

# National Energy Technology Laboratory



## Evaluating the Relationship between Slabbing of $\text{Cr}_2\text{O}_3/\text{MgO}$ Refractories Used in Steelmaking and Spalling of High Chrome Oxide Refractories Used in Gasification

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Office of Fossil Energy



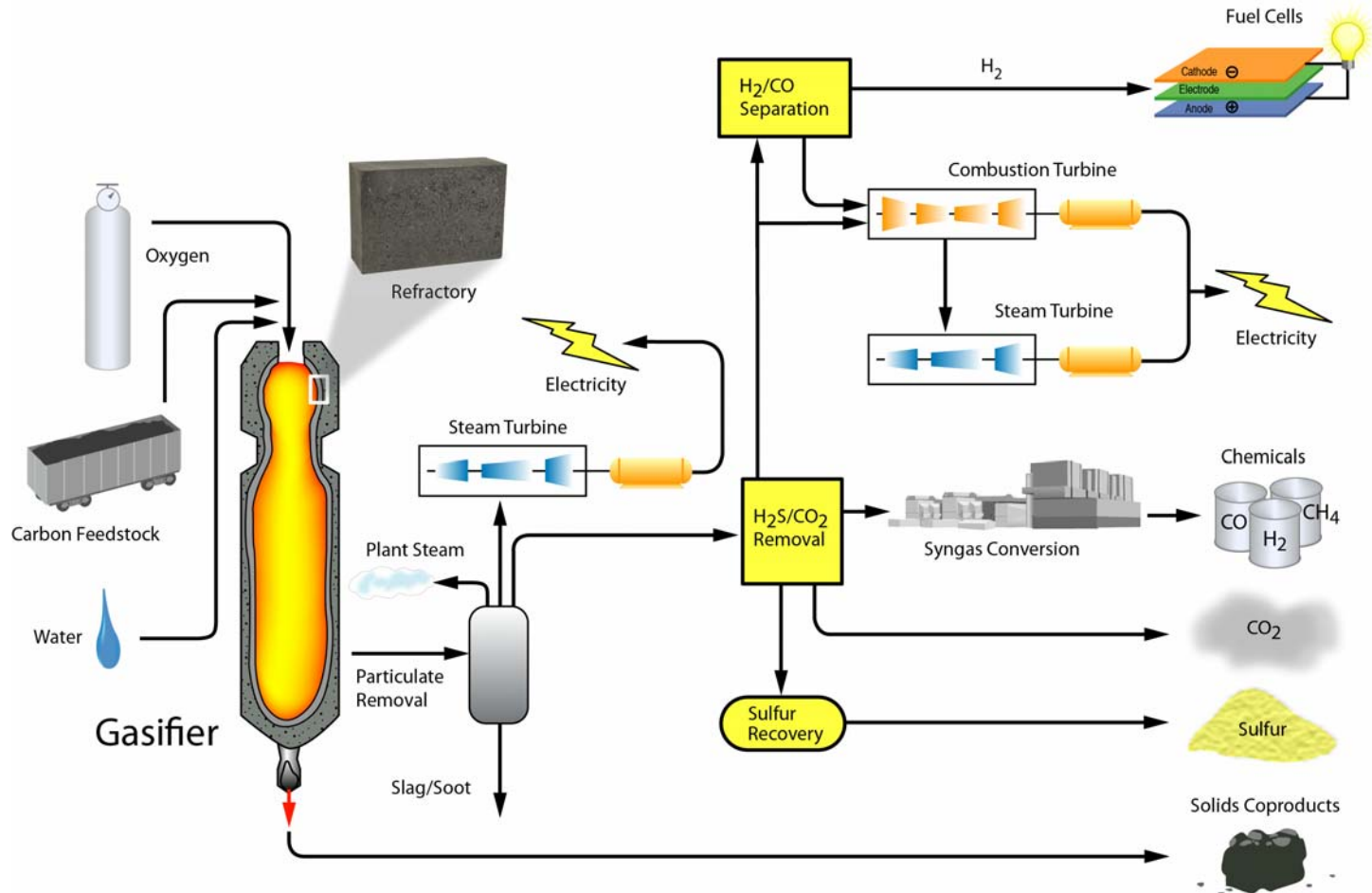
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# Outline

- **Acknowledgement**  
USDOE - NETL  
*Advanced Gasification Technologies*  
*Advanced Metallurgical Processes*
- **Background**
- **Issues/Consequences**
- **Post-Mortem Analysis**
- **Summary**



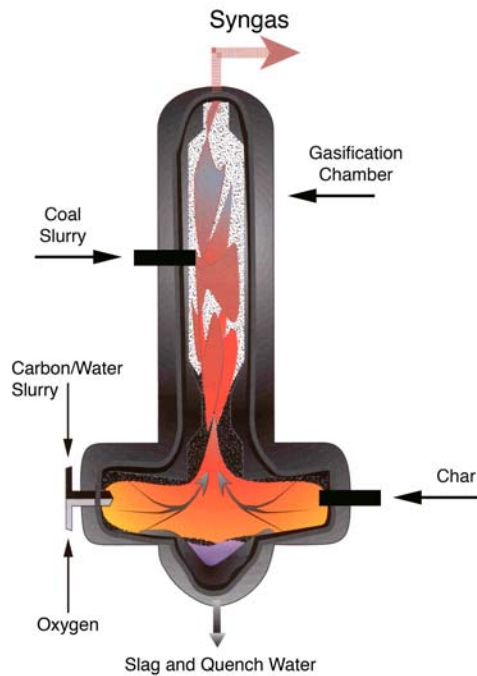
# Gasification Background



## Gasification Reaction

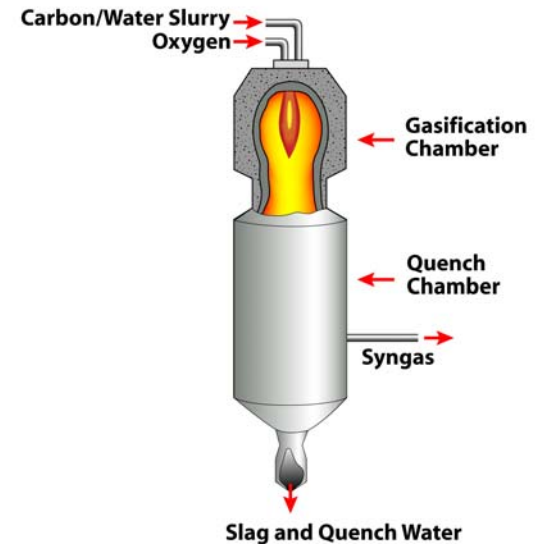


# Material Challenges Inherent to Air-cooled Slagging Gasifiers



***ConocoPhillips (2 stage - syngas cooler)***

- Operating temperatures of 1325° to 1575°C.
- Thermal cycling.
- Alternating reducing and oxidizing environment.
- Corrosive slags of variable chemistry.
- Corrosive gases.
- Pressures  $\geq 400$  psi.



***GE Design (1 stage - syngas cooler or water quench)***

# Refractory Material Issues - Consequences

- 1. Low system reliability, on-line availability**
  - gasifier down as frequently as once/month
  - possible need for “spare” gasifier
- 2. Lost opportunity costs**
- 3. Frequent maintenance; 3-24 months**
- 4. Need for zoning – larger “spare” material inventory**
- 5. High material repair costs**
  - \$1 million for refractory lining
- 6. Excessive safety margins**



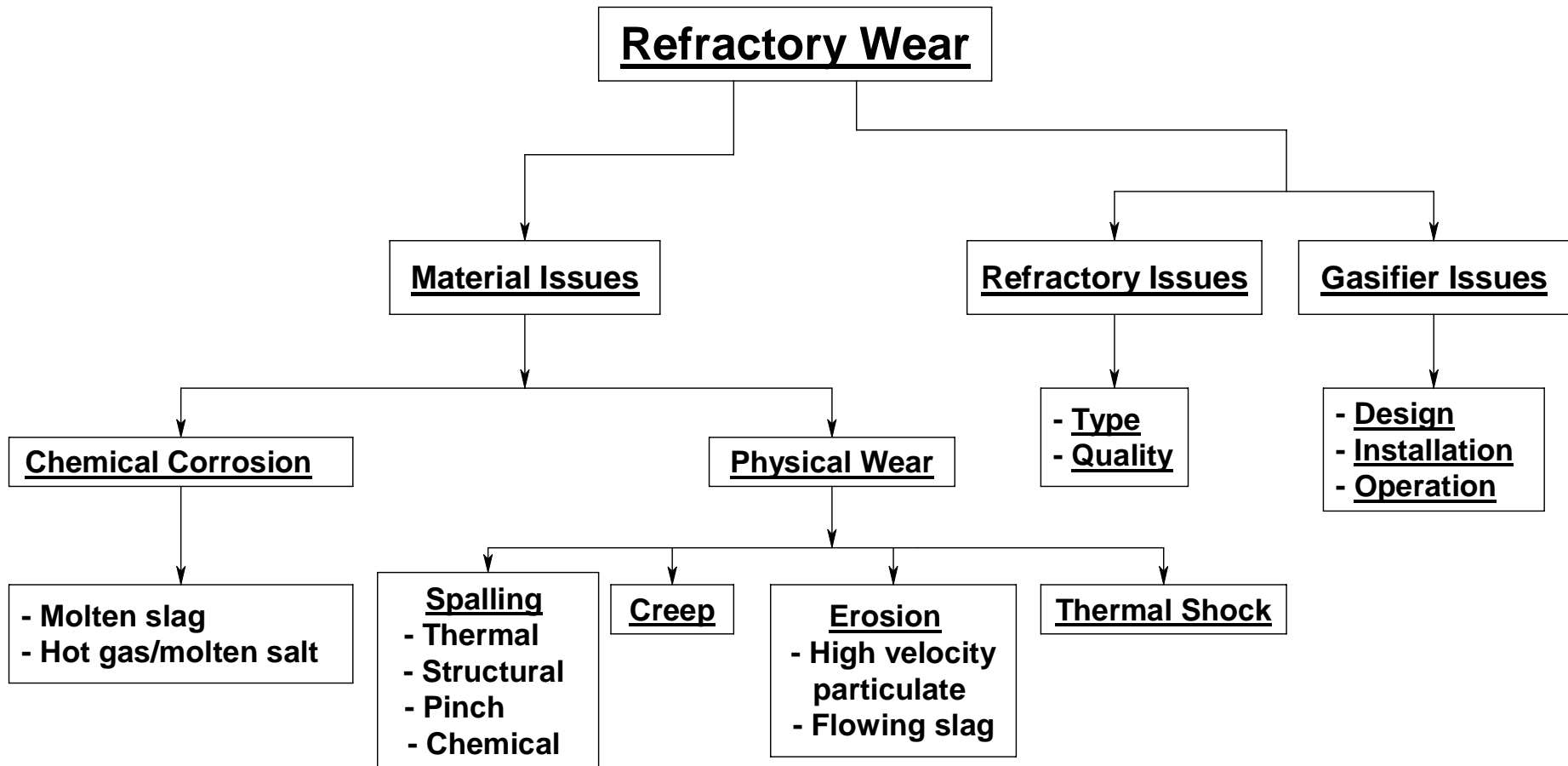
# Research Goal

**Enhance gasifier reliability and economics through the development of improved refractory materials with longer service life**

Two major types of wear in gasifiers: *corrosion and spalling*

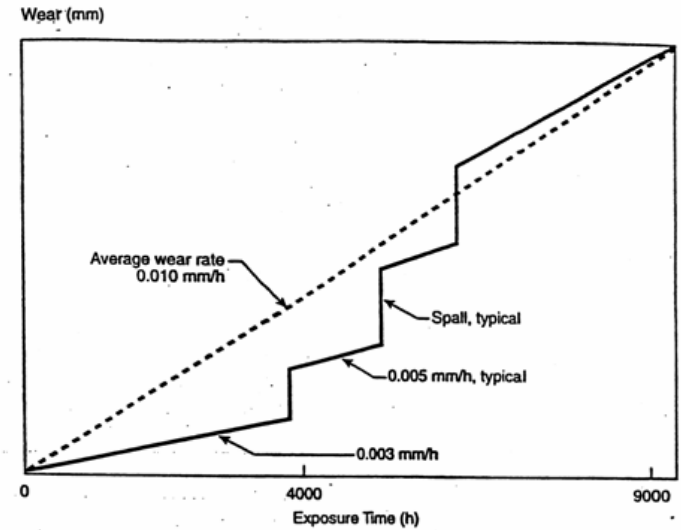
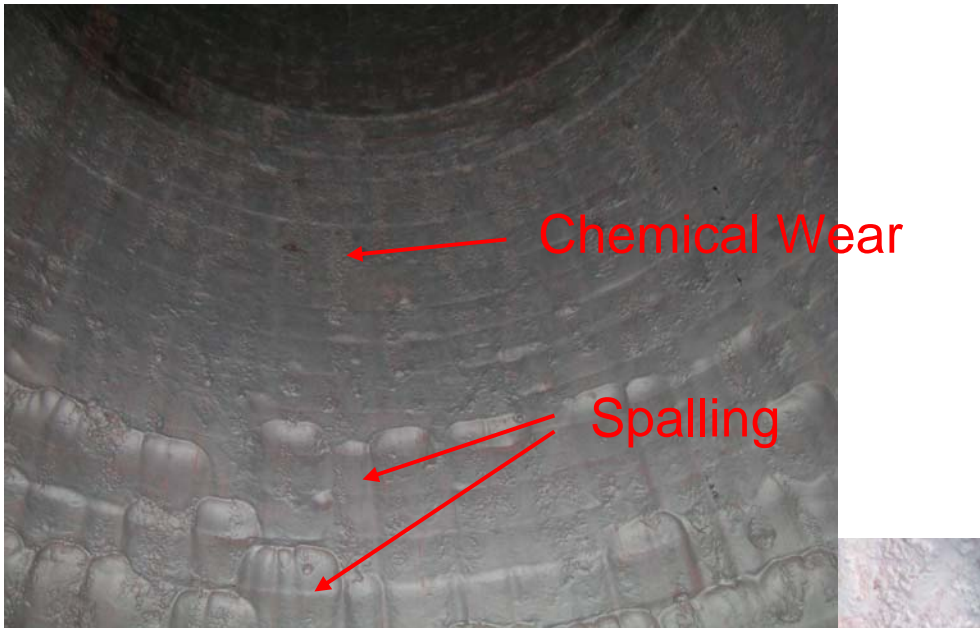
- Research directed towards reducing/eliminating spalling wear of gasifier refractories
- Spalling wear is similar to known wear mechanisms in chromia/magnesia – magnesia/chromia steelmaking refractories (peeling, slabbing, chemical spalling)

# Causes of Refractory Wear in Gasifiers (*High Cr<sub>2</sub>O<sub>3</sub> Materials*)





# Main Wear - Spalling and Chemical Corrosion



Bakker W. T. (1998) "Materials Guidelines for Gasification Plants" EPRI Report TR-110507, Palo Alto, CA.





# Corrosion and Spalling

**Corrosion**

**Spalling**

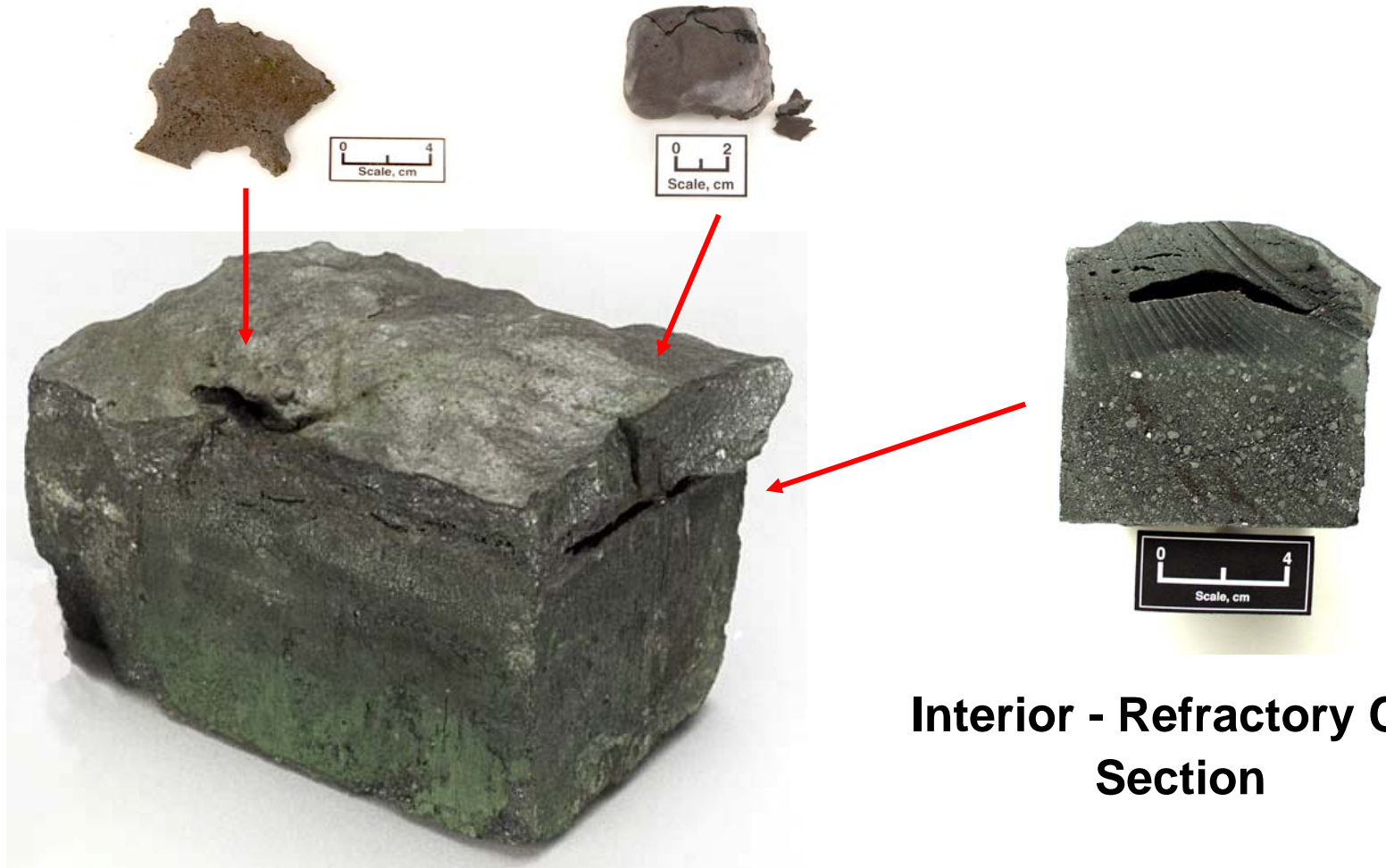


# Corrosion and Spalling





# Spalling – Refractory Surface and Interior

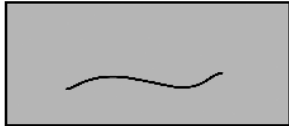
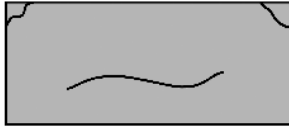






**Interior - Refractory Cross Section**

# Corrosion and Spalling Mechanism



## Stages of Refractory Wear in the Hot Zone of a Slagging Gasifier (Surface Corrosion, Spalling)

Stage	Sample	Description
1		<b>New</b> <ul style="list-style-type: none"> <li>Refractory may contain internal cracks from pressing, firing.</li> </ul>
2		<b>Preheat</b> <ul style="list-style-type: none"> <li>Pinch spalling due to hoop stresses</li> </ul>
3		<b>Infiltration, Corrosion</b> <ul style="list-style-type: none"> <li>Molten slag infiltration on hot face, cracks and pores.</li> <li>Surface corrosion due to slag begins</li> </ul>
4		<b>Horizontal Crack Formation</b> due to: <ul style="list-style-type: none"> <li>Thermal cycling</li> <li>Stress accumulation</li> <li>Creep</li> </ul>
5		<b>Void Formation</b> <ul style="list-style-type: none"> <li>Cracks join</li> <li>Internal void formation</li> <li>Spalling (peeling) begins</li> <li>Creep occurs on slag penetrated hot face</li> <li>Hot face corrosion continues</li> </ul>
6		<b>Renewed Cycle</b> <ul style="list-style-type: none"> <li>Material breakoff on hot face</li> <li>Steps 3-5 repeat</li> </ul>

# Impact of Spalling on Refractory Wear



# Chemical Composition\* of High Cr<sub>2</sub>O<sub>3</sub> Gasifier and Steel Making Refractories

<u>Element</u>	<u>Composition</u>			Chromia-Magnesite Steelmaking Brick
	<u>A</u>	<u>B</u>	<u>C</u>	
Cr <sub>2</sub> O <sub>3</sub>	90.1	87	24.8	
Al <sub>2</sub> O <sub>3</sub>	9.3	3.0	22.3	
ZrO <sub>2</sub>	---	6.5	---	
MgO	---	---	36.4	
Fe <sub>2</sub> O <sub>3</sub> /SiO <sub>2</sub>	---	---	3.3/4.5	

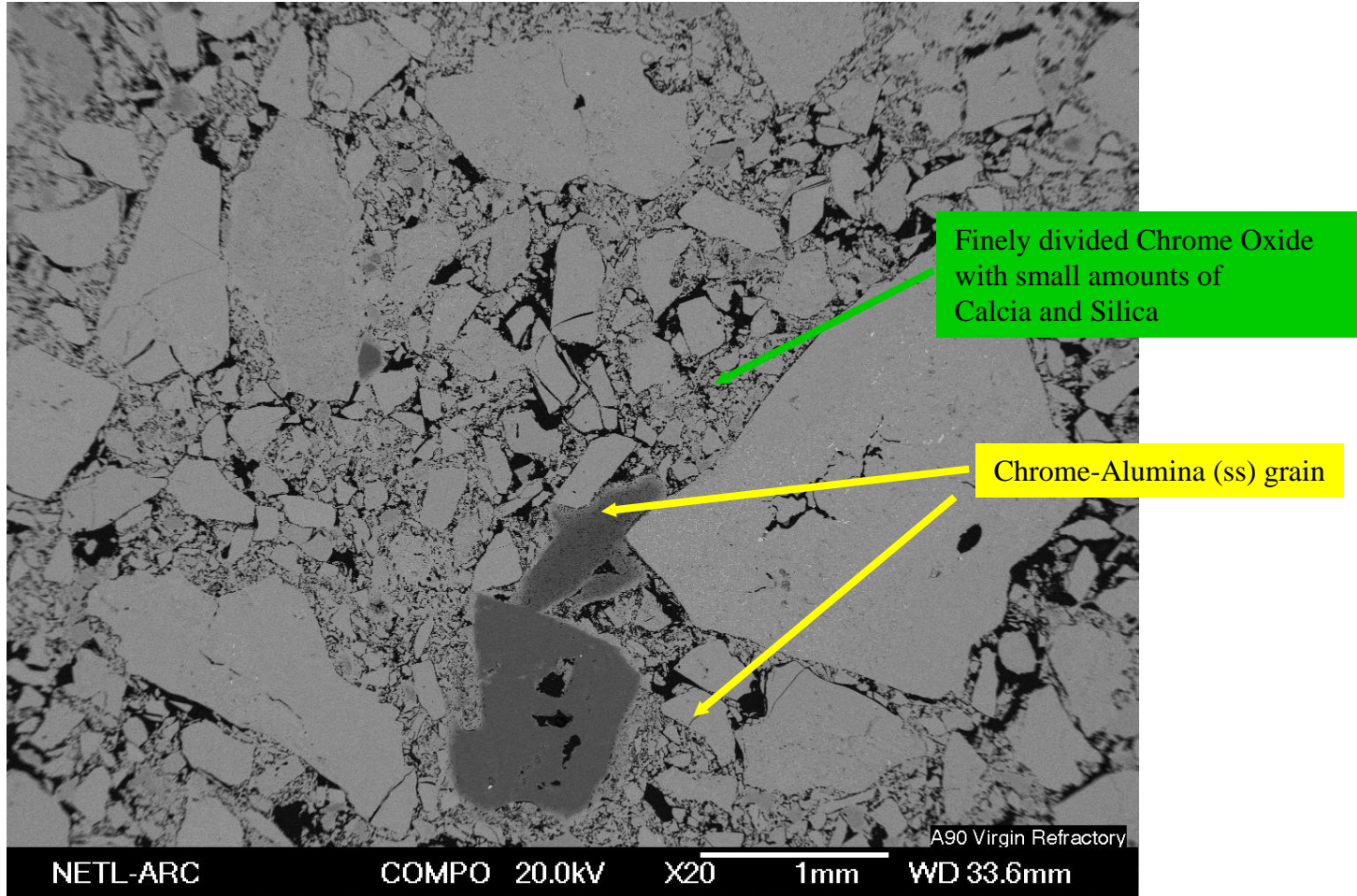
Chrome Gasifier Brick



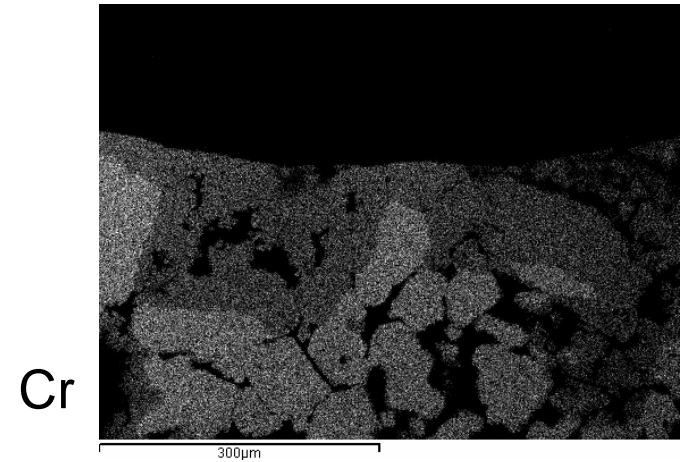
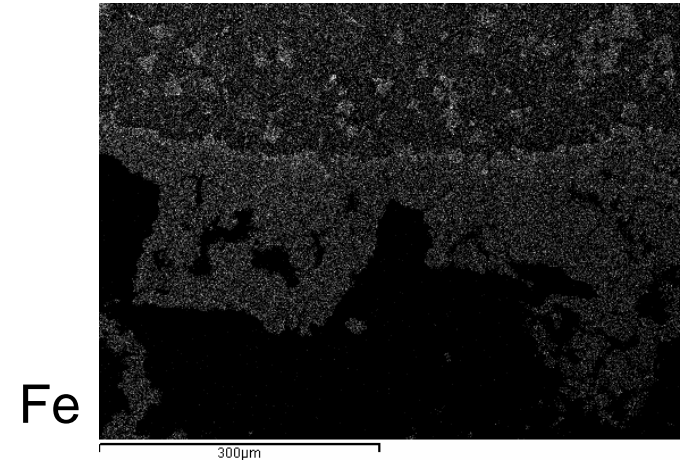
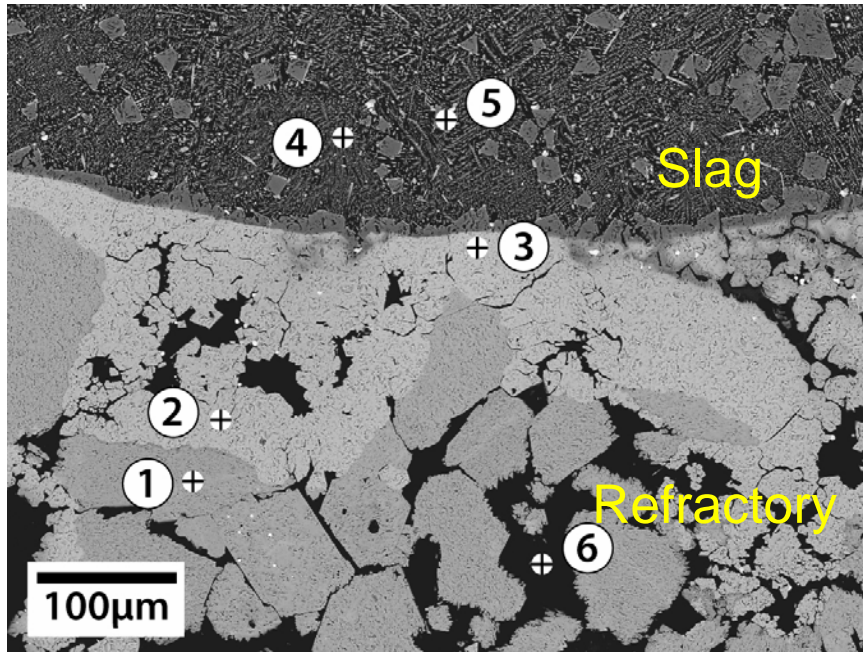
# Causes of Spalling

1. **Density/porosity differences between slag infiltrated/non-infiltrated layers**
2. **Crystalline phase differences between slag infiltrated/non-infiltrated layers**
3. **Bursting expansion observed in chromia/magnesia steelmaking refractories (*reoxidation of FeO*)**
4. **Reaction of alkali (Na and K) with chromia/alumina grain, resulting in the formation of disruptive sodium aluminate phases (vapor interaction)**

# High Chrome Oxide Refractory Microstructure



# High Chrome Oxide Gasifier Refractory Infiltrated with Slag

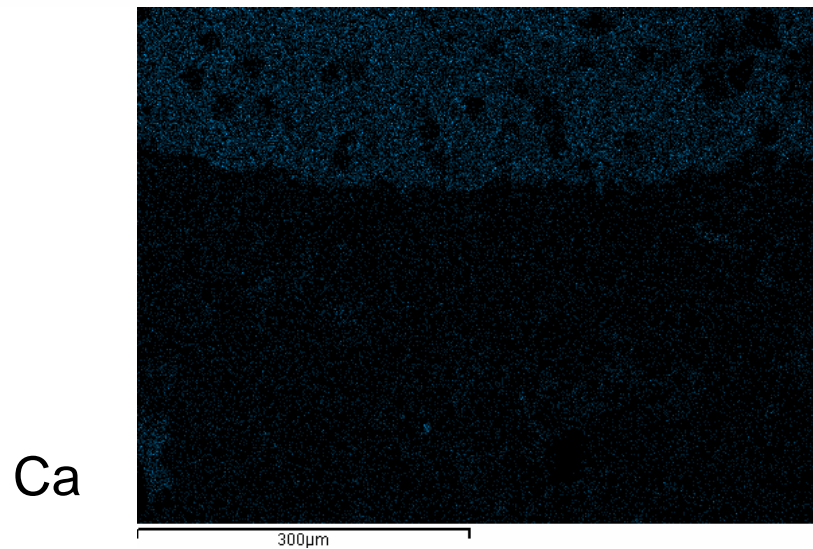
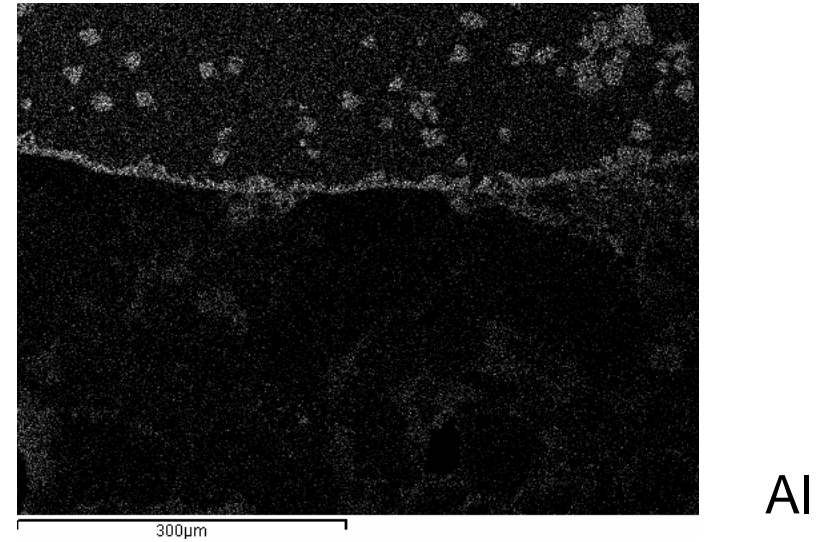
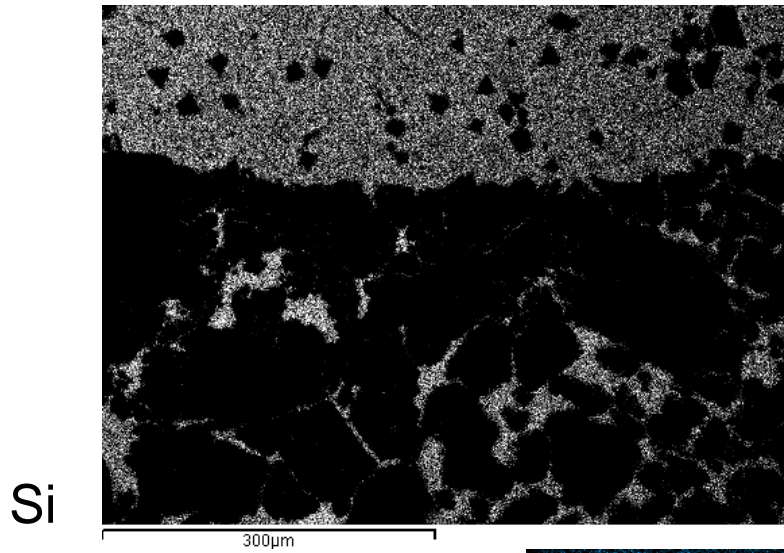


Slag diffusion, corrosion



# Refractory Surface Microstructure – Slag Attack

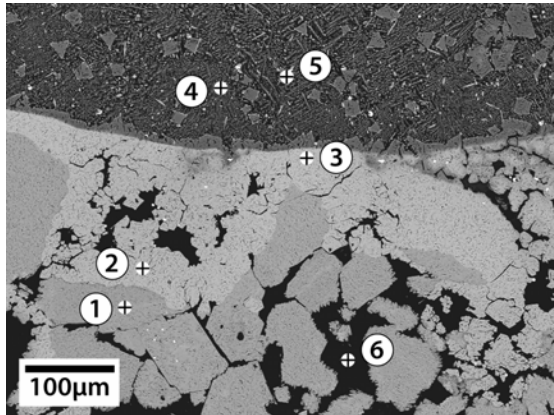
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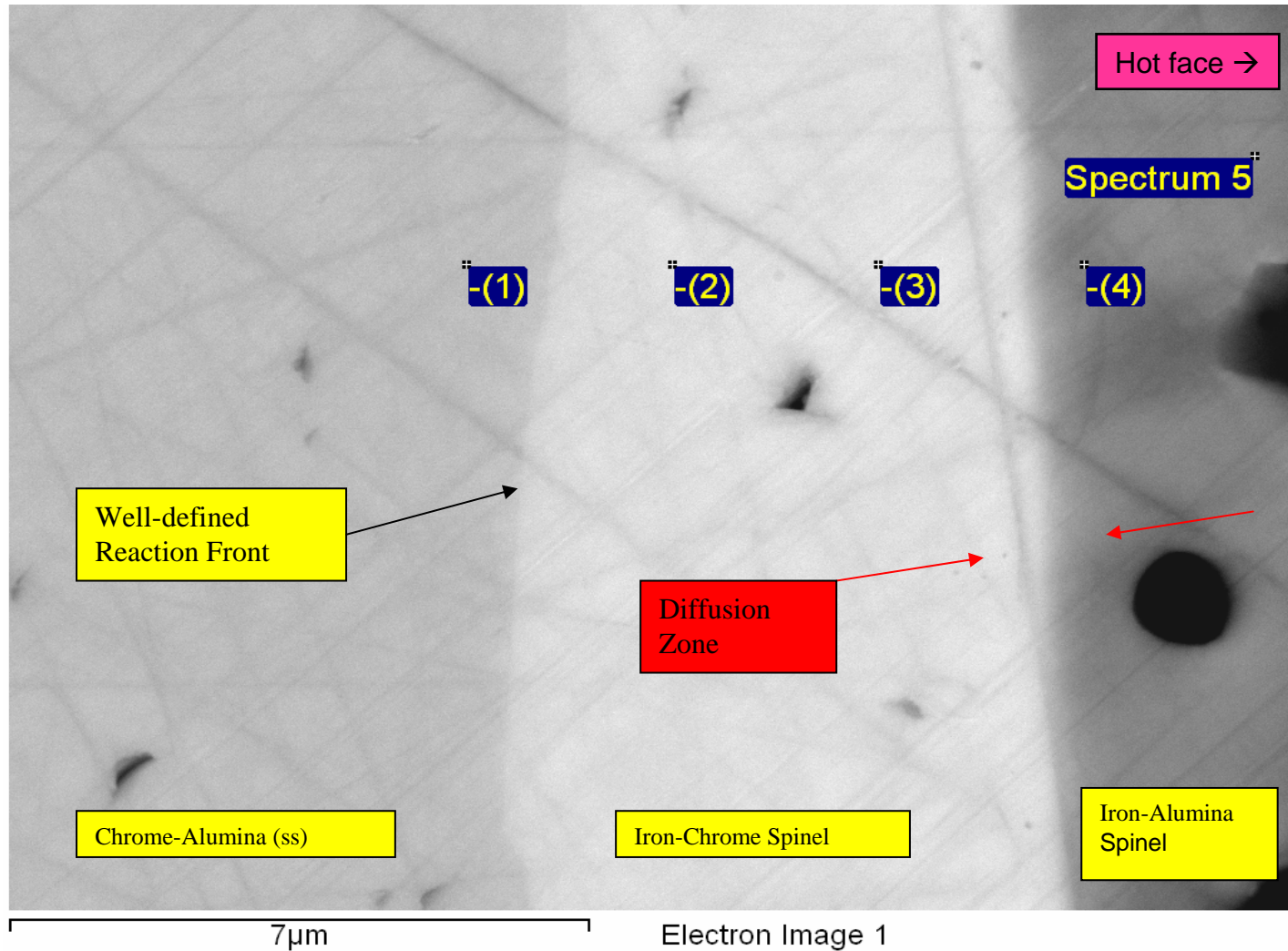
# Elemental Analysis

## *(Refractory/Slag Interface)*

Point	Element (wt pct)				
	Cr	Al	Fe	Si	Ca
1 – int. ref. grain	63.8	5.7	-	-	-
2 – surface ref. grain ( $\text{FeCr}_2\text{O}_4$ )	37.8	5.6	25.2	-	-
3 – inter. ref. grain ( $\text{FeCr}_2\text{O}_4$ )	39.9	3.5	24.0	-	-
4 – slag	-	7.2	21.1	23.2	3.5
5 – crystallized slag ( $\text{FeAl}_2\text{O}_4$ )	-	27.0	31.1	-	-
6 – interior slag	1.3	8.0	1.4	32.7	1.6

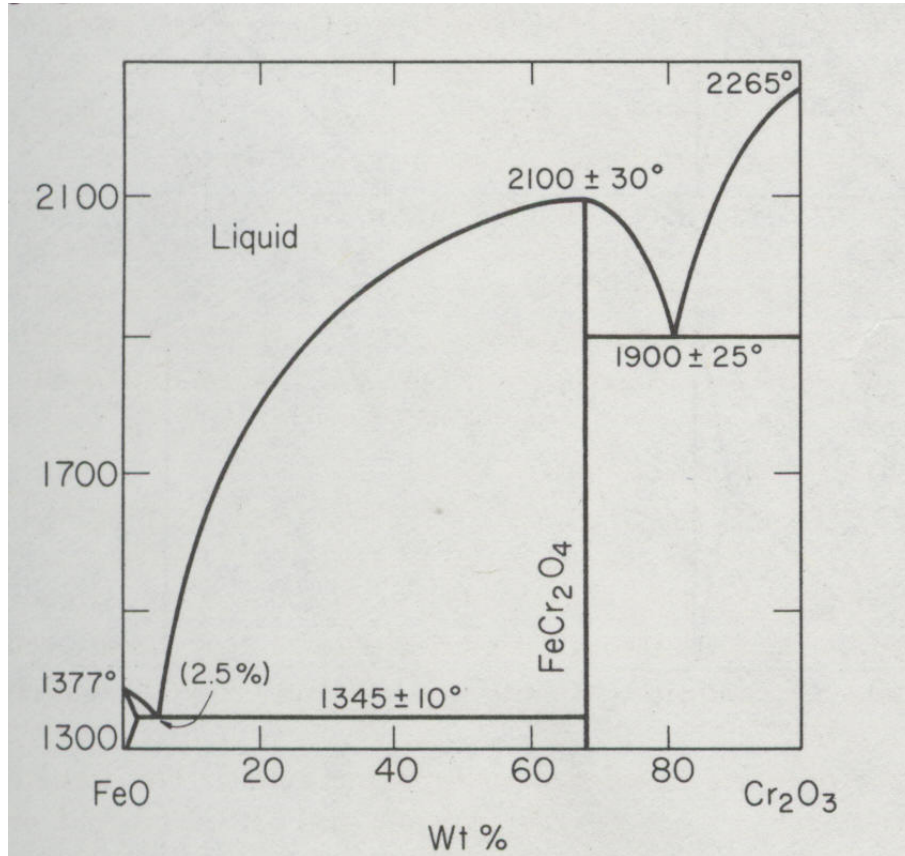


# Iron – Chrome – Alumina Interactions

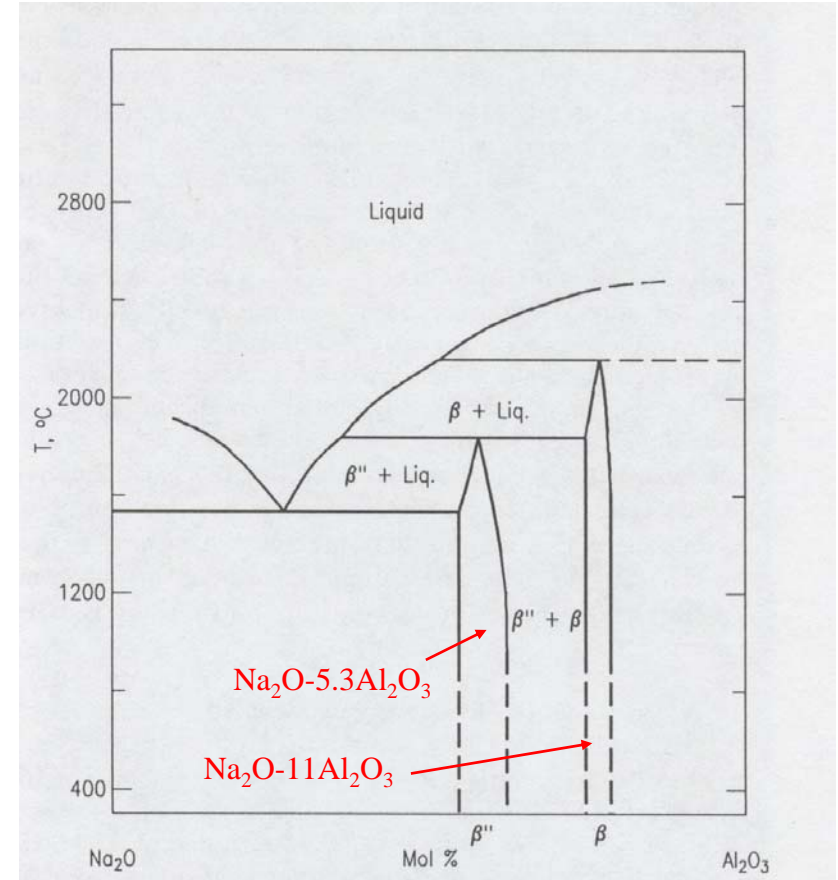




# Known Phase Interactions



FeO - Cr<sub>2</sub>O<sub>3</sub> Phase Diagram (from Phase Diagram for Ceramics 1975 Supplement, # 4187)

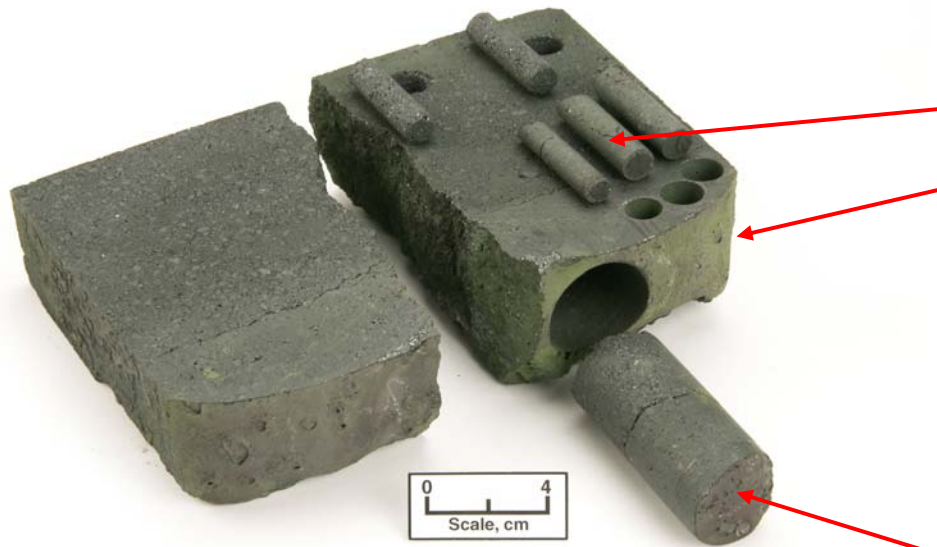


Na<sub>2</sub>O - Al<sub>2</sub>O<sub>3</sub> Phase Diagram (from Phase Equilibria Diagrams Volume XII, # 9884)

# Post-Mortem Analysis

## Thermal Expansion Cores

Cold face:  $7.68 \times 10^{-6}$  mm/mm/°C  
Hot face:  $8.18 \times 10^{-6}$  mm/mm/°C



## Chemical,

X-Ray Crystalline Phases,

Microstructure of Layers

Distance from Hot Face (mm)	Bulk Chemistry (wt pct)					X-Ray Crystalline Phases
	Cr <sub>2</sub> O <sub>3</sub>	Al <sub>2</sub> O <sub>3</sub>	SiO <sub>2</sub>	CaO	Fe	
H.F. to 2.3	80.0	10.8	5.4	0.3	1.6	P - Cr <sub>2</sub> O <sub>3</sub> Tr- FeCr <sub>2</sub> O <sub>4</sub>
6.9	84.2	10.2	3.9	0.3	0.4	P - Cr <sub>2</sub> O <sub>3</sub> Tr- FeCr <sub>2</sub> O <sub>4</sub>
11.4	83.9	10.7	3.2	0.4	0.4	P - Cr <sub>2</sub> O <sub>3</sub> Tr- FeCr <sub>2</sub> O <sub>4</sub>
16.0	87.7	10.3	3.5	0.5	0.3	P - Cr <sub>2</sub> O <sub>3</sub>
20.6	83.9	10.3	3.0	0.5	0.3	Same
25.1	83.5	10.7	3.0	0.6	0.3	Same
29.7	84.3	10.4	2.7	0.6	0.3	Same
34.3	83.5	10.4	2.8	0.6	0.4	Same
38.9	82.7	10.0	1.6	0.6	0.3	Same
43.3	83.9	9.3	2.3	0.5	0.2	Same
48.0	85.4	9.6	0.6	0.1	0.2	Same
52.7	85.7	10.5	0.9	0.2	0.2	Same
57.2	86.1	10.5	0.2	0.0	0.2	Same
61.7	86.1	10.6	0.2	0.0	0.2	Same
127	87.4	9.4	0.2	0.2	0.2	Same

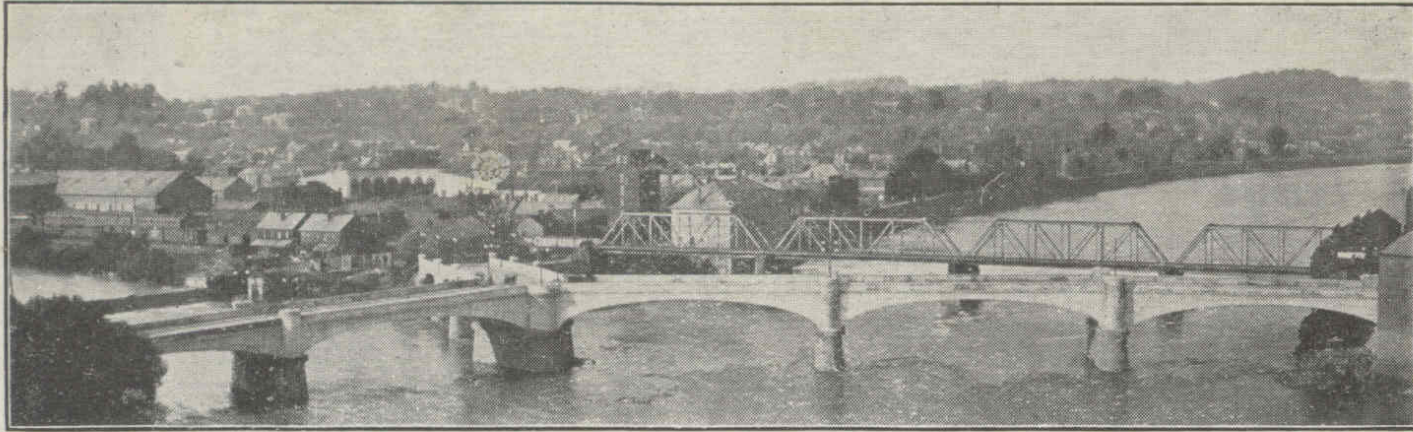
## Core Analysis (Chemical and Crystalline Phases)

# Summary

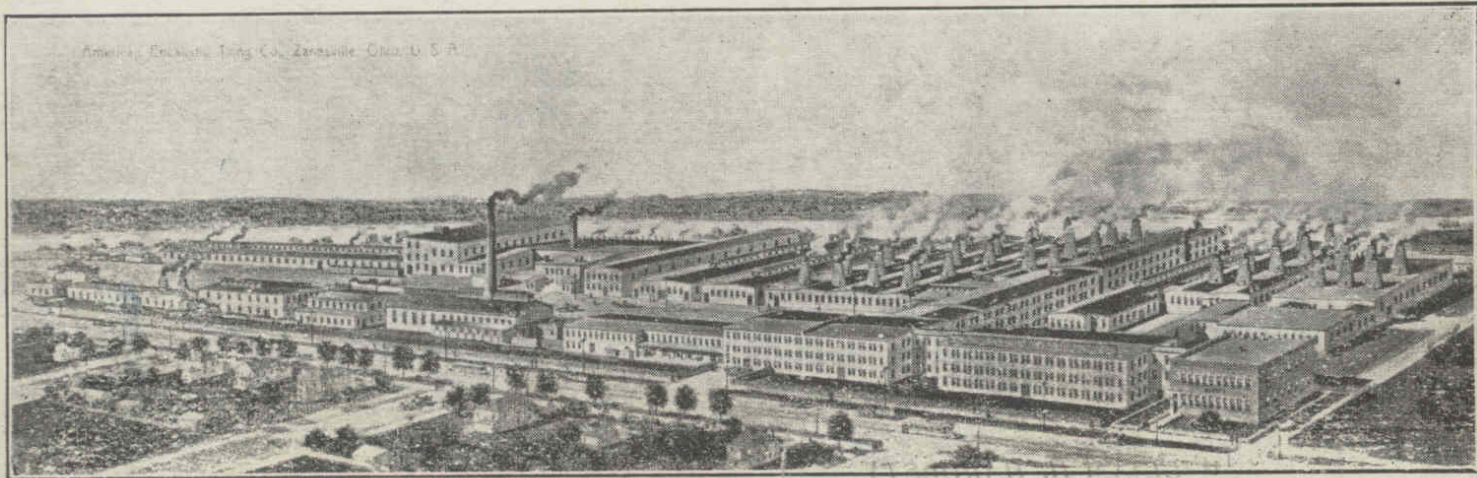
- Slabbing in chromia/magnesia refractories used in steelmaking and spalling in gasifier refractories are similar wear mechanisms
- Issues causing spalling/slabbing are complicated, but include:
  - Reaction between  $\text{Cr}_2\text{O}_3$  and  $\text{FeO}$  to form  $\text{FeCr}_2\text{O}_4$ , increase in  $\text{Al}_2\text{O}_3$  in  $\text{Cr}_2\text{O}_3$ , some structure swelling on surface
  - Microstructure differences between slag infiltrated/non-infiltrated areas
  - Cyclic operation of the gasifier

# American Encaustic Tiling Co., 1920 – Zanesville, OH

*(1<sup>st</sup> Tunnel Kiln in Used in Tile Industry, Largest Tile Producer)*



ZANESVILLE, OHIO.—Only Concrete Y Bridge and Largest Tile Works in the World.





# Mosaic Tile Co., 1906 – Zanesville, OH

*(Also Once the Largest Tile Company in the World)*

