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# Review of Experimental Capabilities and Hydrodynamic Data for Validation of CFD Based Predictions for Slurry Bubble Column Reactors

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### REVIEW OF EXPERIMENTAL CAPABILITIES AND HYDRODYNAMIC DATA FOR VALIDATION OF CFD BASED PREDICTIONS FOR SLURRY BUBBLE COLUMN REACTORS

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#### Abstract

The purpose of this paper is to document the review of several open-literature sources of both experimental capabilities and published hydrodynamic data to aid in the validation of a Computational Fluid Dynamics (CFD) based model of a slurry bubble column (SBC). The review included searching the Web of Science, ISI Proceedings, and Inspec databases, internet searches as well as other open literature sources. The goal of this study was to identify available experimental facilities and relevant data. Integral (i.e., pertaining to the SBC system), as well as fundamental (i.e., separate effects are considered), data are included in the scope of this effort. The fundamental data is needed to validate the individual mechanistic models or closure laws used in a Computational Multiphase Fluid Dynamics (CMFD) simulation of a SBC. The fundamental data is generally focused on simple geometries (i.e., flow between parallel plates or cylindrical pipes) or custom-designed tests to focus on selected interfacial phenomena. Integral data covers the operation of a SBC as a system with coupled effects. This work highlights selected experimental capabilities and data for the purpose of SBC model validation, and is not meant to be an exhaustive summary.

#### Introduction

Slurry Bubble Column Reactors (SBCRs) are used by industry to manufacture liquid hydrocarbon fuels (i.e., diesel or gasoline) via the Fischer Tropsch (FT) process (Figure 1). In the FT process, a synthesis gas comprised of hydrogen and carbon monoxide is sparged through a distributor into a suspension of liquid and solid catalyst particles. Generally, in order to be economically viable, a bubble/slurry bubble column reactor must be operated at high volumetric flow rates. This requires a very active catalyst, high catalyst loading of the slurry, and high gas conversion rates. To achieve a complete catalyst suspension and the desired product reaction, these reactors need to be



**Figure 1**. Churn-turbulent threephase flow in a SBCR.

operated at a high superficial gas velocity in the churn-turbulent flow regime. At these conditions, the two-phase interfacial dynamics dominates the reactor hydrodynamics. Due to the radial gradient of the buoyancy force resulting from non-uniform lateral gas holdup, liquid or slurry recirculation is induced where liquid moves upward in the central region of the bubble column and downward near the wall region of the column. In the case of a SBC, the catalyst particles are very small, hence they closely follow the motion of the liquid flow. The key to successfully modeling this process lies in accurate predictions of the heat and mass transfer along with turbulent mixing, which affects kinetics and thus product yield and selectivity.

Although the FT reaction process was developed in the early 1900's [1], details concerning the hydrodynamic processes, which control the reactor flow, are still poorly understood. For example, significant progress is clearly needed to better understand the unsteady, multiphase fluid dynamics that controls the fluid mixing and interphase transport processes, which in turn determine the overall reactor performance. Due to the complexity of the reaction and hydrodynamic processes occurring in the SBCR, the system performance has traditionally been characterized empirically, rather than from a fundamental physical basis. Since empirical correlations are generally valid only for the (typically narrow) parameter ranges over which they are generated, mechanistic models of the sub-processes occurring in a SBCR as necessary to optimize the overall process and scale laboratory data to industrial applications.

This paper provides a limited summary of institutions with existing experimental facilities capable of providing validation data and sources of validation data published in the open literature. An emphasis is placed on data for churn-turbulent flows, since the hydrodynamics of the flow, as well as the flow mechanisms, change significantly from one flow regime to another [2]. The scope of this effort is limited to hydrodynamics only, without reactions occurring. Once the computational model is validated using suitable hydrodynamic data, reaction kinetics will be incorporated.

#### Hydrodynamic Data in the Open Literature

The Idaho National Laboratory and Rensselaer Polytechnic Institute (RPI) have embarked on a joint effort to develop scientific and technological advances for the design and development of next generation FT reactors. This goal will be achieved by employing state-of-the-art modeling and computational concepts of multidimensional, multiphase reacting flows in complex geometries and linking together multiple scales to upgrade multi-scale simulation capabilities of the computational multiphase fluid dynamic (CMFD) code, NPHASE [3] under development at RPI. An ensembleaveraged, multifield, mechanistic model formulation will be used to develop closure relations for use with the NPHASE software. Once validated and fully functional, the FT SBCR model will be used as a numerical test bed to virtually simulate new concepts and ideas to improve the process.

The complexity in the design of gas-liquid-solid systems lies in the existence of the three phases with mass, momentum and energy transfer occurring between them. The

interfacial structure between the liquid and the gas phase can be considerably different depending upon system configuration and operating conditions. Due to complicated interaction of phases (particularly in churn-turbulent flow regime), the hydrodynamics is not yet fully understood and hence, the reactor design and scale-up are a challenging task. The use of CMFD, rather than empirical correlations, is in principle applicable to a wider range of conditions and design configurations. However, this procedure calls for a solution of the coupled continuity, momentum and energy equations for two fluid phases and a dispersed solid phase. For problems of practical utility, the ensemble-averaged form of the Navier-Stokes equations are solved. This approach necessitates the use of constitutive equations (i.e., closure models) to re-introduce information that was lost through the averaging procedure. Mechanistic-based closure relations have been and are continuing to be developed [4], but require validation with experimental data obtained at conditions applicable to the FT process. To validate a CMFD based model of a SBCR, the important interfacial mass, momentum and energy transfer processes must be identified. For the SBCR under consideration, the transfer processes listed in Table 1 are considered dominant and should be validated with data.

Table 1. Important mass, momentum and energy processes in a SDC			
Mass Transfer	Momentum Transfer	Energy Transfer	
Bubble Coalescence	<ul> <li>Interfacial Drag Force</li> </ul>	Reaction Kinetics	
- small bubble interaction	– bubble drag	• Tube Bank Heat Transfer	
<ul> <li>– cap/slug bubble interaction</li> </ul>	– cap/slug drag	Interfacial Heat Transfer	
Bubble Breakup	– particle drag		
<ul> <li>– cap/slug bubble breakup</li> </ul>	<ul> <li>Interfacial Lift Force</li> </ul>		
Flow Topology Transition	Interfacial Virtual Mass Force		
- interfacial area density	<ul> <li>Interfacial Wall Force</li> </ul>		
transport	Interfacial Turbulence		
	Dispersion Force		
	Liquid Turbulence		
	- shear induced turbulence		
	<ul> <li>bubble induced turbulence</li> </ul>		

**Table 1.** Important mass, momentum and energy processes in a SBCR

The purpose of this review is to identify available experimental facilities, techniques and data to validate CMFD models of gas-liquid-solid flows through SBCRs. The following sections will review results from our database and internet searches. In this paper, we refer to "integral" experiments as pertaining to a system, whereas "fundamental" experiments provide the capability to isolate separate effects occurring within a system.

#### **Experimental Facilities, Techniques and Methods Available**

The experimental facilities, techniques and methods listed cover both integral and fundamental testing. Capabilities span the scale from bench-top or laboratory SBCs to pilot or production scale slurry bubble column reactors. Available integral test facilities and data available in the open literature are shown in Table 2. Testing performed at these facilities has provided gas, liquid, and solid phase velocity field data; turbulence parameters (turbulent dispersion coefficients, Reynolds stresses); effect of process variables and component properties on gas and solids holdup; bubble size, distribution, and frequency data; as well as heat and mass transfer coefficient data. The data includes bubble column flows between large flat plates (i.e., 2D bubble column experiments) and cylindrical test sections. The 2D bubble column experiments listed were typically smaller in scale than the cylindrical SBCs and range in depth (i.e., distance between the large flat plates) from 5.08 to 15.0 cm and width from 15.0 to 30.48 cm. Although the 2D tests are less expensive to build and operate due to lower flow requirements (which translates to reduced pump and compressor requirements), the effects of the large flat plate walls may significantly influence the two-phase flow phenomena.

The cylindrical bubble columns found in the literature ranged in size from benchtop or laboratory scale to pilot and production size units. The laboratory scale integral tests ranged in diameter from 5.0 to 63.0 cm. The pressure and gas flow rate was typically limited by the material structural integrity and the size of gas compressors used. Therefore, the available data in the churn-flow regime was limited and additional experiments may be needed.

Available open source data (integral and fundamental) is listed in Table 3. The fundamental data is needed to validate individual mechanistic models or closure laws used in a CMFD simulation of a SBC. The fundamental data set is generally taken in more simple geometries (i.e., flow between parallel plates or vertical cylindrical pipes) or custom-designed tests to maximize selected interfacial phenomena. Data available in open literature includes gas, liquid, and solid phase velocity fields; turbulence parameters (liquid phase turbulent dispersion coefficients, Reynolds stress, turbulence intensity); gas and solids holdup distributions; bubble rise velocity, bubble drag coefficient, interfacial area; bubble size distribution and frequency; bubble coalescence and break-up parameters; heat and mass transfer coefficients; and residence time distributions. Test configurations were found to range from simple geometries to more prototypic vessels. The fundamental tests are typically done at low pressure and most times used air-water for the modeling fluids. The 2D test sections used for fundamental testing ranged in depth from 1.2 to 4.0 cm and width from 20 to 30 cm. More recent fundamental testing also includes local velocity and void fraction data in cylindrical vessels and pipes. The diameter for these tests ranged from 3.7 cm to 80 cm. The majority of experiments found in the literature were performed using air-water systems. To validate performance of FT SBCs, it will be necessary to include the effect of operating conditions (e.g., temperature, pressure, etc.) on fluid properties (e.g., density, viscosity, surface tension, etc.) and behavior (e.g., turbulent mixing, gas hold up, etc.).

#### **Summary and Conclusions**

This study reviewed existing experimental facilities and available data that can be used to validate a CMFD based model of a SBC. The review included searching the Web of Science, ISI Proceedings, and Inspec databases, internet searches as well as other open literature sources. Experimental facilities, techniques and methods, including bench-top or laboratory scale bubble columns and pilot or production scale slurry bubble column reactors, used to obtain hydrodynamic data relating to SBCs has been summarized. Available experimental data, both integral and fundamental, has also been summarized. Fundamental data is needed to validate individual mechanistic models or closure laws used in a Computational Multiphase Fluid Dynamics (CMFD) simulation of a SBC. The fundamental data set is generally taken in more simple geometries (i.e., flow between parallel plates or cylindrical pipes) or a custom designed tests to maximize selected interfacial phenomena. Integral data can be used to validate the performance of an entire SBC system. The goal of this effort is to produce a validated CMFD based model of a FT SBC that can be used as a numerical test bed. With such a tool, new concepts can be assessed virtually and optimized for FT process improvement.

The authors recognize that an enormous amount of work has been done pertaining to SBCs. The tables provided in this review are meant to be a starting point and are not exhaustive. Future experiments should build upon the work already done and existing data should be used to guide further experimentation.

Table 2. Expe	rimental Capabilities	s to Validate	CMFD-Based Model.

<b>Experimental Facility</b>	Test Section(s)	Measurement	Information Provided	Ref.
		Technique		
Aalborg University Esbjerg, Denmark	$0.15 \text{ m}$ width $\times 0.15 \text{ m}$ depth $\times 1 \text{ m}$ height	Particle Image Velocimetry (PIV), laser Doppler anemometry (LDA)	gas and liquid phase velocity fields, gas and liquid velocity fluctuations	[5]
Advanced Fuels Development Unit (AFDU), LaPorte, TX	0.46 m dia × 15.24 m height	Radioactive tracer measurements employing NaI scintillation detectors	radioactive tracer response, mean liquid phase axial and radial eddy dispersion coefficient, centerline liquid velocity, mean recirculation velocity	[6]
Beijing Institute of Petrochemical Technology, China (Haibo Jin)	0.3 m dia × 6.6 m height, 0.54 m/s gas velocity, 1.0 MPa system pressure	gamma ray attenuation	effects of superficial gas velocity, static liquid height, liquid surface tension, liquid viscosity, acid concentration, solid and antifoam agents, and system pressure on the axial and overall gas holdup	[7]
Delft University of Technology, The Netherlands (Robert F. Mudde, Wouter K. Harteveld)	15.2 cm, 23.4 cm, 38.4 cm dia columns; 15.0 cm dia × 1.2 m column	LDA, glass fiber probe	axial and tangential liquid velocity components, Reynolds stresses, turbulence power spectra, gas fraction profiles, axial mean liquid velocity, axial normal stress profiles	[8] [9] [10]
Florida Atlantic University (D. Moslemian)		Computer Automated Radioactive Particle Tracking (CARPT)	Mean circulation profiles, turbulence parameters (Reynolds normal and shear stresses, turbulent eddy dispersion coefficient),	[11]
Humboldt University Berlin, Germany (U. Kertzscher)	104 mm dia × 100 mm height SBC	XPTV	local solid velocity and the local solid hold-up in three-phase flows	[12]
Illinois Institute of Technology (Dimitri Gidaspow)	30.48 cm width × 5.08 cm depth × 213.36 height	PIV with γ- and x-ray densitometer	Flow profile, particle concentration profiles and Reynolds stresses PIV – time-averaged particle velocities and concentrations	[13]
Iowa State University (Theodore J. Heindel)	32.1 cm dia × 4.88 m height acrylic SBC	X-ray Computed Tomography (CT)	cross-sectional gas holdup distribution	[14]
NASA Glenn Research Center	25 mm dia × 20 cm height tube	MRI	void fraction distributions	[15]
Sandia National Laboratories (Tim O'Hern, Rob Tachau, John Torczynski, Joel Lash)	Industrial-scale riser testbed; 48 cm dia, ~3 m height, T≤200°C, P≤100 psig	Gamma densitometry tomography, electrical impedance tomography, electrical and optical probes	gas volume fraction, three-phase profiles	[16]
Technische Universität Braunschweig, Germany	0.63 m dia × 6.1 m height Plexiglas bubble column	DPM/TDR, DPM/ECM	local gas and solids holdup	[17]

<b>Experimental Facility</b>	Test Section(s)	Measurement Technique	Information Provided	Ref.
Texas A&M University (Dragomir B. Bukur)	0.05 and 0.21 m inside diameter by 3 m tall stainless steel bubble columns	Gamma ray densitometry	radial gas holdup distribution, radial and axial gas and solid volume fraction distributions, flow regime transitions	[18]
Ohio State University (L. S. Fan)	10 cm dia × 100 cm height bubble column; 10.2 cm width × 10.2 cm depth × 150 cm height Plexiglas square SBC	Electrical Capacitance Tomography, LDV, PIV	real-time cross-sectional gas and solids holdup distributions, instantaneous and average velocity distributions, turbulent energy distributions, Reynolds stresses, liquid-phase power spectra, bubble- induced turbulence, effect of gas distributors and solids on the turbulence field, bubble sizes and distributions	[19] [20] [21] [22] [23]
Purdue University (M. Ishii)	10 mm × 2950 mm high	conductance probe, high speed video, LDA	void fraction, interfacial area concentration, bubble size, gas and liquid velocity	[24]
University of Amsterdam, The Netherlands (R. Krishna)	0.1 m dia, 0.174 m dia, 0.19 m dia, 0.38 m dia, 0.63 m × 4 m height polyacrylate columns	dynamic gas disengagement, pitot tube	total gas holdup; small bubble holdup; dense-phase gas voidage; large bubble (dilute phase) holdup; influence of column diameter, liquid properties, gas distributor, and gas density; rise velocity and average size of large bubbles; centerline liquid velocity; radial distribution of liquid velocity; liquid phase dispersion coefficient	[25] [26]
University of Cambridge, UK (A.J. Sederman)	50 mm dia.	magnetic resonance imaging	gas phase volume fractions, distributions of gas bubble length and velocity	[27]
University of Liege, Belguim (E. Fransolet)	0.24 m dia × 2.75 m height	electrical resistance tomography, wall mounted pressure probes	influence of liquid rheology on gas flow pattern in bubble column reactor, average gas holdup, gas phase distribution, bubble size distribution	[28]
University of Mumbai, India (J. B. Joshi)	$\begin{array}{l} 385 \text{ mm dia} \times 3.2 \text{ m} \\ \text{height Perspex bubble} \\ \text{column,} \\ u_{G} = 0.06 - 0.3 \text{ m/s} \end{array}$	gamma ray tomography	radial variation of gas holdup	[29]
Washington University (Muthanna Al-Dahhan, Milorad P. Dudukovic)	0.162 m dia × 2.5 m height SBC, P≤1.2 MPa; 30.5 cm dia plexiglass SBC	Computer Automated Radioactive Particle Tracking (CARPT), Computed Tomography (CT), four-point probe, optical oxygen probe	Axial and radial eddy diffusivities, solids residence time, ensemble- average velocities and eddy diffusivities, incipient particle motion, flow regime identification, cross-sectional gas holdup distribution, bubble frequency, specific interfacial area, bubble chord length, bubble velocity, volumetric gas-liquid mass transfer coefficient ( $k_La$ ), liquid side mass transfer coefficient	[30] [31] [32] [33] [34]

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Air/water, ethanol/solidscm heightcm heightcm heightdimetryl ether synthesis0.46 m dia × 15.24T = 250°C, P = 5.27 MParadioactive tracer response, mean liquid phase axial and radial eddy dispersion coefficient, centerline liquid velocity, mean recirculation velocity[6]BKC reactorm heightP = 5.27 MPa dia × 3 m heightcoefficient, centerline liquid velocity, mean recirculation velocity[37]air/water200, 400, 800 mm dia × 3 m heightT = -20°C, P_{atm,} u_G = 0.27°C, P_{atm,} u_G = 0.25 - 7.5 cm/smass transfer efficiency, volumetric mass transfer coefficient (k_ta)[38]cyclohexane, waterheightu_G = 0.025 - 7.5 cm/sturbulent energy distributions, average velocity and Reynolds stresses, liquid-phase power spectra, bubble-induced turbulence, effect of gas distributors on the turbulence field, effect of solids on the liquid-phase turbulence[39]air/water14 cm diau_G = 0.02 - 0.12.0 cm/sturbulence intensities profiles[40]air/water, K_SSQ_r/poly- hoight Plexiglas bubble column0.63 m dia × 6.1 m dia × 0.50 m heightu_G = 0.02 - 0.15 m/slocal gas and solids holdup[17]air/water0.16 u dia × 2.75 m heightT = 293Kbubble rise velocity, drag coefficient[41]air/water0.16 u dia × 0.50 m heightU_G = 0.02 - 0.15 m/sinfluence of liquid rheology on gas flow pattern, average gas holdup, gas phase distribution, bubble size distribution[34]air/water0.16 u dia × 0.50 m dia × 0.50 m heightP ≤ 1.0 MPaoverall gas holdup, volumetric gas-liquid mass transfer coefficient[42] </td <td>paraffinic wax</td> <td>m height</td> <td>atmospheric pressure</td> <td>flow regime transition</td> <td></td>	paraffinic wax	m height	atmospheric pressure	flow regime transition	
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SBC reactorm height $P = 5.27$ MPacoefficient, centerline liquid velocity, mean recirculation velocityair/water200, 400, 800 mm $u_{g} = 20-90$ mm/sinstantaneous local heat transfer rates[37]nitrogen, carbon dixide/0.2 m dia × 1.6 m $T = \sim 20^{\circ}C, P_{mm},$ mass transfer coefficient, (so that transfer rates[38]cyclohexane, waterheight $T = \sim 20^{\circ}C, P_{mm},$ mass transfer coefficient, (so that transfer rates[39]particles10.2 cm width × $u_{g} = 0.025 \cdot 7.5$ cm/sturbulent energy distributions, average velocity and Reynolds stresses, liquid-phase[39]ait/water (S04/poly-14 cm dia; 19 cm $u_{g} = 2.0-12.0$ cm/sturbulent energy distributions, average velocity and Reynolds stresses, liquid-phase[40]ait/water (S204/poly-0.63 m dia × 6.1 m $u_{g} = 2.0-12.0$ cm/sturbulence intensities profiles[40]ait/water (S204/poly-0.63 m dia × 6.1 m $u_{g} = 0.02-0.09$ m/slocal gas and solids holdup[41]nd non-Newtonian0.24 m dia × 1.6 mT = 293Kbubble rise velocity, drag coefficient[41]fluids0.24 m dia × 2.75 $u_{g} = 0.02-0.15$ m/sinfluence of liquid rheology on gas flow pattern, average gas holdup, gas phase[28]solutionsm height0.24 m dia × 1.6 mT = 293Kbubble rise velocity, drag coefficient[41]fluidsdia × 0.50 m heightT = 293Kug = 0.02-0.15 m/sinfluence of liquid rheology on gas flow pattern, average gas holdup, gas phase[28]solutionsm height0.24 m dia × 2.75 <td>Air/water, ethanol/solids</td> <td>cm height</td> <td></td> <td></td> <td></td>	Air/water, ethanol/solids	cm height			
SBC reactorm height $P = 5.27$ MPacoefficient, centerline liquid velocity, mean recirculation velocityair/water200, 400, 800 mm $u_{g} = 20-90$ mm/sinstantaneous local heat transfer rates[37]nitrogen, carbon dixide/0.2 m dia × 1.6 m $T = \sim 20^{\circ}C, P_{mm},$ mass transfer coefficient, (so that transfer rates[38]cyclohexane, waterheight $T = \sim 20^{\circ}C, P_{mm},$ mass transfer coefficient, (so that transfer rates[39]particles10.2 cm width × $u_{g} = 0.025 \cdot 7.5$ cm/sturbulent energy distributions, average velocity and Reynolds stresses, liquid-phase[39]ait/water (S04/poly-14 cm dia; 19 cm $u_{g} = 2.0-12.0$ cm/sturbulent energy distributions, average velocity and Reynolds stresses, liquid-phase[40]ait/water (S204/poly-0.63 m dia × 6.1 m $u_{g} = 2.0-12.0$ cm/sturbulence intensities profiles[40]ait/water (S204/poly-0.63 m dia × 6.1 m $u_{g} = 0.02-0.09$ m/slocal gas and solids holdup[41]nd non-Newtonian0.24 m dia × 1.6 mT = 293Kbubble rise velocity, drag coefficient[41]fluids0.24 m dia × 2.75 $u_{g} = 0.02-0.15$ m/sinfluence of liquid rheology on gas flow pattern, average gas holdup, gas phase[28]solutionsm height0.24 m dia × 1.6 mT = 293Kbubble rise velocity, drag coefficient[41]fluidsdia × 0.50 m heightT = 293Kug = 0.02-0.15 m/sinfluence of liquid rheology on gas flow pattern, average gas holdup, gas phase[28]solutionsm height0.24 m dia × 2.75 <td>dimethyl ether synthesis</td> <td>0.46 m dia × 15.24</td> <td><math>T = 250^{\circ}C</math>,</td> <td>radioactive tracer response, mean liquid phase axial and radial eddy dispersion</td> <td>[6]</td>	dimethyl ether synthesis	0.46 m dia × 15.24	$T = 250^{\circ}C$ ,	radioactive tracer response, mean liquid phase axial and radial eddy dispersion	[6]
air/water200, 400, 800 mm dia × 3 m height $u_G = 20-90$ mm/sinstantaneous local heat transfer rates[37]introgen, carbon dioxide/ cyclohexane, water0.2 m dia × 1.6 m heightT = ~20°C, P_{atms} u_G ≤ 0.14 m/smass transfer efficiency, volumetric mass transfer coefficient (k_La)[38]air/water/500 µm acetate particles10.2 cm width × 150 cm heightu_G = 0.025-7.5 cm/s power spectra, bubble-induced turbulence, effect of gas distributors on the turbulence field, effect of solids on the liquid-phase turbulence field, effect of solids on the liquid-phase turbulence[39]air/water14 cm dia; 19 cm dia; 44 cm dia u_G = 0.02-0.09 m/su_G = 0.02-0.09 m/s turbulence effect of solids on the liquid-phase turbulence[40]air/water, K <sub>2</sub> SO <sub>4</sub> /poly- polyoxymethylene subble column0.4 m dia × 1.60 m heightu_G = 0.02-0.09 m/s turbulencelocal gas and solids holdup[17]fluids0.24 m dia × 1.60 m heightT = 293Kbubble rise velocity, drag coefficient[41]air/water0.12 m dia × 2.60 m heightU_G = 0.02-0.15 m/s distribution, bubble size distributioninfluence of liquid rheology on gas flow pattern, average gas holdup, gas phase distribution, bubble size distribution[34]air/water0.4 m dia × 9 m heightP $\leq$ 1.0 MPaoverall gas holdup, volumetric gas-liquid mass transfer coefficient (k_La), interfacial area, liquid side mass transfer, total gas hold-up, small bubble hold-up[42]air/water0.4 m dia × 2.8 m heightP $\leq$ 0.8 MPaeffects of gas velocity, system pressure, and catalyst loading on gas holdup[43] <td>SBC reactor</td> <td>m height</td> <td>P = 5.27 MPa</td> <td>coefficient, centerline liquid velocity, mean recirculation velocity</td> <td></td>	SBC reactor	m height	P = 5.27 MPa	coefficient, centerline liquid velocity, mean recirculation velocity	
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air/water $0.4 \text{ m}$ dia × 9 m heightatmospheric pressure gas-liquid mass transfer, total gas hold-up, small bubble hold-up in the liquid-small bubble mixture, large and small bubble hold-up[42] $H_2, N_2, CO, CH_4$ /hexane mixture/FT Fe catalyst $0.316 \text{ m}$ dia × 2.8 m height $P \le 0.8 \text{ MPa}$ effects of gas velocity, system pressure, and catalyst loading on gas holdup[43]air/water, Drakeol 10 $48 \text{ cm}$ dia × $\sim 3 \text{ m}$ height $P \le 50 \text{ psig}$ $u_G \le 0.25 \text{ m/s}$ spatially resolved gas holdup[16]air/water; $0.3 \text{ m}$ dia × $6.6 \text{ m}$ $P \le 1.0 \text{ MPa}$ effects of superficial gas velocity, liquid surface tension, liquid viscosity, and system[7]			_		
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mixture/FT Fe catalyst       m height       Image: Constraint of the second se	H <sub>2</sub> , N <sub>2</sub> , CO, CH <sub>4</sub> /hexane	0.316 m dia × 2.8	P < 0.8 MPa		[43]
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	air/water:	0		effects of superficial gas velocity, liquid surface tension, liquid viscosity, and system	[7]
	air/acetic acid	height	$u_G \leq 0.4 \text{ m/s}$	pressure on the axial gas holdup	L' J

#### Table 3. Experimental Data to Validate CMFD-Based Model.

System	Column	Operating	Data Acquired	Ref.
gas/liquid/solid	Dimensions	Conditions		
nitrogen, helium/ethanol,	0.1 m dia × 2.4 m	T = 293K,	effect of gas density on the gas hold-up structure in a bubble column with organic	[44]
1-butanol, toluene,	height	P = 0.1-4.0 MPa,	liquids	
decalin, tap water		$u_G = 0.01 - 0.20 \text{ m/s}$		
nitrogen/distilled water,	0.3 m width ×	T = 293 K,	local and overall gas holdup in homogenous, transition, and heterogeneous regimes	[45]
sodium gluconate/carbon	$0.015 \text{ m depth} \times 2$	P = 1 bar		
particles	m height			
air/water	0.174, 0.38, 0.63 m	P = 101.3 kPa	radial distribution of liquid velocities, liquid-phase axial dispersion coefficients, bubble	[46]
	dia $\times$ 4 m height		drag coefficient	
air/paraffin oil/porous	0.1, 0.19, 0.38 m	P = 1 bar	gas holdup, gas voidage, rise velocity and average size of large bubbles, centerline	[26]
silica particles	dia		liquid velocity, radial distribution of liquid velocity, liquid phase dispersion coefficient	
air/water, isopropanol	0.10, 0.14, 0.19,	$u_G = 0.02 - 0.12 \text{ m/s}$	effect of column diameter, superficial gas velocity, distributor type, static liquid height,	[47]
	0.26, 0.30 m dia		axial distance from distributor, and liquid properties on gas-holdup	
air/water, inorganic	0.078 m dia × 4.6	T = 45°C,	average gas holdup, bubble size distribution, method to estimate coalescence and	[48]
industrial solution	m height	$u_{\rm G} = 1-8  {\rm cm/s}$	break-up parameters in bubble columns	
air/1.5 wt%	30 cm width $\times$ 1.2		coalescing mechanism of two in-line oblate-cusped bubbles rising in a non-Newtonian	[49]
polyacryamide in water	cm depth $\times$ 120 cm		fluid	
	height			
air/water, glycerin	68 mm width × 88		turbulence intensity, Reynolds stress, liquid velocity profile associated with bubble	[50]
	mm depth × 450		induced flow structure	
	mm height			
nitrogen/water, ethanol,	0.1 m dia × 2.1 m	T = 298-323K,	gas holdup, axial liquid velocity, liquid axial dispersion coefficient, CFD model	[51]
1-butanol	height	P = 0.1 - 0.5 MPa,	developed for the prediction of flow pattern in terms of mean velocity and eddy	
		$u_G = 0.01 - 0.21 \text{ m/s}$	diffusivity profiles. Model also predicts residence time distribution and axial dispersion	
			coefficient.	
air/water/calcium	0.14 m dia		effect of presence of solid phase on homogeneous-heterogeneous flow regime	[52]
alginate beads			transition, regime transition critical point, voidage as function of gas flow rate	<u> </u>
air/water, n-butanol,	$10 \text{ cm width} \times 10$	ambient T and P	effect of liquid properties on average gas holdup values, bubble size distributions and	[53]
glycerin	cm depth $\times$ 150 cm		Sauter diameters	
	height			
air/water	$0.2 \text{ m width} \times 0.04$	ambient T and P	flow regime identification, Reynolds stresses, global gas hold-up, liquid velocities,	[54]
	m depth $\times$ 1.2 m		liquid flow macrostructures	
	height			
coal-to-liquids reaction	1 m dia × 11.8 m	T = 313-733 K,	residence time distribution curves	[55]
• / • •	height pilot plant	P = 16.6 - 16.8 MPa		1.50
air/water, cationic and	0.05 m dia × 0.40	$T = 20^{\circ}C$ , dynamic	effect of liquid properties (surfactants) on bubble generation phenomenon, interfacial	[56]
anionic surfactant	m height glass	bubble regime	area, and liquid-side mass transfer coefficient	1
aqueous solutions	bubble column			-
air/clay particles	$0.3 \text{ m} \text{dia} \times 3 \text{ m}$		effect of power input, fluid phase viscosity and solids loading on the mechanical stress	[57]
suspended in water	height		on suspended particles	

System	Column	Operating	Data Acquired	Ref.
gas/liquid/solid	Dimensions	Conditions		
nitrogen/cyclohexane	0.3 m dia × 4 m	T = 30-160°C	bubble size distribution, mean gas holdup, interfacial area, bubble swarm velocity	[58]
	height	P = 0.2-1.1 MPa		
air/water;	195.3 mm dia	20° C < T < 30° C	gas fraction, gas velocity, bubble size	[59]
steam/water		P = 120  kPa;		
		T=180-280° C		
		1 < P < 6.5 MPa		
air/water/glass beads	0.162 m dia × 2.5	P = 0.1 - 1.0 MPa	effect of reactor pressure and superficial gas velocity on solids phase velocity and shear	[32]
(150 μm)	m height	$u_G = 0.08 - 0.45 \text{ m/s}$	stress	
air/water/glass beads	0.162 m dia × 2.5	P = 0.1-1.0 MPa	cross sectional holdup distribution of gas, liquid, and solid phases	[60]
(150 µm)	m height	$u_G = 0.08 - 0.45 \text{ m/s}$		
air/water	$0.1 \text{ m width} \times 0.1$	$u_{\rm G} = 0-8  {\rm cm/s}$	bubble velocity, diameter, void fraction, and liquid velocity, influence of the void	[61]
	m depth $\times$ 1.0 m		fraction on the relative velocity of a swarm of gas bubbles, experimental drag	
	height		correlation proposed	
nitrogen/water/glass	8.89 cm dia × 290	$u_G = 0.5-12 \text{ cm/s},$	solid holdup	[62]
beads	cm height	5-30 wt% solids		
nitrogen/Tellus oil,	0.15 m dia × 1.22	ambient temperature,	effect of high liquid viscosity, column diameter and operating pressure on the total gas	[63]
aqueous glucose	m height; 0.23 m	P = 0.1-1 MPa	hold-up	
solutions	dia $\times$ 1.22 m height			
air, nitrogen/water,	0.15 m dia	ambient temperature,	influence of particle lyophobicity on gas hold-up, homogeneous to churn-turbulent	[64]
organic oil/activated		P = 0.1-1.3 MPa	regime transition, and gas-liquid mass transfer	
carbon, silica particles				
air/C <sub>9</sub> -C <sub>11</sub> paraffin	0.1 m dia	ambient T and P,	gas holdup and volumetric mass transfer coefficient in homogeneous and churn-	[65]
oil/porous catalyst		$u_{\rm G} = 0-0.4 \ {\rm m/s}$	turbulent flow regimes	
air/aqueous solutions of	0.385 m dia × 3.2	$u_G = 0.06 - 0.24 \text{ m/s}$	effect of superficial gas velocity on radial gas hold-up profiles	[29]
n-butanol	m height column			[66]
air/water	0.1 m dia × 2 m	ambient T and P,	axial and tangential mean and RMS liquid velocity values	[67]
	height column	$u_G = 0.6-15 \text{ cm/s}$		<u> </u>
air/water	0.16 m dia × 2.5 m	$P \le 10$ bar	time-averaged local heat transfer coefficients	[68]
	height column	$u_G = 0.03 - 0.30 \text{ m/s}$		
water, hydrocarbon	5.08 cm dia × 55.88	room temperature,	liquid phase axial dispersion coefficients	[69]
liquids	cm height, 10.16	$P \le 10.3 MPa$		
	cm dia $\times$ 91.44 cm	$u_G \leq 0.4 \text{ cm/s}$		
	height columns			
nitrogen, carbon monox-	$37 \text{ mm dia} \times 480$	$T = RT-300^{\circ}C,$	bubble sizes, interfacial areas and volumetric mass transfer coefficients	[70]
ide, hydrogen/liquid par-	mm height SBC	P = atm-6 MPa		
affin/silica gel powder		$u_G \leq 0.02 \text{ m/s}$		
air/water	10 mm thick $\times$	$T = 70 ^{\circ}C,$	void fraction, interfacial area concentration, interaction mechanisms	[24]
• / .	2950 mm high	P = 24-49  kPa		
air/water	0.114, 0.190, 0.292	$u_{\rm G} = 2-18.4 \ {\rm cm/s}$	axial dispersion coefficients (eddy diffusivities)	[71]
	m dia columns			

### References

- 1. Fischer, F. and H. Tropsch, Synthesis of Methanol for CO and H2, French Patent 540, Editor. 1921.
- 2. Chen, Y., *Modeling Gas-Liquid Flow in Pipes: Flow Regime Transitions and Drift-Flux Modeling*. 2001, Stanford University.
- 3. Antal, S.P., et al., *Development of a Next Generation Computer Code for the Prediction of Multicomponent Multiphase Flows*, in *International Meeting on Trends in Numerical and Physical Modeling for Industrial Multiphase Flow.* 2000: Cargese, France.
- 4. Podowski, M.Z., On the Mechanistic Modeling of Interfacial Phenomena in Gas/Liquid Two-Phase Flows, in 6th International Conference on Multiphase Flows, ICMF 2007. 2007: Leipzig, Germany.
- 5. Deen, N.G., B.H. Hjertager, and T. Solberg, *Comparison of PIV and LDA Measurement Methods Applied to the Gas-Liquid Flow in a Bubble Column*, in 10th International Symposium on *Applications of Laser Techniques to Fluid Mechanics*. 2000: Lisbon, Portugal.
- Chen, P., et al., *Hydrodynamics of slurry bubble column during dimethyl ether (DME) synthesis: Gas-liquid recirculation model and radioactive tracer studies.* Chemical Engineering Science, 2006.
   61(19): p. 6553-6570.
- Jin, H.B., et al., *The axial distribution of holdups in an industrial-scale bubble column with evaluated pressure using gamma-ray attenuation approach*. Chemical Engineering Journal, 2005. 115(1-2): p. 45-50.
- 8. Mudde, R.F., J.S. Groen, and H.E.A.V.D. Akker, *Liquid velocity field in a bubble column: LDA experiments.* Chemical Engineering Science, 1997. **52**(21-22): p. 4217-4224.
- Harteveld, W.K., R.F. Mudde, and H.E.A. Van den Akker, *Estimation of turbulence power spectra for bubbly flows from laser Doppler Anemometry signals*. Chemical Engineering Science, 2005. 60(22): p. 6160-6168.
- Harteveld, W.K., R.F. Mudde, and H.E.A. van den Akker, *Dynamics of a bubble column: Influence of gas distribution on coherent structures*. Canadian Journal of Chemical Engineering, 2003. 81(3-4): p. 389-394.
- Moslemian, D., N. Devanathan, and M.P. Dudukovic, *Radioactive particle tracking technique for investigation of phase recirculation and turbulence in multiphase systems*. Review of Scientific Instruments, 1992. 63(10): p. 4361-4372.
- 12. Seeger, A., et al., *Measurement of the local velocity of the solid phase and the local solid hold-up in a three-phase flow by X-ray based particle tracking velocimetry (XPTV)*. Chemical Engineering Science, 2003. **58**(9): p. 1721-1729.
- 13. Lam, P. and D. Gidaspow, *Computational and Experimental Modeling of Slurry Bubble Column Reactors*. 2000, University of Akron, Illinois Institute of Technology.
- 14. Hubers, J.L., et al., *X-ray computed tomography in large bubble columns*. Chemical Engineering Science, 2005. **60**(22): p. 6124-6133.
- 15. Daidzic, N.E., et al., *Gas-liquid phase distribution and void fraction measurements using MRI*. Nuclear Engineering and Design, 2005. **235**(10-12): p. 1163-1178.
- 16. Jackson, N.B., et al., Sandia Support for PETC Fischer-Tropsch Research: Experimental Characterization of Slurry-Phase Bubble-Column Reactor Hydrodynamics. 1996, Advanced Energy Technology Center, Engineering Sciences Center.
- 17. Dziallas, H., V. Michele, and D.C. Hempel, *Measurement of Local Phase Holdups in a Two- and Three-Phase Bubble Column*. Chemical Engineering & Technology, 2000. **23**(10): p. 877-884.

- Bukur, D.B., J.G. Daly, and S.A. Patel, *Application of gamma-ray attenuation for measurement of gas holdups and flow regime transitions in bubble columns*. Industrial & Engineering Chemistry Research, 1996. **35**(1): p. 70-80.
- 19. Warsito, W. and L.-S. Fan, *Measurement of real-time flow structures in gas-liquid and gas-liquid-solid flow systems using electrical capacitance tomography (ECT)*. Chemical Engineering Science, 2001. **56**(21-22): p. 6455-6462.
- 20. Warsito, W. and L.S. Fan, *3D-ECT velocimetry for flow structure quantification of gas-liquid-solid fluidized beds*. Canadian Journal of Chemical Engineering, 2003. **81**(3-4): p. 875-884.
- Warsito, W. and L.S. Fan, *Dynamics of spiral bubble plume motion in the entrance region of bubble columns and three-phase fluidized beds using 3D ECT*. Chemical Engineering Science, 2005. 60(22): p. 6073-6084.
- 22. Cui, Z. and L.S. Fan, *Energy spectra for interactive turbulence fields in a bubble column*. Industrial & Engineering Chemistry Research, 2005. **44**(5): p. 1150-1159.
- Chen, R.C. and L.-S. Fan, Particle Image Velocimetry for Characterizing the Flow Structure in Three-Dimensional Gas-Liquid-Solid Fluidized Beds. Chemical Engineering Science, 1992.
   47(13/14): p. 3615-3622.
- 24. Sun, X., *Two-Group Interfacial Area Transport Equation for a Confined Test Section*. 2001, Purdue University.
- 25. Krishna, R. and J. Ellenberger, *Gas holdup in bubble column reactors operating in the churnturbulent flow regime*. Aiche Journal, 1996. **42**(9): p. 2627-2634.
- 26. Krishna, R., et al., *Design and scale up of a bubble column slurry reactor for Fischer-Tropsch synthesis*. Chemical Engineering Science, 2001. **56**(2): p. 537-545.
- 27. Sederman, A.J., M.D. Mantle, and L.F. Gladden, *Quantitative 'Real-Time' Imaging of Multi-Phase Flow in Ceramic Monoliths*. Magnetic Resonance Imaging, 2003. **21**: p. 359-361.
- 28. Fransolet, E., et al., *Analysis of gas holdup in bubble columns with non-Newtonian fluid using electrical resistance tomography and dynamic gas disengagement technique.* Chemical Engineering Science, 2005. **60**(22): p. 6118-6123.
- 29. Veera, U.P., K.L. Kataria, and J.B. Joshi, *Gas hold-up profiles in foaming liquids in bubble columns*. Chemical Engineering Journal, 2001. **84**(3): p. 247-256.
- Chen, J., et al., *Particle Motion in Packed/Ebullated Beds by CT and CARPT*. Aiche Journal, 2001.
   47(5): p. 994-1004.
- Nedeltchev, S., S.B. Kumar, and M.P. Dudukovic, *Flow regime identification in a bubble column* based on both Kolmogorov entropy and quality of mixedness derived from CARPT data. Canadian Journal of Chemical Engineering, 2003. 81(3-4): p. 367-374.
- 32. Rados, N., A. Shaikh, and M.H. Al-Dahhan, *Solids flow mapping in a high pressure slurry bubble column*. Chemical Engineering Science, 2005. **60**(22): p. 6067-6072.
- 33. Xue, J.L., et al., *Bubble dynamics measurements using four-point optical probe*. Canadian Journal of Chemical Engineering, 2003. **81**(3-4): p. 375-381.
- 34. Han, L. and M.H. Al-Dahhan, *Gas-liquid mass transfer in a high pressure bubble column reactor with different sparger designs*. Chemical Engineering Science, 2007. **62**(1-2): p. 131-139.
- 35. Al-Masry, W.A., E.M. Ali, and Y.M. Aqeel, *Determination of bubble characteristics in bubble columns using statistical analysis of acoustic sound measurements*. Chemical Engineering Research & Design, 2005. 83(A10): p. 1196-1207.
- 36. Bukur, D.B., D. Petrovic, and J.G. Daly, *Flow Regime Transitions in a Bubble Column with a Parrafin Wax as the Liquid Medium*. Industrial & Engineering Chemistry Research, 1987. **26**: p. 1087-1092.

- 37. Chen, W., et al., *Generalized dynamic modeling of local heat transfer in bubble columns*. Chemical Engineering Journal, 2003. **96**(1-3): p. 37-44.
- Chaumat, H., et al., Mass transfer in bubble column for industrial conditions-effects of organic medium, gas and liquid flow rates and column design. Chemical Engineering Science, 2005. 60(22): p. 5930-5936.
- 39. Cui, Z. and L.S. Fan, *Turbulence energy distributions in bubbling gas-liquid and gas-liquid-solid flow systems.* Chemical Engineering Science, 2004. **59**(8-9): p. 1755-1766.
- 40. Degaleesan, S., M. Dudukovic, and Y. Pan, *Experimental Study of Gas-Induced Liquid-Flow Structures in Bubble Columns*. Aiche Journal, 2001. **447**(9): p. 1913-1931.
- 41. Frank, X., et al., *Bubble motion in non-Newtonian fluids and suspensions*. Canadian Journal of Chemical Engineering, 2003. **81**(3-4): p. 483-490.
- 42. Haut, B. and T. Cartage, *Mathematical modeling of gas-liquid mass transfer rate in bubble columns operated in the heterogeneous regime*. Chemical Engineering Science, 2005. **60**(22): p. 5937-5944.
- 43. Inga, J.R. and B.I. Morsi, *Effect of operating variables on the gas holdup in a large-scale slurry bubble column reactor operating with an organic liquid mixture.* Industrial & Engineering Chemistry Research, 1999. **38**(3): p. 928-937.
- 44. Jordan, U., A.K. Saxena, and A. Schumpe, *Dynamic gas disengagement in a high-pressure bubble column.* Canadian Journal of Chemical Engineering, 2003. **81**(3-4): p. 491-498.
- 45. Kluytmans, J.H.J., et al., *2D slurry bubble column hydrodynamic phenomena clarified with a 3D gas-liquid model.* Canadian Journal of Chemical Engineering, 2003. **81**(3-4): p. 456-464.
- 46. Krishna, R., et al., *Influence of scale on the hydrodynamics of bubble columns operating in the churn-turbulent regime: experiments vs. Eulerian simulations.* Chemical Engineering Science, 1999. 54(21): p. 4903-4911.
- 47. Kumar, S.B., D. Moslemian, and M.P. Dudukovic, *Gas-holdup measurements in bubble columns using computed tomography*. Aiche Journal, 1997. **43**(6): p. 1414-1425.
- 48. Laari, A. and K. Turunen, *Experimental determination of bubble coalescence and break-up rates in a bubble column reactor*. Canadian Journal of Chemical Engineering, 2003. **81**(3-4): p. 395-401.
- 49. Lin, T.J. and G.M. Lin, *The mechanisms of bubble coalescence in a non-Newtonian fluid*. Canadian Journal of Chemical Engineering, 2003. **81**(3-4): p. 476-482.
- 50. Liu, Z., et al., *Study of bubble induced flow structure using PIV*. Chemical Engineering Science, 2005. **60**(13): p. 3537-3552.
- 51. Lorenz, O., et al., *Liquid phase axial mixing in bubble columns operated at high pressures*. Chemical Engineering Science, 2005. **60**(13): p. 3573-3586.
- 52. Mena, P.C., et al., *Effect of solids on homogeneous-heterogeneous flow regime transition in bubble columns*. Chemical Engineering Science, 2005. **60**(22): p. 6013-6026.
- Mouza, A.A., G.K. Dalakoglou, and S.V. Paras, *Effect of liquid properties on the performance of bubble column reactors with fine pore spargers*. Chemical Engineering Science, 2005. 60(5): p. 1465-1475.
- Olmos, E., C. Gentric, and N. Midoux, *Identification of flow regimes in a flat gas-liquid bubble column via wavelet transform*. Canadian Journal of Chemical Engineering, 2003. 81(3-4): p. 382-388.
- 55. Onozaki, M., et al., *Dynamic simulation of gas-liquid dispersion behavior in coal liquefaction reactors*. Chemical Engineering Science, 2000. **55**(21): p. 5099-5113.
- 56. Painmanakul, P., et al., *Effect of surfactants on liquid-side mass transfer coefficients*. Chemical Engineering Science, 2005. **60**(22): p. 6480-6491.

- 57. Pilz, R.D. and D.C. Hempel, *Mechanical stress on suspended particles in two- and three-phase airlift loop reactors and bubble columns.* Chemical Engineering Science, 2005. **60**(22): p. 6004-6012.
- 58. Pohorecki, R., et al., *Hydrodynamics of a pilot plant bubble column under elevated temperature and pressure*. Chemical Engineering Science, 2001. **56**(3): p. 1167-1174.
- 59. Prasser, H.-M., et al., *Evolution of the structure of a gas-liquid two-phase flow in a large vertical pipe*. Nuclear Engineering and Design, 2007: p. doi:10.1016/j.nucengdes.2007.02.018.
- Rados, N., A. Shaikh, and M. Al-Dahhan, *Phase distribution in a high pressure slurry bubble column via a single source computed tomography*. Canadian Journal of Chemical Engineering, 2005. 83(1): p. 104-112.
- 61. Simonnet, M., et al., *Experimental determination of the drag coefficient in a swarm of bubbles*. Chemical Engineering Science, 2007. **62**(3): p. 858-866.
- 62. Soong, Y., et al., *Ultrasonic Characterizations of Slurries in a Bubble Column Reactor*. Industrial & Engineering Chemistry Research, 1999. **38**: p. 2137-2143.
- 63. Urseanu, M.I., et al., *Influence of operating pressure on the gas hold-up in bubble columns for high viscous media.* Chemical Engineering Science, 2003. **58**(3-6): p. 697-704.
- 64. van der Schaaf, J., et al., *Effect of particle lyophobicity in slurry bubble columns at elevated pressures.* Chemical Engineering Science, 2006.
- Vandu, C.O., K. Koop, and R. Krishna, *Volumetric mass transfer coefficient in a slurry bubble column operating in the heterogeneous flow regime*. Chemical Engineering Science, 2004. 59(22-23): p. 5417-5423.
- 66. Veera, U.P., K.L. Kataria, and J.B. Joshi, *Effect of superficial gas velocity on gas hold-up profiles in foaming liquids in bubble column reactors.* Chemical Engineering Journal, 2004. **99**(1): p. 53-58.
- 67. Vial, C., et al., *Influence of gas distribution and regime transitions on liquid velocity and turbulence in a 3-D bubble column*. Chemical Engineering Science, 2001. **56**(3): p. 1085-1093.
- 68. Wu, C., M.H. Al-Dahhan, and A. Prakash, *Heat transfer coefficients in a high-pressure bubble column.* Chemical Engineering Science, 2007. **62**(1-2): p. 140-147.
- Yang, G.Q. and L.S. Fan, *Axial liquid mixing in high-pressure bubble columns*. Aiche Journal, 2003.
   **49**(8): p. 1995-2008.
- Yang, W., et al., Gas-liquid mass transfer in slurry bubble systems. II. Verification and simulation of the model based on the single bubble mechanism. Chemical Engineering Journal, 2003. 96(1-3): p. 29-35.
- 71. Yang, Y.B., N. Devanathan, and M.P. Dudukovic, *Liquid Backmixing in Bubble Columns*. Chemical Engineering Science, 1992. **47**(9-11): p. 2859-2864.