27 (2006) 1101.
- [8] W.M. Faustino, O.L. Malta, G.F. de Sá, *J. Chem. Phys.* 122 (2005) 054109.
- [9] E.E.S. Teotonio, H.F. Brito, M.C.F.C. Felinto, L.C. Thompson, V.G. Young, O.L. Malta, *J. Mol. Struct.* 751 (2005) 85.
- [10] O.L. Malta, *J. Non-Cryst. Solids* 354 (2008) 4770.
- [11] J. Kai, M.C.F.C. Felinto, L.A.O. Nunes, O.L. Malta, H.F. Brito, *J. Mater. Chem.* 21 (2011) 3796.
- [12] A.P. Souza, L.C.V. Rodrigues, H.F. Brito, S. Alves Jr., O.L. Malta, *J. Lumin.* 130 (2010) 181.
- [13] W.M. Faustino, O.L. Malta, G.F. de Sá, *Chem. Phys. Lett.* 429 (2006) 595.
- [14] E.E.S. Teotonio, H.F. Brito, H. Viertler, W.M. Faustino, O.L. Malta, G.F. de Sá, M.C.F.C. Felinto, R.H.A. Santos, M. Cremona, *Polyhedron* 25 (2006) 3488.
- [15] R.O. Freire, F.R.G.E. Silva, M.O. Rodrigues, M.E. de Mesquita, N.B. da Costa Jr., *J. Mol. Model.* 12 (2005) 16.
- [16] E.E.S. Teotonio, G.M. Fett, H.F. Brito, W.M. Faustino, G.F. de Sá, M.C.F.C. Felinto, R.H.A. Santos, *J. Lumin.* 128 (2008) 190.
- [17] R.S. Viana, E.H.L. Falcão, J.D.L. Dutra, N.B. da Costa, R.O. Freire, S. Alves, *J. Photochem. Photobiol. A* 281 (2014) 1.
- [18] E.R. Souza, I.G.N. Silva, E.E.S. Teotonio, M.C.F.C. Felinto, H.F. Brito, *J. Lumin.* 130 (2010) 283.
- [19] A.J. Barbosa, F.A. Dias Filho, Y. Messaddeq, S.J.L. Ribeiro, R.R. Gonçalves, S.R. Lüthi, A.S.L. Gomes, *J. Non-Cryst. Solids* 352 (2006) 3636.
- [20] M.O. Rodrigues, J.D.L. Dutra, L.A.O. Nunes, G.F. de Sá, W.M. de Azevedo, P. Silva, F.A.A. Paz, R.O. Freire, S.A. Júnior, *J. Phys. Chem. C* 116 (2012) 19951.
- [21] E.R. dos Santos, R.O. Freire, N.B. da Costa Jr., F.A.A. Paz, C.A. de Simone, S.A. Júnior, A.A.S. Araújo, L.A.O. Nunes, M.E. de Mesquita, M.O. Rodrigues, *J. Phys. Chem. A* 114 (2010) 7928.
- [22] H.H.S. Oliveira, M.A. Cebim, A.A. Da Silva, M.R. Davolos, *J. Alloy. Compd.* 488 (2009) 619.
- [23] G.S. Maciel, R.B. Guimarães, P.G. Barreto, I.C.S. Carvalho, N. Rakov, *Opt. Mater.* 31 (2009) 1735.
- [24] J. Jakutis, L. Gomes, C.T. Amancio, L.R.P. Kassab, J.R. Martinelli, N.U. Wetter, *Opt. Mater.* 33 (2010) 107.
- [25] L.C. Courrol, R.E. Samad, *Curr. Pharm. Anal.* 4 (2008) 238.
- [26] V.K. Rai, L.S. Menezes, C.B. de Araújo, *J. Appl. Phys.* 102 (2007) 043505.
- [27] N. Rakov, G.S. Maciel, R.B. Guimarães, I.C.S. Carvalho, *Mater. Chem. Phys.* 123 (2010) 199.
- [28] S.P. Vila Nova, H.J. Batista, S. Alves Jr., C.M. Donega, R.L. Longo, G.F. de Sá, L.C. Thompson, *J. Lumin.* 118 (2006) 83.
- [29] A.S. Gouveia-Neto, E.B. da Costa, L.A. Bueno, S.J.L. Ribeiro, Y. Messaddeq, *J. Lumin.* 116 (2006) 52.
- [30] W.G. Quirino, C. Legnani, R.M.B. dos Santos, K.C. Teixeira, M. Cremona, M.A. Guedes, H.F. Brito, *Thin Solid Films* 517 (2008) 1096.
- [31] L.C. Courrol, M.H. Bellini, L.V.G. Tarelho, F.R.O. Silva, R.D. Mansano, L. Gomes, N.D. Vieira Jr., Nestor Shor, *Anal. Biochem.* 355 (2006) 140.
- [32] P.P. Lima, M.M. Nolasco, F.A.A. Paz, R.A.S. Ferreira, R.L. Longo, O.L. Malta, L.D. Carlos, *Chem. Mater.* 25 (2013) 586.
- [33] N. Rakov, J.A.B. Barbosa, R.B. Guimarães, G.S. Maciel, *J. Alloy. Compd.* 534 (2012) 32.
- [34] L.R.P. Kassab, M.E. Camilo, C.T. Amancio, D.M. da Silva, J.R. Martinelli, *Opt. Mater.* 33 (2011) 1948.
- [35] F.A. Bomfim, J.R. Martinelli, L.R.P. Kassab, N.U. Wetter, J.J. Neto, *J. Non-Cryst. Solids* 354 (2008) 4755.
- [36] W.M. Faustino, S.A. Junior, L.C. Thompson, G.F. de Sá, O.L. Malta, A.M. Simas, *J. Quant. Chem.*, 103, (2005) 572.
- [37] J.D.L. Dutra, R.O. Freire, *J. Photochem. Photobiol. A* 256 (2013) 29.
- [38] N.O. Dantas, E.O. Serqueira, M.J.V. Bell, V. Anjos, E.A. Carvalho, S.A. Lourenço, M.A. Pereira-da-Silva, *J. Lumin.* 131 (2011) 1029.
- [39] A.F.H. Librantz, L. Gomes, G. Pairier, S.J.L. Ribeiro, Y. Messaddeq, *J. Lumin.* 128 (2008) 51.
- [40] V.A.G. Rivera, Y. Ledemi, S.P.A. Osorio, D. Manzani, F.A. Ferri, S.J.L. Ribeiro, L.A. O. Nunes, E. Marega Jr., *J. Non-Cryst. Solids* 378 (2013) 126.
- [41] A.S. Gouveia-Neto, L.A. Bueno, R.F. Nascimento, E.A. Silva, E.B. da Costa, *Mater. Lett.* 63 (2009) 1999.
- [42] V.K. Rai, C.B. de Araújo, Y. Ledemi, B. Bureau, M. Poulain, Y. Messaddeq, *J. Appl. Phys.* 106 (2009) 103512.
- [43] R.A.S. Ferreira, M. Nolasco, A.C. Roma, R.L. Longo, O.L. Malta, L.D. Carlos, *Chem. Eur. J.* 18 (2012) 12130.

- 1 [44] I.A.A. Terra, A.S.S. de Camargo, L.A. de, O. Nunes, R.A. Carvalho, M.S. Li, J. Appl. Phys. 100 (2006) 123103.
- 2 [45] P.N.M. dos Anjos, E.C. Pereira, Y.G. Gobato, J. Alloy. Compd. 391 (2005) 277.
- 3 [46] W.M. Faustino, G.F. de Sá, O.L. Malta, M.C.F. Soares, D. Windmüller, J. C. Machado, Chem. Phys. Lett. 424 (2006) 63.
- 4 [47] N.B.D. Lima, S.M.C. Gonçalves, S.A. Junior, A.M. Simas, Sci. Rep. 3 (2013) 2395.
- 5 [48] M.F. Belian, H.J. Batista, A.G.S. Bezerra, W.E. Silva, G.F. de Sá, S. Alves Jr., Chem. Phys. 381 (2011) 29.
- 6 [49] E. Silva, M.F. Belian, R.O. Freire, G.F. de Sá, S. Alves Jr., J. Phys. Chem. A 114 (2010) 10066.
- 7 [50] L.A. Gómez, L.S. Menezes, C.B. de Araújo, R.R. Gonçalves, S.J.L. Ribeiro, Y. Messaddeq, J. Appl. Phys. 107 (2010) 113508.
- 8 [51] R.S. Cunha, J.L. Ferrari, D.C. de Oliveira, L.J.Q. Maia, A.S.L. Gomes, S.J.L. Ribeiro, R.R. Gonçalves, Mater. Chem. Phys. 136 (2012) 120.
- 9 [52] E.O. Serqueira, N.O. Dantas, V. Anjos, M.A. Pereira-da-Silva, M.J.V. Bell, J. Lumin. 131 (2011) 1401.
- 10 [53] A.J. Barbosa, L.J.Q. Maia, B. Montanari, R.R. Gonçalves, Y. Messaddeq, R.A. S. Ferreira, L.D. Carlos, S.J.L. Ribeiro, Langmuir 26 (2010) 14170.
- 11 [54] A.S. Gouveia-Neto, L.A. Bueno, R.F. do Nascimento, E.B. da Costa, S.J.L. Ribeiro, Y. Messaddeq, Phys. Chem. Glasses – B 50 (2009) 37.
- 12 [55] M.E. de Mesquita, L.S. Barreto, M.L. Santos, R.O. Freire, J. Non-Cryst. Solids 351 (2005) 394.
- 13 [56] W.M. Faustino, L.A. Nunes, I.A.A. Terra, M.C.F.C. Felinto, H.F. Brito, O.L. Malta, J. Lumin. 137 (2013) 269.
- 14 [57] A.P. Souza, L.C.V. Rodrigues, H.F. Brito, S. Alves Jr., O.L. Malta, Inorg. Chem. Commun. 15 (2012) 97.
- 15 [58] C.C.C. Bejan, G.B. Rocha, R.Q. Albuquerque, F.W.J. Demnitz, G.F. de Sá, S. Alves, J. Lumin. 113 (2005) 79.
- 16 [59] L.C.V. Rodrigues, J. Hölsä, M. Lastusaari, M.C.F.C. Felinto, H.F. Brito, J. Mater. Chem. C 2 (2014) 1612.
- 17 [60] A.S. Gouveia-Neto, L.A. Bueno, E.B. Costa, E.A. Silva Jr., J.L. Ferrari, K.O. Lima, R. R. Gonçalves, Proc. SPIE 7598 (2010) 759806.
- 18 [61] A.S. Gouveia-Neto, A.F. da Silva, L.A. Bueno, E.B. Costa, Proc. SPIE 7954 (2011) 795412.
- 19 [62] L.F. Marques, A.A.B. Cantaruti Jr., C.C. Correa, M.G. Lahoud, R.R. da Silva, S.J. L. Ribeiro, F.C. Machado, J. Photochem. Photobiol. A 252 (2013) 69.
- 20 [63] E.F. de Melo, N.C. Santana, K.G.B. Alves, G.F. de Sá, C.P. de Melo, M. O. Rodrigues, S. Alves Jr., J. Mater. Chem. C 1 (2013) 7574.
- 21 [64] J.G. Santos, J.D.L. Dutra, S. Alves Jr., G.F. de Sá, N.B. da Costa Jr., R.O. Freire, J. Braz. Chem. Soc. 24 (2013) 236.
- 22 [65] A.F. da Silva, D.S. Moura, A.S. Gouveia-Neto, E.A. Silva Jr., L.A. Bueno, E. B. Costa, E.N. Azevedo, Opt. Commun. 284 (2011) 4501.
- 23 [66] J.D.L. Dutra, T.D. Bispo, R.O. Freire, J. Comput. Chem. 35 (2014) 772.
- 24 [67] N.O. Dantas, E.O. Serqueira, V. Anjos, M.J.V. Bell, Appl. Phys. Lett. 101 (2012) 121903.
- 25 [68] A.F.G. do Monte, D.N. Messias, J. Lumin. 130 (2010) 674.
- 26 [69] V. Pilla, E.F. Chillce, E. Rodriguez, T. Catunda, E. Munin, C.L. Cesar, L. C. Barbosa, Proc. SPIE 6116 (2006) 61160X.
- 27 [70] A.S. Gouveia-Neto, E.B. da Costa, L.A. Bueno, Y. Messaddeq, S.J.L. Ribeiro, J. Alloy. Compd. 394 (2005) 24.
- 28 [71] M.F. Belian, G.F. de Sá, S. Alves Jr., A. Galembeck, J. Lumin. 131 (2011) 856.
- 29 [72] E.R. Souza, C.H.F. Zulato, I.O. Mazali, F.A. Sigoli, J. Fluoresc. 23 (2013) 939.
- 30 [73] L.J.Q. Maia, J. Fick, A.C. Hernandez, V.R. Mastelaro, A. Ibanez, Opt. Mater. 34 (2012) 665.
- 31 [74] M.B.S. Botelho, M.D. Gálvez-López, L. De Cola, R.Q. Albuquerque, A.S.S. de Camargo, Eur. J. Inorg. Chem. 2013 (2013) 5064.
- 32 [75] G.M. de Oliveira, A. Machado, G.W. Gomes, J.H.S.K. Monteiro, M.R. Davolos, U. Abram, A. Jagst, Polyhedron 30 (2011) 851.
- 33 [76] R.F. Martins, C.R. Neri, P.C. de Sousa Filho, O.A. Serra, K.T. de Oliveira, Quim. Nova 33 (2010) 2118.
- 34 [77] R.D. Adati, S.A.M. Lima, M.R. Davolos, M. Jafelizzi Jr., J. Alloy. Compd. 488 (2009) 595.
- 35 [78] E.A. Morais, L.V.A. Scalvi, L.P. Ravaro, Phys. Procedia 2 (2009) 353.
- 36 [79] R.A. Silva, G. Tiraó, C. Cusatis, J.P. Andreetta, J. Cryst. Growth 294 (2006) 447.
- 37 [80] E.A. de Morais, L.V.A. Scalvi, M.R. Martins, S.J.L. Ribeiro, Braz. J. Phys. 36 (2006) 270.
- 38 [81] J.H.S.K. Monteiro, A.L.B. Formiga, F.A. Sigoli, J. Lumin. 154 (2014) 22.
- 39 [82] V.A.G. Rivera, Y. Ledemi, M. El-Amraoui, Y. Messaddeq, E. Marega Jr., Opt. Express 22 (2014) 21122.
- 40 [83] J.R. Silva, L.A. Bueno, A.S. Gouveia-Neto, J. Lumin. 154 (2014) 531.
- 41 [84] N. Rakov, S.A. Vieira, R.B. Guimarães, G.S. Maciel, J. Solid State Chem. 211 (2014) 32.
- 42 [85] I.G.N. Silva, H.F. Brito, E.R. Souza, D. Mustafa, M.C.F.C. Felinto, L.D. Carlos, O. L. Malta, Z. Naturforschungs B 69 (2014) 231.
- 43 [86] A.S. Borges, J.G. da Silva, J.D. Ayala, J.D.L. Dutra, N.L. Speziali, H.F. Brito, M. H. Araújo, Spectrochim. Acta Part A: Mol. Biomol. Spectrosc. 117 (2014) 718.
- 44 [87] M. Seshadri, J.A.P. Ferencz Jr., Y.C. Ratnakaram, L.C. Barbosa, Proc. SPIE 8961 (2014) 896139.
- 45 [88] E.R. Souza, I.O. Mazali, F.A. Sigoli, J. Fluoresc. 24 (2014) 203.
- 46 [89] A.P.A. Marques, M.T.S. Tanaka, E. Longo, E.R. Leite, I.L.V. Rosa, J. Fluoresc. 21 (2011) 893.
- 47 [90] J.R. Silva, A.S. Gouveia-Neto, L.A. Bueno, Proc. SPIE 8964 (2014) 89641A.
- 48 [91] M.E. Camilo, E.O. Silva, T.A.A. de Assumpção, L.R.P. Kassab, C.B. de Araújo, J. Appl. Phys. 114 (2013) 163515.
- 49 [92] N.O. Dantas, E.O. Serqueira, A.C.A. Silva, A.A. Andrade, S.A. Lourenço, Braz. J. Phys. 43 (2013) 230.
- 50 [93] P.L. Forster, D.F. Parra, J. Kai, H.F. Brito, A.B. Lugao, Radiat. Phys. Chem. 84 (2013) 47.
- 51 [94] V.A.G. Rivera, F.A. Ferri, J.L.H. Clabel, M.K. Kawamura, M.A. Pereira-da-Silva, L. A.O. Nunes, M.S. Li, E. Mareg Jr, Proc. SPIE 8621 (2013) 86211K.
- 52 [95] G.S. Maciel, N. Rakov, J. Mater. Chem. C 1 (2013) 3563.
- 53 [96] F.A. Silva Jr., H.A. Nascimento, D.K.S. Pereira, E.E.S. Teotonio, H.F. Brito, M.C.F. C. Felinto, J.G.P. Espinola, G.F. Sá, W.M. Faustino, J. Braz. Chem. Soc. 24 (2013) 601.
- 54 [97] A.S. Gouveia-Neto, N.P.S.M. Rios, L.A. Bueno, Proc. SPIE 8240 (2012) 82401C.
- 55 [98] A.F. da Silva, D.S. Moura, A.S. Gouveia-Neto, E.A. Silva-Jr, L.A. Bueno, E. B. Costa, E.N. Azevedo, Proc. SPIE 7912 (2011) 79122G.
- 56 [99] L.A. Bueno, A.S. Gouveia-Neto, R.F. do Nascimento, E.A. da Silva, V.B. do Nascimento, E.B. Costa, Proc. SPIE 6875 (2008) 68750Q.
- 57 [100] A.S. Gouveia-Neto, L.A. Bueno, R.F. do Nascimento, E.A. da Silva, Proc. SPIE 6875 (2008) 87513.
- 58 [101] E.R. Souza, F.A. Sigoli, Quim. Nova 35 (2012) 1841.
- 59 [102] A.S. Gouveia-Neto, L.A. Bueno, R.F. do Nascimento, E.A. da Silva, Proc. SPIE 6890 (2008) 68900L.
- 60 [103] M.J.S. Brandão, C.B. de Araújo, G. Poirier, Y. Messaddeq, M. Poulain, J. Appl. Phys. 99 (2006) 113525.
- 61 [104] I.G.N. Silva, J. Kai, M.C.F.C. Felinto, H.F. Brito, Opt. Mater. 35 (2013) 978.
- 62 [105] F. Fulgêncio, F.C. de Oliveira, F.F. Ivashita, A. Paesano, D. Windmüller, A. Marques-Netto, W.F. Magalhães, J.C. Machado, Spectrochim. Acta Part A 92 (2012) 415.
- 63 [106] L.C.V. Rodrigues, H.F. Brito, J. Hölsä, R. Stefani, M.C.F.C. Felinto, M. Lastusaari, T. Laamanen, L.A.O. Nunes, J. Phys. Chem. C 116 (2012) 11232.
- 64 [107] H. Gallardo, G. Conte, A.J. Bortoluzzi, I.H. Bechtold, A. Pereira, W.G. Quirino, C. Legnani, M. Cremona, Inorg. Chim. Acta 365 (2011) 152.
- 65 [108] P.C. de Sousa Filho, O.A. Serra, J. Lumin. 129 (2009) 1664.
- 66 [109] H. Gallardo, G. Conte, P. Tuzimoto, A. Bortoluzzi, R.A. Peralta, A. Neves, Inorg. Chem. Commun. 11 (2008) 1292.
- 67 [110] P.C. de Sousa Filho, O.A. Serra, J. Fluoresc. 18 (2008) 329.
- 68 [111] F. Trivinho-Strixino, E.E.G. Guimaraes, E.C. Pereira, Mol. Cryst. Liq. Cryst. 485 (2008) 766.
- 69 [112] N. Rakov, G.S. Maciel, J. Lumin. 127 (2007) 703.
- 70 [113] L.C.C. da Silva, T.S. Martins, M. Santos Filho, E.E.S. Teotônio, P.C. Isolani, H. F. Brito, M.H. Tabacniks, M.C.A. Fantini, J.R. Matos, Microporous Mesoporous Mater. 92 (2006) 94.
- 71 [114] E. Niyama, H.F. Brito, M. Cremona, E.E.S. Teotonio, R. Reyes, G.E.S. Brito, M.C.F. C. Felinto, Spectrochim. Acta Part A 61 (2005) 2643.
- 72 [115] T.S. Martins, J.R. Matos, G. Vicentini, P.C. Isolani, J. Therm. Anal. Calorim. 82 (2005) 77.
- 73 [116] C.C.C. Bejan, F.W.J. Demnitz, G.F. de Sá, O.A. Serra, P.H. Menezes, S.A. Júnior, Inorg. Chem. Commun. 9 (2006) 464.
- 74 [117] A.P. Souza, F.A.A. Paz, R.O. Freire, L.D. Carlos, O.L. Malta, S. Alves Jr., G.F. de Sá, J. Phys. Chem. B 111 (2007) 9228.
- 75 [118] M.E.A. Andrade, W.M. Azevedo, V.S.M. Barros, H.J. Khoury, Radiat. Meas. 46 (2011) 1474.
- 76 [119] J.R. Diniz, J.R. Correa, D.A. Moreira, R.S. Fontenele, A.L. de Oliveira, P. V. Abdelnur, J.D.L. Dutra, R.O. Freire, M.O. Rodrigues, B.A.D. Neto, Inorg. Chem. 52 (2013).
- 77 [120] a) F. Auzel, Comptes Rendus Acad. Sci. 262 (1966) 1016; b) F. Auzel, Comptes Rendus Acad. Sci. 263 (1966) 765.
- 78 [121] A.M. Pires, O.A. Serra, M.R. Davolos, J. Lumin. 113 (2005) 174.
- 79 [122] L.R.P. Kassab, A.O. Preto, W. Lozano, F.X. de Sá, G.S. Maciel, J. Non-Cryst. Solids 351 (2005) 3468.
- 80 [123] D.S. da Silva, T.A.A. de Assumpção, L.R.P. Kassab, C.B. de Araújo, J. Alloy. Compd. 586 (2014) 5516.
- 81 [124] C.B. de Araújo, D.S. da Silva, T.A.A. de Assumpção, L.R.P. Kassab, D.M. da Silva, Sci. World J. 2013 (2013) 385193.
- 82 [125] S. Meruva, B.L. Carlos, J.A.F.J. Peres, Proc. SPIE 8961 (2014) 896132.
- 83 [126] M. Seshadri, E.F. Chillce, J.D. Marconi, F.A. Sigoli, Y.C. Ratnakaram, L. C. Barbosa, J. Non-Cryst. Solids 402 (2014) 141.
- 84 [127] M.J.V. Bell, V. Anjos, L.M. Moreira, R.F. Falcí, L.R.P. Kassab, D.S. da Silva, J. L. Doualan, P. Camy, R. Moncorge, J. Opt. Soc. Am. B 31 (2014) 1590.
- 85 [128] K.U. Kumar, W.F. Silva, K.V. Krishnaiah, C.K. Jayasankar, C. Jacinto, J. Nano-photonics 8 (2014) 083093.
- 86 [129] M.E. Camilo, T.A.A. Assumpção, D.M. da Silva, D.S. da Silva, L.R.P. Kassab, C. B. de Araújo, J. Appl. Phys. 113 (2013) 153507.
- 87 [130] Y. Dwivedi, K. Mishra, S.B. Rai, J. Alloy. Compd. 572 (2013) 90.
- 88 [131] E.O. Serqueira, R.F. de Morais, N.O. Dantas, J. Alloy. Compd. 560 (2013) 200.
- 89 [132] A.C.F.M. Costa, R.H.G.A. Kiminami, P.T.A. Santos, J.F. Silva, J. Mater. Sci. 48 (2013) 172.
- 90 [133] S.A. Camacho, P.H.B. Aoki, C.J.L. Constantino, R.F. Aroca, A.M. Pires, J. Alloy. Compd. 541 (2012) 365.
- 91 [134] A.S. Gouveia-Neto, N.P.S.M. Rios, L.A. Bueno, Opt. Mater. 35 (2012) 126.
- 92 [135] N. Rakov, G.S. Maciel, Proc. Commun. 285 (2012) 5242.
- 93 [136] T.A.A. de Assumpção, D.M. da Silva, M.E. Camilo, L.R.P. Kassab, A.S.L. Gomes, C.B. de Araújo, N.U. Wetter, J. Alloy. Compd. 536 (2012) S504.
- 94 [137] N. Rakov, R.B. Guimarães, D.F. Franceschini, G.S. Maciel, Mater. Chem. Phys. 135 (2012) 317.
- 95 [138] I.T. Weber, I.A.A. Terra, A.J.G. de Melo, M.A.M. Lucena, K.A. Wanderley, C. O. Paiva-Santos, S.G. Antônio, L.A.O. Nunes, F.A.A. Paz, G.F. de Sá, RSC Adv. 2 (2012) 3083.
- 96 [139] P.C. de Sousa Filho, O.A. Serra, J. Phys. Chem. C 115 (2011) 636.
- 97 [140] F.L. Bregolin, U.S. Sias, M. Behar, J. Lumin. 135 (2013) 232.

- [141] J.L. Ferrari, K.O. Lima, L.J.Q. Maia, R.R. Gonçalves, *Thin Solid Films* 519 (2010) 1319.
- [142] A.S. Gouveia-Neto, L.A. Bueno, E.B. da Costa, E.A. Silva Jr., J.L. Ferrari, K.O. Lima, R.R. Gonçalves, *J. Appl. Phys.* 107 (2010) 103539.
- [143] L.A. Bueno, A.S. Gouveia-Neto, A.F. da Silva, D.S. Moura, E.A. da Silva, E.B. Costa, *Proc. SPIE* 7582 (2010) 75821H.
- [144] A.R. Marquesi, J.R.J. Delben, A.A.S.T. Delben, *J. Therm. Anal. Calorim.* 96 (2009) 403.
- [145] L.R.P. Kassab, W.G. Hora, W. Lozano B, M.A.S. de Oliveira, G.S. Maciel, *Opt. Commun.* 269 (2007) 356.
- [146] M.M. Lage, F.M. Matinaga, J.Y. Gesland, R.L. Moreira, *J. Appl. Phys.* 99 (2006) 053510.
- [147] A.S. Gouveia-Neto, A.C.M. Afonso, J.F. Nascimento, E.B. Costa, L.A. Bueno, Y. Messaddeq, S.J.L. Ribeiro, *Proc. SPIE* 6116 (2006) 61160R.
- [148] C. Jacinto, S.L. Oliveira, L.A.O. Nunes, J.D. Myers, M.J. Myers, T. Catunda, *Phys. Rev. B* 73 (2006) 12.
- [149] A.S. Gouveia-Neto, A.F. da Silva, L.A. Bueno, E.B. da Costa, *J. Lumin.* 132 (2012) 299.
- [150] H. Camargo, T.B. Paolini, E. Niyama, H.F. Brito, M. Cremona, *Thin Solid Films* 528 (2013) 36.
- [151] L.R. Tessler, *Nanoscale Lumin. Mater.* 3 (61) (2014) 107.
- [152] N. Rakov, R.B. Guimaraes, W. Lozano B., G.S. Maciel, *J. Appl. Phys.* 114 (2013) 043517.
- [153] G. Gasparotto, M.A. Cebim, M.S. Goes, S.A.M. Lima, M.R. Davolos, J.A. Varela, C.O. Paiva-Santos, M.A. Zaghet, *J. Appl. Phys.* 106 (2009) 063509.
- [154] M.M. Lage, R.L. Moreira, F.M. Matinaga, J.Y. Gesland, *Chem. Mater.* 17 (2005) 4523.
- [155] M.E.G. Valerio, V.G. Ribeiro, A.C.S. de Mello, M.A.C. dos Santos, S.L. Baldochi, V.L. Mazzocchi, C.B.R. Parente, R.A. Jackson, J.B. Amaral, *Opt. Mater.* 30 (2007) 184.
- [156] T.S. Martins, L.C.C. da Silva, J.R. Matos, M.C.A. Fantini, *J. Alloy. Compd.* 560 (2013) 67.
- [157] E.H. de Faria, E.J. Nassar, K.J. Ciuffi, M.A. Vicente, R. Trujillano, V. Rives, P.S. Calefi, *ACS Appl. Mater. Interfaces* 31 (2011) 311.
- [158] D.S. Velasco, A.P. de Moura, A.N. Medina, M.L. Baesso, A.F. Rubira, M. Cremona, A.C. Bento, *J. Phys. Chem. B* 114 (2010) 5657.
- [159] M.A.F. Monteiro, H.F. Brito, M.C.F.C. Felinto, G.E.S. Brito, E.E.S. Teotonio, F.M. Vichi, R. Stefani, *Microporous Mesoporous Mater.* 108 (2008) 237.
- [160] V.S.M. de Barros, W.M. de Azevedo, H.J. Khoury, P.L. Filho, *Radiat. Meas.* 43 (2008) 345.
- [161] J. Kai, D.F. Parra, H.F. Brito, *J. Mater. Chem.* 18 (2008) 4549.
- [162] G.H. Silva, V. Anjos, M.J.V. Bell, A.P. Carmo, A.S. Pinheiro, N.O. Dantas, *J. Lumin.* 154 (2014) 294.
- [163] P.F.S. Pereira, M.G. Matos, C.M.A. Ferreira, E.H. De Faria, P.S. Calefi, L.A. Rocha, K.J. Ciuffi, E.J. Nassar, *J. Lumin.* 146 (2014) 394.
- [164] E.O. Serqueira, N.O. Dantas, *Opt. Lett.* 39 (2014) 131.
- [165] R.F. de Morais, E.O. Serqueira, N.O. Dantas, *Opt. Mater.* 35 (2013) 2122.
- [166] W.S. Souza, R.O. Domingues, L.A. Bueno, E.B. da Costa, A.S. Gouveia-Neto, *J. Lumin.* 144 (2013) 87.
- [167] M.P. Belancon, L.C. Barbosa, *Proc. SPIE* 8601 (2013) 86012F.
- [168] A.S. Gouveia-Neto, W.S. Souza, R.O. Domingues, E.B. da Costa, L.A. Bueno, *Proc. SPIE* 8641 (2013) 86411E.
- [169] R. Schneider, W.H. Schreiner, P.A. Santa-Cruz, *J. Lumin.* 136 (2013) 172.
- [170] S.M.V. Novais, Z.S. Macedo, *Phys. Status Solidi* 10 (2013) 185.
- [171] J.L. Ferrari, M.A. Schiavon, R.R. Gonçalves, A.M. Pires, M.R. Davolos, *Thin Solid Films* 524 (2012) 299.
- [172] N. Rakov, L.R.A. Bispo, G.S. Maciel, *Opt. Commun.* 285 (2012) 1882.
- [173] A.S. Souza, M.A.C. dos Santos, *Chem. Phys. Lett.* 521 (2012) 138.
- [174] A.S. Gouveia-Neto, N.P.S.M. Rios, L.A. Bueno, *Proc. SPIE* 8278 (2012) 82781H.
- [175] J.H.S.K. Monteiro, I.O. Mazali, F.A. Sigoli, *J. Fluoresc.* 21 (2011) 2237.
- [176] R.R. Gonçalves, Y. Messaddeq, M.A. Aegerter, S.J.L. Ribeiro, *J. Nanosci. Nanotechnol.* 11 (2011) 2433.
- [177] L.R.P. Kassab, D.S. da Silva, C.B. de Araújo, *J. Appl. Phys.* 107 (2010) 113506.
- [178] L.J.Q. Maia, V.R. Mastelaro, A.C. Hernandez, J. Fick, A. Ibanez, *Thin Solid Films* 517 (2009) 6584.
- [179] C.N. Santos, D. Mohr, W.F. Silva, A.S.S. de Camargo, H. Eckert, M.S. Li, M.V. D. Vermelho, A.C. Hernandez, A. Ibanez, C. Jacinto, *J. Appl. Phys.* 106 (2009) 023512.
- [180] J. Tronto, S.J.L. Ribeiro, J.B. Valim, R.R. Gonçalves, *Mater. Chem. Phys.* 113 (2009) 71.
- [181] N. Rakov, G.S. Maciel, W. Lozano B, C.B. de Araújo, *J. Appl. Phys.* 101 (2007) 036102.
- [182] V.D. Cacho, L.R.P. Kassab, S.L. Oliveira, R.D. Mansano, P. Verdonck, *Thin Solid Films* 515 (2006) 764.
- [183] E.O. Serqueira, N.O. Dantas, A.F.G. Monte, M.J.V. Bell, *J. Non-Cryst. Solids* 352 (2006) 3628.
- [184] F.H. Cristovan, C.M. Nascimento, M.J.V. Bell, E. Laureto, J.L. Duarte, I.F.L. Dias, W.O. Cruz, A. Marletta, *Chem. Phys.* 326 (2006) 514.
- [185] A.T. de Figueiredo, S. de Lazaro, E. Longo, E.C. Paris, J.A. Varela, M.R. Joya, P.S. Pizani, *Chem. Mater.* 18 (2006) 2904.
- [186] N. Rakov, G.S. Maciel, W. Lozano B, C.B. de Araújo, *Appl. Phys. Lett.* 88 (2006) 081908.
- [187] F.A. Dias Filho, L.D. Carlos, Y. Messaddeq, S.J.L. Ribeiro, *Langmuir* 21 (2005) 1776.
- [188] S.P.A. Osório, E. Fernandez, E. Rodriguez, C.L. Cesar, L.C. Barbosa, *Proc. SPIE* 5733 (2005) 456.
- [189] G. Santos, L.G. Paterno, F.J. Fonseca, A.M. Andrade, L. Pereira, *ECS Trans.* 39 (2011) 307.
- [190] J.R. de Moraes, S.L. Baldochi, L.R.L. Soares, V.L. Mazzocchi, C.B.R. Parente, L. C. Courrol, *MRS Bull.* 47 (2012) 744.
- [191] D.S.L. Figueira, N.C. Frateschi, *J. Appl. Phys.* 103 (2008) 063106.
- [192] S.M.V. Novais, A. Dobrowolska, A.J.J. Bos, P. Dorenbos, Z.S. Macedo, *J. Lumin.* 148 (2014) 353.
- [193] C.M. Abreu, R.S. Silva, M.E.G. Valerio, Z.S. Macedo, *J. Solid State Chem.* 200 (2013) 54.
- [194] C.F. de Almeida, R.C. de Andrade, L.W. Aguiar, F. Caires Jr., E.A. Falcão, C.T. de Carvalho, *J. Therm. Anal. Calorim.* 117 (2014) 251.
- [195] F.F. da Silva, F.L. de Menezes, L.L. da Luz, S. Alves Jr., *New J. Chem.* 38 (2014) 893.
- [196] O.L. Malta, H.F. Brito, J.F.S. Menezes, F.R. Gonçalves e Silva, S. Alves Jr., F.S. Farias Jr., A.V.M. de Andrade, *J. Lumin.* 75 (1997) 255.
- [197] A.V.M. de Andrade, N.B. Da Costa, A.M. Simas, G.F. de Sá, *Chem. Phys. Lett.* 227 (1994) 349.
- [198] J.D.C. Maia, G.A.U. Carvalho, C.P. Manguiera, S.R. Santana, L.A.F. Cabral, G. B. Rocha, *J. Chem. Theory Comput.* 8 (2012) 3072.
- [199] R.O. Freire, S.P. Vila-Nova, E. Brunet, O. Juanes, J.C. Rodríguez-Ubis, S. Alves Jr., *Chem. Phys. Lett.* 443 (2007) 378.
- [200] F.F. da Silva, C.A.F. de Oliveira, E.H.L. Falcão, C.C. Gatto, N.B. da Costa Jr., R.O. Freire, J. Chojnacki, S. Alves Jr., *J. Solid State Chem.* 207 (2013) 132.
- [201] E.M. Rodrigues, E.R. Souza, J.H.S.K. Monteiro, R.D.L. Gaspar, I.O. Mazali, F. A. Sigoli, *J. Mater. Chem.* 22 (2012) 24109.
- [202] A.C.S. Silva, G.G. Souza, M.A.L. Nobre, A.M. Pires, *J. Mater. Sci.* 45 (2010) 4216.
- [203] T.A.C. Colman, D.J.C. Gomes, F.J. Caires, O.T. Filho, R.C. da Silva, M. Ionashiro, *Thermochim. Acta* 591 (2014) 111.
- [204] J.H.S.K. Monteiro, R.D. Adati, M.R. Davolos, J.R.M. Vicenti, R.A. Burrow, *New J. Chem.* 35 (2011) 1234.
- [205] A.S.S. de Camargo, R.A. Silva, J.P. Andreetta, L.A.O. Nunes, *Appl. Phys. B* 80 (2005) 497.
- [206] R.O. Freire, G.B. Rocha, R.Q. Albuquerque, A.M. Simas, *J. Lumin.* 111 (2005) 81.
- [207] J.B. Amaral, J.R. de Moraes, S.L. Baldochi, R.A. Jackson, M.E.G. Valerio, *Phys. Status Solidi C* 10 (2013) 165.
- [208] M.O. Rodrigues, A.M. Brito-Silva, S. Alves Jr., C.A. de Simone, A.A.S. Araújo, P.H.V. de Carvalho, S.C.G. Santos, K.A.S. Aragão, R.O. Freire, M.E. Mesquita, *Quim. Nova* 32 (2009) 286.
- [209] M.A.M. Filho, J.D.L. Dutra, H.L.B. Cavalcanti, G.B. Rocha, A.M. Simas, R.O. Freire, *J. Chem. Theory Comput.* 10 (2014) 3031.
- [210] A.R. Zanatta, *J. Phys. D – Appl. Phys.* 42 (2009) 025109.
- [211] R.F. Martins, R.F. Silva, R.R. Gonçalves, O.A. Serra, *J. Fluoresc.* 20 (2010) 739.
- [212] E.A. de Morais, L.V.A. Scalvi, A.A. Cavalheiro, A. Tabata, J.B.B. Oliveira, *J. Non-Cryst. Solids* 354 (2008) 4840.
- [213] S.A. Camacho, P.H.B. Aoki, C.J.L. Constantino, A.M. Pires, *J. Lumin.* 153 (2014) 272.
- [214] T.F. Pineiz, E.A. de Morais, L.V.A. Scalvi, C.F. Bueno, *Appl. Surf. Sci.* 267 (2013) 200.
- [215] W.S. Lopes, C.R.S. Morais, A.G. de Souza, *Mater. Sci. Forum* 530 (2006) 513.
- [216] D.A. Gálico, M.G. Lahoud, M.R. Davolos, R.C.G. Frem, T.F.C. Fraga-Silva, J. Venturini, M.S.P. Arruda, G. Bannach, *J. Inorg. Biochem.* 140 (2014) 160.
- [217] I.L.V. Rosa, A.V.S. de Lourenço, C.R. Neri, O.A. Serra, *J. Fluoresc.* 16 (2006) 455.
- [218] D.F. Parra, A. Mucciolo, D.G. Duarte, H.F. Brito, A.B. Lugão, *J. Appl. Polym. Sci.* 100 (2006) 406.
- [219] M.F. Belian, R.O. Freire, A. Galembeck, G.F. de Sá, R.F. de Farias, S. Alves Jr., *J. Lumin.* 130 (2010) 1946.
- [220] A.P. Souza, S. Alves Jr., O.L. Malta, *Opt. Mater.* 33 (2011) 402.
- [221] B.S. Barros, C.A.F. de Oliveira, J. Kulesza, S. Alves Jr., M. Bochenska, *Adv. Sci. Technol.* 77 (2013) 132.
- [222] E.R. Botero, J.A. Eiras, R. Guo, A. Bhalla, D. Garcia, *Integr. Ferroelectr.* 131 (2011) 134.
- [223] R.S. Oliveira, D.O. Maia, M.R. Pereira, F.R.G. Silva, *J. Macromol. Sci. A* 47 (2010) 392.
- [224] G.C. Santana, A.C.S. de Mello, M.E.G. Valerio, Z.S. Macedo, *J. Mater. Sci.* 42 (2007) 2231.
- [225] M.F. Belian, G.F. de Sá, S. Alves Jr., R.F. de Farias, *J. Coord. Chem.* 60 (2007) 173.
- [226] E.J.A. Oliveira, S.P.V. Nova, S. Alves-Jr, P. Santa-Cruz, R.J.R. Molica, A. Teixeira, E. Malageno, J.L.L. Filho, *J. Braz. Chem. Soc.* 17 (2006) 243.
- [227] P.P. Lima, O.L. Malta, S. Alves, *Quim. Nova* 28 (2005) 805.
- [228] M.C. Silva, F.H. Cristovan, C.M. Nascimento, M.J.V. Bell, W.O. Cruz, A. Marletta, *J. Non-Cryst. Solids* 352 (2006) 5296.
- [229] R.D.L. Gaspar, I.O. Mazali, F.A. Sigoli, *Colloids Surf. A* 367 (2010) 155.
- [230] P.F.S. Pereira, M.G. Matos, L.R. Ávila, E.C.O. Nassor, A. Cestari, K.J. Ciuffi, P.S. Calefi, E.J. Nassar, *J. Lumin.* 130 (2010) 488.
- [231] I.J.V. Rosa, P.C. de Sousa Filho, C.R. Neri, O.A. Serra, A.T. de Figueiredo, J.A. Varela, E. Longo, *J. Fluoresc.* 21 (2011) 1575.
- [232] A.P. Ramos, R.R. Gonçalves, O.A. Serra, M.E.D. Zaniquelli, K. Wong, *J. Lumin.* 127 (2007) 461.
- [233] T.S. Martins, J.R. Matos, G. Vicentini, P.C. Isolani, *J. Therm. Anal. Calorim.* 86 (2006) 351.
- [234] H.C. Silva, C.R.S. Morais, S.A. Morais, B.F. Lira, *Mater. Sci. Forum* 727 (2012) 1913.
- [235] G. Gasparotto, S.A.M. Lima, M.R. Davolos, J.A. Varela, E. Longo, M.A. Zaghet, *J. Lumin.* 128 (2008) 1606.
- [236] P.J.R. Montes, M.E.G. Valerio, G.M. Azevedo, *Nucl. Instrum. Methods B* 266 (2008) 2923.

1	[237] A.O. Ribeiro, O.A. Serra, J. Braz. Chem. Soc. 18 (2007) 273.	[243] J.L. Ferrari, M.A. Cebim, A.M. Pires, M.A.C. dos Santos, M.R. Davolos, J. Solid	67
2	[238] I.L.V. Rosa, L.H. Oliveira, C.K. Suzuki, J.A. Varela, E.R. Leite, E. Longo, J. Fluoresc.	State Chem. 183 (2010) 2110.	68
3	18 (2008) 541.	[244] A.V.S. Lourenço, C.A. Kodaira, E.M. Ramos-Sanchez, M.C.F.C. Felinto, H. Goto,	69
4	[239] J.C. Batista, P.C. de Sousa Filho, O.A. Serra, Dalton Trans. 41 (2012) 6310.	M. Gidlund, O.L. Malta, H.F. Brito, J. Inorg. Biochem. 123 (2013) 11.	70
5	[240] P.J.R. Montes, M.E.G. Valerio, M.V.S. Rezende, J. Electron Spectrosc. Relat.	[245] C.A. Kodaira, A.V.S. Lourenço, M.C.F.C. Felinto, E.M.R. Sanchez, F.J.O. Rios, L.A.	71
6	Phenom. 189 (2013) 39.	O. Nunes, M. Gidlund, O.L. Malta, H.F. Brito, J. Lumin. 131 (2011) 727.	72
7	[241] M.V.S. Rezende, P.J.R. Montes, F.M.S. Soares, C. dos Santos, M.E.G. Valerio,	[246] L.C.V. Rodrigues, J. Hölsä, J.M. Carvalho, C.C.S. Pedroso, M. Lastusaari, M.C.F.	73
8	J. Synchrotron Radiat. 21 (2014) 143.	C. Felinto, S. Watanabe, H.F. Brito, Physica B 439 (2014) 67.	74
9	[242] V.A.G. Rivera, F.A. Ferri, J.L. Clabel H., M.A. Pereira-da-Silva, L.A.O. Nunes,		75
10	M.S. Li, E. Marega Jr., J. Lumin. 148 (2014) 186.		76
11			77
12			78
13			79
14			80
15			81
16			82
17			83
18			84
19			85
20			86
21			87
22			88
23			89
24			90
25			91
26			92
27			93
28			94
29			95
30			96
31			97
32			98
33			99
34			100
35			101
36			102
37			103
38			104
39			105
40			106
41			107
42			108
43			109
44			110
45			111
46			112
47			113
48			114
49			115
50			116
51			117
52			118
53			119
54			120
55			121
56			122
57			123
58			124
59			125
60			126
61			127
62			128
63			129
64			130
65			131
66			132