SAND99-0125J

Band Structure of InGaAsN Alloys and Effects of Pressure

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(December 22, 1998)

Abstract

InGaAsN is a new semiconductor alloy system with the remarkable property that the inclusion of only 2% nitrogen reduces the bandgap by more than 30%. In order to help understand the physical origin of this extreme deviation from the typically observed nearly linear dependence of alloy properties on concentration, we have investigated the pressure dependence of the excited state energies using both experimental and theoretical methods. We report measurements of the low temperature photoluminescence of the material for pressures between ambient and 110 kbar. We also describe a simple, densityfunctional-theory-based approach to calculating the pressure dependence of low lying excitation energies for low concentration alloys. The theoretically predicted pressure dependence of the bandgap is in excellent agreement with the experimental data. Based on the results of our calculations, we suggest an explanation for the strongly non-linear pressure dependence of the bandgap that, surprisingly, does not involve a nitrogen impurity state.

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A new semiconductor alloy system, InGaAsN, has been identified as a key candidate material for high-efficiency multi-junction solar cells [1,2] and also for long wavelength laser systems [3,4]. The introduction of small amounts of nitrogen ($\approx 2\%$) in GaAs greatly reduces [5,6] the bandgap energy, with reductions approaching 0.5 eV! With the appropriate ratio of indium to nitrogen concentrations, InGaAsN can be lattice matched to GaAs. Lattice matching allows the design of multi-junction solar cells without the inherent problems found in strained cells. Of prime importance is the role of the nitrogen isoelectronic atom: (1) What is the origin of the large bandgap reduction and (2) Are the states extended (bandlike) or localized (impurity-like)? In order to answer these questions, a better understanding of the electronic properties of this type of alloy system is required.

In the past, both first-principles [7] and empirical [8] theoretical treatments for this material system have concentrated on understanding the dependence of the bandgap energy on nitrogen composition. In this paper we present pressure dependent photoluminescence (PL) data together with a first principles local density approximation (LDA) calculation for the band structure and its pressure dependence. It will be shown that, while it is well known that bandgap energies calculated by the LDA method are not accurate, the predicted pressure dependence of the bandgap energy is in excellent agreement with experiment. Because of this good agreement, we have confidence that this technique could be useful for other low concentration alloy systems.

The structures were grown in a vertical flow, high speed rotating disk, EMCORE GS/3200 metalorganic chemical vapor deposition reactor. The $\ln_x Ga_{1-x}As_{1-y}N_y$ films were grown using trimethylindium (TMIn), trimethylgallium (TMG), 100% arsine and dimethylhydrazine (DMHy). Dimethylhydrazine was used as the nitrogen source since it has a lower disassociation temperature than ammonia and has a vapor pressure of approximately 110 torr at 18°C. Unintentionally doped InGaAsN was p-type. InGaAsN films for Hall and optical measurements were grown on semi-insulating GaAs orientated 2° off (100) towards (110). Lattice matched ($\delta a/a < 8 \times 10^{-4}$) films were grown at 600°C and 60 torr using a V/III ratio of 97, a DMHy/V ratio of 0.97 and a TMIn/III ratio of 0.12. The growth rate was 10Å/s.

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Portions of this document may be illegible in electronic image products. Images are produced from the best available original document. These conditions resulted in films with an indium mole fraction of 0.07 ± 0.005 and a nitrogen mole fraction of 0.022 ± 0.003 . The composition was determined by calibration growths of GaAsN and InGaAs along with double crystal x-ray diffraction measurements. The nitrogen composition of bulk films was also confirmed from elastic recoil detection measurements. A significant increase in photoluminescence intensity was observed from these films following a post-growth anneal. Ex-situ, post-growth anneals were carried out in a rapid thermal anneal system under nitrogen using a sacrificial GaAs wafer in close proximity to the InGaAsN sample. The photoluminescence intensity was a maximum for samples annealed either 700°C for 2 minutes or at 650°C for 30 minutes. Similiar observations have been reported by Rao *et.al.* [13] Transmission electron microscopy measurements indicate that the samples are random and no evidence for clustering or phase separation was observed. [9]

The pressure was generated using a small BeCu piston-cylinder diamond anvil cell, 8.75mm-diameter and 12.5-mm-height [10]. Methanol, ethanol, and water in a ratio of 16:3:1 was used for the pressure medium [11]. The shift in the fluorescence of a small chip of ruby placed in the pressurized volume was used to calibrate the pressure at 4K with an accuracy of ± 0.5 kbar [12]. A single 600- μ m-diameter optical fiber, butted up against one of the diamonds, brought the 1 mW of 5145-nm-wavelength laser power to the sample and also collected the PL signal from the sample. A beam splitter system was used to direct the PL signal to an optical monochromator. Depending on the bandgap energy, two liquid-nitrogen-cooled detectors were used to detect the PL signal. For low pressure regimes, where the bandgap energies were near or below 1 eV, a NORTH-COAST EO-817L Ge-detector was employed, while at higher pressures, a standard CCD array was used. A typical low temperature (4K) PL spectrum is shown in the inset of Fig. 1. As can be seen, the 4-K bandgap energy is near 1.15 eV. The pressure dependence of the bandgap energy shift data, as measured by the peak of the PL spectrum, is shown in Fig. 1 as solid circles. The pressure data ranged between ambient and 110 kbar. The solid curve drawn through the data is discussed below.

In order to model the InGaAsN system, we used the Vienna *Ab initio* Simulation Package (VASP) [14] to perform first-principles electronic structure calculations based on the Kohn-

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Sham density functional theory (DFT) with plane wave basis sets, ultrasoft pseudopotentials [15], and the local density approximation for the exchange-correlation functional. In construction of the pseudopotentials, the Ga 3d electrons were treated as valence electrons in order to accurately represent any effects of a near resonance with the N 2s level that has been observed in GaN [16]. We modeled the InGaAsN system in the experimentally relevant concentration range using a series of supercells of the zincblende GaAs structure each with a single As replaced by a N. The lattice constants of the supercells were varied to simulate the effects of pressure, and for each cell, the ionic positions were relaxed using first-principles forces until the residual forces were less than 20 meV per Å. In all cases, we found that the nitrogen atom remained in the symmetric position during relaxation. In order to compare to experimental data, which is taken as a function of pressure, an *ab initio* calculation of the system pressure was performed. We have investigated supercells with the following stoichiometries: Ga₃₂As₃₁N, Ga₆₄As₆₃N, Ga₁₀₈As₁₀₇N, and Ga₁₂₈As₁₂₇N. These cells correspond to nitrogen concentrations of 3.13, 1.56, 0.93 and 0.78%, with the nitrogen atoms ordered in simple cubic, fcc, bcc, and simple cubic lattices respectively. In contrast, the nitrogen atoms in the experimental system are believed to be nearly randomly distributed [9]. However, we found that calculated band structures of our supercells were qualitatively similar despite their differing symmetries, indicating that that the N atoms interact weakly with each other at these low concentrations. Therefore, we believe that our artificially ordered supercells provide an adequate model of the near-band-gap electronic structure of the disordered experimental system. Likewise, we have ignored the presence of In in the experimental system (except for indirect effects due to the change in lattice constant, as will be discussed below). This is justified since experimental studies of InGaAs alloys indicate that the low concentration of indium found in the experimental InGaAsN system has a small effect on the electronic properties [3].

Figure 2 shows a representative band structure for the 3.13% system. It should be noted that the band structure is plotted with respect to the Brillouin zone of a 64-atom cell. Since the nitrogen substitution breaks the symmetry of the underlying zincblende structure,

there is no uniquely defined way to "unfold" the band structure into the Brillouin zone of the primitive 2-atom zincblende unit cell. The high symmetry points of the primitive GaAs cell fold into the Γ -point of the 64-atom cell, and therefore in the presence of a real symmetry breaking term (such as produced by nitrogen substitution), we can expect interaction between the resulting levels. The valence band and the conduction band are indicated with a heavy solid line. The conduction band is well separated from the other bands throughout most of the Brillouin zone, and it is quite dispersive with more than 1 eV of band width. Likewise, the bands above the conduction band show a substantial amount of dispersion, and there is no evidence of a flat impurity-like band anywhere above the conduction band. The absense of a nitrogen derived impurity-like state is supported by a decomposition of the wave-functions in terms of atomic-like orbitals, which shows that the conduction band has about 5% of its weight on the nitrogen atom, which is by far the highest fraction of any of the bands above the gap. The calculated bandgap is only 0.12 eV, while the experimental bandgap is of order 1 eV for this concentration of N. This large error in the bandgap is a well known problem of the LDA. A central result of this paper is that despite this large error in the absolute magnitude of the bandgap, the *change* in bandgap with lattice constant is in excellent agreement with experiment (see Fig. 1).

Figure 1 compares experimental data to results of our theoretical model. The agreement between theory and experiment is excellent. In order to make a meaningful comparison, some nontrivial analysis of the theoretical calculations is required. The basic principle of this analysis is to treat the lattice constant and N concentration as independent variables, while the bandgap and pressure are treated as dependent variables. The dependent variables are then shifted to remove known LDA errors. In order to obtain results applicable to the 2.0% experimental nitrogen concentration, the bandgap and the pressure are linearly interpolated between the results of 128-atom (1.56% N) and 64-atom (3.13% N) supercells for each lattice constant. Then, results obtained at the experimental lattice constant of GaAs are taken as the reference (assumed to correspond to the experimental zero of pressure), and we plot the change in bandgap against the change in pressure. This procedure compensates for two well known errors of the LDA: (1) The bandgap is severely underestimated, as discussed above. (2) The lattice constant is underestimated by about 1%. The second error may seem to be insignificant compared to the errors in LDA results for some other quantities, but it corresponds to about a 20 kbar error in pressure, which is significant on an experimental scale. Since the experimental GaAs lattice constant is used, this procedure also helps to implicitly compensate for neglecting the In, which is added to the experimental system to match the lattice constant to that of GaAs.

The approach desribed here could prove useful for similar systems. However, we believe that a reason for the success of our approach is that, with the exception of the largest N concentration at the largest lattice constant, all of our model systems have a positive bandgap. Previous first-principles calculations for the GaAsN system [7] have used high nitrogen concentrations in order to obtain smaller model systems, which are computationally undemanding. At these higher concentrations, the LDA bandgap error is so severe that computated band structures do not have a bandgap. This artificial bandgap collapse leads to unphysical occupations of the electronic states (i.e. conduction-band-like states are occupied, while valence-band-like states are unoccupied), producing a significant error in the electronic charge density. In this regime of strongly negative LDA bandgaps, we are not confident that our simple LDA-based approach to computing the pressure dependence of excited state energies could be applied fruitfully.

A remarkable feature of Fig. 1 is the strongly non-linear dependence of the gap on the pressure. This is in marked contrast to the parent GaAs material and provides additional evidence, beyond the large reduction in the bandgap, that a few percent of nitrogen is producing remarkable changes in the material. In order to understand this nonlinearity, it is necessary to study additional bands above the conduction band. Figure 3 shows the theoretical pressure dependence of the Γ -point energies of several additional bands treated with the same analysis that was used for the conduction band in Fig. 1. Eight energy bands of the system are shown, but these fall into five degenerate groups. The figure shows three singlets (filled circles), a doublet (open squares), and a triplet (open triangles).

A common origin of nonlinear behavior of energy levels as a function of a parameter (such as pressure) is band replusion. Band repulsion results from the mixing (hybridization) of bands in the same representation of the crystal symmetry group in such a way that level crossings are replaced with non-intersecting horizontal curves separated by a gap-like region. In Fig. 3, the highest singlet on the left and the triplet on the upper right bend downward due to repulsion from higher energy bands that have been omitted from the figure in order to improve its clarity. Whether the upper two singlets cross or repel at about 80 kbar can not be determined from the limited number of points that we have calculated, but if they repel, the effect is not very strong. We have chosen to show the bands as crossing in order to aid the eye in following their relationship with the doublet and the triplet. These groups of bands were chosen by comparing our results for various nitrogen concentrations with results for pure GaAs with the appropriately folded Brillouin zone. This comparison also allows useful, but non-rigorous, assignments of the low energy GaAsN bands at the Γ -point to special points of the primitive 2-atom GaAs Brillouin zone that are folded into the Γ -point. For the bands shown in Fig. 3, we propose the following assignments: (1) The bottom singlet corresponds to the Γ -point of the fundamental call. (2) The second singlet on the left hand side (the third singlet on the right) plus the triplet correspond to a split quartet formed from the 4 L-points (111), $(\overline{111})$, $(\overline{111})$, and $(\overline{111})$. (3) The remaining singlet plus the doublet correspond to a split triplet consisting of the three X-points (100), (010), and (001). In regions of band repulsion, for example, the character of different bands becomes mixed, and thus these assignments should not be taken too literally. However, we feel that they provide useful labels and help in interpreting the data.

With one exception, all of the bands within a few eV of the gap are observed to regain the degeneracies of pure GaAs to within of a few hundredths of an eV by the time our largest cell (0.78% N concentration) is reached. The exception is the L derived singlet, which remains split off from the triplet by about 0.1 eV. This suggests that this singlet may evolve into the impurity state observed at very low N concentrations [17,18]. However, as mentioned above, this state does not act like an impurity state at the technologically interesting con-

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centrations around 2%. Furthermore, the L derived singlet rises faster than the conduction band throughout the studied pressure range, and we do not see the upward curvature that would be expected if it was repelled by the conduction band. Therefore, we believe that repulsion between the L derived singlet and the conduction band contributes at most a small amount to the nonlinearity of the conduction band. In contrast, Fig. 3 demonstrates almost textbook repulsion between the X derived singlet and the conduction band at pressures over 100 kbar, and it is likely that this repulsion continues to lower pressures, even though the effect is obscured by the additional repulsion between the X derived singlet and a higher state. Based on these observations, we propose that repulsion from the X derived singlet is the chief cause of the experimentally observed nonlinear dependence of the bandgap on pressure. Based on the above, we conclude that the nonlinear dependence of the bandgap on pressure does not result from localized nitrogen states. Finally, we note that our results are consistent with recent reflectance measurements which show some of the predicted higher energy states [19,20].

This work was supported in part by the Division of Material Science, Office of Basic Energy Science, U. S. DOE, No. DE-AC04-94AL85000, the National High Magnetic Field Laboratory, which is supported by NSF Cooperative Agreement No. DMR-9016241 and by the State of Florida, and by a grant of HPC time from the DoD HPC ARL and CEWES Major Shared Resource Centers. The *ab initio* total-energy and molecular-dynamics package VASP (Vienna *Ab initio* Simulation Package) and the corresponding ultrasoft pseudopotential database were developed at the Institute für Theoretische Physik of the Technische Universät Wien.

> Sandia is a multiprogram laboratory operated by Sandia Corporation, a Lockheed Martin Company, for the United States Department of Energy under contract DE-AC04-94AL85000.

REFERENCES

- Sarah R. Kurtz, D. Myers, and J. M. Olson, in Proc. 26th IEEE Photovoltaics Spec. Conf. (IEEE, New York, NY, 1997) p. 875.
- [2] Steven R. Kurtz, A. A. Allerman, E. D. Jones, J. M. Gee, J. J. Banas, and B. E. Hammons, (Accepted for Publication, Appl. Phys. Lett.).
- [3] M. Kondow, K. Uomi, A. Niwa, T. Kitatani, S. Watahiki, and Y. Yazawa, Jpn. J. Appl. Phys. 35, 1273 (1996); M. Kondow, T. Kitatani, S. Nakatsuka, M. C. Larson, K. Nakahara, Y. Yazawa, M. Okai, and K. Uomi, IEEE J. Selected Topics in Quantum Electronics 3, 719 (1997).
- [4] T. Miyamoto, K. Takeuchi, F. Koyama, and K. Iga, IEEE Photonics Tech. Lett. 9 1448 (1997)
- [5] W. G. Bi and C. W. Tu, J. Appl. Phys. 80, 1934 (1996); W. G. Bi and C. W. Tu, Appl. Phys. Lett. 72, 1161 (1998); H. P. Xin and C. W. Tu, Appl. Phys. Lett. 72, 2442 (1998)
- [6] L. Malikova, F. H. Pollak, and R. Bhat J. Electronic Materials 27, 484 (1998)
- [7] A. Rubio and M. L. Cohen, Phys. Rev. B 51, 4343 (1995); J. Neugebauer and C. G. Van de Walle, Phys. Rev. B 51, 10568 (1995); S.-H. Wei and A. Zunger, Phys. Rev. Lett., 76, 664 (1996).
- [8] L. Bellaiche, S.-H. Wei, and A. Zunger, Phys. Rev. B 54, 17568 (1996); Appl. Phys. Lett. 70, 3558 (1997); Phys. Rev. B 56, 10233 (1997); L. Bellaiche and A. Zunger, Phys. Rev. B 57, 4425 (1998).
- [9] D. M. Follstaedt (Private Communication).
- [10] S.W. Tozer (Private Communication).
- [11] G. J. Piermarini, S. Block, J. D. Barnett, J. Appl. Phys. 44, 5377 (1973).

- [12] R. A. Forman, G. J. Piermarini, J. D. Barnett, S. Block, Science 176, 284 (1972).
- [13] E. V. K. Rao, A. Ougazzaden, Y. Le Bellego, and M. Juhel, Appl. Phys. Lett. 72, 1409 (1998).
- [14] G. Kresse and J. Hafner, Phys. Rev. B 47, 558 (1993); G. Kresse and J. Hafner, Phys.
 Rev. B 49, 14251 (1994); G. Kresse and J. Furthmüller, Comput. Mat. Sci. 6, 15 (1996);
 G. Kresse and J. Furthmüller, Phys. Rev. B 54, 11169 (1996).
- [15] D. Vanderbilt, Phys. Rev. B 41, 7892 (1990).
- [16] V. Fiorentini, M. Methfessel, and M. Scheffler, Phys. Rev. B 47, 13353 (1993).
- [17] D. J. Wolflord, J. A. Bradley, K. Fry, J. Thompson, in *Physics of Semiconductors*, ed. J. D. Chadi and W. A. Harrison (Springer, New York, 1984) p. 627; D. J. Wolflord, J. A. Bradley, K. Fry, J. Thompson, and H. E. King, in *GaAs and Related Compounds*, Inst. Phys. Conf. Ser. No. 65 (IOP, Bristol & Philadelphia) p. 477.
- [18] X. Liu, M. -E. Pistol, L. Samuelson, and W. Seifert, Appl. Phys. Lett. 56, 1451 (1990).
- [19] J. D. Perkins, A. Mascarenhas, Y. Zhang, J. F. Geisz, D. J. Friedman, J. M. Olson, and S. R. Kurtz (Private Communication).
- [20] W. Shan, W. Walukiewicz, and J. W. Ager III (Private Communication).



(solid line) dependence of the bandgap energy shift versus pressure at 4 K for 2% nitrogen in InGaAsN. The inset shows the 4-K ambient pressure photoluminescence spectrum.

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tion calculation for the band structure for 3.3% nitrogen InGaAsN alloy. The solid lines, top to bottom, are respectively the conduction and valence-band states.



