# **National Energy Technology Laboratory**



Microstructure and high temperature oxidation behavior of Cr-W alloys

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#### National Energy Technology Laboratory



**Office of Fossil Energy** 



# **High Temperature Materials in Power Generation**

- In energy systems, there is an increasing demand for new materials with
  - -higher strength,
  - -creep resistance,
  - -corrosion resistance,
  - -thermal fatigue resistance, and
  - -wear resistance
- to increase generation efficiency,
- to reduce environmental pollution,
- to make new energy generating technologies economically feasible.

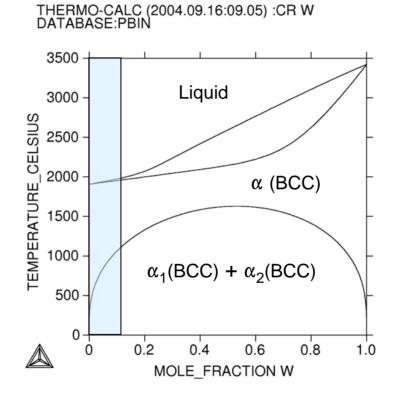


# **High Temperature Materials in Power Generation**

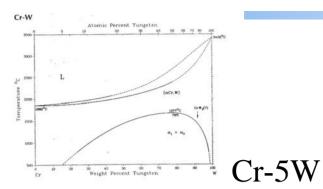
- New technologies such as ultra super critical steam plants, integrated gasification combined cycle, pressurized fluidized bed combustion, and solid oxide fuel cells are being developed to meet this demand.
- One common barrier in the development of these different technologies is insufficiency of existing materials.
- Even the highly alloyed and high cost Ni based superalloys do not have desired properties at the temperatures that some parts of the future energy generating systems will be exposed.
- Therefore, there is an urgent need to develop structural alloys with desirable properties using elements with high melting point.



# **Cr-W Equilibrium Phase Diagram**



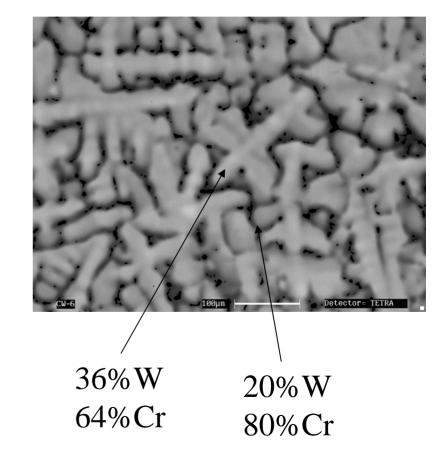




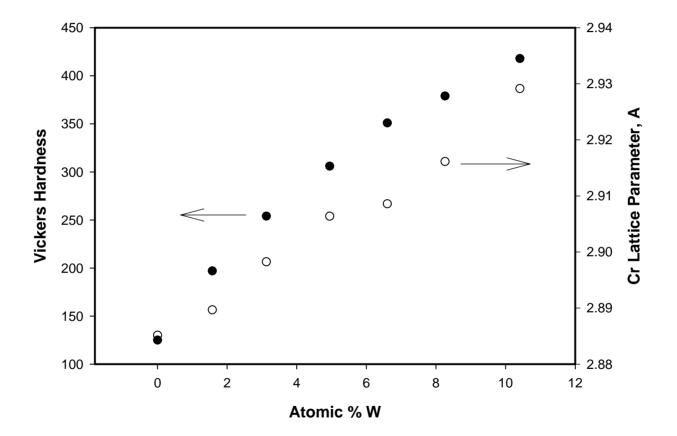
### Microstructure

CW-1 Detector= TETRA 5%W 6%W 95%Cr 94%Cr

#### Cr-30W

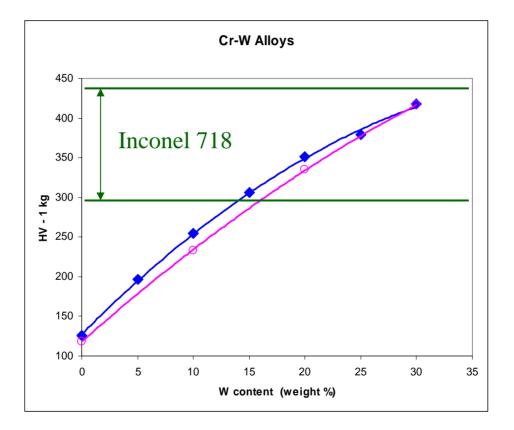


# **Solid solution strengthening**



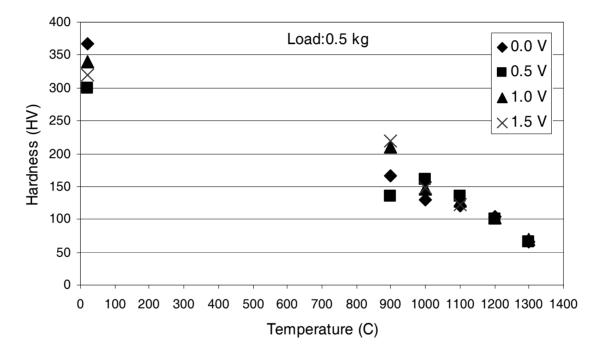


# **Room Temperature Strength**





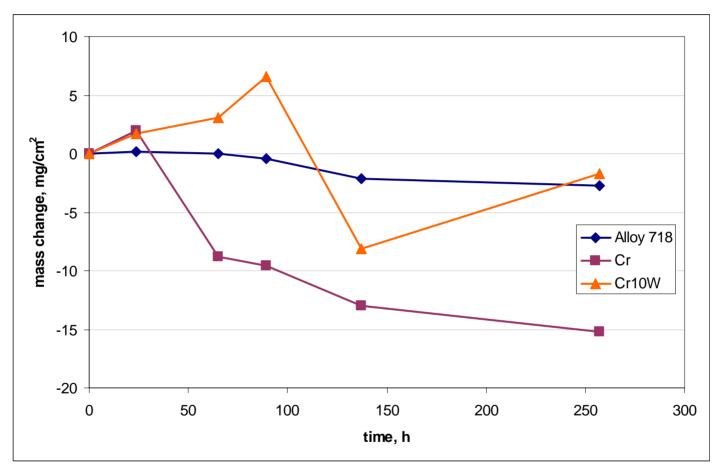
# **Hardness at Elevated Temperatures**



Results of hot hardness tests done using 0.5 kg load on Vicker's indenter.



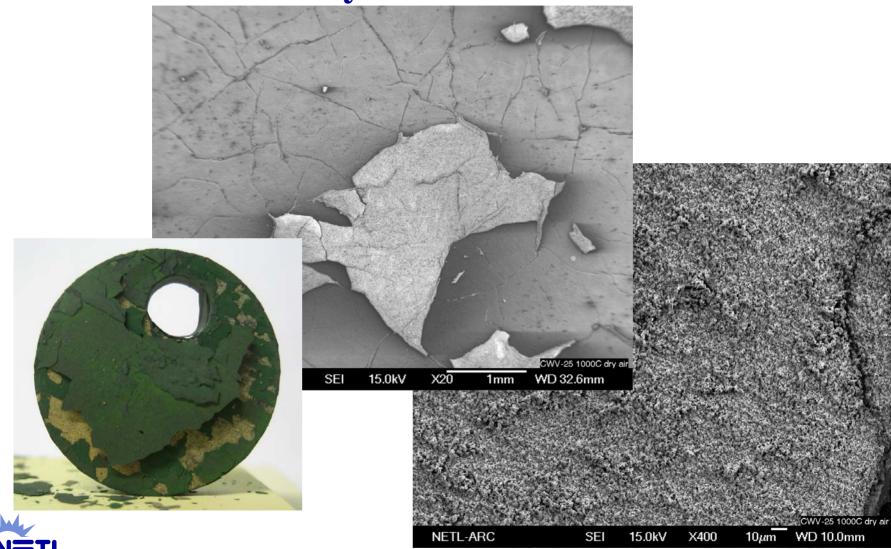
# **High Temperature Oxidation**



## @1000°C in dry air



# Oxidation of Cr-10W in dry air at 1000°C



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# To improve high temperature oxidation resistance of Cr-W alloys

- Prevent Cr<sub>2</sub>O<sub>3</sub> scale from spalling
- Obtain stable scales at temperatures above 900°C



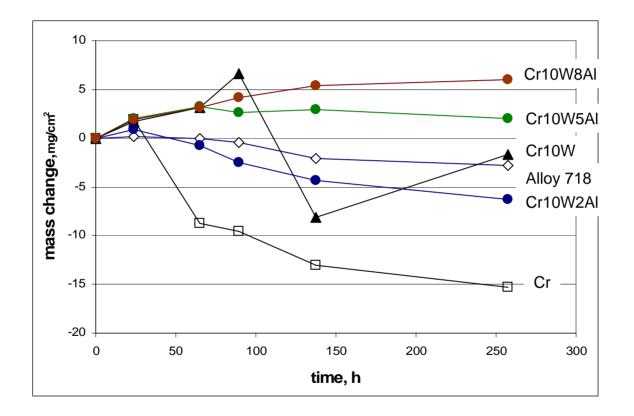
# **Alloying with Aluminum**

### • Aluminum is added to Cr-10W alloy

- -2 wt%
- -5 wt%
- -8 wt%



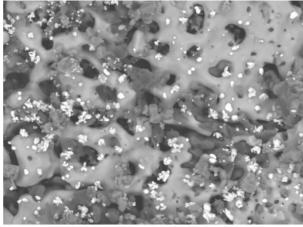
# **Oxidation of Cr-10W-Al Alloys**



@1000°C in dry air

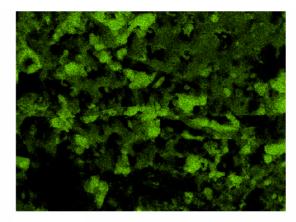


# Elemental X-ray maps of oxidation scale on Cr-10W-8Al after exposure to dry air at 1000°C

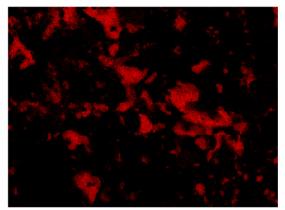


10µm

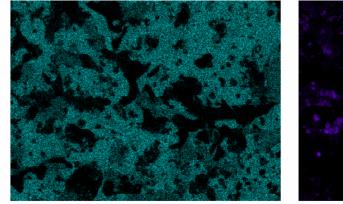
Electron Image 1



Ο Κ<sub>α1</sub>



Al  $K_{\alpha 1}$ 





 $WM_{\alpha 1}$ 



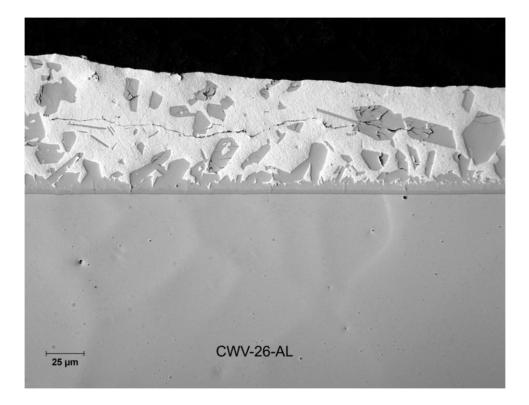
# **Alloying with Aluminum**

### Improved oxidation resistance

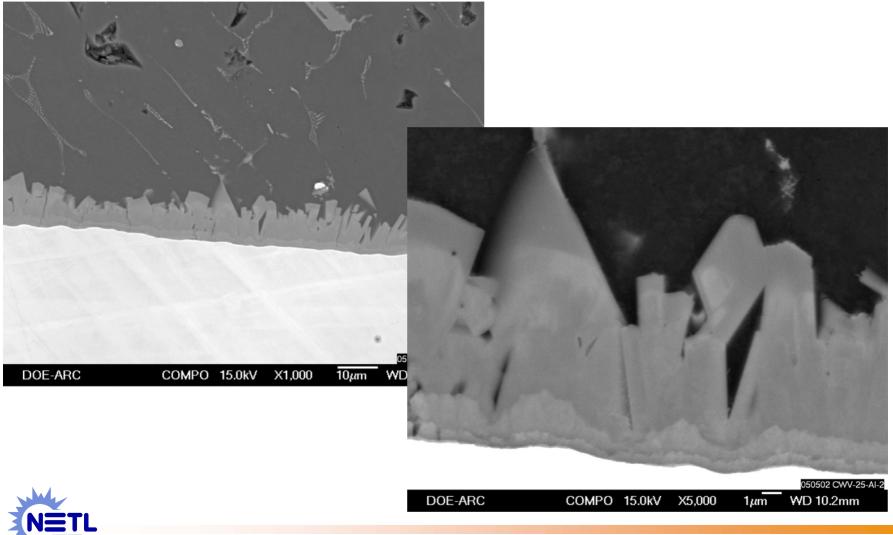
-by reducing spallation drastically -by forming stable Al<sub>2</sub>O<sub>3</sub>

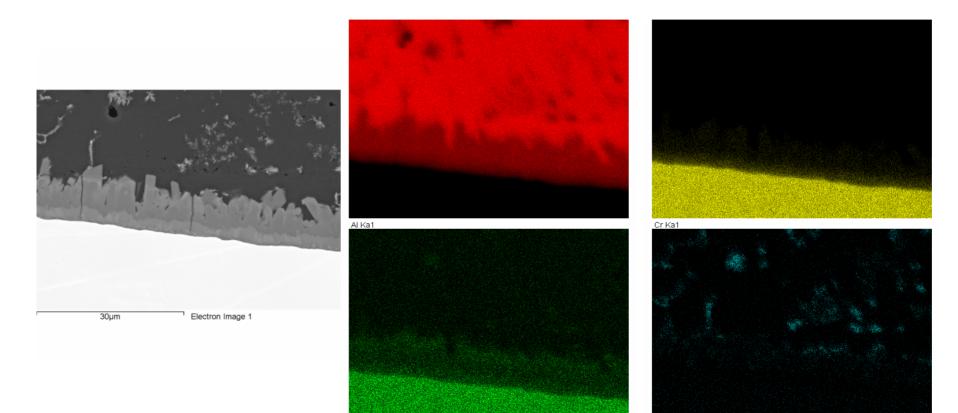
#### Increased brittleness







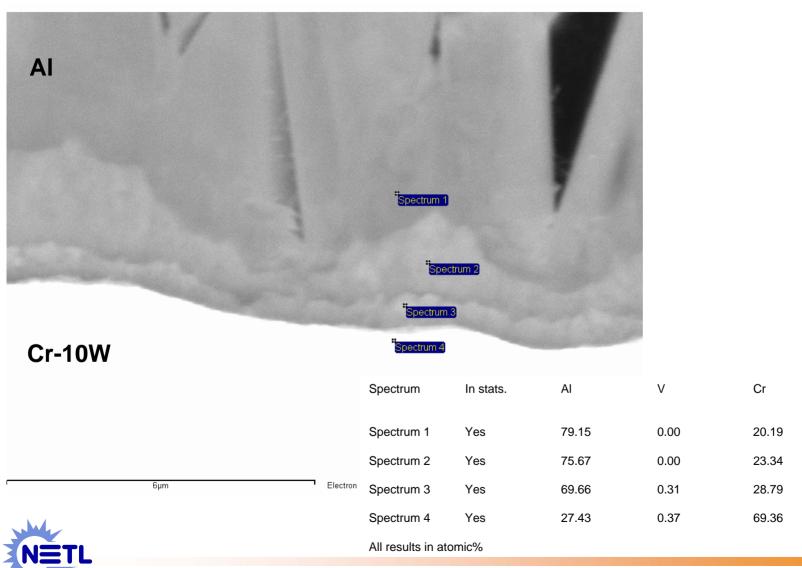






Fe Ka1





W

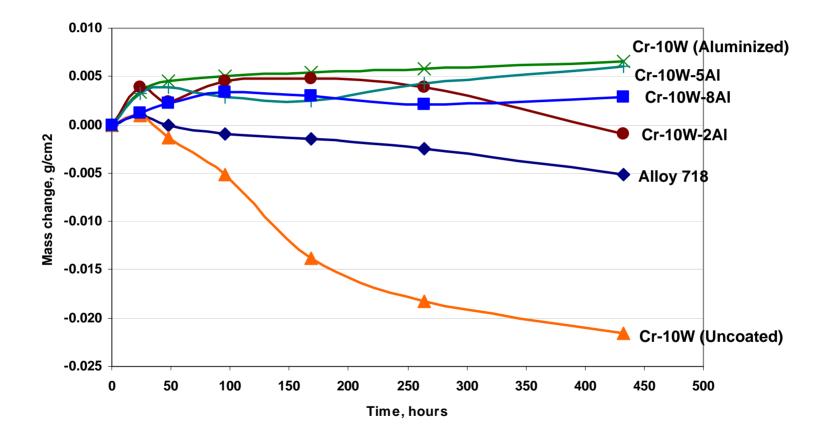
0.65

0.98

1.24

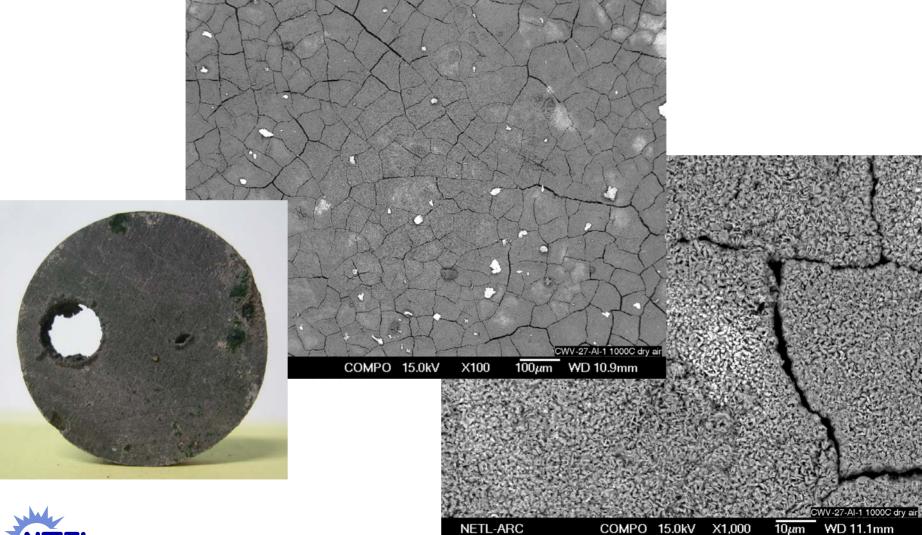
2.83

# **Oxidation of Cr-W in dry air at 1000°C**





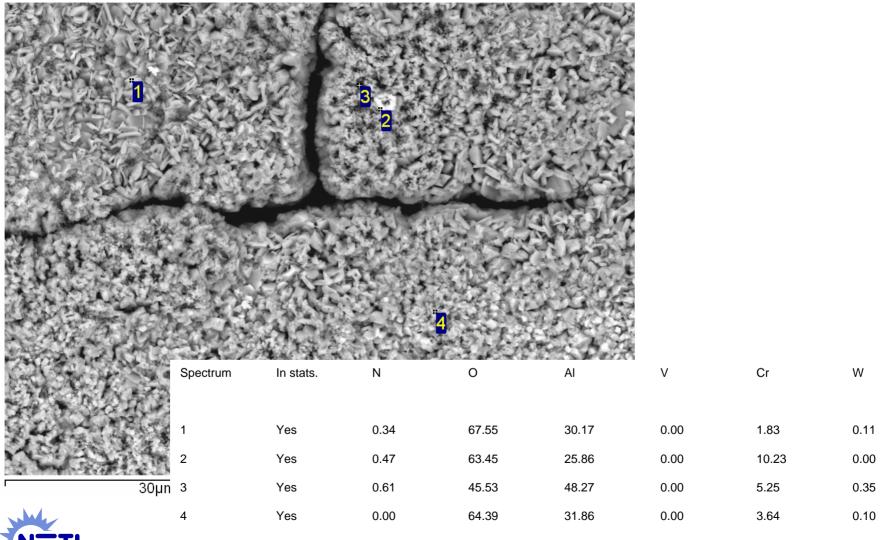
# Oxidation of Aluminized Cr-10W in dry air at 1000°C

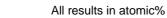




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# Oxidation of Aluminized Cr-10W in dry air at 1000°C





# Summary

- Cr-W alloys have high strength between 1000°C and 1300°C.
- Cr-W alloys have poor oxidation resistance at 1000°C in air.
- Oxidation resistance can be improved by alloying with aluminum.
- However, alloying with Al increases brittleness.
- Therefore, a better approach is aluminizing surface of Cr-W alloys.
- Aluminizing improved the oxidation resistance in dry air at 1000°C dramatically.

