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## Metallic Materials Development for Solid Oxide Fuel Cells

J. Dunning, J. Hawk, D. Alman, P. Jablonski, G. Holcomb, M. Ziomek-Moroz, S. Cramer, A. Petty and R. Walters



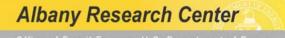
Albany Research Center Albany, OR, 97321 www.alrc.doe.gov

SECA Core Technologies Peer Review Workshop Tampa Bay, Florida, January, 27-28, 2005



## Outline

- 1. Low CTE Nickel Base Alloys (J.Dunning)
  - Composition
  - Production of Strip
- 2. Modifications for Improved Oxidation Resistance (J. Dunning)
  - Nickel-Base and Ferritic Alloys
- 3. Balance of Plant (J. Hawk)





## Low CTE Nickel Alloy Design Concepts

### Oxidation Resistance and Low CTE

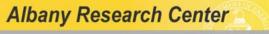
#### Oxidation Resistance: Chromia former required

Cr-Mn Spinel is conductive and minimizes Chrome evaporation

#### CTE vs. Oxidation Resistance: A balancing act

Chrome raises CTE while Mo and W lower CTE Al, Ti and C also lower CTE Fe and Co raise CTE

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## **Alloy Design Concepts**

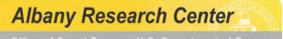
Formulation for CTE
 CTE=13.87 + 7.28x10<sup>-2</sup>[Cr] - 7.96x10<sup>-2</sup> [W]

 - 8.23x10<sup>-2</sup>[Mo] - 1.83x10<sup>-2</sup>[Al]
 - 1.63x10<sup>-1</sup> [Ti]

R. Yamamoto et. al., in Materials for Adavanced Power Engineering – 2002, Proc. 7<sup>th</sup> Leige Conf. Sept 30-Oct 3, 2003, <u>Energy and Technology Vol. 21</u>.

- ThermoCalc software used to verify phases.
- Melted 28 different compositions

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## **J-Series Ni-Cr-Mo Alloys**

Nominal Composition (wt%)

			Mo				
<b>J1</b>	Bal	12	18	1.1	0.9	0	0
<b>J2</b>	Bal	10	22.5	3	0.1	0.5	0.1
J3	Bal	12.5	22.5	3	0.1	0.5	0.1
J4	Bal	15	22.5	3	0.1	0.5	0.1
J5	Bal	12.5	18 22.5 22.5 22.5 22.5	1	0.1	0.5	0.1
J6	Bal	12.5	27.7	0	0	0.5	0.1
<b>J7</b>	Bal	22	36.1	0	0	0.5	0.1

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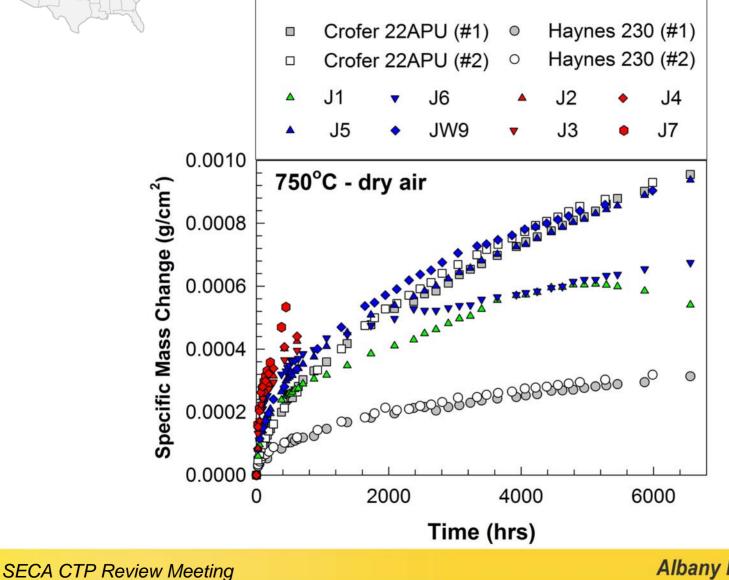
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## **CTE-J series alloys**

			•	
Alloy	Predicted (23-700°C)	<b>Measured</b> (23-700°C)	Measured (23-800°C)	Measured (23-900°C)
J1	13.06	12.9	13.6	14.4
J2	12.25	12.5	13.2	14.0
J3	12.44	12.3	13.4	14.3
J4	12.61	12.7	13.6	14.4
<b>J5</b>	12.71	12.6	13.4	14.0
J6	12.50	13.8	14.6	15.7
J7	12.50	11.2	11.9	12.5
Crofer		11.0	11.9	12.6
Haynes 230	14.2	13.3	14.3	15.4
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## 750°C Oxidation



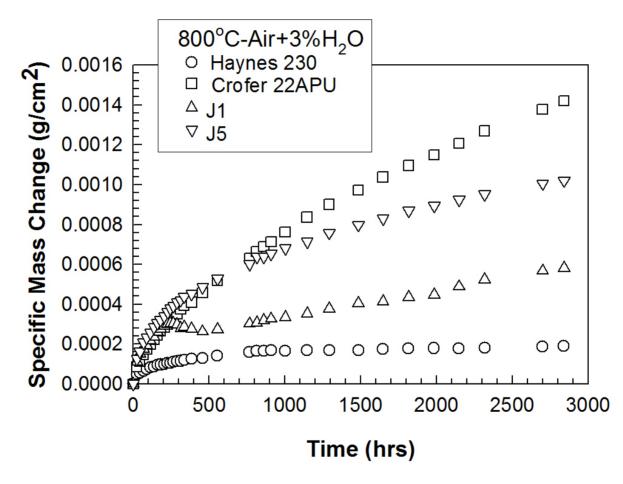
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## 800°C Oxidation

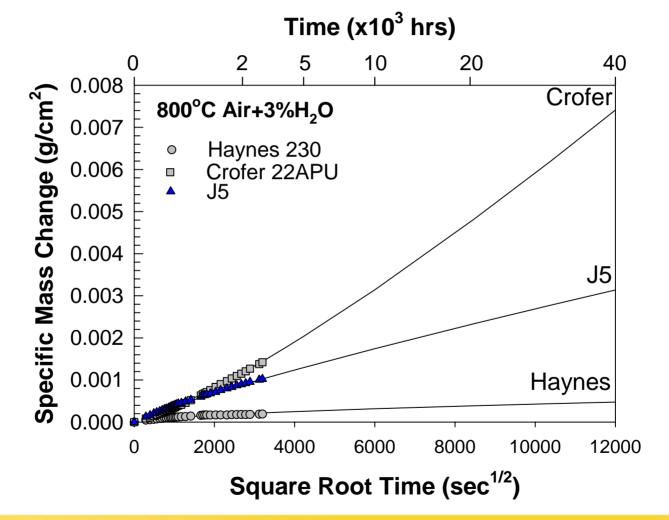


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## 40,000 hr Extrapolated Behavior



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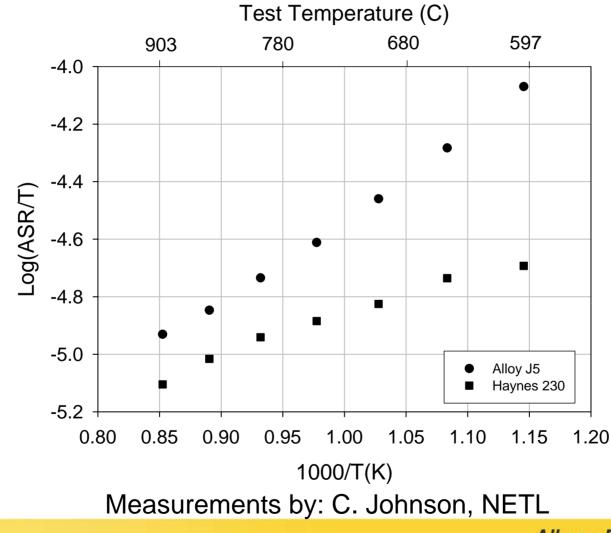
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## ASR

#### 700C/100h/Dry Air



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## Modifications for Improved Oxidation Resistance

Ferritic Steels

Nickel Alloys

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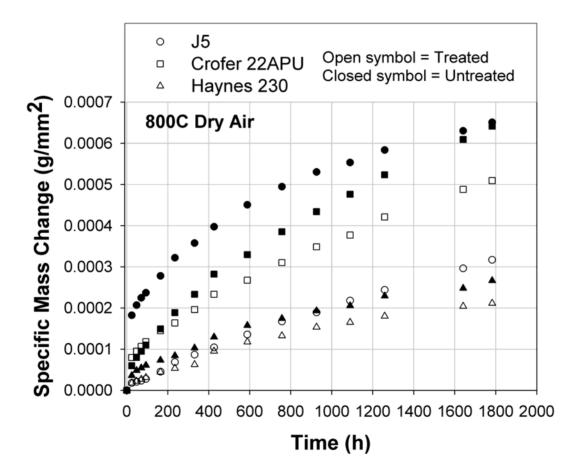


## **Reactive Element Additions**

- Minor additions of rare earth (Ce, La, Y, etc.) improve oxidation resistance.
- Developed method for enhancing rare earth element (RE) content of alloys (patent application filed).
- Comparing with other treatments, such as method described by Hou and Stringer (1987).



## **Treatment to Enhance Oxidation Resistance Via RE Additions**

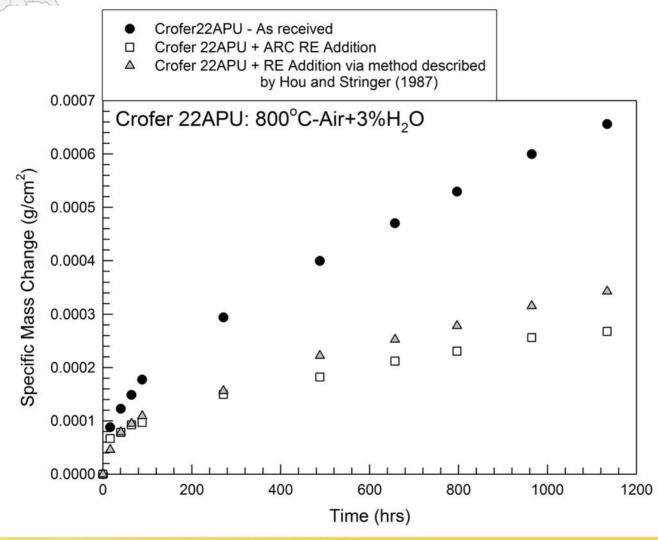


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## **Crofer 22APU**



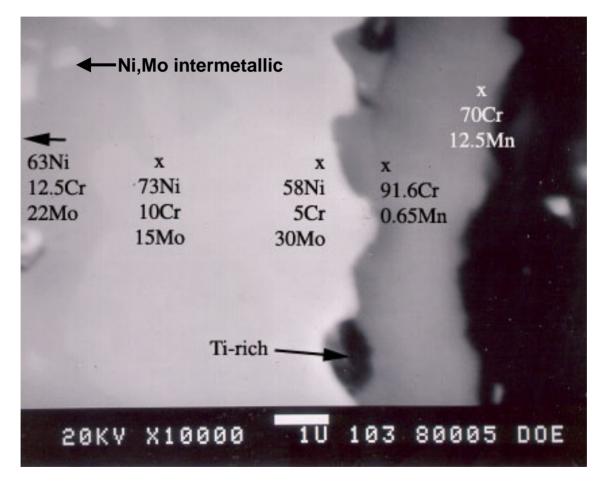
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### Oxide Scale: Alloy J5 500hr - 800°C dry air



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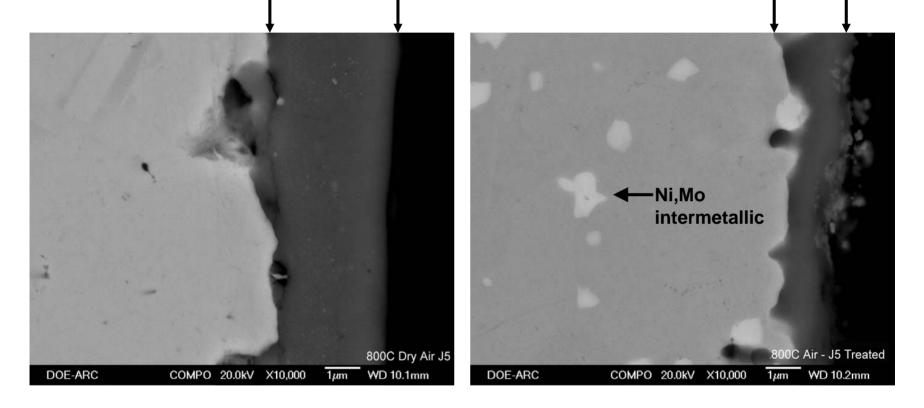
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### Oxide Scale: Alloy J5 1800 hrs - 800°C dry air



#### As polished

+ ARC Treated

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## **Alloy J5: Strip Production**



#### A length of 4" wide x 0.020" thick Alloy J5 prepared by cold rolling

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# Alloy J5 Strip and Treated J5 Strip

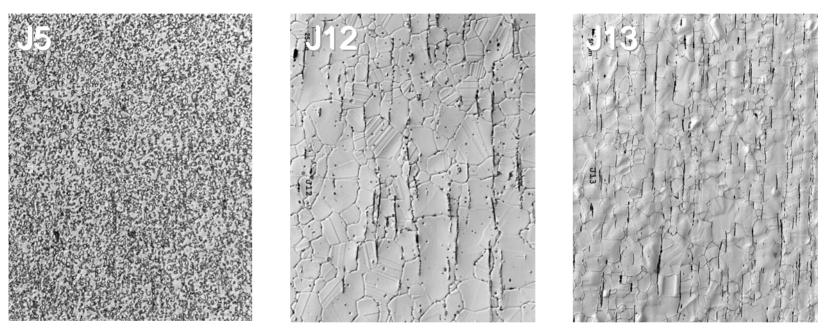
- PNNL (J. Stevenson and G. Yang)
  - Sent for testing
- GE (J. Guan, GE-Energy Systems and K. Browall, GE-GR&D)
   In process of delivering material
- Requests for material from:
  - Versa Power Systems (Canada)
  - Korean Advanced Institute of Science and Technology (Korea)
  - Ikerlan Technical Research Center (Spain)
- Will send sample of J5 to any SECA participant or US entity for evaluation Contact: J. Dunning: dunning@alrc.doe.gov (541) 967-5885

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Alloys J12&J13



- J5 derivates designed using ThermoCalc (minimize Ni-Mo ppt)
- Microstructures after aging at 800°C for 40 hours
   J5 → Ni-Mo ppt prevalent; J12 & J13 → few ppt
- Evaluating corrosion behavior ( $800^{\circ}C$ -Air+3%H<sub>2</sub>O)

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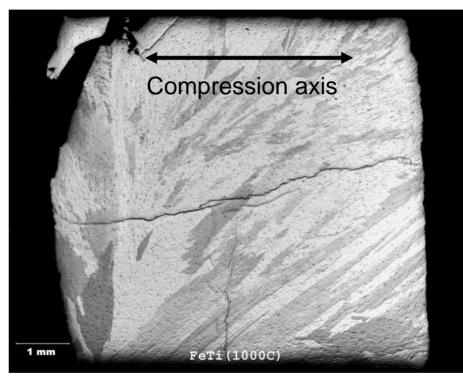
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## Fe-Ti for Argonne National Laboratory

- Two Fe-Ti intermetallic alloys prepared by arc melting.
- Hot-hardness and hotcompression tests to determine formability
  - poor formability
  - p/m alloy
- Ingots sent to ANL (Terry Cruse)



# Sample after compression testing at 1000°C.

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## Materials Performance for Heat Exchangers & Other Balance of Plant (BOP) Components for (SOFC)



### **Generic SOFC System Components**

- 1. Fuel Cell Stack
- 2. Fuel Pre-reformer/Reformer
- 3. Process Gas Heater
- 4. Fuel De-sulfurizer
- 5. Air Pre-heater
- 6. Effluent Burner
- 7. Heat Recovery
- 8. Fuel Management
- 9. Air Blower
- 10. Control Unit
- 11. Power Conversion Unit
- 12. Back-up Power Unit
- 13. Purge Gas
- 14. Water Purification for Start-up Steam

Fontell et al., "Conceptual Study of a 250 kW Planar SOFC System for CHP Application," J. Power Sources, 131 (2004) 49-56.

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#### **Cost Structure for 250 kW SOFC System**

Stack	31%
Fuel System	8%
Air System	6%
Exhaust System	2%
Start-up System	2%
Purge Gas System	0%
System Control	17%
Power Electronics	15%
Insulation	3%
Structure	2%
Labor and Overhead	15%

Fontell *et al.*, "Conceptual Study of a 250 kW Planar SOFC System for CHP Application," *J. Power Sources*, 131 (2004) 49-56.

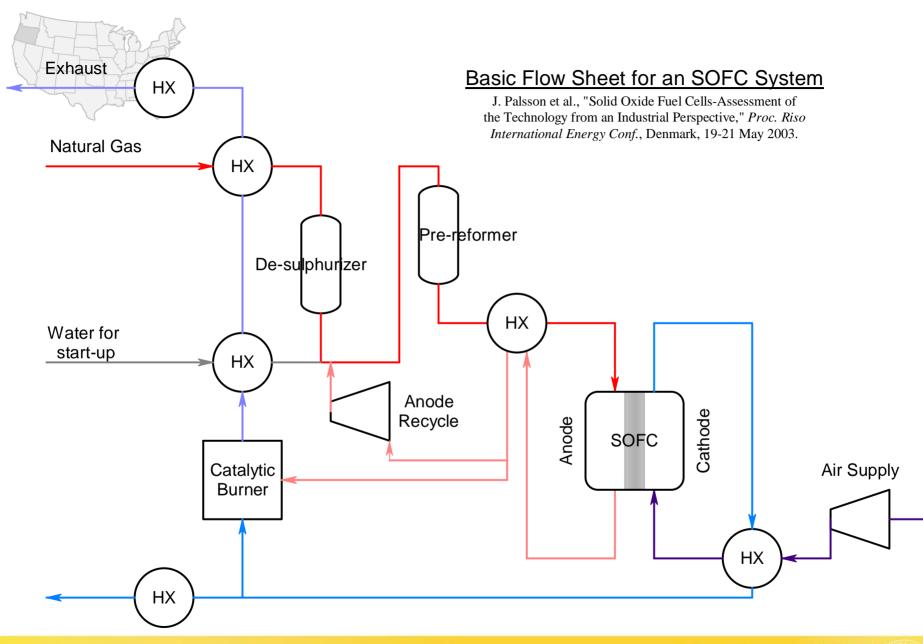


#### **Cost Structure for 250 kW SOFC System**

Stack	31%	
Fuel System	8%	
Air System	6%	
Exhaust System	2%	
Start-up System	2%	
Purge Gas System	0%	<u>= 54%</u>
System Control	17%	
Power Electronics	15%	
Insulation	3%	
Structure	2%	
Labor and Overhead	15%	1

Fontell *et al.*, "Conceptual Study of a 250 kW Planar SOFC System for CHP Application," *J. Power Sources*, 131 (2004) 49-56.

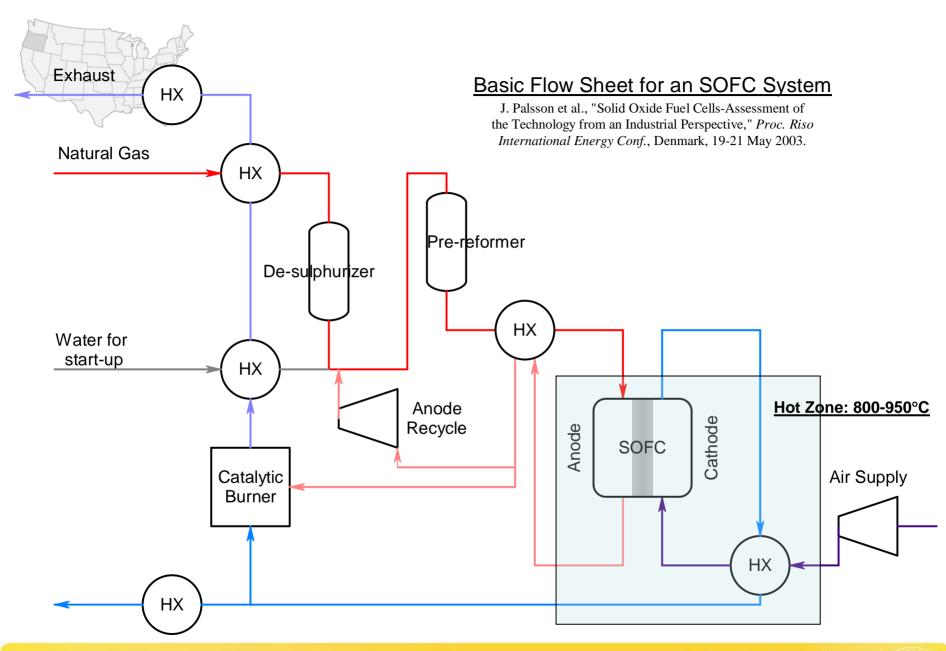




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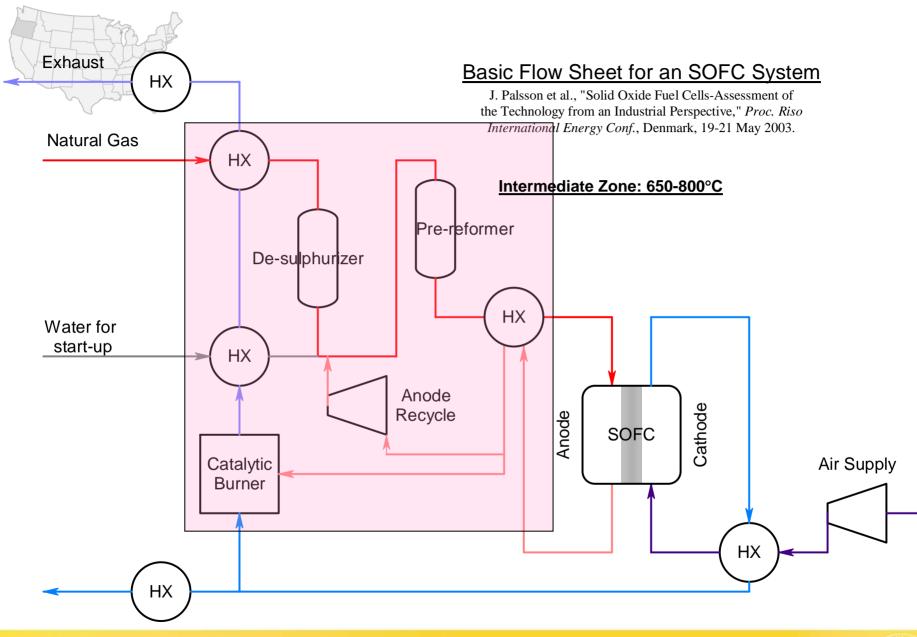
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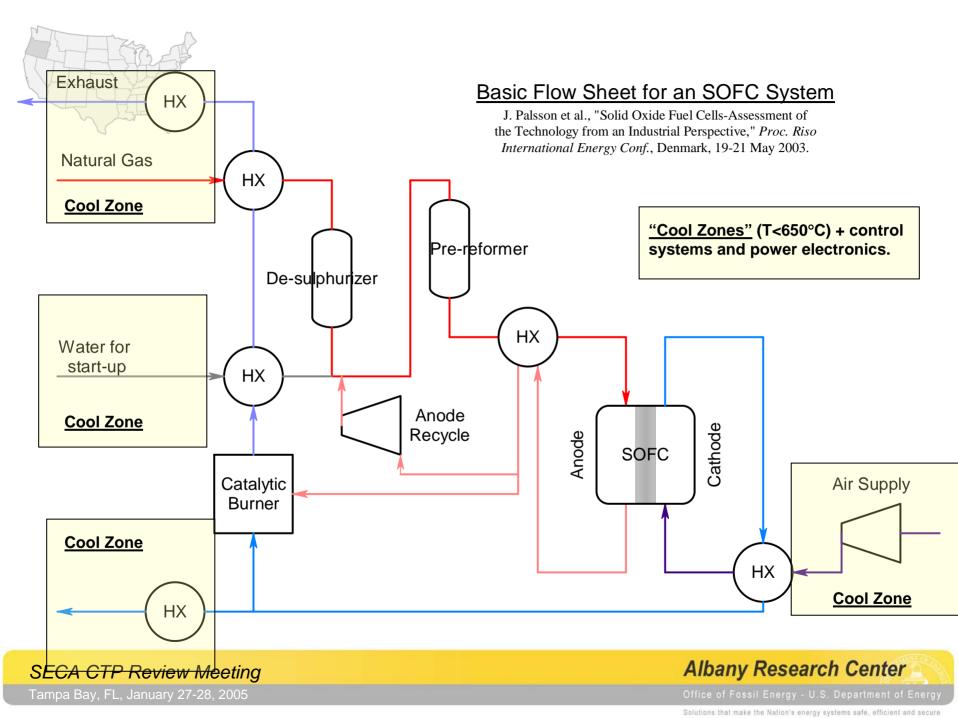
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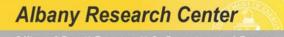
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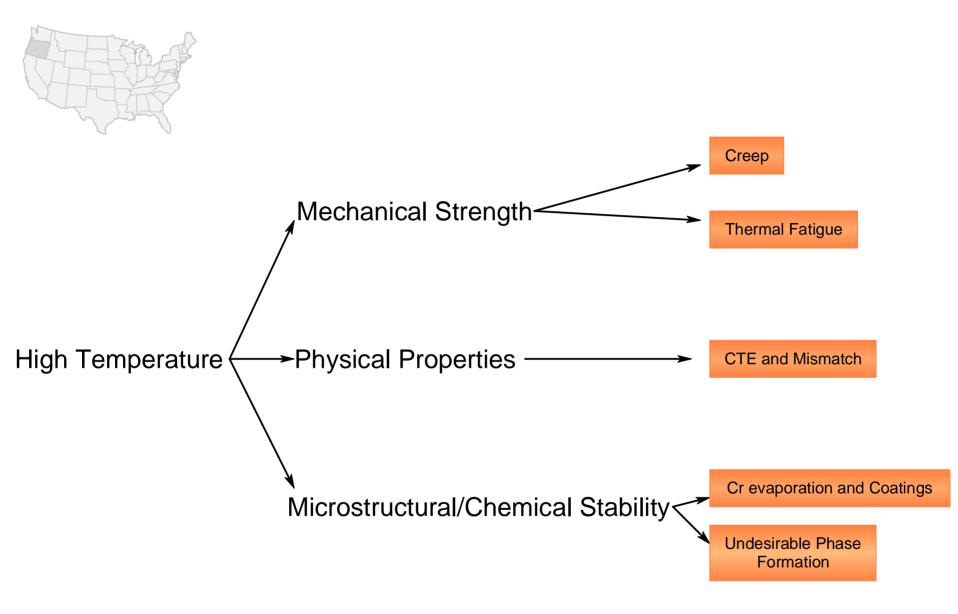




### BOP Component Design and Testing Strategy

- 1. Define the component requirements.
- 2. Identify candidate materials.
- 3. Evaluate materials in depth.
- 4. Specify and select materials.
- 5. Establish a strategy for evaluating generic candidate BOP components.

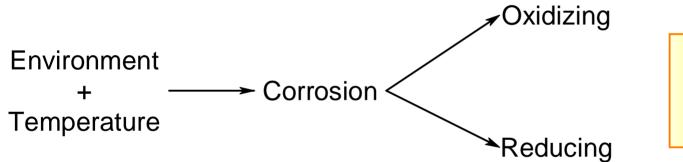




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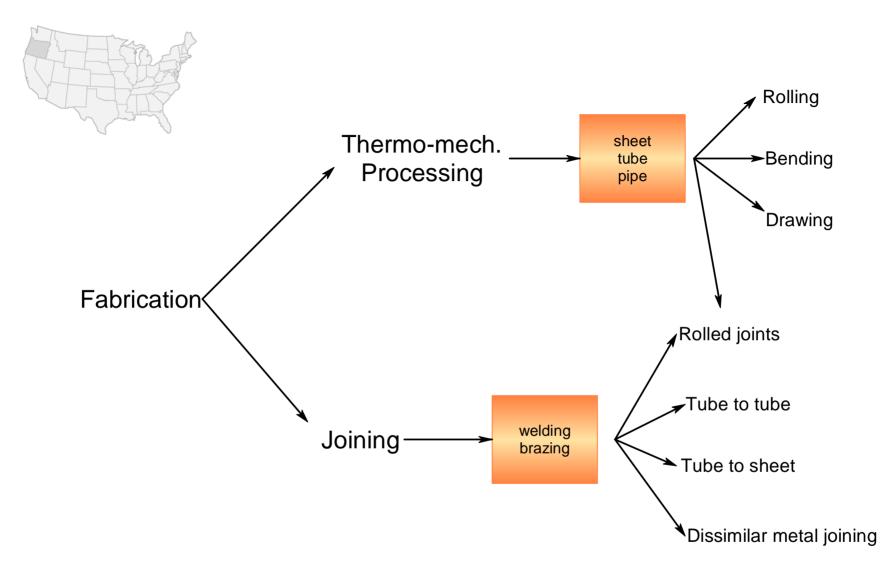
monolithic surface modified composite cladded

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### **Heat Exchangers**

- 1. Plate and fin
- 2. Shell and tube
- 3. Finned tubes
- 4. Tube-in-tube

### **Materials of Construction**

Raw materials Alloy processing Fabrication Joining HX production method

ost

- 1. Ferritic stainless steels
- 2. Austenitic stainless steels
- 3. Nickel alloys
  - . Ni-base superalloys
- 5. Ceramics
- 6. Hi-temp composites

Decreasing temperature

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### BOP Systems Approach (Identify-Evaluate-Specify)

ARC Alloy Design/Development

Laboratory Materials Testing

BOP Prototype Component Testing

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Investigate low cost material alternatives:

> Identify coating/surface modification strategies.

- Develop application strategies for any material/component configuration.
- > Optimize for lowest cost and greatest protection.
- Evaluate efficacy of approach.







### Laboratory Materials Testing

- Exposure to airExposure to fuel
  - gas/effluent
    - (with and without S)
- Exposure to dual
  - environment

Develop empirical equations to quantify material wastage in SOFC environment.

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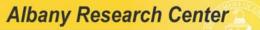
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Prototype Component Testing Serve as a test platform for SOFC-BOP prototype components: >Test single components >Test "system" components • Upstream of the FC stack Downstream of the FC stack

For a set of SOFC conditions: (1) measure component efficiency<br/>(2) determine material wastage<br/>(3) perform forensic analysis of<br/>spent component

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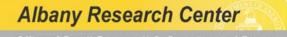
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### **Research Approach**

- 1. Construction of BOP Component Testing Facility
- 2. Material and Component Testing of High Temperature Heat Exchangers and Other BOP Components
- 3. Fuel Chemistry: Effects of Sulfur on BOP Components
- 4. SOFC/BOP Efficiency Optimization







## Construction of BOP Testing Facility

Approach

Simulated combustion environment using a "furnace/hotbox" with feed through connections for air and fuel/effluent gases.

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### **Material and Component Testing of High Temperature Heat Exchanger**

- Mechanical and Physical Property Behavior of BOP Candidate Materials
- Prototype BOP Component Testing
- Microscopic Investigations
- Characterization of Scales





### **Prototype BOP Component Testing**

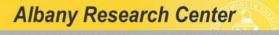
- Test facility must be flexible enough to use: different fuel chemistries different operating temperatures different operating pressures (but not pressurized)
- Must be modular in design to facilitate the easy insertion and removal of BOP components
- Allow easy post mortem analysis of BOP components
- Allow evaluation of operating efficiency of the "system" for both the SOFC and the BOP components





### **General Summary**

- <u>Identify</u>, <u>evaluate</u> (test as needed) and <u>specify</u> materials for use as BOP components in SOFC applications. Explore coating/surface modification strategies to extend operational range.
- Design and construct a BOP component and BOP component system test facility.





### Summary

#### (Mechanical and Physical Property Behavior)

- 1. Physical characterization of the potential materials of construction for BOP components.
- 2. Analysis of mechanical behavior of materials of construction for BOP components.
- 3. Evaluation of the microstructural stability of BOP materials after long-term, high temperature exposure.
- 4. Evaluation of BOP materials after long-term, dualatmosphere, high temperature exposure.
- 5. Characterization of the microstructure and integrity of joints between similar and dissimilar materials in BOP components.





#### Summary

(Proposed BOP Material and Component Test Conditions)

<b><u>Temperature</u></b>		Pressure	<u>Environment</u>	Flow Rate
to	<u>НТ</u> 700°С to 900°С	♀ 110 kPa (internal)	Fuel Gas Effluent Air H <sub>2</sub> O (w/ and w/o S)	100 slpm (or any other suggestions)

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### Summary

(Proposed BOP SOFC Environmental Conditions)

Air	<u>Fuel Gas</u>	<u>Effluent</u>
Laboratory	76.0 N <sub>2</sub>	46.5 N <sub>2</sub>
Air	<b>15.0</b> O <sub>2</sub>	27.0 H <sub>2</sub>
	6.5 H <sub>2</sub> O	6.0 H <sub>2</sub> O
	2.5 CO <sub>2</sub>	3.5 CO <sub>2</sub>
	(Sulfur)	13.0 CO
		4.0 CH <sub>4</sub>
		(SO <sub>2</sub> )

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