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Biodegradation of Trichlorofluoromethane by Sediment Associated Anaerobic Bacteria from an Aquifer Contaminated by Landfill Leachate

by

R. L. Brigmon

Westinghouse Savannah River Company Savannah River Site Aiken, South Carolina 29808

D. J. Altman

A. J. Tien

M. M. Franck

P. C. McKinsey

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Q134 Biodegradation of Trichlorofluoromethane (CFC-11) by Sediment Associated Anaerobic Bacteria from an Aquifer Contaminated by Landfill Leachate

A.J. Tien!, C.J. Berry, M.M. Franck, P.C. McKinsey, R. L. Brigmon

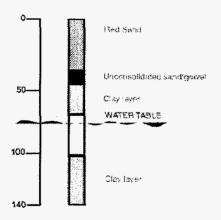
Environmental Biotechnology Section, Westinghouse Savannah River Co. LLC, Aiken SC 29808

ABSTRACT

The halogenated aliphatic hydrocarbon trichlorofluoromethane (CFC-11) has been widely used as a refrigerant and aerosol propellant. CFC-11 and related compounds such as CFC-12 (dichlorodifluoromethane) have been implicated in the depletion of stratospheric ozone and have been systematically replaced in the United States by less reactive compounds. Anaerobic groundwater contaminated by landfill leachate, anaerobic soils and sediments have been shown to transform CFC-11. Recently, CFC-11 was detected in an aquifer under a closed landfill site. In the present study, we report the results of microcosm studies using anaerobic landfill sediments capable of degrading CFC-11 with lactate as an electron donor and nitrate as an electron acceptor under acidic pH conditions characteristic to this aquifer.



TYPICAL BOREHOLE CHARACTERISTICS



SITE CHARACTERISTICS AND SAMPLING METHODS

Cores composed of Late Cretaceous and Tertiary sediments consisting of unconsolidated interbedded sands, silts, and clays, with some gravel and carbonate deposits were collected using a cone penetrometer (CPT) device at a depth of 50 ft to 70 ft below the ground surface of the landfill (i.e. Aiken Plateau portion of the Upper Atlantic Coastal Plain Province). Soil cores were encased in lexan sheaths and sealed with paraffin wax and stored at 4C.

INCUBATION CONDITIONS

Landfill samples composed of Late Cretaceous and Tertiary sediments consisting of unconsolidated interbedded sands, sitts, and clays, with some gravel and carbonate deposits were transported to the laboratory and homogenized within an anaerobic hood. These sediment samples were mixed groundwater and basal salts medium (Sonier et al. 1994) and added to 40 ml amber serum flasks. Alternate carbon sources such as lactate and acetate were evaluated. Results indicate that lacate not acetate could support freon biodegradation. Lactate was added to a final concentration of 1 mM, 2.5 mM, 5mM and 10 mM). Final pH of the medium was 4.0. Sodium azide (1% v/v) treated sets were used as controls.

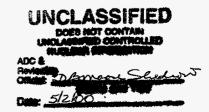
GROUNDWATER CONDITIONS	
·Sulfate Limited	
·Nitrate Limited	
•pH	3.93-4.33
·Conductivity (mS)	0.030-0.036
•Dissolved Oxygen (%)	1.98-8.94
•Temperature (F)	18.7-19.8
•Depth (ft)	65-80



Experimental Set-up under either denitrifying or sulfate reducing conditions

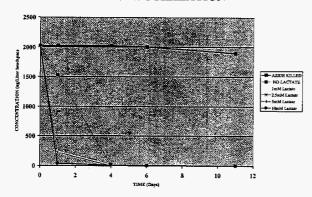
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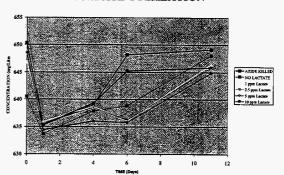


RESULTS

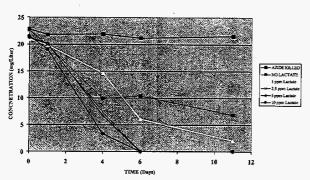
CFC-11 UTILIZATION



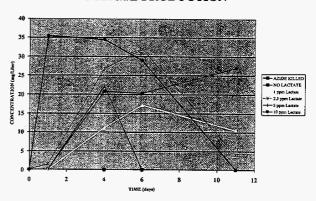
SULFATE UTILIZATION



NITRATE UTILIZATION



NITRITE PRODUCTION

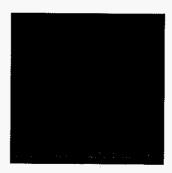


CONCLUSIONS

Denitrification rather than sulfate reduction plays a role in CFC-11 elimination from this contaminated aquifer.

CFC-11 biodegradation rates increased at lactate concentrations greater than $5\ mM$.

This study presents supporting evidence that in situ bioremediation presents a environmentally sound option for treatment of this CFC-11 contaminated site.



REFERENCES

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