

THE STEUBENVILLE COMPREHENSIVE AIR MONITORING PROGRAM (SCAMP)

**Final Report
For the Period February 1, 1999, through June 30, 2006**

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Report Issued: March 2007

**DOE Cooperative Agreement DE-FC26-00NT40771
OCDO Grant Agreement CDO/D-98-2**

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ABSTRACT

In 1999, CONSOL Energy Inc. Research & Development (CONSOL R&D), in cooperation with the research partners and funding agencies acknowledged below, launched the Steubenville Comprehensive Air Monitoring Program (SCAMP) to clarify uncertainties regarding fine particulate matter (PM_{2.5}) concentration, composition, and exposure. The study was conducted in Steubenville, Ohio, a small but heavily industrialized city located in the Upper Ohio River Valley that historically has had elevated particulate matter (PM) concentrations and been a key location for PM health effects research. In addition to providing a much-needed current assessment of PM_{2.5} in Steubenville, which had been studied previously during the late 1970s and 1980s as part of the Harvard Six Cities Study, SCAMP helped to fulfill a number of PM_{2.5} research needs that remained after the U.S. Environmental Protection Agency (EPA) promulgated a new National Ambient Air Quality Standard (NAAQS) for PM_{2.5} in 1997. These included the need for better estimates of the relationship between ambient concentrations of PM_{2.5} and its components and actual human exposures to these species, the need to characterize the potential for confounding by gaseous pollutants in PM_{2.5} epidemiology studies, the need for better characterization of individual chemical components of PM_{2.5}, and the need for data to develop State Implementation Plans (SIPs) in response to the PM_{2.5} NAAQS.

SCAMP included two major interrelated components. As part of the ambient air monitoring program, which was led by CONSOL R&D, PM_{2.5}, PM_{2.5} components, co-pollutants, and meteorological conditions were monitored from May 2000 through May 2002 at a central ambient air monitoring station in Steubenville, and PM_{2.5} and its components were simultaneously monitored at four comparatively remote satellite sites in the surrounding region. As part of the indoor and personal exposure program, which was led by the Harvard School of Public Health (HSPH) under subcontract to CONSOL R&D, personal, indoor, and outdoor exposures to PM_{2.5}, PM_{2.5} components, and gaseous pollutants were characterized during the summer and fall of 2000 for a panel of older adults and during the winter and summer of 2001 for a panel of children living in Steubenville. Smaller studies conducted as part of SCAMP included methods development and validation activities and a concurrent evaluation of the effects of PM_{2.5} and gases on the cardiac health of the older adult panel.

Results from SCAMP reveal the importance of distinct regional- and local-source mechanisms that contribute to ambient PM_{2.5} concentrations in Steubenville. Regional, secondary-sulfate-dominated PM_{2.5}, likely originating from coal-fired power plants to the west and southwest of Steubenville, is the largest contributor to PM_{2.5} mass, accounting for about 40% of the total. Local sources in the immediate Steubenville vicinity, including motor vehicles and iron and steel facilities, account for about 30% of the total PM_{2.5} mass observed in the city. PM_{2.5} from these local sources tends to be enriched in Fe, Mn, Zn, and carbon species and to correlate strongly with primary gaseous pollutants, especially during nocturnal inversion-driven episodes. Results also indicate that central-site ambient PM_{2.5} concentrations are strong proxies of corresponding personal exposures for older adults and, to a lesser extent, for children in Steubenville. The ability of ambient concentrations to reflect personal exposures differed by particle component; for example, ambient concentrations of SO₄²⁻, a predominantly regional pollutant, were better proxies of corresponding personal exposures than were ambient concentrations of elemental carbon, a predominantly local pollutant. Home ventilation conditions also significantly modified personal-ambient associations. Finally, ambient concentration and personal exposure results suggest that time-series health studies based on 24-hour ambient concentrations may not be able to separate the independent effects of particles and gases.

Funding for SCAMP was provided by the U.S. Department of Energy's National Energy Technology Laboratory, Ohio Coal Development Office within the Ohio Air Quality Development Authority, Electric Power Research Institute, American Petroleum Institute, National Mining Association, American Iron and Steel Institute, Edison Electric Institute, National Institute of Environmental Health Sciences, U.S. EPA, and CONSOL Energy Inc. Additional research participants included Ohio University, Franciscan University of Steubenville, Saint Vincent College, Wheeling Jesuit University, Optimal Technologies, Air Quality Sciences, Inc., and Control Analytics, Inc.

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1. INTRODUCTION

Fine particulate matter (PM_{2.5}), an air pollutant class nominally comprising airborne particles with aerodynamic diameters less than 2.5 µm, has garnered substantial attention from the scientific and regulatory communities because of concerns about its possible effects on human health. Driven largely by epidemiology studies that reported associations between fine particulate air pollution and mortality (e.g., Dockery et al., 1993; Pope et al., 1995) or morbidity, especially for susceptible subpopulations such as older adults (e.g., Delfino et al., 1997) and children (e.g., Dockery et al., 1996), the U.S. Environmental Protection Agency (EPA) announced a new National Ambient Air Quality Standard (NAAQS) for PM_{2.5} in July 1997. The standard limits annual mean outdoor PM_{2.5} concentrations to 15 µg/m³ (averaged over three years) and 24-hr mean outdoor PM_{2.5} concentrations to 65 µg/m³ (3-year average of annual 98th percentile 24-hr concentration). More recently, in December 2005, the U.S. EPA proposed reducing the 24-hr standard from 65 µg/m³ to 35 µg/m³, citing “increased protection against health effects associated with short-term exposure (including premature mortality and increased hospital admissions and emergency room visits),” and retaining the annual standard of 15 µg/m³ for “continuing protection against health effects associated with long-term exposure (including premature mortality and development of chronic respiratory disease)” (71 FR 2620).

However, in spite of these epidemiological findings and regulatory actions, substantial uncertainties remain regarding the actual relationship between PM_{2.5} and human health and the best means for reducing ambient PM_{2.5} concentrations to attain regulatory compliance. These uncertainties stem in large part from the fact that PM_{2.5} is a broad class of chemically and physically diverse substances that originate from a variety of natural and anthropogenic sources. Fine particles can be emitted directly to the atmosphere as primary emissions or formed secondarily from gaseous precursors. Primary particles typically include crustal material, elemental carbon (EC), certain organic compounds, fly ash, and trace metal species, whereas secondary particles typically include ammonium sulfate, ammonium nitrate, and secondary organic aerosols; the amounts of these particle components can vary appreciably in time and space. Ambient particles and their precursors are emitted by a variety of sources, such as fossil-fuel-fired power plants, automobiles, diesel engines, metal smelting and processing plants, incinerators, kilns, various industrial facilities, wood burning, forest fires, fugitive dust and soil, and sea salt. PM_{2.5} and PM₁₀ (particles with aerodynamic diameters less than 10 µm), are the only pollutants regulated under the NAAQS that do not comprise a specific chemical composition. Hence, in order to develop State Implementation Plans (SIPs) to reduce ambient mass concentrations of PM_{2.5} in nonattainment areas under the PM_{2.5} NAAQS, it is first important to identify the extents to which various chemical components contribute to PM_{2.5} mass concentrations in these areas and to identify the sources that emit these components or their precursors. Other factors, such as meteorology, that control the effects of these sources, must also be understood.

Moreover, it is still uncertain whether some or all of the health effects that have been attributed to PM_{2.5} in epidemiology studies result from the mass of fine particles in general, or whether they instead result from a specific physically- or chemically-defined type of fine particle or another pollutant that correlates to some extent with total PM_{2.5} mass concentrations. Toxicological evidence suggests that certain chemical components of PM_{2.5} are likely to be more harmful to human health than others. For example, Gurgueira et al. (2002) reported that inhalation of residual oil fly ash (ROFA) particles triggered oxidative stress in rats, whereas inhalation of carbon black aerosols did not, and Medeiros et al. (2004) studied the effects of nasal instillation of various particle surrogates in mice and concluded that acute inflammatory changes were modulated more significantly by particle composition than by particle mass. A

number of toxicology studies (e.g., Ghio et al., 1999a; Zelikoff et al., 2002; Gavett et al., 2003) have reported associations between adverse health effects and the transition metal content of particulate matter (PM). The solubility of compounds containing these transition metals may be of importance; for example, Fernandez et al. (2005) recently reported that lung permeability in mice increased significantly when the mice were exposed to soluble ZnSO₄ aerosol, but not when they were exposed to insoluble ZnO aerosol. In a recent review of toxicological and epidemiological evidence regarding the health risks of PM components, Grahame and Schlesinger (2005) concluded that “toxicology studies have identified mechanisms by which specific components of vehicular, industrial, or urban emissions might be harmful, but find that secondary sulfates and nitrates appear not to be harmful at ambient concentrations.” These authors also reported that recent innovative epidemiological studies that were able to differentiate among the effects of urban and rural air masses generally yielded similar conclusions. Indeed, several epidemiology studies (e.g., Peters et al., 1997; Burnett et al., 2000; Mar et al., 2000; Laden et al., 2000) have discovered associations between particular components, source classes, or size fractions of PM_{2.5} and adverse health effects. However, because of the paucity of ambient PM_{2.5} composition data relative to PM_{2.5} total mass concentration data, many of the epidemiological studies conducted to date have not assessed the acute health risks posed by specific particle components. (Only within the last five years have EPA monitoring sites begun routine monitoring of PM_{2.5} component concentrations on a 1-in-3 or 1-in-6 day frequency). Hence, there is a need to enhance the quantity and quality of data concerning PM_{2.5} chemical composition, especially for components such as trace metals that are difficult to measure but may have important health implications, so that the health risks posed by individual fine particle constituents can be better understood.

It is also possible that the health effects ascribed to PM_{2.5} may instead result to some extent from co-pollutants that vary collinearly with PM_{2.5} concentrations and therefore have the potential to confound the results of PM_{2.5} epidemiology studies. Gaseous pollutants such as CO, NO₂, SO₂, and O₃ are known to pose risks to human health; research (Jakab et al., 1996; Kleinman et al., 2000) has also suggested that synergistic toxicological interactions between gaseous and particle pollutants may be important. A number of early particulate matter epidemiology studies, many of which were instrumental in the promulgation of the PM_{2.5} NAAQS, have been criticized (Moolgavkar and Luebeck, 1996; Lipfert and Wyzga, 1997) for not adequately considering the potential confounding effects of gases such as CO and NO₂. Several recent studies (Moolgavkar, 2003; Villeneuve et al., 2003; Klemm et al., 2004) that considered both particulate matter and gaseous pollutants reported that gases were more strongly associated with health effects than was particulate matter. Stieb et al. (2002) conducted a meta-analysis incorporating the results of 109 time-series studies of air pollution and mortality from around the world and concluded that “PM₁₀, CO, NO₂, O₃, and SO₂ were all positively and significantly associated with all-cause mortality.” However, it remains uncertain whether one or several of the particle or gaseous pollutants, or some other unaccounted factor, is responsible for the observed health effects. Pollen and spores are also generally co-pollutants rather than components of PM_{2.5}, because they typically have diameters on the order of 10 μm to 200 μm (U.S. Geological Survey, 2003). These bioaerosols have been epidemiologically associated with respiratory health effects including asthma (Stieb et al., 2000), although the literature contains mixed conclusions regarding the ability of pollen and spores to confound the results of air pollution epidemiology studies (Neas et al., 1996; Rosas et al., 1998). Possible effects of other particle species with diameters greater than 2.5 μm must also be clarified; coarse particles (PM_{10-2.5}) have been significantly associated with mortality in several studies exploring both PM_{2.5} and PM_{10-2.5} (Mar et al., 2000; Ostro et al., 2000). Hence, there is a need to better elucidate the possibility for confounding by gases and other co-pollutants in PM_{2.5} epidemiology studies.

A further complicating factor in establishing the relationship between $PM_{2.5}$ and human health is the fact that concentrations of $PM_{2.5}$ and co-pollutants measured at centrally-located outdoor ambient air monitoring stations, which have typically been used in epidemiological studies, do not represent actual human exposures to these species. Outdoor pollutant concentrations vary spatially to some extent, depending upon factors like meteorology, topography, proximity to sources, etc., such that concentrations outside of a person's home, for example, may differ appreciably from those measured at a centrally located ambient air monitoring site. Moreover, people in modern urban societies spend a majority of their time in indoor microenvironments; hence, their exposure to particles of outdoor origin is dependent upon these particles' ability to penetrate indoors without depositing or chemically reacting. Because ambient concentrations are only proxies of actual air pollutant exposures, their use in epidemiology studies to represent such exposures introduces error, potentially biasing the results of these studies (Zeger et al., 2000; Meng et al., 2005). Studies that have reported associations between ambient $PM_{2.5}$ concentrations and health effects based solely on central site exposure data have often relied on large sample sizes to compensate for the potential bias to the null that can result from exposure misclassification when central site measurements are used to reflect exposures. However, for epidemiologic panel studies, which include a smaller number of participants, more accurate and precise exposure estimates are needed to provide sufficient power to examine the associations between ambient particle concentrations and health effects. Exposures can be estimated more precisely using personal monitoring and/or home-specific assessment of indoor and outdoor concentrations.

Several studies have suggested that the spatial variability of particle concentrations and the relationship between personal particle exposures and ambient particle concentrations vary appreciably by particle component. Studies conducted in Philadelphia, PA, and Washington, D.C., found that summer PM_{10} and $PM_{2.5}$ concentrations were relatively uniform across these urban areas (Burton et al., 1996). Correlations among various sites' summer PM levels were strong in each city, particularly for $PM_{2.5}$, suggesting that the sources of $PM_{2.5}$ in the summer are regional in nature, which is not surprising given the cities' high summer concentrations of SO_4^{2-} , a known regional pollutant. In contrast, in a study conducted in inner-city Boston, MA (Levy et al., 2000), personal exposures to $PM_{2.5}$ and traffic particles varied considerably, with this variation related to proximity to a bus depot and to diesel traffic. These findings are consistent with findings from several other exposure studies, which also suggest that whereas ambient concentrations of regional $PM_{2.5}$ components such as SO_4^{2-} measured at centrally-located air monitoring sites are strong predictors on average of personal exposures for panels of individuals living nearby (Sarnat et al., 2000; Sarnat et al., 2002), the ability of a single centrally-located ambient monitoring site to reflect exposures for locally-generated particle components such as EC is poor (Janssen et al., 1997; Chang et al., 2000; Koutrakis et al., 1992; Tolocka et al., 2001; Sarnat et al., 2001). In light of the aforementioned need for further research to clarify the link between specific chemical components of ambient $PM_{2.5}$ and human health, additional elucidation of the associations between ambient concentrations of these components and human exposures to them is warranted.

Home ventilation conditions are also suggested to have a significant effect on the relationship between ambient $PM_{2.5}$ concentrations and personal exposures to $PM_{2.5}$. In a panel study of senior citizens in Baltimore, MD (Abt et al., 2000), repeated measurements of personal air pollutant exposures, home ventilation, and ambient air pollutant concentrations were used to investigate the influence of home ventilation on the personal-ambient $PM_{2.5}$ association. Results showed that ventilation (as determined using self-reported information about the fraction of time spent indoors with open windows) was a significant predictor of the personal-ambient

PM_{2.5} relationship, which was strongest for individuals spending most of their time in well-ventilated environments and weakest for individuals spending their time in poorly ventilated environments. Further research is needed to confirm and supplement these findings.

Finally, personal exposure measurements of PM_{2.5} and gaseous pollutants can provide valuable information regarding the potential for confounding in air pollution epidemiology studies. For confounding to occur in studies employing central site measurements as proxies of exposure, ambient concentrations of a given pollutant of interest (e.g., PM_{2.5}) must correlate with personal exposures to concentrations of the possible confounders (e.g., gases and other co-pollutants). Sarnat et al. (2001) examined the potential for confounding of particle health effect results by gaseous pollutants in a study of senior citizens living in Baltimore, MD, and found that although ambient PM_{2.5} concentrations were strongly associated with ambient concentrations of several gaseous pollutants, including O₃, only ambient PM_{2.5} concentrations (and not ambient gaseous pollutant concentrations) were strong indicators of respective personal exposures, suggesting that the gaseous pollutants can not confound observed PM_{2.5}-associated health effects for individuals with similar exposure profiles and living in similar urban locations. There is a need for additional assessments of this type to permit an evaluation of the applicability of these Baltimore results to different locations.

Resolution of the numerous PM_{2.5} research needs identified above is essential to ensure that the regulations written and money spent to control fine particulate matter produce the greatest possible benefits to public health and the environment. For example, if the health effects that have been associated with fine particulate matter are caused by specific chemical components of the particles rather than by total PM_{2.5} mass, then money spent to control PM_{2.5} mass by reducing ambient levels of its most prevalent, but benign, constituents may yield little improvement in public health.

Thus, in response to these PM_{2.5} research needs, CONSOL Energy Inc. Research & Development (CONSOL R&D), in collaboration with the funding agencies and participating groups listed in Table 1, launched the Steubenville Comprehensive Air Monitoring Program (SCAMP) in 1999. As its name suggests, SCAMP focused on characterizing PM_{2.5} and gaseous pollutant concentrations and exposures in Steubenville, OH, a city of ~19,000 people located in a heavily industrialized area along the Ohio River in eastern Ohio. The city is part of the larger Steubenville-Weirton Metropolitan Statistical Area (MSA), which has a population of ~132,000. Despite its small population relative to major U.S. cities, Steubenville is an important location for PM_{2.5} research. Steubenville had the greatest concentrations of PM_{2.5}, total suspended particulate (TSP), inhalable particles, SO₄²⁻, SO₂, and NO₂ of the cities examined by the landmark Harvard Six Cities Study (Dockery et al., 1993), which reported an association between PM_{2.5} and mortality and was one of the “core studies” used by the U.S. EPA to establish the NAAQS for PM_{2.5}. Although ambient PM_{2.5} concentrations in Steubenville have declined appreciably (Connell et al., 2005a) since the Six Cities Study was conducted in the 1970s and 1980s, likely owing to declines in the city’s population (U.S. Census Bureau, 2000) and industrial activity (West Virginia University Bureau of Business and Economic Research, 2001) and advances in the application of air pollution control equipment, Jefferson County, OH, where Steubenville is situated, is a nonattainment area under the PM_{2.5} NAAQS and is one of 18 counties projected by the U.S. EPA to remain in nonattainment in 2015 after the implementation of the Clean Air Interstate Rule (CAIR) (U.S. EPA, 2005a). Steubenville remains an ideal location for examining the impact of coal-fired power plant emissions on ambient PM_{2.5}, as it is situated within ~20 km of two major coal-fired power plants and within 200 km of numerous others located in the Ohio River Valley region. Mass-based PM_{2.5} regulations have particularly large implications for these plants, as sulfate, which results largely

from SO₂ emitted by coal combustion, is the largest contributor to PM_{2.5} mass in many eastern U.S. locations (Frank, 2006). Steubenville also permits an analysis of the effects of iron, steel, and coke facilities on ambient PM_{2.5} concentrations and compositions, as such facilities are located within a few kilometers of the city. Hence, the current assessment of PM_{2.5} concentration, composition, and exposure in Steubenville provided by SCAMP is particularly relevant for aiding implementation plan development and supplementing the results of previous landmark epidemiology studies conducted in that city.

Table 1. SCAMP sponsors and research participants.

Sponsors	Research Participants
U.S. Department of Energy – National Energy Technology Laboratory	CONSOL Energy Inc. Research & Development
Ohio Coal Development Office within the Ohio Air Quality Development Authority	Harvard School of Public Health
Electric Power Research Institute	Ohio University
American Petroleum Institute	Franciscan University of Steubenville
National Mining Association	Saint Vincent College
American Iron and Steel Institute	Wheeling Jesuit University
Edison Electric Institute	Optimal Technologies
National Institute of Environmental Health Sciences	Air Quality Sciences, Inc.
U.S. Environmental Protection Agency	Control Analytics, Inc.
CONSOL Energy Inc.	

SCAMP included two interrelated major components: the ambient air monitoring program and the indoor and personal exposure program. As part of the ambient air monitoring program, which was led by CONSOL R&D, PM_{2.5} and PM₁₀ mass, PM_{2.5} and PM₁₀ composition, gaseous pollutant concentrations, pollen and spore counts, and meteorological conditions were measured from May 2000 through May 2002 at a central outdoor monitoring site in Steubenville. PM_{2.5} mass and composition were also determined during this period at four surrounding satellite sites. As part of the indoor and personal exposure program, which was led by the Harvard School of Public Health (HSPH) under subcontract to CONSOL R&D, personal, indoor, and outdoor exposures to PM_{2.5}, PM_{2.5} components, and gaseous pollutants were characterized during the summer and fall of 2000 for a panel of older adults and during the winter and summer of 2001 for a panel of children living in Steubenville. The ambient air monitoring program and the outdoor monitoring portion of the indoor and personal exposure program were funded by the U.S. Department of Energy’s National Energy Technology Laboratory (DOE-NETL) and CONSOL Energy Inc. The indoor and personal air sampling portions of the indoor and personal exposure program were funded by the consortium of non-DOE sources listed in Table 1, led by the Ohio Coal Development Office (OCDO) within the Ohio Air Quality Development Authority (OAQDA).

The SCAMP ambient air monitoring and indoor and personal exposure programs were designed to provide information to help resolve several of the specific PM_{2.5} research needs identified in the discussion above, including the need for better estimates of the relationship between

ambient concentrations of PM_{2.5} and its components and actual human exposures to these species, the need for better characterization of the potential for confounding by gaseous pollutants in PM_{2.5} epidemiology studies, the need for better characterization of individual chemical components of PM_{2.5}, and the need for data for development of SIPs in response to the PM_{2.5} NAAQS. In addition to its two major components, SCAMP included several smaller studies designed to validate personal sampling methodologies, develop more sensitive analytical methods for determining trace elements in PM_{2.5}, and examine associations between pollutant concentrations and heart rate variability for the older adult panel. Major objectives of the overall project were as follows:

1. To measure PM_{2.5}, PM₁₀, and co-pollutant concentrations and collect hourly meteorological data and daily pollen and spore counts at a central urban site in Steubenville,
2. To measure PM_{2.5} concentrations at four remote sites in the Steubenville region,
3. To measure PM_{2.5} and co-pollutant concentrations outside the homes of participating individuals in Steubenville,
4. To measure indoor PM_{2.5} and co-pollutant concentrations in the living environment of participating individuals in Steubenville,
5. To measure PM_{2.5} and co-pollutant concentrations in the “personal” environment of participating individuals in Steubenville,
6. To characterize the chemical and physical nature of indoor, outdoor, and personal PM_{2.5},
7. To evaluate the relationships between ambient concentrations and personal exposure to PM_{2.5} and co-pollutants,
8. To evaluate the relationships among indoor, outdoor, and personal PM_{2.5} concentrations and indoor, outdoor, and personal co-pollutants,
9. To compare PM_{2.5} collected at urban and remote sites,
10. To determine whether heart rate variability correlates with any or all of the pollutants monitored, and
11. To provide a comprehensive database for use in epidemiological studies, long-range transport studies, and SIP development.

Results from the SCAMP ambient air monitoring program, the SCAMP indoor and personal exposure program, and the various smaller studies conducted as part of the project have been disseminated via a number of publications and presentations, which are listed in the Bibliography included with this report. This final report integrates and summarizes these results and discusses their implications for PM_{2.5} research and policymaking.

2. EXECUTIVE SUMMARY

The Steubenville Comprehensive Air Monitoring Program (SCAMP) was conducted to clarify uncertainties regarding fine particulate matter (PM_{2.5}) concentration, composition, and exposure. As part of the program, PM_{2.5}, PM_{2.5} components, co-pollutants, and meteorological conditions were monitored from May 2000 through May 2002 at a central ambient air monitoring station in Steubenville, OH, and PM_{2.5} and its components were simultaneously monitored at four comparatively remote satellite sites in the surrounding region. Additionally, personal, indoor, and outdoor exposures to PM_{2.5}, PM_{2.5} components, and gaseous pollutants were characterized during the summer and fall of 2000 for a panel of older adults and during the winter and summer of 2001 for a panel of children living in Steubenville. The program also included the successful development of a method for determining PM_{2.5} trace elements by dynamic reaction cell ICP-MS, a study to validate personal sampling methodologies, and a concurrent evaluation of the effects of PM_{2.5} and gases on the cardiac health of the older adult panel. SCAMP succeeded in generating a powerful database of information concerning ambient, outdoor, indoor, and personal-breathing-space concentrations of PM_{2.5}, its components, and co-pollutants in Steubenville and in helping to address a number of PM_{2.5} research needs, including the need for better estimates of the relationship between ambient concentrations of PM_{2.5} and its components and actual human exposures to these species, the need for better characterization of the potential for confounding by gaseous pollutants in PM_{2.5} epidemiology studies, the need for better characterization of individual chemical components of PM_{2.5}, and the need for data for development of State Implementation Plans (SIPs) in response to the PM_{2.5} National Ambient Air Quality Standards (NAAQS). Major conclusions of the program are as follows:

- Average concentrations of PM_{2.5}, as well as those of many PM_{2.5} components and gaseous pollutants, in Steubenville have decreased appreciably since the Harvard Six Cities Study was conducted in the late 1970s and 1980s, but remain elevated relative to the annual PM_{2.5} NAAQS. The mean PM_{2.5} concentration measured during SCAMP (18.4 µg/m³) was 40% less than that measured during the Six Cities Study, but remained more than 3 µg/m³ above the PM_{2.5} NAAQS. Hence, further reductions in PM_{2.5} concentrations in Steubenville will be required; data collected as part of SCAMP are particularly relevant for informing the development of an appropriate PM_{2.5} implementation plan for Steubenville.
- Secondary-sulfate-dominated regional PM_{2.5}, likely resulting from coal-fired power plants to the west and southwest of Steubenville, is the largest contributor to ambient PM_{2.5} in Steubenville, accounting for about 40% of the total mass. Primary PM_{2.5} emissions from coal-fired power plants, on the other hand, appear to contribute relatively little (i.e., < 5%) to the total mass of ambient PM_{2.5}. Hence, whereas reductions in secondary sulfates originating from coal-fired power plants (as required under the Clean Air Interstate Rule) will almost certainly be needed if Steubenville is to attain compliance with the PM_{2.5} NAAQS, further reductions in primary particulate emissions from coal-fired power plants would be of relatively little aid in attaining a mass-based standard.
- Local sources in the immediate Steubenville vicinity, including motor vehicles, iron and steel facilities, and possibly other point and area sources, contribute an estimated 5 µg/m³ (~30%) to the total mass of ambient PM_{2.5} in Steubenville.
- Distinct mechanisms cause local- and regional-scale episodes of elevated PM_{2.5} concentrations in Steubenville. Regional-scale episodes typically occurred during warmer periods of the year and were often characterized by multiple-day periods of consistently and moderately elevated concentrations of secondary-sulfate-dominated PM_{2.5}. Conversely, local-scale episodes typically occurred during cooler portions of the year and were characterized by shorter overnight periods of more acutely elevated concentrations of PM_{2.5} resulting from the buildup of primary particles enriched in trace metals and carbonaceous species during nocturnal temperature inversions. Both of the PM_{2.5} episode types identified during SCAMP must be considered when developing an implementation plan to reduce concentrations of PM_{2.5} in Steubenville.
- Concentrations of certain transition metals, including Fe, Mn, and Zn, observed at Steubenville are appreciably elevated relative to concentrations observed in the surrounding region and in other larger U.S. cities. Further research should be conducted to better elucidate whether these metals play any role in the health effects previously associated with PM_{2.5} in Steubenville.

- Ambient fine particle concentration is a strong proxy of corresponding personal exposure for older adults and, to a lesser extent, for children in Steubenville. Mixed model results indicated that, for every 10 $\mu\text{g}/\text{m}^3$ increase in corresponding ambient $\text{PM}_{2.5}$ concentrations, personal exposures of the older adults increased an average of $7.3 \pm 0.5 \mu\text{g}/\text{m}^3$ during the summer and $6.3 \pm 0.5 \mu\text{g}/\text{m}^3$ during the fall. For the children, for every 10 $\mu\text{g}/\text{m}^3$ increase in corresponding ambient concentrations, personal $\text{PM}_{2.5}$ exposures increased an average of 5.8 ± 0.8 during the summer and $2.5 \pm 1.4 \mu\text{g}/\text{m}^3$ during the winter. Overall, results of SCAMP confirm the results of previous studies showing that ambient $\text{PM}_{2.5}$ concentrations are strong proxies of corresponding personal exposures, especially for older adults.
- The ability of ambient fine particle concentrations to reflect personal exposures differs by particle component, and is greater for SO_4^{2-} , a predominantly regional fine particle, than for elemental carbon (EC), a predominantly locally-emitted fine particle. Personal-ambient associations for SO_4^{2-} , like those for $\text{PM}_{2.5}$, were stronger in summer than in fall or winter, whereas personal-ambient associations for EC followed an opposite seasonal pattern. Again, personal-ambient associations were stronger for the older adults than for the children. Findings are consistent with those of previous studies and provide additional justification for the use of ambient SO_4^{2-} , and to a lesser extent EC, to represent corresponding mean personal exposures in epidemiologic analyses. Future exposure studies should examine factors contributing to the observed differences in pollutant exposures for older adults and children. Further improvements in methods for measuring personal exposures to $\text{PM}_{2.5}$ components (and to gases such as SO_2) are also needed, particularly for trace elements, which often exhibited poor precision or could not be quantified in personal exposure samples during SCAMP, and for organic carbon, which was the second largest component of ambient $\text{PM}_{2.5}$ mass in Steubenville during SCAMP but was not measured in personal exposure samples because of methodological limitations.
- Home ventilation was the primary factor affecting the ability of ambient pollutant concentrations to reflect the personal exposures of older adults and children to these pollutants. For both cohorts, slopes describing the increase in indoor or personal pollutant concentrations for a corresponding increase in ambient concentrations were generally greater when windows were open or when participants spent more time near open windows. Results from SCAMP are the first to show the impact of home ventilation as a modifier of personal-ambient gas associations, and suggest that ventilation may be an important modifier of the magnitude of effect in time-series health studies.
- Time-series health studies based on 24-hour ambient concentrations may not be able to separate the independent effects of particles and gases in Steubenville. Analyses of ambient air monitoring data from the central site suggest that gaseous pollutants such as CO, NO_2 , and to a lesser extent O_3 , may cause seasonally-dependent collinearity problems in $\text{PM}_{2.5}$ time series epidemiology studies focusing on Steubenville. Furthermore, comparisons of central site data with personal exposure data indicate statistically significant associations between ambient particle concentrations and personal O_3 and NO_2 exposures, suggesting that the effects of these pollutants may be difficult to differentiate. It is important to note, however, that ambient particle concentrations were generally better proxies of exposures to the particles themselves rather than the gases. Nevertheless, future epidemiology studies focusing on $\text{PM}_{2.5}$ should carefully consider the possibility for confounding by gaseous co-pollutants such as CO, NO_2 , and O_3 , as well as the possibility that the associations among and effects of these pollutants may be significantly modified by season or by home ventilation status. Previous studies that did not adequately consider these possibilities should be interpreted cautiously.
- Health effect studies conducted using $\text{PM}_{2.5}$ and gaseous pollutant data collected at the SCAMP central monitoring site in Steubenville found that fine particulate sulfate, but not elemental carbon, was significantly associated with decreased heart rate variability and increased odds of supraventricular arrhythmia among members of the older adult cohort. Although these studies were based on a fairly small cohort and did not consider all major chemical components of $\text{PM}_{2.5}$ in Steubenville (e.g., organic carbon, the second largest contributor to $\text{PM}_{2.5}$ mass in Steubenville, was not included in the analyses), they provide evidence that increased levels of ambient fine particulate pollution from non-traffic sources (e.g., sulfate particles) may adversely affect cardiac health in the elderly.

3. EXPERIMENTAL

As discussed in the introduction, the Steubenville Comprehensive Air Monitoring Program included two major complementary components: the ambient air monitoring program and the indoor and personal exposure program. As part of the ambient air monitoring program, which was designed to permit a characterization of $PM_{2.5}$ concentration and composition in Steubenville, OH, and the surrounding region and an assessment of factors that affect $PM_{2.5}$ in Steubenville, $PM_{2.5}$, $PM_{2.5}$ speciation, PM_{10} , PM_{10} speciation, gaseous pollutants, meteorological conditions, and pollen and spores were measured at a central ambient air monitoring site in Steubenville between May 2000 and May 2002, and $PM_{2.5}$ and $PM_{2.5}$ speciation were simultaneously measured at four satellite sites in Ohio, Pennsylvania, and West Virginia. As part of the indoor and personal exposure program, which was designed to assess how indoor and personal exposures to $PM_{2.5}$, $PM_{2.5}$ components, and gaseous pollutants relate to outdoor concentrations of these species in Steubenville for potentially sensitive populations, personal, indoor, and outdoor exposures to $PM_{2.5}$, $PM_{2.5}$ components, and gaseous pollutants were characterized for a panel of older adults and a panel of children living in Steubenville. Each panel was monitored in two distinct seasons during 2000-2001. This section summarizes the experimental methods employed by the SCAMP ambient air monitoring and indoor and personal exposure programs.

SCAMP also included several smaller components designed to supplement the ambient air monitoring and indoor and personal exposure programs. These included a study focused on the development and validation of sampling and analytical methodologies for measuring personal, indoor, and outdoor concentrations of $PM_{2.5}$ and its components, as well as a study exploring associations between pollutant concentrations and heart rate variability for the older adult panel. Summaries of these smaller components of SCAMP, including both methods and key findings, are provided in the Results section of this report.

3.1 Ambient Air Monitoring Program

3.1.1 Study Design

Figure 1 shows the locations of the five sites constituting the SCAMP ambient air monitoring network. The central ambient air monitoring station in Steubenville was located on the campus of Franciscan University of Steubenville ~ 500 m northeast of where the Harvard Six Cities Study central monitoring site in Steubenville had been located. The SCAMP site was situated atop a bluff overlooking the western bank of the Ohio River from a height of ~ 110 m. The river runs from approximately due north to due south in the Steubenville vicinity; the monitoring site was located within a few kilometers of several major industrial facilities and vehicular thoroughfares located along the river. A long-term parking lot near the monitoring station was usually full but experienced only infrequent traffic. Hence, the SCAMP central site was expected to capture the effect of local source emissions in the immediate Steubenville vicinity on ambient air $PM_{2.5}$ and co-pollutant concentrations in the city, without being unduly influenced by any one particular source.

The four satellite monitoring sites were positioned to provide information about $PM_{2.5}$ concentrations and compositions observed in less-industrialized areas surrounding Steubenville for comparison with those observed in Steubenville, and to permit an assessment of the effects of distance and direction on $PM_{2.5}$ concentration and composition in the Steubenville region. Relative to Steubenville, the satellite sites, which were situated approximately at the cardinal compass points around Steubenville, were located about 19 km to the north (N) in Tomlinson

Run State Park near New Manchester, WV; 33 km to the south (S) on the campus of Wheeling Jesuit University in Wheeling, WV; 108 km to the east (E) on the campus of Saint Vincent College in Latrobe, PA; and 22 km to the west (W) in Hopedale, OH. Among the satellite sites, the southern site was most similar to the Steubenville site, as it was located ~3 km from the center of Wheeling, a city of ~30,000 people located along the Ohio River.

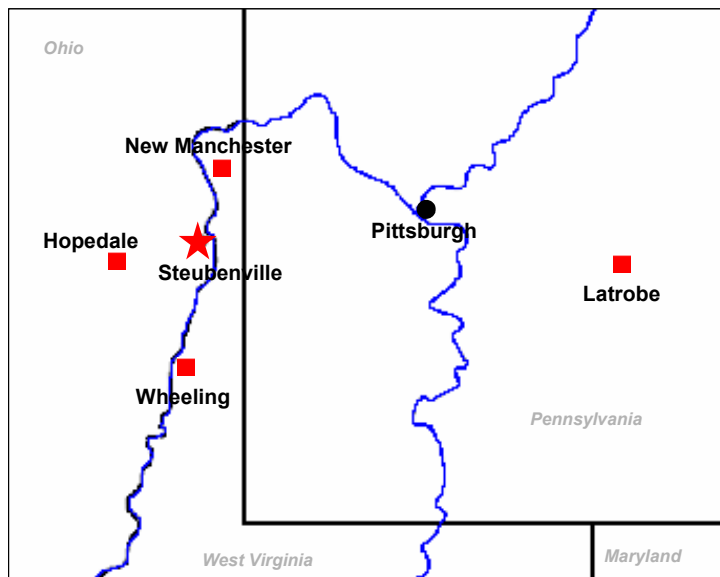


Figure 1. SCAMP ambient air monitoring sites.

Table 2 summarizes the sampling regimen followed by the SCAMP ambient air monitoring sites. The central monitoring station in Steubenville served as the ambient air monitoring “supersite” for the program. There, 24-hr average $PM_{2.5}$ mass concentrations were determined daily using the Federal Reference Method (FRM) for $PM_{2.5}$, and $PM_{2.5}$ component concentrations were determined from 24-hr integrated samples collected every fourth day using the FRM monitor and a speciation sampler. The measured components of $PM_{2.5}$ included sulfate (SO_4^{2-}), nitrate (NO_3^-), chloride (Cl^-), ammonium (NH_4^+), elemental carbon (EC), organic carbon (OC), a suite of elements in the water-extractable $PM_{2.5}$ fraction, and a suite of elements in the acid-digestible $PM_{2.5}$ fraction. Twenty-four-hour average PM_{10} concentrations were also determined each day at the central Steubenville site using the FRM for PM_{10} ; the PM_{10} samples collected every fourth day were further analyzed to determine concentrations of SO_4^{2-} , NO_3^- , Cl^- , NH_4^+ , and a suite of water-extractable elements. In addition to these 24-hr integrated $PM_{2.5}$ and PM_{10} measures, hourly $PM_{2.5}$ mass concentrations were measured using a tapered element oscillating microbalance (TEOM), daily pollen and spore counts were determined from samples collected using a Burkard volumetric spore trap, and hourly gas (SO_2 , O_3 , NO_x , and CO) concentrations and meteorological conditions were monitored.

At each of the four satellite sites, 24-hr average $PM_{2.5}$ mass concentrations were determined daily using the FRM for $PM_{2.5}$, and water extracts from the FRM $PM_{2.5}$ samples collected every fourth day were analyzed for ions (SO_4^{2-} , NO_3^- , Cl^- , NH_4^+) and a suite of elements.

For all five sites, the 24-hr sampling period for a given day occurred from ~9:00 a.m. Eastern Standard Time (EST) on that day until ~9:00 a.m. EST on the following day in order to be consistent with the sampling schedule followed by the SCAMP indoor and personal exposure program. (On every third day from June 30, 2001 – August 2, 2001 and from January 8, 2002 –

Table 2. Sampling regimen for SCAMP ambient air monitoring sites.

Variable	Locations ^a	Sampling Period	Frequency (days ⁻¹)	Resolution (hours)	Sampling Method ^b	Analytical Method
PM _{2.5}	ST, N, S, E, W	5/13/00-5/14/02	1/1	24	PM _{2.5} FRM Sampler, Teflon Filter	Gravimetry (FRM)
Ions (SO ₄ ²⁻ , NO ₃ ⁻ , Cl ⁻ , NH ₄ ⁺)	ST, N, S, E, W	5/16/00-5/14/02	1/4	24	PM _{2.5} FRM Sampler, Teflon Filter	Ion Chromatography
Carbon (EC, OC)	ST	8/12/00-5/14/02	1/4	24	PM _{2.5} Speciation Sampler, Quartz Filter	Thermal Optical Transmittance
Water-Extractable Elements	ST, N, S, E, W	5/16/00-5/14/02	1/4	24	PM _{2.5} FRM Sampler, Teflon Filter	Dynamic Reaction Cell ICP-MS
Acid-Digestible Elements	ST	8/12/00-5/14/02	1/4	24	PM _{2.5} Speciation Sampler, Teflon Filter	Dynamic Reaction Cell ICP-MS
PM ₁₀	ST	5/13/00-5/14/02	1/1	24	PM ₁₀ FRM Sampler, Teflon Filter	Gravimetry
Ions (SO ₄ ²⁻ , NO ₃ ⁻ , Cl ⁻ , NH ₄ ⁺)	ST	5/16/00-5/14/02	1/4	24	PM ₁₀ FRM Sampler, Teflon Filter	Ion Chromatography
Water-Extractable Elements	ST	5/16/00-5/14/02	1/4	24	PM ₁₀ FRM Sampler, Teflon Filter	Dynamic Reaction Cell ICP-MS
PM _{2.5}	ST	6/14/00-5/14/02	1/1	1, 24 ^d	TEOM	N/A
SO ₂	ST	5/16/00-5/14/02	1/1	1, 24 ^d	FEM, UV Fluorescence	N/A
NO/NO ₂ /NO _x	ST	5/16/00-5/14/02	1/1	1, 24 ^d	FRM, Chemiluminescence	N/A
CO	ST	5/16/00-5/14/02	1/1	1, 24 ^d	FRM, Gas Filter Correlation	N/A
O ₃	ST	5/16/00-5/14/02	1/1	1, 24 ^d	FEM, UV	N/A
Weather Conditions ^c	ST	5/16/00-5/14/02	1/1	1, 24 ^d	10-meter Meteorological Tower	N/A
Pollen and Spores	ST	6/21/00-5/13/02	1/1	24	7-Day Burkard Volumetric Spore Trap	Optical Microscopy

^aST is Steubenville; N, S, E, and W are the northern, southern, eastern, and western satellite sites, respectively. ^bFRM = Federal Reference Method, FEM = Federal Equivalent Method. ^cIncludes wind speed, temperature, relative humidity, solar radiation, rainfall, and barometric pressure. ^dMonitored continuously; 1-hr and 24-hr averages were computed.

January 20, 2002, the PM_{2.5} speciation sampler at the central Steubenville site was operated from 12:00 a.m. EST to 12:00 a.m. EST as part of a sampling program being conducted by the U.S. EPA. When these EPA sampling days fell on regularly scheduled SCAMP speciation sampling days, the EPA sampling schedule was followed for the speciation sampler. Speciation sampler results from these EPA sampling days are not included in the analyses presented in this report).

3.1.2 Sampling and Analytical Methods

3.1.2.1 Integrated PM_{2.5} Mass

Twenty-four-hour average PM_{2.5} mass concentrations were determined daily at the central Steubenville site and each of the four satellite sites according to the Federal Reference Method for PM_{2.5}. Each 24-hr integrated PM_{2.5} sample was collected on a 47-mm Teflon filter using a Thermo Andersen PM_{2.5} Sequential Sampler (RAAS2.5-300, U.S. EPA Designation RFPS-0598-120) operating at a flow rate of ~16.67 L/min. Filters were conditioned before and after exposure, and the mass of sampled PM_{2.5} was determined gravimetrically using a Cahn C-33 microbalance in a clean weighing room that was controlled for temperature, humidity, and vibration. Daily average concentrations were determined based upon the mass of sample collected and the volume of air sampled.

3.1.2.2 PM_{2.5} Water-Extractable Ions and Elements

The 24-hr integrated PM_{2.5} FRM samples from every fourth day at each of the five SCAMP monitoring sites were submitted for ionic and water-extractable elemental analyses. The Teflon filters containing these samples were transferred into rinsed polypropylene digestion tubes using Teflon-coated forceps and covered with 25 mL of 18-MΩ deionized water containing 0.2% isopropanol as a wetting agent. The tubes were then sonicated for 30 minutes, and the resulting leachate from each sample was divided into separate aliquots for anionic, cationic, and elemental analyses.

3.1.2.2.1 Ions

SO₄²⁻, NO₃⁻, Cl⁻, and NH₄⁺ were determined by ion chromatography (IC). Anionic species were separated by an anion exchange resin using a sodium carbonate-sodium bicarbonate eluent in a Dionex DX-100 ion chromatograph. Ammonium was separated by a cation exchange resin using a nitric acid-EDTA eluent in a Waters ion chromatograph. Both instruments use a conductivity meter to detect the concentrations of the ionic species in solution, which are then converted to 24-hr average ambient air values. Analytical detection limits, expressed in units of µg/m³ of sampled ambient air for ease of comparison with the ambient air concentration data presented later, were approximately 0.02 µg/m³ for Cl⁻, 0.06 µg/m³ for NO₃⁻, and 0.1 µg/m³ for NH₄⁺ and SO₄²⁻.

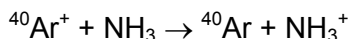
3.1.2.2.2 Water-Extractable Elements

Concentrations of 21 water-extractable elemental components of PM_{2.5} (Al, As, Ba, Ca, Cd, Co, Cr, Cu, Fe, K, Mg, Mn, Na, Ni, Pb, Se, Si, Sn, Ti, V, Zn) were determined by dynamic reaction cell (DRC) inductively coupled plasma-mass spectrometry (ICP-MS). Prior to analysis, an internal standard solution containing Ge and In was added to each sample so that instrument drift could be monitored and corrected.

All elemental determinations were performed using a PerkinElmer ELAN 6100 DRC ICP-MS. Aside from the DRC, the instrument is a conventional quadrupole ICP-MS. The liquid sample to be analyzed is nebulized and passed through an argon plasma, which converts the analyte atoms into primarily singly charged ions. These ions are focused by a lens system and fed to a quadrupole mass spectrometer, which filters the ions according to their mass-to-charge ratio (m/z), allowing only those with a specific m/z to pass through. A discrete dynode detector is used to count the ions passing through the quadrupole.

Interferences limit the ability of conventional ICP-MS to determine certain isotopes (e.g., ^{39}K , ^{40}Ca , ^{51}V , ^{52}Cr , ^{56}Fe , ^{75}As , and ^{80}Se) that are of interest to $\text{PM}_{2.5}$ researchers. These interferences may result from isobaric species (e.g., $^{40}\text{Ar}^+$, which interferes with the determination of ^{40}Ca) or polyatomic species (e.g., $^{40}\text{Ar}^{16}\text{O}^+$, which interferes with the determination of ^{56}Fe) that have the same m/z as the desired analyte. Certain sample matrices can increase the potential for interferences. For example, if the sample contains Cl^- , $^{40}\text{Ar}^{35}\text{Cl}^+$ can interfere with the determination of ^{75}As , and $^{35}\text{Cl}^{16}\text{O}^+$ can interfere with the determination of ^{51}V .

The DRC is employed to minimize these interferences. Prior to entering the analyzing quadrupole, ionic species generated in the plasma pass through a reaction cell that also contains an active quadrupole. A gas such as CH_4 or NH_3 that readily reacts with the interfering species but not with the desired analyte is introduced into the reaction cell. The reaction gas converts interfering species to uncharged species or species with masses that differ from that of the analyte. For example, NH_3 reacts with $^{40}\text{Ar}^+$ to form neutral ^{40}Ar according to the following electron transfer reaction, which has a rate constant of $1.7 \times 10^{-9} \text{ cm}^3 \text{ molecule}^{-1} \text{ s}^{-1}$ (Tanner and Baranov, 1999):



The electron transfer reaction between $^{40}\text{Ca}^+$ and NH_3 is comparatively slow, having a rate constant of less than $10^{-13} \text{ cm}^3 \text{ molecule}^{-1} \text{ s}^{-1}$, and therefore does not occur to any appreciable extent. (Tanner and Baranov, 1999). Hence, NH_3 can be used to minimize the $^{40}\text{Ar}^+$ interference on the determination of ^{40}Ca . The quadrupole present in the reaction cell is used to control the m/z characteristics of the ion beam that is sent to the analyzing quadrupole. Species falling outside of a specific mass bandpass window are ejected from the cell, preventing the formation of new interfering species. Hence, the reaction gas and active quadrupole present in the DRC work together to substantially reduce many of the interferences that limit the capabilities of conventional ICP-MS.

As part of SCAMP, methods for applying DRC ICP-MS to the determination of elements in filter-based $\text{PM}_{2.5}$ samples were developed and validated. The results of this method development are described in the Section 4.1.2 of this report. Table 3 outlines the strategy that was ultimately used as part of SCAMP to determine each of the 21 targeted elements in the water-extractable $\text{PM}_{2.5}$ fraction by DRC ICP-MS. As the table indicates, the method included multiple determinations of many elements, including determinations of different isotopes of the element, determinations of the same isotope using different reaction gases (or no reaction gas), and determinations of the same isotope using different flow rates of the same reaction gas. Because of this strategy, multiple detection limits were obtained for several elements. Detection limits shown in Table 3 were converted to units of ng/m^3 of sampled ambient air for ease of comparison with the ambient air concentration data presented later.

Table 3. Strategy and detection limits for the determination of elements in the water-soluble fraction of PM_{2.5} using DRC ICP-MS.

Element	Isotope	Reaction Gas	Detection Limits (ng/m ³ of sampled air) ^a
Al	27	NH ₃	0.1
As	75	CH ₄	0.04
Ba	138	None	0.01
Ca	40	CH ₄	12
	40	NH ₃	3, 1
	44	None	27
Cd	111	None	0.01
	114	None	0.01
Co	59	None	0.002
Cr	52	CH ₄	0.03
	53	CH ₄	2.16
Cu	63	None	0.04
Fe	54	CH ₄	5.2, 1.8
	56	CH ₄	0.2
K	39	CH ₄	69
	39	NH ₃	0.5, 0.4, 0.3
Mg	24	None	0.1
	25	None	0.2
Mn	55	CH ₄	0.8
Na	23	NH ₃	1
	23	None	2
Ni	58	None	0.1
	60	None	0.2
Pb	208	None	0.1
Se	80	CH ₄	0.05
	82	None	0.26
Si	28	NH ₃	25, 7, 6
Sn	118	None	0.021
Ti	48	None	0.07
V	51	CH ₄	0.08
Zn	66	None	0.2
	68	None	0.2

^aSome isotopes were determined under more than one reaction gas flow rate, resulting in multiple detection limits corresponding to the different flow rates.

3.1.2.3 PM_{2.5} Carbonaceous Species

On every fourth day at the central Steubenville site from August 12, 2000, through May 14, 2002, a 24-hr integrated PM_{2.5} sample was collected on a quartz filter using a Thermo Andersen PM_{2.5} speciation sampler (RAAS2.5-400) with the quartz channel operating at a flow rate of 7.5 to 8.5 L/min. No denuder was used in series with the quartz filter. Prior to use, the quartz filters were prepared by firing to 600 °C for 4 hr to remove carbon background. Filters were sealed and refrigerated during periods of storage. Field blanks were used on each sampling day to assess the carbon background experienced during handling and storage. To determine the 24-hr average ambient concentration of elemental and organic carbon, a square of known area was punched from the exposed filter and analyzed using a Sunset Laboratory Thermal Optical

Transmittance (TOT) Analyzer. The instrument used a thermal program similar to that currently being followed by the U.S. EPA Speciation Trends Network (Research Triangle Institute, 2003), but it was modified slightly in accordance with the instrument manufacturer's recommendations. All results were adjusted to account for background carbon by blank subtraction, and the blank-corrected masses were converted to 24-hr average ambient air values by dividing by the volume of air sampled. Detection limits for EC and OC, expressed in ambient air concentration units, were approximately $0.04 \mu\text{g}/\text{m}^3$ and $0.25 \mu\text{g}/\text{m}^3$, respectively.

3.1.2.4 PM_{2.5} Acid-Digestible Elements

The PM_{2.5} speciation sampler at the central Steubenville site was also used to collect 24-hr integrated PM_{2.5} samples for acid-digestible elemental analysis on a 1-in-4 day frequency from August 12, 2000, through May 14, 2002. All acid-digestible elemental analyses were performed using DRC ICP-MS. To facilitate DRC ICP-MS method development and allow for possible reanalyses required during this process, PM_{2.5} samples were collected in duplicate on Teflon filters using separate channels of the speciation sampler. Both channels operated at volumetric sampling rates of about 16.67 L/min.

Filters were digested and analyzed via DRC ICP-MS in two sequences. All filters from the primary speciation sampler channel were digested and analyzed during the first sequence. Because of several method-related issues encountered during this first succession of analyses, filters from the secondary speciation sampler channel were digested and analyzed in a second sequence in order to improve valid data capture. In cases where valid results for an element were obtained from both the primary filter and the secondary filter for a given day, the results were arithmetically averaged, and this average concentration value is reported. If a valid result was obtained from only one of the two filters collected on a given day, this result is likewise reported.

To prepare the PM_{2.5} samples for elemental analysis, each exposed filter was placed in a cleaned, disposable polypropylene vessel and digested for 16-18 hr in a dilute mixture of HNO₃, HF, and HCl. Although it is hazardous and requires careful handling, HF is required to allow more complete digestion of elements contained in the siliceous matrices commonly encountered in PM_{2.5}. After cooling, a boric acid solution was added to each digestate to complex the HF, and an internal standard solution containing Ge and In was added so that instrument drift could be monitored and corrected. (Isopropyl alcohol, a wetting agent, was also added to each digestate during the second sequence of analyses). The digestates were then heated for an additional hour before submittal for elemental analysis.

All acid-digestible elemental determinations during SCAMP were performed using a PerkinElmer ELAN 6100 DRC ICP-MS. The method was designed to determine 21 elements (i.e., Al, As, Ba, Ca, Cd, Co, Cr, Cu, Fe, K, Mg, Mn, Na, Ni, Pb, Se, Si, Sn, Ti, V, Zn) that are of interest to PM_{2.5} researchers. However, two of the 21 elements, Si and Na, routinely failed to meet quality control criteria and will not be discussed further in this report. Table 4 shows the remaining 19 elements and outlines the general strategy used to determine each of them by DRC ICP-MS.

As the table indicates, the method included multiple determinations of many elements, including determinations of different isotopes of the element, determinations of the same isotope using different reaction gases (or no reaction gas), and determinations of the same isotope using different flow rates of the same reaction gas. Because of this strategy, multiple analytical detection limits were obtained for most elements. Detection limits shown in Table 4 have been

converted to units of ng/m³ of sampled ambient air for ease of comparison with the ambient air concentration data presented later.

Table 4. Strategy and detection limits for the determination of elements in the acid-digestible fraction of PM_{2.5} using DRC ICP-MS.

Element	Isotopes	Reaction Gases	Detection Limits (ng/m ³ of sampled air) ^a
Al	27	NH ₃	13.5, 15.1, 15.9
As	75	CH ₄ , NH ₃ , None	0.19, 0.34, 0.37, 0.43
Ba	138	None	0.1, 0.6
Ca	40	CH ₄	15
	44	None	92
Cd	111	None	0.03
	114	None	0.07
Co	59	None	0.025, 0.027
Cr	52	CH ₄	0.11
	53	NH ₃	0.38
Cu	63	CH ₄ , NH ₃	0.6, 0.8
	65	CH ₄	0.6
Fe	54	CH ₄	2.8, 3.4
	56	CH ₄ , NH ₃	1.1, 10.5
K	39	CH ₄ , NH ₃	1, 4, 60
Mg	24	None	2
	25	None	2
Mn	55	CH ₄ , NH ₃	0.4, 0.7
Ni	58	NH ₃ , None	0.2, 0.3
	60	CH ₄	0.3
Pb	207	None	0.2
	208	None	0.1, 0.2
Se	78	CH ₄	0.17
	80	CH ₄	0.09
Sn	118	None	0.073
	120	None	0.058, 0.283
Ti	48	None	1.48, 4.23
V	51	NH ₃	0.14
Zn	64	NH ₃	1.1
	66	None	0.9, 1.4
	68	None	2

^aSome isotopes were determined under more than one reaction gas, reaction gas flow rate, etc., resulting in multiple detection limits corresponding to the different instrument conditions.

3.1.2.5 Integrated PM₁₀ Mass and Composition

Twenty-four-hour average PM₁₀ mass concentrations were determined daily at the central Steubenville site according to the Federal Reference Method for PM₁₀. Each 24-hr integrated PM₁₀ sample was collected on a 47-mm Teflon filter using a Thermo Andersen PM₁₀ Sequential Sampler (RAAS10-300, U.S. EPA Designation RFPS-0699-0132) operating at a flow rate of 16.67 L/min. Filters were conditioned before and after exposure, and the mass of PM₁₀ was determined gravimetrically using a Cahn C-33 microbalance in a clean weighing room that was

controlled for temperature, humidity, and vibration. Daily average ambient air concentrations were determined based upon the mass of sample collected and the volume of air sampled.

The 24-hr integrated PM₁₀ FRM samples from every fourth day were submitted for ionic and water-extractable elemental analyses. Ions (SO₄²⁻, NO₃⁻, Cl⁻, and NH₄⁺) and 21 water-extractable elements (Al, As, Ba, Ca, Cd, Co, Cr, Cu, Fe, K, Mg, Mn, Na, Ni, Pb, Se, Si, Sn, Ti, V, Zn) were determined by ion chromatography and DRC ICP-MS, respectively, according to the same procedures described in Section 3.1.2.2 above for PM_{2.5}.

3.1.2.6 Continuous PM_{2.5}

Ambient PM_{2.5} mass concentrations were measured continuously from June 14, 2000, through May 14, 2002, at the central Steubenville site using a Rupprecht & Patashnick Series 1400a tapered element oscillating microbalance (TEOM) equipped with a sharp cut inlet for PM_{2.5}. The instrument was configured in the standard way to collect PM_{2.5} from a 3.0 L/min sample stream on a Teflon-coated glass fiber filter. The TEOM filter and sample stream were maintained at 50 °C to remove moisture from the sampled air and prevent condensation on the filter. However, operation at this temperature may also cause the loss of some semi-volatile particulate matter (e.g., ammonium nitrate, semi-volatile organic compounds, and particle-bound water), resulting in an underestimation of total PM_{2.5} mass (Allen et al., 1997). The magnitude of this artifact is dependent upon the composition of the sampled PM_{2.5}, and therefore varies with location and time. TEOMs can be equipped with Nafion diffusion dryers for moisture removal and operated at 30 °C to reduce the loss of semi-volatile material; however, this was not common practice at the time of SCAMP. All TEOM data collected during SCAMP were validated against operational performance criteria established by the instrument manufacturer.

3.1.2.7 Pollen and Spores

Pollen and spores were collected daily at the central Steubenville site between June 21, 2000, and May 13, 2002, using a 7-day Burkard volumetric spore trap. Ambient air was sampled at a constant volumetric flow rate, and pollen and spores in the sampled stream impacted onto an adhesive plastic tape, which rotated once at a constant rate during the course of each weekly sampling interval. Exposed tapes were analyzed by a certified environmental microbiology laboratory (Air Quality Sciences, Marietta, GA), which used optical microscopy counting procedures to determine speciated daily average pollen and spore counts.

3.1.2.8 Gaseous Pollutants

Ambient concentrations of gaseous priority pollutants, including SO₂, NO_x, CO, and O₃, were measured continuously using an Advanced Pollution Instruments Model 900 turnkey ambient air sampling system. The system consisted of individual rack-mounted gaseous pollutant analyzers, which were housed in a mobile trailer and integrated into a dedicated sampling manifold. Gaseous pollutants were collected through a typical glass “candy cane” at a sampling height of approximately 3.5 m above ground level. The gas analyzers were all Advanced Pollution Instruments models, and included an automated federal equivalent method (FEM) ultraviolet fluorescence SO₂ analyzer (Model 100, EQSA-0990-077), an automated FRM chemiluminescence NO_x analyzer (Model 200A, RFNA-1194-099), an automated FEM ultraviolet absorption O₃ analyzer (Model 400, EQOA-0992-087), and an automated FRM gas filter correlation CO analyzer (Model 300, RFCA-1093-093). The NO_x analyzer determined concentrations of NO_x and NO; concentrations of NO₂ were computed by difference.

3.1.2.9 Meteorological Conditions

Meteorological conditions, including wind speed (WS), temperature (Temp), relative humidity (RH), solar radiation (Rad), rainfall (Rain), and barometric pressure (BP), were monitored continuously at the central Steubenville site using standard instrumentation (Met One) atop a 10-m tower. Wind direction was also monitored, but no valid data were obtained because of a data logging error.

3.1.3 Quality Assurance and Quality Control

All sampling activities and laboratory analyses conducted as part of the SCAMP ambient air monitoring program were carried out in accordance with the *CONSOL R&D Quality Assurance Guidelines for the Steubenville Comprehensive Air Monitoring Program* (April 2000) and with applicable standard operating procedures (SOPs), as listed in Table 5.

Table 5. Standard Operating Procedures followed by the SCAMP ambient air monitoring program.

Title	Source
SOP for the RAAS FRM PM _{2.5} and PM ₁₀ Samplers	Optimal Technologies
SOP for Downloading Data off of the FRM PM _{2.5} and PM ₁₀ RAAS Samplers	Optimal Technologies
SOP for the RAAS PM _{2.5} Speciation Sampler	Optimal Technologies
SOP for Auditing and Calibrating Flow on the Speciation Sampler	Optimal Technologies
SOP for the R&P TEOM Sampler (Continuous PM)	Optimal Technologies
SOP for Continuous Gas Analyzers	Optimal Technologies
SOP for the Burkard Pollen/Spore Sampler	Optimal Technologies
Guidelines for Determining Calibration	Optimal Technologies
Particulate Sampling Instrumentation Field Quality Assurance Schedule	Optimal Technologies
Real Time Pollutant and Meteorological Instrumentation Field Quality Assurance Schedule	Optimal Technologies
Mass Flow Calibrator Preventative Maintenance Schedule	Optimal Technologies
Zero Air System Preventative Maintenance Schedule	Optimal Technologies
Ozone Analyzer Preventative Maintenance Schedule	Optimal Technologies
NO ₂ and NH ₃ Analyzers Preventative Maintenance Schedule	Optimal Technologies
SO ₂ Analyzer Preventative Maintenance Schedule	Optimal Technologies
CO Analyzer Preventative Maintenance Schedule	Optimal Technologies
Meteorological Tower Maintenance Schedule	Optimal Technologies
Standard Method for Analyzing Teflon Filters for Anions by Ion Chromatography	CONSOL R&D
Standard Method for Analyzing Teflon Filters for Ammonium by Ion Chromatography	CONSOL R&D
Standard Method for Analyzing Teflon Filters for H ₂ O Soluble Elements by DRC ICP-MS	CONSOL R&D
Standard Method for Analyzing Teflon Filters for Total Elements by DRC ICP-MS	CONSOL R&D
Carbon Analysis of Aerosol Filter Samples for the Pittsburgh PM Supersite	Carnegie Mellon University

Site operators were trained in sampling methodologies and quality assurance and quality control (QA/QC) protocols during a seminar hosted by CONSOL R&D in April 2000. Logbooks were maintained for each monitoring site and participating laboratory to document observations, exceptions, maintenance activities, and QA/QC items. CONSOL R&D periodically audited each monitoring site and participating laboratory to assess instrument performance, review logbooks, and ensure compliance with QA/QC guidelines. Moreover, performance audits of FRM and FEM analyzers were periodically conducted by the Ohio EPA during the program. Corrective actions were taken where necessary. An end-of-program audit of all PM samplers, gas analyzers, and meteorological instruments used during the SCAMP ambient air monitoring program was performed by Optimal Technologies in May 2002; no significant instrument performance issues were detected. Gravimetric determinations of PM mass collected at the five SCAMP monitoring sites were performed by laboratories at CONSOL R&D and Ohio University; both of these laboratories demonstrated acceptable performance for gravimetric analysis of PM filters in an inter-laboratory round robin program conducted by the Harvard School of Public Health in 2000 and 2001. In addition, CONSOL R&D, which performed all PM speciation analyses conducted as part of the SCAMP ambient air monitoring program, demonstrated proficiency for performing IC on filter extracts in an external quality assurance program that was conducted by the Harvard School of Public Health in 2000 and 2001, and demonstrated proficiency for performing carbon analyses in a mini round robin carried out with Desert Research Institute in 2002.

Table A.1 in Appendix A summarizes the QA/QC and maintenance schedule followed for ambient pollutant samplers during the SCAMP ambient air monitoring program. All filter-based PM samples were collected in accordance with the guidelines outlined in EPA Quality Assurance Guidance Document 2.12 (U.S. EPA, 1998). The guidelines outlined in 40 CFR Parts 50, 53, and 58 and in the *Guideline on Speciated Particulate Monitoring* (Chow and Watson, 1998) were also followed where applicable. Field blanks were routinely collected for all integrated PM samplers to permit an assessment of background levels of PM and PM components experienced during filter handling, transport, and storage. At each of the five SCAMP ambient air monitoring sites, one field blank was included with every batch of four FRM filters. Moreover, Teflon and quartz field blanks were collected for every day of PM_{2.5} speciation sampling at the central Steubenville site.

All gas measurements were made in accordance with quality assurance guidelines outlined by the U.S. EPA (U.S. EPA, 1994; 40 CFR 58.61). The performance of each analyzer was monitored by taking daily (shortly after midnight) calibration readings for zero and span concentrations of appropriate certified gas standards. Zero and span control charts were used to determine the validity of measured data, and to determine when physical adjustments to the instruments were required. For each gaseous pollutant, prior to data analysis, hourly average concentrations recorded by the system's data logger were corrected to track instrument performance by linear interpolation between adjacent daily calibration readings.

Table A.2 in Appendix A summarizes the laboratory QA/QC procedures that were followed for the gravimetric analysis of filter-based PM_{2.5} and PM₁₀ samples collected during SCAMP. QA/QC measures were also taken to ensure the quality of ion, carbon, and element concentrations determined from filter-based PM samples. Ion chromatographs were calibrated using serially diluted standards prepared from commercially available stock solutions (Environmental Resource Associates). Analyses were only conducted if a linear or quadratic calibration line having an $R^2 > 0.9995$ was obtained and recovery of a serially diluted quality control standard (Environmental Resource Associates) was 100±10%. This quality control

standard was further analyzed every tenth sample; again, recoveries of $100\pm 10\%$ were required. Duplicate analyses were performed on every tenth filter sample extract. A solution blank was also analyzed with each batch of samples to ensure low background concentrations.

For carbon species, prior to each batch of routine analyses, the TOT analyzer's internal calibration was verified using a sucrose solution containing a known amount of total carbon. Recovery was required to be $100\pm 10\%$. Duplicate analyses were performed on every tenth $PM_{2.5}$ sample. In addition, an instrument blank was run with each batch of samples to ensure a low carbon background, and was required to contain $< 0.2 \mu\text{g}$ of carbon.

Prior to each batch of elemental analyses, the DRC ICP-MS was calibrated using a solution blank consisting of deionized water and internal standards plus three solutions prepared from commercially available stock solutions that covered the concentration range of the elements to be determined. Calibrations were verified by analyzing a tenfold dilution of National Institute of Standards and Technology (NIST) Standard Reference Material (SRM) 1640 (natural water) or 1643d (trace elements in water) for water-extractable elemental analyses or an acid-digestion of NIST SRM 1648 (urban particulate) for acid-digestible elemental analyses. Results were considered to be valid only if these SRM determinations fell within pre-established, isotope-specific acceptable ranges around the NIST-certified values. A quality control solution was analyzed after every ten samples to re-verify calibration; recovery of this quality control sample was required to be $100\pm 20\%$.

Prior to data analysis, all measurements were qualified as being "valid", "flagged", or "invalid" based on the validation schedule detailed in Table A.3 in Appendix A and the analytical criteria described above. Invalid data were excluded from data analyses. All valid and flagged data underwent a second level of quality control checks, in which they were systematically examined using descriptive statistics and graphs. Anomalous data points identified via this procedure were investigated using site and instrument logs and data validation notes (for flagged data), and were invalidated or retained based on an assessment of the available evidence. All actions taken as a result of these second-level checks were logged electronically.

3.1.4 Data Analysis

All data collected as part of the SCAMP ambient air monitoring program were analyzed using descriptive statistics, regression and correlation analyses, and graphs. The statistical significance of seasonal, weekday/weekend, and spatial differences in pollutant concentrations was determined using independent sample t-tests, paired t-tests, or their nonparametric equivalents (i.e., Wilcoxon rank sum tests and Wilcoxon signed rank tests, respectively), as discussed below. Speciated $PM_{2.5}$ data were further analyzed using reconstructed fine mass calculations (to compare measured masses of $PM_{2.5}$ chemical components with total measured $PM_{2.5}$ mass), enrichment factors (to compare the observed abundances of elements in ambient $PM_{2.5}$ with average crustal abundances of these elements), and coefficients of divergence (to quantify the degree of spatial homogeneity in $PM_{2.5}$ component concentrations observed across the SCAMP ambient air monitoring sites). In addition, binary recursive partitioning was employed to search for associations between hourly $PM_{2.5}$ concentrations and hourly gaseous pollutant concentrations and meteorological conditions, and source apportionment techniques including Positive Matrix Factorization (PMF), Unmix, and potential source contribution function (PSCF) modeling were used to estimate the contributions of various emission sources to $PM_{2.5}$ concentrations at the Steubenville site. All of these data analysis techniques are described or referenced, as appropriate, in the Results and Discussion section of this report.

Where possible, actual instrument readings for determinations below the analytical limits of detection (LOD) were used in the statistical analyses presented in this paper in order to prevent the introduction of bias (Gilbert, 1987). However, for ionic and carbonaceous species, the laboratory reported only censored data; hence, determinations below the analytical LOD were set equal to half of the LOD value for these species. For analyses requiring 24-hr average values of variables that were monitored continuously (i.e., gaseous pollutants, meteorological conditions, continuous PM_{2.5}), arithmetic means were computed from hourly concentrations recorded by the data logger and were considered to be valid only if at least 19 out of 24 valid hourly average values were available for the 24-hr time period of interest.

When analyzing data from the SCAMP ambient air monitoring program, special attention was given to statistical issues that are relevant to air monitoring time series data, but have generally not been adequately considered in the PM_{2.5} literature. Specifically, statistical tests commonly applied to air monitoring data, including simple linear regression analyses, t-tests, and Pearson correlation analyses, only provide meaningful results if the assumptions on which they are based are reasonably satisfied. Several common characteristics of air monitoring data that can violate these assumptions are as follows:

1. Autocorrelation – Simple linear regression analysis assumes that the error component of the model is *statistically independent (random)*; independent sample t-tests assume that samples are *randomly* drawn from two independent populations, and paired t-tests assume that the paired differences between the two variables under consideration represent a *random* sample. However, time series of PM_{2.5} (and other air pollutants and meteorological conditions) are frequently autocorrelated, meaning that the concentration at a given time, t , can be predicted to some extent by the concentrations observed at previous times (e.g., for daily data, $t-1$ for 1-day lags; $t-7$ for day-of-week trends; $t-365$ for seasonal trends, etc.). Hence, these time series are not statistically random. The autocorrelated nature of air monitoring data is demonstrated in Figure 2. Figure 2a presents an autocorrelation function (ACF) plot (i.e., a plot showing the correlation of a time series with various lags of itself) for the time series of 24-hr average PM_{2.5} mass concentrations measured daily at the central Steubenville site using the FRM monitor, and illustrates that this time series exhibited statistically significant autocorrelation, especially at a 1-day lag. Figure 2b shows an ACF plot for the time series of NO₃⁻ concentrations measured every fourth day at the central Steubenville site, demonstrating that significant autocorrelation can affect even those variables that are measured on a less-than-daily frequency. Figures 2c and 2d, which present ACF plots for the time series of differences between PM_{2.5} concentrations measured at the central Steubenville site and the northern satellite site and the time series of residuals from simple linear regression of PM_{2.5} as a function of CO at the central Steubenville site, respectively, show how the autocorrelated nature of air monitoring data can cause the assumption of randomness to be violated for paired t-tests and simple linear regression analyses.
2. Non-Normal Distributions – Simple linear regression analyses, independent sample t-tests, and paired t-tests assume that residuals, sample means, and paired differences, respectively, are normally distributed. However, air pollutant concentration data typically are not normally distributed. Rather, these data, which physically must be greater than or equal to zero, often exhibit lognormal distributions, as illustrated in Figure 3 for 24-hour average PM_{2.5} concentrations at the central Steubenville site. Deviations from normality are of greatest concern when sample sizes are small (the central limit theorem applies to larger sample sizes).
3. Outliers – Air pollutant time series often contain outliers resulting from abnormally elevated concentrations observed during air pollution episodes. These outliers can

substantially influence the results of linear regression and Pearson correlation analyses. Moreover, t-tests are not robust to outliers, because outliers have an effect on sample means and sample standard deviations.

4. Heteroscedasticity – Linear regression analysis assumes that the error component of the model is homoscedastic (i.e., has a constant variance at all points along the linear relationship). However, air pollutant data frequently fail to meet this assumption. This is exemplified by the fan-like shape of the scatterplot of NO versus CO concentrations shown in Figure 4.

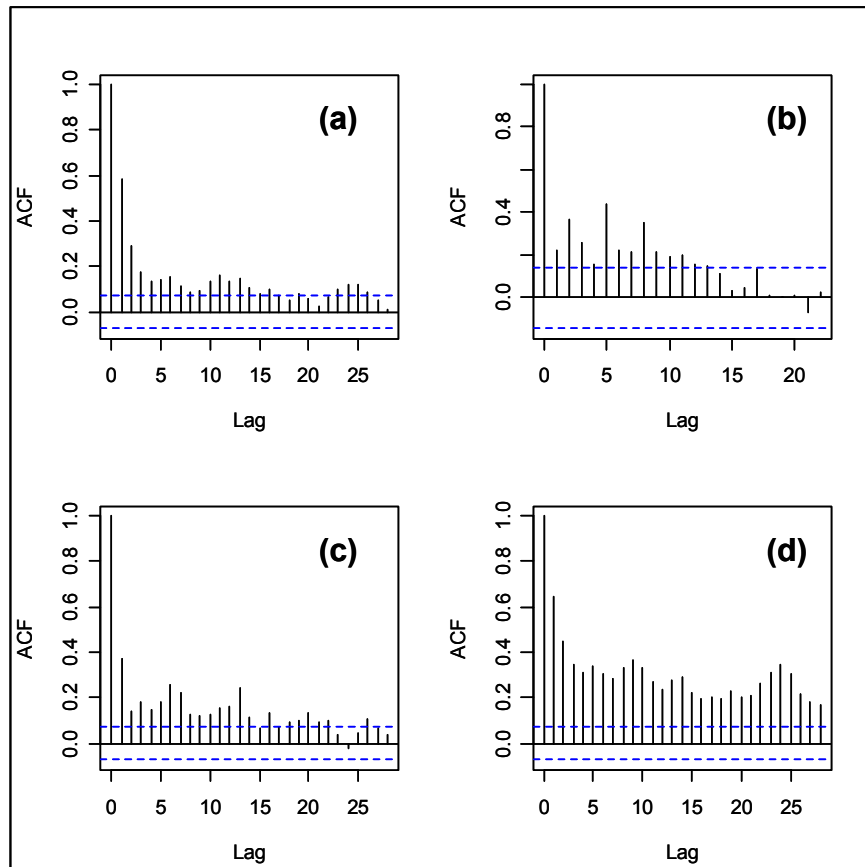


Figure 2. Autocorrelation function (ACF) plots for (a) the time series of 24-hour $PM_{2.5}$ concentrations measured at the central Steubenville site using the FRM monitor, (b) the time series of 24-hour NO_3^- concentrations measured at the central Steubenville site, (c) the time series of paired differences between 24-hour $PM_{2.5}$ concentrations measured at the central Steubenville site and at the northern satellite site, and (d) the time series of residuals from a simple linear regression analysis of 24-hour $PM_{2.5}$ concentrations as a function of 24-hour CO concentrations at the central Steubenville site. Lines extending beyond the dashed blue interval indicate statistically significant autocorrelation. Lags are in units of 1 day for subplots (a), (c), and (d), and in units of 4 days for subplot (b).

The SCAMP ambient air monitoring program included an exploration of the applicability of autoregressive integrated moving average (ARIMA) models (Box and Jenkins, 1976) to time series of daily $PM_{2.5}$, $PM_{2.5}$ ionic and carbonaceous component, gaseous pollutant, and meteorological data as a means for overcoming these statistical limitations of air monitoring data. For select comparisons, prior to performing linear regression analyses or t-tests, data were transformed (e.g., by application of a natural logarithm, square root, fourth root, etc.) to render their distributions more normal, improve homoscedasticity, and reduce the impact of outliers. Autocorrelation was identified by examining ACF and partial autocorrelation function

(PACF) plots, and where statistically significant autocorrelation was discovered, it was modeled by fitting an appropriate ARIMA model. The general form of an ARIMA(p,d,q) model representing a time series, X, of appropriately transformed air pollutant concentrations is:

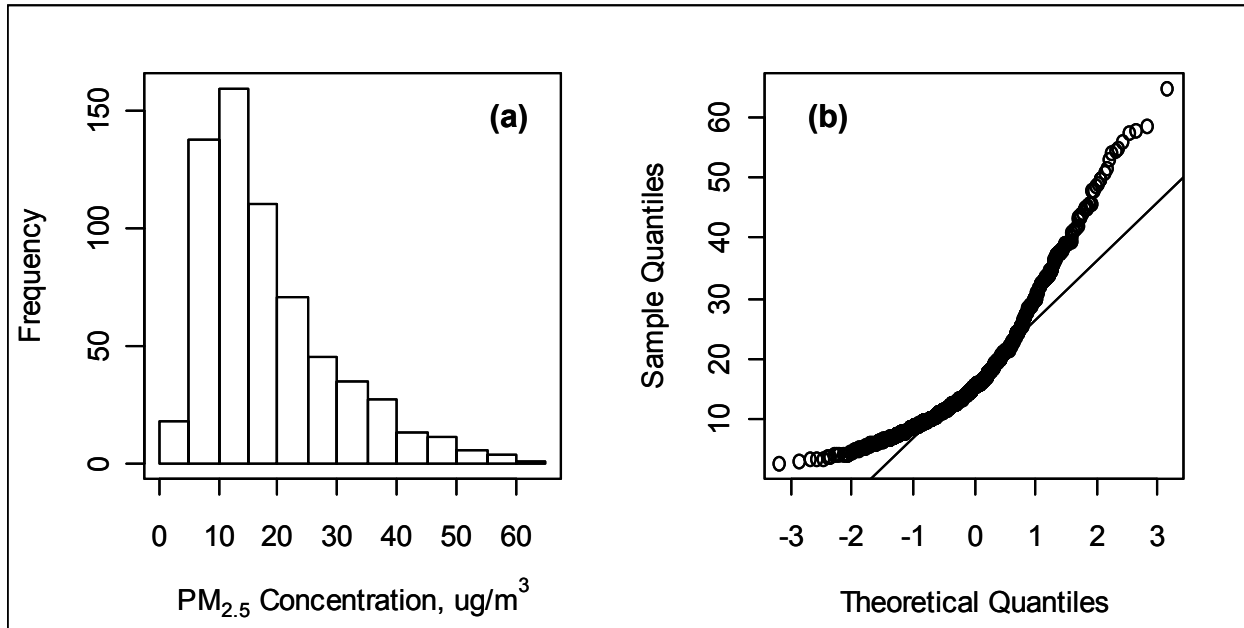


Figure 3. Histogram (a) and quantile-quantile plot (b) showing the distribution of 24-hour average PM_{2.5} concentrations measured at the central Steubenville site during SCAMP using the FRM.

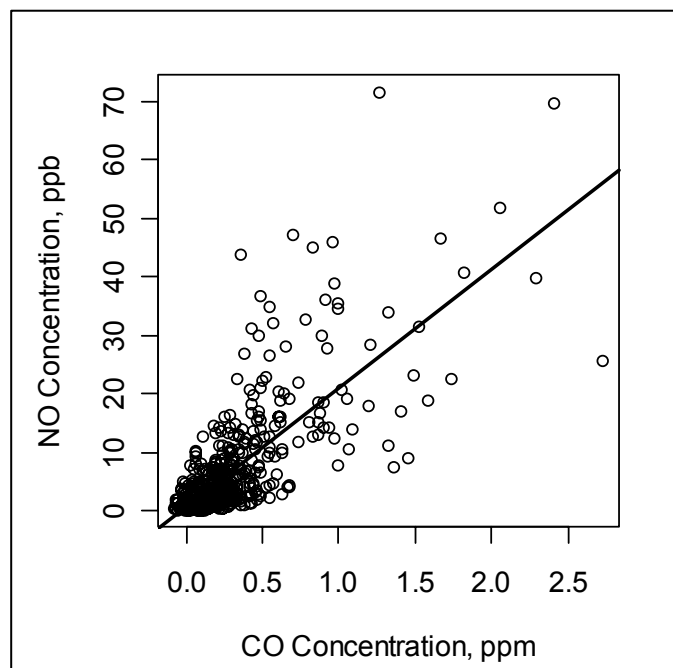


Figure 4. Scatterplot of 24-hr average NO concentrations versus 24-hr average CO concentrations measured at the Central Steubenville site, with linear least squares line.

$$W_t = \eta + \phi_1 W_{t-1} + \phi_2 W_{t-2} + \dots + \phi_p W_{t-p} + \varepsilon_t + \theta_1 \varepsilon_{t-1} + \theta_2 \varepsilon_{t-2} + \dots + \theta_q \varepsilon_{t-q}$$

where

$$W_t = \nabla^d X_t.$$

Here, ∇ is the difference operator and d is the order of differencing such that for $d = 0$, $W_t = X_t$, and for $d = 1$, $W_t = X_t - X_{t-1}$. X_t and ε_t are the transformed air pollutant concentration and random error term, respectively, at time t ; $\phi_1, \phi_2, \dots, \phi_p$ are the autoregressive parameters; $\theta_1, \theta_2, \dots, \theta_q$ are the moving average parameters, and η is the intercept term. Seasonal patterns were removed from the datasets by the inclusion of first-order differencing in the ARIMA model or by subtracting out a 3-month moving average before modeling, when appropriate. The residuals of the ARIMA model (or the transformed pollutant concentration values, for cases in which ARIMA modeling was not required) were then used in the regression analyses and t -tests.

Further details of the ARIMA modeling procedure used in SCAMP and of the results of its application to data from the ambient air monitoring program are reported by Connell et al. (2005a, 2005b). These results are also referenced as appropriate throughout the Results and Discussion section of this report. The authors demonstrated that, although trends observed in SCAMP were generally similar regardless of whether autocorrelation was accounted for, “ p -values can change by several orders of magnitude upon properly accounting for autocorrelation” (Connell et al., 2005a). Because of practical limitations, ARIMA modeling was not incorporated into all of the analyses presented in this report. Hence, these analyses, especially those assessing statistical significance, must be interpreted with some caution. Nevertheless, nonparametric statistics, including Spearman correlation coefficients, Wilcoxon rank sum tests, and Wilcoxon signed rank tests, were used in place of their parametric equivalents where appropriate in a number of the analyses presented in this paper in order to account for skewed distributions, outliers, etc. All other things being equal, such nonparametric tests are generally less powerful (i.e., less likely to reject the null hypothesis when it is false), and hence more conservative, than their parametric counterparts.

3.2 Indoor and Personal Exposure Program

As discussed earlier, the indoor and personal exposure program consisted of two parts focusing on the characterization of exposures to $PM_{2.5}$ and pollutant gases for older adults and children, respectively. The study design was reviewed and approved by the Human Subjects Committees of the Harvard School of Public Health and the Brigham and Women’s Hospital.

3.2.1 Older Adult Cohort

3.2.1.1 Study Design

Exposure monitoring for the older adult cohort was conducted over two 3-month sessions, spanning the summer (June 4 – August 18) and fall (September 25 – December 15) of 2000 in Steubenville, OH. In both seasons, 25 older adults participated in exposure monitoring. These participants also participated in a companion study (described later in Section 4.4 of this report) in which repeated cardiovascular and other health measurements were made.

Air pollutant samples were collected for each participant over 24-hr periods, beginning at 9:00 a.m. (± 2 hours). Indoor samples were collected for all 25 subjects; personal air pollutant

exposures were also collected for 10 of these participants. Each participant was sampled on two consecutive days each week, resulting in up to 24 repeated measures per subject per season. These two weekly sampling days for each participant began on the day prior to the participant's weekly scheduled health measurement. In general, samples were collected for five to ten individuals each day. For the subset of subjects for which personal exposure monitoring was performed, all personal monitoring was conducted simultaneously with monitoring inside of the subjects' homes. Outdoor measurements were additionally collected on up to six days per week on the rooftops of two of the apartment buildings in which many of the subjects resided, and ambient measurements were made at the SCAMP central ambient air monitoring site in Steubenville (described in Section 3.1.1). The overall study area was small, with not greater than 3 km between sampling locations.

3.2.1.2 Participant Recruitment and Selection

The Jefferson County Housing Authority helped identify apartment buildings for our recruitment effort. Access was granted to the J. F. Kennedy (K), Elmer White (E), and Gaylord (G) apartment complexes. Posters, door-to-door solicitation, and group meetings were used to introduce the study and initiate interest. Inclusion criteria for this study included being a non-smoker, over the age of 65, and meeting criteria for the companion health study. Potential participants signed up for screening appointments.

Overall, 53 people were screened for inclusion in the study. Thirty-two of these (29 female, 3 male) were deemed eligible. The majority of the cohort resided in one of three government-subsidized apartment buildings located in Steubenville: Kennedy (n=16), Gaylord (n=5), and Elmer White (n=5). Other participants lived in freestanding houses (n=5) or a townhouse (n=1).

Of these participants, 25 were recruited for the exposure assessment portion of the study. Twenty-two subjects took part in both the summer and fall sessions. Air pollutant sampling was performed inside the homes of all participants; personal air pollutant sampling was also performed for a subset of individuals (n=9 for the summer, n=10 for the fall). Each participant gave written informed consent prior to their participation in the study.

3.2.1.3 Pollutant Sampling

All pollutant samples were collected using Harvard multi-pollutant monitors, which were developed to sample PM_{2.5}, PM_{2.5} components, and gaseous pollutants simultaneously (Demokritou et al., 2001). Each monitor is comprised of an aluminum elutriator designed to hold six individual samplers. The configuration used in Steubenville included four active and two passive badge samplers. The active samplers were outfitted with PM_{2.5} size-selective inlets for measuring fine particles and their components, and included two (duplicate) personal exposure monitors (PEMs) loaded with Teflon filters (37-mm Teflo™, Gelman Sciences) for determination of PM_{2.5} mass by gravimetric analysis and of water-extractable and acid-digestible elements by DRC ICP-MS; a mini-PEM loaded with a 15-mm Fluoropore filter, which was analyzed for SO₄²⁻ by ion chromatography; and a second mini-PEM loaded with a 15-mm quartz filter, which was analyzed for EC using thermal optical transmittance.

Vacuum grease was used as the impaction medium for these four samplers. During sampling, air was pulled through the filters by use of a single vacuum pump located upstream of the inlets; flow restrictors were used to split the inlet air stream into four separate streams with adjustable flow rates. The overall flow rate through the four samplers was 5.2 L/min (1.8 L/min for PEMs

and 0.8 L/min for mini-PEMs) for personal exposure sampling and 9.6 L/min (4.0 L/min for PEMs and 0.8 L/min for mini-PEMs) for indoor and outdoor sampling.

The passive badge samplers were used to measure O₃, SO₂, and NO₂. They were placed in the sides of the elutriator downstream of the PEMs such that the air being pulled through the PEMs moved past the passive badges. This ensured a constant face velocity across the filters to minimize problems that result from variable wind speeds and have previously hindered measurements involving passive methods (Liu et al., 1993; Liu et al., 1994). Each sampler contained a cellulose filter, coated with either nitrite for the collection of O₃ (Koutrakis et al., 1993) or triethanolamine for the collection of NO₂ and SO₂ (Ogawa, 1998).

For indoor and outdoor sampling, the multi-pollutant monitor was affixed to a lab stand (indoors) or a tripod (outdoors). The monitors were connected to Gast pumps running on AC power via amber-colored tubing. For indoor sampling, the set-up was placed in the room where the subject spent most of his or her time, typically the living room, at a height of approximately 1 meter above the floor. The outdoor locations included the rooftops of the Kennedy and Gaylord apartment buildings, as well as the central Steubenville ambient air monitoring site (C). All sampling flow rates were measured before and after sampling with pre-calibrated rotameters (Matheson 603 and 406).

For measuring personal exposures, the multi-pollutant monitor was affixed to the shoulder strap of a small bag used to carry a personal sampling pump (BGI Model 400), battery, and motion sensor. The entire sampling set-up weighed approximately 3 kg. Subjects were asked to wear the sampler over their shoulder for as much time as possible during each monitoring session. Because the personal sampling pump had limited power, the PEMs were sampled at flow rates of 1.8 L/min rather than 4 L/min as used by the indoor and outdoor multi-pollutant monitors. To accommodate this lower flow rate, a specially designed inlet was used. All flow rates were measured in duplicate pre- and post-sampling using a BUCK™ flow meter. Averages of two consecutive measurements within 10 mL/min of each other were used in air volume calculations.

3.2.1.4 Sample Analysis

A chemistry lab at the Franciscan University of Steubenville was used for the assembly, disassembly, cleaning, and temporary storage of filter samples. Gravimetric analyses of Teflon filters were conducted at the Harvard School of Public Health; all other lab analyses were conducted by CONSOL R&D according to the general analytical procedures described in Section 3.1. Teflon filters were analyzed for the following elements in the water-extractable and acid-digestible PM_{2.5} fractions: Al, As, Ba, Ca, Cd, Co, Cr, Cu, Fe, K, Mg, Mn, Na, Ni, Pb, Se, Si, Sn, Ti, V, and Zn.

3.2.1.5 Housing Characteristics Questionnaire

Technicians administered a one-time dwelling questionnaire (included in Appendix B to this report) to study participants prior to the first week of monitoring. This questionnaire asked for information on the location of the apartment, general apartment characteristics, ventilation characteristics (e.g., number of windows, air conditioning units, heating sources, fans), and cooking and fuel characteristics. On days when sampling occurred, an indoor follow-up questionnaire was also given. This follow-up questionnaire was intended to identify indoor pollutant sources and included nine questions on cleaning and cooking; use of candles,

incense, cigars, and cigarettes; and the extent and duration for which doors and windows were open to the outside.

3.2.1.6 Time-Activity Diaries

Subjects who participated in personal exposure monitoring were asked to complete time-activity diaries (TADs) during each 24-hour sampling session. The format for these diaries is provided in Appendix B to this report. Each participant was asked to record information about his or her location, activity, and proximity to pollutant sources in 30-minute time intervals. The diaries were reviewed and verified by field technicians after each sampling period and cross-checked with corresponding motion sensor data.

3.2.1.7 Quality Assurance and Quality Control

Flags were assigned to each sample according to the validity of airflows, sample durations and comments recorded during weigh sessions, assembly/disassembly stages, and field operations. Invalid samples were excluded from all analyses, as were samples (n=10) for subjects who withdrew from the study. Five extreme NO₂ samples were additionally excluded due to presumable lab or field contamination.

The multi-pollutant monitor was configured with two PM_{2.5} samplers; thus, duplicate samples were obtained for each subject-day, and they were used for active sample validation purposes prior to averaging for use in data analyses. A precision criterion of ≤ 0.50 relative difference (RD) between the duplicates was established above which both PM_{2.5} samples and any corresponding SO₄²⁻ or EC samples were invalidated (samples with >0.50 RD were assumed to have been adversely affected at some stage of sampling; e.g., by kinked tubing). After all invalidations, data completeness was calculated as follows: % Completeness = (total valid/total sampled) x 100%.

Field blanks comprised 11-15% of outdoor, indoor, and personal field samples. Blank corrections were not deemed necessary for PM_{2.5} mass. For the other pollutants, blank corrections were applied by season and by microenvironment in cases where the median blank value was greater than its analytical limit of detection. Data below the analytical LOD for SO₄²⁻, EC, O₃, SO₂, NO₂, and the 21 acid-digestible and water-extractable elements were reported as non-detect (ND) by the laboratory. The field LOD for PM_{2.5} mass was calculated as two times the standard deviation of the field blanks. For the other pollutants, because many blanks were ND, field LODs were calculated by using the 95th percentile of field blanks by season (and by microenvironment if they differed sufficiently). Concentration LODs in $\mu\text{g}/\text{m}^3$ and ppb were further calculated based on a 24-hr sampling period and the target sample flow rate (for active samplers) or the predetermined collection rate (for passive samplers). Collection rates were 11 cc/min for O₃ (Chang et al., 1999), 13.3 cc/min for NO₂ (Chang et al., 1999) and 9.9 cc/min for SO₂ (Chang – personal communication).

Imprecision for PM_{2.5} samplers was determined by regression analyses applied to the duplicate filter-based samples collected by the multi-pollutant monitors; for the other pollutants, co-located multi-pollutant monitors (A and B) at the central Steubenville site were used to estimate imprecision. Valid samples greater than the field LOD were paired by date (n = 69, 22, and 44 sample pairs for O₃, SO₂ and NO₂, respectively), and precision was calculated using: $100\% \times [\text{SD}(\text{diff})/\text{Mean}(\text{A,B})] / \sqrt{2}$ (Kinney and Thurston, 1993).

Accuracy was determined by comparing the multi-pollutant monitor measurements made at the central Steubenville site with 24-hr average FRM or FEM measurements made there (reference measurements were not available for SO_4^{2-} , EC, or the elements). For the gases, only data above the field LOD were analyzed, resulting in 91, 31 and 30 sample pairs for O_3 , SO_2 and NO_2 , respectively. Accuracy was calculated as the ratio of mean concentrations from the two methods: $100\% \times (\text{multi-pollutant value} / \text{reference value})$. Regression slopes between the passive and FRM or FEM values were low, suggesting that the collection rates we used were overestimated for field use. However, accuracy of the O_3 and NO_2 concentrations was thought to be sufficient, and use of different estimated collection rates in the calculations did not improve regression fits significantly for these gases. For SO_2 , because so much data were $< \text{LOD}$, only few samples were in effect underestimated.

3.2.1.8 Data Analysis

Prior to statistical analyses, several data points were removed from the dataset, including data for subjects ($n=10$) who withdrew early from the study, as discussed above. In addition, three indoor samples with extremely high concentrations of $\text{PM}_{2.5}$ and SO_4^{2-} were excluded from all data analyses, as were 27 $\text{PM}_{2.5}$ and NO_2 indoor samples collected on days during which indoor questionnaires indicated the presence of environmental tobacco smoke (ETS). For personal exposure data, samples for which participants indicated ETS exposure on their TAD (68 person-days) were left in the data set due to their high number; models using personal $\text{PM}_{2.5}$ and NO_2 gave similar results when these samples were removed. Because concentrations for many of the analyzed elements were extremely low, only those pollutants with at least 50% of values above their respective LODs were included in the data analyses. These included the water-extractable concentrations of As, Ba, Cd, K, Mg, Pb, Se, and V, and the acid-digestible concentrations of Al, Fe, K, Mg, Mn, Pb, Se, V, and Zn. Due to the generally poor precision of these elemental data, however, data analyses were limited to summary statistics and correlation analyses between water-extractable and acid-digestible concentrations.

All data analyses were performed using Excel 2000, SAS Release 8.02 (SAS Institute, Cary, NC) and S-PLUS 2000 Professional Release 3 (Insightful Corporation). Pollutants were first characterized by season and by microenvironment using graphical displays and descriptive statistics, including means, medians, standard deviations (SD), and maximum concentrations. Relationships between ambient $\text{PM}_{2.5}$ concentrations and corresponding personal pollutant exposures were subsequently examined. The effects of ventilation and activities on these relationships were also investigated. Similar analyses were performed using ambient SO_4^{2-} and EC concentrations to determine whether the relationships between ambient concentrations and personal exposures differ for ambient regional $\text{PM}_{2.5}$ pollution and ambient local $\text{PM}_{2.5}$ pollution, respectively. The ambient concentrations used in the analyses were those measured at the central Steubenville site, because these data were the most complete and correspond best to the type of central site data typically used in epidemiologic analyses.

Personal-ambient relationships were examined using general linear mixed-effect models for longitudinal data (PROC MIXED in SAS), which are designed to handle data from multiple subjects and allow subjects to serve as their own controls, and are able to handle unequally-spaced data and to model the covariance structure of the data. Subjects were modeled as random effects using a compound symmetry covariance structure, which generally yielded the lowest Akaike Information Criteria (AIC). All models included a building effect in order to control for differences in building characteristics and pollutant levels where subjects resided. Missing data were assumed to be missing completely at random (MCAR) and thus not likely to affect the maximum likelihood estimates. A Tobit model (Tobin, 1958) was also used to examine

associations with personal and indoor O₃ in the fall and with SO₂ in both seasons, because these pollutants had a large percentage of samples with non-detect values. Tobit models use a linear model to examine data above the pollutant detection limit and a logistic model for data below the detection limit (e.g., $y_i^* = x_i'B + e_i$, where $y_i = (y_i^* \text{ if } y_i^* > nd) \text{ or } (0 \text{ if } y_i^* \leq nd)$). For these models, both fixed effects and random subject effects were specified in S-PLUS using the survReg object.

The dwelling questionnaires provided information about the presence of gas stoves in subjects' apartments. From the indoor questionnaires, we obtained a cleaning factor, a cooking factor, and two ventilation variables using information regarding the hours of AC use and the duration of open windows/doors. All factors were 0/1 variables indicating "none" versus "any," with the exception of the TAD AC use and indoor questionnaire windows use variables. For the TAD AC data (summer only), sample numbers were very unbalanced if categorized as 0/1 (with 16 out of 187 samples falling in the "none" category). Therefore we modified the summer grouping such that "none" represented subjects spending 0-33% of time indoors with AC, and "any" represented subjects spending 33-100% of their time in this environment. For information on windows usage obtained from the indoor questionnaire, a three-level variable was created by summing the number of hours of open windows and/or doors during the sampling period and categorizing by approximate tertiles as follows: "Lo" = no windows/doors open, "Med" = the sum of open windows/doors is > 0 but ≤ 24 hr, and "Hi" = the sum of open windows/doors is > 24 hr.

Finally, models using indoor pollutant data were evaluated for three "subsets" of our study population. "Subset 10" included all subjects for whom personal monitoring data were available, which allowed for direct comparisons between models using personal exposure and indoor concentration data. "Subset 15" included those subjects for whom no personal monitoring data were available. "Subset 25" was the full cohort.

On the basis of previous studies, time spent outdoors and time spent in transit were of particular interest in our analyses of exposure relationships, as were variables assessing ventilation conditions, including time spent indoors with air conditioning (AC) and *time* spent indoors with open windows/doors. Therefore, nine variables (i.e., presence of gas stoves, cooking, cleaning, AC use, open windows/doors, time spent outdoors, time spent in transit, time spent indoors with AC, and time spent indoors with open windows/doors) from the dwelling questionnaires, indoor questionnaires, and time-activity diaries were included as categorical variables in statistical models.

The impact of home ventilation was examined by including home ventilation variables (VENT) in the mixed effect models as main effects and as interaction terms with ambient concentrations:

$$(E)_{ij} = \beta_0 + \beta_1(C_o)_{ij} + \beta_2(\text{BUILDING})_{ij} + \beta_3(\text{VENT})_{ij} + \beta_4(C_o * \text{VENT})_{ij} + b_i + \epsilon_{ij}$$

Home ventilation data were included in models as categorical variables. For open window usage, a value of "1" indicated that the individual spent time in indoor environments with open windows, whereas a value of "0" indicated that the individual spent all of his/her time in indoor environments with closed windows. Similarly, for air conditioning use, values of "1" and "0" indicated that individuals spent time in air conditioned and non-air conditioned environments, respectively. Similar models were also constructed to examine the impact of specific activities, such as cooking and cleaning. Ventilation and activity data were obtained from the time-activity diaries for models of personal exposures and were obtained from housing questionnaires for models of indoor concentrations. Models of indoor concentrations only included data for the

ten subjects for whom personal monitoring data were available to allow results to be compared to those for personal exposures.

All models were stratified by season. Analyses using indoor or personal NO₂ as either dependent or independent variables were stratified by gas stove presence to examine whether the presence of gas stoves would affect observed relationships. Analyses evaluating the effects of ambient SO₂ as the independent variable were run with and without data below the laboratory LOD since these values were replaced with random numbers during data processing stages. For most analyses, results using the full data set and the restricted data set were comparable, with some loss of significance for the restricted data set likely due to the smaller sample size. As a result, only results using the restricted data set are presented. Regression results presented for each relationship include dependent and independent variables, parameter estimates (β), standard errors (SE), t-statistics, and intercepts.

3.2.2 Children's Cohort

3.2.2.1 Study Design

Exposure monitoring for the children's cohort was conducted in Steubenville during the winter (January 20 – March 4) and summer (June 1 – July 3) of 2001. Simultaneous measurements of personal exposures and indoor and outdoor concentrations of PM_{2.5}, EC, SO₄²⁻, NO₂, O₃, and SO₂ were made for 15 children in each season. Monitoring for each child was performed for seven consecutive 24-hr periods. In general, three subjects were sampled simultaneously each week. Corresponding 24-hr air pollutant samples were also collected at the central ambient air monitoring site in Steubenville.

3.2.2.2 Participant Recruitment and Selection

Children between the ages of 9 and 13 were recruited through local community contacts for participation in the study. The children who were identified by these contacts were contacted via telephone. Children who lived in non-smoking households and were willing to participate were included in the study. In all, sixteen children participated in the winter monitoring session. Twelve of these children again participated in the summer monitoring session, as did 3 newly recruited children, for a total of 15 subjects in the summer session. Each participant gave written informed consent prior to his or her participation in the study.

3.2.2.3 Air Pollution and Air Exchange Rate Sampling

As for the older cohort, all air pollutant concentrations for the children's cohort were measured using Harvard multi-pollutant monitors (Demokritou et al., 2001). As described above, the monitor samples fine particles, fine particle components, and gaseous pollutants simultaneously. The personal, indoor, and outdoor monitors used for the children's cohort were configured and operated identically to those used for the older adult cohort, and samples collected by these monitors were analyzed according to the same analytical procedures (see Section 3.2.1.3). In short, the monitors each included two active samplers to measure PM_{2.5} and water-extractable and acid-digestible elements, a mini-sampler to measure SO₄²⁻, a mini-sampler to measure EC, and two passive samplers to measure O₃ and NO₂ and SO₂.

A chemistry lab at the Franciscan University of Steubenville was used for the assembly, disassembly, cleaning, and temporary storage of filter samples. Gravimetric analyses of Teflon filters were conducted at the Harvard School of Public Health; all other lab analyses were

conducted by CONSOL R&D according to the general analytical procedures described in Section 3.1. Teflon filters were analyzed for the following elements in the water-extractable and acid-digestible PM_{2.5} fractions: Al, As, Ba, Ca, Cd, Co, Cr, Cu, Fe, K, Mg, Mn, Na, Ni, Pb, Se, Si, Sn, Ti, V, and Zn.

In addition to air pollutant sampling, daily air exchange rates (AER) were measured in each home on each monitoring day. Air exchange rates were measured using a tracer gas source (perfluorocarbon tracer gas, PFT) and passive samplers (capillary adsorption tubes, CATs). PFT gas was released at a controlled rate from multiple sources within each home, with the sources placed inside the home approximately 24 hours prior to sampling to allow for equilibrium (Dietz et al., 1986). CATs were used for sample collection and were generally placed in the living room, bedroom, and kitchen. In general, 3 to 4 CATs were collected for each house. Additionally, collocated and field blank CAT samples were collected for quality assurance. After sampling, CATs were shipped to HSPH for analysis using a gas chromatograph with electron capture detector (GC/ECD). Due to a problem with sample storage, collected CATs could not be analyzed properly. As a result, air exchange rates could not be determined. Information on home ventilation, however, was captured via questionnaire data (i.e., use of fans and air conditioners, extent of open doors and windows) and was used in place of air exchange rates in our analyses. Open window information has been shown to be a better proxy of home ventilation than air exchange rates (Brown, 2006), suggesting that the lack of air exchange rate data would not impair our ability to achieve study objectives.

3.2.2.4 Housing Characteristics Questionnaire

Information about each subject's home was collected from the subject's parents using a technician-administered questionnaire (Appendix B), which asked for information on heating sources, ventilation sources, cleaning products, cleaning frequency, pets, tobacco usage, occupant information, and proximity to busy roads or other external pollutant sources.

After each 24-hr sampling period, the technician again queried the parents about activities that occurred during that period (see the Daily Follow-up Questionnaire in Appendix B). This included information on tobacco usage, cleaning activities, cooking activities, candle/incense burning, exhaust fan usage, open windows, air conditioner usage, and humidifier usage.

3.2.2.5 Time-Activity Diaries

During each sampling period, subjects kept daily time-activity diaries to record their activity patterns. Each TAD allowed for the entry of activities in half-hour intervals over the course of a waking day (6:30 am to 12:30 am). Technicians trained the subjects in the use of the diaries, and reviewed the completed diaries after each 24-hr sampling period. The diaries documented the locations of the subjects (indoor, outdoor, in transit, type of transit) and the specific activities the subjects were engaged in during each half-hour interval. They also were used to record information that could bias the exposure results, including whether the subject was near a window or near a smoker.

3.2.2.6 Quality Assurance and Quality Control

Flags were assigned to each sample according to the validity of airflows, sample durations, and comments recorded during weigh sessions, assembly/disassembly stages, and field operations. Invalid samples were excluded from all analyses, as were samples for subjects who withdrew

from the study. Extreme NO₂ samples were additionally excluded due to presumable lab or field contamination.

The multi-pollutant monitor was configured with two PM_{2.5} samplers; thus, duplicate samples were obtained for each subject-day and were used for active sample validation purposes prior to averaging for use in data analyses. A precision criterion of ≤ 0.50 relative difference (RD) between the duplicates was established above which both PM_{2.5} samples and any corresponding SO₄²⁻ or EC samples were invalidated (samples with >0.50 RD were assumed to have been adversely affected at some stage of sampling – e.g., by kinked tubing). After all invalidations, data completeness was calculated as follows: % Completeness = (total valid/total sampled) x 100%.

Field blanks comprised 10% of outdoor, indoor and personal field samples. Based on a t-test for significance, blanks for all pollutants were found to be significantly greater than zero, and thus all samples were blank corrected. Data below the analytical LOD for SO₄²⁻, EC, O₃, SO₂, NO₂, and the 21 acid-digestible and water-extractable elements were reported as non-detect (ND) by the laboratory. The field LOD for PM_{2.5} mass was calculated as two times the standard deviation of the field blanks. For the other pollutants, because many blanks were ND, field LODs were calculated by using the 95th percentile of field blanks by season (and by microenvironment if they differed sufficiently). Concentration LODs in µg/m³ and ppb were further calculated based on a 24-hr sampling period and the target sample flow rate (for active samplers) or the predetermined collection rate (for passive samplers). Collection rates were 11 cc/min for O₃ (Chang et al., 1999), 13.3 cc/min for NO₂ (Chang et al., 1999) and 9.9 cc/min for SO₂ (Chang – personal communication). Because a large proportion of SO₂ concentrations were below the LOD, these values are not discussed.

Imprecision for PM_{2.5} samplers was determined by regression analyses applied to the duplicate filter-based samples collected by the multi-pollutant monitors; for the other pollutants, co-located multi-pollutant monitors (A and B) at the central Steubenville site were used to estimate imprecision. Valid samples greater than the field LOD were paired by date (n = 51 sample pairs for both O₃ and NO₂), and precision was calculated using: 100% x [SD(|diff|)/Mean(A,B)] / √2 (Kinney and Thurston, 1993).

Accuracy was determined by comparing the multi-pollutant monitor measurements made at the central Steubenville site with 24-hr average FRM or FEM measurements made there (reference measurements were not available for SO₄²⁻, EC, or the elements). For the gases, only data above the field LOD were analyzed, resulting in 444 and 211 sample pairs for O₃ and NO₂, respectively. Accuracy was calculated as the ratio of mean concentrations from the two methods: 100% x (multi-pollutant value / reference value).

3.2.2.7 Data Analysis

All data analyses were performed using Excel 2000, SAS Release 8.02 (SAS Institute, Cary, NC) and S-PLUS 2000 Professional Release 3 (Insightful Corporation). Pollutants were first characterized by season and by microenvironment using descriptive statistics. Relationships between ambient PM_{2.5} concentrations and corresponding indoor, outdoor, and personal pollutant levels were subsequently examined. The effect of ventilation on these relationships was also investigated. Similar analyses were performed using ambient SO₄²⁻ and EC concentrations to determine whether relationships differ for ambient regional PM pollution and ambient local PM pollution, respectively. The ambient concentrations used in the analyses were those measured at the central Steubenville site.

As for the older adult cohort, relationships among personal, indoor, outdoor, and ambient concentrations for the children's cohort were examined using general linear mixed-effect models for longitudinal data (PROC MIXED in SAS), which are designed to handle data from multiple subjects and allow subjects to serve as their own controls, and are able to handle unequally-spaced data and to model the covariance structure of the data. Subjects were modeled as random effects using a compound symmetry covariance structure, which generally yielded the lowest Akaike Information Criteria (AIC). All models included a building effect in order to control for differences in building characteristics and pollutant levels where subjects resided. Missing data were assumed to be missing completely at random (MCAR) and thus not likely to affect the maximum likelihood estimates.

On the basis of previous studies, time spent outdoors and time spent in transit were of particular interest in our analyses of exposure relationships, as were variables assessing ventilation conditions, including time spent indoors with air conditioning (AC) and time spent indoors with open windows/doors. As with the older adult cohort, nine variables (i.e., presence of gas stoves, cooking, cleaning, AC use, open windows/doors, time spent outdoors, time spent in transit, time spent indoors with AC, and time spent indoors with open windows/doors) from the dwelling questionnaires, daily follow-up questionnaires, and time-activity diaries were included as categorical variables in statistical models for the children's cohort.

The impact of home ventilation was examined by including home ventilation variables (VENT) in the mixed effect models as main effects and as interaction terms with ambient concentrations:

$$(E)_{ij} = \beta_0 + \beta_1(C_o)_{ij} + \beta_3(VENT)_{ij} + \beta_4(C_o*VENT)_{ij} + b_i + \epsilon_{ij}$$

Home ventilation data were included in the models as categorical variables. From the indoor questionnaires, we obtained two ventilation variables using information regarding the hours of AC use and the duration of open windows/doors. For window/door usage, a three-level variable was created by summing the number of hours of open windows and/or doors during the sampling period and categorizing by approximate tertiles as follows: "Lo" = no windows/doors open, "Med" = the sum of open windows/doors is > 0 but ≤ 24 hr, and "Hi" = the sum of open windows/doors is > 24 hr. An AC variable was created where "Any" indicated any use of air conditioning and "None" indicated no air conditioning use. Similar models were also constructed to examine the impact of specific activities, such as cooking and cleaning.

All models were stratified by season. Regression results presented for each relationship include dependent and independent variables, parameter estimates (β), standard errors (SE), t-statistics, and intercepts. Element concentrations were less than their respective LOD in a number of cases; however, a subgroup of elements selected on the basis of their average values and reproducibility were analyzed using summary statistics, correlation analyses among water-extractable and acid-digestible concentrations, mixed effect models, and factor analysis techniques, as described below.

Factor analysis was applied to $PM_{2.5}$, SO_4^{2-} , EC, and certain elemental (either water-extractable or acid-digestible) data collected using the multi-pollutant monitors at the central Steubenville site so that the associations between these central site factors and the exposures of subjects in the children's cohort could be assessed. The subgroup of elements included in the factor analyses was selected based on the completeness and reproducibility of the measurements. For completeness, elements for which at least 80% of the samples collected produced valid determinations were included. For reproducibility, elements were included if their collocated

measurements (at the central Steubenville site) produced a linear regression R^2 value of at least 0.60. For some elements, both the water-extractable and acid-digestible concentrations met these criteria. In these cases, the factor analyses included concentrations for the method for which (1) the sample size or percent completeness was greatest, and (2) the R^2 value for the collocated comparison was greatest. As a result of these criteria, 15 elements were selected for inclusion in the factor analyses: the water-extractable concentrations of As, Ba, Ca, Cd, Fe, K, Mg, Mn, Na, Ni, Sn, and V, and the acid-digestible concentrations of Fe, Pb, and Ti.

The factor analysis method used by the SCAMP indoor and personal exposure program to apportion $PM_{2.5}$ source contributions was described previously (Thurston and Spengler, 1985). This method is based on the premise that daily $PM_{2.5}$ component concentrations can be reduced to a few factors with defined compositional profiles that explain most of the variability in the observed concentrations. For example, for a given day, the concentrations of a total of m components can be explain by p factors, where $p < m$.

Daily factor scores were calculated as:

$$[FS] = [W][Z]$$

where FS are the factor scores, W is the scoring coefficient matrix, and Z consists of standardized concentrations of $PM_{2.5}$ components. The SAS statistical package (PROC FACTOR) was used to solve for W . The initial solution usually has no physical interpretation. To give a physical interpretation, the vectors of the matrices were rotated (i.e., orthogonal rotation) to maximize the loading of each component and establish independent daily scores. This procedure makes each component more strongly associated with one factor rather than spread between several factors.

To derive absolute factor scores, an extra day with zero $PM_{2.5}$ component concentrations was calculated and subtracted from these derived factors as follows:

$$[AFS] = [FS] - [FS^0]$$

where the AFS matrix consists of the absolute factor scores and FS^0 are the factor scores for the day with zero concentrations.

By regressing the daily factor scores (AFS) against the daily $PM_{2.5}$ mass or component concentrations, we can estimate the source contributions from the slopes given by β . The regression equation is as follows:

$$M_k = Intercept + \sum_{j=1}^p \beta_j AFS_{jk}$$

where M_k is the mass or component concentration for day k , β is the estimate of the regression of daily factor scores for source j , and AFS are daily factor scores for factor j for day k . The average source contribution to $PM_{2.5}$ or its components for a given factor was calculated by multiplying the average of the factor scores and the β estimates obtained above. Source profiles (% of $PM_{2.5}$ component mass per total $PM_{2.5}$ mass) were calculated by dividing the average $PM_{2.5}$ component contributions for each factor by the average total mass contribution of the factor.

4. RESULTS AND DISCUSSION

All valid data collected as part of the SCAMP ambient air monitoring program and the SCAMP indoor and personal exposure program are included in Appendix C to this report. This section presents and discusses findings resulting from an analysis of these data. Results from the methods development and validation and cardiovascular health studies conducted as part of SCAMP are also summarized.

4.1 Methods Development and Validation

As mentioned earlier in the Introduction and Experimental sections, a small sub-component of SCAMP involved the development and validation of sampling and analytical methodologies for measuring personal, indoor, and outdoor concentrations of $PM_{2.5}$, $PM_{2.5}$ chemical constituents, and gaseous pollutants. Specifically, this component included a 20-week field comparison of the performance of the Harvard multi-pollutant monitor with the performance of FRM and FEM particle and gas monitors, as well as the development and validation of a method for determining trace elements in ambient, outdoor, indoor, and personal filter-based $PM_{2.5}$ samples by dynamic reaction cell ICP-MS. Results from both of these activities are relevant to assessing the quality of data produced by the SCAMP ambient air monitoring program and indoor and personal exposure program; hence, they are presented first.

4.1.1 Evaluation of the Performance of the Harvard Multi-Pollutant Sampler vs. Ambient FRM or FEM Monitors

One of the research priorities for $PM_{2.5}$ set forth by the National Research Council (NRC) in 1998 was to determine whether $PM_{2.5}$ concentrations measured at a central ambient air monitoring station accurately reflect the breathing zone exposures of individuals to $PM_{2.5}$. However, different sampling equipment and methodologies are typically used to monitor personal exposures to $PM_{2.5}$ than are used to measure ambient concentrations of this pollutant. Therefore, it is important to quantify the comparability of ambient and personal monitoring methods so that possible effects of measurement error on observed differences between personal and ambient concentrations can be understood.

During the summer and fall of 2000, a 20-week study was conducted at the SCAMP central ambient air monitoring site to compare the ambient and personal exposure monitors used by the program. Two Harvard multi-pollutant monitors were collocated with the FRM $PM_{2.5}$ sampler and FRM or FEM SO_2 , NO_2 , and O_3 gaseous pollutant analyzers used by the SCAMP ambient air monitoring program. Various performance metrics, including relative bias, precision, and LODs, were quantified and compared for the various sampling techniques. Sampling, analytical, and statistical methods employed by the sampler comparison study are detailed by Winter et al. (2004).

Estimates of imprecision and limits of detection for the multi-pollutant monitors and the various FRM/FEM ambient air monitors, as well as estimates of the bias of the multi-pollutant monitors relative to the FRM/FEM monitors, are summarized in Tables 6-10. Because the relative biases of the multi-pollutant monitors were in some cases dependent upon the concentrations of the pollutants being measured, these relative biases are shown as a function of ambient pollutant concentration. For each of the pollutants being studied, the shaded cell highlights the bias of the multi-pollutant monitor relative to the FRM or FEM monitor at approximately the mean concentration observed during the 20-week study. For measurements of O_3 and NO_2 made using the multi-pollutant monitor, separate LODs were computed for the summer and fall

seasons because of seasonal differences in the reactivity of the chemically-coated filters used for passive sampling.

Table 6. Imprecision estimates for the Harvard multi-pollutant monitor and FRM or FEM PM_{2.5} and gas monitors.

	Estimated Imprecision				
	PM _{2.5} ($\mu\text{g}/\text{m}^3$)	SO ₄ ²⁻ ($\mu\text{g}/\text{m}^3$)	SO ₂ (ppbv)	O ₃ (ppbv)	NO ₂ (ppbv)
FRM PM _{2.5} Monitor	2.6	0.9	-----	-----	-----
FRM or FEM Gas Analyzer	-----	-----	4.5	4.1	3.9
Multi-Pollutant Monitor	1.9	0.6	2.5	5.7	7.0

Table 7. Estimated bias of the Harvard multi-pollutant monitor relative to the FRM PM_{2.5} monitor for PM_{2.5}.

Ambient Concentration of PM _{2.5} ($\mu\text{g}/\text{m}^3$)	5.0	10.0	20.0	30.0	50.0
Bias ($\mu\text{g}/\text{m}^3$)	-0.8	0.2	2.0	4.5	8.7

Table 8. Estimated bias of the Harvard multi-pollutant monitor relative to the FRM PM_{2.5} monitor for SO₄²⁻.

Ambient Concentration of SO ₄ ²⁻ ($\mu\text{g}/\text{m}^3$)	2.0	4.0	6.0	8.0	10.0
Bias ($\mu\text{g}/\text{m}^3$)	-0.7	-0.4	0.0	0.3	0.6

Table 9. Estimated bias of the Harvard multi-pollutant monitor relative to the FRM or FEM gas analyzers.

Ambient Concentration of Gas (ppbv)	5.0	10.0	25.0	45.0	70.0
Bias for O ₃ (ppbv)	-0.4	0.2	2.2	4.8	9.4
Bias for NO ₂ (ppbv)	2.0	2.0	2.0	-----	-----
Bias for SO ₂ (ppbv)	6.0	7.9	13.4	20.7	-----

Table 10. Estimated limits of detection for the Harvard multi-pollutant monitor and FRM or FEM PM_{2.5} and gas monitors.

	Estimated Limit of Detection				
	PM _{2.5} ($\mu\text{g}/\text{m}^3$)	SO ₄ ²⁻ ($\mu\text{g}/\text{m}^3$)	SO ₂ (ppbv)	O ₃ (ppbv)	NO ₂ (ppbv)
FRM PM _{2.5} Monitor	1.2	0.1	-----	-----	-----
FRM or FEM Gas Analyzer	-----	-----	2.4	2.1	1.2
Multi-Pollutant Monitor ^a	3.0	0.2	6.4	12.7, 10.7	10.8, 6.1

^aFor O₃ and NO₂, the first LOD shown is for the summer, and the second is for the fall.

The Harvard multi-pollutant monitors and the FRM PM_{2.5} monitor performed comparably for measurements of both total PM_{2.5} mass concentrations and fine particulate SO₄²⁻ mass concentrations. The biases of the multi-pollutant monitor relative to the FRM sampler were approximately 10% for PM_{2.5} and 5% for SO₄²⁻ at the mean ambient concentrations of these species. Relative biases increased slightly with increasing concentration but remained less than 20% for PM_{2.5} and less than 10% for SO₄²⁻ at the largest ambient concentrations observed during the study. The imprecisions and LODs of the two monitors for PM_{2.5} and SO₄²⁻ were similar.

Results were less consistent for gaseous pollutant measurements. For O₃ and NO₂ at their average concentrations, the multi-pollutant monitors exhibited biases of approximately 10% and 20%, respectively, relative to the FRM/FEM gas analyzers. The imprecisions of the methods

were also fairly comparable for these two gases. (The imprecisions of the multi-pollutant monitors and the FRM analyzer for NO₂ were large relative to the mean NO₂ concentration measured during the sampler comparison program, but the imprecision of both monitors decreased to approximately 2 ppbv when 4 discordant data points were removed). However, the multi-pollutant monitors' LODs for O₃ and NO₂ were 5-10 times greater than the FRM/FEM gas analyzers' LODs and about 40-110% as great as the mean ambient concentrations observed during the sampler comparison study. The differences in detection limits between the multi-pollutant monitor and the FRM/FEM monitors are likely attributable to the former's large filter blank concentrations for O₃ and NO₂ and its use of passive sampling technology. Concentrations of SO₂ determined by the multi-pollutant monitor exhibited a large bias relative to those determined by the FEM SO₂ analyzer; this relative bias was 80% at the mean concentration measured during the sampler comparison study. Moreover, although the estimated imprecision for the multi-pollutant monitor was slightly better than that for the FEM analyzer for SO₂, the SO₂ detection limit estimated for the multi-pollutant monitor was more than 2.5 times as great as that for the FEM gas analyzer. The detection limits calculated for the multi-pollutant monitor for SO₂ and O₃ were greater than many of the indoor and personal concentrations of these gases measured during SCAMP. Consequently, the interpretation of such data is primarily limited to averages and trends (with appropriate statistical considerations, as discussed in the section describing results from the SCAMP indoor and personal exposure program) and not to single day or individual SO₂ and O₃ measurements.

In summary, the comparison of personal and ambient sampling methodologies conducted as part of SCAMP concluded that the Harvard multi-pollutant monitor performs well in measuring concentrations of PM_{2.5} and fine particulate sulfate, but that its ability to determine concentrations of gaseous pollutants is limited by bias (for SO₂) and high detection limits (for SO₂, O₃, and NO₂). Hence, results for these gaseous pollutants, although generally suitable for use in statistical analyses, must be interpreted with caution.

4.1.2 Development and Evaluation of Methods for Determining Elements in PM_{2.5} Using Dynamic Reaction Cell ICP-MS

An objective of SCAMP was to develop improved methodology for determining trace elements in PM_{2.5}. Although they constitute only a very small portion of total PM_{2.5} mass, trace elements are important for use as tracers in source apportionment analyses of PM_{2.5} and may also play a role in the toxicity of fine particles. Hence, it is important that methods be developed to enable their concentrations to be determined accurately and precisely, even when these concentrations are very low (i.e., < 10 ng/m³ of ambient air).

A majority of the PM_{2.5} trace element determinations to date, including those carried out previously in Steubenville as part of the Six Cities Study and those currently being carried out by the U.S. EPA's Speciation Trends Network, have been performed using X-ray fluorescence spectrometry (XRF). An advantage of this technique is that it is nondestructive, meaning that sample preparation procedures such as dissolution or digestion are not required, and that the PM_{2.5} sample is preserved for future analyses. However, the widespread use of speciation samplers, which simultaneously collect multiple filter-based PM_{2.5} samples for analysis, has made destructive techniques relevant. Moreover, while XRF performs adequately for determining the more abundant elemental components of PM_{2.5}, including many common crustal elements, it lacks the sensitivity required to determine a number of trace elements that are typically present in filter-based PM_{2.5} samples in quantities less than 50-100 ng. This limitation becomes magnified for a program such as SCAMP that includes personal exposure monitoring, because personal samplers operate at ~ 1/10 the volumetric flow rate of ambient

monitors (assuming a standard FRM flow rate of 16.67 L/min), and therefore collect substantially less PM_{2.5}. Finally, SCAMP sought to quantify elements in both the acid-digestible and water-extractable fractions of PM_{2.5}, as these fractions each provide unique information that is useful for drawing inferences about sources and potential health effects of PM_{2.5}. However, given XRF's limited sensitivity for direct analyses, the technique was not considered to be applicable to the analysis of water extracts of PM_{2.5}.

After considering the strengths and weaknesses of the various candidate methods, CONSOL R&D selected dynamic reaction cell ICP-MS for determining PM_{2.5} trace elements as part of SCAMP. DRC ICP-MS exhibited better sensitivity than X-ray techniques such as XRF and proton induced X-ray emission (PIXE), and included measures (i.e., the DRC) to help to overcome the isobaric and polyatomic interferences that limit the capability of conventional ICP-MS to determine isotopes like ³⁹K, ⁴⁰Ca, ⁵¹V, ⁵²Cr, ⁵⁶Fe, ⁷⁵As, and ⁸⁰Se. Another candidate technology, neutron activation analysis, exhibited sensitivities that were similar to or better than those exhibited by DRC ICP-MS; however, this technology was comparably expensive and did not lend itself to the high sample throughput (approximately 5000 filters) required by SCAMP.

The general principles of DRC ICP-MS and the specific methods ultimately employed by SCAMP for PM_{2.5} elemental analyses were described earlier in sections 3.1.2.2 and 3.1.2.4 of this report. This section summarizes the work that was performed to develop and validate those methods.

Prior to the start of the sampling components of SCAMP, CONSOL R&D began to develop methodology for determining elements in both the acid-digestible and water-extractable fractions of PM_{2.5} by DRC ICP-MS. This work, which was conducted from the summer of 1999 through May 2000, focused on evaluating and optimizing sample preparation procedures and analysis parameters. Table 11 lists some of the items that were explored.

Once these and other items were thoroughly evaluated, a comparative study was conducted in 2000 to evaluate the performance of DRC ICP-MS versus that of XRF. The study involved collecting 20 matched pairs of collocated PM_{2.5} filters. The matched pairs included 5 sets of filters collected from personal exposure samplers and 15 sets collected from FRM PM_{2.5} samplers. CONSOL R&D completed the DRC ICP-MS analyses on one filter from each pair, and the collocated filters were sent to a well-respected contract laboratory for XRF analysis. Results of CONSOL R&D's determination of elements in the acid-digestible PM_{2.5} fraction by DRC ICP-MS were then compared with the XRF results, and the relative bias, precision, and sensitivity of the methods were evaluated. The accuracy of the DRC ICP-MS methods for determining acid-digestible and water-extractable elements was also investigated by analyzing NIST Standard Reference Material (SRM) 1648 (urban particulate) and NIST SRM 1643d (trace elements in water).

The results of the study demonstrated the sensitivity advantages that DRC ICP-MS has over XRF. For example, precisions of the XRF and DRC ICP-MS results were evaluated by calculating the percent difference between duplicate analyses of the same sample (for XRF, duplicate analyses were performed on filter-based PM_{2.5} samples; for DRC ICP-MS, analyses were performed on duplicate digestions of NIST SRM 1648, because digesting a filter-based sample in duplicate was not feasible). Figure 5 illustrates the precision of each method, expressed as relative percent difference, as a function of the total mass of element present in the PM_{2.5} sample. The XRF method exhibited good precision for elements that were present in large quantities in the PM_{2.5} samples; however, precision declined markedly for elements that were present in smaller amounts. On average, the relative percent differences for duplicate

XRF determinations of elements present in amounts less than 50 ng were 50% or greater. Elements such as As, Cd, Co, Cr, Ni, V, and Se are typically present in amounts less than 50 ng, especially in personal exposure filter samples. The DRC ICP-MS procedure was able to repeat the analyses of all 21 study elements in two different digestions of NIST SRM 1648 to within 50%. Figure 5 suggests that the precision of DRC ICP-MS does not depend strongly on the quantity of element present in the sample, as it does for XRF.

Table 11. Issues explored during the development of methods for determining trace elements in PM_{2.5} by DRC ICP-MS.

Examined background levels in the digestion. Determined the contribution of elements from various items in the digestion, including acids, DI water, reagents, filters, digestion tubes, and laboratory environment. As a result, used trace grade acids, 18 MΩ DI water, and certified cleaned digestion tubes (Environmental Express) in the method, and used the cleaner chromatography and PM laboratories for sample preparation.
Assessed the degree of isobaric and polyatomic interferences in the MS resulting from variations in the amounts of HF and HCl used in the acid digestion method. Results were used to refine the analytical protocol.
Investigated the effects of varying acid concentrations on digestion efficiency, chloride interferences, and torch bleed. Results were used to refine the analytical protocol.
Examined whether acid-induced corrosion to the electronic pipette's piston was a point of concern for routine sample preparation for the acid-digestible method. Corrosion was determined to be negligible.
Analyzed multi-pollutant sampler filter blanks to determine if washing had a significant effect on the cleanliness of 37-mm Teflon and Fluoropore filters, both of which would be used in the personal exposure samplers. As a result, pre-washing was not employed, and Fluoropore filters were used for sulfate collection.
Optimized the volume of solution required for efficient filter digestion for the acid-digestible method. Digestion volume was selected to be 1.5 ml of a concentrated acid mixture. The volume was sufficient for filter wetting and refluxing.
Explored whether a HNO ₃ -only digestion was adequate for the acid-digestible method. A HNO ₃ digestion was not adequate, as it produced low recoveries for most elements. It was believed that HF was required to release elements from siliceous particulate matrices. In addition, HCl was required for sufficient recovery of certain elements such as Fe and V.
Examined whether it was possible to analyze directly from the digestion tubes, rather than using autosampler tubes, in order to help minimize sample preparation and cost. This was determined to be impractical, and the autosampler tubes were sufficiently clean to present no contamination concerns.
Investigated the addition of high-purity boric acid to stop torch bleed resulting from HF used in the digestion. Torch bleed causes high silicon background concentrations in the analysis. Decided to use a commercially available boron standard made from high-purity boric acid to complex the HF prior to analysis. In addition, an HF-resistant torch and sample introduction system was purchased for routine analysis.

The SCAMP method comparison study concluded that XRF had adequate sensitivity to determine Al, Ca, Fe, K, Pb, Si, and Zn in PM_{2.5}. DRC ICP-MS displayed adequate sensitivity to determine all 21 study elements with the exception of Si, for which uncertainties often exceeded 20%. The XRF and DRC ICP-MS results consistently agreed within 20% for only Se and Pb. Fe and Mn concentrations determined by the two methods generally agreed within 30%; Al, Ca, K, and Zn concentrations typically differed by 30%-70%, and Si concentrations typically differed by a factor of 5-10. In almost all instances, concentrations determined by XRF were lower than those determined by DRC ICP-MS. A sampling of the data is included in Table 12. Results are discussed in greater detail by Conrad et al. (2000).

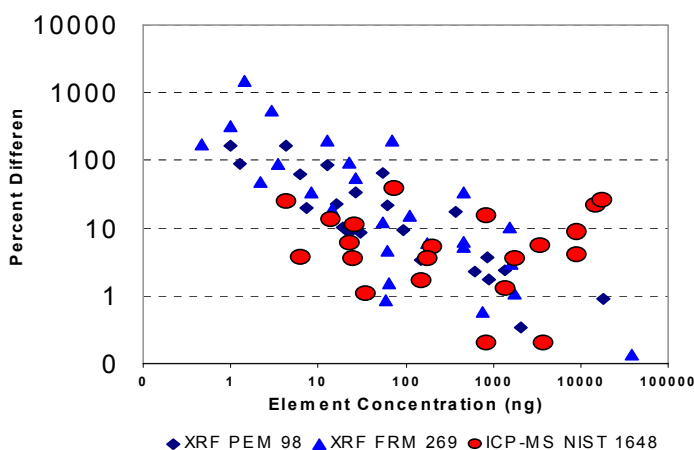


Figure 5. Precision of the DRC ICP-MS and XRF for various elements as a function of the mass of the element present in the PM_{2.5} sample.

Table 12. Concentrations and uncertainties of various elements determined in three sets of duplicate PM_{2.5} samples by XRF and DRC ICP-MS.

Element Isotope	FILTERS 92 & BCH 094 PM				FILTERS 98 & BCH 101 PM				FILTERS 99033 & 272				
	Filter 92 X-ray		Filter BCH 094 PM ICP-MS		Filter 98 X-ray		Filter BCH 101 PM ICP-MS		Filter 99033 X-ray		Filter GA 272 ICP-MS		
	Concentration and Uncertainty (ng) (ng)		Concentration and Uncertainty (ng) (ng)		Concentration and Uncertainty (ng) (ng)		Concentration and Uncertainty (ng) (ng)		Concentration and Uncertainty (ng) (ng)		Concentration and Uncertainty (ng) (ng)		
Filter Wt (mg)													
Ca 40	885	25	2551	16.4	915	25	1435	33	0.924	0.924	0.91	0.91	
Ca 44	885	25	2988	200	915	25	1997	129	1085	36	1499	67	
K 39	521	21	1186	54	633	22	1326	84	1305	44	1978	74	
Al 27	554	48	1234	41.2	963	48	2001	51	1365	117	1729	21	
Cr 50	8.2	29	18.8	1.1	5.8	29.3	23.4	1.9	97.1	11.3	35.3	1.5	
Mn 55	44.4	8.6	106	1.4	50.7	8.7	72.5	4.9	121	9.8	152	8	
Se 80	10	11.1	17.0	0.37	20.1	3.7	23.7	0.78	86.3	5.9	87.1	0.9	
Fe 56	1318	13.4	2149	48.3	1307	13.4	1810	74	2523	21	2612	159	
Na 23	10.4	476	777	57.3	311	433	1667	43	1158	372	1411	62	
Mg 24	87.3	219	408	18.7	75.6	200	515	32	472	514	1061	29	
Si 28	923	41	9114	3031	2063	44.5	8100	1614	2904	101	8298	1479	
V 51	39.8	94.6	44.8	3.1	30.2	95.2	48.5	2.1	21.9	132	49.0	0.5	
Ti 48	65.9	225	115	2.6	108	226	154	5.1	145	324	156	2.7	
Cu 63	7	17.6	33.9	1.1	10.2	18	56.0	0.7	116	6.4	148	2.6	
Co 59	0	26.7	2.0	0.25	0	26.7	2.6	0.3	12.3	44.1	6.5	0.4	
Ni 60	20.2	5.5	50.9	1.6	24.2	5.6	59.8	2.8	46.1	5.7	75.5	4.1	
Zn 66	428	7.9	659	21.2	145	7	345	11	867	10.1	1045	25	
As 75	0.8	21.9	9.6	1.3	0.5	20.6	8.8	0.2	37.9	56.7	54.9	1	
Cd 111	0	97.8	2.4	0.11	0	97.7	2.1	0.4	24.1	144	11.0	0.5	
Cd 114	0	97.8	4.5	0.27	0	97.7	4.1	0.2	24.1	144	13.3	0.3	
Sn 118	55.7	138	5.1	0.77	41.3	139	2.8	0.3	77	206	16.8	0.4	
Sn 120	55.7	138	33.7	49	41.3	139	3.2	1.2	77	206	15.4	0.8	
Ba 135	28.4	566	40.1	2.5	0	568	22.0	1.7	146	820	60.4	3	
Pb 208	56.1	9.8	55.3	3	42	9.5	52.4	1.4	266	17	272	9	

Additional DRC ICP-MS method validation work was completed in late 2000. Acid digestates produced as part of the XRF/DRC ICP-MS comparison study described above were analyzed by Columbia University using a high-resolution ICP-MS. Because high-resolution ICP-MS employs a different technique to resolve interferences than DRC ICP-MS, it served as an independent check on the accuracy of the DRC ICP-MS. Results of the comparison were presented by Winter et al. (2002). In general, element concentrations determined by the two ICP-MS methods agreed well. (DRC ICP-MS performed most poorly for Cu and K; high-resolution ICP-MS performed most poorly for Cu, Cr, and Si). Moreover, as shown in Figures 6 and 7, the methods generally agreed very well for elemental determinations in two different digestions of NIST SRM 1648 (100 µg loaded onto a Teflon filter, as would typically be encountered for PM_{2.5} personal exposure samples).

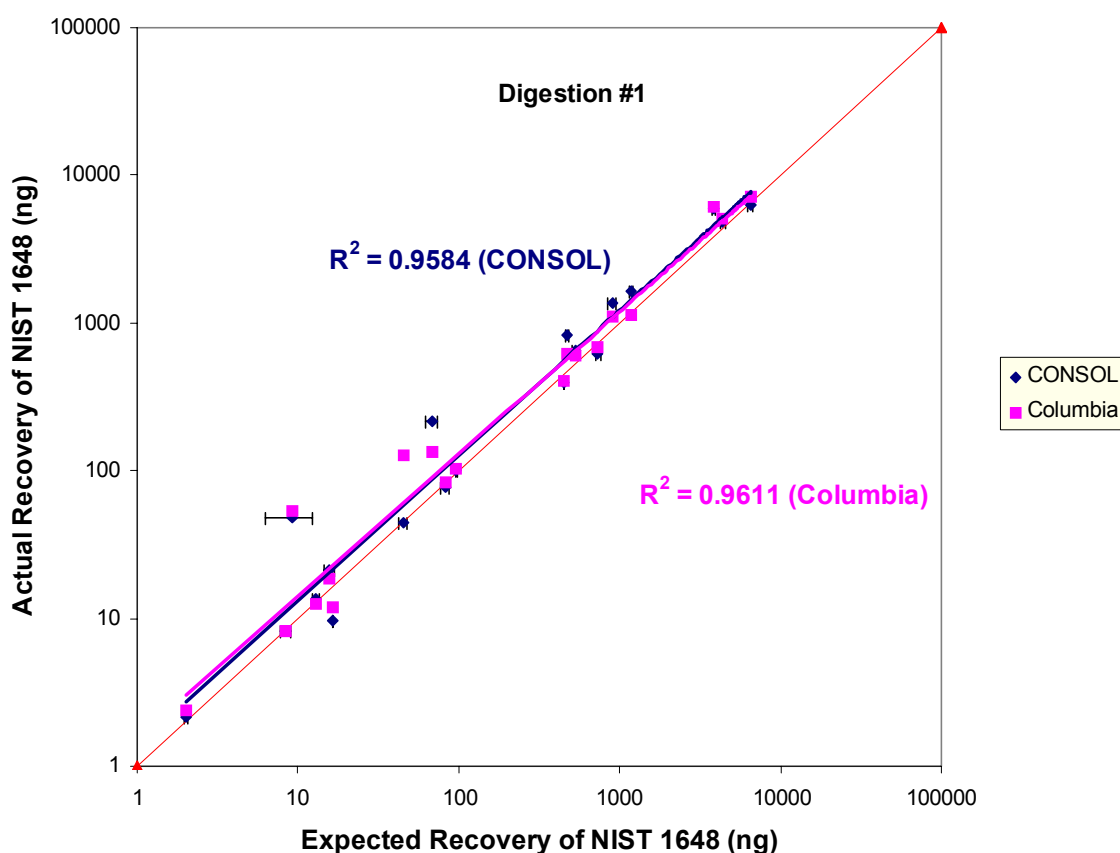


Figure 6. CONSOL R&D (DRC ICP-MS) and Columbia (high-resolution ICP-MS) recoveries of elements in NIST SRM 1648 for digestion #1.

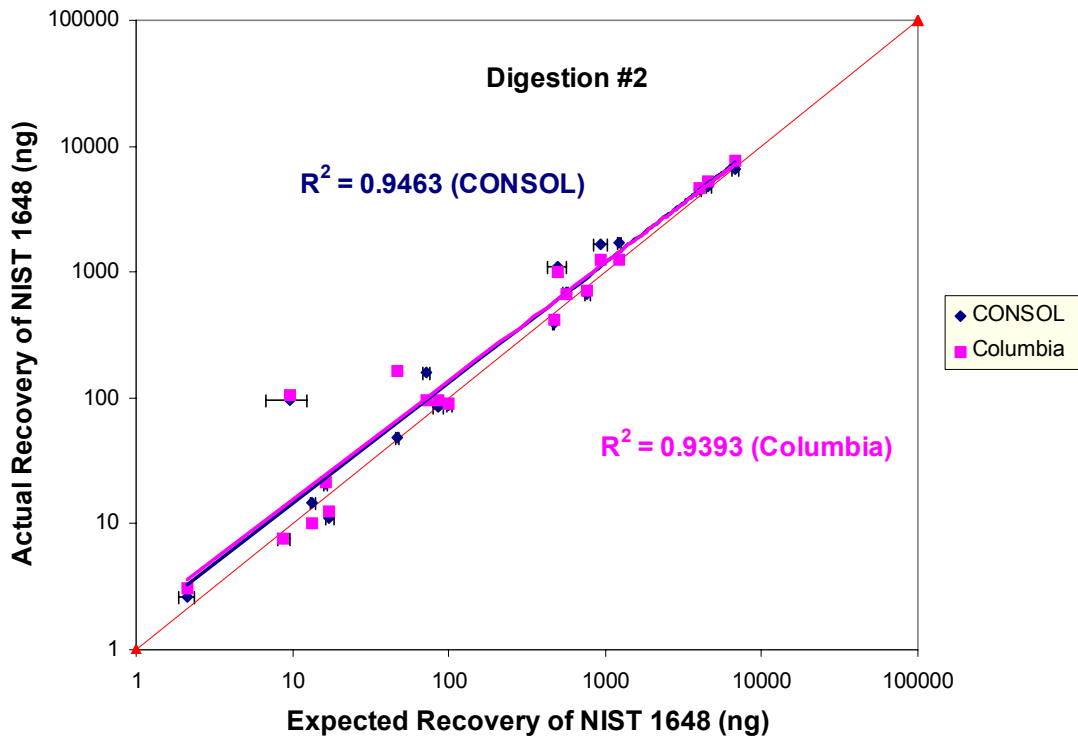


Figure 7. CONSOL R&D (DRC ICP-MS) and Columbia (high-resolution ICP-MS) recoveries of elements in NIST SRM 1648 for digestion #2.

In summary, the comparisons of DRC ICP-MS with XRF and high-resolution ICP-MS confirmed that DRC ICP-MS performs adequately for determining elements in $PM_{2.5}$ samples and that it offers improved sensitivity over XRF for a number of trace elements that are of interest to $PM_{2.5}$ researchers. Additional performance metrics for DRC ICP-MS are presented later in the sections describing results from the SCAMP ambient air monitoring program and indoor and personal exposure program.

4.2 Ambient Air Monitoring Program

4.2.1 Data Overview

The sampling schedule for the five sites comprising the SCAMP ambient air monitoring network was described earlier in section 3.1.1 of this report. Tables 13 and 14, which summarize the data completeness for each parameter monitored at each site, indicate the extents to which the sites were successful in producing valid data.

Table 13. Data completeness for 24-hr integrated PM and PM component measurements at the SCAMP ambient air monitoring sites.

Parameter	No. of Planned Sampling Days		Data Completeness (%) ^a							
			Steubenville PM _{2.5}		Steubenville PM ₁₀	North PM _{2.5}	South PM _{2.5}	East PM _{2.5}	West PM _{2.5}	
PM _{2.5} /PM ₁₀	732		87		92	91	57	92	80	
SO ₄ ²⁻	183		83		88	88	52	86	70	
NO ₃ ⁻	183		83		88	88	52	85	70	
Cl ⁻	183		83		88	88	52	85	70	
NH ₄ ⁺	183		83		88	88	52	85	70	
EC	161		88		NA	NA	NA	NA	NA	
OM	161		88		NA	NA	NA	NA	NA	
	water ^b	acid ^c	water ^b	acid ^d	water ^b	water ^b	water ^b	water ^b	water ^b	
Al	183	161	70	84	69	48	27	74	50	
As	183	161	77	86	78	75	36	79	61	
Ba	183	161	77	77	79	75	37	78	55	
Ca	183	161	61	85	58	64	28	67	40	
Cd	183	161	77	86	79	75	37	77	55	
Co	183	161	78	76	79	77	38	78	60	
Cr	183	161	70	84	70	67	33	69	52	
Cu	183	161	68	81	62	64	33	67	54	
Fe	183	161	68	86	69	66	31	65	43	
K	183	161	69	83	68	64	27	70	52	
Mg	183	161	78	86	79	78	39	79	61	
Mn	183	161	73	86	74	72	35	75	51	
Na	183	161	63	NA	62	65	35	69	51	
Ni	183	161	76	75	75	75	39	77	57	
Pb	183	161	74	86	75	72	35	73	51	
Se	183	161	75	82	75	73	37	72	56	
Si	183	161	66	NA	65	64	26	68	48	
Sn	183	161	78	85	79	76	37	78	55	
Ti	183	161	28	75	45	20	11	19	11	
V	183	161	78	82	79	78	39	79	61	
Zn	183	161	78	86	79	78	39	78	61	

Notes: NA = not applicable; ^aComputed as 100% x number of valid observations / number of planned observations; ^bMeasured in water-extractable PM_{2.5} fraction; ^cMeasured in acid-digestible PM_{2.5} fraction.

The data completeness value shown in the tables for a given parameter at a given site is the number of valid daily or hourly determinations of that parameter as a percentage of the total number of planned daily or hourly determinations (based on the sampling regimen presented in Table 2). Hence, days for which sampling was planned but did not occur (e.g., due to an instrument problem) or for which sampling did occur but data were invalidated because of sampling, analytical, or level 2 quality control checks both count against data completeness. As shown in Table 8, valid FRM-determined PM_{2.5} concentrations were obtained for ≥ 80% of the 732 planned sampling days at all of the SCAMP monitoring sites except for the southern satellite site. This FRM sampler at this site experienced a variety of problems that prevented sampling on a number of days. In many cases, data completeness for PM_{2.5} components was less than that for PM_{2.5} mass because some samples failed to pass QC criteria for laboratory

analyses. With the exception of rainfall, data completeness was $\geq 70\%$ for gaseous pollutants, meteorological conditions, continuous PM_{2.5} mass, and pollen and spores measured at the central Steubenville site. The rain gauge malfunctioned during the first half of the program, and only produced valid data from April 24, 2001, until the end of the program.

Table 14. Data completeness for continuous PM_{2.5}, gaseous pollutants, meteorological conditions, and pollen and spores measured at the central Steubenville site.

Parameter	No. of Planned Sampling Periods	Data Completeness (%) ^a
TEOM PM _{2.5} Mass	16,800	87
SO ₂	17,496	86
NO	17,496	79
NO ₂	17,496	72
NO _x	17,496	74
CO	17,496	85
O ₃	17,496	77
Wind Speed	17,496	96
Temperature	17,496	96
Relative Humidity	17,496	96
Solar Radiation	17,496	84
Barometric Pressure	17,496	96
Rainfall	17,496	52
Pollen	692	85
Spores	692	87

Notes: NA = not applicable; ^aComputed as 100% x number of valid observations / number of planned observations.

All of the 24-hr average PM_{2.5} and PM₁₀ mass concentrations determined at the SCAMP ambient air monitoring sites using the FRMs for these species were greater than the detection limits of the methods. Table 15 summarizes the percentage of valid observations that were below the respective analytical LODs for each PM_{2.5} or PM₁₀ chemical component measured at each SCAMP monitoring site. As shown in the table, a majority of components had only a small percentage (i.e. < 15%) of non-detects. Noteworthy exceptions included Cl⁻, Cr, and Ti in the water-extractable PM_{2.5} fraction. PM_{2.5} and PM₁₀ Cl⁻ concentrations are presented in this report; however, results should be interpreted with caution because of the large number of observations below the LOD. Water-extractable Cr, for which an overwhelming majority of observations were below the LOD, and water-extractable Ti, for which data completeness was poor and a substantial percentage of the few valid determinations were below the LOD, are not included in the analyses presented here.

Because SCAMP included the novel application of dynamic reaction cell ICP-MS to PM_{2.5} elemental analysis, the quality of the resulting elemental data is of particular interest. Quality indicators for the performance of the DRC ICP-MS in determining water-extractable element concentrations are shown in Table 16. As discussed above, water-extractable Cr and Ti were excluded from the analyses presented in this report because of poor data completeness and/or a large number of observations below the LOD. Water-extractable Si was also excluded from the analyses presented here because of high background concentrations, as shown in Table 16. For the remaining 18 elements, the method exhibited good short-term precision (< 10% relative standard deviation (RSD) among replicate determinations), acceptable long-term precision (< 25% RSD among replicate determinations), good accuracy (recovery of NIST 1643d was

100±10% for 15 elements), and small background concentrations (median solution blank ≤ 20% of median exposed filter determination).

Table 15. Summary of the percentage of valid PM and PM component observations at each of the SCAMP ambient air monitoring sites that were below the limits of detection.

Parameter	% of Valid Observations Below LOD						
	Steubenville PM _{2.5}		Steubenville PM ₁₀	North PM _{2.5}	South PM _{2.5}	East PM _{2.5}	West PM _{2.5}
SO ₄ ²⁻	0		0	0	0	0	0
NO ₃ ⁻	2		2	11	6	6	12
Cl ⁻	40		16	60	40	60	55
NH ₄ ⁺	1		1	0	0	0	0
EC	1		NA	NA	NA	NA	NA
OM	1		NA	NA	NA	NA	NA
	water ^a	acid ^b	water ^a	water ^a	water ^a	water ^a	water ^a
Al	0	0	0	0	0	0	0
As	0	7	0	0	0	1	0
Ba	0	1	0	0	0	0	0
Ca	0	6	0	0	0	0	0
Cd	0	1	0	0	0	1	1
Co	1	22	1	3	0	1	0
Cr	75	0	73	73	90	76	90
Cu	1	2	0	0	0	1	0
Fe	4	0	8	8	4	3	4
K	5	1	6	4	4	5	8
Mg	0	0	0	0	0	0	0
Mn	7	1	1	10	13	7	14
Na	0	NA	0	0	0	0	0
Ni	2	27	0	1	1	2	1
Pb	0	0	1	0	0	2	1
Se	0	1	0	0	0	2	1
Si	2	NA	2	4	0	5	0
Sn	5	1	17	9	2	6	1
Ti	25	2	6	38	10	31	38
V	3	1	5	2	1	4	5
Zn	0	0	0	0	0	1	0

Notes: NA = not applicable; ^aWater-extractable PM_{2.5} fraction; ^bAcid-digestible PM_{2.5} fraction.

Table 16 presents data regarding the performance of the DRC ICP-MS method in determining acid-digestible element concentrations. Included among these data are estimates of the relative imprecision of the overall method (i.e., accounting for filter handling, digestion, and instrumental analysis) per element based on pairwise comparisons of determinations from duplicate filters, as well as estimates of the accuracy of the method, which was measured as the average recovery of elements in NIST SRM 2783 (air particulate on filter media). Imprecision data are not available for Ba, Co, and Ni, because valid results were not obtained for these elements during the first sequence of analyses. Accuracy data are not available for Cd, Se, and Sn, because no certified or reference values were provided by NIST for these elements. Based on the quality indicators shown in Table 16, Cr was excluded from the analyses presented in the remainder of this paper because of high field blank loadings (relative to exposed filter loadings)

and poor accuracy (evidenced by an abnormally high recovery of SRM 2783). Performance of the DRC ICP-MS method for the remaining 18 elements was considered acceptable, especially given the low concentrations being measured. The method performed particularly well for As, Cd, Fe, K, Mg, Mn, Pb, Se, and Sn; these elements were characterized by few observations (< 10%) below the analytical LOD (Table 4), low relative imprecisions (< 25%), low background concentrations (mean field blank loading < 25% of mean sample loading), and good accuracy where data were available (recovery of NIST SRM 2783 = 100±10%). Aluminum also met these criteria; however, the elevated standard deviation for recovery of Al in the SRM indicates that results for this element tended to vary across instrument runs and reaction gas conditions.

Table 16. Quality indicators for the determination of elements in the water-extractable fraction of PM_{2.5} via DRC ICP-MS.

Element	Solution Blank as % of Exposed Filter Determination (Based on Median Values)	Short-Term Precision ^a (% RSD)	Long-Term Precision ^b (% RSD)	Accuracy ^a (% Recovery of NIST Water 1643d)
Al	2	4	4	95
As	3	4	10	95
Ba	3	3	4	101
Ca	3	7	24	91
Cd	2	3	1	96
Co	17	2	2	99
Cr	8	7	6	99
Cu	11	3	11	98
Fe	2	8	15	129
K	2	6	11	107
Mg	3	4	9	101
Mn	1	5	6	106
Na	7	2	21	103
Ni	17	4	4	97
Pb	1	3	5	98
Se	0	8	9	86
Si	110	3	4	102
Sn	5	NA	NA	NA
Ti	NA	NA	NA	NA
V	1	6	9	95
Zn	1	7	6	102

Notes: NA = Not applicable; ^aBased upon replicate determinations of a 10x dilution of NIST Water 1643d over a period of 2 hours; ^bBased upon replicate determinations of a 10x dilution of NIST Water 1643d on 3 different days.

FRM and TEOM monitors for measuring PM_{2.5} mass concentrations were collocated at the central Steubenville site throughout SCAMP, permitting an assessment of the performance of the TEOM relative to the FRM. The results of such a comparison were detailed by Connell et al. (2005c). For purposes of the comparison, hourly TEOM data were averaged over 24-hr periods corresponding to the FRM sampling schedule; 515 pairs of collocated measurements were included in the analysis. Both methods exhibited similar imprecisions, ranging from 0.0 to 4.2 µg/m³ for the FRM sampler and from 1.7 to 4.0 µg/m³ for the TEOM monitor for FRM-determined ambient PM_{2.5} concentrations of 6.6 to 43.2 µg/m³. (The FRM imprecisions estimated here from the comparison of the FRM sampler with the TEOM sampler are consistent

with the FRM imprecision of 2.6 $\mu\text{g}/\text{m}^3$ estimated above from the comparison of the FRM sampler with the multi-pollutant sampler). However, as shown in Figure 8, $\text{PM}_{2.5}$ mass concentrations determined by the TEOM were substantially biased in the negative direction relative to those determined by the FRM. As discussed in Section 3.1.2.6, such a negative bias is expected when the TEOM is operated at 50°C, because this temperature is sufficiently high to promote the loss of some semi-volatile material from the mass of $\text{PM}_{2.5}$ that has accumulated on the TEOM filter. TEOM data collected as part of SCAMP were used for trending rather than for determining compliance with mass-based standards, and therefore were not corrected to account for this bias. Nevertheless, the bias should be considered when interpreting the results presented in this report; hourly average $\text{PM}_{2.5}$ concentrations presented here are likely lower than true ambient concentrations.

Table 17. Quality indicators for the determination of elements in the acid-digestible fraction of $\text{PM}_{2.5}$ via DRC ICP-MS.

Element	Field Blank as % of Exposed Filter Determination ^a	Relative Imprecision ^b (%)	Recovery (S.D.) of NIST SRM 2783 ^c (%)	SRM 2783 Blank as % of Loaded Filter Determination ^a
Al	23.0	20.8	99 (34)	2
As	6.6	21.9	110 (7)	9
Ba	9.1	NA	104 (11)	2
Ca	53.6	24.2	104 (6)	11
Cd	4.2	21.5	NA	NA
Co	51.5	NA	115 (12)	-3
Cr	84.8	26.2	163 (4)	50
Cu	42.6	17.9	79 (21)	4
Fe	7.2	19.2	97 (12)	0
K	12.8	15.1	99 (11)	4
Mg	16.5	22.1	97 (5)	2
Mn	1.6	22.6	106 (NA)	3
Ni	9.5	NA	119 (20)	0
Pb	21.9	21.8	101 (5)	-1
Se	14.5	16.8	NA	NA
Sn	5.3	24.0	NA	NA
Ti	52.8	21.3	71 (14)	3
V	35.0	33.6	84 (10)	13
Zn	28.9	18.7	108 (8)	4

Notes: NA = Not available; ^aBased on mean values; ^bRelative imprecision = $100 \cdot (\text{RMSD}/\sqrt{2}) / \text{mean}$, where RMSD is the root mean square difference between duplicate determinations (i.e., from separate analyses of co-collected primary and secondary filters) of a given element, and mean is the overall mean concentration of that element measured at Steubenville during SCAMP; ^cStandard deviation values represent the variability among determinations of the same SRM 2783 digestate across different runs, isotopes, and/or reaction gas conditions.

4.2.2 PM_{2.5} and Co-Pollutant Concentrations at Steubenville

4.2.2.1 Overall Concentrations

4.2.2.1.1 PM_{2.5} and PM_{2.5} Chemical Components

Table 18 presents summary statistics for 24-hr average PM_{2.5} mass concentrations measured using the Federal Reference Method monitor at the central Steubenville site, as well as for 24-hr average mass concentrations of PM_{2.5} chemical components determined at the central site. The overall mean PM_{2.5} mass concentration measured at Steubenville during SCAMP was 18.4 $\mu\text{g}/\text{m}^3$. Sulfate and organic material (OM), which is estimated by multiplying the measured organic carbon mass concentration by a factor of 1.4 to account for the mass of elements such as H, O, and N that are associated with carbon in organic compounds, on average collectively accounted for over half of the total PM_{2.5} mass. Mean ambient air concentrations of these species were 5.8 and 4.5 $\mu\text{g}/\text{m}^3$, respectively. Mean concentrations of fine particulate ammonium and nitrate were 2.3 and 1.16 $\mu\text{g}/\text{m}^3$, respectively; all of the other measured PM_{2.5} components had average ambient air concentrations less than 1 $\mu\text{g}/\text{m}^3$.

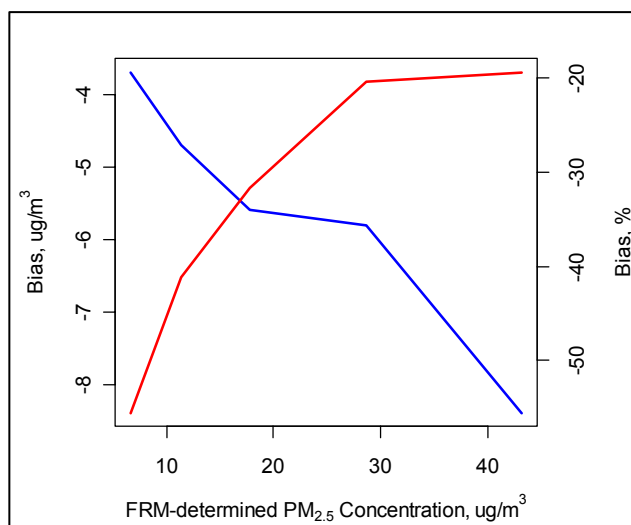


Figure 8. Bias of the TEOM monitor relative to the FRM monitor at the central Steubenville site, as a function of the FRM-determined PM_{2.5} concentration. The blue line represents bias in $\mu\text{g}/\text{m}^3$; the red line represents bias in %.

It is important to note that mass concentrations of nitrate reported in Table 18 and throughout this report, which were determined from PM_{2.5} (or PM₁₀) samples collected on Teflon filters using the FRM monitor, are likely lower than true ambient air concentrations. Biases for NO_3^- , as well as for other semi-volatile species such as OC and NH_4^+ , have been described in the literature (Jansen et al., 2002). As reported by Connell et al. (2005a), NO_3^- concentrations determined from FRM samples at Steubenville were on average 0.79 $\mu\text{g}/\text{m}^3$ less than NO_3^- concentrations determined from samples collected on nylon filters using a collocated speciation sampler equipped with a magnesium oxide denuder. However, because compliance with the NAAQS for PM_{2.5} is based upon measurements made using the FRM, concentrations determined using this method are of greater interest than true ambient concentrations when assessing the impact of PM_{2.5} components on a location's attainment status. Hence, the NO_3^- concentrations determined from FRM samples are reported and used in the statistical analyses

presented throughout this report. All of the other $PM_{2.5}$ components presented here except for elemental and organic carbon were similarly determined from undenuded Teflon-filter-based samples. The carbonaceous species cannot be determined accurately from samples collected on Teflon filters; however, as stated the Experimental section, these species were collected without a denuder to better approximate FRM sampling conditions.

As evidenced by the standard deviations and ranges shown in Table 18, mass concentrations of $PM_{2.5}$ and its components exhibited appreciable day-to-day variability at Steubenville during SCAMP. This variability is visually evident in Figure 9, which is a plot of the time series of daily $PM_{2.5}$ concentrations at the central site. Twenty-four-hour average $PM_{2.5}$ concentrations ranged from 2.7 to 64.8 $\mu\text{g}/\text{m}^3$, with a coefficient of variation (100% x standard deviation / mean) of 61%. All but five of the $PM_{2.5}$ components determined at the central site had coefficients of variation greater than this; the coefficients of variation ranged from 45% for acid-digestible Ca to 242% for Cl⁻. Mean concentrations were greater than median concentrations for $PM_{2.5}$ and each of its components, reflecting the positively skewed distribution of these air pollutant concentrations and the effect of episodic occurrences of high concentrations on the means.

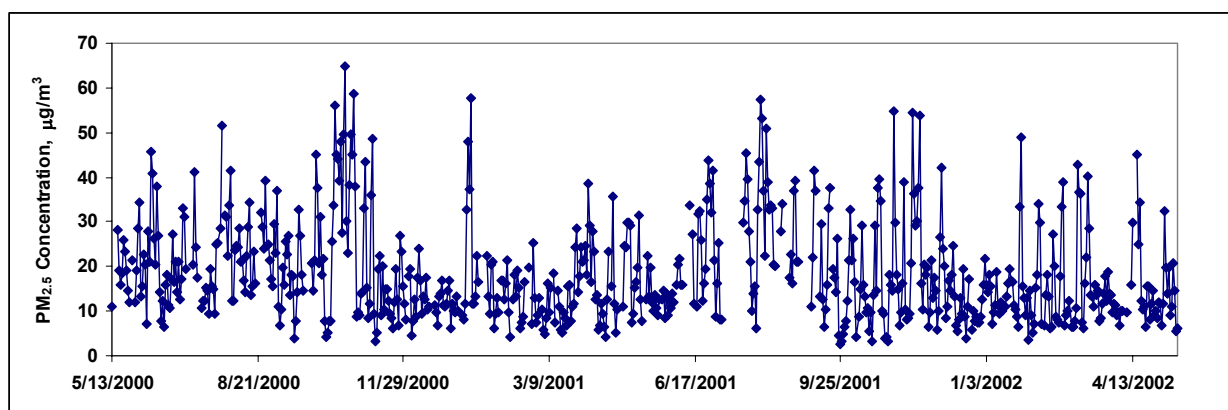


Figure 9. Time series of 24-hr average $PM_{2.5}$ concentrations measured by the FRM at Steubenville.

In addition to measurements made using the FRM monitor, total $PM_{2.5}$ mass concentrations were monitored continuously at the Steubenville site using a TEOM. Table 19 summarizes 1-hr and 24-hr average mass concentrations computed from these continuous measurements. As discussed earlier, $PM_{2.5}$ concentrations measured using the TEOM exhibited an appreciable negative bias relative to those measured using the FRM because of losses of semi-volatile material from the TEOM. Hence, the $PM_{2.5}$ concentrations reported in Table 19 are substantially lower than those reported in Table 18. Hourly $PM_{2.5}$ concentrations measured using the TEOM ranged from -9.0 $\mu\text{g}/\text{m}^3$ to 139.6 $\mu\text{g}/\text{m}^3$, with a mean of 12.7 $\mu\text{g}/\text{m}^3$. Some valid 1-hr average $PM_{2.5}$ concentrations measured using the TEOM were negative, likely because the mass of accumulated $PM_{2.5}$ lost from the filter due to volatilization exceeded the mass of newly collected $PM_{2.5}$ during those measurement periods. Although negative concentrations clearly do not represent physical reality, they are consistent with the known bias of the TEOM, and were used in the analyses presented in this report, as removing them or forcing them to zero would bias the statistical results. Because hourly $PM_{2.5}$ measurements capture intraday as well as interday variability, the coefficient of variation for the hourly measurements (98%) was appreciably greater than that for the daily measurements at Steubenville.

Table 18. Summary statistics for 24-hr average PM_{2.5} and PM_{2.5} component mass concentrations (µg/m³) at the central Steubenville site.

		N	Mean	S.D.	Min	Q25	Q50	Q75	Max
PM_{2.5}		640	18.4	11.2	2.7	10.1	15.3	23.3	64.8
NH₄⁺		151	2.3	1.5	0.0	1.3	1.8	3.1	7.2
SO₄²⁻		151	5.8	4.1	1.0	2.8	4.4	7.3	23.4
NO₃⁻		151	1.16	1.16	0.03	0.35	0.71	1.60	5.34
Cl⁻		151	0.21	0.51	0.01	0.01	0.05	0.14	3.76
EC		142	0.84	0.75	0.00	0.37	0.67	1.02	4.62
OM		142	4.5	3.1	0.2	2.6	3.7	5.8	15.3
Al	tot	135	0.0977	0.0566	0.0297	0.0622	0.0808	0.1233	0.4117
	ws	128	0.0187	0.0185	0.0013	0.0061	0.0132	0.0254	0.1243
As	tot	139	0.00164	0.00164	0.00016	0.00067	0.00120	0.00188	0.01056
	ws	140	0.00259	0.00284	0.00030	0.00105	0.00195	0.00304	0.02227
Ba	tot	124	0.0023	0.0015	0.0003	0.0011	0.0020	0.0032	0.0085
	ws	141	0.0019	0.0013	0.0004	0.0010	0.0016	0.0024	0.0086
Ca	tot	137	0.136	0.061	0.045	0.093	0.118	0.167	0.310
	ws	111	0.078	0.051	0.010	0.044	0.061	0.105	0.241
Cd	tot	139	0.00046	0.00058	0.00001	0.00019	0.00030	0.00051	0.00589
	ws	141	0.00051	0.00067	0.00004	0.00019	0.00036	0.00065	0.00716
Co	tot	122	0.000065	0.000047	-0.000023	0.000031	0.000061	0.000091	0.000209
	ws	142	0.000037	0.000021	-0.000002	0.000024	0.000035	0.000049	0.000132
Cu	tot	130	0.0032	0.0018	0.0002	0.0019	0.0028	0.0040	0.0098
	ws	124	0.0028	0.0027	0.0000	0.0013	0.0020	0.0036	0.0217
Fe	tot	139	0.272	0.308	0.024	0.087	0.173	0.294	1.530
	ws	125	0.019	0.023	0.000	0.005	0.012	0.024	0.136
K	tot	134	0.091	0.055	0.022	0.055	0.073	0.114	0.261
	ws	126	0.099	0.062	0.017	0.059	0.080	0.119	0.314
Mg	tot	138	0.047	0.048	0.009	0.020	0.032	0.057	0.389
	ws	142	0.029	0.036	0.003	0.010	0.019	0.032	0.321
Mn	tot	139	0.0146	0.0154	0.0007	0.0050	0.0096	0.0174	0.0905
	ws	133	0.0078	0.0080	-0.0002	0.0025	0.0053	0.0098	0.0413
Na	tot	NA	NA	NA	NA	NA	NA	NA	NA
	ws	115	0.090	0.074	0.016	0.045	0.069	0.108	0.501
Ni	tot	121	0.0011	0.0015	-0.0003	0.0003	0.0007	0.0015	0.0081
	ws	139	0.0006	0.0006	0.0000	0.0003	0.0005	0.0007	0.0045
Pb	tot	139	0.0153	0.0207	0.0021	0.0070	0.0105	0.0162	0.2183
	ws	136	0.0090	0.0128	0.0004	0.0026	0.0053	0.0098	0.1216
Se	tot	132	0.00330	0.00321	0.00005	0.00149	0.00240	0.00420	0.02469
	ws	137	0.00473	0.00493	0.00034	0.00194	0.00343	0.00573	0.03391
Sn	tot	137	0.00136	0.00119	0.00015	0.00062	0.00097	0.00155	0.00597
	ws	142	0.00022	0.00031	-0.00002	0.00006	0.00011	0.00025	0.00235
Ti	tot	121	0.0169	0.0206	0.0000	0.0089	0.0116	0.0177	0.1590
	ws	NA	NA	NA	NA	NA	NA	NA	NA
V	tot	132	0.00150	0.00149	0.00009	0.00057	0.00106	0.00199	0.01102
	ws	143	0.00105	0.00133	0.00004	0.00025	0.00066	0.00117	0.00780
Zn	tot	139	0.0844	0.0814	0.0199	0.0409	0.0576	0.0898	0.4486
	ws	143	0.0555	0.0745	0.0029	0.0150	0.0280	0.0601	0.4355

NOTE: tot = concentration in the acid-digestible PM_{2.5} fraction, ws = concentration in the water-soluble PM_{2.5} fraction

Table 19. Summary statistics for 1-hr and 24-hr average PM_{2.5} mass concentrations (µg/m³) measured at the central Steubenville site using a TEOM.

	N	Mean	S.D.	Min	Q25	Q50	Q75	Max
1-hr PM _{2.5}	14682	12.7	12.5	-9.0	4.3	8.6	17.0	139.6
24-hr PM _{2.5}	590	12.6	9.7	-0.1	5.7	9.6	16.6	53.9

In light of recent findings suggesting that health effects, including asthma (Delfino et al, 1998) and myocardial infarction (Peters et al., 2001), are associated with short-term transient exposure to PM, it is important to determine whether the 24-hr PM_{2.5} concentrations commonly used in epidemiology studies can serve as surrogates for shorter-term exposures. Chuersuwan et al. (2000) and Weber et al. (2003) reported that in Newark, NJ, and Atlanta, GA, respectively, short-term episodes of elevated PM_{2.5} concentrations were at times masked by the use of 24-hr average data. Connell et al. (2005c) analyzed the potential for such masking to occur at Steubenville using SCAMP data from the central site, and reported that 24-hr average concentrations generally indicated the potential for exposure to high 1-hr average concentrations, although there were some exceptions. For example, 5 days with 24-hr average PM_{2.5} concentrations <30 µg/m³ had 1-hr maximum PM_{2.5} concentrations >65 µg/m³. Figure 10 graphically illustrates another such exception by showing two days with very similar 24-hr average PM_{2.5} concentrations but very different hourly concentration profiles. Although such exceptions did not occur frequently at Steubenville, they should be considered when designing studies to differentiate among the effects of daily and shorter-term exposures to PM_{2.5}.

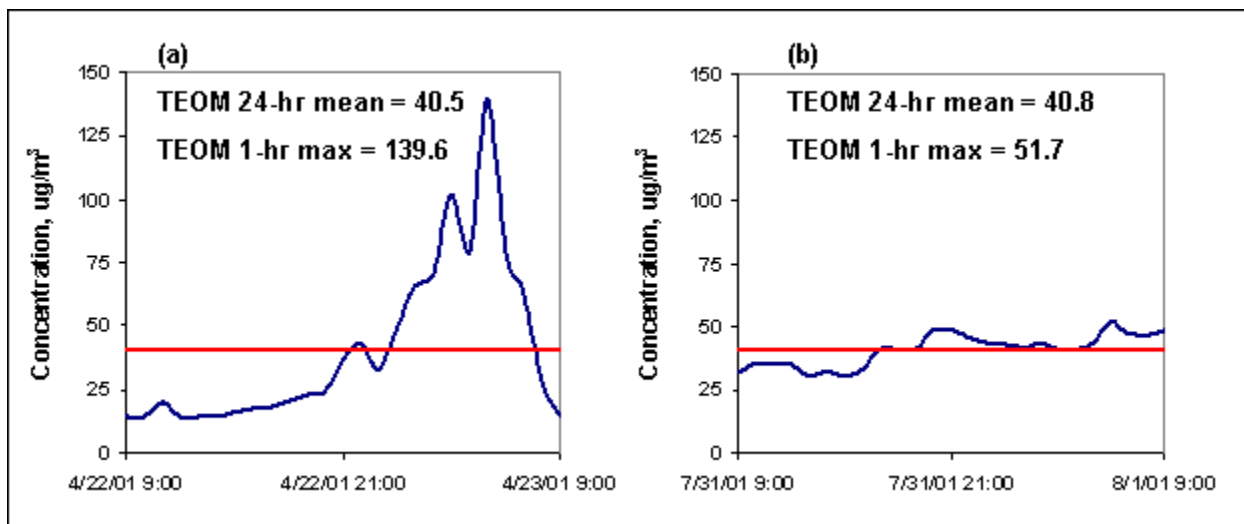


Figure 10. Case study showing two days at Steubenville with similar 24-hr average PM_{2.5} concentrations but different hourly PM_{2.5} concentration profiles, based on TEOM data from the central site. The blue line shows hourly average concentrations; the red line shows the 24-hr mean.

As described earlier and shown in Table 18, ambient air concentrations of a suite of trace and crustal elements were determined for both the acid-digestible and water-extractable fractions of PM_{2.5} samples collected at the Steubenville site. The water-extractability of PM_{2.5} trace elements is of interest to health effects researchers, because it provides some indication of their bioavailability. A number of recent PM toxicology studies (e.g., Fernandez et al., 2005; Frampton et al., 1999; Huang et al., 2003; Zelikoff et al., 2002) have reported associations between soluble transition metals and pulmonary health endpoints. Figure 11 summarizes the distribution of daily fractional solubilities (computed as the concentration determined in the

water-extractable PM_{2.5} fraction divided by the concentration determined in the acid-digestible PM_{2.5} fraction on a given day) observed at Steubenville for the 17 elements for which valid concentrations were determined in both PM_{2.5} fractions during SCAMP. Fe was the most abundant but least water-extractable of the PM_{2.5} elements determined at Steubenville, having a mean ambient air concentration of 272 ng/m³ and a median fractional solubility of 6%. Al and Sn were also largely insoluble, having median fractional solubilities of less than 0.2. Conversely, As, Ba, Cd, K, and Se were largely water-extractable from PM_{2.5} collected at Steubenville; these elements each had a median fractional solubility of greater than 0.8. The data presented in Table 18 and Figure 11 indicate that concentrations of As and Se determined in the water-extractable fraction were frequently greater than concentrations determined in the acid-digestible fraction of PM_{2.5}. This physical incongruity is likely due to a sampling or analytical error; however, as noted by Connell et al. (2006), pairwise concentrations measured in the water-extractable and acid-digestible fractions were well-correlated for each of these elements, suggesting that they are suitable for use in trending and source apportionment analyses.

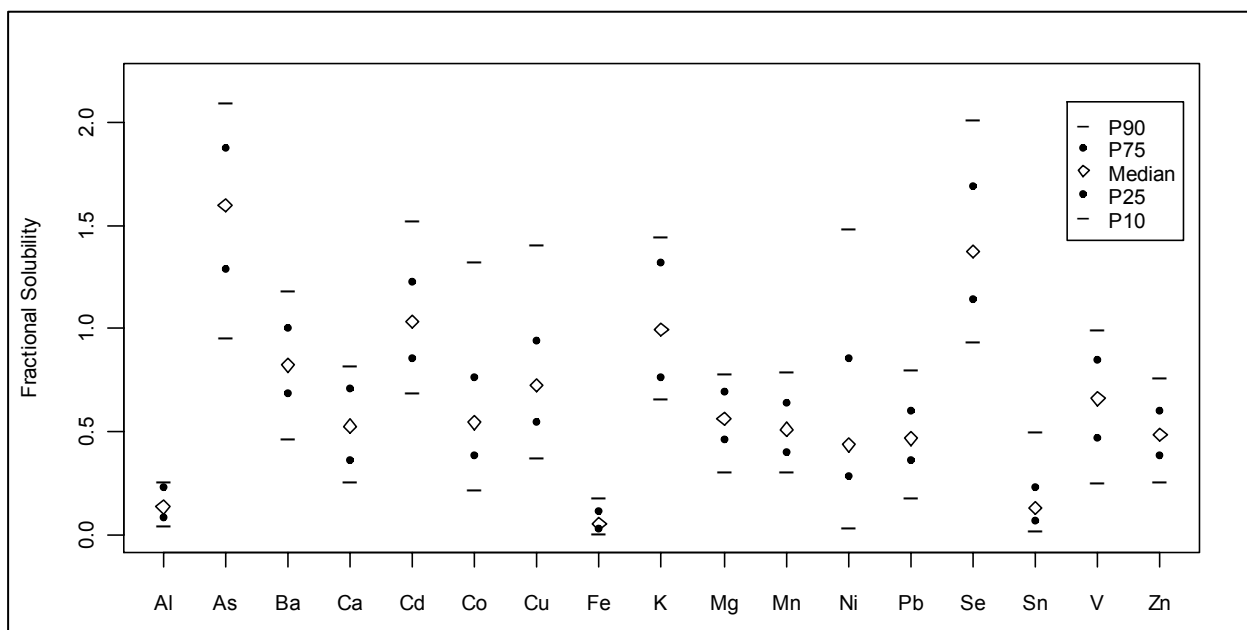


Figure 11. Distribution of daily fractional solubilities (defined in text) computed using 24-hr average PM_{2.5} water-extractable and acid-digestible element concentrations measured at the central Steubenville site.

Because 24-hr average PM_{2.5} chemical component mass concentrations at Steubenville were determined using several different sampling and analytical techniques, mass balances were performed to assess the extent to which these measurements accounted for the total PM_{2.5} mass determined by the FRM. In addition to providing an assessment of the quality of the PM_{2.5} speciation measurements made at the Steubenville site, the mass reconstruction process also provided a means for better evaluating the composition of PM_{2.5} on a percentage basis. (Because of missing values and variations in the sampling schedules for total PM_{2.5} mass concentrations and the mass concentrations of its chemical components, percentages computed using the means shown in Table 18 are only approximate). PM_{2.5} mass balances were performed using a modified version of the reconstructed fine mass (RCFM) formula

commonly employed by Desert Research Institute (Tropp et al., 2003); the formula used in SCAMP is given by:

$$RCFM = SO_4^{2-} + NO_3^- + NH_4^+ + OM + EC + (Na + Cl) + Geological + Trace Elements,$$

where $OM = 1.4 \cdot OC$, as discussed previously; $Na + Cl$ was computed as the sum of the measured masses of water-extractable Na and Cl^- if data were available for water-extractable Na , and as $1.6 \cdot Cl^-$ if water-extractable Na data were not available; Geological was computed as $1.89 \cdot Al + 2.14 \cdot Si + 1.4 \cdot Ca + 1.43 \cdot Fe$ (the multipliers are used to account for the mass of oxides of these elements), and Trace Elements was computed as the sum of the masses of the 15 species other than Al , Ca , and Fe that were determined by DRC ICP-MS. Because Si data were not available from the central Steubenville site, the mass of Si was estimated as $1.52 \cdot Al$, where 1.52 is the ratio of the mass of fine particulate Si to the mass of fine particulate Al measured in Steubenville during the Six Cities Study (Laden et al., 2000).

There were 109 days for which valid data were available for all of the $PM_{2.5}$ components required to compute RCFM. (Because their contribution to total $PM_{2.5}$ mass is very small, it was not required that all trace element species be present). Table 20 summarizes the overall $PM_{2.5}$ composition at Steubenville during these days. Figure 12 shows the distribution of daily mass balance closures (computed as $100\% \cdot RCFM / FRM \text{ Mass}$), both overall and by astronomical season. The median daily mass balance closure was 90.3%, with minimum and maximum values of 63.0 and 119.6%, and 25th and 75th percentiles of 85.0% and 96.1%. The underestimation of total $PM_{2.5}$ mass on average is expected and is likely due in part to the presence of species such as particle-bound water and compounds of trace elements that were not measured. Moreover, the multipliers used to estimate OM from OC and Si from Al for inclusion in the RCFM formula may be low. Turpin and Lim (2001) suggest a factor of 1.4-1.8 for estimating the average molecular weight per carbon weight for urban organic aerosols, and an even greater factor for non-urban aerosols. If a factor of 1.8 is used in the RCFM formula for SCAMP, the median daily mass balance closure increases to 97.7%. In addition, the Si/Al ratio of 1.52 derived from the Laden et al. (2000) data is substantially less than the ratio of 3.41 derived from crustal averages reported by Mason (1966). Use of this larger ratio further increases the median mass balance closure to 100.3%.

Table 20. Overall composition of $PM_{2.5}$ at Steubenville, based on 109 days with complete or nearly complete $PM_{2.5}$ speciation data.

Component	Percent (w/w) of Total $PM_{2.5}$ Mass
Sulfate	30.4
Nitrate	6.7
Ammonium	12.5
Organic Material	25.7
Elemental Carbon	4.8
$Na + Cl^-$	1.9
Geological	6.3
Trace Elements	1.6
Unidentified	10.0

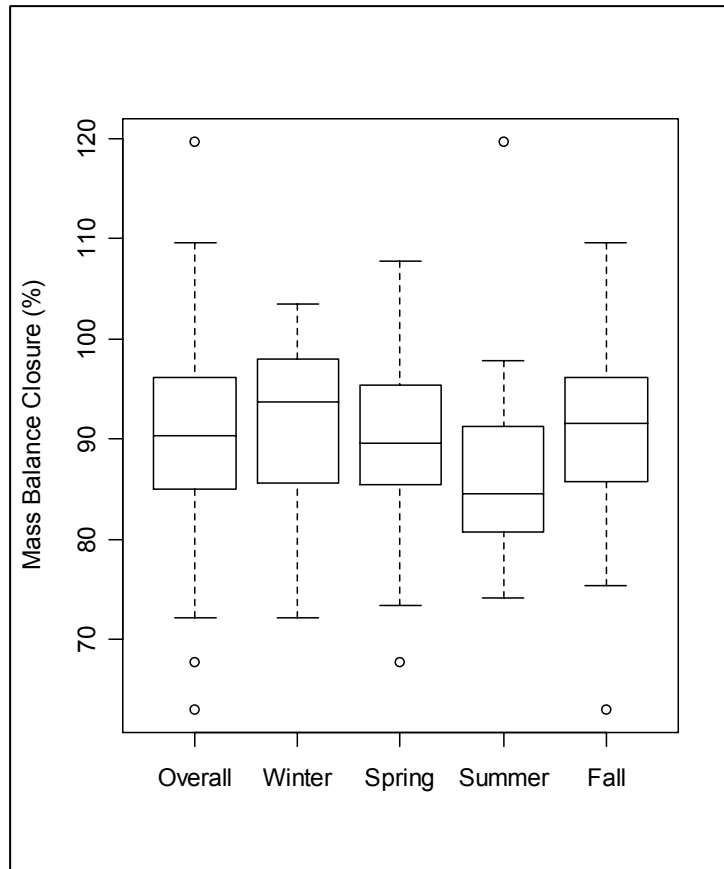


Figure 12. Daily PM_{2.5} mass balance closures at the central Steubenville site, overall and by season.

4.2.2.1.2 PM₁₀ and PM₁₀ Chemical Components

Table 21 presents summary statistics for 24-hr average mass concentrations of PM₁₀ and its chemical components determined at the central Steubenville site. Figure 13 shows the time series of daily PM₁₀ mass concentrations measured during SCAMP. The overall mean PM₁₀ mass concentration observed between May 2000 and May 2002 as part of SCAMP was 27.4 μg/m³. Daily average PM₁₀ concentrations ranged from 4.9 to 102.2 μg/m³, with a coefficient of variation of 58%. Sulfate was the most abundant of the PM₁₀ chemical components that were measured (OM, the second most abundant component of PM_{2.5}, was not measured for PM₁₀ as part of SCAMP), having a mean ambient air concentration of 6.2 μg/m³ (about 22.6% of the total PM₁₀ mass concentration). Coefficients of variation for PM₁₀ components ranged from 49% for water-soluble K to 194% for water-soluble Pb.

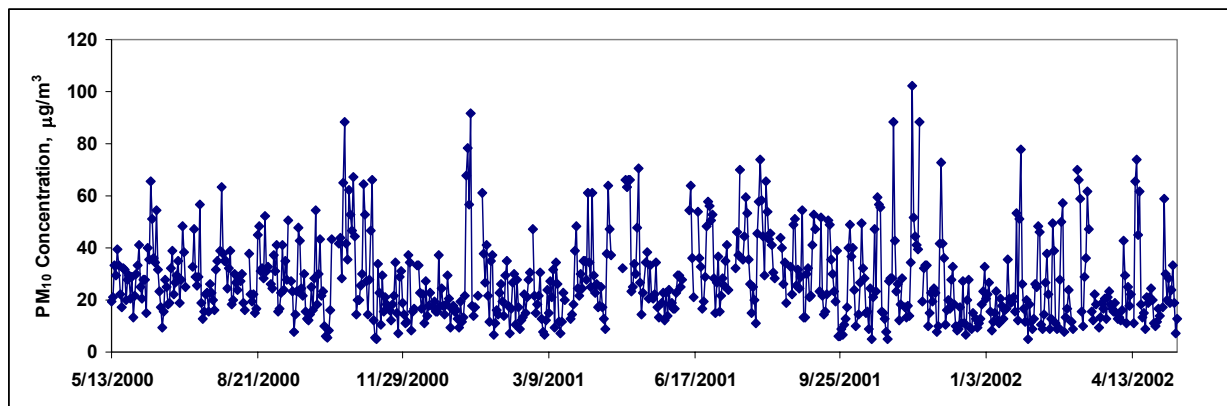


Figure 13. Time series of 24-hr average PM₁₀ concentrations measured by the FRM at Steubenville.

PM₁₀ includes PM_{2.5}, as well as coarser particles with diameters between about 2.5 and 10 µm. Hence, in order to permit an assessment of the extent to which the mass and composition of PM₁₀ at Steubenville is attributable to PM_{2.5}, Table 21 includes quartiles summarizing the daily ratios of the masses of PM and PM components in the PM_{2.5} fraction to those in the PM₁₀ fraction. PM_{2.5} typically accounted for 58-77% (interquartile range) of the mass of PM₁₀ at Steubenville. Median ratios of concentrations of NH₄⁺, SO₄²⁻, and water-extractable As, Cd, Cu, Se, and V in PM_{2.5} to concentrations of these species in PM₁₀ were 1 ± 0.1, suggesting that they are predominantly present in the fine fraction. Water-extractable Pb and Sn had median daily PM_{2.5}/PM₁₀ ratios substantially greater than 1, likely because of sampling or analytical error. Conversely, median PM_{2.5}/PM₁₀ ratios for Cl⁻ and water-extractable Ba, Ca, Co, Mg, and Mn were < 0.5, suggesting that at least half of their total concentration in PM₁₀ is contributed by the coarse fraction.

4.2.2.1.3 Co-Pollutants

Table 22 summarizes 1-hr and 24-hr gaseous pollutant concentrations computed from continuous monitoring data collected at the central Steubenville site. Mean gaseous pollutant concentrations ranged from 6.8 ppb for NO to 0.30 ppm for CO. Coefficients of variation were 45% (O₃) to 137% (NO) for daily data, and 69% (O₃) to 219% (NO) for hourly data. As with PM_{2.5} and PM₁₀, mean concentrations were greater than median concentrations, reflecting the effect of episodic occurrences of high concentrations on the mean. Negative values were obtained for several gas observations that otherwise met all validation criteria, indicating that ambient concentrations were smaller than instrument noise during the periods in which these measurements were made. As with PM_{2.5} TEOM data, these negative values were used in the analyses presented in this report, because adjusting them would introduce bias.

Table 21. Summary statistics for 24-hr average mass concentrations of PM₁₀ and water-extractable PM₁₀ components (µg/m³) at the central Steubenville site. Also shown are quartiles summarizing the distribution of daily ratios of the mass of PM and water-soluble PM components in the PM_{2.5} fraction to the mass of these species in the PM₁₀ fraction.

	PM ₁₀ Summary Statistics								PM _{2.5} / PM ₁₀ Summary Statistics			
	N	Mean	S.D.	Min	Q25	Q50	Q75	Max	N	Q25	Q50	Q75
PM ₁₀	676	27.4	15.9	4.9	15.8	23.1	34.7	102.2	610	0.58	0.68	0.77
NH ₄ ⁺	161	2.2	1.4	0.0	1.2	1.9	2.8	8.5	141	0.94	1.03	1.11
SO ₄ ²⁻	161	6.2	4.5	1.2	3.1	5.2	7.8	31.8	141	0.86	0.94	1.00
NO ₃ ⁻	161	1.76	1.45	0.03	0.68	1.25	2.42	6.55	141	0.42	0.61	0.76
Cl ⁻	161	0.38	0.64	0.01	0.06	0.16	0.38	4.67	141	0.20	0.49	1.00
Al	126	0.0280	0.0266	0.0017	0.0098	0.0184	0.0351	0.1446	116	0.34	0.51	1.00
As	142	0.00222	0.00182	0.00017	0.00113	0.00176	0.00266	0.01242	129	0.93	1.06	1.27
Ba	144	0.0064	0.0045	0.0008	0.0033	0.0052	0.0079	0.0247	130	0.26	0.30	0.36
Ca	106	0.517	0.327	0.036	0.284	0.508	0.655	1.744	95	0.11	0.13	0.20
Cd	144	0.00054	0.00079	0.00004	0.00021	0.00037	0.00064	0.00861	130	0.81	0.96	1.09
Co	144	0.000079	0.000048	-0.000002	0.000044	0.000071	0.000106	0.000271	131	0.40	0.49	0.58
Cu	114	0.0030	0.0035	0.0000	0.0013	0.0022	0.0039	0.0315	108	0.66	0.93	1.26
Fe	127	0.023	0.038	0.000	0.005	0.011	0.023	0.300	112	0.55	0.82	1.52
K	124	0.106	0.052	0.016	0.070	0.100	0.133	0.309	113	0.70	0.87	1.10
Mg	144	0.124	0.108	0.005	0.057	0.097	0.149	0.735	131	0.15	0.21	0.29
Mn	135	0.0214	0.0216	0.0004	0.0080	0.0145	0.0305	0.1660	120	0.25	0.39	0.51
Na	113	0.237	0.314	0.019	0.073	0.107	0.234	1.689	104	0.39	0.58	0.78
Ni	138	0.0010	0.0015	0.0001	0.0004	0.0007	0.0011	0.0141	126	0.50	0.71	1.00
Pb	137	0.0057	0.0111	0.0000	0.0011	0.0028	0.0063	0.1147	125	1.07	1.49	3.20
Se	137	0.00453	0.00468	0.00033	0.00182	0.00324	0.00551	0.02885	124	0.95	1.08	1.24
Sn	144	0.00015	0.00022	0.00000	0.00003	0.00008	0.00018	0.00143	130	1.00	1.57	2.51
V	145	0.00118	0.00148	0.00003	0.00031	0.00068	0.00156	0.01135	132	0.61	0.93	1.11
Zn	145	0.0623	0.0732	0.0039	0.0203	0.0374	0.0699	0.4601	132	0.71	0.84	0.94

Table 22. Summary statistics for 1-hr and 24-hr average gas concentrations measured at the central Steubenville site.

		N	Mean	S.D.	Min	Q25	Q50	Q75	Max
SO₂ (ppb)	1-hr	14969	10.6	14.0	0.4	2.9	6.0	12.4	233.9
	24-hr	624	10.5	8.4	1.3	5.0	8.2	12.9	60.1
NO (ppb)	1-hr	13857	6.8	14.9	-1.4	0.3	1.2	4.9	215.3
	24-hr	569	6.8	9.3	0.1	1.3	3.3	8.1	71.5
NO₂^a (ppb)	1-hr	12562	12.7	8.9	-0.5	5.6	10.1	18.0	52.9
	24-hr	504	12.7	5.8	1.9	8.3	11.8	16.4	33.1
NO_x (ppb)	1-hr	12952	18.8	19.8	-0.8	6.4	11.5	23.5	196.9
	24-hr	525	18.8	12.8	2.4	9.9	15.3	23.4	102.6
CO (ppm)	1-hr	14881	0.30	0.58	-0.19	0.05	0.13	0.29	8.09
	24-hr	620	0.30	0.35	-0.08	0.10	0.20	0.36	2.73
O₃ (ppb)	1-hr	13482	27.9	19.2	-0.4	12.1	25.7	39.5	104.3
	24-hr	580	28.0	12.6	2.4	19.3	26.6	35.5	77.0

^aComputed by differencing measured NO_x and NO concentrations.

Tables 23 and 24 summarize daily counts of spores and pollen, respectively, from the central site. For each type of pollen and spore, the tables show both the total number of daily observations and the number of daily observations with nonzero counts, as there were a considerable number of days for which certain types were not found on the tape from the Burkard sampler. Only four of the spore types (ascospores, basidiospores, cladosporium, and smut/myxo/periconia) were present on greater than 75% of the days studied; each of the specific pollen types was observed in 30% of the samples or less. Hence, because of statistical limitations related to the presence of large numbers of non-detects, the trending and correlation analyses presented in the rest of this paper will be limited to the four prevalent spore types, total spores, and total pollen.

4.2.2.2 Temporal Trends

Temporal trends in ambient concentrations of PM_{2.5} and co-pollutants are induced by temporal trends in factors, such as anthropogenic activities, biogenic activities, and meteorological conditions, that affect these concentrations. Hence, characterizing seasonal, day-of-week, and diurnal trends can provide insight into factors that influence ambient pollutant concentrations, which can prove valuable when developing ways to reduce these concentrations. Moreover, the characterization of temporal trends in pollutant concentrations is important for health effects research, as epidemiological models must be designed to appropriately account for these trends (e.g., to prevent confounding by season), and the potential for exposure to elevated concentrations of ambient pollutants is dependent upon the interaction between the temporal patterns exhibited by these pollutants and temporal patterns of human activity. This section characterizes seasonal, day-of-week, and diurnal trends exhibited by PM_{2.5} and co-pollutants at the central ambient air monitoring site in Steubenville during SCAMP.

Table 23. Summary statistics for 24-hr average spore counts (number/m³) at the central Steubenville site.

Species	N	N _{nonzero}	Mean	S.D.	Min	Q25	Q50	Q75	Max
Alternaria	599	386	52.6	84.9	0	0	11	75	652
Ascospores	599	557	1482.1	2794.7	0	64	417	1551	22929
Basidiospores	599	591	3200.7	4646.4	0	139	1059	4663	35784
Botrytis	599	146	9.9	27.5	0	0	0	0	278
Cercospora	599	23	0.9	5.6	0	0	0	0	75
Chaetomium	599	39	0.8	3.6	0	0	0	0	53
Cladosporium	599	583	1372.7	1843.7	0	128	567	2091	12598
Curvularia	599	80	2.1	6.3	0	0	0	0	64
Drechs/Helminth/Bipolaris	599	79	1.7	5.0	0	0	0	0	32
Epicoccum	599	367	34.0	65.8	0	0	11	43	674
Fusarium	599	5	0.2	3.4	0	0	0	0	75
Hyaline Conidia	599	365	84.6	154.3	0	0	21	107	1283
Penicillium/Aspergillus	599	6	1.5	21.7	0	0	0	0	493
Peronospora	599	13	0.6	6.5	0	0	0	0	118
Pestalotia	599	88	2.6	7.9	0	0	0	0	75
Pithomyces	599	182	10.9	27.7	0	0	0	11	332
Polythrincium	599	141	5.7	14.4	0	0	0	0	139
Rusts	599	161	10.3	30.2	0	0	0	11	353
Smut/Myxo/Periconia	599	463	82.2	146.7	0	11	32	91	1208
Stachybotrys	599	1	0.1	1.3	0	0	0	0	32
Torula	599	65	2.7	14.4	0	0	0	0	267
Ulocladium	599	7	0.3	3.1	0	0	0	0	64
Unidentified Spores	599	37	1.8	9.5	0	0	0	0	131
Hyphal Fragments	599	6	0.4	4.5	0	0	0	0	87
Other Spores	599	240	12.3	49.8	0	0	0	11	898
Total Spores	599	599	6373.5	7832.2	32	551	3101	9711	44596

Table 24. Summary statistics for 24-hr average pollen counts (number/m³) at the central Steubenville site.

Species	N	N_{nonzero}	Mean	S.D.	Min	Q25	Q50	Q75	Max
Ambrosia/Franseria	586	140	6.3	20.4	0	0	0	0	171
Betulaceae	586	73	2.7	20.0	0	0	0	0	441
Carya	586	65	1.1	4.5	0	0	0	0	70
Chenopod/Amaranth	586	36	0.2	1.0	0	0	0	0	13
Compositae	586	3	0.0	0.2	0	0	0	0	3
Cupressaceae	586	88	1.8	12.1	0	0	0	0	243
Cyperaceae	586	0	0.0	0.0	0	0	0	0	0
Fagus	586	7	0.1	0.6	0	0	0	0	11
Fraxinus	586	60	6.8	65.3	0	0	0	0	1294
Gramineae	586	178	4.2	12.5	0	0	0	3	120
Juglans	586	24	0.3	2.4	0	0	0	0	32
Liquidambar	586	18	0.2	1.5	0	0	0	0	21
Morus	586	56	2.5	13.1	0	0	0	0	152
Pinus	586	135	6.7	30.9	0	0	0	0	537
Plantago	586	125	1.2	2.9	0	0	0	0	21
Platanus	586	76	4.6	19.9	0	0	0	0	286
Populus	586	86	2.1	7.8	0	0	0	0	78
Quercus	586	93	25.4	137.2	0	0	0	0	2505
Rumex	586	15	0.2	1.7	0	0	0	0	37
Salix	586	36	0.5	2.2	0	0	0	0	24
Tilia	586	3	0.0	0.6	0	0	0	0	13
Typha	586	13	0.1	1.1	0	0	0	0	21
Ulmus	586	60	5.5	42.0	0	0	0	0	636
Urtica	586	20	0.1	0.8	0	0	0	0	11
Unidentified Pollen	586	164	3.9	12.4	0	0	0	3	123
Other Pollen	586	210	6.4	20.7	0	0	0	3	238
Total Pollen	586	445	83.0	241.0	0	3	13	55	3478

4.2.2.2.1 Seasonal Variability

Figure 14 presents boxplots showing $PM_{2.5}$, PM_{10} , and $PM_{10-2.5}$ concentrations at the central Steubenville site by month. Both $PM_{2.5}$ and PM_{10} exhibited a trend of higher summertime and lower wintertime concentrations. Median monthly $PM_{2.5}$ concentrations ranged from $24.0 \mu\text{g}/\text{m}^3$ in August to $11.3 \mu\text{g}/\text{m}^3$ in December, and median monthly PM_{10} concentrations ranged from $30.2 \mu\text{g}/\text{m}^3$ in August to $17.3 \mu\text{g}/\text{m}^3$ in December. The lowest monthly median $PM_{10-2.5}$ concentration ($5.3 \mu\text{g}/\text{m}^3$) also occurred during December; however, the greatest monthly median ($9.4 \mu\text{g}/\text{m}^3$) occurred during May rather than August. Despite the seasonal pattern exhibited by $PM_{2.5}$ and PM_{10} concentrations on average, episodically high daily concentrations (i.e., those greater than the 90th percentile concentrations of $34.7 \mu\text{g}/\text{m}^3$ and $50.5 \mu\text{g}/\text{m}^3$ for $PM_{2.5}$ and PM_{10} , respectively) were observed in all seasons, and the maximum 24-hr average concentrations observed during SCAMP for both of these particulate matter measures occurred during autumn rather than during summer. The maximum $PM_{10-2.5}$ concentration observed during summer was $23.5 \mu\text{g}/\text{m}^3$, whereas maxima during the other three seasons were all greater than $40 \mu\text{g}/\text{m}^3$.

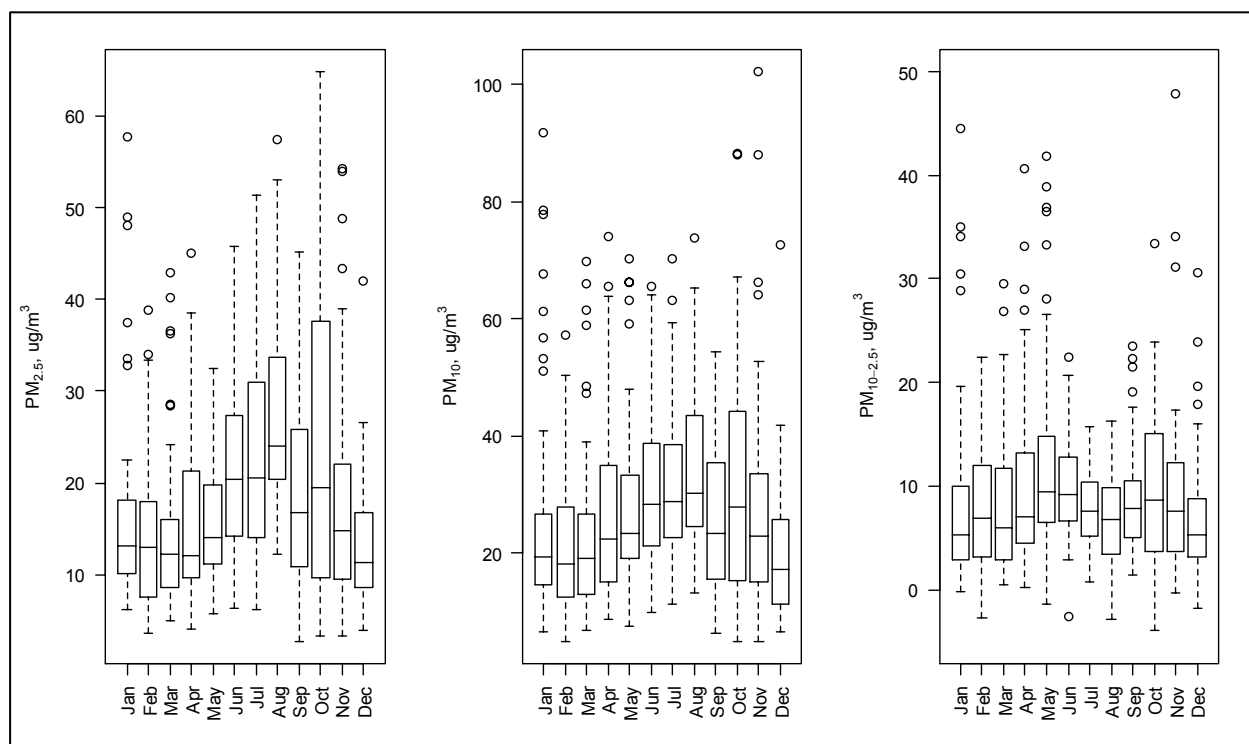


Figure 14. Boxplots summarizing the distributions of 24-hr average $PM_{2.5}$, PM_{10} , and $PM_{10-2.5}$ concentrations measured at the central Steubenville site by month, based on data collected between May 2000 and May 2002.

To further examine seasonal trends in concentrations of PM, PM components, and co-pollutants at the Steubenville site, median 24-hr average concentrations observed during the “warm season” were compared with median 24-hr average concentrations observed during the “cool season.” For the purposes of this comparison, the “warm season” was defined as April through September, and the “cool season” was defined as October through March. Median values were used because they provide an indication of typical concentration levels, but are not inflated by high-concentration outliers. Table 25 summarizes the average meteorological conditions observed at Steubenville during these seasons.

Table 25. Median 24-hr average meteorological conditions at Steubenville during the warm (April-September) and cool (October-March) seasons.

Parameter	Warm Season Median	Cool Season Median
Wind Speed (m/s)	1.99	2.83
Temperature (°C)	18.9	3.6
Relative Humidity (%)	75.19	70.78
Solar Radiation (W/m ²)	230	80
Barometric Pressure (mm Hg)	734.3	732.8

Figure 15 is a plot of the ratios of median warm season concentrations to median cool season concentrations for PM and gaseous pollutant variables that were measured on a daily frequency during SCAMP. For each variable, a nonparametric Wilcoxon rank sum test was applied to test whether the locations of the distributions of 24-hr average concentrations observed during the warm and cool seasons differed significantly (i.e., at a statistical significance level of $\alpha=0.05$). (It should be noted that because of the autocorrelated nature of PM_{2.5} data, the results of the statistical significance tests presented here may be subject to bias, as the tests assume independent, random samples). Concentrations of PM_{2.5}, PM₁₀, and PM_{10-2.5} were significantly higher during the warm season than during the cool season, although the difference was more pronounced for PM_{2.5} and PM₁₀ than for PM_{10-2.5}. These results are consistent with the conclusions drawn above from Figure 15. O₃ concentrations similarly were significantly greater during the warm season than during the cool season, reflecting the influence of seasonal variations in sunlight on concentrations of this photochemically-produced pollutant. Conversely, concentrations of SO₂, NO, and CO were significantly greater during the cool season than during the warm season. These gases are primary pollutants; hence, the seasonal variation in their concentrations may be attributable to seasonal variations in emissions or to seasonal variations in meteorological conditions such as wind direction and mixing height that affect transport and accumulation. Concentrations of NO₂ and NO_x (which includes NO and NO₂) did not vary significantly by season.

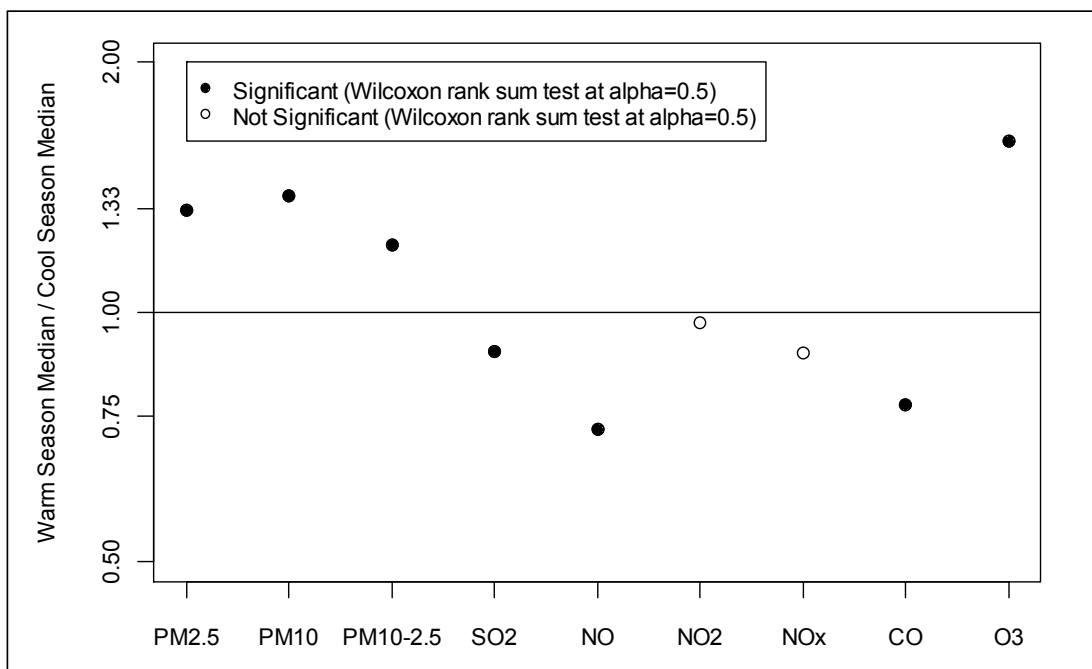


Figure 15. Ratios of warm-to-cool season median concentrations for PM and gaseous pollutants at the central Steubenville site. Solid points indicate statistically significant seasonal differences.

In order to preserve reasonable y-axis limits, pollen and spores are not shown in Figure 16; however, their concentrations were substantially and significantly higher during the warm season than during the cool season. The boxplots in Figure 16 show total pollen and spore counts by month at Steubenville. Pollen counts were greatest in April and May, when many trees and grasses in the region pollinate. A less pronounced local maximum is also evident during weed pollination season in August and September. Spore counts were greatest in August and lowest in January.

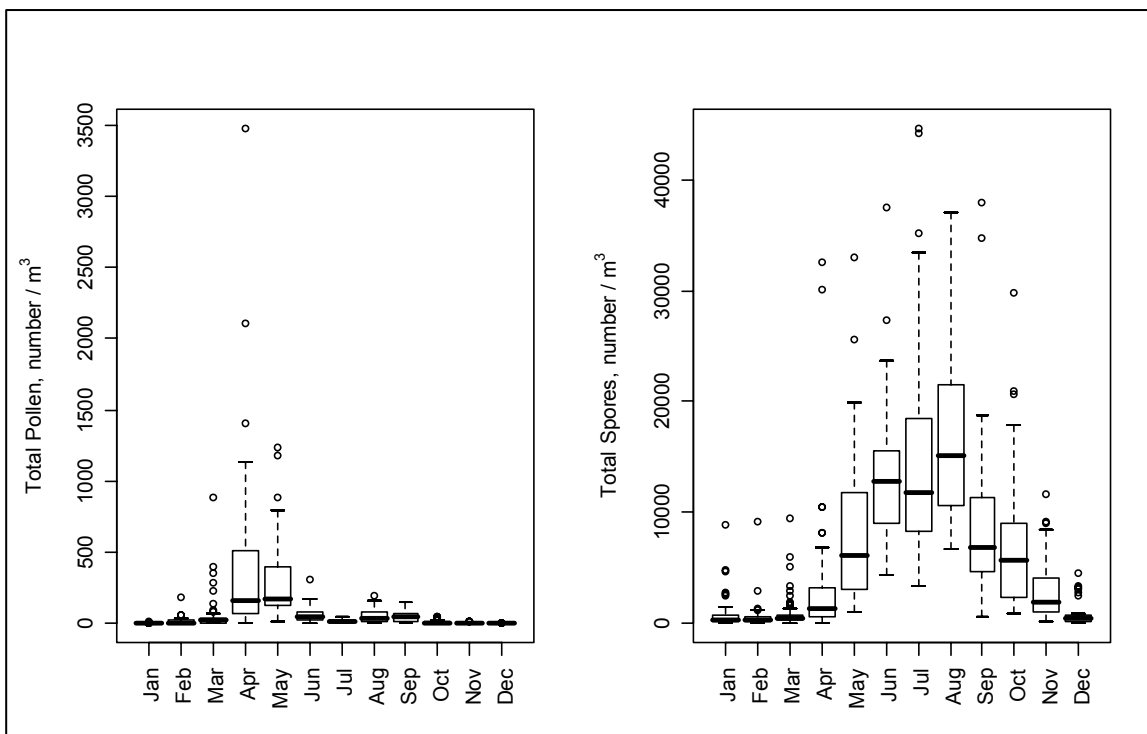


Figure 16. Boxplots summarizing the distributions of 24-hr average pollen and spore number concentrations measured at the central Steubenville site by month, based on data collected between June 2000 and May 2002.

Figure 17 shows ratios of median warm season concentrations to median cool season concentrations for $PM_{2.5}$ and PM_{10} ions and water-soluble elemental components that were measured every fourth day during SCAMP. Figure 18 shows these ratios for fine particulate EC, OM, and acid-digestible elements, which were also measured on a 1-in-4 day frequency. As is typical for the northeastern United States (Sickles, 1999), sulfate concentrations were significantly greater during the warm season than during the cool season at Steubenville, whereas nitrate concentrations were significantly greater during the cool season than during the warm season. The seasonal trend exhibited by sulfate (which is expected to be present primarily as $(NH_4)_2SO_4$ or as NH_4HSO_4) is likely due to the effect of increased summertime photochemical activity on concentrations of this predominantly secondary pollutant (Faust, 1994). Formation of particulate nitrate (which is expected to be present primarily as NH_4NO_3) is thermodynamically favored by low temperatures and high relative humidity and depends upon the availability of NH_3 (Pittsburgh Air Quality Study, 2003), explaining why NO_3^- concentrations are elevated during the cool season, when temperatures favor NH_4NO_3 formation and less NH_3 has reacted to form $(NH_4)_2SO_4$.

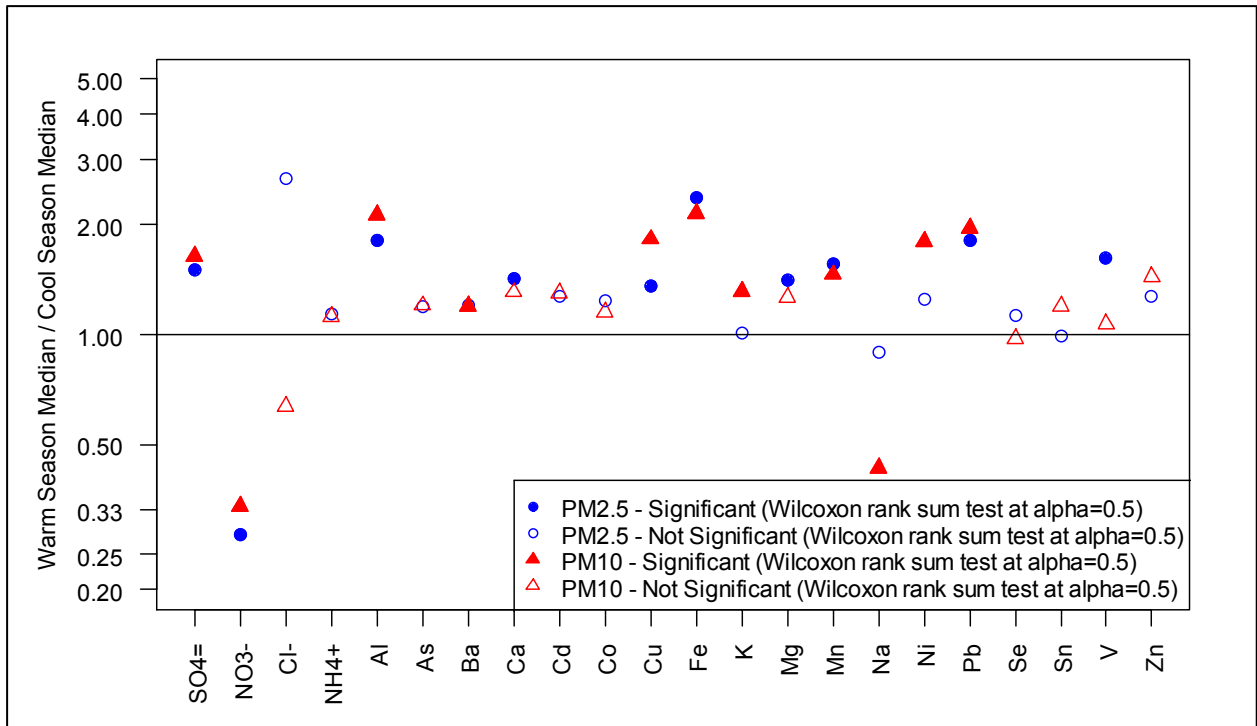


Figure 17. Ratios of warm-to-cool season median concentrations for PM_{2.5} and PM₁₀ water-extractable components at the central Steubenville site. Solid symbols indicate statistically significant seasonal differences.

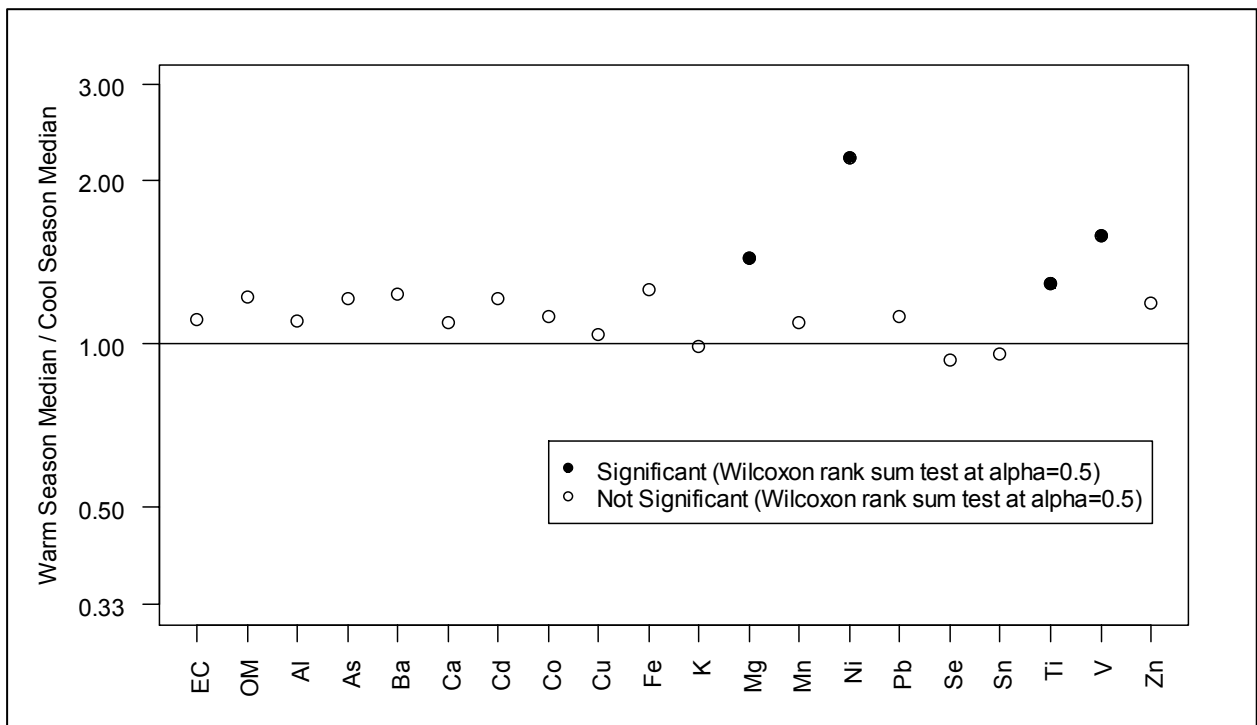


Figure 18. Ratios of warm-to-cool season median concentrations for PM_{2.5} carbonaceous and acid-digestible elemental components at the central Steubenville site. Solid symbols indicate statistically significant seasonal differences.

As a result of seasonal variability in the mass concentrations of its components, the composition of PM_{2.5} (measured on a percentage basis) exhibited similar seasonal variability. Figure 19 shows average PM_{2.5} composition at Steubenville by astronomical season. Calculations were performed using the mass balance procedure and 109-day dataset described earlier in Section 4.2.2.1. Sulfate and the unidentified components of PM_{2.5} showed a pronounced trend of accounting for a greater portion of total PM_{2.5} mass during summer and a lesser portion during winter, whereas nitrate showed the opposite trend. The trends displayed by SO₄²⁻ and NO₃⁻ are explained by the physical mechanisms described above that cause seasonal variations in the mass concentrations of these pollutants. The trend shown by the unidentified components likely reflects the increased contribution of particle-bound water to FRM-determined PM_{2.5} mass in Steubenville during summertime. Ammonium accounted for a slightly larger percentage of PM_{2.5} mass in summer than in the other three seasons, reflecting its association with SO₄²⁻ in ambient PM_{2.5}. The remaining major PM_{2.5} components each exhibited minimum percent contributions to total PM_{2.5} mass during summer. However, as shown in Figures 17 and 18, mass concentrations of these components did not vary significantly by season; hence, the observed compositional trend is instead attributable to increased total PM_{2.5} mass resulting from elevated contributions of sulfate, ammonium, and unidentified components in summertime. Sulfate was by far the most prevalent PM_{2.5} component in summer, when average PM_{2.5} mass concentrations were greatest, accounting for 38.4% of the total PM_{2.5} mass. The second largest contributor in this season was organic material, which accounted for 21.7% of the total mass. Conversely, in winter, when average PM_{2.5} concentrations were lowest, OM accounted for a greater percentage of PM_{2.5} mass (27.5%) than did SO₄²⁻ (24.8%).

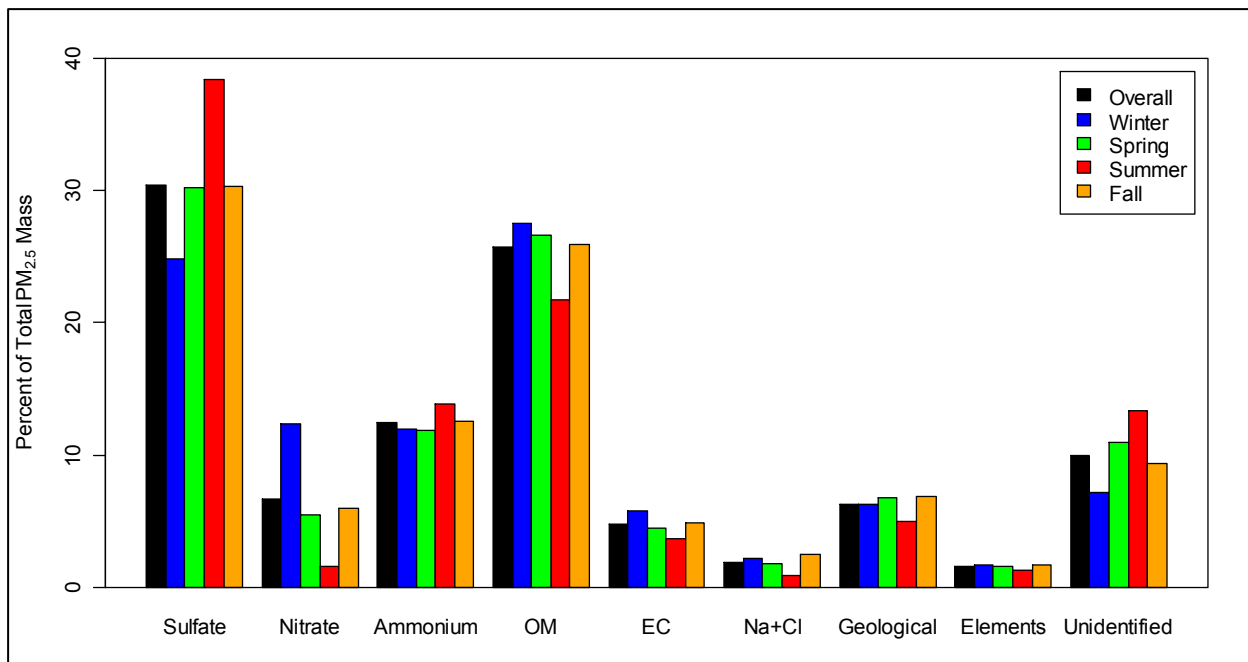


Figure 19. Overall average PM_{2.5} composition by season at the central Steubenville site, based on the application of a mass balance (described earlier) to 109 days with complete or near-complete PM_{2.5} speciation data.

As shown in Figures 17 and 18, concentrations of a number of elemental components of PM_{2.5} and PM₁₀, including Al, Ba, Ca, Cu, Fe, Mg, Mn, Pb, and V in the water-extractable PM_{2.5} fraction, Al, Ba, Cu, Fe, K, Mn, Ni, and Pb in the water-extractable PM₁₀ fraction, and Mg, Ni, Ti, and V in the acid-digestible PM_{2.5} fraction, were significantly greater during the warm season

than during the cool season at Steubenville. This trend may be attributable to seasonal variations in primary emissions of these elements, although it is observed for a diverse array of elements that likely originate from different sources. Alternatively, it may result from seasonal variations in ventilation conditions or transport direction that cause variations in the extent to which one or more source regions impact the monitoring site. For elements in the water-soluble PM fraction, another plausible explanation is that certain PM_{2.5} constituents (e.g., secondary sulfates) that are more prevalent during summertime than during wintertime interact with ambient metals to increase their solubility. Connell et al. (2006) examined seasonal variations in trace element fractional solubilities observed at the Steubenville site during SCAMP, and reported that fractional solubilities of Al, Fe, and Pb were significantly greater during the warm season than during the cool season. Moreover, the fractional solubility of Fe exhibited moderate, positive correlations ($r_s = 0.66-0.72$) with SO₄²⁻ and temperature during the summertime, when photochemical activity and secondary SO₄²⁻ concentrations are maximal, whereas correlations between Fe fractional solubility and temperature and SO₄²⁻ were comparatively weak ($r_s = -0.21-0.17$) during wintertime. Ghio et al. (1999b) reported that in controlled laboratory experiments, aqueous SO₄²⁻ was capable of mobilizing soluble Fe from Fe₂O₃, but only in the presence of light. Results from SCAMP are consistent with the possibility that fine particulate Fe in Steubenville may be solubilized to some extent by interaction with sulfates under suitable photochemical conditions, and reinforce Grahame and Schlesinger's (2005) recommendation for further research to examine the effect of secondary sulfates on the solubility of PM_{2.5} metals.

Aside from NO₃⁻, the only PM component with significantly greater ambient concentrations during the cool season than during the warm season at Steubenville was Na in the water-soluble PM₁₀ fraction. This difference may reflect the use of NaCl to treat icy roads in Steubenville during winter. The median ambient PM₁₀ Cl⁻ concentration was also greater during the cool season than during the warm season in Steubenville, consistent with this hypothesis, although the seasonal difference in Cl⁻ concentrations was not statistically significant. Ambient concentrations of water-soluble Na and Cl⁻ in PM_{2.5} at Steubenville showed a less-pronounced or different seasonal trend, suggesting that different sources account for the presence of these species in the fine and coarse particle fractions.

4.2.2.2 Day-of-Week Variability

Figure 20 presents boxplots showing PM_{2.5}, PM₁₀, and PM_{10-2.5} concentrations at the central Steubenville site by day-of-week. PM_{2.5} concentrations did not exhibit a pronounced day-of-week trend during SCAMP; the greatest median concentration by day-of-week was 16.4 µg/m³ on Monday, and the lowest was 13.9 µg/m³ on Friday. Conversely, PM₁₀ and PM_{10-2.5} concentrations exhibited a trend of higher weekday (i.e., Monday-Friday) and lower weekend (i.e., Saturday-Sunday) concentrations. Median Saturday and Sunday concentrations ranged from 19.0-19.9 µg/m³ for PM₁₀ and from 5.8-5.9 µg/m³ for PM_{10-2.5}, whereas median weekday concentrations ranged from 22.3-26.8 µg/m³ for PM₁₀ and from 7.6-8.6 µg/m³ for PM_{10-2.5}.

Figure 21, which shows ratios of weekday median concentrations to weekend median concentrations for variables that were measured on a daily frequency during SCAMP, echoes these observations. Concentrations of PM₁₀ and PM_{10-2.5}, as well as those of NO and NO_x, were significantly greater on weekdays than on weekends during SCAMP (based on the results of Wilcoxon rank sum tests), whereas concentrations of PM_{2.5}, SO₂, NO₂, CO, O₃, total spores, and total pollen did not show significant weekday/weekend differences. Figure 22 shows ratios of median weekday concentrations to median weekend concentrations for PM_{2.5} and PM₁₀ ions and water-soluble elemental components that were measured every fourth day

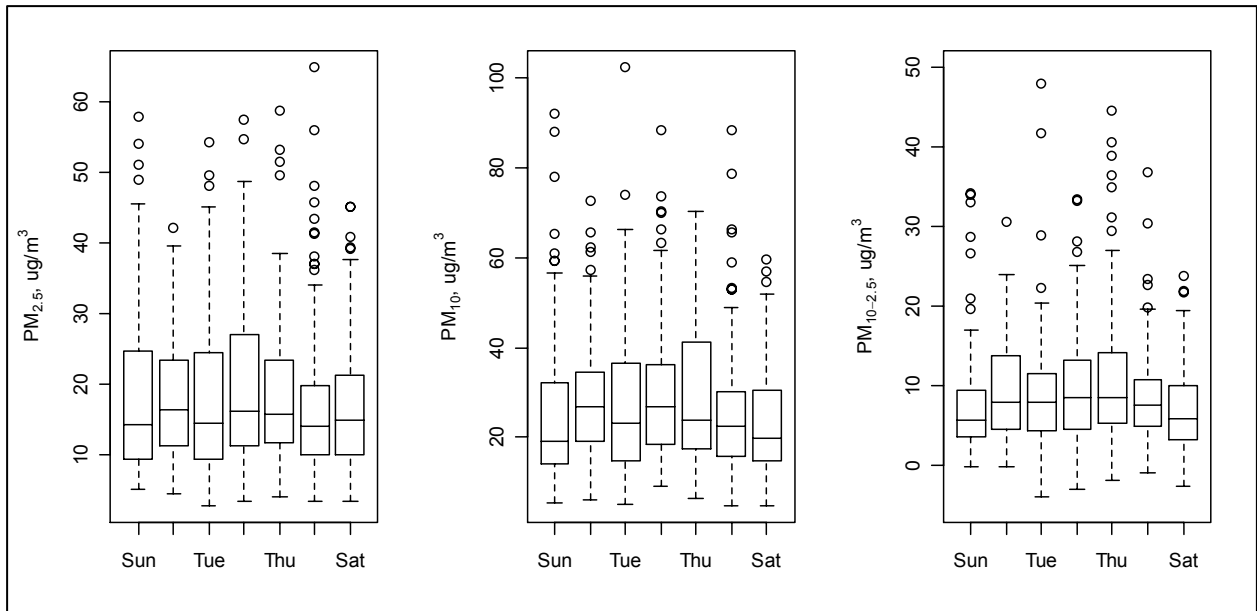


Figure 20. Boxplots summarizing the distributions of 24-hr average PM_{2.5}, PM₁₀, and PM_{10-2.5} concentrations measured at the central Steubenville site by day of week, based on data collected between May 2000 and May 2002.

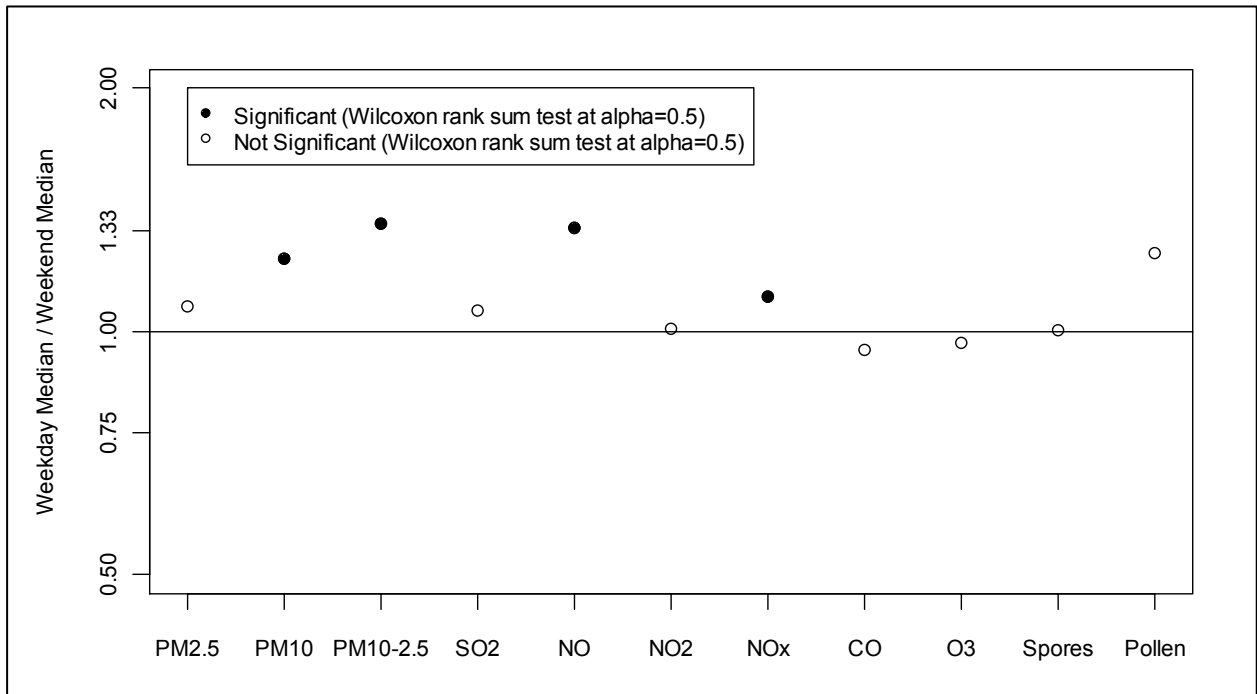


Figure 21. Ratios of weekday-to-weekend median concentrations for PM and gaseous pollutants at the central Steubenville site. Solid points indicate statistically significant weekday/weekend differences.

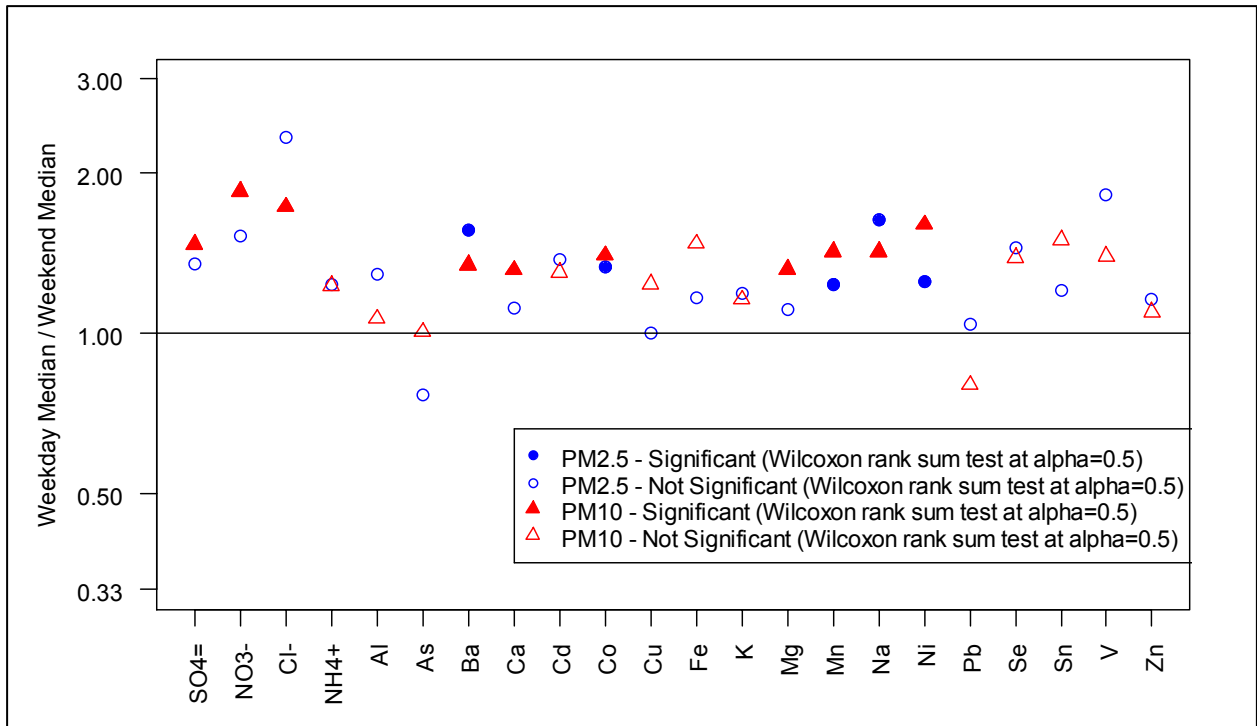


Figure 22. Ratios of weekday-to-weekend median concentrations for PM_{2.5} and PM₁₀ water-extractable components at the central Steubenville site. Solid symbols indicate statistically significant weekday/weekend differences.

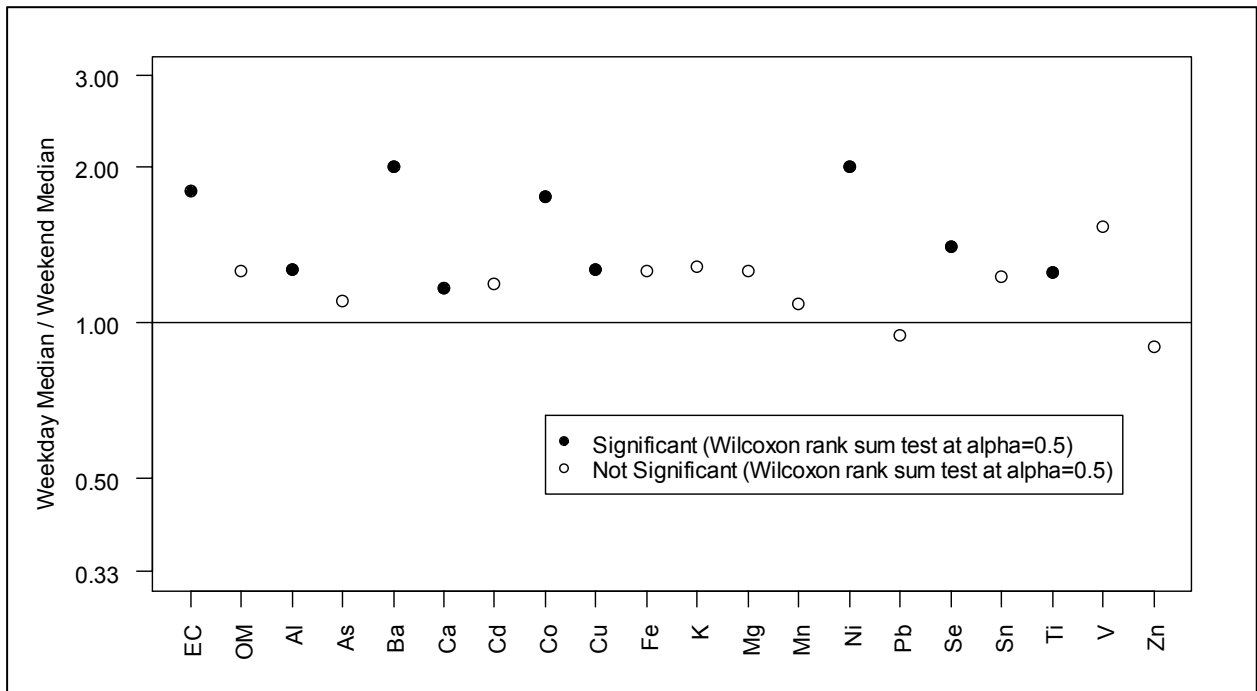


Figure 23. Ratios of weekday-to-weekend median concentrations for PM_{2.5} carbonaceous and acid-digestible elemental components at the central Steubenville site. Solid points indicate statistically significant weekday/weekend differences.

during SCAMP, and Figure 23 shows these ratios for fine particulate EC, OM, and acid-digestible elements, which were also measured on a 1-in-4 day frequency. A majority of the $PM_{2.5}$ and PM_{10} components measured during SCAMP exhibited greater median concentrations during weekdays than during weekends; these weekday/weekend differences were statistically significant in a number of cases. Because of the relatively small number of samples collected for variables that were sampled on a 1-in-4 day frequency, the observed weekend/weekday differences may be an anomaly of the subset of days sampled in some cases. (For example, weekday/weekend differences for $PM_{2.5}$, SO_2 , and NO_2 , which were not statistically significant when all daily data collected during SCAMP were used in the analyses, became statistically significant when only data collected every fourth day were used). Nevertheless, the trend of higher weekday and lower weekend concentrations is likely real for some of the parameters (e.g., for PM_{10} , $PM_{10-2.5}$, and NO_x , which exhibited this trend regardless of whether 1-in-1 or 1-in-4 day data were used), and the direction of this trend was consistent in all cases in which a significant difference was found (i.e., no parameter exhibited significantly greater concentrations during weekends than during weekdays). Unlike seasonal trends, day-of-week trends are not expected to result from variations in meteorology. Weekday and weekend values did not differ significantly for any of the meteorological parameters monitored during SCAMP, regardless of whether daily or 1-in-4 day data were used. Rather, the day-of-week trends observed during SCAMP likely resulted from increased anthropogenic emissions (e.g., vehicular or industrial) during the weekdays.

4.2.2.2.3 Diurnal Variability

As discussed earlier, $PM_{2.5}$ total mass and gaseous pollutant concentrations were monitored continuously at the central site in Steubenville, permitting their diurnal variability to be explored. Figure 24 includes a boxplot showing the distribution of $PM_{2.5}$ concentrations observed at Steubenville for each hour of the day during SCAMP, as well as a line plot showing mean $PM_{2.5}$ concentrations by time-of-day and season. Figure 25 shows mean gaseous pollutant concentrations observed at the central Steubenville site by time-of-day and season.

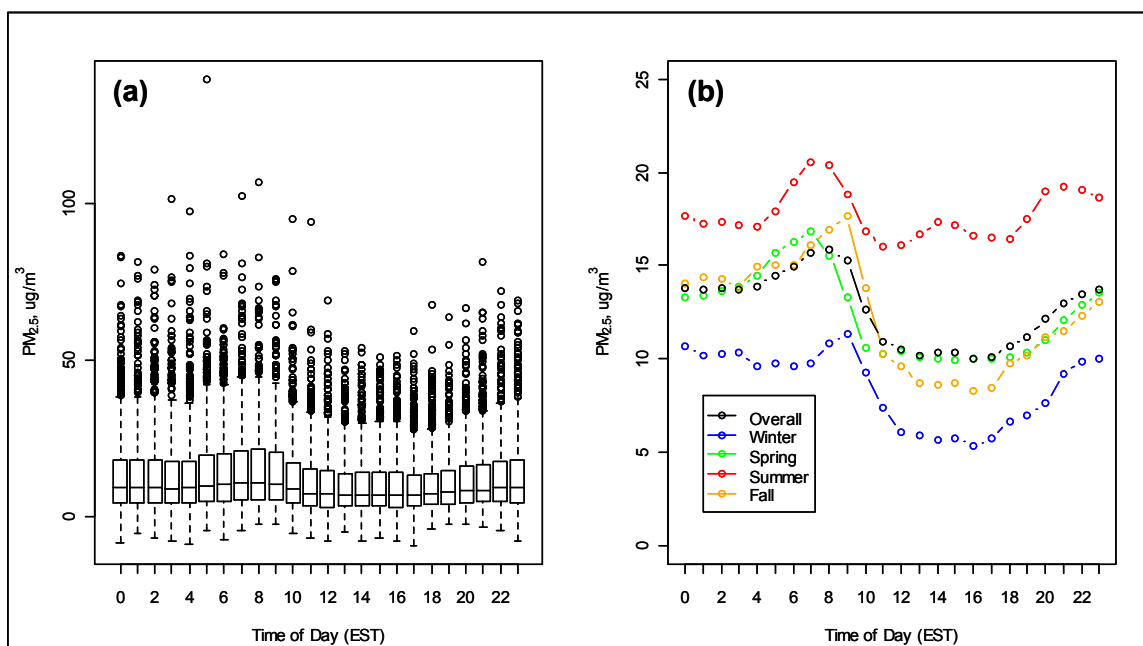


Figure 24. Diurnal profiles of $PM_{2.5}$ at Steubenville: (a) boxplot showing distribution of $PM_{2.5}$ concentrations by hour of day, and (b) mean $PM_{2.5}$ concentrations by hour of day and season.

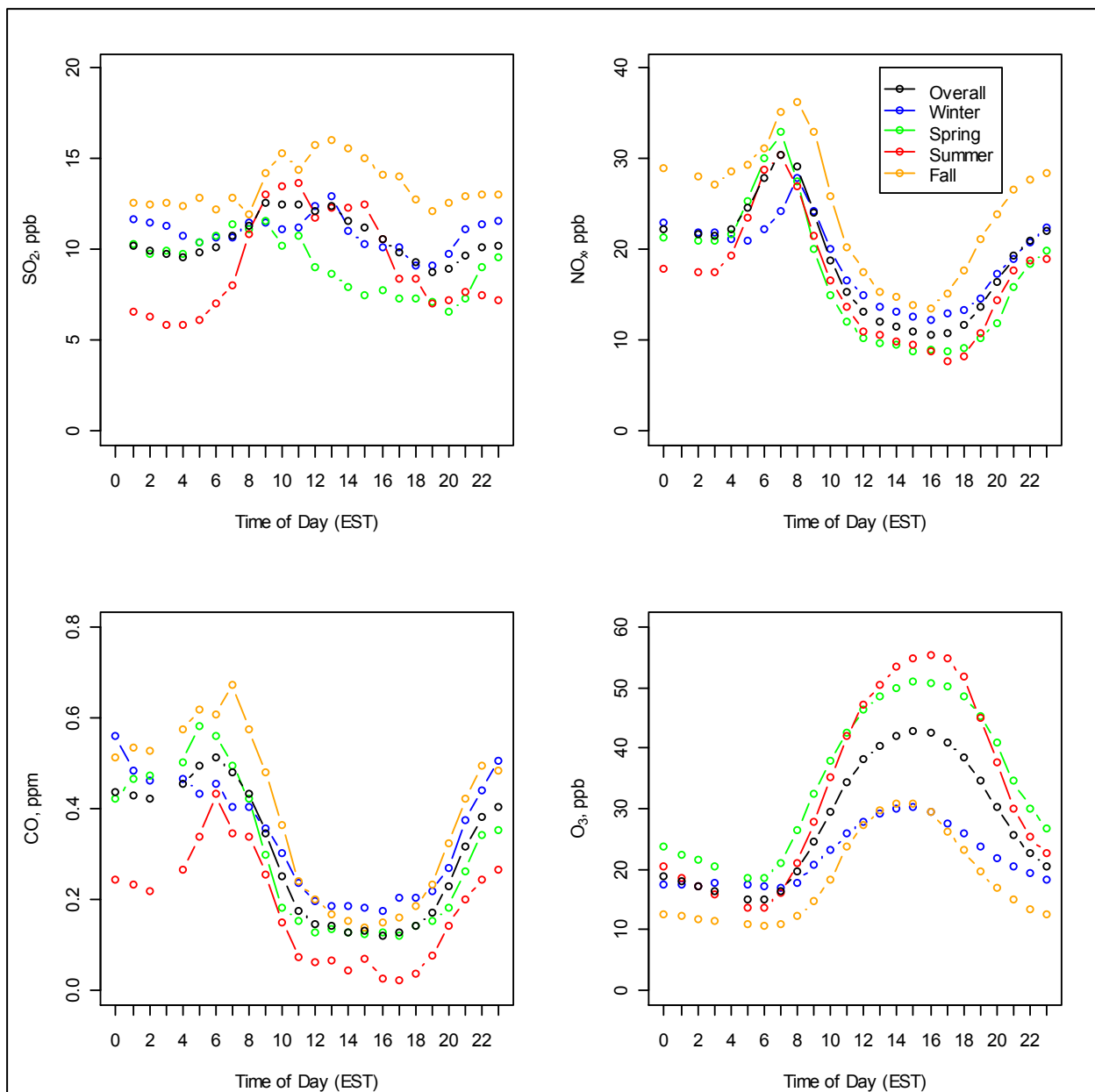


Figure 25. Concentrations of SO_2 , NO_x , CO , and O_3 by hour of day and season at the central Steubenville site.

Although the diurnal profile of $\text{PM}_{2.5}$ concentration was highly variable, as evidenced in Figure 24, on average it exhibited a trend of moderate overnight concentrations building up to a peak concentration of $15.8 \mu\text{g}/\text{m}^3$ (measured using the TEOM) at 8:00 a.m. EST, and then declining to a minimum concentration of $10.0 \mu\text{g}/\text{m}^3$ at 4:00 p.m. EST. (The shift in the timing of the morning maximum in spring and summer relative to fall and winter evident in Figure 24(b) likely reflects Steubenville's shift from Eastern Standard Time to Eastern Daylight Time during most of spring and all of summer). This trend is similar to the diurnal profiles exhibited by NO_x and CO , which on average reached maximum concentrations at 7:00 a.m. and 6:00 a.m. EST, respectively, and minimum concentrations at 4:00 p.m. EST. O_3 , a photochemically-produced secondary pollutant, exhibited an opposite diurnal pattern of maximal concentrations during mid-afternoon, when photochemical activity peaks, and minimal concentrations during early

morning, when photochemical production ceases and O₃ depletion by deposition and reaction with freshly-emitted pollutants such as NO occurs rapidly. Even though a large portion of PM_{2.5} mass during summer in Steubenville is expected to result from secondary sulfates, PM_{2.5} mass concentrations during that season still exhibited a diurnal trend characterized by an early morning peak in concentration; although the minimum concentration occurred during late morning (11:00 a.m. EST) rather than late afternoon, and a small local maximum was observed during mid-afternoon, likely reflecting the photochemical formation of secondary aerosol.

The diurnal behavior exhibited by PM_{2.5} in Steubenville likely results from diurnal variability in both emissions and meteorology. The close semblance between the average diurnal profiles of PM_{2.5}, NO_x, and CO likely reflects the influence of a common emission source of these pollutants. Motor vehicles are one known source of CO, NO_x, and PM_{2.5} emissions; Chuersuwan et al. (2000) attributed an average morning (6:00 a.m. to 9:30 a.m. EST) peak in PM_{2.5} concentrations in New Jersey to the influence of local motor vehicle traffic. Statewide average traffic count data from the Ohio Department of Transportation (2004) suggest that automobile traffic on urban principal arterial roadways increases drastically from about 5:00 a.m. to about 7:00 a.m., similar to the timing of the build-up in concentration leading to the early morning maxima in PM_{2.5}, CO, and NO_x concentrations. Hence, it is likely that automotive emissions contribute at least in part to the average morning peak in PM_{2.5} concentration. However, unlike traffic counts, which remain elevated throughout midday and reach an even greater maximum during late afternoon rush hour before declining dramatically overnight, PM_{2.5}, CO, and NO_x concentrations on average exhibit minima during afternoon and more elevated concentrations overnight. This suggests that factors other than motor vehicles also influence their diurnal profiles. Based on 2001 estimates from the U.S. EPA (2001), the primary metals industry and electric and industrial fossil fuel combustion sources are the largest sources of CO and NO_x, respectively, in Jefferson County, OH, where Steubenville is located. Hence, these sources, which also emit PM_{2.5}, likely influence the diurnal behavior of these pollutants. Based on Connell et al.'s (2005c) analyses of hourly data from the SCAMP central Steubenville site, the trend of higher overnight and lower midday PM_{2.5}, NO_x, and CO concentrations likely results from the effect of nocturnal temperature inversions and reduced overnight mixing heights, which trap primary emissions of these emissions close to ground level, allowing them to accumulate. Greater mixing heights during midday allow pollutants to disperse through a larger volume of air, resulting in lower ambient concentrations. Based on the data plotted in Figure 24, this trend is most pronounced during the fall and winter, when the morning maxima in the diurnal profiles are more than twice as great as the afternoon minima, and least pronounced during the summer, when the maximum and minimum in the diurnal profile differ by a factor of only 1.3.

Connell et al. (2005c) also noted that the highest 1-hr PM_{2.5} concentrations (i.e., those greater than 65 µg/m³) observed during SCAMP occurred during the early morning and, to a lesser extent, the evening hours, but were never observed during the mid-afternoon between 1:00 and 5:00 p.m. EST. Moreover, the authors observed that the number and severity (measured by the peak 1-hr average concentration) of transient episodes of elevated hourly PM_{2.5} concentration were greater during fall than during summer. These results, coupled with the observations made above, are consistent with the hypothesis that whereas secondary sulfates and other photochemically produced PM_{2.5} components in general promote a chronic increase in PM_{2.5} concentrations across all hours of the day (see Figure 24) during summertime in Steubenville, it is primary PM_{2.5} emitted locally during overnight and early morning periods of reduced mixing height that results in the highest short-term (e.g., hourly) PM_{2.5} concentrations in that city.

4.2.2.3 Comparison with Benchmark Values

To place the concentrations of PM_{2.5}, PM_{2.5} component, and co-pollutant concentrations measured at Steubenville in proper perspective, these concentrations must be compared with regulatory, historic, and other benchmarks. Ambient air concentrations of several of the pollutants measured at the Steubenville site are regulated by National Ambient Air Quality Standards. Moreover, concentrations of a number of the species monitored during SCAMP were measured and reported previously at Steubenville as part of the Harvard Six Cities Study, and several of the trace element components of PM_{2.5} are listed in the Clean Air Act as Hazardous Air Pollutants (HAPs) and have established reference concentrations denoting concentration levels at or below which no adverse health effects (non-cancer) are anticipated as a result of chronic inhalation exposure. Finally, all of the elemental components of PM_{2.5} measured at Steubenville are present to some extent in the earth's crust; comparing the abundances of these elements in ambient PM_{2.5} with their crustal abundances can provide some indication of the extent to which their ambient concentrations are enriched by anthropogenic activities. This section compares pollutant concentrations measured at Steubenville during SCAMP with various benchmark values.

4.2.2.3.1 Comparison with National Ambient Air Quality Standards

SCAMP data were not collected for the purpose of determining attainment with National Ambient Air Quality Standards, and did not meet all of the criteria required to determine attainment (e.g., 3-year collection period for certain standards). Nevertheless, comparing concentrations measured during SCAMP with NAAQS values helps to establish a context for air quality in Steubenville for the purposes of this report. Table 26 summarizes the primary National Ambient Air Quality Standards that have been set by the U.S. EPA Office of Air Quality Planning and Standards (OAQPS) (U.S. EPA, 2006), which are designed to protect public health. Secondary standards (designed to protect public welfare) are the same as primary standards for lead, NO₂, annual PM₁₀, annual PM_{2.5}, and O₃; sulfur oxides are subject to a secondary 3-hour standard of 0.5 ppm.

Table 26. Summary of primary National Ambient Air Quality Standards (U.S. EPA, 2006).

Pollutant	Primary Standard	Averaging Time	Notes
Carbon Monoxide	9 ppm	8-hour	Not to be exceeded more than once/year
	35 ppm	1-hour	
Lead	1.5 µg/m ³	Quarterly Average	
Nitrogen Dioxide	0.053 ppm	Annual	Arithmetic Mean
PM ₁₀	50 µg/m ³	Annual	3-yr average of weighted annual mean
	150 µg/m ³	24-hour	Not to be exceeded more than once/year
PM _{2.5}	15.0 µg/m ³	Annual	3-yr average of weighted annual mean
	65 µg/m ³	24-hour	3-yr average of the 98 th percentile of 24-hr concentrations
Ozone	0.08 ppm	8-hour	3-yr average of 4 th highest daily maximum 8-hr average
Sulfur Dioxide	0.03 ppm	Annual	Arithmetic Mean
	0.14 ppm	24-hour	Not to be exceeded more than once/year

Data collected as part of SCAMP suggest that Steubenville is at risk of violating the annual PM_{2.5} NAAQS and the 8-hour ozone NAAQS. Although based on only two years of monitoring data, the mean PM_{2.5} concentration measured at Steubenville during SCAMP (18.4 µg/m³) was 3.4 µg/m³ greater than the annual PM_{2.5} NAAQS of 15 µg/m³, and 8-hour average ozone concentrations greater than 0.08 ppm were observed on nine days during the two-year monitoring period. Steubenville is currently designated as a nonattainment area under these two standards. No 24-hr average PM_{2.5} concentration measured at Steubenville during SCAMP exceeded the 24-hr PM_{2.5} NAAQS of 65 µg/m³; the 98th percentile concentration observed during the program (49.1 µg/m³) was well below the current 24-hr NAAQS, but above the revised 24-hour standard of 35 µg/m³ recently proposed by the U.S. EPA (71 FR 2620).

Concentrations of CO, NO₂, PM₁₀, and SO₂ observed at Steubenville were well below the NAAQS values for these pollutants. As shown in Table 22, no 1-hr average CO concentration during SCAMP exceeded the 1-hr CO standard of 35 ppm or the 8-hr standard of 9 ppm, and the average NO₂ concentration was less than ¼ as much as the annual NO₂ standard of 0.053 ppm. Mean and maximum PM₁₀ concentrations (27.4 and 102.2 µg/m³, respectively) were substantially less than the annual and 24-hour standards for this pollutant. (The maximum 24-hr average PM_{10-2.5} concentration of 47.9 µg/m³ was also substantially less than the recently proposed 24-hr standard of 70 µg/m³ for inhalable coarse particles). The overall average SO₂ concentration at Steubenville was about 65% less than the annual SO₂ NAAQS, and maximum 24-hr and 1-hr average concentrations of 0.0601 and 0.2339 ppm, respectively, were appreciably less than the 24-hr primary NAAQS and 3-hr secondary NAAQS for this pollutant. Lead was not measured in TSP at Steubenville, preventing a comparison with the Pb NAAQS.

4.2.2.3.2 Comparison with Six Cities Study Concentrations

Table 27 compares concentrations of ambient air pollutants measured at Steubenville between 2000 and 2002 as part of SCAMP with concentrations measured there in the late 1970s and 1980s as part of the Harvard Six Cities Study (Dockery et al., 1993; Laden et al., 2000).

Table 27. Comparison of mean pollutant concentrations observed at the SCAMP central Steubenville site (2000-2002) with mean pollutant concentrations observed in Steubenville during the Six Cities Study. PM_{2.5} total mass concentrations are in µg/m³; PM_{2.5} component concentrations are in ng/m³, and gas concentrations are in ppb.

Pollutant	Six Cities Mean	Six Cities Period	SCAMP Mean	% Change
PM _{2.5} Total Mass	30.5 ^a	1979-1988	18.4	-40%
PM _{2.5} Al	186.8 ^a	1979-1988	97.7	-48%
PM _{2.5} Ca	102 ^a	1979-1988	136	+33%
PM _{2.5} Cu	11.9 ^a	1979-1988	3.2	-73%
PM _{2.5} Fe	542 ^a	1979-1988	272	-50%
PM _{2.5} K	344 ^a	1979-1988	91	-74%
PM _{2.5} Mn	30.4 ^a	1979-1988	14.6	-52%
PM _{2.5} Ni	3.7 ^a	1979-1988	1.1	-70%
PM _{2.5} Pb	184.5 ^a	1979-1988	15.3	-92%
PM _{2.5} Se	5.2 ^a	1979-1988	3.3	-37%
PM _{2.5} SO ₄ ²⁻	12745 ^{a,c}	1979-1988	5800	-54%
PM _{2.5} V	10.5 ^a	1979-1988	1.5	-86%
PM _{2.5} Zn	138.4 ^a	1979-1988	84.4	-39%
SO ₂	24.0 ^b	1977-1985	10.5	-56%
NO ₂	21.9 ^b	1977-1985	12.7	-42%
O ₃	22.3 ^b	1977-1985	28.4	+27%

^aSource: Laden et al., 2000; ^bSource: Dockery et al., 1993; ^cEstimated from XRF sulfur

As shown in the table, concentrations of all of the pollutants examined except for O₃ and fine particulate Ca had lower measured means in Steubenville during SCAMP than during the Harvard Six Cities Study. These differences may be due in part to analytical error, as the two monitoring campaigns used different sampling and analytical techniques. However, as discussed by Connell et al. (2005a), the widespread trend of decreasing pollutant concentrations observed at Steubenville is likely related to declines in population and industrial activity in Steubenville, leading to corresponding declines in emissions, as well as to advances in the application of air pollution control technology, spurred in part by the 1990 Clean Air Act amendments and other environmental regulations. Although PM_{2.5} concentrations in Steubenville remain above the annual PM_{2.5} NAAQS, the data presented in Table 27 suggest that the average total mass of ambient PM_{2.5} in that city has decreased by about 40% since the Six Cities Study was conducted. Concentrations of SO₄²⁻, the largest chemical component of PM_{2.5} in Steubenville, and of SO₂, its gaseous precursor, showed even larger decreases of 54-56%. Fine particulate Pb exhibited the largest relative decrease in ambient concentration of all of the pollutants compared in Table 27, likely reflecting the discontinued use of leaded gasoline (Connell et al., 2006).

4.2.2.3.3 Comparison with Health Reference Concentrations

Seven of the trace elements (As, Cd, Co, Mn, Ni, Pb, and Se) determined in PM_{2.5} at Steubenville during SCAMP are listed as Hazardous Air Pollutants (HAPs) in the 1990 Clean Air Act amendments. To obtain some indication of the potential for adverse health effects resulting from these elements in PM_{2.5} in Steubenville, mean ambient air concentrations of these elements measured during SCAMP were compared with non-cancer chronic inhalation reference concentrations (RfCs), minimal risk levels (MRLs), and reference exposure levels (RELs) established by the U.S. Environmental Protection Agency (EPA), U.S. Agency for Toxic Substances and Disease Registry (ATSDR), and California EPA, respectively (U.S. EPA, 2005b). These reference values represent concentration levels at or below which no adverse non-cancer health effects are anticipated as a result of chronic exposure. Although the reference values are based upon total airborne concentrations of the elements of interest, and not only the fine particulate fraction, they provide some basis for making a preliminary assessment.

Table 28 compares mean ambient air concentrations of the seven HAP elements determined in the acid-digestible PM_{2.5} fraction during SCAMP with the most stringent non-cancer reference concentrations established for these elements by one of the agencies listed above. For all seven elements, mean ambient air concentrations were below the non-cancer reference concentrations. Fine particulate manganese had the highest average concentration relative to the reference concentrations; the mean PM_{2.5} Mn concentration at Steubenville was about 37% as great as the chronic inhalation MRL. Mean concentrations of As, Cd, Ni, and Pb were 1/10 to 1/100 as much as the most stringent reference concentrations, and mean concentrations of Co and Se were less than 1/1000 as much as the reference concentrations. Based upon this preliminary analysis, the risk of non-cancer adverse health effects resulting from chronic exposure to any of these single elemental constituents of PM_{2.5} in Steubenville appears to be low. However, it is important to note that based on the data presented in Table 16, fine particles on average contributed only about 39% of the mass of water-extractable Mn measured in PM₁₀ in Steubenville. Hence, it is possible that if all particle sizes were considered, total airborne Mn in Steubenville may approach or exceed the chronic inhalation MRL. Further research should be conducted to determine whether there is any risk of adverse health effects associated with total suspended Mn in Steubenville.

Table 28. Comparison of mean ambient air HAP concentrations in PM_{2.5} at Steubenville during SCAMP with non-cancer chronic inhalation RfCs, MRLs and RELs.

Element	Mean Ambient Air Concentration in Acid-Digestible PM _{2.5} Fraction (ng/m ³)	Most Stringent Non-Cancer Reference Concentration (ng/m ³)
As	1.64	30 (REL)
Cd	0.46	20 (REL)
Co	0.065	100 (MRL)
Mn	14.6	40 (MRL)
Ni	1.1	50 (REL)
Pb	15.3	1,500 (RfC)
Se	3.30	20,000 (REL)

No chronic inhalation reference concentrations have been adopted by the agencies named above for transition metals such as Fe and Zn; however, as discussed by Connell et al. (2006), airborne concentrations of these elements, which have been implicated in PM_{2.5} toxicology studies, in PM_{2.5} at Steubenville during SCAMP were appreciably greater than concentrations reported recently for other U.S. Cities (Appendix D). Further research should be conducted to better elucidate whether these metals play any role in the health effects previously associated with PM_{2.5} in Steubenville.

In addition to the non-cancer reference concentrations discussed above, the U.S. EPA (2005b) identifies As as a carcinogen and Cd as a probable carcinogen. Based upon the chronic inhalation Unit Risk Estimates (UREs) that have been established for these elements and the range of mean ambient concentrations determined in the water-extractable PM_{2.5} fraction, acid-digestible PM_{2.5} fraction, and water-extractable PM₁₀ fraction during SCAMP (both elements appeared to be largely water-extractable and largely present in the fine particle fraction), mean airborne As concentrations observed at Steubenville during SCAMP correspond to a cancer risk of about 7 in 1,000,000 to 11 in 1,000,000, and mean airborne Cd concentrations correspond to a cancer risk of about 1 in 1,000,000 or less. (The risk estimates for As should be interpreted with some caution, as the greatest mean ambient concentration and hence the greatest cancer risk was observed for the water-extractable PM_{2.5} fraction, which is physically expected to have the lowest ambient concentration. This incongruity may indicate the presence of sampling or analytical error, which would bias the risk estimate).

4.2.2.3.4 Comparison with Crustal Abundances

Although natural abundances of trace elements in the Earth's crust are very small, anthropogenic activities (or natural processes such as geologic weathering, volcanic eruptions, wildfires, etc.) can enrich the abundances of these elements in ambient PM_{2.5}. In order to ascertain the extent to which such enrichment occurs in Steubenville, enrichment factors were calculated for each of the 18 elements measured in the acid-digestible PM_{2.5} fraction during SCAMP. For this analysis, the enrichment factor (EF) was computed as:

$$EF(X) = \frac{(X / Al)_{PM2.5}}{(X / Al)_{crustal}},$$

where X is the concentration of the element for which an enrichment factor is being computed, (X / Al)_{PM2.5} is the ratio of that element's mean mass concentration in PM_{2.5} at Steubenville to the

mean mass concentration of Al in PM_{2.5} at Steubenville, and $(X / Al)_{\text{crustal}}$ is the ratio of that element's average crustal concentration to the average crustal concentration of Al. Hence, it is assumed that most of the Al present in PM_{2.5} at Steubenville is associated with crustal material. Lacking data specific to Steubenville, the average crustal composition reported by Mason (1966) was used to compute values of $(X / Al)_{\text{crustal}}$.

Figure 26 shows PM_{2.5} enrichment factors at Steubenville for the 18 elements measured during SCAMP. The EFs for As, Cd, Pb, Sn, Zn, and especially Se were markedly high, ranging from 565 for Sn to almost 55,000 for Se. All of these elements are expected in emissions from industrial sources (e.g., coal-fired power plants, metal smelting and processing plants, coke plants, incinerators, etc.) that are prevalent in the Steubenville area; the large EFs likely reflect major anthropogenic contributions to their ambient concentrations. As expected, the most abundant crustal elements (i.e., Fe, Ca, K, Mg, Ti) had much lower enrichment factors than the trace metal species, ranging from 1.9 for Mg to 4.5 for Fe, although it is noteworthy that the Al-normalized EFs for all of these elements were greater than one. Probable emission sources of these enriched elements are discussed later in this report in the section on source apportionment.

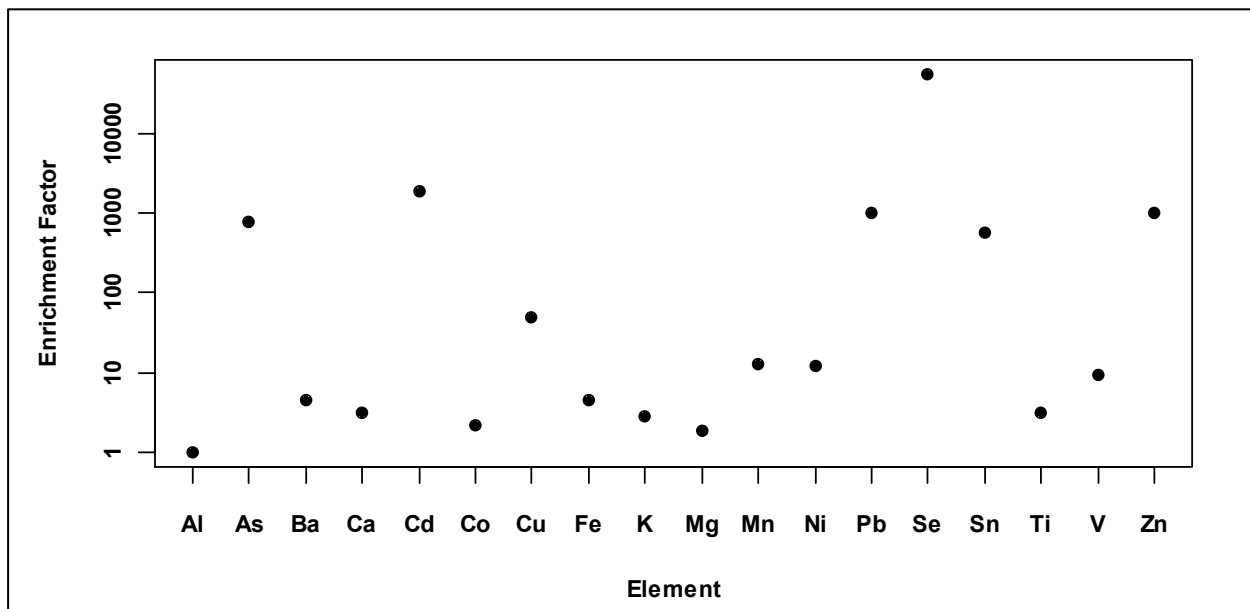


Figure 26. Al-normalized enrichment factors for elements in PM_{2.5} at Steubenville.

4.2.3 Spatial Variability of PM_{2.5} in the Steubenville Region

As discussed earlier, in addition to measurements made at the central site in Steubenville, the ambient air monitoring portion of SCAMP included measurements of PM_{2.5} and its water-extractable ionic and elemental components at four satellite sites surrounding the city, permitting a characterization of the spatial variability of these species in the Steubenville region.

Table 29 summarizes the overall mean 24-hr PM_{2.5} and water-extractable PM_{2.5} component concentrations observed at each of the SCAMP satellite sites. Mean concentrations at the Steubenville site are also provided for reference. The overall mean concentrations observed at the satellite sites were less than that observed at Steubenville, ranging from 13.9 µg/m³ at the northern and western sites to 16.0 µg/m³ at the southern site. The southern and eastern sites had mean concentrations that were greater than the annual PM_{2.5} NAAQS of 15 µg/m³.

Although not shown in Table 29, maximum 24-hr concentrations at the satellite sites were all less than the current 24-hr PM_{2.5} NAAQS of 65 µg/m³, ranging from 51.1 µg/m³ at the southern site to 64.8 µg/m³ at the western site. Ninety-eighth percentile concentrations at the satellite sites ranged from 34.4 µg/m³ at the western site to 40.6 µg/m³ at the eastern site.

As with the Steubenville site, sulfate was the most abundant of the measured PM_{2.5} components at each of the satellite sites, accounting for 4.5-5.9 µg/m³, or about 32-37%, of the total PM_{2.5} mass across the four sites. (Organic material, the second most abundant PM_{2.5} component at Steubenville, was not measured at the satellite sites as part of SCAMP). Ammonium, nitrate, and chloride accounted for about 12-14%, 4.6-6.3%, and 0.35-0.46%, respectively, of the total PM_{2.5} mass at the satellite sites based on the mean concentrations shown in Table 29.

Seasonal and weekday/weekend trends observed at the satellite sites for PM_{2.5} and its major components were generally similar to those observed at the central Steubenville site. Figure 27 is a plot of the ratios of the median warm season (April-September) daily average PM_{2.5} concentration to the median cool season (October-March) daily average PM_{2.5} concentration for each of the five SCAMP ambient monitoring sites. Solid points indicate statistically significant (i.e., at $\alpha=0.05$) differences in the locations of the distributions of 24-hr average concentrations observed during the warm and cool seasons, based on the application of Wilcoxon rank sum tests. Like the Steubenville site, each of the satellite sites exhibited a statistically significant trend of greater warm season and lesser cool season concentrations. Median warm season concentrations were 20-40% greater than median cool season concentrations at all of the SCAMP ambient air monitoring sites.

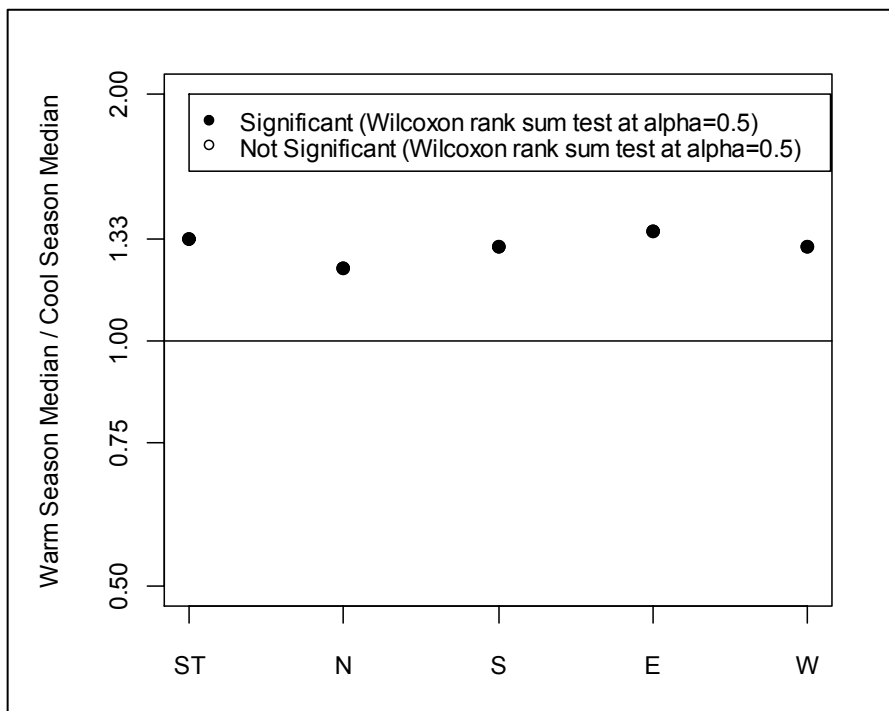


Figure 27. Ratios of warm-to-cool season median concentrations for PM_{2.5} at the SCAMP ambient air monitoring sites. Solid points indicate statistically significant seasonal differences.

Table 29. Summary statistics for 24-hr average mass concentrations of PM_{2.5} and water-extractable PM_{2.5} components (µg/m³) at the five SCAMP ambient air monitoring sites, 2000-2002.

	Steubenville			North			South			East			West		
	N	Mean	SD	N	Mean	SD	N	Mean	SD	N	Mean	SD	N	Mean	SD
PM_{2.5}	640	18.4	11.2	663	13.9	7.7	416	16.0	8.8	672	15.2	8.7	587	13.9	8.3
NH₄⁺	151	2.3	1.5	161	1.7	1.1	96	2.2	1.4	158	1.9	1.4	129	1.8	1.1
SO₄²⁻	151	5.8	4.1	161	4.5	3.8	96	5.9	4.3	155	5.1	4.4	129	4.9	4.1
NO₃⁻	151	1.16	1.16	161	0.71	0.97	96	0.73	0.95	155	0.74	0.93	129	0.88	1.14
Cl⁻	151	0.21	0.51	161	0.06	0.10	96	0.07	0.08	155	0.07	0.18	129	0.05	0.06
Al	128	0.019	0.019	125	0.010	0.008	50	0.012	0.009	136	0.009	0.009	91	0.010	0.008
As	140	0.00259	0.00284	138	0.00173	0.00102	65	0.00193	0.00107	144	0.00172	0.00098	112	0.00139	0.00069
Ba	141	0.0019	0.0013	137	0.0011	0.0007	68	0.0037	0.0033	142	0.0018	0.0012	100	0.0013	0.0015
Ca	111	0.078	0.051	117	0.049	0.033	52	0.059	0.035	123	0.053	0.039	74	0.060	0.031
Cd	141	0.00051	0.00067	137	0.00037	0.00035	68	0.00039	0.00038	141	0.00033	0.00028	100	0.00034	0.00033
Co	142	0.000037	0.000021	141	0.000035	0.000025	70	0.000049	0.000056	143	0.000044	0.000048	109	0.000054	0.000068
Cu	124	0.0028	0.0027	118	0.0022	0.0026	61	0.0030	0.0025	122	0.0028	0.0039	98	0.0021	0.0018
Fe	125	0.019	0.023	120	0.011	0.013	56	0.014	0.012	119	0.012	0.012	78	0.010	0.011
K	126	0.099	0.062	118	0.077	0.047	50	0.085	0.058	129	0.105	0.088	96	0.066	0.033
Mg	142	0.029	0.036	142	0.013	0.017	71	0.011	0.011	144	0.010	0.007	112	0.012	0.010
Mn	133	0.0078	0.0080	131	0.0036	0.0037	64	0.0033	0.0042	137	0.0024	0.0017	93	0.0030	0.0035
Na	115	0.090	0.074	119	0.054	0.030	64	0.060	0.043	126	0.058	0.030	93	0.062	0.057
Ni	139	0.0006	0.0006	138	0.0006	0.0007	71	0.0010	0.0014	141	0.0008	0.0013	105	0.0011	0.0025
Pb	136	0.0090	0.0128	131	0.0046	0.0035	64	0.0047	0.0035	134	0.0047	0.0037	93	0.0042	0.0030
Se	137	0.00473	0.00493	134	0.00516	0.00534	68	0.00410	0.00296	132	0.00557	0.00666	102	0.00375	0.00350
Sn	142	0.00022	0.00031	139	0.00017	0.00019	68	0.00021	0.00023	142	0.00042	0.00066	100	0.00021	0.00023
V	143	0.00105	0.00133	142	0.00075	0.00079	71	0.00079	0.00061	144	0.00056	0.00041	112	0.00056	0.00049
Zn	143	0.0555	0.0745	142	0.0199	0.0142	71	0.0180	0.0083	143	0.0189	0.0158	112	0.0173	0.0134

Figure 28 shows ratios of median warm season concentrations to median cool season concentrations for PM_{2.5} ions and water-soluble elemental components that were measured every fourth day at Steubenville and each of the four satellite sites during SCAMP. Sulfate concentrations were significantly greater during the warm season than during the cool season, and nitrate concentrations were significantly greater during the cool season than during the warm season at all five sites, consistent with the previously discussed effects of meteorology

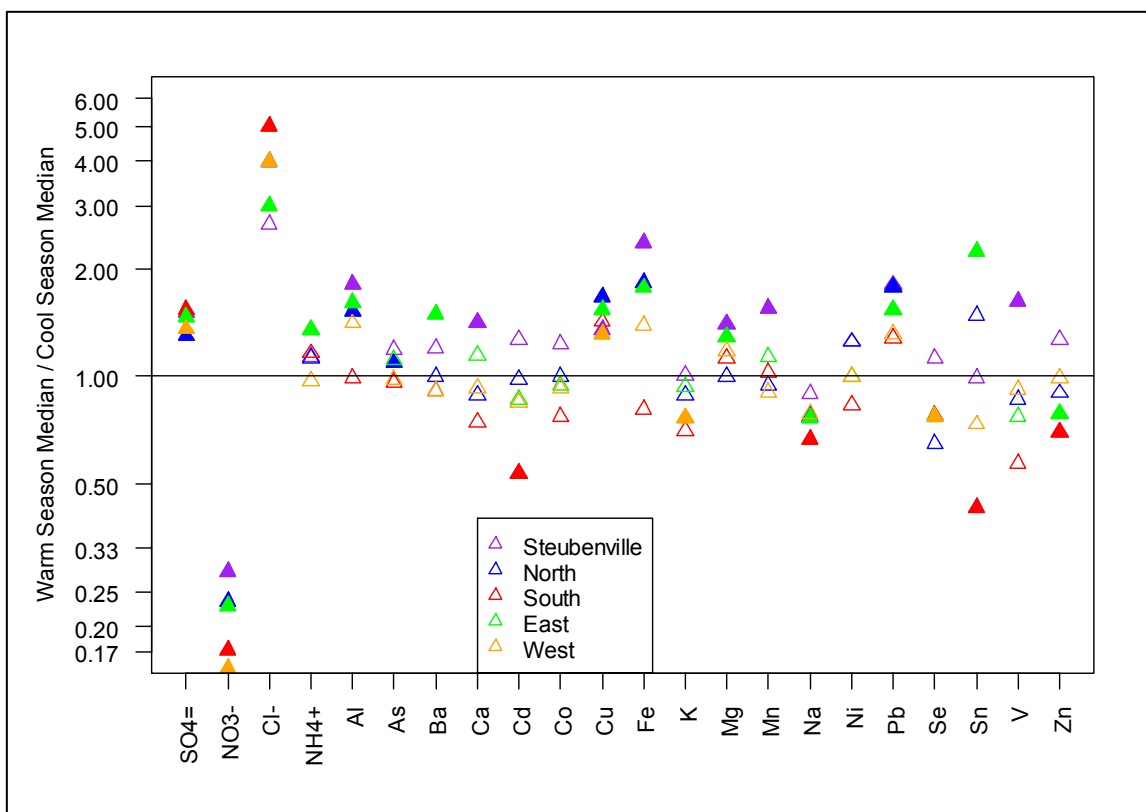


Figure 28. Ratios of warm-to-cool season median concentrations for PM_{2.5} water-extractable ionic and elemental components at the SCAMP ambient air monitoring sites. Solid points indicate statistically significant seasonal differences, based on the results of Wilcoxon rank sum tests at $\alpha=0.05$.

and ammonium availability on concentrations of these PM_{2.5} components. Among the other water-extractable PM_{2.5} components, Cl⁻, Al, Cu, Fe, and Pb had significantly greater ambient concentrations during the warm season than during the cool season at a majority (i.e., at least three out of five) of the sites. These seasonal variations may be due to seasonal differences in emissions or to seasonal differences in meteorology or atmospheric chemistry that affect ambient concentrations or solubility. The list above includes the three elements (Al, Fe, and Pb) for which fractional solubilities were significantly greater during the warm season than during the cool season at the Steubenville site. Conversely, concentrations of water-extractable Na were significantly greater during the cool season than during the warm season at three of the five monitoring sites.

Weekday/weekend differences in PM_{2.5} concentrations observed at the five SCAMP ambient air monitoring sites are summarized in Figure 29. At all of the sites, median 24-hr PM_{2.5} concentrations observed during weekdays (Monday through Friday) were 6-11% greater than

those observed during the weekend (Saturday and Sunday); however, the weekday/weekend difference only reached statistical significance for the northern site. As shown in Figure 30, which summarizes weekday/weekend differences in concentrations of PM_{2.5} water-extractable ionic and elemental components at the five monitoring sites, the only PM_{2.5} component that exhibited a consistent, statistically significant weekday/weekend trend at a majority of the SCAMP ambient air monitoring sites was water-extractable Ba. Median concentrations of water-extractable Ba were 19-57% greater during weekdays than during the weekend at each of the sites, and this difference was statistically significant for four of the five sites. As shown earlier in Figures 22 and 23, ambient concentrations of Ba in the acid-digestible PM_{2.5} fraction and water-extractable PM₁₀ fraction were also significantly greater during weekdays than during the weekend. This weekday/weekend trend may reflect the impact of weekly variations in motor vehicle emissions, which would be expected to impact all of the monitoring sites to some extent, on ambient PM_{2.5} in the Steubenville region. Ba is found in brake wear emissions (Garg et al., 2000) and has been suggested as an elemental tracer for heavy-duty diesel truck emissions (Chellam et al., 2005). As shown in Figure 23, concentrations of elemental carbon, which is also commonly used as a tracer for diesel emissions (Suarez and Ondov, 2002), likewise showed a significant trend of greater weekday and lower weekend concentrations at the Steubenville site.

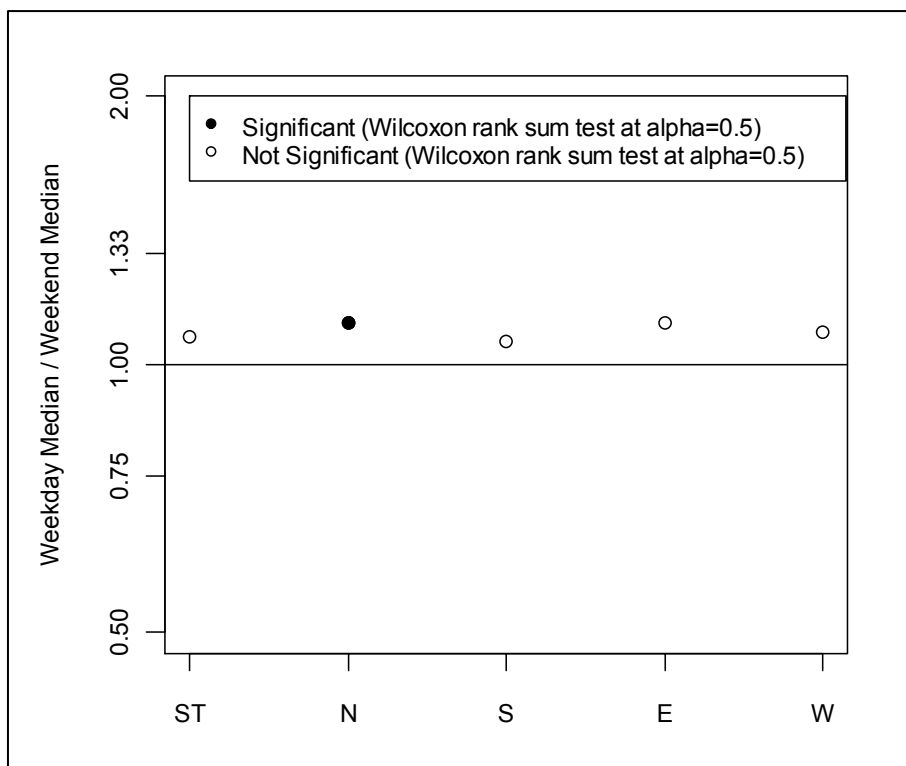


Figure 29. Ratios of weekday-to-weekend median concentrations for PM_{2.5} at the SCAMP ambient air monitoring sites. Solid points indicate statistically significant weekday/weekend differences.

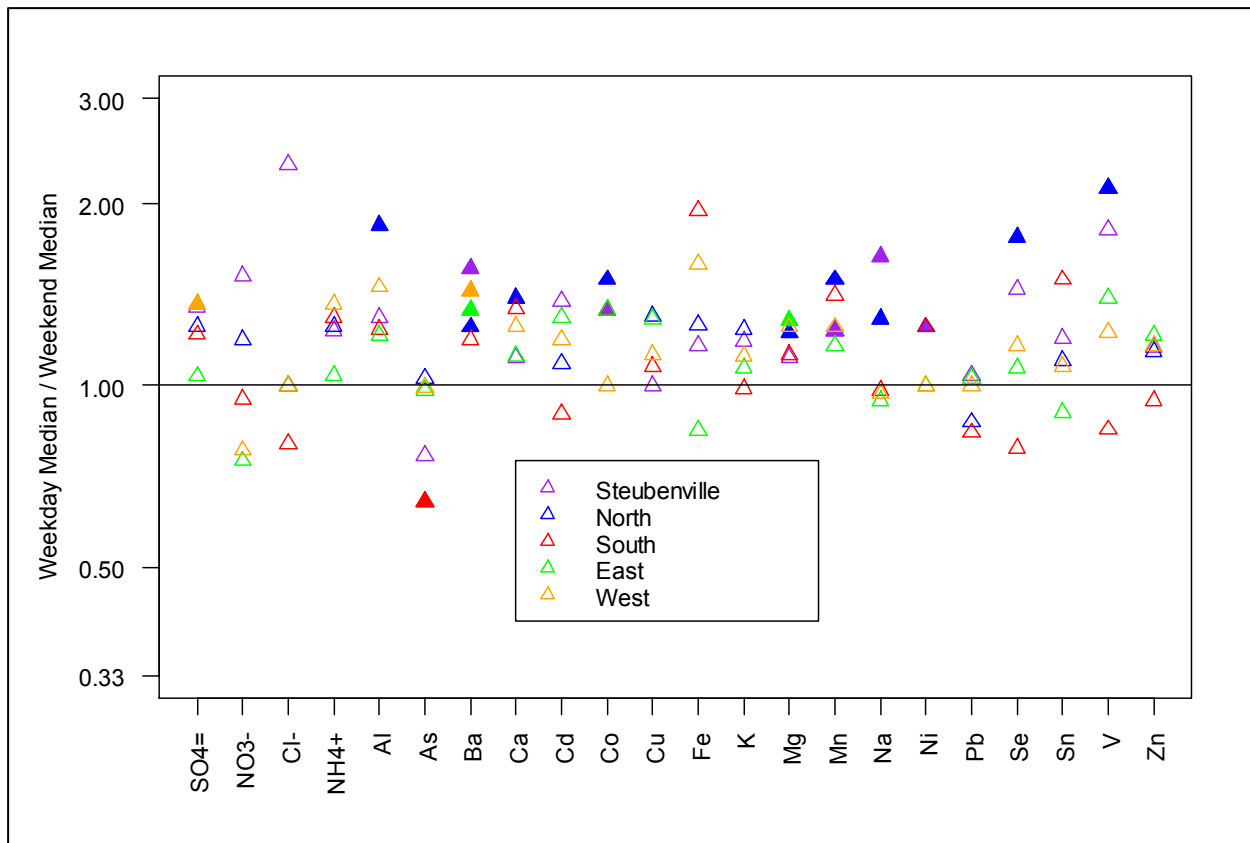


Figure 30. Ratios of weekday-to-weekend median concentrations for PM_{2.5} water-extractable ionic and elemental components at the SCAMP ambient air monitoring sites. Solid points indicate statistically significant weekday/weekend differences, based on the results of Wilcoxon rank sum tests at $\alpha=0.05$.

Having characterized the overall average concentrations of PM_{2.5} and its water-extractable components at the SCAMP satellite sites, as well as the seasonal and weekday/weekend variability in these concentrations, it is important to focus specifically on the following two questions:

1. To what extent are ambient PM_{2.5} and PM_{2.5} chemical component concentrations at Steubenville greater than or less than PM_{2.5} and PM_{2.5} chemical component concentrations in the surrounding region?
2. To what extent do concentrations of PM_{2.5} and its chemical components correlate in space across the Steubenville region?

Answering the first question is important for quantifying the effect of local sources in Steubenville on ambient PM_{2.5} concentrations there, as well as for gaining insights relevant to assessing the possible toxicological potency of the urban increment of PM_{2.5} mass. Answering the second question is important for assessing the extent to which concentrations measured at a given site in the Steubenville region are appropriate for representing day-to-day variations in the exposure of the region's population for the purposes of a time series epidemiology study. These questions are explored in the next two subsections.

4.2.3.1 Spatial Variability in the Magnitude of $PM_{2.5}$ and $PM_{2.5}$ Component Concentrations

Table 29 provides a preliminary basis for comparing the magnitude of $PM_{2.5}$ and $PM_{2.5}$ component concentrations in Steubenville with the magnitude of these concentrations in the surrounding region. Based on the data provided in the table, the overall mean $PM_{2.5}$ concentration at the central Steubenville site was 15-32% greater than the overall mean $PM_{2.5}$ concentrations at each of the four surrounding satellite sites. Mean concentrations at the Steubenville site were greater than those at each of the four satellite sites for 14 of the 22 $PM_{2.5}$ components (i.e., NH_4^+ , NO_3^- , Cl^- , and water-extractable Al, As, Ca, Cd, Fe, Mg, Mn, Na, Pb, V, and Zn) that were measured at all of the sites. The data suggest that Cl^- and water-extractable Mg, Mn, and Zn were particularly enriched at the Steubenville site relative to the surrounding region; overall mean concentrations of these $PM_{2.5}$ components measured at Steubenville were at least twice as great as mean concentrations measured at each of the satellite sites.

Given the presence of missing values in the SCAMP dataset and the appreciable temporal variability in $PM_{2.5}$ and co-pollutant concentrations in the Steubenville region, intersite comparisons are more appropriately made using pairwise daily data than overall means, as comparisons based on the latter may be biased by the presence or absence of missing values during periods of high or low concentrations. Figure 31 presents parity plots comparing 24-hr average $PM_{2.5}$ concentrations measured at each satellite site with corresponding 24-hr average $PM_{2.5}$ concentrations measured at the Steubenville site. For all of the satellite sites, a majority of the points fall below the parity line, suggesting that $PM_{2.5}$ concentrations at the satellite sites tended to be lower than $PM_{2.5}$ concentrations at the Steubenville site. This trend is further quantified in Figure 32, which shows the percentage of days for which $PM_{2.5}$ and water-extractable $PM_{2.5}$ component concentrations observed at the Steubenville site were greater than corresponding concentrations observed at each satellite site. For total $PM_{2.5}$ mass, the trend of greater Steubenville and lesser satellite site concentrations was most pronounced for the northern and western satellite sites, for which about 83% of the daily $PM_{2.5}$ concentrations were less than corresponding concentrations at Steubenville, and least pronounced for the eastern site, for which 64% of daily concentrations were less than those at Steubenville. This is reasonable, as the northern and western sites were the most rural of the four satellite sites and, assuming that transport predominantly occurs from west-to-east in the Steubenville region, tended to be situated upwind or crosswind from major emission sources in Steubenville. Conversely, the eastern satellite site tended to be situated downwind of emission sources in both Steubenville and the Pittsburgh metropolitan area to the east of Steubenville, and therefore was likely more frequently impacted by $PM_{2.5}$ from these sources than the northern and western sites.

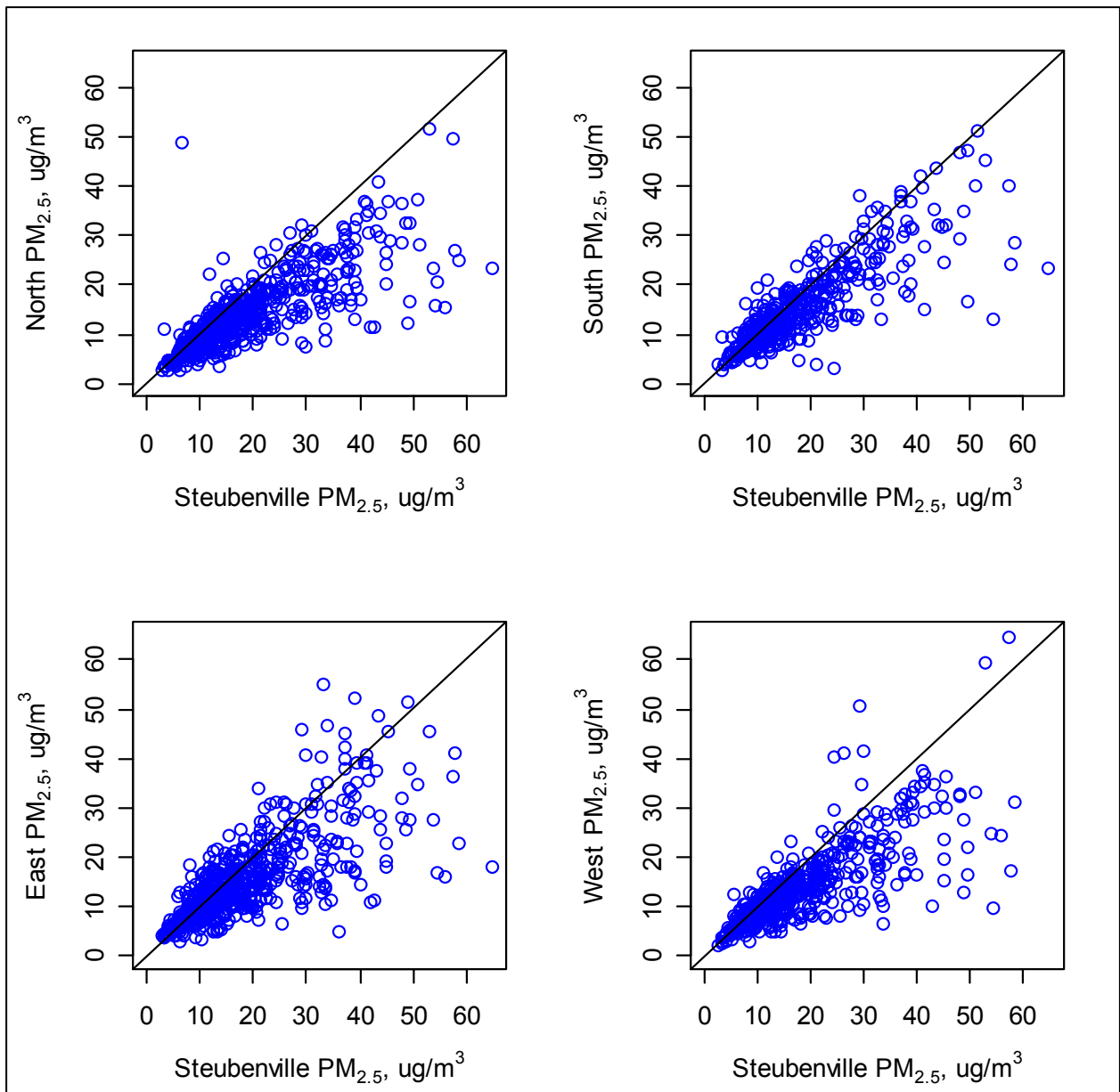


Figure 31. Parity plots showing 24-hr average PM_{2.5} concentrations measured at each of the SCAMP satellite sites versus 24-hr average PM_{2.5} concentrations measured simultaneously at the Steubenville site.

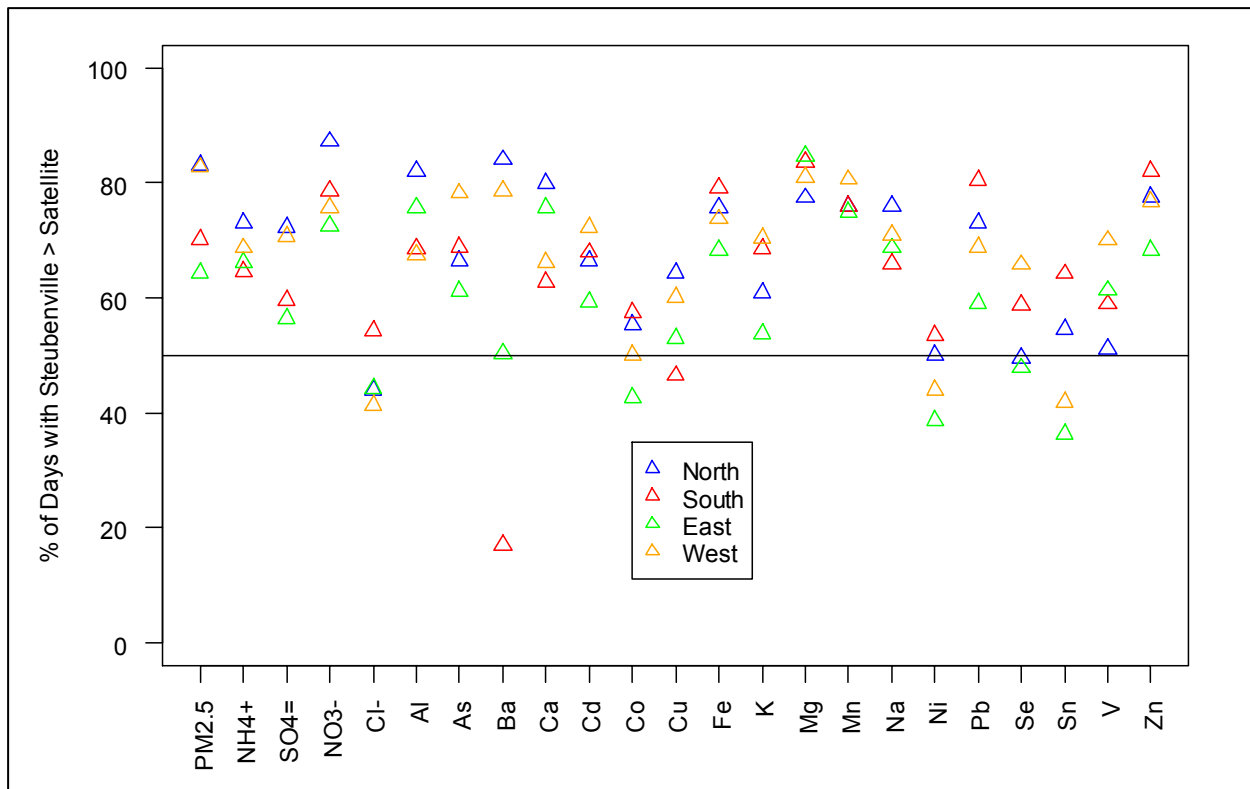


Figure 32. Percentage of days for which 24-hr average PM_{2.5} and water-extractable PM_{2.5} component concentrations measured at each SCAMP satellite site were greater than corresponding concentrations measured at the central Steubenville site.

Figure 32 and Figure 33, which plot median ratios of daily PM_{2.5} and PM_{2.5} component concentrations measured at Steubenville to corresponding daily PM_{2.5} and PM_{2.5} component concentrations measured at each satellite site, generally confirm the observations made above on the basis of overall mean concentrations. The data presented in these figures indicate that, based on pairwise comparisons, 15 of the 22 measured PM_{2.5} components (i.e., NH₄⁺, SO₄²⁻, NO₃⁻, and water-extractable Al, As, Ca, Cd, Fe, K, Mg, Mn, Na, Pb, V, and Zn) had greater concentrations at Steubenville than at each of the four satellite sites on a majority of the days during SCAMP. Again, water-extractable Mg, Mn, and Zn exhibited appreciably higher concentrations at Steubenville than at the satellite sites; median daily Steubenville/satellite site concentration ratios were >1.6 for all comparisons involving these elements. Water-extractable Fe and NO₃⁻ also exhibited markedly greater typical concentrations at Steubenville than at the satellite sites; median daily Steubenville/satellite site concentration ratios were 1.62-1.77 for water-extractable Fe and 1.52-1.89 for NO₃⁻. One noteworthy difference between the comparisons based on overall mean concentrations and the comparisons based on pairwise concentrations involved Cl⁻. Whereas the former comparison suggested that Cl⁻ concentrations at Steubenville were substantially greater than those at the satellite sites, the latter suggests no appreciable difference. This discrepancy likely reflects the occurrence of several pollution episodes at Steubenville marked by abnormally elevated Cl⁻ concentrations, which are captured when the mean is used as the measure of central tendency but not when the median is used as such.

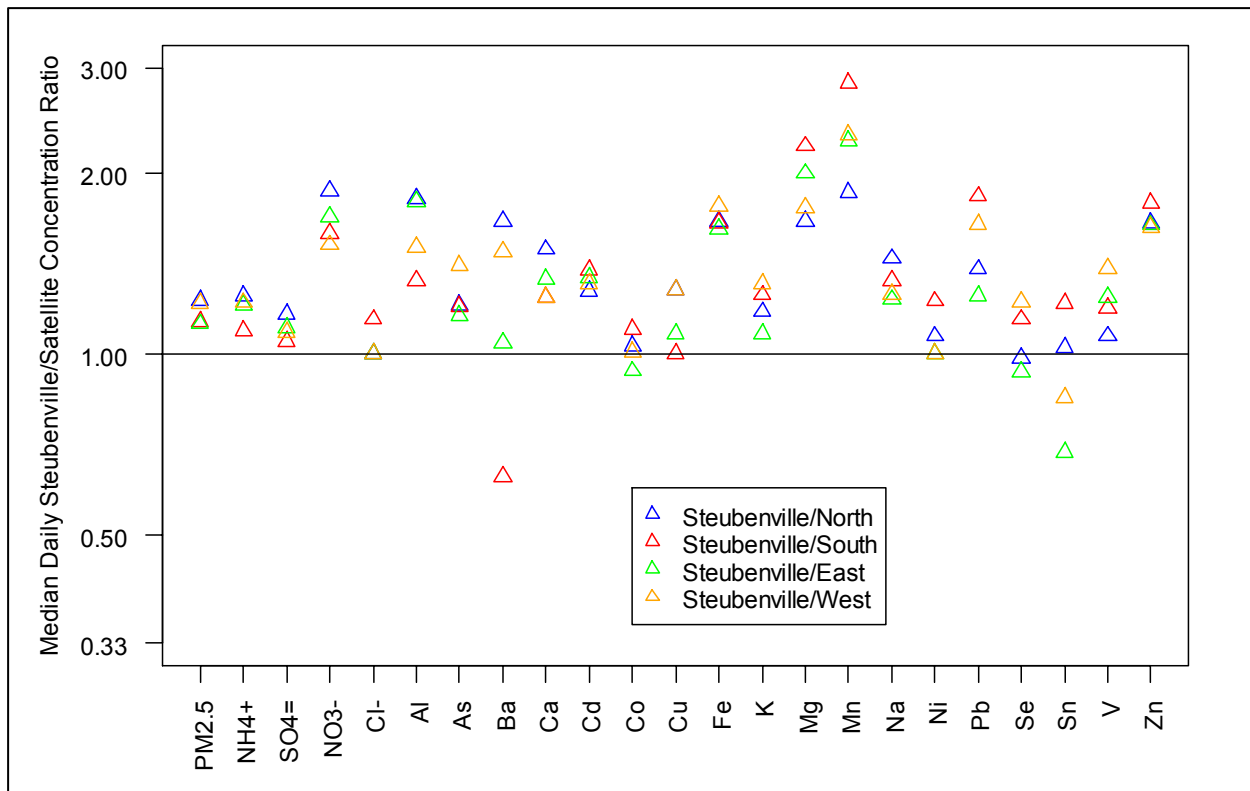


Figure 33. Median daily ratios of PM_{2.5} and water-extractable PM_{2.5} component concentrations measured at the central Steubenville site to corresponding concentrations measured at each SCAMP satellite site.

Connell and co-workers (2005a, 2006) evaluated the statistical significance of differences between PM_{2.5} and PM_{2.5} component concentrations measured at the SCAMP Steubenville site and PM_{2.5} and PM_{2.5} component concentrations measured at the satellite sites. For concentrations of total PM_{2.5} and its ionic components (i.e., SO₄²⁻, NH₄⁺, and NO₃⁻), statistical significance was determined by applying paired t-tests to adjusted paired differences between Steubenville and satellite site concentrations. The concentrations were adjusted by applying a log-transform, which was required because paired differences in raw concentrations were not normally distributed, and by applying ARIMA modeling when necessary to remove autocorrelation from the resulting time series of log-transformed differences. Results indicated that PM_{2.5} concentrations and concentrations of fine particulate NH₄⁺ and NO₃⁻ at the Steubenville site were significantly greater ($p < 0.0001$ for PM_{2.5} and NO₃⁻; $p = 0.003-0.01$ for NH₄⁺) than concentrations of these species at each of the four satellite sites. Concentrations of fine particulate SO₄²⁻ at Steubenville were significantly greater than concentrations at the northern ($p < 0.0001$), eastern ($p = 0.015$), and western sites ($p = 0.002$), but not the southern site ($p = 0.27$).

For ambient concentrations of water-extractable elemental PM_{2.5} components, statistical significance was determined by applying Wilcoxon signed rank tests to paired differences in concentrations measured at the Steubenville site and each satellite site. Of the 18 elements that were determined at all five monitoring sites, ten (Al, As, Ca, Cd, Fe, Mg, Mn, Na, Pb, and Zn) had significantly greater (at $\alpha=0.05$) concentrations at the Steubenville site than at each of the four satellite sites. Satellite site concentrations were significantly greater than Steubenville concentrations only for water-extractable Sn at the eastern site and water-extractable Ba at the southern site.

To further evaluate the degree of homogeneity in PM_{2.5} component concentrations among the five SCAMP monitoring sites, coefficients of divergence (Wongphatarakul et al., 1998) were computed for each site pair using overall mean concentrations of the 22 water-extractable ionic and elemental PM_{2.5} components measured at each site. The coefficient of divergence (CD) is a self-normalizing measure of the degree of similarity or difference between two sites based on a comparison of the overall average concentrations of pollutants (in this case, PM_{2.5} components) measured at both sites. Each component is given equal weight in the computation of the CD, which is defined as:

$$CD_{jk} = \sqrt{\frac{1}{n} \sum_{i=1}^n \left(\frac{x_{ij} - x_{ik}}{x_{ij} + x_{ik}} \right)^2}$$

where j and k represent two sampling sites, x_{ij} represents the mean concentration of component i at site j , and n is the number of chemical components. The CD for a site pair will equal zero if the average PM_{2.5} component concentrations measured at the two sites are identical for each of the components being studied. Greater CD values indicate less similarity among the sites.

Coefficients of divergence computed for each SCAMP site pair are plotted in Figure 34. In all cases, CDs for the Steubenville site / satellite site pairs were greater than those for the satellite site / satellite site pairs. This indicates that PM_{2.5} component concentrations measured at Steubenville were relatively dissimilar to those measured at other sites throughout the region, and suggests the influence of local emission sources on the concentrations of many PM_{2.5} constituents in Steubenville. Among the Steubenville site / satellite site pairs, the Steubenville site was most similar to the southern site, which may be expected, as this site was located in Wheeling, WV, a small city along the Ohio River that is expected to be impacted by local sources similar to those impacting Steubenville. The pair comprising the northern and western satellite sites had the lowest CD, suggesting that these were the most similar of the SCAMP sites. As discussed previously, the northern and western sites also had the lowest overall average PM_{2.5} concentrations (13.9 µg/m³) of the SCAMP sites, despite being located closer to Steubenville than any of the other sites, and were not generally situated downwind of Steubenville. Moreover, as discussed by Connell et al. (2005a), sulfate values for the northern and western sites were more highly correlated than for any other site pair, suggesting that these sites are most representative of regional concentrations of secondary PM_{2.5}.

Hence, for the purposes of this analysis, the northern and western sites are assumed to be representative of regional “background” concentrations in the Steubenville area. The background concentration measured on a given day is taken as the average concentration measured at the northern and western sites on that day. As discussed by Connell et al. (2005a) and summarized in Figure 35, on the basis of this assumption, local sources in the immediate Steubenville vicinity contributed an estimated 4.6 µg/m³ to Steubenville’s overall average PM_{2.5} concentration of 18.4 µg/m³ during SCAMP. (This may be an underestimate of the contribution of local sources to the extent that the northern and western sites were impacted by emissions from Steubenville or by local sources located near these satellite sites). Results suggest that about 20% of the SO₄²⁻ and NH₄⁺ and 30% of the NO₃⁻ observed in Steubenville was produced locally; this is consistent with the expectation that these secondary PM_{2.5} components would be predominantly regional pollutants. The more than half of the estimated local source contribution labeled “other” in Figure 35 likely consists of carbonaceous species, crustal materials, trace elements, and particle-bound water.

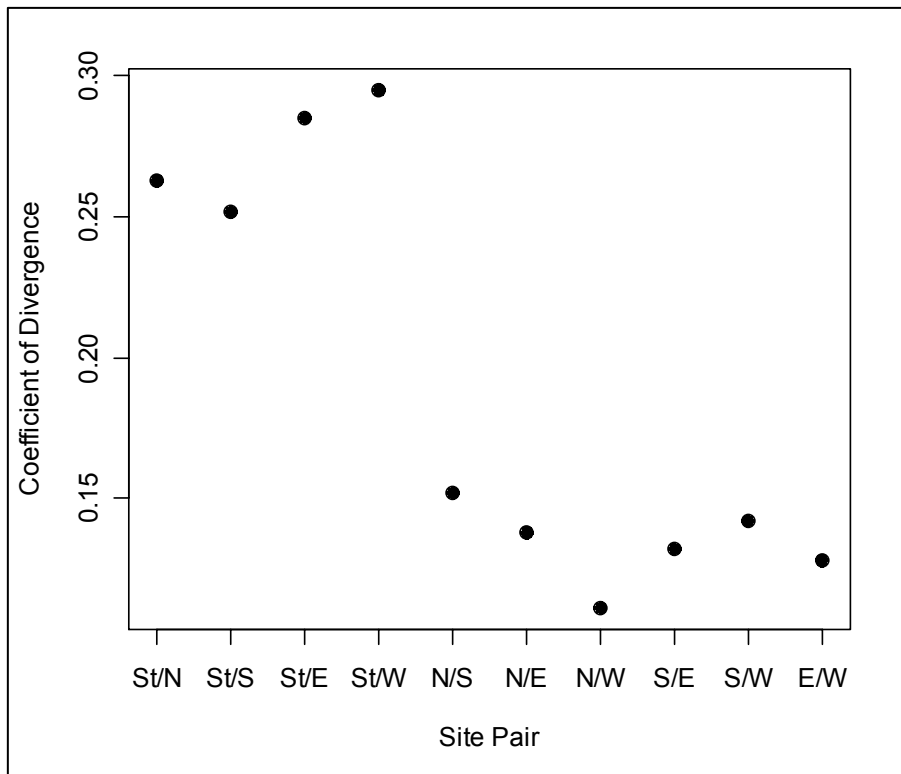


Figure 34. Intersite coefficients of divergence for all possible SCAMP site pairs based on mean concentrations of 22 water-extractable PM_{2.5} components.

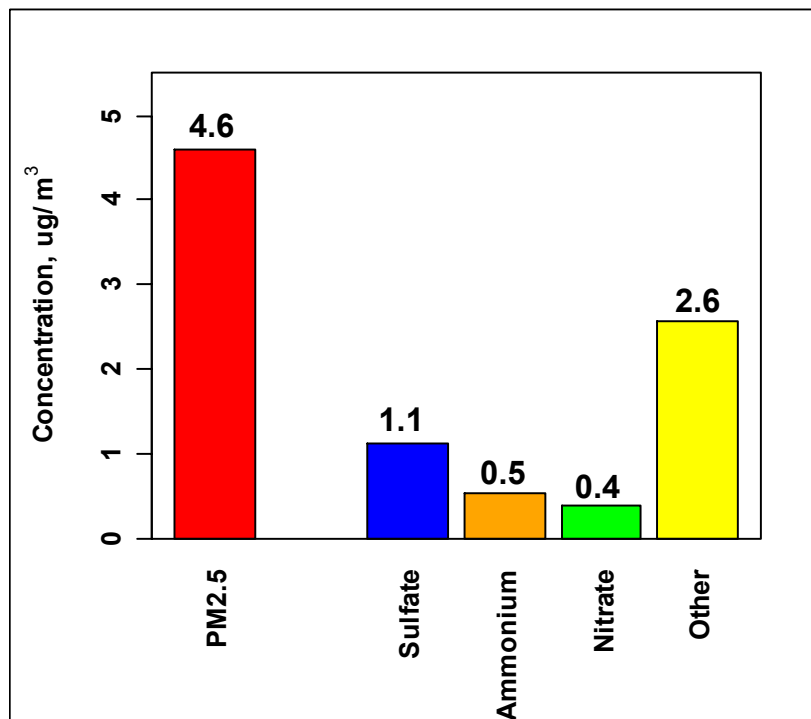


Figure 35. Estimated contribution of local sources to concentrations of PM_{2.5} and its major ionic components at Steubenville during SCAMP.

Although they do not contribute a large percentage to the total mass of PM_{2.5} in Steubenville, trace metals such as Fe, Zn, and Mn, which comprise a portion of the local source contribution labeled “other,” may have implications for the health effects of PM_{2.5} in Steubenville. The results presented earlier in this section suggest that local sources in the immediate Steubenville vicinity contribute substantially to ambient concentrations of these metals in PM_{2.5} in Steubenville. Based on the mean paired differences between Steubenville concentrations and estimated “background” concentrations of these elements, this local source contribution accounted for more than half of the total average mass of water-extractable Fe, Mn, and Zn observed in Steubenville. Mn is a HAP, and Fe and Zn have been implicated in PM toxicology studies (e.g., Adamson et al., 2000; Ghio et al., 1999a); hence, this result again reinforces the need for further research to elucidate the toxicological potency of these trace metals in Steubenville and similar locations.

4.2.3.2 Intersite Correlations

Figure 36 presents a boxplot summarizing intersite Spearman correlation coefficients computed for PM_{2.5} total mass and each of the 22 water-extractable PM_{2.5} components that were measured at all five SCAMP ambient air monitoring sites. The box plotted for each variable represents the distribution of ten correlation coefficients computed using pairwise daily concentrations of that variable for each of the ten possible site pairs.

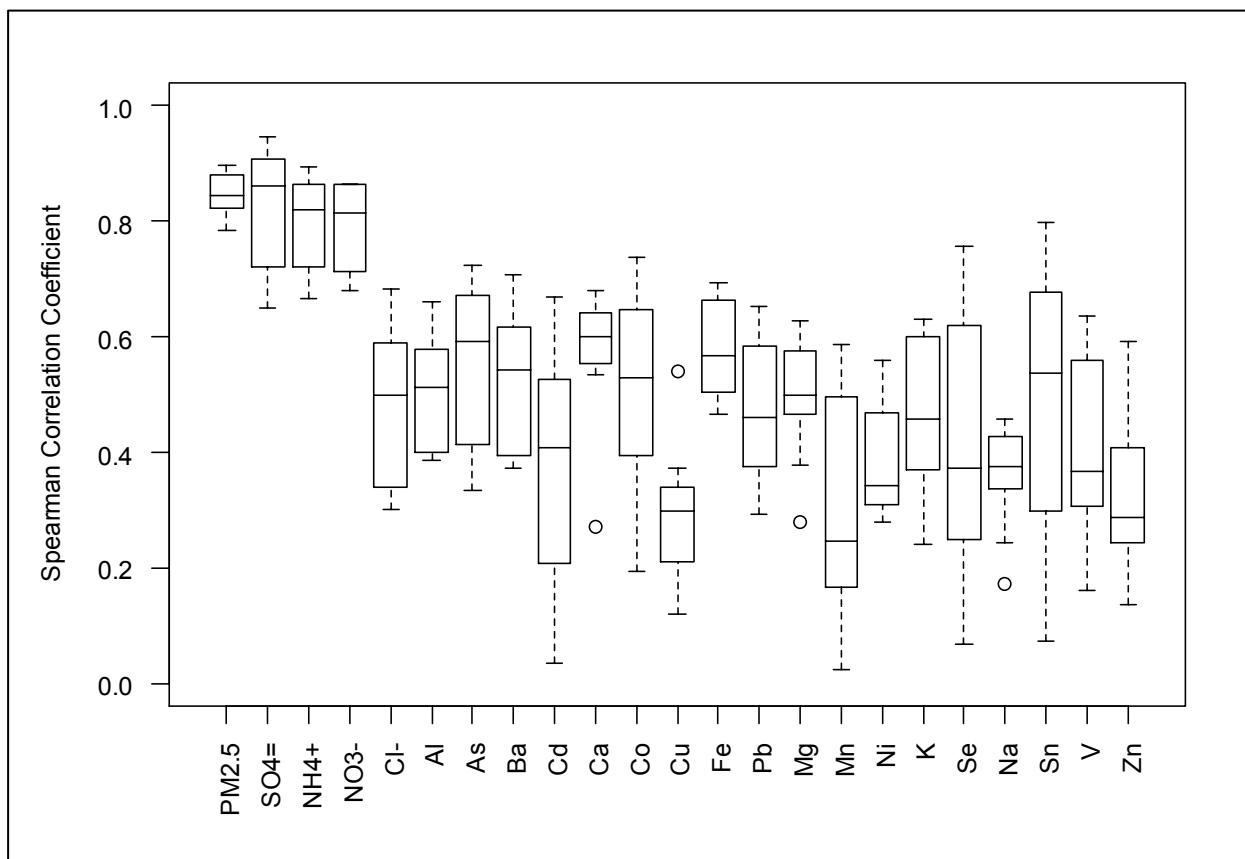


Figure 36. Intersite Spearman correlation coefficients for PM_{2.5} and its water-extractable components in the Steubenville region. Ten correlations are plotted for each variable, corresponding to the ten possible SCAMP site pairs.

As shown in the figure, daily $PM_{2.5}$ concentrations were well-correlated across the SCAMP monitoring sites; Spearman correlation coefficients describing these correlations ranged from 0.78 (for the Steubenville and eastern sites) to 0.90 (for the northern and western sites), with a median of 0.84. Concentrations of SO_4^{2-} , NH_4^+ , and NO_3^- were similarly well-correlated in space throughout the region, with median intersite Spearman correlation coefficients of 0.86 for SO_4^{2-} , 0.82 for NH_4^+ , and 0.81 for NO_3^- . These strong correlations likely reflect the influence of long-range transport and meteorology on concentrations of these secondary $PM_{2.5}$ components throughout the region.

Because of the autocorrelated nature of concentrations of $PM_{2.5}$ and some of its major components measured during SCAMP, it is possible that the associations described above are induced to some extent by seasonal and other temporal patterns in the datasets, and therefore do not reflect the extent to which concentrations at the various sites are associated on a day-to-day basis. To better evaluate the day-to-day associations among concentrations of $PM_{2.5}$, sulfate, ammonium, and nitrate measured at the SCAMP monitoring sites, and to assess the statistical significance of these associations, Connell et al. (2005a) adjusted the SCAMP data to remove seasonal trends and autocorrelation (by applying ARIMA modeling or subtracting a 3-month moving average) and then computed intersite correlations using the resulting adjusted data. Even after accounting for autocorrelation, daily concentrations of $PM_{2.5}$, fine particulate sulfate, and fine particulate ammonium were strongly and significantly associated across all of the monitoring sites. Pearson correlation coefficients describing these associations were 0.66-0.79 for $PM_{2.5}$, and 0.57-0.94 for NH_4^+ and SO_4^{2-} . Intersite NO_3^- associations, while statistically significant for all possible site pairs, were weaker, with Pearson correlation coefficients of 0.33-0.73. This suggests that the NO_3^- correlations reported in Figure 36 are likely induced to some extent by the strong seasonal trend exhibited by this $PM_{2.5}$ component (see Figure 28).

Intersite correlations for Cl^- and the 18 water-extractable $PM_{2.5}$ components measured as part of SCAMP were generally less than those for $PM_{2.5}$, SO_4^{2-} , NH_4^+ , and NO_3^- , as evidenced in Figure 36. Median intersite correlations for these 19 components ranged from 0.25 for Mn to 0.60 for Ca. The lower correlations likely reflect the fact that local source emissions, rather than regional transport and secondary formation, are responsible for much of the variability in concentrations of these primary pollutants at the various SCAMP monitoring sites. They may also be due in part to the greater uncertainty (i.e., noise) in the measured values of these components, which are present in much smaller quantities than $PM_{2.5}$ and its major ionic components.

In general, the strength of correlation between $PM_{2.5}$ and $PM_{2.5}$ component concentrations at the SCAMP monitoring sites decreased as the distance between the sites increased, although there was considerable variability in this trend from species to species. Figure 37 summarizes the overall average trend by showing the 25th, 50th, and 75th percentile intersite Spearman correlations (out of the 23 correlations, corresponding to the 23 measured $PM_{2.5}$ variables, that were computed for each pair of sites) for each SCAMP site pair as a function of the distance separating the sites. Linear least squares trend lines are also shown. For all three percentile measures, the strongest correlation was observed for the Steubenville site / northern site pair, perhaps reflecting the impact of emission sources in the immediate Steubenville vicinity on concentrations measured at both of these sites, which were the most closely situated of the SCAMP ambient air monitoring sites. The weakest correlation for all three percentile measures was observed for the southern site / eastern site pair. This is likely a result of the distance separating the sites and the influence of different types of local emission sources and/or localized meteorological regimes on ambient $PM_{2.5}$ at the sites.

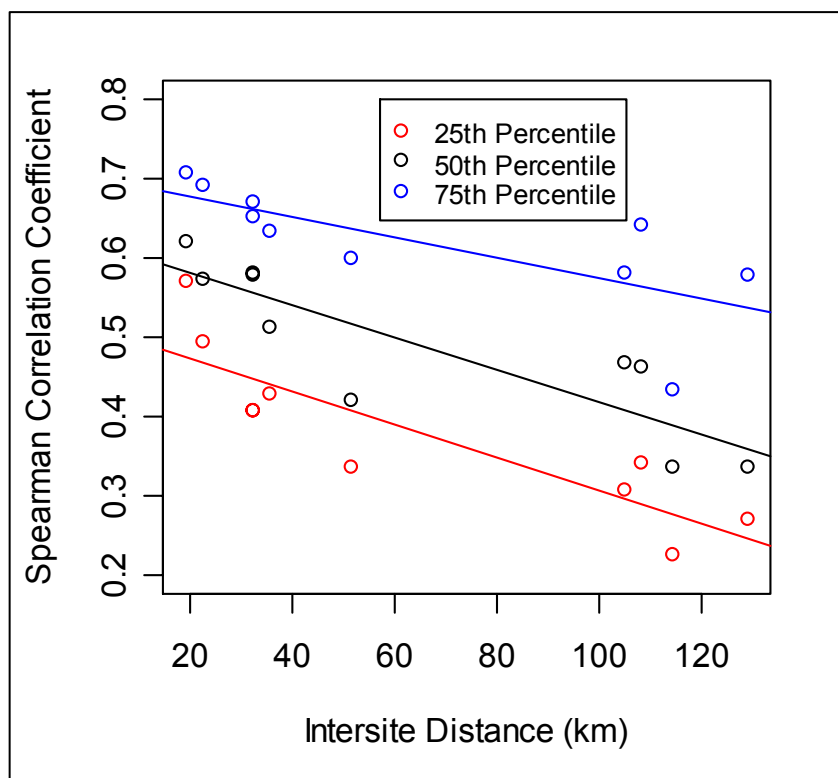


Figure 37. Scatterplot illustrating the association between intersite distance and the strength of association between $PM_{2.5}$ and $PM_{2.5}$ component concentrations measured at SCAMP site pairs. Plotted are the 25th, 50th, and 75th percentile Spearman correlation coefficients for each site pair, with linear least squares lines.

Whereas intersite correlations for total $PM_{2.5}$ mass were consistently strong during all seasons of the year, intersite correlations for several water-extractable $PM_{2.5}$ components exhibited appreciable seasonal variability across the SCAMP monitoring sites. This is illustrated in Figure 38, which shows the distributions of intersite correlations (for all 10 possible SCAMP site pairs) observed for each measured $PM_{2.5}$ component during the warm season (April-September) and during the cool season (October-March). Appreciable seasonal trends were observed in particular for NO_3^- and water-extractable K, Na, V, and Zn, which had 25th percentile intersite Spearman correlation coefficients during the cool season that were greater than their 75th percentile intersite Spearman correlation coefficients during the warm season. The seasonal variation in the strength of association between sites for these components may reflect seasonal variability in the predominance of various emission sources that impact the sites to a greater or lesser extent (e.g., the stronger correlations observed for water-extractable Na during the cool season may reflect the use of rock salt containing Na to treat icy roads throughout the region during this season). Alternatively, this seasonal variation may be due to differences in meteorology or transport conditions that affect the extent to which the sites are impacted by local and regional sources of certain pollutants. For example, the seasonal variation in intersite correlations for NO_3^- reflects the increased regional-scale formation and transport of this $PM_{2.5}$ component during the cool part of the year, when atmospheric production of secondary NH_4NO_3 is thermodynamically favored. To some extent, the greater intersite correlations during the cool season likely result from the larger signal-to-noise ratio (i.e., resulting from the larger average magnitude of ambient NO_3^- concentrations) for measured NO_3^- values during this season.

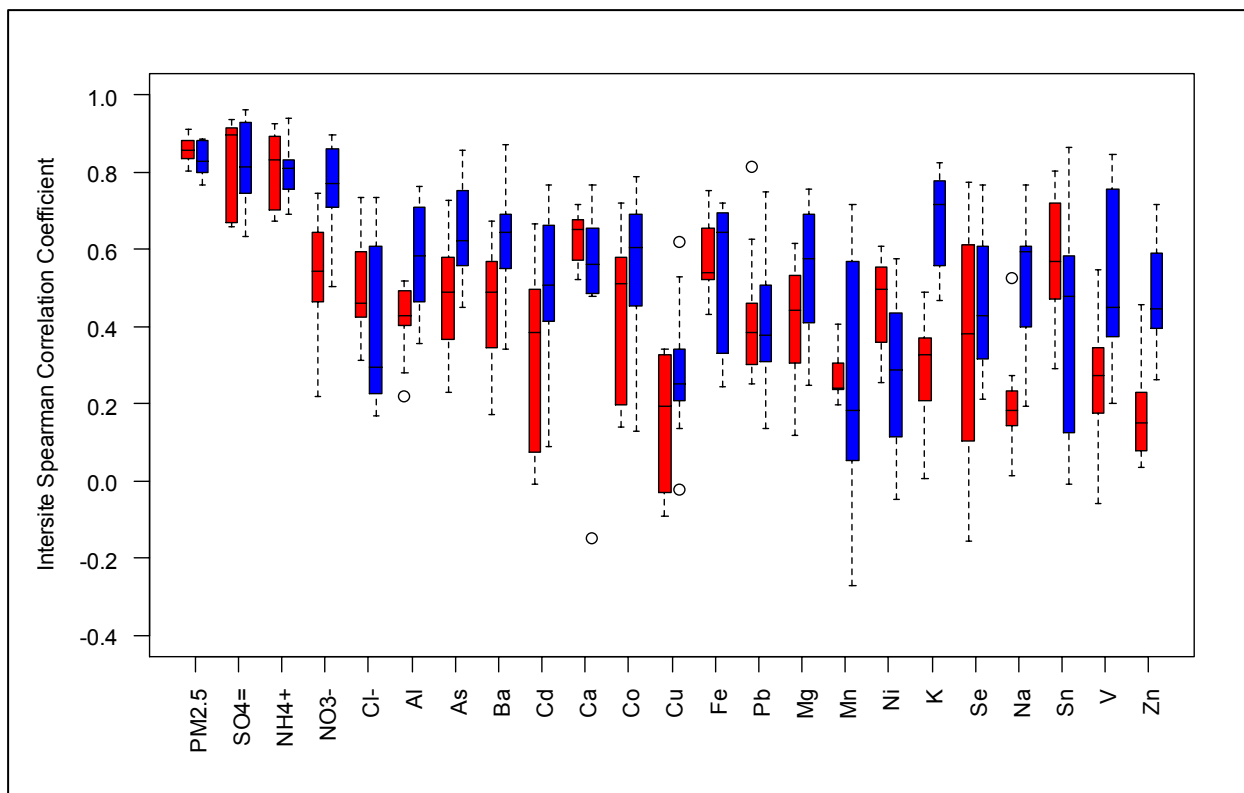


Figure 38. Intersite Spearman correlation coefficients for PM_{2.5} and its water-extractable components in the Steubenville region during the warm season (red; April-September) and the cool season (blue; October-March). Ten correlations are plotted for each variable, corresponding to the ten possible SCAMP site pairs.

In summary, the results presented in this subsection suggest that concentrations of PM_{2.5} and several of its major components (i.e., SO₄²⁻, NH₄⁺, and NO₃⁻) generally tend to vary on a regional scale in the Steubenville area. Although correlations between sites did generally exhibit a downward trend with increasing intersite distance, moderate-to-strong intersite correlations were observed for these variables for all pairs of SCAMP monitoring sites. Even the eastern satellite site, which was located about 100-130 km from the other four SCAMP ambient air monitoring sites and was separated from these sites by the Pittsburgh metropolitan source region, exhibited relatively strong correlations with the other SCAMP sites for total PM_{2.5} ($r_s = 0.78-0.85$) and fine particulate SO₄²⁻ ($r_s = 0.65-0.73$), NH₄⁺ ($r_s = 0.66-0.74$), and NO₃⁻ ($r_s = 0.68-0.74$). This reflects the largely regional nature of PM_{2.5} in the Upper Ohio River Valley region, driven largely by secondary PM_{2.5} components such as NH₄NO₃ and especially (NH₄)₂SO₄, and suggests that for the purposes of time series epidemiological studies, central site measurements of PM_{2.5}, SO₄²⁻, NH₄⁺, and NO₃⁻ are likely sufficient for representing the daily fluctuations in concentrations of these species throughout the Steubenville region. However, intersite correlations for water-extractable elemental components of PM_{2.5}, many of which are present in trace quantities and are expected to be emitted largely by local sources in the region, were considerably weaker and more variable. Caution should be exercised before using central site measurements of these components (or other trace or primary PM_{2.5} components, which may be subject to substantial measurement noise or to spatial variability resulting from localized primary source impacts) to represent daily variations in the region's exposure to them. The seasonal variability in the strength of intersite correlations for certain PM_{2.5} components, as shown in Figure 38, should also be accounted for when designing PM_{2.5} epidemiology studies, as the exposure error resulting from the use of measurements of these components from a

single ambient monitoring site may be much greater during certain times of year than during others.

4.2.4 Associations of PM_{2.5} and PM₁₀ with Co-Pollutants and Meteorological Conditions

Having examined the overall concentrations, composition, and temporal variability of PM_{2.5} and PM₁₀ in Steubenville, as well as the spatial variability of PM_{2.5} in the Steubenville region, it is next important to study the extent to which PM_{2.5} and PM₁₀ in Steubenville are associated with various gaseous pollutants and meteorological conditions. These associations can provide valuable insights into the atmospheric conditions and types of sources that affect PM concentrations in Steubenville. Moreover, both weather conditions (e.g., temperature) and gaseous pollutant (e.g., SO₂, NO₂, CO, O₃) concentrations are important for inclusion in PM_{2.5} epidemiological models as potential explanatory factors, as these variables can affect mortality and/or morbidity.

As discussed in the Introduction, if daily measures of weather or gases are collinear (i.e., correlated) with daily PM concentrations, they may confound the results of PM time series epidemiology studies, making it difficult to separate the effect of PM from the effects of these other variables. Moolgavkar and Luebeck's (1996) and Lipfert and Wyzga's (1997) criticisms of epidemiology studies for not adequately considering the potential confounding effects of gases such as CO and NO₂ cited several studies that were conducted in Steubenville (Schwartz and Dockery, 1992; Schwartz et al., 1996). Direct evidence of confounding by gaseous pollutants has been observed in at least one time series particulate matter epidemiology study in Steubenville; in a reanalysis of Steubenville data originally presented by Schwartz and Dockery (1992), Moolgavkar et al. (1995) reported that, whereas TSP was significantly associated with daily mortality in a single-pollutant model, addition of SO₂ to the model caused the effect of TSP to become statistically insignificant. In nearby Pittsburgh, Chock et al. (2000) reported that although PM₁₀ was significantly associated with daily mortality among those less than 75 years of age in non-seasonal single- and multi-pollutant models, the use of seasonal models revealed collinearity problems among concentrations of PM₁₀, CO, NO₂, and O₃ (spring and summer), casting doubt upon the findings of the non-seasonal models.

This section, by presenting correlation analyses of PM and gaseous pollutant data collected at Steubenville during SCAMP, provides an indication of the potential for such collinearity problems to occur in PM_{2.5} time series epidemiology studies conducted there. Collinearity can also cause problems for time series studies exploring the health effects of PM_{2.5} components if these components are correlated with each other or with co-pollutants. Hence, these correlations are explored as well using the speciated PM_{2.5} data collected at the central Steubenville site during SCAMP. In addition to correlation analyses, this section presents case studies illustrating the effect of meteorology on the behavior of PM_{2.5} during high-concentration episodes at Steubenville, and also presents the results of a binary recursive partitioning analysis, which was applied to identify factors that are associated with elevated hourly PM_{2.5} concentrations at Steubenville.

4.2.4.1 Correlation Analyses

Table 30 shows Spearman correlation coefficients between concentrations of PM_{2.5} (both 1-hr and 24-hr averages), PM₁₀, and PM_{10-2.5} and corresponding concentrations of gaseous pollutants measured at the central Steubenville site. (One-hour average gaseous pollutant concentrations were used when computing the correlations for 1-hr average PM_{2.5} concentrations). Correlation coefficients were computed using all data collected during SCAMP (“overall”), as well as using data stratified by astronomical season. Correlations of the gaseous pollutants (and other variables) with PM₁₀ are expected to be similar to those with PM_{2.5}, because, as discussed earlier, PM_{2.5} typically accounted for about 58-77% (interquartile range) of the mass of PM₁₀ at Steubenville. The Spearman correlation coefficient describing the association between PM_{2.5} and PM₁₀ total mass is 0.91. Hence, correlations with estimated mass concentrations of PM_{10-2.5} (computed by differencing measured PM₁₀ and PM_{2.5} mass concentrations) are shown to illustrate differences between fine and coarse particles. The Spearman correlation coefficient describing the association between PM_{2.5} and PM_{10-2.5} concentrations observed during SCAMP is 0.47, indicating that fine and coarse particle mass concentrations were only moderately collinear at Steubenville.

Table 30. Spearman correlation coefficients between ambient concentrations of various PM measures (i.e., hourly PM_{2.5}, daily PM_{2.5}, daily PM₁₀, and daily PM_{10-2.5}) and ambient concentrations of various gaseous pollutants at the central Steubenville site, both overall and by astronomical season.

Gaseous Pollutant	PM Measure	Spearman Correlation Coefficient				
		Overall	Winter	Spring	Summer	Fall
SO ₂	PM _{2.5} – 1-h	0.50	0.62	0.61	0.48	0.60
	PM _{2.5} – 24-h	0.55	0.64	0.58	0.49	0.69
	PM ₁₀ – 24-h	0.55	0.63	0.58	0.57	0.69
	PM _{10-2.5} – 24-h	0.44	0.54	0.36	0.42	0.51
NO	PM _{2.5} – 1-h	0.33	0.51	0.29	0.09	0.58
	PM _{2.5} – 24-h	0.40	0.74	0.32	-0.09	0.66
	PM ₁₀ – 24-h	0.44	0.76	0.36	0.11	0.66
	PM _{10-2.5} – 24-h	0.48	0.66	0.32	0.50	0.49
NO ₂	PM _{2.5} – 1-h	0.50	0.69	0.55	0.30	0.66
	PM _{2.5} – 24-h	0.51	0.77	0.43	0.29	0.72
	PM ₁₀ – 24-h	0.59	0.80	0.53	0.47	0.74
	PM _{10-2.5} – 24-h	0.58	0.66	0.43	0.68	0.58
NO _x	PM _{2.5} – 1-h	0.49	0.68	0.54	0.27	0.68
	PM _{2.5} – 24-h	0.49	0.77	0.47	0.14	0.72
	PM ₁₀ – 24-h	0.55	0.80	0.55	0.33	0.72
	PM _{10-2.5} – 24-h	0.56	0.67	0.45	0.65	0.54
CO	PM _{2.5} – 1-h	0.46	0.64	0.45	0.53	0.72
	PM _{2.5} – 24-h	0.52	0.74	0.44	0.52	0.78
	PM ₁₀ – 24-h	0.59	0.77	0.45	0.69	0.79
	PM _{10-2.5} – 24-h	0.55	0.65	0.39	0.60	0.60
O ₃	PM _{2.5} – 1-h	-0.16	-0.61	-0.24	0.06	-0.38
	PM _{2.5} – 24-h	0.05	-0.51	-0.04	0.47	-0.21
	PM ₁₀ – 24-h	0.12	-0.47	0.08	0.41	-0.20
	PM _{10-2.5} – 24-h	0.04	-0.28	0.12	-0.07	0.03

As shown in the table, PM concentrations were moderately correlated with primary gaseous pollutant (i.e., SO₂, NO, NO₂, NO_x, and CO) concentrations at Steubenville. Overall Spearman correlations between the various PM measures and these gaseous pollutants ranged from 0.4 to 0.6 for all of the comparisons except that between 1-hr PM_{2.5} and 1-hr NO ($r_s = 0.33$). Conversely, the PM measures were uncorrelated with O₃; Spearman correlation coefficients describing the associations between PM and O₃ ranged from -0.16 to 0.12.

Whereas overall correlations between PM_{2.5} and PM₁₀ and gaseous pollutants at Steubenville were moderate (i.e., for predominantly primary gases) or weak (i.e., for O₃), stronger correlations were observed when the data were stratified by season. Correlations of PM_{2.5} and PM₁₀ with SO₂, CO, and especially NO, NO₂, and NO_x generally were stronger during the fall and winter and weaker during the spring or summer. For 24-hr average PM_{2.5} concentrations, which are of most interest with respect to recent daily time series epidemiology studies, Spearman correlations with NO₂ ranged from 0.77 in the winter to 0.29 in the summer; Spearman correlations with CO ranged from 0.78 in the fall to 0.44 in the spring, and Spearman correlations with SO₂ ranged from 0.69 in the fall to 0.49 in the summer. Seasonal differences in the strength of association between 24-hr average PM_{2.5} and 24-hr average NO₂ are graphically illustrated in Figure 39, which presents scatterplots comparing log-transformed concentrations of these variables during summer and winter, with linear least squares lines. (The log-transform was applied to improve the homoscedasticity of the data). The stronger wintertime association is evidenced by the greater slope and tighter scatter of the data around the least squares line during winter than during summer.

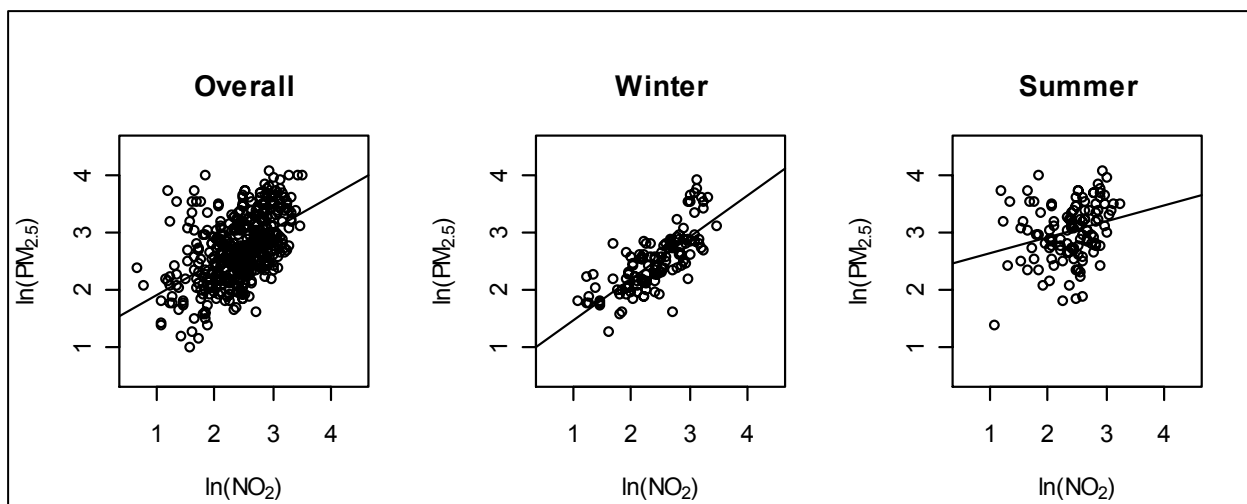


Figure 39. Scatterplots showing the association between log-transformed daily PM_{2.5} concentrations and log-transformed daily NO₂ concentrations measured at the central Steubenville site during SCAMP, with linear least squares lines. The leftmost plot shows all data collected during SCAMP; the center and rightmost plots show only those data that were collected during the winter and summer, respectively.

In contrast to the seasonal trends evident in the correlations between primary gaseous pollutants and PM_{2.5} and PM₁₀, correlations between O₃, a secondary pollutant, and daily measures of these PM variables were positive and moderately strong during summer and negative and moderately strong during winter. This discrepancy likely results from seasonal differences in the factors that influence ambient mass concentrations of PM_{2.5} and PM₁₀ in Steubenville. As discussed earlier, PM_{2.5} mass during summer in Steubenville is dominated by secondary sulfates, which tend to be formed on a regional scale during the same photochemically-active conditions that favor O₃ formation. This is consistent with the moderate,

positive correlation observed between $PM_{2.5}$ and O_3 during summer. It is noteworthy, however, that whereas daily $PM_{2.5}$ concentrations were moderately correlated with daily O_3 concentrations during summer ($r_s = 0.47$), hourly concentrations of these species were uncorrelated ($r_s = 0.06$) during that season, suggesting that peak concentrations tended to occur during different times of the day. NO_x , CO, and SO_2 , which are expected to result predominantly from primary emission sources in the Steubenville area, were more weakly correlated with $PM_{2.5}$ during summer, when secondary production of $PM_{2.5}$ is maximal, than during fall and winter, when a larger percentage of total $PM_{2.5}$ mass is expected to result from local, primary emission sources than from regional, secondary $PM_{2.5}$ formation. Correlations of NO_x , CO, and SO_2 with $PM_{2.5}$ were similar regardless of whether daily or hourly data were used, suggesting some similarity in the diurnal behavior of these pollutants. This is consistent with the average diurnal profiles shown in Figures 24 and 25. Daily and hourly O_3 concentrations were negatively correlated with $PM_{2.5}$ during winter, suggesting that a mechanism other than photochemically-assisted secondary formation is dominant in controlling $PM_{2.5}$ mass concentrations during that season.

Correlations of the gaseous pollutants measured at the central Steubenville site during SCAMP with coarse particle ($PM_{10-2.5}$) mass concentrations estimated for that site are consistent with the above interpretation. Daily concentrations of $PM_{10-2.5}$, which is expected to result largely from primary emissions in the Steubenville area, showed less-pronounced seasonal variations in their associations with NO, NO_2 , and CO concentrations than did daily $PM_{2.5}$ concentrations, and did not exhibit any appreciable correlation with O_3 concentrations during summer. This suggests that the effect of primary emission sources on $PM_{10-2.5}$ mass concentrations is more consistent throughout the year than the effect of these sources on $PM_{2.5}$ mass concentrations. Correlations between daily concentrations of $PM_{2.5}$ and $PM_{10-2.5}$ at Steubenville were strongest during winter ($r_s = 0.62$) and weakest during summer ($r_s = 0.28$), further supporting the hypothesis that primary sources control the day-to-day variability of $PM_{2.5}$ mass concentrations in Steubenville to a greater extent during summer than during winter.

The correlations presented in Table 30 do not account for the autocorrelated nature of the time series of PM and gaseous pollutant data collected at the central Steubenville site. Connell et al. (2005b) employed ARIMA modeling to account for autocorrelation in a more thorough analysis of the associations between daily $PM_{2.5}$ concentrations and daily gaseous pollutant concentrations measured at the Steubenville site during SCAMP. The application of linear regression analysis to ARIMA-adjusted daily $PM_{2.5}$ and gaseous pollutant concentrations indicated statistically significant, moderate, positive overall associations between $PM_{2.5}$ and SO_2 , NO, NO_2 , and CO ($R^2 = 0.25-0.46$), consistent with the overall Spearman correlations reported in Table 30. The overall association between $PM_{2.5}$ and O_3 , while statistically significant, was weak ($R^2 = 0.029$). Moreover, the associations between $PM_{2.5}$ and NO, NO_2 , CO, and O_3 exhibited a statistically significant interaction with season. Associations of NO and NO_2 with $PM_{2.5}$ were strongest during fall ($R^2 = 0.53$ and 0.61 , respectively; slope = 0.39 and 0.57 , respectively) and weakest during summer ($R^2 = 0.0014$ and 0.086 , respectively; slope = -0.015 and 0.20 , respectively). The association between CO and $PM_{2.5}$ was strongest during winter ($R^2 = 0.60$; slope = 0.66) and weakest during summer ($R^2 = 0.24$; slope = 0.27), whereas the association between O_3 and $PM_{2.5}$ was weakly positive during summer ($R^2 = 0.12$; slope = 0.17) and moderately negative during winter ($R^2 = 0.25$; slope = -0.29). Seasonal variations in the strength of association between SO_2 and $PM_{2.5}$ were not statistically significant.

These results generally confirm the observations made on the basis of Table 30 above, and suggest that confounding by criteria pollutants such as NO_2 , CO, and SO_2 , which have well-established effects on human health, is a possibility in time series epidemiology studies conducted in Steubenville. Moolgavkar and Luebeck (1996) state that "seasonal influences are

effect modifiers in epidemiologic studies of air pollution;” results from the ambient air monitoring portion of SCAMP suggest that in addition to possibly modifying the effect of pollutants on health outcomes, as suggested by Moolgavkar and Luebeck, season can significantly modify the degree of collinearity among pollutants. The correlations above suggest that, as observed by Chock et al. (2000) in nearby Pittsburgh, gaseous pollutants such as CO and NO₂ may cause seasonally-dependent multicollinearity problems in PM_{2.5} time series epidemiology studies focusing on Steubenville. This interaction with season is observed even after autocorrelation (including seasonality) has been removed from the individual pollutant time series. Hence, results from SCAMP reinforce the need for future epidemiology studies to employ separate analyses by season in order to better capture and control for the collinearity between PM_{2.5} and co-pollutants, and suggest that caution should be exercised when interpreting the results of previous epidemiology studies in Steubenville (e.g., Schwartz and Dockery, 1992; Schwartz et al., 1996) and similar locations that did not consider the effects of pollutants such as CO and NO₂ and their seasonal interactions with PM.

Table 31 presents Spearman correlation coefficients describing the associations between 24-hr average mass concentrations of PM_{2.5}, PM₁₀, and PM_{10-2.5} and corresponding 24-hr number concentrations of total pollen, total spores, and several prevalent types of spores measured at the central Steubenville site. As with the correlations between PM and gaseous pollutants, correlation coefficients were computed using all data collected during SCAMP (“overall”), as well as using data stratified by astronomical season. No appreciable positive correlations were observed between the various PM measures and the various pollen and spore measures, suggesting little possibility that pollen and spores would confound the results of PM epidemiology studies in Steubenville. (The strongest positive correlation, $r_s = 0.34$, was between total pollen and PM_{10-2.5} in spring, when pollen counts in Steubenville are maximal).

Table 31. Spearman correlation coefficients between daily ambient concentrations of PM_{2.5}, PM₁₀, and PM_{10-2.5} and daily ambient pollen and spore counts at the central Steubenville site, both overall and by astronomical season.

Pollen/Spore Type	PM Measure	Spearman Correlation Coefficient				
		Overall	Winter	Spring	Summer	Fall
Ascospores	PM _{2.5}	0.10	-0.07	0.04	-0.15	-0.20
	PM ₁₀	0.03	-0.13	-0.06	-0.33	-0.30
	PM _{10-2.5}	-0.18	-0.27	-0.10	-0.52	-0.40
Basidiospores	PM _{2.5}	0.28	-0.15	0.18	0.15	0.04
	PM ₁₀	0.21	-0.20	0.13	-0.01	-0.06
	PM _{10-2.5}	-0.03	-0.32	0.11	-0.25	-0.21
Cladosporium	PM _{2.5}	0.29	-0.04	0.16	0.24	0.04
	PM ₁₀	0.27	0.02	0.21	0.12	0.05
	PM _{10-2.5}	0.12	0.02	0.32	-0.02	0.05
Smut/Myxo/Periconia	PM _{2.5}	0.17	0.09	0.06	-0.16	0.06
	PM ₁₀	0.22	0.23	0.22	-0.11	0.16
	PM _{10-2.5}	0.19	0.27	0.25	-0.04	0.32
Total Spores	PM _{2.5}	0.25	-0.07	0.12	0.09	-0.02
	PM ₁₀	0.19	-0.08	0.08	-0.11	-0.07
	PM _{10-2.5}	-0.03	-0.17	0.09	-0.36	-0.17
Total Pollen	PM _{2.5}	0.07	-0.07	0.11	0.06	-0.04
	PM ₁₀	0.16	0.03	0.25	0.07	-0.05
	PM _{10-2.5}	0.17	0.09	0.34	0.23	0.04

Table 32 similarly presents Spearman correlation coefficients describing the overall and seasonal associations between concentrations of PM_{2.5} (both 1-hr and 24-hr), PM₁₀, and PM_{10-2.5} and simultaneously-measured meteorological conditions at the central Steubenville site. (One-hour average meteorological condition values were used when computing the correlations for 1-hr average PM_{2.5} concentrations). As shown in the table, wind speed exhibited a negative association with all of the PM measures, likely reflecting the tendency of air pollutant concentrations to build up during periods of poor ventilation characterized by low wind speeds. As indicated in the table and confirmed by Connell et al. (2005b) using more advanced time series analyses, the association between daily PM_{2.5} and daily average wind speed varied appreciably by season, with the strongest negative association observed during winter (when local primary sources are expected to contribute a larger relative proportion of total PM_{2.5} mass in Steubenville) and a much weaker negative association observed during summer (when increases of locally-emitted PM_{2.5} during periods of stagnation have less of an effect on secondary-sulfate-dominated total PM_{2.5} mass). Correlations of daily PM_{2.5} with daily temperature exhibited an opposite seasonal trend. The strongest association between these variables was observed during the summer, when secondary particle formation is favored on warm, photochemically-active days. The summertime correlation between PM_{2.5} and temperature was substantially stronger for 24-hr average data than for 1-hr average data, though, suggesting that the greatest PM_{2.5} concentrations did not necessarily occur during the warmest times of day.

Table 32. Spearman correlation coefficients between ambient concentrations of various PM measures (i.e., hourly PM_{2.5}, daily PM_{2.5}, daily PM₁₀, and daily PM_{10-2.5}) and meteorological conditions at the central Steubenville site, both overall and by astronomical season.

Meteorological Parameter	PM Measure	Spearman Correlation Coefficient				
		Overall	Winter	Spring	Summer	Fall
Wind Speed	PM _{2.5} – 1-h	-0.38	-0.38	-0.31	-0.08	-0.39
	PM _{2.5} – 24-h	-0.43	-0.48	-0.30	-0.03	-0.47
	PM ₁₀ – 24-h	-0.42	-0.48	-0.25	-0.12	-0.48
	PM _{10-2.5} – 24-h	-0.28	-0.37	-0.20	-0.25	-0.25
Temperature	PM _{2.5} – 1-h	0.42	0.22	0.38	0.36	0.22
	PM _{2.5} – 24-h	0.50	0.33	0.50	0.71	0.40
	PM ₁₀ – 24-h	0.49	0.31	0.53	0.64	0.36
	PM _{10-2.5} – 24-h	0.26	0.29	0.42	0.14	0.24
Relative Humidity	PM _{2.5} – 1-h	0.23	0.17	0.21	0.19	0.20
	PM _{2.5} – 24-h	0.08	-0.10	0.07	0.03	0.04
	PM ₁₀ – 24-h	-0.12	-0.31	-0.20	-0.11	-0.16
	PM _{10-2.5} – 24-h	-0.43	-0.54	-0.39	-0.42	-0.51
Solar Radiation	PM _{2.5} – 1-h	0.03	-0.08	-0.03	-0.04	-0.02
	PM _{2.5} – 24-h	0.34	0.18	0.21	0.24	0.24
	PM ₁₀ – 24-h	0.46	0.38	0.42	0.31	0.38
	PM _{10-2.5} – 24-h	0.47	0.55	0.49	0.50	0.57
Barometric Pressure	PM _{2.5} – 1-h	0.16	0.05	0.11	0.03	0.21
	PM _{2.5} – 24-h	0.29	0.23	0.24	0.07	0.31
	PM ₁₀ – 24-h	0.36	0.35	0.41	0.15	0.35
	PM _{10-2.5} – 24-h	0.37	0.43	0.43	0.38	0.31

As expected, $PM_{10-2.5}$, which consists predominantly of primary rather than secondary particles, did not correlate strongly with temperature during summertime. Rather, coarse particles correlated more strongly with relative humidity (negative association), solar radiation (positive association), and barometric pressure (positive association) than did fine particles. This suggests that elevated ambient concentrations of coarse PM in all seasons tended to occur on clear, dry, high-pressure days. Winds are capable of suspending coarse particles; however, as shown in Table 32, coarse particles at Steubenville exhibited a weakly negative association with wind speed, suggesting that ventilation may have a larger effect on ambient $PM_{10-2.5}$ concentrations than wind-blown dust.

Figure 40 graphically illustrates the seasonal variability in the strength of the association between daily $PM_{2.5}$ total mass concentrations (log-transformed values are plotted in the figure to improve homoscedasticity) and daily average temperatures at Steubenville. Temperature affects mortality and is commonly controlled for in $PM_{2.5}$ epidemiology studies. However, the data presented in Table 32 and Figure 40 suggest that, like the gaseous pollutants discussed earlier, temperature exhibits a statistically significant seasonal interaction with $PM_{2.5}$. This again reinforces the need for time series studies exploring the health effects of $PM_{2.5}$ to use seasonal models so that these seasonal variations in collinearity with potential confounders can be appropriately accounted for.

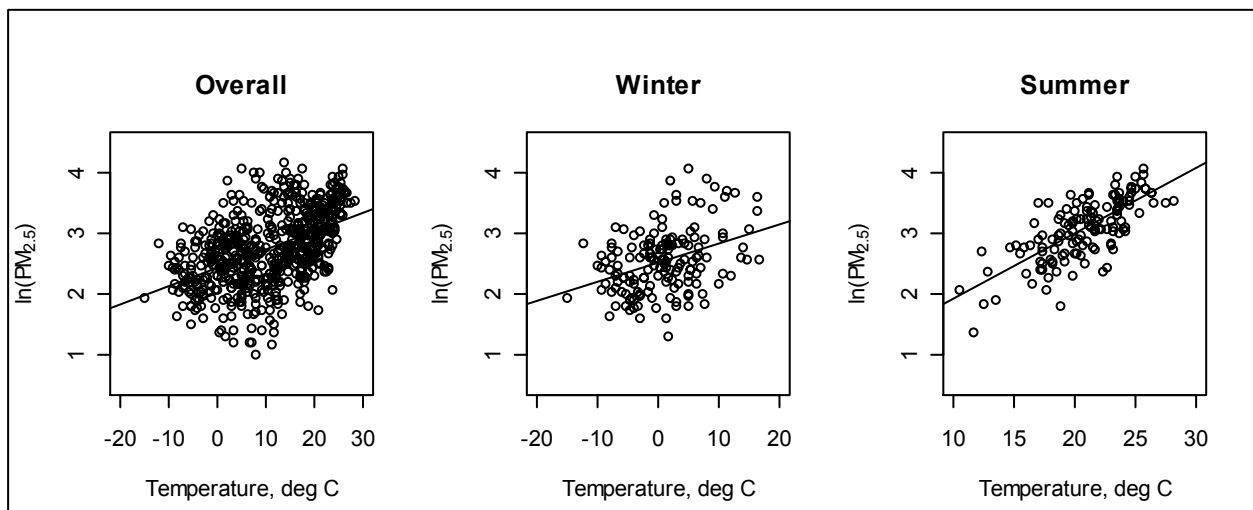


Figure 40. Scatterplots showing the association between log-transformed daily $PM_{2.5}$ concentrations and daily temperatures measured at the central Steubenville site during SCAMP, with linear least squares lines. The leftmost plot shows all data collected during SCAMP; the center and rightmost plots show only those data that were collected during the winter and summer, respectively.

Spearman correlation coefficients describing the overall associations between 1-hr average $PM_{2.5}$, wind speed, and barometric pressure measurements at Steubenville during SCAMP are relatively small, as shown in Table 32. However, the scatterplots presented in Figure 41 provide additional insight into the nature of the associations among these variables. As shown in the figure, 1-hr $PM_{2.5}$ concentrations greater than $50 \mu\text{g}/\text{m}^3$ occurred at Steubenville only when the wind speed was moderate-to-low (i.e., $< 3.2 \text{ m/s}$) and the barometric pressure was moderate-to-high (i.e., $> 727 \text{ mm Hg}$). Hours having the greatest wind speeds and the lowest barometric pressures were characterized by relatively low $PM_{2.5}$ concentrations. This again reflects the effects of ventilation and atmospheric stability on $PM_{2.5}$ concentrations at Steubenville. Figure 42 indicates that the greatest $PM_{2.5}$ concentrations at Steubenville also tended to occur during

periods that were free from precipitation, likely resulting from increased instability and particle removal via wet deposition during these precipitation events.

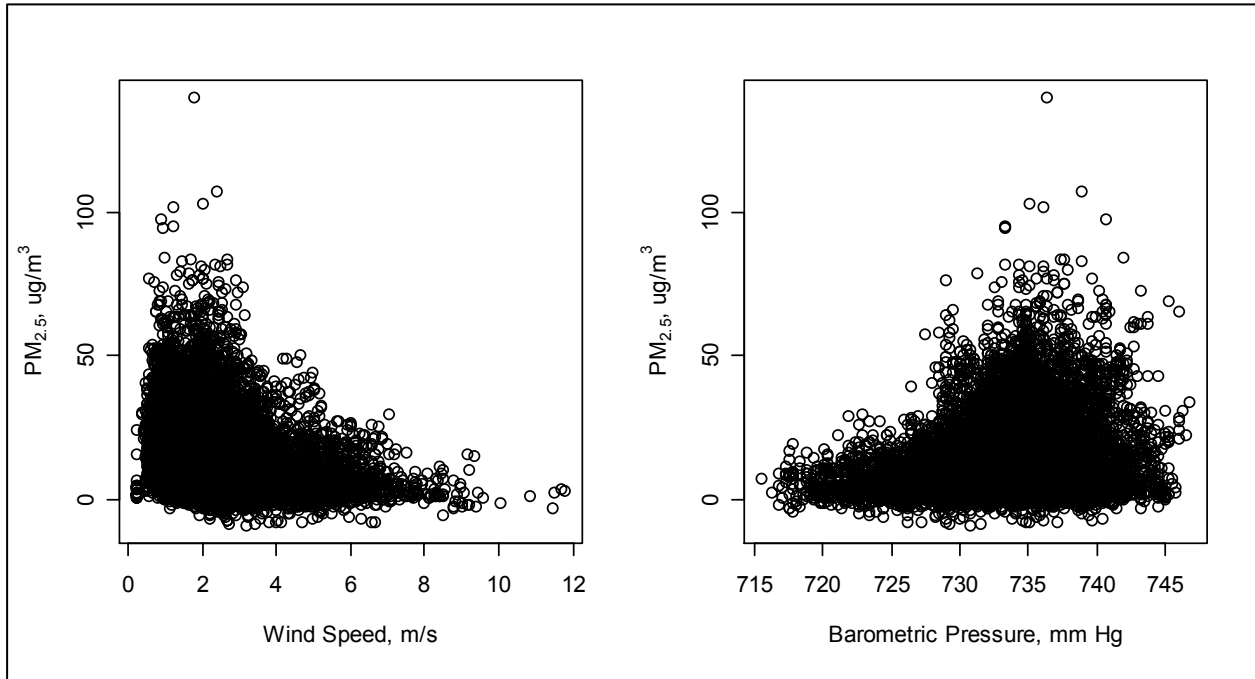


Figure 41. Scatterplots showing 1-hr average PM_{2.5} concentrations versus 1-hr average wind speeds and barometric pressures measured at the central Steubenville site during SCAMP.

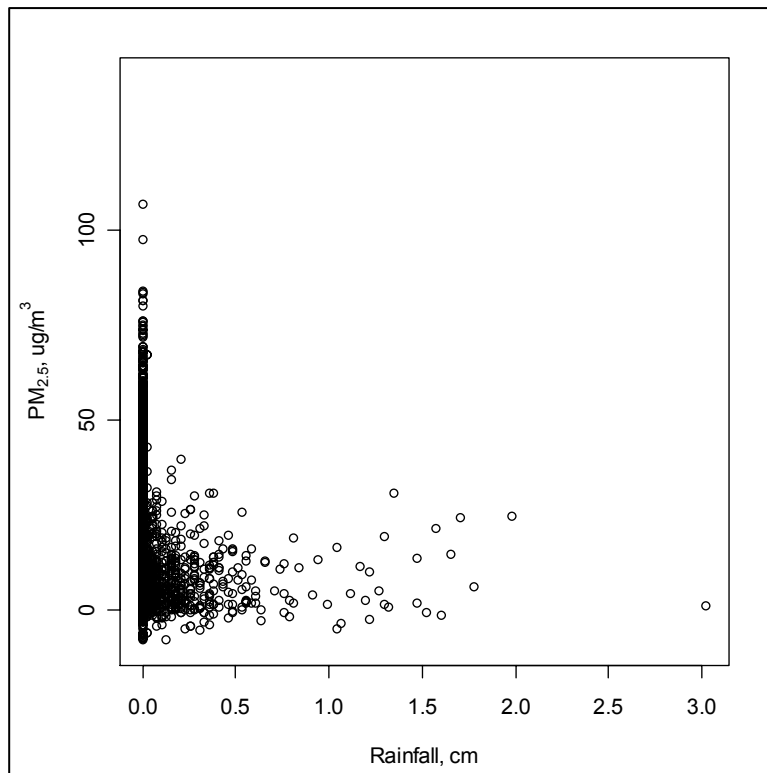


Figure 42. Scatterplot showing 1-hr average PM_{2.5} concentrations versus 1-hr rainfall at the central Steubenville site during SCAMP.

To better understand the associations observed between total PM_{2.5} mass and co-pollutants and meteorological conditions at Steubenville, it is informative to explore correlations involving chemical components of PM_{2.5}. Table 33 presents Spearman correlation coefficients describing associations between ionic, carbonaceous, and acid-digestible elemental components of PM_{2.5} and corresponding daily total mass concentrations of PM_{2.5}, PM₁₀, and PM_{10-2.5} at Steubenville. As expected, NH₄⁺ and SO₄²⁻ correlated strongly with total PM_{2.5} mass, as (NH₄)₂SO₄ is the largest contributor to PM_{2.5} mass at Steubenville. However, a number of other PM_{2.5} components, including organic material, elemental carbon, and even elements such as Fe, K, Mn, Se, Sn, and Zn, exhibited moderately strong (i.e., r_s = 0.66-0.78) correlations with PM_{2.5} mass as well. These correlations allow for the possibility that total PM_{2.5} mass could act as a surrogate not only for its major components (e.g., SO₄²⁻, OM, and NH₄⁺), but also for its minor but toxicologically relevant trace metal constituents. These trace constituents are subject to greater measurement noise than are total PM_{2.5} mass and major PM_{2.5} components; Lipfert and Wyzga (1997) point out that “In joint regressions of two correlated variables with differing exposure errors, the entire effect may be shifted onto the more precisely measured variable.” Hence, the appreciable correlations observed between total PM_{2.5} mass and both major and minor chemical constituents reinforce the need for further research to elucidate the health implications of these individual chemical components.

Table 33. Spearman correlations between concentrations of PM_{2.5} ionic, carbonaceous, and acid-digestible elemental components and concentrations of PM_{2.5}, PM₁₀, and PM_{10-2.5} total mass measured at the central Steubenville site during SCAMP.

	PM_{2.5}	PM₁₀	PM_{10-2.5}
NH₄⁺	0.89	0.75	0.34
SO₄²⁻	0.91	0.79	0.39
NO₃⁻	0.17	0.10	0.03
Cl⁻	0.44	0.46	0.41
EC	0.68	0.73	0.65
OM	0.78	0.84	0.73
Al	0.58	0.67	0.60
As	0.70	0.74	0.64
Ba	0.61	0.73	0.70
Ca	0.55	0.62	0.66
Cd	0.55	0.53	0.47
Co	0.64	0.65	0.60
Cu	0.53	0.54	0.54
Fe	0.67	0.75	0.71
K	0.67	0.75	0.65
Mg	0.57	0.70	0.74
Mn	0.66	0.71	0.70
Ni	0.54	0.56	0.49
Pb	0.63	0.64	0.56
Se	0.71	0.67	0.48
Sn	0.66	0.64	0.59
Ti	0.30	0.37	0.41
V	0.52	0.54	0.52
Zn	0.70	0.70	0.62

Whereas the largely secondary components of PM_{2.5} (i.e., NH₄⁺, SO₄²⁻, NO₃⁻) exhibited relatively weak correlations ($r_s = 0.03-0.39$) with coarse PM mass, a number of other PM_{2.5} components (e.g., EC, OM, Ba, Ca, Fe, K, Mg, Mn) showed moderately strong correlations (i.e., $r_s = 0.65-0.74$) with PM_{10-2.5}. This suggests that common or closely-situated primary emission sources likely affect ambient concentrations of both coarse PM and these PM_{2.5} constituents.

Table 34 presents a matrix showing Spearman correlations among ionic, carbonaceous, and acid-digestible elemental components of PM_{2.5} that were measured at the central Steubenville site. As indicated in the table, a number of the PM_{2.5} components at Steubenville were moderately-to-strongly collinear; all of the components except for NO₃⁻, Cl⁻, Cu, Se, and Ti exhibited a correlation ≥ 0.70 with at least one other component. This suggests the potential for collinearity problems in time series epidemiology studies focusing on chemical components of PM_{2.5} in Steubenville. Several of the correlations reported in Table 34 are readily explainable on the basis of emission profiles and atmospheric chemistry. For example, SO₄²⁻ and NH₄⁺ correlated more strongly with each other ($r_s = 0.91$) than with any other PM_{2.5} components, and Se correlated more strongly with SO₄²⁻ ($r_s = 0.68$) than with any other PM_{2.5} component. These associations are expected, as secondary SO₄²⁻ particles are generally present as (NH₄)₂SO₄ or NH₄HSO₄ in Steubenville, and Se is often used as a tracer for coal-fired power plant emissions, which are a major source of secondary SO₄²⁻ in the Steubenville region.

Many of the strongest intercomponent associations were observed among a group of elements comprising As, Fe, Mn, Pb, Sn, and Zn. Spearman correlations among these components ranged from 0.76 to 0.90 for all possible component pairs. Concentrations of each of these six elements were also more strongly associated with concentrations of elemental carbon and organic material ($r_s = 0.67-0.78$) than with concentrations of SO₄²⁻ ($r_s = 0.50-0.59$). (Concentrations of Fe and Mn were also strongly associated with concentrations of Mg, with Spearman correlation coefficients of 0.85 and 0.80, respectively). These strong associations may indicate a common emission source for these elements, or alternatively may reflect the influence of meteorological conditions on ambient concentrations of pollutants originating from several sources. Elevated concentrations of particulate Fe and Mn are often found near steel production facilities (Dusseldorp et al., 1995), and based on the U.S. EPA's 2001 Toxic Release Inventory (TRI) for air emissions, the primary metals industry accounts for 93% of Zn emissions and 83% of Mn emissions from stationary sources in Jefferson County, OH, where Steubenville is located. Hence, the correlations observed among Fe, Mn, and Zn are likely at least in part attributable to Steubenville's iron and steel facilities. Further analyses of sources of PM_{2.5} in Steubenville are presented in section 4.2.5 of this report.

Table 35 summarizes overall Spearman correlation coefficients between daily concentrations of PM_{2.5} chemical components and daily average gaseous pollutant concentrations and meteorological conditions. SO₂ correlated more strongly with Se ($r_s = 0.78$) than with any of the other PM_{2.5} components, likely reflecting the common emission of these two species from coal-fired electric power plants and other coal combustion sources. Se was also well-correlated with NO₂, which results from coal combustion emissions. Again, many of the strongest correlations involved the group of six collinear elements identified above (i.e., As, Fe, Mn, Pb, Sn, and Zn). These elements, as well as elemental carbon, exhibited appreciable associations with NO, NO₂, NO_x, and CO; pairwise Spearman correlation coefficients describing these associations ranged from 0.62 to 0.84. Correlations of Ba with NO₂ and K with CO were also > 0.70 , perhaps reflecting common combustion sources of these species. None of the PM_{2.5} components correlated well with O₃.

Table 34. Matrix of Spearman correlations computed for pairwise concentrations of PM_{2.5} ionic, carbonaceous, and acid-digestible elemental components measured at the central Steubenville site during SCAMP.

	NH ₄ ⁺	SO ₄ ²⁻	NO ₃ ⁻	Cl ⁻	EC	OM	Al	As	Ba	Ca	Cd	Co	Cu	Fe	K	Mg	Mn	Ni	Pb	Se	Sn	Ti	V	Zn
NH ₄ ⁺		0.91	0.26	0.27	0.46	0.52	0.39	0.46	0.37	0.31	0.40	0.50	0.36	0.40	0.46	0.35	0.39	0.40	0.38	0.58	0.42	0.26	0.38	0.45
SO ₄ ²⁻	0.91		0.01	0.36	0.55	0.60	0.49	0.59	0.48	0.46	0.50	0.59	0.43	0.54	0.54	0.48	0.52	0.56	0.50	0.68	0.53	0.32	0.49	0.56
NO ₃ ⁻	0.26	0.01		0.13	0.02	0.05	-0.04	-0.09	-0.09	-0.02	-0.02	0.08	-0.05	-0.09	0.09	-0.03	-0.06	-0.13	-0.08	0.05	-0.01	-0.05	-0.10	-0.03
Cl ⁻	0.27	0.36	0.13		0.60	0.43	0.41	0.54	0.49	0.47	0.39	0.56	0.48	0.61	0.46	0.58	0.59	0.49	0.55	0.42	0.56	0.32	0.43	0.55
EC	0.46	0.55	0.02	0.60		0.71	0.56	0.75	0.70	0.62	0.51	0.58	0.64	0.75	0.64	0.64	0.73	0.48	0.71	0.55	0.69	0.40	0.48	0.71
OM	0.52	0.60	0.05	0.43	0.71		0.71	0.78	0.75	0.62	0.48	0.52	0.56	0.73	0.78	0.66	0.70	0.49	0.67	0.59	0.69	0.30	0.56	0.72
Al	0.39	0.49	-0.04	0.41	0.56	0.71		0.64	0.65	0.72	0.41	0.42	0.38	0.70	0.78	0.71	0.66	0.52	0.54	0.58	0.57	0.28	0.50	0.62
As	0.46	0.59	-0.09	0.54	0.75	0.78	0.64		0.72	0.59	0.68	0.63	0.65	0.85	0.75	0.72	0.78	0.64	0.86	0.59	0.81	0.38	0.64	0.85
Ba	0.37	0.48	-0.09	0.49	0.70	0.75	0.65	0.72		0.70	0.53	0.64	0.64	0.75	0.64	0.72	0.69	0.60	0.63	0.50	0.61	0.49	0.59	0.67
Ca	0.31	0.46	-0.02	0.47	0.62	0.62	0.72	0.59	0.70		0.53	0.63	0.52	0.73	0.62	0.77	0.78	0.55	0.53	0.54	0.58	0.53	0.55	0.66
Cd	0.40	0.50	-0.02	0.39	0.51	0.48	0.41	0.68	0.53	0.53		0.59	0.66	0.59	0.47	0.52	0.62	0.71	0.70	0.39	0.64	0.38	0.62	0.69
Co	0.50	0.59	0.08	0.56	0.58	0.52	0.42	0.63	0.64	0.63	0.59		0.62	0.67	0.47	0.58	0.67	0.70	0.55	0.61	0.61	0.49	0.57	0.63
Cu	0.36	0.43	-0.05	0.48	0.64	0.56	0.38	0.65	0.64	0.52	0.66	0.62		0.64	0.53	0.52	0.64	0.60	0.65	0.38	0.69	0.51	0.57	0.67
Fe	0.40	0.54	-0.09	0.61	0.75	0.73	0.70	0.85	0.75	0.73	0.59	0.67	0.64		0.77	0.85	0.90	0.62	0.79	0.59	0.81	0.41	0.59	0.87
K	0.46	0.54	0.09	0.46	0.64	0.78	0.78	0.75	0.64	0.62	0.47	0.47	0.53	0.77		0.72	0.71	0.46	0.65	0.56	0.59	0.23	0.47	0.73
Mg	0.35	0.48	-0.03	0.58	0.64	0.66	0.71	0.72	0.72	0.77	0.52	0.58	0.52	0.85	0.72		0.80	0.55	0.65	0.48	0.61	0.39	0.50	0.73
Mn	0.39	0.52	-0.06	0.59	0.73	0.70	0.66	0.78	0.69	0.78	0.62	0.67	0.64	0.90	0.71	0.80		0.61	0.77	0.63	0.76	0.44	0.60	0.85
Ni	0.40	0.56	-0.13	0.49	0.48	0.49	0.52	0.64	0.60	0.55	0.71	0.70	0.60	0.62	0.46	0.55	0.61		0.63	0.56	0.65	0.41	0.70	0.62
Pb	0.38	0.50	-0.08	0.55	0.71	0.67	0.54	0.86	0.63	0.53	0.70	0.55	0.65	0.79	0.65	0.65	0.77	0.63		0.51	0.81	0.28	0.63	0.86
Se	0.58	0.68	0.05	0.42	0.55	0.59	0.58	0.59	0.50	0.54	0.39	0.61	0.38	0.59	0.56	0.48	0.63	0.56	0.51		0.56	0.32	0.49	0.57
Sn	0.42	0.53	-0.01	0.56	0.69	0.69	0.57	0.81	0.61	0.58	0.64	0.61	0.69	0.81	0.59	0.61	0.76	0.65	0.81	0.56		0.42	0.69	0.82
Ti	0.26	0.32	-0.05	0.32	0.40	0.30	0.28	0.38	0.49	0.53	0.38	0.49	0.51	0.41	0.23	0.39	0.44	0.41	0.28	0.32	0.42		0.57	0.36
V	0.38	0.49	-0.10	0.43	0.48	0.56	0.50	0.64	0.59	0.55	0.62	0.57	0.57	0.59	0.47	0.50	0.60	0.70	0.63	0.49	0.69	0.57		0.66
Zn	0.45	0.56	-0.03	0.55	0.71	0.72	0.62	0.85	0.67	0.66	0.69	0.63	0.67	0.87	0.73	0.73	0.85	0.62	0.86	0.57	0.82	0.36	0.66	

Table 35. Spearman correlations between concentrations of PM_{2.5} ionic, carbonaceous, and acid-digestible elemental components and gaseous co-pollutant concentrations and meteorological conditions measured at the central Steubenville site during SCAMP.

	SO ₂	NO	NO ₂	NO _x	CO	O ₃	WS	Temp	RH	Rad	BP
NH ₄ ⁺	0.37	0.22	0.27	0.24	0.45	0.02	-0.30	0.39	0.23	0.17	0.11
SO ₄ ²⁻	0.44	0.22	0.29	0.26	0.43	0.15	-0.35	0.59	0.19	0.33	0.24
NO ₃ ⁻	0.04	0.18	0.04	0.07	0.23	-0.37	0.12	-0.56	0.03	-0.43	-0.27
Cl ⁻	0.24	0.41	0.30	0.36	0.39	-0.01	-0.18	0.21	-0.14	0.29	0.23
EC	0.51	0.76	0.73	0.79	0.73	-0.23	-0.45	0.27	-0.17	0.25	0.33
OM	0.49	0.57	0.66	0.64	0.68	-0.01	-0.47	0.43	-0.25	0.38	0.36
Al	0.52	0.41	0.63	0.58	0.60	0.03	-0.16	0.36	-0.38	0.30	0.29
As	0.52	0.72	0.80	0.80	0.76	-0.06	-0.55	0.41	-0.23	0.43	0.41
Ba	0.43	0.60	0.74	0.68	0.57	-0.02	-0.46	0.41	-0.40	0.41	0.39
Ca	0.57	0.59	0.66	0.63	0.52	-0.18	-0.21	0.23	-0.41	0.27	0.46
Cd	0.35	0.59	0.63	0.63	0.48	-0.22	-0.42	0.23	-0.09	0.19	0.42
Co	0.57	0.58	0.62	0.57	0.53	-0.22	-0.28	0.21	-0.27	0.18	0.43
Cu	0.28	0.60	0.62	0.67	0.44	-0.29	-0.48	0.26	-0.14	0.30	0.38
Fe	0.59	0.71	0.76	0.77	0.75	-0.07	-0.36	0.37	-0.32	0.36	0.47
K	0.52	0.52	0.68	0.62	0.74	0.02	-0.19	0.27	-0.38	0.29	0.30
Mg	0.48	0.55	0.61	0.59	0.67	0.10	-0.26	0.34	-0.47	0.47	0.42
Mn	0.62	0.72	0.73	0.77	0.68	-0.21	-0.32	0.29	-0.30	0.27	0.45
Ni	0.45	0.46	0.58	0.54	0.35	-0.06	-0.34	0.42	-0.15	0.26	0.39
Pb	0.44	0.70	0.74	0.78	0.67	-0.15	-0.49	0.35	-0.17	0.33	0.37
Se	0.78	0.56	0.74	0.65	0.57	-0.27	-0.25	0.24	-0.08	0.06	0.16
Sn	0.45	0.75	0.79	0.84	0.62	-0.26	-0.49	0.34	-0.12	0.22	0.45
Ti	0.23	0.48	0.40	0.40	0.22	-0.12	-0.32	0.22	-0.05	0.27	0.33
V	0.32	0.53	0.63	0.61	0.40	-0.04	-0.44	0.42	-0.09	0.31	0.43
Zn	0.52	0.74	0.75	0.80	0.74	-0.17	-0.43	0.32	-0.21	0.31	0.41

NOTE: WS = wind speed; Temp = temperature; RH = relative humidity; Rad = solar radiation; BP = barometric pressure.

Overall correlations of 24-hr average PM_{2.5} chemical component concentrations with 24-hr weather conditions were in most cases similar to those exhibited by total PM_{2.5} mass. A vast majority of the components exhibited weak-to-moderate negative correlations with wind speed, and weak-to-moderate positive correlations with temperature, solar radiation, and barometric pressure. The strongest correlations were a positive correlation between SO₄²⁻ and temperature ($r_s = 0.59$) and a negative correlation between NO₃⁻ and temperature ($r_s = -0.56$). These correlations reflect the similar seasonal trends exhibited by SO₄²⁻ and temperature and the dissimilar trend exhibited by NO₃⁻, and they may also reflect the thermodynamic and kinetic effects of temperature on the secondary formation of these species.

4.2.4.2 Case Studies of PM_{2.5} Episodes

In order to better understand the factors that affect PM_{2.5} and co-pollutant concentrations at Steubenville, it is informative to perform case studies of episodes characterized by elevated concentrations of these species. Many such case studies were conducted as part of the data analysis portion of the SCAMP ambient air monitoring program; results have been discussed in

depth by Connell et al. (2005b, 2005c). Hence, rather than reiterate the results reported there, this section summarizes important findings via a discussion of two representative $PM_{2.5}$ episodes that were not detailed by Connell et al. but illustrate their key conclusions.

The case studies presented by Connell et al. (2005b, 2005c) suggest that two distinctly different types of $PM_{2.5}$ episode commonly occur at Steubenville. Figure 43 presents an example of each of these episode types. The first episode, which occurred from September 5 through September 11, 2001, is characteristic of several episodes that were observed during the warm part of the year (e.g., summer) in Steubenville. As depicted in the Figure, which shows the time series of hourly $PM_{2.5}$ concentrations observed during each episode, this warm season episode consisted of a prolonged period of moderately elevated $PM_{2.5}$ concentrations. Hourly $PM_{2.5}$ concentrations determined by the TEOM at the central Steubenville site remained between 30.2 and 53.5 $\mu\text{g}/\text{m}^3$ during the 45-hour period from 5:00 a.m. on September 7 through 1:00 a.m. on September 9. Maximum 1-hr average concentrations occurred during the early morning or overnight periods (i.e., 6:00 a.m. on September 7; 9:00 p.m. on September 7 through 12:00 a.m. on September 8), likely driven by reduced mixing heights, but $PM_{2.5}$ concentrations did not exhibit a pronounced diurnal pattern, remaining well above average during mid-afternoon on September 7 and September 8.

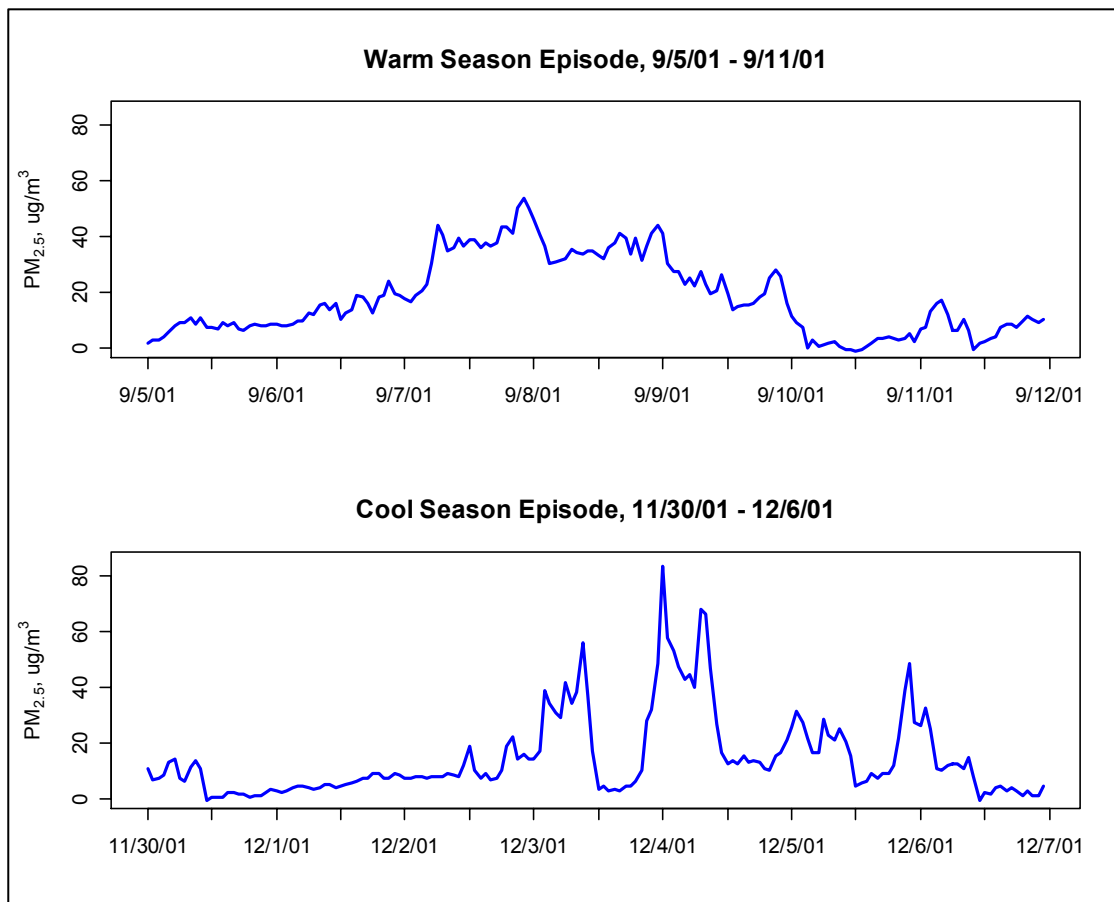


Figure 43. Time series of hourly $PM_{2.5}$ concentrations measured by the TEOM monitor at the central Steubenville site during a warm season episode of elevated $PM_{2.5}$ concentrations, which occurred between September 5 and September 11, 2001, and a cool season episode of elevated $PM_{2.5}$ concentrations, which occurred between November 30 and December 6, 2001.

Conversely, PM_{2.5} concentrations exhibited a very pronounced diurnal profile during the second episode, which occurred from November 30 through December 6, 2001, and is characteristic of a number of episodes that were observed during cooler parts of the year in Steubenville. As shown in Figure 43, PM_{2.5} concentrations during this cool-season episode were acutely elevated during the late evening, overnight, and early morning periods on December 3, December 4, and to a lesser extent, December 5 and December 6, but fell to or below the overall average TEOM-determined PM_{2.5} concentration (12.7 µg/m³) by 12:00 p.m. on each of these days and remained low for the duration of each afternoon. Although the overall average PM_{2.5} concentration during the 3-day “heart” of the warm season episode (i.e., 31.4 µg/m³ determined by the TEOM from 12:00 a.m. on September 7 through 11:00 p.m. on September 9) was greater than that during the 3-day “heart” of the cool season episode (i.e., 24.6 µg/m³ determined by the TEOM from 12:00 a.m. on December 3 through 11:00 p.m. on December 5), the maximum 1-hr concentration during the cool season episode (83.4 µg/m³) was more than 1.5 times as great as that during the warm season episode (53.5 µg/m³). This trend of greater 1-hr maximum PM_{2.5} concentrations during fall, winter, and spring PM_{2.5} episodes than during summer episodes was noted by Connell et al. (2005c); out of 22 episodes characterized by at least one 1-hr PM_{2.5} concentration ≥ 65 µg/m³, the five occurring during June-August had maximum 1-hr PM_{2.5} concentrations ≤ 76.6 µg/m³. Episodes occurring during the non-summer months often had much higher peak concentrations; five episodes observed during the months of October, November, and April had maximum 1-hr PM_{2.5} concentrations ≥ 95.0 µg/m³.

Figures 44 and 45 compare PM_{2.5} concentrations observed during the warm and cool season episodes, respectively, with gaseous pollutant concentrations observed during these episodes. As shown in Figure 45, during the cool season episode, concentrations of PM_{2.5} and concentrations of primary gaseous pollutants, including SO₂, NO, NO₂, and CO, covaried remarkably well, exhibiting very similar diurnal patterns of sharply elevated overnight and morning concentrations and dramatically reduced afternoon concentrations. NO and CO in particular reached very high 1-hr peak concentrations during the episode; maximum concentrations of these species were 17 and 27 times greater, respectively, than their overall mean 1-hr concentrations. Concentrations of O₃, on the other hand, exhibited a diurnal pattern opposite to those displayed by concentrations of PM_{2.5} and the primary gases.

During the warm season episode shown in Figure 44, however, associations between PM_{2.5} concentrations and gaseous pollutant concentrations were much weaker. Concentrations of NO₂, NO (during the early part of the episode), and especially CO continued to show a general diurnal profile of greater nighttime and morning concentrations and lesser afternoon concentrations; however, maximum 1-hr average concentrations of NO and CO were much less than those observed during the cool season episode. The timing of local maxima in the concentration profiles of these gases often corresponded closely to the timing of overnight peaks in PM_{2.5} concentrations during the warm season episode, but PM_{2.5} concentrations remained relatively elevated throughout the afternoon when primary gaseous pollutant concentrations decreased more substantially. O₃ concentrations exhibited greater mid-afternoon maxima during the warm season episode than during the cool season episode, exceeding 70 ppb on three consecutive days, but did not correlate strongly with PM_{2.5} concentrations.

Meteorological conditions are now examined to provide insight into the mechanisms responsible for the different PM_{2.5} and gaseous pollutant behaviors observed during the warm and cool season episodes. Based on case studies of daily data collected at Steubenville during SCAMP, Connell et al. (2005b) concluded that “Episodes of elevated PM_{2.5} concentrations frequently

consisted of an increase in concentration during a period of locally high pressure or elevated temperature, followed by a rapid decrease in concentration with the passage of a frontal system.” The data presented in Figures 46 and 47, which show time series of hourly $PM_{2.5}$ concentrations and meteorological conditions during the warm and cool season episodes being examined, are consistent with this conclusion. As shown in Figure 46, the September 5-11, 2001, episode occurred during a period of locally elevated temperatures. During the 72-hr period of greatest concentration, afternoon maximum temperatures were 28-29 °C, and overnight minima were > 20 °C (with the exception of the very beginning of the period). (To place these temperatures in perspective, maxima were < 25 °C and minima were < 15 °C for days just before and after the episode period). The greatest $PM_{2.5}$ concentrations during the cool season episode, which is depicted in Figure 47, occurred during local minima in temperature; however, this episode happened during a period of high barometric pressure. The association between barometric pressure and $PM_{2.5}$ concentrations during the warm season episode was more complex but generally consistent with the observation of Connell et al. (2005b) that, on the basis of cross-correlation analyses, barometric pressure exhibited a unique lagged relationship with $PM_{2.5}$. Specifically, “an increase in barometric pressure precedes an increase in $PM_{2.5}$ concentration by ~1 day, and a decrease in $PM_{2.5}$ concentration with the passage of a frontal system is followed by an increase in barometric pressure ~1 day later.” The arrows in the bottommost subplots in Figures 46 and 47 denote the passage of frontal systems through Steubenville. Elevated $PM_{2.5}$ concentrations during both the warm and cool season episodes occurred during periods when no frontal systems passed through the area (Unisys Weather, 2006). (The greatest $PM_{2.5}$ concentrations during the September 5-11, 2001, episode occurred when a front was stalled to the west of Steubenville and an area of high pressure was located off the coast of the eastern United States). The episodes ended with the passage of cold fronts through Steubenville on the mornings of September 10, 2001, and December 6, 2001. As shown in Figures 46 and 47, $PM_{2.5}$ concentrations dropped to near-zero with the passage of these frontal systems. These abrupt decreases in concentration could be caused by increased atmospheric instability associated with the frontal systems, increased removal of $PM_{2.5}$ from the atmosphere by precipitation that commonly accompanies the frontal systems, or the arrival of cleaner air masses with the passage of the frontal systems. Similar behavior was observed during many of the episodes of elevated $PM_{2.5}$ concentrations that occurred during SCAMP.

Figure 46 reveals no other noteworthy trends between $PM_{2.5}$ concentrations and meteorological conditions during the September 5-11, 2001, $PM_{2.5}$ episode. However, as shown in Figure 47, during the cool season episode days having the highest $PM_{2.5}$ concentrations (e.g., 12/3/01 and 12/4/01), the diurnal variability of $PM_{2.5}$ concentrations was distinctly similar to that of relative humidity and opposite that of wind speed, temperature, and solar radiation. Elevated $PM_{2.5}$ concentrations during this episode occurred during a period of relatively clear skies, as evidenced by the increased levels of solar radiation during the heart of the episode as compared to the surrounding days. This episode, like a number of others that occurred during cooler portions of the year in Steubenville (Connell et al., 2005c), was likely driven by nocturnal temperature inversions, which trap primary emissions of $PM_{2.5}$ and co-pollutants (e.g., CO, NO_x , and SO_2) in a small volume of air close to ground level and thereby allow their concentrations to build up. (Conversely, concentrations of O_3 are appreciably depleted during nocturnal temperature inversions, as photochemical production of O_3 stops, and O_3 is removed from the atmosphere by deposition and reaction with concentrated levels of primary gases such as NO).

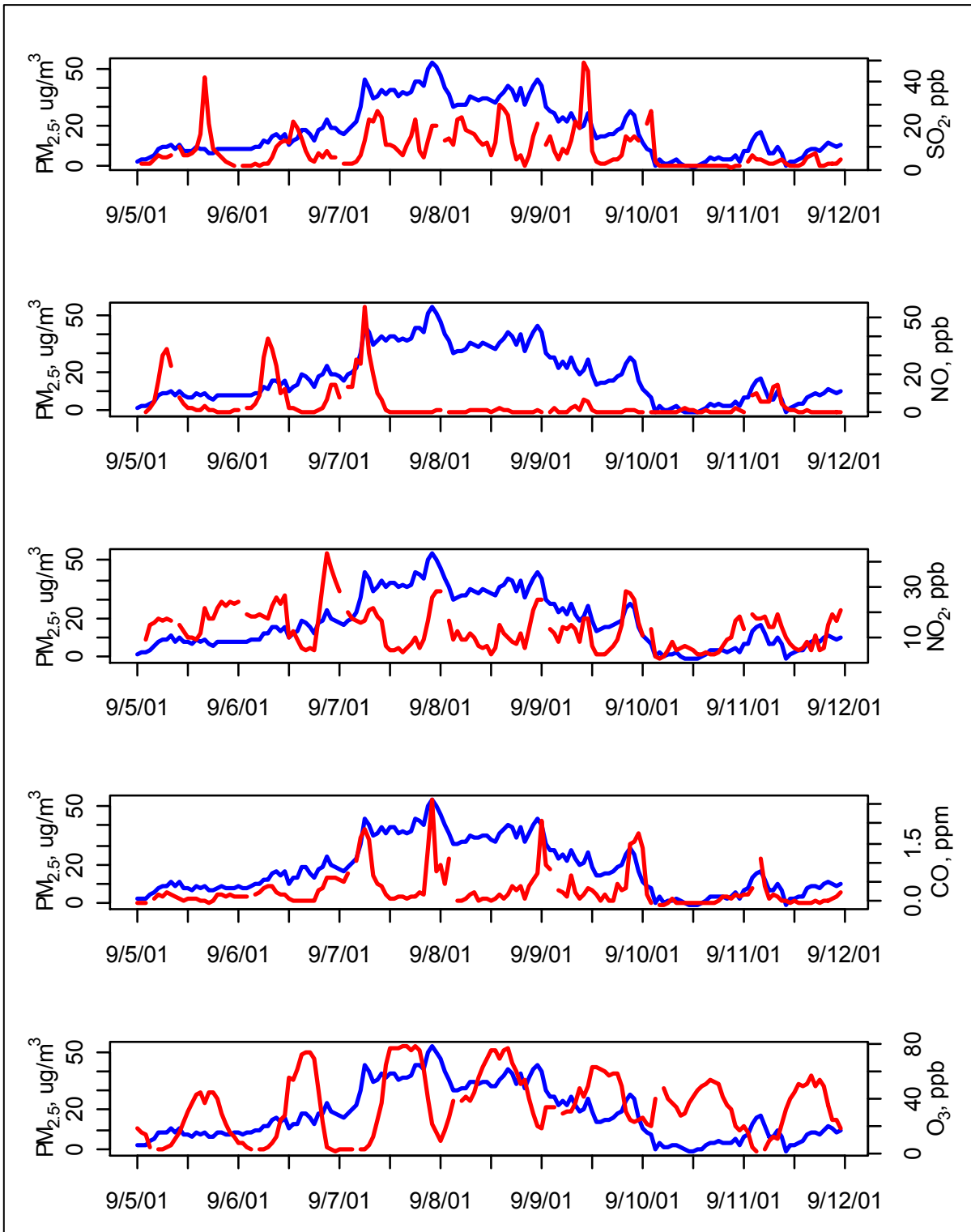


Figure 44. Time series of 1-hr PM_{2.5} concentrations (blue) and gaseous pollutant concentrations (red) observed at the central Steubenville site on September 5-11, 2001.

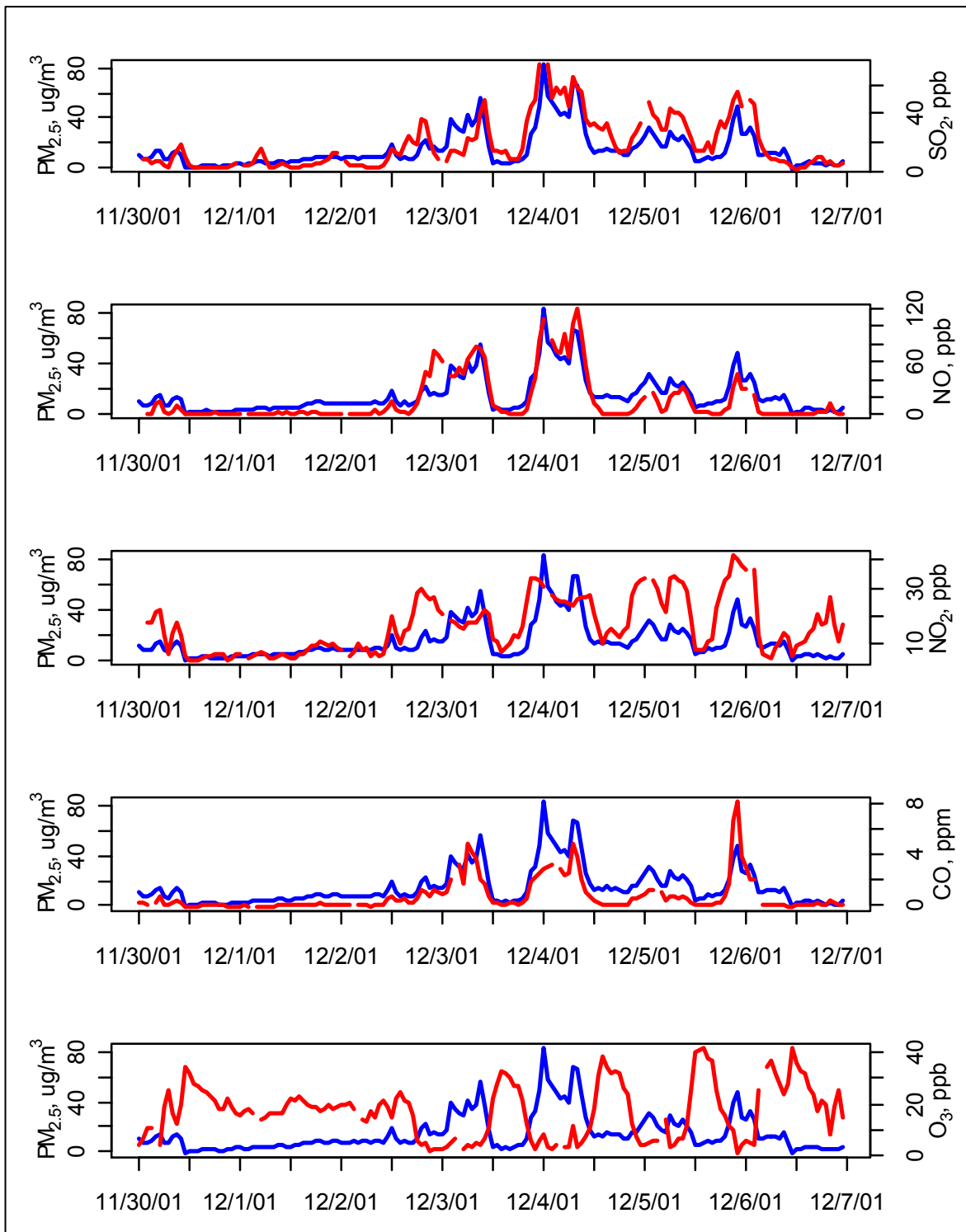


Figure 45. Time series of 1-hr PM_{2.5} concentrations (blue) and gaseous pollutant concentrations (red) observed at the central Steubenville site on November 30 - December 6, 2001.

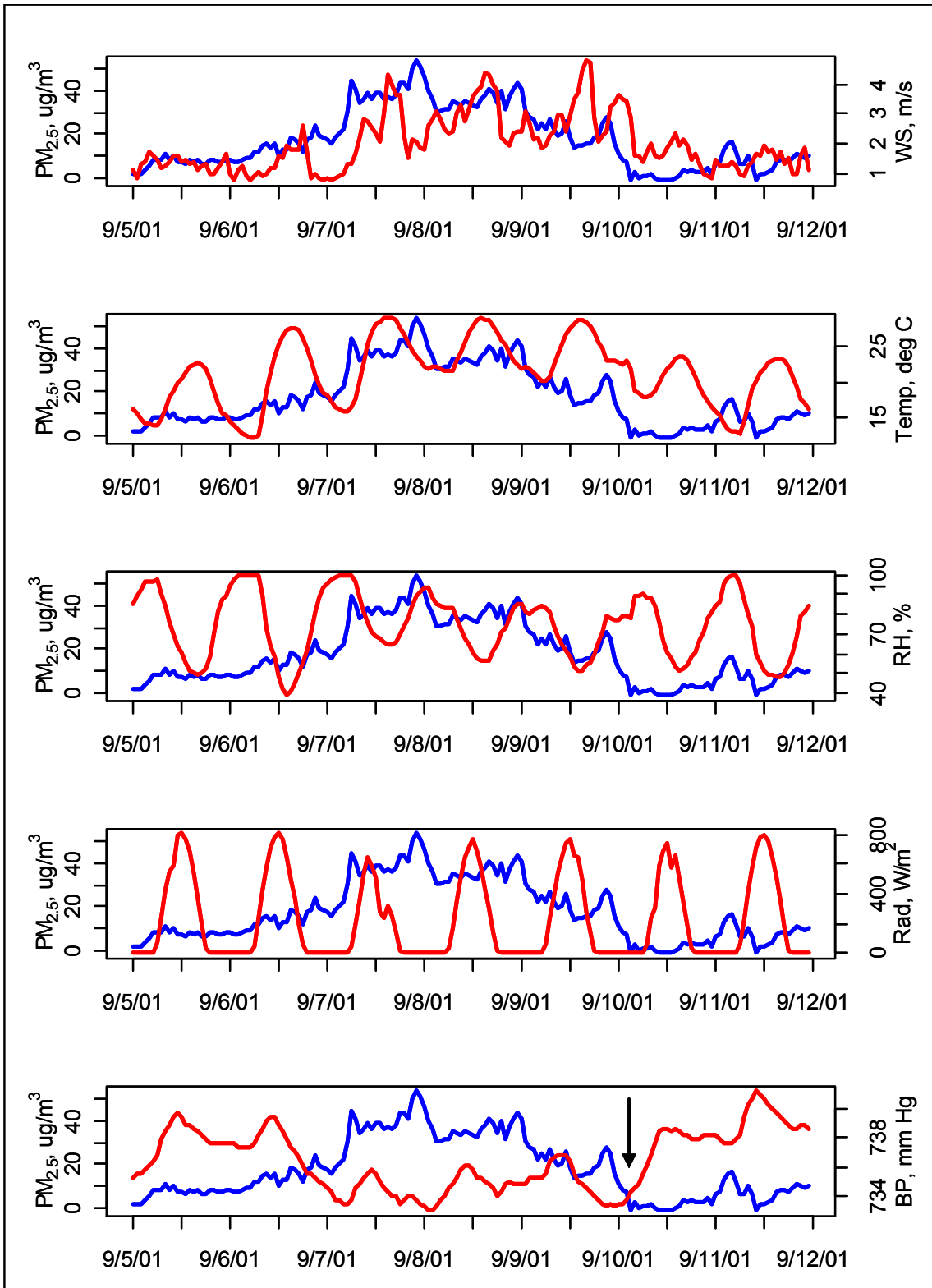


Figure 46. Time series of 1-hr PM_{2.5} concentrations (blue) and meteorological conditions (red) observed at the central Steubenville site on September 5-11, 2001.

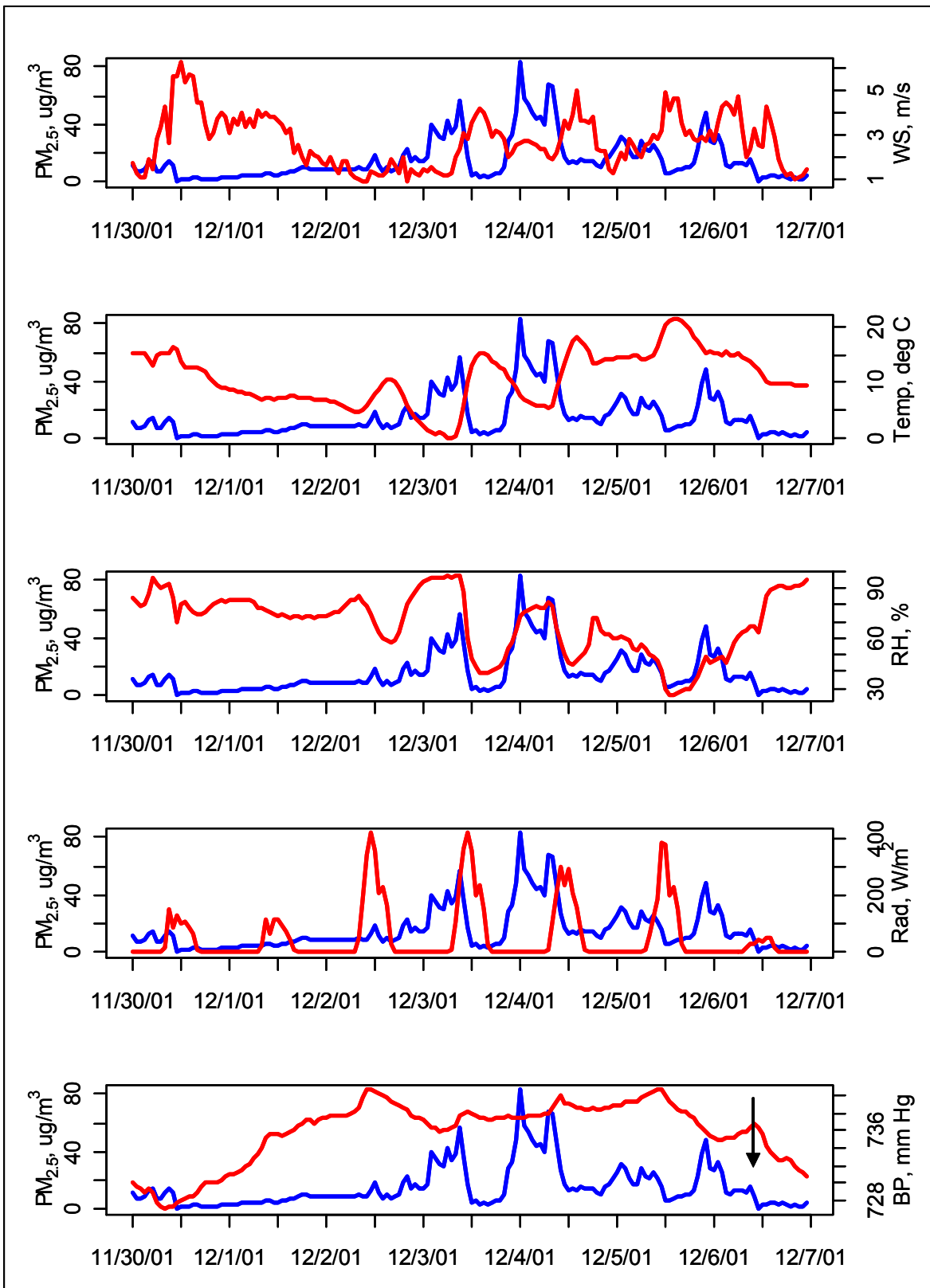


Figure 47. Time series of 1-hr PM_{2.5} concentrations (blue) and meteorological conditions (red) observed at the central Steubenville site on November 30 - December 6, 2001.

Nocturnal inversions result from radiational cooling of the earth's surface, which occurs most readily during periods of clear skies; such cooling typically results in a pronounced diurnal surface temperature profile. Relative humidity near ground level increases during the inversion, whereas wind speed decreases as a result of the vertical instability resulting from the inversion. This is consistent with the behavior depicted in Figure 47. As shown in the figure, the periods of elevated PM_{2.5} concentrations during the December 2001 episode ended abruptly each day during the mid-to-late morning, as increases in solar radiation heated the earth's surface, causing the inversions to break.

Temperature inversions are characterized by negative lapse rates near the surface, meaning that the air temperature increases with increasing height above ground level in the inversion layer. Temperature soundings were not taken at the Steubenville site during SCAMP but were available from the National Weather Service station in Moon Township, PA, which is located about 38 km east-northeast of Steubenville. As shown in Figure 48, which presents vertical temperature soundings taken at Moon Township at 7:00 a.m. EST on each day during the warm and cool season episodes being studied, very strong temperature inversions were observed on the two days during the cool season episode (i.e., 12/3/01 and 12/4/01) for which PM_{2.5} concentrations at 7:00 a.m. were the greatest (i.e., 34.0 and 67.7 µg/m³, respectively). On 12/5/01, the inversion at 7:00 a.m. EST was weaker, and the average PM_{2.5} concentration for that hour was a more modest 23 µg/m³. PM_{2.5} concentrations at 7:00 a.m. EST on 11/30/01, 12/1/01, 12/2/01, and 12/6/01 were ≤ 12.4 µg/m³; these days had only weak surface inversions or no surface inversions at that time.

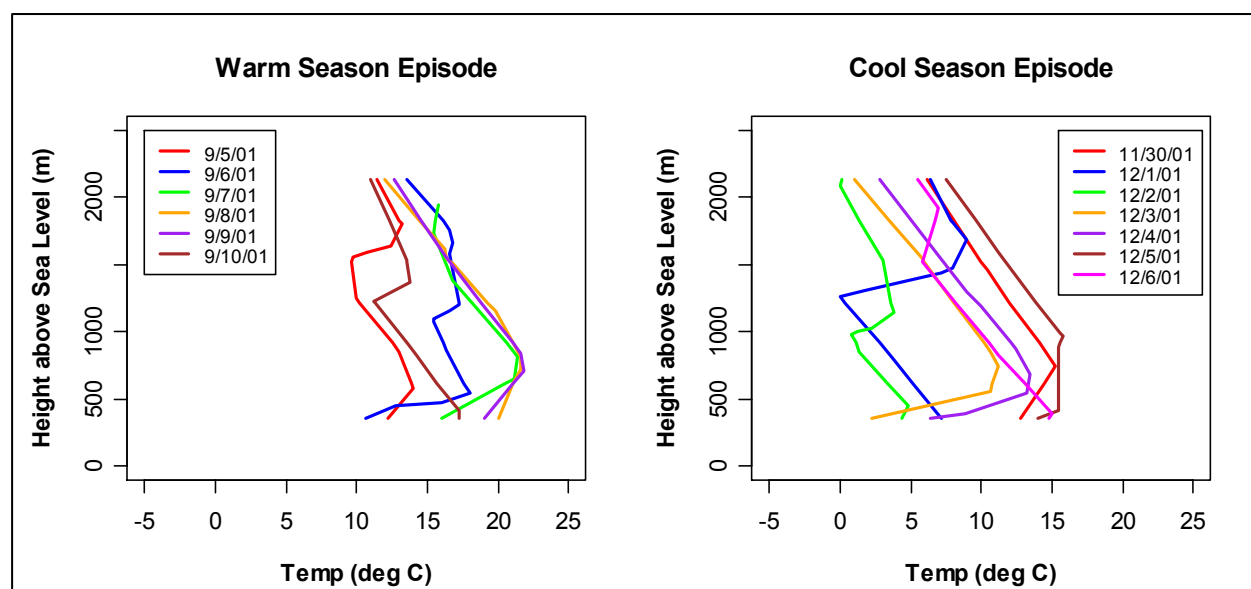


Figure 48. Vertical temperature soundings taken at 7:00 a.m. EST at the Moon Township, PA, National Weather Service station on each day of the September 5-11, 2001, “warm season” PM_{2.5} episode and the November 30-December 6, 2001, “cool season” PM_{2.5} episode.

Surface-based temperature inversions were also evident at 7:00 a.m. on several days during the warm season episode (e.g., 9/6/01, 9/7/01). As shown in Figure 34, these inversions likely helped to elevate concentrations of primary gaseous pollutants such as NO and CO, which exhibited peak concentrations around 7:00 a.m. EST on these days. (On days with weaker inversions at 7:00 a.m., such as 9/8/01 and 9/9/01, CO concentrations peaked earlier during the

overnight period, and NO concentrations did not exhibit appreciable peaks). However, PM_{2.5} concentrations during this warm season episode did not display the inversion-driven behavior characteristic of the November 30-December 6 episode, suggesting that a different mechanism was primarily responsible for the September 2001 episode.

Figure 49 summarizes the composition of PM_{2.5} determined from 24-hr integrated PM_{2.5} samples collected during the warm and cool season episodes. The warm season episode sample was collected from 9:00 a.m. EST on September 8, 2001, through 9:00 a.m. EST on September 9, 2001, and the cool season episode sample was collected from 9:00 a.m. EST on December 5, 2001, through 9:00 a.m. EST on December 6, 2001. To clearly illustrate departures from typical PM_{2.5} composition at Steubenville, Figure 49 shows compositional enrichment factors (CEFs) computed for each PM_{2.5} component during each of the two episode days. The CEF for component *x* on day *i* is defined as:

$$CEF_{x,i} = \frac{w_{x,i}}{w_{x,median}}$$

where $w_{x,i}$ is the weight fraction of component *x* in the PM_{2.5} sample collected on day *i*, and $w_{x,median}$ is the overall median weight fraction of component *x* in 24-hour PM_{2.5} samples collected at Steubenville during SCAMP. As shown in the figure, PM_{2.5} collected during the 24-hr period from the warm season episode was enriched in SO₄²⁻ and NH₄⁺, but contained less NO₃⁻, Cl⁻, carbonaceous material, and trace elements than normal on a percentage basis (valid data were not available for Ba, Co, Ni, Se, and Ti for 9/8/01). This compositional profile suggests that the elevated PM_{2.5} concentrations observed at Steubenville between September 5 and September 11, 2001, were likely driven by a regional episode of secondary ammonium sulfate formation.

The PM_{2.5} compositional profile for the 24-hr period during the cool season episode was largely opposite that observed for the warm season episode. PM_{2.5} collected on 12/5/01 was enriched in organic material, Cl⁻, and a number of trace elements, which are expected to be emitted by primary sources in Steubenville, but depleted in more regional pollutants such as SO₄²⁻, NH₄⁺, and NO₃⁻. In particular, the high CEFs for trace metals such as Fe, Mn, and Zn, which as discussed earlier are particularly enriched at Steubenville relative to the surrounding area, support the interpretation that nocturnal temperature inversions drove this cool season episode in Steubenville by permitting short-term buildups in concentrations of locally-emitted primary pollutants.

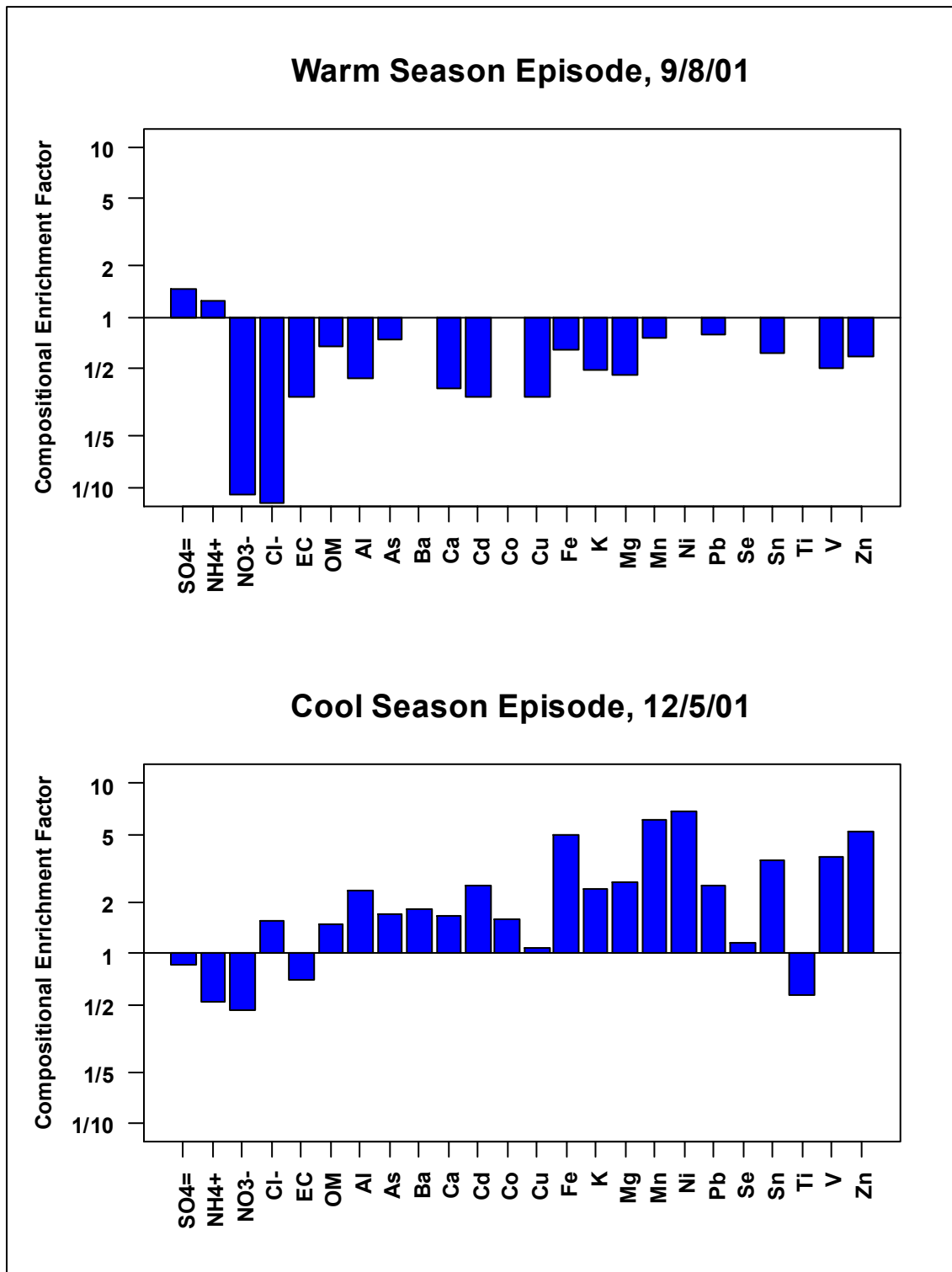


Figure 49. Normalized PM_{2.5} composition (expressed as a “compositional enrichment factor” per the definition given in the text) at Steubenville during the 24-hr period from 9:00 a.m. on 9/8/01 to 9:00 a.m. on 9/9/01 and the 24-hr period from 9:00 a.m. on 12/5/01 to 9:00 a.m. on 12/6/01.

If the September 5-11, 2001, PM_{2.5} episode was a regional episode driven by secondary-sulfate-dominated PM_{2.5}, then it should have been observed on a regional scale across the SCAMP monitoring sites. Similarly, if the November 30-December 6, 2001, PM_{2.5} episode resulted from the nocturnal inversion-driven accumulation of primary PM_{2.5} and co-pollutant emissions, then it should have been more pronounced at the Steubenville site, which was located near a number of major primary emission sources, than at the more remote satellite sites. Figure 50 confirms both of these expectations.

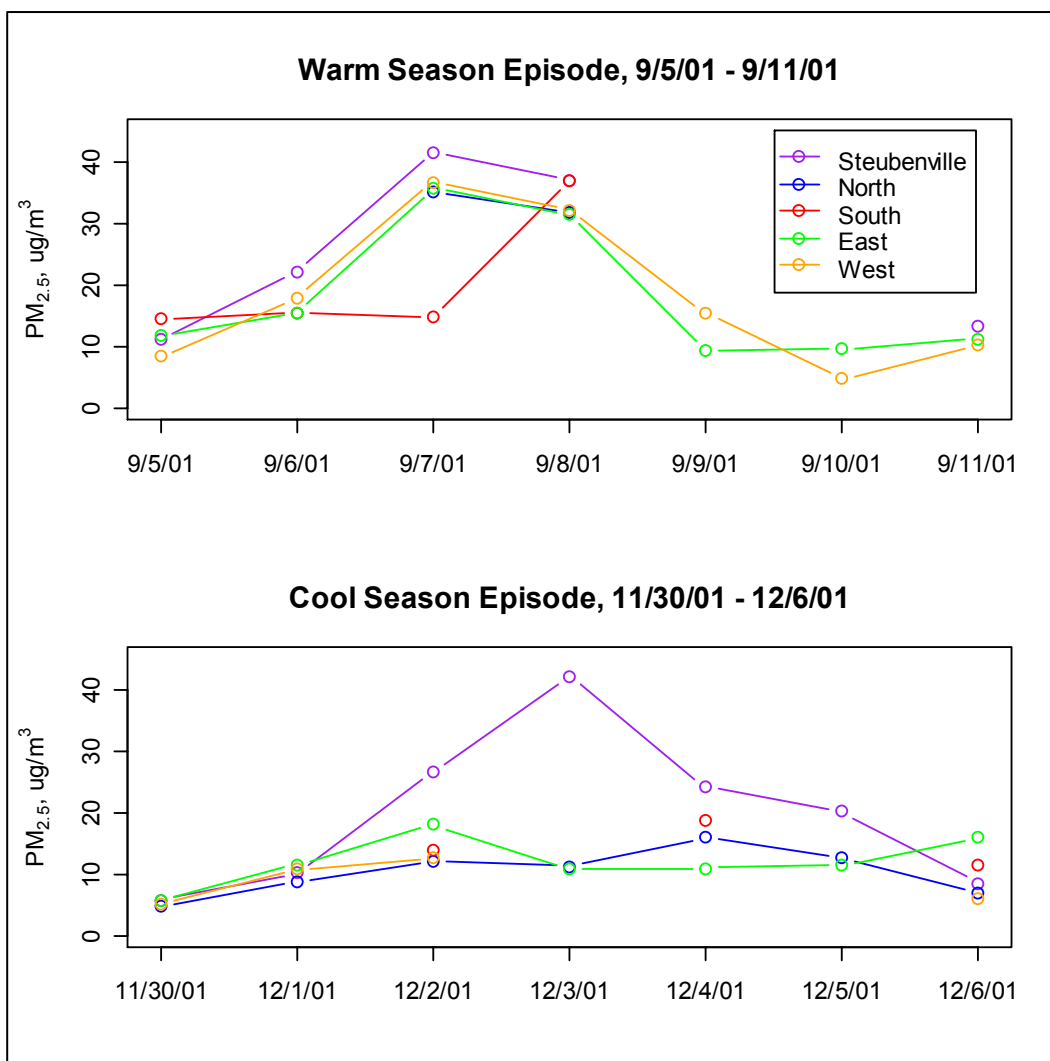


Figure 50. Time series of 24-hr average PM_{2.5} concentrations measured at the five SCAMP monitoring sites from September 5 through September 11, 2001, and from November 30 through December 6, 2001.

The figure, which plots time series of 24-hr PM_{2.5} concentrations (collected from 9:00 a.m. EST on the indicated day to 9:00 a.m. EST on the following day) observed during the warm and cool season episodes, shows that the elevated PM_{2.5} concentrations associated with the warm season episode were generally observed at all of the satellite sites for which data were available (concentrations at the southern site did not increase until 9/8/01), whereas the elevated PM_{2.5} concentrations associated with the cool season episode were unique to Steubenville (data were not available for the southern and western sites on 12/3/01, the date on which the 24-hr PM_{2.5} concentration at Steubenville exhibited a local maximum).

Further support for this interpretation is provided by Figure 51, which shows 24-hr $PM_{2.5}$ and $PM_{10-2.5}$ concentrations observed at the central Steubenville site during each of the episodes. As shown in this figure, during the warm season $PM_{2.5}$ episode that was dominated by secondary ammonium sulfates, concentrations of $PM_{10-2.5}$, which is expected to originate predominantly from primary emission sources in Steubenville, did not show any appreciable increase. Conversely, during the cool season $PM_{2.5}$ episode that was dominated by local emissions in the immediate Steubenville vicinity, concentrations of $PM_{10-2.5}$ tracked concentrations of $PM_{2.5}$ very closely.

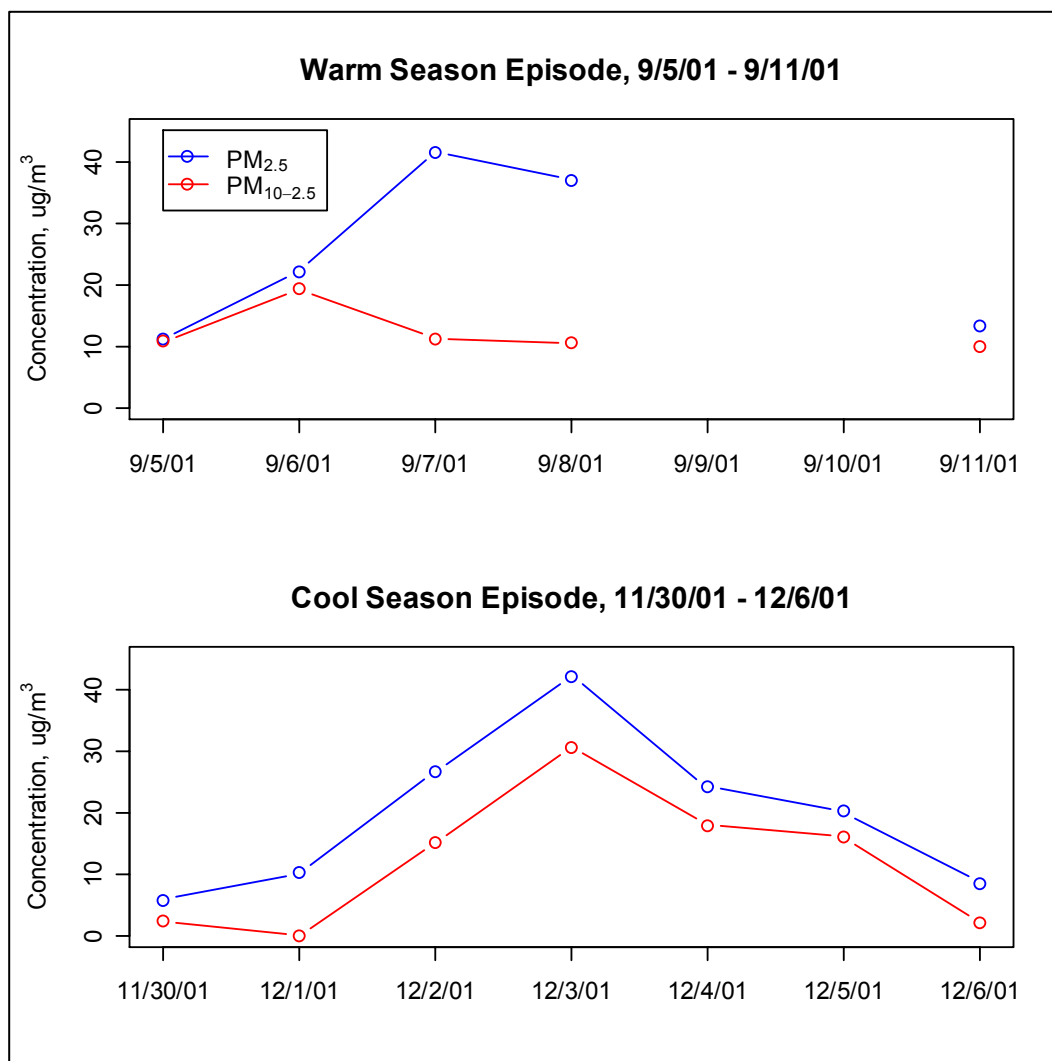


Figure 51. Time series of 24-hr average $PM_{2.5}$ and $PM_{10-2.5}$ concentrations observed at the central Steubenville site from September 5 through September 11, 2001, and from November 30 through December 6, 2001.

In summary, the case studies presented above clearly illustrate two distinctly different types of episode that led to elevated $PM_{2.5}$ concentrations at Steubenville during SCAMP:

1. A regional episode type characterized by a multiple-day period of consistently and moderately elevated concentrations of secondary-sulfate-dominated $PM_{2.5}$, which typically occurred during warmer portions of the year in Steubenville, and
2. A local episode type characterized by shorter overnight periods of more acutely elevated concentrations of $PM_{2.5}$ resulting from the buildup of primary particles enriched in trace

metals and carbonaceous species during nocturnal temperature inversions, which typically occurred during cooler portions of the year in Steubenville.

As mentioned earlier, more examples of these types of episodes can be found in Connell et al. (2005b, 2005c). Further evidence of the distinction between regional daylong episodes of elevated secondary PM_{2.5} concentrations and localized nighttime episodes of elevated primary PM_{2.5} concentrations is provided in Table 36. The first two columns of this table present Spearman correlation coefficients between 24-hr average concentrations of PM_{2.5} chemical components at Steubenville and estimated 24-hr “local” and “background” PM_{2.5} concentrations. “Background” concentrations were estimated as the average PM_{2.5} concentration measured at the northern and western satellite sites on a given day, and “local” concentrations were estimated as the difference between the concentration measured at Steubenville and this “background” concentration. The last two columns of the table present Spearman correlation coefficients between 24-hr concentrations of PM_{2.5} chemical components at Steubenville and 1-hr maximum PM_{2.5} concentrations measured by the TEOM during the “afternoon” (12:00 p.m. EST – 7:00 p.m. EST) and “overnight” (2:00 a.m. EST – 9:00 a.m. EST) portions of the same 24-hr periods. The data in the table show that SO₄²⁻ and NH₄⁺ correlated more strongly with “background” PM_{2.5} concentrations than with “local” PM_{2.5} concentrations and more strongly with “afternoon” maximum concentrations than with “overnight” maximum concentrations, consistent with the first type of episode described above. Conversely, elemental carbon, organic material, and a number of trace elements, including especially those highlighted earlier as being locally enriched at Steubenville (e.g., As, Fe, Mn, Pb, Zn), correlated more strongly with “local” PM_{2.5} concentrations than with “background” concentrations and more strongly with “overnight” maximum PM_{2.5} concentrations than with “afternoon” maximum concentrations, consistent with the second type of episode.

Both of the PM_{2.5} episode types identified during SCAMP must be considered when developing an implementation plan to reduce concentrations of PM_{2.5} in Steubenville, as both can lead to elevated 24-hr PM_{2.5} concentrations (e.g., concentrations exceeding the recently proposed 24-hr standard of 35 µg/m³), which contribute to Steubenville’s high annual average PM_{2.5} concentration. As shown above, these episode types result from very different mechanisms; therefore, reducing their occurrences will require the application of unique strategies. The distinct differences between the behavior and composition of PM_{2.5} during the episode types also reinforces the need for further research to clarify several uncertainties related to possible effects of fine particulate matter on human health. These include:

1. Whether health effects are more likely to result from prolonged (e.g., multiple-day) exposure to chronically elevated PM_{2.5} concentrations, as were observed during the warm season episode, or from shorter-term (e.g., < ½ day) exposure to more acutely elevated PM_{2.5} concentrations, as were observed during the cool season episode;
2. Whether the composition of PM_{2.5} influences its effect on human health (e.g., whether particles enriched in trace metals and carbonaceous species, which were prevalent during the cool season episode, are more toxic than the secondary-sulfate-dominated particles that were prevalent during the warm season episode, or whether the mere existence of particles, regardless of composition, triggers the observed health effects); and
3. Whether primary gases such as CO, NO₂, and SO₂, which exhibited elevated concentrations and correlated strongly with PM_{2.5} during the cool season episode, account for some or all of the health effects commonly attributed to PM_{2.5} or, as suggested by several studies (Jakab et al., 1996; Kleinman et al., 2000; Schlesinger, 1995), exhibit synergistic toxicological interactions with PM_{2.5}.

Table 36. Spearman correlation coefficients between 24-hr PM_{2.5} chemical component (i.e., ion, carbon, and acid-digestible element) concentrations at the central Steubenville site and measures of the intraday timing of PM_{2.5} episodes and the impact of local versus regional sources on PM_{2.5} concentrations in Steubenville.

PM _{2.5} Component	Local PM _{2.5} ^a	Background PM _{2.5} ^b	Afternoon Maximum PM _{2.5} ^c	Overnight Maximum PM _{2.5} ^d
SO ₄ ²⁻	0.68	0.84	0.77	0.61
NH ₄ ⁺	0.61	0.87	0.67	0.49
NO ₃ ⁻	0.04	0.17	-0.08	-0.10
Cl ⁻	0.46	0.10	0.06	0.53
EC	0.70	0.35	0.24	0.73
OM	0.67	0.55	0.41	0.71
Al	0.65	0.39	0.39	0.50
As	0.71	0.38	0.26	0.70
Ba	0.63	0.41	0.29	0.57
Ca	0.55	0.21	0.28	0.51
Cd	0.58	0.27	0.17	0.33
Co	0.45	0.38	0.27	0.45
Cu	0.53	0.19	0.08	0.43
Fe	0.75	0.33	0.24	0.69
K	0.62	0.46	0.33	0.55
Mg	0.56	0.25	0.23	0.59
Mn	0.74	0.28	0.22	0.63
Ni	0.48	0.35	0.39	0.36
Pb	0.71	0.29	0.21	0.61
Se	0.56	0.63	0.50	0.45
Sn	0.73	0.22	0.27	0.58
Ti	0.33	0.14	0.22	0.43
V	0.50	0.23	0.37	0.49
Zn	0.80	0.35	0.26	0.66

^aDifference between 24-hr concentration measured at the central Steubenville site and the average of the 24-hr concentrations measured at the northern and western satellite sites; ^bAverage of the 24-hr concentrations measured at the northern and western satellite sites; ^cMaximum 1-hr concentration measured between 12:00 p.m. and 7:00 p.m. EST; ^dMaximum 1-hr concentration measured between 2:00 a.m. and 9:00 a.m. EST

The diurnal behavior of PM_{2.5} during the different episode types also suggests a need for further research using time-resolved measurements to elucidate the effect of the intraday timing of PM_{2.5} episodes on actual human exposure to PM_{2.5}.

4.2.4.3 Binary Recursive Partitioning of Hourly PM_{2.5} Data

To help to validate the conclusions drawn from the correlation analyses and case studies presented above, binary recursive partitioning (rpart) was applied to the 14,682 valid hourly PM_{2.5} observations made at the central Steubenville site during SCAMP. Binary recursive partitioning (Breiman et al., 1984; Therneau and Atkinson, 1997), which has been used to study the influence of meteorological conditions on daily PM_{2.5} concentrations (Kenski, 2003) but has not been discussed extensively in the PM_{2.5} literature, classifies a categorical dependent variable via a series of binary splits based upon “yes/no” questions about predictor (splitting)

variables (e.g., is $\text{NO} < 40 \text{ ppm}$?). As a result, a classification tree is developed, which identifies factors that tend to be associated with the various categories of the dependent variable and which may also be used as a predictive tool. The technique is nonparametric and handles both continuous and categorical splitting variables. For each node of the classification tree, the rpart algorithm considers all possible binary divisions of the data based upon all values of all splitting variables specified in the analysis and chooses the best split, which minimizes the impurity or heterogeneity (measured using the Gini diversity index) in the resulting nodes while causing the greatest improvement in the number of correct classifications of the dependent variable. Missing values in the primary splitting variable are handled via secondary splitting rules based upon surrogate variables that show strong agreement with the primary variable. Trees are validated using cross-validation; to avoid over- or under-fitting the data, the tree size is selected that minimizes the cross-validation error. Generally, the simplest tree having a cross-validation error within one standard error of the minimum is selected as the “optimal” tree.

In order to use rpart to classify hourly $\text{PM}_{2.5}$ data collected during SCAMP, a categorical variable describing hourly $\text{PM}_{2.5}$ concentrations was defined. Class A includes all $\text{PM}_{2.5}$ concentrations less than $10.0 \mu\text{g}/\text{m}^3$; class B includes concentrations greater than or equal to $10.0 \mu\text{g}/\text{m}^3$ and less than $30.0 \mu\text{g}/\text{m}^3$; class C includes concentrations greater than or equal to $30.0 \mu\text{g}/\text{m}^3$ and less than $50.0 \mu\text{g}/\text{m}^3$, and class D includes all concentrations of $50.0 \mu\text{g}/\text{m}^3$ or greater. All gas and weather variables specified in Tables 22 and 25 (except for NO_x , because NO and NO_2 are included) were specified as predictor variables. In addition, to account for the possibility of lagged relationships, 1, 3, 12, and 24-hr lags of each of these variables were included, as were 1, 12, and 24-hr changes in the value of each variable. Moreover, categorical variables indicating astronomical season, time of day (in 3-hr increments), and time of week (weekday versus weekend) were specified. Because of the large number of observations, splits were only attempted for nodes having at least 75 observations, and terminal nodes having less than 25 observations were not allowed. Up to five surrogate rules were allowed for each split, and 60 cross-validations were used. According to the criterion discussed above, the “optimal” tree had approximately 76 terminal nodes; however, for ease of presentation, the tree was pruned to 37 terminal nodes. Pruning was done in the standard way based upon the complexity parameter (Therneau and Atkinson, 1997).

This tree is shown in Figure 52. For each split, cases satisfying the primary splitting criterion (shown) or a surrogate criterion are sent to the left; all other cases are sent to the right. Table 37 shows the number of observations falling in each terminal node, as well as the percent composition of each node on the basis of the $\text{PM}_{2.5}$ classes defined above. Although not perfectly homogenous, the homogeneity of each terminal node with respect to its predicted class is greater than the homogeneity of the root node. The model correctly classified 11460, or 78.05 %, of the 1-hr $\text{PM}_{2.5}$ concentrations observed during SCAMP, compared with 55.62 % correct classification in the root node (assuming its predicted class is A). A clearer depiction of the performance of the model is provided in Figure 53, which shows the distribution of actual $\text{PM}_{2.5}$ concentrations ($\mu\text{g}/\text{m}^3$) falling in each terminal node. Nodes are sorted by increasing median concentration.

As shown in Figure 53, only one terminal node (Node 31) had a median concentration and a majority of its observations in class D ($\text{PM}_{2.5}$ concentration $\geq 50 \mu\text{g}/\text{m}^3$). Cases falling in this node were characterized by a CO concentration greater than 2.67 ppm and a NO concentration greater than 35.8 ppb. This is consistent with the previous observation that above-average

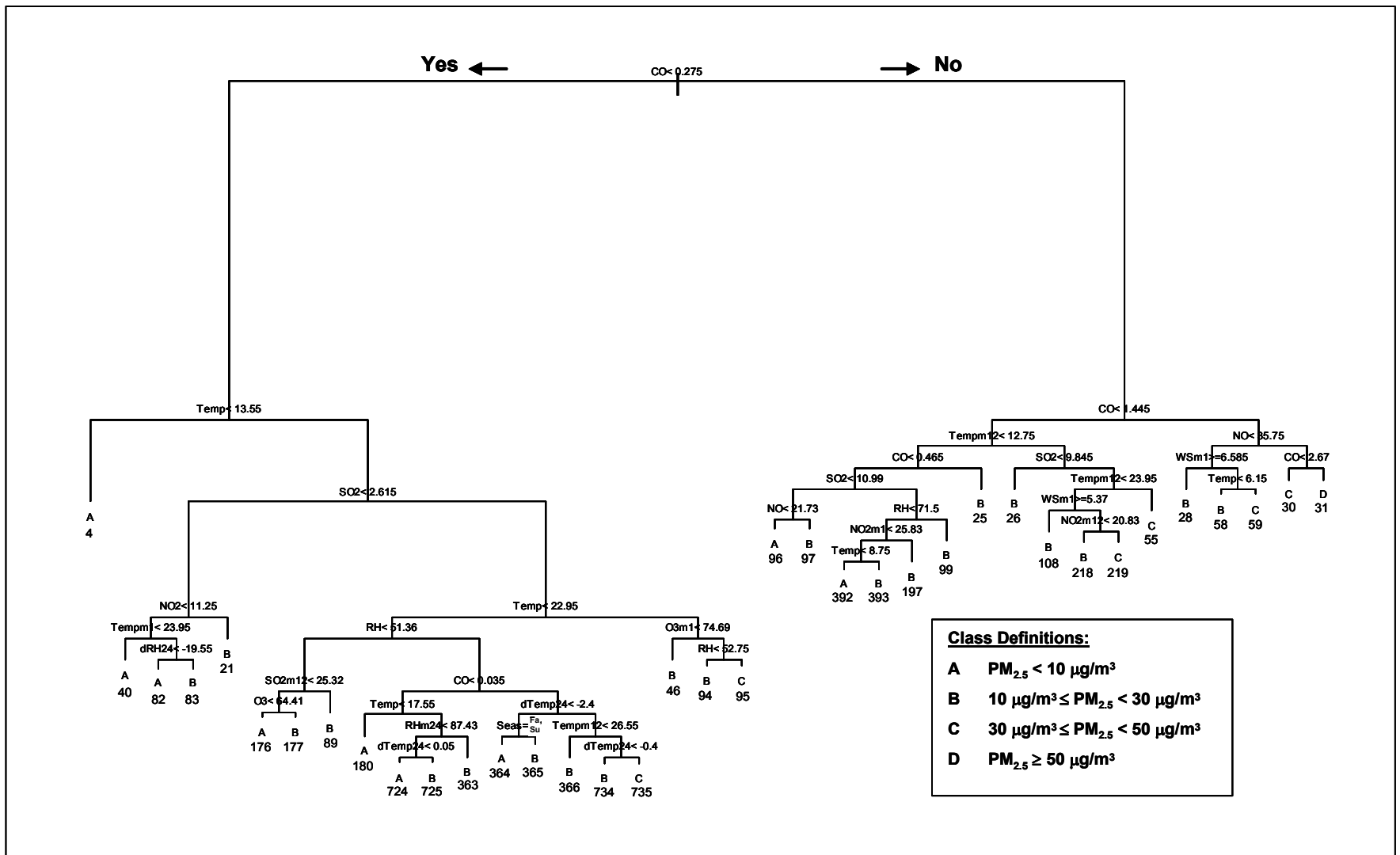


Figure 52. Binary recursive partitioning classification tree constructed using all valid 1-hr average $PM_{2.5}$ concentrations measured at Steubenville during SCAMP. For each splitting node, observations satisfying the splitting criterion are sent to the left; all other observations are sent to the right. Each terminal node is labeled with its predicted class as well as an identification number. Units are ppm for CO, ppb for all other gases, °C for temperature (Temp), % for relative humidity (RH), mph for wind speed (WS), kW/m^2 for solar radiation (Rad), and inches of Hg for barometric pressure (BP). For a splitting variable V, V_{mN} represents the 1-hr average value of that variable observed N hours earlier; dVN represents the N-hr change in that variable's value. Seas is season; Fa is Fall, and Su is Summer.

Table 37. Summary of terminal nodes of the binary recursive partitioning tree shown in Figure 52.

Node	Predicted Class ^a	N	Distribution of Observations by Class ^a (%)			
			A	B	C	D
Root	A	14682	55.62	34.72	7.87	1.79
4	A	6121	88.09	11.24	0.44	0.23
21	B	212	38.68	58.02	3.30	0.00
25	B	850	16.35	74.12	8.59	0.94
26	B	585	7.18	76.07	15.73	1.03
28	B	28	3.57	92.86	3.57	0.00
30	C	180	0.00	14.44	66.11	19.44
31	D	109	0.00	3.67	26.61	69.72
40	A	774	82.82	17.18	0.00	0.00
46	B	893	10.97	64.28	22.51	2.24
55	C	134	0.00	15.67	66.42	17.91
58	B	32	0.00	78.13	15.63	6.25
59	C	155	3.23	30.97	49.68	16.13
82	A	36	94.44	5.56	0.00	0.00
83	B	193	35.23	61.14	3.63	0.00
89	B	100	36.00	62.00	1.00	1.00
94	B	71	0.00	66.20	33.80	0.00
95	C	108	0.00	13.89	78.70	7.41
96	A	412	68.93	30.58	0.24	0.24
97	B	77	23.38	74.03	1.30	1.30
99	B	131	11.45	86.26	2.29	0.00
108	B	126	8.73	80.95	6.35	3.97
176	A	426	79.11	20.42	0.47	0.00
177	B	27	14.81	85.19	0.00	0.00
180	A	230	80.43	19.57	0.00	0.00
197	B	101	29.70	69.31	0.99	0.00
218	B	231	2.16	65.80	26.84	5.19
219	C	147	0.68	31.29	59.18	8.84
363	B	164	20.73	74.39	4.88	0.00
364	A	155	65.81	29.68	4.52	0.00
365	B	83	31.33	67.47	1.20	0.00
366	B	1227	24.69	67.40	7.17	0.73
392	A	104	74.04	25.96	0.00	0.00
393	B	69	28.99	69.57	1.45	0.00
724	A	142	70.42	28.17	1.41	0.00
725	B	151	43.05	56.95	0.00	0.00
734	B	35	22.86	71.43	5.71	0.00
735	C	63	4.76	20.63	69.84	4.76

^aClass definitions: A is $PM_{2.5} < 10 \mu g/m^3$; B is $10 \mu g/m^3 \leq PM_{2.5} < 30 \mu g/m^3$; C is $30 \mu g/m^3 \leq PM_{2.5} < 50 \mu g/m^3$; D is $PM_{2.5} \geq 50 \mu g/m^3$.

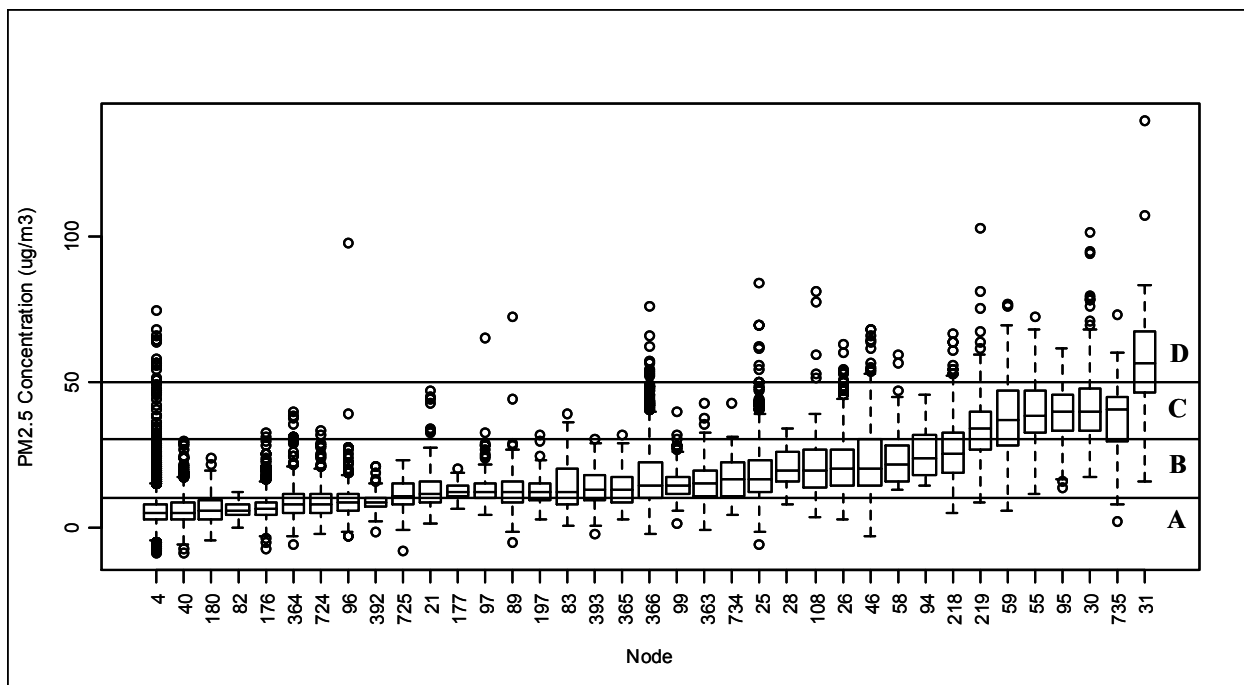


Figure 53. Boxplot showing the distribution of actual $PM_{2.5}$ concentrations ($\mu\text{g}/\text{m}^3$) falling in each terminal node of the classification tree presented in Figure 42. Nodes are sorted by increasing median concentration. Regions corresponding to classes A through D are labeled.

$PM_{2.5}$ concentrations at Steubenville were frequently accompanied by elevated concentrations of primary pollutant gases such as CO , NO_x , and SO_2 , especially during nocturnal temperature inversions. Moreover, the choice of CO and NO , rather than SO_2 , as splitting variables is consistent with the observation that $PM_{2.5}$, CO , and NO_x exhibited similar overall average diurnal profiles, while SO_2 displayed a somewhat different profile.

Six nodes had median concentrations in class C. Conditions associated with these nodes also generally reflected scenarios observed in the case studies discussed previously. For instance, Nodes 59 and 219 both suggest that $PM_{2.5}$ concentrations between 30 and $49.9 \mu\text{g}/\text{m}^3$ often occurred 1 hr after a period of low wind speeds (less than 2.95 or 2.40 m/s, respectively). Nodes 55 and 219 both suggest a positive association between 1-hr $PM_{2.5}$ concentrations and temperatures observed 12 hr earlier, likely reflecting the opposing diurnal patterns that were exhibited by $PM_{2.5}$ and temperature during some episodes, as well as the tendency of $PM_{2.5}$ episodes to occur during periods of locally warm temperatures (Connell et al., 2005b). These nodes also indicate a positive association between $PM_{2.5}$ and SO_2 concentrations. Node 95 shows that moderately elevated $PM_{2.5}$ concentrations (i.e., between 30 and $49.9 \mu\text{g}/\text{m}^3$) at Steubenville tended to occur when the concentration of CO was less than 0.28 ppm, but the concentration of O_3 1 hr earlier was greater than 74.7 ppb, the temperature was greater than $23.0 \text{ }^\circ\text{C}$, and the relative humidity was greater than 52.75%. These conditions are likely indicative of summertime episodes of secondary $PM_{2.5}$ formation like the September 5-11, 2001, episode discussed in the preceding section. The emergence of relative humidity as an important predictor of $PM_{2.5}$ concentration when it was included with other independent variables in binary recursive partitioning is noteworthy, as this variable showed relatively weak associations with hourly $PM_{2.5}$ concentrations in overall and seasonal bivariate correlation analyses ($r_s \leq 0.23$). Connell et al. (2005b) similarly reported that whereas daily average relative humidity was not significantly associated with daily average $PM_{2.5}$ concentration in simple linear regression analyses performed using the SCAMP data, relative humidity was a

statistically significant, positive predictor of PM_{2.5} when included with NO_x, CO, SO₂, O₃, temperature, and wind speed in a multiple regression time series model.

Node 4 had the lowest median PM_{2.5} concentration and the greatest number of observations of all terminal nodes. It included 14 class-D and 27 class-C PM_{2.5} concentrations; however, these represented less than 1% of the total observations in the node. Observations in this node occurred during hours when the concentration of CO was less than 0.28 ppm and the temperature was less than 13.6 °C. Hence, the node probably largely represents low background PM_{2.5} concentrations experienced during cool portions of the year. The eight other class-A nodes also tended to occur when temperatures or gaseous co-pollutant concentrations were below various threshold values, again suggesting the strong association between PM_{2.5} and these potential epidemiologic confounders.

Thus, the results presented here demonstrate the usefulness of binary recursive partitioning for systematically analyzing large quantities of PM_{2.5} air monitoring data and also affirm the conclusions drawn from the correlation analyses and examination of PM_{2.5} episodes presented earlier.

4.2.5 Source Apportionment of PM_{2.5} at Steubenville

The case studies and correlation analyses presented above provide insights into factors that affect PM_{2.5} concentrations at Steubenville. However, to more systematically identify sources of PM_{2.5} that impacted the central Steubenville monitoring site during SCAMP, receptor modeling was applied to PM_{2.5} mass and speciation data collected at that site during the program. The general receptor modeling problem is given by:

$$x_{ij} = \sum_{k=1}^p g_{ik} f_{kj} + e_{ij} ,$$

where x_{ij} is the ambient mass concentration of the j th chemical species from the i th PM_{2.5} sample collected at the receptor site; g_{ik} is the PM_{2.5} mass concentration from the k th source contributing to the i th sample; f_{kj} is the mass fraction of the j th chemical species in PM_{2.5} from the k th source; e_{ij} is the residual associated with x_{ij} , and p is the total number of sources contributing to PM_{2.5} at the receptor site. If the compositional profiles of the various sources impacting the receptor site are known (i.e., if the matrix of source compositions, F , can be specified), then the receptor modeling problem can be solved for the matrix of source contributions, G , via the commonly used Chemical Mass Balance (CMB) technique (Hidy and Friedlander, 1971). In many cases, though, source compositions are not known; hence, there are an infinite number of possible solutions to the receptor modeling equation corresponding to different rotations of the F and G matrices. Such was the case in SCAMP. Several different factor analytic techniques have been proposed for use in resolving physically reasonable combinations of the matrix of source contributions, G , and the matrix of source compositions, F , for any given set of speciated PM_{2.5} data, X , measured at a receptor site. Two of the most commonly used techniques, Positive Matrix Factorization (PMF) and Unmix, were applied to the SCAMP central site data. Results are described in the following two subsections.

4.2.5.1 Positive Matrix Factorization Results

Positive Matrix Factorization (Paatero, 1997) solves the general receptor modeling problem given above for the F and G matrices by imposing non-negativity constraints on these matrices

and employing an iterative least squares procedure in which observations are weighted by the inverse of their estimated uncertainties. PMF was applied to PM_{2.5} mass, ion, carbon, and acid-digestible elemental data collected at the SCAMP central Steubenville site. Only the 115 days for which data were available for total mass, all four ions (NH₄⁺, SO₄²⁻, NO₃⁻, and Cl⁻), both elemental and organic carbon, and at least 10 out of the 18 routinely determined acid-digestible elements were included in the model runs. Details of the application of PMF to these data are provided by Connell et al. (2006); a copy of this manuscript is included as Appendix D to this report. As discussed in the manuscript, a final 7-source PMF model for the Steubenville site was selected based on an evaluation of the model goodness-of-fit and the physical reasonableness of the source profiles obtained from several candidate PMF solutions. Potential source contribution function (PSCF) modeling was applied to the source contribution time series resolved by PMF to help identify the probable geographic locations of each of the seven PMF sources; these PSCF results are also presented in Appendix D.

Table 38 shows the source profiles resolved for the seven PMF source factors, and Figure 54 presents boxplots showing the seasonal distributions of daily source contributions estimated for each source factor. Table 39 summarizes probable identities of the seven source factors based on the interpretation provided in Appendix D. The PMF and PSCF results suggest that secondary sulfates likely originating from coal-fired power plants located to the west and southwest of Steubenville (Factor 1) are the largest contributor to PM_{2.5} in Steubenville, accounting for about 7.4 μg/m³, or about 42%, of the total PM_{2.5} mass. Correlation analyses similar to those presented in Table 36 for individual PM_{2.5} components indicate that elevated concentrations of this ammonium sulfate-dominated source typically were observed on a regional scale and tended to occur during warm summertime afternoons, when photochemical activity would have been maximal. Local sources in the immediate Steubenville vicinity, represented by Factors 2 and 3, collectively accounted for about 5.3 μg/m³, or 30%, of the total PM_{2.5} mass. These sources include motor vehicles and other local sources of carbonaceous aerosols (Factor 2), which contributed ~20% of the total PM_{2.5} mass, as well as iron and steel production facilities (Factor 3), which contributed ~10% of the total PM_{2.5} mass. The PMF-estimated local source contribution of 5.3 μg/m³ is similar to the local source contribution of 4.6 μg/m³ reported above on the basis of pairwise comparisons of PM_{2.5} concentrations measured at the Steubenville site and PM_{2.5} concentrations measured at the northern and western satellite sites. The estimated contributions of Factors 2 and 3 to total PM_{2.5} mass were more strongly correlated with overnight maximum PM_{2.5} concentrations than with afternoon maximum concentrations and with estimated local source PM_{2.5} mass concentrations than with estimated background concentrations, consistent with the previously discussed effects of nocturnal temperature inversions and reduced nighttime mixing heights on locally-emitted pollutant concentrations. Both sources also exhibited moderate-to-strong correlations with concentrations of CO, NO, and NO₂ consistent with the cool-season case study presented above.

Other sources of PM_{2.5} resolved by PMF include a nitrate source (Factor 4), which accounted for about 15% of the total PM_{2.5} mass and was associated with source regions in the vicinity of urban areas located to the west of Steubenville, a crustal source (Factor 5, 6% of total mass), a mixed nonferrous metals and industrial source (Factor 6, 3% of total mass), for which the source region estimated by PSCF was consistent with the locations of metal smelting and processing plants and industrial facilities that reported air emissions of Pb, Cd, and other nonferrous trace metals to the EPA's Toxics Release Inventory (TRI), and a primary coal combustion source (Factor 7, 3% of total mass), for which PSCF identified a source region comprising portions of

the Monongahela and Ohio River Valleys, where the region's major coal-fired power plants are located.

Table 38. Source profile (ng/m³) for the seven-source PMF model run using PM_{2.5} total mass, ion, carbon, and acid-digestible element data collected at the central Steubenville site during SCAMP. Estimated concentrations with magnitudes at least twice as great as their standard deviations are highlighted in bold.

	Factor 1	Factor 2	Factor 3	Factor 4	Factor 5	Factor 6	Factor 7
PM _{2.5}	7400	3500	1800	2700	1000	600	600
NH ₄ ⁺	1500	0	100	400	0	100	100
NO ₃ ⁻	0	50	0	1100	0	0	0
SO ₄ ²⁻	3800	600	300	300	0	100	100
Cl ⁻	0	0	0	0	0	0	0
EC	0	630	0	0	0	70	50
OM	400	2800	400	400	300	0	0
Al	1.5	0.7	13.5	14.0	44.5	5.5	9.6
As	0.09	0.35	0.43	0.06	0.14	0.30	0.12
Ba	0	1.1	0	0.2	0.1	0.0	0.6
Ca	2	14	18	25	53	5	7
Cd	0.01	0.06	0.01	0	0	0.2	0.02
Co	0.008	0.023	0.009	0.002	0	0.001	0.007
Cu	0.2	0.7	0.3	0.4	0.3	0.7	0
Fe	10	48	184	2	11	0	19
K	0	5	23	15	45	0	2
Mg	0	5	17	4	13	0	0
Mn	0	3.3	8.6	0	0.2	0.1	1.4
Ni	0.1	0.1	0.1	0	0	0	0
Pb	0	1.4	4	0.8	1	7.3	0
Se	0.11	0	0.04	0.2	0	0	2.96
Sn	0.08	0.15	0.44	0.11	0.02	0.39	0.12
Ti	1.6	0	0	1	4.2	2	0
V	0.05	0.17	0	0	0.22	0.33	0.11
Zn	7.2	0	47.3	7.5	2.2	18.1	3.8

Table 39. Summary of results of the seven-source PMF model run using PM_{2.5} total mass, ion, carbon, and acid-digestible element data collected at the central Steubenville site during SCAMP.

Factor #	Source	Enriched Species	Contribution to Total PM _{2.5} Mass
1	Secondary Sulfates	SO ₄ ²⁻ , NH ₄ ⁺	42%
2	Vehicular / Local Carbon	EC, OM, Ba	20%
3	Iron and Steel	Fe, Mg, Mn, Sn, Zn	10%
4	Nitrates	NO ₃ ⁻	15%
5	Crustal	Al, Ca, K, Mg, Ti, V	6%
6	Nonferrous Metals / Industrial	Cd, Pb, V, Sn, Cu, Ti, Zn, As	3%
7	Primary Coal Combustion	Se, Ba, Co, V, Al, Mn	3%

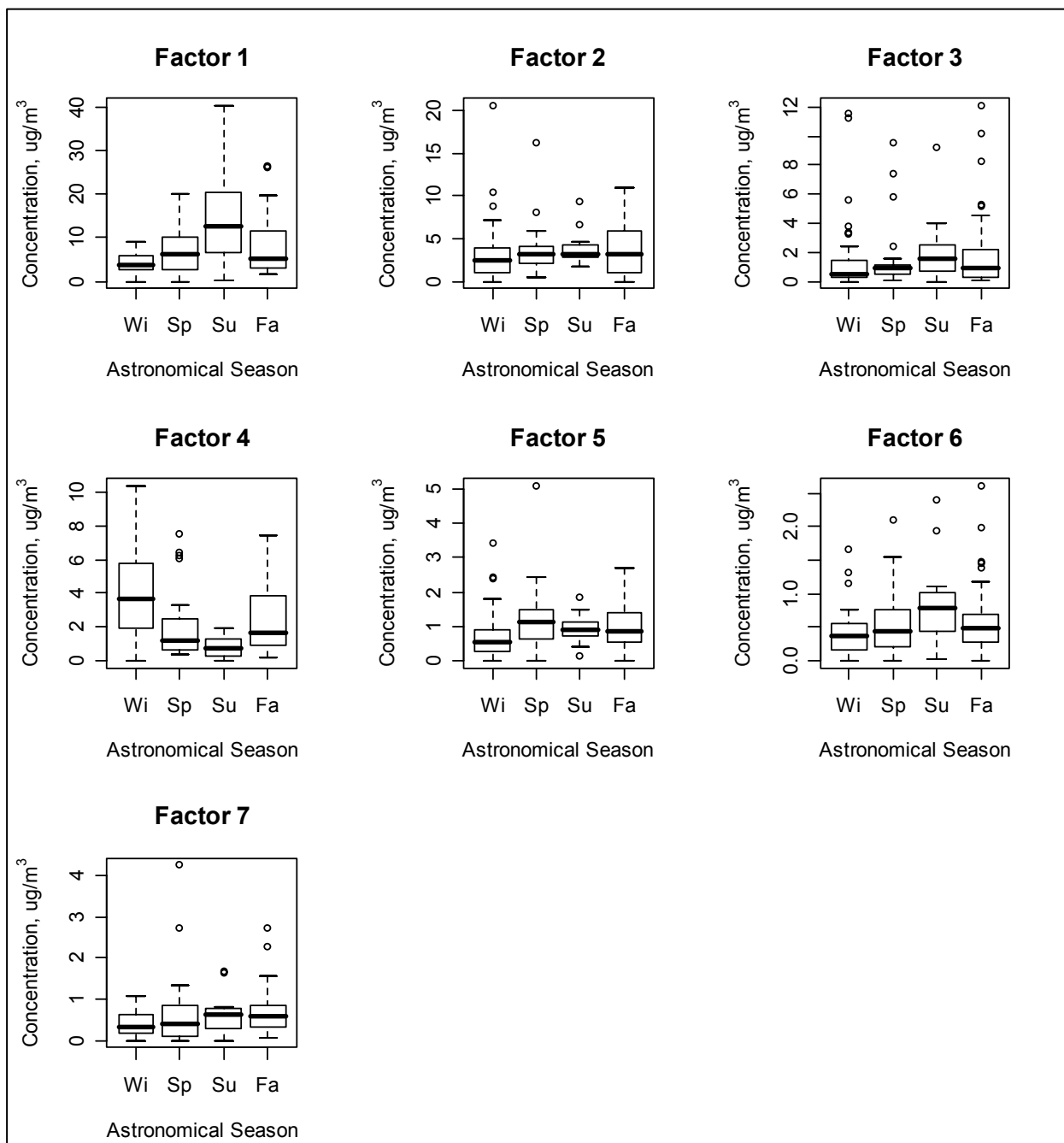


Figure 54. Boxplots showing the distributions of estimated source contributions ($\mu\text{g}/\text{m}^3$) by astronomical season for each of the seven sources resolved by PMF for the central Steubenville site.

4.2.5.2 Unmix Results

Source apportionment was also performed on $\text{PM}_{2.5}$ mass and speciation data from the central Steubenville site using the EPA Unmix 2.3 receptor model. Like PMF, the Unmix model solves the general receptor modeling problem given above for the F and G matrices; however, unlike PMF, it does this by first estimating the number of sources, k , that can be resolved from the

dataset being modeled, based on signal-to-noise ratios, and then employing an algorithm to detect “edges” (i.e., constant ratios among species concentrations) in $(k-1)$ -dimensional space. These edges are used to determine the source compositions and contributions. Bootstrapping is then employed to estimate the uncertainties in the results (Lewis et al., 2003). Unmix does not utilize user-specified measurement uncertainties as PMF does, and it often fails to resolve as many sources as PMF. Moreover, feasible solutions are typically obtained only when a subset of the measured species (rather than all of these species) are included in the model. Hence, the results are less detailed than those produced by PMF. Nevertheless, the application of Unmix to the SCAMP central site $PM_{2.5}$ speciation data provided a means for independently validating the PMF source apportionment results presented above.

A number of Unmix runs were performed (using the software’s “Unmix Overnight” utility) to systematically explore the effects of including various ionic, carbonaceous, and acid-digestible elemental components of $PM_{2.5}$. Several satisfactory five-source solutions were obtained for which diagnostic indicators met or exceeded specifications provided with the software (i.e., minimum $R^2 \geq 0.8$; minimum signal-to-noise ratio ≥ 2). Strength, a third indicator that measures confidence in the solution, is recommended to be greater than 3; however, as discussed by researchers from Carnegie Mellon University (2005), strengths less than 3 may be acceptable for some datasets. The Carnegie Mellon researchers reported a strength of 1.41 for a six-source Unmix model of speciated $PM_{2.5}$ data in Pittsburgh, PA. Strengths for the five-source SCAMP model runs ranged from 1.45 to 3.65. To further aid in the selection of the best five-source Unmix solutions, the satisfactory models obtained from the Unmix Overnight process were re-run with the inclusion of $PM_{2.5}$ speciation data from EPA sampling intensives that were conducted at the Steubenville site during June-August 2001 and January 2002. (These data have otherwise been excluded from the analyses presented in this paper). Based on an assessment of the stability of the models when these additional days were included, three models were determined to be superior to the others. (No further distinction among the merits of these three could be made, as they included different combinations of $PM_{2.5}$ chemical species). These three models are now discussed in greater detail (for the solutions that did not include data from the EPA sampling intensives).

Table 40 summarizes model specifications and diagnostics for each of the three best five-source Unmix solutions. Table 41 presents Unmix-derived source profiles (in ng/m^3) for each of these solutions. As shown in Figure 55, which compares 24-hr average $PM_{2.5}$ concentrations predicted by the Unmix models with actual 24-hr average $PM_{2.5}$ concentrations measured at Steubenville, the models were generally successful in predicting total $PM_{2.5}$ mass concentrations at Steubenville. R^2 values resulting from simple linear regression of predicted versus observed values ranged from 0.96 to 0.98.

Table 40. Model specifications and diagnostics for the three best five-source Unmix model runs using $PM_{2.5}$ total mass, ion, carbon, and acid-digestible element data collected at the central Steubenville site during SCAMP.

	Solution # 1	Solution # 2	Solution # 3
Number of Days	108	104	113
Species	$PM_{2.5}$, NH_4^+ , NO_3^- , SO_4^{2-} , OM, Al, K, Pb, V	$PM_{2.5}$, NH_4^+ , NO_3^- , SO_4^{2-} , OM, Fe, K, Mn, Se, Zn	$PM_{2.5}$, NH_4^+ , SO_4^{2-} , Al, Cd, Fe, Mn, V, Zn
Minimum R^2	0.86	0.88	0.94
Minimum Signal / Noise	2.23	2.15	2.01
Strength	2.55	2.58	3.61

Table 41. Source profiles (ng/m³) for the three best five-source Unmix model runs using PM_{2.5} total mass, ion, carbon, and acid-digestible element data collected at the central Steubenville site during SCAMP. Estimated concentrations with magnitudes at least twice as great as their uncertainties are highlighted in bold.

UNMIX Solution # 1					
	Factor 1	Factor 2	Factor 3	Factor 4	Factor 5
PM_{2.5}	1100	3600	1200	4000	7700
NH₄⁺	100	600	100	200	1200
NO₃⁻	-10	1260	-20	40	-70
SO₄²⁻	200	600	400	500	3500
OM	500	400	200	2100	1200
Al	9.1	4.9	3.4	63.5	17.1
K	6	9	9	51	16
Pb	1.0	0.7	10.7	1.3	2.2
V	1.07	-0.01	0.06	0.20	0.21
UNMIX Solution # 2					
	Factor 1	Factor 2	Factor 3	Factor 4	Factor 5
PM_{2.5}	2900	2200	3500	1000	8500
NH₄⁺	500	200	100	100	1400
NO₃⁻	1130	30	110	-70	20
SO₄²⁻	400	400	300	400	4000
OM	300	600	2200	300	1200
Fe	-4	217	25	31	41
K	8	23	44	4	15
Mn	0.3	10.5	3.8	1.1	0.3
Se	0.22	0.40	-0.03	2.42	0.53
Zn	5.2	59.4	6.2	3.8	12.6
UNMIX Solution # 3					
	Factor 1	Factor 2	Factor 3	Factor 4	Factor 5
PM_{2.5}	1500	1900	4800	1300	7400
NH₄⁺	100	300	600	0	1200
SO₄²⁻	400	700	1100	100	3100
Al	10.7	3.9	58	10.1	15.8
Cd	0.06	0.36	-0.03	0.08	0.01
Fe	27	19	-2	196	32
Mn	2.0	0.0	2.4	9.6	0.4
V	1.07	0.04	0.20	0.16	0.03
Zn	8.4	5.3	11.9	49.2	9.2

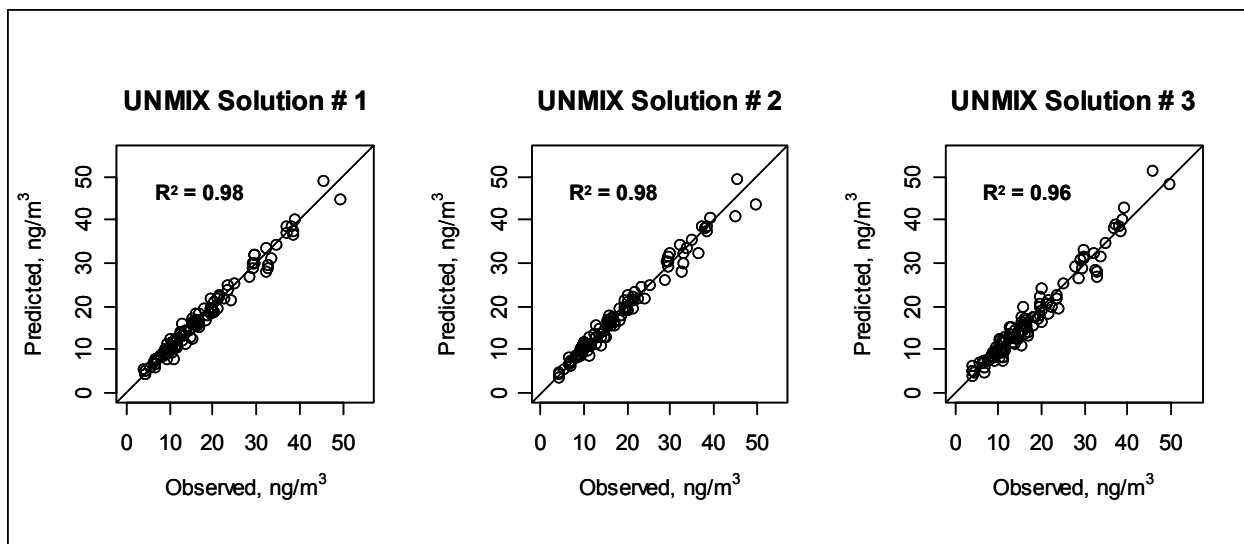


Figure 55. PM_{2.5} concentrations at Steubenville predicted by each of the three best five-source Unmix model runs vs. corresponding PM_{2.5} concentrations measured at the central Steubenville site.

Table 42 presents Spearman correlation coefficients comparing the estimated contributions of the Unmix sources to total PM_{2.5} mass with the estimated source contributions of the seven PMF sources discussed in the previous subsection. Figure 56 presents boxplots showing the seasonal distributions of daily source contributions estimated for each Unmix source factor, and Tables 43 and 44 show Spearman correlation coefficients comparing estimated daily Unmix source contributions with daily average gaseous pollutant concentrations and meteorological conditions and daily measures of the impact of local versus regional sources on PM_{2.5} concentrations in Steubenville.

Table 42. Spearman correlation coefficients comparing estimated daily source contributions (to total PM_{2.5} mass at Steubenville) for the three best five-source Unmix solutions with estimated daily source contributions for the seven-source PMF model.

		PMF Source 1	PMF Source 2	PMF Source 3	PMF Source 4	PMF Source 5	PMF Source 6	PMF Source 7
Unmix Solution #1	Source 1	0.12	0.42	0.36	-0.10	0.28	0.50	0.36
	Source 2	0.11	0.03	-0.13	0.93	-0.41	-0.08	0.02
	Source 3	0.01	0.48	0.67	-0.12	0.24	0.79	0.29
	Source 4	-0.13	0.52	0.56	0.02	0.51	0.04	0.36
	Source 5	0.95	0.34	0.15	-0.03	0.26	0.25	0.40
Unmix Solution #2	Source 1	-0.01	-0.09	-0.24	0.93	-0.50	-0.11	-0.03
	Source 2	0.14	0.67	0.95	-0.16	0.38	0.48	0.45
	Source 3	-0.08	0.63	0.53	-0.02	0.58	0.08	0.28
	Source 4	0.32	0.45	0.29	-0.01	0.32	0.31	0.94
	Source 5	0.96	0.39	0.19	-0.08	0.31	0.35	0.44
Unmix Solution #3	Source 1	0.29	0.37	0.25	-0.19	0.34	0.50	0.33
	Source 2	0.22	0.29	0.24	-0.01	0.22	0.58	0.34
	Source 3	-0.20	0.20	0.25	0.06	0.39	-0.11	0.25
	Source 4	0.17	0.69	0.87	-0.08	0.39	0.35	0.50
	Source 5	0.80	0.24	0.01	0.26	0.01	0.16	0.20

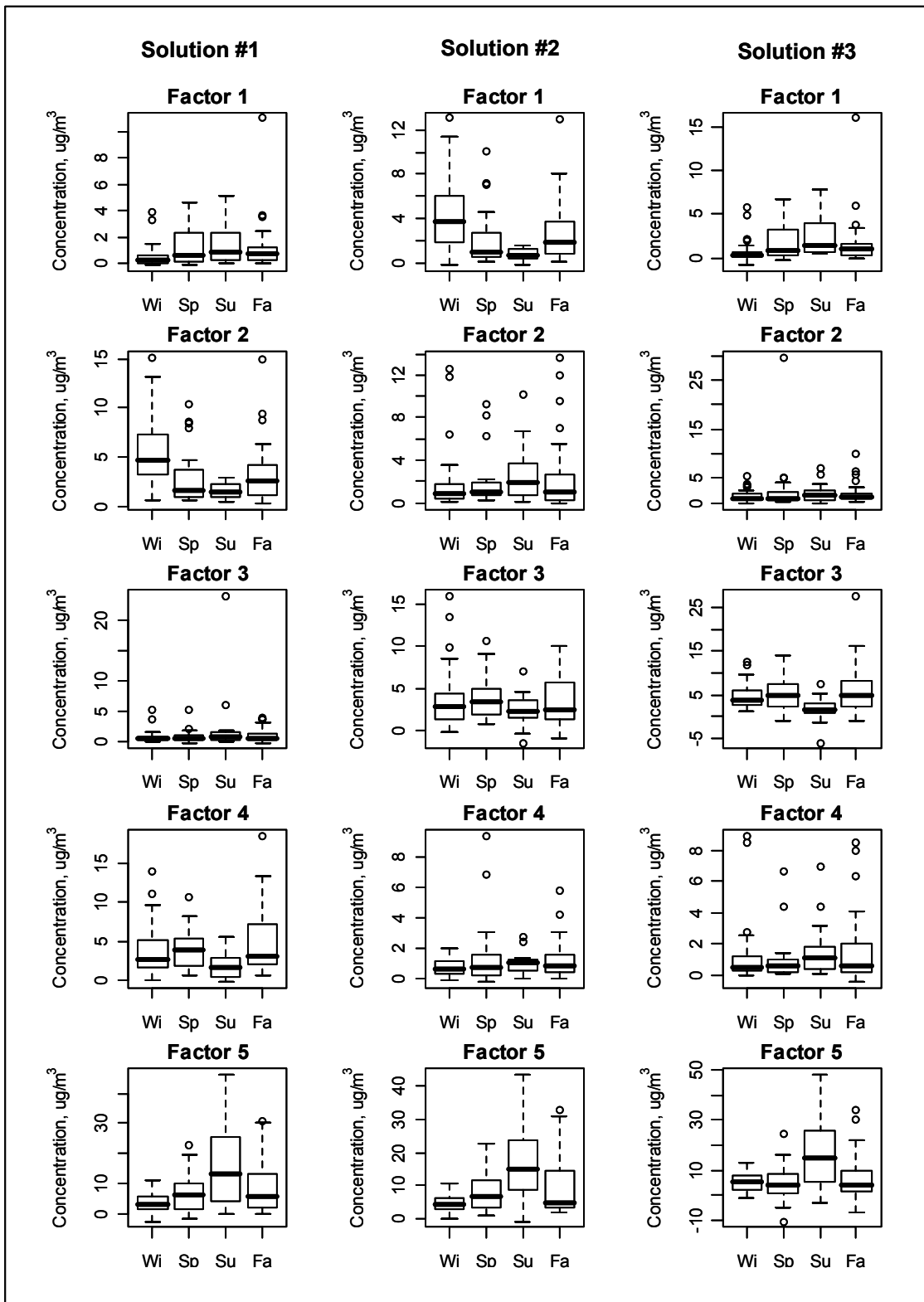


Figure 56. Boxplots showing the distributions of estimated source contributions ($\mu\text{g}/\text{m}^3$) by astronomical season for each of the source factors resolved by the three best five-source Unmix solutions.

Table 43. Spearman correlation coefficients comparing estimated daily source contributions (to total PM_{2.5} mass at Steubenville) for the three best five-source Unmix solutions with daily average gaseous pollutant concentrations and meteorological conditions.

		SO ₂	NO	NO ₂	CO	O ₃	WS	Temp	RH	BP
Unmix Solution #1	Source 1	0.23	0.51	0.58	0.32	-0.17	-0.41	0.26	-0.01	0.29
	Source 2	0.06	0.14	0.01	0.17	-0.36	0.04	-0.47	0.17	-0.23
	Source 3	0.32	0.67	0.68	0.62	-0.20	-0.36	0.20	-0.22	0.29
	Source 4	0.31	0.34	0.45	0.43	-0.09	-0.05	0.13	-0.53	0.26
	Source 5	0.30	0.06	0.09	0.28	0.17	-0.22	0.49	0.23	0.07
Unmix Solution #2	Source 1	-0.02	0.10	-0.02	0.06	-0.41	0.07	-0.58	0.19	-0.26
	Source 2	0.50	0.71	0.70	0.75	-0.09	-0.28	0.31	-0.42	0.46
	Source 3	0.28	0.32	0.46	0.39	0.09	-0.19	0.24	-0.52	0.36
	Source 4	0.70	0.41	0.63	0.40	-0.27	-0.22	0.17	-0.10	0.13
	Source 5	0.27	0.09	0.18	0.33	0.24	-0.30	0.55	0.21	0.15
Unmix Solution #3	Source 1	0.21	0.37	0.52	0.27	-0.01	-0.45	0.41	0.10	0.31
	Source 2	0.24	0.36	0.44	0.37	-0.15	-0.25	0.17	-0.03	0.21
	Source 3	0.21	0.14	0.31	0.22	-0.10	0.18	-0.03	-0.36	0.08
	Source 4	0.57	0.70	0.68	0.70	-0.15	-0.29	0.31	-0.36	0.40
	Source 5	0.20	0.09	0.08	0.27	0.09	-0.24	0.22	0.31	-0.03

Table 44. Spearman correlation coefficients between estimated daily source contributions (to total PM_{2.5} mass at Steubenville) for the three best five-source Unmix solutions and estimates of the impact of local versus regional sources on PM_{2.5} concentrations in Steubenville.

		Local PM _{2.5} ^a	Background PM _{2.5} ^b
Unmix Solution #1	Source 1	0.28	0.01
	Source 2	0.17	0.34
	Source 3	0.44	0.01
	Source 4	0.28	-0.02
	Source 5	0.47	0.72
Unmix Solution #2	Source 1	0.05	0.26
	Source 2	0.70	0.09
	Source 3	0.32	0.05
	Source 4	0.34	0.44
	Source 5	0.52	0.74
Unmix Solution #3	Source 1	0.38	0.28
	Source 2	0.33	0.36
	Source 3	0.12	-0.11
	Source 4	0.65	0.28
	Source 5	0.43	0.78

^aDifference between 24-hr concentration measured at the central Steubenville site and the average of the 24-hr concentrations measured at the northern and western satellite sites; ^bAverage of the 24-hr concentrations measured at the northern and western satellite sites

The results presented in Table 42 indicate that the results of the Unmix models were generally similar to those of the PMF model. Estimated source contributions for each of the seven PMF source factors exhibited a Spearman correlation coefficient of ≥ 0.58 with estimated source contributions for at least one of the 15 source factors resolved by the three best Unmix solutions.

Source 5 from each of the three Unmix solutions correlated well with Source 1 from the seven-source PMF model ($r_s = 0.80-0.96$). As with Source 1 from the PMF model, which was comprised of ~ 71% (w/w) NH_4^+ and SO_4^{2-} and accounted for an estimated $7.4 \mu\text{g}/\text{m}^3$ of the total $\text{PM}_{2.5}$ mass in Steubenville, these Unmix sources were dominated by NH_4^+ and SO_4^{2-} (these species in combination accounted for 58-64% of the total source mass) and contributed an estimated $7.4-8.5 \mu\text{g}/\text{m}^3$ to the total mass of $\text{PM}_{2.5}$ at Steubenville. The slightly greater $\text{PM}_{2.5}$ mass contributions and lower NH_4^+ and SO_4^{2-} weight percentages for the Unmix sources compared to the PMF source likely result from the fact that Unmix was able to resolve fewer sources than PMF; hence, the Unmix sources incorporate some $\text{PM}_{2.5}$ mass that was able to be further resolved and apportioned to additional sources by PMF. As shown in Figure 56, source contributions estimated for Source 5 from each of the three Unmix solutions exhibited pronounced seasonal patterns of higher summertime and lower wintertime concentrations, and as indicated in Table 44, these source contributions correlated more strongly with regional background $\text{PM}_{2.5}$ concentrations than did source contributions from any of the other Unmix sources. Hence, consistent with the interpretation provided earlier for Source 1 from the seven-source PMF model, these Unmix sources represent the contribution of regional, sulfate-dominated secondary particles to $\text{PM}_{2.5}$ in Steubenville.

Unmix Solution #2, Source 2, and Unmix Solution #3, Source 4, each correlated strongly with PMF Source 3 ($r_s = 0.87-0.95$). As shown in Tables 41 and 43 respectively, both of these Unmix sources are enriched in Fe, Mn, and Zn, and correlate strongly with concentrations of CO, NO, and NO_2 . According to the U.S. EPA's 2001 TRI for air emissions, the primary metals industry accounts for 93% of Zn emissions and 83% of Mn emissions from stationary sources in Jefferson County, OH, where Steubenville is located. (Although not reported to the TRI, substantial Fe emissions would also be expected from these facilities). The approximate ratio of the mass of Mn to the mass of Zn emitted by these facilities is 0.16 based upon the TRI data. Mn/Zn ratios computed for Unmix Solution #2, Source 2, and Unmix Solution #3, Source 4 of 0.18 and 0.20, respectively, agree remarkably well with the TRI value of 0.16. Moreover, as shown in Table 44, these two source factors correlated more strongly than any of the other Unmix source factors with estimated concentrations of $\text{PM}_{2.5}$ from local sources in the immediate Steubenville vicinity. Hence, they likely represent the contribution of the local iron and steel industry to ambient $\text{PM}_{2.5}$ concentrations in Steubenville. The total average $\text{PM}_{2.5}$ mass contributions of $2.2 \mu\text{g}/\text{m}^3$ and $1.3 \mu\text{g}/\text{m}^3$ estimated for Unmix Solution #2, Source 2, and Unmix Solution #3, Source 4, respectively (average of these two solutions = $1.8 \mu\text{g}/\text{m}^3$), are comparable to the contribution of $1.8 \mu\text{g}/\text{m}^3$ estimated for PMF Factor 3, which is likewise believed to represent local iron and steel emissions.

Other common sources resolved by both the Unmix and PMF models are as follows:

- A source enriched in Pb, which contributed an estimated $0.6 \mu\text{g}/\text{m}^3$ (PMF Source 6) to $1.2 \mu\text{g}/\text{m}^3$ (Unmix Solution #1, Source 3) to the total mass of $\text{PM}_{2.5}$ in Steubenville. As discussed in the previous subsection, the PMF source profile and PSCF results suggest that this source includes contributions from various nonferrous metal smelting and processing plants and other industrial facilities located in southwestern Pennsylvania, northern West Virginia, and eastern Ohio.
- A nitrate-dominated source (PMF Source 4; Unmix Solution #1, Source 2; Unmix Solution #2, Source 1), which contributed an estimated $2.7-3.6 \mu\text{g}/\text{m}^3$ to the total mass of $\text{PM}_{2.5}$ in Steubenville and, as shown in Figures 54 and 56, exhibited the expected seasonal trend of greater wintertime contributions and lesser summertime contributions.

- A source enriched in Se, which contributed an estimated 0.6 $\mu\text{g}/\text{m}^3$ (PMF Source 7) to 1.0 $\mu\text{g}/\text{m}^3$ (Unmix Solution #2, Source 4) to the total mass of $\text{PM}_{2.5}$ in Steubenville and, as shown in Table 43 and Appendix D, correlated well with SO_2 and NO_2 . Per the discussion in the previous section, this source likely represents the contribution of primary coal combustion emissions to $\text{PM}_{2.5}$ in Steubenville.

The remaining two PMF sources, the vehicular / local carbon source (Source 2) and the crustal source (Source 5), were not clearly resolved by the three best five-source Unmix solutions. The lack of a clearly resolved vehicular source in the Unmix solutions is not surprising, because the Unmix model runs that produced these solutions included neither EC nor Ba, two of the important tracers used by PMF to resolve this source. The lack of a clearly resolved crustal source in the Unmix solutions is also not surprising, because the Unmix model specifications each included only one or two of the five major species (Al, Ca, K, Mg, Ti) that were abundant in the PMF crustal source. These species cannot be expected to individually serve as tracers for a unique crustal source, because they are each found in significant quantities in the compositional profiles of a number of other sources. Two of the Unmix sources, however, appear to represent a blending of vehicular / local carbon particles and crustal particles. These are Unmix Solution #2, Source 3, which correlated moderately with PMF Sources 2 and 5 ($r_s = 0.63$ and 0.58 , respectively) and was enriched in OM and K, and Unmix Solution #1, Source 4, which also correlated moderately with both of these PMF sources ($r_s = 0.51$ - 0.52) and was enriched in OM, Al, and K. The estimated contributions of these Unmix sources to total $\text{PM}_{2.5}$ mass were within 25% of the combined estimated mass contribution of PMF Sources 2 and 5.

Hence, the Unmix results, although limited by their inability to utilize all of the available $\text{PM}_{2.5}$ speciation data from the central Steubenville site, confirm the reasonableness of the seven sources resolved by PMF. Results from both source apportionment models confirm the existence of distinct regional and local source mechanisms that contribute to ambient $\text{PM}_{2.5}$ concentrations in Steubenville, consistent with the case studies presented earlier in this report. These mechanisms will need to be considered independently when developing an implementation plan for Steubenville. The various model runs concur that regional, secondary-sulfate dominated $\text{PM}_{2.5}$ is the largest contributor to total $\text{PM}_{2.5}$ mass in Steubenville. These secondary sulfates likely originate primarily from SO_2 emitted by coal-fired power plants (and other more minor sources) in the eastern United States. However, source apportionment results suggest that primary particle emissions from coal-fired power plants contribute very little (i.e., $\leq 1 \mu\text{g}/\text{m}^3$) to total $\text{PM}_{2.5}$ mass in Steubenville. Local sources, including motor vehicles and iron and steel facilities, also contribute appreciably to $\text{PM}_{2.5}$ mass concentrations in the immediate Steubenville vicinity. Unlike secondary sulfate particles, the particles produced by these sources are enriched in carbon species and metals such as Fe, Mn, and Zn, and elevated concentrations of these particles tend to be accompanied by elevated concentrations of primary pollutant gases such as NO_2 and CO. Further research is recommended to elucidate possible differences in the health implications of these chemically dissimilar types of particles.

4.3 Indoor and Personal Exposure Program

4.3.1 Older Adult Cohort

4.3.1.1 Participant Summary and Characteristics

Twenty-eight subjects (25 female, 3 male) were involved in the indoor and personal monitoring components of the study. The average age was 71.8 years (range: 53-90 years). No

participants were current smokers, although four participants reported guests or family visitors who smoked occasionally.

4.3.1.2 Housing Summary and Characteristics

Results from the one-time dwelling questionnaire are presented in Table 45. About half of the subjects lived in the Kennedy complex, and approximately one-sixth lived in each of the other apartment complexes and in other houses. Locations of the apartment complexes relative to the SCAMP central ambient air monitoring site are shown in Figure 57.

Table 45. Housing characteristics for the older adult cohort.

	Apartment*	E	G	K	O
	Subjects	5	4	15	4
Building Characteristics	Date Built (year)	1962	1975	1964	n/a
	Units	88	100	155	n/a
	Floors	6	8	9	2
Ventilation	AC	4	4	14	3
	Whole House Fan	0	0	0	1
	Indoor Fan, Summer	4	3	14	2
	Indoor Fan, Fall	3	2	7	4
Heat Source	Radiator	5	0	15	3
	Forced Air	0	0	0	2
	Open Stove	0	0	0	1
	Electric Heater	0	4	0	1
	Gas Heater	0	0	0	1
	Fireplace	0	0	0	1
Cooking Fuel	Electric	0	4	15	1
	Gas	5	0	0	3

*E = Elmer White, G = Gaylord, K = J.F. Kennedy, O = other homes.
Data obtained from housing questionnaires. Values for ventilation, heating source, and cooking fuel expressed as number of participants.

Air conditioning was present in the majority of buildings, whereas whole house fans were rare, and indoor fans were more prevalent in the summer compared to the fall. Gas stoves were prevalent in the Elmer White building and in other homes, and electric stoves were found in the Gaylord and Kennedy apartments. Heat was primarily dissipated via radiators, except in the Gaylord apartments where electric heaters were used.

4.3.1.3 Time-Activity Data Summary

Results from the time-activity questionnaires are summarized in Tables 46 through 49. Table 46 displays statistics for the number of 30-minute segments per personal sampling day spent in various locations. During both seasons, participants spent the majority of segments in the home, although more time was spent at home during the fall season. This difference is

accounted for by more segments being spent outdoors in the summer. Other significant locations frequented during the summer season included others' homes, the apartment's community room, church, and other public buildings. In the fall, others' homes, public buildings, and health testing rooms were frequently visited, as were outdoor locations.

Table 46. Average number of 30-minute segments per day spent in various locations for the older adult cohort.*

General Location	Specific Location	Summer					Fall				
		Mean	SD	Median	Min	Max	Mean	SD	Median	Min	Max
Own Home	Indoors at Home	31.6	8.1	34.8	15.0	40.6	36.5	5.5	35.8	28.8	46.7
Other's Home	Same Building	1.2	0.7	1.2	0.6	1.7	1.5	1.4	0.8	0.6	3.1
	Laundry Room	0.3	0.4	0.2	0.0	0.9	0.7	0.5	0.8	0.1	1.4
	Community Room	1.4	1.8	0.4	0.1	4.7	2.3	3.0	1.0	0.0	8.4
	Indoor Lobby	0.5	0.3	0.5	0.0	1.0	0.9	1.2	0.4	0.0	3.5
	Hallway	0.3	0.4	0.2	0.0	1.0	0.3	0.2	0.3	0.1	0.5
	Health Testing Room	0.9	0.2	1.0	0.7	1.2	0.9	0.2	0.9	0.7	1.2
Transit	Car	0.9	1.2	0.4	0.0	3.1	0.9	0.8	0.6	0.1	2.6
	Truck						0.8	0.6	1.0	0.1	1.3
	Bus	0.1	0.1	0.1	0.0	0.3	0.2	0.1	0.2	0.2	0.3
	Walking	0.1		0.1	0.1	0.1	0.5	0.6	0.4	0.0	1.3
Other Indoor	Different Building	1.4	2.3	0.6	0.1	6.0	1.7	2.1	0.9	0.0	5.6
	Public Building	1.1	1.0	0.8	0.0	3.3	1.1	1.0	0.8	0.0	3.3
	Bar/Night Club	0.3	0.2	0.3	0.2	0.5	0.4	0.1	0.4	0.4	0.5
	Health Care	0.5	0.8	0.2	0.0	2.1	1.0	1.5	0.5	0.1	3.7
	Beauty Parlor	0.0		0.0	0.0	0.0	0.1	0.0	0.1	0.0	0.1
	Restroom	0.5	0.6	0.2	0.1	1.4	0.3	0.2	0.3	0.0	0.5
Outdoors (Not Home)	Church	1.2	1.5	0.5	0.0	2.9	0.7	0.6	0.6	0.1	1.4
	Other Outside						0.2		0.2	0.2	0.2
	Sidewalk/Street, Parking Lot	0.0	0.0	0.0	0.0	0.0	0.1		0.1	0.1	0.1
	Park, Pool, River, Lake	0.9		0.9	0.9	0.9					
Outdoors (Near Home)	Yard/Patio/Other Outside	0.3	0.0	0.3	0.3	0.3	0.4		0.4	0.4	0.4
	Other Outside						0.7	0.4	0.7	0.4	1.0
	Sidewalk/Street	2.1	3.3	0.8	0.1	8.7	0.6	0.5	0.5	0.2	1.4
	Parking Lot	0.0		0.0	0.0	0.0	0.1	0.1	0.1	0.0	0.2
	Yard/Patio/Other Outside	1.8	2.1	0.9	0.3	5.3	0.5	0.6	0.1	0.0	1.4

*Total number of 30-minute segments in a day is 48. Data obtained from participant-completed time-activity diaries for the ten subjects participating in personal monitoring each season. Five subjects participated in both seasons. Blank cells indicate subjects did not visit the particular location.

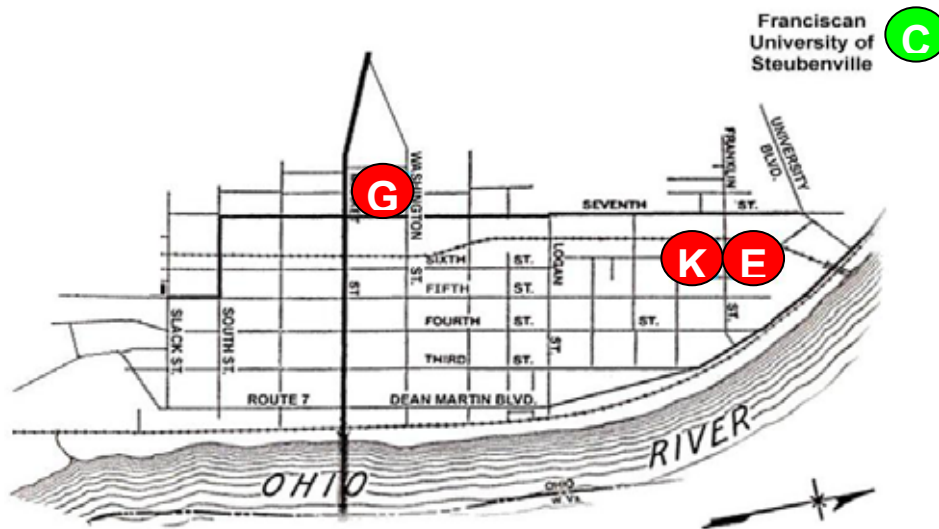


Figure 57. Locations of the apartment complexes where many of the older adult participants resided (“G”, “K”, “E”) and the central ambient air monitoring site (“C”) in Steubenville, OH. “G” represents the location of the Gaylord apartment building, “K” the Kennedy apartment building, “E” the Elmer White apartment building, and “C” the central ambient site located at the Franciscan University of Steubenville.

Table 47 gives the average count of 30-minute segments per sampling day spent doing various activities. Personal needs and care, communications, and household work were the most frequent activities in both the summer and fall. Overall, the personal needs and care category, which included bathing, eating, sleeping, and personal hygiene activities, had the greatest average count, with 17.3 segments per day in the summer and 20.3 in the fall. All categories, with the exception of child care, in transit, and educational, had higher mean counts in the fall season than in the summer season. Median values for the communications, child care, in transit, organizational, and obtaining goods and services activity categories, however, were greater in the summer than in the fall.

Table 47. Average number of 30-minute segments per day spent doing various activities for the older adult cohort.*

Activity	Summer					Fall				
	Mean	SD	Median	Min	Max	Mean	SD	Median	Min	Max
Household Work	2.6	1.4	3.0	0.8	4.3	4.3	3.1	4.1	0.0	9.5
Child Care	2.6	1.8	2.6	1.3	3.8	0.5	0.4	0.3	0.2	0.9
Obtaining Goods, Services	1.1	0.8	0.9	0.3	2.7	1.3	1.3	0.8	0.3	3.8
Personal Needs and Care	17.3	4.4	17.5	7.0	22.2	20.3	3.4	21.3	15.6	24.3
Educational	0.4		0.4	0.4	0.4	0.1	0.1	0.1	0.0	0.1
Organizational	1.3	1.3	0.9	0.0	3.4	1.4	1.7	0.6	0.0	5.3
Entertainment/Social	1.2	0.7	1.3	0.3	2.7	2.6	2.6	1.3	0.3	6.3
Recreation	1.1	1.0	1.2	0.0	2.4	1.5	1.1	1.8	0.1	3.6
Communications	8.6	4.3	8.2	2.9	17.0	8.8	4.5	7.7	4.3	19.4
In Transit	2.1	2.3	1.7	0.0	8.2	1.6	0.7	1.6	0.3	2.5

*Total number of 30-minute segments in a day is 48. Data obtained from participant-completed time-activity diaries for the ten subjects participating in personal monitoring each season. Five subjects participated in both seasons.

The proximity of the monitored individuals to particle sources and the indoor ventilation conditions that they encountered during monitoring were recorded as well. Summary statistics are given in Table 48. Subjects on average spent more time near busy roads and in indoor environments with air conditioners (AC) and open windows in the summer season than they did in the fall season. Subjects on average spent slightly more time near cooking, cleaning, and smoking in the fall season than in the summer season, however. Overall, subjects were in indoor environments with air conditioning and open windows more frequently than they were near busy roads, cooking, cleaning, and smoking.

As shown in Table 49, more time was spent in transit, on average, during the summer as compared to the fall (24.91 versus 19.89 min/day, respectively). Similarly, subjects spent more time outdoors in the summer, with an average value more than three times as great as that in fall.

Table 48. Proximity to sources/ventilation, average count of 30-minute segments per day for the older adult cohort.*

Proximity to:	Summer					Fall				
	Mean	SD	Median	Min	Max	Mean	SD	Median	Min	Max
Busy Road	3.3	4.1	1.7	0.5	12.9	1.3	1.0	1.0	0.2	3.2
AC	17.0	11.9	17.9	2.4	31.9	5.4	6.8	3.5	0.0	19.8
Window	16.3	10.8	18.2	1.0	27.3	11.1	13.8	7.0	0.0	43.7
Cooking	1.9	1.3	1.8	0.7	5.2	3.3	2.1	3.2	0.3	6.9
Cleaning	0.8	1.0	0.6	0.1	3.3	1.0	1.1	0.6	0.0	3.6
Smoker	0.5	0.5	0.4	0.0	1.4	0.9	1.0	0.8	0.0	3.0

*Total number of 30-minute segments in a day is 48. Data obtained from participant-completed time-activity diaries for the ten subjects participating in personal monitoring each season. Five subjects participated in both seasons

Table 49. Average duration (min/day) in transit and outdoors (n = 10).

Microenvironment	Summer					Fall				
	Mean	SD	Median	Min	Max	Mean	SD	Median	Min	Max
In Transit	24.9	26.4	13.5	1.6	72.8	19.9	16.3	18.1	2.7	56.7
Outdoors	83.9	113.3	43.2	1.1	358.0	24.6	21.0	19.6	8.7	81.5

Note: Data are for 10 subjects each season; 5 subjects participated in both seasons.

4.3.1.4 Pollutant Summary

4.3.1.4.1 Precision, Accuracy, and Data Completeness

Sample validity, limits of detection (LOD), completeness, imprecision, and accuracy for pollutant measurements made for the older adult cohort are presented in Table 50. There were 7-22 valid repeated measures per subject per pollutant in the summer season. In the fall, there were 13-24 valid samples per subject for indoor monitoring and 16-24 valid samples for personal monitoring.

Relatively few EC and NO₂ samples were below the LODs for these species, whereas a considerable percentage of fall O₃ indoor/personal, summer and fall SO₂, and summer and fall elemental data were non-detects, thus making their distributions left-censored. This issue was handled by the selection of appropriate statistical analysis techniques (detailed below) and by the insertion of random numbers to represent the noise in the data. Values were randomly chosen from data below the analytical LODs that were collected during a previous field study in

Baltimore, MD. Two hundred ninety-one, 755, and 10 observations were required for O₃, SO₂, and NO₂, respectively. In addition, 45 observations were required for EC; these were obtained from a random number generator using a uniform distribution. The random number insertions were verified by conducting three trials. Different combinations of random numbers had no effect on overall pollutant concentrations and other summary statistics.

Table 50. Quality assurance parameters for pollutant data collected for the older adult cohort.

Pollutant	Season	N	Field LOD*			Completeness	Imprecision	Accuracy
			Ambient (Outdoor)	Indoor	Personal			
PM _{2.5}	Summer	214	3.0	3.0	6.6	93%	1 - 2%	93%
	Fall	256	2.9	2.9	5.7	94%	0 - 2%	
SO ₄ ²⁻	Summer	106	0.2	0.3	0.4	92%	10.8%	NA
	Fall	127	0.2	0.2	0.2	93%		
EC	Summer	100	0.55	0.55	0.55	90%	14.5%	NA
	Fall	128	0.04	0.04	0.04	94%		
O ₃	Summer	98	12.7	12.7	12.7	97%	10.4%	92%
	Fall	121	10.7	10.7	10.7	99%		
NO ₂	Summer	95	10.8	10.8	10.8	96%	17.0%	106%
	Fall	120	6.4	6.4	6.4	99%		
SO ₂	Summer	97	5.5	5.5	5.5	98%	24.9%	73%
	Fall	121	3.8	3.8	3.8	99%		

*Limit of detection for PM_{2.5}, SO₄²⁻ and EC in µg/m³; for gases in ppb; "N" is the number of blanks; EC LOD was lower in the fall due to the use of a new measurement method for EC; NA = not available.

4.3.1.4.2 Ambient Concentrations

Table 51 presents descriptive statistics for ambient pollutant concentrations measured at the central monitoring site on the campus of Franciscan University of Steubenville using the Harvard multi-pollutant monitor. Ambient PM_{2.5} concentrations were comparable during both of the seasons in which monitoring was conducted, with averages of 20.1 (± 9.3) µg/m³ during the summer and 19.3 (± 12.2) µg/m³ during the fall. Ambient SO₄²⁻ (expressed as ammonium sulfate) comprised a large fraction of the total PM_{2.5} mass, with contributions of 52% and 43% in the summer and fall, respectively. Ambient EC, in contrast, comprised only 6% of the PM_{2.5} mass in either season. The composition of PM_{2.5} reflects the pollutant sources in the Steubenville region, which include numerous coal-fired power plants that contribute to SO₄²⁻, but relatively little motor vehicle traffic, which is a source of EC.

Among the gases, ambient O₃ concentrations showed the greatest seasonal differences, with considerably higher mean concentrations during the summer (29.3 ± 13.4 ppb) as compared to the fall (16.0 ± 8.1 ppb). The higher summertime concentrations likely reflect the importance of photochemical processes for O₃ production. For both SO₂ and NO₂, mean summer concentrations were low, falling below their respective field LODs, and fall concentrations were slightly higher.

Table 51. Ambient and outdoor concentrations by season for the older adult cohort.*

Location	Pollutant	Season	N	#ND	Mean	Median	SD	Max
Ambient	PM _{2.5}	Summer	65	0	20.1	18.5	9.3	46.6
		Fall	72	0	19.3	16.1	12.2	50.7
	SO ₄ ⁻²	Summer	61	0	7.7	7.2	4.8	25.0
		Fall	72	0	6.2	5.2	4.7	22.4
	EC	Summer	61	5	1.0	1.0	0.6	2.9
		Fall	71	0	1.1	0.9	0.7	3.6
	O ₃	Summer	62	0	29.3	26.4	13.4	74.8
		Fall	72	0	16.0	14.4	8.1	42.4
	SO ₂	Summer	63	23	2.7	2.2	3.9	21.9
		Fall	71	24	5.4	2.6	9.6	63.6
	SO ₂ > ND**	Summer	40	0	4.6	3.7	3.6	21.9
		Fall	47	0	8.2	4.2	10.8	63.6
	NO ₂	Summer	62	1	9.5	8.5	7.4	37.9
		Fall	71	0	11.3	10.1	6.0	27.9
Kennedy Roof	PM _{2.5}	Summer	61	0	21.6	21.0	9.2	47.3
		Fall	70	0	21.1	18.0	11.8	64.7
	SO ₄ ⁻²	Summer	58	0	7.7	7.2	4.3	24.4
		Fall	69	0	6.3	4.7	4.4	18.1
	EC	Summer	55	2	1.2	1.1	0.5	2.4
		Fall	68	0	1.3	1.2	0.7	3.8
	O ₃	Summer	64	0	30.6	29.0	13.5	87.3
		Fall	71	1	15.2	13.9	8.3	46.2
	SO ₂	Summer	64	26	2.6	1.9	3.9	22.4
		Fall	69	17	4.4	3.5	5.9	38.1
	SO ₂ > ND	Summer	38	0	4.7	3.9	3.7	22.4
		Fall	52	0	6.0	4.0	6.0	38.1
	NO ₂	Summer	64	3	9.6	9.7	6.6	34.8
		Fall	69	0	12.9	11.8	5.6	26.9
Gaylord Roof	PM _{2.5}	Summer	30	0	19.3	17.1	10.2	51.5
		Fall	34	0	19.5	17.8	11.3	61.7
	SO ₄ ⁻²	Summer	28	0	7.7	6.2	5.7	26.2
		Fall	33	0	5.9	5.1	3.7	16.5
	EC	Summer	25	0	1.1	1.0	0.5	2.4
		Fall	32	0	1.3	1.0	0.8	3.7
	O ₃	Summer	32	0	31.6	30.0	10.9	60.3
		Fall	35	1	15.4	14.3	9.2	42.9
	SO ₂	Summer	32	14	3.3	1.7	5.6	27.9
		Fall	35	5	5.8	3.5	7.5	37.1
	SO ₂ > ND	Summer	18	0	6.2	5.2	6.0	27.9
		Fall	30	0	6.8	4.2	7.7	37.1
	NO ₂	Summer	32	0	9.3	9.4	5.0	20.0
		Fall	35	0	12.3	11.1	6.2	26.4

*PM_{2.5}, SO₄⁻² and EC in µg/m³; gases in ppb

**"N" represents the number of valid samples; "ND" represents the number of samples with concentrations below the analytical detection limits; "SD" is the standard deviation; "max" is the maximum measured concentration.

Correspondingly, we found significant summertime associations among ambient PM_{2.5}, SO₄²⁻, and O₃ concentrations, which were likely due to the common effects of photochemistry on concentrations of these pollutants. Table 52 shows associations among the ambient pollutants by season. During the fall, associations between ambient particles and O₃ were negative, which may be due to meteorological conditions during this season. As discussed earlier in the section describing results from the ambient air monitoring portion of SCAMP, Steubenville experiences considerable inversions during the fall, which can trap PM_{2.5} and local pollutants close to the ground while preventing mixing with the air aloft containing regional pollutants, such as O₃. Associations between ambient particles and NO₂ and SO₂ were only significant during the fall. In this season, associations between ambient EC and NO₂ concentrations were particularly strong and positive (t-value = 11.39). This strong association is likely explained by the fact that both EC and NO₂ are emitted by motor vehicles and other similar sources.

Table 52. Associations (analyzed via mixed models) among ambient pollutants measured at the central Steubenville site during the summer and fall of 2000 using Harvard multi-pollutant monitors.*

Comparison		Summer					Fall				
		N	β	SE	t-stat	R ²	N	β	SE	t-stat	R ²
O ₃	PM _{2.5}	62	0.74	0.16	4.55	0.26	72	-0.2	0.08	-2.41	0.07
NO ₂		62	-0.01	0.11	-0.1	0	71	0.38	0.04	9.75	0.61
SO ₂		63	0.07	0.05	1.37	0.03	71	0.4	0.1	4.14	0.22
O ₃	SO ₄ ²⁻	58	1.45	0.28	5.09	0.27	72	-0.52	0.23	-2.24	0.07
NO ₂		58	-0.17	0.21	-0.79	0.01	71	0.96	0.12	7.9	0.49
SO ₂		59	0.18	0.11	1.66	0.05	71	1.38	0.25	5.45	0.33
O ₃	EC	53	-6.98	3.9	-1.79	0.06	71	-3.18	1.44	-2.2	0.06
NO ₂		53	3.76	2.19	1.72	0.06	70	7.01	0.62	11.39	0.68
SO ₂		54	-0.65	0.81	-0.8	0.01	70	9.39	1.56	6.03	0.34

*bold numbers indicate p-values less than or equal to 0.05

4.3.1.4.3 Outdoor Concentrations

Mean and median outdoor pollutant concentrations were comparable at the Kennedy and Gaylord apartment buildings in both the summer and the fall. In addition, seasonal trends in outdoor pollutant concentrations were comparable at the sites (Table 51). As with the ambient concentrations, mean outdoor O₃ concentrations in summer were approximately two times as great as those in fall. For example, at the Kennedy site, the mean summer outdoor O₃ concentration was 30.6 ppb, while the mean fall concentration was 15.2 ppb. Mean and median outdoor PM_{2.5} and EC concentrations did not vary seasonally, whereas outdoor SO₄²⁻ concentrations were slightly greater in the summer than in the fall (7.7 versus 6.3 μg/m³, respectively, at the Kennedy site; 7.7 versus 5.9 μg/m³, respectively, at the Gaylord site). Mean SO₂ and NO₂ concentrations, on the other hand, were slightly greater in the fall than in the summer.

Outdoor pollutant concentrations were also comparable to those measured at the central ambient air monitoring site at Franciscan University of Steubenville. Table 53 shows correlations between ambient and outdoor concentrations for each of the major pollutants

measured using the Harvard multi-pollutant monitors at these sites. Figures 58 through 63 show pollutant-specific scatterplots of ambient vs. outdoor concentrations. A parity line is also shown on each plot for reference. These correlations and scatterplots indicate little spatial variability across the study area in outdoor concentrations of PM_{2.5}, SO₄²⁻, and O₃ in both seasons, and in outdoor concentrations of EC and NO₂, especially during the fall. For these pollutants and seasons, strong correlations justify using the ambient data to represent outdoor concentrations in longitudinal analyses.

Table 53 Correlations between ambient pollutant concentrations at the SCAMP central site and outdoor concentrations measured at apartment complexes in Steubenville using Harvard multi-pollutant monitors in Summer and Fall of 2000.*

Pollutant	Location	Summer		Fall	
		Kennedy	Gaylord	Kennedy	Gaylord
PM _{2.5}	Ambient	0.98⁺	0.94⁺	0.95⁺	0.95⁺
	Kennedy		0.96⁺		0.96⁺
O ₃	Ambient	0.83⁺	0.82⁺	0.74⁺	0.72⁺
	Kennedy		0.79[*]		0.75[*]
EC	Ambient	0.53⁺	0.64	0.93⁺	0.89⁺
	Kennedy		0.83[*]		0.91[*]
NO ₂	Ambient	0.55⁺	0.58	0.86⁺	0.73⁺
	Kennedy		0.68[*]		0.65[*]
SO ₄ ²⁻	Ambient	0.91⁺	0.97⁺	0.93⁺	0.95⁺
	Kennedy		0.98[*]		0.88[*]
SO ₂	Ambient	0.20	0.37	0.68⁺	0.81⁺
	Kennedy		0.41		0.74⁺

*bold numbers indicate p-values ≤ 0.05

+ indicate p-values ≤ 0.0001

Note: Kennedy and Gaylord are rooftop concentrations

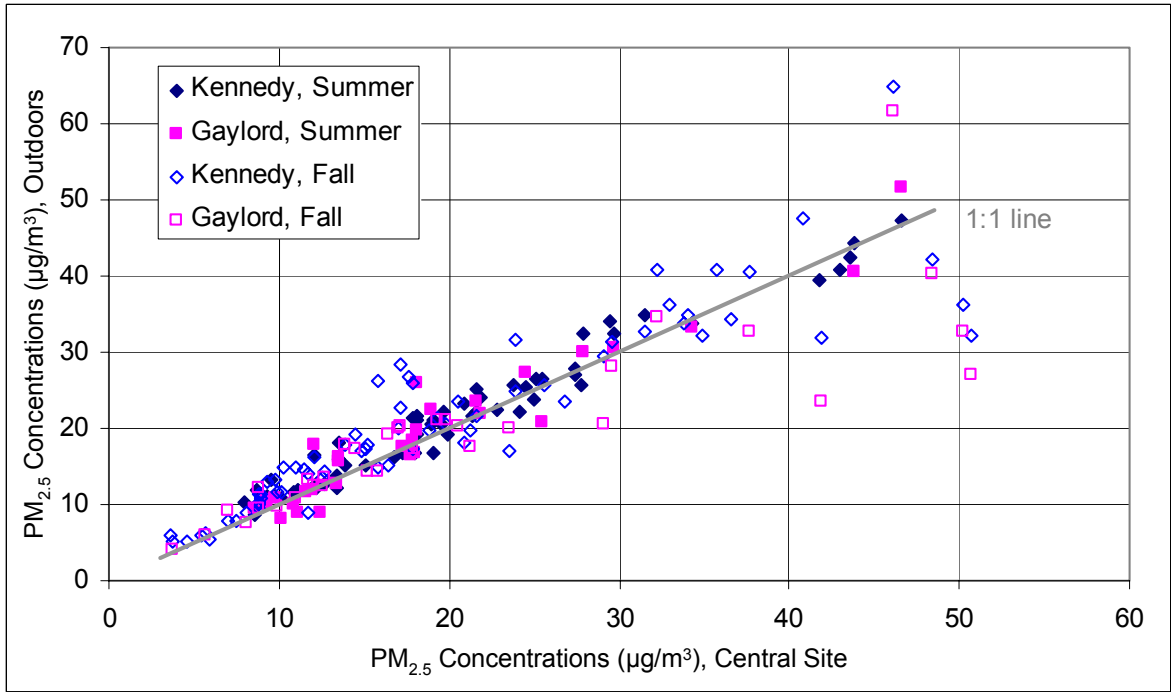


Figure 58. Ambient and outdoor $PM_{2.5}$ concentrations in Steubenville measured using Harvard multi-pollutant monitors in the summer and fall of 2000.

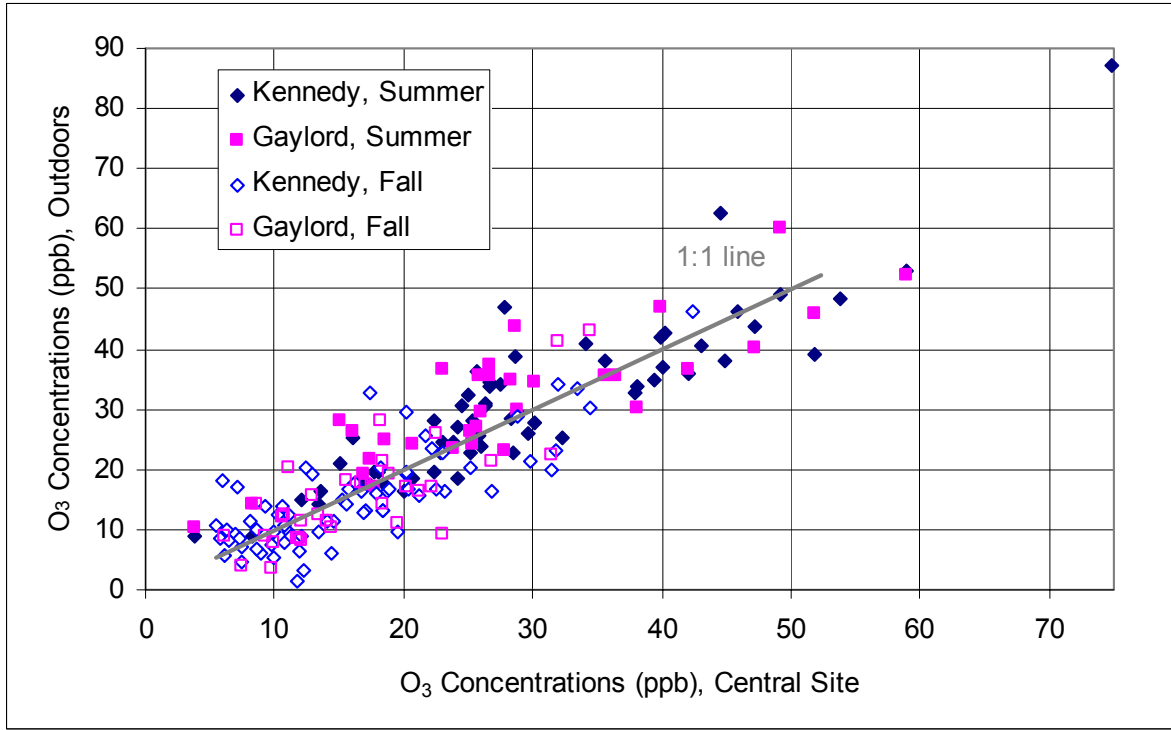


Figure 59. Ambient and outdoor O_3 concentrations in Steubenville measured using Harvard multi-pollutant monitors in the summer and fall of 2000.

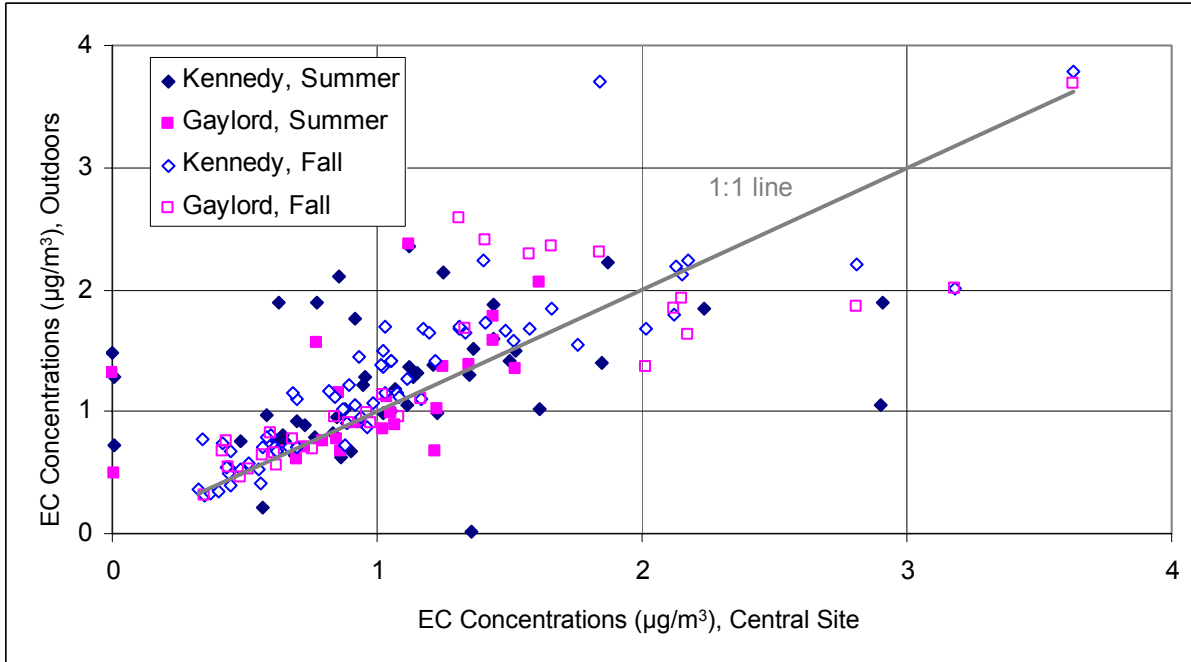


Figure 60. Ambient and outdoor EC concentrations in Steubenville measured using Harvard multi-pollutant monitors in the summer and fall of 2000.

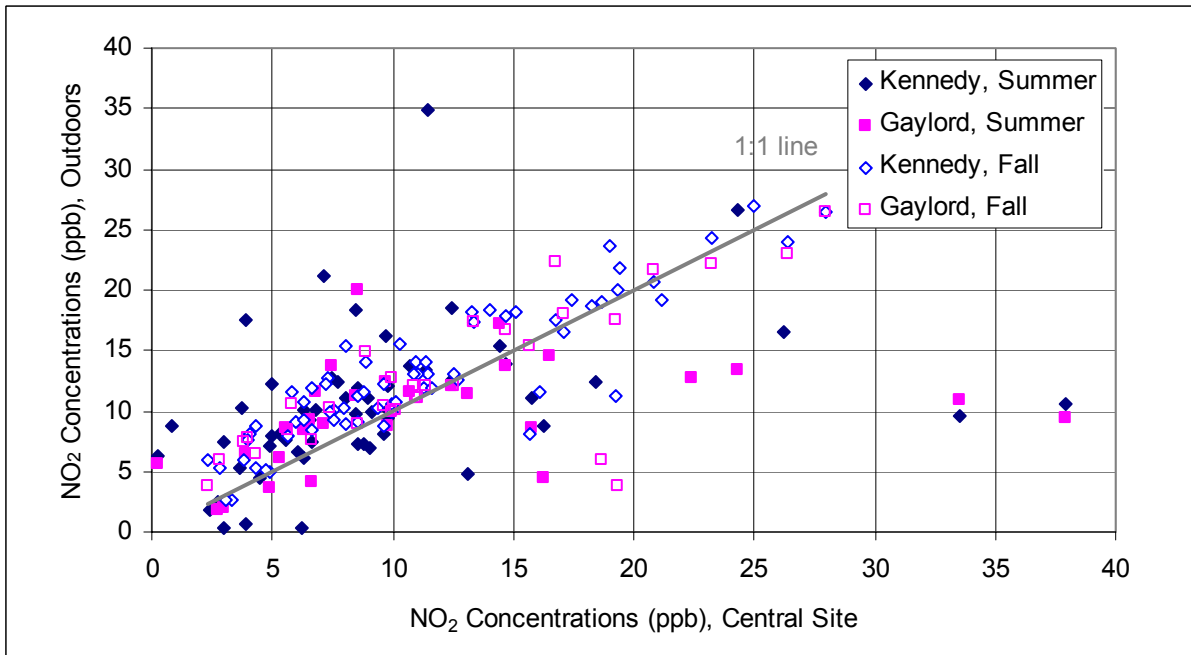


Figure 61. Ambient and outdoor NO_2 concentrations in Steubenville measured using Harvard multi-pollutant monitors in the summer and fall of 2000.

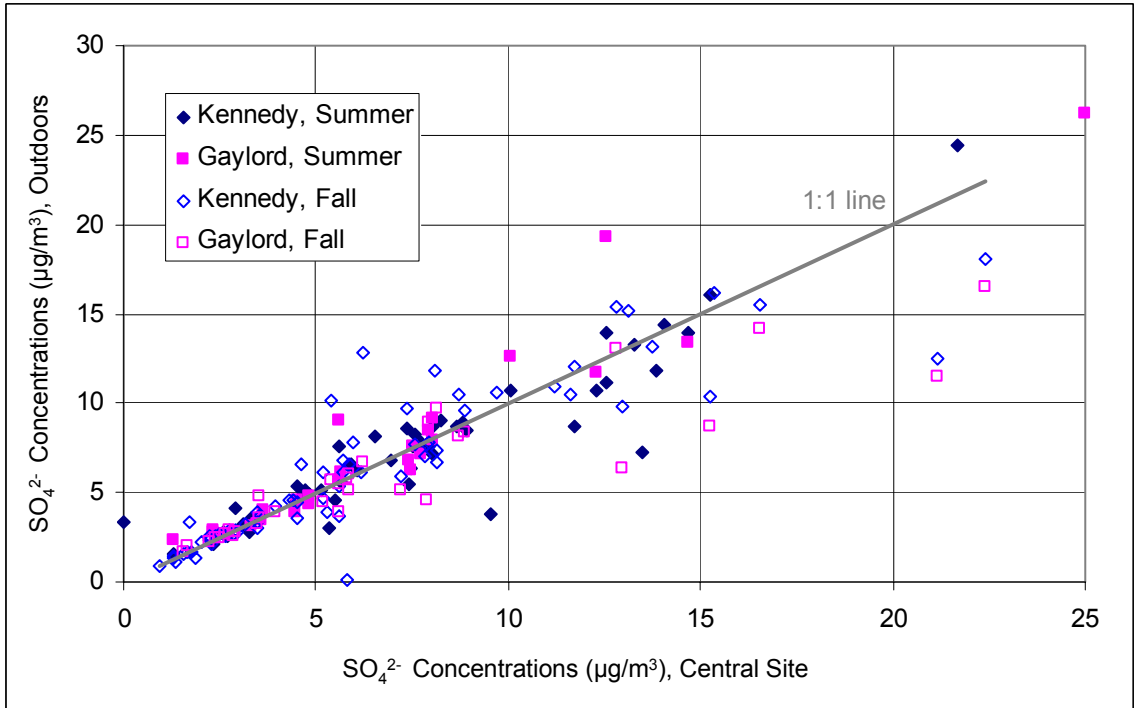


Figure 62. Ambient and outdoor SO_4^{2-} concentrations in Steubenville measured using Harvard multi-pollutant monitors in the summer and fall of 2000.

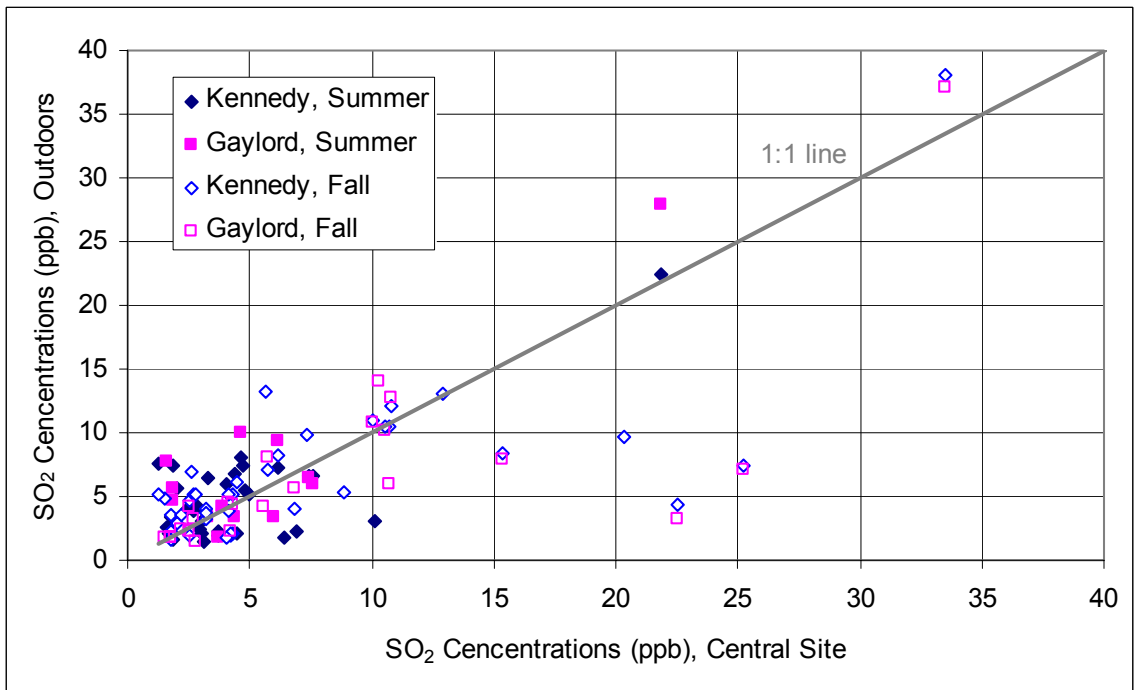


Figure 63. Ambient and outdoor SO_2 concentrations in Steubenville measured using Harvard multi-pollutant monitors in the summer and fall of 2000.

As would be expected given the similarities between outdoor and ambient pollutant concentrations, associations among the pollutants outside Gaylord and Kennedy buildings were similar to those reported in Table 52 among ambient pollutant concentrations.

4.3.1.4.4 Personal and Indoor Concentrations

4.3.1.4.4.1 Summary Concentrations

Summary statistics for personal and indoor concentrations for the older adult cohort are given in Table 54. Means and standard deviations for the indoor and personal microenvironments for all pollutants were comparable. Mean indoor and personal PM_{2.5}, EC, and NO₂ concentrations were similar across seasons, while SO₄²⁻, O₃, and SO₂ mean indoor and personal levels were higher in the summer than in the fall.

On average, personal PM_{2.5}, EC, and NO₂ exposures were slightly higher than corresponding ambient levels. Mean personal/ambient ratios for PM_{2.5} (ratio = 1.14), EC (ratio = 1.15) and NO₂ (ratio = 2.05 and 1.27, for subjects with and without gas stoves in their homes, respectively) all exceeded 1.00, likely due to influences of indoor sources. For pollutants without significant indoor sources, including SO₄²⁻ and in particular O₃, mean personal/ambient ratios were lower than 1.00 (0.75 and 0.24, respectively).

4.3.1.4.4.2 Associations among Pollutants

Tables 55 and 56 display the relationships (by season) among PM_{2.5} and gas levels for personal exposures and indoor concentrations for the older adult cohort. Personal PM_{2.5} and SO₄²⁻ (but not personal EC) were significantly associated with personal O₃ in both seasons. The slopes describing the summer particle-O₃ relationships for personal exposures (0.34 for PM_{2.5}, 0.17 for SO₄²⁻) were similar to the slopes describing corresponding summer relationships among ambient concentrations of these pollutants (0.36 for PM_{2.5}, 0.21 for SO₄²⁻), although model fits were weaker for personal data. The indoor particle-O₃ relationships were similar to those observed for personal exposures. Neither personal nor indoor concentrations of SO₂ were associated with any of the personal or indoor particle measures. In the summer, personal NO₂ was associated only with personal EC. (This result did not hold for corresponding indoor concentrations, though). In the fall, however, for both indoor and personal exposure measurements, concentrations of NO₂ were significantly associated with corresponding concentrations of PM_{2.5}, SO₄²⁻, and EC. (Note that the personal SO₄²⁻-NO₂ association was significant only when homes with gas stoves were excluded.) This result is similar to that observed among ambient concentrations of these pollutants. The associations among indoor concentrations were stronger than were those among personal exposures (as evidenced by the greater slopes and t-statistics for the indoor associations), which may be a result of indoor sources affecting both indoor particle and indoor NO₂ concentrations to a greater extent than they affect personal exposures to these species.

Table 54. Personal and indoor pollutant concentrations by season for the older adult cohort.*

	Pollutant	Season	N	#ND	Mean	Median	SD	Max	
Personal	PM _{2.5}	Summer	169	0	19.9	17.9	9.4	59	
		Fall	204	0	20.1	16.8	11.6	66	
	SO ₄ ⁻²	Summer	165	0	5.9	5	4.2	25.6	
		Fall	188	0	4.4	3.4	3.3	16.3	
	EC	Summer	166	7	1.1	1	0.6	4.6	
		Fall	197	1	1.2	1.1	0.7	6.2	
	O ₃	Summer	183	2	5.3	4.3	5.2	35.7	
		Fall	226	84	3.9	3.5	4.4	21.3	
	SO ₂	Summer	185	99	1.5	1.3	3.3	30.4	
		Fall	228	72	0.7	0.4	1.9	14.2	
	NO ₂	Summer	183	1	9.9	9	6	38.9	
		Fall	228	1	12.1	10.8	6.1	38.8	
	Indoor (subset 10)**	PM _{2.5}	Summer	178	0	20	17.8	10.1	54.5
			Fall	216	0	19.3	15.9	12.2	79.9
SO ₄ ⁻²		Summer	183	0	6.4	5.4	4.6	26.9	
		Fall	223	0	5.1	3.9	3.9	20.8	
EC		Summer	179	14	1	1	0.5	3.7	
		Fall	223	0	1.2	1.1	0.8	8.5	
O ₃		Summer	190	3	6.2	5.1	6	41	
		Fall	228	73	4.3	3.6	4.1	23.8	
SO ₂		Summer	190	109	1.5	1.1	2.7	18.5	
		Fall	227	74	1.1	0.7	2.3	19	
NO ₂		Summer	185	5	11.3	9.5	7.1	38.4	
		Fall	219	1	13.2	11.9	6.8	51.9	
Indoor (subset 15)		PM _{2.5}	Summer	291	0	18.6	17.8	7.4	59.3
			Fall	297	0	19.8	16.6	14.2	131.4
	SO ₄ ⁻²	Summer	292	0	6.2	5.6	3.8	21.5	
		Fall	303	0	4.8	3.5	3.8	18.9	
	EC	Summer	285	19	1	1	0.5	2.9	
		Fall	311	2	1.2	1.1	0.7	5.6	
	O ₃	Summer	296	8	5.6	4.8	5.4	32.3	
		Fall	320	127	3.6	3.1	4.6	27.2	
	SO ₂	Summer	303	169	2.1	1.1	7.3	84.1	
		Fall	320	98	0.8	0.7	1.8	10.9	
	NO ₂	Summer	299	3	10.2	9.3	8.2	61.7	
		Fall	308	0	10.6	8.8	7.6	60.3	
	Indoor (All)	PM _{2.5}	Summer	469	0	19.1	17.8	8.5	59.3
			Fall	513	0	19.6	16.4	13.4	131.4
SO ₄ ⁻²		Summer	475	0	6.2	5.5	4.2	26.9	
		Fall	526	0	4.9	3.6	3.8	20.8	
EC		Summer	464	33	1	1	0.5	3.7	
		Fall	534	2	1.2	1.1	0.7	8.5	
O ₃		Summer	486	11	5.8	4.9	5.7	41	
		Fall	548	200	3.9	3.3	4.4	27.2	
SO ₂		Summer	493	278	1.9	1.1	6	84.1	
		Fall	547	172	0.9	0.7	2	19	
NO ₂		Summer	484	8	10.6	9.3	7.8	61.7	
		Fall	527	1	11.7	10.2	7.4	60.3	

*PM_{2.5}, SO₄²⁻ and EC in µg/m³; gases in ppb

** “subset 10” includes indoor concentrations for individuals with personal monitoring data; “subset 15” includes indoor concentrations for individuals without personal monitoring data; “All” includes indoor concentrations for every study participant. “N” represents the number of valid samples; “ND” represents the number of samples with concentrations below the analytical detection limits; “SD” is the standard deviation; “max” is the maximum measured concentration.

Table 55. Associations between personal particle concentrations and personal gas concentrations, by season, for the older adult cohort.*

Comparison		Summer					Fall				
		N	β	SE	t-stat	Int.	N	β	SE	t-stat	Int.
PM _{2.5}	O ₃	163	0.36	0.16	2.28	17.97	202	0.52	0.18	2.91	18.09
	O ₃ > ND	NA	NA	NA	NA	NA	129	0.36	0.28	1.27	20.29
	SO ₂	166	-0.05	0.21	-0.22	19.87	204	-0.47	0.41	-1.13	20.42
	SO ₂ > ND	77	-0.11	0.31	-0.35	21.36	139	-0.44	0.46	-0.96	19.09
	NO ₂	164	-0.07	0.13	-0.52	20.43	204	0.37	0.13	2.73	15.54
	NO ₂ (no gas stove)	118	-0.06	0.16	-0.35	21.02	122	0.67	0.19	3.59	12.28
SO ₄ ²⁻	O ₃	159	0.17	0.04	4.9	5.03	186	0.19	0.05	3.7	3.62
	O ₃ > ND	NA	NA	NA	NA	NA	119	0.2	0.09	2.35	3.67
	SO ₂	162	-0.24	0.1	-2.53	6.24	188	-0.05	0.12	-0.43	4.44
	SO ₂ > ND	74	-0.15	0.11	-1.4	5.96	133	0.16	0.11	1.38	3.65
	NO ₂	160	-0.09	0.06	-1.57	6.78	188	0.04	0.04	1.05	3.9
	NO ₂ (no gas stove)	113	-0.1	0.07	-1.42	6.81	111	0.15	0.07	2.29	3.08
EC	O ₃	160	-0.02	0.01	-1.7	1.15	195	0.02	0.01	1.54	1.14
	O ₃ > ND	NA	NA	NA	NA	NA	122	0.01	0.02	0.42	1.25
	SO ₂	162	-0.24	0.1	-2.53	6.24	188	-0.05	0.12	-0.43	4.44
	SO ₂ > ND	74	-0.15	0.11	-1.4	5.96	133	0.16	0.11	1.38	3.65
	NO ₂	161	0.03	0.01	3.67	0.77	197	0.04	0.01	4.87	0.75
	NO ₂ (no gas stove)	114	0.04	0.01	4.02	0.73	119	0.06	0.01	6.76	0.55

* Bold represents p-values ≤ 0.05 . "ND" indicates data below detection limit. NA = not applicable.

Table 56. Associations between indoor particle concentrations and indoor gas concentrations, by season, for the older adult cohort.*

Comparison		Cohort	Summer					Fall					
			N	β	SE	t-stat	Int.	N	β	SE	t-stat	Int.	
PM _{2.5}	O ₃	subset 10	175	0.58	0.15	3.97	16.29	216	0.61	0.19	3.14	16.69	
	O ₃ > ND		n/a	n/a	n/a	n/a	n/a	149	0.36	0.3	1.18	18.97	
	O ₃	subset 15	283	0.35	0.09	3.96	16.71	296	0.23	0.17	1.36	19.39	
	O ₃ > ND		n/a	n/a	n/a	n/a	n/a	181	0.1	0.22	0.45	20.33	
	O ₃	All	458	0.44	0.08	5.59	16.6	512	0.37	0.13	2.95	18.4	
	O ₃ > ND		n/a	n/a	n/a	n/a	n/a	330	0.2	0.18	1.09	19.8	
	SO ₂	subset 10	175	0.21	0.27	0.78	19.65	215	-0.4	0.35	-1.14	19.81	
	SO ₂ > ND		80	0.45	0.46	0.98	18.76	146	-0.08	0.42	-0.18	19.08	
	SO ₂	subset 15	290	-0.02	0.06	-0.31	18.69	295	-0.17	0.44	-0.39	20.37	
	SO ₂ > ND		127	-0.03	0.05	-0.47	18.69	208	0.28	0.56	0.5	18.69	
	SO ₂	All	465	0	0.06	-0.05	19.17	510	-0.3	0.28	-1.05	20.16	
	SO ₂ > ND		207	-0.01	0.07	-0.15	19.32	354	0.09	0.34	0.26	18.91	
	NO ₂		subset 10	175	0.23	0.11	2.05	17.38	215	0.85	0.13	6.7	8.13
			subset 15	288	-0.08	0.05	-1.52	19.48	295	0.95	0.11	8.36	9.84
			All	463	0.01	0.05	0.13	19.11	510	0.9	0.08	10.64	9.15
NO ₂ (no gas stove)		subset 10	119	0.1	0.15	0.69	19.34	133	1.28	0.18	7.33	5.44	
		subset 15	228	-0.05	0.06	-0.9	19.25	220	1.38	0.17	7.98	7.36	
		All	347	-0.02	0.06	-0.35	19.52	353	1.32	0.12	10.68	6.87	
SO ₄ ²⁻	O ₃	subset 10	180	0.23	0.04	6.49	4.98	223	0.24	0.06	4.01	4.05	
	O ₃ > ND		n/a	n/a	n/a	n/a	n/a	152	0.19	0.1	1.95	4.48	
	O ₃	subset 15	284	0.27	0.04	6.62	4.71	302	0.18	0.05	3.98	4.16	
	O ₃ > ND		n/a	n/a	n/a	n/a	n/a	185	0.14	0.08	1.81	4.59	
	O ₃	All	464	0.27	0.03	8.77	4.69	525	0.2	0.04	5.49	4.15	
	O ₃ > ND		n/a	n/a	n/a	n/a	n/a	337	0.16	0.06	2.64	4.55	
	SO ₂	subset 10	180	0.07	0.13	0.54	6.26	222	-0.16	0.11	-1.48	5.3	
	SO ₂ > ND		81	0.18	0.19	0.91	5.73	151	0.02	0.11	0.19	4.63	
	SO ₂	subset 15	291	-0.03	0.03	-0.87	6.25	301	-0.11	0.01	-0.92	4.93	
	SO ₂ > ND		129	-0.03	0.03	-0.96	6.25	211	0.14	0.13	1.03	4.07	
	SO ₂	All	471	-0.02	0.03	-0.54	6.28	523	-0.14	0.08	-1.76	5.09	
	SO ₂ > ND		210	-0.02	0.03	-0.59	6.27	362	0.07	0.08	0.8	4.32	
	NO ₂		subset 10	175	0.08	0.05	1.51	5.51	214	0.12	0.04	2.81	3.49
			subset 15	287	-0.05	0.03	-1.77	6.65	291	0.22	0.03	6.32	2.51
			All	462	-0.01	0.02	-0.22	6.3	505	0.17	0.03	6.63	2.89
NO ₂ (no gas stove)		subset 10	119	0	0.07	0.06	6.25	132	0.41	0.07	6.01	1.11	
		subset 15	227	-0.03	0.03	-1.05	6.44	217	0.5	0.04	11.21	0.6	
		All	346	-0.02	0.03	-0.79	6.4	349	0.44	0.03	12.91	1	

Table 56. (continued)

Comparison		Cohort	Summer					Fall				
			N	β	SE	t-stat	Int.	N	β	SE	t-stat	Int.
EC	O ₃	subset 10	176	0	0.01	-0.64	1.04	223	0.02	0.01	1.43	1.15
	O ₃ > ND		n/a	n/a	n/a	n/a	n/a	152	0	0.02	0.16	1.28
	O ₃	subset 15	277	0.01	0.01	1.44	0.98	310	0.01	0.01	1.39	1.13
	O ₃ > ND		n/a	n/a	n/a	n/a	n/a	189	-0.03	0.01	-2.14	1.46
	O ₃	All	453	0	0	0.58	1	533	0.01	0.01	2.06	1.14
	O ₃ > ND		n/a	n/a	n/a	n/a	n/a	341	-0.01	0.01	-1.31	1.39
	SO ₂	subset 10	176	0	0.02	0.25	1	222	-0.03	0.02	-1.42	1.27
	SO ₂ > ND		78	-0.05	0.04	-1.3	1.24	151	0	0.03	-0.09	1.14
	SO ₂	subset 15	284	0	0	-0.32	1.03	309	-0.04	0.02	-1.7	1.21
	SO ₂ > ND		123	0	0	-0.28	1.02	215	-0.02	0.03	-0.85	1.14
	SO ₂	All	460	0	0	-0.18	1.02	531	-0.03	0.02	-2.12	1.23
	SO ₂ > ND		201	0	0	-0.55	1.04	366	-0.01	0.02	-0.5	1.14
	NO ₂	subset 10	171	0	0.01	-0.76	1.06	214	0.08	0.01	10.07	0.11
		subset 15	280	0	0	0.85	0.99	298	0.04	0.01	7.95	0.74
		All	451	0	0	0.38	1	512	0.06	0	12.38	0.51
	NO ₂ (no gas stove)	subset 10	116	-0.01	0.01	-0.67	1.1	132	0.11	0.01	12.17	0.1
subset 15		222	0	0	1.01	0.98	223	0.09	0.01	10.58	0.41	
All		338	0	0	0.57	1	355	0.09	0.01	15.96	0.29	

*bold = $p \leq 0.05$; "subset 10" includes data for subjects with personal monitoring data; "subset 15" includes data for subjects without personal monitoring data. "N" is the valid sample number; "ND" the number of samples with values below the LOD; "SD" the standard deviation; "t-stat" the t-statistic of the slope.

4.3.1.5 Associations among Personal, Indoor, and Ambient Pollutant Concentrations

4.3.1.5.1 Personal Exposures vs. Ambient Concentrations

Table 57 presents the slopes from regressions of ambient concentrations on corresponding personal exposures for PM_{2.5}, SO₄²⁻, and EC. Associations between ambient PM_{2.5} concentrations and corresponding personal exposures were strong, with high slopes and t-statistics (t-value > 13.32). The association varied slightly by season, with slopes of 0.73 (\pm 0.05) in the summer and 0.63 (\pm 0.05) in the fall. Personal-ambient SO₄²⁻ slopes were similar to those for PM_{2.5} during both seasons (summer slope = 0.74 \pm 0.02; fall slope = 0.64 \pm 0.02), with stronger associations than those found for PM_{2.5} (t-value > 26.36). The strong SO₄²⁻ associations are consistent with previous findings (Sarnat et al., 2000; Ebelt et al., 2000) and are likely due to the fact that SO₄²⁻ is a stable particle with few indoor sources. The slope describing the personal-ambient EC association for the fall (0.69 \pm 0.06) was also similar to that for total PM_{2.5} in the fall, but the slope for EC in the summer (0.33 \pm 0.10) was substantially lower than that for PM_{2.5}. The lower summertime slope suggests a lower effective penetration efficiency for EC as compared to other particle measures in the summer. Reasons for this lower association are unclear; however, it should be noted that greater noise in the personal and ambient EC measurements during the summer likely decreased the strength of the summertime

EC association. Summertime EC measurements showed a very high field LOD, which was approximately 50% as great as mean EC exposures, which likely contributed to the low t-statistic of the personal-ambient EC association during the summer (t-value = 3.24) as compared to the fall (t-value = 12.47).

For each of the measured gases, slopes of personal-ambient regressions were statistically significant in both seasons, with the exception of summertime SO₂. For O₃ and NO₂, fall slopes were almost two times as great as summer slopes. The personal-ambient O₃ slope was 0.15 (±0.02) in the summer and 0.27 (±0.03) in the fall. For NO₂, the summer and fall slopes were 0.25 (± 0.06) and 0.49 (± 0.05), respectively. For all gases in both seasons, however, slopes were generally lower than those found for particles, particularly for SO₂.

4.3.1.5.2 Indoor Concentrations vs. Ambient Concentrations

For all participants, indoor-ambient slopes were greater in the summer than in the fall for PM_{2.5}, SO₄²⁻, and SO₂. Slopes for EC, O₃, and NO₂ were greater in the fall than in the summer, with the largest seasonal differences observed for EC and NO₂ (slopes approximately three times greater in fall than in summer). Similar seasonal variability in the relationships between indoor and ambient concentrations was found for all measured pollutants when the data set was limited to individuals participating in personal monitoring, although seasonal differences in the indoor-ambient associations were even greater for EC and NO₂ when this subset of participants was used. The association between indoor and ambient NO₂ concentrations followed a similar seasonal trend for individuals living in homes with and without gas stoves; however, as shown by the lower and less-significant slopes, the indoor-ambient NO₂ association was weaker for individuals living in homes with gas stoves than for those living in homes without gas stoves.

When compared to the corresponding associations between personal exposures and ambient concentrations, the associations between indoor and ambient concentrations (for individuals with simultaneous personal monitoring - subset 10) were slightly greater for PM_{2.5}, SO₄²⁻, EC (fall), O₃ (summer), and NO₂ (all homes in fall and homes without gas stoves) (Table 57). Slopes were similar, however, when they were normalized to mean personal or indoor levels. Similar normalized slopes suggest that ambient pollutants predicted personal and indoor levels equally well and suggest a high influence of housing factors on personal exposures. The greater slopes for indoor-ambient as compared to personal-ambient associations for PM_{2.5} and SO₄²⁻ suggest that individuals spent time in other environments for which the indoor-ambient slopes were lower.

4.3.1.5.3 Personal Exposures vs. Indoor Concentrations

All slopes describing personal-indoor associations were significant except for those for SO₂ and NO₂ (in homes with gas stoves) in the summer season. In both seasons, slopes describing personal-indoor associations tended to be greater than those describing personal-ambient associations. For example, slopes and t-statistics (t-value > 19.44) describing the relationship between personal and indoor PM_{2.5} were higher when compared to personal-ambient values (Table 57). Similarly, personal-indoor SO₄²⁻ and EC slopes were greater than corresponding personal-ambient SO₄²⁻; the strengths of the personal-indoor associations for SO₄²⁻ (t-value > 30.6) were even greater than those for PM_{2.5}. These high and significant slopes for the particulate measures indicate that indoor concentrations are strong proxies of corresponding personal particulate exposure.

Table 57. Comparison of personal-ambient, personal-indoor, and indoor-ambient relationships (by season) for the older adult cohort via mixed models.*

Pollutant	Cohort**	Comparison	Summer					Fall					
			N	β	SE	t-stat	Int.	N	β	SE	t-stat	Int.	
PM _{2.5}	subset 10	personal vs. indoor	155	0.83	0.03	29.28	2.80	193	0.73	0.04	19.44	5.56	
		personal vs. ambient	167	0.73	0.05	16.08	4.83	204	0.63	0.05	13.32	7.42	
	subset 15	indoor vs. ambient	176	0.83	0.04	21.97	2.97	216	0.70	0.05	14.70	5.68	
			288	0.69	0.02	30.08	4.82	297	0.60	0.05	11.44	8.08	
			464	0.75	0.02	36.35	4.05	513	0.64	0.04	17.65	7.05	
SO ₄ ²⁻	subset 10	personal vs. indoor	156	0.84	0.03	30.60	0.52	183	0.88	0.02	58.36	0.03	
		personal vs. ambient	150	0.74	0.02	32.35	-0.10	188	0.64	0.02	26.36	0.45	
	subset 15	indoor vs. ambient	168	0.86	0.03	31.58	-0.59	223	0.76	0.02	31.77	0.39	
			268	0.68	0.02	38.30	0.68	303	0.69	0.03	27.38	0.48	
			436	0.75	0.02	47.79	0.18	526	0.72	0.02	40.64	0.43	
EC	subset 10	personal vs. indoor	137	0.64	0.08	7.90	0.41	191	0.75	0.02	34.89	0.27	
		personal vs. ambient (> ND ^s)	146	0.33	0.10	3.24	0.74	197	0.69	0.06	12.47	0.44	
	subset 15	indoor vs. ambient (> ND)	156	0.23	0.07	3.11	0.74	223	0.83	0.06	12.93	0.32	
			247	0.26	0.05	4.83	0.73	311	0.67	0.04	16.58	0.42	
			403	0.24	0.04	5.65	0.73	534	0.74	0.04	20.40	0.38	
O ₃	subset 10	personal vs. indoor	180	0.64	0.05	12.65	1.49	117	0.58	0.08	7.28	2.68	
		personal vs. ambient	174	0.15	0.02	7.21	0.88	226	0.27	0.03	8.64	-0.30	
	subset 15	indoor vs. ambient	180	0.17	0.02	7.59	1.31	228	0.23	0.03	7.46	0.79	
			282	0.17	0.02	8.87	0.91	320	0.29	0.03	10.16	-0.92	
			462	0.17	0.01	11.66	1.07	548	0.26	0.02	12.64	-0.25	
SO ₂	subset 10	personal vs. indoor	56	0.34	0.22	1.54	2.74	131	0.49	0.05	9.52	0.55	
		personal vs. ambient	179	0.14	0.06	2.16	1.17	227	0.10	0.01	6.92	0.25	
		personal vs. ambient (> ND)	106	0.03	0.10	0.29	2.05	153	0.08	0.02	4.98	0.49	
		indoor vs. ambient	184	0.15	0.05	3.03	1.13	226	0.09	0.02	5.26	0.66	
	subset 15	indoor vs. ambient (> ND)	111	0.07	0.07	1.00	1.71	152	0.08	0.02	3.35	0.92	
			291	0.16	0.11	1.43	1.81	318	0.02	0.01	1.73	0.74	
		all	indoor vs. ambient (> ND)	173	0.03	0.16	0.20	2.62	216	-0.01	0.01	-0.88	1.31
				475	0.15	0.07	2.18	1.55	544	0.05	0.01	4.84	0.71
284	0.05	0.11	0.46	2.27	368	0.02	0.01	1.87	1.16				

Table 57. (continued)

Pollutant	Cohort**	Comparison	Summer					Fall				
			N	β	SE	t-stat	Int.	N	β	SE	t-stat	Int.
NO ₂	subset 10	personal (all) vs. indoor	174	0.23	0.06	3.78	7.45	218	0.66	0.04	17.48	3.33
		personal (all) vs. ambient	173	0.23	0.05	4.67	7.62	227	0.39	0.05	7.56	7.72
	subset 15	indoor (all) vs. ambient	175	0.22	0.06	3.74	9.42	218	0.43	0.05	8.11	8.43
			282	0.09	0.06	1.51	9.10	306	0.38	0.04	8.52	6.55
	all	457	0.14	0.04	3.28	9.19	524	0.40	0.03	11.76	7.30	
	subset 10	personal (no gas stove) vs. indoor	125	0.22	0.07	2.96	7.01	135	0.72	0.06	11.94	2.49
		personal (no gas stove) vs. ambient	122	0.25	0.06	4.30	6.45	138	0.49	0.05	10.09	4.27
	subset 15	indoor (no gas stove) vs. ambient	121	0.32	0.07	4.76	6.65	135	0.59	0.04	15.28	3.57
			226	0.09	0.07	1.21	8.39	230	0.35	0.03	11.37	4.33
	all	347	0.17	0.05	3.13	7.77	365	0.44	0.02	17.64	4.02	
	subset 10	personal (gas stove) vs. indoor	49	0.17	0.11	1.50	10.00	83	0.60	0.06	10.11	4.72
		personal (gas stove) vs. ambient	51	0.19	0.09	2.03	10.25	89	0.25	0.10	2.39	12.84
	subset 15	indoor (gas stove) vs. ambient	54	0.02	0.11	0.21	15.01	83	0.18	0.12	1.56	15.88
			56	0.07	0.10	0.78	12.12	76	0.44	0.15	3.02	12.62
	all	110	0.05	0.07	0.63	13.56	159	0.31	0.09	3.33	14.31	

* Tobit model results were similar to mixed model results; bold numbers indicate p-values less than or equal to 0.05

** "subset 10" – indoor concentrations for individuals with personal monitoring data; "subset 15" – indoor concentrations for individuals without personal monitoring data; "all" – full cohort; § "ND" – non-detectable concentrations.

In contrast, the slopes describing personal-indoor gas associations were lower than those describing personal-indoor particle associations, especially for NO₂ and SO₂ during summertime. These results suggest that indoor concentrations are poorer proxies of corresponding personal exposures for gases than for particles.

4.3.1.5.4 Cross-Pollutant Associations between Personal Exposures and Ambient Concentrations

Table 58 shows results from cross-pollutant analyses examining associations between ambient particle concentrations and personal gas exposures. Associations between ambient PM_{2.5} concentrations and personal gas exposures were significant for O₃ in both seasons and for NO₂ in the fall. While significant, however, the slopes for the associations were quite low (slopes < 0.17) and indicate that 24-hr personal O₃ and NO₂ exposures increased on average by only 1.1 and 1.7 ppb with every 10 µg/m³ increase in ambient PM_{2.5}. Associations were also significant between the specific ambient particle components and personal O₃ and NO₂ exposures. Ambient particles were not significant predictors of personal SO₂ levels.

Table 59 shows results from cross-pollutant analyses examining associations between ambient gas concentrations and personal particle exposures. Associations between ambient O₃ concentrations and personal particle exposures were significant in the summer, although the slopes and R² values were low (R² < 0.16). Associations between ambient NO₂ concentrations and personal particle exposures were significant in the fall, in particular for EC (t-value = 13.6, R² = 0.49). The slopes for the associations with ambient NO₂ were moderate, suggesting that 24-hour personal exposures to PM_{2.5} increased by 9.3 µg/m³ for each 10 ppb increase in ambient NO₂.

Table 58. Associations between personal gas exposures and ambient particle concentrations for the older adult cohort, by season.*

Comparison		Summer					Fall				
		N	β	SE	t-stat	R ²	N	β	SE	t-stat	R ²
O ₃	PM _{2.5}	181	0.11	0.03	3.46	0.06	226	0.1	0.02	4.24	0.07
NO ₂ **		128	-0.01	0.05	-0.24	0.00	139	0.17	0.03	5.82	0.21
SO ₂		183	-0.0004	0.03	-0.02	0.00	228	0.0005	0.01	0.05	0.00
O ₃	SO ₄ ²⁻	168	0.16	0.06	2.58	0.04	226	0.27	0.06	4.42	0.08
NO ₂ **		118	-0.09	0.10	-0.86	0.01	139	0.34	0.08	4.14	0.12
SO ₂		169	-0.06	0.05	-1.22	0.01	228	0.007	0.03	0.27	0.00
O ₃	EC	154	-0.81	0.64	-1.28	0.01	222	1.27	0.44	2.92	0.04
NO ₂ **		107	1.81	0.91	1.99	0.03	136	3.71	0.51	7.32	0.32
SO ₂		157	0.59	0.52	1.14	0.01	224	-0.11	0.20	-0.57	0.00

*bold numbers indicate p-values ≤ 0.05

**models predicting NO₂ exposures restricted to subjects residing in homes without gas stoves

Table 59. Associations between personal particle exposures and ambient gas concentrations for the older adult cohort, by season.*

Comparison		Summer					Fall				
		N	β	SE	t-stat	R ²	N	β	SE	t-stat	R ²
PM _{2.5}	O ₃	159	0.28	0.05	5.46	0.16	204	0.08	0.10	0.78	0.00
	NO ₂	159	-0.07	0.09	-0.80	0.00	203	0.93	0.11	8.25	0.25
	SO ₂ **	95	0.73	0.27	2.70	0.07	136	0.18	0.11	1.60	0.02
SO ₄ ²⁻	O ₃	155	0.14	0.02	5.56	0.16	188	0.01	0.03	0.49	0.00
	NO ₂	155	-0.06	0.04	-1.55	0.01	187	0.28	0.04	7.78	0.27
	SO ₂ **	93	0.21	0.12	1.70	0.03	125	0.07	0.03	2.48	0.06
EC	O ₃	157	-0.01	0.004	-2.60	0.04	197	-0.02	0.006	-3.00	0.04
	NO ₂	157	0.02	0.006	3.45	0.07	196	0.08	0.006	13.60	0.49
	SO ₂ **	92	0.02	0.02	0.88	0.01	135	0.02	0.008	2.47	0.05

*bold numbers indicate p-values ≤ 0.05

**models using ambient SO₂ restricted to data greater than the analytical LOD

4.3.1.6 Factors Affecting Relationships

Mixed model results showing the effects of activity patterns (i.e., time spent outdoors, time spent in transit) and ventilation conditions (i.e., presence of open windows, AC use) on associations between ambient pollutant concentrations and personal exposures are presented in Tables 60 – 67. Time spent outdoors generally increased the personal-ambient slopes, except for SO₄²⁻ in summer and PM_{2.5} and EC in fall. The overall effect of this variable, however, was slight. For example, the only significant increase in slope was observed for NO₂ in the fall, which was attributed primarily to individuals living in homes with gas stoves. Further, with the exception of the personal-ambient association for EC in summertime, all significant associations between personal exposures and corresponding ambient concentrations remained significant even when individuals spent no time outdoors. Time spent in transit tended to decrease personal-ambient associations for O₃ in the summer and for SO₄²⁻ and SO₂ in fall. Weaker associations for O₃ could arise because its concentrations are lower near traffic; however, this theory is not supported by the similarly weaker associations that were observed for the traffic-related pollutants EC and NO₂. The weaker associations for SO₄²⁻ and SO₂ are unexpected and may be spurious due to the relatively low fraction of time individuals spent in transit.

Home ventilation was an important modifying factor for many of the personal-ambient relationships, with the greatest slopes and strongest associations observed for subjects spending time indoors with open windows. The influence of home ventilation was particularly evident in the summer for SO₄²⁻ and for O₃. The slope of the regression line of ambient O₃ concentrations on corresponding personal O₃ exposures for individuals spending time in indoor environments with open windows (slope = 0.18 ± 0.03 , t-value = 7.34), for example, was twice that of individuals spending no time indoors with open windows (slope = 0.08 ± 0.04 , t-value = 1.89). The stronger associations and higher slopes during conditions in which homes were well ventilated was probably due to the fact that O₃, a reactive pollutant, could penetrate indoors more efficiently during these conditions.

The effects of ventilation on the associations between indoor and corresponding ambient pollutant concentrations (Tables 68 - 75) were similar to its effects on associations between

personal exposures and ambient concentrations, as the high ventilation category (either any windows use or no AC use) resulted in increased indoor-ambient slopes and increased statistical significance relative to the corresponding low ventilation category. The differences between levels of each ventilation factor, however, were more striking for the indoor-ambient associations than for the personal-ambient associations; statistically significant differences between the levels of each factor were observed more frequently for the indoor-ambient associations. For NO₂ in the fall, ventilation effects were more apparent for indoor-ambient associations than for corresponding personal-ambient associations (especially when excluding gas stove homes). Stronger effects of ventilation on indoor-ambient associations as compared to personal-ambient associations may be explained by the fact that individuals do not spend all of their time indoors at home.

Table 60. Personal vs. ambient PM_{2.5} concentrations by activity and proximity to ventilation for the older adult cohort.*

Modifier	Status	Summer				Fall			
		β	SE	t-stat	Int.	β	SE	t-stat	Int.
Activity Pattern									
Time Spent Outdoors	none	0.69	0.07	10.22	5.81	0.63	0.05	11.70	9.06
	any	0.77	0.06	12.35	2.92	0.62	0.10	5.97	8.97
Time Spent in Transit	none	0.72	0.07	10.04	4.78	0.69	0.07	9.92	7.65
	any	0.73	0.06	12.27	4.03	0.57	0.06	8.83	10.35
Ventilation									
Open Windows	none [†]	0.59	0.12	5.14	3.82	0.53	0.07	7.22	9.29
	any	0.76	0.05	15.39	4.34	0.65	0.06	10.14	9.96
Air Conditioner	none	0.84	0.06	13.57	3.34	0.69	0.06	10.87	7.86
	any	0.60	0.07	9.08	6.22	0.54	0.07	7.35	11.07

*shaded cells represent significant interaction effect; bold numbers indicate p-values ≤ 0.05

[†]for summer season only: none = subjects spend < 33% of time; any = subjects spend > 33% of time

Table 61. Personal vs. ambient SO₄²⁻ concentrations by activity and proximity to ventilation for the older adult cohort.*

Modifier	Status	Summer				Fall			
		β	SE	t-stat	Int.	β	SE	t-stat	Int.
Activity Pattern									
Time Spent Outdoors	none	0.80	0.03	24.78	-0.47	0.63	0.03	22.57	0.02
	any	0.68	0.03	20.99	0.21	0.68	0.05	13.97	0.02
Time Spent in Transit	none	0.71	0.04	18.91	0.10	0.75	0.04	19.69	-0.45
	any	0.75	0.03	25.76	-0.33	0.58	0.03	19.26	0.37
Ventilation									
Open Windows	none [†]	0.51	0.06	8.32	1.40	0.57	0.04	14.86	0.11
	any	0.77	0.02	32.81	-0.36	0.67	0.03	21.31	0.12
Air Conditioner	none	0.80	0.03	28.82	-0.56	0.68	0.03	20.83	-0.12
	any	0.62	0.04	16.75	0.62	0.60	0.04	16.62	0.13

*shaded cells represent significant interaction effect; bold numbers indicate p-values ≤ 0.05

[†]for summer season only: none = subjects spend < 33% of time; any = subjects spend > 33% of time

Table 62. Personal vs. ambient EC (> ND**) concentrations by activity and proximity to ventilation for the older adult cohort.*

Modifier	Status	Summer				Fall			
		β	SE	t-stat	Int.	β	SE	t-stat	Int.
Activity Pattern									
Time Spent Outdoors	none	0.26	0.16	1.60	0.73	0.72	0.06	11.31	0.40
	any	0.33	0.13	2.55	0.75	0.62	0.12	5.04	0.47
Time Spent in Transit	none	0.41	0.14	3.04	0.60	0.83	0.09	9.61	0.37
	any	0.18	0.15	1.14	0.86	0.62	0.07	8.31	0.46
Ventilation									
Open Windows	none [†]	0.14	0.19	0.72	0.65	0.66	0.08	8.74	0.41
	any	0.40	0.12	3.39	0.67	0.73	0.08	8.87	0.42
Air Conditioner	none	0.36	0.14	2.50	0.74	0.69	0.07	9.62	0.35
	any	0.28	0.14	1.97	0.69	0.66	0.09	7.34	0.58

*shaded cells represent significant interaction effect; bold numbers indicate p-values ≤ 0.05

**ND – non detectable concentrations

[†]for summer season only: none = subjects spend < 33% of time; any = subjects spend > 33% of time

Table 63. Personal vs. ambient O₃ concentrations by activity and proximity to ventilation for the older adult cohort.*

Modifier	Status	Summer				Fall			
		β	SE	t-stat	Int.	β	SE	t-stat	Int.
Activity Pattern									
Time Spent Outdoors	none	0.12	0.03	3.70	0.72	0.23	0.04	6.24	-0.38
	any	0.18	0.03	6.24	-0.18	0.35	0.06	6.09	-1.84
Time Spent in Transit	none	0.19	0.03	6.55	-1.39	0.29	0.04	7.60	-1.70
	any	0.11	0.03	3.42	1.90	0.22	0.05	4.14	0.27
Ventilation									
Open Windows	none [†]	0.08	0.04	1.89	1.73	0.20	0.05	3.90	-1.17
	any	0.18	0.03	7.34	-0.54	0.27	0.04	7.38	0.52
Air Conditioner	none	0.25	0.03	8.79	-1.82	0.24	0.04	6.14	-1.35
	any	0.06	0.03	2.06	2.60	0.30	0.05	6.25	0.02

*shaded cells represent significant interaction effect; bold numbers indicate p-values ≤ 0.05

[†]for summer season only: none = subjects spend < 33% of time; any = subjects spend > 33% of time

Table 64. Personal vs. ambient SO₂ concentrations by activity and proximity to ventilation for the older adult cohort.*

Modifier	Status	Summer				Fall			
		β	SE	t-stat	Int.	β	SE	t-stat	Int.
Activity Pattern									
Time Spent Outdoors	none	0.09	0.09	1.07	0.74	0.09	0.02	5.75	0.14
	any	0.19	0.10	1.92	1.32	0.14	0.04	3.88	0.05
Time Spent in Transit	none	0.09	0.08	1.04	1.21	0.20	0.03	7.43	-0.13
	any	0.20	0.11	1.83	0.83	0.06	0.02	3.87	0.25
Ventilation									
Open Windows	none [†]	0.16	0.10	1.55	0.91	0.08	0.02	5.00	0.37
	any	0.11	0.09	1.26	1.09	0.15	0.03	4.68	-0.14
Air Conditioner	none	0.18	0.10	1.69	0.92	0.10	0.02	6.62	0.20
	any	0.10	0.09	1.13	1.10	0.06	0.03	1.87	0.13

*shaded cells represent significant interaction effect; bold numbers indicate p-values ≤ 0.05

[†]for summer season only: none = subjects spend < 33% of time; any = subjects spend > 33% of time

Table 65. Personal vs. ambient SO₂ (> ND**) concentrations by activity and proximity to ventilation for the older adult cohort.*

Modifier	Status	Summer				Fall			
		β	SE	t-stat	Int.	β	SE	t-stat	Int.
Activity Pattern									
Time Spent Outdoors	none	-0.03	0.14	-0.22	1.77	0.08	0.02	4.18	0.31
	any	0.10	0.17	0.59	1.89	0.15	0.04	3.35	-0.26
Time Spent in Transit	none	0.01	0.13	0.05	2.01	0.22	0.03	6.64	-0.52
	any	0.03	0.18	0.15	1.75	0.05	0.02	2.58	0.59
Ventilation									
Open Windows	none [†]	0.07	0.15	0.46	2.01	0.07	0.02	3.90	0.48
	any	-0.06	0.15	-0.39	2.13	0.13	0.04	3.15	-0.14
Air Conditioner	none	-0.03	0.16	-0.18	2.08	0.10	0.02	5.37	0.22
	any	0.05	0.15	0.32	1.76	0.01	0.05	0.26	0.67

*shaded cells represent significant interaction effect; bold numbers indicate p-values ≤ 0.05

**ND – non detectable concentrations

[†]for summer season only: none = subjects spend < 33% of time; any = subjects spend > 33% of time

Table 66. Personal vs. ambient NO₂ concentrations by activity and proximity to ventilation for the older adult cohort.*

Modifier	Status	Summer				Fall			
		β	SE	t-stat	Int.	β	SE	t-stat	Int.
Activity Pattern									
Time Spent Outdoors	none	0.16	0.07	2.52	8.69	0.33	0.06	5.63	9.77
	any	0.32	0.07	4.29	8.41	0.57	0.11	5.28	8.49
Time Spent in Transit	none	0.24	0.07	3.43	9.10	0.38	0.07	5.36	9.41
	any	0.20	0.07	2.85	8.33	0.39	0.08	5.24	9.29
Ventilation									
Open Windows	none [†]	0.25	0.11	2.32	7.33	0.52	0.08	6.60	7.55
	any	0.23	0.06	4.05	8.88	0.24	0.07	3.43	11.75
Air Conditioner	none	0.18	0.07	2.42	10.18	0.41	0.07	6.02	9.23
	any	0.26	0.07	3.94	7.70	0.36	0.08	4.30	9.50

*shaded cells represent significant interaction effect; bold numbers p-values ≤ 0.05;

[†]for summer: none = subjects spend <33% of time; any = subjects spend > 33% of time

Table 67. Personal vs. ambient NO₂ concentrations by activity and proximity to ventilation for individuals without gas stoves for the older adult cohort.*

Modifier	Status	Summer				Fall			
		β	SE	t-stat	Int.	β	SE	t-stat	Int.
Activity Pattern									
Time Spent Outdoors	none	0.18	0.07	2.45	5.53	0.45	0.06	7.79	4.08
	any	0.34	0.09	3.87	5.66	0.51	0.09	6.00	5.12
Time Spent in Transit	none	0.24	0.07	3.50	5.78	0.48	0.06	8.32	3.76
	any	0.27	0.11	2.31	5.23	0.54	0.08	6.43	4.85
Ventilation									
Open Windows	none [†]	0.24	0.11	2.26	3.94	0.44	0.07	6.83	3.84
	any	0.27	0.07	3.88	5.77	0.46	0.07	6.15	5.86
Air Conditioner	none	0.34	0.11	3.04	5.55	0.45	0.06	8.04	4.50
	any	0.23	0.07	3.39	4.92	0.57	0.10	5.65	3.71

*Bold numbers indicate p-values ≤ 0.05;

[†]for summer: none = subjects spend <33% of time; any = subjects spend > 33% of time

Table 68. Indoor vs. ambient PM_{2.5} concentrations by ventilation conditions for the older adult cohort.*

Modifier	Status	Summer				Fall			
		β	SE	t-stat	Int.	β	SE	t-stat	Int.
Open Windows	Lo**	0.54	0.05	10.95	7.89	0.66	0.06	11.22	6.71
	Med	0.64	0.04	16.93	5.71	0.56	0.06	8.63	8.52
	Hi	0.86	0.03	32.04	2.53	0.7	0.07	9.74	6.85
AC	None	0.84	0.05	18.48	3.12	0.65	0.04	17.03	6.73
	Any	0.71	0.03	21.98	5.35	0.46	0.11	4.31	12.77

*shaded cells represent significant interaction effect; bold numbers indicate p-values ≤ 0.05

**Lo = no windows open, Med = the sum of open windows up to 24-hours, Hi = the sum of open windows > 24-hours

Table 69. Indoor vs. ambient SO₄²⁻ concentrations by ventilation conditions for the older adult cohort.*

Modifier	Status	Summer				Fall			
		β	SE	t-stat	Int.	β	SE	t-stat	Int.
Open Windows	Lo**	0.52	0.04	14.41	1.12	0.66	0.03	23.75	0.03
	Med	0.66	0.03	23.17	0.7	0.61	0.03	22.26	0.69
	Hi	0.86	0.02	45.16	-0.28	0.9	0.03	26.01	-0.49
AC	None	0.9	0.03	28.34	-0.57	0.71	0.02	37.06	0.08
	Any	0.74	0.02	30.9	0.38	0.72	0.05	14.8	-0.24

*shaded cells represent significant interaction effect; bold numbers indicate p-values ≤ 0.05

**Lo = no windows open, Med = the sum of open windows up to 24-hours, Hi = the sum of open windows > 24-hours

Table 70. Indoor vs. ambient EC (> ND**) concentrations by ventilation conditions for the older adult cohort.*

Modifier	Status	Summer				Fall			
		β	SE	t-stat	Int.	β	SE	t-stat	Int.
Open Windows	Lo***	0.19	0.09	2.04	0.61	0.71	0.05	13.4	0.32
	Med	0.14	0.09	1.6	0.75	0.67	0.07	10.07	0.54
	Hi	0.3	0.06	5.12	0.72	0.96	0.08	11.53	0.12
AC	None	0.4	0.07	5.81	0.59	0.76	0.04	19.52	0.32
	Any	0.32	0.04	7.32	0.58	0.71	0.14	5.11	0.48

*shaded cells represent significant interaction effect; bold numbers indicate p-values ≤ 0.05

**ND – non detectable concentrations

***Lo = no windows open, Med = the sum of open windows up to 24-hours, Hi = the sum of open windows > 24-hours

Table 71. Indoor vs. ambient O₃ concentrations by ventilation conditions for the older adult cohort.*

Modifier	Status	Summer				Fall			
		β	SE	t-stat	Int.	β	SE	t-stat	Int.
Open Windows	Lo**	0.07	0.03	2.3	2.32	0.18	0.03	5.54	-0.32
	Med	0.14	0.03	4.41	0.47	0.3	0.04	7.18	-0.82
	Hi	0.22	0.02	12.01	0.42	0.31	0.03	9.49	0.24
AC	None	0.27	0.03	7.88	-0.38	0.26	0.02	11.22	-0.67
	Any	0.17	0.02	8.03	0.8	0.27	0.05	5.37	0.56

*shaded cells represent significant interaction effect; bold numbers indicate p-values ≤ 0.05

**Lo = no windows open, Med = the sum of open windows up to 24-hours, Hi = the sum of open windows > 24-hours

Table 72. Indoor vs. ambient SO₂ concentrations by ventilation conditions for the older adult cohort.*

Modifier	Status	Summer				Fall			
		β	SE	t-stat	Int.	β	SE	t-stat	Int.
Open Windows	Lo**	0.12	0.13	0.9	0.44	0.07	0.02	4.95	0.63
	Med	0.18	0.16	1.17	0.81	0.02	0.02	1.4	0.47
	Hi	0.18	0.11	1.65	2	0.12	0.04	3.09	0.17
AC	None	0.1	0.18	0.56	0.79	0.06	0.01	5.23	0.58
	Any	0.05	0.13	0.37	1.74	0.01	0.06	0.09	0.18

*shaded cells represent significant interaction effect; bold numbers indicate p-values ≤ 0.05

**Lo = no windows open, Med = the sum of open windows up to 24-hours, Hi = the sum of open windows > 24-hours

Table 73. Indoor vs. ambient SO₂ (> ND**) concentrations by ventilation conditions for the older adult cohort.*

Modifier	Status	Summer				Fall			
		β	SE	t-stat	Int.	β	SE	t-stat	Int.
Open Windows	Lo***	0.14	0.19	0.74	0.49	0.05	0.02	2.82	0.94
	Med	0.03	0.23	0.12	2	0	0.02	0.18	0.89
	Hi	0.01	0.18	0.07	3.11	0.08	0.05	1.53	0.43
AC	None	-0.09	0.27	-0.32	1.81	0.04	0.01	2.59	0.89
	Any	-0.3	0.21	-1.43	3.3	-0.08	0.09	-0.9	0.99

*shaded cells represent significant interaction effect; bold numbers indicate p-values ≤ 0.05

**ND – non detectable concentrations

***Lo = no windows open, Med = the sum of open windows up to 24-hours, Hi = the sum of open windows > 24-hours

Table 74. Indoor vs. ambient NO₂ concentrations by ventilation conditions for the older adult cohort.*

Modifier	Status	Summer				Fall			
		β	SE	t-stat	Int.	β	SE	t-stat	Int.
Open Windows	Lo**	0.26	0.1	2.64	8.74	0.38	0.06	6.79	6.48
	Med	0.16	0.09	1.81	8.58	0.34	0.06	5.22	8.85
	Hi	0.13	0.06	1.98	9.97	0.32	0.07	4.86	10.32
AC	None	0.16	0.07	2.29	8.9	0.45	0.04	12.24	7.03
	Any	0.14	0.05	3.09	9.16	-0.03	0.11	-0.29	13.19

*shaded cells represent significant interaction effect; bold numbers indicate p-values ≤ 0.05

**Lo = no windows open, Med = the sum of open windows up to 24-hours, Hi = the sum of open windows > 24-hours

Table 75. Indoor vs. ambient NO₂ concentrations by ventilation conditions for homes without gas stoves for the older adult cohort.*

Modifier	Status	Summer				Fall			
		β	SE	t-stat	Int.	β	SE	t-stat	Int.
Open Windows	Lo**	0.34	0.12	2.91	5.57	0.33	0.04	9.34	3.02
	Med	0.15	0.09	1.62	7.07	0.4	0.04	9.49	3.43
	Hi	0.15	0.08	1.82	7.33	0.53	0.05	9.68	3.77
AC	None	0.21	0.09	2.43	6.38	0.44	0.03	17.21	2.68
	Any	0.19	0.05	3.98	5.16	0.24	0.12	1.93	6.17

*shaded cells represent significant interaction effect; bold numbers indicate p-values ≤ 0.05

**Lo = no windows open, Med = the sum of open windows up to 24-hours, Hi = the sum of open windows > 24-hours

In addition to home ventilation, we also examined whether stove type modified the association between ambient pollutant concentrations and both indoor concentrations and personal exposures. Results for indoor-ambient concentration comparisons are presented in Table 76. The intercepts for models of indoor NO₂ as a function of ambient NO₂ were significantly higher in both seasons for participants living in homes with gas as compared to electric stoves; this was expected, because gas stoves are known sources of NO₂ indoors. Stove type did not modify the indoor-ambient intercepts for the other measured pollutants, as significant differences in the intercepts for gas as compared to electric stoves were not found. Stove type was found to modify the indoor-ambient slopes for PM_{2.5} and SO₄²⁻ in the summer and EC and O₃ in the fall, with this effect generally resulting in greater slopes for homes with gas stoves as compared to homes with electric stoves (except for O₃ in the fall, which exhibited the opposite trend). Greater slopes suggest greater effective penetration efficiencies of ambient pollutants; however, it is not clear how stove type would impact the effective penetration efficiency of any pollutant. It is possible that the results reflect the effect of different buildings rather than the effect of stove type, as stove type and place of residence were directly correlated. Electric stoves were present only in residences in the Gaylord and Kennedy buildings, whereas gas stoves were present only in the Elmer White residences and in private homes. As a result, it is not possible to separate the effects of cooking fuel use from those of building of residence.

Table 76. Indoor vs. ambient pollutant concentrations by stove type for the older adult cohort.*

Pollutant	Fuel	Summer					Fall				
		N	β	SE	t-stat	Int.	N	β	SE	t-stat	Int.
PM _{2.5}	gas	118	0.82	0.04	20.38	2.84	158	0.69	0.06	10.71	8.04
	electric	351	0.72	0.02	30.32	4.51	355	0.62	0.04	14.04	6.59
SO ₄ ²⁻	gas	112	0.86	0.03	28.22	-0.22	171	0.71	0.03	23.40	0.33
	electric	345	0.72	0.02	39.89	0.34	355	0.73	0.02	33.19	0.47
EC	gas	116	0.15	0.08	1.96	0.87	173	0.87	0.06	14.53	0.33
	electric	348	0.15	0.04	3.65	0.86	361	0.66	0.05	14.67	0.41
O ₃	gas	113	0.15	0.03	4.62	2.80	174	0.19	0.04	4.68	0.71
	electric	357	0.17	0.17	10.73	0.55	374	0.29	0.02	11.96	-0.60
SO ₂	gas	116	0.30	0.17	1.77	1.24	174	0.02	0.02	1.12	0.81
	electric	363	0.12	0.08	1.57	1.64	373	0.06	0.01	4.95	0.67
NO ₂	gas	113	0.04	0.09	0.49	13.64	159	0.31	0.06	5.21	14.32
	electric	357	0.17	0.05	3.37	7.76	368	0.44	0.04	10.70	4.02

*shaded cells represent significant interaction effect; bold numbers indicate p-value ≤ 0.05

As shown in Table 77, stove type also modified associations between personal pollutant exposures and corresponding ambient concentrations. As with indoor-ambient associations, intercepts for models of personal NO₂ as a function of ambient NO₂ were significantly higher in the fall for participants living in homes with gas stoves as compared to those living in homes with electric stoves. Similar trends were observed in the summer, although the intercepts did not differ significantly by stove type. Again, higher NO₂ intercepts for individuals with gas stoves likely resulted from the fact that gas stoves are major sources of NO₂ indoors. Slopes of personal-ambient associations were also significantly modified by stove type; however, this modification was inconsistent and did not generally agree with that observed for indoor-ambient concentration comparisons. Inconsistent personal-ambient and indoor-ambient results suggest that results are highly dependent on the participants, as individuals with personal monitoring data include only a subset of individuals for whom indoor concentrations were measured.

Table 77. Personal vs. ambient pollutant concentrations by stove type for the older adult cohort.*

Pollutant	Fuel	Summer					Fall				
		N	β	SE	t-stat	Int.	N	β	SE	t-stat	Int.
PM _{2.5}	Gas	48	0.68	0.08	8.19	5.27	82	0.58	0.07	8.21	9.67
	electric	121	0.75	0.05	13.78	4.61	122	0.68	0.06	10.52	5.82
SO ₄ ²⁻	Gas	49	0.81	0.04	21.34	-0.43	77	0.56	0.03	17.11	0.40
	electric	116	0.70	0.03	25.23	0.11	111	0.72	0.03	21.26	0.36
EC	Gas	49	0.28	0.19	1.51	0.66	78	0.69	0.08	8.56	0.45
	electric	117	0.29	0.09	3.16	0.85	119	0.69	0.08	8.94	0.43
O ₃	Gas	53	0.07	0.05	1.43	3.40	89	0.19	0.05	3.54	0.36
	electric	130	0.18	0.02	7.44	0.15	137	0.31	0.04	8.10	-0.57
SO ₂	Gas	54	0.59	0.18	3.30	0.72	89	0.05	0.02	2.36	0.15
	electric	131	0.08	0.07	1.13	1.24	139	0.14	0.02	7.60	0.28
NO ₂	Gas	53	0.17	0.08	2.07	10.40	89	0.25	0.08	3.16	12.85
	electric	130	0.26	0.06	4.20	6.39	139	0.49	0.07	7.31	4.28

*shaded cells represent significant interaction effect; bold numbers indicate p-value ≤ 0.05

Cooking and cleaning were additionally tested in the personal-ambient and indoor-ambient models. Interaction terms with these factors were not significant; these activities were not expected to vary the slopes of the associations but only possibly to increase the intercepts of the associations. Indeed, when included as main effects only, the overall trends were for cooking and cleaning to cause very small increases in intercepts; significant differences in indoor concentrations between any cooking and no cooking were found in the fall for indoor PM_{2.5} and NO₂. However, the presence of these factors, whether significant or not, did not affect the t-statistics of the indoor-ambient or personal-ambient slopes.

4.3.1.7 Summary of Elemental Data

Table 78 summarizes water-extractable element concentrations measured as part of the adult cohort exposure assessment. As shown in the table, water-extractable concentrations of many elements were below their field limits of detection in personal, indoor, outdoor, and ambient environments. For eight elements, including As, Ba, Cd, Pb, Mg, K, Se, and V, water-extractable concentrations were above the respective LODs for more than 50% of the samples. Median water-extractable concentrations of these elements were generally highest outdoors, with the exceptions of water-extractable K and Cd for which indoor and outdoor concentrations were comparable. Median personal exposures to water-extractable Ba, Cd, Pb, Se, and V were below the LOD. For water-extractable Pb, As, and Cd, median indoor and outdoor concentrations were greater in the fall than in the summer, and for water-extractable As, median personal exposures were also greater in the fall than in the summer.

Table 78. Water-extractable elements concentrations (ng/m³) measured in personal, indoor, outdoor, and ambient environments for the older adult cohort.

Element	Location	Summer					Fall				
		N	Field LOD	10th	50th	90th	N	Field LOD	10th	50 th	90th
Aluminum	Personal	183	287.93	<LOD	<LOD	<LOD	210	287.93	<LOD	<LOD	<LOD
	Indoor	483	129.14	<LOD	<LOD	<LOD	495	129.14	<LOD	<LOD	<LOD
	Gaylord*	32	129.14	<LOD	<LOD	<LOD	32	129.14	<LOD	<LOD	<LOD
	Kennedy	62	129.14	<LOD	<LOD	<LOD	65	129.14	<LOD	<LOD	<LOD
	Ambient	101	129.14	<LOD	<LOD	<LOD	116	129.14	<LOD	<LOD	<LOD
Arsenic	Personal	188	0.05	<LOD	0.5	2.1	214	0.05	<LOD	1.2	3.8
	Indoor	494	0.02	0.2	1.0	2.4	519	0.02	0.2	1.5	4.4
	Gaylord	33	0.02	0.3	1.5	3.6	33	0.02	0.6	1.6	5.3
	Kennedy	65	0.02	0.4	1.4	2.9	68	0.02	0.5	2.1	5.3
	Ambient	103	0.02	0.2	1.3	2.9	115	0.02	0.5	2.3	7.3
Barium	Personal	180	0.38	<LOD	<LOD	30.6	167	0.38	<LOD	<LOD	5.0
	Indoor	448	0.17	<LOD	0.9	15.5	387	0.17	<LOD	1.5	10.5
	Gaylord	29	0.17	0.4	1.4	11.3	27	0.17	<LOD	3.1	18.9
	Kennedy	57	0.17	0.4	3.5	33.0	52	0.17	<LOD	2.7	11.1
	Ambient	93	0.17	<LOD	1.6	14.8	89	0.17	<LOD	1.2	6.7
Cadmium	Personal	170	0.10	<LOD	<LOD	2.3	162	0.10	<LOD	<LOD	1.4
	Indoor	428	0.04	<LOD	0.2	1.6	401	0.04	<LOD	0.4	1.9
	Gaylord	29	0.04	<LOD	0.1	1.2	25	0.04	<LOD	0.5	1.9
	Kennedy	56	0.04	<LOD	0.2	3.0	54	0.04	<LOD	0.4	2.0
	Ambient	91	0.04	<LOD	0.2	2.0	89	0.04	<LOD	0.5	2.6
Calcium	Personal	181	141.59	<LOD	<LOD	<LOD	176	141.59	<LOD	<LOD	288.3
	Indoor	459	63.50	<LOD	<LOD	<LOD	404	63.50	<LOD	<LOD	652.8
	Gaylord	30	63.50	<LOD	<LOD	<LOD	28	63.50	<LOD	<LOD	630.3
	Kennedy	62	63.50	<LOD	<LOD	<LOD	53	63.50	<LOD	<LOD	916.2
	Ambient	95	63.50	<LOD	<LOD	<LOD	95	63.50	<LOD	<LOD	542.0
Cobalt	Personal	188	0.19	<LOD	<LOD	<LOD	214	0.19	<LOD	<LOD	<LOD
	Indoor	490	0.09	<LOD	<LOD	<LOD	518	0.09	<LOD	<LOD	0.1
	Gaylord	33	0.09	<LOD	<LOD	<LOD	33	0.09	<LOD	<LOD	0.1
	Kennedy	65	0.09	<LOD	<LOD	<LOD	68	0.09	<LOD	<LOD	<LOD
	Ambient	102	0.09	<LOD	<LOD	<LOD	115	0.09	<LOD	<LOD	0.1
Copper	Personal	179	447.02	<LOD	<LOD	<LOD	212	447.02	<LOD	<LOD	<LOD
	Indoor	470	200.49	<LOD	<LOD	<LOD	511	200.49	<LOD	<LOD	<LOD
	Gaylord	33	200.49	<LOD	<LOD	<LOD	33	200.49	<LOD	<LOD	<LOD
	Kennedy	62	200.49	<LOD	<LOD	<LOD	68	200.49	<LOD	<LOD	<LOD
	Ambient	96	200.49	<LOD	<LOD	<LOD	115	200.49	<LOD	<LOD	<LOD
Iron	Personal	178	201.37	<LOD	<LOD	<LOD	136	201.37	<LOD	<LOD	<LOD
	Indoor	430	90.31	<LOD	<LOD	<LOD	314	90.31	<LOD	<LOD	123.0
	Gaylord	27	90.31	<LOD	<LOD	<LOD	21	90.31	<LOD	<LOD	185.9
	Kennedy	58	90.31	<LOD	<LOD	<LOD	39	90.31	<LOD	<LOD	145.6
	Ambient	88	90.31	<LOD	<LOD	<LOD	76	90.31	<LOD	<LOD	130.6

Table 78. (continued)

Element	Location	Summer					Fall				
		N	Field LOD	10th	50th	90th	N	Field LOD	10th	50 th	90th
Lead	Personal	125	6.85	<LOD	<LOD	44.1	167	6.85	<LOD	<LOD	28.7
	Indoor	338	3.07	<LOD	4.8	20.7	411	3.07	<LOD	8.6	39.8
	Gaylord	24	3.07	<LOD	5.7	34.5	27	3.07	<LOD	16.4	33.0
	Kennedy	49	3.07	<LOD	7.0	42.0	55	3.07	<LOD	11.1	46.9
	Ambient	68	3.07	<LOD	7.8	24.6	93	3.07	<LOD	12.3	51.2
Magnesium	Personal	188	12.41	<LOD	22.4	43.7	209	12.41	<LOD	13.0	61.6
	Indoor	494	5.56	<LOD	15.6	38.2	497	5.56	<LOD	17.5	65.1
	Gaylord	33	5.56	8.2	18.9	40.8	32	5.56	<LOD	20.7	88.7
	Kennedy	65	5.56	10.8	23.1	54.4	66	5.56	<LOD	22.7	91.1
	Ambient	103	5.56	8.8	19.0	40.1	112	5.56	<LOD	20.2	83.8
Manganese	Personal	181	17.40	<LOD	<LOD	<LOD	131	17.40	<LOD	<LOD	<LOD
	Indoor	442	7.80	<LOD	<LOD	8.4	296	7.80	<LOD	<LOD	17.8
	Gaylord	28	7.80	<LOD	<LOD	13.0	21	7.80	<LOD	<LOD	18.3
	Kennedy	60	7.80	<LOD	<LOD	12.7	38	7.80	<LOD	<LOD	21.8
	Ambient	91	7.80	<LOD	<LOD	11.6	74	7.80	<LOD	<LOD	17.5
Nickel	Personal	188	2.80	<LOD	<LOD	<LOD	213	2.80	<LOD	<LOD	<LOD
	Indoor	494	1.25	<LOD	<LOD	<LOD	513	1.25	<LOD	<LOD	<LOD
	Gaylord	33	1.25	<LOD	<LOD	<LOD	32	1.25	<LOD	<LOD	1.9
	Kennedy	65	1.25	<LOD	<LOD	<LOD	68	1.25	<LOD	<LOD	1.4
	Ambient	103	1.25	<LOD	<LOD	<LOD	114	1.25	<LOD	<LOD	1.3
Potassium	Personal	147	44.57	<LOD	60.7	219.0	192	44.57	<LOD	67.5	225.8
	Indoor	388	19.99	28.1	61.3	214.1	458	19.99	<LOD	79.6	196.1
	Gaylord	25	19.99	30.4	60.1	148.2	28	19.99	24.8	56.8	157.2
	Kennedy	48	19.99	41.1	68.2	222.7	60	19.99	29.5	74.5	177.7
	Ambient	83	19.99	25.0	46.8	133.0	102	19.99	<LOD	61.1	145.9
Selenium	Personal	188	1.52	<LOD	<LOD	3.3	183	1.52	<LOD	<LOD	7.8
	Indoor	494	0.68	<LOD	1.0	3.7	460	0.68	<LOD	1.4	7.2
	Gaylord	33	0.68	0.8	2.7	7.0	30	0.68	<LOD	3.6	15.2
	Kennedy	65	0.68	<LOD	2.0	6.8	61	0.68	<LOD	3.6	16.1
	Ambient	103	0.68	<LOD	2.3	6.3	104	0.68	<LOD	4.0	16.5
Sodium	Personal	183	97.31	<LOD	<LOD	460.7	214	97.31	<LOD	<LOD	152.1
	Indoor	457	43.65	<LOD	<LOD	175.6	507	43.65	<LOD	<LOD	140.0
	Gaylord	31	43.65	<LOD	<LOD	72.2	32	43.65	<LOD	<LOD	116.4
	Kennedy	62	43.65	<LOD	<LOD	105.0	64	43.65	<LOD	<LOD	186.0
	Ambient	95	43.65	<LOD	<LOD	114.4	111	43.65	<LOD	<LOD	128.3
Tin	Personal	157	0.29	<LOD	<LOD	<LOD	163	0.29	<LOD	<LOD	<LOD
	Indoor	415	0.13	<LOD	<LOD	<LOD	386	0.13	<LOD	<LOD	<LOD
	Gaylord	29	0.13	<LOD	<LOD	<LOD	26	0.13	<LOD	<LOD	<LOD
	Kennedy	56	0.13	<LOD	<LOD	<LOD	52	0.13	<LOD	<LOD	<LOD
	Ambient	89	0.13	<LOD	<LOD	<LOD	91	0.13	<LOD	<LOD	<LOD
Titanium	Personal	188	0.00	<LOD	<LOD	<LOD	214	0.00	<LOD	<LOD	<LOD
	Indoor	493	0.00	<LOD	<LOD	0.2	517	0.00	<LOD	<LOD	1.4
	Gaylord	33	0.00	<LOD	<LOD	0.6	33	0.00	<LOD	<LOD	<LOD
	Kennedy	65	0.00	<LOD	<LOD	0.5	68	0.00	<LOD	<LOD	2.4
	Ambient	103	0.00	<LOD	<LOD	1.6	115	0.00	<LOD	<LOD	4.0

Table 78. (continued)

Element	Location	Summer					Fall				
		N	Field LOD	10th	50th	90th	N	Field LOD	10th	50 th	90th
Vanadium	Personal	188	0.00	<LOD	<LOD	0.8	214	0.00	<LOD	<LOD	3.2
	Indoor	494	0.00	<LOD	0.3	1.2	518	0.00	<LOD	0.4	3.7
	Gaylord	33	0.00	<LOD	0.4	1.5	33	0.00	<LOD	0.6	3.0
	Kennedy	65	0.00	<LOD	0.5	1.5	68	0.00	<LOD	0.9	5.0
	Ambient	103	0.00	<LOD	0.5	2.3	115	0.00	<LOD	1.0	8.2
Zinc	Personal	188	417.00	<LOD	<LOD	<LOD	211	417.00	<LOD	<LOD	<LOD
	Indoor	489	187.03	<LOD	<LOD	<LOD	506	187.03	<LOD	<LOD	<LOD
	Gaylord	33	187.03	<LOD	<LOD	<LOD	33	187.03	<LOD	<LOD	207.1
	Kennedy	65	187.03	<LOD	<LOD	<LOD	67	187.03	<LOD	<LOD	<LOD
	Ambient	101	187.03	<LOD	<LOD	<LOD	113	187.03	<LOD	<LOD	<LOD

*Gaylord and Kennedy are rooftop monitors. “<LOD” indicates concentrations that were below the corresponding field LOD.

Similarly, median acid-digestible concentrations were above their field LOD for several of the measured elements, including Al, Cu (fall only), Fe, K, Mg, Mn, Pb, Se, V, and Zn (Table 79). Median acid-digestible Al, Fe, Pb, Se, and Zn concentrations tended to be higher outdoors than in indoor and personal environments. Median acid-digestible aluminum and zinc concentrations were higher in summer as compared to fall, while the opposite was true for copper and iron. Seasonal variation in the median acid-digestible concentrations for the other elements was not apparent.

Table 79. Water-extractable elements concentrations (ng/m³) measured in personal, indoor, outdoor, and ambient environments for the older adult cohort.

Element	Location	Summer					Fall				
		N	Field LOD	10th	50th	90th	N	Field LOD	10th	50th	90th
Aluminum	Personal	84	0.02	<LOD	87.9	323.8	154	0.02	<LOD	5.4	486.1
	Indoor	198	0.05	<LOD	55.5	241.5	352	0.05	<LOD	19.2	186.4
	Gaylord*	12	0.05	<LOD	60.9	197.8	23	0.05	<LOD	30.2	202.4
	Kennedy	29	0.05	11.0	103.7	258.0	39	0.05	<LOD	33.5	166.5
	Ambient	41	0.05	<LOD	88.8	233.3	78	0.05	<LOD	29.7	171.8
Arsenic	Personal	157	0.12	<LOD	<LOD	<LOD	198	0.12	<LOD	<LOD	0.4
	Indoor	410	0.30	<LOD	<LOD	1.0	470	0.30	<LOD	<LOD	1.8
	Gaylord	24	0.30	<LOD	<LOD	1.8	30	0.30	<LOD	<LOD	2.5
	Kennedy	56	0.30	<LOD	<LOD	1.2	59	0.30	<LOD	1.0	2.9
	Ambient	82	0.30	<LOD	<LOD	1.0	107	0.30	<LOD	0.5	2.9
Barium	Personal	144	0.17	<LOD	<LOD	5.5	167	0.17	<LOD	<LOD	2.0
	Indoor	382	0.27	<LOD	<LOD	4.4	401	0.27	<LOD	<LOD	5.0
	Gaylord	23	0.27	<LOD	0.5	2.9	27	0.27	<LOD	0.7	10.1
	Kennedy	52	0.27	<LOD	0.3	6.0	50	0.27	<LOD	0.7	6.6
	Ambient	72	0.27	<LOD	<LOD	5.5	91	0.27	<LOD	<LOD	2.6
Cadmium	Personal	133	0.17	<LOD	<LOD	0.9	174	0.17	<LOD	0.3	2.0
	Indoor	355	0.37	<LOD	<LOD	0.7	409	0.37	<LOD	<LOD	1.3
	Gaylord	23	0.37	<LOD	<LOD	<LOD	29	0.37	<LOD	<LOD	1.2
	Kennedy	48	0.37	<LOD	<LOD	0.6	52	0.37	<LOD	<LOD	1.3
	Ambient	67	0.37	<LOD	<LOD	1.0	94	0.37	<LOD	<LOD	1.4

Table 79. (continued)

Element	Location	Summer					Fall				
		N	Field LOD	10th	50th	90th	N	Field LOD	10th	50th	90th
Calcium	Personal	3	0.07	<LOD	<LOD	<LOD	96	0.07	<LOD	<LOD	82.3
	Indoor	41	0.05	<LOD	<LOD	<LOD	232	0.05	<LOD	<LOD	610.3
	Gaylord	1	0.05	<LOD	<LOD	<LOD	18	0.05	<LOD	<LOD	<LOD
	Kennedy	7	0.05	<LOD	<LOD	<LOD	27	0.05	<LOD	<LOD	235.3
	Ambient	6	0.05	<LOD	<LOD	72.4	57	0.05	<LOD	<LOD	537.2
Cobalt	Personal	89	0.02	<LOD	<LOD	<LOD	135	0.02	<LOD	<LOD	1.9
	Indoor	230	0.05	<LOD	<LOD	<LOD	318	0.05	<LOD	<LOD	1.1
	Gaylord	12	0.05	<LOD	<LOD	<LOD	24	0.05	<LOD	<LOD	<LOD
	Kennedy	31	0.05	<LOD	<LOD	<LOD	41	0.05	<LOD	<LOD	0.3
	Ambient	47	0.05	<LOD	<LOD	<LOD	79	0.05	<LOD	<LOD	<LOD
Copper	Personal	145	0.02	<LOD	<LOD	9.5	186	0.02	<LOD	3.6	33.0
	Indoor	391	0.02	<LOD	<LOD	6.1	442	0.02	<LOD	2.7	13.6
	Gaylord	21	0.02	<LOD	<LOD	2.8	29	0.02	<LOD	2.7	22.9
	Kennedy	53	0.02	<LOD	<LOD	2.7	55	0.02	<LOD	3.9	23.6
	Ambient	77	0.02	<LOD	<LOD	4.3	101	0.02	<LOD	2.6	13.5
Iron	Personal	149	0.33	<LOD	114.8	449.6	204	0.33	<LOD	132.4	573.5
	Indoor	375	0.54	21.7	114.1	383.3	454	0.54	<LOD	167.6	590.9
	Gaylord	23	0.54	43.4	167.0	503.8	30	0.54	4.3	193.9	876.1
	Kennedy	51	0.54	41.1	146.8	483.0	58	0.54	<LOD	208.1	771.4
	Ambient	76	0.54	20.5	119.2	325.5	104	0.54	<LOD	169.9	861.2
Lead	Personal	141	0.09	0.4	5.2	14.0	169	0.09	<LOD	6.5	19.4
	Indoor	382	0.16	2.1	5.8	14.2	385	0.16	1.5	7.0	20.3
	Gaylord	23	0.16	2.7	8.1	17.6	25	0.16	1.6	10.4	27.3
	Kennedy	52	0.16	2.6	8.5	17.6	49	0.16	2.7	9.7	20.6
	Ambient	73	0.16	2.9	7.4	15.7	91	0.16	2.0	8.2	27.6
Magnesium	Personal	151	0.16	<LOD	24.3	66.4	194	0.16	<LOD	24.9	82.6
	Indoor	389	0.33	3.0	20.8	52.6	461	0.33	4.4	24.6	76.9
	Gaylord	23	0.33	<LOD	30.4	53.7	30	0.33	2.6	24.1	85.4
	Kennedy	53	0.33	6.7	27.9	55.4	58	0.33	7.5	29.7	98.1
	Ambient	77	0.33	3.0	26.3	63.3	105	0.33	4.3	21.0	104.0
Manganese	Personal	140	0.10	<LOD	5.7	20.7	155	0.10	<LOD	3.5	24.8
	Indoor	352	0.30	1.5	6.5	18.2	378	0.30	<LOD	8.2	27.3
	Gaylord	23	0.30	2.6	12.4	21.1	23	0.30	<LOD	9.8	40.2
	Kennedy	48	0.30	2.8	9.1	21.9	44	0.30	<LOD	12.9	38.0
	Ambient	69	0.30	1.8	8.0	22.6	83	0.30	<LOD	11.8	33.6
Nickel	Personal	70	0.00	<LOD	<LOD	7.9	151	0.00	<LOD	<LOD	<LOD
	Indoor	198	0.01	<LOD	0.3	3.8	344	0.01	<LOD	<LOD	<LOD
	Gaylord	11	0.01	<LOD	<LOD	3.4	23	0.01	<LOD	<LOD	<LOD
	Kennedy	26	0.01	<LOD	0.1	3.5	46	0.01	<LOD	<LOD	4.9
	Ambient	44	0.01	<LOD	0.8	8.1	83	0.01	<LOD	<LOD	<LOD
Potassium	Personal	117	0.16	<LOD	144.0	296.1	169	0.16	<LOD	54.7	165.2
	Indoor	286	0.33	27.4	67.4	234.6	394	0.33	<LOD	65.5	199.3
	Gaylord	17	0.33	35.4	65.1	128.6	24	0.33	<LOD	62.6	172.7
	Kennedy	39	0.33	8.6	67.8	138.3	45	0.33	<LOD	57.4	185.0
	Ambient	55	0.33	12.2	67.0	199.2	88	0.33	<LOD	62.8	161.4

Table 79. (continued)

Element	Location	Summer					Fall				
		N	Field LOD	10th	50th	90th	N	Field LOD	10th	50th	90th
Selenium	Personal	88	0.03	<LOD	<LOD	3.8	80	0.03	<LOD	<LOD	8.3
	Indoor	221	0.05	<LOD	1.0	3.4	214	0.05	<LOD	0.7	5.0
	Gaylord	10	0.05	0.9	1.9	7.3	15	0.05	<LOD	2.1	9.1
	Kennedy	28	0.05	0.2	1.6	4.8	29	0.05	<LOD	2.3	13.4
	Ambient	46	0.05	<LOD	1.8	6.3	56	0.05	0.2	3.5	13.8
Sodium	Personal	108	0.07	<LOD	<LOD	967.7	0	0.07	NA	NA	NA
	Indoor	287	0.05	<LOD	<LOD	345.7	0	0.05	NA	NA	NA
	Gaylord	18	0.05	<LOD	<LOD	103.0	0	0.05	NA	NA	NA
	Kennedy	38	0.05	<LOD	<LOD	212.2	0	0.05	NA	NA	NA
	Ambient	48	0.05	<LOD	<LOD	287.7	0	0.05	NA	NA	NA
Tin	Personal	118	0.04	<LOD	<LOD	4.0	167	0.04	<LOD	<LOD	1.4
	Indoor	311	0.05	<LOD	0.2	3.0	395	0.05	<LOD	0.1	2.5
	Gaylord	18	0.05	<LOD	0.1	1.4	28	0.05	<LOD	0.1	4.6
	Kennedy	41	0.05	<LOD	0.1	1.2	49	0.05	<LOD	<LOD	3.0
	Ambient	59	0.05	<LOD	0.1	2.2	88	0.05	<LOD	<LOD	2.6
Titanium	Personal	130	0.02	<LOD	<LOD	79.1	138	0.02	<LOD	<LOD	46.3
	Indoor	326	0.06	<LOD	<LOD	32.2	331	0.06	<LOD	<LOD	45.2
	Gaylord	19	0.06	<LOD	<LOD	15.0	22	0.06	<LOD	<LOD	28.5
	Kennedy	48	0.06	<LOD	<LOD	25.0	43	0.06	<LOD	<LOD	45.6
	Ambient	64	0.06	<LOD	<LOD	74.7	81	0.06	<LOD	<LOD	60.9
Vanadium	Personal	130	0.02	<LOD	2.3	9.7	148	0.02	<LOD	0.3	4.7
	Indoor	336	0.05	<LOD	1.3	3.9	369	0.05	<LOD	1.0	5.4
	Gaylord	18	0.05	<LOD	1.0	2.1	28	0.05	<LOD	1.2	2.7
	Kennedy	48	0.05	<LOD	1.3	3.5	45	0.05	<LOD	1.8	6.3
	Ambient	63	0.05	<LOD	1.8	4.2	85	0.05	<LOD	1.5	10.1
Zinc	Personal	140	0.05	2.5	64.4	211.8	169	0.05	<LOD	12.0	171.2
	Indoor	370	0.08	9.8	47.3	125.2	398	0.08	<LOD	26.9	129.2
	Gaylord	22	0.08	19.4	72.5	171.0	27	0.08	<LOD	61.0	344.8
	Kennedy	52	0.08	12.9	66.9	141.7	50	0.08	<LOD	46.6	205.3
	Ambient	74	0.08	9.0	43.3	122.5	92	0.08	<LOD	27.0	185.1

*Gaylord and Kennedy are rooftop monitors. "<LOD" indicates concentrations that were below the corresponding field LOD.

4.3.1.8 Associations Among Personal, Indoor, and Ambient Elemental Concentrations

Associations among personal, indoor, and ambient concentrations were examined for a subset of six elements, including Cu, Fe, K, Pb, Se, and V. These elements were selected based on their water-extractable and acid-digestible concentrations, their relevance to health studies, and their precision (as determined from QA/QC results). Despite being selected on the basis of these considerations, it is important to note that measurements of these elements were nevertheless relatively imprecise, and their concentrations were still low relative to the sensitivity of the method. Hence, the results presented below should be interpreted with caution.

4.3.1.8.1 Association between Personal Exposures and Indoor Concentrations

Associations between personal exposures and indoor concentrations differed by element, extraction method, and season (Table 80). For acid-digestible concentrations, intercepts of the regression lines for personal concentrations as a function of indoor concentrations were generally significant in both seasons (with the exception of Pb in the fall), with very large intercepts for Fe and K. These large intercepts suggest that personal exposures to Fe and K are affected by a source that is independent of the indoor microenvironment. Slopes describing the personal-indoor associations tended to be significant in the summer, with the exception of Cu. Slopes were close to 1.00 for V, and to a lesser extent for Fe, which is consistent with the fact that individuals spend most of their time indoors. In the fall, slopes describing the personal-indoor associations were significant only for V and Fe, with slopes for both elements lower than those in the summer.

Table 80. Personal vs. indoor concentrations of select elements (acid-digestible) for the older adult cohort.*

Pollutant	Summer					Fall				
	N	β	SE	t-stat	Int.	N	β	SE	t-stat	Int.
Cu	57	0.20	0.15	1.35	10.89	186	0.13	0.10	1.28	13.91
Fe	149	0.75	0.12	6.33	155.19	204	0.57	0.04	15.11	112.87
K	117	0.25	0.04	5.64	151.94	169	0.04	0.03	1.10	132.56
Pb	141	0.58	0.08	7.49	3.57	169	0.38	0.24	1.58	7.52
Se	78	0.42	0.09	4.59	1.90	60	0.31	0.19	1.67	4.23
V	130	0.96	0.49	1.96	2.92	148	0.39	0.08	4.86	1.69

*bold numbers indicate p-value ≤ 0.05

The personal-indoor associations for the six water-extractable elements showed a different pattern, although they too varied by element and season (Table 81). Intercepts of the personal-on-indoor regression lines for the water-extractable elements were largely significant, especially in the fall. Intercepts, however, were lower for water-soluble K and especially Fe in both seasons as compared to those for their acid-digestible counterparts. Slopes describing the personal-indoor associations were generally well below 1.00, even when significant. In contrast to the acid-digestible elements, slopes for the six elements tended to be significant in the fall as compared to summer, as all but Cu had significant fall slopes. However, for V, Se, and K, for which the slopes were significant in both seasons, the slopes were greater in the summer than in the fall.

Table 81. Personal vs. indoor concentrations of select elements (water-extractable) for the older adult cohort.*

Pollutant	Summer					Fall				
	N	β	SE	t-stat	Int.	N	β	SE	t-stat	Int.
Cu	11	n/a	n/a	n/a	n/a	84	-0.02	0.46	0.96	50.36
Fe	88	0.04	0.02	1.67	29.67	76	0.38	0.04	9.11	34.73
K	83	0.40	0.07	5.76	54.47	102	0.07	0.04	2.00	89.65
Pb	68	0.10	0.09	1.10	16.34	93	0.13	0.02	5.86	12.99
Se	103	0.75	0.04	19.77	0.25	104	0.47	0.06	8.41	1.58
V	103	0.60	0.04	14.46	0.27	115	0.40	0.04	10.47	1.08

*bold numbers indicate p-value ≤ 0.05

4.3.1.8.2 Association between Indoor Concentrations and Ambient Concentrations

Indoor concentrations of all of the selected acid-digestible elements (except for Cu in the summer) were significantly associated with corresponding ambient concentrations in both seasons, as evidenced by the significant slopes shown in Table 82. As was the case with personal-indoor associations, intercepts for the indoor-ambient associations were largest for Fe and K in both seasons. Although significant, indoor-ambient slopes for the acid-digestible elements varied around 0.50, with K having the greatest slope in both seasons. Slopes did not show a consistent seasonal pattern.

Water-extractable elements also tended to exhibit significant associations between indoor and ambient concentrations (Table 83). Indoor-ambient slopes for the water-extractable elements tended to be more strongly significant than those for the acid-digestible elements, as evidenced by the higher t-statistics. Furthermore, indoor-ambient slopes for the water-extractable elements, particularly for Cu in the fall and for Fe and K in both seasons, tended to be closer to 1.00 than slopes for corresponding acid-digestible elements. Associations between indoor and ambient concentrations for both the acid-extractable and water-soluble elements differed substantially from those observed between personal and indoor levels. Factors contributing to these differences are not known.

Table 82. Indoor vs. ambient concentrations of select elements (acid-digestible) for the older adult cohort.*

Pollutant	Summer					Fall				
	N	β	SE	t-stat	Int.	N	β	SE	t-stat	Int.
Cu	57	0.02	0.07	0.31	6.08	101	0.04	0.02	2.13	8.62
Fe	76	0.28	0.04	6.90	110.54	104	0.41	0.03	13.88	134.17
K	55	0.70	0.06	10.98	40.86	88	0.65	0.19	3.38	66.08
Pb	73	0.34	0.06	5.75	4.04	91	0.43	0.03	14.86	3.99
Se	46	0.39	0.03	11.62	0.76	56	0.26	0.03	10.21	0.89
V	63	0.50	0.04	14.11	0.85	85	0.47	0.04	11.29	0.82

*bold numbers indicate p-value \leq 0.05

Table 83. Indoor vs. ambient concentrations of select elements (water-extractable) for the older adult cohort.*

Pollutant	Summer					Fall				
	N	β	SE	t-stat	Int.	N	β	SE	t-stat	Int.
Cu	21	0.04	0.07	0.59	22.24	101	1.07	0.44	2.42	9.33
Fe	76	0.25	0.11	2.24	18.35	104	0.84	0.05	16.60	4.16
K	55	0.81	0.03	28.77	29.16	88	0.85	0.14	6.12	49.72
Pb	73	0.37	0.06	5.73	5.48	91	0.18	0.06	2.90	13.90
Se	46	0.44	0.02	22.39	0.30	56	0.42	0.03	13.28	0.42
V	63	0.36	0.02	22.69	0.18	85	0.48	0.03	17.43	0.58

*bold numbers indicate p-value \leq 0.05

4.3.1.8.3 Association between Personal Exposures and Ambient Concentrations

Summertime personal exposures were significantly associated with corresponding ambient concentrations of all the selected acid-digestible elements, except for Cu, as evidenced by the significant slopes shown in Table 84. In the fall, personal-ambient slopes were significant for Fe, Pb, and V, with significant intercepts for all selected elements except for Se. As was the case with personal-indoor and indoor-ambient associations, intercepts for the personal-ambient

associations were largest for Fe and K in both seasons. Slopes for the acid-digestible elements did not show a pattern consistent with season or with personal-indoor or indoor-ambient associations.

Water-extractable elements also tended to exhibit significant associations between personal exposures and ambient concentrations (Table 85); however, associations were unrelated to those for acid-digestible elements. Associations between personal and ambient concentrations for both the acid-digestible and water-extractable elements differed substantially from those observed between personal and indoor levels and between indoor and ambient levels. Factors contributing to these differences are not known.

Table 84. Personal vs. ambient concentrations of select elements (acid-digestible) for the older adult cohort.*

Pollutant	Summer					Fall				
	N	β	SE	t-stat	Int.	N	β	SE	t-stat	Int.
Cu	47	NA	NA	NA	NA	101	0.11	0.06	1.84	14.41
Fe	76	0.27	0.10	2.59	213.85	104	0.42	0.02	16.94	125.23
K	55	0.39	0.05	7.39	142.12	88	-0.36	0.18	-2.02	181.59
Pb	73	0.20	0.06	3.36	5.04	91	0.43	0.02	19.19	2.69
Se	46	0.15	0.05	3.19	1.87	56	NA	NA	NA	NA
V	63	0.73	0.10	7.26	2.06	85	0.33	0.08	4.06	1.79

*bold numbers indicate p-value \leq 0.05

Table 85. Personal vs. ambient concentrations of select elements (water-extractable) for the older adult cohort.*

Pollutant	Summer					Fall				
	N	β	SE	t-stat	Int.	N	β	SE	t-stat	Int.
Cu	15	NA	NA	NA	NA	95	-1.20	1.24	-0.96	53.34
Fe	76	0.21	0.04	5.92	18.09	104	0.45	0.05	9.02	26.75
K	55	0.27	0.23	1.17	92.74	88	0.45	0.11	3.97	74.44
Pb	73	0.40	0.14	2.95	14.14	91	0.14	0.03	5.44	12.12
Se	46	0.51	0.02	30.04	-0.21	56	0.31	0.04	7.10	1.03
V	63	0.30	0.02	15.64	0.27	85	0.32	0.02	13.27	0.95

*bold numbers indicate p-value \leq 0.05

4.3.2 Children's Cohort

4.3.2.1 Participant Summary and Characteristics

A total of 19 children participated in the study. Their homes are identified in Figure 64. Sixteen subjects between the ages of 10 and 12 years participated in the winter 2001 season; 62.5% were female. Fifteen participants between the ages of 9 and 13 years, including 12 who had participated in the winter season, were sampled in the summer; 66.7% of the participants in the summer season were female.

4.3.2.2 Housing Summary and Characteristics

Housing characteristics for the cohort of children are summarized in Table 86. A large fraction of the children's homes had electric stoves and central heating systems. Central heating systems were also supplemented with gas and space heaters in a few homes. All but one

home had a source of ventilation in the kitchen area (i.e., fan over the cooking stove, range, oven, or elsewhere). Fourteen of the 19 homes had central air conditioning. Window air conditioning units were found in nine homes, with four of these homes having at least two units. Identifiable sources of dust (i.e., busy road, dirt road, other source of dust) were found at six homes. Two homes were located next to both a busy road and a dirt road.



Figure 64. Locations of the residences of participants in the children's cohort. The central ambient air monitoring site was located at the Franciscan University of Steubenville.

Table 86. Housing characteristics of the children's cohort.

Housing Characteristic		Count
Cooking Fuel	Gas	5
	Electric	14
Heating Source	Central	16
	Radiator	3
	Space	2
	Fireplace	1
	Gas	1
	Electric Coil	1
Ventilation	Kitchen Fan	18
	Central AC Units	14
	Window Units	9
Outdoor Sources	Busy Road	4
	Dirt Road	4

4.3.2.3 Time-Activity Data Summary

Results from the children's time-activity questionnaires are summarized in Table 87. This table displays the mean fraction of time spent in various microenvironments, in transit, or near pollutant sources or ventilation over the 24-hr periods when personal measurements were collected. When averaged over all sampling days, subjects spent the vast majority of their time

indoors at home. In the winter, 71% of the time was spent indoors at home, while in the summer, the mean percent of time spent indoors at home equaled 81%. Subjects spent the next largest fraction of time indoors but away from home. On average, subjects spent 15% of the day indoors but away from home (24% in winter; 8% in the summer). The children's cohort spent, on average, 2% and 9% of their time outdoors for the winter and summer seasons, respectively. Only two participants, on average, spent greater than 10% of their time outdoors. Relatively little time was spent in transit, with the highest average travel time being spent in a car when compared to other modes of transportation (i.e., bus, walking, bicycle).

Table 87. Fraction of time spent in various locations, in transit, and near sources for the children's cohort.

Location	Winter					Summer				
	Mean	SD	Median	Min	Max	Mean	SD	Median	Min	Max
Indoors, Home and Away	0.95	0.02	0.96	0.88	0.99	0.88	0.06	0.88	0.78	0.99
Indoors, Home	0.71	0.07	0.71	0.59	0.79	0.81	0.09	0.81	0.68	0.98
Indoors, Away from Home	0.24	0.06	0.23	0.16	0.36	0.08	0.05	0.08	0.01	0.19
Outdoors	0.02	0.01	0.02	0.00	0.04	0.09	0.06	0.09	0.00	0.21
Travel by Car	0.02	0.01	0.02	0.01	0.07	0.02	0.01	0.02	0.01	0.05
Travel by Bus	0.00	0.01	0.00	0.00	0.04	0.00	0.00	0.00	0.00	0.00
Travel by Walking	0.00	0.00	0.00	0.00	0.01	0.00	0.00	0.00	0.00	0.00
Travel by Bicycle	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Near Smoker	0.00	0.00	0.00	0.00	0.02	0.00	0.01	0.00	0.00	0.02
Near Open Window	0.01	0.02	0.00	0.00	0.08	0.32	0.29	0.28	0.00	0.82

Note: Data obtained from time-activity diaries. Values represent mean fraction of time over 24-hr period for all person-days.

4.3.2.4 Pollutant Summary

4.3.2.4.1 Precision, Accuracy, and Data Completeness

Sample validity, limits of detection, completeness, and imprecision for the major pollutant data collected for the children's cohort are presented in Table 88.

Pollutant concentrations were measured successfully during the study, as indicated by the high rates of completeness found in this study. The field LODs were calculated as the 95th percentile of the net mass of the blank samples. SO₄²⁻ and indoor and outdoor PM_{2.5} concentrations were typically greater than the limit of detection. Note that SO₂ values were not reported, because they were largely below the LOD.

Table 88. Quality assurance parameters – children's cohort.*

Pollutant	N	Field LOD	% > Field LOD	Completeness	Precision
PM _{2.5} (personal)	509	18.71	83%	100%	8%
PM _{2.5} (indoor/outdoor)	1024	8.36	99%	100%	9%
SO ₄ ²⁻	769	0.18	100%	99%	7%
EC	762	0.33	75%	100%	17%
O ₃	803	7.03	58%	98%	24%
NO ₂	801	10.36	53%	99%	29%

*Limit of detection for PM_{2.5}, SO₄²⁻, and EC in µg, for gases in ng

4.3.2.4.2 Ambient Concentrations

Table 89 shows summary statistics for ambient pollutant concentrations (by season) measured by the Harvard multi-pollutant monitor at the SCAMP central ambient air monitoring site. Mean ambient concentrations were higher in the summer than in the winter, particularly for the regional pollutants $\text{PM}_{2.5}$, SO_4^{2-} , and O_3 . For example, the ratio of summer/winter means for SO_4^{2-} was about 2.5. In contrast, ambient EC and NO_2 concentrations were only slightly higher in the summer as compared to winter (summer/winter ratio = 1.1 for both pollutants). O_3 and $\text{PM}_{2.5}$ concentrations were the most variable, with standard deviations of 10.04 (winter) and 27.26 (summer), and 7.22 (winter) and 13.96 (summer), respectively.

4.3.2.4.3 Outdoor Concentrations

Mean concentrations of each of the pollutants (except for NO_2) measured outside of the participants' homes were greater in the summer than in the winter (Table 89). As was the case with ambient pollution, the greatest seasonal difference was found for SO_4^{2-} , with the average summer concentration almost three times higher than that in the winter. The mean value for $\text{PM}_{2.5}$ was almost two times higher in the summer as compared to the winter. Values in the summer were also more variable (with the exception of EC values), as shown by the greater standard deviations during that season.

4.3.2.4.4 Personal and Indoor Concentrations

As was the case outdoors, mean concentrations of all of the major pollutants except for NO_2 inside the participants' homes were higher in the summer than in the winter (Table 89), with a mean SO_4^{2-} concentration three times higher in the summer as compared to the winter (ratios of summer-to-winter means were 2.0 for $\text{PM}_{2.5}$; 1.2 for EC; 2.2 for O_3 ; and 0.7 for NO_2). Mean indoor concentrations, however, were less than outdoor concentrations with the exception of NO_2 in the winter and EC in the summer (indoor/outdoor ratio is 1.1 for both exceptions). Indoor O_3 concentrations were substantially less than corresponding outdoor values. In the winter and summer, the mean outdoor O_3 concentration was eight and four times greater than that indoors, respectively. Similarly, mean personal exposures for all pollutants except NO_2 were higher in the summer season as compared to the winter (ratios of summer-to-winter means: $\text{PM}_{2.5}$ = 1.7; SO_4^{2-} = 3.3; EC = 1.9; O_3 = 1.8; NO_2 = 0.7).

Personal exposures were similar to, although typically greater than, indoor concentrations. For example, personal-indoor ratios were 1.6 ± 1.0 , 1.1 ± 0.4 , 1.3 ± 0.5 , 1.8 ± 3.6 , and 1.2 ± 0.9 for $\text{PM}_{2.5}$, SO_4^{2-} , EC, O_3 , and NO_2 , respectively, in the winter, and 1.1 ± 0.4 , 1.2 ± 0.6 , 1.2 ± 0.5 , 1.2 ± 1.7 , and 1.3 ± 2.1 , respectively, for these pollutants in the summer.

Mean personal exposures were typically less than outdoor concentrations. Exceptions include EC (personal/outdoor ratio = 1.2 ± 0.8 in the winter, 2.2 ± 12 in the summer) and NO_2 in the winter (personal/outdoor ratio = 1.6 ± 2.3), for which mean personal exposures were slightly greater than mean outdoor concentrations.

Table 89. Personal, indoor, outdoor, and ambient pollutant concentrations by season for the children's cohort.*

Location	Pollutant	Season	N	#ND	Mean	Median	SD	Max
Ambient	PM _{2.5}	Winter	56	0	15.41	14.97	7.22	37.47
		Summer	58	0	25.14	21.23	13.96	58.10
	SO ₄ ²⁻	Winter	28	0	3.84	3.71	2.11	10.84
		Summer	29	0	9.39	7.80	6.58	25.34
	EC	Winter	28	1	0.74	0.53	0.43	1.55
		Summer	29	0	0.81	0.79	0.31	1.42
	O ₃	Winter	28	0	24.80	21.13	10.04	62.82
		Summer	29	1	35.17	33.63	27.26	148.09
	NO ₂	Winter	28	0	8.12	7.25	4.27	17.43
		Summer	29	2	9.04	6.21	10.38	46.47
Outdoor	PM _{2.5}	Winter	220	0	13.44	12.62	6.10	36.38
		Summer	210	0	24.23	20.93	13.57	57.44
	SO ₄ ²⁻	Winter	110	0	3.39	2.96	1.68	10.68
		Summer	105	0	9.03	8.13	5.88	24.79
	EC	Winter	110	2	0.58	0.47	0.32	1.45
		Summer	105	1	0.67	0.61	0.30	1.58
	O ₃	Winter	110	0	20.17	19.53	7.07	45.13
		Summer	105	0	26.13	26.44	10.05	45.10
	NO ₂	Winter	110	0	7.48	6.97	4.36	21.46
		Summer	105	11	6.43	5.93	5.91	23.79
Indoor	PM _{2.5}	Winter	220	0	9.12	7.15	5.90	31.20
		Summer	210	0	18.50	15.09	12.42	60.85
	SO ₄ ²⁻	Winter	110	0	1.93	1.46	1.23	7.09
		Summer	105	0	5.84	4.38	4.71	23.19
	EC	Winter	110	2	0.59	0.41	0.68	4.86
		Summer	105	0	0.71	0.50	1.57	16.14
	O ₃	Winter	110	4	2.67	2.69	1.65	6.11
		Summer	105	48	5.95	1.70	8.61	48.68
	NO ₂	Winter	110	0	8.08	6.31	5.56	27.20
		Summer	105	18	5.44	3.28	5.43	21.66
Personal	PM _{2.5}	Winter	220	0	11.58	10.64	7.00	43.80
		Summer	210	0	19.63	16.25	12.86	63.89
	SO ₄ ²⁻	Winter	110	0	1.91	1.44	1.25	6.50
		Summer	105	1	6.19	4.96	4.65	22.88
	EC	Winter	110	1	0.61	0.51	0.33	1.82
		Summer	105	2	1.14	0.54	4.02	35.46
	O ₃	Winter	110	4	3.11	2.79	2.33	17.79
		Summer	105	25	5.56	3.46	7.12	43.34
	NO ₂	Winter	110	0	7.65	6.93	4.33	31.73
		Summer	105	6	5.63	4.01	5.76	33.00

Note: Values represent summary statistics over all person-days. N represents the total number of collected valid samples. Concentrations expressed in $\mu\text{g}/\text{m}^3$ for PM_{2.5}, EC, and SO₄²⁻, and in ppb for O₃ and NO₂.

4.3.2.5 Associations among Pollutants

Spearman correlation coefficients were very similar for between-pollutant comparisons using ambient (i.e., central site) and outdoor (i.e., outside of the participants' homes) concentrations (Table 90). At both the ambient and outdoor monitoring locations, concentrations of PM_{2.5} were strongly associated with concentrations of its components, SO₄²⁻ and EC. Correlations between concentrations of PM_{2.5} and gases were weaker, with Spearman correlation coefficients of 0.26 and 0.41 for pairwise comparisons between ambient PM_{2.5} and ambient O₃ and NO₂, respectively. Correlations of SO₄²⁻ with the gases were similar to those for PM_{2.5}, whereas EC correlated more strongly with NO₂ than did PM_{2.5}, but did not show any significant correlation with O₃.

Table 90. Correlations among outdoor and ambient PM_{2.5}, SO₄²⁻, EC, O₃, and NO₂ concentrations for the children's cohort.*

	PM _{2.5}	SO ₄ ²⁻	EC	O ₃	NO ₂
PM _{2.5}	--	0.89	0.68	0.26	0.41
SO ₄ ²⁻	0.88	--	0.51	0.16	0.32
EC	0.65	0.55	--	0.03	0.59
O ₃	0.26	0.26	0.01	--	0.03
NO ₂	0.42	0.32	0.60	-0.12	--

*Values are Spearman correlation coefficients for pairwise comparisons among the pollutants. A total of 202 PM_{2.5}, 203 SO₄²⁻, 200 EC, 214 O₃, and 210 NO₂ outdoor samples and 52 PM_{2.5}, 51 SO₄²⁻, 52 EC, 53 O₃, and 54 NO₂ ambient samples were included in the analysis. Outdoor correlations shown in shaded boxes (lower left); ambient correlations shown in clear boxes (upper right).

In both indoor and personal microenvironments, correlations among the measured pollutants generally followed similar trends, although associations between NO₂ and PM_{2.5}, SO₄²⁻, and EC were much lower than those observed outdoors (Tables 91 and 92). In contrast, O₃ and NO₂ were more strongly associated indoors as compared to outdoors, with Spearman correlation coefficients of 0.26 indoors and -0.12 outdoors. The correlation between PM_{2.5} and NO₂ personal exposures was slightly greater than that between indoor concentrations of these pollutants; otherwise, between-pollutant correlations for personal exposures mirrored those observed indoors.

Table 91. Correlations among indoor PM_{2.5}, SO₄²⁻, EC, O₃, and NO₂ concentrations for the children's cohort.*

	PM _{2.5}	SO ₄ ²⁻	EC	O ₃	NO ₂
PM _{2.5}	--				
SO ₄ ²⁻	0.79	--			
EC	0.57	0.48	--		
O ₃	0.30	0.22	0.30	--	
NO ₂	0.04	-0.02	0.29	0.26	--

*Values are Spearman correlation coefficients for pairwise comparisons among the pollutants. A total of 209 PM_{2.5}, 205 SO₄²⁻, 207 EC, 211 O₃, and 211 NO₂ samples were included in the analysis.

Table 92. Correlations among personal PM_{2.5}, SO₄²⁻, EC, O₃, and NO₂ exposures for the children's cohort.*

	PM _{2.5}	SO ₄ ²⁻	EC	O ₃	NO ₂
PM _{2.5}	--				
SO ₄ ²⁻	0.70	--			
EC	0.51	0.38	--		
O ₃	0.29	0.30	0.16	--	
NO ₂	0.26	-0.01	0.25	0.19	--

*Values are Spearman correlation coefficients for pairwise comparisons among the pollutants. A total of 206 PM_{2.5}, 208 SO₄²⁻, 205 EC, 210 O₃, and 215 NO₂ samples were included in the analysis.

Table 93 displays the associations among pollutants within each microenvironment that were estimated using mixed models. Outdoors in the winter, particle and gas concentrations were significantly associated. In the summer, outdoor PM_{2.5} and SO₄²⁻ were significantly associated with outdoor O₃ and NO₂. The association between outdoor EC and NO₂ in the summer was also significant. Similarly, indoors, PM_{2.5} and SO₄²⁻ were significantly associated with both O₃ and NO₂ in the summer season. The only significant association observed within the indoor environment in the winter season was that between EC and NO₂, however. Personal PM_{2.5} exposures were also significantly related to personal O₃ and NO₂ exposures in the summer, while personal SO₄²⁻ was only significantly associated with personal O₃ (and not NO₂) during that season. In the winter, personal exposures to the particle measures were significantly associated with personal exposures to NO₂.

4.3.2.6 Associations among Personal, Outdoor, Indoor, and Ambient Pollutant Concentrations

Associations between personal exposures and ambient concentrations for the children's cohort differed by particulate measure and by season (Table 94). Personal-ambient associations were strong for SO₄²⁻, a regional pollutant with few indoor sources, with greater slopes observed during the summer than during the winter. Significant personal-ambient associations were also found for PM_{2.5} when data from both seasons were analyzed together, with slopes again greater in summer as compared to winter. For EC, however, associations between personal exposures and corresponding ambient concentrations were only significant in the winter season. Greater slopes for SO₄²⁻ and PM_{2.5} in the summer as compared to the winter are consistent with the fact that the effective penetration efficiency of ambient particles is generally greater during the summer due to increased indoor ventilation. Stronger summertime associations for SO₄²⁻ and PM_{2.5} as compared to EC may be attributed the fact that SO₄²⁻ comprises a large fraction of PM_{2.5}, especially in the summer, and to the fact that EC tends to be smaller and thus may have a lower effective penetration efficiency. Both summer and winter slopes in Steubenville are lower than those found in our exposure studies of children in Baltimore and in Boston.

Personal-ambient associations for O₃ and NO₂ also varied seasonally, with similar patterns as observed for PM_{2.5} and SO₄²⁻. In the summer, personal O₃ and NO₂ exposures were significantly associated with their respective ambient concentrations, although the slope was low for O₃. In contrast, in the winter, the slopes describing the personal-ambient associations were insignificant for both O₃ and NO₂. Since ventilation was greater during the summer months, significant slopes in the summer but not in the winter suggest that ambient gas concentrations are strong proxies of their corresponding gaseous exposures only when individuals spend time in well-ventilated environments.

Table 93. Mixed model results describing associations among particle and gas concentrations measured in the indoor, personal, and outdoor microenvironments for the children’s cohort.*

Comparison			Winter					Summer				
			Slope	SE	t-stat	Int.	SE	Slope	SE	t-stat	Int.	SE
Indoor	PM _{2.5}	O ₃	0.51	0.31	1.66	7.90	1.02	0.36	0.13	2.71	17.27	1.59
		NO ₂	0.09	0.11	0.82	8.64	1.10	0.48	0.19	2.51	16.51	1.76
		SO ₂	0.44	0.66	0.67	9.57	0.70	0.09	0.34	0.27	18.66	1.70
	SO ₄ ²⁻	O ₃	0.03	0.06	0.44	1.93	0.24	0.16	0.05	3.13	5.20	0.59
		NO ₂	0.00	0.03	-0.10	2.04	0.28	0.32	0.07	4.66	4.25	0.66
		SO ₂	0.23	0.15	1.56	2.13	0.18	-0.13	0.13	-0.95	6.00	0.62
	EC	O ₃	0.02	0.04	0.66	0.54	0.12	-0.01	0.02	-0.37	0.75	0.18
		NO ₂	0.03	0.01	1.99	0.39	0.13	-0.01	0.03	-0.51	0.77	0.20
		SO ₂	-0.03	0.08	-0.43	0.59	0.09	0.08	0.05	1.70	0.61	0.17
Personal	PM _{2.5}	O ₃	-0.09	0.29	-0.30	12.79	1.22	0.56	0.16	3.47	17.58	1.64
		NO ₂	0.43	0.17	2.58	9.18	1.52	0.50	0.19	2.61	17.07	1.83
		SO ₂	0.07	0.83	0.09	12.56	0.94	-0.42	0.29	-1.47	20.65	1.69
	SO ₄ ²⁻	O ₃	0.04	0.04	0.98	1.82	0.21	0.35	0.05	6.70	4.71	0.49
		NO ₂	0.06	0.03	2.06	1.52	0.26	0.11	0.07	1.62	5.55	0.72
		SO ₂	0.11	0.13	0.84	2.00	0.16	-0.17	0.10	-1.70	6.50	0.64
	EC	O ₃	0.00	0.01	0.18	0.62	0.06	-0.06	0.06	-0.99	1.43	0.49
		NO ₂	0.01	0.01	1.99	0.52	0.07	-0.07	0.06	-1.11	1.53	0.52
		SO ₂	0.00	0.03	0.05	0.63	0.05	0.13	0.09	1.40	0.90	0.42
Outdoor	PM _{2.5}	O ₃	-0.25	0.07	-3.41	18.62	1.63	0.57	0.13	4.34	9.61	3.72
		NO ₂	0.79	0.10	7.59	7.67	0.93	0.66	0.18	3.64	20.31	2.03
		SO ₂	0.50	0.17	2.98	12.93	0.67	-0.63	0.37	-1.70	25.33	1.87
	SO ₄ ²⁻	O ₃	-0.06	0.02	-3.06	4.72	0.45	0.18	0.06	3.06	4.20	1.68
		NO ₂	0.19	0.03	6.46	2.13	0.29	0.20	0.08	2.39	7.56	0.93
		SO ₂	0.13	0.05	2.94	3.41	0.24	-0.14	0.16	-0.89	9.00	0.84
	EC	O ₃	-0.02	0.00	-4.30	0.94	0.09	0.00	0.00	0.26	0.65	0.09
		NO ₂	0.05	0.01	10.52	0.19	0.04	0.03	0.00	10.55	0.47	0.03
		SO ₂	0.04	0.01	4.81	0.53	0.03	-0.01	0.01	-0.74	0.69	0.04

*bold numbers indicate p-values ≤ 0.05. Slopes and intercepts were calculated using season-specific mixed models.

When analyzed by participant, considerable inter-personal variability in the associations between personal exposures and ambient concentrations was found for each of the particulate measures, even for SO₄²⁻ and even during the summer when overall associations were strong. Median individual-specific R² values differed by participant (Figures 65-67). Results indicate that the ability of ambient concentrations to reflect personal exposures varies by child and by particulate component.

Table 94. Associations between personal exposures and ambient concentrations of particle and gaseous pollutants monitored for the children’s cohort.

Pollutant		Both seasons			Winter			Summer		
		Estimate	StdErr	p-value	Estimate	StdErr	p-value	Estimate	StdErr	p-value
PM _{2.5} (ug/m ³)	Intercept	6.01	1.65	0.00	10.16	2.47	0.00	5.32	2.45	0.05
	Slope	0.54	0.06	<.0001	0.25	0.14	0.08	0.58	0.08	<.0001
SO ₄ ²⁻ (ug/m ³)	Intercept	0.08	0.34	0.81	0.43	0.27	0.14	0.58	0.58	0.34
	Slope	0.60	0.03	<.0001	0.37	0.05	<.0001	0.59	0.04	<.0001
EC (ug/m ³)	Intercept	1.07	0.63	0.10	0.37	0.07	0.00	2.10	1.32	0.13
	Slope	-0.05	0.67	0.94	0.41	0.07	<.0001	-0.94	1.42	0.51
O ₃ (ppb)	Intercept	1.54	0.75	0.05	2.94	0.67	0.00	2.13	1.24	0.11
	Slope	0.08	0.02	<.0001	-0.01	0.03	0.71	0.09	0.03	0.0009
NO ₂ (ppb)	Intercept	4.59	0.57	<.0001	7.27	0.81	<.0001	3.39	0.78	0.00
	Slope	0.31	0.05	<.0001	0.10	0.07	0.12	0.36	0.06	<.0001

Note: Slopes and intercepts calculated using mixed effect models.

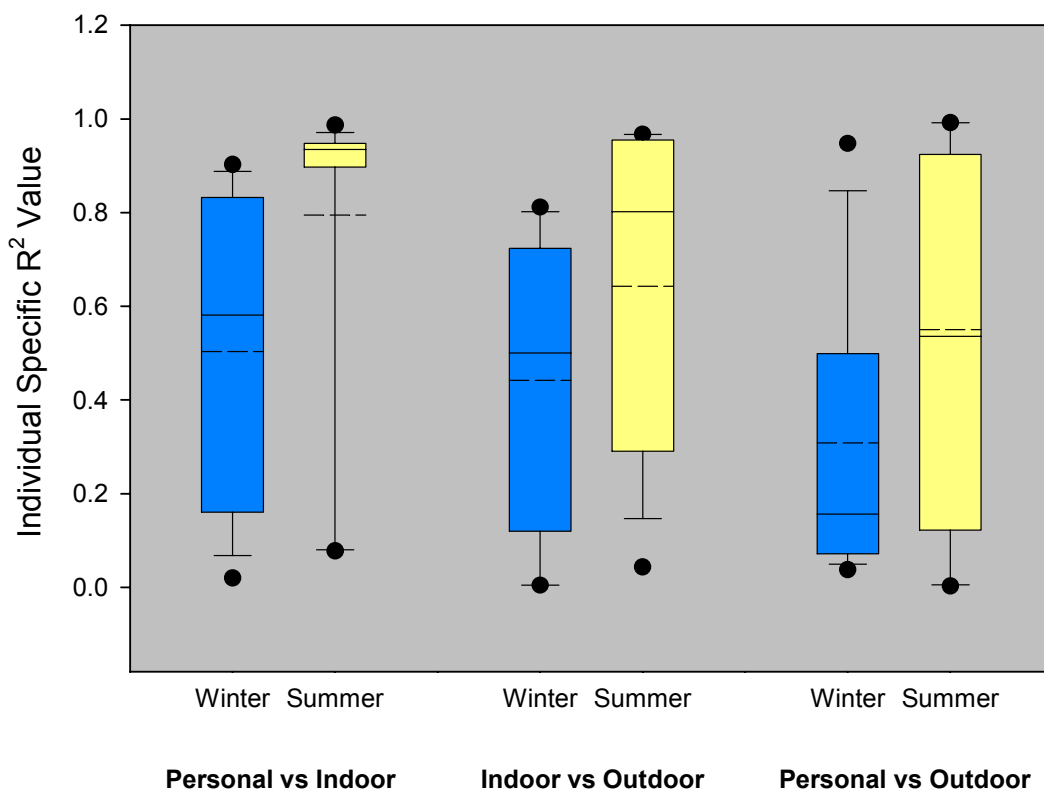


Figure 65. Individual-specific R² values for associations among personal, indoor, and outdoor PM_{2.5} concentrations measured for the children’s cohort.

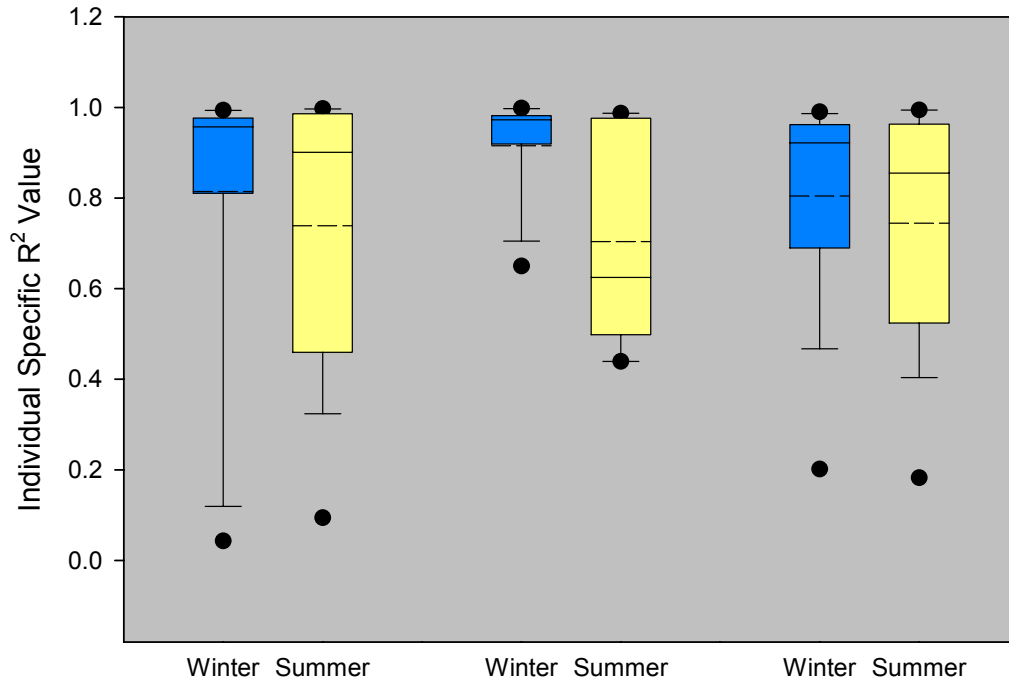


Figure 66. Individual-specific R² values for associations among personal, indoor, and outdoor SO₄²⁻ concentrations measured for the children's cohort.

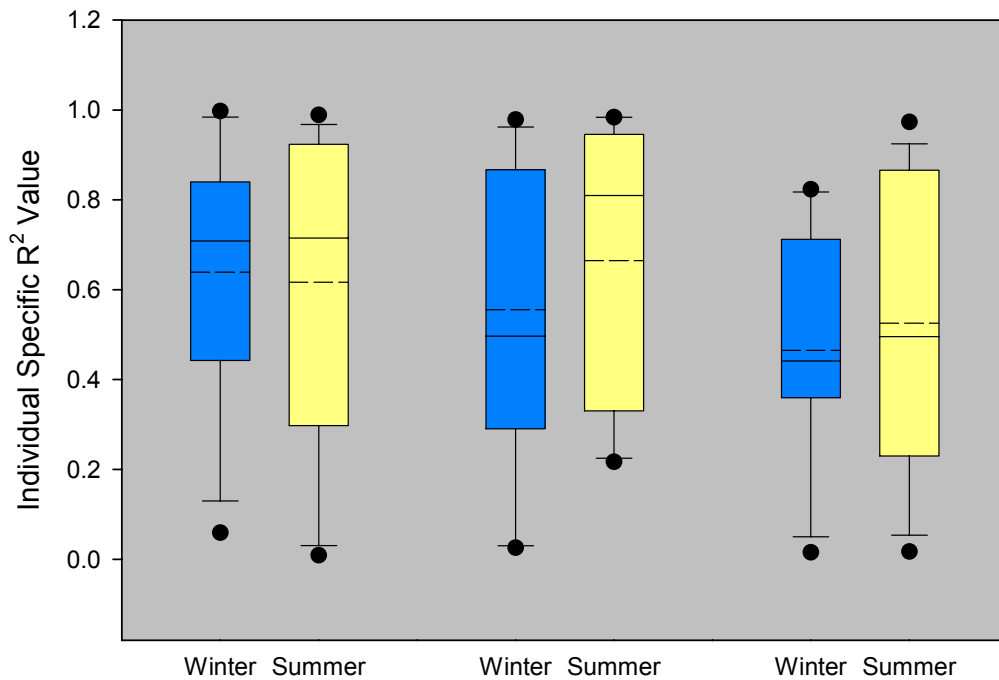


Figure 67. Individual-specific R² values for associations among personal, indoor, and outdoor EC concentrations measured for the children's cohort.

4.3.2.7 Cross-Pollutant Associations

Associations between ambient particle concentrations and personal gas exposures for the children are given in Table 95. Significant associations were found between personal O₃ and ambient PM_{2.5}, personal O₃ and ambient SO₄²⁻, and personal NO₂ and ambient EC, but all occurred only in the summer season. All winter cross-pollutant associations of ambient particle concentrations with personal gas exposures were not statistically significant. As shown with the adult data, slopes were relatively low, with the possible exception of associations between the gases and EC.

In contrast to the summertime associations observed between ambient particle measures and personal gas exposures, ambient gas concentrations were not significantly associated with corresponding personal exposures to any of the particle measures in the summer. In the winter, however, associations between ambient NO₂ concentrations and personal particle exposures were significant. Ambient O₃ concentrations were not significantly associated with personal exposures to any of the particle measures in either season (Table 95).

Results indicate that during the summer, which was characterized by more frequent open window usage by the participants, ambient particle concentrations were strong proxies of children's personal exposures to gases from similar sources, such as was observed with the traffic-related pollutants NO₂ and EC and with the regional pollutants O₃, SO₄²⁻, and PM_{2.5} (likely because of its large mass fraction of SO₄²⁻). Despite this, the low slopes further suggest that ambient particle concentrations are much better proxies of children's exposures to the particles themselves. Ambient gas concentrations, with the exception of ambient NO₂ in the winter, were poor proxies of personal particle exposures.

Table 95. Cross-pollutant associations between ambient particle and gas concentrations and personal gas and particle exposures for the children's cohort.

Comparison		Winter					Summer				
Personal	Ambient	N	β	SE	t-stat	Int.	N	β	SE	t-stat	Int.
O ₃	PM _{2.5}	69	0.02	0.04	0.44	2.94	66	0.25	0.06	4.11	0.81
NO ₂		75	0.06	0.04	1.24	6.65	81	0.04	0.04	0.99	5.34
O ₃	SO ₄ ²⁻	69	0.18	0.14	1.28	2.55	63	0.46	0.13	3.53	2.65
NO ₂		75	0.05	0.15	0.30	7.34	78	0.02	0.08	0.22	6.34
O ₃	EC	67	0.37	0.78	0.48	3.01	66	1.02	3.07	0.33	6.61
NO ₂		73	1.53	0.78	1.96	6.29	81	4.03	1.80	2.25	3.12
PM _{2.5}	O ₃	76	0.12	0.08	1.47	11.11	87	-0.01	0.05	-0.23	21.81
	NO ₂	79	0.46	0.19	2.35	9.89	76	0.06	0.14	0.43	20.84
SO ₄ ²⁻	O ₃	78	-0.01	0.01	-0.94	2.08	87	0.01	0.02	0.55	6.47
	NO ₂	81	0.12	0.02	4.98	0.80	77	-0.06	0.06	-1.12	7.56
EC	O ₃	78	0.00	0.00	0.13	0.60	85	-0.02	0.02	-1.29	2.10
	NO ₂	81	0.03	0.01	5.50	0.35	74	-0.01	0.03	-0.45	1.06

4.3.2.8 Factors Affecting Associations

4.3.2.8.1 Ventilation

Table 96 presents mixed model results showing the effects of ventilation (window and air conditioning use) on the associations between indoor and ambient pollutant concentrations in the summer season. A three-level variable for window use was created by using information obtained from questionnaires. The number of hours of open windows and doors over the sampling period were summed and categorized by approximate tertiles. The levels included “Lo,” representing that no windows or doors were open during the sampling period, “Med,” representing that the sum of time for which windows and doors were open during the sampling period was 24 hours or less, and “Hi,” representing that the sum of time for which windows and doors were open during the sampling period was greater than 24 hours. The use of air conditioning was dichotomized into “any use” and “no use.”

Results for the children’s cohort are similar to those discussed earlier for the older adult cohort in that indoor-ambient slopes increased with a change from “Lo” to “Hi” ventilation. The slopes for all pollutants were significant and statistically greater for indoor environments characterized by open windows. These results indicate that the effective penetration efficiency for all pollutants was higher for well-ventilated as compared to poorly ventilated indoor environments.

Results were less consistent when AC use was used as the measure of ventilation. For $PM_{2.5}$, SO_4^{2-} , and to a lesser extent O_3 , the use of air conditioning increased the slopes of the indoor-ambient associations. This is contrary to previous findings and expected findings. The results may indicate that the air conditioner use categories did not accurately reflect home ventilation conditions. For EC and NO_2 , the presence of AC decreased the slope of the indoor-ambient associations.

4.3.2.8.2 Stove Type

The types of stoves (gas or electric) found in the participants’ homes were examined to determine whether stove type affected the associations between indoor and ambient pollutant concentrations and between personal exposures and ambient concentrations. Results for the mixed models, stratified by season, are presented in Tables 97 and 98, respectively.

As was the case for the older adults, intercepts of the indoor-on-ambient associations for NO_2 were greater in the winter for participants living in homes with gas stoves as compared to those living in homes with electric stoves; however, these differences were not significant. Indoor-ambient NO_2 intercepts differed by stove type in the summer as well, although differences were extremely small. Differences between the intercepts of indoor-on-ambient regression lines for homes with gas stoves and homes with electric stoves were significant for $PM_{2.5}$, SO_4^{2-} , and EC in both seasons, with greater intercepts generally found for homes with gas stoves than for those with electric stoves, especially in the winter. Intercepts were often insignificant, however, making these results questionable.

Table 96. Effect of ventilation on indoor-ambient associations in the summer season for the children's cohort.*

Pollutant	Factor	Level	Slope	SE	t-stat	Int.
PM _{2.5}	Window Use	Lo	0.28	0.13	2.17	5.29
		Med	0.27	0.27	0.88	7.09
		Hi	0.73	0.08	9.42	3.24
SO ₄ ²⁻		Lo	0.26	0.08	3.12	1.17
		Med	0.28	0.21	1.33	2.22
		Hi	0.87	0.05	16.61	-1.14
EC		Lo	0.71	0.51	1.40	-0.10
		Med	-0.68	0.16	-4.14	0.97
		Hi	0.75	0.09	8.69	0.11
O ₃		Lo	0.00	0.33	0.00	-2.84
		Med	0.00	0.13	-0.01	-2.77
		Hi	0.63	2.51	6.62	-6.87
NO ₂	Lo	0.19	0.59	0.32	1.06	
	Med	0.02	0.14	0.16	4.26	
	Hi	0.80	0.07	11.42	0.88	
PM _{2.5}	AC Use	None	0.40	0.10	4.14	6.26
		Any	0.76	0.09	8.45	3.57
SO ₄ ²⁻		None	0.43	0.07	5.93	1.47
		Any	0.88	0.06	14.15	-0.92
EC		None	0.62	0.14	4.51	0.08
		Any	0.33	0.12	2.69	0.46
O ₃		None	0.22	0.11	1.98	-3.40
		Any	0.23	0.16	1.45	2.47
NO ₂		None	0.72	0.11	6.69	-0.20
		Any	0.52	0.09	5.65	3.02

*Bold numbers indicate p-values ≤ 0.05. Slopes and intercepts calculated using season-specific mixed models that included an interaction term for ventilation. Air conditioner use was classified as “none” if air conditioners were not used during the 24-hr monitoring period and “any” if air conditioners were used at all during the 24-hr monitoring period.

Table 97. Indoor vs. ambient concentrations by stove type for the children's cohort.*

Pollutant	Fuel	Winter					Summer				
		N	β	SE	t-stat	Int.	N	β	SE	t-stat	Int.
PM _{2.5}	gas	19	0.59	0.23	2.63	7.57	24	0.77	0.24	3.21	8.99
	electric	56	0.35	0.10	3.44	2.45	74	0.50	0.07	6.75	3.40
SO ₄ ²⁻	gas	19	0.59	0.07	8.50	0.63	24	0.90	0.18	5.05	0.21
	electric	56	0.45	0.03	13.70	-0.10	74	0.50	0.05	10.81	0.58
EC	gas	19	0.46	0.14	3.16	0.38	24	3.04	1.18	2.57	-0.95
	electric	56	0.36	0.05	6.72	0.22	76	0.45	0.62	0.72	0.21
O ₃	gas	21	0.00	0.03	-0.08	2.33	28	0.27	0.07	3.89	0.26
	electric	60	-0.04	0.02	-2.04	3.15	71	0.14	0.02	6.52	0.82
NO ₂	gas	21	0.03	0.13	0.23	9.66	26	0.73	0.17	4.21	2.78
	electric	63	0.20	0.07	2.70	6.42	71	0.33	0.06	5.93	2.68

*shaded cells represent significant interaction effect; bold numbers indicate p-value ≤ 0.05

Stove type modified the slopes of the indoor-on-ambient associations for SO_4^{2-} and NO_2 in the summer, with higher slopes observed for homes with gas stoves. Reasons for these higher slopes are not known, but may suggest that the results are spurious or that homes with gas stoves had different characteristics that were associated with increased penetration efficiency.

Table 98. Personal vs. ambient concentrations by stove type for the children's cohort.*

Pollutant	Fuel	Winter					Summer				
		N	β	SE	t-stat	Int.	N	β	SE	t-stat	Int.
PM _{2.5}	gas	19	0.62	0.27	2.35	9.44	24	0.43	0.24	1.76	14.53
	electric	56	0.17	0.15	1.13	9.88	75	0.57	0.07	7.84	3.39
SO ₄ ²⁻	gas	19	0.62	0.08	7.33	0.57	24	0.86	0.14	6.08	-0.02
	electric	56	0.33	0.04	7.48	0.27	74	0.55	0.04	15.32	0.57
EC	gas	19	0.70	0.14	5.10	0.30	24	-3.89	3.18	-1.23	6.79
	electric	56	0.37	0.07	5.37	0.34	74	0.59	1.32	0.45	0.12
O ₃	gas	21	0.03	0.04	0.81	1.89	28	0.15	0.08	1.91	-1.49
	electric	60	-0.05	0.03	-1.42	3.83	71	0.06	0.03	2.38	4.58
NO ₂	gas	21	0.11	0.14	0.80	9.66	26	0.78	0.20	3.96	1.66
	electric	63	0.10	0.08	1.31	6.45	71	0.32	0.06	5.09	3.54

*shaded cells represent significant interaction effect; bold numbers indicate p-value ≤ 0.05

Stove type had similar confusing effects on the associations between personal exposures and ambient concentrations, with individuals living in homes with gas stoves having higher intercepts for NO_2 in the winter but not summer and higher slopes for SO_4^{2-} in both seasons, EC in the winter, and NO_2 in the summer. The interpretation of these results is unclear.

4.3.2.8.3 Activity Patterns

The impact of time spent in different microenvironments on the relationship between personal exposures and ambient concentrations was examined for the children's cohort. Because participants spent the vast majority of their time indoors, time spent in other micro-environments was generally extremely low. For example, the median fraction of time spent outdoors equaled only 0.03 (or 43 minutes over 24 hours); even less time was spent in other microenvironments. As a result, findings of modification by time spent in microenvironments should be interpreted with caution.

As shown in Table 99, time spent outdoors modified the personal-ambient associations for the particle measures, as evidenced by the significantly greater slopes in both seasons for SO_4^{2-} for individuals spending a larger fraction of their time outdoors (as compared to the median). Although differences were not statistically significant, similar trends were found for PM_{2.5} in the summer and EC in the winter. Personal-ambient intercepts for PM_{2.5} and EC in both seasons were also higher, although not statistically significant, for individuals who spent more time outdoors. Time spent outdoors did not have consistent effects on the associations between personal exposures and ambient concentrations for the gases O₃ and NO_2 , although intercepts for NO_2 in both seasons and for O₃ in the summer were significantly higher for individuals spending more time outdoors.

In the summer, time spent indoors near an open window was found to modify the association between personal exposures and ambient concentrations for most of the measured pollutants, with significantly higher personal-ambient PM_{2.5}, SO_4^{2-} , and O₃ slopes for individuals who spent greater-than-average (median) amounts of time indoors near open windows (Table 100).

Table 99. Personal vs. ambient concentrations by fraction of time spent outdoors for the children's cohort.*

Pollutant	Status	Winter					Summer				
		N	β	SE	t-stat	Int.	N	β	SE	t-stat	Int.
PM _{2.5}	Lo**	87	0.17	0.13	1.30	8.67	20	0.24	0.28	0.86	6.23
	Hi	14	0.06	0.29	0.19	18.64	80	0.57	0.07	8.11	7.19
SO ₄ ²⁻	Lo	88	0.51	0.05	10.23	0.10	19	0.25	0.16	1.56	1.43
	Hi	14	0.80	0.13	6.30	-0.23	81	0.64	0.04	16.89	0.83
EC	Lo	85	0.34	0.09	3.57	0.42	19	0.05	3.73	0.01	0.52
	Hi	14	0.56	0.16	3.37	0.51	79	-1.22	1.27	-0.95	2.13
O ₃	Lo	96	-0.05	0.03	-1.37	3.57	21	0.06	0.17	0.36	-0.05
	Hi	14	0.07	0.08	0.83	1.52	79	0.01	0.08	0.11	6.33
NO ₂	Lo	91	0.10	0.09	1.09	6.91	21	0.53	0.20	2.62	0.39
	Hi	14	-0.01	0.22	-0.04	11.05	84	0.58	0.08	7.63	2.62

*shaded cells represent significant interaction effect; bold numbers indicate p-value ≤ 0.05

** "Lo" \leq median fraction of time outside (0.03); "Hi" $>$ median fraction of time outside

Table 100. Personal vs. ambient concentrations by fraction of time spent indoors near open windows for the children's cohort.*

Pollutant	Status	Winter					Summer				
		N	β	SE	t-stat	Int.	N	β	SE	t-stat	Int.
PM _{2.5}	Lo	66	0.22	0.16	1.37	9.00	26	0.29	0.12	2.37	7.07
	Hi	34	0.08	0.18	0.45	10.93	74	0.68	0.08	9.02	5.22
SO ₄ ²⁻	Lo	67	0.55	0.06	8.72	-0.05	27	0.30	0.05	5.91	1.65
	Hi	34	0.53	0.07	7.14	0.35	73	0.79	0.04	22.07	-0.19
EC	Lo	65	0.49	0.11	4.56	0.35	26	0.24	2.96	0.08	0.35
	Hi	33	0.28	0.13	2.19	0.49	73	-1.20	1.30	-0.93	2.17
O ₃	Lo	75	-0.06	0.04	-1.61	3.83	28	-0.12	0.13	-0.99	6.72
	Hi	35	0.03	0.05	0.51	2.31	72	0.23	0.09	2.52	0.93
NO ₂	Lo	70	0.09	0.11	0.80	6.84	28	0.19	0.23	0.81	3.16
	Hi	35	0.08	0.13	0.64	8.66	77	0.64	0.07	8.80	2.41

*shaded cells represent significant interaction effect; bold numbers indicate p-value ≤ 0.05

** "Lo" \leq than median fraction of time near an open window (0.01); "Hi" $>$ median fraction of time

Although not significant, spending time near an open window was found to have a similar effect on the personal-ambient NO₂ association. Spending time near open windows in the winter had no significant effects on the associations between personal exposures and ambient concentrations for any of the measured pollutants, which may result from the fact that individuals tended to spend little time near open windows in this season. The summertime results followed a pattern similar to that observed in Table 96 for window usage, where the indoor-outdoor slope was greatest for indoor environments with open windows.

As shown in Table 101, children spending a greater fraction of time traveling by car (as compared to the median amount of time) had significantly lower personal-ambient slopes for PM_{2.5}, SO₄²⁻, and NO₂ in the summer and for EC in the winter than did those spending lesser amounts of time in cars. Lower personal-ambient slopes for individuals spending more time in transit suggest that, for a given increase in ambient concentrations, the increase in personal exposures is smaller for these individuals. However, since children spent a very small fraction of time traveling by car, it is likely that time in transit is acting as a proxy for general activity or another modifying variable.

Table 101. Personal vs. ambient concentrations by fraction of time spent traveling by car for the children’s cohort.*

Pollutant	Status	Winter					Summer				
		N	β	SE	t-stat	Int.	N	β	SE	t-stat	Int.
PM _{2.5}	Lo	53	0.19	0.17	1.11	12.54	48	0.78	0.09	8.32	1.81
	Hi	48	0.11	0.17	0.66	7.43	53	0.36	0.09	4.05	9.17
SO ₄ ²⁻	Lo	52	0.59	0.07	8.00	-0.12	48	0.85	0.05	17.97	-0.72
	Hi	49	0.52	0.06	8.27	0.18	53	0.44	0.04	10.40	1.40
EC	Lo	50	0.52	0.10	5.30	0.35	47	0.89	1.80	0.50	0.01
	Hi	49	0.13	0.14	0.93	0.55	53	-2.17	1.59	-1.36	2.91
O ₃	Lo	54	-0.05	0.04	-1.15	3.80	47	-0.06	0.10	-0.61	7.77
	Hi	56	0.01	0.05	0.15	2.28	53	0.15	0.11	1.37	1.05
NO ₂	Lo	54	0.22	0.11	1.92	7.19	49	0.77	0.11	6.80	1.28
	Hi	51	-0.08	0.13	-0.63	7.95	56	0.48	0.09	5.46	2.37

*shaded cells represent significant interaction effect; bold numbers indicate p-value ≤ 0.05

** “Lo” ≤ median fraction of time in car (0.02); “Hi” > median fraction of time in car

4.3.2.9 Summary of Elemental Data

Table 102 summarizes water-extractable element concentrations measured as part of the children’s exposure assessment. Water-extractable concentrations of several elements were below their respective limits of detection. Those elements that had median water-extractable concentrations below the LOD included: Al (winter only), Cu, Mn, Ti, and Zn (winter only). For many elements, especially those associated with smaller particle sizes, median outdoor and ambient concentrations tended to be greater than indoor and personal levels. Personal exposures to elements that are commonly associated with larger particle sizes, such as Ca and K, were generally greater than indoor concentrations of these elements and comparable to or greater than outdoor concentrations.

Table 102. Water-extractable element concentrations (ng/m³) measured in personal, indoor, outdoor, and ambient environments for the children’s cohort.

Element	Location	Winter				Summer			
		N	10th	50th	90th	N	10th	50th	90 th
Al	Personal	103	< LOD	< LOD	32.14	101	< LOD	9.75	101.92
	Indoor	107	< LOD	< LOD	23.62	98	< LOD	15.91	57.35
	Outdoor	103	< LOD	< LOD	21.48	99	< LOD	17.81	47.70
	Central	25	< LOD	1.71	33.89	26	< LOD	20.14	67.20
As	Personal	103	< LOD	0.57	1.26	102	< LOD	0.87	2.77
	Indoor	107	< LOD	0.59	1.29	103	0.43	1.43	3.15
	Outdoor	103	< LOD	0.94	2.49	103	0.76	1.91	3.96
	Central	25	< LOD	1.14	4.20	27	0.51	2.09	4.33
Ba	Personal	103	0.35	1.14	4.05	84	0.72	1.72	4.10
	Indoor	107	0.16	0.80	1.76	79	0.33	1.08	3.80
	Outdoor	103	0.52	1.22	2.41	75	0.83	1.89	4.54
	Central	25	0.49	1.61	2.62	23	1.00	2.36	4.77
Cd	Personal	103	0.11	0.24	0.54	84	< LOD	0.18	0.60
	Indoor	101	0.06	0.15	0.36	79	< LOD	0.19	0.75
	Outdoor	95	0.11	0.22	0.62	75	< LOD	0.32	0.78
	Central	25	0.12	0.28	0.95	23	0.05	0.34	0.88

Table 102. (continued)

Element	Location	Winter				Summer			
		N	10th	50th	90th	N	10th	50th	90 th
Ca	Personal	85	31.72	149.09	541.47	95	< LOD	92.36	208.02
	Indoor	89	31.93	60.29	115.95	93	< LOD	57.88	127.38
	Outdoor	85	< LOD	42.11	88.50	91	< LOD	71.64	145.72
	Central	20	20.98	50.35	122.52	23	< LOD	82.96	222.78
Co	Personal	102	< LOD	< LOD	0.07	102	< LOD	0.02	0.07
	Indoor	106	< LOD	0.01	0.06	103	< LOD	0.02	0.05
	Outdoor	102	< LOD	0.02	0.09	103	0.01	0.04	0.07
	Central	25	< LOD	0.02	0.09	27	0.02	0.04	0.07
Cu	Personal	103	< LOD	< LOD	< LOD	102	< LOD	< LOD	6.41
	Indoor	107	< LOD	< LOD	< LOD	103	< LOD	< LOD	2.87
	Outdoor	103	< LOD	< LOD	0.04	103	< LOD	< LOD	2.89
	Central	25	< LOD	< LOD	< LOD	27	< LOD	< LOD	9.57
Fe	Personal	103	< LOD	< LOD	56.70	81	< LOD	18.72	58.31
	Indoor	107	< LOD	1.99	35.44	73	< LOD	20.57	70.63
	Outdoor	103	< LOD	11.94	41.98	75	7.44	31.84	156.00
	Central	25	< LOD	23.52	70.34	21	17.47	54.20	186.26
Pb	Personal	103	1.51	2.78	9.01	70	1.54	4.91	14.01
	Indoor	107	1.06	2.71	8.00	64	1.60	4.47	14.12
	Outdoor	103	2.23	4.60	14.23	68	2.39	6.98	25.24
	Central	25	2.32	5.43	17.44	17	3.10	7.05	14.65
Mg	Personal	103	4.08	12.35	45.93	102	7.52	15.27	39.54
	Indoor	107	3.49	8.87	27.63	103	5.02	11.42	38.55
	Outdoor	103	5.11	11.98	34.99	103	6.80	21.84	48.34
	Central	25	5.23	14.25	62.22	27	8.62	27.24	71.95
Mn	Personal	103	< LOD	< LOD	9.49	102	< LOD	< LOD	11.03
	Indoor	107	< LOD	< LOD	7.68	103	< LOD	< LOD	10.15
	Outdoor	103	< LOD	3.75	17.56	103	< LOD	4.19	15.52
	Central	25	< LOD	6.20	24.32	27	< LOD	5.30	13.15
Ni	Personal	103	< LOD	0.78	5.69	98	< LOD	0.40	2.94
	Indoor	107	< LOD	0.48	4.08	95	< LOD	0.35	0.91
	Outdoor	103	< LOD	0.50	1.86	95	< LOD	0.37	1.42
	Central	25	< LOD	0.35	1.18	25	< LOD	0.36	1.59
K	Personal	103	12.50	78.43	191.78	93	< LOD	61.20	160.92
	Indoor	107	20.99	53.59	116.43	92	< LOD	43.57	121.75
	Outdoor	103	27.27	61.86	126.11	93	< LOD	72.85	147.53
	Central	25	24.83	77.01	172.99	24	< LOD	59.77	153.11
Se	Personal	103	< LOD	1.19	2.89	102	< LOD	1.50	4.51
	Indoor	107	< LOD	1.04	2.48	103	0.40	1.37	4.37
	Outdoor	103	0.99	2.74	9.06	103	1.24	2.80	6.14
	Central	25	1.43	3.11	9.29	27	1.69	3.21	5.94
Na	Personal	103	< LOD	53.89	193.53	102	< LOD	47.41	193.77
	Indoor	107	< LOD	36.16	71.45	103	8.62	43.13	171.42
	Outdoor	103	< LOD	25.12	81.48	103	9.20	34.55	185.47
	Central	25	< LOD	37.35	133.43	27	8.89	60.15	314.98

Table 102. (continued)

Element	Location	Winter				Summer			
		N	10th	50th	90th	N	10th	50th	90 th
Sn	Personal	95	< LOD	0.25	0.83	84	< LOD	0.24	0.90
	Indoor	91	< LOD	0.12	0.83	79	< LOD	0.18	0.70
	Outdoor	87	< LOD	< LOD	0.24	75	< LOD	0.14	0.72
	Central	23	< LOD	< LOD	0.21	23	< LOD	0.15	0.62
Ti	Personal	103	< LOD	< LOD	< LOD	102	< LOD	< LOD	< LOD
	Indoor	107	< LOD	< LOD	< LOD	103	< LOD	< LOD	< LOD
	Outdoor	103	< LOD	< LOD	< LOD	103	< LOD	< LOD	0.50
	Central	25	< LOD	< LOD	< LOD	27	< LOD	< LOD	3.25
V	Personal	103	< LOD	< LOD	1.20	102	< LOD	< LOD	1.54
	Indoor	107	< LOD	< LOD	1.46	103	< LOD	0.63	1.67
	Outdoor	103	< LOD	0.64	3.00	103	0.45	1.25	4.07
	Central	25	< LOD	0.61	3.26	27	0.47	1.61	4.17
Zn	Personal	103	< LOD	< LOD	39.49	102	< LOD	2.93	70.13
	Indoor	107	< LOD	< LOD	42.48	103	< LOD	13.50	85.73
	Outdoor	103	< LOD	5.51	57.95	103	< LOD	26.39	123.35
	Central	25	< LOD	11.00	58.82	27	2.03	36.79	112.71

Trends exhibited by acid-digestible element concentrations were similar to those exhibited by water-extractable element concentrations, with some elements again having median concentrations below their respective limits of detection (Table 103). The number of acid-digestible elements with median values below the LOD was larger than the number of water-extractable elements, and included As, Ba, Cd, and Pb in the winter, Ca, Co, and Ni in the summer, and Cu in both seasons. Median personal exposures to the measured elements tended to be greater than or comparable to concentrations measured indoors and outdoors.

Table 103. Acid-digestible element concentrations (ng/m³) measured in personal, indoor, outdoor, and ambient environments for the children's cohort.

Element	Location	Winter				Summer			
		N	10th	50th	90th	N	10th	50 th	90th
Al	Personal	68	< LOD	196.47	518.63	67	< LOD	145.52	418.54
	Indoor	76	< LOD	99.13	223.05	62	< LOD	126.52	301.15
	Outdoor	74	< LOD	92.97	203.77	59	< LOD	80.03	336.96
	Central	22	< LOD	42.89	135.26	16	< LOD	120.44	431.20
As	Personal	105	< LOD	< LOD	< LOD	101	< LOD	< LOD	3.91
	Indoor	107	< LOD	< LOD	< LOD	100	< LOD	1.44	2.90
	Outdoor	100	< LOD	< LOD	1.12	101	< LOD	1.81	3.61
	Central	25	< LOD	< LOD	3.13	27	< LOD	2.16	3.32
Ba	Personal	105	< LOD	< LOD	8.09	101	< LOD	< LOD	7.85
	Indoor	107	< LOD	< LOD	5.74	100	< LOD	0.34	4.95
	Outdoor	100	< LOD	< LOD	3.63	101	< LOD	2.23	5.48
	Central	25	< LOD	< LOD	3.99	27	< LOD	2.99	7.45
Cd	Personal	105	< LOD	< LOD	1.00	101	< LOD	0.34	1.63
	Indoor	107	< LOD	< LOD	0.50	100	< LOD	0.41	1.06
	Outdoor	100	< LOD	0.39	0.80	101	< LOD	0.53	1.19
	Central	25	< LOD	0.30	0.99	27	< LOD	0.62	1.80

Table 103. (continued)

Element	Location	Winter				Summer			
		N	10th	50th	90th	N	10th	50 th	90th
Ca	Personal	53	< LOD	156.26	738.69	24	< LOD	< LOD	35.87
	Indoor	44	< LOD	52.74	418.70	25	< LOD	< LOD	101.59
	Outdoor	40	< LOD	56.91	617.96	22	< LOD	< LOD	30.05
	Central	10	< LOD	80.16	340.79	7	< LOD	< LOD	52.27
Co	Personal	18	< LOD	< LOD	1.82	38	< LOD	< LOD	0.33
	Indoor	25	0.13	0.44	0.79	34	< LOD	< LOD	0.14
	Outdoor	23	< LOD	0.43	0.52	32	< LOD	< LOD	< LOD
	Central	5	0.19	0.44	0.90	9	< LOD	< LOD	0.25
Cu	Personal	105	< LOD	< LOD	0.81	100	< LOD	< LOD	29.37
	Indoor	107	< LOD	< LOD	1.26	98	< LOD	< LOD	18.14
	Outdoor	100	< LOD	< LOD	4.62	97	< LOD	< LOD	21.12
	Central	25	< LOD	< LOD	1.67	26	< LOD	< LOD	10.09
Fe	Personal	104	51.01	187.47	485.14	87	< LOD	243.28	654.12
	Indoor	105	44.67	92.07	257.11	89	48.88	163.57	665.93
	Outdoor	98	50.27	119.21	471.04	87	73.02	215.22	771.94
	Central	24	52.45	149.39	551.35	22	101.84	256.18	748.65
Pb	Personal	105	< LOD	< LOD	26.12	101	< LOD	8.52	39.97
	Indoor	107	< LOD	< LOD	16.08	100	< LOD	7.97	24.21
	Outdoor	100	< LOD	0.58	18.20	101	< LOD	10.58	26.82
	Central	25	< LOD	< LOD	15.44	27	< LOD	9.88	29.21
Mg	Personal	93	3.94	35.25	74.43	79	< LOD	22.08	59.02
	Indoor	101	1.70	17.88	40.03	81	< LOD	12.15	46.17
	Outdoor	91	3.45	17.41	34.77	83	1.79	25.93	58.17
	Central	24	< LOD	21.64	67.18	22	10.19	37.73	69.06
Mn	Personal	74	< LOD	11.69	32.52	83	< LOD	7.49	20.04
	Indoor	78	2.39	7.47	16.24	82	< LOD	5.50	19.76
	Outdoor	75	2.60	10.55	26.69	80	< LOD	8.23	25.28
	Central	21	3.15	14.03	45.69	20	5.57	13.97	33.54
Ni	Personal	16	< LOD	3.85	32.10	69	< LOD	< LOD	12.59
	Indoor	15	< LOD	< LOD	1.91	65	< LOD	< LOD	3.56
	Outdoor	13	< LOD	1.43	3.27	66	< LOD	< LOD	13.88
	Central	2	< LOD	3.75	7.50	18	< LOD	< LOD	2.71
K	Personal	68	< LOD	118.82	317.29	61	31.07	119.70	256.52
	Indoor	72	5.09	82.24	198.04	54	21.27	90.25	204.13
	Outdoor	70	< LOD	87.09	189.89	55	33.35	98.63	249.74
	Central	20	< LOD	97.06	175.72	14	44.84	125.30	227.46
Se	Personal	0	n/a	n/a	n/a	21	< LOD	< LOD	4.47
	Indoor	0	n/a	n/a	n/a	24	< LOD	0.62	3.21
	Outdoor	0	n/a	n/a	n/a	24	< LOD	2.05	3.32
	Central	0	n/a	n/a	n/a	6	< LOD	2.46	6.17
Na	Personal	40	75.09	307.46	1083.92	59	< LOD	113.95	998.73
	Indoor	44	8.55	174.48	579.76	61	< LOD	89.98	303.28
	Outdoor	45	23.18	120.88	266.87	63	< LOD	62.49	243.36
	Central	14	24.95	103.09	271.22	17	< LOD	68.22	462.68

Table 103. (continued)

Element	Location	Winter				Summer			
		N	10th	50th	90th	N	10th	50 th	90th
Sn	Personal	105	< LOD	1.81	5.03	101	< LOD	2.52	6.43
	Indoor	107	< LOD	1.07	4.02	100	0.50	1.46	6.67
	Outdoor	100	< LOD	0.76	1.71	101	0.63	1.28	3.69
	Central	25	< LOD	1.04	3.16	27	0.69	1.45	3.72
Ti	Personal	101	< LOD	10.42	48.78	101	< LOD	< LOD	94.98
	Indoor	106	< LOD	6.42	25.05	100	< LOD	< LOD	63.81
	Outdoor	97	< LOD	5.89	24.42	101	< LOD	5.30	65.45
	Central	25	< LOD	4.70	25.66	27	< LOD	6.76	58.86
V	Personal	88	< LOD	0.33	4.73	49	< LOD	1.27	5.18
	Indoor	80	< LOD	0.06	1.63	43	< LOD	0.64	2.62
	Outdoor	80	< LOD	0.25	3.37	42	< LOD	0.78	2.78
	Central	20	< LOD	1.16	3.11	12	< LOD	1.21	3.68
Zn	Personal	102	< LOD	6.64	242.20	101	< LOD	34.49	247.39
	Indoor	103	< LOD	9.95	81.62	100	< LOD	38.92	160.34
	Outdoor	97	< LOD	17.99	79.31	101	< LOD	33.27	242.09
	Central	24	< LOD	24.11	113.15	27	< LOD	68.01	164.79

4.3.2.10 Associations between Personal Exposures, Indoor Concentrations, and Ambient Concentrations for Selected Elements

4.3.2.10.1 Personal Exposures and Ambient Elemental Concentrations

The associations between personal exposures and corresponding ambient concentrations for PM_{2.5} elements for the children’s cohort differed by element, extraction method, and season (Tables 104-105). These differences, however, did not follow a distinct pattern, as slopes and intercepts were higher in summer than in winter for some elements but were lower in summer than in winter for other elements. Furthermore, seasonal patterns in the personal-ambient association for the same element depended on whether the elemental concentration was determined in the water-extractable or acid-digestible PM_{2.5} fraction. It is likely that the observed inconsistencies in our findings largely reflect errors in the measurement and/or analysis techniques, as evidenced by the poor precisions and low concentrations observed for many of the elements.

4.3.2.10.2 Indoor and Ambient Elemental Concentrations

The associations between indoor concentrations and corresponding ambient concentrations for PM_{2.5} elements for the children’s cohort also differed by element, extraction method, and season (Tables 106-107). These differences, however, did not follow a distinct pattern, as slopes and intercepts were higher in summer than in winter for some elements but were lower in summer than in winter for other elements. Furthermore, indoor-ambient associations observed for the elements often differed from corresponding personal-ambient associations. These differences may reflect differences in measurement and method performance for the determination of indoor and personal elemental concentrations.

Table 104. Associations between personal exposures and ambient concentrations for select water-extractable elements for the children's cohort.*

Water-Extractable Element		Both seasons			Winter			Summer		
		estimate	s.e.	p-value	estimate	s.e.	p-value	estimate	s.e.	p-value
As	Intercept	0.13	0.12	0.31	0.26	0.12	0.06	-0.01	0.19	0.95
	Slope	0.39	0.04	<.0001	0.19	0.03	<.0001	0.52	0.05	<.0001
Ba	Intercept	3.70	1.67	0.04	1.06	0.98	0.30	6.81	3.14	0.05
	Slope	-0.24	0.65	0.71	0.50	0.48	0.30	-0.95	1.04	0.36
Cd	Intercept	0.14	0.04	0.00	0.07	0.03	0.06	0.18	0.07	0.02
	Slope	0.32	0.06	<.0001	0.50	0.07	<.0001	0.23	0.10	0.02
Fe	Intercept	18.39	5.87	0.00	29.69	10.82	0.02	6.56	4.24	0.14
	Slope	0.22	0.08	0.01	0.03	0.19	0.89	0.33	0.04	<.0001
Pb	Intercept	1.33	0.51	0.01	1.45	0.49	0.01	1.72	0.84	0.06
	Slope	0.35	0.03	<.0001	0.24	0.04	<.0001	0.42	0.04	<.0001
Mg	Intercept	15.24	9.19	0.11	29.39	10.51	0.02	-6.79	15.00	0.66
	Slope	0.57	0.20	0.00	0.03	0.20	0.86	1.23	0.32	0.00
Ni	Intercept	0.89	0.28	0.00	1.14	0.48	0.04	0.79	0.23	0.00
	Slope	0.75	0.32	0.02	1.46	0.57	0.01	0.19	0.26	0.48
K	Intercept	32.38	9.66	0.00	41.59	19.18	0.05	29.50	10.95	0.02
	Slope	0.64	0.09	<.0001	0.57	0.17	0.00	0.66	0.11	<.0001
Se	Intercept	0.57	0.57	0.33	0.38	0.36	0.31	-0.86	1.07	0.43
	Slope	0.39	0.12	0.00	0.24	0.06	0.00	0.98	0.28	0.00
Na	Intercept	80.88	18.60	0.00	103.20	42.84	0.03	54.70	12.16	0.00
	Slope	0.32	0.10	0.00	0.43	0.41	0.30	0.36	0.06	<.0001
V	Intercept	0.33	0.07	<.0001	0.27	0.07	0.00	0.38	0.12	0.01
	Slope	0.15	0.10	0.15	0.07	0.13	0.60	0.21	0.14	0.16
Zn	Intercept	-0.19	0.07	0.01	-0.25	0.06	0.00	-0.14	0.12	0.27
	Slope	0.46	0.03	<.0001	0.50	0.03	<.0001	0.44	0.04	<.0001

*All concentrations in ng/m³. Bold values indicate significance at 0.05 level.

Table 105. Associations between personal exposures and ambient concentrations for select acid-digestible elements for the children's cohort.*

Acid-Digestible Element		Both seasons			Winter			Summer		
		estimate	s.e.	p-value	estimate	s.e.	p-value	estimate	s.e.	p-value
Al	Intercept	134.64	24.68	<.0001	51.88	32.92	0.15	91.47	26.10	0.00
	Slope	0.43	0.10	<.0001	2.22	0.32	<.0001	0.38	0.09	0.00
As	Intercept	0.38	0.19	0.06	0.32	0.19	0.13	0.34	0.31	0.29
	Slope	0.26	0.06	<.0001	-0.02	0.05	0.70	0.39	0.10	0.00
Ba	Intercept	4.55	2.54	0.08	3.11	4.63	0.52	5.28	2.47	0.05
	Slope	0.50	0.80	0.53	2.51	1.91	0.19	-0.21	0.68	0.76
Cd	Intercept	0.17	0.07	0.02	0.09	0.10	0.39	0.26	0.10	0.02
	Slope	0.53	0.06	<.0001	0.45	0.12	0.00	0.53	0.08	<.0001
Fe	Intercept	132.45	31.99	0.00	190.17	48.53	0.00	94.76	43.34	0.05
	Slope	0.53	0.07	<.0001	0.37	0.13	0.00	0.61	0.08	<.0001
Pb	Intercept	5.94	9.22	0.53	-0.12	0.83	0.89	14.79	18.34	0.43
	Slope	1.08	0.50	0.03	0.61	0.07	<.0001	0.97	0.84	0.25
Mg	Intercept	18.66	9.95	0.07	43.49	10.96	0.00	-19.99	17.93	0.28
	Slope	0.74	0.21	0.00	0.00	0.23	0.99	1.75	0.33	<.0001
Mn	Intercept	5.43	2.66	0.05	12.54	5.31	0.04	-0.40	1.64	0.81
	Slope	0.44	0.12	0.00	0.29	0.21	0.17	0.60	0.08	<.0001
K	Intercept	53.50	22.10	0.02	62.98	32.87	0.08	50.29	28.87	0.10
	Slope	0.77	0.16	<.0001	0.81	0.26	0.00	0.72	0.18	0.00
Na	Intercept	450.43	79.87	<.0001	571.21	129.34	0.00	406.87	110.36	0.00
	Slope	-0.48	0.44	0.28	-1.05	0.94	0.28	-0.42	0.53	0.43
Sn	Intercept	6.18	4.83	0.21	1.98	1.16	0.12	11.21	9.52	0.26
	Slope	0.00	2.47	1.00	0.83	0.64	0.20	-1.56	4.68	0.74
V	Intercept	1.18	0.37	0.01	0.59	0.54	0.30	1.50	0.46	0.01
	Slope	0.50	0.14	0.00	0.79	0.22	0.00	0.39	0.17	0.03
Zn	Intercept	69.72	13.64	<.0001	105.64	29.36	0.00	53.96	15.04	0.00
	Slope	0.39	0.11	0.00	-0.20	0.40	0.62	0.51	0.10	<.0001

*All concentrations in ng/m³. Bold values indicate significance at 0.05 level.

Table 106. Associations between indoor and ambient concentrations for select water-extractable elements for the children's cohort.*

Water-Extractable Element		Both seasons			Winter			Summer		
		estimate	s.e.	p-value	estimate	s.e.	p-value	estimate	s.e.	p-value
As	Intercept	0.61	0.14	0.00	0.41	0.08	0.00	0.74	0.21	0.00
	Slope	0.33	0.04	<.0001	0.15	0.03	<.0001	0.43	0.07	<.0001
Ba	Intercept	2.86	1.64	0.09	0.38	0.12	0.01	6.63	3.25	0.06
	Slope	-0.22	0.69	0.75	0.25	0.04	<.0001	-1.05	1.13	0.36
Cd	Intercept	0.05	0.02	0.05	0.03	0.02	0.25	0.07	0.04	0.11
	Slope	0.48	0.04	<.0001	0.47	0.05	<.0001	0.47	0.06	<.0001
Fe	Intercept	4.95	2.84	0.09	5.98	4.20	0.18	5.27	3.33	0.14
	Slope	0.35	0.04	<.0001	0.24	0.06	<.0001	0.38	0.04	<.0001
Pb	Intercept	1.29	0.45	0.01	1.59	0.37	0.00	1.52	0.83	0.09
	Slope	0.35	0.03	<.0001	0.22	0.03	<.0001	0.44	0.04	<.0001
Mg	Intercept	11.11	4.19	0.01	17.69	7.50	0.04	0.43	4.19	0.92
	Slope	0.32	0.09	0.00	0.10	0.16	0.54	0.62	0.08	<.0001
Ni	Intercept	1.13	0.62	0.08	1.91	1.20	0.14	0.54	0.15	0.00
	Slope	0.53	0.72	0.46	0.89	1.31	0.50	0.07	0.19	0.73
K	Intercept	17.25	8.17	0.04	28.71	17.93	0.14	13.53	6.34	0.05
	Slope	0.63	0.07	<.0001	0.50	0.09	<.0001	0.68	0.08	<.0001
Se	Intercept	0.49	0.19	0.02	0.40	0.16	0.03	-0.12	0.30	0.70
	Slope	0.29	0.03	<.0001	0.21	0.03	<.0001	0.57	0.07	<.0001
Na	Intercept	45.80	14.23	0.00	48.62	32.61	0.16	36.64	10.02	0.00
	Slope	0.34	0.09	0.00	0.38	0.32	0.23	0.38	0.06	<.0001
V	Intercept	-0.10	0.06	0.09	-0.12	0.05	0.05	-0.11	0.11	0.30
	Slope	0.52	0.03	<.0001	0.55	0.03	<.0001	0.52	0.04	<.0001
Zn	Intercept	6.64	3.80	0.09	4.83	2.53	0.08	13.04	6.30	0.06
	Slope	0.54	0.04	<.0001	0.37	0.05	<.0001	0.53	0.06	<.0001

*All concentrations in ng/m³. Bold values indicate significance at 0.05 level.

Table 107. Associations between indoor and ambient concentrations for select acid-digestible elements for the children's cohort.*

Acid-Digestible Element		Both seasons			Winter			Summer		
		estimate	s.e.	p-value	estimate	s.e.	p-value	estimate	s.e.	p-value
Al	Intercept	85.60	15.78	<.0001	41.25	16.90	0.03	105.55	29.04	0.00
	Slope	0.44	0.07	<.0001	0.90	0.17	<.0001	0.39	0.09	0.00
As	Intercept	0.27	0.12	0.04	0.03	0.05	0.52	0.33	0.19	0.11
	Slope	0.49	0.05	<.0001	0.10	0.03	0.00	0.66	0.07	<.0001
Ba	Intercept	2.47	1.13	0.04	0.68	0.57	0.26	4.93	2.11	0.04
	Slope	-0.01	0.35	0.98	0.19	0.17	0.26	-0.42	0.58	0.48
Cd	Intercept	0.09	0.03	0.01	0.05	0.03	0.13	0.14	0.06	0.03
	Slope	0.44	0.04	<.0001	0.42	0.05	<.0001	0.41	0.06	<.0001
Fe	Intercept	32.75	21.59	0.14	51.90	15.42	0.01	13.07	41.01	0.75
	Slope	0.54	0.05	<.0001	0.38	0.04	<.0001	0.66	0.07	<.0001
Pb	Intercept	0.45	3.54	0.90	-1.96	6.81	0.78	1.64	1.01	0.13
	Slope	0.89	0.20	<.0001	1.91	0.58	0.00	0.62	0.04	<.0001
Mg	Intercept	9.55	5.08	0.07	19.53	7.84	0.03	-8.13	6.57	0.24
	Slope	0.49	0.11	<.0001	0.25	0.18	0.18	0.92	0.10	<.0001
Mn	Intercept	-0.47	1.35	0.73	-0.68	2.46	0.79	-0.20	1.47	0.89
	Slope	0.61	0.06	<.0001	0.63	0.10	<.0001	0.58	0.07	<.0001
K	Intercept	53.04	15.89	0.00	54.59	23.39	0.04	58.25	24.36	0.03
	Slope	0.48	0.08	<.0001	0.50	0.10	<.0001	0.43	0.14	0.00
Na	Intercept	159.29	25.39	<.0001	162.15	55.32	0.02	127.27	24.68	0.00
	Slope	-0.01	0.14	0.96	0.34	0.40	0.40	-0.03	0.10	0.74
Sn	Intercept	0.89	0.56	0.12	0.38	0.84	0.66	1.21	0.91	0.20
	Slope	1.05	0.27	0.00	1.33	0.52	0.01	1.06	0.16	<.0001
V	Intercept	0.35	0.12	0.01	0.87	0.30	0.02	0.32	0.17	0.11
	Slope	0.36	0.05	<.0001	-0.05	0.08	0.54	0.41	0.07	<.0001
Zn	Intercept	13.02	8.41	0.13	7.49	4.40	0.12	23.02	14.59	0.14
	Slope	0.58	0.06	<.0001	0.44	0.06	<.0001	0.58	0.09	<.0001

*All concentrations in ng/m³. Bold values indicate significance at 0.05 level.

4.3.2.11 Associations between PM_{2.5} Source Factors at the Central Site and Outdoor, Indoor, and Personal PM_{2.5} Concentrations

Sources of PM_{2.5} were determined by applying factor analysis to PM_{2.5} samples collected with the Harvard multi-pollutant monitor at the central ambient air monitoring site in Steubenville. Average PM_{2.5}, PM_{2.5} component, and gaseous pollutant concentrations observed at the site are reported in Table 108. The mean PM_{2.5} concentration at the central site was 20.8 µg/m³.

Factor analysis was performed using several different models. For the initial model, all species (including elements) that met the completeness and reproducibility requirements described in section 3.2.2.7 were used. Further models were then tested by removing elements individually or by exchanging between the water-extractable and the acid-digestible concentrations until an optimized solution was found. The performance of the model was evaluated using results of the regression of daily scores vs. daily PM_{2.5}, including (1) the R² value (which should be high), (2) the intercept (which should be low), (3) positive regression estimates for the sources, and (4) positive source contributions and source profiles.

Using this approach, it was found that the best model was obtained using concentrations of SO_4^{2-} , EC, NO_2 , CO, water-extractable As, Ba, Cd, Fe, K, Mg, Na, and V, and acid-digestible Pb. Using these data, four factors or source types were identified. The regression plot showing observed $\text{PM}_{2.5}$ concentrations vs. concentrations predicted by the model is presented in Figure 68.

Table 108. Summary statistics for central site data considered for use in factor analysis.

Pollutant	N	Mean	SD
$\text{PM}_{2.5}$ ($\mu\text{g}/\text{m}^3$)	52	20.72	12.09
SO_4^{2-} ($\mu\text{g}/\text{m}^3$)	51	6.67	5.62
EC ($\mu\text{g}/\text{m}^3$)	52	0.78	0.37
O_3 (ppb)	54	8.38	8.01
NO_2 (ppb)	53	29.84	20.94
CO (ppm)	49	0.28	0.24
Water-Extractable Elements (ng/m^3)			
As	52	1.9	1.4
Ba	48	2.1	1.4
Cd	48	0.4	0.4
Ca	43	78.4	73.5
Fe	46	35.5	54.1
Mg	52	30.5	26.9
Mn	52	6.9	8.2
Ni	50	0.4	0.8
K	49	69.0	58.5
Na	52	86.2	120.8
Sn	46	0.2	0.3
V	52	1.6	1.5
Acid-Digestible Elements (ng/m^3)			
Fe	46	280.4	287.4
Pb	52	7.3	14.0
Ti	52	4.4	29.9

Factor analysis can only be performed with samples that have no missing values for any species. From a total of 52 samples collected, only 32 samples had no missing values for all analyzed species. The regression parameters for the model using these 32 samples are shown in Table 109.

Table 109. Parameters from the regression of $\text{PM}_{2.5}$ vs. daily source scores.

Parameter	Estimate	t-value	p-value
Intercept	3394	3.03	0.0054
Factor 1	2318	4.30	0.0002
Factor 2	3343	6.20	<.0001
Factor 3	8271	15.33	<.0001
Factor 4	1582	2.93	0.006

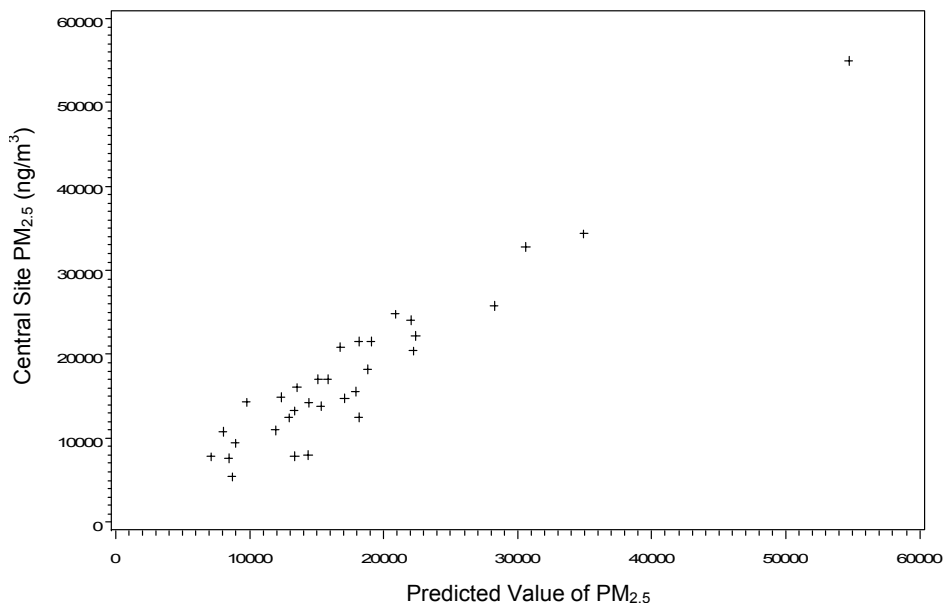


Figure 68. PM_{2.5} concentrations observed at the central Steubenville site vs. PM_{2.5} concentrations predicted by the factor analysis model.

As shown in Table 109, all of the source factors were significant at the 0.05 level. The intercept was significant indicating that approximately 3.4 $\mu\text{g}/\text{m}^3$ of PM_{2.5} mass was left unexplained. Considering only the samples that have no missing values, the average PM_{2.5} mass concentration is 17.7 $\mu\text{g}/\text{m}^3$. Table 110 shows the estimated source contribution for each species for each factor. Table 111 shows the source profile for each factor.

Table 110. Average source contributions (in ng/m^3) for the four-source factor analysis model.

Species	Factor 1	Factor 2	Factor 3	Factor 4
PM _{2.5}	2718.64	945.164	9391.29	1225.91
SO ₄ ²⁻	55.8	330.257	4607.42	434.81
EC	358.7	47.259	122.98	84.64
NO ₂	1.76	2.019	1.99	0.81
CO	0.26	0.029	0.01	0
Water-Extractable Elements				
As	1	0.213	0.19	0.45
Ba	0.49	0.084	0.77	0.23
Cd	0.27	0.035	-0.02	0.11
Fe	18.45	9.615	7.96	3.28
Mg	21.36	4.843	4.18	-1.25
Mn	7.96	1.622	1.2	-0.19
K	57.51	-2.459	9.49	3.01
Na	2.27	32.652	25.2	5.83
V	0.17	0.019	0.41	1.09
Acid-Digestible Elements				
Pb	4.62	2.111	2.9	0.67

NOTE: The maximum value for each species is shown in bold.

Table 111. Source profiles (species mass per total mass associated with the source factor, expressed as a percentage) for the four-source factor analysis model.

Species	Factor 1	Factor 2	Factor 3	Factor 4
SO ₄ ²⁻	2.052	34.942	49.061	35.468
EC	13.194	5.000	1.310	6.904
NO ₂	0.065	0.214	0.021	0.066
CO	0.010	0.003	0.000	0.000
Water-Extractable Elements				
As	0.037	0.023	0.002	0.037
Ba	0.018	0.009	0.008	0.019
Cd	0.010	0.004	0.000	0.009
Fe	0.679	1.017	0.085	0.268
Mg	0.786	0.512	0.044	-0.102
Mn	0.293	0.172	0.013	-0.016
K	2.115	-0.260	0.101	0.245
Na	0.083	3.455	0.268	0.475
V	0.006	0.002	0.004	0.089
Acid-Digestible Elements				
Pb	0.170	0.223	0.031	0.055

Results of source apportionment using factor analysis have been described in the past for Steubenville (Koutrakis and Spengler, 1987). In that paper, six sources of PM_{2.5} were resolved, including sources that appeared to represent secondary sulfates, coal and oil combustion, iron and steel production, titanium production, soil, and motor vehicles. Two of these six sources cannot be accounted for in our analysis. Typical tracers of soil, such as Al and Si, did not meet our analysis requirements and therefore were not used in the analysis. Average titanium concentrations in the 1987 report were 190 ng/m³. However, the average concentration during the period covered by our factor analysis was 30 ng/m³. This large concentration drop suggests either that local sources of Ti no longer exist or that industry has installed new particle control technology. As a result, our solution with four sources instead of six is not inconsistent with the results presented by Koutrakis and Spengler.

Factor 1 is associated with EC, CO, As, Cd, Fe, Mg, Mn, K and Pb. EC, Pb, and CO are tracers that are often associated with vehicular emissions. Factor 1 contributed an average of 2.7 µg/m³ (15%) of the total PM_{2.5} mass. A similar concentration (2.6 µg/m³) was associated with mobile sources in the previous report. This factor is likely associated with vehicular emissions, resuspended road dust, and other local sources.

Factor 2 is associated with NO₂ and Na but also with Fe, Pb, and some sulfate. It is possible that this is the iron and steel production source determined previously. Factor 2 contributed 0.95 µg/m³ (5.3%) of the total PM_{2.5} mass. In the previous report, the similar factor contributed 2.3 µg/m³.

Factor 3 is associated with sulfate and Ba, but also with some As, Fe and K, all of which are species associated with secondary sulfates. This source contributed 9.4 µg/m³ (54%) of the total PM_{2.5} mass, which is consistent with the contribution of the secondary sulfate source (9.7 µg/m³) reported previously.

The last source is associated mainly with V but also with some As and sulfate. This source likely represents coal or oil combustion. The source contributed 1.2 µg/m³ (6.9%) of the total

PM_{2.5} mass. This contribution is much less than the 8.6 µg/m³ previously reported for coal and oil combustion. It is possible that new particle control technologies have decreased the contribution of this source. Alternatively, it is possible that this factor is not entirely separated from the secondary sulfate factor, as a major tracer for this source, Ni, did not meet the requirements for analysis.

The masses contributed by each factor at the central site were examined as predictors of PM_{2.5} exposures for children. Because repeated measurements were made for each child, two different regression models were used to determine the predictive ability of the central site factors: (1) a simple regression model that does not consider repeated measures, and (2) a model that considers repeated measures and a covariance structure. Specifically, the models were applied to examine the associations between the PM_{2.5} factors from the central site and the PM_{2.5} concentrations observed inside the children’s homes. For the repeated measurement models, R² values were calculated by regressing the predicted values against the measured PM_{2.5} data. Table 112 shows the results from each model in terms of R² values, intercepts, and regression estimates.

Table 112. Comparison of results from two different models of indoor PM_{2.5} concentrations vs. PM_{2.5} factors from the central ambient air monitoring site.

Model	Parameter	Value	SE	t-stat	p-value
1	R ²	0.27			
	Intercept	7235	1868	3.87	0.0002
	Factor 1	-0.53	0.39	-1.34	0.1831
	Factor 2	0.30	0.31	0.96	0.34
	Factor 3	0.5182	0.098	5.29	<0.0001
2	Factor 4	1.50	0.53	2.81	0.006
	R ²	0.24			
	Intercept	7501	2299	3.26	0.0034
	Factor 1	0.06	0.43	0.14	0.88
	Factor 2	-0.07	0.31	-0.23	0.82
Factor 3	0.47	0.11	4.39	<0.0001	
Factor 4	0.93	0.52	1.77	0.081	

As shown in Table 112, Model 1 (the simple regression) explained more of the variability (R²=0.27) in indoor PM_{2.5} concentrations. Intercept values are similar for both approaches. The estimates, however, were more significant (lower p-values) for Model 1. Despite these differences, both approaches were relatively consistent, showing significant associations for Factors 3 and 4 only. Because Model 1 explained more of the mass, this approach was used for future analyses.

To determine the contributions of the different factors to the PM_{2.5} concentrations observed in various microenvironments for the children’s cohort, daily outdoor, indoor, or personal PM_{2.5} levels were regressed against each of the four factors using random effects models. Table 113 shows the results of these regressions.

For outdoor concentrations, the model performed well, with an overall R² value of 0.90. The intercept is slightly higher than in the model used for the central site (which had an intercept of 3.6 µg/m³), indicating that a larger portion of the mass is left unexplained in this model. All factors in the model were significant except for Factor 1. This insignificant value may result from spatial variability of the sources associated with this factor. The estimates shown in Table 113 represent the amount of outdoor PM_{2.5} that can be explained by each 1 µg/m³ increase in each factor at the central site. For instance, for Factors 3 and 4, a 1 µg/m³ increase in PM_{2.5}

concentrations at the central site corresponds to an identical 1 ug/m³ increase in PM_{2.5} outdoors at the homes. However, for Factor 2, the estimate of approximately 0.5 suggests that only about half of the mass at the central site is captured by samples collected outside of the participants' homes. These results may suggest that Factor 2 represents a local source that has the greatest effect on the central ambient air monitoring site or that spatial and temporal variability in this Factor reduces our ability to estimate it outdoors at the homes.

Table 113. Regression results describing the ability of central site factors to predict 24-hr outdoor, indoor, or personal PM_{2.5} concentrations for the children's cohort.

Model	Parameter	Value	SE	T	P
Outdoor	R ²	0.90			
	Intercept	3635	686	5.29	<0.0001
	Factor 1	0.24	0.15	1.61	0.11
	Factor 2	0.66	0.12	5.69	<0.0001
	Factor 3	1.02	0.035	28.7	<0.0001
	Factor 4	1.14	0.19	5.99	<0.0001
Indoor	R ²	0.28			
	Intercept	7235	1868	3.87	0.0002
	Factor 1	-0.53	0.39	-1.34	0.18
	Factor 2	0.30	0.32	0.96	0.34
	Factor 3	0.52	0.098	5.29	<0.0001
	Factor 4	1.50	0.53	2.81	0.006
Personal	R ²	0.22			
	Intercept	11102	1438	7.72	<0.0001
	Factor 1	-0.65	0.46	-1.42	0.16
	Factor 2	0.51	0.10	5.17	<0.0001
	Factor 3	0.92	0.39	2.35	0.02
	Factor 4	-0.55	0.61	-0.91	0.36

Using the same approach, the factors at the central site were found to explain 28% of the variability in indoor PM_{2.5} concentrations. The intercept of the model was higher for indoor as compared to outdoor concentrations, indicating that the model explained a smaller amount of the total indoor PM_{2.5} mass. The model shows that Factors 1 and 2 were not significantly associated with indoor PM_{2.5} mass. It was expected that Factor 1 would be a poor predictor, because it was not able to predict outdoor concentrations. The poor predictive ability of Factor 2 may be due to poorer infiltration or greater indoor deposition of the types of particles associated with this factor. Additionally, these particles may contain more volatile material that can be lost indoors (like semi-volatile organic components). Factors 3 and 4 remained significantly associated with indoor PM_{2.5} concentrations. However, the slope estimated for Factor 3 for the indoor samples was half as great as that estimated for the outdoor samples, whereas the slope estimated for Factor 4 for the indoor samples was greater than that estimated for the outdoor samples. Lower slope estimates may be due to variable penetration of particles from outdoor to indoor environments. Higher estimates may be due to closer proximity of subjects' homes to the sources associated with Factor 4.

The model incorporating source factors resolved for the central site predicted 22% of the variability in personal PM_{2.5} exposures. The intercept for the model of personal exposures (11.1 ug/m³) was much higher than the intercepts observed for the other microenvironments, indicating that a large portion of the personal PM_{2.5} exposures was left unexplained by the model. Factors 1 and 4 were not significant. However, the contributions of Factors 2 and 3 were significantly associated with personal PM_{2.5} exposures. The slopes estimated for Factors 2 and 3 for personal exposures (0.51 and 0.92, respectively) were greater than those estimated

for these factors for indoor concentrations (0.30 and 0.52, respectively), but less than those estimated for these factors for outdoor concentrations (0.66 and 1.02, respectively). These factors can represent the contribution of both indoor and outdoor concentrations to personal exposures.

4.3.3 Discussion of Results from the Older Adult and Children's Cohorts

For the older adult cohort, we found 24-hour ambient particle concentrations to be consistently strong proxies of corresponding personal exposures, regardless of the particle species (i.e., PM_{2.5}, SO₄²⁻, or EC), season (i.e., summer or fall), and ventilation status. Associations between ambient concentrations and corresponding personal exposures were strongest for SO₄²⁻, a regional pollutant with no major indoor sources. Ambient concentrations of EC were also significant proxies of corresponding exposures, although associations were weaker, likely due to the influence of local sources such as traffic and cooking. Our findings are consistent with those of previous studies (Janssen et al., 2000; Sarnat et al., 2000; Rojas-Bracho et al., 2000; Ebel, et al., 2000; Rojas-Bracho et al., 2004) and provide additional justification for the use of ambient PM_{2.5}, SO₄²⁻, and to a lesser extent EC, to represent corresponding mean personal exposures in epidemiologic analyses.

Contrary to previous findings (Liu et al., 1997; Linaker et al., 2000; Brauer et al., 1989; Patterson et al., 2000; Sarnat et al., 2001), however, associations between ambient concentrations and personal exposures for O₃ and NO₂ were also strong, suggesting that ambient O₃ and NO₂ concentrations are suitable proxies of their respective personal exposures. Significant associations in Steubenville may be due in part to the differences in study design, where the number of collected samples was larger – and, for the gases, collected with greater sensitivity – in Steubenville than in previous studies, thus improving the power to detect associations between ambient and personal gas concentrations. Our current results support this theory, as personal-ambient gas associations were stronger in the fall, when field LODs were lower, as compared to the summer.

As was the case for PM, home ventilation levels were shown to be an important modifier of the association between ambient concentrations and personal exposures for the gases. In Steubenville, personal-ambient gas associations, in particular for O₃, were highest for subjects spending time indoors with windows open as compared to for those spending time indoors with windows closed. Consistent with our findings, a recent follow-up study of 43 children and healthy senior citizens in Boston, MA (Sarnat et al., 2005), found significant personal-ambient associations for O₃ and NO₂ in the summer but not the winter, which may be due to greater home ventilation in the summer in Boston. Results of previous indoor-outdoor monitoring studies are also consistent with our findings (Gold et al., 1996). However, no other personal monitoring studies show the impact of home ventilation as a modifier of personal-ambient gas associations, and further investigation is needed to determine its importance.

We also found significant associations between ambient particle concentrations and personal O₃ and NO₂ exposures, with associations varying with particle species, season, and ventilation status. Specifically, we found 24-hr ambient particle concentrations to be significantly and positively correlated with personal O₃ exposures in the summer and with both personal O₃ and NO₂ exposures in the fall. These associations reflected similarly significant associations a) between ambient PM and ambient gas concentrations and b) between ambient gas concentrations and their corresponding personal exposures. Our results indicate that ambient particle and gas concentrations are strong proxies of personal gaseous exposures during seasons in which homes tend to be well-ventilated or in locations characterized by well-

ventilated homes. The findings have important implications for air pollution health effect studies based on ambient concentrations, as results suggest that the health effects of particles and gases will be difficult to separate, particularly for individuals living in well-ventilated homes. This separation will be particularly difficult for pollutants originating from similar sources, as shown by the fact that associations between ambient PM and personal gases were strongest for pollutants originating from similar sources; associations between ambient SO_4^{2-} and personal O_3 , both regional pollutants, and between ambient EC and personal NO_2 , both emitted primarily from motor vehicles, were significant in both seasons. Our results correspond to results from the Boston study (Sarnat et al., 2005) which showed significant associations between ambient $\text{PM}_{2.5}$ and personal O_3 in the summer only, when photochemical formation of $\text{PM}_{2.5}$ and O_3 is greatest. Similarly, the Boston study found significant associations between ambient $\text{PM}_{2.5}$ and personal NO_2 in only the fall, a season characterized more by local sources such as motor vehicle traffic that contribute to $\text{PM}_{2.5}$ and NO_2 .

It is important to note that personal O_3 and NO_2 exposures were found to increase by only 1.1 and 1.7 ppb, respectively, with every $10 \mu\text{g}/\text{m}^3$ increase in ambient $\text{PM}_{2.5}$. These observed increases in personal O_3 and NO_2 are extremely small and have not been shown to elicit adverse health effects in controlled laboratory studies (Gong et al., 1998; Devlin et al., 1997; Frampton et al., 1991). Despite this, numerous epidemiological studies have linked 24-hr ambient O_3 and NO_2 concentrations to adverse health impacts, suggesting that these gases may elicit health impacts along with or in combination with other pollutants, or that 24-hr concentrations of these gases are acting as proxies for shorter-term exposures.

Children were monitored when mean ambient concentrations of regional pollution were higher, while mean ambient concentrations of locally-generated pollution were lower. In the summer in which the children were monitored, for example, the mean and maximum ambient $\text{PM}_{2.5}$, SO_4^{2-} , and O_3 concentrations were higher than in the previous summer when the older adults were monitored. Mean ambient concentrations of EC and NO_2 , generally markers of locally generated traffic pollution, were lower in the summer when the children were monitored. Although not directly comparable, mean ambient concentrations of $\text{PM}_{2.5}$ and especially SO_4^{2-} were lower in the winter when the children were monitored as compared to when the older adults were monitored, while wintertime mean ambient concentrations of all other measured pollutants, even O_3 , observed when the children were monitored tended to be comparable to mean concentrations when the older adults were monitored.

For the children, ambient concentrations were poorer proxies of personal exposures for the gases and to a lesser extent for the particle species (i.e., $\text{PM}_{2.5}$, SO_4^{2-} , and EC), with personal-ambient slopes generally lower and less significant for children as compared to older adults. In the summer, personal-ambient slopes for $\text{PM}_{2.5}$ and SO_4^{2-} were approximately 20% lower for the children as compared to the adults. For every $10 \mu\text{g}/\text{m}^3$ increase in corresponding ambient concentrations, summertime personal $\text{PM}_{2.5}$ and SO_4^{2-} exposures increased an average of $5.8 \mu\text{g}/\text{m}^3$ for children and $7.5 \mu\text{g}/\text{m}^3$ for older adults. Personal-ambient $\text{PM}_{2.5}$ and SO_4^{2-} slopes were even lower in the winter, consistent with the poorer ventilation during the winter months and the observed effect modification by open window use. The opposite seasonal pattern was found for personal-ambient associations for EC, which for the children were not significant in the summer, but were significant in the winter with a slope of 0.41 ± 0.07 . A similar seasonal pattern in the association between personal and ambient EC was found for the older adults, with a greater slope observed in the fall than in the summer. For O_3 , the personal-ambient slope for the children was highest in the summer, but it was still lower than that found for older adults in the summer, with values of 0.09 ± 0.03 and 0.15 ± 0.02 for the children and older adults, respectively. In contrast to the other measured pollutants, the summertime personal-ambient

NO₂ slope of 0.36±0.06 for children was approximately 33% higher than that for older adults (0.23±0.05). Results suggest that the ability of ambient concentrations to reflect exposures for children was the greatest for PM_{2.5} and SO₄²⁻ and generally during the summer months; however, ambient particle and gas concentrations were generally poorer proxies of corresponding exposures for children as compared to older adults.

As was the case with the older adults, home ventilation was the primary factor affecting the ability of ambient pollutant concentrations to reflect the personal exposures of children to these pollutants. Personal-ambient and indoor-ambient slopes were higher when windows were open or when participants spent more time near open windows, both conditions consistent with increased ventilation. Other factors, such as stove type or time spent outdoors, were significant but relatively minor modifiers of the relationship between ambient concentrations and either indoor or personal exposures.

Our findings also suggest that ventilation was an important determinant of the ability of ambient particles to act as proxies of personal gaseous exposures for children. During the summer, which was characterized by a greater percentage of open window usage, ambient particle concentrations were strong proxies of children's personal exposures to gases from similar sources, such as was observed with the traffic-related pollutants NO₂ and EC and with the regional pollutants O₃, SO₄²⁻, and PM_{2.5} (likely because of its large mass fraction of SO₄²⁻). Since these associations were not significant in the winter, results suggest that ambient particles can represent personal gaseous exposures only when homes are well-ventilated. However, even under these well-ventilated conditions, ambient particle concentrations were better proxies of children's exposures to the particles themselves rather than the gases, as evidenced by the observed low slopes for ambient particle-personal gas associations. Ambient O₃ concentrations in both seasons and ambient NO₂ in the summer were poor proxies of personal particulate exposures, which is contrary to findings for older adults in Steubenville and Baltimore.

This study provided a powerful database for characterizing the particulate and gaseous exposures of older adults and children and examining their associations with corresponding indoor and ambient concentrations. Results confirmed the results of previous studies showing that ambient PM_{2.5} and SO₄²⁻ concentrations were strong proxies of corresponding personal exposures, especially for older adults, whereas ambient EC and gas concentrations were generally weaker proxies of corresponding personal exposures to these pollutants. Results also verified that ambient gas concentrations could act as proxies of their own exposures and also as proxies of personal particulate exposures. However, when interpreting these findings, it should be noted that results were based on a relatively small number of individuals. Hence, these results are not generalizable to other cohorts, including other cohorts of older adults or of children. Furthermore, the sensitivity and precision of the measurement and analytical methods used in the indoor and personal exposure program, particularly for the elements and SO₂, were insufficient to measure pollutant concentrations in Steubenville. As a result, findings should be interpreted with caution. Similarly, the effects of stove type and activity patterns on personal-ambient and indoor-ambient relationships should also be interpreted cautiously. For the older adult cohort, separation of the effect of stove type from building was not possible, and relatively few children lived in homes with gas stoves. Both the older adult and child cohorts were fairly inactive, limiting the ability to explore possible effect modifications by activity patterns.

Results from our study indicate that ambient fine particles are strong proxies of corresponding exposures for older adults and to a lesser extent for children. The ability of ambient fine particle concentrations to reflect exposures differed by particle component and was greatest for

the regional fine particle SO_4^{2-} and poorest for the traffic-related particle EC. Furthermore, results indicate that ambient concentrations may represent personal exposures to both $\text{PM}_{2.5}$ and gases, especially for pollutants from common sources and for individuals living in well-ventilated environments. This suggests that time-series health studies based on 24-hour ambient concentrations may not be able to separate the independent effects of particles and gases. Future exposure studies should examine factors contributing to differences in pollutant exposures for older adults and children.

4.4 Associations between Air Pollution and Cardiac Health Effects

As discussed earlier, the SCAMP exposure study of older adults was conducted in conjunction with a companion study of cardiac health. Participants in the companion health study included the same cohort of older adults participating in the exposure study. As mentioned in Section 3.2.1.2, prior to their inclusion in the study, participants were screened to collect information on cardiovascular and respiratory health and medications and to obtain baseline electrocardiogram (12-lead MAC6, Marquette Medical Systems Inc.) recordings. Exclusion criteria included smoking, having a pacemaker, a recent acute coronary syndrome, atrial flutter, or atrial fibrillation.

For each participant, health measurements were performed weekly on the same day of the week and at the same time during summer (June 4 – August 18) and fall (September 25 – December 15) of 2000. A short questionnaire on recent symptoms, hospital or doctor visits, and medication use was administered, followed by Holter electrocardiogram monitoring (SEER MC, GE Medical Systems) with electrodes in a modified V5 and AVF position. The Holter monitoring protocol included: (1) five minutes of rest in a supine position; (2) three supine blood pressure measurements (NIBP Vital Signs Monitor, Welch Allyn); (3) five minutes of standing with three standing blood pressure measurements taken after two minutes; (4) five minutes of exercise (walking) outdoors (weather and health permitting); (5) five minutes of rest in a supine position; and (6) two minutes and 20 seconds of paced breathing.

The Holter tapes were analyzed using a Marquette MARS Workstation (GE Medical Systems), with a 125 samples/second sampling rate following heart rate variability (HRV) guidelines (Task Force of the European Society of Cardiology and the North American Society of Pacing and Electrophysiology 1996). The Holter tapes were analyzed for heart rate, arrhythmias, and time and frequency domain HRV measures. HRV, average heart rate, and arrhythmia counts were obtained for the whole protocol and for each separate interval. A detailed description of health measurements and pollution-association cardiac health effects are provided in the manuscripts included in Appendices E and F to this report. Results generally suggest that in Steubenville, increased levels of ambient particulate pollution from non-traffic sources (e.g., sulfate particles) may adversely affect autonomic function and increase the risk of supraventricular arrhythmia in the elderly.

5. CONCLUSIONS

The Steubenville Comprehensive Air Monitoring Program generated a powerful database of information concerning ambient, outdoor, indoor, and personal-breathing-space concentrations of PM_{2.5}, its components, and co-pollutants in Steubenville, Ohio. Results provide valuable information in response to a number of PM_{2.5} research needs, including the need for better estimates of the relationship between ambient concentrations of PM_{2.5} and its components and actual human exposures to these species, the need for better characterization of the potential for confounding by gaseous pollutants in PM_{2.5} epidemiology studies, the need for better characterization of individual chemical components of PM_{2.5}, and the need for data for development of State Implementation Plans (SIPs) in response to the PM_{2.5} NAAQS. Moreover, SCAMP provides a much-needed current assessment of PM_{2.5} in Steubenville, Ohio, which historically has been a key location for air pollution research, and remains a nonattainment area under the PM_{2.5} NAAQS despite major changes in industry and population during the past two decades. Major conclusions of the program are as follows:

- **Average concentrations of PM_{2.5}, as well as those of many PM_{2.5} components and gaseous pollutants, in Steubenville have decreased appreciably since the Harvard Six Cities Study was conducted in the late 1970s and 1980s, but remain elevated relative to the annual PM_{2.5} NAAQS.**

A comparison of mean ambient concentrations of PM_{2.5} total mass, twelve PM_{2.5} components (Al, Ca, Cu, Fe, K, Mn, Ni, Pb, Se, SO₄²⁻, V, and Zn), and three gaseous pollutants (SO₂, NO₂, and O₃) measured during the SCAMP ambient air monitoring program with mean concentrations measured during the Six Cities Study indicated that ambient concentrations of all but two (fine particulate Ca and O₃) of these species in Steubenville have decreased by more than 30% since the Six Cities Study was conducted. The largest decrease in concentration (-92%) was observed for fine particulate Pb, likely resulting from the discontinued use of leaded gasoline. The mean PM_{2.5} mass concentration measured during SCAMP (18.4 µg/m³) was about 40% less than that measured during the Six Cities Study. Mean concentrations of fine particulate SO₄²⁻, a major component of PM_{2.5} in Steubenville, and of SO₂, its gaseous precursor, were 54% and 56% lower, respectively, during SCAMP than during the Six Cities Study. Although these differences may in part be due to differences in the sampling and analytical techniques employed by the two programs, the consistent downward trend strongly suggests that appreciable improvements in Steubenville's air quality have been made over the past 25 years. Nevertheless, Steubenville remains a nonattainment area under the PM_{2.5} NAAQS; the overall mean PM_{2.5} concentration measured there during SCAMP was 3.4 µg/m³ greater than the annual standard of 15 µg/m³. Hence, further reductions in PM_{2.5} concentrations in Steubenville will be required; data collected as part of SCAMP are particularly relevant for informing the development of an appropriate PM_{2.5} implementation plan for Steubenville.

- **Secondary sulfates, likely resulting to a great extent from coal-fired power plants to the west and southwest of Steubenville, are the largest contributor to ambient PM_{2.5} mass in Steubenville. Primary PM_{2.5} emissions from coal-fired power plants appear to contribute relatively little to the total mass of ambient PM_{2.5}, however.**

The overall mean SO₄²⁻ concentration measured at the central Steubenville site during the SCAMP ambient air monitoring program was 5.8 µg/m³. PM_{2.5} mass balance calculations performed using 109 days with complete or nearly complete PM_{2.5} speciation data indicated

that, on average, SO_4^{2-} accounted for 30.4% of the total mass of ambient $\text{PM}_{2.5}$ at Steubenville. (The second largest contributor to total $\text{PM}_{2.5}$ mass was organic material, which accounted for an estimated 25.7% by weight). As expected for this predominantly secondary pollutant, SO_4^{2-} concentrations at Steubenville exhibited a statistically significant seasonal pattern of greater concentrations during the summer, when meteorological conditions favor secondary sulfate formation, and lesser concentrations during the winter. This variability drove a similar statistically significant seasonal pattern in total $\text{PM}_{2.5}$ mass concentrations. Sulfate accounted for 38.4% of the total mass of ambient $\text{PM}_{2.5}$ during the summer at Steubenville, but only for 24.8% of the total mass during the winter. Source apportionment analysis performed by applying Positive Matrix Factorization to 115 days of speciated $\text{PM}_{2.5}$ data from the central site suggested that a factor dominated by secondary sulfates (including NH_4^+ and SO_4^{2-} at a molar ratio of 2:1, consistent with the presence of secondary ammonium sulfate aerosol) was the largest contributor to $\text{PM}_{2.5}$ mass at Steubenville, accounting for about 42% by weight. Potential Source Contribution Function modeling indicated a likely source region to the west and southwest of Steubenville, consistent with the general location of numerous coal-fired power plants. Conversely, the primary coal combustion source factor resolved by PMF (which was strongly enriched in Se, well-correlated with SO_2 and NO_2 , and had a probable source region including portions of the Ohio and Monongahela River valleys in the Steubenville region), contributed $< 1 \mu\text{g}/\text{m}^3$ to the total mass of ambient $\text{PM}_{2.5}$. Thus, results from the SCAMP ambient air monitoring program suggest that whereas reductions in secondary sulfates originating from coal-fired power plants (as required under the Clean Air Interstate Rule) will almost certainly be needed if Steubenville is to attain compliance with the $\text{PM}_{2.5}$ NAAQS, further reductions in primary particulate emissions from coal-fired power plants would be of relatively little aid in attaining a mass-based standard.

- **Local sources in the immediate Steubenville vicinity, including motor vehicles, iron and steel facilities, and other point and area sources, contribute an estimated $5 \mu\text{g}/\text{m}^3$ (~30%) to the total mass of ambient $\text{PM}_{2.5}$ in Steubenville.**

Similar estimates of the contribution of local sources in the immediate Steubenville vicinity to the city's mean ambient $\text{PM}_{2.5}$ concentration were obtained via two independent analyses, improving our confidence in this estimate. The first analysis considered pairwise differences between daily $\text{PM}_{2.5}$ concentrations measured by the FRM monitor at the central ambient air monitoring site in Steubenville and daily $\text{PM}_{2.5}$ concentrations measured at the northern and western satellite sites, which were considered to be most representative of regional background $\text{PM}_{2.5}$ concentrations in the Steubenville area. Based on this pairwise comparison, the estimated contribution of local sources in the immediate Steubenville vicinity to the total mass of ambient $\text{PM}_{2.5}$ observed there during SCAMP was $4.6 \mu\text{g}/\text{m}^3$. The second analysis – source apportionment using PMF and PSCF – identified two factors representing local sources in Steubenville, which together accounted for $5.3 \mu\text{g}/\text{m}^3$ of the total mass of $\text{PM}_{2.5}$. The compositional profiles of these two local source factors suggest that they likely include motor vehicles and other local sources of carbonaceous aerosols, which contributed ~20% of the total $\text{PM}_{2.5}$ mass and more than half of the mass of fine particulate EC, OM, and Ba observed at the Steubenville site, as well as iron and steel production facilities, which contributed ~10% of the total $\text{PM}_{2.5}$ mass and more than half of the mass of fine particulate Fe, Mn, and Zn observed there. The identity of the local iron and steel production source and its estimated contribution to total $\text{PM}_{2.5}$ mass were further validated by Unmix source apportionment modeling, which estimated a $\text{PM}_{2.5}$ mass contribution of $1.3 - 2.2 \mu\text{g}/\text{m}^3$ (average = $1.8 \mu\text{g}/\text{m}^3$) from this source, and by comparing source apportionment results with data from the U.S. EPA's Toxic Release Inventory.

According to the 2001 TRI for air emissions, the primary metals industry accounts for 93% of Zn emissions and 83% of Mn emissions from stationary sources in Jefferson County, OH, where Steubenville is located. The Mn/Zn mass ratios of 0.18-0.20 computed for the iron and steel source factors resolved by the PMF and Unmix models agree reasonably well with the ratio of 0.16 computed using the TRI data.

- **Distinct mechanisms cause local- and regional-scale episodes of elevated PM_{2.5} concentrations in Steubenville.**

Case studies of episodes of elevated PM_{2.5} concentration observed during the SCAMP ambient air monitoring program revealed two distinctly different types of episodes that led to elevated PM_{2.5} concentrations at the central monitoring site in Steubenville:

1. A regional episode type characterized by a multiple-day period of consistently and moderately elevated concentrations of secondary-sulfate-dominated PM_{2.5}, which typically occurred during warmer portions of the year, and
2. A local episode type characterized by shorter overnight periods of more acutely elevated concentrations of PM_{2.5} resulting from the buildup of primary particles enriched in trace metals and carbonaceous species during nocturnal temperature inversions, which typically occurred during cooler portions of the year.

During local, inversion-driven episodes, hourly PM_{2.5} concentrations were typically very strongly correlated with hourly concentrations of primary gaseous pollutants (e.g., CO, NO_x, and SO₂), whereas PM_{2.5} concentrations generally did not correlate well with gaseous pollutant concentrations during regional, secondary-sulfate-dominated episodes. Moreover, the greatest hourly PM_{2.5} concentrations observed during SCAMP occurred in the nighttime or early morning during cool-season episodes; none of the 76 1-hr average concentrations greater than 65 µg/m³ were observed during the mid-afternoon (between 1:00 p.m. and 5:00 p.m. EST), and none of the 20 greatest 1-hr average concentrations were observed during the summer. Both of the PM_{2.5} episode types identified during SCAMP must be considered when developing an implementation plan to reduce concentrations of PM_{2.5} in Steubenville, as both can lead to elevated 24-hr PM_{2.5} concentrations (e.g., concentrations exceeding the recently proposed 24-hr standard of 35 µg/m³), which in turn contribute to Steubenville's high annual average PM_{2.5} concentration. As the data collected as part of SCAMP illustrate, these episode types result from very different mechanisms; therefore, reducing their occurrences will require the application of unique strategies.

- **Concentrations of certain transition metals, including Fe, Mn, and Zn, observed at Steubenville are appreciably elevated relative to concentrations observed in the surrounding region and in other larger U.S. cities.**

Median daily ratios of concentrations of water-extractable Fe, Mn, and Zn measured at the central Steubenville site to concentrations of these elements measured at each of the four SCAMP satellite sites were > 1.5, suggesting that local sources in the immediate Steubenville vicinity contribute substantially to their ambient concentrations there. In fact, an analysis of mean paired differences between Steubenville concentrations and estimated "regional background" concentrations of these elements indicates that their mean concentrations at Steubenville are at least twice as great as their mean concentrations in the surrounding area. Mean acid-digestible concentrations of Fe, Mn, and Zn observed at Steubenville during SCAMP were also substantially greater than concentrations measured in other, much larger U.S. cities during comparable time periods. (With the exception of Fe

in St. Louis, mean concentrations of Fe, Mn, and Zn in Steubenville were 1.9-7.6 times as great as those reported for Atlanta, GA, Baltimore, MD, Boston, MA, Chicago, IL, New York, NY, St. Louis, MO, and Washington, DC, where data were available). Mn is defined as a Hazardous Air Pollutant in the 1990 Clean Air Act amendments (the mean concentration of fine particulate Mn, which accounts for only a portion of total airborne Mn, at Steubenville during SCAMP was about 37% as great as the U.S. ATSDR's MRL for Mn), and Fe and Zn have both been associated with pulmonary health endpoints in PM toxicology studies. Hence, further research should be conducted to better elucidate whether these metals play any role in the health effects previously associated with PM_{2.5} in Steubenville.

- **Ambient fine particles are strong proxies of corresponding personal exposures for older adults and, to a lesser extent, for children in Steubenville.**

For the older adult cohort, personal exposures to PM_{2.5} were significantly associated with corresponding ambient PM_{2.5} concentrations measured at the central ambient air monitoring site in Steubenville during both the summer and fall. Mixed model results indicated that, for every 10 µg/m³ increase in corresponding ambient PM_{2.5} concentrations, personal exposures of the older adults increased an average of 7.3±0.5 µg/m³ during the summer and 6.3±0.5 µg/m³ during the fall. Personal-ambient slopes (i.e., average increase in personal exposure per unit increase in corresponding ambient concentration) for PM_{2.5} for the children's cohort were lower and less significant than those for the older adult cohort. For every 10 µg/m³ increase in corresponding ambient concentrations, summertime personal PM_{2.5} exposures increased an average of 5.8±0.8 for the children, which is about 20% less than the summertime increase in exposures observed for the older adults. In winter, personal PM_{2.5} exposures for the children increased an average of only 2.5±1.4 µg/m³ for every 10 µg/m³ increase in ambient concentration. Overall, results of SCAMP confirm the results of previous studies showing that ambient PM_{2.5} concentrations are strong proxies of corresponding personal exposures, especially for older adults.

- **The ability of ambient fine particle concentrations to reflect personal exposures differs by particle component, and is greater for SO₄²⁻, a predominantly regional fine particle, than for EC, a predominantly locally-emitted fine particle.**

For both the older adult cohort and the children's cohort, associations between ambient concentrations and corresponding personal exposures were stronger for fine particulate SO₄²⁻, a regional pollutant with no major indoor sources, than they were for total PM_{2.5} mass. As with total PM_{2.5}, the associations were stronger in summer than in fall or winter and were stronger for the older adults than for the children. Personal-ambient slopes describing these associations were 0.74±0.02 and 0.59±0.04 for the older adults and children, respectively, in the summer, 0.64±0.02 for the older adults in the fall, and 0.37±0.05 for the children in the winter.

Ambient concentrations of EC were also significant proxies of corresponding exposures for the older adults in both seasons (slope = 0.33±0.10 in summer, 0.69±0.06 in fall) and for the children in winter (slope=0.41±0.07), although personal-ambient associations were weaker for EC than for SO₄²⁻, likely due to the influence of local sources such as traffic and cooking. As the slopes presented above indicate, personal-ambient associations for EC were weaker during summer and stronger during fall and winter, contrary to the trend exhibited by PM_{2.5} and SO₄²⁻. Findings are consistent with those of previous studies and provide additional justification for the use of ambient SO₄²⁻, and to a lesser extent EC, to represent corresponding mean personal exposures in epidemiologic analyses.

Personal-ambient associations for water-extractable and acid-digestible elemental components of PM_{2.5}, although positive and statistically significant in many instances, exhibited inconsistencies by element, extraction method, cohort, and season. Hence, no definitive conclusions could be drawn regarding these results.

- **Home ventilation was the primary factor affecting the ability of ambient pollutant concentrations to reflect the personal exposures of older adults and children to these pollutants.**

For both cohorts, indoor-ambient and personal-ambient slopes were generally higher when windows were open or when participants spent more time near open windows, both conditions consistent with increased ventilation. Other factors, such as stove type or time spent outdoors, were significant but relatively minor modifiers of the relationship between ambient concentrations and either indoor or personal exposures. Results from SCAMP are the first to show the impact of home ventilation as a modifier of personal-ambient gas associations, and they suggest that ventilation may be an important modifier of the magnitude of effect in time-series health studies.

- **Time-series health studies based on 24-hour ambient concentrations may not be able to separate the independent effects of particles and gases in Steubenville.**

An analysis of data collected at the central ambient air monitoring site in Steubenville as part of the SCAMP ambient air monitoring program revealed that, after accounting for autocorrelation, concentrations of PM_{2.5} exhibited statistically significant ($\alpha = 0.05$), positive associations with concentrations of CO, NO, NO₂, and SO₂ ($R^2 = 0.35-0.46$). Associations of NO and NO₂ with PM_{2.5} were strongest during fall ($R^2 = 0.53$ and 0.61 , respectively; slope = 0.39 and 0.57 , respectively) and weakest during summer ($R^2 = 0.0014$ and 0.086 , respectively; slope = -0.015 and 0.20 , respectively). The association between CO and PM_{2.5} was strongest during winter ($R^2 = 0.60$; slope = 0.66) and weakest during summer ($R^2 = 0.24$; slope = 0.27), whereas the association between O₃ and PM_{2.5} was weakly positive during summer ($R^2 = 0.12$; slope = 0.17) and moderately negative during winter ($R^2 = 0.25$; slope = -0.29). These associations suggest that gaseous pollutants such as CO, NO₂, and to a lesser extent O₃, may cause seasonally-dependent collinearity problems in PM_{2.5} time series epidemiology studies focusing on Steubenville. For associations between PM_{2.5} and these gases, an interaction with season is observed even after autocorrelation (including seasonality) has been removed from the individual pollutant time series. Hence, results from the SCAMP ambient air monitoring program reinforce the need for future epidemiology studies to employ separate analyses by season or to include appropriate interaction terms in order to better capture and control for the collinearity between PM_{2.5} and co-pollutants, and suggest that caution should be exercised when interpreting the results of previous epidemiology studies that did not consider the effects of pollutants such as CO and NO₂ and their seasonal interactions with PM.

Cross-pollutant analyses conducted using data from both the older adult and children's cohorts indicated statistically significant associations between ambient particle concentrations and personal O₃ and NO₂ exposures, with associations varying by particle species, season, and ventilation status. For the older adult cohort, 24-hr ambient particle concentrations were significantly and positively correlated with personal O₃ exposures in the summer and with both personal O₃ and NO₂ exposures in the fall. For the children's cohort during the summer, which was characterized by a greater percentage of open window

usage, ambient particle concentrations were strong proxies of children's personal exposures to gases from similar sources, such as was observed with the traffic-related pollutants NO₂ and EC and with the regional pollutants O₃, SO₄²⁻, and PM_{2.5} (likely because of its large mass fraction of SO₄²⁻). Since these associations were not significant in the winter, results suggest that ambient particles can represent personal gaseous exposures only when homes are well-ventilated. Hence, results from the SCAMP indoor and personal exposure program suggest that ambient pollution may represent exposures to both PM_{2.5} and gases in Steubenville, especially for pollutants from common sources and for individuals living in well-ventilated environments, making the health effects of particles and gases difficult to separate. It is important to note, however, that ambient particle concentrations were generally better proxies of exposures to the particles themselves rather than the gases. For instance, for the older adult cohort, personal O₃ and NO₂ exposures were found to increase by only 1.1 and 1.7 ppb with every 10 µg/m³ increase in ambient PM_{2.5}, respectively. These observed increases in personal O₃ and NO₂ are extremely small and have not been shown to elicit adverse health effects in controlled laboratory studies. Nevertheless, numerous epidemiologic studies have linked both 24-hour ambient O₃ and NO₂ concentrations to adverse health impacts, suggesting that these gases may elicit health impacts along with or in combination with other pollutants, or that 24-hour ambient concentrations of these gases are acting as proxies for shorter-term pollutant exposures.

- **Health effect studies conducted using PM_{2.5} and gaseous pollutant data collected at the SCAMP central monitoring site in Steubenville found that fine particulate sulfate, but not elemental carbon, was significantly associated with decreased heart rate variability and increased odds of supraventricular arrhythmia among members of the older adult cohort.**

Although these studies were based on a fairly small cohort and did not consider all major chemical components of PM_{2.5} in Steubenville (e.g., organic carbon, the second largest contributor to PM_{2.5} mass in Steubenville, was not included in the analyses), they provide evidence that increased levels of ambient fine particulate pollution from non-traffic sources (e.g., sulfate particles) may adversely affect cardiac health in the elderly.

In addition to producing the comprehensive database and key findings described above, SCAMP made several other important contributions to advancing PM_{2.5} research. The program identified sensitivity issues that limited the ability to determine certain PM_{2.5} trace elements by conventionally-used XRF techniques, and it pioneered the development of methods for applying dynamic reaction cell ICP-MS to enable these elements to be determined with improved sensitivity. The program is one of the few that has explored both water-extractable and acid-digestible (total) PM_{2.5} trace element concentrations, and it is the first to do so in Steubenville. Papers published as part of SCAMP also identified statistical concerns that are common to air monitoring data but are commonly overlooked in the PM_{2.5} literature and provided recommendations, such as the application of ARIMA modeling, for properly accounting for these concerns. Moreover, SCAMP helped to seed several additional studies that collectively add to the understanding of air quality in the Upper Ohio River Valley region, factors that affect it, and its effects on human health. Two studies led by Ohio University, including "The Evaluation of the Emission, Transport and Deposition of Mercury, PM_{2.5}, and Arsenic in the Ohio Valley from Coal-Fired Power Plants," which is being funded by the U.S. DOE-NETL (agreement No. DE-FC26-03NT41723), and "Proposed Research and Support Activities for Air Quality Initiatives in Ohio and Surrounding States," which is being funded by U.S. EPA Region V, are using the DRC ICP-MS methodology developed during SCAMP, as well as monitoring equipment from the program. Data and statistical concepts from SCAMP are being applied in a third study, titled

“Design and Feasibility Assessment of a Retrospective Epidemiologic Study of Coal-Fired Power Plant Emissions in the Pittsburgh, Pennsylvania, Region,” which is being led by the University of Pittsburgh Graduate School of Public Health and funded by DOE-NETL (agreement No. DE-FC26-05NT42302). Data from SCAMP are also being incorporated into the database being developed as part of a DOE-NETL-funded project (agreement No. DE-FC26-02NT41476) titled “Database and Analytical Tool for Air Quality in the Upper Ohio River Valley,” which is being led by ATS-Chester Engineers.

Finally, results from SCAMP have important implications for guiding future PM_{2.5} research. Specific recommendations are as follows:

1. Future epidemiology studies focusing on PM_{2.5} should carefully consider the possibility for confounding by gaseous co-pollutants such as CO, NO₂, and O₃, as well as the possibility that the associations among and effects of these pollutants may be significantly modified by season or by home ventilation status. Previous studies that did not adequately consider these possibilities should be interpreted cautiously.
2. Further research is required to determine whether the composition of PM_{2.5} impacts its effect on human health (e.g., whether particles enriched in trace metals and carbonaceous species, which were prevalent during the cool season episodes typically observed during SCAMP, are more toxic than the secondary-sulfate-dominated particles that were typically prevalent during summertime episodes, or whether the mere existence of particles, regardless of composition, triggers the observed health effects). In particular, given the elevated concentrations of fine particulate Fe, Mn, and Zn observed at Steubenville, the PM health effects that have previously been observed there, and recent toxicological findings implicating these transition metals, further research should be conducted to elucidate what role, if any, Fe, Mn, and Zn play in the health effects that have been attributed to PM_{2.5}.
3. Data from SCAMP are consistent with the hypothesis that ambient secondary sulfates, under suitable photochemical conditions, may play a role in increasing the solubility of fine particulate Fe. Further research should be conducted to confirm or refute this hypothesis.
4. Findings from SCAMP illustrating that PM_{2.5} and co-pollutants exhibit markedly different diurnal variability during local, inversion-driven episodes as compared to regional, secondary-sulfate-driven episodes suggest a need for further research using time-resolved measurements to elucidate the effect of the intraday timing and short-term intensity of pollution episodes on human exposures and possible health effects.
5. Additional improvements are needed in methods for measuring human exposures to gaseous pollutants, especially SO₂, and to fine particulate trace elements and carbonaceous species. The sensitivity and precision of sampling and analytical methods used during SCAMP were insufficient to measure personal exposures to SO₂ and fine particulate trace elements. Moreover, organic carbon, the second largest contributor to the mass of ambient PM_{2.5} in Steubenville during SCAMP, could not be measured in personal exposure samples because of methodological limitations. Given current interest in the health implications of specific PM_{2.5} components, including trace metal and carbonaceous species, further method development is warranted to allow human exposures to these species to be better quantified.
6. Associations between ambient pollutant concentrations and corresponding personal exposures observed during SCAMP differed by cohort. Future exposure studies should examine factors contributing to differences in pollutant exposures for older adults and children.

REFERENCES

- Abt, E.; Suh, H.; Catalana, P.; Koutrakis, P. (2000) Relative contribution of outdoor and indoor particle sources to indoor concentrations. *Environ. Sci. Technol.* 34: 3579-3587.
- Adamson, I.Y.; Prieditis, H.; Hedgecock, C.; Vincent, R. (2000) Zinc is the toxic factor in the lung response to an atmospheric particulate sample. *Toxicol. Appl. Pharmacol.* 166: 111-119.
- Allen, G.; Sioutas, C.; Koutrakis, P.; Reiss, R.; Lurmann, F.W.; Roberts, P.T. (1997) Evaluation of the TEOM method for measurement of ambient particulate mass in urban areas. *J. Air & Waste Manage. Assoc.* 47: 682-689.
- Box, G.E.P.; Jenkins, G.M. (1976) *Times series analysis: forecasting and control*. San Francisco: Holden-Day.
- Brauer, M.; Koutrakis, P.; Spengler, J.D. (1989) Personal exposure to acidic aerosols and gases. *Environ. Sci. Technol.* 23: 1408-1412.
- Breiman, L.; Friedman, J.H.; Olshen, R.A.; Stone, C.J. (1984) *Classification and Regression Trees*. New York: Chapman & Hall.
- Brown, K. (2006) Characterization of particulate and gaseous exposures of sensitive populations living in Baltimore and Boston. Department of Environmental Health. Boston, MA, Harvard School of Public Health: 146.
- Burnett, R.T.; Brook, J.; Dann, T.; Delocla, C.; Philips, O.; Cakmak, S.; Vincent, R.; Goldberg, M.S.; Krewski, D. (2000) Association between particulate- and gas phase components of urban air pollution and daily mortality in eight Canadian cities. *Inhalation Toxicol.* 12(suppl. 4): 15-39.
- Burton, R.M.; Suh, H.H.; Koutrakis, P. (1996) Spatial variation in particulate concentrations within metropolitan Philadelphia. *Environ. Sci. Technol.* 30: 400-407.
- Carnegie Mellon University. (2005) Atmospheric aerosol source-receptor relationships: the role of coal-fired plants. Semi-Annual Technical Progress Report to the U.S. Department of Energy for the Period September 2004-February 2005. Available: http://www.netl.doe.gov/technologies/coalpower/ewr/air_quality_research/am_monitor/pubs/41017R08.rev.pdf [24 May 2006].
- Chang, L.T.; Sarnat, J.A.; Wolfson, J.M.; Rojas-Bracho L.; Suh, H.H.; Koutrakis, P. (1999) Development of a personal multi-pollutant exposure sampler for particulate matter and criteria gases.
- Chang, L.T.; Koutrakis, P.; Catalano, P.; Suh, H.H. (2000) Hourly personal exposures to fine particles and gaseous pollutants-results from Baltimore, MD. *J. Air & Waste Manage. Assoc.* 50: 1223-1235.
- Chellam, S.; Kulkarni, P.; Fraser, M.P. (2005) Emissions of organic compounds and trace metals in fine particulate matter from motor vehicles: a tunnel study in Houston, Texas. *J. Air & Waste Manage. Assoc.* 55: 60-72.
- Chock, D.P.; Winkler, S.L.; Chen, C. (2000) A study of the association between daily mortality and ambient air pollutant concentrations in Pittsburgh, Pennsylvania. *J. Air & Waste Manage. Assoc.* 50: 1481-1500.
- Chow, J.C.; Watson, J.G. (1998) Guideline on speciated particulate monitoring. Reno, NV: Desert Research Institute. Available: <http://www.epa.gov/ttnamti1/files/ambient/pm25/spec/drispec.pdf> [22 May 2006].
- Chuersuwan, N.; Turpin, B.J.; Pietarinen, C. (2000) Evaluation of time-resolved PM_{2.5} data in urban/suburban areas of New Jersey. *J. Air & Waste Manage. Assoc.* 50: 1780-1789.
- Connell, D.P.; Withum, J.A.; Winter, S.E.; Statnick, R.M.; Bilonick, R.A. (2005a) The Steubenville comprehensive air monitoring program (SCAMP): overview and statistical considerations. *J. Air & Waste Manage. Assoc.* 55: 467-480.
- Connell, D.P.; Withum, J.A.; Winter, S.E.; Statnick, R.M.; Bilonick, R.A. (2005b) The

- Steubenville comprehensive air monitoring program (SCAMP): associations among fine particulate matter, co-pollutants, and meteorological conditions. *J. Air & Waste Manage. Assoc.* 55: 481-496.
- Connell, D.P.; Withum, J.A.; Winter, S.E.; Statnick, R.M.; Bilonick, R.A. (2005c) The Steubenville comprehensive air monitoring program (SCAMP): analysis of short-term and episodic variations in PM_{2.5} concentrations using hourly air monitoring data. *J. Air & Waste Manage. Assoc.* 55: 559-573.
- Connell, D.P.; Winter, S.E.; Conrad, V.B.; Kim, M.; Crist, K.C. (2006) The Steubenville comprehensive air monitoring program (SCAMP): concentrations and solubilities of PM_{2.5} trace elements and their implications for source apportionment and health research. *J. Air & Waste Manage. Assoc.* 56: 1750-1766.
- Conrad, V.B.; Winter, S.E.; Krofcheck, D.S. (2000) ICP-MS and x-ray fluorescence: a comparative study for the elemental analysis of fine particulate matter. Presented at the International Symposium on Measurement of Toxic and Related Air Pollutants, Research Triangle Park, NC, September 2000.
- Delfino, R.J.; Murphy-Moulton, A.M.; Burnett, R.T.; Brook, J.R.; Becklake, M.R. (1997) Effects of air pollution on emergency room visits for respiratory illnesses in Montreal, Quebec. *Am. J. Respir. Crit. Care Med.* 155: 568-576.
- Delfino, R.J.; Zeiger, R.S.; Seltzer, J.M.; Street, D.H. (1998) Symptoms in pediatric asthmatics and air pollution: differences in effects by symptom severity, anti-inflammatory medication use and particulate averaging time. *Environ. Health Perspect.* 106: 751-761.
- Demokritou, P.; Kavouras, I.G.; Ferguson, S.T.; Koutrakis, P. (2001) Development and laboratory performance evaluation of a personal multipollutant sampler for simultaneous measurements of particulate and gaseous pollutants. *Aerosol Sci. Technol.* 35: 741-752.
- Devlin, R.B.; Folinsbee, L.J.; Biscardi, F.; Hatch, G.; Becker, S.; Madden, M.C.; et al. (1997) Inflammation and cell damage induced by repeated exposure of humans to ozone. *Inhal. Toxicol.* 9: 211-235.
- Dockery, D.W.; Pope, C.A.; Xu, X.; Spengler, J.D.; Ware, J.H.; Fay, M.E.; Ferris, B.G.; Speizer, F.E. (1993) An association between air pollution and mortality in six U.S. cities. *N. Engl. J. Med.* 329: 1753-1759.
- Dockery, D.W.; Cunningham, J.; Damokosh, A.I.; Neas, L.M.; Spengler, J.D.; Koutrakis, P.; Ware, J.H.; Raizenne, M.; Speizer, F.E. (1996) Health effects of acid aerosols on North American children: respiratory symptoms. *Environ. Health Perspect.* 104: 500-505.
- Dusseldorp, A.; Kruize, H.; Brunekreef, B.; Hofschreuder, P.; de Meer, G.; van Oudvorst, A.B. (1995) Associations of PM₁₀ and airborne iron with respiratory health of adults living near a steel factory. *Am. J. Respir. Crit. Care Med.* 152: 1932-1939.
- Ebelt, S.T.; Petkau, A.J.; Vedal, S.; Fisher, T.V.; Brauer, M. (2000) Exposure of chronic obstructive pulmonary disease patients to particulate matter: relationships between personal and ambient air concentrations. *J. Air Waste Manage. Assoc.* 50: 1081-1094.
- Faust, B.C. (1994) Photochemistry of clouds, fogs, and aerosols. *Environ. Sci. Technol.* 28: 217A-222A.
- Fernandez, A.; Wendt, J.O.L.; Witten, M.L. (2005) Health effects engineering of coal and biomass combustion particulates: influence of zinc, sulfur and process changes on potential lung injury from inhaled ash. *Fuel* 85: 1320-1327.
- Frampton, M.W.; Morrow, P.E.; Cox, C.; Gibb, F.R.; Speers, D.M.; Utell, M.J. (1991) Effects of nitrogen dioxide exposure on pulmonary function and airway reactivity in normal humans. *Am. Rev. Respir. Dis.* 143: 522-527.
- Frampton, M.W.; Ghio, A.J.; Samet, J.M.; Carson, J.L.; Carter, J.D.; Devlin, R.B. (1999) Effects of aqueous extracts of PM₁₀ filters from the Utah Valley on human airway epithelial cells. *Am. J. Physiol.* 277: L960-L967.
- Frank, N.H. (2006) Retained nitrate, hydrated sulfates, and carbonaceous mass in federal

- reference method fine particulate matter for six eastern U.S. cities. *J. Air & Waste Manage. Assoc.* 56: 500-511.
- Garg, B.D.; Cadle, S.H.; Mulawa, P.A.; Groblicki, P.J.; Laroo, C.; Parr, G.A. (2000) Brake wear particulate matter emissions. *Environ. Sci. Technol.* 34: 4463-4469.
- Gavett, S.H.; Haykal-Coates, N.; Copeland, L.B.; Heinrich, J.; Gilmour, M.I. (2003) Metal composition of ambient PM_{2.5} influences severity of allergic airways disease in mice. *Environ. Health Perspect.* 111: 1471-1477.
- Ghio, A.J.; Carter, J.D.; Dailey, L.A.; Devlin, R.B.; Samet, J.M. (1999a) Respiratory epithelial cells demonstrate lactoferrin receptors that increase after metal exposure. *Am. J. Physiol.* 276: L933-L940.
- Ghio, A.J.; Stoneheurner, J.; McGee, J.K.; Kinsey, J.S. (1999b) Sulfate content correlates with iron concentrations in ambient air pollution particles. *Inhal. Toxicol.* 11: 293-307.
- Gilbert, R.O. (1987) *Statistical Methods for Environmental Pollution Monitoring*. New York: Van Nostrand Reinhold. pp. 177-178.
- Gold, D.R.; Allen, G.; Damokosh, A.; Serrano, P.; Hayes, C.; Castillejos, M. (1996) Comparison of outdoor and classroom ozone exposures for school children in Mexico City. *J. Air. Waste Manage. Assoc.* 46: 335-342.
- Gong, H.; Wong, R.; Sarma, R.J.; Linn, W.S.; Sullivan, E.D.; Shamo, D.A.; et al. (1998) Cardiovascular effects of ozone exposure in human volunteers. *Am. J. Respir. Crit. Care Med.* 158: 538-546.
- Grahame, T.J.; Schlesinger, R.B. (2005) Evaluating the health risk from secondary sulfates in eastern North American regional ambient air particulate matter. *Inhal. Toxicol.* 17: 15-27.
- Gurgueira, S.A.; Lawrence, J.; Coull, B.; Krishna Murthy, G.G.; Gonzalez-Flecha, B. (2002) Rapid increases in the steady-state concentration of reactive oxygen species in the lungs and heart after particulate air pollution inhalation. *Environ. Health Perspect.* 110: 749-755.
- Hidy, G.M.; Friedlander, S.K. (1971) *The Nature of the Los Angeles Aerosol*. Englund, H.M.; Beery, W.T., Eds. New York: Academic Press. pp. 391-404.
- Huang, Y.C.; Ghio, A.J.; Stoneheurner, J.; McGee, J.; Carter, J.D.; Grambow, S.C.; Devlin, R.B. (2003) The role of soluble components in ambient fine particles – induced changes in human lungs and blood. *Inhal. Toxicol.* 15: 327-342.
- Jakab, G.J.; Clarke, R.W.; Hemenway, D.R.; Longphre, M.V.; Kleeberger, S.R.; Frank, R. (1996) Inhalation of acid coated carbon black particles impairs alveolar macrophage phagocytosis. *Toxicol. Lett.* 88: 243-248.
- Jansen, J.J.; Edgerton, E.S.; Hansen, D.A.; Hartsell, B.E. (2002) Sampling artifacts in the federal reference method for PM_{2.5}. In Proceedings of the International Conference on Air Quality III, Arlington, VA, September 9-12.
- Janssen, N.; Van Mansom, D.; Van Der Jagt, K.; Harssema, H.; Hoek, G. (1997) Mass concentration and elemental composition of airborne particulate matter at street and background locations. *Atmos. Envir.* 8: 1185-93.
- Janssen, N.A.; de Hartog, J.J.; Hoek, G.; Brunekreef, B.; Lanki, T.; Timonen, K.L.; et al. (2000) Personal exposure to fine particulate matter in elderly subjects: relation between personal, indoor, and outdoor concentrations. *J. Air. Waste Manage. Assoc.* 50: 1133-1143.
- Kenski, D.M. (2003) Meteorological conditions and haze: a CART analysis. Available: http://www.ladco.org/tech/monitoring/docs_gifs/cart3.pdf [5 March 2004].
- Kinney, P.L.; Thurston, G.D. (1993) Field evaluation of instrument performance: statistical considerations. *Appl. Occup. Environ. Hyg.* 8: 267-271.
- Kleinman, M.T.; Bufalino, C.; Rasmussen, R.; Hyde, D.; Bhalla, D.K.; Mautz, W.J. (2000) Toxicity of chemical components of ambient fine particulate matter (PM_{2.5}) inhaled by aged rats. *J. Appl. Toxicol.* 20: 357-364.

- Klemm, R.J.; Lipfert, F.W.; Wyzga, R.E.; Gust, C. (2004) Daily mortality and air pollution in Atlanta: two years of data from ARIES. *Inhal. Toxicol.* 16(suppl. 1): 131-141.
- Koutrakis, P.; Spengler, J.D. (1987) Source apportionment of ambient particles in Steubenville, OH using specific rotation factor analysis. *Atmos. Environ.* 21: 1511-1519.
- Koutrakis, P.; Briggs, S.L.K.; Leaderer, B.P. (1992) Source apportionment of indoor aerosols in Suffolk and Onondaga Counties, New York. *Environ. Sci. Technol.* 26: 521-527.
- Koutrakis, P.; Wolfson, J.M.; Bunyaviroch, A.; Froehlich, S.E.; Hirano, K.; Mulik, J.D. (1993) Measurement of ambient ozone using a nitrite-coated filter. *Anal. Chem.* 65: 209-214.
- Laden, F.; Neas, L.M.; Dockery, D.W.; Schwartz, J. (2000) Association of fine particulate matter from different sources with daily mortality in six U.S. cities. *Environ. Health Perspect.* 108: 941-947.
- Levy, J.I.; Houseman, E.A.; Ryan, L.; Richardson, D.; Spengler, J. (2000) Particle concentrations in urban microenvironments. *Environ. Health Perspect.* 108: 1051-1057.
- Lewis, C.W.; Norris, G.A.; Conner, T.L.; Henry, R.C. (2003) Source apportionment of Phoenix PM_{2.5} aerosol with the Unmix Receptor Model. *J. Air & Waste Manage. Assoc.* 53: 325-338.
- Linaker, C.H.; Chauhan, A.J.; Inskip, H.M.; Holgate, S.T.; Coggon, D. (2000) Personal exposures of children to nitrogen dioxide relative to concentrations in outdoor air. *Occup. Environ. Med.* 57: 472-476.
- Lipfert, F.W.; Wyzga, R.E. (1997) Air pollution and mortality: the implications of uncertainties in regression modeling and exposure measurement. *J. Air & Waste Manage. Assoc.* 47: 517-523.
- Liu, L.J.; Koutrakis, P.; Suh, H.H.; Mulik, J.D.; Burton, R.M. (1993) Use of personal measurements for ozone exposure assessment: a pilot study. *Environ. Health Perspect.* 101(4): 318-324.
- Liu, L.J.; Koutrakis, P.; Olson, M.P.; Allen, G.A.; McDonnell, W.F.; Gerrity, T.R. (1994) Evaluation of the Harvard ozone passive sampler on human subjects indoors. *Environ. Sci. & Technol.* 28: 915-923.
- Liu, L.J.; Delfino, R.; Koutrakis, P. (1997) Ozone exposure assessment in a southern California community. *Environ. Health Perspect.* 105: 58-65.
- Mar, T.F.; Norris, G.A.; Koenig, J.Q.; Larson, T.V. (2000) Associations between air pollution and mortality in Phoenix, 1995-1997. *Environ. Health Perspect.* 108: 347-353.
- Mason, B. (1966) *Principles of Geochemistry*. New York: John Wiley & Sons. pp 45-46.
- Medeiros, N.; Rivero, D.H.; Kasahara, D.I.; Saiki, M.; Godleski, J.J.; Koutrakis, P.; Capelozzi, V.L.; Saldiva, P.H.; Antonangelo, L. (2004) Acute pulmonary and hematological effects of two types of particle surrogates are influenced by their elemental composition. *Environ. Res.* 95: 62-70.
- Meng, Q.Y.; Turpin, B.J.; Polidori, A.; Lee, J.H.; Weisel, C.; Morandi, M.; Colome, S.; Stock, T.; Winer, A.; Zhang, J. (2005) PM_{2.5} of ambient origin: estimates and exposure errors relevant to PM epidemiology. *Environ. Sci. Technol.* 39: 5105-5112.
- Moolgavkar, S.H. (2003) Air pollution and daily mortality in two U.S. counties: season-specific analyses and exposure-response relationships. *Inhal. Toxicol.* 15: 877-907.
- Moolgavkar, S.H.; Luebeck, E.G. (1996) A critical review of the evidence on particulate air pollution and mortality. *Epidemiology* 7: 420-428.
- Moolgavkar, S.H.; Luebeck, E.G.; Hall, T.A.; Anderson, E.L. (1995) Particulate air pollution, sulfur dioxide, and daily mortality: a reanalysis of the Steubenville data. *Inhal. Toxicol.* 7: 35-44.

- Neas, L.M.; Dockery, D.W.; Burge, H.; Koutrakis, P.; Speizer, F.E. (1996) Fungus spores, air pollutants, and other determinants of peak expiratory flow rate in children. *Am. J. Epidemiol.* 143: 797-807.
- Ogawa, C. (1998) NO, NO₂, NO_x and SO₂ sampling protocol using the Ogawa sampler, Version 3.
- Ohio Department of Transportation. (2004) Traffic Monitoring Reports. <http://www.dot.state.oh.us/techservsite/offceorg/traffmonit/TMReports/> [17 March 2006].
- Ostro, B.D.; Broadwin, R.; Lipsett, M.J. (2000) Coarse and fine particles and daily mortality in the Coachella Valley, California: a follow-up study. *J. Exop. Anal. Environ. Epidemiol.* 10: 412-419.
- Paatero, P. (1997) Least square formulation of robust nonnegative factor analysis. *Chemomet. Intell. Lab. Systems* 37: 23-35.
- Patterson, E.; Eatough, D.J. (2000) Indoor/outdoor relationships for ambient PM_{2.5} and associated pollutants: epidemiological implications in Lindon, Utah. *J. Air Waste Manage. Assoc.* 50: 103-110.
- Peters, A.; Wichmann, H.E.; Tuch, T.; Heinrich, J.; Heyder, J. (1997) Respiratory effects are associated with the number of ultrafine particles. *Am. J. Respir. Crit. Care Med.* 155: 1376-1383.
- Peters, A.; Dockery, D.W.; Muller, J.E.; Mittleman, M.A. (2001) Increased particulate air pollution and the triggering of myocardial infarction. *Circulation* 103: 2810-2815.
- Pittsburgh Air Quality Study. (2003) PM_{2.5} concentrations and composition in Pittsburgh: summary of ambient measurements from Pittsburgh Air Quality Study (PAQS). http://www.netl.doe.gov/coalpower/environment/air_q/docs/Ambient_summary.pdf [20 October 2003].
- Pope, C.A.; Thun, M.J.; Namboodiri, M.M.; Dockery, D.W.; Evans, J.S.; Speizer, F.E.; Heath, C.W. (1995) Particulate air pollution as a predictor of mortality in a prospective study of U.S. adults. *Am. J. Respir. Crit. Care Med.* 151: 669-674.
- Research Triangle Institute. (2003) Standard operating procedure for the determination of organic, elemental, and total carbon in particulate matter using a thermal/optical-transmittance carbon analyzer. Research Triangle Park, NC: Environmental & Industrial Sciences Division.
- Rojas-Bracho, L.; Suh, H.; Koutrakis, P. (2000) Relationship among personal, indoor, and outdoor fine and coarse particulate concentrations for individuals with COPD. *J. Exposure Anal. Environ. Epidemiol.* 10: 294-306.
- Rojas-Bracho, L.; Suh, H.H.; Catalano, P.J.; Koutrakis, P. (2004) Personal exposures to particles and their relationships with personal activities for chronic obstructive pulmonary disease patients living in Boston. *J. Air Waste Manage. Assoc.* 54: 207-217.
- Rosas, I.; McCartney, H.A.; Payne, R.W.; Calderon, C.; Lacey, J.; Chapela, R.; Ruiz-Velazco, S. (1998) Analysis of the relationships between environmental factors (aeroallergens, air pollution, and weather) and asthma emergency admissions to a hospital in Mexico City. *Allergy* 53: 394-401.
- Sarnat, J.A.; Koutrakis, P.; Suh, H.H. (2000) Assessing the relationship between personal particulate and gaseous exposures of senior citizens living in Baltimore, MD. *J. Air & Waste Manage. Assoc.* 50: 1184-98.
- Sarnat, J.A.; Schwartz, J.; Catalano, P.J.; Suh, H.H. (2001) Gaseous pollutants in particulate matter epidemiology: confounders or surrogates? *Environ. Health Perspect.* 109: 1053-1061.
- Sarnat, J.A.; Koutrakis, P.; Long, C.M.; Coull, B.; Schwartz, J.; Suh, H.H. (2002) Using sulfur as a tracer for outdoor fine particulate matter. *Environ. Sci. Technol.* 36: 5305-5314.
- Sarnat, J.A.; Brown, K.W.; Schwartz, J.; Coull, B.A.; Koutrakis, P. (2005) Ambient gas

- concentrations and personal particulate matter exposures: implications for studying the health effects of particles. *Epidemiology* 16: 385-395.
- Schlesinger, R.B. (1995) Interaction of gaseous and particulate pollutants in the respiratory tract: mechanisms and modulators. *Toxicology* 105: 315-325.
- Schwartz, J.; Dockery, D. (1992) Particulate air pollution and daily mortality in Steubenville, Ohio. *Am. J. Epidemiol.* 135: 12-19.
- Schwartz, J.; Dockery, D.W.; Neas, L.M. (1996) Is daily mortality associated specifically with fine particles? *J. Air & Waste Manage. Assoc.* 46: 927-939.
- Sickles, J.E. (1999) A summary of airborne concentrations of sulfur- and nitrogen-containing pollutants in the northeastern United States. *J. Air & Waste Manage. Assoc.* 49: 882-893.
- Stieb, D.M.; Beveridge, R.C.; Brook, J.R.; Smith-Doiron, M.; Burnett, R.T.; Dales, R.E.; Beaulieu, S.; Judek, S.; Mamedov A. (2000) Air pollution, aeroallergens and cardiorespiratory emergency department visits in Saint John, Canada. *J. Expo. Anal. Environ. Epidemiol.* 10: 461-477.
- Stieb, D.M.; Judek, S.; Burnett, R.T. (2002) Meta-analysis of time-series studies of air pollution and mortality: effects of gases and particles and the influence of cause of death, age, and season. *J. Air & Waste Manage. Assoc.* 52: 470-484.
- Suarez, A.E.; Ondov, J.M. (2002) Ambient aerosol concentrations of elements resolved by size and source: contributions of some cytokine-active metals from coal- and oil-fired power plants. *Energy & Fuels* 16: 562-568.
- Tanner, S.D.; Baranov, V.I. (1999) Theory, design, and operation of a dynamic reaction cell for ICP-MS. *Atomic Spectroscopy* 20(2): 45-52.
- Task Force of the European Society of Cardiology and the North American Society of Pacing and Electrophysiology. (1996) Heart rate variability: standards of measurement, physiological interpretation, and clinical use. *Circulation* 93: 1043-1065.
- Therneau, T.M.; Atkinson, E.J. (1997) An introduction to recursive partitioning using the RPART routines. Technical Report #61. Rochester, MN: Mayo Foundation, September.
- Thurston, G.D.; Spengler, J.D. (1985) A quantitative assessment of source contributions to inhalable particulate matter pollution in metropolitan Boston. *Atmos. Environ.* 19: 9-26.
- Tobin, J. (1958) Estimation of relationships for limited dependent variables. *Econometrica* 26: 24-36.
- Tolocka, M.P.; Solomon, P.A.; Mitchell, W.; Norris, G.A.; Gemmill, D.B.; Wiener, R.W.; Vanderpool, R.W.; Homolya, J.B.; Rice, J. (2001) East vs. west in the US: chemical characteristics of PM_{2.5} during the winter of 1999. *J. Aerosol Sci. Technol.* 34, 88-96.
- Tropp, R.J.; Engelbrecht, J.; Lowenthal, D.H.; Kohl, S.D.; Chow, J.C.; Watson, J.G.; Dubois, D.W.; Countess, R.; Countess, S.J.; Michel, E.L. (2003) Preliminary assessment of PM_{2.5} chemical speciation results for Texas. In Proceedings of the International Conference on Air Quality IV, Arlington, VA, September 22-24.
- Turpin, B.J.; Lim, H.-J. (2001) Species contributions to PM_{2.5} mass concentrations: revisiting common assumptions for estimating organic mass. *Aerosol Sci. Technol.* 35: 602-610.
- Unisys Weather. (2006) Image and Map Archive. <http://weather.unisys.com/archive/index.html> [24 May 2006].
- U.S. Census Bureau. (2000) Census 2000 Gateway. <http://www.census.gov/main/www/cen2000.html> [26 February 2004].
- U.S. Environmental Protection Agency. (1994) Quality assurance handbook for air pollution measurement systems, volume II: ambient air specific methods. EPA 600/R-94-038b.
- U.S. Environmental Protection Agency. (1998) Quality assurance guidance document

- 2.12: monitoring PM_{2.5} in ambient air using designated reference or class I equivalent methods. Research Triangle Park, NC: Human Exposure and Atmospheric Sciences Division.
- U.S. Environmental Protection Agency. (2001) AirData: Access to air pollution data. <http://www.epa.gov/air/data/index.html> [29 December 2005].
- U.S. Environmental Protection Agency. (2005a) Technical support document for the final Clean Air Interstate Rule – air quality modeling. Research Triangle Park, NC: Office of Air Quality Planning and Standards. Available: <http://www.epa.gov/interstateairquality/pdfs/finaltech02.pdf> [18 November 2005].
- U.S. Environmental Protection Agency. (2005b) Technology Transfer Network, Air Toxics Website, Sources of Chronic Dose-Response Information. <http://www.epa.gov/ttn/atw/toxsource/chronicsources.html> [1 December 2005].
- U.S. Environmental Protection Agency. (2006) National Ambient Air Quality Standards. <http://www.epa.gov/air/criteria.html> [21 March 2006].
- U.S. Geological Survey. (2003) Fossil Groups – Spores and Pollen. <http://geology.er.usgs.gov/paleo/sporepollen.shtml>
- Villeneuve, P.J.; Burnett, R.T.; Shi, Y.; Krewski, D.; Goldberg, M.S.; Hertzman, C.; Chen, Y.; Brook, J. (2003) A time-series study of air pollution, socioeconomic status, and mortality in Vancouver, Canada. *J. Expo. Anal. Environ. Epidemiol.* 13: 427-435.
- Weber, R., et al. (2003) Short-term temporal variation in PM_{2.5} mass and chemical composition during the Atlanta supersite experiment, 1999. *J. Air & Waste Manage. Assoc.* 53: 84-91.
- West Virginia University Bureau of Business and Economic Research. (2001) Steubenville-Weirton MSA outlook. Morgantown, WV. Available: http://www.bber.wvu.edu/pdf_files/BBER-2001-11.pdf [3 May 2004].
- Winter, S.E.; Conrad, V.B.; Ross, J. (2002) Direct reaction cell ICP-MS vs. XRF: which is the superior technique for the analysis of water soluble and total elements in fine particulate matter? Presented at the NETL PM_{2.5} and Electricity Power Generation: Recent Findings and Implications conference, Pittsburgh, PA, April 9-10, 2002.
- Winter, S.E.; Bilonick, R.A.; Conrad, V.B.; Withum, J.A.; Statnick, R.M.; Ebel, S.T.; Koutrakis, P.; Suh, H.H. (2004) A field evaluation to compare the performance of personal exposure multi-pollutant samplers vs. federal method monitors at a central ambient air monitoring station in Steubenville, Ohio. In *Proceedings of Air Quality IV*, Arlington VA, September 22-24, 2004.
- Wongphatarakul, V.; Friedlander, S.K.; Pinto, J.P. (1998) A comparative study of PM_{2.5} ambient aerosol chemical databases. *Environ. Sci. Technol.* 32: 3926-3934.
- Zeger, S.L.; Thomas, D.; Dominici, F.; Samet, J.M.; Schwartz, J.; Dockery, D.; Cohen, A. (2000) Exposure measurement error in time-series studies of air pollution: concepts and consequences. *Environ. Health Perspect.* 108: 419-426.
- Zelikoff, J.T.; Schermerhorn, K.R.; Fang, K.; Cohen, M.D.; Schlesinger, R.B. (2002) A role for associated transition metals in the immunotoxicity of inhaled ambient particulate matter. *Environ. Health Perspect.* 110(suppl. 5): 871-875.

BIBLIOGRAPHY

Papers and Presentations Produced as Part of SCAMP

Peer-Reviewed Journal Articles

Adamkiewicz, G.; Ebelt, S.; Syring, M.; Slater, J.; Speizer, F.E.; Schwartz, J.; Suh, H.H.; Gold, D.R. Association Between Air pollution Exposure and Exhaled Nitric Oxide in an Elderly Population; *Thorax*. **2004**, 59, 204–209.

Connell, D.P.; Withum, J.A.; Winter, S.E.; Statnick, R.M.; Bilonick, R.A. The Steubenville Comprehensive Air Monitoring Program (SCAMP): Overview and Statistical Considerations; *J. Air & Waste Manage. Assoc.* **2005**, 55, 467-480.

Connell, D.P.; Withum, J.A.; Winter, S.E.; Statnick, R.M.; Bilonick, R.A. The Steubenville Comprehensive Air Monitoring Program (SCAMP): Associations among Fine Particulate Matter, Co-Pollutants, and Meteorological Conditions; *J. Air & Waste Manage. Assoc.* **2005**, 55, 481-496.

Connell, D.P.; Withum, J.A.; Winter, S.E.; Statnick, R.M.; Bilonick, R.A. The Steubenville Comprehensive Air Monitoring Program (SCAMP): Analysis of Short-Term and Episodic Variations in PM_{2.5} Concentrations Using Hourly Air Monitoring Data; *J. Air & Waste Manage. Assoc.* **2005**, 55, 559-573.

Sarnat, S.E.; Coull, B.A.; Schwartz, J.; Gold, D.R.; Suh, H.H. Factors Affecting the Association Between Ambient Concentrations and Personal Exposures to Particles and Gases; *Environ Health Perspect.* **2006**, 114, 649-654.

Luttmann-Gibson, H.; Suh, H.H.; Coull, B.A.; Dockery, D.W.; Sarnat, S.E.; Schwartz, J.; Stone, P.H.; Gold, D.R. Short-term Effects of Air Pollution on Heart Rate Variability in Senior Adults in Steubenville, OH; *J. Occup. Environ. Med.* **2006**, 48, 780-788.

Sarnat, S.E.; Suh, H.H.; Coull, B.A.; Schwartz, J.; Stone, P.H.; Gold, D.R. Ambient Particulate Air Pollution and Cardiac Arrhythmia in a Panel of Older Adults in Steubenville, OH; *Occup. Environ. Med.* **2006**, 63, 700-706.

Connell, D.P.; Winter, S.E.; Conrad, V.B.; Kim, M.; Crist, K.C. The Steubenville Comprehensive Air Monitoring Program (SCAMP): Concentrations and Solubilities of PM_{2.5} Trace Elements and their Implications for Source Apportionment and Health Research; *J. Air & Waste Manage Assoc.* **2006**, 56, 1750-1766.

Conference Proceedings

Conrad, V.B.; Winter, S.E.; Krofcheck, D.S. ICP-MS and X-Ray Fluorescence: A Comparative Study for the Elemental Analysis of Fine Particulate Matter; Presented at the *International Symposium on Measurement of Toxic and Related Air Pollutants, Research Triangle Park, NC, September 2000*.

Ebelt, S.; Koutrakis, P.; Winter, S.; Conrad, V.; Statnick, R.; Suh, H. Assessment of Exposures of an Elderly Cohort to Fine PM Concentrations in Steubenville, Ohio; Presented at the *Annual Conference of the International Society of Exposure Analysis, Charleston, South Carolina, November 4-8, 2001*.

Withum, J.A.; Winter, S.E.; Conrad, V.B.; Statnick, R.M. The Steubenville Comprehensive Air Monitoring Program (SCAMP): Initial Ambient Air Results; Presented at the *NETL PM_{2.5} and Electricity Power Generation: Recent Findings and Implications, Pittsburgh, PA, April 9-10, 2002*.

Winter, S.E.; Conrad, V.B.; Ross, J. Direct Reaction Cell ICP-MS vs. XRF: Which is the Superior Technique for the Analysis of Water Soluble and Total Elements in Fine Particulate Matter?; Presented at the *NETL PM_{2.5} and Electricity Power Generation: Recent Findings and Implications, Pittsburgh, PA, April 9-10, 2002*.

Ebelt, S.T.; Koutrakis, P.; Winter, S.; Conrad, V.; Statnick, R.; Suh, H.H. Ambient Fine Particulate Matter Exposure Estimation: A Comparison of Approaches; Presented at the *Annual Conference of the International Society of Exposure Analysis, Vancouver, Canada, August 11-15, 2002*.

Ebelt, S.T.; Schwartz, J.; Coull, B.; Koutrakis, P.; Suh, H.H. Examining Factors that Influence the Potential for Confounding in PM Epidemiology; Presented at the *Annual Conference of the International Society of Exposure Analysis, Vancouver, Canada, August 11-15, 2002*.

Adamkiewicz, G.; Ebelt, S.T.; Syring, M.; Slater, J.; Speizer, F.; Schwartz, J.; Suh, H.H.; Gold, D.R. Association between Air Pollution Exposure and Exhaled Nitric Oxide in an Elderly Panel; Presented at the *14th Annual Conference of the International Society for Environmental Epidemiology, Vancouver, Canada August 11-15, 2002*.

Withum, J.A.; Winter, S.E.; Conrad, V.B.; Statnick, R.M. The Steubenville Comprehensive Air Monitoring Program (SCAMP): Initial Ambient Air Results; In *Proceedings of Air Quality III, Arlington, VA, September 10-12, 2002*.

Ebelt, S.T.; Schwartz, J.; Coull, B.; Koutrakis, P.; Suh, H.H. Examining Factors that Influence the Potential for Confounding in PM Epidemiology; Presented at the *American Association for Aerosol Research Conference Particulate Matter: Atmospheric Sciences, Exposure and the Fourth Colloquium on PM and Health, Pittsburgh, PA, March 31-April 4, 2003*.

Winter, S.E.; Bilonick, R.A.; Conrad, V.B.; Withum, J.A.; Statnick, R.M.; Ebelt, S.T.; Koutrakis, P.; Suh, H.H. A Field Evaluation to Compare the Performance of Personal Exposure Multi-Pollutant Samplers vs. Federal Method Monitors at a Central Ambient Air Monitoring Station in Steubenville, Ohio; Presented at the *American Association for Aerosol Research Conference Particulate Matter: Atmospheric Sciences, Exposure and the Fourth Colloquium on PM and Health, Pittsburgh, PA, March 31-April 4, 2003*.

Bilonick, R.A. Assessing the Precision and Accuracy of the Federal Method PM_{2.5} and Personal Exposure Multi-Pollutant Samplers Used in the Steubenville Comprehensive Air Monitoring Program; Presented at the *Joint Statistical Meetings, San Francisco, California, August 3-7, 2003*.

Connell, D.P.; Withum, J.A.; Winter, S.E.; Bilonick, R.A. The Steubenville Comprehensive Air Monitoring Program (SCAMP): An Overview of Outdoor Ambient Results; In *Proceedings of A&WMA's 97th Annual Conference & Exhibition, Indianapolis, IN, June 22-25, 2004*; A&WMA: Pittsburgh, PA, 2004; Paper 619.

Winter, S.E.; Bilonick, R.A.; Conrad, V.B.; Withum, J.A.; Statnick, R.M.; Ebelt, S.T.; Koutrakis, P.; Suh, H.H. A Field Evaluation to Compare the Performance of Personal Exposure Multi-Pollutant Samplers vs. Federal Method Monitors at a Central Ambient Air Monitoring Station in Steubenville, Ohio; In *Proceedings of Air Quality IV, Arlington VA, September 22-24, 2004*.

Connell, D.P.; Winter, S.E.; Withum, J.A. A Current Assessment of the Nature of PM_{2.5} in Steubenville, Ohio, Using SCAMP Monitoring Data; Presented at the *2004 Ohio Air Quality and Coal Research Symposium; Athens, Ohio, December 2-3, 2004*.

Ebelt, S.T.; Suh, H.H.; Coull, B.A.; Schwartz, J.; Stone, P.H.; Gold, D.R. Associations Between Cardiac Arrhythmia And Ambient Air Pollutants In An Elderly Population; Presented at the *16th Annual Conference of the International Society for Environmental Epidemiology, New York, NY, August 1-4, 2004*.

Sullivan, J.R.; Ebelt, S.T.; Suh, H.H. Preliminary Findings from the Steubenville Multi-pollutant Exposure Study of Children; Presented at the *Annual Conference of the International Society of Exposure Analysis, Philadelphia, Pennsylvania, October 17-21, 2004*.

Sullivan J.R.; Slater, J.; Koutrakis, P.; Suh, H.H. Multi-Pollutant Exposure Assessment Study of Children, Steubenville, OH; Presented at the *Annual Conference of the International Society of Exposure Analysis, Philadelphia, Pennsylvania, October 17-21, 2004*.

Ebelt, S.T.; Suh, H.H.; Coull, B.A.; Schwartz, J.; Stone, P.H.; Gold, D.R. Associations between Cardiac Arrhythmia and Ambient Particulate Air Pollution in a Panel of Senior Adults in Steubenville, OH; Presented at the *2005 International Conference of the American Thoracic Society (ATS) Annual Conference, San Diego, CA, May 20-25, 2005*.

Connell, D.P.; Winter, S.E.; Conrad, V.B. The Elemental Composition of PM_{2.5} Collected During the Steubenville Comprehensive Air Monitoring Program (SCAMP); In *Proceedings of A&WMA's 98th Annual Conference & Exhibition, Minneapolis, MN, June 21-24, 2005*; A&WMA: Pittsburgh, PA, 2005; Paper 582.

Luttmann-Gibson, H.; Suh, H.H.; Dockery, D.W.; Coull, B.A.; Ebelt, S.T.; Stone, P.H.; Gold, D.R. Short-term Effects of Air Pollution on Heart Rate Variability in Senior Adults in Steubenville, OH; Presented at the *18th Annual Conference of the International Society for Environmental Epidemiology, Johannesburg, South Africa, September 13-16, 2005*.

Connell, D.P.; Winter, S.E.; Conrad, V.B. Characterization of PM_{2.5} Trace Elements in Steubenville, Ohio, Using Dynamic Reaction Cell ICP-MS; In *Proceedings of Air Quality V, Arlington, VA, September 19-21, 2005*.

Connell, D.P.; Winter, S.E.; Sarnat, S.E.; Sullivan, J.; Koutrakis, P.; Suh, H.H. Key Findings from the Steubenville Comprehensive Air Monitoring Program; In *Proceedings of A&WMA's 99th Annual Conference & Exhibition, New Orleans, LA, June 20-23, 2006*.

FINAL BUDGET

Steubenville Comprehensive Air Monitoring Program Overall Program Budget

(Includes OCDO Grant Agreement CDO/D-98-2
and DOE Cooperative Agreement DE-FC26-00NT40771)

Contributions (through June 2006)

Organization	Amount Obligated, \$	Amount Contributed, \$ (through 6/06)
OCDO	600,000	594,949
CONSOL	100,000	1,094,252
NIEHS, API, EPRI, and EPA - Direct to HSPH	1,105,818	1,108,794
NMA	100,000	99,183
AISI	100,000	99,183
EEI	82,000	81,316
DOE	3,265,500	3,233,825
Total	5,353,318	6,311,502

Expenditures (through June 2006)

Item	Cost, \$ (through 6/06)
Materials	161,447
Subcontracts	2,867,408
Direct Labor	967,520
Labor Overhead	517,569
Travel	21,980
Other Directs (Including Equipment)	385,383
Other	34,486
G&A	1,355,709
Total	6,311,502

**Steubenville Comprehensive Air Monitoring Program
Budget for OCDO Grant Agreement CDO/D-98-2**

Contributions (through June 2006)

Organization	Amount Contributed, \$	% of Total
OCDO	594,949	21.9
CONSOL	732,640	27.0
NIEHS, API, EPRI, and EPA - Direct to HSPH	1,108,794	40.8
NMA	99,183	3.65
AISI	99,183	3.65
EEL	81,316	3.0
Total	2,716,065	100.0

Expenditures (through June 2006)

OCDO, CDO/D-98-2 Item	As per CDO/D-98-2, \$			Actual, \$		
	OCDO	Non-OCDO	Total	OCDO	Non- OCDO	Total
Materials	37,735	17,210	54,945	12,343	45,427	57,770
Subcontract (Harvard)	161,483	1,488,562	1,650,045	377,570	1,428,864	1,806,434
Labor (includes fringe benefits)	204,938	93,466	298,404	106,319	332,851	439,170
Travel	5,443	2,480	7,923	1,369	2,910	4,279
Other Directs (includes equip.)	0	0	0	0	0	0
Other	0	0	0	521	1,959	2,480
G&A	190,401	86,844	277,245	96,827	309,105	405,932
Total	600,000	1,628,562	2,288,562	594,949	2,121,116	2,716,065

ACRONYMS AND ABBREVIATIONS

°C	- temperature in degrees Celsius
µg	- microgram, 10 ⁻⁶ gram
µg/m ³	- micrograms per cubic meter
µm	- micrometer
AC	- air conditioning
ACF	- autocorrelation function
AER	- air exchange rate
AIC	- Akaike Information Criterion
AISI	- American Iron and Steel Institute
API	- American Petroleum Institute
ARIMA	- autoregressive integrated moving average
ATSDR	- Agency for Toxic Substances and Disease Registry
BP	- barometric pressure
CAIR	- Clean Air Interstate Rule
CATs	- capillary adsorption tubes
cc/min	- cubic centimeters per minute
CD	- coefficient of divergence
CEF	- compositional enrichment factor
cm	- centimeter
CMB	- chemical mass balance
CONSOL R&D	- CONSOL Energy Inc., Research and Development
degC	- temperature in degrees Celsius
DI	- deionized
DOE	- U.S. Department of Energy
DRC	- dynamic reaction cell
E	- eastern satellite site
EC	- elemental carbon
EDTA	- ethylenediaminetetraacetic acid
EEI	- Edison Electric Institute
EF	- enrichment factor
EPA	- Environmental Protection Agency
EPRI	- Electric Power Research Institute
EST	- Eastern Standard Time
ETS	- environmental tobacco smoke
Fa	- fall
FEM	- Federal Equivalent Method
FRM	- Federal Reference Method

ACRONYMS AND ABBREVIATIONS

GC/ECD	- gas chromatograph with electron capture detector
HAP	- Hazardous Air Pollutant
hr	- hour
HRV	- heart rate variability
HSPH	- Harvard School of Public Health
IC	- ion chromatography
ICP-MS	- inductively coupled plasma-mass spectrometry
km	- kilometers
kW/m ²	- kilowatts per square meter
L/min	- liters per minute
LOD	- limit of detection
MΩ	- mega-ohm
Max	- Maximum
MCAR	- missing completely at random
mg	- milligram
Min	- Minimum
min	- minute
mL	- milliliter
mmHg	- millimeters of mercury
MRL	- minimal risk level
m/s	- meters per second
MSA	- Metropolitan Statistical Area
m/z	- mass-to-charge ratio
N	- northern satellite site
NA	- not applicable or not available
NAAQS	- National Ambient Air Quality Standards
ND	- non-detect
NETL	- National Energy Technology Laboratory
ng/m ³	- nanograms per cubic meter
NIEHS	- National Institute of Environmental Health Sciences
NIST	- National Institute of Standards and Technology
NMA	- National Mining Association
NRC	- National Research Council
OAQDA	- Ohio Air Quality Development Authority
OAQPS	- Office of Air Quality Planning and Standards
OC	- organic carbon
OCDO	- Ohio Coal Development Office
OM	- organic material

ACRONYMS AND ABBREVIATIONS

PACF	- partial autocorrelation function
PEM	- personal exposure monitor
PFT	- perfluorocarbon tracer gas
PIXE	- proton induced X-ray emission
PM	- particulate matter
PM ₁₀	- particulate matter < 10 µm in diameter
PM _{10-2.5}	- particulate matter > 2.5 µm and < 10 µm in diameter (coarse particulate matter)
PM _{2.5}	- particulate matter < 2.5 µm in diameter (fine particulate matter)
PMF	- Positive Matrix Factorization
ppb	- parts per billion
ppbv	- parts per billion by volume
ppm	- parts per million
PSCF	- potential source contribution function
Q25	- 25th percentile
Q50	- 50th percentile
Q75	- 75th percentile
QA/QC	- quality assurance and quality control
Rad	- solar radiation
RCFM	- reconstructed fine mass
RD	- relative difference
REL	- Reference Exposure Level
RfC	- chronic inhalation reference concentration
RH	- relative humidity
RMSD	- root mean square difference
ROFA	- residual oil fly ash
RSD	- relative standard deviation
S	- southern satellite site
SCAMP	- Steubenville Comprehensive Air Monitoring Program
SD	- standard deviation
SE	- standard error
s.e.	- standard error
Seas	- season
SIP	- State Implementation Plan
SOP	- standard operating procedure
Sp	- spring
SRM	- Standard Reference Material
ST	- Steubenville central monitoring site

ACRONYMS AND ABBREVIATIONS

StdErr	- standard error
Su	- summer
TAD	- time-activity diary
Temp	- temperature
TEOM	- tapered element oscillating microbalance
TOT	- thermal optical transmittance
tot	- acid-digestible fraction
TRI	- Toxics Release Inventory
TSP	- total suspended particulate
URE	- Unit Risk Estimate
UV	- ultraviolet
VENT	- home ventilation variables
vs.	- versus
W	- western satellite site
Wi	- winter
W/m ²	- watts per square meter
WS	- wind speed
ws	- water-extractable fraction
XRF	- X-ray fluorescence spectrometry

APPENDIX A

**AMBIENT AIR MONITORING PROGRAM
SUPPLEMENTAL QA/QC INFORMATION**

Table A.1. Schedule of QA/QC and maintenance activities for SCAMP ambient air pollutant samplers.

Unit	Daily	Per Visit	Weekly / As Needed	Monthly	Bi-monthly	Quarterly	Semi-annually	Annually
RAAS PM ₁₀ and PM _{2.5} FRM Samplers		Wipe carousel, cover plate Exchange impactor Leak test		Verify temperature, pressure, flow	Change intake filter	Clean PM ₁₀ inlet, downtube Change exhaust filter		Calibrate
PM _{2.5} Speciation Sampler		Wipe interior of cabinet				Change denuder Verify temperature, pressure, flow Change intake filter		
Burkard Pollen/Mold Spore Sampler			Check flows Check timing					
TEOM		Check filter load	Change filter Clean cyclone, PM ₁₀ head				Change bypass filters Test batteries, pump Flow control calibration	Clean MFC, and orifices Clean air inlet system Calibrations in Section 3 Change flow controller filters
O ₃ Monitor	Zero / Span check		Change particulate filter			Multi-point cal	TEST functions	Leak check Zero/Span cal Clean lines Sample flow Rebuild pump Clean sample cell Factory calibration
CO Monitor	Zero / Span check		Change particulate filter			Multi-point cal	TEST functions	Leak check Zero/Span cal Clean lines Sample flow Rebuild pump Perma Pure drier Reaction cell

Table A.1 (continued)

Unit	Daily	Per Visit	Weekly / As Needed	Monthly	Bi-monthly	Quarterly	Semi-annually	Annually
NO _x , NO, and NO ₂ Monitor	Zero / Span check		Change particulate filter			Multi-point cal Charcoal on pump	TEST functions Moly converter efficiency	Leak check Zero/Span cal Clean lines Sample flow Rebuild pump
SO ₂ Monitor	Zero / Span check		Change particulate filter			Multi-point cal	TEST functions Sample flow	Leak check Zero/Span cal Clean lines Sample flow Rebuild pump Clean reaction cell
Zero Air System							NO scrubber Charcoal scrubber	CO Scrubber
Mass Flow Calibrator						Leak Check Calibrate O ₃	Inspect Lines Check MFC flows	Check O ₃ perm and bench flows. Clean orifices, replace filters and o-rings

Table A.2. QA/QC for gravimetric analysis of filter-based PM_{2.5} and PM₁₀ samples

Activity	Method and Frequency	Requirements	Out of Specification Action
Microbalance Calibration	1 per year by microbalance maintenance company	Manufacturer's specification.	Repair.
Calibration Mass Standards	1 per year by microbalance maintenance company	NIST traceability and within 0.025 mg.	Replace standard weights.
Temperature and Humidity Instruments	Every 6 months.	Within 2°C, Within 2% RH.	Re-calibrate temp/humidity logger.
Filter Conditioning for Pre-Weighing	Minimum 24 hours (In addition, filters are to be exposed partially open for 1- 2 months.)	Conditioning to the weighing room environment	Repeat conditioning.
Filter Conditioning for Post –Weighing	Minimum 24 hours.	Conditioning to the weighing room environment	Repeat conditioning.
Filter Integrity (Pre-Weighing)	Every filter.	Check for visual defects.	Discard filter.
Filter Integrity (Post-Weighing)	Every filter.	Check for visual defects.	Flag data point.
Balance Calibration Verification	200 mg calibration mass standard weighed every 10 filters.	Within +- 3 ug accepted weight.	Re-zero, re-calibrate with mass standards and re-weigh filters up to the last acceptable balance calibration verification.
Field Blank	Weigh one with every batch of four filters.	Pre and Post weights agree within +-30 ug.	Flag all four filters associated with this blank.
Dedicated Lab Blank	Weigh at the beginning and end of each weighing session.	Within +- 15 ug of accepted weight.	Check for contamination source and re-weigh filters if warranted.
Lot Blank	Choose 1 filter from each lot of filters. Weigh once each session until criteria is met.	Less than 15ug difference for 3 consecutive 24 hour weighing sessions.	Revise conditioning of lot of filters until difference is less than 15 ug for 3 consecutive (24 hour conditioning) weighing sessions.
Duplicate Filter Weighing	Re-weigh 1 filter in duplicate every 10 filters.	Duplicate weighings agree within +- 15 ug.	Re-weigh. Troubleshoot and re-calibrate if necessary.

Table A.3. SCAMP ambient air monitoring program data validation schedule

Parameter	Rule	Action
PM _{2.5} and PM ₁₀ FRM Samples	1. Flow rate coefficient of variation $\geq 2\%$	Invalid
	2. Elapsed sampling time < 23 or > 25 hours	Invalid
	3. Filter temperature errors	Invalid
	4. Average flow not within 5% of 16.67	Invalid
	5. Filters removed > 96 hours from end of run	Invalid
	6. Power failures, pressure anomalies, visible filter/sample defects (e.g., rips, drops, particle anomalies, bugs, water droplets)	Flagged and Noted.
	7. Filter shipping temperature $> 4^{\circ}\text{C}$	Flagged and Noted.
	8. Weighing room: Temperature $\neq 68 - 74^{\circ}\text{F}$ or RH $\neq 20-50\%$ (30-40% target)	Flagged and Noted.
	9. Field blank pre- and post-weights differ by $> 30\ \mu\text{g}$.	Flagged and Noted.
PM _{2.5} Speciation Sampler Samples	1. Elapsed sampling time < 23 or > 25 hours	Invalid
	2. Filter temperature errors	Invalid
	3. Power failures, pressure anomalies, visible filter/sample defects (e.g., rips, drops, particle anomalies, bugs, water droplets)	Flagged and Noted.
	4. Filter shipping temperature $> 4^{\circ}\text{C}$	Flagged and Noted.
	5. Weighing room: Temperature $\neq 68 - 74^{\circ}\text{F}$ or RH $\neq 20-50\%$ (30-40% target)	Flagged and Noted.
	6. Field blank pre- and post-weights differ by $> 30\ \mu\text{g}$.	Flagged and Noted.
SO ₂	1. $\{2\ \text{ppb} < D_z \leq 3\ \text{ppb}\}$ OR $\{5\% < D_s \leq 7.5\%\}$	Flagged
	2. $\{ D_z > 3\ \text{ppb}\}$ OR $\{ D_s > 7.5\%\}$	Invalid
NO	1. $\{0.5\ \text{ppb} < D_z \leq 0.75\ \text{ppb}\}$ OR $\{2.5\% < D_s \leq 3.75\%\}$	Flagged
	2. $\{ D_z > 0.75\ \text{ppb}\}$ OR $\{ D_s > 3.75\%\}$	Invalid
NO _x	1. $\{1.5\ \text{ppb} < D_z \leq 2.25\ \text{ppb}\}$ OR $\{2.5\% < D_s \leq 3.75\%\}$	Flagged
	2. $\{ D_z > 2.25\ \text{ppb}\}$ OR $\{ D_s > 3.75\%\}$	Invalid
NO ₂	1. Corresponding NO or NO _x data are flagged	Flagged
	2. Corresponding NO or NO _x data are invalid	Invalid
CO	1. $\{1\ \text{ppm} < D_z \leq 1.5\ \text{ppm}\}$ OR $\{2.5\% < D_s \leq 3.75\%\}$	Flagged
	2. $\{ D_z > 1.5\ \text{ppm}\}$ OR $\{ D_s > 3.75\%\}$	Invalid
O ₃	1. $\{2.5\ \text{ppb} < D_z \leq 3.75\ \text{ppb}\}$ OR $\{2\% < D_s \leq 3\%\}$	Flagged
	2. $\{ D_z > 3.75\ \text{ppb}\}$ OR $\{ D_s > 3\%\}$	Invalid
TEOM	TEOMSTAT $\neq 4.0$ ("normal") OR TEOMFILT > 95	Invalid
Meteorological Data	Examined and validated by staff meteorologist at Optimal Technologies	
Pollen and Mold Spore Samples	Examined and validated by Air Quality Sciences Laboratory	

NOTE: D_z = zero drift of analyzer (checked daily) = Z' ; D_s = span drift of analyzer (checked daily) = $(m_c - 1) \times 100$, where Z' = unadjusted zero reading (ppm or ppb), S' = unadjusted span reading (ppm or ppb), C = span gas concentration (ppm or ppb), and m_c = current analyzer calibration slope = $(S' - Z')/C$.

APPENDIX B

INDOOR AND PERSONAL EXPOSURE PROGRAM QUESTIONNAIRE AND TIME-ACTIVITY DIARY FORMS

One-time Dwelling Questionnaire

Harvard School of Public Health, Steubenville, OH – Fall 2000

Date: _____

Participant ID: _____

Operator Initials:

1. Location of apartment

a) Where is your home located? *(circle one)*

Kennedy

Gaylord

Other: _____ *(address)*

If living in Kennedy or Gaylord:

b) On which floor is the apartment located? _____

c) On which side of the building? *(circle one)*

North

South

East

West

d) Is this a corner unit?

Yes

No

2. General apartment characteristics

a) How many rooms does your home have? _____

b) Percent of carpeting and/or floor rugs in each room: *(check the applicable box for each)*

Room	Carpeting?	Rugs?	None?
Kitchen			
Living room			
Dining room			
Bedroom			
Bathroom			
Hallways			
Other: _____			
Other: _____			
Other: _____			

3. Ventilation Characteristics

a) How many window/wall AC units are in the home?

of window/wall units: _____

b) What are the heating sources in the home? (*circle or specify*)

radiators (steam or hot water)

forced air (vents)

open stove

electric space heater

gas space heater

kerosene space heater

wood burning stove

fireplace

other: _____

c) Do you tend to use any heating sources during the fall? Yes No

d) What is the thermostat setting(s)? _____

e) Is there a whole-house fan? Yes No

f) Do you use any individual fans? Yes No

g) Are there storm windows? Yes No

h) Does the apartment have a balcony? Yes No

i) How would you best describe the VENTILATION FACTOR in this unit? _____
(*Specify a number between 0 and 3 (0.5 is fresh, 2.5 is very stuffy)*)

4. Cooking/Fuel Characteristics

a) What type of cooking fuel is used? (*circle or specify*)

Gas

Electric

Other: _____

b) Is there a fan over the cooking stove or elsewhere in the kitchen area? Yes No

c) If so, how does this fan work? (*circle or specify*)

Kitchen exhaust vented outside

Recirculation of indoor air

Charcoal filter

Other: _____

Do not know

d) Is there a clothes dryer? Yes No

e) If yes, is the clothes dryer unvented? Yes No

f) Are there any pilot lights on the: (*circle all applicable*)

Gas range

Oven

Clothes dryer

g) What type of filter bag is used in the vacuum cleaner? (*circle or specify*)

Standard filter

High efficiency filter (HEPA)

Other:

5. Placement of indoor sampling set up in apartment

(*specify room, location within room, approximate height, proximity to potential sources ie TV, kitchen, windows and doors, candles etc...*)

DAYTIME

SUMMER 2000 TIME-ACTIVITY DIARY – STEUBENVILLE, OH

PARTICIPANT ID: _____

Date:

Sampling Day: 1 2 (circle one)

PAGE ONE

Time	Activity	Location	<u>In transit:</u>	<u>Outdoors:</u>	Near busy road?	<u>Ventilation:</u>		Cooking or near cooking?	Dusting, sweeping, vacuuming?	Near smoker?
			If yes, how many min?	If yes, how many min?		A/C or window?				
7:00 - 7:30 am			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
7:30 - 8:00 am			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
8:00 - 8:30 am			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
8:30 - 9:00 am			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
9:00 - 9:30 am			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
9:30 - 10:00 am			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
10:00- 10:30 am			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
10:30 - 11:00 am			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
11:00 - 11:30 am			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
11:30 am -Noon			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
Noon - 12:30 pm			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
12:30 - 1:00 pm			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
1:00 - 1:30 pm			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
1:30 - 2:00 pm			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
2:00 - 2:30 pm			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
2:30 - 3:00 pm			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
3:00 - 3:30 pm			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
3:30 - 4:00 pm			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
4:00 - 4:30 pm			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
4:30 - 5:00 pm			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
5:00 - 5:30 pm			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
5:30 - 6:00 pm			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
6:00 - 6:30 pm			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
6:30 - 7:00 pm			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>

Date: _____

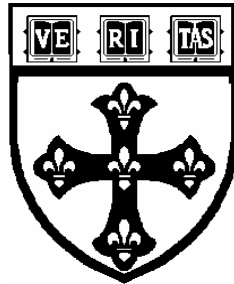
Time	Activity	Location Please specify where you are during this activity	In transit: If yes, how many min?	Outdoors: If yes, how many min?	Near busy road?	Ventilation:		Cooking or near cooking?	Dusting, sweeping, vacuuming?	Near smoker?
						A/C or	Window?			
7:00 - 7:30 pm			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
7:30 - 8:00 pm			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
8:00 - 8:30 pm			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
8:30 - 9:00 pm			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
9:00 - 9:30 pm			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
9:30 - 10:00 pm			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
10:00- 10:30 pm			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
10:30 - 11:00 pm			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
11:00 - 11:30 pm			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
11:30 - Midnight			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
Midnight - 12:30			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
12:30 - 1:00 am			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
1:00 - 1:30 am			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
1:30 - 2:00 am			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
2:00 - 2:30 am			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
2:30 - 3:00 am			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
3:00 - 3:30 am			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
3:30 - 4:00 am			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
4:00 - 4:30 am			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
4:30 - 5:00 am			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
5:00 - 5:30 am			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
5:30 - 6:00 am			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
6:00 - 6:30 am			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
6:30 - 7:00 am			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>

Time	Activity	Location Please specify where you are during this activity	In transit: If yes, how many min?	Outdoors: If yes, how many min?	Near busy road?	Ventilation: A/C or window?		Cooking or near cooking?	Dusting, sweeping, vacuuming?	Near smoker?
						A/C	WIN			
7:00 - 7:30 am			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
7:30 - 8:00 am			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
8:00 - 8:30 am			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
8:30 - 9:00 am			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
9:00 - 9:30 am			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
9:30 - 10:00 am			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
10:00- 10:30 am			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
10:30 - 11:00 am			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
11:00 - 11:30 am			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
11:30 am -Noon			<input type="checkbox"/> /	<input type="checkbox"/> /	<input type="checkbox"/>	A/C	WIN	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>

Indoors at Home	Indoors Away from Home	In Transit	Outdoors
Kitchen	Grocery store/Pharmacy	Car	Gas station
Bedroom	Hospital/Doctor's office	Bus	Park
Bathroom	Senior Center	Train	Bus stop
Dining room	Hair dresser		Beach
Living room	Shopping mall/Bank		Someone else's yard

For use by study team only:
 Initials of collector: _____
 Start time: _____ AM
 End time: _____ AM
 Comments: _____

HARVARD SCHOOL OF PUBLIC HEALTH
Steubenville, OH - Winter, 2001



QUESTIONNAIRE ON HOUSEHOLD AND PARTICIPANT'S CHARACTERISTICS

I. HOUSEHOLD IDENTIFICATION:

1. Address

Number Street Apartment

City Zip Code

--	--	--	--	--

2. Telephone:

			-				-				
--	--	--	---	--	--	--	---	--	--	--	--

3. Social Security Number:

				-			-				
--	--	--	--	---	--	--	---	--	--	--	--

The purpose of this questionnaire is to obtain information about you and your residence. All the information will be kept confidential.

II. HOUSEHOLD CHARACTERISTICS:

TYPE OF RESIDENCE:

1. What type of cooking fuel do you use?	Gas.....1 Electric.....2 Other.....3
2. Does your house have storm windows?	Yes.....1 No.....2 IF YES, ON WHAT % OF WINDOWS _____
2.1 What type of heating do you use during the winter?	Central.....1 Radiators.....2 Kerosene/Space.....3 Wood fireplace.....4 Other _____.....9
2.2 Do you use a wood-burning fireplace?	Yes.....1 No.....2 IF YES, HOW FREQUENTLY _____
3. Is there a fan over the cooking stove, range, oven, or elsewhere in the kitchen area?	Yes.....1 No.....2 IF NO, GO TO 6
4. How does this fan work?	Kitchen exhaust vented to the outside.....1 Recirculation of indoor air.2 Charcoal filter.....3 Other, please specify _____ 8 Don't know.....9 Not applicable.....0
5. How often is this fan used?	Most of the time when the stove is used.....1 Occasionally.....2 Rarely.....3 Never.....4 Don't know.....9 Not applicable.....0
6. Do you have: Circle all that apply	An unvented clothes dryer located in the house or an attached structure, such as a garage?.....1 An unvented kerosene heater in the house or an attached structure?.....2 A fireplace or wood stove in the house or an attached structure?.....3 A whole-house or attic fan?.....4

7. Do you have any air cleaning devices in your home?	Yes.....1 No.....2
8. What type of air cleaning device(s) do you have at home? Circle all that apply	Ion generator.....1 Filter.....2 Electrostatic precipitator.....3 Other, please specify...8 _____ Not applicable.....0
9. What type of garage, if any do you have?	None, detached or separate carport.....1 Attached.....2 Underneath.....3 Don't Know.....9 IF NONE, GO TO 12
10. Is this garage used for:	Parking one car.....1 Parking two cars.....2 Storage only.....3
11. Does this garage:	Have a pull-down door.....1 Electric door.....2 No door.....3
12. Where do you store your paints and solvents?	Indoors.....1 Outdoors.....2 In the garage.....3 No paints/solvents4 Other, please specify...8 _____
13. How many separate central AC or window/wall units do you have in your home?	_____ number of central AC units _____ number of window/wall units
14. Do you have a humidifier in your home?	Yes.....1 No.....2
15. How frequently is your home <u>dusted</u> ?	Once per day.....1 Once per week.....2 Once every two weeks.....3 Once per month.....4 Less than once per month.....5 Twice per week.....6
16. How frequently is your home <u>vacuumed</u> ?	Once per day.....1 Once per week.....2 Once every two weeks.....3 Once per month.....4 Less than once per month.....5
17. Is your vacuum in good working order?	Yes.....1 No.....2 Don't know.....9
18. Do you have a gas-powered lawn mower?	Yes.....1 No.....2 IF NO, GO TO 20
19a. If yes, how many times a month do you use this mower?	Number of times/month _____

19b. How many months a year do you use this mower?	Number of months/year _____
20. Do you have any pets, such as dogs or cats or other furry animals, which usually spend some time each day <u>in</u> your home?	Yes.....1 No.....2 IF NO, GO TO 23
21. If so, how many are there?	One.....1 Two.....2 Three to five.....3 More than five.....4 Not applicable.....0
22. Could you tell me which pets you have? <p style="text-align: center;">Circle all that apply</p>	Cat(s).....1 Dog(s).....2 Bird(s).....3 Other, please specify...8 _____ Not applicable.....0
23. How many people regularly smoke <u>inside</u> your home?	0.....1 1.....2 2.....3 3 or more.....4
24. Did anyone smoke any tobacco products in your home during the last seven days?	Yes.....1 No.....2 Don't Know.....9

IV. HOUSEHOLD RESIDENTS:

25. How many people currently live in this home?	Number of People ()
--------------------------------------------------	--------------------------

For each person living in this house, including yourself, please list the following information in the table below.

- a) Age, in years at last birthday
 - b) Type of work of each person currently employed outside of the home for 30 hours a week or more.
 - c) Current smoking status (yes/no).
- (The person is considered a smoker if he/she smoke at least 1 cigarette, cigar, or pipeful per day.)

(26) Household Member's Name	(27) Age	(28) Type of Work Outside the Home (30 hrs. a Week or More)	(29) Smoker (Y/N)
1			
2			
3			
4			
5			
6			
7			

31. Is the house located within 100 yards of a busy roadway	YES.....1 NO.....2
32. Is there a dirt drive located within 100 yards of this house?	YES.....1 NO.....2
33. Are there any other sources of dirt located within 100 yards of the house?	YES.....1 NO.....2 If the answer is YES, please specify :

List each room in the house and estimate the percentage of usable floor space covered by rugs or carpets.

	(34) Room	(35) % of floor covered by rug or carpet
1		
2		
3		
4		
5		
6		
7		
8		
9		
10		
11		

THANK YOU VERY MUCH FOR YOUR TIME AND COOPERATION!

THIS PAGE TO BE COMPLETED BY INTERVIEWER ONLY

Interviewer Initials: _____

36. How would you best describe the VENTILATION FACTOR in this unit?	Codes from 0 to 3: 0.5 is fresh 2.5 is stuffy
37. How would you best describe the DUST FACTOR in this unit?	Codes from 0 to 3: 0.5 is immaculate 2.5 is clutter

END OF QUESTIONNAIRE

INDOOR SAMPLING: Daily Follow-up Questionnaire

Harvard School of Public Health – Steubenville, OH Winter 2001

(Daily_Follow_Up_Questionnaire.doc 1/26/01)

Date: _____ Participant ID: _____ Day: 2 3 4 5 6 7 8 (circle) Operator Initials: _____

1.	How many people spent > 4 hours in your house in the last 24 hours?		
2.	How many cigarettes or cigars were smoked inside your home in the last 24 hours?		
3.	Was any cleaning done in your home in the last 24 hours? (No = 0, Yes = 1)		
4.	Were any meals cooked in your home in the last 24 hours? (No = 0, Yes = 1)		
	<i>What kind of cooking was done? Who cooked? How many times?</i>	<i>Self</i>	<i>Other</i>
4a.	Frying, grilling, sautéing, broiling? (circle which apply)		
4b.	Baking?		
4c.	Toasting?		
4d.	Boiling? (<i>circle: water, soup, reheating liquid</i>)		
4e.	Microwave?		
4f.	Was any food burned in your home in the last 24 hours? (No = 0, Yes = 1)		
4g.	Was the exhaust fan used for any cooking activity? (No = 0, Yes = 1)		
5.	For how long were candles burned in your house in the last 24 hours? (# of hours)		
5a.	For how long was incense used in you home in the last 24 hours? (# of hours)		
6.	Were any outdoor windows or doors open in the last 24 hours? (No = 0, Yes = 1)		
	<i>How many inches wide were they open and how many hours were they open?</i>	<i>Inches</i>	<i>Hours</i>
	Door 1		
	Door 2		
	Window 1		
	Window 2		
	Window 3		
	Window 4		
7.	For how long was an air conditioner used in your home in the last 24 hours? (# hrs)		
8.	For how long was the heating turned on in your home in the last 24 hours? (# hrs)		
8a.	For how long was a space heater used in your home in the last 24 hours? (# hrs) (<i>circle: electric, kerosene, gas</i>)		
9.	For how long was a humidifier used in your home in the last 24 hours? (# of hours)		

APPENDIX C

**DATA - (SEPARATE COMPACT DISC CONTAINING
AIR MONITORING AND EXPOSURE DATA)**

APPENDIX D

THE STEUBENVILLE COMPREHENSIVE AIR MONITORING PROGRAM (SCAMP): CONCENTRATIONS AND SOLUBILITIES OF PM_{2.5} TRACE ELEMENTS AND THEIR IMPLICATIONS FOR SOURCE APPORTIONMENT AND HEALTH RESEARCH

*MANUSCRIPT SUBMITTED TO THE JOURNAL OF THE AIR & WASTE
MANAGEMENT ASSOCIATION*

1 **THE STEUBENVILLE COMPREHENSIVE AIR MONITORING PROGRAM (SCAMP):**
2 **CONCENTRATIONS AND SOLUBILITIES OF PM_{2.5} TRACE ELEMENTS AND**
3 **THEIR IMPLICATIONS FOR SOURCE APPORTIONMENT AND HEALTH**
4 **RESEARCH**

5
6 **Daniel P. Connell, Stephen E. Winter, Vincent B. Conrad**
7 *CONSOL Energy Inc. Research & Development, South Park, Pennsylvania*

8
9 **Myoungwoo Kim, Kevin C. Crist**
10 *Center for Air Quality, Ohio University, Athens, Ohio*

11
12 **ABSTRACT**

13 The elemental compositions of the water-soluble and acid-digestible fractions of 24-hr integrated
14 PM_{2.5} samples collected in Steubenville, OH, from 2000-2002 were determined using dynamic
15 reaction cell ICP-MS. The water-soluble elemental compositions of PM_{2.5} samples collected at
16 four satellite monitoring sites in the surrounding region were also determined. Fe was the most
17 abundant but least water-soluble of the elements determined at the Steubenville site, having a
18 mean ambient concentration of 272 ng/m³ and a median fractional solubility of 6%. Fe solubility
19 and its correlations with SO₄²⁻ and temperature varied significantly by season, consistent with the
20 hypothesis that secondary sulfates may help to mobilize soluble Fe under suitable summertime
21 photochemical conditions. Significantly higher ambient concentrations were observed at
22 Steubenville than at each of the four satellite sites for 10 of the 18 elements (Al, As, Ca, Cd, Fe,
23 Mg, Mn, Na, Pb, and Zn) determined in the water-soluble PM_{2.5} fraction. Concentrations of Fe,
24 Mn, and Zn at Steubenville were substantially higher than concentrations recently reported for
25 larger U.S. cities. Receptor modeling identified seven sources affecting the Steubenville site.
26 An (NH₄)₂SO₄-dominated source, likely representing secondary PM_{2.5} from coal-fired plants to
27 the west and southwest of Steubenville, accounted for 42% of the PM_{2.5} mass, and two sources
28 likely dominated by emissions from motor vehicles and from iron and steel facilities in the
29 immediate Steubenville vicinity accounted for 20% and 10%, respectively. Other sources
30 included a NH₄NO₃ source (15%), a crustal source (6%), a mixed nonferrous metals and
31 industrial source (3%), and a primary coal combustion source (3%). Results suggest the

32 importance of very different regional and local source mechanisms in contributing to PM_{2.5} mass
33 at Steubenville, and reinforce the need for further research to elucidate whether metals such as
34 Fe, Mn, and Zn play a role in the PM_{2.5} health effects previously observed there.

35

36 **IMPLICATIONS**

37 Steubenville, OH, is located in one of the 18 counties projected by the U.S. EPA to remain in
38 nonattainment under the annual PM_{2.5} NAAQS in 2015 after the implementation of CAIR. This
39 paper presents source apportionment results that will be useful when developing an
40 implementation plan to address Steubenville's nonattainment status. Moreover, elevated
41 concentrations of PM_{2.5} in Steubenville have been epidemiologically associated with increased
42 mortality. Given recent toxicological evidence linking certain soluble particulate transition
43 metals with health endpoints, the characterization of transition metal concentrations and
44 solubilities presented here provides important insights that can be utilized by future studies to
45 help determine whether these metals play any role in the health effects previously attributed to
46 PM_{2.5} in Steubenville.

47

48 **INTRODUCTION**

49 Although trace elements generally account for only a few percent of the total mass of ambient
50 fine particulate matter (PM_{2.5}) in the United States, their characterization is an essential part of
51 PM_{2.5} source apportionment and health effects research. Numerous recent receptor modeling
52 studies¹⁻⁴ have utilized fine particulate trace element concentration data to help identify sources
53 that emit PM_{2.5} or its precursors. These studies increasingly are employing techniques such as
54 backward trajectory analysis and potential source contribution function (PSCF) modeling⁵⁻⁶ to
55 geographically corroborate the compositional source profiles derived from multivariate models
56 like Positive Matrix Factorization (PMF) or Unmix. The results are particularly relevant given
57 the need to develop State Implementation Plans (SIPs) to reduce ambient PM_{2.5} concentrations in
58 nonattainment areas under the National Ambient Air Quality Standard (NAAQS) for PM_{2.5}.⁷

59 Moreover, toxicology and epidemiology studies have reported associations between
60 adverse health effects and the trace metal content of particulate matter (PM).⁸⁻¹¹ Growing
61 evidence suggests that transition metals such as Fe, V, and Zn, and especially soluble forms of
62 these metals, play a role in inducing pulmonary responses to PM. For instance, Zelikoff et al.¹²

63 found that exposure to soluble divalent Fe resulted in significant increases in bacterial burdens in
64 rats with preexisting *Streptococcus pneumoniae* infections; the result was similar to that
65 observed when the rats were exposed to concentrated ambient PM_{2.5} from New York City.
66 Residual oil combustion emissions, characterized by elevated concentrations of V and Ni, have
67 been associated with cellular stress responses in human bronchial epithelial (HBE) cells;¹³ Carter
68 et al.¹⁴ reported that residual oil fly ash (ROFA) and vanadium-containing compounds, but not
69 nickel sulfates, triggered the production of inflammatory cytokines by normal HBE cells. In a
70 different study,¹⁵ both iron (III) sulfate and vanadyl sulfate, but not nickel sulfate, were
71 demonstrated to increase the expression of lactoferrin receptors on respiratory epithelial cells,
72 possibly indicating a response to oxidative stress induced by these catalytically active metals.
73 Huang et al.¹⁶ observed that a SO₄²⁻/Fe/Se factor was associated with pulmonary neutrophilic
74 inflammation and a Cu/Zn/Fe factor was associated with increased blood fibrinogen in healthy
75 humans exposed to concentrated ambient particles. Zinc particles also have been implicated in
76 pulmonary health effect studies.¹⁷ For example, Adamson et al.¹⁸ discovered that soluble Zn
77 salts stimulated increases in inflammatory cells and protein in lung lavage fluid from exposed
78 mice, whereas solutions containing Cu, Fe, Al, Pb, Mg, and Ni induced minimal lung effects.

79 Current interest in the source apportionment and health effects of ambient PM_{2.5} and its
80 chemical constituents stems in part from research conducted in Steubenville, OH. The landmark
81 Harvard Six Cities Study,¹⁹ which discovered an association between Steubenville's elevated
82 PM_{2.5} concentrations and mortality using data from 1979-1985, partially drove the promulgation
83 of the PM_{2.5} NAAQS. In addition to having the highest atmospheric concentrations of PM_{2.5} and
84 SO₄²⁻ of the six cities studied, Spengler and Thurston²⁰ reported that concentrations of Fe, Mn,
85 and Se measured in Steubenville between 1979 and 1981 were greater than those measured in the
86 other five cities during comparable time periods, and concentrations of V and Ni in Steubenville
87 were second highest among the cities. Koutrakis and Spengler²¹ applied specific rotation factor
88 analysis to PM_{2.5} elemental data collected in Steubenville in 1984 and identified six sources of
89 PM_{2.5}. Fe and Mn (as well as Zn and K) were primarily apportioned to iron and steel production;
90 Se and Ni were primarily apportioned to coal/oil combustion, and V was primarily apportioned
91 to titanium production and coal/oil combustion. More recently, Laden et al.²² used elemental
92 data collected in the six cities between 1979 and 1988 in an epidemiology study examining
93 associations between daily mortality and three generic PM_{2.5} source factors: a Si factor (crustal

94 material), a Pb factor (automobiles), and a Se factor (coal combustion). None of the factors were
95 significantly associated with mortality in Steubenville, although associations between mortality
96 and the Pb and Se factors were statistically significant in a combined analysis across all six
97 cities. Grahame and Hidy²³ challenged these findings, arguing that the Se factor in Boston likely
98 included substantial residual oil emissions that were mislabeled as coal combustion emissions,
99 perhaps causing an overstatement of the risk associated with the latter. Grahame and Hidy also
100 suggest that much of the V and Se attributed by Koutrakis and Spengler to coal and oil
101 combustion in Steubenville instead likely originated from coke and steel production facilities.
102 Hence, there is still considerable uncertainty regarding which particular components or sources
103 of PM_{2.5} in Steubenville, if any, are responsible for the association between PM_{2.5} and mortality
104 observed in the Six Cities Study.

105 As discussed by Connell et al.,²⁴ Steubenville's population, industrial activity, and air
106 pollution have declined appreciably since the Six Cities Study was conducted. Average PM_{2.5}
107 concentrations in Steubenville today are more than 35% lower than the 29.6 µg/m³ reported by
108 Dockery et al.¹⁹ for the 1979-1985 time period. Nevertheless, Jefferson County, OH, where
109 Steubenville is located, is a nonattainment area under the annual NAAQS for PM_{2.5} and is one of
110 the 18 counties projected by the U.S. Environmental Protection Agency (EPA) to remain in
111 nonattainment in 2015 after implementation of the Clean Air Interstate Rule (CAIR).²⁵ Hence,
112 given Steubenville's nonattainment status, the substantial changes that have occurred there since
113 the Six Cities Study, and the recent findings concerning possible health implications of PM_{2.5}
114 transition metals, a current characterization of PM_{2.5} trace elements in Steubenville is warranted.

115 This paper provides such an assessment based on elemental analyses of PM_{2.5} samples
116 collected between May 2000 and May 2002 as part of the Steubenville Comprehensive Air
117 Monitoring Program (SCAMP). Whereas previous PM_{2.5} studies²⁰⁻²² in Steubenville used X-ray
118 fluorescence spectroscopy (XRF) to determine the elemental composition of PM_{2.5}, SCAMP
119 used dynamic reaction cell (DRC)²⁶ inductively coupled plasma-mass spectrometry (ICP-MS)
120 for this purpose. ICP-MS has been employed by several recent PM_{2.5} studies²⁷⁻²⁸ because it
121 exhibits better sensitivity than conventional energy dispersive XRF for determining certain
122 elements that are present in part-per-billion to part-per-trillion amounts. However, the
123 capabilities of conventional low-resolution ICP-MS are limited by polyatomic and isobaric
124 interferences that impede the determination of isotopes such as ³⁹K, ⁴⁰Ca, ⁵¹V, ⁵²Cr, ⁵⁶Fe, ⁷⁵As,

125 and ⁸⁰Se. SCAMP is among the first PM_{2.5} studies to utilize ICP-MS equipped with a DRC in
126 order to minimize these interferences.

127 Also, unlike previous studies in Steubenville, which determined only the elemental
128 composition of the total PM_{2.5} sample, SCAMP included determinations of the composition of
129 both the water-soluble and acid-digestible PM_{2.5} fractions, allowing the fractional solubility of
130 PM_{2.5} trace elements to be estimated. These solubility results, presented here, provide some
131 indication of the bioavailability of fine particulate trace metals, and are particularly relevant
132 given the aforementioned toxicological evidence linking soluble metals with pulmonary health
133 endpoints. In addition to data collected at the central SCAMP monitoring site in Steubenville,
134 the elemental composition of the water-soluble PM_{2.5} fraction was determined at four
135 comparatively remote satellite sites surrounding Steubenville, permitting an assessment of the
136 impact of local sources in Steubenville on soluble PM_{2.5} trace element concentrations. This
137 paper also presents a recent source apportionment of PM_{2.5} in Steubenville based on the
138 application of PMF and PSCF to compositional data from the central SCAMP monitoring site;
139 results will be useful in the development of an implementation plan to solve Steubenville's
140 NAAQS nonattainment status.

141

142 **METHODS**

143 **Sample Collection and Analysis**

144 Details of SCAMP, including results for total PM_{2.5} mass, major PM_{2.5} components, gaseous
145 pollutants, and meteorological conditions, have been presented in three previous papers.^{24,29,30}
146 Figure 1 shows the locations of the five SCAMP outdoor ambient air monitoring sites. The
147 central site in Steubenville was situated on the campus of Franciscan University of Steubenville
148 (40.362° N, 80.615° W) atop a bluff overlooking the Ohio River, which flows from
149 approximately due north to due south in the Steubenville vicinity. Many of the region's major
150 industrial facilities, including iron and steel production facilities, coke plants, coal-fired electric
151 power plants, and other manufacturing facilities, are located along the river's banks. The four
152 satellite sites were located approximately 19 km to the north near New Manchester, WV; 33 km
153 to the south in Wheeling, WV; 108 km to the east in Latrobe, PA; and 22 km to the west in
154 Hopedale, OH.

155 At each site, 24-hr integrated PM_{2.5} samples were collected daily between May 13, 2000,
156 and May 14, 2002, using an Andersen RAAS2.5-300 Federal Reference Method (FRM) sampler.
157 All samples were collected from ~ 9:00 a.m. to 9:00 a.m. EST on 47-mm Teflon filters at a
158 sampling flow rate of ~16.67 L/min. The PM_{2.5} samples collected every fourth day at each site
159 were analyzed to determine the elemental composition of their water-soluble fractions. The
160 Teflon filters containing these samples were transferred into rinsed polypropylene digestion
161 tubes using Teflon-coated forceps and covered with 25 mL of 18-MΩ deionized water
162 containing 0.2% isopropanol as a wetting agent. The tubes were then sonicated for 30 minutes,
163 and an aliquot of the resulting leachate from each sample was submitted for elemental analysis.
164 An internal standard solution containing Ge and In was added to each aliquot so that instrument
165 drift could be monitored and corrected.

166 In addition, 24-hr integrated PM_{2.5} samples were collected on a 1-in-4 day frequency at
167 the central site in Steubenville between August 12, 2000, and May 14, 2002, using an Andersen
168 RAAS2.5-400 PM_{2.5} speciation sampler. To facilitate DRC ICP-MS method development and
169 allow for reanalyses required during this process, PM_{2.5} samples for acid-digestible elemental
170 analysis were collected in duplicate on 47-mm Teflon filters using separate channels of the
171 speciation sampler. Both channels sampled from ~ 9:00 a.m. to 9:00 a.m. EST at sampling flow
172 rates of ~16.67 L/min. All filters from the primary channel were digested and analyzed first;
173 however, because of several method-related issues encountered during this succession of
174 analyses, all filters from the backup channel were subsequently digested and analyzed in order to
175 improve valid data capture. In cases where valid results for an element were obtained from both
176 the primary and backup filters for a given day, the arithmetic mean concentration is reported. If
177 a valid result was obtained from only one of the two filters collected on a given day, this single
178 valid result is reported.

179 To prepare the PM_{2.5} samples for acid-digestible elemental analysis, each exposed filter
180 was placed in a rinsed polypropylene digestion tube, and an acid solution containing 1.39 mL of
181 Fisher Scientific TraceMetal grade HNO₃ (67-70%) and 0.11 mL of an 80:35 mixture of
182 TraceMetal grade HF (47-51%) and HCl (34-37%) was added. Although it requires careful
183 handling, the HF is required for more complete digestion³¹ of elements contained in the siliceous
184 matrices commonly encountered in PM_{2.5}. The samples were then digested at 90°C for 16-18 hr.
185 After cooling, 48.5 mL of a dilute boric acid solution (used to complex the HF) and 0.5 mL of an

186 internal standard solution containing Ge and In were added to each sample. Isopropyl alcohol, a
187 wetting agent, was also added to each digestate during the second sequence of analyses. The
188 digestates were heated to 90°C for an additional hour before submittal for elemental analysis.

189 Concentrations of elements in the water extracts and acid digestates were determined
190 using a PerkinElmer ELAN 6100 DRC ICP-MS equipped with a cross-flow nebulizer and HF-
191 resistant torch. The instrument operates in the same way as a conventional quadrupole ICP-MS
192 except that a reaction cell located between the ion optics system and the analyzing quadrupole is
193 employed to selectively remove species that would otherwise interfere with the determination of
194 the desired analytes. Interferences in ICP-MS can result from isobaric species (e.g., $^{40}\text{Ar}^+$, which
195 interferes with the determination of ^{40}Ca) or polyatomic species (e.g., $^{40}\text{Ar}^{16}\text{O}^+$, which interferes
196 with the determination of ^{56}Fe) that are formed from plasma gases or chemical components of
197 the sample matrix and have the same mass-to-charge ratio (m/z) as the desired analyte.³² In the
198 reaction cell, a gas that readily reacts with the interfering species but not with the desired analyte
199 is introduced when necessary in order to alter the m/z of these interfering species. An active
200 filtering quadrupole located in the reaction cell is tuned to eject species falling outside of a
201 specific mass bandpass window, thereby preventing the introduction of new interferants and
202 controlling the m/z characteristics of the ion beam that is sent to the analyzing quadrupole. Thus,
203 the reaction gas and active quadrupole in the reaction cell collectively reduce the number of
204 interfering species that pass to the analyzing quadrupole. Additional details concerning the
205 theory and performance of DRC ICP-MS have been presented previously.^{26,33}

206 In this study, CH_4 and NH_3 were used as reaction gases in the DRC when necessary.
207 Table 1 summarizes the general strategy used to determine elements in the water-soluble and
208 acid-digestible $\text{PM}_{2.5}$ fractions. Because of the nature of DRC ICP-MS, the same element at
209 times was determined using different instrumental methods, which included variations in the
210 isotope being determined, the reaction gas (if any) being used, and the reaction gas flow rate. All
211 elemental determinations were validated based upon determinations of quality control standards,
212 including NIST Standard Reference Material (SRM) 1643d or NIST SRM 1640 for water-
213 soluble elements and NIST SRM 1648 for acid-digestible elements, that were included during
214 the course of each run. Due to the use of multiple methods to determine certain elements,
215 multiple analytical limits of detection (LOD) were obtained for these elements; these LODs have
216 been reported previously.^{34,35} To avoid the introduction of bias,³⁶ actual instrument readings for

217 determinations below the LOD were used in the statistical analyses presented in this paper. As
218 shown in Table 1, most elements had only a small percentage of observations below the LOD,
219 indicating that DRC ICP-MS exhibited sufficient sensitivity for the concentrations being
220 examined in this study.

221 Table 1 also presents statistics describing method performance, as measured by the
222 recovery of elements in NIST standard reference materials. The NIST SRM 2783 was used to
223 assess method performance for the acid-digestible fraction because it allows an evaluation of the
224 success of the digestion procedure for a PM_{2.5}-impregnated filter. Mean recoveries were within
225 15% of target values for 16 elements in SRM 1643d (analyzed by the water-soluble method) and
226 11 elements in SRM 2783 (analyzed by the acid-digestible method), and were within 30% of
227 target values for all elements having NIST-certified values. The overall imprecision of the
228 method for elements in the acid-digestible fraction, as computed using results from analyses of
229 collocated filters collected during SCAMP, ranged from 15-34% depending upon the element.
230 Results for Ti in the water-soluble fraction and Na in the acid-digestible fraction routinely failed
231 to pass quality control criteria; hence, these results are excluded from Table 1 and the remainder
232 of this paper. Si and Cr were also routinely determined, but are excluded from the analyses
233 presented here because of high blank concentrations (for water-soluble Si and acid-digestible
234 Cr), a frequent failure to meet quality control criteria (for acid-digestible Si), or a high
235 percentage of observations below the LOD (for water-soluble Cr).

236 Source Apportionment

237 Source apportionment was performed using Positive Matrix Factorization (specifically, PMF2)
238 and potential source contribution function modeling. PMF, as developed by Paatero,³⁷ is a factor
239 analysis algorithm for solving the general receptor modeling problem given by

$$240 \quad x_{ij} = \sum_{k=1}^p g_{ik} f_{kj} + e_{ij} \quad (1)$$

241 where x_{ij} is the ambient mass concentration of the j th species from the i th PM_{2.5} sample collected
242 at the receptor site; g_{ik} is the PM_{2.5} mass concentration from the k th source contributing to the i th
243 sample; f_{kj} is the mass fraction of the j th species in PM_{2.5} from the k th source; e_{ij} is the residual
244 associated with x_{ij} , and p is the total number of sources contributing to PM_{2.5} at the receptor site.
245 Because values of both g_{ik} and f_{kj} are unknown, there are an infinite number of solutions to eq 1
246 corresponding to different rotations of the matrix of source contributions, G , and matrix of

247 source compositions, F. Hence, the selection of a final solution necessarily involves some
248 subjectivity. PMF imposes non-negativity constraints to help limit the number of possible
249 solutions, and solves for G and F by using an iterative weighted least-squares procedure to
250 minimize an objective function, $Q(E)$, which is defined as

$$251 \quad Q(E) = \sum_{i=1}^n \sum_{j=1}^m \left[\frac{x_{ij} - \sum_{k=1}^p g_{ik} f_{kj}}{u_{ij}} \right]^2 \quad (2)$$

252 Here, u_{ij} is the uncertainty in the j th species measured in the i th sample; n is the total number of
253 samples, and m is the total number of species. Additional details of PMF have been described
254 extensively in the literature.^{2-4,37}

255 The receptor modeling results presented in this paper are based on the application of PMF
256 to PM_{2.5} mass, ion, carbon, and acid-digestible elemental data collected at the central monitoring
257 site in Steubenville. Only days for which data were available for mass, ions (NH₄⁺, SO₄²⁻, NO₃⁻,
258 and Cl⁻), elemental and organic carbon (EC and OC), and at least 10 out of the 18 acid-digestible
259 elements were used; as a result, 115 days were included in the model runs. Because analytical
260 uncertainties were not determined in a consistent manner for all of the included species,
261 uncertainty values for use in PMF were assigned based on the analytical LODs for each species.
262 For determinations above the LOD, concentration-dependent uncertainties were assigned by
263 multiplying each determined concentration by the ratio of its LOD to the overall 25th percentile
264 species concentration. This procedure inherently down-weighted species such as Cl⁻, Co, and Ni
265 for which determined concentrations were often not much greater than the detectable limits of
266 the analytical method being used. For determinations below the LOD, the uncertainty was set
267 equal to 5/6 times the LOD. Any missing values for a species during the 115 days being
268 modeled were replaced with the overall median concentration of that species, and the
269 uncertainties for these missing values were set at four times the median species concentration in
270 order to decrease their weight in the model. To account for possible underestimation of
271 uncertainties according to the procedure above and possible violations of the assumption that the
272 chemical profiles of sources impacting the Steubenville site do not change with time, an extra
273 5% uncertainty was added to all observations prior to modeling.

274 PMF was run in robust mode to reduce the impact of extreme values, which were defined
275 as data points for which the model residual was greater than four times the error estimate, on the
276 solution. Instead of employing the normal multiple regression procedure to determine PM_{2.5}
277 mass apportionment, PM_{2.5} mass concentration was included as an independent variable in the
278 model runs to directly obtain this apportionment.² PMF was run several times to examine the
279 effects of altering the specified number of sources and the value of the FPEAK parameter, which
280 is used to control rotations of the F and G matrices. The final 7-source PMF model that is
281 presented in this paper was selected based on an evaluation of the model goodness-of-fit (e.g.,
282 robust Q, scaled residuals) and the physical reasonableness of the source profiles obtained from
283 the various candidate solutions. The robust Q for this 7-source solution, Q = 3231, agreed
284 reasonably well with the theoretical Q of 2875 for the dataset being modeled, and the FPEAK
285 value (FPEAK = 0.0) occurred in a range of FPEAK values for which $Q(E)$ exhibited a minimum
286 and was relatively insensitive to changes in FPEAK (Figure 2). Five different seeds of initial
287 values were used to test the optimality of the PMF solutions. Uncertainties in the elements of the
288 F and G matrices were estimated by PMF2 in the standard way from the uncertainties in the
289 ambient concentration values via alternating least squares fits.³⁸

290 Potential source contribution function modeling was employed to help identify probable
291 geographic locations of the sources resolved by PMF. Backward trajectories from the
292 Steubenville site were computed for every hour of each of the 115 days included in the PMF
293 model. All trajectory calculations were performed using the HYSPLIT model³⁹ with
294 NCEP/NCAR Reanalysis data.⁴⁰ Because many sources of PM_{2.5} in Steubenville are expected to
295 be located within about 100 km of the monitoring site, 24-hr trajectories with a starting height of
296 50 m above model ground level were typically used. These backward trajectories were utilized
297 to compute the PSCF for each of the seven sources resolved by PMF.

298 The geographical region covered by the backward trajectories was divided into grid cells
299 with a resolution of 0.1° latitude by 0.1° longitude. For a given source, the potential source
300 contribution function for the ij th grid cell, $PSCF_{ij}$, is computed as

$$301 \quad PSCF_{ij} = \frac{m_{ij}}{n_{ij}} \quad (3)$$

302 where m_{ij} is the number of trajectories passing through the cell that are associated with a “high
303 concentration” of PM_{2.5} from that source observed at the receptor site, and n_{ij} is the total number

304 of trajectories passing through the cell. Hence, a $PSCF_{ij}$ value close to one indicates a high
 305 probability that the ij th cell either contains the source under consideration or is situated along a
 306 commonly occurring transport route that results in the source impacting the receptor site. For the
 307 analyses presented in this paper, all of the 24 backward trajectories computed for a given day
 308 were considered to be associated with the 24-hr average source contributions resolved by PMF
 309 for that day. A source was considered to have a “high concentration” at the receptor site on a
 310 given day if its estimated contribution to $PM_{2.5}$ at the receptor site on that day was greater than
 311 the overall geometric mean of its estimated daily contributions. To prevent the occurrence of
 312 large PSCF values with large uncertainties that result from small values of n_{ij} , all PSCF values
 313 were multiplied by an arbitrary weight function factor, $W(n_{ij})$, which served to down-weight
 314 those values for which n_{ij} was less than three times the average n_{ij} :

$$315 \quad W(n_{ij}) = \begin{cases} 1.0 & 42 < n_{ij} \\ 0.7 & 10 < n_{ij} \leq 42 \\ 0.5 & 6 < n_{ij} \leq 10 \\ 0.3 & 4 < n_{ij} \leq 6 \\ 0.1 & 0 < n_{ij} \leq 4 \end{cases} \quad (4)$$

316

317 **RESULTS AND DISCUSSION**

318 **Ambient Concentrations of Elements in $PM_{2.5}$ at Steubenville**

319 Table 2 summarizes the 24-hr average concentrations of elements in the acid-digestible $PM_{2.5}$
 320 fraction measured at Steubenville during SCAMP. Based upon the mean concentrations
 321 observed during the program, these 18 elements collectively accounted for about $0.8 \mu\text{g}/\text{m}^3$ or
 322 4.3% of the total $PM_{2.5}$ mass at Steubenville. Fe was the most abundant of the elements that
 323 were determined, as it alone constituted about $0.3 \mu\text{g}/\text{m}^3$ or 1.5% of the total average $PM_{2.5}$ mass.
 324 Twenty-four-hour average Fe concentrations greater than $1.0 \mu\text{g}/\text{m}^3$ were measured nine times
 325 during the program.

326 Average concentrations of 11 of the $PM_{2.5}$ elements (Al, Ca, Cu, Fe, K, Mn, Ni, Pb, Se,
 327 V, and Zn) measured during SCAMP were also determined in Steubenville between 1979 and
 328 1988 as part of the Harvard Six Cities Study.²² Apart from Ca, elemental concentrations
 329 measured during SCAMP were 37% (Se) to 92% (Pb) less than those measured during the Six
 330 Cities Study. These differences may in part be due to analytical error; however, the trend of

331 decreasing ambient concentrations is consistent with trends observed for total PM_{2.5} mass.^{24,41}
332 The large decrease in Pb concentrations likely reflects the discontinued use of leaded gasoline.
333 The comparatively small decrease in Se concentrations is also reasonable given the continued
334 prevalence of coal combustion and coking plants in the Steubenville region. These plants are a
335 major source of Se, a volatile element that is present in coal and tends to be less effectively
336 controlled by particulate control devices than less-volatile elements.^{42,43}

337 Despite the decreases that have occurred since the Six Cities Study, concentrations of
338 certain PM_{2.5} transition metals measured in Steubenville during SCAMP remained elevated
339 relative to concentrations reported for other U.S. cities during comparable time periods. For
340 example, although concentrations of Fe and Zn measured during SCAMP were 50% and 39%
341 lower, respectively, than those reported by Laden et al.²² for Steubenville, concentrations of
342 these elements at Steubenville during SCAMP were 1.2–3.2 (Fe) and 2.0-7.0 (Zn) times as great
343 as concentrations summarized by Grahame⁴⁴ (using AIRS data) for New York, NY, Baltimore,
344 MD, St. Louis, MO, Boston, MA, and Chicago, IL, for the year 2001. Moreover, mean
345 concentrations of Fe and Zn observed during SCAMP were 2.4-3.0 (Fe) and 5.2-5.3 (Zn) times
346 greater than concentrations measured during recent sampling campaigns in Washington, DC,⁴⁵
347 and Atlanta, GA.⁴⁶ Mean concentrations of Mn measured during SCAMP were 6.6-7.6 times as
348 high as those reported for Washington and Atlanta. Conversely, concentrations of fine
349 particulate Cu, Ni, and V at Steubenville during SCAMP were less than the average
350 concentrations reported for most of the seven cities named above. Again, these comparisons
351 likely include some error resulting from differences in the analytical methods used in the various
352 cities, but the consistently observed differences for Fe, Mn, and Zn are larger than would be
353 expected to result solely from analytical error.

354 Seven of the 18 elements listed in Table 2 (As, Cd, Co, Pb, Mn, Ni, and Se) are defined
355 as Hazardous Air Pollutants (HAPs) in the Clean Air Act. Mean airborne concentrations of all of
356 these elements in PM_{2.5} at Steubenville during SCAMP were less than the non-cancer chronic
357 inhalation reference concentrations (RfCs),⁴⁷ minimal risk levels (MRLs),⁴⁸ and reference
358 exposure levels (RELs)⁴⁹ established by the U.S. EPA, U.S. Agency for Toxic Substances and
359 Disease Registry, and California EPA. The mean Mn concentration at Steubenville was about
360 37% of the chronic inhalation MRL; concentrations of the other six elements were at least an
361 order of magnitude less than the most stringent reference concentrations. However, total

362 ambient concentrations of these elements at Steubenville may be greater than the concentrations
363 reported in Table 2, especially for elements such as Mn that are more abundant in crustal
364 material,⁵⁰ as SCAMP measured PM_{2.5} rather than total suspended particulate (TSP). Fe and Zn
365 are not classified as HAPs; however, given their high concentrations at Steubenville compared to
366 other U.S. cities and the aforementioned toxicological findings concerning their possible
367 association with pulmonary health endpoints, further study of these metals is warranted to
368 determine whether they play any role in the observed association between PM_{2.5} and mortality in
369 Steubenville.

370 Solubility of Elements in PM_{2.5} at Steubenville

371 Table 2 also shows the median and 25th and 75th percentile daily fractional solubilities for each
372 element that was determined in both the water-soluble and acid-digestible PM_{2.5} fractions at the
373 Steubenville site. Based on these data, fine particulate As, Ba, Cd, K, and Se tended to be
374 present in predominantly water-soluble forms at Steubenville; each of these elements had a
375 median fractional solubility > 0.8. This list of highly soluble elements includes three of the
376 seven HAPs discussed above, including the two (As and Cd) that are listed as carcinogens by the
377 U.S. EPA.⁴⁷ Conversely, fine particulate Al, Fe, and Sn were largely present in insoluble forms
378 at Steubenville; these elements had median fractional solubilities < 0.2.

379 As shown in Table 2, fractional solubilities computed for As and Se at Steubenville were
380 often substantially > 1.0. This apparent physical incongruity may be the result of a sampling
381 bias, as samples for water-soluble and acid-digestible analyses were collected using separate
382 samplers. It could also be attributable to an analytical error (e.g., ⁷⁵As and ⁸⁰Se are prone to
383 interferences by species such as ⁴⁰Ar³⁵Cl⁺ and ⁴⁰Ar₂⁺, respectively)³² or digestion error (e.g.,
384 volatile losses of AsCl₃ and SeOCl₂ can occur after digestion in hot HCl).⁵¹ These potential
385 errors are guarded against by use of the DRC and a capped digestion procedure with a post-
386 digestion cooling period, though, and are not supported by the performance data presented in
387 Table 1. Pairwise concentrations of As and Se measured in the water-soluble and acid-digestible
388 PM_{2.5} fractions were well correlated (Spearman correlation coefficients, r_s, were 0.89 for As and
389 0.88 for Se), indicating that these concentrations are suitable for use in trending and source
390 apportionment analyses. It is noteworthy, however, that the mean concentrations of acid-
391 digestible As and Se reported in Table 2 may be biased low. Mean ambient air concentrations of
392 these elements in the water-soluble PM_{2.5} fraction were 2.59 ng/m³ for As and 4.73 ng/m³ for Se.

393 Given the toxicological findings concerning possible health implications of soluble PM_{2.5}
394 transition metals, there is considerable interest in elucidating factors that affect the solubility of
395 these metals. Figure 3 shows the ratio of the median warm season (April – September) 24-hr
396 average fractional solubility to the median cool season (October – March) solubility for each of
397 the 17 elements for which data were available. Warm-to-cool season ratios are also shown for
398 concentrations of these elements in the water-soluble and acid-digestible PM_{2.5} fractions. As
399 shown in the figure, fractional solubilities of Al, Fe, and Pb were significantly (i.e., at a statistical
400 significance level of $\alpha = 0.05$) greater during the warm season than during the cool season.
401 Fractional solubilities of the remaining 14 elements did not differ significantly by season.

402 Although Fe is the least soluble of the 17 elements studied as part of SCAMP, its
403 fractional solubility exhibited the greatest variation by season at Steubenville. The fractional
404 solubility of Fe is of particular interest given evidence⁵² that even a very small amount of soluble
405 iron in particles (i.e., 0.036% of the total Fe content) may be capable of inducing oxidant
406 generation leading to inflammation. The trend of higher Fe fractional solubility during the warm
407 season than during the cool season at Steubenville could be attributable to seasonal variations in
408 primary emissions of soluble forms of this metal. Alternatively, the increased fractional
409 solubilities observed during the warm season may be related to atmospheric chemistry. Grahame
410 and Schlesinger⁵³ recently recommended further research to examine the extent to which ambient
411 PM_{2.5} metals are made soluble by contact with secondary sulfates. Previous studies^{54,55} have
412 illustrated the photoreduction of Fe(III) to Fe(II), which is generally more mobile than Fe(III),⁵⁶
413 in atmospheric aerosols. Ghio et al.⁵⁷ hypothesized that SO₄²⁻ may participate in the
414 solubilization of particulate Fe by serving as a ligand for Fe after its mobilization by
415 photoreduction, and demonstrated via controlled laboratory experiments that aqueous SO₄²⁻
416 (from CaSO₄·2H₂O) is capable of mobilizing soluble Fe from Fe₂O₃, but only in the presence of
417 light. Results from SCAMP are consistent with the possibility that atmospheric Fe in
418 Steubenville may be solubilized to some extent by interaction with sulfates under suitable
419 photochemical conditions. The fractional solubility of Fe in Steubenville was greatest during the
420 warm season, when photochemical activity and secondary SO₄²⁻ concentrations are maximal.
421 Moreover, as shown in Table 3, 24-hr average Fe fractional solubilities exhibited moderate,
422 positive correlations with SO₄²⁻ and temperature during summertime in Steubenville, whereas
423 correlations among these variables were comparatively weak or negative during other, less

424 photochemically active seasons. (Correlations of Cu, Ni, and Sn with SO_4^{2-} and temperature
425 exhibited similar trends). During the summer, Fe fractional solubility correlated more strongly
426 with SO_4^{2-} than it did with any of the other measured $\text{PM}_{2.5}$ components. These results reinforce
427 the need for further research to investigate possible effects of secondary sulfates on the solubility
428 of Fe in ambient $\text{PM}_{2.5}$. Such research should also examine whether aerosol acidity, which was
429 not directly measured as part of SCAMP, plays a role in solubilizing Fe. Additional research on
430 the health effects of soluble fine particulate Fe is also warranted, as several studies^{58,59} have
431 failed to discover any statistically significant associations between Fe and adverse health
432 endpoints, and another⁶⁰ has suggested that both the soluble and insoluble fractions of PM
433 contain ionizable metals that can catalyze oxidative stress.

434 Although fine particulate Al and Pb tended to be more soluble during the warm season
435 than during the cool season at Steubenville, fractional solubilities of these two metals were not
436 well correlated with SO_4^{2-} or temperature during summertime. Conversely, Al fractional
437 solubilities exhibited moderately strong positive correlations with SO_4^{2-} and temperature during
438 winter. Moreover, fractional solubilities of both Al and Zn were well correlated with SO_4^{2-} but
439 not temperature during autumn. Although the results are not shown, Al and Zn fractional
440 solubilities in autumn also correlated moderately-to-strongly (i.e., $r_s > 0.7$) with concentrations of
441 a number of other $\text{PM}_{2.5}$ chemical components, including elemental and organic carbon and
442 several trace elements. These results are more suggestive of primary emissions of soluble forms
443 of these metals, perhaps during inversion-driven pollution episodes that have been shown³⁰ to
444 promote buildups in the concentrations of locally-emitted pollutants during non-summer months
445 in Steubenville, than of a photochemically-driven mechanism of solubilization. Soluble Zn may
446 have health implications; Fernandez et al.⁶¹ recently reported that exposure to ZnSO_4 aerosol led
447 to a significant increase in lung permeability in mice, whereas exposure to ZnO aerosol did not
448 induce such a response.

449 **Spatial Variability of Elements in the Water-Soluble $\text{PM}_{2.5}$ Fraction**

450 Table 4 shows the median 24-hr average ambient air concentrations of elements in the
451 water-soluble $\text{PM}_{2.5}$ fraction determined at each of the SCAMP monitoring sites between May
452 2000 and May 2002. The two-tailed p-values listed for each satellite site indicate whether
453 element concentrations measured at the satellite site differed significantly from those measured
454 at the Steubenville site, based on the application of a Wilcoxon signed rank test to paired

455 concentrations measured at the two sites. Ambient concentrations at Steubenville were
456 significantly greater (at $\alpha = 0.05$) than those measured at all four satellite sites for 10 of the 18
457 elements studied. The northern and western sites are most representative of background $PM_{2.5}$
458 concentrations in the Steubenville region;²⁴ concentrations of 14 elements at the northern site and
459 15 elements at the western site were significantly less than corresponding concentrations
460 measured at Steubenville. Water-soluble Fe, Mg, and Mn were particularly enriched at
461 Steubenville; median ambient air concentrations of these elements at the Steubenville site were
462 at least 1.9 times as great as median concentrations at the northern and western satellite sites.
463 This enrichment likely reflects the impact of the local iron and steel industry in the immediate
464 Steubenville vicinity, as elevated airborne concentrations of particulate Fe and Mn are
465 commonly found near steel production facilities.⁶² Ambient concentrations of Al and Zn in the
466 water-soluble $PM_{2.5}$ fraction were also substantially greater at Steubenville than at the
467 background sites.

468 Figure 4 shows Spearman correlation coefficients describing the associations between
469 water-soluble element concentrations measured at Steubenville and corresponding water-soluble
470 element concentrations measured at the satellite sites. Concentrations at the satellite sites were
471 generally moderately correlated with those at Steubenville, and for many of the elements, the
472 strength of association tended to decrease with increasing inter-site distance. These results likely
473 reflect the impact of local sources in Steubenville on $PM_{2.5}$ trace element concentrations at
474 nearby monitoring sites, as well as the effect of regional-scale meteorology on localized ambient
475 concentrations of trace elements emitted by local sources situated near each satellite site. The
476 weakest correlations were between Mn concentrations at the Steubenville site and Mn
477 concentrations at the southern and eastern sites. This, coupled with the data in Table 4, likely
478 indicates the very localized impact of substantial Mn emissions from sources in Steubenville.

479 **Source Apportionment at Steubenville**

480 Figures 5 and 6, respectively, present the source profiles and source contribution time series for
481 each of the seven sources identified by applying PMF to $PM_{2.5}$ mass and speciation data
482 collected at the Steubenville site. As shown in Figure 7, which compares daily reconstructed
483 $PM_{2.5}$ mass concentrations estimated from the PMF source contributions with actual measured
484 $PM_{2.5}$ concentrations, the model succeeded in accounting for much of the variation in $PM_{2.5}$ mass
485 at Steubenville. Although results are not presented here, data from the Steubenville site were

486 also modeled using the EPA Unmix 2.3 receptor model.⁶³ Each of the seven sources resolved by
487 PMF resembled a source identified in one or more of the three best 5-source Unmix solutions,
488 and source contributions estimated by the two different models for these pairs of similar sources
489 were moderately-to-strongly correlated ($r_s = 0.58-0.96$), providing further evidence of the
490 validity of the results.

491 Moreover, as discussed earlier, PMF was run a number of times to explore the effects of
492 altering the specified number of sources on the solution. The best eight-source solution, which
493 like the seven-source solution had a robust Q value ($Q = 2601$) near the theoretical Q value of
494 2875, resolved two separate carbon-dominated sources rather than the single source (Source 2)
495 resolved by the seven-source model. However, the identities of these two sources, which also
496 included small amounts of the mass apportioned to Sources 1, 3, 4, 5, and 7 by the seven-source
497 model, were not individually interpretable; one was dominated by EC but contained no OC, and
498 the other was dominated by OC but contained no EC. The six-source solution had a robust Q
499 value that was nearly 50% greater than the theoretical Q value, and among other problems, it
500 blended various elements of Sources 2, 3, and 6 from the seven-source model into two sources
501 such that the identities of these mixed sources were difficult to ascertain. Overall, the seven-
502 source solution appeared to have the most appropriate degree of source resolution given the size
503 of the dataset being modeled and known characteristics of $PM_{2.5}$ sources in the Steubenville area.

504 Among the sources identified by this final seven-source PMF solution, Source 1 is the
505 largest contributor to $PM_{2.5}$ mass at Steubenville, accounting for $7.4 \mu\text{g}/\text{m}^3$, or ~ 42%, of the total
506 apportioned mass. SO_4^{2-} and NH_4^+ account for ~ 71% of the mass associated with this source,
507 whereas the fractional contribution of trace elements to the total source mass is an order of
508 magnitude less than for any other source. The molar ratio of NH_4^+ to SO_4^{2-} is 2.0 for Source 1
509 (the mass of NO_3^- associated with this source is negligible), suggesting that this source represents
510 the contribution of secondary $(\text{NH}_4)_2\text{SO}_4$ -dominated aerosol to $PM_{2.5}$ mass at Steubenville. As
511 shown in Figure 6a and Table 5, respectively, the $PM_{2.5}$ attributed to Source 1 exhibited a
512 pronounced seasonal pattern of higher summertime and lower wintertime concentrations and
513 displayed a moderate, positive correlation with temperature, reflecting the effect of increased
514 summertime photochemical activity on secondary SO_4^{2-} formation. SO_2 emitted by coal-fired
515 power plants is expected to be the dominant precursor of secondary SO_4^{2-} in the upper Ohio
516 River valley. PSCF results for Source 1, shown in Figure 8a for 24-hr backward trajectories with

517 a 50-m starting height, suggest a probable source region to the south of Steubenville, including
518 the Ohio River valley, where several coal-fired power plants are located. For a regional
519 secondary pollutant such as $(\text{NH}_4)_2\text{SO}_4$, however, a larger PSCF domain is warranted to allow
520 time for atmospheric formation and transport. Figure 8h shows PSCF results calculated for
521 Source 1 using 3-day backward trajectories with a 500-m starting height, which was selected for
522 longer-range transport calculations because it approximately represents the middle of the mixed
523 layer. Although the areas of high probability do not all correspond to locations of major power
524 plants, the results generally suggest transport from the west and southwest of Steubenville, where
525 many such plants are situated.

526 Sources 2 and 3 are the most similar of the seven sources resolved by PMF. The source
527 contributions from these sources correlate more strongly ($r_s = 0.63$) than do those from any other
528 pair of sources, and exhibit moderate-to-strong correlations with CO, NO, and NO₂ (Table 5).
529 Source 2, which accounts for ~ 20% of the total apportioned PM_{2.5} mass at Steubenville, is
530 characterized by the largest mass fractions of EC and organic material ($\text{OM} = 1.4 \cdot \text{OC}$) of the
531 seven sources. This source also contains more of the total apportioned mass of Ba and Cu than
532 any of the other sources. EC is often attributed to diesel emissions,⁶⁴ and Ba, which is used as a
533 smoke suppressant in diesel fuel and as a filler in brake linings, has been suggested as a tracer for
534 heavy-duty diesel trucks.⁶⁵ Moreover, Cu is found in engine oils and brake linings,⁶⁵ and other
535 elements present in the Source 2 profile, including Fe, Ca, and Pb, are expected in fine particle
536 emissions from motor vehicles.⁶⁶ The EC/OC ratio for Source 2 (0.32) is considerably less than
537 the ratio of ≥ 1 expected for diesel emissions,⁶⁷ but is only slightly less than the average EC/OC
538 ratio of 0.44 reported by Lough et al.⁶⁶ for mixed vehicular emissions in two Milwaukee, WI,
539 tunnels. Hence, given its source profile and its strong correlations with NO_x and CO, both of
540 which are emitted by motor vehicles, Source 2 likely represents the mixed contribution of local
541 gasoline- and diesel-fueled vehicle emissions to ambient PM_{2.5} concentrations in Steubenville.
542 The absence of Zn in the Source 2 profile is unexpected, as Zn is commonly found in engine oils,
543 tires, and brake linings;^{65,66} however, this may be due to modeling error, especially given the
544 presence of several other strong sources of Zn affecting the Steubenville site. It is also possible
545 that Source 2, which contains 84% of the total mass of apportioned EC and 66% of the total mass
546 of apportioned OM, includes some contribution from non-vehicular sources of fine particulate
547 carbon. Possible other sources include iron and steel facilities,⁶⁸ coke production facilities,⁶⁹ and

548 vegetative burning; estimated source contributions for Source 2 were more highly correlated ($r_s =$
549 0.73) with water-soluble K, which is commonly used as a tracer for vegetative burning,⁷⁰ than
550 were contributions for any of the other sources resolved by PMF ($r_s = 0.06-0.70$).

551 As discussed above, the iron and steel industry is expected to be a major source of Fe and
552 Mn in Steubenville. Source 3, which accounts for ~ 10% of the total apportioned PM_{2.5} mass at
553 Steubenville, is characterized by the largest mass fractions of Fe and Mn of the seven sources
554 resolved by PMF, and likely represents the contribution of local iron and steel emissions to PM_{2.5}
555 in Steubenville. More than half of the total mass of Fe, Mn, and Zn and more than 40% of the
556 total mass of Mg measured at the central Steubenville site were apportioned to this source. Iron
557 oxide emissions are expected from blast furnaces, basic oxygen furnaces, and sintering
558 operations, and magnesium oxide emissions are expected from blast furnaces.⁷¹ Moreover,
559 based on 2001 Toxics Release Inventory (TRI) data from the U.S. EPA,⁷² the primary metals
560 industry is the largest point source emitter of Mn and Zn in Jefferson County, OH, where
561 Steubenville is located. The Mn/Zn mass ratio of 0.18 in the Source 3 profile agrees reasonably
562 well with the ratio of 0.16 computed using the Jefferson County TRI data for the primary metals
563 industry. Koutrakis and Spengler²¹ also resolved an iron and steel production source in
564 Steubenville that was characterized by high mass concentrations of Fe, Mn, and Zn; their source
565 accounted for ~ 9% of total PM_{2.5} mass and had a Mn/Zn mass ratio of 0.15, similar to our
566 Source 3. As shown in Table 5, Source 3 exhibited the strongest correlation with CO of the
567 seven sources resolved by PMF. Based on 1999 data from the U.S. EPA,⁷³ the ferrous metals
568 processing industry accounts for ~ 47% of estimated CO emissions from point and area sources
569 in the Steubenville-Weirton Metropolitan Statistical Area, consistent with our interpretation of
570 the identity of Source 3.

571 The aforementioned similarities between Sources 2 and 3 may result from the fact that
572 these factors both represent the contribution of emission sources located predominantly in the
573 immediate Steubenville vicinity to the mass of ambient PM_{2.5} in Steubenville. The correlations
574 presented in Table 6 support this hypothesis. As shown in the table, the estimated daily source
575 contributions for Sources 2 and 3 correlate more strongly than do the source contributions for
576 any other sources with “local source” PM_{2.5} concentrations estimated by subtracting daily
577 averages of the PM_{2.5} concentrations measured at the northern and western SCAMP satellite sites
578 from daily PM_{2.5} concentrations measured at the Steubenville site. Moreover, the source

579 contributions for Sources 2 and 3 exhibit the strongest correlations with nighttime maximum
580 PM_{2.5} concentrations of the seven PMF sources, and correlate much more strongly with
581 nighttime maximum PM_{2.5} concentrations than with daytime maxima. These correlations are
582 consistent with our interpretation of these two sources, as build-ups in the emissions from
583 vehicles and from iron and steel facilities, which have relatively short stacks, would tend to be
584 promoted by reductions in mixing height overnight and during early morning rush hour. PSCF
585 results for Source 2, shown in Figure 8b, show a high-probability source region surrounding
586 Steubenville in all directions, consistent with the hypothesis that vehicular and other local carbon
587 emissions impacted the Steubenville site regardless of the prevailing transport direction in the
588 surrounding region. The PSCF plot for Source 3 shown in Figure 8c suggests that this source
589 tended to impact the monitoring site when transport was from the south, consistent with the
590 location of major iron and steel production facilities just south of Steubenville.

591 Source 4 accounts for ~15% of the total apportioned PM_{2.5} mass and represents the
592 contribution of NH₄NO₃-dominated aerosol to PM_{2.5} in Steubenville during SCAMP. As shown
593 in Figure 6d, the mass contributed by this source exhibited a marked seasonal pattern of higher
594 wintertime and lower summertime concentrations, reflecting the influence of temperature,
595 humidity, and NH₃ availability on secondary NO₃⁻ formation. The PSCF results shown in Figure
596 8i (3-day backward trajectories, 500-m starting height) suggest that NO_x emissions from the
597 Columbus, Dayton, Cincinnati, and Indianapolis areas to the west⁷³ likely affect NH₄NO₃
598 concentrations in Steubenville.

599 Source 5, which accounts for ~6% of the total apportioned PM_{2.5} mass at Steubenville
600 and has a source profile that is characterized by the largest mass fractions of Al, Ca, K, Mg, and
601 Ti of the seven sources resolved by PMF, likely represents the contribution of crustal material to
602 Steubenville's ambient PM_{2.5}. The Mg/Al and Ti/Al mass ratios of 0.30 and 0.09, respectively,
603 computed from the source profile for Source 5 agree well with corresponding ratios of 0.26 and
604 0.05 computed from the average composition of the Earth's crust reported by Mason.⁵⁰ The
605 ratios of Ca/Al and K/Al computed from the Source 5 profile were greater and the ratio of Fe/Al
606 was less than corresponding ratios computed from Mason's data. These differences may result
607 from disparities between the composition of the crustal material affecting the Steubenville site
608 and the Earth's overall average crustal composition or from modeling or measurement error. As
609 shown in Table 5, daily source contributions for Source 5, like those for the secondary

610 $(\text{NH}_4)_2\text{SO}_4$ and NH_4NO_3 sources, did not correlate strongly with any gaseous pollutants.
611 However, Source 5 daily contributions correlated more strongly with temperature than did
612 contributions from any other source. This may reflect the increased tendency for crustal particles
613 to become airborne during warm periods when dust-producing outdoor activities are common
614 and the ground is dry and free from snow cover.

615 The remaining two sources, Source 6 and Source 7, each accounted for $\sim 0.6 \mu\text{g}/\text{m}^3$ ($\sim 3\%$)
616 of the total apportioned $\text{PM}_{2.5}$ mass at Steubenville. The profile for Source 6 is characterized by
617 the largest mass fractions of As, Cu, Sn, V, Zn, and especially Cd and Pb of the seven sources
618 resolved by PMF, but contains no Fe or Mn. As shown in Figure 8f, PSCF results for this source
619 indicate a probable source region including areas to the east and north of Steubenville. This
620 region is consistent with the locations of a number of metal smelting and processing plants that
621 reported air emissions of Pb, Cu, and Zn to the TRI, as well as with the approximate locations of
622 several sources, including a zinc smelter, glass plant, waste incinerator, and resins compounding
623 plant, that reported Cd emissions to the TRI in 2000 through 2002.⁷² In their source
624 apportionment of $\text{PM}_{2.5}$ in Pittsburgh, Pekney et al.⁷⁴ separately resolved a Cd source, a Pb
625 source, and a source characterized by As, Cu, and V (as well as Ga, which was not determined as
626 part of SCAMP, and Ni, which was assigned a large uncertainty in our PMF model). The PSCF
627 results reported for these three sources affecting Pittsburgh agree reasonably well with those
628 presented here for Source 6. Hence, Source 6 likely represents the combined impact of various
629 nonferrous metal smelting and processing plants and other industrial point sources located to the
630 east and north of Steubenville on ambient $\text{PM}_{2.5}$ in that city.

631 Source 7 includes $\sim 90\%$ of the total apportioned mass of Se, which is commonly used as
632 a tracer for primary $\text{PM}_{2.5}$ emitted by coal-fired power plants,⁶⁴ and also correlates well (Table 5)
633 with SO_2 and NO_2 , which are produced in large quantities by these plants. In addition, the
634 profile for Source 7 is characterized by the largest mass fractions of Ba, Co, and Ni of the seven
635 sources resolved from the SCAMP data. All of these elements, as well as Al, As, Ca, K, Fe, Mn,
636 V, and Zn, which also appear in the Source 7 profile (Figure 5f), are expected in PM emissions
637 from coal-fired plants.^{72,75} The largest contributor to the mass associated with Source 7 was
638 SO_4^{2-} , likely reflecting some primary sulfate emissions as well as some secondary SO_4^{2-} formed
639 by the atmospheric conversion of SO_2 during transport from the stack to the receptor site. As
640 expected for coal-fired power plant emissions,^{68,75} carbon species (i.e., EC + OM) accounted for

641 very little (~8%) of the mass associated with Source 7. PSCF results for Source 7 (Figure 8g)
642 show a probable source area comprising the Ohio River valley to the south of Steubenville, as
643 well as portions of the Ohio River valley to the north of Steubenville and the Monongahela River
644 valley to the south of Pittsburgh, where many of the region's major coal-fired power plants are
645 located. In contrast to the source apportionment modeling conducted by Koutrakis and
646 Spengler²¹ using PM_{2.5} speciation data collected in Steubenville in 1984, which attributed 8.6
647 µg/m³ (35%) of Steubenville's total PM_{2.5} mass to local coal/oil combustion emissions, our
648 source apportionment results suggest that the contribution of primary coal-fired power plant
649 emissions to ambient PM_{2.5} in Steubenville is small (i.e., ~ 0.6 µg/m³ or 3% of the total mass) in
650 spite of the continued prevalence of these plants in the surrounding area. This small contribution
651 reflects the widespread use of particulate control equipment at the plants,⁴³ and suggests that
652 whereas reductions in secondary sulfates originating from coal-fired plants (e.g., Source 1) will
653 almost certainly be required if Steubenville is to attain compliance with the PM_{2.5} NAAQS,
654 further reductions in primary particulate emissions from these plants would be of little aid in
655 attaining a mass-based standard.

656

657 **CONCLUSIONS**

658 The source apportionment results presented here suggest that ~70% of the total mass of
659 PM_{2.5} in Steubenville during SCAMP resulted from very distinct regional, secondary-source and
660 local, primary-source mechanisms that produced particles with dissimilar chemical
661 compositions. Secondary (NH₄)₂SO₄-dominated aerosol, which likely originated from SO₂
662 emitted by coal-fired power plants located to the west and southwest of Steubenville, contributed
663 ~42% of the apportioned PM_{2.5} mass. Correlation analyses presented in Tables 5 and 6 suggest
664 that elevated concentrations of PM_{2.5} from this source (Source 1), which contained very low
665 amounts of trace metals, typically were observed on a regional scale in the Steubenville area and
666 tended to occur during warm summertime afternoons, when photochemical activity would have
667 been maximal. Conversely, major local sources in the immediate Steubenville vicinity likely
668 accounted for ~30% of the total apportioned mass. These sources include motor vehicles and
669 other local sources of carbonaceous aerosols (Source 2), which contributed ~20% of the total
670 PM_{2.5} mass and more than half of the mass of fine particulate EC, OM, and Ba observed at the
671 Steubenville site, as well as iron and steel production facilities (Source 3), which contributed

672 ~10% of the total PM_{2.5} mass and more than half of the mass of fine particulate Fe, Mn, and Zn
673 observed there. Based on correlation analyses, elevated concentrations of PM_{2.5} from these
674 sources were frequently accompanied by elevated concentrations of CO and NO_x, and often
675 occurred overnight when mixing heights tend to be minimal. The effect of the iron and steel
676 production source in particular appeared to be very localized in Steubenville; median ambient
677 concentrations of Fe and Mn observed in the water-soluble PM_{2.5} fraction at Steubenville were at
678 least 1.9 times as great as concentrations observed at satellite sites located 19-22 km away. As
679 shown in Figure 1, the overall mean PM_{2.5} concentration measured at these satellite sites was
680 ~24% less than that measured at Steubenville, which is reasonably consistent with the 30% local
681 contribution estimated by PMF for Sources 2 and 3 (assuming that the satellite sites are impacted
682 by local sources from Steubenville or by their own local sources to some extent). In addition to
683 these major regional and local contributors to PM_{2.5} in Steubenville, PMF resolved a NH₄NO₃
684 source, a crustal source, a mixed nonferrous metals and industrial source, and a primary coal
685 combustion source, which contributed an estimated 15%, 6%, 3%, and 3%, respectively, to the
686 total PM_{2.5} mass observed during SCAMP.

687 Hence, these results suggest that the diverse mechanisms associated with regional and
688 local source contributions to PM_{2.5} in Steubenville must be carefully considered when
689 developing a SIP to address the city's nonattainment status under the PM_{2.5} NAAQS. Moreover,
690 the results provide an impetus for further health effects research. As discussed in the
691 introduction, a number of recent toxicology studies have discovered associations between soluble
692 PM transition metals such as Fe, V, and Zn, and pulmonary health endpoints. Trace metal
693 concentrations determined during SCAMP suggest that although ambient levels of these metals
694 have decreased appreciably since the Six Cities Study examined PM_{2.5} in Steubenville between
695 1979 and 1988, concentrations of fine particulate Fe, Mn, and Zn at Steubenville remain
696 substantially higher than concentrations of these elements recently observed in other U.S. cities.
697 The mean fine particulate Mn concentration at Steubenville during SCAMP was about 37% as
698 great as the U.S. ATSDR's chronic inhalation MRL for total airborne Mn; however, no chronic
699 inhalation RfCs, MRLs, or RELs have been established for Fe or Zn, which were implicated in
700 many of the aforementioned toxicology studies. Hence, further research is warranted to
701 elucidate whether these elements play any role in the health effects that previously¹⁹ have been
702 attributed to PM_{2.5} in Steubenville. Whereas fine particulate Zn in Steubenville is moderately

703 soluble and appears to be emitted in soluble form by local sources, fine particulate Fe in the city
704 is largely insoluble. These results are consistent with PM₁₀ trace metal solubility observations
705 made in Redcar, UK, before, during, and after the temporary closure of a steel-mill.⁷⁶ However,
706 results from SCAMP are also consistent with the hypothesis that secondary sulfates may play a
707 role in mobilizing soluble fine particulate Fe under suitable photochemical conditions,
708 supporting Grahame and Schlesinger's⁵³ recent call for additional research to examine this
709 possibility, as well as additional research to better elucidate the importance of solubility for
710 determining the biological activity of transition metals such as Fe. Ultimately, such research is
711 required to ascertain whether reducing ambient concentrations of particular components of PM_{2.5}
712 might produce a greater public health benefit than reducing total PM_{2.5} mass.

713

714 **ACKNOWLEDGMENTS**

715 The authors thank the groups who participated in the collection of data presented here, including
716 Optimal Technologies, Control Analytics, Inc., Franciscan University of Steubenville, Wheeling
717 Jesuit University, and Saint Vincent College. The outdoor ambient air monitoring portion of
718 SCAMP was funded by the United States Department of Energy's National Energy Technology
719 Laboratory. Additional funders of SCAMP include the Ohio Coal Development Office within
720 the Ohio Air Quality Development Authority, Electric Power Research Institute, American
721 Petroleum Institute, National Mining Association, American Iron and Steel Institute, Edison
722 Electric Institute, National Institute of Environmental Health Sciences, U. S. Environmental
723 Protection Agency, and CONSOL Energy Inc.

724 This paper was prepared with the support of the U. S. Department of Energy, under
725 Award No. DE-FC26-00NT40771. However, any opinions, findings, conclusions, or
726 recommendations expressed herein are those of the authors and do not necessarily reflect the
727 views of the DOE.

728

729 **REFERENCES**

- 730 1. Lewis, C.W.; Norris, G.A.; Conner, T.L.; Henry, R.C. Source Apportionment of Phoenix
731 PM_{2.5} Aerosol with the Unmix Receptor Model; *J. Air & Waste Manage. Assoc.* **2003**, *53*,
732 325-338.

- 733 2. Kim, E.; Hopke, P.K.; Edgerton, E.S. Source Identification of Atlanta Aerosol by Positive
734 Matrix Factorization; *J. Air & Waste Manage. Assoc.* **2003**, *53*, 731-739.
- 735 3. Maykut, N.N.; Lewtas, J.; Kim, E.; Larson, T.V. Source Apportionment of PM_{2.5} at an Urban
736 IMPROVE Site in Seattle, Washington; *Environ. Sci. Technol.* **2003**, *37*, 5135-5142.
- 737 4. Kim, E.; Hopke, P.K.; Pinto, J.P.; Wilson, W.E. Spatial Variability of Fine Particle Mass,
738 Components, and Source Contributions during the Regional Air Pollution Study in St. Louis;
739 *Environ. Sci. Technol.* **2005**, *39*, 4172-4179.
- 740 5. Kim, E.; Hopke, P.K.; Kenski, D.M.; Koerber, M. Sources of Fine Particles in a Rural
741 Midwestern U.S. Area; *Environ. Sci. Technol.* **2005**, *39*, 4953-4960.
- 742 6. Poirot, R.L.; Wishinski, P.R.; Hopke, P.K.; Polissar, A.V. Comparative Application of
743 Multiple Receptor Methods to Identify Aerosol Sources in Northern Vermont; *Environ. Sci.*
744 *Technol.* **2001**, *35*, 4622-4636.
- 745 7. U.S. Environmental Protection Agency. National Ambient Air Quality Standards For
746 Particulate Matter: Final Rule; *Fed. Regist.* **1997**, *62*, 38651-38760.
- 747 8. Pritchard, R.J.; Ghio, A.J.; Lehmann, J.R.; Winsett, D.W.; Tepper, J.S.; Park, P.; Gilmour,
748 M.I.; Dreher, K.L.; Costa, D.L. Oxidant Generation and Lung Injury after Particulate Air
749 Pollutant Exposure Increase with the Concentrations of Associated Metals; *Inhal. Toxicol.*
750 **1996**, *8*, 457-477.
- 751 9. Frampton, M.W.; Ghio, A.J.; Samet, J.M.; Carson, J.L.; Carter, J.D.; Devlin, R.B. Effects of
752 Aqueous Extracts of PM₁₀ Filters from the Utah Valley on Human Airway Epithelial Cells;
753 *Am. J. Physiol.* **1999**, *277*, L960-L967.
- 754 10. Molinelli, A.R.; Madden, M.C.; McGee, J.K.; Stonehuerner, J.G.; Ghio, A.J. Effect of Metal
755 Removal on the Toxicity of Airborne Particulate Matter from the Utah Valley; *Inhal.*
756 *Toxicol.* **2002**, *14*, 1069-1086.
- 757 11. Gavett, S.H.; Haykal-Coates, N.; Copeland, L.B.; Heinrich, J.; Gilmour, M.I. Metal
758 Composition of Ambient PM_{2.5} Influences Severity of Allergic Airways Disease in Mice;
759 *Environ. Health Perspect.* **2003**, *111*, 1471-1477.
- 760 12. Zelikoff, J.T.; Schermerhorn, K.R.; Fang, K.; Cohen, M.D.; Schlesinger, R.B. A Role for
761 Associated Transition Metals in the Immunotoxicity of Inhaled Ambient Particulate Matter;
762 *Environ. Health Perspect.* **2002**, *110* (suppl. 5), 871-875.

- 763 13. Maciejczyk, P.; Chen, L.C. Effects of Subchronic Exposures to Concentrated Ambient
764 Particles (CAPs) in Mice: VIII. Source-Related Daily Variations in In Vitro Responses to
765 CAPs; *Inhal. Toxicol.* **2005**, *17*, 243-253.
- 766 14. Carter, J.D.; Ghio, A.J.; Samet, J.M.; Devlin, R.B. Cytokine Production by Human Airway
767 Epithelial Cells After Exposure to an Air Pollution Particle is Metal-Dependent; *Toxicol.*
768 *Appl. Pharmacol.* **1997**, *146*, 180-188.
- 769 15. Ghio, A.J.; Carter, J.D.; Dailey, L.A.; Devlin, R.B.; Samet, J.M. Respiratory Epithelial Cells
770 Demonstrate Lactoferrin Receptors that Increase After Metal Exposure; *Am. J. Physiol.* **1999**,
771 *276*, L933-L940.
- 772 16. Huang, Y.C.; Ghio, A.J.; Stoneheurner, J.; McGee, J.; Carter, J.D.; Grambow, S.C.; Devlin,
773 R.B. The Role of Soluble Components in Ambient Fine Particles – Induced Changes in
774 Human Lungs and Blood; *Inhal. Toxicol.* **2003**, *15*, 327-342.
- 775 17. Fernandez, A.; Davis, S.B.; Wendt, J.O.L.; Cenni, R.; Young, R.S.; Witten, M.L. Particulate
776 Emission from Biomass Combustion; *Nature* **2001**, *409*, 998.
- 777 18. Adamson, I.Y.; Prieditis, H.; Hedgecock, C.; Vincent, R. Zinc is the Toxic Factor in the Lung
778 Response to an Atmospheric Particulate Sample; *Toxicol. Appl. Pharmacol.* **2000**, *166*, 111-
779 119.
- 780 19. Dockery, D.W.; Pope, C.A.; Xu, X.; Spengler, J.D.; Ware, J.H.; Fay, M.E.; Ferris, B.G.;
781 Speizer, F.E. An Association Between Air Pollution and Mortality in Six U.S. Cities; *N.*
782 *Engl. J. Med.* **1993**, *329*, 1753-1759.
- 783 20. Spengler, J.D.; Thurston, G.D. Mass and Elemental Composition of Fine and Coarse
784 Particles in Six U.S. Cities; *J. Air Pollution Control Assoc.* **1983**, *33*, 1162-1171.
- 785 21. Koutrakis, P.; Spengler, J.D. Source Apportionment of Ambient Particles in Steubenville,
786 OH Using Specific Rotation Factor Analysis; *Atmos. Environ.* **1987**, *21*, 1511-1519.
- 787 22. Laden, F.; Neas, L.M.; Dockery, D.W.; Schwartz, J. Association of Fine Particulate Matter
788 from Different Sources with Daily Mortality in Six U.S. Cities; *Environ. Health Perspect.*
789 **2000**, *108*, 941-947.
- 790 23. Grahame, T.; Hidy, G. Using Factor Analysis to Attribute Health Impacts to Particulate
791 Pollution Sources; *Inhal. Toxicol.* **2004**, *16(suppl. 1)*, 143-152.

- 792 24. Connell, D.P.; Withum, J.A.; Winter, S.E.; Statnick, R.M.; Bilonick, R.A. The Steubenville
793 Comprehensive Air Monitoring Program (SCAMP): Overview and Statistical
794 Considerations; *J. Air & Waste Manage. Assoc.* **2005**, *55*, 467-480.
- 795 25. U.S. Environmental Protection Agency. *Technical Support Document for the Final Clean Air*
796 *Interstate Rule – Air Quality Modeling*. Office of Air Quality Planning and Standards:
797 Research Triangle Park, NC, 2005. Available: [http://www.epa.gov/interstateairquality/pdfs/](http://www.epa.gov/interstateairquality/pdfs/finaltech02.pdf)
798 [finaltech02.pdf](http://www.epa.gov/interstateairquality/pdfs/finaltech02.pdf) (accessed 11/18/05).
- 799 26. Tanner, S.D.; Baranov, V.I. Theory, Design, and Operation of a Dynamic Reaction Cell for
800 ICP-MS; *Atomic Spectroscopy*, **1999**, *20*(2), 45-52.
- 801 27. Graney, J.R.; Landis, M.S.; Norris, G.A. Concentrations and Solubility of Metals from
802 Indoor and Personal Exposure PM_{2.5} Samples; *Atmos. Environ.* **2003**, *38*, 237-247.
- 803 28. Pekney, N.J.; Davidson, C.I. Determination of Trace Elements in Ambient Aerosol Samples;
804 *Anal. Chim. Acta* **2005**, *540*, 269-277.
- 805 29. Connell, D.P.; Withum, J.A.; Winter, S.E.; Statnick, R.M.; Bilonick, R.A. The Steubenville
806 Comprehensive Air Monitoring Program (SCAMP): Associations among Fine Particulate
807 Matter, Co-Pollutants, and Meteorological Conditions; *J. Air & Waste Manage. Assoc.* **2005**,
808 *55*, 481-496.
- 809 30. Connell, D.P.; Withum, J.A.; Winter, S.E.; Statnick, R.M.; Bilonick, R.A. The Steubenville
810 Comprehensive Air Monitoring Program (SCAMP): Analysis of Short-Term and Episodic
811 Variations in PM_{2.5} Concentrations Using Hourly Air Monitoring Data; *J. Air & Waste*
812 *Manage. Assoc.* **2005**, *55*, 559-573.
- 813 31. Kingston, H.M.; Jassie, L.B.; In *Introduction to Microwave Sample Preparation: Theory and*
814 *Practice*; Kingston, H.M., Jassie, L.B., Eds.; American Chemical Society: Washington, DC,
815 1988; pp. 93-154.
- 816 32. May, T.W.; Wiedmeyer, R.H. A Table of Polyatomic Interferences in ICP-MS; *Atomic*
817 *Spectroscopy* **1998**, *19*(5), 150-155.
- 818 33. Conrad, V.B.; Winter, S.E.; Krofcheck, D.S. ICP-MS and X-Ray Fluorescence: A
819 Comparative Study for the Elemental Analysis of Fine Particulate Matter; Presented at the
820 *International Symposium on Measurement of Toxic and Related Air Pollutants*, Research
821 Triangle Park, NC, September 2000.

- 822 34. Connell, D.P.; Winter, S.E.; Conrad, V.B. The Elemental Composition of PM_{2.5} Collected
823 During the Steubenville Comprehensive Air Monitoring Program (SCAMP); In *Proceedings*
824 *of A&WMA's 98th Annual Conference & Exhibition, Minneapolis, MN, June 21-24, 2005*;
825 A&WMA: Pittsburgh, PA, 2005; Paper 582.
- 826 35. Connell, D.P.; Winter, S.E.; Conrad, V.B. Characterization of PM_{2.5} Trace Elements in
827 Steubenville, Ohio, Using Dynamic Reaction Cell ICP-MS; In *Proceedings of Air Quality V*,
828 *Arlington, VA, September 19-21, 2005*.
- 829 36. Gilbert, R.O. *Statistical Methods for Environmental Pollution Monitoring*; Van Nostrand
830 Reinhold: New York, 1987; pp. 177-178.
- 831 37. Paatero, P. Least Square Formulation of Robust Nonnegative Factor Analysis; *Chemomet.*
832 *Intell. Lab. Systems* **1997**, *37*, 23-35.
- 833 38. Paatero, P. *User's Guide for Positive Matrix Factorization Programs PMF2 and PMF3, Part*
834 *2: Reference*; December 12, 2002.
- 835 39. Draxler, R.R.; Rolph, G.D. HYSPLIT (Hybrid Single-Particle Lagrangian Integrated
836 Trajectory) Model. NOAA Air Resources Laboratory: Silver Spring, MD, 2003. Available:
837 <http://www.arl.noaa.gov/ready/hysplit4.html> (accessed 12/27/05).
- 838 40. NOAA ARL – Gridded Meteorological Data Archives; [http://www.arl.noaa.gov/ss/transport/](http://www.arl.noaa.gov/ss/transport/archives.html)
839 [archives.html](http://www.arl.noaa.gov/ss/transport/archives.html) (accessed 12/27/05).
- 840 41. Lipfert, F.W. Trends in Airborne Particulate Matter in the United States; *Appl. Occup.*
841 *Environ. Hyg.* **1998**, *13*, 370-384.
- 842 42. Yan, R.; Gauthier, D.; Flamant, G.; Peraudeau, G.; Lu, J.; Zheng, C. Fate of Selenium in
843 Coal Combustion: Volatilization and Speciation in the Flue Gas; *Environ. Sci. Technol.* **2001**,
844 *35*, 1406-1410.
- 845 43. Brown, T.D.; Smith, D.N.; Hargis, R.A.; O'Dowd, W.J. Mercury Measurement and Its
846 Control: What We Know, Have Learned, and Need to Further Investigate; *J. Air & Waste*
847 *Manage. Assoc.* **1999**, *49*, 628-640.
- 848 44. Grahame, T.J. What Types of PM_{2.5} (Or Associated Pollutants) May Be More (Or Less)
849 Likely to Cause Premature Mortality? In *Proceedings of the Valuing Externalities Workshop*,
850 *McLean, VA, February 20-21, 2003*. Available: [http://www.netl.doe.gov/publications/](http://www.netl.doe.gov/publications/proceedings/03/valuing-ext/v-ext03.html)
851 [proceedings/03/valuing-ext/v-ext03.html](http://www.netl.doe.gov/publications/proceedings/03/valuing-ext/v-ext03.html) (accessed 12/27/05).

- 852 45. Begum, B.A.; Hopke, P.K.; Zhao, W. Source Identification of Fine Particles in Washington,
853 DC, by Expanded Factor Analysis Modeling; *Environ. Sci. Technol.* **2005**, *39*, 1129-1137.
- 854 46. Marmur, A.; Unal, A.; Mulholland, J.A.; Russell, A.G. Optimization-Based Source
855 Apportionment of PM_{2.5} Incorporating Gas-to-Particle Ratios; *Environ. Sci. Technol.* **2005**,
856 *39*, 3245-3254.
- 857 47. U.S. Environmental Protection Agency, Technology Transfer Network, Air Toxics Website,
858 Sources of Chronic Dose-Response Information. [http://www.epa.gov/ttn/atw/toxsource/
859 chronicsources.html](http://www.epa.gov/ttn/atw/toxsource/chronicsources.html) (accessed 12/1/05).
- 860 48. Agency for Toxic Substances and Disease Registry, Minimal Risk Levels (MRLs) for
861 Hazardous Substances. <http://www.atsdr.cdc.gov/mrls.html> (accessed 12/1/05).
- 862 49. California Office of Environmental Health Hazard Assessment, Air – Chronic RELs.
863 http://www.oehha.ca.gov/air/chronic_rels/AllChrels.html (accessed 12/1/05).
- 864 50. Mason, B. *Principles of Geochemistry*; John Wiley & Sons: New York, 1966; pp 45-46.
- 865 51. Skoog, D.A.; West, D.M.; Holler, F.J. *Fundamentals of Analytical Chemistry*, 7th ed.;
866 Harcourt: Fort Worth, 1996; p 750.
- 867 52. Lay, J.C.; Bennett, W.D.; Ghio, A.J.; Bromberg, P.A.; Costa, D.L.; Kim, C.S.; Koren, H.S.;
868 Devlin, R.B. Cellular and Biochemical Response of the Human Lung After Intrapulmonary
869 Instillation of Ferric Oxide Particles; *Am. J. Respir. Cell Mol. Biol.* **1999**, *20*, 631-642.
- 870 53. Grahame, T.J.; Schlesinger, R.B. Evaluating the Health Risk from Secondary Sulfates in
871 Eastern North American Regional Ambient Air Particulate Matter; *Inhal. Toxicol.* **2005**, *17*,
872 15-27.
- 873 54. Faust, B.C.; Hoigne, J. Photolysis of Fe(III)-Hydroxy Complexes as Sources of OH Radicals
874 in Clouds, Fog and Rain; *Atmos. Environ.* **1990**, *24A*, 79-89.
- 875 55. Zhu, X.; Prospero, J.M.; Savoie, D.L.; Millero, F.J.; Zika, R.G.; Saltzman, E.S.
876 Photoreduction of Iron(III) in Marine Mineral Aerosols; *J. Geophys. Res. [Atmos.]* **1993**, *98*,
877 9039-9046.
- 878 56. Majestic, B.J.; Schauer, J.J.; Shafer, M.M.; Turner, J.R.; Fine, P.M.; Singh, M.; Sioutas, C.
879 Development of a Wet-Chemical Method for the Speciation of Iron in Atmospheric Aerosols;
880 *Environ. Sci. Technol.* **2006**, *40*, 2346-2351.
- 881 57. Ghio, A.J.; Stoneheurner, J.; McGee, J.K.; Kinsey, J.S. Sulfate Content Correlates with Iron
882 Concentrations in Ambient Air Pollution Particles; *Inhal. Toxicol.* **1999**, *11*, 293-307.

- 883 58. Lay, J.C.; Zeman, K.L.; Ghio, A.J.; Bennett, W.D. Effects of Inhaled Iron Oxide Particles on
884 Alveolar Epithelial Permeability in Normal Subjects; *Inhal. Toxicol.* **2001**, *13*, 1065-1078.
- 885 59. Sackner, M.A.; Dougherty, R.L.; Chapman, G.A.; Ciple, J.; Perez, D.; Kwoka, M.;
886 Reinhart, M.; Brito, M.; Schreck, R. Effects of Brief and Intermediate Exposures to Sulfate
887 Submicron Aerosols and Sulfate Injections and Cardiopulmonary Function of Dogs and
888 Tracheal Mucous Velocity of Sheep; *J. Toxicol. Environ. Health*; **1981**, *7*, 951-972.
- 889 60. Ghio, A.J.; Stonehuerner, J.; Dailey, L.A.; Carter, J.D. Metals Associated with Both the
890 Water-Soluble and Insoluble Fractions of an Ambient Air Pollution Particle Catalyze an
891 Oxidative Stress; *Inhal. Toxicol.* **1999**, *11*, 37-49.
- 892 61. Fernandez, A.; Wendt, J.O.L.; Witten, M.L. Health Effects Engineering of Coal and Biomass
893 Combustion Particulates: Influence of Zinc, Sulfur and Process Changes on Potential Lung
894 Injury from Inhaled Ash; *Fuel* **2005**, *85*, 1320-1327.
- 895 62. Dusseldorp, A.; Kruize, H.; Brunekreef, B.; Hofschreuder, P.; de Meer, G.; van Oudvorst,
896 A.B. Associations of PM₁₀ and Airborne Iron with Respiratory Health of Adults Living Near
897 a Steel Factory; *Am. J. Respir. Crit. Care Med.* **1995**, *152*, 1932-1939.
- 898 63. Henry, R.C.; Norris, G.A. *EPA Unmix 2.3 User Guide*; March 2002.
- 899 64. Suarez, A.E.; Ondov, J.M. Ambient Aerosol Concentrations of Elements Resolved by Size
900 and Source: Contributions of Some Cytokine-Active Metals from Coal- and Oil-Fired Power
901 Plants; *Energy & Fuels* **2002**, *16*, 562-568.
- 902 65. Chellam, S.; Kulkarni, P.; Fraser, M.P. Emissions of Organic Compounds and Trace Metals
903 in Fine Particulate Matter from Motor Vehicles: A Tunnel Study in Houston, Texas; *J. Air &*
904 *Waste Manage. Assoc.* **2005**, *55*, 60-72.
- 905 66. Lough, G.C.; Schauer, J.J.; Park, J.-S.; Shafer, M.M.; Deminter, J.T.; Weinstein, J.P.
906 Emissions of Metals Associated with Motor Vehicle Roadways; *Environ. Sci. Technol.* **2005**,
907 *39*, 826-836.
- 908 67. Coutant, B.W.; Holloman, C.H.; Swinton, K.E.; Hafner, H.R. *Eight-Site Source*
909 *Apportionment of PM_{2.5} Speciation Trends Data*; Final Report to the U.S. EPA Office of Air
910 Quality Planning and Standards, Emissions, Monitoring, and Analysis Division; Contract No.
911 68-D-02-061, September 2003.

- 912 68. Cabada, J.C.; Pandis, S.N.; Robinson, A.L. Sources of Atmospheric Carbonaceous
913 Particulate Matter in Pittsburgh, Pennsylvania; *J. Air & Waste Manage. Assoc.* **2002**, *52*,
914 732-741.
- 915 69. Weitkamp, E.A.; Lipsky, E.M.; Pancras, P.J.; Ondov, J.M.; Polidori, A.; Turpin, B.J.;
916 Robinson, A.L. Fine Particle Emission Profile for a Large Coke Production Facility Based on
917 Highly Time-Resolved Fence Line Measurements; *Atmos. Environ.* **2005**, *39*, 6719-6733.
- 918 70. Gordon, G.E. Receptor Models; *Environ. Sci. Technol.* **1988**, *22*, 1132-1142.
- 919 71. U.S. Environmental Protection Agency; *Compilation of Air Pollutant Emission Factors*,
920 *Volume 1: Stationary Point and Area Sources*, 5th ed. Available: [http://www.epa.gov/ttn/](http://www.epa.gov/ttn/chief/ap42/index.html)
921 [chief/ap42/index.html](http://www.epa.gov/ttn/chief/ap42/index.html) (accessed 12/29/05).
- 922 72. U.S. Environmental Protection Agency, Toxics Release Inventory (TRI) Program, Get TRI
923 Data. <http://www.epa.gov/tri/tridata/index.htm> (accessed 12/29/05).
- 924 73. U.S. Environmental Protection Agency, AirData: Access to Air Pollution Data.
925 <http://www.epa.gov/air/data/index.html> (accessed 12/29/05).
- 926 74. Pekney, N.J.; Davidson, C.I.; Zhou, L.; Hopke, P.K. Identifying Sources of PM_{2.5} in
927 Pittsburgh Using PMF and PSCF; Presented at the *AAAR Specialty Conference: Particulate*
928 *Matter, Supersites Program & Related Studies, Atlanta, GA, February 7-11, 2005*.
- 929 75. Lee, S.W. Source Profiles of Particulate Matter Emissions from a Pilot-Scale Boiler Burning
930 North American Coal Blends; *J. Air & Waste Manage. Assoc.* **2001**, *51*, 1568-1578.
- 931 76. Hutchinson, G.R.; Brown, D.M.; Hibbs, L.R.; Heal, M.R.; Donaldson, K.; Maynard, R.L.;
932 Monaghan, M.; Nicholl, A.; Stone, V. The Effect of Refurbishing a UK Steel Plant on PM₁₀
933 Metal Composition and Ability to Induce Inflammation; *Respir. Res.* **2005**, *6*, 43.

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Table 1. Summary of methods and method performance for the determination of elements in the water-soluble and acid-digestible PM_{2.5} fractions by DRC ICP-MS.

Element	Water-Soluble PM _{2.5} Fraction			Acid-Digestible PM _{2.5} Fraction			Method Imprecision ^c (%)
	Methods ^a	% of Obs. < LOD ^b	Recovery of NIST SRM 1643d (Mean ± SD, %)	Methods ^a	% of Obs. < LOD ^b	Recovery of NIST SRM 2783 (Mean ± SD, %)	
Al	27(NH ₃)	0	94 ± 4	27(NH ₃)	0	99 ± 34	20.8
As	75(CH ₄)	0	90 ± 9	75, 75(CH ₄), 75(NH ₃)	7	110 ± 7	21.9
Ba	138	0	102 ± 4	138	1	104 ± 11	NA
Ca	40(CH ₄), 40(NH ₃), 44	0	98 ± 24	40(CH ₄), 44	6	104 ± 6	24.2
Cd	111, 114	0	97 ± 1	111, 114	1	NA	21.5
Co	59	1	99 ± 2	59	22	115 ± 12	NA
Cu	63	0	95 ± 10	63(CH ₄), 63(NH ₃), 65(CH ₄)	2	79 ± 21	17.9
Fe	54(CH ₄), 56(CH ₄)	4	121 ± 18	54(CH ₄), 56(CH ₄), 56(NH ₃)	0	97 ± 12	19.2
K	39(CH ₄), 39(NH ₃)	5	115 ± 13	39(CH ₄), 39(NH ₃)	1	99 ± 11	15.1
Mg	24, 25	0	100 ± 9	24, 25	0	97 ± 5	22.1
Mn	55(CH ₄)	9	111 ± 7	55(CH ₄), 55(NH ₃)	1	106	22.6
Na	23, 23(NH ₃)	0	105 ± 23	-	-	-	-
Ni	58, 60	2	98 ± 3	58, 58(NH ₃), 60(CH ₄)	27	119 ± 20	NA
Pb	208	1	101 ± 5	207, 208	0	101 ± 5	21.8
Se	80(CH ₄), 82	1	92 ± 8	78(CH ₄), 80(CH ₄)	1	NA	16.8
Sn	118	5	NA	118, 120	1	NA	24.0
Ti	-	-	-	48	2	71 ± 14	21.3
V	51(CH ₄)	3	99 ± 9	51(NH ₃)	1	84 ± 10	33.6
Zn	66, 68	0	100 ± 6	64(NH ₃), 66, 68	0	108 ± 8	18.7

NOTE: NA = No data available; ^aIndicates the isotopes and reaction gases (where applicable) used to produce valid determinations of the listed element by DRC ICP-MS; ^bPercentage of observations that were less than the analytical limit of detection; ^cMethod imprecision = 100 · (RMSD/√2) / mean, where RMSD is the root mean square difference between duplicate determinations (i.e., from separate analyses of collocated filters) of a given element, and mean is the overall mean concentration of that element measured at Steubenville during SCAMP.

Table 2. Summary statistics for 24-hr average ambient air concentrations and fractional solubilities of elements in PM_{2.5} collected at Steubenville during SCAMP.

Element	Concentration (ng/m ³)							Fractional Solubility ^a		
	N	Mean	SD	Q25	Q50	Q75	Max	Q25	Q50	Q75
Al	135	97.7	56.6	62.2	80.8	123.3	411.7	0.09	0.14	0.23
As	139	1.64	1.64	0.67	1.20	1.88	10.56	1.29	1.60	1.88
Ba	124	2.3	1.5	1.1	2.0	3.2	8.5	0.69	0.82	1.00
Ca	137	136	61	93	118	167	310	0.37	0.53	0.71
Cd	139	0.46	0.58	0.19	0.30	0.51	5.89	0.86	1.04	1.23
Co	122	0.065	0.047	0.030	0.061	0.091	0.209	0.38	0.55	0.77
Cu	130	3.2	1.8	1.9	2.8	4.0	9.8	0.55	0.73	0.94
Fe	139	272	308	87	173	294	1530	0.03	0.06	0.11
K	134	91	55	55	73	114	261	0.76	1.00	1.32
Mg	138	47	48	20	32	57	389	0.47	0.56	0.70
Mn	139	14.6	15.4	5.0	9.6	17.4	90.5	0.40	0.51	0.64
Ni	121	1.1	1.5	0.3	0.7	1.5	8.1	0.29	0.44	0.86
Pb	139	15.3	20.7	7.0	10.5	16.2	218.3	0.37	0.47	0.60
Se	132	3.30	3.21	1.49	2.40	4.20	24.69	1.14	1.38	1.69
Sn	137	1.36	1.19	0.62	0.97	1.55	5.97	0.07	0.13	0.23
Ti	121	16.9	20.6	8.9	11.6	17.7	159.0	NA	NA	NA
V	132	1.50	1.49	0.57	1.06	1.99	11.02	0.47	0.66	0.85
Zn	139	84.4	81.4	40.9	57.6	89.8	448.6	0.39	0.49	0.60

NOTE: NA = No data available; ^aComputed as the ratio of the concentration of the element in the water-soluble PM_{2.5} fraction to the concentration of the element in the acid-digestible PM_{2.5} fraction for a given day.

Table 3. Spearman correlations between 24-hr average PM_{2.5} trace element fractional solubilities and 24-hr sulfate concentrations or temperatures at Steubenville, by astronomical season.

Element	Sulfate				Temperature			
	Winter	Spring	Summer	Fall	Winter	Spring	Summer	Fall
Al	0.75	0.59	-0.01	0.68	0.64	0.63	0.17	0.28
As	0.05	0.02	-0.01	-0.11	-0.17	-0.24	0.32	-0.45
Ba	-0.26	-0.06	-0.13	-0.06	-0.37	0.21	-0.29	-0.48
Ca	0.46	-0.15	-0.01	0.34	0.63	0.14	-0.25	0.12
Cd	0.47	0.26	-0.15	0.18	0.34	0.05	0.07	-0.13
Co	0.32	0.02	0.11	-0.15	0.19	-0.13	-0.18	-0.09
Cu	0.25	-0.07	0.83	-0.18	0.07	0.29	0.82	0.43
Fe	0.17	0.33	0.66	0.07	-0.21	0.39	0.72	-0.34
K	0.24	0.19	-0.52	0.24	-0.06	0.10	-0.37	-0.41
Mg	0.13	0.43	-0.36	0.33	0.02	0.47	-0.51	-0.31
Mn	0.30	0.40	0.39	0.02	0.21	0.30	0.47	-0.13
Ni	0.08	-0.11	0.46	0.14	-0.07	-0.05	0.60	-0.53
Pb	0.27	0.42	-0.06	0.02	0.21	0.45	0.18	-0.16
Se	0.31	-0.31	-0.32	0.33	0.18	0.01	0.25	-0.04
Sn	-0.22	-0.17	0.51	-0.43	-0.34	0.27	0.79	-0.49
V	0.12	0.11	-0.05	0.12	0.24	-0.10	0.39	-0.31
Zn	0.50	0.55	0.32	0.73	0.47	0.17	-0.02	0.16

Table 4. Median 24-hr ambient air concentrations of elements in the water-soluble PM_{2.5} fraction at the five SCAMP monitoring sites, as well as significance of differences between concentrations at each Satellite site and concentrations at the Steubenville site.

Element	Steubenville Median (ng/m ³)	Northern Site		Southern Site		Eastern Site		Western Site	
		Median (ng/m ³)	P-value ^a	Median (ng/m ³)	P-value ^a	Median (ng/m ³)	P-value ^a	Median (ng/m ³)	P-value ^a
Al	13.2	6.9	0.000 (<)	9.9	0.001 (<)	6.7	0.000 (<)	7.4	0.001 (<)
As	1.95	1.52	0.000 (<)	1.73	0.009 (<)	1.51	0.000 (<)	1.29	0.000 (<)
Ba	1.6	0.9	0.000 (<)	2.8	0.000 (>)	1.5	0.116	1.1	0.000 (<)
Ca	61	42	0.000 (<)	50	0.003 (<)	43	0.000 (<)	54	0.000 (<)
Cd	0.36	0.24	0.000 (<)	0.28	0.011 (<)	0.25	0.002 (<)	0.25	0.000 (<)
Co	0.035	0.029	0.007 (<)	0.038	0.285	0.035	0.407	0.033	0.361
Cu	2.0	1.6	0.000 (<)	2.2	0.756	1.9	0.427	1.7	0.020 (<)
Fe	12.4	6.7	0.000 (<)	11.9	0.000 (<)	7.0	0.000 (<)	6.4	0.000 (<)
K	80	70	0.000 (<)	71	0.007 (<)	77	0.557	58	0.000 (<)
Mg	19	10	0.000 (<)	9	0.000 (<)	8	0.000 (<)	10	0.000 (<)
Mn	5.3	2.5	0.000 (<)	2.0	0.000 (<)	2.0	0.000 (<)	2.0	0.000 (<)
Na	69	46	0.000 (<)	52	0.013 (<)	53	0.000 (<)	47	0.000 (<)
Ni	0.5	0.5	0.077	0.5	0.564	0.5	0.131	0.5	0.319
Pb	5.3	3.8	0.000 (<)	3.8	0.000 (<)	3.4	0.000 (<)	3.5	0.000 (<)
Se	3.43	3.72	0.963	3.28	0.100	3.48	0.278	2.83	0.003 (<)
Sn	0.11	0.11	0.215	0.14	0.129	0.19	0.000 (>)	0.14	0.225
V	0.66	0.55	0.068	0.66	0.126	0.50	0.000 (<)	0.46	0.000 (<)
Zn	28.0	16.4	0.000 (<)	15.6	0.000 (<)	15.7	0.000 (<)	14.1	0.000 (<)

^aTwo-tailed p-value resulting from a Wilcoxon signed rank test applied to paired differences in concentrations measured at the satellite site and at the Steubenville site. For P-values indicating statistically significant differences at $\alpha = 0.05$, the parenthetical indicates whether concentrations measured at the satellite site tended to be greater than (>) or less than (<) those measured at Steubenville.

Table 5. Spearman correlations between 24-hr PMF-estimated source contributions and 24-hr average gaseous pollutant concentrations and meteorological conditions at Steubenville.

	Source 1	Source 2	Source 3	Source 4	Source 5	Source 6	Source 7
SO₂	0.33	0.54	0.50	-0.05	0.34	0.26	0.71
NO	0.03	0.71	0.65	0.06	0.11	0.55	0.54
NO₂	0.12	0.77	0.65	-0.08	0.45	0.56	0.70
CO	0.27	0.73	0.76	0.05	0.35	0.49	0.54
O₃	0.17	-0.14	-0.05	-0.37	0.35	-0.16	-0.30
WS	-0.21	-0.52	-0.32	0.07	-0.08	-0.47	-0.29
Temp	0.51	0.35	0.37	-0.54	0.58	0.30	0.28
RH	0.31	-0.36	-0.39	0.10	-0.32	0.03	-0.10
BP	0.07	0.40	0.47	-0.28	0.37	0.25	0.26

Note: WS = wind speed; Temp = temperature; RH = relative humidity; BP = barometric pressure.

Table 6. Spearman correlation coefficients between 24-hr PMF-estimated source contributions and measures of the intraday timing of PM_{2.5} episodes and impact of local vs. regional sources on PM_{2.5} concentrations in Steubenville.

	Nighttime Max. PM_{2.5}^a	Daytime Max. PM_{2.5}^b	Local PM_{2.5}^c	Regional PM_{2.5}^d
Source 1	0.35	0.71	0.44	0.76
Source 2	0.73	0.27	0.63	0.38
Source 3	0.65	0.09	0.67	0.06
Source 4	-0.13	-0.06	0.06	0.22
Source 5	0.42	0.37	0.36	0.24
Source 6	0.36	0.08	0.50	0.15
Source 7	0.45	0.38	0.44	0.45

^aMaximum 1-hr concentration measured between 2:00 a.m. and 9:00 a.m. EST;
^bMaximum 1-hr concentration measured between 12:00 p.m. and 7:00 p.m. EST;
^cDifference between 24-hr concentration measured at Steubenville and average of 24-hr concentrations measured at the northern and western sites; ^dAverage of 24-hr concentrations measured at the northern and western sites.

FIGURE CAPTIONS

Figure 1. Average $PM_{2.5}$ concentrations ($\mu\text{g}/\text{m}^3$) measured at the five SCAMP monitoring sites between May 2000 and May 2002.

Figure 2. Sensitivity of robust Q to changes in FPEAK for final 7-source PMF model.

Figure 3. Ratio of warm season (April – September) to cool season (October – March) median concentrations (in both the water-soluble and acid-digestible $PM_{2.5}$ fractions) and fractional solubilities measured at the Steubenville site. Solid symbols indicate statistically significant (at $\alpha = 0.05$) seasonal differences, based on the results of Wilcoxon rank sum tests.

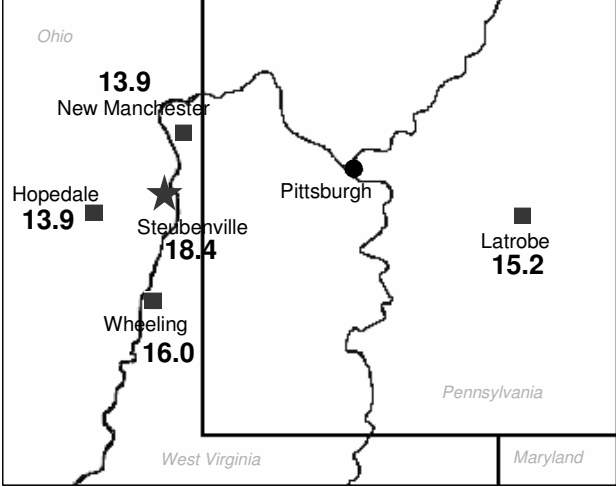
Figure 4. Spearman correlations between concentrations measured at Steubenville and concentrations measured at each satellite site for elements in the water-soluble $PM_{2.5}$ fraction.

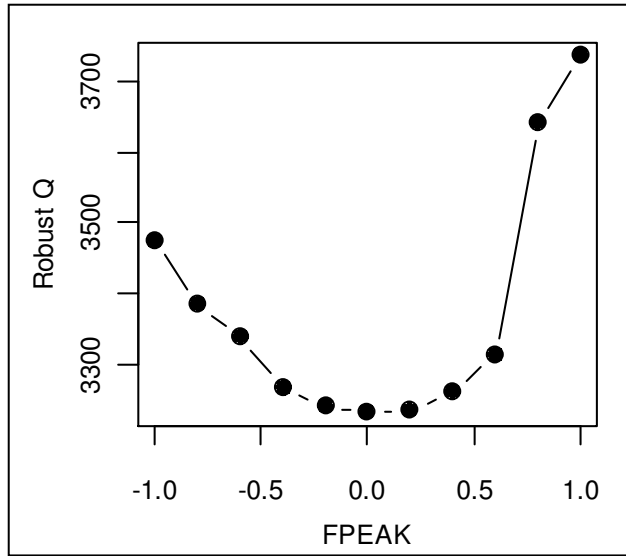
Figure 5. Source profiles (with 1σ uncertainties) estimated by applying PMF to $PM_{2.5}$ mass and composition data measured at Steubenville during SCAMP.

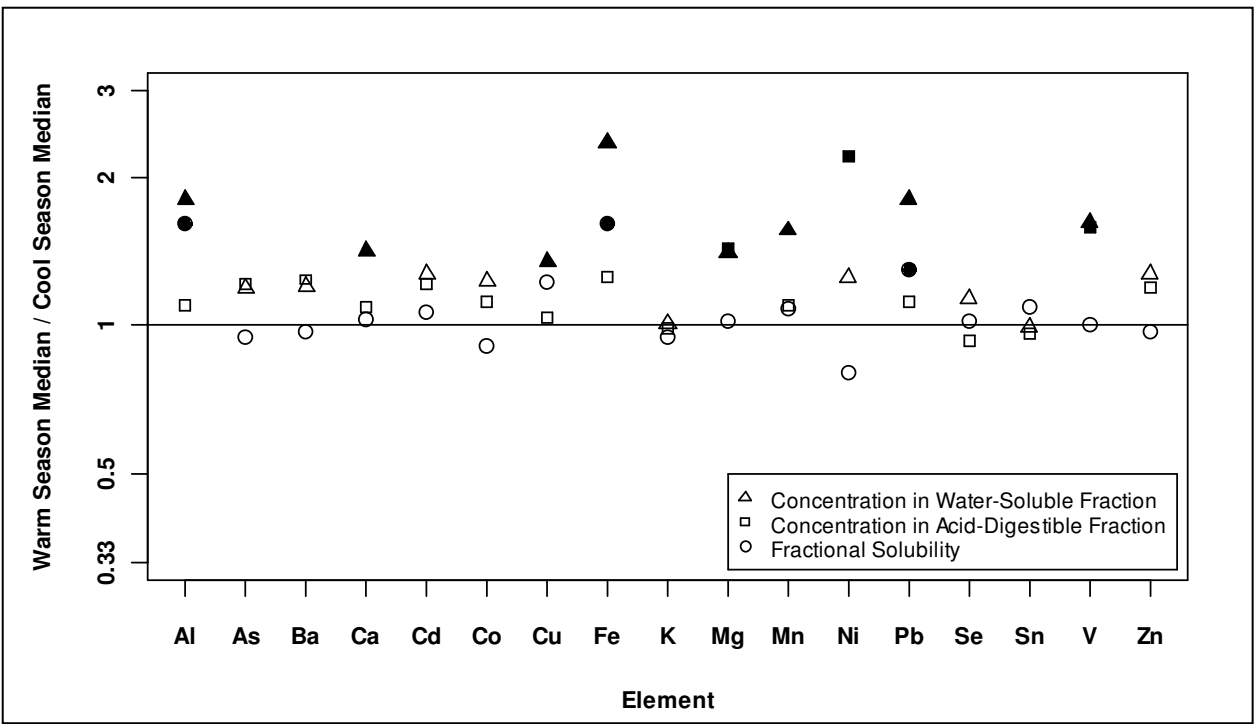
Figure 6. Time series of source contributions ($\mu\text{g } PM_{2.5} / \text{m}^3$) at Steubenville estimated by the seven-factor PMF model. The overall mean contribution from each source is also indicated.

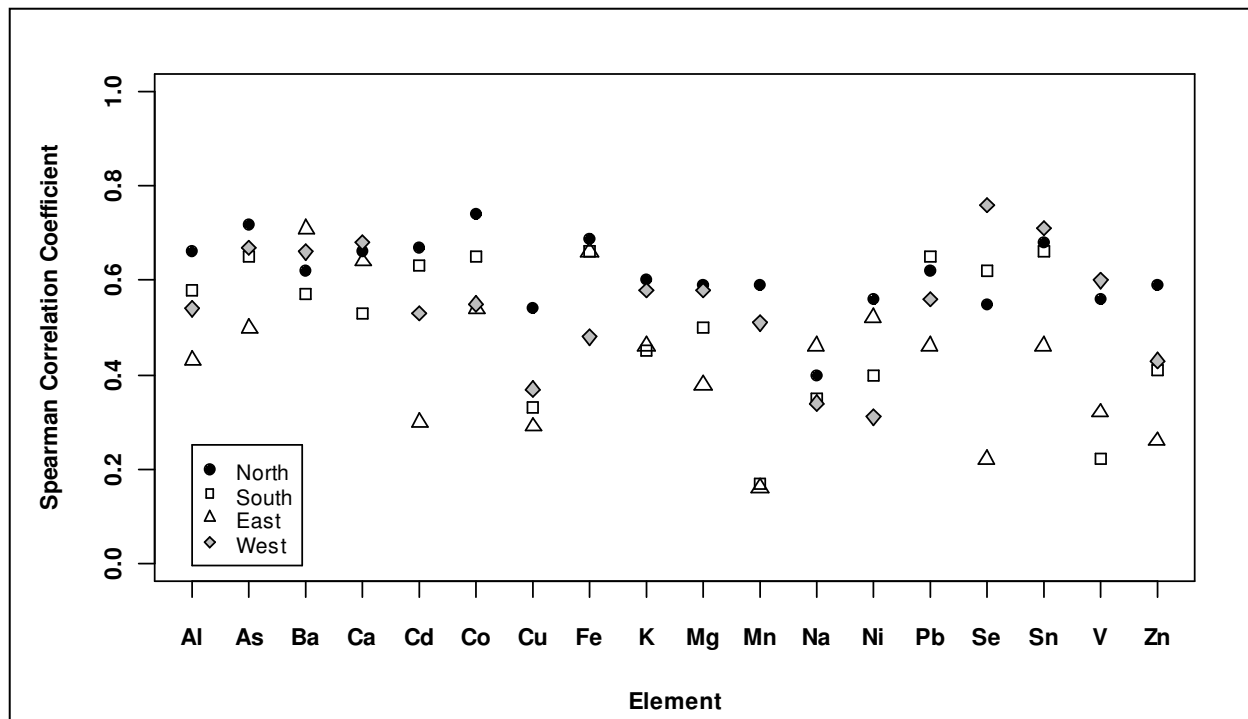
Figure 7. $PM_{2.5}$ concentrations predicted by the seven-factor PMF model vs. $PM_{2.5}$ concentrations measured at Steubenville, with parity line and linear regression line (bold).

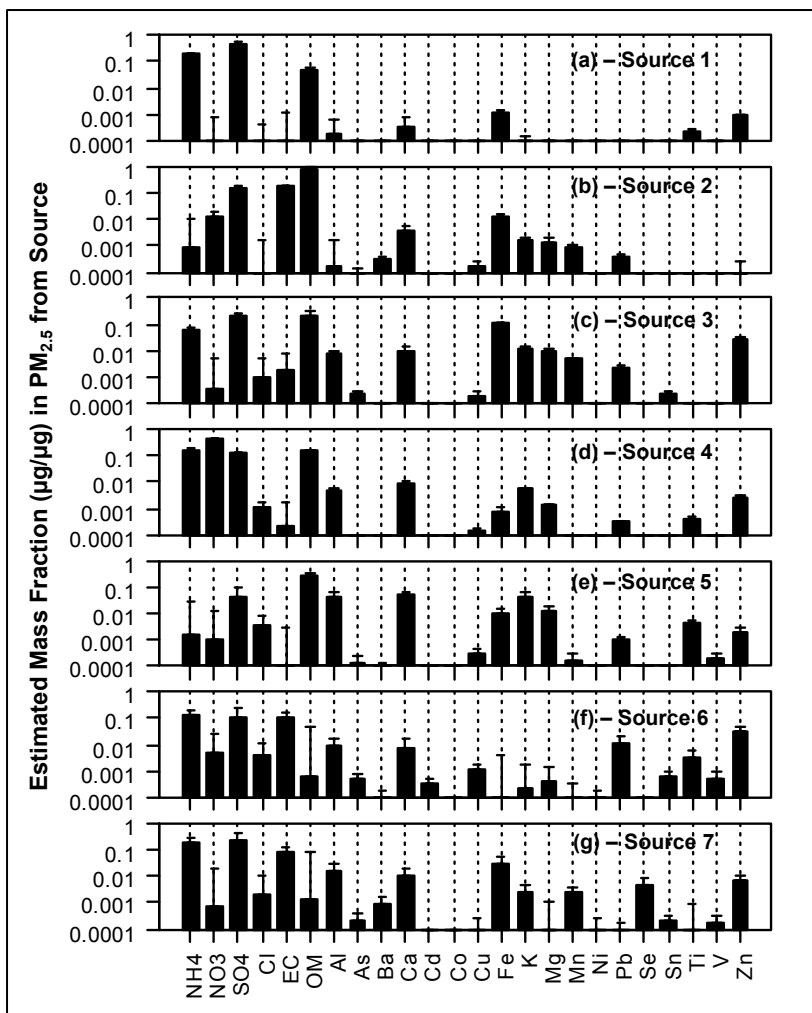
Figure 8. PSCF plots showing possible source regions for each of the seven factors resolved by PMF. Subplots (a) through (g) show results obtained for each of the factors using 24-hr backward trajectories with starting heights of 50 m; subplots (h) and (i) show results obtained for the sulfate and nitrate sources, respectively, using 72-hr backward trajectories with starting heights of 500 m.

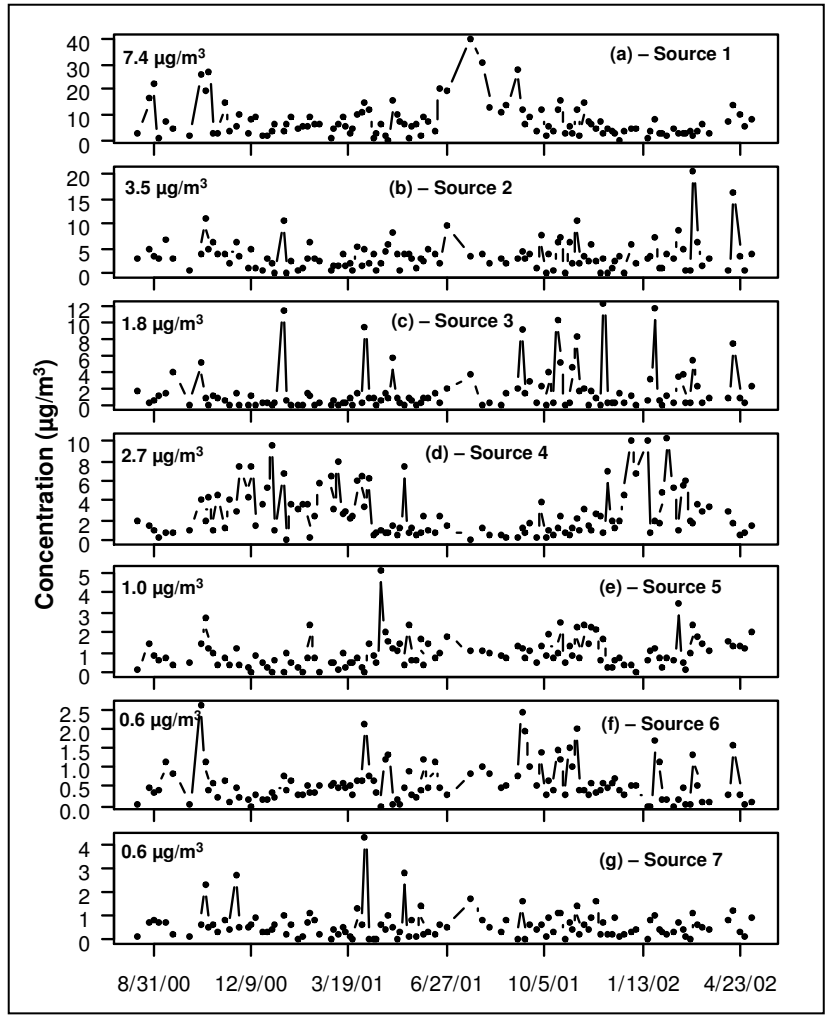


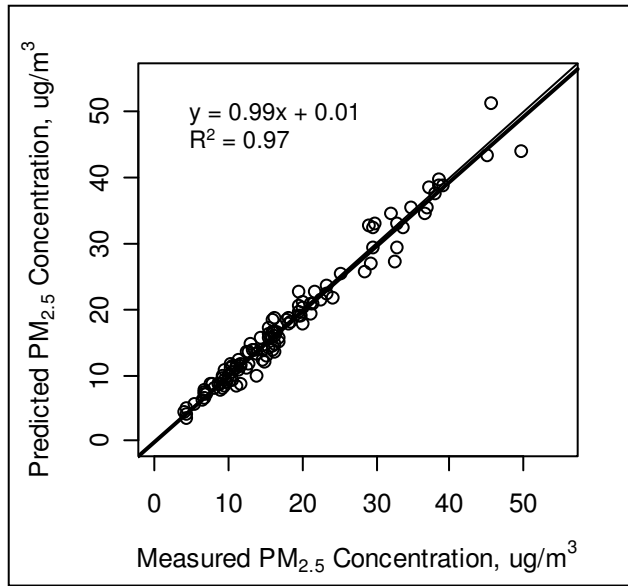


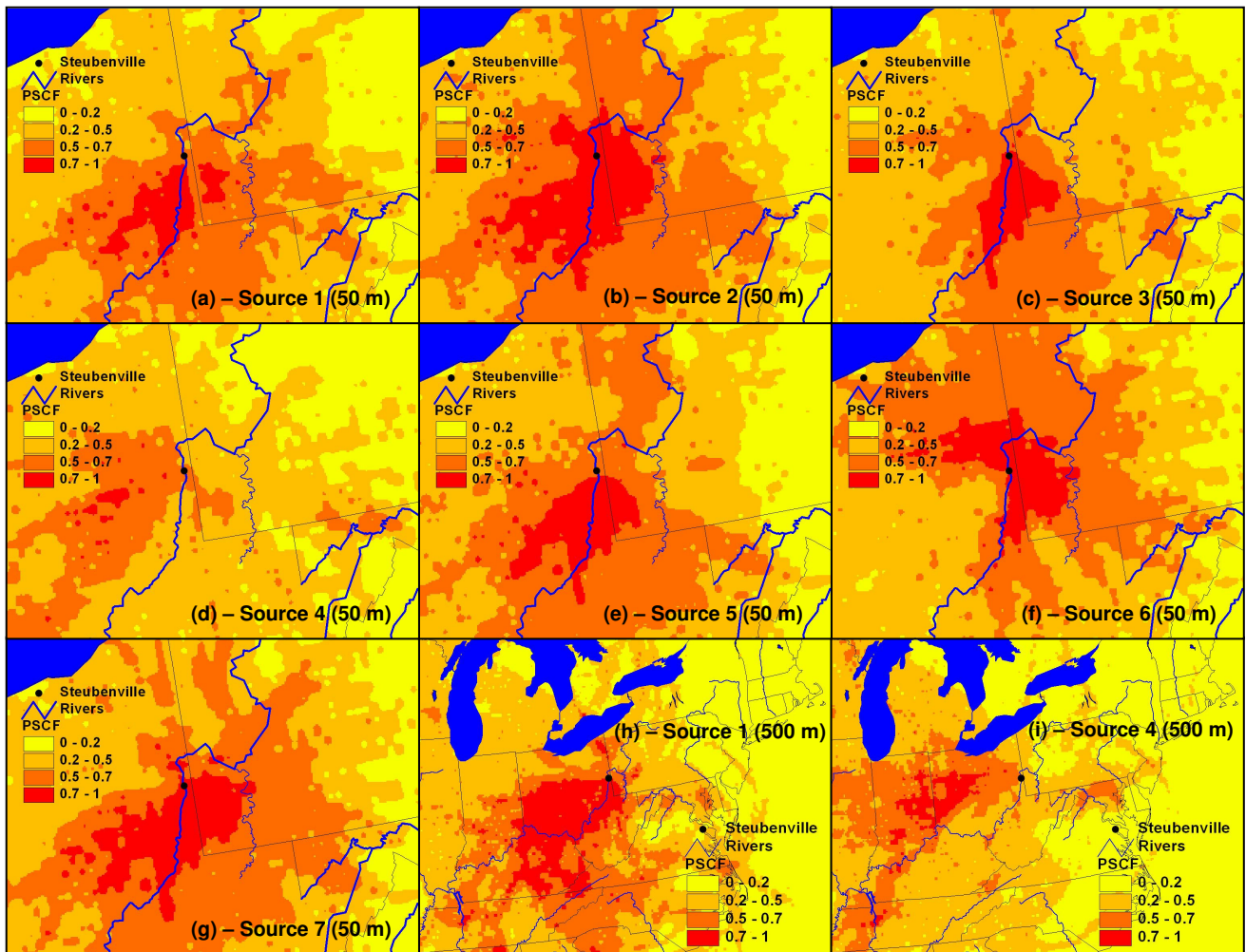












APPENDIX E

AMBIENT PARTICULATE AIR POLLUTION AND CARDIAC ARRHYTHMIA IN A PANEL OF OLDER ADULTS IN STEUBENVILLE, OHIO

*MANUSCRIPT SUBMITTED TO OCCUPATIONAL AND
ENVIRONMENTAL MEDICINE*

**Ambient particulate air pollution and cardiac arrhythmia in a panel of older adults
in Steubenville, Ohio**

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Word count: text=4312 (including tables), abstract=296

Running Title: Ambient air pollution and cardiac arrhythmia

Keywords: Air pollution, epidemiology, cardiovascular disease, arrhythmia

ABSTRACT

Objectives: Ambient particulate air pollution has been associated with increased risk of cardiovascular morbidity and mortality. Pathways by which particles may act involve autonomic nervous system dysfunction or inflammation, which can affect cardiac rate and rhythm. The importance of these pathways may vary by particle component or source. In an eastern U.S. location with significant non-traffic pollution, we examine the association of air pollution and odds of cardiac arrhythmia in older adults.

Methods: We evaluated 32 non-smoking older adults on a weekly basis for 24-weeks during the summer and fall of 2000 with a standardized 30-minute protocol that included continuous electrocardiogram measurements. A central ambient monitoring station provided daily concentrations of fine particles (PM_{2.5}, sulfate, elemental carbon) and gases. Sulfate was used as a marker of regional, non-traffic pollution. We used logistic mixed effects regression to examine the odds of having any supraventricular ectopy (SVE) or ventricular ectopy (VE) in association with increases in air pollution for moving average pollutant concentrations up to ten days prior to the health assessment.

Results: Participant-specific mean counts of arrhythmia over the protocol varied between 0.1 – 363 for SVE and 0 – 350 for VE. We observed odds ratios for having SVE over the length of the protocol of 1.42 (95% CI: 0.99, 2.04), 1.70 (95% CI: 1.12, 2.57) and 1.78 (95% CI: 0.95, 3.35) for 10.0 µg/m³, 4.2 µg/m³ and 14.9 ppb increases in 5-day moving average PM_{2.5}, sulfate and ozone concentrations, respectively. The other pollutants, including elemental carbon, showed no effect on arrhythmia. Participants reporting cardiovascular conditions (e.g., previous myocardial infarction or hypertension)

were the most susceptible to pollution-induced SVE. We found no association of pollution with VE.

Conclusion: Increased levels of ambient particulate air pollution from non-traffic sources may increase the risk of supraventricular arrhythmia in the elderly.

INTRODUCTION

Several recent studies provide evidence of the relationship between ambient particle concentrations and cardiovascular morbidity[1][2][3][4] and mortality.[5][6][7]

Autonomic nervous system dysfunction, as well as inflammation, may be pathways by which particles affect cardiac rate and rhythm.

Findings from recent panel studies suggest that particulate air pollution can affect the risk of ectopy,[8][9][10][11] defined as extra cardiac depolarizations within either the atria (i.e., supraventricular ectopy [SVE]) or the ventricles (i.e., ventricular ectopy [VE]). For example, a study examining chronic obstructive pulmonary disease (COPD) patients in Vancouver found a 22% increased rate of SVE for each $5.8 \mu\text{g}/\text{m}^3$ increase in fine particles ($\text{PM}_{2.5}$).[11] Riediker and colleagues (2004) found 23.0% and 19.1% increased rates of SVE and VE, respectively, for $10 \mu\text{g}/\text{m}^3$ increases in $\text{PM}_{2.5}$ in a study of North Carolina highway patrolmen.[10] In contrast to these two studies, where the primary sources of air pollution were traffic, this paper reports the effects particle pollution on the risk of ectopy in a population of older adults in the small town of Steubenville, OH. Steubenville is located in an industrial area of the Ohio River Valley, characterized by coal-fired power plants and steel mills.

METHODS

Study design and recruitment

We recruited 32 non-smoking older adults to participate in a cardiovascular health and air pollution exposure study in Steubenville, OH. Most participants lived in one of three centrally located apartment buildings. Sampling was conducted over two 12-week

sampling sessions during the summer (June 4 – August 18) and fall (September 25 – December 15) of 2000. Of the 32 participants, 28 participated in both seasons. Prior to the study, we held screening appointments to obtain information on the health status of each participant and baseline resting electrocardiograms (ECG) (12-lead MAC6, Marquette Medical Systems Inc.). Since heart rate variability (HRV) was an endpoint of interest, exclusion criteria for this study included having a pacemaker, a recent acute coronary syndrome, atrial flutter or atrial fibrillation. The research protocol was approved by Institutional Review Boards of the Harvard School of Public Health and the Brigham and Women's Hospital in Boston, MA. Informed consent was obtained from all individuals prior to their participation in the study.

Health measurements

Participants visited one of two clinic rooms, set-up in rooms of two main apartment buildings where subjects resided, on a weekly basis following a regular schedule (Monday-Friday, 8am-4pm). During each visit, we administered a questionnaire regarding symptoms, doctor and hospital visits, changes in medication and participant medication use that morning. Holter monitoring (SEER MC, GE Medical Systems) with electrodes in a modified V5 and AVF position provided continuous ECG data throughout a 30-minute protocol, comprised of: (1) five minutes of rest in a supine position; (2) three supine blood pressure (BP) measurements (NIBP Vital Signs Monitor, Welch Allyn); (3) five minutes of standing with three standing BP measurements taken after two minutes; (4) five minutes of exercise (walking) outdoors (weather and health permitting);

(5) five minutes of rest in a supine position and; (6) two minutes and 20 seconds of paced breathing.

Research technicians downloaded the ECG recordings onto a Sparq station in the field. Cardiology technicians at the Brigham & Women's Hospital analyzed the ECG files using a MARS Marquette Workstation. For each clinic visit, we obtained data on heart rate and counts of SVE and VE, as compiled over the whole protocol and for each separate interval.

Exposure measurements

We measured 24-hour integrated fine particulate ($PM_{2.5}$, sulfate [SO_4^{2-}], elemental carbon [EC]) and gaseous (ozone [O_3], nitrogen dioxide [NO_2], sulfur dioxide [SO_2]) pollutant concentrations at a central ambient site at the Franciscan University of Steubenville beginning at 9am each day (except Saturday) using a Harvard multi-pollutant monitor (HMPM). CONSOL Energy Inc. Research & Development analyzed the SO_4^{2-} , O_3 , NO_2 and SO_2 filters using ion chromatography and EC filters using thermal optical transmission. A complete description of the HMPM and its performance in the current study has been described previously.[12][13] Due to the high number of HMPM SO_2 samples below the limit of detection,[13] these data were not used in the current analysis.

The HMPM data were supplemented with continuous measurements of $PM_{2.5}$ (R&P, TEOM 1400A [50°C]), SO_2 (API, Model 100UV Fluorescent), carbon monoxide (CO)(API, Model 300 GFC) and meteorological variables (Met-1, 10 meter station) collected by CONSOL Energy Inc. Research & Development as part of a broader

monitoring effort in Steubenville.[14][15] The $PM_{2.5}$ data were used in sensitivity analyses only, to fill in missing Saturday values in the HMPM $PM_{2.5}$ data series. The SO_2 and CO data supplemented the pollutants measured with the HMPM sampler. Using the meteorological variables, we calculated apparent temperature as: $-2.653 + (0.994 \times Ta) + (0.0153 \times Td^2)$, where Ta is the air temperature and Td is the dew point.

Data analysis

We excluded one participant who dropped out of the study after four weeks due to the limited number of samples, and another participant who had atrial fibrillation. Intervals or sessions with less than 50% valid ECG data were invalidated.

We used logistic regression to examine the odds of having at least one arrhythmia (i.e., during the 30-minute protocol or for specific protocol intervals) associated with increases in air pollution. In these analyses, we excluded participants who experienced no change in their binomial (0/1) outcome, as they would have no variation in the outcome. Excluded from primary analyses were participants who experienced a) ectopic beats during all clinic visits, or b) no ectopic beats during any clinic visit. Sensitivity analyses were conducted that included all participants regardless of their binomial outcomes.

We additionally examined the relative rate of arrhythmia (e.g. # beats/30-minute protocol) associated with increases in air pollution by fitting overdispersed Poisson regression models, otherwise known as negative binomial regression, to the arrhythmia counts. These analyses included the entire cohort.

We conducted all analyses using PROC NLMIXED (SAS Release 8.02; SAS Institute, Cary, NC), controlling for a random subject effect and the fixed effects of apparent temperature, season (summer vs. fall), day of week, hour of day, age and body mass index (BMI). We chose ‘current hour of the clinic visit’, *a priori*, as the most appropriate exposure window for apparent temperature since participants went outside during the ECG protocol. A sensitivity analysis, conducted by including a categorical “building of residence” variable, did not affect model results.

We created separate models to examine the relative effects of particles (PM_{2.5}), particle components (SO₄²⁻ and EC) and gases (O₃, NO₂, SO₂ and CO) on cardiac arrhythmia. Based on autonomic nervous system dysfunction, which may be acute, and inflammation, which may take several days to develop, as potential mechanisms by which particles could influence cardiac arrhythmia, we examined the relative effects of 1-day to 5-day moving average concentrations of each pollutant prior to the health assessment. Moving averages up to 10-days prior to the health assessment were examined in sensitivity analyses. Effect estimates are presented for interquartile range (IQR) increases in pollutant concentrations at each exposure period evaluated.

Finally, we explored the potential for heterogeneity in response across the cohort by including participant characteristics, medications use (recorded at each clinic visit) and participant-specific mean ectopy levels as potential effect modifiers of the PM_{2.5} associations. These factors were included as main effects and interaction terms with PM_{2.5} in the models.

RESULTS

Participant characteristics and arrhythmia occurrence

Participants (mean age: 71.2 years, range: 53.5 – 90.3 years) were predominantly female (n = 28) and over the age of 60 (n = 27) (Table 1). The majority of participants (n = 29) reported at least one of the listed cardiovascular or respiratory illnesses. 23 participants reported multiple diagnoses. Of the 6 participants that reported angina, for example, 4 also reported having at least one of the respiratory diseases. Similarly, of the 9 participants that reported having asthma, 6 also reported having COPD. Participants with cardiovascular conditions (e.g., previous myocardial infarction (MI), heart failure and/or hypertension) reported taking cardiovascular medications more frequently than those without the condition. For example, of the 22 subjects who reported ever having had hypertension, 80% were on beta-blockers, Ca²⁺ channel blockers and/or ACE inhibitors compared to only 5% in those without hypertension. Similarly, participants with respiratory illness (bronchitis, COPD and/or asthma) reported taking respiratory medications (bronchodilators, beta-agonists and/or steroids) more frequently (43-57%) than those without the condition (7-12%).

Over the 24-week study period, 623 clinic visits provided valid ECG data. The mean duration of the structured ECG protocol session was 33 (\pm 3) minutes and mean counts of arrhythmia for the whole cohort were 27.2 (\pm 87.8) beats/protocol session (or 50.0 \pm 162.8 beats/hr) for SVE and 28.0 (\pm 93.3) beats/protocol session (or 52.6 \pm 173.5 beats/hr) for VE. There was considerable variability in participant-specific SVE and VE counts, as participant-specific mean counts (over the whole protocol) varied between 0.1 – 363 for SVE and 0 – 350 for VE. The correlation between the two arrhythmia

measures was low, with counts of SVE and VE over the whole protocol showing a Spearman correlation coefficient (r) of 0.03.

Table 1. Participant characteristics, disease diagnoses and medications use.

			Whole Cohort		SVE Models		VE Models	
			N (%)	# Obs.	N (%)	# Obs.	N (%)	# Obs.
OVERALL			30 (100)	630	23 (100)	482	18 (100)	374
ARRHYTHMIA	SVE (in beats/hr)		50.0 ± 162.8		8.9 ± 47.1		---	
	VE (in beats/hr)		52.6 ± 173.5		---		7.4 ± 22.9	
GENERAL	Age	< 65	10 (33)	202	9 (39)	178	6 (33)	116
		65 – 74	9 (30)	196	6 (26)	138	4 (22)	92
		≥ 75	11 (37)	232	8 (35)	166	8 (44)	166
	Male		2 (7)	33	1 (4)	22	0 (0)	0
	Black	(vs. white)	9 (30)	184	7 (30)	137	6 (33)	121
	Ever Smoker	(vs. never)	14 (47)	289	9 (39)	186	8 (44)	167
DIAGNOSES*								
Cardiovascular	Angina Pectoris		6 (20)	116	5 (22)	93	4 (22)	73
	Previous MI		6 (20)	118	5 (22)	97	6 (33)	118
	Heart Failure		5 (17)	103	5 (22)	103	3 (17)	65
	Hypertension		22 (73)	464	15 (65)	316	14 (78)	294
	Diabetes		7 (23)	149	7 (30)	149	6 (33)	125
	Obesity		20 (67)	388	17 (74)	342	11 (61)	210
Respiratory	Bronchitis		9 (30)	173	5 (22)	96	6 (33)	120
	COPD		9 (30)	177	5 (22)	97	6 (33)	118
	Asthma		9 (30)	170	6 (26)	114	5 (28)	97
MEDICATIONS†								
Cardiovascular	Beta-blockers		12 (40)	184	8 (35)	108	8 (44)	136
	Ca²⁺ Channel Blockers		11 (37)	204	8 (35)	161	6 (33)	106
	ACE Inhibitors		11 (37)	176	8 (35)	120	7 (39)	96
	Digoxin		3 (10)	56	3 (13)	56	3 (17)	56
	Statins		8 (27)	129	7 (30)	109	7 (39)	105
	Anti-coagulants		5 (17)	82	3 (13)	49	3 (17)	50
Respiratory	Bronchodilator‡		8 (27)	123	5 (22)	64	5 (28)	64
	Asthma Steroids		8 (27)	96	5 (22)	48	4 (22)	46

*Diagnoses are self-reported doctor's diagnoses except for: COPD and asthma are self-reported; diabetes determined via diabetic medications use, obesity defined as BMI > 30, bronchitis defined as chronic cough or phlegm on most days for 3 consecutive months or more during the year.

†Medications use lists participants who reported taking the medication at least once throughout the study on the day of their clinic visit.

‡Bronchodilator grouping includes beta-agonist and atrovent medications.

Air quality data

Table 2 and Table 3 present descriptive statistics and correlations for air pollutants, respectively. During the study, 24-hr PM_{2.5} concentrations averaged approximately 19 µg/m³ and ranged from 3.6 to 48.4 µg/m³. The correlations between pollutants showed similar patterns whether examining 1-day or 5-day moving averages. Ambient PM_{2.5} and SO₄²⁻ were highly correlated ($r = 0.90$ for 5-day moving averages), with slightly lower correlations between PM_{2.5} and the other pollutants (range of $r = 0.33 - 0.75$ for 5-day moving averages). Correlations were also relatively strong between apparent temperature (at current hour) and SO₄²⁻ and O₃ concentrations ($r = 0.57$ and 0.77 , respectively), likely due to the direct influence of meteorology on the formation of these secondary pollutants. In contrast, correlations were weaker between apparent temperature and EC, NO₂, SO₂ and CO, which are all primary pollutants ($r = 0.40, 0.10, -0.19$ and -0.32 , respectively). SO₄²⁻ comprised approximately 50% of the ambient PM_{2.5} mass, similar to other locations in the eastern U.S.; this suggests that the regional influence on PM_{2.5} in Steubenville was relatively high. EC comprised approximately 6% of the ambient PM_{2.5} mass, which also compares well to other areas of the country.

Table 2. Summary of air pollutant concentrations from the central monitoring site for 1- and 5-day moving average concentrations prior to the first participant visit each clinic day.

Average	Pollutant	N	Mean ± SD	Min	25th	50th	75th	Max	IQR*
1-day	PM _{2.5} (µg/m ³)	113	19.6 ± 10.4	3.6	11.7	17.7	25.0	48.4	13.3
	SO ₄ ²⁻ (µg/m ³)	109	6.8 ± 4.7	0.0	3.3	5.7	8.5	25.0	5.2
	EC (µg/m ³)	103	1.1 ± 0.6	0.3	0.7	1.0	1.3	3.6	0.6
	O ₃ (ppb)	111	21.8 ± 12.6	-0.8	12.0	20.2	28.5	74.8	16.5
	NO ₂ (ppb)	111	10.7 ± 6.7	-0.9	6.3	9.6	14.0	37.9	7.7
	SO ₂ (ppb)	106	10.4 ± 8.3	1.8	5.2	8.1	12.4	58.3	7.2
	CO (ppm)	110	0.2 ± 0.3	-0.1	0.0	0.1	0.3	1.5	0.3

5-day	PM _{2.5} (µg/m ³)	107	19.8 ± 7.5	4.9	13.9	19.0	23.9	38.7	10.0
	SO ₄ ²⁻ (µg/m ³)	97	6.7 ± 3.1	1.5	4.2	6.1	8.4	14.8	4.2
	EC (µg/m ³)	80	1.0 ± 0.4	0.4	0.7	0.9	1.2	2.2	0.5
	O ₃ (ppb)	101	22.2 ± 9.1	6.5	14.2	20.7	29.1	44.0	14.9
	NO ₂ (ppb)	98	10.5 ± 4.2	4.5	7.4	9.2	12.9	20.4	5.5
	SO ₂ (ppb)	103	10.7 ± 5.5	2.4	7.4	9.1	12.8	31.3	5.4
	CO (ppm)	106	0.2 ± 0.2	0.0	0.1	0.2	0.3	1.3	0.2

*IQR: interquartile range (75th – 25th percentile).

Table 3. Spearman correlations between all air parameters using 5-day moving average concentrations prior to first participant visit each clinic day.

	PM _{2.5}	SO ₄ ²⁻	EC	O ₃	NO ₂	SO ₂	CO
App Temp ‡	0.46 [†]	0.57 [†]	0.40*	0.77 [†]	0.10	-0.19	-0.32*
PM_{2.5}		0.90 [†]	0.75 [†]	0.33*	0.46 [†]	0.51 [†]	0.47 [†]
SO₄²⁻			0.60 [†]	0.53 [†]	0.37*	0.35*	0.26*
EC				0.15	0.66 [†]	0.46 [†]	0.55 [†]
O₃					-0.19	-0.21	-0.35*
NO₂						0.40*	0.50 [†]
SO₂							0.66 [†]

*p<0.05; [†]p<0.0001; ‡ Apparent temperature at current hour of clinic visit.

Associations between air pollution and cardiac arrhythmia

Results of the logistic regression models showed that air pollution increased the odds of having at least one SVE, but not VE, over the entire protocol (Table 4). The results for SVE were greatest for PM_{2.5}, SO₄²⁻ and O₃, while EC, NO₂, SO₂ and CO demonstrated weaker and null effects. The impacts of air pollution on the odds of SVE and VE did not differ when examining individual protocol intervals (i.e., resting, standing, exercise, etc.) in separate models.

Table 4. Odds ratios for SVE and VE for IQR increases in 5-day mean pollutant concentrations.*

Pollutant	5-day IQR	SVE			VE		
		N	Odds Ratio (95% CI)	p-value	N	Odds Ratio (95% CI)	p-value
Particles							
PM _{2.5}	10.0 µg/m ³	400	1.42 (0.99, 2.04)	0.07	314	1.02 (0.63, 1.65)	0.93
SO ₄ ²⁻	4.2 µg/m ³	356	1.70 (1.12, 2.57)	0.02	286	1.08 (0.65, 1.80)	0.78
EC	0.5 µg/m ³	310	1.15 (0.73, 1.81)	0.57	243	1.00 (0.57, 1.75)	0.99
Gases							
O ₃	14.9 ppb	387	1.78 (0.95, 3.35)	0.09	304	1.43 (0.63, 3.27)	0.41
NO ₂	5.5 ppb	379	0.89 (0.61, 1.28)	0.53	296	0.61 (0.36, 1.03)	0.08
SO ₂	5.4 ppb	397	1.04 (0.78, 1.39)	0.78	306	1.28 (0.85, 1.92)	0.25
CO	0.2 ppm	410	0.99 (0.76, 1.29)	0.93	317	1.05 (0.75, 1.46)	0.80

*Models control for apparent temperature at hour of clinic visit, season, day of week, hour of day, age and BMI.

Over the entire protocol, the OR for having at least one air pollution-related SVE was greatest at long-term moving average concentrations prior to the health assessment (Figure 1). For example, the OR for SVE was 1.42 (95% CI: 0.99, 2.04) for a 10 µg/m³ increase in 5-day PM_{2.5} as compared to 1.04 (95% CI: 0.73, 1.47) for a 13.3 µg/m³ increase in 1-day PM_{2.5}. The observed particle effects remained elevated or increased at moving average concentrations longer than 5-days. In these analyses, the effects of PM_{2.5} on the odds of SVE were not different when the HMPM PM_{2.5} time-series was filled in for missing Saturday data using regressions with 24-hour averaged TEOM data.

Including heart rate in the models resulted in more precise estimation of effects, with the 5-day moving average PM_{2.5} effect on the odds of SVE (OR: 1.48, 95% CI: 1.02, 2.15) becoming significant. Inclusion of 5-day O₃ in models together with either 5-day PM_{2.5} or SO₄²⁻ reduced the effects of both pollutants slightly. For example, with 5-day

SO_4^{2-} in the model, the 5-day O_3 effect fell to 1.57 (95% CI: 0.74, 3.35) and the SO_4^{2-} effect fell to 1.54 (95% CI: 1.00, 2.36).

In sensitivity analyses, the logistic model results were similar when all 30 participants were included in the analyses. In order to use the information provided by the counts of SVE and VE in our dataset, we additionally modeled the outcome as a count variable using negative binomial regression. Estimated associations between pollution levels and the rate of arrhythmias per 30-minute interval from this model were not significantly different from zero (results not shown).

Examination of effect modification

Figure 2 presents the results of interactions between 5-day moving average $\text{PM}_{2.5}$ concentrations and participant characteristics on the odds of SVE. Pollution effects were greatest for participants with cardiovascular conditions. Participants who reported a previous MI, for example, had an OR for SVE of 2.35 (95% CI: 1.03, 5.39) as compared to an OR of 1.31 (95% CI: 0.89, 1.91) in those without a previous MI, each for a $10.0 \mu\text{g}/\text{m}^3$ increase in 5-day $\text{PM}_{2.5}$ (test of significant difference, $p = 0.20$). In addition, participants who reported ever having hypertension showed a significantly higher effect (OR: 1.88, 95% CI: 1.21, 2.92) compared to those without hypertension (OR: 0.94, 95% CI: 0.56, 1.59) (test of significant difference, $p = 0.04$). Medications use showed collinear effects with diagnosis; participants on medications for cardiac conditions, such as beta blockers or Ca^{2+} channel blockers, experienced greater air pollution effects than those not on cardiac medications (corresponding to those without diagnosed cardiac disease). In contrast, participants with respiratory conditions (e.g. bronchitis, asthma) did

not influence the effect of PM_{2.5} on the odds of SVE. The null effect of PM_{2.5} on participants with angina is likely influenced by the fact that a majority of these participants (4 of 6) also had respiratory disease.

The temporal association between PM_{2.5} and SVE noted above was similar when considering only those participants with cardiac diagnoses. For example, subsetting the analyses only on those participants with hypertension, the OR for having at least one air pollution-related SVE was greater at 5-day moving average concentrations (2.07 [95% CI: 1.29, 3.32] for a 10 µg/m³ increase in 5-day PM_{2.5}) prior to the health assessment as compared to 1-day moving average concentrations (1.28 [95% CI: 0.82, 1.99] for a 13.3 µg/m³ increase in 1-day PM_{2.5}).

We also examined whether pollution effects were greater in participants with higher levels of ectopy. We observed that 5-day PM_{2.5} effects on the odds of SVE were significantly higher for participants with mean SVE rates greater than the population median rate of 1.5 beats/hr (OR: 1.99, 95% CI: 1.20, 3.29) as compared to participants with mean rates < 1.5 beats/hr (OR: 1.03, 95% CI: 0.65, 1.64) (test of significant difference, p = 0.05).

DISCUSSION

In a community with significant industrial sources for air pollution, our study demonstrates an association of particle pollution with increased odds of supraventricular arrhythmia in a cohort of older adults, with findings of 42%, 70% and 78% increases in odds of SVE associated with IQR increases in 5-day moving average PM_{2.5}, SO₄²⁻ and O₃, respectively. Air pollution effects were greatest for participants with a history of

clinically significant cardiac disease. Since two-pollutant models demonstrated stability in the effects of both particles and O₃, collectively our results may provide evidence of the combined effect of the secondary pollutant mix in Steubenville on cardiac arrhythmia. Specifically, the strong effects found with SO₄²⁻ are interesting as Steubenville is located in an industrial area of the Ohio River Valley, with little traffic but with a number of coal-fired power plants, which are the major source of SO₂, a SO₄²⁻ precursor. A previous study conducted in Boston, reporting on patients with implantable cardioverter defibrillators (ICD), found that traffic-related pollutants, particularly NO₂, showed the greatest odds of arrhythmia.[16] Our data suggest that pollution in an industrial location may also contribute to the risk of arrhythmia and they indicate the potential for varying impacts of air pollution by geographical location and source contributions.

Recent air pollution health effects studies have suggested mechanisms to support the association between ambient particle levels and SVE occurrence.[8][9][10][11] A leading hypothesis regarding the biological mechanism of air pollution health effects centers on the relationship between increased air pollution levels and autonomic nervous system imbalance,[17][18][19][20] which is known to aggravate SVE.[21] Our observation that long-term moving average pollutant concentrations exerted the greatest impact on the odds of SVE is consistent with the previous ICD patient studies,[16] [22] and the results may suggest the presence of an inflammatory mechanism as well.[23] Stone and Godleski suggest that elevated circulating cytokine levels as part of the systemic response can act to impair cardiac myocyte and electrophysiologic function,[24][25] which could cause cardiac irritability, thus increasing SVE occurrence.

We identified several factors that increased the odds of air pollution-related SVE within our cohort. Participants reporting previous cardiovascular conditions, including MI or hypertension, or individuals with mean SVE rates > 1.5 beats/hr were most susceptible to pollution-induced SVE. Participants on cardiac medications, including beta blockers or Ca^{2+} channel blockers, also showed sensitivity to air pollution mediated SVE. The association of these cardiac medications and increased sensitivity to air pollution is likely due to these medications serving as a surrogate for the presence of the clinical cardiovascular conditions and due not to the pharmacologic effect of the drug per se.

It has also been hypothesized that individuals with chronic lung disease comprise a susceptible subgroup, given their tendency for loss of vagal restraint, thereby leading to an increased heart rate and conditions suitable for ectopic tachycardias.[26] However, air pollution did not influence the odds of SVE in these participants. Our results differ from previous studies reporting significant effects of air pollution on SVE for groups of COPD patients.[8] [9] [11] A study of 16 COPD patients in Vancouver, whose mean rate of SVE was 33 beats/hr, reported a 22% increased rate of SVE for each $5.8 \mu\text{g}/\text{m}^3$ increase in $\text{PM}_{2.5}$. [11] Since the number of participants with bronchitis, COPD and/or asthma ($n = 5 - 6$) in our analyses was low, the lack of observed effects for these groups is likely a function of weak statistical power. Moreover, the small size ($n = 23$) of the cohort used in the SVE models may limit the generalizability of our findings regarding effect modification by diagnosis.

This study has several limitations. We recognize that the occurrence of at least one SVE or VE is a subclinical outcome. While we were able to detect associations between

pollution levels and the occurrence of at least one SVE during the protocol, we did not detect significant associations between pollution levels and the actual counts of arrhythmias. Thus, incorporating this extra information did not improve our ability to detect effects over the binary analysis. The large amount of overdispersion in the non-zero counts could have reduced our power to detect associations between pollution levels and the arrhythmia counts.

Effect sizes of air pollution on the odds of VE were extremely small. The lack of observed effects may have been due to a lack of vulnerability of our small ($n = 18$), and therefore limited, study population. Dockery and colleagues (2004) demonstrated in their study of ICD patients that ventricular arrhythmias within three days of a prior event presented significantly higher effects (OR: 1.60, 95% CI: 1.25, 2.04) for 2-day mean $PM_{2.5}$ than those more than three days since a previous event (OR: 0.98, 95% CI: 0.84, 1.14).[22] In a discussion of their findings, the authors suggest that air pollution may act in combination with acutely predisposing conditions, which increase ventricular electrical instability and that could lead to arrhythmia occurrence.[22] It is plausible that our study population did not experience such acutely predisposing conditions that would enhance the effects of air pollution on VE.

Our results are consistent with a previous study examining the effects of air pollution on both SVE and VE in the elderly, where significant effects were found with SVE but not VE.[8] In contrast, Riediker and colleagues (2004) found a significant effect of air pollution on both SVE and VE in their study of highway patrolmen, reporting a 19.1% increased rate of VE for each $10 \mu g/m^3$ increase in $PM_{2.5}$. [10] Studies of associations of $PM_{2.5}$ with risk of VE require a sufficient number of observations on a vulnerable study

population, and may, as in the case of the patrolman, require stress during the period of observation.

The relatively brief period of observation (i.e., 30-minute ECG recording per participant-visit) may have further limited our ability to capture the air pollution effect on the odds of VE, a relatively rare outcome. Recordings of longer length (e.g. 24-hour) as conducted in other panel studies[8][9][10] may provide outcome data for which associations with air pollution are better detected.

CONCLUSION

In summary, our results suggest that increased levels of ambient particulate air pollution associated with non-traffic sources may increase the risk of supraventricular arrhythmia in the elderly. Findings of long-term air pollution effects (\geq 5-day moving average concentrations prior to the health assessment) as the highest and most significant, may suggest that a long-acting mechanism promoted the ectopic beats in our population. Furthermore, the results suggest that individuals with a history of clinically significant cardiac disease may be at particular risk of particle health effects.

ACKNOWLEDGEMENTS

The authors wish to thank all of the participants of the study as well as those involved in project coordination (Monique Verrier, Meghan Syring, Marisa Barr), data collection (field team members from the Franciscan University of Steubenville) and data processing (Bruce Nearing, Gail McCallum, Marina Jacobson-Canner). The authors are also grateful for CONSOL Energy Inc. Research & Development's laboratory analysis of air pollutant samples and for the provision of continuous ambient monitoring data. Finally, the authors are appreciative of the insightful comments provided by Jeremy Sarnat and Petros Koutrakis.

FUNDING AND COMPETING INTEREST STATEMENT

This work is supported by funding from the National Institute of Environmental Health Sciences (ES-09825), the U.S. Environmental Protection Agency (R826780-01-0, R827353-01-0), the Ohio Coal Development Office within the Ohio Air Quality Development Authority (CDO/D-98-2), the Electric Power Research Institute (EP-P4464/C2166), the American Petroleum Institute (#78142), and the United States Department of Energy's National Energy Technology Laboratory (DE-FC26-00NT40771). These sponsors did not play a role in the study design, collection, analysis, nor interpretation of the data, writing of the report, nor the decision to submit the paper for publication. Any opinions, findings, conclusions, or recommendations expressed herein are those of the authors and do not necessarily reflect the views of these funding sources. The authors have no competing financial interests to declare.

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REFERENCES

1. Schwartz J. Air pollution and hospital admissions for heart disease in eight U.S. counties. *Epidemiology* 1999;**10**:17-22.
2. Zanobetti A, Schwartz J, Gold D. Are there sensitive subgroups for the effects of airborne particles? *Environ Health Perspect* 2000;**108**:841-845.
3. Mann JK, Tager IB, Lurmann F, *et al.* Air pollution and hospital admissions for ischemic heart disease in persons with congestive heart failure or arrhythmia. *Environ Health Perspect* 2002;**110**:1247-1252.
4. Metzger KB, Tolbert PE, Klein M, *et al.* Ambient air pollution and cardiovascular emergency department visits. *Epidemiology* 2004;**15**:46-56.
5. Samet JM, Dominici F, Curriero FC, *et al.* Fine particulate air pollution and mortality in 20 U.S. cities, 1987-1994. *N Engl J Med* 2000;**343**:1742-1749.
6. Hoek G, Brunekreef B, Fischer P, *et al.* The association between air pollution and heart failure, arrhythmia, embolism, thrombosis, and other cardiovascular causes of death in a time series study. *Epidemiology* 2001;**12**:355-357.
7. Pope CA, 3rd, Burnett RT, Thurston GD, *et al.* Cardiovascular mortality and long-term exposure to particulate air pollution: Epidemiological evidence of general pathophysiological pathways of disease. *Circulation* 2004;**109**:71-77.
8. Linn WS, Gong H, Jr., Clark KW, *et al.* Day-to-day particulate exposures and health changes in Los Angeles area residents with severe lung disease. *J Air Waste Manage Assoc* 1999;**49**:108-115.

9. Brauer M, Ebelst ST, Fisher TV, *et al.* Exposure of chronic obstructive pulmonary disease patients to particles: Respiratory and cardiovascular health effects. *J Expo Anal Environ Epidemiol* 2001;**11**:490-500.
10. Riediker M, Cascio WE, Griggs TR, *et al.* Particulate matter exposure in cars is associated with cardiovascular effects in healthy young men. *Am J Respir Crit Care Med* 2004;**169**:934-940.
11. Ebelst ST, Wilson WE, Brauer M. Exposure to ambient and nonambient components of particulate matter: A comparison of health effects. *Epidemiology* 2005;**16**:396-405.
12. Demokritou P, Kavouras IG, Ferguson ST, *et al.* Development and laboratory performance evaluation of a personal multipollutant sampler for simultaneous measurements of particulate and gaseous pollutants. *Aerosol Sci Technol* 2001;**35**:741-752.
13. Sarnat SE, Coull BA, Schwartz J, *et al.* Factors affecting the association between ambient concentrations and personal exposures to particles and gases. *Environ Health Perspect* 2006;In press.
14. Connell DP, Withum JA, Winter SE, *et al.* The Steubenville Comprehensive Air Monitoring Program (SCAMP): Analysis of short-term and episodic variations in PM_{2.5} concentrations using hourly air monitoring data. *J Air Waste Manage Assoc* 2005;**55**:559-573.
15. Connell DP, Withum JA, Winter SE, *et al.* The Steubenville Comprehensive Air Monitoring Program (SCAMP): Associations among fine particulate matter, co-

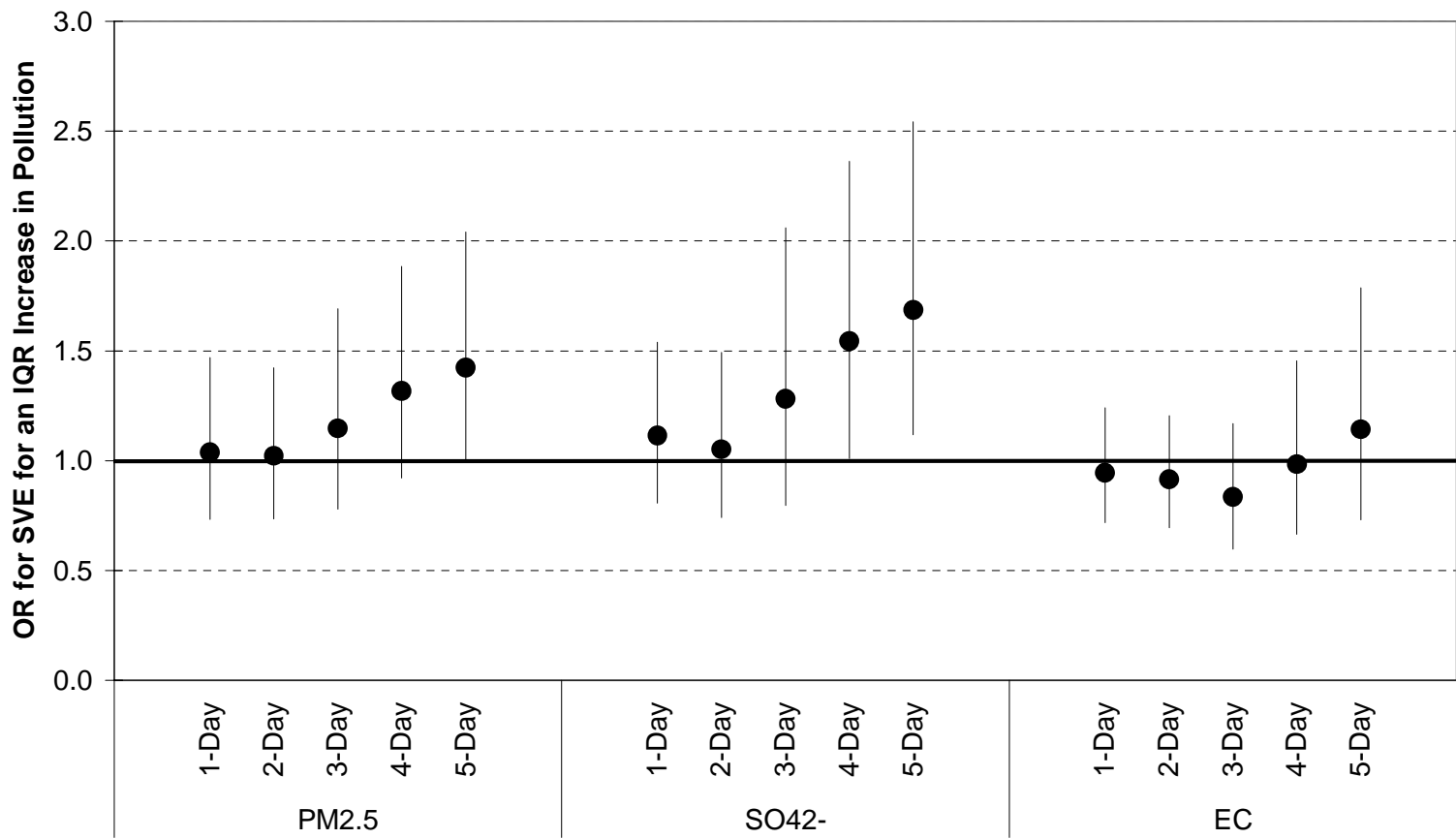
- pollutants, and meteorological conditions. *J Air Waste Manage Assoc* 2005;**55**:481-496.
16. Peters A, Liu E, Verrier RL, *et al.* Air pollution and incidence of cardiac arrhythmia. *Epidemiology* 2000;**11**:11-17.
 17. Pope CA, 3rd, Verrier RL, Lovett EG, *et al.* Heart rate variability associated with particulate air pollution. *Am Heart J* 1999; **138**:890-899.
 18. Gold DR, Litonjua A, Schwartz J, *et al.* Ambient pollution and heart rate variability. *Circulation* 2000;**101**:1267-1273.
 19. Creason J, Neas L, Walsh D, *et al.* Particulate matter and heart rate variability among elderly retirees: The Baltimore 1998 PM study. *J Expo Anal Environ Epidemiol* 2001;**11**:116-122.
 20. Liao DP, Duan YK, Whitsel EA, *et al.* Association of higher levels of ambient criteria pollutants with impaired cardiac autonomic control: A population-based study. *Am J Epidemiol* 2004;**159**:768-777.
 21. Ho JE, Stevenson WG, Strichartz GR, *et al.* Mechanisms of cardiac arrhythmias. In: Lilly LS, ed. *Pathophysiology of heart disease: A collaborative project of medical students and faculty, 3rd Edition*. Baltimore, MD: Lippincott Williams & Wilkins 2003:253-268.
 22. Dockery DW, Luttmann-Gibson H, Rich DQ, *et al.* Particulate air pollution and nonfatal cardiac events. Part II: Association of air pollution with confirmed arrhythmias recorded by implanted defibrillators. *Health Effects Institute Research Report Number 124* 2005:83-126.

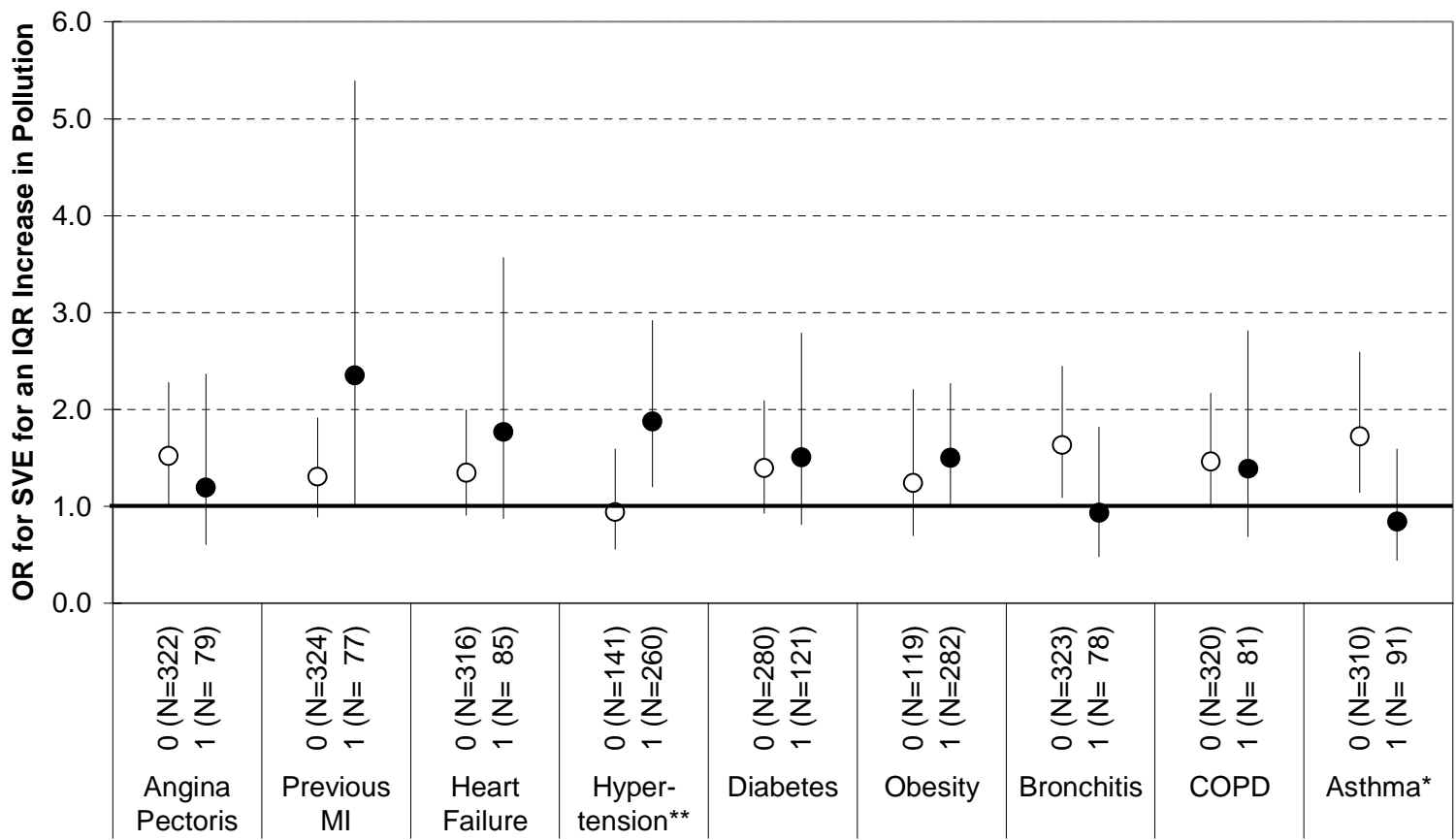
23. Seaton A, MacNee W, Donaldson K, *et al.* Particulate air pollution and acute health effects. *Lancet* 1995;**345**:176-178.
24. Stone PH, Godleski JJ. First steps toward understanding the pathophysiologic link between air pollution and cardiac mortality. *Am Heart J* 1999;**138**:804-807.
25. Brook RD, Brook JR, Rajagopalan S. Air pollution: The "heart" of the problem. *Curr Hypertens Rep* 2003;**5**:32-39.
26. Levine PA, Klein MD. Mechanisms of arrhythmias in chronic obstructive lung disease. *Geriatrics* 1976;**November**:47-56.

FIGURE LEGENDS

Figure 1. Odds ratios for SVE for IQR increases in pollutant concentrations at 1 to 5-day moving average concentrations (N=310-421). Models control for apparent temperature at hour of clinic visit, season, day of week, hour of day, age and BMI. Error bars indicate 95% confidence intervals.

Figure 2. Participant characteristics as effect modifiers of the 5-day moving average PM_{2.5} SVE association. 0=participants without the characteristic, 1=participants with the characteristic. Difference in pollution effect is significant at **p<0.05, *p<0.10. Models control for apparent temperature at hour of clinic visit, season, day of week, hour of day, age and BMI. Error bars indicate 95% confidence intervals.





APPENDIX F

SHORT-TERM EFFECTS OF AIR POLLUTION ON HEART RATE VARIABILITY IN SENIOR ADULTS IN STEUBENVILLE, OH

*MANUSCRIPT SUBMITTED TO JOURNAL OF OCCUPATIONAL AND
ENVIRONMENTAL MEDICINE*

Short-term Effects of Air Pollution on Heart Rate Variability in Senior Adults in Steubenville, OH

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Acknowledgements

This work is supported by funding from the National Institute of Environmental Health Sciences (ES-09825 and ES-00002), the U.S. Environmental Protection Agency (R826780-01-0, R827353-01-0), the Ohio Coal Development Office (CDO/D-98-2) and the United States Department of Energy's National Energy Technology Laboratory Award No. DE-FC26-00NT40771. Any opinions, findings, conclusions, or recommendations expressed herein are those of the authors and do not necessarily reflect the views of the U.S. Department of Energy. The authors are also grateful for CONSOL Energy Inc. Research and Development's laboratory analysis of air pollutant samples and for the provision of continuous ambient monitoring data.

Short-term Effects of Air Pollution on Heart Rate Variability in Senior Adults in Steubenville, OH

Key words: Air Pollution, short-term effects, autonomic function, heart rate variability, fine particles, sulfates

Running Title: Short-term Effects of Air Pollution on Autonomic Function

Abbreviations:

BC = Black Carbon

CI = confidence interval

CO = carbon monoxide

EC = elemental carbon

HF = high frequency power

HRV = heart rate variability

IQR = interquartile range

LF = low frequency power

NO₂ = nitrogen dioxide

O₃ = ozone

PM_{2.5} = particulate matter < 2.5 μm aerodynamic diameter

PNN50 = percent of adjacent RR intervals more than 50 msec different from each other

r-MSSD = square root of the mean of the squared differences between adjacent normal NN intervals

SDNN = standard deviation of normal RR Intervals

SO_2 = sulfur dioxide

SO_4^{2-} = sulfate

T_a = air temperature

T_d = dew point temperature

Abstract

Objective: We examined the association between ambient air pollution levels and heart rate variability (HRV) in a panel study of 32 subjects.

Methods: We used linear mixed models to analyze the effects of fine particles (PM_{2.5}), sulfate (SO₄²⁻), elemental carbon (EC), and gases on log-transformed standard deviation of normal RR intervals (SDNN), mean square of differences between adjacent RR intervals (r-MSSD), and high and low frequency power (HF, LF).

Results: An interquartile range (IQR) increase of 5.1 µg/m³ in SO₄²⁻ on the previous day was associated with a decrease of -3.3% SDNN (95% CI: -6.0%,-0.5%), -5.6% r-MSSD (-10.7%,-0.2%), and -10.3% HF (-19.5%,-0.1%). Associations with total PM_{2.5} were similar. HRV was not associated with EC, NO₂, SO₂ or O₃.

Conclusion: In addition to traffic-related particles, elevated levels of sulfate particles may also adversely affect autonomic function.

Introduction

Acute exposure to increases in ambient air pollution have been associated with increased cardiovascular mortality¹⁻⁴ and morbidity^{5,6} including increased subclinical indicators such as heart rate⁷⁻¹⁰ and blood pressure,¹¹⁻¹⁴ increased risk of cardiac arrhythmias^{15,16} and myocardial infarction,¹⁷⁻²⁰ ST-segment depression,²¹ and decreased HRV.^{8,10,22-28} HRV varies with autonomic tone, and reduced HRV is believed to be either in the pathway leading to increased cardiac risk or, alternatively, a marker of poor health.

Reduced HRV as a patient characteristic has been prospectively associated with the development of clinically significant cardiovascular disease,²⁹⁻³¹ and short-term HRV changes may also have clinical significance. HRV was decreased in patients in the period immediately preceding ischemic sudden death,³² and increased LF/HF ratios have been reported immediately preceding the onset of ventricular tachycardias,³³⁻³⁵ suggesting that shorter-term increased sympathetic and reduced parasympathetic activity may trigger cardiac arrhythmias.

While some studies investigating the relation of ambient pollution with HRV have specifically connected traffic-associated particle pollution with reduced HRV,¹⁰ less is known about the specific effects of non-traffic related pollution on HRV. In a study of elders living near a major inner city roadway in Boston, the effect of traffic-related black carbon (BC) on reduced HRV predominated.¹⁰ In this study of elders in the small industrial town of Steubenville, OH we examined the association of acute non-traffic as well as traffic-related air pollution exposures with repeatedly measured HRV, using sulfate (SO_4^{2-}) as a marker of non-traffic pollution.³⁶

Subjects and Methods

Study Population and Protocol

Thirty-two non-smoking senior adults from Steubenville, OH were recruited to participate in a study on air pollution and cardiovascular health during the summer and fall of 2000. Most participants lived in one of three centrally located, subsidized apartment buildings. Prior to inclusion in the study, a screening appointment was held to collect information on cardiovascular and respiratory health and medications and to obtain baseline electrocardiogram (12-lead MAC6, Marquette Medical Systems Inc.) recordings. Exclusion criteria included smoking, having a pacemaker, a recent acute coronary syndrome, atrial flutter or atrial fibrillation.

The study design was reviewed and approved by the Human Subjects Committees of the Brigham and Women's Hospital and the Harvard School of Public Health.

Study participants were seen weekly on the same day of the week and at the same time during summer (June 4 – August 18) and fall (September 25 – December 15) of 2000. A short questionnaire on recent symptoms, hospital or doctor's visits and medication use was administered, followed by Holter electrocardiogram monitoring (SEER MC, GE Medical Systems) with electrodes in a modified V5 and AVF position. The Holter monitoring protocol included: (1) five minutes of rest in a supine position; (2) three supine blood pressure (BP) measurements (NIBP Vital Signs Monitor, Welch Allyn); (3) five minutes of standing with three standing BP measurements taken after two minutes; (4) five minutes of exercise (walking) outdoors (weather and health permitting); (5) five minutes of rest in a supine position and; (6) two minutes and 20 seconds of paced breathing.

The Holter tapes were analyzed using a Marquette MARS Workstation (GE Medical Systems), with a 125 samples/second sampling rate following HRV guidelines³⁷. The Holter tapes were analyzed for time domain HRV measures including the standard deviation of normal RR intervals (SDNN), the mean square of differences between adjacent RR intervals (r-MSSD), and the percent of RR intervals more than 50 msec different from the prior interval (PNN50), and frequency domain HRV measures including low frequency power (LF) and high frequency power (HF). HRV and average heart rate was obtained for the whole protocol and for each separate interval.

Air Pollution Concentration and Meteorological Measurements

Hourly PM_{2.5} (TEOM 1400A, Rupprecht and Patashnick, East Greenbush, NY), ozone (O₃; API, Model 400 UV), nitrogen dioxide (NO₂; API, Model 200A Chemiluminescent), sulfur dioxide (SO₂; API, Model 100UV Fluorescent), and carbon monoxide (CO; API, Model 300 GFC) concentrations, temperature, relative humidity, and dew point were obtained from a centrally located ambient monitoring site at the Franciscan University of Steubenville operated by CONSOL Energy Inc. Research & Development.

We calculated apparent temperature as: $-2.653 + (0.994 \times Ta) + (0.0153 \times Td^2)$, where Ta is the air temperature and Td is the dew point.

We also collected 24-hour integrated PM_{2.5}, SO₄²⁻, EC and gaseous pollutant (O₃, NO₂, SO₂) samples beginning at 9 a.m. each day (except Saturday) using a Harvard multi-pollutant monitor³⁸. PM_{2.5} was measured gravimetrically. CONSOL analyzed the SO₄²⁻, O₃, NO₂ and SO₂ filters using ion chromatography and EC filters using thermal optical transmission. The

concentration of non-sulfate $PM_{2.5}$ was calculated as the difference between the measured $PM_{2.5}$ and sulfate concentrations, with sulfate assumed to be in the form of ammonium sulfate ($(NH_4)_2SO_4$), that is: non-sulfate $PM_{2.5} = PM_{2.5} - (SO_4^{2-} \cdot 132/96)$.

Statistical Methods

We examined the associations between air pollution and log-transformed HRV parameters and HR over the entire 30-minute protocol with linear mixed models including random subject effects, fixed effects of pollution, age, gender, race, obesity (defined as Body Mass Index > 30), season, time of day, apparent temperature, and a first-order auto-regressive process for the within-subject residuals.³⁹ In the main analysis 24-hour integrated pollution measurements (ending at 9 a.m. on the day of the electrocardiogram recording) were used as the exposure variable. Exposures also were assessed using moving averages (4 to 72 hours) of $PM_{2.5}$ and gaseous pollutant hourly concentrations before the electrocardiogram.

We assessed effect modification by medical conditions and medication intake in regression models including interaction terms between air pollution effects and potential effect modifiers. Results are restricted to r-MSSD as an outcome and $PM_{2.5}$ and SO_4^{2-} as exposure variables. Hypertension, coronary artery disease (angina or myocardial infarction), and congestive heart failure were defined as subject-specific conditions. Medication intake was defined as a time-varying condition. Results were similar when we defined medication intake as subject-specific (subject defined as “on medication” when medication was taken on most days).

Results are reported as estimated percent differences and 95% confidence interval (CI) in HRV or heart rate associated with an IQR increase in each pollutant. All analyses were done with SAS software version 9.1 (SAS Institute, Inc; Cary, NC).

Results

Study Population

Of 53 subjects screened, thirty-two were recruited for the study. There were 645 health visits, with an average of 20 (range 4-24) per subject (Table 1). The mean age at screening was 71 years; 91% of the participants were female, and 69% were white. Most participants (84%) had one or more cardiovascular condition, and two thirds were on one or more medications (Table 1). Twenty subjects (63%) with body mass index greater than 30 kg/m² were classified as obese. Average HRV and heart rate measurements are listed in Table 2.

Ambient Air Pollutant Concentrations

Daily mean ambient air pollution concentrations are summarized in Table 3 for PM_{2.5}, SO₄²⁻, non-sulfate PM_{2.5}, EC, NO₂, SO₂, and O₃ during the study period. Ambient concentrations of the measured criteria gases were moderate in Steubenville while ambient fine particle concentrations were relatively high, with some daily means approaching the 24-h National Ambient Air Quality Standards (65 µg/m³ for PM_{2.5}). On six of the 118 days (5%) the daily mean PM_{2.5} (based on the hourly data averaged over the day) exceeded 50 µg/m³ (maximum 50.7 µg/m³). The correlation coefficients between daily particulate and gas concentrations are shown in Table 4.

Associations between Air Pollution and Heart Rate Variability Measures

Time and frequency domain HRV parameters including SDNN, r-MSSD, HF and LF were negatively associated with the mean PM_{2.5} concentration during the day before the HRV

measurement (Table 5). For example, R-MSSD was -6.5% (CI -12.1% to -0.6%, $p=0.03$) lower with each IQR range increase in mean $PM_{2.5}$ ($13.4 \mu\text{g}/\text{m}^3$) on the day prior to the measurement. We found no suggestion of an association of HRV with mean NO_2 , SO_2 , or O_3 ($p>0.10$). For PNN50, the results were similar in magnitude but with wider confidence intervals (data not shown). In contrast to HRV, we found that heart rate was elevated with increased $PM_{2.5}$. For example, heart rate was elevated by 1.1% (CI 0.2% to 2.1%, $p=0.02$) with each IQR range increase in $PM_{2.5}$ on the day before the measurement.

We compared the associations of $PM_{2.5}$ with those for its major components (EC, SO_4^{2-} and non-sulfate $PM_{2.5}$). We found that SO_4^{2-} was significantly associated with decreased SDNN, r-MSSD, HF, and LF and increased heart rate, while non-sulfate $PM_{2.5}$ was moderately associated ($p<0.10$) with decreased SDNN, r-MSSD, and increased heart rate. In contrast, we found no associations with HRV for the EC fraction of $PM_{2.5}$ or for any of the gaseous pollutants ($p>0.10$).

We compared the association between HRV and ambient $PM_{2.5}$ for averaging periods at 4, 6, 12, 24, 48 and 72 hours prior to the electrocardiogram recordings. For all HRV measures, the associations between the daily mean $PM_{2.5}$ and HRV were similar to those for the 24- and 48-hour moving averages (Figure 1).

Effect Modification

We found no consistent effect modification of the air pollution effect by congestive heart failure, coronary artery disease, or statin intake (Table 6). However, we did find that subjects with hypertension showed very little response to increased levels of $PM_{2.5}$, while subjects without hypertension showed a change of -17.8% in r-MSSD [CI -25.8 to -8.9] associated with increased

PM_{2.5} (p-value of interaction = 0.002). We also found stronger effects of PM_{2.5} on days when subjects were not on beta-blocker medication (-10.0% r-MSSD [CI -16.0 to -3.4] for a PM_{2.5} increase of 13.4 µg/m³) as compared to no effect (1.6%, [CI -7.6 to 11.9]) on days when subjects were on beta-blockers (interaction p-value = 0.03). Similarly, we found higher PM_{2.5} effects on days when subjects were not on statins (-8.4% r-MSSD [CI -14.4 to -2.0]) as compared to days when they were (interaction p-value = 0.15). While 52% of those with hypertension and 0% of those without hypertension were on beta blockers, the effect modification by hypertension remained significant (p<0.05) in the 20 subjects not on beta-blockers. Associations between sulfates and HRV were similarly modified by hypertension and beta blocker intake, with slightly higher interaction p-values.

Discussion

We found reduced HRV in a group of elderly subjects living in Steubenville, OH associated with increased total PM_{2.5} and SO₄²⁻ during the previous day. Reduced HRV was more weakly related to the non-sulfate component of PM_{2.5}, and was not related to increases in pollutants generally associated with traffic sources such as EC, O₃, or NO₂. Increases in overall PM_{2.5} and SO₄²⁻ were associated not only with decreased r-MSSD and HF, suggesting reduction in vagal tone, but also with decreased LF and increased heart rate, suggesting increases in sympathetic tone.

Our sulfate effects suggest that in an environment where non-traffic sources of pollution (e.g. coal combustion at power plants and steel mills) contribute significantly to ambient pollution in the community, elevated levels of particles from these sources may also adversely affect cardiac autonomic function and clinical cardiac outcomes. This conclusion is consistent with the American Cancer Society's cohort study⁴⁰ in which sulfate concentrations were a significant predictor of cardiovascular mortality, the time-series results within the Six-Cities Study (which included Steubenville) by Laden et al.,⁴¹ who found sulfur was predictive of short-term mortality, and the findings by O'Neill et al.,⁴² who found increased sulfate concentrations to be a significant predictor of decreased vascular reactivity in diabetics.

Within the framework of the 2000-2002 Steubenville Comprehensive Air Monitoring Program (SCAMP), Connell et al.⁴³ studied fine particulate matter concentration and composition and found that SO₄²⁻ is the major constituent of PM_{2.5} in Steubenville, accounting for approximately 31% of total mass. Within the period of this study, ambient SO₄²⁻ comprised between 52% and 43% of the total PM_{2.5} mass (as (NH₄)₂SO₄).⁴⁴ The SCAMP analyses⁴³ suggest

that local sources in Steubenville contribute only approximately $4.6 \mu\text{g}/\text{m}^3$ (25% of the annual mean $18.4 \mu\text{g}/\text{m}^3$) to the city's $\text{PM}_{2.5}$ concentrations. The main source of $\text{PM}_{2.5}$ and sulfate in Steubenville is fossil fuel (primarily coal) combustion,³⁶ most likely produced by the steel mills and power plants located in the Ohio River Valley.

Traffic pollutant effects predominated in our two previous Boston studies conducted with a similar study design in a community of elders living next to a city roadway.^{8,10} In the first study, SO_4^{2-} and BC measurements were not available, but we found associations of ozone as well as overall $\text{PM}_{2.5}$ with reduced HRV.⁸ In the second study, reduced HRV was most strongly and consistently associated with increased BC, which is a marker for traffic exposure¹⁰. Mean air pollution levels during the study period from June to December 2000 show that $\text{PM}_{2.5}$ and SO_4^{2-} were twice as high in Steubenville than in Boston ($\text{PM}_{2.5}$: 20 vs. $10 \mu\text{g}/\text{m}^3$, SO_4^{2-} : 6.9 vs. $3.0 \mu\text{g}/\text{m}^3$) whereas the traffic-related pollutant NO_2 was lower in Steubenville (10.0 vs. 21.5 ppb) (Steubenville: Table 3, Boston: data not shown).

Our results for Steubenville and Boston elders, suggesting pollution-associated decreases in vagal stimulation and/or increases in sympathetic tone, are consistent with findings from many other studies that did not separate out components of $\text{PM}_{2.5}$ that might enable the investigators to distinguish traffic from non-traffic $\text{PM}_{2.5}$ effects (Table 7).

Elevation of overall particle pollution ($\text{PM}_{2.5}$ or PM_{10}) has been related to reduction in HRV or increases in heart rate in repeated measures elders studies from Baltimore,^{23,24} Utah Valley,^{22,25} and Mexico City.²⁶ While studies have not consistently found that elevated pollution levels increase HR at the same time as they reduce HRV, it has been proposed that the association of increased pollution with reduced HRV may be mediated, in part, through the

increase in heart rate that results from increased sympathetic tone. For all HRV parameters in our study, controlling for heart rate led to small changes in the magnitude of the effect estimates (Figure 2), suggesting that a portion, but not all, of the HRV associations related to the pathways reflected by heart rate.

Similar associations of particle mass with HRV have been reported in a chamber study on elderly volunteers exposed for two hours to concentrated air pollution particles, in which significant decrements in HRV in both time and frequency domain were found immediately and 24 hours after exposure.⁴⁵ While the majority of the studies point to a significant association of particle pollution with reduced HRV, effects of PM_{2.5} on HRV have not been found in panel studies conducted in Northwestern US and Canada, perhaps because of differences in particle composition or, alternatively, low power.⁴⁶⁻⁴⁸ Also, in contrast to the negative associations noted above, increases in traffic pollution were associated with increased HRV in a study of a younger healthy population of traffic police in North Carolina.⁴⁹ The investigators hypothesize that the nature of the pollution-related autonomic dysfunction may depend on the age and underlying disease states of the study participants.

Between-person comparisons of HRV show that reduced HRV may be a marker for overall ill health.²⁹ While the significance of within-person short-term changes in HRV is somewhat less well established, a study of ischemic events showed that they were preceded by decreased high frequency HRV in the hour before the event, and in low frequency HRV in the 4 minutes before the event.⁵⁰ Similarly, a significant decrease in heart HRV has been reported to precede paroxysmal atrial fibrillation.⁵¹

We found effect modification of hypertension and beta-blocker intake in this study, with stronger associations between particles and HRV among the “healthier” subjects (no hypertension, not on beta-blocker). These findings contrast with several previous studies which showed stronger adverse effects of air pollution on HRV in subjects with preexisting cardiovascular conditions,²³ previous myocardial infarction,¹⁰ or hypertension.²⁶⁻²⁸ Two of those studies were cross-sectional analyses of large cohorts, with much larger numbers of subjects.^{27,28} The small numbers of subjects in panel studies such as this one make examination of effect modification difficult, and this result may be a chance finding. Alternatively, it is possible that in sicker populations other influences on HRV (for example medication) predominate, making it more difficult to detect pollution effects. In support of our findings, a panel study of elderly subjects in Seattle also found associations between particles and heart rate primarily in healthy subjects.⁵²

Personal exposures to particles and gases were measured in a subset of the study population. Associations between ambient particle concentrations and corresponding personal exposures were strong, whereas associations between ambient gases and their corresponding personal exposures were much weaker.⁴⁴ We conclude that PM_{2.5} and SO₄²⁻ ambient concentrations are very good proxies for the personal exposure of the study participants.

Recently, the scientific and regulatory communities have focused their attention on the adverse cardiac effects of ambient particles from mobile sources. This study supports the hypothesis that in an environment such as Steubenville, Ohio, where non-traffic sources of pollution contribute significantly to ambient pollution in the community, particle pollution from these sources, particularly sulfate particles, can also have adverse effects on autonomic function.

Acknowledgements

The authors wish to thank all of the participants of the study as well as Monique Verrier, Meghan Syring, Bruce Nearing, Gail McCallum, Marisa Barr and Marina Jacobson-Canner. We also thank the field team members from Franciscan University of Steubenville, in particular Kim Agnew, Michelle Bellini, Karen Gay, Sara Krich, Debra Santarelli, Steven Shea, and Teresa Sobie. The authors are also grateful for CONSOL Energy Inc. Research and Development's laboratory analysis of air pollutant samples and for the provision of continuous ambient monitoring data. This work is supported by funding from the National Institute of Environmental Health Sciences (ES-09825 and ES-00002), the U.S. Environmental Protection Agency (R826780-01-0, R827353-01-0), the Ohio Coal Development Office (CDO/D-98-2) and the United States Department of Energy's National Energy Technology Laboratory Award No. DE-FC26-00NT40771. Any opinions, findings, conclusions, or recommendations expressed herein are those of the authors and do not necessarily reflect the views of the U.S. Department of Energy.

References

1. Schwartz J, Dockery DW. Particulate air pollution and daily mortality in Steubenville, Ohio. *Am J Epidemiol* 1992;135:12-9.
2. Dockery DW, Pope CA, 3rd, Xu X, et al. An association between air pollution and mortality in six U.S. cities. *N Engl J Med* 1993;329:1753-9.
3. Pope CA, 3rd, Thun MJ, Namboodiri MM, et al. Particulate air pollution as a predictor of mortality in a prospective study of U.S. adults. *Am J Respir Crit Care Med* 1995;151:669-74.
4. Samet JM, Dominici F, Currier FC, Coursac I, Zeger SL. Fine particulate air pollution and mortality in 20 U.S. cities, 1987-1994. *N Engl J Med* 2000;343:1742-9.
5. Zanobetti A, Schwartz J. The effect of particulate air pollution on emergency admissions for myocardial infarction: a multicity case-crossover analysis. *Environ Health Perspect* 2005;113:978-82.
6. Wellenius GA, Schwartz J, Mittleman MA. Air pollution and hospital admissions for ischemic and hemorrhagic stroke among medicare beneficiaries. *Stroke* 2005;36:2549-53.
7. Peters A, Perz S, Doring A, Stieber J, Koenig W, Wichmann HE. Increases in heart rate during an air pollution episode. *Am J Epidemiol* 1999;150:1094-8.
8. Gold DR, Litonjua A, Schwartz J, et al. Ambient pollution and heart rate variability. *Circulation* 2000;101:1267-1273.

9. Pope CA, 3rd, Dockery DW, Kanner RE, Villegas GM, Schwartz J. Oxygen Saturation, Pulse Rate, and Particulate Air Pollution. A Daily Time-Series Panel Study. *American Journal of Respiratory & Critical Care Medicine* 1999;159:365-372.
10. Schwartz J, Litonjua A, Suh H, et al. Traffic related pollution and heart rate variability in a panel of elderly subjects. *Thorax* 2005;60:455-61.
11. Ibald-Mulli A, Stieber J, Wichmann HE, Koenig W, Peters A. Effects of air pollution on blood pressure: a population-based approach. *Am J Public Health* 2001;91:571-7.
12. Ibald-Mulli A, Timonen KL, Peters A, et al. Effects of particulate air pollution on blood pressure and heart rate in subjects with cardiovascular disease: a multicenter approach. *Environ Health Perspect* 2004;112:369-77.
13. Zanobetti A, Canner MJ, Stone PH, et al. Ambient pollution and blood pressure in cardiac rehabilitation patients. *Circulation* 2004;110:2184-9.
14. de Paula Santos U, Braga AL, Giorgi DM, et al. Effects of air pollution on blood pressure and heart rate variability: a panel study of vehicular traffic controllers in the city of Sao Paulo, Brazil. *Eur Heart J* 2005;26:193-200.
15. Peters A, Liu E, Verrier RL, et al. Air pollution and incidence of cardiac arrhythmia. *Epidemiology* 2000;11:11-7.
16. Dockery DW, Luttmann-Gibson H, Rich DQ, et al. Association of air pollution with increased incidence of ventricular tachyarrhythmias recorded by implanted cardioverter defibrillators. *Environ Health Perspect* 2005;113:670-4.

17. Peters A, Dockery DW, Muller JE, Mittleman MA. Increased particulate air pollution and the triggering of myocardial infarction. *Circulation* 2001;103:2810-5.
18. Grazuleviciene R, Maroziene L, Dulskiene V, et al. Exposure to urban nitrogen dioxide pollution and the risk of myocardial infarction. *Scand J Work Environ Health* 2004;30:293-8.
19. Peters A, von Klot S, Heier M, et al. Exposure to traffic and the onset of myocardial infarction. *N Engl J Med* 2004;351:1721-30.
20. D'Ippoliti D, Forastiere F, Ancona C, et al. Air pollution and myocardial infarction in Rome: a case-crossover analysis. *Epidemiology* 2003;14:528-35.
21. Gold DR, Litonjua AA, Zanobetti A, et al. Air pollution and ST-segment depression in elderly subjects. *Environ Health Perspect* 2005;113:883-7.
22. Pope CA, 3rd, Verrier RL, Lovell EG, et al. Heart rate variability associated with particulate air pollution. *American Heart Journal* 1999;138:890-899.
23. Liao D, Creason J, Shy C, Williams R, Watts R, Zweidinger R. Daily variation of particulate air pollution and poor cardiac autonomic control in the elderly. *Environ Health Perspect* 1999;107:521-5.
24. Creason J, Neas L, Walsh D, et al. Particulate matter and heart rate variability among elderly retirees: the Baltimore 1998 PM study. *J Expo Anal Environ Epidemiol* 2001;11:116-22.

25. Pope CA, 3rd, Hansen ML, Long RW, et al. Ambient particulate air pollution, heart rate variability, and blood markers of inflammation in a panel of elderly subjects. *Environ Health Perspect* 2004;112:339-45.
26. Holguin F, Tellez-Rojo MM, Hernandez M, et al. Air pollution and heart rate variability among the elderly in Mexico City. *Epidemiology* 2003;14:521-7.
27. Liao D, Duan Y, Whitsel EA, et al. Association of higher levels of ambient criteria pollutants with impaired cardiac autonomic control: a population-based study. *Am J Epidemiol* 2004;159:768-77.
28. Park SK, O'Neill MS, Vokonas PS, Sparrow D, Schwartz J. Effects of air pollution on heart rate variability: the VA normative aging study. *Environ Health Perspect* 2005;113:304-9.
29. Tsuji H, Larson MG, Venditti FJ, Jr., et al. Impact of reduced heart rate variability on risk for cardiac events. The Framingham Heart Study. *Circulation* 1996;94:2850-5.
30. La Rovere MT, Bigger JT, Jr., Marcus FI, Mortara A, Schwartz PJ. Baroreflex sensitivity and heart-rate variability in prediction of total cardiac mortality after myocardial infarction. ATRAMI (Autonomic Tone and Reflexes After Myocardial Infarction) Investigators. *Lancet* 1998;351:478-84.
31. Schroeder EB, Liao D, Chambless LE, Prineas RJ, Evans GW, Heiss G. Hypertension, blood pressure, and heart rate variability: the Atherosclerosis Risk in Communities (ARIC) study. *Hypertension* 2003;42:1106-11.

32. Pozzati A, Pancaldi LG, Di Pasquale G, Pinelli G, Bugiardini R. Transient sympathovagal imbalance triggers "ischemic" sudden death in patients undergoing electrocardiographic Holter monitoring. *J Am Coll Cardiol* 1996;27:847-52.
33. Lombardi F, Porta A, Marzegalli M, et al. Heart rate variability patterns before ventricular tachycardia onset in patients with an implantable cardioverter defibrillator. Participating Investigators of ICD-HRV Italian Study Group. *Am J Cardiol* 2000;86:959-63.
34. Pruvot E, Thonet G, Vesin JM, et al. Heart rate dynamics at the onset of ventricular tachyarrhythmias as retrieved from implantable cardioverter-defibrillators in patients with coronary artery disease. *Circulation* 2000;101:2398-404.
35. Huikuri HV, Valkama JO, Airaksinen KE, et al. Frequency domain measures of heart rate variability before the onset of nonsustained and sustained ventricular tachycardia in patients with coronary artery disease. *Circulation* 1993;87:1220-8.
36. Koutrakis P, Spengler JD. Source Apportionment of Ambient Particles in Steubenville, OH using Specific Rotation Factor Analysis. *Atmospheric Environment* 1987;21:1511-1519.
37. Task Force of the European Society of Cardiology and the North American Society of Pacing and Electrophysiology. Heart rate variability. Standards of measurements, physiological interpretation, and clinical use. *Circulation* 1996;93:1043-1065.
38. Demokritou P, Kavouras IG, Ferguson ST, Koutrakis P. Development and Laboratory Performance Evaluation of a Personal Multipollutant Sampler for Simultaneous

Measurements of Particulate and Gaseous Pollutants. *Aerosol Science and Technology* 2001;35:741-752.

39. Diggle P, Heagerty P, Liang K-Y, Zeger S. *Analysis of Longitudinal Data, 2nd ed.* New York: Oxford University Press; 2002.
40. Pope CA, 3rd, Burnett RT, Thun MJ, et al. Lung cancer, cardiopulmonary mortality, and long-term exposure to fine particulate air pollution. *Jama* 2002;287:1132-41.
41. Laden F, Neas LM, Dockery DW, Schwartz J. Association of fine particulate matter from different sources with daily mortality in six U.S. cities. *Environ Health Perspect* 2000;108:941-7.
42. O'Neill MS, Veves A, Zanobetti A, et al. Diabetes enhances vulnerability to particulate air pollution-associated impairment in vascular reactivity and endothelial function. *Circulation* 2005;111:2913-20.
43. Connell DP, Withum JA, Winter SE, Statnick RM, Bilonick RA. The Steubenville comprehensive air monitoring program (SCAMP): overview and statistical considerations. *J Air Waste Manag Assoc* 2005;55:467-80.
44. Sarnat SE, Coull BA, Schwartz J, Gold DR, Suh HH. Factors affecting the association between ambient concentrations and personal exposures to particles and gases. *Environ Health Perspect* 2006;114:649-54.

45. Devlin RB, Ghio AJ, Kehrl H, Sanders G, Cascio W. Elderly humans exposed to concentrated air pollution particles have decreased heart rate variability. *Eur Respir J Suppl* 2003;40:76s-80s.
46. Brauer M, Ebelst ST, Fisher TV, Brumm J, Petkau AJ, Vedal S. Exposure of chronic obstructive pulmonary disease patients to particles: respiratory and cardiovascular health effects. *J Expo Anal Environ Epidemiol* 2001;11:490-500.
47. Sullivan JH, Schreuder AB, Trenga CA, et al. Association between short term exposure to fine particulate matter and heart rate variability in older subjects with and without heart disease. *Thorax* 2005;60:462-6.
48. Ebelst ST, Wilson WE, Brauer M. Exposure to ambient and nonambient components of particulate matter: a comparison of health effects. *Epidemiology* 2005;16:396-405.
49. Riediker M, Cascio WE, Griggs TR, et al. Particulate matter exposure in cars is associated with cardiovascular effects in healthy young men. *Am J Respir Crit Care Med* 2004;169:934-40.
50. Kop WJ, Verdino RJ, Gottdiener JS, O'Leary ST, Bairey Merz CN, Krantz DS. Changes in heart rate and heart rate variability before ambulatory ischemic events. *J Am Coll Cardiol* 2001;38:742-9.
51. Bettoni M, Zimmermann M. Autonomic tone variations before the onset of paroxysmal atrial fibrillation. *Circulation* 2002;105:2753-9.

52. Mar TF, Koenig JQ, Jansen K, et al. Fine particulate air pollution and cardiorespiratory effects in the elderly. *Epidemiology* 2005;16:681-7.

Legends

Figure 1: Estimated percent changes [95% Confidence intervals] in SDNN and r-MSSD associated with an interquartile range increase in PM_{2.5} mean air pollution for different exposure metrics: moving averages for 4, 6, 12, 24, 48, and 72 hours (based on continuous data) and 1-day lag (based on filter data, 9 a.m. to 9 a.m.) IQRs are: 17.7, 17.5, 16.6, 15.2, 12.6, 11.4, and 13.4 µg/m³, respectively.

Figure 2: Estimated percent changes [95% Confidence intervals] in HRV parameters associated with an interquartile range increase in PM_{2.5} and SO₄²⁻ (previous day) in the base model and adjusted for heart rate.

Table 1: Participant Characteristics (n=32)

Characteristic	Mean	range
Number of visits	20.2	4-24
Age (years)	70.8	54-90
Body Mass Index (kg/m ²)	31.4	21-47

Characteristic	N	%
Gender		
Male	3	9
Female	29	91
Race/Ethnicity		
Black	10	31
White	22	69
Diagnoses		
Angina	6	19
Myocardial Infarct	7	22
Coronary Artery Disease ^a	11	34
Congestive Heart Failure	6	19
Diabetes	7	22
Hypertension	23	72
Chronic Obstructive Pulmonary Disease	9	28
Medications Use ^b		
Beta Blocker	12	38
Calcium Channel Blocker	11	34
Statin	9	28
Angiotensin Converting Enzyme Inhibitors	11	34
Digoxin	3	9
Any of the above listed medications	21	66

^a Angina or Myocardial Infarct

^b Medication use is defined as reported at least once during the study

Table 2: Average Heart Rate Variability (HRV) and average Heart Rate during 30-minute protocol for 32 subjects (638 valid measurements)

	N	Mean (standard deviation)
SDNN (ms)	638	86.4 (32.6)
r-MSSD (ms)	638	36.3 (37.4)
PNN50 (%)	638	7.7 (13.5)
HF (ms ²)	634	239 (425)
LF (ms ²)	635	355 (589)
Heart Rate (beats/min)	638	79.7 (12.0)

Table 3: Average air pollution concentrations during the two study periods (Jun 01 – Aug 25, 2000, Sep 20 – Dec 15, 2000)

	Pollutant	N	Mean	25th percentile	75th	IQR
Daily data	PM _{2.5} (µg/m ³)	137	19.7	11.6	25.0	13.4
9 a.m. to 9 a.m.	SO ₄ ²⁻ (µg/m ³)	133	6.9	3.3	8.5	5.1
	Non-sulfate PM _{2.5} (µg/m ³)	133	10.0	6.3	11.5	5.3
	EC (µg/m ³)	127	1.1	0.7	1.3	0.6
	NO ₂ (ppb)	133	10.5	6.0	13.4	7.3
	SO ₂ (ppb)	134	4.1	0.4	4.7	4.3
	O ₃ (ppb)	134	22.2	12.1	28.5	16.4
Hourly data	PM _{2.5} (µg/m ³)	118	20.0	11.3	26.5	15.2
averaged over	NO ₂ (ppb)	113	10.9	7.4	14.7	7.3
Calendar days	SO ₂ (ppb)	151	10.8	5.4	13.2	7.8
	O ₃ (ppb)	144	27.3	15.7	36.3	20.6

Table 4: Pearson correlation coefficients between daily air pollution concentrations during the two study periods (Jun 01 – Aug 25, 2000, Sep 20 – Dec 15, 2000)

	SO ₄ ²⁻	non-sulfate PM _{2.5}	EC	NO ₂	SO ₂	O ₃
PM _{2.5}	0.90	0.87	0.59	0.40	0.33	0.23
SO ₄ ²⁻		0.57	0.47	0.24	0.31	0.36
non-sulfate PM _{2.5}			0.62	0.54	0.31	0.01
EC				0.54	0.38	-0.10
NO ₂					0.26	-0.29
SO ₂						-0.03

Table 5: Estimated percent changes [95% Confidence intervals] in HRV and HR associated with an interquartile range increase in mean air pollution on the day before^a the HRV measurement.

	IQR	n	SDNN		r-MSSD		HF		LF		Heart Rate	
			% change	95% CI	% change	95% CI	% change	95% CI	% change	95% CI	% change	95% CI
PM_{2.5}	13.4 μg/m ³	559	-4.0*	[-7.0,-0.9]	-6.5*	[-12.1,-0.6]	-11.4*	[-21.5,-0.1]	-10.7*	[-20.1,-0.3]	1.1*	[0.2,2.1]
SO₄²⁻	5.1 μg/m ³	538	-3.3*	[-6.0,-0.5]	-5.6*	[-10.7,-0.2]	-10.3*	[-19.5,-0.1]	-8.4 ⁺	[-17.0,1.2]	0.8 ⁺	[-0.02,1.7]
non-sulfate PM_{2.5}	5.3 μg/m ³	538	-2.1 ⁺	[-4.3,0.1]	-3.8 ⁺	[-8.0,0.5]	-5.9	[-13.7,2.6]	-7.2	[-14.3,0.5]	0.7 ⁺	[-0.01,1.31]
EC	0.6 μg/m ³	511	1.5	[-1.1,4.1]	-1.1	[-5.8,3.9]	3.1	[-6.4,13.5]	-4.7	[-12.7,4.0]	-0.2	[-0.9,0.6]
NO₂	7.3 ppb	552	0.1	[-2.2,2.4]	-1.3	[-5.6,3.2]	-1.3	[-9.4,7.7]	-2.9	[-10.4,5.2]	0.0	[-0.7,0.7]
SO₂	4.3 ppb	558	0.7	[-1.0,2.5]	0.5	[-2.8,4.0]	1.7	[-4.9,8.7]	4.9	[-1.4,11.5]	0.3	[-0.2,0.8]
O₃	16.4 ppb	552	-1.1	[-4.2,2.1]	-2.5	[-8.4,3.9]	-3.8	[-14.9,8.7]	-3.9	[-14.2,7.6]	0.7	[-0.3,1.6]

^a previous day refers to the measurement cycle 9 a.m. of the previous day to 9. a.m. of the current day

⁺ p<0.10, * p<0.05

Table 6: Effect modification of association of r-MSSD with PM_{2.5} (13.4 µg/m³) and SO₄²⁻ (5.1 µg/m³) on the previous day by chronic cardiovascular diagnoses and medication intake

		N (Subjects)	N (obs)	PM _{2.5}		SO ₄ ²⁻	
				% change	95% CI	% change	95% CI
CVD Diagnoses							
Hypertension	Yes	23	406	-2.0	[-8.5, 4.8]	-3.0	[-8.6, 3.1]
	No	9	153	-17.8**	[-25.8, -8.9]	-14.2**	[-22.6, -5.0]
		<i>p-value interaction</i>		0.002		0.03	
Coronary Artery Disease	Yes	11	178	-2.9	[-11.6, 6.7]	-1.5	[-9.5, 7.1]
	No	21	381	-8.3*	[-14.6, -1.6]	-7.7*	[-13.6, -1.5]
		<i>p-value interaction</i>		0.29		0.20	
Congestive Heart Failure	Yes	6	100	-13.2*	[-23.8, -1.2]	-9.5	[-20.2, 2.6]
	No	26	459	-5.2	[-11.2, 1.1]	-4.9	[-10.3, 0.8]
		<i>p-value interaction</i>		0.20		0.46	
Medication							
Beta-Blocker	Yes		164	1.6	[-7.6, 11.9]	-0.3	[-8.1, 8.2]
	No		395	-10.0**	[-16.0, -3.4]	-8.3*	[-14.2, -2.1]
		<i>p-value interaction</i>		0.03		0.09	
Statins	Yes		120	0.0	[-10.5, 11.8]	-0.9	[-9.7, 8.7]
	No		418	-8.4*	[-14.4, -2.0]	-7.5*	[-13.2, -1.4]
		<i>p-value interaction</i>		0.15		0.20	

⁺ p<0.10, * p<0.05, ** p<0.01

Table 7: Estimated effects of 10 µg/m³ PM_{2.5} increase on HRV measures in selected studies of elderly

Study	N Subjects	N observations	metric	SDNN		r-MSSD		HF		LF	
				% change	95% CI	% change	95% CI	% change	95% CI	% change	95% CI
Boston Elders, 1997 (Gold et al. 2000) ⁸	20	163	4 hrs	-2.8	(-6.4,0.8)	-10.0	(-15.4,-4.6)				
Baltimore PM Study, 1998 (Creason et al. 2001) ²⁴	56	up to 12 per subject	1 day					-14.9	(-25.9,-2.3)	-12.9	(-20.6,-4.5)
Vancouver 1998 (Ebelt et al. 2005) ⁴⁸	19	76	1 day	-8.0	(-23.0,6.9)	-13.4	(-42.6,15.9)				
Boston Elders, 1999 (Schwartz et al. 2005) ¹⁰	28	310	24 hrs	-2.6	(-6.0,0.8)	-10.1	(-16.9,-2.8)				
Utah, 1990-2001 (Pope et al. 2004) ²⁵	88	250	1 day	-2.7	(-3.9,-1.4)	-6.1	(-9.3,-3.0)				
Mexico City, 2002 (Holguin et al. 2003) ²⁶	34	384	1 day					-19.3	(-29.2,-8.0)	-8.4	(-19.3,4.0)
Normative Aging Study, 2000-03 (Park et al. 2005) ²⁸	497	497	24 hrs	-2.7	(-9.5,4.5)			-16.2	(-28.9,-1.2)	0.8	(-12.0,15.3)
Steubenville 2000	32	638	1 day	-4.0	(-7.0,-0.9)	-6.5	(-12.1,-0.6)	-11.4	(-21.5,-0.1)	-10.7	(-20.1,-0.3)

Figure1
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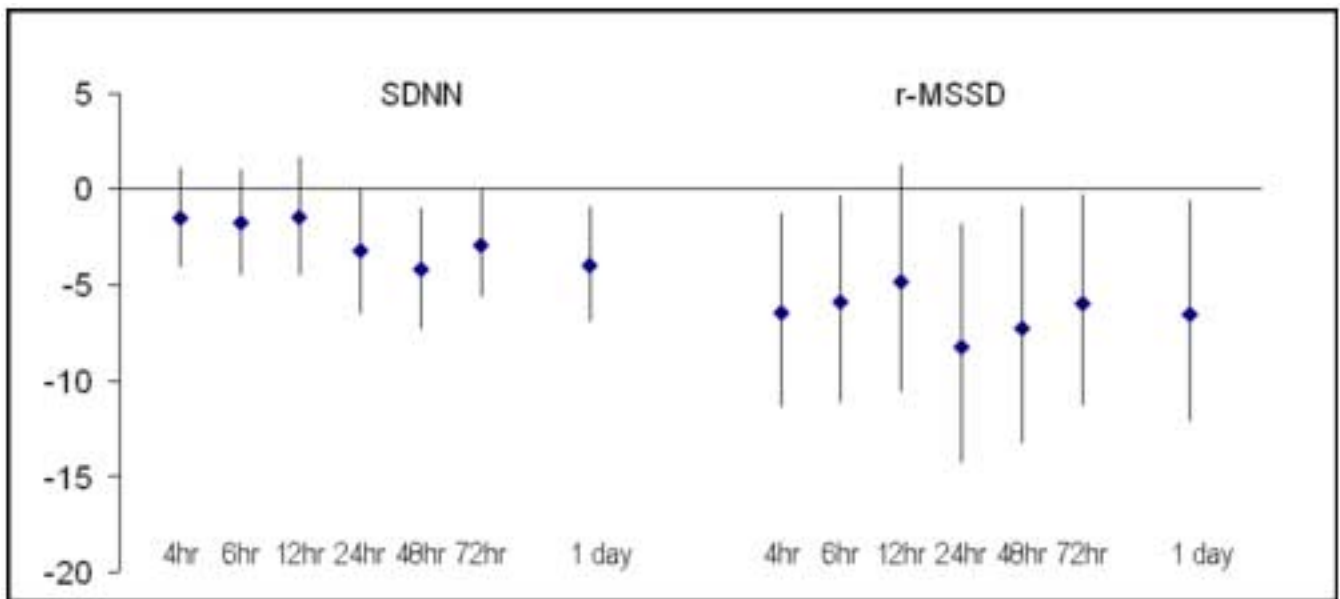


Figure2

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