

SUPPLEMENTAL MATERIAL

ENTHALPY OF SOLVATION CORRELATIONS FOR GASEOUS SOLUTES
DISSOLVED IN LINEAR ALKANES (C₅ thru C₁₆) BASED ON THE ABRAHAM
MODEL

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TABLE S1. Experimental Enthalpies of Solvation (kJ/mole) of Gaseous Solutes in

Linear Alkane Solvents

Solute	E	S	A	B	L	Exp	Ref
Pentane							
Pentane	0.000	0.000	0.000	0.000	2.162	-26.74	10
Methyl methacrylate	0.245	0.510	0.000	0.440	2.880	-32.17	42
Pentyl acetate	0.067	0.600	0.000	0.450	3.844	-43.96	2
Butyl acetate	0.071	0.600	0.000	0.450	3.353	-40.09	3
Butyl propanoate	0.058	0.560	0.000	0.470	3.833	-44.17	3
Propyl acetate	0.092	0.600	0.000	0.450	2.819	-34.04	4
Propyl propanoate	0.070	0.560	0.000	0.450	3.338	-38.58	4
Propyl butanoate	0.050	0.560	0.000	0.450	3.783	-41.90	4
Ethyl butanoate	0.068	0.580	0.000	0.450	3.271	-38.29	5
Ethyl pentanoate	0.049	0.580	0.000	0.450	3.769	-43.46	5
Ethyl hexanoate	0.043	0.580	0.000	0.450	4.251	-48.71	5
Xenon	0.000	0.000	0.000	0.000	0.378	-10.95	8
1,1,2-Tetrachloroethane	0.595	0.760	0.160	0.120	3.803	-40.64	7
Propanenitrile	0.162	0.900	0.020	0.360	2.082	-26.10	43
Acetophenone	0.818	1.010	0.000	0.480	4.501	-44.93	9
Methyl acrylate	0.254	0.660	0.000	0.420	2.360	-29.79	6
1,1,1-Trichloroethane	0.369	0.410	0.000	0.090	2.733	-30.86	104
Hexane							
Methane	0.000	0.000	0.000	0.000	-0.323	-2.25	11
Ethane	0.000	0.000	0.000	0.000	0.492	-8.33	11
Propane	0.000	0.000	0.000	0.000	1.050	-14.10	11
Butane	0.000	0.000	0.000	0.000	1.615	-20.50	11
Pentane	0.000	0.000	0.000	0.000	2.162	-26.74	11
Hexane	0.000	0.000	0.000	0.000	2.668	-31.54	11
Heptane	0.000	0.000	0.000	0.000	3.173	-36.56	11
Octane	0.000	0.000	0.000	0.000	3.677	-41.50	11
Decane	0.000	0.000	0.000	0.000	4.686	-50.84	12
Undecane	0.000	0.000	0.000	0.000	5.191	-56.22	13
Dodecane	0.000	0.000	0.000	0.000	5.696	-61.60	27
Hexadecane	0.000	0.000	0.000	0.000	7.714	-81.34	29
2-Methylpropane	0.000	0.000	0.000	0.000	1.409	-18.87	11
2,2-Dimethylpropane	0.000	0.000	0.000	0.000	1.820	-21.55	11
2,2-Dimethylbutane	0.000	0.000	0.000	0.000	2.352	-27.80	14
2,4-Dimethylpentane	0.000	0.000	0.000	0.000	2.809	-32.67	16
2,2,4-Trimethylpentane	0.000	0.000	0.000	0.000	3.106	-35.24	15
Cyclopentane	0.263	0.100	0.000	0.000	2.477	-28.28	27
Cyclohexane	0.305	0.100	0.000	0.000	2.964	-32.43	11
Adamantane	0.760	0.570	0.000	0.040	4.934	-48.40	39
Ethene	0.107	0.100	0.000	0.070	0.289	-7.46	36

Benzene	0.610	0.520	0.000	0.140	2.786	-30.70	40
Toluene	0.601	0.520	0.000	0.140	3.325	-34.11	28
Naphthalene	1.340	0.920	0.000	0.200	5.161	-51.20	18
Bromobenzene	0.882	0.730	0.000	0.090	4.041	-41.44	35
Chlorobenzene	0.718	0.650	0.000	0.070	3.657	-38.02	40
1,2-Dichlorobenzene	0.872	0.780	0.000	0.040	4.518	-44.35	40
1,2,4-Trichlorobenzene	0.980	0.810	0.000	0.000	5.248	-51.94	34
Nitrobenzene	0.871	1.110	0.000	0.280	4.557	-40.44	40
Methanol	0.278	0.440	0.430	0.470	0.970	-15.10	32
Ethanol	0.246	0.420	0.370	0.480	1.485	-19.30	32
1-Propanol	0.236	0.420	0.370	0.480	2.031	-22.06	33
1-Butanol	0.224	0.420	0.370	0.480	2.601	-28.57	33
1-Hexanol	0.210	0.420	0.370	0.480	3.610	-34.44	19
2-Ethyl-1-butanol	0.231	0.390	0.370	0.480	3.523	-34.43	19
2-Methyl-1-pentanol	0.211	0.390	0.370	0.480	3.530	-33.51	19
1-Nonanol	0.193	0.420	0.370	0.480	5.124	-57.67	37
1-Undecanol	0.181	0.420	0.370	0.480	6.130	-63.33	37
1,1,1,3,3,3-hexafluoro-2-propanol	-0.240	0.550	0.770	0.100	1.392	-20.03	38
Benzaldehyde	0.820	1.000	0.000	0.390	4.008	-40.89	41
Aniline	0.955	0.960	0.260	0.410	3.934	-46.80	40
Pyridine	0.631	0.840	0.000	0.520	3.022	-32.08	17
2-Methylpyridine	0.598	0.750	0.000	0.580	3.422	-36.75	17
3-Methylpyridine	0.631	0.810	0.000	0.540	3.631	-37.91	25
4-Methylpyridine	0.630	0.820	0.000	0.540	3.640	-37.88	26
2,4-Dimethylpyridine	0.634	0.760	0.000	0.630	4.006	-41.53	17
2,6-Dimethylpyridine	0.607	0.700	0.000	0.630	3.760	-40.16	17
2-Chloropyridine	0.738	1.030	0.000	0.370	3.875	-39.70	17
3-Chloropyridine	0.732	0.830	0.000	0.400	3.783	-42.00	17
3-Cyanopyridine	0.750	1.260	0.000	0.620	4.164	-40.40	17
4-Cyanopyridine	0.750	1.210	0.000	0.590	4.033	-39.00	17
Methyl methacrylate	0.245	0.510	0.000	0.440	2.880	-33.24	42
Propionitrile	0.162	0.900	0.020	0.360	2.082	-24.09	30
Butyronitrile	0.188	0.900	0.000	0.360	2.548	-31.68	43
Acetone	0.179	0.700	0.040	0.490	1.696	-21.84	30
2-Butanone	0.166	0.700	0.000	0.510	2.287	-27.44	30
2-Pentanone	0.143	0.680	0.000	0.510	2.755	-32.50	20
3-Pentanone	0.154	0.660	0.000	0.510	2.811	-33.08	21
Dipropyl ether	0.008	0.250	0.000	0.450	2.954	-35.07	30
Butyl methyl ether	0.045	0.250	0.000	0.440	2.658	-31.29	22
Methyl <i>tert</i> -butyl ether	0.024	0.210	0.000	0.590	2.380	-29.18	44
Ethyl <i>tert</i> -butyl ether	-0.020	0.160	0.000	0.600	2.720	-32.50	46
Methyl <i>tert</i> -amyl ether	0.050	0.210	0.000	0.600	2.916	-34.79	48
Furan	0.369	0.510	0.000	0.130	1.913	-23.83	49
Tetrahydrofuran	0.289	0.520	0.000	0.480	2.636	-28.81	47
2-Methyltetrahydrofuran	0.241	0.480	0.000	0.530	2.820	-31.64	45
Tetrahydropyran	0.275	0.470	0.000	0.550	3.057	-32.55	49
1,4-Dioxane	0.329	0.750	0.000	0.640	2.892	-31.15	30

Dimethoxymethane	0.099	0.460	0.000	0.520	1.894	-23.85	50
1,2-Dimethoxyethane	0.116	0.670	0.000	0.680	2.654	-31.36	30
2,5,8-Trioxanonane	0.113	0.760	0.000	1.170	3.920	-40.30	30
2,5,8,11,14-Pentaoxapentadecane	-0.020	1.110	0.000	1.790	6.498	-67.61	51
Paraldehyde	0.136	0.680	0.000	0.680	3.169	-34.66	30
Anisole	0.710	0.750	0.000	0.290	3.890	-40.75	30
Acetal	-0.020	0.560	0.000	0.620	3.066	-37.51	30
Quinoline	1.268	0.970	0.000	0.540	5.457	-50.19	30
Methyl acetate	0.142	0.640	0.000	0.450	1.911	-21.40	24
Ethyl acetate	0.106	0.620	0.000	0.450	2.314	-28.28	24
Propyl acetate	0.092	0.600	0.000	0.450	2.819	-33.45	24
Butyl acetate	0.071	0.600	0.000	0.450	3.353	-38.40	24
Pentyl acetate	0.067	0.600	0.000	0.450	3.844	-42.50	24
Ethyl propionate	0.087	0.580	0.000	0.450	2.807	-32.52	24
Ethyl butanoate	0.068	0.580	0.000	0.450	3.271	-36.71	24
Ethyl hexanoate	0.043	0.580	0.000	0.450	4.251	-47.24	24
Propyl propionate	0.070	0.560	0.000	0.450	3.338	-37.65	52
Helium	0.000	0.000	0.000	0.000	-1.741	8.03	11
Neon	0.000	0.000	0.000	0.000	-1.575	5.44	11
Argon	0.000	0.000	0.000	0.000	-0.688	-2.72	11
Krypton	0.000	0.000	0.000	0.000	-0.211	-4.73	11
Xenon	0.000	0.000	0.000	0.000	0.378	-10.71	11
Radon	0.000	0.000	0.000	0.000	0.877	-12.68	11
Hydrogen	0.000	0.000	0.000	0.000	-1.200	5.10	11
Nitrogen	0.000	0.000	0.000	0.000	-0.978	0.79	11
Oxygen	0.000	0.000	0.000	0.000	-0.723	-0.96	11
Nitric Oxide	0.370	0.020	0.000	0.090	-0.590	-2.23	31
Tetrafluoromethane	-0.580	-0.260	0.000	0.000	-0.817	-1.46	11
Sulfur Hexafluoride	-0.600	-0.200	0.000	0.000	-0.120	-8.28	11
Triethylamine	0.101	0.150	0.000	0.790	3.040	-34.58	53
Penylamine	0.211	0.350	0.160	0.610	3.139	-35.01	54
Hexylamine	0.197	0.350	0.160	0.610	3.655	-40.61	55
Heptylamine	0.197	0.350	0.160	0.610	4.153	-45.61	56
Octylamine	0.187	0.350	0.160	0.610	4.520	-50.68	57
Decylamine	0.182	0.350	0.160	0.610	5.606	-60.18	58
Nitromethane	0.313	0.950	0.060	0.310	1.892	-22.59	59
Nitroethane	0.270	0.950	0.020	0.330	2.414	-27.61	28
1-Nitropropane	0.242	0.950	0.000	0.310	2.894	-32.66	28
2-Nitropropane	0.216	0.920	0.000	0.330	2.550	-31.23	60
Diethyl sulfide	0.373	0.380	0.000	0.320	3.104	-33.64	61
1,2-Dichloroethane	0.416	0.640	0.100	0.110	2.573	-29.23	62
Chloroform	0.425	0.490	0.150	0.020	2.480	-29.90	63
Ethyl iodide	0.640	0.400	0.000	0.150	2.573	-28.64	28
1-Chlorobutane	0.210	0.400	0.000	0.100	2.722	-34.21	28
2-Chloro-2-methylpropane	0.142	0.300	0.000	0.030	2.273	-27.07	28
Carbon tetrachloride	0.458	0.380	0.000	0.000	2.823	-32.40	28
1-Chloronaphthalene	1.417	1.000	0.000	0.140	5.856	-61.74	64

Acetophenone	0.818	1.010	0.000	0.480	4.501	-44.77	23
Ethyl benzoate	0.689	0.850	0.000	0.460	5.075	-54.11	65
Methyl benzoate	0.733	0.850	0.000	0.460	4.704	-49.08	65
Heptane							
Methane	0.000	0.000	0.000	0.000	-0.323	-3.81	1
Ethane	0.000	0.000	0.000	0.000	0.492	-11.17	1
Pentane	0.000	0.000	0.000	0.000	2.162	-26.53	1
Hexane	0.000	0.000	0.000	0.000	2.668	-31.55	1
Heptane	0.000	0.000	0.000	0.000	3.173	-36.57	1
Octane	0.000	0.000	0.000	0.000	3.677	-41.51	1
Nonane	0.000	0.000	0.000	0.000	4.182	-46.40	1
Decane	0.000	0.000	0.000	0.000	4.686	-51.38	1
Dodecane	0.000	0.000	0.000	0.000	5.696	-61.17	1
Hexadecane	0.000	0.000	0.000	0.000	7.714	-80.92	1
3-Ethylpentane	0.000	0.000	0.000	0.000	3.091	-35.19	1
2-Methylpentane	0.000	0.000	0.000	0.000	2.503	-30.05	1
3-Methylpentane	0.000	0.000	0.000	0.000	2.581	-30.43	1
2,2-Dimethylbutane	0.000	0.000	0.000	0.000	2.352	-27.85	1
2,3-Dimethylbutane	0.000	0.000	0.000	0.000	2.495	-29.30	1
2,4-Dimethylpentane	0.000	0.000	0.000	0.000	2.809	-32.92	1
2,2,4-Trimethylpentane	0.000	0.000	0.000	0.000	3.106	-35.06	1
Cyclohexane	0.305	0.100	0.000	0.000	2.964	-32.29	1
Cycloheptane	0.350	0.100	0.000	0.000	3.704	-37.81	1
Cyclooctane	0.413	0.100	0.000	0.000	4.329	-42.66	1
Cyclodecane	0.474	0.100	0.000	0.000	5.340	-52.10	1
Methylcyclopentane	0.225	0.100	0.000	0.000	2.816	-31.39	1
Methylcyclohexane	0.244	0.100	0.000	0.000	3.323	-35.37	1
<i>cis</i> -1,2-Dimethylcyclohexane	0.281	0.240	0.000	0.000	3.847	-39.76	1
<i>trans</i> -1,2-Dimethylcyclohexane	0.227	0.200	0.000	0.000	3.722	-38.69	1
<i>cis</i> Decalin	0.550	0.250	0.000	0.000	5.156	-50.82	1
<i>trans</i> Decalin	0.467	0.230	0.000	0.000	4.984	-49.95	1
Bicyclohexyl	0.531	0.330	0.000	0.070	5.922	-57.84	1
Tetralin	0.891	0.650	0.000	0.170	5.203	-52.80	1
1-Hexene	0.078	0.080	0.000	0.070	2.572	-30.38	1
1-Octyne	0.155	0.220	0.090	0.100	3.521	-40.04	1
2-Octyne	0.225	0.300	0.000	0.100	3.850	-42.63	1
Propanal	0.196	0.650	0.000	0.450	1.815	-20.19	1
Butanal	0.187	0.650	0.000	0.450	2.270	-25.19	1
Pentanal	0.163	0.650	0.000	0.450	2.770	-31.15	1
Hexanal	0.146	0.650	0.000	0.450	3.370	-36.45	1
Heptanal	0.140	0.650	0.000	0.450	3.860	-41.59	1
Octanal	0.160	0.650	0.000	0.450	4.380	-45.76	1
Nonanal	0.150	0.650	0.000	0.450	4.856	-50.84	1
Isobutyraldehyde	0.144	0.620	0.000	0.450	2.120	-26.30	1
Acetone	0.179	0.700	0.040	0.490	1.696	-21.59	1
2-Butanone	0.166	0.700	0.000	0.510	2.287	-26.53	1

2-Heptanone	0.123	0.680	0.000	0.510	3.760	-41.09	1
4-Heptanone	0.110	0.660	0.000	0.510	3.705	-41.51	1
2-Nonanone	0.119	0.680	0.000	0.510	4.735	-50.12	1
5-Nonanone	0.103	0.660	0.000	0.510	4.698	-50.08	1
2,2,4,4-Tetramethyl-3-pentanone	0.099	0.560	0.000	0.520	4.370	-42.13	1
Cyclohexanone	0.403	0.860	0.000	0.560	3.792	-37.78	1
Butyl acetate	0.071	0.600	0.000	0.450	3.353	-38.72	1
Butyl propionate	0.058	0.560	0.000	0.470	3.833	-44.94	1
Butyl butanoate	0.044	0.560	0.000	0.450	4.275	-47.61	1
Methyl methacrylate	0.245	0.510	0.000	0.440	2.880	-33.11	1
Diethyl ether	0.041	0.250	0.000	0.450	2.015	-24.31	1
Dibutyl ether	0.000	0.250	0.000	0.450	3.924	-44.22	1
Dipentyl ether	0.000	0.250	0.000	0.450	4.875	-52.54	1
Diisopropyl Ether	-0.060	0.160	0.000	0.580	2.530	-30.96	1
n-Butyl methyl ether	0.045	0.250	0.000	0.440	2.658	-30.79	1
Methyl heptyl ether	0.048	0.250	0.000	0.450	4.088	-46.11	1
Methyl <i>tert</i> -butyl ether	0.024	0.210	0.000	0.590	2.380	-24.81	1
Ethyl <i>tert</i> -butyl ether	-0.020	0.160	0.000	0.600	2.720	-32.28	1
Methyl <i>tert</i> -amyl ether	0.050	0.210	0.000	0.600	2.916	-34.61	1
Tetrahydrofuran	0.289	0.520	0.000	0.480	2.636	-29.32	1
2-Methyltetrahydrofuran	0.241	0.480	0.000	0.530	2.820	-31.59	1
1-Fluorooctane	-0.020	0.350	0.000	0.100	3.850	-47.86	1
Tetrachloromethane	0.458	0.380	0.000	0.000	2.823	-31.21	1
1,2-Dichloroethane	0.416	0.640	0.100	0.110	2.573	-27.98	1
1-Chlorooctane	0.191	0.400	0.000	0.090	4.708	-50.88	1
Diiodomethane	0.714	0.690	0.110	0.070	2.886	-37.74	1
Methanol	0.278	0.440	0.430	0.470	0.970	-14.90	1
Ethanol	0.246	0.420	0.370	0.480	1.485	-18.60	1
Butan-1-ol	0.224	0.420	0.370	0.480	2.601	-27.87	1
Pentan-1-ol	0.219	0.420	0.370	0.480	3.106	-34.14	1
1-Octanol	0.199	0.420	0.370	0.480	4.619	-47.57	1
1-Nonanol	0.193	0.420	0.370	0.480	5.124	-53.66	1
1-Undecanol	0.181	0.420	0.370	0.480	6.130	-61.83	1
3-Methyl-1-butanol	0.192	0.390	0.370	0.480	3.011	-35.63	1
2-Pentanol	0.195	0.360	0.330	0.560	2.840	-35.90	1
3-Methyl-2-butanol	0.194	0.330	0.330	0.560	2.793	-31.93	1
2-Methyl-2-butanol	0.194	0.300	0.310	0.600	2.630	-26.44	1
Cyclohexanol	0.460	0.540	0.320	0.570	3.758	-37.50	1
2-Methoxyethanol	0.269	0.500	0.300	0.840	2.490	-34.22	1
Triethylamine	0.101	0.150	0.000	0.790	3.040	-34.45	1
Propylamine	0.225	0.350	0.160	0.610	2.141	-25.14	1
Isopropylamine	0.181	0.320	0.160	0.610	1.908	-22.87	1
Butylamine	0.224	0.350	0.160	0.610	2.618	-29.97	1
sec-Butylamine	0.170	0.320	0.160	0.630	2.410	-27.89	1
iso-Butylamine	0.121	0.290	0.160	0.710	2.493	-28.11	1
<i>tert</i> -Butylamine	0.121	0.290	0.160	0.710	2.493	-24.98	1
Pentylamine	0.211	0.350	0.160	0.610	3.139	-34.12	1

Hexylamine	0.197	0.350	0.160	0.610	3.655	-39.47	1
Heptylamine	0.197	0.350	0.160	0.610	4.153	-44.97	1
Octylamine	0.187	0.350	0.160	0.610	4.520	-49.41	1
Nonylamine	0.187	0.350	0.160	0.610	5.100	-54.09	1
Decylamine	0.182	0.350	0.160	0.610	5.606	-58.69	1
3-Methylphenol	0.822	0.880	0.570	0.340	4.310	-39.92	1
Benzene	0.610	0.520	0.000	0.140	2.786	-30.54	1
Toluene	0.601	0.520	0.000	0.140	3.325	-35.94	1
Mesitylene	0.649	0.520	0.000	0.190	4.344	-46.23	1
Naphthalene	1.340	0.920	0.000	0.200	5.161	-51.82	1
Trifluorotoluene	0.225	0.480	0.000	0.100	2.894	-33.35	1
Nitrobenzene	0.871	1.110	0.000	0.280	4.557	-44.73	1
Anisole	0.710	0.750	0.000	0.290	3.890	-41.21	1
Aniline	0.955	0.960	0.260	0.410	3.934	-40.75	1
Acetonitrile	0.237	0.900	0.070	0.320	1.739	-17.56	1
Helium	0.000	0.000	0.000	0.000	-1.741	7.72	1
Neon	0.000	0.000	0.000	0.000	-1.575	5.56	1
Argon	0.000	0.000	0.000	0.000	-0.688	-1.22	1
Krypton	0.000	0.000	0.000	0.000	-0.211	-5.51	1
Xenon	0.000	0.000	0.000	0.000	0.378	-10.08	1
Hydrogen	0.000	0.000	0.000	0.000	-1.200	3.78	1
Carbon dioxide	0.000	0.280	0.050	0.100	0.058	-9.66	1
Tetrafluoromethane	-0.580	-0.260	0.000	0.000	-0.817	-1.59	1
Sulfur Hexafluoride	-0.600	-0.200	0.000	0.000	-0.120	-8.28	1
Difluorodichloromethane	0.037	0.130	0.000	0.000	1.124	-18.91	1
Hexafluorobenzene	0.088	0.560	0.000	0.010	2.345	-31.62	1
Chlorotrifluoromethane	-0.247	-0.046	0.000	0.000	0.209	-12.84	1
Nitromethane	0.313	0.950	0.060	0.310	1.892	-24.40	1
1-Methylnaphthalene	1.337	0.940	0.000	0.220	5.802	-54.06	1
1-Butanethiol	0.382	0.350	0.000	0.240	3.243	-33.81	1
1,2,4-Trimethylbenzene	0.677	0.560	0.000	0.190	4.441	-46.76	1
2,5,8,11,14-Pentaoxopentadecane	-0.020	1.110	0.000	1.790	6.498	-68.10	1
2,5,8,11-Tetraoxododecane	0.000	0.980	0.000	1.440	5.157	-54.15	1
2,5,8-Trioxanonane	0.113	0.760	0.000	1.170	3.920	-39.94	1
Diethoxymethane	0.010	0.490	0.000	0.540	2.789	-33.41	1
1,2-Dimethoxyethane	0.116	0.670	0.000	0.680	2.654	-31.76	1
Dimethoxymethane	0.099	0.460	0.000	0.520	1.894	-26.06	1
1-Propanol	0.236	0.420	0.370	0.480	2.031	-22.10	1
2-Butanol	0.217	0.360	0.330	0.560	2.338	-26.70	1
2-Methyl-1-propanol	0.217	0.390	0.370	0.480	2.413	-26.60	1
2-Methyl-2-propanol	0.180	0.300	0.310	0.600	1.963	-22.60	1
1-Hexanol	0.210	0.420	0.370	0.480	3.610	-38.32	1
Sulfur dioxide	0.370	0.660	0.280	0.100	0.778	-15.90	1
Pentyl acetate	0.067	0.600	0.000	0.450	3.844	-44.52	1
Cyclopentanol	0.427	0.540	0.320	0.560	3.241	-37.21	1
Ethyl benzoate	0.689	0.850	0.000	0.460	5.075	-53.08	1
Ethyl butanoate	0.068	0.580	0.000	0.450	3.271	-38.07	1

Propyl butanoate	0.050	0.560	0.000	0.450	3.783	-43.10	1
Pyridine	0.631	0.840	0.000	0.520	3.022	-31.76	68
3-Methylpyridine	0.631	0.810	0.000	0.540	3.631	-37.77	66
4-Methylpyridine	0.630	0.820	0.000	0.540	3.640	-37.34	69
Propane	0.000	0.000	0.000	0.000	1.050	-17.13	78
1,2,4-Trichlorobenzene	0.980	0.810	0.000	0.000	5.248	-52.36	34
Octane							
Methane	0.000	0.000	0.000	0.000	-0.323	-4.06	36
Octane	0.000	0.000	0.000	0.000	3.677	-41.51	10
Dodecane	0.000	0.000	0.000	0.000	5.696	-61.62	79
2,2,4-Trimethylpentane	0.000	0.000	0.000	0.000	3.106	-35.09	15
Diethyl ether	0.041	0.250	0.000	0.450	2.015	-25.90	83
Dipropyl ether	0.008	0.250	0.000	0.450	2.954	-34.76	80
Dibutyl ether	0.000	0.250	0.000	0.450	3.924	-44.37	85
n-Butyl methyl ether	0.045	0.250	0.000	0.440	2.658	-31.22	81
Methyl <i>tert</i> -butyl ether	0.024	0.210	0.000	0.590	2.380	-28.60	83
Ethyl <i>tert</i> -butyl ether	-0.020	0.160	0.000	0.600	2.720	-32.19	46
Tetrahydropyran	0.275	0.470	0.000	0.550	3.057	-33.31	84
1,4-Dioxane	0.329	0.750	0.000	0.640	2.892	-31.02	87
2-Methyltetrahydrofuran	0.241	0.480	0.000	0.530	2.820	-31.43	45
2,5,8-Trioxanonane	0.113	0.760	0.000	1.170	3.920	-40.80	88
Methanol	0.278	0.440	0.430	0.470	0.970	-15.30	32
Ethanol	0.246	0.420	0.370	0.480	1.485	-19.50	32
1-Propanol	0.236	0.420	0.370	0.480	2.031	-25.00	82
1-Butanol	0.224	0.420	0.370	0.480	2.601	-29.90	82
1-Nonanol	0.193	0.420	0.370	0.480	5.124	-55.74	86
1-Undecanol	0.181	0.420	0.370	0.480	6.130	-64.40	86
1-Hexene	0.078	0.080	0.000	0.070	2.572	-30.36	89
Chlorotrifluoromethane	-0.247	-0.046	0.000	0.000	0.209	-9.41	36
Dichlorodifluoromethane	0.037	0.130	0.000	0.000	1.124	-17.07	36
1-Chloronaphthalene	1.417	1.000	0.000	0.140	5.856	-61.57	90
Helium	0.000	0.000	0.000	0.000	-1.741	8.06	36
Neon	0.000	0.000	0.000	0.000	-1.575	6.94	36
Argon	0.000	0.000	0.000	0.000	-0.688	-0.36	36
Krypton	0.000	0.000	0.000	0.000	-0.211	-5.00	36
Xenon	0.000	0.000	0.000	0.000	0.378	-10.16	8
Hydrogen	0.000	0.000	0.000	0.000	-1.200	4.04	36
Tetrafluoromethane	-0.580	-0.260	0.000	0.000	-0.817	-0.09	36
Sulfur hexafluoride	-0.600	-0.200	0.000	0.000	-0.120	-8.34	36
Carbon monoxide	0.000	0.000	0.000	0.040	-0.836	-1.13	94
Carbon dioxide	0.000	0.280	0.050	0.100	0.058	-8.11	94
Butyronitrile	0.188	0.900	0.000	0.360	2.548	-30.26	43
Acetophenone	0.818	1.010	0.000	0.480	4.501	-44.68	9
Ethyl acetate	0.106	0.620	0.000	0.450	2.314	-29.14	65
Ethyl benzoate	0.689	0.850	0.000	0.460	5.075	-52.89	65
Pyridine	0.631	0.840	0.000	0.520	3.022	-31.50	70

3-Methylpyridine	0.631	0.810	0.000	0.540	3.631	-37.61	67
4-Methylpyridine	0.630	0.820	0.000	0.540	3.640	-37.70	71
Nonane							
Nonane	0.000	0.000	0.000	0.000	4.182	-46.44	10
Cycloheptane	0.350	0.100	0.000	0.000	3.704	-37.71	95
Cyclooctane	0.413	0.100	0.000	0.000	4.329	-42.53	95
Bromobenzene	0.882	0.730	0.000	0.090	4.041	-40.82	35
Helium	0.000	0.000	0.000	0.000	-1.741	9.69	36
Neon	0.000	0.000	0.000	0.000	-1.575	6.29	36
Argon	0.000	0.000	0.000	0.000	-0.688	-1.46	36
Krypton	0.000	0.000	0.000	0.000	-0.211	-4.64	36
Xenon	0.000	0.000	0.000	0.000	0.378	-9.95	8
Ethyl acetate	0.106	0.620	0.000	0.450	2.314	-28.61	105
Methyl nonanoate	0.056	0.600	0.000	0.450	5.321	-58.51	106
Methyl decanoate	0.053	0.600	0.000	0.450	5.806	-63.41	106
Decane							
Methane	0.000	0.000	0.000	0.000	-0.323	-4.31	11
Ethane	0.000	0.000	0.000	0.000	0.492	-7.78	11
Propane	0.000	0.000	0.000	0.000	1.050	-13.72	11
Butane	0.000	0.000	0.000	0.000	1.615	-20.29	11
Hexane	0.000	0.000	0.000	0.000	2.668	-31.49	96
Decane	0.000	0.000	0.000	0.000	4.686	-51.38	10
Dodecane	0.000	0.000	0.000	0.000	5.696	-61.68	79
2-Methylpropane	0.000	0.000	0.000	0.000	1.409	-16.65	11
2-Methylpentane	0.000	0.000	0.000	0.000	2.503	-29.73	12
3-Methylpentane	0.000	0.000	0.000	0.000	2.581	-30.14	12
2,2-Dimethylbutane	0.000	0.000	0.000	0.000	2.352	-27.43	12
2,3-Dimethylbutane	0.000	0.000	0.000	0.000	2.495	-28.83	12
2,2,4-Trimethylpentane	0.000	0.000	0.000	0.000	3.106	-34.94	15
1-Hexene	0.078	0.080	0.000	0.070	2.572	-30.26	89
Diethyl ether	0.041	0.250	0.000	0.450	2.015	-25.63	97
Dipropyl ether	0.008	0.250	0.000	0.450	2.954	-34.66	80
Dibutyl ether	0.000	0.250	0.000	0.450	3.924	-44.26	98
n-Butyl methyl ether	0.045	0.250	0.000	0.440	2.658	-31.05	81
Ethyl <i>tert</i> -Butyl ether	-0.020	0.160	0.000	0.600	2.720	-32.22	99
Tetrahydrofuran	0.289	0.520	0.000	0.480	2.636	-28.73	47
2,5,8,11-Tetraoxadodecane	0.000	0.980	0.000	1.440	5.157	-53.14	100
2,5,8,11,14-Pentaoxapentadecane	-0.020	1.110	0.000	1.790	6.498	-56.58	51
Methyl methacrylate	0.245	0.510	0.000	0.440	2.880	-33.18	42
Methanol	0.278	0.440	0.430	0.470	0.970	-15.82	33
1-Nonanol	0.193	0.420	0.370	0.480	5.124	-54.96	37
1-Undecanol	0.181	0.420	0.370	0.480	6.130	-63.87	37
2-Methyl-2-butanol	0.194	0.300	0.310	0.600	2.630	-28.30	101
1-Chloronaphthalene	1.417	1.000	0.000	0.140	5.856	-61.92	91
Helium	0.000	0.000	0.000	0.000	-1.741	6.86	11

Neon	0.000	0.000	0.000	0.000	-1.575	6.53	11
Argon	0.000	0.000	0.000	0.000	-0.688	-1.63	11
Krypton	0.000	0.000	0.000	0.000	-0.211	-4.87	36
Xenon	0.000	0.000	0.000	0.000	0.378	-9.84	8
Nitrogen	0.000	0.000	0.000	0.000	-0.978	-0.29	11
Carbon monoxide	0.000	0.000	0.000	0.040	-0.836	0.38	11
Carbon dioxide	0.000	0.280	0.050	0.100	0.058	-6.92	94
Oxygen	0.000	0.000	0.000	0.000	-0.723	-0.25	11
Sulfur hexafluoride	-0.600	-0.200	0.000	0.000	-0.120	-7.91	11
Acetophenone	0.818	1.010	0.000	0.480	4.501	-44.75	9
Pyridine	0.631	0.840	0.000	0.520	3.022	-31.43	72
3-Methylpyridine	0.631	0.810	0.000	0.540	3.631	-37.43	73
4-Methylpyridine	0.630	0.820	0.000	0.540	3.640	-37.57	74
Undecane							
Undecane	0.000	0.000	0.000	0.000	5.191	-56.30	10
Xenon	0.000	0.000	0.000	0.000	0.378	-9.64	8
1-Hexanol	0.210	0.420	0.370	0.480	3.610	-37.91	102
Dodecane							
Methane	0.000	0.000	0.000	0.000	-0.323	-3.95	36
Propane	0.000	0.000	0.000	0.000	1.050	-15.28	78
Hexane	0.000	0.000	0.000	0.000	2.668	-31.43	27
Heptane	0.000	0.000	0.000	0.000	3.173	-36.49	27
Octane	0.000	0.000	0.000	0.000	3.677	-41.43	27
Decane	0.000	0.000	0.000	0.000	4.686	-51.36	79
Dodecane	0.000	0.000	0.000	0.000	5.696	-61.70	10
2,4-Dimethylpentane	0.000	0.000	0.000	0.000	2.809	-32.32	16
1-Hexene	0.078	0.080	0.000	0.070	2.572	-30.15	89
Diethyl ether	0.041	0.250	0.000	0.450	2.015	-25.42	97
Dipropyl ether	0.008	0.250	0.000	0.450	2.954	-34.49	80
Dibutyl ether	0.000	0.250	0.000	0.450	3.924	-44.10	98
n-Butyl methyl ether	0.045	0.250	0.000	0.440	2.658	-31.09	81
Ethyl <i>tert</i> -Butyl ether	-0.020	0.160	0.000	0.600	2.720	-32.02	99
2-Methyltetrahydrofuran	0.241	0.480	0.000	0.530	2.820	-31.27	45
2,5,8,11-Tetraoxadodecane	0.000	0.980	0.000	1.440	5.157	-53.61	100
Methyl methacrylate	0.245	0.510	0.000	0.440	2.880	-32.80	42
1,4-Difluorobenzene	0.384	0.600	0.000	0.060	2.766	-32.48	103
1-Chloronaphthalene	1.417	1.000	0.000	0.140	5.856	-62.37	92
Helium	0.000	0.000	0.000	0.000	-1.741	7.21	36
Neon	0.000	0.000	0.000	0.000	-1.575	6.97	36
Argon	0.000	0.000	0.000	0.000	-0.688	-0.64	36
Krypton	0.000	0.000	0.000	0.000	-0.211	-4.25	36
Xenon	0.000	0.000	0.000	0.000	0.378	-9.53	8
Carbon dioxide	0.000	0.280	0.050	0.100	0.058	-6.00	78
Butyronitrile	0.188	0.900	0.000	0.360	2.548	-32.86	43
1-Propanol	0.236	0.420	0.370	0.480	2.031	-22.62	102
Pyridine	0.631	0.840	0.000	0.520	3.022	-31.86	75

3-Methylpyridine	0.631	0.810	0.000	0.540	3.631	-37.52	76
4-Methylpyridine	0.630	0.820	0.000	0.540	3.640	-37.55	77
Tridecane							
Tridecane	0.000	0.000	0.000	0.000	6.200	-66.50	10
Xenon	0.000	0.000	0.000	0.000	0.378	-9.53	8
Tetradecane							
Tetradecane	0.000	0.000	0.000	0.000	6.705	-71.40	10
Diethyl ether	0.041	0.250	0.000	0.450	2.015	-25.20	83
Dipropyl ether	0.008	0.250	0.000	0.450	2.954	-33.68	83
Methyl tert-butyl ether	0.024	0.210	0.000	0.590	2.380	-26.52	83
1-Hexene	0.078	0.080	0.000	0.070	2.572	-30.06	89
Benzene	0.610	0.520	0.000	0.140	2.786	-30.58	27
Hexafluorobenzene	0.088	0.560	0.000	0.010	2.345	-32.33	103
3-Pentanone	0.154	0.660	0.000	0.510	2.811	-31.69	96
Helium	0.000	0.000	0.000	0.000	-1.741	5.89	36
Neon	0.000	0.000	0.000	0.000	-1.575	5.88	36
Argon	0.000	0.000	0.000	0.000	-0.688	-1.47	36
Krypton	0.000	0.000	0.000	0.000	-0.211	-5.32	36
Xenon	0.000	0.000	0.000	0.000	0.378	-9.36	8
15-Crown-5	0.410	1.200	0.000	1.750	6.770	-75.76	107,108
Pentadecane							
Pentadecane	0.000	0.000	0.000	0.000	7.209	-74.50	10
1-Chloronaphthalene	1.417	1.000	0.000	0.140	5.856	-62.39	93
Xenon	0.000	0.000	0.000	0.000	0.378	-9.34	8
Hexadecane							
Methane	0.000	0.000	0.000	0.000	-0.323	-3.97	1
Ethane	0.000	0.000	0.000	0.000	0.492	-11.51	1
Propane	0.000	0.000	0.000	0.000	1.050	-15.94	1
Butane	0.000	0.000	0.000	0.000	1.615	-20.79	1
Pentane	0.000	0.000	0.000	0.000	2.162	-25.94	1
Hexane	0.000	0.000	0.000	0.000	2.668	-31.04	1
Heptane	0.000	0.000	0.000	0.000	3.173	-36.15	1
Octane	0.000	0.000	0.000	0.000	3.677	-41.13	1
Hexadecane	0.000	0.000	0.000	0.000	7.714	-81.38	1
2-Methylpropane	0.000	0.000	0.000	0.000	1.409	-18.74	1
Cyclopentane	0.263	0.100	0.000	0.000	2.477	-27.66	1
Cyclohexane	0.305	0.100	0.000	0.000	2.964	-31.50	1
Ethene	0.107	0.100	0.000	0.070	0.289	-11.17	1
Propene	0.103	0.080	0.000	0.070	0.946	-13.35	1
Acetone	0.179	0.700	0.040	0.490	1.696	-21.42	1
2-Butanone	0.166	0.700	0.000	0.510	2.287	-26.48	1
2-Pentanone	0.143	0.680	0.000	0.510	2.755	-31.05	1
2-Hexanone	0.136	0.680	0.000	0.510	3.286	-35.77	1
2-Heptanone	0.123	0.680	0.000	0.510	3.760	-40.46	1

4-Heptanone	0.110	0.660	0.000	0.510	3.705	-44.89	1
2-Octanone	0.108	0.680	0.000	0.510	4.257	-44.89	1
2-Nonanone	0.119	0.680	0.000	0.510	4.735	-49.37	1
Cyclohexanone	0.403	0.860	0.000	0.560	3.792	-36.48	1
Diethyl ether	0.041	0.250	0.000	0.450	2.015	-25.19	1
Dipropyl ether	0.008	0.250	0.000	0.450	2.954	-34.15	1
Dibutyl ether	0.000	0.250	0.000	0.450	3.924	-43.42	1
Butyl Methyl Ether	0.045	0.250	0.000	0.440	2.658	-30.53	1
Tetrahydrofuran	0.289	0.520	0.000	0.480	2.636	-28.53	1
2-Methyltetrahydrofuran	0.241	0.480	0.000	0.530	2.820	-30.78	1
Dichloromethane	0.387	0.570	0.100	0.050	2.019	-23.18	1
Chloroform	0.425	0.490	0.150	0.020	2.480	-28.07	1
Carbon tetrachloride	0.458	0.380	0.000	0.000	2.823	-30.92	1
1-Chlorobutane	0.210	0.400	0.000	0.100	2.722	-30.88	1
Tetrachloroethene	0.639	0.440	0.000	0.000	3.584	-38.41	1
Nitromethane	0.313	0.950	0.060	0.310	1.892	-25.36	1
2-Nitropropane	0.216	0.920	0.000	0.330	2.550	-30.15	1
Acetonitrile	0.237	0.900	0.070	0.320	1.739	-19.08	1
Methanol	0.278	0.440	0.430	0.470	0.970	-13.35	1
Ethanol	0.246	0.420	0.370	0.480	1.485	-16.32	1
1-Propanol	0.236	0.420	0.370	0.480	2.031	-21.17	1
2-Propanol	0.212	0.360	0.330	0.560	1.764	-22.38	1
1-Butanol	0.224	0.420	0.370	0.480	2.601	-28.07	1
1-Pentanol	0.219	0.420	0.370	0.480	3.106	-31.34	1
1-Hexanol	0.210	0.420	0.370	0.480	3.610	-39.79	1
Heptan-1-ol	0.211	0.420	0.370	0.480	4.115	-44.43	1
1-Octanol	0.199	0.420	0.370	0.480	4.619	-49.07	1
1-Nonanol	0.193	0.420	0.370	0.480	5.124	-52.81	1
1-Undecanol	0.181	0.420	0.370	0.480	6.130	-61.39	1
tert-Butanol	0.180	0.300	0.310	0.600	1.963	-23.01	1
Cyclohexanol	0.460	0.540	0.320	0.570	3.758	-47.53	1
Ethyl acetate	0.106	0.620	0.000	0.450	2.314	-27.99	1
Butyl acetate	0.071	0.600	0.000	0.450	3.353	-38.49	1
Methyl benzoate	0.733	0.850	0.000	0.460	4.704	-48.37	1
Benzene	0.610	0.520	0.000	0.140	2.786	-30.38	1
Toluene	0.601	0.520	0.000	0.140	3.325	-35.90	1
Ethylbenzene	0.613	0.510	0.000	0.150	3.778	-40.12	1
Propylbenzene	0.604	0.500	0.000	0.150	4.230	-44.14	1
m-Xylene	0.623	0.520	0.000	0.160	3.839	-41.37	1
p-Xylene	0.613	0.520	0.000	0.160	3.839	-41.51	1
Mesitylene	0.649	0.520	0.000	0.190	4.344	-46.56	1
Acetophenone	0.818	1.010	0.000	0.480	4.501	-47.36	1
Anisole	0.710	0.750	0.000	0.290	3.890	-41.24	1
Benzaldehyde	0.820	1.000	0.000	0.390	4.008	-41.17	1
Benzonitrile	0.742	1.110	0.000	0.330	4.039	-41.25	1
Chlorobenzene	0.718	0.650	0.000	0.070	3.657	-38.24	1

Fluorobenzene	0.477	0.570	0.000	0.100	2.788	-31.05	1
1,4-Difluorobenzene	0.384	0.600	0.000	0.060	2.766	-32.19	1
Aniline	0.955	0.960	0.260	0.410	3.934	-41.80	1
Nitrobenzene	0.871	1.110	0.000	0.280	4.557	-45.65	1
N,N-Dimethylaniline	0.957	0.810	0.000	0.410	4.701	-48.36	1
Pyridine	0.631	0.840	0.000	0.520	3.022	-32.64	1
2-Methylpyridine	0.598	0.750	0.000	0.580	3.422	-35.90	1
3-Methylpyridine	0.631	0.810	0.000	0.540	3.631	-37.49	1
4-Methylpyridine	0.630	0.820	0.000	0.540	3.640	-36.78	1
Propylamine	0.225	0.350	0.160	0.610	2.141	-23.97	1
Butylamine	0.224	0.350	0.160	0.610	2.618	-29.41	1
Pentylamine	0.211	0.350	0.160	0.610	3.139	-34.81	1
Hexylamine	0.197	0.350	0.160	0.610	3.655	-39.46	1
Heptylamine	0.197	0.350	0.160	0.610	4.153	-45.27	1
Nonylamine	0.187	0.350	0.160	0.610	5.100	-55.03	1
Decylamine	0.182	0.350	0.160	0.610	5.605	-60.02	1
tert-Butylamine	0.121	0.290	0.160	0.710	2.493	-26.15	1
Diethylamine	0.154	0.300	0.080	0.690	2.395	-24.60	1
Triethylamine	0.101	0.150	0.000	0.790	3.040	-34.14	1
Helium	0.000	0.000	0.000	0.000	-1.741	8.24	1
Neon	0.000	0.000	0.000	0.000	-1.575	6.78	1
Argon	0.000	0.000	0.000	0.000	-0.688	-0.79	1
Krypton	0.000	0.000	0.000	0.000	-0.211	-5.02	1
Xenon	0.000	0.000	0.000	0.000	0.378	-10.08	1
Radon	0.000	0.000	0.000	0.000	0.877	-14.18	1
Hydrogen	0.000	0.000	0.000	0.000	-1.200	4.56	1
Diiodomethane	0.714	0.690	0.110	0.070	2.886	-38.95	1
2,2,2-Trifluoroethanol	0.015	0.600	0.570	0.250	1.224	-20.88	1
1,1,1,3,3,3-Hexafluoropropan-2-ol	-0.240	0.550	0.770	0.100	1.392	-22.09	1
Benzyl alcohol	0.803	0.870	0.330	0.560	4.221	-42.38	1
Thiophene	0.687	0.560	0.000	0.150	2.819	-29.92	1
Benzyl chloride	0.821	0.820	0.000	0.330	4.384	-43.32	1
1-Chloronaphthalene	1.417	1.000	0.000	0.140	5.856	-61.74	1
1-Methylnaphthalene	1.337	0.940	0.000	0.220	5.802	-55.80	1
Diisopropyl ether	-0.060	0.160	0.000	0.580	2.530	-30.67	1
Methyl tert-amyl ether	0.050	0.210	0.000	0.600	2.916	-34.12	1
2,2-Dimethylpropane	0.000	0.000	0.000	0.000	1.820	-21.14	1

References to Table S1.

1. Literature references for the heptane and hexadecane data are listed in C. Mintz, M. Clark, K. Burton, W. E. Acree, Jr., M. H. Abraham, Enthalpy of solvation correlations for gaseous solutes dissolved in benzene and in alkane solvents based on the Abraham model. QSAR & Comb. Sci., accepted for publication.
2. Chaar, M.; Ortega, J.; Toledo-Marante, F. J.; Gonzalez, C. Thermodynamic properties of (a pentyl ester + a n-alkane). XIV. The H_m^E and V_m^E for (an ester + a n-alkane). J. Chem. Thermodyn. (2001) 33, 689-710.
3. Toledo-Marante, F. J.; Ortega, J.; Chaar, M.; Vidal, M. Thermodynamic properties of (a butyl ester + an n-alkane). XIII. H_m^E and V_m^E for $\{x\text{CH}_3(\text{CH}_2)_u\text{-CO}_2(\text{CH}_2)_3\text{CH}_3 + (1-x)\text{CH}_3(\text{CH}_2)_{2v+1}\text{CH}_3\}$, where $u = 1$ to 3 and $v = 1$ to 7. J. Chem. Thermodyn. (2000) 32, 1013-1036.
4. Ortega, J.; Vidal, M.; Toledo-Marante, F. J.; Placido, J. Thermodynamic properties of (a propyl ester + an n-alkane). XII. Excess molar enthalpies and excess molar volumes for $\{x\text{CH}_3(\text{CH}_2)_{u-1}\text{COO}(\text{CH}_2)_2\text{CH}_3 + (1-x)\text{CH}_3(\text{CH}_2)_{2v+1}\text{CH}_3\}$ with $u = (1$ to 3), and $v = (1$ to 7). J. Chem. Thermodyn. (1999) 31, 1025-1044.
5. Vidal, M.; Ortega, J.; Placido, J. Thermodynamic properties of (an ethyl ester + an n-alkane) IX. H_m^E and V_m^E for $\{x\text{CH}_3(\text{CH}_2)_u\text{COOCH}_2\text{CH}_3 + (1-x)\text{CH}_3(\text{CH}_2)_{2v+1}\text{CH}_3\}$ with $u = 0$ to 5, and $v = 1$ to 7. J. Chem. Thermodyn. (1997) 29, 47-74.

6. Placido, J.; Ortega, J.; Toledo, F. J. Thermodynamic properties of (an ethyl ester + an n-alkane)X. Mixing enthalpies of (methyl, or ethyl acrylate + an n-alkane). *J. Chem. Thermodyn.* (1998) 30, 805-813.
7. Ortega, J.; Placido, J. Excess enthalpies of 1,1,2,2-tetrachloroethane + n-alkanes. Measurement and comparison with the DISQUAC model. *ELDATA: Int. Elect. J. Physico-Chem. Data* (1995) 1, 69-77.
8. Pollack, G. L.; Himm, J. F. Solubility of xenon in liquid n-alkanes: temperature dependence and thermodynamic functions. *J. Chem. Phys.* (1982) 77, 3221-3229.
9. Urdaneta, O.; Hamam, S.; Handa, Y. P.; Benson, G. C. Thermodynamic properties of binary mixtures containing ketones. IV. Excess enthalpies of acetophenone + an n-alkane and phenylacetone + an n-alkane. *J. Chem. Thermodyn.* (1979) 11, 851-856.
10. Chickos, J. S.; Acree, W. E., Jr. Enthalpies of Vaporization of Organic and Organometallic Compounds, 1880-2002. *J. Phys. Chem. Ref. Data* (2003) 32, 519-878.
11. Abraham, M. H. Free energies, enthalpies, and entropies of solution of gaseous nonpolar nonelectrolytes in water and nonaqueous solvents. The hydrophobic effect. *J. Amer. Chem. Soc.* (1982) 104, 2085-2094.
12. Hamam, S. E. M.; Benson, G. C. Excess enthalpies of binary mixtures of n-decane with hexane isomers. *J. Chem. Eng. Data* (1986) 31, 45-47.
13. Marsh, K. N.; Ott, J. B.; Richards, A. E. Excess enthalpies, excess volumes, and excess Gibbs free energies for n-hexane + n-undecane at 298.15 and 308.15 K. *J. Chem. Thermodyn.* (1980) 12, 897-902.

14. Peng, D.-Y.; Benson, G. C.; Lu, B. C.-Y. Excess enthalpies of {2,2-dimethylbutane + hexane + (octane or dodecane)} at the temperature 298.15 K. *J. Chem. Thermodyn.* (2000) 32, 539-549.
15. Peng, D.-Y.; Horikawa, Y.; Wang, Z.; Benson, G. C.; Lu, B. C.-Y. Excess enthalpies of 2,2,4-trimethylpentane + n-alkane binary mixtures at 298.15 K. *J. Chem. Eng. Data* (2001) 46, 237-238.
16. Hamam, S. E. M.; Kumaran, M. K.; Zhang, D.; Benson, G. C. Excess enthalpies of binary mixtures of 2,4-dimethylpentane with n-hexane, n-heptane, n-octane and n-dodecane. *J. Chem. Eng. Data* (1985) 30, 222-224.
17. Uruska, I.; Koschmidder, M. Enthalpies of transfer of pyridines from the gas phase to solution in weakly and moderately polar aprotic solvents. *J. Chem. Soc., Perkin Trans. 2* (1989) 1845-1848.
18. Figeys, D.; Koschmidder, M.; Benoit, R. L. Enthalpies of solution of naphthalene, N,N-dimethyl-1-naphthylamine, and 1,8-bis(dimethylamino)-naphthalene in 16 organic solvents. *Can. J. Chem.* (1992) 70, 1586-1589.
19. Kumaran, M. K.; Halpin, C. J.; Benson, G. C. Limiting excess partial molar enthalpies of hexan-1-ol, 2-methylpentan-1-ol, and 2-ethylbutan-1-ol in n-hexane at 298.15 K. *J. Chem. Thermodyn.* (1983) 15, 249-52.
20. Benson, G. C.; Handa, Y. P. Excess enthalpy. 2-Pentanone-hexane system. *Int. DATA Ser., Sel. Data Mix., Ser. A* (1980) 72.
21. Benson, G. C.; Handa, Y. P. Excess enthalpy. 3-Pentanone-hexane system. *Int. DATA Ser., Sel. Data Mix., Ser. A* (1980) 77.

22. Benson, G. C. Excess enthalpy. Butyl methyl ether-hexane system. *Int. DATA Ser., Sel. Data Mix., Ser. A* (1990) 296.
23. Grolier, J. P. E.; Kiyohara, O.; Benson, G. C. Thermodynamic properties of binary mixtures containing ketones. II. Excess enthalpies of some aromatic ketones + n-hexane, + benzene, and + tetrachloromethane. *J. Chem. Thermodyn.* (1977) 9, 697-703.
24. Airoidi, C.; Roca, S. Standard molar enthalpies of solvation of aliphatic esters in 1,2-dichloroethane and n-hexane. *J. Solution Chem.* (1993) 22, 707-713.
25. Kechavarz, R.; Dubes, J. P.; Tachoire, H. Excess enthalpy. 3-Methylpyridine - hexane system. *Int. DATA Ser., Sel. Data Mix., Ser. A* (1992) 18.
26. Kechavarz, R.; Dubes, J. P.; Tachoire, H. Excess enthalpy. 4-Methylpyridine-hexane system. *Int. DATA Ser., Sel. Data Mix., Series A* (1992) 25.
27. J. J. Christensen, R. L. Rowley, R. M. Izatt, *Handbook of Heats of Mixing: Supplementary Volume*, John Wiley and Sons, Inc., New York, NY, 524 (1988).
28. E. R. Thomas, B. A. Newman, G. L. Nicolaidis, and C. A. Eckert, Limiting activity coefficients from differential ebulliometry. *J. Chem. Eng. Data* (1982) 17, 233-240.
29. Solomonov, B. N.; Antipin, I. S.; Gorbachuk, V. V.; Konovalov, A. I. Investigation of solvation effects in organic reactions using figures for the enthalpies of dissolution. *Dokl. Akad. Nauk SSSR* (1978) 243, 1499-1502.
30. Solomonov, B. N.; Borisover, M. D.; Konovalov, A. I. Enthalpy of solvation of nonelectrolyte organic compounds in associated solutions. *Russ. J. Gen. Chem.* (1985) 56, 1-11.

31. Shaw, A. W.; Vosper, A. J. Solubility of nitric oxide in aqueous and nonaqueous solvents. *J. Chem. Soc., Faraday Trans. 1* (1977) 73, 1239-1244.
32. Vrbka, P.; Hauge, B.; Frydendal, L.; Dohnal, V. Limiting activity coefficients of lower 1-alkanols in n-alkanes: variation with chain length of solvent alkane and temperature. *J. Chem. Eng. Data* (2002) 47, 1521-1525.
33. Dohnal, V.; Vrbka, P. Limiting activity coefficients in the 1-alkanol + n-alkane systems: survey, critical evaluation and recommended values, interpretation in terms of association models. *Fluid Phase Equilibr.* (1997) 1331, 73-87.
34. Wilhelm, E. Enthalpy of mixing of 1,2,4-trichlorobenzene with n-alkanes: a novel effect. *Ber. Bunsen-Ges.* (1977) 81, 1150-1154.
35. Wilhelm, E.; Inglese, A.; Grolier, J. P. E.; Kehiaian, H. V. Enthalpy of mixing of bromobenzene with n-alkanes, with cyclohexane, and with benzene. *Thermochim. Acta* (1979) 31, 85-92.
36. Wilhelm, E.; Battino, R. Thermodynamic functions of the solubilities of gases in liquids at 25°. *Chem. Rev.* (1973), 73, 1-10.
37. Klofutar, C.; Paljk, S.; Domanska, U. Heats of solution of 1-nonanol and 1-undecanol in n-hexane, n-heptane, n-decane and n-hexadecane at 298.15 K. *Thermochimica Acta* (1990), 158, 301-315.
38. Guidry, R. M.; Drago, R. S. Extension of models for evaluating solvent transfer enthalpies. *J. Phys. Chem.* (1974) 78, 454-459.
39. Morel-Desrosiers, N.; Morel, J. P. Standard molar enthalpies, volumes, and heat capacities of adamantane in cyclohexane, n-hexane, and carbon tetrachloride.

- Interpretation using the scaled-particle theory. *J. Solution Chem.* (1979) 8, 579-592.
40. Korolev, V. P.; Smirnova, N. L.; Batov, D. V. Calorimetric study of the solvation of benzene and its chloro-, nitro-, and aminoderivatives by n-hexane, 1-butanol, and mixed solvent n-hexane-1-butanol. *Russ. J. Appl. Chem.* (2006) 79, 213-218.
 41. Ferino, I.; Marongiu, B.; Solinas, V.; Torrazza, S. Excess enthalpies of some aromatic aldehydes in n-hexane, n-heptane and benzene. *Thermochim. Acta* (1982) 57, 147-154.
 42. Sastry, N. V.; Patel, S. R.; Prasad, D. H. L. Measurements and theoretical analysis of excess enthalpies and excess volumes of methyl methacrylate + n-alkanes (n-pentane, n-hexane, n-heptane, n-decane and n-dodecane). *Thermochim. Acta* (2000) 359, 169-180.
 43. McLure, I. A.; Trejo Rodriguez, A. Excess functions for (n-alkanenitrile + n-alkane) liquid mixtures. 2. Excess enthalpies at 298.15 K for propanenitrile and n-butanenitrile with some C5 to C14 n-alkanes. *J. Chem. Thermodyn.* (1982) 14, 439-445.
 44. Mato, M. M.; Cebreiro, S. M.; Verdes, P. V.; Legido, J. L.; Paz Andrade, M. I. Thermodynamic properties of the ternary system MTBE+1-propanol+hexane application of different group contribution models and empirical methods. *J. Therm. Anal. Calorim.* (2005) 80, 303-309.

45. Wang, Z.; Benson, G. C.; Lu, B. C.-Y. Excess enthalpies of 2-methyltetrahydrofuran + n-alkane binary mixtures at 298.15 K. *J. Chem. Eng. Data* (2001) 46, 1188-1189.
46. Peng, D.-Y.; Wang, Z.; Benson, G. C.; Lu, B. C.-Y. Excess enthalpies and (vapour + liquid) equilibrium for (oxygenated additive + n-alkane). *J. Chem. Thermodyn.* (2001) 33, 83-93.
47. Wang, Z.; Benson, G. C.; Lu, B. C.-Y. Excess enthalpies of the ternary mixtures: tetrahydrofuran + (hexane or cyclohexane) + decane at 298.15 K. *J. Chem. Eng. Data* (2003) 48, 190-194.
48. Alonso, C.; Chamorro, C. R.; Segovia, J. J.; Martin, M. C.; Montero, E. A.; Villamanan, M. A. Excess enthalpies of binary and ternary mixtures containing tert-amyl methyl ether (TAME), tert-amyl alcohol (TAOH) and hexane at 298.15 and 313.15 K. *Fluid Phase Equilibr.* (2004) 217, 145-155.
49. Guillen, M. D.; Gutierrez Losa, C. Excess enthalpies and excess volumes of n-hexane and of tetrachloromethane with furan, 1,4-dioxane, tetrahydrofuran, and tetrahydropyran. *J. Chem. Thermodyn.* (1978) 10, 567-76.
50. K. V. N. Suresh Reddy, Y. V. L. Ravi Kumar, D. H. L. Prasad, and A. Krishnaiah, Vapor-Liquid Equilibria and Excess Enthalpies for Binary Systems of Dimethoxymethane with Hydrocarbons. *J. Chem. Eng. Data* (2006) 51, 326-329.
51. Treszczanowicz, T.; Benson, G. C.; Lu, B. C. Y. Excess enthalpies for binary mixtures of 2,5,8-trioxanonane or 2,5,8,11,14-pentaoxapentadecane with n-alkanes at 298.15 K. *J. Chem. Eng. Data* (1988) 33, 379-81.

52. Casas, H.; Segade, L.; Franjo, C.; Jimenez, E.; Paz Andrade, M. I. Excess molar enthalpies of propyl propanoate + hexane + benzene at 298.15 K and 308.15 K. *J. Chem. Eng. Data* (2000) 45, 445-449.
53. Kwaterski, Matthias; Rezanova, Elena N.; Lichtenthaler, R. N. Excess molar volumes and excess molar enthalpies of binary and ternary mixtures of (ethanol or 1-butanol), triethylamine and n-hexane. *Fluid Phase Equilibr.* (2005) 237, 170-185.
54. Fernandez, J.; Velasco, I.; Otin, S. Excess enthalpy. Pentylamine-hexane system. *Int. DATA Ser., Sel. Data Mix., Ser. A* (1987) 206.
55. Fernandez, J.; Velasco, I.; Otin, S. Excess enthalpy. Hexylamine-hexane system. *Int. DATA Series, Sel. Data Mix., Ser. A* (1987) 207.
56. Fernandez, J.; Velasco, I.; Otin, S. Excess enthalpy. Heptylamine-hexane system. *Int. DATA Ser., Sel. Data Mix., Ser. A* (1987) 208.
57. Fernandez, J.; Velasco, I.; Otin, S. Excess enthalpy. Octylamine-hexane system. *Int. DATA Ser., Sel. Data Mix., Ser. A* (1987) 209.
58. Fernandez, J.; Velasco, I.; Otin, S. Excess enthalpy. Decylamine-hexane system. *Int. DATA Ser., Sel. Data Mix., Ser. A* (1987) 211.
59. Marsh, K. N. Excess enthalpies and excess volumes of nitromethane +, and nitroethane + each of several non-polar liquids. *J. Chem. Thermodyn.* (1985) 17, 29-42.
60. Marsh, K. N.; Allan, W. A.; Richards, A. E. Excess enthalpies and excess volumes of 1-nitropropane +, and 2-nitropropane +, each of several non-polar liquids. *J. Chem. Thermodyn.* (1984) 16, 1107-1120.

61. Sapei, E.; Zaytseva, A.; Uusi-Kyyny, P.; Keskinen, K. I.; Aittamaa, J. Vapor-liquid equilibrium for binary system of diethyl sulfide + n-hexane at (338.15 and 323.15) K and diethyl sulfide + 1-hexene at (333.15 and 323.15) K. *J. Chem. Eng. Data* (2007) 52, 571-576.
62. Chaudhari, S. K.; Katti, S. S. Excess enthalpies of n-alcohols (C1-C4) and n-alkanes (C6-C8) with 1,2-dichloroethane at 303.15 K. *Thermochim. Acta* (1990) 158, 99-106.
63. Sharma, B. R.; Pundeer, G. S.; Singh, P. P. Thermodynamics of weak interactions. Excess enthalpies and excess Gibbs free energies of mixing. *Thermochim. Acta* (1975) 11, 105-114.
64. Inglese, A.; Grolier, J. P. E. Excess heat capacity. 1-Chloronaphthalene-hexane system. *Int. DATA Ser., Sel. Data Mix., Ser. A* (1987) 299.
65. Grolier, J. P. E.; Ballet, D.; Viallard, A. Thermodynamics of ester containing mixtures. Excess enthalpies and excess volumes for alkyl acetates and alkyl benzoates + alkanes, + benzene, + toluene, and + ethylbenzene. *J. Chem. Thermodyn.* (1974) 6, 895-908.
66. Kechavarz, R.; Dubes, J. P.; Tachoire, H. Excess enthalpy. 3-Methylpyridine-heptane system. *Int. DATA Ser., Sel. Data Mix., Ser. A* (1992) 19.
67. Kechavarz, R.; Dubes, J. P.; Tachoire, H. Excess enthalpy. 3-Methylpyridine-octane system. *Int. DATA Ser., Sel. Data Mix., Ser. A* (1992) 20.
68. Kechavarz, R.; Dubes, J. P.; Tachoire, H. Excess enthalpy. Pyridine-heptane system. *Int. DATA Ser., Sel. Data Mix., Ser. A* (1992) 12.

69. Kechavarz, R.; Dubes, J. P.; Tachoire, H. Excess enthalpy. 4-Methylpyridine-heptane system. *Int. DATA Series, Sel. Data Mix., Ser. A* (1992) 26.
70. Kechavarz, R.; Dubes, J. P.; Tachoire, H. Excess enthalpy. Pyridine-octane system. *Int. DATA Ser., Sel. Data Mix., Ser. A* (1992) 13.
71. Kechavarz, R.; Dubes, J. P.; Tachoire, H. Excess enthalpy. 4-Methylpyridine-octane system. *Int. DATA Ser., Sel. Data Mix., Ser. A* (1992) 27.
72. Kechavarz, R.; Dubes, J. P.; Tachoire, H. Excess enthalpy. Pyridine-decane system. *Int. DATA Ser., Sel. Data Mix., Ser. A* (1992) 14.
73. Kechavarz, R.; Dubes, J. P.; Tachoire, H. Excess enthalpy. 3-Methylpyridine-decane system. *Int. DATA Ser. Sel. Data Mix., Ser. A* (1992) 21.
74. Kechavarz, R.; Dubes, J. P.; Tachoire, H. Excess enthalpy. 4-Methylpyridine-decane system. *Int. DATA Ser., Sel. Data Mix., Series A* (1992) 28.
75. Kechavarz, R.; Dubes, J. P.; Tachoire, H. Excess enthalpy. Pyridine-dodecane system. *Int. DATA Ser., Sel. Data Mix., Ser. A* (1992) 15.
76. Kechavarz, R.; Dubes, J. P.; Tachoire, H. Excess enthalpy. 3-Methylpyridine-dodecane system. *Int. DATA Ser., Sel. Data Mix., Ser. A* (1992) 22.
77. Kechavarz, R.; Dubes, J. P.; Tachoire, H. Excess enthalpy. 4-Methylpyridine-dodecane system. *Int. DATA Ser., Sel. Data Mix., Ser. A* (1992) 29.
78. Hayduk, W.; Walter, E. B.; Simpson, P. Solubility of propane and carbon dioxide in heptane, dodecane, and hexadecane. *J. Chem. Eng. Data* (1972) 17, 59-61.
79. Kumaran, M. K.; Benson, G. C. Excess enthalpies of n-dodecane + n-heptane, + n-octane, and + n-decane at 298.15 K. *J. Chem. Thermodyn.* (1986) 18, 993-996.

80. Wang, L.; Benson, G. C.; Lu, B. C. Y. Excess enthalpies for (di-n-propyl ether + n-alkane) at 298.15 K. *J. Chem. Thermodyn.* (1988) 20, 975-979.
81. Wang, L.; Benson, G. C.; Lu, B. C. Y. Excess enthalpies for (n-butyl methyl ether + n-alkane) at 298.15 K. *J. Chem. Thermodyn.* (1990) 22, 173-179.
82. Vrbka, P.; Dohnal, V.; Arlt, W. Limiting activity coefficients by comparative tensimetry: 1-propanol and 1-butanol in neptane and in octane. *J. Chem. Eng. Data* (2004) 49, 867-871.
83. K. Fukuchi, K. Miyoshi and Y. Arai, Measurement and correlation of infinite dilution activity coefficients of ethers in alkanes. *Fluid Phase Equilibr.* (1997) 136, 135-139.
84. Castro, I.; Pintos, M.; Amigo, A.; Bravo, R.; Paz Andrade, M. I. Excess enthalpies of (tetrahydrofuran or tetrahydropyran + an n-alkane) at the temperature 298.15 K. *J. Chem. Thermodyn.* (1994) 26, 29-33.
85. Jimenez, E.; Franjo, C.; Segade, L.; Legido, J. L.; Paz Andrade, M. I. Excess molar enthalpies for di-n-butyl ether + 1-propanol + n-octane at 298.15 K. *Fluid Phase Equilibr.* (1997) 133, 179-185.
86. Domanska, U.; Domanski, K.; Klofutar, C.; Paljk, S. Excess enthalpies of nonan-1-ol and undecan-1-ol with octane at high dilutions and at 298.15 K. *Thermochim. Acta* (1990) 164, 227-236.
87. Hamoudi, Z.; Belaribi, F. B.; Ait-Kaci, A.; Boukais-Belaribi, G. Experimental and predicted excess molar enthalpies for 1,4-dioxane + octane + cyclohexane at 303.15K. *Fluid Phase Equilibr.* (2006) 244, 62-67.

88. H. Nakai, H. Soejima, K. Tamura, H. Ogawa, S. Murakami and Y. Toshiyasu, Thermodynamic properties of 2,5,8-trioxanonane + n-octane mixture at 298.15 K. *Thermochim. Acta* (1991) 183, 15-27.
89. Wang, Z.; Benson, G. C.; Lu, B. C.-Y. Excess enthalpies of binary mixtures of 1-hexene with some n-alkanes at 298.15 K. *J. Chem. Eng. Data* (2004) 49, 311-312.
90. Inglese, A.; Grolier, J. P. E. Excess enthalpy. 1-Chloronaphthalene-octane system. *Int. DATA Ser., Sel. Data Mix., Ser. A* (1987) 288.
91. Inglese, A.; Grolier, J. P. E. Excess enthalpy. 1-Chloronaphthalene-decane system. *Int. DATA Ser., Sel. Data Mix., Ser. A* (1987) 289.
92. Inglese, A.; Grolier, J. P. E. Excess enthalpy. 1-Chloronaphthalene-dodecane system. *Int. DATA Ser., Sel. Data Mix., Ser. A* (1987) 290.
93. Inglese, A.; Grolier, J. P. E. Excess enthalpy. 1-Chloronaphthalene-pentadecane system. *Int. DATA Ser., Sel. Data Mix., Ser. A* (1987) 291.
94. Wilcock, R. J.; Battino, R.; Danforth, W. F.; Wilhelm, E. Solubilities of gases in liquids. II. The solubilities of helium, neon, argon, krypton, oxygen, nitrogen, carbon monoxide, carbon dioxide, methane, tetrafluoromethane, and sulfur fluoride (SF₆) in n-octane 1-octanol, n-decane, and 1-decanol. *J. Chem. Thermodyn.* (1978) 10, 817-822.
95. Wilhelm, E.; Inglese, A.; Grolier, J. P. E. Excess enthalpies of cycloheptane + n-alkane and cyclooctane + n-alkane. *J. Chem. Eng. Data* (1983) 28, 202-204.

96. M. López, M. I Paz Andrade, J. Peleteiro, J. L. Legido, L. Romaní and E. Pérez Martell, Excess molar enthalpies of the ternary systems pentan-3-one + n-hexane + n-decane and n-tetradecane at 298.15 K. *Thermochim. Acta* (1992) 211, 33-42.
97. Luo, B.; Benson, G. C.; Lu, B. C. Y. Excess enthalpies for (diethyl ether + n-alkane) at 298.15 K. *J. Chem. Thermodyn.* (1988) 20, 267-271.
98. Benson, G. C.; Luo, B.; Lu, Benjamin C. Y. Excess enthalpies of dibutyl ether-n-alkane mixtures at 298.15 K. *Can. J. Chem.* (1988) 66, 531-534.
99. Wang, Z.; Horikawa, Yoshiteru; Benson, G. C.; Lu, B. C.-Y. Excess enthalpies of ethyl tert-butylether + 2,2,4-trimethylpentane + decane or dodecane at 25° C. *J. Solution Chem.* (2001) 30, 401-410.
100. Treszczanowicz, T.; Wang, L.; Benson, G. C.; Lu, B. C. Y. Excess enthalpies for binary mixtures formed by 2,5,8,11-tetraoxadodecane with homologous n-alkanes. *Thermochim. Acta* (1991) 189, 255-259.
101. Luo, B.; Benson, G. C.; Lu, B. C. Y. Excess enthalpies and excess volumes for {2-methylbutan-2-ol + (n-decane or n-hexadecane)}. *J. Chem. Thermodyn.* (1987) 19, 785-789.
102. Woycicka, M. K.; Recko, W. M. Heats of mixing of n-propanol and n-hexanol with n-hydrocarbons at high dilutions. *Bull. l'Acad. Polonaise des Sci., Ser. Sci. Chim.* (1972) 20, 783-788.
103. Wilhelm, E.; Inglese, A.; Lainez, A.; Roux, A. H.; Grolier, J.-P. E. Thermodynamics of (1,4-difluorobenzene + an n-alkane) and of (hexafluorobenzene + an n-alkane). *Fluid Phase Equilib.* (1995) 110, 299-313.

104. Gmehling, J. Excess enthalpies for 1,1,1-trichloroethane with alkanes, ketones, and esters. *J. Chem. Eng. Data* (1993) 38, 143 – 146.
105. Ortega, J.; Espiau, F.; Toledo, F. J. Thermodynamic properties of (an ester + an alkane). XVI. Experimental HmE and VmE values and a new correlation method for (an alkyl ethanoate + an n-alkane) at 318.15 K. *J. Chem. Thermodyn.* (2004) 36, 193-209.
106. Ortega, J.; Matos, J. S.; Pena, J. A. Excess molar enthalpies of methyl alkanoates + n-nonane at 298.15 K. *Thermochimica Acta* (1990) 160, 337-342.
107. Takeda, Y.; Watanabe, T.; Yamada, H.; Katsuta, S. Thermodynamic study on transfer between polar solvents of 15-crown-5, benzo-15-crown-5, and their 1:1 complexes with alkali metal ions. *J. Mol. Liq.* (2003) 108, 151-173.
108. Mintz, C.; Clark, M.; Acree, W. E., Jr.; Abraham, M. H. Enthalpy of solvation correlations for gaseous solutes dissolved in water and in 1-octanol based on the Abraham model. *J. Chem. Inf. Model.* (2007) 47, 115-121.