

Contents lists available at [SciVerse ScienceDirect](http://SciVerse.Sciencedirect.com)

Tetrahedron Letters

journal homepage: www.elsevier.com/locate/tetlet

Simple and efficient protocol for the synthesis of novel dihydro-1*H*-pyrano[2,3-*c*]pyrazol-6-ones via a one-pot four-component reaction

Sadeq Hamood Saleh Azzam, M. A. Pasha*

Department of Studies in Chemistry, Central College Campus, Palace Road, Bangalore University, Bengaluru 560001, India

ARTICLE INFO

Article history:

Received 5 July 2012

Revised 3 October 2012

Accepted 5 October 2012

Available online 13 October 2012

Keywords:

Pyranopyrazole-6-ones

Aromatic aldehydes

Meldrums acid

Ethyl acetoacetate

Hydrazine hydrate

Ba(OH)₂

Water

ABSTRACT

Ba(OH)₂ catalyzed simple and efficient one-pot four-component reaction of Meldrums acid, ethyl acetoacetate, hydrazine hydrate, and aromatic aldehydes to give 3-methyl-4-aryl-4,5-dihydro-1*H*-pyrano[2,3-*c*]pyrazol-6-ones in refluxing water is reported. The yields are high and the reactions go to completion in 1–2 h.

© 2012 Elsevier Ltd. All rights reserved.

Multicomponent reaction (MCR) is a process in which three or more accessible components are combined together in one-pot to produce a final product which shows the features of all the input reactants and therefore, offers the greatest possibilities for molecular diversity in one step with minimum synthetic time and effort.¹ As MCRs are one-pot reactions, they are easier to carry out than the multistep syntheses. This strategy is an important development in the drug discovery in the context of rapid identification and optimization of biologically active lead compounds.² In addition, MCRs are environmentally friendly, and often proceed with excellent chemoselectivities.³ There are three wings (techniques) of green chemistry which, if combined, would result in an excellent green chemistry protocol.⁴ These techniques are: the efficient use of solvent-free reactions,⁵ reusability of heterogeneous catalysts,⁶ and use of multicomponent reactions.^{6a}

One of the most challenging aspects in the medicinal chemistry is the design and synthesis of biologically active compounds,⁷ and dihydro-1*H*-pyrano[2,3-*c*]pyrazoles represent an interesting template for medicinal chemistry and play an essential role as biologically active molecules.⁸ Many of the pyrano[2,3-*c*]pyrazoles are known for their antimicrobial,⁹ insecticidal,¹⁰ anti-inflammatory,¹¹ anticancer,¹² and molluscicidal activities.¹³

During the last few years, synthesis of dihydropyrano[2,3-*c*]pyrazoles has received great interest.¹⁴ Pyranopyrazoles

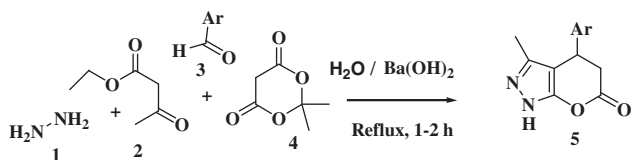
are also used as pharmaceutical ingredients and biodegradable agrochemicals.¹⁵ The first reported pyranopyrazole was synthesized by the reaction between 3-methyl-1-phenyl-pyrazolin-5-one and tetracyanoethylene.¹⁵

Another attractive area in green chemistry is designing organic reactions in aqueous media.¹⁶ Water offers several benefits such as control over exothermic reactions, salting out, and salting in, as well as variation of pH.¹⁷ We have earlier reported the synthesis of 6-amino-3-methyl-4-aryl-1,4-dihydropyrano[2,3-*c*]pyrazol-5-carbonitriles using glycine,¹⁸ iodine,¹⁹ and imidazole²⁰ as catalysts in water. In continuation of our efforts to develop methods for the synthesis of novel heterocyclic compounds using readily available, inexpensive, and environmentally friendly catalysts,^{21–27} herein, we report a rapid and efficient one-pot four-component synthesis of some novel 3-methyl-4-aryl-4,5-dihydro-1*H*-pyrano[2,3-*c*]pyrazol-6-ones²⁸ by the reaction of aromatic aldehydes, Meldrums acid, hydrazine hydrate, and ethyl acetoacetate in the presence of readily available, inexpensive, mild, green, and common laboratory chemical Ba(OH)₂ as a basic catalyst in water (Scheme 1).

In order to optimize the reaction conditions, we carried out the reaction between 3,4,5-trimethoxybenzaldehyde, ethyl acetoacetate, Meldrums acid, and hydrazine hydrate in the presence of 10 mol % of Ba(OH)₂ at reflux in different solvents such as EtOH, MeOH, H₂O, and CH₃CN; among all these solvents, H₂O was found to be the best in terms of the yield of the product and time of completion compared to common organic solvents. The results of this study are presented in Table 1.

* Corresponding author. Tel./fax: +91 80 22961337.

E-mail address: mafasha@gmail.com (M.A. Pasha).



Scheme 1. Synthesis of 3-methyl-4-aryl-4,5-dihydro-1H-pyrano[2,3-c]pyrazol-6-ones

Table 1
Effect of solvent on the synthesis of 3-methyl-4-(3',4',5'-trimethoxyphenyl)-4,5-dihydro-1H-pyrano[2,3-c]pyrazol-6-one (**5a**)^a

Entry	Solvent	Time (h)	Yield (%)
1	Ethanol	5	90
2	Methanol	6	87
3	CH ₃ CN	4	85
4	H ₂ O	1.5	93

^a Reactions are performed on a 1 mmol scale of the reactants.

Table 2
Influence of various catalysts on the synthesis of 3-methyl-4-(3',4',5'-trimethoxyphenyl)-4,5-dihydro-1H-pyrano[2,3-c]pyrazol-6-one (**5a**)^a

Entry	Catalyst (10 mol %)	Time (h)	Yield (%)
1	K ₂ CO ₃	8	30
2	Piperidine	7	40
3	NaOH	3	65
4	Ba(OH) ₂	1.5	93

^a Reactions are performed on a 1 mmol scale of the reactants.

Table 3
Effect of the amount of Ba(OH)₂ on the synthesis of (**5a**)^a

Entry	Ba(OH) ₂ (mol %)	Yield (%)
1	5	50
2	7	65
3	10	93
4	12	93

^a Reactions are performed on a 1 mmol scale of the reactants.

Table 4
Synthesis of 3-methyl-4-aryl-4,5-dihydro-1H-pyrano[2,3-c]pyrazol-6-ones

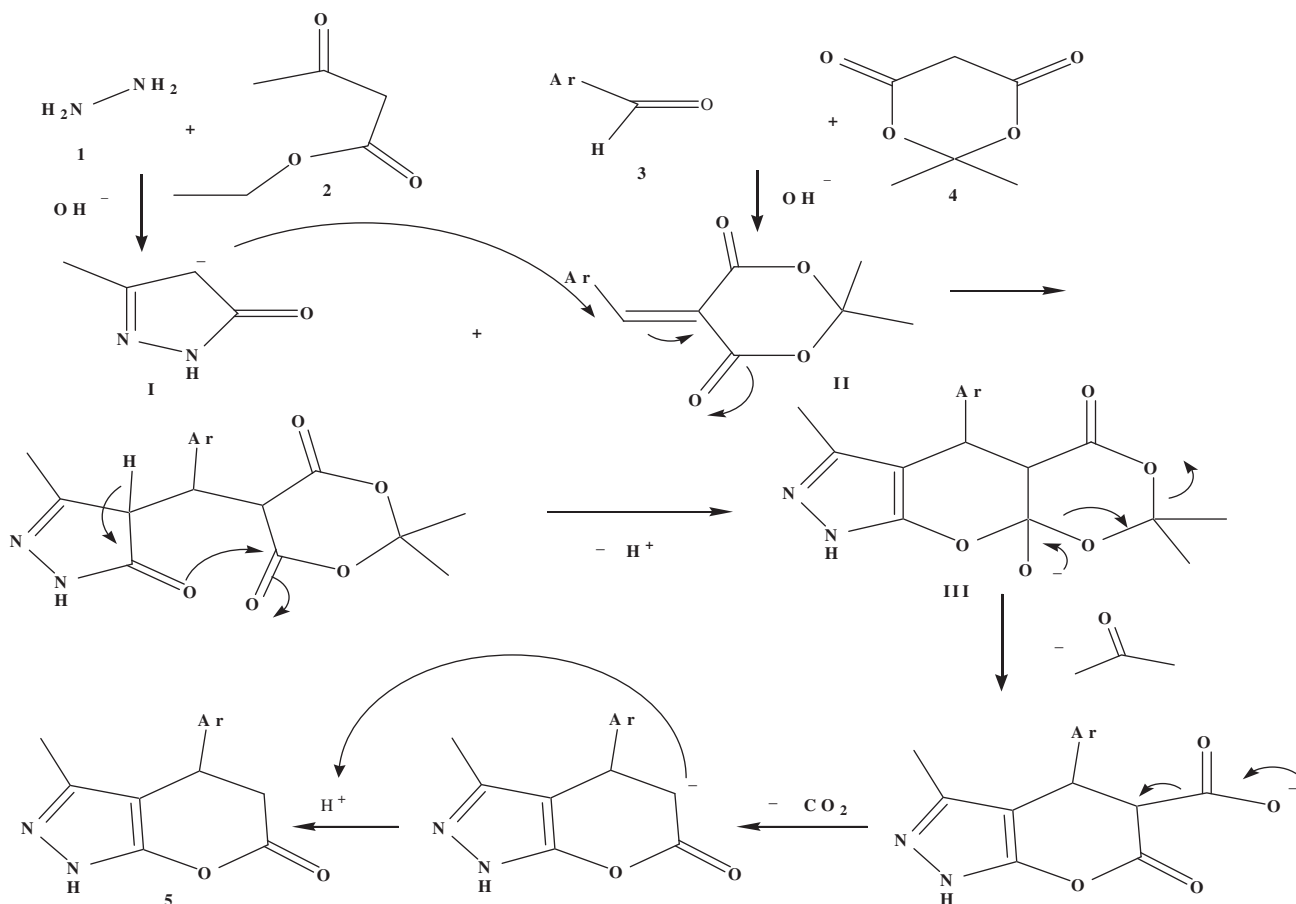
Entry	Aldehyde (3)	Product ^a	Time (h)	Yield ^b (%)	MP (°C)
1	3,4,5-(MeO) ₃ C ₆ H ₂ CHO	5a	1.5	93	205
2	4-MeOC ₆ H ₄ CHO	5b	1	93	157–160
3	3-MeOC ₆ H ₄ CHO	5c	1.5	93	133–135
4	2-ClC ₆ H ₄ CHO	5d	2	90	142–145
5	2-HOC ₆ H ₄ CHO	5e	2	92	235–237
6	4-FC ₆ H ₄ CHO	5f	2	90	178–180
7	3-NO ₂ C ₆ H ₄ CHO	5g	2	91	212
8	HCHO	5h	10	ND	–
9	CH ₃ CHO	5i	10	ND	–
10	CH ₃ CH ₂ CHO	5j	10	ND	–
11	CH ₃ CH ₂ CH ₂ CHO	5k	10	ND	–

ND-not detected.

^a All isolated products are new and were characterized by IR, ¹H NMR, and ¹³C NMR spectral analyses and CHN analysis.

^b Isolated yields.

To select the best catalyst, we carried out the reaction between 3,4,5-trimethoxybenzaldehyde, ethyl acetoacetate, Meldrum's acid,



Scheme 2. Mechanism of formation of 3-methyl-4-aryl-4,5-dihydro-1H-pyrano[2,3-c]pyrazol-6-ones.

and hydrazine hydrate in the presence of 10 mol % of different basic catalysts such as Ba(OH)₂, K₂CO₃, piperidine, and NaOH. We found that, K₂CO₃ did not afford the product in good yield and reaction time was very long, similar results were obtained with piperidine. The yield of the desired product improved to a very less extent when NaOH was used as a basic catalyst and the product was a mixture and a sticky mass. When the same reaction was carried out in the presence of Ba(OH)₂, the product was obtained in very high yield (93%) within 1.5 h (Table 2, entry 4). The results of this study are presented in Table 2.

We have also varied the amount of Ba(OH)₂ from 5, 7, and 10 to 12 mol % and the results revealed that, 10 mol % gives excellent yield of the product in a short duration as shown in Table 3.

After optimizing the conditions, the generality of this method was examined by the reaction of different substituted aldehydes with ethyl acetoacetate, Meldrums acid, and hydrazine hydrate in the presence of 10 mol % Ba(OH)₂ in water under reflux. We also examined the use of aliphatic aldehydes to get the corresponding products (Table 4 entries 8–11) but there was no product formation even after 10 h under the optimized reaction conditions, and the results of this study are shown in Table 4.

It is found that, various aromatic aldehydes containing electron-donating or electron-withdrawing functional groups at different positions did show a difference in the reaction time but the yields of products were almost same (Table 4).

The formation of the product in the present reaction is expected to involve the following tandem reaction mechanism:

Formation of pyrazolone I by the reaction between **1** and **2** and Knoevenagel condensation between **3** and **4** to give **II**. Michael addition of **I** with **II** followed by cyclization is expected to give a tricyclic intermediate **III** which may lose a molecule of acetone and a molecule of CO₂ in subsequent steps to give the final product **5** as shown in Scheme 2. In order to establish the mechanism of the reaction, the intermediates-pyrazolone²⁹ and the Knoevenagel adduct³⁰ were prepared separately (characterized by the ¹H NMR and ¹³C NMR spectral analysis) and were treated with each other to get the product **5a** under the standardized reaction condition, which clearly indicates that the intermediates I and II are formed during the course of the present reaction.

In summary, we have demonstrated a simple, efficient, and a novel one-pot four-component protocol for the synthesis of some new pyranopyrazol-6-one derivatives in water using Ba(OH)₂ as a readily available, inexpensive, and efficient catalyst. The advantages offered by this method are: simple reaction condition, short reaction time, ease of product isolation, and excellent yields. We wish to state that this method involves environmentally friendly procedure, and is the first procedure for the synthesis of novel 3-methyl-4-aryl-4,5-dihydro-1H-pyrano[2,3-c]pyrazol-6-one derivatives.

Acknowledgments

S.H.S.A. gratefully acknowledges the Sana University, Yemen for a fellowship; and the authors thank the VGST, Dept. of Science & Technology, Government of Karnataka for the CESEM Award Grant No. 24 (2010–2011).

Supplementary data

Supplementary data associated with this article can be found, in the online version, at <http://dx.doi.org/10.1016/j.tetlet.2012.10.025>.

References and notes

- (a) Ugi, I. *Pure Appl. Chem.* **2001**, *73*, 187; (b) Dömling, A.; Ugi, I. *Angew. Chem., Int. Ed.* **2000**, *39*, 3168.
- (a) Newton, R. sp2. 2002 (Nov.), 16–18 (see: www.sp2.uk.com); (b) Teague, S. J.; Davis, A. M.; Leeson, P. D.; Oprea, T. *Angew. Chem., Int. Ed.* **1999**, *38*, 3743; (c) Armstrong, R. W.; Combs, A. P.; Tempest, P. A.; Brown, S. D.; Keating, T. A. *Acc. Chem. Res.* **1996**, *29*, 123; (d) Schreiber, S. L. *Science* **1964**, *2000*, 287.
- (a) Tietze, L. F. *Chem. Rev.* **1996**, *96*, 115; (b) Tietze, L. F.; Haunert, F. In *Stimulating Concepts in Chemistry*; Votle, F.; Stoddart, J. F.; Shibasaki, M., Eds.; Wiley: Weinheim, 2000; pp 39–64; (c) Enders, D.; Huttli, M. R. M.; Grondal, C.; Raabe, G. *Nature* **2006**, 861.
- Sheldon, R. A. *Green Chem.* **2005**, *7*, 267.
- (a) Wang, G.-W.; Komatsu, K.; Murata, Y.; Shiro, M. *Nature* **1997**, *387*, 583; (b) Varma, R. S.; Kumar, D.; Dahiya, R. *J. Chem. Res.* **1998**, *s*, 324; (c) Nozoe, T.; Tanimoto, K.; Takemitsu, T.; Kitamura, T.; Harada, T.; Osawa, T.; Takayasu, O. *Solid State Ionics* **2001**, *141*, 695; (d) Lamartine, R.; Perrin, R. In *Spill Over of Adsorbed Species*; Pajonk, G. M., Teichner, S. J., Germain, J. E., Eds.; Elsevier: Amsterdam, 1983; p 251.
- (a) Tsukinoki, T.; Nagashima, S.; Mitoma, Y.; Tashiro, M. *Green Chem.* **2000**, *2*, 117; (b) Ballini, R.; Bosica, G.; Maggi, R.; Ricciutelli, M.; Righi, P.; Sartori, G.; Sartorio, R. *Green Chem.* **2001**, *3*, 178; (c) Amantini, D.; Fringuelli, F.; Piermatti, O.; Pizzo, F.; Vaccaro, L. *Green Chem.* **2001**, *3*, 229.
- Hailes, H. C. *Org. Process Res. Dev.* **2007**, *11*, 114.
- Kuppusamy, K.; Kasi, P. *Tetrahedron Lett.* **2010**, *51*, 3312.
- El-Tamany, E. S.; El-Shahed, F. A.; Mohamed, B. H. *J. Serb. Chem. Soc.* **1999**, *64*, 9.
- Ismail, Z. H.; Aly, G. M.; El-Degwi, M. S.; Heiba, H. I.; Ghorab, M. M. *J. Egypt Biot.* **2003**, *13*, 73.
- Zaki, M. E. A.; Soliman, H. A.; Hiekal, O. A.; Rashad, A. E. Z. *Naturforsch., C.* **2006**, *61*, 1.
- Wang, J. L.; Liu, D.; Zhang, Z. J.; Shan, S.; Han, X.; Srinivasula, S. M.; Croce, C. M.; Alnemri, E. S.; Huang, Z. *Proc. Natl. Acad. Sci. U.S.A.* **2000**, *97*, 7124.
- (a) Abdelrazek, F. M.; Metz, P.; Metwally, N. H.; El-Mahrouky, S. F. *Arch. Pharm.* **2006**, *339*, 456; (b) Abdelrazek, F. M.; Metz, P.; Kataeva, O.; Jaeger, A.; El-Mahrouky, S. F. *Arch. Pharm.* **2007**, *340*, 543.
- (a) Sharanin, Y. A.; Sharanina, L. G.; Puzanova, V. V. *Zh. Org. Khim.* **1983**, *19*, 2609; (b) Rodinovskaya, L. A.; Gromova, A. V.; Shestopalov, A. M.; Nesterov, V. N. *Russ. Chem. Bull., Int. Ed.* **2003**, *52*, 2207.
- Junek, H.; Aigner, H. *Chem. Ber.* **1973**, *106*, 914.
- (a) Herrerias, C. I.; Yao, X.; Li, Z.; Li, C. *Chem. Rev.* **2007**, *107*, 2546; (b) Li, C. J.; Chan, T. H. *Comprehensive Organic Reactions in Aqueous Media*; John Wiley & Sons, 2007; (c) Grieco, P. A. *Organic Reactions in Water*; Thomson R Science Glasgow: Scotland, 1998.
- Gnanasambandam, V.; Kandhasamy, K. *Tetrahedron Lett.* **2008**, *49*, 5636.
- Madhusudana Reddy, M. B.; Pasha, M. A.; Jayashankara, V. P. *Synth. Commun.* **2010**, *40*, 2930.
- Madhusudana Reddy, M. B.; Pasha, M. A. *J. Indian Chem.* **2012**, *51B*, 537.
- Siddekha, A.; Nizam, A.; Pasha, M. A. *Spectrochim. Acta, Part A* **2011**, *81*, 431.
- Madhusudana Reddy, M. B.; Pasha, M. A. *Synth. Commun.* **1995**, *2010*, 40.
- Pasha, M. A.; Jayashankara, V. P. *Bioorg. Med. Chem. Lett.* **2007**, *17*, 621.
- Pasha, M. A.; Jayashankara, V. P. *J. Indian Chem.* **2007**, *46B*, 1328.
- Madhusudana Reddy, M. B.; Pasha, M. A. *Chin. Chem. Lett.* **2010**, *21*, 1025.
- Pasha, M. A.; Jayashankara, V. P. *J. Pharm. Toxicol.* **2006**, *1*, 573.
- Pasha, M. A.; Aatika, N. *Synth. Commun.* **2010**, *40*, 2864.
- Aatika, N.; Pasha, M. A. *J. Saudi Chem. Soc.* **2011**, *15*, 55.
- General procedure for the synthesis of 3-methyl-4-aryl-4,5-dihydro-1H-pyrano[2,3-c]pyrazol-6-ones*: In a 50 mL round bottom flask, a mixture of hydrazine hydrate (2 mmol), ethyl acetoacetate (2 mmol), and Ba(OH)₂ (10 mol %) were taken, 10 mL water was added to the mixture and stirred in an oil bath at reflux for 15 min; aromatic aldehyde (2 mmol) and Meldrums acid (2 mmol) were then added and refluxing was continued for the remaining time (Table 4). The crude product thus separated was filtered, washed with water, the solid was dried, and subjected to silica gel column chromatography [silica gel G; 100–200 mesh, using EtOAc–hexane (1: 9) as eluent] to get the pure products. The structures of all the products were established by IR, ¹H NMR, ¹³C NMR spectral, and elemental analyses. *Spectral data*: 3-methyl-4-(3',4',5'-trimethoxyphenyl)-4,5-dihydro-1H-pyrano[2,3-c]pyrazol-6-one (**5a**): Yellow crystalline solid (93%, 0.590 g); mp 205 °C; IR (KBr) v: 3387 (br), 2968 (s), 1734 (vs), 1694 (s), 1622 (s), 1569 (vs), 1498 (s), 1448 (s), 1375 (s), 1236 (s), 1173 (vs), 1033 (s) cm⁻¹; ¹H NMR (400 MHz, DMSO-d₆): δ 2.88 (d, J = 8.0 Hz, 2H, CH₂), 3.43 (s, 3H, Me), 3.59 (s, 3H, OCH₃), 3.62 (s, 3H, OCH₃), 3.82 (s, 3H, OCH₃), 4.81 (t, J = 8.0 Hz, 1H, CH), 6.61 (s, 2H, Ph), 10.85 (s, 1H, NH); ¹³C NMR (100 MHz, DMSO-d₆): δ 169.8 (O=C=O), 154.6, 147.3, 132.8, 128.8, 114.8, 108.1 (all ArCs), 145.0 (C=C=N pyrazole), 136.8 (O=C=C pyrazole), 117.8 (C=C pyrazole), 60.1 (2 × OCH₃), 56.5 (OCH₃), 54.7 (CH₂), 29.7 (CH), 14.7 (CH₃); Anal. Calcd for C₁₆H₁₈N₂O₅: C, 60.37; H, 5.70; N, 8.80%. Found: C, 60.36; H, 5.70; N, 8.80%. 3-methyl-4-(4'-methoxyphenyl)-4,5-dihydro-1H-pyrano[2,3-c]pyrazol-6-one (**5b**): Orange amorphous solid (93%, 0.474 g); mp 157–160 °C; IR (KBr) v: 3401 (br), 2939 (s), 1734 (vs), 1704 (s), 1663 (s), 1504 (vs), 1447 (s), 1373 (vs), 1249 (vs), 1173 (s), 1034 (vs) cm⁻¹; ¹H NMR (400 MHz, DMSO-d₆): δ 3.19 (d, J = 8.4 Hz, 2H, CH₂), 3.34 (s, 3H, CH₃), 3.84 (s, 3H, OCH₃), 4.29 (t, J = 8.0 Hz, 1H, CH), 6.74 (m, 2H, Ph), 7.04–7.21 (dd, J₁ = 8.4 Hz, J₂ = 8.8 Hz, 2H, Ph), 11.10 (s, 1H, NH); ¹³C NMR (100 MHz, DMSO-d₆): δ 169.9 (O=C=O), 154.9, 142.9, 129.9, 127.0, 113.9 (all ArCs), 148.9 (C=C=N pyrazole), 138.8 (O=C=C pyrazole), 114.3 (C=C pyrazole), 55.8 (OCH₃), 52.0 (CH₂), 29.9 (CH), 13.6 (CH₃); Anal. Calcd for

- $C_{14}H_{14}N_2O_3$: C, 65.11; H, 5.46; N, 10.85%. Found: C, 65.12; H, 5.47; N, 10.84%. **3-methyl-4-(3'-methoxyphenyl)-4,5-dihydro-1H-pyrano[2,3-c]pyrazol-6-one (5c)**: White amorphous solid (93%, 0.474 g); mp 133–135 °C: IR (KBr) ν : 3356 (br), 2968 (s), 1733 (vs), 1704 (s), 1668 (s), 1596 (vs), 1508 (s), 1448 (s), 1375 (s), 1283 (vs), 1152 (vs), 1060 (s) cm^{-1} ; 1H NMR (400 MHz, DMSO- d_6): δ 3.09 (d, $J = 8.4$ Hz, 2H, CH₂), 3.54 (s, 3H, CH₃), 3.84 (s, 3H, OCH₃), 4.36 (t, $J = 8.0$ Hz, 1H, CH), 6.99 (s, 1H, Ph), 7.37–7.42 (m, 2H, Ph), 7.68 (d, $J = 5.6$ Hz, 1H, Ph), 11.13 (s, 1H, NH); ^{13}C NMR (100 MHz, DMSO- d_6): δ 169.7 (O=C=O), 153.3, 141.6, 129.7, 122.2, 115.7, 112.6 (all ArCs), 144.5 (C=C=N pyrazole), 135.7 (O=C=C pyrazole), 116.0 (C=C pyrazole), 55.7 (OCH₃), 52.7 (CH₂), 32.9 (CH), 12.6 (CH₃); Anal. Calcd for $C_{14}H_{14}N_2O_3$: C, 65.11; H, 5.46; N, 10.85%. Found: C, 65.12; H, 5.47; N, 10.84%. **3-methyl-4-(2'-chlorophenyl)-4,5-dihydro-1H-pyrano[2,3-c]pyrazol-6-one (5d)**: Pale orange crystalline solid (90%, 0.4672 g); mp 142–145 °C: IR (KBr) ν : 3394 (br), 2959 (s), 1734 (vs), 1704 (s), 1666 (vs), 1615 (s), 1508 (s), 1447 (s), 1380 (s), 1276 (vs), 1173 (vs), 1034 (s), 746 (vs) cm^{-1} ; 1H NMR (400 MHz, CDCl₃): δ 3.19 (d, $J = 8.4$ Hz, 2H, CH₂), 3.54 (s, 3H, CH₃), 4.69 (t, $J = 8$ Hz, 1H, CH), 7.06–7.25 (dd, $J_1 = 6.4$ Hz, $J_2 = 6.4$ Hz, 2H, Ph), 7.29 (d, $J = 7.6$ Hz, 1H, Ph), 7.59 (d, $J = 7.2$ Hz, 2H, Ph), 11.15 (s, 1H, NH); ^{13}C NMR (100 MHz, CDCl₃): δ 170.1 (O=C=O), 141.1, 129.9, 128.9, 126.9, 122.1, 117.1 (all ArCs), 149.9 (C=C=N pyrazole), 135.3 (O=C=C pyrazole), 114.1 (C=C pyrazole), 52.1 (CH₂), 32.1 (CH), 10.7 (CH₃); Anal. Calcd for $C_{13}H_{11}ClN_2O_2$: C, 59.44; H, 4.22; N, 10.66%. Found: C, 59.12; H, 4.22; N, 10.65%. **3-methyl-4-(2'-hydroxyphenyl)-4,5-dihydro-1H-pyrano[2,3-c]pyrazol-6-one (5e)**: Pale yellow amorphous solid (92%, 0.444 g); mp 235–237 °C: IR (KBr) ν : 3562 (br), 3362 (s), 2968 (s), 1733 (vs), 1694 (w), 1622 (s), 1508 (w), 1448 (s), 1375 (s), 1269 (s), 1163 (s), 1033 (s) cm^{-1} ; 1H NMR (400 MHz, DMSO- d_6): δ 2.99 (d, $J = 6.8$ Hz, 2H, CH₂), 3.29 (s, 1H, OH), 3.69 (s, 3H, CH₃), 4.29 (t, $J = 6.8$ Hz, 1H, CH), 6.94–6.98 (m, 2H, Ph), 7.39 (d, $J = 7.2$ Hz, 1H, Ph), 7.67 (d, $J = 8$ Hz, 2H, Ph) 11.09 (s, 1H, NH); ^{13}C NMR (100 MHz, DMSO- d_6): δ 169.7 (O=C=O), 159.5, 131.7, 120.5, 119.9, 117.4 (all ArCs), 144.7 (C=C=N pyrazole), 134.1 (O=C=), 114.7 (C=C pyrazole), 52.9 (CH₂), 32.9 (CH), 12.3 (CH₃); Anal. Calcd for $C_{13}H_{12}N_2O_3$: C, 63.93; H, 4.95; N, 11.47%. Found: C, 63.93; H, 4.95; N, 11.49%. **3-methyl-4-(4'-fluorophenyl)-4,5-dihydro-1H-pyrano[2,3-c]pyrazol-6-one (5f)**: Pale orange crystalline solid (91%, 0.442 g); mp 178–180 °C: IR (KBr) ν : 3398 (br), 2959 (s), 1734 (vs), 1704 (s), 1681 (s), 1544 (vs), 1504 (s), 1448 (s), 1354 (vs), 1223 (vs), 1152 (vs), 1034 (s) cm^{-1} ; 1H NMR (400 MHz, DMSO- d_6): δ 3.17 (d, $J = 8.0$ Hz, 2H, CH₂), 3.64 (s, 3H, CH₃), 4.29 (t, $J = 6.0$ Hz, 1H, CH), 7.00–7.03 (m, 2H, Ph), 7.11–7.28 (m, 2H, Ph), 11.23 (s, 1H, NH); ^{13}C NMR (100 MHz, DMSO- d_6): δ 171.8 (O=C=O), 152.5, 130.1, 130.0, 115.0, 105.0 (all ArCs), 160.1 (C=C=N pyrazole), 140.2 (O=C=C pyrazole), 115.2 (C=C pyrazole), 52.3 (CH₂), 32.9 (CH), 11.2 (CH₃); Anal. Calcd for $C_{13}H_{11}FN_2O_2$: C, 63.41; H, 4.50; N, 11.38%. Found: C, 63.39; H, 4.50; N, 11.39%. **3-methyl-4-(3'-nitrophenyl)-4,5-dihydro-1H-pyrano[2,3-c]pyrazol-6-one (5g)**: Pale orange crystalline solid (90%, 0.486 g); mp 212 °C: IR (KBr) ν : 3389 (br), 2968 (s), 1733 (vs), 1704 (s), 1668 (w), 1596 (s), 1538 (vs), 1448 (s), 1375 (w), 1348 (s), 1256 (s), 1173 (s), 1033 (s) cm^{-1} ; 1H NMR (400 MHz, DMSO- d_6): δ 3.10 (d, $J = 8.0$ Hz, 2H, CH₂), 3.50 (s, 3H, CH₃), 4.29 (t, $J = 8.0$ Hz, 1H, CH), 7.54 (d, $J = 8.0$ Hz, 1H, Ph), 7.79 (d, $J = 7.6$ Hz, 1H, Ph), 8.01 (d, $J = 8.0$ Hz, 1H, Ph), 8.18 (d, $J = 3.2$ Hz, 1H, Ph), 10.99 (s, 1H, NH); ^{13}C NMR (100 MHz, DMSO- d_6): δ 172.6 (O=C=O), 148.6, 147.7, 130.6, 122.7, 122.0, 102.3 (all ArCs), 152.8 (C=C=N pyrazole), 135.2 (O=C=C pyrazole), 112.3 (C=C pyrazole), 52.2 (CH₂), 36.6 (CH), 10.8 (CH₃); Anal. Calcd for $C_{13}H_{11}N_3O_4$: C, 57.14; H, 4.06; N, 15.38%. Found: C, 57.14; H, 4.06; N, 15.38%.
29. **Synthesis of 5-methyl-2,4-dihydro-pyrazol-3-one (I)**: A mixture of hydrazine hydrate (10 mmol), ethyl acetoacetate (10 mmol), and Ba(OH)₂ (10 mol %) were taken, 5 mL water was added to the mixture and stirred at 26 °C for 45 min (TLC), the crude solid thus separated was filtered, washed with water, and dried to get 5-methyl-2,4-dihydro-pyrazol-3-one in quantitative yield whose structure was established by 1H NMR and ^{13}C NMR spectral analysis. White solid, mp: 215–216 °C; 1H NMR (400 MHz, CDCl₃): δ 1.55 (s, 3H, CH₃), 3.14 (s, 2H, CH₂), 6.80 (s, 1H, NH); ^{13}C NMR (100 MHz, CDCl₃): δ 168.1 (NH-C=O), 153.9 (-C=N), 43.5 (CH₂), 20.6 (CH₃).
30. **Synthesis of 2,2-dimethyl-5-(3',4',5'-trimethoxybenzylidene)-[1,3]-dioxane-4,6-dione (II)**: A mixture of 3,4,5-trimethoxy benzaldehyde (2 mmol), Meldrum's acid (2 mmol), and Ba(OH)₂ (10 mol %) were taken in 5 mL water and stirred at reflux for 15 min (TLC), the solid thus separated was filtered, washed with water, and dried to get 2,2-dimethyl-5-(3',4',5'-trimethoxybenzylidene)-[1,3]-dioxane-4,6-dione in almost pure form. The structure was confirmed by 1H NMR and ^{13}C NMR spectral analysis. Pale green solid, mp: 118 °C; 1H NMR (400 MHz, CDCl₃): δ 1.76 (s, 6H, 2 × CH₃), 3.10 (s, 6H, OCH₃), 3.98 (s, 3H, OCH₃), 7.61 (s, 2H, ArH), 8.32 (s, 1H, =CH); ^{13}C NMR (100 MHz, CDCl₃): δ 167.8 (2 × O=C=O), 148.8 (H=C=C), 144.9, 131.4, 129.9, 111.1, 108.2, 106.2 (all ArCs), 117.8 (CO=C=C), 103.3 (>C-O), 57.8, 55.8, 53.5 (3 × OCH₃), 30.1, 27.8 (2 × CH₃).