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Abstract: In the context of environmental remediation of contaminated sites and safety assessment of nuclear waste disposal in the near-surface zone, we investigate the leaching and non-Fickian dispersive migration with sorption of strontium (mocking strontium-90) through columns packed with sand and clay. Analysis is based on breakthrough curves (BTCs) from column experiments, which simulated rainfall infiltration and source term release scenario, rather than applying constant tracer solution at the inlet as commonly used. BTCs are re-evaluated and transport parameters are estimated by inverse modelling using two approaches: (1) equilibrium advection-dispersion equation (ADE); and (2) continuous time random walk (CTRW). Firstly, based on a method for calculating leach concentration, the inlet condition with an exponential decay input is identified. Secondly, the results show that approximately 39%~58% of bromine and $16\%\sim49\%$ of strontium are eluted from the columns at the end of the breakthrough experiments. This suggests that trapping mechanisms, including diffusion into immobile zones and attachment of tracer on mineral surfaces, are more pronounced for strontium than for bromine. Thirdly, we demonstrate robustness of CTRW-based truncated power-law (TPL) model in capturing non-Fickian reactive transport with $0<\beta<2$, and Fickian transport with β >2. The non-Fickian dispersion observed experimentally is explained by variations of local flow field from preferential flow paths due to physical heterogeneities. Particularly, the additional sorption process of strontium on clay minerals contributes to the delay of the peak concentration and the tailing features, which leads to enhanced non-Fickian transport for strontium. Finally, the ADE and CTRW approaches to environmental modelling are evaluated. It is shown that CTRW with a sorption term can describe non-Fickian dispersive transport of strontium at laboratory scale by identifying appropriate parameters, while the traditional ADE with a retardation factor fails to reproduce the complex non-Fickian transport of strontium with remarkable sorption on clay surface.

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Please, consider the attached manuscript "Non-Fickian dispersive transport of strontium in laboratory-scale columns: Modelling and evaluation" for publication in the Journal of Hydrology.

It contains original research that has not been publishedelsewhere.We propose the integration of non-Fickian transport modelling andlaboratory dynamic column experiments mocking rainfall infiltration scenarios, and analyzenon-Fickiandispersive transportfor strontiumdue to variation of flow fields and sorption retardation processes, which is significativetoadvance the understanding and prediction of radionuclide transport through hydrologic paths and heterogeneous systems.

We believe that the methodology and the findings of this work will be of interest to the readership of your Journal.

Respectfully yours,

DongxuLiu Andrey PJivkov Lichun Wang Gaohua Si Jing Yu

- Integrated methodology for coupling release and transport of strontium
- Experimental data with Fickian and non-Fickian transport re-evaluated
- Advection-DispersionEquation and Continuous Time Random Walk examined
- ADE suitable for Fickian transport only; CTRWcapturesboth regimes
- Effects of chemical and physical heterogeneities on transport analysed

1	Non-Fickian dispersive transport of strontium in laboratory-scale columns:
2	Modelling and evaluation
3	
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36

37 1. Introduction

Radioactive strontium (Sr-90), a by-product of the fission of uranium and plutonium, is found in 38 waste from nuclear activities. Since atmospheric nuclear tests conducted in the 1950s and 1960s and 39 the accident occurred at the Chernobyl nuclear power plant, it is well known that, like many other 40 radioactive species, large amounts of strontium-90 were produced and dispersed worldwide in fallout 41 42 into the environment (EPA, 2015). Strontium-90 is considered as one of the hazardous constituents of nuclear wastes, due to its environmental mobility, radiation exposure and biologic toxicity. 43 Specifically, strontium-90 has chemical properties similar to calcium and therefore it tends to 44 45 concentrate in the bones and teeth, and can form many chemical compounds, including halides, oxides, and sulphide. This increases the risk of bone cancer, cancer of the soft tissue near the bone, 46 and leukaemia from internal exposure (EPA, 2015). Environmental behaviour of radioactive 47

strontium has therefore attracted increasing attention over the last three decades, focused on 48 understanding migration of Sr-90 and strontium isotopes through field in situ tests, laboratory 49 experiment analysis, natural analogue studies and computer modelling. However, accurate 50 51 assessment and prediction of Sr-90 transport in complex environments remains unresolved, with limited number of investigations devoted to leaching, non-Fickian migration and sorption processes 52 (Bencala, 1984; Bruno et al., 2002; Bugai et al., 2012; Eagling et al., 2013; Hull and Schafer, 2008; 53 Kosakowski and Smith, 2005; Maderich et al., 2014; Miller et al., 1994; Qian et al., 2009; Rumynin, 54 2011; Volkova, 2009; Yin et al., 2014). Combining laboratory dynamic column experiments and 55 56 theoretical modelling would contribute significantly to fundamental understanding and physically realistic description of strontium transport in porous media. 57

Understanding species (contaminants, tracers, solutes, nuclides or particles) fate and transport in 58 59 the subsurface environment is of central importance in the frame of remediation schemes for 60 contaminated sites and nuclear waste disposal (Bijeljic et al., 2004; Levchuk et al., 2012). The problem is challenging due to perennial complexities and strong heterogeneities in geological mass. 61 62 From systematic viewpoint, a persistent issue is how to explore appropriate predictive models in the absence of sufficient information on heterogeneous hydraulic-physical-chemical properties. 63 Traditionally, species migration in geological systems from the pore scale upwards is described by 64 Fickian-based advection-dispersion equation (ADE) models, adopting deterministic and/or stochastic 65 66 approaches (Bear and Cheng, 2010; Berkowitz et al., 2001; Rumynin, 2011). However, the ADE-67 based approaches often fail to capture and predict non-Fickian (or anomalous) transport of species in the subsurface. This is commonly observed both in the laboratory and in the field (Berkowitz et al., 68 2001; Bijeljic et al., 2013; Neuman and Tartakovsky, 2009), and now recognized as ubiquitous in the 69 70 transport in fractured porous media.

Non-Fickian transport behaviour presents different forms and traits, commonly referred to as
 scale-dependent dispersion, but manifests through "unusual" early breakthrough times and long late

73 time tails in measured breakthrough curves (BTCs), which deviate from the Gaussian distributions of species concentration (Bakshevskaia and Pozdniakov, 2015; Berkowitz et al., 2006; Edery et al., 74 2014). The continuous time random walk (CTRW) has been shown to be a successful framework to 75 quantify non-Fickian transport with capacity to capture the transport transition from non-Fickian to 76 Fickian (Berkowitz et al., 2006; Hansen and Berkowitz, 2015). The CTRW approach describes the 77 random movement of solute particles in an Eulerian-Lagrangian framework (Neuman and 78 Tartakovsky, 2009). The use of CTRW for species transport in porous and fractured media across 79 different scales have been presented elsewhere (e.g. Berkowitz et al., 2006; Berkowitz and Scher, 80 81 2009; Blunt et al., 2013; Cortis and Berkowitz, 2004; Edery et al., 2013; Rhodes et al., 2008; Wang and Cardenas, 2014). 82

In the CTRW framework, the conservative tracer (e.g. bromide) parameters, including mean fluid 83 84 velocity and dispersion coefficient, are inappropriate to describe transport of reactive or sorbing tracers (e.g. atrazine, TBNPA) through laboratory columns (Li and Ren, 2009; Rubin et al., 2012). 85 Although models for describing non-Fickian transport, including mobile-immobile, fractional 86 derivative, and multi-rate mass transport, have been applied to simulate certain pollutants transport 87 process (e.g. pesticides), few studies focused on the application of those models to the fate of 88 reactive nuclides (e.g. strontium) (Dentz et al., 2011; Kapetas et al., 2014; Neuman and Tartakovsky, 89 2009). With respect to reactive transport it should be mentioned, that separation of migration and 90 91 sorption into independent sets of measured/fitted parameters is not justifiable, although such 92 separation has been used in the traditional ADE formulation (Rubin et al., 2012). This makes the prediction of reactive nuclide transport with pervasive non-Fickian phenomenon even more 93 challenging. Thus, for a special reactive species, strontium in this study, there is a continuing need to 94 95 advance the understanding and prediction of transport phenomena involving natural tracer release, non-Fickian migration and sorption. 96

97 The objectives of this paper are: (1) to re-analyse BTCs from column experiments simulating rainfall infiltration scenarios and transport behaviour of strontium (Gaohua et al., 2013); (2) to apply 98 the ADE and CTRW models to characterize coupled effects of release, transport and sorption 99 100 processes; (3) to identify transport parameters by inverse modelling against experimental data; (4) to evaluate the applicability of ADE and CTRW models in capturing reactive species transport with 101 pervasive non-Fickian transport. This study advances the understanding and prediction of 102 radionuclide transport through heterogeneous systems, which is critical for environmental 103 remediation of contaminated sites and the safety assessment of nuclear waste subsurface disposal. 104

105

106 2. Methodology

107 2.1. Column experiments

108 Safety assessment of near-surface zone and planning remedial options for contaminated sites require detailed studies of release and transport processes. Of particular interest for near-surface 109 nuclear waste storage is the radioactive fallout dispersed to surface in a rainfall scenario. The column 110 experiments described here served this purpose. These were mocking rainfall infiltration with a 111 contaminant (source) located at the inlet and slowly released to the environment, rather than the 112 traditional supply of a tracer concentration through columns over time. Firstly, examining the tracer 113 concentrations leaching from the source layer is essential for identifying and then modelling the inlet 114 condition. Secondly, measuring the tracer mass eluted from the columns is necessary for the 115 116 quantitative analysis of mass transport and models validation.

117 Column experiments, with setup shown in Fig. 1, were performed to investigate strontium (Sr) 118 transport with varying sprinkling volumetric flow rates, and different packed materials, including 119 sand and clay. Bromine (Br) was used as a conservative tracer for water flow. Full details of the 120 experimental setup and primary results are described by Gaohua et al. (2013); only information 121 relevant to this work is given below. 122 The tracer experiments were conducted in a set of cylindrical organic glass columns (3cm diameter and 20cm length) packed with sand and/or clay. Eight different columns were prepared: 123 124 columns a, b, c, d were 18-cm long packed by medium sand of $0.25 \sim 0.50$ mm diameter; columns e, f, g, h were packed by a mixture of clay and the same sand with 25%, 50%, 75% and 100% clay 125 content, respectively. The packing materials were taken from the sediments in northwest China. Neat 126 tracer sources were prepared by the following the steps: (1) Sr stock solution was made by dissolving 127 128 100g SrCl2•6H2O into 250mL distilled water; Br stock solution was made by dissolving 66g NaBr in 1000mL distilled water; and (2) 250g pure quartz sand was placed in 1000mL beaker, to which 129 105mL Sr solution and 50mL Br solution were added, respectively; Beaker was stirred well and 130 dried until moisture content reduced to 3%. 131

Columns a-d were sprinkled with distilled water under different volumetric flow rates: 100mL/d, 50mL/d, 20mL/d and 2mL/d, respectively. We note that although a-d were packed with sand with the same particle size distribution, the packing structure might have varied due to preparation uncertainty. Columns e-h were sprinkled with the same flow rate 2mL/d of distilled water. To prevent water and soil loss, a polyester fibre membrane with an aperture 0.45µm was placed at the bottom of column. Sprinkling volumetric flow rate was controlled by injection pump and peristaltic pump as a water supply device.

During experiments, columns were saturated slowly by leaching with distilled water until steadystate fluid velocity was achieved. The first inlet section was occupied by a 0.8-cm thick tracer source layer (Br 135.5mg and Sr 718mg, approximately), which was overlain by a 0.5cm thick sand. At the column outlet, effluent water was collected automatically over experimental duration. The concentration of Br and Sr was determined by ion chromatography and inductively coupled plasmaatomic emission spectrometry (ICP-AES), respectively. Thus, the breakthrough curves (temporal profile of concentration) of Br and Sr were obtained.

147 2.2. Transport model with Advection-Dispersion Equation (ADE)

163

Assuming steady-state flow in homogeneous porous media, the classical ADE for one-dimensional transport of species, subject to equilibrium isotherm adsorption and radioactive decay, can be written as (Šimůnek et al., 1999; Toride et al., 1995)

151
$$R\frac{\partial C}{\partial t} = D\frac{\partial^2 C}{\partial t^2} - v\frac{\partial C}{\partial x} - \lambda C, \qquad (1)$$

where C is species concentration in the liquid (ML⁻³), R is dimensionless retardation factor, D is 152 dispersion coefficient (L^2T^{-1}), v is average pore water velocity (LT^{-1}), λ is the first-order decay 153 coefficient (T^{-1}) , t is time (T), and x is transport distance (L). R depends on the relative transport 154 velocity, v_s , such that $R = v / v_s$, and reflects the strength of sorption, R = 1 when no sorption occurs. 155 D can be represented by $D = v\alpha + \tau D_w$ (Bear, 1972), where D_w is molecular diffusion coefficient in 156 free water (L²T⁻¹), α is dispersivity (L), τ is a dimensionless tortuosity factor. For stable species $\lambda = 0$. 157 For an initially species-free and semi-finite column, where tracer is placed in a fixed layer at the 158 inlet of column and is eluted with free water, the leach concentration of effluent will decrease over 159 time as the tracer mass in the source layer reduces due to mass loss through leaching. The leach 160 concentration can be estimated via an exponential function using the leaching rate coefficient (Chen 161 et al., 2007) 162

$$C(t,0) = \frac{\lambda_L m_0}{Q} e^{-(\lambda_L + \lambda)t}$$
(2)

where C(t,0) is the leach concentration (ML⁻³), m_0 is the total mass of tracer (M), Q is the sprinkling volumetric flow rate (L³T⁻¹), λ_L is a leaching rate coefficient (T⁻¹). A first-order decay rate of the source term, similar to radioactive decay, was given by (Baes and Sharp, 1983; Chen et al., 2007)

167
$$\lambda_L = \frac{Q}{\varepsilon R V_t} = \frac{I}{\varepsilon R d}$$
(3)

168 where ε , V_t , d are effective porosity (L³L⁻³), volume (L³), and depth (L) of the tracer source layer, 169 respectively, I is infiltration rate of water (LT⁻¹), R is the retardation factor of tracer in the source 170 layer (-).

The model of leach concentration represented by Eqs.(2)~(3) is a relatively simple approximation 171 of the complex transport mechanism involving physical and chemical processes. This model assumes 172 that equilibrium adsorption is achieved instantaneously between the passing water and the source 173 174 layer when the percolating water comes into contact with the tracer source layer. The equilibrium is generally described by a linear adsorption relationship using a distribution coefficient (K_d) estimated 175 176 from batch experiments (Baes and Sharp, 1983; Chen et al., 2007). R can be thus calculated based on K_d , such that $R = l + \rho_b \cdot K_d / n_e$, where ρ_b is bulk density, n_e is effective porosity. Moreover, combining 177 with the inverse modelling, R can be obtained directly from column experiments (Rod et al., 2010). 178

The simulated BTCs can be solved with given boundary conditions. Here, Dirichlet boundary with prescribed concentration, which is determined by the leach concentration curve as described by Eq.(2), is set for the inlet boundary of the column. Further, Neuman boundary condition with prescribed flux is set for the outlet boundary.

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184 2.3. Transport model with Continuous Time Random Walk (CTRW)

The concept and the formulation of the CTRW method can be found in various references (e.g. Berkowitz et al., 2006; Cortis et al., 2004; Dentz et al., 2004; Edery et al., 2015; Edery et al., 2013; Neuman and Tartakovsky, 2009; Scher and Lax, 1973). Here, we sketch the key features of CTRW for the purpose of this study.

In the CTRW framework, transport processes are conceptualized as a series of particle transitions by statistically characterizing tracer motion. The main feature of CTRW lies in quantifying temporal transition probability of particle transport induced by spatial heterogeneities via a memory function (Berkowitz et al., 2006). The one-dimensional (1D) CTRW for macroscopic transport through a uniform domain is given in
Laplace space by Eqs. (4)-(7) (Cortis and Berkowitz, 2005; Edery et al., 2014)

195
$$u\tilde{c}(x,u) - c_0(x) = -\tilde{M}(u) \left[v_{\psi} \frac{\partial}{\partial x} \tilde{c}(x,u) - D_{\psi} \frac{\partial}{\partial x^2} \tilde{c}(x,u) \right]$$
(4)

196
$$\tilde{M}(u) \equiv \bar{t}u \frac{\tilde{\psi}(u)}{1 - \tilde{\psi}(u)}$$
(5)

197
$$v_{\psi} = \frac{1}{n\overline{t}} \sum_{s} p(s)s \tag{6}$$

198
$$D_{\psi} = \frac{1}{n\overline{t}} \sum_{s} \frac{1}{2} p(s) ss$$
(7)

where u is the Laplace variable and the tilde represents the Laplace transformed variable; c_0 is the 199 initial concentration and $\tilde{c}(x,u)$ is the Laplace-transformed normalized concentration; $\tilde{M}(u)$ is a 200 201 memory function; $\psi(t)$ is a probability density function (PDF), defined as the probability rate for a transition time t between transport sites and $\tilde{\psi}(u)$ is the Laplace-transformed form of $\psi(t)$; v_{ψ} and D_{ψ} 202 are the tracer transport velocity and dispersion coefficient, respectively; \overline{t} is a characteristic time; n is 203 204 porosity, and p(s) is the probability distribution of the length of transitions. Note that both v_{ψ} and D_{ψ} depend on $\psi(t)$, which is fundamentally different from the average pore water velocity (v) and 205 206 dispersion coefficient (D) in the classical ADE (Berkowitz et al., 2006).

The PDF, $\psi(t)$, is the core of the CTRW formulation as it depicts the nature of solute plume migration patterns (Berkowitz et al., 2001). Depending on the choice of $\psi(t)$, the CTRW can capture a broad range of transport regimes, including the transition from non-Fickian to Fickian transport. There are several compelling forms of $\psi(t)$ that have been proposed in the literature (Berkowitz et al., 2006; Cortis and Berkowitz, 2005; Cortis et al., 2004; Margolin et al., 2003). Here, we present one specific form for conservative tracer transport, which is widely used to interpret transport phenomena at different scales: the truncated power law (TPL) model

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$$\tilde{\psi}(u) = (1 + \tau_2 u t_1)^{\beta} \exp(t_1 u) \Gamma(-\beta, \tau_2^{-1} + t_1 u) / \Gamma(-\beta, \tau_2^{-1}), 0 < \beta < 2$$
(8)

where β is a critical parameter characterizing the regimes of the dispersive transport, t_1 is a characteristic transition time for the onset of the power law region, t_2 is a "cut-off" time, $\tau_2 = t_2 / t_1$. The memory function is determined by setting $\bar{t} = t_1$.

The TPL model allows for a systematic investigation of non-Fickian transport observed for 218 transition times $t_1 < t < t_2$, and for analysing the transition from non-Fickian to Fickian transport for 219 $t > t_2$ (Dentz et al., 2004). The t_1 sets the lower limit from which the power law behaviour begins, so 220 that the time range of interest is $t > t_1$. The time t_2 governs the crossover from power law to a 221 decreasing exponential function (Edery et al., 2014; Gao et al., 2009). Overall, the $\psi(t)$ exhibits a 222 223 wide change in behaviour as a function of β that largely determines $\psi(t)$ (Edery et al., 2014). That is, β controls the particle transport and thus functionally captures the dispersion regimes (i.e. Fickian or 224 non-Fickian) (Cortis et al., 2004). The relative shapes of the anomalous transport regimes are 225 226 strongly dependent on β , as shown in Table 1.

In this study, the TPL model is used to interpret the measured BTCs of strontium, since the TPL model has been proved to be effective in describing non-Fickian transport for numerous laboratory and field observations (Edery et al., 2014). Specifically, the Laplace transformed analytical solution to the CTRW equations is obtained for a set of given boundary conditions: Dirichlet boundary condition (prescribed concentration) with an exponential decay function of concentration is specified at the inlet of the columns according to the Eq.(2); Neumann (prescribed flux) boundary condition is specified at the outlet boundary.

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235 2.4. Inverse modelling with ADE and CTRW-TPL methods

The experimental BTCs are analyzed using the equilibrium ADE and CTRW-TPL methods described above. Inverse modelling for parameters identification is implemented by fitting analytical solutions with the models to the observed BTCs. The inversion processes are conducted through the non-linear least squares inversion program, CXTFIT (Šimůnek et al., 1999; Toride et al., 1995), and 240 the CTRW MATLAB toolbox which is available from the website
241 http://www.weizmann.ac.il/ESER/People/Brian/CTRW.

From the measured effluent concentration C_f at the outlet boundary, and the concentration at the onset of leaching process C_0 (both in ML⁻³), a normalized concentration is defined by

$$c = \frac{C_f}{C_0},\tag{9}$$

Further, the ratio of eluted tracer mass to the total tracer mass m_0 , i.e. the cumulative mass fraction eluted from columns is calculated by

247
$$f = \frac{Q \int_{0}^{t} C_{f} dt}{m_{0}}.$$
 (10)

248 The two parameters, *c* and *f*, are used to constrain the inversion process.

The inverse modelling with the equilibrium ADE for bromide (Br) and strontium (Sr) is performed with CXTFIT. The inverse problem is solved by minimizing an objective function of root mean square error between observed and fitted concentrations (Šimůnek et al., 1999). For bromide, the retardation coefficient R is set to 1, because it is assumed that bromide ions have minimal sorption onto sand and clay (Srinivasan and Sarmah, 2014). The parameters identified for Br, v and D, are then used as input values for the measured strontium BTCs, so that R for strontium can be estimated.

Further, the inversion of 1D TPL model for strontium is implemented by comparing the predicted 255 BTCs from the CTRW to the normalized BTCs from tracer experiment. The inverse modelling 256 requires initial guesses of seven fitted parameters, including v_{ψ} , D_{ψ} , β , t_1 , t_2 , Λ and α . Depending on 257 how close the initial values of the parameters are to the optimized ones, the inverse estimation of 258 parameters can be time-consuming. Here the values of v and D from CXTFIT are used as initial 259 260 guesses of v_{ψ} and D_{ψ} . Considering that the TPL model is sensitive to the exponent β while relatively insensitive to t_1 and t_2 (Cortis and Berkowitz, 2005), a broad range for β is tested, from 0.8 to 1.8, 261 while t_1 and t_2 are estimated from global error minimisation. Further, Λ and α , which influence the 262

degree of retardation and the power law slope of the BTC tailing (Cortis et al., 2006; Margolin et al., 2003), are initially set to 0.01~0.3 and 0.01~0.2, respectively. The approach to estimate initial values and to perform adjustments using trial and error method corresponds to the successful sensitivity analysis by Cortis et al.(2006). This allowed for achieving good agreement between experimental and modelling results with the CTRW toolbox.

268

269 **3. Results and Discussion**

270 3.1. Tracer leach concentration and eluted mass fraction

271 Leach concentration of bromine and strontium in the experiments are estimated by Eqs. (2) and (3), and presented in Fig. 2. Assuming that there is no strontium sorption to quartz in the source layer 272 (Rod et al., 2010), strontium concentration demonstrates an exponential decay over the experimental 273 274 duration, similar to bromine. Because the leaching rate, λ_L , is proportional to the volumetric flow rate, Q, the leach concentration is inversely exponentially dependent on Q, i.e. very strongly. The higher 275 the flow rate is, the more easily the tracer particles transfer to aqueous phase and are subjected to 276 stronger hydrodynamic dispersion with time. This eventually leads to faster falloff of leach 277 concentration with shorter residence time, and vice versa (Fig. 2). Moreover, as more mass of 278 279 strontium than bromine is located at the source layer, the strontium concentration is greater than that of bromine at the onset of leaching. The release of source term is characterized as a continuous and 280 281 decreasing process rather than an instantaneous process. This is due to mass loss through leaching 282 over time. As a result, the inlet condition of the leaching release process should be treated as an exponential decay function, instead of a simple instantaneous pulse or Dirac delta function, which is 283 commonly employed for experimental and numerical simplicity. This treatment of tracer release is 284 285 closer to the natural condition in simulating rainfall infiltration, and is critical for dynamic simulation of tracer transport, e.g. in the CTRW modelling. 286

287 The cumulative eluted mass fraction for bromine is approximately 39%~58% for the observed columns, as given in Fig. 3. The results show that around half of the mass remains in the source 288 layers and soil columns. This indicates that a fraction of pore water is immobile due to the presence 289 of "dead pore" in relation to fluid flow. For strontium, the cumulative eluted mass fraction is 290 approximately 16%~49% (Fig. 3), which is much less than that of bromine. This demonstrates that, 291 in addition to the effect of immobile water on mass transport (i.e. diffusion into immobile zones), the 292 sorption of strontium on clay minerals is significant and results in extra mass loss as it is transported 293 through packed columns. Expectedly, more clay content results in larger sorption of strontium and 294 295 less aqueous strontium, as seen for columns e-h. The effect of volumetric flow rate on transport is excluded in the study since the packing structures for columns a-d may vary from one to another due 296 297 to uncertainty in the experimental design. It is worth noting, however, that larger eluted mass fraction 298 is observed for bromide and strontium in column d, which can be attributed to relatively larger heterogeneity in this packing resulting in local preferential flows and non-Fickian transport, as 299 discussed in the next sub-section. 300

301

302 3.2. Observations of non-Fickian transport

The typical non-Fickian features, asymmetric BTCs with early breakthrough and late tailings, are 303 observed in varying degrees in our study. Particularly pronounced is the experiment with column d. 304 This is illustrated by the concentration results shown in Fig. 4, which are also used to show the 305 306 predictions of the two methods after parameter estimation, i.e. best fits. Although column d is packed with "homogeneous" sand and sprinkled with a relatively slow flow rate, 2mL/d, the "anomalous" 307 transport phenomenon is remarkable. This might be due to the occurrence of preferential flow path 308 309 caused by the heterogeneous packing structure. Nevertheless, non-Fickian transport of strontium is observed in all BTCs, which implies that physical-chemical heterogeneities contribute strongly via 310 311 preferential flow, diffusion in/out of immobile zones and complex sorption.

The estimated parameters of the two tested models are given in Table 2. Estimated $0 < \beta < 1$ for 312 columns d, g, and h, indicates highly dispersive transport leading to longer tailing. The early 313 breakthrough time observed for column d (Fig. 4d) is not expected in view of the lower flow rate 314 compared to columns a-c. This suggests the existence of stronger preferential flow path as well as 315 larger immobile zones. Potentially, a fraction of strontium is trapped or diffuses into low 316 permeability lenses and remains there for long times. The strontium particles that firstly reach the 317 318 outlet with short residence time are those that propagate through the faster moving regions along preferential flow paths. Thus, the "homogeneous" column d with a much smaller β , larger transport 319 320 velocity and dispersion coefficient, fails to comply with the Fick transport theory and exhibits significant non-Fickian dispersion. 321

For columns g and h (Fig. 4g and h), in addition to diffusion into immobile zones, sorption is a reasonable explanation for the non-Fickian migration based on the conclusion of Rubin et al. (2012). The reduction and delay of the peak concentration and the tailing features are affected by the additional processes of adsorption/desorption of strontium onto/from the mineral surface. This partitioning mechanism between aqueous and mineral phases enhances the non-Fickian transport and causes the failure of Fickian-based equilibrium ADE model.

Comparatively, columns a, e and f with $1 < \beta < 2$, present moderate non-Fickian processes at 328 different water flow fields and particles residence time. Note that in Fig. 4a, the BTCs demonstrate 329 330 early arrival and late tail, corresponding to the larger transport velocity and dispersion coefficient, 331 which are similar to the phenomenon of column d and distinct from columns e and f. This can be explained by the higher pore water velocities enhancing the preferential flow in the presence of 332 immobile water (Gaber et al., 1995; Li and Ren, 2009). For columns e and f, sorption retardation 333 334 process (e.g. strontium cations exchange with inorganic cations in clay), serves as an additional factor contributing weakly to the non-Fickian dispersive transport. The contribution of sorption to the 335 336 non-Fickian features depends on the amount of clay as seen in Fig. 4e-h.

Unlike column a with high volumetric flow rate, Fickian transport at medium volumetric flow rate is predicted fairly well for column b and c with values of β slightly larger than 2 (Fig. 4b and c). This suggests that the assumption of equilibrium mass transfer between spectrums of different mobility is adequate for moderate flow rates within relatively homogeneous structures. The results also supports our assumption that little sorption/desorption occurs for sand-packed columns.

Our results strengthen the understanding that both physical and chemical heterogeneities may 342 influence strongly species transport (Mohamed et al., 2010). For laboratory scale disturbed and 343 undisturbed soil columns, the role of physical non-equilibrium transport stemming from physical 344 345 heterogeneities, i.e. different pore scale structures, is critical for inducing non-Fickian phenomena: macro-pores commonly control the preferential flow process, while some pore spaces serve as tracer 346 347 sinks at early times, and then release tracer slowly. This drives non-Fickian transport with late tailing 348 feature (Dousset et al., 2007; Swanson et al., 2015). It is known that preferential flow and transport 349 may exist in unconsolidated sandy soils. This applies to soils with available structural voids as a result of textural shift horizontally, funneling effect, or fluid instabilities due to viscosity or density 350 351 differences between the invading and resident water (Kamra et al., 2001). Further, chemical heterogeneities across various scales, which control chemical equilibrium (instantaneous sorption) or 352 non-equilibrium (kinetic sorption) transport processes, apparently contribute the non-Fickian 353 mechanisms. 354

355

356 3.3. Physical heterogeneity effects on non-Fickian transport

Due to minimal sorption in the sand-packed columns, column experiments a-d allow for evaluating transport behaviour quantitatively due to physical aspects alone – advection and dispersion. The main parameters influencing advective-dispersive transport are the fluid velocity and dispersion coefficient. Although a correlation between the tracer transport velocities and the volumetric flow rates is found by the ADE model (Table 2), the resultant fitting parameters do not

follow the expected Fickian trend, because the ADE is insufficient to capture the highly non-Fickian transport in column d. In contrast, transport velocity and dispersion coefficient for column d estimated by the CTRW-TPL are greater than those for columns b and c. This implies that columns b and c are likely forming more homogeneous packed structures, while column d is likely to have a relatively heterogeneous one. That is, the presence of macro-pores in column d may lead to overall faster transport and high dynamic dispersion.

368 Similar to fracture-matrix systems, where fracture leads to fast breakthrough and matrix contributes to the long tailing of BTCs (Kosakowski and Smith, 2005), the eluted mass fraction and 369 370 inverse modelling for the BTCs indicate that the presence of interconnected open channels, e.g. macro-pores or preferential flow paths, in conjunction with low velocity regions, e.g. in dead-end 371 channels or immobile zones, is responsible for the non-Fickian dispersion transport in heterogeneous 372 373 columns. Specifically, local flow velocity variations, i.e. low-velocity fluid near channel walls and 374 higher-velocity fluid near the centre of the channel, as well as the matrix diffusion, result in a skewed breakthrough curve with an earlier peak and longer tail. 375

Pore-scale heterogeneity and variations of fluids velocity in the pore space are known to yield 376 enhanced dispersive transport, associated with the Peclet number (Pe) (Dentz et al., 2011). 377 Depending on the selected length scale, Pe is presented in different forms (Bear and Cheng, 2010). 378 At the macroscopic level, $Pe = v L / D = L / \alpha$, where L is a characteristic length, represents the ratio 379 between the advective and dispersive transport terms. It increases when the advection dominates over 380 381 dispersion (Simunek et al., 2008). For $Pe \ll 1$, the advective mass flux may be neglected as much smaller than the dispersive one, and vice versa (Bear and Cheng, 2010). In this study, based on the 382 fitted parameters (Table 2) and the column length, L = 18cm, Pe is calculated for the experimental 383 384 columns with results presented in Table 3. The results, 1 < Pe < 200, show that both dispersion and advection are operating, e.g. there is a transition region to a power-law regime (Dentz et al., 2011). 385

In this case, the CTRW-based TPL model is valid for transport regimes. The possible effects of *Pe*on dispersive transport are not analysed here, but are subject of further investigations.

388

389 3.4. Chemical heterogeneity effects on non-Fickian transport

The column experiments e-h are designed to evaluate the effect of sorption on strontium transport. Expectedly, the tracer transport velocities decrease with increased clay content (Table 2), as strontium transport is significantly retarded. The emergent behaviour is non-Fickian dispersive transport, i.e. reduction and delay of the peak concentration and elongation of tailing. The larger the clay content, the greater the retardation and the reduction of strontium (Fig. 4e-h, Table 2). Inversely, the degree of non-Fickian transport is negatively associated with clay content (see Tables 1 and 2). Those phenomena are fundamentally due to the sorption/desorption of strontium on clay surface.

397 Sorption coefficients, represented in the form of R or K_d , are commonly used to approximate the extent of species sorption on various minerals, where typical sorption reactions are surface 398 complexation and ion exchange (Altmann et al., 2001). As sorption is commonly a complex function 399 400 of chemical properties, e.g. pH, ionic strength, mineral surface area, etc., the integral (constant) coefficient R or K_d used in ADE model often fail to capture such complexities. This is demonstrated 401 by our results that the equilibrium ADE is insufficient to fully capture measured BTCs. Moreover, 402 for spatially heterogeneous retardation properties, which can give rise to distinct non-Fickian 403 404 transport behaviours, the sorption coefficient R may not exist at relevant time and length scales 405 (Dentz and Castro, 2009). Consequently, the prediction of increasing R with increasing clay content by the ADE model (Table 2) is not sufficient to describe strontium transport in clayed media using 406 the equilibrium sorption model (Fig. 4e-h), because R reflects the simplified kinetic sorption process 407 408 to some degree but fails to capture the chemical heterogeneities, e.g. spatially varying sorption properties and specific reactive surface areas (Dentz et al., 2011). 409

410 In contrast, the CTRW framework with a sticking rate term is found capable of representing strontium transport (Fig. 4e-h), where slow and fast species transport is manipulated by the random 411 spatial and time increment as a stochastic process, and heterogeneity is mapped on the joint 412 413 distribution of transition length and times (Dentz et al., 2011). As expected, the average "sticking" rate Λ increases with increasing clay content (Table 2). This demonstrates that Λ represents 414 415 physically the degree of sorption of strontium on clay minerals, and shows that our CTRW-TPL model is robust in capturing non-Fickian transport when sorption is operating (Fig. 4e-h). Overall, 416 417 the CTRW model is of the greater flexibility to describe the local scale transport with spatially 418 varying equilibrium mass transfer and sorption-desorption reactions (Bromly and Hinz, 2004; Dentz and Castro, 2009), and thus can capture observed non-Fickian transport features, e.g. highly non-419 420 Gaussian BTCs.

421

422 3.5. Performance of ADE and CTRW methods in environmental modelling

Environmental modelling uses mathematics and computers to simulate physical and chemical phenomena across different environmental scales (Holzbecher, 2012). For nuclide transport related to contaminated sites remediation and safety assessments of a nuclear waste disposal, a number of macroscopic models based on ADE have been proposed, e.g. dual-porosity model, dual permeability model, two-phase model (Kapetas et al., 2014; Šimůnek and van Genuchten, 2008). The CTRW framework has mainly been applied for reactive transport, e.g. chemical sorption on minerals (Deng et al., 2013; Dentz et al., 2011; Edery et al., 2015).

In this work, inverse modelling by CXTFIT and CTRW toolbox have been used to obtain the relevant model parameters (Table 2). Overall, the CTRW performs better than the ADE, specifically in capturing measured BTCs, even though there should be little difference between them for column d, due to negligible sorption (Fig. 4). The equilibrium ADE modelling leads to broad range of parameters in capturing asymmetric BTCs by CXTFIT (Table 2), as the simple average deterministic 435 values may be insufficient to capture pervasive non-Fickian transport and estimate the randomness of tracer transference. The CTRW-TPL model appears to be the preferred method to describe this non-436 Fickian transport. Further, the CTRW-TPL is found insensitive to the magnitude of the Peclet 437 438 number and offers an efficient alternative for modelling species transport in convection-dominant flows (COMSOL, 2013). It is noted that some BTCs fitted with the ADE through CXTFIT are 439 similar to ones with TPL through CTRW toolbox (Fig. 4). This can be explained by the relative 440 flexibility of CXTFIT to fit certain BTCs. However, this flexibility appears to have limited validity, 441 excluding strongly preferential flow conditions (Kamra et al., 2001). 442

443 Both ADE and CTRW methods have been widely credited as effective tools to simulate species transport in heterogeneous conditions, utilising Lagrangian (particle tracing) and Eulerian (integral-444 445 differential equation) frameworks (Bakshevskaia and Pozdniakov, 2015; Hansen and Berkowitz, 446 2015; Rumynin, 2011). In fact, CTRW offers a general formalism which encompasses the ADE 447 models, and thus has comprehensive ability in treating non-Fickian dispersion transport (Berkowitz et al., 2006; Gao et al., 2009). Moreover, the probabilistic base of CTRW provides a more realistic 448 449 mathematical framework to model the advection-dispersion in porous media than the deterministic base of ADE (Kulasiri, 2013). However, there are unresolved issues with respect to the non-450 uniqueness of fitted parameters and inverse solution because of internal correlations between CTRW 451 parameters. This results in the existence of many local minima in the parameter estimation method 452 453 (Kulasiri, 2013). To surmount the problem, apart from collecting supplementary information which 454 is independent of the fitted parameter and reflects the actual migration processes, it is suggested to integrate the ADE and CTRW approaches to find optimal parameters to improve inversion problem. 455 For example, in this work we set admissible limits to the effective transport velocity and dispersion 456 457 coefficient found by ADE model as constraints to the CTRW. Then, in the CTRW framework, β can be estimated through analysis of the velocity field (Gao et al., 2009). Finally, the optimization 458 scheme allows for calculating reliable parameter values, which provide robust model prediction. 459

From a practical perspective, however, a disadvantage of CTRW is the lack of well-established public domain or commercial package; only CTRW MATLAB toolbox is available as mentioned in the section 2. In contrast, there are widely available user-friendly software packages for ADE, such as CTXFIT, HYDRUS (Šimůnek et al., 2008), MODFLOW (Chiang and Kinzelbach, 2005), upon which customized models can be built without sophisticated programming and coding.

465

466 **4. Conclusions**

Understanding release and transport of radioactive strontium is important in the context of 467 468 remedial options for contaminated sites and safety assessment of disposal in near-surface zone. Few investigations have focused on strontium non-Fickian transport with coupled advection, dispersion 469 470 and sorption processes, through an integrated approach of laboratory dynamic column experiments 471 and computer modelling. By re-analyzing BTCs from realistic column experiments of strontium 472 transport behaviour, comprehensive analysis for release and non-Fickian dispersive migration of strontium was carried out using the ADE-based CTXFIT and TPL-based CTRW toolbox. The results 473 474 allow for the following conclusions:

(1) Tracer release is found to be exponentially decaying from experimentally-informed
understanding that it is a continuous and decreasing process (Fig. 2); this is essential for defining
appropriate inlet conditions in future modelling.

(2) Observed reduction of eluted tracer mass (Fig. 3), 39%~58% of bromine and 16%~49% of
strontium, results from heterogeneities in the physical and chemical properties and processes:
molecular diffusion into/out immobile zone, strong preferential flow paths, sorption on the clay
surface, etc; this leads to non-Fickian behaviour with variable strength shown by early breakthroughs
and late tails.

(3) The CTRW-TPL model can capture non-Fickian transport with $0 < \beta < 2$ and Fickian transport with $\beta > 2$ (Table 2, Fig. 4). The non-Fickian dispersive transport is due to variation of flow fields and sorption retardation processes.

(4) The CTRW-TPL performs better than the ADE model in terms of capturing pervasive nonFickian transport, however, both ADE and CTRW models have positive and negative aspects for
practical use: integration of ADE and CTRW is suggested to obtain reliable parameters and provide
robust model prediction for both conservative and reactive transport.

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646 Tables

Table 1 Relationship between β and transport regimes¹

Value	Characteristics of parameters	Transport feature
0<\$<1	The first and the second moment of the $\psi(t)$ is	Indicates a highly non-Fickian
	inexistence; both transport velocity and dispersion	dispersion transport.
	coefficient vary with time as power laws.	
1<\$<2	The second moment of the $\psi(t)$ does not exist,	Presents moderate non-Fickian
	where transport velocity is constant but dispersion	transport or moderately
	coefficient still vary with time.	dispersive systems.
<i>β</i> ≥2	The first and second moments of the $\psi(t)$ exist, both	Displays Fickian transport,
	transport velocity and dispersion coefficient are	Gaussian distributions of plume
	constant.	

- ¹ Summarized from Li and Ren (2009), Wang and Cardenas (2014).
- 649
- 650

Table 2 Fitted values of parameters for ADE and CTRW

	ADE (CXTFIT)				СТ	RW (N	/IATLAB t	oolbox)		
	v^{I}	D^{I}		v_{ψ}	D_{ψ}	0				
	(cm/d)	(cm^2/d)	R ²	(cm/d)	(cm^2/d)	β	<i>t</i> ₁ (d)	<i>t</i> ₂ (d)	Λ	μ
a	79.78±5.78	286.56±1.44	0.997±0.111	55.34	178.56	1.80	1.48E-2	1.09	0.05	0.01
b	46.30±3.22	67.25±15.50	1.017±0.069	11.23	30.65	2.16	4.32E-2	5.93	0.04	0.01
c	22.50±0.59	11.10±2.20	1.039±0.027	10.8	3.44	2.27	4.25E-2	4.60	0.03	0.03
d	9.21±9.20	22.37 ±4.79	0.999±0.111	20.25	215.94	0.31	2.05E-2	4.06E3	0.09	0.08
e	1.05 ± 0.01	0.57 ± 0.03	1.053±0.027	0.52	0.12	1.43	2.53 E-1	3.62E3	0.12	0.12

	f	0.83±0.0	1 0.55	5±0.04 1	.076±0.048	0.47	0.10	1.26	3.69 E-1	6.59E3	0.13	0.09
	g	0.67±0.0	1 0.47	/±0.04 1	.093±0.029	0.38	0.12	0.62	6.24E-1	6.89E3	0.15	0.15
	h	0.58±0.0	1 0.43	8±0.06 1	.194±0.017	0.31	0.03	0.87	9.11E-1	1.07E4	0.18	0.16
652	(Columns a	ı-d: pack	ed by san	d, with spri	nkling volu	metric flo	w rat	e Q =100	0, 50, 20), 2 mI	_/d,
653	respectively; Columns e-h: $Q=2$ mL/d, packed by sand+clay (3:1), sand+clay (1:1), sand+clay (1:3),											
654	clay, respectively.											
655	¹ : The parameters values with 95% confidence limits.											
656												
657												
658		Table 3	Estimated	values of	Peclet numbe	er for the col	umns					
	-	column	a	b	c	d	e	f	g	5	h	
	-	Pe ¹	4.6~5.6	9.4~17.2	29.6~56.1	0.01~18.8	31.2~78	3 25.	1~84.6 2	3.3~57	20.9~1	86
659	-	¹ : The va	alues wer	e estimated	using the in	verse modell	ling resul	ts liste	d in Table	e 2.		
660												

661 **Figure captions**

662

Fig. 1. Schematic illustration of the experimental setup

664

Fig. 2. Simulated leach concentrations of bromine (a) and strontium (b) over column experimental duration under different sprinkling volumetric flow rate (Q) conditions according to equations (2~3).

667

Fig. 3. Observed cumulative mass fraction of bromine (a) and strontium (b) eluted from corresponding columns, calculated by Equation (13). Columns a-d: packed by sand, sprinkling volume flow rate Q = 100, 50, 20, 2 mL/d, respectively; Columns e-h: Q=2 mL/d, packed by sand and clay (3:1), sand and clay (1:1), sand and clay (1:3), clay, respectively.

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Fig. 4. Measured and best fitted BTCs of strontium using ADE/CTRW for column experiments. The corresponding fitted values of parameters for the columns are shown in Table 2. The first four columns (a-d) are packed with sand, with sprinkling volume flow rate Q = 100, 50, 20, 2 mL/d, respectively; the next three columns (e-h) are packed with different ratios of sand to clay: e 3:1, f 1:1, g 1:3, respectively. The column-h was packed with clay. Note that columns e-h are used by applying the same volumetric flow rate 2 mL/d. Dashed line is the equilibrium ADE model fit and solid line is the CTRW-based TPL model fit.

- 680
- 681



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Fig. 2b Click here to download high resolution image



Fig. 3a Click here to download high resolution image



Fig. 3b Click here to download high resolution image



Fig. 4a Click here to download high resolution image



Fig. 4b Click here to download high resolution image



Fig. 4c Click here to download high resolution image



Fig. 4d Click here to download high resolution image



Fig. 4e Click here to download high resolution image



Fig. 4f Click here to download high resolution image



Fig. 4g Click here to download high resolution image



Fig. 4h Click here to download high resolution image



Value	Characteristics of parameters	Transport feature
0<β<1	The first and the second moment of the $\psi(t)$ is	Indicates a highly non-Fickian
	inexistence; both transport velocity and	dispersion transport.
	dispersion coefficient vary with time as power	
	laws.	
1<β<2	The second moment of the $\psi(t)$ does not exist,	Presents moderate non-Fickian
	where transport velocity is constant but	transport or moderately
	dispersion coefficient still vary with time.	dispersive systems.
<i>β</i> ≥2	The first and second moments of the $\psi(t)$ exist,	Displays Fickian transport,
	both transport velocity and dispersion	Gaussian distributions of
	coefficient are constant.	plume

Table 1 Relationship between β and transport regimes¹

¹ Summarized from Li and Ren (2009), Wang and Cardenas (2014).

			CTRW (MATLAB toolbox)							
	v^{l} (cm/d)	D^{I} (cm ² /d)	R^{I}	$\frac{v_{\psi}}{(\text{cm/d})}$	D_{ψ} (cm ² /d)	β	<i>t</i> ₁ (d)	$t_2(\mathbf{d})$	Λ	μ
a	79.78±5.78	286.56±1.44	0.997±0.111	55.34	178.56	1.80	1.48E-2	1.09	0.05	0.01
b	46.30±3.22	67.25±15.50	1.017±0.069	11.23	30.65	2.16	4.32E-2	5.93	0.04	0.01
c	22.50±0.59	11.10±2.20	1.039±0.027	10.8	3.44	2.27	4.25E-2	4.60	0.03	0.03
d	9.21±9.20	22.37 ±4.79	0.999±0.111	20.25	215.94	0.31	2.05E-2	4.06E3	0.09	0.08
e	1.05 ± 0.01	0.57 ± 0.03	1.053±0.027	0.52	0.12	1.43	2.53 E-1	3.62E3	0.12	0.12
f	0.83±0.01	0.55 ± 0.04	1.076±0.048	0.47	0.10	1.26	3.69 E-1	6.59E3	0.13	0.09
g	0.67 ± 0.01	0.47 ± 0.04	1.093±0.029	0.38	0.12	0.62	6.24E-1	6.89E3	0.15	0.15
h	0.58±0.01	0.43±0.06	1.194±0.017	0.31	0.03	0.87	9.11E-1	1.07E4	0.18	0.16
	Columns a-d	: packed by s	and, with sprin	nkling vol	umetric fl	ow rat	$\overline{Q} = \overline{100}$, 50, 20,	, 2 mI	L/d,

Table 2 Fitted values of parameters for ADE and CTRW

Columns a-d: packed by sand, with sprinkling volumetric flow rate Q = 100, 50, 20, 2 mL/d, respectively; Columns e-h: Q=2 mL/d, packed by sand+clay (3:1), sand+clay (1:1), sand+clay (1:3), clay, respectively.

¹: The parameters values with 95% confidence limits.

ns
I

column	a	b	С	d	e	f	g	h
Pe^{I}	4.6~5.6	9.4~17.2	29.6~56.1	0.01~18.8	31.2~78	25.1~84.6	23.3~57	20.9~186

¹: The values were estimated using the inverse modelling results listed in Table 2.