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MAY 8, 2013

2013 AICHE CONTEST PROBLEM UNIVERSITY OF NEW HAMPSHIRE

COMPARISON OF BIOMASS TO BIO-OILS REACTOR SYSTEMS: DIRECT CONVERSION VS. COMPANION COAL GASIFICATION

ALEXANDRA EICHER

MAY 8, 2013

2013 AIChE DESIGN PROBLEM

UNIVERSITY OF NEW HAMPSHIRE

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SUMMARY

It is well known that the United States' dependence on crude-oil negatively affects its economy, safety, and environment. To alleviate these negative consequences, a more economical and environmentally-friendly source of fuel, such as biomass, should be explored. The conversion of biomass to bio-oils involves the pyrolysis of biomass at about 500°C, thus requiring a great deal of heat. This heat source could be the excess waste heat from a coal gasifier.

As such, this report specifies the design of an industrial plant that produces bio-oils from biomass by using the waste heat from a coal gasifier. It is designed to produce 2.24×10^8 kg/yr of bio-oil that can be sold at \$0.79/kg. This plant involves coal and biomass solids handling, a coal gasification reactor, a biomass pyrolysis reactor, and a series of separation units to remove waste products from the syngas and isolate the bio-oil. The syngas contains methane, hydrogen, and carbon monoxide and is sold as a by-product credit. The plant is expected to run on feeds of 1.5×10^{11} kg/yr of coal and 5.4 $\times 10^{8}$ kg/yr of raw biomass.

The coal gasification reactor was sized based on the heating duty of steam at 273000 kJ/s and the biomass pyrolysis reactor was sized based on a heating duty of 7026 kJ/s. The plant's operating factor (POF) is 0.9 at 7884 hrs/yr running 24 hrs/day and 328.5 days/yr. The total bare module equipment cost, including all pumps, heat exchangers, grinders, separators, absorber, and reactors is \$93 million.

The safety considerations include:

- Insulation on gasification product lines due to high temperatures.
- Spring loaded relief values and bursting discs on all vessels to prevent built up pressure and on storage vessels to prevent backflow.
- Pressure relief valves on the discharge side of pumps, compressors, and turbines .
- Double block and bleeds on feed lines and cool lines exposed to heat.
- Control system, with PID controls, on gasification and pyrolysis reactor to monitor and control their pressure and temperature by adjusting the cooling water to reactor jackets and the feed flow rates.
- Additional temperature and pressure gauges on the reactor for manual monitoring by the operator.

The key, innovative design features include:

- Waste heat, in the form of steam, from the gasifier used to heat the biomass.
- Bio-char from the pyrolysis of biomass recycled to the gasifier to decrease the amount of raw coal fed and, consequently, the cost of raw materials.
- Bio-gas from the pyrolysis of biomass used to dry the raw biomass to a moisture content of 10% as well as to blow the biomass up through the fluidized bed.

The total capital investment of the plant is \$173 million. The DCFRR and NRR are 12.59% and 20% respectively.

Given that selling price of bio-oil ((0.79/kg)) associated with this plant is about six times more expensive that the average cost of bio-oil ((0.13/kg)), it is not recommended that a Class – 1 Estimate be conducted. Before a Class – 1 Estimate can be conducted, the unnecessary costs associated with this proposed plant must be addressed and reduced. Specific attention must be paid to the following two heat exchangers, E-127 and E-129. Additionally, attention should be given to discover a cheaper source of industrial, liquid oxygen.

INRODUCTION

It is well known that the United States is heavily dependent on crude-oil; annually, Americans consumes 180 billion gallons of gasoline and diesel (Ackerson, 2012). Due to the shortage of crude-oil that can be found and harvested in America, our heavy dependence on crude-oil forces the United States to import vast quantities of it. This results in three major, negative consequences: 1) the United States' economy becomes tied to the price of crude-oil, which it controlled by foreign nations; 2) the United States must trade with nations that, traditionally, do not hold America in very high regard; and 3) the carbon dioxide emissions, produced from consuming crude-oil, accumulates in our atmosphere and greatly contributes to global warming (Ackerson, 2012).

To alleviate these negative consequences, a more economical and environmentallyfriendly source of fuel that does not necessitate importation should be explored. One abundant resource found in the United States is biomass. "It is estimated that there are at least 500 million dry tones of biomass available in the U.S. annually in the form of forest residues, mill residues, potential dedicated energy crops, urban wood wastes, and agricultural residues" (Ackerson, 2012). While using biomass as a source of fuel would be more economical and environmentallyfriendly than using crude-oil, the technical specifics of industrial conversion of biomass to biooil has yet to be fully explored.

The conversion of biomass to bio-oils involves the pyrolysis of biomass at about 500°C (Ackerson, 2012). This requires a great deal of heat. One source of heat is hot sand. Another is waste heat from a separate reaction, such as coal gasification to generate syngas. To yield syngas with an acceptable amount of CO as compared to CO_2 , a coal gasifier must be operated at high temperatures, thus producing a significant amount of waste heat and low thermal efficiencies. Utilizing the waste heat from coal gasification to drive the pyrolysis of biomass would not only increase the thermal efficiency of the coal gasifier, but also allow for a maximum conversion of energy to liquid fuels (Ackerson, 2012). As such, this report proposes an industrial plant that follows this "married" process for the conversion of biomass to bio-oils using the waste heat from coal gasification.

The primary reaction for this plant is the pyrolysis of biomass. There are three different pyrolysis processes: slow pyrolysis, fast pyrolysis, and flash pyrolysis (Goteti, 2010). Fast pyrolysis produces the most bio-liquid of all three processes, and, as such, is the most common thermal process for converting biomass to bio-oil (Ackerson, 2012). Fast pyrolysis yields 75% bio-liquid, 12% bio-char, and 13% bio-gas and is characterized by moderate temperatures of about 500°C and short residence times of less than 2 seconds (Goteti, 2010). To ensure that fast pyrolysis occurs, there must be a very high heat flux to the biomass (i.e. a high heat transfer rate to the biomass) and a very short residence time (i.e. the heat transfer rate must occur in a very short amount of time) (Ringer, 2006). For these reasons, the pyrolysis reactor in the proposed married process is a fluidized bed reactor. This ensures direct contact with the biomass, and thus, a high heat transfer rate to the biomass.

Immediately after the pyrolysis reaction, the resulting gaseous mixture must be quenched to prevent the compounds from further cracking into permanent gases or polymerizing into char (Ringer, 2006). In this process, quenching is accomplished through fractional condensation using a series of shell and tube heat exchangers. The bio-oil that condenses is a mixture of more than 100 chemical species that have a wide range of molecular weights (Ackerson, 2012). "A typical

elemental composition shows 44-47% carbon, 6-7% hydrogen, 46-48% oxygen, and 0-0.2% nitrogen" (Ackerson, 2012).

The secondary reaction for this plant is the pyrolysis of coal. Since the United States has over 273.6 billion short tons of coal reserves (the largest in the world), there is no shortage of raw material for this married process (Ackerson, 2012). This proposed process follows the Texaco process because it is the most widely used and the most economically preferred (Ackerson, 2012). Additionally, "the low capital investment for this process makes it ideal to integrate it with a coal and biomass production" (Ackerson, 2012). For the Texaco process, coal is feed as a slurry into the gasifier. For combustion to occur properly, this slurry must have a high enough coal concentration so that the reactor operates at the appropriate temperature but a high enough water concentration so that the slurry's viscosity is low enough to ensure an ease of transport. Since the major hazard for the pyrolysis of coal is the high gasification temperature, the coal concentration in the slurry was set around 60% to ensure a low flow rate of water entering and vaporizing in the gasifier. The pyrolysis of coal ultimately produces synthesis gas, or syngas, which is a fuel gas mixture consisting hydrogen, carbon monoxide, and methane.

The following report gives a detailed description of a plant that uses the married process to produce 2.24×10^8 kg/yr of bio-oil. There are three main, novel features of this plant: 1) waste heat, in the form of steam, from the gasifier is used to heat the biomass; 2) bio-char from the pyrolysis of biomass is recycled to the gasifier to decrease the amount of raw coal fed and, consequently, the cost of raw materials; and 3) the bio-gas from the pyrolysis of biomass is used to dry the raw biomass to a moisture content of 10% as well as to blow the biomass up through the fluidized bed. The major objective of this married process design is to determine its economic feasibility, i.e. the selling price of crude bio-oil: \$0.79/kg.

CONCLUSIONS

To produce bio-oil from biomass, safety, environmental, process, and economic issues must all be addressed.

First, to ensure the safety of the operators and the surrounding area, the plant was designed with numerous safety features. The key safety features include the use of safety relief valves, temperature and pressure gauges for manual monitoring, and control systems for automatic vessel and reactor monitoring. Safety relief valves were placed on all vessels to inhibit pressure build up. These valves were also placed on storage vessels to prevent back flow. Safety valves were also included on all pumps, compressors, and turbines on the discharge side to release pressure in case of excessive pressure caught between the compressor and valve. A double block and bleed system, which involves two valves with a relief valve in between, is in place on all feed lines and lines exposed to heat. The double block and bleed prevents pressure build up in the line. A control system, in place on the reactor, monitors and adjusts its temperature and pressure. The supply of cooling water to the reactor jacket is varied by the control system to control the reactor temperature. Feed flow is varied by the control system to control the reactor pressure and temperature gauges are in place on all units for manual monitoring by the operator. Manual monitoring is an extra level of safety in case of control system malfunction.

A key environmental consideration tied to this project is reforestation. The only way to avoid carbon dioxide accumulation when burning bio-oils is to ensure the growth of more biomass. As biomass grows, it absorbs carbon dioxide from the atmosphere. If fact, if the amount biomass grown equaled the amount of bio-oil burnt, the whole process would be carbon-neutral.

This plant design features a few key, innovative systems. First, steam from the gasifier is used to heat the biomass. This takes advantage of excess heat, in the form of steam, from the coal gasifier. Not only does this increase the thermal efficiency of the gasifier but it also provides a convenient method to heat the biomass. Second, bio-char from the pyrolysis of biomass is recycled to the gasifier. This decreases the amount of raw coal needed to be fed to the coal gasifier, thus, decreasing the cost of raw materials. Third, the bio-gas from the pyrolysis of biomass is used to dry the raw biomass to a moisture content of 10% as well as to blow the biomass up through the fluidized bed. This decreases waste steams leaving the plant.

There are a few other noteworthy features. The MEA stream loop has no inlet and outlet streams because all of the MEA is assumed to be recycled though the process. It is regenerated on a yearly basis. This keeps the costs of solvents and, thus, manufacturing costs down. It also minimizes MEA waste. Also to minimize consumption and cost, the process water used by the Venturi scrubber to remove slag from the product stream is recycled with only a minimum of water being released and wasted. Before the product stream is cleaned of its waste gases by the MEA absorber the temperature of the stream is lowered though use of scrubber. This scrubber was used instead of a heat exchanger to take advantage of direct contact with water. This allows for a more efficient cooling of product gases, when compared to that of a heat exchanger, and, therefore reduces cost of equipment and electricity consumption.

Given a typical cost of bio-oil of \$44 per barrel equivalent and a higher heating value for bio-oil of 7,554 Btu/lb, the average cost of bio-oil is \$0.13/kg. The selling price of bio-oils derived from this study is \$0.79/kg. This price is about six times more than the average cost of bio-oil. Therefore, it does not seem economically sound to produce bio-oil from biomass given the proposed plant.

RECCOMENDATIONS

The recommendations for this Class – 4 Estimate are the following:

- 1. This plant should not be constructed as specified.
- 2. Means of cutting cost need to be considered.
- 3. Specific attention should be paid to E-127, the heat exchanger designed to cool down the scrubber recycle water, and to E-129, the heat exchanger designed to cool down the recycled MEA to the absorber while heating up the saturated MEA to the stripper. Both of these heat exchangers are designed with higher-than-anticipated surface areas, which necessitate the use of multiple heat exchangers. This drastically increases the Total Bare Module Cost and, consequently, all other costs associated with it.
- 4. Another large source of expense is the cost of raw materials, specifically, the cost of liquid oxygen. Attention should be given to discover a cheaper source of industrial, liquid oxygen.
- 5. Given that selling price of bio-oil is about six times more expensive that the average cost of bio-oil, it is not recommended that a Class -1 Estimate be conducted until the unnecessary costs associated with this proposed plant are addressed.

PROJECT PREMISES

1. Start and Completion Dates

- a. Effective Start Date: May 18, 2014
- b. Mechanical Completion Date: May 18, 2017

2. Feed Specifications

- a. Coal
 - i. Proximate Analysis (wt%): 8.9% Moisture, 10.7% Ash, 32.5% Volatiles, 47.9% Fixed Carbon (Zheng, 2005)
 - ii. Ultimate Analysis (wt%): 69.71% Carbon, 4.8% Hydrogen, 0.82% H/C, 1.4% Nitrogen, 3.64% Sulfur, 7.83% Oxygen, 11.8% Ash (Zheng, 2005)
 - iii. Higher Heating Value (HHV): 29.4 MJ/kg (Zheng, 2005)
 - iv. Heat Capacity: 31400 kJ/kg
 - v. Temperature (°C): 25
 - vi. Pressure (bara): 1
 - vii. Mass Flow Rate (kg/s): 52.04
- b. Water
 - i. Concentration in Slurry (wt%): 40%
 - ii. Heat Capacity (kJ/kg/K): 4.181 (Goteti, 2010)
 - iii. Heat of Vaporization at 75 °C (kJ/kg): 2322.87 (Goteti, 2010)
 - iv. Temperature (°C): 25
 - v. Pressure (bara): 16
- c. Liquid Oxygen
 - i. Mass flow rate: 0.97% of coal mass flow rate (Zheng, 2005)
 - ii. Temperature (°C): -183
 - iii. Pressure (bara): 16
- d. Biomass
 - i. Heat Capacity at 298 K (assuming the heat capacity in the given temperature range does not change significantly): 1.2 KJ/kg/K (Goteti, 2010)
 - ii. Temperature (°C): 25
 - iii. Pressure (bara): 1

Product Specifications and Quality Considerations

- e. Bio-oil
 - i. Mean Specific Heat Capacity (kJ/kg/K): 2.435 (Goteti, 2010)
 - ii. Mean Heat of Vaporizaiton (kJ/kg): 609.9 (Goteti, 2010)
 - iii. Temperature (°C): 352.2
 - iv. Pressure (bara): 12.5

3. Manufacturing Costs

- a. Capital Expenses
 - i. Fixed Capital: \$144,000,000/yr.
 - ii. Working Capital: \$28,800,000/yr.
- b. Manufacturing Expenses
 - i. Direct
 - 1. Raw Materials: \$697,000,000/yr.

- 2. By-product Credits: \$841,000,00/yr.0
- 3. Solvents: \$49,400/yr.
- 4. Operating Labor: \$1,260,000/yr.
- 5. Supervisory and Clerical Labor: \$251,000/yr.
- 6. Utilities
 - a. Steam: \$301,000/yr.
 - b. Electricity: \$12,30,000/yr.
 - c. Cooling Water: \$679,000/yr.
 - d. Waste Disposal: \$4,210,000/yr.
- 7. Maintenance and Repairs: \$14,400,000/yr.
- 8. Operating Supplies: \$2,880,000/yr.
- 9. Laboratory Charges: \$252,000/yr.
- 10. Patents and Royalties: \$8,440,000/yr.
- ii. Indirect
 - 1. Overhead, Packaging, and Storage: \$11,100,000/yr.
 - 2. Local Taxes: \$4,310,000/yr.
 - 3. Insurance: \$2,880,000/yr.
- c. General Expenses
 - i. Administrative Costs: \$2,780,000/yr.
 - ii. Distribution and Selling: \$14,100,000/yr.
 - iii. Research and Development: \$7,040,000/yr.

4. Product Selling Price

a. Bio-oil: \$0.79/kg

5. Economic Parameters

- a. Project Life: 10 years
- b. Depreciation Schedule: \$14,400,000/yr.
- c. Total Expenses: \$141,000,000/yr.
- d. Net Annual Profit: \$35,400,000/yr.
- e. Income Taxes: \$12,400,000/yr.
- f. Net Annual Profit After Taxes: \$23,000,000/yr.
- g. Revenue from Sales: \$176,000,000/yr.
- h. Net Rate of Return (NRR): 20%
- i. Payback Period (PBP): 4.25 years
- j. Discounted Break-even Point (DBEP): 11.5 years
- k. Discounted Cash Flow Rate of Return (DCFRR): 12.59%

6. Environmental Requirements

- a. De Minimis Emission Levels
 - i. Carbon Monoxide: 100 tons/year (De Minimis Levels, 2013)

7. Processing Limitations

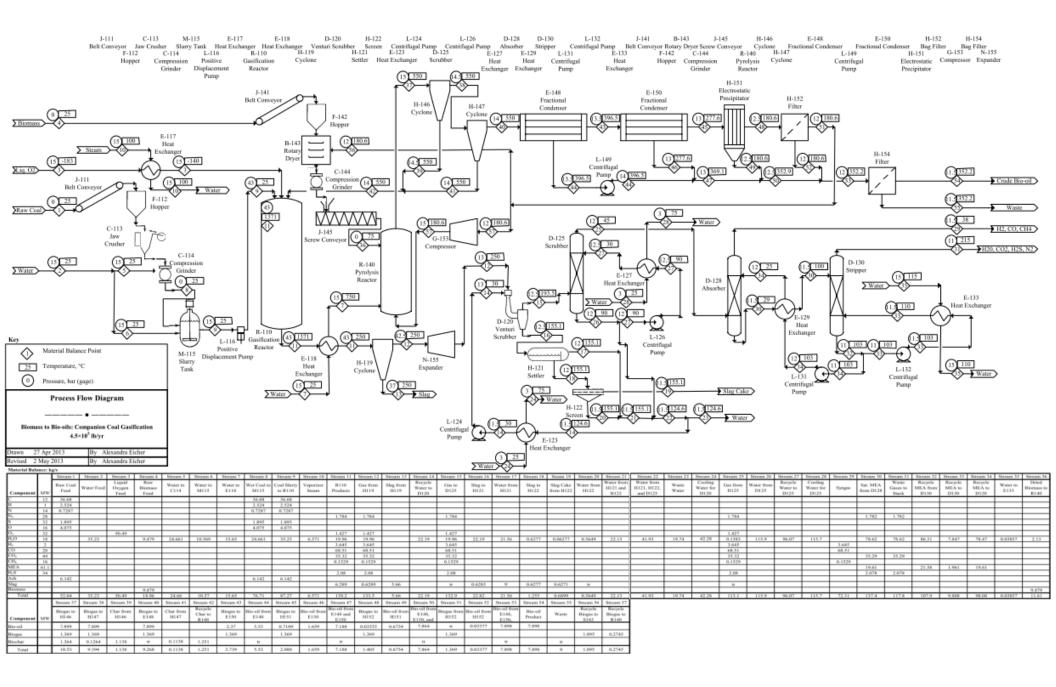
a. The mass flow rate of the biomass rate is tied to the steam mass flow rate from the heat exchanger in the coal gasifier. This limitation is designed to take full advantage

of all of the waste heat from the coal gasifier. Due to this, the mass flow rate of the resulting bio-oil is also limited.

b. The mass flow rate of the feed water is limited to 40% of the slurry mass flow rate to ensure that the temperature in the coal gasifier does not exceed optimal operating temperatures.

PROCESS FLOW DIAGRAM (PFD)

The PFD is attached on the next page. It includes all process equipment items, process streams, and utility streams. Each process equipment item is named and numbered. Each process and utility stream is numbered and labeled with its operating temperature and pressure. The Steam Attributes table is included on the PFD. This table lists each stream number and name, the mass flow rate of each component in each stream, and the total mass flow rate of each stream.



PROCESS DESCRIPTION

Raw coal enters the plant and travels through a series of solids handling equipment to prepare the coal for the gasification reactor. The coal is conveyed via a belt conveyor to a hopper where it is stored until it is released into a jaw crusher. After the jaw crusher, the coal is fed into a rolling compression grinder where it is wet-milled to 100 µm. Once the desired particle size is reached, the coal is mixed with water to form a slurry with a coal concentration of 60%. About 70% of the feed water is used to wet-mill the coal and the rest of the water is added to the slurry tank to mix the coal. From there, the slurry is charged to the reactor using a positive displacement pump. The slurry, oxygen, and recycled bio-char are introduced into the gasifier through the feed-injector at the top of the reactor (Higman, 2008). The reactor is a top-fired, coal-water slurry feed, slagging, entrained-flow gasifier (Higman, 2008). The AFT of the reactor using a heat exchanger that converts cooling water to steam. This steam is then used to heat the biomass pyrolysis reactor.

The gas from the gasification reactor then goes through a series of separation processes to isolate methane, carbon monoxide, and hydrogen. First, the gasification products travel through a cyclone to remove 90% of the slag. Then, the product steam is sent through an expander to reduce its pressure from 43.5 bara to 14 bara. From theer, the product stream goes through a Venturi scrubber. The Venturi scrubber removes 99.9% of the remaining slag by cycling water through the gas. After the venture scrubber, only a negligible amount of slag is left in the product stream. The water and slag from the Venturi scrubber flows through a settler and filter to remove the slag from the waste water. The slag is pressed into an slag cake and some of the waste water is recycled back through the process. The product stream with trace slag is sent through a scrubber to further cool the stream to 45°C. Most of the water from the scrubber is recycled; some of it is purged and added to the Venturi scrubber waste water recycle stream. The product stream then goes through an absorber to remove hydrogen sulfide, carbon dioxide, and nitrogen. The absorber uses an aqueous solution of monoethanolamine (MEA) to absorb these waste gases. The amount of MEA required to remove the waste gasses is determined by equations 1 through 5.

$$2C_2H_7NO + H_2S \rightleftharpoons (C_2H_7NO)_2S \tag{1}$$

$$(C_2H_7NO)_2S + H_2S \rightleftharpoons (2C_2H_8NO)HS \tag{2}$$

$$2C_2H_7NO + CO_2 + H_2O \rightleftharpoons (C_2H_7NO)_2CO_2$$
(3)

$$(C_2 H_7 N O)_2 C O_2 + C O_2 + H_2 O \rightleftharpoons (2C_2 H_8 N O) H C O_2$$
(4)

$$2C_2H_7NO + CO_2 \rightleftharpoons C_2H_7NO_3C_2H_8NO \tag{5}$$

The product stream leaves the absorber, ready to be sent to Fisher Tropsch processing. The waste gas and MEA solution is sent through a stripper to separate the waste gases from the aqueous MEA. The MEA is then recycled back through the absorber. After the stripper, the waste gases do not need to be process further and are sent through flue-gas stack.

Raw biomass enters the plant in a similar fashion as the raw coal. It is conveyed via a belt conveyor to a hopper where it is stored until it is released into a rotary dryer. Recycled biogas is used to heat the raw biomass and dry it until the biomass has a moisture content of 10 wt%. From there, the dried biomass is fed into a rolling compression grinder where it is ground to a particle size of 100 μ m. Then, a screw conveyor conveys the biomass to the biomass pyrolysis reactor where the steam from the coal gasifier's heat exchanger rapidly heats the biomass. Recycled bio-gas blows the biomass up through the fluidized bed.

After fast pyrolysis, bio-char is removed from the bio-gas and bio-vapors via two cyclones in series. Each cyclone is 90% efficient. The bio-char that is removed by the cyclones is sent back to the coal gasifier. The bio-gas and bio-vapors are then sent through a series of shell-and-tube condensers to condenser the bio-vapors into bio-oils. Each condenser is 70% efficient. The rest of the bio-gas is sent through an electrostatic precipitator to remove any bio-vapor aerosols and a filter to remove any final bio-oil particulates. Both the electrostatic precipitator and the filter are 95% efficient. The bio-gas is recycled to dry the raw biomass to a moisture content of 10% and blow the dried biomass through the fluidized bed reactor. Finally, the bio-oils are sent through one final filter to remove any bio-char particulates. The bio-oils are ready to be hydrotreated.

MAJOR EQUIPMENT

Mixers: The slurry tank (M-115) is an axial turbine agitator made of carbon steel. Its purpose is to mix the wet, ground coal with additional water to form a coal-water slurry that has a coal concentration of about 60 wt%. The turbine impeller is mounted on a shaft that is powered by a motor. Axial turbine impellers are characterized by blades that are pitched at about 45° (Ulrich, 2004). This angle forces the liquid to flow downward, parallel to the shaft, across the bottom of the vessel, and then upward along the wall of the tank (Ulrich, 2004). Given a residence time of 2 hours, the slurry tank has a volume of 766.9 m³ and a corresponding power consumption of 203.1 kW. At these conditions, the slurry tank mixers the slurry with vigorous agitation of 0.2649 kW/m³. The slurry tank mixes 52.04 kg/s of coal and 35.23 kg/s of water into a 60 wt% coal slurry.

Pumps: There are five radial centrifugal pumps (L-124, L-126, L-131, L-132, and L-149), all made of cast iron, in this process design. The purpose of these pumps is to overcome pressure drops over the length of long, narrow pipes and to pump fluids, such as water and MEA, through recycle streams. The shaft power of each pump, respectfully, is 3.196 kW, 5.561 kW, 11.51 kW, 0.5767 kW, and 0.2706 kW. The volumetric flow rates though each pump, respectfully, is 2219 cm³/s, 96070 cm³/s, 97850 cm³/s, 9785 cm³/s, and 4601 cm³/s. There is one reciprocating, positive displacement pump (L-116) in this process. It is also made of cast iron. The purpose of this pump is to pump the viscous coal slurry to the coal gasifier. This is accomplished via a piston system that physically pushes the slurry through the pipe to the coal gasifier. This pump has a shaft power of 350.9 kW and a volumetric flow rate of 0.1065 m³/s flows through this pump.

Heat Exchangers: There are eight shell and tube heat exchangers (E-117, E-118, E-123, E-127, E-129, E-133, E-148, and E-150), all made of carbon steel, in this process. They are used to either cool down or heat up the various process streams. This is accomplished by

flow either cooling water (to cool down the stream) or steam (to heat up the stream). E-118 was designed as a fixed tube sheet and U- tube heat exchanger with a total heat transfer area of 1079 m². E-118 cools down the 139.2 kg/s of gas from the coal gasifier from 1371°C to 250°C using 14.65 kg/s of cooling water at 25°C. The rest of the heat exchangers were designed as double pipe heat exchangers. E-117 has a total heat transfer area of 23.88 m^2 and heats up 50.49 kg/s of liquid oxygen from -183°C to -140°C using 6.571 kg/s of steam at 100°C. E-123 has a total heat transfer area of 477.7 m² and cools down 22.19 kg/s of water from 124.6°C to 30°C using 42.28 kg/s of cooling water at 25°C. E-127 has a total heat transfer area of 2791 m² and cools down 96.07 kg/s of water from 90°C to 30°C using 115.7 kg/s of cooling water at 25°C. E-129 has a total heat transfer area of 6542 m² and heats up 137.4 kg/s of saturated MEA from 29°C to 100°C by cooling down 98.08 of recycled MEA from 103°C to 25°C. E-133 has a total heat transfer area of 46.82 m² and heats up 9.808 kg/s of recycled MEA from 103°C to 110°C using 38.58 g/s of water at 115°C. E-148 has a total heat transfer area of 12.15 m² and condenses 5.53 kg/s of bio-oil from 550°C to 396.5°C. E-150 has a total heat transfer area of 6.386 m^2 and condenses 1.659 kg/s of bio-oil from 396.5°C to 277.6°C. Both condensers are 70% efficient.

Gas Movers and Compressors: The axial compressor (G-153) is designed to compress the recycled bio-gas to the coal gasifier from 13 bara to 16 bara and is made of carbon steel. This is accomplished through a series of rotating blades that increase the velocity of the bio-gas and, therefore, its pressure. This compressor has a fluid power of 13.27 W and compresses 274.5 g/s of bio-gas.

Gas-Solid Contacting Equipment: The rotary dryer (B-143) is a direct contact dryer, made of carbon steel, that is designed to dry the raw biomass to a moisture content of 10%. This is accomplished by feeding the biomass into the elevated side of the rotary dryer and allowing it to slowly migrate to the discharge end while hot bio-gas flows, counter-currently, up through the dryer (Ulrich, 2004). This dryer has an internal volume of 4.731 m³ and heats 18.96 kg/s of raw biomass from 25°C to 75°C using 1.095 kg/s of bio-gas at 180.6°C.

Drivers and Power Recovery Machines: The radial gas expander (N-155) is designed to expand the gas from the coal gasifier from 43.5 bara to 14 bara and is made of carbon steel. This is accomplished by decreasing the velocity of the gas in a manner opposite of the axial compressor. This expander has a shaft power of 14,000 kW and expands 133.5 kg/s of gas.

Separators: There are three cyclones (H-119, H-146, and H-147), all made of carbon steel, that are used in this process. These cyclones use centrifugal forces to remove particulates (slag for H-119 and bio-char for H-146 and H-147) from gaseous streams (gas for H-119 and bio-gas for H-146 and H-147) and are 90% efficient. The volumetric flow rates through each cyclone, respectfully, are $6.244 \text{ m}^3/\text{s}$, $0.3285 \text{ m}^3/\text{s}$, and $0.3389 \text{ m}^3/\text{s}$. The settler (H-121) is a single-compartment drum made of carbon steel. It is designed to remove slag from water. It has a filter area of 13.17 m^2 and 22.82 kg/s of water and slag flow through it. The vibratory screen (H-122) is also made of carbon steel and it designed to remove slag from water. Its power consumption is 23.85 kW and 1.255 kg/s of water and slag flows through it. There are two gas bag filters (H-152 and H-154), both made of carbon steel, that are designed to remove either bio-oils from bio-gas (H-152) or bio-char from bio-oil (H-154). The

volumetric flow rate of bio-oils and bio-gas through H-152 is 0.3821 m^3 /s the volumetric flow rate of bio-char and bio-oil through H-154 is 6571 cm³/s. They are both 95% efficient. The electrostatic precipitator (H-151) uses an electric field to apply a charge to the aerosol bio-vapors so that they can be collected and removed from the bio-gas (Ulrich, 2004). The volumetric flow rate of bio-vapors and bio-gas through the electrostatic precipitator is 0.3689 m³/s. It is made of carbon steel. It is 95% efficient.

Crushers, Mills, and Grinders: The jaw crusher (C-113) is designed to crush the raw coal into a particle size of about 1 cm. It is ideal for hard solids and abrasive materials (Ulrich, 2004). The power consumption of the jaw crusher is 131.7 kW and the mass flow rate of raw coal through it is 52.04 kg/s. There are two rolling compression grinders (C-114 and C-144) that are designed to crush either coal (C-114) or biomass (C-144) to a particle size of about 100 μ m. Rolling compression grinders are ideal for soft materials and wet grinding (Ulrich, 2004). The power consumption of C-114 is 238.7 kW and the mass flow rate of coal through it is 52.04 kg/s. The power consumption of C-144 is 52.34 kW and the mass flow rate of soft materials is 11.61 kg/s.

Conveyors: There are two belt conveyors (J-111 and J-141) that are designed to convey raw coal (J-111) or raw biomass (J-141) to a storage hopper. Both conveyors have a height of 25 meters, a width of one meter, a conveying distance of 50 meters, and an incline angle of 45°. J-111 conveys 52.04 kg/s of coal and J-141 conveys 18.96 kg/s of raw biomass. The screw conveyor (J-145) is designed to convey dried biomass to the biomass pyrolysis reactor. It has a conveying distance of 25 meters and a width of 0.325 meters. It conveys 11.61 kg/s of dried biomass.

Storage Vessels: There are two atmospheric bins (F-112 and F-142) that are designed to store coal (F-112) or biomass (F-142). Both are made of carbon steel. Given a residence time of eight hours, F-112 has a volume of 2053 m^3 and F-142 has a volume of 1021 m^3 .

Reactors: The gasification reactor (R-110) is a top-fired, coal-water slurry feed, slagging, entrained-flow gasifier. Essentially, it acts as a pyrolysis furnace with a large operating temperature to allow for the conversion of coal into syngas. This process needs five gasification reactors to accommodate the extraordinary heating duty needed to cool the syngas from 1371°C to 250°C. Each unit has a heating duty of 55,000 kJ/s and the cooling pipes are made of stainless steel. They all operate at a pressure of 44 bara and 1371°C. The biomass pyrolysis reactor (R-140) is a fluidized bed reactor. It also acts as a pyrolysis furnace. Steam from R-110 is used to heat the dried biomass within the pyrolysis reactor. It has a heating duty of 7026 kJ/s and is made of stainless steel. It operates at a pressure of 16 bara and 550°C.

Process Vessels: The Venturi scrubber (D-120) is used to remove solid particles from largevolume gas streams via water. The advantage of Venturi scrubbers is that there is a minimal pressure drop across the unit. For this process, the Venturi scrubber is designed to remove slag from the product gas stream. It is made of carbon steel and the volumetric flow rate of the gas through it is 22.21 m³/s. The general scrubber (D-125) is designed to further cool down the gas stream through direct contact with water. Given a residence time of ten seconds, the column has a diameter of 3.589 meters and a height of 17.95 meters. The scrubber is made of carbon steel and the mass flow rate of gas through it is 132.9 kg/s. It is assumed to be 99.9% efficient. The absorber (D-128) is designed to remove unwanted components from a gas stream. This is accomplished by flowing a solvent, such as MEA, across the gas stream and allowing certain components of the gas stream to diffusive into the liquid stream. Using solubility properties, it is possible to determine the composition of the liquid and vapor phases as well as the number of theoretical trays needed to achieve a certain separation. The absorber in this process has a diameter of 3.001 meters and a height of 15.01 meters. It is made of carbon steel and the total mass flow rate through it is 211.2 kg/s. The stripper (D-130) is designed to strip away waste gases from a solvent, such as MEA, so that the solvent can be re-used. Stripping is a physical separation process where one or more components are removed from liquid streams by vapor streams. In this process, the stripper has a diameter of 3.086 meters and a height of 15.43 meters. It is made of carbon steel and the total mass flow rate through it is 99% efficient.

PROCESS CONTROL STRATEGY

A control system has been added to the gasification reactor to monitor the pressure and temperature within the reactor. If the temperature is too high, the coal slurry stream can be adjusted to prevent further reaction and, therefore, higher temperatures. The system can also be designed to supply cooling water to the reactor jackets if the reaction temperature exceeds the recommended operating temperature. If the pressure is too high, the liquid oxygen feed can be adjusted to lower the pressure. Temperature and pressure gauges have been installed on the reactor so operators can monitor them and take appropriate action if the temperature or pressure of the reactor exceeds the bounds of the design parameters.

The reactor has been fitted with the following controls, as shown in Figure 1:

- 1.) Pneumatic valves (K-156, and K-157) have been installed to control the flow rate of the liquid oxygen feed and the coal slurry stream to the gasification reactor (G-110). The flow rate of the liquid oxygen feed adjusts the pressure within the reactor and the flow rate of the coal slurry stream controls the temperature of the reactor.
- 2.) PID controllers (K-158, K-159, K-160, and K-161) have been installed to monitor the reactor's temperature and pressure as well as to control the pneumatic valves. Temperature and pressure readings from sensors located within the reactor are converted into an electrical signal that travels to the respective PID controller. Each controller then manipulates the air pressure to the pneumatic valve, thus opening or closing the pneumatic valve. This affects the flow rate of each stream and, thereby, adjusts either the temperature or the pressure within the reactor.
- 3.) A similar process can be applied to other parts of the process. For example, to lower the temperature of the reactor by adjusting the flow rate of cooling water entering the reactor jacket.

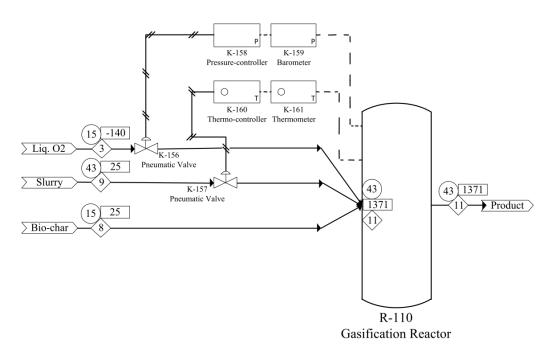


Figure 1: A diagram of possible controls surrounding the Gasification Reactor (R-110).

SAFETY

This safety review consists of two sections: 1.) Chemical Hazards & Safety Measures and 2.) Process Hazards & Safety Measures. The first section discusses the dangers of the chemicals used in this process and the safety measures that have been taken to prevent accidents from occurring. For specific and detailed safety precautions and health hazards of each chemical please referrer to the referenced MSDS sheets. The second section discusses the dangers inherent in this process and the associated safety measures that have been taken to prevent accidents from occurring.

1.) Chemical Hazards and Safety Measures

a.) Hazards

- 1.) Liquid Oxygen:
 - i. Oxidizer.
 - ii. Contact with combustible material may cause fire.
 - iii. Extremely cold liquid and gas under pressure.
 - iv. May cause severe frostbite.
 - v. Refer to Reference No. 9.
- 2.) Hydrogen Sulfide:
 - i. Flammable gas.
 - ii. May cause flash fire.
 - iii. May be fatal if inhaled.

- iv. May cause eye and skin irritation.
- v. Keep away from heat, sparks, and flame.
- vi. Do not breathe gas and avoid contact with eyes, skin, and clothing.
- vii. Use only with adequate ventilation.
- viii. Refer to Reference No. 6.
- 3.) Methane:
 - i. Extremely flammable.
 - ii. May cause flash fire.
 - iii. Can cause rapid suffocation.
 - iv. May cause severe frostbite.
 - v. Refer to Reference No. 7.
- 4.) Monoethanolamine (MEA):
 - i. Causes eye and skin burns.
 - ii. Harmful if inhaled or absorbed through skin.
 - iii. Harmful if swallowed.
 - iv. If exposure occurs, evacuate area and keep upwind of spill.
 - v. Refer to Reference No. 8.
- 5.) Carbon Monoxide:
 - i. Flammable gas.
 - ii. May cause flash fire.
 - iii. May be fatal if inhaled.
 - iv. Keep away from heat, sparks, and flame.
 - v. Avoid breathing gas.
 - vi. Use only with adequate ventilation.
 - vii. Refer to Reference No. 5.

b.) Safety Measures

To address the previously mentioned hazards the following safety measures should be followed:

- 1.) All Material Data and Safety Sheets should be printed out and kept within easy access.
- 2.) Safety glasses should be worn at all times.
- 3.) All operators should know the location of all eye-wash stations, emergency showers, and fire extinguishers.

2.) Process Hazards and Safety Measures

a.) Hazards

- 1.) Exothermic chemical reaction.
- 2.) Consequences of deviation from designed operation:
 - i. Loss of water to heat exchanger. May result in a runaway reaction.
 - ii. Reduction in reactor temperature. May result in a runaway reaction.
 - iii. Piping leaks. May results in contamination, fire, or explosion.

- iv. Spring loaded or pressure relief valves leak. May result in safety features not operating under specification.
- v. Power failure to cooling pumps. May result in a runaway reaction.

b.) Safety Measures

To address the previously mentioned hazards the following safety measures have been added to this design:

1.) A spring loaded relief valve and a bursting disc have been installed on both reactors to allow for a release of excess pressure in the event of a runaway reaction. Figure 2 shows a spring loaded relief valve and a bursting disc fitted to the gasification reactor (R-110).

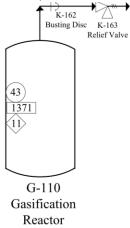


Figure 2: A spring loaded relief valve (K-163) and a bursting disc (K-162) for the Gasification Reactor (G-110).

- 2.) Cooling water is supplied to the reactor jackets to prevent the reactors from operating above the recommended temperature.
- 3.) All storage vessels have been fitted with pressure relief valves to protect against back flow. For example, the stream line from the liquid oxygen storage vessel must have a pressure relief valve in order to prevent any back flow from the Gasification Reactor (G-110). A representation of this system is shown in Fig. 3.

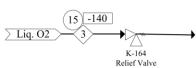


Figure 3: A pressure relief valve (K-164) after the liquid oxygen storage vessel.

4.) Pressure relief valves have also been installed on the discharge side of all pumps, compressors, and expanders. Figure 4 shows an example of a compressor with a pressure relief valve on its discharge side.

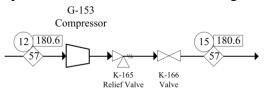


Figure 4: A pressure relief valve (K-165) on the discharge side of a compressor (G-112).

5.) Double block and bleeds have been installed to blocked in sections of cool liquid-liquid lines that are exposed to heat as well as feed lines. Figure 5 shows an example of a double block and bleed that consists of two valves and a relief valve.

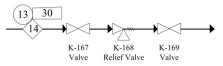


Figure 5: An example of a double-block-and-bleed system.

- 6.) Line and equipment checks should be performed by the operators periodically.
- 7.) Generators should be installed to ensure that necessary processes, like cooling water, still operate during a power failure.
- 8.) Sensors should be installed on the reactors to alert the operators if the temperature or pressure of the reactors exceeds design specifications.

ENVIRONMENTAL

The main environmental advantage of this plant is the reduction of carbon dioxide emissions. By burning bio-oils derived from biomass, instead of burning crude-oil or even coal, less carbon dioxide gas will accumulate in the atmosphere. However, this is only true if the biomass source is replenished. As biomass grows, it absorbs as much carbon dioxide from the atmosphere as it emits when it is burned. If the biomass source is not replenished, then burning bio-oils could have an even greater effect on the accumulation of carbon dioxide in the atmosphere then burning either crude-oil or coal. As such, it is imperative that, if industrial biooil plants become a more economically attractive option than industrial syngas plants, a concerted effort for reforestation is established and maintained.

UTILITY SUMMARY

Table 1 itemizes each utility by user. The table gives (1) the user, (2) the specific utility, (3) mass flow rate in kg/s, (4) mass flow rate in kg/yr, (5) price of utility in k/kg, (6) volumetric flow rate in m^3/yr , (8) price of utility in m^3/yr , (9) electricity usage in kW, (10) electricity usage in kWh, (11) price of electricity usage in k/kWh, and (12) price of utility in y/yr.

User	Utility	kg/s	kg/yr	\$/kg	m ³ /s	m ³ /yr	\$/m ³	kW	kWh	\$/kWh	\$/yr
E-117	Steam	6.571	1.87E+08	\$0.0161							\$3,010,036
E-118	Cooling Water	15.65	4.44E+08	\$0.0001	0.01565	4.44E+05	\$0.1378				\$61,218
E-123	Cooling Water	42.28	1.20E+09	\$0.0001	0.04228	1.20E+06	\$0.1378				\$165,386
E-127	Cooling Water	115.7	3.28E+09	\$0.0001	0.1157	3.28E+06	\$0.1378				\$452,582
Waste	Waste Water	19.74	5.60E+08	\$0.0013	0.01974	5.60E+05	\$1.2746				\$714,096
Disposal	Slag Cake	0.6899	1.96E+07	\$0.1785							\$3,495,212
C-113	Electricity							131.7	1038323	\$0.104	\$107,934
C-114	Electricity							238.7	1881911	\$0.104	\$195,625
C-144	Electricity							52.34	412649	\$0.104	\$42,895
G-153	Electricity							0.01327	104.621	\$0.104	\$11
H-122	Electricity							23.85	188033	\$0.104	\$19,546
J-111	Electricity							6.306	49716.5	\$0.104	\$5,168
J-141	Electricity							2.713	21389.3	\$0.104	\$2,223
J-145	Electricity							14.09	111086	\$0.104	\$11,547
L-116	Electricity							350.9	2766496	\$0.104	\$287,577
L-124	Electricity							3.916	30873.7	\$0.104	\$3,209
L-126	Electricity							5.651	44552.5	\$0.104	\$4,631
L-131	Electricity							11.51	90744.8	\$0.104	\$9,433
L-132	Electricity							0.5756	4538.03	\$0.104	\$472
L-149	Electricity							0.2706	2133.41	\$0.104	\$222
M-115	Electricity							203.1	1601240	\$0.104	\$166,449
N-155	Electricity							13961	1.1E+08	\$0.104	\$11,441,623
•		•	•	•	•	•	•	•		TOTAL	\$20,197,093

Table 1: Utility Table Itemizing All Major Sources of Utility Usage

OPERATING COST SUMMARY

Table 2 itemizes the cost of each operating cost. The tables gives each category's annual cost along with, when applicable, its cost per kg, m³, or kWh. The table breaks down the utilities into mass and volumetric flow rates of material usage along with kWh of electricity usage.

EQUIPMENT INFORMATION SUMMARY

Table 3 itemizes the cost of each process equipment item. The table gives the (1) name and type, (2) number, (3) capacity or size specification as well as the material of construction, (4) purchase cost of the base material, (5) base bare module factor, (6) material factor, (7) pressure factor, and (8) actual bare module cost for each equipment.

CAPITAL ESTIMATE

Table 4 presents the total capital investment for this plant. The Total Bare Module Cost is the sum of process equipment costs from Table 3. Contingency and Fee is 18% of the Total Bare Module Cost. Total Module Cost is the sum of Total Bare Module Cost and Contingency and Fee. Auxiliary Facilities is 30% of the Total Module Cost. Grass Roots Capital is the sum of Total Bare Module Cost and Auxiliary Facilities. Working Capital is 20% of the Grass Roots Capital. Total Capital Investment is the sum of Grass Roots Capital and Working Capital.

Total Bare Module Cost	\$93,700,00
Contingency and Fee	\$16,900,00
Total Module Cost	\$111,000,000
Auxiliary Facilities	\$33,100,000
Grass Roots Capital	\$144,000,000
Working Capital	\$28,800,000
Total Capital Investment	\$173,000,000

Table 4: Total Capital Investment

ECONOMIC ANALYSIS

Figure 6 shows the Yearly Discounted Cash Flow (\$) over Time (years). It displays the cash flow for an interest rate of 0%, 10%, and 13%. The Net Present Value of the plant is \$194 Million with a 0% interest rate, \$21 Million with a 10% interest rate, and \$0 with a 13% interest rate. Figure 6 also shows the Net Payout Time (8.25 years), the Payback Period (4.5 years), and the Discounted Break-even Point (11.5 years).

Table 5 summaries the economic parameters derived from Graph 1 along with the Net Rate of Return (20%), the Discounted Cash Flow Rate of Return (12.59%), and the Selling Price of Bio-oils (\$0.79/kg).

TABLE 2: OPERATING COST SUMMARY			Estimate Applies: 18			Page: 20 of 51	
Job Title: Biomass to Bio-oils: Companion Coal Gasificat	Cost	, ,	y: Alexandra Eicher				
Capacity: 2.24×10 ⁸ kg/yr of bio-oils		Cos	st Index Value: 595		Da	Date: 08 May 2013	
Capital							
Fixed capital, C _{FC}					\$143,757,255		
Working capital (10-20% of fixed capital), Cwc					\$28,751,451		
Total capital investment, CTC					\$172,508,705		
Manufacturing Expenses							
Direct					\$/yr	\$/kg	
Raw materials					\$696,735,641	\$0.63	
By-product credits					(\$841,455,851)	\$0.41	
Catalysts and solvents					\$49,399	\$1.91	
Operating labor					\$1,257,906		
Supervisory and clerical labor (10-20% of operating labor)					\$251,581		
Utilities							
Steam		186,500,750 kg/yr	15 barg @	\$0.0161 \$/kg	\$3,010,036	\$0.0161	
Electricity		118,312,314 kWh @	\$0.104 \$/kWh		\$12,298,565		
Cooling water		4,928,036 m ³ /yr @	\$0.14 \$/m ³		\$679,088	\$0.0001	
Waste disposal		19,581,018 kg/yr @	\$0.18 \$/kg	\$3,495,212 \$/yr	\$4,209,307	\$0.1798	
		560,269 m ³ /yr @	\$1.27 \$/m3	\$714,096 \$/yr			
Maintenance and repairs (2-10% of fixed capital)		econers milita	Griner Grin	<i>willi</i> ,050 <i>wiji</i>	\$14,375,725		
Operating supplies (10-20% of maint & repairs)					\$2,875,145		
Laboratory charges (10-20% of operating labor)					\$251,581		
Patents and royalties (0-6% of total expense)					\$8,442,000		
	A DME				(\$97,019,876)		
Indirect	To Della						
Overhead (payroll and plant), packaging,							
storage (50-70% of op. Labor+supervision+							
maint.)					\$11,119,649		
Local taxes (1-3% of fixed capital)					\$4,312,718		
Insurance (1-2% of fixed capital)					\$2,875,145		
Total	, A IME				\$18,307,512		
Total manufacturing expense, A ME = A DME + A DME					(\$78,712,365)		
General Expenses							
Administrative costs (25% of overhead)					\$2,779,912		
Distribution and selling (10% of total expense)					\$14,070,000		
Research and development (5% of total expense)					\$7,035,000		
Total general expens	e, A _{GE}				\$23,884,912		
Depreciation (approximately 10% of fixed capital), A BD					\$14,375,725		
Total Expenses, A TE					\$140,700,000		
Revenue from Sales	(224,000,000 kg/yr @	\$0.79 \$/kg), As		\$176,086,401	\$0.79	
Net annual profit, A NP			- were 0		\$35,386,401		
Income taxes (net annual profit times the tax rate), A_{II}					\$12,385,240		
Net annual profit after taxes $(A_{NP}-A_{TT}), A_{NNP}$					\$23,001,161		
Provide the second seco			Aftertax rate o	f return, i = (1.5 <i>A</i>		20 %	
					10.1.10.1.2.2.2	20 /0	

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-		INFORMATION SUMMARY		ch estimate appl st Index Type: (2014	Page: 21 of 51
Job Title: Bioma		By: Alexandra Eicher					
Cap	vacity: 2.24×	10° kg/yr of bio-oils		Cost Index Valu	ue: <u>595</u>		Date: 08 May 2013
Equipment Identification	Number	Capacity or Size Specifications (Material of Construction)	Purchased Equipment Cost (base material) Year 2013	Base Bare Module Factor, F _{BM}	Material Factor, F _M	Pressure or other Factors, F _p	Actual Bare Module Cost, C _{BM}
Mixers		Barrow Communities (IAD) 101-66 (Surplus) (Coders Parch	6331 0.CO				0.4/3.050
Slurry Tank: Agitator (Axial Turbine)	M-115	Power Consumption (kW): 101.55 (2 units) (Carbon Steel)	\$231,868	2.0			\$463,737 \$463,737
Total Mixers	1						\$40.5,737
Pumps Centrifugal (Radial)	L-124	Shaft Power (kW): 3.196 (Cast Iron)	\$8,337		1.00	1.09	\$29,968
Centrifugal (Radial)	L-126	Shaft Power (kW): 5.651 (Cast Iron)	\$10,320		1.00	1.11	\$37,488
Centrifugal (Radial)	L-131	Shaft Power (kW): 11.51 (Cast Iron)	\$13,671		1.00	1.06	\$48,607
Centrifugal (Radial)	L-132	Shaft Power (kW): 0.5756 (Cast Iron)	\$4,733		1.00	1.06	\$16,829
Centrifugal (Radial)	L-149	Shaft Power (kW): 0.2706 (Cast Iron)	\$3,846		1.00	1.18	\$14,385
Reciprocating (Positive Displacement)	L-116	Shaft Power (kW): 175.45 (2 units) (Cast Iron)	\$678,063		1.00	1.15	\$2,504,289
Total Pumps	6						\$2,651,566
Heat Exchangers							
Shell and Tube (Double Pipe)	E-117	Surface Area (m ²): 23.88 (Carbon Steel/Carbon Steel)	\$15,323		1.00	1.00	\$48,730
Shell and Tube (Double Pipe)	E-123	Surface Area (m ²): 477.7 (Carbon Steel/Carbon Steel)	\$241,627		1.00	1.00	\$768,396
Shell and Tube (Double Pipe)	E-127	Surface Area (m ²): 558.2 (5 units) (Carbon Steel/Carbon Steel)	\$1,407,529		1.00	1.00	\$4,476,083
Shell and Tube (Double Pipe)	E-129	Surface Area (m ²): 594.818 (11 units) (Carbon Steel/Carbon Steel)	\$3,295,698		1.00	1.00	\$10,480,648
Shell and Tube (Double Pipe)	E-133	Surface Area (m ²): 46.82 (Carbon Steel/Carbon Steel)	\$25,102		1.00	1.00	\$79,827
Condenser: Shell and Tube (Double Pipe)	E-148	Surface Area (m ²): 12.15 (Carbon Steel/Carbon Steel)	\$9,945		1.00	1.00	\$31,625
Condenser: Shell and Tube (Double Pipe)	E-150	Surface Area (m ²): 6.386 (Carbon Steel/Carbon Steel)	\$6,838		1.00	1.00	\$21,745
Shell and Tube (Fixed Tube Sheet and U-tube)	E-118	Surface Area (m ²): 539.5 (2 units) (Carbon Steel/Carbon Steel)	\$112,505		1.00	1.05	\$365,028
Total Heat Exchangers	8						\$16,272,082
Gas Movers and Compressors		Th. 3 Berry d. 80, 0.01222 (Casher Starb)	633.344		<u> </u>		201 222
Compressor (Axial)	G-153	Fluid Power (kW): 0.01327 (Carbon Steel)	\$23,364	3.5	<u> </u>		\$81,775
Total Gas Movers and Compressors	1				<u> </u>		\$81,775
Gas-Solid Contacting Equipment Rotary Dryer (Direct)	B-143	Internel Mahama (m ³): 4 231 (Cashan Study	\$90,770	2.3	<u> </u>		\$208,771
Total Gas-Solid Contacting Equipment	B-145	Internal Volume (m2): 4.731 (Carbon Steel)	\$90,770	2.3	<u> </u>		\$208,771
Drivers and Power Recovery Machines	1						\$200,771
Radial Gas Expander	N-155	Shaft Power (kW): 1269.182 (11 units) (Carbon Steel)	\$2,575,076	3.0			\$7,725,229
Total Drivers and Power Recovery Machines	1		02,010,010				\$7,725,229
Separators							
Cyclone	H-119	Volumetric Flow Rate (m ³ /s): 6.244(Carbon Steel)	\$13,448	3.0			\$40,343
Cyclone	H-146	Volumetric Flow Rate (m3/s): 0.3285 (Carbon Steel)	\$3,138	3.0			\$9,415
Cyclone	H-147	Volumetric Flow Rate (m ³ /s): 0.3389 (Carbon Steel)	\$3,167	3.0			\$9,502
Settler (Single-compartment Drum)	H-121	Filter Area (m2): 13.17 (Carbon Steel)	\$293,802	2.4			\$705,125
Vibratory Screen	H-122	Power Consumption (kW): 23.85 (Carbon Steel)	\$4,958	2.8			\$13,882
Gas Bag Filter	H-152	Volumetric Flow Rate (m ³ /s): 0.3821 (Carbon Steel)	\$44,282	2.2			\$97,421
Gas Bag Filter	H-154	Volumetric Flow Rate (m ³ /s): 0.006571 (Carbon Steel)	\$44,282	2.2			\$97,421
Electrostatic Precipitator	H-151	Volumetric Flow Rate (m ² /s): 0.3689 (Carbon Steel)	\$455,900	2.3			\$1,046,271
Total Separators	8						\$2,019,380
Crushers, Mills, and Grinders	0.115	Connection Restate 83 Feb	8010 A.C.				8000 000
Jaw Crusher	C-113	Capacity (kg/s): 52.04 Capacity (kg/s): 17.247 (3 ymits)	\$269,188	2.1			\$565,294 \$3,952,603
Rolling Compression Grinder Rolling Compression Grinder	C-114 C-144	Capacity (kg/s): 17.347 (3 units) Capacity (kg/s): 11.61	\$1,411,644 \$363,767	2.8	<u> </u>		\$3,952,603 \$1,018,549
Total Crushers, Mills, and Grinders	3	and and the state of the state	4303,107	6.0			\$5,536,446
Conveyors	,						401000,110
Belt	J-111	Conveying Distance: 50 m (Height: 25m, Width: 1m)	\$73,572	2.4			\$176,574
Beh	J-141	Conveying Distance: 50 m (Height: 25m, Width: 1m)	\$73,572	2.4			\$176,574
Screw: Auger	J-145	Conveying Distance: 25m, (Width: 0.325 m)	\$13,814	2.2			\$30,391
Total Conveyors	3						\$383,539
Storage Vessels							
Hopper (Atmospheric Bin)	F-112	Volume (m ²): 2053 (Carbon Steel)	\$9,375	2.1			\$19,688
Hopper (Atmospheric Bin)	F-142	Volume (m3): 1021 (Carbon Steel)	\$5,826	2.1			\$12,235
Total Storage Vessels	2						\$31,923
Process Vessels							
Gasification Reactor	R-110	Heating Duty (kJ/s): 54695.4 (5 units) (Stainless Steel)	\$19,309,589	2.7		1.03	\$53,905,852
Pyrolysis Reactor	R-140	Heating Duty (kJ/s): 7026 (Stainless Steel)	\$720,158	2.7	<u> </u>	0.99	\$1,921,394
AL 1 A 11	D-120	Volumetric Flow Rate (m ³ /s): 22.21 (Carbon Steel)	\$170,011 \$151,290	2.5	1.00	1.70	\$425,027
Venturi Scrubber	ID 2.2.4					1.70	\$830,690
Scrubber	D-125	Column Diameter (m): 3.589; Height (m): 17.95 (Carbon Steel) Column Diameter (m): 3.001; Height (m): 15.01 (Carbon Steel)					
Scrubber Absorber	D-128	Column Diameter (m): 3.001; Height (m): 15.01 (Carbon Steel)	\$113,818		1.00	1.67	\$617,895
Scrubber							

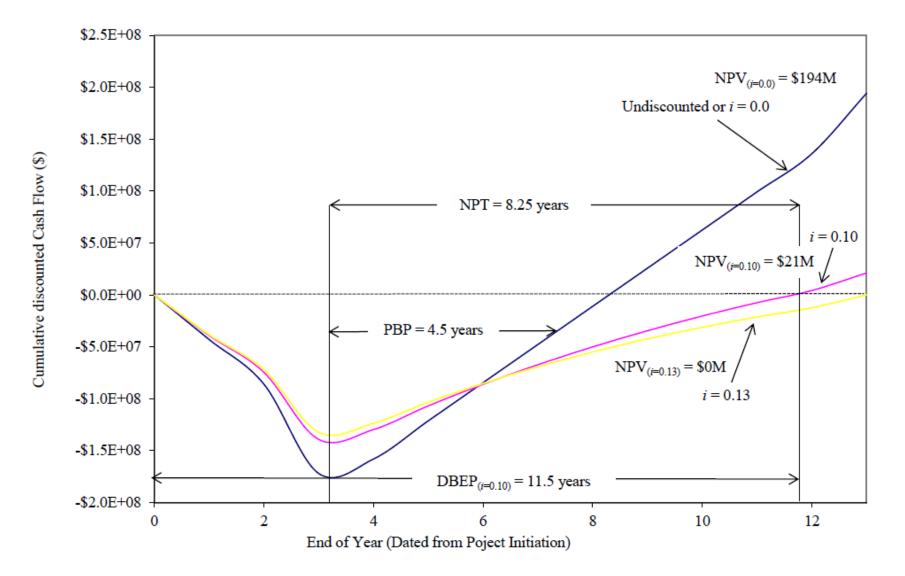


Figure 6: Yearly Discounted Cash Flow (\$) per Year

Selling Price of Bio-oils	\$0.79/kg
NPV (<i>i</i> = 0.13)	\$0
NPV $(i = 0.10)$	\$21,000,000
NPV $(i = 0.10)$	\$21,000,000
DCFRR	12.59%
DBEP	11.5 years
NRR	20%
NPT	8.25 years
PBP	4.5 years

Table 5: Summarization of Economic Parameters

INNOVATION AND OPTIMIZATION

This plant design features a few key, innovative systems. First, steam from the gasifier is used to heat the biomass. This takes advantage of excess heat, in the form of steam, from the coal gasifier. Not only does this increase the thermal efficiency of the gasifier but it also provides a convenient and inexpensive method to heat the biomass. Second, bio-char from the pyrolysis of biomass is recycled to the gasifier. This decreases the amount of raw coal needed to be fed to the coal gasifier, thus, decreasing the cost of raw materials. Third, the bio-gas from the pyrolysis of biomass is used to dry the raw biomass to a moisture content of 10% as well as to blow the biomass up through the fluidized bed. This decreases waste streams, and the cost associated with processing waste streams, leaving the plant.

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COMPUTER PROGRAMS

MATERIAL AND ENERGY BALANCES - EES

Input:

"This unit system is set to temperature in K, pressure in Bara, energy in kJ, and specific properties to a molar basis"

"Ultimate Compostion of Coal A in wt% (Zheng, 2005)" wt%_C_1=0.7048 wt%_H_1=0.0485 wt%_N_1=0.014 wt%_S_1=0.0364 wt%_O_1=0.0783 wt%_Ash_1=0.118

"Given a Coal Feed to Gasifier Flow Rate of 53.3 kg/s (Zheng, 2005)"

Coal_1=53.3-Biochar_42 C_1=wt%_C_1*Coal_1 H_1=wt%_H_1*Coal_1 N_1=wt%_N_1*Coal_1 S_1=wt%_S_1*Coal_1 O_1=wt%_O_1*Coal_1 Ash_1=wt%_Ash_1*Coal_1

"Other Feed Rates in kg/s (Zheng, 2005)" O2_3=0.97*Coal_1

"Composition of Raw Gases from Texaco Process in mol% (Zheng, 2005)"

mol%_O2_11=0.007 mol%_N2_11=0.01 mol%_H2_11=0.286 mol%_CO_11=0.384 mol%_CO2_11=0.126 mol%_H2O_11=0.174 mol%_CH4_11=0.0015 mol%_H2S_11=0.0096

"Molecular Weights of Raw Gases from Texaco Process in kg/kmol"

MW_02=32 MW_N2=28 MW_H2=2 MW_CO=28 MW_CO2=44 MW_H2O=18 MW_CH4=16 MW_H2S=34

"Other Molecular Weights in kg/kmol"

MW_C=12 MW_H=1 MW_N=14 MW_S=32 MW_O=16 "Composition of Raw Gases from Texaco Process in kg using a 1 kmol basis"

mass_O2_11=mol%_O2_11*MW_O2

mass_N2_11=mol%_N2_11*MW_N2

mass_H2_11=mol%_H2_11*MW_H2

mass_CO_11=mol%_CO_11*MW_CO

mass_CO2_11=mol%_CO2_11*MW_CO2

mass_H2O_11=mol%_H2O_11*MW_H2O

mass_CH4_11=mol%_CH4_11*MW_CH4

mass_H2S_11=mol%_H2S_11*MW_H2S

mass_Total_11=mass_O2_11+mass_N2_11+mass_H2_11+mass_CO_11+mass_CO2_11+mass_H2O_11+mass_CH4_11+mass_H2S_11

"Composition of Raw Syngas from Texaco Process in wt%"

wt%_O2_11=mass_O2_11/mass_Total_11 wt%_N2_11=mass_N2_11/mass_Total_11 wt%_H2_11=mass_H2_11/mass_Total_11 wt%_CO_11=mass_CO_11/mass_Total_11 wt%_H2O_11=mass_H2O_11/mass_Total_11 wt%_CH4_11=mass_CH4_11/mass_Total_11 wt%_H2S_11=mass_H2S_11/mass_Total_11

 $\begin{array}{l} Product_11=Coal_1+Biochar_42+H2O_2+O2_3-Ash_1\\ O2_11=wt\%_O2_11^*Product_11\\ N2_11=wt\%_N2_11^*Product_11\\ H2_11=wt\%_H2_11^*Product_11\\ CO_11=wt\%_CO_11^*Product_11\\ CO2_11=wt\%_CO2_11^*Product_11\\ H2O_11=wt\%_H2O_11^*Product_11\\ H2O_11=wt\%_H2O_11^*Product_11\\ CH4_11=wt\%_CH4_11^*Product_11\\ H2S_11=wt\%_H2S_11^*Product_11\\ Slag_11=Ash_1+Ash_42\\ \end{array}$

"------H-119------"

Slag_13=Slag_11*0.9 Slag_12=Slag_11*0.1

"------" " "Assume the liquid oxygen feed is at 90 K." "At 90 K, the enthalpy of vaporization of O2 212.98 kJ/kg." "The heat capacity of oxygen gas is 29 kJ/kmol*K." H_E117=(O2_3*212.98)+((O2_3/MW_O2)*29*(133-90)) Q_E117=O2_3*212.98 "cooling steam" H_E117=(mol_H2O_10)*(enthalpy(Steam,x=1,P=16)-enthalpy(Steam,x=0,P=16)) H2O_10=mol_H2O_10*MW_H2O

"------R-110------"

T_ref=298.15 "reference temperature" T_3=133 T_9=298.15 T_11_1=1644 "(Ackerson, 2012)"

P_ref=1 "reference temperature" P_3=16 P_9=44

call JANAF('H2O(I)',T_ref:cph,hh,sh)

"Reactants" C_9=C_1 H_9=H_1 N_9=N_1 S_9=S_1 O_9=O_1 Ash_9=Ash_1 H2O_9=H2O_2

mol_C_9=C_9/MW_C mol_H_9=H_9/MW_H mol_N_9=N_9/MW_N mol_S_9=S_9/MW_S mol_O_9=O_9/MW_O mol_H2O_9=H2O_9/MW_H2O mol_O2_3=O2_3/MW_O2

C_42=wt%_C_1*Biochar_42 H_42=wt%_H_1*Biochar_42 N_42=wt%_N_1*Biochar_42 S_42=wt%_S_1*Biochar_42 O_42=wt%_O_1*Biochar_42 Ash_42=wt%_Ash_1*Biochar_42

 $\begin{array}{l} mol_C_42=C_42/MW_C\\ mol_H_42=H_42/MW_H\\ mol_N_42=N_42/MW_N\\ mol_S_42=S_42/MW_S\\ mol_O_42=O_42/MW_O\\ \end{array}$

 $\label{eq:hardware} \begin{array}{l} h_C_42 = mol_C_42^*(enthalpy(C,T=T_37)-enthalpy(C,T=T_ref)) \\ h_H_42 = mol_H_42^*(enthalpy(H,T=T_37)-enthalpy(H,T=T_ref)) \\ h_S_42 = mol_S_42^*(enthalpy(S,T=T_37)-enthalpy(S,T=T_ref)) \\ h_O_42 = mol_O_42^*(enthalpy(O,T=T_37)-enthalpy(O,T=T_ref)) \\ \end{array}$

h_H2O_9_ref=enthalpy(Water,T=T_ref,P=P_ref) h_H2O_9_at=enthalpy(Water,T=T_9,P=P_9) h_H2O_9=mol_H2O_9*(h_H2O_9_at-h_H2O_9_ref)

h_O2_3_ref=enthalpy(O2,T=T_ref) h_O2_3_at=enthalpy(O2,T=T_3) h_O2_3=mol_O2_3*(h_O2_3_at-h_O2_3_ref)

h_reactants=h_H2O_9+h_O2_3+h_C_42+h_H_42+h_N_42+h_S_42+h_O_42

"Products"

mol_N2_11=N2_11/MW_N2 mol_H2_11=H2_11/MW_H2 mol_CO_11=CO_11/MW_CO mol_CO2_11=CO2_11/MW_CO2 mol_H2O_11=H2O_11/MW_H2O mol_CH4_11=CH4_11/MW_CH4 mol_H2S_11=H2S_11/MW_H2S

 $\label{eq:h_N2_11=mol_N2_11*enthalpy(N2,T=T_ref) \\ h_H2_11=mol_H2_11*enthalpy(N2,T=T_ref) \\ h_CO_11=mol_CO_11*enthalpy(CO,T=T_ref) \\ h_CO2_11=mol_CO2_11*enthalpy(CO2,T=T_ref) \\ h_H2O_11=mol_H2O_11*enthalpy(H2O,T=T_ref) \\ h_CH4_11=mol_CH4_11*enthalpy(CH4,T=T_ref) \\ h_H2S_11=mol_H2S_11*enthalpy(H2S,T=T_ref) \\ \end{tabular}$

"Heat of Combustion"

"Heat Capacity of Coal is 31400 kJ/kg"

 $\label{eq:h_combustion} h_combustion = (mol_C_9+mol_C_42)^* enthalpy (CO2,T=T_ref) + (mol_H_9+mol_H_42)^* 0.5^* hh + (mol_S_9+mol_S_42)^* enthalpy (SO2,T=T_ref) + 31400^* (Coal_1+Biochar_42)$

"Heat of Reaction"

 $\label{eq:h_reaction_h_N2_11+h_H2_11+h_CO_11+h_CO2_11+h_H2O_11+h_CH4_11+h_H2S_11-h_combustion-mol_H2O_9*hh$

"Heating products from 298 to 1644"

"Heat Capacity of Slag is 2 kJ/kg" $lh_H2O_11=(mol_H2O_11)*(enthalpy(steam,x=1,P=44)-enthalpy(steam,x=0,P=44))$ $h_products=mol_N2_11*(enthalpy(N2,T=T_11_1)$ $enthalpy(N2,T=T_ref))+mol_H2_11*(enthalpy(H2,T=T_11_1)$ $enthalpy(H2,T=T_ref))+mol_CO_11*(enthalpy(CO,T=T_11_1)$ $enthalpy(CO,T=T_ref))+mol_H2O_11*(enthalpy(CO2,T=T_11_1)$ $enthalpy(CO2,T=T_ref))+mol_H2O_11*(enthalpy(H2O,T=T_11_1)$ $enthalpy(H2O,T=T_ref))+mol_CH4_11*(enthalpy(CH4,T=T_11_1)$ $enthalpy(H2O,T=T_ref))+mol_CH4_11*(enthalpy(CH4,T=T_11_1)$ $enthalpy(H2S,T=T_ref))+Slag_11*2*(T_11_1-T_ref)$ $O=h_products+h_reactants+h_reaction+lh_H2O_11$

"------E-118------"

T_11_2=523

T_7_2=1023 "This is the maximum temperature that you can create steam at from water through a heat exchanger."

T_7_1=298

 $\begin{array}{l} H_E118=+mol_H2_11^*(enthalpy(H2,T=T_11_2)-enthalpy(H2,T=T_11_1))+mol_CO_11^*(enthalpy(CO,T=T_11_2)-enthalpy(CO2,T=T_11_1))+mol_CO2_11^*(enthalpy(CO2,T=T_11_2)-enthalpy(CO2,T=T_11_1))+mol_H2O_11^*(enthalpy(H2O,T=T_11_2)-enthalpy(H2O,T=T_11_1))+mol_CH4_11^*(enthalpy(CH4,T=T_11_2)-enthalpy(H2S,T=T_11_1))+mol_H2S_11^*(enthalpy(H2S,T=T_11_2)-enthalpy(H2S,T=T_11_1))+Slag_11^*2^*(T_11_2-T_11_1)\\ H_E118=mol_H2O_7^*(enthalpy(H2O,T=T_7_1)-enthalpy(Steam,T=T_7_2,P=16))\\ H2O_7=mol_H2O_7^*MW_H2O \end{array}$

"-----D-120------"

"Material"

T_12=523 P_12=14 H2S_rho=Density(H2S,T=T_12,P=P_12) N2_rho=Density(N2,T=T_12,P=P_12) CO_rho=Density(CO,T=T_12,P=P_12) CO2_rho=Density(CO2,T=T_12,P=P_12) CH4_rho=Density(CH4,T=T_12,P=P_12) H2O_rho=Density(H2O,T=T_12,P=P_12) H2_rho=Density(H2,T=T_12,P=P_12) Slag_16=Slag_12*0.999 Slag_15=Slag_12*0.001 H2 12=H2 11 CO_12=CO_11 CH4_12=CH4_11 CO2_12=CO2_11 H2S_12=H2S_11 N2_12=N2_11 H2O_12=H2O_11 Gas 12=H2S 12+N2 12+CO 12+CO2 12+CH4 12+H2O 12+H2 12 Vol_Gas_12=(H2S_12/MW_H2S/H2S_rho)+(N2_12/MW_H2/N2_rho)+(CO_12/MW_CO/CO_rh o)+(CO2_12/MW_CO2/CO2_rho)+(CH4_12/MW_CH4/CH4_rho)+(H2O_12/MW_H2O/H2O_rh o)+(H2 12/MW H2/H2 rho)+Slag 12/2114 H2O 14=Vol Gas 12 Vol_H2O_14=H2O_14/1000 Vol D120=Vol Gas 12+Vol H2O 14 H2O 15=H2O 12 mol_H2O_15=H2O_15/MW_H2O H2O_16=H2O_14 mol_H2O_16=H2O_16/MW_H2O H2_15=H2_12 CO 15=CO 12 CH4 15=CH4 12 CO2 15=CO2 12 H2S 15=H2S 12 N2_15=N2_12 "Energy" T_15=T_sat(Water, P=13.5) mol_N2_15=N2_15/MW_N2 mol H2 15=H2 15/MW H2 mol_CO_15=CO_15/MW_CO mol_CO2_15=CO2_15/MW_CO2 mol CH4 15=CH4 15/MW CH4 mol_H2S_15=H2S_15/MW_H2S H D120=mol N2 15*(enthalpy(N2,T=T 15)enthalpy(N2,T=T 12))+mol H2 15*(enthalpy(H2,T=T 15)enthalpy(H2,T=T_12))+mol_CO_15*(enthalpy(CO,T=T_15)enthalpy(CO,T=T_12))+mol_CO2_15*(enthalpy(CO2,T=T_15)enthalpy(CO2,T=T_12))+mol_H2O_15*(enthalpy(H2O,T=T_15)enthalpy(H2O,T=T_12))+mol_CH4_15*(enthalpy(CH4,T=T_15)-enthalpy(CH4, T=T_12))+mol_H2S_15*(enthalpy(H2S,T=T_15)-enthalpy(H2S,T=T_12))+Slag_15*2*(T_15-T 12) H_D120=mol_H2O_16*(Enthalpy(Water,T=303,P=14)-Enthalpy(Water,T=T 16,P=13.5))+Slag 16*2*(T 12-T 16)

"------"

Slag_18=Slag_16*0.999 Slag_17=Slag_16*0.001 H2O_18=Slag_18 H2O_17=H2O_16-H2O_18

"-----"

Slag_19=Slag_18*0.999

Slag_20=Slag_18*0.001 H2O_19=H2O_18*.1 H2O_20=H2O_18*.9 H2O_21=H2O_20+H2O_17 mol_H2O_21=H2O_21/MW_H2O H2O_22=H2O_21+(H2O_26-H2O_27) mol_H2O_22=H2O_22/MW_H2O mol_H2O_2627=(H2O_26-H2O_27)/MW_H2O H2O_23=H2O_22-H2O_14

"-----E-123------"

H_22=(mol_H2O_21*Enthalpy(Water,T=T_16,P=12.5))+(mol_H2O_2627*Enthalpy(Water,T=T_ 26,P=13)) H_22=mol_H2O_22*Enthalpy(Water,T=T_22,P=12.5) T_14=T_22 T_24=348 H_E123=mol_H2O_14*(Enthalpy(Water,T=T_14,P=12.5)-Enthalpy(Water,T=303,P=12.5)) H_E123=mol_H2O_24*(Enthalpy(Water,T=T_24,P=4)-Enthalpy(Water,T=T_ref,P=4)) mol_H2O_14=H2O_14/MW_H2O mol_H2O_24=H2O_24/MW_H2O

"-----D-125------"

"Material"

P H2O sat=P sat(Steam,T=T 25) H2O_factor=P_H2O_sat/12 H2O_25=H2O_15*H2O_factor H2O 26=(H2O 15*(1-H2O factor))+H2O 27 mol_H2O_25=H2O_25/MW H2O mol_H2O_26=H2O_26/MW_H2O H2 25=H2 15 CO_25=CO_15 CH4 25=CH4 15 CO2 25=CO2 15 H2S_25=H2S_15 N2 25=N2 15 Slag 25=Slag 15 mol_N2_25=N2_25/MW_N2 mol H2 25=H2 25/MW H2 mol CO 25=CO 25/MW CO mol_CO2_25=CO2_25/MW_CO2 mol_CH4_25=CH4_25/MW_CH4 mol H2S 25=H2S 25/MW H2S

"Energy"

 $\label{eq:relation} \begin{array}{l} T_{2} = 318 \\ T_{2} = 363 \\ H_{D12} = mol_N2_{2} 5^{*} (enthalpy(N2,T=T_{2}5)- enthalpy(N2,T=T_{1}5)) + mol_H2_{2} 5^{*} (enthalpy(H2,T=T_{2}5)- enthalpy(N2,T=T_{1}5)) + mol_CO_{2} 5^{*} (enthalpy(CO,T=T_{2}5)- enthalpy(CO2,T=T_{1}5)) + mol_CO2_{2} 5^{*} (enthalpy(CO2,T=T_{2}5)- enthalpy(CO2,T=T_{1}5)) + mol_H2O_{2} 5^{*} (enthalpy(H2O,T=T_{2}5)- enthalpy(H2O,T=T_{1}5)) + mol_CH4_{2} 5^{*} (enthalpy(CH4,T=T_{2}5)- enthalpy(H2O,T=T_{1}5)) + mol_CH4_{2} 5^{*} (enthalpy(CH4,T=T_{2}5)- enthalpy(CH4,T=T_{1}5)) + mol_H2S_{2} 5^{*} (enthalpy(H2S,T=T_{2}5)- enthalpy(H2S,T=T_{1}5)) + Slag_{2} 5^{*} 2^{*} (T_{2}5-T_{1}5) \\ H_{D12} = mol_H2O_{2} 7^{*} (Enthalpy(Water,T=303,P=13.5)- Enthalpy(Water,T=T_{2}6,P=13.5)) \\ H2O_{2} 7 = mol_H2O_{2} 7^{*} MW_{H2O} \end{array}$

"------E-127------"

 $\label{eq:tau} \begin{array}{l} T_27=T_26\\ T_28=348\\ H_E127=mol_H2O_27^*(Enthalpy(Water,T=T_27,P=13.5)\text{-}Enthalpy(Water,T=303,P=13.5))\\ H_E127=mol_H2O_28^*(Enthalpy(Water,T=T_28,P=4)\text{-}Enthalpy(Water,T=T_ref,P=4))\\ mol_H2O_28=H2O_28/MW_H2O\\ \end{array}$

"-----D-128------"

"Material"

absorb eff=.999 H2 29=H2 25 CO 29=CO 25 CH4 29=CH4 25 mol MEA 34=.4*mol CO2 25 MEA 34=mol MEA 34*61.08 MEA_30=MEA_34 mol MEA 30=mol MEA 34 MEAaqu_34=MEA_34/0.2 "Assume 20wt% MEA aqueous feed." aquFORmea_34=MEAaqu_34-MEA_34 H2O 34=aquFORmea 34 mol H2O 34=H2O 34/MW H2O mol_CO2_30=absorb_eff*mol_CO2_25 CO2 30=mol CO2 30*MW CO2 mol H2S 30=absorb eff*mol H2S 25 H2S 30=mol H2S 30*MW H2S H2O 30=H2O 34+H2O 25 mol_H2O_30=H2O_30/MW_H2O mol_N2_30=absorb_eff*mol_N2_25 N2 30=mol N2 30*MW N2

"Energy"

 $T_{29=311}$ $T_{34=298.15} "Entering Absorber, After Heat Exchanger"$ $0=mol_CO_{25}^{(enthalpy(CO,T=T_{29})-}$ enthalpy(CO,T=T_{25}))+mol_H2_25^{(enthalpy(H2,T=T_{29})-}
enthalpy(H2,T=T_25))+mol_CH4_25^{(enthalpy(CH4,T=T_{25})-enthalpy(CH4, T=T_ref))+mol_CO2_30^{(enthalpy(CO2,T=T_{30})-}
enthalpy(CO2,T=T_25))+mol_N2_30^{(enthalpy(N2,T=T_{30})-} enthalpy(N2,T=T_{25}))+mol_H2S_30^{*(enthalpy(H2S,T=T_{30})-}
enthalpy(H2S,T=T_25))+mol_MEA_30^{*140.508^{*}}(T_{30}- $T_{34})+mol_H2O_{25}^{*}(Enthalpy(Water,T=T_{30},P=12.5)-$ Enthalpy(Water,T=T_25,P=13))+mol_H2O_34^{*}(Enthalpy(Water,T=T_{30},P=12.5)Enthalpy(Water,T=T_34,P=13))

"------E-129------" $T_30_1=373 \text{ "Entering Stipper, After Heat Exchanger"}$ $H_E129=mol_CO2_30^*(enthalpy(CO2,T=T_30_1)-enthalpy(CO2,T=T_30))+mol_N2_30^*(enthalpy(N2,T=T_30_1)-enthalpy(N2,T=T_30))+mol_H2S_30^*(enthalpy(H2S,T=T_30_1)-enthalpy(H2S,T=T_30))+mol_MEA_30^*140.508^*(T_30_1-T_30)+mol_H2O_30^*(Enthalpy(Water,T=T_30_1,P=12.5)-Enthalpy(Water,T=T_30,P=12.5))$ $H_E129=mol_MEA_34^*140.508^*(T_34_1-T_34)+mol_H2O_34^*(Enthalpy(Water,T=T_34_1,P=13)-Enthalpy(Water,T=T_34,P=13))$

"-----"

"Material"

H2S_31=H2S_30 mol_H2S_31=H2S_31/MW_H2S CO2_31=CO2_30 mol_CO2_31=CO2_31/MW_CO2 N2_31=N2_30 mol_N2_31=N2_31/MW_N2 H2O_31=H2O_30 mol_H2O_31=H2O_31/MW_H2O

mol_MEA_33=mol_MEA_34*0.1 MEA_33=mol_MEA_33*61.08 MEAaqu_33=MEA_33/0.2"this is an assumption" aquFORmea_33=MEAaqu_33-MEA_33 H2O_33=aquFORmea_33 mol_H2O_33=H2O_33/MW_H2O

mol_MEA_32=mol_MEA_34+mol_MEA_33 MEA_32=mol_MEA_32*61.08 H2O_32=H2O_34+H2O_33 mol_H2O_32=H2O_32/MW_H2O

"Energy"

T 32=T 34 1 T_33=T_34_1 T 33 2=383 H D130=mol MEA 30*140.508*(T 30 1-T ref)+mol MEA 33*140.508*(T 33 2-T ref)+mol H2O 30*(Enthalpy(Water,T=T 30 1,P=1)-Enthalpy(Water,T=T_ref,P=112.5))+mol_H2O_33*(Enthalpy(Steam,T=T_33_2,P=1)-Enthalpy(steam,T=T_ref,P=12.5))+mol_CO2_30*(enthalpy(CO2,T=T_30_1)enthalpy(CO2,T=T_ref))+mol_N2_30*(enthalpy(N2,T=T_30)enthalpy(N2,T=T_ref))+mol_H2S_30*(enthalpy(H2S,T=T_30)-enthalpy(H2S,T=T_ref)) h_MEA_32=mol_MEA_32*140.508*(T_32-T_ref) h_H2O_32=mol_H2O_32*(Enthalpy(Water,T=T_32,P=12)-Enthalpy(Water,T=T_ref,P=1)) h_CO2_31=mol_CO2_31*(enthalpy(CO2,T=T_31)-enthalpy(CO2,T=T_30_1)) h_N2_31=mol_N2_31*(enthalpy(N2,T=T_31)-enthalpy(N2,T=T_30_1)) h_H2S_31=mol_H2S_31*(enthalpy(H2S,T=T_31)-enthalpy(H2S,T=T_30_1)) h H2O 31=mol H2O 31*(Enthalpy(Water,T=T 31,P=12)-Enthalpy(Water,T=T 30 1,P=12.5)) H D130=h CO2 31+h N2 31+h H2S 31+h H2O 31+h MEA 32+h H2O 32

"------E-133------"

 $\begin{array}{l} T_35_1=388\\ T_35_2=383\\ H_E133=mol_MEA_33^*140.508^*(T_33_2-T_33)+mol_H2O_33^*(Enthalpy(Water,T=T_33_2,P=12.5)-Enthalpy(Water,T=T_33,P=12.5))\\ H_E133=mol_H2O_35^*(Enthalpy(Water,T=T_35_1,P=16)-Enthalpy(Water,T=T_35_2,P=16))\\ H2O_35=mol_H2O_35/MW_H2O\\ \end{array}$

"------R-140------"

T_36=348 T_37=823 H2O_Biomass_4=Biomass_4 Biomass_36=Biomass_4 H2O_Biomass_36=Biomass_36/9 "assuming that the biomass has a moisture content of 10%" H2O_36=H2O_Biomass_36+Biogas_56

"Heat Capacity of Biomass is 531 kJ/kg"

 $\label{eq:2.1} \begin{array}{l} 0 = Biomass_36^*531 + H2O_36/MW_H2O^*(enthalpy(H2O,T=T_37)-enthalpy(H2O,T=T_36)) + mol_H2O_7^*(enthalpy(H2O,T=T_37)-enthalpy(H2O,T=T_36)) \\ H_R140 = Biomass_36^*531 + H2O_36/MW_H2O^*(enthalpy(H2O,T=T_37)-enthalpy(H2O,T=T_37)) \\ enthalpy(H2O,T=T_36)) \end{array}$

"Given a yield of 75% bio-oil, 12% biochar, and 13% biogas (Ringer, 2006)"

Biototal_36=Biomass_36+H2O_Biomass_36 Biooil_37=Biototal_36*0.75 Biochar_37=Biototal_36*0.12 Biogas_37=Biototal_36*0.13

"------"

Biochar_39=Biochar_37*0.9 Biochar_38=Biochar_37*0.1 Biochar_41=Biochar_38*0.9 Biochar_40=Biochar_38*0.1 Biochar_42=Biochar_39+Biochar_41

"-----E-148------"

Biooil_44=Biooil_37*0.7 Biooil_43=Biooil_37*0.3 T_40=T_37 Biogas_43=Biogas_37 "Heat capacity of the biogas is assumed to be the heat capacity of steam which is 2 kJ/kg/K." "Heat capacity of bio-oil is 2.435 kJ/kg/K (Goteti, 2010)." "Heat of vaporization for bio-oil is 609.9 kJ/kg (Goteti, 2010)." $T_44=T_43$ 0=Biogas_43*2*(T_43-T_40)+Biooil_44*(609.9+(2.435*(T_44-T_40)))+Biooil_43*2.435*(T_43-T_40) H_E148=Biooil_44*(609.9+(2.435*(T_44-T_40)))

"-----E-150------"

 $\begin{array}{l} \text{Biooil}_{46} = \text{Biooil}_{43}^{*0.7} \\ \text{Biooil}_{45} = \text{Biooil}_{43}^{*0.3} \\ \text{Biogas}_{45} = \text{Biogas}_{43} \\ \text{T}_{46} = \text{T}_{45} \\ 0 = \text{Biogas}_{45}^{*2*}(\text{T}_{45} - \text{T}_{43}) + \text{Biooil}_{46}^{*}(609.9 + (2.435^{*}(\text{T}_{46} - \text{T}_{43}))) + \text{Biooil}_{45}^{*2.435*}(\text{T}_{45} - \text{T}_{43}) \\ \text{H}_{E} 150 = \text{Biooil}_{46}^{*}(609.9 + (2.435^{*}(\text{T}_{46} - \text{T}_{43}))) \\ \text{Biooil}_{47} = \text{Biooil}_{44}^{+2.435*}(\text{T}_{44}) + (\text{Biooil}_{46}^{*2.435*}\text{T}_{46}) \\ \text{H}_{47} = (\text{Biooil}_{47}^{*2.435*}\text{T}_{47}^{-47}) \end{array}$

"------H-151------"

 $\begin{array}{l} Biooil_49 = Biooil_45^{*}0.95\\ Biooil_48 = Biooil_45^{*}0.05\\ Biogas_48 = Biogas_45\\ T_49 = T_48\\ 0 = Biogas_48^{*}2^{*}(T_48 - T_45) + Biooil_49^{*}(609.9 + (2.435^{*}(T_49 - T_45))) + Biooil_48^{*}(609.9 + (2.435^{*}(T_49 - T_45)))\\ Biooil_50 = Biooil_47 + Biooil_49\\ H_50 = (Biooil_49^{*}2.435^{*}T_49) + (Biooil_47^{*}2.435^{*}T_47)\\ H_50 = Biooil_50^{*}2.435^{*}T_50\\ \end{array}$

"------H-152------"

 $\begin{array}{l} Biooil_52{=}Biooil_48{*}0.95\\ Biogas_51{=}Biogas_48\\ T_52{=}T_51\\ 0{=}Biogas_51{*}2{*}(T_51{-}T_48){+}Biooil_52{*}2.435{*}(T_52{-}T_48)\\ Biooil_53{=}Biooil_50{+}Biooil_52\\ H_53{=}(Biooil_52{*}2.435{*}T_52){+}(Biooil_50{*}2.435{*}T_50)\\ H_53{=}Biooil_53{*}2.435{*}T_53\\ \end{array}$

"-----B-143------"

"Heat capacity of biomass is 1.2 KJ/kg/K (Goteti, 2010)" "Heat capacity of water is 4.181 kJ/kg/K (Goteti, 2010)" "Heat of vaporization of the biogas is assumed to be the heat of vaporization of water at 75C is 2322.87 kJ/kg (Goteti, 2010)" 0=Biomass_4*1.2*(T_36-T_ref)+H2O_Biomass_4*4.181*(T_36-T_ref)-Biogas_56*2322.8 Biogas_57=Biogas_51-Biogas_56 H_B143=Biomass_4*1.2*(T_36-T_ref)+H2O_Biomass_4*4.181*(T_36-T_ref) T_57=T_51

"-----"" mol_Biogas_57=Biogas_57/MW_H2O H_G153=mol_Biogas_57*(enthalpy(Steam,T=T_57,P=16)-enthalpy(Steam,T=T_57,P=13))

"------"

 $\begin{array}{l} H2S_rho_H119=Density(H2S,T=T_12,P=44)^*MW_H2S\\ N2_rho_H119=Density(N2,T=T_12,P=44)^*MW_N2\\ CO_rho_H119=Density(CO,T=T_12,P=44)^*MW_CO\\ CO2_rho_H119=Density(CO2,T=T_12,P=44)^*MW_CO2\\ CH4_rho_H119=Density(CH4,T=T_12,P=44)^*MW_CH4\\ H2O_rho_H119=Density(H2O,T=T_12,P=44)^*MW_H2O\\ H2_rho_H119=Density(H2,T=T_12,P=44)^*MW_H2\\ V_H119=H2S_11/H2S_rho_H119+N2_11/N2_rho_H119+CO_11/CO_rho_H119+CO2_11/CO2\\ _rho_H119+CH4_11/CH4_rho_H119+H2O_11/H2O_rho_H119+H2_11/H2_rho_H119+Slag_1\\ 1/2114\\ \end{array}$

"------N-155------"

H2S_rho_N155=Density(H2S,T=T_12,P=42.5)*MW_H2S N2_rho_N155=Density(N2,T=T_12,P=42.5)*MW_N2 CO_rho_N155=Density(CO,T=T_12,P=42.5)*MW_CO CO2_rho_N155=Density(CO2,T=T_12,P=42.5)*MW_CO2 CH4_rho_N155=Density(CH4,T=T_12,P=42.5)*MW_CH4 H2O_rho_N155=Density(H2O,T=T_12,P=42.5)*MW_H2O H2_rho_N155=Density(H2,T=T_12,P=42.5)*MW_H2 V_N155=H2S_11/H2S_rho_N155+N2_11/N2_rho_N155+CO_11/CO_rho_N155+CO2_11/CO2 _rho_N155+CH4_11/CH4_rho_N155+H2O_11/H2O_rho_N155+H2_11/H2_rho_N155+Slag_1 2/2114 M_N155=H2S_11+N2_11+CO_11+CO2_11+CH4_11+H2O_11+H2_11+Slag_12 rho_N155=M_N155/V_N155

"-----D-125------"

 $\begin{array}{l} H2S_rho_D125=Density(H2S,T=T_15,P=13.5)*MW_H2S\\ N2_rho_D125=Density(N2,T=T_15,P=13.5)*MW_N2\\ CO_rho_D125=Density(CO,T=T_15,P=13.5)*MW_CO\\ CO2_rho_D125=Density(CO2,T=T_15,P=13.5)*MW_CO2\\ CH4_rho_D125=Density(CH4,T=T_15,P=13.5)*MW_CH4\\ H2O_rho_D125=Density(H2O,T=T_15,P=13.5)*MW_H2O\\ H2_rho_D125=Density(H2,T=T_15,P=13.5)*MW_H2\\ \end{array}$

V_D125=H2S_11/H2S_rho_D125+N2_11/N2_rho_D125+CO_11/CO_rho_D125+CO2_11/CO2 _rho_D125+CH4_11/CH4_rho_D125+H2O_11/H2O_rho_D125+H2_11/H2_rho_D125

"------D-128------"

H2S_rho_D128=Density(H2S,T=T_25,P=13)*MW_H2S N2_rho_D128=Density(N2,T=T_25,P=13)*MW_N2 CO_rho_D128=Density(CO,T=T_25,P=13)*MW_CO CO2_rho_D128=Density(CO2,T=T_25,P=13)*MW_CO2 CH4_rho_D128=Density(CH4,T=T_25,P=13)*MW_CH4 H2O_rho_D128=Density(H2O,T=T_25,P=13)*MW_H2O H2_rho_D128=Density(H2,T=T_25,P=13)*MW_H2 V_D128=H2S_25/H2S_rho_D128+N2_25/N2_rho_D128+CO_25/CO_rho_D128+CO2_25/CO2 _rho_D128+CH4_25/CH4_rho_D128+H2O_25/H2O_rho_D128+H2_25/H2_rho_D128

"-----D-128------"

H2S_rho_D130=Density(H2S,T=T_30,P=12.5)*MW_H2S N2_rho_D130=Density(N2,T=T_30,P=12.5)*MW_N2 CO2_rho_D130=Density(CO2,T=T_30,P=12.5)*MW_CO2 H2O_rho_D130=Density(H2O,T=T_30,P=12.5)*MW_H2O V_D130=H2S_30/H2S_rho_D130+N2_30/N2_rho_D130+CO2_30/CO2_rho_D130+H2O_30/H 2O_rho_D130+MEA_30/1010+H2O_33/H2O_rho_D130+MEA_33/1010

Output:

absorb eff=0.999 aquFORmea_33=7.847 aquFORmea_34=78.47 Ash 1=6.143 Ash 42=0.1464 Ash 9=6.143 Biochar_37=1.253 Biochar 38=0.1253 Biochar_39=1.128 Biochar 40=0.01253 Biochar_41=0.1128 Biochar 42=1.241 Biogas 37=1.358 Biogas_43=1.358 Biogas_45=1.358 Biogas 48=1.358 Biogas_51=1.358 Biogas_56=1.085 Biogas_57=0.2722 Biomass 36=9.399 Biomass 4=9.399 Biooil 37=7.833 Biooil 43=2.35 Biooil_44=5.483 Biooil 45=0.705 Biooil 46=1.645 Biooil_47=7.128

Biooil 48=0.03525 Biooil_49=0.6697 Biooil 50=7.798 Biooil 52=0.03349 Biooil_53=7.831 Biototal 36=10.44 CH4_11=0.1529 CH4 12=0.1529 CH4_15=0.1529 CH4 25=0.1529 CH4 29=0.1529 CH4_rho=0.322 CH4 rho D125=5.569 CH4_rho_D128=7.867 CH4_rho_H119=16.19 CH4 rho N155=15.64 CO2_11=35.33 CO2 12=35.33 CO2_15=35.33 CO2 25=35.33 CO2_30=35.29 CO2 31=35.29 CO2 rho=0.322 CO2_rho_D125=15.31 CO2 rho D128=21.63 CO2 rho D130=21.91 CO2_rho_H119=44.52

CO2_rho_N155=43
Coal 1=52.06
CO_11=68.51
CO_12=68.51
CO_15=68.51
CO_25=68.51
CO_29=68.51
CO_rho=0.322
CO_rho_D125=9.745
CO_rho_D128=13.77
CO_rho_H119=28.33
CO_rho_N155=27.37
cph=75.3
C_1=36.69
C 42=0.8745
C_9=36.69
Gas_12=131.5
H2O_10=6.571
H2O_11=19.96
H2O_12=19.96
H2O_14=22.19
H2O_15=19.96
H2O_16=22.19
H2O_17=21.56
H2O 18=0.6277
H2O_19=0.06277
H2O_2=35.23
H2O_20=0.5649
H2O_20=0.3049 H2O_21=22.13
H2O_22=41.93
H2O_23=19.74
H2O_24=42.28
H2O_25=0.1583
H2O_26=115.9
H2O_27=96.07
H2O_28=115.7
H2O 30=78.62
H2O 31=78.62
H2O 32=86.31
H2O_33=7.847
H2O_34=78.47
H2O_35=0.03857
H2O_36=2.13
H2O_7=15.65
H2O_9=35.23
H2O_Biomass_36=1.044

H2O_Biomass_4=9.399 H2O_factor=0.00793 H2O_rho=0.322 H2O_rho_D125=6.265 H2O_rho_D128=8.85 H2O_rho_D130=8.962 H2O_rho_H119=18.21 H2O_rho_N155=17.59 H2S_11=2.08 H2S_12=2.08 H2S_15=2.08 H2S_25=2.08 H2S_30=2.078 H2S_31=2.078 H2S_rho=0.322 H2S_rho_D125=11.83 H2S_rho_D128=16.72 H2S_rho_D130=16.93 H2S_rho_H119=34.4 H2S_rho_N155=33.23 H2_11=3.645 H2 12=3.645 H2_15=3.645 H2_25=3.645 H2_29=3.645 H2_rho=0.322 H2_rho_D125=0.6961 H2_rho_D128=0.9834 H2_rho_H119=2.024 H2 rho N155=1.955 hh=-285830 H 1=2.525 H_22=21963 H_42=0.06018 H_47=11144 H_50=11884 H_53=11921 H_9=2.525 H_B143=2521 h CH4 11=-713 h_CO2_11=-315930 h_CO2_31=3915 h_combustion=54381 h CO 11=-270456 h_C_42=795.7 H_D120=-11606

H D125=-24137
H_D130=227926
H_E117=12723
H E118=-273477
H_E123=8822
H_E127=24137
H_E129=29108
H_E133=264.7
H_E148=1295
H_E150=527
H_G153=0.04109
h_H2O_11=-268112
h_H2O_31=191434
h_H2O_32=28245
h_H2O_9=140.1
h_H2O_9_at=1960
h_H2O_9_ref=1889
h_H2S_11=-1260
h_H2S_31=254
h_H2_11=0
h_H_42=656.5
h_MEA_32=3863
h_N2_11=0
h_N2_31=214.9
h_N_42=13.54
h_O2_3=-7827
h_O2_3_at=-4960
h_O2_3_ref=0
h_O_42=67.72
h_products=323866
H_R140=7026
h_reactants=-6137
—
h_reaction= -351349
h_S_42=16.73
lh_H2O_11=33620
mass_CH4_11=0.024
mass_CO2_11=5.544
mass_CO_11=10.75
mass_H2O_11=3.132
mass_H2S_11=0.3264
mass_H2_11=0.572
mass_N2_11=0.28
mass_O2_11=0.224
mass_Total_11=20.85
MEAaqu_33=9.808
MEAaqu_33=98.08
1111/10/00

MEA_30=19.62 MEA_32=21.58 MEA_33=1.962 MEA_34=19.62 mol%_CH4_11=0.0015 mol%_CO2_11=0.126 mol%_CO_11=0.384 mol%_H2O_11=0.174 mol%_H2S_11=0.0096 mol%_H2_11=0.286 mol%_N2_11=0.01 mol%_O2_11=0.007 mol_Biogas_57=0.01512 mol_CH4_11=0.009558 mol_CH4_15=0.009558 mol_CH4_25=0.009558 mol CO2 11=0.8029 mol_CO2_15=0.8029 mol_CO2_25=0.8029 mol_CO2_30=0.8021 mol_CO2_31=0.8021 mol CO 11=2.447 mol_CO_15=2.447 mol_CO_25=2.447 mol_C_42=0.07287 mol_C_9=3.058 mol_H2O_10=0.365 mol_H2O_11=1.109 mol_H2O_14=1.233 mol H2O 15=1.109 mol_H2O_16=1.233 mol_H2O_21=1.229 mol_H2O_22=2.329 mol_H2O_24=2.349 mol_H2O_25=0.008793 mol_H2O_26=6.437 mol_H2O_2627=1.1 mol_H2O_27=5.337 mol_H2O_28=6.426 mol H2O 30=4.368 mol_H2O_31=4.368 mol H2O 32=4.795 mol_H2O_33=0.4359 mol H2O 34=4.359 mol_H2O_35=0.6942 mol_H2O_7=0.8695

mol_H2O_9=1.957 mol_H2S_11=0.06117 mol_H2S_15=0.06117 mol_H2S_25=0.06117 mol H2S 30=0.06111 mol_H2S_31=0.06111 mol H2 11=1.822 mol_H2_15=1.822 mol_H2_25=1.822 mol H 42=0.06018 mol_H_9=2.525 mol_MEA_30=0.3212 mol_MEA_32=0.3533 mol_MEA_33=0.03212 mol_MEA_34=0.3212 mol_N2_11=0.06372 mol N2 15=0.06372 mol_N2_25=0.06372 mol_N2_30=0.06366 mol_N2_31=0.06366 mol_N_42=0.001241 mol N 9=0.05206 mol_O2_3=1.578 mol_O_42=0.006072 mol_O_9=0.2548 $mol_S_{42}=0.001411$ mol_S_9=0.05922 $MW_C=12$ MW_CH4=16 MW CO=28 MW CO2=44 MW H=1 $MW_H2=2$ MW_H2O=18 MW H2S=34 MW N=14 MW_N2=28 $MW_O=16$ MW_02=32 MW S=32 M_N155=132.1 N2 11=1.784 N2_12=1.784 N2 15=1.784 N2_25=1.784 N2_30=1.782

N2_31=1.782 N2_rho=0.322 N2 rho D125=9.745 N2_rho_D128=13.77 N2_rho_D130=13.94 N2_rho_H119=28.33 N2 rho N155=27.37 N_1=0.7288 N 42=0.01737 N 9=0.7288 O2_11=1.427 O2 3=50.5 O 1=4.076 O_42=0.09715 O 9=4.076 Product_11=132.9 P 12=14 P_3=16 P_9=44 P_H2O_sat=0.09516 P ref=1 Q E117=10755 rho_N155=20.44 sh=69.94 Slag_11=6.289 Slag_12=0.6289 Slag_13=5.66 Slag_15=0.0006289 Slag_16=0.6283 Slag 17=0.0006283 Slag_18=0.6277 Slag 19=0.6271 Slag_20=0.0006277 Slag_25=0.0006289 S 1=1.895 S 42=0.04516 S 9=1.895 T_11_1=1644 T_11_2=523 T 12=523 T_14=397.6 T 15=466.5 T_16=428.1 T 22=397.6 T_24=348 T 25=318

T_26=363 T_27=363 T_28=348	T_53=625.2 T_57=453.6 T_7_1=298
T 29=311	T 7 2=1023
T 3=133	T 9=298.2
T 30=302	$T_{29290.2}$ T ref=298.2
T 30 1=373	Vol D120=22.21
T 31=488	Vol Gas 12=22.19
T 32=376	Vol_H2O_14=0.02219
T_33=376	V_D125=18.15
T_33_2=383	V_D128=10.61
T_34=298.2	V_D130=11.53
T_34_1=376	V_H119=6.244
T_35_1=388	V_N155=6.462
T_35_2=383	wt%_Ash_1=0.118
T_36=348	wt%_CH4_11=0.001151
T_37=823	wt%_CO2_11=0.2658
T_40=823	wt%_CO_11=0.5156
T_43=669.5	wt%_C_1=0.7048
T_44=669.5	wt%_H2O_11=0.1502
T_45=550.6	wt%_H2S_11=0.01565
T_46=550.6	wt%_H2_11=0.02743
T_47=642.1	wt%_H_1=0.0485
T_48=453.6	wt%_N2_11=0.01343
T_49=453.6	wt%_N_1=0.014
T_50=625.9	wt%_O2_11=0.01074
T_51=453.6	wt%_O_1=0.0783
T_52=453.6	

SIZING - EES

Input:

"-----M-115-----" rho_coal=730 "kg/m3" rho_water=1000 "kg/m3" t_M115=2*3600 V_coal_M115=52.04*t_M115/rho_coal V_water_M115=35.23*t_M115/rho_water V_M115=V_coal_M115+V_water_M115 P_M115=V_M115^0.8 ratio=P_M115/V_M115

"------L-124------"

q_water_L124=22.19/rho_water dP_L124=150 "kPa" epsilon_i_L124=0.85 P_L124=(q_water_L124*dP_L124)/epsilon_i_L124

```
"------L-126------"
```

```
q_water_L126=96.07/rho_water
dP_L126=50 "kPa"
epsilon_i_L126=0.85
P_L126=(q_water_L126*dP_L126)/epsilon_i_L126
```

"------L-131------"

rho_MEA=1012 "kg/m3" q_L131=(78.47/rho_water)+(19.61/rho_MEA) dP_L131=100 "kPa" epsilon_i_L131=0.85 P_L131=(q_L131*dP_L131)/epsilon_i_L131

"------L-132------"

q_L132=(7.847/rho_water)+(1.961/rho_MEA) dP_L132=50 "kPa" epsilon_i_L132=0.85 P_L132=(q_L132*dP_L132)/epsilon_i_L132

"-----L-149------"

rho_biooil=4.55/0.00378541 "kg/m3" q_L149=5.53/rho_biooil dP_L149=50 "kPa" epsilon_i_L149=0.85 P L149=(q L149*dP L149)/epsilon i L149

"------L-116------"

q_L116=(52.04/rho_coal)+(35.23/rho_water) dP_L116=2800 "kPa" epsilon_i_L116=0.85 P_L116=(q_L116*dP_L116)/epsilon_i_L116

"-----E-117------"

Q_E117=10753*1000 "J/s" U_E117=1300 "J/m2-s-K" T_O2_E117=T_sat(Oxygen,P=16) T_H2O_E117=T_sat(Water,P=16) A_E117=Q_E117/(U_E117*(T_H2O_E117-T_O2_E117))

"------E-123------"

Q_E123=8821*1000 "J/s" U_E123=950 "J/m2-s-K" T1_E123=397.6 T2_E123=348 T3_E123=303 T4_E123=298 dTh_E123=T1_E123-T2_E123 dTc_E123=T3_E123-T4_E123 MTD_E123=(dTh_E123-dTc_E123)/ln(dTh_E123/dTc_E123) A_E123=Q_E123/(U_E123*MTD_E123)

"------E-127------"

Q_E127=24135*1000 "J/s" U_E127=950 "J/m2-s-K" T1_E127=363 T2_E127=348 T3_E127=303 T4_E127=298 dTh_E127=T1_E127-T2_E127 dTc_E127=T3_E127-T4_E127 MTD_E127=(dTh_E127-dTc_E127)/ln(dTh_E127/dTc_E127) A_E127=Q_E127/(U_E127*MTD_E127)

"-----E-129------"

Q_E129=29105*1000 "J/s" U_E129=950 "J/m2-s-K" T1_E129=376 T2_E129=373 T3_E129=304.9 T4_E129=298 dTh_E129=T1_E129-T2_E129 dTc_E129=T3_E129-T4_E129 MTD_E129=(dTh_E129-dTc_E129)/ln(dTh_E129/dTc_E129) A_E129=Q_E129/(U_E129*MTD_E129)

"-----E-133------"

Q_E133=264.4*1000 "J/s" U_E133=950 "J/m2-s-K" T1_E133=388 T2_E133=383 T3_E133=383 T4_E133=376 dTh_E133=T1_E133-T2_E133 dTc_E133=T3_E133-T4_E133 MTD_E133=(dTh_E133-dTc_E133)/ln(dTh_E133/dTc_E133) A_E133=Q_E133/(U_E133*MTD_E133)

"------E-148------"

Q_E148=1306*1000 "J/s" U_E148=700 "J/m2-s-K" T1_E148=823 T2_E148=669.5 A_E148=Q_E148/(U_E148*(T1_E148-T2_E148))

"-----E-150------"

Q_E150=531.5*1000 "J/s" U_E150=700 "J/m2-s-K" T1_E150=669.5 T2_E150=550.6 A_E150=Q_E150/(U_E150*(T1_E150-T2_E150))

"------E-118------"

Q_E118=273477*1000 "J/s" U_E118=650 "J/m2-s-K" T1_E118=1644 T2_E118=1023 T3_E118=523 T4_E118=298 dTh_E118=T1_E118-T2_E118 dTc_E118=T3_E118-T4_E118 MTD_E118=(dTh_E118-dTc_E118)/ln(dTh_E118/dTc_E118) A_E118=Q_E118/(U_E118*MTD_E118) "-----G-153------"

Biogas_G153=0.2745 "kg/s" H_G153=0.04109 "kJ/s" epsilon_i_G153=0.85 P_G153=(Biogas_G153*H_G153)/epsilon_i_G153

"-----B-143------"

Q_B143=2521 "kJ/s" U_B143=60*(A_B143^0.67) T1_B143=453.6 T2_B143=348 T3_B143=348 T4_B143=298 dTh_B143=T1_B143-T2_B143 dTc_B143=T3_B143-T4_B143 MTD_B143=(dTh_B143-dTc_B143)/ln(dTh_B143/dTc_B143) A_B143=Q_B143/(U_B143*MTD_B143) A_B143=(((D_B143/2)^2)*3.14) L_B143=D_B143*7 V_B143=A_B143*L_B143

"-----"

epsilon_N155=.75 M_N155=133.5 "kg/s" P1_N155=4250 "kPa" P2_N155=1400 "kPa" rho_N155=20.44 "kg/m3" P_N155=epsilon_N155*M_N155*(P1_N155-P2_N155)/rho_N155

"------H-147-------" rho_biogas_H147=density(steam,T=823,P=15.5) q_H147=7.899/rho_biooil+1.369/rho_biogas_H147

"------"

rho_slag=2114 "kg/m3" t_H121=2*3600 V_H121=((22.19*t_H121)/rho_water)+(0.6283*t_H121/rho_slag) V_H121=A_H121*L_H121 L_H121=3*D_H121 A_H121=(((D_H121/2)^2)*3.14)

"-----H-122------" P_H122=(5*(10^(-6)))*(rho_slag^2)*4.25/(100^0.3)

"-----H-152------" rho_biogas_H152=density(steam,T=823,P=13.5) q_H152=0.03555/rho_biooil+1.369/rho_biogas_H152

"------H-154------" q_H154=7.898/rho_biooil

"------H-151-------"

rho_biogas_H151=density(steam,T=823,P=14) g H151=0.7109/rho biooil+1.369/rho biogas H151 "-----C-113------" P_C113=0.5*(53.04^(0.88))*8 "-----C-114------" P_C114=0.3*53.04*15 "-----C-144------" P C144=0.3*11.63*15 "-----J-111------" P_J111=0.0027*(53.04^0.82)*90 "-----J-141------" P_J141=0.0027*(18.96^0.82)*90 "-----J-145------" P_J145=0.07*(11.63^0.85)*25 "-----F-112------" t F112=8*3600 V F112=(52.04*t F112)/rho coal "-----F-142------" rho biomass=365 "kg/m3" t F142=8*3600 V_F142=((9.479*t_F142)/rho_water)+((9.479*t_F142)/rho_biomass) "-----D-125------" t D125=10 V D125=18.15*t D125 V_D125=A_D125*L_D125 L_D125=5*D_D125 A D125=(((D D125/2)^2)*3.14) "------D-128------" t D128=10 V_D128=10.61*t_D128 V_D128=A_D128*L_D128 L_D128=5*D_D128 A_D128=(((D_D128/2)^2)*3.14) "------D-130------" t D130=10 V D130=11.53*t D130 V D130=A D130*L D130 L_D130=5*D_D130 A_D130=(((D_D130/2)^2)*3.14) Output: A_B143=0.7104 A D130=7.474 A_D125=10.11 A_E117=23.88

A_D128=7.071

A_E118=1079

A_E123=477.7
A_E127=2791
A_E129=6543
A_E133=46.82
A_E148=12.15
A E150=6.386
A_H121=13.17
Biogas_G153=0.2745
dP_L116=2800
dP_L124=150
dP_L126=50
dP_L131=100
dP L132=50
dP_L149=50
dTc_B143=50
dTc E118=225
dTc_E123=5
dTc_E127=5
dTc_E129=6.9
dTc_E133=7
dTh_B143=105.6
dTh_E118=621
dTh E123=49.6
dTh_E127=15
dTh_E129=3
dTh E133=5
D B143=0.9513
D_D125=3.589
D_D128=3.001
D_D130=3.086
D_H121=4.097
epsilon_i_G153=0.85
epsilon_i_L116=0.85
epsilon_i_L124=0.85
epsilon_i_L126=0.85
epsilon_i_L131=0.85
epsilon_i_L132=0.85
epsilon_i_L149=0.85
epsilon_N155=0.75
1
H_G153=0.04109
L_B143=6.659
L_D125=17.95
L_D128=15.01
L_D130=15.43
L_H121=12.29
MTD_B143=74.37
—

MTD_E118=390.1 MTD_E123=19.44 MTD_E127=9.102 MTD_E129=4.682 MTD_E133=5.944 M_N155=133.5 P1 N155=4250 P2_N155=1400 P_C113=131.7 P_C114=238.7 P_C144=52.34 P_G153=0.01327 P_H122=23.85 P_J111=6.306 P_J141=2.713 P_J145=14.09 P_L116=350.9 P_L124=3.916 P_L126=5.651 P_L131=11.51 P_L132=0.5756 P L149=0.2706 P_M115=203.1 P_N155=13961 Q_B143=2521 Q_E117=1.075E+07 Q_E118=2.735E+08 Q_E123=8.821E+06 Q_E127=2.414E+07 Q_E129=2.911E+07 Q_E133=264400 Q_E148=1.306E+06 Q_E150=531500 q_H146=0.3285 q_H147=0.3389 q_H151=0.3689 q_H152=0.3821 q_H154=0.006571 q_L116=0.1065 q_L131=0.09785 q_L132=0.009785 q_L149=0.004601 q_water_L124=0.02219 q_water_L126=0.09607 ratio=0.2649 rho_biogas_H146=4.253

T4_E118=298
T4_E123=298
T4_E127=298
T4_E129=298
T4_E133=376
t_D125=10
t_D128=10
t_D130=10
t_F112=28800
t_F142=28800
t_H121=7200
T_H2O_E117=474.6
t_M115=7200
T_O2_E117=128.2
U_B143=47.72
U_E117=1300
U_E118=650
U_E123=950
U_E127=950
U_E129=950
U_E133=950
U_E148=700
U_E150=700
V_B143=4.731
V_coal_M115=513.3
V_D125=181.5
V_D128=106.1
V_D130=115.3
V_F112=2053
V_F142=1021
V_H121=161.9
V_M115=766.9
V_water_M115=253.7

CAPITAL COST ESTIMATION - ECONEXPERT

An Expert System for Capital Cost Estimation Developed by P.T. Vasudevan and T. Ulrich © Copyright 2000, All rights reserved. DISCLAIMER: We accept no liability for potential errors in the program beyond refunding the fee if a subscriber cancels within thirty days.

Cost Summary

The cost index is 595.0

Mixers : Agitators and Inline Mixers : Agitator open tank

Total purchased cost = \$231868The bare module cost is = \$463737

Pumps : Centrifugal Total purchased cost = \$ 8337 Material factor = 1.00 Pressure factor = 1.09 The bare module cost (incl. electric motor drive) is = \$ 29968

Pumps : Centrifugal Total purchased cost = \$ 10320 Material factor = 1.00 Pressure factor = 1.11 The bare module cost (incl. electric motor drive) is = \$ 37488

Pumps : Centrifugal Total purchased cost = \$ 13671 Material factor = 1.00 Pressure factor = 1.06 The bare module cost (incl. electric motor drive) is = \$ 48607

Pumps : Centrifugal Total purchased cost = \$ 4733 Material factor = 1.00 Pressure factor = 1.06 The bare module cost (incl. electric motor drive) is = \$ 16829

Pumps : Centrifugal Total purchased cost = \$ 3846 Material factor = 1.00 Pressure factor = 1.18 The bare module cost (incl. electric motor drive) is = \$ 14385

Pumps : Reciprocating Purchased cost for one unit = \$ 339031 Total purchased cost = \$ 678063 Material factor = 1.00 Pressure factor = 1.15 The bare module cost (incl. electric motor drive) is = \$ 2504289

Heat Exchangers : Shell and Tube : Double pipe (multiple pipe for areas > 50 sq.m.) Total purchased cost = \$ 15323 Material factor = 1.00 Pressure factor = 1.00 The bare module cost is = \$ 48730 Heat Exchangers : Shell and Tube : Double pipe (multiple pipe for areas > 50 sq.m.) Total purchased cost = \$ 241627 Material factor = 1.00 Pressure factor = 1.00 The bare module cost is = \$ 768396

Heat Exchangers : Shell and Tube : Double pipe (multiple pipe for areas > 50 sq.m.) Purchased cost for one unit = \$281506Total purchased cost = \$1407529Material factor = 1.00 Pressure factor = 1.00 The bare module cost is = \$4476083

Heat Exchangers : Shell and Tube : Double pipe (multiple pipe for areas > 50 sq.m.) Purchased cost for one unit = \$ 299609 Total purchased cost = \$ 3295698 Material factor = 1.00 Pressure factor = 1.00 The bare module cost is = \$ 10480648

Heat Exchangers : Shell and Tube : Double pipe (multiple pipe for areas > 50 sq.m.) Total purchased cost = \$ 25102 Material factor = 1.00 Pressure factor = 1.00 The bare module cost is = \$ 79827

Heat Exchangers : Shell and Tube : Double pipe (multiple pipe for areas > 50 sq.m.) Total purchased cost = \$ 9945 Material factor = 1.00 Pressure factor = 1.00 The bare module cost is = \$ 31625

Heat Exchangers : Shell and Tube : Double pipe (multiple pipe for areas > 50 sq.m.) Total purchased cost = \$ 6838 Material factor = 1.00 Pressure factor = 1.00 The bare module cost is = \$ 21745

Heat Exchangers : Shell and Tube : Fixed tube sheet and U-tube Purchased cost for one unit = \$ 56252 Total purchased cost = \$ 112505 Material factor = 1.00 Pressure factor = 1.05 The bare module cost is = \$ 365028

Gas Movers and Compressors : Blowers and compressors (cost of drive excluded) : Axial

Total purchased cost = \$23364The bare module cost is = \$81775

Gas-Solid Contacting Equipment : Rotary and vertical tower contactors (incl. drive) : Rotary dryer (direct) Total purchased cost = \$ 90770 The bare module cost is = \$ 208771

Drives and Power Recovery Machines : Power recovery machines Purchased cost for one unit = Total purchased cost = The bare module cost is =

Separators : Dust collectors : Gas multi-cyclone Total purchased cost = \$13448The bare module cost is = \$40343

Separators : Dust collectors : Gas multi-cyclone Total purchased cost = \$ 3138 The bare module cost is = \$ 9415

Separators : Dust collectors : Gas multi-cyclone Total purchased cost = \$3167The bare module cost is = \$9502

Separators : Liquid filters : Single-compartment drum (pressure) Total purchased cost = \$ 293802 The bare module cost is = \$ 705125

Separators : Vibratory screens Total purchased cost = \$4958The bare module cost is = \$13882

Separators : Dust collectors : Gas bag filters Total purchased cost = \$44282The bare module cost is = \$97421

Separators : Dust collectors : Gas bag filters Total purchased cost = \$ 44282 The bare module cost is = \$ 97421

Separators : Dust collectors : Electrostatic precipitators (wet) Total purchased cost = \$454900The bare module cost is = \$1046271

Crushers : Jaw

Total purchased cost = \$269188The bare module cost (incl. electric motor drive) is = \$565294

Grinders : Rolling compression (bowl, pan, ring-roll) Purchased cost for one unit = \$ 470548 Total purchased cost = \$ 1411644 The bare module cost (incl. electric motor drive) is = \$ 3952603

Grinders : Rolling compression (bowl, pan, ring-roll) Total purchased cost = \$ 363767 The bare module cost (incl. electric motor drive) is = \$ 1018549

Conveyors : Belt Total purchased cost = \$73572The bare module cost is = \$176574

Conveyors : Belt Total purchased cost = \$73572The bare module cost is = \$176574

Conveyors : Auger Total purchased cost = \$ 13814 The bare module cost is = \$ 30391

Storage Vessels : Atmospheric pressure-Bins Total purchased cost = \$9375The bare module cost is = \$19688

Storage Vessels : Atmospheric pressure-Bins Total purchased cost = \$5826The bare module cost is = \$12235

Furnaces : Process heaters : Reactive - Pyrolysis furnace Purchased cost for one unit = \$ 3861918 Total purchased cost = \$ 19309589 Pressure factor = 1.03 The bare module cost is = \$ 53905852

Furnaces : Process heaters : Reactive - Pyrolysis furnace Total purchased cost = \$720158Pressure factor = 0.99 The bare module cost is = \$1921394

Separators : Dust collectors : Venturi scrubber Total purchased cost = \$170011The bare module cost is = \$425027 Process Vessels (including towers) : Vertically oriented : No packing or trays Total purchased cost = \$ 151290 Material factor = 1.00 Pressure factor = 1.70 The bare module cost is = \$ 830690

Process Vessels (including towers) : Vertically oriented : No packing or trays Total purchased cost = \$ 113818 Material factor = 1.00 Pressure factor = 1.67 The bare module cost is = \$ 617895

Process Vessels (including towers) : Vertically oriented : No packing or trays Total purchased cost = \$ 119038 Material factor = 1.00 Pressure factor = 1.63 The bare module cost is = \$ 638680

Total Bare Module Cost = \$93713986Contingency and Fee = \$16868517Total Module Cost = \$110582503Auxiliary Facilities = \$33174751Grass Roots Capital = \$143757255

CASH FLOW ANALYSIS - EXCEL

					Expenses excluding Depreciation						
	Annual Capital	Sales	Expenses	Depreciation	Ame+Age	Cash Income	Allowance	ANP	Income Tax	ANNP	Cash Income
								Profit BT		Profit after tax	
1	(\$43,000,000)	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	(\$43,000,000)
2	(\$43,000,000)	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	(\$43,000,000)
3	(\$86,000,000)	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	(\$86,000,000)
4	\$0	\$140,800,000	\$141,000,000	\$14,000,000	\$127,000,000	\$13,800,000	\$0	(\$200,000)	(\$70,000)	(\$130,000)	\$13,870,000
5	\$0	\$176,000,000	\$141,000,000	\$14,000,000	\$127,000,000	\$49,000,000	\$0	\$35,000,000	\$12,250,000	\$22,750,000	\$36,750,000
6	\$0	\$176,000,000	\$141,000,000	\$14,000,000	\$127,000,000	\$49,000,000	\$0	\$35,000,000	\$12,250,000	\$22,750,000	\$36,750,000
7	\$0	\$176,000,000	\$141,000,000	\$14,000,000	\$127,000,000	\$49,000,000	\$0	\$35,000,000	\$12,250,000	\$22,750,000	\$36,750,000
8	\$0	\$176,000,000	\$141,000,000	\$14,000,000	\$127,000,000	\$49,000,000	\$0	\$35,000,000	\$12,250,000	\$22,750,000	\$36,750,000
9	\$0	\$176,000,000	\$141,000,000	\$14,000,000	\$127,000,000	\$49,000,000	\$0	\$35,000,000	\$12,250,000	\$22,750,000	\$36,750,000
10	\$0	\$176,000,000	\$141,000,000	\$14,000,000	\$127,000,000	\$49,000,000	\$0	\$35,000,000	\$12,250,000	\$22,750,000	\$36,750,000
11	\$0	\$176,000,000	\$141,000,000	\$14,000,000	\$127,000,000	\$49,000,000	\$0	\$35,000,000	\$12,250,000	\$22,750,000	\$36,750,000
12	\$0	\$176,000,000	\$141,000,000	\$14,000,000	\$127,000,000	\$49,000,000	\$0	\$35,000,000	\$12,250,000	\$22,750,000	\$36,750,000
13	\$21,500,000	\$176,000,000	\$141,000,000	\$14,000,000	\$127,000,000	\$49,000,000	\$0	\$35,000,000	\$12,250,000	\$22,750,000	\$58,250,000
											12.59%

IRR

Net Present Value								
	0% rate		10% rate		13%			
0	0	0	0	0	0	0%	0	0
1	(\$43,000,000)	(43,000,000)	1	(39,090,909)	(39,090,909)	1	(38,189,996)	(38,189,996)
2	(\$43,000,000)	(86,000,000)	1	(35,537,190)	(74,628,099)	1	(33,918,042)	(72,108,038)
3	(\$86,000,000)	(172,000,000)	1	(64,613,073)	(139,241,172)	1	(60,247,902)	(132,355,940)
4	13,870,000	(158,130,000)	1	9,473,397	(129,767,775)	1	8,629,807	(123,726,133)
5	36,750,000	(121,380,000)	1	22,818,859	(106,948,917)	1	20,307,812	(103,418,321)
6	36,750,000	(84,630,000)	1	20,744,417	(86,204,500)	0	18,036,169	(85,382,152)
7	36,750,000	(47,880,000)	1	18,858,561	(67,345,939)	0	16,018,633	(69,363,518)
8	36,750,000	(11,130,000)	0	17,144,146	(50,201,793)	0	14,226,780	(55,136,739)
9	36,750,000	25,620,000	0	15,585,587	(34,616,205)	0	12,635,364	(42,501,374)
10	36,750,000	62,370,000	0	14,168,716	(20,447,489)	0	11,221,966	(31,279,409)
11	36,750,000	99,120,000	0	12,880,651	(7,566,839)	0	9,966,670	(21,312,738)
12	36,750,000	135,870,000	0	11,709,683	4,142,844	0	8,851,793	(12,460,945)
13	58,250,000	194,120,000	0	16,872,950	21,015,794	0	12,460,945	(0)
	\$194,120,000.00			\$21,015,794.03			(\$0.00)	