STUDIES OF FERROELECTRIC HETEROSTRUCTURE THIN FILMS AND INTERFACES VIA IN SITU ANALYTICAL TECHNIQUES*

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The science and technology of ferroelectric thin films has experienced an explosive development during the last ten years. Low-density non-volatile ferroelectric random access memories (NVFRAMs) are now incorporated in commercial products such as "smart cards", while high permittivity capacitors are incorporated in cellular phones. However, substantial work is still needed to develop materials integration strategies for high-density memories. We have demonstrated that the implementation of complementary *in situ* characterization techniques is critical to understand film growth and interface processes, which play critical roles in film microstructures and properties.

We are using uniquely integrated time of flight ion scattering and recoil spectroscopy (TOF-ISARS) and spectroscopic ellipsometry (SE) techniques to perform in situ, real-time studies of film growth processes in the high background gas pressure required to growth ferroelectric thin films. TOF-ISARS provides information on surface processes, while SE permits the investigation of buried interfaces as they are being formed. Recent studies on SrBi₂Ta₂O₉ (SBT) and Ba_xSr_{1-x}TiO₃ (BST) film growth and interface processes are discussed.

Keywords: in situ characterization, ferroelectric films, ion scattering, recoil spectroscopy, spectroscopic ellipsometry, SFM piezoresponse, X-ray scattering

INTRODUCTION

Ferroelectric and high dielectric constant (k) thin films, relevant to non-volatile ferroelectric random access memories (NVFRAMs) and high (k) dynamic random access memories (DRAMs), respectively, will be very thin (< 100 nm) for the next generation of high density, low voltage devices. In addition, as the heterostructure ferroelectric capacitors needed for these memories and

processing conditions become increasingly more complex, there is a growing need for *in situ*, real-time, surface-specific analytical tools to characterize phenomena occurring at the surface of the growing films and heterostructure interfaces. These analytical tools must not be destructive, should provide a wide range of surface compositional and structural information on a time scale commensurate with the deposition rate, and must be compatible with the geometric constraints of the deposition process and the temperatures and ambient gas pressures required by the thin film growth environment.

We are focusing our research on investigating two materials that play critical roles in NVFRAMs [SrBi₂Ta₂O₉ and DRAMs and high frequency

devices [Ba_xSr_{1-x}TiO₃ (BST)].

We have demonstrated that an appropriate combination of three low-keV ion beam spectroscopies, x-ray photoelectron spectroscopy, and/or spectroscopic ellipsometry can be used to provide an exceptionally wide range of surface and interface compositional, structural and chemical information related to the synthesis of ferroelectric films. In addition, we have demonstrated that an appropriate combination of *in situ* or *in situ*, real-time characterization techniques can provide valuable information to aid in the understanding of polarization dynamics critical to the operation of ferroelectric thin film-based devices. This review provides information related to recent work that demonstrate the power of using *in situ* and/or *in situ*, real-time characterization techniques to elucidate film growth and surface and interface related processes that are critical to the development of materials integration strategies for the present and next generation of ferroelectric thin film-based devices.

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Background on TOF-ISARS

The fundamental and experimental bases of the complementary time-of-flight ion scattering and recoil spectroscopy techniques (TOF-ISARS) discussed in this paper have been described in detail in our prior publications ^[1, 2]. Briefly, an ion beam with a kinetic energy of several keV is incident upon the surface to be studied. At these kinetic energies, the collision kinematics are essentially those of classical two body elastic collisions.

In low energy ion scattering spectroscopy (ISS), the mass of a surface atom is determined by measuring the kinetic energy loss of a primary ion of known mass, which scatters from a single surface atom. ISS is a well-established surface analysis technique [3, 4], and is the most surface-specific of the surface analytical methods, with a depth sensitivity, which can be limited to one atomic layer.

In our system, the kinetic energy of the backscattered primary particle is measured by pulsing the beam and using a time of flight (ToF) detection scheme. A typical analysis beam dose for ToF detection is ~10¹¹-10¹² ions/cm², removing or displacing 1 part in 10³-10⁴ of the near-surface atoms and making the ToF scheme essentially non-destructive [1, 5].

Direct Recoil Spectroscopy (DRS) has been in use in a number of laboratories for several years [6-8], and benefited from recent improvements in

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time of flight instrumentation^[9]. DRS employs relatively grazing entrance and exit angles to eject surface atoms via a single collision event with the primary ion. The ejected surface atoms are detected by a particle counter such as a channel plate multiplier, and the beam forming and detection hardware are very similar to that of ISS. The principal advantage of DRS compared with ISS is the ability of the former method to detect all atomic species, even those, which are lighter than the primary ion, including hydrogen.

Mass Spectroscopy of Recoiled Ions (MSRI)^[2, 5, 10] is a variation of DRS in which only the ion fraction of the direct recoil spectrum is detected. It is similar to secondary ion mass spectroscopy (SIMS) except that the DRS geometry is chosen to emphasize single collision ejection events, rather than the multiple collision cascade mechanism associated with the SIMS process. Consequently, the kinetic energy and ion-fraction of the ejected surface atoms

are much higher than for SIMS.

The TOF-ISARS techniques are based on classical collision theory discussed in prior publications ^[1, 2, 10]. A schematic of the fundamental principles is shown in Fig. 1, and a detailed discussion can be found in prior publications ^[2, 10] and in these proceedings ^[11]. Briefly, the chamber is split into two sections. Auger electron spectroscopy (AES) and x-ray photoelectron spectroscopy (XPS) are included in the upper chamber to perform immediate post-deposition surface analysis as a complement to the TOF-ISARS studies of growth processes. Thin film deposition is accomplished in the lower chamber by means of a 3-cm Kaufmann ion gun which is focused onto a target mounted on a rotating carousel with multiple targets. The use of differential pumping methods to perform TOF-ISARS characterization of film growth processes in high-pressure environments was described in one of our prior papers ^[12].

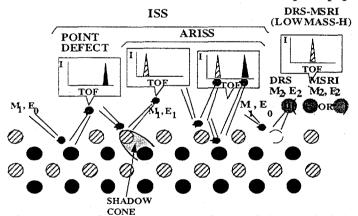


FIGURE 1. Schematic of the fundamental principles underlying the TOF-ISARS techniques: ISS; including angular-resolved ion scattering spectroscopy (ARISS), which provides information about surface structure; DRS; and MSRI.

The combined use of ISS and DRS techniques described above provides the means for measuring the surface composition ^[7] with excellent sensitivity for trace elements ^[1], including hydrogen ^[13], and determination of the local surface geometry ^[14] and chemical phase. It is also possible to distinguish layer- by –layer growth in thin films from 2D and 3D island formation^[15]. It is also

possible to obtain information on surface disorder, including the presence of point defects and surface phonon dispersion, under special analysis conditions.

Although the experimental work described in this chapter is focused on multicomponent oxide ferroelectric thin films and heterostructures, the TOF-ISARS technique has also been successfully used to study surface processes of materials such as high temperature superconductors [16], semiconductors [16] and diamond [17].

Studies of Film Growth and Interface Processes Relevant to Ferroelectric Capacitors

Due to space constrains, we can only present selected examples of TOF-ISARS studies performed in the last five years. Data related to the growth of SBT and BST films is presented and discussed as a means of illustrating previously discussed capabilities of the methods to study ferroelectric film growth and surface and interface processes.

In Situ Analysis of SBT/Bottom Electrode Interfaces and SBT Initial Growth Stages

Recently, there has been extensive interest in the synthesis of ferroelectric layered perovskite thin films such as SrBi₂Ta₂O₉ (SBT) for application to NVFRAMs. Pt/SBT/Pt capacitors show practically no polarization fatigue, little imprint, and memory-compatible retention^[18]. Platinum is simple to deposit and has a much lower electrical resistivity than oxide electrodes needed for fatiguefree PZT-based capacitors[18], resulting in short RC time constants and high device speed. The ability to use Pt electrodes and the properties already demonstrated for Pt/SBT/Pt capacitors make them a favored choice for the first generation of commercial NVFRAMs. Because of the high crystallization temperature of SBT films (> 700 °C), integration of SBT- based capacitors with Si based technology requires a stable bottom electrode / barrier heterostructure. Pt/Ti/SiO₂/Si has been widely used as a bottom electrode heterostructure for ferroelectric capacitors. The Ti layer enhances the adhesion of Pt on SiO₂. Prior studies on the high temperature stability of Pt/Ti layers in an oxygen environment revealed that Ti diffuses into the Pt layer along grain boundaries during heating [19.20]. If the Ti segregates to the Pt surface and is incorporated into the SBT film, it may have an adverse affect on composition, microstructure, and initial stages of film growth, which affect the properties of the SBT-based capacitors. In fact, it has been reported that the electrical properties of SBT capacitors are degraded by the diffused Ti through Pt electrodes[21].

Conventional analytical techniques have not been able to clearly describe the state of the Pt surface after heating, or explain how the Pt surface condition (on the monolayer scale) affects the growth of the subsequent SBT film. For example conventional Rutherford Backscattering Spectroscopy (RBS) was used to study the near-surface composition of the Pt/Ti/SiO₂/Si heterostructure substrate during heating to 700 °C in vacuum. Figure 2 shows the RBS spectra of the Pt/Ti/SiO₂/Si layered heterostructure before and after heating to 700 °C. The spectra clearly show the broadening of the Ti peak, indicating diffusion of

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Ti into the Pt layer. However, the depth resolution of RBS is insufficient to disclose whether Ti appears at the surface of the Pt layer.

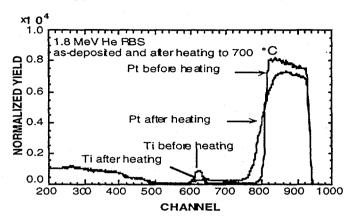


FIGURE 2. RBS spectrum for a Pt/Ti electrode before and after 700 °C anneal.

MSRI was used to obtain a clearer picture of surface segregation phenomena than that provided by RBS. As shown in Fig. 3, the surface composition in vacuum at room temperature is dominated by H, C, O, Pt, Ar (backscattered primary ions), and a small amount of Ti. Currently, the MSRI peak intensity cannot be correlated to the surface concentration. In addition, because of its relative chemical inertness and high mass, the sensitivity for Pt is lower than that for Ti, Si, and H. However, as the temperature increases (Fig. 3(b)) and 3(c)), the Pt signal decreases and nearly disappears, while at the same time the Ti signal increases significantly and a Si peak appears. From the decrease in the Pt signal, it may be estimated that at least 80% of the Pt surface consists of segregated Ti and Si due to heating at 700 °C.

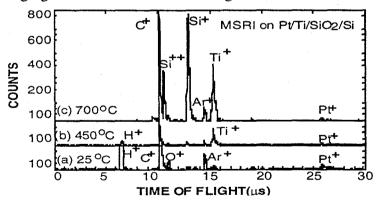


FIGURE 3. Segregation processes in the Pt/Ti/SiO₂/Si heterostructure bottom electrode system heated in vacuum, as observed using MSRI analysis.

When the Pt/Ti/SiO₂/Si heterostructure is heated in $5x10^{-4}$ Torr of oxygen, Ti appears on the Pt surface at ~ 400 °C, but very little Si appears even

at 700 °C. We attributed this effect to the formation of a TiO_x layer at the Ti/Si interface as a result of oxygen diffusion via Pt grain boundaries. The TiO_x layer provides a good diffusion barrier for Si^[20]. To test this hypothesis, a TiO₂ film was grown on a SiO₂/Si substrate and subsequently covered with a Pt layer. The Pt/TiO₂/SiO₂/Si structure was subsequently heated in 5x10⁻⁴ Torr of oxygen. The MSRI spectra in Fig. 4 show that there is neither Ti nor Si segregation to the Pt surface up to 700 °C, indicating the stability of the Pt/TiO₂ electrode, as shown also by other groups^[20,22].

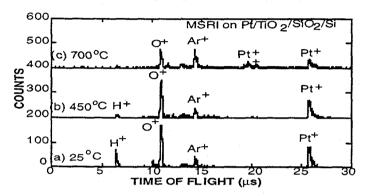


FIGURE 4. MSRI analysis of a Pt /TiO₂/SiO₂/Si heterostructure during heating to 700 °C in 5x10⁻⁴ Torr of oxygen.

We also investigated the thermal stability of Pt/Ta/SiO₂/Si, Ir/SiO₂/Si and RuO₂/SiO₂ heterostructures, in order to see if Ta, Ir, and RuO₂ provide more stable adhesion-diffusion barrier layers than Ti for the integration of Pt with Si. Additionally, Ta is not an impurity for SBT films even if Ta diffuses to the Pt surface during heating, although excess Ta may affect the stoichiometry and properties of the SBT films. The use of Ta may facilitate elimination of the oxidation processing step required to stabilize the Ti layer in the Pt/Ti/SiO₂/Si heterostructure. MSRI analysis showed that Ta diffuses to the surface of Pt at 700 °C in an oxygen environment, but no Si migration to the Pt surface was observed (Fig. 5(a). In the cases of the Ir/SiO₂/Si (Fig. 5(b) and RuO₂/SiO₂ (Fig. 5)c) heterostructures, also no Si diffusion to the Pt surface was observed even if heated up to 700 °C in oxygen. Fig 5 reveals that the Ta, Ir, and RuObased heterostructure electrodes described above are stable and therefore can provide good alternatives for the integration of SBE thin films with Si for the fabrication of NVFRAMs. However, further work, particularly involving electrical characterization of SBT capacitors using these electrodes, is necessary to evaluate their suitability for integration into NVFRAMs.

Studies of the Initial Stages of SBT Film Growth via Physical Vapor-Deposition

Since it has been reported that Ti degrades the electrical properties of SBT based capacitors^[21], we studied the initial growth of SBT films on the Pt/Ti and Pt/TiO₂/SiO₂/Si electrode layers discussed above. These experiments

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demonstrated that for SBT films sputter-deposited on Pt/Ti electrodes in 5x10⁻⁴ Torr of oxygen at 700 °C, there is negligible incorporation of Bi, in the first 20 nm of SBT, and significant segregation of Ti and Si at the Pt-SBT interface. Once the Ti and Si species have been covered by Sr and Ta atoms of the growing film, Bi begins to incorporate and the SBT film composition becomes closer to stoichiometric. On the other hand, when SBT is deposited at 700 °C in 5x10⁻⁴ Torr of oxygen, on a Pt/TiO₂/SiO₂/Si heterostructure. MSRI analysis reveals that there is neither Ti nor Si segregation on the surface of the Pt layer, and Bi is efficiently incorporated in all stages of the growing SBT film. The effect of Ti and Si surface species on Bi incorporation can be explained by the fact that both Ti and Si have larger (negative) magnitude of oxide formation energy than Bi^[23,24]. Due to this difference, the presence of Ti and Si at the surface of the Pt layer can reduce bismuth oxide to Bi, which is volatile at 700 °C. This hypothesis was confirmed by investigating the growth of SBT films directly on pure Ti and Si surfaces at 700 °C in oxygen. Under these conditions, there was practically no Bi incorporation during film growth. On the other hand, we demonstrated that there is an efficient Bi incorporation from the beginning of SBT growth on Pt/MgO substrates where there is no possibility that Ti or Si would appear on the Pt surface^[23].

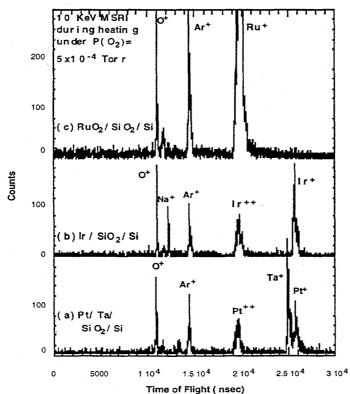


FIGURE 5. MSRI spectra from the analysis of Pt /Ta/SiO₂/Si, Ir/SiO₂/Si, RuO₂/SiO₂/Si heterostructures during heating up to 700 °C in oxygen ($P(O_2)$ = $5x10^{-4}$ Torr); the spectra were obtained using a 10 keV Ar⁺ primary ion beam.

We also investigated the growth of sputter-deposited SBT films on Pt/Ta/SiO₂/Si, Ir/SiO₂/Si, and RuO₂/SiO₂/Si at 700 °C in oxygen (P(O₂)= 5x10⁻⁴). MSRI analysis revealed that Bi is efficiently incorporated on RuO₂ and Ir surfaces at all stages of the film growth, while the incorporation on Pt/Ta electrodes is intermediate between that observed for Pt/Ti and that for Ir and RuO₂ surfaces. Based on thermodynamic considerations, our interpretation is that RuO₂ and Ir do not compete for oxygen with Bi species as effectively as Ti and Si species do, while Ta which has a smaller affinity for oxygen than Ti, is less effective than Ti in competing with Bi for the available oxygen. This analysis correlates with our prior work, which demonstrated that RuO₂ films are oxygen terminated^[25].

COMPLEMENTARY IN SITU TOF-ISARS / SPECTROSCOPIC ELLIPSOMETRY (SE) STUDIES OF FILM SURFACE AND BURIED INTERFACES

In this section, we discuss the synthesis of Ba_xSr_{1-x}TiO₃ (BST) as the vehicle to demonstrate the power of complementary in situ TOF-ISARS/SE characterization of film growth and interface processes. Research on BaxSr1. TiO, thin films has recently intensified because of the multiple potential applications in high dielectric constant DRAMs, high frequency devices, and as possible gate oxides in metal-oxide-semiconductor field effect transistors (MOSFETs)^[26,27]. Due to the progressive reduction in dielectric film thickness necessitated by ever-shrinking feature size in Si microcircuits, it is becoming more important to perform in situ characterization of the surface of films and buried interfaces of heterostructures, during film deposition, in order to understand and control critical growth processes. TOF-ISARS provides a wide range of information on the physics and chemistry of surfaces, but not on processes occurring at buried interfaces during film growth. Information on buried interface processes can be obtained with SE. Therefore, we have developed an integrated sputter-deposition/TOF-ISARS/SE system to investigate the processes described above [28].

The dielectric properties of BST-based capacitors have been extensively investigated in recent years^[29,30]. The work published in the literature indicate that the BST/bottom electrode interface and diffusion barriers underneath this electrode may play significant roles in controlling critical properties such as leakage and breakdown of BST capacitors. Therefore, it is important to understand bottom electrode/diffusion barrier heterostructure processes before and during BST deposition. For this purpose, we recently initiated a systematic series of studies on the growth of BST on different bottom electrode/barrier heterostructures. We discuss here initial results from studies of BST growth on Ir/TiN heterostructures^[28], considered as candidates for integration of BST films with Si CMOS devices.

Synthesis of BST Thin Films

(Ba_{0.5}, Sr_{0.5})TiO₃ films were deposited by reactive ion beam sputtering using Ar⁺ ions generated by a 3 cm Kaufman RF plasma ion source. The substrates used were <001> MgO single crystals and Ir/TiN/SiO₂/Si, the latter supplied by

Motorola, with the Ir electrode deposited by RF magnetron sputtering and the TiN layer by chemical vapor deposition (CVD). The substrate temperature was kept at 700 °C during BST growth and subsequent post-deposition annealing. The BST films were synthesized via ion beam sputtering of a sintered Ba_{0.5}Sr_{0.5}TiO₃ ceramic target. Details on substrate cleaning and preparation, and film deposition conditions can be found elsewhere^[28].

TOF-ISARS and SE Characterization of Ir/TiN Electrodes and BST Growth Processes

We studied the stability of the Ir/TiN/SiO₂/Si heterostructure (a candidate for integration of BST films with Si CMOS devices). The tests included annealing in 1x10⁻⁴ Torr of oxygen at 700⁰C before deposition of the BST thin film. Fig 6 (a) shows an MSRI spectrum of the as-deposited Ir/TiN heterostructure. The spectrum revealed the presence of H⁺, C⁺, Na⁺, and other impurities, due mainly to atmospheric exposure. In addition, a small Ti peak and the Ir host peaks are present, indicating that there was slight segregation of Ti during the deposition of the Ir layer. MSRI spectra taken after 30 minutes annealing in oxygen (Fig. 6 (b)) show the disappearance of the Ir signal and strong reduction of all peaks shown in Fig. 6(a), simultaneously with the appearance of dominant Ti signal, indicative of a strong segregation of Ti to the surface of the Ir layer and that the Ir/TiN structure is not stable under heating in oxygen.

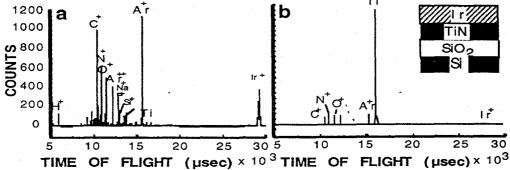


FIGURE 6. MSRI spectra of as-deposited (a) and oxygen annealed (5 x 10⁻⁴ Torr) at 700 °C (b) Ir /TiN/SiO₂/Si heterostructures^[28].

SE System Calibration for Characterization of BST Film Growth

In situ SE characterization of BST film growth processes requires an accurate determination of the optical constant of the BST film. Although some work has been reported [31] on the optical properties of BST thin films, the results showed that the refractive index of BST films varies with different deposition techniques, and deposition and post-annealing temperature, which results in different densities and degrees of crystallization of the BST films. Therefore, it was necessary to characterize the optical properties of the BST films grown in our ion beam sputter-deposition system, before undertaking the studies of growth processes. To avoid the SE modeling complexity that may arise from the possible formation of a reaction interface layer between the BST film and substrate, (Ba_{0.5}Sr_{0.5})TiO₃ was deposited on MgO <001> single crystal substrates, resulting in a sharp BST/MgO interface [32] because of the good thermal and chemical stability of MgO. The oxygen deficiency and

microstructural stability of BST films grown on MgO were investigated using sequential annealing in vacuum and oxygen at 700 °C. Real time SE was performed during each annealing cycle. No obvious changes in the SE spectra were observed before and after annealing. XRD analysis revealed that the ion beam sputter deposition technique produces good polycrystalline BST films at 700 °C. The surface roughness of the films was about 0.6 nm, as measured by AFM. This roughness value was used in the SE modeling to extract the optical constants of the BST films on MgO. The refractive index (n) and extinction coefficient (k) of our BST films agree with those expected for dense BST films^[31]. We observed low optical absorption until near the band gap (about 3.4 eV), as expected.

In Situ Spectroscopic Ellipsometry Studies of Buried Interfaces during Growth of BST on Ir/TiN/SiO₂/Si Substrates

We used *in situ* spectroscopic ellipsometry to study the buried interface between a BST film and an Ir/TiN/SiO₂/Si substrate during ion beam sputter-deposition of BST at 700 °C in 10 ⁻⁴ Torr of oxygen. The best fit of the SE model to the experimental data for the Ψ and Δ ellipsometric parameters related to the BST/Ir system is achieved with the incorporation of an effective medium approximation (EMA) interface layer involving a mixture of IrO₂ and Ir. This indicates that the Ir electrode is oxidized during the deposition of BST under the conditions specified above. The SE modeling results yield a thickness of 9.5 nm for the EMA interface layer formed after 60 minutes of BST deposition (Fig. 7). No evidence for a TiO₂ layer was found, which indicates that the Ti that diffused to the Ir surface prior to BST deposition (Fig. 6(b)) was incorporated into the growing BST film. As a result, a slightly higher Ti content than stoichiometric composition could be expected in BST deposited on this substrate structure.

Because SE analysis is largely based on model-fitting to experimental data, it is necessary to use complementary techniques to confirm the information provided by SE. In this respect, the *in situ* MSRI analysis and complementary AFM characterization performed by us confirmed the SE analysis. Cross-section TEM studies and XPS depth profiling can also contribute to confirm the information provided by the SE studies and we will be doing these in the future.

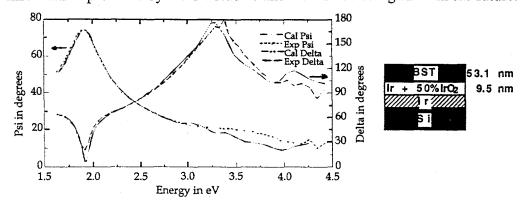


FIGURE 7. Experimental and calculated SE data for an as-deposited BST/Ir sample based on the best fit model, including the mean squared error (MSE) and the Ir layer thick enough to be considered as an opaque substrate.

CONCLUSIONS

We have demonstrated that *in situ* and/or *in situ*, real-time characterization techniques can provide a wide range of information on thin film growth and interface processes critical to the development of materials integration strategies for the next generation of ferroelectric thin film-based devices.

TOF-ISARS demonstrated that: (a) Ti and Si atoms segregate to the surface of a Pt/Ti/SiO₂/Si heterostructure when heated above 450 °C, while no such segregation is observed for the Pt/TiO₂/SiO₂/Si heterostructure; (b) Some Ta segregation is observed in Pt/Ta/SiO₂/Si heterostructures, but the extent of Ta segregation is much smaller than that of Ti in the Pt/Ti/SiO₂/Si heterostructure; (c) The presence of segregated Ti and Si on the surface of the Pt/Ti/SiO₂/Si heterostructure is associated with reduction and evaporative loss of metallic Bi, resulting in inefficient incorporation of Bi during the early stages of SBT film growth; (d) an intermediate level of Bi incorporation is observed for the intial stages of SBT growth on Pt/Ta/SiO₂/Si heterostructures; (e) Bi is incorporated very efficiently on RuO₂/SiO₂/Si structures. The results indicate that the choice of bottom electrode layers is critical for the growth of SBT films and their integration into Si substrates for the fabrication of NVFRAMs..

TOF-ISARS showed that there is a strong segregation of Ti to the surface of Ir in Ir/TiN bottom electrode heterostructures proposed as candidates for integration of BST with Si substrates, making the Ir/TiN an unsuitable electrode heterostructure. In addition, the use of the complementary spectroscopic ellipsometry method revealed a partial oxidation of Ir bottom electrodes during growth of BST films by sputter-deposition in oxygen.

Our work has demonstrated that complementary in situ and in situ, realtime characterization techniques will play an ever increasing role in basic and applied research, not only in the field of ferroelectrics, but also in many others.

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