Chemical Engineering Journal 240 (2014) 344-351

Contents lists available at ScienceDirect



Chemical Engineering Journal

Chemical Engineering Journal

journal homepage: www.elsevier.com/locate/cej

Carbon-based materials prepared from pine gasification residues for acetaminophen adsorption



Margarida Galhetas ^{a,b}, Ana S. Mestre ^a, Moisés L. Pinto ^c, Ibrahim Gulyurtlu ^b, Helena Lopes ^b, Ana P. Carvalho ^{a,*}

^a Departamento de Química e Bioquímica and CQB, Faculdade de Ciências da Universidade de Lisboa, Ed. C8, Campo Grande, 1749-016 Lisboa, Portugal

^b LNEG, Estrada do Paço do Lumiar 22, 1749-032 Lisboa, Portugal

^c Departamento de Química, CICECO, Universidade de Aveiro, 3810-193 Aveiro, Portugal

HIGHLIGHTS

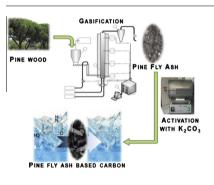
- Fly ash from pine gasification was used as precursor of carbon-based materials.
- Materials with $A_{\rm BET}$ of 1171 $m^2\,g^{-1}$ were obtain by chemical activation with $K_2CO_3.$
- The solids were tested for the removal of acetaminophen from aqueous solution.
- Molecular dimensions of monomer and dimer were considered in the discussion.
- The affinity of paracetamol for the carbon is maximized by pores with width of 0.7 nm.

ARTICLE INFO

Article history: Received 2 October 2013 Received in revised form 22 November 2013 Accepted 25 November 2013 Available online 4 December 2013

Keywords: Pine gasification Fly ash Carbon materials Micropore size distribution Acetaminophen adsorption

GRAPHICAL ABSTRACT



ABSTRACT

Fly ash, a residue produced from pine gasification, was used as precursor of carbon-based materials assayed in acetaminophen adsorption. Materials prepared by activation with K₂CO₃, presented high porosity development ($A_{BET} \approx 1200 \text{ m}^2 \text{ g}^{-1}$) and samples calcined at 900 °C presented high volumes of large micropores and mesopores. Kinetic and equilibrium acetaminophen adsorption data showed that the process obeys to the pseudo-second order kinetic equation and Langmuir model, respectively. The rate of acetaminophen adsorption depends of the presence of larger micropores. For the lab-made samples monolayer adsorption capacities attained values similar to those of commercial carbons. The influence of the micropore size distribution of the carbons in the acetaminophen adsorption process justified the lower adsorption affinities of the lab-made carbons. The influence of the affinity of the molecule towards the carbon surface was demonstrated. The increase of temperature lead to higher monolayer adsorption capacities, most likely due to the easier accessibility of the acetaminophen species to the narrowest micropores.

© 2013 Elsevier B.V. All rights reserved.

1. Introduction

In the industrialized countries energy production is, to a large extent, based on fossil fuels, which leads to the release of

greenhouse gases into the atmosphere, causing serious and well known environmental problems. In this context, energy production through biomass gasification is nowadays considered a low carbon emission technology being an alternative to combustion, since it is,

^{*} Corresponding author. Tel.: +351 217500897; fax: +351 217500888.

E-mail addresses: amgalhetas@fc.ul.pt (M. Galhetas), asmestre@fc.ul.pt (A.S. Mestre), moises.pinto@ua.pt (M.L. Pinto), ibrahim.gulyurtlu@lneg.pt (I. Gulyurtlu), helena.lopes@lneg.pt (H. Lopes), ana.carvalho@fc.ul.pt (A.P. Carvalho).

^{1385-8947/\$ -} see front matter @ 2013 Elsevier B.V. All rights reserved. http://dx.doi.org/10.1016/j.cej.2013.11.067