Comparison Of Cleaning Methods For Analysis Of Underground Beryllium Corrosion

National Association of Corrosion Engineers 2006 Annual Conference

M. K. Adler Flitton T. S. Yoder

March 2006

The INL is a U.S. Department of Energy National Laboratory operated by Battelle Energy Alliance



This is a preprint of a paper intended for publication in a journal or proceedings. Since changes may not be made before publication, this preprint should not be cited or reproduced without permission of the author. This document was prepared as an account of work sponsored by an agency of the United States Government. Neither the United States Government nor any agency thereof, or any of their employees, makes any warranty, expressed or implied, or assumes any legal liability or responsibility for any third party's use, or the results of such use, of any information, apparatus, product or process disclosed in this report, or represents that its use by such third party would not infringe privately owned rights. The views expressed in this paper are not necessarily those of the United States Government or the sponsoring agency.

COMPARISON OF CLEANING METHODS FOR ANALYSIS OF UNDERGROUND BERYLLIUM CORROSION

M. K. Adler Flitton Idaho National Laboratory P. O. Box 1625 Idaho Falls, ID 83415

T. S. Yoder Idaho National Laboratory P.O. Box 1625 Idaho Falls, ID 83415

ABSTRACT

The subsurface radioactive disposal site located at the Idaho National Laboratory contains neutron-activated beryllium metals from non-fuel nuclear-reactor-core components. A long-term underground corrosion test is being conducted to obtain site-specific corrosion rates of the disposed beryllium to support efforts to more accurately estimate the transfer of activated elements in the surrounding arid vadose zone environment. During the corrosion analysis, two cleaning methods were used. This paper describes the cleaning methods and presents a comparison of the results.

Keywords: beryllium, vadose zone, neutron-activated metals, nuclear reactor components, underground corrosion.

INTRODUCTION

The long-term corrosion test is designed to assist in the determination of site-specific corrosion rates of neutron-irradiated beryllium buried in an arid vadose zone environment at the radioactive disposal site at Idaho National Laboratory. Corrosion rates are based on mass loss from nonradioactive beryllium coupons exposed to underground site conditions. The corrosion rates, once determined, reduce the uncertainty of the site-specific transfer of radioactive isotopes to the environment (radiological release rates). Of interest are the disposed beryllium reflector blocks and outer shim control cylinders. These components, when exposed to high neutron fluxes in a reactor environment, become activated with long-lived radioactive isotopes. After disposal, corrosion processes can cause these radioactive isotopes to be released from the irradiated beryllium waste to the environment.

The long-term corrosion testing implemented direct corrosion testing (i.e., burying beryllium coupons in the soil). ^{2,3,4,5} Previous beryllium corrosion studies do not include underground corrosion testing. ⁶ For this study, the corrosion analysis applies Test Method ASTM G 1-03⁷ for cleaning and mass loss determination, but beryllium is a metal excluded from the ASTM G 1 procedures. During the corrosion analysis, two cleaning methods were used: ASTM G 1-C.5.1 *Magnesium and Magnesium Alloys* and ASTM G 1-C.1.1 *Aluminum and Aluminum Alloys*. This paper presents a comparison of the results.

EXPERIMENTAL PROCEDURE

The long-term corrosion testing implements direct corrosion testing (i.e., burying beryllium coupons in the soil, ASTM G 4). The beryllium corrosion coupons are retrieved after exposure to the underground environment. Following the procedures from Test Method ASTM G 1, the beryllium corrosion product and adjacent soils tightly adheres to the metal base requiring chemical cleaning after the simple brush-wash and rinse cycle. Figures 1 and 2 show the corroded beryllium coupon after exposure to soil environment and after cleaning.

Based on the beryllium manufacturer's recommendations, two chemical cleaning procedures are applicable for cleaning the beryllium corrosion coupons: ASTM G 1-C.5.1 *Magnesium and Magnesium Alloys* and ASTM G 1-C.1.1 *Aluminum and Aluminum Alloys*. Each of these procedures was used during the study. Following the procedures, multiple cleaning cycles were performed on corroded coupons and blank (archived) coupons until mass loss results indicated the successful removal of the corrosion products. Microscopic examination followed the cleaning process to visually verify the results. In addition, archived coupons (in an asreceived state from the manufacturer) and cleaned coupons (from each of the two chemical procedures) were then further examined using the scanning electron microscope and compared.

RESULTS

The exposed beryllium coupons were cleaned following ASTM G 1. First cycle cleaning, water wash and brush, did not remove all of the adhering corrosion products. Chemical cleaning was then applied to all exposed coupons following ASTM G 1-C.5.1 *Magnesium and Magnesium Alloys* or ASTM G 1-C.1.1 *Aluminum and Aluminum Alloys*. Figure 3 shows the typical cleaning process for the beryllium coupons. The cleaning curves do not indicate a difference in method effectiveness.

To establish another basis of comparison, archived beryllium coupons were examined using the scanning electron microscope. The archived beryllium coupons provide the baseline. Baseline scans are shown in Figures 4 and 5 and the baseline spectrum is shown in Figure 6. The spectrum clearly shows peaks of the elemental impurities of aluminum, silicon, and iron.

The first set of beryllium corrosion coupons were chemically cleaned following the ASTM G 1-C.5.1 *Magnesium and Magnesium Alloys* procedure. Based on cleaning curves and visual examination with a microscope, corrosion products were successfully removed using this procedure, Figure 7. ASTM G 1-C.5.1 uses silver chromate to precipitate chloride as a silver salt. Upon closer examination using the scanning electron microscope, residual, crystalline precipitates remain in the corroded areas after thorough cleaning; Figures 8 and 9. Comparing the baseline spectrum (Figure 6) to the coupon spectrum (Figure 10), verifies the precipitates are identifiable as silver chloride. This precipitate strongly adheres to the beryllium base metal even after multiple cleaning cycles.

The second set of beryllium corrosion coupons were chemically cleaned following the ASTM G 1-C.1.1 *Aluminum and Aluminum Alloys* procedure. Based on cleaning curves and visual examination with a microscope, corrosion products were successfully removed using this procedure, Figure 11. The coupons were more closely examined using the scanning electron microscope, Figures 12 and 13. By comparing the baseline spectrum (Figure 6) to the coupon spectrum (Figure 14), residual phosphorous does remain on the coupon from the cleaning solution. However, the coupon is free from any residual precipitates.

CONCLUSIONS

Two procedures were used to clean beryllium corrosion coupons after underground exposure: ASTM G 1-C.5.1 *Magnesium and Magnesium Alloys* and ASTM G 1-C.1.1 *Aluminum and Aluminum Alloys*. Both procedures resulted in coupons clean of adhering soils and corrosion products. The coupons were then further examination with scanning electron microscopy and compared with an archived coupon. Both cleaning procedures left residuals from the cleaning solutions, but the coupons cleaned using ASTM G 1-C.5.1 *Magnesium and Magnesium Alloys* left silver chloride precipitates predominately in the corroded areas of the beryllium. Based on the results of this study, ASTM G 1-C.1.1 *Aluminum and Aluminum Alloys* is the preferred procedure for chemically cleaning underground beryllium corrosion coupons.

ACKNOWLEDGEMENTS

This work was carried out under U.S. Department of Energy (DOE) contracts DE-AC07-05ID14516 and DE-AC07-05ID14517, with funding provided by the DOE Environmental Management, Low Level Waste Management Program. The support necessary to successfully pursue the objectives of this test has been vast, and the authors wish to thank the many Idaho National Laboratory personnel who have made significant contributions and made this project both safe and successful.



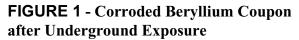




FIGURE 2 - Corroded Beryllium Coupon after Cleaning

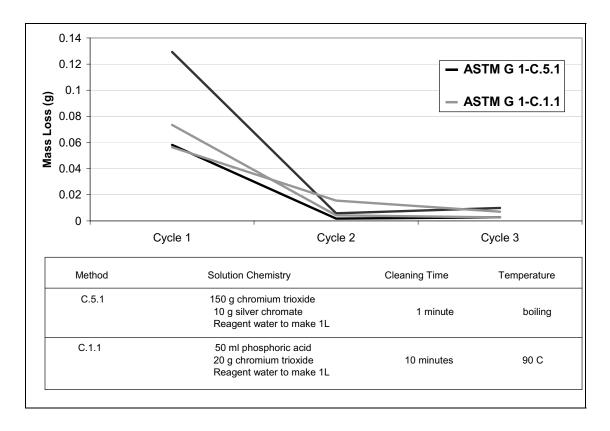


FIGURE 3 – Beryllium Coupon Cleaning Curves and Procedures

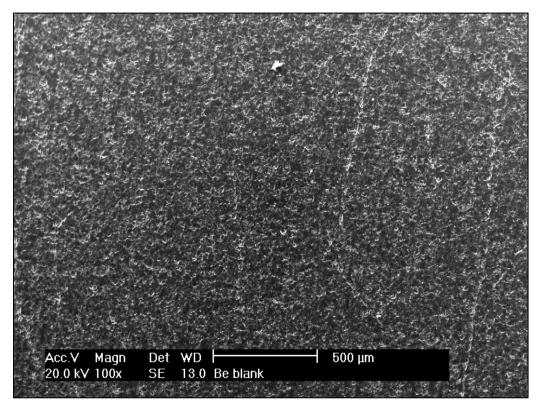


FIGURE 4 - Archived Beryllium Coupon (As-Received Condition), Scanning Electron Microscope Image 500 µm

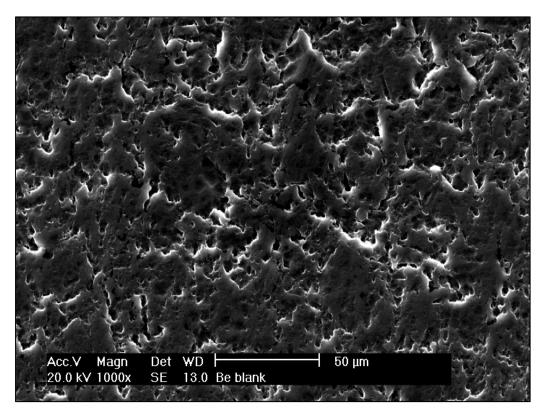


FIGURE 5 - Archived Beryllium Coupon (As-Received Condition), Scanning Electron Microscope Image 50 µm

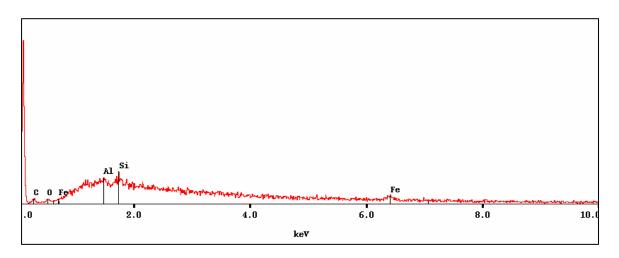


FIGURE 6 - Archived Beryllium Coupon (As Received Condition), Scanning Electron Microscope Spectrum

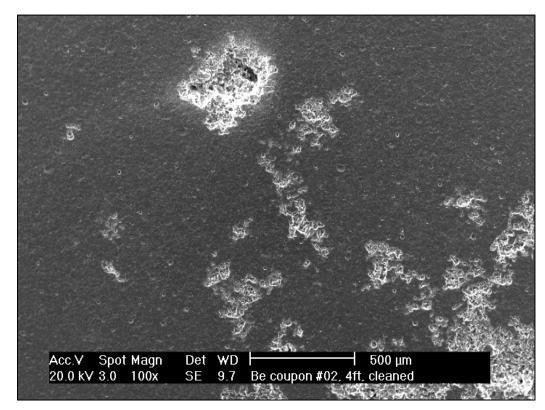


FIGURE 7 - Beryllium Corrosion Coupon Cleaned Using ASTM G 1-C.5.1, Scanning Electron Microscope Image 500 μm

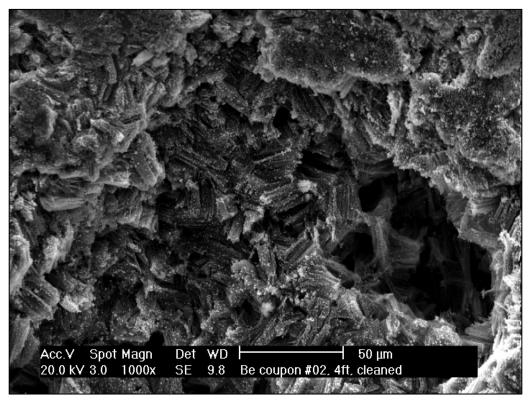


FIGURE 8 - Beryllium Corrosion Coupon Cleaned Using ASTM G 1-C.5.1, Scanning Electron Microscope Image 50 µm

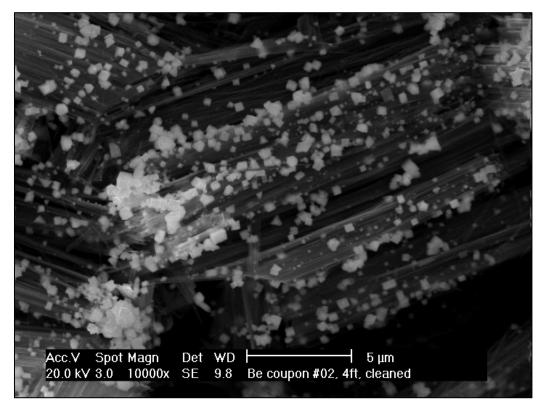


FIGURE 9 - Beryllium Corrosion Coupon Cleaned Using ASTM G 1-C.5.1, Scanning Electron Microscope Image 5 µm

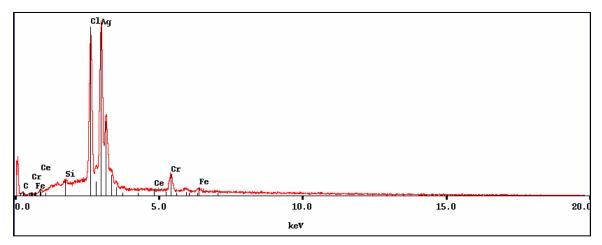


FIGURE 10 - Beryllium Corrosion Coupon Cleaned Using ASTM G 1-C.5.1, Scanning Electron Microscope Spectrum

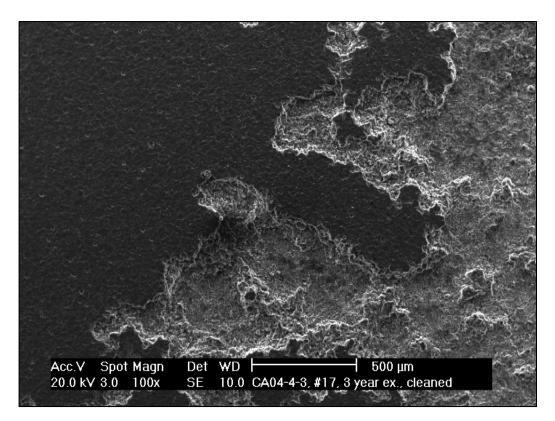


FIGURE 11 - Beryllium Corrosion Coupon Cleaned Using ASTM G 1-C.1.1, Scanning Electron Microscope Image 500 μm

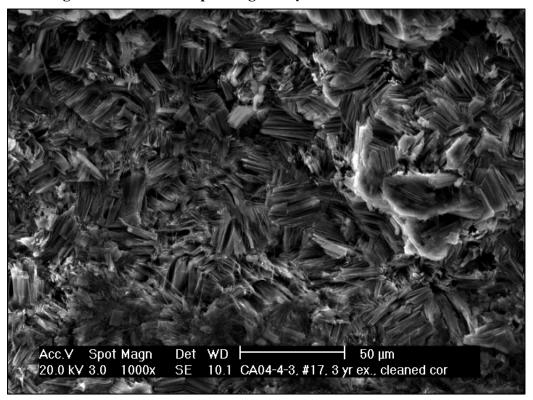


FIGURE 12 - Beryllium Corrosion Coupon Cleaned Using ASTM G 1-C.1.1, Scanning Electron Microscope Image 50 μm

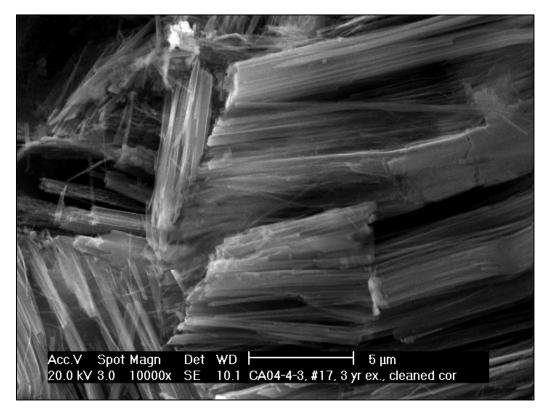


FIGURE 13 - Beryllium Corrosion Coupon Cleaned Using ASTM G 1-C.1.1, Scanning Electron Microscope Image 5 μm

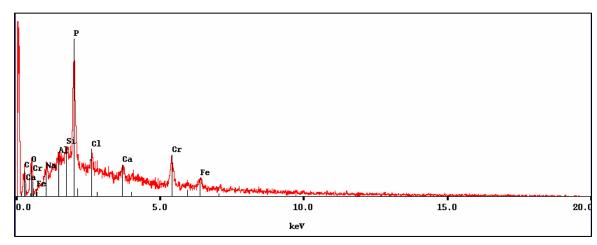


FIGURE 14 - Beryllium Corrosion Coupon Cleaned Using ASTM G 1-C.1.1, Scanning Electron Microscope Spectrum

REFERENCES

- 1. Adler Flitton, M. K., C. W. Bishop, M. E. Delwiche, and T. S. Yoder, *Long Term Corrosion/Degradation Test Six Year Results*, INEEL/EXT-04-02335, Idaho National Engineering and Environmental Laboratory, Bechtel BWXT Idaho, LLC, Idaho Falls, Idaho, 2004.
- 2. Romanoff, M., *Underground Corrosion*, NBS 579, NTS PB 168350, National Bureau of Standards, 1957.
- 3. Adler Flitton, M. K. and E. Escalante, "Simulated Service Testing in Soil," American Society of Metals Handbook, Volume 13A: *Corrosion: Fundamentals, Testing and Protection*, sub-section S-3c, American Society for Metals International, Metals Park, Ohio, August 2003.
- 4. Miller, P. D., and W. K. Boyd, *Corrosion of Beryllium*, Defense Metals Information Center Report 242, Battelle Memorial Institute, December 11, 1967.
- 5. Test Method G 4, Standard Guide for Conducting Corrosion Coupon Tests in Field Applications, American Society for Testing and Materials, West Conshohocken, Pennsylvania, 2001.
- 6. Escalante, E., "Soils," *Corrosion Tests and Standards*, American Society for Testing and Materials, West Conshohocken, Pennsylvania, 1995.
- 7. Test Method G 1-90, Practice for Preparing, Cleaning, and Evaluating Corrosion Test Specimens, American Society for Testing and Materials, West Conshohocken, Pennsylvania, 1999.