

# Phase Change Enthalpies and Entropies of Liquid Crystals

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The thermochemical behavior of more than 3000 organic compounds known to form liquid crystals is reported along with references to the original literature. A group additivity approach used to estimate total phase change entropies of organic molecules applied to 627 of these liquid crystals is found to significantly overestimate their total phase change entropies. Comparison of experimental and estimated values also show significant scatter relative to database compounds. The origins of these discrepancies are discussed in terms of a model used to explain liquid crystal formation. © 2006 American Institute of Physics. [DOI: 10.1063/1.1901689]

Key words: fusion enthalpy; fusion entropy; liquid crystals; liquid crystal compendium; phase transitions; total phase change entropy.

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## 1. Introduction

Since their first discovery back in 1888, interest in the properties and practical applications of liquid crystals has increased dramatically.<sup>1,2</sup> General acceptance of liquid crystals as a distinct phase of matter was slow, occurring some 30 years since they were first reported. Liquid crystalline behavior is found among numerous classes of compounds that include biphenyls, cholesterol esters, soaps, lipids, polymers, and elastomers. More than 76 000 compounds have been identified as exhibiting liquid crystalline behavior.<sup>3(a)–(c)</sup> This paper will review the total phase change enthalpies and entropies of the more than 3000 compounds whose condensed phase thermochemical properties have been studied.

Unlike most small molecules that behave isotropically upon liquefaction, many molecules that are highly nonspherical in shape, exhibit marked self-assembly in the liquid phase that persists even upon continued heating. Cylindrical rod, disk, and banana shaped molecular structures are among those most frequently encountered exhibiting this behavior. Loss of self-assembly can occur in stages and can be monitored by changes in a variety of physical properties. In some cases, liquid crystalline behavior is observed, only upon supercooling of the melt. In these instances, the melting of the solid first produces an isotropic liquid; self-association is observed upon supercooling below the melting temperature. Although self-association may be prompted by the supercooling of many nonspherically shaped molecular liquids, the scattering of visible light by liquid crystals requires a higher level of self-association not observed with most substances.

Though liquid crystals can be considered as a separate phase of matter,<sup>2</sup> they exhibit properties intermediate be-

tween anisotropic solids that are rigidly and uniquely arranged in a lattice with very little mobility, plastic crystals that flow under stress and usually characterized by rotational motion within a lattice, and isotropic liquids characterized by free rotational and translational motion. If the forces of interaction are sufficiently strong, a more limited form of self-association can also be detected in the gas phase, as exemplified by the dimerization observed with some carboxylic acids.

Of all the techniques used to study liquid crystals, thermal analysis, while perhaps not the most sensitive, provides a quantitative measure of the magnitude of the interactions responsible for self-assembly. A study of the thermal behavior exhibited by liquid crystals may also provide insight into the associative behavior of other molecules that behave nonisotropically but do not form liquid crystals. This study reports the thermal behavior of some 3000+ liquid crystals and compares the total molar phase change entropy ( $\Delta_0^{T_{\text{iso}}}S_{\text{tpce}}$ ) of a representative number of them to the total molar phase change entropy of substances that are believed to melt to isotropic liquids. In view of the large number of compounds in the database, the calculated and experimental total molar phase change entropy of some 667 entries on approximately 600 different compounds were compared. Compounds were selected to include a variety of functional groups and structures. In order to simplify the calculations, members of homologous series were frequently chosen. If the compounds selected included multiple independent determinations of their thermal properties, all entries for the compound were included in the analysis.

Throughout this article,  $T_{\text{fus}}$ ,  $T_{\text{cld}}$ , and  $T_{\text{iso}}$  are used to distinguish between slightly different events and conditions. The temperature,  $T_{\text{fus}}$ , refers to the temperature at which a solid is converted to either an isotropic liquid or to a liquid crystal.  $T_{\text{cld}}$  is used to refer to the clearing temperature if the isotropic liquid is converted to the liquid crystal by supercooling the isotropic liquid below  $T_{\text{fus}}$ ;  $T_{\text{cld}}$  may be observed experimentally at the temperature the supercooled liquid becomes cloudy. The term  $T_{\text{iso}}$  has been used to refer to temperatures at which the liquid becomes isotropic above  $T_{\text{fus}}$ . The relationship between these terms is defined as follows:  $T_{\text{cld}} < T_{\text{fus}} \leq T_{\text{iso}}$ .

### 1.1. Phase Change Enthalpies

An examination of the phase change enthalpies of liquid crystals reveals that these substances exhibit several thermal transitions that can be detected. For most substances, the largest enthalpic change occurs upon conversion of the solid to a nematic or smectic phase. In a few cases, some highly substituted anthraquinones (for example, see  $C_{124}H_{216}O_{18}$ ,  $C_{132}H_{232}O_{18}$  in Table 10), the largest enthalpy change observed is associated with changes occurring at the mesomorphic stage.

It has been previously shown that the total molar phase change enthalpy ( $\Delta_0^{T_{\text{iso}}}H_{\text{tpce}}$ ) associated in going from the rigid solid to the isotropic liquid at the melting temperature

TABLE 1. Contributions to  $\Delta_0^{T_{\text{iso}}}S_{\text{tpce}}$  by the hydrocarbon portion of acyclic and aromatic molecules

Acyclic and aromatic carbon groups	Group	Group value $G_i$ , (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	Group coefficient $C_i$
primary $sp^3$	$\text{CH}_3-$	17.6	
secondary $sp^3$	$>\text{CH}_2$	7.1	1.31 <sup>a</sup>
tertiary $sp^3$	$-\text{CH}<$	-16.4	0.60
quaternary $sp^3$	$>\text{C}<$	-34.8	0.66
secondary $sp^2$	$=\text{CH}_2$	17.3	
tertiary $sp^2$	$=\text{CH}-$	5.3	0.75
quaternary $sp^2$	$=\text{C}(\text{R})-$	-10.7	
tertiary $sp$	$\text{H}-\text{C}\equiv$	14.9	
quaternary $sp$	$-\text{C}\equiv$	-2.8	
aromatic tertiary $sp^2$	$=\text{C}_a\text{H}-$	7.4	
quaternary aromatic $sp^2$ carbon adjacent to an $sp^3$ atom	$=\text{C}_a(\text{R})-$	-9.6	
peripheral quaternary aromatic $sp^2$ carbon adjacent to an $sp^2$ atom	$=\text{C}_a(\text{R})-$	-7.5	
internal quaternary aromatic $sp^2$ carbon adjacent to an $sp^2$ atom	$=\text{C}_a(\text{R})-$	-0.7	

<sup>a</sup>The group coefficient of 1.31 for  $\text{C}_{\text{CH}_2}$  is applied only when the number of consecutive methylene groups equals or exceeds the sum of the remaining groups; see the discussion in text; R: any alkyl or aryl group unless specified otherwise.

of most organic compounds is not as good of a group property as the corresponding total molar phase change entropy ( $\Delta_0^{T_{\text{iso}}}S_{\text{tpce}}$ ); this is particularly true for molecules that have multiple phase changes occurring at various temperatures intermediate between  $T=0$  K and  $T=T_{\text{fus}}$ .<sup>4</sup> A comparison of  $\Delta_0^{T_{\text{iso}}}S_{\text{tpce}}$  calculated for molecules believed to be isotropic in the liquid phase to the experimental total phase change entropy of molecules forming liquid crystals would be informative with regards to the quantitative nature of the phase changes occurring in liquid crystals.

## 2. Phase Change Entropies

### 2.1. Estimation of Total Phase Change Entropy

A method for estimation of the total phase change entropy of a wide variety of organic molecules has appeared recently.<sup>5</sup> The method is based on group additivity. Only a brief description of the details of the estimations are given here; additional details can be found in the literature.<sup>5,6</sup> In the discussion that follows, it is important to bear in mind that the terms primary, secondary, tertiary, and quaternary, are based solely on the number of hydrogens attached to carbon, 3, 2, 1, 0, respectively. A listing of the group values used in the estimations of the total phase change entropy of the test compounds is provided in Tables 1–5. Group values for sev-

eral functional groups encountered in this study are not currently available.

Table 1 provides the group values used to estimate the acyclic hydrocarbon portion of the molecule. The contributions of the hydrocarbon portion of the molecule are obtained by simply adding up the contribution of each group present. The only exceptions to this rule occur whenever a functional group listed in Tables 2 and 3 is attached directly to either a tertiary or quaternary  $sp^3$  carbon or to a tertiary  $sp^2$  carbon; in this case the contribution of each carbon is attenuated as the product of the group coefficient and group value ( $C_iG_i$ ). Additionally, if the number of consecutive methylene groups equals or exceeds the sum of the remaining groups, the contribution of each  $\text{CH}_2$  is also evaluated as the product of  $C_iG_i$ ; other methylene groups in the molecule are not included in this count and are treated normally.

The contribution of each functional group is treated in much the same way. Functional groups are defined in Tables 2, 3, and 5. The contribution of each function group defined in Table 2 depends on the total number of functional groups present in the molecule. Each of these functional groups contribute  $C_kG_k$ . The appropriate value of  $C_k$  is chosen from the table on the basis of the total number of functional groups

TABLE 2. Contributions of acyclic functional groups used in estimating  $\Delta_0^{T_{\text{iso}}}S_{\text{tpce}}$  of liquid crystals; functional groups dependent on the substitution patterns

Functional groups <sup>a</sup>	Total number of functional groups; $k=$	Group value ( $G_k$ ) <sup>a</sup> (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	Group Coefficient ( $C_k$ )				
			2	3	4	5	6
chlorine	R-Cl	10.8	1.5	1.5	1.5	1.5	1.5
two fluorines on an $sp^3$ carbon	R-CF <sub>2</sub> -R	13.2 <sup>b</sup>	1.06	1.06	1.06	1.06	1.06
hydroxyl group	R-OH	1.7	10.4	9.7	13.1	12.1	13.1
carboxylic acid	R-C(=O)OH	13.4	1.21	2.25	2.25	2.25	2.25

<sup>a</sup>R: any alkyl or aryl group unless specified otherwise.

<sup>b</sup>This group value has been modified from the value reported in an earlier version;<sup>6</sup> use this value for each fluorine; see Table 3 for the fluorine value to use in perfluorinated compounds and Ref. 3 for examples.

TABLE 3. Contributions of the remaining acyclic functional groups used in estimating  $\Delta_0^{T_{iso}}S_{\text{tpcc}}$  of liquid crystals

Functional groups <sup>a</sup>	Abbreviated structure	Group value ( $G_k$ ) <sup>a</sup> ( $\text{J} \cdot \text{K}^{-1} \cdot \text{mol}^{-1}$ )
bromine	<b>R-Br</b>	17.5
fluorine on an $sp^2$ carbon <sup>b</sup>	$\text{R}_2\text{C}=\text{CRF}; \text{R}_2\text{C}=\text{CF}_2$	19.5
fluorine on an aromatic carbon	$=\text{CF}-$	16.6
three fluorines on an $sp^3$ carbon <sup>b</sup>	$\text{CF}_3-\text{R}$	13.2
one fluorine on an $sp^3$ carbon <sup>b</sup>	$\text{R}-\text{CF}-\text{(R)}_2$	12.7
fluorine in perfluorinated compounds <sup>c</sup>	$\text{C}_n\text{F}_{2n+2}$	15.2
one fluorine on a ring carbon <sup>b</sup>	$-\text{CHF}-;$	[17.5]
two fluorines on a ring carbon <sup>b</sup>	$-\text{CF}_2-$	[17.5]
iodine	<b>R-I</b>	19.4
phenol	$=\text{C}-(\text{OH})-$	20.3
ether	$\text{R}-\text{O}-\text{R}$	4.71
aldehyde	$\text{R}-\text{CH}(=\text{O})$	21.5
ketone	$\text{R}-\text{C}(=\text{O})-\text{R}$	4.6
ester	$\text{R}-\text{C}(=\text{O})\text{O}-\text{R}$	7.7
aromatic heterocyclic amine	$=\text{N}-$	[10.9]
acyclic $sp^2$ nitrogen	$=\text{N}-$	[-1.8]
tertiary amine	$\text{R}-\text{N}(\text{R}_2)$	-22.2
secondary amine	$\text{R}-\text{NH}-\text{R}$	-5.3
primary amine	$\text{R}-\text{NH}_2$	21.4
nitro group	$\text{R}-\text{NO}_2$	17.7
azoxy nitrogen	$\text{N}=\text{N}(-\text{O})-$	[6.8]
nitrile	$\text{R}-\text{C}\equiv\text{N}$	17.7
isocyanide	$\text{R}-\text{NC}$	[17.5]
tertiary amides	$\text{R}-\text{C}(=\text{O})\text{NR}_2$	-11.2
secondary amides	$\text{R}-\text{C}(=\text{O})\text{NH}-\text{R}$	1.5
primary amide	$\text{R}-\text{CONH}_2$	27.9
N,N-dialkylformamide	$\text{HC}(=\text{O})\text{NR}_2$	[6.9]
sulfides	$\text{R}-\text{S}-\text{R}$	2.1
disulfides	$\text{R}-\text{SS}-\text{R}$	9.6
thiols	$\text{R}-\text{SH}$	23.0

<sup>a</sup>R: any alkyl or aryl group unless specified otherwise; values in brackets are tentative assignments; all group coefficients can be assumed to be 1; the functional groups are in bold.

<sup>b</sup>The value in column 3 is the contribution of each fluorine.

<sup>c</sup>The contribution of each F in perfluorinated compounds is the same as previously reported (see Chickos and Acree<sup>5</sup>); it is now treated as a new group value.

present. For the functional groups in Table 2, the group coefficient  $C_6$  should be used for all molecules containing more than six functional groups. The contribution of each functional group in Tables 3 and 5 is given by the corresponding

group value.<sup>7</sup> The group coefficient,  $C_k$ , for all of the groups in these tables can be assumed to be one. Equation (1) summarizes estimations of acyclic and aromatic hydrocarbons and their derivatives (aah)

TABLE 4. Contributions of the cyclic hydrocarbon portions of the molecule

Contributions of cyclic carbons	Groups	Group value ( $G_c$ ) ( $\text{J} \cdot \text{K}^{-1} \cdot \text{mol}^{-1}$ )	Group coefficient— $C_c$
cyclic tertiary $sp^3$ carbon	$>\text{C}_c\text{H}(\text{R})$	-14.7	
cyclic quaternary $sp^3$ carbon	$>\text{C}_c(\text{R})_2$	-34.6	
cyclic tertiary $sp^2$ carbon	$=\text{C}_c\text{H}-$	-1.6	1.92
cyclic quaternary $sp^2$ carbon	$=\text{C}_c(\text{R})-$	-12.3	
cyclic quaternary $sp$ carbon	$=\text{C}_c=; \text{R}-\text{C}\equiv$	-4.7	

All group values in this table are only to be used with the ring equations (3) or (4); R: any alkyl or aryl group unless specified otherwise.

TABLE 5. Contributions of the cyclic functional groups used in estimating  $\Delta_0^{T_{\text{iso}}}S_{\text{tpce}}$  of liquid crystals

Heteroatoms and functional groups comprising a portion of a ring <sup>a,b</sup>	Abbreviated structure	Group value ( $G_c$ ) <sup>a</sup> ( $J \cdot K^{-1} \cdot mol^{-1}$ )
cyclic ketone	$R-\mathbf{C}(=\mathbf{O})-R$	-1.4
cyclic $sp^2$ nitrogen	$R=\mathbf{N}-R$	0.5
cyclic tertiary amine	$R_2>\mathbf{N}-R$	-19.3
cyclic secondary amine	$R-\mathbf{NH}-R$	2.2
cyclic sulfide	$R-\mathbf{S}-R$	2.9

<sup>a</sup>R: any alkyl or aryl group unless specified otherwise; values in brackets are tentative assignments; all group coefficients can be assumed to be 1; all group values in this table are only to be used with the ring equations (3) or (4).

<sup>b</sup>The R groups that are a part of the ring structure are designated by italics.

$$\Delta_0^{T_{\text{iso}}}S_{\text{tpce}}(aaH) = \sum_i n_i C_i G_i + n_i C_i G_i + n_i C_i G_{CH_2} \\ + \sum_k n_k C_k G_k. \quad (1)$$

Values of  $\Delta_0^{T_{\text{iso}}}S_{\text{tpce}}$  for nonbenzenoid cyclic hydrocarbons are calculated using Eq. (2)

$$\Delta_0^{T_{\text{iso}}}S_{\text{tpce}}(c) = \Delta S_{\text{ring}} + \sum_c n_i C_c G_c + \sum_c n_i C_i G_i \\ + n_i C_i G_{CH_2} + \sum_k n_k C_k G_k. \quad (2)$$

The contribution of the ring,  $\Delta S_{\text{ring}}$ , is based on size and is given by Eqs. (3) and (4):

ring equation for nonaromatic cyclic compounds

$$\Delta S_{\text{ring}} = [33.4] + [3.7][n - 3]; \\ n = \text{number of ring atoms} \quad (3)$$

ring equation for nonaromatic polycyclic compounds

$$\Delta S_{\text{ring}} = [33.4]N + [3.7][RA - 3N]; \\ RA = \text{total number of ring atoms}; \\ N = \text{number of rings} \quad (4)$$

for cyclic and polycyclic compounds, respectively. The contributions of any ring carbons atoms differing in substitution or hybridization from secondary  $sp^3$  are adjusted by adding the contributions listed in Table 4 for typical substitution and hybridization patterns. If any of the functional groups listed in Tables 2, 3, or 5 are also attached to a cyclic tertiary  $sp^2$  carbon, then the contribution of the cyclic tertiary  $sp^2$  carbon is modified as  $C_c G_c$ . Functional groups that make up part of a ring are treated in a similar fashion. The group coefficient for all cyclic groups can also be assumed to be one. If additional acyclic components are attached to the ring, their estimation is included as described above.

## 2.2. Some Estimations of $\Delta_0^{T_{\text{iso}}}S_{\text{tpce}}$ of Liquid Crystals

Some examples of the estimation of  $\Delta_0^{T_{\text{iso}}}S_{\text{tpce}}$  as applied to liquid crystals are illustrated in Table 6. The total phase change entropy of a liquid crystal  $\Delta_0^{T_{\text{iso}}}S_{\text{tpce}}$  is defined as the sum of all the entropy changes associated with phase transitions occurring from  $T=0\text{ K}$  to the clearing temperature,  $T=T_{\text{iso}}$ . Details about the phase transitions observed for these compounds can be found in Table 10 according to their molecular formula.

### 2.2.1. 1,2,13,14-Tetrahydroxytetradecane

This molecule is composed of a total of three different groups, four hydroxy groups, two types of methylene groups, and two tertiary  $sp^3$  carbons. The two terminal methylene groups are isolated and contribute at total of  $2(7.1)\text{ J} \cdot \text{K}^{-1} \cdot \text{mol}^{-1}$ . The methine carbons contribute  $2(-16.4)$  and the ten consecutive methylene groups contribute  $10(7.1)(1.31)$  since only a total of six other groups are present in the molecule. The contribution of the hydroxyl groups depends on the total number of functional groups present in the molecule. In this case there are four; the hydroxyls contribute  $4(1.7)(13.1)\text{ J} \cdot \text{K}^{-1} \cdot \text{mol}^{-1}$ .

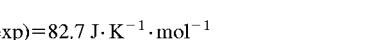
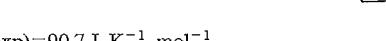
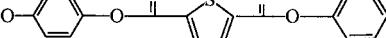
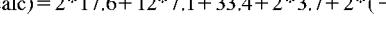
### 2.2.2. 4-Butylcyclohexyl 4-methoxycinnamate

Estimation of the alcohol portion of the ester requires the use of the ring equation. The ring is a six membered ring and therefore contributes  $(33.4 + 3(3.7))$ ; the two cyclic tertiary ring carbons contribute  $2(-14.7)$  and the methyl and methylene groups contribute  $17.6$  and  $3(7.1)\text{ J} \cdot \text{K}^{-1} \cdot \text{mol}^{-1}$ , respectively. The phenyl group of the carboxylic acid portion contains two types of carbons, four tertiary aromatic carbons, and two quaternary aromatic carbons. Since both substituents attached at the quaternary aromatic carbons can be conjugated with the ring, they each contribute  $-7.5\text{ J} \cdot \text{K}^{-1} \cdot \text{mol}^{-1}$ . If the methyl group had been attached directly to the benzene ring, the quaternary aromatic carbon would have contributed  $(-9.6)\text{ J} \cdot \text{K}^{-1} \cdot \text{mol}^{-1}$ . In general, a value of  $-7.5$  is used whenever the group attached to the quaternary aromatic carbon can conjugate with the ring through available empty orbitals or lone pairs of electrons. The contributions of the two tertiary  $sp^2$  carbons and the methyl group round out the contributions of the hydrocarbon backbone. Addition of the contributions of the two functional groups, an ether, and ester complete the estimation.

### 2.2.3. Diphenyl 4,4'-biphenylidicarboxylate

The estimation of this molecule is very similar to the one above. The molecule contains three different groups, tertiary aromatic carbons  $(18(7.4))$ , quaternary aromatic carbons  $(6(-7.5))$  and an ester functional group  $(2(7.7))$ . The value for a quaternary aromatic carbon adjacent to an  $sp^2$  atom is selected because each of them extends the conjugation.

TABLE 6. Application of group additivity toward the estimation of  $\Delta_0^{T_{\text{iso}}} S_{\text{tpcc}}$  of liquid crystals<sup>a</sup>

$\text{CH}_2\text{OHCHOH}(\text{CH}_2)_{10}\text{CHOHCH}_2\text{OH}$	$\text{C}_{14}\text{H}_{30}\text{O}_4$
$\Delta_0^{T_{\text{iso}}}S_{\text{tpce}}(\text{exp})=131.3 \text{ J}\cdot\text{K}^{-1}\cdot\text{mol}^{-1}$	$\Delta_0^{T_{\text{iso}}}S_{\text{tpce}}(\text{calc})=176.6 \text{ J}\cdot\text{K}^{-1}\cdot\text{mol}^{-1}$
$\Delta_0^{T_{\text{iso}}}S_{\text{tpce}}(\text{calc})=2(7.1)+2(-16.4)(0.6)+4(1.7)(13.1)+10(1.31)(7.1)$	
	$\text{C}_{20}\text{H}_{28}\text{O}_3$
$\Delta_0^{T_{\text{iso}}}S_{\text{tpce}}(\text{exp})=82.7 \text{ J}\cdot\text{K}^{-1}\cdot\text{mol}^{-1}$	$\Delta_0^{T_{\text{iso}}}S_{\text{tpce}}(\text{calc})=107.9 \text{ J}\cdot\text{K}^{-1}\cdot\text{mol}^{-1}$
$\Delta_0^{T_{\text{iso}}}S_{\text{tpce}}(\text{calc})=2*17.6+3*7.1+33.4+3*3.7+2*(-14.7)+5.3*0.75+5.3+4*7.4+2(-7.5)+7.7+4.7$	
	$\text{C}_{26}\text{H}_{18}\text{O}_4$
$\Delta_0^{T_{\text{iso}}}S_{\text{tpce}}(\text{exp})=90.7 \text{ J}\cdot\text{K}^{-1}\cdot\text{mol}^{-1}$	$\Delta_0^{T_{\text{iso}}}S_{\text{tpce}}(\text{calc})=103.6 \text{ J}\cdot\text{K}^{-1}\cdot\text{mol}^{-1}$
$\Delta_0^{T_{\text{iso}}}S_{\text{tpce}}(\text{calc})=18*7.4+6(-7.5)+2*7.7$	
	$\text{C}_{32}\text{H}_{40}\text{O}_6\text{S}$
$\Delta_0^{T_{\text{iso}}}S_{\text{tpce}}(\text{exp})=190.3 \text{ J}\cdot\text{K}^{-1}\cdot\text{mol}^{-1}$	$\Delta_0^{T_{\text{iso}}}S_{\text{tpce}}(\text{calc})=190.3 \text{ J}\cdot\text{K}^{-1}\cdot\text{mol}^{-1}$
$\Delta_0^{T_{\text{iso}}}S_{\text{tpce}}(\text{calc})=2*17.6+12*7.1+33.4+2*3.7+2*(-12.3)+2*(-1.6)+8*7.4+4(-7.5)+2*7.7+2*4.7+2.9$	
	$\text{C}_{60}\text{H}_{102}\text{O}_{12}$
$\Delta_0^{T_{\text{iso}}}S_{\text{tpce}}(\text{exp})=291.7 \text{ J}\cdot\text{K}^{-1}\cdot\text{mol}^{-1}$	$\Delta_0^{T_{\text{iso}}}S_{\text{tpce}}(\text{calc})=405 \text{ J}\cdot\text{K}^{-1}\cdot\text{mol}^{-1}$
$\Delta_0^{T_{\text{iso}}}S_{\text{tpce}}(\text{calc})=6*[17.6+7*7.1+7.7]+6*(-7.5)$	

<sup>a</sup>See Tables 10 and 11 for references.

#### 2.2.4. *bis*(4-Heptyloxyphenyl) 2,5-thiophenedicarboxylate

Estimation of the thiophene ring begins by using the ring equation,  $33.4 + 2(3.7)$ . In addition, the cyclic sulfide contributes  $2.9 \text{ J}\cdot\text{K}^{-1}\cdot\text{mol}^{-1}$  with the cyclic tertiary and quaternary  $sp^2$  carbons contributing  $2(-1.6)$  and  $2(-12.3) \text{ J}\cdot\text{K}^{-1}\cdot\text{mol}^{-1}$ , respectively. The contributions of the ester functional groups ( $2(7.7)$ ) complete the estimation of the acid portion of the molecule. The two alcohols contribute methyl groups ( $2(17.6)$ ), methylene groups ( $2(7(7.1))$ ), tertiary aromatic carbons ( $2(4(7.4))$ ), and conjugated quaternary aromatic carbons ( $2(2(-7.5))$ ).

### **2.2.5. Benzene hexa-nanoate**

This molecule is an example of a disk shape molecule in contrast to the estimations of the previous molecules that would be considered more rod shaped. The molecule contains four different types of groups, methyls (6(17.6)), secondary  $sp^3$  carbons (6(7(7.1))), conjugated quaternary aromatic carbons (6(-7.5)), and esters (6(7.7)).

### **2.3. Statistics of the Correlations of Total Phase Change Entropy**

### 2.3.1. Database Compounds

It is immediately apparent from the sample calculations in Table 6 that the experimental values for  $\Delta_0^{T_{\text{iso}}} S_{\text{tpce}}$  are considerably smaller than the values estimated by the protocol just described. The group values used in the estimations were

derived from a large collection of experimental fusion enthalpies of compounds that do not form liquid crystals upon melting. Some idea of how well Eqs. (1)–(4) are capable of estimating  $\Delta_{\text{tpce}}^{T_{\text{iso}}} S$  is illustrated in Fig. 1. This figure com-

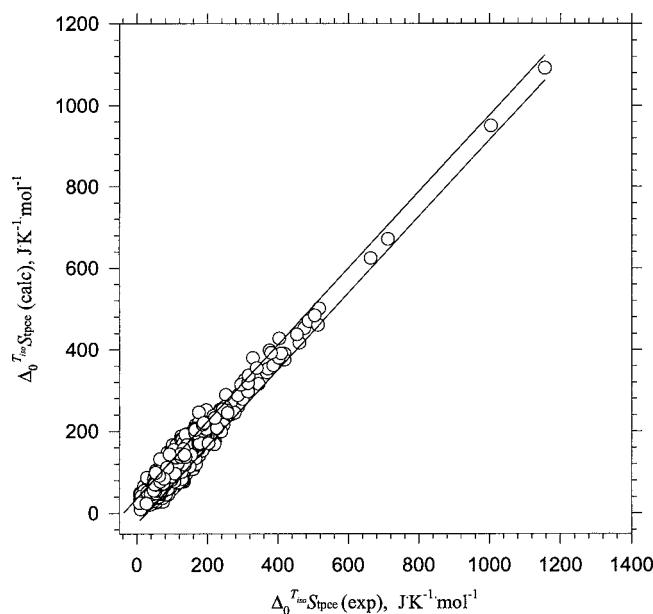


FIG. 1. A comparison of the experimental and calculated total phase change entropies of 2637 compounds.

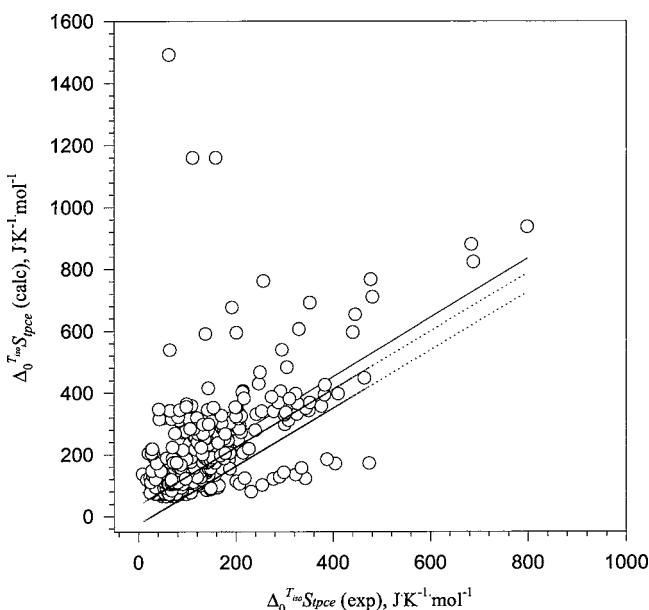


FIG. 2. A comparison of calculated and experimental  $\Delta_0^{T_{iso}}S_{tpce}$  for 627 liquid crystals.

pares calculated and experimental total phase change entropies of some 2637 different database compounds. The correlation equation is given by

$$\begin{aligned} \Delta_0^{T_{iso}}S_{tpce}(\text{calcd})/\text{J}\cdot\text{K}^{-1}\cdot\text{mol}^{-1} \\ = (0.9408 \pm 0.0048)\Delta_0^{T_{iso}}S_{tpce}(\text{exp}) + (4.71 \pm 14.9); \\ r^2 = 0.9370. \end{aligned} \quad (5)$$

The standard deviation was  $\pm 15.3 \text{ J}\cdot\text{K}^{-1}\cdot\text{mol}^{-1}$  and the fractional error was 0.169; the errors of 69 compounds were greater than 3 standard deviations ( $\pm 3\sigma$ ) and were not included in these statistics. The area between the two lines in Fig. 1 represents  $\pm 2\sigma$ .

### 2.3.2. Calculations of $\Delta_0^{T_{iso}}S_{tpce}$ for Liquid Crystals

When the protocol, used to estimate the total phase change entropy of the database compounds, was applied to the estimation of 627 liquid crystals selected from Table 11, the resulting correlation obtained is shown in Fig. 2. Calculated  $\Delta_0^{T_{iso}}S_{tpce}$  values (ordinate) are compared to experimental values (abscissa). The area between the two dotted lines represents  $\pm 2\sigma$ ; the 95% confidence level associated with estimation of the database compounds that do not form liquid crystals. The solid line represents the equation of the line derived from a linear regression analysis of calculated versus experimental values. The statistics of the correlation are given by

$$\begin{aligned} \Delta_0^{T_{iso}}S_{tpce}(\text{calcd})/\text{J}\cdot\text{K}^{-1}\cdot\text{mol}^{-1} \\ = (0.956 \pm 0.052)\Delta_0^{T_{iso}}S_{tpce}(\text{exp}) + (70.1 \pm 112); \\ r^2 = 0.35. \end{aligned} \quad (6)$$

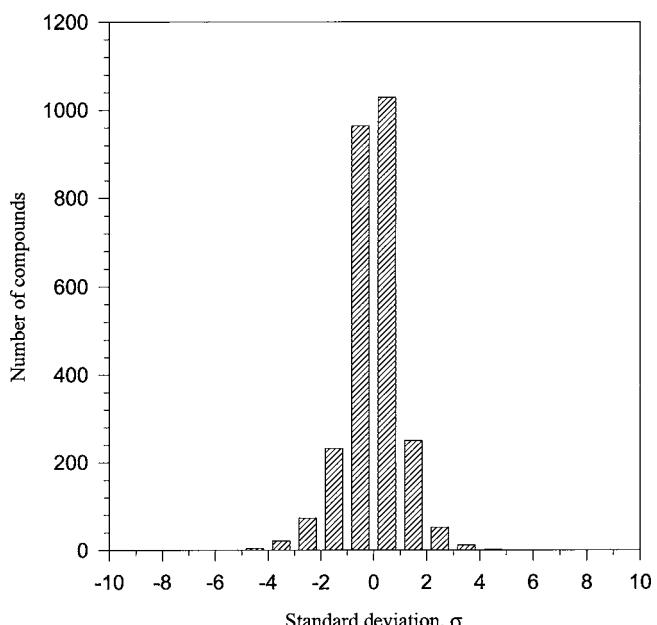


FIG. 3. A histogram of the distribution of errors in  $\Delta_0^{T_{iso}}S_{tpce}(\text{exp}) - \Delta_0^{T_{iso}}S_{tpce}(\text{calc})$  for 2637 compounds used in deriving and validating group parameters for estimating total phase change entropies.

Two trends are immediately evident in this figure; the scatter associated with compounds forming liquid crystals is considerably greater than with those compounds that do not. A standard deviation of  $\pm 112 \text{ J}\cdot\text{K}^{-1}\cdot\text{mol}^{-1}$  is obtained when comparing calculated and experimental total phase change entropies. In addition, the total phase change entropy of compounds forming liquid crystals is greatly overestimated by this protocol. These two trends are also evident by comparing Figs. 3 and 4. Figure 3 is a histogram of the errors associated with the database compounds. Each bar represents one standard deviation  $15.3 \text{ J}\cdot\text{K}^{-1}\cdot\text{mol}^{-1}$ . The errors associated with similar estimations for liquid crystals are represented graphically in Fig. 4. How the magnitude of this error is influenced by the size of the molecule is illustrated in Fig. 5. Figure 5 compares the difference between calculated and experimental values. It appears from this figure that the error is somewhat dependent on size.

Estimations for approximately one third of the liquid crystals fall within  $\pm 2\sigma$  ( $30.6 \text{ J}\cdot\text{K}^{-1}\cdot\text{mol}^{-1}$ ). Two-thirds of the estimations fall outside the anticipated error. Only 3% of the total estimations are less than experimental values. The bulk, approximately 63% of the estimations, is considerably larger. When this histogram is compared to Fig. 3,<sup>2</sup> the errors appear skewed in one direction; experimental values are clearly overestimated by the protocol described. Yet despite the fact that the total phase change entropies of liquid crystals are overestimated, the slope of the line obtained by linear regression is remarkably near one. This suggests that the protocol used to estimate  $\Delta_0^{T_{iso}}S_{tpce}$  remains valid despite the large uncertainty in the prediction.

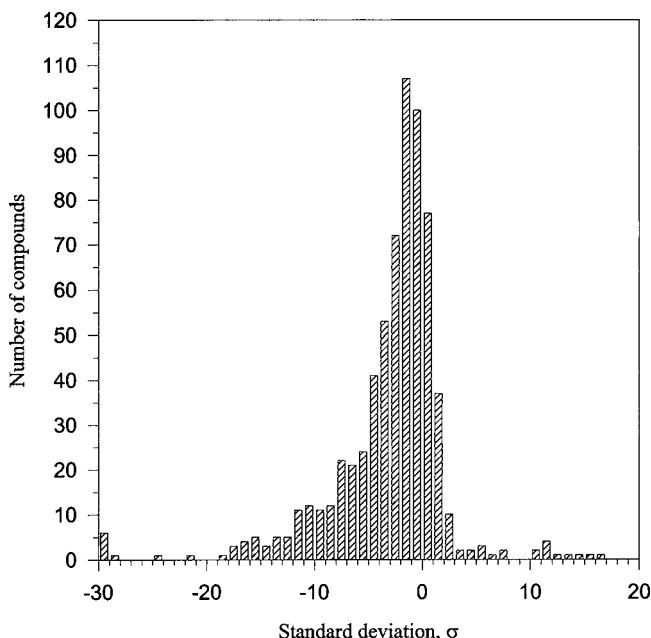


FIG. 4. A histogram of the distribution of errors in  $\Delta_0^{T_{\text{iso}}}S_{\text{tpce}}(\text{exp}) - \Delta_0^{T_{\text{iso}}}S_{\text{tpce}}(\text{calc})$  for liquid crystals.

### 3. A Discussion of $\Delta_0^{T_{\text{iso}}}S_{\text{tpce}}$ for Liquid Crystals

There are several possible reasons responsible for both the scatter and overestimations observed in Fig. 3. The possibility that some residual molecular association exists at the clearing temperature  $T_{\text{iso}}$ , is certainly one conceivable explanation. A second possibility is that the attenuation in total phase change entropy is compensated for by increases in the heat capacity of the crystal over some temperature range. Over a significant temperature range small changes in heat capacity may make a significant contribution to the total entropy. Polymorphism may also play a role. Calorimetric measurements on nonliquid crystalline compounds have shown that there can be substantial differences in the fusion enthalpies of different polymorphic forms.<sup>5,6</sup> The value estimated by the group additivity approach described above gives the total phase change entropy associated with the most stable solid modification at the melting point. The purity of the material is also known to have a large effect on both the enthalpies and temperatures of transition.<sup>8,9</sup> For some compounds the solid phase may not possess total crystallinity. Studies of polymeric materials have shown that the measured enthalpies of fusion depend on the degree of crystallinity. Finally the occurrence of undetected solid–solid phase transitions at low temperatures is certainly another possibility. Many calorimetric measurements on liquid crystalline compounds have generally been performed at ambient temperatures and above. Solid–solid phase transitions below room temperature would not have been detected in most of the calorimetric studies used to create this liquid crystal database. Solid–solid phase transitions are also known to occur

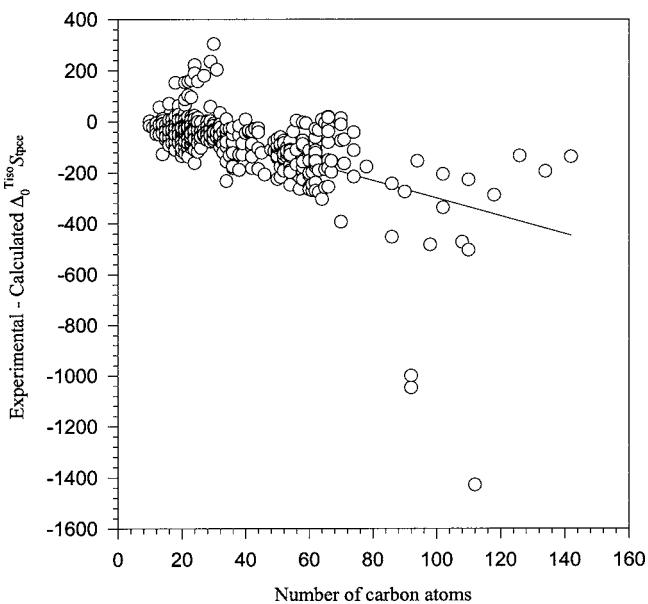


FIG. 5. A plot of the difference between experimental and calculated  $\Delta_0^{T_{\text{iso}}}S_{\text{tpce}}$  as a function of the number of carbon atoms.

with regular frequency in the database compounds of Fig. 1. Although low temperature studies of some of the compounds of Fig. 1 have been reported, most of these are also studies conducted above room temperature. Is there any reason to expect solid–solid phase transitions to occur at any higher frequency in liquid crystals?

Sorai and co-workers have measured heat capacities ( $C_p$ ) and phase transitions of a series of benzene-hexa-alkanoates varying the alkanoate side chain sequentially from pentanoate to decanoate from about  $T = 15$  K to temperatures above the clearing point.<sup>10–14</sup> Table 7 summarizes the phase transitions enthalpies and entropies and total entropy measured for these compounds. The pentanoate, hexanoate, and decanoate do not form liquid crystals. The nonanoate does so only on supercooling the isotropic liquid below the melting temperature. Both the hexa-heptanoate and hexa-octanoate derivatives form liquid crystals. Agreement between estimated and experimental total phase change entropies is within the noise level expected for these estimations for the pentanoate, hexanoate, octanoate, and decanoate. Values for the heptanoate and nonanoate are clearly overestimated. Even for the octanoate, although the two values are within experimental uncertainty, the calculated value is somewhat larger.

If the total entropies at  $T = 385$  K obtained experimentally by combining phase change entropies with heat capacity contributions are now compared, the total entropies increase in a linear fashion with the total number of carbon atoms  $C$ . The dependence of total entropy for the isotropic liquid at  $T = 385$  K is illustrated in Fig. 6. The equation of the line obtained by a linear regression analysis is

TABLE 7. A series of compounds including liquid crystals whose heat capacity and phase transitions of the crystalline phase have been studied over most of the experimentally accessible region<sup>a</sup>

Molecular formula	Transition	Compound			$\Delta_0^{T_{iso}} S_{pcce}$ (exp)	$\Delta_0^{T_{iso}} S_{pcce}$ (estimated)	$S^\circ$ (385 K)
		T (K)	$\Delta H_{pcce}$	$\Delta S_{pcce}$			
$C_{36}H_{54}O_{12}$		benzene hexa-n-pentanoate					
	Sol/Sol	173.1	8.8	50.83			
	Sol/Sol	313.2	15.3	48.85			
$C_{42}H_{66}O_{12}$	Sol/Liq	379.5	30.3	79.84	183.5	234.6	1629.9
		benzene hexa-n-hexanoate					
	Sol/Sol	251.6	25.67	102.0			
	Sol/Sol	291.5	12.27	42.1			
$C_{48}H_{78}O_{12}$	Sol/Sol	348.3	16.26	46.7			
	Sol/Liq	368.7	33.50	90.9	281.7	277	1877.1
		benzene hexa-n-heptanoate					
	Sol/Sol	129	1.1	8.5			
$C_{54}H_{90}O_{12}$	Sol/Meso	353.8	32.2	91.1			
	Meso/Liq	359.3	21.5	59.9	159.5	319.8	2114.4
		benzene hexa-n-octanoate					
$C_{60}H_{102}O_{12}$	Sol/Sol	301.9	49.0	164.0			
	Sol/Meso	355.1	46.1	129.8			
	Meso/Liq	357.1	19.2	53.8	347.6	362.4	2361.3
$C_{66}H_{114}O_{12}$		benzene hexa-n-nonanoate					
	Sol/Sol	248.3	19.4	78.13			
	Sol/Sol	278.3	4.9	17.60			
	Sol/Liq	353.1	69.2	195.98	291.7	405	2597.8
	Liq/Meso	350	NA				
		benzene hexa-n-decanoate					
	Sol/Sol	330.8	75.7	228.9			
	Sol/Liq	360.9	91.8	254.3	483.2	448	2861.5

<sup>a</sup> $C_p$  measurements of the solid from approximately 15 K to the isotropic liquid at 385 K;  $^{10-14}$  units for entropy:  $J \cdot K \cdot mol^{-1}$ ; enthalpy:  $kJ \cdot mol^{-1}$ .

$$\begin{aligned} S^\circ(385 \text{ K}) / J \cdot K^{-1} \cdot mol^{-1} &= (40.8 \pm 0.29)C \\ &\quad + (159.8 \pm 7.27); \\ r^2 &= 0.9998. \end{aligned} \quad (7)$$

The linearity observed in this correlation suggests that the attenuated total phase change entropies observed for the heptanoate and nonanoate have been compensated for by higher heat capacity contributions to the entropy over a portion of the temperature range investigated. Since the hexaheptanoate and hexa-nonanoate derivatives only differ in the number of methylene groups, these results can be interpreted in terms of packing of the methylene groups of the alkyl chains in a shallower potential energy minimum such that the density of states available for molecular motion of these groups in the crystal exceeds the number found in the other homologues. A cause of the discrepancy observed between calculated and experimental total phase change entropy both in those members of the series that melt to isotropic liquids and those that form liquid crystals can be explained.

Extrapolating the results of Sorai *et al.*, <sup>10-14</sup> that the total entropies of members of homologous series that do and do not form liquid crystals correlate as a function of the number of repeat units in the series at a temperature where all members exist as isotropic liquids ( $S^\circ(T_{iso})$ ), then the statistics of Fig. 4 suggest that the alkyl chains in the solid state of those members of the series destined to form liquid crystals are more likely to lie in shallower potential energy minima than

members that melt directly to isotropic liquids. In addition, alkyl chains held in place by shallow minima in the crystal may be more prone to experience solid–solid phase transitions.

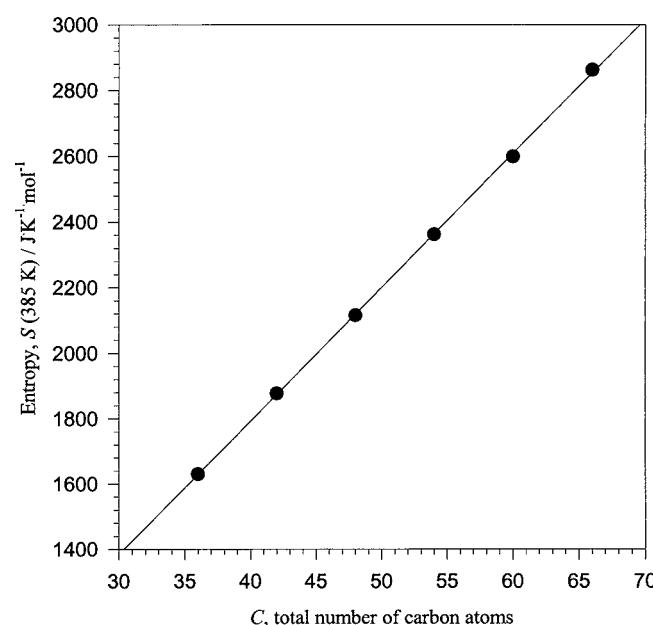


FIG. 6. A plot of the entropy of the benzene hexa-n-alkanoates as a function of the number of carbon atoms.

#### 4. A Discussion of $T_{\text{fus}}$ , $T_{\text{cld}}$ and $T_{\text{iso}}$ for Liquid Crystals

It has been found that the melting temperature ( $T_{\text{fus}}$ ) of many homologous series that do not form liquid crystals correlate with each other as a function of the number of repeat units and asymptotically approach the melting temperature of the polymer formed upon convergence.<sup>15</sup> Even branched series such as the symmetrical triglycerides can be correlated in this fashion. Most liquid crystals studied thus far contain cyclic group(s) with pendant alkyl chains of varying length. In the limit, continuous elongation of these chains ultimately results in a molecule that closely resembles polyethylene. The melting temperature of polyethylene is  $T_{\text{fus}} = 411 \text{ K}$  and according to the model proposed,<sup>15</sup> homologues of molecules that form liquid crystals would be expected to approach this limiting temperature. Furthermore, since polyethylene does not exhibit liquid crystalline properties, liquid crystal behavior would be expected to cease at some stage of homologation.

Some empirical evidence that  $T_{\text{fus}}$  in liquid crystals may actually be depressed relative to homologues that melt directly to isotropic liquids, has recently been reported for a series of *trans* 4-*n*-alkoxy-3-chlorocinnamic acids. This conclusion is based on the correlation observed between  $T_{\text{fus}}$  for members of the series that melt directly to isotropic liquids, and  $T_{\text{iso}}$  for others that first melt to form liquid crystals.<sup>15</sup>

In liquid crystals, it is frequently observed that the melting temperature of a homologous series initially decreases relative to the parent compound as the chain length increases. According to the model proposed,<sup>15</sup> if the melting temperature drops below  $T = 411 \text{ K}$ , this decrease would be attenuated with increasing chain length and as the chain length continues to increase and becomes more like polyethylene, the melting temperature is predicted to begin to increase and approach  $T_{\text{fus}} = 411 \text{ K}$ . In the cases studied previously, the increase or decrease in melting temperature with increasing chain length could be modeled effectively by the following hyperbolic functions:

$$T_{\text{fus}}/\text{K} = 411 \cdot [1 - 1/(mN + b)] \quad \text{for increasing } T_{\text{fus}}(N), \quad (8)$$

$$T_{\text{fus}/\text{iso}}/\text{K} = T_{\text{fus}/\text{iso}}^{\min} / [1 - 1/(mN + b)] \quad \text{for decreasing } T_{\text{fus}/\text{iso}}(N). \quad (9)$$

In Eq. (8), 411 K represents the melting temperature of polyethylene. In Eq. (9),  $T_{\text{fus}/\text{iso}}$  represents either the melting temperature to an isotropic liquid, or  $T_{\text{iso}}$  as previously defined for each member of the series. The term  $T_{\text{fus}/\text{iso}}^{\min}$  represents the minimum melting or clearing temperature experienced by the series prior to its convergence to  $T_{\text{fus}}(\infty)$ , 411 K for convergence to polyethylene. In both equations  $N$  represents the number of repeat units in the series, and  $m$  and  $b$  are two adjustable parameters.  $T_{\text{fus}/\text{iso}}^{\min}$  was evaluated by a nonlinear least squares fit of the experimental data for each series ex-

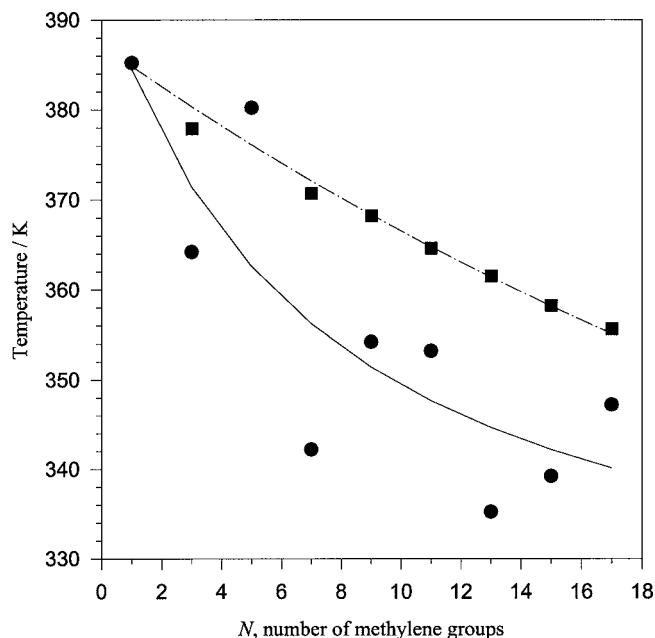


FIG. 7. A comparison of the melting and clearing temperatures of the odd thiocholesteryl *n*-alkanoates.

amined. Since the melting properties of the even and odd series differ,  $T_{\text{fus}/\text{iso}}$  were segregated into two groups based on the number of repeat units present and were modeled by Eqs. (8) and/or (9) separately.<sup>15</sup>

Most of the experimental phase transition data available on liquid crystals show an overall decrease in temperature in  $T_{\text{fus}/\text{iso}}$  associated with increases in chain length of the pendant alkyl groups; only Eq. (9) is applicable in these cases. As examples, Figs. 7, 8, and 9 illustrate melting temperature

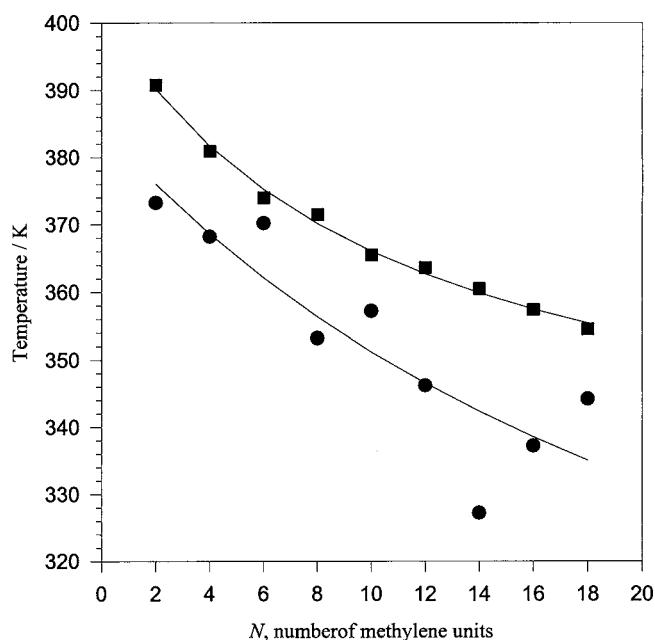


FIG. 8. A comparison of the melting and clearing temperatures of the even thiocholesteryl *n*-alkanoates.

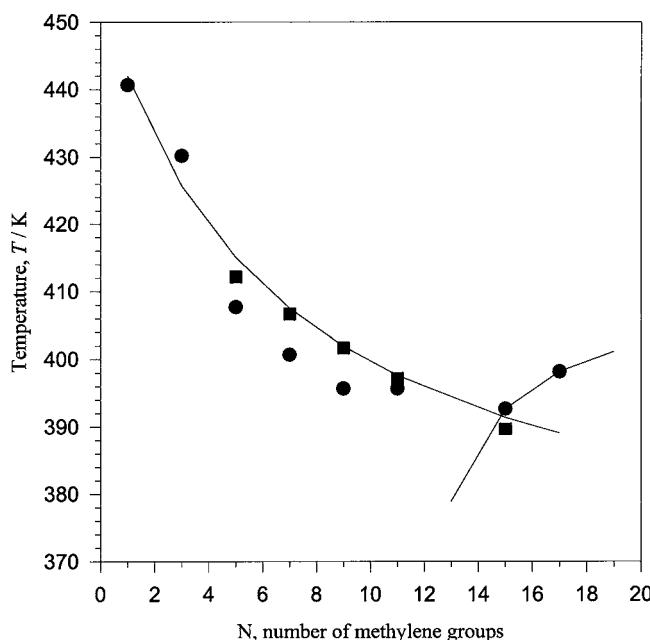


FIG. 9. A comparison of the melting and clearing temperatures of the alkyl 4'-methoxybiphenyl-4-carboxylates.

(circles) and clearing temperature (squares) behavior of the series of odd and even thiocholesteryl *n*-alkanoates<sup>16</sup> and of the odd series of the alkyl 4'-methoxybiphenyl-4-carboxylates,<sup>17</sup> respectively. In Fig. 7, the phase change behavior of members of the odd series, cholestrylo thiopropanoate, thiopentanoate, ... are plotted against the number of methylene groups. The solid circles represent  $T_{\text{fus}}$  and the solid squares represent  $T_{\text{iso}}$ . Two of the compounds, the first two members of the odd series melt directly to the isotropic liquid; the other members form liquid crystals upon melting. Examination of this figure and Fig. 8 clearly illustrates the more erratic nature of melting temperature ( $T_{\text{fus}}$ ) as a function of the number of repeat units. In comparison, the clearing temperatures ( $T_{\text{iso}}$ ) correlate better regardless of whether or not liquid crystal behavior is observed. The solid lines were calculated from a nonlinear least squares fit to Eq. (9) using the melting and clearing temperatures in separate calculations; the lines are described by Eqs. (10)–(13). The best value of  $T_{\text{iso}}^{\min}$  was obtained from the experimental data by a least squares fit by treating  $T_{\text{iso}}^{\min}$  as a variable odd series:

$$T_{\text{fus}}(N)/\text{K} = 317/[1 - 1/(0.564N + 5.133)]; \\ r^2 = 0.6586, \quad (10)$$

$$T_{\text{fus}}(N)/\text{K} = 233/[1 - 1/(0.0238N + 2.528)]; \\ r^2 = 0.9661, \quad (11)$$

even series:

$$T_{\text{fus}}(N)/\text{K} = 259/[1 - 1/(0.0745N + 3.065)]; \\ r^2 = 0.7508, \quad (12)$$

$$T_{\text{fus}}(N)/\text{K} = 328/[1 - 1/(0.4185N + 5.436)];$$

$$r^2 = 0.9931. \quad (13)$$

The more erratic behavior in melting temperature can be explained by packing of the methylene groups in shallow but variable potential energy minima, resulting in variability in the density of states as described above. The clearing temperatures,  $T_{\text{iso}}$ , on the other hand are more continuous and appear to correlate better to the melting temperature of homologues melting directly to isotropic liquids. This suggests that the melting temperatures of liquid crystals,  $T_{\text{fus}}$ , have been lowered relative to those homologues that do not form liquid crystals and that  $T_{\text{iso}}$ , is the thermodynamic function that correlates best within a homologous series.

Although not much experimental data is available, Fig. 9 also illustrates a portion of the melting behavior expected of a homologous series after the minimum is reached. The members of the series exhibiting liquid crystalline behavior occur on the descending portion of the curve. Liquid crystalline behavior appears to cease before the minimum. Once the minimum is reached, the melting temperatures are predicted to increase gradually to  $T = 411$  K. The descending portion of the curve was fit using Eq. (14). The ascending portion of the curve was fit using Eq. (15) using the protocol described above

odd series:

$$T_{\text{iso}}(N)/\text{K} = 364/[1 - 1/(0.6166N + 5.0517)]; \\ r^2 = 0.9827, \quad (14)$$

$$T_{\text{fus}}(N)/\text{K} = 411/[1 - 1/(4.93N - 49.5)]. \quad (15)$$

Both the ascending and descending hyperbolic behavior of melting temperature as a function of the number of methylene groups of homologous series approaching the melting temperature of polyethylene ( $T_{\text{fus}} = 411$  K), can be understood by means of modeling this behavior. Both total phase change enthalpy and total phase change entropy of many homologous series not forming liquid crystals can be modeled as linear properties of the total number of methylene units,  $N_{\text{CH}_2}$ . Their ratio, given by Eq. (16), is a hyperbola that asymptotically

$$T_{\text{fus}} = \Delta H_{\text{fus}} / \Delta S_{\text{fus}} \\ \approx \Delta_0^{T_{\text{iso}}} H_{\text{tpce}} / \Delta_0^{T_{\text{iso}}} S_{\text{tpce}} \\ = [A_H N_{\text{CH}_2} + B_H] / [A_S N_{\text{CH}_2} + B_S] \quad (16)$$

approaches a limiting value as the number of methylene groups increase. It may be an ascending or descending function depending on the magnitude of the group values  $A_H$ ,  $A_S$ .<sup>15</sup> In liquid crystals,  $\Delta_0^{T_{\text{iso}}} H_{\text{tpce}}$  and  $\Delta_0^{T_{\text{iso}}} S_{\text{tpce}}$  are not always well modeled by group methods as shown in Fig. 2. However for the series examined thus far, a hyperbolic function appears quite good at modeling their ratio,  $T_{\text{fus}/\text{iso}}$ .

TABLE 8. Some examples of compounds forming liquid crystals upon supercooling the melt

Molecular formula	Transition	Compound	$\Delta H_{\text{pce}}$	$\Delta S_{\text{pce}}$	$\Delta_0^{T_{\text{iso}}}S_{\text{tpce}}$ (exp)	$\Delta_0^{T_{\text{iso}}}S_{\text{tpce}}$ (estimated)
$C_{22}H_{28}O_2$	Sol/Liq	4-methoxy-4'-heptoxy- <i>trans</i> -stilbene				
	Nem/Liq <sup>a</sup>	423	42.76	101.1	101.1	127.0
$C_{27}H_{46}O$	cholesterol <sup>b</sup>	421	421	1.6		
	Sol/Sol	304.8	2.5	8.2		
$C_{33}H_{58}O_2$	Sol/Liq <sup>b</sup>	420.2	27.41	65.22	73.42	73.7
	Sol/Liq	360.15	29.6	82.1	82.1	124.9
$C_{37}H_{56}O_2$	Chol/Liq <sup>a</sup>	340.15	0.63	1.8		
	cholesteryl $\omega$ -phenylbutyrate	364.1	28.20	77.45	77.45	128.4
$C_{37}H_{66}O_2$	Sol/Liq	299.3	0.84	0.3		
	Liq/Chol <sup>a</sup>	354.65	51.5	145.2	145.2	153.3
$C_{38}H_{66}O_2$	Sol/Liq	326.15	0.36	1.1		
	Smec/Chol	344.15	1.42	4.1		
$C_{38}H_{66}O_2$	Chol/Liq <sup>a</sup>	364.65	26.4	72.4	72.4	161.2
	cholesteryl undecanoate	351.85	1.55	4.4		
$C_{39}H_{60}O_2$	Smec/Liq <sup>a</sup>	361.05	1.34	3.7		
	cholesteryl $\omega$ -phenylhexanoate	354.5	27.74	78.25	78.25	142.6
$C_{39}H_{68}O_2$	Sol/Liq	318.4	0.29	0.9		
	Liq/Chol <sup>a</sup>	364.45	31.8	87.4	87.4	168.3
$C_{39}H_{70}O_2$	Chol/Smec	353.35	0.95	2.7		
	Smec/Liq <sup>a</sup>	360.35	0.74	2.0		
$C_{43}H_{76}O_2$	5 $\alpha$ -cholestane-3 $\beta$ -ol dodecanoate	360.65	57.3	159.0	159.0	167.5
	Sol/Liq	331.15	3.50	1.1		
$C_{46}H_{82}O_2$	Chol/Liq <sup>a</sup>	340.15	1.71	5.0		
	cholesterol palmitate	350.5	58.58	167.13		
$C_{45}H_{80}O_2$	Chol/Liq <sup>a</sup>	354.8	1.18	3.33	170.46	196.7
	cholesteryl nonadecanoate	353.55	73.3	207.3	207.3	218
$C_{58}H_{69}ClO_{10}$	Sol/Liq	344.95	1.9	5.5		
	Chol/Smec	348.75	1.7	4.8		
$C_{58}H_{69}ClO_{10}$	Chol/Liq <sup>a</sup>	354.95	70.38	198.3	198.3	210.9
	bis-4-[[4-(dodecyloxy)benzoyl]oxy]benzoic acid, 4-chloro-1,3-phenylene ester	342.75	1.64	4.8		
$C_{58}H_{69}ClO_{10}$	Sol/Liq	347.55	1.64	4.7		
	Meso/Liq	371.2	38.7	104.26	131.8	305.9
	Nem/Liq <sup>c</sup>	353.2	9.1	25.76		
		368.2	0.65	1.77		

<sup>a</sup>Transition to a liquid crystal observed on cooling.<sup>b</sup>Melts directly to an isotropic liquid; see Chickos *et al.*<sup>6</sup> for details on the estimation and Tables 10 and 11 for references and units.

## 5. Why Do Liquid Crystals Form?

Sorai *et al.*<sup>10-14</sup> have demonstrated that although the phase change entropy of compounds may be attenuated in a homologous series, the total entropy remains a group property (Fig. 6). Similarly, the total phase change entropy of most liquid crystals is smaller than for compounds melting directly to isotropic liquids. The clearing temperature of liquid crystals,  $T_{\text{iso}}$ , appears to correlate best with the melting temperatures of homologues that melt directly to isotropic li-

uids, suggesting that if compounds forming liquid crystals were to melt directly to isotropic liquids, they would do so at  $T_{\text{iso}}$ .

Table 8 includes a list of compounds that melt directly to isotropic liquids containing the two structural units common to most liquid crystals, a somewhat rigid ring and pendant alkyl groups of various lengths. Note that  $\Delta_0^{T_{\text{iso}}}S_{\text{tpce}}$  is overestimated for many of these compounds as well. Cholesterol, which does not form an ordered liquid crystal, is included in this table simply to illustrate that the protocol used in the

TABLE 9. Group values for estimating heat capacities of solids at  $T=298$  K

Hydrocarbon groups	Group values <sup>a</sup> $\Gamma(c)$	Hydrocarbon groups	Group values <sup>a</sup> $\Gamma(c)$
primary $sp^3$ C	36.6	tertiary aromatic $sp^2$ C	17.5
secondary $sp^3$ C	26.9	quaternary aromatic C	8.5
tertiary $sp^3$ C	9.0	internal quaternary aromatic $sp^2$ C	[9.1]
quaternary $sp^3$ C	-4.98	cyclic secondary $sp^3$ C	24.6
secondary $sp^2$ C	[46.0]	cyclic tertiary $sp^3$ C	11.7
tertiary $sp^2$ C	21.4	cyclic quaternary $sp^3$ C	6.1
quaternary $sp^2$ C	[6.86]	cyclic tertiary $sp^2$ C	15.9
tertiary $sp$ C	[23.9]	cyclic quaternary $sp^2$ C	[4.73]
quaternary $sp$ C	[15.2]		
Functional groups			
alcohols, phenols	23.5	cyclic ether	9.71
fluorine	[24.8]	isocyanate	[52.7]
chlorine	28.7	nitro group	56.1
bromine	32.4	thiols	[51.9]
nitrile	[42.3]	primary $sp^3$ N	21.6
carboxylic acid	53.1	secondary $sp^3$ N	[-0.29]
aldehyde	[84.5]	tertiary $sp^3$ N	31.5
ketone	[28.0]	tertiary $sp^2$ N	10.7
cyclic ketone	34.3	cyclic secondary $sp^3$ N	23.9
ester	40.3	cyclic tertiary $sp^3$ N	1.21
lactone	[45.2]	cyclic tertiary $sp^2$ N	13.9
ether	49.8	sulfides	[116]
		cyclic sulfides	18.2

<sup>a</sup>Values in brackets are considered as tentative assignments only; there are no other corrections to be applied; this is only a partial listing of group values available; units:  $J \cdot K^{-1} \cdot mol^{-1}$ .

estimations is not biased toward the steroid nucleus. The compounds chosen for this table do form liquid crystals upon supercooling the isotropic liquid below the melting temperature. The results from Fig. 6 suggest that the heat capacity of the solid phase of these compounds at some temperature regions would be underestimated by group values evaluated at these temperatures from compounds melting directly to isotropic liquids. Furthermore, a large discrepancy between estimated and experimental  $\Delta_0^{T_{iso}}S_{tpce}$  appears to be a good indicator of a substance's propensity to form liquid crystals upon supercooling.

Table 9 lists a series of group values used to estimate heat capacities of organic solids that melt directly to isotropic liquids at  $T=298.15$  K.<sup>17</sup> These group values were derived from compounds melting directly to isotropic liquids. It is interesting to compare the group values for a methylene and ether group, two structural units found in all or most liquid crystals, to the values derived for other common structural units. With the exception of a few terminal groups, these two group values on an atom basis (ignoring hydrogens) make the largest contribution to  $C_p(c)$  at  $T=298.15$  K. Even in compounds that melt to isotropic liquids, the  $CH_2$  and O groups make significant contributions to the heat capacity.

Dialkyl ethers and *n*-alkanes themselves do not form liquid crystals. In addition, their heat capacities at  $T=298.15$  K are generally well modeled by the group values listed in Table 9. Why is it that they do not form liquid crystals and why are the  $C_p(c)$  values for these molecules at

various temperatures less than the values found for compounds forming liquid crystals?

It can be argued that in both simple ethers and alkanes, the density of vibrational states available in the crystal are uniformly distributed throughout the crystal. As thermal energy is supplied, a point is reached when thermal motion is sufficient to overcome the intracrystalline forces and the crystal melts to an isotropic liquid.<sup>18</sup>

With regards to why  $C_p(c)$  values of liquid crystals are larger, the answer probably lies in problems associated with packing a substance that contains two very different structural components, a somewhat rigid cyclic component, and one or more flexible alkyl chain. In our previous study of melting points, we found that the melting temperature of the first few members of many homologous series deviated from the hyperbolic behavior observed for the remaining members of the series. This could be interpreted as due to a change in crystal packing from one dominated by the functional group that characterizes the series, to one dominated by the pendent group(s).<sup>15</sup> This region appears exaggerated in homologous series that form liquid crystals. In the interim, packing of either group(s) is likely to be less than ideal and it is in this region that the heat capacity as a function of temperature is likely to be anomalously large. According to this model, this is a region at which various physical properties are in transition. Liquid crystal formation appears to lie early in this region of transition during the period at which  $T_{iso}$  is decreasing. At this point the crystal lattice should still be domi-

TABLE 10. Abbreviations and notations used in Table 11

Blue:	Blue phase; original blue phase was nematic with long range, three-dimensional crystalline order. Blue phases with smectic and cholesteric layering in addition to long range three-dimensional crystalline cubic and hexagonal ordering have also been observed.
Chol:	Cholesteric phase; a spiraling or helical arrangement of the local orientational order perpendicular to the long axis of the molecules; characteristic of chiral molecules.
Col:	Columnar phase; one of the specific liquid crystalline phases formed by disk-shaped molecules. Columnar phases are the discotic equivalent of the smectic phase. In the simplest case the short axes of the molecules are parallel to the axis of the vertical columns, which are randomly distributed in space.
Cube:	Cubic liquid crystalline phase; generally occurs in between liquid crystalline phases, and is believed to possess a labyrinth structure characterized by either a $Ia3d$ or $Im3m$ space group.
$\Delta H_{\text{fus}}$ :	Enthalpy of fusion
$\Delta H_{\text{pcc}}$ :	The enthalpy associated with a phase change.
$\Delta_0^{T_{\text{iso}}}H_{\text{tpcc}}$ :	$\Delta_0^{T_{\text{iso}}}H_{\text{tpcc}}$ : The total molar phase change enthalpy associated in going from a rigid solid at $T=0$ K to the isotropic liquid at temperature $T=T_{\text{iso}}$ .
$\Delta S_{\text{fus}}$ :	Entropy of fusion
$\Delta S_{\text{pcc}}$ :	The entropy associated with a phase change.
$\Delta_0^{T_{\text{iso}}}S_{\text{tpcc}}$ :	$\Delta_0^{T_{\text{iso}}}S_{\text{tpcc}}$ : The total molar phase change entropy associated in going from a rigid solid at $T=0$ K to the isotropic liquid at temperature $T=T_{\text{iso}}$ .
$\Delta_0^{T_{\text{fus}}}S_{\text{tpcc}}$ :	$\Delta_0^{T_{\text{fus}}}S_{\text{tpcc}}$ : The total molar phase change entropy associated in going from a rigid solid at $T=0$ K to the isotropic liquid at temperature $T=T_{\text{fus}}$ .
Disc:	Discotic phase, broad classification of liquid crystalline phases formed by disk-shaped molecules. The classification is further subdivided into discotic nematic, discotic chiral, and columnar.
Gel:	A highly viscous liquid of undefined mesostructure.
Liq:	Liquid phase; generally associated with isotopic behavior on a molecular level.
Meso:	Mesophase: a thermodynamically stable intermediate phase between crystal and isotropic liquid; first observed by Austrian botanist Frederick Reinitzer in 1888.
NA:	Values for one of the groups present is not currently available.
Nem:	Nematic phase; a state in which the centers of mass of the molecules are disordered but where there is order in molecular orientation.
$S^*(T)$ :	Standard entropy at temperature $T$ .
Smec:	Smectic phase; a phase in which there is order in molecular orientation and also some order in translation associated with planarity or layered structures; as many as 12 smectic phases have been identified.
Sol:	Solid phase; generally associated with regularity in molecular packing.
Sol/Sol:	Solid to solid transition; generally associated with either the onset of molecular motion as found in many plastic crystals or in a phase change associated with a change in lattice parameters; frequently associated with a reorganization in molecular packing.
$T_{\text{cld}}$ :	The temperature in K of those materials forming liquid crystals upon supercooling the melt; $T_{\text{cld}} < T_{\text{fus}}$ .
$T_{\text{fus}}$ :	Melting to either an isotropic or liquid crystal; temperature in K.
$T_{\text{iso}}$ :	Defined as $T_{\text{iso}} \geq T_{\text{fus}}$ ; the temperature at which the liquid becomes isotropic.

nated by the rigid cyclic component with the pendent alkyl groups more loosely packed. This view is consistent with solid state nuclear magnetic resonance studies.<sup>19</sup> As the crystalline form of the compound destined to exhibit liquid crystalline behavior is heated, thermal energy (motion) may not be distributed evenly in the crystal but may tend to accumulate in those regions of the crystal with the highest density of low energy states, most likely the pendant  $-\text{CH}_2-$  and  $-\text{O}-$  groups. Some point is reached at a temperature below the expected melting temperature ( $T_{\text{iso}}$ ) at which sufficient thermal motion is present to cause flow but not sufficient to overcome all the attractive forces still present.<sup>20</sup> Whether a compound forms a liquid crystal or not is likely to be a delicate balance between how evenly or unevenly thermal motion is distributed within the crystal lattice. According to the results illustrated in Figs. 7–9, the effect of loose packing on those substances destined to form liquid crystals is somewhat similar to the effects of an impurity dissolved in nonstoichiometric amounts within a crystal lattice; a depression of the melting temperature results. In liquid crystals, the pendant alkyl chains are present stoichiometrically.

## 6. A Compendium of the Thermochemical Behavior of Liquid Crystals

As noted above, the thermochemical behavior of more than 3000 organic compounds known to form strong associations in the liquid phase have been reported. Table 10 provides a summary of the terms used in Table 11 and their definitions. Table 11 contains the thermochemical data that have been reported, estimations of total phase change entropies of a representative number of compounds, and references to the original literature. In addition abbreviated comments are provided for some entries. Table 12 contains references to Table 11.

The authors have made an effort to present the data accurately and without error. Table 11 has been compiled over a period of 10 years and has gone through several revisions. Some errors have been noted and corrected; however, in a compilation of this magnitude it is unlikely that all the errors have been detected and corrected. The reader is encouraged to consult the original literature when using this compendium.

TABLE 11. The solid-liquid phase change properties of liquid crystals

Molecular formula	transition	Compound		$\Delta S_{\text{pce}}$	$\Delta_0^{T_{\text{fus}}} S_{t \text{pce}} (\text{exp})$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{t \text{pce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )				
$\text{C}_{10}\text{H}_{12}\text{O}_3$	Sol/Nem	4-propoxybenzoic acid					
		423.8	14.38	33.93			
		432.0	1.80	4.17	38.10	67.3	[64]
	Independent values from another reference						
	Sol/Sol	394.2	7.95	20.17			
	Sol/Nem	419.7	16.74	39.89			
	Nem/Liq	426.7	2.51	5.88	65.94	67.3	[156]
$\text{C}_{10}\text{H}_{22}\text{O}_2\text{S}$	3-(heptylthio)-1,2-propanediol						
	Sol/SmeC	289.5	27.30	94.30			
	SmeC/Liq	292.5	1.70	5.81	100.1	112.9	[161]
$\text{C}_{11}\text{H}_{14}\text{O}_2$	4-butylbenzoic acid						
	Sol/Sol	279.0	0.70	2.51			
	Sol/Sol	313.0	7.80	24.92			
	Sol/Meso	377.0	13.30	35.28			
	Meso/Liq	387.0	1.30	3.36	66.07	64.8	[163]
	Independent values from another reference						
	Sol/Meso	309.6	4.70	15.18			
	Sol/Meso	374.6	11.58	30.91			
	Meso/Liq	388.0	0.76	1.96	48.05	64.8	[410]
	4-butoxybenzoic acid						
$\text{C}_{11}\text{H}_{14}\text{O}_3$	Sol/Nem	423.1	18.01	42.57			
	Nem/Liq	433.7	2.45	5.65	48.22	74.4	[64]
$\text{C}_{12}\text{H}_{16}\text{O}_2$	Independent values from another reference						
	Sol/Nem	420.9	19.10	45.38			
	Nem/Liq	432.2	2.93	6.78	52.16	74.4	[160]
	4-pentylbenzoic acid						
	Sol/Sol	252.0	2.60	10.32			
$\text{C}_{12}\text{H}_{16}\text{O}_2$	Sol/Nem	362.0	9.90	27.35			
	Nem/Liq	395.0	1.50	3.80	41.47	71.9	[163]
	Independent values from another reference						
	Sol/Meso	362.5	11.38	31.39			
	Meso/Liq	399.3	1.10	2.75	34.14	71.9	[410]
$\text{C}_{12}\text{H}_{16}\text{O}_3$	4-pentyloxybenzoic acid						
	Sol/Nem	396.8	19.24	48.49			
	Nem/Liq	426.8	2.19	5.13	53.62	81.5	[64]
$\text{C}_{12}\text{H}_{26}\text{O}_2\text{S}$	3-(nonylthio)-1,2-propanediol						
	Sol/SmeC	303.2	23.00	75.86			
	SmeC/Liq	304.1	2.00	6.58	82.44	131.5	[161]
$\text{C}_{13}\text{H}_{13}\text{N}_5\text{O}$	3-(4-butoxyphenyl)-6-cyano-1,2,4,5-tetrazine						
	Sol/SmeC	379.5	14.99	39.50			
	SmeC/Liq	420.6	6.61	15.72	55.22	104.5	[97]
$\text{C}_{13}\text{H}_{16}\text{O}_2$	4-hexylbenzoic acid						
	Sol/Nem	371.0	17.40	46.90			
	Nem/Liq	380.0	2.4	6.32	53.22	79.0	[163]
	Independent values from another reference						
	Sol/Meso	370.6	12.88	34.75			
$\text{C}_{13}\text{H}_{16}\text{O}_2$	Meso/Liq	385.9	0.95	2.46	37.21	79.0	[410]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta H_{\text{pcc}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pcc}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)						
$C_{13}H_{16}O_3$	Sol/Nem	427.2	51.5		120.55			
	Nem/Liq	458.2	8.8		19.20	139.75	83.7	[22]
$C_{13}H_{18}O_3$	Sol/Nem	382.5	12.47		32.60			
	Nem/Liq	428.3	2.45		5.72	38.32	88.6	[64]
Independent values from another reference								
$C_{13}H_{18}O_3$	Sol/Nem	377.7	12.11		32.06			
	Nem/Liq	421.8	2.16		5.12	37.18	88.6	[71, 326]
$C_{13}H_{28}O_4$	Sol/Meso	383.2	42.4		110.65			
	Meso/Liq	388.2	17.2		44.31	154.96	167.3	[145]
$C_{13}H_{28}O_6$	Sol/Smec	421.2	41.30		98.05			
	Smec/Liq	466.2	Not reported in paper				151	[183]
$C_{14}H_8N_2O_2$	Sol/Nem	380.2	28.1		73.91			
	Nem/Liq	404.2	0.52		1.29	75.20	64.2	[108]
$C_{14}H_{10}O_4$	Sol/Nem	502.7	26.35		52.41			
	Nem/Liq	Not observed by dsc				52.41	75.5	[140]
$C_{14}H_{11}ClNO_2S$	Sol/Nem	394.8	21.35		54.08			
	Nem/Liq	410.6	0.22		0.54	54.62	76.3	[75]
$C_{14}H_{14}ClNO$	Sol/Smec	314.5	0.92		2.92			
	Smec/Liq	350.7	16.18		46.14	49.06		[402]
$C_{14}H_{14}N_2O_2$	Sol/Nem	390.8	31.13		79.66			
	Nem/Liq	409.6	0.74		1.81	81.47	78.8	[164]
Independent values from another reference								
$C_{14}H_{14}N_2O_2$	Sol/Nem	391.4	29.57		75.55			
	Nem/Liq	408.5	0.57		1.40	76.95	78.8	[179]
Independent values from another reference								
$C_{14}H_{14}N_2O_2$	Sol/Nem	391.0	29.2		74.68			
	Nem/Liq	409.0	0.73		1.78	76.46	78.8	[440]
$C_{14}H_{20}O_2$	Sol/Sol	238.0	0.50		2.10			
	Sol/Sol	273.0	0.40		1.47			
$C_{14}H_{20}O_2$	Sol/Sol	321.0	6.30		19.63			
	Sol/Nem	376.0	13.60		36.17			
$C_{14}H_{20}O_2$	Nem/Liq	391.0	1.30		3.32	62.69	91.0	[163]
Independent values from another reference								
$C_{14}H_{20}O_2$	Sol/Sol	320.2	7.40		23.11			
	Sol/Meso	374.5	14.33		38.26			
$C_{14}H_{20}O_2$	Meso/Liq	393.1	1.00		2.54	63.91	91.0	[410]
	Sol/Smec	368.0	24.67		67.04			

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta S_{\text{pce}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.			
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )							
$\text{C}_{14}\text{H}_{26}\text{O}_7$	Smec/Nem	372.4	1.23	3.30	74.08	95.7	[64]			
	Nem/Liq	421.9	1.58	3.74						
Independent values from another reference										
$\text{C}_{14}\text{H}_{26}\text{O}_7$	Sol/Smec	367.2	19.25	52.42	87.54	95.7	[156]			
	Smec/Nem	373.2	10.88	29.15						
	Nem/Liq	420.2	2.51	5.97						
$\text{C}_{14}\text{H}_{26}\text{O}_7$	6-O-octanoyl- $\alpha$ -D-galactopyranose		358.2	2.48	6.92	136.3	[39]			
	Sol/Smec	358.2	2.48	6.92						
	Note: Above value is far out of line with others in series.									
$\text{C}_{14}\text{H}_{28}\text{O}_6$	Smec/Liq	406.2	0.61	1.50	8.42	143.4	[39]			
$\text{C}_{14}\text{H}_{28}\text{O}_6$	6-O-octyl- $\alpha$ -D-galactopyranose		358.2	34.24	95.59	143.4	[39]			
	Sol/Smec	358.2	34.24	95.59						
$\text{C}_{14}\text{H}_{28}\text{O}_6$	Smec/Liq	442.2	2.27	5.13	100.72					
$\text{C}_{14}\text{H}_{28}\text{O}_6$	octyl- $\beta$ -D-glucopyranoside		382.0	1.26	3.30	[421]				
	Sol/Smec	Not reported in paper								
	Smec/Liq	382.0	1.26	3.30						
$\text{C}_{14}\text{H}_{30}\text{O}_4$	1,2,13,14-tetrahydroxytetradecane		358.2	30.0	83.75	176.6	[145]			
	Sol/Meso	358.2	30.0	83.75						
	Meso/Liq	397.2	18.9	47.58	131.33					
$\text{C}_{14}\text{H}_{30}\text{O}_6$	6-O-octyl-D-galacitol		420.2	60.56	144.12	158.2	[183]			
	Sol/Smec	420.2	60.56	144.12						
	Smec/Liq	427.2	1.99	4.66	148.78					
$\text{C}_{15}\text{H}_8\text{F}_{13}\text{NO}_3\text{S}$	2-(perfluorohexyl)ethyl 4-nitrothiobenzoate		360.8	32.92	91.24	NA	[426]			
	Sol/Smec	360.8	32.92	91.24						
	Smec/Liq	363.1	Not reported in paper		91.24					
Note: Smec/Liq transition enthalpy is included in Sol/Smec value.										
$\text{C}_{15}\text{H}_{11}\text{NO}_4$	4-methoxybiphenyl 4-isocyanatobenzoate		363.2	22.9	63.05	96.9	[108]			
	Sol/Nem	363.2	22.9	63.05						
	Nem/Liq	369.2	0.13	0.35	63.40					
$\text{C}_{15}\text{H}_{14}\text{ClNO}_2\text{S}$	1-[2-(5-chlorothienyl)]-3-(4-ethoxyphenylamino)-2-propen-1-one		402.6	16.65	41.36	83.4	[75]			
	Sol/Smec	402.6	16.65	41.36						
	Smec/Liq	438.3	4.87	11.11	52.47					
$\text{C}_{15}\text{H}_{14}\text{F}_2$	3,5-difluoro-4'-propyl-[1,1'-biphenyl]		298.1	15.26	51.19	[437]				
	Sol/Nem	298.1	15.26	51.19						
	Nem/Liq	Not reported in paper								
$\text{C}_{15}\text{H}_{13}\text{NS}$	4-ethyl-4'-thiocyanatobiphenyl		329.5	7.10	21.55	NA	[84]			
	Sol/Meso	329.5	7.10	21.55						
	Meso/Liq	357.3	11.72	32.80	54.35					
$\text{C}_{15}\text{H}_{21}\text{NO}_5$	4-octyloxy-3-nitrobenzoic acid		393.2	29.6	75.28	104.3	[1]			
	Smec/Liq	393.2	29.6	75.28	75.28					
$\text{C}_{15}\text{H}_{22}\text{O}_2$	4-octylbenzoic acid		305.5	5.40	1.77	[410]				
	Sol/Sol	305.5	5.40	1.77						
	Sol/Sol	366.6	0.47	1.28						
	Sol/Meso	373.3	14.32	38.36						
$\text{C}_{15}\text{H}_{22}\text{O}_3$	Meso/Liq	385.5	1.20	3.11	44.52					
	4-octyloxybenzoic acid		377.7	10.26	27.16					
	Sol/Smec	377.7	10.26	27.16						
	Smec/Nem	373.5	1.00							

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.			
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$					
Note: Smec/Nem phase transition temperature was below the Sol/Smec transition temperature.									
Nem/Liq		423.6	2.10	4.96	32.12	102.8			
Independent values from another reference									
Sol/Smec		347.1	1.43	4.12					
Note: Cooling curve indicated a significantly larger transition enthalpy of 12.09.									
Smec/Nem		371.4	11.46	30.86					
Nem/Liq		413.2	1.80	4.35	39.33	102.8			
Independent values from another reference									
Sol/Sol		348.2	17.99	51.67					
Sol/Smec		374.2	10.88	29.08					
Smec/Nem		381.2	1.26	3.31					
Nem/Liq		419.2	2.51	5.99	90.05	102.8			
[156]									
C <sub>15</sub> H <sub>28</sub> O <sub>5</sub>	pentaerythritol mono-[ <i>trans</i> -4-propylcyclohexane]-1-carboxylate								
Sol/Smec		355.5	27.2	76.51					
Smec/Liq		391.3	1.4	3.58	80.09	116.3			
[21]									
C <sub>15</sub> H <sub>28</sub> O <sub>7</sub>	6-O-nonanoyl- $\alpha$ -D-galactopyranose								
Sol/Smec		363.2	32.65	89.90					
Smec/Liq		388.2	0.61	1.57	91.47	150.5			
[39]									
C <sub>15</sub> H <sub>30</sub> O <sub>6</sub>	nonyl- $\beta$ -D-glucopyranoside								
Sol/Smec	Not reported in paper								
Smec/Liq		400.0	1.68	4.20					
[421]									
C <sub>15</sub> H <sub>32</sub> O <sub>6</sub>	6-O-nonyl-D-galacitol								
Sol/Smec		417.2	52.09	124.86					
Smec/Liq		468.2	Not reported in paper			165.3			
[183]									
C <sub>16</sub> H <sub>14</sub> N <sub>2</sub> O <sub>2</sub> S <sub>2</sub>	2,5-bis(4-methoxyphenyl)thiazolo[5,4-d]dithiazole								
Sol/Nem		547.1	41.77	76.35					
Nem/Liq		622.5	0.36	0.58	76.93				
[269]									
C <sub>16</sub> H <sub>15</sub> BrO	4-bromo-4'-(3-butenyloxy)biphenyl								
Sol/Meso		324.2	13.3	41.02					
Meso/Liq		396.8	15.8	39.82	80.84	88.2			
[118]									
C <sub>16</sub> H <sub>15</sub> NO <sub>3</sub>	4-[( <i>E</i> )-[(4-methoxyphenyl)methylene]amino]phenol, acetate (ester)								
Sol/Nem		358.1	23.6	65.90					
Nem/Liq		382.7	0.65	1.70	67.60				
[365]									
C <sub>16</sub> H <sub>15</sub> NS	4-propyl-4'-thiocyanatobiphenyl								
Sol/Meso		304.9	9.16	30.04					
Meso/Liq		361.4	14.77	40.87	70.91	NA			
[84]									
C <sub>16</sub> H <sub>16</sub> N <sub>2</sub> O <sub>2</sub>	anisaldazine								
Sol/Nem		442.1	29.7	67.2					
Nem/Liq		453.7	0.66	1.5	68.7	78.2			
[309]									
C <sub>16</sub> H <sub>17</sub> NO <sub>3</sub>	4-(4-methoxybenzylidene)-ethoxyaniline-N-oxide								
Sol/Liq		419.2	37.66	89.84					
Liq/Nem		411.2	0.72	1.75	91.59	NA			
[162]									
C <sub>16</sub> H <sub>18</sub> N <sub>2</sub> O	4-propyl-4'-methoxyazobenzene								
Sol/Nem		334.3	52.93	158.33					
Nem/Liq		341.1	1.56	4.57	162.90	92.6			
[153]									
C <sub>16</sub> H <sub>18</sub> N <sub>2</sub> O <sub>2</sub>	4,4'-diethoxyazobenzene								
Sol/Nem		406.5	27.00	66.42					
Nem/Liq		438.5	1.70	3.88	70.30	84.4			
[157]									

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta S_{\text{pce}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.	
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )					
<chem>C16H18N2O3</chem>	Sol/Nem	4,4'-diethoxyazoxybenzene						
		411.4	25.8	62.71				
		441.4	1.75	3.96	66.67	93.0	[9]	
	Independent values from another reference							
	Sol/Nem	409.8	26.87	65.57				
	Nem/Liq	440.7	1.37	3.11	68.68	93.0	[179]	
	Independent values from another reference							
	Sol/Sol	360.0	2.09	5.81				
	Sol/Nem	410.0	26.8	65.37				
	Nem/Liq	441.0	1.66	3.76	74.94	93.0	[440]	
<chem>C16H19N5</chem>	3-cyano-6-(4-heptylphenyl)-1,2,4,5-tetrazine							
	Sol/SmeC	327.7	20.04	61.15				
	SmeC/Liq	372.4	5.42	14.55	75.70	119	[97]	
<chem>C16H20O5</chem>	4-(6-acryloyloxyhexyloxy)benzoic acid							
	Sol/SmeC	365.2	29.91	81.89				
	SmeC/Nem	375.2	0.88	2.35				
	Nem/Liq	382.2	1.70	4.45	88.69	121.0	[222]	
<chem>C16H21N</chem>	4-( <i>trans</i> -4-propylcyclohexyl)benzonitrile							
	Sol/Nem	316.3	20.4	64.50				
	Nem/Liq	319.1	1.1	3.45	67.95	84.2	[148]	
<chem>C16H21NS</chem>	4-( <i>trans</i> -4'-propylcyclohexyl)isothiocyanatobenzene							
	Sol/Nem	312.2	13.1	41.96				
	Nem/Liq	315.2	0.90	2.86	44.82	NA	[151]	
<chem>C16H23NO5</chem>	4-nonyloxy-3-nitrobenzoic acid							
	Sol/SmeC	369.2	24.6	66.63				
	SmeC/Liq	Unable to be calculated				126.7	[1]	
<chem>C16H24O3</chem>	4-nonyloxybenzoic acid							
	Sol/SmeC	368.3	39.28	106.66				
	SmeC/Nem	386.4	1.64	4.24				
	Nem/Liq	414.5	2.51	6.06	116.96	109.9	[64]	
	Independent values from another reference							
	Sol/SmeC	365.2	33.47	91.65				
	SmeC/Nem	391.2	1.68	4.29				
	Nem/Liq	418.2	2.51	6.00	101.94	109.9	[156]	
	<i>trans,trans</i> -4'-propyl-[1,1']-bicyclohexyl-4-carbonitrile							
	Sol/Nem	330.7	27.0	81.64				
<chem>C16H27N</chem>		Nem/Liq	353.8	1.8	5.09	86.73	79.7	[15]
6-O-decanoyl- $\alpha$ -D-galactopyranose								
Sol/SmeC	360.2	28.59	79.37					
<chem>C16H30O7</chem>	SmeC/Liq	448.2	0.77	1.72	81.09	150.5	[39]	
	6-S-decyl-6-thio- $\alpha$ -D-galactopyranose							
	Sol/SmeC	377.2	30.45	80.73				
<chem>C16H32O5S</chem>	SmeC/Liq	464.2	3.84	8.27	89.00	152.1	[39]	
	6-O-decyl- $\alpha$ -D-galactopyranose							
	Sol/SmeC	390.2	23.29	59.69				
<chem>C16H32O6</chem>	SmeC/Liq	445.2	0.60	1.35	61.04	154.6	[39]	
	decyl- $\beta$ -D-glucopyranoside							

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.	
		T (K)							
$C_{16}H_{34}O_6$	Sol/Smec	Not reported in paper		416.0	2.15	5.17	134.24	172.4	[421]
	Smec/Liq						137.71		
$C_{17}H_{16}N_2$	Sol/Smec	6-O-decyl-D-galactitol		418.2	56.14	134.24			[183]
	Smec/Liq			443.2	1.54	3.47	59.65		
	Sol/Nem	4-[(4-propylphenyl)methylene]amino]benzonitrile		336.2	19.5	58.00			
$C_{17}H_{17}NO_3$	Nem/Liq			344.6	0.57	1.65			[365]
	Sol/Sol	4-[(4-propoxyphenyl)methylene]amino]benzoic acid		429.2	4.35	10.14			
	Sol/Nem			469.2	14.64	31.20			
	Nem/Liq			548.2	6.36	11.60	52.94		
$C_{17}H_{18}BrNOS$	Sol/Smec	1-[2-(5-butylthienyl)]-3-(4-bromophenylamino)-2-propen-1-one		392.2	17.29	44.08			[75]
	Smec/Liq			432.4	6.30	14.57	58.65	94.2	
	Sol/Smec	1-[2-(5-butylthienyl)]-3-(4-chlorophenylamino)-2-propen-1-one		375.8	15.82	42.10			
$C_{17}H_{18}ClNOS$	Smec/Liq			426.3	5.94	13.93	56.03	92.9	[75]
	Sol/Smec	1-[2-(5-butylthienyl)]-3-(4-fluorophenylamino)-2-propen-1-one		354.0	24.72	69.83			
	Smec/Liq			359.8	2.85	7.92	77.75	93.3	
$C_{17}H_{18}INOS$	Sol/Smec	1-[2-(5-butylthienyl)]-3-(4-iodophenylamino)-2-propen-1-one		409.5	18.86	46.06			[75]
	Smec/Liq			427.3	6.08	14.23	60.29	96.1	
	Sol/Liq	4-(4-ethoxybenzylidene)-ethoxyaniline-N-oxide		451.2	43.68	96.8			
$C_{17}H_{19}NO_3$	Liq/Nem			431.2	1.45	3.36	100.16	NA	[162]
	Sol/Smec	6-heptyl-2-chloro-3-quinoliniccarboxaldehyde		305.2	6.52	21.36			
	Smec/Liq			336.2	22.60	67.22	88.58		
$C_{17}H_{20}ClNO$	Sol/Smec	4-chloro-2'-hydroxy-4'-pentyloxyazobenzene		342.1	26.0	76.00			[402]
	Smec/Nem			355.9	1.7	4.78			
	Nem/Liq			365.0	0.4	1.10	81.88	97.9	
$C_{17}H_{20}N_2O$	Sol/Nem	4-butyl-4'-methoxyazobenzene		305.7	16.85	55.12			[229]
	Nem/Liq			321.8	0.33	1.03	56.15	99.7	
	Sol/Nem	4-propyl-4'-ethoxyazobenzene		361.6	52.15	144.22			
$C_{17}H_{20}N_2O$	Nem/Liq			371.5	2.74	7.38	151.60	99.7	[153]
	Sol/Nem	4-methyl-2'-hydroxy-4'-butoxyazobenzene		357.6	29.6	82.77			
	Nem/Liq			359.9	0.77	2.14	84.91	105.1	
Independent values from another reference									
	Sol/Nem			359.2	4.41	12.28			[282]
	Nem/Liq			363.2	0.27	0.74	13.02	105.1	
Note: Experimental transition enthalpy data seems much too small.									

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{17}H_{23}NO_2S$	Sol/Smec	310.6	25.6	82.42	NA	[407]
	Smec/Liq	354.0	3.80	10.73		
$C_{17}H_{23}NS$	4-( <i>trans</i> -4'-butylcyclohexyl)isothiocyanatobenzene					
	Sol/Nem	307.2	25.2	82.03	NA	[151]
$C_{17}H_{24}O_3$	Nem/Liq	308.2	0.45	1.46		
	4-n-octyloxycinnamic acid					
$C_{17}H_{25}NO_5$	Sol/Nem	414.2	43.1	104.06	112.1	[22]
	Nem/Liq	445.2	8.3	18.64		
$C_{17}H_{26}O_3$	4-decyloxy-3-nitrobenzoic acid					
	Sol/Smec	371.2	20.8	56.03	133.8	[1]
$C_{17}H_{26}O_3$	Smec/Smec	372.2	1.2	3.22		
	Smec/Nem	396.9	1.25	3.15		
$C_{17}H_{26}O_3$	Nem/Liq	415.6	2.14	5.15	76.93	136.8
	Independent values from another reference					
$C_{17}H_{26}O_3$	Sol/Smec	368.9	7.04	19.08	NA	[53]
	Note: Above value is quite low compared to values from three other references. Also above value might be for Smec/Smec transition, rather than the Sol/Smec transition as indicated in the paper.					
$C_{17}H_{26}O_3$	Smec/Nem	384.6	0.92	2.39	117.0	[64]
	Nem/Liq	408.3	2.61	6.39		
$C_{17}H_{26}O_3$	Independent values from another reference					
	Sol/Smec	359.1	21.13	58.84	117.0	[71]
$C_{17}H_{26}O_3$	Smec/Smec	369.2	11.61	31.45		
	Smec/Nem	395.0	1.59	4.03		
$C_{17}H_{26}O_3$	Nem/Liq	414.0	2.45	5.92	100.24	117.0
	Independent values from another reference					
$C_{17}H_{26}O_3$	Sol/Smec	359.2	21.76	60.58	117.0	[156]
	Smec/Smec	370.2	10.46	28.25		
$C_{17}H_{32}O_7$	Smec/Nem	398.2	1.67	4.19		
	Nem/Liq	416.2	2.93	7.04	100.06	
$C_{17}H_{32}O_7$	6-O-undecanoyl- $\alpha$ -D-galactopyranose					
	Sol/Smec	388.2	25.78	66.41	117.0	[39]
$C_{17}H_{34}O_4$	Smec/Liq	454.2	3.00	6.61	73.02	
$C_{17}H_{34}O_4$	pentaerythritol mono-[ <i>trans</i> -4-pentylcyclohexylmethyl] ether					
	Sol/Smec	348.8	41.1	117.83	115.7	[21]
$C_{17}H_{34}O_6$	Smec/Liq	355.6	2.2	6.19	124.02	
$C_{17}H_{34}O_6$	undecyl- $\beta$ -D-glucopyranoside					
	Sol/Smec	Not reported in paper			115.7	[421]
$C_{17}H_{34}O_6S$	Smec/Liq	412.0	1.55	3.76		
$C_{17}H_{34}O_6S$	6-O-(propylene-[3'-S-octyl])- $\alpha$ -D-galactopyranose					
	Sol/Smec	368.2	47.10	127.92	115.7	[39]
$C_{17}H_{36}O_6$	Smec/Liq	429.2	1.43	3.33	131.25	
$C_{17}H_{36}O_6$	6-O-undecyl-D-galacitol					
	Sol/Smec	419.2	66.72	159.16	115.7	[183]
$C_{18}H_{13}F_{13}O_2S$	Smec/Liq	466.2	1.68	3.60	162.76	
	2-(perfluorohexyl)ethyl 4-allyloxythiobenzoate					

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{18}H_{15}NO$	Sol/Smec	302.5	21.34	70.54		
	Smec/Liq	371.0	7.56	20.38	90.92	NA [134]
	Sol/Nem	448.5	21.1	47.04		
	Nem/Liq	486.2	Not reported in paper		90.4	[210]
$C_{18}H_{16}O_3$	6-(4-pentenyl)-2-naphthoic acid					
	Sol/Nem	424.7	85.77	201.95		
	Nem/Liq	470.9	13.81	29.33	231.28	79.2 [68]
$C_{18}H_{18}N_2O$	N-(4-butoxybenzylidene)-4-aminobenzonitrile					
	Sol/Nem	337.7	25.7	76.10		
	Nem/Liq	379.4	0.58	1.53	77.63	[365]
$C_{18}H_{18}N_2O_2S_2$	2,5-bis(4-ethoxyphenyl)thiazolo[5,4-d]dithiazole					
	Sol/Nem	511.6	30.86	60.32		
	Nem/Liq	601.4	0.76	1.26	61.58	[269]
$C_{18}H_{18}N_2O_3$	4-[{(4-ethoxyphenyl)azo}phenyl 2-butenoate					
	Sol/Nem	382.9	23.6	61.63		
	Nem/Liq	469.9	1.92	4.09	65.72	[365]
$C_{18}H_{18}N_2O_5$	ethyl azoxybenzenedicarboxylate					
	Sol/Meso	386.9	20.50	52.99		
	Meso/Liq	395.7	5.44	13.75	66.74	99.0 [157]
$C_{18}H_{19}BrO$	4-bromo-4'-(5-hexenyl)bisphenyl					
	Sol/Meso	308.2	13.8	44.78		
	Meso/Liq	393.9	16.0	40.62	85.40	102.4 [118]
$C_{18}H_{19}ClO_2S$	4-chlorophenyl 4-pentyloxythiobenzoate					
	Sol/Nem	352.4	32.80	93.08		
	Nem/Liq	358.9	0.71	1.98	95.06	NA [383]
$C_{18}H_{19}N$	4'-pentyl-4-cyanobiphenyl					
	Sol/Nem	295.7	17.15	58.00		
	Nem/Liq	308.2	Value was not reported			90.8 [158]
Independent values from another reference						
	Sol/Nem	287.9	15.95	55.40		
	Nem/Liq	306.9	0.54	1.76	57.16	90.8 [233]
Independent values from another reference						
	Sol/Nem	297.0	13.39	45.08		
	Nem/Liq	309.0	0.33	1.07	46.15	90.8 [237]
$C_{18}H_{19}NO$	4'-pentyloxy-4-cyanobiphenyl					
	Sol/Nem	321.2	28.90	89.98		
	Nem/Liq	340.7	Value was not reported			97.6 [158]
Independent values from another reference						
	Sol/Nem	320.5	29.58	92.29		
	Nem/Liq	340.8	0.42	1.23	93.52	97.6 [233]
Independent values from another reference						
	Sol/Nem	325.5	35.80	109.98		
	Nem/Liq	341.2	0.40	1.17	111.15	97.6 [326]
$C_{18}H_{19}NO_3$	4-[(4-methoxyphenyl)methylene]amino]phenyl butanoate					
	Sol/Nem	323.7	23.3	71.98		
	Nem/Liq	385.0	0.60	1.56	73.54	[365]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.	
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$			
$C_{18}H_{19}NS$	Sol/Meso	326.3	11.60	35.55	NA	[84]	
	Meso/Liq	346.8	10.84	31.26			
$C_{18}H_{20}BrNOS$	Sol/Smec	378.8	17.08	45.09	101.3	[75]	
	Smec/Liq	436.0	6.88	15.78			
$C_{18}H_{20}ClNOS$	Sol/Smec	360.9	14.21	39.39	100.0	[75]	
	Smec/Liq	429.5	6.24	14.53			
$C_{18}H_{20}ClNO_2S$	1-[2-(5-pentylthienyl)]-3-(4-chlorophenylamino)-2-propen-1-one						
	Sol/Meso	347.5	10.94	31.48	104.7	[75]	
	Meso/Smec	369.2	0.12	0.33			
	Smec/Smec	376.0	0.97	2.58			
$C_{18}H_{20}FNOS$	Smec/Liq	455.7	6.64	14.57	100.4	[75]	
	1-[2-(5-pentylthienyl)]-3-(4-fluorophenylamino)-2-propen-1-one						
	Smec/Smec	338.2	17.85	52.78			
	Smec/Liq	365.4	3.23	8.84			
$C_{18}H_{20}N_2O$	Sol/Nem	335.4	18.14	54.08	108.5	[17]	
	Nem/Liq	368.8	0.55	1.49			
$C_{18}H_{20}OS$	Sol/Meso	358.4	11.6	32.37	107.9	[118]	
	Meso/Liq	384.6	13.6	35.36			
$C_{18}H_{21}NO$	N-(4-methoxybenzylidene)-4-n-butyylaniline						
	Sol/Nem	295.7	18.03	60.97	90.5	[157]	
$C_{18}H_{21}NO$	Nem/Liq	317.0	0.62	1.96			
	Independent values from another reference						
$C_{18}H_{21}NO$	Sol/Nem	294.1	13.80	46.92	90.5	[376, 381]	
	Nem/Liq	326.1	0.31	0.95			
Note: Temperature was calculated from published enthalpy and entropy values in paper.							
$C_{18}H_{21}NO_2$	Independent values from another reference						
	Sol/Nem	295.7	12.93	43.72	90.5	[406]	
$C_{18}H_{21}NO_2$	Nem/Liq	320.1	0.28	0.87			
	N-(2-hydroxy-4-methoxybenzylidene)-4-butylaniline						
$C_{18}H_{21}NO_2$	Sol/Nem	314.5	24.41	77.62	95.9	[157, 341]	
	Nem/Liq	335.7	0.89	2.65			
$C_{18}H_{22}N_2O$	4-pentyl-4'-methoxyazobenzene						
	Sol/Nem	313.2	18.67	59.61	95.4	[141]	
$C_{18}H_{22}N_2O$	Nem/Liq	340.3	0.77	2.26			
	4-butyl-4'-ethoxyazobenzene						
$C_{18}H_{22}N_2O$	Sol/Nem	316.5	14.79	46.73	95.4	[141]	
	Nem/Liq	351.3	0.80	2.28			
$C_{18}H_{22}N_2O_2$	4-methyl-2'-hydroxy-4'-pentyloxyazobenzene						
	Sol/Nem	339.9	27.7	81.49	97.2	[73]	
$C_{18}H_{22}N_2O_2$	Nem/Liq	353.2	1.0	2.83			
$C_{18}H_{22}N_2O_3$	4,4'-dipropoxyazoxybenzene						
	Sol/Sol	372.5	20.60	55.30	NA	[84]	
	Sol/Nem	390.7	27.40	70.13			

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta S_{\text{pcc}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pcc}}$ (kJ·mol <sup>-1</sup> )				
	Nem/Liq	397.8	0.92	2.31	127.74	107.2	[9]
Independent values from another reference							
Sol/Nem	388.7	26.90		69.21			
Nem/Liq	396.8	0.67		1.69	70.90	107.2	[179]
Note: Sol/Sol phase transition likely occurs and was not included in the authors' measured values.							
Independent values from another reference							
Sol/Sol	376.0	20.7		55.05			
Sol/Nem	389.0	27.5		70.69			
Nem/Liq	398.0	1.03		2.59	128.33	107.2	[440]
$C_{18}H_{24}O_4$	4,4'-di(2-methoxyethoxy)biphenyl						
Sol/Smec	409.5	17.53		42.80			
Smec/Liq	412.4	22.67		54.97	97.77	111.6	[102]
$C_{18}H_{26}O_4$	4-decyloxy carbonylbenzoic acid						
Sol/Smec	342.2	9.71		28.38			
Smec/Liq	386.5	29.72		76.90	105.28	140.8	[53]
$C_{18}H_{27}NO_5$	4-undecyloxy-3-nitrobenzoic acid						
Sol/Smec	364.2	22.6		62.05			
Smec/Liq	365.2	2.2		6.02	68.07	162.9	[1]
$C_{18}H_{28}O_3$	4-undecyloxybenzoic acid						
Sol/Smec	369.2	40.17		108.80			
Smec/Nem	402.2	2.09		5.20			
Nem/Liq	413.2	2.51		6.07	120.07	146.1	[156]
Independent values from another reference							
Sol/Smec	378.4	34.23		90.46			
Smec/Nem	404.7	1.99		4.92			
Nem/Liq	415.6	2.37		5.70	101.08		[418]
$C_{18}H_{34}O_7$	6-O-dodecanoyl- $\alpha$ -D-galactopyranose						
Sol/Smec	367.2	23.16		63.07			
Smec/Liq	455.2	4.02		8.83	71.90	164.7	[39]
$C_{18}H_{36}O_5S$	6-S-dodecyl-6-thio- $\alpha$ -D-galactopyranose						
Sol/Smec	382.2	44.07		115.31			
Smec/Liq	458.2	2.84		6.20	121.51	192.6	[39]
$C_{18}H_{36}O_6$	6-O-dodecyl- $\alpha$ -D-galactopyranose						
Sol/Smec	392.2	44.53		113.54			
Smec/Liq	444.2	1.29		2.90	116.44	195.2	[39]
$C_{18}H_{36}O_6$	dodecyl- $\beta$ -D-glucopyranoside						
Sol/Smec	Not reported in paper						
Smec/Liq	427.0	1.55		3.63			[421]
$C_{18}H_{38}O_4$	1,2,17,18-tetrahydroxyoctadecane						
Sol/Meso	352.2	21.5		61.04			
Meso/Meso	407.2	11.0		27.01			
Meso/Liq	410.2	14.6		35.59	123.64	234.8	[145]
$C_{18}H_{38}O_6$	6-O-dodecyl-D-galacitol						
Sol/Smec	415.2	66.42		159.97			
Smec/Liq	444.2	1.49		3.35	163.32	215.2	[183]
$C_{19}H_{13}F_9SO$	2-(perfluorobutyl)ethyl 4-phenylthiobenzoate						
Sol/Smec	319.7	24.9		77.89			
Smec/Liq	406.2	15.7		38.65	116.54	NA	[38]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)						
$C_{19}H_{15}F$	Sol/Nem	4-propyl-4'-fluorodiphenyldiacetylene		367.6	20.27	55.14	64.3	[270]
	Nem/Liq	375.4	0.59		1.57	56.71		
$C_{19}H_{16}N_2O$	Sol/Nem	4'-methoxy-N-[2-(pyridinyl)methylene]-1,1-biphenyl-4-amine		390.0	20.46	52.46	86.7	[314, 155]
	Nem/Liq	392.0	0.20		0.51	52.96		
$C_{19}H_{16}N_2O$	Sol/Nem	4'-methoxy-N-[3-(pyridinyl)methylene]-1,1-biphenyl-4-amine		448.9	15.48	34.49	86.7	[314, 155]
	Nem/Liq	468.4	0.40		0.85	35.33		
$C_{19}H_{16}N_2O$	Sol/Nem	4'-methoxy-N-[4-(pyridinyl)methylene]-1,1-biphenyl-4-amine		467.0	36.19	77.50	86.7	[314, 155]
	Nem/Liq	454.5	0.35		0.77	78.27		
$C_{19}H_{17}FNOS$	Sol/Nem	4'-butoxy-3'-fluoro-4-isothiocyanatotolane		337.4	23.93	70.92	NA	[135]
	Nem/Liq	344.8	0.54		1.57	72.49		
$C_{19}H_{17}N$	Sol/Nem	N,N-dimethyl-4-[4-(4-methylphenyl)-1,3-butadinyl]benzenamine		401.3	27.9	69.52	NA	[216]
	Nem/Liq	406.5	1.24		3.05	72.57		
$C_{19}H_{17}NOS$	Sol/SmeC	4'-butoxy-4-isothiocyanatotolane		385.3	22.43	58.21	NA	[135, 238]
	SmeC/Nem	385.7	13.22		34.28			
	Nem/Liq	389.2	0.80		2.06	94.55		
$C_{19}H_{17}NOS$	Sol/Nem	4'-methoxy-N-[5-methyl-2-thienyl)methylene]-1,1-biphenyl-4-amine		433.3	31.42	72.52	NA	[314, 155]
	Nem/Liq	484.2	0.64		1.32	73.84		
$C_{19}H_{18}O_3$	Sol/Sol	6-(5-hexenoxy)-2-naphthoic acid		303.2	2.93	9.66	86.3	[68]
	Sol/Nem	421.2	50.62		120.21			
	Nem/Liq	464.3	8.37		18.03	147.90		
$C_{19}H_{20}N_2O$	Sol/Nem	N-(4-pentyloxybenzylidene)-4-aminobenzonitrile		335.9	24.4	72.64	NA	[365]
	Nem/Liq	361.4	0.45		1.25	73.89		
$C_{19}H_{20}N_2O_2S$	Sol/SmeC	1-[2-(5-cyanothienyl)]-3-(4-pentyloxyphenylamino)-2-propen-1-one		404.2	8.40	20.78	106.2	[75]
	SmeC/Liq	455.2	4.49		9.86	30.64		
$C_{19}H_{20}N_2O_3$	Sol/Sol	4-pantanoyl-4'-ethanoyloxyazobenzene		367.5	6.84	18.61	94.4	[157]
	Sol/Nem	387.4	26.97		69.62			
	Nem/Liq	400.1	0.67		1.67	89.90		
$C_{19}H_{21}ClO_2S$	Sol/SmeC	4-chlorophenyl 4-hexyloxythiobenzoate		340.8	21.84	64.08	NA	[383]
	SmeC/Nem	359.9	2.47		6.86			
	Nem/Liq	364.2	0.75		2.06	73.00		
$C_{19}H_{21}N$	Sol/Nem	4-cyano-4'-hexylbiphenyl		286.7	24.27	84.65	97.9	[158]
	Nem/Liq	300.2	Value was not reported					

Independent values from another reference

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.		
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$				
<chem>C19H21NO</chem>	Sol/Nem	286.5	16.53	57.70	59.36	97.9 [233]		
	Nem/Liq	301.2	0.50	1.66				
	Independent values from another reference			72.64	73.60	97.9 [237]		
	Sol/Nem	288.0	20.92					
	Nem/Liq	303.0	0.29	0.96				
	Independent values from another reference			89.67	104.7	[158]		
	Sol/Nem	331.2	29.70					
	Nem/Liq	349.2	Value was not reported					
	Independent values from another reference			83.35	104.7	[222]		
	Sol/Nem	332.2	27.69					
<chem>C19H21NS</chem>	Nem/Liq	350.2	0.50	1.43	84.78	104.7		
	Independent values from another reference			103.54	104.7	[233]		
	Sol/Nem	330.1	34.18					
	Nem/Liq	349.2	0.63	1.80	105.34	104.7		
	Independent values from another reference			36.50	NA	[84]		
	Sol/Meso	307.1	11.21					
	Meso/Liq	347.0	11.21	32.31	68.81	NA		
	Independent values from another reference			85.08	108.4	[75]		
	Sol/SmeC	362.2	24.07					
<chem>C19H22BrNOS</chem>	SmeC/SmeC	367.5	0.90	2.45	100.7	[70]		
	SmeC/Liq	438.9	7.10	16.18				
	Independent values from another reference			69.93	105.4	[157,336]		
	Sol/SmeC	311.2	14.41					
	SmeC/Liq	329.2	7.78	23.63				
	Independent values from another reference			33.20	NA	[84]		
	Sol/SmeC	327.7	10.88					
	SmeC/SmeC	333.9	12.35	36.99				
	SmeC/Nem	363.0	3.39	9.34	111.8	[75]		
	Nem/Liq	370.4	5.79	15.63	95.16	105.4		
<chem>C19H22ClNOS</chem>	Independent values from another reference			46.30	107.5	[75]		
	Sol/SmeC	311.2	14.41					
	SmeC/Liq	329.2	7.78	23.63	69.93	100.7		
	Independent values from another reference			32.98	NA	[84]		
	Sol/SmeC	346.6	11.43					
	SmeC/SmeC	350.2	0.55	1.57	105.4	[157,336]		
	SmeC/Liq	432.6	6.57	15.19	49.74	107.5		
	Independent values from another reference			14.45	NA	[84]		
	Sol/Meso	341.8	4.94					
<chem>C19H22ClNO2S</chem>	Meso/SmeC	369.3	0.05	0.14	28.65	111.8		
	SmeC/SmeC	377.5	0.86	2.28				
	SmeC/Liq	458.4	5.4	11.78	28.65	111.8		
	Independent values from another reference			9.23	NA	[84]		
	Sol/SmeC	328.1	23.22					
	SmeC/SmeC	330.3	3.05	9.23	105.8	[157,335]		
	SmeC/Nem	334.9	3.41	10.18				
	Nem/Liq	336.3	1.17	3.48	93.66	105.8		
	Independent values from another reference			10.18	NA	[84]		
	Sol/SmeC	353.3	32.90					
<chem>C19H22FNOS</chem>	SmeC/Liq	371.3	3.64	9.80	102.92	107.5		
	Independent values from another reference			102.92	NA	[75]		
	Independent values from another reference							
	Sol/Nem	385.2	29.08	75.49				
<chem>C19H22FNO2</chem>	Nem/Liq	402.2	Not reported in paper		78.5	[197]		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)						
$C_{19}H_{22}N_2O$	Sol/Nem	4-( $\omega$ -aminohexyloxy)-4'-cyanobiphenyl		340.9	10.28	30.16	30.68	115.6 [17]
	Nem/Liq	365.3	0.19		0.52			
$C_{19}H_{22}O_2$	Liq/Nem	4-butoxy-4'-methoxy- <i>trans</i> -stilbene		435	0.76	1.7	89.7	105.7 [157]
	Sol/Liq	422	39.67			89.7		
$C_{19}H_{22}O_4$	4,4'- <i>bis</i> ( $\omega$ -hydroxyethoxy)- $\alpha$ -methylstilbene						123.8 [56]	
	Sol/Sol	376.2	10.35		27.5			
	Sol/SmeC	422.5	22.35		52.9			
	SmeC/Nem	433.6	4.68		10.8			
$C_{19}H_{23}NO$	Nem/Liq	438.2	Not reported in paper				97.6 [376, 381]	
	N-(4-methoxybenzylidene)-4-pentylaniline							
	Sol/Nem	312.9	22.7		72.55			
	Nem/Liq	333.3	0.67		2.01	74.56		
	Note: Temperature calculated from published enthalpy and entropy data in paper.							
$C_{19}H_{23}NO$	N-(4-ethoxybenzylidene)-4-butylaniline						97.6 [157, 340]	
	Sol/Nem	305.6	27.09		88.65			
	Nem/Liq	349.1	1.55		4.44	93.09	97.6 [376, 381]	
	Independent values from another reference							
	Sol/Nem	308.5	17.2		55.75			
	Nem/Liq	357.1	0.52		1.46	57.21		
$C_{19}H_{23}NO_3$	N-(4-methoxyphenyl)- $\alpha$ -(4-pentyloxyphenyl)nitronate						NA [162]	
	Sol/Nem	385.2	31.25		81.13			
	Nem/Liq	393.2	0.70		1.78	82.91		
$C_{19}H_{24}ClN_2O_2$	4-chloro-2'-hydroxy-4'-heptyloxyazobenzene						127 [229]	
	Sol/SmeC	340.2	34.6		101.70			
	SmeC/Nem	367.5	2.0		5.44			
	Nem/Liq	369.7	0.7		1.89	109.03		
$C_{19}H_{24}FNO_2$	4-cyano-2-fluorophenyl <i>trans</i> -4-pentylcyclohexane-1-carboxylate						102.8 [197]	
	Sol/Nem	348.7	30.10		86.32			
	Nem/Liq	366.7	Not reported in paper					
$C_{19}H_{24}N_2O$	4-pentyl-4'-ethoxyazobenzene						98.9 [141]	
	Sol/Nem	344.7	13.78		39.98			
	Nem/Liq	364.9	0.86		2.36	42.34		
$C_{19}H_{24}N_2O$	4-propyl-4'-butoxyazobenzene						98.9 [153]	
	Sol/Nem	344.4	41.11		119.37			
	Nem/Liq	358.0	2.64		7.37	126.74		
$C_{19}H_{26}O_3$	4-propylcyclohexyl 4-methoxycinnamate						100.8 [5]	
	Sol/Nem	340.2	19.5		57.32			
	Nem/Liq	364.2	0.9		2.47	59.79		
$C_{19}H_{27}NO_2S$	5-octyl-2-(4-isothiocyanatophenyl)-1,3-dioxane						NA [407]	
	Sol/SmeC	322.8	28.4		87.98			
	SmeC/Liq	356.1	3.82		10.73	98.71		
$C_{19}H_{27}NS$	4-( <i>trans</i> -4'-hexylcyclohexyl)isothiocyanatobenzene						NA [151]	
	Sol/Nem	285.7	26.80		93.80			
	Nem/Liq	315.8	1.60		5.07	98.87		
$C_{19}H_{28}O_3$	4-n-decyloxyacrylic acid							

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{19}H_{29}NO_5$	Sol/Smec	404.2	39.6	97.97		
	Smec/Nem	423.2	4.9	11.58		
	Nem/Liq	442.2	10.4	23.52	133.07	126.3 [22]
$C_{19}H_{29}NO_5$	4-dodecyloxy-3-nitrobenzoic acid					
	Sol/Smec	367.2	30.8	83.88		
	Smec/Liq	368.2	2.3	6.25	90.13	172.2 [1]
$C_{19}H_{30}O_3$	4-dodecyloxybenzoic acid					
	Sol/Smec	363.7	35.6	97.88		
	Smec/Nem	405.2	1.8	4.44		
$C_{19}H_{30}O_3$	Nem/Liq	413.2	2.0	4.84	107.16	155.4 [104–106]
	Independent values from another reference					
	Sol/Sol	356.2	17.99	50.51		
	Sol/Smec	363.2	16.32	44.93		
	Smec/Nem	406.2	3.35	8.25		
$C_{19}H_{30}O_4$	Nem/Liq	412.2	4.18	10.14	113.83	155.4 [156]
	4-dodecyloxy-2-hydroxybenzoic acid					
	Sol/Smec	372.6	30.9	82.93		
$C_{19}H_{38}O_6S$	Smec/Nem	380.0	2.9	7.63		
	Nem/Liq	383.1	3.3	8.61	99.17	160.8 [104]
	6-O-(propylene-[3'-S-decyl])-α-D-galactopyranose					
$C_{19}H_{38}O_6S$	Sol/Smec	374.2	54.29	145.08		
	Smec/Liq	439.2	0.55	1.25	146.33	178 [39]
$C_{19}H_{40}O_4$	1,2,18,19-tetrahydroxynonadecane					
	Sol/Meso	362.2	45.4	125.35		
	Meso/Meso	408.2	12.4	30.38		
$C_{19}H_{40}O_6$	Meso/Liq	412.2	15.0	36.39	192.12	244.1 [145]
	6-O-tridecyl-D-galacitol					
	Sol/Smec	415.2	59.27	142.75		
$C_{20}H_7ClF_{13}NO_2$	Smec/Liq	458.2	Not reported in paper			224.5 [183]
	5-chloro-2-(4-n-tridecafluorohexylphenyl)isoindole-1,3-dione					
	Sol/Sol	417.1	3.2	7.67		
$C_{20}H_7ClF_{13}NO_2$	Sol/Smec	428.4	25.4	59.29		
	Smec/Liq	438.0	6.5	14.84	81.80	102.4 [32]
	5-fluoro-2-(4-n-tridecafluorohexylphenyl)isoindole-1,3-dione					
$C_{20}H_7F_{14}NO_2$	Sol/Smec	399.6	26.9	67.32		
	Smec/Liq	410.7	4.9	11.93	79.25	102.8 [32]
$C_{20}H_{14}N_2O_4$	4-[(4-nitrophenyl)methylene]amino]phenol, benzoate (ester)					
	Sol/Nem	478.2	33.0	69.01		
	Nem/Liq	496.2	0.7	1.41	70.42	[346]
$C_{20}H_{14}N_2O_4$	4-[(phenylmethylene)amino]phenyl 4-nitrobenzoate					
	Sol/Nem	467.2	39.0	83.48		
	Nem/Liq	472.2	0.7	1.48	84.96	[346]
$C_{20}H_{14}N_2O_4$	4-[(4-nitrophenyl)imino]methyl]phenol, benzoate (ester)					
	Sol/Nem	473.2	43.0	90.87		
	Nem/Liq	494.2	0.9	1.82	92.69	[346]
$C_{20}H_{14}N_2O_4$	phenyl 4-[(4-nitrophenyl)methylene]amino]benzoate					
	Sol/Nem	474.2	40.0	84.35		
	Nem/Liq	483.2	0.7	1.45	85.80	[346]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated) (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{20}H_{14}N_2O_4$	Sol/Nem	439.2	37.0	84.24	85.95	[346]
	Nem/Liq	469.2	0.8	1.71		
$C_{20}H_{15}F$	Sol/Nem	366.2	23.68	64.66	66.07	[212]
	Nem/Liq	438.9	0.62	1.41		
$C_{20}H_{15}NO$	Sol/Nem	481.2	25.94	53.90	93.74	[277]
	Nem/Liq	575.2	Decomposed			
$C_{20}H_{16}N_2O_3$	Sol/Nem	408.2	38.0	93.09	93.74	[291]
	Nem/Liq	460.2	0.3	0.65		
$C_{20}H_{16}N_2O_3$	Sol/Nem	406.2	41.0	100.94	102.20	[291]
	Nem/Liq	475.2	0.6	1.26		
$C_{20}H_{16}N_2O_3$	Sol/Nem	394.2	31.0	78.64	79.73	[291]
	Nem/Liq	460.2	0.5	1.09		
$C_{20}H_{16}N_2O_3$	Sol/Nem	405.2	35.0	86.38	88.10	[291]
	Nem/Liq	464.2	0.8	1.72		
$C_{20}H_{17}F$	Sol/Nem	363.3	26.56	73.11	74.13	[212]
	Nem/Liq	363.9	0.37	1.02		
$C_{20}H_{17}F$	Sol/Nem	358.6	19.10	53.26	55.64	71.4
	Nem/Liq	360.9	0.86	2.38		
$C_{20}H_{17}F_9OS$	Sol/Sol	358.9	2.3	6.41	99.7	[66]
	Sol/Smec	367.7	1.3	3.53		
	Smec/Liq	413.3	8.4	20.32		
$C_{20}H_{17}NO$	Sol/Nem	443.2	35.15	79.30	83.2	[314, 155]
	Nem/Liq	448.7	0.39	0.88		
$C_{20}H_{18}FN$	Sol/Nem	375.2	33.5	89.29	68.2	[216]
	Nem/Liq	406.3	Value was not reported			
$C_{20}H_{19}FNOS$	Sol/Nem	334.3	30.33	90.73	NA	[135]
	Nem/Liq	338.6	3.47	10.25		
$C_{20}H_{19}NOS$	Sol/Sol	342.8	24.39	71.15	NA	[135, 238]
	Sol/Liq	386.0	13.81	35.77		
$C_{20}H_{20}O_3$	Sol/Sol	307.2	18.41	59.93	88.7	[68]
	Sol/Nem	421.2	50.63	120.23		
	Nem/Liq	464.4	8.37	18.03		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$\text{C}_{20}\text{H}_{21}\text{NO}_2$	Sol/Nem	318.5	37.03	116.26	105.5	[34]
	Nem/Liq	321.5	0.66	2.05		
$\text{C}_{20}\text{H}_{21}\text{NO}_2$	ethyl 4-(4-ethoxybenzylideneamino)cinnamate					
	Sol/Smec	354.2	27.30	77.08		
	Smec/Smec	391.7	2.10	5.36		
	Smec/Nem	429.7	5.10	11.87		
$\text{C}_{20}\text{H}_{21}\text{NO}_2$	Nem/Liq	432.2	0.50	1.16	95.47	[393]
	(E)-4-[2-[4-(5-hydroxypentyloxy)phenyl]vinyl]benzonitrile					
	Sol/Smec	362.0	12.4	34.25		
	Smec/Liq	368.1	16.3	44.28		
$\text{C}_{20}\text{H}_{21}\text{NO}_2$	Nem/Liq	378.0	5.4	14.29	92.82	[435]
$\text{C}_{20}\text{H}_{22}\text{N}_2\text{O}$	4-[hexyloxybenzylideneamine]-4'-benzonitrile					
	Sol/Sol	307.0	5.11	16.64		
	Sol/Nem	334.1	23.77	71.15		
	Nem/Liq	375.0	1.75	4.67	92.46	106.9
$\text{C}_{20}\text{H}_{22}\text{N}_2\text{O}$	N-(hexyloxybenzylidene)-4-aminobenzonitrile					
	Sol/Nem	329.7	25.8	78.25		
	Nem/Liq	375.4	0.64	1.70	79.95	[365]
$\text{C}_{20}\text{H}_{22}\text{N}_2\text{O}_2\text{S}$	1-[2-(5-cyanothienyl)]-3-(4-hexyloxyphenylamino)-2-propen-1-one					
	Sol/Smec	399.0	9.47	23.73		
	Smec/Liq	456.8	5.13	11.23	34.96	113.3
$\text{C}_{20}\text{H}_{22}\text{N}_2\text{O}_2\text{S}_2$	2,5-bis(4-propoxyphenyl)thiazolo[5,4-d]dithiazole					
	Sol/Nem	486.6	33.97	69.81		
	Nem/Liq	579.1	1.15	1.99	71.80	[269]
$\text{C}_{20}\text{H}_{22}\text{N}_2\text{O}_3$	4-pentanoyl-4'-propanoiloxyazobenzene					
	Sol/Sol	345.1	2.63	7.62		
	Sol/Smec	391.5	24.17	61.74		
	Smec/Nem	394.1	3.27	8.30		
$\text{C}_{20}\text{H}_{22}\text{N}_2\text{O}_3$	Nem/Liq	406.1	0.69	1.70	79.36	101.5
$\text{C}_{20}\text{H}_{22}\text{N}_4\text{O}_2\text{S}$	5-(4-pyridyl)-2-(4-hexyloxy)phenylamido-1,3,4-thiadiazole					
	Sol/Smec	484.5	18.7	38.60		
	Smec/Liq	523.8	6.7	12.79	51.39	[31]
$\text{C}_{20}\text{H}_{22}\text{O}_6$	butyl-4-(4-ethoxyphenylcarbonyl)phenylcarbonate					
	Sol/Nem	328.3	23.40	71.27		
$\text{C}_{20}\text{H}_{22}\text{O}_6$	Nem/Liq	358.1	Not reported in paper			[157]
$\text{C}_{20}\text{H}_{23}\text{ClO}_2\text{S}$	4-chlorophenyl 4-heptyloxythiobenzoate					
	Sol/Smec	344.7	27.74	80.48		
$\text{C}_{20}\text{H}_{23}\text{ClO}_2\text{S}$	Smec/Liq	365.9	4.56	12.46	92.94	NA
$\text{C}_{20}\text{H}_{23}\text{F}_9\text{S}$	4-propylcyclohexylbenzyl perfluorobutyl thioether					
	Sol/Smec	302.8	13.1	43.26		
$\text{C}_{20}\text{H}_{23}\text{F}_9\text{S}$	Smec/Liq	312.2	8.1	25.94	69.2	98.2
$\text{C}_{20}\text{H}_{23}\text{N}$	4'-heptyl-4-cyanobiphenyl					
	Sol/Nem	301.7	25.94	85.98		
$\text{C}_{20}\text{H}_{23}\text{N}$	Nem/Liq	315.2	Value was not reported			
Independent values from another reference						
Sol/Nem	303.0	28.49	94.03			
Nem/Liq	315.6	0.75	2.38	96.41	105	[233]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta S_{\text{pce}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.			
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )							
Independent values from another reference										
C <sub>20</sub> H <sub>23</sub> NO	Sol/Nem	303.0	25.52	84.22	86.21	105	[237]			
	Nem/Liq	316.0	0.63	1.99						
C <sub>20</sub> H <sub>23</sub> NO <sub>5</sub>	4'-heptyloxy-4-cyanobiphenyl		326.7	28.87	88.37	109.7	[158]			
	Sol/Nem	348.2	Value was not reported							
	Independent values from another reference		Sol/Nem	323.8	28.07	86.69	[233]			
C <sub>20</sub> H <sub>23</sub> NO <sub>3</sub>	Nem/Liq	349.4	0.54	1.55	88.24	109.7	[233]			
	4-(4-heptyloxyphenyleniminoethyl)benzoic acid		Sol/SmeC	462.2	16.0	34.62	[416]			
	SmeC/Nem	531.2	3.0	5.65						
C <sub>20</sub> H <sub>23</sub> NS	Nem/Liq	539.2	8.5	15.76	56.03					
	4-heptyloxyphenyl 4-nitrobenzoate		Sol/Nem	336.2	10.75	31.98	[236]			
	Nem/Liq	342.2	1.88	5.49	37.47	119.5				
C <sub>20</sub> H <sub>23</sub> NS	4-heptyl-4'-thiocyanatobiphenyl		Sol/Meso	331.0	14.81	44.74	NA			
	Meso/Liq	345.6	7.70	22.28	67.02					
	4-chlorobenzylidene 4'-heptylaniline		Sol/SmeC	357.5	20.02	56.00	[84]			
C <sub>20</sub> H <sub>24</sub> BrNOS	SmeC/SmeC	364.4	0.57	1.56	72.94	115.5	[75]			
	SmeC/Liq	440.7	6.78	15.38						
	1-[2-(5-heptylthienyl)]-3-(4-bromophenylamino)-2-propen-1-one									
C <sub>20</sub> H <sub>24</sub> CIN	Sol/SmeC	316.2	19.64	62.11	86.79	105.7	[70]			
	SmeC/Liq	332.7	8.21	24.68						
	1-[2-(5-heptylthienyl)]-3-(4-chlorophenylamino)-2-propen-1-one		Sol/SmeC	353.9	25.20	71.21	114.2			
C <sub>20</sub> H <sub>24</sub> CINOS	SmeC/Liq	432.8	7.05	16.29	87.50					
	1-[2-(5-chlorothienyl)]-3-(4-heptyloxyphenylamino)-2-propen-1-one		Sol/Meso	342.5	24.30	70.95	[75]			
	Meso/SmeC	369.7	0.03	0.08						
C <sub>20</sub> H <sub>24</sub> ClNO <sub>2</sub> S	SmeC/SmeC	378.7	1.89	4.99						
	SmeC/Liq	458.1	7.47	16.31	92.33	118.9				
	1-[2-(5-chlorothienyl)]-3-(4-heptyloxyphenylamino)-2-propen-1-one									
C <sub>20</sub> H <sub>24</sub> FNOS	Sol/SmeC	334.8	25.70	76.76	87.29	114.6	[75]			
	SmeC/Liq	373.3	3.93	10.53						
	1-[2-(5-heptylthienyl)]-3-(4-fluorophenylamino)-2-propen-1-one									
C <sub>20</sub> H <sub>24</sub> FNO <sub>2</sub>	Sol/Nem	364.7	29.71	81.46	85.6	[197]	[197]			
	Nem/Liq	398.2	Not reported in paper							
	4-cyano-2-fluorophenyl 4-butylbicyclo[2.2.2]octane-1-carboxylate									
C <sub>20</sub> H <sub>24</sub> N <sub>2</sub> O	Sol/Nem	342.2	21.17	61.86	63.96	122.7	[17]			
	Nem/Liq	361.2	0.76	2.10						
	4-( $\omega$ -aminoheptyloxy)-4'-cyanobiphenyl									
C <sub>20</sub> H <sub>24</sub> N <sub>2</sub> O <sub>3</sub>	Sol/Nem	328.2	39	118.83	121.45	[339]	[339]			
	Nem/Liq	343.2	0.9	2.62						
	4-pentanoyloxy-2-methyl-4'-ethoxyazobenzene									
C <sub>20</sub> H <sub>24</sub> N <sub>2</sub> O <sub>3</sub>	Sol/Nem	349.2	40	114.55						

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound			$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$			
$C_{20}H_{24}N_2O_3$	Nem/Liq	351.2	0.8	2.28	116.83		[339]
	Sol/Nem	347.8	35.9	103.22			
	Nem/Liq	403.6	1.29	3.20	106.42		[365]
$C_{20}H_{24}O_2$	2-methoxy-4'-pentyloxy- <i>trans</i> -stilbene						
	Sol/Nem	435.0	41.17	94.64			
	Nem/Liq	427.0	0.78	1.83	96.47	112.8	[157]
$C_{20}H_{24}O_3$	4-ethoxyphenyl 4'-pentybenzoate						
	Sol/Nem	334.4	33.35	99.73			
	Nem/Liq	336.6	0.82	2.44	102.2	110.2	[218]
$C_{20}H_{25}NO$	4-hexyloxybenzylidene-4'-toluidine						
	Sol/Sol	317.5	5.04	15.87			
	Sol/Nem	334.3	25.04	74.90			
$C_{20}H_{25}NO$	Nem/Liq	346.9	1.37	3.95	94.72	104.7	[157, 334]
	N-(4-methoxybenzylidene)-4-hexylaniline						
	Sol/Nem	308.5	27.5	89.14			
$C_{20}H_{25}NO$	Nem/Liq	326.5	0.67	2.05	91.19		[376, 381]
Note: Temperature calculated from published enthalpy and entropy data in paper.							
N-(4-ethoxybenzylidene)-4-pentylaniline							
$C_{20}H_{25}NO$	Sol/Nem	336.5	24.6	73.11			
	Nem/Liq	361.1	1.09	3.02	76.13		[376, 381]
Note: Temperature calculated from published enthalpy and entropy data in paper.							
$C_{20}H_{25}NO$	N-(4-propoxybenzylidene)-4-butylaniline						
	Sol/Nem	314.3	27.3	86.86			
	Nem/Liq	322.6	0.42	1.30	88.16		[376, 381]
Note: Temperature calculated from published enthalpy and entropy data in paper.							
$C_{20}H_{25}NO_2S$	4-isothiocyanatophenyl 4-butylbicyclo[2.2.2]octane-1-carboxylate						
	Sol/Nem	345.7	20.5	59.30			
	Nem/Liq	373.2	Not reported in paper			NA	[357]
$C_{20}H_{25}NO_3$	N-(4-methoxyphenyl)- $\alpha$ -(4-hexyloxyphenyl)nitron						
	Sol/Nem	380.2	27.87	73.30			
	Nem/Liq	398.2	0.82	2.06	75.36	NA	[162]
$C_{20}H_{26}ClN_2O_2$	4-chloro-2'-hydroxy-4'-octyloxyazobenzene						
	Sol/Smec	334.6	27.8	83.08			
	Smec/Nem	367.6	1.2	3.26			
$C_{20}H_{26}N_2O$	Nem/Liq	371.4	0.6	1.62	87.96	119.2	[229]
	4-pentyl-4'-propoxyazobenzene						
	Sol/Nem	329.9	16.10	48.80			
$C_{20}H_{26}N_2O$	Nem/Liq	348.1	0.95	2.73	51.53	106	[141]
	4-propyl-4'-pentyloxyazobenzene						
	Sol/Nem	326.6	35.02	107.23			
$C_{20}H_{26}N_2O$	Nem/Liq	346.8	2.96	8.54	115.77	106	[153]
	4-butyl-4'-butoxyazobenzene						
	Sol/Nem	338.2	14.98	44.29			
$C_{20}H_{26}N_2O$	Nem/Liq	347.7	0.65	1.87	46.16	106	[141]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound			$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$			
$C_{20}H_{26}N_2O$	Sol/Nem	4-heptyl-4'-methoxyazobenzene					
		307.2	20.12	65.49			
	Nem/Liq	336.1	0.81	2.41	67.90	106	[390]
$C_{20}H_{26}N_2O$	Sol/Nem	bis(4-butylphenyl)diazene N-oxide					
		292.8	13.4	45.77			
	Nem/Liq	305.7	0.35	1.14	46.91		[365]
$C_{20}H_{26}N_2O_3$	Sol/Sol	4,4'-dibutoxyazoxybenzene					
		338.0	5.20	15.38			
		360.6	1.37	3.80			
		377.5	19.50	51.66			
	Nem/Liq	411.9	1.25	3.03	73.87	121.4	[9]
Independent values from another reference							
	Sol/Sol	315.5	10.44	33.09			
	Sol/Sol	323.4	0.93	2.88			
	Sol/Sol	360.5	1.57	4.36			
	Sol/Nem	376.7	20.96	55.64			
	Nem/Liq	409.9	1.03	2.51	98.48	121.4	[155,157]
Independent values from another reference							
$C_{20}H_{26}N_2O_3$	Sol/Nem	375.2	20.94	55.81			
	Nem/Liq	409.9	1.03	2.51	58.32	121.4	[179]
Independent values from another reference							
	Sol/Sol	321.0	7.18	22.37			
	Sol/Sol	358.0	1.19		3.32		
	Sol/Nem	377.0	20.3	53.84			
	Nem/Liq	411.0	1.29	3.14	82.67	121.4	[440]
$C_{20}H_{26}O_2S$	Sol/Meso	2-octanoyl-5-(4-ethoxyphenyl)thiophene					
		394.0	31.78	80.66			
	Meso/Liq	396.3	5.29	13.35	94.01	127.8	[18]
$C_{20}H_{27}NS$	1-(4-isothiocyanantophenyl)-4-pentylbicyclo[2.2.2]octane						
	Sol/Sol	337.7	5.0	14.81			
	Sol/Nem	347.2	15.5	44.64			
	Nem/Liq	372.7	Not reported in paper				[357]
$C_{20}H_{28}$	4-ethyl-4'-(5-octenyl)biphenyl						
	Sol/SmeC	282.6	4.80	16.99			
	SmeC/Liq	301.4	7.90	26.21	43.20	113.4	[165]
$C_{20}H_{28}$	4-butyl-4'-(5-hexenyl)biphenyl						
	Sol/SmeC	248.6	2.20	8.85			
	SmeC/Liq	315.6	9.60	30.41	39.26	114.9	[165]
$C_{20}H_{28}O_3$	4-butylcyclohexyl 4-methoxycinnamate						
	Sol/Nem	330.6	26.7	80.76			
	Nem/Liq	360.4	0.7	1.94	82.52	107.9	[5]
$C_{20}H_{29}NS$	4-( <i>trans</i> -4'-heptylcyclohexyl)isothiocyanatobenzene						
	Sol/Nem	311.3	32.1	103.12			
	Nem/Liq	325.3	1.10	3.38	106.50	NA	[151]
$C_{20}H_{31}NO_5$	4-tridecyloxy-3-nitrobenzoic acid						
	Sol/SmeC	362.2	33.8	93.32			
	SmeC/Liq	368.2	1.9	5.16	98.48	181.5	[1]
$C_{20}H_{32}N_2O_2$	N,N'-dipentanoyl-3,4,5,6-tetramethylbenzene-1,2-diamine						
	Sol/Meso	401.2	8	19.94			

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound			$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$			
$C_{20}H_{32}O_3$	Meso/Meso	447.2	4	8.94	76.80	97.8	[61]
	Meso/Meso	483.2	5	10.34			
	Meso/Liq	532.2	20	37.58			
$C_{20}H_{38}O_7$	4-tridecyloxybenzoic acid						
	Sol/Smec	373.2	46.44	124.44			
	Smec/Nem	408.2	4.18	10.24			
	Nem/Liq	410.2	2.51	6.12	140.80	164.7	[156]
$C_{20}H_{40}O_7S$	6-O-tetradecanoyl- $\alpha$ -D-galactopyranose						
	Sol/Smec	390.2	27.45	70.35	72.23	207.5	[39]
	Smec/Liq	457.2	0.86	1.88			
$C_{20}H_{40}O_6$	6-S-tetradecyl-6-thio- $\alpha$ -D-galactopyranose						
	Sol/Smec	373.2	41.26	110.56	118.00	211.2	[39]
	Smec/Liq	456.2	2.94	6.44			
$C_{20}H_{42}O_4$	1,2,19,20-tetrahydroxyeicosane						
	Sol/Meso	357.2	24.0	67.19	138.57	253.4	[145]
	Meso/Meso	411.2	14.0	34.05			
$C_{20}H_{42}O_6$	Meso/Liq	415.2	15.5	37.33			
	6-O-tetradecyl-D-galactitol						
	Sol/Smec	411.2	52.73	128.23	128.71	233.8	[183]
$C_{21}H_{10}F_{13}NO_2$	Smec/Liq	440.2	0.21	0.48			
	5-methyl-2-(4-n-tridecafluorohexylphenyl)isoindole-1,3-dione						
	Sol/Smec	418.3	50.6	120.97	141.96	101.7	[32]
$C_{21}H_{10}F_{13}NO_3$	Smec/Liq	438.4	9.2	20.99			
	5-methoxy-2-(4-n-tridecafluorohexylphenyl)isoindole-1,3-dione						
	Sol/Sol	407.0	5.5	13.51	109.42	108.5	[32]
$C_{21}H_{13}F_{13}SO$	Sol/Smec	423.1	34.9	82.49			
	Smec/Liq	454.7	6.1	13.42			
	2(perfluorohexyl)ethyl 4-phenylthiobenzoate						
$C_{21}H_{14}Cl_3NO_2$	Sol/Smec	324.8	27.6	85.90	120.64	NA	[38]
	Smec/Liq	423.2	14.7	34.74			
	2,3,4-trichloro-N-[4-(4-methylbenzoyloxy)benzylidene]aniline						
$C_{21}H_{14}Cl_3NO_3$	Sol/Nem	449.2	33.0	73.46	74.54	72.5	[191]
	Nem/Liq	463.2	0.5	1.08			
	2,3,4-trichloro-N-[4-(4-methoxybenzoyloxy)benzylidene]aniline						
$C_{21}H_{14}Cl_3NO_3$	Sol/Nem	456.2	42.0	92.06	92.87	77.2	[191]
	Nem/Liq	495.2	0.4	0.81			
	4-[(phenylmethylene)amino]phenyl 4-cyanobenzoate						
$C_{21}H_{14}N_2O_2$	Sol/Nem	425.2	28.0	65.85	67.28	[346]	
	Nem/Liq	491.2	0.7	1.43			
	4-[(4-cyanophenyl)methylene]amino]phenol, benzoate (ester)						
$C_{21}H_{14}N_2O_2$	Sol/Nem	437.2	36.0	82.34	83.72	[346]	
	Nem/Liq	506.2	0.7	1.38			
	4-[(phenylimino)methyl]phenyl 4-cyanobenzoate						
$C_{21}H_{14}N_2O_2$	Sol/Nem	433.2	31.0	71.56	73.18	[346]	
	Nem/Liq	494.2	0.8	1.62			

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)						
$C_{21}H_{14}N_2O_2$	Sol/Nem	phenyl 4-[[4-(cyanophenyl)imino]phenyl]benzoate		456.2	32.0	70.14		[346]
	Nem/Liq			491.2	0.7	1.43	71.57	
$C_{21}H_{14}N_2O_2$	Sol/Nem	4-[[4-(cyanophenyl)imino]methyl]phenol, benzoate (ester)		452.2	37.0	81.82		[346]
	Nem/Liq			499.2	0.8	1.60	83.42	
$C_{21}H_{14}N_2O_2$	Sol/Nem	phenyl 4-[[4-(cyanophenyl)methylene]amino]benzoate		447.2	27.0	60.38		[346]
	Nem/Liq			496.2	0.7	1.41	61.79	
$C_{21}H_{14}N_2O_2$	Sol/Nem	4-cyanophenyl 4-[(phenylamino)methyl]benzoate		407.2	34.0	83.50		[346]
	Nem/Liq			484.2	0.8	1.65	85.15	
$C_{21}H_{14}N_2O_2$	Sol/Nem	4-cyanophenyl 4-[(phenylmethylene)amino]benzoate		458.2	40.0	87.30		[346]
	Nem/Liq			476.2	0.6	1.26	88.56	
$C_{21}H_{15}Cl_2NO_2$	Sol/Nem	2,4-dichloro-N-[4-(4-methylbenzoyloxy)benzylidene]aniline		439.2	37.0	84.24		[191]
	Nem/Liq			470.2	0.6	1.28	85.52	
$C_{21}H_{15}Cl_2NO_2$	Sol/Nem	3,4-dichloro-N-[4-(4-methylbenzoyloxy)benzylidene]aniline		425.2	41.0	96.43		[191]
	Nem/Liq			463.2	0.5	1.08	97.51	
$C_{21}H_{15}Cl_2NO_3$	Sol/Nem	2,4-dichloro-N-[4-(4-methoxybenzoyloxy)benzylidene]aniline		416.2	43.0	103.32		[191]
	Nem/Liq			504.2	2.1	4.17	107.49	
$C_{21}H_{15}NO_3S$	Sol/Smec	4'-(2-propenyl)oxobiphenyl 5-cyano-2-thiophenecarboxylate		404.1	60.25	149.10		[63]
	Smec/Nem			417.6	5.44	13.03		
	Nem/Liq			481.2	1.26	2.62	164.75	
$C_{21}H_{16}BrNO_3$	Sol/Nem	4-bromo-N-[4-(4-methoxybenzoyloxy)benzylidene]aniline		460.2	38.0	82.61		[191]
	Nem/Liq			553.2	0.9	1.63	84.24	
$C_{21}H_{16}BrNO_3$	Sol/Nem	4-methoxyphenyl 4-[4-(4-bromophenyliminomethyl)]benzoate		442.2	42.0	94.98		[203]
	Nem/Liq			553.2	0.7	1.27	96.25	
$C_{21}H_{16}ClNO_2$	Sol/Nem	4-chloro-N-[4-(4-methylbenzoyloxy)benzylidene]aniline		442.2	29.0	65.58		[191]
	Nem/Liq			530.2	0.6	1.13	66.71	
$C_{21}H_{16}ClNO_3$	Sol/Nem	4-chloro-N-[4-(4-methoxybenzoyloxy)benzylidene]aniline		445.2	27.0	60.65		[191]
	Nem/Liq			558.2	1.6	2.87	63.52	
$C_{21}H_{16}ClNO_3$	Sol/Nem	4-methoxyphenyl 4-[4-(4-chlorophenyliminomethyl)]benzoate		441.2	39.0	88.40		[203]
	Nem/Liq			557.2	0.8	1.44	89.54	
$C_{21}H_{16}FNO_3$	Sol/Nem	3-fluoro-N-[4-(4-methoxybenzoyloxy)benzylidene]aniline		408.2	39.0	95.54		[191]
	Nem/Liq			423.2	0.4	0.95	96.49	

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)						
$\text{C}_{21}\text{H}_{16}\text{FNO}_3$		4-fluoro-N-[4-(4-methoxybenzoyloxy)benzylidene]aniline						[191]
	Sol/Nem	404.2	32.0		79.17			
	Nem/Liq	528.2	0.8		1.51	80.68		
$\text{C}_{21}\text{H}_{16}\text{FNO}_3$		4-methoxyphenyl 4-[4-(2-fluorophenyliminomethyl)]benzoate						[203]
	Sol/Nem	391.2	31.0		79.24			
	Nem/Liq	425.2	0.3		0.71	79.95		
$\text{C}_{21}\text{H}_{16}\text{FNO}_3$		4-methoxyphenyl 4-[4-(4-fluorophenyliminomethyl)]benzoate						[203]
	Sol/Nem	419.2	36.0		85.88			
	Nem/Liq	524.2	0.8		1.53	87.41		
$\text{C}_{21}\text{H}_{16}\text{F}_2\text{O}$		1-(6-propoxy-2-naphthyl)-2-(3,4-difluorophenyl)acetylene						[95]
	Sol/Nem	362.8	25.6		70.56			
	Nem/Liq	364.5	0.4		1.10	71.66		
$\text{C}_{21}\text{H}_{16}\text{INO}_3$		4-iodo-N-[4-(4-methoxybenzoyloxy)benzylidene]aniline						[191]
	Sol/Nem	475.2	21.0		44.19			
	Nem/Liq	547.2	0.4		0.73	44.92		
$\text{C}_{21}\text{H}_{16}\text{INO}_3$		4-methoxyphenyl 4-[4-(4-iodophenyliminomethyl)]benzoate						[202]
	Sol/Nem	466.2	46.0		98.67			
	Nem/Liq	547.2	0.5		0.91	99.58		
$\text{C}_{21}\text{H}_{17}\text{F}$		4-[(4-penten-1-yl)phenyl-1,3-butadiynyl]-4-fluorobenzene						[212]
	Sol/Nem	358.8	19.56		54.52			
	Nem/Liq	442.7	0.79		1.78	56.30		
$\text{C}_{21}\text{H}_{17}\text{FO}$		1-(6-propoxy-2-naphthyl)-2-(4-fluorophenyl)acetylene						[95]
	Sol/Nem	395.2	27.8		70.34			
	Nem/Liq	404.7	0.5		1.23	71.57		
$\text{C}_{21}\text{H}_{17}\text{NO}$		4-ethoxy-4"-cyano-p-terphenyl						[277]
	Sol/SmeC	438.2	4.49		10.25			
	SmeC/Nem	494.2	16.74		33.87			
$\text{C}_{21}\text{H}_{17}\text{NO}_3$		Nem/Liq Decomposed						[277]
	Sol/Nem	409.2	34.0		83.09			
	Nem/Liq	444.2	0.3		0.68	83.77		
$\text{C}_{21}\text{H}_{17}\text{NO}_3$		phenyl 4-(4-methoxybenzylideneamino)benzoate						[292]
	Sol/Nem	391.2	33.0		84.36			
	Nem/Liq	452.2	0.5		1.11	85.47		
$\text{C}_{21}\text{H}_{17}\text{NO}_3$		4-[(4-methoxyphenyl)methylene]amino]phenol, benzoate (ester)						[346,347]
	Sol/Nem	412.2	33.0		80.06			
	Nem/Liq	443.2	0.6		1.35	81.41		
$\text{C}_{21}\text{H}_{17}\text{NO}_3$		4-[(phenylmethylene)amino]phenyl 4-methoxybenzoate						[346,347]
	Sol/Nem	402.2	31.0		77.08			
	Nem/Liq	450.2	0.4		0.89	77.98		
$\text{C}_{21}\text{H}_{17}\text{NO}_3$		phenyl 4-[[4-methoxyphenyl]imino]phenylbenzoate						[346,347]
	Sol/Nem	418.2	34.0		81.30			
	Nem/Liq	460.2	0.3		0.65	81.95		
$\text{C}_{21}\text{H}_{17}\text{NO}_3$		phenyl 4-[[4-methoxyphenyl]imino]methylphenol, benzoate (ester)						[346,347]
	Sol/Nem	417.2	34.0		81.50			
	Nem/Liq	460.2	0.3		0.65	82.15		
$\text{C}_{21}\text{H}_{17}\text{NO}_3$		phenyl 4-[[4-methoxyphenyl)methylene]amino]benzoate						[346,347]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated) (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{21}H_{17}NO_3$	Sol/Nem	412.2	32.0	77.63		
	Nem/Liq	447.2	0.4	0.89	78.52	[346,347]
$C_{21}H_{17}NO_3$	Sol/Nem	424.2	40.0	94.30		
	Nem/Liq	446.2	0.3	0.67	94.97	[346,347]
$C_{21}H_{17}NO_3$	Sol/Nem	426.2	33.0	77.43		
	Nem/Liq	431.2	0.5	1.16	78.59	[346,347]
$C_{21}H_{17}NO_4$	Sol/Nem	499.2	56.0	112.18		
	Nem/Liq	603.2		Decomposed prior to transition		[203]
$C_{21}H_{17}NO_4$	Sol/Nem	413.2	35.0	84.70		
	Nem/Liq	426.2	0.3	0.70	85.40	[203]
$C_{21}H_{17}NO_4$	Sol/Nem	476.2	35.0	73.50		
	Nem/Liq	597.2		Decomposed prior to transition		[203]
$C_{21}H_{17}O_4S$	Sol/Nem	387.0	73.64	190.28		
	Nem/Liq	484.1	3.77	7.79	198.07	[63]
$C_{21}H_{18}N_2O_3$	Sol/Nem	411.2	31.0	75.39		
	Nem/Liq	468.2	0.4	0.85	76.24	[291]
$C_{21}H_{18}N_2O_3$	Sol/Nem	406.2	34.0	83.70		
	Nem/Liq	481.2	0.9	1.87	85.57	[291]
$C_{21}H_{18}N_2O_3$	Sol/Nem	401.2	38.0	94.72		
	Nem/Liq	469.2	0.6	1.28	96.00	[291]
$C_{21}H_{18}N_2O_3$	Sol/Nem	389.2	34.0	87.36		
	Nem/Liq	471.2	0.8	1.70	89.06	[291]
$C_{21}H_{19}F$	Sol/Nem	361.6	24.69	68.28		
	Nem/Liq	368.8	0.53	1.44	69.72	[212]
$C_{21}H_{19}F$	Sol/Nem	358.9	26.04			
	Nem/Liq	353.3	0.21			[270]
Note: Nem/Liq transition temperature was reported in the paper to be lower than the Sol/Nem transition temperature.						
$C_{21}H_{19}NS$						
	Sol/Nem	428.5	11.2	26.14		
	Nem/Liq	474.0	0.83	1.75	27.89	101.7
$C_{21}H_{21}FNOS$						
	Sol/Nem	322.9	29.50	91.36		
	Nem/Liq	344.9	0.48	1.39	92.75	NA
$C_{21}H_{21}N$						
	Sol/Meso	353.6	2.31	6.53		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{21}H_{21}NO$	Meso/Liq	386.6	7.54	19.50	26.03	99
	Sol/Nem	414.7	25.8	64.21		[260]
	Nem/Liq	482.3	0.4	0.83	65.04	[210]
$C_{21}H_{21}NOS$	4'-hexyloxy-4-isothiocyanatotolane					
	Sol/Sol	331.6	26.28	79.25		
	Sol/Nem	385.9	12.97	33.61		
	Nem/Liq	386.8	1.17	3.02	115.88	NA [135]
$C_{21}H_{22}N_2O_2$	4-[(4-cyanophenyl)imino]methyl]phenyl heptanoate					
	Sol/Nem	328.6	30.1	91.60		
	Nem/Liq	368.7	0.63	1.71	93.31	[365]
$C_{21}H_{22}O_3$	6-(7-octenylxyloxy)-2-naphthoic acid					
	Sol/Sol	331.1	27.20	82.15		
	Sol/Sol	378.9	0.84	2.22		
	Sol/Nem	420.8	64.02	152.14		
$C_{21}H_{23}NO_2$	Nem/Liq	454.1	7.95	17.51	254.03	100.5 [68]
	4-cyanophenyl 4-(heptyl)benzoate					
	Sol/Nem	316.7	35.48	112.03		
$C_{21}H_{24}N_2O$	Nem/Liq	329.5	0.94	2.85	114.88	112.7 [34]
	N-(4-heptyloxybenzylidene)-4-aminobenzonitrile					
	Sol/Nem	340.6	34.3	100.70		
$C_{21}H_{24}N_2O_2S$	Nem/Liq	369.7	0.71	1.92	102.62	[365]
	1-[2-(5-cyanothienyl)]-3-(4-heptyloxyphenylamino)-2-propen-1-one					
	Sol/Smec	397.4	11.18	28.13		
$C_{21}H_{24}N_2O_3$	Smec/Smec	459.5	5.65	12.30	40.43	120.4 [75]
	Nem/Liq	373.3	14.23	38.12		
	Smec/Smec	378.1	14.24	37.66		
	Smec/Nem	398.4	3.41	8.56		
$C_{21}H_{24}N_2O_3$	Nem/Liq	404.7	0.77	1.90	86.24	108.6 [157]
	4-propanoyl-4'-hexanoyloxyazobenzene					
	Sol/Smec	372.2	29.79	80.04		
$C_{21}H_{24}N_4O_2S$	Smec/Nem	411.7	3.93	9.55		
	Nem/Liq	420.7	0.88	2.09	91.68	108.6 [157]
	5-(4-pyridyl)-2-(4-heptyloxy)phenylamido-1,3,4-thiadiazole					
$C_{21}H_{24}N_4O_2S$	Sol/Smec	482.7	19.8	41.02		
	Smec/Liq	529.3	8.5	16.06	57.08	[31]
$C_{21}H_{25}ClN_2O_4$	4-(4-propyloxyphenylazoxy)phenyl 2S,3S-2-chloro-3-methylpentanoate					
	Sol/Nem	323.5	25.43	78.61		
	Nem/Liq	325.9	0.28	0.86	79.47	[47]
$C_{21}H_{25}ClO_2S$	4-chlorophenyl 4-octyloxythiobenzoate					
	Smec/Smec	337.6	30.17	89.37		
$C_{21}H_{25}FO_3S$	Smec/Liq	368.8	3.60	9.76	99.13	NA [383]
	S-(2-fluoro-4-hexyloxyphenyl) 4-ethoxythiobenzoate					
	Sol/Nem	351.2	45.2	128.70		
$C_{21}H_{25}N$	Nem/Liq	352.2	1.7	4.83	133.53	NA [4]
	4-octyl-4'-cyanobiphenyl					
$C_{21}H_{25}N$	Smec/Smec	294.5	22.62	76.81		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta S_{\text{pce}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )				
$\text{C}_{21}\text{H}_{25}\text{NO}$	Smec/Nem	306.4	0.02	0.07	78.89	112.1	[33, 157]
	Nem/Liq	313.8	0.63	2.01			
Independent values from another reference							
Sol/Smec	292.9	29.16		99.56			
Smec/Nem	306.3	0.08		0.26			
Nem/Liq	312.4	0.88		2.82	102.64	112.1	[233]
Independent values from another reference							
Sol/Smec	295.0	23.43		79.42			
Smec/Nem	307.0	0.04		0.13			
Nem/Liq	313.0	0.67		2.14	81.69	112.1	[237]
Independent values from another reference							
Sol/Smec	294.4	25.7		87.30			
Smec/Nem	307.0	0.40		1.30			
Nem/Liq	313.9	0.61		1.94	90.54	112.1	[358]
Independent values from another reference							
Sol/Smec	297.2	25.3		85.13			
Smec/Nem	307.2	0.13		0.42			
Nem/Liq	315.8	0.97		3.07	88.62	112.1	[365]
Independent values from another reference							
Sol/Smec	294.4	25.7		87.30			
Smec/Nem	307.2	Too small to be measured					
Nem/Liq	313.7	0.61		1.94	89.24	112.1	[387]
$\text{C}_{21}\text{H}_{25}\text{NO}$	4-octyloxy-4'-cyanobiphenyl						
	Sol/Smec	327.7	26.65	81.32			
	Smec/Nem	340.4	0.03	0.09			
	Nem/Liq	353.4	0.63	1.78	83.19	116.8	[33]
Independent values from another reference							
Sol/Smec	326.1	29.79		91.35			
Smec/Nem	339.9	0.04		0.12			
Nem/Liq	352.6	0.88		2.50	93.97	116.8	[233]
Independent values from another reference							
Sol/Smec (I)	330.2	43.93		133.04			
Sol/Smec(II)	323.2	49.37		152.75			
Smec/Liq	Not reported in paper					116.8	[372]
$\text{C}_{21}\text{H}_{25}\text{NO}_3$	4-(4-heptyloxyphenyliminomethyl)benzoic acid						
	Sol/Smec	463.2	14.5	31.30			
	Smec/Nem	528.2	Not reported in paper				
$\text{C}_{21}\text{H}_{25}\text{NO}_5$	Nem/Liq	534.2	16.3	30.51			[416]
	4-nitrophenyl-4'-octyloxybenzoate						
	Sol/Smec	323.2	34.72	107.43			
$\text{C}_{21}\text{H}_{25}\text{NO}_5$	Smec/Nem	334.0	0.09	0.27			
	Nem/Liq	341.0	0.29	0.85	108.55	126.6	[157]
	4-octyloxyphenyl 4-nitrobenzoate						
$\text{C}_{21}\text{H}_{25}\text{NO}_5$	Sol/Smec	337.2	22.80	67.62			
	Smec/Nem	344.2	0.04	0.12			
	Nem/Liq	346.4	2.30	6.64	74.38	126.6	[236]
$\text{C}_{21}\text{H}_{25}\text{NS}$	4-octyl-4'-thiocyanatobiphenyl						
	Sol/Meso	303.6	18.40	60.61			
	Meso/Liq	342.6	10.13	29.57	90.18	NA	[84]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.		
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$				
Independent values from another reference								
C <sub>21</sub> H <sub>26</sub> BrNO	Sol/Smec	301.5	16.4	54.39	NA	[407]		
	Smec/Liq	345.0	10.3	29.86				
C <sub>21</sub> H <sub>26</sub> BrNOS	4-octyloxy-N-(4-bromobenzylidene)aniline							
	Sol/Smec	365.5	29.34	80.27	120.9	[204]		
	Smec/Smec	384.8	3.31	8.60				
	Smec/Liq	388.2	7.69	19.80				
C <sub>21</sub> H <sub>26</sub> BrNOS	1-[2-(5-octylthienyl)]-3-(4-bromophenylamino)-2-propen-1-one							
	Sol/Smec	363.7	28.73	78.99	122.6	[75]		
	Smec/Liq	440.1	7.73	17.56				
C <sub>21</sub> H <sub>26</sub> ClN	4-chlorobenzylidene 4'-octylaniline							
	Sol/Smec	315.7	27.37	86.70	112.8	[70]		
	Smec/Liq	330.7	7.66	23.16				
C <sub>21</sub> H <sub>26</sub> ClNO	4-octyloxy-N-(4-chlorobenzylidene)aniline							
	Sol/Smec	363.3	30.77	84.7	119.6	[204]		
	Smec/Smec	370.4	2.80	7.55				
C <sub>21</sub> H <sub>26</sub> ClNO	Smec/Liq	381.4	6.22	16.3	108.55			
	1-[2-(5-octylthienyl)]-3-(4-chlorophenylamino)-2-propen-1-one							
	Sol/Smec	361.3	30.68	84.92	121.3	[75]		
C <sub>21</sub> H <sub>26</sub> ClNO	Smec/Liq	432.3	7.13	16.49	101.41			
	1-[2-(5-chlorothienyl)]-3-(4-octyloxyphenylamino)-2-propen-1-one							
	Sol/Meso	338.7	21.61	63.80	126.0	[75]		
C <sub>21</sub> H <sub>26</sub> ClNO <sub>2</sub> S	Meso/Smec	368.8	0.02	0.05				
	Smec/Smec	379.0	1.96	5.17				
	Smec/Liq	457.8	7.74	16.91	85.93			
	Sol/Smec	356.6	41.90	117.50				
C <sub>21</sub> H <sub>26</sub> FNOS	Smec/Liq	375.1	4.20	11.20	128.70	121.7		
	1-[2-(5-octylthienyl)]-3-(4-fluorophenylamino)-2-propen-1-one							
	Sol/Nem	357.2	32.20	90.15	[75]			
C <sub>21</sub> H <sub>26</sub> FNO <sub>2</sub>	Nem/Liq	406.2	Not reported in paper					
	4-cyano-2-fluorophenyl 4-pentylbicyclo[2.2.2]octane-1-carboxylate							
	Sol/Nem	336.7	27.4	81.38	NA	[357]		
C <sub>21</sub> H <sub>26</sub> FNO <sub>2</sub> S	Nem/Liq	375.7	Not reported in paper					
	3-fluoro-4-isothiocyanatophenyl 4-pentylbicyclo[2.2.2]octane-1-carboxylate							
	Sol/Nem	346.7	26.15	75.43	129.8	[17]		
C <sub>21</sub> H <sub>26</sub> N <sub>2</sub> O	Nem/Liq	359.3	0.74	2.06	77.49			
	4-( $\omega$ -aminooctyloxy)-4'-cyanobiphenyl							
	Sol/Nem	349.2	31	88.77	121.0	[204]		
C <sub>21</sub> H <sub>26</sub> N <sub>2</sub> O <sub>3</sub>	Nem/Liq	374.2	0.6	1.60	90.37			
	4-pentanoyloxy-2,3-dimethyl-4'-ethoxyazobenzene							
	Sol/Nem	330.5	23.3	70.50	110.1	[339]		
C <sub>21</sub> H <sub>27</sub> ClN <sub>2</sub> O <sub>2</sub>	Smec/Smec	371.2	4.0	10.78	81.28			
	Smec/Liq	4-chloro-2'-hydroxy-4'-nonyloxyazobenzene						
	Sol/Smec	330.5	23.3	70.50	110.1	[229]		
C <sub>21</sub> H <sub>26</sub> N <sub>2</sub> O <sub>3</sub>	4-[(4-ethoxyphenyl)azo]phenyl heptanoate							

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.	
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$			
$C_{21}H_{27}NO$	Sol/Nem	338.8	21.7	64.05	67.14	[365]	
	Nem/Liq	391.8	1.21	3.09			
$C_{21}H_{27}NO$	N-(4-methoxybenzylidene)-4-heptylaniline					[376, 381]	
	Sol/Nem	308.2	33.7	109.34			
	Nem/Liq	333.3	0.63	1.89	111.23		
Note: Temperature was calculated from published enthalpy and entropy data in paper.							
$C_{21}H_{27}NO$	N-(4-ethoxybenzylidene)-4-hexylaniline					[376, 381]	
	Sol/Nem	312.8	25.9	82.80			
$C_{21}H_{27}NO$	Nem/Liq	350.0	0.88	2.51	85.31	[376, 381]	
	Note: Temperature was calculated from published enthalpy and entropy data in paper.						
	N-(4-propoxybenzylidene)-4-pentylaniline						
$C_{21}H_{27}NO$	Sol/Smec	Not reported in paper				[376]	
	Smec/Nem	295.0	4.94	16.75			
$C_{21}H_{27}NO$	Nem/Liq	338.5	0.92	2.72		[376]	
	Note: Temperatures calculated from published enthalpy and entropy data in paper.						
	N-(4-butoxybenzylidene)-4-butylaniline						
$C_{21}H_{27}NO$	Sol/Sol	281.2	1.7	6.05	26.90	[376, 381]	
	Smec/Smec	285.2	1.9	6.66			
$C_{21}H_{27}NO$	Smec/Smec	317.2	3.85	12.14	26.90	[376, 381]	
	Smec/Nem Peak was not resolved						
$C_{21}H_{27}NO$	Nem/Liq	346.9	0.71	2.05	26.90	[376, 381]	
	Note: Temperatures calculated from published enthalpy and entropy data in paper.						
$C_{21}H_{27}NO_2$	<i>trans</i> -4-octyloxy-3'-stilbazole-N-oxide					[278]	
	Sol/Smec	351.3	0.68	1.94			
$C_{21}H_{27}NO_2$	Smec/Smec	375.5	6.44	17.15		[278]	
	Smec/Liq	384.0	19.92	51.88	70.97		
$C_{21}H_{27}NO_2$	<i>trans</i> -4-octyloxy-4'-stilbazole-N-oxide					[278]	
	Sol/Smec	385.6	29.00	75.21			
$C_{21}H_{27}NO_2S$	Smec/Liq	389.9	2.73	7.00	82.21	[278]	
	4-isothiocyanatophenyl 4-pentylbicyclo[2.2.2]octane-1-carboxylate						
$C_{21}H_{27}NO_3$	Sol/Nem	347.7	18.0	51.77	NA	[357]	
	Nem/Liq	386.7	Not reported in paper				
$C_{21}H_{27}NO_3$	N-(4-methoxyphenyl)- $\alpha$ -(4-heptyloxyphenyl)nitrone					[162]	
	Sol/Nem	393.2	34.86	88.66			
$C_{21}H_{28}FNO_2$	Nem/Liq	398.2	0.92	2.31	90.97	NA	[162]
	4-cyano-2-fluorophenyl <i>trans</i> -4-heptylcyclohexane-1-carboxylate					[197]	
$C_{21}H_{28}N_2O$	Sol/Nem	349.2	35.10	100.52	113.1	[141]	
	Nem/Liq	365.2	Not reported in paper				
$C_{21}H_{28}N_2O$	4-butyl-4'-pentyloxyazobenzene					[141]	
	Sol/Nem	315.7	11.57	36.65			
$C_{21}H_{28}N_2O$	Nem/Liq	338.6	0.52	1.54	38.19	NA	[141]
	4-propyl-4'-hexyloxyazobenzene					[153]	
$C_{21}H_{28}N_2O$	Sol/Nem	336.0	64.52	192.02	113.1	[141]	
	Nem/Liq	354.3	3.01	8.50	200.52		
$C_{21}H_{28}N_2O$	4-pentyl-4'-butoxyazobenzene					[141]	
	Sol/Nem	341.0	15.32	44.93			
$C_{21}H_{28}N_2O$	Nem/Liq	360.8	1.08	2.99	47.92	NA	[141]
	4-heptyl-4'-ethoxyazobenzene					[141]	
$C_{21}H_{28}N_2O$	Sol/Nem	320.2	19.11	59.68	113.1	[141]	

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{21}H_{28}N_2O_2$	Nem/Liq	361.3	0.71	1.97	61.65	113.1
	Sol/Nem	342.1	29.9	87.40		[390]
	Nem/Liq	358.9	1.12	3.12	90.52	118.5
Independent value from another reference						
$C_{21}H_{28}O_3$	Sol/Nem	344.2	5.59	16.24		[282]
	Nem/Liq	361.2	0.42	1.16	17.40	
Note: Experimental enthalpies seem abnormally low for this compound.						
$C_{21}H_{28}O_3$	6-decyloxy-2-naphthoic acid					
	Sol/Sol	378.0	0.23	0.61		
	Sol/Smec	409.5	12.94	31.60		
	Smec/Nem	413.7	0.95	2.30		
$C_{21}H_{29}F_3$	Nem/Liq	449.1	2.53	5.63	40.14	116.8
	1-( <i>trans</i> 4'-propylbicyclohexyl)-2,3,4-trifluorobenzene					
	Sol/Nem	310.6	24.41	78.59		101.6
$C_{21}H_{29}F_3$	Nem/Liq	390.0	0.45	1.15	79.74	[27]
	1-( <i>trans</i> 4'-propylbicyclohexyl)-3,4,5-trifluorobenzene					
	Sol/Nem	307.6	21.35	69.41		101.6
$C_{21}H_{29}NS$	Nem/Liq	368.4	0.53	1.44	70.85	[27]
	1-hexyl-4-(4-isothiocyanatophenyl)-bicyclo[2.2.2]octane					
	Sol/Nem	323.7	13.5	41.70		NA
$C_{21}H_{29}NS$	Nem/Liq	362.2	Not reported in paper			[357]
	1-butyl-4-[2-(4-isothiocyanatophenyl)ethyl]bicyclo[2.2.2]octane					
	Sol/Nem	337.2	10.5	31.14		NA
$C_{21}H_{30}F_2$	Nem/Liq	378.7	Not reported in paper			[357]
	1-( <i>trans</i> 4'-propylbicyclohexyl)-3,4-difluorobenzene					
	Sol/Nem	317.9	27.54	86.63		99.9
$C_{21}H_{30}O_3$	Nem/Liq	391.7	0.31	0.79	87.42	[27]
	4-n-dodecyloxycinnamic acid					
	Sol/Smec	400.2	40.6	101.45		
$C_{21}H_{30}O_3$	Smec/Nem	429.2	5.3	12.35		
	Nem/Liq	437.2	13.3	30.42	144.22	164
	4-pentyloxyphenyl 4'-pentybenzoate					
$C_{21}H_{30}O_3$	Sol/Nem	315.2	19.58	62.12		
	Nem/Liq	325.0	0.61	1.88	64.00	131.5
$C_{21}H_{31}NO_2S$	5-decyl-2-(4-isothiocyanatophenyl)-1,3-dioxane					
	Sol/Sol	317.7	7.37	23.20		
	Sol/Smec	336.1	29.4	87.47		
$C_{21}H_{31}NS$	Smec/Liq	354.8	3.79	10.68	121.35	NA
	4-( <i>trans</i> -4'-octylcyclohexyl)isothiocyanatobenzene					
	Sol/Nem	301.0	35.3	117.28		NA
$C_{21}H_{31}NS$	Nem/Liq	318.6	1.10	3.45	120.73	[151]
	3-(4-heptylphenyl)-6-hexyloxy-1,2,4,5-tetrazine					
	Sol/Nem	321.1	30.12	93.80		
$C_{21}H_{32}N_4O$	Nem/Liq	325.8	1.85	5.68	99.48	159.1
	1-methoxy-4-[ <i>(E</i> )-3-( <i>trans</i> -4-pentylcyclohexyl)allyloxy]benzene					
	Sol/Nem	317.2	27.5	86.70		[97]
$C_{21}H_{32}O_2$	Nem/Liq	319.2	Not reported in paper			[198]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)						
$C_{21}H_{33}F_2N$	Sol/Sol	310.2	3.47	11.19				
	Note: Sol/Sol transition was observed only in the first heating cycle.							
	Sol/Nem	348.5	17.9	51.35				[325]
$C_{21}H_{33}NO_5$	Nem/Liq	352.6	1.8	5.10	67.64			
	4-tetradecyloxy-3-nitrobenzoic acid							
	Sol/Smec	369.2	45.9	124.32				[1]
$C_{21}H_{33}N_3O_3$	Smec/Liq	370.2	2.7	7.29	131.61		190.8	
	N,N',N"-tributanoyl-2,4,6-trimethyl-1,3,5-benzenetriamine							
	Sol/Meso	653.2	4.0	6.12				[190]
$C_{21}H_{34}O_3$	Meso/Liq	683.2	35.0	51.23	57.35		101.4	
	4-tetradecyloxybenzoic acid							
	Smec/Smec	369.2	41.42	112.19				[156]
$C_{21}H_{42}O_6S$	Smec/Liq	408.2	8.37	20.50	132.69		174	
	6-O-(propylene-[3'-S-dodecyl])- $\alpha$ -D-galactopyranose							
	Sol/Smec	372.2	59.08	158.73				[39]
$C_{21}H_{44}O_6$	Smec/Liq	438.2	0.80	1.83	160.56		192.2	
	6-O-pentadecyl-D-galacitol							
	Smec/Smec	416.2	68.03	163.46				[183]
$C_{22}H_{14}FNO_5S$	Smec/Liq	445.2	0.44	0.99	164.45		243.1	
	4-methoxyphenyl 3-fluoro-4-thiocyanatophenyl terephthalate							
	Sol/Nem	422.2	38.5	91.19				[37]
$C_{22}H_{15}F_7O$	Nem/Liq	438.2	0.1	0.23	91.42		NA	
	4-[difluoro(3,4,5-trifluorophenoxy)methyl]-3,5-difluoro-4'-propyl-[1,1'-biphenyl]							
	Sol/Nem	315.5	31.62	100.22				[437]
$C_{22}H_{15}NO_5S$	Nem/Liq	Not reported in paper						
	4-thiocyanophenyl 4-(4-methoxybenzoyloxy)benzoate							
	Sol/Nem	422.2	41.3	97.82				[114]
$C_{22}H_{16}N_2O_3$	Nem/Liq	477.2	0.4	0.84	98.66		NA	
	4-methoxyphenyl 4-[4-(4-cyanophenyliminomethyl)]benzoate							
	Sol/Nem	477.2	36.0	75.44				[203]
$C_{22}H_{16}N_2O_3$	Nem/Liq	549.2	0.6	1.09	76.53			
	N-[4-(4-methoxybenzoyloxy)benzylidene]-4-cyanoaniline							
	Sol/Nem	453.2	38.0	83.85				[203]
$C_{22}H_{17}F_3$	Nem/Liq	597.2	0.8	1.34	85.19		93.7	
	4-[4-penten-1-yl]phenyl-1,3-butadiynyl]-4-(trifluoromethyl)benzene							
	Sol/Nem	403.8	17.43	43.16				[212]
$C_{22}H_{17}F_{13}OS$	Nem/Liq	413.6	0.92	2.22	45.38		74.3	
	4-propoxyphenylbenzyl perfluorohexyl thioether							
	Sol/Sol	341.4	7.5	21.97				
	Smec/Smec	361.3	2.4	6.64				
$C_{22}H_{17}F_{17}O_7$	Smec/Smec	373.6	2.1	5.62				
	Smec/Liq	433.3	10.8	24.92	59.15		107.4	
	4-CF <sub>3</sub> CF <sub>2</sub> CF <sub>2</sub> OCF(CF <sub>3</sub> )CF <sub>2</sub> OCF(CF <sub>3</sub> )CH <sub>2</sub> OO(CH <sub>2</sub> ) <sub>5</sub> O-C <sub>6</sub> H <sub>4</sub> COOH							
$C_{22}H_{17}F_{17}O_7$	Sol/Smec	268.2	39.1	145.79				[3]
	Smec/Liq	381.2	10.4	27.28	173.07			
$C_{22}H_{17}F_{17}O_7$	4-[5-[1H,1H-2,5-di(fluoromethyl)-3,6-dioxaundecafluorononyloxy]carbonyl]pentyloxy]benzoic acid							

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.	
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$			
$C_{22}H_{17}N$	Sol/Smec	268.2	19.6	73.08	87.51	[221]	
	Smec/Liq	381.2	5.5	14.43			
$C_{22}H_{17}N$		4-[(4-penten-1-yl)phenyl-1,3-butadiynyl]-4-cyanobenzene					
	Sol/Nem	448.0	30.88	68.93			
	Nem/Liq	501.0		Decomposed prior to transition		[212]	
$C_{22}H_{17}N$		2-(4-cyanophenyl)-7-ethylfluorene					
	Sol/Nem	474.5	11.2	23.60			
	Nem/Liq	518.2	0.46	0.89	24.49	75.9	
$C_{22}H_{18}ClNO_3$		2-methyl-3-chloro-N-[4-(4-methoxybenzoyloxy)benzylidene]aniline					
	Sol/Nem	412.2	31.0	75.21			
	Nem/Liq	422.2	0.4	0.95	76.16	92.8	
$C_{22}H_{18}ClNO_3$		3-chloro-4-methyl-N-[4-(4-methoxybenzoyloxy)benzylidene]aniline					
	Sol/Nem	432.2	44.0	101.80			
	Nem/Liq	464.2	0.4	0.86	102.66	92.8	
$C_{22}H_{18}O_4$		4,4'-dipropoxyoxydiphenyldiacetylene					
	Sol/Sol	351.0	0.59	1.68			
	Sol/Sol	359.0	7.53	20.97			
	Sol/Nem	430.0	19.40	45.12			
$C_{22}H_{19}F$		Nem/Liq	470.0	1.38	2.94	70.71	82.8
		4-[(4-hexen-1-yl)phenyl-1,3-butadiynyl]-4-fluorobenzene					
	Sol/Nem	355.9	24.28	68.22			
	Nem/Liq	423.2	0.77	1.82	70.04	82	
$C_{22}H_{19}FO$		1-(6-butoxy-2-naphthyl)-2-(4-fluorophenyl)acetylene					
	Sol/Nem	383.9	27.8	72.41			
	Nem/Liq	407.2	0.7	1.72	74.13	83.6	
$C_{22}H_{19}F_{15}O_5$		4-[5-[(2-perfluoro-5-methylhexyl)ethoxycarbonyl]pentyloxy]benzoic acid					
	Sol/Smec	360.2	17.0	47.20			
	Smec/Liq	415.2	11.4	27.46	74.66	[221]	
$C_{22}H_{19}N$		4-[4-pentylphenyl-1,3-butadiynyl]-4-cyanobenzene					
	Sol/Nem	422.2	33.11	78.42			
	Nem/Liq	435.1	0.38	0.87	79.29	79.6	
$C_{22}H_{19}NO$		4-propoxy-4"-cyano-p-terphenyl					
	Sol/Smec	463.2	2.82	6.09			
	Smec/Nem	480.2	19.11	39.80			
	Nem/Liq	555.2		Decomposed		[277]	
$C_{22}H_{19}NO_3$		4-methyl-N-[4-(4-methoxybenzoyloxy)benzylidene]aniline					
	Sol/Nem	402.2	28.0	69.62			
	Nem/Liq	551.2	1.1	2.00	71.62	106.5	
$C_{22}H_{19}NO_3$		4-methoxyphenyl 4-[4-(4-methylphenyliminomethyl)]benzoate					
	Sol/Nem	422.2	36.0	85.27			
	Nem/Liq	548.2	0.9	1.64	86.91	106.5	
$C_{22}H_{19}NO_3$		phenyl 4-(4-ethoxybenzylideneamino)benzoate					
	Sol/Nem	436.2	46.0	105.46			
	Nem/Liq	457.2	0.4	0.87	106.33	[292]	
$C_{22}H_{19}NO_3$		4-[(4-ethoxyphenyl)methylene]amino]phenol, benzoate (ester)					
	Sol/Nem	416.2	33.0	79.29			
	Nem/Liq	461.2	0.8	1.73	81.02	[346]	

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{22}H_{19}NO_3$	Sol/Nem	420.2	33.0	78.53		[346]
	Nem/Liq	458.2	0.6	1.31	79.84	
$C_{22}H_{19}NO_3$	Sol/Nem	416.2	33.0	79.29		[346]
	Nem/Liq	456.2	1.0	2.19	81.48	
$C_{22}H_{19}NO_3$	Sol/Nem	433.2	37.0	85.41		[346,347]
	Nem/Liq	466.2	0.5	1.07	86.48	
$C_{22}H_{19}NO_3$	Sol/Nem	419.2	37.0	88.26		[346]
	Nem/Liq	463.2	0.7	1.51	89.77	
$C_{22}H_{19}NO_3$	Sol/Nem	434.2	36.0	82.91		[346]
	Nem/Liq	455.2	0.5	1.10	84.01	
$C_{22}H_{19}NO_3$	Sol/Nem	434.2	38.0	87.52		[346]
	Nem/Liq	455.2	0.5	1.10	88.62	
$C_{22}H_{19}NO_3$	Sol/Nem	418.2	35.0	83.69		[346]
	Nem/Liq	440.2	0.6	1.36	85.05	
$C_{22}H_{19}N_3O$	Sol/Smec	363.2	19.12	52.64		[279]
	Smec/Nem	418.2	0.65	1.55		
	Nem/Liq	438.2	0.27	0.62	54.81	
$C_{22}H_{20}$	Sol/Nem	387.2	27.90	72.09		[212]
	Nem/Liq	454.0	1.28	2.82	74.91	
$C_{22}H_{20}$	Sol/Nem	377.6	19.37	51.30		[212]
	Nem/Liq	447.3	0.85	1.90	53.20	
$C_{22}H_{20}N_2O_2$	Sol/Nem	498.3	45.4	91.11		[365]
	Nem/Liq	Not reported in paper				
$C_{22}H_{20}N_2O_3$	Sol/Smec	409.2	31.0	75.76		[291]
	Smec/Nem	416.2	2.3	5.53		
	Nem/Liq	449.2	0.3	0.67	81.96	
$C_{22}H_{20}N_2O_3$	Sol/Nem	393.2	36.0	91.56		[291]
	Nem/Liq	458.2	0.6	1.31	92.87	
$C_{22}H_{20}N_2O_3$	Sol/Nem	395.2	39.0	98.68		[291]
	Nem/Liq	447.2	0.4	0.89	99.57	
$C_{22}H_{20}N_2O_3$	Sol/Nem	391.2	39.0	99.69		[291]
	Nem/Liq	448.2	0.7	1.56	101.25	

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.	
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$			
$C_{22}H_{21}F$	Sol/Nem	349.2	31.10	89.06	85.6	[212]	
	Nem/Liq	353.3	0.21	0.59			
$C_{22}H_{21}F$	Sol/Nem	349.2	31.06	88.95	85.6	[270]	
	Nem/Liq	353.3	0.21	0.59			
$C_{22}H_{21}F_9OS$	4-pentyloxyphenylbenzyl perfluorobutyl thioether						
	Sol/Sol	320.3	7.3	22.79	58.53	[66]	
	Sol/Smec	350.2	4.5	12.85			
$C_{22}H_{21}NS$	Smec/Liq	391.7	7.4	18.89			
	2-pentyl-5-(4'-cyanobiphenyl-4-yl)thiophene					[20]	
	Sol/Nem	423.1	12.0	28.36			
$C_{22}H_{22}$	Nem/Liq	470.3	0.66	1.40	29.76	[212]	
	4-[4-butylphenyl-1,3-butadiynyl]-4-ethylbenzene						
	Sol/Nem	317.6	22.34	70.34	77.4	[212]	
$C_{22}H_{22}$	Nem/Liq	371.6	0.60	1.61	71.95		
	4-[4-pentylphenyl-1,3-butadiynyl]-4-methylbenzene					[212]	
	Sol/Nem	360.9	22.06	61.12	77.4		
$C_{22}H_{22}N_2O_2$	Nem/Liq	381.6	1.01	2.65	63.77		
	N-hexyl-4-[4-(4-nitrophenyl)-1,3-butadiinyl]benzenamine					[216]	
	Sol/Nem	403.2	39.8	98.71	99.49		
$C_{22}H_{23}F_{13}S$	Nem/Liq	422.4	0.33	0.78	[66]		
	4-propylcyclohexylbenzyl perfluorohexyl thioether						
	Smec/Smec	326.7	22.3	68.26	108.2	[66]	
$C_{22}H_{23}NO$	Smec/Liq	345.0	13.2	38.26	106.52		
	2-(4-pentylphenyl)-5-phenylpyridine					[260]	
	Sol/Meso	339.8	2.17	6.39	24.13		
$C_{22}H_{23}NO$	Meso/Liq	385.0	6.83	17.74	24.13	[260]	
Note: Experimental enthalpies seem abnormally low for this compound.							
$C_{22}H_{23}NOS$	4'-heptyloxy-4-isothiocyanatotolane					[135]	
	Sol/Sol	333.5	29.54	88.58	NA	[135]	
	Smec/Smec	384.8	1.55	4.03			
$C_{22}H_{23}NO_2$	Smec/Liq	386.1	12.01	31.11	123.72		
	1-[2-( <i>trans</i> -4-ethylcyclohexyl)]-4-[(4-nitrophenyl)ethynyl]benzene					[238]	
	Sol/Nem	437.1	25.65	58.68	79		
$C_{22}H_{23}NO_2$	Nem/Liq	483.3	0.59	1.22	59.90		
	(E)-4-[2-[4-(5-vinyloxypentyl)oxy]phenyl]vinylbenzonitrile					[435]	
	Smec/Smec	343.6	7.1	20.66	74.75		
$C_{22}H_{23}NO_2S$	Smec/Nem	347.2	8.5	24.48	[435]		
	Nem/Liq	347.9	10.3	29.61			
	4-isothiocyanatophenyl 4-( <i>trans</i> -4-ethylcyclohexyl)benzoate						
$C_{22}H_{23}NO_2S$	Sol/Nem	416.2	32.22	77.41	NA	[356]	
	Nem/Liq	493.2	0.79	1.60	79.01		
$C_{22}H_{24}FNO_2S$	S-(2-fluoro-4-octyloxyphenyl) 4-cyanothiobenzoate						
	Sol/Nem	351.2	40.4	115.03	NA	[4]	
$C_{22}H_{24}O_3$	Nem/Liq	352.2	0.9	2.56	117.59		
	6-(8-nonyloxy)-2-naphthoic acid						
$C_{22}H_{24}O_3$	Sol/Sol	307.2	5.86	19.08	NA	[4]	

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{22}H_{24}O_4$	Sol/Sol	375.6	57.32	152.61		
	Sol/Nem	412.3	54.39	131.92		
	Nem/Liq	454.4	9.20	20.25	323.86	107.6 [68]
$C_{22}H_{24}O_6$	di(4'-methylphenyl)- <i>trans</i> -cyclohexane-1,4-dicarboxylate					
	Sol/Nem	450.4	35.5	78.82		
	Nem/Liq	466.3	1.07	2.29	81.11	90.7 [215]
$C_{22}H_{24}O_6$	di(4'-methoxyphenyl)- <i>trans</i> -cyclohexane-1,4-dicarboxylate					
	Sol/Sol	411.6	4.81	11.69		
	Sol/Nem	416.2	31.48	75.64		
$C_{22}H_{24}O_6$	Nem/Liq	516.0	2.87	5.56	92.89	104.3 [157, 405]
Independent values from another reference						
Sol/Nem	416.2	36.65	88.06			
Note: The reported enthalpy might include the Sol/Sol phase transition that is indicated in the first set of values.						
$C_{22}H_{25}NO_2$	Nem/Liq	516.2	2.87	5.56	93.62	104.3 [220]
	(E)-4-[2-[4-(7-hydroxyheptyloxy)phenyl]vinyl]benzonitrile					
	Sol/Nem	356.6	15.6	43.75		
$C_{22}H_{25}N_3O$	Nem/Liq	364.5	9.6	26.34	70.09	[435]
	4-[5-[4-( <i>trans</i> -4-propylcyclohexyl)phenyl]-1,2,4-oxadiazol-3-yl]pyridine					
	Sol/Nem	367.2	23.52	64.05		
$C_{22}H_{25}N_3O$	Nem/Liq	430.2	0.21	0.49	64.54	110.8 [279]
	3-[5-[4-( <i>trans</i> -4-propylcyclohexyl)phenyl]-1,2,4-oxadiazol-3-yl]pyridine					
	Sol/Sol	345.2	10.11	29.29		
$C_{22}H_{25}N_3O$	Sol/Smec	346.2	4.10	11.84		
	Smec/Nem	349.2	0.03	0.09		
	Nem/Liq	463.2	0.35	0.76	41.98	110.8 [279]
$C_{22}H_{25}N_3O$	2-[5-[4-( <i>trans</i> -4-propylcyclohexyl)phenyl]-1,2,4-oxadiazol-3-yl]pyridine					
	Sol/Sol	393.2	20.64	52.49		
	Sol/Nem	400.2	2.60	6.50		
$C_{22}H_{26}N_2O$	Nem/Liq	427.2	0.17	0.40	59.39	110.8 [279]
	4-octyloxy-N-(4-cyanobenzylidene)aniline					
	Sol/Sol	339.9	11.28	33.19		
$C_{22}H_{26}N_2O$	Smec/Nem	347.0	29.70	85.6		
	Nem/Liq	356.2	To small to be measured			
	Nem/Liq	382.1	0.87	2.27	121.06	121.1 [204]
$C_{22}H_{26}N_2O$	N-(4-octyloxybenzylidene)-4-aminobenzonitrile					
	Sol/Nem	353.1	37.6	106.49		
	Nem/Liq	366.6	0.83	2.26	108.75	[365]
$C_{22}H_{26}N_2O_2S$	1-[2-(5-cyanothienyl)]-3-(4-octyloxyphenylamino)-2-propen-1-one					
	Smec/Smec	391.8	14.69	37.49		
	Smec/Liq	456.6	5.85	12.81	50.30	[75]
$C_{22}H_{26}N_2O_2S_2$	2,5-bis(4-butoxyphenyl)thiazolo[5,4-d]dithiazole					
	Sol/Sol	455.8	1.75	3.84		
	Smec/Smec	470.9	19.15	40.67		
$C_{22}H_{26}N_2O_2S_2$	Smec/Nem	483.8	3.45	7.13		
	Nem/Liq	567.4	1.62	2.86	54.50	[269]
	4-pentanoyl-4'-pentanoyloxyazobenzene					
$C_{22}H_{26}N_2O_3$	Smec/Smec	355.1	7.10	19.99		
	Smec/Smec	377.1	4.69	12.44		
	Smec/Smec	377.5	7.68	20.34		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{22}H_{26}N_2O_3$	Smec/Liq	399.5	6.09	15.24	68.01	115.7
	Sol/Smec	365.2	24.81	67.94		[157]
	Smec/Nem	414.7	5.31	12.80		
	Nem/Liq	416.7	0.92	2.21	82.95	115.7
$C_{22}H_{26}N_4OS$	4-(4-propanoyl)-4'-heptanoyloxyazobenzene					
	Sol/Smec	468.6	29.8	63.59		[31]
	Smec/Liq	480.6	4.4	9.16	72.75	
$C_{22}H_{26}N_4OS$	5-(4-pyridyl)-2-(4-n-octyl)phenylamido-1,3,4-thiadiazole					
	Sol/Smec	420.7	21.9	52.06		[79]
	Smec/Liq	434.0	2.5	5.76	57.82	
$C_{22}H_{26}N_4O_2S$	5-(4-pyridyl)-2-(4-octyloxy)benzylideneamino-1,3,4-thiadiazole					
	Sol/Smec	484.4	14.1	29.11		[31]
	Smec/Liq	523.5	5.3	10.12	39.23	
$C_{22}H_{27}ClO_2S$	4-chlorophenyl 4-nonyloxythiobenzoate					
	Sol/Smec	343.5	34.69	100.99		
	Smec/Liq	370.2	5.23	14.13	115.12	NA
$C_{22}H_{27}N$	4'-nonyl-4-cyanobiphenyl					
	Sol/Smec	313.7	33.48	106.73		
	Smec/Nem	320.8	Value is very small			
$C_{22}H_{27}N$	Nem/Liq	322.7	1.20	3.72	110.45	119.2
	Independent values from another reference					
	Sol/Smec	313.4	37.57	119.88		
	Smec/Nem	320.4	0.29	0.91		
$C_{22}H_{27}N$	Nem/Liq	322.1	1.44	4.47	125.26	119.2
	Independent values from another reference					
	Sol/Smec	305.0	30.96	101.51		
	Smec/Nem	321.0	0.25	0.78		
$C_{22}H_{27}N$	Nem/Liq	323.0	1.00	3.10	105.39	119.2
	Independent values from another reference					
	Sol/Smec	314.9	34.7	110.19		
	Smec/Nem	320.2	0.5	1.56		
$C_{22}H_{27}N$	Nem/Liq	322.2	1.63	5.06	116.81	119.2
	Independent values from another reference					
	Sol/Smec	315.7	34.5	109.28		
	Smec/Nem	320.8	0.01	0.03		
$C_{22}H_{27}NO$	Nem/Liq	322.8	1.20	3.72	113.03	119.2
	Independent values from another reference					
	Sol/Smec	335.7	35.48	105.69		
	Smec/Nem	353.2	0.38	1.08		
$C_{22}H_{27}NO$	Nem/Liq	353.3	0.92	2.60	109.37	126
	Independent values from another reference					
	Sol/Smec	Not reported in paper				
	Smec/Nem	351.4	0.16	0.46		
$C_{22}H_{27}NO$	Nem/Liq	353.2	0.84	2.38	126	[232]
	Independent values from another reference					
	Sol/Smec	456.2	16.0	35.07		
	Smec/Nem	531.2	Not reported in paper			
$C_{22}H_{27}NO_3$	Nem/Liq	534.2	19.7	36.88		[416]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta S_{\text{pce}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.	
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )					
$C_{22}H_{27}NO_5$	Sol/Nem	4-nonyloxyphenyl 4-nitrobenzoate	340.2	43.51	127.90	133.7	[236]	
	Nem/Liq		350.2	4.06	11.59	139.49		
$C_{22}H_{27}NS$	Sol/Meso	4-nonyl-4'-thiocyanatobiphenyl	323.4	31.55	97.56	NA	[84]	
	Meso/Liq		340.4	10.28	30.20	127.76		
$C_{22}H_{28}BrNOS$	Sol/SmeC	1-[2-(5-nonylthienyl)]-3-(4-bromophenylamino)-2-propen-1-one	362.5	28.64	79.01	129.7	[75]	
	SmeC/SmeC		363.8	0.82	2.25			
	SmeC/Liq		440.6	7.77	17.64	98.90		
$C_{22}H_{28}ClN$	Sol/SmeC	4-chlorobenzylidene 4'-nonylaniline	320.2	27.70	86.51	106.58	[31]	
	SmeC/SmeC		329.7	2.68	8.13			
	SmeC/Liq		331.6	3.96	11.94			
$C_{22}H_{28}ClNOS$	Sol/SmeC	1-[2-(5-nonylthienyl)]-3-(4-chlorophenylamino)-2-propen-1-one	358.2	29.52	82.41	128.4	[75]	
	SmeC/Liq		431.6	6.93	16.06	98.47		
$C_{22}H_{28}FNOS$	Sol/SmeC	1-[2-(5-nonylthienyl)]-3-(4-fluorophenylamino)-2-propen-1-one	343.8	28.61	83.22	128.8	[75]	
	SmeC/Liq		374.9	4.10	10.94	94.16		
	SmeC/Liq							
$C_{22}H_{28}INOS$	Sol/SmeC	1-[2-(5-nonylthienyl)]-3-(4-iodophenylamino)-2-propen-1-one	374.8	12.89	34.39	56.87	[75]	
	SmeC/SmeC		387.1	1.73	4.47			
	SmeC/Liq		440.8	7.94	18.01	131.6		
$C_{22}H_{28}N_2O$	Sol/Nem	4-( $\omega$ -aminononyloxy)-4'-cyanobiphenyl	349.1	33.5	95.96	98.29	[17]	
	Nem/Liq		356.4	0.83	2.33	136.9		
$C_{22}H_{28}N_2O_2$	Sol/SmeC	4-nitrobenzylidene 4'-nonylaniline	319.2	7.53	23.59	61.05	[235]	
	SmeC/Nem		323.2	0.42	1.30			
	Nem/Liq		325.2	11.76	36.16			
$C_{22}H_{28}N_2O_3$	Sol/Nem	4-hexanoyloxy-2,3-dimethyl-4'-ethoxyazobenzene	349.2	32	91.64	92.74	[339]	
	Nem/Liq		362.2	0.4	1.10			
$C_{22}H_{28}N_2O_3$	Sol/Nem	4-heptanoyloxy-2-methyl-4'-ethoxyazobenzene	324.2	38	117.21	119.86	[339]	
	Nem/Liq		340.2	0.9	2.65			
$C_{22}H_{28}N_2O_3$	Sol/Nem	4-heptanoyloxy-3-methyl-4'-ethoxyazobenzene	335.2	39	116.35	118.67	[339]	
	Nem/Liq		345.2	0.8	2.32			
$C_{22}H_{28}O_2$	Nem/Liq	4-methoxy-4'-heptoxy- <i>trans</i> -stilbene	421	421	1.6	101.1	127.0	[157]
	Sol/Liq		423	42.76	101.1			
Note: Nem/SmeC transition observed on cooling.								
$C_{22}H_{28}O_2S$	Sol/Nem	4-pentylphenyl 4'-butyloxythiobenzoate	336.1	23.43	69.71	71.78	NA	[409]
	Nem/Liq		362.9	0.75	2.07			

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.		
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$				
$C_{22}H_{28}O_3$	Sol/Nem	321.6	26.07	81.06	124.4	[218]		
	Nem/Liq	330.9	0.73	2.21				
$C_{22}H_{29}ClN_2O_2$	Sol/SmeC	333.7	28.4	85.11	133.4	[229]		
	SmeC/Liq	372.1	4.4	11.82				
$C_{22}H_{29}NO$	Sol/SmeC	326.2	22.91	70.23	93.30	[242]		
	SmeC/SmeC	331.8	4.04	12.18				
	SmeC/Nem	337.8	2.77	8.20				
	Nem/Liq	342.2	0.92	2.69				
$C_{22}H_{29}NO$	Sol/Nem	322.5	30.9	95.81	118.9	[376, 381]		
	Nem/Liq	343.8	0.46	1.34				
	Note: Temperature calculated from published enthalpy and entropy data in paper.							
$C_{22}H_{29}NO$	Sol/Nem	325.7	26.3	80.75	118.9	[376, 381]		
	Nem/Liq	365.4	0.79	2.16				
	Note: Temperature calculated from published enthalpy and entropy data in paper.							
$C_{22}H_{29}NO$	Sol/Nem	313.9	27.2	86.65	118.9	[376, 381]		
	Nem/Liq	323.5	0.42	1.30				
	Note: Temperature calculated from published enthalpy and entropy data in paper.							
$C_{22}H_{29}NO$	Sol/SmeC	301.2	4.2	13.94	118.9	[376, 318]		
	SmeC/Nem	Error in published data						
	Nem/Liq	363.6	1.17	3.22	118.9			
Note: Temperature calculated from published enthalpy and entropy data in paper.								
$C_{22}H_{29}NO$	Sol/SmeC	299.7	22.68	75.68	118.9	[157]		
	SmeC/Nem	325.7	7.11	21.83				
	Nem/Liq	342.5	1.78	5.20	102.71			
Independent values from another reference								
	Sol/SmeC	285.2	8.4	29.45	118.9	[376, 381]		
	SmeC/SmeC	325.6	5.86	18.00				
	SmeC/Nem	Peak was not resolved						
$C_{22}H_{29}NO_2S$	Nem/Liq	343.3	0.96	2.80	118.9	[376, 381]		
	Note: Temperatures calculated from published enthalpy and entropy data in paper.							
	4-isothiocyanatophenyl 4-hexylbicyclo[2.2.2]octane-1-carboxylate							
$C_{22}H_{29}NO_3$	Sol/Sol	312.1	7.1	22.75	NA	[357]		
	Sol/Nem	324.7	19.2	59.13				
	Nem/Liq	379.2	Not reported in paper					
$C_{22}H_{29}NO_3$	Sol/Nem	385.2	49.37	128.17	NA	[162]		
	Nem/Liq	401.2	1.12	2.79				
	130.96							
$C_{22}H_{30}$	Sol/SmeC	298.3	7.10	23.80	77.2	[90]		
	SmeC/SmeC	319.3	0.25	0.78				
	SmeC/SmeC	320.3	2.03	6.34				
	Nem/Liq	325.5	9.56	29.37	60.29			

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.			
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$					
$\text{C}_{22}\text{H}_{30}\text{F}_3$	Sol/Nem	314.8	25.05	79.57	101.6	[27]			
	Nem/Liq	391.5	0.31	0.79					
$\text{C}_{22}\text{H}_{30}\text{N}_2\text{O}$	4-propyl-4'-heptyloxyazobenzene				120.2	[153]			
	Sol/Nem	337.0	90.71	269.17					
$\text{C}_{22}\text{H}_{30}\text{N}_2\text{O}$	Nem/Liq	349.6	2.70	7.72	276.89	Note: Sol/Nem transition enthalpy is out of line with other derivatives in this series.			
	4-butyl-4'-hexyloxyazobenzene								
$\text{C}_{22}\text{H}_{30}\text{N}_2\text{O}$	Sol/Nem	316.0	22.79	72.12	120.2	[141]			
	Nem/Liq	346.5	0.89	2.57					
$\text{C}_{22}\text{H}_{30}\text{N}_2\text{O}$	4-pentyl-4'-pentylloxyazobenzene				120.2	[141]			
	Sol/Nem	328.1	13.93	42.46					
$\text{C}_{22}\text{H}_{30}\text{N}_2\text{O}$	Nem/Liq	352.5	1.04	2.95	45.41	[390]			
	4-heptyl-4'-propoxyazobenzene								
$\text{C}_{22}\text{H}_{30}\text{N}_2\text{O}$	Sol/Nem	319.4	9.46	29.62	120.2	[390]			
	Nem/Liq	346.1	0.48	1.39	31.01				
$\text{C}_{22}\text{H}_{30}\text{N}_2\text{O}$	bis(4-pentylphenyl)diazene N-oxide				[365]				
	Sol/Nem	299.8	15.7	52.42					
$\text{C}_{22}\text{H}_{30}\text{N}_2\text{O}_2$	Nem/Liq	341.1	0.95	2.79	55.21				
	4-methyl-2'-hydroxy-4'-nonyloxyazobenzene								
$\text{C}_{22}\text{H}_{30}\text{N}_2\text{O}_2$	Sol/Nem	339.7	31.9	93.91	125.6	[73]			
	Nem/Liq	355.3	0.83	2.34	96.25				
$\text{C}_{22}\text{H}_{30}\text{N}_2\text{O}_3$	4,4'-dipentyloxyazoxybenzene				135.6	[9]			
	Sol/Sol	342.0	22.0	64.33					
$\text{C}_{22}\text{H}_{30}\text{N}_2\text{O}_3$	Sol/Nem	350.2	15.80	45.12					
	Nem/Liq	397.2	1.50	3.78	113.23				
Independent values from another reference									
Sol/Nem	348.7	14.59	41.84						
Note: Three other independent studies report that this compound exhibits a Sol/Sol transition.									
Nem/Liq	396.4	0.72	1.82	43.66	135.6	[179]			
Independent values from another reference									
Sol/Sol	340.7	22.8	66.92						
Sol/Nem	350.2	15.4	43.97						
Nem/Liq	395.2	1.10	2.78	113.67	135.6	[365]			
Independent values from another reference									
Sol/Sol	341.0	23.0	67.45						
Sol/Nem	349.0	13.8	39.54						
Nem/Liq	397.0	0.86	2.17	109.16	135.6	[440]			
$\text{C}_{22}\text{H}_{30}\text{O}_2$	4,4'-bis(2-methylbutoxy)biphenyl				104.6	[172]			
	Sol/Smec	356.8	8.43	23.63					
$\text{C}_{22}\text{H}_{30}\text{O}_2$	Smec/Liq	360.2	36.29	100.75	124.38				
	Sol/Smec	346.4	20.96	60.51					
$\text{C}_{22}\text{H}_{30}\text{O}_2$	Smec/Liq	349.9	51.45	147.04	207.55	104.6			
	Sol/Meso	400.8	24.57	61.30					
$\text{C}_{22}\text{H}_{30}\text{O}_2\text{S}$	Meso/Liq	404.1	8.42	20.84	82.14	138.9			
	2-octanoyl-5-(4-butoxyphenyl)thiophene								
C <sub>22</sub> H <sub>31</sub> NS									
4-( <i>trans, trans</i> -4-propylbicyclohexyl)benzene-isothiocyanate									

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.		
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$				
$C_{22}H_{32}N_6OS$	Sol/Nem	349.7	10.45	29.88	NA	[52]		
	Nem/Liq	521.9	1.46	2.80				
	Sol/Smec	417.9	16.33	39.08				
$C_{22}H_{33}NS$	Smec/Liq	428.6	1.97	4.60	43.68	[384]		
	1-(1-oxodecyl)-4-[4-[(1E)-1,3,4-thiadiazol-2-ylazo]phenyl]piperazine							
	Sol/Nem	312.3	40.0	128.08	NA	[151]		
$C_{22}H_{34}F_2O_2$	Nem/Liq	326.6	1.20	3.67	131.75			
	1-ethoxy-2,3-difluoro-4-[3-( <i>trans</i> -4-pentylcyclohexyl)propoxybenzene							
	Sol/Nem	298.2	25.4	85.18	Not reported in paper	[198]		
$C_{22}H_{34}O$	Nem/Liq	299.2	Not reported in paper					
	1-methoxy-(4-[(E)-3-( <i>trans</i> -4-pentylcyclohexyl)-1-butenyl])-benzene							
	Sol/Nem	298.2	17.5	58.69	Not reported in paper	[198]		
$C_{22}H_{35}NO_5$	Nem/Liq	307.2	Not reported in paper					
	4-pentadecyloxy-3-nitrobenzoic acid							
	Smec/Smec	368.2	50.1	136.07	200.1	[1]		
$C_{22}H_{36}N_2O_2$	Smec/Liq	370.2	1.9	5.13	141.20			
	$N,N'$ -dihexanoyl-3,4,5,6-tetramethylbenzene-1,2-diamine							
	Sol/Meso	387.2	10	25.83	112.0	[61]		
	Meso/Meso	400.2	5	12.49				
$C_{22}H_{36}O_2$	Meso/Meso	480.2	6	12.49				
	Meso/Liq	534.2	24	44.93	95.74	[198]		
	Sol/Nem	323.2	22.6	69.93				
$C_{22}H_{36}O_3$	Nem/Liq	330.2	Not reported in paper					
	4-pentadecyloxybenzoic acid							
	Smec/Smec	375.2	51.88	138.27	183.3	[156]		
$C_{22}H_{42}O_7$	Smec/Liq	407.2	9.62	23.62	161.89			
	6-O-hexadecanoyl- $\alpha$ -D-galactopyranose							
	Smec/Smec	391.2	43.78	111.91	226.1	[39]		
$C_{22}H_{44}O_5S$	Smec/Liq	455.2	0.67	1.47	113.38			
	6-S-hexadecyl-6-thio- $\alpha$ -D-galactopyranose							
	Smec/Smec	376.2	58.89	156.54	229.8	[39]		
$C_{22}H_{44}O_6$	Smec/Liq	454.2	3.45	7.60	164.14			
	6-O-hexadecyl- $\alpha$ -D-galactopyranose							
	Smec/Smec	394.2	51.42	130.44	232.4	[39]		
$C_{22}H_{46}O_4$	Smec/Liq	440.2	0.64	1.45	131.89			
	1,2,21,22-tetrahydroxydocosane							
	Smec/Smec	360.2	20.3	56.36	272.0	[145]		
$C_{22}H_{46}O_6$	Meso/Meso	413.2	17.0	41.14				
	Meso/Liq	416.2	15.2	36.52	134.02			
	Smec/Smec	419.2	69.28	165.27				
$C_{23}H_{13}F_{17}SO$	Smec/Liq	438.2	0.82	1.87	167.14	252.4	[183]	
	2(perfluorooctyl)ethyl 4-phenylthiobenzoate							
	Smec/Smec	355.5	25.3	71.17				
$C_{23}H_{13}F_{17}SO$	Smec/Liq	432.5	9.2	21.27	92.44	NA	[38]	

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated) (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{23}H_{16}FNO_5S$	Sol/Nem	412.2	19.7	47.79	NA	[37]
	Nem/Liq	439.2	0.3	0.68		
$C_{23}H_{16}O_5$	Sol/Nem	459.3	13.23	28.80	29.18	[14]
	Nem/Liq	499.0	0.19	0.38		
$C_{23}H_{16}F_2O$	Sol/Nem	413.6	31.9	77.13	72.6	[95]
	Nem/Liq	419.9	0.1	0.24		
$C_{23}H_{17}ClO_8$	Sol/Nem	459.2	41.4	90.16	91.34	[174]
	Nem/Liq	515.2	0.61	1.18		
$C_{23}H_{17}FO$	Sol/Nem	405.6	25.4	62.62	62.84	70.9
	Nem/Liq	460.8	0.1	0.22		
$C_{23}H_{17}NO_5S$	Sol/Nem	431.2	40.1	93.00	93.62	NA
	Nem/Liq	481.2	0.3	0.62		
$C_{23}H_{18}ClNO_4$	Sol/Smec	494.2	48.0	97.13	102.3	[196]
	Smec/Nem	491.2	11.0	22.39		
	Nem/Liq	539.2	0.4	0.74		
$C_{23}H_{18}N_2O_6$	Sol/Nem	448.2	40.0	89.25	Not reported in paper	[196]
	Nem/Liq	553.2	Not reported in paper			
$C_{23}H_{19}N$	Sol/Nem	459.8	15.77	34.30	93.5	[2]
	Nem/Liq	512.8	0.4	0.78		
$C_{23}H_{19}NO_3S$	Sol/Smec	386.5	16.32	42.23	115.9	[63]
	Smec/Smec	408.9	17.15	41.94		
	Smec/Nem	419.6	2.15	5.12		
	Nem/Liq	451.5	2.51	5.56		
$C_{23}H_{19}NO_5$	Sol/Nem	483.2	55.0	113.82	Not reported in paper	[195]
	Nem/Liq	589.2	Not reported in paper			
$C_{23}H_{19}NO_5$	Sol/Nem	485.2	56.0	115.42	116.81	[195]
	Nem/Liq	574.2	0.8	1.39		
$C_{23}H_{20}F_2O$	Sol/Nem	347.7	18.0	51.77	92.4	[95]
	Nem/Liq	358.6	0.2	0.56		
$C_{23}H_{21}FO$	Sol/Nem	373.6	21.1	56.48	90.7	[95]
	Nem/Liq	396.2	0.4	1.01		
$C_{23}H_{21}NO$	Sol/Smec	394.2	4.91	12.46	25.73	
	Smec/Nem	458.2	11.79	25.73		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{23}H_{21}NO_3$	Nem/Liq	545.2	Decomposed		105.1	[277]
	Sol/Nem	399.2	31.0	77.66		
	Nem/Liq	431.2	0.4	0.93	78.59	92.1 [191]
$C_{23}H_{21}NO_3$	Sol/Nem	415.2	36.0	86.71		
	Nem/Liq	495.2	0.8	1.62	88.33	92.1 [191]
	Sol/Nem	391.2	26.0	66.46		
$C_{23}H_{21}NO_3$	Nem/Liq	455.2	0.5	1.10	67.56	92.1 [191]
$C_{23}H_{21}NO_3$	Sol/Nem	394.2	39.0	98.93		
	Nem/Liq	436.2	0.3	0.69	99.62	[292]
$C_{23}H_{21}NO_3$	Sol/Nem	411.2	31.0	75.39		
	Nem/Liq	528.2	0.9	1.70	77.09	[292]
	Sol/Nem	394.2	39.0	98.93		
$C_{23}H_{21}NO_3$	Nem/Liq	436.2	0.3	0.69	99.62	105.1 [292]
$C_{23}H_{21}NO_3$	Sol/Nem	411.2	31.0	75.39		
	Nem/Liq	528.2	0.9	1.70	77.09	115.6 [292]
$C_{23}H_{21}O_4S$	Sol/Nem	398.2	83.26	209.09		
	Nem/Liq	468.7	4.18	8.92	218.01	[63]
	Sol/Nem	355.5	23.17	65.18		
$C_{23}H_{22}$	Nem/Liq	450.2	1.12	2.49	67.67	82.4 [212]
$C_{23}H_{22}N_2O_3$	Sol/Nem	411.7	40.9	99.34		
	Nem/Liq	433.2	0.4	0.92	100.26	[265, 266]
$C_{23}H_{22}N_2O_3$	Sol/SmeC	391.2	36.0	92.02		
	SmeC/Nem	424.2	1.8	4.24		
	Nem/Liq	452.2	0.5	1.11	97.37	115.9 [291]
$C_{23}H_{22}N_2O_3$	Sol/Nem	384.2	41.0	106.72		
	Nem/Liq	460.2	0.6	1.30	108.02	[291]
	Sol/Nem	374.2	35.0	93.53		
$C_{23}H_{22}N_2O_3$	Nem/Liq	451.2	0.7	1.55	95.08	[291]
$C_{23}H_{22}N_2O_3$	Sol/Nem	390.2	41.0	105.07		
	Nem/Liq	450.2	0.8	1.78	106.85	[291]
$C_{23}H_{23}NS$	2-hexyl-5-(4'-cyanobiphenyl-4-yl)thiophene					
	Sol/Sol	336.6	0.71	2.11		
	Sol/Nem	419.1	10.6	25.29		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{23}H_{23}NS$	Nem/Liq	462.2	0.81	1.75	29.15	115.9
	Sol/Nem	413.4	25.69	62.14		[20]
	Nem/Liq	516.2	0.84	1.63	63.77	NA
$C_{23}H_{24}$	4'- <i>(4</i> -ethylcyclohexyl)-4-isothiocyanatotolane					[135, 238]
	Sol/Nem	317.9	17.85	56.15		[212]
	Nem/Liq	374.9	Not reported in paper			
$C_{23}H_{25}NO_2$	6-n-pentyloxy-2-[4-methoxystyryl]quinoline					
	Sol/Sol	354.6	1.19	3.36		[112]
	Sol/Nem	401.6	30.65	76.32		
$C_{23}H_{25}NO_2$	Nem/Liq	455.4	0.56	1.23	80.91	116.1
	1-[2-( <i>trans</i> -4-propylcyclohexyl)]-4-[ <i>(4</i> -nitrophenyl)ethynyl]-benzene					
	Sol/Nem	437.4	28.07	64.17		[238]
$C_{23}H_{26}N_2O$	Nem/Liq	504.9	0.92	1.82	65.99	
	3-phenyl-5-[4-( <i>trans</i> -4-propylcyclohexyl)phenyl]-1,2,4-oxadiazole					
	Sol/Nem	361.2	20.23	56.01		[279]
$C_{23}H_{26}O_3$	Nem/Liq	439.2	0.31	0.71	56.72	107.3
	6-(9-decenoxy)-2-naphthoic acid					
	Sol/SmeC	402.7	50.21	124.68		[68]
$C_{23}H_{28}N_2O_2S$	SmeC/Nem	407.1	3.35	8.22		
	Nem/Liq	447.4	8.79	19.65	152.55	114.7
	1-[2-(5-cyanothienyl)]-3-(4-nonyloxyphenylamino)-2-propen-1-one					
$C_{23}H_{28}N_2O_2S$	Sol/SmeC	395.2	14.42	36.49		[75]
	SmeC/Liq	459.0	6.59	14.36	50.85	134.6
$C_{23}H_{28}N_2O_3$	4-pentanoyl-4'-hexanoyloxyazobenzene					
	Sol/Sol	323.8	4.67	14.42		
	Sol/SmeC	365.1	7.12	19.50		
	SmeC/SmeC	375.2	3.36	8.96		
	SmeC/SmeC	377.5	9.57	25.35		
$C_{23}H_{28}N_2O_3$	SmeC/Liq	401.2	5.81	14.48	82.71	122.8
	4-propanoyl-4'-octanoyloxyazobenzene					
	Sol/SmeC	369.7	27.49	101.63		[157]
$C_{23}H_{28}N_4OS$	SmeC/Meso	416.2	5.31	12.76		
	Meso/Liq	416.7	1.00	2.40	116.79	122.8
	5-(4-pyridyl)-2-(4-nonyloxy)benzylideneamino-1,3,4-thiadiazole					
$C_{23}H_{28}N_4OS$	Sol/SmeC	415.1	23.6	56.85		[79]
	SmeC/Liq	436.5	3.7	8.48	65.33	
$C_{23}H_{28}N_4O_2$	1-[4-[ <i>(1E</i> )-(4-formylphenyl)azo]phenyl]-4-(1-oxohexyl)piperazine					
	Sol/Nem	417.5	19.9	47.66		[345]
	Nem/Liq	450.6	0.2	0.44	48.10	
$C_{23}H_{29}ClO_2S$	4-chlorophenyl 4-decyloxythiobenzoate					
	Sol/SmeC	341.2	30.21	88.54		[383]
$C_{23}H_{29}N$	SmeC/Liq	366.3	4.85	13.24	101.78	NA
	4-decyl-4'-cyanobiphenyl					
$C_{23}H_{29}N$	Sol/SmeC	320.9	37.87	118.01		[233]
	SmeC/Liq	327.8	3.05	9.30	127.31	126.3
Independent values from another reference						
Sol/SmeC	317.0	33.47	105.58			

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)						
	Smec/Liq	322.0	2.68		8.32	113.90	126.3	[237]
Independent values from another reference								
Sol/Smec		316.9	37.2		117.39			
Smec/Liq		323.6	3.05		9.43	126.82	126.3	[350]
Independent values from another reference								
Sol/Smec		317.7	36.0		113.31			
Smec/Liq		324.5	2.83		8.72	122.02	126.3	[387]
C <sub>23</sub> H <sub>29</sub> NO		4-decyloxy-4'-cyanobiphenyl						
Sol/Smec		330.8	36.07		109.04			
Smec/Liq		357.2	3.31		9.27	118.31	133.1	[233]
C <sub>23</sub> H <sub>29</sub> NO <sub>3</sub>		4-(4-nonyloxyphenyliminomethyl)benzoic acid						
Sol/Smec		453.2	16.2		35.75			
Smec/Liq		533.2	19.6		36.76	72.51		[416]
C <sub>23</sub> H <sub>29</sub> NO <sub>5</sub>		4-decyloxyphenyl 4-nitrobenzoate						
Sol/Smec		322.2	5.90		18.31			
Smec/Nem		326.2	16.44		50.40			
Nem/Liq		347.2	4.39		12.64	81.35		[236]
C <sub>23</sub> H <sub>29</sub> NS		4-decyl-4'-thiocyanatobiphenyl						
Sol/Meso		305.4	27.4		89.72			
Meso/Liq		338.1	10.04		29.70	119.42	NA	[84]
C <sub>23</sub> H <sub>30</sub> ClNO <sub>2</sub> S		1-[2-(5-chlorothienyl)]-3-(4-decyloxyphenylamino)-2-propen-1-one						
Sol/Smec		344.7	29.19		84.68			
Smec/Smec		378.2	1.57		4.15			
Smec/Liq		455.5	8.21		18.02	106.85		[75]
C <sub>23</sub> H <sub>30</sub> N <sub>2</sub> O <sub>2</sub>		4-nitrobenzylidene 4'-decylaniline						
Sol/Nem		325.2	34.81		107.04			
Nem/Liq		327.2	1.72		5.26	112.30		[235]
C <sub>23</sub> H <sub>30</sub> N <sub>2</sub> O <sub>3</sub>		4-heptanoyloxy-2,3-dimethyl-4'-ethoxyazobenzene						
Sol/Nem		346.2	45		129.99			
Nem/Liq		365.2	0.6		1.64	131.63		[339]
C <sub>23</sub> H <sub>30</sub> N <sub>2</sub> O <sub>3</sub>		4-octanoyloxy-2-methyl-4'-ethoxyazobenzene						
Sol/Nem		327.2	40		122.25			
Nem/Liq		333.2	0.7		2.10	124.35		[339]
C <sub>23</sub> H <sub>30</sub> N <sub>2</sub> O <sub>3</sub>		4-octanoyloxy-3-methyl-4'-ethoxyazobenzene						
Sol/Nem		323.2	43		133.04			
Nem/Liq		339.2	0.6		1.77	134.81		[339]
C <sub>23</sub> H <sub>30</sub> O <sub>3</sub>		4-pentyloxyphenyl 4'-pentylnbenzoate						
Sol/Nem		315.2	19.58		62.12			
Nem/Liq		325.0	0.61		1.88	64.00	131.5	[218]
C <sub>23</sub> H <sub>30</sub> O <sub>3</sub> S		S-(4-octyloxyphenyl) 4-butoxythiobenzoate						
Sol/Nem		333.2	40.1		120.35			
Nem/Liq		378.2	1.7		4.49	124.84	NA	[4]
C <sub>23</sub> H <sub>30</sub> O <sub>4</sub>		4,4'-bis( $\omega$ -hydroxybutoxy)- $\alpha$ -methylstilbene						
Sol/Sol		348.5	5.82		16.7			
Sol/Smec		412.4	20.70		50.2			
Smec/Nem		426.6	5.20		12.2			
Nem/Liq		432.4		Not reported in paper				[56]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{23}H_{31}NO$		4-propyl-N-[[4-(heptyloxy)phenyl]methylene]benzenamine				
	Sol/SmeC	312.3	22.29	71.37		
	SmeC/SmeC	337.4	4.64	13.75		
	SmeC/Liq	346.7	1.29	3.72		
$C_{23}H_{31}NO$	Nem/Liq	354.3	0.99	2.79	91.63	[242]
		N-(4-ethoxybenzylidene)-4-octylaniline				
	Sol/Nem	320.8	31.0	96.63		
	Nem/Liq	340.9	0.63	1.85	98.48	126 [376, 381]
Note: Temperature calculated from published enthalpy and entropy data in paper.						
$C_{23}H_{31}NO$		N-(4-propoxybenzylidene)-4-heptylaniline				
	Sol/Nem	303.8	29.2	96.12		
	Nem/Liq	350.0	0.88	2.51	98.63	126 [376, 381]
		Note: Temperature calculated from published enthalpy and entropy data in paper.				
$C_{23}H_{31}NO$		N-(4-butoxybenzylidene)-4-hexylaniline				
	Sol/Sol	280.2	4.2	14.99		
	Sol/Sol	295.2	0.2	0.68		
	Sol/SmeC	299.2	8.9	29.75		
$C_{23}H_{31}NO$	SmeC/SmeC	310.5	2.47	7.95		
	SmeC/Nem	322.6	0.84	2.60		
	Nem/Liq	360.7	0.92	2.55	58.52	126 [376, 381]
		Note: Temperatures calculated from published enthalpy and entropy data in paper.				
$C_{23}H_{31}NO$		N-(4-pentyloxybenzylidene)-4-pentylaniline				
	Sol/SmeC	302.2	7.1	23.49		
	SmeC/SmeC	312.5	2.09	6.69		
	SmeC/Nem	333.3	0.21	0.63		
$C_{23}H_{31}NO$	Nem/Liq	355.9	0.88	2.47	33.28	126 [376, 381]
		Note: Temperatures calculated from published enthalpy and entropy data in paper.				
$C_{23}H_{31}NO$		N-(4-hexyloxybenzylidene)-4-n-butylaniline				
	Sol/SmeC	306.6	23.29	75.96		
	SmeC/SmeC	331.6	0.80	2.42		
	SmeC/SmeC	332.9	3.37	10.12		
$C_{23}H_{31}NO$	SmeC/SmeC	343.2	3.20	9.32		
	Nem/Liq	350.9	1.89	5.38	103.22	126 [157]
		Independent values from another reference				
	Sol/SmeC	283.2	1.1	3.88		
$C_{23}H_{32}$	SmeC/SmeC	328.6	2.89	8.79		
	SmeC/Nem	337.5	2.26	6.70		
	Nem/Liq	348.6	1.59	4.56	23.93	126 [376, 381]
		Note: Temperatures calculated from published enthalpy and entropy data in paper.				
The Sol/SmeC transition enthalpy is unusually low, and is likely in error.						
$C_{23}H_{32}$		4-pentyl-4'-hexylbiphenyl				
	Sol/SmeC	Not reported in paper				
	SmeC/SmeC	284.9	0.34	1.19		
	SmeC/SmeC	314.9	0.20	0.64		
	SmeC/SmeC	315.8	1.62	5.13		
$C_{23}H_{32}ClN_2O_2$	SmeC/Liq	326.9	9.99	30.56		[355]
		4-chloro-2'-hydroxy-4'-undecyloxyazobenzene				
	Sol/SmeC	340.6	37.3	109.51		
$C_{23}H_{32}FNO_2$	SmeC/Liq	369.5	4.6	12.45	121.96	[229]
		4-cyano-2-fluorophenyl <i>trans</i> -4-nonylcyclohexane-1-carboxylate				
	Sol/Nem	352.7	41.00	116.25		
$C_{23}H_{32}FNO_2$	Nem/Liq	364.7	Not reported in paper			[197]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$\text{C}_{23}\text{H}_{32}\text{N}_2\text{O}$	Sol/Nem	338.9	94.93	280.11	127.3	[153]
	Nem/Liq	354.2	3.39	9.57		
$\text{C}_{23}\text{H}_{32}\text{N}_2\text{O}$	Sol/Nem	326.2	28.28	86.70	127.3	[141]
	Nem/Liq	343.3	0.72	2.10		
$\text{C}_{23}\text{H}_{32}\text{N}_2\text{O}$	Sol/Nem	320.2	20.81	64.99	127.3	[131]
	Nem/Liq	359.0	1.19	3.31		
$\text{C}_{23}\text{H}_{32}\text{N}_2\text{O}$	Sol/Nem	312.8	13.22	42.26	127.3	[390]
	Nem/Liq	354.6	1.30	3.67		
$\text{C}_{23}\text{H}_{32}\text{N}_2\text{O}$	Sol/Smec	342.2	4.86	14.20	110	[279]
	Smec/Nem	343.2	3.35	9.76		
	Nem/Liq	349.2	0.42	1.20		
$\text{C}_{23}\text{H}_{32}\text{N}_2\text{O}_2$	Sol/Nem	341.8	31.9	93.33	132.7	[73]
	Nem/Liq	358.1	1.1	3.07		
Independent values from another reference						
	Sol/Nem	338.2	10.96	32.41	132.7	[282]
	Nem/Liq	357.2	0.60	1.68		
Note: Transition enthalpies seem abnormally low for this compound.						
$\text{C}_{23}\text{H}_{32}\text{O}_2\text{S}$	Sol/Meso	393.7	22.16	56.29	146.0	[18]
	Meso/Liq	400.6	7.53	18.80		
$\text{C}_{23}\text{H}_{33}\text{F}_3$	Sol/Nem	314.9	35.79	113.66	114.90	[27]
	Nem/Liq	395.7	0.49	1.24		
$\text{C}_{23}\text{H}_{33}\text{F}_3$	Sol/Nem	309.2	27.17	87.87	89.29	[27]
	Nem/Liq	373.2	0.53	1.42		
$\text{C}_{23}\text{H}_{33}\text{NS}$	Sol/Nem	322.3	22.57	70.03	NA	[52]
	Nem/Liq	512.6	1.67	3.26		
$\text{C}_{23}\text{H}_{33}\text{NS}$	Sol/Nem	323.7	19.2	59.31	NA	[357]
	Nem/Liq	360.7	Not reported in paper			
$\text{C}_{23}\text{H}_{33}\text{NS}$	Sol/Nem	334.2	15.5	46.38	NA	[357]
	Nem/Liq	378.7	Not reported in paper			
$\text{C}_{23}\text{H}_{34}\text{F}_2$	Sol/Nem	327.4	22.18	67.75	68.73	[27]
	Nem/Liq	398.2	0.39	0.98		
$\text{C}_{23}\text{H}_{34}\text{N}_2\text{O}_6$	Sol/Smec	483.2	53.0	109.69	117.55	[284]
	Smec/Liq	509.2	4.0	7.86		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{23}H_{34}O$		4-hexyl-4'-[S-(+)-2-methylbutoxy]biphenyl				
	Sol/Smec	308.8	6.97	22.57		
	Smec/Liq	318.8	32.35	101.47	124.04	117.9 [172]
$C_{23}H_{34}O_2$		4-propylcyclohexyl 4-pentylcinnamate				
	Sol/Nem	318.5	18.2	57.14		
	Nem/Liq	338.6	1.0	2.95	60.09	122.4 [5]
$C_{23}H_{34}O_3$		4-heptylcyclohexyl 4-methoxycinnamate				
	Sol/Nem	325.9	36.0	110.46		
	Nem/Liq	368.5	1.3	3.53	113.99	129.2 [5]
$C_{23}H_{35}F_2N$		N-[2-[4-(4-pentylcyclohexyl)cyclohexyl]ethyl]-3,4-difluoropyrrole				
	Sol/Nem	343.7	31.6	91.94		
	Nem/Liq	363.4	0.75	2.06	94.00	[325]
$C_{23}H_{35}NS$		4-( <i>trans</i> -4'-decylcyclohexyl)isothiocyanatobenzene				
	Sol/Nem	314.9	37.80	120.04		
	Nem/Liq	323.9	0.80	2.47	122.51	NA [151]
$C_{23}H_{36}O$		1-ethoxy-(4-[ <i>E</i> -3-( <i>trans</i> -4-pentylcyclohexyl)-1-butenyl])benzene				
	Sol/Nem	307.2	17.9	58.27		
	Nem/Liq	324.2	Not reported in paper			[198]
$C_{23}H_{37}NO_5$		4-hexadecyloxy-3-nitrobenzoic acid				
	Sol/Smec	371.2	57.7	155.44		
	Smec/Liq	371.2	2.5	6.73	162.17	209.4 [1]
$C_{23}H_{38}N_2O_2$		N,N'-diheptanoyl-2,3,5-trimethylbenzene-1,4-diamine				
	Sol/Meso	453.2	14	30.89		
	Meso/Liq	512.2	24	46.86	77.75	[36]
$C_{23}H_{38}O$		1-ethoxy-4-[4-( <i>trans</i> -4-pentylcyclohexyl)-1-butyl]benzene				
	Sol/Smec	291.2	14.6	50.14		
	Smec/Liq	304.2	Not reported in paper			[198]
$C_{23}H_{38}O_3$		4-hexadecyloxybenzoic acid				
	Sol/Smec	375.2	48.53	129.34		
	Smec/Liq	406.2	9.62	23.68	153.02	192.6 [156]
$C_{23}H_{46}O_6S$		6-O-(propylene-[3'-S-tetradecyl])- $\alpha$ -D-galactopyranose				
	Sol/Smec	382.2	66.70	174.52		
	Smec/Liq	436.2	0.68	1.56	176.08	[39]
$C_{24}H_{18}FNO_5S$		4-propoxyphenyl 3-fluoro-4-thiocyanatophenyl terephthalate				
	Sol/Nem	422.9	39.8	94.11		
	Nem/Liq	425.2	0.2	0.47	94.58	NA [37]
$C_{24}H_{18}F_2O$		1-(6-butoxy-2-naphthyl)-4-(3,4-difluorophenyl)diacetylene				
	Sol/Nem	399.2	34.5	86.42		
	Nem/Liq	427.8	0.1	0.23	86.65	[95]
$C_{24}H_{18}N_2O_4$		ethyl 4-[4-(4-cyanobenzoyloxy)benzylideneamino]benzoate				
	Sol/Nem	454.2	36.0	79.26		
	Nem/Liq	579.2	0.3	0.52	79.78	[196]
$C_{24}H_{18}O_5$		7-(4'-ethoxybenzoyloxy)isoflavone				
	Sol/Nem	468.2	28.26	60.36		
	Nem/Liq	503.5	0.72	1.43	61.79	[14]
$C_{24}H_{18}O_6S_2$	Sol/Nem	bis(4-methoxyphenyl) 2,2'-bithiophene-5,5'-dicarboxylate	519.4	57.6	110.90	

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{24}H_{18}O_8$	Nem/Liq	556.3	1.1	1.98	112.88	121.0
	Sol/Nem	520.2	58.0	111.50		[12]
	Nem/Liq	574.2	1.01	1.76	113.26	109.8
$C_{24}H_{19}FO$		<i>bis</i> (4-methoxycarbonylphenyl) terephthalate				
	Sol/Nem	380.5	25.4	66.75		[95]
	Nem/Liq	459.3	0.2	0.44	67.19	
$C_{24}H_{19}NO_5S$		1-(6-butoxy-2-naphthyl)-4-(4-fluorophenyl)diacetylene				
	Sol/Smec	412.2	30.5	73.99		[114]
	Smec/Nem	417.2	Not reported in paper			
$C_{24}H_{20}ClNO_4$	Nem/Liq	467.2	0.4	0.86		NA
		isopropyl 4-[4-(4-chlorobenzoyloxy)benzylideneamino]benzoate				
	Sol/Smec	429.2	26.0	60.58		[196]
$C_{24}H_{20}ClNO_4$	Smec/Liq	481.2	8.3	17.25	77.83	
		propyl 4-[4-(4-chlorobenzoyloxy)benzylideneamino]benzoate				
	Sol/Smec	447.2	38.0	84.97		[196]
$C_{24}H_{20}N_2O_4$	Smec/Nem	472.2	2.8	5.93	90.90	
		dimethyl N,N'-[1,4-phenylene- <i>bis</i> (methylidyne)]- <i>bis</i> [aminobenzoate]				
	Sol/Nem	516.2	53.0	102.67		[192]
$C_{24}H_{20}N_2O_6$	Nem/Liq	612.2	Not reported in paper			
		isopropyl 4-[4-(4-nitrobenzoyloxy)benzylideneamino]benzoate				
	Sol/Nem	451.2	42.0	93.09		[196]
$C_{24}H_{20}N_2O_6$	Nem/Liq	504.2	0.6	1.19	94.28	
		propyl 4-[4-(4-nitrobenzoyloxy)benzylideneamino]benzoate				
	Sol/Nem	414.2	32.0	77.26		[196]
$C_{24}H_{21}N$	Nem/Liq	544.2	Not reported in paper			
		2-(4-cyanophenyl)-7-butylfluorene				
	Sol/Nem	431.1	18.92	43.89		[2]
$C_{24}H_{21}NO_4$	Nem/Liq	492.8	0.41	0.83	44.72	
		ethyl 4-[4-(4-methylbenzoyloxy)benzylideneamino]benzoate				
	Sol/Nem	454.2	46.0	101.28		[196]
$C_{24}H_{21}NO_5$	Nem/Liq	523.2	0.5	0.96	102.24	
		ethyl 4-[4-(4-methoxybenzoyloxy)benzylideneamino]benzoate				
	Sol/Nem	474.2	52.0	109.66		[195]
$C_{24}H_{21}NO_5$	Nem/Liq	555.2	0.6	1.08	110.74	
		ethyl 4-[4-(4-methoxybenzylideneamino)benzoyloxy]benzoate				
	Sol/Nem	458.2	53.0	115.67		[195]
$C_{24}H_{22}F_2O$	Nem/Liq	541.2	0.6	1.11	116.78	
		1-(6-hexyloxy-2-naphthyl)-2-(3,4-difluorophenyl)acetylene				
	Sol/Nem	346.3	18.6	53.71		[95]
$C_{24}H_{23}FO$	Nem/Liq	361.4	0.4	1.11	54.82	
		1-(6-hexyloxy-2-naphthyl)-2-(4-fluorophenyl)acetylene				
	Sol/Nem	370.4	23.0	62.10		[95]
$C_{24}H_{23}F_7O_5$	Nem/Liq	397.2	0.5	1.26	63.36	
		4-(2,2,3,3,4,4,4-heptafluorobutyloxycarbonyl)phenyl-4-(hexyloxy) benzoate				
	Sol/Smec	336.9	19.49	57.86		
	Smec/Smec	345.2	0.31	0.90		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{24}H_{23}F_{17}S$	Smec/Liq	372.4	6.18	16.60	75.36	[125]
	Sol/Smec	358.6	21.8	60.79		
	Smec/Liq	366.2	12.7	34.68	95.47	[66]
$C_{24}H_{23}N$	4-(4-pentylphenyl)-4'-cyanobiphenyl					
	Sol/Sol	353.2	0.79	2.24		
	Sol/Sol	388.2	6.61	17.03		
	Sol/Nem	404.2	9.00	22.27		
$C_{24}H_{23}NO$	Nem/Liq	513.2	0.93	1.81	43.35	[365]
	4-pentyloxy-4"-cyano-p-terphenyl					
	Sol/Smec	377.2	7.17	19.01		
	Smec/smec	403.2	1.20	2.98		
	Smec/Nem	445.2	12.29	27.61		
$C_{24}H_{23}NO_3$	Nem/Liq	526.2	1.37	2.60	52.20	[277]
	phenyl 4-(4-butoxybenzylideneamino)benzoate					
	Sol/Smec	388.2	32.0	82.43		
	Smec/Nem	408.2	1.3	3.18		
	Nem/Liq	441.2	0.5	1.13	86.74	[292]
$C_{24}H_{23}NO_3$	4-ethylphenyl 4-(4-ethoxybenzylideneamino)benzoate					
	Sol/Nem	429.2	40.0	93.20		
	Nem/Liq	534.2	1.3	2.43	95.63	[292]
$C_{24}H_{23}NO_3$	4-isopropylphenyl 4-(4-methoxybenzylideneamino)benzoate					
	Sol/Nem	404.2	30.0	74.22		
	Nem/Liq	479.2	0.5	1.04	75.26	[292]
$C_{24}H_{23}NO_8$	4-[(4-hexyloxy)phenoxy]carbonylphenyl 5-nitro-2-furancarboxylate					
	Sol/Nem	420.2	Value not reported in paper			
	Nem/Liq	445.2	1.1	2.47		[352]
$C_{24}H_{24}N_2O_3$	4-[(pyridine-4-ylmethylene)amino]phenyl 4-pentyloxybenzoate					
	Sol/Nem	386.2	24.7	63.96		
	Nem/Liq	431.3	0.2	0.46	64.42	[265]
$C_{24}H_{25}NS$	4'-(4-propylcyclohexyl)-4-iso thiocyanatotolane					
	Sol/Nem	410.4	26.5	64.57		
	Nem/Liq	538.6	1.51	2.80	67.37	NA
$C_{24}H_{27}NO_2$	6-n-pentyloxy-2-[4-ethoxystyryl]quinoline					
	Sol/Sol	374.8	11.75	31.35		
	Sol/Nem	396.7	28.84	72.70		
	Nem/Liq	465.9	0.54	1.16	105.21	[112]
$C_{24}H_{27}NO_2$	1-[2-( <i>trans</i> -4-butylcyclohexyl)]-4-[(4-nitrophenyl)ethynyl]benzene					
	Sol/Nem	409.4	23.68	57.84		
	Nem/Liq	497.9	0.88	1.77	59.61	[238]
$C_{24}H_{27}NO_2$	(E)-4-{2-[4-(7-vinyl oxyheptyloxy)phenyl]vinyl}benzonitrile					
	Sol/Smec	334.7	5.3	15.84		
	Smec/Nem	345.1	16.3	47.23		
	Nem/Liq	351.3	8.5	24.20	87.25	[435]
$C_{24}H_{27}NO_2S$	4-iso thiocyanatophenyl 4-( <i>trans</i> -4-butylcyclohexyl)benzoate					
	Sol/Smec	382.2	25.94	67.87		
	Smec/Nem	397.7	Too small to be measured			
	Nem/Liq	512.2	0.92	1.80	69.67	NA
						[356]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)						
$C_{24}H_{28}O_3$		6-(10-undecyloxy)-2-naphthoic acid						
	Sol/Sol	393.8	126.4		320.98			
	Sol/Smec	399.7		Not completely resolved				
	Note: Sol/Smec transition enthalpy is included in the value for the Sol/Sol transition.							
	Smec/Nem	410.4	1.67		4.07			
$C_{24}H_{28}O_4$	Nem/Liq	444.2	7.95		17.90	342.95	121.8	[68]
		di(4'-ethylphenyl)- <i>trans</i> -cyclohexane-1,4-dicarboxylate						
	Smec/Nem	384.2	30.4		79.13			
$C_{24}H_{28}O_6$	Nem/Liq	434.2	0.51		1.17	80.30	104.9	[215]
		di(4'-ethoxyphenyl)- <i>trans</i> -cyclohexane-1,4-dicarboxylate						
	Smec/Nem	414.2	52.59		126.97			
$C_{24}H_{29}NO_2$	Nem/Liq	517.2	1.44		2.78	129.75	118.5	[220]
		(E)-4-[2-[4-(9-hydroxynonyloxy)phenyl]vinyl]benzonitrile						
	Smec/Nem	364.2	28.8		79.08			
$C_{24}H_{29}N_3O$	Nem/Liq	371.6	12.0		32.29	111.37		[435]
		4-[5-[4-( <i>trans</i> -4-pentylcyclohexyl)phenyl]-1,2,4-oxadiazol-3-yl]pyridine						
	Smec/Nem	351.2	16.52		47.04			
$C_{24}H_{29}N_3O$	Nem/Liq	431.2	0.30		0.70	47.74	125	[279]
		2-[5-[4-( <i>trans</i> -4-pentylcyclohexyl)phenyl]-1,2,4-oxadiazol-3-yl]pyridine						
	Smec/Nem	368.2	21.22		57.63			
$C_{24}H_{29}NO_5$	Nem/Liq	425.2	0.34		0.80	58.43	125	[279]
		butyl 4-[[[4-[(pentyloxy)carbonyl]oxy]phenyl]imino]methyl]benzoate						
	Smec/Nem	321.7	25.8		80.20			
$C_{24}H_{30}N_2O_2S$	Nem/Liq	369.2	0.77		2.09	82.29		[365]
		1-[2-(5-cyanothienyl)]-3-(4-decyloxyphenylamino)-2-propen-1-one						
	Smec/Smec	386.7	7.11		18.39			
$C_{24}H_{30}N_2O_2S_2$	Smec/Liq	461.1	7.03		15.25	33.64		[75]
		2,5-bis(4-pentyloxyphenyl)thiazolo[5,4-d]dithiazole						
	Smec/Smec	362.9	4.98		13.72			
	Smec/Smec	392.3	4.69		11.96			
	Smec/Smec	453.0	26.52		58.54			
$C_{24}H_{30}N_2O_3$	Smec/Smec	497.3	5.43		10.92			
	Smec/Smec	543.9	1.74		3.20	98.34		[269]
		4-n-pentanoyl-4-n'-heptanoyloxyazobenzene						
	Smec/Smec	347.1	7.51		21.63			
$C_{24}H_{30}N_2O_4$	Smec/Smec	371.4	13.27		35.74			
	Smec/Smec	375.6	8.71		23.18			
	Smec/Smec	401.7	6.83		17.01	97.56	130.0	[157]
	Smec/Smec							
$C_{24}H_{30}N_2O_5$		4-n-propanoyl-4-n'-nonanoyloxyazobenzene						
	Smec/Smec	367.2	3.02		8.22			
	Smec/Smec	361.1	1.76		4.87			
	Smec/Smec	417.2	1.76		4.87			
$C_{24}H_{30}N_4OS$	Smec/Smec	417.2	7.03		16.84	29.9	130.0	[157]
	Smec/Smec							
		5-(4-pyridyl)-2-(4-decyloxy)benzylideneamino-1,3,4-thiadiazole						
$C_{24}H_{31}ClN_2O_4$	Smec/Smec	415.3	25.3		60.92			
	Smec/Smec	438.8	5.0		11.39	72.31		[79]
$C_{24}H_{31}ClN_2O_4$		4-(4-hexyloxyphenylazoxy)phenyl 2S,3S-2-chloro-3-methyl-pentanoate						
	Smec/Smec	326.4	29.14		89.28			
	Smec/Smec	329.2	0.36		1.09			

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta S_{\text{pce}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )				
$\text{C}_{24}\text{H}_{31}\text{ClO}_2\text{S}$	Nem/Liq	346.8	0.85	2.45	92.82		[47]
	Sol/Smec	348.4	34.10	97.88		NA	[383]
	Smec/Liq	370.9	4.39	11.84	109.72		
$\text{C}_{24}\text{H}_{31}\text{N}$	4-chlorophenyl 4-undecyloxythiobenzoate						
	Sol/Smec	325.3	46.90	144.17		133.4	[233]
	Smec/Liq	329.3	4.18	12.69	156.86		
Independent values from another reference							
	Sol/Smec	326.0	38.07	116.78		133.4	[237]
	Smec/Liq	330.0	3.39	10.27	127.05		
Independent values from another reference							
	Sol/Smec	325.5	42.7	131.18		133.4	[350]
	Smec/Liq	329.5	3.85	11.68	142.86		
Independent values from another reference							
	Sol/Smec	326.2	43.2	132.43		133.4	[387]
	Smec/Liq	330.2	3.8	11.50	143.93		
$\text{C}_{24}\text{H}_{31}\text{NO}$	4-cyano-4'-undecyloxybiphenyl						
	Sol/Smec	342.2	37.12	108.47		140.2	[222]
	Smec/Liq	359.2	3.32	9.24	117.71		
Independent values from another reference							
	Sol/Smec	341.8	44.73	130.87		140.2	[233]
	Smec/Liq	359.8	4.10	11.40	142.27		
$\text{C}_{24}\text{H}_{31}\text{NO}_3$	4-(4-decyloxyphenyliminomethyl)benzoic acid						
	Sol/Smec	448.2	29.1	64.93		127.50	[416]
	Smec/Liq	532.2	33.3	62.57			
$\text{C}_{24}\text{H}_{31}\text{NO}_5$	4-undecyloxyphenyl 4-nitrobenzoate						
	Sol/Smec	346.2	46.02	132.93		155.5	[236]
	Smec/Liq	358.2	3.56	9.94	142.87		
$\text{C}_{24}\text{H}_{31}\text{NO}_5$	4'-undecyloxy-3'-nitrobiphenyl-4-carboxylic acid						
	Sol/Smec	397.0	27.15	68.39		77.10	[87]
	Smec/Smec	473.0	0.33	0.70			
$\text{C}_{24}\text{H}_{32}\text{ClNO}_2\text{S}$	4-[2-(5-chlorothienyl)]-3-(4-undecyloxyphenylamino)-2-propen-1-one						
	Sol/Smec	348.3	29.89	85.82		110.04	[75]
	Smec/Smec	377.5	2.20	5.83			
$\text{C}_{24}\text{H}_{32}\text{FNO}_2\text{S}$	3-fluoro-4-isothiocyanatophenyl 4-octylbicyclo[2.2.2]octane-1-carboxylate						
	Sol/Sol	332.2	5.0	15.05		NA	[357]
	Sol/Nem	341.7	33.8	98.92			
$\text{C}_{24}\text{H}_{32}\text{N}_2\text{O}_2$	4-nitrobenzylidene 4'-undecylaniline						
	Sol/Nem	325.2	15.94	49.02		135.6	[235]
	Nem/Liq	334.2	1.09	3.26	52.28		
$\text{C}_{24}\text{H}_{32}\text{N}_4\text{OS}_2$	2-(4-pentyloxyphenylazo)-5-(5'-heptyl-2'-thienyl)-1,3,4-thiadiazole						
	Sol/Nem	416.0	31.7	76.20		76.68	[396]
	Nem/Liq	418.9	0.2	0.48			
$\text{C}_{24}\text{H}_{32}\text{N}_2\text{O}_3$	4-octanoyloxy-2,3-dimethyl-4'-ethoxyazobenzene						

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$\text{C}_{24}\text{H}_{32}\text{N}_2\text{O}_3$	Sol/Nem	342.2	32	93.51	94.63	[339]
	Nem/Liq	358.2	0.4	1.12		
$\text{C}_{24}\text{H}_{32}\text{N}_2\text{O}_3$	4-nonenoyloxy-2-methyl-4'-ethoxyazobenzene					
	Sol/Nem	335.2	51	152.15	155.12	[339]
	Nem/Liq	336.2	1.0	2.97		
$\text{C}_{24}\text{H}_{32}\text{N}_2\text{O}_3$	4-nonenoyloxy-3-methyl-4'-ethoxyazobenzene					
	Sol/Nem	324.2	48	148.06	150.71	[339]
	Nem/Liq	339.2	0.9	2.65		
$\text{C}_{24}\text{H}_{32}\text{O}_3$	4-hexyloxyphenyl 4'-pentylbenzoate					
	Sol/Nem	307.4	15.44	50.23	53.02	138.6
	Nem/Liq	333.1	0.93	2.79		
$\text{C}_{24}\text{H}_{32}\text{O}_6\text{S}$	<i>bis</i> (4-pentyloxyphenyl) 2,5-thiophenedicarboxylate					
	Sol/Sol	365.1	13.7	37.52	144.62	[12]
	Sol/Nem	409.8	43.1	105.17		
$\text{C}_{24}\text{H}_{33}\text{NO}$	N-(4-propoxybenzylidene)-4-octylaniline					
	Sol/Sol	302.2	3.4	11.25	127.72	[376, 381]
	Sol/Sol	305.2	1.9	6.23		
	Sol/Sol	309.2	16.5	53.36		
	Sol/Sol	311.2	10.3	33.10		
$\text{C}_{24}\text{H}_{33}\text{NO}$	Sol/Nem	312.1	6.6	21.14	127.72	[376, 381]
	Nem/Liq	333.3	0.88	2.64		
Note: Temperature calculated from published enthalpy and entropy data in paper.						
$\text{C}_{24}\text{H}_{33}\text{NO}$	N-(4-butoxybenzylidene)-4-heptylaniline					
	Sol/Smec	293.2	6.6	22.51	34.11	[376, 381]
	Smec/Smec	317.6	2.26	7.12		
	Smec/Nem	333.3	0.42	1.26		
	Nem/Liq	363.6	1.17	3.22		
Note: Temperatures calculated from published enthalpy and entropy data in paper.						
$\text{C}_{24}\text{H}_{33}\text{NO}$	N-(4-pentyloxybenzylidene)-4-hexylaniline					
	Sol/Smec	309.2	23.1	74.71	89.91	[376, 381]
	Smec/Smec	316.7	2.38	7.51		
	Smec/Nem	329.4	1.17	3.55		
	Nem/Liq	343.4	1.42	4.14		
Note: Temperatures calculated from published enthalpy and entropy data in paper.						
$\text{C}_{24}\text{H}_{33}\text{NO}$	N-(4-hexyloxybenzylidene)-4-pentylaniline					
	Sol/Smec	313.2	9.7	30.97	48.94	[376, 381]
	Smec/Smec	342.0	2.72	7.95		
	Smec/Nem	350.0	1.46	4.17		
	Nem/Liq	364.3	2.13	5.85		
Note: Temperatures calculated from published enthalpy and entropy data in paper.						
$\text{C}_{24}\text{H}_{33}\text{NO}$	N-(4-heptyloxybenzylidene)-4-butylaniline					
	Sol/Smec	305.4	30.2	98.89	Not reported in paper	[381]
	Smec/Liq					
$\text{C}_{24}\text{H}_{33}\text{NO}$	N-(4-nonyloxybenzylidene)-4-ethylaniline					
	Sol/Smec	309.1	28.51	92.24	121.69	[65]
	Smec/Smec	333.9	3.41	10.21		
	Smec/Liq	343.6	6.61	19.24		
$\text{C}_{24}\text{H}_{33}\text{NO}_2\text{S}$	4-isothiocyanatophenyl 4-octylbicyclo[2.2.2]octane-1-carboxylate					
	Sol/Sol	325.7	2.3	7.06		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$\text{C}_{24}\text{H}_{33}\text{NO}_3$	Sol/Nem	335.7	22.2	66.13	NA	[357]
	Nem/Liq	376.2	Not reported in paper			
$\text{C}_{24}\text{H}_{34}\text{ClN}_2\text{O}_2$	N-(4-methoxyphenyl)- $\alpha$ -(4-pentyloxyphenyl)nitrone					
	Sol/Smec	381.2	48.37	126.89	NA	[162]
	Smec/Nem	382.2	Not reported in paper			
	Smec/Liq	402.2	1.56	3.88		
$\text{C}_{24}\text{H}_{34}\text{F}_2\text{O}$	4-chloro-2'-hydroxy-4'-dodecyloxyazobenzene				[229]	
	Sol/Smec	333.9	33.5	100.33		
	Smec/Liq	369.2	4.9	13.27	113.60	
$\text{C}_{24}\text{H}_{34}\text{N}_2\text{O}$	1,2-difluoro-4-[[3-(4'-propyl[1,1'-bicyclohexyl]-4-yl)-2-propenyl]oxy]benzene				[198]	
	Sol/Nem	321.2	27.0	84.06		
	Nem/Liq	374.2	Not reported in paper			
$\text{C}_{24}\text{H}_{34}\text{N}_2\text{O}$	4-propyl-4'-nonyloxyazobenzene				[153]	
	Sol/Nem	336.2	105.1	312.61		
	Nem/Liq	351.9	3.64	10.34	322.95	
$\text{C}_{24}\text{H}_{34}\text{N}_2\text{O}$	4-butyl-4'-octyloxyazobenzene				[141]	
	Sol/Smec	328.3	32.93	100.30		
	Smec/Nem	331.2	0.80	2.42		
$\text{C}_{24}\text{H}_{34}\text{N}_2\text{O}$	Nem/Liq	349.0	1.11	3.18	105.90	[141]
	4-pentyl-4'-heptyloxyazobenzene					
	Sol/Nem	320.2	21.64	67.58		
$\text{C}_{24}\text{H}_{34}\text{N}_2\text{O}$	Nem/Liq	355.0	1.22	3.44	71.02	[141]
	4-heptyl-4'-pentyloxyazobenzene					
	Sol/Nem	312.7	16.19	51.77		
$\text{C}_{24}\text{H}_{34}\text{N}_2\text{O}$	Nem/Liq	348.3	1.22	3.50	55.27	
	<i>bis</i> (4-hexylphenyl)diazene N-oxide				[365]	
	Sol/Nem	298.7	17.2	57.58		
$\text{C}_{24}\text{H}_{34}\text{N}_2\text{O}$	Nem/Liq	327.5	0.69	2.11	59.69	
	2-[4-(5-hexenoxy)phenyl]-5-[(S)-5-methylheptyl]pyrimidine				[201]	
	Sol/Chol	303.2	7.9	26.06		
$\text{C}_{24}\text{H}_{34}\text{N}_2\text{O}_2$	Chol/Liq	304.2	Not reported in paper			
	4-methyl-2'-hydroxy-4'-undecyloxyazobenzene				[73]	
	Sol/Nem	339.6	38.4	113.07		
$\text{C}_{24}\text{H}_{34}\text{N}_2\text{O}_3$	Nem/Liq	350.6	1.0	2.85	115.92	
	4,4'-dihexyloxyazoxybenzene				[9]	
	Sol/Sol	352.0	3.90	11.08		
	Sol/Nem	355.5	37.50	105.49		
	Nem/Liq	403.5	1.35	3.35	119.92	
	Independent values from another reference					
	Sol/Nem	354.5	41.39	116.76	149.8	[179]
	Nem/Liq	402.3	1.05	2.61	119.37	
	Independent values from another reference				149.8	[440]
	Sol/Sol	349.0	3.84	11.00		
	Sol/Nem	355.0	36.3	102.25		
$\text{C}_{24}\text{H}_{34}\text{O}_2\text{S}$	Nem/Liq	403.0	1.35	3.35	116.60	149.8
	2-octanoyl-5-(4-hexyloxyphenyl)thiophene					
	Sol/Meso	393.2	22.99	58.47		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{24}H_{35}FO$	Meso/Liq	403.0	7.09	17.59	76.06	153.1
	Sol/Nem	348.2	29.9	85.87		[18]
	Nem/Liq	400.2	Not reported in paper			[198]
$C_{24}H_{35}NS$			4-( <i>trans, trans</i> -4-pentylbicyclohexyl)benzene-isothiocyanate			
	Sol/Sol	411.6	12.5	30.37		
	Sol/Nem	328.7	10.45	31.79		
$C_{24}H_{36}F_2O$	Nem/Liq	510.2	1.46	2.86	65.02	NA
			1,2-difluoro-4-[3-(4'-propyl[1,1'-bicyclohexyl]-4-yl)propoxy]benzene			
	Sol/Nem	352.2	35.1	99.66		[198]
$C_{24}H_{36}O_3$	Nem/Liq	380.2	Not reported in paper			
			4-heptylcyclohexyl 4-ethoxycinnamate			
	Sol/Nem	372.0	30.6	82.26		
$C_{24}H_{37}FO$	Nem/Liq	377.5	1.4	3.71	85.97	136.3
			1-fluoro-4-[3-(4'-propyl[1,1'-bicyclohexyl]-4-yl)propoxy]benzene			
	Sol/Nem	351.2	35.8	101.94		[198]
$C_{24}H_{37}NS$	Nem/Liq	397.2	Not reported in paper			
			4-( <i>trans</i> -4'-undecylcyclohexyl)isothiocyanatobenzene			
	Sol/Nem	321.3	41.7	129.79		
$C_{24}H_{38}F_2O_2$	Nem/Liq	327.2	1.2	3.67	133.46	NA
			1-butoxy-2,3-difluoro-4-[3-( <i>trans</i> -4-pentylcyclohexyl)propoxy]benzene			
	Sol/Nem	293.2	34.6	118.00		[198]
$C_{24}H_{38}O$	Nem/Liq	295.2	Not reported in paper			
			1-propoxy-(4-[ <i>(E</i> )-3-( <i>trans</i> -4-pentylcyclohexyl)-1-butenyl])benzene			
	Sol/Smec	305.2	10.2	33.42		
$C_{24}H_{39}N_3O_3$	Smec/Liq	315.2	Not reported in paper			[198]
			N,N',N"-tripentanoyl-2,4,6-trimethyl-1,3,5-benzenetriamine			
	Sol/Meso	588.2	16.0	27.20		
$C_{24}H_{40}N_2O_2$	Meso/Liq	653.2	34.0	52.05	79.25	122.7
			N,N'-diheptanoyl-2,3,5,6-tetramethylbenzene-1,4-diamine			
	Sol/Meso	570.2	20	35.08		
$C_{24}H_{40}O_3$	Meso/Liq	582.2	25	42.94	78.02	[36]
			4-heptadecyloxybenzoic acid			
	Smec/Liq	377.2	54.39	144.19		
$C_{24}H_{46}O_7$	Smec/Liq	405.2	9.20	22.70	166.89	201.9
			6-O-octadecanoyl- $\alpha$ -D-galactopyranose			
	Sol/Smec	387.2	33.36	86.16		[156]
$C_{24}H_{48}O_5S$	Smec/Liq	454.2	0.80	1.76	87.92	244.7
			6-S-octadecyl-6-thio- $\alpha$ -D-galactopyranose			
	Sol/Smec	387.2	68.47	176.83		
$C_{24}H_{48}O_6$	Smec/Liq	450.2	1.48	3.29	180.12	248.4
			6-O-octadecyl- $\alpha$ -D-galactopyranose			
	Sol/Smec	392.2	54.34	138.55		
$C_{24}H_{50}O_6$	Smec/Liq	437.2	0.50	1.14	139.69	251
			6-O-octadecyl-D-galacitol			
	Sol/Smec	412.2	44.12	107.04		
	Smec/Liq	422.2	0.62	1.47	108.51	271
						[183]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{24}H_{70}Si_{10}$	Sol/Meso	323.2	11.9	36.82	74.10	[110]
	Meso/Liq	378.2	14.1	37.28		
$C_{25}H_{16}F_{16}O_5$	Sol/SmeC	353.7	22.17	62.68	74.60	[128]
	SmeC/SmeC	360.6	1.05	2.91		
	SmeC/Liq	364.1	3.28	9.01		
$C_{25}H_{17}ClF_4O_3$	Sol/Nem	398.6	24.89	62.44	64.12	[101]
	Nem/Liq	459.5	0.77	1.68		
	Sol/Nem	364.4	30.00	82.33		
$C_{25}H_{17}ClF_4O_3$	Nem/Liq	486.8	0.76	1.56	83.89	[101]
$C_{25}H_{17}F_{13}O_4S$	Sol/SmeC	309.5	7.60	24.56	NA	[134]
	SmeC/Liq	490.8	9.38	19.11		
	Sol/Nem	484.2	37.0	76.41		
$C_{25}H_{18}N_2O_2$	Nem/Liq	502.2	0.46	0.92	77.33	[338]
$C_{25}H_{18}N_2O_2$	Sol/Nem	485.2	41.0	84.50	84.90	[338]
	Nem/Liq	503.2	0.20	0.40		
$C_{25}H_{18}N_4O_2$	Sol/Nem	487.2	51.0	104.68		[338]
	Nem/Liq	525.2	0.25	0.48	105.16	
	Sol/Nem	523.2	39.0	74.54		
$C_{25}H_{19}N$	Nem/Liq	524.2	0.43	0.82	75.36	[338]
$C_{25}H_{20}FNO_5S$	Sol/SmeC	401.2	34.4	85.74	[37]	
	SmeC/Nem	Not reported in paper				
	Nem/Liq	426.2	0.3	0.70		
$C_{25}H_{20}F_2O$	Sol/Nem	378.3	32.6	86.17	[95]	
	Nem/Liq	414.2	0.1	0.24		
	Sol/Nem	441.2	38.0	86.13		
$C_{25}H_{20}N_2O_4$	Nem/Liq	519.2	0.3	0.58	86.71	[196]
$C_{25}H_{20}N_2O_4$	Sol/Nem	406.2	34.0	83.70	[196]	
	Nem/Liq	531.2	Not reported in paper			
$C_{25}H_{20}O_5$	Sol/Nem	467.8	36.89	78.86	[14]	
	Nem/Liq	486.4	0.62	1.27		
	Sol/Nem	412.2	38.8	94.13		
$C_{25}H_{21}ClO_8$	Nem/Liq	435.2	0.35	0.80	94.93	[174]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{25}H_{21}FO$	Sol/Nem	1-(6-pentyloxy-2-naphthyl)-4-(4-fluorophenyl)diacetylene				
	380.0	37.1	97.63			
	Nem/Liq	451.1	0.6	1.33	98.96	[95]
$C_{25}H_{21}NO$	Sol/Nem	5-(4'-methoxybiphenyl-4-yl)-2-(4-methylphenyl)pyridine				
	537.0	23.0	42.83			
	Nem/Liq	Decomposed prior to transition				
$C_{25}H_{21}NO_5S$	Sol/SmeC	4-thiocyanophenyl 4-(4-butoxybenzoyloxy)benzoate				
	402.2	42.1	104.67			
	SmeC/Nem	429.2	1.1	2.56		
$C_{25}H_{21}NO_6$	Nem/Liq	464.2	0.4	0.86	108.09	[114]
	Sol/SmeC	ethyl 4-[4-(4-acetoxybenzoyloxy)benzylideneamino]benzoate				
	419.2	23.0	54.87			
$C_{25}H_{22}ClNO_4$	SmeC/Nem	423.2	0.1	0.24		
	Nem/Liq	Not reported in paper				
	Sol/SmeC	419.2	23.0	54.87		[196]
$C_{25}H_{22}ClNO_4$	SmeC/Nem	423.2	0.1	0.24		
	Nem/Liq	464.2	0.4	0.86	108.09	[114]
	Sol/SmeC	<i>tert</i> -butyl 4-[4-(4-chlorobenzoyloxy)benzylideneamino]benzoate				
$C_{25}H_{22}ClNO_4$	484.2	57.0	117.72			
	SmeC/Nem	491.2	11.0	22.39		
	Nem/Liq	536.2	0.5	0.93	141.04	[196]
$C_{25}H_{22}ClNO_4$	Sol/SmeC	1-methylpropyl 4-[4-(4-chlorobenzoyloxy)benzylideneamino]benzoate				
	403.2	27.0	66.96			
	SmeC/Liq	460.2	5.7	12.39	79.35	[196]
$C_{25}H_{22}ClNO_4$	Sol/SmeC	2-methylpropyl 4-[4-(4-chlorobenzoyloxy)benzylideneamino]benzoate				
	405.2	54.0	133.27			
	SmeC/Nem	451.2	1.6	3.55		
$C_{25}H_{22}ClNO_4$	Nem/Liq	511.2	0.3	0.59	137.41	[196]
	Sol/SmeC	butyl 4-[4-(4-chlorobenzoyloxy)benzylideneamino]benzoate				
	428.2	37.0	86.41			
$C_{25}H_{22}N_2O_6$	SmeC/Nem	474.2	0.4	0.84		
	Nem/Liq	497.2	0.3	0.60	87.85	[196]
	Sol/Nem	<i>tert</i> -butyl 4-[4-(4-nitrobenzoyloxy)benzylideneamino]benzoate				
$C_{25}H_{22}N_2O_6$	448.2	42.0	93.71			
	Nem/Liq	458.2	0.1	0.22	93.93	[196]
	Sol/Nem	1-methylpropyl 4-[4-(4-nitrobenzoyloxy)benzylideneamino]benzoate				
$C_{25}H_{22}N_2O_6$	422.2	18.0	42.63			
	Nem/Liq	434.2	0.5	1.15	43.78	[196]
	Sol/Nem	2-methylpropyl 4-[4-(4-nitrobenzoyloxy)benzylideneamino]benzoate				
$C_{25}H_{22}N_2O_6$	415.2	37.0	89.11			
	Nem/Liq	531.2	0.4	0.75	89.86	[196]
	Sol/Nem	5-(2-chlorophenyl)-3-(4'-pentyl)[1,1'-biphenyl]-4-yl)-1,2,4-oxadiazole				
$C_{25}H_{23}ClN_2O$	353.9	27.16	76.74			
	Nem/Liq	359.0	0.20	0.56	77.30	[392]
	Sol/Nem	2-(4-cyanophenyl)-7-pentyfluorene				
$C_{25}H_{23}N$	401.8	14.54	36.19			
	Nem/Liq	489.5	0.43	0.88	37.07	[2]
	Sol/Nem	4-thiocyanophenyl 4-pentyloxybiphenyl-4'-carboxylate				
$C_{25}H_{23}NO_3S$	382.2	30.7	80.32			
	SmeC/Nem	444.2	2.2	4.95		
	Nem/Liq	450.2	0.3	0.67	85.94	[114]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta H_{\text{pcc}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pcc}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)						
$C_{25}H_{23}NO_4$	Sol/Nem	propyl 4-[4-(4-methylbenzoyloxy)benzylideneamino]benzoate		423.2	40.0	94.52		[196]
	Nem/Liq	409.2	0.3		0.59	95.11		
$C_{25}H_{23}NO_5$	Sol/Nem	isopropyl 4-[4-(4-methoxybenzoyloxy)benzylideneamino]benzoate		434.2	39.0	89.82		[195]
	Nem/Liq	485.2	0.1		0.21	90.03		
$C_{25}H_{23}NO_5$	Sol/Nem	propyl 4-[4-(4-methoxybenzoyloxy)benzylideneamino]benzoate		438.2	46.0	104.97		[195]
	Nem/Liq	540.2	0.5		0.93	105.90		
$C_{25}H_{23}NO_5$	Sol/SmeC	propyl 4-[4-(4-methoxybenzylideneamino)benzoyloxy]benzoate		432.2	52.0	120.31		[195]
	SmeC/Nem	434.2	0.1		0.23			
	Nem/Liq	531.2	0.3		0.56	121.10		
$C_{25}H_{23}NO_5$	Sol/SmeC	isopropyl 4-[4-(4-methoxybenzylideneamino)benzoyloxy]benzoate		426.2	38.0	89.16		[195]
	SmeC/Nem	450.2	1.7		3.78			
	Nem/Liq	478.2	0.4		0.84	93.78		
$C_{25}H_{23}NO_6$	Sol/Nem	4-[4-(hexyloxy)phenoxy]carbonylphenyl 5-cyano-2-furancarboxylate		412.2	Not reported in paper			[352]
	Nem/Liq	426.2	0.9		2.11			
$C_{25}H_{24}F_8O_5$	Sol/SmeC	4-(2,2,3,3,4,4,5,5-octafluoropentyloxycarbonyl)phenyl 4-(hexyloxy)benzoate		333.0	37.31	112.04		[125]
	SmeC/Liq	344.4	5.56		16.14	128.18		
$C_{25}H_{25}F_7O_5$	Sol/SmeC	4-(2,2,3,3,4,4,4-heptafluorobutyloxycarbonyl)phenyl 4-(heptyloxy)benzoate		332.5	21.47	64.57		[125]
	SmeC/SmeC	356.8	0.32		0.90			
	SmeC/Liq	368.6	6.08		16.49	81.96		
$C_{25}H_{25}IO_3S$	Sol/SmeC	4'-(7-octenyl)oxo-2-thiophenecarboxylate		418.9	23.85	56.93		[63]
	SmeC/Liq	428.1	15.99		37.35	94.28		
$C_{25}H_{25}NO$	4-hexyloxy-4"-cyano-p-terphenyl			363.2	2.45	6.75		[277]
	Sol/SmeC	388.2	2.24		5.77			
	SmeC/SmeC	440.2	8.89		20.20			
	SmeC/Nem	524.2	1.17		2.23	34.95	119.3	
	Nem/Liq							
$C_{25}H_{25}NO_3$	phenyl 4-(4-pentyloxybenzylideneamino)benzoate			389.2	41.0	105.34		[292]
	Sol/SmeC	408.2	1.4		3.43			
	Nem/Liq	431.2	0.4		0.93	109.70		
$C_{25}H_{25}NO_3$	Sol/Nem	4-isopropylphenyl 4-(4-ethoxybenzylideneamino)benzoate		419.2	37.0	88.26		[292]
	Nem/Liq	490.2	0.7		1.43	89.69		
$C_{25}H_{25}NO_3$	Sol/Nem	4-ethylphenyl 4-(4-propoxybenzylideneamino)benzoate		398.2	33.0	82.87		[292]
	Nem/Liq	516.2	1.1		2.13	85.00		
$C_{25}H_{26}$	4-[(4-hexen-1-yl)phenyl-1,3-butadiynyl]-4-propylbenzene			326.8	24.89	76.16		[212]
	Sol/Nem	436.5	1.09		2.50	78.66		
	Nem/Liq							

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{25}H_{26}FNS$	Sol/Nem	4'- <i>(4</i> -butylcyclohexyl)-3-fluoro-4-isothiocyanatotolane 350.3	20.75	59.23	NA	[135]
	Nem/Liq	512.1	1.21	2.36		
				61.59		
$C_{25}H_{26}N_2O_3$	Sol/Nem	4-[ <i>(pyridine-4-ylmethylene)amino</i> ]phenyl 4-hexyloxybenzoate 384.4	27.2	70.76	[265]	
	Nem/Liq	420.4	0.2	0.48		
				71.24		
$C_{25}H_{26}N_2O_3$	Sol/Smec	3-pyridyl 4-(4-hexyloxybenzylideneamino)benzoate 373.2	39.0	104.50	[291]	
	Smec/Nem	427.2	1.9	4.45		
	Nem/Liq	442.2	0.6	1.36		
$C_{25}H_{26}N_2O_3$	Sol/Nem	4[(E)- <i>(4</i> -hexyloxyphenyl)methylene]amino]phenyl 3-pyridinecarboxylate 383.2	45.0	117.43	[291]	
	Nem/Liq	448.2	0.7	1.56		
				118.99		
$C_{25}H_{26}N_2O_3$	Sol/Nem	4-[ <i>(E)-(3-pyridimino)methyl</i> ]phenyl 4-hexyloxybenzoate 378.2	35.0	92.54	[291]	
	Nem/Liq	441.2	0.6	1.36		
				93.90		
$C_{25}H_{26}N_2O_3$	Sol/Nem	4-[ <i>(E)-(3-pyridinylmethylene)amino</i> ]phenyl 4-hexyloxybenzoate 384.2	42.0	109.32	[291]	
	Nem/Liq	438.2	0.8	1.83		
				111.15		
$C_{25}H_{26}O_3$	Sol/Nem	4-biphenyl 4"-hexyloxybenzoate 405.7	39.33	Not reported in paper	[425]	
	Nem/Liq	409.7				
Note: Authors report only an enthalpy of fusion, and state in a footnote that Nem/Liq transition enthalpies for the compounds studied were in the 0.84–2.09 kJ·mol <sup>-1</sup> range.						
$C_{25}H_{27}NS$	4'- <i>(4</i> -butylcyclohexyl)-4-isothiocyanatotolane		NA	[135]		
	Sol/Sol	349.3	4.69			
	Smec/Nem	380.1	21.05			
	Nem/Liq	413.1	0.59			
$C_{25}H_{27}NS$	Sol/Smec	530.6	1.09	2.05	72.29	[238]
	Smec/Nem					
	Nem/Liq					
$C_{25}H_{28}$	1- <i>(trans-4-ethylcyclohexyl)</i> -4-[ <i>(4-isothiocyanatophenyl)ethynyl</i> ]benzene		NA	[212]		
	Sol/Nem	380.1	21.88			
	Nem/Liq	413.1	0.08	0.19		
$C_{25}H_{29}BrN_2O$	Sol/Nem	530.6	0.83	1.56	59.31	[392]
	Nem/Liq					
$C_{25}H_{29}ClN_2O$	Sol/Smec					[392]
	Smec/Nem					
	Nem/Liq					
$C_{25}H_{29}ClN_2O$	Sol/Sol	315.8	0.82	2.60	[392]	
	Smec/Nem	324.5	29.06	89.55		
	Nem/Liq	370.4	0.27	0.73		
				92.88		
$C_{25}H_{29}ClN_2O$	Sol/Smec	317.1	22.12	69.76	[392]	
	Smec/Nem	336.5	0.16	0.48		
	Nem/Liq	379.3	0.53	1.40		
				71.64		
$C_{25}H_{29}ClN_2O$	Sol/Sol	319.2	0.12	0.38	[392]	
	Smec/Nem	320.3	0.12	0.37		
	Nem/Liq	329.3	7.52	22.84		
		344.6	17.79	51.63		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{25}H_{29}N$	Nem/Liq	370.5	0.29	0.78	76.00	[392]
	Sol/Nem	370.9	23.9	64.44		
	Nem/Liq	416.2	1.16	2.79	67.23	[216]
$C_{25}H_{29}NO$	4-(4-cyanophenyl)-1-(4-octyloxyphenyl)-buta-1E,3E-diene					
	Sol/SmeC	378.5	36.2	95.64		
	SmeC/Nem	432.8	Not reported in paper			
$C_{25}H_{29}NO_2$	Nem/Liq	459.8	0.4	0.87	96.51	[210]
	6-n-pentyloxy-2-[4-propoxystyryl]quinoline					
	Sol/Nem	393.9	30.71	77.96		
$C_{25}H_{29}NO_2$	Nem/Liq	449.3	0.65	1.45	79.41	[112]
	1-[2-( <i>trans</i> -4-pentylcyclohexyl)]-4-[ <i>(4</i> -nitrophenyl)ethynyl]benzene					
	Sol/Nem	408.3	27.24	66.72		
$C_{25}H_{29}NO_2S$	Nem/Liq	497.5	0.88	1.77	68.49	[238]
	4-isothiocyanatophenyl 4-( <i>trans</i> -4-pentylcyclohexyl)benzoate					
	Sol/Sol	339.7	0.79	2.33		
	Sol/SmeC	390.2	22.18	56.84		
$C_{25}H_{30}BrNO_2$	SmeC/Nem	402.2	Too small to be measured			
	Nem/Liq	508.2	1.26	2.48	61.65	NA
	3-(4-decyloxyphenyl)-5-(4-bromophenyl)isoxazole					[356]
$C_{25}H_{30}ClNO_2$	Sol/Sol	350.2	28.7	81.95		
	Sol/SmeC	385.5	12.4	32.17		
	SmeC/Liq	459.9	7.9	17.18	131.30	[138]
$C_{25}H_{30}N_2O_2$	3-(4-decyloxyphenyl)-5-(4-chlorophenyl)isoxazole					
	Sol/Sol	367.7	37.9	103.07		
	Sol/SmeC	370.6	6.3	17.00		
$C_{25}H_{31}BrN_2O$	SmeC/Liq	448.4	Not reported in paper			[138]
	5-methyl-5'-[2-(4-hexyloxyphenyl)-2-hydroxyethyl]-2,2'-bipyridine					
	Sol/SmeC	371.0	21.48	57.90		
$C_{25}H_{31}ClN_2O$	SmeC/Nem	396.1	0.27	0.68		
	Nem/Liq	416.4	3.24	7.78	66.36	[48]
	3-(4-decyloxyphenyl)-5-(4-bromophenyl)pyrazole					
$C_{25}H_{31}NO_5$	Sol/Sol	385.9	32.1	83.18		
	Sol/SmeC	436.5	10.5	24.05		
	SmeC/Liq	497.2	8.2	16.49	123.72	[138]
$C_{25}H_{31}N_2O$	3-(4-decyloxyphenyl)-5-(4-chlorophenyl)pyrazole					
	Sol/Sol	372.2	32.6	87.59		
	Sol/SmeC	427.3	9.8	22.93		
$C_{25}H_{31}NO_5$	SmeC/Liq	494.9	8.2	16.57	127.09	[138]
	4-[ <i>(4</i> -butoxyphenyliminomethyl)-3-hydroxyphenoxy]butylmethacrylate					
	Sol/SmeC	312.0	30.25	96.96		
$C_{25}H_{31}N_3O_2S$	SmeC/SmeC	331.1	Not reported in paper			
	SmeC/Nem	355.7	2.08	5.85		
	Nem/Liq	361.7	4.76	13.16	115.97	[294]
$C_{25}H_{32}N_2O_2S$	5-(4-pentyloxyphenyl)-N-[[4-(2S)-2-methylbutoxy]phenyl)methylene-1,3,4-thiadiazol-2-amine					
	Sol/SmeC	393.9	21.42	54.38		
	SmeC/Nem	402.7	2.24	5.56		
$C_{25}H_{32}N_2O_2S$	Nem/Liq	447.1	0.82	1.83	61.77	[167]
	1-[2-(5-cyanothienyl)]-3-(4-undecyloxyphenylamino)-2-propen-1-one					

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.		
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$				
$C_{25}H_{32}N_2O_3$	Sol/Smec	389.6	27.52	70.64	86.44	[75]		
	Smec/Liq	459.4	7.26	15.80				
$C_{25}H_{32}N_2O_3$		4-pentanoyl-4'-octanoyloxyazobenzene						
	Sol/Sol	333.0	5.23	15.71				
	Sol/Smec	366.1	16.31	44.56				
	Smec/Smec	370.1	7.26	19.61				
	Smec/Liq	403.8	6.94	17.19	97.07	[157]		
$C_{25}H_{32}N_2O_3$		4-[ <i>(4</i> -ethoxyphenyl)azo]phenyl 10-undecenoate						
	Sol/Nem	336.7	44.8	133.06				
	Nem/Liq	382.2	1.63	4.26	137.32	[365]		
$C_{25}H_{32}N_4O_2$		1-[4-[ <i>(1E</i> )-(4-formylphenyl)azo]phenyl]-4-(1-oxooctyl)piperazine						
	Sol/Nem	398.6	19.4	48.67				
	Nem/Liq	431.9	0.3	0.69	49.36	[345]		
$C_{25}H_{32}N_4O_3$		4-[ <i>(1E</i> )-[4-[4-(1-oxohexyl)-1-piperazinyl]phenyl]azo]benzoic acid, ethyl ester						
	Sol/Smec	388.9	6.55	16.84				
	Smec/Smec	422.6	0.13	0.31				
$C_{25}H_{32}O_4$		4-(4-pentenyl)phenyl 4-[ <i>(S</i> )-4-methylhexyloxy]benzoate						
	Sol/Chol	314.2	24.2	77.02				
	Chol/Liq	331.2	Not reported in paper			[200]		
$C_{25}H_{33}FO_3S$		S-(2-fluoro-4-hexyloxyphenyl) 4-hexyloxythiobenzoate						
	Sol/Nem	330.2	38.0	115.08				
$C_{25}H_{33}FO_3S$		Nem/Liq	347.2	1.7	4.90	119.98	NA	[4]
		S-(2-fluoro-4-octyloxyphenyl) 4-butoxythiobenzoate						
	Sol/Nem	321.2	43.1	134.18				
$C_{25}H_{33}N$		Nem/Liq	345.2	1.6	4.63	138.81	NA	[4]
		4-dodecyl-4'-cyanobiphenyl						
	Sol/Smec	318.9	39.71	124.52				
$C_{25}H_{33}N$		Smec/Liq	331.5	4.77	14.39	138.91	140.5	[233]
		Independent values from another reference						
	Sol/Smec	321.0	33.47	104.27				
$C_{25}H_{33}N$		Smec/Liq	332.0	4.06	12.23	116.50	140.5	[237]
		Independent values from another reference						
	Sol/Smec	318.9	34.9	109.44				
$C_{25}H_{33}NO$		Smec/Liq	331.0	4.57	13.81	123.25	140.5	[350]
		4-dodecyloxy-4'-cyanobiphenyl						
	Sol/Smec	340.7	43.18	126.74				
$C_{25}H_{33}NO_5$		Smec/Liq	361.7	4.77	13.19	139.93	147.3	[233]
		4'-dodecyloxy-3'-nitrobiphenyl-4-carboxylic acid						
	Sol/Sol	365.0	3.39	9.28				
$C_{25}H_{33}NO_5$		Sol/Smec	372.0	37.80	101.61			
		Smec/Smec	473.0	0.57	1.21			
		Smec/Liq	482.0	3.37	6.99	119.09	162.6	[87]
$C_{25}H_{33}NO_5$		4-dodecyloxyphenyl 4-nitrobenzoate						
	Sol/Smec	340.2	54.39	159.88				
	Smec/Liq	359.2	3.59	9.99	169.87	155	[236]	
$C_{25}H_{34}ClNO_2S$		1-[2-(5-chlorothienyl)]-3-(4-dodecyloxyphenylamino)-2-propen-1-one						
	Sol/Smec	352.7	35.97	101.98				

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound			$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$			
$C_{25}H_{34}N_2O_2$	Smec/Smec	377.4	1.34	3.55	124.03	[75]	
	Smec/Liq	451.4	8.35	18.50			
$C_{25}H_{34}N_2O_3$	4-nitrobenzylidene 4'-dodecylaniline						
	Sol/Nem	335.2	37.40	111.58			[235]
	Nem/Liq	336.2	2.20	6.54	118.12		
$C_{25}H_{34}N_2O_3$	4-nonanoyloxy-2,3-dimethyl-4'-ethoxyazobenzene						
	Sol/Nem	344.2	43.0	124.93			[339]
	Nem/Liq	358.2	0.7	1.95	126.88		
$C_{25}H_{34}N_2O_3$	4-decanoyloxy-2-methyl-4'-ethoxyazobenzene						
	Sol/Nem	335.2	46.0	137.23			[339]
	Nem/Liq	336.2	1.0	2.97	140.20		
$C_{25}H_{34}N_2O_3$	4-decanoxy-3-methyl-4'-ethoxyazobenzene						
	Sol/Nem	328.2	52.0	158.44			[339]
	Nem/Liq	337.2	0.8	2.37	160.81		
$C_{25}H_{34}N_4OS_2$	2-(4-hexyloxyphenylazo)-5-(5'-heptyl-2'-thienyl)-1,3,4-thiadiazole						
	Sol/Nem	420.9	34.1	81.02			[396]
	Nem/Liq	422.7	0.28	0.66	81.68		
$C_{25}H_{34}O_2S$	4-pentylphenyl 4'-heptyloxythiobenzoate						
	Sol/Sol	183.5	1.17	6.38			[157]
	Sol/Sol	272.0	0.23	0.85			
	Sol/Nem	325.9	28.51	87.48			
$C_{25}H_{34}O_2S$	Nem/Liq	352.2	2.55	7.24	101.95	NA	[217]
	4-pentylbenzenethio 4'-heptyloxybenzoate						
	Sol/Nem	328.2	26.99	82.24			
$C_{25}H_{34}O_3$	Nem/Liq	356.9	1.02	2.86	85.10	NA	[218]
	4-heptyloxyphenyl 4'-pentylbenzoate						
	Sol/Nem	313.7	25.15	80.17			
$C_{25}H_{34}O_3S$	Nem/Liq	330.6	0.90	2.72	82.89		[4]
	S-(4-octyloxyphenyl) 4-hexyloxythiobenzoate						
	Sol/Nem	331.2	35.4	106.88			
$C_{25}H_{35}NO$	Nem/Liq	378.2	1.7	4.49	111.37	NA	[376, 381]
	N-(4-butoxybenzylidene)-4-octylaniline						
	Sol/Smec	306.2	37.2	121.49			
	Smec/Smec	312.5	2.09	6.69			
	Smec/Nem	320.0	0.33	1.03			
$C_{25}H_{35}NO$	Nem/Liq	354.4	1.17	3.30	132.51		[376, 381]
	N-(4-pentyloxybenzylidene)-4-heptylaniline						
	Sol/Sol	285.2	7.3	25.60			
	Sol/Sol	292.2	5.8	19.85			
	Sol/Sol	298.2	1.0	3.35			
	Sol/Smec	302.7	2.7	8.92			
	Smec/Smec	323.5	2.30	0.71			
$C_{25}H_{35}NO$	Smec/Nem	329.5	0.41	1.24			[376, 381]
	Nem/Liq	350.5	1.42	4.05	63.72		
	N-(4-hexyloxybenzylidene)-4-hexylaniline						
	Sol/Smec	288.2	11.9	41.29			
$C_{25}H_{35}NO$	Smec/Smec	325.0	2.72	8.37			[376, 381]
	Smec/Nem	352.9	2.51	7.11			

Note: Temperatures calculated from published enthalpy and entropy data in paper.

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TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound			$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$			
	Nem/Liq	359.4	1.92	5.34	62.11		[376, 381]
Note: Temperatures calculated from published enthalpy and entropy data in paper.							
C <sub>25</sub> H <sub>35</sub> NO							
	Sol/Smec	296.2	13.5	45.58			[381]
	Smec/Liq						
C <sub>25</sub> H <sub>35</sub> NO <sub>2</sub>							
	Sol/Smec	326.2	1.83	5.61			
	Smec/Smec	338.3	1.76	5.20			
	Smec/Smec	362.9	7.40	20.39			
	Smec/Smec	375.2	21.79	58.08			
	Smec/Liq	402.8	4.35	10.80	100.08		[278]
C <sub>25</sub> H <sub>35</sub> NO <sub>2</sub>							
	Sol/Smec	381.2	36.90	96.80			
	Smec/Liq	411.5	4.20	10.21	107.01		[278]
C <sub>25</sub> H <sub>36</sub> F <sub>2</sub>							
	Sol/Smec	300.2	20.4	67.95			
	Smec/Nem	323.2					
	Nem/Liq	392.2					[198]
C <sub>25</sub> H <sub>36</sub> N <sub>2</sub> O							
	Sol/Smec	337.3	91.25	270.53			
	Smec/Nem	341.9	5.09	14.89			
	Nem/Liq	354.0	4.71	13.31	298.73	141.5	[153]
C <sub>25</sub> H <sub>36</sub> N <sub>2</sub> O							
	Sol/Smec	334.6	36.81	110.01			
	Smec/Nem	337.0	1.71	5.07			
	Nem/Liq	346.5	1.02	2.94	118.02	141.5	[141]
C <sub>25</sub> H <sub>36</sub> N <sub>2</sub> O							
	Sol/Smec	332.2	35.55	107.01			
	Smec/Smec	334.3					
	Smec/Nem	337.3	0.91	2.70			
	Nem/Liq	359.0	1.97	5.49		141.5	[141]
C <sub>25</sub> H <sub>36</sub> N <sub>2</sub> O							
	Sol/Nem	322.8	23.05	71.41			
	Nem/Liq	356.0	1.67	4.69	76.10	141.5	[390]
C <sub>25</sub> H <sub>36</sub> N <sub>2</sub> O							
	Sol/Chol	286.2	9.8	34.24			
	Chol/Liq	294.2					[201]
C <sub>25</sub> H <sub>36</sub> N <sub>2</sub> O <sub>2</sub>							
	Sol/Nem	335.4	24.1	71.85			
	Nem/Liq	350.4	1.7	4.85	76.70	146.9	[73]
Independent value from another reference							
	Sol/Nem	345.2	8.86	25.67			
	Nem/Liq	353.2	0.74	2.10	27.77	146.9	[282]
Note: Transition enthalpies seem too low for this compound.							
C <sub>25</sub> H <sub>36</sub> O <sub>2</sub> S							
	Sol/Meso	382.2	21.49	56.23			
	Meso/Liq	400.6	7.68	19.17	75.40	160.2	[18]
C <sub>25</sub> H <sub>37</sub> F <sub>3</sub> O <sub>2</sub>							
	Sol/Smec	317.2	25.5	80.39			

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.		
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$				
$C_{25}H_{37}NS$	Smec/Nem	335.2	Not reported in paper		NA	[198]		
	Nem/Liq	390.2	Not reported in paper					
$C_{25}H_{38}F_2$	Sol/Nem	315.4	26.33	83.48	NA	[52]		
	Nem/Liq	496.3	1.3	2.62				
	Sol/Smec	312.2	30.5	97.69				
$C_{25}H_{38}N_2O_6$	Smec/Nem	342.2	Not reported in paper		[284]	[198]		
	Sol/Smec	483.2	48.0	99.34				
	Smec/Liq	519.2	12.0	23.11				
$C_{25}H_{38}O_2$	Sol/Nem	309.7	22.5	72.65	[5]	[5]		
	Nem/Liq	342.2	0.9	2.63				
	Sol/Nem	339.3	20.1	59.24				
$C_{25}H_{38}O_3$	Nem/Liq	368.5	1.2	3.26	62.50	[5]		
	Sol/Nem	358.2	34.0	94.92	[5]	[5]		
	Nem/Liq	366.0	1.1	3.01	97.93			
$C_{25}H_{39}F$	Sol/Smec	329.2	29.6	89.91	[198]	[198]		
	Smec/Nem	360.2	Not reported in paper					
	Nem/Liq	389.2	Not reported in paper					
	Sol/Nem	324.0	56.7	175.00				
$C_{25}H_{39}NS$	Nem/Liq	326.0	1.3	3.99	179.0	NA		
	Sol/Nem	324.0	4-( <i>trans</i> -4'-dodecylcyclohexyl)isothiocyanatobenzene		[151]	[151]		
	Nem/Liq	326.0	4-( <i>trans</i> -4'-dodecylcyclohexyl)isothiocyanatobenzene					
$C_{25}H_{40}F_2O_2$	Sol/Nem	293.2	20.3	69.24	[198]	[198]		
	Nem/Liq	295.2	Not reported in paper					
	Sol/Smec	297.2	12.1	40.71				
$C_{25}H_{40}O$	Smec/Liq	326.2	Not reported in paper		[198]	[198]		
	Sol/Smec	297.2	1-butoxy-(4-[ <i>E</i> -3-( <i>trans</i> -4-pentylcyclohexyl)-1-butenyl])benzene					
	Smec/Liq	326.2	1-butoxy-(4-[ <i>E</i> -3-( <i>trans</i> -4-pentylcyclohexyl)-1-butenyl])benzene					
$C_{25}H_{40}O_2$	Smec/Nem	314.2	17.2	54.74	[198]	[198]		
	Nem/Liq	319.2	Not reported in paper					
	Nem/Liq	327.2	Not reported in paper					
$C_{25}H_{41}DO_3$	Smec/Liq	374.2	65.70	175.57	[157]	[157]		
	Sol/Smec	402.5	14.40	35.78				
	Sol/Smec	402.5	4-octadecyloxybenzoic acid-d <sub>1</sub>					
$C_{25}H_{42}N_2O_2$	Meso/Liq	399.2	23	57.62	[36]	[36]		
	Sol/Meso	500.2	20	39.98				
	Sol/Meso	500.2	N,N'-dioctanoyl-2,3,5-trimethylbenzene-1,4-diamine					
$C_{25}H_{42}O$	Smec/Liq	302.2	31.4	103.90	[198]	[198]		
	Smec/Liq	316.2	Not reported in paper					
	Sol/Smec	316.2	1-butoxy-4-[ <i>trans</i> -4-pentylcyclohexyl]-1-butyl]benzene					
$C_{25}H_{42}O_2$			1-pentyloxy-4-[3-( <i>trans</i> -4-pentylcyclohexyl)-1-propoxy]benzene					

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{25}H_{42}O_3$	Sol/Smec	315.2	22.2	70.43		
	Smec/Nem	321.2	Not reported in paper			[198]
	Nem/Liq	326.2	Not reported in paper			
$C_{25}H_{50}O_6S$	4-octadecyloxybenzoic acid					
	Sol/Smec	378.2	55.65	147.14		
	Smec/Liq	405.2	12.13	29.94	177.08	211.2
	Independent values from another reference					
	Sol/Smec	379.5	67.30	177.34		
$C_{26}H_{18}F_{16}O_5$	Smec/Liq	408.5	13.30	32.56	209.90	211.2
	6-O-(propylene-[3'-S-hexadecyl])- $\alpha$ -D-galactopyranose					
	Sol/Smec	363.2	53.23	146.56		
$C_{26}H_{18}N_2O$	Smec/Liq	435.2	0.53	1.22	147.78	[39]
	4-(2,2,3,3,4,4,4-heptafluorobutyloxycarbonyl)phenyl 4-[(perfluorobutyl)-butoxy]benzoate					
	Sol/Smec	361.6	43.43	120.11		
$C_{26}H_{18}N_2O_4$	Smec/Liq	368.3	3.83	10.40	130.51	[128]
	2-(p-terphenyl-4-yl)-5-phenyl-1,3,4-oxadiazole					
	Sol/Nem	499.2	41.0	82.13		
$C_{26}H_{18}N_2O_4$	Nem/Liq	526.2	0.1	0.19	82.32	[35]
	1,4-benzenedicarboxylic acid, phenyl, 4-(phenylazo)phenyl ester					
	Sol/Nem	493.2	53.0	107.46		
$C_{26}H_{18}N_2O_4$	Nem/Liq	519.2	0.38	0.73	108.19	[338]
	4-(phenylazo)phenyl 4-(benzoyloxy)benzoate					
	Sol/Nem	453.2	40.0	88.26		
$C_{26}H_{18}N_2O_4$	Nem/Liq	525.2	0.54	1.03	89.29	[338]
	4-(phenoxy carbonyl)phenyl 4-(phenylazo)benzoate					
	Sol/Nem	458.2	49.0	106.94		
$C_{26}H_{18}N_2O_4$	Nem/Liq	527.2	0.27	0.51	107.45	[338]
	4-(benzoyloxy)phenyl 4-(phenylazo)benzoate					
	Sol/Nem	497.2	45.0	90.51		
$C_{26}H_{18}O_4$	Nem/Liq	529.2	0.40	0.76	91.27	[338]
	diphenyl 4,4'-biphenyldicarboxylate					
	Sol/Sol	453.0	4.8	10.60		
$C_{26}H_{18}O_4$	Sol/Nem	485.0	38.4	79.18		
	Nem/Liq	518.6	0.5	0.96	90.74	[12]
	[1,1'-biphenyl]-4-yl 4-(benzoyloxy)benzoate					
$C_{26}H_{18}O_4$	Sol/Nem	483.2	43.0	88.99		
	Nem/Liq	497.2	0.41	0.82	89.81	[338]
	4-(phenoxy carbonyl)phenyl [1,1'-biphenyl]-4-carboxylate					
$C_{26}H_{18}O_4$	Sol/Nem	459.2	45.0	98.00		
	Nem/Liq	501.2	0.29	0.58	98.58	[338]
	4-(benzoyloxy)phenyl [1,1'-biphenyl]-4-carboxylate					
$C_{26}H_{18}O_4$	Sol/Nem	487.2	51.0	104.68		
	Nem/Liq	506.2	0.53	1.05	105.73	[338]
	4-(4-pentyloxyphenyl)acetylene-2,3,5,6-tetrafluorophenyl 4'-chlorobenzoate					
$C_{26}H_{19}ClF_4O_3$	Sol/Nem	364.8	32.99	90.43		
	Nem/Liq	456.4	1.33	2.91	93.34	[101]
$C_{26}H_{19}ClF_4O_3$	4-(4-pentyloxy-2,3,5,6-tetrafluorophenylacetylenyl)phenyl 4'-chlorobenzoate					

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{26}H_{19}NO_2$	Sol/Nem	340.0	19.34	56.88	57.95	[101]
	Nem/Liq	450.2	0.48	1.07		
$C_{26}H_{19}NO_2$	phenyl {[1,1'-biphenyl]-4-ylimino)methyl]benzoate					
	Sol/Nem	472.2	44.0	93.18	94.06	[338]
	Nem/Liq	531.2	0.47	0.88		
$C_{26}H_{19}NO_2$	4-[{[1,1'-biphenyl]-4-ylimino)methyl]phenol, benzoate (ester)					
	Sol/Nem	473.2	47.0	99.32	100.25	[338]
	Nem/Liq	525.2	0.49	0.93		
$C_{26}H_{19}NO_2$	4-[{[1,1'-biphenyl]-ylmethylene)amine]benzoic acid, phenyl ester					
	Sol/Nem	484.2	46.0	95.00	95.83	[338]
	Nem/Liq	518.2	0.43	0.83		
$C_{26}H_{19}NO_2$	4-[{[1,1'-biphenyl]-4-ylmethylene)amino]phenol, benzoate (ester)					
	Sol/Nem	473.2	36.0	76.08	76.69	[338]
	Nem/Liq	524.2	0.32	0.61		
$C_{26}H_{19}NO_2$	[1,1'-biphenyl]-4-yl 4-[(phenylimino)methyl]benzoate					
	Sol/Nem	481.2	44.0	91.44	92.06	[338]
	Nem/Liq	516.2	0.32	0.62		
$C_{26}H_{19}NO_2$	[1,1'-biphenyl]-4-yl 4-[(phenylmethylene)amino]benzoate					
	Sol/Nem	474.2	43.0	90.68	91.40	[338]
	Nem/Liq	499.2	0.36	0.72		
$C_{26}H_{19}NO_2$	4-[(phenylimino)methyl]phenyl [1,1'-biphenyl]-4-carboxylate					
	Sol/Nem	465.2	36.0	77.39	78.28	[338]
	Nem/Liq	515.2	0.46	0.89		
$C_{26}H_{19}NO_2$	4-[(phenylmethylene)amino]phenyl [1,1'-biphenyl]-4-carboxylate					
	Sol/Nem	493.2	44.0	89.21	90.30	[338]
	Nem/Liq	506.2	0.55	1.09		
$C_{26}H_{19}N_3O_2$	phenyl 4-[[4-(phenylazo)phenyl]imino)methyl]benzoate					
	Sol/Nem	478.2	38.0	79.46	79.81	[338]
	Nem/Liq	541.2	0.19	0.35		
$C_{26}H_{19}N_3O_2$	4-[[[4-(phenylazo)phenyl]imino)methyl]phenol, benzoate (ester)					
	Sol/Nem	461.2	40.0	86.73	87.27	[338]
	Nem/Liq	540.2	0.29	0.54		
$C_{26}H_{19}N_3O_2$	4-(phenylazo)phenyl 4-[(phenylimino)methyl]benzoate					
	Sol/Nem	484.2	52.0	107.39	108.03	[338]
	Nem/Liq	529.2	0.34	0.64		
$C_{26}H_{19}N_3O_2$	4-(phenylazo)phenyl 4-[(phenylmethylene)amino]benzoate					
	Sol/Nem	458.2	40.0	87.30	87.97	[338]
	Nem/Liq	522.2	0.35	0.67		
$C_{26}H_{19}N_3O_2$	4-[(phenylimino)methyl]phenyl 4-(phenylazo)benzoate					
	Sol/Nem	463.2	44.0	94.99	95.47	[338]
	Nem/Liq	538.2	0.26	0.48		
$C_{26}H_{19}N_3O_2$	4-[(phenylmethylene)amino]phenyl 4-(phenylazo)benzoate					
	Sol/Nem	479.2	45.0	93.91	94.75	[338]
	Nem/Liq	524.2	0.44	0.84		
$C_{26}H_{20}N_2O_2$	[1,1'-biphenyl]-4-carboxylic acid, 4-[(4-methylphenyl)azo]phenyl ester					
	Sol/Nem	426.6	13.47	31.58	32.29	[57]
	Nem/Liq	534.0	0.38	0.71		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{26}H_{20}N_2O_3$	Sol/Nem	[1,1'-biphenyl]-4-carboxylic acid, 4-[(4-methoxyphenyl)azo]phenyl ester 447.4	15.87	35.47	36.13	[57]
	Nem/Liq	546.7	0.36	0.66		
$C_{26}H_{20}N_2O_6$	Sol/SmeC	bis(4-methoxyphenyl) 2,2'-bipyridine-5,5'-dicarboxylate 493.2	40.4	81.91	Sample decomposed before isotropic liquid	[93]
	SmeC/Liq					
$C_{26}H_{21}F_{17}O_4$	Sol/SmeC	3-[3-(4,4,5,5,6,6,7,7,8,8,9,9,10,10,11,11,11-heptadecafluoro-undecyl)- 4'-hydroxy[1,1'-biphenyl]-4-yl]oxy]-1,2-propanediol 391.2	0.8	2.04	Note: Authors report that sample was not totally crystalline. SmeC/Liq	[404]
		412.2	6.0	14.56		
$C_{26}H_{22}$	Sol/Nem	4'-[2-(4-methylphenyl)-1-ethynyl]-2'-methyl-4-ethyltolane 416.4	30.8	73.97	75.90	[55, 430]
	Nem/Liq	465.6	0.9	1.93		
$C_{26}H_{22}$	Sol/Nem	4'-[2-(4-ethylphenyl)-1-ethynyl]-2'-methyl-4-methyltolane 409.7	29.8	72.74	74.34	[55, 430]
	Nem/Liq	462.8	0.74	1.60		
$C_{26}H_{22}FNO_5S$	Sol/SmeC	4-pentyloxyphenyl 3-fluoro-4-thiocyanatophenyl terephthalate 395.2	23.9	60.48	0.66	[37]
	SmeC/Nem	Not reported in paper				
	Nem/Liq	452.2	0.3	0.66		
$C_{26}H_{22}F_2O$	Sol/Nem	1-(6-hexyloxy-2-naphthyl)-4-(3,4-difluorophenyl)diacetylene 364.7	21.4	58.68	59.40	[95]
	Nem/Liq	415.9	0.3	0.72		
$C_{26}H_{22}N_2O_4$	Sol/Nem	tert-butyl 4-[4-(4-cyanobenzoyloxy)benzylideneamino]benzoate 478.2	26.0	54.37	Not reported in paper	[196]
	Nem/Liq	485.2				
$C_{26}H_{22}N_2O_4$	Sol/Nem	1-methylpropyl 4-[4-(4-cyanobenzoyloxy)benzylideneamino]benzoate 426.2	38.0	89.16	0.21	[196]
	Nem/Liq	486.2	0.1	0.21		
$C_{26}H_{22}N_2O_4$	Sol/Nem	2-methylpropyl 4-[4-(4-cyanobenzoyloxy)benzylideneamino]benzoate 399.2	37.0	92.69	0.18	92.87
	Nem/Liq	549.2	0.1	0.18		
$C_{26}H_{22}O_2$	Sol/Sol	4,4'''-dimethoxy-p-quaterphenyl 604.0	10.27	17.00	Not reported in paper	[111]
	Sol/SmeC	626.0	13.15	21.01		
	SmeC/Liq	665.0				
$C_{26}H_{22}O_5$	Sol/Nem	7-(4'-butoxybenzoyloxy)isoflavone 452.7	34.15	75.44	0.88	76.32
	Nem/Liq	486.7	0.43	0.88		
$C_{26}H_{22}O_6S_2$	Sol/Nem	bis(4-ethoxyphenyl) 2,2'-bithiophene-5,5'-dicarboxylate 491.6	47.4	96.42	98.58	135.2
	Nem/Liq	555.3	1.2	2.16		
$C_{26}H_{22}O_8$	Sol/SmeC	bis(4-ethoxycarbonylphenyl) terephthalate 466.2	43.0	92.24	95.11	124
	SmeC/Nem	477.2	0.9	1.89		
	Nem/Liq	509.2	0.5	0.98		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{26}H_{23}FO$	Sol/Nem	386.3	25.0	64.72		[95]
	Nem/Liq	447.1	0.6	1.34	66.06	
$C_{26}H_{23}NO$	Sol/Nem	541.0	20.0	36.97		[185]
	Nem/Liq			Decomposed prior to transition		
$C_{26}H_{23}NO_5S$	Sol/SmeC	393.2	29.4	74.77		[114]
	SmeC/Nem	435.2	1.3	2.99		
	Nem/Liq	457.2	0.4	0.87	78.63	
$C_{26}H_{23}NO_6$	Sol/SmeC	420.2	31.0	73.77		[196]
	SmeC/Nem	441.2	0.1	0.23		
	Nem/Liq	529.2	0.1	0.19	74.19	
$C_{26}H_{23}NO_6$	Sol/SmeC	405.2	21.0	51.83		[196]
	SmeC/Nem	428.2	0.1	0.23		
	Nem/Liq			Not reported in paper		
$C_{26}H_{24}ClNO_4$	Sol/SmeC	481.2	43.0	89.36		[196]
	SmeC/Nem	492.2	1.0	2.03		
	Nem/Liq	566.2	0.1	0.18	91.57	
$C_{26}H_{24}ClNO_4$	Sol/SmeC	411.2	33.0	80.25		[196]
	SmeC/Nem	438.2	3.0	6.85		
	Nem/Liq	442.2	0.3	0.68	87.78	
$C_{26}H_{24}ClNO_4$	Sol/SmeC	419.2	30.0	71.56		[196]
	SmeC/Nem	427.2	10.0	23.41		
	Nem/Liq	483.2	0.3	0.62	95.59	
$C_{26}H_{24}ClNO_4$	Sol/SmeC	395.2	26.0	65.79		[196]
	SmeC/Liq	457.2	6.3	13.78	79.57	
$C_{26}H_{24}ClNO_4$	Sol/SmeC	391.2	28.0	71.57		[196]
	SmeC/Nem	452.2	2.2	4.87		
	Nem/Liq	485.2	0.3	0.62	77.06	
$C_{26}H_{24}ClNO_4$	Sol/SmeC	421.2	30.0	71.23		[196]
	SmeC/Nem	480.2	4.0	8.33		
	Nem/Liq	488.2	0.3	0.61	80.17	
$C_{26}H_{24}ClNO_4$	Sol/SmeC	418.2	38.0	90.87		[196]
	SmeC/Nem	474.2	2.7	5.69		
	Nem/Liq	494.2	0.4	0.81	97.37	
$C_{26}H_{24}N_2O_4$	Sol/Nem	483.2	31.0	64.16		[192]
	Nem/Liq	568.2	0.21	0.37	64.53	
$C_{26}H_{24}N_2O_6$				1,2-dimethylpropyl 4-[4-(4-nitrobenzoyloxy)benzylideneamino]benzoate		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{26}H_{24}N_2O_6$	Sol/Nem	418.2	32.0	76.52	77.17	[196]
	Nem/Liq	459.2	0.3	0.65		
$C_{26}H_{25}ClO_4S$	Sol/Nem	451.2	41.0	90.87	91.07	[196]
	Nem/Liq	492.2	0.1	0.20		
	Sol/Smec	391.8	35.31	90.12		
$C_{26}H_{25}N$	Smec/Nem	473.5	0.92	1.94	NA	[383]
	Nem/Liq	522.8	0.75	1.43		
	Sol/Nem	387.4	18.5	47.75		
$C_{26}H_{25}NO_3S$	Nem/Liq	476.5	0.41	0.86	48.61	[2]
	Sol/Smec	379.2	32.7	86.23	91.38	[114]
	Smec/Nem	446.2	2.0	4.48		
$C_{26}H_{25}NO_3S$	Nem/Liq	450.2	0.3	0.67		
	Sol/Smec	385.7	71.12	184.39	196.24	[63]
	Smec/Nem	410.8	2.93	7.13		
$C_{26}H_{25}NO_4$	Nem/Liq	442.8	2.09	4.72		
	Sol/Nem	409.2	37.0	90.42	91.02	[196]
	Nem/Liq	502.2	0.3	0.60		
$C_{26}H_{25}NO_4$	Sol/Smec	409.2	37.0	90.42	91.28	[196]
	Smec/Nem	422.2	0.1	0.24		
	Nem/Liq	482.2	0.3	0.62		
$C_{26}H_{25}NO_5$	Sol/Nem	470.2	57.0	121.23	122.14	[195]
	Nem/Liq	551.2	0.5	0.91		
	Sol/Smec	424.2	42.0	99.01		
$C_{26}H_{25}NO_5$	Nem/Liq	457.2	0.1	0.22	99.23	[195]
	Sol/Nem	412.2	39.0	94.61	95.37	[195]
	Nem/Liq	528.2	0.4	0.76		
$C_{26}H_{25}NO_5$	Sol/Nem	444.2	58.0	130.58	131.17	[195]
	Nem/Liq	510.2	0.3	0.59		
	Sol/Smec	417.2	36.0	86.29		
$C_{26}H_{25}NO_5$	Smec/Nem	441.2	1.4	3.17	89.68	[195]
	Nem/Liq	457.2	0.1	0.22		
	Sol/Smec	407.2	41.0	100.69		
$C_{26}H_{25}NO_5$	Smec/Nem	429.2	0.3	0.70	103.12	[195]
	Nem/Liq	519.2	0.9	1.73		
	Sol/Smec	407.2	41.0	100.69		
$C_{26}H_{25}NO_5$	Sol/Smec	407.2	41.0	100.69	butyl 4-[4-(4-methoxybenzylideneamino)benzoyloxy]benzoate	

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated) (estimated)	Ref.	
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$			
$C_{26}H_{25}NO_6$	Sol/Smec	430.2	55.0	127.85	129.97	[195]	
	Smec/Nem	443.2	0.5	1.13			
	Nem/Liq	504.2	0.5	0.99			
$C_{26}H_{25}NO_6$	Sol/Nem	410.2	Value was not reported in paper		2.14	[352]	
	Nem/Liq	420.2	0.9				
	Sol/Nem	356.8	40.0	112.11			
$C_{26}H_{26}F_2O$	Nem/Liq	359.4	0.5	1.39	113.50	[95]	
	Sol/Smec	316.3	17.16	54.25	71.74	[125]	
	Smec/Liq	324.7	5.68	17.49			
$C_{26}H_{26}FNS$	Sol/Sol	336.2	3.81	11.33	NA	[135]	
	Sol/Smec	358.4	18.91	52.76			
	Smec/Nem	385.8	Too small to be measured				
	Nem/Liq	516.4	1.21	2.34	66.43		
$C_{26}H_{26}O_4$	Sol/Sol	290.0	1.24	4.28	111.2	[157]	
	Sol/Nem	405.0	24.70	60.99			
	Nem/Liq	434.0	2.30	5.30	70.57		
$C_{26}H_{27}FO$	Sol/Nem	366.9	31.8	86.67	88.46	[95]	
	Nem/Liq	390.6	0.7	1.79			
	Sol/Smec	338.7	26.83	79.21			
$C_{26}H_{27}F_7O_5$	Smec/Smec	358.4	0.44	1.23	97.66	[125]	
	Smec/Liq	365.2	6.29	17.22			
	Sol/Smec	338.7	26.83	79.21			
$C_{26}H_{27}NO$	Sol/Smec	356.2	19.95	56.01	126.4	[277]	
	Smec/Smec	373.2	4.06	10.88			
	Smec/Nem	442.2	9.61	21.73			
	Nem/Liq	513.2	1.22	2.38	91.00		
$C_{26}H_{27}NO_3$	Sol/Smec	376.2	38.0	101.01	106.26	[292]	
	Smec/Nem	414.2	1.6	3.86			
	Nem/Liq	433.2	0.6	1.39	79.57		
$C_{26}H_{27}NO_3$	Sol/Smec	379.2	29.0	76.48	98.29	[292]	
	Smec/Nem	406.2	0.3	0.74			
	Nem/Liq	511.2	1.2	2.35	79.57		
$C_{26}H_{27}NO_3$	Sol/Nem	413.2	40.0	96.81	98.29	[292]	
	Nem/Liq	474.2	0.7	1.48			
	Sol/Smec	412.2	Value not reported in paper				
$C_{26}H_{27}N_1O_8$	Smec/Nem	438.2	0.1	0.23	4.93	[352]	
	Nem/Liq	446.2	2.2	4.93			
	Sol/Smec	412.2	Value not reported in paper				
$C_{26}H_{28}O_4S$	4'-(7-octenyl)oxy)biphenyl 5-methoxy-2-thiophenecarboxylate						

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.	
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$			
$C_{26}H_{28}$	Sol/Nem	378.9	72.38	191.03	198.67	[63]	
	Nem/Liq	438.3	3.35	7.64			
$C_{26}H_{28}$	4-[ <i>(4-hexen-1-yl)phenyl-1,3-butadiynyl</i> ]-4-butylbenzene					[212]	
	Sol/Nem	328.0	9.70	29.57	31.92		
	Nem/Liq	426.4	1.00	2.35			
$C_{26}H_{28}$	1-( <i>trans-4-ethylcyclohexylethynyl</i> )-4-(4-ethylphenylethynyl)benzene					[436]	
	Sol/Nem	396.3	16.9	42.64	45.28		
	Nem/Liq	454.3	1.2	2.64			
$C_{26}H_{28}N_2O_3$	4-[ <i>(pyridine-4-ylmethylene)amino</i> ]phenyl 4-heptyloxybenzoate					[265]	
	Sol/SmeC	384.2	28.8	74.96	76.24		
	SmeC/Nem	391.0	0.22	0.56			
	Nem/Liq	415.0	0.3	0.72			
$C_{26}H_{28}N_2O_3$	3-pyridyl 4-(4-heptyloxybenzylideneamino)benzoate					[291]	
	Sol/SmeC	360.2	42.0	116.60	123.52		
	SmeC/Nem	432.2	2.4	5.55			
$C_{26}H_{28}N_2O_3$	Nem/Liq	439.2	0.6	1.37	123.52		
	4[(E)-[(4-heptyloxyphenyl)methylene]amino]phenyl 3-pyridinecarboxylate					[291]	
	Sol/Nem	389.2	53.0	136.18	138.00		
$C_{26}H_{28}N_2O_3$	Nem/Liq	440.2	0.8	1.82			
	4-[ <i>(E)-(3-pyridimino)methyl</i> ]phenyl 4-heptyloxybenzoate					[291]	
	Sol/SmeC	367.2	46.0	125.27	127.65		
$C_{26}H_{28}N_2O_3$	SmeC/Nem	390.2	0.3	0.77			
	Nem/Liq	434.2	0.7	1.61	127.65		
$C_{26}H_{28}N_2O_3$	4-[ <i>(E)-(3-pyridinylmethylene)amino</i> ]phenyl 4-heptyloxybenzoate						
	Sol/Nem	370.2	31.0	83.74	85.59	[291]	
	Nem/Liq	432.2	0.8	1.85			
$C_{26}H_{28}O_3$	4-biphenyl 4"-heptyloxybenzoate					[425]	
	Sol/Nem	399.2	39.75	Not reported in paper	NA		
	Nem/Liq	402.7					
Note: Authors report only an enthalpy of fusion, and state in a footnote that Nem/Liq transition enthalpies for the compounds studied were in the 0.84–2.09 kJ·mol <sup>-1</sup> range.							
$C_{26}H_{29}F_3O$	1-[ <i>4-(ω-undecyloxy)phenyl</i> ]-2-(4'-trifluoromethylphenyl)acetylene					[92]	
	Sol/SmeC	340.1	16.97	49.90	88.76		
	SmeC/Liq	351.5	13.66	38.86			
$C_{26}H_{29}NS$	4'-(4-pentylcyclohexyl)-4-isothiocyanatotolane					[135]	
	Sol/SmeC	392.7	26.86	68.40	71.89		
	SmeC/Nem	420.8	0.63	1.50			
	Nem/Liq	528.1	1.05	1.99			
Independent values from another reference							
$C_{26}H_{29}NS$	Sol/SmeC	392.7	26.32	67.02	70.30	NA	
	SmeC/Nem	420.8	0.25	0.59			
	Nem/Liq	528.1	1.42	2.69			
$C_{26}H_{29}NS$	4'-(4-propylcyclohexylethyl)-4-isothiocyanatotolane					[135, 238]	
	Sol/Nem	403.2	26.02	64.53	70.06		
	Nem/Liq	484.6	2.68	5.53			
$C_{26}H_{30}$	4-[ <i>4-hexylphenyl-1,3-butadiynyl</i> ]-4-butylbenzene					[212]	
	Sol/Nem	312.3	10.26	32.85	Not reported in paper		
	Nem/Liq	355.8					

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.	
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$			
$C_{26}H_{31}NO_2$	Sol/Nem	385.7	25.33	65.67	67.50	[112]	
	Nem/Liq	453.3	0.83	1.83			
$C_{26}H_{31}NO_2$	Sol/Nem	376.0	21.97	58.43	63.21	[238]	
	Nem/Liq	455.9	2.18	4.78			
$C_{26}H_{31}NO_2$	Sol/Nem	358.6	11.1	30.95	61.97	[435]	
	Nem/Liq	361.1	11.2	31.02			
$C_{26}H_{31}NO_2S$	4-isothiocyanatophenyl 4-( <i>trans</i> -4-hexylcyclohexyl)benzoate					[356]	
	Sol/Sol	321.5	12.59	39.16	NA		
	Sol/Smec	377.2	24.85	65.88			
	Smec/Nem	419.7	0.21	0.50			
$C_{26}H_{31}N_3O$	Nem/Liq	498.2	2.01	4.03	109.57	[138]	
	3-(4-decyloxyphenyl)-5-(4-cyanophenyl)pyrazole						
	Smec/Smec	433.1	35.5	81.97			
$C_{26}H_{31}N_2O_2$	Smec/Liq	459.2	3.1	6.75	88.72	[138]	
	3-(4-decyloxyphenyl)-5-(4-chlorophenyl)isoxazole						
	Sol/Sol	338.0	0.5	1.48	115.97	[138]	
	Sol/Sol	350.9	7.8	22.23			
$C_{26}H_{32}N_2O$	Sol/Smec	365.0	30.8	84.38	NA	[392]	
	Smec/Smec	456.8	3.6	7.88			
	Nem/Liq	401.1	0.47	1.17	83.46		
$C_{26}H_{32}N_2O_2$	5-(2-methylphenyl)-3-[4-( <i>trans</i> -4-pentylcyclohexyl)phenyl]-1,2,4-oxadiazole					[112]	
	Sol/Sol	317.8	0.31	0.98	100.83		
	Sol/Nem	348.4	28.33	81.31			
	Nem/Liq	414.6	0.47	1.17			
$C_{26}H_{32}O_4$	di(4'-propylphenyl) <i>trans</i> -cyclohexane-1,4-dicarboxylate					[215]	
	Sol/Nem	389.1	34.9	89.69	NA		
	Nem/Liq	446.7	0.86	1.93	91.62		
$C_{26}H_{32}O_6$	di(4'-propoxyphenyl) <i>trans</i> -cyclohexane-1,4-dicarboxylate					[220]	
	Sol/Nem	410.2	53.01	129.23	NA		
	Nem/Liq	487.2	0.44	0.90	130.13		
$C_{26}H_{33}NOS$	3-[4-( <i>trans</i> -4-heptylcyclohexyl)phenyl]-5-(2-thienyl)-isoxazole					[131]	
	Sol/Smec	380.2	27.05	71.15	NA		
	Smec/Nem	403.2	1.32	3.27			
$C_{26}H_{33}NO_2$	Nem/Liq	459.2	1.14	2.48	76.90	NA	
	(E)-4-[2-[4-(11-hydroxyundecyloxy)phenyl]vinyl]benzonitrile						
	Sol/Nem	371.4	14.2	38.23	NA		
$C_{26}H_{33}NO_3$	Nem/Liq	374.4	26.4	70.51	108.74		
	6-hexyloxynaphth-2-yl 5-butoxy-2-methypyridyl ketone					NA	
	Sol/Sol	330.2	15.10	45.73	NA		
$C_{26}H_{33}NO_3$	Sol/Nem	365.2	18.63	51.01	NA		
	Nem/Liq	377.2	3.30	8.75	105.49		
	3-(4-decyloxyphenyl)-5-(4-methoxyphenyl)isoxazole						
$C_{26}H_{33}NO_3$	Sol/Nem	374.8	43.7	116.60	NA	[123]	

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.	
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$			
$C_{26}H_{33}NO_5$	Nem/Liq	425.7	0.6	1.41	118.01	[138]	
	Sol/Smec	314.6	45.00	143.04			
	Smec/Smec	339.8	Not reported in paper				
	Smec/Nem	357.0	2.50	7.00			
	Nem/Liq	360.3	1.54	4.27	154.31		
$C_{26}H_{33}N_2O_2S$	1-[2-(5-cyanothienyl)]-3-(4-dodecyloxyphenylamino)-2-propen-1-one					[294]	
	Sol/Smec	378.4	7.74	20.45			
	Smec/Liq	455.6	7.17	15.74	36.19		
$C_{26}H_{33}N_2O_2S_2$	2,5-bis(4-hexyloxyphenyl)thiazolo[5,4-d]dithiazole					[269]	
	Sol/Sol	392.4	8.60	21.92			
	Sol/Smec	440.6	22.27	50.54			
	Smec/Smec	505.8	4.83	9.55			
	Smec/Liq	535.9	2.09	3.90	85.91		
$C_{26}H_{33}N_3O_2S$	5-(4-hexyloxyphenyl)-N-[[4-(2S)-2-methylbutoxy]phenyl]methylene-1,3,4-thiadiazol-2-amine					[167]	
	Sol/Smec	383.5	18.47	48.16			
	Smec/Nem	402.7	1.73	4.30			
$C_{26}H_{34}N_2O_2$	Nem/Liq	429.8	0.95	2.21	54.67	[138]	
	3-(4-decyloxyphenyl)-5-(4-methoxyphenyl)pyrazole						
	Sol/Smec	379.9	39.0	102.66			
$C_{26}H_{34}N_2O_3$	Smec/Liq	454.1	3.7	8.15	110.81	[157]	
	4-n-pentanoyl-4-n'-nonanoyloxyazobenzene						
	Sol/Sol	352.2	4.33	12.29			
	Sol/Smec	367.2	24.81	67.56			
	Smec/Liq	404.3	7.57	18.72	98.57		
$C_{26}H_{34}O_4$	Smec/Smec	366.8	6.12	16.68		[200]	
	4-[(S)-2-methylbutoxy]phenyl 4-(7-octenoxy)benzoate						
	Sol/Smec	317.2	25.3	79.76			
$C_{26}H_{34}O_4$	Smec/Liq	322.2	Not reported in paper			[200]	
	4-(5-hexenoxy)phenyl 4-[(S)-4-methylhexyloxy]benzoate						
	Sol/Chol	311.2	20.7	66.52			
$C_{26}H_{34}O_4$	Chol/Liq	323.2	Not reported in paper			[200]	
	4-(3-butenoxy)phenyl 4-[(S)-6-methyloctyloxy]benzoate						
	Sol/Chol	316.2	26.8	84.76			
$C_{26}H_{35}ClN_2O_4$	Chol/Liq	325.2	Not reported in paper			[200]	
	4-(4-octyloxyphenylazoxy)phenyl 2S,3S-2-chloro-3-methylpentanoate						
	Sol/Smec	330.5	37.76	114.25			
$C_{26}H_{35}NO_3$	Smec/Nem	333.8	0.33	0.99		[416]	
	Nem/Liq	343.7	1.05	3.05	118.29		
	4-(4-dodecyloxyphenyliminomethyl)benzoic acid						
$C_{26}H_{35}NO_5$	Sol/Smec	428.2	30.8	71.93		[87]	
	Smec/Liq	528.2	35.8	67.78	139.71		
	4'-tridecyloxy-3'-nitrobiphenyl-4-carboxylic acid						
$C_{26}H_{35}N_3OS$	Sol/Smec	392.0	39.83	101.61		[87]	
	Smec/Smec	472.0	0.76	1.61			
	Smec/Liq	480.0	4.32	9.00	112.22		
$C_{26}H_{35}N_3OS$	5-[4-(pentyloxy)phenyl]-N-[(5-octyl-2-thienyl)methylene]-1,2,4-thiadiazole-2-amine					[416]	
	Sol/Smec	387.4	36.4	93.96			
	Smec/Nem	389.0	4.8	1.23			

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated) (estimated)	Ref.		
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$				
$C_{26}H_{35}N_3O_2S$	Nem/Liq	403.7	0.7	1.73	96.92	[396,397]		
	Sol/Nem	354.0	32.6	92.09				
	Nem/Liq	385.9	0.9	2.33	94.42			
$C_{26}H_{36}N_2O_3$	6-n-decyloxy-2-(4-propoxyphenylazo)benzothiazole					[41]		
	Sol/Nem	341.2	44.0	128.96				
	Nem/Liq	354.2	0.6	1.69	130.65			
$C_{26}H_{36}N_2O_3$	4-decanoxyloxy-2,3-methyl-4'-ethoxyazobenzene					[339]		
	Sol/Nem	323.2	49.0	151.61				
	Nem/Liq	340.2	1.0	2.94	154.55			
$C_{26}H_{36}N_4OS_2$	4-undecanoyloxy-3-methyl-4'-ethoxyazobenzene					[339]		
	Sol/Nem	423.8	35.9	84.71				
	Nem/Liq	426.0	0.38	0.89	85.60			
$C_{26}H_{36}O_2S$	4-pentylbenzenethio-4'-octyloxybenzoate					[217]		
	Sol/Smec	332.3	33.10	99.61				
	Smec/Nem	336.3	0.15	0.45				
$C_{26}H_{36}O_3$	Nem/Liq	360.1	1.38	3.83	103.89	NA	[217]	
	4-octyloxyphenyl 4'-pentylbenzoate							
	Sol/Nem	322.2	27.70	85.97				
$C_{26}H_{36}O_3S$	Nem/Liq	333.8	1.04	3.12	89.09		[218]	
	S-(2-methyl-4-octyloxyphenyl) 4-butoxythiobenzoate							
	Sol/Nem	325.2	42.0	129.15				
$C_{26}H_{37}F_3O$	Nem/Liq	326.2	1.7	5.21	134.36	NA	[4]	
	1-[4-(4'-propyl[1,1'-bicyclohexyl]-4-yl)-3-but enyl]-4-(trifluoromethoxy)benzene							
	Smec/Smec	314.2	3.1					
Note: Enthalpy of fusion is significantly small compared with other compounds in series.						[198]		
$C_{26}H_{37}NO$	Smec/Liq	401.2	Not reported in paper					
	N-(4-pentyloxybenzylidene)-4-octylaniline					[376, 381]		
	Sol/Smec	316.4	35.7	112.83				
$C_{26}H_{37}NO$	Smec/Smec	327.5	2.26	6.90				
	Smec/Nem	333.3	0.71	2.13				
	Nem/Liq	345.7	1.17	3.38	125.24			
Note: Temperatures calculated from published enthalpy and entropy data in paper.						[376, 381]		
$C_{26}H_{37}NO$	N-(4-hexyloxybenzylidene)-4-heptylaniline							
	Sol/Smec	300.2	10.3	34.31				
	Smec/Smec	331.8	3.05	9.19				
$C_{26}H_{37}NO$	Smec/Nem	354.2	2.13	6.01				
	Nem/Liq	359.1	2.43	6.77	56.28		[376, 381]	
	Note: Temperatures calculated from published enthalpy and entropy data in paper.							
$C_{26}H_{37}NO$	N-(4-heptyloxybenzylidene)-4-hexylaniline					[381]		
	Sol/Smec	313.6	36.6	116.71				
	Smec/Liq	Not reported in paper						
$C_{26}H_{37}NO$	N-(p-n-octyloxybenzylidene)-4-(n-pentyl)aniline					[11]		
	Sol/Smec	314.7	29.34	93.23				
	Smec/Smec	343.8	3.42	9.95				
$C_{26}H_{38}N_2O$	Smec/Liq	358.7	5.43	15.14	118.32		[11]	
	4-propyl-4'-undecyloxyazobenzene							
	Sol/Smec	343.9	83.72	243.44				
$C_{26}H_{38}N_2O$	Smec/Nem	345.5	5.57	16.12				

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{26}H_{38}N_2O$	Nem/Liq	352.3	4.79	13.60	273.16	148.6 [153]
	Sol/Smec	327.4	26.97	82.38		
	Smec/Smec	337.2	Not reported in paper			
	Smec/Nem	344.1	3.30	9.59		
	Nem/Liq	349.6	1.84	5.26	148.6	[141]
$C_{26}H_{38}N_2O$	4-pentyl-4'-nonyloxyazobenzene					
	Sol/Smec	324.5	25.04	77.16		
	Smec/Smec	338.3	Not reported in paper			
	Smec/Nem	345.6	1.61	4.66		
	Nem/Liq	357.2	2.06	5.77	148.6	[141]
$C_{26}H_{38}N_2O$	4-heptyl-4'-heptyloxyazobenzene					
	Sol/Smec	323.4	28.03	86.67		
	Smec/Nem	325.4	0.34	1.04		
	Nem/Liq	353.0	1.55	4.39	92.10	148.6
						[390]
$C_{26}H_{38}N_2O$	2-[4-(7-octenyoxy)phenyl]-5-[(S)-5-methylheptyl]pyrimidine					
	Sol/Chol	302.2	11.4	37.72		
	Chol/Liq	313.2	Not reported in paper			[201]
$C_{26}H_{38}N_2O$	2-[4-(6-heptenyoxy)phenyl]-5-[(S)-6-methyloctyl]pyrimidine					
	Sol/Smec	280.2	9.7	34.62		
	Smec/Chol	292.2	Not reported in paper			
	Chol/Liq	312.2	Not reported in paper			[201]
$C_{26}H_{38}N_2O_3$	4,4'-bis(heptyloxy)azoxybenzene					
	Sol/Smec	347.8	27.45	78.92		
	Smec/Nem	368.2	0.55	1.49		
	Nem/Liq	397.3	0.83	2.09	82.50	164
						[157]
	Independent values from another reference					
	Sol/Smec	347.6	40.92	117.72		
	Smec/Nem	368.6	1.59	4.31		
	Nem/Liq	397.4	1.02	2.57	124.60	164
						[179]
	Independent values from another reference					
	Sol/Sol	346.0	1.12	3.24		
	Sol/Smec	348.0	41.0	117.82		
	Smec/Nem	368.0	1.13	3.07		
	Nem/Liq	398.0	1.24	3.12	127.25	164
						[440]
$C_{26}H_{38}O_2$	4,4'-bis[2-(S-(+)-2-methylbutoxy)ethoxy]biphenyl					
	Sol/Smec	342.1	26.79	78.31		
	Smec/Liq	345.2	71.47	207.03	285.34	172]
$C_{26}H_{38}O_2S$	2-octanoyl-5-(4-octyloxyphenyl)thiophene					
	Sol/Meso	379.3	18.07	47.64		
	Meso/Liq	399.4	6.32	15.82	63.46	167.3
$C_{26}H_{39}F_3O$	1-(trifluoromethoxy)-4-[4-(4'-propyl[1,1'-bicyclohexyl]-4-yl)butyl]benzene					
	Sol/Smec	320.2	27.2	84.95		
	Smec/Nem	359.2	Not reported in paper			
	Nem/Liq	384.2	Not reported in paper			[198]
$C_{26}H_{39}NS$	4-(trans, trans-4-heptylcyclohexyl)benzene-isothiocyanate					
	Sol/Nem	328.1	33.86	103.20		
	Nem/Liq	491.0	1.67	3.40	106.60	NA
$C_{26}H_{40}O_3$	4-heptylcyclohexyl 4-butoxycinnamate					

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.	
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$			
$C_{26}H_{42}F_2O_2$	Sol/Nem	349.0	24.6	70.49	74.00	[5]	
	Nem/Liq	370.7	1.3	3.51			
$C_{26}H_{42}O$	1-hexyloxy-2,3-difluoro-4-[3-( <i>trans</i> -4-pentylcyclohexyl)propoxybenzene					[198]	
	Sol/Nem	293.2	30.9	105.39			
	Nem/Liq	301.2	Not reported in paper				
$C_{26}H_{42}O$	1-pentyloxy-(4-[( <i>E</i> )-3-( <i>trans</i> -4-pentylcyclohexyl)-1-butenyl])benzene					[198]	
	Sol/Smec	301.2	9.9	32.87			
	Smec/Liq	325.2	Not reported in paper				
$C_{26}H_{42}O$	<i>trans</i> -1-(4-heptanoylphenyl)-4-heptylcyclohexane					[313]	
	Sol/Sol	313.35	4.33	13.81			
	Sol/Smec	343.15	16.47	47.98			
	Smec/Liq	344.65	7.70	22.33	84.13		
$C_{26}H_{42}O_2$	1-hexyloxy-4-[( <i>E</i> )-3-( <i>trans</i> -4-pentylcyclohexyl)allyloxy]benzene					[198]	
	Sol/Smec	323.2	28.4	87.87			
	Smec/Nem	328.2	Not reported in paper				
	Nem/Liq	333.2	Not reported in paper				
$C_{26}H_{44}N_2O_2$	N,N'-dioctanoyl-3,4,5,6-tetramethylbenzene-1,2-diamine					[61]	
	Sol/Meso	391.2	8	20.45			
	Meso/Meso	493.2	2	4.06			
$C_{26}H_{44}N_2O_2$	Meso/Liq	517.2	22	42.54	67.05	[36]	
	N,N'-dioctanoyl-2,3,5,6-tetramethylbenzene-1,4-diamine						
	Sol/Sol	408.2	6.0	14.70			
	Sol/Meso	529.2	16.0	30.23			
	Meso/Meso	538.2	7.0	13.01			
$C_{26}H_{44}O$	Meso/Liq	573.2	23.0	40.13	98.07	[198]	
	1-pentyloxy-4-[4-( <i>trans</i> -4-pentylcyclohexyl)-1-butyl]benzene						
	Sol/Smec	304.2	27.1	89.09			
$C_{26}H_{44}O_2$	Smec/Liq	312.2	Not reported in paper			[198]	
	1-hexyloxy-4-[3-( <i>trans</i> -4-pentylcyclohexyl)-1-propoxy]benzene						
	Sol/Smec	317.2	28.8	90.79			
	Smec/Nem	327.2	Not reported in paper				
$C_{26}H_{74}Si_{10}$	Nem/Liq	330.2	Not reported in paper			[198]	
	1,10-dipropyl(permethyl)decasilane						
	Sol/Meso	295.2	22.1	74.86			
$C_{27}H_{18}O_6$	Meso/Liq	360.2	14.2	39.42	114.28	[110]	
	4-(phenoxy carbonyl)phenyl 4-(benzoyloxy)benzoate						
	Sol/Nem	451.2	44.0	97.52			
$C_{27}H_{18}O_6$	Nem/Liq	527.2	0.59	1.12	98.64	[338]	
	1,4-benzenedicarboxylic acid, 4-(phenoxy carbonyl)phenyl, phenyl ester						
	Sol/Nem	496.2	51.0	102.78			
$C_{27}H_{18}O_6$	Nem/Liq	528.2	0.48	0.91	103.69	[338]	
	4-(benzoyloxy)phenyl 4-(benzoyloxy)benzoate						
	Sol/Nem	499.2	40.0	80.13			
$C_{27}H_{18}O_6$	Nem/Liq	524.2	0.68	1.30	81.43	[338]	
	1,4-benzenedicarboxylic acid, 4-(benzoyloxy)phenyl, phenyl ester						
	Sol/Nem	511.2	51.0	99.77			
$C_{27}H_{18}O_6$	Nem/Liq	526.2	0.39	0.74	100.51	[338]	

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{27}H_{19}NO_4$	Sol/Nem	464.2	39.0	84.02		[338]
	Nem/Liq	541.2	0.27	0.50	84.52	
$C_{27}H_{19}NO_4$	Sol/Nem	462.2	44.0	95.20		[338]
	Nem/Liq	544.2	0.60	1.10	96.30	
$C_{27}H_{19}NO_4$	Sol/Nem	488.2	46.0	94.22		[338]
	Nem/Liq	547.2	0.08	0.15	94.37	
$C_{27}H_{19}NO_4$	Sol/Nem	475.2	40.0	84.18		[338]
	Nem/Liq	541.2	1.2	2.22	86.40	
$C_{27}H_{19}NO_4$	Sol/Nem	465.2	42.0	90.28		[338]
	Nem/Liq	532.2	0.22	0.41	90.69	
$C_{27}H_{19}NO_4$	Sol/Nem	489.2	48.0	98.12		[338]
	Nem/Liq	524.2	0.26	0.50	98.62	
$C_{27}H_{19}NO_4$	Sol/Nem	492.2	56.0	113.77		[338]
	Nem/Liq	532.2	0.32	0.60	114.37	
$C_{27}H_{19}NO_4$	Sol/Nem	484.2	50.0	103.26		[338]
	Nem/Liq	523.2	0.32	0.61	103.87	
$C_{27}H_{19}NO_4$	Sol/Nem	462.2	42.0	90.87		[338]
	Nem/Liq	539.2	0.53	0.98	91.85	
$C_{27}H_{19}NO_4$	Sol/Nem	474.2	37.0	78.03		[338]
	Nem/Liq	520.2	0.31	0.60	78.63	
$C_{27}H_{19}NO_4$	Sol/Nem	482.2	49.0	101.62		[338]
	Nem/Liq	528.2	0.48	0.91	102.53	
$C_{27}H_{19}NO_4$	Sol/Nem	459.2	36.0	78.40		[338]
	Nem/Liq	525.2	0.48	0.91	79.31	
$C_{27}H_{20}F_{16}O_5$		4-(2,2,3,3,4,4,4-heptafluorobutyloxycarbonyl)phenyl 4-[(perfluorobutyl)-pentyloxy]benzoate				
	Sol/SmeC	344.1	33.23	96.57		[128]
	SmeC/SmeC	354.1	0.68	1.92		
$C_{27}H_{20}N_2O_2$	SmeC/Liq	362.8	3.31	9.12	107.61	
	Sol/Nem	464.2	33.0	71.09		[338]
	Nem/Liq	546.2	0.71	1.30	72.39	
$C_{27}H_{20}N_2O_2$	Sol/Nem	469.2	45.0	95.91		[338]
	Nem/Liq	566.2	Not reported in paper			

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{27}H_{20}N_2O_2$	Sol/Nem	451.2	45.0	99.73		[338]
	Nem/Liq	550.2	Not reported in paper			
$C_{27}H_{20}N_2O_2$	Sol/Nem	470.2	43.0	91.45		[338]
	Nem/Liq	529.2	1.1	2.08	93.53	
$C_{27}H_{20}F_{16}O_5$	Sol/SmeC	344.1	33.23	96.57		[128]
	SmeC/SmeC	354.1	0.68	1.92		
	SmeC/Liq	362.8	3.31	9.12	107.61	
$C_{27}H_{21}ClF_4O_3$	Sol/Nem	364.5	34.13	93.64		[101]
	Nem/Liq	446.2	2.22	4.98	98.62	
$C_{27}H_{21}ClF_4O_3$	Sol/Nem	355.5	34.34	96.60		[101]
	Nem/Liq	471.0	0.70	1.49	98.09	
$C_{27}H_{22}ClF_4O_4$	Sol/Nem	354.5	23.24	65.56		[124]
	Nem/Liq	422.0	0.31	0.73	66.29	
$C_{27}H_{22}N_2O_3$	Sol/Nem	329.1	13.75	41.78		[57]
	Nem/Liq	570.5	0.17	0.30	42.08	
$C_{27}H_{24}$	Sol/Nem	412.3	27.8	67.43		[55, 430]
	Nem/Liq	460.6	0.73	1.58	69.01	
$C_{27}H_{24}$	Sol/Nem	417.7	26.4	63.20		[55, 430]
	Nem/Liq	459.6	0.45	0.98	64.18	
$C_{27}H_{24}FNO_5S$	Sol/SmeC	378.2	20.8	55.00		[37]
	SmeC/SmeC	389.2	0.1	0.26		
	SmeC/Nem	393.2	0.1	0.25		
	Nem/Liq	424.2	0.2	0.47	55.98	
$C_{27}H_{24}N_2O_4$	Sol/Nem	414.2	22.0	53.11		[196]
	Nem/Liq	476.2	0.4	0.84	53.95	
$C_{27}H_{24}N_2O_4$	Sol/Nem	423.2	32.0	75.61		[196]
	Nem/Liq	521.2	0.4	0.77	76.38	
$C_{27}H_{24}O_5$	Sol/SmeC	445.0	33.14	74.47		[14]
	SmeC/Nem	450.5	1.36	3.03		
	Nem/Liq	477.8	0.44	0.92	78.42	
$C_{27}H_{25}ClO_8$	Sol/SmeC	359.2	29.0	80.73		[174]
	SmeC/Nem	378.2	0.31	0.82		
	Nem/Liq	414.2	0.16	0.39	81.94	

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{27}H_{25}NO$		5-(4'-propoxybiphenyl-4-yl)-2-(4-methylphenyl)pyridine				
	Sol/Sol	522.0	2.8	5.36		
	Sol/Smec	530.0	6.6	12.45		
	Smec/Smec	549.0	6.4	11.66		
	Smec/Nem	555.0	0.4	0.72		
$C_{27}H_{25}NO_5S$	Nem/Liq	Decomposed prior to transition				
		4-thiocyanophenyl 4-(4-hexyloxybenzoyloxy)benzoate				
$C_{27}H_{25}NO_6$	Smec/Nem	387.2	28.9	74.64		
	Nem/Liq	443.2	1.3	2.93		
		456.2	0.5	1.10	78.67	NA
$C_{27}H_{25}NO_6$	Smec/Nem	417.2	34.0	81.50		
	Nem/Liq	438.2	0.5	1.14		
		500.2	0.3	0.60	83.24	[196]
$C_{27}H_{25}NO_6$	Smec/Nem	406.2	32.0	78.78		
	Nem/Liq	419.2	0.1	0.24		
		548.2	0.3	0.55	79.57	[196]
$C_{27}H_{25}NO_6$	Smec/Nem	424.2	41.0	96.65		
	Nem/Liq	435.2	0.2	0.46		
		529.2	0.4	0.76	97.87	[196]
$C_{27}H_{26}ClNO_4$	Smec/Nem	482.2	54.0	111.99		
	Nem/Liq	492.2	12.0	24.38		
		539.2	0.6	1.11	137.48	[196]
$C_{27}H_{26}ClNO_4$	Smec/Liq	408.2	34.0	83.29		
		447.2	5.9	13.19	96.48	[196]
$C_{27}H_{26}ClNO_4$	Smec/Nem	372.2	31.0	83.29		
	Nem/Liq	443.2	0.9	2.03		
		481.2	0.3	0.62	85.94	[196]
$C_{27}H_{26}ClNO_4$	Smec/Liq	432.2	36.0	83.29		
		471.2	5.9	12.52	95.81	[196]
$C_{27}H_{26}ClNO_4$	Smec/Nem	422.2	41.0	97.11		
	Nem/Liq	471.2	4.4	9.34		
		478.2	0.4	0.84	107.29	[196]
$C_{27}H_{26}N_2O_6$	Smec/Nem	450.2	44.0	97.73		
	Nem/Liq	559.2	Not reported in paper			
						[196]
$C_{27}H_{26}O_8$	Smec/Nem	398.2	38.1	95.68		
	Nem/Liq	481.2	Not reported in paper			
		499.2	4.2	8.41	104.09	[174]
$C_{27}H_{27}ClO_4S$	Smec/Nem	394.1	41.97	106.50		
	Nem/Liq	477.1	1.59	3.33		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta S_{\text{pce}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )				
$C_{27}H_{27}N$	Nem/Liq	513.8	0.67	1.30	111.13	NA	[383]
	Sol/Smec	390.7	19.85	50.81			
	Smec/Nem			Weak transition not detected by dsc			
	Nem/Liq	475.0	0.49	1.03	51.84		[2]
$C_{27}H_{27}NO_3S$							
	Sol/Smec	375.2	40.0	106.61			
	Smec/Liq	449.2	3.6	8.01	114.62		[114]
$C_{27}H_{27}NO_4$							
	Sol/Smec	392.2	26.0	66.29			
	Smec/Nem	409.2	0.9	2.20			
$C_{27}H_{27}NO_4$	Nem/Liq	472.2	0.4	0.85	69.34		[196]
	Sol/Smec	408.2	28.0	68.59			
$C_{27}H_{27}NO_4$	Smec/Nem	437.2	1.9	4.34			
	Nem/Liq	467.2	0.2	0.43	73.36		[196]
$C_{27}H_{27}NO_4$							
	Sol/Smec	412.2	41.0	99.47			
	Smec/Nem	420.2	0.5	1.19			
$C_{27}H_{27}NO_5$	Nem/Liq	479.2	0.5	1.04	101.70		[196]
	Sol/Nem	414.2	37.0	89.33			
$C_{27}H_{27}NO_5$	Nem/Liq	431.2	0.1	0.23	89.56		[195]
$C_{27}H_{27}NO_5$	Sol/Nem	395.2	29.0	73.38			
	Nem/Liq	496.2	0.3	0.60	73.98		[195]
$C_{27}H_{27}NO_5$	Sol/Nem	428.2	46.0	107.43			
	Nem/Liq	486.2	0.5	1.03	108.46		[195]
$C_{27}H_{27}NO_5$	Sol/Nem	466.2	52.0	111.54			
	Nem/Liq	563.2	0.5	0.89	112.43		[195]
$C_{27}H_{27}NO_5$	Sol/Nem	408.2	30.0	73.49			
	Nem/Liq	503.2	0.2	0.40	73.89		[195]
$C_{27}H_{27}NO_5$	Sol/Nem	434.2	54.0	124.37			
	Nem/Liq	510.2	0.5	0.98	125.35		[195]
$C_{27}H_{27}NO_5$							
	Sol/Smec	420.2	27.0	64.26			
	Smec/Nem	435.0	0.7	1.61			
$C_{27}H_{27}NO_5$	Nem/Liq	455.2	0.1	0.22	66.09		[195]
$C_{27}H_{27}NO_5$	Sol/Smec	418.2	39.0	93.26			
	Smec/Nem	437.2	0.7	1.60			
	Nem/Liq	497.2	0.8	1.61	96.47		[195]
$C_{27}H_{27}NO_5$							
	Sol/Smec	418.2	33.0	78.91			

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{27}H_{27}NO_5$	Smec/Liq	426.2	2.6	6.10	85.01	[195]
	Sol/Smec	398.2	31.0	77.85		
	Smec/Liq	442.2	3.3	7.46	85.31	
$C_{27}H_{27}NO_5$	2-methylbutyl 4-[4-(4-methoxybenzylideneamino)benzoyloxy]benzoate					[195]
	Sol/Smec	399.2	34.0	85.17		
	Smec/Nem	436.2	0.5	1.15		
$C_{27}H_{27}NO_5$	Nem/Liq	494.2	0.4	0.81	87.13	[195]
	3-methylbutyl 4-[4-(4-methoxybenzylideneamino)benzoyloxy]benzoate					
	Sol/Smec	418.2	41.0	98.04		
$C_{27}H_{27}NO_5$	Smec/Nem	458.2	1.5	3.27		[195]
	Nem/Liq	482.2	0.5	1.04	102.35	
	pentyl 4-[4-(4-methoxybenzylideneamino)benzoyloxy]benzoate					
$C_{27}H_{27}NO_5$	Sol/Smec	423.2	29.0	68.53		[195]
	Smec/Nem	445.2	0.4	0.90		
	Nem/Liq	501.2	0.8	1.60	71.03	
$C_{27}H_{27}NO_6$	4-[[4-(octyloxy)phenoxy]carbonyl]phenyl 5-cyano-2-furancarboxylate					[352]
	Sol/Smec	408.2	Value not reported in paper			
	Smec/Smec	413.2	Value not reported in paper			
$C_{27}H_{27}NO_6$	Smec/Nem	414.2	3.3	7.97		[352]
	Note: Smec/Smec transition enthalpy was included in the Smec/Nem value.					
	Nem/Liq	420.2	1.0	2.38		
$C_{27}H_{28}F_8O_5$	4-(2,2,3,3,4,4,5,5-octafluoropentyloxycarbonyl)phenyl 4-(octyloxy)benzoate					[125]
	Sol/Smec	330.5	21.61	65.39		
	Smec/Liq	336.6	5.20	15.45	80.84	
$C_{27}H_{29}F_7O_5$	4-(2,2,3,3,4,4,4-heptafluorobutyloxycarbonyl)phenyl 4-(nonyloxy)benzoate					[125]
	Sol/Smec	337.8	17.72	52.46		
	Smec/Smec	350.9	0.45	1.28		
$C_{27}H_{29}N$	Smec/Liq	357.5	6.28	17.57	74.31	[125]
	4-octyl-4"-cyano-p-terphenyl					
	Sol/Meso	355.2	16.33	45.97		
$C_{27}H_{29}NO_2$	Note: Transition enthalpy includes a Sol/Sol transition.					[382]
	Meso/Meso	391.7	5.55	14.17		
	Smec/Nem	470.7	0.18	0.38		
$C_{27}H_{29}NO_2$	Nem/Liq	486.9	0.95	1.95	62.47	[8]
	4'-cyano[1,1'-biphenyl]-4-yl 8-propyltricyclo[4.4.0.03,8]deccane-1-carboxylate					
	Sol/Chol	418.2	26.15	62.53		
$C_{27}H_{29}NO_2$	Chol/Liq	470.2	Not reported in paper			[382]
	Note: Authors report only a single transition enthalpy, which we have assumed is for the Sol/Chol transition.					
$C_{27}H_{29}NO_3$	phenyl 4-(4-heptyloxybenzylideneamino)benzoate					[292]
	Sol/Smec	361.2	41.0	113.51		
	Smec/Nem	417.2	2.1	5.03		
$C_{27}H_{29}NO_3$	Nem/Liq	429.2	0.6	1.40	119.94	[292]
	4-ethylphenyl 4-(4-pentyloxybenzylideneamino)benzoate					
	Sol/Smec	362.2	22.0	60.74		
$C_{27}H_{29}NO_3$	Smec/Nem	418.2	0.5	1.20		[292]
	Nem/Liq	498.2	1.2	2.41	54.35	
	4-isopropylphenyl 4-(4-butoxybenzylideneamino)benzoate					
$C_{27}H_{29}NO_3$	Sol/Smec	406.2	27.0	66.47		
	Smec/Nem	426.2	0.4	0.94		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated) (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{27}H_{30}$	Nem/Liq	473.2	0.7	1.48	68.89	[292]
	Sol/Nem	352.6	17.29	49.04		
	Nem/Liq	429.1	1.66	3.87	52.91	[212]
$C_{27}H_{30}$	Sol/Nem	345.8	11.85	34.27		
	Nem/Liq	433.0	1.13	2.61	36.88	[212]
	Sol/Nem	388.0	15.3	39.43		
$C_{27}H_{30}$	Nem/Liq	461.8	1.3	2.82	42.26	[436]
	Sol/Nem	405.4	14.1	34.78		
	Nem/Liq	481.2	0.9	1.87	36.65	[436]
$C_{27}H_{30}N_2O_3$	Sol/SmeC	372.2	26.7	71.74		
	SmeC/Nem	398.7	0.17	0.43		
	Nem/Liq	425.5	0.4	0.94	73.11	[265]
$C_{27}H_{30}N_2O_3$	Sol/SmeC	364.2	33.0	90.61		
	SmeC/Nem	434.2	3.3	7.60		
	Nem/Liq	438.2	1.0	2.28	100.49	[291]
$C_{27}H_{30}N_2O_3$	Sol/SmeC	386.2	53.0	137.23		
	SmeC/Nem	397.2	0.6	1.51		
	Nem/Liq	438.2	0.8	1.83	140.57	[291]
$C_{27}H_{30}N_2O_3$	Sol/SmeC	364.2	35.0	96.10		
	SmeC/Nem	400.2	0.4	1.00		
	Nem/Liq	433.2	0.8	1.85	98.95	[291]
$C_{27}H_{30}N_2O_3$	Sol/SmeC	352.2	39.0	110.73		
	SmeC/Nem	381.2	0.3	0.79		
	Nem/Liq	431.2	1.0	2.32	113.84	[291]
$C_{27}H_{30}O_3$	Sol/Nem	392.2	37.66			
	Nem/Liq	403.7	Not reported in paper			[425]
	Note: Authors report only an enthalpy of fusion, and state in a footnote that Nem/Liq transition enthalpies for the compounds studied were in the 0.84–2.09 kJ·mol <sup>-1</sup> range.					
$C_{27}H_{31}NS$	$4'-(4\text{-butylcyclohexylethyl})-4\text{-isothiocyanatotolane}$					
	Sol/Sol	343.6	17.41	50.67		
	Sol/Nem	375.5	22.97	61.17		
	Nem/Liq	480.2	2.18	4.54	116.38	NA
	Independent values from another reference					
	Sol/Nem	374.0	22.64	60.53		
	Nem/Liq	478.9	2.59	5.41	65.94	NA
	Note: The compound might exhibit a Sol/Sol transition as indicated in the data set immediately above.					
	Sol/Nem	333.2	14.62	43.88		
$C_{27}H_{32}$	Nem/Liq	371.0	Not reported in paper			[212]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{27}H_{32}F_4O$						[72]
	Sol/Smec	367.2	20.00	54.47		
	Smec/Smec	383.2	1.21	3.16		
$C_{27}H_{33}ClO_5$	Smec/Liq	434.2	4.58	10.55	68.18	[257]
	Sol/Smec	354.7	25.15	70.90		
$C_{27}H_{33}ClO_5$	Smec/Liq	366.7	4.02	10.96	81.86	[257]
	Sol/Smec	354.2	25.10	70.86		
$C_{27}H_{33}F_3O$	Smec/Chol	363.7	0.54	1.48		[257]
	Chol/Liq	371.7	0.59	1.59	73.93	
$C_{27}H_{33}NO_2$						[72]
	Sol/Smec	364.2	17.52	48.11		
	Smec/Smec	440.2	2.15	4.88		
$C_{27}H_{33}NO_2$	Smec/Liq	482.2	7.79	16.16	69.15	[112]
	Sol/Sol	367.1	8.64	23.54		
$C_{27}H_{33}NO_2S$	Sol/Nem	389.8	18.10	46.43		[238]
	Nem/Liq	445.3	0.47	1.06	71.03	
$C_{27}H_{34}F_4O$						[356]
	Sol/Smec	349.2	19.33	55.36		
	Smec/Smec	392.2	1.44	3.67		
$C_{27}H_{34}F_4O$	Smec/Nem	440.2	4.73	10.75		[72]
	Nem/Liq					
$C_{27}H_{33}NO_3$						[117]
	Sol/Smec	371.2	18.5	49.84		
	Smec/Nem	395.3	0.6	1.52		
$C_{27}H_{34}N_2O_2$	Nem/Liq	425.3	1.1	2.59	53.95	[48]
	Sol/Smec	353.2	10.46	29.61		
$C_{27}H_{35}F_3O$	Smec/Nem	411.5	1.84	4.47		[72]
	Nem/Liq	420.6	5.36	12.74	46.82	
$C_{27}H_{35}NO_5$						[294]
	Sol/Smec	303.5	45.66	150.44		
$C_{27}H_{35}NO_5$	Smec/Smec	331.9	Not reported in paper			
	Smec/Liq	361.0	5.12	14.18		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated) (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{27}H_{35}N_3O_2S$			5-(4-heptyloxyphenyl)-N-[[4-(2S)-2-methylbutoxy]phenyl]-methylene-1,3,4-thiadiazol-2-amine			
	Sol/SmeC	384.4	19.29	50.18		
	SmeC/Nem	432.9	1.96	4.53		
$C_{27}H_{35}N_3O_2S$	Nem/Liq	445.1	1.38	3.10	57.81	[167]
			N-[[4-(butoxy)phenyl]methylene]-5-[4-(octyloxy)phenyl]-1,2,4-thiadiazol-2-amine			
	Sol/SmeC	373.3	33.82	90.60		
$C_{27}H_{35}N_3O_2S$	SmeC/Nem	423.1	1.16	2.74		
	Nem/Liq	484.1	1.23	2.54	95.88	[283]
			2-[(E)-[5-[4-(butoxy)phenyl]-1,3,4-thiadiazol-2-yl]imino]methyl]-5-(octyloxy)phenol			
$C_{27}H_{36}$	Sol/SmeC	381.2	27.22	71.41		
	SmeC/Nem	458.7	2.66	5.80		
	Nem/Liq	496.3	2.25	4.53	81.74	[283]
$C_{27}H_{36}N_2O_2S$			1-(trans-4-ethylcyclohexylethynyl)-4-(trans-propylcyclohexylethynyl)benzene			
	Sol/Nem	408.8	15.8	38.65		
	Nem/Liq	441.8	1.4	3.17	41.82	[436]
$C_{27}H_{36}N_2O_2S$	Sol/SmeC	395.6	3.65	9.23		
	SmeC/Liq	457.6	8.01	17.50	26.73	[75]
			1-[2-(5-cyanothienyl)]-3-(4-tridecyloxyphenylamino)-2-propen-1-one			
$C_{27}H_{36}N_2O_2S$	Sol/Sol	348.2	Not given in paper			
	Sol/Nem	352.9	25.9	73.39		
	Nem/Liq	376.4	0.5	1.33		[41]
$C_{27}H_{36}N_2O_3$			6-n-decyloxy-2-(4-propoxybenzylidenamino)benzothiazole			
	Sol/Sol	338.2	4.31	12.74		
	Sol/SmeC	368.6	28.75	78.00		
$C_{27}H_{36}N_2O_3$	SmeC/Liq	405.2	8.00	19.74	110.48	151.2
			4-propanoyl-4'-dodecanoyloxyazobenzene			
	Sol/SmeC	373.7	38.07	101.87		
$C_{27}H_{36}N_2O_3$	SmeC/Liq	416.2	7.87	18.91	120.78	151.2
			1-[4-[(1E)-(4-formylphenyl)azo]phenyl]-4-(1-oxodecyl)piperazine			
	Sol/Nem	422.5	21.4	50.65		
$C_{27}H_{36}N_4O_2$	Nem/Liq	437.9	0.3	0.69	51.34	[345]
			4-[(E)-but-2-enyloxy]-4'-(trans-4-pentylcyclohexyl)biphenyl			
	Sol/SmeC	354.2	25.53	72.08		
$C_{27}H_{36}O$	SmeC/SmeC	433.2	2.64	6.09		
	SmeC/Nem	463.2	1.85	3.99		
	Nem/Liq	481.2	0.45	0.94	83.10	[72]
$C_{27}H_{36}O_3$			4-(trans-4-propylcyclohexyl)phenyl 4-[(S)-2-methylbutyl]oxybenzoate			
	Sol/Chol	386.2	22.02	57.02		
	Chol/Liq	441.2	Not reported in paper			[208]
$C_{27}H_{36}O_4$			4-[(S)-2-methylbutoxy]phenyl 4-(8-nonenyloxy)benzoate			
	Sol/SmeC	309.2	25.3	81.82		
	SmeC/Liq	331.2	Not reported in paper			[200]
$C_{27}H_{36}O_4$			4-[(S)-4-methylhexyloxy]phenyl 4-(6-heptenoxy)benzoate			
	Sol/SmeC	337.2	23.9	70.88		
	SmeC/Chol	338.2	Not reported in paper			
	Chol/Liq	342.2	Not reported in paper			[200]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{27}H_{36}O_4$	Sol/Smec	321.2	20.0	62.27		[200]
	Smec/Chol	327.2		Not reported in paper		
	Chol/Liq	339.2		Not reported in paper		
$C_{27}H_{36}O_4$	Sol/Smec	306.2	16.5	53.89		[200]
	Smec/Chol	310.2		Not reported in paper		
	Chol/Liq	334.2		Not reported in paper		
$C_{27}H_{37}ClN_2O_4$		4-(4-nonyloxyphenylazoxy)phenyl 2S,3S-2-chloro-3-methylpentanoate				
	Sol/Smec	333.7	37.66	112.86		
	Smec/Nem	340.5	0.24	0.70		
$C_{27}H_{37}ClO_2S$	Nem/Liq	346.3	1.27	3.67	117.23	[47]
		4-chlorophenyl 4-tetradecyloxythiobenzoate				
	Sol/Sol	340.5	2.38	6.99		
$C_{27}H_{37}FO$	Sol/Smec	354.5	50.67	142.93		
	Smec/Liq	369.0	5.98	16.21	166.13	NA
	Nem/Liq	444.2	0.79	1.78	40.14	[72]
$C_{27}H_{37}FO_3S$		4-butoxy-3-fluoro-4'-( <i>trans</i> -4-pentylcyclohexyl)biphenyl				
	Sol/Smec	327.2	9.04	27.63		
	Smec/Smec	387.2	1.39	3.59		
$C_{27}H_{37}N$	Smec/Nem	389.2	2.78	7.14		
	Nem/Liq	444.2	0.79	1.78	40.14	[72]
		4-cyano-4'-tetradecylbiphenyl				
$C_{27}H_{37}NO_5$	Sol/Smec	328.4	41.7	126.98		
	Smec/Liq	335.1	5.72	17.07	144.05	[351]
		Independent values from another reference				
$C_{27}H_{37}N_3OS$	Sol/Smec	331.4	41.4	124.92		[407]
	Smec/Liq	338.0	5.4	15.98	140.90	
		4'-tetradecyloxy-3'-nitrobiphenyl-4-carboxylic acid				
$C_{27}H_{37}NO_5$	Sol/Smec	392.0	36.50	93.11		
	Smec/Smec	470.0	0.80	1.70		
	Smec/Liq	479.0	5.75	12.00	106.81	175.8
$C_{27}H_{37}N_3OS$		5-[4-(hexyloxy)phenyl]-N-[5-octyl-2-thienyl)methylene]-1,2,4-thiadiazole-2-amine				
	Sol/Smec	390.1	21.9	56.14		
	Smec/Nem	400.8	3.25	8.11		
$C_{27}H_{37}N_3O_2S$	Nem/Liq	409.6	1.1	2.69	66.94	[396,397]
		6-n-decyloxy-2-(4-butyloxyphenylazo)benzothiazole				
	Sol/Nem	357.0	50.7	142.02		
$C_{27}H_{38}N_2O_3$	Nem/Liq	395.1	0.8	2.02	144.04	[41]
		4-undecanoyloxy-2,3-dimethyl-4'-ethoxyazobenzene				
	Sol/Nem	340.2	39	114.64		
$C_{27}H_{38}N_2O_3$	Nem/Liq	355.2	0.7	1.97	116.61	[339]
		4-dodecanoyloxy-3-methyl-4'-ethoxyazobenzene				
	Sol/Nem	325.2	49	150.68		
$C_{27}H_{38}N_2O_3$	Nem/Liq	337.2	0.8	2.37	153.05	[339]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.		
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$				
$C_{27}H_{38}N_4OS_2$	Sol/Nem	425.1	36.1	84.92	85.89	[396]		
	Nem/Liq	430.8	0.42	0.97				
$C_{27}H_{38}O$	4-butoxy-4'-( <i>trans</i> -4-pentylcyclohexyl)biphenyl					[72]		
	Sol/Smec	337.2	27.03	80.16	95.32			
	Smec/Smec	449.2	3.48	7.45				
	Smec/Nem	464.2	2.80	6.03				
$C_{27}H_{38}O_2S$	Nem/Liq	475.2	0.80	1.68	NA	[217]		
	4-pentylbenzenethio-4'-nonyloxybenzoate							
	Sol/Smec	336.7	35.65	105.88				
	Smec/Nem	347.4	0.38	1.09				
$C_{27}H_{38}O_3S$	Nem/Liq	359.2	1.46	4.06	111.03	NA		
	S-(4-octyloxyphenyl) 4-octyloxythiobenzoate							
	Sol/Nem	332.2	40.5	121.91	128.84			
	Nem/Liq	375.2	2.6	6.93				
$C_{27}H_{38}O_4$	4,4'-bis( $\omega$ -hydroxyhexyloxy)- $\alpha$ -methylstilbene					[56]		
	Sol/Sol	350.8	17.43	49.7	Not reported in paper			
	Smec/Smec	406.5	33.33	82.0				
	Smec/Nem	411.0	8.55	20.8				
$C_{27}H_{39}NO$	Nem/Liq	413.7	Not reported in paper		148.73	[242]		
	4-decyl-N-[[(4-(butoxy)phenyl)methylene]benzenamine							
	Smec/Smec	329.6	48.17	146.15				
	Nem/Liq	349.1	0.9	2.58				
$C_{27}H_{39}NO$	4-hexyl-N-[[(4-(octyloxy)phenyl)methylene]benzenamine					[242]		
	Smec/Smec	317.6	40.64	127.96	158.56			
	Smec/Smec	338.1	0.06	0.18				
	Smec/Smec	346.1	4.07	11.76				
$C_{27}H_{39}NO$	Smec/Liq	359.0	6.70	18.66	NA	[243]		
	4-nonyl-N-[[(4-(pentyloxy)phenyl)methylene]benzenamine							
	Smec/Smec	311.2	10.44	33.55	48.52			
	Smec/Smec	319.3	2.73	8.55				
$C_{27}H_{39}NO$	Smec/Nem	336.4	0.53	1.58	112.26	[243]		
	Nem/Liq	347.1	1.68	4.84				
	4-pentyl-N-[[(4-(nonyloxy)phenyl)methylene]benzenamine							
	Smec/Smec	307.0	25.75	83.88				
$C_{27}H_{39}NO$	Smec/Smec	335.2	0.04	0.12	117.26	[376, 381]		
	Smec/Smec	346.0	6.31	18.24				
	Smec/Liq	359.4	3.60	10.02				
$C_{27}H_{39}NO$	N-(4-hexyloxybenzylidene)-4-octylaniline							
	Smec/Smec	268.2	6.4	23.86	NA	[376, 381]		
	Smec/Smec	302.2	26.6	88.02				
	Smec/Nem	320.0	0.67	2.09				
	Nem/Liq	352.1	0.42	1.19				
$C_{27}H_{40}N_2O$	Nem/Liq	356.4	0.75	2.10	117.26			
	4-propyl-4'-dodecyloxyazobenzene					[153]		
	Smec/Smec	342.5	105.3	307.45	155.7			
	Smec/Nem	350.3	10.50	29.97				
$C_{27}H_{40}N_2O$	Nem/Liq	354.0	6.01	16.98	334.93			
	4-butyl-4'-undecyloxyazobenzene					NA		
	Smec/Smec	338.8	34.39	101.51	Not reported in paper			
	Smec/Smec	340.4	Not reported in paper					

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound			$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$			
$C_{27}H_{40}N_2O$	Smec/Nem	347.6	4.49	12.92			
	Nem/Liq	349.5	2.25	6.44		155.7	[141]
$C_{27}H_{40}N_2O$	4-pentyl-4'-decyloxyazobenzene						
	Sol/Smec	325.1	23.96	73.70			
	Smec/Smec	343.0	Not given in paper				
	Smec/Nem	352.6	2.60	7.37			
	Nem/Liq	359.0	2.51	6.99		155.7	[141]
$C_{27}H_{40}N_2O$	4-heptyl-4'-octyloxyazobenzene						
	Sol/Smec	391.3	37.45	95.71			
	Smec/Nem	344.2	1.01	2.93			
	Nem/Liq	356.9	2.36	6.61	105.25	155.7	[390]
$C_{27}H_{40}N_2O$	2-[4-(8-nonenyloxy)phenyl]-5-[S-5-methylheptyl]pyrimidine						
	Sol/Smec	289.2	10.4	35.96			
	Smec/Chol	308.2	Not reported in paper				
	Chol/Liq	321.2	Not reported in paper				[201]
$C_{27}H_{40}O_2S$	2-octanoyl-5-(4-nonyloxyphenyl)thiophene						
	Sol/Meso	378.3	21.31	56.33			
	Meso/Liq	397.0	7.03	17.71	74.04	174.4	[18]
$C_{27}H_{42}N_2O_6$	4,4'-bis[4-(4-propoxybenzylideneamino)benzoyloxy]diphenylmethane						
	Sol/Smec	472.2	53.0	112.24			
	Smec/Liq	515.2	14.0	27.17	139.41		[284]
$C_{27}H_{42}O_3$	4-heptylcyclohexyl 4-pentyloxycinnamate						
	Sol/Nem	334.2	47.2	141.23			
	Nem/Liq	365.7	1.4	3.83	145.06		[5]
$C_{27}H_{44}O$	1-hexyloxy-(4-[ <i>E</i> -3-( <i>trans</i> -4-pentylcyclohexyl)-1-butenyl])benzene						
	Sol/Smec	305.2	32.9	107.80			
	Smec/Liq	329.2	Not reported in paper			145.3	[198]
$C_{27}H_{44}O_2$	1-heptyloxy-4-[ <i>E</i> -3-( <i>trans</i> -4-pentylcyclohexyl)allyloxy]benzene						
	Sol/Smec	315.2	32.0	101.52			
	Smec/Nem	330.2	Not reported in paper				
	Nem/Liq	331.2	Not reported in paper			163	[198]
$C_{27}H_{45}N_3O_3$	N,N',N''-trihexanoyl-2,4,6-trimethyl-1,3,5-benzenetriamine						
	Sol/Meso	573.2	22.0	38.38			
	Meso/Liq	630.2	30.0	47.60	85.98	144	[190]
$C_{27}H_{46}N_2O_2$	N,N'-dinonanoyl-2,3,5-trimethylbenzene-1,4-diamine						
	Sol/Meso	410.2	31	75.57			
	Meso/Liq	487.2	19	39.00	114.57	154	[36]
$C_{27}H_{46}O_2$	1-heptyloxy-4-[3-( <i>trans</i> -4-pentylcyclohexyl)-1-propoxy]benzene						
	Sol/Smec	324.2	42.0	129.55			
	Smec/Liq	330.2	Not reported in paper				[198]
$C_{27}H_{46}O_6$	3'-pentadecylphenyl- $\beta$ -D-glucopyranoside						
	Sol/Meso	415.9	Not reported in paper				
	Meso/Liq	418.2	22.22	53.13	53.13		[290]
Note: Reported enthalpy is a combination of the Sol/Meso and Meso/Liq values.							
$C_{27}H_{54}O_6S$	6-O-(propylene-[3'-S-octadecyl])- $\alpha$ -D-galactopyranose						
	Sol/Smec	377.2	70.54	187.01			
	Smec/Liq	427.2	0.41	0.96	187.97		[39]
$C_{28}H_{14}F_4N_2O_5$	2,3-difluorobenzoic acid, 1,2,4-oxadiazole-3,5-diyl-4,1-phenylene ester						

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$\text{C}_{28}\text{H}_{16}\text{F}_2\text{N}_2\text{O}_5$	Sol/Nem	459.9	44.01	95.69	96.13	[264]
	Nem/Liq	475.7	0.21	0.44		
$\text{C}_{28}\text{H}_{16}\text{F}_2\text{N}_2\text{O}_5$	Sol/Nem	489.2	22.37	45.73	Decomposed prior to transition	[264]
	Nem/Liq					
	Sol/Smec	371.6	46.32	124.65		
$\text{C}_{28}\text{H}_{18}\text{F}_{20}\text{O}_5$	Smec/Liq	384.5	4.29	11.16	135.81	[128]
	Sol/Nem	654.2	22.0	33.63	34.63	[35]
	Nem/Liq	698.2	0.70	1.00		
$\text{C}_{28}\text{H}_{20}\text{S}$	Sol/Smec	562.2	22.0	39.13	45.01	[35]
	Smec/Nem	589.2	3.0	5.09		
	Nem/Liq	629.2	0.5	0.79		
$\text{C}_{28}\text{H}_{21}\text{F}_{17}\text{O}_5$	Sol/Smec	342.7	40.09	116.98	126.70	[128]
	Smec/Liq	346.6	3.37	9.72		
	Sol/Smec	334.2	9.3	27.83	34.16	[221]
$\text{C}_{28}\text{H}_{21}\text{F}_{17}\text{O}_8$	Smec/Smec	361.2	1.6	4.43		
	Smec/Liq	369.2	0.7	1.90		
	Sol/Smec	408.3	37.2	91.11		
Note: Sol/Smec transition enthalpy may include Sol/Sol transition(s).						
$\text{C}_{28}\text{H}_{21}\text{F}_{21}\text{O}_4$	Smec/Smec	425.7	2.0	4.70	99.77	[404]
	Smec/Liq	429.2	1.7	3.96		
	Sol/Smec					
$\text{C}_{28}\text{H}_{22}\text{O}_6$	Sol/Sol	455.6	3.3	7.24	Not reported in paper	[12]
	Sol/Smec	466.3	28.7	61.55		
	Smec/Nem	504.5	0.03	0.06		
	Nem/Liq					
$\text{C}_{28}\text{H}_{23}\text{ClF}_4\text{O}_3$	Sol/Nem	382.8	33.71	88.06	90.83	[101]
	Nem/Liq	448.4	1.24	2.77		
	Sol/Nem	353.8	41.59	117.55		
$\text{C}_{28}\text{H}_{23}\text{ClF}_4\text{O}_3$	Nem/Liq	458.4	0.99	2.16	119.71	[101]
	Sol/Nem					
	Sol/Smec					
$\text{C}_{28}\text{H}_{23}\text{F}_{15}\text{O}_6$	Sol/Smec	371.2	34.9	94.02	106.22	[221]
	Smec/Liq	385.2	4.7	12.20		
	Sol/Nem					
$\text{C}_{28}\text{H}_{23}\text{NO}$	Sol/Nem	472.2	18.8	39.82	42.45	[82]
	Nem/Liq	570.2	1.5	2.63		
	Sol/Nem					
$\text{C}_{28}\text{H}_{24}\text{F}_4\text{O}_4$	Sol/Nem	365.0	31.37	85.95	87.00	[24]
	Nem/Liq	440.0	0.46	1.05		
	Sol/Nem					

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{28}H_{24}N_6O_6$	Sol/Nem	$\alpha,\omega$ -bis(4-nitroazobenzene-4'-oxy)butane	524.2	50.99	97.27	[422]
	Nem/Liq		526.2	2.89	5.49	
$C_{28}H_{25}F_{13}O_5$	Sol/SmeC	4-(3,3,4,4,5,5,6,6,7,7,8,8,8-tridecafluoroctyloxycarbonyl)phenyl 4-(hexyloxy)benzoate	356.6	30.55	85.68	[125]
	SmeC/SmeC		382.6	0.07	0.18	
	SmeC/Liq		414.4	9.36	22.59	
$C_{28}H_{26}$	Sol/Nem	4'-[2-(4-methylphenyl)-1-ethynyl]-2'-methyl-4-butyltolane	393.2	22.3	56.71	[55, 430]
	Nem/Liq		457.8	0.90	1.97	
$C_{28}H_{26}$	Sol/Nem	4'-[2-(4-propylphenyl)-1-ethynyl]-2'-methyl-4-ethyltolane	388.5	22.5	57.92	[55, 430]
	Nem/Liq		463.2	1.01	2.18	
$C_{28}H_{26}FNO_5S$	4-heptyloxyphenyl 3-fluoro-4-thiocyanatophenyl terephthalate					[37]
	Sol/SmeC	374.2	21.3	56.92		
	SmeC/SmeC	390.2	Too small to be measured			
	SmeC/Nem	400.2	0.2	0.50		
$C_{28}H_{26}F_2O$	Nem/Liq		427.2	0.3	0.70	[95]
	1-(6-octyloxy-2-naphthyl)-4-(3,4-difluorophenyl)diacetylene					
	Sol/SmeC	361.8	26.8	74.07		
$C_{28}H_{26}O_2$	SmeC/Nem	380.0	0.3	0.79		[111]
	Nem/Liq		406.3	0.3	0.74	
					75.60	
$C_{28}H_{26}O_2$	Sol/Sol	548.0	8.77	16.00		[111]
	Sol/SmeC	637.0	10.19	16.00		
	SmeC/Liq	678.0	Not reported in paper			
$C_{28}H_{26}O_5$	7-[(4'-hexyloxy)benzoyloxy]isoflavone					[14]
	Sol/SmeC	432.9	21.98	50.77		
	SmeC/Nem	455.3	1.82	4.00		
$C_{28}H_{26}O_6$	Nem/Liq	477.8	0.15	0.31	55.08	
	<i>bis</i> (4-pentyloxyphenyl) terephthalate					[12]
	Sol/Sol	413.0	6.7	16.22		
	Sol/SmeC	439.6	38.8	88.26		
$C_{28}H_{26}O_6S_2$	SmeC/Nem	447.0	0.1	0.22		[12]
	Nem/Liq	485.4	1.1	2.27	106.97	
					160.6	
$C_{28}H_{26}O_6S_2$	<i>bis</i> (4-propoxyphephenyl) 2,2'-bithiophene-5,5'-dicarboxylate					[12]
	Sol/SmeC	473.1	43.7	92.37		
	SmeC/Nem	488.4	0.7	1.43		
$C_{28}H_{26}O_8$	Nem/Liq	525.9	1.1	2.09	95.89	
	<i>bis</i> (4-propoxycarbonylphenyl) terephthalate					[194]
	Sol/SmeC	403.2	35.0	86.81		
$C_{28}H_{27}FO$	SmeC/Nem	480.2	1.9	3.96		[95]
	Nem/Liq	490.2	0.6	1.22	91.99	
					138.2	
$C_{28}H_{27}NO$	1-(6-octyloxy-2-naphthyl)-4-(4-fluorophenyl)diacetylene					[95]
	Sol/Nem	378.0	24.6	65.08		
	Nem/Liq	433.2	0.8	1.85	66.93	

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{28}H_{27}NO_5S$	Smec/Smec	548.0	4.5	8.21		[185]
	Smec/Nem	574.0	0.8	1.39		
	Nem/Liq	668.0	0.8	1.20	33.43	
$C_{28}H_{27}NO_5S$	4-thiocyanophenyl 4-(4-heptyloxybenzoyloxy)benzoate					
	Sol/Smec	385.2	31.3	81.26		[114]
	Smec/Nem	445.2	2.5	5.62		
	Nem/Liq	451.2	0.2	0.44	87.32	
$C_{28}H_{27}NO_6$	1,2-dimethylpropyl 4-[4-(4-acetoxybenzoyloxy)benzylideneamino]benzoate					
	Sol/Nem	429.2	36.0	83.88		[196]
	Nem/Liq	486.2	0.2	0.41	84.29	
$C_{28}H_{27}NO_6$	2,2-dimethylpropyl 4-[4-(4-acetoxybenzoyloxy)benzylideneamino]benzoate					
	Sol/Smec	433.2	28.0	64.64		[196]
	Smec/Liq	526.2	0.1	0.19	64.83	
$C_{28}H_{27}NO_6$	1-methylbutyl 4-[4-(4-acetoxybenzoyloxy)benzylideneamino]benzoate					
	Sol/Smec	408.2	30.0	73.49		[196]
	Smec/Nem	440.2	1.5	3.41		
$C_{28}H_{27}NO_6$	Nem/Liq	453.2	0.1	0.22	77.12	
$C_{28}H_{27}NO_6$	2-methylbutyl 4-[4-(4-acetoxybenzoyloxy)benzylideneamino]benzoate					
	Sol/Smec	395.2	29.0	73.38		[196]
	Smec/Nem	416.2	0.1	0.24		
$C_{28}H_{27}NO_6$	Nem/Liq	506.2	0.3	0.59	74.21	
$C_{28}H_{27}NO_6$	3-methylbutyl 4-[4-(4-acetoxybenzoyloxy)benzylideneamino]benzoate					
	Sol/Smec	414.2	34.0	82.09		[196]
	Smec/Nem	443.2	0.4	0.90		
$C_{28}H_{27}NO_6$	Nem/Liq	506.2	0.3	0.59	83.58	
$C_{28}H_{27}NO_6$	pentyl 4-[4-(4-acetoxybenzoyloxy)benzylideneamino]benzoate					
	Sol/Smec	398.2	32.0	80.36		[196]
	Smec/Nem	434.2	0.1	0.23		
$C_{28}H_{27}NO_6$	Nem/Liq	515.2	0.4	0.78	81.37	
$C_{28}H_{28}N_2O_4$	dipropyl N,N'-(1,4-phenylene-bis(methylidyne)]-bis[amino benzoate]					
	Sol/Smec	426.2	39.0	91.50		[192]
	Smec/Liq	472.3	0.21	0.44	91.94	
$C_{28}H_{28}O_2$	3-butyxo-6-[4-(4-butoxyphenyl)buta-1,3-diynyl]naphthalene					
	Sol/Sol	379.2	29.7	78.32		[225]
	Sol/Nem	384.1	0.8	2.08		
$C_{28}H_{28}O_2$	Nem/Liq	498.6	5.6 (Some decomposition)			
$C_{28}H_{29}ClO_2$	1-(4-heptylbiphenyl)-3-(4-chlorophenyl)propane-1,3-dione					
	Sol/Smec	411.7	27.61	67.06		[250]
	Smec/Liq	417.7	4.31	10.32	77.38	
$C_{28}H_{29}F_3$	3-methyl-4-[2-( <i>trans</i> -4-pentylcyclohexyl)ethynyl]-1-[(3,4,5-trifluorophenyl)ethynyl]benzene					
	Sol/Nem	329.2	21.9	66.52		[76]
	Nem/Liq	371.2	0.4	1.08	67.60	
$C_{28}H_{29}N$	2-(4-cyanophenyl)-7-octylfluorene					
	Sol/Smec	387.7	21.14	54.53		[2]
	Smec/Nem	451.1	0.55	1.22		
$C_{28}H_{29}N$	Nem/Liq	466.5	0.41	0.88	56.63	
$C_{28}H_{29}NO_4$	2-methylpentyl 4-[4-(4-methylbenzoyloxy)benzylideneamino]benzoate					
	Sol/Smec	394.2	32.0	81.18		[2]
	Smec/Nem	402.2	1.7	4.23		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$\text{C}_{28}\text{H}_{29}\text{NO}_4$	Nem/Liq	469.2	0.4	0.85	86.26	[196]
	Sol/SmeC	420.2	35.0	83.29		
	SmeC/Nem	434.2	1.6	3.68		
	Nem/Liq	457.2	0.3	0.66	87.63	
$\text{C}_{28}\text{H}_{29}\text{NO}_4$	Sol/SmeC	409.2	30.0	73.31		[196]
	SmeC/Nem	435.2	0.8	1.84		
	Nem/Liq	471.2	0.3	0.64	75.79	
	Sol/Nem	454.2	52.0	114.49		
$\text{C}_{28}\text{H}_{29}\text{NO}_4$	Nem/Liq	525.2	0.4	0.76	115.25	[196]
	Sol/Nem	402.2	38.0	94.48		
	Nem/Liq	413.2	0.2	0.48	94.96	
	Sol/Nem	427.2	48.0	112.36		
$\text{C}_{28}\text{H}_{29}\text{NO}_5$	Nem/Liq	436.2	0.2	0.46	112.82	[195]
	Sol/SmeC	384.2	21.0	54.66		
	SmeC/Liq	393.2	2.4	6.10	60.76	
	Sol/SmeC	416.2	30.0	72.08		
$\text{C}_{28}\text{H}_{29}\text{NO}_5$	SmeC/Liq	425.2	0.3	0.71	72.79	
	Sol/SmeC	412.2	39.0			[195]
	SmeC/Liq	399.2	2.3			
	Note: There is likely an error in the published temperatures of the two phase transitions.					
$\text{C}_{28}\text{H}_{29}\text{NO}_5$	Sol/Nem	395.2	28.0	70.85		[195]
	Nem/Liq	466.2	0.2	0.43	71.28	
	Sol/SmeC	406.2	33.0	81.24		
	SmeC/Nem	414.2	0.7	1.69		
$\text{C}_{28}\text{H}_{29}\text{NO}_5$	Nem/Liq	423.2	0.2	0.47	83.40	[195]
	Sol/Nem	383.2	38.0	99.16		
	Nem/Liq	492.2	0.3	0.61	99.77	
	Sol/Nem	433.2	48.0	110.80		
$\text{C}_{28}\text{H}_{29}\text{NO}_5$	Nem/Liq	476.2	0.2	0.42	111.22	[195]
	Sol/Nem	426.2	44.0	103.24		
	Nem/Liq	490.2	0.5	1.02	104.26	
	Sol/Nem	424.2	57.0	134.37		
$\text{C}_{28}\text{H}_{29}\text{NO}_5$	Nem/Liq	490.2	0.5	1.02	135.39	[195]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{28}H_{29}NO_5$	Sol/Smec	394.2	34.0	86.25		[195]
	Smec/Liq	444.2	4.2	9.46	95.71	
$C_{28}H_{29}NO_5$	Sol/Smec	397.2	32.0	80.56		[195]
	Smec/Liq	424.2	3.9	9.19	89.75	
$C_{28}H_{29}NO_5$	Sol/Smec	387.2	21.0	54.24		[195]
	Smec/Nem	432.9	0.6	1.39		
	Nem/Liq	467.2	0.2	0.43	56.06	
$C_{28}H_{29}NO_5$	Sol/Smec	392.2	32.0	81.59		[195]
	Smec/Liq	437.8	3.3	7.54	89.13	
$C_{28}H_{29}NO_5$	Sol/Smec	378.2	34.0	89.90		[195]
	Smec/Nem	431.2	0.3	0.70		
	Nem/Liq	490.2	1.2	2.45	93.05	
$C_{28}H_{29}NO_5$	Sol/Smec	419.2	40.0	95.42		[195]
	Smec/Nem	457.2	1.1	2.41		
	Nem/Liq	476.2	0.4	0.84	98.67	
$C_{28}H_{29}NO_5$	Sol/Smec	423.2	46.0	108.70		[195]
	Smec/Nem	455.2	0.8	1.76		
	Nem/Liq	485.2	0.7	1.44	111.90	
$C_{28}H_{29}NO_5$	Sol/Smec	418.2	50.0	119.56		[195]
	Smec/Nem	451.2	0.5	1.11		
	Nem/Liq	486.2	0.7	1.44	122.11	
$C_{28}H_{29}NO_6$	Sol/Smec	407.2	Value not reported in paper			[352]
	Smec/Smec	415.2	1.9	4.58		
	Smec/Nem	417.2	1.5	3.60		
	Nem/Liq	418.2	1.7	4.07		
$C_{28}H_{30}F_8O_5$	Sol/Smec	321.8	16.74	52.02		[125]
	Smec/Liq	326.8	4.37	13.37	65.39	
$C_{28}H_{30}O_4$	Sol/Sol	343.0	19.00	55.39		[157]
	Sol/Sol	396.0	1.46	3.69		
	Sol/Nem	407.0	26.30	64.62		
	Nem/Liq	430.0	1.71	3.98	127.68	
					125.4	
$C_{28}H_{30}O_7$	Sol/Nem	343.4	8.91	25.95		[103]
	Nem/Liq	433.2	0.39	0.90	26.85	
$C_{28}H_{31}F_7O_5$	Sol/Smec	333.1	17.86	53.62		[125]
	Smec/Smec	353.7	0.46	1.30		
	Smec/Liq	356.3	6.15	17.26	72.18	

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{28}H_{31}IO_3S$		4'-(10-undecyloxy)biphenyl 5-iodo-2-thiophenecarboxylate				
	Sol/Smec	375.0	2.51	6.69		
	Smec/Smec	397.6	40.17	101.03		
	Smec/Smec	410.8	5.86	14.26		
$C_{28}H_{31}N$	Smec/Liq	427.0	7.53	17.63	139.61	[63]
		4-nonyl-4"-cyano-p-terphenyl				
	Sol/Meso	358.3	20.19	56.35		
	Meso/Meso	383.3	1.74	4.54		
	Meso/Smec	391.0	3.27	8.36		
$C_{28}H_{31}NO_3$	Smec/Nem	478.7	0.51	1.07		
	Nem/Liq	484.8	1.00	2.06	72.38	126.4
		phenyl 4-(4-octyloxybenzylideneamino)benzoate				[8]
$C_{28}H_{31}NO_3$	Sol/Smec	362.2	41.0	113.20		
	Smec/Nem	421.2	2.5	5.94		
	Nem/Liq	429.2	0.8	1.86	121.00	140.6
$C_{28}H_{31}NO_3$		4-ethylphenyl 4-(4-hexyloxybenzylideneamino)benzoate				
	Sol/Smec	366.2	33.0	90.11		
	Smec/Nem	432.2	0.7	1.62		
	Nem/Liq	492.2	1.2	2.44	94.17	141.2
$C_{28}H_{31}NO_3$		4-isopropylphenyl 4-(4-pentyloxybenzylideneamino)benzoate				
	Sol/Smec	390.2	35.0	89.70		
	Smec/Nem	426.2	0.5	1.17		
	Nem/Liq	461.2	0.7	1.52	92.39	121.1
$C_{28}H_{31}NO_3$		4-(4-octyloxybenzylideneamino)phenyl benzoate				
	Sol/Nem	383.2	42.0	109.60		
	Nem/Liq	424.2	0.8	1.89	111.49	140.6
$C_{28}H_{32}$		1-( <i>trans</i> -4-ethylcyclohexylethynyl)-4-(4-butylphenylethynyl)benzene				
	Sol/Nem	375.8	15.4	40.98		
	Nem/Liq	448.5	1.3	2.90	43.88	[436]
$C_{28}H_{32}$		1-( <i>trans</i> -4-propylcyclohexylethynyl)-4-(4-propylphenylethynyl)benzene				
	Sol/Nem	379.1	16.0	42.21		
	Nem/Liq	485.9	1.1	2.26	44.47	[436]
$C_{28}H_{32}$		1-( <i>trans</i> -4-butylcyclohexylethynyl)-4-(4-ethylphenylethynyl)benzene				
	Sol/Nem	386.2	18.5	47.90		
	Nem/Liq	468.0	1.3	2.78	50.68	[436]
$C_{28}H_{33}NS$		4'-(4-pentylcyclohexylethyl)-4-isothiocyanatotolane				
	Sol/Nem	387.3	27.53	71.08		
	Nem/Liq	480.2	2.93	6.10	77.18	NA
$C_{28}H_{35}ClO_5$		2-chloro-4-methylpentyl 4-[3-[4-(hexyloxy)phenyl]-1-oxo-2-propenyl]oxy]benzoate				
	Sol/Smec	324.7	38.07	117.25		
	Smec/Chol	351.2	1.46	4.16		
	Chol/Liq	353.7	0.84	2.37	123.78	[257]
$C_{28}H_{35}ClO_5$		2-chloro-3-methylpentyl 4-[3-[4-(hexyloxy)phenyl]-1-oxo-2-propenyl]oxy]benzoate				
	Sol/Smec	344.2	21.76	63.22		
	Smec/Chol	356.7	1.67	4.68		
	Chol/Liq	357.7	0.42	1.17	69.07	[257]
$C_{26}H_{35}ClO_5$		4-[3-[4-(heptyloxy)phenyl]-1-oxo-2-propenyl]oxy]phenyl 2-chloro-4-methylpentanoate				
	Sol/Smec	328.2	34.14	104.02		
	Smec/Smec	340.2	0.21	0.62		
	Smec/Liq	363.7	3.39	9.32	113.96	[257]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.	
		T (K)							
$C_{28}H_{35}ClO_5$		4-[[3-[4-(heptyloxy)phenyl]-1-oxo-2-propenyl]oxy]phenyl 2-chloro-3-methylpentanoate							
	Sol/Smec	334.7	36.40		108.75			[257]	
	Smec/Smec	341.7	0.25		0.73				
	Smec/Chol	361.2	0.50		1.38				
$C_{28}H_{35}NO_2$	Chol/Liq	368.2	0.88		2.39	113.25		[112]	
		6-n-pentyloxy-2-[4-hexyloxystyryl]quinoline							
	Sol/Smec	383.0	21.74		56.76				
	Smec/Nem	392.3	12.00		30.59				
$C_{28}H_{35}NO_2$	Nem/Liq	444.1	0.88		1.98	89.33		[435]	
		(E)-4-{2-[4-(11-vinyloxyundecyloxy)phenyl]vinyl}benzonitrile							
	Sol/Nem	350.7	2.2		6.27				
	Nem/Liq	369.5	21.8		59.00	65.27			
$C_{28}H_{35}NO_2S$		4-isothiocyanatophenyl 4-( <i>trans</i> -4-octylcyclohexyl)benzoate							
	Sol/Sol	340.5	16.95		49.78			[356]	
	Smec/Smec	372.2	23.85		64.08				
	Smec/Nem	443.7	0.17		0.38				
$C_{28}H_{35}N_5O$	Nem/Liq	487.2	1.09		2.24	116.48	NA	[403]	
		4-[4-[(1E)-(4-cyanophenyl)azo]phenyl]-4-(1-oxo-10-undecenyl)piperazine							
	Sol/Smec	403.4	39.58		98.12				
	Smec/Nem	409.7	0.05		0.12				
$C_{28}H_{35}NO_3$	Nem/Liq	431.7	0.55		1.27	99.51		[117]	
		3(5)-(4-hex-5'-enyl-1'-oxyphenyl)-5(3)-(4-heptyl-1'-oxyphenyl)isoxazole							
	Sol/Smec	362.1	18.4		50.81				
	Smec/Nem	400.9	0.6		1.50				
$C_{28}H_{36}O_4$	Nem/Liq	423.2	1.2		2.84	55.15		[44]	
		7-propoxy-3-(4-decyloxyphenyl)-3 <i>H</i> -1-benzopyran-4-one							
	Sol/Sol	353.5	1.6		4.53				
	Smec/Nem	377.5	30.2		80.00				
$C_{28}H_{36}O_4$	Nem/Liq	405.8	0.9		2.22	86.75		[215]	
		di(4'-butylphenyl)- <i>trans</i> -cyclohexane-1,4-dicarboxylate							
	Sol/Smec	364.8	27.0		74.01				
	Smec/Nem	380.9	0.36		0.95				
$C_{28}H_{36}O_6$	Nem/Liq	423.9	0.56		1.32	76.28	133.3	[220]	
		di(4'-butoxyphenyl)- <i>trans</i> -cyclohexane-1,4-dicarboxylate							
	Sol/Smec	380.2	28.69		75.46				
	Smec/Nem	429.2	0.96		2.24				
$C_{28}H_{37}NO_3$	Nem/Liq	482.2	0.89		1.85	79.55	146.9	[123]	
		6-hexyloxy naphth-2-yl 5-hexyloxy-2-methylpyridyl ketone							
	Sol/Sol	349.2	0.72		2.06				
	Smec/Nem	383.2	46.01		120.07				
$C_{28}H_{37}NO_5$	Nem/Liq	392.2	0.35		0.89	123.02		[294]	
		4-[4-(4-heptyloxyphenyliminomethyl)-3-hydroxyphenoxy]butylmethacrylate							
	Sol/Smec	313.5	48.52		154.77				
	Smec/Smec	344.2	Not reported in paper						
$C_{28}H_{37}N_3O_2S$	Smec/Liq	359.2	6.63		18.45			[167]	
		5-(4-octyloxyphenyl)-N-[[4-(2S)-2-methylbutoxy]phenyl]methylene-1,3,4-thiadiazol-2-amine							
	Sol/Smec	374.6	20.72		55.31				
	Smec/Nem	412.6	2.83		6.86				
	Nem/Liq	426.0	1.65		3.87	66.04			

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.	
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$			
$C_{28}H_{37}N_3O_2S$	Sol/Smec	396.7	36.4	91.76		[79,396,397]	
	Smec/Nem	449.0	2.4	5.35			
	Nem/Liq	483.0	1.7	3.52	100.63		
$C_{28}H_{37}N_3O_3S$	Sol/Smec	438.0	20.7	47.26		[31]	
	Smec/Nem	515.0	3.8	7.38			
	Nem/Liq	533.7	1.2	2.25	56.89		
$C_{28}H_{38}$	Sol/Nem	386.5	18.0	46.57		[436]	
	Nem/Liq	451.3	1.1	2.44	49.01		
$C_{28}H_{38}$	Sol/Nem	438.1	14.0	31.96		[436]	
	Nem/Liq	471.7	1.3	2.76	34.72		
$C_{28}H_{38}N_2O_2S$	Sol/Nem	358.2	40.7	113.62		[41]	
	Nem/Liq	392.1	1.1	2.81	116.43		
$C_{28}H_{38}N_2O_2S$	Sol/Smec	379.8	7.70	20.27		[75]	
	Smec/Liq	457.2	7.70	16.84	37.11		
$C_{28}H_{38}N_2O_2S_2$	Sol/Sol	386.6	32.52	84.11		[269]	
	Sol/Smec	433.9	44.68	102.97			
	Smec/Smec	509.3	6.24	12.25			
	Smec/Liq	523.5	2.05	3.92	203.25		
$C_{28}H_{38}N_2O_3$	Sol/Sol	354.4	4.68	13.21		[157]	
	Sol/Smec	368.7	31.36	85.06			
	Smec/Liq	404.6	8.54	21.11	119.38		
					158.3		
$C_{28}H_{38}N_2O_3$	Sol/Smec	374.7	44.77	119.48		[157]	
	Smec/Liq	416.7	8.08	19.39	138.87		
					158.3		
$C_{28}H_{38}O_4$	Sol/Smec	312.2	26.9	86.16		[200]	
	Smec/Liq	328.2	Not reported in paper				
$C_{28}H_{38}O_4$	Sol/Smec	321.2	30.4	94.65		[200]	
	Smec/Chol	334.2	Not reported in paper				
	Chol/Liq	336.2	Not reported in paper				
$C_{28}H_{38}O_4$	Sol/Smec	318.2	18.6	58.45		[200]	
	Smec/Chol	324.2	Not reported in paper				
	Chol/Liq	338.2	Not reported in paper				
$C_{28}H_{38}O_4$	Sol/Smec	303.2	11.7	38.59		[200]	
	Smec/Chol	313.2	Not reported in paper				
	Chol/Liq	329.2	Not reported in paper				
$C_{28}H_{38}O_4$	Sol/Smec	312.2	24.0	76.87			

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated) (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{28}H_{38}O_3$	Smec/Chol	324.2	Not reported in paper		[200]	
	Chol/Liq	333.2	Not reported in paper			
$C_{28}H_{39}ClN_2O_4$	Sol/Chol	365.2	23.05	63.12	[208]	
	Chol/Liq	417.2	Not reported in paper			
	Sol/Smec	338.8	44.17	130.37		
$C_{28}H_{39}ClO_2S$	Smec/Nem	343.6	0.15	0.44	[47]	
	Nem/Liq	347.5	2.87	8.26		
	Sol/Smec	358.3	57.32	159.98		
$C_{28}H_{39}NO_3$	Smec/Liq	366.7	4.90	13.36	[383]	
	Sol/Smec	405.2	10.8	26.65		
	Nem/Liq	515.2	0.97	1.92		
$C_{28}H_{39}NO_5$	Nem/Liq	13.8	26.79	55.36	[416]	
	Sol/Smec	400.0	47.24	118.10		
	Smec/Smec	460.0	0.32	0.70		
	Smec/Smec	471.0	1.32	2.80		
$C_{28}H_{39}N_3OS$	Smec/Liq	479.0	6.23	13.01	[87]	
	Sol/Smec	399.2	24.6	61.62		
	Smec/Nem	406.5	2.39	5.88		
	Nem/Liq	408.2	1.37	3.36		
$C_{28}H_{39}N_3O_2S$	Sol/Nem	335.5	21.4	63.79	[41]	
	Nem/Liq	388.6	0.8	2.06		
	Sol/Nem	406.5	2.39	5.88		
$C_{28}H_{40}N_2O_2$	Sol/Chol	392.2	13.8	35.19	[201]	
	Chol/Liq	426.2	Not reported in paper			
	Sol/Nem	340.2	45.0	132.28		
$C_{28}H_{40}N_2O_3$	Nem/Liq	350.2	0.8	2.28	[339]	
	Sol/Nem	323.2	42.0	129.95		
	Nem/Liq	337.2	1.0	2.97		
$C_{28}H_{40}N_4OS_2$	Sol/Smec	425.9	38.3	89.93	[396]	
	Nem/Liq	431.7	0.88	2.04		
	Sol/Smec	425.9	38.3	89.93		
$C_{28}H_{40}O_2S$	Sol/Smec	338.9	38.74	114.31	[217]	
	Smec/Nem	354.3	0.97	2.73		
	Nem/Liq	360.3	1.87	5.19		
$C_{28}H_{40}O_2S$	Sol/Meso	303.4	3.2	10.55	[217]	
	Meso/Smec	312.6	6.1	19.51		
	Smec/Smec	320.1	3.3	10.31		
	Smec/Smec	331.2	Too small to be measured			

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.			
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$					
$\text{C}_{28}\text{H}_{41}\text{NO}$	Smec/Nem	352.5	1.0	2.84	47.39	[371]			
	Nem/Liq	358.6	1.5	4.18					
$\text{C}_{28}\text{H}_{41}\text{NO}$	N-(4-n-octyloxybenzylidene)-4-(n-heptyl)aniline					[11]			
	Sol/Smec	302.3	29.37	97.16	125.64				
	Smec/Smec	347.4	3.59	10.33					
	Smec/Liq	359.8	6.53	18.15					
$\text{C}_{28}\text{H}_{41}\text{NO}$	N-(4-tridecyloxybenzylidene)-4-ethylaniline					[65]			
	Sol/Smec	324.4	29.92	92.23	109.65				
	Smec/Smec	337.5	1.55	4.59					
	Smec/Liq	346.0	4.44	12.83					
$\text{C}_{28}\text{H}_{41}\text{NO}$	4-dodecyloxybenzylidene 4'-isopropylaniline					[273]			
	Sol/Smec	327.7	10.95	33.41	37.01				
	Smec/Liq	333.2	1.20	3.60					
$\text{C}_{28}\text{H}_{41}\text{NO}_3$	N-(4-methoxyphenyl)- $\alpha$ -(4-tetradecyloxyphenyl)nitrone					[162]			
	Sol/Smec	385.2	64.94	168.59	NA				
	Smec/Liq	397.2	3.88	9.77	178.36				
$\text{C}_{28}\text{H}_{41}\text{NO}_5$	4-hexadecyloxyphenyl 4-nitrobenzoate					[144]			
	Sol/Smec	355.6	60.2	169.29	183.4				
	Smec/Liq	364.9	4.9	13.43	182.72				
$\text{C}_{28}\text{H}_{41}\text{NO}_5$	4-nitrophenyl 4-hexadecyloxybenzoate					[144]			
	Sol/Smec	352.6	54.5	154.57	183.4				
	Smec/Liq	361.9	3.6	9.95	164.52				
$\text{C}_{28}\text{H}_{42}\text{N}_2\text{O}$	4-butyl-4'-dodecyloxyazobenzene					[141]			
	Sol/Smec	337.0	32.74	97.15	162.8				
	Smec/Smec	341.1	Not reported in paper						
	Smec/Nem	349.5	7.86	22.49					
	Nem/Liq	349.9	Not detected by dsc Nem/Liq transition enthalpy is likely included in the Smec/Nem value.						
	4-pentyl-4'-undecyloxyazobenzene								
$\text{C}_{28}\text{H}_{42}\text{N}_2\text{O}$	Sol/Smec	331.9	25.18	75.87	162.8	[141]			
	Smec/Smec	335.1	1.57	4.69					
	Smec/Nem	355.0	3.40	9.58					
	Nem/Liq	357.4	2.43	6.80	96.94				
$\text{C}_{28}\text{H}_{42}\text{N}_2\text{O}$	4-heptyl-4'-nonyloxyazobenzene					[390]			
	Sol/Smec	312.4	34.54	110.56	124.16				
	Smec/Smec	320.5	0.65	2.03					
	Smec/Nem	349.4	1.93	5.52					
$\text{C}_{28}\text{H}_{42}\text{N}_2\text{O}$	Nem/Liq	355.3	2.15	6.05	124.16	162.8			
	2-[4-(9-decenoxy)phenyl]-5-[ $\text{S}$ -(5-methylheptyl)]pyrimidine								
	Sol/Smec	290.2	14.2	48.93	Not reported in paper				
	Smec/Chol	307.2	Not reported in paper						
$\text{C}_{28}\text{H}_{42}\text{N}_2\text{O}$	Chol/Liq	317.2	Not reported in paper			[201]			
	2-[4-(8-nonenyloxy)phenyl]-5-[ $\text{S}$ -(6-methyloctyl)]pyrimidine								
	Sol/Smec	272.2	9.1	33.43	Not reported in paper				
	Smec/Chol	301.2	Not reported in paper						
$\text{C}_{28}\text{H}_{42}\text{N}_2\text{O}$	Chol/Liq	315.2	Not reported in paper			[201]			
	4,4'-dioctyloxyazoxybenzene								
	Sol/Smec	352.7	42.18	119.59	126.47	178.22			
	Smec/Nem	360.9	1.18	3.27					
$\text{C}_{28}\text{H}_{42}\text{N}_2\text{O}_3$	Nem/Liq	399.3	1.44	3.61	178.22	[179]			

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated) (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{28}H_{44}O_3$	Sol/Nem	4-heptylcyclohexyl 4-hexyloxycinnamate				
	Nem/Liq	342.9	48.1	140.27	144.89	[5]
$C_{28}H_{45}N_3O_6$	Sol/Nem	4,4'-bis[4-(4-butoxybenzoyloxy)benzylideneamino]diphenylamine				
	Nem/Liq	504.2	46.0	91.23	95.06	[284]
$C_{28}H_{46}$	<i>trans, trans</i> -4,4'-bis[2-(4-propylcyclohexyl)ethyl]benzene					
	Sol/Smec	337.2	2.21	6.55		
	Smec/Smec	375.7	7.53	20.04		
	Smec/Nem	377.0	7.49	19.87		
$C_{28}H_{46}O$	Nem/Liq	388.2	3.22	8.29	54.75	[348]
	<i>trans</i> -1-heptyl-4-(4-nonanoylphenyl)cyclohexane					
	Smec/Smec	343.35	19.52	56.86		
$C_{28}H_{48}N_2O_2$	Smec/Liq	353.25	10.62	30.07	86.92	159.7
	N,N'-dinonanoyl-2,3,5,6-tetramethylbenzene-1,4-diamine					
	Sol/Sol	425.2	15.0	35.28		
	Sol/Meso	491.2	12.0	24.43		
$C_{28}H_{48}N_2O_2$	Meso/Meso	527.2	10.0	18.97		
	Meso/Liq	568.2	23.0	40.48	119.16	[36]
	N,N'-dinonanoyl-3,4,5,6-tetramethylbenzene-1,2-diamine					
$C_{28}H_{48}N_2O_2$	Sol/Meso	396.2	11.0	27.76		
	Meso/Meso	483.2	2.0	4.14		
	Meso/Liq	512.2	19.0	37.09	63.99	[61]
$C_{28}H_{48}N_2O_2$	N,N'-didecylbenzene-1,2-dicarboxamide					
	Sol/Meso	349.2	39.0	111.68		
	Meso/Liq	373.2	11.0	29.47	141.15	[61]
$C_{28}H_{49}NO_4$	3,4,5-tris(heptyloxy)benzamide					
	Sol/Meso	353.2	17.9	50.68		
	Meso/Liq	362.2	6.1	16.84	67.52	[378]
$C_{28}H_{78}Si_{10}$	1,10-dibutyl(permethyl)decasilane					
	Sol/Meso	296.2	28.6	96.56		
	Meso/Liq	342.2	12.5	36.53	133.09	[110]
$C_{29}H_{18}F_{17}O_5$	4-(3,3,4,4,5,5,6,6,7,7,8,8,8-tridecafluoroctyloxycarbonyl)phenyl 4-[(perfluorobutyl)propoxy]benzoate					
	Sol/Smec	377.3	32.4	85.87		
	Smec/Liq	389.0	6.44	16.56	102.43	[128]
$C_{29}H_{19}F_{21}O_5$	4-(2,2,3,3,4,4,5-octafluoropentyloxycarbonyl)phenyl 4-[(perfluorohexyl)butoxy]benzoate					
	Sol/Smec	365.0	45.04	123.40		
	Smec/Liq	372.9	4.56	12.23	135.63	[128]
$C_{29}H_{20}F_{20}O_5$	4-(2,2,3,3,4,4,4-heptafluorobutyloxycarbonyl)phenyl 4-[(perfluorohexyl)pentyloxy]benzoate					
	Sol/Smec	346.9	35.80	103.20		
	Smec/Smec	365.5	0.73	2.00		
$C_{29}H_{21}F_{17}O_9$	Smec/Liq	377.2	3.73	9.89	115.09	[128]
	4-(carboxy)phenyl 4-{5-[1H,1H-2,5-di(trifluoromethyl)-3,6-dioxaundecafluorononyloxycarbonyl]-pentyloxy}benzoate					
	Sol/Smec	374.2	4.4	11.76		
$C_{29}H_{23}F_{15}O_7$	Smec/Liq	491.2	11.2	22.80	34.56	[221]
	4-(carboxy)phenyl 4-{5-[2-(perfluoro-5-methylhexyl)ethoxycarboxy]pentyloxy}benzoate					
	Sol/Smec	406.2	10.1	24.86		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.	
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$			
$C_{29}H_{24}F_4O_5$	Smec/Liq	496.2	18.2	36.68	61.54	[221]	
	Sol/Smec	368.6	30.80	83.56			
	Smec/Chol	383.5	0.18	0.47			
	Chol/Liq	447.9	0.42	0.94	84.97		
$C_{29}H_{25}ClF_4O_3$	4-[ <i>(S</i> )-2-methylbutoxycarbonyl]phenyl 4-[ <i>(4</i> -ethoxy-2,3,4,5-tetrafluorophenyl)ethynyl]benzoate					[91]	
	Sol/Nem	388.4	32.51	83.70			
	Nem/Liq	440.2	1.08	2.45	86.15		
$C_{29}H_{25}ClF_4O_3$	4-(4-octyloxyphenyl)acetylene-2,3,5,6-tetrafluorophenyl 4'-chlorobenzoate					[101]	
	Sol/Nem	354.0	25.23	71.27			
	Nem/Liq	452.9	0.74	0.16	71.43		
$C_{29}H_{26}F_4O_4$	4-[ <i>(S</i> )-2-methylbutoxyl]phenyl 4-[ <i>(4</i> -propoxo-2,3,5,6-tetrafluorophenyl)ethynyl]benzoate					[124]	
	Sol/Nem	381.3	25.47	66.80			
	Nem/Liq	433.3	0.26	0.60	67.40		
$C_{29}H_{27}F_{13}O_5$	4-(3,3,4,4,5,5,6,6,7,7,8,8,8-tridecafluoroctyloxycarbonyl)phenyl 4-(heptyloxy)benzoate					[125]	
	Sol/Smec	362.3	29.83	82.34			
	Smec/Smec	391.9	0.07	0.18			
$C_{29}H_{28}$	Smec/Liq	407.8	8.92	21.87	104.39	[55, 430]	
	4'-[2-(4-methylphenyl)-1-ethynyl]-2'-methyl-4-pentyltolane						
	Sol/Nem	384.5	22.5	58.52			
$C_{29}H_{28}$	Nem/Liq	458.9	1.05	2.29	60.81		
	4'-[2-(4-propylphenyl)-1-ethynyl]-2'-methyl-4-propyltolane					[55, 430]	
	Sol/Nem	396.6	21.0	52.95			
$C_{29}H_{28}$	Nem/Liq	473.6	1.05	2.22	55.17		
	4'-[2-(4-butylphenyl)-1-ethynyl]-2'-methyl-4-ethyltolane					[55, 430]	
	Sol/Nem	345.4	17.5	50.67			
$C_{29}H_{28}$	Nem/Liq	450.6	0.98	2.17	52.84		
	4'-[2-(4-ethylphenyl)-1-ethynyl]-2'-ethyl-4-propyltolane					[55]	
	Sol/Nem	347.1	14.0	40.33			
$C_{29}H_{28}$	Nem/Liq	414.7	0.94	2.27	42.60		
	4'-[2-(4-propylphenyl)-1-ethynyl]-2'-ethyl-4-ethyltolane					[55]	
	Sol/Nem	310.4	13.3	42.85			
$C_{29}H_{28}FNO_5S$	Nem/Liq	409.2	0.83	2.03	44.88		
	4-octyloxyphenyl 3-fluoro-4-thiocyanatophenyl terephthalate					[37]	
	Sol/Smec	377.2	24.6	65.22			
$C_{29}H_{28}O_5$	Smec/Smec	382.2	Too small to be measured				
	Smec/Nem	411.2	0.2	0.49			
	Nem/Liq	429.2	0.7	1.63	67.34		
$C_{29}H_{28}O_5$	7-(4'-heptyloxybenzoyloxy)isoflavone					[14]	
	Sol/Smec	434.8	33.71	77.53			
	Smec/Nem	460.8	1.67	3.62			
$C_{29}H_{29}ClO_8$	Nem/Liq	472.9	0.49	1.04	82.19	[174]	
	di-(4-butoxycarbonylphenyl) 2-chloroterephthalate						
	Sol/Smec	366.2	26.3	71.82			
$C_{29}H_{29}NO$	Smec/Liq	372.2	1.99	5.35	77.17		
	5-(4'-pentyloxybiphenyl-4-yl)-2-(4-methylphenyl)pyridine					[14]	
	Sol/Sol	445.0	6.3	14.16			
$C_{29}H_{29}NO$	Sol/Smec	518.0	4.0	7.72			

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$\text{C}_{29}\text{H}_{29}\text{NO}_5\text{S}$	Smec/Smec	548.0	3.6	6.57		[185]
	Smec/Nem	589.0	1.4	2.38		
	Nem/Liq	654.0	0.3	0.46	31.29	
$\text{C}_{29}\text{H}_{29}\text{NO}_5\text{S}$	Sol/Smec	383.2	38.8	101.25		[114]
	Smec/Liq	452.2	3.8	8.40	109.65	
$\text{C}_{29}\text{H}_{29}\text{NO}_6$	Sol/Nem	455.2	42.0	92.27		[196]
	Nem/Liq	472.2	0.3	0.64	92.91	
$\text{C}_{29}\text{H}_{29}\text{NO}_6$	Sol/Smec	396.2	25.0	63.10		[196]
	Smec/Nem	445.2	1.5	3.37		
	Nem/Liq	451.2	0.2	0.44	66.91	
$\text{C}_{29}\text{H}_{29}\text{NO}_6$	Sol/Smec	400.2	31.0	77.46		[196]
	Smec/Nem	423.2	0.1	0.24		
	Nem/Liq	511.2	0.5	0.98	78.68	
$\text{C}_{29}\text{H}_{29}\text{NO}_6$	Sol/Smec	397.2	28.0	70.49		[196]
	Smec/Nem	441.2	0.5	1.13		
	Nem/Liq	483.2	0.3	0.62	72.24	
$\text{C}_{29}\text{H}_{29}\text{NO}_6$	Sol/Smec	388.2	28.0	72.13		[196]
	Smec/Nem	423.2	0.1	0.24		
	Nem/Liq	485.2	0.1	0.21	72.58	
$\text{C}_{29}\text{H}_{31}\text{N}$	Sol/Smec	386.2	21.87	56.63		[2]
	Smec/Nem	455.1	0.93	2.04	58.67	
$\text{C}_{29}\text{H}_{30}\text{N}_4\text{O}_2\text{S}$	Sol/Nem	435.3	51.95	119.34		[79]
	Nem/Liq	451.5	0.79	1.75	121.09	
$\text{C}_{29}\text{H}_{30}\text{O}_8$	Sol/Smec	414.2	40.1	96.81		[174]
	Smec/Nem	467.2	Not reported in paper			
	Nem/Liq	499.2	4.1	8.21	105.02	
$\text{C}_{29}\text{H}_{31}\text{ClO}_2$	Sol/Smec	410.7	19.25	46.87		[250]
	Smec/Liq	419.2	4.60	10.97	57.84	
$\text{C}_{29}\text{H}_{31}\text{F}_3\text{O}$	Sol/Smec	338.2	24	70.96		[76]
	Smec/Nem	366.2	0.09	0.25		
	Nem/Liq	440.2	1.0	2.27	73.48	
$\text{C}_{29}\text{H}_{31}\text{N}$	Sol/Nem	351.2	29	82.57		[76]
	Nem/Liq	501.2	3.6	7.18	89.75	
$\text{C}_{29}\text{H}_{31}\text{N}$	Sol/Nem	361.2	25.5	70.60		[76]
	Nem/Liq	485.2	1.0	2.06	72.66	

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.	
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$			
$\text{C}_{29}\text{H}_{31}\text{NO}_3\text{S}$	Sol/Smec	400.7	92.47	230.77		[63]	
	Smec/Nem	428.5		Not given in paper			
	Nem/Liq	432.8	4.60	10.63			
$\text{C}_{29}\text{H}_{31}\text{NO}_5$	2-methylhexyl 4-[4-(4-methoxybenzoyloxy)benzylideneamino]benzoate					[195]	
	Sol/Smec	381.2	35.0	91.82			
	Smec/Nem	396.2	0.1	0.25			
$\text{C}_{29}\text{H}_{31}\text{NO}_5$	Nem/Liq	477.2	0.3	0.63	92.70	[195]	
	heptyl 4-[4-(4-methoxybenzoyloxy)benzylideneamino]benzoate						
	Sol/Nem	423.2	66.0	155.95			
$\text{C}_{29}\text{H}_{31}\text{NO}_5$	Nem/Liq	487.2	0.5	1.03	156.98	[195]	
	2-methylhexyl 4-[4-(4-methoxybenzylideneamino)benzoyloxy]benzoate						
	Sol/Smec	365.2	26.0	71.19			
$\text{C}_{29}\text{H}_{31}\text{NO}_5$	Smec/Nem	438.2	0.6	1.37		[195]	
	Nem/Liq	480.2	0.6	1.25	73.81		
	heptyl 4-[4-(4-methoxybenzylideneamino)benzoyloxy]benzoate						
$\text{C}_{29}\text{H}_{31}\text{NO}_5$	Sol/Smec	415.2	51.0	122.83		[195]	
	Smec/Nem	454.2	0.5	1.10			
	Nem/Liq	482.2	0.7	1.45	125.38		
$\text{C}_{29}\text{H}_{31}\text{NO}_6$	ethoxyethyl 4-(4'-butoxybenzoyloxybenzylidene)-4"-aminobenzoate					[58]	
	Sol/Smec	430.2	6.04	14.04			
	Smec/Nem	450.2	0.04	0.09			
$\text{C}_{29}\text{H}_{31}\text{NO}_6$	Nem/Liq	480.2	0.55	1.15	15.33	[58]	
	4-[[4-(decyloxy)phenoxy]carbonyl]phenyl 5-cyano-2-furancarboxylate					[352]	
	Sol/Smec	408.2	Value not reported in paper				
$\text{C}_{29}\text{H}_{32}\text{F}_8\text{O}_5$	Smec/Smec	417.2	2.8	6.71		[352]	
	Smec/Liq	421.2	3.3	7.83			
	4-(2,2,3,3,4,4,5,5-octafluoropentyloxycarbonyl)phenyl 4-(decyloxy)benzoate						
$\text{C}_{29}\text{H}_{32}\text{O}_7$	Sol/Smec	324.0	21.60	66.67		[125]	
	Smec/Liq	331.3	5.14	15.51	82.18		
$\text{C}_{29}\text{H}_{33}\text{F}_7\text{O}_5$	4-(2,2,3,3,4,4,4-heptafluorobutyloxycarbonyl)phenyl 4-(undecyloxy)benzoate					[125]	
	Sol/Smec	332.9	25.78	77.44			
	Smec/Smec	350.0	0.48	1.37			
$\text{C}_{29}\text{H}_{33}\text{N}$	Smec/Liq	352.4	6.12	17.37	96.18	[125]	
	4-decyl-4"-cyano-p-terphenyl						
	Sol/Meso	365.2	28.52	78.09			
Note: Transition enthalpy includes a Sol/Sol transition.							
$\text{C}_{29}\text{H}_{33}\text{NO}_3$	Meso/Meso	380.9	1.53	4.02		[8]	
	Meso/Smec	390.4	3.24	8.30			
	Smec/Liq	478.5	4.15	8.67	99.08		
					133.5		
$\text{C}_{29}\text{H}_{33}\text{NO}_3$	4-ethylphenyl 4-(4-heptyloxybenzylideneamino)benzoate					[292]	
	Sol/Smec	453.2	39.0	86.05			
	Smec/Nem	440.2	0.9	2.04			
$\text{C}_{29}\text{H}_{33}\text{NO}_3$	Nem/Liq	483.2	1.2	2.48	90.57	[292]	
	4-(4'-octyloxybenzoyloxybenzylidene)-2"-aniline						
$\text{C}_{29}\text{H}_{33}\text{NO}_3$	Sol/Nem	352.1	35.00	99.40		[292]	

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$\text{C}_{29}\text{H}_{33}\text{NO}_3$	Nem/Liq	384.4	0.55	1.43	100.83	[399]
	Sol/SmeC	385.2	32.0	83.07		
	SmeC/Nem	435.2	0.9	2.07		
	Nem/Liq	460.2	0.9	1.96	87.10	128.2 [292]
$\text{C}_{29}\text{H}_{33}\text{O}_4\text{S}$	4'- <i>(10</i> -undecenoxy)biphenyl 5-methoxy-2-thiophenecarboxylate					
	Sol/Nem	362.4	92.47	255.16		
	Nem/Liq	426.4	3.77	8.84	264.00	[63]
$\text{C}_{29}\text{H}_{34}$	1-( <i>trans</i> -4-ethylcyclohexylethynyl)-4-(4-pentylphenylethynyl)benzene					
	Sol/Nem	360.6	16.9	46.87		
	Nem/Liq	444.0	0.7	1.58	48.45	[436]
$\text{C}_{29}\text{H}_{34}$	1-( <i>trans</i> -4-propylcyclohexylethynyl)-4-(4-butylphenylethynyl)benzene					
	Sol/Nem	375.1	17.7	47.19		
	Nem/Liq	472.5	1.6	3.39	50.58	[436]
$\text{C}_{29}\text{H}_{34}$	1-( <i>trans</i> -4-butylcyclohexylethynyl)-4-(4-propylphenylethynyl)benzene					
	Sol/Nem	386.8	17.3	44.73		
	Nem/Liq	477.7	1.4	2.93	47.66	[436]
$\text{C}_{29}\text{H}_{34}$	1-( <i>trans</i> -4-pentylcyclohexylethynyl)-4-(4-ethylphenylethynyl)benzene					
	Sol/Nem	388.8	9.0	23.15		
	Nem/Liq	480.2	0.6	1.25	24.40	[436]
$\text{C}_{29}\text{H}_{34}\text{N}_2\text{O}_3$	4-[ <i>(pyridine-4-yl)methylene</i> amino]phenyl 4-decyloxybenzoate					
	Sol/SmeC	367.2	40.9	111.38		
	SmeC/SmeC	406.2	Not reported in paper			
	SmeC/Nem	416.2	1.6	3.84		
$\text{C}_{29}\text{H}_{34}\text{N}_2\text{O}_3$	Nem/Liq	418.2	0.8	1.91		[265, 266]
		3-pyridyl 4-(4-decyloxybenzylideneamino)benzoate				
		Sol/SmeC	356.2	51.0	143.18	
		SmeC/Liq	437.2	5.5	12.58	155.76 [291]
$\text{C}_{29}\text{H}_{34}\text{N}_2\text{O}_3$	4[ <i>(E)</i> -[(4-decyloxyphenyl)methylene]amino]phenyl 3-pyridinecarboxylate					
	Sol/SmeC	390.2	57.0	146.08		
	SmeC/Nem	407.2	0.7	1.72		
	Nem/Liq	432.2	0.9	2.08	149.88	[291]
$\text{C}_{29}\text{H}_{34}\text{N}_2\text{O}_4$	4-[ <i>(4</i> -ethoxyphenyl)azo]phenyl 4-(octyloxy)benzoate					
	Sol/Nem	390.2	40.0	102.51		
	Nem/Liq	515.2	1.5	2.91	105.42	[339]
$\text{C}_{29}\text{H}_{34}\text{O}_3$	4-biphenyl 4"-decyloxybenzoate					
	Sol/Nem	383.2	53.14	138.67		
	Nem/Liq	400.2	Not reported in paper			[425]
Note: Authors report only an enthalpy of fusion, and state in a footnote that Nem/Liq transition enthalpies for the compounds studied were in the 0.84–2.09 kJ·mol <sup>-1</sup> range.						
$\text{C}_{29}\text{H}_{34}\text{O}_7$	4-propoxyphephenyl 7-decanoxyloxychromone-2-carboxylate					
	Sol/SmeC	392.0	16.3	41.58		
$\text{C}_{29}\text{H}_{37}\text{ClO}_5$	SmeC/Liq	417.6	2.9	6.94	48.52	[286]
		2-chloro-4-methylpentyl 4-[ <i>[3-[4-(heptyloxy)phenyl]-1-oxo-2-propenyl]oxy</i> ]benzoate				
$\text{C}_{29}\text{H}_{37}\text{ClO}_5$	Sol/SmeC	330.7	33.05	99.94		
	SmeC/Chol	347.7	1.05	3.02		
	Chol/Liq	350.2	0.92	2.63	105.59	[257]
$\text{C}_{29}\text{H}_{37}\text{ClO}_5$	2-chloro-3-methylpentyl 4-[ <i>[3-[4-(heptyloxy)phenyl]-1-oxo-2-propenyl]oxy</i> ]benzoate					

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.	
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$			
$C_{29}H_{37}ClO_5$	Sol/Smec	337.7	33.89	100.36		[257]	
	Smec/Chol	353.7	0.84	2.37			
	Chol/Liq	355.7	0.59	1.66	102.86		
$C_{29}H_{37}ClO_5$	4-[[3-[4-(octyloxy)phenyl]-1-oxo-2-propenyl]oxy]phenyl 2-chloro-4-methylpentanoate					[257]	
	Sol/Smec	335.2	24.27	72.40			
	Smec/Smec	346.2	0.21	0.61			
	Smec/Liq	365.7	4.06	11.10	84.11		
$C_{29}H_{37}ClO_5$	4-[[3-[4-(octyloxy)phenyl]-1-oxo-2-propenyl]oxy]phenyl 2-chloro-3-methylpentanoate					[257]	
	Sol/Smec	334.2	18.83	56.34			
	Smec/Smec	346.2	0.21	0.61			
	Smec/Chol	366.2	0.42	1.15			
$C_{29}H_{37}NO$	Chol/Liq	369.7	0.75	2.03	60.13	[257]	
	4-(4-cyanophenyl)-1-(4-dodecyloxyphenyl)-buta-1E,3E-diene						
	Sol/Smec	370.1	46.2	124.83			
	Smec/Nem	445.1	3.4	7.64	132.47		
$C_{29}H_{37}NO_2$	3-[4-(trans-4-heptylcyclohexyl)phenyl]-5-(4-methoxyphenyl)isoxazole					[131]	
	Sol/Nem	408.2	25.04	61.34			
	Nem/Liq	551.2	0.98	1.78	63.12		
$C_{29}H_{37}NO_2S$	4-isothiocyanatophenyl 4-(trans-4-nonylcyclohexyl)benzoate					[356]	
	Sol/Smec	373.7	35.98	96.28			
	Smec/Nem	450.7	0.13	0.29			
	Nem/Liq	484.2	1.76	3.63	100.20		
$C_{29}H_{37}N_3O_3$	1-[4-(N-ethoxycarbonylpiperazinyl)phenyl]-3-(4-heptyloxy phenyl)(1-propargyl-3-imine)					[154]	
	Sol/Smec	410.7	15.3	37.25			
	Smec/Liq	454.5	6.0	13.20	50.45		
$C_{29}H_{37}NO_3$	3(5)-(4-hex-5'-enyl-1'-oxyphenyl)-5(3)-(4-octyl-1'-oxyphenyl)isoxazole					[117]	
	Sol/Smec	350.0	15.7	44.86			
	Smec/Nem	405.2	0.4	0.99			
	Nem/Liq	422.4	1.4	3.31	49.16		
$C_{29}H_{38}N_2O$	5-methyl-5'-[2-(4-dodecyloxyphenyl)ethyl]-2,2'-bipyridine					[48]	
	Sol/Smec	333.9	25.8	77.27			
	Smec/Smec	346.0	4.5	13.01			
	Smec/Nem	361.8	0.5	1.38			
	Nem/Liq	379.8	4.6	12.11	103.77		
$C_{29}H_{38}N_2O_2$	5-methyl-5'-[2-(4-decyloxyphenyl)-2-hydroxyethyl]-2,2'-bipyridine					[48]	
	Sol/Smec	337.5	21.39	63.38			
	Smec/Liq	402.1	6.83	16.99	80.37		
$C_{29}H_{38}O_3$	4-pentylphenyl 4-[4-(1-oxopentyl)cyclohexyl]benzoate					[389]	
	Sol/Smec	349.4	76.2	218.09			
	Smec/Smec	393.7	0.85	2.16			
	Smec/Nem	430.8	0.46	1.07			
	Nem/Liq	445.0	0.15	0.34	221.66		
$C_{29}H_{38}O_4$	7-butoxy-3-(4-decyloxyphenyl)-3H-1-benzopyran-4-one					[44]	
	Sol/Smec	380.2	28.0	73.65			
	Smec/Nem	411.2	Not given in paper				
	Nem/Liq	417.0	1.9	4.56			
$C_{29}H_{39}NO_3$	6-hexyloxynaphth-2-yl 5-heptyloxy-2-methylpyridyl ketone						
	Sol/Sol	337.2	5.59	16.58			
	Sol/Sol	353.2	0.79	2.24			
	Sol/Smec	376.2	34.50	91.71			

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated) (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{29}H_{39}NO_5$	Smec/Liq	384.2	6.85	17.83	128.36	[123]
	Sol/Smec	311.5	50.26	161.35		
	Smec/Smec	343.6	Not reported in paper			
	Smec/Liq	365.7	6.83	18.68	180.03	[294]
$C_{29}H_{39}NO_5$	Sol/Smec	333.2	38.72	116.21		
	Smec/Smec	342.2	Not reported in paper			
	Smec/Liq	369.6	5.53	14.96	131.17	[294]
$C_{29}H_{39}N_3O_2S$	Sol/Smec	379.3	21.48	56.63		
	Smec/Nem	413.8	2.28	5.51		
	Nem/Liq	427.4	1.31	3.07	65.21	[167]
$C_{29}H_{39}N_3O_2S$	Sol/Smec	389.3	33.4	85.80		
	Smec/Nem	457.5	2.8	6.12		
	Nem/Liq	481.9	1.5	3.11	95.03	[79,396,397]
$C_{29}H_{39}N_3O_2S$	Sol/Smec	381.7	35.90	94.05		
	Smec/Nem	452.4	2.76	6.10		
	Nem/Liq	479.9	1.69	3.52	103.67	[283]
$C_{29}H_{39}N_3O_3S$	Sol/Smec	427.1	23.1	54.09		
	Smec/Nem	519.5	5.0	9.62		
	Nem/Liq	532.8	1.3	2.44	66.15	[31]
$C_{29}H_{39}N_3O_3S$	Sol/Smec	371.2	22.45	60.48		
	Smec/Nem	484.4	4.40	9.08		
	Nem/Liq	496.1	2.81	5.66	75.22	[283]
$C_{29}H_{40}N_2O_2S$	Sol/Nem	349.1	35.6	101.98		
	Nem/Liq	385.4	1.2	3.11	105.09	[41]
$C_{29}H_{40}$	Sol/Nem	371.1	13.7	36.92		
	Nem/Liq	442.8	0.8	1.81	38.73	[436]
$C_{29}H_{40}$	Sol/Nem	421.1	16.0	38.00		
	Nem/Liq	468.1	1.3	2.78		[436]
$C_{29}H_{40}N_2O_3$						
$C_{29}H_{40}N_2O_3$	Sol/Sol	315.6	3.03	9.60		
	Sol/Sol	347.5	5.23	15.05		
	Smec/Smec	371.0	33.62	90.62		
	Smec/Liq	404.4	8.87	21.93	137.20	165.4
$C_{29}H_{40}N_2O_3$						[157]
$C_{29}H_{40}N_4OS$						
$C_{29}H_{40}N_4OS$	Sol/Nem	435.3	51.95	119.34		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$\text{C}_{29}\text{H}_{40}\text{N}_4\text{O}_2$	Nem/Liq	451.5	0.79	1.75	121.09	[396]
	Sol/Nem	394.7	16.0	40.54		
	Nem/Liq	435.9	1.5	3.44	43.98	
$\text{C}_{29}\text{H}_{40}\text{N}_4\text{O}_3$		1-[4-[(1E)-(4-formylphenyl)azo]phenyl]-4-(1-oxododecyl)piperazine				
	Sol/Smec	374.9	29.56	78.85		
	Smec/Smec	390.6	0.69	1.77		
	Smec/Smec	407.3	1.18	2.89		
$\text{C}_{29}\text{H}_{40}\text{O}_3$	Smec/Liq	476.6	11.28	23.67	107.18	[344]
		4-[(1E)-[4-[4-(1-oxohexyl)-1-piperazinyl]phenyl]azo]benzoic acid, hexyl ester				
	Sol/Chol	353.2	25.01	70.81		
$\text{C}_{29}\text{H}_{40}\text{O}_3$	Chol/Liq	443.2	Not reported in paper			[208]
		4-( <i>trans</i> -4-pentylcyclohexyl)phenyl 4-[[S]-4-methylhexyl]oxy]benzoate				
	Sol/Chol	373.2	23.27	62.35		
$\text{C}_{29}\text{H}_{40}\text{O}_4$	Chol/Liq	434.2	Not reported in paper			[208]
		4-[(S)-2-methylbutoxy]phenyl 4-(10-undecenoxy)benzoate				
	Sol/Smec	321.2	32.9	102.43		
$\text{C}_{29}\text{H}_{40}\text{O}_4$	Smec/Liq	333.2	Not reported in paper			[200]
		4-[(S)-4-methylhexyloxy]phenyl 4-(8-nonenyloxy)benzoate				
	Sol/Smec	314.2	24.1	76.70		
$\text{C}_{29}\text{H}_{40}\text{O}_4$	Smec/Liq	341.2	Not reported in paper			[200]
		4-[(S)-6-methyloctyloxy]phenyl 4-(6-heptenoxy)benzoate				
	Sol/Smec	315.2	31.5	99.94		
$\text{C}_{29}\text{H}_{40}\text{O}_4$	Smec/Chol	347.2	Not reported in paper			[200]
	Chol/Liq	348.2	Not reported in paper			
		4-(8-nonenyloxy)phenyl 4-[(S)-4-methylhexyloxy]benzoate				
$\text{C}_{29}\text{H}_{40}\text{O}_4$	Sol/Smec	293.2	8.1	27.63		[200]
	Smec/Chol	314.2	Not reported in paper			
	Chol/Liq	336.2	Not reported in paper			
$\text{C}_{29}\text{H}_{41}\text{BrN}_2\text{O}_4$		4-(4-undecyloxyphenylazoxy)phenyl 2S,3S-2-bromo-3-methylpentanoate				
	Sol/Smec	326.8	37.62	115.12		
	Smec/Nem	337.3	0.17	0.50		
$\text{C}_{29}\text{H}_{41}\text{BrO}_3\text{S}$	Nem/Liq	347.5	2.25	6.47	122.09	[47]
		S-(2-bromo-4-octyloxyphenyl) 4-octyloxythiobenzoate				
	Sol/Nem	307.2	46.2	150.39		
$\text{C}_{29}\text{H}_{41}\text{ClN}_2\text{O}_4$	Nem/Liq	308.2	2.0	6.49	156.88	NA
		4-(4-undecyloxyphenylazoxy)phenyl 2S,3S-2-chloro-3-methylpentanoate				
$\text{C}_{29}\text{H}_{41}\text{ClO}_2\text{S}$	Sol/Smec	319.7	35.94	112.42		
	Smec/Smec	342.1	0.31	0.91		
	Smec/Nem	349.2	0.16	0.46		
$\text{C}_{29}\text{H}_{41}\text{ClO}_3\text{S}$	Nem/Liq	350.8	3.41	9.72	123.51	[47]
		4-chlorophenyl 4-hexadecyloxythiobenzoate				
	Sol/Smec	358.9	58.83	163.92		
$\text{C}_{29}\text{H}_{41}\text{ClO}_3\text{S}$	Smec/Liq	365.7	5.69	15.56	179.48	NA
		S-(2-chloro-4-octyloxyphenyl) 4-octyloxythiobenzoate				
	Sol/Nem	316.2	50.6	160.03		
$\text{C}_{29}\text{H}_{41}\text{FO}_3\text{S}$	Nem/Liq	317.2	2.6	8.20	168.23	NA
		S-(2-fluoro-4-octyloxyphenyl) 4-octyloxythiobenzoate				

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.		
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$				
$\text{C}_{29}\text{H}_{41}\text{NO}_3$	Sol/Nem	317.2	44.6	140.61	NA	[4]		
	Nem/Liq	347.2	1.9	5.47	146.08			
$\text{C}_{29}\text{H}_{41}\text{NO}_3$	4-hexadecyloxyphenyl 4-cyanobenzoate							
	Sol/Smec	361.4	49.4	136.69	183.4	[144]		
	Smec/Liq	380.0	5.9	15.53	152.22			
$\text{C}_{29}\text{H}_{41}\text{NO}_3$	4-cyanophenyl 4-hexadecyloxybenzoate							
	Sol/Smec	358.6	61.1	170.38	183.4	[144]		
	Smec/Liq	369.5	5.7	15.42	185.80			
$\text{C}_{29}\text{H}_{41}\text{NO}_5$	4'-hexadecyloxy-3'-nitrobiphenyl-4-carboxylic acid							
	Sol/Sol	320.7	0.6	1.87	191.0	[133]		
	Sol/Sol	347.7	2.3	6.61				
	Sol/Sol	362.6	0.2	0.55				
	Sol/Smec	398.2	38.6	96.94				
	Smec/Cube	447.7	0.5	1.12				
	Cube/Smec	471.1	1.5	3.18				
	Smec/Liq	472.0	0.9	1.91	112.18			
Independent values from another reference								
Sol/Sol	369.2	16.59	44.93					
Note: Value includes two neighboring Sol/Sol transition enthalpies.								
Sol/Smec	401.1	38.92	97.03					
Smec/Smec	445.9	0.73	1.64					
Smec/Smec	470.8	1.90	4.04					
Smec/Liq	472.0	4.65	9.85	157.49	191.0	[40]		
Independent values from another reference								
Sol/Sol	352.0	4.29	12.18					
Sol/Smec	401.0	42.67	106.41					
Smec/Smec	450.0	0.63	1.40					
Smec/Smec	472.0	1.37	2.90					
Smec/Liq	475.0	6.65	14.00	136.89	191.0	[87]		
$\text{C}_{29}\text{H}_{41}\text{N}_3\text{OS}$	5-[4-(octyloxy)phenyl]-N-[(5-octyl-2-thienyl)methylene]-1,2,4-thiadiazole-2-amine							
	Sol/Smec	396.6	22.5	56.73	72.56	[396,397]		
	Smec/Liq	410.6	6.5	15.83				
$\text{C}_{29}\text{H}_{41}\text{N}_3\text{O}_2\text{S}$	6-n-decyloxy-2-(4-hexyloxyphenylazo)benzothiazole							
	Sol/Sol	338.6	6.8	20.08	[41]			
	Sol/Smec	321.5	Not given in paper					
	Smec/Nem	348.4	34.5	99.02				
	Nem/Liq	394.9	1.1	2.79				
$\text{C}_{29}\text{H}_{42}\text{N}_2\text{O}_2$	4-[5-[(S)-6-methyloctyl]-2-pyrimidinyl]phenyl <i>trans</i> -4-propylcyclohexane-1-carboxylate							
	Sol/Chol	371.2	12.1	32.60	Not reported in paper	[201]		
	Chol/Liq	414.2	Not reported in paper					
$\text{C}_{29}\text{H}_{42}\text{N}_2\text{O}_3$	4-dodecanoxy-2,3-dimethyl-4'-ethoxyazobenzene				130.32	[339]		
	Sol/Nem	352.2	45	127.77				
	Nem/Liq	352.2	0.9	2.55				
$\text{C}_{29}\text{H}_{42}\text{N}_4\text{OS}_2$	2-(4-decyloxyphenylazo)-5-(5'-heptyl-2'-thienyl)-1,3,4-thiadiazole				90.42	[396]		
	Sol/Nem	427.0	35.9	84.07				
	Nem/Liq	433.0	2.75	6.35				
$\text{C}_{29}\text{H}_{42}\text{O}_2\text{S}$	4-pentylbenzenethio-4'-undecyloxybenzoate				123.88	[217]		
	Sol/Smec	338.1	38.12	112.75				
	Smec/Nem	358.3	1.95	5.44				
	Nem/Liq	360.0	2.05	5.69				

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{29}H_{43}NO$	Sol/Smec	328.0	37.32	113.78	148.04	[65]
	Smec/Smec	340.0	3.79	11.15		
	Smec/Liq	348.3	8.05	23.11		
$C_{29}H_{43}NO$	N-(4-tetradecyloxybenzylidene)-4-ethylaniline					
	Sol/Smec	328.3	29.36	89.43		
	Smec/Smec	345.8	0.07	0.20		
	Smec/Smec	348.5	4.68	13.43		
$C_{29}H_{43}NO_2$	Smec/Liq	358.8	8.06	22.46	125.52	[256]
	N-(4-dodecyloxybenzylidene)-4-butylaniline					
	Sol/Smec	328.3	29.36	89.43		
	Smec/Smec	345.8	0.07	0.20		
$C_{29}H_{43}NO_2$	Smec/Smec	348.5	4.68	13.43		
	Smec/Liq	358.8	8.06	22.46	125.52	[256]
	<i>trans</i> -4-hexadecyloxy-3'-stilbazole-N-oxide					
	Sol/Smec	375.3	43.05	114.71		
$C_{29}H_{43}NO_2$	Smec/Liq	400.1	3.50	8.75	123.46	[278]
	<i>trans</i> -4-hexadecyloxy-4'-stilbazole-N-oxide					
	Sol/Smec	373.8	42.42	113.48		
	Smec/Liq	406.7	3.88	9.54	123.02	[278]
$C_{29}H_{44}N_2O$	5-pentyl-2-[4-[3-[( <i>trans</i> -4-pentylcyclohexyl)oxy]propyl]phenyl]pyrimidine					
	Sol/Nem	355.2	19.2	54.05		
	Nem/Liq	412.2	Not reported in paper			[198]
$C_{29}H_{44}N_2O$	4-propyl-4'-tetradecyloxyazobenzene					
	Sol/Smec	344.8	113.7	329.76		
	Smec/Liq	352.7	26.36	74.73	404.49	169.9
$C_{29}H_{44}N_2O$	4-pentyl-4'-dodecyloxyazobenzene					
	Sol/Smec	333.6	27.63	82.82		
	Smec/Smec	341.0	2.96	8.68		
	Smec/Nem	357.4	8.40	23.50		
	Nem/Liq	358.0	Not detected by dsc		115.00	169.9
Note: Nem/Liq transition enthalpy is likely included in the Smec/Nem value.						
$C_{29}H_{44}N_2O$	4-heptyl-4'-decyloxyazobenzene					
	Sol/Smec	310.6	30.22	97.30		
	Smec/Smec	328.5	1.61	4.90		
	Smec/Nem	353.6	2.19	6.19		
$C_{29}H_{44}N_2O$	Nem/Liq	355.6	2.62	7.37	115.76	169.9
	2-[4-(10-undecenoxy)phenyl]-5-[(S)-5-methylheptyl]pyrimidine					
	Sol/Smec	293.2	11.1	37.86		
	Smec/Chol	313.2	Not reported in paper			
$C_{29}H_{44}N_2O$	Chol/Liq	322.2	Not reported in paper			[201]
	2-[4-(9-decenoxy)phenyl]-5-[(S)-6-methyloctyl]pyrimidine					
	Sol/Smec	285.2	21.7	76.09		
	Smec/Chol	300.2	Not reported in paper			
$C_{29}H_{46}N_2O_6$	Chol/Liq	311.2	Not reported in paper			[201]
	4,4'- <i>bis</i> [4-(4-butoxybenzylideneamino)benzoyloxy]diphenylmethane					
	Sol/Smec	448.2	27.0	60.24		
	Smec/Liq	518.2	17.0	32.81	93.05	[284]
$C_{29}H_{46}O_3$	4-heptylcyclohexyl 4-heptyloxy cinnamate					
	Sol/Nem	350.4	40.8	116.44		
	Nem/Liq	365.7	1.6	4.38	120.82	[5]
$C_{29}H_{48}OS$	cholesteryl thioacetate					
	Sol/Liq	399.2	27.3	68.4	68.4	NA
	Chol/Liq	385.2	0.3	0.8		[155,312]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
Note: Liquid crystalline behavior was observed upon cooling.						
$\text{C}_{29}\text{H}_{48}\text{O}_6$	Sol/Meso	336.2	16.6	49.38		
		368.7	0.9	2.44	51.82	[98]
	Meso/Liq					
$\text{C}_{29}\text{H}_{50}\text{N}_2\text{O}_2$	Sol/Meso	394.2	30.0	76.10		
		477.2	17.0	35.62	111.72	[36]
	Meso/Liq					
$\text{C}_{30}\text{H}_{20}\text{F}_2\text{N}_2\text{O}_7$	Sol/Nem	460.0	47.11	102.41		
						[264]
	Nem/Liq					
$\text{C}_{30}\text{H}_{21}\text{F}_{21}\text{O}_5$	Sol/Smec	349.5	32.20	92.13		
		350.2	0.34	0.97		
		368.4	4.32	11.73	104.83	[128]
$\text{C}_{30}\text{H}_{21}\text{F}_{21}\text{O}_5$	Smec/Smec	349.5	32.20	92.13		
		350.2	0.34	0.97		
		368.4	4.32	11.73	104.83	[128]
$\text{C}_{30}\text{H}_{21}\text{F}_{25}\text{O}_4$	Smec/Smec	349.5	32.20	92.13		
		350.2	0.34	0.97		
	Smec/Liq	368.4	4.32	11.73	104.83	[128]
$\text{C}_{30}\text{H}_{21}\text{F}_{25}\text{O}_4$	Sol/Smec	427.2	18.6	43.54		
		461.2	2.0	4.34	47.88	[404]
	Smec/Liq					
$\text{C}_{30}\text{H}_{22}$	p-quinquephenyl					
	Sol/Nem	661.2	44.0	66.55		
	Nem/Liq	693.2	0.70	1.01	67.56	102.8
$\text{C}_{30}\text{H}_{22}$	Independent values from another reference					
	Sol/Nem	659.6	42.30	64.13		
	Nem/Liq	688.1	0.92	1.34	65.47	102.8
$\text{C}_{30}\text{H}_{22}$	Independent values from another reference					
	Sol/Sol	264.0	0.45	1.70		
						[386]
Note: Authors reported results of low temperature adiabatic calorimetric measurements from 6 to 303 K.						
$\text{C}_{30}\text{H}_{22}\text{F}_{20}\text{O}_5$	4-(2,2,3,3,4,4,4-heptafluorobutyloxycarbonyl)phenyl 4-[(perfluorohexyl)hexyloxy]benzoate					
	Sol/Smec	378.0	47.24	124.97		
	Smec/Liq	380.3	4.35	11.44	136.41	[128]
$\text{C}_{30}\text{H}_{26}\text{F}_4\text{O}_5$	4-[(S)-2-methylbutoxycarbonyl]phenyl 4-[4-propoxy-2,3,4,5-tetrafluorophenyl]ethynyl]benzoate					
	Sol/Smec	366.8	24.19	65.95		
	Smec/Chst	391.2	0.29	0.74		
$\text{C}_{30}\text{H}_{26}\text{O}_8$	Chst/Liq	437.6	0.44	1.01	67.70	[91]
	Sol/Nem	464.5	62.6	134.77		
	Nem/Liq	588.1	Not reported in paper			[271]
$\text{C}_{30}\text{H}_{26}\text{O}_8$	1,4-benzenedicarboxylic acid, bis[4-[3-(ethoxy)-3-oxo-1-propenyl]phenyl ester					
	Sol/Sol	443.3	5.8	13.08		
	Sol/Smec	463.2	40.0	86.36		
	Smec/Nem	502.8	0.7	1.39		
$\text{C}_{30}\text{H}_{27}\text{BrN}_2\text{O}_4$	Nem/Liq	Not observed by dsc		100.83		[271]
		3'-bromo-4'-methoxy- [1,1'-biphenyl]-4-carboxylic acid, 4-[(4-butoxyphenyl)azo]phenyl ester				

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated) (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$\text{C}_{30}\text{H}_{27}\text{ClF}_4\text{O}_3$	Sol/Nem	449.5	23.15	51.50	52.50	[57]
	Nem/Liq	587.8	0.59	1.00		
$\text{C}_{30}\text{H}_{27}\text{NO}$	Sol/Nem	385.4	30.89	80.15	82.47	[101]
	Nem/Liq	435.5	1.01	2.32		
$\text{C}_{30}\text{H}_{27}\text{NO}$	Sol/Nem	379.2	39.0	102.85	104.36	[82]
	Nem/Liq	464.2	0.7	1.51		
$\text{C}_{30}\text{H}_{28}\text{F}_4\text{O}_4$	Sol/Nem	422.2	37.6	89.06	90.21	[82]
	Nem/Liq	433.2	0.5	1.15		
$\text{C}_{30}\text{H}_{28}\text{N}_6\text{O}_6$	Sol/Nem	371.9	22.20	59.69	61.00	[124]
	Nem/Liq	435.4	0.57	1.31		
$\text{C}_{30}\text{H}_{29}\text{F}_{13}\text{O}_5$	Sol/SmeC	476.2	46.72	98.11	104.01	[422]
	SmeC/SmeC	500.2	2.95	5.90		
$\text{C}_{30}\text{H}_{30}$	Sol/Nem	359.4	28.14	78.30	100.65	[125]
	Nem/Liq	395.5	0.21	0.53		
$\text{C}_{30}\text{H}_{30}$	Sol/Nem	403.8	8.81	21.82	62.58	[55, 430]
	Nem/Liq	435.9	0.97	2.23		
$\text{C}_{30}\text{H}_{30}$	Sol/Nem	406.8	21.2	60.35	39.38	[55]
	Nem/Liq	464.3	1.05	2.26		
$\text{C}_{30}\text{H}_{30}$	Sol/Nem	451.3	18.9	50.94	53.08	[55, 430]
	Nem/Liq	453.1	0.97	2.14		
$\text{C}_{30}\text{H}_{30}$	Sol/Nem	371.0	22.0	59.19	62.20	[55, 430]
	Nem/Liq	464.4	1.40	3.01		
$\text{C}_{30}\text{H}_{30}$	Sol/Nem	334.8	27.0	78.76	81.23	[55]
	Nem/Liq	446.7	0.98	2.47		
$\text{C}_{30}\text{H}_{30}\text{FNO}_5\text{S}$	Sol/Nem	342.8	14.8	44.21	46.31	[55, 430]
	Nem/Liq	397.0	0.94	2.10		
$\text{C}_{30}\text{H}_{30}\text{O}_2$	Sol/SmeC	374.2	41.3	110.37	112.24	[37]
	SmeC/SmeC	375.2	Too small to be measured			
$\text{C}_{30}\text{H}_{30}\text{O}_2$	Sol/Nem	423.2	0.3	0.71		
	Nem/Liq	430.3	0.5	1.16		
$\text{C}_{30}\text{H}_{30}\text{O}_2$	Sol/Sol	529.0	8.46	15.99	16.01	
	Sol/SmeC	631.0	10.10			

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated) (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{30}H_{30}O_5$	Smec/Liq	680.0	Not reported in paper			[111]
	Sol/Smec	421.1	29.10	69.10		
	Smec/Nem	464.5	1.71	3.68		
	Nem/Liq	472.3	1.14	2.41	75.19	[14]
$C_{30}H_{30}O_6$	<i>bis</i> (4-hexyloxyphenyl) terephthalate					
	Sol/Sol	401.0	6.0	14.96		
	Sol/Smec	433.0	38.7	89.38		
	Smec/Smec	447.5	0.2	0.45		
	Smec/Nem	453.3	0.4	0.88		
$C_{30}H_{30}O_6S_2$	Nem/Liq	479.3	1.4	2.92	108.59	174.8
	<i>bis</i> (4-butoxyphenyl) 2,2'-bithiophene-5,5'-dicarboxylate					
	Sol/Sol	379.4	13.2	34.79		
	Sol/Smec	443.8	30.6	68.95		
	Smec/Nem	505.8	1.3	2.57		
$C_{30}H_{30}O_8$	Nem/Liq	523.3	1.0	1.91	108.22	163.6
	<i>bis</i> (4-butoxycarbonylphenyl) terephthalate					
	Sol/Smec	410.2	31.0	75.57		
$C_{30}H_{31}NO$	Smec/Liq	455.2	4.0	8.79	84.36	152.4
	5-(4'-hexyloxybiphenyl-4-yl)-2-(4-methylphenyl)pyridine					
	Sol/Sol	435.0	9.9	22.76		
	Sol/Smec	514.0	4.1	7.98		
	Smec/Smec	546.0	4.6	8.42		
$C_{30}H_{31}NO_5S$	Smec/Nem	590.0	1.5	2.54		
	Nem/Liq	642.0	1.1	1.71	43.41	[185]
	4-thiocyanophenyl 4-(4-nonyloxybenzoyloxy)benzoate					
	Sol/Smec	378.3	35.3	93.31		
	Smec/Liq	453.2	4.1	9.05	102.36	[114]
$C_{30}H_{32}N_2O_4$	dibutyl N,N'-[1,4-phenylene- <i>bis</i> (methylidyne)]- <i>bis</i> [aminobenzoate]					
	Sol/Smec	365.2	32.0	87.62		
	Smec/Smec	410.2	Not reported in paper			
	Smec/Nem	463.2	0.84	1.81		
	Nem/Liq	482.2	0.38	0.08		[192]
$C_{30}H_{32}N_3O_2S$	5-(4-hexyloxyphenyl)-5-(4-decyloxy)phenylazo-1,3,4-thiadiazole					
	Sol/Nem	428.8	50.06	116.74		
	Nem/Liq	452.2	0.66	1.46	118.20	[79]
$C_{30}H_{32}O_7$	2-acetyl-1,4-phenylene 4-butoxybenzoate					
	Sol/Nem	383.0	34.6	90.34		
	Nem/Liq	428.2	2.2	5.14	95.48	[432]
$C_{30}H_{33}ClO_2$	1-(4-nonylbiphenyl)-3-(4-chlorophenyl)propane-1,3-dione					
	Sol/Smec	409.7	27.61	67.39		
	Smec/Liq	419.2	4.60	10.97	78.36	[250]
$C_{30}H_{33}ClO_4$	4-pentylphenyl 2-chloro[(4-pentylbenzoyl)oxy]benzoate					
	Sol/Nem	312.8	23.8	76.09		
	Nem/Liq	396.2	1.18	2.98	79.08	[365]
$C_{30}H_{34}F_8O_5$	4-(2,2,3,3,4,4,5,5-octafluoropentyloxycarbonyl)phenyl 4-(undecyloxy)benzoate					
	Sol/Smec	324.1	15.84	48.87		
	Smec/Smec	325.1	0.31	0.95		
	Smec/Liq	330.4	5.32	16.10	65.92	[125]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{30}H_{34}O_4$	Sol/Sol	318.0	18.90	59.43		
	Sol/Nem	402.0	25.50	63.43		
	Nem/Liq	411.0	1.17	2.85	125.71	139.6 [157]
$C_{30}H_{34}O_7$	Sol/Nem	355.8	9.44	26.53		
	Nem/Liq	422.2	0.38	0.90	27.43	[103]
$C_{30}H_{35}F_7O_5$	Sol/Smec	338.9	27.00	79.67		
	Smec/Smec	346.2	0.36	1.04		
	Smec/Liq	348.8	6.02	17.26	97.97	[125]
$C_{30}H_{35}NO_3$	Sol/Smec	381.2	13.61	35.70		
	Smec/Nem	391.2	1.04	2.66		
	Nem/Liq	455.2	0.65	1.43	39.79	[273]
$C_{30}H_{35}NO_3$	Sol/Smec	365.2	48.0	131.43		
	Smec/Liq	424.2	5.1	12.02	143.45	154.8 [292]
$C_{30}H_{35}NO_3$	Sol/Smec	360.2	35.0	97.17		
	Smec/Nem	446.2	1.1	2.47		
	Nem/Liq	479.2	1.2	2.50	102.14	155.4 [292]
$C_{30}H_{35}NO_3$	Sol/Smec	370.2	37.0	99.95		
	Smec/Nem	436.2	1.0	2.29		
	Nem/Liq	453.2	1.1	2.43	104.67	135.3 [292]
$C_{30}H_{35}NO_3$	Sol/Nem	369.2	31.0	83.97		
	Nem/Liq	482.2	1.7	3.53	87.50	148.3 [292]
$C_{30}H_{35}NO_3$	Sol/Nem	378.2	55.0	145.43		
	Nem/Liq	419.2	1.0	2.39	147.82	154.8 [292]
$C_{30}H_{35}NO_3$	Sol/Nem	343.8	29.26	85.11		
	Nem/Liq	384.8	0.63	1.64	86.75	[399]
$C_{30}H_{36}$	Sol/Nem	363.6	13.6	37.40		
	Nem/Liq	471.4	1.3	2.76	40.16	[436]
$C_{30}H_{36}$	Sol/Nem	367.1	17.5	47.67		
	Nem/Liq	465.7	1.5	3.22	50.89	[436]
$C_{30}H_{36}$	Sol/Nem	389.3	13.1	33.65		
	Nem/Liq	479.9	1.1	2.29	35.94	[436]
$C_{30}H_{36}N_2O_4$	Sol/Nem	361.2	50	138.43		
	Nem/Liq	474.2	1.7	3.58	142.01	[339]
$C_{30}H_{36}N_2O_4$						

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$\text{C}_{30}\text{H}_{36}\text{O}_5$	Sol/Nem	365.2	42	115.01		[339]
	Nem/Liq	468.2	1.6	3.41	118.42	
$\text{C}_{30}\text{H}_{36}\text{O}_5$		2-hydroxyethyl 4-[[[4'-(octyloxy)[1,1'-biphenyl]-4-yl]oxy]methyl]benzoate				
	Sol/Sol	414.4	12.68	30.60		[304]
	Sol/SmeC	444.9	6.20	13.94		
	SmeC/SmeC	455.9	4.57	10.02		
	SmeC/SmeC	457.4	0.24	0.52		
	SmeC/Liq	470.5	18.73	39.81	94.89	
$\text{C}_{30}\text{H}_{36}\text{O}_6\text{S}$		<i>bis</i> (4-hexyloxyphenyl) 2,5-thiophenedicarboxylate				
	Sol/Sol	379.2	11.2	29.53		[12]
	Sol/Nem	403.0	40.0	99.26		
	Nem/Liq	416.8	1.0	2.40	131.19	
$\text{C}_{30}\text{H}_{36}\text{O}_7$		4-butoxyphenyl 7-decanoyloxychromone-2-carboxylate				
	Sol/SmeC	397.0	20.0	50.38		[286]
	SmeC/SmeC	423.2	Not observed by dsc			
	SmeC/Liq	434.6	3.5	8.05	58.43	
$\text{C}_{30}\text{H}_{37}\text{N}_5\text{O}_3$		2-cyano-3-[4-[1E]-[4-[4-(1-oxohexyl)-1-piperazinyl]phenyl]azo]phenyl]-2-propenoic acid, butyl ester				
	Sol/SmeC	459.9	23.61	51.34		[326]
	SmeC/Liq	485.2	Not observed by dsc			
$\text{C}_{30}\text{H}_{38}\text{N}_2\text{O}_2\text{S}$		2-(4-butoxyphenyl)-5-[4-[(2-methyl-2,3-undecadienyl)oxy]phenyl]-1,3,4-thiadiazole				
	Sol/SmeC	333.2	27.4	82.23		[398]
	SmeC/Nem	364.2	1.54	4.23		
	Nem/Liq	374.2	0.89	2.38	88.84	
$\text{C}_{30}\text{H}_{38}\text{N}_2\text{O}_3$		1-[4-(N-ethoxycarbonylpiperazinyl)phenyl]-4-(4-heptyloxyphenyl)but-3-en-1-yne				
	Sol/SmeC	448.5	32.7	72.91		[154]
	SmeC/Nem	475.2	2.0	4.21		
	Nem/Liq	483.6	1.4	2.89	80.01	
$\text{C}_{30}\text{H}_{39}\text{ClO}_5$		2-chloro-4-methylpentyl 4-[[3-[4-(octyloxy)phenyl]-1-oxo-2-propenyl]oxy]benzoate				
	Sol/SmeC	323.2	24.27	75.09		[257]
	SmeC/SmeC	329.7	0.33	1.00		
	SmeC/Chol	352.7	0.88	2.50		
	Chol/Liq	353.4	0.67	1.90	80.49	
$\text{C}_{30}\text{H}_{39}\text{ClO}_5$		2-chloro-3-methylpentyl 4-[[3-[4-(octyloxy)phenyl]-1-oxo-2-propenyl]oxy]benzoate				
	Sol/SmeC	330.2	30.54	92.49		[257]
	SmeC/SmeC	332.2	0.17	0.51		
	SmeC/Chol	357.7	1.26	3.52		
	Chol/Liq	358.2	0.84	2.35	98.87	
$\text{C}_{30}\text{H}_{39}\text{ClO}_5$		4-[[3-[4-(nonyloxy)phenyl]-1-oxo-2-propenyl]oxy]phenyl 2-chloro-4-methylpentanoate				
	Sol/SmeC	346.2	25.65	74.09		[257]
	SmeC/SmeC	348.7	0.25	0.72		
	SmeC/Liq	363.7	4.18	11.49	86.30	
$\text{C}_{30}\text{H}_{39}\text{ClO}_5$		4-[[3-[4-(nonyloxy)phenyl]-1-oxo-2-propenyl]oxy]phenyl 2-chloro-3-methylpentanoate				
	Sol/SmeC	342.2	30.54	89.25		[257]
	SmeC/SmeC	351.2	0.25	0.71		
	SmeC/Chol	368.2	0.21	0.57		
	Chol/Liq	370.2	0.96	2.59	93.12	
$\text{C}_{30}\text{H}_{39}\text{NO}_2$		6-n-pentyloxy-2-[4-octyloxystyryl]quinoline				
	Sol/Sol	366.1	7.15	19.53		[112]
	Sol/SmeC	375.0	21.45	57.20		
	SmeC/Nem	402.4	1.99	4.95		
	Nem/Liq	436.9	0.98	2.24	83.92	

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{30}H_{39}NO_2S$			4-isothiocyanatophenyl 4-( <i>trans</i> -4-decyloxyphenyl)benzoate			
	Sol/Sol	357.2	15.48	43.34		
	Sol/Smec	368.7	32.38	87.82		
	Smec/Nem	453.2	0.42	0.93		
$C_{30}H_{39}N_3O_3$	Nem/Liq	473.2	0.71	1.50	133.59	NA
			1-[4-(N-ethoxycarbonylpiperazinyl)phenyl]-3-(4-octyloxyphenyl)(1-propargyl-3-imine)			
	Smec/Smec	407.4	19.0	46.64		
	Smec/Liq	455.3	4.1	9.01	55.65	[154]
$C_{30}H_{40}N_2O_2$			6-n-decyloxy-2-[ <i>(4'-N-butoxyphenylimino)methyl</i> ]quinoline			
	Smec/Smec	352.3	34.37	97.56		
	Smec/Nem	372.9	0.19	0.51		
	Nem/Liq	411.1	1.48	3.60	101.67	[112]
$C_{30}H_{40}N_2O_3$			N-4-butoxyphenyl-6-decyloxyquinoline-2-carboxamide			
	Smec/Smec	367.0	45.6	124.25		
	Smec/Liq	384.4	5.1	13.27	137.52	[43]
			4-[ <i>(1E)-[4-[4-(1-oxo-10-undecenyl)-1-piperazinyl]phenyl]azo</i> ]benzoic acid, ethyl ester			
$C_{30}H_{40}N_4O_3$	Smec/Smec	341.4	2.12	6.21		
	Smec/Smec	353.6	0.35	0.99		
	Smec/Smec	395.9	0.50	1.26		
	Smec/Smec	434.3	0.25	0.58		
	Smec/Liq	473.6	8.13	17.17	26.21	[403]
$C_{30}H_{40}O_4$			7-pentyloxy-3-(4-decyloxyphenyl)-3 <i>H</i> -1-benzopyran-4-one			
	Smec/Smec	385.2	28.1	72.95		
	Smec/Liq	415.5	4.5	10.83	83.78	[44]
$C_{30}H_{40}O_4$			di(4'-pentyloxyphenyl)- <i>trans</i> -cyclohexane-1,4-dicarboxylate			
	Smec/Smec	364.2	35.2	96.65		
	Smec/Smec	368.1	2.06	5.60		
	Smec/Nem	401.9	0.65	1.62		
	Nem/Liq	426.5	0.85	1.99	105.86	147.5
$C_{30}H_{40}O_6$			di(4'-pentyloxyphenyl)- <i>trans</i> -cyclohexane-1,4-dicarboxylate			
	Smec/Smec	374.2	29.08	77.71		
	Smec/Smec	378.2	1.16	3.07		
	Smec/Nem	446.2	0.73	1.64		
	Nem/Liq	466.2	1.04	2.23	84.65	161.1
$C_{30}H_{41}NO_3$			6-hexyloxyphenyl-2-yl 5-octyloxy-2-methylpyridyl ketone			
	Smec/Smec	326.2	0.86	2.64		
	Smec/Smec	347.2	4.67	13.45		
	Smec/Smec	372.2	35.12	94.36		
	Smec/Liq	384.2	9.60	24.99	135.44	[123]
$C_{30}H_{41}NO_3$			4'-hexadecyloxy-3'-cyanobiphenyl-4-carboxylic acid			
	Smec/Smec	385.7	26.34	68.29		
	Smec/Smec	397.2	14.98	37.71		
	Smec/Smec	399.2	21.64	54.21		
	Smec/Cube	431.2	1.15	2.67		
	Cube/Liq	474.3	3.96	8.35	171.23	176.8
$C_{30}H_{41}NO_5$			6-[4-(4-heptyloxyphenyliminomethyl)-3-hydroxyphenoxy]hexylmethacrylate			
	Smec/Smec	324.6	27.72	85.40		
	Smec/Smec	343.0	Not reported in paper			
	Smec/Liq	364.6	5.00	13.71	99.11	[294]
$C_{30}H_{41}N_3O_2S$		5-(4-decyloxyphenyl)-N-[[4-(2S)-2-methylbutoxy]phenyl]methylene-1,3,4-thiadiazol-2-amine				

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$\text{C}_{30}\text{H}_{41}\text{N}_3\text{O}_2\text{S}$	Sol/Smec	379.0	33.45	88.26	97.15	[167]
	Smec/Nem	416.4	2.33	5.60		
	Nem/Liq	434.3	1.43	3.29		
$\text{C}_{30}\text{H}_{41}\text{N}_3\text{O}_2\text{S}$	Sol/Smec	386.0	31.3	81.09	91.79	[79,396,397]
	Smec/Nem	462.7	3.6	7.78		
	Nem/Liq	479.5	1.4	2.92		
$\text{C}_{30}\text{H}_{41}\text{N}_3\text{O}_2\text{S}$	Sol/Smec	393.2	41.27	104.96	116.82	[283]
	Smec/Nem	460.8	3.56	7.73		
	Nem/Liq	478.4	2.07	4.33		
$\text{C}_{30}\text{H}_{41}\text{N}_3\text{O}_3\text{S}$	Sol/Smec	422.5	19.9	47.10	61.06	[31]
	Smec/Nem	520.7	5.6	10.75		
	Nem/Liq	529.6	1.7	3.21		
$\text{C}_{30}\text{H}_{41}\text{N}_3\text{O}_3\text{S}$	Sol/Smec	373.2	29.68	79.53	89.72	[59]
	Smec/Nem	478.2	3.24	6.78		
	Nem/Liq	487.2	1.66	3.41		
$\text{C}_{30}\text{H}_{41}\text{N}_3\text{O}_3\text{S}$	Sol/Smec	375.3	25.03	66.69	83.13	[283]
	Smec/Nem	487.1	4.82	9.90		
	Nem/Liq	492.6	3.22	6.54		
$\text{C}_{30}\text{H}_{42}$	Sol/Nem	389.2	19.8	50.87	54.07	[436]
	Nem/Liq	468.9	1.5	3.20		
	Sol/Nem	411.9	9.3	22.57		
$\text{C}_{30}\text{H}_{42}\text{N}_2\text{O}_2\text{S}$	Nem/Liq	463.6	0.9	1.94	24.51	[436]
	Sol/Smec	355.9	39.4	110.71	115.92	[41]
	Smec/Nem	362.1	0.6	1.66		
$\text{C}_{30}\text{H}_{42}\text{N}_2\text{O}_2\text{S}_2$	Nem/Liq	394.3	1.4	3.55		
	Sol/Sol	383.5	29.26	76.30	187.58	[269]
	Smec/Nem	426.3	37.88	88.86		
$\text{C}_{30}\text{H}_{42}\text{N}_2\text{O}_3$	Smec/Smec	510.3	8.47	16.60		
	Smec/Liq	517.2	3.01	5.82		
	Sol/Sol	361.5	6.09	16.85		
$\text{C}_{30}\text{H}_{42}\text{N}_2\text{O}_3$	Smec/Liq	371.6	37.07	99.76	172.5	[157]
	Smec/Liq	402.9	9.23	22.91		
	Sol/Smec	376.7	51.51	136.74		
$\text{C}_{30}\text{H}_{42}\text{N}_4\text{OS}$	Smec/Liq	412.2	8.45	20.50	157.24	[157]
	Sol/Nem	428.8	50.06	116.74	118.20	[396]
	Nem/Liq	452.2	0.66	1.46		
$\text{C}_{30}\text{H}_{42}\text{O}_4$				4-[(S)-2-methylbutoxy]phenyl 4-(11-dodecenoxy)benzoate		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$\text{C}_{30}\text{H}_{42}\text{O}_4$	Sol/Smec	318.2	27.1	85.17		[200]
	Smec/Liq	331.2		Not reported in paper		
$\text{C}_{30}\text{H}_{42}\text{O}_4$		4-[(S)-4-methylhexyloxy]phenyl 4-(9-decenoxy)benzoate				
	Sol/Smec	302.2	28.9	95.63		[200]
	Smec/Smec	327.2		Not reported in paper		
	Smec/Liq	338.2		Not reported in paper		
$\text{C}_{30}\text{H}_{42}\text{O}_4$		4-[(S)-6-methyloctyloxy]phenyl 4-(7-octenoxy)benzoate				
	Sol/Smec	327.2	34.0	103.91		[200]
	Smec/Smec	329.2		Not reported in paper		
	Smec/Chol	344.2		Not reported in paper		
$\text{C}_{30}\text{H}_{42}\text{O}_4$	Chol/Liq	344.2		Not reported in paper		[200]
		4-(9-decenoxy)phenyl 4-[(S)-4-methylhexyloxy]benzoate				
	Sol/Smec	298.2	9.8	32.86		
	Smec/Chol	313.2		Not reported in paper		
$\text{C}_{30}\text{H}_{42}\text{O}_4$	Chol/Liq	329.2		Not reported in paper		[200]
		4-(7-octenoxy)phenyl 4-[(S)-6-methyloctyloxy]benzoate				
	Sol/Smec	311.2	24.0	77.12		
	Smec/Chol	330.2		Not reported in paper		
$\text{C}_{30}\text{H}_{42}\text{O}_4$	Chol/Liq	338.2		Not reported in paper		[200]
		4-[(trans-4-propylcyclohexyl)methoxy]phenyl 4-[(S)-4-methylhexyl]oxy}benzoate				[208]
	Sol/Chol	352.2	22.36	63.49		
	Chol/Liq	421.2		Not reported in paper		
$\text{C}_{30}\text{H}_{42}\text{O}_4$		4-[(trans-4-pentylcyclohexyl)methoxy]phenyl 4-[(S)-2-methylbutyl]oxy}benzoate				
	Sol/Chol	361.2	23.15	64.09		[208]
	Chol/Liq	414.2		Not reported in paper		
		4-(4-dodecyloxyphenylazoxy)phenyl 2S,3S-2-bromo-3-ethylpentanoate				
$\text{C}_{30}\text{H}_{43}\text{BrN}_2\text{O}_4$	Sol/Smec	330.0	45.82	138.85		[47]
	Smec/Nem	338.8	0.17	0.50		
	Nem/Liq	347.4	2.30	6.62	145.97	
		4-(4-dodecyloxyphenylazoxy)phenyl 2S,3S-2-chloro-3-methylpentanoate				
$\text{C}_{30}\text{H}_{43}\text{ClN}_2\text{O}_4$	Sol/Smec	326.1	2.02	6.19		[47]
		Note: Value is abnormally small compared to other compounds in series.				
	Smec/Smec	339.3	0.16	0.47		
	Smec/Liq	349.3	3.82	10.94	17.60	
$\text{C}_{30}\text{H}_{43}\text{NO}$		N-(4-pentyloxybenzylidene)-4-dodecyaniline				
	Sol/Smec	303.7	38.86	127.96		[147]
	Smec/Smec	327.5	2.23	6.81		
	Smec/Nem	344.6	0.40	1.16		
$\text{C}_{30}\text{H}_{43}\text{N}_3\text{OS}$	Nem/Liq	347.5	1.38	3.97	139.90	
		5-[4-(4-(nonyloxy)phenyl]-N-[5-octyl-2-thienyl)methylene]-1,2,4-thiadiazole-2-amine				
	Sol/Smec	394.5	37.0	93.79		[396,397]
	Smec/Liq	411.9	7.1	17.24	111.03	
$\text{C}_{30}\text{H}_{43}\text{N}_3\text{O}_2\text{S}$		6-n-decyloxy-2-(4-heptyloxyphenylazo)benzothiazole				
	Sol/Smec	347.6	47.0	135.21		[41]
	Smec/Nem	352.7		Not given in paper		
	Nem/Liq	394.4	1.2	3.04		
$\text{C}_{30}\text{H}_{44}\text{N}_2\text{O}$		5-hexyl-2-[4-[3-[(trans-4-pentylcyclohexyl)oxy]-1-propenyl]phenyl]pyrimidine				
	Sol/Nem	356.2	18.1	50.81		[198]
	Nem/Liq	412.2		Not reported in paper		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$\text{C}_{30}\text{H}_{44}\text{N}_2\text{O}_2$	Sol/Chol	405.2	15.4	38.00		[201]
	Chol/Liq	427.2		Not reported in paper		
$\text{C}_{30}\text{H}_{44}\text{N}_2\text{O}_4$						
			1,2-bis(4-n-octyloxybenzoyl)hydrazine			
	Sol/Sol	405.2	21.05	51.95		
	Sol/Cube	410.6	27.6	67.22		
$\text{C}_{30}\text{H}_{44}\text{O}_2\text{S}$	Cube/Smec	433.3	0.98	2.26		
	Smec/Liq	437.5	9.97	22.79	144.22	176.2
						[49]
$\text{C}_{30}\text{H}_{44}\text{O}_2\text{S}$			4-pentylbenzenethio 4'-dodecyloxybenzoate			
	Sol/Smec	335.3	34.10	101.70		
	Smec/Liq	360.2	6.95	19.29	120.99	NA
$\text{C}_{30}\text{H}_{44}\text{O}_3\text{S}$			S-(2-methyl-4-octyloxyphenyl) 4-octyloxythiobenzoate			
	Sol/Nem	302.2	39.9	132.03		
	Nem/Liq	322.2	1.8	5.59	137.62	NA
$\text{C}_{30}\text{H}_{45}\text{NO}$			N-(p-n-nonyloxybenzylidene)-4-(n-octyl)aniline			
	Smec/Smec	323.2	47.6	147.28		
	Smec/Liq	348.5	3.91	11.22		
						[11]
$\text{C}_{30}\text{H}_{45}\text{NO}$			N-(4-pentadecyloxybenzylidene)-4-ethylaniline			
	Smec/Smec	338.5	48.32	142.75		
	Smec/Liq	341.0	4.05	11.88		
						[65]
$\text{C}_{30}\text{H}_{45}\text{NO}$			4-ethyl-N-[[(4-(pentadecyloxy)phenyl)methylene]benzenamine			
	Smec/Smec	340.2	57.52	169.08		
	Smec/Liq	347.2	9.39	27.04	196.12	
$\text{C}_{30}\text{H}_{45}\text{NO}$			4-dodecyl-N-[[(4-(pentyloxy)phenyl)methylene]benzenamine			
	Smec/Smec	322.7	42.12	130.52		
	Smec/Nem	325.5	6.98	21.44		
	Nem/Liq	343.7	0.95	2.76		
						[242]
$\text{C}_{30}\text{H}_{45}\text{NO}$			N-(4-dodecyloxybenzylidene)-4-pentylaniline			
	Smec/Smec	325.0	28.86	88.80		
	Smec/Liq	355.6	5.18	14.57		
						[256]
$\text{C}_{30}\text{H}_{45}\text{NO}_3$			N-(4-methoxyphenyl)- $\alpha$ -(4-hexadecyloxyphenyl)nitrone			
	Smec/Smec	388.2	66.48	171.25		
$\text{C}_{30}\text{H}_{46}\text{N}_2\text{O}$			Smec/Liq	401.2	4.48	182.42
						NA
						[162]
$\text{C}_{30}\text{H}_{46}\text{N}_2\text{O}$			4-butyl-4'-tetradecyloxyazobenzene			
	Smec/Smec	334.6	28.68	85.71		
$\text{C}_{30}\text{H}_{46}\text{N}_2\text{O}$			Smec/Liq	343.3	7.92	23.07
						108.78
						177
						[141]
$\text{C}_{30}\text{H}_{46}\text{N}_2\text{O}$			4-heptyl-4'-undecyloxyazobenzene			
	Smec/Smec	324.7	32.49	100.06		
	Smec/Nem	336.7	2.40	7.13		
	Nem/Liq	355.4	3.16	8.89		
						[390]
$\text{C}_{30}\text{H}_{46}\text{N}_2\text{O}$			4-[4-(11-dodecenoxy)phenyl]-5-[(S)-5-methylheptyl]pyrimidine			
	Smec/Smec	308.2	27.1	87.93		
	Smec/Chol	313.2		Not reported in paper		
	Chol/Liq	320.2		Not reported in paper		
						[201]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$\text{C}_{30}\text{H}_{46}\text{N}_2\text{O}$	Sol/Smec	290.2	19.1	65.82		[201]
	Smec/Chol	309.2		Not reported in paper		
	Chol/Liq	318.2		Not reported in paper		
$\text{C}_{30}\text{H}_{46}\text{N}_2\text{O}_3$	4,4'-dinonyloxyazoxybenzene					
	Sol/Smec	348.7	38.17	109.46		[179]
	Nem/Liq	386.2	1.65	4.27		
	Nem/Liq	394.7	1.76	4.46	118.19	
$\text{C}_{30}\text{H}_{49}\text{BrO}_2$	cholesterol $\beta$ -bromopropionate					
	Sol/Chol	375.0	20.8	55.47		[182]
	Chol/Liq	390.0	0.49	1.26	56.73	
$\text{C}_{30}\text{H}_{49}\text{ClO}_2$	cholesterol $\beta$ -chloropropionate					
	Sol/Chol	359.9	19.5	54.18		[182]
	Chol/Liq	398.0	0.74	1.86	56.04	
$\text{C}_{30}\text{H}_{49}\text{N}_3\text{O}_6$	4,4'-bis[4-(4-pentyloxybenzoyloxy)benzylideneamino]diphenylamine					
	Sol/Nem	504.2	42.0	83.30		[284]
	Nem/Liq	608.2	1.8	2.96	86.26	
$\text{C}_{30}\text{H}_{50}\text{OS}$	cholesteryl thiopropionate					
	Sol/Liq	385.2	24.6	63.7	64.3	NA
	Chol/Liq	384.3	0.3	0.6		
Note: Liquid crystalline behavior was observed upon cooling.						
$\text{C}_{30}\text{H}_{50}\text{O}_2$	cholesterol propionate					
	Sol/Chol	370.4	24.82	67.01		[166, 170]
	Chol/Liq	386.2	0.31	0.80	67.81	
Independent values from another reference						
	Sol/Chol	374.8	24.08	64.25		[169]
	Chol/Liq	388.4	0.43	1.11	65.36	
Independent values from another reference						
	Sol/Chol	372.8	21.8	58.48		[180]
	Chol/Liq	387.3	0.69	1.78	60.26	
Independent values from another reference						
	Sol/Chol	368.8	22.7	61.55		[182]
	Chol/Liq	384.1	0.25	0.65	62.20	
$\text{C}_{30}\text{H}_{50}\text{O}_3$	cholesterol ethyl carbonate					
	Sol/Chol	356.9	21.1	59.12		[180]
	Chol/Liq	378.9	0.70	1.85	60.97	
$\text{C}_{30}\text{H}_{51}\text{N}_3\text{O}_3$	$\text{N},\text{N}',\text{N}''$ -triheptanoyl-2,4,6-trimethyl-1,3,5-benzenetriamine					
	Sol/Meso	530.2	28.0	52.81		[190]
	Meso/Liq	630.2	16.0	25.39	78.20	
$\text{C}_{30}\text{H}_{52}\text{N}_2\text{O}_2$	$\text{N},\text{N}'$ -didecanoyl-2,3,5,6-tetramethylbenzene-1,4-diamine					
	Sol/Sol	421.2	22	52.23		[36]
	Sol/Meso	486.2	15	30.85		
	Meso/Meso	522.2	10	19.15		
	Meso/Liq	560.2	22	39.27	141.50	
$\text{C}_{30}\text{H}_{52}\text{N}_2\text{O}_2$	$\text{N},\text{N}'$ -didecanoyl-3,4,5,6-tetramethylbenzene-1,2-diamine					
	Sol/Meso	395.2	15	37.96		[61]
	Meso/Meso	475.2	2	4.21		
	Meso/Liq	508.2	21	41.32	83.49	

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{30}H_{52}N_2O_2$	Sol/Meso	355.2	49	137.95		[61]
	Meso/Liq	373.2	11	29.47	167.42	
$C_{31}H_{22}F_{22}O_5$	Sol/Smec	366.0	29.12	79.56		[128]
	Smec/Liq	387.9	6.72	17.32	96.88	
$C_{31}H_{23}F_{21}O_5$	Sol/Smec	368.7	45.99	124.74		[128]
	Smec/Liq	373.2	5.17	13.85	138.59	
$C_{31}H_{28}F_4O_5$						
	Sol/Smec	363.6	26.61	73.18		
	Smec/Chol	409.2	0.60	1.47		
$C_{31}H_{30}F_4O_4$	Chol/Liq	442.4	0.45	1.02	75.67	[91]
	Sol/Smec	366.2	28.14	76.84		
	Smec/Nem	367.8	Not detected by dsc			
$C_{31}H_{30}F_4O_4$	Nem/Liq	375.5	0.80	2.13	78.97	[96]
	Sol/Nem	375.6	28.22	75.13		
	Nem/Liq	423.8	0.46	1.09	76.22	
$C_{31}H_{30}F_4O_4$	Sol/Smec	383.3	39.06	101.90		[96]
	Smec/Nem	400.5	0.15	0.37		
	Nem/Liq	413.7	1.32	3.19	105.46	
$C_{31}H_{30}N_6O_6$						
	Sol/Nem	449.2	68.71	152.96		[422]
$C_{31}H_{31}F_{13}O_5$	Nem/Liq	464.2	1.12	2.41	155.37	
	Sol/Smec	357.5	29.49	82.49		[125]
	Smec/Smec	386.7	0.29	0.75		
$C_{31}H_{32}$	Smec/Liq	393.2	8.39	21.34	104.58	
	Sol/Nem	347.1	21.6	62.23		[55, 430]
	Nem/Liq	439.7	1.01	2.30	64.43	
$C_{31}H_{32}$	Sol/Nem	359.4	15.1	42.01		[55, 430]
	Nem/Liq	456.0	1.45	3.18	45.19	
$C_{31}H_{32}$	Sol/Nem	330.0	14.4	43.64		[55]
	Nem/Liq	433.3	0.85	1.96	45.60	
$C_{31}H_{32}$	Sol/Nem	331.6	11.9	35.89		[55]
	Nem/Liq	397.8	1.09	2.74	38.63	
$C_{31}H_{32}$	Sol/Nem	309.1	18.6	60.17		[55]
	Nem/Liq	399.9	1.25	3.13	63.30	
$C_{31}H_{32}$	Sol/Nem	302.6	13.6	44.94		[55]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{31}H_{32}$	Nem/Liq	401.2	1.11	2.76	47.70	[55]
	Sol/Nem	420.2	18.6	44.26		
	Nem/Liq	482.2	1.2	2.49	46.75	[82]
$C_{31}H_{32}N_2O_2$		2,3-dicyano-4-pentylphenyl 4'-pentyl-4-biphenyl-1-carboxylate				
		372.2	30.54	82.05		
		385.7	Not reported in paper			[202]
$C_{31}H_{32}N_2O_5$		4-methyl-2-oxo-2H-1-benzopyran-7-yl 4-[(1E)-4-(octyloxy)phenyl]azo]benzoate				
		403.2	5.53	13.72		
		436.2	0.29	0.66		
	Nem/Liq	478.2	0.52	1.09	15.47	[320]
$C_{31}H_{32}N_4O_2$		$\alpha$ -(4'-methylazobenzene-4-oxy)- $\omega$ -(azobenzene-4-oxy)hexane				
		422.2	47.39	112.25		
		432.2	4.31	9.97	122.22	[67]
$C_{31}H_{32}O_5$		7-[(4'-nonyloxy)benzoyloxy]isoflavone				
		423.5	26.55	62.69		
		467.4	1.99	4.26		
	Nem/Liq	468.5	0.29	0.62	67.57	[14]
$C_{31}H_{33}ClF_2O_5$		4-[(2S)-2-chloro-3-methylbutanoyloxy]biphenyl-4'-yl 4-heptyloxy-2,3-difluorobenzoate				
		348.2	18.15	52.13		
		408.9	1.6	3.91		
	Nem/Liq	445.0	0.6	1.35	57.39	[121]
$C_{31}H_{33}NO$		5-(4'-heptyloxybiphenyl-4-yl)-2-(4-methylphenyl)pyridine				
		424.0	9.1	21.46		
		512.0	4.0	7.81		
		545.0	4.0	7.34		
		597.0	1.2	2.01		
	Nem/Liq	633.0	1.1	1.74	40.36	[185]
$C_{31}H_{33}O$		1-[(4-pentyloxyphenyl)ethynyl]-4-[(4-butylphenyl)ethynyl]benzene				
		435.2	19.4	44.58		
		507.2	1.4	2.76	47.34	[82]
$C_{31}H_{34}N_4O_2S$		5-(4-heptyloxyphenyl)-5-(4-decyloxy)phenylazo-1,3,4-thiadiazole				
		426.9	42.49	99.53		
		449.7	0.58	1.29	100.82	[79]
$C_{31}H_{35}ClO_2$		1-(4-decylbiphenyl)-3-(4-chlorophenyl)propane-1,3-dione				
		406.2	44.18	108.76		
		418.2	5.15	12.31	121.07	[250]
$C_{31}H_{35}NO_6$		4-nitrophenyl 3-[[4'-(decyloxy)[1,1'-biphenyl]-4-yloxy]-2-propenoate				
		368.8	36.5	98.97		
		403.3	3.8	9.42	108.39	[305]
$C_{31}H_{36}ClNO_4$		3-hydroxy-4-[{(E)}-[(4-pentylphenyl)imino]methyl]phenyl 4-butoxy- $\alpha$ -chlorobenzene propanoate				
		324.2	Not reported in paper			
		328.8	2.30	7.00		
		359.5	0.34	0.95		
	Nem/Liq	385.5	5.42	14.06		[249]
$C_{31}H_{36}F_8O_5$		4-(2,2,3,3,4,4,5,5-octafluoropentyloxycarbonyl)phenyl 4-(dodecyloxy)benzoate				
		319.4	15.87	49.69		
		325.2	0.27	0.83		
	Nem/Liq	327.2	5.30	16.20	66.72	[125]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{pce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{pce}}$ (estimated) (estimated)	Ref.	
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$			
$C_{31}H_{36}O_7$	Sol/Nem	4-(4-nonyloxybenzoyloxy)phenyl (3,4-dimethoxy)benzoate	357.2	10.49	29.37	[103]	
	Nem/Liq		415.3	0.27	0.65		
$C_{31}H_{37}NO_3$	Sol/SmeC	4-isopropylphenyl 4-(4-octyloxybenzylideneamino)benzoate	373.2	29.0	77.71	[292]	
	SmeC/Nem		439.2	1.1	2.50		
	Nem/Liq		451.2	1.1	2.44		
$C_{31}H_{37}NO_3$	Sol/Nem	4-(4-octyloxybenzylideneamino)phenyl 4-isopropylbenzoate	384.2	34.0	88.50	[292]	
	Nem/Liq		449.2	1.5	3.34		
					91.84		
$C_{31}H_{37}NO_3$	Sol/SmeC	4-(4'-decyloxybenzoyloxybenzylidene)-2''-aniline	343.1	26.84	78.23	[399]	
	SmeC/Nem		359.4	0.36	1.00		
	Nem/Liq		386.7	0.64	1.66		
$C_{31}H_{38}$	Sol/Nem	1-( <i>trans</i> -4-butylcyclohexylethynyl)-4-(4-pentylphenylethynyl)benzene	358.3	15.7	43.82	[436]	
	Nem/Liq		466.9	1.4	3.00		
					46.82		
$C_{31}H_{38}$	Sol/Nem	1-( <i>trans</i> -4-butylcyclohexylethynyl)-4-(4-pentylphenylethynyl)benzene	358.3	15.7	43.82	[436]	
	Nem/Liq		466.9	1.4	3.00		
					46.82		
$C_{31}H_{38}$	Sol/Nem	1-( <i>trans</i> -4-pentylcyclohexylethynyl)-4-(4-butylphenylethynyl)benzene	368.3	12.4	33.67	[436]	
	Nem/Liq		468.5	0.9	1.92		
					35.59		
$C_{31}H_{38}N_2O_3$	Sol/Chol	4-[5-[S]-5-methylheptyl]-2-pyrimidinyl]phenyl 4-(5-hexenyl)benzoate	281.2	28.2	100.28	[201]	
	Chol/Liq		416.2	Not reported in paper			
$C_{31}H_{38}N_2O_4$	Sol/Nem	4-[[(4-ethoxyphenyl)azo]-2,3-methylphenyl 4-(octyloxy)benzoate	377.2	39.0	103.39	[339]	
	Nem/Liq		492.2	1.5	3.05		
					106.44		
$C_{31}H_{38}N_2O_4$	Sol/Nem	4-[[(4-ethoxyphenyl)azo]phenyl 4-(decyloxy)benzoate	381.2	37.0	97.06	[339]	
	Nem/Liq		501.2	1.3	2.59		
					99.65		
$C_{31}H_{38}O_5$	Sol/Sol	2-hydroxypropyl 4-[[[4'-(octyloxy)[1,1'-biphenyl]-4-yl]oxy]methyl]benzoate	408.6	25.07	61.36	[304]	
	Sol/SmeC		422.0	3.73	8.84		
	SmeC/SmeC		440.9	3.58	8.12		
	SmeC/SmeC		446.1	0.25	0.56		
	SmeC/Liq		459.8	16.63	36.17		
$C_{31}H_{38}O_5$	Sol/Sol	3-hydroxypropyl 4-[[[4'-(octyloxy)[1,1'-biphenyl]-4-yl]oxy]methyl]benzoate	409.8	10.55	25.74	[304]	
	Sol/SmeC		423.2	2.85	6.73		
	SmeC/SmeC		448.2	3.78	8.43		
	SmeC/SmeC		454.9	0.69	1.52		
	SmeC/Liq		461.8	18.50	40.06		
$C_{31}H_{38}O_7$	Sol/SmeC	4-pentyloxyphenyl 7-decanoxyoxochromone-2-carboxylate	390.1	22.3	57.16	[286]	
	SmeC/SmeC		428.2	Not observed by dsc			
	SmeC/Liq		434.1	4.6	10.60		
					67.76		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{31}H_{40}O$		4-octyloxy-4"-pentylterphenyl				
	Sol/Smec	369.9	12.55	33.93		
	Smec/Smec	457.2	5.02	10.98		
	Smec/Smec	483.7	6.69	13.83		
$C_{31}H_{40}N_2O_2S$	Smec/Nem	494.7	11.30	22.84	81.58	[50]
		2-(4-butoxyphenyl)-5-[4-[(2-ethyl-2,3-undecadienyl)oxy]phenyl]-1,3,4-thiadiazole				
	Sol/Smec	360.2	35.5	98.56		
	Smec/Nem	364.2	1.41	3.87		
$C_{31}H_{40}N_2O_2S$	Nem/Liq	366.2	1.02	2.79	105.22	[398]
		2-(4-butoxyphenyl)-5-[4-[(3S)-3-methyl-3,4-dodecadienyl]oxy]phenyl]-1,3,4-thiadiazole				
	Sol/Smec	321.2	13.6	42.34		
	Smec/Nem	356.2	1.8	5.05		
$C_{31}H_{40}N_2O_3$	Nem/Liq	377.2	0.5	1.33	48.72	[398]
		4-[5-[(S)-5-methylheptyl]-2-pyrimidinyl]phenyl 4-hexyloxybenzoate				
	Sol/Chol	329.2	18.5	56.20		
	Chol/Liq	425.2	Not reported in paper			[201]
$C_{31}H_{40}O_2$		8-propyltricyclo[4.4.0.0 <sup>3,8</sup> ]dec-1-yl 4'-pentyl[1,1'-biphenyl]-4-carboxylate				
	Sol/Chol	365.2	14.98	41.02		
	Chol/Liq	395.2	Not reported in paper			[382]
Note: Authors report only a single transition enthalpy, which we have assumed is for the Sol/Chol transition.						
$C_{31}H_{40}O_4$		4-[(4-propylcyclohexyl)carbonyl]oxy]phenyl 4-(4-ethylcyclohexyl)benzoate				
	Sol/Smec	365.2	12.34	33.79		
	Smec/Nem	407.7	13.39	32.84		
	Nem/Liq	596.2	Not reported in paper			[348]
$C_{31}H_{41}ClN_2O_3$		3-[4-[(2S,3S)-2-chloro-3-methylpentanoyloxy]phenyl]-5-(4-decyloxyphenyl)pyrazole				
	Sol/Sol	381.6	6.7	17.56		
	Smec/Smec	398.4	16.2	40.66		
	Smec/Liq	408.0	Not reported in paper			[137]
$C_{31}H_{41}ClO_5$		2-chloro-4-methylpentyl 4-[[3-[4-(nonyloxy)phenyl]-1-oxo-2-propenyl]oxy]benzoate				
	Sol/Smec	335.2	41.84	124.82		
	Smec/Chol	351.2	0.84	2.39		
	Chol/Liq	352.2	0.84	2.39	129.60	[257]
$C_{31}H_{41}ClO_5$		2-chloro-3-methylpentyl 4-[[3-[4-(nonyloxy)phenyl]-1-oxo-2-propenyl]oxy]benzoate				
	Sol/Smec	325.2	28.45	87.48		
	Smec/Smec	337.2	0.21	0.62		
	Smec/Liq	358.2	3.35	9.35	97.45	[257]
$C_{31}H_{41}ClO_5$		4-[[3-[4-(decyloxy)phenyl]-1-oxo-2-propenyl]oxy]phenyl 2-chloro-4-methylpentanoate				
	Sol/Smec	342.2	23.39	68.35		
	Smec/Smec	351.2	0.21	0.60		
	Smec/Liq	365.7	3.85	10.53	79.48	[257]
$C_{31}H_{41}ClO_5$		4-[[3-[4-(decyloxy)phenyl]-1-oxo-2-propenyl]oxy]phenyl 2-chloro-3-methylpentanoate				
	Sol/Smec	329.2	33.4	101.46		
	Smec/Smec	353.2	0.17	0.48		
	Smec/Chol	371.7	0.29	0.78		
$C_{31}H_{41}NO_2$	Chol/Liq	372.2	0.88	2.36	105.08	[257]
		6-n-pentyloxy-2-[4-nonyloxystyryl]quinoline				
	Sol/Smec	370.2	39.83	107.59		
	Smec/Nem	405.1	2.07	5.11		
$C_{31}H_{41}NO_2$	Nem/Liq	433.7	1.00	2.31	115.01	[112]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{31}H_{41}N_3O_3$	Sol/Smec	405.8	34.4	84.77		[154]
	Smec/Liq	456.8	9.1	19.92	104.69	
$C_{31}H_{42}N_2O_2$		6-n-decyloxy-2-[ $(4'$ -N-pentyloxyphenylimino)methyl]quinoline				
	Sol/Smec	353.1	27.95	79.16		[112]
	Smec/Nem	375.9	0.10	0.27		
$C_{31}H_{42}N_2O_2$	Nem/Liq	401.1	1.29	3.22	82.65	
		5-methyl-5'-[2-(4-dodecyloxyphenyl)-2-hydroxyethyl]-2,2'-bipyridine				
	Sol/Smec	355.0	29.15	82.11		[48]
$C_{31}H_{42}N_2O_3$	Smec/Liq	413.5	4.84	11.70	93.81	
		N-4-pentyloxyphenyl-6-decyloxyquinoline-2-carboxamide				
	Sol/Smec	364.1	42.3	116.18		[43]
$C_{31}H_{42}O_4$	Smec/Liq	385.5	5.7	14.79	130.97	
		7-hexyloxy-3-(4-decyloxyphenyl)-3 <i>H</i> -1-benzopyran-4-one				
	Sol/Smec	390.4	31.6	80.94		[44]
$C_{31}H_{43}NO_5$	Smec/Liq	421.7	6.3	14.94	95.88	
		4-[4-(4-decyloxyphenyliminomethyl)-3-hydroxyphenoxy]butyl methacrylate				
	Sol/Smec	332.5	56.09	168.69		[294]
$C_{31}H_{43}NO_5$	Smec/Smec	351.6	Not reported in paper			
	Smec/Liq	371.2	7.23	19.48	188.17	
$C_{31}H_{43}N_3O_2S$		6-[4-(4-octyloxyphenyliminomethyl)-3-hydroxyphenoxy]hexyl methacrylate				
	Sol/Smec	327.9	39.38	120.10		[294]
	Smec/Smec	353.2	Not reported in paper			
$C_{31}H_{43}N_3O_2S$	Smec/Liq	371.1	5.80	15.63	135.73	
$C_{31}H_{43}N_3O_2S$		5-(4-octyloxyphenyl)-2-(4-octyloxy)benzylideneamino-1,3,4-thiadiazole				
	Sol/Smec	381.0	33.8	88.71		[79,396,397]
	Smec/Nem	467.1	4.6	9.85		
$C_{31}H_{43}N_3O_2S$	Nem/Liq	478.4	1.1	2.30	100.86	
$C_{31}H_{43}N_3O_2S$		N-[[4-(octyloxy)phenyl]methylene]-5-[4-(octyloxy)phenyl]-1,2,4-thiadiazol-2-amine				
	Sol/Smec	383.5	39.27	102.40		[283]
	Smec/Nem	462.8	4.07	8.79		
$C_{31}H_{43}N_3O_2S$	Nem/Liq	474.8	2.43	5.12	116.31	
$C_{31}H_{43}N_3O_2S$		5-(4-octyloxy)phenyl-2-(4-n-octyloxy)phenylamido-1,3,4-thiadiazole				
	Sol/Smec	419.8	17.0	40.50		[31]
	Smec/Nem	520.9	5.0	9.60		
$C_{31}H_{43}N_3O_2S$	Nem/Liq	527.5	1.3	2.46	52.56	
$C_{31}H_{43}N_3O_3S$		5-(4-decyloxy)phenyl-2-(2-hydroxy-4-hexyloxy)benzylideneamino-1,3,4-thiadiazole				
	Sol/Smec	370.2	28.90	78.07		[59]
	Smec/Nem	482.2	7.66	15.89		
$C_{31}H_{43}N_3O_3S$	Nem/Liq	486.2	0.07	0.14	94.10	
$C_{31}H_{43}N_3O_3S$		2-[(E)-[(5-[4-(octyloxy)phenyl]-1,3,4-thiadiazol-2-yl)imino]methyl]-5-(octyloxy)phenol				
	Sol/Smec	379.6	26.63	70.15		[283]
	Smec/Nem	489.5	6.18	12.63		
$C_{31}H_{44}$	Nem/Liq	490.4	2.98	6.08	88.86	
$C_{31}H_{44}N_2O_2$		1-( <i>trans</i> -4-butylcyclohexylethynyl)-4-( <i>trans</i> -pentylcyclohexylethynyl)benzene				
	Sol/Nem	394.8	17.0	43.06		[436]
	Nem/Liq	465.4	1.2	2.58	45.64	
$C_{31}H_{44}N_2O_2$		3,5-di(4-n-octyloxyphenyl)pyrazole				
	Sol/Sol	334.9	26.9	80.32		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.		
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$				
$\text{C}_{31}\text{H}_{44}\text{N}_2\text{O}_2\text{S}$	Sol/Smec	416.5	14.3	34.33	129.06	[16]		
	Smec/Smec	452.6	0.3	0.66				
	Smec/Liq	465.4	6.4	13.75				
$\text{C}_{31}\text{H}_{44}\text{N}_2\text{O}_3$	6-n-decyloxy-2-(4-heptyloxybenzylidenamino)benzothiazole					[41]		
	Sol/Smec	354.0	36.7	103.67				
	Smec/Nem	374.4	0.9	2.40				
	Nem/Liq	393.5	1.8	4.57	110.64			
$\text{C}_{31}\text{H}_{44}\text{N}_2\text{O}_3$	4-pentanoyl-4'-tetradecanoyloxyazobenzene					[157]		
	Sol/Sol	325.1	3.23	9.94				
	Sol/Sol	354.4	6.00	16.93				
	Sol/Smec	373.5	38.12	102.06				
$\text{C}_{31}\text{H}_{44}\text{N}_2\text{O}_3$	Smec/Liq	402.6	9.54	23.70	152.63			
	4-propanoyl-4'-hexadecanoyloxyazobenzene					[157]		
	Sol/Smec	378.7	53.01	139.98				
	Smec/Liq	410.7	8.62	20.99	160.97			
$\text{C}_{31}\text{H}_{44}\text{N}_2\text{O}_3\text{S}$	N-[2-(6-decyloxybenzothiazolyl)]-4-heptyloxybenzamide					[41]		
	Sol/Sol	357.1	13.2	36.96				
	Sol/Smec	378.1	28.2	74.58				
	Smec/Liq	410.6	4.9	11.93	123.47			
$\text{C}_{31}\text{H}_{44}\text{N}_4\text{OS}$	2-(4-decanoxyphenylazo)-5-(4-heptyloxy)phenyl-1,3,4-thiadiazole					[396]		
	Sol/Nem	426.9	48.49	113.59				
	Nem/Liq	449.7	0.58	1.29	114.88			
	4-[(1E)-[4-[4-(1-oxohexyl)-1-piperazinyl]phenyl]azo]benzoic acid, octyl ester							
$\text{C}_{31}\text{H}_{44}\text{N}_4\text{O}_3$	Sol/Smec	373.9	22.65	60.58	[344]			
	Smec/Smec	382.2	2.71	7.09				
	Smec/Smec	405.9	0.83	2.04				
	Smec/Liq	473.6	9.84	20.78	90.49			
$\text{C}_{31}\text{H}_{44}\text{O}_3$	4-( <i>trans</i> -4-propylcyclohexyl)phenyl 4-[(S)-6-methyloctyl]oxybenzoate					[208]		
	Sol/Smec	358.2	37.55	104.83				
	Smec/Chol	359.2	Not reported in paper					
	Chol/Liq	436.2	Not reported in paper					
$\text{C}_{31}\text{H}_{44}\text{O}_3$	4-( <i>trans</i> -4-pentylcyclohexyl)phenyl 4-[(S)-4-methylhexyl]oxybenzoate					[208]		
	Sol/Chol	358.2	38.51	107.51				
	Chol/Liq	438.2	Not reported in paper					
	4-[(S)-4-methylhexyloxy]phenyl 4-(10-undecenyloxy)benzoate							
$\text{C}_{31}\text{H}_{44}\text{O}_4$	Sol/Smec	313.2	39.8	127.08	[200]			
	Smec/Smec	329.2	Not reported in paper					
	Smec/Liq	343.2	Not reported in paper					
	4-[(S)-6-methyloctyloxy]phenyl 4-(8-nonyloxy)benzoate							
$\text{C}_{31}\text{H}_{44}\text{O}_4$	Sol/Smec	325.2	31.6	97.17	[200]			
	Smec/Liq	351.2	Not reported in paper					
	4-(10-undecenyloxy)phenyl 4-[(S)-4-methylhexyloxy]benzoate					[200]		
	Sol/Smec	307.2	27.5	89.52				
$\text{C}_{31}\text{H}_{44}\text{O}_4$	Smec/Chol	315.2	Not reported in paper					
	Chol/Liq	334.2	Not reported in paper					
	4-(8-nonyloxy)phenyl 4-[(S)-6-methyloctyloxy]benzoate					[200]		
	Sol/Smec	311.2	27.0	86.76				
$\text{C}_{31}\text{H}_{44}\text{O}_4$	Smec/Chol	334.2	Not reported in paper					
	Chol/Liq	343.2	Not reported in paper					

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$\text{C}_{31}\text{H}_{45}\text{FO}_3\text{S}$	Sol/Nem	335.2	45.6	136.04	NA	[4]
	Nem/Liq	346.2	2.0	5.78	141.82	
$\text{C}_{31}\text{H}_{45}\text{NO}_5$	4'-n-octadecyloxy-3'-nitrobiphenyl-4-carboxylic acid					[46]
	Sol/Sol	324.8	11.42	35.16		
	Sol/Sol	337.2	3.41	10.11		
	Sol/Smec	399.4	42.5	106.41		
	Smec/Smec	428.1	1.14	2.66		
$\text{C}_{31}\text{H}_{45}\text{N}_3\text{OS}$	Smec/Liq	464.0	1.88	4.05	158.39	[396,397]
	5-[4-(4-(decyloxy)phenyl)-N-[(5-octyl-2-thienyl)methylene]-1,2,4-thiadiazole-2-amine					
$\text{C}_{31}\text{H}_{45}\text{N}_3\text{O}_2\text{S}$	Sol/Smec	391.7	39.2	100.08	[42]	
	Smec/Liq	414.5	8.5	20.51	120.59	
	6-n-decyloxy-2-(4-octyloxyphenylazo)benzothiazole					
$\text{C}_{31}\text{H}_{46}\text{N}_2\text{O}$	Sol/Smec	343.9	44.9	130.56	[198]	
	Smec/Nem	368.2	0.8	2.17		
	Nem/Liq	397.0	1.8	4.53	137.26	
$\text{C}_{31}\text{H}_{46}\text{N}_2\text{O}_2$	Sol/Nem	364.2	18.5	50.80	[201]	
	Nem/Liq	414.2	Not reported in paper			
	4-[5-[(S)-6-methyloctyl]-2-pyrimidinyl]phenyl <i>trans</i> -4-pentylcyclohexane-1-carboxylate					
$\text{C}_{31}\text{H}_{46}\text{O}_2$	Sol/Chol	383.2	31.0	80.90	[382]	
	Chol/Liq	420.2	Not reported in paper			
	4-( <i>trans</i> -4-pentylcyclohexyl)phenyl 8-propyltricyclo[4.4.0.0 <sup>3,8</sup> ]decane-1-carboxylate					
$\text{C}_{31}\text{H}_{46}\text{O}_2\text{S}$	Sol/Chol	374.2	21.25	56.79	[217]	
	Chol/Liq	401.2	Not reported in paper			
	Note: Authors report only a single transition enthalpy, which we have assumed is for the Sol/Chol transition.					
$\text{C}_{31}\text{H}_{47}\text{NO}$	Sol/Smec	339.2	37.66	111.03	NA	[11]
	Smec/Liq	361.9	7.78	21.50	132.53	
	N-(p-n-nonyloxybenzylidene)-4-(n-nonyl)aniline					
$\text{C}_{31}\text{H}_{47}\text{NO}$	Sol/Smec	320.2	42.20	131.79	[65]	
	Smec/Smec	351.3	4.31	12.27		
	Smec/Liq	360.1	8.42	23.38	167.44	
$\text{C}_{31}\text{H}_{47}\text{NO}$	Sol/Smec	336.2	21.05	62.61	[256]	
	Smec/Smec	341.6	4.09	11.97		
	Smec/Liq	345.6	2.92	8.45	83.03	
$\text{C}_{31}\text{H}_{47}\text{NO}$	Sol/Smec	322.4	25.70	79.71	[240]	
	Smec/Smec	356.7	6.18	17.33		
	Smec/Liq	363.0	9.00	24.79	121.83	
$\text{C}_{31}\text{H}_{47}\text{NO}$	Sol/Smec	344.0	25.06	72.82	[65]	
	Smec/Smec	352.8	2.90	8.22		
	Smec/Liq	359.3	5.39	15.00	96.04	
$\text{C}_{31}\text{H}_{47}\text{NO}$	Sol/Smec	322.2	45.60	141.53	[65]	
	Smec/Liq	342.3	2.16	6.31	147.84	
	5-hexyl-2-[4-( <i>trans</i> -4-pentylcyclohexyl)butyl]phenyl pyrimidine					

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{31}H_{48}N_2O$	Sol/Nem	352.2	18.9	53.66		
	Nem/Liq	386.2		Not reported in paper		[198]
$C_{31}H_{48}N_2O$	4-propyl-4'-hexadecyloxyazobenzene					
	Sol/Smec	347.1	107.9	310.86		
	Smec/Liq	352.4	26.92	76.39	387.25	184.1 [153]
$C_{31}H_{48}N_2O$	4-pentyl-4'-tetradecyloxyazobenzene					
	Sol/Smec	335.2	29.13	86.90		
	Smec/Smec	346.1	4.57	13.20		
$C_{31}H_{48}N_2O$	Smec/Liq	357.6	10.56	29.53	129.63	184.1 [141]
	4-heptyl-4'-dodecyloxyazobenzene					
	Sol/Smec	323.4	27.54	85.16		
$C_{31}H_{48}N_2O$	Smec/Smec	343.7	3.39	9.86		
	Smec/Liq	358.7	9.93	27.68	122.70	184.1 [390]
	5-heptyl-2-[4-[3-[ <i>trans</i> -4-pentylcyclohexyl]oxy]propyl]phenyl]pyrimidine					
$C_{31}H_{48}N_2O$	Sol/Nem	357.2	16.1	45.07		
	Nem/Liq	410.2		Not reported in paper		[198]
$C_{31}H_{48}N_2O$	2-[4-(11-dodecenoxy)phenyl]-5-[(S)-6-methyloctyl]pyrimidine					
	Sol/Smec	306.2	27.9	91.11		
	Smec/Chol	310.2		Not reported in paper		
$C_{31}H_{50}N_2O_6$	Chol/Liq	316.2		Not reported in paper		[201]
	4,4'-bis[4-(4-pentyloxybenzylideneamino)benzoyloxy]diphenylmethane					
	Sol/Smec	438.2	21.0	47.92		
$C_{31}H_{50}O_2$	Smec/Liq	503.2	17.0	33.78	81.70	[284]
	cholesteryl crotonate					
	Sol/Chol	385.95	24.73	64.07		
$C_{31}H_{52}OS$	Chol/Liq	435.15	1.26	2.88	66.95	107.9 [155,310]
	cholesteryl thiobutyrate					
	Sol/Chol	373.2	23.6	63.2		
$C_{31}H_{52}O_6$	Chol/Liq	390.8	0.36	0.9	64.1	NA [155,312]
	<i>cis, cis</i> -(3,5-dihydroxycyclohexyl) 3,4-bis(nonyloxy)benzoate					
	Sol/Meso	352.2	10.9	30.95		
$C_{31}H_{53}BrO_2$	Meso/Liq	382.7	0.9	2.35	33.30	[98]
	3-[2-(2-bromoethoxy)ethoxy]-cholest-5-ene					
	Sol/Nem	313.0	20.9	66.77		
$C_{31}H_{53}ClO_2$	Nem/Liq	319.7	0.43	1.35	68.12	[316]
	3-[2-(2-chloroethoxy)ethoxy]-cholest-5-ene					
	Sol/Nem	312.4	3.26	10.44		
$C_{31}H_{53}IO_2$	Nem/Liq	319.2	0.42	1.32	11.76	[316]
	3-[2-(2-iodoethoxy)ethoxy]-cholest-5-ene					
	Sol/Nem	314.3	26.4	84.00		
$C_{31}H_{54}O_2$	Nem/Liq	322.7	1.26	3.90	87.90	[322]
	$5\alpha$ -cholesta-3 $\beta$ -ol butyrate					
	Sol/Liq	361.15	26.9	74.6		
$C_{31}H_{54}N_2O_2$	Chol/Liq	356.15	0.34	1.0		[311]
	Note: Liquid crystalline phase detected upon cooling.					
	$N,N'$ -diundecanoyl-2,3,5-trimethylbenzene-1,4-diamine					
$C_{31}H_{54}N_2O_2$	Sol/Meso	398.2	38	95.43		
	Meso/Nem	467.2	17	36.39		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{31}H_{55}NO_4$	Nem/Liq	468.2	To small to be measured	130.92		[36]
	Sol/Meso	355.2	43.0	121.06		
	Meso/Liq	362.2	6.8	18.77	139.83	[378]
$C_{32}H_{20}F_{26}O_5$	4-(3,3,4,4,5,5,6,6,7,7,8,8,8-tridecafluoroctyloxycarbonyl)phenyl 4-[ (perfluorohexyyl)butoxy]benzoate					
	Sol/Smec	399.2	49.33	123.57		
	Smec/Smec	399.6	1.86	4.65		
	Smec/Liq	404.3	4.11	10.17	138.39	[128]
$C_{32}H_{22}F_4N_2O_7$	4-ethoxy-2,3-difluorobenzoic acid, 1,2,4-oxadiazole-3,5-diyl-4,1-phenylene ester					
	Sol/Sol	453.5	25.33	55.85		
	Sol/Nem	461.3	17.84	38.67		
	Nem/Liq	Decomposed prior to transition				[264]
$C_{32}H_{30}F_4O_5$	4-[ (S)-2-methylbutoxycarbonyl]phenyl 4-[ (4-pentyloxy-2,3,4,5-tetrafluorophenyl)ethynyl]benzoate					
	Sol/Smec	362.7	26.08	71.91		
	Smec/Smec	363.2	Not detected by dsc			
	Smec/Chol	397.9	0.36	0.90		
$C_{32}H_{30}O_8$	Chol/Liq	426.3	0.41	0.96	73.77	[91]
	benzoic acid, 4,4'-[1,4-phenylenebis[(1-oxo-2-propene-3,1-diyl)oxo]]bis, dipropyl ester					
	Sol/Nem	487.6	60.6	124.28		
$C_{32}H_{30}O_8$	Nem/Liq	571.1	2.0	3.50	127.78	[271]
	1,4-benzenedicarboxylic acid, bis[4-[3-(propoxy)-3-oxo-1-propenyl]phenyl ester					
	Sol/Sol	386.2	1.2	3.11		
	Smec/Smec	407.6	31.9	78.26		
$C_{32}H_{32}F_4O_4$	Smec/Nem	526.6	0.8	1.52		
	Nem/Liq	575.9	0.8	1.39	84.28	[271]
	pentyl 4-[4-((4-pentyloxy-2,3,5,6-tetrafluorophenyl)ethynyl)benzyloxy]benzoate					
	Sol/Smec	361.3	22.59	62.52		
$C_{32}H_{32}F_4O_4$	Smec/Nem	367.4	1.00	2.72		
	Nem/Liq	371.6	0.83	2.23	67.47	[96]
	ethyl 4-[4-((4-octyloxy-2,3,5,6-tetrafluorophenyl)ethynyl)benzyloxy]benzoate					
	Sol/Smec	367.6	43.36	117.95		
$C_{32}H_{32}F_4O_4$	Smec/Nem	376.0	0.41	1.09		
	Nem/Liq	388.4	1.56	4.02	123.06	[96]
	4-[ (S)-2-methylbutoxy]phenyl 4-[ (4-hexyloxy-2,3,5,6-tetrafluorophenyl)ethynyl]benzoate					
	Sol/Nem	354.6	33.13	93.43		
$C_{32}H_{32}F_4O_4$	Nem/Liq	409.5	0.54	1.32	94.75	[124]
	$bis(4\text{-butyoxypyhenyl})\text{2,2}'\text{-bipyridine-5,5}'\text{-dicarboxylate}$					
	Smec/Smec	467.2	31.5	67.42		
	Nem/Liq	Sample decomposed before isotropic liquid.				[93]
$C_{32}H_{32}N_4O_6S_3$	$bis[4\text{-}(5\text{-n-pentylthio-1,3,4-oxadiazole-2-yl)phenyl}]2,5\text{-thiophenedicarboxylate}$					
	Smec/Smec	416.2	27.7	66.55		
	Smec/Liq	501.2	5.8	11.57	78.12	[23]
$C_{32}H_{32}N_6O_6$	$\alpha,\omega\text{-bis}(4\text{-nitroazobenzene-4'-oxy)octane}$					
	Sol/Nem	462.2	54.56	118.04		
	Nem/Liq	477.2	3.25	6.81	124.85	[422]
$C_{32}H_{33}F_{13}O_5$	4-(3,3,4,4,5,5,6,6,7,7,8,8,8-tridecafluoroctyloxycarbonyl)phenyl 4-(decyloxy)benzoate					
	Sol/Smec	355.4	27.45	77.24		
	Smec/Smec	382.2	0.30	0.78		
	Smec/Liq	385.7	8.41	21.80	99.82	[125]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{32}H_{34}$	Sol/Nem	4'-[2-(4-propylphenyl)-1-ethynyl]-2'-methyl-4-hexyltolane				
		351.1	15.7	44.72		
	Nem/Liq	443.2	1.17	2.64	47.36	[55, 430]
$C_{32}H_{34}$	Sol/Nem	4'-[2-(4-hexylphenyl)-1-ethynyl]-2'-methyl-4-propyltolane				
		330.5	13.7	31.45		
	Nem/Liq	441.9	1.7	3.85	35.30	[55]
$C_{32}H_{34}$	Sol/Nem	4'-[2-(4-propylphenyl)-1-ethynyl]-2'-ethyl-4-pentyltolane				
		329.3	12.5	37.96		
	Nem/Liq	408.1	1.26	3.08	41.04	[55]
$C_{32}H_{34}$	Sol/Nem	4'-[2-(4-hexylphenyl)-1-ethynyl]-2'-ethyl-4-ethyltolane				
		304.2	13.3	43.72		
	Nem/Liq	380.0	0.75	1.97	45.69	[55]
$C_{32}H_{34}$	Sol/Nem	1-[(4-hexylphenyl)ethynyl]-3-methyl-4-[(4-propylphenyl)ethynyl]benzene				
		330.2	14.4	43.61		
	Nem/Liq	442.2	1.3	2.94	46.55	[82]
$C_{32}H_{34}$	Sol/Nem	1-[(4-hexylphenyl)ethynyl]-4-[(2-methyl-4-propylphenyl)ethynyl]benzene				
		340.2	13.5	39.68		
	Nem/Liq	449.2	1.0	2.23	41.91	[82]
$C_{32}H_{34}$	Sol/Nem	1-[(4-hexylphenyl)ethynyl]-4-[(3-methyl-4-propylphenyl)ethynyl]benzene				
		358.2	30.9	86.26		
	Nem/Liq	416.2	0.9	2.16	88.42	[82]
$C_{32}H_{34}N_2O_2$	Sol/Nem	2,3-dicyano-4-pentylphenyl 4'-hexyl-4-biphenyl-1-carboxylate				
		366.2	31.80	86.84		
	Nem/Liq	382.2	Not reported in paper			[202]
$C_{32}H_{34}N_2O_2$	Sol/Nem	2,3-dicyano-4-heptylphenyl 4'-butyl-4-biphenyl-1-carboxylate				
		360.7	29.29	81.20		
	Nem/Liq	373.2	Not reported in paper			[202]
$C_{32}H_{34}N_2O_2$	Sol/Nem	$\alpha$ -(4'-methylazobenzene-4-oxy)- $\omega$ -(4'-methylazobenzene-4-oxy)hexane				
		458.2	53.33	116.39		
	Nem/Liq	481.2	6.96	14.46	130.85	[67]
$C_{32}H_{34}N_4O_2$	Sol/Nem	$\alpha$ -(4'-ethylazobenzene-4-oxy)- $\omega$ -(azobenzene-4-oxy)hexane				
		415.2	53.51	128.88		
	Nem/Liq	426.2	3.44	8.07	136.95	[67]
$C_{32}H_{34}O$	Sol/Nem	1-[(4-pentyloxyphenyl)ethynyl]-3-methyl-4-[(4-butylphenyl)ethynyl]benzene				
		356.2	16.4	46.04		
	Nem/Liq	474.2	1.4	2.95	48.99	[82]
$C_{32}H_{34}O_2$	4,4"-dibutoxy-p-quaterphenyl					
	Sol/Sol	520.0	8.84	17.00		
	Sol/Sol	610.0	4.88	8.00		
	Sol/Smec	616.0	3.70	6.01		
	Smec/Liq	670.0	10.05	15.00	46.01	145.6
$C_{32}H_{34}O_5$	7-[(4'-decyloxy)benzoyloxy]isoflavone					
	Smec/Smec	413.1	8.03	19.44		
	Note: Solid/Smectic transition enthalpy is abnormally low compared to other compounds in this series.					
	Smec/Liq	469.6	3.42	7.28		[14]
$C_{32}H_{34}O_6$	bis(4-heptyloxyphenyl) terephthalate					
	Sol/Smec	425.5	56.2	132.08		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{32}H_{34}O_6S_2$	Smec/Smec	449.2	0.3	0.67		
	Smec/Nem	454.0	0.9	1.98		
	Nem/Liq	468.0	1.7	3.63	138.36	189.0
	bis(4-pentyloxyphenyl) 2,2'-bithiophene-5,5'-dicarboxylate					
	Sol/Sol	372.8	2.0	5.36		
	Sol/Sol	378.9	5.2	13.72		
$C_{32}H_{34}O_8$	Sol/Sol	403.4	2.4	5.95		
	Sol/Smec	422.8	29.6	70.01		
	Smec/Nem	504.6	1.8	3.57		
	Nem/Liq	510.0	1.3	2.55	101.16	177.8
	bis[4-(2-methylbutoxycarbonyl)phenyl] terephthalate					
	Sol/Smec	406.2	35.0	86.16		
$C_{32}H_{34}O_8$	Smec/Nem	426.2	0.6	1.41		
	Nem/Liq	438.2	0.4	0.91	88.48	126.4
	bis[4-(3-methylbutoxycarbonyl)phenyl] terephthalate					
	Sol/Smec	426.2	47.0	110.28		
	Smec/Nem	440.2	4.4	10.00	120.28	126.4
	[194]					
$C_{32}H_{34}O_8$	bis(4-pentyloxycarbonylphenyl) terephthalate					
	Sol/Smec	418.2	40.0	95.65		
	Smec/Liq	463.2	5.3	11.44	107.09	166.6
	[194]					
	$C_{32}H_{35}ClF_2O_5$					
	Sol/Smec	347.9	16.08	46.22		
$C_{32}H_{35}ClF_2O_5$	Smec/Nem	397.4	1.7	4.28		
	Nem/Liq	425.4	0.7	1.65	52.15	[121]
	4-[(2S)-2-chloro-3-methylpentanoyloxy]biphenyl-4'-yl 4-heptyloxy-2,3-difluorobenzoate					
	Sol/Smec	342.7	23.01	67.14		
	Smec/Nem	412.9	1.4	3.39		
	Nem/Liq	443.3	0.86	1.94	72.47	[121]
$C_{32}H_{35}NO$	5-(4'-octyloxybiphenyl-4-yl)-2-(4-methylphenyl)pyridine					
	Sol/Sol	415.0	8.5	20.48		
	Sol/Smec	510.0	4.0	7.84		
	Smec/Smec	543.0	4.0	7.37		
	Smec/Nem	599.0	1.6	2.67		
	Nem/Liq	623.0	1.2	1.93	40.29	[185]
$C_{32}H_{35}NO_4$	4-cyanophenyl 3-[[4'-(decyloxy)[1,1'-biphenyl]-4-yloxy]-2-propenoate					
	Sol/Smec	358.6	38.8	108.20		
	Smec/Liq	362.8	4.9	13.51	121.71	[305]
	$C_{32}H_{35}NO_4$					
	Sol/Smec	364.1	35.5	97.50		
	Smec/Liq	391.2	4.1	10.48	107.98	[305]
$C_{32}H_{36}N_2O_4$	dipentyl N,N'-[1,4-phenylene- <i>bis</i> (methylidyne)]- <i>bis</i> [aminobenzoate]					
	Sol/Smec	373.2	34.0	91.10		
	Smec/Nem	479.2	1.3	2.71		
	Nem/Liq	489.2	0.46	0.94	94.75	[192]
	$C_{32}H_{36}N_4O_2S$					
	Sol/Smec	423.1	37.46	88.54		
$C_{32}H_{36}O_7$	Smec/Nem	429.4	1.13	2.63		
	Nem/Liq	449.6	1.15	2.56	93.97	[79]
	5-(4-octyloxyphenyl)-5-(4-decyloxy)phenylazo-1,3,4-thiadiazole					
	Sol/Nem	363.3	28.3	77.90		
	2-acetyl-1,4-phenylene 4-(pentyloxy)benzoate					
	[79]					

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)						
$C_{32}H_{37}ClO_2$	Nem/Liq	408.1	1.8		4.41	82.31		[432]
	Sol/SmeC	397.7	35.48		89.21			
	SmeC/Liq	417.2	5.15		12.34	101.55		[250]
$C_{32}H_{38}N_2O_2$	Sol/Nem	396.2	21.33		53.84			
	Nem/Liq	398.2	Not reported in paper					[202]
	Sol/Sol	359.0	35.40		98.61			
$C_{32}H_{38}O_4$	Sol/Nem	406.0	34.00		83.74			
	Nem/Liq	412.0	2.18		5.29	187.64	153.8	[157]
	Sol/Sol	359.0	35.40		98.61			
$C_{32}H_{38}O_7$	Sol/Nem	361.7	9.39		25.96			
	Nem/Liq	412.0	0.39		0.95	26.91		[103]
	Sol/Nem	376.7	9.53		25.30			
$C_{32}H_{38}O_7$	Nem/Liq	397.0	0.34		0.86	26.16		[103]
	Sol/SmeC	356.2	37.0		103.87			
	SmeC/Liq	452.2	1.7		3.76	107.63	169.6	[292]
$C_{32}H_{39}NO_3$	Sol/SmeC	404.2	37.0		91.54			
	SmeC/Liq	425.2	4.4		10.35	101.89	141.6	[292]
	Sol/SmeC	348.4	33.41		95.90			
$C_{32}H_{39}NO_3$	SmeC/Nem	359.8	0.63		1.75			
	Nem/Liq	380.0	0.90		2.37	100.02		[399]
	Sol/Nem	361.2	32.0		88.59			
$C_{32}H_{39}NO_3$	Nem/Liq	469.2	1.7		3.62	92.21	162.6	[292]
$C_{32}H_{40}$	Sol/Nem	368.0	11.8		32.07			
	Nem/Liq	461.6	0.8		1.73	33.80		[436]
	Sol/Chol	339.2	29.4		86.67			
$C_{32}H_{40}N_2O_3$	Chol/Liq	412.2	Not reported in paper					[201]
	Sol/SmeC	317.2	20.7		65.26			
	SmeC/Chol	321.2	Not reported in paper					
$C_{32}H_{40}N_2O_3$	Chol/Liq	422.2	Not reported in paper					[201]
	Sol/Chol	339.2	29.4		86.67			
	Chol/Liq	412.2	Not reported in paper					[201]
$C_{32}H_{40}N_2O_4$	Sol/Chol	365.2	51.0		139.65			
	Nem/Liq	459.2	1.5		3.27	142.92		[339]
	Sol/Nem	365.2	51.0		139.65			
$C_{32}H_{40}N_2O_4$	4-[(4-ethoxyphenyl)azo]-3-methylphenyl 4-(decyloxy)benzoate							

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$\text{C}_{32}\text{H}_{40}\text{O}_5$	Sol/Nem	366.2	54.0	147.46	150.52	[339]
	Nem/Liq	458.2	1.4	3.06		
$\text{C}_{32}\text{H}_{40}\text{O}_5$		364.4	22.86	62.73		
	Sol/Sol	413.0	2.67	6.46		
	Sme/Sme	435.4	3.89	8.93		
	Sme/Sme	444.0	0.56	1.26		
	Sme/Liq	447.7	16.65	37.19	116.57	[304]
		394.8	24.38	61.75		
$\text{C}_{32}\text{H}_{40}\text{O}_5$	Sol/Sol	414.2	2.57	6.20		
	Sme/Sme	442.1	3.99	9.03		
	Sme/Sme	447.9	0.35	0.78		
	Sme/Liq	452.9	17.46	38.55	116.31	[304]
		388.9	12.3	31.63		
$\text{C}_{32}\text{H}_{40}\text{O}_6\text{S}$	Sol/Sol	395.5	7.6	19.22		
	Sol/Nem	403.9	39.7	98.29		
	Nem/Liq	408.9	1.4	3.42	152.56	190.3
		385.9	25.7	66.60		
$\text{C}_{32}\text{H}_{40}\text{O}_7$	Sme/Sme	430.2	Not observed by dsc			
	Sme/Liq	435.6	5.2	11.94	78.54	[286]
		386.2	41.84	108.34		
$\text{C}_{32}\text{H}_{41}\text{NO}_2$	Sme/Sme	489.7	1.67	3.41		
	Nem/Liq	493.7	1.17	2.37	114.12	[323]
		339.9	39.86	117.27		
$\text{C}_{32}\text{H}_{41}\text{NO}_2$	Sme/Liq	437.4	3.91	8.94	126.21	[139]
		2-cyano-3-[4-[(1E)-[4-[4-(1-oxooctyl)-1-piperazinyl]phenyl]azo]phenyl]-2-propenoic acid, butyl ester				
	Sol/Sme	456.3	17.94	39.32		
$\text{C}_{32}\text{H}_{41}\text{N}_5\text{O}_3$	Sme/Liq	479.9	2.45	5.11	44.43	[326]
		361.0	14.14	39.17		
	Sme/Nem	404.7	13.68	33.80		
$\text{C}_{32}\text{H}_{42}\text{O}_4$	Nem/Liq	592.2	Not reported in paper			[348]
		4-[(4-butylcyclohexyl)carbonyl]oxy]phenyl 4-(4-ethylcyclohexyl)benzoate				
	Sol/Sme	349.6	11.72	33.52		
$\text{C}_{32}\text{H}_{42}\text{O}_4$	Sme/Nem	418.7	17.24	41.18		
	Nem/Liq	598.2	Not reported in paper			[348]
		4-[(4-propylcyclohexyl)carbonyl]oxy]phenyl 4-(4-propylcyclohexyl)benzoate				
$\text{C}_{32}\text{H}_{43}\text{ClO}_5$	Sol/Sme	317.2	32.22	101.58		
	Sme/Sme	336.7	0.38	1.13		
	Sme/Liq	354.3	3.77	10.64	113.35	[257]
$\text{C}_{32}\text{H}_{43}\text{ClO}_5$		2-chloro-4-methylpentyl 4-[[3-[4-(decyloxy)phenyl]-1-oxo-2-propenyl]oxy]benzoate				
	Sol/Sme	315.7	30.12	95.41		
	Sme/Sme	339.7	0.21	0.62		
$\text{C}_{32}\text{H}_{43}\text{ClO}_5$	Sme/Liq	360.2	3.77	10.47	106.50	[257]
		4-[[3-[4-(undecyloxy)phenyl]-1-oxo-2-propenyl]oxy]phenyl 2-chloro-4-methylpentanoate				

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.		
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$				
$C_{32}H_{43}ClO_5$	Sol/Smec	327.2	26.90	82.21	94.60	[257]		
	Smec/Smec	352.2	0.17	0.48				
	Smec/Liq	365.2	4.35	11.91				
$C_{32}H_{43}NO_2$	4-[[3-[4-(undecyloxy)phenyl]-1-oxo-2-propenyl]oxy]phenyl 2-chloro-3-methylpentanoate							
	Sol/Smec	323.2	22.72	70.30	80.98	[257]		
	Smec/Smec	356.7	0.25	0.70				
	Smec/Liq	372.7	3.72	9.98				
$C_{32}H_{43}NO_2$	6-n-pentyloxy-2-[4-decyloxystyryl]quinoline							
	Sol/Sol	367.6	19.56	53.21	118.84	[112]		
	Sol/Smec	375.0	22.22	59.25				
	Smec/Nem	405.6	1.60	3.94				
$C_{32}H_{43}NO_2S$	Nem/Liq	431.1	1.05	2.44	NA	[356]		
	4-isothiocyanatophenyl 4-( <i>trans</i> -4-dodecylcyclohexyl)benzoate							
	Sol/Smec	363.7	53.14	146.11				
	Smec/Nem	455.2	0.88	1.93				
$C_{32}H_{43}N_3O_3$	Nem/Liq	461.2	1.30	2.82	150.86	[154]		
	1-[4-(N-ethoxycarbonylpiperazinyl)phenyl]-3-(4-decyloxyphenyl)(1-propargyl-3-imine)							
	Sol/Smec	399.0	26.2	65.66	85.32			
	Smec/Liq	457.7	9.0	19.66				
$C_{32}H_{44}N_2O_2$	6-n-decyloxy-2-[(4'-N-hexyloxyphenylimino)methyl]quinoline							
	Sol/Smec	341.6	42.09	123.21	118.53	[112]		
	Smec/Nem	388.2	Not detected by dsc					
	Nem/Liq	406.0	1.90	4.68				
$C_{32}H_{44}N_2O_3$	N-4-hexyloxyphenyl-6-decyloxyquinoline-2-carboxamide							
	Sol/Smec	367.1	42.6	116.04	131.53	[43]		
	Smec/Liq	387.4	6.0	15.49				
$C_{32}H_{44}N_2O_4$	3-[4-[(2S)-2-butyloxypropanoyloxy]phenyl]-5-(4-decyloxyphenyl)pyrazole							
	Sol/Smec	396.8	15.9	40.07	7.37	[137]		
	Smec/Smec	400.8	Not reported in paper					
	Smec/Liq	406.8	3.0	7.37				
$C_{32}H_{44}N_4O_3$	4-[(1E)-[4-[4-(1-oxo-10-undecenyl)-1-piperazinyl]phenyl]azo]benzoic acid, butyl ester							
	Sol/Smec	333.3	1.81	5.43	28.59	[403]		
	Smec/Smec	398.4	1.01	2.54				
	Smec/Liq	462.2	9.53	20.62				
$C_{32}H_{44}O_4$	7-heptyloxy-3-(4-decyloxyphenyl)-3 <i>H</i> -1-benzopyran-4-one							
	Sol/Smec	393.6	35.0	88.92	107.28	[44]		
	Smec/Liq	419.3	7.7	18.36				
$C_{32}H_{44}O_4$	di(4'-hexylphenyl)- <i>trans</i> -cyclohexane-1,4-dicarboxylate							
	Sol/Smec	347.2	30.0	86.41	103.07	161.7		
	Smec/Smec	379.6	2.94	7.74				
	Smec/Nem	411.5	2.65	6.44				
$C_{32}H_{44}O_6$	Nem/Liq	415.5	1.03	2.48	175.3	[215]		
	di(4'-hexyloxyphenyl)- <i>trans</i> -cyclohexane-1,4-dicarboxylate							
	Sol/Smec	371.2	37.03	99.76				
	Smec/Smec	381.2	1.21	3.17				
$C_{32}H_{45}NO_3$	Smec/Nem	454.2	1.91	4.21	109.70	[220]		
	Nem/Liq	461.2	1.18	2.56				
	6-hexyloxynaphth-2-yl 5-decyloxy-2-methylpyridyl ketone							
$C_{32}H_{45}NO_3$	Sol/Sol	329.2	10.06	30.56	0.85			
	Sol/Sol	341.2	0.29					

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.		
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$				
$\text{C}_{32}\text{H}_{45}\text{NO}_5$	Sol/Smec	372.2	Not resolved by disc		126.84	[123]		
	Smec/Liq	385.2	36.76	95.43				
Note: Sol/Smec transition enthalpy is included in Smec/Liq value.								
$\text{C}_{32}\text{H}_{45}\text{NO}_5$	$4-[4-(4-\text{undecyloxyphenyliminomethyl})-3-\text{hydroxyphenoxy}]$ butyl methacrylate							
	Sol/Smec	334.6	60.25	180.07		[294]		
	Smec/Smec	351.1	Not reported in paper					
	Smec/Liq	366.7	6.90	18.82				
$\text{C}_{32}\text{H}_{45}\text{NO}_5$	$6-[4-(4-\text{nonyloxyphenyliminomethyl})-3-\text{hydroxyphenoxy}]$ hexyl methacrylate							
	Sol/Smec	328.4	29.29	89.19		[294]		
	Smec/Smec	343.4	Not reported in paper					
	Smec/Liq	364.5	6.01	16.49				
$\text{C}_{32}\text{H}_{45}\text{N}_3\text{O}_2\text{S}$	$5-(4-\text{nonyloxyphenyl})-2-(4-\text{octyloxy})\text{benzylideneamino}-1,3,4-\text{thiadiazole}$							
	Sol/Smec	380.0	35.4	93.16		[79,396,397]		
	Smec/Nem	466.7	3.9	8.36				
	Nem/Liq	474.3	1.4	2.95	104.47			
$\text{C}_{32}\text{H}_{45}\text{N}_3\text{O}_2\text{S}$	$\text{N}-[[4-(\text{nonyloxy})\text{phenyl}]\text{methylene}]-5-[4-(\text{octyloxy})\text{phenyl}]-1,2,4-\text{thiadiazol}-2\text{-amine}$							
	Sol/Smec	376.6	32.14	85.34		[283]		
	Smec/Liq	432.2	8.08	18.70	104.04			
	$5-(4-\text{nonyloxy})\text{phenyl}-2-(4-\text{n-octyloxy})\text{phenylamido}-1,3,4-\text{thiadiazole}$							
$\text{C}_{32}\text{H}_{45}\text{N}_3\text{O}_3\text{S}$	Sol/Smec	412.7	14.1	34.17		[31]		
	Smec/Nem	519.4	3.6	6.93				
	Nem/Liq	524.0	0.8	1.53	42.63			
	$5-(4-\text{decyloxy})\text{phenyl}-2-(2-\text{hydroxy}-4-\text{heptyloxy})\text{benzylideneamino}-1,3,4-\text{thiadiazole}$							
$\text{C}_{32}\text{H}_{45}\text{N}_3\text{O}_3\text{S}$	Sol/Smec	369.2	30.98	83.91		[59]		
	Smec/Liq	486.2	6.26	12.88	96.79			
$\text{C}_{32}\text{H}_{45}\text{N}_3\text{O}_3\text{S}$	$2-[(\text{E})-[(5-[4-(\text{nonyloxy})\text{phenyl}]-1,3,4-\text{thiadiazol}-2\text{-yl}]\text{jimino}]\text{methyl}]-5-(\text{octyloxy})\text{phenol}$							
	Sol/Smec	388.5	16.08	41.39		[283]		
	Smec/Liq	452.2	8.11	17.93	59.32			
	$1-(\text{trans}-4-\text{pentylcyclohexylethyne})-4-(\text{trans}-\text{pentylcyclohexylethyne})\text{benzene}$							
$\text{C}_{32}\text{H}_{46}$	Sol/Nem	405.6	15.6	38.46		[436]		
	Nem/Liq	465.0	1.4	3.01	41.47			
$\text{C}_{32}\text{H}_{46}\text{N}_2\text{O}_2\text{S}$	$6-\text{n-decyloxy}-2-(4-\text{octyloxybenzylidenamino})\text{benzothiazole}$							
	Sol/Smec	355.8	40.1	112.70		[41]		
	Smec/Nem	385.3	1.2	3.11				
	Nem/Liq	396.2	2.5	6.31	122.12			
$\text{C}_{32}\text{H}_{46}\text{N}_2\text{O}_2\text{S}_2$	$2,5\text{-bis}(4-\text{nonyloxyphenyl})\text{thiazolo}[5,4-\text{d}]$ dithiazole							
	Sol/Sol	394.1	41.43	105.13		[269]		
	Sol/Smec	420.8	51.19	121.65				
	Smec/Liq	506.9	11.64	22.96	249.74			
$\text{C}_{32}\text{H}_{46}\text{N}_2\text{O}_3$	$4-\text{propanoyl}-4'\text{-heptadecanoyloxyazobenzene}$							
	Sol/Smec	379.7	58.74	154.70		[157]		
	Smec/Liq	409.7	8.70	21.24	175.94			
	$4-\text{pentanoyl}-4'\text{-pentadecanoyloxyazobenzene}$							
$\text{C}_{32}\text{H}_{46}\text{N}_2\text{O}_3$	Sol/Sol	368.9	9.12	24.72		[157]		
	Sol/Smec	374.8	42.70	113.93				
	Smec/Liq	401.2	9.84	24.53	163.18			
	$\text{N}-[2-(6-\text{decyloxybenzothiazolyl})]-4-\text{octyloxybenzamide}$							
$\text{C}_{32}\text{H}_{46}\text{N}_2\text{O}_3\text{S}$	Sol/Sol	361.2	Not reported in paper					
	Sol/Smec	373.2	32.1	86.01				

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)						
$\text{C}_{32}\text{H}_{46}\text{N}_4\text{OS}$	Smec/Liq	406.0	4.0		9.85	95.86		[41]
	Sol/Smec	423.1	37.46		88.54			
	Smec/Nem	429.4	1.13		2.63			
	Nem/Liq	449.6	1.15		2.56	93.73		[396]
$\text{C}_{32}\text{H}_{46}\text{O}_3$	2-(4-decanoxyphenylazo)-5-(4-octyloxy)phenyl-1,3,4-thiadiazole							
	Sol/Chol	313.2	27.07		86.43			
	Chol/Liq	347.2	Not reported in paper					[208]
$\text{C}_{32}\text{H}_{46}\text{O}_4$	4-[ <i>trans</i> -4-propylcyclohexyl]ethyl]phenyl 4-[(S)-1-methylheptyl]oxy]benzoate							
	Sol/Smec	353.2	27.94		79.11			
	Smec/Nem	363.2	Not reported in paper					
	Nem/Liq	436.2	Not reported in paper					[207]
$\text{C}_{32}\text{H}_{46}\text{O}_4$	4-[ <i>trans</i> -4-pentylcyclohexyl)methoxy]phenyl 4-heptyloxybenzoate							
	Sol/Smec	323.2	35.4		109.53			
	Smec/Smec	334.2	Not reported in paper					
	Smec/Liq	342.2	Not reported in paper					[200]
$\text{C}_{32}\text{H}_{46}\text{O}_4$	4-[(S)-4-methylhexyloxy]phenyl 4-(11-dodecenoxy)benzoate							
	Sol/Smec	323.2	35.4		109.53			
	Smec/Smec	334.2	Not reported in paper					
	Smec/Liq	342.2	Not reported in paper					[200]
$\text{C}_{32}\text{H}_{46}\text{O}_4$	4-[(S)-6-methyloctyloxy]phenyl 4-(9-decenoxy)benzoate							
	Sol/Smec	336.2	43.2		128.49			
	Smec/Smec	340.2	Not reported in paper					
	Smec/Liq	348.2	Not reported in paper					[200]
$\text{C}_{32}\text{H}_{46}\text{O}_4$	4-(9-decenoxy)phenyl 4-[(S)-6-methyloctyloxy]benzoate							
	Sol/Smec	318.2	21.8		68.51			
	Smec/Chol	333.2	Not reported in paper					
	Chol/Liq	340.2	Not reported in paper					[200]
$\text{C}_{32}\text{H}_{46}\text{O}_4$	4-[ <i>trans</i> -4-propylcyclohexyl)methoxy]phenyl 4-[(S)-6-methyloctyl]oxy]benzoate							
	Sol/Smec	356.2	28.82		80.91			
	Smec/Chol	367.2	Not reported in paper					[208]
	Sol/Chol	359.2	26.12		72.72			
$\text{C}_{32}\text{H}_{46}\text{O}_4$	4-[ <i>trans</i> -4-pentylcyclohexyl)methoxy]phenyl 4-[(S)-4-methylhexyl]oxy]benzoate							
	Smec/Smec	340.2	Not reported in paper					
	Chol/Liq	421.2	Not reported in paper					[208]
	Sol/Smec	333.5	46.96		140.81			
$\text{C}_{32}\text{H}_{47}\text{BrN}_2\text{O}_4$	4-(4-tetradecyloxyphenylazoxy)phenyl 2S,3S-2-bromo-3-methylpentanoate							
	Smec/Smec	339.1	0.18		0.53			
	Smec/Liq	346.5	4.10		11.83	153.17		[47]
	Sol/Smec	328.2	43.33		132.02			
$\text{C}_{32}\text{H}_{47}\text{ClN}_2\text{O}_4$	4-(4-tetradecyloxyphenylazoxy)phenyl 2S,3S-2-chloro-3-methylpentanoate							
	Smec/Smec	339.2	0.11		0.32			
	Smec/Liq	348.6	4.31		12.36	144.70		[47]
	Sol/Smec	316.2	45.47		143.80			
$\text{C}_{32}\text{H}_{47}\text{NO}$	N-(4-pentyloxybenzylidene)-4-tetradecylaniline							
	Smec/Smec	324.9	2.48		7.63			
	Smec/Nem	342.5	0.72		2.10			
	Nem/Liq	344.7	1.83		5.31	158.84		[147]
$\text{C}_{32}\text{H}_{47}\text{N}_3\text{O}_2\text{S}$	6-n-decyloxy-2-(4-nonyloxyphenylazo)benzothiazole							
	Smec/Smec	347.7	22.7		65.29			
	Smec/Nem	389.4	0.9		2.31			
	Nem/Liq	396.5	1.8		4.54	72.14		[41]
$\text{C}_{32}\text{H}_{48}\text{N}_2\text{O}$	5-octyl-2-[4-[3-[ <i>trans</i> -4-pentylcyclohexyl]oxy]-1-propenyl]phenyl]pyrimidine							
	Smec/Smec	360.2	17.2		47.75			
	Smec/Nem	361.2	Not reported in paper					

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated) (estimated)	Ref.	
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$			
$C_{32}H_{48}N_2O_2$	Nem/Liq	411.2	Not reported in paper			[198]	
	Sol/Chol	387.2	13.8	35.64			
	Chol/Liq	438.2	Not reported in paper			[201]	
$C_{32}H_{48}O_2S$			4-pentylbenzenethio-4'-tetradecyloxybenzoate				
	Sol/SmeC	342.0	45.02	131.64			
	SmeC/Liq	361.7	7.82	21.62	153.26	NA	[217]
$C_{32}H_{49}NO$			4-butyl-N-[[4-(pentadecyloxy)phenyl]methylene]benzenamine				
	Sol/SmeC	340.9	32.48	95.28			
	SmeC/SmeC	347.8	0.02	0.06			
	SmeC/SmeC	349.5	3.43	9.81			
$C_{32}H_{49}NO$	SmeC/Liq	356.1	6.57	18.45	123.55		[240]
			N-(p-n-nonyloxybenzylidene)-4-(n-decyl)aniline				
	Sol/SmeC	329.4	58.0	176.08			
	SmeC/SmeC	348.5	4.16	11.94			
$C_{32}H_{49}NO_3$	SmeC/Liq	356.7	8.58	24.05	212.07		[11]
			N-(4-methoxyphenyl)- $\alpha$ -(4-octadecyloxyphenyl)nitrone				
	Sol/SmeC	392.2	79.50	202.70			
$C_{32}H_{50}N_2$	SmeC/Liq	400.2	4.94	12.34	215.04		[162]
			5-heptyl-2-[4-[4-(trans-4-pentylcyclohexyl)butyl]phenyl]pyrimidine				
	Sol/Nem	350.2	18.6	53.11			
$C_{32}H_{50}N_2O$	Nem/Liq	389.2	Not reported in paper				[198]
			4-butyl-4'-hexadecyloxyazobenzene				
	Sol/SmeC	344.1	35.58	103.40			
$C_{32}H_{50}N_2O$	SmeC/Liq	349.6	11.12	31.81	135.21		[141]
			5-octyl-2-[4-[3-[(trans-4-pentylcyclohexyl)oxy]propyl]phenyl]pyrimidine				
	Sol/SmeC	352.2	25.3	71.83			
$C_{32}H_{50}N_2O_3$	SmeC/Nem	361.2	Not reported in paper				
	Nem/Liq	406.2	Not reported in paper				[198]
			4,4'-didecyloxyazoxybenzene				
$C_{32}H_{50}N_2O_3$	Sol/SmeC	351.4	38.58	109.79			
	SmeC/Nem	393.8	2.32	5.89			
	Nem/Liq	396.6	3.15	7.94	123.62		[179]
$C_{32}H_{52}O_6$			(S)-4-(2-ethylheptyloxycarbonyl)phenyl 4-undecanoyloxytolane-4'-carboxylate				
	Sol/SmeC	329.6	34.8	105.58			
	SmeC/SmeC	330.1	2.47	7.48			
	SmeC/SmeC	354.9	0.43	1.21			
$C_{32}H_{53}N_3O_6$	SmeC/Liq	370.1	1.72	4.65	118.92		[173]
			4,4'-bis[4-(4-hexyloxybenzoyloxy)benzylideneamino]diphenylamine				
	Sol/Nem	499.2	42.0	84.13			
$C_{32}H_{54}OS$	Nem/Liq	604.2	1.6	2.65	86.78		[284]
			cholesteryl thiopentanoate				
	Sol/Chol	364.2	23.8	65.5			
$C_{32}H_{55}IO_2$	Chol/Liq	377.9	0.24	0.6	66.1	NA	[155,312]
			3-[2-(2-iodopentyl)oxy]-cholest-5-ene				
	Sol/Nem	336.5	18.41	54.71			
$C_{32}H_{56}N_2O_2$	Nem/Liq	340.0	0.42	1.24	55.95		[316]
			N,N'-didodecanoyl-2,6-dimethylbenzene-1,4-diamine				

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.		
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$				
$\text{C}_{32}\text{H}_{56}\text{N}_2\text{O}_2$	Sol/Meso	389.2	47.0	120.76	53.16	[36]		
	Meso/Liq	401.2	13.0	32.40				
$\text{C}_{32}\text{H}_{56}\text{N}_2\text{O}_2$	N,N'-diundecanoyl-2,3,5,6-tetramethylbenzene-1,4-diamine					[36]		
	Sol/Meso	425.2	41.0	96.43	148.79			
	Meso/Meso	515.2	11.0	21.35				
	Meso/Liq	548.2	17.0	31.01				
$\text{C}_{32}\text{H}_{56}\text{N}_2\text{O}_2$	N,N'-diundecanoyl-3,4,5,6-tetramethylbenzene-1,2-diamine					[61]		
	Sol/Meso	398.2	17.0	42.69	78.52			
	Meso/Meso	467.2	2.0	4.28				
$\text{C}_{32}\text{H}_{56}\text{N}_2\text{O}_2$	Meso/Liq	507.2	16.0	31.55	177.76	[61]		
	N,N'-didodecylbenzene-1,2-dicarboxamide							
	Sol/Meso	364.2	53.0	145.52				
$\text{C}_{33}\text{H}_{22}\text{F}_{26}\text{O}_5$	Meso/Liq	372.2	12.0	32.24	110.45	[128]		
	4-(3,3,4,4,5,5,6,6,6,7,7,8,8,8-tridecafluoroctyloxycarbonyl)phenyl 4-[(perfluorohexyl)pentyloxy]benzoate							
	Sol/Smec	376.1	34.11	90.69				
$\text{C}_{33}\text{H}_{32}\text{F}_4\text{O}_5$	Smec/Liq	399.8	7.90	19.76	69.19	[91]		
	(S)-2-methylbutoxycarbonyl]phenyl 4-[(4-hexyloxy-2,3,4,5-tetrafluorophenyl)ethynyl]benzoate							
	Sol/Smec	362.0	24.15	66.71				
	Smec/Smec	363.0	Not detected by dsc					
$\text{C}_{33}\text{H}_{34}\text{F}_4\text{O}_4$	Smec/Chol	406.0	0.60	1.48	110.10	[96]		
	Chol/Liq	427.9	0.43	1.00				
	pentyl 4-[4-((4-hexyloxy-2,3,5,6-tetrafluorophenyl)ethynyl)benzyloxy]benzoate							
$\text{C}_{33}\text{H}_{34}\text{F}_4\text{O}_4$	Sol/Smec	363.1	37.74	103.94	103.94	[96]		
	Smec/Nem	368.8	1.29	3.50				
	Nem/Liq	372.0	0.99	2.66				
$\text{C}_{33}\text{H}_{34}\text{F}_4\text{O}_4$	propyl 4-[4-((4-octyloxy-2,3,5,6-tetrafluorophenyl)ethynyl)benzyloxy]benzoate					[96]		
	Sol/Smec	364.2	40.25	110.52	115.25			
	Smec/Nem	375.1	0.62	1.65				
$\text{C}_{33}\text{H}_{34}\text{F}_4\text{O}_4$	Nem/Liq	382.5	1.18	3.08	73.30	[124]		
	(S)-2-methylbutoxy]phenyl 4-[(4-heptyloxy-2,3,5,6-tetrafluorophenyl)ethynyl]benzoate							
	Sol/Nem	371.8	26.88	72.30				
$\text{C}_{33}\text{H}_{34}\text{N}_6\text{O}_6$	Nem/Liq	411.0	0.41	1.00	108.91	[422]		
	$\alpha,\omega$ -bis(4-nitroazobenzene-4'-oxy)nonane							
	Sol/Nem	424.2	44.79	105.59				
$\text{C}_{33}\text{H}_{35}\text{ClN}_2\text{O}_3$	Nem/Liq	457.2	1.52	3.32	66.25	[274]		
	2"-[4-[4'-decyloxybenzoyloxy)-3-chlorophenylazo]naphthalene							
	Sol/Nem	369.2	24.06	65.17				
$\text{C}_{33}\text{H}_{35}\text{F}_{13}\text{O}_5$	Nem/Liq	443.2	0.48	1.08	124.15	[125]		
	4-(3,3,4,4,5,5,6,6,7,7,8,8,8-tridecafluoroctyloxycarbonyl)phenyl 4-(undecyloxy)benzoate							
	Sol/Smec	362.4	35.55	98.10				
$\text{C}_{33}\text{H}_{36}$	Smec/Liq	387.0	10.08	26.05	41.21	[55]		
	4'-[2-(4-propylphenyl)-1-ethynyl]-2'-ethyl-4-hexyltolane							
	Sol/Nem	330.1	12.7	38.47				
$\text{C}_{33}\text{H}_{36}$	Nem/Liq	393.6	1.08	2.74	53.18	[55]		
	4'-[2-(4-hexylphenyl)-1-ethynyl]-2'-ethyl-4-propyltolane							
	Sol/Nem	293.2	14.7	50.14				
$\text{C}_{33}\text{H}_{36}\text{N}_2\text{O}_2$	Nem/Liq	381.0	1.16	3.04	53.18	[55]		
	2,3-dicyano-4-pentylphenyl 4'-heptyl-4-biphenyl-1-carboxylate							

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$\text{C}_{33}\text{H}_{36}\text{N}_2\text{O}_2$	Sol/Nem	371.2	31.38	84.54		
	Nem/Liq	386.2		Not reported in paper		[202]
$\text{C}_{33}\text{H}_{36}\text{N}_4\text{O}_2$	Sol/Nem	366.2	31.80	86.84		
	Nem/Liq	383.2		Not reported in paper		[202]
	Sol/Nem	415.2	52.47	126.37		
$\text{C}_{33}\text{H}_{36}\text{N}_4\text{O}_2$	Nem/Liq	434.2	4.66	10.73	137.10	[67]
	Sol/Nem	429.2	51.03	118.90		
	Nem/Liq	474.2	6.11	12.88	131.78	[67]
$\text{C}_{33}\text{H}_{37}\text{ClF}_2\text{O}_5$	Sol/SmeC	334.5	13.05	39.01		
	SmeC/Nem	401.6	1.4	3.49		
	Nem/Liq	424.9	0.8	1.88	44.38	[121]
	Sol/SmeC	371.3	49.2	132.51		
$\text{C}_{33}\text{H}_{37}\text{ClF}_2\text{O}_5$	SmeC/Nem	416.0	1.4	3.37		
	Nem/Liq	439.7	0.82	1.86	137.74	[121]
	Sol/Sol	400.0	7.7	19.25		
$\text{C}_{33}\text{H}_{37}\text{NO}$	Sol/SmeC	506.0	3.9	7.71		
	SmeC/SmeC	540.0	4.2	7.78		
	SmeC/Nem	598.0	2.0	3.34		
	Nem/Liq	612.0	1.5	2.45	40.53	[185]
	Sol/SmeC	422.7	44.41	105.06		
$\text{C}_{33}\text{H}_{38}\text{N}_4\text{O}_2\text{S}$	SmeC/Nem	432.7	2.17	5.02		
	Nem/Liq	447.2	1.09	2.44	112.52	[79]
	Sol/SmeC	373.7	45.06	120.58		
$\text{C}_{33}\text{H}_{39}\text{BrO}_2$	SmeC/Liq	382.7	4.22	11.03	131.61	[297]
	Sol/SmeC	400.7	47.57	118.72		
	SmeC/Liq	416.2	4.39	10.55	129.27	[250]
$\text{C}_{33}\text{H}_{39}\text{ClO}_2$	Sol/SmeC	371.7	34.98	94.11		
	SmeC/Liq	383.7	4.18	10.89	105.00	[297]
	Sol/SmeC	385.2	10.29	26.71		
$\text{C}_{33}\text{H}_{39}\text{NO}_6$	SmeC/Liq	455.2	1.01	2.22	28.93	[58]
	Sol/SmeC	323.2		Not reported in paper		
	SmeC/SmeC	336.0	1.64	4.88		
$\text{C}_{33}\text{H}_{40}\text{ClNO}_4$	SmeC/SmeC	367.3	0.26	0.71		
	SmeC/Liq	389.2	5.14	13.21		[249]
	Sol/Nem	404.2	30.96	76.60		
	Nem/Liq	411.2		Not reported in paper		[202]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)						
$C_{33}H_{41}NO_3$		4-isopropylphenyl 4-(4-decyloxybenzylideneamino)benzoate						
	Sol/Smec	363.2	38.0		104.63			
	Smec/Nem	440.2		Not reported in paper				
	Nem/Liq	453.2	4.6		10.15	114.78	156.6	[292]
Note: Smec/Nem transition enthalpy is included in the Nem/Liq value.								
$C_{33}H_{41}NO_3$		4-(4'-dodecyloxybenzoyloxybenzylidene)-2"-aniline						
	Sol/Smec	350.4	34.48		98.40			
	Smec/Nem	362.9	0.78		2.15			[399]
$C_{33}H_{41}NO_3$	Nem/Liq	379.3	0.79		2.08	102.63		
		4-(4-decyloxybenzylideneamino)phenyl 4-isopropylbenzoate						
	Sol/Nem	375.2	26.0		69.30		156.6	[292]
$C_{33}H_{41}O$	Nem/Liq	442.2	1.5		3.39	72.69		
		3-methyl-4-[2-( <i>trans</i> -4-pentylcyclohexyl)ethynyl]-1-[(4-pentyloxyphenyl)ethynyl]benzene						
	Sol/Nem	352.2	26.9		76.38			[76]
$C_{33}H_{42}N_2O_2$	Nem/Liq	464.2	0.4		0.86	77.24		
		2-(4-butoxyphenyl)-5-[4-[(4-ethyl-2,3-undecadienyl)oxy]phenyl]pyrimidine						
	Sol/Smec	342.2	28.7		83.87			
$C_{33}H_{42}N_2O_2$	Smec/Nem	347.2	4.8		13.82			[398]
	Nem/Liq	351.7	0.6		1.71	99.40		
		2-(4-butoxyphenyl)-5-[4-[(3-methyl-3,4-dodecadienyl)oxy]phenyl]pyrimidine						
$C_{33}H_{42}N_2O_2$	Sol/Smec	305.2	18.5		60.62			
	Smec/Nem	374.2		Not reported in paper				[398]
	Nem/Liq	394.2		Not reported in paper				
$C_{33}H_{42}N_2O_3$		4-[5-[(S)-6-methyloctyl]-2-pyrimidinyl]phenyl 4-(6-heptenoxy)benzoate						
	Sol/Smec	318.2	32.3		101.51			
	Smec/Chol	321.2		Not reported in paper				[201]
$C_{33}H_{42}N_2O_4$	Chol/Liq	417.2		Not reported in paper				
		4-[(4-ethoxyphenyl)azo]-2,3-dimethylphenyl 4-(decyloxy)benzoate						
	Sol/Nem	386.2	35.0		90.63			
$C_{33}H_{42}N_2O_4$	Nem/Liq	479.2	1.4		2.92	93.55		[339]
		4-[(4-ethoxyphenyl)azo]-3,6-dimethylphenyl 4-(decyloxy)benzoate						
	Sol/Nem	378.2	70.0		185.09			
$C_{33}H_{42}N_2O_4$	Nem/Liq	414.2	2.1		5.07	190.16		[339]
		4-[(4-ethoxyphenyl)azo]-3,5-dimethylphenyl 4-(decyloxy)benzoate						
	Sol/Nem	363.2	42.0		115.64			
$C_{33}H_{42}N_2O_4$	Nem/Liq	427.2	2.0		4.68	120.32		[339]
		4-[(4-ethoxyphenyl)azo]-2,6-dimethylphenyl 4-(decyloxy)benzoate						
	Sol/Nem	369.2	45.0		121.89			
$C_{33}H_{42}N_2O_4$	Nem/Liq	419.2	2.0		4.77	126.66		[339]
		4-[(4-ethoxyphenyl)azo]phenyl 4-(dodecyloxy)benzoate						
	Sol/Nem	375.2	50.0		133.26			
$C_{33}H_{42}O_5$	Nem/Liq	490.2	1.3		2.65	135.91		[339]
		4-hydroxypentyll 4-[[4'-(octyloxy)[1,1'-biphenyl]-4-yl]oxy]methyl]benzoate						
	Sol/Sol	354.4	19.92		56.21			
	Sol/Smec	399.0	2.33		5.84			
	Smec/Smec	429.5	3.73		8.69			
$C_{33}H_{42}O_5$	Smec/Smec	436.8	0.41		0.94			
	Smec/Liq	441.0	16.55		37.53	109.21		[304]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{33}H_{42}O_7$	Sol/Smec	385.6	20.4	52.90		
	Smec/Smec	429.2	Not observed by dsc			
	Smec/Liq	434.5	5.0	11.51	64.41	[286]
$C_{33}H_{43}NO_2$	N-(2-hydroxy-4-ethoxybenzylidene)-4"-dodecylphenylaniline					
	Sol/Sol	347.7	Not reported in paper			
	Sol/Smec	371.2	17.78	47.90		
	Smec/Smec	421.7	0.92	2.18		
$C_{33}H_{43}NO_4$	Smec/Liq	503.2	6.11	12.14		[323]
	4'-(2-diethylamino-ethoxy)biphenyl 4-octyloxybenzoate					
	Sol/Smec	340.4	5.34	15.69		
	Smec/Smec	382.9	2.35	6.14		
$C_{33}H_{44}N_2O_2S$	Smec/Liq	415.2	0.97	2.34	24.17	[427]
	2-(4-butoxyphenyl)-5-[4-[(2-butyl-2,3-undecadienyl)oxy]phenyl]-1,3,4-thiadiazole					
	Sol/Smec	341.2	36.4	106.68		
	Smec/Nem	347.2	1.68	4.84		
$C_{33}H_{44}N_2O_3$	Nem/Liq	351.7	0.98	2.78	114.30	[398]
	1-[4-(N-ethoxycarbonylpiperazinyl)phenyl]-4-(4-decyloxyphenyl)-but-3-en-1-yne					
	Sol/Smec	427.1	17.7	41.44		
	Smec/Liq	471.0	4.0	0.85	42.29	[154]
$C_{33}H_{44}N_2O_3$	4-[5-[(S)-5-methylheptyl]-2-pyrimidinyl]phenyl 4-octyloxybenzoate					
	Sol/Chol	344.2	25.4	73.79		
	Chol/Liq	415.2	Not reported in paper			[201]
$C_{33}H_{44}N_2O_3$	4-[5-[(S)-6-methyloctyl]-2-pyrimidinyl]phenyl 4-heptyloxybenzoate					
	Sol/Smec	329.2	33.9	102.98		
	Smec/Chol	337.2	Not reported in paper			
	Chol/Liq	417.2	Not reported in paper			[201]
$C_{33}H_{44}O_2$	8-pentyltricyclo[4.4.0 <sup>3,8</sup> ]dec-1-yl 4'-pentyl[1,1'-biphenyl]-4-carboxylate					
	Sol/Chol	341.2	12.97	38.01		
	Chol/Liq	401.2	Not reported in paper			[382]
Note: Authors report only a single transition enthalpy, which we have assumed is for the Sol/Chol transition.						
$C_{33}H_{44}O_4$	4-[[4-pentylcyclohexyl]carbonyl]oxy]phenyl 4-(4-ethylcyclohexyl)benzoate					
	Sol/Smec	361.2	13.39	37.07		
	Smec/Nem	401.9	15.94	39.66		
	Nem/Liq	592.2	Not reported in paper			[348]
$C_{33}H_{44}O_4$	4-[[4-butylcyclohexyl]carbonyl]oxy]phenyl 4-(4-propylcyclohexyl)benzoate					
	Sol/Smec	380.1	10.50	27.62		
	Smec/Nem	405.9	12.05	29.69		
	Nem/Liq	599.2	Not reported in paper			[348]
$C_{33}H_{45}ClO_5$	2-chloro-4-methylpentyl 4-[[3-[4-(undecyloxy)phenyl]-1-oxo-2-propenyl]oxy]benzoate					
	Sol/Smec	328.7	38.07	115.82		
	Smec/Smec	337.7	0.25	0.74		
	Smec/Liq	353.7	3.77	10.66	127.22	[257]
$C_{33}H_{45}ClO_5$	2-chloro-3-methylpentyl 4-[[3-[4-(undecyloxy)phenyl]-1-oxo-2-propenyl]oxy]benzoate					
	Sol/Smec	335.7	38.49	114.66		
	Smec/Smec	341.7	0.17	0.50		
	Smec/Liq	361.2	4.06	11.24	126.40	[257]
$C_{33}H_{45}ClO_5$	4-[[3-[4-(dodecyloxy)phenyl]-1-oxo-2-propenyl]oxy]phenyl 2-chloro-4-methylpentanoate					
	Sol/Smec	335.2	31.38	93.62		
	Smec/Smec	353.2	0.21	0.59		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.	
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$			
$C_{33}H_{45}ClO_5$	Smec/Liq	365.7	4.18	11.43	105.64	[257]	
	Sol/Smec	329.7	42.17	127.90			
	Smec/Smec	357.7	0.21	0.59			
	Smec/Liq	373.2	4.18	11.20	139.69		
$C_{33}H_{46}N_2O_2$	4-[3-[4-(dodecyloxy)phenyl]-1-oxo-2-propenyl]oxy]phenyl 2-chloro-3-methylpentanoate					[257]	
	Sol/Smec	344.2	69.9	203.08			
	Smec/Nem	383.1	0.19	0.50			
	Nem/Liq	405.5	1.72	4.24	207.82		
$C_{33}H_{46}N_2O_3$	N-4-heptyloxyphenyl-6-decyloxyquinoline-2-carboxamide					[43]	
	Sol/Smec	372.5	48.6	130.47			
	Smec/Liq	389.4	6.5	16.69	147.16		
$C_{33}H_{46}N_2O_3$	N-[4-[5-oxo-4-[(1-oxodecyl)amino]-1,3,6-cycloheptatrien-1-yl]-phenyl]decamide					[251]	
	Sol/Sol	418.2	6.9	16.50			
	Sol/Smec	424.2	Not reported in paper				
	Smec/Liq	427.2	8.4	19.66			
$C_{33}H_{46}O_4$	7-octyloxy-3-(4-decyloxyphenyl)-3 <i>H</i> -1-benzopyran-4-one					[44]	
	Sol/Smec	394.1	30.9	78.41			
	Smec/Liq	423.2	7.4	17.49	95.90		
$C_{33}H_{47}NOS$	3-[4-( <i>trans</i> -4-heptylcyclohexyl)phenyl]-5-(5-heptyl-2-thienyl)isoxazole					[131]	
	Sol/Smec	355.8	24.09	67.71			
	Smec/Nem	438.4	1.81	4.13			
	Nem/Liq	451.6	1.87	4.14	75.98		
$C_{33}H_{47}NO_3$	6-hexyloxynaphth-2-yl 5-undecyloxy-2-methypyridyl ketone					[123]	
	Sol/Sol	368.2	0.27	0.73			
	Sol/Smec	385.2	38.41	99.71			
	Smec/Liq	391.2	10.26	26.23	126.67		
$C_{33}H_{47}NO_3$	pentyl 4-(4-dodecyloxybenzalmino)cinnamate					[181]	
	SolSmec	347.8	28.0	80.51			
	Smec/Smec	369.5	5.48	14.83			
	Smec/Smec	380.5	0.61	1.60			
$C_{33}H_{47}NO_5$	Smec/Liq	406.3	8.44	20.77	117.71		[294]
	4-[4-(4-dodecyloxyphenyliminomethyl)-3-hydroxyphenoxy]butyl methacrylate						
	Sol/Smec	329.7	59.12	179.31			
	Smec/Smec	351.4	Not reported in paper				
$C_{33}H_{47}NO_5$	Smec/Liq	367.5	7.95	21.63			
	6-[4-(4-decyloxyphenyliminomethyl)-3-hydroxyphenoxy]hexyl methacrylate					[294]	
	Sol/Smec	321.5	30.02	93.34			
	Smec/Smec	356.5	Not reported in paper				
$C_{33}H_{47}NO_5$	Smec/Liq	369.8	6.82	18.44			
$C_{33}H_{47}N_3O_2S$	5-(4-decyloxyphenyl)-2-(4-octyloxy)benzylideneamino-1,3,4-thiadiazole					[79,396,397]	
	Sol/Smec	376.5	38.2	101.46			
	Smec/Nem	467.1	4.1	8.78			
	Nem/Liq	472.0	1.3	2.75	112.99		
$C_{33}H_{47}N_3O_3S$	5-(4-decyloxyphenyl)-2-(4-n-octyloxy)phenylamido-1,3,4-thiadiazole					[31]	
	Sol/Smec	413.9	16.3	39.38			
	Smec/Nem	518.2	3.8	7.33			
	Nem/Liq	521.5	0.9	1.73	48.44		
$C_{33}H_{47}N_3O_3S$		5-(4-decyloxy)phenyl-2-(2-hydroxy-4-octyloxy)benzylideneamino-1,3,4-thiadiazole					

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.	
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$			
$C_{33}H_{47}N_3O_3S$	Sol/Smec	371.2	26.60	71.66	88.44	[59]	
	Smec/Liq	486.2	8.16	16.78			
$C_{33}H_{48}N_2O_2S$	2-[{(E)-[(5-[4-(decyloxy)phenyl]-1,3,4-thiadiazol-2-yl]imino)methyl]-5-(octyloxy)phenol}					[283]	
	Sol/Smec	368.5	26.66	72.35	92.60		
	Smec/Liq	488.9	9.90	20.25			
$C_{33}H_{48}N_2O_2S$	6-n-decyloxy-2-(4-nonyloxybenzylidenamino)benzothiazole					[41]	
	Sol/Smec	354.0	38.3	108.19	120.41		
	Smec/Nem	389.6	1.9	4.88			
$C_{33}H_{48}N_2O_3$	Nem/Liq	394.9	2.9	7.34	120.41	[41]	
	4-propanoyl-4'-octadecanoyloxyazobenzene						
	Sol/Smec	380.7	59.62	156.61	178.51		
$C_{33}H_{48}N_2O_3$	Smec/Liq	408.7	8.95	21.90	[157]		
	4-pentanoyl-4'-hexadecanoyloxyazobenzene						
	Sol/Sol	333.9	4.69	14.05		179.91	
$C_{33}H_{48}N_2O_3$	Sol/Sol	361.8	7.15	19.76	[157]		
	Sol/Smec	377.4	45.63	120.90		179.91	
	Smec/Liq	399.6	10.07	25.20			
$C_{33}H_{48}N_2O_3S$	N-[2-(6-decyloxybenzothiazolyl)]-4-nonyloxybenzamide					[41]	
	Sol/Sol	346.7	7.9	22.79	111.49		
	Smec/Liq	408.3	4.6	11.27			
$C_{33}H_{48}N_4OS$	2-(4-decanoxyphenylazo)-5-(4-nonyloxy)phenyl-1,3,4-thiadiazole					[396]	
	Sol/Smec	422.7	44.41	105.06	112.52		
	Smec/Nem	432.7	2.17	5.02			
$C_{33}H_{48}N_4O_3$	Nem/Liq	447.2	1.09	2.44	112.52	[344]	
	4-[(1E)-[4-[4-(1-oxohexyl)-1-piperazinyl]phenyl]azo]benzoic acid, decyl ester						
	Sol/Smec	364.9	23.32	63.91	89.83		
$C_{33}H_{48}N_4O_3$	Smec/Smec	405.6	1.48	3.65	[344]		
	Smec/Liq	470.6	10.48	22.27			
$C_{33}H_{48}O$	4-octyloxy-4'-(pentylbicyclo[2.2.2]octan-1-yl)biphenyl					[50]	
	Sol/Smec	362.4	16.32	45.03	Not given in paper		
	Smec/Smec	463.1	3.35	7.23			
	Smec/Nem	489.6					
	Nem/Liq	490.2	6.69	13.65	65.91		
Note: Enthalpy for Smec/Nem transition is included in the Nem/Liq value.							
$C_{33}H_{48}O_3$	4-(2-methylbutoxy)benzoic acid, 4-[2-( <i>trans</i> -4-heptylcyclohexyl)ethyl]phenyl ester					[199, 208]	
	Sol/Chol	339.2	21.91	64.59	Not reported in paper		
	Chol/Liq	404.2					
$C_{33}H_{48}O_3$	4-(heptyloxy)benzoic acid, 4-[2-( <i>trans</i> -4-pentylcyclohexyl)ethyl]phenyl ester					[199, 207]	
	Sol/Smec	340.2	23.19	68.17	Not reported in paper		
	Smec/Nem	341.2					
$C_{33}H_{48}O_3$	Nem/Liq	433.2			Not reported in paper	[199, 207]	
	4-( <i>trans</i> -4-pentylcyclohexyl)phenyl 4-[(S)-6-methylhexyl]oxy]benzoate						
	Sol/Smec	348.2	17.98	51.64	Not reported in paper		
$C_{33}H_{48}O_3$	Smec/Chol	379.2			[208]		
	Chol/Liq	436.2					
$C_{33}H_{48}O_4$	4-[(S)-6-methyloctyloxy]phenyl 4-(10-undecenyloxy)benzoate						
	Sol/Smec	326.2	41.7	127.84	Not reported in paper	[208]	
	Smec/Smec	342.2					

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$\text{C}_{33}\text{H}_{48}\text{O}_4$	Smec/Liq	351.2	Not reported in paper			[200]
	Sol/Smec	318.2	35.3	110.94		
	Smec/Chol	336.2	Not reported in paper			
	Chol/Liq	343.2	Not reported in paper			[200]
$\text{C}_{33}\text{H}_{48}\text{O}_4$	Sol/Smec	354.2	19.23	54.29		
	Smec/Nem	380.2	Not reported in paper			
	Nem/Liq	436.2	Not reported in paper			[207]
	Sol/Sol	350.1	6.9	19.71		
$\text{C}_{33}\text{H}_{49}\text{NO}_5$	Sol/Smec	382.0	44.7	117.02		
	Smec/Cube	417.7	1.3	3.11		
	Cube/Liq	465.8	2.7	5.80	145.64	[133]
	Sol/Sol	329.2	2.6	7.90		
$\text{C}_{33}\text{H}_{49}\text{N}_3\text{O}_2\text{S}$	Sol/Smec	351.1	21.5	61.24		
	Smec/Nem	387.9	6.0	15.47		
	Nem/Liq	396.3	Not given in paper			[41]
	Sol/Chol	378.2	17.1	45.21		
$\text{C}_{33}\text{H}_{50}\text{N}_2\text{O}_2$	Chol/Liq	414.2	Not reported in paper			[201]
	Sol/Chol	333.7	21.55	64.58		
	Chol/Liq	398.7	Not reported in paper			[382]
Note: Authors report only a single transition enthalpy, which we have assumed is for the Sol/Chol transition.						
$\text{C}_{33}\text{H}_{51}\text{NO}$	Sol/Smec	338.0	41.68	123.31		
	Smec/Smec	358.3	6.28	17.53		
	Smec/Liq	362.8	10.26	28.28	169.12	[240]
	Sol/Smec	327.5	38.91	118.81		
$\text{C}_{33}\text{H}_{51}\text{NO}$	Smec/Smec	361.5	5.13	14.19		
	Smec/Liq	364.3	7.59	20.83	153.83	[256]
	Sol/Nem	342.2	15.5	45.30		
	Nem/Liq	383.2	Not reported in paper			[189]
$\text{C}_{33}\text{H}_{52}\text{N}_2\text{O}$	Sol/Smec	339.0	28.30	83.48		
	Smec/Smec	350.4	4.16	11.87		
	Smec/Liq	358.1	8.95	24.99	120.34	198.3
	Sol/Smec	337.0	27.99	83.06		
$\text{C}_{33}\text{H}_{52}\text{N}_2\text{O}$	Smec/Smec	354.4	2.64	7.44		
	Smec/Liq	363.1	6.82	18.78	109.28	198.3
	Sol/Smec	339.5	7.53	22.18		
	Smec/Nem	428.5	10.63	24.81		
$\text{C}_{33}\text{H}_{54}$	Nem/Liq	434.2	2.13	0.91	51.90	[348]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.		
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$				
$C_{33}H_{54}N_2O_6$	Sol/Smec	432.2	32.0	74.04	110.54	[284]		
	Smec/Liq	493.2	18.0	36.50				
$C_{33}H_{54}O_4$	Sol/Nem	407.2	24.5	60.17	63.73	[265, 266]		
	Nem/Liq	421.2	1.5	3.56				
$C_{33}H_{54}O_6$	(S)-4-(2-ethylheptyloxycarbonyl)phenyl 4-dodecanoyloxytolane-4'-carboxylate					[173]		
	Sol/Smec	347.5	44.1	126.91	32.11			
	Smec/Smec	354.2	0.54	1.52				
$C_{33}H_{56}N_2O_4$	Smec/Liq	366.6	1.35	3.68	[193]			
	N,N'-dioctanoyl-2,5,6-trimethyl-4-octanoyloxy-1,3-benzenediamine							
	Sol/Meso	403.2	8.0	19.84		192.8		
$C_{33}H_{56}OS$	Meso/Meso	459.2	20.0	43.55	[155,312]			
	Meso/Liq	474.2	4.0	8.44		71.83		
	Sol/Liq	368.2	26.2	71.2		71.2		
$C_{33}H_{56}O_6$	Chol/Liq	380.9	0.42	1.1	74.66	[98]		
	<i>cis, cis</i> -(3,5-dihydroxycyclohexyl) 3,4-bis(decyloxy)benzoate							
	Sol/Sol	302.2	11.2	37.06				
$C_{33}H_{57}N_3O_3$	Sol/Meso	345.2	12.0	34.76	82.79	[190]		
	Meso/Liq	386.7	1.1	2.84				
	N,N',N"-trioctanoyl-2,4,6-trimethyl-1,3,5-benzenetriamine							
$C_{33}H_{58}N_2O_2$	Sol/Meso	512.2	30.0	58.57	130.64	[36]		
	Meso/Nem	462.2	15.0	32.45				
	Nem/Liq	463.2	Very small					
$C_{33}H_{58}O_2$	5 $\alpha$ -cholestane-3 $\beta$ -ol hexanoate					[311]		
	Sol/Liq	360.15	29.6	82.1	124.9			
	Chol/Liq	340.15	0.63	1.8	84.0			
Note: Liquid crystalline phase detected upon cooling.								
$C_{34}H_{24}F_{26}O_5$	4-(3,3,4,4,5,5,6,6,7,7,8,8,8-tridecafluoroctyloxycarbonyl)phenyl 4-[(perfluorohexyl)hexyloxy]benzoate					[128]		
	Sol/Smec	393.5	41.60	105.72	121.01			
	Smec/Smec	395.9	2.25	5.68				
$C_{34}H_{31}NO_2$	Smec/Liq	399.5	3.84	9.61	[254]			
	[1-(4-hexylbiphenyl)-3-(4"-cyanobiphenyl)]-propane-1,3-dione							
	Sol/Nem	463.7	37.70	81.30		81.58		
$C_{34}H_{32}N_2O_3$	Nem/Liq	467.2	0.13	0.28	[385]			
	4-[(1E)-2-naphthalenylazo]-1-naphthalenyl 4-(heptyloxy)benzoate							
	Sol/Nem	408.2	12.65	30.99		31.60		
$C_{34}H_{34}F_4O_5$	Nem/Liq	443.2	0.27	0.61	[91]			
	4-[(S)-2-methylbutoxycarbonyl]phenyl 4-[(4-heptyloxy-2,3,4,5-tetrafluorophenyl)ethynyl]benzoate							
	Sol/Smec	351.0	23.22	66.15		68.16		
$C_{34}H_{34}N_4O_6$	Smec/Smec	351.8	Not detected by dsc			[91]		
	Smec/Chol	402.8	0.53	1.32				
	Chol/Liq	420.7	0.29	0.69				
$\alpha,\omega$ -bis(4-ethoxyazobenzene-4'-carbonyloxy)butane								

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.		
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$				
$\text{C}_{34}\text{H}_{34}\text{N}_4\text{O}_6\text{S}_2$	Sol/Nem	441.3	47.7	108.09	122.46	[107]		
	Nem/Liq	494.0	7.1	14.37				
	Sol/Smec	402.2	37.1	92.24		[23]		
	Smec/Nem	552.2	1.5	2.72				
$\text{C}_{34}\text{H}_{34}\text{O}_6$	Nem/Liq	565.2	0.6	1.06	96.02			
	<i>bis</i> [4-(5-n-pentylthio-1,3,4-oxadiazole-2-yl)phenyl] terephthalate							
	Sol/Smec	444.4	34.2	76.96	317.0	[12]		
	Smec/Nem	612.8	1.6	2.61				
$\text{C}_{34}\text{H}_{34}\text{O}_8$	Nem/Liq	639.7	2.4	3.75	83.32			
	<i>bis</i> (4-butoxyphenyl) 4,4'-biphenylenedicarboxylate							
	Sol/Smec	304.6	1.7	5.58	122.61	[271]		
	Smec/Nem	470.5	54.1	114.98				
$\text{C}_{34}\text{H}_{34}\text{O}_8$	Note: Smec/Nem was not observed by dsc.							
	Nem/Liq	535.5	1.1	2.05	122.61	[271]		
	benzoic acid, 4,4'-[1,4-phenylene <i>bis</i> [(1-oxo-2-propene-3,1-diyl)oxo]] <i>bis</i> , dibutyl ester							
	Sol/Sol	382.9	33.3	86.97				
$\text{C}_{34}\text{H}_{34}\text{O}_8$	Smec/Nem	526.6	1.6	3.04	91.48	[271]		
	Nem/Liq	542.4	0.8	1.47				
$\text{C}_{34}\text{H}_{35}\text{N}_3\text{O}_6$	1,4-benzeneddicarboxylic acid, <i>bis</i> [4-[3-(butoxy)-3-oxo-1-propenyl]phenyl ester							
	Sol/Smec	430.2	45.78	106.42	112.08	[109]		
	Nem/Liq	562.2	3.18	5.66				
	4-(4'-ethoxybenzoyloxy)-2-ethoxy-4'-(4-butyoxysalicylaldimine)azobenzene							
$\text{C}_{34}\text{H}_{36}\text{F}_4\text{O}_4$	Sol/Nem	361.2	36.05	99.80	106.31	[96]		
	Smec/Nem	357.9	1.38	3.86				
	Nem/Liq	369.4	0.98	2.65				
	pentyl 4-[4-((4-heptyloxy-2,3,5,6-tetrafluorophenyl)ethynyl)-benzyloxy]benzoate							
$\text{C}_{34}\text{H}_{36}\text{F}_4\text{O}_4$	Sol/Smec	361.9	29.21	80.71	87.48	[96]		
	Smec/Nem	363.8	1.32	3.63				
	Nem/Liq	366.4	1.15	3.14				
	butyl 4-[4-((4-octyloxy-2,3,5,6-tetrafluorophenyl)ethynyl)benzyloxy]benzoate							
$\text{C}_{34}\text{H}_{36}\text{F}_4\text{O}_4$	Sol/Smec	358.8	35.06	97.71	99.18	[124]		
	Nem/Liq	409.5	0.60	1.47				
$\text{C}_{34}\text{H}_{36}\text{N}_2$	4-[ <i>(S</i> )-2-methylbutoxy]phenyl 4-[ <i>(4</i> -octyloxy-2,3,5,6-tetrafluorophenyl)ethynyl]benzoate							
	Sol/Nem	461.2	18.1	39.25	91.58	[258]		
	Smec/Nem	502.2	24.9	49.58				
	Nem/Liq	568.2	1.56	2.75				
$\text{C}_{34}\text{H}_{36}\text{N}_4\text{O}_6\text{S}_3$	<i>bis</i> [4-(5-n-hexylthio-1,3,4-oxadiazole-2-yl)phenyl]-2,5-thiophenedicarboxylate							
	Sol/Smec	409.2	23.7	57.92	70.95	[23]		
	Smec/Liq	499.2	6.5	13.02				
	$\alpha,\omega$ - <i>bis</i> (4-nitroazobenzene-4'-oxy)decane							
$\text{C}_{34}\text{H}_{36}\text{N}_4\text{O}_6$	Sol/Nem	462.2	57.64	124.71	132.60	[422]		
	Nem/Liq	466.2	3.68	7.89				
$\text{C}_{34}\text{H}_{37}\text{FO}_7\text{S}$	1-(butoxycarbonyl)ethyl 5-[ <i>(3</i> -fluoro-4-heptyloxybenzoylphenyl)ethynyl]thiophene-2-carboxylate							
	Sol/Nem	351.2	40.24	114.58	116.06	[267]		
	Nem/Liq	358.0	0.53	1.48				
$\text{C}_{34}\text{H}_{37}\text{F}_{13}\text{O}_5$	4-(3,3,4,4,5,5,6,6,7,7,8,8,8-tridecafluoroctyloxycarbonyl)phenyl 4-(dodecyloxy)benzoate							
	Sol/Smec	358.4	16.44	45.87	73.62	[125]		
	Smec/Liq	381.3	10.58	27.75				

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta H_{\text{pcc}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pcc}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)						
$\text{C}_{34}\text{H}_{37}\text{NO}_7$		5-[[[4'-(decyloxy)[1,1'-biphenyl]-4-yl]oxy]carbonyl]-1,3-dihydro-1,3-dioxo-2 <i>H</i> -isoindole-2-acetic acid, methyl ester						
	Sol/Sol	336.7	10.3		30.59			
	Sol/Smec	387.2	22.2		57.33			[280]
$\text{C}_{34}\text{H}_{38}\text{N}_2\text{O}_2$	Smec/Liq	428.6	4.92		11.48	99.40		
		2,3-dicyano-4-pentylphenyl 4'-octyl-4-biphenyl-1-carboxylate						
	Sol/Nem	367.7	33.47		91.03			[202]
$\text{C}_{34}\text{H}_{38}\text{N}_2\text{O}_2$	Nem/Liq	382.7	Not reported in paper					
		2,3-dicyano-4-heptylphenyl 4'-hexyl-4-biphenyl-1-carboxylate						
	Sol/Nem	356.7	11.72		32.86			[202]
$\text{C}_{34}\text{H}_{38}\text{N}_4\text{O}_2$	Nem/Liq	381.2	Not reported in paper					
		$\alpha$ -(4'-ethylazobenzene-4-oxy)- $\omega$ -(4'-ethylazobenzene-4-oxy)hexane						
	Sol/Nem	442.2	39.34		88.96			[67]
$\text{C}_{34}\text{H}_{38}\text{N}_4\text{O}_2$	Nem/Liq	468.2	5.33		11.38	100.34		
		$\alpha$ -(4'-butylazobenzene-4-oxy)- $\omega$ -(azobenzene-4-oxy)hexane						
	Sol/Nem	412.2	51.75		125.55			[67]
$\text{C}_{34}\text{H}_{38}\text{N}_4\text{O}_2$	Nem/Liq	427.2	4.26		9.97	135.52		
		$\alpha$ -(4'-propylazobenzene-4-oxy)- $\omega$ -(4'-methylazobenzene-4-oxy)hexane						
	Sol/Nem	424.2	46.91		110.58			[67]
$\text{C}_{34}\text{H}_{38}\text{O}_2$	Nem/Liq	479.2	7.13		14.88	125.46		
		4,4'''-dipentyloxy-p-quaterphenyl						
	Sol/Sol	335.0	9.72		29.01			
$\text{C}_{34}\text{H}_{38}\text{O}_5$	Sol/Sol	493.0	7.89		16.00			
	Sol/Sol	594.0	7.72		13.00			
	Sol/Smec	600.0	5.4		9.00			
$\text{C}_{34}\text{H}_{38}\text{O}_5$	Smec/Liq	653.0	9.14		14.00	81.01	159.8	[111]
		7-[(4'-dodecyloxy)benzoyloxy]isoflavone						
	Smec/Liq	415.2	30.78		74.13			[14]
$\text{C}_{34}\text{H}_{38}\text{O}_6\text{S}_2$		bis(4-hexyloxyphenyl) 2,2'-bithiophene-5,5'-dicarboxylate						
	Sol/Sol	384.3	3.8		9.89			
	Sol/Sol	398.8	6.9		17.30			
	Sol/Smec	413.0	22.5		54.48			
	Smec/Nem	505.8	3.7		7.32			
$\text{C}_{34}\text{H}_{38}\text{O}_8$	Nem/Liq	507.3	2.3		4.53	93.52	192.0	[12]
		bis(4-hexyloxycarbonylphenyl) terephthalate						
	Sol/Smec	423.2	41.0		96.88			
$\text{C}_{34}\text{H}_{39}\text{ClF}_2\text{O}_5$	Smec/Liq	452.2	5.6		12.38	109.26	180.8	[194]
		4-[(2S)-2-chloro-3-methylpentanoyloxy]biphenyl-4'-yl 4-nonyloxy-2,3-difluorobenzoate						
	Smec/Nem	404.5	1.6		3.96			
$\text{C}_{34}\text{H}_{39}\text{ClF}_2\text{O}_5$	Nem/Liq	421.6	0.8		1.90	112.40		[121]
		4-[(2S)-2-chloro-3-methylbutanoyloxy]biphenyl-4'-yl 4-decyloxy-2,3-difluorobenzoate						
	Smec/Nem	423.2	24.67		68.78			
$\text{C}_{34}\text{H}_{39}\text{NO}$	Nem/Liq	437.5	1.25		2.99			
		5-(4'-decyloxybiphenyl-4-yl)-2-(4-methylphenyl)pyridine						
	Sol/Sol	395.0	9.9		25.06			
$\text{C}_{34}\text{H}_{39}\text{NO}$	Smec/Nem	506.0	3.7		7.31			

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.		
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$				
$C_{34}H_{40}N_2O_4$	Smec/Smec	539.0	3.5	6.49	44.32	[185]		
	Smec/Nem	600.0	2.0	3.33				
	Nem/Liq	609.0	1.3	2.13				
$C_{34}H_{40}N_4O_2S$	dihexyl N,N'-[1,4-phenylene- <i>bis</i> (methylidyne)]- <i>bis</i> [aminobenzoate]					[192]		
	Sol/Smec	386.2	27.0	69.91				
	Smec/Smec	421.2	Not reported in paper					
	Smec/Liq	462.2	2.9	6.27				
$C_{34}H_{40}N_4O_2S$	5-(4-decyloxyphenyl)-5-(4-decyloxy)phenylazo-1,3,4-thiadiazole					[79]		
	Sol/Smec	420.0	46.92	111.71				
	Smec/Nem	435.2	2.76	6.34				
	Nem/Liq	446.1	1.45	3.25	121.30			
$C_{34}H_{42}N_2O_2$	2,3-dicyano-4-heptylphenyl 4-[4-hexylbicyclo[2.2.2]oct-1-yl]benzoate					[202]		
	Sol/Nem	395.2	25.10	63.51				
	Nem/Liq	403.2	Not reported in paper					
$C_{34}H_{42}O_2$	1-(4'-dodecylbiphenyl-4-yl)-3-(3-methylphenyl)propane-1,3-dione					[297]		
	Sol/Smec	362.7	34.14	94.13				
	Smec/Liq	367.7	4.02	10.93	105.06			
$C_{34}H_{42}O_3$	1-(4'-dodecylbiphenyl-4-yl)-3-(2-methoxyphenyl)propane-1,3-dione					[297]		
	Sol/Nem	352.7	21.71	61.55				
	Nem/Liq	362.2	0.67	1.85	63.40			
$C_{34}H_{42}O_4$	4,4'-dinonanoyloxydiphenyldiacetylene					[157]		
	Sol/Sol	326.0	19.50	59.82				
	Sol/Nem	400.0	33.50	83.75				
	Nem/Liq	401.0	14.60	36.41	179.98			
$C_{34}H_{42}O_5S$	(R)-4-(3-ethylmercapto-2-methylpropionyl)phenyl 4'-nonyloxybiphenyl-4-carboxylate					[100]		
	Sol/Smec	334.6	30.78	91.99				
	Smec/Smec	339.4	3.88	11.43				
	Smec/Smec	357.1	3.33	9.33				
	Smec/Smec	407.9	0.17	0.42				
	Smec/Liq	430.7	5.85	13.58	126.75			
$C_{34}H_{42}O_6$	4-(2-butoxy-1-methyl-2-oxoethoxy)phenyl [1,1'-biphenyl]-4-carboxylate					[428]		
	Sol/Smec	284.5	0.78	2.74				
	Smec/Smec	329.6	1.93	5.86				
	Smec/Liq	392.5	3.74	9.53	18.13			
$C_{34}H_{42}O_7$	4-(4-dodecyloxybenzoyloxy)phenyl (3,4-dimethoxy)benzoate					[103]		
	Sol/Nem	367.8	11.44	31.10				
	Nem/Liq	406.7	0.51	1.25	32.35			
$C_{34}H_{42}O_7$	4-(4-butoxybenzoyloxy)phenyl (3,4-dipentyloxy)benzoate					[103]		
	Sol/Nem	368.9	12.09	32.77				
	Nem/Liq	390.2	0.37	0.95	33.72			
$C_{34}H_{43}NO_3$	4- <i>tert</i> -butylphenyl 4-(4-decyloxybenzylideneamino)benzoate					[292]		
	Sol/Smec	402.2	31.0	77.08				
	Smec/Liq	421.2	4.1	9.73	86.81			
$C_{34}H_{44}N_2O_3$	4-[5-[(S)-5-methylheptyl]-2-pyrimidinyl]phenyl 4-(8-nonenyloxy)benzoate					[201]		
	Sol/Chol	335.2	25.2	75.18				
	Chol/Liq	415.2	Not reported in paper					
$C_{34}H_{44}N_2O_3$	4-[5-[(S)-6-methyloctyl]-2-pyrimidinyl]phenyl 4-(7-octenyloxy)benzoate					[201]		
	Sol/Smec	323.2	33.3	103.00				
	Smec/Chol	333.2	Not reported in paper					

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated) (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{34}H_{44}N_2O_4$	Chol/Liq	410.2	Not reported in paper			[201]
	Sol/Nem	395.2	44.0	111.34		
	Nem/Liq	431.2	1.9	4.41	115.75	[339]
$C_{34}H_{44}N_2O_4$	Sol/Nem	390.2	61.0	156.33		
	Nem/Liq	445.2	2.0	4.49	160.82	[339]
	Sol/Nem	360.2	37.0	102.72		
$C_{34}H_{44}N_2O_4$	Nem/Liq	450.2	1.4	3.11	105.83	[339]
	Sol/Nem	368.2	55.0	149.38		
	Nem/Liq	446.2	1.3	2.91	152.29	[339]
$C_{34}H_{44}O_3S$	S-4-heptyloxybiphenyl (S)-4-(1-methylheptyloxy)benzoate					
	Sol/Smec	345.3	29.3	84.85		
	Smec/Nem	363.4	2.2	6.05		
$C_{34}H_{44}O_6S$	Nem/Liq	407.9	1.1	2.70	93.60	[281]
	bis(4-octyloxyphenyl) 2,5-thiophenedicarboxylate					
	Sol/Sol	390.6	13.1	33.54		
$C_{34}H_{44}O_7$	Meso/Smec	397.1	44.6	112.31		
	Smec/Nem	408.8	3.3	8.07		
	Nem/Liq	411.4	2.5	6.08	160.00	204.5
$C_{34}H_{44}O_7$	Sol/Smec	383.3	22.5	58.70		
	Smec/Liq	436.4	6.1	13.98	72.68	[286]
	Sol/Smec	364.2	34.85	95.69		
$C_{34}H_{45}NO_2$	Smec/Smec	422.2	1.00	2.37		
	Smec/Smec	444.7	0.29	0.65		
	Smec/Liq	479.2	7.82	16.32	115.03	[323]
$C_{34}H_{45}NO_2$	7-(undec-10-enyloxy)-3-(4'-hexyloxystyryl)quinoline					
	Sol/Smec	330.5	27.93	84.51		
	Smec/Liq	438.6	5.50	12.54	97.05	[139]
$C_{34}H_{45}NO_4$	4'-(2-diethylamino-ethoxy)biphenyl 4-nonyloxybenzoate					
	Sol/Smec	346.8	8.60	24.80		
	Smec/Smec	390.5	2.39	6.12		
$C_{34}H_{45}NO_4$	Smec/Liq	422.6	0.97	2.30	33.22	[427]
	Sol/Smec	333.2	7.77	23.32		
	Smec/Smec	394.5	2.99	7.58		
$C_{34}H_{45}NO_4$	Smec/Liq	424.4	1.18	2.78	33.68	[427]
$C_{34}H_{45}N_5O_3$	2-cyano-3-[4-[1(E)-[4-[4-(1-oxodecyl)-1-piperazinyl]phenyl]azo]phenyl]-2-propenoic acid, butyl ester					
	Sol/Smec	453.9	17.60	38.78		
	Smec/Liq	479.9	3.54	7.38	46.16	[326]
$C_{34}H_{46}N_2O_2S$	2-[4-(2-methyl-2,3-undecadienyl)oxy]phenyl]-5-[4-(octyloxy)phenyl]-1,3,4-thiadiazole					
	Sol/Smec	337.2	29.9	88.67		
	Smec/Nem	371.2	1.9	5.12		
$C_{34}H_{46}N_2O_2S$	Nem/Liq	374.2	1.37	3.66	97.45	[398]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{fus}}S_{tpce}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{fus}}S_{tpce}$ (estimated)	Ref.
		T (K)	$\Delta H_{pce}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{pce}$		
$C_{34}H_{46}N_2O_2S$	Sol/Smec	327.2	30.6	93.52		
	Smec/Nem	352.2	1.3	3.69		
	Nem/Blue	359.2	0.98	2.73		
	Blue/Liq	359.8	Not reported in paper			[398]
$C_{34}H_{46}N_2O_3$	Sol/Chol	351.2	28.6	81.44		
	Chol/Liq	415.2	Not reported in paper			[201]
	Sol/Smec	329.2	33.6	102.07		
$C_{34}H_{46}N_2O_3$	Smec/Chol	345.2	Not reported in paper			
	Chol/Liq	415.2	Not reported in paper			[201]
	Sol/Disc	362.2	9.0	24.85		
$C_{34}H_{46}N_2O_6$	Disc/Liq	397.2	19.0	47.84	72.69	
	Sol/Chol	367.7	22.80	62.01		
$C_{34}H_{46}O_2$	Chol/Liq	376.9	Not reported in paper			[382]
	Note: Authors report only a single transition enthalpy, which we have assumed is for the Sol/Chol transition.					
	Sol/Smec	377.5	15.73	41.67		
	Smec/Nem	408.5	14.02	34.32		
$C_{34}H_{46}O_4$	Nem/Liq	603.2	Not reported in paper			[348]
	Sol/Smec	359.1	11.63	32.39		
	Smec/Nem	402.7	15.86	39.38		
	Nem/Liq	603.2	Not reported in paper			[348]
$C_{34}H_{47}ClO_5$	Sol/Smec	325.2	33.89	104.21		
	Smec/Smec	338.7	0.42	1.24		
	Smec/Liq	354.2	4.18	11.80	117.25	
	Sol/Smec	326.7	43.93	134.47		
$C_{34}H_{47}ClO_5$	Smec/Smec	342.2	0.21	0.61		
	Smec/Liq	361.7	5.86	16.20	151.28	
	Sol/Smec	346.6	50.10	144.55		
$C_{34}H_{48}N_2O_2$	Smec/Nem	386.6	0.13	0.34		
	Nem/Liq	405.7	1.77	4.36	149.25	
	Sol/Smec	373.3	48.6	130.19		
$C_{34}H_{48}N_2O_3$	Smec/Liq	389.9	6.5	16.67	146.85	
	Sol/Smec	384.4	16.04	41.73		
$C_{34}H_{48}N_4O_3$	Smec/Liq	405.4	11.38	28.07	69.80	
	Sol/Smec	395.3	34.8	88.03		
$C_{34}H_{48}O_4$	Smec/Liq	421.4	7.9	18.75	106.78	
	$di(4'-heptylphenyl)-trans-cyclohexane-1,4-dicarboxylate$					

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{34}H_{48}O_6$	Sol/Smec	342.4	30.0	87.62		
	Smec/Smec	384.0	2.67	6.95		
	Smec/Liq	414.1	5.45	13.16	107.73	175.9 [215]
$C_{34}H_{48}BrO_2$	di(4'-heptyloxyphenyl)- <i>trans</i> -cyclohexane-1,4-dicarboxylate					
	Sol/Smec	369.2	46.2	125.14		
	Smec/Smec	380.2	1.09	2.87		
	Smec/Smec	384.2	Not reported in paper			[220]
$C_{34}H_{49}NO_5$	Smec/Liq	453.2	5.70	12.58	189.4	
	4-bromophenyl cholest-5-ene-3 $\beta$ -carboxylate					
	Sol/Nem	382.4	23.31	60.96		
$C_{34}H_{49}IO_2$	Nem/Liq	517.6	0.78	1.51	62.47	[369]
	4-iodophenyl cholest-5-ene-3 $\beta$ -carboxylate					
	Sol/Nem	401.0	10.02	24.99		
$C_{34}H_{49}NO_5$	Nem/Liq	526.3	0.92	1.75	26.74	[369]
	6-[4-(4-undecyloxyphenyliminomethyl)-3-hydroxyphenoxy]hexyl methacrylate					
	Sol/Smec	331.3	41.99	126.74		
$C_{34}H_{50}$	Smec/Smec	347.8	Not reported in paper			
	Smec/Smec	365.2	7.27	19.91		[294]
	<i>trans, trans</i> -4,4'-bis[2-(4-propylcyclohexyl)ethyl]-1,1'-biphenyl					
$C_{34}H_{50}N_2O_2S$	Sol/Smec	339.2	8.87	26.15		
	Smec/Smec	395.2	4.77	12.07		
	Smec/Smec	423.7	3.68	8.69		
	Smec/Smec	447.2	5.02	11.23		
	Smec/Nem	470.7	0.46	0.98		
	Nem/Liq	504.2	4.81	9.54	68.66	[348]
$C_{34}H_{50}N_2O_2S$	6-n-decyloxy-2-(4-decyloxybenzylidenamino)benzothiazole					
	Sol/Smec	354.2	40.0	112.93		
	Smec/Nem	394.8	Not given in paper			
	Nem/Liq	395.9	8.1	20.46		[41]
$C_{34}H_{50}N_2O_2S_2$	2,5-bis(4-decyloxyphenyl)thiazolo[5,4-d]dithiazole					
	Sol/Sol	389.0	35.62	91.57		
	Sol/Smec	413.8	44.68	107.97		
	Smec/Liq	500.0	11.88	23.76	223.30	[269]
$C_{34}H_{50}N_2O_3$	4-pentanoyl-4'-heptadecanoyloxyazobenzene					
	Sol/Sol	371.5	9.22	24.82		
	Sol/Smec	377.4	46.50	123.21		
	Smec/Liq	397.0	9.98	25.14	173.17	[157]
$C_{34}H_{50}N_2O_3S$	N-[2-(6-decyloxybenzothiazolyl)]-4-decyloxybenzamide					
	Sol/Sol	373.8	24.4	65.28		
	Sol/Smec	389.2	18.3	47.02		
	Smec/Nem	397.7	0.8	2.01		
	Nem/Liq	400.7	2.1	5.24	119.55	[41]
$C_{34}H_{50}N_4OS$	2-(4-decanoxyphenylazo)-5-(4-decyloxy)phenyl-1,3,4-thiadiazole					
	Sol/Smec	420.0	46.92	111.71		
	Smec/Nem	435.2	2.76	6.34		
	Nem/Liq	446.1	1.45	3.25	121.30	[396]
$C_{34}H_{50}O_2$	cholesterol benzoate					
	Sol/Chol	422.3	29.16	69.05		
	Chol/Liq	450.0	0.51	1.13	70.18	109.2 [166]

Independent values from another reference

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$\text{C}_{34}\text{H}_{50}\text{O}_2$	Sol/Chol	418.6	32.72	78.16	109.2	[155, 310]
	Chol/Liq	453.9	0.71	1.57		
			phenyl cholest-5-ene-3 $\beta$ -carboxylate			
$\text{C}_{34}\text{H}_{50}\text{O}_3$	Sol/Nem	379.9	16.37	43.09	44.23	[369]
	Nem/Liq	448.4	0.51	1.14		
			4-[2-( <i>trans</i> -4-pentylcyclohexyl)ethyl]phenyl 4-[(S)-1-methylheptyl]oxybenzoate			
$\text{C}_{34}\text{H}_{50}\text{O}_3$	Sol/Chol	301.2	25.44	84.46	Not reported in paper	[208]
	Chol/Liq	351.2				
$\text{C}_{34}\text{H}_{50}\text{O}_3$	Sol/SmeC	341.2	32.83	96.22	Not reported in paper	[199, 207]
	SmeC/Nem	360.2				
	Nem/Liq	431.2				
$\text{C}_{34}\text{H}_{50}\text{O}_4$	Sol/SmeC	325.2	34.9	107.32	Not reported in paper	[200]
	SmeC/SmeC	344.2				
	SmeC/Liq	348.2				
$\text{C}_{34}\text{H}_{50}\text{O}_4$	Sol/SmeC	348.2	26.32	75.59	Not reported in paper	[207]
	SmeC/Nem	383.2				
	Nem/Liq	429.2				
$\text{C}_{34}\text{H}_{50}\text{O}_4$	Sol/SmeC	346.2	23.63	68.26	Not reported in paper	[207]
	SmeC/Chol	381.2				
	Chol/Liq	419.2				
$\text{C}_{34}\text{H}_{50}\text{O}_4$	Sol/SmeC	343.2	34.07	99.27	Not reported in paper	[208]
	SmeC/SmeC	350.2				
	SmeC/Chol	366.2				
$\text{C}_{34}\text{H}_{51}\text{BrN}_2\text{O}_4$	Sol/SmeC	332.6	38.34	115.27	Not reported in paper	[47]
	SmeC/Liq	345.7	4.36	12.61		
$\text{C}_{34}\text{H}_{51}\text{ClN}_2\text{O}_4$	Sol/SmeC	336.0	33.35	99.26	Not reported in paper	[47]
	SmeC/SmeC	346.0	0.12	0.35		
	SmeC/Liq	351.3	4.76	13.55		
$\text{C}_{34}\text{H}_{51}\text{NO}$	Sol/SmeC	327.8	57.57	175.63	Not reported in paper	[147]
	SmeC/Nem	340.9	0.64	1.88		
	Nem/Liq	342.2	1.87	5.46		
$\text{C}_{34}\text{H}_{52}\text{N}_2\text{O}_2$	Sol/Chol	377.2	18.7	49.58	Not reported in paper	[201]
	Chol/Liq	414.2				
$\text{C}_{34}\text{H}_{52}\text{N}_2\text{O}_2$	Sol/Sol	395.2	24.3	61.49	146.88	[434]
	Sol/Sol	400.3	4.9	12.24		
	Sol/Cube	414.4	23.1	55.74		
	Cube/SmeC	426.2	0.4	0.94		
	SmeC/Liq	437.1	7.2	16.47		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{34}H_{53}NO$		4-hexyl-N-[[4-(pentadecyloxy)phenyl]methylene]benzenamine				
	Sol/SmeC	336.4	35.75	106.27		
	SmeC/SmeC	358.6	6.53	18.21		
$C_{34}H_{53}NO$	SmeC/Liq	360.7	8.90	24.67	149.15	[240]
		4-propyl-N-[[4-(octadecyloxy)phenyl]methylene]benzenamine				
	Sol/SmeC	349.7	55.37	158.34		
$C_{34}H_{54}$	SmeC/SmeC	351.7	5.18	14.73		
	SmeC/Liq	355.3	10.06	28.31	201.38	[240]
		1-ethyl-4-[2-[4-[2-(4-hexylbicyclo[2.2.2]oct-1-yl)ethyl]phenyl]ethyl]bicyclo[2.2.2]octane				
$C_{34}H_{54}$	Sol/SmeC	335.9	10.25	30.52		
	SmeC/SmeC	391.1	5.52	14.11		
	SmeC/Nem	428.2	7.61	17.77		
	Nem/Liq	452.9	1.21	2.67	65.07	[348]
$C_{34}H_{54}N_2$		5-nonyl-2-[4-[4-(trans-4-pentylcyclohexyl)butyl]phenyl]pyrimidine				
	Sol/SmeC	347.2	15.9	45.79		
	SmeC/SmeC	355.2	Not reported in paper			
	SmeC/Nem	370.2	Not reported in paper			
$C_{34}H_{54}N_2O$	Nem/Liq	386.2	Not reported in paper			[198]
		4-butyl-4'-octadecyloxyazobenzene				
	Sol/SmeC	342.8	59.56	173.75		
$C_{34}H_{54}N_2O$	SmeC/Liq	343.5	14.08	40.99	214.74	205.4
		5-decyl-2-[4-[3-[ <i>trans</i> -4-pentylcyclohexyl]oxy]propyl]phenyl]pyrimidine				
	Sol/SmeC	342.2	32.2	94.10		
$C_{34}H_{54}N_2O$	SmeC/Nem	383.2	Not reported in paper			
	Nem/Liq	403.2	Not reported in paper			[198]
$C_{34}H_{54}N_2O_3$		4,4'-diundecyloxyazoxybenzene				
	Sol/SmeC	354.0	41.09	116.07		
	SmeC/Nem	394.6	10.07	25.52	141.59	220.8
$C_{34}H_{56}O_6$		(S)-4-(2-ethylheptyloxycarbonyl)phenyl 4-tridecanoyloxytolane-4'-carboxylate				
	Sol/SmeC	353.3	49.6	140.39		
	SmeC/SmeC	356.2	0.68	1.91		
	SmeC/Liq	364.9	0.94	2.58	144.88	[173]
$C_{34}H_{57}N_3O_6$		4,4'-bis[4-(4-heptyloxybenzoyloxy)benzylideneamino]diphenylamine				
	Sol/SmeC	494.2	41.0	82.96		
	SmeC/Nem	496.2	0.6	1.21		
	Nem/Liq	593.2	1.8	3.03	87.20	[284]
$C_{34}H_{58}OS$		cholesteryl thioheptanoate				
	Sol/Liq	380.2	32.6	85.6	85.6	NA
	SmeC/Chol	346.3	0.1	0.3		
	Chol/Liq	375.3	0.3	0.9		[155,312]
Note: Liquid crystalline behavior observed on cooling.						
$C_{34}H_{60}N_2O_2$		N,N'-didodecanoyl-2,3,5,6-tetramethylbenzene-1,4-diamine				
	Sol/Meso	417.2	40	95.88		
	Meso/Meso	510.2	11	21.56		
	Meso/Liq	536.2	16	29.84	147.28	[36]
$C_{34}H_{60}N_2O_2$		N,N'-didodecanoyl-3,4,5,6-tetramethylbenzene-1,2-diamine				
	Sol/Meso	401.2	18	44.87		
	Meso/Meso	460.2	1	2.17		
	Meso/Liq	505.2	22	43.55	90.59	[61]
$C_{34}H_{61}NO_4$		3,4,5-tris(nonyloxy)benzamide				

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.			
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$					
$C_{35}H_{33}NO_2$	Sol/Meso	354.2	37.9	107.00	125.27	[378]			
	Meso/Liq	361.2	6.6	18.27					
$C_{35}H_{34}N_2O_2$	[1-(4-heptylbiphenyl)-3-(4"-cyanobiphenyl)]-propane-1,3-dione					[254]			
	Sol/Nem	459.7	27.15	59.06	59.34				
	Nem/Liq	469.2	0.13	0.28					
$C_{35}H_{34}N_2O_2$	$\alpha,\omega$ -bis[(4,4'-cyanobiphenyl)oxy]nonane					[127]			
	Sol/Nem	406.8	47.0	115.54	123.96				
	Nem/Liq	445.5	3.75	8.42					
$C_{35}H_{35}F_7O_6$	4-[(1-methylheptyloxy)carbonyl]phenyl 4'-(3-(2,2,3,3,4,4,4-heptafluoro-1-oxobutoxy)propyl)[1,1'-biphenyl]-4-carboxylate					[394]			
	Sol/SmeC	359.2	25.36	70.60	84.59				
	SmeC/SmeC	395.8	0.11	0.28					
	SmeC/SmeC	398.3	1.50	3.77					
	SmeC/Liq	403.6	4.01	9.94					
$C_{35}H_{35}F_7O_7$	(S)-4-(1-methylheptyloxycarbonyl)phenyl 4'-(3-perfluoro-butanoyloxyprop-1-oxy)biphenyl-4-carboxylate					[390]			
	Sol/SmeC	356.7	18.49	51.84	66.32				
	SmeC/SmeC	394.2	1.26	3.20					
	SmeC/SmeC	397.1	1.34	3.37					
	SmeC/Liq	402.2	3.18	7.91					
$C_{35}H_{36}F_4O_5$	4-[(S)-2-methylbutoxycarbonyl]phenyl 4-[(4-octyloxy-2,3,4,5-tetrafluorophenyl)ethynyl]benzoate					[91]			
	Sol/SmeC	348.6	25.49	73.16	75.98				
	SmeC/SmeC	350.9	Not detected by dsc						
	SmeC/Chst	402.6	0.70	1.74					
	Chst/Liq	418.0	0.45	1.08					
$C_{35}H_{36}N_4O_6$	$\alpha,\omega$ -bis(4-ethoxyazobenzene-4'-carbonyloxy)pentane					[107]			
	Sol/SmeC	377.1	18.8	49.85	81.30				
	SmeC/Liq	397.5	12.5	31.45					
$C_{35}H_{37}NO_9$	5-[[4-[(4-decyloxy)benzoyl]oxy]phenoxy]carbonyl]-1,3-dihydro-1,3-dioxo-2 <i>H</i> -isoindole-2-acetic acid, methyl ester					[280]			
	Sol/Nem	404.1	33.9	83.89	87.20				
	Nem/Liq	435.3	1.44	3.31					
$C_{35}H_{38}F_4O_4$	pentyl 4-[4-((4-octyloxy-2,3,5,6-tetrafluorophenyl)ethynyl)benzyloxy]benzoate					[96]			
	Sol/SmeC	357.5	48.62	136.00	146.24				
	SmeC/Nem	369.8	1.99	5.38					
	Nem/Liq	370.5	1.8	4.86					
$C_{35}H_{38}F_4O_4$	4-[(S)-2-methylbutoxy]phenyl 4-[(4-nonyloxy-2,3,5,6-tetrafluorophenyl)ethynyl]benzoate					[124]			
	Sol/Nem	355.4	26.65	74.99	76.52				
	Nem/Liq	404.5	0.62	1.53					
$C_{35}H_{38}N_2O_7$	5-[[4-[(E)-[[4-(decyloxy)phenyl]methylene]amino]phenoxy]carbonyl]-1,3-dihydro-1,3-dioxo-2 <i>H</i> -isoindole-2-acetic acid, methyl ester					[280]			
	Sol/Nem	403.6	33.8	83.75	85.50				
	Nem/Liq	416.4	0.73	1.75					
$C_{35}H_{38}N_6O_6$	$\alpha,\omega$ -bis(4-nitroazobenzene-4'-oxy)undecane					[422]			
	Sol/Nem	421.2	64.43	152.97	158.37				
	Nem/Liq	450.2	2.43	5.40					
$C_{35}H_{39}FO_7S$	1-(butoxycarbonyl)ethyl 5-[(3-fluoro-4-octyloxybenzoylphenyl)ethynyl]thiophene-2-carboxylate					[267]			
	Sol/SmeC	340.9	31.33	91.90	100.3				
	SmeC/Nem	354.7	2.33	6.57					
	Nem/Liq	359.8	0.66	1.83					

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{35}H_{39}NO_7$		5-[[[4'-(decyloxy)[1,1'-biphenyl]-4-yl]oxy]carbonyl]-1,3-dihydro-1,3-dioxo-2H-isoindole-2-propanoic acid, methyl ester				
	Sol/Smec	440.5	10.9	24.74		
	Smec/Liq	458.4	7.55	16.47	41.21	[280]
$C_{35}H_{40}N_4O_2$		$\alpha$ -(4'-pentylazobenzene-4-oxy)- $\omega$ -(azobenzene-4-oxy)hexane				
	Sol/Nem	413.2	52.56	127.20		
	Nem/Liq	431.2	4.84	11.22	138.42	[67]
$C_{35}H_{40}N_4O_2$		$\alpha$ -(4'-butylazobenzene-4-oxy)- $\omega$ -(4'-methylazobenzene-4-oxy)hexane				
	Sol/Nem	403.2	48.94	121.38		
	Nem/Liq	470.2	6.23	13.25	144.63	[67]
$C_{35}H_{40}N_4O_2$		$\alpha$ -(4'-propylazobenzene-4-oxy)- $\omega$ -(4'-ethylazobenzene-4-oxy)hexane				
	Sol/Nem	431.2	35.49	82.31		
	Nem/Liq	473.2	6.33	13.38	95.69	[67]
$C_{35}H_{41}ClF_2O_5$		4-[(2S)-2-chloro-3-methylpentanoyloxy]biphenyl-4'-yl 4-decyloxy-2,3-difluorobenzoate				
	Sol/Smec	348.5	28.88	82.87		
	Smec/Nem	405.7	1.28	3.16		
	Nem/Blue	420.1	0.9	2.14		
$C_{35}H_{41}ClF_2O_5$	Blue/Liq	420.8	Not detected by dsc	88.17		[121]
		4-[(2S)-2-chloro-3-methylbutanoyloxy]biphenyl-4'-yl 4-undecyloxy-2,3-difluorobenzoate				
	Sol/Smec	354.9	16.01	45.11		
	Smec/Nem	419.9	1.15	2.74		
$C_{35}H_{41}NO_5$	Nem/Liq	434.2	0.88	2.03	49.88	[121]
		2-butyl-2,3-dihydro-1,3-dioxo-1 <i>H</i> -isoindole-5-carboxylic acid, 4'-(decyloxy)[1,1'-biphenyl]-4-yl ester				
	Sol/Smec	404.6	26.8	66.24		
	Smec/Smec	451.4	0.72	1.60		
$C_{35}H_{42}N_2O_7$	Smec/Liq	459.9	5.07	11.02	78.86	[280]
		4-butoxy-2'-(4-methacryloyloxybutoxy)-4'-(4-butoxybenzoyloxy)azobenzene				
	Sol/Nem	356.4	50.21	140.88		
	Nem/Liq	358.7	0.64	1.78	142.66	[62]
$C_{35}H_{42}N_2O_7$		4-[[[(1S)-1-methylheptyl]oxy]carbonyl]phenyl 4'-[3-(1-oxobutoxy)propoxy][1,1'-biphenyl]-4-carboxylate				
	Sol/Smec	340.4	22.01	64.66		
	Smec/Smec	365.7	0.19	0.52		
$C_{35}H_{42}O_7$	Smec/Liq	390.1	5.02	12.87	78.05	[391]
		(1S)-2-butoxy-1-methyl-2-oxoethyl 4-[[2(E)-3-[6-octyloxy)-2-naphthalenyl]-1-oxo-2-propenyl]oxy]benzoate				
	Sol/Meso	344.6	35.54	103.13		
$C_{35}H_{43}ClO_4$	Meso/Liq	390.9	3.59	9.18	112.31	[401]
		4-octylphenyl 2-chloro[(4-heptylbenzoyl)oxy]benzoate				
	Sol/Nem	312.9	30.4	97.16		
$C_{35}H_{43}ClO_6$	Nem/Liq	376.2	1.03	2.74	99.90	[365]
		[S]-4'-decyloxybiphenyl 4-(2-chloro-3-methylbutanoyloxy)-3-methylbenzoate				
	Sol/Smec	340.2	23.99	70.52		
	Smec/Smec	384.0	0.18	0.47		
$C_{35}H_{43}NO_6$	Smec/Nem	407.0	0.71	1.74		
	Nem/Liq	412.3	0.83	2.01	74.74	[45]
		ethoxyethyl 4-(4'-decyloxybenzoyloxybenzylidene)-4"-aminobenzoate				
$C_{35}H_{44}ClNO_4$	Sol/Smec	365.7	9.45	25.84		
	Smec/Liq	431.2	0.94	2.18	28.02	[58]
$C_{35}H_{44}ClNO_4$		3-hydroxy-4-[(E)-[(4-nonylphenyl)imino]methyl]phenyl 4-butoxy- $\alpha$ -chlorobenzenepropanoate				
	Sol/Smec	332.2	Not reported in paper			

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.	
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$			
$C_{35}H_{44}N_2O_2$	Smec/Smec	332.7	3.27	9.82		[249]	
	Smec/Smec	361.6	0.29	0.80			
	Smec/Liq	387.5	7.56	19.51			
$C_{35}H_{44}O_2$	2,3-dicyano-4-heptylphenyl 4-[4-heptylbicyclo[2.2.2]oct-1-yl]benzoate					[202]	
	Sol/Nem	398.2	30.33	76.17			
	Nem/Liq	407.2	Not reported in paper				
$C_{35}H_{44}O_3$	1-(4'-dodecylbiphenyl-4-yl)-3-(2-ethoxyphenyl)propane-1,3-dione					[297]	
	Sol/Nem	343.7	27.02	78.62			
	Nem/Liq	366.7	0.33	0.90	79.52		
$C_{35}H_{44}O_5S$	(R)-4-(3-ethylmercapto-2-methylpropionyl)phenyl 4'-decyloxybiphenyl-4-carboxylate					[100]	
	Smec/Smec	338.1	26.39	78.05			
	Smec/Smec	355.4	2.44	6.87			
	Smec/Smec	410.6	0.07	0.17			
	Smec/Smec	429.4	4.08	9.50	94.59		
$C_{35}H_{44}O_6$	4-[(1R)-2-(1-ethylpropoxy)-1-methyl-2-oxoethoxy]phenyl 4'-(octyloxy)[1,1'-biphenyl]-4-carboxylate					[239]	
	Sol/Smec	302.3	16.04	53.06			
	Smec/Meso	385.9	1.46	3.78			
	Meso/Nem	387.5	Too small to be measured				
	Nem/Meso	393.7	Too small to be measured				
$C_{35}H_{45}NO_3$	4-(4'-dodecyloxybenzoyloxy)benzylidene 4"-isopropylaniline					[273]	
	Sol/Smec	383.2	12.81	33.43			
	Nem/Liq	435.2	0.65	1.49	37.50		
$C_{35}H_{46}N_2O_3$	4-[5-[(S)-6-methyloctyl]-2-pyrimidinyl]phenyl 4-(8-nonyloxy)benzoate					[201]	
	Sol/Smec	326.2	27.2	83.38			
	Smec/Chol	340.2	Not reported in paper				
	Chol/Liq	412.2	Not reported in paper				
$C_{35}H_{46}N_2O_4$	4-[(4-isobutyloxyphenyl)azo]phenyl 4-dodecyloxybenzoate					[319]	
	Sol/Smec	374.9	27.96	74.58			
	Smec/Nem	385.2	0.30	0.78			
	Nem/Liq	456.1	1.11	2.43	77.79		
$C_{35}H_{46}N_2O_4$	4-[(4-ethoxyphenyl)azo]phenyl 4-(tetradecyloxy)benzoate					[339]	
	Sol/Nem	371.2	54.0	145.47			
	Nem/Liq	482.2	1.1	2.28	147.45		
$C_{35}H_{46}N_2O_4$	4-[(4-ethoxyphenyl)azo]-2,3,5,6-tetramethylphenyl 4-(decyloxy)benzoate					[339]	
	Sol/Nem	425.2	53.0	124.65			
	Nem/Liq	458.2	1.7	3.71	128.36		
$C_{35}H_{46}N_2O_4$	4-[(4-ethoxyphenyl)azo]-2,3-dimethylphenyl 4-(dodecyloxy)benzoate					[339]	
	Sol/Nem	383.2	37.0	96.56			
	Nem/Liq	468.2	1.3	2.78	99.34		
$C_{35}H_{46}N_2O_4$	4-[(4-ethoxyphenyl)azo]-3,6-dimethylphenyl 4-(dodecyloxy)benzoate					[339]	
	Sol/Nem	375.2	63.0	167.91			
	Nem/Liq	405.2	1.8	4.44	172.35		
$C_{35}H_{46}N_2O_4$	4-[(4-ethoxyphenyl)azo]-3,5-dimethylphenyl 4-(dodecyloxy)benzoate					[339]	
	Sol/Nem	362.2	59.0	162.89			
	Nem/Liq	421.2	1.8	4.27	167.16		
$C_{35}H_{46}N_2O_4$	4-[(4-ethoxyphenyl)azo]-2,6-dimethylphenyl 4-(dodecyloxy)benzoate					[339]	
	Sol/Nem	361.2	52.0	143.96			

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta S_{\text{pce}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )				
$C_{35}H_{46}O_3S$	Nem/Liq	408.2	2.0	4.90	148.86		[339]
	Sol/Smec	339.0	23.8	70.21			
	Smec/Nem	371.9	2.3	6.18			
	Nem/Liq	406.2	1.5	3.69	80.08		[281]
$C_{35}H_{46}O_7$	4-nonyloxyphenyl 7-decanoxyloxychromone-2-carboxylate						
	Sol/Smec	377.9	23.3	61.66			
	Smec/Liq	436.5	6.3	14.43	76.09		[286]
$C_{35}H_{47}NO_2$	N-(2-hydroxy-4-butoxybenzylidene)-4"-dodecylphenylaniline						
	Sol/Smec	360.7	38.53	106.82			
	Smec/Smec	421.2	0.88	2.09			
	Smec/Smec	462.2	Too small to be measured				
$C_{35}H_{47}NO_4$	Smec/Liq	495.7	8.12	16.39	125.30		[323]
	4'-(2-diethylamino-ethoxy)biphenyl 4-decyloxybenzoate						
	Sol/Smec	356.7	7.87	22.06			
	Smec/Smec	392.8	1.72	4.38			
$C_{35}H_{47}NO_4$	Smec/Liq	415.8	0.87	2.09	28.53		[427]
	4'-(3-diethylamino-propoxy)biphenyl 4-nonyloxybenzoate						
	Sol/Smec	336.5	9.15	27.19			
	Smec/Smec	395.7	2.66	6.72			
$C_{35}H_{48}N_2O_2S$	Smec/Liq	418.5	1.04	2.49	36.40		[427]
	2-[4-(2-ethyl-2,3-undecadienyl)oxy]phenyl]-5-[4-(octyloxy)phenyl]-1,3,4-thiadiazole						
	Sol/Smec	323.2	27.9	86.32			
	Smec/Nem	365.2	2.24	6.13			
$C_{35}H_{48}N_2O_2S$	Nem/Liq	368.2	1.68	4.56	97.01		[398]
	2-[4-(4-ethyl-2,3-undecadienyl)oxy]phenyl]-5-[4-(octyloxy)phenyl]-1,3,4-thiadiazole						
	Sol/Smec	318.2	25.5	80.14			
	Smec/Smec	347.2	0.10	0.29			
$C_{35}H_{48}N_2O_3$	Smec/Nem	354.2	1.15	3.25			
	Nem/Liq	361.2	0.8	2.21	85.89		[398]
	4-[5-[(S)-6-methyloctyl]-2-pyrimidinyl]phenyl 4-nonyloxybenzoate						
	Sol/Smec	341.2	34.9	102.29			
$C_{35}H_{48}O_4$	Smec/Chol	353.2	Not reported in paper				
	Chol/Liq	411.2	Not reported in paper				[201]
	4-[[4-butylcyclohexyl]carbonyl]oxy]phenyl 4-(4-pentylcyclohexyl)benzoate						
$C_{35}H_{48}O_4$	Sol/Smec	384.5	13.51	35.14			
	Smec/Nem	404.7	11.55	28.54			
	Nem/Liq	601.2	Not reported in paper				[348]
$C_{35}H_{49}NO_2$	3-[4-( <i>trans</i> -4-heptylcyclohexyl)phenyl]-5-[4-(heptyloxy)phenyl]isoxazole						
	Sol/Smec	385.5	26.02	67.50			
	Smec/Nem	498.5	1.20	2.41			
$C_{35}H_{49}N_3O_3S$	Nem/Liq	507.2	1.12	2.21	72.12		[131]
	5-(4-decyloxy)phenyl-2-(2-hydroxy-4-nonyloxy)benzylideneamino-1,3,4-thiadiazole						
	Sol/Smec	374.2	26.04	69.59			
$C_{35}H_{50}N_2O_2$	Smec/Liq	486.2	9.42	19.37	85.96		[59]
	6-n-decyloxy-2-[(4'-N-nonyloxyphenylimino)methyl]quinoline						
	Sol/Smec	373.6	61.38	164.29			
$C_{35}H_{50}N_2O_2$	Smec/Nem	389.8	0.20	0.51			
	Nem/Liq	404.3	1.84	4.55	169.35		[112]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{35}H_{50}N_2O_3$	Sol/Smec	371.2	48.7	131.20		[43]
	Smec/Liq	389.7	6.8	17.45	148.65	
$C_{35}H_{50}O_4$	Sol/Smec	392.8	34.9	88.85		[44]
	Smec/Liq	421.3	9.7	23.02	111.87	
$C_{35}H_{51}NO_5$	Sol/Smec	319.3	33.28	104.23		[294]
	Smec/Smec	361.1	Not reported in paper			
	Smec/Liq	369.8	7.12	19.25		
$C_{35}H_{51}N_3O_2S$	N-[4-(4-dodecyloxyphenyl)phenyl]methylene-5-[4-(octyloxy)phenyl]-1,2,4-thiadiazol-2-amine					[283]
	Sol/Sol	369.2	11.5	31.15		
	Sol/Smec	391.5	32.76	83.68		
	Smec/Smec	462.9	9.81	21.19		
$C_{35}H_{52}N_2O_2$	Smec/Liq	466.9	Not reported in paper			[16]
	3,5-di(4-n-decyloxyphenyl)pyrazole					
	Sol/Sol	352.5	31.8	90.21		
	Smec/Smec	403.5	13.1	32.47		
$C_{35}H_{52}N_2O_3$	Smec/Liq	453.7	0.3	0.66		[157]
	Sol/Sol	456.1	6.1	13.37	136.71	
	Smec/Liq					
$C_{35}H_{52}N_2O_3$	4-pentanoyl-4'-octanodecanoxyloxyazobenzene					[251]
	Sol/Smec	339.6	5.31	15.64		
	Smec/Liq	367.5	8.27	22.50		
$C_{35}H_{52}O_3$	Smec/Liq	379.3	51.83	136.65	174.79	
	Sol/Smec					[199, 208]
	Smec/Chol					
	Chol/Liq					
$C_{35}H_{52}O_3$	Sol/Smec	335.3	8.32	24.81		[199, 208]
	Smec/Smec	343.2	Not reported in paper			
	Smec/Chol	348.2	Not reported in paper			
	Chol/Liq	409.2	Not reported in paper			
$C_{35}H_{52}O_3$	Sol/Smec	331.2	19.95	60.24		[199, 208]
	Smec/Smec	332.2	Not reported in paper			
	Smec/Chol	366.2	Not reported in paper			
	Chol/Liq	413.2	Not reported in paper			
$C_{35}H_{52}O_3$	Sol/Smec	341.2	34.21	100.26		[199, 207]
	Smec/Nem	374.2	Not reported in paper			
	Nem/Liq	427.2	Not reported in paper			
	Sol/Chol					
$C_{35}H_{52}O_3$	Chol/Liq	452.4	35.31	78.07		[155, 310]
		5.38.1	1.21	2.26	80.32	
$C_{35}H_{52}O_4$	Sol/Smec	345.2	25.40	73.58		[207]
	Smec/Smec	351.2	Not reported in paper			
	Smec/Nem	387.2	Not reported in paper			
	Nem/Liq	427.2	Not reported in paper			
$C_{35}H_{53}NO_5$	4'-docosanyloxy-3'-nitrobiphenyl-4-carboxylic acid					

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.		
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$				
$C_{35}H_{54}N_2O_2$	Sol/Smec	370.4	29.5	79.64	83.70	[133]		
	Smec/Cube	406.8	0.6	1.47				
	Cube/Liq	464.0	1.2	2.59				
Independent set of measurements								
$C_{35}H_{54}N_2O_2$	Sol/Sol	375.7	37.62	100.13	210.37	[337]		
	Smec/Smec	377.0	36.34	96.39				
	Smec/Cube	407.9	1.98	4.85				
	Cube/Cube	454.1	0.17	0.37				
	Cube/Liq	466.8	4.03	8.63				
Note: The authors observed a liquid phase anomaly in the dsc curve at 475 K, with a corresponding enthalpy of 5.6 kJ·mol <sup>-1</sup> .								
$C_{35}H_{54}N_2O_2$	4-[5-[(S)-6-methyloctyl]-2-pyrimidinyl]phenyl <i>trans</i> -4-nonylcyclohexane-1-carboxylate							
	Smec/Smec	355.2	19.3	54.34	[201]	[201]		
	Smec/Smec	360.2	Not reported in paper					
	Smec/Chol	365.2	Not reported in paper					
	Chol/Liq	409.2	Not reported in paper					
$C_{35}H_{55}NO$	4-heptyl-N-[[4-(pentadecyloxy)phenyl]methylene]benzenamine							
	Smec/Smec	333.1	30.30	90.96	217.17	[240]		
	Smec/Liq	364.0	45.94	126.21				
$C_{35}H_{55}NO$	4-butyl-N-[[4-(octadecyloxy)phenyl]methylene]benzenamine							
	Smec/Smec	346.9	56.10	161.70	190.74	[240]		
	Smec/Smec	349.7	4.35	12.44				
	Smec/Liq	352.4	5.85	16.60				
$C_{35}H_{56}N_2O$	4-heptyl-4'-hexadecyloxyazobenzene							
	Smec/Smec	288.9	29.37	101.66	178.76	[390]		
	Smec/Smec	290.7	0.47	1.62				
	Smec/Smec	309.0	22.06	71.39				
	Smec/Liq	337.0	1.38	4.09				
$C_{35}H_{56}N_2O$	4-pentyl-4'-octadecyloxyazobenzene							
	Smec/Smec	339.3	37.66	110.99	258.7	[141]		
	Smec/Liq	353.8	15.66	44.26				
$C_{35}H_{58}N_2O_6$	4,4'-bis[4-(4-heptyloxybenzylideneamino)benzoyloxy]diphenylmethane							
	Smec/Smec	429.2	33.0	76.89	112.66	[284]		
	Smec/Liq	475.2	17.0	35.77				
$C_{35}H_{58}OS$	cholesteryl thiooctanoate							
	Smec/Chol	370.2	29.0	78.3	NA	[155,312]		
	Smec/Chol	349.8	3.1	0.01				
	Chol/Liq	374	0.67	1.8				
Note: Smec/Chol behavior observed on cooling.								
$C_{35}H_{62}O_2$	$5\alpha$ -cholestan-3 $\beta$ -ol octanoate							
	Smec/Smec	349.15	40.6	116.4	139.1	[311]		
	Smec/Smec	342.15	0.35	1.0				
Note: Liquid crystalline phase detected upon cooling.								
$C_{36}H_{26}Cl_2N_2O_6$	4,5-dichloro-1,3-phenylene bis[4-(4-methoxybenzylideneamino)benzoate]							
	Smec/Smec	489.2	47	96.08	98.63	[86]		
	Smec/Smec	503.2	1.3	2.58				
$C_{36}H_{27}ClN_2O_6$	4-chloro-1,3-phenylene bis[4-(4-methoxybenzylideneamino)benzoate]							
	Smec/Smec	459.2	43	93.64	94.28	[86]		
	Smec/Smec	472.2	0.3	0.64				
$C_{36}H_{32}N_2O_4$	3,3',4,4'-biphenyltetracarboxy-N,N'-bis(butylphenyl)diimide							
	Smec/Smec	537.9	33.5	62.28	[311]	[311]		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{36}H_{32}N_2O_6$	Smec/Liq	556.3	8.4	15.10	77.38	[94]
	Sol/Smec	545.1	36.8	67.51		
	Smec/Liq	587.3	8.2	13.96	81.47	[94]
$C_{36}H_{33}F_{13}O_5$	(S)-4-(1-methylheptyloxycarbonyl)phenyl 4-(1H,1H,2H,2H-perfluoroctyloxy)biphenyl-4'-carboxylate					
	Sol/Sol	350.0	13.06	37.31		
	Sol/Smec	371.0	19.60	52.83		
	Smec/Smec	421.3	1.05	2.49		
$C_{36}H_{35}NO_2$	Smec/Liq	456.8	4.65	10.18	102.81	[255]
	[1-(4-octylbiphenyl)-3-(4"-cyanobiphenyl)]-propane-1,3-dione					
	Sol/Smec	456.7	22.97	50.30		
	Smec/Nem	465.2	0.42	0.90		
$C_{36}H_{36}N_2O_2$	Nem/Liq	468.2	0.13	0.28	51.48	[254]
	$\alpha,\omega$ -bis[(4,4'-cyanobiphenyl)oxy]decane					
	Sol/Nem	438.7	53.1	121.04		
$C_{36}H_{36}N_2O_6$	Nem/Liq	457.6	8.2	17.92	138.96	[127]
	$\alpha,\omega$ -bis[4-(4-ethoxybenzoyloxy)benzylideneamino]butane					
	Sol/Nem	437.2	54.0	123.51		
$C_{36}H_{38}F_4O_5$	Nem/Liq	504.2	9.2	18.24	141.75	[293]
	4-[(S)-2-methylbutoxycarbonyl]phenyl 4-[(4-nonyloxy-2,3,4,5-tetrafluorophenyl)ethynyl]benzoate					
	Sol/Smec	347.2	27.02	77.82		
	Smec/Smec	362.2	Not detected by dsc			
$C_{36}H_{38}N_4O_6S_2$	Smec/Chst	401.5	0.64	1.59		
	Chst/Liq	412.5	0.49	1.19	80.60	[91]
	$\alpha,\omega$ -bis(4-ethoxyazobenzene-4'-carbonyloxy)hexane					
$C_{36}H_{38}N_4O_6S_2$	Sol/Nem	405.6	47.7	117.60		
	Nem/Liq	460.9	7.5	16.27	133.87	[107]
$C_{36}H_{38}O_6$	bis[4-(5-n-hexylthio-1,3,4-oxadiazole-2-yl)phenyl] terephthalate					
	Sol/Smec	407.2	48.7	119.60		
	Smec/Nem	552.2	1.7	3.08		
	Nem/Liq	556.2	0.8	1.44	124.12	[23]
$C_{36}H_{38}O_6$	bis(4-pentyloxyphenyl) 4,4'-biphenylenedicarboxylate					
	Sol/Sol	378.6	19.4	51.24		
	Sol/Smec	418.6	30.6	73.10		
	Smec/Nem	608.6	2.7	4.44		
$C_{36}H_{38}O_8$	Nem/Liq	619.2	2.1	3.39	132.17	
	benzoic acid, 4,4'-[1,4-phenylenebis[(1-oxo-2-propene-3,1-diyl)oxo]]-bis, dipentyl ester					
	Sol/Sol	390.7	3.8	9.73		
	Sol/Smec	447.0	47.0	105.15		
$C_{36}H_{38}O_8$	Smec/Nem	503.5	0.7	1.39		
	Nem/Liq	527.9	0.7	1.33	117.60	[271]
	1,4-benzenedicarboxylic acid, bis[4-[3(pentyloxy)-3-oxo-1-propenyl]phenyl ester					
$C_{36}H_{38}O_8$	Sol/Smec	386.0	36.6	94.82		
	Smec/Nem	526.1	2.2	4.18		
	Nem/Liq	532.6	0.9	1.69	100.69	[271]
$C_{36}H_{39}N_3O_6$	4-(4'-ethoxybenzoyloxy)-2-butoxy-4'-(4-butyoxysalicylaldimine)azobenzene					
	Sol/Nem	431.2	42.30	98.10		
	Nem/Liq	540.2	2.43	4.50	102.60	[109]
$C_{36}H_{40}F_4O_4$	pentyl 4-[4-((4-nonyloxy-2,3,5,6-tetrafluorophenyl)ethynyl)benzyloxy]benzoate					

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{36}H_{40}N_4O_6S_3$	Sol/Smec	357.9	47.62	133.05		[96]
	Smec/Liq	369.3	5.86	15.87	148.92	
$C_{36}H_{41}FO_7S$	<i>bis</i> [4-(5-n-heptylthio-1,3,4-oxadiazole-2-yl)phenyl]-2,5-thiophene-dicarboxylate					
	Sol/Smec	409.2	25.2	61.58		[23]
	Smec/Liq	498.2	7.5	15.05	76.63	
$C_{36}H_{41}FO_7S$	1-(butoxycarbonyl)ethyl 5-[{(3-fluoro-4-nonyloxybenzoylphenyl)ethynyl]thiophene-2-carboxylate}					
	Sol/Smec	338.7	36.13	106.67		[267]
	Smec/Nem	348.2	1.35	3.88		
$C_{36}H_{41}F_{13}O_5$	Nem/Liq	354.8	0.62	1.75	112.3	
	4-(3,3,4,4,5,5,6,6,7,7,8,8,8-tridecafluoroctyloxycarbonyl)phenyl 4-(tetradecyloxy)benzoate					
	Sol/Smec	365.5	23.84	65.23		[125]
$C_{36}H_{41}NO_3$	Smec/Liq	374.9	11.12	29.66	94.89	
	4-(4'-dodecyloxybenzoyloxybenzylidene)-2"-aminonaphthalene					
	Sol/Smec	379.2	10.77	28.40		[186]
$C_{36}H_{41}NO_3S$	Smec/Nem	430.2	0.12	0.28		
	Nem/Liq	471.2	0.35	0.74	29.42	
	4-(4'-dodecyloxybenzoyloxybenzylidene)-2"-aminonaphthalene-1"-thiol					[186]
$C_{36}H_{41}NO_7$	Sol/Nem	413.2	21.48	51.98		
	Nem/Liq	488.2	0.23	0.47	52.45	
	$5-[[4'-(decyloxy)[1,1'-biphenyl]-4-yl]oxy]carbonyl]-1,3-dihydro-1,3-dioxo-2H-isoindole-2-butanoic acid, methyl ester$					[280]
$C_{36}H_{42}F_2O_2S$	Sol/Sol	408.1	8.82	21.61		
	Sol/Smec	421.1	34.4	81.69		
	Smec/Liq	440.4	6.30	14.31	117.61	
$C_{36}H_{42}F_2O_7$	3,4-difluoro-2,5- <i>bis</i> [[4-(octyloxy)phenyl]ethynyl]thiophene					[373]
	Sol/Smec	338.4	38.9	114.95		
	Smec/Liq	382.9	1.4	3.66	118.61	
$C_{36}H_{42}N_2O_2$	(R)-1-methylheptyl 4-[4-(4-heptyloxy-3-fluorobenzoyloxy)-3-fluorobenzoyloxy] benzoate					[69]
	Sol/Smec	341.7	17.94	52.50		
	Smec/Smec	343.6	0.08	0.23		
$C_{36}H_{42}N_2O_2$	Smec/Liq	381.2	4.99	13.09	65.82	
$C_{36}H_{42}N_2O_2$	2,3-dicyano-4-heptylphenyl 4'-octyl-4-biphenyl-1-carboxylate					[202]
	Sol/Nem	357.2	11.30	31.63		
	Nem/Liq	383.7	Not reported in paper			
$C_{36}H_{42}N_2O_2S$	2-[4-(octyloxy)phenyl]-5-[[4-(octyloxy)phenyl]ethynyl]-3,4-thiophenedicarbonitrile					[373]
	Sol/Smec	346.2	29.6	85.50		
	Smec/Nem	353.8	0.4	1.13		
$C_{36}H_{42}N_2O_2$	Nem/Liq	368.6	1.5	4.07	90.70	
$C_{36}H_{42}N_4O_2$	$\alpha$ -(4'-propylazobenzene-4-oxy)- $\omega$ -(4'-propylazobenzene-4-oxy)hexane					[67]
	Sol/Nem	438.2	37.2	84.89		
	Nem/Liq	476.2	7.32	15.37	99.86	
$C_{36}H_{42}N_4O_2$	$\alpha$ -(4'-hexylazobenzene-4-oxy)- $\omega$ -(azobenzene-4-oxy)hexane					[67]
	Sol/Nem	409.2	52.73	128.86		
	Nem/Liq	424.2	4.16	9.81	138.67	
$C_{36}H_{42}N_4O_2$	$\alpha$ -(4'-pentylazobenzene-4-oxy)- $\omega$ -(4'-methylazobenzene-4-oxy)hexane					[67]
	Sol/Nem	406.2	45.59	112.23		
	Nem/Liq	471.2	6.78	14.39	126.62	
$C_{36}H_{42}N_4O_2$	$\alpha$ -(4'-butylazobenzene-4-oxy)- $\omega$ -(4'-ethylazobenzene-4-oxy)hexane					

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{36}H_{42}O_2$	Sol/Smec	413.2	30.92	74.38	86.93	[67]
	Smec/Nem	416.2	Too small to be measured			
	Nem/Liq	465.2	5.84	12.55		
$C_{36}H_{42}O_5$	$4,4''$ -dihexyloxy-p-quaterphenyl					
	Sol/Sol	344.0	5.84	16.98		
	Sol/Sol	476.0	6.66	13.99		
	Sol/Smec	587.0	11.74	20.00		
	Smec/Liq	640.0	8.96	14.00	64.97	174.0
$C_{36}H_{42}O_5$	$7$ -[(4'-tetradecyloxy)benzoyloxy]isoflavone					
	Sol/Smec	407.3	29.82	3.21		
	Smec/Liq	466.6	4.75	10.18	83.39	[14]
$C_{36}H_{42}O_6S_2$	<i>bis</i> (4-heptyloxyphenyl) 2,2'-bithiophene-5,5'-dicarboxylate					
	Sol/Sol	389.2	9.8	24.06		
	Sol/Sol	400.9	14.6	36.41		
	Sol/Smec	410.9	24.6	59.87		
	Smec/Liq	498.0	7.1	14.26	134.60	206.2
$C_{36}H_{42}O_8$	<i>bis</i> (4-heptyloxy carbonylphenyl) terephthalate					
	Sol/Smec	418.2	41.0	98.04		
	Smec/Liq	452.2	6.3	13.93	111.97	195.0
$C_{36}H_{43}ClF_2O_5$	4-[(2S)-2-chloro-3-methylpentanoyloxy]biphenyl-4'-yl 4-undecyloxy-2,3-difluorobenzoate					
	Sol/Smec	340.0	36.57	107.56		
	Smec/Nem	407.8	1.19	2.92		
	Nem/Blue	417.3	1.01	2.42		
	Blue/Liq	417.4	Not detected by dsc112.90			[121]
$C_{36}H_{43}ClF_2O_5$	4-[(2S)-2-chloro-3-methylbutanoyloxy]biphenyl-4'-yl 4-dodecyloxy-2,3-difluorobenzoate					
	Sol/Smec	347.8	26.96	77.52		
	Smec/Nem	422.4	1.04	2.46		
	Nem/Liq	432.6	1.22	2.82	82.80	[121]
	$2,3$ -dihydro-2[(1S)-1-methylpropyl]-1,3-dithioxo-1 <i>H</i> -isoindole-5-carboxylic acid, 4'-(undecyloxy)[1,1'-biphenyl]-4-yl ester					
$C_{36}H_{43}NO_4S_2$	Sol/Sol	236.5	14.4	60.89		
	Sol/Smec	324.9	19.5	60.02		
	Smec/Smec	412.8	0.80	1.94		
	Smec/Liq	419.1	1.51	3.60	126.45	[306]
	$2,3$ -dihydro-2[(1S)-1-methylpropyl]-1-oxo-3-thioxo-1 <i>H</i> -isoindole-5-carboxylic acid, 4'-(undecyloxy)[1,1'-biphenyl]-4-yl ester					
$C_{36}H_{43}NO_4S$	Sol/Sol	233.6	2.22	9.50		
	Sol/Smec	488.2	26.2	53.67		
	Smec/Liq	517.9	4.48	8.65	71.82	[306]
	$2,3$ -dihydro-2[(1S)-1-methylpropyl]-3-oxo-1-thioxo-1 <i>H</i> -isoindole-5-carboxylic acid, 4'-(undecyloxy)[1,1'-biphenyl]-4-yl ester					
	Sol/Smec	489.1	36.7	75.04		
$C_{36}H_{43}NO_5$	Smec/Liq	505.8	2.84	5.61	80.65	[306]
	$2,3$ -dihydro-2[(1S)-1-methylpropyl]-1,3-dioxo-1 <i>H</i> -isoindole-5-carboxylic acid, 4'-(undecyloxy)[1,1'-biphenyl]-4-yl ester					
	Sol/Sol	446.3	7.53	16.87		
	Sol/Smec	529.6	20.6	38.90		
	Smec/Liq	570.6	2.34	4.10	59.87	[306]
$C_{36}H_{44}N_2O_4$	diheptyl N,N'-[1,4-phenylene- <i>bis</i> (methylidyne)]- <i>bis</i> [aminobenzoate]					
	Sol/Smec	365.2	41.0	112.27		
	Smec/Smec	413.2	Not reported in paper			

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{36}H_{44}O_2$	Smec/Liq	469.2	4.5	9.59		[192]
	Sol/Nem	358.1	35.4	98.86		
	Nem/Liq	445.7	3.2	7.18	106.04	[225]
$C_{36}H_{44}O_5$	2-acetyl-1,4-phenylene 4-heptylbenzoate					
	Sol/Nem	340.2	20.6	60.55		
	Nem/Liq	366.8	3.21	8.75	69.30	[432]
$C_{36}H_{44}O_5S$	(S)-1-methylheptyl 5-[(4-octyloxybenzoyloxyphenyl)ethynyl]thiophene-2-carboxylate					
	Sol/Smec	321.1	30.63	95.39		
	Smec/Liq	369.5	5.41	14.64	110.03	[267]
$C_{36}H_{44}O_6S$	4-[[4-(heptyloxy)benzoyl]thio]benzoic acid, 4-[[[(1-methylheptyl)oxy]carbonyl]phenyl ester					
	Sol/Smec	356.9	28.6	80.13		
	Smec/Liq	434.9	6.0	13.80	93.93	[85]
$C_{36}H_{44}O_7$	(1S)-2-butoxy-1-methyl-2-oxoethyl 4-[(2E)-3-[6-nonyloxy)-2-naphthalenyl]1-oxo-2-propenyl]oxy]benzoate					
	Sol/Meso	342.9	27.82	81.13		
	Meso/Liq	383.4	3.38	8.82	89.95	[401]
$C_{36}H_{44}O_7$	2-acetyl-1,4-phenylene 4-(heptyloxy)benzoate					
	Sol/Nem	369.9	51.2	138.42		
	Nem/Liq	401.9	2.6	6.47	144.89	[432]
$C_{36}H_{45}F_3O_3$	4-(1,1,1-trifluoro-2-octyl)phenyl 4'-nonyloxybiphenyl-4-carboxylate					
	Sol/Smec	337.2	33.22	98.52		
	Smec/Liq	356.2	2.86	8.03	106.55	[142]
$C_{36}H_{46}O_3$	1-(4'-dodecylbiphenyl-4-yl)-3-(2-propoxyphephenyl)propane-1,3-dione					
	Sol/Nem	333.2	44.18	132.59		
	Nem/Liq	339.2	0.37	1.09	133.68	[297]
$C_{36}H_{46}O_5S$	(R)-4-(3-ethylmercapto-2-methylpropionyl)phenyl 4'-undecyloxy-biphenyl-4-carboxylate					
	Sol/Smec	353.8	34.43	97.31		
	Smec/Smec	406.5	0.06	0.15		
$C_{36}H_{46}O_6$	4-[(1R)-2-(1-ethylpropoxy)-1-methyl-2-oxoethoxy]phenyl 4'-(nonyloxy)[1,1'-biphenyl]-4-carboxylate					
	Sol/Smec	302.5	19.83	65.55		
	Smec/Meso	302.5	0.99	3.27		
$C_{36}H_{46}O_7$	Too small to be measured					
	Meso/Nem	381.1				
	Nem/Meso	388.2				
$C_{36}H_{46}O_7$	Too small to be measured					
	Meso/Liq	389.2	1.47	3.78	72.60	[239]
$C_{36}H_{46}O_7$	4-(4-tetradecyloxybenzoyloxy)phenyl (3,4-dimethoxy)benzoate					
	Sol/Nem	368.9	10.49	28.44		
	Nem/Liq	401.0	0.60	1.50	29.94	203.5
$C_{36}H_{46}O_7$	203.5					[103]
	Sol/Nem	360.7	7.08	19.63		
	Nem/Liq	388.6	0.31	0.80	20.43	[103]
$C_{36}H_{46}O_7$	4-(4-butoxybenzoyloxy)phenyl (3,4-dihexyloxy)benzoate					
	Sol/Nem	371.6	9.50	25.57		
	Nem/Liq	390.2	0.41	1.05	26.62	203.5
$C_{36}H_{47}ClN_2O_2$	4-(4-octyloxybenzoyloxy)phenyl (3,4-dibutoxy)benzoate					
	Sol/Nem	331.9	49.5	149.14		
	Nem/Liq	453.6	1.68	3.70	152.84	[365]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{fus}} S_{tpce}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{fus}} S_{tpce}$ (estimated)	Ref.
		T (K)	$\Delta H_{pce}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{pce}$		
$C_{36}H_{48}N_2O_3$	Sol/Smec	340.2	34.7	102.00		[201]
	Smec/Chol	348.2		Not reported in paper		
	Chol/Liq	409.2		Not reported in paper		
$C_{36}H_{48}N_2O_3$	Sol/Smec	340.2	35.5	104.35		[201]
	Smec/Chol	348.2		Not reported in paper		
	Chol/Liq	406.2		Not reported in paper		
$C_{36}H_{48}N_2O_4$						
	Sol/Smec	367.1	13.46	36.67		[319]
	Smec/Nem	390.1	0.17	0.44		
$C_{36}H_{48}N_2O_4$	Nem/Liq	464.4	0.57	1.23	38.34	
	Sol/Nem	357.2	35.0	97.98		[339]
$C_{36}H_{48}N_2O_4$	Nem/Liq	441.2	1.3	2.95	100.93	
	Sol/Nem	370.2	53.0	143.17		[339]
$C_{36}H_{48}N_2O_4$	Nem/Liq	437.2	1.2	2.74	145.91	
$C_{36}H_{48}N_2O_4$	Sol/Nem	383.2	48.0	125.26		[339]
	Nem/Liq	421.2	1.9	4.51	129.77	
$C_{36}H_{48}N_2O_4$	Sol/Nem	374.2	53.0	141.64		[339]
	Nem/Liq	434.2	1.8	4.15	145.79	
$C_{36}H_{48}N_2O_4$	Sol/Nem	379.5	59.2	155.99		[388]
	Nem/Liq	387.2	0.9	2.32	158.31	
$C_{36}H_{48}O_3S$	Sol/Smec	330.0	31.5	95.45		[281]
	Smec/Nem	376.7	2.6	6.90		
	Nem/Liq	404.6	1.6	3.95	106.3	
$C_{36}H_{48}O_6S$						
	Sol/Sol	393.8	8.6	21.84		[12]
	Smec/Smec	399.7	60.6	151.61		
$C_{36}H_{48}O_7$	Smec/Liq	409.6	9.1	22.22	195.67	
					218.7	
	Sol/Smec	375.1	22.1	58.72		[286]
$C_{36}H_{48}O_7$	Smec/Liq	434.9	5.8	13.34	72.06	
$C_{36}H_{49}NO_2$						
	Sol/Smec	331.5	29.40	88.69		[139]
	Smec/Smec	338.3	0.69	2.04		
	Smec/Smec	342.8	0.37	1.08		
	Smec/Smec	430.4	0.21	0.49		
$C_{36}H_{49}NO_2$	Smec/Liq	441.1	8.76	19.86	112.16	
	Sol/Smec	355.7	39.58	111.27		[323]
	Smec/Smec	418.2	1.13	2.70		
$C_{36}H_{49}NO_2$	Smec/Smec	477.7		Too small to be measured		
	Smec/Liq	492.7	8.20	16.64	130.61	

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.	
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$			
$C_{36}H_{49}NO_4$	Sol/Smec	348.6	11.70	33.56		[427]	
	Smec/Smec	388.8	2.70	6.94			
	Smec/Liq	406.1	1.24	3.05	43.55		
$C_{36}H_{49}NO_4$	Sol/Smec	343.7	6.99	20.34		[427]	
	Smec/Smec	396.7	2.61	6.58			
	Smec/Liq	416.5	0.83	1.99	28.91		
$C_{36}H_{49}N_5O_3$	Sol/Smec	443.6	15.24	34.36		[326]	
	Smec/Liq	472.6	1.62	3.43	37.79		
	Sol/Smec	351.2	31.2	88.84			
$C_{36}H_{50}N_2O_3$	Smec/Chol	358.2	Not reported in paper			[201]	
	Chol/Liq	409.2	Not reported in paper				
	Sol/Smec	373.5	15.36	41.12			
$C_{36}H_{50}O_4$	Smec/Nem	407.6	9.58	23.50		[348]	
	Nem/Liq	594.2	Not reported in paper				
	Sol/Smec	371.2	41.8	112.61			
$C_{36}H_{50}S_4$	Smec/Liq	441.2	36.9	8.36	120.97	[303]	
	Sol/Smec	380.2	30.98	81.48		[131]	
	Smec/Nem	495.1	1.58	3.19			
$C_{36}H_{51}NO_2$	Nem/Liq	502.3	1.00	1.99	86.66		
	Sol/Smec	355.6	60.54	170.25		[112]	
	Smec/Nem	391.9	0.33	0.84			
$C_{36}H_{52}N_2O_2$	Nem/Liq	404.2	2.75	6.80	177.89		
	Sol/Smec	372.6	50.2	134.73		[43]	
	Smec/Liq	390.1	7.1	18.20	152.93		
$C_{36}H_{52}O_2$	Sol/Sol	360.3	24.4	67.72		[225]	
	Sol/Smec	365.0	9.2	25.21			
	Smec/Nem	384.1	0.9	2.34			
	Nem/Liq	413.3	2.3	5.56	100.83		
	Note: Enthalpies of the Sol/Sol and Sol/Smec transitions are from cooling measurements.						
$C_{36}H_{52}O_2$	cholesteryl cinnamate						
	Sol/Chol	435.75	28.74	65.96		[155, 310]	
	Chol/Liq	488.15	0.71	1.46	67.42		
$C_{36}H_{52}O_4$	di(4'-octylphenyl)- <i>trans</i> -cyclohexane-1,4-dicarboxylate					[215]	
	Sol/Smec	343.5	28.0	81.51			
	Smec/Smec	387.5	4.09	10.55			
$C_{36}H_{52}O_6$	Smec/Liq	412.4	6.95	16.85	108.91		190.1
	Sol/Smec	366.2	31.2	85.20			
	Smec/Smec	384.2	1.28	3.33			
	Smec/Smec	392.2	Not reported in paper				

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{36}H_{53}NOS$	Smec/Liq	451.2	6.75	14.96	203.7	[220]
	Sol/Smec	347.8	22.25	63.97		
	Smec/Nem	441.2	1.74	3.94		
	Nem/Liq	444.2	1.54	3.47	71.38	[131]
$C_{36}H_{54}$	3-[4-( <i>trans</i> -4-heptylcyclohexyl)phenyl]-5-(5-decyl-2-thienyl)isoxazole					
	Sol/Smec	349.2	6.07	17.38		
	Smec/Smec	421.2	2.09	4.96		
	Smec/Smec	463.7	6.28	13.54		
	Smec/Nem	488.2	1.55	3.17		
$C_{36}H_{54}O_2$	Nem/Liq	511.2	5.44	10.64	49.69	[348]
	4-[2-(4-pentylcyclohexyl)ethyl]-4'-(2-(4-propylcyclohexyl)ethyl)-1,1'-biphenyl					
	Sol/Smec	384.7	36.69	95.38		
	Chol/Liq	387.3	0.50	1.30	96.68	121.3
$C_{36}H_{54}O_3$	cholesteryl $\omega$ -phenylpropionate					
	Sol/Chol	333.2	24.89	74.70		
	Smec/Nem	380.2	Not reported in paper			
	Nem/Liq	426.2	Not reported in paper			[199, 207]
$C_{36}H_{54}O_4$	4-(decyloxy)benzoic acid, 4-[2-( <i>trans</i> -4-pentylcyclohexyl)ethyl]phenyl ester					
	Sol/Smec	339.2	28.85	85.05		
	Smec/Smec	349.2	Not reported in paper			
	Smec/Nem	398.2	Not reported in paper			
$C_{36}H_{55}ClN_2O_4$	Nem/Liq	427.2	Not reported in paper			[207]
	4-[( <i>trans</i> -4-pentylcyclohexyl)methoxy]phenyl 4-undecyloxybenzoate					
	Sol/Smec	334.8	38.70	115.60		
	Smec/Liq	348.7	5.23	15.00	130.60	[47]
$C_{36}H_{56}N_2O_{10}$	N,N'-dipentanoyl-2,3,5,6-tetrakis(pentanoyloxy)-1,4-benzenediamine					
	Sol/Disc	407.2	26.0	63.85		
	Disc/Liq	481.2	45.0	93.52	157.4	[188]
$C_{36}H_{57}NO$	4-octyl-N-[[4-(pentadecyloxy)phenyl]methylene]benzenamine					
	Sol/Smec	341.9	53.10	155.31		
	Smec/Liq	363.4	18.49	50.88	206.19	[240]
$C_{36}H_{57}NO$	4-pentyl-N-[[4-(octadecyloxy)phenyl]methylene]benzenamine					
	Sol/Smec	343.4	55.88	162.73		
	Smec/Liq	358.7	13.06	36.41	199.14	[240]
$C_{36}H_{58}N_2O_3$	4,4'-didodecyloxyazoxybenzene					
	Sol/Smec	354.9	42.06	118.51		
	Smec/Liq	395.2	11.97	30.29	121.80	[179]
$C_{36}H_{60}O_6$	(S)-4-(2-ethylheptyloxy carbonyl)phenyl 4-pentadecanoyloxytolane-4'-carboxylate					
	Sol/Smec	354.8	52.3	147.41		
	Smec/Liq	370.2	2.6	7.02	154.43	[173]
$C_{36}H_{61}N_3O_6$	4,4'-bis[4-(4-octyloxybenzoyloxy)benzylideneamino]diphenylamine					
	Sol/Smec	488.2	40.0	81.93		
	Smec/Smec	497.2	0.7	1.41		
$C_{36}H_{62}N_2O_4$	Smec/Nem	505.2	0.3	0.59	83.93	[284]
	N,N'-dinonanoyl-2,5,6-trimethyl-4-nanonoyloxy-1,3-benzenediamine					
	Sol/Meso	399.2	14.0	35.07		
	Meso/Meso	442.2	21.0	47.49		
	Meso/Liq	471.2	4.0	8.49	91.05	214.1
[193]						

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)						
$C_{36}H_{62}OS$		cholesteryl thiononanoate						
	Sol/SmeC	342.2	22.3		65.3			
	SmeC/Chol	357.2	0.6		1.6			
$C_{36}H_{62}O_2$	Chol/Liq	370.7	0.5		1.4	68.3	NA	[155,312]
		cholesterol nonanoate						
	Sol/Chol	351.0	25.13		71.60			[166, 170]
$C_{36}H_{63}N_3O_3$	Chol/Liq	364.9	0.55		1.51	73.11		
		Independent values from another reference						
	Sol/Chol	353.9	22.27		62.93			[169]
$C_{36}H_{64}N_2O_2$	Chol/Liq	366.2	0.51		1.39	64.32		
		N,N',N"-trinonanoyl-2,4,6-trimethyl-1,3,5-benzenetriamine						
	Sol/Meso	473.2	8.0		16.91			[190]
$C_{36}H_{64}N_2O_2$	Meso/Liq	495.2	19.0		38.37	55.28		
		N,N'-ditetradecanoyl-2,6-dimethylbenzene-1,4-diamine						
	Sol/Meso	386.2	47.0		121.70			[36]
$C_{36}H_{64}N_2O_2$	Meso/Liq	394.2	12.0		30.44	152.14		
$C_{36}H_{82}Si_{12}$		N,N'-ditetradecylbenzene-1,2-dicarboxamide						
	Sol/Meso	372.2	84.0		225.69			
	Meso/Liq	373.2	12.0		32.15	257.84		[61]
$C_{37}H_{37}NO_2$		1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,9,9,10,10,11,11,12,12-tetracosomethyl-1,12-diphenyldodecasilane						
	Sol/Meso	399.2	38.1		95.44			
	Meso/Liq	412.2	24.2		58.71	154.15		[400]
$C_{37}H_{39}F_3O$		[1-(4-nonylbiphenyl)-3-(4"-cyanobiphenyl)]-propane-1,3-dione						
	Sol/SmeC	458.7	22.97		50.08			
	SmeC/Liq	472.7	2.13		4.51	54.59		[254]
$C_{37}H_{39}FeNO$		2-methyl-4-[2-[ <i>trans</i> -4-pentylcyclohexyl]ethyl]phenylethyneyl]-1-[(4-trifluoromethoxyphenyl)ethynyl]benzene						
	Sol/SmeC	370.2	26.4		71.31			
	SmeC/Nem	489.2	0.17		0.35			
$C_{37}H_{40}N_4O_6$	Nem/Liq	532.2	2.9		5.45	77.11		[76]
		[4'-(E)-[[4-(octyloxy)phenyl]imino]methyl][1,1'-biphenyl]-4-yl]ferrocene						
	Sol/SmeC	450.2	17.94		39.85			
$C_{37}H_{43}ClN_2O_3$	SmeC/Nem	456.2	17.94		39.32			
	Nem/Liq	474.2	0.68		1.43	80.60		[433]
		$\alpha,\omega$ -bis(4-butoxyazobenzene-4'-carbonyloxy)propane						
$C_{37}H_{43}FO_7S$	Sol/SmeC	406.9	31.8		78.15			
	SmeC/Liq	423.4	15.9		37.55	115.70		[107]
$C_{37}H_{44}F_2O_7$		2"-[[4-[4'-tetradecyloxybenzoyloxy]-3-chlorophenylazo]naphthalene						
	Sol/Nem	367.2	40.77		111.03			
	Nem/Liq	422.2	0.55		1.30	112.33		[274]
$C_{37}H_{44}FO_7S$		1-(butoxycarbonyl)ethyl 5-[(3-fluoro-4-decyloxybenzoylphenyl)ethynyl]thiophene-2-carboxylate						
	Sol/SmeC	334.6	35.90		107.29			
	SmeC/Nem	348.8	1.16		3.33			
$C_{37}H_{44}FO_7S$	Nem/Liq	355.1	0.77		2.17	112.79		[267]
$C_{37}H_{44}F_2O_7$		(R)-1-methylheptyl 4-[4-(4-octyloxy-3-fluorobenzoyloxy)-3-fluorobenzoyloxy] benzoate						
	Sol/SmeC	335.1	23.67		70.64			
	SmeC/SmeC	340.7	Too small to be measured					
	SmeC/SmeC	342.6	Too small to be measured					

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.		
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$				
$C_{37}H_{44}N_4O_2$	Smec/Smec	354.0	0.10	0.28	84.03	[69]		
	Smec/Liq	379.9	4.98	13.11				
$C_{37}H_{44}N_4O_2$	$\alpha$ -(4'-hexylazobenzene-4-oxy)- $\omega$ -(4'-methylazobenzene-4-oxy)hexane					[67]		
	Sol/Nem	412.2	45.24	109.75	122.89			
	Nem/Liq	464.2	6.10	13.14				
$C_{37}H_{44}N_4O_2$	$\alpha$ -(4'-pentylazobenzene-4-oxy)- $\omega$ -(4'-ethylazobenzene-4-oxy)hexane					[67]		
	Sol/Smec	393.2	37.92	96.44	110.66			
	Smec/Nem	432.2	0.21	0.49				
$C_{37}H_{44}N_4O_2$	Nem/Liq	466.2	6.40	13.73	96.37	[67]		
	$\alpha$ -(4'-butylazobenzene-4-oxy)- $\omega$ -(4'-propylazobenzene-4-oxy)hexane							
	Sol/Smec	427.2	34.81	81.48				
$C_{37}H_{44}O_7$	Smec/Nem	434.2	0.29	0.67	102.81	[149]		
	Nem/Liq	469.2	6.67	14.22				
	(S)-(+) -1-methylheptyl 4-[4-( <i>trans</i> -4-hexyloxy)cinnamoyloxy]benzoate							
$C_{37}H_{45}ClF_2O_5$	Sol/Smec	380.4	34.68	91.17	132.97	[121]		
	Smec/Nem	391.2	0.03	0.08				
	Nem/Liq	451.6	5.22	11.56				
$C_{37}H_{45}NO_3S_2$	4-[(2S)-2-chloro-3-methylpentanoyloxy]biphenyl-4'-yl 4-dodecyloxy-2,3-difluorobenzoate					[306]		
	Sol/Smec	334.2	42.5	127.17	73.83			
	Smec/Nem	409.7	1.23	3.00				
$C_{37}H_{45}NO_3S_2$	Nem/Liq	417.2	1.17	2.80	70.63	[306]		
	2,3-dihydro-2[(2S)-2-methylbutyl]-1,3-dithioxo-1 <i>H</i> -isoindole-5-carboxylic acid, 4'-(undecyloxy)[1,1'-biphenyl]-4-yl ester							
	Sol/Smec	361.3	24.6	68.09				
$C_{37}H_{45}NO_3S_2$	Smec/Smec	548.9	0.13	0.24	56.81	[306]		
	Smec/Nem	616.7	1.26	2.04				
	Nem/Liq	627.6	2.17	3.46				
$C_{37}H_{45}NO_3S$	2,3-dihydro-2[(2S)-2-methylbutyl]-1-oxo-3-thioxo-1 <i>H</i> -isoindole-5-carboxylic acid, 4'-(undecyloxy)[1,1'-biphenyl]-4-yl ester					[306]		
	Sol/Smec	418.8	26.3	62.80	71.55			
	Smec/Smec	539.7	0.24	0.44				
$C_{37}H_{45}NO_3S$	Smec/Liq	645.6	4.77	7.39	114.78	[306]		
	2,3-dihydro-2[(2S)-2-methylbutyl]-3-oxo-1-thioxo-1 <i>H</i> -isoindole-5-carboxylic acid, 4'-(undecyloxy)[1,1'-biphenyl]-4-yl ester							
	Sol/Sol	402.8	5.32	13.21				
$C_{37}H_{45}NO_3S$	Sol/Smec	437.5	16.7	38.17	101.25	[306]		
	Smec/Smec	597.9	0.20	0.33				
	Smec/Smec	730.0	3.72	5.10				
$C_{37}H_{45}NO_3S$	2,3-dihydro-2[(2S)-2-methylbutyl]-1,3-dioxo-1 <i>H</i> -isoindole-5-carboxylic acid, 4'-(undecyloxy)[1,1'-biphenyl]-4-yl ester					128.56		
	Sol/Smec	443.8	27.7	62.42	115.59	[267]		
	Smec/Smec	615.1	0.13	0.21				
$C_{37}H_{46}O_5S$	Smec/Liq	750.3	6.69	8.92	115.59	[85]		
	(S)-1-methylheptyl 5-[(4-nonyloxybenzoyloxyphenyl)ethynyl]thiophene-2-carboxylate							
	Sol/Smec	328.2	37.67	114.78				
$C_{37}H_{46}O_6S$	Smec/Liq	365.8	5.04	13.78	2-(1-oxopropyl)-1,4-phenylene 4-(heptyloxy)benzoate	[85]		
	4-[[4-(octyloxy)benzoyl]thio]benzoic acid, 4-[[1-methylheptyl]oxy]carbonyl]phenyl ester							
	Sol/Smec	359.5	36.4	101.25				
$C_{37}H_{46}O_7$	Smec/Liq	432.4	6.2	14.34	115.59	[85]		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.				
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$						
$\text{C}_{37}\text{H}_{48}\text{O}_3$	Sol/Nem	361.3	41.9	115.97	120.46	[432]				
	Nem/Liq	378.3	1.7	4.49						
$\text{C}_{37}\text{H}_{48}\text{O}_3$	1-(4'-dodecylbiphenyl-4-yl)-3-(2-butoxyphenyl)propane-1,3-dione					[297]				
	Sol/Sol	315.2	12.76	40.48	128.55					
	Sol/Nem	320.7	27.78	86.62						
	Nem/Liq	343.7	0.50	1.45						
$\text{C}_{37}\text{H}_{48}\text{O}_3$	1-(4'-dodecylbiphenyl-4-yl)-3-(3-butoxyphenyl)propane-1,3-dione					[297]				
	Sol/Sol	352.7	19.30	54.72	154.49					
	Sol/SmeC	366.2	29.70	81.10						
$\text{C}_{37}\text{H}_{48}\text{O}_5$	SmeC/Liq	371.7	6.94	18.67	103.57	[255]				
	(S)-4-(1-methyloctyloxycarbonyl)phenyl 4-octyloxybiphenyl-4'-carboxylate									
	Sol/Sol	331.0	7.49	22.63						
	Sol/SmeC	353.0	23.4	66.29						
	SmeC/SmeC	392.0	0.12	0.31	143.3					
$\text{C}_{37}\text{H}_{48}\text{O}_5$	SmeC/SmeC	392.6	0.04	0.10						
	SmeC/Liq	417.2	5.94	14.24						
	Sol/SmeC	339.7	41.0	120.69	120.69	[184]				
$\text{C}_{37}\text{H}_{48}\text{O}_5\text{S}$	SmeC/SmeC	342.6	0.02	0.06						
	SmeC/Liq	363.2	8.2	22.58						
	(R)-4-(1-methylheptyloxycarbonyl)phenyl 4-(4-heptyloxyphenyl)ethylbenzoate									
$\text{C}_{37}\text{H}_{48}\text{O}_6$	Sol/Sol	362.2	4.95	13.67	90.91	[227]				
	Sol/SmeC	380.2	23.49	61.78						
	SmeC/Liq	415.2	6.42	15.46						
Note: Enthalpies were determined from cooling measurements.										
$\text{C}_{37}\text{H}_{48}\text{O}_6$	4'-(decyloxy)[1,1'-biphenyl]-4-yl 4-[(2S)-2-methylbutoxy]-1-oxopropoxy]benzoate					[227]				
	Sol/SmeC	371.2	28.38	76.45	92.36					
	SmeC/Liq	414.2	6.59	15.91						
Note: Enthalpies were determined from cooling measurements.										
$\text{C}_{37}\text{H}_{48}\text{O}_6$	4'-(decyloxy)[1,1'-biphenyl]-4-yl 4-[(2S)-1-oxo-2-(pentyloxy)propoxy]benzoate					[239]				
	Sol/SmeC	300.7	37.42	124.44	128.61					
	SmeC/Meso	381.0	0.85	2.23						
	Meso/Nem	382.8	Too small to be measured							
	Nem/Meso	386.5	Too small to be measured							
$\text{C}_{37}\text{H}_{49}\text{FO}_7$	Meso/Liq	387.4	0.75	1.94	117.58	[51]				
	[(R)-4-(1-methylheptyloxycarbonyl)]phenyl 4-(4-decyloxy-3-fluorobenzoyloxy)benzoate									
	Sol/SmeC	362.2	37.8	104.36						
	SmeC/SmeC	386.0	0.24	0.63						
$\text{C}_{37}\text{H}_{50}\text{N}_2\text{O}_3$	SmeC/Liq	400.2	5.04	12.59	91.44	[201]				
	4-[5-[(S)-5-methylheptyl]-2-pyrimidinyl]phenyl 4-(11-dodecenoxy)benzoate									
	Sol/SmeC	341.2	31.2	Not reported in paper						
	SmeC/Chol	354.2	Not reported in paper							
$\text{C}_{37}\text{H}_{50}\text{N}_2\text{O}_3$	Chol/Liq	404.2	Not reported in paper							
	4-[5-[(S)-6-methyloctyl]-2-pyrimidinyl]phenyl 4-(10-undecenoxy)benzoate					[201]				
	Sol/SmeC	334.2	34.9	104.43	104.43					
	SmeC/Chol	353.2	Not reported in paper							
$\text{C}_{37}\text{H}_{50}\text{N}_2\text{O}_3$	Chol/Liq	405.2	Not reported in paper							

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)						
$C_{37}H_{50}N_2O_4$	Sol/Nem	4-[[(4-ethoxyphenyl)azo]-2,3-dimethylphenyl 4-(tetradecyloxy)benzoate	371.2	48.0	129.31			[339]
	Nem/Liq		460.2	1.2	2.61	131.92		
$C_{37}H_{50}N_2O_4$	Sol/Nem	4-[[(4-ethoxyphenyl)azo]-2,3,5,6-tetramethylphenyl 4-(dodecyloxy)benzoate	420.2	55.0	130.89			[339]
	Nem/Liq		443.2	1.6	3.61	134.50		
$C_{37}H_{50}O_3S$	Sol/SmeC	S-4-decyloxybiphenyl (S)-4-(1-methylheptyloxy)benzoate	341.8	35.6	104.15			[281]
	SmeC/Nem		379.6	2.7	7.11			
	Nem/Liq		402.8	1.6	3.97	115.23		
$C_{37}H_{51}NO_2$	Sol/Sol	N-(2-hydroxy-4-hexyloxybenzylidene)-4"-dodecylphenylaniline	347.7	Not reported in paper				[323]
	Sol/SmeC		354.2	18.87	53.27			
	SmeC/SmeC		418.2	2.34	5.60			
	SmeC/SmeC		482.2	0.17	0.35			
	SmeC/Liq		491.2	8.12	16.53	75.75		
$C_{37}H_{51}NO_4$	Sol/SmeC	4'--(2-diethylamino-ethoxy)biphenyl 4-dodecyloxybenzoate	351.0	11.14	31.74			[427]
	SmeC/SmeC		392.8	2.74	6.97			
	SmeC/Liq		408.2	1.27	3.11	41.82		
$C_{37}H_{51}NO_4$	Sol/SmeC	4'-(2-diethylamino-propoxy)biphenyl 4-undecyloxybenzoate	345.9	9.88	28.56			[427]
	SmeC/SmeC		397.5	2.99	7.52			
	SmeC/Liq		411.7	1.17	2.84	38.92		
$C_{37}H_{52}N_2O_3$	Sol/SmeC	4-[5-[(S)-5-methylheptyl]-2-pyrimidinyl]phenyl 4-dodecyloxybenzoate	356.2	32.0	89.84			[201]
	SmeC/Chol		361.2	Not reported in paper				
	Chol/Liq		406.2	Not reported in paper				
$C_{37}H_{52}N_2O_3$	Sol/SmeC	4-[5-[(S)-6-methyloctyl]-2-pyrimidinyl]phenyl 4-undecyloxybenzoate	356.2	35.8	100.51			[201]
	SmeC/Chol		363.2	Not reported in paper				
	Chol/Liq		405.2	Not reported in paper				
$C_{37}H_{53}NO_2$	Sol/SmeC	3-[4-( <i>trans</i> -4-heptylcyclohexyl)phenyl]-5-[4-(nonyloxy)phenyl]isoxazole	371.9	24.47	65.80			[131]
	SmeC/SmeC		496.2	1.87	3.77			
	SmeC/Nem		502.9	1.23	2.45	72.02		
$C_{37}H_{54}$	Sol/SmeC	4-[2-(4-butylcyclohexyl)ethyl]-4'-(2-(4-pentylcyclohexyl)ethyl)-1,1-biphenyl	341.2	2.30	6.74			[348]
	SmeC/SmeC		346.2	3.64	10.51			
	SmeC/SmeC		431.2	1.46	3.39			
	SmeC/SmeC		470.2	6.69	14.23			
	SmeC/Nem		492.2	2.93	5.95			
	Nem/Liq		498.2	5.86	11.76	52.58		
$C_{37}H_{56}N_2O_3$	Sol/Sol	N-[4-[5-oxo-4-[(1-oxododecyl)amino]-1,3,6-cycloheptatrien-1-yl]phenyl]dodecamide	406.2	6.4	15.76			[251]
	Sol/SmeC		423.2	36.1	85.30			
	SmeC/Liq		430.2	7.3	16.97	118.03		
$C_{37}H_{56}O_2$	Sol/Liq	cholesteryl $\omega$ -phenylbutyrate	364.1	28.20	77.45	77.45	128.4	[155, 169]
	Liq/Chol		299.3	0.84	0.3			

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
Note: Enthalpy of the Liq/Chol transition is from cooling measurements.						
$C_{37}H_{56}O_3$			4-(6-methyloctyloxy)benzoic acid, 4-[2-( <i>trans</i> -4-heptylcyclohexyl)ethyl]phenyl ester			
	Sol/Smec	315.2	24.97	79.22		
	Smec/Smec	317.2	Not reported in paper			
	Smec/Smec	342.2	Not reported in paper			
	Smec/Chol	376.2	Not reported in paper			
$C_{37}H_{56}O_3$	Chol/Liq	409.2	Not reported in paper			[199, 208]
			4-(undecyloxy)benzoic acid, 4-[2-( <i>trans</i> -4-pentylcyclohexyl)ethyl]phenyl ester			
	Sol/Smec	327.2	32.97	100.76		
	Smec/Smec	341.2	Not reported in paper			
	Smec/Smec	393.2	Not reported in paper			
$C_{37}H_{56}O_3$	Smec/Nem	398.2	Not reported in paper			
	Nem/Liq	422.2	Not reported in paper			[199, 207]
$C_{37}H_{58}N_4O_7$			3,4,5-trioctyloxybenzaldehyde-2',4'-dinitrophenyl hydrazone			
	Sol/Meso	369.2	29.2	79.09		
	Meso/Liq	409.2	3.8	9.29	88.48	[13]
$C_{37}H_{58}N_4O_6$			3,4-didodecyloxybenzaldehyde-2',4'-dinitrophenyl hydrazone			
	Sol/Meso	367.2	48.5	132.08		
	Meso/Liq	403.2	3.0	7.44	139.52	[13]
$C_{37}H_{59}NO$			4-nonyl-N-[[4-(pentadecyloxy)phenyl]methylene]benzenamine			
	Sol/Smec	338.1	39.97	118.22		
	Smec/Liq	364.9	14.77	40.48	158.70	[240]
$C_{37}H_{59}NO$			4-hexyl-N-[[4-(octadecyloxy)phenyl]methylene]benzenamine			
	Sol/Smec	342.6	42.46	123.93		
	Smec/Liq	358.5	14.54	40.56	164.49	[240]
$C_{37}H_{59}NO$			N-(4-dodecyloxybenzylidene)-4-dodecylaniline			
	Sol/Smec	341.7	55.13	161.34		
	Smec/Liq	363.5	17.31	47.62	208.96	[256]
$C_{37}H_{60}N_2O$			4-heptyl-4'-octadecyloxyazobenzene			
	Sol/Smec	299.5	40.76	136.09		
	Smec/Smec	317.7	25.10	79.01		
	Smec/Liq	335.1	1.91	5.70	220.80	[390]
$C_{37}H_{62}N_2O_6$			4,4'-bis[4-(4-octyloxybenzylideneamino)benzoyloxy]diphenylmethane			
	Sol/Smec	416.2	46.0	110.52		
	Smec/Liq	469.2	20.0	42.63	153.15	[284]
$C_{37}H_{64}OS$			cholesteryl thiodecanoate			
	Sol/Smec	353.2	23.4	66.2		
	Smec/Chol	360.6	1.8	2.0		
	Chol/Liq	371.5	0.8	2.0	73.1	NA
$C_{37}H_{64}O_2$			cholesterol decanoate			
	Sol/Chol	360.4	30.10	83.52		
	Chol/Liq	364.6	0.63	1.73	85.25	[169]
Independent values from another reference						
	Sol/Smec	342.2	22.3	65.3		
	Smec/Chol	357.2	0.6	1.6		
	Chol/Liq	370.7	0.5	1.4	68.3	[155,312]
$C_{37}H_{64}O_6$			<i>cis, cis</i> -(3,5-dihydroxycyclohexyl) 3,4-bis(dodecyloxy)benzoate			
	Sol/Sol	320.7	25.0	77.95		
	Sol/Meso	352.7	21.6	61.24		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$\text{C}_{37}\text{H}_{66}\text{O}_2$	Meso/Liq	397.2	1.4	3.52	142.71	[98]
	Sol/Liq	354.65	51.5	145.2	145.2	153.3
	Smec/Chol	326.15	0.36	1.1		[311]
	Chol/Liq	344.15	1.42	4.1		
Note: Liquid crystalline phase detected upon cooling.						
$\text{C}_{37}\text{H}_{66}\text{N}_2\text{O}_2$	$\text{N},\text{N}'$ -ditetradecanoyl-2,3,5-trimethylbenzene-1,4-diamine					
	Sol/Meso	399.2	50	125.25		
	Meso/Meso	447.2	17	38.01		
	Meso/Nem	447.2	0.6	1.34	164.60	[36]
$\text{C}_{37}\text{H}_{67}\text{NO}_4$	3,4,5-tris(decyloxy)benzamide					
	Sol/Meso	331.2	9.8	29.59		
	Meso/Liq	358.2	6.1	17.03	46.62	[378]
$\text{C}_{38}\text{H}_{30}\text{Cl}_2\text{N}_2\text{O}_6$	4,5-dichloro-1,3-phenylene bis[4-(4-ethoxybenzylideneamino)benzoate]					
	Sol/Nem	471.2	38	80.65		
	Nem/Liq	502.2	2.3	4.58	85.23	[86]
$\text{C}_{38}\text{H}_{32}\text{N}_2\text{O}_6$	1,3-phenylene bis[4-(4-ethoxyphenyliminomethyl)benzoate]					
	Sol/Smec	467.2	36	77.05		
	Smec/Liq	476.2	14	29.40	106.45	[86]
$\text{C}_{38}\text{H}_{32}\text{N}_2\text{O}_6$	1,3-phenylene bis[4-(4-ethoxybenzylideneamino)benzoate]					
	Sol/Smec	450.2	26	57.75		
	Smec/Liq	461.2	5.4	11.71	69.46	[86]
$\text{C}_{38}\text{H}_{36}\text{N}_2$	3,8-bis[(4-pentylphenyl)ethynyl]-1,10-phenanthroline					
	Sol/Nem	431.2	28.0	64.94		
	Nem/Liq	Decomposed prior to transition				[258]
$\text{C}_{38}\text{H}_{36}\text{N}_2\text{O}_4$	3,3',4,4'-biphenyltetracarboxy-N,N'-bis(pentylphenyl)diimide					
	Sol/Sol	364.0	5.7	15.66		
	Sol/Sol	392.9	5.5	14.00		
	Sol/Smec	521.4	25.9	49.67		
	Smec/Liq	566.0	10.6	18.73	98.06	[94]
$\text{C}_{38}\text{H}_{36}\text{N}_2\text{O}_6$	3,3',4,4'-biphenyltetracarboxy-N,N'-bis(pentyloxyphenyl)diimide					
	Sol/Smec	523.5	25.2	48.14		
	Smec/Liq	586.1	9.9	16.89	65.03	[94]
$\text{C}_{38}\text{H}_{38}\text{FeO}_4$	[4-[[4'-(octyloxy)carbonyl][1,1'-biphenyl]-4-yloxy]carbonyl]phenyl]ferrocene					
	Sol/Nem	419.2	41.9	99.95		
	Nem/Liq	434.2	1.02	2.35	102.30	NA
$\text{C}_{38}\text{H}_{38}\text{N}_2\text{O}_4$	$\text{N},\text{N}'$ -diheptyl-3,4,9,10-perylenebis(carboxamide)					
	Sol/Meso	486.2	24.3	49.98		
	Meso/Meso	518.2	0.5	0.96		
	Meso/Meso	658.2	21.8	33.12		
	Meso/Liq	677.2	8.8	12.99	97.05	[263]
$\text{C}_{38}\text{H}_{39}\text{NO}_2$	[1-(4-decylbiphenyl)-3-(4"-cyanobiphenyl)]-propane-1,3-dione					
	Sol/Smec	455.2	34.64	76.10		
	Smec/Liq	477.2	2.13	4.46	80.56	[254]
$\text{C}_{38}\text{H}_{39}\text{NO}_3$	[1-(4-decyloxybiphenyl)-3-(4"-cyanobiphenyl)]-propane-1,3-dione					
	Sol/Smec	488.7	50.00	102.31		
	Smec/Liq	490.2	2.09	4.26	106.57	[254]
$\text{C}_{38}\text{H}_{39}\text{NO}_4$	4'-cyano[1,1'-biphenyl]-4-yloxy-3-[[4'-(decyloxy)[1,1'-biphenyl]-4-yloxy]-2-propenoate					

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$\text{C}_{38}\text{H}_{40}\text{N}_2\text{O}_6$	Sol/Smec	433.9	33.1	76.28	86.06	[305]
	Smec/Liq	460.2	4.5	9.78		
$\text{C}_{38}\text{H}_{40}\text{N}_2\text{O}_6$	$\alpha,\omega$ -bis[4-(4-propoxybenzoyloxy)benzylideneamino]butane					
	Sol/Nem	455.2	61.0	134.01		[293]
	Nem/Liq	478.2	7.8	16.31	150.32	
$\text{C}_{38}\text{H}_{40}\text{N}_2\text{O}_6$	$\alpha,\omega$ -bis[4-(4-ethoxybenzoyloxy)benzylideneamino]hexane					
	Sol/Nem	424.2	51.0	120.23		[293]
	Nem/Liq	475.2	8.8	18.52	138.75	
$\text{C}_{38}\text{H}_{42}\text{N}_2\text{O}_2\text{S}$	2,5-bis[[4-(octyloxy)phenyl]ethynyl]-3,4-thiophenedicarbonitrile					
	Sol/Sol	339.0	12.1	35.69		[373]
	Sol/Nem	372.8	31.7	85.03		
	Nem/Liq	400.6	1.2	3.00	123.72	
$\text{C}_{38}\text{H}_{42}\text{N}_4\text{O}_6$	$\alpha,\omega$ -bis(4-ethoxyazobenzene-4'-carbonyloxy)octane					
	Sol/Nem	402.2	47.2	117.35		[107]
	Nem/Liq	439.2	8.4	19.13	136.48	
$\text{C}_{38}\text{H}_{42}\text{N}_4\text{O}_6$	$\alpha,\omega$ -bis(4-butoxyazobenzene-4'-carbonyloxy)butane					
	Sol/Smec	418.3	36.8	87.98		[107]
	Smec/Nem	462.1	5.0	10.82		
$\text{C}_{38}\text{H}_{42}\text{N}_4\text{O}_6\text{S}_2$	bis[4-(5-n-heptylthio-1,3,4-oxadiazole-2-yl)phenyl] terephthalate					
	Sol/Smec	389.2	21.6	55.50		[23]
	Smec/Liq	554.2	5.5	9.92	65.42	
$\text{C}_{38}\text{H}_{42}\text{O}_6$	bis(4-hexyloxyphenyl) 4,4'-biphenylenedicarboxylate					
	Sol/Sol	375.4	18.2	48.48		[12]
	Sol/Smec	412.5	31.2	75.64		
	Smec/Nem	602.4	4.5	7.47		
	Nem/Liq	605.8	2.5	4.13	135.72	
$\text{C}_{38}\text{H}_{42}\text{O}_8$	benzoic acid, 4,4'-(1,4-phenylenebis[(1-oxo-2-propene-3,1-diyl)oxo])bis, dihexyl ester					
	Sol/Sol	327.6	12.0	36.63		[271]
	Sol/Smec	442.1	60.0	135.72		
	Smec/Smec	474.9	0.2	0.42		
	Smec/Nem	498.4	1.1	2.21		
$\text{C}_{38}\text{H}_{42}\text{O}_8$	1,4-benzeneddicarboxylic acid, bis[4-[3(hexyloxy)-3-oxo-1-propenyl]phenyl ester					
	Sol/Sol	318.8	3.8	11.92		[271]
	Sol/Smec	381.8	35.9	94.03		
	Smec/Liq	518.1	5.2	10.04	115.99	
$\text{C}_{38}\text{H}_{43}\text{N}_3\text{O}_6$	4-(4'-ethoxybenzoyloxy)-2-hexyloxy-4'-(4-butyoxysalicylaldimine)azobenzene					
	Sol/Sol	422.2	9.13	21.62		[109]
	Sol/Nem	426.2	35.43	83.13		
	Nem/Liq	530.2	2.03	3.83	108.58	
$\text{C}_{38}\text{H}_{44}\text{F}_2\text{O}_5$	1-methylheptyl 4'-(4-octyloxy-2',3"-difluorobenzoyloxy)tolan-4-carboxylate					
	Sol/Smec	347.2	29.89	86.09		[213]
	Smec/Liq	397.3	4.33	10.90	96.99	
$\text{C}_{38}\text{H}_{44}\text{N}_4\text{O}_6\text{S}_3$	bis[4-(5-n-octylthio-1,3,4-oxadiazole-2-yl)phenyl]-2,5-thiophenedicarboxylate					
	Sol/Smec	404.2	25.1	62.10		[23]
	Smec/Liq	497.2	8.2	16.49	78.59	
$\text{C}_{38}\text{H}_{45}\text{FO}_5$	1-methylheptyl 4'-(4-octyloxy-3-fluorobenzoyloxy)tolan-4-carboxylate					
	Sol/Smec	353.7	24.93	70.48		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{38}H_{45}FO_7S$	Smec/Liq	392.2	5.14	13.11	83.59	[213]
	Sol/Smec	342.4	40.70	118.87		
	Smec/Nem	345.4	0.80	2.32		
	Nem/Liq	351.8	0.69	1.68	122.87	[267]
$C_{38}H_{45}F_3O_7$	(S)-1-methylheptyl 4-[4-(4-octyloxy-3-fluorobenzoyloxy)-3-fluorobenzoyloxy]-2-fluorobenzoate					
	Sol/Smec	322.3	18.66	57.90		
	Smec/Smec	347.9	0.41	1.18		
$C_{38}H_{46}F_2O_7$	Smec/Liq	363.6	3.90	10.73	69.81	[80]
	(R)-1-methylheptyl 4-[4-(4-nonyloxy-3-fluorobenzoyloxy)-3-fluorobenzoyloxy] benzoate					
	Sol/Smec	338.4	37.67	111.32		
	Smec/Smec	364.0	0.26	0.71		
$C_{38}H_{46}N_4O_2$	Smec/Liq	376.8	4.70	12.47	124.50	[69]
	$\alpha$ -(4'-pentylazobenzene-4-oxy)- $\omega$ -(4'-pentylazobenzene-4-oxy)butane					
	Sol/Smec	391.2	29.60	75.66		
	Smec/Nem	491.2	2.61	5.31		
$C_{38}H_{46}N_4O_2$	Nem/Liq	493.2	8.65	17.54	98.51	[67]
	$\alpha$ -(4'-butylazobenzene-4-oxy)- $\omega$ -(4'-butylazobenzene-4-oxy)hexane					
	Sol/Smec	421.2	31.17	74.00		
	Smec/Nem	441.2	0.59	1.34		
$C_{38}H_{46}N_4O_2$	Nem/Liq	460.2	5.89	12.80	88.14	[67]
	$\alpha$ -(4'-hexylazobenzene-4-oxy)- $\omega$ -(4'-ethylazobenzene-4-oxy)hexane					
	Sol/Smec	403.2	36.20	89.78		
	Smec/Nem	438.2	0.33	0.75		
$C_{38}H_{46}N_4O_2$	Nem/Liq	456.2	5.99	13.13	103.66	[67]
	$\alpha$ -(4'-pentylazobenzene-4-oxy)- $\omega$ -(4'-propylazobenzene-4-oxy)hexane					
	Sol/Smec	409.2	28.92	70.67		
	Smec/Nem	447.2	Too small to be measured			
$C_{38}H_{46}O_2$	Nem/Liq	470.2	7.27	15.46	86.13	[67]
	1,4-phenylene bis[3-methyl-4-(heptyloxy)ethynylbenzene]					
	Sol/Nem	419.2	54.8	130.73		
	Nem/Liq	435.2	1.8	4.14	134.87	[82]
$C_{38}H_{46}O_2$	4,4"-diheptyloxy-p-quaterphenyl					
	Sol/Sol	380.0	14.82	39.00		
	Sol/Sol	454.0	2.27	5.00		
	Sol/Sol	463.0	6.95	15.01		
	Sol/Smec	580.0	13.34	23.00		
$C_{38}H_{46}O_5$	Smec/Liq	623.0	9.97	16.00	98.01	188.2
	7-[(4'-hexadecyloxy)benzoyloxy]isoflavone					
	Sol/Smec	406.6	61.53	151.33		
	Nem/Liq	464.3	8.20	17.66	168.99	[14]
$C_{38}H_{46}O_7$	(S)-(+)-1-methylheptyl 4-[4-( <i>trans</i> -4-heptyloxycinnamoyloxy)benzoyloxy]benzoate					
	Sol/Smec	365.7	32.17	87.97		
	Smec/Smec	406.2	0.03	0.07		
$C_{38}H_{48}N_2O_2$	Smec/Liq	445.2	4.85	10.89	98.93	[149]
	2,5-dihydro-2,5-dimethyl-3,6-bis[4-( <i>trans</i> -4-propylcyclohexyl)phenyl]pyrrolo[3,4-c]pyrrole-1,4-dione					
	Sol/Nem	464.2	23.3	50.19		
$C_{38}H_{48}O_5S$	Nem/Liq	631.2	1.7	2.69	52.88	[388]
	(S)-1-methylheptyl 5-[(4-decyloxybenzoyloxyphenyl)ethynyl]thiophene-2-carboxylate					

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated) (estimated)	Ref.			
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$					
$C_{38}H_{48}O_6S$	Sol/Smec	338.7	32.42	95.72	110.63	[267]			
	Smec/Liq	366.3	5.46	14.91					
$C_{38}H_{48}O_7$	4-[[4-(nonyloxy)benzoyl]thio]benzoic acid, 4-[[[(1-methylheptyl)oxy]carbonyl]phenyl ester					[85]			
	Sol/Smec	374.5	40.1	107.08	121.46				
	Smec/Smec	388.7	0.06	0.15					
	Smec/Liq	428.6	6.1	14.23					
$C_{38}H_{48}O_7$	2-acetyl-1,4-phenylene 4-(octyloxy)benzoate					[432]			
	Sol/Nem	374.8	39.6	105.66	110.19				
	Nem/Liq	397.4	1.8	4.53					
$C_{38}H_{49}F_{13}O_7$	3,3'-[[3-[(6,6,7,7,8,8,9,9,10,10,11,11,11-tridecafluoro-2-nonylundecyl)oxy]oxy]-[1,1'-biphenyl]-4,4'-diyl]bis(oxy)-bis(1,2-propanediol)					[374]			
	Sol/Col	339.2	21.6	63.68	81.74				
	Col/Liq	404.2	7.3	18.06					
	3-hydroxy-4-[(E)-[(4-dodecylphenyl)imino]methyl]phenyl 4-butoxy- $\alpha$ -chlorobenzenepropanoate								
$C_{38}H_{50}ClNO_4$	Sol/Smec	331.2	Not reported in paper			[249]			
	Smec/Smec	332.0	3.38	10.18	15.93				
	Smec/Smec	362.7	0.24	0.66					
	Smec/Liq	387.2	6.17	15.93					
$C_{38}H_{50}O_3$	1-(4'-dodecylbiphenyl-4-yl)-3-(2-pentyloxyphenyl)propane-1,3-dione					[297]			
	Sol/Nem	326.7	40.58	124.21	125.45				
	Nem/Liq	338.2	0.42	1.24					
$C_{38}H_{50}O_3$	1-(4'-dodecylbiphenyl-4-yl)-3-(3-pentyloxyphenyl)propane-1,3-dione					[297]			
	Sol/Sol	351.2	20.29	57.77	161.51				
	Sol/Smec	363.2	30.37	83.62					
	Smec/Liq	372.2	7.49	20.12					
$C_{38}H_{50}O_5$	(R)-4-(1-methylheptyloxy carbonyl)phenyl 4-(4-octyloxyphenyl)ethylbenzoate					[184]			
	Sol/Smec	327.2	25.0	76.41	101.7				
	Smec/Smec	350.2	0.10	0.29					
	Smec/Liq	364.4	9.10	24.97					
$C_{38}H_{50}O_5S$	(R)-4-(3-ethylmercapto-2-methylpropionyl)phenyl 4'-tridecyloxybiphenyl-4-carboxylate					[100]			
	Sol/Smec	359.5	37.32	103.81	112.45				
	Smec/Smec	413.2	0.06	0.15					
	Smec/Liq	424.7	3.58	8.49					
$C_{38}H_{50}O_6$	4-[(1R)-2-(1-ethylpropoxy)-1-methyl-2-oxoethoxy]phenyl 4'-(undecyloxy)[1,1'-biphenyl]-4-carboxylate					[239]			
	Sol/Smec	306.1	49.90	163.02	165.38				
	Smec/Meso	380.0	0.47	1.24					
	Meso/Nem	381.8	Too small to be measured						
	Nem/Meso	383.7	Too small to be measured						
	Meso/Liq	384.6	0.43	1.12					
$C_{38}H_{50}O_7$	4-(4-butoxybenzoyloxy)phenyl (3,4-diheptyloxy)benzoate					[103]			
	Sol/Nem	360.5	9.59	26.60	217.7				
	Nem/Liq	383.4	0.33	0.86					
$C_{38}H_{50}O_7$	4-(4-octyloxybenzoyloxy)phenyl (3,4-dipentyloxy)benzoate					[103]			
	Sol/Nem	365.4	9.94	27.20	217.7				
	Nem/Liq	384.2	0.45	1.17					
$C_{38}H_{52}N_2O_3$	4-[5-[(S)-6-methyloctyl]-2-pyrimidinyl]phenyl 4-(11-dodecenoxy)benzoate					[201]			
	Sol/Smec	338.2	37.2	109.99	Not reported in paper				
	Smec/Chol	357.2	Not reported in paper						
	Chol/Liq	400.2	Not reported in paper						

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{38}H_{52}O_3S$	Sol/Smec	329.7	20.9	63.39		[281]
	Smec/Nem	379.3	2.8	7.38		
	Nem/Liq	397.1	1.4	3.53	74.30	
$C_{38}H_{52}O_5$	Sol/Sol	388.5	48.52	124.89		[304]
	Smec/Smec	389.0	2.47	6.35		
	Smec/Liq	427.6	28.56	66.79	198.03	
$C_{38}H_{52}O_6S$	Sol/Sol	391.5	11.4	29.12		[12]
	Smec/Smec	397.6	60.4	151.91		
	Smec/Liq	410.5	10.6	25.82	206.85	
$C_{38}H_{53}NO_2$	Sol/Smec	356.2	55.56	155.98		[323]
	Smec/Smec	416.2	2.01	4.83		
	Smec/Smec	483.2	Too small to be measured			
	Smec/Liq	485.7	8.62	17.75	178.56	
$C_{38}H_{53}NO_2$	Sol/Smec	344.4	42.24	122.65		[139]
	Smec/Liq	437.7	10.51	24.01	146.66	
	Sol/Smec	349.4	8.84	25.30		
$C_{38}H_{53}NO_4$	Smec/Smec	399.9	3.29	8.23		[427]
	Smec/Liq	410.9	1.42	3.46	36.99	
	Sol/Smec	335.2	39.8	118.74		
$C_{38}H_{54}N_2O_2S$	Smec/Nem	353.2	0.3	0.85		[398]
	Nem/Liq	361.2	0.82	2.27	121.86	
	Sol/Smec	342.2	30.0	87.67		
$C_{38}H_{54}N_2O_3$	Smec/Chol	367.2	Not reported in paper			[201]
	Chol/Liq	405.2	Not reported in paper			
	Sol/Disc	360.2	10.0	27.76		
$C_{38}H_{54}N_2O_6$	Disc/Liq	395.2	18.0	45.55	73.31	[189]
	Sol/Smec	383.2	10.4	27.14		
	Smec/Smec	397.2	4.1	10.32		
$C_{38}H_{56}N_2O_3$	Col/Smec	402.2	5.5	13.67		[231]
	Smec/Liq	405.2	3.8	9.38	60.51	
	Sol/Smec	329.3	51.3	155.78		
$C_{38}H_{56}O_4$	Smec/Smec	392.2	4.82	12.29		[215]
	Smec/Liq	412.1	8.00	19.41	187.48	
	Sol/Smec	352.6	30.12	85.42		
$C_{38}H_{57}NOS$	Smec/Smec	440.0	4.95	11.25	96.67	[131]
	Smec/Liq	402.2	53.56	133.17		
	Sol/Smec	395.7	46.60	117.77		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{38}H_{58}O_2$	Smec/Liq	494.3	11.49	23.24	274.18	[269]
	Sol/Liq	371.5	31.38	84.47	84.47	135.5
	Liq/Chol	364.5	0.63	1.7		[155, 169]
Note: Enthalpy of the Liq/Chol transition is from cooling measurements.						
$C_{38}H_{58}O_3$	cholesteryl $\omega$ -phenylpentanoate					
	Sol/Smec	331.2	23.84	71.98		
	Smec/Smec	351.2	Not reported in paper			
	Smec/Smec	391.2	Not reported in paper			
	Smec/Nem	405.2	Not reported in paper			
$C_{38}H_{61}NO$	Nem/Liq	420.2	Not reported in paper			[199, 207]
	4-(dodecyloxy)benzoic acid, 4-[2-( <i>trans</i> -4-pentylcyclohexyl)ethyl]phenyl ester					
	Sol/Smec	343.8	61.20	178.01		
$C_{38}H_{61}NO$	Smec/Liq	359.7	20.47	56.91	234.92	[240]
	4-decyl-N-[[4-(pentadecyloxy)phenyl]methylene]benzenamine					
	Sol/Smec	338.5	33.48	98.91		
$C_{38}H_{64}O_6$	Smec/Liq	360.3	16.37	45.43	144.34	[240]
	(S)-4-(2-ethylheptyloxycarbonyl)phenyl 4-heptadecanoyloxytolane-4'-carboxylate					
	Sol/Smec	354.5	54.6	154.02		
$C_{38}H_{66}OS$	Smec/Liq	369.2	3.5	9.48	163.50	[173]
	cholesteryl thioundecanoate					
	Sol/Smec	354.2	26.2	73.8		
$C_{38}H_{66}O_2$	Smec/Chol	361.3	1.1	1.0		
	Chol/Liq	368.2	0.6	1.7	78.4	NA
	cholesteryl undecanoate					
$C_{38}H_{66}O_2$	Sol/Liq	364.65	26.4	72.4	72.4	161.2
	Chol/Smec	351.85	1.55	4.4		
	Smec/Liq	361.05	1.34	3.7		[170]
Note: Liquid crystalline phase detected upon cooling.						
$C_{38}H_{67}BrO$	3-[(11-bromoundecyl)oxy'-cholest-5-ene					
	Sol/Nem	345.9	Not reported in paper			
	Nem/Liq	348.4	22.59	58.77	58.77	[323]
Note: Sol/Nem enthalpy is included in the Nem/Liq value.						
$C_{38}H_{68}N_2O_2$	N,N'-ditetradecanoyl-2,3,5,6-tetramethylbenzene-1,4-diamine					
	Sol/Meso	414.2	52.0	125.54		
	Meso/Meso	508.2	11.0	21.65		
$C_{38}H_{68}N_2O_2$	Meso/Liq	514.2	12.0	23.33	170.52	[36]
	N,N'-ditetradecanoyl-3,4,5,6-tetramethylbenzene-1,2-diamine					
	Sol/Meso	396.2	18.0	45.43		
$C_{38}H_{68}N_2O_2$	Meso/Meso	441.2	1.0	2.27		
	Meso/Liq	498.2	21.0	42.15	89.85	[61]
	1,1,2,2,3,3,4,4,5,5,6,6,7,7,8,8,9,9,10,10,11,11,12,12,13,13-tetracoso-methyl-1,13-diphenyltridecasilane					
$C_{38}H_{88}Si_{13}$	Sol/Meso	400.2	37.7	94.20		
	Meso/Liq	428.2	26.1	60.95	155.55	[400]
$C_{39}H_{38}N_2$	2-methyl-3,8-bis[(4-pentylphenyl)ethynyl]-1,10-phenanthroline					
	Sol/Nem	439.2	21.5	48.95		
$C_{39}H_{38}N_2$	Nem/Liq	Decomposed prior to transition				[258]
$C_{39}H_{40}FeO_4$	[4-[[4'-(nonyloxy)carbonyl][1,1'-biphenyl]-4-yl]oxy]carbonylphenylferrocene					
	Sol/Nem	400.2	33.9	84.71		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$\text{C}_{39}\text{H}_{41}\text{NO}_2$	Nem/Liq	430.2	1.12	2.60	87.31	NA
	Sol/Smec	453.2	28.70	63.33		[375]
	Smec/Liq	480.2	3.22	6.71	70.04	[254]
$\text{C}_{39}\text{H}_{41}\text{NO}_3$	[1-(4-undecylbiphenyl)-3-(4"-cyanobiphenyl)]-propane-1,3-dione					
	Sol/Smec	485.7	35.94	74.00		[254]
	Smec/Liq	493.2	3.26	6.61	80.61	
$\text{C}_{39}\text{H}_{42}\text{N}_2\text{O}_6$	$\alpha,\omega$ -bis[4-(4-ethoxybenzoyloxy)benzylideneamino]heptane					
	Sol/Smec	367.2	40.0	108.93		
	Smec/Nem	379.2	0.1	0.26		
	Nem/Liq	403.2	0.5	1.24	110.43	[293]
$\text{C}_{39}\text{H}_{44}\text{F}_4\text{O}_5$	4-[(S)-2-methylbutoxycarbonyl]phenyl 4-[(4-dodecyloxy-2,3,4,5-tetrafluorophenyl)ethynyl]benzoate					
	Sol/Smec	346.1	34.15	98.67		
	Smec/Smec	354.8	Not detected by dsc			
	Smec/Chol	400.6	1.44	3.59		
	Chol/Liq	405.3	0.77	1.90	104.19	[91]
$\text{C}_{39}\text{H}_{44}\text{N}_4\text{O}_6$	$\alpha,\omega$ -bis(4-butoxyazobenzene-4'-carbonyloxy)pentane					
	Sol/Smec	353.0	18.0	50.99		
	Smec/Liq	410.1	13.0	31.70	82.69	[107]
$\text{C}_{39}\text{H}_{46}\text{F}_2\text{O}_5$	1-methylheptyl 4'-(4"-nonyloxy-2",3"-difluorobenzoyloxy)tolan-4-carboxylate					
	Sol/Smec	342.4	38.85	113.46		
	Smec/Liq	394.2	4.25	10.78	124.24	[213]
$\text{C}_{39}\text{H}_{47}\text{FO}_5$	1-methylheptyl 4'-(4"-nonyloxy-3"-fluorobenzoyloxy)tolan-4-carboxylate					
	Sol/Smec	353.4	27.24	77.08		
	Smec/Liq	388.9	4.98	12.81	89.89	[213]
$\text{C}_{39}\text{H}_{47}\text{FO}_5$	[(R)-1-methylheptyl] 4-[[2-fluoro-4-[[4-(nonyloxy)benzoyl]oxy]phenyl]ethynyl]benzoate					
	Sol/Smec	343.0	24.44	71.25		
	Smec/Liq	393.3	5.07	12.89	84.14	[211]
$\text{C}_{39}\text{H}_{47}\text{FO}_7\text{S}$	1-(butoxycarbonyl)ethyl 5-[(3-fluoro-4-dodecyloxybenzoylphenyl)ethynyl]thiophene-2-carboxylate					
	Sol/Meso	349.1	42.22	120.94		
	Meso/Liq	352.2	0.56	1.59	122.53	[267]
$\text{C}_{39}\text{H}_{47}\text{F}_3\text{O}_7$	(S) 1-ethylheptyl 4-[4-(4-nonyloxy-3-fluorobenzoyloxy)-3-fluorobenzoyloxy]-2-fluorobenzoate					
	Sol/Smec	335.6	15.10	44.99		
	Smec/Smec	346.8	0.55	1.59		
	Smec/Liq	360.2	3.72	10.33	56.91	[80]
$\text{C}_{39}\text{H}_{48}\text{F}_2\text{O}_7$	(R)-1-methylheptyl 4-[4-(4-decyloxy-3-fluorobenzoyloxy)-3-fluorobenzoyloxy] benzoate					
	Sol/Smec	330.9	31.80	96.10		
	Smec/Smec	367.1	0.37	1.01		
	Smec/Liq	376.9	4.97	13.19	110.30	[69]
$\text{C}_{39}\text{H}_{48}\text{N}_4\text{O}_2$	$\alpha$ -(4'-hexylazobenzene-4-oxy)- $\omega$ -(4'-propylazobenzene-4-oxy)hexane					
	Sol/Smec	398.2	37.74	94.78		
	Smec/Nem	452.2	0.79	1.75		
	Nem/Liq	464.2	6.68	14.39	110.92	[67]
$\text{C}_{39}\text{H}_{48}\text{N}_4\text{O}_2$	$\alpha$ -(4'-pentylazobenzene-4-oxy)- $\omega$ -(4'-butylazobenzene-4-oxy)hexane					
	Sol/Smec	414.2	30.65	74.00		
	Smec/Nem	453.2	1.09	2.41		
	Nem/Liq	464.2	7.18	15.47	91.88	[67]
$\text{C}_{39}\text{H}_{48}\text{N}_4\text{O}_2$		$\alpha$ -(4'-butylazobenzene-4-oxy)- $\omega$ -(4'-butylazobenzene-4-oxy)heptane				

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{39}H_{48}N_4O_2$	Sol/Smec	384.2	39.6	103.07		[67]
	Smec/Nem	394.2	Too small to be measured			
	Nem/Liq	415.2	1.62	3.90	106.97	
$C_{39}H_{48}O_7$	$\alpha$ -(4'-pentylazobenzene-4-oxy)- $\omega$ -(4'-pentylazobenzene-4-oxy)pentane					
	Sol/Smec	388.2	35.18	90.62		[67]
	Smec/Nem	412.2	1.68	4.08		
	Nem/Liq	420.2	1.68	4.00	98.70	
$C_{39}H_{48}O_7$	(S)-(+)-1-methylheptyl 4-[4-(trans-4-octyloxycinnamoyloxy)benzoyloxy]benzoate					
	Sol/Smec	352.7	41.95	118.94		[149]
	Smec/Smec	415.2	0.12	0.29		
	Smec/Liq	441.7	4.45	10.07	129.30	
$C_{39}H_{50}O_5S$	(S)-1-methylheptyl 5-[(4-undecyloxybenzoyloxyphenyl)ethynyl]thiophene-2-carboxylate					
	Sol/Smec	341.4	32.81	96.10		[267]
	Smec/Liq	363.4	4.97	13.68	109.78	
$C_{39}H_{50}O_6S$	4-[[4-(dodecyloxy)benzoyl]thio]benzoic acid, 4-[[1-methylheptyl)oxy]carbonyl]phenyl ester					
	Sol/Smec	382.9	45.5	118.83		[85]
	Smec/Smec	396.8	0.06	0.15		
	Smec/Liq	425.8	6.2	14.56	133.54	
$C_{39}H_{50}O_7$	(1S)-2-butoxy-1-methyl-2-oxoethyl 4-[[2(E)-3-[6-dodecyloxy)-2-naphthalenyl]-1-oxo-2-propenyl]oxy]benzoate					
	Sol/Smec	343.2	37.06	107.98		[401]
	Smec/Smec	354.4	0.39	1.10		
	Smec/Liq	384.7	3.80	9.88	118.96	
$C_{39}H_{50}O_7$	2-(1-oxyheptyl)-1,4-phenylene 4-(hexyloxy)benzoate					
	Sol/Nem	354.7	49.5	139.55		[432]
	Nem/Liq	361.0	0.96	2.66	142.21	
$C_{39}H_{50}O_7$	2-(1-oxopropyl)-1,4-phenylene 4-(octyloxy)benzoate					
	Sol/Nem	358.8	54.3	151.34		[432]
	Nem/Liq	375.6	1.6	4.26	155.60	
$C_{39}H_{52}O_3$	1-(4'-dodecylbiphenyl-4-yl)-3-(2-hexyloxyphenyl)propane-1,3-dione					
	Sol/Nem	324.7	43.01	132.46		[297]
	Nem/Liq	340.7	0.42	1.23	133.69	
$C_{39}H_{52}O_3$	1-(4'-dodecylbiphenyl-4-yl)-3-(3-hexyloxyphenyl)propane-1,3-dione					
	Sol/Sol	352.2	19.70	55.93		[297]
	Smec/Smec	358.7	24.52	68.36		
$C_{39}H_{52}O_5$	Smec/Liq	372.7	8.16	21.89	146.18	
	(R)-4-(1-methylheptyloxy carbonyl)phenyl 4-(4-nonyloxyphenyl)ethylbenzoate					
	Sol/Smec	328.2	27.0	82.27		[184]
$C_{39}H_{52}O_5S$	Smec/Smec	355.5	0.17	0.48		
	Smec/Liq	362.2	8.9	24.57	107.3	
	(R)-4-(3-ethylmercapto-2-methylpropionyl)phenyl 4'-tetradecyloxybiphenyl-4-carboxylate					
$C_{39}H_{52}O_5S$	Sol/Smec	359.7	40.30	112.04		[100]
	Smec/Smec	413.1	0.03	0.07		
	Smec/Liq	424.1	4.06	9.57	121.68	
$C_{39}H_{52}O_6$	4-[(1R)-2-(1-ethylpropoxy)-1-methyl-2-oxoethoxy]phenyl 4'-(dodecyloxy)[1,1'-biphenyl]-4-carboxylate					
	Sol/Smec	308.0	59.20	192.21		[239]
	Smec/Meso	379.1	0.35	0.92		
	Meso/Nem	380.9	Too small to be measured			
	Nem/Meso	381.8	Too small to be measured			
	Meso/Liq	382.8	0.03	0.08	193.21	

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{39}H_{53}NO_4$	Sol/SmeC	330.8	52.75	159.46	[175]	
	SmeC/Liq	333.7	3.39	10.15		
$C_{39}H_{54}O_3S$	Sol/SmeC	334.5	23.5	70.25	[281]	
	SmeC/Nem	379.3	3.4	8.96		
	Nem/Liq	395.4	1.4	3.54		
$C_{39}H_{55}NO_2$	Sol/Sol	344.7	N-(2-hydroxy-4-octyloxybenzylidene)-4"-dodecylphenylaniline Not reported in paper		[323]	
	Sol/SmeC	349.7	16.95	48.47		
	SmeC/SmeC	414.7	2.72	6.56		
	SmeC/Liq	481.7	9.16	19.02		
$C_{39}H_{60}N_2O_3$	Sol/Sol	395.2	6.0	15.18	[251]	
	Sol/SmeC	423.2	38.1	90.03		
	SmeC/Liq	431.2	7.7	17.86		
$C_{39}H_{60}N_2O_3$	Sol/Sol	405.2	3.8	9.38	[251]	
	Sol/SmeC	418.2	39.7	94.93		
	SmeC/Liq	431.2	8.5	19.71		
$C_{39}H_{60}N_2O_3$	Sol/SmeC	424.2	46.0	108.44	[251]	
	SmeC/Liq	429.2	8.0	18.64		
	Sol/Sol	424.2	46.0	108.44		
$C_{39}H_{60}O_2$	Sol/Liq	354.5	27.74	78.25	142.6	[155, 169]
	Liq/Chol	318.4	0.29	0.9		
Note: Enthalpy of the Liq/Chol transition is from cooling measurements.						
$C_{39}H_{63}NO$	Sol/SmeC	339.7	25.00	73.58	[240]	
	SmeC/Liq	358.8	20.71	57.72		
$C_{39}H_{63}NO$	Sol/SmeC	347.2	69.85	201.18	[256]	
	SmeC/Liq	363.1	19.68	54.20		
$C_{39}H_{68}OS$	Sol/SmeC	357.2	34.9	97.6	[155,312]	
	SmeC/Chol	359.9	1.5	4.2		
	Chol/Liq	365.5	0.8	2.2		
$C_{39}H_{68}O_2$	Sol/Liq	364.45	31.8	87.4	168.3	[170]
	Chol/SmeC	353.35	0.95	2.7		
	SmeC/Liq	360.35	0.74	2.0		
Note: Liquid crystalline phase detected upon cooling.						
$C_{39}H_{68}N_2O_4$	Sol/Meso	381.2	19.0	49.84	[193]	
	Meso/Meso	432.2	21.0	48.59		
	Meso/Liq	468.2	3.0	6.41		
$C_{39}H_{69}N_3O_3$	Sol/Meso	462.2	5.0	10.82		
	Meso/Meso	479.2	16.0	33.39		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$\text{C}_{39}\text{H}_{70}\text{O}_2$	Meso/Liq	615.2	15.0	31.30	75.51	[190]
	Sol/Liq	360.65	57.3	159.0	159.0	167.5
	Smec/Chol	331.15	3.50	1.1		
	Chol/Liq	340.15	1.71	5.0		[311]
Note: Liquid crystalline phase detected upon cooling.						
$\text{C}_{40}\text{H}_{34}\text{Cl}_2\text{N}_2\text{O}_6$	4,5-dichloro-1,3-phenylene bis[4-(4-propoxybenzylideneamino)benzoate]					
	Sol/Nem	427.2	36.0	84.27		
	Nem/Liq	471.2	2.1	4.46	88.73	[86]
$\text{C}_{40}\text{H}_{40}\text{N}_2\text{O}_4$	3,3',4,4'-biphenyltetracarboxy-N,N'-bis(hexylphenyl)diimide					
	Sol/Sol	432.4	19.9	46.02		
	Sol/Smec	513.8	26.2	50.99		
	Smec/Liq	563.3	9.5	16.86	113.87	[94]
$\text{C}_{40}\text{H}_{40}\text{N}_2\text{O}_6$	3,3',4,4'-biphenyltetracarboxy-N,N'-bis(hexyloxyphenyl)diimide					
	Sol/Sol	479.1	12.7	26.51		
	Sol/Sol	496.7	1.3	2.62		
	Smec/Smec	517.0	28.2	54.55		
	Smec/Liq	586.4	12.2	20.80	104.48	[94]
$\text{C}_{40}\text{H}_{42}\text{FeO}_4$	[4-[[4'-(decyloxy)carbonyl][1,1'-biphenyl]-4-yl]oxy]carbonyl]phenyl]ferrocene					
	Sol/Nem	408.2	48.8	119.55		
	Nem/Liq	428.2	0.21	0.49	120.04	NA
$\text{C}_{40}\text{H}_{42}\text{O}_9$	4-[[[(1S)-2-butoxy-1-methyl-2-oxoethoxy]carbonyl]phenyl] 4-[[[(2E)-3-[6-(hexyloxy)-2-naphthyl]-1-oxo-2-propenyl]oxy]benzoate					
	Smec/Smec	271.7	20.97	77.18		
	Smec/Smec	382.2	Not reported in paper			
	Smec/Liq	503.6	3.26	6.47	83.65	[401]
$\text{C}_{40}\text{H}_{43}\text{NO}_2$	[1-(4-dodecylbiphenyl)-3-(4"-cyanobiphenyl)]-propane-1,3-dione					
	Smec/Smec	452.7	28.03	61.92		
	Smec/Liq	482.7	4.31	8.93	70.85	[254]
$\text{C}_{40}\text{H}_{43}\text{NO}_3$	[1-(4-dodecyloxybiphenyl)-3-(4"-cyanobiphenyl)]-propane-1,3-dione					
	Smec/Smec	483.2	52.34	108.32		
	Smec/Liq	496.2	4.94	9.96	118.28	[254]
$\text{C}_{40}\text{H}_{44}\text{N}_2\text{O}_2$	3,6-bis[(4'-butyl)[1,1'-biphenyl]-4-yl]-2,5-dihydro-2,5-dimethylpyrrolo[3,4-c]pyrrole-1,4-dione					
	Sol/Nem	493.2	4.2	8.52		
	Nem/Liq	603.2	0.1	0.17		[388]
Note: Experimental values are extremely small. The values are either in error or there is an unmeasured Sol/Sol transition at a lower temperature.						
$\text{C}_{40}\text{H}_{44}\text{N}_2\text{O}_6$	$\alpha,\omega$ -bis[4-(4-butoxybenzoyloxy)benzylideneamino]butane					
	Sol/Nem	442.2	55.0	124.38		
	Nem/Liq	474.2	8.8	18.56	142.94	[293]
$\text{C}_{40}\text{H}_{44}\text{N}_2\text{O}_6$	$\alpha,\omega$ -bis[4-(4-propoxybenzoyloxy)benzylideneamino]hexane					
	Sol/Nem	413.2	51.0	123.43		
	Nem/Liq	453.2	7.9	17.43	140.86	[293]
$\text{C}_{40}\text{H}_{44}\text{N}_2\text{O}_6$	$\alpha,\omega$ -bis[4-(4-ethoxybenzoyloxy)benzylideneamino]octane					
	Sol/Nem	428.2	47.0	109.76		
	Nem/Liq	457.2	8.8	19.25	129.01	[293]
$\text{C}_{40}\text{H}_{44}\text{O}_{13}$	1,4,7,10,13,20,28,31,34,37,45-decaoxa-21,44-dioxo[13.2.10.2]paracyclophane					
	Sol/Nem	418.2	55.3	132.23		
	Nem/Liq	445.2	1.8	4.04	136.27	[113]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{40}H_{46}N_4O_6$	$\alpha,\omega$ -bis(4-butoxyazobenzene-4'-carbonyloxy)hexane					
	Sol/Smec	395.9	41.4	104.57		
	Smec/Nem	439.1	4.6	10.48		
$C_{40}H_{46}N_4O_6S_2$	Nem/Liq	442.5	9.1	20.56	135.61	[107]
	bis[4-(5-n-octylthio-1,3,4-oxadiazole-2-yl)phenyl]terephthalate					
	Sol/Smec	389.2	49.2	126.41		
$C_{40}H_{46}O_6$	Smec/Liq	548.2	10.2	18.61	145.02	[23]
	bis(4-heptyloxyphenyl) 4,4'-biphenylenedicarboxylate					
	Sol/Sol	388.8	28.4	73.05		
	Sol/Smec	416.2	29.6	71.12		
	Smec/Smec	484.2	0.8	1.65		
$C_{40}H_{46}O_8$	Smec/Nem	591.4	2.6	4.40		
	Nem/Liq	592.7	4.7	7.93	158.15	203.6
	benzoic acid, 4,4'-[1,4-phenylenebis[(1-oxo-2-propene-3,1-diyl)oxo]]bis, diheptyl ester					
$C_{40}H_{46}O_8$	Sol/Smec	436.3	59.1	135.46		
	Smec/Smec	482.4	0.2	0.41		
	Smec/Nem	502.0	1.1	2.19		
	Nem/Liq	506.7	1.1	2.17	140.23	[271]
$C_{40}H_{46}O_8$	1,4-benzeneddicarboxylic acid, bis[4-[3-(heptyloxy)]-3-oxo-1-propenyl]phenyl ester					
	Sol/Smec	381.7	38.1	99.82		
	Smec/Liq	511.0	5.8	11.35	111.17	[271]
$C_{40}H_{47}N_3O_6$	4-(4'-ethoxybenzoyloxy)-2-octyloxy-4'-(4-butoxysalicylaldimine)azobenzene					
	Sol/Sol	372.2	3.40	9.13		
	Sol/Nem	403.2	44.92	111.41		
	Nem/Liq	520.2	1.69	3.25	123.79	[109]
$C_{40}H_{48}F_2O_5$	1-methylheptyl 4'-(4"-decyloxy-2",3"-difluorobenzoyloxy)tolan-4-carboxylate					
	Sol/Smec	347.1	32.47	93.55		
	Smec/Liq	394.6	4.52	11.45	105.00	[213]
$C_{40}H_{48}N_2$	3,8-bis(4-octylphenyl)-1,10-phenanthroline					
	Sol/Smec	395.2	12.9	32.64		
	Smec/Liq	545.2	4.8	8.80	41.44	[258]
$C_{40}H_{48}N_2O_6$	bis(4-octyloxyphenyl) 2,2'-bipyridine-5,5'-dicarboxylate					
	Sol/Sol	408.2	18.9	46.30		
	Sol/Smec	428.2	30.5	71.23		
	Smec/Nem	553.2	Not detected by dsc			
$C_{40}H_{48}N_2O_6S_3$	Nem/Liq	563.2	1.23	2.18	119.71	[93,115]
	bis[4-(5-nonylthio-1,3,4-oxadiazole-2-yl)phenyl]-2,5-thiophene-dicarboxylate					
	Sol/Smec	403.2	26.3	65.23		
$C_{40}H_{49}FO_5$	Smec/Liq	493.2	8.5	17.23	82.46	[23]
	[(R)-1-methylheptyl] 4-[[2-fluoro-4-[[4-(decyloxy)benzoyl]oxy]phenyl]ethynyl]benzoate					
	Sol/Smec	338.3	26.81	79.25		
$C_{40}H_{49}FO_5$	Smec/Liq	392.8	5.32	13.54	92.79	[211]
	1-methylheptyl 4'-(4"-decyloxy-3"-fluorobenzoyloxy)tolan-4-carboxylate					
	Sol/Smec	366.2	30.56	83.45		
$C_{40}H_{49}FO_5$	Smec/Liq	388.7	5.16	13.28	96.73	[213]
$C_{40}H_{49}FO_7S$	1-(butoxycarbonyl)ethyl 5-[(3-fluoro-4-tridecyloxybenzoylphenyl)ethynyl]thiophene-2-carboxylate					
	Sol/Meso	347.4	41.73	120.12		
	Meso/Liq	349.6	0.43	1.23	121.35	[267]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{40}H_{49}F_3O_7$	Sol/Smec	(S) 1-ethylheptyl 4-[4-(4-decyloxy-3-fluorobenzoyloxy)-3-fluorobenzoyloxy]-2-fluorobenzoate				
		344.5	20.28	58.87		
		349.9	0.71	2.03		
$C_{40}H_{50}$	Smec/Liq	358.8	3.51	9.78	70.68	[80]
		1,8-bis[4-4'-(butylbiphenyl)]octane				
		398.2	13.0	32.6		
$C_{40}H_{50}F_2O_5S$	Smec/Liq	414.2	27.0	65.2	97.8	[165]
		(S)-1-methylheptyl 5-[(2,3-difluoro-4-dodecyloxybenzoyloxyphenyl)ethynyl]-thiophene-2-carboxylate				
		333.1	41.19	123.66		
$C_{40}H_{50}F_2O_7$	Smec/Liq	355.6	4.52	12.71	136.37	[267]
		(R)-1-methylheptyl 4-[4-(4-undecyloxy-3-fluorobenzoyloxy)-3-fluorobenzoyloxy] benzoate				
		342.5	35.19	102.74		
$C_{40}H_{50}N_2O_2S$	Smec/Nem	368.5	0.36	0.98		
		374.9	4.96	13.23	116.95	[69]
		338.7	20.9	61.71		
$C_{40}H_{50}N_4O_2$	Nem/Liq	368.7	0.8	2.17		
		372.0	2.5	6.72	70.60	[373]
		2-[4-(decyloxy)phenyl]-5-[[4-(decyloxy)phenyl]ethynyl]-3,4-thiophene-dicarbonitrile				
$C_{40}H_{50}N_4O_2$	Sol/Smec	399.2	39.83	99.77		
		455.2	1.89	4.15		
		458.2	7.16	15.63	119.55	[67]
$C_{40}H_{50}N_4O_2$	Smec/Nem	419.2	33.46	79.82		
		460.2	1.65	3.59		
		465.2	8.01	17.22	100.63	[67]
$C_{40}H_{50}O_2$	Smec/Liq	$\alpha$ -(4'-hexylazobenzene-4-oxy)- $\omega$ -(4'-butylazobenzene-4-oxy)hexane				
		353.0	3.53	10.00		
		481.0	3.84	7.98		
		575.0	13.8	24.00		
$C_{40}H_{50}O_5$	Smec/Liq	618.0	11.75	19.01	60.99	[111]
		4,4'''-dioctyloxy-p-quaterphenyl				
		397.8	46.23	116.21		
$C_{40}H_{50}O_7$	Smec/Liq	461.2	5.46	11.84	128.05	[14]
		(S)-(+)-1-methylheptyl 4-[4-(trans-4-nonyloxy)cinnamoyloxy]benzoxyloxy]benzoate				
		354.2	26.00	73.40		
$C_{40}H_{50}O_7$	Smec/Liq	418.7	0.22	0.53		
		437.7	4.1	9.37	83.30	[149]
		392.7	5.58	14.21	113.49	[149]
$C_{40}H_{51}FO_5S$	Smec/Liq	(S)-(+)-1-methylheptyl 4-[4-(trans-4-octyloxy- $\alpha$ -methylcinnamoyloxy)benzoxyloxy]benzoate				
		331.7	32.93	99.28		
		392.7	5.58	14.21	113.49	[149]
$C_{40}H_{52}O_2$	Smec/Liq	(S)-1-methylheptyl 5-[(3-fluoro-4-dodecyloxybenzoyloxyphenyl)ethynyl]-thiophene-2-carboxylate				
		332.4	35.55	106.95		
		347.3	4.82	13.88	120.83	[267]
$C_{40}H_{52}O_2$	Smec/Liq	2-decyloxy-6-[4-(4-decyloxyphenyl)buta-1,3-diynyl]naphthalene				
		349.2	23.8	68.16		
		359.1	0.7	1.95		
$C_{40}H_{52}O_5S$	Smec/Liq	426.3	2.4	5.63	75.74	[225]
		(S)-1-methylheptyl 5-[(4-dodecyloxybenzoyloxyphenyl)ethynyl]-thiophene-2-carboxylate				

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.	
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$			
$\text{C}_{40}\text{H}_{52}\text{O}_6\text{S}$	Sol/Smec	336.1	53.32	158.64	173.36	[267]	
	Smec/Smec	348.3	0.10	0.29			
	Smec/Liq	362.5	5.23	14.43			
$\text{C}_{40}\text{H}_{52}\text{O}_6\text{S}$	4-[[4-(undecyloxy)benzoyl]thio]benzoic acid, 4-[[[(1-methylheptyl)oxy]carbonyl]phenyl ester					[85]	
	Sol/Smec	368.7	49	132.90			
	Smec/Smec	401.2	0.1	0.25			
	Smec/Liq	422.2	5.8	13.74	146.89		
$\text{C}_{40}\text{H}_{52}\text{O}_7$	2-(1-oxooctyl)-1,4-phenylene 4-(hexyloxy)benzoate					[432]	
	Sol/Nem	354.3	53.3	150.44			
	Nem/Liq	360.6	0.76	2.11	152.55		
$\text{C}_{40}\text{H}_{52}\text{O}_7$	2-(1-oxybutyl)-1,4-phenylene 4-(octyloxy)benzoate					[432]	
	Sol/Nem	356.1	44.8	125.81			
	Nem/Liq	367.2	1.8	4.90	130.71		
$\text{C}_{40}\text{H}_{52}\text{O}_7$	2-acetyl-1,4-phenylene 4-(nonyloxy)benzoate					[432]	
	Sol/Nem	374.9	40.2	107.23			
	Nem/Liq	390.8	1.4	3.58	110.81		
$\text{C}_{40}\text{H}_{52}\text{O}_8$	propyl 2,5-bis[[4-(octyloxy)benzoyl]oxy]benzoate					[432]	
	Sol/Nem	363.6	27.9	76.73			
	Nem/Liq	373.2	2.0	5.36	82.09		
$\text{C}_{40}\text{H}_{52}\text{S}_5$	5,5'''-didecyl-2,2':5',5":5",2"-quinquethiophene					[303]	
	Sol/Smec	345.2	10.7	31.00			
	Smec/Liq	498.2	41.5	83.30	114.30		
$\text{C}_{40}\text{H}_{53}\text{ClO}_5$	(R)-1-methylheptyl 4'-(3"-chloro-4"-dodecyloxybenzoyloxy)biphenyl-4-carboxylate					[226]	
	Sol/Smec	323.5	22.7	70.17			
	Smec/Smec	351.1	0.32	0.91			
	Smec/Meso	353.6	Could not be measured				
	Meso/Liq	356.8	2.1	5.89	76.97		
$\text{C}_{40}\text{H}_{54}\text{ClNO}_4$	3-hydroxy-4-[{(E)-[(4-tetradecylphenyl)imino]methyl}phenyl 4-butoxy- $\alpha$ -chlorobenzenepropanoate					[249]	
	Sol/Smec	332.2	Not reported in paper				
	Smec/Smec	340.0	4.41	12.97			
	Smec/Smec	366.4	0.44	1.20			
$\text{C}_{40}\text{H}_{54}\text{O}_2$	2-(decyloxy)-6-[4-(decyloxy)phenyl]anthracene					[302]	
	Sol/Smec	456.2	38.38	84.13			
	Smec/Liq	475.2	7.48	15.74	99.87		
$\text{C}_{40}\text{H}_{54}\text{O}_3$	1-(4'-dodecylbiphenyl-4-yl)-3-(2-heptyloxyphenyl)propane-1,3-dione					[297]	
	Sol/Nem	321.7	50.42	156.73			
	Nem/Liq	339.2	0.33	0.97	157.70		
$\text{C}_{40}\text{H}_{54}\text{O}_3$	1-(4'-dodecylbiphenyl-4-yl)-3-(3-heptyloxyphenyl)propane-1,3-dione					[297]	
	Sol/Smec	351.2	49.29	140.35			
	Smec/Liq	372.2	8.20	22.03	62.36		
$\text{C}_{40}\text{H}_{54}\text{O}_5$	(R)-4-(1-methylheptyloxy carbonyl)phenyl 4-(4-decyloxyphenyl)ethylbenzoate					[184]	
	Sol/Smec	322.2	23.7	73.56			
	Smec/Smec	358.0	0.18	0.50			
$\text{C}_{40}\text{H}_{54}\text{O}_7$	4'-(decyloxy)[1,1'-biphenyl]-4-yl 4-[(2S)-2-(heptyloxy)-1-oxopropoxy]-3-methoxybenzoate					[227]	
	Sol/Sol	323.2	13.13	40.63			
	Sol/Smec	336.2	3.75	11.15			
	Smec/Liq	376.2	6.08	16.16	67.94		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated) (estimated)	Ref.	
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$			
Note: Enthalpies determined from cooling measurements.							
$C_{40}H_{54}O_7$	Sol/Nem	4-(4-butoxybenzoyloxy)phenyl (3,4-dioctyloxy)benzoate	366.7	9.97	27.19	[103]	
			381.7	0.25	0.65		
					27.84		
$C_{40}H_{54}O_7$	Sol/Smec	4-(4-octyloxybenzoyloxy)phenyl (3,4-dihexyloxy)benzoate	357.4	6.43	17.99	[103]	
			366.7	1.78	4.85		
			384.5	0.41	1.07		
$C_{40}H_{54}O_7$	Nem/Liq	4-(4-dodecyloxybenzoyloxy)phenyl (3,4-dibutoxy)benzoate	365.7	8.63	23.60	[103]	
			380.2	0.30	0.79		
					24.39		
$C_{40}H_{55}NO_4$	Sol/Smec	4-(3-decyloxyphenyliminomethyl)phenyl 4-decyloxybenzoate	334.2	44.4	132.85	[175]	
			337.2	4.0	11.86		
					144.71		
$C_{40}H_{56}N_2O_2$	Sol/Liq	cholesteryl p-phenylazobenzoate	465.6	29.62	63.6	[155, 310]	
			458.5	1.51	3.29		
$C_{40}H_{56}N_2O_4$	Sol/Smec	2,5-dihydro-2,5-dimethyl-3,6-bis[4-(decyloxy)phenyl]pyrrolo[3,4-c]pyrrole-1,4-dione	379.5	68.9	181.55	[388]	
			392.2	3.0	7.65		
					189.20		
$C_{40}H_{57}NO_2$	Sol/Smec	7-(undec-10-enyloxy)-3-(4'-dodecyloxystyryl)quinoline	356.4	48.29	135.49	[139]	
			433.6	10.04	23.15		
					158.64		
$C_{40}H_{57}NO_2$	Smec/Smec	N-(2-hydroxy-4-nonyloxybenzylidene)-4"-dodecylphenylaniline	358.2	65.10	181.74	[323]	
			413.2	1.55	3.75		
			480.2	10.79	22.47		
$C_{40}H_{59}NO_7Si_2S$	Sol/Smec	(R)-6-[11-(1,1,3,3-pentamethyldisiloxanyl)undecyloxy]-5'-nitro-6'-(oct-2-yloxy)- $\Delta^{2,2'}\text{-bi-1-benzothiophene}$ -3(2H),3'(2'H)dione	393.2	19.26	48.98	[316]	
			416.2	2.43	5.84		
					54.82		
$C_{40}H_{60}N_2O_3$	Sol/Sol	pentyldecyl 2-[(4-decyloxy)phenyl]-6-cycloheptimidazolecarboxylate	383.2	8.0	20.88	[231]	
			389.2	8.0	20.55		
			394.2	4.1	10.40		
			400.2	3.2	8.00		
$C_{40}H_{60}O_4$	Smec/Smec	di(4'-decylphenyl)-trans-cyclohexane-1,4-dicarboxylate	338.0	64.8	191.72	[215]	
			390.4	5.36	13.73		
			407.8	8.63	21.16		
$C_{40}H_{60}O_6$	Smec/Smec	di(4'-decyloxyphenyl)-trans-cyclohexane-1,4-dicarboxylate	360.2	41.66	115.66	[220]	
			384.2	2.71	7.05		
			419.2	Not reported in paper			
			443.2	8.24	8.59		
$C_{40}H_{62}O_2$	Smec/Liq	cholesteryl $\omega$ -phenylheptanoate	370.6	39.58	106.8	[155, 169]	
			355.9	0.79	2.23		
Note: Enthalpy of the Liq/Chol transition is from cooling measurements.							

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.		
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$				
$C_{40}H_{65}NO$	Sol/SmeC	N-(4-undecyloxybenzylidene)-4-hexadecylaniline 317.5	41.35	130.24	153.60	[65]		
	SmeC/Liq	352.7	8.24	23.36				
$C_{40}H_{65}NO$	Sol/SmeC	4-dodecyl-N-[4-(pentadecyloxy)phenyl]methylene]benzenamine 347.9	44.28	127.28	163.92	[240]		
	SmeC/Liq	364.1	13.34	36.64				
$C_{40}H_{65}NO$	Sol/SmeC	4-nonyl-N-[4-(octadecyloxy)phenyl]methylene]benzenamine 338.9	48.58	143.35	194.56	[240]		
	SmeC/Liq	356.8	18.27	51.21				
$C_{40}H_{70}OS$	cholesteryl thiotridecanoate							
	Sol/SmeC	353.2	34.9	98.7	NA	[155,312]		
	SmeC/Chol	359.7	1.3	3.6				
	Chol/Liq	364.6	0.6	1.7	104.0			
$C_{40}H_{73}NO_4$	Sol/Meso	330.2	14.3	43.31	60.30	[378]		
	Meso/Liq	353.2	6.0	16.99				
$C_{41}H_{39}N_3O_6$	Sol/Nem	418.2	40.0	95.65	96.60	[205]		
	Nem/Liq	420.2	0.4	0.95				
$C_{41}H_{46}N_2O_6$	Sol/SmeC	365.2	42.0	115.01	116.97	[293]		
	SmeC/Nem	381.2	Not reported in paper					
	Nem/Liq	407.2	0.8	1.96				
$C_{41}H_{46}N_2O_8$	Sol/SmeC	407.2	33.0	81.04	108.74	[412]		
	SmeC/Liq	433.2	12.0	27.70				
$C_{41}H_{48}F_2O_7$	4-[[4-[(2-ethoxy-1-methyl-2-oxyethoxy)carbonyl]phenyl]ethynyl]phenyl 2,3-difluoro-4-(tetradecyloxy)-benzoate							
	Sol/SmeC	363.7	53.4	146.82	149.90	[268]		
	SmeC/Nem	384.0	0.6	1.56				
	Nem/Liq	394.4	0.6	1.52				
$C_{41}H_{48}N_4O_6$	Sol/SmeC	351.7	25.1	71.37	98.90	[107]		
	SmeC/Liq	395.9	10.9	27.53				
$C_{41}H_{49}FO_7$	4-[[4-[(2-ethoxy-1-methyl-2-oxyethoxy)carbonyl]phenyl]ethynyl]phenyl 3-fluoro-4-(tetradecyloxy)benzoate							
	Sol/SmeC	351.7	50.7	144.16	149.59	[268]		
	SmeC/Nem	379.4	0.4	1.05				
	Nem/Liq	387.7	1.7	4.38				
$C_{41}H_{50}F_2O_5$	Sol/SmeC	354.3	40.71	114.90	126.37	[213]		
	SmeC/Liq	392.2	4.50	11.47				
$C_{41}H_{50}O$	Sol/Nem	381.2	26.2	68.73	74.28	[76]		
	Nem/Liq	540.2	3.0	5.55				
$C_{41}H_{50}O_7$	2-methyl-4{4-[2-(trans-pentylcyclohexyl)ethyl]phenylethynyl}-1-[(4-pentyloxyphenyl)ethynyl]benzene							
	Sol/SmeC	351.0	41.8	119.09	125.54	[268]		
	SmeC/Nem	398.1	0.5	1.26				
	Nem/Liq	404.3	2.1	5.19				

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)						
$C_{41}H_{51}FO_5$	Sol/Smec	[R]-1-methylheptyl 4-[[2-fluoro-4-[[4-(undecyloxy)benzoyl]oxy]phenyl]ethynyl]benzoate 341.3	40.74	119.37				[211]
	Smec/Liq	389.6	4.96	12.73	132.10			
$C_{41}H_{51}FO_5$	Sol/Smec	1-methylheptyl 4'-(4"-undecyloxy-3"-fluorobenzoyloxy)tolan-4-carboxylate 369.7	35.16	95.10			[213]	
	Smec/Liq	385.9	5.08	13.16	108.26			
$C_{41}H_{51}FO_7S$	Sol/Meso	1-(butoxycarbonyl)ethyl 5-[(3-fluoro-4-tetradecyloxybenzoylphenyl)ethynyl]thiophene-2-carboxylate 343.9	38.60	112.24			[267]	
	Meso/Liq	350.0	0.80	2.29	114.53			
$C_{41}H_{51}F_3O_7$	Sol/Smec	(S)-1-ethylheptyl 4-[4-(4-undecyloxy-3-fluorobenzoyloxy)-3-fluorobenzoyloxy]-2-fluorobenzoate 336.4	28.01	83.26			[80]	
	Smec/Smec	348.7	0.87	2.49				
	Smec/Liq	356.2	3.28	9.21	94.96			
$C_{41}H_{52}F_2O_7$	Sol/Smec	(R)-1-methylheptyl 4-[4-(4-dodecyloxy-3-fluorobenzoyloxy)-3-fluorobenzoyloxy]benzoate 341.1	33.68	98.74			[69]	
	Smec/Smec	370.3	0.33	0.89				
	Smec/Liq	374.7	5.03	13.42	113.05			
$C_{41}H_{52}N_4O_2$	Sol/Smec	$\alpha$ -(4'-hexylazobenzene-4-oxy)- $\omega$ -(4'-pentylazobenzene-4-oxy)hexane 400.2	45.25	113.07			[67]	
	Smec/Liq	461.2	12.85	27.86	140.93			
$C_{41}H_{52}N_4O_2$	Sol/Smec	$\alpha$ -(4'-pentylazobenzene-4-oxy)- $\omega$ -(4'-pentylazobenzene-4-oxy)heptane 391.2	39.35	100.59			[67]	
	Smec/Nem	404.2	0.54	1.34				
	Nem/Liq	420.2	2.97	7.07	109.00			
$C_{41}H_{52}O_7$	Sol/Smec	(S)-(+)1-methylheptyl 4-[4-(trans-4-decyloxycinnamoyloxy)-benzoyloxy]benzoate 359.2	34.83	96.97			[149]	
	Smec/Smec	421.7	0.52	1.23				
	Smec/Liq	434.2	3.87	8.91	107.11			
$C_{41}H_{52}O_7$	Sol/Smec	(S)-(+)-1-methylheptyl 4-[4-(trans-4-nonyloxy- $\alpha$ -methylcinnamoyloxy)benzoyloxy]benzoate 330.7	43.36	131.12			[149]	
	Smec/Smec	332.8	0.09	0.27				
	Smec/Liq	388.7	4.85	12.48	143.87			
$C_{41}H_{53}NO_6$	Sol/Smec	4-cyanobenzyl 5-(4-octyloxybenzoyloxy)-2-undecyloxybenzoate 334.2	33.7	100.84			[407]	
	Smec/Liq	349.2	8.16	23.37	124.21			
Independent values from another reference								
	Sol/Smec	331.4	34.0	102.60			[407, 408]	
	Smec/Liq	348.3	8.0	22.97	125.57			
$C_{41}H_{54}O_5S$	Sol/Smec	(S)-1-methylheptyl 5-[(4-tridecyloxybenzoyloxyphenyl)ethynyl]-thiophene-2-carboxylate 328.3	35.47	108.04			[267]	
	Smec/Smec	348.0	0.15	0.43				
	Smec/Liq	359.8	4.27	11.87	120.34			
$C_{41}H_{54}O_6S$	Sol/Smec	4-[[4-(dodecyloxy)benzoyl]thio]benzoic acid, 4-[[1-methylheptyl]oxy]carbonyl]phenyl ester 355.6	27.4	77.05			[85]	
	Smec/Smec	388.2	0.01	0.03				
	Smec/Smec	403.2	0.27	0.67				
	Smec/Liq	419.2	5.9	14.07	91.82	NA		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{41}H_{54}O_7$			(1S)-2-butoxy-1-methyl-2-oxoethyl 4-[[[(2E)-3-[6-tetradecyloxy)-2-naphthalenyl]-1-oxo-2-propenyl]oxy]benzoate			
	Sol/Smec	344.8	38.36	111.25		
	Smec/Smec	358.9	0.37	1.03		
$C_{41}H_{54}O_7$	Smec/Liq	385.9	3.62	9.38	121.66	[401]
			2-(1-oxoheptyl)-1,4-phenylene 4-(heptyloxy)benzoate			
	Sol/Nem	354.4	53.0	149.54		
$C_{41}H_{54}O_7$	Nem/Liq	355.8	1.16	3.26	152.80	[432]
			2-(1-oxopropyl)-1,4-phenylene 4-(nonyloxy)benzoate			
	Sol/Nem	350.7	60.3	171.94		
$C_{41}H_{54}O_8$	Nem/Liq	369.3	1.6	4.33	176.27	[432]
			butyl 2,5-bis[[4-(octyloxy)benzoyl]oxy]benzoate			
	Sol/Nem	355.4	27.2	76.53		
$C_{41}H_{56}O_3$	Nem/Liq	371.8	1.5	4.03	80.56	[432]
			1-(4'-dodecylbiphenyl-4-yl)-3-(2-octyloxyphenyl)propane-1,3-dione			
	Sol/Nem	333.7	35.77	107.19		
$C_{41}H_{56}O_3$	Nem/Liq	344.2	0.42	1.22	108.41	[297]
			1-(4'-dodecylbiphenyl-4-yl)-3-(3-octyloxyphenyl)propane-1,3-dione			
	Sol/Smec	349.7	48.07	137.46		
$C_{41}H_{56}O_5$	Smec/Liq	372.7	8.70	23.34	160.80	[297]
			(R)-4-(1-methylheptyloxy carbonyl)phenyl 4-(4-undecyloxyphenyl)ethylbenzoate			
	Sol/Smec	319.2	41.0	128.45		
$C_{41}H_{56}O_5$	Smec/Liq	359.2	10.0	27.84	156.29	[184]
			(R)-4-(3-ethylmercapto-2-methylpropionyl)phenyl 4'-hexadecyloxy-biphenyl-4-carboxylate			
	Sol/Smec	364.7	35.36	96.96		
$C_{41}H_{56}O_7$	Smec/Smec	413.8	0.02	0.05		
	Smec/Liq	423.8	3.63	8.57	105.58	[100]
			4'-(decyloxy)[1,1'-biphenyl]-4-yl 4-[(2S)-2-(octyloxy)-1-oxopropoxy]-3-methoxybenzoate			
$C_{41}H_{56}O_7$	Sol/Sol	325.2	15.00	46.13		
	Sol/Smec	335.2	2.58	7.70		
	Smec/Liq	375.2	6.15	16.39	70.22	[227]
Note: Enthalpies were determined from cooling measurements.						
$C_{41}H_{57}NO_4$			4-(3-decyloxyphenyliminomethyl)phenyl 4-undecyloxybenzoate			
	Sol/Smec	330.2	61.0	184.74		
	Smec/Smec	333.7	0.05	0.15		
	Smec/Smec	335.7	0.25	0.74		
$C_{41}H_{59}NO_2$	Smec/Liq	336.7	4.30	12.77	198.40	[175]
			N-(2-hydroxy-4-decyloxybenzylidene)-4''-dodecylphenylaniline			
	Sol/Smec	352.7	52.51	148.88		
$C_{41}H_{64}N_2O_3$	Smec/Smec	411.2	2.30	5.59		
	Smec/Liq	476.7	12.01	25.19	179.66	[323]
			N-[4-[5-oxo-4-[(1-oxotetradecyl)amino]-1,3,6-cycloheptatrien-1-yl]phenyl]tetradecamide			
$C_{41}H_{64}N_2O_3$	Sol/Smec	421.2	37.2	88.32		
	Smec/Liq	428.2	6.3	14.71	103.03	[251]
$C_{41}H_{64}N_2O_3$			N-[7-oxo-4-[4-[(1-oxohexadecyl)amino]phenyl]-1,3,5-cycloheptatrien-1-yl]dodecamide			
	Sol/Smec	416.2	45.7	109.80		
$C_{41}H_{64}O_2$	Smec/Liq	429.2	7.7	17.94	127.74	[251]
			cholesteryl $\omega$ -phenyloctanoate			
$C_{41}H_{64}O_2$	Sol/Liq	364.4	38.16	104.71	104.71	156.8
						[155, 169]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$\text{C}_{41}\text{H}_{67}\text{NO}$	Liq/Chol	329.8	3.51	10.66		
	Chol/Smec	307.9	0.59	1.9		
Note: Enthalpies of the Liq/Chol and Chol/Smec transitions are from cooling measurements.						
$\text{C}_{41}\text{H}_{67}\text{NO}$	4-decyl-N-[[4-(octadecyloxy)phenyl]methylen]benzenamine					
	Sol/Smec	345.6	46.93	135.79		[240]
	Smec/Liq	363.5	16.01	44.04	179.83	
$\text{C}_{41}\text{H}_{67}\text{NO}$	N-(4-dodecyloxybenzylidene)-4-hexadecylaniline					
	Sol/Smec	351.3	75.49	214.89		[256]
	Smec/Liq	361.3	20.37	56.38	271.27	
$\text{C}_{41}\text{H}_{70}\text{N}_2\text{O}_6$	4,4'-bis[4-(4-decyloxybenzylideneamino)benzoyloxy]diphenylmethane					
	Sol/Smec	419.2	55.0	131.20		[284]
	Smec/Liq	456.2	22.0	48.22	179.42	
$\text{C}_{41}\text{H}_{72}\text{OS}$	cholesteryl thiotetradecanoate					
	Sol/Smec	346.2	29.6	85.5		
	Smec/Chol	358.7	1.4	4.0		
$\text{C}_{41}\text{H}_{72}\text{O}_2$	Chol/Liq	363.6	0.8	2.2	91.7	NA
	cholesterol myristate (tetradecanoate)					[155,312]
	Sol/Smec	346.8	46.71	134.69		
$\text{C}_{41}\text{H}_{72}\text{O}_2$	Smec/Chol	353.2	1.30	3.68		
	Chol/Liq	358.8	1.02	2.84	141.21	182.5
	Independent values from another reference					[169]
$\text{C}_{41}\text{H}_{72}\text{O}_2$	Sol/Smec	343.7	46.46	135.18		
	Smec/Chol	351.0	1.40	3.99		
	Chol/Liq	356.4	1.02	2.86	142.03	182.5
$\text{C}_{41}\text{H}_{72}\text{O}_2$	Independent values from another reference					[170]
	Sol/Smec	344.2	47.5	138.00		
	Smec/Chol	352.3	1.78	5.05		
$\text{C}_{41}\text{H}_{72}\text{O}_2$	Chol/Liq	354.8	1.51	4.26	147.31	182.5
	Independent values from another reference					[180]
	Sol/Smec	346.8	46.6	133.5		
$\text{C}_{41}\text{H}_{74}\text{N}_2\text{O}_2$	Smec/Chol	352.9	1.3	3.7		
	Chol/Liq	358.7	1.0	2.9	141.0	182.5
	Independent values from another reference					[306]
$\text{C}_{41}\text{H}_{74}\text{N}_2\text{O}_2$	$\text{N},\text{N}'\text{-dihexadecanoyl-2,3,5-trimethylbenzene-1,4-diamine}$					
	Sol/Meso	400.2	61	152.42		
	Meso/Nem	435.2	18	41.36		
$\text{C}_{42}\text{H}_{36}\text{N}_2\text{O}_5$	Nem/Liq	436.2	0.6	1.38	195.16	[36]
$\text{C}_{42}\text{H}_{36}\text{N}_2\text{O}_5$	4-[4-[(4-hexyloxybenzoyloxy)benzoyloxy]phenyl]-6'-phenyl-2,2'-bipyridine					
	Sol/Nem	414.2	21.2	51.18		
	Nem/Liq	446.2	0.40	0.90	52.08	[171]
$\text{C}_{42}\text{H}_{38}\text{Cl}_2\text{N}_2\text{O}_6$	4,5-dichloro-1,3-phenylene bis[4-(4-butoxybenzylideneamino)benzoate]					
	Sol/Nem	423.2	36.0	85.07		
	Nem/Liq	460.2	2.3	5.00	90.07	[86]
$\text{C}_{42}\text{H}_{40}\text{N}_2\text{O}_6$	1,3-phenylene bis[4-(4-butoxyphenyliminomethyl)benzoate]					
	Sol/Meso	455.2	14.0	30.76		
	Meso/Smec	458.2	16.0	34.92		
$\text{C}_{42}\text{H}_{40}\text{N}_2\text{O}_6$	Smec/Liq	461.2	17.0	36.86	102.54	[86]
$\text{C}_{42}\text{H}_{40}\text{N}_2\text{O}_6$	1,3-phenylene bis[4-(4-butoxybenzylideneamino)benzoate]					
	Sol/Smec	435.2	14.0	32.17		
	Smec/Liq	442.2	12.0	27.14	59.31	[86]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)						
$C_{42}H_{40}N_2O_8$	Sol/Meso	461.2	1,3-bis[4-(2-hydroxy-4-butoxybenzylideneamino)benzoyloxy]benzene	36.0	78.06			[412]
	Meso/Liq	469.2		13.0	27.71	105.77		
$C_{42}H_{42}O_4$	Sol/Nem	394.1	bis(4-heptylbutadiynylphenyl) terephthalate	27.0	68.51			[176]
	Nem/Liq	469.2			Not reported in paper			
$C_{42}H_{44}N_2O_4$	Sol/Sol	411.0	3,3',4,4'-biphenyltetracarboxy-N,N'-bis(heptylphenyl)diimide	18.8	45.74			[94]
	Sol/Smec	504.9		24.7	48.92			
	Smec/Liq	560.8		11.1	19.79	114.45		
$C_{42}H_{44}N_2O_6$	Sol/Sol	375.9	3,3',4,4'-biphenyltetracarboxy-N,N'-bis(heptyloxyphenyl)diimide	21.1	56.13			[94]
	Sol/Sol	498.3		7.5	15.05			
	Sol/Smec	518.6		25.1	48.40			
	Smec/Liq	579.5		9.9	17.08	136.66		
$C_{42}H_{46}FeO_4$	Sol/Nem	396.2	[4-[[[4'-(dodecyloxy)carbonyl][1,1'-biphenyl]-4-yl]oxy]carbonyl]phenyl]ferrocene	28.6	72.19			[376]
	Nem/Liq	406.2		0.75	1.85	74.04	NA	
$C_{42}H_{46}O_9$	Sol/Smec	353.9	4-[(1S)-2-butoxy-1-methyl-2-oxoethoxy]carbonyl]phenyl	18.43	52.08			[401]
	Smec/Smec	430.6		0.17	0.39			
	Smec/Liq	497.3		2.73	5.49	57.96		
$C_{42}H_{48}N_2O_6$	Sol/Nem	436.2	$\alpha,\omega$ -bis[4-(4-pentyloxybenzoyloxy)benzylideneamino]butane	60.0	137.55			[293]
	Nem/Liq	460.2		7.8	16.95	154.50		
$C_{42}H_{48}N_2O_6$	Sol/Nem	436.2	$\alpha,\omega$ -bis[4-(4-butoxybenzoyloxy)benzylideneamino]hexane	56.0	128.38			[293]
	Nem/Liq	454.2		8.7	19.16	147.54		
$C_{42}H_{48}N_2O_6$	Sol/Nem	406.2	$\alpha,\omega$ -bis[4-(4-propoxybenzoyloxy)benzylideneamino]octane	38.0	93.55			[293]
	Nem/Liq	436.2		8.3	10.03	103.58		
$C_{42}H_{48}N_2O_6$	Sol/Nem	432.2	$\alpha,\omega$ -bis[4-(4-ethoxybenzoyloxy)benzylideneamino]decane	54.0	124.94			[293]
	Nem/Liq	439.2		7.7	17.53	142.47		
$C_{42}H_{50}N_2O_2S$	Sol/Smec	362.2	2,5-bis[[4-(decyloxy)phenyl]ethynyl]-3,4-thiophenedicarbonitrile	45.2	124.79			[373]
	Smec/Nem	384.5		1.9	4.94			
	Nem/Liq	394.9		1.9	4.81	134.54		
$C_{42}H_{50}N_4O_6$	Sol/Smec	391.9	$\alpha,\omega$ -bis(4-butoxyazobenzene-4'-carbonyloxy)octane	52.7	134.47			[107]
	Smec/Nem	421.1		3.8	9.02			
	Nem/Liq	425.4		10.0	23.51	167.00		
$C_{42}H_{50}N_4O_6S_2$	Sol/Smec	386.2	bis[4-(5-n-nonylthio-1,3,4-oxadiazole-2-yl)phenyl]terephthalate	44.3	114.71			[23]
	Smec/Liq	541.2		8.1	14.97	129.68		
$C_{42}H_{50}O_8$	Sol/Sol	318.9	benzoic acid, 4,4'-[1,4-phenylenebis[(1-oxo-2-propene-3,1-diyl)oxo]]bisdioctyl ester					

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated) (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{42}H_{50}O_8$	Sol/Sol	325.2	23.9	73.49		
	Note: Both Sol/Sol transition enthalpies are included in the latter value.					
	Sol/Smec	432.3	54.8	126.76		
	Smec/Smec	485.4	0.2	0.41		
$C_{42}H_{51}N_3O_6$	Smec/Liq	499.5	6.7	13.41	214.07	[271]
	1,4-benzenedicarboxylic acid, bis[4-[3-(octyloxy)-3-oxo-1-propenyl]phenyl ester					
	Sol/Sol	318.9	10.2	31.98		
	Sol/Sol	369.8	1.1	2.97		
$C_{42}H_{52}F_2O_5$	Smec/Smec	381.4	35.6	93.34		
	Smec/Liq	504.1	6.0	11.90	140.19	[271]
	4-(4'-ethoxybenzoyloxy)-2-decyloxy-4'-(4-butyoxysalicylaldimine)azobenzene					
	Sol/Sol	358.7	4.77	13.30		
$C_{42}H_{52}N_4O_6S_3$	Sol/Nem	379.7	35.99	94.79		
	Nem/Liq	507.7	1.60	3.15	111.24	[109]
	1-methylheptyl 4'-(4"-dodecyloxy-2",3"-difluorobenzoyloxy)tolan-4-carboxylate					
	Sol/Smec	338.8	35.57	104.99		
$C_{42}H_{53}FO_5$	Smec/Liq	390.5	4.27	10.93	115.92	[213]
	[(R)-1-methylheptyl] 4-[[2-fluoro-4-[[4-(dodecyloxy)benzoyl]oxy]phenyl]ethynyl]benzoate					
	Sol/Smec	323.8	36.54	112.85		
	Smec/Smec	358.7	0.03	0.08		
$C_{42}H_{53}FO_5$	Smec/Liq	387.0	5.02	12.97	125.90	[211]
	1-methylheptyl 4'-(4"-dodecyloxy-3"-fluorobenzoyloxy)tolan-4-carboxylate					
	Sol/Smec	358.7	28.77	80.21		
	Smec/Liq	383.9	4.40	11.46	91.67	[213]
$C_{42}H_{53}F_3O_7$	(S)-1-ethylheptyl 4-[4-(4-dodecyloxy-3-fluorobenzoyloxy)-3-fluorobenzoyloxy]-2-fluorobenzoate					
	Sol/Smec	314.1	22.49	71.60		
	Smec/Smec	350.0	0.84	2.4		
	Smec/Liq	356.5	3.35	9.40	83.40	[80]
$C_{42}H_{54}F_2O_5S$	(S)-1-methylheptyl 5-[(2,3-difluoro-4-tetradecyloxybenzoyloxy-phenyl)ethynyl]-thiophene-2-carboxylate					
	Sol/Smec	334.3	62.00	185.46		
	Smec/Smec	343.0	0.12	0.35		
	Smec/Liq	352.1	4.00	11.36	197.17	[267]
$C_{42}H_{54}F_2O_7$	(R)-1-methylheptyl 4-[4-(4-tridecyloxy-3-fluorobenzoyloxy)-3-fluorobenzoyloxy]benzoate					
	Sol/Smec	336.1	31.79	94.58		
	Smec/Meso	370.0	0.37	1.00		
	Meso/Liq	373.2	4.67	12.51	108.09	[69]
$C_{42}H_{54}N_4O_2$	$\alpha$ -(4'-hexylazobenzene-4-oxy)- $\omega$ -(4'-hexylazobenzene-4-oxy)hexane					
	Sol/Smec	415.2	47.98	115.56		
	Smec/Liq	458.2	14.78	32.26	187.82	[67]
	$\alpha$ -(4'-pentylazobenzene-4-oxy)- $\omega$ -(4'-pentylazobenzene-4-oxy)octane					
$C_{42}H_{54}N_4O_2$	Sol/Smec	393.2	50.34	128.03		
	Smec/Nem	427.2	0.46	1.08		
	Nem/Liq	444.2	7.57	17.04	146.15	[67]
	4,4'''-dinonyloxy-p-quaterphenyl					
$C_{42}H_{54}O_2$	Sol/Sol	396.0	22.57	56.99		
	Sol/Sol	439.0	2.20	5.01		
	Sol/Sol	453.0	6.34	14.00		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{42}H_{54}O_7$	Sol/Smec	566.0	12.45	22.00		
	Smec/Smec	597.0	1.79	3.00		
	Smec/Liq	600.0	9.00	15.00	116.00	216.6 [111]
$C_{42}H_{54}O_7$	(S)-(+)-1-methylheptyl 4-[4-( <i>trans</i> -4-undecyloxy)cinnamoyloxy]benzoyloxy]benzoate					
	Sol/Smec	360.2	34.03	94.48		
	Smec/Smec	420.5	0.60	1.43		
	Smec/Liq	430.2	2.94	6.83	102.74	[149]
$C_{42}H_{54}O_7$	(S)-(+)-1-methylheptyl 4-[4-( <i>trans</i> -4-decyloxy- $\alpha$ -methylcinnamoyloxy]benzoyloxy]benzoate					
	Sol/Smec	335.7	44.47	132.47		
	Smec/Smec	342.7	0.01	0.03		
	Smec/Smec	346.7	0.02	0.06		
$C_{42}H_{55}FO_5S$	Smec/Liq	387.2	5.89	15.21	147.77	[149]
	(S)-1-methylheptyl 5-[(3-fluoro-4-tetradecyloxybenzoyloxyphenyl)ethynyl]thiophene-2-carboxylate					
	Sol/Smec	315.7	27.10	85.84		
	Smec/Smec	338.4	0.16	0.47		
$C_{42}H_{56}O_2$	Smec/Liq	344.4	4.60	13.36	99.67	[267]
	2-decyloxy-6-[4-(4-dodecyloxyphenyl)buta-1,3-diynyl]naphthalene					
	Sol/Smec	347.1	16.6	47.82		
	Smec/Nem	378.3	1.3	3.44		
$C_{42}H_{56}O_5S$	Nem/Liq	424.8	2.6	6.12	57.38	[225]
	(S)-1-methylheptyl 5-[(4-tetradecyloxybenzoyloxyphenyl)ethynyl]thiophene-2-carboxylate					
	Sol/Smec	326.0	41.50	127.30		
	Smec/Smec	352.0	0.24	0.68		
$C_{42}H_{56}O_7$	Smec/Liq	359.3	5.20	14.47	142.45	[267]
	2-(1-oxohexyl)-1,4-phenylene 4-(octyloxy)benzoate					
	Sol/Nem	356.7	79.1	221.75		
	Nem/Liq	358.4	1.5	4.18	225.93	[432]
$C_{42}H_{57}ClO_5$	(R)-1-methylheptyl 4'-(3"-chloro-4"-tetradecyloxybenzoyloxy)-biphenyl-4-carboxylate					
	Sol/Smec	315.2	22.5	71.38		
	Smec/Smec	349.7	0.33	0.94		
	Smec/Meso	350.2	Could not be measured			
$C_{42}H_{57}ClO_5$	Meso/Liq	355.0	2.3	6.48	78.80	[226]
	(S)-1-methylheptyl 4'-(3"-chloro-4"-tetradecyloxybenzoyloxy)-biphenyl-4-carboxylate					
	Sol/Smec	312.2	24.8	79.44		
	Smec/Smec	348.5	0.34	0.98		
$C_{42}H_{58}O_3$	Smec/Liq	353.7	2.3	6.50	86.92	[226]
	1-(4'-dodecylbiphenyl-4-yl)-3-(2-nonyloxyphenyl)propane-1,3-dione					
	Sol/Nem	339.7	38.78	114.16		
	Nem/Liq	343.7	0.37	1.08	115.24	[297]
$C_{42}H_{58}O_3$	1-(4'-dodecylbiphenyl-4-yl)-3-(3-nonyloxyphenyl)propane-1,3-dione					
	Sol/Sol	333.7	8.95	26.82		
	Sol/Smec	347.7	46.77	134.51		
	Smec/Liq	371.7	8.37	22.52	183.85	[297]
$C_{42}H_{58}O_5$	(R)-4-(1-methylheptyloxycarbonyl)phenyl 4-(4-dodecyloxyphenyl)ethylbenzoate					
	Sol/Smec	325.2	40.0	123.00		
	Smec/Liq	359.2	9.4	26.17	149.17	[184]
$C_{42}H_{58}O_6$	4'-(decyloxy)[1,1'-biphenyl]-4-yl 4-[(2 <i>S</i> )-2-(decyloxy)-1-oxo-propoxy]benzoate					
	Sol/Smec	373.2	51.72	138.59		
	Smec/Liq	408.2	10.94	26.80	165.39	[227]

Note: Enthalpies were determined from cooling measurements.

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)						
$C_{42}H_{58}O_7$	Sol/SmeC	4'--(decyloxy)[1,1'-biphenyl]-4-yl 4-[(2S)-2-(nonyloxy)-1-oxo-propoxy]-3-methoxybenzoate		301.2	14.58	48.41		[227]
				371.2	5.67	15.27	63.68	
		Note: Enthalpies were determined from cooling measurements.						
$C_{42}H_{58}O_7$	Sol/SmeC	4-(4-octyloxybenzoyloxy)phenyl (3,4-diheptyloxy)benzoate		364.6	7.47	20.49		[103]
				370.2	1.87	5.05		
				381.5	0.40	1.05	26.59	
$C_{42}H_{58}O_7$	Sol/SmeC	4-(4-dodecyloxybenzoyloxy)phenyl (3,4-dipentyloxy)benzoate		353.6	8.03	22.71		[103]
				364.4	2.07	5.68		
				376.9	0.47	1.25	29.64	
$C_{42}H_{59}NO_4$	Sol/SmeC	4-(3-decyloxyphenyliminomethyl)phenyl 4-dodecyloxybenzoate		338.2	55.2	163.22		[175]
				339.2	5.2	15.33	178.55	
$C_{42}H_{60}O_6$	Sol/Col	2,6,10-tris(pentyloxy)-3,7,11-tris(propoxy)triphenylene		341.2	30.71	90.01		[261]
				378.2	5.44	14.38	104.39	
	Independent values from another reference							
		Sol/Col	Not reported in paper					
		Col/Liq	378.2	5.4	14.28			[368]
$C_{42}H_{61}NO_2$	Sol/SmeC	7-(undec-10-enyloxy)-3-(4'-tetradecyloxystyryl)quinoline		346.7	35.37	102.02		[139]
				352.5	2.51	7.12		
				427.2	9.79	22.92	132.06	
$C_{42}H_{61}NO_2$	Sol/SmeC	N-(2-hydroxy-4-undecyloxybenzylidene)-4''-dodecylphenylaniline		356.2	70.63	198.29		[323]
				409.7	2.72	6.64		
				474.7	10.50	22.12	227.05	
$C_{42}H_{62}N_2O_6$	Sol/Disc	N,N'-dinonanoyl-2,4-bis(nanonoyloxy)-1,3-benzenediamine		363.2	16.0	44.05		[189]
				394.2	18.0	45.66	89.71	
$C_{42}H_{64}N_2O_3$	Sol/Sol	pentyldecyl 2-[(4-dodecyloxy)phenyl]-6-cycloheptimidazolecarboxylate		398.2	6.7	16.83		[231]
				399.2	13.6	34.07		
				402.2	1.5	3.73	54.63	
$C_{42}H_{64}O_4$	Sol/SmeC	di(4'-undecylphenyl)-trans-cyclohexane-1,4-dicarboxylate		334.2	35.5	106.22		[215]
				392.7	6.12	15.58		
				406.2	9.51	23.41	145.21	
$C_{42}H_{65}NO_7Si_3S$	Sol/SmeC	(R)-6-[11-(1,1,1,3,3,5,5-heptamethyltrisiloxanyl)undecyloxy]-5'-nitro-6'-(oct-2-yloxy) [ $\Delta^{2,2'} - \text{bi}-1\text{-benzothiophene}] - 3(2\text{H}),3'(2'\text{H})\text{-dione}$		383.2	18.32	47.81		[316]
				419.2	2.75	6.56	54.37	
$C_{42}H_{68}N_2O_{10}$	Sol/Disc	N,N'-dihexanoyl-2,3,5,6-tetrakis(hexanoyloxy)-1,4-benzenediamine		344.2	18.0	52.30		[188]
				482.2	45.0	93.32	145.62	
$C_{42}H_{69}NO$		4-tetradecyl-N-[[4-(pentadecyloxy)phenyl]methylene]benzenamine						

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{42}H_{74}N_2O_4$	Sol/Smec	353.1	42.10	119.23	168.97	[240]
	Smec/Liq	364.3	18.12	49.74		
$C_{42}H_{74}N_2O_4$	Sol/Meso	396.2	25.0	63.10	256.7	[193]
	Meso/Meso	425.2	21.0	49.39		
	Meso/Liq	465.2	3.0	6.45		
				118.94		
$C_{42}H_{74}OS$						
	cholesteryl thiopentadecanoate					
	Sol/Smec	335.2	36.5	109.0	NA	[155,312]
	Smec/Chol	357.5	4.2	11.7		
$C_{42}H_{74}O_2$	Chol/Liq	361.5	0.8	2.3	91.7	
	cholesterol pentadecanoate					
	Sol/Smec	343.5	49.60	144.40	189.6	[170]
$C_{42}H_{74}O_2$	Smec/Chol	350.3	1.61	4.60		
	Chol/Liq	355.0	0.97	2.73	151.73	
$C_{42}H_{75}N_3O_3$						
	N,N',N"-triundecanoyl-2,4,6-trimethyl-1,3,5-benzenetriamine					
	Sol/Meso	458.2	15.0	32.74	79.78	[190]
	Meso/Meso	464.2	15.0	32.31		
$C_{42}H_{76}N_2O_2$	Meso/Liq	611.2	9.0	14.73		
	N,N'-dihexadecanoyl-2,3,5,6-tetramethylbenzene-1,4-diamine					
	Sol/Meso	412.2	59	143.13	186.29	[36]
$C_{42}H_{76}N_2O_2$	Meso/Liq	498.2	22	44.16		
$C_{43}H_{44}N_3O_6$						
	4-[3,4-bis(4-pentyloxybenzoyloxy)benzylideneamino]azobenzene					
	Sol/Nem	399.2	51.0	127.76	128.74	[205]
	Nem/Liq	408.2	0.4	0.98		
$C_{43}H_{44}N_2O_8$						
	bis[4-ethoxyphenyl] 2-{3-[4-(4-butylphenylazo)phenoxy]propyloxy}terephthalate					
	Sol/Nem	392.2	22.6	57.62	62.55	[54]
	Nem/Liq	426.2	2.1	4.93		
$C_{43}H_{48}O_9$						
	4-[(1S)-2-butoxy-1-methyl-2-oxoethoxy]carbonylphenyl 4-[(2E)-3-[6-(nonyloxy)-2-naphthyl]-1-oxo-2-propenyl]oxy]benzoate					
	Sol/Smec	353.4	20.97	59.34	67.27	[401]
	Smec/Smec	445.3	0.23	0.52		
$C_{43}H_{50}O_3$	Smec/Liq	501.0	3.71	7.41		
	4-[(1E)-2-naphthalenylazo]-1-naphthalenyl 4-(hexadecyloxy)benzoate					
	Sol/Nem	368.2	39.19	106.44	107.98	[385]
$C_{43}H_{50}O_3$	Nem/Liq	408.2	0.63	1.54		
$C_{43}H_{50}N_2O_6$						
	$\alpha,\omega$ -bis[4-(4-butoxybenzoyloxy)benzylideneamino]heptane					
	Sol/Nem	382.2	50.0	130.82	132.11	[293]
	Nem/Liq	386.2	0.5	1.29		
$C_{43}H_{50}N_2O_6$						
	$\alpha,\omega$ -bis[4-(4-propoxybenzoyloxy)benzylideneamino]nonane					
	Sol/Nem	369.2	31.0	83.97	84.75	[293]
	Nem/Liq	384.2	0.3	0.78		
$C_{43}H_{50}N_2O_8$						
	malonic acid, bis{3-[4-[(4-butoxyphenylimino)methyl]phenoxy]propyl} ester					
	Sol/Nem	394.7	31.1	78.79	83.02	[244]
	Nem/Liq	401.8	1.7	4.23		
$C_{43}H_{50}N_2O_8$						
	1,5-[4-(2-hydroxy-4-pentyloxybenzylideneamino)benzoyloxy]pentane					
	Sol/Smec	415.2	41.0	98.75	127.04	[412]
	Smec/Liq	424.2	12.0	28.29		
$C_{43}H_{50}N_2O_8$						
	1,5-[4-(2-hydroxy-4-butoxybenzylideneamino)benzoyloxy]-2,2-dimethylpentane					

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.	
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$			
$C_{43}H_{50}N_2O_8$	Sol/Smec	370.7	27.0	72.84	97.07	[412]	
	Smec/Liq	412.7	10.0	24.23			
$C_{43}H_{51}FeNO$	1,5-[4-(2-hydroxy-4-butoxybenzylideneamino)benzoyloxy]-3,3-dimethylpentane					[412]	
	Sol/Smec	375.2	43.0	114.61	141.96		
	Smec/Smec	394.2	0.7	1.78			
	Smec/Liq	430.2	11.0	25.57			
$C_{43}H_{51}FeNO$	[4'-[(E)-[[4-(tetradecyloxy)phenyl]imino]methyl][1,1'-biphenyl]-4-yl]ferrocene					[433]	
	Sol/Smec	419.2	3.01	7.18	NA		
	Smec/Nem	437.2	10.39	23.76			
	Nem/Liq	455.2	0.63	1.38			
$C_{43}H_{52}N_4O_6$	$\alpha,\omega$ -bis(4-butoxyazobenzene-4'-carbonyloxy)nonane					[107]	
	Sol/Smec	336.9	26.3	78.06	103.77		
	Smec/Liq	389.0	10.0	25.71			
$C_{43}H_{55}FO_5$	[(R)-1-methylheptyl] 4-[[2-fluoro-4-[[4-(tridecyloxy)benzoyl]oxy]phenyl]ethynyl]benzoate					[211]	
	Sol/Smec	334.6	43.08	128.75	141.37		
	Smec/Smec	365.7	0.09	0.25			
	Smec/Liq	384.8	4.76	12.37			
$C_{43}H_{55}FO_7S$	1-(butoxycarbonyl)ethyl 5-[(3-fluoro-4-hexadecyloxybenzoylphenyl)ethynyl]thiophene-2-carboxylate					[267]	
	Sol/Meso	342.1	38.35	112.10	116.27		
	Meso/Nem	345.4	0.27	0.78			
	Nem/Liq	348.3	1.18	3.39			
$C_{43}H_{56}F_2O_7$	(R)-1-methylheptyl 4-[4-(4-tetradecyloxy-3-fluorobenzoyloxy)-3-fluorobenzoyloxy]benzoate					[69]	
	Sol/Smec	327.9	32.74	99.85	114.04		
	Smec/Meso	371.7	0.36	0.97			
	Meso/Liq	374.5	4.95	13.22			
$C_{43}H_{56}N_4O_2$	$\alpha$ -(4'-pentyiazobenzene-4-oxy)- $\omega$ -(4'-pentyiazobenzene-4-oxy)nonane					[67]	
	Sol/Smec	377.2	39.20	103.92	111.97		
	Smec/Nem	385.2	0.32	0.83			
	Nem/Liq	412.9	2.98	7.22			
$C_{43}H_{56}N_4O_5$	1,3-bis[4-[(1E)-[4-(octyloxy)phenyl]azo]phenoxy]-2-propanol					[230]	
	Sol/Smec	428.4	44.09	102.92	122.42		
	Smec/Liq	443.7	8.65	19.50			
	(S)-(+)1-methylheptyl 4-[4-(trans-4-dodecyloxy)cinnamoyloxy]benzoyloxy]benzoate						
$C_{43}H_{56}O_7$	Sol/Smec	361.2	35.49	98.26	106.71	[149]	
	Smec/Smec	419.0	0.66	1.58			
	Smec/Liq	427.7	2.94	6.87			
	(S)-(+)1-methylheptyl 4-[4-(trans-4-undecyloxy- $\alpha$ -methylcinnamoyloxy)benzoyloxy]benzoate						
$C_{43}H_{56}O_7$	Sol/Smec	333.9	33.0	98.83	112.85	[149]	
	Smec/Smec	357.4	0.01	0.03			
	Smec/Smec	361.2	0.01	0.03			
	Smec/Liq	386.9	5.40	13.96			
$C_{43}H_{58}O_5S$	(S)-1-methylheptyl 5-[(4-pentadecyloxybenzoyloxyphenyl)ethynyl]-thiophene-2-carboxylate					[267]	
	Sol/Smec	337.2	51.72	153.38	169.35		
	Smec/Smec	353.1	0.37	1.05			
	Smec/Liq	357.9	5.34	14.92			
$C_{43}H_{58}O_7$	2-(1-oxoheptyl)-1,4-phenylene 4-(octyloxy)benzoate					[432]	
	Sol/Nem	349.6	61.8	176.77	180.11		
	Nem/Liq	359.8	1.2	3.34			
	hexyl 2,5-bis[4-(octyloxy)benzoyl]oxy]benzoate						

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.			
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$					
$C_{43}H_{60}O_3$	Sol/Nem	333.1	31.7	95.17	98.95	[432]			
	Nem/Liq	370.8	1.4	3.78					
$C_{43}H_{60}O_3$	1-(4'-dodecylbiphenyl-4-yl)-3-(2-decyloxyphenyl)propane-1,3-dione					[297]			
	Sol/Nem	340.2	36.73	107.97	109.03				
	Nem/Liq	349.7	0.37	1.06					
$C_{43}H_{60}O_3$	1-(4'-dodecylbiphenyl-4-yl)-3-(3-decyloxyphenyl)propane-1,3-dione					[297]			
	Sol/Sol	327.7	10.42	31.80	191.66				
	Sol/SmeC	346.7	46.98	135.51					
$C_{43}H_{60}O_6$	SmeC/Liq	371.2	9.04	24.35	88.19	[227]			
	4'-(decyloxy)[1,1'-biphenyl]-4-yl 4-[{(2S)-2-(decyloxy)-1-oxopropoxy}-3-methylbenzoate								
	Sol/SmeC	330.2	23.82	72.14					
$C_{43}H_{60}O_7$	SmeC/Liq	381.2	6.12	16.05	64.90	[227]			
	Note: Enthalpies were determined from cooling measurements.								
	4'-(decyloxy)[1,1'-biphenyl]-4-yl 4-[{(2S)-2-(decyloxy)-1-oxopropoxy}-3-methoxybenzoate								
$C_{43}H_{61}BrO_4S$	Sol/SmeC	302.2	15.02	49.70	NA	[29]			
	SmeC/Liq	367.2	5.58	15.20					
	Note: Enthalpies were determined from cooling measurements.								
$C_{43}H_{63}NO_2$	2-bromo-6,7,10,11-tetrakis(pentyloxy)-3-(pentylsulfanyl)triphenylene					[323]			
	Sol/Chol	348.8	18.6	53.32	235.49				
	Chol/Liq	418.2	6.8	16.26					
$C_{43}H_{71}NO$	N-(2-hydroxy-4-dodecyloxybenzylidene)-4''-dodecylphenylaniline					[240]			
	Sol/SmeC	353.7	71.09	200.99	254.58				
	SmeC/SmeC	408.2	3.89	9.53					
$C_{43}H_{76}OS$	SmeC/Liq	472.7	11.80	24.97	NA	[155,312]			
	4-dodecyl-N-[[4-(octadecyloxy)phenyl]methylene]benzenamine								
	Sol/SmeC	351.4	64.69	184.09					
$C_{43}H_{76}O_2$	SmeC/Liq	366.0	25.80	70.49	170.46	[170]			
	cholesterol palmitate								
	Sol/Chol	350.5	58.58	167.13	196.7				
$C_{43}H_{79}NO_4$	Chol/Liq	354.8	1.18	3.33					
	Note: Liquid crystalline phase detected on cooling.								
	3,4,5-tris(dodecyloxy)benzamide								
$C_{44}H_{40}N_2O_5$	Sol/Meso	331.2	26.4	79.71	95.85	[378]			
	Meso/Liq	353.2	5.7	16.14					
$C_{44}H_{42}Cl_2N_2O_6$	4-{4-[(4-octyloxybenzoyloxy)benzoyloxy]phenyl}-6'-phenyl-2,2'-bipyridine					[171]			
	Sol/Nem	413.2	44.2	106.97	108.35				
	Nem/Liq	435.4	0.60	1.38					
$C_{44}H_{44}N_2O_6$	4,5-dichloro-1,3-phenylene bis[4-(4-pentyloxybenzylideneamino)benzoate]					[86]			
	Sol/Nem	409.2	37	90.42	94.33				
	Nem/Liq	435.2	1.7	3.91					
$C_{44}H_{44}N_2O_6$	1,3-phenylene bis[4-(4-pentyloxyphenyliminomethyl)benzoate]					[86]			
	Sol/Meso	444.2	16	36.02	107.17				
	Meso/SmeC	448.2	15	33.47					
$C_{44}H_{44}N_2O_6$	SmeC/Liq	451.2	17	37.68	NA	[86]			

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$\text{C}_{44}\text{H}_{44}\text{N}_2\text{O}_6$	Sol/Meso	423.2	30.9	73.02		[78]
	Meso/Liq	425.2	6.8	15.99	89.01	
$\text{C}_{44}\text{H}_{44}\text{N}_2\text{O}_8$	Sol/Meso	433.2	32.0	73.87		[412]
	Meso/Liq	452.2	16.0	35.38	109.25	
$\text{C}_{44}\text{H}_{46}\text{N}_2\text{O}_8$	Sol/Nem	384.2	41.5	108.02		[54]
	Nem/Liq	440.2	2.3	5.22	113.24	
$\text{C}_{44}\text{H}_{46}\text{O}_4$	Sol/Nem	378.2	26.0	68.75		[176]
	Nem/Liq	483.2	Not reported in paper			
$\text{C}_{44}\text{H}_{48}\text{N}_2$	Sol/SmeC	417.2	22.7	54.41		[258]
	SmeC/Nem	505.2	1.1	2.18		
	Nem/Liq	550.2	0.9	1.64	58.23	
$\text{C}_{44}\text{H}_{48}\text{N}_2\text{O}_4$	Sol/Sol	428.5	36.1	84.25		[94]
	SmeC/Nem	499.2	21.3	42.67		
	SmeC/Liq	553.5	8.5	15.36	142.28	
$\text{C}_{44}\text{H}_{48}\text{N}_2\text{O}_4$	Sol/Nem	478.2	44.3	92.64		[388]
	Nem/Liq	608.2	1.2	1.97	94.61	
$\text{C}_{44}\text{H}_{48}\text{N}_2\text{O}_6$	Sol/Sol	341.9	6.8	19.89		[94]
	Sol/Sol	477.1	8.5	17.82		
	SmeC/Nem	517.3	25.0	48.33		
	SmeC/Liq	574.4	10.8	18.80	104.84	
$\text{C}_{44}\text{H}_{50}\text{FeO}_4$	Sol/Nem	392.2	27.2	69.35		[376]
	Nem/Liq	409.2	0.69	1.69	71.04	
					NA	
$\text{C}_{44}\text{H}_{50}\text{N}_4\text{O}_{10}\text{S}_2$	Sol/SmeC	412.2	30.5	73.99		[113]
	SmeC/Nem	428.2	3.6	8.41		
	Nem/Liq	480.2	2.1	4.37	86.77	
$\text{C}_{44}\text{H}_{50}\text{O}_9$	Sol/SmeC	353.6	19.17	54.21		[401]
	SmeC/SmeC	433.7	0.18	0.42		
	SmeC/Liq	486.1	3.77	7.76	62.39	
$\text{C}_{44}\text{H}_{51}\text{NO}_{10}$	Sol/SmeC	366.9	29.2	79.59		[407]
	SmeC/Liq	436.1	9.6	22.01	101.60	
$\text{C}_{44}\text{H}_{52}\text{N}_2\text{O}_6$	Sol/Nem	433.2	63.0	145.43		[293]
	Nem/Liq	456.2	9.2	20.17	165.60	
$\text{C}_{44}\text{H}_{52}\text{N}_2\text{O}_6$	Sol/Nem	423.2	51.0	120.51		[293]
	Nem/Liq	436.2	7.5	17.19	137.70	

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{44}H_{52}N_2O_6$	Sol/Nem	397.2	41.0	103.22		[293]
	Nem/Liq	436.2	8.8	20.17	123.39	
$C_{44}H_{54}N_2O_2S$	Sol/SmeC	368.9	45.5	123.34		[373]
	SmeC/Liq	387.6	6.8	17.54	140.88	
$C_{44}H_{54}N_4O_6S_2$	Sol/SmeC	383.2	33.2	86.64		[23]
	SmeC/Liq	517.2	10.4	20.11	106.75	
$C_{44}H_{54}O_8$	Sol/SmeC	435.2	50.8	116.73		[271]
	SmeC/SmeC	485.8	2.2	4.53		
	SmeC/Liq	493.2	5.3	10.75	132.01	
$C_{44}H_{54}O_8$	Sol/Sol	338.7	12.7	37.50		[271]
	Sol/SmeC	380.1	40.6	106.81		
	SmeC/Liq	498.9	6.5	13.03	157.34	
$C_{44}H_{54}S_6$	Sol/SmeC	381.2	21.9	57.45		[303]
	SmeC/SmeC	554.2	34.0	61.35		
	SmeC/Liq	563.2	17.3	30.72	149.52	
$C_{44}H_{55}N_3O_6$	Sol/Sol	379.7	6.95	18.30		[109]
	Sol/Nem	399.2	51.11	128.03		
	Nem/Liq	494.7	1.32	2.67	149.00	
$C_{44}H_{56}F_2O_5$	Sol/SmeC	339.7	46.53	136.97		[213]
	SmeC/SmeC	371.7	Too small to be detected by dsc			
	SmeC/Liq	385.7	3.85	9.98	146.95	
$C_{44}H_{57}FO_5$	Sol/SmeC	339.5	49.86	146.86		[211]
	SmeC/SmeC	351.6	0.01	0.03		
	SmeC/SmeC	369.0	0.2	0.54		
	SmeC/Liq	383.2	4.69	12.24	159.67	
$C_{44}H_{57}FO_5$	Sol/SmeC	339.7	35.62	104.86		[146, 213]
	SmeC/Meso	368.1	0.16	0.43		
	Meso/Liq	378.1	3.63	9.60	114.89	
$C_{44}H_{57}FO_5$	Sol/SmeC	309.9	22.18	71.57		[214]
	SmeC/Liq	347.7	1.11	3.19		
	Liq/Liq	352.9	1.40	3.97	78.73	
$C_{44}H_{57}FO_5$	Sol/SmeC	319.2	24.64	77.19		[214]
	SmeC/Liq	347.5	1.23	3.54		
	Liq/Liq	352.7	1.63	4.62	85.35	
$C_{44}H_{57}FO_5$	Sol/SmeC	307.1	34.52	112.41		
	SmeC/SmeC	348.5	0.17	0.49		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{44}H_{57}FO_5$	Smec/Liq	358.1	5.41	15.11	128.01	[214]
	Sol/Smec	306.8	34.15	111.31		
	Smec/Smec	348.9	0.12	0.34		
	Smec/Liq	358.4	5.64	15.74	127.39	[214]
$C_{44}H_{57}F_3O_7$	(S)-1-ethylheptyl 4-[4-(4-tetradecyloxy-3-fluorobenzoyloxy)-3-fluorobenzoyloxy]-2-fluorobenzoate					
	Sol/Smec	327.7	31.61	96.46		
	Smec/Smec	338.3	0.02	0.06		
	Smec/Smec	350.1	0.33	0.94		
$C_{44}H_{58}O_2$	Smec/Liq	358.7	3.42	9.53	106.99	[80]
	4,4'''-didecyloxy-p-quaterphenyl					
	Sol/Sol	397.0	21.04	53.00		
	Sol/Sol	433.0	5.63	13.00		
	Sol/Sol	449.0	6.29	14.01		
$C_{44}H_{58}F_2O_2S$	Sol/Smec	558.0	12.83	22.99		
	Smec/Liq	589.0	12.96	22.00	125.00	230.8
	3,4-difluoro-2,5-bis[[4-(dodecyloxy)phenyl]ethynyl]thiophene					
	Sol/Nem	342.0	78.8	230.41		
	Nem/Liq	372.0	2.0	5.38	235.79	[373]
$C_{44}H_{58}F_2O_5S$	(S)-1-methylheptyl 5-[(2,3-difluoro-4-hexadecyloxybenzoyloxy-phenyl)ethynyl]thiophene-2-carboxylate					
	Sol/Smec	332.6	44.42	133.55		
	Smec/Smec	346.8	0.17	0.49		
	Smec/Liq	351.0	4.28	12.19	146.23	[267]
$C_{44}H_{58}N_2O_2S$	2-[4-(dodecyloxy)phenyl]-5-[[4-(dodecyloxy)phenyl]ethynyl]-3,4-thiophenedicarbonitrile					
	Sol/Smec	342.6	22.7	66.26		
	Smec/Liq	373.2	6.9	18.49	84.75	[373]
$C_{44}H_{58}O_7$	(S)-(+)1-methylheptyl 4-[4-(trans-4-dodecyloxy- $\alpha$ -methylcinnamoyloxy)benzoyloxy]benzoate					
	Sol/Smec	330.2	52.76	159.78		
	Smec/Smec	335.3	0.28	0.84		
	Smec/Smec	363.8	0.02	0.05		
	Smec/Smec	366.2	0.02	0.05		
$C_{44}H_{59}FO_5S$	Smec/Liq	386.2	5.44	14.09	174.81	[149]
	(S)-1-methylheptyl 5-[(3-fluoro-4-hexadecyloxybenzoyloxy-phenyl)ethynyl]thiophene-2-carboxylate					
	Sol/Smec	313.0	39.20	125.24		
	Smec/Smec	337.3	Not reported in paper			
	Smec/Liq	342.3	5.26	15.37	140.61	[267]
Note: Smec/Smec transition enthalpy is included in the Smec/Liq value.						
$C_{44}H_{60}O_5S$	(S)-1-methylheptyl 5-[(4-hexadecyloxybenzoyloxyphenyl)ethynyl]-thiophene-2-carboxylate					
	Sol/Smec	333.0	46.31	139.07		
	Smec/Smec	353.1	0.30	0.85		
$C_{44}H_{60}O_7$	Smec/Smec	357.0	5.52	15.46	155.38	[267]
	2-(1-oxooctyl)-1,4-phenylene 4-(octyloxy)benzoate					
	Sol/Nem	350.1	60.5	172.81		
$C_{44}H_{60}O_8$	Nem/Liq	360.0	1.1	3.06	175.87	[432]
	heptyl 2,5-bis[[4-(octyloxy)benzoyl]oxy]benzoate					
	Sol/Nem	327.6	29.0	88.52		
$C_{44}H_{61}ClO_5$	Nem/Liq	368.6	2.0	5.43	93.95	[432]
	(R)-1-methylheptyl 4'-(3"-chloro-4"-hexadecyloxybenzoyloxy)biphenyl-4-carboxylate					
	Sol/Smec	314.2	44.6	141.95		
	Smec/Smec	344.1	0.21	0.61		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{44}H_{61}ClO_7$	Smec/Meso	348.2	Could not be measured			[226]
	Meso/Liq	352.2	2.5	7.10	149.66	
$C_{44}H_{61}ClO_7$	Sol/Meso		2-chloro-4-methylpentanoic acid, 3,6,7,10,11-pentabutoxy-2-triphenylenyl ester			[324]
	Meso/Liq	473.9	21.8	46.00		
$C_{44}H_{61}NO_4S$	Sol/Meso		2-chloro-3-methylpentanoic acid, 3,6,7,10,11-pentabutoxy-2-triphenylenyl ester			[324]
	Meso/Liq	465.2	21.8	46.86		
$C_{44}H_{62}O_3$	Sol/Col	394.2	30.0	76.10		[29]
	Col/Liq	467.0	3.3	7.07	83.17	
$C_{44}H_{62}O_3$	Sol/Nem	336.2	37.2	110.65		[297]
	Nem/Liq	350.7	0.46	1.31	111.96	
$C_{44}H_{62}O_3$	Sol/Smec	348.2	64.68	185.76		[297]
	Smec/Liq	370.7	9.54	25.74	211.50	
$C_{44}H_{62}O_6$	Sol/Smec	372.2	38.47	103.36		[227]
	Smec/Liq	402.2	7.49	18.62	121.98	
Note: Enthalpies were determined from cooling measurements.						
$C_{44}H_{62}O_7$			4-(4-octyloxybenzoyloxy)phenyl (3,4-dioctyloxy)benzoate			
	Sol/Smec	368.7	10.5	28.48		[103]
$C_{44}H_{62}O_7$	Sme/Nem	370.8	1.72	4.64		
	Nem/Liq	380.4	0.51	1.34	34.46	
$C_{44}H_{62}O_7$	Sol/Smec	355.7	6.34	17.83		[103]
	Sme/Nem	369.7	1.82	4.92		
$C_{44}H_{64}N_2O_4$	Nem/Liq	378.4	0.31	0.82	23.57	
$C_{44}H_{64}N_2O_4$	Sol/Smec	382.0	81.9	214.40		[388]
	Smec/Liq	395.9	4.7	11.87	226.27	
$C_{44}H_{66}N_2O_5$			di-n-dodecyl 4,4'-azoxy- $\alpha$ -methylcinnamate			
	Sol/Smec	352.2	75.20	213.52		[393]
$C_{44}H_{66}N_2O_5$	Smec/Smec	355.7	0.10	0.28		
	Smec/Liq	360.7	8.80	24.40	238.20	
$C_{44}H_{68}O_4$			di(4'-dodecylphenyl)- <i>trans</i> -cyclohexane-1,4-dicarboxylate			
	Sol/Smec	341.8	57.0	166.76		[215]
$C_{44}H_{68}O_4$	Smec/Smec	392.4	6.41	16.34		
	Smec/Liq	403.9	8.71	21.56	204.66	
$C_{44}H_{68}O_6$			di(4'-dodecylphenyl)- <i>trans</i> -cyclohexane-1,4-dicarboxylate			
	Sol/Smec	363.2	38.71	106.58		[220]
$C_{44}H_{70}N_2O_6$	Smec/Smec	385.2	3.95	10.25		
	Smec/Smec	429.2	Not reported in paper			
$C_{44}H_{70}N_2O_6$	Smec/Liq	436.2	8.49	19.46	260.5	
	Sol/Col	327.6	32.3	98.60		[178]
	Col/Liq	343.8	2.44	7.10	105.70	

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{44}H_{73}NO$		4-hexadecyl-N-[4-(pentadecyloxy)phenyl]methylene]benzenamine				
	Sol/Smec	348.8	86.30	247.42		[240]
	Smec/Liq	363.9	25.06	68.87	316.29	
$C_{44}H_{78}OS$		cholesteryl thioheptadecanoate				
	Sol/Smec	339.2	42.7	125.8		[155,312]
	Smec/Chol	354.2	1.7	4.7		
$C_{44}H_{78}O_2$	Chol/Liq	358.2	0.8	2.3	132.9	
		cholesterol heptadecanoate				
	Sol/Chol	349.5	60.96	174.42		[166, 170]
$C_{44}H_{78}N_3O_6$	Chol/Liq	352.9	1.39	3.94	178.36	
		4-[3,4-bis(4-hexyloxybenzoyloxy)benzylideneamino]azobenzene				
	Sol/Nem	380.2	55.0	144.66		[205]
$C_{45}H_{47}N_3O_6$	Nem/Liq	409.2	0.4	0.98	145.64	
		4-[(4-butoxyphenyl)azo]phenyl 3-[[4-[(4-octyloxy)benzoyl]oxy]phenyl]methylene]amino]benzoate				
	Sol/Meso	424.5	19.4	45.70		[296]
$C_{45}H_{48}N_2O_8$	Meso/Liq	431.6	17.0	39.39	85.09	
		bis[4-ethoxyphenyl] 2-{5-[4-(4-butylphenylazo)phenoxy]pentyloxy}terephthalate				
	Sol/Nem	383.2	50.3	131.26		[54]
$C_{45}H_{51}NO_8$	Nem/Liq	418.2	2.4	5.74	137.00	
		3-cyanophenyl 2,5-bis(4-octyloxybenzoyloxy)benzoate				
	Sol/Smec	363.4	42.4	116.68		[407]
$C_{45}H_{51}NO_8$	Smec/Liq	374.1	3.1	8.29	124.97	
		3-cyanophenyl 2,5-bis(4-octyloxybenzoyloxy)benzoate				
	Sol/Smec	360.0	25.6	71.11		[407]
$C_{45}H_{54}N_2O_6$	Smec/Liq	423.0	8.1	19.15	90.26	
		$\alpha,\omega$ -bis[4-(4-heptyloxybenzoyloxy)benzylideneamino]propane				
	Sol/Smec	381.2	35.0	91.82		[293]
$C_{45}H_{54}N_2O_6$	Smec/Liq	386.2	17.0	36.26	128.08	
		$\alpha,\omega$ -bis[4-(4-pentyloxybenzoyloxy)benzylideneamino]heptane				
	Sol/Smec	367.2	34.0	92.59		[293]
$C_{45}H_{54}N_2O_6$	Smec/Liq	375.2	8.9	23.72	116.31	
		$\alpha,\omega$ -bis[4-(4-butoxybenzoyloxy)benzylideneamino]nonane				
	Sol/Nem	361.2	18.0	49.83		[293]
$C_{45}H_{54}N_2O_8$	Nem/Liq	391.2	0.8	2.04	51.87	
		malonic acid, bis{4-[4-(4-butoxyphenylimino)methyl]phenoxy}butyl ester				
	Sol/Nem	390.9	52.7	134.82		[244, 298]
$C_{45}H_{54}N_2O_8$	Nem/Liq	396.1	1.2	3.03	137.85	
		1,5-[4-(2-hydroxy-4-hexyloxybenzylideneamino)benzoyloxy]pentane				
	Sol/Smec	415.2	41.0	98.75		[412]
$C_{45}H_{54}N_2O_8$	Smec/Liq	420.2	13.0	30.94	129.69	
		1,5-[4-(2-hydroxy-4-pentyloxybenzylideneamino)benzoyloxy]-2,2-dimethylpentane				
	Sol/Smec	372.2	23.0	61.79		[412]
$C_{45}H_{54}N_2O_8$	Smec/Liq	408.2	12.0	29.40	91.19	
		1,5-[4-(2-hydroxy-4-pentyloxybenzylideneamino)benzoyloxy]-3,3-dimethylpentane				
	Sol/Smec	391.2	36.0	92.02		[412]
$C_{45}H_{54}N_2O_8$	Smec/Liq	426.2	12.0	28.16	120.18	

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{45}H_{56}F_2O_7$			4-[[4-[(2-ethoxy-1-methyl-2-oxoethoxy)carbonyl]phenyl]ethynyl]phenyl 2,3-difluoro-4-octadecyloxybenzoate			
	Sol/SmeC	349.3	52.6	150.59		
	SmeC/Nem	381.6	1.3	3.41		
$C_{45}H_{57}FO_7$	Nem/Liq	386.7	0.8	2.07	156.07	[268]
			4-[[4-[(2-ethoxy-1-methyl-2-oxoethoxy)carbonyl]phenyl]ethynyl]phenyl 3-fluoro-4-octadecyloxybenzoate			
	Sol/SmeC	334.8	35.2	105.44		
Note: Value is abnormally low compared to values for other compounds in this series.						
$C_{45}H_{58}N_2O_6$	SmeC/Nem	377.0	0.3	0.80		
	Nem/Liq	380.1	1.8	4.74	110.98	[268]
			$\alpha,\omega$ -bis{4-[(2-hydroxy-4-octyloxyphenyl)iminomethyl]phenyloxy}propane			
$C_{45}H_{58}O_7$	Sol/SmeC	427.1	36.3	84.99		
	SmeC/SmeC	431.9		Not detected by dsc		
	SmeC/Liq	456.8	9.8	21.45	106.44	[224]
$C_{45}H_{58}O_7$			4-[[4-[(2-ethoxy-1-methyl-2-oxoethoxy)carbonyl]phenyl]ethynyl]phenyl 4-octadecyloxybenzoate			
	Sol/SmeC	363.4	61.3	168.68		
	SmeC/Nem	392.1	0.04	0.10		
$C_{45}H_{59}FO_5$	Nem/Liq	394.6	3.2	8.11	176.89	[268]
			[(R)-1-methylheptyl] 4-[[2-fluoro-4-[(4-(pentadecyloxy)benzoyl)oxy]phenyl]ethynyl]benzoate			
	Sol/SmeC	333.2	47.30	141.96		
$C_{45}H_{59}FO_7S$	SmeC/SmeC	370.1	0.25	0.68		
	SmeC/Liq	381.2	4.53	11.88	154.52	[211]
			1-(butoxycarbonyl)ethyl 5-[(3-fluoro-4-octadecyloxybenzoyl-phenyl)ethynyl]thiophene-2-carboxylate			
$C_{45}H_{60}F_2O_7$	Sol/Meso	342.1	43.00	125.69		
	Meso/Liq	347.1	2.61	7.52	133.21	[267]
$C_{45}H_{60}N_4O_2$	Sol/SmeC	325.3	29.16	89.64		
	SmeC/Meso	367.8		Not reported in paper		
	Meso/Liq	370.2	5.58	15.07		[69]
$C_{45}H_{60}O_7$			(R)-1-methylheptyl 4-[4-(4-hexadecyloxy-3-fluorobenzoyloxy)-3-fluorobenzoyloxy]benzoate			
	Sol/Nem	387.2	60.52	156.30		
	Nem/Liq	409.2	3.33	8.14	164.44	[67]
$C_{45}H_{60}O_7$	Sol/SmeC	358.7	46.63	130.00		
	SmeC/SmeC	416.9	0.70	1.68		
	SmeC/Liq	424.2	2.42	5.70	137.38	[149]
$C_{45}H_{62}O_4S$			2-ethynyl-6,7,10,11-tetrakis(pentyloxy)3-(pentylsulphanyl)-triphenylene			
	Sol/Colh	330.9	24.3	73.44		
	Colh/Liq	398.6	2.8	7.02	80.46	[29]
$C_{45}H_{62}O_6S$			4-[[4-(hexadecyloxy)benzoyl]thio]benzoic acid, 4-[[1-methylheptyl)oxy]carbonyl]phenyl ester			
	Sol/SmeC	340.2	41.0	120.52		
	SmeC/SmeC	399.2	0.44	1.10		
$C_{45}H_{62}O_7$	SmeC/Liq	407.2	5.5	13.51	135.13	NA
			2-(1-oxononyl)-1,4-phenylene 4-(octyloxy)benzoate			
	Sol/Nem	354.0	62.1	175.42		
$C_{45}H_{62}O_7$	Nem/Liq	356.9	1.9	5.32	180.74	[432]
			2-(1-oxoheptyl)-1,4-phenylene 4-(nonyloxy)benzoate			
	Sol/Nem	333.6	51.7	154.98		
$C_{45}H_{62}O_7$	Nem/Liq	355.5	1.3	3.66	158.64	[432]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{45}H_{62}O_8$	Sol/Nem	octyl 2,5-bis[[4-(octyloxy)benzoyl]oxy]benzoate	324.8	35.5	109.30	[432]
			366.5	1.7	4.64	
	Nem/Liq				113.94	
$C_{45}H_{64}O_3$	Sol/Nem	1-(4'-dodecylbiphenyl-4-yl)-3-(2-dodecyloxyphenyl)propane-1,3-dione	336.7	32.55	96.67	[297]
			354.7	0.46	1.30	
	Nem/Liq				97.97	
$C_{45}H_{64}O_3$	Sol/SmeC	1-(4'-dodecylbiphenyl-4-yl)-3-(3-dodecyloxyphenyl)propane-1,3-dione	346.7	62.30	179.69	[297]
			370.7	9.12	24.60	
	SmeC/Liq				204.29	
$C_{45}H_{64}O_6$	Sol/SmeC	4'-(decyloxy)[1,1'-biphenyl]-4-yl 4-[{(2S)-2-(dodecyloxy)-1-oxopropoxy}-3-methylbenzoate	381.2	42.27	110.89	[227]
			372.2	3.29	8.84	
					119.73	
Note: Enthalpies were determined from cooling measurements.						
$C_{45}H_{64}O_7$	Sol/SmeC	4'-(decyloxy)[1,1'-biphenyl]-4-yl 4-[{(2S)-2-(dodecyloxy)-1-oxopropoxy}-3-methoxybenzoate	313.2	15.70	50.13	[227]
			371.2	5.45	14.68	
					64.81	
Note: Enthalpies were determined from cooling measurements.						
$C_{45}H_{66}O_6$	Sol/Col	2,6,10-tris(pentyloxy)-3,7,11-tris(butoxy)triphenylene	328.2	21.71	66.15	[261]
			402.2	10.09	25.09	
					91.24	
Independent values from another reference						
	Sol/Col	Not reported in paper				[368]
	Col/Liq	402.2	9.9		24.61	
$C_{45}H_{72}N_2O_3$	N-[4-[5-oxo-4-[(1-oxohexadecyl)amino]-1,3,6-cycloheptatrien-1-yl]phenyl]hexadecamide					
	Sol/Sol	403.2	1.4	3.47		[251]
	Sol/SmeC	415.2	41.5	99.95		
	SmeC/Liq	427.2	7.0	16.39	119.81	
$C_{45}H_{74}N_4O_6$	Sol/Meso	3,4-dihexadecyloxy cinnamic aldehyde-2',4'-dinitrophenylhydrazone	358.2	52.4	146.29	[13]
			388.2	2.3	0.59	
	Meso/Liq				146.88	
$C_{45}H_{74}N_4O_6$	Sol/Meso	3,4-dihexadecyloxybenzaldehyde-2',4'-dinitrophenyl hydrazone	397.2	59.4	149.55	[13]
			403.2	3.9	9.67	
	Meso/Liq				159.22	
$C_{45}H_{75}NO$	N-(4-hexadecyloxybenzylidene)-4-hexadecylaniline					
	Sol/SmeC	335.3	38.83	115.81		[65]
	SmeC/Liq	354.6	16.48	46.47	162.28	
$C_{45}H_{75}NO$	4-tetradecyl-N-[4-(octadecyloxy)phenyl]methylene]benzenamine					
	Sol/SmeC	355.0	46.31	130.45		[240]
	SmeC/Liq	364.8	15.22	41.72	172.17	
$C_{45}H_{80}OS$	cholesteryl thiooctadecanoate					
	Sol/SmeC	337.2	40.9	121.2		[155,312]
	SmeC/Chol	353.4	1.7	4.9		
	Chol/Liq	357.4	0.5	1.5	127.6	
$C_{45}H_{80}O_2$	cholesteryl octadecanoate					
	Sol/Liq	354.95	70.38	198.3	198.3	[170]
	Chol/SmeC	342.75	1.64	4.8		
	Chol/Liq	347.55	1.64	4.7		
Note: Liquid crystalline phase detected on cooling.						

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{45}H_{80}N_2O_4$	Sol/Meso	384.2	32.0	83.29		
	Meso/Meso	416.2	20.0	48.05		
	Meso/Liq	461.2	2.0	4.34	135.68	278
$C_{45}H_{81}N_3O_3$	Sol/Meso	456.2	14.0	30.69		
	Meso/Meso	462.2	7.0	15.14		
	Meso/Liq	615.2	11.0	17.88	63.71	[190]
$C_{45}H_{82}N_2O_2$	Sol/Meso	395.2	60.0	151.82		
	Meso/Nem	426.2	17.0	39.89		
	Nem/Liq	427.2	0.5	1.17	192.88	[36]
$C_{46}H_{44}N_2O_6$	Sol/Smec	312.8	2.8	8.95		
	Smec/Smec	386.6	47.2	122.09		
	Smec/Liq	406.0	9.8	24.14	155.18	[25]
$C_{46}H_{46}Cl_2N_2O_6$	Sol/Nem	403.2	48.0	119.05		
	Nem/Liq	430.2	2.0	4.65	123.70	[86]
	Sol/Nem	400.2	46.2	115.44		
$C_{46}H_{46}Cl_2N_2O_6$	Nem/Liq	438.2	1.8	4.11	119.55	[416]
$C_{46}H_{46}N_2O_9$	Sol/Nem	474.2	34.4	72.54		
	Nem/Liq	485.2	0.5	1.03	73.57	[289]
	Sol/Meso	393.2	14.68	37.33		
$C_{46}H_{46}O_8S_2$	Meso/Liq	433.2	12.0	27.70	65.03	NA
	Sol/Meso	430.2	20	46.49		
	Smec/Smec	444.2	16	36.02		
$C_{46}H_{48}N_2O_6$	Smec/Liq	448.2	20	44.62	127.13	[86]
	Sol/Meso	409.2	37.7	92.13		
	Meso/Meso	415.7	0.1	0.24		
$C_{46}H_{48}N_2O_6$	Meso/Liq	422.7	14.0	33.12	125.49	[78]
$C_{46}H_{48}N_2O_8$	Sol/Meso	416.2	11.0	26.43		
	Meso/Liq	447.2	17.0	38.01	64.44	[412]
	Sol/Nem	374.2	55.2	147.51		
$C_{46}H_{50}N_2O_8$	Nem/Liq	422.2	2.6	6.15	153.66	[54]
	Sol/Nem	427.2	68.19	159.62		
	Nem/Liq	442.2	3.3	7.46	167.08	[126]
$C_{46}H_{54}O_2$	Sol/Smec	483.2	29.20	60.43		
	Smec/Smec	506.2	0.35	0.69		
	Smec/Liq	526.2	5.08	9.65	70.77	[302]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{46}H_{54}O_4$						
	Sol/Sol	324.2	10	30.85		
	Sol/Sol	398.7	19	47.65		
	Sol/Smec	449.2	47	104.63		
	Smec/Meso	459.2	1.3	2.83		
	Meso/Nem	463.7	Not detected by dsc			[120]
$C_{46}H_{54}O_9$	Nem/Liq	469.2	8.7	18.54	204.50	
$C_{46}H_{54}O_9$						
$C_{46}H_{56}N_2O_6$	Sol/Smec	354.0	17.18	48.53		
	Smec/Smec	451.8	0.37	0.82		
	Smec/Liq	482.7	2.86	5.93	55.28	[401]
$C_{46}H_{56}N_2O_6$						
$C_{46}H_{56}N_2O_6$	Sol/Nem	421.2	65.0	154.32		
	Nem/Liq	448.2	9.3	20.75	175.07	[293]
$C_{46}H_{56}N_2O_6$						
$C_{46}H_{56}N_2O_6$	Sol/Nem	414.2	54.0	130.37		
	Nem/Liq	436.2	9.8	22.47	152.84	[293]
$C_{46}H_{56}N_2O_6$						
$C_{46}H_{56}N_2O_6$	Sol/Nem	398.2	58.0	145.66		
	Nem/Liq	424.2	8.6	20.27	165.93	[293]
$C_{46}H_{56}N_2O_8$						
$C_{46}H_{56}N_2O_8$	Sol/Sol	374.3	25.1	67.06		
	Smec/Smec	380.8	4.9	12.87		
	Smec/Liq	402.4	10.9	27.09	107.02	[298]
$C_{46}H_{58}N_4O_6$						
$C_{46}H_{58}O_2$						
$C_{46}H_{58}O_2$	Sol/Smec	510.2	32.02	62.76		
	Smec/Smec	551.2	0.48	0.87		
	Smec/Liq	560.2	6.43	11.48	75.11	[302]
$C_{46}H_{58}O_8$						
$C_{46}H_{58}O_8$	Sol/Sol	375.5	13.2	35.15		
	Smec/Smec	430.0	51.3	119.30		
	Smec/Smec	483.9	0.5	1.03		
	Smec/Liq	487.5	5.2	10.67	166.15	[271]
$C_{46}H_{58}O_8$						
$C_{46}H_{60}F_2O_5$						
$C_{46}H_{60}F_2O_5$	Sol/Smec	338.8	49.05	144.78		
	Smec/Smec	375.2	0.23	0.61		
	Smec/Liq	382.4	3.74	9.78	155.17	[213]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{46}H_{60}N_2O_6$	Sol/Sol	337.2	14.3	42.41		
	Sol/Smec	443.1	64.4	145.34		
	Smec/Smec	489.3	Not detected by dsc			
	Smec/Liq	514.6	21.9	42.56	230.31	[224]
$C_{46}H_{61}FO_5$	1-methylheptyl 4'-(4"-hexadecyloxy-3"-fluorobenzoyloxy)tolan-4-carboxylate					
	Sol/Smec	334.2	50.41	150.84		
	Smec/Smec	368.8	0.30	0.81		
	Smec/Liq	374.8	5.06	13.50	165.15	[213]
$C_{46}H_{62}N_4O_2$	$\alpha$ -(4'-pentylazobenzene-4-oxy)- $\omega$ -(4'-pentylazobenzene-4-oxy)dodecane					
	Sol/Nem	396.2	65.89	166.30		
	Nem/Liq	419.2	8.26	19.70	186.00	[67]
$C_{46}H_{62}O_2$	4,4'''-diundecyloxy-p-quaterphenyl					
	Sol/Sol	409.0	29.45	72.00		
	Sol/Sol	427.0	6.83	16.00		
	Sol/Sol	447.0	6.26	14.00		
	Smec/Smec	549.0	13.73	25.01		
$C_{46}H_{62}O_7$	Smec/Liq	579.0	14.48	25.01	152.02	[111]
	(S)-(+) -1-methylheptyl 4-[4-(trans-4-tetradecyloxy- $\alpha$ -methylcinnamoyloxy)benzoyloxy]benzoate					
	Smec/Smec	337.4	41.23	122.20		
	Smec/Liq	367.2	0.15	0.41		
$C_{46}H_{64}O_5S$	Smec/Liq	383.2	4.28	11.17	133.78	[149]
	(S)-1-methylheptyl 5-[(4-octadecyloxybenzoyloxyphenyl)ethynyl]-thiophene-2-carboxylate					
	Smec/Smec	319.3	35.22	110.30		
	Smec/Smec	351.2	0.22	0.63		
$C_{46}H_{64}O_7$	Smec/Smec	354.7	4.43	12.49	123.42	[267]
	2-(1-oxodecyl)-1,4-phenylene 4-(octyloxy)benzoate					
	Smec/Smec	344.7	56.2	163.04		
$C_{46}H_{64}O_7$	Smec/Smec	357.9	1.3	3.63	166.67	[432]
	2-(1-oxooctyl)-1,4-phenylene 4-(nonyloxy)benzoate					
	Smec/Smec	347.2	76.3	219.75		
$C_{46}H_{64}O_8$	Smec/Smec	356.0	1.2	3.37	223.12	[432]
	nonyl 2,5-bis[[4-(octyloxy)benzoyl]oxy]benzoate					
	Smec/Smec	323.4	41.5	128.32		
$C_{46}H_{65}ClO_5$	Smec/Smec	363.4	2.0	5.50	133.82	[432]
	(R)-1-methylheptyl 4'-(3"-chloro-4"-octadecyloxybenzoyloxy)biphenyl-4-carboxylate					
	Smec/Smec	316.6	47.3	149.40		
	Smec/Smec	344.2	0.23	0.67		
$C_{46}H_{66}O_7$	Smec/Meso	350.4	Could not be measured			
	Meso/Liq	352.2	2.6	7.38	157.45	[226]
	4-(4-dodecyloxybenzoyloxy)phenyl (3,4-diheptyloxy)benzoate					
	Smec/Meso	359.5	7.79	21.67		
$C_{46}H_{67}NO_4$	Smec/Meso	373.5	1.91	5.11		
	Nem/Liq	376.7	0.32	0.85	27.63	[103]
	4-(3-decyloxyphenyliminomethyl)phenyl 4-hexadecyloxybenzoate					
$C_{46}H_{67}NO_4$	Smec/Smec	342.2	79.22	231.50		
	Smec/Smec	344.8	6.66	19.32	250.82	[175]
$C_{46}H_{70}N_2O_6$	N,N'-didecanoyl-2,4-bis(decanoxyloxy)-1,3-benzenediamine					
	Smec/Smec	366.2	18.0	49.15		
	Smec/Smec	395.2	17.0	43.02	92.17	300.8
	Smec/Smec					[189]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)						
$\text{C}_{46}\text{H}_{82}\text{OS}$		cholesteryl thiononadecanoate						
	Sol/Smec	347.2	41.6		119.8			
	Smec/Chol	351.5	1.7		4.8			
$\text{C}_{46}\text{H}_{82}\text{O}_2$	Chol/Liq	355.6	0.5		1.5	126.1	NA	[155,312]
		cholesteryl nonadecanoate						
	Sol/Liq	353.55	73.3		207.3	207.3	218	[170]
$\text{C}_{46}\text{H}_{84}\text{N}_2\text{O}_2$	Chol/Smec	344.95	1.9		5.5			
	Chol/Liq	348.75	1.7		4.8			
	Note: Liquid crystalline phase detected on cooling.							
$\text{C}_{46}\text{H}_{84}\text{N}_2\text{O}_2$		$\text{N,N}'\text{-dioctadecanoyl-2,3,5,6-tetramethylbenzene-1,4-diamine}$						
	Sol/Meso	386.2	21.0		54.38			
	Meso/Liq	489.2	24.0		49.06	103.44		[36]
$\text{C}_{47}\text{H}_{36}\text{FNO}_8$		4-cyanophenyl 4-[3-[4-(4'-hexylbiphenyl-4-carbonyloxy)-2-fluorobenzoyloxy]benzoyloxy]benzoate						
	Sol/Nem	448.7	57.8		128.82			
	Nem/Liq	470.4	0.42		0.89	129.71		[411]
$\text{C}_{47}\text{H}_{36}\text{FNO}_8$		4-cyanophenyl 4-[3-[4-(4'-hexylbiphenyl-4-carbonyloxy)-3-fluorobenzoyloxy]benzoyloxy]benzoate						
	Sol/Nem	432.7	54.0		124.80			
	Nem/Liq	470.7	0.59		1.25	126.05		[411]
$\text{C}_{47}\text{H}_{37}\text{NO}_8$		4-cyanophenyl 4-[3-[4-(4'-hexylbiphenyl-4-carbonyloxy)benzoyloxy]benzoyloxy]benzoate						
	Sol/Nem	429.2	46.7		108.81			
	Nem/Liq	477.2	0.48		1.01	109.82		[411]
$\text{C}_{47}\text{H}_{44}\text{FNO}_9$		3-[4-[4-(cyanophenoxy)carbonyl]phenoxy]carbonylphenyl 3-fluoro-4-[[4-(dodecyloxy)benzoyl]oxy]benzoate						
	Sol/Smec	392.7	80.68		205.45			
	Smec/Liq	393.2	3.86		9.82	215.27		[248]
$\text{C}_{47}\text{H}_{44}\text{N}_2\text{O}_8$		bis[4-ethoxyphenyl] 2-[5-[4-(4-biphenylazo)phenoxy]pentyloxy]terephthalate						
	Sol/Nem	417.2	47.0		112.66			
	Nem//Liq	463.2	3.0		6.48	119.14		[54]
$\text{C}_{47}\text{H}_{45}\text{NO}_9$		4-[4-(cyanophenoxy)carbonyl]phenyl 3-[4-[4-(dodecyloxy)benzoyl]oxy]benzoyloxy]benzoate						
	Sol/Smec	402.7	41.11		102.09			
	Smec/Smec	403.7	0.17		0.42			
$\text{C}_{47}\text{H}_{47}\text{N}_3\text{O}_6$	Smec/Liq	404.0	3.48		8.61	111.12		[348]
		2-cyano-1,3-phenylene bis[4-(4-hexyloxyphenyliminomethyl)benzoate]						
	Sol/Meso	450.2	14.5		32.21			
$\text{C}_{47}\text{H}_{47}\text{N}_3\text{O}_6$	Meso/Liq	463.2	26.7		57.64	89.85		[246]
		4-cyano-1,3-phenylene bis[4-(4-hexyloxyphenyliminomethyl)benzoate]						
	Sol/Meso	395.2	32.2		81.48			
$\text{C}_{47}\text{H}_{51}\text{N}_3\text{O}_6$	Meso/Smec	406.2	1.8		4.43			
	Smec/Nem	429.2	0.6		1.40			
	Nem/Liq	438.2	0.9		2.05	89.36		[252]
$\text{C}_{47}\text{H}_{51}\text{N}_3\text{O}_6$		4-[3,4-bis(4-heptyloxybenzoyloxy)benzylideneamino]azobenzene						
	Sol/Nem	383.2	61.0		159.19			
	Nem/Liq	405.2	0.4		0.99	160.18		[205]
$\text{C}_{47}\text{H}_{51}\text{N}_3\text{O}_6$		4-[4-hexyloxyphenyl]azo]phenyl 3-[4-[4-(octyloxy)benzoyl]oxy]phenyl)methylene]amino]benzoate						
	Sol/Meso	420.1	25.8		61.41			
	Meso/Liq	430.4	22.9		53.21	114.62		[296]
$\text{C}_{47}\text{H}_{56}\text{O}_4$		1,7-bis(4-(4-hexyloxyphenylethyanyl)phenoxy)heptane						
	Sol/Sol	312.2	9.2		29.46			

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{47}H_{58}N_2O_6$	Sol/Nem	416.2	63.0	151.37		[120]
	Nem/Liq	430.2	3.4	7.90	188.73	
$C_{47}H_{58}N_2O_6$	$\alpha,\omega$ -bis[4-(4-octyloxybenzoyloxy)benzylideneamino]propane					
	Sol/Smec	377.2	50.0	132.56		[293]
	Smec/Liq	383.2	18.0	46.97	179.53	
$C_{47}H_{58}N_2O_6$	$\alpha,\omega$ -bis[4-(4-heptyloxybenzoyloxy)benzylideneamino]pentane					
	Sol/Smec	382.2	49.0	128.21		[293]
	Smec/Liq	392.2	15.0	28.25	156.46	
$C_{47}H_{58}N_2O_6$	$\alpha,\omega$ -bis[4-(4-hexyloxybenzoyloxy)benzylideneamino]heptane					
	Sol/Smec	370.2	35.0	94.54		[293]
	Smec/Liq	387.2	13.0	33.57	128.11	
$C_{47}H_{58}N_2O_6$	$\alpha,\omega$ -bis[4-(4-pentyloxybenzoyloxy)benzylideneamino]nonane					
	Sol/Nem	371.2	55.0	148.17		[293]
	Nem/Liq	383.2	0.7	1.83	150.00	
$C_{47}H_{58}N_2O_8$	malonic acid, bis{5-[4-[(4-butoxyphénylimino)methyl]phenoxy]pentyl} ester					
	Sol/Nem	396.2	63.9	161.28		[244]
	Nem/Liq	400.4	2.0	5.00	166.28	
$C_{47}H_{58}N_2O_8$	ethylpropanedioic acid, bis[4-[4-[(E)-[(4-butoxyphenyl)imino]methyl]phenoxy]butyl] ester					
	Sol/Sol	382.5	40.4	105.62		[298]
	Smec/Liq	402.2	13.9	34.56	140.18	
$C_{47}H_{58}N_2O_8$	1,5-[4-(2-hydroxy-4-hexyloxybenzylideneamino)benzoyloxy]-2,2-dimethylpentane					
	Sol/Smec	370.2	24.0	64.83		[412]
	Smec/Liq	408.2	12.0	29.40	94.23	
$C_{47}H_{58}N_2O_8$	1,5-[4-(2-hydroxy-4-hexyloxybenzylideneamino)benzoyloxy]-3,3-dimethylpentane					
	Sol/Smec	385.2	33.0	85.67		[412]
	Smec/Liq	424.2	12.0	28.29	113.96	
$C_{47}H_{62}N_2O_6$	$\alpha,\omega$ -bis{4-[(2-hydroxy-4-octyloxyphenyl)iminomethyl]phenoxy}pentane					
	Sol/Smec	427.7	73.8	172.55		[224]
	Smec/Liq	454.8	11.1	24.41	196.96	
$C_{47}H_{63}FO_7S$	1-(butoxycarbonyl)ethyl 5-[(3-fluoro-4-eicosyloxybenzoylphenyl)-ethynyl]thiophene-2-carboxylate					
	Sol/Meso	341.8	52.0	152.14		[267]
	Meso/Liq	347.0	3.85	11.10 1	63.24	
$C_{47}H_{64}F_2O_7$	(R)-1-methylheptyl 4-[4-(4-octadecyloxy-3-fluorobenzoyloxy)-3-fluorobenzoyloxy]benzoate					
	Sol/Smec	323.8	31.28	96.60		[69]
	Smec/Meso	366.9	0.23	0.63		
$C_{47}H_{64}N_4O_5$	369.8	5.00	13.52	110.75		
	Sol/Smec	423.5	52.61	124.23		[230]
	Smec/Liq	446.2	12.08	27.07	151.30	
$C_{47}H_{64}O_7$	(S)-(+)-1-methylheptyl 4-[4-(trans-4-hexadecyloxy-4-cinnamoyloxy)benzoyloxy]benzoate					
	Sol/Smec	356.2	32.93	92.45		[149]
	Smec/Smec	414.2	1.25	3.02		
$C_{47}H_{65}NO_{10}$	Smec/Liq	420.2	1.77	4.21	99.68	
	Sol/Meso	Not reported in paper				
	Col/Col	383.2	13.4	34.97		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)						
$C_{47}H_{66}O_6S$	Col/Liq	436.4	14.0		32.08			[363]
	Sol/SmeC	335.2	41		122.32			
	SmeC/SmeC	397.2	0.26		0.65			
	SmeC/Liq	405.2	5.3		13.08	136.05		[85]
$C_{47}H_{66}O_7$	2-(1-oxoundecyl)-1,4-phenylene 4-(octyloxy)benzoate							
	Sol/Nem	349.1	53.2		152.39			[432]
	Nem/Liq	356.0	1.2		3.27	155.76		
$C_{47}H_{66}O_7$	2-(1-oxononyl)-1,4-phenylene 4-(nonyloxy)benzoate							
	Sol/Nem	349.8	44.20		126.36			[432]
	Nem/Liq	355.5	0.64		1.80	128.16		
$C_{47}H_{66}O_8$	decyl 2,5-bis[[4-(octyloxy)benzoyl]oxy]benzoate							
	Sol/Nem	326.8	58.8		179.93			[432]
	Nem/Liq	363.6	1.6		4.40	184.33		
$C_{47}H_{79}NO$	N-(4-octadecyloxybenzylidene)-4-hexadecylaniline							
	Sol/Meso	329.5	55.21		167.56			[65]
	Meso/Liq	360.0	30.88		85.78	253.33		
$C_{47}H_{79}NO$	4-hexadecyl-N-[[4-(octadecyloxy)phenyl]methylene]benzenamine							
	Sol/SmeC	361.3	115.30		319.13			[240]
	SmeC/Liq	363.7	Not reported in paper					
Note: SmeC/Liq transition enthalpy is likely included in the Sol/SmeC value.								
$C_{47}H_{84}OS$	cholesteryl thioeicosanoate							
	Sol/SmeC	344.2	47.3		137.4			
	SmeC/Chol	350.3	1.8		5.0			
$C_{48}H_{42}N_4O_4$	Sol/SmeC	468.2	38.6		82.44			
	SmeC/Liq	476.2	9.8		20.58	103.02		[331]
	m-terphenyl-4,4"-diyl bis[4-(5-butylpyrimidin-2-yl)benzoate]							
$C_{48}H_{46}FNO_9$	Sol/SmeC	393.2	82.75		210.45			
	SmeC/SmeC	396.6	0.02		0.05			
	SmeC/Liq	398.7	4.22		10.58	221.08		[248]
$C_{48}H_{46}N_2O_8$	Sol/Nem	403.2	50.0		124.01			
	Nem/Liq	466.2	3.7		7.94	131.95		[54]
	bis[4-ethoxyphenyl] 2-{6-[4-(4-biphenylazo)phenoxy]hexyloxy}terephthalate							
$C_{48}H_{47}NO_9$	Sol/SmeC	402.2	80.86		201.04			
	Note: Sol/Sol transition enthalpy is included in the Sol/SmeC value.							
	SmeC/SmeC	408.2	0.09		0.22			
$C_{48}H_{47}NO_9$	SmeC/Liq	408.7	4.24		10.37	211.63		[248]
	4-[(4-cyanophenoxy)carbonyl]phenyl 3-[[4-[(4-(tridecyloxy)benzoyl)oxy]benzoyl]oxy]benzoate							
	Sol/Sol	369.0	4.0		10.84			
$C_{48}H_{48}N_2O_5$	Sol/Nem	404.4	34.4		85.06			
	Nem/Liq	417.9	0.60		1.44	97.34		[171]
	4-{{4-[(4-dodecyloxybenzoyloxy)benzoyloxy]phenyl}-6'-phenyl-2,2'-bipyridine}							
$C_{48}H_{50}Cl_2N_2O_6$	Sol/Nem	418.2	43.0		102.82			
	Nem/Liq	419.2	1.5		3.58	106.40		[86]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{48}H_{50}Cl_2N_2O_6$	Sol/Nem	388.2	35.9	92.48	95.53	[416]
	Nem/Liq	426.2	1.3	3.05		
$C_{48}H_{50}N_2O_9$	Sol/Nem	460.2	46.7	101.48	102.34	[289]
	Nem/Liq	465.2	0.4	0.86		
$C_{48}H_{50}O_8S_2$		1,3-phenylene 4-[4-(heptyloxy)benzoyl]thio]benzoate				
	Sol/Meso	393.2	13.9	35.35	NA	[318]
	Meso/Meso	412.2	0.18	0.44		
$C_{48}H_{52}N_2O_6$	Meso/Liq	419.2	14.0	33.40		
		1,3-phenylene bis[4-(4-heptyloxyphenylminomethyl)benzoate]				
	Sol/Meso	429.2	21	48.93	126.08	[86]
$C_{48}H_{52}N_2O_8$	Meso/SmeC	436.2	12	27.51		
	SmeC/Liq	443.2	22	49.64		
		N,N'-bis[4-(4-heptyloxybenzoyloxy)benzylidene]-phenylene-1,3-diamine				
$C_{48}H_{52}N_2O_6$	Sol/Sol	379.7	13.7	36.08	127.02	[78]
	Sol/Meso	387.2	20.5	52.94		
	Meso/Liq	413.2	15.7	38.00		
$C_{48}H_{52}N_2O_8$		1,3-bis[4-(2-hydroxy-4-heptyloxybenzylideneamino)benzoyloxy]benzene				
	Sol/Meso	401.2	20.0	49.85	87.91	[412]
	Meso/Liq	446.7	17.0	38.06		
$C_{48}H_{58}N_2O_4$		N,N'-didodecyl-3,4,9,10-perylenebis(carboxamide)				
	Sol/Sol	360.2	6.8	18.88	132.31	[263]
	Meso/Meso	434.2	14.4	33.16		
	Meso/Meso	453.2	13.8	30.45		
	Meso/Meso	486.2	1.2	2.47		
	Meso/Liq	619.2	20.1	32.46		
$C_{48}H_{58}O_4$		<i>bis</i> (4-decylbutadiinylphenyl) terephthalate				
	Sol/Nem	376.9	28.0	74.29	Not reported in paper	[176]
	Nem/Liq	487.2				
$C_{48}H_{58}O_4$		1,8-bis(4-(4'-hexyloxyphenylethynyl)phenoxy)octane				
	Sol/Sol	322.2	9.9	30.73	216.83	[120]
	Sol/Sol	420.2	31	73.77		
	Sol/Nem	434.2	40	92.12		
	Nem/Liq	450.2	9.1	20.21		
$C_{48}H_{58}O_9$		4-[[(1S)-2-butoxy-1-methyl-2-oxoethoxy]carbonyl]phenyl 4-[[2(E)-3-[6-(tetradecyloxy)-2-naphthyl]-1-oxo-2-propenyl]oxy]benzoate				
	Sol/SmeC	351.4	21.50	61.18	68.23	[401]
	SmeC/SmeC	445.3	0.31	0.70		
	SmeC/Liq	489.4	3.11	6.35		
$C_{48}H_{60}N_2O_6$		$\alpha,\omega$ -bis[4-(4-octyloxybenzoyloxy)benzylideneamino]butane				
	Sol/Nem	395.2	60.0	151.82	173.53	[293]
	Nem/Liq	442.2	9.6	21.71		
$C_{48}H_{60}N_2O_6$		$\alpha,\omega$ -bis[4-(4-heptyloxybenzoyloxy)benzylideneamino]hexane				
	Sol/Nem	411.2	61.0	148.35	171.18	[293]
	Nem/Liq	429.2	9.8	22.83		
$C_{48}H_{60}N_2O_6$		$\alpha,\omega$ -bis[4-(4-hexyloxybenzoyloxy)benzylideneamino]octane				
	Sol/Nem	407.2	51.0	125.25	146.99	[293]
	Nem/Liq	423.2	9.2	21.74		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)						
$C_{48}H_{60}N_2O_6$	Sol/Nem	$\alpha,\omega$ -bis[4-(4-pentyloxybenzoyloxy)benzylideneamino]decane		384.2	44.0	114.52		[293]
	Nem/Liq	413.2	8.1		19.60	134.12		
$C_{48}H_{62}N_4O_6$	Sol/SmeC	374.7	48.5		129.44			[107]
	SmeC/Nem	419.1	7.5		17.90			
	Nem/Liq	420.2	12.5		29.75	177.09		
$C_{48}H_{62}O_6$	Sol/Col	420.9	27.7		65.81			[6]
	Col/Liq	510.1	2.9		5.69	71.50		
$C_{48}H_{63}N_9O_3$	Sol/SmeC	441.1	38.72		87.78			[187]
	SmeC/Liq	481.5	4.71		9.78	97.56		
$C_{48}H_{64}F_2O_5$	Sol/SmeC	341.2	48.58		142.37			[213]
	SmeC/SmeC	376.2	0.20		0.53			
	SmeC/Liq	380.2	3.80		9.99	152.89		
$C_{48}H_{64}N_2O_6$	Sol/SmeC	413.2	33.5		81.07			[115]
	SmeC/Nem	533.2	1.67		3.13			
	Nem/Liq	538.2	2.03		3.77	87.97		
$C_{48}H_{64}N_2O_6$	Sol/Sol	342.8	14.4		2.01			[224]
	Sol/SmeC	436.9	76.4		174.87			
	SmeC/SmeC	456.9	Not detected by dsc					
	SmeC/Liq	486.3	21.0		43.18	260.06		
$C_{48}H_{65}FO_5$	Sol/SmeC	316.7	49.65		156.77			[211]
	SmeC/SmeC	368.6	0.23		0.62			
	SmeC/Liq	375.6	3.92		10.43	167.82		
$C_{48}H_{65}FO_5$	Sol/SmeC	340.2	49.65		145.94			[146]
	SmeC/Meso	366.2	0.21		0.57			
	Meso/Liq	370.2	4.05		10.94	157.45		
$C_{48}H_{65}FO_5$	Sol/SmeC	340.2	49.65		145.95			[213]
	SmeC/SmeC	366.2	0.21		0.57			
	SmeC/Liq	370.2	4.05		10.94	157.46		
$C_{48}H_{66}F_2O_5$	Sol/Nem	414.2	42.7		103.09			[333]
	Note: Sol/Nem transition enthalpy includes the enthalpy of a Sol/Sol transition.							
	Nem/Liq	Decomposed prior to transition						
$C_{48}H_{66}O_2$	$4,4''$ -didecyloxy-p-quaterphenyl							
	Sol/Sol	403.0	27.40		67.99			
	Sol/Sol	422.0	6.75		16.00			
	Sol/Sol	446.0	5.80		13.00			
	Sol/SmeC	545.0	14.17		26.00			
$C_{48}H_{68}O_5S$	SmeC/Liq	573.0	14.90		26.00	148.99	259.2	[111]
	2-methyl-4-[6,7,10,11-tetrakis(pentyloxy)-3-(pentylsulphanyl)-2-triphenylenyl]-3-butyn-2-ol							

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$\text{C}_{48}\text{H}_{68}\text{O}_7$	Sol/Col	355.0	23.8	67.04	80.19	[29]
	Col/Liq	441.0	5.8	13.15		
$\text{C}_{48}\text{H}_{68}\text{O}_7$	2-(1-oxododecyl)-1,4-phenylene 4-(octyloxy)benzoate					
	Sol/Nem	347.4	55.1	158.61	162.56	[432]
	Nem/Liq	354.0	1.4	3.95		
$\text{C}_{48}\text{H}_{68}\text{O}_7$	2-(1-oxodecyl)-1,4-phenylene 4-(nonyloxy)benzoate					
	Sol/Nem	345.8	59.5	172.06	175.73	[432]
	Nem/Liq	354.2	1.3	3.67		
$\text{C}_{48}\text{H}_{70}\text{O}_4$	cholesterly 5-[4-(4'-butoxy)biphenyloxy]pentanoate					
	Sol/Nem	377.8	18.35	48.57	50.92	[74]
	Nem/Liq	426.3	1.00	2.35		
$\text{C}_{48}\text{H}_{70}\text{O}_7$	4-(4-dodecyloxybenzoyloxy)phenyl (3,4-dioctyloxy)benzoate					
	Sol/Smec	357.7	9.16	25.61	31.74	[103]
	Smec/Nem	374.8	1.92	5.12		
$\text{C}_{48}\text{H}_{71}\text{BrO}_5$	Nem/Liq	376.8	0.38	1.01		
	2-cyano-3,6,7,10,11-pentakis(hexyloxy)triphenylene					
	Sol/Sol	348.2	6.79	19.50	117.83	[364]
$\text{C}_{48}\text{H}_{72}\text{N}_2$	Sol/Col	366.2	28.66	78.26		
	Col/Liq	488.2	9.80	20.07		
	terephthalylidene-bis-(tetradecylaniline)					
$\text{C}_{48}\text{H}_{72}\text{O}_6$	Sol/Smec	359.0	65.40	182.17	227.30	[116]
	Smec/Smec	417.2	7.17	17.19		
	Smec/Liq	442.7	12.37	27.94		
$\text{C}_{48}\text{H}_{72}\text{O}_6$	2,3,6,7,10,11-hexakis(pentyloxy)triphenylene				122.96	[143]
	Sol/Col	341.3	33.94	99.44		
	Col/Liq	395.8	9.31	23.52		
$\text{C}_{48}\text{H}_{72}\text{O}_6$	Independent values from another reference				121.51	[261]
	Sol/Col	342.2	34.48	100.76		
	Col/Liq	395.2	8.20	20.75		
$\text{C}_{48}\text{H}_{72}\text{O}_6$	Independent values from another reference				20.75	[368]
	Sol/Col	Not reported in paper				
	Col/Liq	395.2	8.2	20.75		
$\text{C}_{48}\text{H}_{72}\text{O}_6$	2,6,10-tris(hexyloxy)-3,7,11-tris(butoxy)triphenylene				120.77	[261]
	Sol/Col	331.2	38.95	117.60		
	Col/Liq	369.2	1.17	3.17		
$\text{C}_{48}\text{H}_{72}\text{O}_6$	Independent values from another reference				3.25	[368]
	Sol/Col	Not reported in paper				
	Col/Liq	369.2	1.2	3.25		
$\text{C}_{48}\text{H}_{72}\text{O}_{12}$	2,3,6,7-tetrakis(pentyloxy)-1,5-bis[2-[(tetrahydro-2H-pyran-2-yl)oxy]ethoxy]-9,10-anthracenedione				100.77	[275]
	Sol/Col	352.7	21.96	62.26		
	Col/Liq	374.7	14.43	38.51		
$\text{C}_{48}\text{H}_{76}\text{O}_6$	di(4'-tetradecyloxyphenyl) <i>trans</i> -cyclohexane-1,4-dicarboxylate				183.31	[220]
	Sol/Smec	367.2	52.89	144.04		
	Smec/Smec	387.2	6.41	16.55		
	Smec/Smec	428.2	Not reported in paper			
$\text{C}_{48}\text{H}_{78}\text{O}_{12}$	Smec/Liq	429.2	9.75	22.72	183.31	[220]
	Sol/Sol	129	1.1	8.5		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.		
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$				
$\text{C}_{48}\text{H}_{80}\text{N}_2\text{O}_{10}$	Sol/Meso	353.8	32.2	91.1	319.8	[342]		
	Meso/Liq	359.3	21.5	59.9				
$\text{C}_{48}\text{H}_{82}\text{N}_2\text{O}_3$	N,N'-diheptanoyl-2,3,5,6-tetrakis(heptanoyloxy)-1,4-benzenediamine					[188]		
	Sol/Disc	337.2	28.0	83.04	166.17			
	Disc/Liq	481.2	40.0	83.13				
$\text{C}_{48}\text{H}_{86}\text{N}_2\text{O}_4$	4,4'-dioctadecyloxyazoxybenzene					[181]		
	Sol/SmeC	367.3	79.0	215.08	301.62			
	SmeC/SmeC	372.2	10.56	28.37				
$\text{C}_{49}\text{H}_{38}\text{FNO}_8$	SmeC/Liq	388.5	22.6	58.17	124.88	[193]		
	N,N'-ditridecanoyl-2,5,6-trimethyl-4-tridecanoxyloxy-1,3-benzenediamine							
	Sol/Meso	390.7	Not reported in paper					
$\text{C}_{49}\text{H}_{38}\text{FNO}_8$	Meso/Meso	410.2	21.0	51.19	[411]	[411]		
	Meso/Liq	458.2	2.0	4.36				
	Sol/Nem	428.2	53.1	124.01				
$\text{C}_{49}\text{H}_{41}\text{NO}_8$	Nem/Liq	458.1	0.4	0.87	94.46	[411]		
	4-cyanophenyl 4-[3-[4-(4'-octylbiphenyl-4-carbonyloxy)-2-fluorobenzoyloxy]benzoyloxy}benzoate							
	Sol/Nem	430.2	40.2	93.44				
$\text{C}_{49}\text{H}_{48}\text{FNO}_9$	Nem/Liq	459.6	0.47	1.02	114.23	[411]		
	4-cyanophenyl 4-[3-[4-(4'-octylbiphenyl-4-carbonyloxy)benzoyloxy]benzoyloxy}benzoate							
	Sol/Nem	443.2	50.2	113.27				
$\text{C}_{49}\text{H}_{48}\text{N}_2\text{O}_8$	Nem/Liq	468.2	0.45	0.96	229.00	[248]		
	3-[4-[4-(4-cyanophenoxy)carbonyl]phenoxy]carbonyl]phenyl 3-fluoro-4-[[4-(tetradecyloxy)benzoyl]oxy]-benzoate							
	Sol/SmeC	393.7	85.46	217.07				
$\text{C}_{49}\text{H}_{49}\text{NO}_9$	SmeC/SmeC	399.4	0.04	0.10	87.21	[54]		
	SmeC/Liq	403.2	4.77	11.83				
	Sol/Nem	416.2	35.1	84.33				
$\text{C}_{49}\text{H}_{51}\text{N}_3\text{O}_6$	Nem/Liq	451.2	1.3	2.88	203.86	[248]		
	Note: Sol/Sol transition enthalpy is included in Sol/SmeC value.							
	SmeC/SmeC	411.2	0.13	0.32				
$\text{C}_{49}\text{H}_{55}\text{N}_3\text{O}_6$	SmeC/Liq	413.4	4.81	11.64	98.37	[246]		
	4-[3,4-bis(4-octyloxybenzoyloxy)benzylideneamino]azobenzene							
	Sol/Nem	375.2	59.0	157.25				
$\text{C}_{49}\text{H}_{55}\text{N}_3\text{O}_6$	Nem/Liq	404.2	0.4	0.99	158.24	[205]		
	4-[[(4-octyloxyphenyl)azo]phenyl 3-[[4-[[(4-(octyloxy)benzoyl]oxy]phenyl]methylene]amino]benzoate					[296]		
	Sol/Meso	411.8	21.0	51.00	93.66			
$\text{C}_{49}\text{H}_{60}\text{O}_4$	Meso/Liq	426.6	18.2	42.66				
	1,9-bis(4-(4'-hexyloxyphenylethyanyl)phenoxy)nonane					[120]		
	Sol/Sol	312.2	7.8	24.98	194.72			
$\text{C}_{49}\text{H}_{60}\text{O}_4$	Sol/Nem	407.2	65.0	159.63				
	Nem/Liq	425.2	4.3	10.11				

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$\text{C}_{49}\text{H}_{62}\text{N}_2\text{O}_6$	Sol/Smec	364.2	67.0	183.96		[293]
	Smec/Liq	391.2	16.0	40.90	224.86	
$\text{C}_{49}\text{H}_{62}\text{N}_2\text{O}_6$	Sol/Smec	388.2	68.0	175.17		[293]
	Smec/Liq	390.2	15.0	38.44	213.61	
$\text{C}_{49}\text{H}_{62}\text{N}_2\text{O}_6$	Sol/Smec	356.2	37.0	103.87		[293]
	Smec/Liq	383.2	12.0	31.32	135.19	
$\text{C}_{49}\text{H}_{62}\text{N}_2\text{O}_8$						
	Sol/Smec	372.5	70.8	190.07		[244]
$\text{C}_{49}\text{H}_{62}\text{N}_2\text{O}_8$	Smec/Nem	393.5	4.6	11.69		
	Nem/Liq	396.7	1.7	4.29	206.05	
$\text{C}_{49}\text{H}_{62}\text{N}_2\text{O}_8$	Sol/Smec	368.2	37.0	100.49		[412]
	Smec/Liq	406.2	12.0	29.54	130.03	
$\text{C}_{49}\text{H}_{62}\text{N}_2\text{O}_9$						
	Sol/Smec	386.2	39.0	100.98		[412]
$\text{C}_{49}\text{H}_{62}\text{N}_2\text{O}_9$	Smec/Liq	422.2	13.0	30.79	131.77	
$\text{C}_{49}\text{H}_{64}\text{N}_4\text{O}_6$						
	Sol/Smec	347.2	28.0	80.65		[107]
$\text{C}_{49}\text{H}_{64}\text{N}_4\text{O}_6$	Smec/Liq	398.2	20.5	51.48	132.13	
$\text{C}_{49}\text{H}_{66}\text{N}_2\text{O}_6$						
	Sol/Sol	353.1	12.0	33.98		[224]
$\text{C}_{49}\text{H}_{66}\text{N}_2\text{O}_6$	Sol/Smec	417.2	79.3	190.08		
	Smec/Liq	440.5	10.1	22.93	246.99	
$\text{C}_{49}\text{H}_{68}\text{F}_2\text{O}_5$						
	Sol/Smec	388.2	23.5	60.54		[333]
	Smec/Meso	394.2	Could not be measured			
$\text{C}_{49}\text{H}_{68}\text{F}_2\text{O}_5$	Meso/Nem	400.2	0.26	0.65		
	Nem/Liq					
$\text{C}_{49}\text{H}_{68}\text{O}_3$						
	Sol/Smec	428.2	32.45	75.79		[245]
$\text{C}_{49}\text{H}_{68}\text{O}_3$	Sme/Nem	460.0	1.06	2.30		
	Nem/Liq	477.9	4.22	8.84	86.93	
$\text{C}_{49}\text{H}_{70}\text{O}_7$						
	Sol/Nem	347.5	59.63			[432]
$\text{C}_{49}\text{H}_{70}\text{O}_7$	Nem/Liq	352.3	0.93			
$\text{C}_{49}\text{H}_{71}\text{ClO}_7$						
	Sol/Meso	Not reported in paper				[324]
$\text{C}_{49}\text{H}_{71}\text{ClO}_7$	Meso/Liq	464.3	20.2	43.51		
$\text{C}_{49}\text{H}_{71}\text{NO}_5$						
	Sol/Sol	348.2	6.79	19.50		[364]
$\text{C}_{49}\text{H}_{71}\text{NO}_5$	Sol/Col	366.2	28.66	78.26		
	Col/Liq	488.2	9.80	20.07	117.83	
$\text{C}_{49}\text{H}_{72}\text{O}_4$						
	Sol/Smec	392.8	35.89	91.37		[364]
$\text{C}_{49}\text{H}_{72}\text{O}_4$	Sme/Meso	397.6	Not reported in paper			
	Meso/Nem	402.8	0.51	1.27		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
Note: Smec/Meso transition enthalpy is included in Meso/Nem value.						
	Nem/Liq	423.2	1.52	3.59	96.23	[74]
<i>C</i> <sub>49</sub> H <sub>72</sub> O <sub>4</sub> cholesterly 5-[4-(4'-butyloxy)biphenyloxy]hexanoate						
	Sol/Smec	389.8	22.41	57.49		
	Smec/Nem	439.7	0.94	2.14		
	Nem/Liq	463.8	4.71	10.16	69.79	186.3
						[74]
<i>C</i> <sub>49</sub> H <sub>75</sub> NO <sub>2</sub> N-(2-hydroxy-4-octadecyloxybenzylidene)-4"-dodecylphenylaniline						
	Sol/Smec	360.2	76.27	211.74		
	Smec/Smec	401.7	0.96	2.39		
	Smec/Liq	452.7	11.09	24.50	238.63	[323]
<i>C</i> <sub>49</sub> H <sub>80</sub> N <sub>2</sub> O <sub>3</sub> N-[4-[5-oxo-4-[1-(oxooctadecyl)amino]-1,3,6-cycloheptatrien-1-yl]phenyl]octadecamide						
	Sol/Sol	402.2	2.0	4.97		
	Sol/Smec	415.2	51.5	124.04		
	Smec/Liq	424.2	6.0	14.14	143.15	[251]
<i>C</i> <sub>49</sub> H <sub>82</sub> N <sub>4</sub> O <sub>7</sub> 3,4,5-tridodecyloxybenzaldehyde-2',4'-dinitrophenyl hydrazone						
	Sol/Meso	370.2	43.2	116.69		
	Meso/Liq	410.2	4.6	11.22	127.91	[13]
<i>C</i> <sub>49</sub> H <sub>91</sub> NO <sub>4</sub> 3,4,5-tris(tetradecyloxy)benzamide						
	Sol/Meso	352.2	14.2	40.32		
	Meso/Liq	353.2	5.7	16.14	56.46	[378]
<i>C</i> <sub>50</sub> H <sub>42</sub> FNO <sub>8</sub> 4-cyanophenyl 4-[3-[4-(4'-nonylbiphenyl-4-carbonyloxy)-2-fluorobenzoyloxy]benzoyloxy]benzoate						
	Sol/Nem	433.2	52.3	120.73		
	Nem/Liq	456.2	0.40	0.88	121.61	[411]
<i>C</i> <sub>50</sub> H <sub>42</sub> FNO <sub>8</sub> 4-cyanophenyl 4-[3-[4-(4'-nonylbiphenyl-4-carbonyloxy)-3-fluorobenzoyloxy]benzoyloxy]benzoate						
	Sol/Nem	427.2	35.9	84.04		
	Nem/Liq	456.3	0.35	0.77	84.81	[411]
<i>C</i> <sub>50</sub> H <sub>43</sub> NO <sub>8</sub> 4-cyanophenyl 4-[3-[4-(4'-nonylbiphenyl-4-carbonyloxy)benzoyloxy]benzoyloxy]benzoate						
	Sol/Nem	433.2	39.2	90.49		
	Nem/Liq	463.0	0.43	0.93	91.42	[411]
<i>C</i> <sub>50</sub> H <sub>50</sub> FNO <sub>9</sub> 3-[4-[4-(4-cyanophenoxy)carbonyl]phenoxy]carbonyl]phenyl 3-fluoro-4-[[4-(pentadecyloxy)benzoyl]oxy]benzoate						
	Sol/Smec	394.2	72.64	184.27		
	Smec/Smec	401.2	0.09	0.22		
	Smec/Liq	407.0	4.95	12.16	196.65	[248]
<i>C</i> <sub>50</sub> H <sub>50</sub> N <sub>2</sub> O <sub>8</sub> bis[4-ethoxyphenyl] 2-{8-[4-(4-biphenylazo)phenoxy]octyloxy}terephthalate						
	Sol/Nem	400.2	44.7	111.69		
	Nem/Liq	455.2	3.9	8.57	120.26	[54]
<i>C</i> <sub>50</sub> H <sub>51</sub> NO <sub>9</sub> 4-[4-(4-cyanophenoxy)carbonyl]phenyl 3-[[4-[4-(pentadecyloxy)benzoyl]oxy]benzoyl]oxy]benzoate						
	Sol/Smec	402.7	89.07	221.18		
Note: Sol/Sol transition enthalpy is included in Sol/Smec value.						
	Smec/Smec	413.4	0.09	0.22		
	Smec/Liq	417.2	5.27	12.63	234.03	[248]
<i>C</i> <sub>50</sub> H <sub>52</sub> Cl <sub>2</sub> F <sub>2</sub> N <sub>2</sub> O <sub>6</sub> 4,6-dichloro-1,3-phenylene bis[4-(4-octyloxy-3-fluorophenyl-iminomethyl)benzoate]						
	Sol/Nem	396.0	32.69	82.55		
	Nem/Liq	398.3	0.97	2.43	84.98	[150]
Independent values from another reference						
	Sol/Smec	400.2	49.0	122.44		
	Smec/Nem	403.0	Not reported in paper			
	Nem/Liq	403.7	2.0	4.95	127.39	[415]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
Note: Smec/Nem transition enthalpy is included in the Nem/Liq value.						
$C_{50}H_{52}Cl_4N_2O_6$	Sol/Nem	4,6-dichloro-1,3-phenylene bis[4-(4-octyloxy-3-chlorophenyliminomethyl)benzoate]				
		377.2	47.59	126.17		
		387.3	0.37	0.96	127.13	[150]
$C_{50}H_{52}N_2O_6$	Sol/Smec	1,3-phenylene bis[4-(4,7-octenyloxyphenyliminomethyl)benzoate]				
		404.0	27.2	67.33		
		416.1	17.0	40.86		
$C_{50}H_{54}Cl_2N_2O_6$	Smec/Smec	433.0	25.4	58.66	166.85	[25]
		387.2	49	126.55		
		416.2	1.8	4.32	130.87	256.6
$C_{50}H_{54}Cl_2N_2O_6$	Nem/Liq	364.0	10.45	28.71		
		395.0	33.06	83.70		
		421.4	1.02	2.42	114.83	256.6
Independent values from another reference						
$C_{50}H_{54}N_2O_9$	Sol/Nem	399.2	45.7	114.48		
	Nem/Liq	421.2	0.9	2.14	116.62	256.6
$C_{50}H_{54}N_2O_9$	Sol/Nem	449.2	41.5	92.39		
		461.2	0.5	1.08	93.47	[289]
		408.2	43.6	106.81		
$C_{50}H_{54}N_4O_{10}$	Meso/Liq	418.2	12.18	29.12	135.93	[321]
		393.2	16.1	40.95		
$C_{50}H_{54}O_8S_2$	Sol/Meso	408.2	16.3	39.93	80.88	NA
		429.2	22.0	51.26		
		433.2	14.0	32.32		
$C_{50}H_{55}ClN_2O_6$	Meso/Liq	448.2	25.0	55.78	139.36	251.4
		393.2	33.2	84.44		
		406.2	16.8	41.36	125.80	254.0
$C_{50}H_{55}ClN_2O_6$	Sol/Nem	378.2	53.0	140.14		
		381.2	0.6	1.57	141.71	254.0
		399.2	31.7	79.41		
$C_{50}H_{56}N_2O_6$	Smec/Smec	421.2	13.4	31.81		
		425.2	54.1	127.23	238.45	[417]
		429.2	22.0	51.26		
$C_{50}H_{56}N_2O_6$	Sol/Meso	448.2	25.0	55.78	139.36	251.4
		433.2	14.0	32.32		
		439.2	26.7	61.92		
$C_{50}H_{56}N_2O_6$	Meso/Liq	448.2	24.6	54.89	140.02	[424]
		377.2	28.6	75.82		
		387.7	20.3	52.36		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated) (estimated)	Ref.	
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$			
$C_{50}H_{56}N_2O_8$	Meso/Liq	407.7	16.0	39.24	167.42	[78]	
	Sol/Meso	399.2	20.0	50.10			
	Meso/Liq	447.7	18.0	40.21	90.31		
$C_{50}H_{58}N_2O_8$	bis[4-ethoxyphenyl] 2-{10-[4-(4-butylphenylazo)phenoxy]decyloxy}terephthalate					[54]	
	Sol/Nem	363.2	63.3	174.28			
	Nem/Liq	397.3	3.2	8.05	182.33		
$C_{50}H_{60}O_6$	2,7-bis(dodecyloxy)phenanthro[3,4-c]phenanthrene-9,12,13,16-tetraone					[420]	
	Sol/Meso	Not reported in paper					
	Meso/Liq	481.7	54.70	113.56			
$C_{50}H_{62}O_4$	1,10-bis(4-(4'-hexyloxyphenylethylynol)phenoxy)decane					[120]	
	Sol/Sol	322.2	9.6	29.80			
	Sol/Sol	421.2	46	109.21			
	Sol/Nem	431.2	48	111.32			
$C_{50}H_{64}N_2O_6$	Nem/Liq	440.2	9.5	21.58	271.91	[293]	
	$\alpha,\omega$ -bis[4-(4-octyloxybenzoyloxy)benzylideneamino]hexane						
	Sol/Nem	407.2	61.0	149.80			
$C_{50}H_{64}N_2O_6$	Nem/Liq	427.2	9.7	22.71	172.51	[293]	
	$\alpha,\omega$ -bis[4-(4-heptyloxybenzoyloxy)benzylideneamino]octane						
	Sol/Nem	397.2	51.0	128.40			
$C_{50}H_{64}N_2O_6$	Nem/Liq	418.2	9.9	23.67	152.07	[293]	
	$\alpha,\omega$ -bis[4-(4-hexyloxybenzoyloxy)benzylideneamino]decane						
	Sol/Nem	384.2	44.0	114.52			
$C_{50}H_{66}N_4O_6$	Nem/Liq	414.2	8.8	21.25	135.77	[293]	
	$\alpha,\omega$ -bis(4-octyloxyazobenzene-4'-carbonyloxy)octane						
	Sol/SmeC	379.1	48.1	126.88			
$C_{50}H_{68}F_2O_5$	SmeC/Liq	412.7	25.9	62.76	189.64	[107]	
	1-methylheptyl 4'-(4"-eicosyloxy-2",3"-difluorobenzoyloxy)tolan-4-carboxylate						
	Sol/SmeC	338.6	49.90	147.37			
$C_{50}H_{68}N_2O_6$	SmeC/SmeC	371.2	0.15	0.40		[213]	
	SmeC/Liq	376.2	3.85	10.23	158.00		
	$\alpha,\omega$ -bis{4-[(2-hydroxy-4-octyloxyphenyl)iminomethyl]phenyloxy}octane						
$C_{50}H_{68}N_2O_6$	Sol/Sol	342.4	10.6	30.96		[224]	
	Sol/SmeC	434.4	79.8	183.70			
	SmeC/SmeC	439.5	Not detected by dsc				
$C_{50}H_{69}FO_5$	SmeC/Liq	462.3	18.5	40.02	254.68	[213]	
	1-methylheptyl 4'-(4"-eicosyloxy-3"-fluorobenzoyloxy)tolan-4-carboxylate						
	Sol/SmeC	345.2	68.45	198.29			
$C_{50}H_{70}F_2O_5$	SmeC/SmeC	366.2	0.18	0.49		[333]	
	SmeC/Liq	369.7	4.00	10.82	209.60		
	cholesteryl 2-fluoro-4-(3-fluoro-4-nonyloxybenzoyloxy)benzoate						
$C_{50}H_{70}O_2$	Sol/SmeC	388.7	28.8	74.09		[333]	
	Note: Sol/Sol transition enthalpy is included in the Sol/SmeC value.						
	SmeC/Meso	414.2	Could not be measured				
$C_{50}H_{70}O_2$	Meso/Nem	422.7	0.25	0.59		[333]	
	Nem/Liq	Decomposed prior to transition					
$C_{50}H_{70}O_2$	4,4"-ditridecyloxy-p-quaterphenyl					[333]	
	Sol/Sol	412.0	31.3	75.97			
$C_{50}H_{70}O_2$	Sol/Sol	418.0	2.93	7.01			

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{50}H_{70}O_3$	Sol/Sol	445.0	3.11	6.99		
	Sol/Smec	539.0	13.47	24.99		
	Smec/Liq	565.0	13.56	24.00	138.96	273.4 [111]
$C_{50}H_{70}O_3$		cholesteryl $\omega$ -[4-(4-pentylphenylethynyl)phenoxy]butanoate				
	Sol/Smec	426.8	27.14	63.59		
	Smec/Nem	451.3	0.57		1.27	
	Nem/Liq	468.8	3.30	7.05	71.9	173.2 [245]
$C_{50}H_{70}O_3$		cholesteryl $\omega$ -[4-(4-butylphenylethynyl)phenoxy]pentanoate				
	Sol/Smec	391.4	29.22	74.66		
	Smec/Nem	400.0	0.57	1.44		
	Nem/Liq	421.5	2.4	2.04	78.14	173.2 [245]
$C_{50}H_{72}O_7$		2-(1-oxododecyl)-1,4-phenylene 4-(nonyloxy)benzoate				
	Sol/Nem	337.1	92.2	273.51		
	Nem/Liq	351.2	1.4	3.99	277.50	[432]
$C_{50}H_{74}O_4$		cholesterlyl 5-[4-(4'-hexyloxy)biphenyloxy]pentanoate				
	Sol/Smec	400.4	50.04	124.98		
	Smec/Meso	409.3	Not reported in paper			
	Meso/Nem	410.8	1.11	2.70		
	Note: Smec/Meso transition enthalpy is included in Meso/Nem value.					
$C_{50}H_{74}O_4$	Nem/Liq	423.9	1.63	3.85	131.53	199.8 [74]
		cholesterlyl 5-[4-(4'-pentyloxy)biphenyloxy]hexanoate				
	Sol/Smec	418.4	34.89	83.39		
	Smec/Meso	444.0	Not reported in paper			
	Meso/Nem	444.2	1.48	3.33		
$C_{50}H_{74}O_4$	Note: Smec/Meso transition enthalpy is included in Meso/Nem value.					
	Nem/Liq	461.7	4.80	10.40	97.12	193.4 [74]
$C_{50}H_{78}N_2O_6$		N,N'-diundecanoyl-2,4-bis(undecanoyloxy)-1,3-benzenediamine				
	Sol/Disc	368.2	23.0	62.47		
	Disc/Liq	393.2	16.0	40.69	103.16	329.2 [189]
$C_{50}H_{80}O_{10}$		2,3,6,7-tetrakis(octyloxy)-1,5-bis(2-hydroxyethoxy)-9,10-anthracenedione				
	Sol/Col	345.7	56.76	164.19		
	Col/Liq	403.7	20.26	50.19	214.38	[275]
$C_{50}H_{82}N_2O_6$		2,5-bis(3,4,5-trihexyloxyphenyl)-1,3,4-oxadiazole				
	Sol/Col	Below ambient temperature				
	Col/Liq	341.5	5.33	15.61		[178]
$C_{51}H_{44}FNO_8$		4-cyanophenyl 4-{3-[4-(4'-decylbiphenyl-4-carbonyloxy)-2-fluorobenzoyloxy]}benzoate				
	Sol/Smec	434.7	48.2	110.88		
	Smec/Smec	435.7	0.20	0.46		
	Smec/Nem	439.2	0.10	0.23		
	Nem/Liq	453.0	0.38	0.84	112.41	[411]
$C_{51}H_{44}FNO_8$		4-cyanophenyl 4-{3-[4-(4'-decylbiphenyl-4-carbonyloxy)-3-fluorobenzoyloxy]}benzoate				
	Sol/Smec	431.2	40.2	93.23		
	Smec/Smec	435.2	0.13	0.30		
	Smec/Nem	440.1	0.10	0.23		
	Nem/Liq	453.0	0.69	1.52	95.28	[411]
$C_{51}H_{45}FNO_8$		4-cyanophenyl 4-{3-[4-(4'-decylbiphenyl-4-carbonyloxy)benzoyloxy]}benzoate				
	Sol/Smec	442.2	56.8	128.45		
	Smec/Smec	444.5	0.15	0.34		
	Smec/Nem	448.0	0.20	0.45		
	Nem/Liq	462.2	0.47	1.02	130.26	[411]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.		
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$				
$C_{51}H_{49}N_5O_4$	Sol/Meso	497.2	35.5	71.40	112.35	[331]		
	Meso/Liq	515.2	21.1	40.95				
$C_{51}H_{52}FNO_9$		3-[[4-[(4-cyanophenoxy)carbonyl]phenoxy]carbonyl]phenyl 2-fluoro-4-[[4-(hexadecyloxy)benzoyl]oxy]-benzoate						
	Sol/SmeC	404.7	60.30	149.00	161.10	[248]		
$C_{51}H_{52}FNO_9$	SmeC/Liq	410.7	4.97	12.10				
		3-[[4-[(4-cyanophenoxy)carbonyl]phenoxy]carbonyl]phenyl 3-fluoro-4-[[4-(hexadecyloxy)benzoyl]oxy]-benzoate						
	Sol/SmeC	394.2	67.15	170.35	183.88	[248]		
	SmeC/SmeC	402.2	0.08	0.20				
$C_{51}H_{52}N_2O_8$	SmeC/Liq	410.2	5.47	13.33				
		bis[4-ethoxyphenyl] 2-{9-[4-(4-biphenylazo)phenoxy]nonyloxy}-terephthalate						
	Sol/Nem	391.2	52.2	133.44	139.16	[54]		
$C_{51}H_{53}NO_9$	Nem/Liq	437.2	2.5	5.72				
		4-[(4-cyanophenoxy)carbonyl]phenyl 3-[[4-[[4-(hexadecyloxy)benzoyl]oxy]benzoyl]oxy]benzoate						
	Sol/SmeC	403.2	86.67	214.96	228.91	[248]		
	Note: Sol/Sol transition enthalpy is included in Sol/SmeC value.							
$C_{51}H_{53}NO_9$	SmeC/SmeC	415.1	0.08	0.19				
	SmeC/Liq	420.7	5.79	13.76				
		2-cyano-1,3-phenylene bis[4-(4-octyloxyphenyliminomethyl)benzoate]						
$C_{51}H_{55}N_3O_6$	Sol/Meso	443.2	14.5	32.72	90.36	[246]		
	Meso/Liq	466.7	26.9	57.64				
$C_{51}H_{55}N_3O_6$		4-cyano-1,3-phenylene bis[4-(4-octyloxyphenyliminomethyl)benzoate]						
	Sol/Meso	370.2	37.4	101.03	121.33	[252]		
	Meso/SmeC	415.2	1.9	4.58				
	SmeC/SmeC	419.2	2.1	5.01				
	SmeC/Liq	448.2	4.8	10.71				
$C_{51}H_{56}F_2N_2O_6$		2-methyl-1,3-phenylene bis[4-(3-fluoro-4-n-octyloxyphenyliminomethyl)benzoate]						
	Sol/Meso	376.2	11.4	30.30	75.05	[24]		
	Meso/Meso	385.2	0.5	1.30				
	Meso/Meso	409.2	0.4	0.98				
	Meso/Liq	426.2	18.1	42.47				
$C_{51}H_{58}N_2O_6$		2-methyl-1,3-phenylene 4-[[4-(octyloxy)phenyl]imino]methylbenzoate						
	Sol/Sol	350.2	1.0	2.86	100.21	[424]		
	Sol/Sol	363.2	3.6	9.91				
	Sol/Sol	395.2	2.3	5.82				
	Sol/Meso	434.2	13.3	30.63				
$C_{51}H_{59}N_3O_5$	Meso/Liq	445.2	22.7	50.99	70.35	[296]		
		4-[(4-decylophenyl)azo]phenyl 3-[[4-[[4-(octyloxy)benzoyl]oxy]phenyl]methylene]amino]benzoate						
	Sol/Meso	404.2	15.9	39.34				
$C_{51}H_{59}N_3O_6$	Meso/Liq	409.6	12.7	31.01	97.77	[296]		
		4-[(4-decyloxyphenyl)azo]phenyl 3-[[4-[[4-(octyloxy)benzoyl]oxy]phenyl]methylene]amino]benzoate						
	Sol/Meso	407.7	22.3	54.70				
$C_{51}H_{59}N_3O_6$	Meso/Liq	420.2	18.1	43.07	181.1	[205]		
	Note: Paper reported a value of 181.1 kJ·mol <sup>-1</sup> for the Meso/Liq phase transition enthalpy, which is believed to be in error. Transition enthalpies for other derivatives in this series are in the range of 18 kJ·mol <sup>-1</sup> .							
		4-[3,4-bis(4-nonyloxybenzoyloxy)benzylideneamino]azobenzene						
$C_{51}H_{59}N_3O_6$	Sol/Nem	372.2	68.0	182.70	245.2	[205]		
	Nem/Liq	400.2	0.4	1.00				

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.		
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$				
$C_{51}H_{66}N_2O_6$	Sol/Smec	383.2	51.0	133.09	271.9	[293]		
	Smec/Liq	391.2	16.0	40.90				
$C_{51}H_{66}N_2O_6$	Sol/Smec	361.2	63.0	174.42	271.9	[293]		
	Smec/Liq	387.2	14.0	36.16				
$C_{51}H_{66}N_2O_8$	malonic acid, bis{7-[4-[(4-butoxyphenylimino)methyl]phenoxy]heptyl} ester							
	Sol/Smec	391.8	75.0	191.42	207.07	[244]		
	Smec/Nem	394.8	3.5	8.87				
$C_{51}H_{66}N_2O_8$	Nem/Liq	398.2	2.7	6.78				
	Sol/Smec	370.2	38.0	102.65	132.27	[412]		
	Smec/Liq	405.2	12.0	29.62				
$C_{51}H_{66}N_2O_8$	Sol/Smec	382.2	40.0	104.66	135.52	[412]		
	Smec/Liq	421.2	13.0	30.86				
$C_{51}H_{68}N_4O_6$	$\alpha,\omega$ -bis(4-octyloxyazobenzene-4'-carbonyloxy)nonane							
	Sol/Smec	351.4	21.7	61.76	274.5	[107]		
$C_{51}H_{70}N_2O_6$	Smec/Liq	396.5	13.8	34.80				
	Sol/Smec	411.6	78.6	191.0	290.9	[224]		
$C_{51}H_{70}N_2O_6$	Smec/Smec	394.7	6.5	16.5				
	Note: Smec/Smec transition is reported to occur at a temperature lower than the Sol/Smec transition.							
$C_{51}H_{72}F_2O_5$	Smec/Nem	427.4	1.8	4.21				
	Nem/Liq	432.4	2.8	6.48				
$C_{51}H_{72}F_2O_5$	cholesteryl 2-fluoro-4-(3-fluoro-4-decyloxybenzoyloxy)benzoate							
	Sol/Smec	374.2	22.4	59.86	290.9	[333]		
	Smec/Meso	428.2	Could not be measured					
	Meso/Nem	438.2	0.34	0.78				
$C_{51}H_{72}N_4O_5$	Nem/Liq	Decomposed prior to transition						
	Sol/Smec	421.5	119.8	284.22	318.23	[230]		
	Smec/Liq	445.1	15.14	34.01				
Note: Sol/Smec transition enthalpy is abnormal large compared with values for other derivatives in series.								
$C_{51}H_{72}O_3$	cholesteryl $\omega$ -[4-(4-hexylphenylethynyl)phenoxy]butanoate							
	Sol/Smec	392.5	31.08	79.18	89.23	[245]		
	Smec/Nem	469.6	0.81	1.72				
$C_{51}H_{72}O_3$	Nem/Liq	475.6	3.96	8.33	68.17	[245]		
	Sol/Smec	389.0	25.29	65.01				
	Smec/Nem	407.8	0.51	1.25				
$C_{51}H_{72}O_3$	Nem/Liq	423.6	0.81	1.91	68.17	[245]		
	Sol/Sol	366.9	Not reported in paper					
	Sol/Smec	372.7	2.57	6.90				
	Smec/Nem	425.7	0.73	1.71				
$C_{51}H_{73}FO_5$	Nem/Liq	459.3	3.74	8.14	89.23	[245]		
	cholesteryl 2-fluoro-4-(4-decyloxybenzoyloxy)benzoate							
	Sol/Smec	356.2	34.1	95.73				
$C_{51}H_{73}FO_5$	Smec/Nem	391.2	0.27	0.69	89.23	[245]		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.			
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$					
	Smec/Liq	Decomposed prior to transition					[333]		
$C_{51}H_{74}F_2O_4$		cholesteryl 4-[4'-(2,3-difluoro-4-octyloxy)biphenyloxy]butanoate							
	Sol/Smec	400.2	47.5	118.69					
	Smec/Meso	450.3	2.5	5.55					
	Meso/Liq	453.1	4.1	9.05	133.29		[132]		
$C_{51}H_{76}O_4$		cholesterlyl 5-[4-(4'-heptyloxy)biphenyloxy]pentanoate							
	Sol/Smec	385.5	38.86	100.80					
	Smec/Meso	412.6	Not reported in paper						
	Meso/Nem	413.7	1.28	3.09					
	Nem/Liq	419.3	1.36	3.24	107.13		[74]		
$C_{51}H_{76}O_4$		cholesterlyl 5-[4-(4'-hexyloxy)biphenyloxy]hexanoate							
	Sol/Smec	407.5	30.58	75.04					
	Smec/Meso	431.5	Not reported in paper						
	Meso/Nem	437.5	0.60	1.37					
	Nem/Liq	455.3	4.67	10.26	86.67		[74]		
$C_{51}H_{78}O_6$		2,6,10-tris(pentyloxy)-3,7,11-tris(hexyloxy)triphenylene							
	Sol/Col	321.2	28.28	88.04					
	Col/Liq	379.2	5.76	15.19	103.23		[261]		
Independent values from another reference									
$C_{51}H_{78}O_6$	Sol/Col	Not reported in paper							
	Col/Liq	379.2	5.8	15.30			[368]		
$C_{51}H_{92}N_2O_4$		N,N'-ditetradecanoyl-2,5,6-trimethyl-4-tetradecanoyloxy-1,3-benzenediamine							
	Sol/Meso	383.2	Not reported in paper						
	Meso/Meso	401.2	22.0	54.84					
	Meso/Liq	453.2	2.0	4.41			[193]		
$C_{51}H_{93}N_3O_3$		N,N',N"-tritetradeccanoyl-2,4,6-trimethyl-1,3,5-benzenetriamine							
	Sol/Meso	391.2	16.0	40.90					
	Meso/Meso	455.2	19.0	41.74					
	Meso/Liq	661.2	9.0	13.61	96.25	314.4	[190]		
$C_{52}H_{46}$		9,10-bis(4'-pentylbiphenyl-4-ylethynyl)anthracene							
	Sol/Nem	514.2	37.8	73.51					
	Nem/Liq	543.2	Decomposed				[429]		
$C_{52}H_{46}FNO_8$		4-cyanophenyl 4-[3-[4-(4'-undecylbiphenyl-4-carbonyloxy)-2-fluorobenzoyloxy]benzoyloxy]benzoate							
	Sol/Smec	434.7	58.9	135.50					
	Smec/Smec	441.2	0.18	0.41					
	Smec/Nem	450.2	0.30	0.67					
	Nem/Liq	453.4	0.57	1.26	137.84		[411]		
$C_{52}H_{46}FNO_8$		4-cyanophenyl 4-[3-[4-(4'-undecylbiphenyl-4-carbonyloxy)-3-fluorobenzoyloxy]benzoyloxy]benzoate							
	Sol/Smec	431.2	37.4	86.73					
	Smec/Smec	440.2	0.13	0.30					
	Smec/Nem	450.8	0.48	1.06					
	Nem/Liq	453.0	0.58	1.28	89.37		[411]		
$C_{52}H_{47}NO_8$		4-cyanophenyl 4-[3-[4-(4'-undecylbiphenyl-4-carbonyloxy)benzoyloxy]benzoyloxy]benzoate							
	Sol/Smec	442.2	44.7	101.09					
	Smec/Smec	449.7	0.13	0.29					
	Smec/Nem	458.5	0.40	0.87					
	Nem/Liq	460.5	0.57	1.24	103.49		[411]		
$C_{52}H_{50}N_2O_8S$		(3,4-dicyano-2,5-thiophenediyi)bis(2,1-ethynediyl-4,1-phenylene) 3,4-bis(butoxy)benzoate							

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{fus}} S_{tpce}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{fus}} S_{tpce}$ (estimated)	Ref.
		T (K)	$\Delta H_{pce}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{pce}$		
$C_{52}H_{50}N_4O_4$	Sol/Nem	421.6	44.4	105.31		[234]
	Nem/Liq	435.7	1.14	2.62	107.93	
$C_{52}H_{52}$	Sol/Smec	482.2	32.4	67.19		[331]
	Smec/Liq	495.2	15.5	31.30	98.49	
	Sol/Sol	399.2	13.2	33.07		
$C_{52}H_{54}F_4O_{10}$	Sol/Nem	508.2	34.5	67.89		[429]
	Nem/Liq	548.2	Decomposed			
	Sol/Meso	380.2	74.8	196.74		
$C_{52}H_{54}N_2O_8$	Meso/Liq	383.7	19.7	51.34	248.08	[247]
	Sol/Nem	391.2	42.6	108.90		
	Nem/Liq	441.2	3.6	8.16	117.06	
$C_{52}H_{56}Cl_2F_2N_2O_6$	Sol/Smec	386.2	44.4	114.97		[415]
	Smec/Liq	406.2	4.2	10.34	125.31	
	Sol/Meso	398.0	20.86	52.41		
$C_{52}H_{56}Cl_2O_6$	Meso/Liq	412.9	16.03	38.82	91.23	[177]
	Sol/Meso	4-[1E]-[3-chloro-4-octyloxyphenyl]ethenyl]benzoic acid, 1,3-phenylene ester				
	Meso/Liq	398.0	20.86	52.41		
$C_{52}H_{57}ClO_{10}$	Sol/Nem	359.2	36.6	101.89		[438]
	Nem/Liq	368.2	0.5	1.36	103.25	
	Sol/Nem	374.2	44	117.58		
$C_{52}H_{58}Cl_2N_2O_6$	Nem/Liq	404.2	1.5	3.71	121.29	[86]
	Sol/Nem	379.2	13.9	36.66		
	Nem/Liq	416.2	0.5	1.20	37.86	
Note: Sol/Nem transition enthalpy of this compound is much lower than value for other compounds in this series.						
$C_{52}H_{58}N_2O_9$	Sol/Nem	442.2	23.2	52.46		[289]
	Nem/Liq	459.2	0.4	0.87	53.33	
	Sol/Meso	386.2	14.9	38.58		
$C_{52}H_{58}O_8S_2$	Meso/Liq	396.2	14.6	36.85	75.43	[318]
	Sol/Nem	362.2	48.0	132.52		
	Nem/Liq	373.2	0.5	1.34	133.86	
$C_{52}H_{59}ClN_2O_6$	Sol/Meso	398.2	15.5	38.93		[42]
	Meso/Liq	409.2	14.8	36.17	75.10	
	Sol/Nem	362.2	48.0	132.52	268.2	
$C_{52}H_{59}ClN_2O_6$	Sol/Meso	398.2	15.5	38.93		[42]
	Meso/Liq	409.2	14.8	36.17	75.10	
	Sol/Nem	362.2	48.0	132.52	268.2	
$C_{52}H_{60}N_2O_4S_2$	Sol/Meso	398.5	15.11	37.92		[272]
	Meso/Meso	411.1	13.73	33.40		
	Meso/Liq	422.0	21.11	50.02	121.34	

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.	
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$			
$C_{52}H_{60}N_2O_6$	Sol/SmeC	417.2	35.0	83.89	251.0	[86]	
	SmeC/Liq	442.2	23.0	52.01			
$C_{52}H_{60}N_2O_6$	Sol/Sol	354.7	29.8	84.01	251.0	[78]	
	Sol/Meso	392.2	20.7	52.78			
	Meso/Liq	398.2	15.6	39.18	175.97		
$C_{52}H_{60}N_2O_6$	Sol/Meso	405.2	45.8	113.03	230.40	[417]	
	Meso/Meso	418.2	37.2	88.95			
	Meso/Liq	422.2	12.0	28.42			
$C_{52}H_{60}N_2O_8$	Sol/Meso	392.2	19.0	48.44	90.69	[412]	
	Meso/Liq	449.7	19.0	42.25			
	Sol/SmeC	405.2	54.0	133.27			
$C_{52}H_{68}N_2O_6$	SmeC/Nem	410.2	1.8	4.39	137.66	279.0	[293]
	Sol/SmeC	389.2	49.0	125.90	151.84	279.0	[293]
	SmeC/Nem	401.2	0.6	1.50			
$C_{52}H_{68}N_2O_6$	Nem/Liq	409.2	10.0	24.44			
	Sol/Meso	422.2	32.64	77.31	147.80	293.2	[295]
	Meso/Meso	450.6	10.98	24.37			
$C_{52}H_{68}N_2O_6$	Meso/Liq	498.5	22.99	46.12			
$C_{52}H_{68}N_2O_6$	Sol/Meso	407.2	36.20	88.90	161.60	293.2	[295]
	Meso/Meso	446.7	8.76	19.61			
	Meso/Liq	483.7	25.68	53.09			
$C_{52}H_{71}N_9O_3$	Sol/Col	432.8	43.41	100.30	15.04	[187]	
	Col/Liq	477.0	7.03	14.74			
	Sol/SmeC	433.6	88.6	204.33			
$C_{52}H_{72}N_2O_6$	SmeC/Nem	443.0	2.6	5.87	227.50	[224]	
	Nem/Liq	445.0	7.7	17.30			
	Sol/SmeC	438.8	Not reported in paper				
$C_{52}H_{74}ClNO_5$	Meso/Chol	438.9	0.30	0.68	54.99	[362]	
	Note: SmeC/Meso transition enthalpy is included in the Meso/Chol value.						
	Chol/Liq	458.6	3.81	8.31			
	Sol/SmeC	393.9	17.19	43.64			
	SmeC/SmeC	423.7	1.00	2.36			
$C_{52}H_{74}F_2O_5S$	SmeC/Meso	438.8	Not reported in paper		309.78	[373]	
	Meso/Chol	438.9	0.30	0.68			
	Chol/Liq	458.6	3.81	8.31			
$C_{52}H_{74}F_2O_5$	Sol/Nem	354.2	107.3	302.94	88.90	[373]	
	Nem/Liq	365.6	2.5	6.84			
$C_{52}H_{74}F_2O_5$	Sol/SmeC	380.2	33.8	88.90	Note: Sol/Sol transition enthalpy is included in the Sol/SmeC value.		
	Sol/Sol	380.2	33.8	88.90			

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{52}H_{74}O_2$	Smec/Meso	441.2	Could not be measured			[333]
	Meso/Nem	451.7	0.50	1.11		
	Nem/Liq	Decomposed prior to transition				
$C_{52}H_{74}O_3$	$4,4''$ -ditetradecyloxy-p-quaterphenyl					
	Sol/Sol	411.0	4.80	109.00		
	Sol/Sol	445.0	4.90	11.01		
	Sol/Smec	534.0	13.35	25.00		
$C_{52}H_{74}O_3$	Smec/Liq	558.0	13.39	24.00	169.01	287.6
	cholesteryl $\omega$ -[4-(4-heptylphenylethynyl)phenoxy]butanoate					[111]
	Sol/Smec	356.0	62.76	176.29		
	Smec/Nem	466.5	Not observed by dsc			
$C_{52}H_{74}O_3$	Nem/Liq	469.5	5.60	11.93	188.22	[245]
	cholesteryl $\omega$ -[4-(4-hexylphenylethynyl)phenoxy]pentanoate					
	Sol/Smec	379.1	20.55	54.21		
	Smec/Nem	411.3	0.97	2.36		
$C_{52}H_{74}O_3$	Nem/Liq	419.3	0.82	1.96	58.53	[245]
	cholesteryl $\omega$ -[4-(4-pentylphenylethynyl)phenoxy]hexanoate					
	Sol/Smec	388.1	26.67	68.72		
	Smec/Nem	434.6	0.37	0.85		
$C_{52}H_{75}FO_4$	Nem/Liq	458.6	3.96	8.63	78.20	[245]
	cholesteryl 2-fluoro-4-(4-undecylbenzoyloxy)benzoate					
	Sol/Smec	389.7	25.6	65.69		
	Smec/Meso	397.2	Could not be measured			
$C_{52}H_{75}FO_5$	Meso/Nem	406.2	0.19	0.47		
	Nem/Liq	Decomposed prior to transition				[333]
	cholesteryl 2-fluoro-4-(4-undecyloxybenzoyloxy)benzoate					
	Sol/Smec	367.7	27.2	73.97		
$C_{52}H_{75}NO_5$	Smec/Meso	401.2	Could not be measured			
	Meso/Nem	408.7	0.29	0.71		
	Nem/Liq	Decomposed prior to transition				[333]
	cholest-5-en-3-ol ( $3\beta$ ), 6-[4-[(E)-[[4-[(3S)-3-methyl-1-oxopentyl]oxy]phenyl]imino]methyl]phenoxy]-hexanoate					
$C_{52}H_{76}F_2O_4$	Sol/Smec	400.8	16.15	40.29		
	Smec/Meso	442.3	Not reported in paper			
	Meso/Chol	442.7	0.45	1.02		
	Note: Smec/Meso transition enthalpy is included in the Meso/Chol value.					
$C_{52}H_{76}F_2O_4$	Chol/Liq	466.4	3.11	6.67	47.98	[362]
	cholesteryl 4-{4'-(2,3-difluoro-4-octyloxy)biphenyloxy]pentanoate					
	Sol/Smec	353.3	45.4	128.50		
	Smec/Meso	392.9	1.0	2.55		
$C_{52}H_{76}O_{10}$	Meso/Liq	398.7	0.7	1.76	132.81	[132]
	$5,5'$ -[[6,7,10,11-tetrakis(hexyloxy)]-2,3-triphenylenediy]bis(oxy)-bis-pentanoic acid					
	Sol/Col	367.2	51.8	141.07		
$C_{52}H_{78}O_4$	Col/Liq	378.5	6.1	16.12	157.19	[353]
	cholesterly 5-[4-(4'-octyloxy)biphenyloxy]pentanoate					
	Sol/Smec	381.0	38.36	100.68		
$C_{52}H_{78}O_4$	Smec/Meso	417.9	Not reported in paper			
	Meso/Nem	418.4	1.99	4.76		
	Note: Smec/Meso transition enthalpy is included in Meso/Nem value.					
	Nem/Liq	421.0	1.15	2.73	108.17	[74]
$C_{52}H_{78}O_4$		cholesterly 5-[4-(4'-heptyloxy)biphenyloxy]hexanoate				

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{52}H_{80}N_2$	Sol/Smec	412.9	28.54	69.12		
	Smec/Meso	440.4	Not reported in paper			
	Meso/Nem	441.2	0.84	1.90		
	Note: Smec/Meso transition enthalpy is included in Meso/Nem value.					
$C_{52}H_{80}O_{12}$	Nem/Liq	452.6	4.22	9.32	80.34	[74]
	terephthalylidene- <i>bis</i> -(hexadecylaniline)					
	Sol/Smec	365.1	62.88	172.23		
	Smec/Smec	413.6	6.61	15.98		
$C_{52}H_{84}O_6$	Smec/Liq	434.9	8.47	19.48	207.69	292.2
	2,3,6,7-tetrakis(hexyloxy)-1,5- <i>bis</i> [2-[ <i>trans</i> -(tetrahydro-2 <i>H</i> -pyran-2-yl)oxy]ethoxy]-9,10-anthracenedione					
	Sol/Col	356.7	34.41	96.47		
	Col/Liq	364.2	11.81	32.43	128.90	[275]
$C_{53}H_{48}FNO_8$	di(4'-hexadecyloxyphenyl) <i>trans</i> -cyclohexane-1,4-dicarboxylate					
	Sol/Smec	369.2	51.54	139.60		
	Smec/Smec	388.2	8.99	23.16		
	Smec/Liq	423.2	11.32	26.75	189.51	317.3
$C_{53}H_{48}FNO_8$	4-cyanophenyl 4-{3-[4-(4'-dodecylbiphenyl-4-carbonyloxy)-2-fluorobenzoyloxy]benzoyloxy}benzoate					
	Sol/Smec	433.2	45.6	105.26		
	Smec/Smec	445.7	0.10	0.22		
	Smec/Liq	456.2	2.70	5.92	111.40	[411]
$C_{53}H_{49}NO_8$	4-cyanophenyl 4-{3-[4-(4'-dodecylbiphenyl-4-carbonyloxy)-3-fluorobenzoyloxy]benzoyloxy}benzoate					
	Sol/Smec	427.7	38.7	90.48		
	Smec/Smec	443.5	0.11	0.25		
	Smec/Liq	455.2	3.15	6.92	97.65	[411]
$C_{53}H_{54}N_2O_9$	4-cyanophenyl 4-{3-[4-(4'-dodecylbiphenyl-4-carbonyloxy)benzoyloxy]benzoyloxy}benzoate					
	Sol/Smec	431.2	53.6	124.30		
	Smec/Smec	454.2	0.25	0.55		
	Smec/Liq	465.2	2.9	6.23	131.08	[411]
$C_{53}H_{56}FNO_9$	1,3-{4-[4-(4-pentyloxybenzoyloxy)benzylidene]aminophenoxy}-propan-2-ol					
	Sol/Nem	457.3	33.50	73.26		
	Nem/Liq	570.2	2.20	3.86	77.12	[413]
	3-[[4-[(4-cyanophenoxy)carbonyl]phenoxy]carbonyl]phenyl 2-fluoro-4-[[4-(octadecyloxy)benzoyl]oxy]benzoate					
$C_{53}H_{56}FNO_9$	Sol/Smec	404.2	60.26	149.08		
	Smec/Smec	405.7	0.06	0.15		
	Smec/Liq	416.7	5.95	14.28	163.51	266.2
	3-[[4-[(4-cyanophenoxy)carbonyl]phenoxy]carbonyl]phenyl 3-fluoro-4-[[4-(octadecyloxy)benzoyl]oxy]benzoate					
$C_{53}H_{57}NO_9$	Sol/Smec	393.2	44.8	113.94		
	Smec/Smec	403.2	0.11	0.27		
	Smec/Liq	415.5	5.85	14.08	128.29	266.2
	4-[(4-cyanophenoxy)carbonyl]phenyl 3-[[4-[(4-(octadecyloxy)benzoyl]oxy]benzoyl]oxy]benzoate					
$C_{53}H_{59}N_3O_6$	Sol/Smec	402.7	60.57	150.41		
	Note: Sol/Sol transition enthalpy is included in Sol/Smec value.					
	Smec/Smec	416.4	0.11	0.26		
	Smec/Liq	425.7	6.17	14.49	165.16	[248]
$C_{53}H_{59}N_3O_6$	2-cyano-1,3-phenylene <i>bis</i> [4-(4-nonyloxyphenyliminomethyl)benzoate]					
	Sol/Meso	436.2	16.6	38.06		
	Meso/Liq	466.7	28.2	60.42	98.48	[246]
	4-cyano-1,3-phenylene <i>bis</i> [4-(4-nonyloxyphenyliminomethyl)benzoate]					

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.	
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$			
$C_{53}H_{62}N_2O_6$	Sol/Meso	335.2	45.9	136.93		[252]	
	Meso/SmeC	407.2	3.5	8.60			
	SmeC/SmeC	411.7	2.7	6.56			
	SmeC/Liq	453.2	5.8	12.80	164.89		
$C_{53}H_{62}N_2O_6$	2-methyl-1,3-phenylene 4-[[[4-(nonyloxy)phenyl]imino]methyl]benzoate					[424]	
	Sol/Sol	348.2	4.8	13.79			
	Sol/Sol	374.2	2.9	7.75			
	Sol/Sol	405.2	13.6	33.56			
	Sol/Meso	430.2	8.5	19.76			
	Meso/Meso	436.2	17.8	40.81			
$C_{53}H_{63}N_3O_5$	Meso/Liq	441.2	25.2	57.12	172.79	[296]	
	4-[[(4-dodecylphenyl)azo]phenyl 3-[[[4-[[(4-octyloxy)benzoyl]oxy]phenyl]methylene]amino]benzoate						
	Sol/Meso	397.2	23.4	58.91			
	Meso/Liq	409.2	18.4	44.97	103.88		
	$\alpha,\omega$ -bis[4-[4-(4-octyloxybenzoyloxy)benzylideneamino]nonane						
	Sol/Meso	398.6	58.8	147.52			
$C_{53}H_{63}N_3O_6$	Meso/Liq	416.0	20.0	48.08	195.60	[296]	
	4-[[(4-dodecyloxyphenyl)azo]phenyl 3-[[[4-[[(4-octyloxy)benzoyl]oxy]phenyl]methylene]amino]benzoate						
	Sol/Meso	398.6	58.8	147.52			
	Meso/Liq	416.0	20.0	48.08	195.60		
	4-[3,4-bis(4-decyloxybenzoyloxy)benzylideneamino]azobenzene						
	Sol/Nem	364.2	67.0	183.96			
$C_{53}H_{70}N_2O_6$	Nem/Liq	398.2	0.5	1.26	185.22	[205]	
	$\alpha,\omega$ -bis[4-(4-octyloxybenzoyloxy)benzylideneamino]nonane						
	Sol/SmeC	384.2	51.0	132.74			
	SmeC/Liq	392.2	15.0	38.25	170.99		
	heptanedioic acid, bis[4-[[(4-decyloxy)phenyl]imino]methyl]phenyl] ester						
	Sol/SmeC	438.2	61.0	139.21			
$C_{53}H_{70}N_2O_6$	SmeC/Liq	455.7	16.97	37.24	176.45	[295]	
	malonic acid, bis{8-[4-[[(4-butyoxyphenylimino)methyl]phenoxy]octyl} ester						
	Sol/Sol	371.7	44.2	118.91			
	Sol/SmeC	381.4	36.8	96.49			
	SmeC/Liq	396.8	11.2	28.23	243.63		
	1,5-[4-(2-hydroxy-4-nonyloxybenzylideneamino)benzoyloxy]-2,2-dimethylpentane						
$C_{53}H_{70}N_2O_8$	Sol/SmeC	370.2	39.0	105.35		[412]	
	SmeC/Liq	405.2	12.0	29.62	134.97		
	1,5-[4-(2-hydroxy-4-nonyloxybenzylideneamino)benzoyloxy]-3,3-dimethylpentane						
	Sol/SmeC	375.2	56.0	149.26			
	SmeC/Liq	420.2	13.0	30.94	180.20		
	$\alpha,\omega$ -bis{4-[[(2-hydroxy-4-octyloxyphenyl)iminomethyl]phenoxy]undecane}						
$C_{53}H_{74}N_2O_6$	Sol/SmeC	405.6	82.0	202.17		[224]	
	SmeC/Nem	410.1	Not detected by dsc				
	Nem/Liq	428.8	3.2	7.46	209.63		
	cholesteryl 2-fluoro-4-(3-fluoro-4-dodecyloxybenzoyloxy)benzoate						
	Sol/SmeC	383.2	70.5	183.98			
	Note: Sol/Sol transition enthalpy is included in the Sol/SmeC value.						
$C_{53}H_{76}F_2O_5$	SmeC/Mes	449.7	Could not be measured			[333]	
	Mes/Nem	459.2	0.50	1.09			
	Nem/Liq	Decomposed prior to transition					
	cholesteryl $\omega$ -[4-(4-heptylphenylethynyl)phenoxy]pentanoate						
	Sol/SmeC	376.2	36.76	97.71			
	SmeC/Nem	415.4	1.52	3.66			
$C_{53}H_{76}O_3$	Nem/Liq	419.7	0.91	2.17	103.54	[245]	

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{53}H_{76}O_3$			cholesteryl $\omega$ -[4-(4-hexylphenylethynyl)phenoxy]hexanoate			
	Sol/Smec	362.0	23.22	64.14		
	Smec/Smec	373.6	Not reported in paper			
	Smec/Nem	439.9	0.76	1.73		
$C_{53}H_{76}O_3$	Nem/Liq	453.5	3.81	8.40		[245]
			cholesteryl $\omega$ -[4-(4-butylphenylethynyl)phenoxy]octanoate			
	Sol/Smec	420.1	44.22	105.26		
	Note: Smec/Nem not observed by dsc.					
$C_{53}H_{77}FO_4$	Nem/Liq	442.0	3.81	8.62	113.88	[245]
			cholesteryl 2-fluoro-4-(4-dodecylbenzoyloxy)benzoate			
	Sol/Smec	384.2	24.4	63.51		
	Smec/Meso	417.2	Could not be measured			
$C_{53}H_{77}FO_5$	Meso/Nem	419.7	0.23	0.55		
	Nem/Liq		Decomposed prior to transition			[333]
			cholesteryl 2-fluoro-4-(4-dodecyloxybenzoyloxy)benzoate			
	Sol/Smec	384.2	26.4	68.71		
$C_{53}H_{78}F_2O_4$	Smec/Meso	405.7	Could not be measured			
	Meso/Nem	419.7	0.21	0.50		
	Nem/Liq		Decomposed prior to transition			[333]
			cholesteryl 4-[4'-(2,3-difluoro-4-decyloxy)biphenyloxy]butanoate			
$C_{53}H_{78}F_2O_4$	Sol/Smec	354.6	44.4	125.21		
	Smec/Smec	379.0	Enthalpy not reported in paper			
	Smec/Liq	447.5	10.7	23.91		[132]
$C_{53}H_{78}F_2O_4$			cholesteryl 4-[4'-(2,3-difluoro-4-octyloxy)biphenyloxy]hexanoate			
	Sol/Smec	340.6	20.3	59.60		
	Smec/Smec	365.7	Enthalpy not reported in paper			
	Smec/Meso	424.4	0.7	1.65		
$C_{53}H_{80}O_4$	Meso/Liq	434.3	4.3	9.90		[132]
			cholesterily 5-[4-(4'-octyloxy)biphenyloxy]hexanoate			
	Sol/Smec	384.7	24.69	64.18		
	Smec/Meso	445.4	Not reported in paper			
$C_{53}H_{80}O_4$	Meso/Nem	445.6	1.05	2.36		
	Note: Smec/Meso transition enthalpy is included in Meso/Nem value.					
	Nem/Liq	450.7	3.12	6.92	73.46	[74]
$C_{53}H_{82}O_8$			2-(1,4,7-trioxaoctyl)-3,6,7,10,11-pentakis(hexyloxy)triphenylene			
	Sol/Meso	328.6	38.55	117.32		
$C_{54}H_{58}F_4O_{10}$	Meso/Liq	342.5	1.86	5.43	122.75	[83]
$C_{54}H_{58}F_4O_{10}$			1,3-phenylene bis[4-(3-fluoro-4-decyloxybenzoyloxy)-2-fluoro-benzoate]			
	Sol/Meso	378.2	57.1	150.98		
	Meso/Liq	386.7	21.6	55.86	206.84	283.0
$C_{54}H_{58}N_2O_8$						[247]
	Sol/Nem	393.2	52.0	132.25		
	Nem/Liq	420.2	2.6	6.19	138.44	[54]
$C_{54}H_{60}Cl_2F_2N_2O_6$			bis[4-ethoxyphenyl] 2-{12-[4-(4-biphenylazo)phenoxy]dodecyloxy}-terephthalate			
	Sol/Smec	381.2	50.5	132.48		
	Smec/Liq	410.2	5.1	12.43	144.91	[415]
$C_{54}H_{60}Cl_2O_6$			4,6-dichloro-1,3-phenylene bis[4-(4-decyloxy-3-fluoro-phenyliminomethyl)benzoate]			
	Sol/Meso	394.9	17.89	45.30		
	Meso/Liq	410.6	13.84	33.71	79.01	[177]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{54}H_{61}ClO_{10}$	Sol/Nem	362.2	62.5	172.56		[438]
	Nem/Liq	369.2	0.6	1.63	174.19	
$C_{54}H_{62}Cl_2N_2O_6$	Sol/Nem	357.2	61	170.77		[86]
	Nem/Liq	405.2	1.7	4.20	174.97	
$C_{54}H_{62}Cl_2N_2O_6$	Sol/Nem	384.3	43.8	113.97		[416]
	Nem/Liq	413.2	1.5	3.63	117.60	
$C_{54}H_{62}N_2O_9$						
	Sol/Smec	450.2	15.4	34.21		
	Smec/Nem	453.2	12.8	28.24		
	Nem/Liq	453.7	Not reported in paper	46.45		[289]
Note: Nem/Liq transition enthalpy is included in the Smec/Nem value.						
$C_{54}H_{62}N_4O_{10}$						
	Sol/Sol		Not reported in paper			
	Sol/Meso	386.2	25.74	66.65		[321]
$C_{54}H_{63}Cl N_2O_6$	Meso/Liq	427.2	15.86	37.13	103.78	
	Sol/Nem	365.2	40	109.53		[86]
$C_{54}H_{64}N_2O_2$	Nem/Liq	377.2	0.6	1.59	111.12	
	Sol/Nem	521.7	31.3	60.00		[388]
$C_{54}H_{64}N_2O_6$	Nem/Liq	600.2	0.3	0.50	60.50	
	Sol/Smec	417.2	37.0	88.69		[86]
$C_{54}H_{64}N_2O_6$	Smec/Liq	441.2	24.0	54.40	143.09	
	Sol/Meso	404.2	43.5	107.62		[417]
$C_{54}H_{64}N_2O_8$	Meso/Meso	421.2	39.4	93.54		
	Meso/Liq	426.2	15.8	37.07	238.23	
$C_{54}H_{70}O_4$	Sol/Meso	390.2	20.0	51.26		[412]
	Meso/Liq	451.7	21.0	46.49	97.75	
$C_{54}H_{72}N_2O_6$						
	Sol/Sol	383.2	22	57.41		
	Sol/Sol	390.2	18	46.13		
	Sol/Sol	425.2	7.0	16.46		
	Sol/Smec	430.2	29	67.41		
	Smec/Smec	438.2	Not detected by dsc			
	Smec/Smec	447.2	10.5	23.48		
$C_{54}H_{74}N_4O_4$	Smec/Liq	460.7	22	47.75	258.64	[120]
$C_{54}H_{72}N_2O_6$						
	Sol/Smec	397.2	50.0	125.88		[293]
$C_{54}H_{74}N_4O_4$	Smec/Liq	408.2	15.9	38.95	164.83	
$C_{54}H_{74}N_4O_4$						
	Sol/Smec		Not reported in paper			
	Smec/Nem	317.4	0.23	0.72		[360]
	Nem/Liq	357.5	2.5	6.99		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)						
$C_{54}H_{74}N_4O_6$	Sol/Nem	$\alpha,\omega$ -bis(4-dodecyloxyazobenzene-4'-carbonyloxy)butane		416.0	122	293.27		
	Nem/Liq	424.2	17.6		41.49	334.76		[107]
$C_{54}H_{74}O_6$	Sol/Col	393.7	52.1		132.33			
	Col/Liq	496.3	4.1		8.26	140.59		[6]
$C_{54}H_{78}O_2$	4,4''-dipentadecyloxy-p-quaterphenyl							
	Sol/Sol	415.0	50.6		121.93			
	Sol/Sol	444.0	4.88		10.99			
	Sol/SmeC	529.0	13.75		25.99			
$C_{54}H_{78}O_3$	SmeC/Liq	550.0	15.40		28.00	186.91	301.8	
	cholesteryl $\omega$ -[4-(4-heptylphenylethylnyl)phenoxy]hexanoate							
	Sol/SmeC	381.2	26.74		70.15			
	SmeC/Nem	444.0	1.16		2.61			
$C_{54}H_{78}O_3$	Nem/Liq	451.9	4.03		8.92	81.68		[245]
	cholesteryl $\omega$ -[4-(4-pentylphenylethylnyl)phenoxy]octanoate							
	Sol/Nem	418.9	43.88		104.75			
	Nem/Liq	441.6	3.57		8.08	112.83		[245]
$C_{54}H_{80}D_4N_4$	7,12-dihydro-5,7-di(octadecyl-1,1-d <sub>2</sub> )quinoxalino[2,3-b]-phenazinium (inner salt)							
	Sol/Meso	395.2	45.5		115.13			
	Meso/.Meso	439.8	19.3		43.88			
$C_{54}H_{80}F_2O_4$	Meso/Liq	459.0	28.8		62.75	221.76		[288]
	cholesteryl 4-[4'-(2,3-difluoro-4-decyloxy)biphenyloxy]pentanoate							
	Sol/SmeC	356.6	47.1		132.08			
	SmeC/Meso	394.9	0.7		1.77			
$C_{54}H_{84}O_6$	Meso/Liq	394.9	0.7		1.77	135.62		[132]
	2,3,6,7,10,11-hexa-n-hexyloxytriphenylene							
	Sol/Sol	105.9	1.16		10.95			
	Sol/Sol	220.9	1.95		8.83			
$C_{54}H_{84}O_6$	Sol/Sol	233.9	1.30		5.56			
	Sol/Sol	330.8	3.06		9.25			
	Sol/Sol	337.1	2.5		7.42			
	Sol/Meso	340.3	39.05		114.75			
$C_{54}H_{84}O_6$	Meso/Liq	371.2	5.24		14.12	170.88	301.3	
	Independent values from another reference							
	Sol/Col	Not reported in paper						
	Col/Liq	364.2	4.9		13.45			[368]
$C_{54}H_{86}N_2O_6$	$N,N'$ -didodecanoyl-2,4-bis(dodecanoyloxy)-1,3-benzenediamine							
	Sol/Disc	370.2	26.0		70.23			
	Disc/Liq	390.2	15.0		38.44	108.67	357.6	
$C_{54}H_{90}O_{12}$	benzene hexa-n-octanoate							
	Sol/Sol	301.9	49.0		164.0			
	Sol/Meso	355.1	46.1		129.8			
$C_{54}H_{92}N_2O_{10}$	Meso/Liq	357.1	19.2		53.8	347.6	362.4	
	Sol/Disc	348.2	39.0		112.00			[343]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{54}H_{98}N_2O_4$	Disc/Liq	478.2	37.0	77.38	189.38	[188]
	Sol/Meso	381.3	49.0	128.51		
	Meso/Meso	399.2	4.0	10.02		
	Meso/Liq	451.2	2.0	4.43	142.96	341.9
$C_{55}H_{52}FNO_8$	N,N'-dipentadecanoyl-2,5,6-trimethyl-4-pentadecanoyloxy-1,3-benzenediamine					
	Sol/SmeC	432.7	42.8	98.91		
	SmeC/SmeC	450.0	0.14	0.31		
	SmeC/Liq	464.2	4.54	9.78	109.00	[411]
$C_{55}H_{52}FNO_8$	4-cyanophenyl 4-[3-[4-(4'-tetradecylbiphenyl-4-carbonyloxy)-2-fluorobenzoyloxy]benzoyloxy]benzoate					
	Sol/SmeC	426.2	43.4	101.83		
	SmeC/SmeC	447.2	0.10	0.22		
	SmeC/Liq	462.7	4.80	10.37	112.42	[411]
$C_{55}H_{53}NO_8$	4-cyanophenyl 4-[3-[4-(4'-tetradecylbiphenyl-4-carbonyloxy)-3-fluorobenzoyloxy]benzoyloxy]benzoate					
	Sol/SmeC	430.2	58.5	135.98		
	SmeC/SmeC	457.5	0.16	0.35		
	SmeC/Liq	472.4	4.76	10.08	146.41	[411]
$C_{55}H_{58}N_2O_9$	1,3-[4-[4-(4-hexyloxybenzoyloxy)benzylidene]aminophenoxy]propan-2-ol					
	Sol/Nem	457.2	38.42	84.03		
	Nem/Liq	571.1	3.08	5.39	89.42	[413]
$C_{55}H_{59}N_5O_4$	4-[[(1E)-(4-octylphenyl)azo]phenyl 3-[{(E)-[4-[4-(1E)-(4-octylphenyl)azo]phenoxy]carbonyl]phenyl]-methylene]amino]benzoate					
	Sol/Sol	Not reported in paper				
	Sol/Meso	425.2	43.7	101.36		
	Meso/Liq	493.2	18.03	36.56		[431]
$C_{55}H_{59}N_5O_6$	4-[[(1E)-(4-octyloxyphenyl)azo]phenyl 3-[{(E)-[4-[4-(1E)-(4-octyloxyphenyl)azo]phenoxy]carbonyl]phenyl]-methylene]amino]benzoate					
	Sol/Sol	Not reported in paper				
	Sol/Meso	473.7	59.74	126.11		
	Meso/Liq	503.7	19.6	38.91		[431]
$C_{55}H_{62}Cl_2O_6$	4-[[(1E)-[3-chloro-4-nonyloxyphenyl]ethenyl]benzoic acid, 2-methyl-1,3-phenylene ester					
	Sol/Meso	381.5	26.34	69.04		
	Meso/Liq	393.0	12.55	31.93	100.97	[177]
$C_{55}H_{63}N_3O_6$	2-cyano-1,3-phenylene bis[4-(4-decyloxyphenyliminomethyl)benzoate]					
	Sol/Meso	436.2	35.1	80.47		
	Meso/Liq	466.7	27.9	59.78	140.25	[246]
$C_{55}H_{67}N_3O_6$	4-[3,4-bis(4-undecyloxybenzoyloxy)benzylideneamino]azobenzene					
	Sol/Nem	355.2	65.0	183.00		
	Nem/Liq	394.2	0.5	1.27	184.27	[205]
$C_{55}H_{72}O_4$	1,7-bis(4-(4-decyloxyphenylethynyl)phenoxy)heptane					
	Sol/Sol	383.2	22	57.41		
	Sol/SmeC	404.2	62	153.39		
	SmeC/Liq	419.2	12	28.63	239.43	278.7
$C_{55}H_{74}N_2O_8$	1,5-[4-(2-hydroxy-4-decyloxybenzylideneamino)benzoyloxy]-2,2-dimethylpentane					
	Sol/SmeC	369.2	37.0	100.22		
	SmeC/Liq	406.2	12.0	29.54	129.76	[412]
$C_{55}H_{74}N_2O_8$	1,5-[4-(2-hydroxy-4-decyloxybenzylideneamino)benzoyloxy]-3,3-dimethylpentane					
	Sol/SmeC	373.2	59.0	158.09		
	SmeC/Liq	420.2	14.0	33.32	191.41	[412]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta H_{\text{pcc}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pcc}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)						
$C_{55}H_{76}N_4O_6$	Sol/Smec	366.9	$\alpha,\omega$ -bis(4-dodecyloxyazobenzene-4'-carbonyloxy)pentane	36.4	99.21			[107]
	Smec/Liq	380.8		22.2	58.30	157.51		
$C_{55}H_{77}NO_9$	Sol/Smec	378.6	cholest-5-en-3-ol (3 $\beta$ , 6-[4-[E]-[[4-[4(R,5R)-4,5-bis(ethoxycarbonyl)-1,3-dioxolan-2-yl]phenyl]-imino]methyl]phenoxy]hexanoate	15.89	41.97			[362]
	Smec/Chol	396.7		0.12	0.30			
	Chol/Liq	421.7		2.55	6.05	48.32		
$C_{55}H_{80}F_2O_5$	Sol/Smec	386.2	cholesteryl 2-fluoro-4-(3-fluoro-4-tetradecyloxybenzoyloxy)benzoate	31.8	82.34			[333]
	Smec/Meso	464.2		Could not be measured				
	Meso/Nem	472.2		0.10	0.21			
	Nem/Liq			Decomposed prior to transition				
$C_{55}H_{80}O_3$	Sol/Sol	359.8	cholesteryl $\omega$ -[4-(4-hexylphenylethynyl)phenoxy]octanoate		Not reported in paper			[245]
	Sol/Smec	364.1		13.18	36.20			
	Smec/Smec	371.1		Not reported in paper				
	Smec/Nem	440.7		0.87	1.97			
	Nem/Liq	453.5		4.10	9.04			
$C_{55}H_{81}FO_4$	Sol/Smec	372.7	cholesteryl 2-fluoro-4-(4-tetradecylbenzoyloxy)benzoate	37.8	101.42			[333]
	Note: Sol/Smec transition enthalpy includes the enthalpy of a Sol/Sol transition.							
	Sol/Meso	431.7		Could not be measured				
	Meso/Liq			Decomposed prior to transition				
$C_{55}H_{81}FO_5$	Sol/Smec	358.7	cholesteryl 2-fluoro-4-(4-tetradecyloxybenzoyloxy)benzoate	27.2	75.83			[333]
	Smec/Meso	436.2		Could not be measured				
	Meso/Nem	445.2		0.35	0.79			
	Nem/Liq			Decomposed prior to transition				
	Sol/Smec	335.3	cholesteryl 4-[4'-(2,3-difluoro-4-decyloxy)biphenyloxy]hexanoate	36.9	110.05			
$C_{55}H_{82}F_2O_4$	Smec/Smec	376.8		Enthalpy not reported in paper				[132]
	Smec/Meso	428.1		2.5	5.84			
	Meso/Liq	430.0		4.1	9.53			
	Sol/Sol	327.9		0.5	1.52			
$C_{55}H_{85}N_3O_9$	Sol/Gel	359.5		30.7	85.40			[379]
	Gel/Liq	377.6		0.7	1.85	88.77		
	Sol/Meso	437.7	1,3-phenylene bis[4-(3-chlorobenzoyloxy)] 4'-pentylbiphenyl-4-carboxylate	54.78	125.15			
$C_{56}H_{48}Cl_2O_8$	Meso/Meso	441.2		0.64	1.45			[276]
	Meso/Liq	451.7		12.51	27.70	154.30		
	Sol/Meso	416.7	1,3-phenylene bis[4-(4'-n-pentylbiphenyl-4-carbonyloxy)-2-fluorobenzoate]	29.61	71.06			
$C_{56}H_{50}F_2O_8$	Meso/Meso	505.2		0.02	0.04			[26]
	Meso/Liq	507.2		16.87	33.26	104.36		
	Sol/Meso	430.2	1,3-phenylene bis[4-(4'-n-pentylbiphenyl-4-carbonyloxy)-3-fluorobenzoate]	32.59	75.76			
$C_{56}H_{50}F_2O_8$	Meso/Meso	473.2		0.41	0.87			[26]
	Meso/Liq	500.7		15.0	29.96	106.59		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula	Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
		T (K)	$\Delta H_{\text{pce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pce}}$		
$C_{56}H_{50}O_8$	Sol/Meso	448.2	36.22	80.81		
	Meso/Meso	519.2	0.41	0.79		
	Meso/Liq	530.2	18.79	35.44	117.04	[26]
$C_{56}H_{58}N_2O_8S$	Sol/Nem	406.7	40.33	99.16		
	Nem/Liq	420.8	0.94	2.23	101.39	[234]
$C_{56}H_{58}N_4O_4$	Sol/Sol	378.2	10.2	26.97		
	Sol/SmeC	451.2	32.1	71.14		
	SmeC/Liq	479.2	16.5	34.43	132.54	[331]
$C_{56}H_{62}Cl_2O_6$	Sol/Meso	391.2	18.80	48.06		
	Meso/Liq	413.6	16.18	39.12	87.18	[177]
$C_{56}H_{62}F_4O_{10}$	Sol/Meso	351.2	23.9	68.05		
	Meso/Liq	360.7	21.6	59.88	127.93	297.2
						[247]
$C_{56}H_{62}F_4O_{10}$	Sol/Sol	Not reported in paper				
	Sol/Meso	379.2	92.4	243.67		
	Note: Sol/Sol transition enthalpy is included in the Sol/Meso value.					
$C_{56}H_{64}Cl_2F_2N_2O_6$	Meso/Liq	388.7	22.1	56.86	300.53	297.2
						[247]
$C_{56}H_{64}Cl_2F_2N_2O_6$	Sol/SmeC	380.2	52.8	138.87		
	SmeC/SmeC	389.2	Not observed by dsc			
	SmeC/Liq	412.2	5.7	13.82	152.69	[415]
$C_{56}H_{64}O_{12}$	Sol/SmeC	427.2	62.16	145.51		
	SmeC/SmeC	474.2	0.08	0.17		
	SmeC/Liq	484.2	16.91	34.92	180.60	[119]
$C_{56}H_{65}ClO_{10}$	Sol/Nem	361.2	66.0	182.72		
	Nem/Liq	368.2	0.6	1.63	184.35	[438]
$C_{56}H_{66}N_2O_9$	Sol/SmeC	440.2	15.4	34.98		
	SmeC/Liq	458.2	15.2	33.17	68.15	[289]
$C_{56}H_{66}O_8S_2$	Sol/Meso	377.2	31.7	84.04		
	Meso/Liq	385.2	16.3	42.32	126.36	NA
						[318]
$C_{56}H_{68}N_2O_2$	Sol/Nem	517.3	37.7	72.88		
	Nem/Liq	622.6	0.2	0.32	73.20	[388]
$C_{56}H_{68}N_2O_8$	Sol/Meso	387.2	20.0	51.65		
	Meso/Liq	450.2	20.0	44.42	96.07	[412]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula Transition	Compound					
	T (K)	$\Delta H_{\text{pcc}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pcc}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (estimated)	Ref.
$C_{56}H_{68}O_7$	7-[4-((4-[4-nonylphenoxy]carbonyl)phenoxy)carbonyl]phenolheptyl] 4'-heptyl-1,1'-biphenylcarboxylate					
	Sol/Sol	352.1	12.36	35.10		
	Sol/Smec	401.2	46.50	115.90		
	Smec/Liq	454.2	12.09	26.61	117.61	[228]
$C_{56}H_{74}O_4$	1,8-bis(4-(4'-decyloxyphenylethynyl)phenoxy)octane					
	Sol/Sol	382.2	23.0	60.18		
	Sol/Sol	410.2	40.0	97.51		
	Sol/Smec	418.2	19.0	45.43		
	Smec/Smec	434.2	11.0	25.33		
$C_{56}H_{76}N_2O_6$	hexanedioic acid, bis[4-[[4-(dodecyloxy)phenyl]imino]methyl]phenyl] ester					
	Sol/Meso	418.0	35.80	85.65		
	Meso/Meso	446.2	10.17	22.79		
	Meso/Liq	495.2	23.28	47.01	155.45	[295]
$C_{56}H_{76}N_2O_6$	decanedioic acid, bis[4-[[4-(decyloxy)phenyl]imino]methyl]phenyl] ester					
	Sol/Meso	412.2	36.93	89.59		
	Meso/Meso	446.2	15.31	34.31		
	Meso/Smec	449.4	0.28	0.62		
$C_{56}H_{76}N_2O_6$	Smec/Liq	454.7	23.84	52.43	176.95	[295]
	Sol/Meso	406.5	44.70	109.96		
	Meso/Meso	442.9	9.24	20.86		
	Meso/Liq	482.7	27.5	56.97	187.79	[295]
$C_{56}H_{76}N_2O_6$	hexanedioic acid, bis[4-[4-(dodecyloxy)benzylideneamino]phenyl] ester					
	Sol/Meso	397.7	40.92	102.89		
	Meso/Meso	443.3	Not observed by dsc			
	Meso/Liq	444.2	40.52	91.22	194.11	[295]
$C_{56}H_{78}N_4O_6$	$\alpha,\omega$ -bis(4-dodecyloxyazobenzene-4'-carbonyloxy)hexane					
	Sol/Nem	391.1	185	473.02		
$C_{56}H_{82}O_2$	Nem/Liq	406.0	20.1	49.51	522.53	[107]
	Sol/Sol	414.0	56.30	135.99		
$C_{56}H_{82}O_2$	Sol/Sol	444.0	4.88	10.99		
	Sol/Sol	524.0	14.67	28.00		
	Sol/Smec	533.0	2.67	5.01		
	Smec/Liq	546.0	13.65	25.00	204.99	
					316.0	[111]
$C_{56}H_{82}O_3$	cholesteryl $\omega$ -[4-(4-heptylphenylethynyl)phenoxy]octanoate					
	Sol/Smec	401.5	38.07	94.82		
	Smec/Meso	413.0	Not reported in paper			
	Meso/Nem	415.6	Not reported in paper			
$C_{56}H_{88}O_{12}$	Nem/Liq	437.3	4.90	11.21		[245]
	Sol/Col	353.7	38.46	108.74		
	Col/Liq	358.7	10.25	28.58	137.32	[275]
	Sol/Col	Below ambient room temperature				
$C_{56}H_{94}N_2O_6$	Col/Liq	345.3	5.26	15.23		[178]
	Sol/Col	2,5-bis(3,4,5-triheptyloxyphenyl)-1,3,4-oxadiazole				
	Col/Liq	Below ambient room temperature				
$C_{57}H_{56}FNO_8$	Sol/Smec	431.2	48.9	113.40		
	Smec/Smec	451.4	0.06	0.13		
	Smec/Liq	470.2	5.8	12.34	125.87	[411]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula Transition	Compound					Ref.
	T (K)	$\Delta H_{\text{pcc}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pcc}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (estimated)	
$C_{57}H_{56}FNO_8$		4-cyanophenyl 4-[3-[4-(4'-hexadecylbiphenyl-4-carbonyloxy)-3-fluorobenzoyloxy]benzoyloxy]benzoate				
	Sol/Smec	421.7	47.0	111.45		
	Smec/Smec	433.2	0.34	0.78		
	Smec/Smec	449.4	0.17	0.38		
$C_{57}H_{57}NO_8$	Smec/Liq	468.2	5.8	12.39	125.00	[411]
		4-cyanophenyl 4-[3-[4-(4'-hexadecylbiphenyl-4-carbonyloxy)benzoyloxy]benzoyloxy]benzoate				
	Sol/Smec	429.2	38.4	89.47		
	Smec/Smec	458.2	0.1	0.22		
$C_{57}H_{62}N_2O_9$	Smec/Liq	477.7	5.8	12.14	101.83	[411]
		1,3-{4-[4-(4-heptyloxybenzoyloxy)benzylidene]aminophenoxy}propan-2-ol				
	Sol/Smec	453.0	19.37	42.76		
	Smec/Nem	469.8	0.27	0.57		
$C_{57}H_{63}N_5O_5$	Nem/Liq	553.0	1.68	3.04	46.37	[413]
		4-[{(E)-(4-octyloxyphenyl)azo}phenyl 3-[(E)-[4-[4-(1E)-(4-decylphenyl)azo]phenoxy]carbonyl]phenyl]-methylene]amino]benzoate				
	Sol/Sol		Not reported in paper			
	Sol/Meso	442.7	54.98	124.19		
$C_{57}H_{66}Cl_2O_6$	Meso/Meso	505.2	19.26	38.12		[431]
		4-[{(1E)-[3-chloro-4-decyloxyphenyl]ethenyl]benzoic acid, 2-methyl-1,3-phenylene ester}				
	Sol/Meso	380.4	31.67	83.25		
	Meso/Liq	399.2	11.57	28.98	112.23	[177]
$C_{57}H_{66}O_{12}$		(R)-4-[4-(1-methylheptyloxy carbonyl)phenyloxycarbonyl]phenyl 4-{5-[4-(4-decyloxybenzoyloxy)-benzoyloxy]butoxy}benzoate				
	Sol/Smec	393.2	45.11	114.73		
	Smec/Liq	413.2	15.12	36.59	151.32	[119]
		(R)-4-[4-(1-methylheptyloxy carbonyl)phenyloxycarbonyl]phenyl 4-{5-[4-(4-decyloxybenzoyloxy)-benzoyloxy]pentyloxy}benzoate				
$C_{57}H_{66}O_{12}$	Sol/Smec	421.2	62.33	147.98		
	Smec/Liq	458.7	15.86	34.58	182.56	[119]
		2-cyano-1,3-phenylene bis[4-(4-undecyloxyphenyliminomethyl)benzoate]				
	Sol/Meso	435.2	42.8	98.35		
$C_{57}H_{71}N_3O_6$	Meso/Liq	466.2	30.8	66.07	164.42	[246]
		4-[3,4-bis(4-dodecyloxybenzoyloxy)benzylideneamino]azobenzene				
	Sol/Nem	355.2	73.0	205.52		
	Nem/Liq	394.2	0.6	1.52	207.04	[205]
$C_{57}H_{76}O_4$		1,9-bis[4-(4'-decyloxyphenylethynyl)phenoxy]nonane				
	Sol/Sol	381.2	21	55.09		
	Sol/Smec	397.2	45	113.29		
	Smec/Smec	401.2	13	32.40		
$C_{57}H_{78}N_2O_6$	Smec/Liq	410.2	12	29.25	230.03	[120]
		heptanedioic acid, bis[4-[[4-(dodecyloxy)phenyl]imino]methyl]phenyl ester				
	Sol/Meso	428.7	11.98	27.94		
	Meso/Smec	435.9	52.41	120.23		
$C_{57}H_{78}N_2O_8$	Smec/Liq	455.5	18.45	40.50	188.67	[295]
		malonic acid, bis{10-[4-(4-butyoxypyrenylimino)methyl]phenoxy}decyl ester				
	Sol/Smec	379.0	49.6	130.87		
	Smec/Liq	394.4	15.4	39.05	169.92	[244, 298]
$C_{57}H_{78}N_2O_8$		1,5-[4-(2-hydroxy-4-undecyloxybenzylideneamino)benzoyloxy]-2,2-dimethylpentane				
	Sol/Smec	369.2	37.0	100.22		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
	T (K)	$\Delta H_{\text{pcc}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pcc}}$		
Smec/Liq	406.2	12.0	29.54	129.76	[412]
$C_{57}H_{78}N_2O_8$					
	Sol/Smec	372.2	55.0	147.77	
	Smec/Liq	421.2	15.0	35.61	183.38
$C_{57}H_{80}N_4O_6$					
	Sol/Smec	358.2	48.1	134.28	
	Smec/Liq	378.2	23.5	62.14	196.42
$C_{57}H_{82}N_2O_8$					
	Sol/Meso	411.2	42.0	102.14	
	Meso/Liq	411.2	17.0	41.34	143.48
Note: Paper reported identical temperatures for the Sol/Meso and Meso/Liq transitions.					
$C_{57}H_{84}F_2O_5$					
	Sol/Smec	373.5	33.7	90.23	
	Smec/Meso	468.2	Could not be measured		
	Meso/Meso	469.2	Could not be measured		
	Meso/Nem	480.2	Could not be measured		
$C_{57}H_{85}FO_4$	Nem/Liq		Decomposed prior to transition		[333]
	Sol/Smec	375.7	29.0	77.19	
	Smec/Meso	443.7	Could not be measured		
	Meso/Nem	450.7	0.05	0.11	
$C_{57}H_{85}FO_5$	Nem/Liq		Decomposed prior to transition		[333]
	Sol/Smec	375.2	28.1	74.89	
	Smec/Meso	448.7	Could not be measured		
	Meso/Nem	457.2	0.24	0.52	
$C_{57}H_{86}F_2O_4$	Nem/Liq		Decomposed prior to transition		[333]
	Sol/Smec	375.1	41.8	111.44	
	Smec/Meso	405.6	0.6	1.48	
	Meso/Liq	415.2	4.4	10.60	123.52
$C_{57}H_{89}N_3O_6$					
	Sol/Meso		Not reported in paper		
	Meso/Liq	334.7	3.25	9.71	[259]
$C_{57}H_{90}O_6$					
	Sol/Col	319.2	21.61	67.70	
	Col/Liq	340.2	2.11	6.20	[261]
Note: Col/Liq transition enthalpy was determined from cooling measurements.					
Independent values from another reference					
Sol/Col					
	Col/Liq	341.2	1.7	4.98	[368]
Note: Col/Liq transition enthalpy was determined from cooling measurements.					
$C_{57}H_{104}N_2O_4$					
	Sol/Meso	382.3	34.0	88.94	
	Meso/Meso	395.2	2.0	5.06	
	Meso/Liq	447.2	2.0	4.47	98.47
$C_{57}H_{105}N_3O_3$				363.2	[193]
	Sol/Meso	393.2	28.0	71.21	
	Meso/Meso	448.2	12.0	26.77	
	Meso/Liq	601.2	9.0	14.97	112.95
					[190]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula Transition	Compound					Ref.
	T (K)	$\Delta H_{\text{pcce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pcce}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	
$C_{58}H_{52}Cl_2O_8$	Sol/Meso	436.2	56.07	128.54		
	Meso/Liq	446.2	15.47	34.67	163.21	237.6 [276]
$C_{58}H_{52}F_2O_8$	Sol/Meso	407.2	37.55	92.22		
	Meso/Liq	495.7	21.58	43.53	135.75	[26]
$C_{58}H_{52}F_2O_8$	Sol/Meso	427.2	23.28	54.49		
	Meso/Liq	493.2	19.58	39.70	94.19	[26]
$C_{58}H_{54}O_8$	Sol/Meso	438.2	39.55	90.26		
	Meso/Liq	513.2	23.62	46.02	136.28	[26]
$C_{58}H_{54}O_8$	Sol/Meso	416.2	52.33	125.73		
	Meso/Liq	474.2	13.43	28.32	154.05	[26]
$C_{58}H_{54}O_8$	Sol/Meso	434.7	43.65	100.41		
	Meso/Liq	439.7	12.52	28.47	128.88	222 [276]
$C_{58}H_{66}F_4O_{10}$	Sol/Sol	1,3-phenylene bis[4-(3-fluoro-4-dodecyloxybenzoyloxy)-2-fluorobenzoate]				
	Sol/Meso	372.7	92.8	248.99		
	Note: Sol/Sol transition enthalpy is included in Sol/Meso value.					
	Meso/Liq	390.7	23.1	59.12	308.11	311.4 [247]
$C_{58}H_{68}Cl_2F_2N_2O_6$	Sol/Smec	376.2	57.1	151.78		
	Smec/Smec	398.2	Not observed by dsc			
	Smec/Liq	412.2	6.1	14.80	166.58	[415]
$C_{58}H_{68}Cl_2O_6$	Sol/Meso	388.7	17.80	45.79		
	Meso/Liq	409.8	17.52	42.75	88.54	[177]
$C_{58}H_{68}O_{12}$	(R)-4-[4-(1-methylheptyloxycarbonyl)phenyloxycarbonyl]phenyl 4-{5-[4-(4-decyloxybenzoyloxy)-benzoyloxy]pentyloxy}benzoate					
	Sol/Smec	409.2	54.43	133.02		
	Smec/Smec	415.2	0.01	0.02		
	Smec/Smec	425.2	0.04	0.09		
	Smec/Liq	457.7	15.98	34.91	168.04	[119]
$C_{58}H_{69}ClO_{10}$	bis-4-[4-(dodecyloxy)benzoyl]oxy]benzoic acid, 4-chloro-1,3-phenylene ester					
	Sol/Liq	371.2	38.7	104.26		
	Meso/Liq	353.2	9.1	25.76		
	Nem/Liq	368.2	0.65	1.77		[300]
Note: Authors report melting to an isotropic liquid, which was immediately followed by a spontaneous ordering to a chiral mesophase.						
$C_{58}H_{70}Cl_2N_2O_6$	4,5-dichloro-1,3-phenylene bis[4-(4-dodecyloxybenzylideneamino)benzoate]					
	Sol/Nem	363.2	41.0	112.89		
$C_{58}H_{70}Cl_2N_2O_6$	Nem/Liq	397.2	1.6	4.03	116.92	313.4 [86]
	Sol/Smec	384.2	47.2	122.85		
	Smec/Nem	394.2	0.9	2.28		
	Nem/Liq	410.2	1.9	4.63	129.76	[416]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula Transition	Compound					Ref.
	T (K)	$\Delta H_{\text{pcc}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pcc}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (estimated)	
$C_{58}H_{70}N_2O_9$	Sol/Smec	435.2	16.9	38.83		[289]
	Smec/Liq	464.2	16.9	36.41	75.24	
$C_{58}H_{70}N_4O_{10}$	Sol/Sol	1,3-phenylene bis[[4-(methylamino)phenyl]-3-nitro-4-dodecyloxybenzoate]				[321]
	Not reported in paper					
	Sol/Meso	384.9	28.44	73.89		
	Meso/Liq	431.2	18.62	43.18		
$C_{58}H_{70}O_8S_2$	Sol/Meso	1,3-phenylene 4-[[4-(dodecyloxy)benzoyl]thio]benzoate				[318]
	382.2	38.48	100.68			
	387.2	18.6	48.04	148.72	NA	
$C_{58}H_{71}ClN_2O_6$	Sol/Meso	4-chloro-1,3-phenylene bis[4-(4-dodecyloxyphenyliminomethyl)benzoate]				[42]
	388.2	16.8	43.28			
	415.2	16.5	39.74	83.02	310.8	
$C_{58}H_{71}ClN_2O_6$	Sol/Nem	4-chloro-1,3-phenylene bis[4-(4-dodecyloxybenzylideneamino)benzoate]				[86]
	369.2	48.0	130.01			
	Nem/Liq	374.2	0.6	1.60	131.61	
$C_{58}H_{72}N_2O_6$	Sol/Smec	1,3-phenylene bis[4-(4-dodecyloxyphenyliminomethyl)benzoate]				[86]
	416.2	38.0	91.30			
	441.2	26.0	58.93	150.23	308.2	
$C_{58}H_{72}N_2O_6$	Sol/Meso	isophthalidene bis[4-(4-dodecyloxyphenyloxycarbonyl)aniline]				[417]
	409.2	53.5	130.74			
	Meso/Meso	418.2	38.1	91.10		
	Meso/Liq	427.2	15.4	36.05	257.89	
$C_{58}H_{72}N_2O_8$	Sol/Meso	1,3-bis[4-(2-hydroxy-4-dodecyloxybenzylideneamino)benzoyloxy]benzene				[412]
	387.2	21.0	54.24			
	Meso/Liq	451.2	22.0	48.76	103.00	
$C_{58}H_{74}O_7$	Sol/Smec	7-[4-((4-[4-nonylphenoxy]carbonyl)phenoxy)carbonyl)phenolheptyl] 4'-nonyl-1,1'-biphenylcarboxylate				[228]
	393.5	45.53	115.71			
	Smec/Liq	455.4	17.59	38.63	54.34	
$C_{58}H_{78}O_4$	Sol/Sol	1,10-bis(4-(4'-dodecyloxyphenylethynyl)phenoxy)decane				[120]
	384.2	23	59.86			
	Sol/Sol	410.2	33	80.56		
	Sol/Smec	423.2	32	75.61		
	Smec/Liq	428.2	31	72.40	288.43	
$C_{58}H_{82}N_4O_6$	Sol/Nem	$\alpha,\omega$ -bis(4-dodecyloxyazobenzene-4'-carbonyloxy)octane				[107]
	386.6	82.4	213.14			
	Nem/Liq	396.5	15.9	40.10	253.24	
$C_{58}H_{86}O_2$	Sol/Sol	4,4'''-diheptadecyloxy-p-quaterphenyl				[111]
	416.0	67.39	162.00			
	Sol/Sol	444.0	4.44	10.00		
	Sol/Smec	519.0	20.24	39.00		
	Smec/Liq	542.0	16.80	31.00	242.00	
$C_{58}H_{89}NO_3$	Sol/Smec	cholest-5-en-3-ol ( $3\beta$ ), 6-[4-[(E)-[(4-dodecylphenyl)imino]]methyl-phenoxy]hexanoate				[362]
	336.8	24.21	71.88			
	Smec/Smec	378.0	Too small to be measured			
	Smec/Liq	447.9	11.21	25.03	96.91	
$C_{58}H_{96}O_{10}$	Sol/Col	2,3,6,7-tetrakis(decyloxy)-1,5-bis(2-hydroxyethoxy)-9,10-anthracenedione				
		355.2	89.30	251.41		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.	
	T (K)	$\Delta H_{\text{pcc}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pcc}}$			
C <sub>59</sub> H <sub>60</sub> FNO <sub>8</sub>	Col/Col	380.9	1.33	3.49	[275]	
	Col/Liq	395.7	12.38	31.29		
C <sub>59</sub> H <sub>60</sub> FNO <sub>8</sub>	Sol/Smec	430.7	49.9	115.86	[411]	
	Smec/Smec	452.4	0.05	0.11		
	Smec/Liq	475.2	6.60	13.89		
				129.86		
C <sub>59</sub> H <sub>60</sub> FNO <sub>8</sub>	Sol/Smec	422.2	44.5	105.40	[411]	
	Smec/Smec	439.2	0.10	0.23		
	Smec/Smec	449.7	0.12	0.27		
	Smec/Liq	472.7	6.36	13.45		
C <sub>59</sub> H <sub>61</sub> NO <sub>8</sub>	Sol/Smec	427.2	65.4	153.09	[411]	
	Smec/Smec	457.2	0.09	0.20		
	Smec/Smec	479.9	6.60	13.75		
				167.04		
C <sub>59</sub> H <sub>66</sub> N <sub>2</sub> O <sub>9</sub>	Sol/Smec	449.7	45.20	100.51	[413]	
	Smec/Nem	497.6	0.92	1.85		
	Nem/Liq	549.5	4.24	7.72		
				110.08		
C <sub>59</sub> H <sub>67</sub> N <sub>5</sub> O <sub>4</sub>	Sol/Sol	Not reported in paper			[431]	
	Sol/Meso	419.7	45.1	107.58		
	Meso/Liq	489.2	17.12	35.00		
C <sub>59</sub> H <sub>67</sub> N <sub>5</sub> O <sub>5</sub>	Sol/Sol	Not reported in paper			[431]	
	Sol/Meso	445.7	58.42	131.07		
	Meso/Meso	460.2	Not reported in paper			
	Meso/Liq	503.2	21.23	42.19		
C <sub>59</sub> H <sub>67</sub> N <sub>5</sub> O <sub>6</sub>	Sol/Sol	Not reported in paper			[431]	
	Sol/Meso	462.2	50.25	108.72		
	Meso/Meso	479.2	Not reported in paper			
	Meso/Liq	505.7	17.02	33.66		
C <sub>59</sub> H <sub>68</sub> O <sub>8</sub>	Sol/Meso	4-[[4'-(octyloxy)[1,1'-biphenyl]-4-yl]oxy]methyl]benzoic acid, 1,3-propanediyl ester			[304]	
	Meso/Liq	483.7	38.29	79.16		
		490.1	23.99	48.95		
C <sub>59</sub> H <sub>70</sub> Cl <sub>2</sub> O <sub>6</sub>	Sol/Meso	384.8	39.36	102.29	[177]	
	Meso/Liq	394.5	3.27	8.29		
				110.58		
C <sub>59</sub> H <sub>70</sub> O <sub>12</sub>	Sol/Smec	386.2	44.31	114.73	[119]	
	Smec/Liq	405.2	16.17	39.91		
				154.64		
C <sub>59</sub> H <sub>70</sub> O <sub>12</sub>	Sol/Smec	395.2	62.43	157.97	[119]	
	Smec/Smec	423.2	0.07	0.17		
	Smec/Liq	454.2	15.18	33.42		
				191.56		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
	T (K)	$\Delta H_{\text{pcc}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pcc}}$		
$C_{59}H_{71}N_3O_4$		4-cyano-1,3-phenylene bis[4-(4-dodecylyphenyliminomethyl)benzoate]			
	Sol/Meso	353.2	13.2	37.37	
	Meso/Smec	397.2	1.3	3.27	
$C_{59}H_{71}N_3O_6$	Sol/Meso	433.2	33.6	77.56	
	Meso/Liq	465.2	33.6	72.23	149.79
					[246]
$C_{59}H_{71}N_3O_6$		4-cyano-1,3-phenylene bis[4-(4-dodecyloxyphenyliminomethyl)benzoate]			
	Sol/Meso	338.2	14.7	43.47	
	Meso/Smec	395.2	4.8	12.15	
	Smec/Smec	414.2	Not reported in paper		
$C_{59}H_{82}N_2O_8$	Smec/Liq	461.2	7.2	15.61	71.23
					[252]
$C_{59}H_{82}N_2O_8$		1,5-[4-(2-hydroxy-4-dodecyloxybenzylideneamino)benzoyloxy]-2,2-dimethylpentane			
	Sol/Smec	374.2	44.0	117.58	
	Smec/Liq	407.2	13.0	31.93	149.51
$C_{59}H_{84}N_4O_6$		1,5-[4-(2-hydroxy-4-dodecyloxybenzylideneamino)benzoyloxy]-3,3-dimethylpentane			
	Sol/Smec	372.2	65.0	174.64	
	Smec/Liq	422.2	15.0	35.53	210.17
$C_{59}H_{84}N_4O_6$		$\alpha,\omega$ -bis(4-dodecyloxyazobenzene-4'-carbonyloxy)nonane			
	Sol/Smec	366.3	89.9	245.43	
	Smec/Liq	382.0	30.5	79.84	325.27
$C_{59}H_{85}FO_4$		cholesteryl 2-fluoro-4-(4-octadecylbenzoyloxy)benzoate			
	Sol/Smec	373.2	30.4	81.46	
	Smec/Meso	450.2	Could not be measured		
	Meso/Nem	458.2	Not reported in paper		
$C_{59}H_{88}F_2O_5$	Nem/Liq	Decomposed prior to transition			[333]
$C_{59}H_{89}FO_5$		cholesteryl 2-fluoro-4-(3-fluoro-4-octadecyloxybenzoyloxy)benzoate			
	Sol/Smec	376.2	36.4	96.76	
	Smec/Smec	466.2	Could not be measured		
	Smec/Meso	482.7	Could not be measured		
$C_{60}H_{56}Cl_2O_8$	Meso/Nem	485.7	Could not be measured		
	Nem/Liq	Decomposed prior to transition			[333]
$C_{60}H_{56}F_2O_8$		cholesteryl 2-fluoro-4-(4-octadecyloxybenzoyloxy)benzoate			
	Sol/Meso	359.2	27.8	77.39	
	Smec/Meso	454.7	Not reported in paper		
	Meso/Nem	462.2	Not reported in paper		
$C_{60}H_{56}F_2O_8$	Nem/Liq	Decomposed prior to transition			[333]
$C_{60}H_{56}F_2O_8$		1,3-phenylene bis[4-(3-chlorobenzoyloxy)] 4'-heptylbiphenyl-4-carboxylate			
	Sol/Meso	398.2	50.19	126.04	
	Meso/Liq	444.2	17.04	38.36	164.40
$C_{60}H_{56}F_2O_8$		1,3-phenylene bis[4-(4'-n-heptylbiphenyl-4-carbonyloxy)-2-fluorobenzoate]			
	Sol/Meso	398.2	29.71	74.61	
	Meso/Liq	488.2	22.46	46.01	120.62
$C_{60}H_{56}F_2O_8$		1,3-phenylene bis[4-(4'-n-heptylbiphenyl-4-carbonyloxy)-3-fluorobenzoate]			
	Sol/Meso	426.7	21.08	49.40	
	Meso/Liq	488.2	20.90	42.81	92.21
$C_{60}H_{58}O_8$		1,3-phenylene bis[4-(4'-n-heptylbiphenyl-4-carbonyloxy)benzoate]			
	Sol/Meso	436.7	50.56	115.78	
	Meso/Liq	508.2	23.46	46.16	161.94
					[26]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.	
	T (K)	$\Delta H_{\text{pcc}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pcc}}$			
$C_{60}H_{58}O_8$		1,3-phenylene bis[4-(4'-n-hexylbiphenyl-4-carbonyloxy)-2-methylbenzoate]				
	Sol/Meso	416.7	43.32	103.96		
	Meso/Meso	428.2	0.11	0.26	[26]	
$C_{60}H_{58}O_8$	Meso/Liq	463.2	14.54	31.39	135.61	[26]
		1,3-phenylene bis[4-(4'-n-pentylbiphenyl-4-carbonyloxy)-2-ethylbenzoate]				
	Sol/Meso	405.2	40.63	100.27		
$C_{60}H_{66}N_2O_8S$	Meso/Liq	415.7	9.95	23.94	124.21	[26]
		(3,4-dicyano-2,5-thiophenediyl)bis(2,1-ethynediyl-4,1-phenylene) 3,4-bis(hexyloxy)benzoate				
	Sol/Smec	387.1	32.37	83.62		
$C_{60}H_{70}O_8$	Smec/Nem	394.8	7.08	17.93		
	Nem/Liq	417.6	0.96	2.30	103.85	[234]
		4-[[[4'-(octyloxy)[1,1'-biphenyl]-4-yloxy]methyl]benzoic acid, 1-methyl-1,3-propanediyl ester				
$C_{60}H_{72}Br_2O_6$	Sol/Sol	440.0	29.87	67.89		
	Sol/Meso	453.3	7.17	15.82		
	Meso/Smec	469.9	4.41	9.39		
	Smec/Liq	475.4	27.85	58.58	151.7	[304]
$C_{60}H_{72}Cl_2O_6$		4-[(1E)-[3-bromo-4-dodecyloxyphenyl]ethenyl]benzoic acid, 1,3-phenylene ester				
	Sol/Meso	391.5	19.41	49.58		
	Meso/Liq	404.5	16.36	40.44	90.02	[177]
$C_{60}H_{72}Cl_2O_6$		4-[(1E)-[3-chloro-4-dodecyloxyphenyl]ethenyl]benzoic acid, 1,3-phenylene ester				
	Sol/Meso	389.6	18.15	46.59		
	Meso/Liq	417.6	17.47	41.83	88.42	[177]
$C_{60}H_{72}ClNO_8$		3-[[4-{(E)-2-[4-(dodecyloxy)-3-nitrophenyl]-1-ethenyl}benzoyl]oxy]phenyl 4-{(E)-2-[3-chloro-4-(dodecyloxy)phenyl]-1-ethenyl}benzoate				
	Sol/Meso	386.8	12.80	33.09		
	Meso/Liq	420.1	17.0	40.39	73.5	331.1
$C_{60}H_{72}FNO_8$		3-[[4-{(E)-2-[4-(dodecyloxy)-3-nitrophenyl]-1-ethenyl}benzoyl]oxy]phenyl 4-{(E)-2-[3-fluoro-4-(dodecyloxy)phenyl]-1-ethenyl}benzoate				
	Sol/Meso	400.9	24.2	60.38		
	Meso/Liq	430.5	18.68	43.39	103.8	314.9
$C_{60}H_{72}INO_8$		3-[[4-{(E)-2-[4-(dodecyloxy)-3-nitrophenyl]-1-ethenyl}benzoyl]oxy]phenyl 4-{(E)-2-[3-iodo-4-(dodecyloxy)phenyl]-1-ethenyl}benzoate				
	Sol/Meso	386.3	5.73	14.83		
	Meso/Liq	408.0	14.96	36.66	51.5	317.7
$C_{60}H_{72}N_2O_{10}$		4-[(1E)-[3-nitro-4-dodecyloxyphenyl]ethenyl]benzoic acid, 1,3-phenylene ester				
	Sol/Meso	415.0	27.47	66.19		
	Meso/Meso	423.0	1.30	3.07		
$C_{60}H_{72}O_{12}$	Meso/Liq	423.8	14.72	34.73	103.99	[177]
		(R)-4-[4-(1-methylheptyloxy carbonyl)phenyloxycarbonyl]phenyl 4-{5-[4-(4-decyloxybenzoyloxy)-benzoyloxy]heptyloxy}benzoate				
	Sol/Smec	416.7	64.09	153.80		
$C_{60}H_{72}O_{12}$	Smec/Liq	432.2	15.45	35.75	189.55	[119]
		(R)-4-[4-(1-methylheptyloxy carbonyl)phenyloxycarbonyl]phenyl 4-{5-[4-(4-decyloxybenzoyloxy)-benzoyloxy]pentyl}benzoate				
	Sol/Smec	393.2	51.98	132.20		
$C_{60}H_{74}N_2O_9$	Smec/Smec	436.2	0.10	0.23		
	Smec/Liq	456.2	15.78	34.59	167.02	[119]
		4-[4-[4-(4-undecyloxybenzoyl)oxy]benzoyl]-1-piperazinyl]phenyl 4-[4-(4-undecyloxybenzoyl)oxy]benzoate				
$C_{60}H_{74}N_2O_9$	Sol/Smec	434.2	17.9	41.23		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula Transition	Compound				
	T (K)	$\Delta H_{\text{pcc}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pcc}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (estimated)
Smec/Liq	467.2	18.2	38.96	80.19	[289]
$C_{60}H_{74}O_6$		4-[1E]-[4-dodecyloxyphenyl]ethenyl]benzoic acid, 1,3-phenylene ester			
	Sol/Meso	456.1	10.07	22.08	
	Meso/Liq	473.2		Decomposed prior to transition	[177]
$C_{60}H_{74}O_8S_2$		1,3-phenylene 4-[[4-(tridecyloxy)benzoyl]thio]benzoate			
	Sol/Meso	382.2	43.6	114.08	
	Meso/Liq	389.2	19.8	50.87	NA [318]
$C_{60}H_{82}N_2O_4$		N,N'-dioctadecyl-3,4,9,10-perylenebis(carboxamide)			
	Sol/Sol	384.2	16.4	42.69	
	Sol/Sol	414.2	18.4	44.42	
	Sol/Meso	453.2	15.5	34.20	
	Meso/Meso	489.2	0.9	1.84	
$C_{60}H_{84}N_2O_6$	Meso/Liq	585.2	30.7	52.46	175.61
		decanedioic acid, bis[4-[[4-(dodecyloxy)phenyl]imino]methyl]phenyl] ester			
	Sol/Meso	408.7	43.49	106.41	
	Meso/Meso	443.2	15.05	33.96	
$C_{60}H_{84}N_2O_6$	Meso/Liq	454.2	27.12	59.71	200.08 [295]
		decanedioic acid, bis[4-[4-(dodecyloxy)benzylideneamino]phenyl] ester			
	Sol/Meso	397.2	45.98	115.76	
$C_{60}H_{84}N_2O_6$	Meso/Meso	441.5	10.99	24.89	
	Meso/Liq	445.9	28.70	64.36	205.01 [295]
$C_{60}H_{84}N_6O_6$		hexakis(hexyloxy)diquinoxalino[2,3-a:2',3'-c]phenazine			
	Sol/Sol	398.1	15.51	38.96	
	Sol/Meso	460.3	1.32	2.87	
$C_{60}H_{87}N_9O_3$	Meso/Liq	503.5	0.52	1.03	42.86 [209]
		2,4-bis[4-hexadecyloxy-4'-aminoazobenzene]-6-methoxy-1,3,5-triazine			
	Sol/Smec	419.0	79.49	189.71	
$C_{60}H_{90}O_2$	Smec/Liq	471.1	11.30	23.99	213.70 [187]
		4,4'''-dioctadecyloxy-p-quaterphenyl			
	Sol/Sol	404.0	29.90	74.01	
$C_{60}H_{90}O_2$	Sol/Sol	464.0	4.18	9.01	
	Sol/Smec	525.0	17.33	33.01	
	Smec/Liq	542.0	14.09	26.00	142.03 344.4 [111]
$C_{60}H_{96}O_6$		2,6,10-tris(pentyloxy)-3,7,11-tris(nonyloxy)triphenylene			
	Sol/Col	317.2	61.27	193.16	
	Col/Liq	332.2	3.15	9.48	202.64 [261]
$C_{60}H_{96}O_6$		Independent values from another reference			
	Sol/Col	Not reported in paper			
	Col/Liq	332.2	3.2	9.63	[368]
$C_{60}H_{96}O_{12}$		2,3,6,7-tetrakis(octyloxy)-1,5-bis[2-[tetrahydro-2H-pyran-2-yl]oxy]ethoxy]-9,10-anthracenedione			
	Sol/Col	337.2	34.48	102.25	
	Col/Liq	349.2	9.99	28.61	130.86 [275]
$C_{60}H_{98}O_4$		dicholesteryl adipate			
	Sol/Chol	468.7	29.29	62.49	
	Chol/Liq	498.7	3.64	7.30	69.79 223.0 [155, 310]
$C_{60}H_{100}O_8$		1,5-diheptyloxy-2,3,6,7-tetraoctyloxy-9,10-anthraquinone			
	Sol/Meso	264.6	13.1	49.51	
	Meso/Col	318.0	2.8	8.81	
	Col/Liq	366.6	11.8	32.19	90.51 [395]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula Transition	Compound					Ref.
	T (K)	$\Delta H_{\text{pcc}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pcc}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (estimated)	
$C_{60}H_{102}O_{12}$		benzene hexa-n-nonanoate				
	Sol/Sol	248.3	19.4	78.13		
	Sol/Sol	278.3	4.9	17.60		
	Sol/Liq	353.1	69.2	195.98	291.71	405 [129]
Note: Calorimeter could only be used in the heating mode; authors did state that liquid crystalline phases were formed upon cooling.						
$C_{60}H_{104}N_2O_{10}$		N,N'-dinonanoyl-2,3,5,6-tetrakis(nonanoyloxy)-1,4-benzenediamine				
	Sol/Disc	350.2	50.0	142.78		
	Disc/Liq	473.2	36.0	76.08	218.86	[188]
$C_{61}H_{70}N_2O_9$		1,3-{4-[4-(4-nonyloxybenzoyloxy)benzylidene]aminophenoxy}propan-2-ol				
	Sol/Smec	442.0	41.76	94.48		
	Smec/Nem	508.9	1.18	2.32		
	Nem/Liq	540.6	3.98	7.36	104.16	[413]
$C_{61}H_{72}O_8$		4-[[[4'-(octyloxy)[1,1'-biphenyl]-4-yl]oxy]methyl]benzoic acid, 1,5-pentanediyil ester				
	Sol/Smec	444.1	36.21	81.54		
	Smec/Smec	449.5	3.36	7.48		
	Smec/Liq	471.8	24.64	52.27	141.2	[304]
$C_{61}H_{72}O_8$		4-[[[4'-(octyloxy)[1,1'-biphenyl]-4-yl]oxy]methyl]benzoic acid, 1,3-dimethyl-1,3-propanediyl ester				
	Sol/Smec	446.6	33.13	74.18		
	Smec/Smec	453.4	4.20	9.26		
	Smec/Liq	468.7	33.04	70.49	153.9	[304]
$C_{61}H_{72}O_8$		4-[[[4'-(octyloxy)[1,1'-biphenyl]-4-yl]oxy]methyl]benzoic acid, 1-methyl-1,4-butanediyl ester				
	Sol/Smec	443.3	20.53	46.31		
	Smec/Smec	462.1	0.28	0.61		
	Smec/Liq	470.3	23.52	50.01	96.93	[304]
$C_{61}H_{74}ClFO_6$		3-[[4-(E)-2-[4-(dodecyloxy)-3-chlorophenyl]-1-ethenyl]benzoyl]oxy]-2-methylphenyl 4-{{(E)-2-[3-fluoro-4-(dodecyloxy)phenyl]-1-ethenyl}benzoate				
	Sol/Meso	408.3	0.96	2.3		
	Meso/Liq	415.2	16.83	40.5	42.9	
					314.0	[253]
$C_{61}H_{74}ClNO_8$		3-[[4-(E)-2-[4-(dodecyloxy)-3-nitrophenyl]-1-ethenyl]benzoyl]oxy]-2-methylphenyl 4-{{(E)-2-[3-chloro-4-(dodecyloxy)phenyl]-1-ethenyl}benzoate				
	Sol/Meso	394.9	15.44	39.1		
	Meso/Meso	401.0	0.89	2.2		
	Meso/Liq	410.7	11.02	26.8	68.1	
$C_{61}H_{74}INO_8$		3-[[4-(E)-2-[4-(dodecyloxy)-3-nitrophenyl]-1-ethenyl]benzoyl]oxy]-2-methylphenyl 4-{{(E)-2-[3-iodo-4-(dodecyloxy)phenyl]-1-ethenyl}benzoate				
	Sol/Meso	362.4	34.07	94.02		
	Meso/Meso	386.1	1.07	2.78		
	Meso/Liq	398.8	9.24	23.20	120	
					318.3	[253]
$C_{61}H_{74}O_{12}$		(R)-4-[4-(1-methylheptyloxy carbonyl)phenyloxycarbonyl]phenyl 4-{{5-[4-(4-decyloxybenzoyloxy)-benzoyloxy]octyloxy}benzoate				
	Sol/Smec	376.2	42.54	113.08		
	Smec/Liq	397.2	15.52	39.07	152.15	
						[119]
$C_{61}H_{74}O_{12}$		(R)-4-[4-(1-methylheptyloxy carbonyl)phenyloxycarbonyl]phenyl 4-{{5-[4-(4-tridecyloxybenzoyloxy)-benzoyloxy]pentyloxy}benzoate				
	Sol/Smec	390.2	55.80	143.00		
	Smec/Smec	437.2	0.20	0.05		
	Smec/Smec	454.2	15.60	34.35	177.40	
$C_{61}H_{75}ClINO_6$		3-[[4-(E)-2-[4-(dodecyloxy)-3-iodophenyl]-1-ethenyl]benzoyl]oxy]-5-methylpyridyl-2-methyl-4-{{(E)-2-[3-chloro-4-(dodecyloxy)phenyl]-1-ethenyl}benzoate				

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
	T (K)	$\Delta H_{\text{pcc}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pcc}}$		
C <sub>61</sub> H <sub>75</sub> ClN <sub>2</sub> O <sub>8</sub> Sol/Meso	391.8	8.0	20.42	106.7	[253]
Meso/Liq	405.9	35.01	86.25		
C <sub>61</sub> H <sub>82</sub> F <sub>3</sub> N <sub>7</sub> O <sub>11</sub> Sol/Meso	391.2	2.00	5.10	62.9	[253]
Meso/Meso	411.6	2.30	5.58		
Meso/Liq	416.9	21.77	52.21		
C <sub>61</sub> H <sub>86</sub> N <sub>2</sub> O <sub>8</sub> Sol/Meso	406.2	42.0	103.40	144.94	[412]
Meso/Liq	409.2	17.0	41.54		
409.2	17.0	41.54			
C <sub>61</sub> H <sub>106</sub> N <sub>4</sub> O <sub>7</sub> Sol/Meso	379.2	65.3	172.20	184.72	[13]
Meso/Liq	407.2	5.1	12.52		
407.2	5.1	12.52			
C <sub>62</sub> H <sub>58</sub> O <sub>7</sub> Sol/Disc	405.2	37.0	91.31	91.70	[414]
Disc/Liq	519.2	0.2	0.39		
519.2	0.2	0.39			
C <sub>62</sub> H <sub>60</sub> Cl <sub>2</sub> O <sub>8</sub> Sol/Meso	404.7	35.04	86.58	124.74	[276]
Meso/Liq	438.7	16.74	38.16		
438.7	16.74	38.16			
C <sub>62</sub> H <sub>60</sub> F <sub>2</sub> O <sub>8</sub> Sol/Meso	397.7	29.78	74.88	266.8	[26]
Meso/Liq	478.2	22.57	47.20		
478.2	22.57	47.20			
C <sub>62</sub> H <sub>60</sub> F <sub>2</sub> O <sub>8</sub> Sol/Meso	429.7	56.74	132.05	266.8	[26]
Meso/Liq	482.2	22.12	45.87		
482.2	22.12	45.87			
C <sub>62</sub> H <sub>62</sub> O <sub>8</sub> Sol/Meso	426.7	36.14	84.70	296.6	[26]
Meso/Liq	499.2	23.81	47.70		
499.2	23.81	47.70			
C <sub>62</sub> H <sub>62</sub> O <sub>8</sub> Sol/Meso	391.7	30.91	78.91	283.6	[26]
Meso/Meso	448.2	0.12	0.27		
Meso/Liq	455.2	15.63	34.34		
C <sub>62</sub> H <sub>62</sub> O <sub>8</sub> Sol/Meso	402.2	32.55	80.93	106.04	[26]
Meso/Liq	406.2	10.20	25.11		
406.2	10.20	25.11			
C <sub>62</sub> H <sub>62</sub> O <sub>8</sub> Sol/Meso	419.2	46.2	110.21	139.68	[276]
Meso/Liq	420.7	12.4	29.47		
420.7	12.4	29.47			
C <sub>62</sub> H <sub>62</sub> O <sub>8</sub> Sol/Sol	446.2	11.3	25.32	149.85	[331]
Sol/Meso	453.2	36.9	81.42		
Meso/Liq	480.2	20.7	43.11		
C <sub>62</sub> H <sub>62</sub> O <sub>10</sub> Sol/Meso	433.2	20.6	47.55	93.22	[331]
Meso/Liq	499.2	22.8	45.67		
499.2	22.8	45.67			

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula Transition	Compound					Ref.	
	T (K)	$\Delta H_{\text{pcc}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pcc}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (estimated)		
$C_{62}H_{74}F_4O_{10}$	Sol/Meso	360.2	69.1	191.84		[247]	
	Meso/Liq	370.7	22.7	61.24	253.08		
$C_{62}H_{74}F_4O_{10}$	Sol/Sol	1,3-phenylene bis[4-(2-fluoro-4-tetradecyloxybenzoyloxy)-2-fluorobenzoate]					
	Sol/Meso	376.2	104.5	277.78		[247]	
	Note: Sol/Sol transition enthalpy is included in Sol/Meso value.						
	Meso/Liq	393.7	24.2	61.47	277.8		
$C_{62}H_{74}O_8$	Sol/Smec	451.3	29.84	66.12		[304]	
	Smec/Liq	460.6	19.13	41.53	107.65		
	4-[[[4'-(octyloxy)[1,1'-biphenyl]-4-yl]oxy]methyl]benzoic acid, 1,4-dimethyl-1,4-butanediyl ester						
$C_{62}H_{76}ClNO_8$	Sol/Smec	381.9	68.83	180.2		[253]	
	Smec/Liq	415.3	12.67	30.5	210.7		
	3-{{[(E)-2-[4-(dodecyloxy)-3-nitrophenyl]-1-ethenyl]benzoyl]oxo}methyl}benzyl 4-{{(E)-2-[3-chloro-4-(dodecyloxy)phenyl]-1-ethenyl}benzoate}						
$C_{62}H_{76}Cl_2O_6$	Sol/Smec	382.6	68.10	178.0		[253]	
	Smec/Liq	392.3	8.69	22.14	200.1		
$C_{62}H_{76}F_2O_{10}$	Sol/Meso	397.2	97.9	246.48		[301]	
	Meso/Liq	401.7	23.3	58.00	304.48		
	Note: Sol/Meso transition enthalpy includes the enthalpy of a Sol/Sol transition.						
$C_{62}H_{76}N_2O_{10}$	Sol/Smec	396.1	53.93	136.1		[253]	
	Smec/Liq	425.9	14.62	34.32	170.5		
	3-{{[(E)-2-[4-(dodecyloxy)-3-nitrophenyl]-1-ethenyl]benzoyl]oxo}methyl}benzyl 4-{{(E)-2-[3-nitro-4-(dodecyloxy)phenyl]-1-ethenyl}benzoate}						
$C_{62}H_{76}O_8S$	Sol/Nem	410.8	69.88	170.11		[234]	
	Nem/Liq	417.3	1.11	2.66	172.77		
$C_{62}H_{76}O_{12}$	Sol/Smec	416.2	70.94	170.45		[119]	
	Smec/Liq	416.2	13.84	33.25	203.70		
	Note: Paper reports that both transitions occur at 416.2 K; transition enthalpies were calculated from published transition entropies.						
$C_{62}H_{76}O_{12}$	Sol/Smec	388.2	60.68	156.31		[119]	
	Smec/Smec	441.2	0.15	0.34			
	Smec/Liq	453.2	16.20	35.75	192.40		
	(R)-4-[4-(1-methylheptyloxycarbonyl)phenyloxycarbonyl]phenyl 4-{5-[4-(4-decyloxybenzoyloxy)-benzoyloxy]nonyloxy}benzoate						
$C_{62}H_{77}NO_9$	Sol/Meso	371.4	19.78	53.25		[253]	
	Meso/Meso	379.4	0.88	2.32			
	Meso/Liq	390.8	10.08	25.80	81.4		
$C_{62}H_{78}Cl_2N_2O_6$	Sol/Nem	372.2	51.0	137.02		[86]	
	Nem/Liq	393.2	1.6	4.07	141.09		
	4,5-dichloro-1,3-phenylene bis[4-(4-tetradecyloxybenzylideneamino)benzoate]						

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.	
	T (K)	$\Delta H_{\text{pcc}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pcc}}$			
$\text{C}_{62}\text{H}_{78}\text{Cl}_2\text{N}_2\text{O}_6$	Sol/Smec	367.2	33.5	91.23	[416]	
	Smec/Liq	390.2	6.1	15.63		
$\text{C}_{62}\text{H}_{78}\text{N}_2\text{O}_9$	Sol/Smec	438.2	16.6	37.88	[289]	
	Smec/Liq	474.2	17.9	37.75		
$\text{C}_{62}\text{H}_{78}\text{O}_8\text{S}_2$	Sol/Meso	384.2	43.3	112.71	[318]	
	Meso/Liq	391.2	20.8	53.17		
$\text{C}_{62}\text{H}_{80}\text{N}_2\text{O}_6$	Sol/Smec	415.2	41	98.75	[86]	
	Smec/Liq	440.2	27	61.34		
$\text{C}_{62}\text{H}_{80}\text{N}_2\text{O}_8$	Sol/Meso	385.2	21.0	54.52	[412]	
	Meso/Liq	448.2	23.0	51.32		
$\text{C}_{62}\text{H}_{84}\text{O}_7$	Sol/Meso	2,4,7-tris(dodecyloxy)phenanthro[3,4-c]phenanthrene-9,12,13,16-tetraone			[420]	
	Meso/Liq	493.0	56.36	114.32		
$\text{C}_{62}\text{H}_{86}\text{O}_4$	1,6-bis(4-(4'-tetradecyloxyphenylethynyl)phenoxy)hexane			[120]		
	Sol/Sol	395.2	56.0	141.70		
	Sol/Sol	410.2	19.0	46.32		
	Sol/Smec	419.2	17.0	40.55		
	Smec/Smec	435.2	9.0	20.68		
$\text{C}_{62}\text{H}_{102}\text{N}_2\text{O}_4$	Smec/Liq	453.2	23.0	50.75	[189]	
	Sol/Disc	374.2	41.0	109.57		
$\text{C}_{62}\text{H}_{104}\text{O}_8$	Disc/Liq	387.2	13.0	33.57	[395]	
	Sol/Meso	279.9	21.2	75.74		
	Meso/Meso	295.4	0.9	3.05		
	Meso/Col	308.4	1.2	3.89		
$\text{C}_{62}\text{H}_{106}\text{N}_2\text{O}_6$	Col/Liq	370.1	13.2	35.67	[178]	
	Sol/Col	2,5-bis(3,4,5-trioctyloxyphenyl)-1,3,4-oxadiazole				
	Col/Liq	343.7	5.97	17.37		
$\text{C}_{63}\text{H}_{59}\text{F}_2\text{NO}_8$	Sol/Meso	415.2	24.6	59.25	[130]	
	Meso/Liq	420.2	20.9	49.74		
$\text{C}_{63}\text{H}_{62}\text{F}_2\text{O}_8$	Sol/Meso	441.2	28.5	64.60	[168]	
	Sol/Meso	481.2	Decomposed			
$\text{C}_{63}\text{H}_{65}\text{NO}_{10}$	Sol/Meso	452.2	25.5	56.39	[331]	
	Meso/Liq	519.2	22.9	44.11		
$\text{C}_{63}\text{H}_{74}\text{N}_2\text{O}_9$	1,3-{4-[4-(4-decyloxybenzoyloxy)benzylidene]aminophenoxy}propan-2-ol			[413]		
	Sol/Smec	442.7	48.43	109.40		
	Smec/Nem	518.0	1.90	3.67		
	Nem/Liq	532.7	5.06	9.50		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula Transition	Compound					Ref.	
	T (K)	$\Delta H_{\text{pcc}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pcc}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (estimated)		
$C_{63}H_{78}O_{12}$		(R)-4-[4-(1-methylheptyloxy carbonyl)phenyloxycarbonyl]phenyl 4-{5-[4-(4-pentadecyloxybenzoyloxy)-benzoyloxy]pentyloxy}benzoate					
	Sol/SmeC	386.2	60.36	156.29		[119]	
	SmeC/SmeC	444.2	0.18	0.41			
$C_{63}H_{79}ClN_2O_8$	SmeC/Liq	453.2	16.24	35.83	192.53		
		3-[[4-{(E)-2-[4-(dodecyloxy)-3-nitrophenyl]-1-ethenyl}benzoyl]oxy]-5-methylpyridyl-2-methyl 4-{(E)-2-[3-chloro-4-(tetradecyloxy)phenyl]-1-ethenyl}benzoate					
	Sol/Meso	389.6	1.85	4.74			
$C_{63}H_{79}N_3O_6$	Meso/Meso	411.8	2.87	6.98		[253]	
	Meso/Liq	418.8	22.17	52.9	64.7		
		4-cyano-1,3-phenylene bis[4-(4-tetradecylphenyliminomethyl)benzoate]					
$C_{63}H_{79}N_3O_6$	Sol/Meso	363.2	19.5	53.69		[252]	
	Meso/SmeC	401.2	1.8	4.49			
	SmeC/SmeC	415.2	Not reported in paper				
$C_{63}H_{79}N_3O_6$	SmeC/Liq	439.2	7.6	17.30			
		2-cyano-1,3-phenylene bis[4-(4-tetradecyloxyphenyliminomethyl)benzoate]					
	Sol/Meso	429.2	43.8	102.05		[246]	
$C_{63}H_{80}O_9$	Meso/Liq	463.2	33.9	73.19	175.24		
		1-{6-[3,6,7,10,11-pentakis(pentyloxy)triphenyl-2-yloxy]hexyloxy}-9,10-anthraquinone					
	Sol/Meso	316.0	22.7	71.84		[7]	
$C_{63}H_{81}N_3O_3$	Meso/Liq	340.0	2.83	8.32	80.16		
		N,N',N"-tri-1-dodecynyl-N,N',N"-triphenyl-1,3,5-benzene-tricarboxamide					
	Sol/Meso	369.2	4.5	12.19		[241]	
$C_{63}H_{88}O_4$	Meso/Liq	498.2	Decomposed				
		1,7-bis(4-(4'-tetradecyloxyphenylethynyl)phenoxy)heptane					
	Sol/SmeC	396.2	90.0	227.16		[120]	
$C_{63}H_{90}N_2O_8$	SmeC/SmeC	400.2	13.0	32.48			
	SmeC/Liq	418.2	18.0	43.04	302.68		
		1,5-[4-(2-hydroxy-4-hexadecyloxybenzylideneamino)benzoyloxy]pentane					
$C_{63}H_{90}N_2O_8$	Sol/Meso	404.2	41.0	101.43		[412]	
	Meso/Liq	408.2	16.0	39.20	140.63		
		1,5-[4-(2-hydroxy-4-tetradecyloxybenzylideneamino)benzoyloxy]-2,2-dimethylpentane					
$C_{63}H_{90}N_2O_8$	Sol/SmeC	382.2	56.0	146.52		[412]	
	SmeC/Liq	408.2	13.0	31.85	178.37		
		1,5-[4-(2-hydroxy-4-tetradecyloxybenzylideneamino)benzoyloxy]-3,3-dimethylpentane					
$C_{63}H_{90}N_2O_8$	Sol/SmeC	374.2	105.0	280.60		[412]	
	SmeC/Liq	422.2	15.0	35.53	316.13		
$C_{63}H_{101}N_3O_6$		4-(phenylmethyl)-3,5-bis[3,4,5-tris(heptyloxy)phenyl]-4H-1,2,4-triazole					
	Sol/Meso	Not reported in paper				[259]	
	Meso/Liq	341.7	2.22	6.50			
$C_{63}H_{116}N_2O_4$		N,N'-dioctadecanoyl-2,5,6-trimethyl-4-octadecanoyloxy-1,3-benzenediamine					
	Sol/Meso	377.2	80	212.09		[193]	
	Meso/Liq	438.2	1.0	2.28	214.37		
$C_{64}H_{64}Cl_2O_8$		1,3-phenylene bis[4-(3-chlorobenzoyloxy)] 4'-nonylbiphenyl-4-carboxylate					
	Sol/Meso	406.7	46.72	114.88		[276]	
	Meso/Liq	436.7	17.21	39.41	154.29		
$C_{64}H_{64}F_2O_8$		1,3-phenylene bis[4-(4'-n-nonylbiphenyl-4-carboxyloxy)-2-fluorobenzoate]					
	Sol/Meso	383.2	28.21	73.62		[276]	

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula Transition	Compound					Ref.
	T (K)	$\Delta H_{\text{pcc}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pcc}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (estimated)	
Meso/Liq	469.7	21.49	47.75	121.37		[26]
$C_{64}H_{64}F_2O_8$		1,3-phenylene bis[4-(4'-n-nonylbiphenyl-4-carbonyloxy)-3-fluorobenzoate]				
	Sol/Meso	429.2	57.54	134.06		
	Meso/Meso	476.7	22.21	46.59	180.65	[26]
$C_{64}H_{65}NO_{12}$		2'-nitro-m-terphenyl-4,4"-diyl bis[4-(4-nonyloxybenzoyloxy)benzoate]				
	Sol/Sol	381.2	4.0	10.49		
	Sol/Meso	422.2	21.2	50.21		
	Meso/Nem	470.2	10.5	22.33		
$C_{64}H_{66}O_8$	Nem/Liq	490.2	0.4	0.82	83.85	[331]
		1,3-phenylene bis[4-(4'-nonylbiphenyl-4-carbonyloxy)benzoate]				
	Sol/Meso	414.2	50.98	123.08		
$C_{64}H_{66}O_8$	Meso/Liq	492.2	24.24	49.25	172.33	[26]
		1,3-phenylene bis[4-(4'-n-octylbiphenyl-4-carbonyloxy)-2-methylbenzoate]				
	Sol/Meso	377.2	27.96	74.13		
$C_{64}H_{66}O_8$	Meso/Liq	445.2	18.67	41.94	116.07	[26]
		1,3-phenylene bis[4-(4'-n-heptylbiphenyl-4-carbonyloxy)-2-ethylbenzoate]				
	Sol/Meso	386.2	30.05	77.81		
$C_{64}H_{66}O_{10}$	Meso/Liq	401.7	11.41	28.40	106.21	[26]
		m-terphenyl-4,4"-diyl bis[4-(4-nonyloxybenzoyloxy)benzoate]				
	Sol/Meso	434.2	27.3	62.87		
$C_{64}H_{74}N_2O_8S$	Meso/Liq	492.2	25.1	51.00	113.87	[331]
		(3,4-dicyano-2,5-thiophenediyl)bis(2,1-ethyndiyl-4,1-phenylene) 3,4-bis(heptyloxy)benzoate				
	Sol/Smec	400.3	46.46	115.51		
$C_{64}H_{80}INO_8$	Smec/Liq	402.2	Shoulder of preceding peak			[234]
		3-[[4-{(E)-2-[4-(dodecyloxy)-3-nitrophenyl]-1-ethenyl}benzoyl]oxy]phenyl 4-{(E)-2-[3-iodo-4-(hexadecyloxy)phenyl]-1-ethenyl}benzoate				
	Sol/Meso	369.7	5.47	14.8		
$C_{64}H_{80}O_{12}$	Meso/Meso	402.9	2.57	6.4		
	Meso/Liq	409.8	8.38	20.4	41.6	346.1
		(R)-4-[4-(1-methylheptyloxy carbonyl)phenyloxycarbonyl]phenyl 4-{5-[4-(4-hexadecyloxybenzoyloxy)-benzoyloxy]pentyloxy}benzoate				
$C_{64}H_{80}O_{12}$	Sol/Smec	386.2	60.36	156.29		
	Smec/Smec	444.2	0.21	0.47		
	Smec/Liq	453.2	16.58	36.58	193.34	[119]
$C_{64}H_{82}O_8S_2$		1,3-phenylene 4-[[4-(pentadecyloxy)benzoyl]thio]benzoate				
	Sol/Meso	386.2	38.3	99.17		
	Meso/Liq	391.2	16.7	42.69	141.86	NA
$C_{64}H_{90}O_4$		1,8-bis(4-(4'-tetradecyloxyphenylethynyl)phenoxy)octane				
	Sol/Sol	398.2	37	92.92		
	Sol/Smec	408.2	66	161.69		
	Smec/Smec	429.2	13	30.29		
$C_{64}H_{92}N_2O_6$	Smec/Liq	437.2	28	64.04	349.94	342.6
		hexanedioic acid, bis[4-[[4-(hexadecyloxy)phenyl]imino]methyl]phenyl] ester				
	Sol/Meso	413.4	52.59	127.21		
$C_{64}H_{92}N_2O_6$	Meso/Meso	435.9	11.0	25.24		
	Meso/Liq	485.9	27.15	55.88	208.33	[295]
		hexanedioic acid, bis[4-[4-(hexadecyloxy)benzylideneamino]phenyl] ester				
$C_{64}H_{92}N_2O_6$	Sol/Meso	403.5	59.45	147.34		
	Meso/Meso	431.2	8.41	19.50		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula Transition	Compound					Ref.
	T (K)	$\Delta H_{\text{pcc}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pcc}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (estimated)	
Meso/Liq	476.2	28.55	59.95	226.79		[295]
$C_{64}H_{104}O_{12}$		2,3,6,7-tetrakis(nonyloxy)-1,5-bis[2-[(tetrahydro-2H-pyran-2-yl)oxy]ethoxy]-9,10-anthracenedione				
	Sol/Col	331.7	57.72	174.01		[275]
	Col/Liq	343.2	8.86	25.82	199.83	
$C_{64}H_{106}O_4$		dicholesteryl sebacate				
	Sol/Liq	452.25	44.35	98.07		[155, 310]
	Liq/Chol	448.95	3.68	8.20	106.27	
$C_{64}H_{108}O_8$		1,5-dinonyloxy-2,3,6,7-tetraoctyloxy-9,10-anthraquinone				
	Sol/Col	327.1	113.8	347.91		[395]
	Col/Liq	369.8	11.4	30.83	378.74	
$C_{65}H_{63}F_2NO_8$		2-cyano-1,3-phenylene bis[4-(4-nonylbiphenyl-4'-carbonyloxy)-3-fluorobenzoate]				
	Sol/Meso	413.7	45.6	110.22		[130]
	Meso/Liq	419.2	21.8	52.00	162.22	
$C_{65}H_{63}F_2NO_8$		2-cyano-1,3-phenylene bis[4-(4-nonylbiphenyl-4'-carbonyloxy)-2-fluorobenzoate]				
	Sol/Meso	409.7	30.2	73.71		[130]
	Meso/Liq	410.7	23.4	56.98	130.69	
$C_{65}H_{66}F_2O_8$		4'-nonyl-[1,1'-biphenyl]-4-carboxylic acid, (2-methyl-1,3-phenylene)bis[oxycarbonyl (3-fluoro-4,1-phenylene)] ester				
	Sol/Meso	423.2	29.1	68.76		[168]
	Meso/Liq	470.2	Decomposed			
$C_{65}H_{82}ClNO_8$		3-[4-{(E)-2-[4-(tetradecyloxy)-3-nitrophenyl]-1-ethenyl}benzoyl]oxy]-2-methylphenyl 4-{(E)-2-[3-chloro-4-(tetradecyloxy)phenyl]-1-ethenyl}benzoate				
	Sol/Meso	389.7	22.97	58.95		[253]
	Meso/Meso	400.9	1.04	2.59		
	Meso/Liq	414.5	9.46	22.82	84.4	343.5
$C_{65}H_{92}O_4$		1,9-bis(4-(4'-tetradecyloxyphenylethynyl)phenoxy)nonane				
	Sol/Smec	394.2	98	248.60		[120]
	Smec/Smec	404.2	17	42.06		
$C_{65}H_{94}N_2O_6$		heptanedioic acid, bis[4-[[4-(hexadecyloxy)phenyl]imino]methyl]phenyl] ester				
	Sol/Meso	422.7	20.42	48.31		[295]
	Meso/Smec	429.4	51.66	120.31		
$C_{66}H_{68}Cl_2O_8$		1,3-phenylene bis[4-(3-chlorobenzoyloxy)] 4'-decylbiphenyl-4-carboxylate				
	Sol/Meso	405.2	20.20	49.85		[276]
	Meso/Liq	429.7	16.05	37.35	87.20	
$C_{66}H_{68}F_2O_8$		1,3-phenylene bis[4-(3-fluorobenzoyloxy)] 4'-decylbiphenyl-4-carboxylate				
	Sol/Meso	384.2	29.23	76.08		[276]
	Meso/Liq	464.2	22.90	49.33	125.41	
$C_{66}H_{68}F_2O_8$		1,3-phenylene bis[4-(4'-n-decylbiphenyl-4-carbonyloxy)-2-fluorobenzoate]				
	Sol/Meso	384.2	29.23	76.08		[26]
	Meso/Liq	464.2	22.9	49.33	125.41	
$C_{66}H_{68}F_2O_8$		1,3-phenylene bis[4-(4'-n-decylbiphenyl-4-carbonyloxy)-3-fluorobenzoate]				
	Sol/Meso	414.7	27.96	67.42		[26]
	Meso/Liq	470.2	22.70	48.28	115.70	
$C_{66}H_{70}O_8$		1,3-phenylene bis[4-(4'-n-decylbiphenyl-4-carbonyloxy)benzoate]				
	Sol/Meso	432.2	42.87	99.19		[26]
	Meso/Liq	484.2	23.76	49.07	148.26	

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula Transition	Compound					Ref.
	T (K)	$\Delta H_{\text{pcc}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pcc}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (estimated)	
$C_{66}H_{70}O_8$	Sol/Meso	374.2	24.83	66.35		[26]
	Meso/Liq	440.7	19.87	45.09	111.44	
$C_{66}H_{70}O_8$	Sol/Meso	372.2	27.55	74.02		[26]
	Meso/Liq	391.7	11.29	28.82	102.84	
$C_{66}H_{72}N_2O_6$	Sol/Meso	471.7	31.6	66.99		[299]
	Meso/Liq	541.2	33.5	61.90	128.89	
$C_{66}H_{78}O_{10}$	(S,S)-4,4'-bis{5-[4'-(1-methylheptyloxy carbonyl)-1",1"-biphenyl-4"-y]oxycarbonyl}pentyloxy biphenyl					
	Sol/SmeC	392.2	30.09	76.72		[287]
	SmeC/SmeC	395.6	1.81	4.58		
$C_{66}H_{82}F_4O_{10}$	SmeC/Liq	441.0	24.25	54.99	136.29	
	Sol/Meso	366.7	81.8	223.07		[247]
	Meso/Liq	374.2	24.0	64.14	287.21	
$C_{66}H_{82}F_4O_{10}$	Sol/Sol	Not reported in paper				
	Sol/Meso	378.2	109.6	289.79		
	Note: Sol/Sol transition enthalpy is included in Sol/Meso value.					
	Meso/Liq	395.2	24.6	62.25	352.04	368.2
$C_{66}H_{82}O_8$	1,3-phenylene bis[4-(2-fluoro-4-hexadecyloxybenzoyloxy)-2-fluorobenzoate]					
	Sol/Sol	426.0	2.31	5.42		[304]
	Sol/SmeC	446.5	5.72	12.81		
$C_{66}H_{84}ClNO_8$	SmeC/Liq	465.8	31.91	68.51	86.74	
	4-[[[4'-(octyloxy)[1,1'-biphenyl]-4-yl]oxy]methyl]benzoic acid, 1,10-decanediyl ester					
	Sol/Sol	382.6	65.42	171.0		[253]
$C_{66}H_{84}Cl_2O_6$	Sol/SmeC	418.2	12.75	30.48	201.5	
	SmeC/Liq				352.9	
	Sol/Meso	380.3	17.33	45.57		[177]
$C_{66}H_{84}F_2O_{10}$	Meso/Liq	410.8	16.92	41.19	86.76	
	Sol/Meso	394.2	104.9	266.11		[301]
	Meso/Liq	402.2	24.3	60.42	326.53	
Note: Sol/Meso transition enthalpy includes the enthalpy of a Sol/Sol transition.						
$C_{66}H_{84}N_2O_{10}$	3-[[4-{(E)-2-[4-(tetradecyloxy)-3-nitrophenyl]-1-ethenyl}benzoyl]oxo]methyl]benzyl					
	4-{(E)-2-[3-chloro-4-(dodecyloxy)phenyl]-1-ethenyl}benzoate					[253]
	Sol/SmeC	385.8	24.37	63.16		
$C_{66}H_{84}O_8S$	SmeC/Liq	431.4	15.11	35.02	98.1	354.4
	2,5-thiophenediyl bis(2,1-ethylenediyl-4,1-phenylene) 3,4-bis(octyloxy)benzoate					[234]
	Sol/Nem	390.8	73.14	187.15		
$C_{66}H_{86}Cl_2N_2O_6$	Nem/Liq	406.4	1.22	3.00	190.15	
	Sol/Nem	372.2	57	153.14		[86]
	Nem/Liq	388.2	1.5	3.86	157.00	
$C_{66}H_{86}N_4O_{10}$	1,3-phenylene bis[[4-(methylamino)phenyl]-3-nitro-4-hexadecyloxybenzoate]					
	Sol/Meso	383.7	68.48	178.47		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula Transition	Compound					Ref.
	T (K)	$\Delta H_{\text{pcc}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pcc}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (estimated)	
Meso/Liq	433.7	17.06	39.33	217.80		[321]
$C_{66}H_{86}O_8S_2$	1,3-phenylene	4-[4-(hexadecyloxy)benzoyl]thio]benzoate				
	Sol/Meso	381.2	44.0	115.42		
	Meso/Liq	392.2	16.0	40.80	156.22	NA [318]
$C_{66}H_{88}N_2O_6$	1,3-phenylene	bis[4-(4-hexadecyloxyphenyliminomethyl)benzoate]				
	Sol/Smec	409.2	41	100.20		
	Smec/Liq	436.2	28	64.19	164.39	[86]
$C_{66}H_{88}N_2O_8$	1,3-bis[4-(2-hydroxy-4-hexadecyloxybenzylideneamino)benzoyloxy]benzene					
	Sol/Meso	386.2	76.0	196.79		
	Meso/Liq	445.2	23.0	51.66	248.45	[412]
Note: Sol/Meso transition enthalpy for this compound is significantly larger than values for the other compounds in this series.						
$C_{66}H_{94}O_4$	1,10-bis(4-(4'-tetradecyloxyphenoylethynyl)phenoxy)decane					
	Sol/Sol	399.2	39	97.70		
	Sol/Smec	411.2	72	175.10		
	Smec/Liq	427.2	44	103.00	375.80	356.8 [120]
$C_{66}H_{98}O_6$	2,5,6,9,12,13-hexahexyloxydibenzo[fg,op]naphthacene					
	Sol/Col	416.6	86.2	206.91		
	Col/Liq	452.9	2.8	6.18	213.09	[6]
$C_{66}H_{112}O_8$	1,5-didecyloxy-2,3,6,7-tetraoctyloxy-9,10-anthraquinone					
	Sol/Col	321.7	106.3	330.43		
	Col/Liq	368.5	11.7	31.75	362.18	[395]
$C_{66}H_{114}O_{12}$	benzene hexa-n-decanoate					
	Sol/Sol	330.8	75.7	209.75		
	Sol/Liq	360.9	91.8	254.36	464.11	447.6 [129]
$C_{66}H_{116}N_2O_{10}$	N,N'-didecanoyl-2,3,5,6-tetrakis(decanoxy)-1,4-benzenediamine					
	Sol/Disc	350.2	62.0	177.04		
	Disc/Liq	472.2	33.0	69.89	246.93	429 [188]
$C_{67}H_{67}F_2NO_8$	2-cyano-1,3-phenylene bis[4-(4-decylbiphenyl-4'-carbonyloxy)-3-fluorobenzoate]					
	Sol/Meso	398.2	31.2	78.35		
	Meso/Liq	421.2	25.2	59.83	138.18	[130]
$C_{67}H_{67}F_2NO_8$	2-cyano-1,3-phenylene bis[4-(4-decylbiphenyl-4'-carbonyloxy)-2-fluorobenzoate]					
	Sol/Meso	403.7	33.7	83.48		
	Meso/Liq	414.2	25.0	60.36	143.84	298 [130]
$C_{67}H_{68}O$	hexadecyl pentakis[4-methylphenylethynyl]phenyl ether					
	Sol/Meso	435.5	55.7	127.90		
	Meso/Liq	465.7	0.2	0.43	128.33	[88]
$C_{67}H_{70}F_2O_8$	4'-decyl-[1,1'-biphenyl]-4-carboxylic acid, (2-methyl-1,3-phenylene)bis[oxycarbonyl (3-fluoro-4,1-phenylene)] ester					
	Sol/Meso	421.0	38.4	91.21		
	Note: Sol/Meso enthalpy includes other Sol/Sol transitions.					
$C_{67}H_{86}Cl_2O_6$	Meso/Liq	467.4	Decomposed			[168]
	4-[1(E)-[3-chloro-4-pentadecyloxyphenyl]ethenyl]benzoic acid, 2-methyl-1,3-phenylene ester					
	Sol/Meso	374.9	55.88	149.05		
$C_{67}H_{86}F_2O_8$	Meso/Liq	395.9	13.65	34.48	183.53	[177]
	2-fluoro-4-[4-hexadecylbenzoyloxy]benzoic acid, 2-methyl-1,3-phenylene ester					
	Sol/Meso	377.7	33.6	88.96		
	Meso/Liq	379.2	24.5	64.61	153.57	351.8 [168]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula Transition	Compound				Ref.
	T (K)	$\Delta H_{\text{pcc}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pcc}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	
$C_{67}H_{87}N_3O_6$	Sol/Meso	425.2	52.5	123.47	[246]
	Meso/Liq	461.2	34.7	75.24	
$C_{67}H_{89}N_5O_7S_2$	Sol/Smec	327.2	57.3	175.12	[99]
	Smec/Liq	376.2	17.4	46.25	
$C_{67}H_{98}N_2O_8$	Sol/Smec	385.2	78.0	202.49	[412]
	Smec/Liq	408.2	14.0	34.30	
$C_{67}H_{98}N_2O_8$	Sol/Smec	377.2	125.0	331.39	[412]
	Smec/Liq	421.2	17.0	40.36	
$C_{68}H_{72}Cl_2O_8$	Sol/Meso	401.2	15.91	39.66	[276]
	Meso/Liq	426.2	15.36	36.04	
$C_{68}H_{72}F_2O_8$	Sol/Meso	387.2	31.78	82.08	[276]
	Meso/Liq	467.2	22.96	49.14	
$C_{68}H_{72}F_2O_8$	Sol/Meso	387.2	31.78	82.08	[26]
	Meso/Liq	467.2	22.96	49.14	
$C_{68}H_{72}F_2O_8$	Sol/Meso	415.2	35.86	86.37	[26]
	Meso/Liq	464.2	22.13	47.67	
$C_{68}H_{74}O_8$	Sol/Meso	424.2	41.42	97.64	[26]
	Meso/Liq	482.2	24.45	50.71	
$C_{68}H_{74}O_8$	Sol/Meso	392.7	39.59	100.81	[26]
	Meso/Liq	435.2	19.89	45.70	
$C_{68}H_{74}O_8$	Sol/Meso	375.7	36.70	97.68	[26]
	Meso/Liq	385.7	12.34	31.99	
$C_{68}H_{90}O_9$	Sol/Meso	325.0	14.6	44.92	[7]
	Meso/Liq	334.0	0.70	2.10	
$C_{68}H_{82}N_2O_8S$	Sol/Smec	404.4	39.89	98.64	[234]
	Smec/Liq	408.5	6.92	16.94	
$C_{68}H_{100}N_2O_6$	Sol/Meso	406.0	58.77	144.75	[295]
	Meso/Meso	436.0	14.60	33.49	
	Meso/Liq	450.5	30.55	67.81	
$C_{68}H_{100}N_2O_6$	Sol/Meso	397.0	66.04	166.35	[295]
	Meso/Meso	431.9	10.93	25.31	
	Meso/Liq	443.2	32.62	73.60	

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula Transition	Compound				Ref.
	T (K)	$\Delta H_{\text{pcc}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pcc}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	
$C_{68}H_{106}O_6$ Sol/Nem Nem/Liq	cholest-5-en-3-ol ( $3\beta$ ), 4,4'-[1,4-phenylenebis(oxy)]bis[butanoate]	412.2 452.2	16.3 3.3	39.54 7.30	[152]
	452.2	3.3	7.30	46.84	
$C_{68}H_{112}O_{12}$ Sol/Col Col/Liq	2,3,6,7-tetrakis(decyloxy)-1,5-bis[2-[(tetrahydro-2H-pyran-2-yl)oxy]ethoxy]-9,10-anthracenedione	325.2 335.7	56.61 7.84	174.08 23.35	[275]
	335.7	7.84	23.35	197.43	
$C_{68}H_{116}O_8$ Sol/Col Col/Liq	1,5-diundecyloxy-2,3,6,7-tetraoctyloxy-9,10-anthaquinone	295.4 365.1	29.9 12.5	101.22 34.24	[395]
	365.1	12.5	34.24	135.46	
$C_{69}H_{71}F_2NO_8$ Sol/Meso Meso/Liq	2-cyano-1,3-phenylene bis[4-(4-undecylbiphenyl-4'-carbonyloxy)-3-fluorobenzoate]	401.2 424.2	28.1 25.7	70.04 60.58	[130]
	424.2	25.7	60.58	130.62	
$C_{69}H_{71}F_2NO_8$ Sol/Meso Meso/Liq	2-cyano-1,3-phenylene bis[4-(4-undecylbiphenyl-4'-carbonyloxy)-2-fluorobenzoate]	411.2 417.7	38.5 26.6	93.63 63.68	[130]
	417.7	26.6	63.68	157.31	
$C_{69}H_{74}F_2O_8$ Sol/Meso Meso/Liq	4'-undecyl-[1,1'-biphenyl]-4-carboxylic acid, (2-methyl-1,3-phenylene)bis[oxy carbonyl (3-fluoro-4,1-phenylene)] ester	420.8 465.2	43.5 Decomposed	103.37	[168]
	465.2	Decomposed			
$C_{69}H_{93}N_3O_3$ Sol/Meso Meso/Liq	N,N',N"-tridodecyl-2,4,6-tris(phenylethynyl)-1,3,5-benzenetricarboxamide	414.2 498.2	9.0 Decomposed	21.73	[241]
	498.2	Decomposed			
$C_{69}H_{93}N_5O_2S_2$ Sol/SmeC SmeC/Liq	1,2-bis-5-[4-(5-undecyl-1,3,4-thiadiazol-2-yl)phenoxy]pentanoyloxy]-6-[4'-cyanobiphenyl-4-yloxy]octane	337.2 383.2	57.2 20.4	163.63 53.24	[99]
	383.2	20.4	53.24	216.87	
$C_{69}H_{102}N_2O_6$ Sol/Sol Sol/SmeC SmeC/Liq	malonic acid, bis{10-[4-[(4-decylphenylimino)methyl]phenoxy]decyl} ester	338.8 361.6 365.0	38.8 13.5 13.3	114.52 37.33 36.44	[244]
	361.6	13.5	37.33		
	365.0	13.3	36.44	188.29	
$C_{69}H_{102}N_2O_8$ Sol/SmeC SmeC/Liq	malonic acid, bis{10-[4-[(4-decyloxyphenylimino)methyl]phenoxy]decyl} ester	381.3 389.7	94.8 13.1	248.62 33.62	[244]
	389.7	13.1	33.62	282.24	
$C_{69}H_{113}N_3O_6$ Sol/Meso Meso/Liq	4-(phenylmethyl)-3,5-bis[3,4,5-tris(octyloxy)phenyl]-4H-1,2,4-triazole	Not reported in paper			[259]
	342.3	4.18	12.21		
$C_{70}H_{76}Cl_2O_8$ Sol/Meso Meso/Liq	1,3-phenylene bis[4-(3-chlorobenzoyloxy)] 4'-dodecylbiphenyl-4-carboxylate	404.2 421.2	15.15 14.34	37.48 34.05	[276]
	421.2	14.34	34.05	71.53	
$C_{70}H_{76}F_2O_8$ Sol/Meso Meso/Liq	1,3-phenylene bis[4-(3-fluorobenzoyloxy)] 4'-dodecylbiphenyl-4-carboxylate	380.2 467.7	50.36 22.46	132.46 48.02	[276]
	467.7	22.46	48.02	180.48	
$C_{70}H_{76}F_2O_8$ Sol/Meso Meso/Liq	1,3-phenylene bis[4-(4'-n-dodecylbiphenyl-4-carbonyloxy)-2-fluorobenzoate]	380.2 467.7	50.36 22.46	132.46 48.02	[26]
	467.7	22.46	48.02	180.48	
$C_{70}H_{76}F_2O_8$ Sol/Meso Meso/Liq	1,3-phenylene bis[4-(4'-n-dodecylbiphenyl-4-carbonyloxy)-3-fluorobenzoate]	419.2 458.7	40.04 22.07	95.52 48.11	[26]
	458.7	22.07	48.11	143.63	

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
	T (K)	$\Delta H_{\text{pcc}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pcc}}$		
$C_{70}H_{78}O_8$	Sol/Meso	394.2	209.0	530.19	[26]
	Meso/Liq	531.4	26.98	50.77	
$C_{70}H_{78}O_8$	Sol/Meso	385.7	40.53	105.08	[26]
	Meso/Liq	430.2	21.58	50.16	
$C_{70}H_{78}O_8$	Sol/Meso	377.5	36.03	95.44	[26]
	Meso/Liq	379.2	12.75	33.62	
$C_{70}H_{80}N_2O_6$	Sol/Meso	403.2	15.1	37.45	[299]
	Meso/Meso	469.7	41.8	88.99	
	Meso/Liq	535.7	27.2	50.77	
$C_{70}H_{86}O_{10}$	Sol/Smec	374.7	26.83	71.60	[287]
	Smec/Smec	399.7	4.27	10.68	
	Smec/Liq	414.6	26.37	63.60	
$C_{70}H_{86}O_{10}$	Sol/Smec	374.2	25.92	69.27	[287]
	Smec/Smec	399.7	3.96	9.91	
	Smec/Liq	414.4	25.47	61.51	
$C_{70}H_{90}F_4O_{10}$	Sol/Meso	370.2	93.8	253.38	[247]
	Meso/Liq	375.7	26.1	69.47	
				322.85	
$C_{70}H_{90}F_4O_{10}$	Sol/Sol	Not reported in paper			[247]
	Sol/Meso	380.2	131.7	346.40	
	Meso/Liq	395.2	25.2	63.77	
$C_{70}H_{92}F_2O_{10}$	Sol/Meso	391.2	125.7	321.32	[301]
	Meso/Liq	401.7	24.5	60.99	
		Note: Sol/Meso transition enthalpy includes the enthalpy from a Sol/Sol transition.			
$C_{70}H_{92}O_8S$	Sol/Smec	385.2	62.11	161.24	[234]
	Smec/Nem	388.0	6.92	17.84	
	Nem/Liq	399.1	1.09	2.73	
$C_{70}H_{94}N_2O_9$	Sol/Smec	415.2	17.0	40.94	[289]
	Smec/Liq	466.2	18.8	40.33	
		Note: Sol/Meso transition enthalpy includes the enthalpy from a Sol/Sol transition.			
$C_{70}H_{94}N_4O_{10}$	Sol/Meso	376.7	21.72	57.66	[321]
	Meso/Liq	431.9	15.1	34.96	
		Note: Sol/Meso transition enthalpy includes the enthalpy from a Sol/Sol transition.			
$C_{70}H_{110}O_6$	Sol/Nem	403.2	24.9	61.76	[152]
	Nem/Liq	418.2	0.3	0.72	
		Note: Sol/Meso transition enthalpy includes the enthalpy from a Sol/Sol transition.			

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula Transition	Compound				Ref.
	T (K)	$\Delta H_{\text{pcc}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pcc}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	
$C_{70}H_{111}BrN_2O_6$	Sol/Sol	270.2	10.09	37.34	[439]
	Sol/Col	282.6	1.34	4.74	
	Col/Liq	417.6	7.60	18.20	
$C_{70}H_{111}ClN_2O_6$	Sol/Sol	274.9	17.44	63.44	[439]
	Sol/Col	287.1	7.61	26.51	
	Col/Liq	414.1	7.28	17.58	
$C_{70}H_{111}FN_2O_6$	Sol/Sol	283.2	24.96	88.14	[439]
	Sol/Col	294.6	1.43	4.85	
	Col/Liq	401.4	6.21	15.47	
$C_{70}H_{111}IN_2O_6$	Sol/Col	280.5	17.70	63.10	[439]
	Col/Liq	411.2	7.09	17.24	
	Sol/Disc	376.2	60.0	159.49	
$C_{70}H_{118}N_2O_6$	Disc/Liq	387.2	16.0	41.32	[189]
	Sol/Meso	404.2	44.6	110.34	
$C_{71}H_{71}F_2NO_8$	Meso/Liq	425.7	26.6	62.49	[28, 130]
	Sol/Meso	407.7	36.7	90.02	
$C_{71}H_{75}F_2NO_8$	Meso/Liq	419.2	26.7	63.69	[130]
	Sol/Meso	409.2	43.3	105.82	
$C_{71}H_{78}F_2O_8$	Note: Sol/Meso enthalpy includes other Sol/Sol transitions.				[168]
	Meso/Liq	466.2	Decomposed		
$C_{71}H_{94}F_2O_8$	Sol/Meso	380.2	91.9	241.71	[168]
	Note: Sol/Meso enthalpy includes other Sol/Sol transitions.				
	Meso/Liq	381.2	26.1	68.47	
$C_{71}H_{95}N_3O_6$	Sol/Meso	419.7	58.5	139.39	[246]
	Meso/Liq	458.7	35.3	76.96	
$C_{71}H_{96}O_{10}S_3$	Sol/Meso	446.3	18.69	41.88	[420]
	Not reported in paper				
	Meso/Liq	422.8	9.8	23.18	
$C_{71}H_{96}O_{10}S_3$	Sol/Meso	434.4	12.81	29.49	[420]
	Not reported in paper				
	Meso/Liq	422.8	9.8	23.18	
$C_{71}H_{112}F_3N_3O_8S$					
	4-(4-trifluoromethanesulphonylaminophenyl)-2,6-bis(3,4,5-trioctyloxyphenyl)pyrimidine				

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.	
	T (K)	$\Delta H_{\text{pcc}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pcc}}$			
Sol/Col	303.2	8.07	26.62	41.56	[439]	
Col/Liq	421.6	6.30	14.94			
C <sub>71</sub> H <sub>114</sub> N <sub>2</sub> O <sub>6</sub>	4-(4-methylphenyl)-2,6-bis(3,4,5-trioctyloxyphenyl)pyrimidine					
Sol/Col	270.3	15.38	56.90	71.71	[439]	
Col/Liq	378.1	5.60	14.81			
C <sub>71</sub> H <sub>114</sub> N <sub>2</sub> O <sub>7</sub>	4-(4-methoxyphenyl)-2,6-bis(3,4,5-trioctyloxyphenyl)pyrimidine					
Sol/Col	309.6	49.92	161.24	174.58	[439]	
Col/Liq	377.7	5.04	13.34			
C <sub>72</sub> H <sub>82</sub> O <sub>8</sub>	1,3-phenylene bis[4-(4'-n-dodecylbiphenyl-4-carbonyloxy)-2-methylbenzoate]					
Sol/Meso	388.2	61.96	159.61	213.62	[26]	
Note: A Sol/Sol transition enthalpy is included in Sol/Meso value.	Meso/Liq	427.7	23.10	54.01		
C <sub>72</sub> H <sub>90</sub> N <sub>2</sub> O <sub>8</sub> S	(3,4-dicyano-2,5-thiophenediyi)bis(2,1-ethylenediyi-4,1-phenylene) 3,4-bis(nonyloxy)benzoate					
Sol/Smec	392.5	35.27	89.86	103.47	[234]	
Smec/Liq	404.1	5.50	13.61			
C <sub>72</sub> H <sub>94</sub> N <sub>4</sub> O <sub>10</sub>	bis[(S)-2-methylbutyl] 3,3'-[m-phenylene-bis(iminocarbonyloxyundecamethyleneoxy-p-phenylenemethylidynenitrilo-p-phenylene)] bis[2-propenoate]				[206]	
Sol/Smec	357.2	51.8	145.02	158.24		
Smec/Liq	378.2	5.0	13.22			
Note: Transition enthalpies were determined from a cooling curve.						
C <sub>72</sub> H <sub>99</sub> N <sub>3</sub> O <sub>3</sub>	2,4,6-tri-1-dodecynyl-N,N',N''-tri[(2S)-2-phenylpropyl]-1,3,5-benzenetricarboxamide				[241]	
Sol/Meso	383.2	69.0	180.06	211.71		
Meso/Liq	480.2	15.2	31.65			
C <sub>72</sub> H <sub>114</sub> O <sub>6</sub>	cholest-5-en-3-ol (3 $\beta$ ), 4,4'-[1,4-phenylenebis(oxy)]bis[hexanoate]					
Sol/Nem	418.2	23.1	55.24	62.30	[152]	
Nem/Liq	453.2	3.2	7.06			
C <sub>72</sub> H <sub>116</sub> N <sub>2</sub> O <sub>6</sub>	4-(4-ethylphenyl)-2,6-bis(3,4,5-trioctyloxyphenyl)pyrimidine					
Sol/Col	285.7	32.37	113.30	128.54	[439]	
Col/Liq	374.7	5.71	15.24			
C <sub>72</sub> H <sub>128</sub> N <sub>2</sub> O <sub>10</sub>	N,N'-diundecanoyl-2,3,5,6-tetrakis(undecanoyloxy)-1,4-benzenediamine					
Sol/Disc	354.2	65.0	183.51	465.8	[188]	
Disc/Liq	471.2	31.0	65.79	249.30		
C <sub>73</sub> H <sub>96</sub> N <sub>4</sub> O <sub>10</sub>	bis[(S)-2-methylbutyl] 3,3'-[4-methyl-1,3-phenylene-bis(iminocarbonyloxyundecamethyleneoxy-p-phenylenemethylidynenitrilo-p-phenylene)] bis[2-propenoate]				[206]	
Sol/Smec	327.2	39.2	119.80	13.16		
Smec/Smec	338.2	Not reported in paper				
Smec/Liq	372.2	4.9	13.16	Note: Transition enthalpies were determined from a cooling curve.		
C <sub>74</sub> H <sub>84</sub> F <sub>2</sub> O <sub>8</sub>	1,3-phenylene bis[4-(4'-n-tetradecylbiphenyl-4-carbonyloxy)-2-fluorobenzoate]					
Sol/Meso	376.2	72.55	192.85	251.82	[26]	
Meso/Liq	469.2	27.67	58.97			
C <sub>74</sub> H <sub>84</sub> F <sub>2</sub> O <sub>8</sub>	1,3-phenylene bis[4-(4'-n-tetradecylbiphenyl-4-carbonyloxy)-3-fluorobenzoate]					
Sol/Meso	415.2	47.36	114.07	166.15	[26]	
Meso/Liq	456.2	23.76	52.08			
C <sub>74</sub> H <sub>86</sub> O <sub>8</sub>	1,3-phenylene bis[4-(4'-n-tetradecylbiphenyl-4-carbonyloxy)benzoate]					
Sol/Meso	391.2	45.05	115.16	181.25	[26]	
Meso/Liq	481.7	27.67	66.09			
C <sub>74</sub> H <sub>98</sub> F <sub>4</sub> O <sub>10</sub>	1,3-phenylene bis[4-(3-fluoro-4-eicosyloxybenzyloxy)-2-fluorobenzoate]					

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula Transition	Compound					Ref.
	T (K)	$\Delta H_{\text{pcce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pcce}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	
Sol/Sol	Not reported in paper					
Sol/Meso	381.2	120.9	317.16			
Note: Sol/Sol transition enthalpy is included in Sol/Meso value.						
Meso/Liq	394.2	26.0	65.96	383.12	425.0	[247]
C <sub>74</sub> H <sub>100</sub> O <sub>8</sub> S		2,5-thiophenediylbis(2,1-ethynediyl-4,1-phenylene) 3,4-bis(dodecyloxy)benzoate				
Sol/Nem	390.1	87.66	224.71			
Nem/Liq	393.7	1.26	3.20	227.91		[234]
C <sub>74</sub> H <sub>100</sub> I <sub>2</sub> O <sub>6</sub>		3-[{[4-(E)-2-[4-(hexadecyloxy)-3-iodophenyl]-1-ethenyl]benzoyl}oxy]methyl]benzyl 4-[{(E)-[2-iodo-4-(hexadecyloxy)-phenyl]-1-ethenyl}benzoate				
Sol/SmeC	338.7	87.49	258.3			
SmeC/Liq	356.4	5.35	15.0	273.3	386.2	[253]
C <sub>74</sub> H <sub>108</sub> O <sub>8</sub>		2,4,5,7-tetrakis(dodecyloxy)phenanthro[3,4-c]phenanthrene-9,12, 13,16-tetraone				
Sol/Meso	Not reported in paper					
Meso/Liq	509.2	45.0	88.37			[420]
C <sub>74</sub> H <sub>118</sub> O <sub>6</sub>		cholest-5-en-3-ol (3 $\beta$ ), 4,4'-[1,4-phenylenebis(oxy)]bis[heptanoate]				
Sol/Nem	406.2	31.4	77.30			
Nem/Liq	421.2	0.4	0.95	78.25		[152]
C <sub>74</sub> H <sub>130</sub> N <sub>20</sub> S		2,5-bis[3,4,5-tris(decyloxyphenyl)]-1,3,4-oxadiazole				
Sol/Col	Below ambient room temperature					
Col/Liq	337.0	5.53	16.41			[178]
C <sub>74</sub> H <sub>130</sub> O <sub>9</sub>		3,4,5-tris(decyloxy)benzoic acid anhydride				
Sol/Col	262.2	19.3	73.61			
Col/Liq	326.8	8.2	25.09	98.70		[366]
C <sub>75</sub> H <sub>79</sub> F <sub>2</sub> NO <sub>8</sub>		2-cyano-1,3-phenylene bis[4-(4'-tetradecylbiphenyl-4-carbonyloxy)-3-fluorobenzoate]				
Sol/Meso	395.7	35.9	90.73			
Meso/Liq	427.2	27.4	64.14	154.87		[28, 130]
C <sub>75</sub> H <sub>83</sub> F <sub>2</sub> NO <sub>8</sub>		2-cyano-1,3-phenylene bis[4-(4-tetradecylbiphenyl-4-carbonyloxy)-2-fluorobenzoate]				
Sol/Meso	404.2	148.0	366.16			
Meso/Liq	413.2	28.5	68.97	435.13		[130]
C <sub>75</sub> H <sub>88</sub> F <sub>2</sub> O <sub>8</sub>		4'-tetradecyl-[1,1'-biphenyl]-4-carboxylic acid, (2-methyl-1,3-phenylene)bis[oxycarbonyl (3-fluoro-4,1-phenylene)] ester				
Sol/Meso	409.0	48.6	118.83			
Note: Sol/Meso enthalpy includes other Sol/Sol transitions.						
Meso/Liq	467.7	Decomposed				[168]
C <sub>76</sub> H <sub>86</sub> N <sub>4</sub> O <sub>8</sub>		5,10,15,20-tetrakis(4-hydroxyphenyl)porphyrin				
Sol/Col	363.8	15.27	41.97			
Col/Liq	407.1	12.77	31.37	73.34		[285]
C <sub>76</sub> H <sub>86</sub> O <sub>8</sub>		1,3-phenylene bis[4-(4-n-tetradecylbiphenyl-4-carbonyloxy)-2-methylbenzoate]				
Sol/Meso	376.2	28.63	76.10			
Meso/Liq	429.7	24.18	56.27	132.37		[26]
C <sub>76</sub> H <sub>90</sub> O <sub>8</sub>		1,3-phenylene bis[4-(3-methylbenzoyloxy)] 4'-dodecylbiphenyl-4-carboxylate				
Sol/Meso	387.2	56.38	145.61			
Meso/Liq	401.7	16.27	40.50	186.11		[276]
C <sub>76</sub> H <sub>98</sub> N <sub>2</sub> O <sub>8</sub> S		(3,4-dicyano-2,5-thiophenediyl)bis(2,1-ethynediyl-4,1-phenylene) 3,4-bis(decyloxy)benzoate				
Sol/SmeC	373.7	50.88	136.15			
SmeC/Col	396.6	2.95	7.44			
Col/Liq	404.6	3.95	9.76	153.35		[234]
C <sub>76</sub> H <sub>106</sub> N <sub>2</sub> O <sub>4</sub>		2,5,2',5'-tetrakis(4-decyloxyphenyl)azobenzene				

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
	T (K)	$\Delta H_{\text{pcc}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pcc}}$		
C <sub>76</sub> H <sub>122</sub> O <sub>6</sub>	Sol/Smec	451.2	73.3	162.46	[122]
	Smec/Liq	510.2	18.1	35.48	
C <sub>78</sub> H <sub>92</sub> F <sub>2</sub> O <sub>8</sub>	cholest-5-en-3-ol (3 $\beta$ ), 4,4'-[1,4-phenylenebis(oxy)]bis[octanoate]				
	Sol/Nem	401.2	19.2	47.86	[152]
	Nem/Liq	434.2	3.8	8.75	
C <sub>78</sub> H <sub>92</sub> F <sub>2</sub> O <sub>8</sub>	1,3-phenylene bis[4-(4'-n-hexadecylbiphenyl-4-carbonyloxy)-2-fluorobenzoate]				
	Sol/Meso	373.2	85.09	228.00	[26]
	Meso/Liq	468.7	29.03	61.94	
C <sub>78</sub> H <sub>92</sub> F <sub>2</sub> O <sub>8</sub>	1,3-phenylene bis[4-(4'-n-hexadecylbiphenyl-4-carbonyloxy)-3-fluorobenzoate]				
	Sol/Meso	409.2	49.15	120.11	[26]
	Meso/Liq	455.2	21.27	46.73	
C <sub>78</sub> H <sub>94</sub> O <sub>8</sub>	1,3-phenylene bis[4-(4'-n-hexadecylbiphenyl-4-carbonyloxy)benzoate]				
	Sol/Meso	390.7	43.78	112.06	[26]
	Meso/Liq	479.7	28.72	59.87	
C <sub>78</sub> H <sub>96</sub> N <sub>2</sub> O <sub>6</sub>	1,3-phenylenebis[(E)-methylidynenitrilo-4,1-phenylene] 4'-(hexadecyloxy)[1,1'-biphenyl]-4-carboxylate				
	Sol/Meso	415.7	14.9	35.84	[299]
	Meso/Meso	465.7	46.3	99.42	
	Meso/Liq	524.7	30.2	57.56	192.82
C <sub>78</sub> H <sub>102</sub> O <sub>12</sub>	hexa(4-pentyloxyphenoxyethyl)benzene				
	Sol/Sol	270.1	14.9	55.16	[136]
	Sol/Meso	337.7	14.3	42.34	
	Meso/Liq	369.5	24.6	66.58	164.08
C <sub>78</sub> H <sub>106</sub> O <sub>7</sub>	2,5-bis-(4-decyloxyphenyl)benzoic acid anhydride				
	Sol/Smec	417.2	49.5	118.65	[122]
	Smec/Liq	445.2	26.3	59.07	
C <sub>78</sub> H <sub>108</sub> N <sub>2</sub> O <sub>6</sub>	N,N'-bis-(4,4"-didecyloxy-p-terphenyl-2'-carbonyl)hydrazine				
	Sol/Smec	414.2	22.5	54.32	[122]
	Smec/Liq	432.2	12.0	27.76	
C <sub>78</sub> H <sub>110</sub> O <sub>9</sub>	1-[6-[3,6,7,10,11-pentakis(octyloxy)triphenyl-2-yloxy]hexyloxy]-9,10-anthraquinone				
	Sol/Meso	313.0	30.4	97.12	[7]
	Meso/Liq	326.0	3.04	9.33	
C <sub>78</sub> H <sub>120</sub> O <sub>12</sub>	4,7-dimethyloctanoic acid, 2,3,6,7,10,11-triphenylenehexyl ester				
	Sol/Meso	Not reported in paper			[324]
	Meso/Liq	385.3	4.8	12.46	
C <sub>78</sub> H <sub>120</sub> O <sub>18</sub>	octakis(octanoyloxy)-9,10-anthraquinone				
	Sol/Meso	358.7	90.2	251.46	[89]
	Meso/Liq	423.2	22.6	53.40	
C <sub>78</sub> H <sub>126</sub> O <sub>6</sub>	cholest-5-en-3-ol (3 $\beta$ ), 4,4'-[1,4-phenylenebis(oxy)]bis[nonanoate]				
	Sol/Nem	386.2	36.7	95.03	[152]
	Nem/Liq	404.2	1.6	3.96	
C <sub>78</sub> H <sub>140</sub> N <sub>2</sub> O <sub>10</sub>	N,N'-didodecanoyl-2,3,5,6-tetrakis(dodecanoyloxy)-1,4-benzenediamine				
	Sol/Disc	362.2	82.0	226.39	[188]
	Disc/Liq	470.2	28.0	59.55	
C <sub>79</sub> H <sub>87</sub> F <sub>2</sub> NO <sub>8</sub>	2-cyano-1,3-phenylene bis[4-(4-hexadecylbiphenyl-4-carbonyloxy)-3-fluorobenzoate]				
	Sol/Meso	390.2	47.5	121.73	[28,130]
	Meso/Liq	428.2	29.1	67.96	
C <sub>79</sub> H <sub>91</sub> F <sub>2</sub> NO <sub>8</sub>	2-cyano-1,3-phenylene bis[4-(4-hexadecylbiphenyl-4-carbonyloxy)-2-fluorobenzoate]				

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula Transition	Compound				Ref.
	T (K)	$\Delta H_{\text{pcc}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pcc}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	
Sol/Meso	397.2	58.6	147.53		
Note: Sol/Sol transition enthalpy is included in the Sol/Meso value.					
Meso/Liq	422.2	28.7	67.98	215.51	[130]
C <sub>79</sub> H <sub>94</sub> F <sub>2</sub> O <sub>8</sub>					
	4'-hexadecyl-[1,1'-biphenyl]-4-carboxylic acid, (2-methyl-1,3-phenylene)bis[oxycarbonyl (3-fluoro-4,1-phenylene)] ester				
Sol/Meso	405.2	56.8	140.18		
Note: Sol/Meso enthalpy includes other Sol/Sol transitions.					
Meso/Liq	467.2	Decomposed			[168]
C <sub>79</sub> H <sub>110</sub> O <sub>7</sub>					
	bis-[(4,4"-didecyloxy-p-terphenyl-2'-yl)methyl]carbonate				
Sol/SmeC	378.2	84.2	222.63		
SmeC/Liq	389.2	15.9	40.85	263.48	[122]
C <sub>80</sub> H <sub>94</sub> O <sub>8</sub>					
	1,3-phenylene bis[4-(4'-n-hexadecylbiphenyl-4-carbonyloxy)-2-methylbenzoate]				
Sol/Meso	376.7	74.06	199.60		
Note: Sol/Sol transition enthalpy is included in the above value.					
Meso/Liq	429.2	25.03	58.32	257.92	[26]
C <sub>80</sub> H <sub>106</sub> N <sub>2</sub> O <sub>8</sub> S					
	(3,4-dicyano-2,5-thiophenediyi)bis(2,1-ethynediyl-4,1-phenylene) 3,4-bis(undecyloxy)benzoate				
Sol/Col	379.4	59.99	158.12		
Col/Liq	410.7	5.29	12.88	171.00	[234]
C <sub>80</sub> H <sub>120</sub> O <sub>4</sub>					
	2,3,5,6,8,9,11,12-octaheptyltetraoxy[8]circulene				
Sol/Meso	466.2	28.0	60.06		
Meso/Liq	493.2	30.0	60.83	120.89	[262]
C <sub>80</sub> H <sub>130</sub> O <sub>6</sub>					
	cholest-5-en-3-ol (3 $\beta$ ), 4,4'-[1,4-phenylenebis(oxy)]bis[decanoate]				
Sol/Nem	380.2	27.3	71.80		
Nem/Liq	415.2	3.4	8.19	79.99	[152]
C <sub>81</sub> H <sub>122</sub> F <sub>3</sub> N <sub>7</sub> O <sub>11</sub>					
	bis[2-[3,4-bis(undecyloxy)phenyl]ethyl] N-[N <sup>10</sup> -[(trifluoroacetyl)pteroyl]-L-glutamate				
Sol/SmeC	260.2	10.0	38.43		
SmeC/Liq	508.2	34.0	66.90	105.33	[317]
C <sub>81</sub> H <sub>137</sub> N <sub>3</sub> O <sub>6</sub>					
	4-(phenylmethyl)-3,5-bis[3,4,5-tris(decyloxy)phenyl-4H-1,2,4-triazole]				
Sol/Meso	Not reported in paper				
Sol/Meso	335.9	3.61	10.75		[259]
C <sub>82</sub> H <sub>100</sub> F <sub>2</sub> O <sub>8</sub>					
	1,3-phenylene bis[4-(4'-n-octadecylbiphenyl-4-carbonyloxy)-2-fluorobenzoate]				
Sol/Meso	377.2	73.96	196.08		
Meso/Liq	467.7	29.92	63.97	260.05	[26]
C <sub>82</sub> H <sub>100</sub> F <sub>2</sub> O <sub>8</sub>					
	1,3-phenylene bis[4-(4'-n-octadecylbiphenyl-4-carbonyloxy)-3-fluorobenzoate]				
Sol/Meso	408.2	57.35	140.49		
Meso/Liq	453.7	24.20	53.34	193.83	[26]
C <sub>82</sub> H <sub>96</sub> O <sub>3</sub>					
	11-{pentakis[(4-pentylphenyl)ethynyl]}phenoxy}undecanoic acid				
Sol/Meso	356.4	43.5	122.05		
Meso/Liq	364.4	0.2	0.55	122.60	[19]
C <sub>82</sub> H <sub>98</sub> O <sub>2</sub>					
	11-{pentakis[(4-pentylphenyl)ethynyl]}phenoxy}undecan-1-ol				
Sol/Meso	339.6	25.8	75.97		
Meso/Liq	367.6	0.2	0.54	76.51	[19]
C <sub>82</sub> H <sub>98</sub> O <sub>8</sub>					
	1,3-phenylene bis[4-(4'-n-octadecylbiphenyl-4-carbonyloxy)benzoate]				
Sol/Meso	390.2	48.05	123.14		
Meso/Liq	477.2	28.47	59.66	182.80	[26]
C <sub>82</sub> H <sub>104</sub> N <sub>2</sub> O <sub>6</sub>					
	1,3-phenylenebis[(E)-methylidynenitrilo-4,1-phenylene] 4'-(octadecyloxy)[1,1'-biphenyl]-4-carboxylate				
Sol/Meso	425.2	13.9	32.69		
Meso/Meso	463.2	44.8	96.72		

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula Transition	Compound				Ref.
	T (K)	$\Delta H_{\text{pcce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pcce}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	
Meso/Liq	518.7	31.8	61.31	190.72	[299]
$\text{C}_{82}\text{H}_{114}\text{N}_2\text{O}_8\text{S}$		4-[5-[4-[3,4-bis(dodecyloxy)benzoyl]oxy]phenyl]-3,4-dicyano-2-thienyl]ethynyl]phenyl, 3,4-bis(dodecyloxy)benzoate			
	Sol/Col	385.9	54.5	141.23	[373]
	Col/Liq	394.7	4.4	11.15	
$\text{C}_{82}\text{H}_{118}\text{O}_7$		1,9-bis(4,4"-didecyloxy-p-terphenyl-2'-yl)-2,5,8-trioxanonane			
	Sol/Smec	347.2	72.8	209.68	[122]
	Smec/Liq	363.2	8.0	22.03	
$\text{C}_{82}\text{H}_{134}\text{O}_6$		cholest-5-en-3-ol ( $3\beta$ ), 4,4'-[1,4-phenylenebis(oxy)]bis[undecanoate]			
	Sol/Nem	383.2	29.2	76.20	[152]
	Nem/Liq	395.2	0.2	0.51	
$\text{C}_{82}\text{H}_{136}\text{N}_2\text{O}_6$		2,4-bis(3',4',5'-tridecyloxyphenyl)-6-phenylpyrimidine			
	Sol/Col	318.9	36.0	112.89	[30]
	Col/Liq	356.1	2.70	7.58	
$\text{C}_{83}\text{H}_{95}\text{F}_2\text{NO}_8$		2-cyano-1,3-phenylene bis[4-(4'-octadecylbiphenyl-4-carbonyloxy)-3-fluorobenzoate]			
	Sol/Meso	385.2	38.4	99.69	[28, 130]
	Meso/Liq	427.7	29.2	68.27	
$\text{C}_{83}\text{H}_{99}\text{F}_2\text{NO}_8$		2-cyano-1,3-phenylene bis[4-(4-octadecylbiphenyl-4-carbonyloxy)-2-fluorobenzoate]			
	Sol/Meso	396.2	59.4	149.92	[130]
	Note: Sol/Sol transition enthalpy is included in the Sol/Meso value.				
$\text{C}_{83}\text{H}_{102}\text{F}_2\text{O}_8$	Meso/Liq	422.7	30.2	71.45	[130]
				221.37	
$\text{C}_{83}\text{H}_{120}\text{O}_9$		4'-octadecyl-[1,1'-biphenyl]-4-carboxylic acid, (2-methyl-1,3-phenylene)bis[oxycarbonyl(3-fluoro-4,1-phenylene)] ester			
	Sol/Meso	400.7	44.0	109.81	[168]
	Meso/Liq	465.7	Decomposed		
$\text{C}_{84}\text{H}_{100}\text{O}_3$		1-[6-[3,6,7,10,11-pentakis(nonyloxy)triphenyl-2-yloxy]hexyloxy]-9,10-anthraquinone			
	Sol/Sol	294.0	1.11	3.78	[7]
	Sol/Meso	304.0	0.26	0.86	
$\text{C}_{84}\text{H}_{102}\text{N}_4\text{O}_8$	Meso/Liq	317.0	21.3	67.19	[7]
				71.83	
$\text{C}_{84}\text{H}_{106}\text{O}_8$		11-[pentakis[(4-pentylphenyl)ethynyl]phenoxy]undecanoic acid, ethyl ester			
	Sol/Meso	329.8	51.6	156.46	[19]
	Meso/Liq	342.2	0.3	0.88	
$\text{C}_{84}\text{H}_{102}\text{N}_4\text{O}_8$		5,10,15,20-tetrakis(4-octanoyloxyphenyl)porphyrin			
	Sol/Col	297.2	20.59	69.27	[285]
	Col/Col	321.6	15.46	48.07	
$\text{C}_{84}\text{H}_{114}\text{N}_2\text{O}_8\text{S}$	Col/Liq	397.9	12.15	30.54	[285]
				147.88	
$\text{C}_{84}\text{H}_{114}\text{O}_{12}$		1,3-phenylene bis[4-(4'-n-octadecylbiphenyl-4-carbonyloxy)-2-methylbenzoate]			
	Sol/Meso	376.2	84.40	224.35	[26]
	Meso/Liq	428.2	24.18	56.47	
$\text{C}_{84}\text{H}_{122}\text{O}_8$		1,12-bis(4,4"-didecyloxy-p-terphenyl-2'-yl)-2,5,8,11-tetraoxadodecane			
	Sol/Sol	271.4	0.4	1.47	[136]
	Sol/Sol	287.1	1.2	4.18	
$\text{C}_{84}\text{H}_{114}\text{O}_{12}$	Sol/Meso	349.8	20.2	57.75	[136]
	Meso/Liq	356.5	20.4	57.22	
				120.62	

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula Transition	Compound					Ref.
	T (K)	$\Delta H_{\text{pcce}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pcce}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	
C <sub>84</sub> H <sub>138</sub> O <sub>6</sub>	Sol/Smec	343.2	73.4	213.87		[122]
	Smec/Liq	353.2	7.9	22.37	236.24	
C <sub>86</sub> H <sub>136</sub> O <sub>18</sub>	Sol/Nem	378.2	11.2	29.61		[152]
	Nem/Liq	407.2	3.0	7.37	36.98	
	Sol/Meso	359.7	85.4	237.42		
C <sub>86</sub> H <sub>154</sub> N <sub>2</sub> O <sub>6</sub>	Meso/Liq	421.2	21.6	57.11	294.53	538.9
	Sol/Col	305.1	36.7	120.29		[178]
	Col/Liq	332.5	5.60	16.84	137.13	
C <sub>88</sub> H <sub>130</sub> O <sub>9</sub>	Sol/Sol	310.0	4.01	12.94		[7]
	Sol/Meso	322.0	13.7	42.55		
	Meso/Liq	336.0	0.50	1.49	56.98	
	Sol/Meso	433.2	27.0	62.33		
C <sub>88</sub> H <sub>136</sub> O <sub>4</sub>	Meso/Liq	456.2	16.0	35.07	97.40	[262]
	Sol/Col	308.2	63.94	207.46		[349]
	Col/Liq	361.2	1.24	3.43	210.89	
C <sub>90</sub> H <sub>126</sub> O <sub>12</sub>	Sol/Sol	207.2	5.0	24.13		[136]
	Sol/Sol	259.5	12.2	47.01		
	Sol/Meso	340.0	29.8	87.65		
	Meso/Liq	344.4	17.1	49.65	208.44	
	Sol/Meso	357.4	66.2	185.23		
C <sub>90</sub> H <sub>134</sub> N <sub>2</sub> O <sub>7</sub>	Meso/Liq	391.8	10.4	26.54	211.77	[77]
	Sol/Meso	399.4	103.4	258.89		[77]
	Meso/Liq	421.4	18.1	42.95	301.84	
C <sub>90</sub> H <sub>164</sub> N <sub>2</sub> O <sub>10</sub>	Sol/Disc	365.2	103.0	282.04		[188]
	Disc/Liq	462.2	22.0	47.60	329.64	
	Sol/Col	299.2	36.00	120.32		
C <sub>92</sub> H <sub>118</sub> N <sub>4</sub> O <sub>8</sub>	Col/Liq	407.7	12.98	31.84	152.16	[285]
	Sol/Meso	357.2	11	30.80		[332]
	Meso/Liq	484.2	36	74.35	105.15	
C <sub>92</sub> H <sub>156</sub> N <sub>2</sub> O <sub>7</sub>	Sol/Col	332.9	32.5	97.63		[30]
	Col/Liq	378.2	4.96	13.11	110.74	
	Sol/Col	329.7	49.0	148.62		[30]
C <sub>92</sub> H <sub>156</sub> N <sub>2</sub> O <sub>7</sub>	Col/Liq	387.2	3.70	9.56	158.18	

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula Transition	Compound					Ref.
	T (K)	$\Delta H_{\text{pcc}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pcc}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (estimated)	
$C_{93}H_{146}F_3N_7O_{11}$	bis{2-[3,4-bis(tetradecyloxy)phenyl]ethyl} N-[N <sup>10</sup> -[(trifluoroacetyl)pteroyl]-L-glutamate					
	Sol/Meso	Not reported in paper				
	Meso/Meso	309.2	40.0	129.39		
	Meso/Meso	499.2	9.0	18.03		
$C_{93}H_{161}N_3O_6$	Meso/Liq	505.2	9.0	17.81		[317]
	Sol/Meso	4-(phenylmethyl)-3,5-bis[3,4,5-tris(dodecyloxy)phenyl]-4H-1,2,4-triazole				
	Meso/Liq	Not reported in paper				
	Meso/Liq	340.3	5.21	15.31		[259]
$C_{94}H_{152}O_{18}$	Sol/Meso	octakis(decanoxyloxy)-9,10-anthraquinone				
	Meso/Liq	371.9	146.2	393.12		
	Sol/Meso	420.4	19.9	47.34	440.46	595.7
	Meso/Liq					[89]
$C_{95}H_{132}O_{11}$	Sol/Nem	4-[3',4',5'-tri(4-dodecyloxybenzoyloxy)]-benzoyloxy-4''-(4-dodecyloxybenzoyloxy)biphenyl				
	Nem/Liq	356.4	69.04	193.71		
	Sol/Nem	365.7	0.41	1.12	194.83	
	Nem/Liq					[219]
$C_{96}H_{152}O_4$	Sol/Meso	2,3,5,6,8,9,11,12-octanoyltetraoxy[8]circulene				
	Meso/Liq	421.2	38.0	90.22		
	Sol/Meso	442.2	14.0	31.66	121.88	
	Meso/Liq					[262]
$C_{97}H_{140}O_{12}$	Sol/Col	2,2',3,3',6,6',7,7'-octakis(heptyloxy)-12,12'(13H,13'H)-spiro[11H-triphenyleno[2,3-b][1,4]dioxepin]				
	Col/Liq	343.2	16.6	48.37		
	Sol/Col	376.2	1.7	4.52	52.89	
	Col/Liq					[367]
$C_{98}H_{178}N_2O_6$	Sol/Col	2,5-bis[3,4,5-tris(tetradecyloxyphenyl)]-1,3,4-oxadiazole				
	Col/Liq	307.1	53.0	172.58		
	Sol/Col	331.8	6.38	19.23	191.81	575.7
	Col/Liq					[178]
$C_{102}H_{158}N_2O_7$	Sol/Meso	bis[10-[4-(4-decylphenyliminomethynyl)phenoxy]decyl] 2-[6-(cholesteryloxy)hexyl]malonate				
	Meso/Liq	355.7	80.4	226.03		
	Sol/Meso	383.7	19.0	49.52	275.55	
	Meso/Liq					[77]
$C_{102}H_{158}N_2O_9$	Sol/Meso	bis[10-[4-(4-decyloxyphenyliminomethynyl)phenoxy]decyl] 2-[6-(cholesteryloxy)hexyl]malonate				
	Meso/Liq	382.5	89.8	234.77		
	Sol/Meso	401.6	18.5	46.07	280.84	
	Meso/Liq					[77]
$C_{102}H_{168}O_{18}$	Sol/Meso	octakis(undecanoyloxy)-9,10-anthraquinone				
	Meso/Liq	375.2	149.5	398.45		
	Sol/Meso	416.6	19.6	47.05	445.50	652.5
	Meso/Liq					[89]
$C_{102}H_{176}N_2O_8$	Sol/Chol	2,4-bis(3',4',5'-tridecyloxyphenyl)-6-(3',4'-didecyloxyphenyl)pyrimidine				
	Chol/Liq	309.4	26.2	84.68		
	Sol/Chol	404.6	5.03	12.43	97.11	1325
	Chol/Liq					[30]
$C_{102}H_{166}N_2O_{10}$	Sol/Disc	N,N'-dihexadecanoyl-2,3,5,6-tetrakis(hexadecanoyloxy)-1,4-benzenediamine				
	Disc/Liq	367.2	115.0	313.18		
	Sol/Disc	463.2	18.0	38.86	352.04	690.8
	Disc/Liq					[188]
$C_{103}H_{124}O_5$	Sol/Sol	ethyl 102-hydroxy-17,80-dioxaoctacyclo-[85.2.2.27,10.213,16.243,46.251,54.281,84.12,6]dohecta-2,4,6(102),8,9,13,15,43,51,53,81,83,87,89,90,92,94,96,98,100-heneicosane-11,41,47,49,55,85-hexayne-4-carboxylate				
	Sol/Nem	407.2	Not reported in paper			
	Sol/Nem	415.2	Not reported in paper			
	Nem/Liq	434.2	0.20	0.46		[327]
$C_{103}H_{281}N_3O_9$	Sol/Sol	N,N-bis[2-[[3,4-bis(tridecyloxy)benzoyl]amino]ethyl]-3,4-bis(tridecyloxy)benzamide				
	Sol/Sol	318.6	40.1	125.86		
	Note: Sol/Sol transition was observed only on the first heating cycle.					
	Sol/Gel	354.6	55.8	157.36		
	Gel/Liq	373.3	1.3	3.48	286.70	
						[379]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula Transition	Compound					Ref.
	T (K)	$\Delta H_{\text{pcc}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pcc}}$	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}} \text{ (exp)}$ (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpcc}}$ (estimated)	
$C_{104}H_{168}O_4$ Sol/Meso	2,3,5,6,8,9,11,12-octadecyltetraoxy[8]circulene					[262]
	406.2	53.0	130.48			
$C_{105}H_{124}F_6O_{11}$ Sol/Sol	Meso/Liq	438.2	26.0	59.33	189.81	
	Sol/SmeC	356.2	5.5	15.44		
	SmeC/SmeC	396.2	35.0	88.34		
	SmeC/Liq	438.2	0.3	0.68		
$C_{105}H_{156}O_{12}$ Sol/Col	478.2	18.0	37.64	142.10		[419]
	338.2	21.6	63.87			
	Col/Liq	382.2	2.0	5.23	69.10	
$C_{108}H_{132}O_{12}$ Sol/Meso	2,3,6,7,10,11-hexakis(4-octyloxybenzoyloxy)triphenylene					[81]
	Meso/Meso	423.2	21	49.62		
	Meso/Liq	440.2	6.0	13.63		
$C_{108}H_{162}N_2O_{10}$ Sol/Meso	N,N-di(3,4,5-tridodecyloxyphenyl)perylene-3,4:9,10-tetracarboxylic acid bisimide					[329]
	Meso/Liq	Not reported in paper				
	646.2	8.9	13.77			
$C_{109}H_{178}F_3N_7O_{11}$ Sol/Meso	bis{2-[3,4-bis(octadecyloxy)phenyl]ethyl} N-{N <sup>10</sup> -(trifluoroacetyl)pteroyl]-L-glutamate					[317]
	Not reported in paper					
	Meso/Meso	335.2	81.0	241.65		
	Meso/Liq	480.2	5.0	10.41		
		496.2	8.0	16.12	268.18	
$C_{109}H_{193}N_3O_9$ Sol/Sol	N,N-bis[2-[3,4-bis(tetradecyloxy)benzoyl]amino]ethyl]-3,4-bis(tetradecyloxy)benzamide					[379]
	323.6	36.2	111.87			
	Sol/Gel	359.4	50.9	141.62		
	Gel/Liq	379.0	0.9	2.37	255.86	
$C_{110}H_{84}O_{18}$ Sol/Meso	octakis(dodecanoyloxy)-9,10-anthraquinone					[89]
	Meso/Liq	376.2	164.2	436.47		
		413.2	18.3	44.29	480.76	
$C_{110}H_{146}FeO_{10}$ Sol/Cube	octakis(dodecanoyloxy)-9,10-anthraquinone					[308]
	Cube/Liq	436.1	69.4	159.14		
		446.1	7.2	16.14	175.28	
$C_{110}H_{202}N_2O_6$ Sol/Chol	bis{4-[4'-(3",4",5"-tris(octyloxy)phenylcarbonyloxy)biphenyloxycarbonyl]phenyl} ferrocene-1,1'-dicarboxylate					[178]
	324.6	79.5	244.92			
	Chol/Liq	327.3	3.73	11.40	256.32	
$C_{112}H_{196}N_2O_9$ Sol/Chol	2,5-bis[3,4,5-tris(hexadecyloxyphenyl)]-1,3,4-oxadiazole					[30]
	335.3	15.6	46.53			
	Chol/Liq	413.7	6.52	15.76	62.29	
$C_{113}H_{84}F_{104}O_{16}$ Sol/Col	2,4,6-tris(3',4',5'-tridecyloxyphenyl)pyrimidine					[324]
	361.2	86.5	239.48			
	Col/Liq	404.2	5.6	13.85	253.33	
$C_{113}H_{116}F_{72}O_{16}$ Sol/Meso	2,2-bis[[3,4-bis[(5,5,6,6,7,7,8,8,9,9,10,10,10,10-tridecafluorodecyl)oxy]benzoyl]oxy]methyl]-1,3-propanediyl 3,4-bis[(5,5,6,6,7,7,8,8,9,9,10,10,10-tridecafluorodecyl)oxy]benzoate					[330]
	Not observed					
	Col/Liq	373.2	4.1	10.99		

Note: Authors report that a crystalline phase has yet to be observed.

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
	T (K)	$\Delta H_{\text{pcc}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pcc}}$		
$\text{C}_{113}\text{H}_{136}\text{F}_{52}\text{O}_{16}$					
	Sol/Meso	Not observed			
	Meso/Liq	381.2	5.6	14.69	[330]
	Note: Authors report that a crystalline phase has not yet been observed.				
$\text{C}_{113}\text{H}_{172}\text{O}_{12}$					
	Sol/Col	333.2	19.1	57.32	
	Col/Liq	394.2	5.2	13.19	70.51 [367]
$\text{C}_{114}\text{H}_{150}$					
	Sol/Disc	279.3	18.1	64.80	
	Disc/Liq	342.5	0.3	0.88	65.68 [354]
$\text{C}_{115}\text{H}_{205}\text{N}_3\text{O}_9$					
	Sol/Sol	327.4	50.2	153.33	
	Sol/Gel	357.3	45.5	127.34	
	Gel/Liq	380.6	1.0	2.63	283.30 [379]
$\text{C}_{117}\text{H}_{195}\text{N}_3\text{O}_7$					
	Sol/Col	320.1	55.7	174.01	
	Col/Liq	366.7	9.0	24.54	198.55 [366]
$\text{C}_{118}\text{H}_{100}\text{O}_{18}$					
	Sol/Meso	375.2	163.8	436.57	
	Meso/Liq	411.2	16.7	40.61	477.18 766.1 [89]
$\text{C}_{120}\text{H}_{144}\text{N}_9\text{O}_{12}\text{P}_3$					
	Sol/Smec	460.0	74.0	160.87	
	Smec/Smec	482.0	1.0	2.07	
$\text{C}_{120}\text{H}_{150}\text{N}_3\text{O}_{16}\text{P}_3$					
	Smec/Smec	499.0	0.5	1.00	
	Smec/Liq	512.0	18.0	35.16	199.10 [361]
$\text{C}_{121}\text{H}_{188}\text{O}_{12}$					
	Sol/Col	329.2	30.0	91.13	
	Col/Liq	379.2	2.3	6.07	97.20 [367]
$\text{C}_{121}\text{H}_{217}\text{N}_3\text{O}_9$					
	Sol/Sol	316.2	21.2	67.05	
	Sol/Gel	358.5	58.0	161.79	
	Gel/Liq	380.1	1.0	2.63	231.47 [379]
$\text{C}_{122}\text{H}_{170}\text{FeO}_{10}$					
	Sol/Cube	399.7	128.7	322.0	
$\text{C}_{126}\text{H}_{116}\text{O}_{18}$	Cube/Liq	447.8	4.4	9.83	331.82 NA [308]
$\text{C}_{128}\text{H}_{186}\text{N}_2\text{O}_{12}$					
	Sol/Meso	Not reported in paper			
	Meso/Liq	556.2	6.2	11.15	[323]

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
	T (K)	$\Delta H_{\text{pcc}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pcc}}$		
$\text{C}_{134}\text{H}_{132}\text{O}_{18}$	<i>octakis(pentadecanoxyloxy)-9,10-anthraquinone</i>				
	Sol/Meso	375.7	50.5	134.42	
	Meso/Meso	382.7	198	517.38	
$\text{C}_{134}\text{H}_{194}\text{FeO}_{10}$	Meso/Liq	405.5	13	32.06	683.86
	Sol/Col	390.3	98.9	253.39	NA
	Col/Liq	445.5	1.6	3.59	256.98
$\text{C}_{135}\text{H}_{237}\text{N}_3\text{O}_{12}$	<i>bis[4-[4'-(3',4'',5''-tris(dodecyloxy)phenylcarbonyloxy)biphenyloxycarbonyl]phenyl]ferrocene-1,1'-dicarboxylate</i>				
	Sol/Meso	301.2	107.0	355.24	
	Meso/Liq	437.2	53.0	121.23	476.47
$\text{C}_{138}\text{H}_{186}\text{S}_6$	<i>hexakis[4-(5'dodecyl-2'-thienyl)phenyl]benzene</i>				
	Sol/Sol	302.4	Not reported in paper		
	Sol/Meso	308.9	58.19	118.38	
	Note: Sol/Sol transition enthalpy is included in the Sol/Meso value.				
	Meso/Liq	428.7	41.14	95.96	214.34
$\text{C}_{141}\text{H}_{132}\text{N}_6\text{O}_{18}$	<i>7-[4'(4'-cyano[1,1'-biphenyl]-4-yl)oxy]heptanoic acid 10,15-dihydro-5<i>H</i>-tribenzo[a,d,g]cyclononene-2,3,7,8,12,13-hexyl ester</i>				
	Sol/Sol	338.2	18.8	55.59	
	Sol/Smec	385.2	7.0	18.17	
$\text{C}_{142}\text{H}_{148}\text{O}_{18}$	Smec/Liq	396.2	8.2	20.70	94.46
	<i>octakis(hexadecanoxyloxy)-9,10-anthraquinone</i>				
	Sol/Meso	376.9	57.7	153.09	
$\text{C}_{146}\text{H}_{218}\text{FeO}_{10}$	Meso/Meso	386.2	236.7	612.89	
	Meso/Liq	404.3	13	32.15	798.13
	Sol/Col	385.8	82.9	212.29	NA
	Col/Liq	443.9	1.8	4.05	216.34
	<i>bis[4-[4'-(3',4'',5''-tris(tetradecyloxy)phenylcarbonyloxy)biphenyloxycarbonyl]phenyl]ferrocene-1,1'-dicarboxylate</i>				
	Sol/Smec	385.8	82.9	212.29	
$\text{C}_{146}\text{H}_{242}\text{O}_{12}$	Smec/Liq	443.9	1.8	4.05	216.34
	<i>2,3,6,7,10-pentakis(decyloxy)-11-[[10-[[3,6,7,10,11-pentakis[(3R)-3,7-dimethyloctyl]oxy]-2-triphenylenyl]-oxy]decyl]oxy]triphenyl</i>				
	Sol/Meso	307.2	21.4	69.66	
	Meso/Liq	318.2	3.9	12.26	81.92
	Note: The above values were determined from the first heating cycle. On subsequent heating cycles the authors observed only a Meso/Liq phase transition at 314.2 K, with an enthalpy of transition of 8.5 kJ·mol <sup>-1</sup> . The authors reported the mesophase failed to crystallize.				
	Sol/Smec	318.2	3.9	12.26	
$\text{C}_{148}\text{H}_{210}\text{N}_2\text{O}_{14}$	Sol/Meso	619.2	15.0	24.22	
	Meso/Liq	619.2	15.0	24.22	
	Sol/Smec	619.2	15.0	24.22	
$\text{C}_{158}\text{H}_{242}\text{FeO}_{10}$	<i>bis[4-[4'-(3',4'',5''-tris(hexadecyloxy)phenylcarbonyloxy)biphenyloxycarbonyl]phenyl]ferrocene-1,1'-dicarboxylate</i>				
	Sol/Col	393.6	110.0	279.47	
	Col/Liq	445.1	1.4	3.15	282.62
$\text{C}_{160}\text{H}_{192}\text{N}_{12}\text{O}_{16}\text{P}_4$	<i>2,2,4,4,6,6,8,8-octakis[4-(E)-[[4-(heptyloxy)phenyl]imino]methylphenoxy]-2,2,4,4,6,6,8,8-octydro-1,3,5,7,2,4,6,8-tetraazatetraphosphocine</i>				
	Sol/Smec	428.0	77.0	179.91	
	Smec/Liq	430.0	28.0	65.12	245.03
$\text{C}_{164}\text{H}_{292}\text{N}_2\text{O}_{14}$	<i>N,N-di(3,4,5-tridodecyloxyphenyl)-1,6,7,12-tetra(4-(1,1,3,3-tetramethylbutyl)phenoxy)perylene-3,4:9,10-tetracarboxylic acid bisimide</i>				
	Sol/Smec	430.0	28.0	65.12	
	Smec/Liq	430.0	28.0	65.12	

TABLE 11. The solid-liquid phase change properties of liquid crystals—Continued

Molecular Formula Transition	Compound		$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (exp) (J·K <sup>-1</sup> ·mol <sup>-1</sup> )	$\Delta_0^{T_{\text{fus}}} S_{\text{tpce}}$ (estimated)	Ref.
	T (K)	$\Delta H_{\text{pcc}}$ (kJ·mol <sup>-1</sup> )	$\Delta S_{\text{pcc}}$		
Sol/Meso Meso/Liq	Not reported in paper				
	556.2	23.9	42.97		[329]
$C_{165}H_{176}F_6O_{17}$	<i>3,4,5-tris[[6-[[2',3'-difluoro-4'-(octyloxy)[1,1',4',1"-terphenyl]-4-yl]oxy]hexyl]oxy]benzoic acid, 11-[pentakis[(4-methoxyphenyl)ethynyl]phenoxy]undecyl ester</i>				
	Sol/Sol	356.2	10.5	29.48	
	Sol/Smec	388.2	31.0	79.86	
	Smec/Smec	407.2	17.0	41.75	
	Nem/Liq	408.2	0.2	0.49	
$C_{165}H_{180}N_6O_{18}$	432.2	1.5	3.47	155.05	[419]
	<i>4-[(4'-cyano[1,1'-biphenyl]-4-yl)oxy]undecanoic acid 10,15-dihydro-5<i>H</i>-tribenzo[a,d,g]cyclononene-2,3,7,8,12,13-hexyl ester</i>				
	Sol/Smec	368.2	58.3	158.34	
$C_{170}H_{266}FeO_{10}$	Smec/Liq	389.2	21.3	54.73	213.07
	<i>bis{4-[4'-(3",4",5"-tris(octadecyloxy)phenylcarbonyloxy)biphenyloxycarbonyl]phenyl}ferrocene-1,1'-dicarboxylate</i>				
	Sol/Col	394.1	120.4	305.51	
$C_{194}H_{324}O_{12}$	Col/Liq	446.1	1.6	3.59	309.10
	<i>tetrakis[4-(3,4,5-trisdodecyloxyphenyl)phenyl]ethane</i>				
	Sol/Col	322.2	131.2	407.20	
$C_{240}H_{280}N_8O_{32}$	Col/Liq	338.2	10.5	31.05	438.25
	<i>4,6,10,12,16,18,22,24-octakis{2-[11[(4'-cyano-1,1'-biphenyl-4-yl)oxy]undecanoyloxy]ethoxy}-2,8,14,10-tetramethylcalix[4]arene</i>				
	Sol/Nem	Not reported in paper			
Nem/Liq	349.2	20.0	57.27		[307]

TABLE 12. References for Table 11

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## 7. Acknowledgment

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