M. González-Pajuelo · J. C. Andrade · I. Vasconcelos

Production of 1,3-Propanediol by *Clostridium butyricum* VPI 3266 in continuous cultures with high yield and productivity

Abstract The effects of dilution rate and substrate feed concentration on continuous glycerol fermentation by Clostridium butyricum VPI 3266, a natural 1,3-propanediol producer, were evaluated in this work. A high and constant 1,3-propanediol yield (around 0.65 mol/ mol), close to the theoretical value, was obtained irrespective of substrate feed concentration or dilution rate. Improvement of 1,3-propanediol volumetric productivity was achieved by increasing the dilution rate, at a fixed feed substrate concentration of 30, 60 or 70 g l^{-1} . Higher 1,3-propanediol final concentrations and volumetric productivities were also obtained when glycerol feed concentration was increased from 30 to 60 g l⁻¹, at D = 0.05-0.3 h⁻¹, and from 60–70 g l⁻¹, at D = 0.05 and 0.1 h⁻¹ · 30 g l⁻¹ of 1,3-propanediol and the highest reported value of productivity, $10.3 \text{ g } \text{l}^{-1} \text{ h}^{-1}$, was achieved at D = 0.30 h⁻¹ and 60 g l⁻¹ of feed glycerol. A switch to an acetate/butyrate ratio higher than one was observed for 60 g l⁻¹ of feed glycerol and a dilution rate higher than 0.10 h⁻¹; moreover, at D = 0.30 h⁻¹ 3-hydroxypropionaldehyde accumulation was observed for the first time in the fermentation broth of C. butyricum.

Keywords *Clostridium butyricum* · 1,3-propanediol · Continuous cultures · 3-hydroxypropionaldehyde

Introduction

1,3-propanediol is a versatile degradable intermediate compound for the synthesis of heterocycles and a monomer for the production of polymers, such as

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polyesters and polyurethanes. The classic route to produce this monomer is the chemical process from acrolein, a very harmful reagent. As an alternative, it has been shown that some bacteria are able to produce 1,3-propanediol from glycerol [2, 7–9, 15, 16]. The recent development of PTT, a new polyester based on 1,3-propanediol and terephtalic acid, has increased the number of studies on microbial conversion of glycerol to 1,3-propanediol [3]. Yield, final product concentration and volumetric productivity are the main issues for the feasibility of an industrial production process. A high final product concentration in the effluent turns the down-stream separation process less expensive and, at the same time, contributes to achieve a high volumetric productivity [3]. Clostridium butyricum VPI 3266 was described as probably the best natural 1,3-propanediol producer since, unlike other microorganisms, production of 1,3-propanediol by this strain is not a B_{12} -vitamin dependent process, which is clearly an economical advantage for an industrial application [13]. Furthermore, production of 1,3-propanediol by this strain using a synthetic medium and low-price raw glycerol has been recently reported [4]. Glycerol is metabolised by C. butyricum following two pathways. One pathway leads to glycerol oxidation to dihydroxyacetone (DHA) by a NAD⁺ -dependent glycerol dehydrogenase, followed by DHA phosphorilation to DHA-phosphate (DHAP) by a DHA kinase; DHAP enters the glycolytic pathway. In the other pathway glycerol is dehydrated to 3-hydroxipropionaldehyde (3-HPA) via a glycerol dehydratase; 3-HPA is then reduced to 1,3-propanediol by a 1,3-propanediol dehydrogenase with NADH consumption, leading to a theoretical yield of 0.70 mol/mol in conditions of no butyrate and no H_2 formation [2, 3, 18].

Since an industrial utilisation of *C. butyricum* VPI 3266 may be suggested, continuous glycerol fermentations of this strain have been carried out in the present work in order to increase 1,3-propanediol concentration and productivity by manipulating dilution rate and glycerol feed concentration.

Materials and methods

Organism

Clostridium butyricum VPI 3266 (Virginia Polytechnic Institute Culture Collection, Blacksburg, Va.) was maintained in the synthetic medium described below, in spore form, at -20° C. This strain is available from other culture collections as *C. butyricum* NCIMB 7423 (National Collections of Industrial and Marine Bacteria Ltd., Aberdeen, Scotland, United Kingdom) and *C. butyricum* CECT 361 (Colección Española de Cultivos Tipo; Universitat de Valencia, Valencia, Spain).

Culture media

The synthetic medium used in the experiments contained per litre of deionized water: glycerol, 30–80 g; KH₂PO₄, 0.5 g; K₂HPO₄, 0.5 g; MgSO₄.7H₂O, 0.2 g; CoCl₂.6-H₂O, 0.01 g; FeSO₄.7H₂O, 0.01 g; biotin, 0.04 mg; *p*-aminobenzoic acid, 8 mg; acetic acid, 2 g. The medium pH was adjusted to 6.5 with 6N NH₄OH. The feed medium for continuous cultures was the synthetic medium described above, without acetic acid, and with 0.028 g 1^{-1} of FeSO₄.7H₂O (instead of 0.01 g 1^{-1}), 1.5 g 1^{-1} of NH₄Cl and 1 ml of H₂SO₄17.4 M; medium pH was not adjusted in this case.

Continuous cultures conditions

Continuous cultures were performed in a 2-1 bioreactor (Biostat MD; Braun, Melsungen, Germany), with a working volume of 1,250 ml, and in a 400 ml glass bioreactor, with a working volume of 300 ml. The culture was stirred at 200 rpm, temperature was set to 35°C and pH was maintained constant by automatic addition of 6N NH₄OH. To create anaerobic conditions, the sterilised medium in the vessel was flushed with sterile O₂-free nitrogen until room temperature was attained. A growing culture taken at the early exponential growth phase was used as inoculum (10% v/v). The culture was first grown batchwise and continuous feeding was started once the exponential growth phase was reached. After sterilisation, the feed medium was sparged with sterile O2-free nitrogen, until room temperature was reached. During the experiments, the feed medium was maintained under nitrogen at 30 mbar, to avoid O₂ entry. All tubing was made of butyl rubber and the bioreactor gas outlet was protected with a pyrogallol arrangement [17].

Analytical procedures

Cell concentration was measured turbidometrically, at 620 nm, and correlated with cell dry weight determined

directly. Glycerol, 1,3-propanediol, ethanol and acetic, butyric and lactic acid concentrations were determined by HPLC (System Gold; Beckman, Fullerton, CA, USA). Separation was performed on a Biorad Aminex HPX-87H column (300×7.8 mm; Bio-Rad, Richmond, CA, USA) and detection was achieved by refractive index. Operating conditions were as follows: mobile phase, sulphuric acid 0.5 mM; flow rate, 0.5 ml/min; temperature, 30°C. A qualitative HPLC analysis of 3-hydroxypropionaldehyde in fermentation broth was also performed in the conditions described. As 3-HPA is not commercially available, it was chemically synthesised following the method described by Hall and Stern [5].

1,3-propanediol volumetric productivity ($Q_{1,3-\text{propanediol}}$) and specific 1,3-propanediol formation or glycerol consumption rate ($q_{1,3-\text{propanediol}}$ or glycerol) in chemostat cultures were calculated as $Q_{1,3-\text{propanediol}} = C_{1,3-\text{propanediol}} \times D$ and $q_{1,3-\text{propanediol}}$ or glycerol = $C_{1,3-\text{propanediol}} \times D/X$ respectively, where $C_{1,3-\text{propanediol}}$ is the 1,3-propanediol mass concentration, C_{glycerol} is the consumed glycerol mass concentration, D is the dilution rate of the chemostat and X is the cell mass concentration.

Results

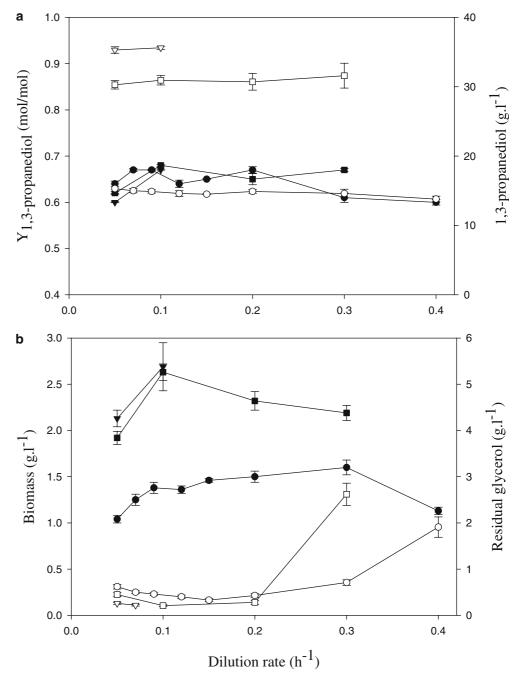
Effect of dilution rate on continuous 1,3-propanediol production

Dilution rate was increased from 0.05 to 0.50 h^{-1} in chemostat cultures of C butyricum VPI 3266, with a constant glycerol feed of 30 g l⁻¹. 1,3-propanediol was the major fermentation end-product, attaining 15 g l^{-1} (Fig. 1a), and acetate and butyrate were the main byproducts. No ethanol or lactate, other possible fermentation products, were synthesised. Residual glycerol exceeded 1 g l⁻¹ only when dilution rate increased above 0.3 h⁻¹ (Fig. 1b). At higher dilution rate of 0.5 h⁻¹, residual glycerol had reached 12.5 g l^{-1} and severe decreases in biomass $(0.2 \text{ g } \text{l}^{-1})$ and 1,3-propanediol $(8.1 \text{ g } 1^{-1})$ were observed (data not shown). Up to a dilution rate of 0.40 h^{-1} , 1,3-propanediol concentration was constant, leading to an increase in the volumetric productivity, with a maximum value of 5.5 g l^{-1} h^{-1} (Fig. 2). For every D value, conversion of glycerol into 1,3-propanediol was between 0.60 and 0.68 mol 1,3propanediol/mol glycerol consumed (Fig. 1a), indicating that the carbon flux through the 1,3-propanediol pathway is under a strong regulation.

Effect of glycerol feed concentration on continuous 1,3-propanediol production

The effect of glycerol feed concentration on the fermentation pattern of *C. butyricum* VPI 3266 in chemostat was also studied. Feed glycerol concentration varied from 30 to 70 g l^{-1} at different dilution rates (0.05–

Fig. 1 Influence of glycerol feed concentration and dilution rate on a- 1,3-propanediol molar yield (*solid symbols*) and 1,3-propanediol concentration (*open symbols*) and on b-biomass concentration (*solid symbols*) and residual glycerol (*open symbols*) in continuous cultures of *C. butyricum* VPI 3266 (pH 6.5, 35°C). *circle, filled circle* 30 g 1⁻¹ of feed glycerol; *square, filled square* 60 g 1⁻¹ of feed glycerol; *inverted triangle, closed inverted triangle* 70 g 1⁻¹ of feed glycerol. *Vertical bars* represent standard deviation values



0.30 h^{-1}). However, when 70 g l^{-1} of glycerol were fed at a dilution rate of 0.20 h^{-1} washout of the reactor occurred.

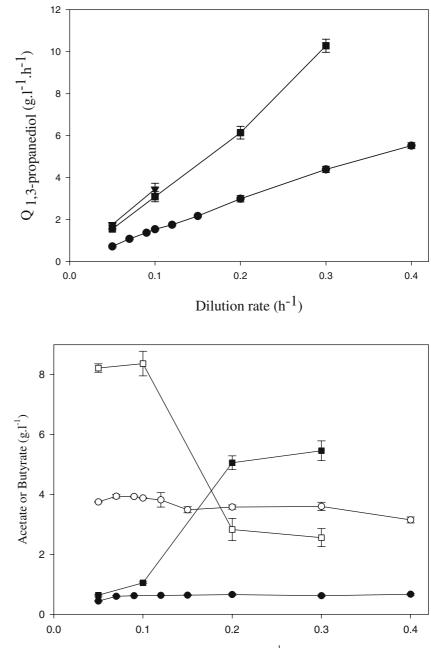
For every dilution rate, an increase in glycerol feed concentration led to an increase in 1,3-propanediol concentration, which achieved 35 g l^{-1} for 70 g l^{-1} of feed glycerol, and 30 g l^{-1} for 60 g l^{-1} of feed glycerol (Fig. 1a). 1,3-propanediol yield was not affected by increasing glycerol feed concentration and was always around 0.65 mol 1,3-propanediol/mol glycerol consumed. Therefore, an improvement of volumetric productivity was observed with increasing dilution rate or glycerol feed concentration (Fig. 2). The highest volu-

metric productivity value, $10.3 \text{ g } \text{l}^{-1} \text{ h}^{-1}$, was obtained at a dilution rate of 0.30 h^{-1} with $60 \text{ g } \text{l}^{-1}$ of feed glycerol. Biomass concentration was also affected by glycerol feed concentration, since up to a two fold increase was achieved when substrate concentration was changed from 30 to 60 and 70 g l^{-1} (Fig. 1b). For every glycerol feed concentration, a decrease of biomass concentration was associated with an increase of residual glycerol in fermentation broth.

For a 30 g l^{-1} glycerol feed, acetate and butyrate concentrations were constant, although the dilution rate increased; furthermore, butyrate concentration was always higher than acetate concentration (Fig. 3). How-

Fig. 2 Influence of glycerol feed concentration and dilution rate on 1,3-propanediol volumetric productivity in continuous cultures of *C. butyricum* VPI 3266 (pH 6.5, 35°C). *Filled circle* 30 g 1^{-1} of feed glycerol, *filled square* 60 g 1^{-1} of feed glycerol, *inverted triangle* 70 g 1^{-1} of feed glycerol. *Vertical bars* represent standard deviation values

Fig. 3 Influence of glycerol feed concentration and dilution rate on acetate (*solid symbols*) and butyrate (*open symbols*) production in continuous cultures of *C. butyricum* VPI 3266 (pH 6.5, 35°C). *circle*, *filled circle* 30 g 1^{-1} of feed glycerol; *square*, *filled square* 60 g 1^{-1} of feed glycerol. *Vertical bars* represent standard deviation values



Dilution rate (h^{-1})

ever, a switch in the acetate/butyrate ratio occurred for a glycerol feed of 60 g l^{-1} , at dilution rates of 0.20 and 0.30 h⁻¹, indicating a change in the metabolic carbon flux. At D=0.30 h⁻¹, 3-HPA was detected for the first time in the fermentation broth of *C. butyricum*, although quantification was not possible, as the pure compound was not available.

Discussion

Clostridium butyricum VPI 3266 has been described as the best candidate for 1,3-propanediol production,

since it was demonstrated that this strain carries out a B_{12} -vitamin independent process [13]. It has also been shown that this strain is able to produce 30 g l⁻¹ of 1,3-propanediol, with a molar yield of 0.61, in continuous cultures using a synthetic medium and raw glycerol [4]. However, there are no reports in literature establishing high substrate consumption values for high growth rate continuous cultures, as pointed out by Biebl et al. [3]. In chemostat cultures with a feed glycerol concentration of 10 g l⁻¹, *C. butyricum* DSM 5431 could grow up to a maximum growth rate of 0.26 h⁻¹, with exhaustive glycerol consumption [1]; biomass decreased markedly at 0.30 h⁻¹, corresponding to glycerol accumulation.

Klebsiella pneumoniae has also been considered a good 1,3-propanediol producer. Menzel et al. [9] showed that this microorganism was able to produce 35.2-48.5 g 1^{-1} of 1,3-propanediol in continuous fermentation and volumetric productivities of 4.9–8.8 g l^{-1} h^{-1} were obtained at dilution rates between 0.10 and 0.25 h^{-1} ; however, in these experiments more than 45 g l⁻¹ of residual glycerol were observed in the fermentation broth. Moreover, K. pneumoniae is an opportunist pathogen microorganism and special care must be taken when used in fermentation. In another attempt to increase the volumetric productivity, C. butyricum DSM 5431 was cultivated in a cell recycle bioreactor with 55 g l⁻¹ of feed glycerol [12]. In this case, a 1,3-propanediol concentration of 26.5 g l⁻¹ was maintained up to a dilution rate of 0.50 h⁻¹, which corresponds to a high volumetric productivity (13.3 g l^{-1} h⁻¹). Nevertheless, this process could only be run for short time periods as the microfiltration membrane was prone to clogging. Other bacteria genus have been used for 1,3propanediol production. Pflugmacher and Gottschalk [11] developed an immobilised cell reactor for 1,3-propanediol production by Citrobacter freundii. However, 10 g l^{-1} of residual glycerol were observed and only 18.7 g l^{-1} of 1,3-propanediol were produced with 60 g l⁻¹ of feed glycerol and D = 0.30 h⁻¹. These results lead to a relatively low 1,3-propanediol yield (0.46 mol/ mol) and a poor productivity (5.6 g l^{-1} h^{-1}). Hartlep et al. [6] reported a two step-process for 1,3-propanediol production, wherein glucose was first converted to glycerol by a recombinant Escherichia coli strain; in a second stage, glycerol was converted to 1,3-propanediol by K. pneumoniae. A fed-batch cultivation under limiting glucose supply resulted in a production of 14 g l^{-1} of 1,3-propanediol in the second stage, leading to a pro-ductivity of 2 g 1^{-1} h⁻¹. The 1,3-propanediol productivity observed in the present work, $10.3 \text{ g } 1^{-1} \text{ h}^{-1}$, is the highest value reported for chemostat cultures.

The switch in acetate/butyrate ratio observed in this work was reported before for C. butyricum DSM 5431 growing in glycerol [1]: whereas at D values from 0.05 to 0.26 h^{-1} cells produced more butyrate than acetate, at a D value of 0.30 h^{-1} a switch was observed and the cells produced more acetate than butyrate. C. tyrobutyricum grown on glucose also showed a decrease of the selectivity for butyrate with an increase in glucose concentration [10]; this decrease was observed at both $D=0.10 \text{ h}^{-1}$ and $D=0.20 \text{ h}^{-1}$, but was faster at D = 0.20 h⁻¹. Up to now, accumulation of 3-HPA, described as a toxic compound to the cells [1], was never reported for C. butyricum. In this work, 3-HPA was detected in fermentation broth from chemostat cultures with 60 g l^{-1} of feed glycerol at a dilution rate of 0.30 h^{-1} , when the acetate/butyrate ratio was higher than one. Formation of butyrate is redox-neutral, but acetate synthesis generates NADH excess, which could be used for the production of 1,3-propanediol from 3-HPA. The 1,3-propanediol yield observed in this case, 0.68 mol/mol, is very close to the theoretical maximum yield (0.70 mol/mol) calculated by Zeng [18] in conditions of no butyrate and no H_2 formation. The strain VPI 3266 is known to produce no molecular H_2 when grown on glycerol as the sole carbon and energy source [14]. Therefore, the inversion of the acetate/butyrate rate may be a mechanism to avoid accumulation 3-HPA by increasing NADH availability.

In this work it was shown that *C. butyricum* VPI 3266 is able to produce up to 30 g l⁻¹ of 1,3-propanediol in continuous cultures from 60 g l⁻¹ of feed glycerol, at high dilution rate, leading to a volumetric productivity of 10.3 g l⁻¹ h⁻¹. This value is the highest ever reported for a chemostat culture of *C. butyricum*. A constant propanediol yield, close to the theoretical value, was also obtained in this work, irrespective of substrate concentration or dilution rate. Since an economic production of 1,3-propanediol from glycerol requires high final concentration and productivity and considering that 1,3propanediol production by *C. butyricum* VPI 3266 is a B₁₂-vitamin independent process, this strain seems to be the best natural candidate for a 1,3-propanediol industrial process.

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