

Demonstration of Mer-Cure™ Technology for Enhanced Mercury Control

FINAL REPORT

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Public Abstract

Alstom Power Inc. has completed a DOE/NETL-sponsored program (under DOE Cooperative Agreement No. DE-FC26-07NT42776) to demonstrate Mer-Cure™, one of Alstom's mercury control technologies for coal-fired boilers. The Mer-Cure™ system utilizes a small amount of Mer-Clean™ sorbent that is injected into the flue gas stream for oxidation and adsorption of gaseous mercury. Mer-Clean™ sorbents are carbon-based and prepared with chemical additives that promote oxidation and capture of mercury. The Mer-Cure™ system is unique in that the sorbent is injected into an environment where the mercury capture kinetics is accelerated.

The full-scale demonstration program originally included test campaigns at two host sites: LCRA's 480-MW_e Fayette Unit #3 and Reliant Energy's 190-MW_e Shawville Unit #3. The only demonstration tests actually done were the short-term tests at LCRA due to budget constraints. This report gives a summary of the demonstration testing at Fayette Unit #3.

The goals for this Mercury Round 3 program, established by DOE/NETL under the original solicitation, were to reduce the uncontrolled mercury emissions by 90% at a cost significantly less than 50% of the previous target of \$60,000/lb mercury removed.

The results indicated that Mer-Cure[™] technology could achieve mercury removal of 90% based on uncontrolled stack emissions. The estimated costs for 90% mercury control, at a sorbent cost of \$0.75 to \$2.00/lb respectively, were \$13,400 to \$18,700/lb Hg removed.

In summary, the results from demonstration testing show that the goals established by DOE/NETL were met during this test program. The goal of 90% mercury reduction was achieved. Estimated mercury removal costs were 69-78% lower than the benchmark of \$60,000/lb mercury removed, significantly less than 50% of the baseline removal cost.



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List of Abbreviations

BOP balance of plant

CFD computational fluid dynamics
CMM continuous mercury monitor
DOE U.S. Department of Energy

UND EERC University of North Dakota Energy and Environmental Research Center

ESP electrostatic precipitator

LCRA Lower Colorado River Authority

LOI loss on ignition

NETL National Energy Technology Laboratory

OH Ontario Hydro

PAC powdered activated carbon

PC pulverized coal PRB Powder River Basin

PSA PS Analytical
QA quality assurance
QC quality control

SCA specific collection area

UBC unburned carbon



Executive Summary

Alstom was awarded a DOE/NETL-sponsored project under "Mercury Round 3" (DOE Cooperative Agreement No. DE-FC26-07NT42776) to perform full-scale demonstration of its Mer-Cure™ technology in two coal-fired boilers burning coals of various ranks. The full-scale demonstration program originally included test campaigns at two host sites: Lower Colorado River Authority's (LCRA's) 480-MW_e Fayette Unit #3 and Reliant Energy's 190-MW_e Shawville Unit #3. The only demonstration tests actually done were the short-term tests LCRA due to budget constraints. This report gives a summary of the demonstration testing at Fayette Unit #3.

Mer-Cure™ technology is a treated sorbent-based mercury control technology. This technology has been developed based on a mechanistic understanding of the critical areas that need to be augmented for mercury oxidation and capture. The technology has progressed through well-controlled laboratory-scale testing, pilot-scale testing, and DOE/NETL-sponsored full-scale test programs.

In Alstom's Mer-Cure™ mercury control technology, a small amount of sorbent is injected into the flue gas stream for oxidation and adsorption of gaseous mercury. The sorbents are activated carbon-based and prepared with chemical additives that promote oxidation and capture of elemental mercury. The technology is unique in that the sorbent is injected into a high temperature (above 700°F) environment where the mercury oxidation/removal kinetics are accelerated.

The report concludes that Mer-Cure[™] technology, using Mer-Clean[™] sorbent(s), is both a cost and performance effective means of controlling mercury emissions. Mer-Cure[™] technology was also successfully applied to alternative sorbent products in this project.

Background

This field demonstration program had two host sites for full-scale demonstration to obtain the required information for assessing the technical feasibility and determining the mercury capture performance and costs of controlling mercury in coal-fired utility plants using Mer-Cure™ technology. The two sites were selected to represent boilers with specific mercury capture issues. Fayette Unit #3 was selected to evaluate the effect of mercury control on fly ash utilization for concrete, and Shawville Unit #3 was selected to evaluate the effect of high SO₃ levels on mercury capture performance. The program was designed to document the performance of the Mer-Cure™ technology using both Mer-Clean™ sorbent and alternative sorbents to determine their SO₃ tolerance and "concrete-friendly" characteristics.

The overall objective of the work was to perform two- to six-month, full-scale demonstrations of Mer-Cure[™] technology in two coal-fired boilers. The goals of the Mercury Round 3 program, as originally defined by DOE/NETL in the solicitation, are to demonstrate that Mer-Cure[™] technology can achieve:

- Greater than 90% reduction of gaseous mercury in the flue gas;
- At a mercury control cost significantly lower than 50% of \$60,000/lb mercury removed.



Alstom-PPL also collected performance data to evaluate the long-term impact of mercury control on the operation of power plants and overall plant economics, including information desired by LCRA on the maximum mercury reduction possible while maintaining the ability to sell their fly ash for concrete applications.

Work Scope

In order to accomplish the technical objectives, a technical approach was utilized to effectively demonstrate and evaluate Alstom-PPL mercury control technology. Testing at each demonstration site was planned to include both a five week long short-term and two to six month long long-term, full-scale test campaign. The work required for the test campaigns can be broken down to the following tasks:

- Task 1. Design, Engineering and Fabrication of System
- Subtask 1.1 Design and Engineering of System Architecture
- Subtask 1.2 Component Fabrication and System Assembly
- Task 2. Short-Term Demonstration
- Subtask 2.1 Project Planning
- Subtask 2.2 Design, Fabrication and Maintenance of Site-Specific System
- Subtask 2.3 Installation of Mobile Mer-Cure[™] System
- Subtask 2.4 Parametric Testing
- Task 3 Long-Term Field Demonstration
- Subtask 3.1 Installation, Checkout and Commissioning
- Subtask 3.2 Field Testing and Measurement
- Subtask 3.3 System Removal
- Subtask 3.4 Data Analysis and Site Report
- Task 4. Technology Transfer
- Task 5. Program Management and Reporting.

Alstom-PPL engineers at Alstom-PPL's Windsor site performed most of the Task 1 work. Tasks 2 and 3 were to be repeated for both host sites. Alstom-PPL engineers performed Task 2 with the assistance of host site personnel and EERC measurement crew. The five week short-term test campaign at Fayette Unit #3 took place in March and April of 2007. Tasks 4 and 5 were performed by Alstom-PPL throughout the performance period to support the entire demonstration program. The EERC and LCRA also provided input to the tasks.

Tasks 2 and 3 at Shawville Unit #3 and Task 3 at Fayette Unit #3 were never performed due to funding constraints.

Field Demonstration Mercury Capture Results

The first performance goal for this project was at least 90% reduction in uncontrolled stack emissions of mercury. The mercury capture results for Fayette Unit #3 are shown in Figure i-1. Two of the four sorbents tested achieved 90% mercury reductions, and the other two sorbents were not tested at injection rates above 0.5 lb/MMacf because of time constraints. In the ash evaluation tests, Foam



Index Test measurements indicated that the ash generated with sorbent injection to achieve 90% mercury reductions would likely be unsuitable for concrete applications. Figure i summarizes the parametric test mercury reduction results, the maximum mercury reduction demonstrated, and the maximum mercury reduction possible while maintaining ash quality for sale.

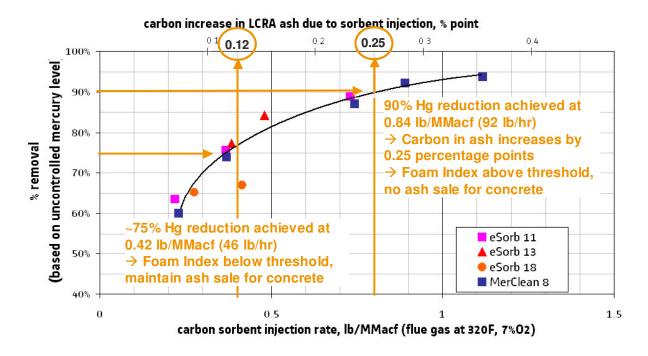


Figure i-1 Mercury Capture Results for Demonstration Project Host Sites

Note that mercury reductions above 90% were demonstrated, achieving the original project goals. Mercury reductions of 75% were also demonstrated while maintaining fly ash quality for concrete applications.

Engineering System Analysis and Economics

The second performance goal was achieving 90% mercury reduction at a mercury removal cost significantly lower that 50% of the assumed baseline cost of \$60,000/lb mercury removed.

The mercury removal costs in \$/lb Hg removed are summarized in Figure i-2. The figure shows very low removal costs for 75% reduction because only 46 lb/hr sorbent is required and testing indicated that the fly ash sale for concrete could be maintained. The removal costs at 90% reduction levels are significantly higher. Sorbent costs are doubled, because 92 lb/hr sorbent flow is required to achieve 90% reduction, but the inability to sell ash for concrete accounts for the majority of the cost increase. Even with the ash effects, the removal costs at 90% reduction are over 70% below the baseline cost of \$60,000/lb Hg removed at sorbent costs below \$2.00/lb. Even at \$2.00/lb sorbent, the removal cost is \$18,700/lb or 68.8% below \$60,000/lb Hg removed.

Mercury levels were reduced 90% at a cost at least 68.8% below the baseline cost, achieving the second performance goal for this project.



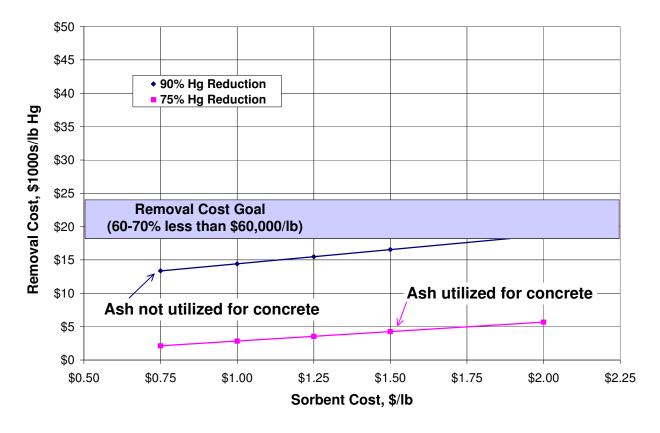


Figure i-2 Mercury Removal Costs for Demonstration Project Host Sites

Balance of Plant Impacts

Observations throughout the testing suggest that injection of Mer-Clean™ sorbent did not increase the stack opacity during long-term testing. Ash leaching test results indicated that no detectable mercury leached from the fly ash.

The unburned carbon in the fly ash increased, and the increase was related to the amount of sorbent injected. At 46 lb/hr sorbent injection rate, unburned carbon in the ash increased by 0.12 percentage points. At 92 lb/hr sorbent injection rate, unburned carbon in the ash increased by 0.25 percentage points.



1 Introduction

Various methods for mercury control in coal-fired power plants are being developed and demonstrated to meet current and impending mercury emission regulations.^{1, 2} These technologies range from activated carbon (AC) injection, coal and flue gas additives, catalytic and electro-catalytic mercury oxidation with subsequent capture in scrubbers, and in-situ mercury sorbent generation from coal. Among these, powdered AC injection is one of the more mature technologies for mercury control.

In previous Round 1 demonstration projects, it was observed that very high injection rates (6-15 lb/MMacf) ¹ of plain AC were needed to achieve reasonable levels of mercury removal, particularly for low-chlorine containing sub-bituminous (Powder River Basin or PRB) and lignite coals. The low removal levels could be ascribed to a high proportion of elemental mercury in the flue gas when firing these coals. Researchers and technology developers have since used halogenated carbon sorbents to address this problem, with improved performance compared to plain AC (3.3 to 6 lb/MMacf injection rate for 90% removal) ^{1, 2}

Alstom was awarded a DOE/NETL-sponsored project (DOE Cooperative Agreement DE-FC26-07NT42776) to perform full-scale demonstration of its Mer-Cure™ technology in two coal-fired boilers burning coals of various ranks. In the full-scale demonstration program, Alstom-PPL teams up with the Lower Colorado River Authority (LCRA), Reliant Energy, and the University of North Dakota-Energy and Environmental Research Center (UND-EERC).

The overall objective of the work is to perform two- to six-month, full-scale demonstrations of Mer-Cure™ technology in two coal-fired boilers. The goals of the Mercury Round 3 program, as originally defined by DOE/NETL in the solicitation, are to demonstrate that Mer-Cure™ technology can achieve:

- Greater than 90% reduction of gaseous mercury in the flue gas;
- At a mercury control cost significantly lower than 50% of \$60,000/lb mercury removed.

Alstom-PPL also collected performance data to evaluate the long-term impact of mercury control on the operation of power plants and overall plant economics, including information desired by LCRA on the maximum mercury reduction possible while maintaining the ability to sell their fly ash for concrete applications.

The two test sites selected for the demonstration program are LCRA's Fayette Unit #3 and Reliant Energy's Shawville Unit #3. Detailed descriptions of the two proposed host sites are provided in Table 1.1-1. LCRA's Fayette Unit #3 is located in La Grange, TX, and fires PRB coal blends. The unit has a 480-MW_e boiler and is equipped with two Ljungstrom™ air heaters, two electrostatic precipitators (ESPs) and followed by a wet scrubber system with three modules. The ESP ash from the unit is currently sold to ash marketers. In the Fayette Unit #3 test campaign, Alstom-PPL will address the effect of sorbent injection on ash utilization.

Reliant Energy's test site is Shawville Unit #3 in Shawville, PA. The unit is a 190-MW $_{\rm e}$ (gross) boiler with two Ljungstrom air heaters and two ESP's for particulate control. The unit currently fires



medium-sulfur Eastern bituminous coals with average sulfur content of 2%. The mercury level at the stack ranged from 15 to 20 mg/Nm3. In the Shawville Unit #3 campaign, Alstom-PPL will focus on the effect of SO₃ in the flue gas on mercury capture.

The EERC will provide mercury measurement expertise. Continuous mercury monitors (CMMs) will be used throughout the test period to provide both elemental and oxidized mercury concentrations in the flue gas. Ontario Hydro method will be used for some of the key test conditions to verify CMM data.

Table 1.1-1 Host Site, Coal and Emission Data for the Field Demonstration

	LCRA	Reliant Energy
Unit	Fayette Unit #3	Shawville Unit #3
Location	La Grange, TX	Shawville, PA
Capacity (MW _e Gross)	480	190
Operation	Baseload	Cycling
NO _x and SO ₂ control	Low NOx firing/	Low NOx
	wet scrubber	firing/SNCR
Air Heater	Two Ljungstrom™	Two Ljungstrom™
Particulate control	Two ESP's	Four ESP's
(SCA in ft ² /kacfm)	(640)	(312 total)
Ash utilization	Sold	Landfilled
Coal	PRB blends	E. bituminous coal
Higher Heating Value,	8,000-8,500	11,000-13,000
As-rcvd (lb/MMBtu)		
Sulfur (%)	0.3-0.5	1.8-2.2
Ash (%)	4-6	12-15
Cl in coal (ppmwd)	20-50	720-1320
Hg in coal (ppmwd)	0.1-0.2	0.3-0.63
Uncontrolled Hg Stack	11.4 (total),	18 – 22 (total)
(Hg ^T , Hg ^p , Hg ^{ox} , Hg ^{el})	(97% as elemental	(50-70% oxidized)
(μg/Nm³)	Hg)	
Carbon-in-ash (%)	0.18	4-6
Flue gas temperature	310	280
(ESP Inlet,°F)		

Each of these boilers is equipped with a cold-side electrostatic precipitator (ESP) for particulate control. This report gives a summary of the demonstration testing at Fayette Unit #3, the only site tested because of budgetary constraints of the Mercury Round 3 program.

Mer-Cure™ is a treated sorbent-based mercury control technology. This technology has been developed based on a mechanistic understanding of the critical areas that need to be augmented for mercury oxidation and capture and scaled up from well-controlled laboratory-scale testing, pilot-



scale testing, and through a short-term demonstration at a 220 MW_e pulverized coal-fired boiler. Alstom is also developing other technologies for mercury control in coal-fired boilers, including coal additives (KNX[™]) and traditional carbon injection with the addition of a polishing fabric filter ⁶

1.1 Background

In Alstom's Mer-Cure™ mercury control technology, a small amount of sorbent is injected into the flue gas stream for oxidation and adsorption of gaseous mercury. The sorbents are activated carbon-based and prepared with chemical additives that promote oxidation and capture of elemental mercury. The technology is unique in that the sorbent is injected into a high temperature (above 700°F) environment where the mercury oxidation/removal kinetics are accelerated.

Mer-Cure™ mercury control technology applied to coal-fired power generation has the potential to be a cost-effective mercury control technology for the entire range of coals (bituminous, subbituminous, and lignite). As the technology is based on oxidation and adsorption of mercury, it is also applicable to all air pollution control configurations including wet scrubber and spray dryer-ESP/baghouse units. The main focus of this project, however, is with a cold-side ESP as the particulate control device. Cold-side ESPs represent over 70% of the coal-fired boilers in the United States. The mercury control technology has low-capital costs (\$5-10/kW_e). It also requires a very small amount of low-cost additives for treatment, which results in low operating costs (0.5-0.75 mills/kWh) and minimal balance-of-plant (BOP) impact.

The Mer-Cure™ system takes advantage of a mechanistic understanding of mercury oxidation and capture to identify the critical areas that need to be augmented to maximize mercury oxidation and capture. The approach for enhanced mercury capture is:

- Pre-treatment of activated carbon (AC)-based sorbent with proprietary additive and its
 injection into the high temperature region (400 to 800°F). This activates the additive on the
 carbon and makes it reactive with elemental mercury;
- Accelerated reaction (due to the high temperature) of adsorbed mercury with the activated additive to an oxidized product and its retention; and
- Injection methodology that ensures dispersion and uniform distribution of sorbent in the flue gas and minimizes mixing limitation;
- Capture of sorbent particle with adsorbed mercury in an ESP or fabric filter.

Mer-Clean™ is a carbon-based sorbent with a proprietary additive that accelerates mercury oxidation by providing the oxidation reaction partner in high concentrations on the carbonaceous particle. Mer-Cure™ technology has also been successfully applied to alternative commercially available sorbent products.

More importantly, Mer-Cure™ technology involves injection at an "ideal" temperature location, which allows contact with the elemental mercury to begin as early as it can be effective. This is typically in the duct leading to the air heater. Higher temperature also increases reaction rates exponentially. This enables maximum utilization of the entire effective temperature range for oxidation with the added sorbent. In contrast to the Mer-Cure™ system, current practice involves



the injection of carbon sorbents at low temperatures downstream an air heater, where typical temperatures are between 250 and 350°F. In this case, the contact time between the sorbent and the mercury in the flue gas is significantly shorter and the added benefit of accelerated kinetics at a higher temperature is not available.

In connection with injection at higher temperatures is the unique and proprietary nature of Mer-Clean™. Careful selection of the additive and method by which it is administered to the sorbent ensures that the sorbent-additive combination becomes active and retains its integrity through the high temperature range. In addition, Mer-Cure™ technology has also been successfully applied to alternative commercially available sorbent products.

Finally, detailed computational flow modeling of the injection region is performed to match the injection system to local flow conditions. In this manner, uniform distribution of sorbent is achieved before the flue gases enter the air heater. Sorbent distribution is critical to minimize the amount of sorbent required for a given performance level.

1.2 Project Overview

The full-scale demonstration program originally included test campaigns at two host sites: LCRA's 480-MW_e Fayette Unit #3 (FPP3) burning PRB blends, and equipped with an ESP and a wet scrubber; and Reliant Energy's 190-MW_e Shawville Unit #3 burning Eastern bituminous coal, and equipped with two Ljungstrom air heaters and two ESP's in series. The demonstration program as proposed included a two- to five-week short-term field test followed by two-month long-term demonstration for each of the two selected sites.

Project funding was approved by the DOE/NETL for budget period 1. During budget period 1, a short-term demonstration at the LCRA's Fayette Unit #3 was performed. Unfortunately, the remainder of the Mercury Round 3 project was cancelled due to budgetary constraints. Therefore, the only tests performed under this project were the short term tests at Fayette Unit #3. Table 1.2-1 shows the overall program short-term and long-term testing schedule as originally defined, with the cancelled tasks in red. A detailed description of the statement of work actually performed is included for completeness in Section 3.

 Site
 2007
 3008

 Fayette Unit #3
 3008

 Fayette Unit #3
 3008

 Shawville Unit #3

Table 1.2-1 Overall Program Schedule



2 Project Objectives

The overall objective of the proposed work is to develop a mercury control technology in coal-fired boilers that can achieve more than 90% mercury capture. In Alstom's Mer-Cure™ mercury control technology, sorbents with chemical additives that promote oxidation and capture of elemental mercury are injected into an environment where the kinetics are favorable. Installation of Mer-Cure™ technology has low-capital costs (\$5-10/kW_e). The mercury control technology also requires a very small amount of sorbent, which results in low operating costs (0.5-0.75 mills/kWh) and minimal Balance of Plant (BOP) impact.

The goals of the program, as originally defined by DOE/NETL in the solicitation, are to demonstrate that Mer-Cure™ technology can achieve:

- Greater than 90% reduction of gaseous mercury in the flue gas;
- At a mercury control cost significantly lower than 50% of \$60,000/lb mercury removed.

Alstom-PPL also collected performance data to evaluate the long-term impact of mercury control on the operation of power plants and overall plant economics, including information desired by LCRA on the maximum mercury reduction possible while maintaining the ability to sell their fly ash for concrete applications.

Alstom-PPL is committed to offer the most economically viable mercury control solution to our utility customers while minimizing the impact on the overall power plant economics, and Mer-Cure™ technology offers a great opportunity for utility companies to control mercury in the most cost-effective manner. Alstom-PPL is committed to commercialize the technology and believes that the completed full-scale demonstration of the technology was a critical step to successful commercialization.



3 Statement of Work

In order to accomplish the technical objectives, a technical approach was utilized to effectively demonstrate and evaluate Alstom-PPL mercury control technology. Testing at each demonstration site was planned to include both a five week short-term and two month long-term, full-scale test campaign. The work required for the test campaigns can be broken down to the following tasks:

- Task 1. Design, Engineering and Fabrication of System
- Subtask 1.1 Design and Engineering of System Architecture
- Subtask 1.2 Component Fabrication and System Assembly
- Task 2. Short-Term Demonstration
- Subtask 2.1 Project Planning
- Subtask 2.2 Design, Fabrication and Maintenance of Site-Specific System
- Subtask 2.3 Installation of Mobile Mer-Cure[™] System
- Subtask 2.4 Parametric Testing
- Task 3 Long-Term Field Demonstration
- Subtask 3.1 Installation, Checkout and Commissioning
- Subtask 3.2 Field Testing and Measurement
- Subtask 3.3 System Removal
- Subtask 3.4 Data Analysis and Site Report
- Task 4. Technology Transfer
- Task 5. Program Management and Reporting.

Alstom-PPL engineers at Alstom-PPL's Windsor site performed most of the Task 1 work. Tasks 2 and 3 were to be repeated for both host sites. Alstom-PPL engineers performed Task 2 with the assistance of host site personnel and EERC measurement crew. The five week short-term test campaign at Fayette Unit #3 took place in March and April of 2007. Tasks 4 and 5 were performed by Alstom-PPL throughout the performance period to support the entire demonstration program. The EERC and the participating utility companies also provided input to the tasks.

Tasks 2 and 3 at Shawville Unit #3 and Task 3 at Fayette Unit #3 were never performed due to funding constraints. Therefore, Task 3 is not described in the following section.

3.1 Tasks Performed

Detailed description of the tasks and subtasks listed in Section 3 are presented in the following paragraphs.

<u>Task 1</u> Design, Engineering and Fabrication of System

Alstom-PPL has a mobile Mer-Cure™ unit developed under Mercury Round 2 that was employed for the Round 3 demonstration. In this task, the system was modified and upgraded to provide adequate capacity and enhanced performance for the project.



<u>Subtask 1.1</u> Design and Engineering of System Architecture

The current mobile Mer-Cure™ unit was designed for shorter term testing at smaller units of up to 220 MW_e. The testing at LCRA's Fayette Unit #3 required treatment of the entire flue gas stream coming from a 480-MW_e boiler. This was accomplished by designing an upgraded the transport air system. A blower system was designed as part of the upgraded transport air system for increased pneumatic capacity. This allowed injection of sorbent into a larger volume of flue gas while not requiring any additional electrical services installed at the host sites.

Subtask 1.2 Component Fabrication and System Assembly

System modifications and upgrades identified in Subtask 1.1 were implemented into the existing mobile unit. Most of the system components were fabricated and assembled at Alstom – PPL's site in Windsor, CT. The system design was modular so that maintenance and modification, if necessary, could be made quickly.

Task 2 Short-Term Demonstration

In this task, Alstom-PPL collected short-term performance data of our mercury control technology from the host site boilers for evaluation by DOE/NETL and the team. The task was further broken down to the following.

Subtask 2.1 Project Planning

The success of the project was very dependent on close and smooth coordination with various team members throughout the performance period. In the planning stage, Alstom-PPL communicated closely with the plant personnel of the host sites and the mercury measurement crew of the EERC.

Alstom-PPL held meetings with plant and corporate personnel to design and discuss the detailed test plan. In these meetings, any potential operational issues of the plant and plant equipment were thoroughly discussed. Also, plant data acquisition was coordinated with the operational crew. Alstom worked with the environmental personnel to help obtain permits for the testing from the state Department of Environmental Protection. Alstom also coordinated closely with EERC for a sound measurement plan. The EERC provided personnel and equipment to obtain continuous mercury concentrations using two CMMs and Ontario Hydro (OH) measurements upstream of injection locations and downstream of the ESPs and/or scrubber. The team members collaborated to develop a QA/QC plan on the overall test program.

Subtask 2.2 Design, Fabrication and Maintenance of Site-Specific System

The mobile Mer-Cure™ system developed under Mercury Round 2 was employed for this task. As part of the task, the mobile Mer-Cure™ system was modified to accommodate site-specific circumstances. For example, the injection system, e.g., lances and nozzles, were designed based on computational fluid dynamics (CFD) studies of the flue gas flows in the sorbent injection area.

The subtask began with an initial visit to the host sites for preliminary site evaluation. At the sites, Alstom-PPL engineers and site personnel thoroughly reviewed plant arrangements, accessibility of testing-related sections of the plant, site operations, and other plant data available. This initial



review of site-specific information led to an optimized design of the mobile Mer-Cure™ system, the test plan, and its ensured adequacy and integrity at the specified sites. Any safety issues were also addressed.

Subtask 2.3 Installation of Mobile Mer-Cure™ System

The mobile Mer-Cure™ system, mounted on a 48-ft trailer, was shipped and installed at the test site. Any required plant modifications (such as installation of injection/sampling ports or catwalks, etc.) were completed during plant outage prior to testing. Structural and electrical interfaces with the boiler were made. Finally, system shakedown was conducted to ensure system operability.

While the Mer-Cure™ system was installed, the EERC crew set up the CMMs at the predetermined locations. For continuous measurement of gaseous mercury, the flue gas was sampled with inertial separation probes (ISP) for removing flyash and analyzed by CMMs such as PS Analytical, Tekran, and Horiba. The CMM assembly was housed in a clean, temperature-controlled environment. Operated under a well-developed QA/QC plan, they provided rapid and accurate measurements at various test conditions.

Subtask 2.4 Parametric Testing

Once the preparations were completed, Alstom-PPL conducted the test campaign by first performing baseline mercury measurement for a week, followed by four weeks of parametric testing.

In order to obtain comprehensive data for go/no go decision, a test program was constructed as listed in Table 3.1-1. During the first week, uncontrolled mercury levels were established. The next week was dedicated to testing at a range of feed rates (0.5 to 5 lb/MMacf) with Mer-Clean™ sorbents. The test conditions factored in variations in boiler load and flue gas flow, and selected conditions were repeated for data quality control and assurance.

During parametric testing for a given test condition, sorbent injection was performed over an 8-12 hour period, and the mercury concentrations from the unit allowed to recover for the subsequent 12-16 hours before the next test condition. The feed rates were varied to achieve several target removal efficiencies, e.g., 70%, 90%, and 95% based on baseline mercury level.

		CMM	Coal and Ash
Activity	Duration	Measurements	Sampling
Baseline measurement	1 week	AH inlet/stack	Daily
Parametric Testing	1 week	AH inlet/ESP outlet	Daily
Ash Evaluation Testing	2 weeks	AH inlet/ESP outlet	Daily

Table 3.1-1 Summary of Short-Term Demonstration Plan

One of the objectives of LCRA's Fayette Unit #3 testing was to determine the maximum level of mercury capture possible while still permitting continued ash sales. In order to achieve this, an injection rate was selected that allowed this, and then "ash evaluation" testing was conducted for



the subsequent two weeks. Towards the end of the ash evaluation testing, a large quantity of ash from the ash silo was collected for detailed ash characterization testing by LCRA and their ash marketers.

OH measurements for mercury concentration and speciation were performed for selected conditions upstream of the sorbent injection location and at the ESP outlet to substantiate the mercury CMM data. A bulk quantity of by-product samples was also collected by plant personnel as directed by the Program Solicitation. Representative 5-gal bucket samples were taken from the ESP as well as from the scrubbers over the course of the test campaign. These were sent out to DOE contractors for by-product characterization. Independent analysis was also carried out by Alstom-PPL and EERC for selected samples. The analysis of collected samples included mercury content, LOI, leachability and foam index tests. The plant operating data were obtained from the plant personnel. These included the temperature profile, oxygen concentration profile, SO₂, NO_x, and stack opacity at the exit of the ESP.

Throughout the testing, the test crew adhered to procedures that ensured tight QC and QA. The CMM system was calibrated before and after tests with zero and span drift checks and OH measurements were made for selected test conditions to ensure QA/QC.

<u>Task 4</u> Technology Transfer

A number of utility companies have shown a great interest in the Mer-Cure™ mercury control technology because of its potential as a low capital and operating cost option for mercury control. In addition, under Mercury Round 2 it delivered high performance with respect to one of the most difficult configurations: low rank coals (PRB and lignite) with an ESP. Alstom-PPL disseminated project results to the power generation industry by attending and making presentations at Electric Power 2008 and A&WMA's 2008 Mega Symposium. An accompanying technical paper was also written for publication at A&WMA's 2008 Mega Symposium.

<u>Task 5</u> Project Management and Reporting

Throughout the project, the project manager for Alstom-PPL communicated closely with DOE's program manager and other team members. Close contact was maintained with the DOE project manager to report any progresses or issues as well as to obtain feedback and overall direction. Quarterly progress reports were submitted as required under the solicitation.



4 Technology Description

4.1 Mer-Cure™ Process

Alstom's Mer-Cure™ technology (Figure 4.1-1) employs a patented sorbent preparation and injection system that has several unique features to enhance its mercury control performance. First, sorbent injection is at an "ideal" temperature location (600-800°F), which allows contact with the elemental mercury to begin as early as it can be effective. Unlike more conventional AC injection systems, the Mer-Cure™ system injects the sorbent into the duct upstream of the air heater as shown in Figure 4.1-1. This enables maximum utilization of the entire effective temperature range for oxidation with the added sorbent. In current practice, carbon sorbents are often injected at low temperatures, such as downstream an air heater (typically between 250 and 350°F), where oxidation kinetics are significantly slower. Also, the sorbent-mercury contact time becomes significantly shorter.

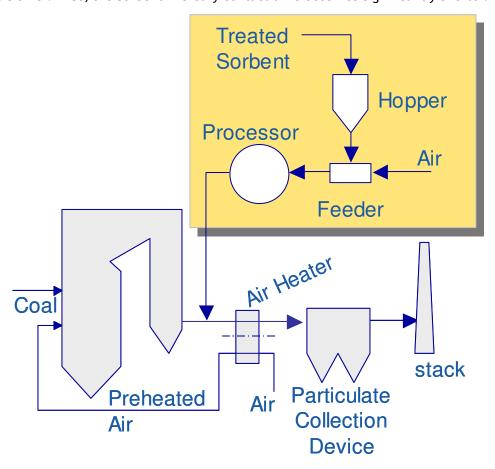


Figure 4.1-1 Alstom's Mer-Cure™ Technology for Mercury Capture

Second, the Mer-Cure™ system utilizes a carbon-based sorbent (called Mer-Clean™) with a proprietary additive that has been designed to accelerate mercury oxidation. Carefully designing the additive and its impregnation method to the sorbent ensures that Mer-Clean™ sorbent is active and retains its integrity throughout the temperature range. Additionally, Mer-Cure™ technology has been successfully applied to alternative commercially available sorbent products.



A third distinct feature of Mer-Cure™ is the unique injection methodology. The system is designed to remove any mass transfer limitations of mercury vapor to sorbent by de-agglomerating the sorbent into individual particles and uniformly distributing them to flue gas flow based on local flow conditions. CFD tools are employed to guide optimum design of the injection system, as sorbent distribution is critical to minimize its consumption for a given performance level.

4.2 Mobile Mer-Cure™ Demonstration System

As part of the Mercury Round 2 test program, Alstom designed and fabricated a mobile Mer-Cure™ system. The mobile Mer-Cure™ demonstration system (Figure 4.2-1) is composed of three components mounted on a 48-foot trailer: a sorbent storage system, a sorbent processing/delivery system, and a sorbent distribution system.



Figure 4.2-1 Mobile Mer-Cure™ Demonstration Trailer Schematic



For this test program, this mobile unit has been redesigned for the Round 3 program to enable a longer-term demonstration at larger units. The two-month testing proposed at LCRA's Fayette Unit #3 required treatment of the entire flue gas stream coming from a 480-MW_e boiler. This was accomplished by upgrading the transport air system. In preparation for the test campaign at LCRA, a blower system was employed in an effort to upgrade the transport air system for increased pneumatic capacity. The added blowers are shown in Figure 4.2-2. This allowed injection of sorbent into a larger volume of flue gas while not requiring any additional electrical services installed at the host sites.

Figure 4.2-2 Mobile Mer-Cure™ System Upgraded Transport Air Blowers

The sorbent storage system is a portable two-piece silo, which can be easily assembled and requires a relatively small footprint. The sorbent storage system, when assembled, is capable of loading powdered material of up to three 900-lb super-sack bags at the same time and will allow



uninterrupted operation for 24 hours at a typical injection rate. Due to height limitations of the trailer during transportation, the storage system is delivered unassembled. The top piece of the two-piece storage system is attached to the base piece at the test site, as shown in Figure 4.2-3.



Figure 4.2-3 On Site Assembly of Mobile Mer-Cure™ Demonstration Trailer

The sorbent processing/delivery system consists of a variable screw feeder for metering the sorbent and an eductor for its pneumatic transport, a processor that de-agglomerates sorbent particles, and a compressed air system that supplies dry air for pneumatic transport. This system is mounted next to the storage system and is completely connected to the other subsystems.

The sorbent distribution system consists of flexible hoses, interconnecting pipes, distribution manifolds and injection lances. The injection lances designed based on CFD analysis of the duct flow conditions. They typically are 1 ½-inch pipes with multiple nozzles for even sorbent distribution throughout the duct cross-section.

The process equipment trailer is also equipped with control and data acquisition systems for the Mer-Cure™ system. The control system allows remote monitoring of operating conditions of the three Mer-Cure™ subsystems. It also automatically shuts down the system in case of a boiler trip or internal system failure. The control system is designed to receive load signals from the boiler and to vary sorbent injection rate as a function of boiler load. The data acquisition system stores system operating data such as injection rates, pressure levels of various sections of the sorbent delivery, processing and injections systems, and temperatures.

Appropriate distribution of sorbent into the flue gas stream is critical for maximum contact between the sorbent and the mercury in the flue gas stream. Flow modeling studies using Fluent CFD package were conducted using specific boiler design data to better determine the location and the number of injection lances.



Based on calculated flow distributions, the number of injection ports, diameter and length of injection lances, and the number, orientation and size of injection lance nozzles were determined. The injection system was designed to proportionally distribute sorbent with flue gas volumetric flow.

The completed mobile Mer-Cure™ demonstration trailer and custom fabricated injection system was shipped to each site. The sorbent storage and delivery system was assembled at the site; the sorbent processing system was placed close to the sorbent injection system and was connected to the storage system by 4-inch flexible hoses. The sorbent distribution system was installed at the sorbent injection location and connected to the delivery system by 1.25-inch flexible hoses. The mobile Mer-Cure™ system is shown installed at the Fayette test site in Figure 4.2-4.



Figure 4.2-4 Moblie Mer-Cure™ Trailer Installed at Fayette Unit #3



5 Test Planning and Mercury Instrumentation

In general, the test program at each site consisted of three major components: baseline testing, parametric testing and long-term testing. During baseline testing, the existing mercury emissions were characterized for the unit, and the extent of native mercury capture was documented. During parametric testing, the sorbents to be evaluated were injected at a constant rate for 8-16 hours. For a given sorbent, several sorbent flow rates were tested to develop a parametric curve of sorbent injection rate versus mercury capture performance. For example, a sorbent could be injected for 12 hours each at 0.5, 1.0, 1.5 and 2.0 lb/MMacf injection rate. That set of tests would constitute a parametric test run for a given sorbent. After roughly 12 hours without sorbent injection to allow the boiler to recover and return to its baseline condition, a parametric test run was performed on a second sorbent. Four different sorbent formulations were tested parametrically at Fayette Unit #3.

In addition to documenting 90% mercury capture, one of the objectives of LCRA's Fayette Unit #3 testing was to determine the maximum level of mercury capture possible while still permitting continued ash sales. The two best performing sorbents from parametric testing were used for ash evaluation testing.

5.1 Mercury Measurements

Mercury concentrations in the flue gas (including speciation) were determined using CMMs and the OH method at (i) the air heater inlet, upstream of sorbent injection and (ii) the stack, downstream of the ESP (and wet scrubber, if present) after the particulate has been removed from the flue gas. The University of North Dakota – Energy and Environmental Research Center (UND-EERC) used PS Analytical and Tekran CMMs to perform the on-line mercury concentration measurements. In this set-up, the flue gas is sampled through an inertial separation probe (Figure 5.1-1) to remove any suspended particulate matter, before it is directed to a wet conditioning system and subsequently to a mercury analyzer.

The inertial separation probe enables a particle-free stream to be sampled from the dust-laden flue gas, while minimizing the contact of the flue gas with suspended particulate during the sampling process. This is particularly important at the inlet sampling location where a high concentration of ash is present compared to the stack where a relatively dust-free flue gas is present. Ash components and unburned carbon in the ash can modify the chemical speciation of mercury or adsorb mercury species during the sampling process upon intimate contact with the sampled gas. Such contact would occur, for example, when the sampled gas passes through a layer of ash collected on filters within the sampling system.

The mercury analyzer consists of a cold-vapor atomic adsorption spectrometer (CVAAS) coupled with a gold amalgamation system. The system is calibrated using vapor phase elemental mercury. The analyzers are capable of measuring both vapor-phase elemental and total mercury. The analyzer determines total vapor-phase mercury concentrations by reducing all of the oxidized mercury to the elemental form in the wet conditioning system. To measure elemental mercury, the oxidized mercury



species are removed in the conditioning system, while the elemental mercury is not removed by the conditioning system.



Figure 5.1-1 Inertial Separation Probe at Air Preheater inlet CMM Location

The results of the CMMs were validated by selective measurements using the OH method performed at the same location. With the OH method, flue gas is sampled through a filter (to capture particulate) and an impinger train that is designed to first capture oxidized mercury in the first three impingers (followed by a blank impinger) and then capture elemental mercury in the next three impingers (followed by a moisture removal impinger). A schematic of an OH sampling train is shown in Figure 5.1-2. Analysis of the filter and the impinger solutions provides speciated mercury concentration data.



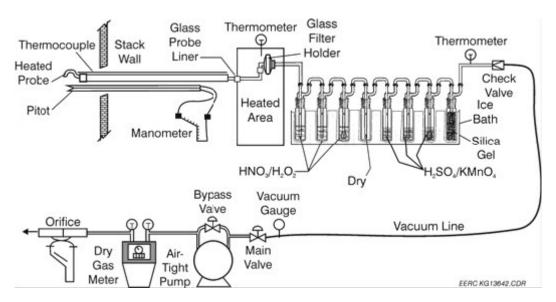


Figure 5.1-2 Ontario Hydro Sample Train Schematic

In addition to sampling for mercury concentrations using CMM and OH methods, samples of coal (from the feeders) and fly ash (from ESP hoppers) were obtained on a regular basis. The coal samples were analyzed for mercury and chlorine content in addition to obtaining their proximate and ultimate analysis. Ash samples were analyzed for unburned carbon content, and selected ash samples underwent foam index testing. Boiler operation data, including load, excess air, ESP operation and air heater outlet flue gas temperature were also collected throughout the test period.



6 Fayette Unit #3 Demonstration Program

This section gives a detailed summary of the field demonstration program Fayette Unit #3 In addition to DOE/NETL's project goals under the original solicitation, another objective of LCRA for Fayette Unit #3 testing was to determine the maximum level of mercury capture possible while still permitting continued ash sales.

6.1 Site Description and Test Preparation

LCRA's Fayette Unit #3 is located in La Grange, TX, and fires PRB blends. The unit is a 480-MW_e tangentially-fired, pulverized coal (PC) boiler and is equipped with low NO_x burners, Ljungstrom™ air heaters, and two electrostatic precipitators (ESP), followed by a wet scrubber system with three modules. The ESP ash from the unit is currently sold to ash marketers. The specific collection area (SCA) of the ESP is 640 ft²/kacfm. Table 1.1-1 previously showed the unit configuration and coal and emissions data for Fayette Unit #3.

Fayette Unit #3 was chosen for this evaluation because it fires blends of PRB coal and is equipped with an ESP. Low rank coals such as lignite and PRB coals have been shown to be difficult to treat for mercury removal in past demonstration programs. ^{1, 2}

A coal-fired boiler firing low rank coals with a cold-side ESP has emerged as a preferred combination in the US utility industry for several reasons. Utilities are switching to these lower sulfur Western coals to lower SO_2 emissions. Second, when the boilers are operated in a low- NO_x configuration (air staging), very low emissions can be achieved with these coals (< 0.1 lb/MMBtu), particularly with tangentially-fired boilers. Fuel cost, which is the predominant operating cost for coal-fired power plants, is also lower with the use of low rank coals. These coals are low cost because they are typically surface mined with low mining costs.

Previous tests at plants with PRB coal and a cold-side ESP using plain activated carbon sorbents indicated that the mercury removal was limited to about 70 percent. For example, in a Round 1 mercury control technology demonstration sponsored by US DOE, testing was conducted at We Energies Pleasant Prairie power plant, also a PRB-fired unit equipped with a cold-side ESP. The average mercury control efficiency ranged from 46 percent at an injection rate of 1.0 lb/MMacf to 50-55 percent at an injection rate of 5.0 lb/MMacf. Increasing the injection rate to 10 lb/MMacf increased mercury capture across the ESP to 60-65 percent. Almost no further increase was observed at higher injection rates. ^{1, 2}

In preparation for the design of the site-specific portion of the Mer-Cure™ system, site visits were made to Fayette Station. During the visits, more detailed information was collected such as that on the injection location (e.g., duct dimensions, turning vane arrangement), workspace, sampling port locations, trailer placement, equipment placement and the availability of utilities at work locations.

Extensive CFD studies were conducted to design the injection lances for the Mer-Cure™ system. The flue gas stream from Fayette Unit #3 boiler is split into two streams (see Figure 6.1-5). Based on the calculated flue gas flow distributions, the numbers of injection ports, injection lances and their nozzles were determined. LCRA completed installation of the injection ports in late 2006.



Appropriate distribution of sorbent into the flue gas stream is critical for maximum contact between the sorbent and the mercury in the flue gas stream. Flow modeling studies using the Fluent CFD package were conducted using specific boiler design data to better determine the location and the number of injection lances. Mer-Cure™ technology requires injection of sorbent into a location upstream air heaters. In order for maximum in-flight capture of mercury, the mixing characteristics of sorbent with flue gas at the boiler exit have been carefully examined by CFD.

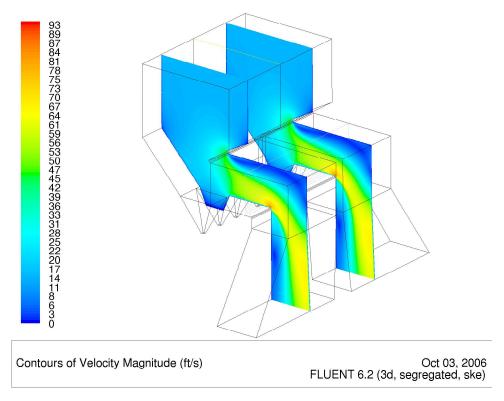


Figure 6.1-1 Flue Gas Flow Distribution at Fayette Unit #3 Injection Location

As shown in Figure 6.1-1, the flue gas flow at Fayette Unit #3 is concentrated more towards the floor of the horizontal section of the economizer outlet. A recirculation zone has been built at the top half of the section. This mal-distribution was been taken into account in designing the injection lances. More injection ports were placed at the bottom half of the injection lances, for example, so that the injected sorbent loading was uniform throughout the cross section of the flue gas ducts.

The CFD tool has also been used for other aspects of field testing preparation. Impact of installing and operating sorbent injection lances upstream of the O_2 probes have been investigated, as shown in Figure 6.1-2. From the CFD work, it was concluded that transport air would not affect O_2 readings by the existing probes.

Sampling locations were also identified using the tool. After evaluating various sampling options for the FGD geometry shown in Figure 6.1-3, the FGD outlet CMM was located at the stack.



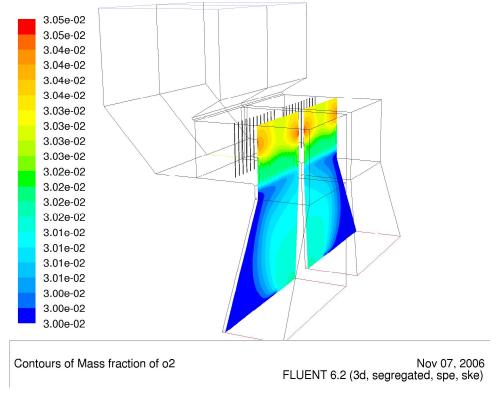


Figure 6.1-2 O2 Distribution at O2 Probe Location With Injection Lances in Operation

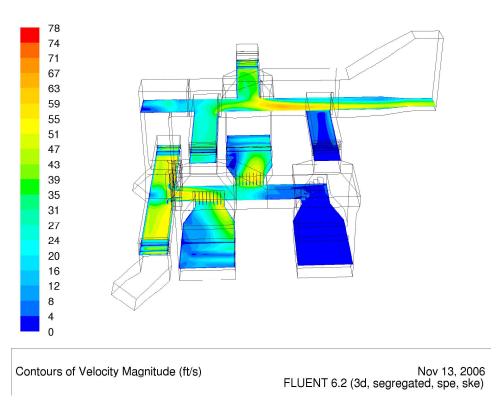


Figure 6.1-3 Flow Distribution around Fayette Unit #3 FGD Modules



Based on the CFD results and these flow distributions, a lance injection system was designed for Fayette Unit #3 targeting proportional distribution of sorbent with the flue gas volumetric flow. The lance design parameters such as the number of lances, nozzle size, number and location for each of the lances were determined based on these studies to create immediate and uniform mixing of sorbent as it is introduced into the flue gas stream. Lances for the demonstration project were fabricated according to the final design.

The sorbent storage and delivery system was assembled at the site; the sorbent processing system was placed close to the sorbent injection system and was connected to the storage system by 4-inch flexible hoses. The sorbent injection and distribution systems are shown installed at the air heater inlet in Figure 6.1-4, and were connected to the sorbent delivery system by 4-inch flexible hoses. Shakedown of the assembled system was carried out when installation was complete.

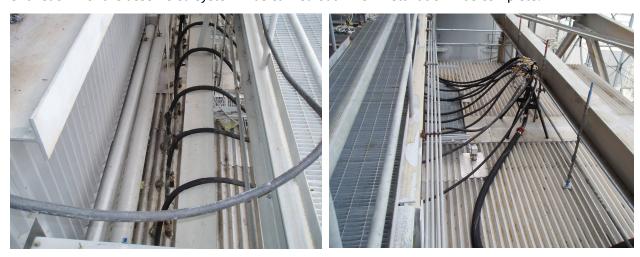


Figure 6.1-4 Fayette Unit #3 Injection Lances and Distributor Installed at AH Inlet

Figure 6.1-5 shows a schematic diagram of the Fayette Unit #3 layout and sampling locations for both gas and solids. As shown, the unit has two Ljungstrom™ air heaters in parallel. The particulate matter in the flue gas is collected by two ESPs downstream of the air heaters. The ash from the two ESPs is retrieved and collected in silo B by a pneumatic ash transfer system. The collected ash is currently sold for use in ready-mix concrete. Fayette Unit #3 has three wet scrubber modules, two of which are in operation for SO₂ control at any point in time. For flue gas temperature control, 15-25% of the total flue gas flow bypasses the scrubber modules.

For the test campaign, three CMMs were installed by the EERC at the Fayette Unit #3 plant: one upstream of the injection location (upstream of an air preheater), one upstream of the wet scrubber modules, and one at the stack. The CMMs were in operation throughout the program except the one upstream of the wet scrubber modules, which was on line only during the baseline test period. Coal samples were taken upstream of one of the eight (8) pulverizers on a daily basis. Fly ash samples were collected from various hoppers of the ESPs. Silo B ash was sampled every day. Wet scrubber samples were taken during baseline measurements.



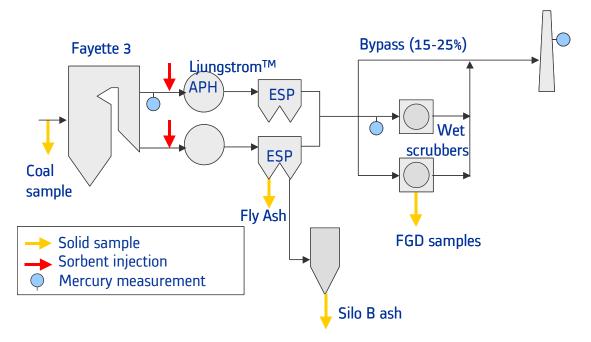


Figure 6.1-5 Sorbent Treatment and Sampling Locations, Fayette Unit #3

One of the unique aspects of Alstom's Mer-Cure™ technology is that the sorbent injection is performed between the economizer and the air heater. In typical utility boilers, temperatures at this location are around 600 to 800°F. The sorbents used in the Mer-Cure™ technology are activated carbon-based. Different halogenated components are added to the activated carbon and the material processed to make the final sorbent. An additional objective of LCRA for Fayette Unit #3 testing was to determine the maximum level of mercury capture possible while still permitting continued ash sales. To determine the effect on ash properties, four sorbent formulations were tested (Mer-Clean™ 8, eSorb™ 11, eSorb™ 13 and eSorb™ 18). These formulations differed from each other in terms of the added components, amounts of the added components, and manufacturing process conditions.

Fayette Unit #3 fires blends of PRB coals such as Black Thunder, Caballo, Jacobs Ranch, and others. Twenty-six (26) daily coal samples were collected during the test program, and Table 6.1-1 lists the average coal properties at Fayette Unit #3 as determined by analysis of these samples. The average mercury content of the coal and its standard deviation were 0.075 μ g/g dry coal and 0.010 μ g/g dry coal, respectively. The average chlorine content of the coal was 105 \pm 11 ppm on a dry basis.

The minimum and maximum mercury contents in the samples analyzed were 0.04 ppm and 0.10 ppm, respectively, indicating a fairly wide variation in the quantity of mercury entering the boiler with the coal.



Table 6.1-1 LCRA Fayette Unit #3 Average Coal Properties

Proximate analysis	As received	Dry basis
Total moisture (%)	29.4	N/A
Volatile matter (%)	30.38	43.03
Fixed Carbon (%)	35.36	50.08
Ash (%)	4.86	6.88
HHV (Btu/lb)	8,366	11,848
Ultimate analysis	As received	Dry basis
Moisture (%)	29.4	N/A
Carbon (%)	48.39	68.54
Hydrogen (%)	3.37	4.77
Oxygen (%)	12.93	18.31
Nitrogen (%)	0.73	1.03
Sulfur (%)	0.32	0.46
Chlorine (ppm)	74 ± 8	105 ± 11
Ash (%)	4.86	6.88
Total (%)	100.00	100.00
Hg (μg/g dry coal)	N/A	0.075 ± 0.010
Hg (lb/TBtu)	6.35 ± 0.89	6.35 ± 0.89

Table 6.1-2 lists the tasks carried out to complete the LCRA Fayette Unit #3 test campaign and their associated timeline.

Table 6.1-2 LCRA Fayette Unit #3 Test Campaign Timeline

Tasks	Timeline
Baseline Measurement	March 5-11, 2007
Parametric Testing	March 12-18, 2007
Ash Evaluation Testing	March 19-31, 2007
System Removal	April 9-12, 2007

Baseline measurements of mercury concentration in the flue gas were performed at the air heater inlet (pre-sorbent injection) and precipitator outlet. Validation data using the OH method was also obtained for the baseline condition. Subsequent to the baseline testing, approximately seven days of parametric testing was conducted. Parametric tests included four enhanced activated carbon sorbents. From parametric tests results, Mer-Clean™ 8 and eSorb™ 18 were chosen for the ash evaluation testing.

6.2 Baseline Testing Results

Baseline CMM measurements of gaseous mercury were made at various locations. Figure 6.2-1 shows a comparison between the measured gaseous mercury at the air heater inlet and the gaseous



mercury level calculated from mercury content in the fuel. The total mercury at the air heater inlet fluctuated between 8 and 16 μ g/m³ (corrected to 3% O_2) and averaged 11.7 μ g/m³, while the calculated values from the mercury content in the fuel fluctuated from 8.5 μ g/m³ to 13.2 μ g/m³. The two mercury concentrations generally agree with each other. The comparison suggests that the fluctuation of the mercury level at the air heater inlet is largely due to the variation of mercury content in the fuel. It also shows that there is no inherent mercury capture between boiler and air heater inlet.

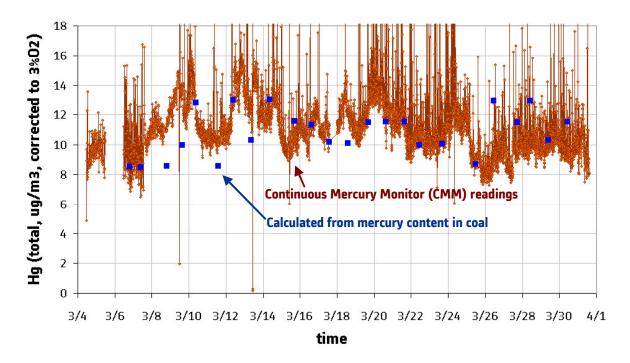


Figure 6.2-1 Comparison Between Coal Mercury Content and CMM Reading at AH Inlet

Figure 6.2-2 shows baseline CMM measurement results of gaseous mercury made at all three locations. The total mercury at the stack varied from 3 to 7 μ g/m³ with an average of 5.9 μ g/m³. The mercury level at the ESP outlet varied with time between those of the air heater inlet and stack and with an average of 8.5 μ g/m³. Comparison of the mercury levels in Figure 6.2-2 indicates that a significant amount of total mercury from the coal is captured across various air pollution control devices installed at Fayette Unit #3. About 27% of the gaseous mercury is currently being removed between air heater inlet and ESP outlet, and another 23% of the air heater inlet gaseous mercury in removed across the wet scrubber modules.

Figure 6.2-3 shows mercury speciation data from CMM readings at the ESP outlet and at the stack. Elemental and oxidized mercury at the ESP outlet averaged 1.7 and 6.8 μ g/m³, respectively. Nearly 80% of the total gaseous mercury was in oxidized form at the ESP outlet and before the wet scrubber modules. At the stack, elemental and oxidized mercury averaged 4.2 and 1.7 μ g/m³, respectively. The increase in elemental mercury indicates significant reemission of elemental mercury by the FGD modules.



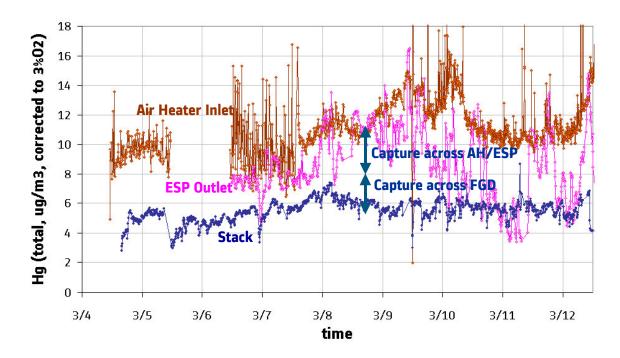


Figure 6.2-2 Baseline Total Mercury Readings from CMMs at Three Locations

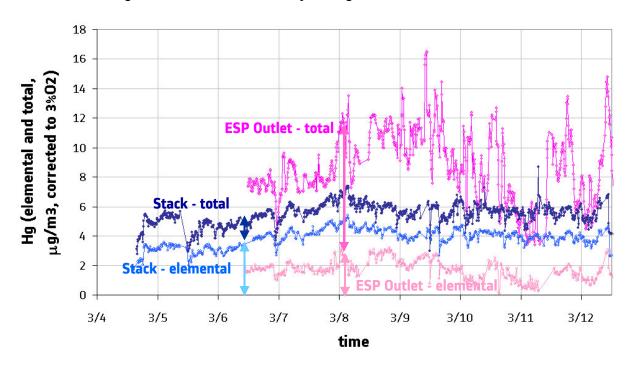


Figure 6.2-3 Mercury Speciation Data at the ESP outlet and Stack

Previous studies and field reports suggest that wet scrubbers capture most of the oxidized mercury, often called "co-benefit." The CMM data, however, does not indicate much co-benefit taking place at Fayette Unit #3. The <u>oxidized</u> mercury at the ESP outlet is 6.8 μ g/m³. If the scrubber modules captured all of the oxidized mercury at the ESP outlet, the expected stack mercury would be only the



elemental portion of the total mercury at the ESP outlet, i.e., 1.7 μ g/m³. The measured <u>total</u> mercury at the stack was 5.9 μ g/m³, and the elemental mercury increased from 1.7 to 4.2 μ g/m³ across the wet scrubber. The wet scrubber modules captured only 38% of the oxidized mercury entering the modules. Figure 6.2-3 suggests that over 60% of the incoming oxidized mercury, in fact, has been reemitted as elemental mercury by the wet FGD modules.

6.3 Parametric Testing Results

Figure 8 shows a typical response of the mercury levels from Fayette Unit #3 with Mer-CureTM system sorbent injection during parametric testing. In this run, Mer-CleanTM 8 was injected using the mobile Mer-CureTM system at a concentration of 1.3 lb/MMacf, which decreased the stack total mercury level from 5.9 μ g/m³ to 0.36 μ g/m³. Notice that by injecting the sorbent, the mercury level at the ESP outlet also decreased to approximately the same level as that for the stack. This clearly shows that all of the mercury capture as a result of sorbent injection took place between the injection point and the ESP outlet and that there was no additional capture of mercury across wet scrubber modules. The figure also shows the mercury reduction in two stages. Within an hour, the mercury level decreased from 8.5 g/m³ to 3.5 μ g/m³ (in flight capture) followed by a slow decrease to 0.5 μ g/m³ over a 12 hour period (wall effects).

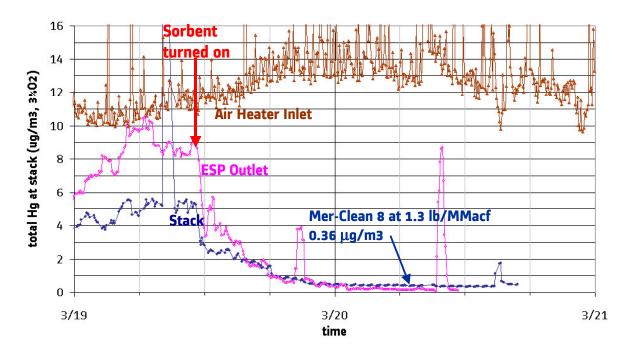


Figure 6.3-1 CMM Response to Mer-Cure™ Sorbent Injection at Fayette Unit #3

Figure 6.3-2 shows the performance of Mer-Cure[™] system at Fayette Unit #3 during individual parametric test runs. The system was operated with four sorbents: Alstom's Mer-Clean[™] 8, and Envergex's eSorb[™] 11, 13 and 18. In the figure, mercury removal efficiencies are reported for various sorbents as a function of sorbent injection rate in pounds per million actual cubic feet (lb/MMacf, where the "actual" condition refers to the FGD inlet conditions of 320°F, 7% O₂), as well



as the percentage points of carbon increase in the ash due to sorbent injection. The removal efficiency has been calculated based on the uncontrolled stack concentrations of gaseous mercury measured during baseline testing. At 0.8 lb/MMacf sorbent injection rate, the system achieved 90% mercury reduction with a 0.25 percentage point increase in the fly ash carbon level.

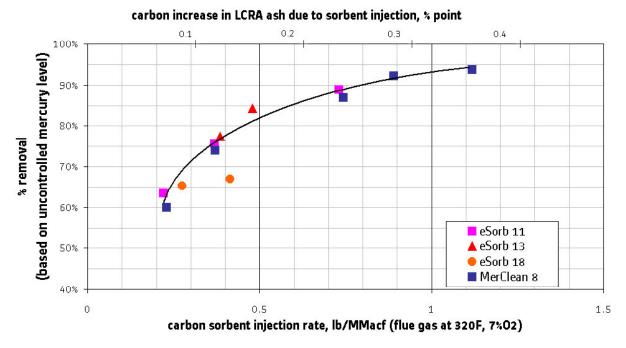


Figure 6.3-2 Parametric Test Results of Mer-Cure™ System at Fayette Unit #3 (based on uncontrolled stack rate)

6.4 Ash Evaluation Testing Results and Balance of Plant Impacts

One of the objectives of the test program at Fayette Unit #3 was to investigate ash salability with sorbent injection for mercury control. Fayette Unit #3 currently sells ash for concrete applications. From the viewpoint of concrete applications, one of the most critical ash properties affected by sorbent injection is the adsorption characteristic of air entraining agent (AEA) by the ash. ⁷

Previous studies by other researchers showed that even at a low rate, injection of activated carbon-based sorbent renders ash not salable. ^{8, 9} In order to evaluate ash salability, ash-sorbent mixtures of various proportions have been prepared and foam index-tested. The fly ash used for mixtures has been collected from the silo B (a silo dedicated to Fayette Unit #3) during baseline measurement. The sorbent used to mix with the fly ash has been collected from one of the ports of the distributors. Mixtures were prepared to simulate silo ash that would be produced as a result of sorbent injection. MRT (Mineral Resource Technologies, Inc.), LCRA's ash marketer, conducted foam index tests according to their test procedure. MRT's test procedure determines the AEA adsorption tendency by adding drops of AEA solutions to ash samples in water until foam starts to cover the entire surface of the sample jar. The number of drops is then converted to a predefined foam index value.

Figure 6.4-1 shows such results from the foam index tests using the simulated ash-sorbent mixtures of various sorbents tested in the program. It also shows the threshold value of 312 above which ash



may not be sold. The foam index value of the LCRA ash collected during baseline testing (no sorbent injection) is 156. As sorbent loading increases in the ash, the foam index increases in a linear fashion with the relative amount of carbon in the ash.

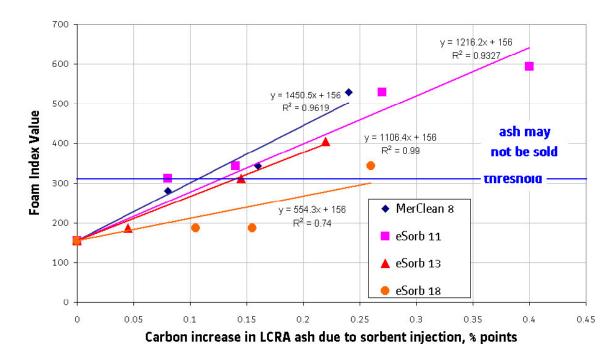


Figure 6.4-1 Foam Index Chart for LCRA Ash-sorbent Samples

Foam index test results show that Mer-Clean™ 8, eSorb™ 11 and eSorb™ 13 sorbents performed similar to one another, with eSorb™ 18 performing better than the other three. Data indicate that the maximum usage of the three similarly performing sorbents consistent with continued ash sale is a sorbent injection rate that results in a fly ash carbon increase of 0.10 to 0.15 percentage points.

Among the four sorbents foam index-tested, eSorb™ 18 gave the lowest foam index values for a given amount of carbon increase. However, the eSorb™ 18 sorbent did not perform as well as either the other two eSorb™ sorbents or the Mer-Clean™ sorbent during the limited number of mercury reduction parametric test runs with eSorb™ 18 (Figure 6.3-2). Even though the eSorb™ 18 was more "concrete-friendly", more of it would need to be injected to obtain the same mercury reduction results as the other three sorbents tested.

Figure 6.3-2 and Figure 6.4-1 can be used to evaluate the overall performance of sorbents for mercury reduction and ash quality combined. Figure 6.4-1 shows that Mer-Clean™ 8 may be employed and injected using the Mer-Cure™ system until the incremental carbon in the ash increases by 0.1 percentage points. This corresponds to injection rate of approximately 0.4 lb/MMacf. At this injection rate, the current mercury emission is expected to decrease by 75%. eSorb™ 13 is more ash friendly and can be injected until the carbon in the ash increases by 0.15 percentage points. This corresponds to an 85% reduction in baseline mercury emissions.

Foam index tests were performed using simulated ash-sorbent mixtures to give a general idea of how a sorbent would impact the quality of the ash. In order to verify the findings, a weeklong



continuous sorbent injection testing was carried out for each of the two sorbents. A third week of testing with a third sorbent was originally scheduled but had to be cancelled due to a forced outage towards the end of the test program. In this test period, sorbents were injected to achieve 90% reduction of the baseline mercury level. Ash samples were taken every day for ash evaluation.

Figure 6.4-2 shows the stack mercury level during Mer-Clean™ 8 testing. As shown, the mercury level was maintained at 0.59 mg/m3 (for a 90% reduction from baseline) by injecting Mer-Clean™ 8 at about 0.9 lb/MMacf, corresponding to an ash carbon increase of 0.28 percentage points. Bulk ash samples were taken every morning during the test week and analyzed for foaming tendency. The foam index values for those four days were 375, 562, 531, and 562, showing that after three days of continuous injection, representative bulk ash samples could be taken. The final value of 562 was very close to the value from the simulated ash mixture tests (Figure 6.4-1) but was higher than the threshold of 312, potentially rendering the ash not salable for concrete application.

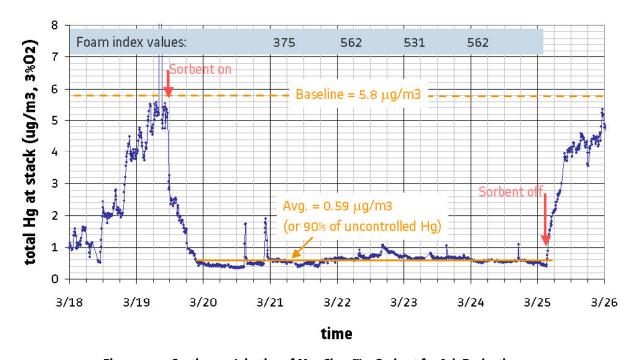


Figure 6.4-2 Continuous Injection of Mer-Clean™ 8 Sorbent for Ash Evaluation

The subsequent week of ash evaluation testing was conducted with the second sorbent, eSorb™ 11 from Envergex, LLC. It was also injected at about 0.9 lb/MMacf, corresponding to an ash carbon increase of 0.28 percentage points, and again targeting 90% reduction of the baseline mercury at the stack. Figure 6.4-3 shows the stack mercury level during eSorb™ 11 testing. The foam index value for the bulk ash sample taken towards the end of the test week was 470, again very close to the value from the simulated ash mixture tests (Figure 6.4-1). This value was also higher than the threshold value of 312 and again potentially renders the ash not salable for concrete application.

From the ash evaluation tests, it is clear that neither of the two sorbents tested can simultaneously achieve both 90% reduction of baseline mercury and ash salability requirements. As mentioned previously, the maximum baseline mercury reduction achievable was 75% for Mer-Clean™ 8 and 85%



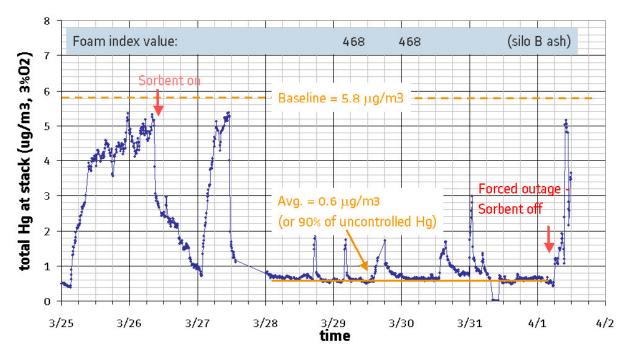


Figure 6.4-3 Continuous Injection of eSorb™ 11 Sorbent for Ash Evaluation

for eSorb™ 13. In terms of percentage removal based on the coal input mercury, this performance translates to an 88% and 92% reduction, respectively, as shown in Figure 6.4-4.

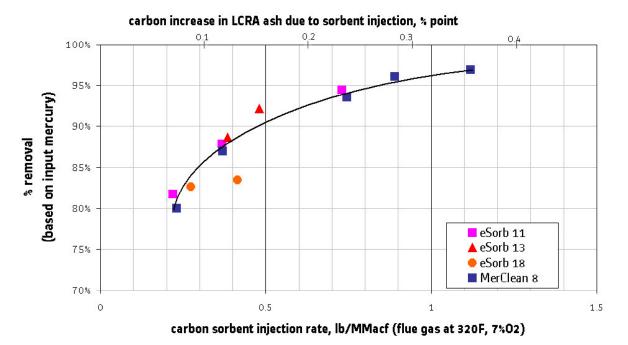


Figure 6.4-4 Parametric Test Results of Mer-Cure™ System at Fayette Unit #3 (based on coal mercury content)



The performance data presented demonstrates that Mer-Cure™ technology offers a solution to effectively control mercury emissions from coal-fired boilers. Sorbent injection may be of operational concern to boiler operators. In this section, the impact of sorbent injection on ash leaching properties and on opacity is discussed.

The ash samples from ash evaluation period have been tested for leaching. The TCLP extraction test results are listed in Table 6.4-1. As listed, no detectable mercury has been leached out. Previous studies have also shown that mercury captured on activated carbon has very low leachability. ^{10, 11}

Sample	Date	Time	μg			TCI	LP Extra	ction (m	g/liter)		
location			Hg/l	As	Ba	Cd	Cr	Pb	Hg	Se	Ag
Reclaim	3/12	8:57	<0.2								
pond											
LS slurry	3/12	8:28	<5								
tank											
Silo ash	3/13			<1	3.5	<0.1	0.3	<0.5	<0.001	<0.5	<0.1
Silo 3B-01	3/23	17:15		<1	3.3	<0.1	0.29	<0.5	<0.001	<0.5	<0.1
Silo 3B-02	3/23	17:15		<1	2.9	<0.1	0.28	<0.5	<0.001	<0.5	<0.1
Silo 3B-01	3/30	16:15		<1	4.3	<0.1	0.32	<0.5	<0.001	<0.5	<0.1
Silo 3B-02	3/30	16:15		<1	3.6	<0.1	0.28	<0.5	<0.001	<0.5	<0.1
Silo B	3/28	15:05		<1	3.7	<0.1	0.31	<0.5	<0.001	<0.5	<0.1
Silo B	3/29	11:35		<1	3.6	<0.1	0.32	<0.5	<0.001	<0.5	<0.1
Silo B	3/30	9:00		<1	3.6	<0.1	0.32	<0.5	<0.001	<0.5	<0.1

Table 6.4-1 LCRA Fayette Unit #3 Ash Leaching Test Results

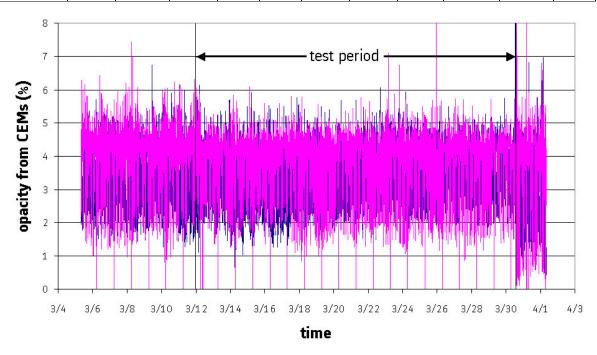


Figure 6.4-5 Measured Fayette Unit #3 Opacity During Test Period



Stack opacity was monitored during the entire test program. Plant opacity data obtained from the plant data historian are plotted in Figure 6.4-5. As shown, no change was observed for opacity during the test period. Observations throughout the Mercury Round 3 testing program suggest that injection of Mer-Clean™ sorbent did not increase the stack opacity during testing.

6.5 Economics

Capital and operating costs for the installation and operation of a Mer-CureTM system for LCRA Fayette Unit #3 are presented in Table 3, along with the assumptions included in the calculations. Capital costs include those for sorbent storage, sorbent processing and injection equipment, and were determined from vendor quotes. Equipment costs including all piping, materials and controls are estimated to be \$1.32 million. Installation, including site integration, is estimated around \$518,000. Adding in start-up support and a contingency of 10 percent of capital costs, the overall capital cost of the Mer-CureTM system is estimated to be about \$2.062 million or about \$4.30/kW_e.

Table 6.5-1 Capital, Operating and Maintenance Cost Estimate for Alstom's Mer-Cure™ Technology Implementation at Fayette Unit #3 for 90 Percent Mercury Reduction

Capital Costs Summary	
Equipment	\$ 1,320,000
Installation including site integration (materials and labor)	\$ 518,000
Start-up support	\$ 18,000
Contingency	\$ 206,000
Total Capital Required	\$ 2,062,000
\$/kW (480 MW _e gross)	4.30
Operating and Maintenance Costs Summary	1
Sorbent (\$1.25/lb, 90% reduction, 92 lb/h, 7,500 hrs)	\$ 862,500
Power, Labor and Maintenance	\$ 115,000
Loss of Ash Sale for Concrete (at \$20/ton ash)	\$ 2,090,000
Annual O&M	\$3,067,500
Levelized – 20 year O&M Cost (\$/kW)	8.41
\$/MWh	0.87

Notes:

- a) 90 percent mercury removal target
- b) Unit is assumed to be base loaded
- c) Ash is no longer utilized for concrete
- d) Installation is with union labor
- e) Contingency is calculated at 10% of capital cost
- f) Levelized costs are calculated assuming 3% increase per year
- g) 480 MW_e gross capacity with 7500 hours operation per year



Normally, the predominant component of the operating cost is the sorbent cost. Costs were estimated based on a long-term Mer-Clean™ 8 sorbent injection rate conservatively assumed to be 0.84 lb/MMacf or 92 lb/hr. Unit operation is assumed at 85 percent capacity factor. With these assumptions, the annual sorbent costs are \$862,500, assuming a delivered sorbent cost of \$1.25/lb. There is a significant uncertainty in the cost for the sorbent because of anti-dumping tariffs being imposed on imported activated carbon and because of anticipated increased demand for activated carbon. The tariff amounts are expected to be around \$0.25 to \$0.30/lb. This almost doubles the cost of imported activated carbon which is currently around \$0.30-0.40/lb. The tariffs and increased demand for active carbon may both lead to an increase in the custom-enhanced sorbent costs, even for carbon produced domestically.

At 90% mercury reduction, the demonstration program results show that Fayette Unit #3 would no longer be able to sell their fly ash for concrete applications. In this case, the lost revenue from the ash sale becomes the predominant component of operation cost. With the assumptions shown in Table 6.5-1 and the average coal quality from Table 6.1-1, the annual ash generated is 1.045 million tons. At \$20/ton, the operating cost associated with the lost ash sale is \$2.09 million.

The other component of the operating cost is parasitic power, labor and maintenance and this is estimated to be \$115,000 annually. The overall operating cost for Fayette Unit #3 for 90 percent mercury control is \$3,067,500 or \$0.87/MWH.

Table 6.5-2 can be used to calculate the operating cost if lower levels of mercury removal are required. For example, if 70 percent reduction is required the sorbent injection rate is 0.33 lbs/MMacf (36 lb/hr). If 80 percent reduction is required, the sorbent injection rate is 0.50 lbs/MMacf (55 lb/hr).

Table 6.5-2 Sorbent Consumption Rates as a Function of Mercury Removal Level for Mer-Cure™ Technology Implementation at Fayette Unit #3

% Mercury Removal	Sorbent (lb/MMacf)	Sorbent (lb/hr)
60	0.22	25
70	0.33	36
75	0.40	44
80	0.50	55
90	0.84	92
95	1.17	129

Sensitivity analysis for Fayette Unit #3 with respect to the mercury removal level required and unit sorbent costs are shown in Figure 6.5-1. The line on this figure labeled "Sorbent price = \$1.25" shows the big impact that ash sales have on the overall economics of mercury reduction. At 75% reduction and below, the ash sale is maintained and the mercury removal costs are \$0.16/MWH or less at a sorbent cost of \$1.25/lb. (All assumptions are the same as those listed in Table 6.5-1).



Once sufficient sorbent is injected to achieve above 75% mercury reduction, sale of the ash for concrete is not longer possible and operating costs rise significantly, to \$0.57/MWH at 80% reduction and increasing from there.

The lines labeled "Injection for 75% removal" and "Injection for 90% removal" also show the dramatic effect of ash sale economics on operating costs. These calculations indicate that at 75% mercury control, the operating cost ranges from \$0.10/MWH to \$0.26/MWH. However, at 90% mercury control, the operating cost increases to a range of \$0.76/MWH to \$1.05/MWH.

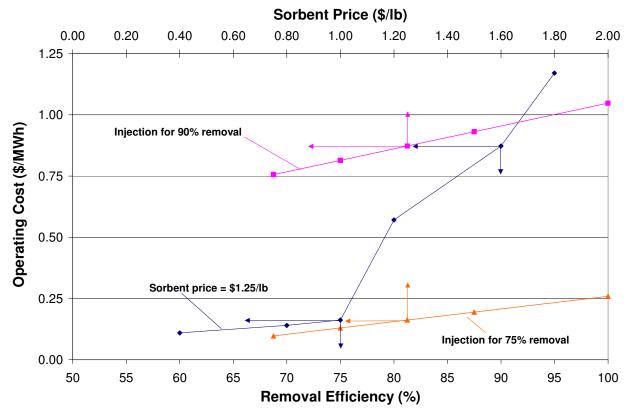


Figure 6.5-1 Sensitivity analysis: Effect of Sorbent Unit Cost and Mercury Removal Required on Operating Cost for Mercury Control for LCRA Fayette Unit #3 using Mer-Cure™ technology

6.6 Conclusions

Field demonstration of the Mer-Cure™ system has been completed at Fayette Unit #3. The goals of the program, as originally defined by DOE/NETL in the solicitation, are to demonstrate that Mer-Cure™ technology can achieve:

- Greater than 90% reduction of gaseous mercury in the flue gas;
- At a mercury control cost significantly lower than 50% of \$60,000/lb mercury removed.

Alstom-PPL also collected performance data to evaluate the long-term impact of mercury control on the operation of power plants and overall plant economics. LCRA was also interested in determining



the maximum mercury reduction possible while maintaining the capability to sell their fly ash for concrete applications.

The first performance goal was achieving 90% mercury reductions. Two of the four sorbents tested achieved 90% mercury reductions, and the other two sorbents were not tested at injection rates above 0.5 lb/MMacf because of time constraints. In the ash evaluation tests, foam index test measurements indicated that the ash generated with sorbent injection to achieve 90% mercury reductions would likely be unsuitable for concrete applications. Figure 6.6-1 summarizes the parametric test mercury reduction results and the maximum mercury reduction possible while maintaining ash quality for sale.

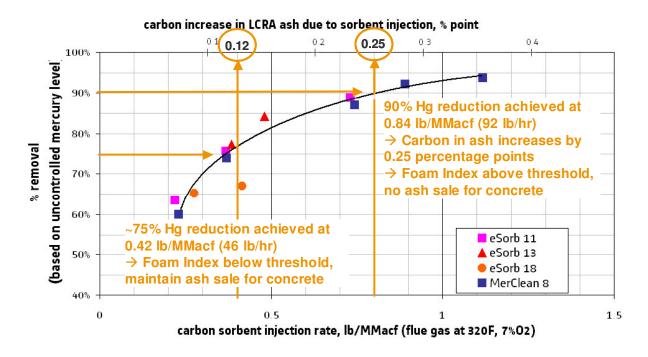


Figure 6.6-1 Mercury Reduction Performance Summary for Fayette Unit #3

Figure 6.6-1 shows that the performance goal of 90% mercury reduction could be achieved if sale of the fly ash were sacrificed. A 75% mercury reduction could be achieved while maintaining sale of the fly ash for concrete applications.

The second performance goal was achieving 90% mercury reduction at a mercury removal cost significantly lower that 50% of the assumed baseline cost of \$60,000/lb mercury removed. For the purposes of this discussion, "significantly lower than 50%" is defined as at least 60-70% below the assumed baseline cost of \$60,000/lb mercury removed.

Figure 6.6-2 shows mercury removal costs (in \$/lb Hg removed) at sorbent injection rates to achieve both 75% and 90% mercury reduction. The figure shows very low removal costs for 75% reduction (\$2100-5700/lb Hg removed at \$0.75-2.00/lb sorbent) because only 46 lb/hr sorbent is required to achieve 75% reduction, and testing indicated that the fly ash sale for concrete could be maintained. The removal costs at 90% reduction levels are significantly higher (\$13,400-18,700/lb Hg removed)



at \$0.75-2.00/lb sorbent). Sorbent costs are doubled, because 92 lb/hr sorbent flow is required to achieve 90% reduction, but the inability to sell ash for concrete accounts for the majority of the cost increase. Even with the ash effects, the removal costs at 90% reduction are over 70% below the baseline cost of \$60,000/lb Hg removed at sorbent costs below \$2.00/lb. Even at \$2.00/lb sorbent, the removal cost is \$18,700/lb or 68.8% below \$60,000/lb Hg removed.

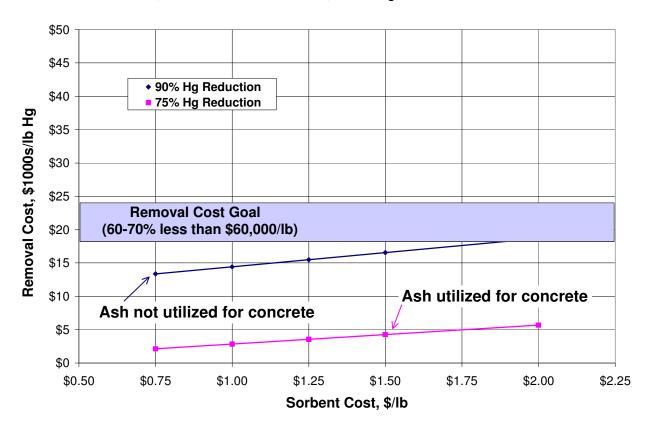


Figure 6.6-2 Mercury Removal Costs for LCRA Fayette Unit #3



7 Commercialization of Mer-Cure™ Process

The timing of commercialization for mercury capture technology is driven largely by mercury reduction regulations. Regulations have consisted of proposed federal regulations plus a variety of proposed and enacted state regulations. On February 8, 2008, the DC Circuit Court of Appeals struck down EPA rules known as the Clean Air Mercury Rule or CAMR, in response to an appeal by 18 states and several environmental groups. ¹² The petitioners questioned EPA's decision to delist coal and oil-fired power plants from the list of sources that produce hazardous air pollutants (including mercury). The court agreed, finding that EPA did not go through the proper process before delisting. As a result, mercury emissions trading is illegal; instead mercury will be subject to yet-to-be-specified command-and-control regulations.

On July 11, 2008, the DC Circuit Court of Appeals also struck down the Clean Air Interstate Rule (CAIR), saying it contains "several fatal flaws." ¹³ That regulation was aimed at reducing emissions of sulfur dioxide and nitrogen oxides from power plants in 28 eastern states and the District of Columbia, and would have required most large utility boilers to install scrubbers for SO_2 control and selective catalytic reduction (SCR) systems for NO_x control. When the two pollution control devices are combined, the SCR often oxidizes the elemental mercury in the gas phase and the wet scrubber often collects a significant portion of the water-soluble oxidized mercury. The CAIR requirements were expected to result in a co-benefit of significant mercury capture because of this characteristic.

The effective result is that state regulations are now driving the near-term commercialization of mercury capture technology. For utilities with units currently under state regulation, there is an understanding that new federal regulations are forthcoming, but they are required to implement solutions to meet the current state regulations.

7.1 Commercial Mer-Cure™ System Design

Alstom's Mer-Cure™ technology employs a sorbent preparation and injection system consisting of four (4) major subsystems as follows:

- Bulk Storage and Feed
- Transport
- Processing
- Distribution and Injection

The bulk storage and feed system receives PAC sorbent delivered by tank trucks. Sorbent is conveyed pneumatically from the truck into a fabricated storage silo. The sorbent is fed into a hopper where it is temporarily stored until fed into the drop tube by the variable speed screw feeder and into an eductor inlet.

Motive air, supplied from air compressors, passes through the eductor nozzle and creates a vacuum in the eductor inlet, resulting in the sorbent being drawn into the mixing zone and exiting the transport system. Sorbent is conveyed via conventional piping to the sorbent processor where high-pressure air from the compressors is used to process the sorbent before it enters the distributors and the injection lances.



Additional motive air from the compressors is added downstream of the sorbent processor in order to ensure that the processed sorbent continues to be entrained and enters the distribution at the correct velocity. Distribution is achieved via splitters that divide the sorbent flow between the flue gas ducts (typically, two) and additional splitters that direct the sorbent flow to the different injection lances.

Overall system operation is controlled via programmable logic controllers for the silo and feed system, and Alstom-supplied control logic for the overall control system, which can be programmed into the plant's existing Distributed Control System (DCS).

All components integrated into an assembled Mer-Cure[™] system are commercially available for systems of various sizes and capacities, with one exception. The injection lances and nozzles are designed and fabricated specifically for each site.

7.2 Status of Mer-Cure™ Commercialization Efforts

Since completion of the field testing for this DOE project, Alstom has conducted multiple commercial demonstrations at electric utility boilers for several generating companies. One of the demonstrations has resulted in the first large commercial installation contract for Mer-Cure™ technology. At least one other multiple unit installation of Mer-Cure™ technology is under consideration by a second utility.

Alstom and Reliant Energy, Inc., one of the largest independent power producers in the United States, have announced the contract signing for the purchase of Alstom's advanced mercury removal system, Mer-Cure™. ¹⁴ The deal follows the successful launch of Alstom's Mer-Cure™ product line in 2007, marking Alstom's entry into this new market. The Mer-Cure™ system delivers reductions in flue gas mercury emissions of up to 90% and more, at low sorbent consumption rates, thus trimming operation and maintenance costs.

The system will be installed at Reliant's Shawville, Conemaugh, Titus, Portland and New Castle power stations in Pennsylvania. Delivery is scheduled for early 2009. This is in response to Pennsylvania's mercury capture regulations, Phase I of which take effect on January 1, 2010.



8 Summary of Demonstration Program Results

Following is the summary of the Fayette Unit #3 test campaign results discussed in this final report.

Baseline

- Baseline measurements showed that the native capture of mercury at Fayette Unit #3 is approximately 50%. About 27% capture was observed across air heaters and ESP's and 23% across wet scrubbers.
- During the baseline measurement period, only 36% of the oxidized mercury was captured by wet scrubbers. A significant amount of oxidized mercury was reduced to elemental mercury in the wet scrubber modules.
- At the ESP outlet, gaseous mercury was 20% elemental / 80% oxidized. At the wet scrubber outlet, gaseous mercury was 72% elemental / 28% oxidized. Absolute levels of elemental mercury increased from 1.7 to 4.2 μ g/m³, indicating reemission of elemental mercury by the scrubber.

Mercury Capture Performance

- Mer-Cure[™] performance was determined for four sorbents: Mer-Clean[™] 8, eSorb[™] 11,
 eSorb[™] 13, and eSorb[™] 18. eSorb[™] 11 performed better than Mer-Clean[™] 8 or eSorb[™] 13.
 eSorb[™] 18 performed the worst.
- Mercury capture performance at Fayette Unit #3 showed that at an injection rate of 0.84
 Ib/MMacf, the mercury removal rate was greater than 90% based on the <u>uncontrolled</u>
 mercury level at the stack. About 0.5 lb/MMacf was needed to achieve 90% reduction based
 on the <u>input</u> mercury.
- At an injection rate of 0.42 lb/MMacf, the mercury removal rate was greater than 75% based on the <u>uncontrolled</u> mercury level at the stack and greater than 88% based on the <u>input</u> mercury.

Ash Evaluation

- Foam index tests were conducted by MRT with simulated ash-sorbent mixtures. ESorb™ 18 performed the best followed by eSorb™ 13, eSorb™ 11, and Mer-Clean™ 8.
- Continuous injection testing was conducted with two sorbents for ash evaluation: Mer-Clean™ 8 and eSorb™ 11. Ninety percent capture of baseline mercury was achieved for both sorbents during week long continuous injection testing.
- The silo ash collected during continuous injection and characterized for foaming tendency verified the results from the simulated ash-sorbent testing.
- Analysis indicates that for Mer-Clean™ 8 sorbent, a reduction of 75% of baseline mercury or 88% of input mercury can be achieved while still allowing continued ash sales.
- Analysis indicates that for eSorb™ 13 sorbent, a reduction of 85% of baseline mercury or 92% of input mercury can be achieved while still allowing continued ash sales.



Economics

- Overall capital cost of the Mer-Cure[™] system is estimated to be about \$2.06 million or \$4.30/kWe.
- The overall operating cost for 75 percent mercury removal is \$546,250 or \$0.16/MWh, assuming a delivered sorbent price of \$1.25/lb.
- The overall operating cost for 75 percent mercury removal is \$3,067,500 or \$0.87/MWh, assuming a delivered sorbent price of \$1.25/lb and ash valued at \$20/ton.
- If sorbent price increases to \$2.00/lb, the Mer-Cure™ operating cost increases to \$0.26/MWh and \$1.05/MWh for 75 and 90 percent mercury removal, respectively.

Balance of Plant Impacts:

- Observations throughout the long-term testing suggest that injection of Mer-Clean™ sorbent did not increase the stack opacity during long-term testing.
- Ash leaching test results indicate that no detectable mercury leached out.



9 References

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Appendices

Appendix A Ontario Hydro Test Results

Appendix B CMM Screens

Appendix C Coal and Ash Analyses



Appendix A Ontario Hydro and Sorbent Trap Test Results

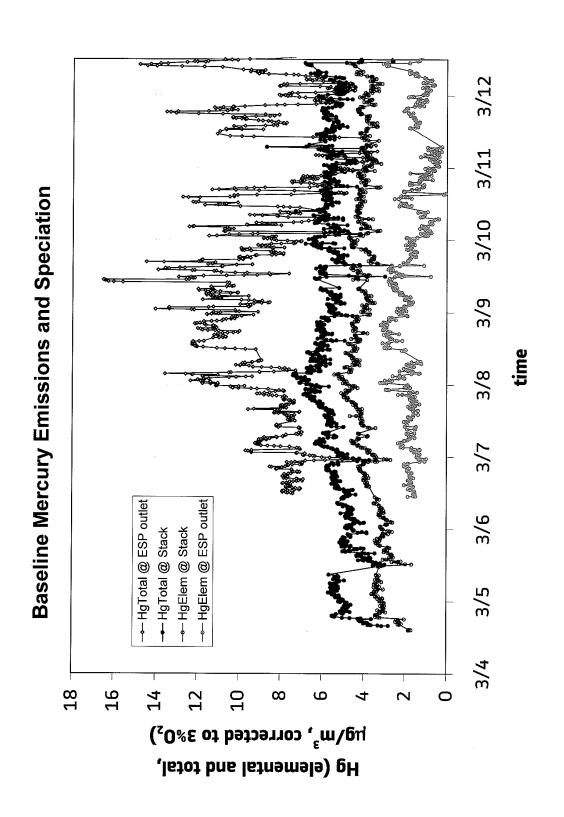
I FAYETTE	ĮĮ)	Ē	ng concentration summary	mary		hoped	adillon	Shadad traffice are halour 110						
				ſ	_									
NOTE	DAT	NOTE: DATA NOT CORRECTED FOR 02 (DRY BASIS)	R O2 (DRY BASIS)			Blan	k cell: I	Blank cell: No analysis	/sis	DATA	CORRI	DATA CORRECTED TO 3%	TO 3%	
				d 6H	Hg	+	Hg	Нg°		Hg	+ H	νв	Total	
	START		SAMPLE	FILTER	PROBE	KCI	H202	KMn04	TOTAL					
DATE	TIME	SAMPLE I.D.	PORT	CONC.	CONC.	CONC.	CONC.	CONC.	CONC.	CONC. CONC.	CONC.	CONC.	CONC.	
				(ug/m3)	(ng/m3)	(ng/m3)	(ng/m3)	(ng/m3)	(ng/m3)	(ug/m3) (ug/m3)		(ng/m3)	(ng/m3)	
3/8/2007	15:15	15:15 FAY-U3-D1-STK-OH-1	Stack-east side	0.01		0.63	0.75	4.40	5.80	0.01	0.54	4.42	4.97	baselino
3/10/2007	ppr 14:0	3/10/2007 ppr 14:0 FAY-U3-D3-AHIN-OH-1	7th port from north	0.20	90.0	99.0	0.79	10.52	12.23	0.17	0.61	69.6	10.48	basetme
														2
3/12/2007	12:30	3/12/2007 12:30 FAY-U3-D4-STK-OH-1	Stack-east port	0.014		96.0	0.35	2.77	4.10	0.01	0.83	2.68	3.51	
3/12/2007	15:42	3/12/2007 15:42 FAY-U3-D4-AHIN-OH-1	AH IN	90.0	0.04	0.87	0.50	12.43	13.89	0.05	0.78	11.08	11.91	
3/12/2007	15:52	15:52 FAY-U3-D4-ESPOUT-OH-1	ESPOUT 2nd from east	0.005		6.13	0.26	flool	6.39	00.0	5.25	######	5.48	
3/13/2007	13:55	3/13/2007 13:55 FAY-U3-D5-AHIN-OH-1		0.02		1.54	0.33	14.95	16.84	0.02	1.32	13.09	14.43	
3/13/2007		14:00 FAY-U3-D5-ESPOUT-OH-1	FOR OUT IN SIDE, ZITO	0.01		0.34	0.11	90.0	0.53	0.01	0.30	0.15	0.45	
3/14/2007	11:56	3/14/2007 11:56 FAY-U3-D6-ESPOUT-OH-1	ESPOOT, N SIDE, ZIIG	0.02		0.57	0.30	0.11	0.99	0.02	0.49	0.35	0.85	58%
3/14/2007	20:00	20:00 FAY-U3-D6-STK-OH-1	Stack-east port	0.02		0.97	0.13	1.01	2.13	0.02	0.83	0.98	1.83	64
														ç

LCRA OH Summary

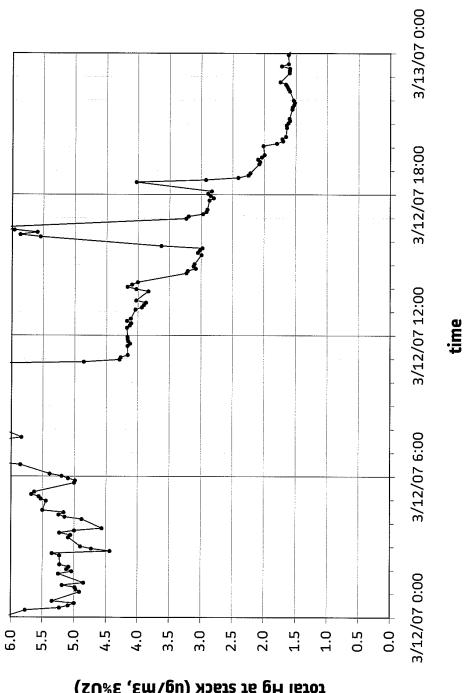
Alstom						
Fayette Carbon Trap Data						
DMC 05/22/07						
						Corrected
						to 3% O2
Sample Id.	Date	Time	Duration 02	02	£	Hg
		Start	Minutes	%	hg/m3	
FAY-U3-D1-Stk-Carbon-1	3/8/2007	8:30	120	6.90	7 4.35	5.55
FAY-U3-D2-ESPout-Carbon-1	3/9/2007	13:20	120	6.80	3 8.47	10.74
FAY-U3-D2-AHin-Carbon-1	3/9/2007	13:50	31	4.00	11.03	11.68
FAY-U3-D2-AHin-Carbon-2	3/9/2007	15:20	09	4.00	10.92	11.56



Appendix B CMM Screens

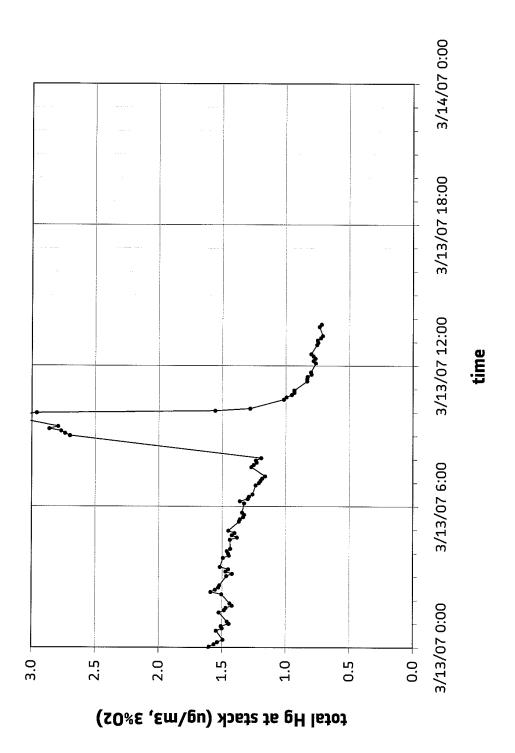




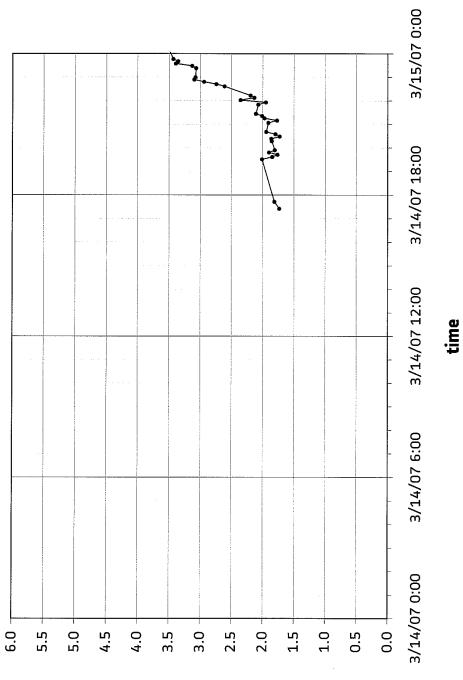


total Hg at stack (ug/m3, 3%02)



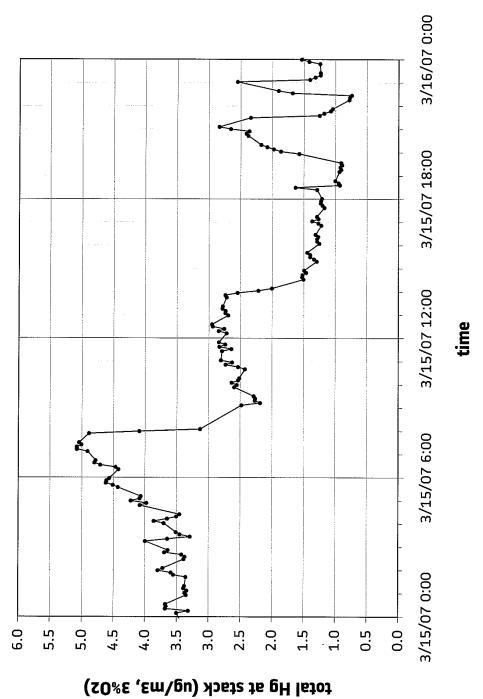




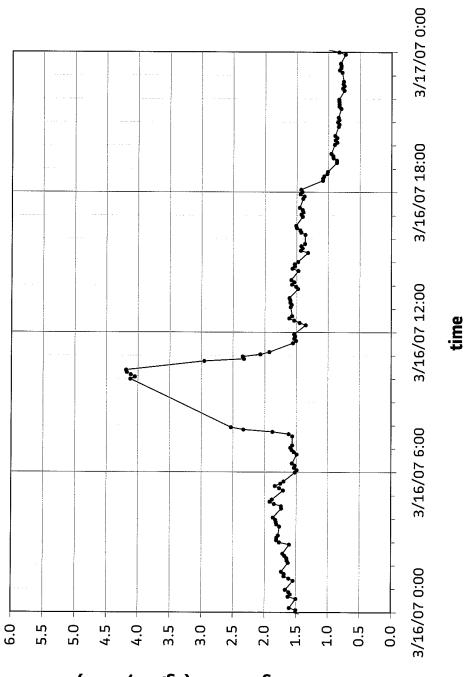


total Hg at stack (ug/m3, 3%02)



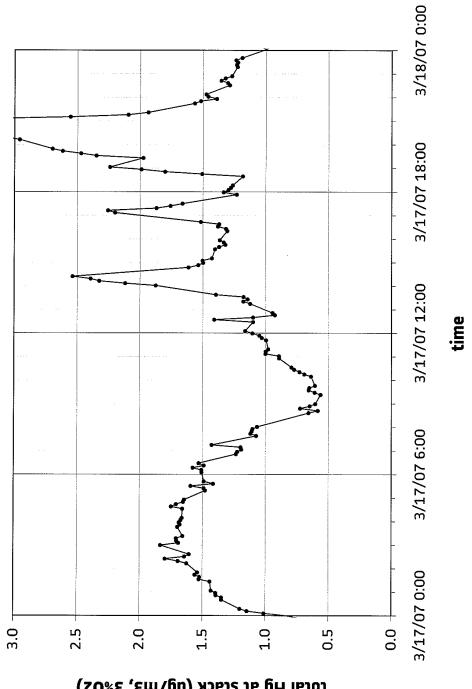






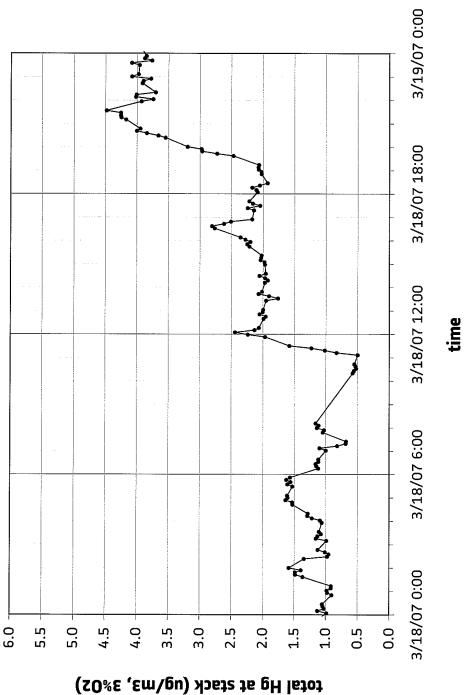
total Hg at stack (ug/m3, 3%02)



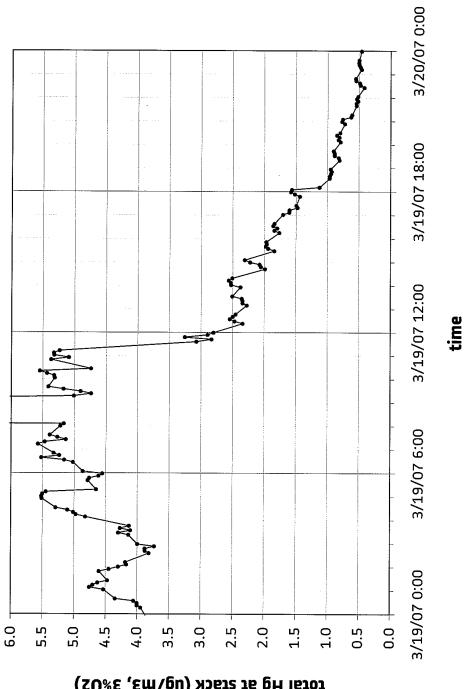


total Hg at stack (ug/m3, 3%02)



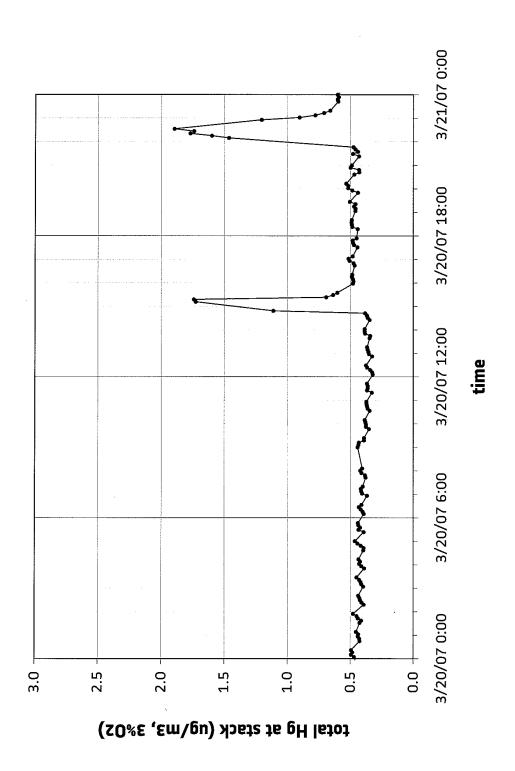






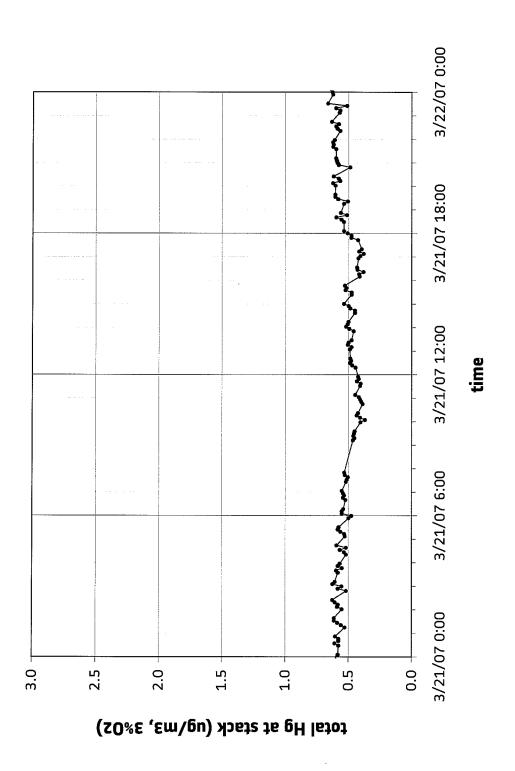
total Hg at stack (ug/m3, 3%02)



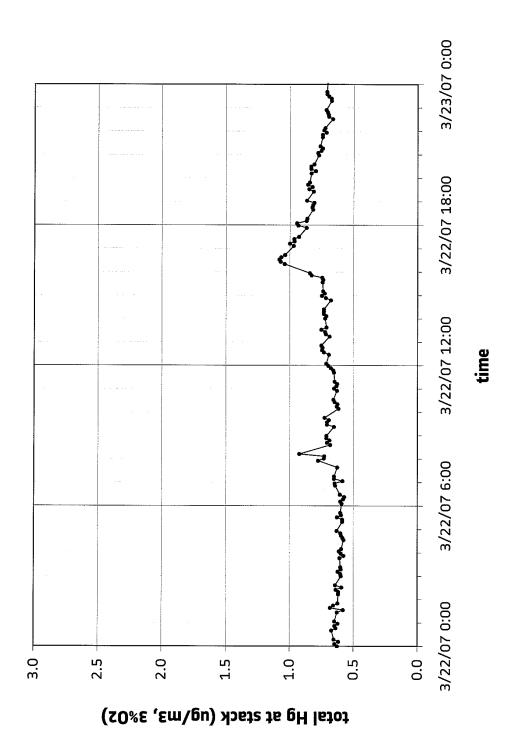


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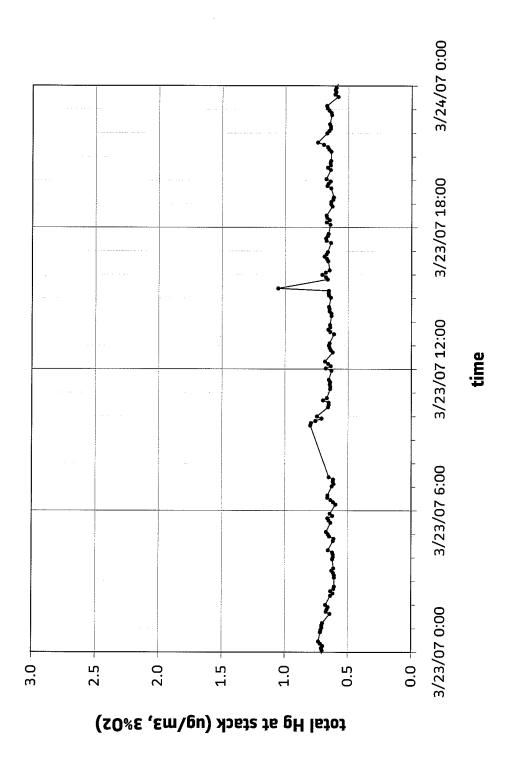




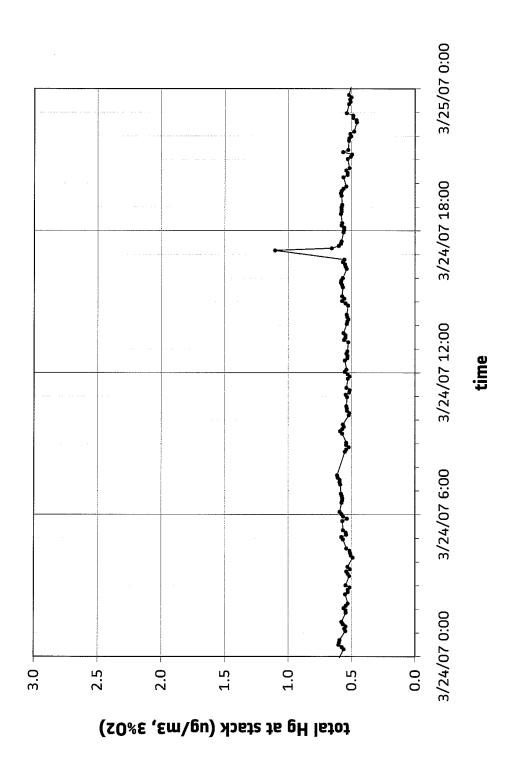




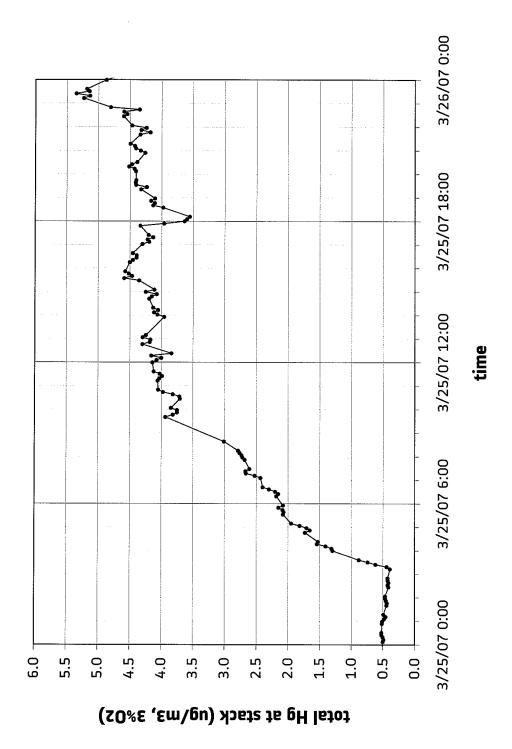




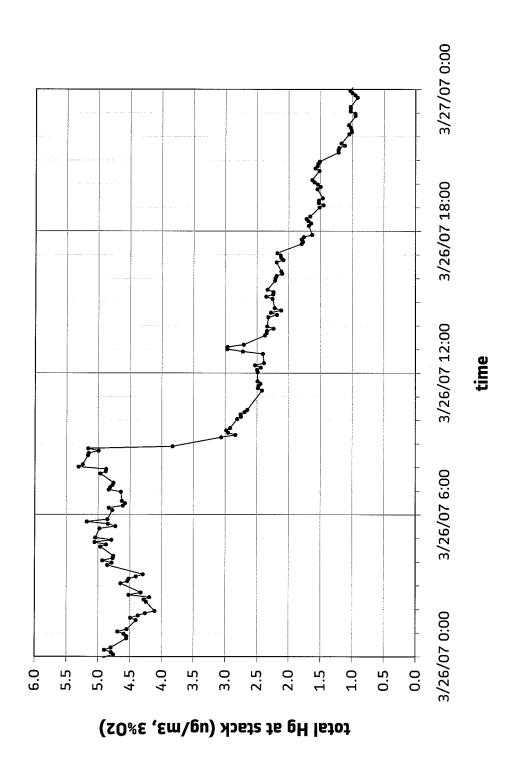




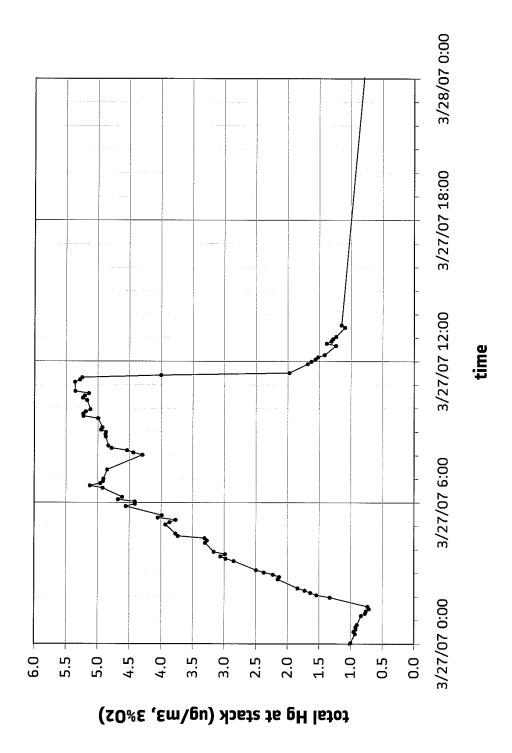






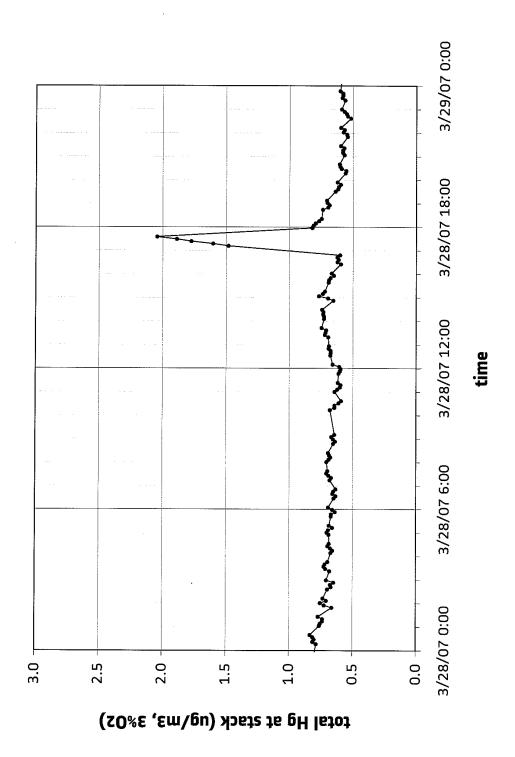




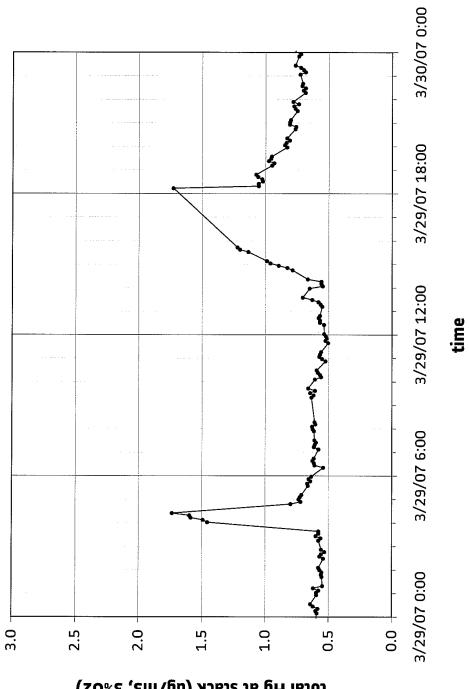


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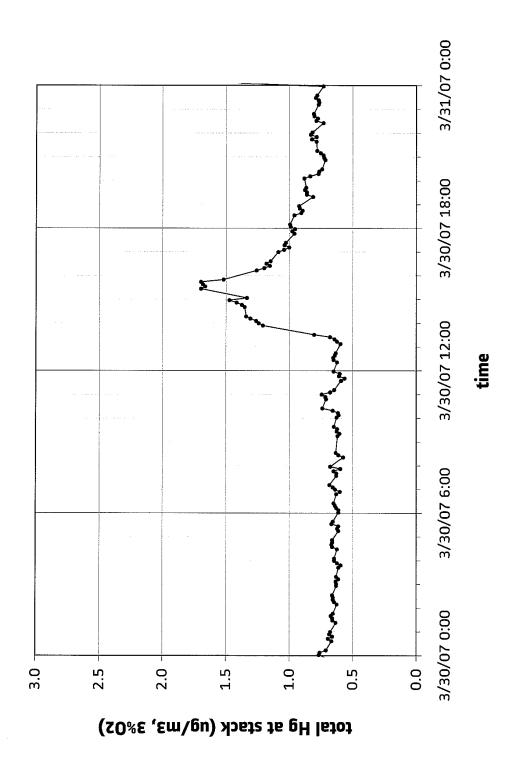






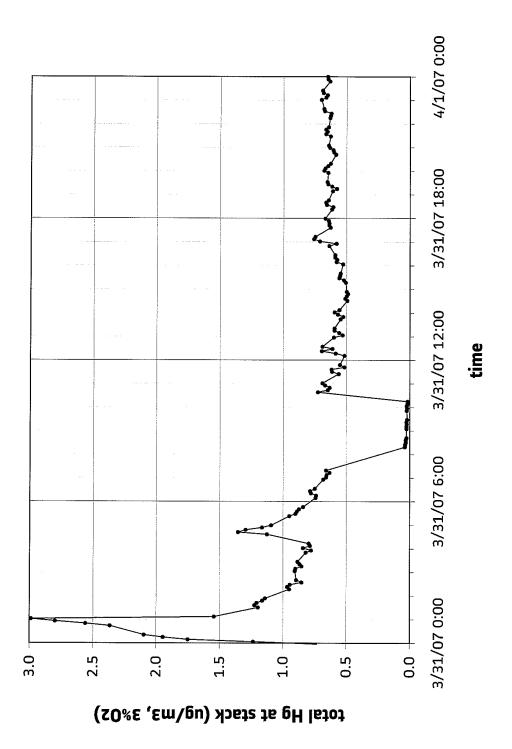
total Hg at stack (ug/m3, 3%02)





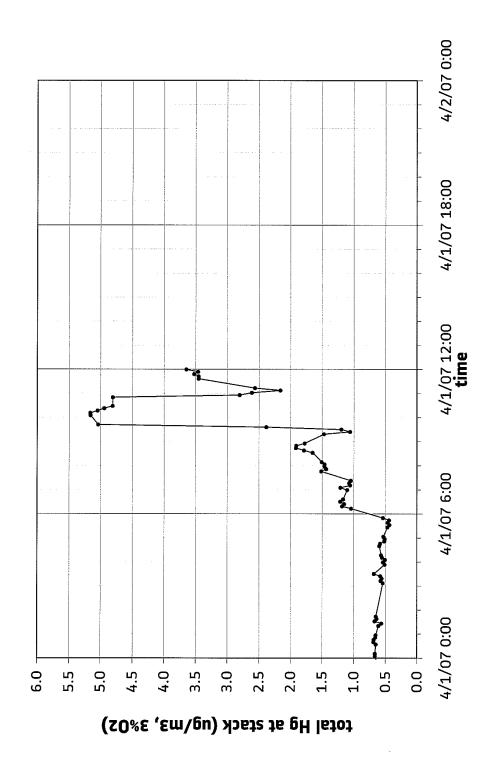
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Appendix C Test Schedule, Coal and Ash Analyses

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Test Schedule for LCRA Fayette Unit 3

3/3	3/10	3/17	3/24	3/31	4/7	4/14
Sat			tative bulk ash) day 6 MerClean	day 6 eSorb 11		
Fri 3/2	3/9 Kick-off Meeting	3/16	3/23 ampling of represen day 5 MerClean	3/30 day 5 eSorb 11	4/6	Demobilize 4/13
Thu 3/1	3/8	3/15 njection rates)	3/22 fixed lb/MMacf for s. day 4 MerClean	3/29 day 4 eSorb 11	4/5	4/12 Demobilize rts
Wed 2/28	3/7 ction of sorbents)	3/14 sorbents at several i	3/21 ion of sorbents at a day 3 MerClean	3/28 day 3 eSorb 11	4/4	Demobilize Described outage starts
Tue 2/27	3/6 3/6 3/6 8aseline mercury measurement (no injection of sorbents) Mer-Cure installation week	3/12 3/14 critic testing (injection of various sorbents at several injection rates)	3/20 s. (continuous inject day 2 MerClean	3/27 day 2 eSorb 11	4/3	4/10
Mon 2/26	3/5 Baseline mercury m Mer-Cure installatio	3/12 Parametric testing (3/19 3/20 3/21 3/22 3/23 Ash evaluation starts (continuous injection of sorbents at a fixed lb/MMacf for sampling of representative bulk ash) day 1 day 2 day 3 day 4 day 5 day 6 MerClean MerClea		4/2	4/9 Demobilize mobile ur Demobilize
Sun 2/25	3/4	3/11	3/18	3/25 day 7 MerClean recovery, PM	forced outage	4/8



Tentative Test Schedule for LCRA Fayette Unit 3

as of Mar 1, 2007

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tasks by	Sun	Mon	Tüe	Wed	Thu	Fri	Sat
	2/25		2/27	2/28		3/2	3/3
	3/4	3/5 Baseline mercury n Mer-Cure installation	ieasurement (no inje	3/7 ection of sorbents)	3/8	3/9 sorbent arrival Kick-off Meeting	
ALS		ALS on site Mer-Cure installn	Mer-Cure installn	Mer-Cure instalin	Mer-Cure installn	Mer-Cure installn	Mer-Cure installn
EERC				Method 29	OH/c trap (stack)	c trap (AHin/FGDin)	SO3 (AH in/out)
LCRA/ MRT		Mer-Cure installn coal, ESP ash	Mer-Cure instalin coal	Mer-Cure installn coal, ESP ash	Mer-Cure installn coal, ESPash, FGD samples bucket of silo "B" ash	Mer-Cure installn coal, ESP ash unload sorbent bags	coal
	3/11	3/12 Parametric testing (3/13 Injection of various	3/14	3/15		
EERC		CEGRIT samples OH	Method 29 OH			CEGRIT samples	CEGRIT samples
LCRA/ MRT		coal	coal foam index	coal foam index	coal foam index	coal foam index	coal foam index
	3/18	3/19	3/20	3/21	3/22	3/23	3/24
		lday 1	day 2 sorbent 1	day 3 sorbent 1	fixed lb/MMacf for s day 4 sorbent 1	day 5 sorbent 1	ltative bulk ash) day 6 sorbent 1
EERC	CEGRIT	continue operation of work on ALS PSA un					
LCRA/ MRT	foam index	coal, ESP ash	coal	coal, ESP ash	coal draw down silo B	coal, ESP ash silo B ash from last l foam index	
	3/25	3/26	3/27	3/28	3/29	3/30	3/31
ALS	sorbent 1 stopped at 1315 recovery	day 1 sorbent 2 eSorb 11	day 2 sorbent 2	day 3 sorbent 2	day 4 sorbent 2	day 5 sorbent 2	day 6 sorbent 2
EERC LCRA/ MRT	coal	coal, ESP ash	coal	coal, ESP ash	coal draw down silo B	coal, ESP ash silo B ash from last li foam index	coal pad
	4/1	4/2	4/3	4/4	4/5		4/7
ALS			FORCED OL day 3 sorbent 3	JTAGE: NO FURTH day 4 sorbent 3	day 5 sorbent 3	day 6 sorbent 3	day 7 sorbent 3
EERC LCRA/ MRT	coal	coal, ESP ash	coal	coal, ESP ash, FGD samples	coal draw down silo B	coal, ESP ash	 Coal coad
	4/8	4/9 Demobilize mobile ur	4/10 Demobilize	4/1-1 Demobilize	4/12 Demobilize	foam index 4/13 Demobilize	4/14
ALS EERC				scheduled outage sta (outage ends May 14	laris D		
LCRA/ MRT		***************************************					
(VII VI							

Note: *During test period, LCRA will provide control room data on a daily basis, as requested previously.

sample type sample size silo B ash

one 5-gal bucket (from the last load of the day)

ESP ash one pint per hopper (six hoppers from North side; four hoppers from South)

coal one gal (one pulverizer) (*) FGD samples: limestone slurry

one quart jar (after mill) one quart jar (after vacuum filter) one quart jar (upstream reaction vessel) gypsum makeup water

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7-2601-C Raw Coal -Slic 2 North 3/16/07 14:00	20.74 34.65 39.10 5.51	54.22 3.66 14.69 0.34 0.34 5.51 100.00 1.00	0.08	5.89 0.72 0.36 1.13 8805 12689 747 9063 57.4 6.78E-06	43.72 49.33 6.95	11807	4.62 68.41 0.42 1.05 0.008 18.53 6.95	0.08 0.000 0.000
7-2599-C 7-2600-C 7-2601-C Raw Coal -Silo Raw Coal -Silo Raw Coal -Silo Ray Coal -Silo Ray Coal -Silo 2 North 2 North 3/14/07 8:20 3/15/07 16:25 3/16/07 14:00	30.59 30.71 34.24 4.46	47.78 3.25 12.91 0.73 0.27 4.46 30.59 100.00 1.00	0.08	5.41 0.65 0.33 1.11 7629 12894 748 8003 72.6 6.73	44.25 49.33 6.42	11879	4.69 68.84 0.39 1.06 0.009 18.59 6.42	0.08 0.001 0.000
7-2599-C Raw Coal -Silo 2 North 3/14/07 8:20	29.97 31.14 32.84 6.05	47.43 3.32 11.93 0.75 0.54 6.05 29.97 100.00 1.00	0.09	7.34 1.32 0.66 1.05 7631 752 8083 7.64E-06 7.64E-06	44.46 46.89 8.64	11777	4.75 67.73 0.78 1.07 0.010 17.02 8.64 100.00	0.09
7-2598-C Raw Coal -Silo 2 North 3/13/07 8:30	33.82 28.94 32.04 5.20	45.00 3.11 11.80 0.77 0.31 5.20 33.82 100.00 1.00	0.07 10.32	6.68 0.79 0.39 1.11 7149 12767 750 7596 7596 7994 6.95E-06 6.95	43.73 48.41 7.86	11764	4.70 67.99 0.46 1.16 0.010 17.82 7.86	0.00 0.001 0.000
7-2597-C Raw Coal -Silo 2 North 3/12/07 8:50	28.18 31.26 35.34 5.22	49.03 3.37 13.11 0.75 0.34 5.22 28.18 100.00 1.00 8445	0.09	6.18 0.80 0.40 1.13 7843 752 8244 693 7.65E-06 7.65	43.53 49.20 7.27	11759	4.69 68.26 0.47 1.04 0.008 18.25 7.27	0.09
7-2596-C Raw Coal -Silo 2 North 3/11/07 14:00	27.65 31.87 36.27 4.22	49.98 3.43 13.73 0.72 0.27 4.22 27.65 100.00 1.00 8630	0.06 8.57	4.89 0.63 0.31 1.14 8027 12666 747 8372 67.8 5.03E-06	44.05 50.12 5.83	11928	4.74 69.08 0.37 1.00 0.008 18.97 5.83	0.06 0.001 0.000
7-2591-C 7-2592-C 7-2593-C 7-2594-C 7-2595-C 7-2594-C 7-2595-C 7-2594-C 7-2	29.55 31.13 34.93 4.39	48.84 3.34 12.85 0.73 0.29 4.39 29.55 100.00 1.00 8385	0.09	5.24 0.68 0.34 1.12 7771 12693 754 8217 7.56E-06 7.56	44.19 49.57 6.23	11902	69.33 0.41 1.04 0.009 18.24 6.23	0.09 0.001 0.000
7-2594-C Raw Coal -Sllo 2 North 3/9/07 14:50	29.62 30.78 34.76 4.83	48.50 3.44 12.57 0.75 0.28 4.83 29.62 100.00 1.00 8318	0.07 9.99	5.81 0.68 0.34 1.13 7694 12691 761 8253 72.9 5.92E-06	43.74 49.39 6.87	11819	4.89 68.91 0.40 1.07 0.009 17.85 6.87	0.07 0.000 0.000
7-2593-C Raw Coal -Silo 2 North 3/8/07 18:45	29.53 31.05 34.28 5.14	48.34 3.47 12.50 0.74 0.27 5.14 29.53 100.00 1.00	0.06 8.58	6.18 0.65 0.32 1.10 7695 12736 760 8254 73.1 5.08E-06	44.06 48.64 7.30	11807	4.93 68.60 0.38 1.05 0.008 17.73 7.30	0.00
7-2592-C Raw Coal -Silo 2 North 3/7/07 8:50	30.05 31.00 34.77 4.19	48.57 3.49 12.70 0.74 0.27 4.19 30.05 100.00 1.00	0.06 8.49	5.01 0.64 0.32 1.12 7731 729 759 8280 735 5.02E-06 5.02	44.31 49.70 5.98	11956	4.99 69.43 0.38 1.06 0.009 18.15 5.98 100.00	0.06 0.000 0.000
7-2591-C Raw Coal -Silo 2 North 3/6/07 18:50	29.57 30.43 35.48 4.52	48.58 3.47 12.86 0.76 0.24 4.52 29.57 100.00 1.00	0.06 8.56	5.43 0.56 0.28 1.17 7708 12846 760 8256 72.9 5.07E-06	43.20 50.37 6.42	11833	4.92 68.98 0.33 1.08 0.008 18.26 6.42 100.00	0.06 0.001 0.000
PPL Sample No. Sample Date Sample I.D.	Fuel Properties - As Received % Total Moisture % Volatile Matter % Fixed Carbon % Ash	% Carbon % Cathon % Uydrogen % Oxygen (diff) % Nitrogen % Sulfur % Sulfur % Ash % Total HHV Btulb	Hg (µgrams/gram dry coal) Hg (µg/m3, 3% O2)	lbs AstivMM Btu bs SOZAMA Btu bs SMAM Btu bs SMAM Btu FC/VM Ratio LHV Q,(net) Btutb HHV (MAF basis) bs Adiven/MMBtu bo LDubng HHV estimate Btufb bs Hg/MM Btu lbs Hg/MM Btu lbs Hg/MB Btu	Fuel Properties - Dry Basis % Volatile Matter % Fixed Carbon % Ash	HHV Btu/lb	% Hydrogen % Carbon % Sulfur % Sulfur % Willogen % Chlorine % Chlorine % Ash % Ash % Total	Hg (ugrams/gram dry coal) % F % Br



7-2612-C Raw Coal -Silc 2 North 3/28/07 10:00	28.82 31.25 35.42 4.51	48.73 3.44 13.50 0.69 0.30 4.51 28.82 100.00 1.00	0.09	5.34 0.71 0.35 1.13 7829 12667 748 8211 70.7 7.59E-06 7.59	43.90 49.76 6.34	11864	4.83 68.47 0.42 0.03 0.008 18.97 6.34	0.09 0.001 0.000
7-2611-C Raw Coal -Silo 2 North 3/27/07 17:10	28.78 31.43 35.42 4.38	48.85 3.42 13.61 0.70 0.27 4.38 28.78 100.00 1.00	0.08 11.53	5.17 0.63 0.32 1.13 7849 12659 747 8210	44.13 49.73 6.14		4.80 68.59 0.38 0.08 19.10 6.14	0.08 0.001 0.000
7-2610-C Raw Coal -Silo 2 North 3/26/07 10:45	28.84 14.81 51.69 4.66	48.96 3.33 13.24 0.72 0.25 4.66 28.84 100.00 1.00	0.09	5.52 0.59 0.29 3.49 7838 12697 748 8195 69.6 7.59E-06 7.59	20.82 72.63 6.55	11865	4.67 68.80 0.35 1.01 0.008 18.60 6.55	0.09 0.001 0.000
7-2609-C Raw Coal -Silo 2 North 3/25/07 10:57	29.87 30.62 34.60 4.90	47.96 3.31 12.98 0.70 0.26 4.90 29.87 100.00 1.00	0.06 8.7	5.93 0.64 0.32 1.13 7657 12681 750 8060 72.1 5.09E-06 5.09	43.67 49.34 6.99	11795	4.72 68.39 0.38 1.01 0.009 18.50 6.99	0.00 0.002 0.000
7-2602-C 7-2603-C 7-2604-C 7-2605-C 7-2606-C 7-2606-C 7-2608-C 7-2609-C 7-2610-C 7-2610-C 7-2611-C 7-2611-C 7-2612-C Raw Coal-Silo Raw Coal-Si	28.93 30.79 35.43 4.85	48.82 3.32 13.11 0.72 0.26 4.85 28.93 100.00 1.00	10.07	5.77 0.61 0.30 1.15 12698 749 8178 69.78 5.92E-06	43.32 49.85 6.83	11831	4.67 68.69 0.36 1.01 0.008 18.43 6.83	0.07 0.001 0.000
7-2607-C Raw Coal -Silo 2 North 3/22/07 9:45	28.57 31.24 35.85 4.35	49.40 3.40 13.35 0.73 0.20 4.35 28.57 100.00 1.00 8548	0.07	5.08 0.46 0.23 1.15 7839 12742 747 8294 69.2 5.85E-06	43.73 50.18 6.08	11967	4.76 69.16 0.28 1.03 0.009 18.68 6.08	0.00 0.000 0.000
7-2606-C Raw Coal -Silo 2 North 3/21/07 14:35	29.58 31.10 34.12 5.20	48.08 3.38 12.62 0.70 0.45 5.20 29.58 100.00 1.00	0.08 11.54	6.24 1.08 0.54 1.10 7714 12775 752 8154 72.0 6.76E-06 6.76	44.17 48.45 7.38	11832	4.80 68.27 0.64 0.99 0.008 17.91 7.38	0.08 0.001 0.000
7-2605-C Raw Coal -Silo 2 North 3/20/07 14:20	29.74 31.04 33.84 5.38	47.84 3.38 12.48 0.71 0.47 5.38 29.74 100.00 1.00	0.08 11.56	6.46 1.14 0.57 1.09 7706 12832 750 8134 72.3 6.75E-06 6.75	44.18 48.16 7.66	11850	4.81 68.09 0.67 1.01 0.009 17.75 7.66	0.08 0.002 0.000
7-2604-C Raw Coal -Silo 2 North 3/19/07 16:37	27.44 32.10 34.69 5.77	49.25 3.47 12.81 0.74 0.52 5.77 27.44 100.00 1.00	0.08	6.72 1.21 0.60 1.08 1975 12843 749 8369 68.4 6.77E-06	44.24 47.81 7.95	11823	4.78 67.87 0.72 1.02 0.009 17.65 7.95	0.08 0.003 0.000
7-2603-C Raw Coal -Silo 2 North 3/18/07 13:35	30.30 30.74 33.40 5.57	47.34 3.38 12.16 0.74 0.52 5.57 30.30 1.00 8231	0.07 10.13	6.77 1.25 0.63 1.09 7607 753 8083 73,7 5.93E-06 5.93	44.10 47.91 7.99	11809	4.84 67.92 0.74 1.06 0.010 17.44 7.99	0.00 0.000 0.000
7-2602-C Raw Coal -Silo 2 North 3/17/07 14:55	31.21 30.05 33.93 4.81	47.05 3.21 12.68 0.73 0.32 4.81 31.21 100.00 1.00	0.07 10.22	5.92 0.78 0.39 1.13 7506 12697 748 7889 74.0 5.93E-06	43.69 49.32 6.99	11810	4.66 68.39 0.46 1.07 0.008 18.42 6.39 100.00	0.000
PPL Sample No. Sample Date Sample I.D.	Fuel Properties - As Received % Total Moisture % Voiatile Matter % Fixed Carbon % Ash	% Carbon % Hydrogen % Oxygen (diff) % Nitrogen % Sulfur % Ash % Ash % Ash % Ash HV Btu/lb	Hg (µgrams/gram dry coal) Hg (µg/m3, 3% O2)	bs AshMM Btu bs SOZNAM Btu bs SOAM Btu FCVM Ralu FCVM Ralu LHV Q _p (nat) Btu/b HHV (MAF Passis) bs Altryvet)MMBtu Dulong HHV estimate Btu/b bs HgVMM Btu bs HgVMM Btu bs HgVMM Btu	Fuel Properties - Dry Basis % Volatile Matter % Fixed Carbon % Ash	HHV Btu/lb	% Hydrogen % Carbon % Sulfur % Nitrogen % Chlorine % Coygen (diff) % Ash % Total	Hg (µgrams/gram dry coal) % F % Br



PPL Sample No.	7-2613-C Raw Coal -Silo	7-2614-C Raw Coal -Silo	7-2613-C 7-2614-C 7-2615-C 7-2624-C Raw Coal -Silo Raw Coal -Silo Raw Coal -Silo Ray Coal -Silo	7-2624-C Raw Coal -Silo
Sample I.D.	2 North 3/29/07 9:15	2 North 3/30/07 10:10	2 North 3/31/07 14:35	2 North 3/24/07 10:45
First Drongeries . As Booghed				
% Total Moisture	34.36	28.48	31.36	28.94
% Votatile Matter	28.76	31.38	30.11	31.39
% Fixed Carbon	32.69	35.40	33.89	35.06
% Asn	4.19	4.74	4.65	4.61
% Carbon	44.89	48.89	47.06	48.74
% Hydrogen	3.12	3.41	3.24	3.35
% Oxygen (dill)	0.00	0.40	27.72	0.50
% Suffur	0.27	0.31	0.27	0.33
% Ash	4.19	4.74	4.65	4.61
% Moisture	34.36	28.48	31.36	28.94
% 10tal	0.00	0.00	0.00	100.00
HHV Btu/lb	7788	8503	8146	8471
Hg (µgrams/gram dry coal)	0.07	0.08	0.08	0.08
(20 %¢ ,em/g4) gn	10.33	8	<u> </u>	66.
lbs Ash/MM Btu	5.38	5.58	5.70	5.45
ibs SOZIMM Btu	0.00	0.73	0.00	0.39
FC/VM Ratio	1.14	1.13	1.13	1.12
LHV Qp(net) Btu/lb	7145	7894	7523	7862
HHV (MAF basis)	12673	12734	12730	12748
Doubord HHV settmate Bruth	75.78	8224	7007	1 5
lbs H2O/MM Btu	80.2	9.69	74.3	69.8
lbs Hg/MM Btu	5.90E-06	6.73E-06	6.74E-06	6.71E-06
lbs Hg/TBtu	5.90	6.73	6.74	6.71
Fuel Properties - Dry Basis	;	!	!	!
% Volatile Matter % Fixed Carbon	43.82	43.87	43.86	44.17
% Ash	6.38	6.63	6.77	6.49
HHV Btu/lb	11864	11890	11869	11921
% Hydrogen	4.75	4.77	4.73	4.72
% Carbon % Suffix	68.39	68.36	68.56	68.59
% Nitrogen	75.0	0.97	103	10.5
% Chlorine	0.007	0.008	0.008	0.008
% Oxygen (diff)	19.08	18.83	18.52	18.72
% Asn % Total	100.00	100.00	100.00	100.00
Hg (µgrams/gram dry coal) % F	0.07	90.0	90:0	90.0
. ii %	0.000	0.000	0.000	0.000



PPL Sample No.	7-2591-C	7-2592-C	7-2593-C	7-2594-C	7-2595-C	7-2596-C	7-2597-C
Sample Date	Raw Coal -Silo 2 North	Raw Coal -Sile 2 North					
Sample I.D.	3/6/07 18:50	3/7/07 8:50	3/8/07 18:45	3/9/07 14:50	3/10/07 8:20	3/11/07 14:00	3/12/07 8:50
Fuel Properties - As Rec	eived						
% Total Moisture	29.57	30.05	29.53	29.62	29.55	27.65	28.18
% Volatile Matter	30.43	31.00	31.05	30.78	31.13	31.87	31.26
% Fixed Carbon	35.48	34.77	34.28	34.76	34.93	36.27	35.34
% Ash	4.52	4.19	5.14	4.83	4.39	4.22	5.22
HHV Btu/lb	8334	8364	8321	8318	8385	8630	8445
% Moisture	29.57	30.05	29.53	29.62	29.55	27.65	28.18
% Hydrogen	3.47	3.49	3.47	3.44	3.34	3.43	3.37
% Carbon	48.58	48.57	48.34	48.50	48.84	49.98	49.03
% Sulfur	0.24	0.27	0.27	0.28	0.29	0.27	0.34
% Nitrogen	0.76	0.74	0.74	0.75	0.73	0.72	0.75
% Oxygen (diff)	12.86	12.70	12.50	12.57	12.85	13.73	13,11
% Ash	4.52	4.19	5.14	4.83	4.39	4.22	5.22
% Total	100.00	100.00	100.00	100.00	100.00	100.00	100.00
lbs Ash/MM Btu	5.43	5.01	6.18	5.81	5.24	4.89	6.18
lbs SO2/MM Btu	0.56	0.64	0.65	0.68	0.68	0.63	0.80
FC/VM Ratio	1.17	1.12	1.10	1.13	1.12	1.14	1.13
LHV Q _o (net) Btu/lb	7708	7731	7695	7694	7771	8027	7843
HHV (MAF basis)	12646	12717	12736	12691	12693	12666	12681
lbs Air(wet)/MMBtu	760	759	760	761	754	747	752
Dulong HHV estimate Btu/		8280	8254	8253	8217	8372	8244
lbs H2O/MM Btu	72.9	73.5	73.1	72.9	71.1	67.8	69.3
lbs Hg/MM Btu	5.07E-06	5.02E-06	5.08E-06	5.92E-06	7.56E-06	5.03E-06	7.65E-06
Fuel Properties - Dry Bas	is						
% Volatile Matter	43.20	44.31	44.06	43.74	44.19	44.05	43.53
% Fixed Carbon	50.37	49.70	48.64	49.39	49.57	50.12	49.20
% Ash	6.42	5.98	7.30	6.87	6.23	5.83	7.27
HHV Btu/lb	11833	11956	11807	11819	11902	11928	11759
% Hydrogen	4.92	4.99	4.93	4.89	4.74	4.74	4.69
% Carbon	68.98	69.43	68.60	68.91	69.33	69.08	68.26
% Sulfur	0.33	0.38	0.38	0.40	0.41	0.37	0.47
% Nitrogen	1.08	1.06	1.05	1.07	1.04	1.00	1.04
% Chlorine	0.008	0.009	0.008	0.009	0.009	0.008	0.008
% Oxygen (diff)	18.26	18.15	17.73	17.85	18.24	18.97	18.25
% Ash	6.42	5.98	7.30	6.87	6.23	5.83	7.27
% Total	100.00	100.00	100.00	100.00	100.00	100.00	100.00
Hg (μgrams/gram dry coal)		0.06	0.06	0.07	0.09	0.06	0.09
% F	0.001	0.000	0.000	0.000	0.001	0.001	0.000
% Br	0.000	0.000	0.000	0.000	0.000	0.000	0.000



PPL Sample No.	7-2598-C	7-2599-C	7-2600-C	7-2601-C	7-2602-C	7-2603-C	7-2604-C
Sample Date	Raw Coal -Silo 2 North						
Sample i.D.	3/13/07 8:30	3/14/07 8:20	3/15/07 16:25	3/16/07 14:00	3/17/07 14:55	3/18/07 13:35	3/19/07 16:37
Fuel Properties - As Rece	•						
% Total Moisture	33.82	29.97	30.59	20.74	31.21	30.30	27.44
% Volatile Matter	28.94	31.14	30.71	34.65	30.05		
% Fixed Carbon	32.04	32.84	34.24	39.10	33.93	30.74	32.10 34.69
% Ash	5.20	6.05	34.24 4.46	59.10 5.51	33.93 4.81	33.40	
70 ASII	5.20	6.05	4.40	5.51	4.61	5.57	5.77
HHV Btu/lb	7785	8247	8245	9358	8124	8231	8579
% Moisture	33.82	29.97	30.59	20.74	31.21	30.30	27.44
% Hydrogen	3.11	3.32	3.25	3.66	3.21	3.38	3.47
% Carbon	45.00	47.43	47.78	54.22	47.05	47.34	49.25
% Sulfur	0.31	0.54	0.27	0.34	0.32	0.52	0.52
% Nitrogen	0.77	0.75	0.73	0.83	0.73	0.74	0.74
% Oxygen (diff)	11.80	11.93	12.91	14.69	12.68	12.16	12.81
% Ash	5.20	6.05	4.46	5.51	4.81	5.57	5.77
% Total	100.00	100.00	100.00	100.00	100.00	100.00	100.00
lbs Ash/MM Btu	6.68	7.34	5.41	5.89	5.92	6.77	6.72
lbs SO2/MM Btu	0.79	1.32	0.65	0.72	0.78	1.25	1.21
FC/VM Ratio	1.11	1.05	1.11	1.13	1.13	1.09	1.08
LHV Qp(net) Btu/lb	7149	7631	7629	8805	7506	7607	7975
HHV (MAF basis)	12767	12891	12694	12689	12697	12834	12843
lbs Air(wet)/MMBtu	750	752	748	747	748	753	749
Dulong HHV estimate Btu/		8083	8003	9063	7889	8083	8369
lbs H2O/MM Btu	79.4	72.6	72.6	57.4	74.0	73.7	68.4
lbs Hg/MM Btu	5.95E-06	7.64E-06	6.73E-06	6.78E-06	5.93E-06	5.93E-06	6.77E-06
•							
Fuel Properties - Dry Bas	;						
% Volatile Matter	43.73	44.46	44.25	43.72	43.69	44.10	44.24
% Fixed Carbon	48.41	46.89	49.33	49.33	49.32	47.91	47.81
% Ash	7.86	8.64	6.42	6.95	6.99	7.99	7.95
HHV Btu/lb	11764	11777	11879	11807	11810	11809	11823
% Hydrogen	4.70	4.75	4.69	4.62	4.66	4.84	4.78
% Carbon	67.99	67.73	68.84	68.41	68.39	67.92	67.87
% Sulfur	0.46	0.78	0.39	0.42	0.46	0.74	0.72
% Nitrogen	1.16	1.07	1.06	1.05	1.07	1.06	1.02
% Chlorine	0.010	0.010	0.009	0.008	0.008	0.010	0.009
% Oxygen (diff)	17.82	17.02	18.59	18.53	18.42	17.44	17.65
% Ash	7.86	8.64	6.42	6.95	6.99	7.99	7.95
% Total	100.00	100.00	100.00	100.00	100.00	100.00	100.00
70 TOWN	100.00	100.00	100.00	100.00	100.00	100.00	100.00
Hg (µgrams/gram dry coal)	0.07	0.09	0.08	0.08	0.07	0.07	0.08
% F	0.001	0.000	0.001	0.000	0.000	0.006	0.003
% Br	0.000	0.000	0.000	0.000	0.000	0.000	0.000



PPL Sample No.	7-2605-C	7-2606-C	7-2607-C	7-2608-C	7-2609-C	7-2610-C	7-2611-C
Sample Date	Raw Coal -Silo 2 North						
Sample I.D.	3/20/07 14:20	3/21/07 14:35	3/22/07 9:45	3/23/07 16:00		3/26/07 10:45	
Fuel Properties - As Reco	•						
% Total Moisture	29.74	29.58	28.57	28.93	29.87	00.04	20.70
% Volatile Matter	31.04	31.10	31.24	30.79	29.67 30.62	28.84 14.81	28.78
% Fixed Carbon	33.84	34.12	35.85	35.43	34.60	51.69	31.43 35.42
% Ash	5.38	5.20	4.35	4.85	4.90		
70 F-311	3.30	3.20	4.33	4.65	4.90	4.66	4.38
HHV Btu/lb	8326	8332	8548	8408	8271	8443	8462
% Moisture	29.74	29.58	28.57	28.93	29.87	28.84	28.78
% Hydrogen	3.38	3.38	3.40	3.32	3.31	3.33	3.42
% Carbon	47.84	48.08	49.40	48.82	47.96	48.96	48.85
% Sulfur	0.47	0.45	0.20	0.26	0.26	0.25	0.27
% Nitrogen	0.71	0.70	0.73	0.72	0.70	0.72	0.70
% Oxygen (diff)	12.48	12.62	13.35	13.11	12.98	13.24	13.61
% Ash	5.38	5.20	4.35	4.85	4.90	4.66	4.38
% Total	100.00	100.00	100.00	100.00	100.00	100.00	100.00
lbs Ash/MM Btu	6.46	6.24	5.08	5.77	5.93	5.52	5.17
lbs SO2/MM Btu	1.14	1.08	0.46	0.61	0.64	0.59	0.63
FC/VM Ratio	1.09	1.10	1.15	1.15	1.13	3.49	1.13
LHV Q _p (net) Btu/lb	7706	7714	7939	7802	7657	7838	7849
HHV (MAF basis)	12832	12775	12742	12698	12681	12697	12659
lbs Air(wet)/MMBtu	750	752	747	749	750	748	747
Dulong HHV estimate Btu/l	8134	8154	8294	8178	8060	8195	8210
lbs H2O/MM Btu	72.3	72.0	69.2	69.9	72.1	69.6	70.4
lbs Hg/MM Btu	6.75E-06	6.76E-06	5.85E-06	5.92E-06	5.09E-06	7.59E-06	#REF!
Fuel Properties - Dry Bas							
% Volatile Matter	44.18	44.17	43.73	43.32	43.67	20.82	44.13
% Fixed Carbon	48.16	48.45	50.18	49.85	49.34	72.63	49.73
% Ash	7.66	7.38	6.08	6.83	6.99	6.55	6.14
		1.00	0.00	0.00	0.00	0.00	0.14
HHV Btu/lb	11850	11832	11967	11831	11795	11865	#REF!
% Hydrogen	4.81	4.80	4.76	4.67	4.72	4.67	4.80
% Carbon	68.09	68.27	69.16	68.69	68.39	68.80	68.59
% Sulfur	0.67	0.64	0.28	0.36	0.38	0.35	0.38
% Nitrogen	1.01	0.99	1.03	1.01	1.01	1.01	0.98
% Chlorine	0.009	0.008	0.009	0.008	0.009	0.008	0.008
% Oxygen (diff)	17.75	17.91	18.68	18.43	18.50	18.60	19.10
% Ash	7.66	7.38	6.08	6.83	6.99	6.55	6.14
% Total	100.00	100.00	100.00	100.00	100.00	100.00	100.00
							100.00
Hg (µgrams/gram dry coal)	0.08	0.08	0.07	0.07	0.06	0.09	0.08
% F	0.002	0.001	0.000	0.001	0.002	0.001	0.001
% Br	0.000	0.000	0.000	0.000	0.000	0.000	0.000
							-



PPL Sample No.	7-2612-C	7-2613-C	7-2614-C	7-2615-C	7-2624-C
Sample Date	2 North	Raw Coal -Silo 2 North	2 North	2 North	Raw Coal -Silo 2 North
Sample I.D.	3/28/07 10:00	3/29/07 9:15	3/30/07 10:10	3/31/07 14:35	3/24/07 10:45
Fuel Properties - As Rec	•				
% Total Moisture	28.82	34.36	28.48	31.36	28.94
% Volatile Matter	31.25	28.76	31.38	30.11	31.39
% Fixed Carbon	35.42	32.69	35.40	33.89	35.06
% Ash	4.51	4.19	4.74	4.65	4.61
HHV Btu/lb	8445	7788	8503	8146	8471
% Moisture	28.82	34.36	28.48	31.36	28.94
% Hydrogen	3.44	3.12	3.41	3.24	3.35
% Carbon	48.73	44.89	48.89	47.06	48.74
% Sulfur	0.30	0.27	0.31	0.27	0.33
% Nitrogen	0.69	0.64	0.69	0.70	0.72
% Oxygen (diff)	13.50	12.53	13.48	12.72	13.30
% Ash	4.51	4.19	4.74	4.65	4.61
% Total	100.00	100.00	100.00	100.00	100.00
lbs Ash/MM Btu	5.34	5.38	5.58	5.70	5.45
lbs SO2/MM Btu	0.71	0.70	0.73	0.66	0.77
FC/VM Ratio	1.13	1.14	1.13	1.13	1.12
LHV Q _p (net) Btu/lb	7829	7145	7894	7523	7862
HHV (MAF basis)	12667	12673	12734	12730	12748
lbs Air(wet)/MMBtu	748	745	744	747	744
Dulong HHV estimate Btu/	8211	7528	8221	7907	8177
lbs H2O/MM Btu	70.7	80.2	69.6	74.3	69.8
lbs Hg/MM Btu	7.59E-06	5.90E-06	6.73E-06	6.74E-06	6.71E-06
Fuel Properties - Dry Bas	;				
% Volatile Matter	43.90	43.82	43.87	43.86	44.17
% Fixed Carbon	49.76	49.80	49.50	49.37	49.34
% Ash	6.34	6.38	6.63	6.77	6.49
HHV Btu/lb	11864	11864	11890	11869	11921
% Hydrogen	4.83	4.75	4.77	4.73	4.72
% Carbon	68.47	68.39	68.36	68.56	68.59
% Sulfur	0.42	0.42	0.43	0.39	0.46
% Nitrogen	0.97	0.97	0.97	1.03	1.01
% Chlorine	0.008	0.007	800.0	0.008	0.008
% Oxygen (diff)	18.97	19.08	18.83	18.52	18.72
% Ash	6.34	6.38	6.63	6.77	6.49
% Total	100.00	100.00	100.00	100.00	100.00
Hg (μgrams/gram dry coal)		0.07	0.08	0.08	0.08
% F	0.001	0.001	0.000	0.000	0.000
% Br	0.000	0.000	0.000	0.000	0.000



PPL # Sample Location Sample ID Date Time	7-2464-A	7-2465-A	7-2466-A	7-2467-A	7-2468-A	7-2469-A	7-2470-A
	North ESP	North ESP	North ESP				
	Hopper 6	Hopper 6	Hopper 6				
	A 13 N	A 23 N	A 33 N	A 43 N	A 53 N	A 63 N	B 13 S
	03/06/07	03/06/07	03/06/07	03/06/07	03/06/07	03/06/07	03/06/07
	15:10	15:15	15:20	15:25	15:25	15:30	15:37
% LOI % Total C μg Hg/gram ash	0.09 0.11	0.03 0.20	0.02 0.26	0.03 0.28	0.03 0.21	0.01 0.23	0.11 0.05 0.06
PPL # Sample Location Sample ID Date Time	7-2471-A	7-2472-A	7-2473-A	7-2475-A	7-2476-A	7-2477-A	7-2478-A
	North ESP	North ESP	North ESP				
	Hopper 6	Hopper 6	Hopper 6				
	B 23 S	B 33 S	B 43 S	A 13 N	A 23 N	A 33 N	A 43 N
	03/06/07	03/06/07	03/06/07	03/07/07	03/07/07	03/07/07	03/07/07
	15:35	15:33	15:32	9:30	9:30	9:30	9:30
% LOI % Total C μg Hg/gram ash	0.15 0.02 0.12	0.02 0.21	0.01 0.25	0.18 0.09 0.08	0.18 0.02 0.15	0.01 0.21	0.01 0.23
PPL # Sample Location Sample ID Date Time	7-2479-A	7-2480-A	7-2481-A	7-2482-A	7-2483-A	7-2484-A	7-2485-A
	North ESP	North ESP	North ESP				
	Hopper 6	Hopper 6	Hopper 6				
	A 53 N	A 63 N	B 13 S	B 23 S	B 33 S	B 43 S	3 A 13 W
	03/07/07	03/07/07	03/07/07	03/07/07	03/07/07	03/07/07	03/09/07
	9:35	9:40	9:45	9:45	9:40	9:40	8:25
% LOI % Total C μg Hg/gram ash	0.02 0.22	0.01 0.15	0.12 0.06 0.06	0.02 0.06	0.0 4 0.11	0.01 0.19	0.09 0.09
PPL# Sample Location Sample ID Date Time	7-2486-A	7-2487-A	7-2488-A	7-2489-A	7-2490-A	7-2491-A	7-2492-A
	North ESP	North ESP	North ESP				
	Hopper 6	Hopper 6	Hopper 6				
	3A 23	3A 33	3A 43	3A 53	3A 63 E	3B 43	3B 33
	03/09/07	03/09/07	03/09/07	03/09/07	03/09/07	03/09/07	03/09/07
	8:28	8:30	8:33	8:36	8:38	8:43	8:47
% LOI % Total C μg Hg/gram ash	0.03 0.14	0.02 0.15	0.01 0.17	0.01 0.20	0.05 0.12	0.01 0.12	0.02 0.10
PPL # Sample Location Sample ID Date Time	7-2493-A	7-2494-A	7-2495-A	7-2496-A	7-2497-A	7-2498-A	7-2499-A
	North ESP	North ESP	North ESP				
	Hopper 6	Hopper 6	Hopper 6				
	3B 23	3B 13	A13 N	A23 N	A33 N	A43 N	A53 N
	03/09/07	03/09/07	03/19/07	03/19/07	03/19/07	03/19/07	03/19/07
	8:50	8:52	15:11	15:13	15:17	15:20	15:22
% LOI % Total C µg Hg/gram ash	0.03 0.09	0.30 0.26 0.05	0.13 0.34	0.09 0.84	0.07 0.86	0.08 1.00	0.07 0.83



PPL#	7-2500-A North ESP	7-2501-A North ESP	7-2502-A North ESP	7-2503-A North ESP	7-2504-A North ESP	7-2505-A North ESP	7-2506-A North ESP
Sample Location	Hopper 6						
Sample ID	A63 N	B13 S 03/19/07	B23 S 03/19/07	B33 S 03/19/07	B43 S 03/19/07	A13 N 03/20/07	A23 N 03/20/07
Date Time	03/19/07 15:24	15:21	15:14	15:16	15:20	14:25	14:23
111110	no sample	.0.21	10.71	101.10	.0.20		
% LOI							
% Total C μg Hg/gram ash		0.2 0.42	0.08 0.42	0.07 0.66	0.08 0.82	0.19 0.62	0.5 1.90
µg rigigiani don		0.72	U.7E	0.00	0.02	0.02	1.00
PPL#	7-2507-A	7-2508-A	7-2509-A	7-2510-A	7-2511-A	7-2512-A	7-2513-A
Sample Location	North ESP						
Sample ID	Hopper 6 A33 N	Hopper 6 A43 N	Hopper 6 A53 N	Hopper 6 A63 N	Hopper 6 B13	Hopper 6 B23	Hopper 6 B33
Date	03/20/07	03/20/07	03/20/07	03/20/07	03/20/07	03/20/07	03/20/07
Time	14:22	14;21	14:19	14:17	14;19	14:18	14:16
% LOI							
% Total C	0.21	0.63	0.25	0.15	0.25	0.15	0.29
μg Hg/gram ash	1.70	2.90	2.00	1.20	0.54	1.20	2.00
PPL#	7-2514-A North ESP	7-2515-A North ESP	7-2516-A North ESP	7-2517-A North ESP	7-2518-A North ESP	7-2519-A North ESP	7-2520-A North ESP
Sample Location	Hopper 6						
Sample ID	B43	A13 -2N	A23 - 2N	A33 - 2N	A43 - 2N	A53 - 2N	A63 - 2N
Date Time	03/20/07 14:14	03/21/07 11:00	03/21/07 11:00	03/21/07 11;00	03/21/07 11:00	03/21/07 11:00	03/21/07 11:00
rime	14,14	11.00	11.00	11,00	11.00	11.00	11.00
% LOI							
% Total C μg Hg/gram ash	0.33 2.10	0.33 0.96	0.67 2.40	1.05 4.00	1.10 4.50	0.62 3.60	0.21 1.30
μg rig/grain asis	2.10	0.50	2.40	4.00	4.50	3.00	1.50
PPL#	7-2521-A	7-2522-A	7-2523-A	7-2524-A	7-2525-A	7-2526-A	7-2527-A
Sample Location	North ESP						
Sample ID	Hopper 6 B13 - 2S	Hopper 6 B23 - 2S	Hopper 6 B33 - 2S	Hopper 6 B43 - 2S	Hopper 6 A13 N	Hopper 6 A23 N	Hopper 6 A33 N
Date	03/21/07	03/21/07	03/21/07	03/21/07	03/21/07	03/21/07	03/21/07
Time	11:00	11:00	11:00	11:00	11:00	11:00	11:00
% LOI							
% Total C	0.30	0.47	0.71	0.80	0.37	0.68	1.04
μg Hg/gram ash	0.87	1.60	2.80	3.30	0.94	1.90	3.20



PPL# Sample Location Sample ID Date Time	7-2528-A	7-2529-A	7-2530-A	7-2531-A	7-2532-A	7-2533-A	7-2534-A
	North ESP						
	Hopper 6						
	A43 N	A53 N	A63 N	B13 S	B23 S	B33 S	B43 S
	03/21/07	03/21/07	03/21/07	03/21/07	03/21/07	03/21/07	03/21/07
	11:00	11:00	11:00	11:00	11:00	11:00	11:00
% LOI % Total C μg Hg/gram ash	1.10 3.60	0.61 2.80	0.21 1.40	0.28 0.76	0.46 1.60	0.65 2.80	0.80 3.40
PPL# Sample Location Sample ID Date Time % LOI	7-2535-A	7-2536-A	7-2537-A	7-2538-A	7-2539-A	7-2540-A	7-2541-A
	North ESP						
	Hopper 6						
	hopper B	hopper B	hopper B	hopper C	A13	A13 - 2	A23
	03/22/07	03/22/07	03/22/07	03/22/07	03/23/07	03/23/07	03/23/07
	2:42	2:45	base line	base line	10:45	10:45	10:43
% Total C	0.08	0.08	1.17	0.15	0.29	0.30	0.53
μg Hg/gram ash	0.16	0.16	0.05	0.07	0.67	0.70	1.80
PPL# Sample Location Sample ID Date Time	7-2542-A	7-2543-A	7-2544-A	7-2545-A	7-2546-A	7-2547-A	7-2548-A
	North ESP						
	Hopper 6						
	A23 - 2	A33	A33 - 2	A43	A43 - 2	A53	A53 - 2
	03/23/07	03/23/07	03/23/07	03/23/07	03/23/07	03/23/07	03/23/07
	10:44	10:47	10:47	10:50	10;50	10:57	10:57
% LOI % Total C μg Hg/gram ash	0.51 1.70	0.80 2.70	0.77 2.70	0.78 2.90	0.84 3.10	0.50 3.80	0.52 2.60
PPL# Sample Location Sample ID Date Time	7-2549-A	7-2550-A	7-2551-A	7-2552-A	7-2553-A	7-2554-A	7-2555-A
	North ESP						
	Hopper 6						
	A63	A63 - 2	B13	B13 - 2	B23	B23 - 2	B33
	03/23/07	03/23/07	03/23/07	03/23/07	03/23/07	03/23/07	03/23/07
	11:02	11:02	11:20	11:20	11:15	11:15	11:09
% LOI % Total C μg Hg/gram ash	0.26 1.20	0.28 1.20	0.27 0.56	0.25 0.65	0.46 1.50	0.48 1.70	0.59 2.70
PPL # Sample Location Sample ID Date Time	7-2556-A	7-2557-A	7-2558-A	7-2559-A	7-2560-A	7-2561-A	7-2562-A
	North ESP						
	Hopper 6						
	B33 - 2	B43	B43 - 2	hopper B	hopper B2	A13	A23
	03/23/07	03/23/07	03/23/07	03/23/07	03/23/07	03/28/07	03/28/07
	11:09	11:04	11:04	11:30	11:30	11:45	11:45
% LOI % Total C μg Hg/gram ash	0.69 2.90	0.65 3.00	0.67 3.30	0.09 0.09	0.11 0.07	0.33 0.89	0.48 1.60



PPL # Sample Location Sample ID Date Time % LOI	7-2563-A North ESP Hopper 6 A33 03/28/07 11:45	7-2564-A North ESP Hopper 6 A43 03/28/07 11:45	7-2565-A North ESP Hopper 6 A53 03/28/07 11:45	7-2566-A North ESP Hopper 6 A63 03/28/07 11:45	7-2567-A North ESP Hopper 6 B13 03/28/07 11:45	7-2568-A North ESP Hopper 6 B23 03/28/07 11:45	7-2569-A North ESP Hopper 6 B33 03/28/07 11:45
% Total C μg Hg/gram ash	0.84 2.60	0.51 2.50	0.42 2.50	0.23 1.30	0.35 0.54	0.34 1.50	0.48 2.40
PPL # Sample Location Sample ID Date Time	7-2570-A North ESP Hopper 6 B43 03/28/07 11:45	7-2571-A North ESP Hopper 6 hopper B 03/28/07 11:47	7-2572-A North ESP Hopper 6 hopper C 03/28/07 11;47	7-2573-A North ESP Hopper 6 A13 03/30/07 14:15	7-2574-A North ESP Hopper 6 A23 03/30/07 14:15	7-2575-A North ESP Hopper 6 A33 03/30/07 14:15	7-2576-A North ESP Hopper 6 A43 03/30/07 14:15
% LOI % Total C μg Hg/gram ash	0.47 3.00	0.07 0.12	0.12 0.06	0.23 0.67	0.39 1.60	0.61 2.60	0.67 3.20
PPL# Sample Location Sample ID Date Time	7-2577-A North ESP Hopper 6 A53 03/30/07 14:15	7-2578-A North ESP Hopper 6 A63 03/30/07 14:15	7-2579-A North ESP Hopper 6 B13 03/30/07 14:15	7-2580-A North ESP Hopper 6 B23 03/30/07 14:15	7-2581-A North ESP Hopper 6 B33 03/30/07 14:15	7-2582-A North ESP Hopper 6 B43 03/30/07 14:15	7-2583-A North ESP Hopper 6 hopper B 03/30/07 14:15
% LOI % Total C µg Hg/gram ash	0.49 2.80	0.27 1.70	0.17 0.45	0.38 1.70	0.53 2.80	0.58 3.50	0.12 0.11
PPL#	7-2584-A	7-2586-A	7-2587-A	7-2588-A	7-2589-A	7-2590-A	
Sample Location	North ESP Hopper 6	Fly Ash - Silo 3A	Fly Ash - Silo 3B-01	Fly Ash - Silo 3B-02	Fly Ash - Silo 3B (1 of 2)	Fly Ash - Silo 3B (2 of 2)	
Sample ID Date Time	hopper C 03/30/07 14:15	03/13/07	03/23/07 17:15	03/23/07 17:15	03/30/07 16:15	03/30/07 16:15	
% LOI % Total C μg Hg/gram ash	0.08 0.09	0.05 0.28	0.26 1.30	0.26 1.30	0.25 1.40	0.26 1.40	
PPL#	7-2616-A	7-2617-A	7-2618-A	7-2676-M	7-2677-M	7-2678-M FGD Filter	7-2679-M
Sample Location	Fly Ash - Silo B	Fly Ash - Silo B	Fly Ash - Silo B			Cake - Vacuum Filter Belt	FGD Filter Cake - Pile
Sample ID Date Time	hopper C 03/28/07 15:05	3/29/2007 11:35	3/30/2007 9:00			3/12/2007 17:00	3/12/2007 9:29
% LOI % Total C μg Hg/gram ash	0.26 1.40	0.28 1.50	0.29 1.60			1.70	1.80



-A 7-2618-A silo B h fly ash 07 03/30/07 5 9:00				3.6						
7-2617-A silo B fly ash 03/29/07 11:35			^	3.6	< 0.1	0.32	< 0.5	< 0.00	< 0.5	× 0.1
7-2616-A silo B fly ash 03/28/07 15:05			^	3.7	< 0.1	0.31	< 0.5	< 0.001	< 0.5	< 0.1
7-2590-A) silo 3B (2 of 2) fly ash 03/30/07 16:15			^	3.6	< 0.1	0.28	< 0.5	< 0.001	< 0.5	< 0.1
7-2589-A silo 3B (1 of 2) fly ash 03/30/07 16:15			^	4.3	< 0.1	0.32	< 0.5	< 0.001	< 0.5	< 0.1
7-2588-A silo 38-02 fly ash 03/23/07 17:15			^ 1	2.9	< 0.1	0.28	< 0.5	< 0.001	< 0.5	< 0.1
7-2587-A silo 3B-01 fly ash 3/23/2007 17:15			, ,	3.3	< 0.1	0.29	< 0.5	< 0.001	< 0.5	< 0.1
7-2586-A silo ash fly ash 03/13/07			^	3.5	< 0.1	0.3	< 0.5	< 0.001	< 0.5	< 0.1
LS SI Lia 100	۷ ئ									
7-2676-M Reclaim Pond FGD Makeup Water 03/12/07 8:57	< 0.2									
PPL # Sample Location Sample ID Date Time	μg Hg/liter	TCLP Extraction	mg/l Arsenic	mg/l Barium	mg/l Cadmium	mg/l Chromium	mg/l Lead	mg/I Mercury	mg/l Selenium	mg/l Silver



ALPHA ANALYTICAL LABORATORIES

Eight Walkup Drive Westborough, Massachusetts 01581-1019 (508) 898-9220 www.alphalab.com

MA:M-MA086 NH:200301-A CT:PH-0574 ME:MA086 RI:65 NY:11148 NJ:MA935 Army:USACE

CERTIFICATE OF ANALYSIS

Client: Alstom Power Laboratory Job Number: L0707573

Address: 2000 Day Hill Road

Windsor, CT 06095 Date Received: 24-MAY-2007

Attn: Mr. Kurt Johnson Date Reported: 08-JUN-2007

Project Number: Delivery Method: UPS

Site: USDOE MERCURY FIELD DEMO

ALPHA SAMPLE NUMBER	CLIENT IDENTIFICATION	SAMPLE LOCATION
L0707573-01	7-2676-M FGD MAKEUP WATE	LCRA-FAYETTE UNIT 3
L0707573-02	7-2677-M FGD LS SLURRY	LCRA-FAYETTE UNIT 3
L0707573-03	7-2586-A FLYASH	LCRA-FAYETTE UNIT 3
L0707573-04	7-2587-A FLYASH 3B1	LCRA-FAYETTE UNIT 3
L0707573-05	7-2588-A FLYASH 3B2	LCRA-FAYETTE UNIT 3
L0707573-06	7-2589-A FLYASH 3B1	LCRA-FAYETTE UNIT 3
L0707573-07	7-2590-A FLYASH 3B2	LCRA-FAYETTE UNIT 3
L0707573-08	7-2616-A FLYASH	LCRA-FAYETTE UNIT 3
L0707573-09	7-2617-A FLYASH	LCRA-FAYETTE UNIT 3
L0707573-10	7-2618-A FLYASH	LCRA-FAYETTE UNIT 3

Authorized by:

Technical Director

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I, the undersigned, attest under the pains and penalties of perjury that, based upon my personal inquiry of those responsible for obtaining the information, the material contained in this report is, to the best of my knowledge and belief, accurate and complete. This certificate of analysis is not complete unless this page accompanies any and all pages of this report.



ALPHA ANALYTICAL LABORATORIES NARRATIVE REPORT

Laboratory Job Number: L0707573

The samples were received in accordance with the chain of custody and no significant deviations were encountered during preparation or analysis unless otherwise noted below.

Sample Receipt

The analysis of Mercury (Total and TCLP) was received with the method required holding time exceeded and was performed at the client's request.

The samples were received at the laboratory above the required temperature range. The samples were not on ice.

Total Mercury

L0707573-02 has an elevated detection limit due to the 25x prep dilution performed. This was done because of the sample matrix.

TCLP Metals

The MS % recovery for Silver is below method acceptance criteria. A post analytical spike was performed with an acceptable recovery of 97%.

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MA:M-MA086 NH:200301-A CT:PH-0574 ME:MA086 RI:65 NY:11148 NJ:MA935 Army:USACE

Laboratory Sample Number: L0707573-01

Date Collected: 12-MAR-2007 08:57

Sample Matrix:

7-2676-M FGD MAKEUP WATE WATER

Date Received : 24-MAY-2007 Date Reported: 08-JUN-2007

Condition of Sample:

Satisfactory

Field Prep: None

Number & Type of Containers: 1-Plastic

PARAMETER	RESULT	UNITS	RDL	RDL REF 1	F METHOD	DAT	e ii
						PREP	ANAL
Total Metals							:
Total Metals							

 $\hbox{{\tt Comments: Complete list of References and Glossary of Terms found in $\tt Addendum I$}\\$

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MA:M-MA086 NH:200301-A CT:PH-0574 ME:MA086 RI:65 NY:11148 NJ:MA935 Army:USACE

Laboratory Sample Number: L0707573-02

Date Collected: 12-MAR-2007 08:28

Sample Matrix:

7-2677-M FGD LS SLURRY

Date Received : 24-MAY-2007

WATER

Date Reported: 08-JUN-2007

Condition of Sample: Satisfactory

Field Prep: None

Number & Type of Containers: 1-Plastic

PARAMETER	RESULT	UNITS	RDL	REF METHOD	DATE PREP ANAL	ID
Total Metals				Sha ka basa		
Mercury, Total	ND	mg/l	0.0050	3 245.1	0604 15:05 0605 14:20) DM

Comments: Complete list of References and Glossary of Terms found in Addendum I

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MA:M-MA086 NH:200301-A CT:PH-0574 ME:MA086 RI:65 NY:11148 NJ:MA935 Army:USACE

Laboratory Sample Number: L0707573-03

Date Collected: 13-MAR-2007 00:00

7-2586-A FLYASH

Date Received: 24-MAY-2007 Date Reported: 08-JUN-2007

Sample Matrix:

Condition of Sample:

SOLID

Satisfactory

Field Prep:

Number & Type of Containers: 1-Can

PARAMETER	RESULT	UNITS	RDL	REF	METHOD	DAT PREP	E ANAL	ID
TCLP Metals TCLP Extraction					e Pojetoja i			٠.
TCLP Extraction				1	1311	0605 17:30		
Arsenic, TCLP	ND	mg/l	1.0	1	6010B	0606 11:20 0	0607 13:30) MG
Barium, TCLP	3.5	mg/l	0.50	1	6010B	0606 11:20 0	607 13:30) MG
Cadmium, TCLP	ND	mg/l	0.10	1	6010B	0606 11:20 0	607 13:30) MG
Chromium, TCLP	0.30	mg/l	0.20	1	6010B	0606 11:20 0	607 13:30) MG
Lead, TCLP	ND	mg/l	0.50	1	6010B	0606 11:20 0	607 13:30) MG
Mercury, TCLP	ND	mg/l	0.0010	1	7470A	0606 15:30 0	607 10:05	5 DM
Selenium, TCLP	ND	mg/l	0.50	1	6010B	0606 11:20 0	607 13:30) MG
Silver, TCLP	ND	mg/l	0.10	1	6010B	0606 11:20 0	607 13:30) MG

Comments: Complete list of References and Glossary of Terms found in Addendum I

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MA:M-MA086 NH:200301-A CT:PH-0574 ME:MA086 RI:65 NY:11148 NJ:MA935 Army:USACE

Laboratory Sample Number: L0707573-04 T-2587-A FLYASH 3B1 Date Collected: 23-MAR-2007 17:15

Date Received: 24-MAY-2007 Date Reported: 08-JUN-2007

Sample Matrix:

SOLID

Condition of Sample: Satisfactory Field Prep: None

Number & Type of Containers: 1-Can

PARAMETER	RESULT	UNITS	RDL	REF	METHOD	DA'			ID
		*				PREP	AN	IAL	
TCLP Metals	(13.775FFEEEE355EE5;				nan yanan i		W		
TCLP Extraction				1	1311	0605 17:30			
Arsenic, TCLP	ND	mg/l	1.0	1	6010B	0606 11:20	0607	13:39	MG
Barium, TCLP	3.3	mg/1	0.50	1	6010B	0606 11:20	0607	13:39	MG
Cadmium, TCLP	ND	mg/l	0.10	1	6010B	0606 11:20	0607	13:39	MG
Chromium, TCLP	0.29	mg/l	0.20	1	6010B	0606 11:20	0607	13:39	MG
Lead, TCLP	ND	mg/l	0.50	1	6010B	0606 11:20	0607	13:39	MG
Mercury, TCLP	ND	mg/l	0.0010	1	7470A	0606 15:30	0607	10:07	DM
Selenium, TCLP	ND	mg/l	0.50	1	60108	0606 11:20	0607	13:39	MG
	ND	mq/l	0.10		6010B	0606 11:20			

Comments: Complete list of References and Glossary of Terms found in Addendum I

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MA:M-MA086 NH:200301-A CT:PH-0574 ME:MA086 RI:65 NY:11148 NJ:MA935 Army:USACE

Laboratory Sample Number: L0707573-05

Date Collected: 23-MAR-2007 17:15

Sample Matrix:

7-2588-A FLYASH 3B2

Date Received: 24-MAY-2007 Date Reported: 08-JUN-2007

SOLID

Condition of Sample:

Satisfactory

Field Prep: 1

None

Number & Type of Containers: 1-Can

PARAMETER	RESULT	UNITS	RDL	REF	METHOD	DA' PREP	PE ANAL	ID
TCLP Metals TCLP Extraction				1.1.2.3	1311	0605 17:30		
Arsenic, TCLP	ND	mg/l	1.0		6010B	0606 11:20	0607 13.4	2 MC
Barium, TCLP	2.9	mg/l	0.50		6010B	0606 11:20		
Cadmium, TCLP	ND	mg/l	0.10	1		0606 11:20		
Chromium, TCLP	0.28	mg/l	0.20	1	6010B	0606 11:20	0607 13:4	2 MG
Lead, TCLP	ND	mg/l	0.50	1	6010B	0606 11:20	0607 13:4	2 MG
Mercury, TCLP	ND	mg/l	0.0010	1	7470A	0606 15:30	0607 10:0	8 DM
Selenium, TCLP	ND	mg/l	0.50	1	6010B	0606 11:20	0607 13:4	2 MG
Silver, TCLP	ND	mg/l	0.10	1	6010B	0606 11:20	0607 13:4	2 MG

Comments: Complete list of References and Glossary of Terms found in Addendum I

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MA:M-MA086 NH:200301-A CT:PH-0574 ME:MA086 RI:65 NY:11148 NJ:MA935 Army:USACE

Laboratory Sample Number: L0707573-06

Date Collected: 30-MAR-2007 16:15

7-2589-A FLYASH 3B1

Date Received : 24-MAY-2007

Sample Matrix:

SOLID

Date Reported: 08-JUN-2007

Condition of Sample:

Satisfactory

Field Prep:

None

Number & Type of Containers: 1-Can

PARAMETER	RESULT	UNITS	RDL	REF	METHOD	DAT	E	ID
						PREP	ANAL	
TCLP Metals						ng sayarayaa		
TCLP Extraction				1	1311	0605 17:30		**********
Arsenic, TCLP	ND	mg/l	1.0	1	6010B	0606 11:20 (607 13:	44 MG
Barium, TCLP	4.3	mg/l	0.50	1	6010B	0606 11:20	607 13:	44 MG
Cadmium, TCLP	ND	mg/l	0.10	1	6010B	0606 11:20	607 13:	44 MG
Chromium, TCLP	0.32	mg/l	0.20	1	6010B	0606 11:20	0607 13:	44 MG
Lead, TCLP	ND	mg/l	0.50	1	6010B	0606 11:20	607 13:	44 MG
Mercury, TCLP	ND	mg/l	0.0010	1	7470A	0606 15:30	607 10:	14 DM
Selenium, TCLP	ND	mg/l	0.50	1	6010B	0606 11:20	0607 13:	44 MG
Silver, TCLP	ND	mg/l	0.10	1	6010B	0606 11:20	1607 13:	44 MG

Comments: Complete list of References and Glossary of Terms found in Addendum I

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MA:M-MA086 NH:200301-A CT:PH-0574 ME:MA086 RI:65 NY:11148 NJ:MA935 Army:USACE

Laboratory Sample Number: L0707573-07

Date Collected: 20-MAR-2007 16:15

Date Received : 24-MAY-2007

Sample Matrix:

7-2590-A FLYASH 3B2 SOLID

Date Reported: 08-JUN-2007

Condition of Sample:

Satisfactory

Field Prep:

None

Number & Type of Containers: 1-Can

PARAMETER	RESULT	UNITS	RDL	REF	METHOD	DA: PREP	E ANAL	ID
CLP Metals				1	1311	0605 17:30		: , .
Arsenic, TCLP	ND	mg/l	1.0	1	6010B	0606 11:20		7 MG
Barium, TCLP	3.6	mg/l	0.50	1	6010B	0606 11:20	0607 13:4	7 MG
Cadmium, TCLP	ND	mg/l	0.10	1	6010B	0606 11:20	0607 13:4	7 MG
Chromium, TCLP	0.28	mg/l	0.20	1	6010B	0606 11:20	0607 13:4	7 MG
Lead, TCLP	ND	mg/l	0.50	1	6010B	0606 11:20	0607 13:4	7 MG
Mercury, TCLP	ND	mg/l	0.0010	1	7470A	0606 15:30	0607 10:1	6 DM
Selenium, TCLP	ND .	mg/l	0.50	1	6010B	0606 11:20	0607 13:4	7 MG
	ND	mq/l	0.10		6010B	0606 11:20		_

Comments: Complete list of References and Glossary of Terms found in Addendum I

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MA:M-MA086 NH:200301-A CT:PH-0574 ME:MA086 RI:65 NY:11148 NJ:MA935 Army:USACE

Laboratory Sample Number: L0707573-08

Date Collected: 28-MAR-2007 15:05

7-2616-A FLYASH

Date Received : 24-MAY-2007

Sample Matrix:

SOLID

Date Reported: 08-JUN-2007

Condition of Sample:

Satisfactory

Field Prep:

None

Number & Type of Containers: 1-Can

PARAMETER	RESULT	UNITS	RDL	REF	METHOD	DAT	E	ID
						PREP	ANAI	•
TCLP Metals								
TCLP Extraction	FOR THE FLORES LANDS AND	ruhau nitum 6,45		1	1311	0605 17:30		
Arsenic, TCLP	ND	mg/l	1.0	1	6010B	0606 11:20	0607 13	50 MG
Barium, TCLP	3.7	mg/l	0.50	1	6010B	0606 11:20	0607 13	50 MG
Cadmium, TCLP	ND	mg/l	0.10	1	6010B	0606 11:20	0607 13	50 MG
Chromium, TCLP	0.31	mg/l	0.20	1	6010B	0606 11:20	0607 13	:50 MG
Lead, TCLP	ND	mq/l	0.50	1	6010B	0606 11:20	0607 13	:50 MG
Mercury, TCLP	ND	mq/l	0.0010	1	7470A	0606 15:30	0607 10	:17 DM
Selenium, TCLP	ND	mg/l	0.50	1	6010B	0606 11:20	0607 13	:50 MG
		-	0.10					

Comments: Complete list of References and Glossary of Terms found in Addendum ${\tt I}$

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MA:M-MA086 NH:200301-A CT:PH-0574 ME:MA086 RI:65 NY:11148 NJ:MA935 Army:USACE

Laboratory Sample Number: L0707573-09

Date Collected: 29-MAR-2007 11:35

7-2617-A FLYASH

Date Received: 24-MAY-2007

Sample Matrix:

SOLID

Date Reported: 08-JUN-2007

Condition of Sample:

Satisfactory

Field Prep:

Number & Type of Containers: 1-Can

PARAMETER	RESULT	UNITS	RDL	REF	METHOD	DA'	ΓE	ID
						PREP	ANAI	
TCLP Metals TCLP Extraction		ereer Marieta		elej.		agginas d	r "". :	
TCLP Extraction	May lagunda (H. St.)			1	1311	0605 17:30		
Arsenic, TCLP	ND	mg/l	1.0	1	6010B	0606 11:20	0607 13	:53 MG
Barium, TCLP	3.6	mg/l	0.50	1	6010B	0606 11:20	0607 13	:53 MG
Cadmium, TCLP	ND	mg/l	0.10	1	6010B	0606 11:20	0607 13	:53 MG
Chromium, TCLP	0.32	mg/l	0.20	1	6010B	0606 11:20	0607 13	:53 MG
Lead, TCLP	ND	mg/l	0.50	1	6010B	0606 11:20	0607 13	:53 MG
Mercury, TCLP	ND	mg/l	0.0010	1	7470A	0606 15:30	0607 10	:19 DM
Selenium, TCLP	ND	mg/l	0.50	1	6010B	0606 11:20	0607 13	:53 MG
Silver, TCLP	ND	mg/l	0.10	1	6010B	0606 11:20	0607 13	:53 MG

Comments: Complete list of References and Glossary of Terms found in Addendum I

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MA:M-MA086 NH:200301-A CT:PH-0574 ME:MA086 RI:65 NY:11148 NJ:MA935 Army:USACE

Laboratory Sample Number: L0707573-10

Date Collected: 30-MAR-2007 09:00

7-2618-A FLYASH Sample Matrix: SOLID

Satisfactory

Date Received: 24-MAY-2007 Date Reported: 08-JUN-2007

Condition of Sample:

Field Prep:

None

Number & Type of Containers: 1-Can

PARAMETER	RESULT	UNITS	RDL	REF	METHOD	DA!	E	ID
						PREP	ANAL	
TCLP Metals		Lui edjirijinijni,	- (148) 1883 (B. 1883) - (148) 1883 (B. 1884)					
ICLP Metals ICLP Extraction				1	1311	0605 17:30		
Arsenic, TCLP	ND	mg/l	1.0	1	6010B	0606 11:20	0607 13:	56 MG
Barium, TCLP	3.6	mg/l	0.50	1	6010B	0606 11:20	0607 13:	56 MG
Cadmium, TCLP	ND	mg/l	0.10	1	6010B	0606 11:20	0607 13:	56 MG
Chromium, TCLP	0.32	mg/l	0.20	1	6010B	0606 11:20	0607 13:	56 MG
Lead, TCLP	ND	mg/l	0.50	1	6010B	0606 11:20	0607 13:	56 MG
Mercury, TCLP	0.0015	mg/l	0.0010	1	7470A	0606 15:30	0607 10:	21 DM
	1770	mg/l	0.50	1	6010B	0606 11:20	0607 13:	56 MG
Selenium, TCLP	ND	mg/x	0.50					

Comments: Complete list of References and Glossary of Terms found in Addendum ${\tt I}$

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ALPHA ANALYTICAL LABORATORIES QUALITY ASSURANCE BATCH DUPLICATE ANALYSIS

Laboratory Job Number: L0707573

arameter	Value 1	Value 2	Units	RPD	RPD Limits
Total Met	als for sam	ole(s) 01-0)2 (L0707424	-01, WG2	82529-3)
ercury, Total	ND	ND	mg/l	NC	
TCLP Meta	ls for samp	le(s) 03-10	(L0707573-	03, WG28	2759-1)
rsenic, TCLP	ND	ИD	mg/l	NC	20
arium, TCLP	3.5	3.4	mg/l	3	20
admium, TCLP	ND	ND	mg/l	NC	20
hromium, TCLP	0.30	0.29	mg/l	3	20
ead, TCLP	ND	ND	mg/l	NC	20
elenium, TCLP	ND	ND	mg/l	NC	20
ilver, TCLP	ND	ND	mg/l	NC	20

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ALPHA ANALYTICAL LABORATORIES QUALITY ASSURANCE BATCH SPIKE ANALYSES

Laboratory Job Number: L0707573

Parameter	% Recovery	QC Criteria
Total Metals LCS f	or sample(s) 01-02 (WG282529+1)
Mercury, Total	100	
TCLP Metals LCS f	or sample(s) 03-10 (wG282759-4)
Arsenic, TCLP	110	75-125
Barium, TCLP	99	75–125
Cadmium, TCLP	110	75-125
Chromium, TCLP	100	75-125
Lead, TCLP	100	75-125
Selenium, TCLP	105	75-125
Silver, TCLP	100	75-125
Mercury, TCLP Total Metals SPIKE for sa Mercury, Total	104 mple(s) 01-02 (L0707 131	424-01, WG282529-2)
TCLP Metals SPIKE for sam	ple(s) 03-10 (L07075	73-03, WG282759-2)
Arsenic, TCLP	100	75-125
Barium, TCLP	96	75-125
Cadmium, TCLP	100	75-125
Chromium, TCLP	97	75-125
Lead, TCLP	100	75-125
Selenium, TCLP	100	75-125
Silver, TCLP	5	75–125
TCLP Metals SPIKE for sam	mle(s) 03-10 (1.07078	89-01. WG282796-2)
Mercury, TCLP	127	,

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ALPHA ANALYTICAL LABORATORIES QUALITY ASSURANCE BATCH BLANK ANALYSIS

Laboratory Job Number: L0707573

PARAMETER	RESULT	UNITS	RDL	REF	METHOD	DAT	
						PREP	ANAL
Blank	Analysis for sam	ple(s) 01-	02 (WG282	529-4			
Cotal Metals							
Mercury, Total	ND	mg/l	0.0002	3	245.1	0604 15:05	0605 13:58 DM
Blank	Analysis for sam	ple(s) 03-	10 (WG282	759-3	3) <u> </u>		
CCLP Metals							
CLP Extraction	The state of the s	Maria de la constança de la calenda de l		1	1311	0605 17:30	100000000000000000000000000000000000000
Arsenic, TCLP	ND	mg/l	1.0	1	6010B	0606 11:20	0607 13:13 MG
Barium, TCLP	ND	mg/l	0.50	1	6010B	0606 11:20	0607 13:13 MG
Cadmium, TCLP	ND	mg/l	0.10	1	6010B	0606 11:20	0607 13:13 MG
Chromium, TCLP	ND	mg/l	0.20	1	6010B	0606 11:20	0607 13:13 MG
Lead, TCLP	ND	mg/l	0.50	1	6010B	0606 11:20	0607 13:13 MG
Selenium, TCLP	ND	mg/l	0.50	1	6010B	0606 11:20	0607 13:13 MG
Silver, TCLP	ND	mg/l	0.10	1	6010B	0606 11:20	0607 13:13 MG
Blank	Analysis for sam	ple(s) 03-	10 (WG282	796-4	4-):		<u>Eschant</u>
CLP Metals		BEHERRANA					
CLP Extraction				1	1311	0605 17:30	
Mercury, TCLP	ND	mg/l	0.0010	,	7470A	0606 15.30	0607 10:01 DM

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ALPHA ANALYTICAL LABORATORIES ADDENDUM I

REFERENCES

- Test Methods for Evaluating Solid Waste: Physical/Chemical Methods. EPA SW-846. Third Edition. Updates I - IIIA, 1997.
- 3. Methods for the Determination of Metals in Environmental Samples, Supplement I. EPA/600/R-94/111. May 1994.

GLOSSARY OF TERMS AND SYMBOLS

REF Reference number in which test method may be found.

METHOD Method number by which analysis was performed.

ID Initials of the analyst.

ND Not detected in comparison to the reported detection limit.

NI Not Ignitable.

ug/cart Micrograms per Cartridge.

LIMITATION OF LIABILITIES

Alpha Analytical, Inc. performs services with reasonable care and diligence normal to the analytical testing laboratory industry. In the event of an error, the sole and exclusive responsibility of Alpha Analytical, Inc., shall be to re-perform the work at it's own expense. In no event shall Alpha Analytical, Inc. be held liable for any incidental consequential or special damages, including but not limited to, damages in any way connected with the use of, interpretation of, information or analysis provided by Alpha Analytical, Inc.

We strongly urge our clients to comply with EPA protocol regarding sample volume, preservation, cooling, containers, sampling procedures, holding times and splitting of samples in the field.

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AT-THA		Project Information	nation	PAGE 1 OF 1		Repo	Report Inform) (tion Data Delive	Report Information Data Deliverables		Billing Information	Billing Information
Eight Walk Up Drive Westborough, MA 0	Eight Walk Up Drive Westborough, MA 01581	Project Name: USDOE Mercury Field Demo	SDOE Merci	ury Field Dem	_	Z	~ ŭ		IXI EMAIL. ☐ Add'i Deliverables	eliverables	8 3	Same as Cilent Into	2
Offent Information		Project Location: LCRA - Fayette Unit 3	CCRA - Fay	ette Unit 3		Redii	atory F	ecuire	nents/Re	Regulatory Bequirements/Report Limits	ts.		
Client: ALSTOM POWER, INC.	VER, INC	Project #:				State/F	State/Fed Program	8			Criteria		
Address: 2000 Day Hill Road	III Road	Project Manager: Kurt W. Johnson	: Kurt W. Joh	usou		3	0		4		1000	OT THE OWN CITY	T DE ANSWEDEN
Windsor, CT 06095		ALPHA Quote #:					PRESU	M P No		Am MCP Analytical Methods Required?	SE GUE	S HOUNS MUS	MICH PRESUMPTIVE CERTAINTY-THESE COESTIONS MUST BE ANSWERED TO A MANAGE Analytical Methods Benitred?
Phone: (860) 285-5747	47	Turn-Around Time	Time			, Yes		2	Are	Drinking Wa	ter Sample	Are Drinking Water Samples Submitted?	
Fax: (860) 285-4033		Standard Standard	□ Rus	☐ Rush (ONLY IF PRE-APPROVED)	(PPROVED)	X Yes	⊠ Yes	£ □	Hav	e you met m	inimum fiek	Have you met minimum field QC requirements?	
nail: kurt.w.johnso	Email: kurt.w.johnson@power.akstom.co	,					2			_			SAMPLE HANDLING
These samples have bu	These semples have been Prowously analyzed by Alpha	Due Date: 6 Bl OTime:	S(O)Time:										Filtration Done
ther Project Speril to my AMEX card	e se	nts/Detection Limit 08/09	isi										paged do
) Combustion parti	(1) Combustion particulates (fly ash) from WY coal						action						Lab to do
						6 ₁							pelow
ALPHA Lab ID	Sample ID	Special Control	Collection	Sample Matrix	Sampler's Initials	H lato]	AROF						Sample Specific
(Late Cae Cyry)						-	-		-	_	1	-	Comments
2130	7-2876-M FGD Makeup Wate	3/12/07	8:57	Liquid	ZW2	×	+		+	$\frac{1}{2}$			
-02	7-2677-M FGD LS Slurry	3/12/07	8238	Liquid	KWJ	×	1		+				
-	7-2586-A Flyash	3/13/07		Solid(1)	KWJ		×						
1	7-2587-A Flvash 3B1	3/23/07	17:15	Solid(1)	KWJ		×						
 	7-2588-A Flyash 3B2	3/23/07	17:15	Solid (1)	K₩J		×						
90-	7-2589-A Flyash 3B1	2/30/07	16:15	Solid(1)	KW		×			1			
P	7-2590-A Flyash 3B2	3/20/07	16:15	Solid(1)	KWJ		×						
þ	7-2616-A Flyash	3/28/07	15:05	Solid(1)	KWJ		×						
8	7-2617-A Flvash	3/29/07	11:35	Solid(1)	KW		×						
7	7-2618-A Flyash	3/30/07	9:00	Solid(1)	KWJ		×						
IFSTIONS ABOVE	VERED FOR	PRESUMPTIVE CERTAINTY	RTAINTY	CO	Container Type	۵			,	•	·		Agent stants
					Preservative			-					and completely. Samples
	IS YOUR		Relin	Relinquished By:		Dak	Date/Time		Rece	Received By:		Date/Time	tumanound time clock will a
	PROJECT	ਨ ਨ	1787	180	}	162/2	<i>55.11 co</i> [52]	I	Fleer			2/24/07/00	resolved. All samples submitted are subject to
FORM NO. DI-OT Informal	MCP.			,				4	,				April Paymen terms