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AMERICAN NATIONAL STANDARD ANSI/ANS-8.12-1987: CRITICALITY CONTROL OF PLUTONIUM-URANIUM MIXTURES

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American National Standard ANSI/ANS-8.12-1987 (Ref. 1) was approved for use on September 11, 1987. The history of the development of the standard is discussed in Ref. 2. The first version of this standard, which only included subcritical limits on homogeneous plutonium-uranium fuel mixtures, was approved July 17, 1978. The current version was revised to include limits on heterogeneous systems as well (Ref. 3). This paper provides additional information on the limits presented in the standard.

As stated in its forward, the standard "... provides guidance for the prevention of criticality accidents in the handling, storing, processing, and transporting of plutonium-uranium fuel mixtures outside reactors and is applicable to all operations involving mixtures of plutonium and natural uranium. It constitutes an extension of the American National Standard for Nuclear Criticality Safety in Operations with Fissionable Materials Outside Reactors, ANSI/ANS-8.1-1983 (Ref. 4).

The standard includes subcritical limits¹ for both homogeneous mixtures and heterogeneous lattices of plutonium and uranium. The uranium is assumed to be natural or depeleted. Limits

¹The limiting value assigned to a controlled parameter that results in a system known to be subcritical, provided the limiting value of no other controlled parameter of the system is violated. The subcritical limit allows for uncertainties in the calculations used in its derivation, but not for contingencies, e.g., double batching or failure of analytical techniques to yield accurate values of process variables.

are provided for a variety of mixtures, including aqueous solutions, dry [H/(Pu + U) = 0] mixed oxides at theoretical density, and damp $[H/(Pu + U) \le 0.45]$ mixed oxides at both theoretical and one-half theoretical density. The limits for homogeneous mixtures shown in Table I cover four different plutonium contents (3, 8, 15, and 30 wt% PuO₂ in PuO₂ + UO₂) and three different isotopic compositions (²⁴⁰Pu > ²⁴¹Pu; ²⁴⁰Pu ≥ 15 wt% and ²⁴¹Pu ≤ 6 wt%; and ²⁴⁰Pu ≥ 25 wt% and ²⁴¹Pu ≤ 15 wt%).

Heterogeneous lattice limits for mass, spherical volume, cylinder diameter, and slab thickness are also presented in the standard. Figure 1 shows one such set of limits: the limiting volume curves for heterogeneous systems. Anomalous behavior was noted for the heterogeneous cases with 30 wt% plutonium in $PuO_2 + UO_2$ with 25 wt% ²⁴⁰Pu and 15 wt% ²⁴¹Pu. For these cases, the minima calculated for the lattices were greater than the minima for dry theoretical oxide. The lattice minima occurred with very small rod diameters, and it is believed that the resonance absorption treatment employed could not adequately handle the unusual geometry (Ref 5). Thus, the standard does not extrapolate beyond 15 wt% plutonium in the mixed oxide for the isotopic mixture of 25 wt% ²⁴⁰Pu and 15 wt% ²⁴¹Pu.

The current version of the standard has now been available for 13 years. The last reaffirmation was on February 17, 1993. The standard provides useful subcritical limits for mixed-oxide systems in the range of 3 to 30 wt% PuO_2 in $PuO_2 + UO_2$. To cover interest in metal fuels (such as the uranium, plutonium, and zirconium fuel used in the integral fast reactor concept [Ref. 6]), a work group is likely to be formed to establish limits applicable to these fuel types.

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TABLE I

Subcritical Limits for Uniform Aqueous Mixtures of the Oxides of Plutonium and Natural Uranium*

PuO_2 in $(PuO_2 + UO_2)$ (wt%)													
	3	1		8			15			30 ª			
- 	Plutonium Isotopic Composition ^b												
I	II	III	· I	II	III	I	II	III	<u> </u>	II	III		
0.73	1.35	2.00	0.61	1.06	1.53	0.54	0.94	1.28	0.50	0.87	1.16		
27.5	51.3	75.9	8.6	15.1	21.7	4.1	7.1	9.7	1.9	3.3	4.4		
24.3	30.8	34.8	19.8	24.9	27.5	17.8	22.5	24.8	16.2	21.0	23.4		
11.0	14.9	17.4	8.2	11.2	12.9	6.9	9.6	11.0	5.9	8.7	9.9		
23.5	44.8	63.4	14.0	25.9	34.4	11.0	20.4	26.6	8.5	16.8	21.6		
6.8°	8.1	9.3	6.9	8.2	9.4	7.0	8.2	9.4	7.0	8.1	9.3		
257° 3780	305 3203	351 2780	97.3 3780	116 3210	134 2790	52.9 3780	61.7 3237	71.0 2818	26.5 3780	30.7 3253	35.2 2848		
0.27	0.38	0.47	0.25	0.34	0.42	0.25	0.33	0.41	0.24	0.32	0.37		
10.2	14.4	17.7	3.5	4.8	5.9	1.9	2.5	3.1	0.9	1.2	1.4		
	I 0.73 27.5 24.3 11.0 23.5 6.8 ^c 257 ^c 3780 0.27 10.2	3III0.731.3527.551.324.330.811.014.923.544.86.8°8.1257°305378032030.270.3810.214.4	IIIIII 0.73 1.35 2.00 27.5 51.3 75.9 24.3 30.8 34.8 11.0 14.9 17.4 23.5 44.8 63.4 6.8^{c} 8.1 9.3 257^{c} 305 351 3780 3203 2780 0.27 0.38 0.47 10.2 14.4 17.7	JIIIIII 0.73 1.35 2.00 0.61 27.5 51.3 75.9 8.6 24.3 30.8 34.8 19.8 11.0 14.9 17.4 8.2 23.5 44.8 63.4 14.0 6.8^{c} 8.1 9.3 6.9 257^{c} 305 351 97.3 3780 3203 2780 3780 0.27 0.38 0.47 0.25 10.2 14.4 17.7 3.5	Trice 8 Plutoni I II II I II 0.73 1.35 2.00 0.61 1.06 27.5 51.3 75.9 8.6 15.1 24.3 30.8 34.8 19.8 24.9 11.0 14.9 17.4 8.2 11.2 23.5 44.8 63.4 14.0 25.9 6.8° 8.1 9.3 6.9 8.2 257° 305 351 3780 3203 2780 0.27 0.38 0.47 0.25 0.34 10.2 14.4 17.7 3.5 4.8	Image: Second	THE 3 Ref 100_2 in $(r H O_2^{-1} + O O_2)$ (where 1 Plutonium Isotopic Composition I II III IIII IIII IIIII IIIIII IIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIII	Fully, in (Fully, + 005) (W178)3815Plutonium Isotopic Composition*IIIIIIIII0.731.352.000.611.061.530.540.9427.551.375.98.615.121.74.17.124.330.834.819.824.927.517.822.511.014.917.48.211.212.96.99.623.544.863.414.025.934.411.020.4 6.8° 8.19.36.98.29.47.08.2257^{\circ}30535197.311613452.961.7378032032780378032102790378032370.270.380.470.250.340.420.250.3310.214.417.73.54.85.91.92.5	Indep in (Fide) (W18) Is Is Plutonium Isotopic Composition ^b I II II II II II II II 0.73 1.35 2.00 0.61 1.06 1.53 0.54 0.94 1.28 27.5 51.3 75.9 8.6 15.1 21.7 4.1 7.1 9.7 24.3 30.8 34.8 19.8 24.9 27.5 17.8 22.5 24.8 11.0 14.9 17.4 8.2 11.2 12.9 6.9 9.6 11.0 23.5 44.8 63.4 14.0 25.9 34.4 11.0 20.4 26.6 6.8° 8.1 9.3 6.9 8.2 9.4 7.0 8.2 9.4 257° 305 351 97.3 116 134 52.9 61.7 71.0 3780 3203 2780 3780 32	Trick of the formation of the formatio	The probability in (FR0, III (FR0, IIII (FR0, IIII (FR0, IIII (FR0, IIII (FR0, IIII (FR0, IIII (FR0, IIIIIIIIII)))))))))))))))))))))))))))		

*All values are upper limits except atomic ratios, which are lower limits.

^aDimensional and volume limits do not apply for isotopic compositions II and III unless, for II, the concentration of oxides is < 5700 g/l and, for III, < 4500 g/l.

^bPlutonium isotopic composition: I -- ²⁴⁰Pu > ²⁴¹Pu, II -- ²⁴⁰Pu ≥ 15 wt% and ²⁴¹Pu ≤ 6 wt%, and III -- ²⁴⁰Pu ≥ 25 wt% and ²⁴¹Pu ≤ 15 wt%.

^cThis concentration limit is not applicable to oxide mixtures in which the $PuO_2/(PuO_2 + UO_2)$ ratio is < 3 wt% because of the increased relative importance of ²³⁵U in high-uranium-bearing materials.



Fig. 1. Limiting volume of heterogeneous mixtures of the oxides of plutonium and natural uranium in water as a function of the plutonium oxide content.