



23 changing the pore space, which is more realistic compared to existing  
24 reactive transport models with assumed constant pore geometry. Anion  
25 exclusion effects are simulated by introducing diffuse double layer (DDL)  
26 in the model. Results agree well with experimentally measured diffusion  
27 coefficients for transport parallel and perpendicular to the bedding direction.  
28 The proposed model is applicable to a large class of geo-materials and  
29 suitable for linking to lattice models for deformation and damage.

30

31 **Keywords:** Opalinus Clay, pore network model, diffusion, anion  
32 exclusion, sorption

33

## 34 **1. Introduction**

35 A thorough understanding of the porous media mass transport is of prime  
36 importance in various science and engineering applications, e.g., in  
37 radioactive waste disposal (Yu and Neretnieks, 1997; Bourg *et al.*, 2003), in  
38 remediation of contaminated groundwater (Grathwohl, 1998), in tracer  
39 studies in oil recovery (Whitaker, 1967). The transport properties such as  
40 permeability and diffusivity vary with pore space changes due to various  
41 mechanical, chemical, physical and thermal processes. These processes can  
42 be simulated and analysed by developing appropriate physical and  
43 microstructure-informed models. Such models are required to be able to  
44 predict measurable transport properties at a macro-scale (considering

45 engineering and geological aspects of the formation) from measurable pore  
46 space characteristics such as pore shapes, pore size distribution. Pore-scale  
47 level approaches, such as Smoothed Particle Hydrodynamics (SPH) (Zhu  
48 and Fox, 2001) and Lattice Boltzmann (LB) (Kang *et al.*, 2006), have been  
49 used for computing single and multiphase flow directly on pore-space  
50 images. These methods are capable of treating complicated geometries and  
51 are very useful for understanding pore morphology effects on transport.  
52 However, they are computationally expensive and the conclusions that can  
53 be drawn are limited to the imaged pore systems, which are not necessarily  
54 representative for a larger volumes of the material analysed. In order to  
55 perform simulations on representative elementary volumes (REV) of porous  
56 media, such direct methods would require substantial computational effort.  
57 Furthermore, pore space information for some materials (such as Opalinus  
58 Clay (OPA)) and transport behaviour in these materials is dominated by  
59 meso-porosity (pores with diameters between 2nm and 50nm) and micro-  
60 porosity (pores with diameters smaller than 2nm) (NAGRA, 2002; Keller *et*  
61 *al.*, 2011). The existing imaging techniques do not have sufficient resolution  
62 to obtain detailed 3D images of such materials, to which methods, such as  
63 SPH and LB, can be applied.

64

65 On the other hand, pore network models (PNM) offer simplicity and  
66 computational efficiency that make them very attractive for modelling

67 transport through larger pore structures, e.g. of the order of tens to hundreds  
68 of inter-pore distances. It should be noted that PNM are conceptually scale  
69 indifferent, i.e. they can be applied to any length-scale interval where the  
70 structure of the pore space has been experimentally observed and analysed.  
71 For example, if a particular experimental technique allows for characterising  
72 pore features of sizes between 0.1 nm and 50 nm, the corresponding PNM  
73 can be constructed to capture effects within this length interval. In addition  
74 they provide a suitable representation of mutable pore space structures. In  
75 pore network models, the pore space is approximated by a set of sites and a  
76 set of bonds connecting some of the sites (Meyers and Liapis, 1999; Dillard  
77 and Blunt, 2000). Pore network models have to reflect the basic properties  
78 of porous media, such as shape and size distribution of pores and throats, as  
79 well as the pore coordination spectrum, i.e. the percentages of pores  
80 coordinated by different number of throats (Gao *et al.*, 2012; Jivkov *et al.*,  
81 2013). These basic properties can be obtained in structures with  
82 distinguishable pores and pore throats. However, in materials such as  
83 Opalinus Clay (OPA), the pore space is dominated by pores smaller than 50  
84 nm (NAGRA, 2002; Keller *et al.*, 2011), and due to the limitation resolution  
85 of current techniques the pore connectivity data cannot be extracted (Jivkov  
86 and Xiong, 2014). Hence, for such cases with limited pore space structure  
87 information, a different approach is required to construct effective pore  
88 networks. Previous approaches to tackle such incomplete pore space

89 information, including predefined connectivity to calculate length scale  
90 (Xiong *et al.*, 2014), and variable length scale to calculate connectivity  
91 (Jivkov and Xiong, 2014; Xiong and Jivkov, 2015), suffer from lack of an  
92 additional constraint. Notably, the length scale refers to the distance  
93 between the centres of neighbouring lattice sites, which corresponds to the  
94 average inter-pore distances in the porous medium. This can be overcome  
95 by considering the solid-phase structure of the material, e.g. the shape and  
96 size distribution of mineral grains. The solid phase characteristics are  
97 incorporated in this work to improve the realism of the constructed pore  
98 network model (PNM). The new method developed in the work can be  
99 coupled directly to the existing lattice models of the solid-phase previously  
100 developed for analysis of damage evolution via micro-cracking (Jivkov and  
101 Yates, 2012; Zhang and Jivkov, 2014).

102

103 The first objective of the work is to develop a methodology of pore network  
104 construction for materials with partially available experimental data. The  
105 second objective is to develop the model to account for anion exclusion and  
106 adsorption effects on the mass transport. The third objective is to validate  
107 the method with experimental data, for which OPA is selected.

## 108 **2. Methodology**

### 109 **2.1. Experimental data**

110 Opalinus Clay displays anisotropic responses to deformation and transport  
111 due to preferred orientation (or texture) of clay minerals attained during  
112 sedimentation and compaction (Wenk *et al.*, 2008). Specifically,  
113 experiments indicated anisotropic diffusion of solute species with slow  
114 diffusion perpendicular and fast diffusion parallel to the bedding plane. The  
115 goal is to construct a regular PNM in line with available structural data and  
116 macroscopic observations.

117

118 The pore space used in this work is OPA with sandy facies, labelled BDR in  
119 Keller *et al.* (Keller *et al.*, 2011; Keller *et al.*, 2013a), in which a large  
120 number of pores are located predominantly within the fine-grained clay  
121 mineral matrix. These pores with sizes  $> 10$  nm, called meso-pores, were  
122 elongated in the bedding plane, which was resolved by Focused Ion Beam  
123 nano-tomography (FIB-nt). The porosity of meso-pores was  $\theta_{mes} = 0.018$ .  
124 Pores with sizes  $< 10$ nm, called hereafter micro-pores, occupied  
125 approximately 9.7 vol.% (obtained from N<sub>2</sub> adsorption analysis). The  
126 porosity of micro-pores is thus  $\theta_{mic} = 0.097$ . Further, the meso-pores were  
127 largely isolated and did not provide a percolating network through the  
128 sample. These definitions of micro- and meso-pores are aligned with the

129 commonly used in physical chemistry and may differ from other fields of  
130 study.

131

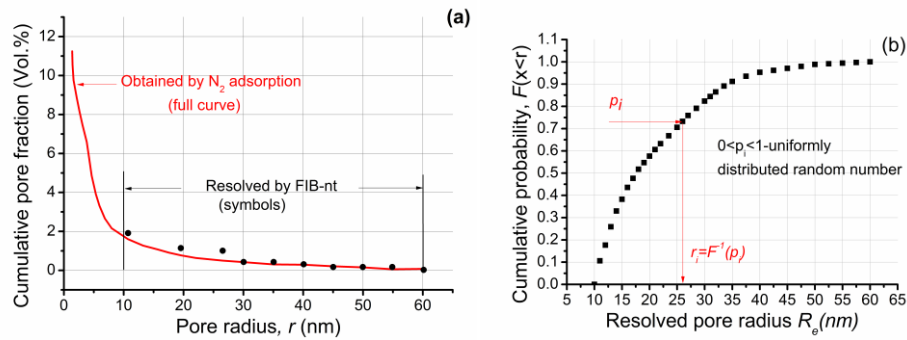
132 The above two measurements were combined into a single 'cumulative pore  
133 volume fraction – pore radius' curve given in Fig 1(a) (Keller *et al.*, 2011;  
134 Jivkov and Xiong, 2014). For model construction the experimental  
135 distribution of Fig. 1(a) is re-evaluated as cumulative probability separately  
136 for meso- and micro-pores. These are shown in Fig. 1(b) and Fig. 1(c),  
137 respectively.

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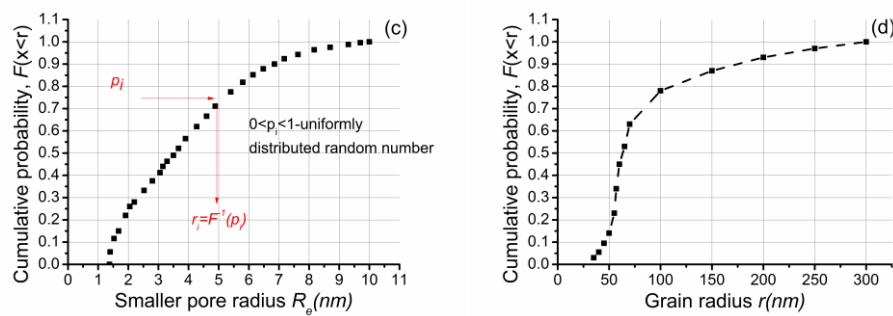
139 Regarding the solid phase, Keller et al. (Keller *et al.*, 2013b) reported 18  
140 vol.% non-porous carbonates with grain sizes ranging between 100 nm and  
141 300 nm and 17 vol.% of non-porous quartz, the grain size distribution of  
142 which was undetermined. For constructing the model for this study, the  
143 reported data were converted into cumulative probability of carbonate grain  
144 sizes as shown in Fig. 1(d). As both carbonates and quartz are non-porous,  
145 the quartz is assumed to follow the size distribution of the carbonate  
146 particles due to lack of quartz-specific experimental data.

147

148



149



150 Fig. 1 Microstructure characteristics of Opalinus Clay: (a) cumulative pore  
 151 volume fraction versus pore size determined by FIB-nt and N<sub>2</sub> adsorption  
 152 analyses of sample BDR (Keller *et al.*, 2011); (b) cumulative distribution of  
 153 meso-pore sizes; (c) cumulative distribution of micro-pore sizes with  
 154 assumed probability density (Jivkov and Xiong, 2014); (d) cumulative  
 155 distribution of carbonate grain sizes (Keller *et al.*, 2013b).

156

## 157 2.2. Pore network construction

158 Many pore networks are constructed based on different length scales due to  
 159 the broad range of pore size distribution. A short overview of existing multi-  
 160 scale pore network models is presented here.

161



162 Jiang et al. (Jiang *et al.*, 2013) developed a methodology to integrate  
163 networks extracted from images at distinct length scales. The pore network  
164 model was generated at each scale and then was integrated into a single two-  
165 scale network by characterizing the cross-scale connection structure  
166 between the two networks. The shortcoming of this method is that it is  
167 computationally costly due to the number of network elements (Bultreys *et*  
168 *al.*, 2015). Recognizing the computational problems when single micro-  
169 pores are taken into account, Mehmani and Prodanović (Mehmani and  
170 Prodanović, 2014) proposed a two-scale pore network by packing  
171 algorithms. The macro-network is constructed by Delaunay tessellation of  
172 the grain centres. Micro-porous networks are generated by downscaling  
173 existing networks extracted from macro-pores. This approach was capable  
174 of investigating fundamental two-phase flow properties of multi-scale  
175 porous media. A clear difference was observed between the behaviour of  
176 systems where micro-porosity was able to act in series with the macro-pores  
177 (intergranular or pore-filling micro-porosity) and systems where macro-  
178 porosity was able to act in parallel the meso-pores (intra-granular or  
179 dissolution micro-porosity). However, in the construction process, distorted  
180 pores were produced when many small grains touched a large grain. In  
181 addition, the ratio between macro and micro length scales needs to be  
182 determined for micro-porous regions. This ratio is difficult to obtain from  
183 image analysis, specifically for clays. Bultreys et al. (Bultreys *et al.*, 2015)

184 developed a workflow to integrate networks of macro-pores and micro-  
185 porous regions extracted from micro-CT images. This methodology allowed  
186 micro-porosity to act both in parallel and in series with the macro-pore  
187 network. However, a representative network for the micro-porosity is  
188 necessary. In addition, the pore networks from Jiang et al. (Jiang *et al.*, 2013)  
189 and Bultreys et al. (Bultreys *et al.*, 2015) are based on experimental data  
190 from micro-CT images which did not take into account the micro-pores that  
191 cannot be resolved by micro-CT. As the truncated cone shape is used to  
192 connect two neighbouring macro-pores, the tortuosity of the connection and  
193 geometric details about the bulk of the micro-porous cluster are neglected,  
194 which can lead to erroneous local conductivities. In this work, regular  
195 networks are used to generate models based on meso-porosity and micro-  
196 porosity.

197

198 The workflow is as follows: Firstly, the cellular basis is selected for  
199 allocation of grains and pores, which results in complementary (dual)  
200 lattices for solid and pore systems. In this work, the material was subdivided  
201 into cells, in which the truncated octahedrons represented the  
202 neighbourhoods of particles in the OPA. The truncated octahedron was the  
203 unit cell of a regular space tessellation, proposed for site-bond modelling of  
204 solids (Jivkov and Yates, 2012), which was used successfully for  
205 mechanical analysis (Jivkov *et al.*, 2012; Zhang and Jivkov, 2014) as well as

206 in the previous works on transport problems (Jivkov *et al.*, 2013; Jivkov and  
207 Xiong, 2014; Xiong *et al.*, 2014).

208

209 The particles or grains are associated with cell centres (interiors). This is  
210 illustrated in Fig. 2(a) for cells with equal distances between the three pairs  
211 of square faces, a setup used in previous works (Jivkov and Xiong, 2014;  
212 Xiong *et al.*, 2014). The geometry of the cellular assembly is described by  
213 three length parameters,  $S_1$ ,  $S_2$ , and  $S_3$ , measuring the distances between the  
214 square faces in directions  $(1, 0, 0)$ ,  $(0, 1, 0)$  and  $(0, 0, 1)$ , respectively. In an  
215 assembly of  $N_c$  cells, the particle radius,  $r_i$  from Fig. 1(d), is assigned in  
216 each cell. The volume of all allocated particles is required to be equal to the  
217 experimentally-measured particle volume fraction,  $\phi$ . From this requirement  
218 the volume of a cell assembly is calculated by:

$$219 \quad V_c = \frac{1}{\phi N_c} \sum_{i=1}^{N_c} \frac{4\pi r_i^3}{3} \quad \left( = \frac{1}{2} S_1 S_2 S_3 \right) \quad (1)$$

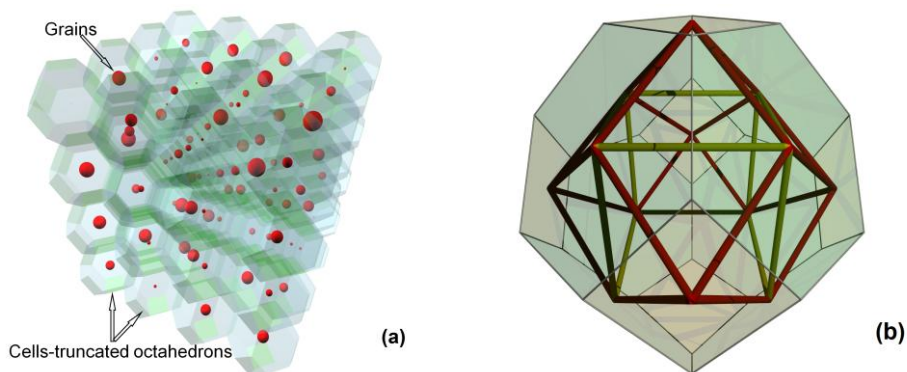
220 where  $V_c$  is the volume of a cell. The calculation of the three length  
221 parameters from Eq. (1) depends on the selection of their ratios used here to  
222 represent texture (a non-textured medium would have  $S_1 = S_2 = S_3$ ).

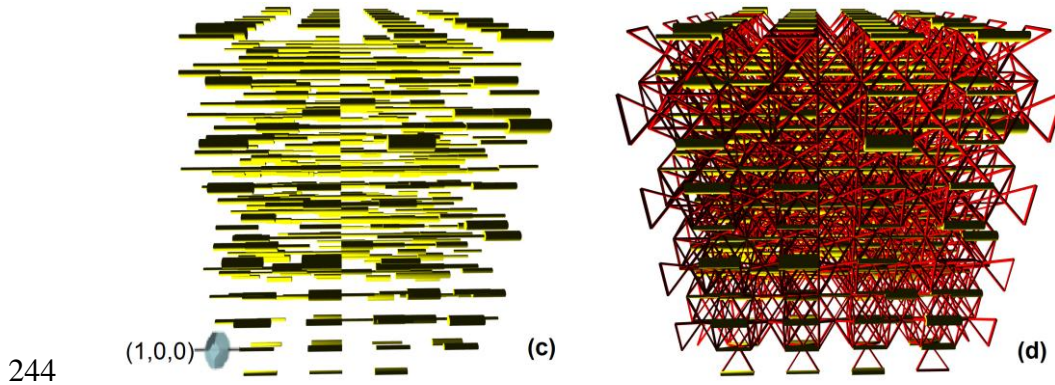
223

224 Considering the fact that the reported pore radii are derived with the  
225 assumption of cylindrical pore shapes (Keller *et al.*, 2011), a new pore  
226 network structure is proposed. Firstly, a skeleton is formed using sites at the

227 centres of cell faces and bonds between neighbouring faces. This is  
228 illustrated in Fig. 2(b) on a single cell with  $S_1 = S_2 = S_3$ . The bonds show the  
229 positions where elongated cylindrical pores are allowed to reside, i.e. they  
230 represent potential diffusion pathways. Since pores reside on bonds, the  
231 sites at face centres represent pore junctions – volume-less containers  
232 redirecting mass transport. Then the experimentally measured porosity is  
233 mapped to the proposed skeleton. The meso-pores are allocated along cell  
234 boundaries according to their size distribution, relative porosity and in  
235 preferred direction(s), while the micro-pores with radii selected from Fig.  
236 1(c) are assigned to bonds not already occupied by the meso-pores. The  
237 process terminates when the total pore volume fraction attains the  
238 experimental porosity. An example of a pore network model assigning  
239 meso-pores and micro-pore sets is shown in Fig. 2(c) and Fig. 2(d). More  
240 details about meso- and micro-pore allocation can be found in Jivkov and  
241 Xiong (Jivkov and Xiong, 2014).

242





245 Fig. 2 (a) a cellular assembly, illustrating particles of variable sizes allocated  
 246 to cell interiors; (b) unit cell illustrating potential diffusion paths (bonds) in  
 247 the neighbourhood of central particle; (c) exclusive assignment of meso-  
 248 pores along (1, 0, 0) taken as the clay bedding direction; (d) pore network  
 249 model after assignment of meso-pores (yellow) and micro-pores (red).  
 250 Examples are given on lattice with equal length parameters for bedding and  
 251 out-of-bedding directions.

252

### 253 2.3. Pore diffusivities

254 Real pores and throats have complex and highly irregular geometrical  
 255 profiles. In this work, a dimensionless circularity  $G$  was introduced to  
 256 approximate pores as cylindrical capillaries with a constant cross section:

$$257 \quad G = \frac{4\pi A_e}{P^2} \quad (2)$$

258 where  $A_e = \pi R_e^2$  [nm<sup>2</sup>] is the cross-section area,  $R_e$  = radius [nm] and  $P$  =  
 259 perimeter [nm]. The overall pore shape in the clay matrix parallel and

260 perpendicular to the bedding is elongated with a mean circularity of 0.45  
261 (Houben *et al.*, 2013).

262

263 The diffusion in the constructed model is driven by the concentration  
264 gradient along the cylindrical pores. For a pore with radius  $R_e$ , connecting  
265 nodes  $n$  and  $m$ , the mass flux of diffusing species,  $J_e$  [kg/m<sup>2</sup>s], is described  
266 by Fick's first law:

$$267 \quad J_e = -D_p G \Delta C_{nm} / L_e \quad (3)$$

268 where  $D_p$  [m<sup>2</sup>/s] is the pore diffusivity,  $L_e$  [m] is the pore length, and  $\Delta C_{nm} =$   
269  $C_n - C_m$  [kg/m<sup>3</sup>] is the concentration difference between the ends of a throat.

270 It is noted that in the pore bodies of this model are the cylindrically shaped  
271 network bonds, while the network nodes are only pore junctions, where  
272 concentrations are the main unknowns. Hence the gradient along the pore  
273 bodies is defined by the difference between the concentrations of the  
274 connected nodes.

275

276 The pore diffusivity of species of radius  $r_0$  is defined by taking into account  
277 the steric hindrance effects (Bryntesson, 2002):

$$278 \quad D_p = D_0 (1 - r_0 / R_e)^4 \quad (4)$$

279 where  $D_0$  is the free molecular diffusion coefficient of the species in the  
280 medium filling the pore system. For a spherical molecule in a circular  
281 cylinder, the term,  $(1 - r_0 / R_e)^4$ , is a geometric hindrance factor based on the

282 molecule size and pore size and shape (Bryntesson, 2002). Pore diffusivities  
283 enter a graph-theoretical representation of PNM, allowing for efficient  
284 solution of the transport through the network via discrete analysis on graphs  
285 as detailed by Jivkov and Xiong (Jivkov and Xiong, 2014). The mass flow  
286 through a pore,  $W$ , is then  $W = J_e \times A_e$ .

287

#### 288 **2.4. Anion exclusion effects**

289 The effective diffusion coefficients, which are a function of the tortuosity  
290 and constrictivity in the clay, and the accessible porosity, are different for  
291 various solutes (Appelo and Postma, 2005). In OPA, diffusion of anions is  
292 reduced relative to a neutral tracer such as tritiated water (HTO). The  
293 decreased diffusion is explained by anion exclusion effects. The anion  
294 exclusion effects are caused by the diffuse double layer (DDL) that  
295 surrounds the negatively charged clay surface. To calculate effective  
296 diffusion coefficients for anions, e.g.  $Cl^-$  and  $Br^-$ , the amount of DDL needs  
297 to be modelled firstly. According to Appelo and Wersin (2007), iodide  
298 cannot access half of the available porosity due to anion exclusion effect.  
299 Hence the amount of DDL-water is approximately equal to half of the  
300 porosity and the volume of DDL is achieved by increasing the layer  
301 thickness of DDL,  $r_t$ , in each pore. If  $R_e < r_t$ , the anions cannot pass through  
302 this pore, otherwise, the anion-accessible pore radius will be  $R^* = (R_e - r_t)$ .

303

304 In this work, the diffusion of HTO is firstly simulated with the total  
305 porosity, and then, model the  $Cl^-$  to find the anion-accessible porosity. The  
306 difference gives the amount of DDL-water which is assigned to be devoid of  
307 porosity. Iodide has the same accessible porosity as  $Cl^-$ . However, the flux  
308 front for  $I^-$  is retarded compared with that of  $Cl^-$  and  $Br^-$ . A likely  
309 explanation of this retardation is the linear adsorption of  $I^-$  onto the clay  
310 surface (Appelo *et al.*, 2010; Altmann *et al.*, 2012).

311

## 312 **2.5. Adsorption effects**

313 OPA has large specific surface area (Altmann *et al.*, 2012), high ion-  
314 exchange capacity (Altmann *et al.*, 2012), and adsorption affinity for ions,  
315 suggesting that long-term diffusivity of such species can be significantly  
316 affected by adsorption (Aytas *et al.*, 2009). The amount of species sorbed  
317 onto the solid,  $\beta$ , is described by the adsorption isotherm,  $\beta = \beta(C_M)$ , where  
318  $C_M$  is the local concentration of species M in the solution. Different species  
319 have linear or nonlinear adsorption processes (MontTerriProject, 2010).

320

321 In order to simulate adsorption effect, the adsorption of species needs to be  
322 judged firstly. In this work, the species are assumed to be strongly adsorbed  
323 so that the estimate of the adsorption obstruction effect can be obtained by  
324 considering the adsorbate to be uniformly smeared into a layer of thickness,  
325  $t$ , as described in Xiong *et al.* (Meyers and Liapis, 1999; Xiong *et al.*, 2014):



326 
$$t^2 - 2tR_e + 8\theta R_e r_M / 3 = 0 \quad (5)$$

327 where  $r_M$  is the radius of the diffusing species. If  $t < r_M$ , the species should  
 328 be weakly adsorbed. This means that the above adsorption equation (Eq. 5)  
 329 underestimates the adsorption obstruction effect (Xiong *et al.*, 2014). In this  
 330 case, the adsorption effects can be achieved by reducing the radii of throats  
 331 by a layer of thickness  $r_M$  (Meyers and Liapis, 1999). The radius of a throat  
 332 after adsorption becomes  $R^* = R_e - r_M$ . The distribution coefficient  $K_d$  of  $\Gamma$  is  
 333  $0.02 \text{ cm}^3/\text{g}$ , thus the linear adsorption isotherm,  $\beta$ , equals  $K_d C_M$  (Van Loon  
 334 *et al.*, 2003).

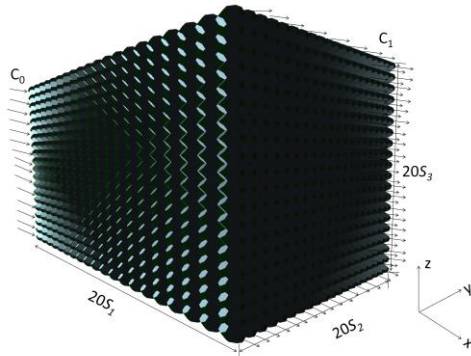
335

### 336 **3. Results and discussion**

337 With respect to a coordinate system  $(X_1, X_2, X_3)$  normal to the square faces  
 338 of the unit cell (see Fig. 2b), a pore network skeleton within the boxed  
 339 region ( $0 \leq X_1 \leq 20S_1, 0 \leq X_2 \leq 20S_2, 0 \leq X_3 \leq 20S_3$ ) was used. Tests with  
 340 increasing models sizes were initially performed. When the model size was  
 341  $20S_1 \times 20S_2 \times 20S_3$ , the variation of calculated diffusivity with different model  
 342 realisations (grain size distribution and pore size distribution) reduced to  
 343 under 10%, which was accepted as reasonable accuracy.

344 The selection of the two boundaries depends on the macroscopic diffusivity  
 345 being analysed. Specifically, the boundary conditions used to calculate the  
 346 macroscopic diffusivity  $D_1$  parallel to bedding direction  $(1, 0, 0)$  are: zero  
 347 flux through all nodes on planes  $X_2 = 0, X_2 = 20S_2, X_3 = 0, X_3 = 20S_3$ . For  
 348 calculating the macroscopic diffusivity  $D_2$  perpendicular to the bedding  
 349 direction, say  $(0, 1, 0)$ , the boundary conditions are: prescribed

350 concentration  $C_0$  in all nodes on plane  $X_2 = 0$ ; prescribed concentration  $C_1$  in  
 351 all nodes on plane  $X_2 = 20S_2$ ; zero flux through all nodes on planes  $X_1 = 0$ ,  
 352  $X_1 = 20S_1$ ,  $X_3 = 0$ ,  $X_3 = 20S_3$ . An illustration is given in Fig. 3.  
 353



354  
 355 Fig. 3. Example of a pore network model ( $S_1=2S_2=2S_3$ ) with applied  
 356 boundary conditions: prescribed concentration  $C_0$  in all nodes on plane  $X_1 = 0$ ;  
 357 prescribed concentration  $C_1$  in all nodes on plane  $X_1 = 20S_1$ .

358

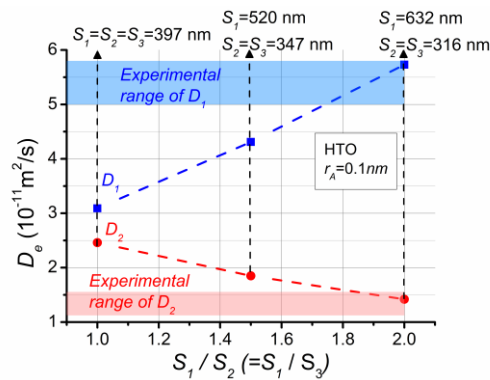
### 359 3.1. Texture effect on diffusivity

360 The pore networks constructed exhibit macroscopic tortuosity (path  
 361 lengthening), introduced by the selection of diffusion pathways along the  
 362 interfaces between solid-phase regions. This tortuosity depends on the  
 363 material texture, represented here by the ratios of the cell length parameters  
 364 in three perpendicular directions. Experimental results show that tortuosity  
 365 is larger in a direction perpendicular to the bedding plane (Van Loon *et al.*,  
 366 2004). Therefore, the effects of larger cell length along the bedding plane  
 367 and smaller cell lengths perpendicular to the bedding plane are investigated.

368

369 The out-of-bedding directions are not differentiated, i.e. the case of having  
370  $S_2 = S_3$  is assumed. The ratio  $S_1 / S_2$  cannot be calculated directly from the  
371 currently available experimental information on the clay pore space. Since a lot of  
372 experimental data of HTO species diffusion has been published, we compare HTO  
373 experimental diffusivity to model predictions with variable  $S_1 / S_2$  to find the ratio  
374 yielding best agreement. Once  $S_1 / S_2$  is decided, the model can be used to simulate  
375 transport of other species such as neutral species and ions. Three ratios of the cell  
376 length parameters are considered:  $S_1 / S_2 = 2, 1.5, 1$ . For each cellular  
377 assembly of given ratio, 10 realisations of pore spatial distributions were  
378 analysed to obtain the diffusion coefficients along the bedding plane,  $D_1$ ,  
379 and out-of-bedding directions,  $D_2$ . The results reported in Fig. 3 are the  
380 averaged values of these analyses on the basis of containing 35 vol.% solid  
381 particles. The calculated diffusion coefficients of HTO are in the following  
382 ranges:  $D_1 = 3.09 \times 10^{-11} \sim 5.73 \times 10^{-11} \text{ m}^2/\text{s}$ ;  $D_2 = 1.42 \times 10^{-11} \sim 2.46 \times 10^{-11} \text{ m}^2/\text{s}$   
383 (see Fig. 3). Reported experimentally obtained values for HTO diffusion in  
384 OPA shown in Fig. 3 are  $D_1 = (5.4 \pm 0.4) \times 10^{-11} \text{ m}^2/\text{s}$  (Van Loon *et al.*, 2004;  
385 MontTerriProject, 2010) and  $D_2 = 1.13 \times 10^{-11} \sim 1.55 \times 10^{-11} \text{ m}^2/\text{s}$  (Van Loon  
386 *et al.*, 2003; Joseph *et al.*, 2013). When  $S_1/S_2 = 2$ , the simulated results are  
387 closest to the experimental values. This ratio is used to calculate  
388 diffusivities for anions in the following section. The difference between the  
389 computational and the experimental results could be partially due to a

390 difference between the microstructure characteristics of OPA obtained by  
 391 Keller et al. (Keller *et al.*, 2011; Keller *et al.*, 2013a) and used for model  
 392 construction, and the clays used for experimental measurement of effective  
 393 diffusion coefficients (Van Loon *et al.*, 2003; Joseph *et al.*, 2013).  
 394



395  
 396 Fig. 4. Calculated macroscopic (effective) diffusivity of HTO in OPA. Clay  
 397 textures are measured by the ratio of length parameters in the bedding,  $S_1$ ,  
 398 and the out-of-bedding,  $S_2$ , directions.  $D_1$  and  $D_2$  are the diffusion  
 399 coefficients parallel and perpendicular to the bedding plane.

400

### 401 3.2. Anion exclusion and adsorption effect on diffusivity

402 The macroscopic diffusivities of  $Cl^-$  and  $I^-$  are shown in Table 1. The  
 403 simulated results of  $Cl^-$  are very close to the experimental effective  
 404 diffusivities (Van Loon *et al.*, 2003). The predicted  $D_1$  of anions is a little  
 405 overestimated and  $D_2$  of anions is underestimated. This is in agreement with  
 406 the expectations for the following reasons. Firstly, allocation of  $r_t$  leads to  
 407 larger number of micro-pores belonging to DDL and a higher increase of

408 macroscopic tortuosity in out-of-bedding directions. Since pores allocated in  
409 out-of-bedding directions are only micro-pores and the layer thickness of  
410 DDL,  $r_t$ , assigned in each pore is the same. A larger fraction of micro-pores  
411 is not accessible for the anions in out-of-bedding direction compared to the  
412 number of pores in bedding plane direction. As a result, the effective  
413 diffusivities in out-of-bedding directions are under estimated. Because the  
414 total DDL volume is 50 vol.% of the pores and the micro-pores in out-of-  
415 bedding directions contribute more to the DDL volume, the effective  
416 diffusivities in the bedding plane direction will be over estimated. Secondly,  
417 the accessible porosity for anions should be in a small range (Van Loon *et*  
418 *al.*, 2003; MontTerriProject, 2010). This model can be used to calculate  
419 other anion effective diffusivities which only have anion exclusion effects.

420

421 The diffusion behaviour of  $I$  is the effect of anion exclusion and adsorption.  
422 Thus, the adsorption is simulated on the basis of the same accessible  
423 porosity as of  $Cl$ . The predicted effective diffusivities of  $I$  are in agreement  
424 with measured values as shown in Table 1. The over/under estimation of  
425 experimental value,  $D_1$  and  $D_2$ , is mainly due to the same reason (exclusion  
426 effects) as for  $Cl$ . Another reason for difference between predicted results  
427 and experimental values could be that the adsorption isotherm measured by  
428 through-diffusion technique (Van Loon *et al.*, 2003) is not quite accurate to  
429 describe adsorption onto pore walls.

430

431 Table1. Comparison of experimental (EXP) and computational (COM)  
432 results for diffusion of different species:  $D_0$ =free molecular diffusion  
433 coefficient;  $D_1$ = diffusivity in OPA bedding direction;  $D_2$ = diffusivity in  
434 OPA out-of-bedding direction. All diffusion coefficients are scaled by  $10^{-12}$   
435  $m^2/s$ .

Species	$D_0$	$D_1$		$D_2$	
		EXP	MOD	EXP	MOD
$Cl$	2030	15~17	18	5.31~5.77	3.82
$I$	2000	10~14	16	4.28~4.86	3.36

436 Note: The free molecular diffusion coefficients  $D_0$  at room temperature are  
437 obtained from Mont Terri Project (MontTerriProject, 2010). Experimental  
438 effective diffusivities are obtained from Van Loon, LR (Van Loon *et al.*,  
439 2003).

440

#### 441 **4. Conclusions**

442 This work presented:

- 443 • A new method for pore network construction, where solid-phase  
444 characteristics of a porous media are used to balance incomplete  
445 knowledge of geometrical and topological pore system  
446 characteristics;
- 447 • A framework for modelling the diffusion anion species due to purely  
448 anion exclusion effects in clay. This could be used to clarify the  
449 kinds of species transport (anion, cation or neutral species) in OPA,

450 as in most experiments, only species in the source reservoir solution  
451 are known. For example, if neutral species with comparable  
452 molecular size of HTO are assumed to diffuse in OPA, the  
453 accessible porosity applied to HTO could be used to simulate the  
454 diffusion of this species. If the simulated effective diffusivities are in  
455 good agreement with experimental values, the assumption is correct.

- 456 • An algorithm to simulate adsorption by changing pore space, which  
457 is more realistic compared to simulations based on constant pore  
458 geometry.

459

460 With simulations for three different solutes, it has been demonstrated that  
461 the outcomes are within the ranges of reported experimental data,  
462 particularly when clay heterogeneity is simulated via different length  
463 parameters along bedding plane and in directions perpendicular to the  
464 bedding plane. In principle, this model can also be applied to model other  
465 neutral species and anions.

466

467 The methodology is particularly suitable for modelling and simulation of  
468 micro- and meso-porous materials, such as bentonite and shale rocks. From  
469 one side, the existing experimental techniques do not allow for complete  
470 quantitative analysis of their pore systems. From another, the effect of

471 damage (micro-cracking) on mass transport in such materials is fundamental  
472 to the understanding of their retention or release functions.

473

474 As the description of the network topology is very important for accurately  
475 predicting residual phase saturation (Sok *et al.*, 2002), the feasibility of  
476 simulating multi-phase flow by the proposed PNM needs further study.

477

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483

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