

4-FLUOROCINNAMIC ACID BIODEGRADATION BY A RHODOCOCCUS STRAIN

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Fluorinated organic compounds

- Highly toxic to living organisms;
- Carcinogenic and mutagenic effects;
- Extremely persistent;



Hazardous environmental pollutants







Accumulation of fluorinated organic compounds in soils and water is a major environmental problem



BIODEGRADATION





Promising approach for the removal from environment What is the potential for biodegradation from environmental compartments?



Chemical processes

- energy-intensive;
- expensive;
- can produce toxic byproducts;

Biodegradation

- cost effective
- allows total mineralization of the toxic

Emerging technology for the degradation of halogenated organic compounds



4-Fluorocinnamic acid

- Intermediate in the synthesis of:
 - Agrochemicals;
 - Pharmaceuticals;
 - Fine chemicals;

4-FCA

Scant information on biodegradation is available;

Biodegradation of this compound by a single strain not reported in the literature



Objective



Assessment of 4-FCA degradation by a pure bacterium



Experimental approach

- Isolation of a 4-FCA degrading strain
- Classification of the 4-FCA degrading strain
- Degradation Assays
 - Biodegradation behaviour in the absence/presence of a conventional carbon source
 - Effect of the initial 4-FCA concentration on the degradation process



Isolation of a 4-FCA degrading strain



Recovery of strains from a Rotating Biological Contactor (RBC) fed with 2- fluorophenol (2-FP)



Isolation of microbial strains using the streak-plate procedure



Inoculation of the isolated strains with 4-FCA



Isolation of a 4-FCA degrading strain

Five microbial strains recovered from a RBC fed with 2- FP



Capacity to grow in batch cultures with 4-FCA supplied as the sole carbon and energy source at concentration of 0.5mM (cultures were incubated on an orbital shaker at 25°C)



Selection of a 4-FCA degrading strain



Classification of the 4-FCA degrading strain

The degrading strain (**Strain S2**) was classified by 16S rRNA gene analysis



Strain S2 constitutes a member of the genus

Rhodococcus





- metabolic versatility
- robustness
- persistence in adverse conditions



Degradation assays

Batch experiments were performed aerobically at 25 °C with constant shaking

- Biodegradation behaviour of 0.5mM of 4-FCA in the absence/presence of a conventional carbon source (acetate 20 mM)
- Effect of the initial 4-FCA concentration in the degradation process (in the presence of 20 mM acetate)

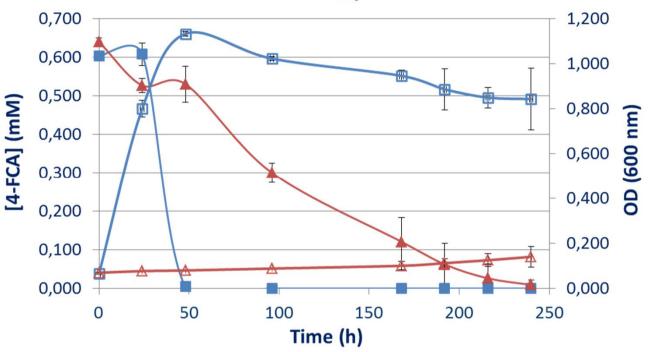


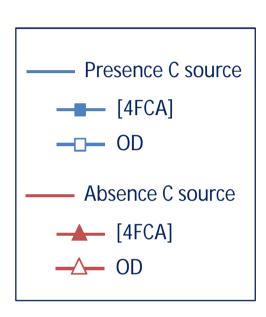
Monitored parameters:

- Growth of the strain
- Consumption of target compound
- Liberation of fluorine ion



Effect of the absence/presence of a conventional carbon source





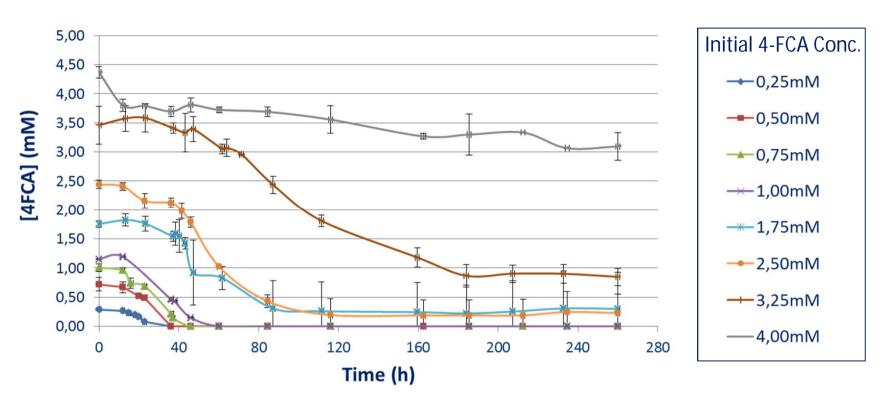
- Complete mineralization of 4-FCA
- Carbon source enhances the rate of biodegradation



The additional substrate promotes the adaptation process towards 4-FCA degradation



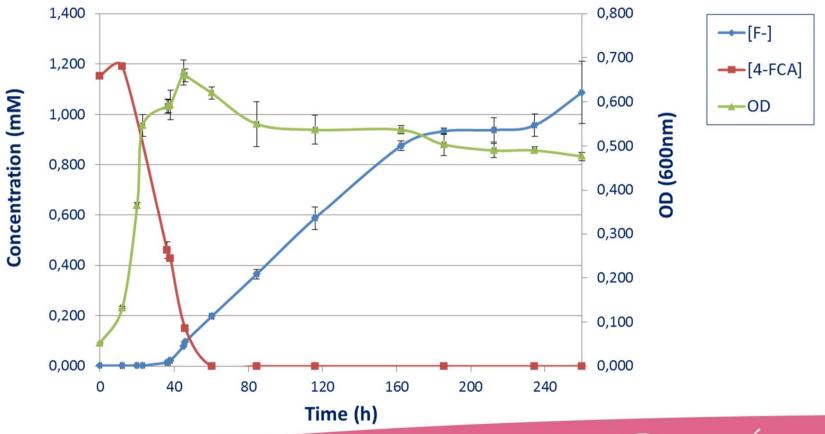
Effect of the initial 4-FCA concentration



- The increase of the initial 4-FCA concentration leads to an increase of the lag phase
- Total degradation of 4-FCA up to 1 mM

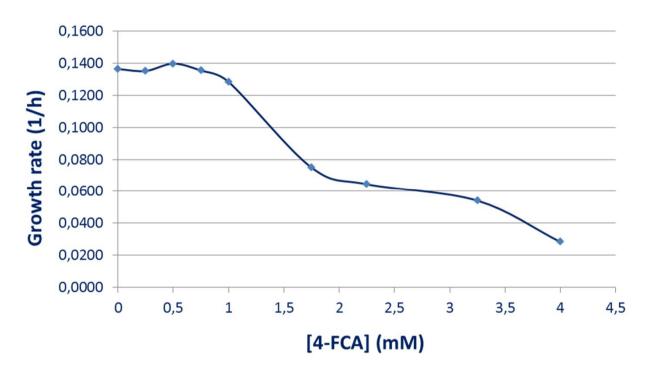


Typical example of batch growth of strain S2 with a mixture of acetate and 4-FCA





Growth kinetics of strain S2 on 4-FCA with a conventional C source



An inhibitory effect with increasing 4-FCA concentration was observed at concentrations higher than 1mM



Conclusions

- Complete mineralization of 0.5 mM of 4-FCA as a sole carbon and energy source was achieved;
- The presence of conventional source of carbon improved the degradation rate of 4-FCA;
- Complete biodegradation of the target compound, in the presence of acetate, within a concentration range between 0 and 1 mM was observed;



Conclusions

Strain S2

- Adaptable strain: able to completely mineralize 4-FCA as a sole carbon and energy source or in the presence of a conventional carbon source
- Robust strain: able to degrade this recalcitrant and toxic compound



Candidate for use in biological treatment plants and bioremediation of contaminated sites containing 4-FCA



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Thank you for your attention!

