

# Investigation of Aluminum Equation of State Generation

Aaron Ward  
*Marquette University*

---

## Recommended Citation

Ward, Aaron, "Investigation of Aluminum Equation of State Generation" (2011). *Master's Theses (2009 -)*. Paper 124.  
[http://epublications.marquette.edu/theses\\_open/124](http://epublications.marquette.edu/theses_open/124)

INVESTIGATION OF ALUMINUM EQUATION OF STATE GENERATION

by

Aaron J. Ward

A Thesis submitted to the Faculty of the Graduate School,  
Marquette University,  
in Partial Fulfillment of the Requirements for  
the Degree of Master of Science

Milwaukee, Wisconsin

December 2011

ABSTRACT  
INVESTIGATION OF ALUMINUM EQUATION OF STATE GENERATION

Aaron J. Ward

Marquette University, 2011

There are many forms and methods to construct equations of state, EOSs. These methods are usually tailored for the particular problem of interest. Here, the EOSs of interest are those used in modeling shock responses. These EOSs cover a wide range of physical characteristics such as detonation and explosions, armor and anti-armor materials, and space structures protection. Aluminum will be the primary focus of this work. Aluminum was chosen because it has been studied in great length in the shock regime and is a common component in shock experiments and space type vehicles.

# Contents

<b>List of Figures</b>	<b>iii</b>
<b>List of Tables</b>	<b>v</b>
<b>1 Introduction</b>	<b>1</b>
1.1 Shock Physics . . . . .	1
1.1.1 Hugoniot Calculations . . . . .	5
1.1.2 Release Isentrope Calculations . . . . .	6
1.1.3 Reshock Calculations . . . . .	7
1.1.4 Multiphase Calculations . . . . .	8
<b>2 Equation of State</b>	<b>10</b>
2.1 Mie-Grüneisen EOS . . . . .	11
2.2 Tillotson EOS . . . . .	12
2.3 Multi-Branch Analytical EOS . . . . .	15
2.4 Bushman-Lomonosov EOS . . . . .	18
<b>3 Results</b>	<b>24</b>
3.1 Mie-Grüneisen EOS . . . . .	25
3.2 Tillotson EOS . . . . .	34
3.3 Multi-Branch Analytical EOS . . . . .	41
3.4 Bushman-Lomonosov EOS . . . . .	44
<b>4 Conclusions</b>	<b>52</b>
<b>5 Future Work</b>	<b>57</b>
<b>Bibliography</b>	<b>59</b>
<b>A Equation of State Source Code</b>	<b>64</b>
<b>Appendix A. Equation of State Source Code</b>	<b>64</b>
<b>B Mie-Grüneisen EOS Source Code</b>	<b>86</b>

---

Appendix B. Mie-Grüneisen EOS Source Code	86
C Tillotson EOS Source Code	94
Appendix C. Tillotson EOS Source Code	94
D Multi-Branch Analytical EOS Source Code	102
Appendix D. Multi-Branch Analytical EOS Source Code	102
E Bushman-Lomonosov EOS Source Code	112
Appendix E. Bushman-Lomonosov EOS Source Code	112

# List of Figures

1.1	Experimental Hugoniot Data from References [5-28] . . . . .	3
1.2	Experimental Release Data from References [21,29] . . . . .	4
1.3	Nellis Mie-Grüneisen Reshock Results from Reference [30] . . . . .	5
1.4	Maxwell Construct for 5000 K Isotherm . . . . .	9
2.1	Tillotson EOS phase regions . . . . .	13
2.2	Hugoniot and Release Isentropes Results from Tillotson [33] . . . . .	15
2.3	MBEOS Phase Space from Reference [3] . . . . .	16
2.4	$U_s-u_p$ Hugoniots from (a) Bushman [1] and (b) Lomonosov [35] . . . . .	21
2.5	Release Isentropes Results from Lomonosov [35] . . . . .	22
2.6	Saturation Domes from (a) Bushman [1] and (b) Lomonosov [35] . . . . .	23
3.1	Zero Kelvin Mie-Grüneisen Hugoniot Plots. Experimental data from References [5-28] . . . . .	27
3.2	Room Temperature Mie-Grüneisen Hugoniot Plots. Experimental data from References [5-28] . . . . .	29
3.3	Mie-Grüneisen Hugoniot Plots. Experimental data from References [5-28] . . . . .	31
3.4	Mie-Grüneisen Release Isentrope Results. Experimental data from References [21-29] . . . . .	32
3.5	Mie-Grüneisen Reshock Results. Experimental data from Reference [30].	33
3.6	Nellis Mie-Grüneisen Reshock Results. Experimental data from Nellis [30]. . . . .	34
3.7	Tillotson Hugoniot comparison with numerical data . . . . .	36
3.8	Tillotson release isentrope comparison with numerical data . . . . .	37
3.9	Tillotson Hugoniot Plots. Experimental data from References [5-28] . . . . .	38
3.10	Tillotson EOS Grüneisen parameter . . . . .	39
3.11	Tillotson Release Isentrope Results. Experimental data from References [21-29] . . . . .	40
3.12	Tillotson Reshock Results. Experimental data from Reference [30]. . . . .	40
3.13	MBEOS Hugoniot Plots. Experimental data from References [5-28] . . . . .	42
3.14	MBEOS Release Isentropes Results. Experimental data from References [21-29] . . . . .	43

---

3.15	MBEOS Phase Diagram Results. . . . .	43
3.16	MBEOS Reshock Results. Experimental data from Reference [30]. . .	44
3.17	Isotherms compared to Lomonosov Data . . . . .	47
3.18	Bushman-Lomonosov EOS Contributions . . . . .	47
3.19	Bushman-Lomonosov Hugoniot Plots. Experimental data from Refer- ences [5-28] . . . . .	48
3.20	Bushman - Lomonosov Release Isentrope Results. Experimental data from References [21-29] . . . . .	49
3.21	Bushman-Lomonosov Reshock Results. Experimental data from Ref- erence [30]. . . . .	50
3.22	Bushman-Lomonosov Phase Diagram Results . . . . .	51
4.1	$U_s-u_p$ Results . . . . .	53
4.2	$P-u_p$ Results . . . . .	53
4.3	$P-V$ Results . . . . .	54
4.4	Residual pressure along the principle Hugoniot . . . . .	55

## List of Tables

3.1	Aluminum Zero-Kelvin Mie-Grüneisen EOS parameters . . . . .	26
3.2	Aluminum Room Temperature Mie-Grüneisen EOS parameters . . . . .	28
3.3	Aluminum Mie-Grüneisen EOS parameters . . . . .	30
3.4	Aluminum Mie-Grüneisen EOS parameters from Reference [30] . . . . .	33
3.5	Aluminum Tillotson EOS parameters . . . . .	35
3.6	Aluminum MBEOS parameters . . . . .	41
3.7	Aluminum Bushman-Lomonosov EOS parameters . . . . .	46



# Chapter 1

## Introduction

The purpose of this study is to examine how different equations of state (EOSs) perform in the shock regime. The EOSs will be evaluated against available experimental data to determine their performance. The experimental data includes shock Hugoniot, release isentropes, reshock experiments and phase diagrams. This study includes investigations into complete and incomplete EOSs. Here incomplete EOSs refer to an EOS that requires an additional relationship to fully describe the thermodynamic state. The simulated EOS results were generated with a Fortran code written specifically for this project. This study will focus on aluminum, but methods presented throughout can be applied to most metallic materials.

### 1.1 Shock Physics

An understanding of the behavior of materials under high-strain rates is necessary to grasp the EOS concepts as they relate to shock physics. While not the focus of this research, a short review of shock physics follows. For a more in-depth overview of

shock physics see references [1–4].

Shock waves represent a rapid change between the undisturbed state and the shock excited state. Conservation of mass, momentum, and energy are performed across the shock wave. The relationships derived from the conservation equations are commonly referred to as the Rankine-Hugoniot or jump equations shown below.

$$\frac{U_s - u_{p0}}{V_1} = \frac{U_s - u_{p1}}{V_0} \quad (1.1)$$

$$P_1 - P_0 = \frac{U_s u_{p1}}{V_0} \quad (1.2)$$

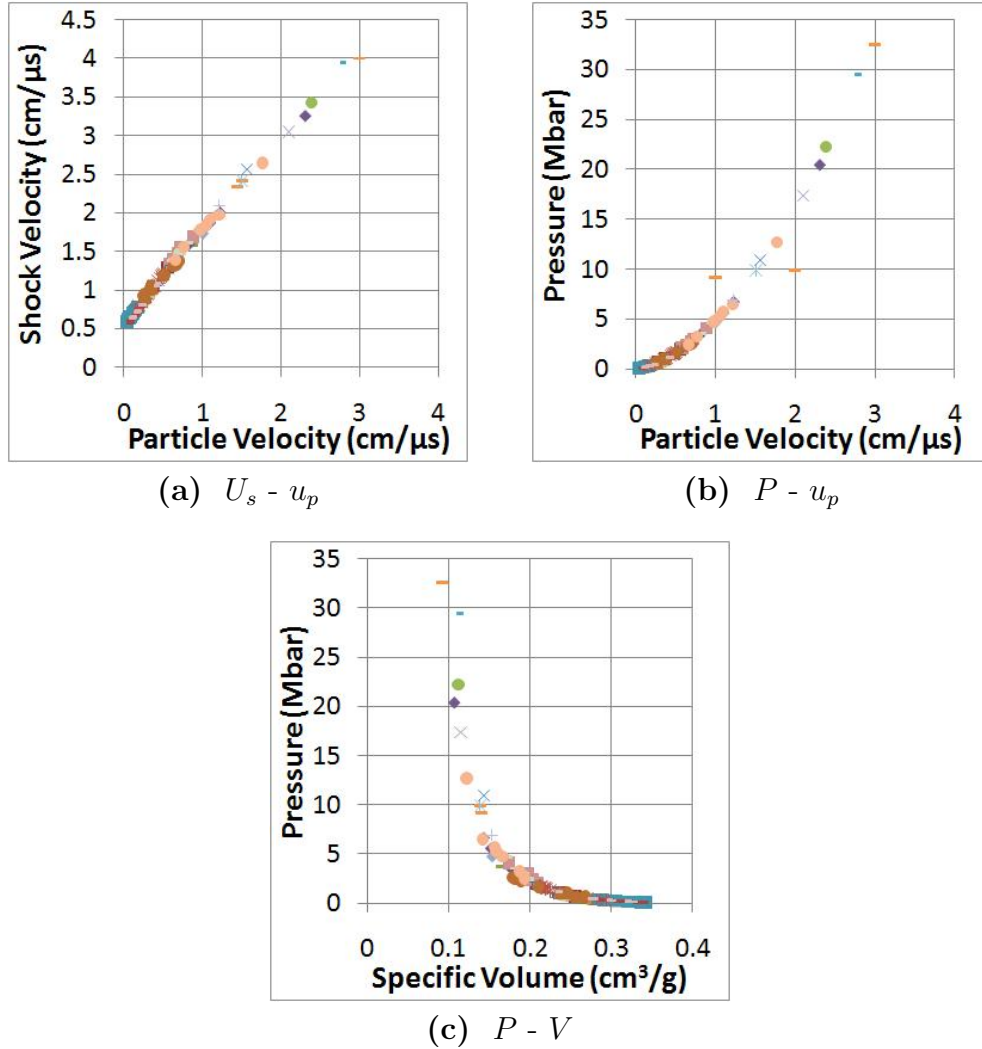
$$E_1 - E_0 = \frac{1}{2}(P_1 + P_0)(V_0 - V_1) \quad (1.3)$$

Along the principle Hugoniot  $P_0$ ,  $E_0$ , and  $u_{p0}$  are zero and  $V_0$  is the initial known density. This leaves three equations with four unknowns ( $u_p$ ,  $U_s$ ,  $P$ ,  $V$ ). A fourth relationship is necessary to solve for the shock state. A common assumption is  $U_s = f(u_p)$ . This relationship represents the locus of possible shock states, it is not a path-dependent process. The  $U_s$ - $u_p$  relationship is commonly determined through extensive experimental studies. A common linear fit to the  $U_s$ - $u_p$  is shown in Equation 1.4

$$U_s = C_0 + S_1 u_p \quad (1.4)$$

Only the three conservation equations and the particle-shock velocity relationship are needed to begin solving problems. This is the foundation to solve shock physics problems. At this point it is helpful to manipulate the above equations to form a series of curves. The three common Hugoniot curves are Shock Velocity ( $U_s$ ) - Particle Velocity ( $u_p$ ), Pressure ( $P$ ) - Particle Velocity ( $u_p$ ), and Pressure ( $P$ ) - Specific Volume ( $V$ ). Experimental shock Hugoniots from references [5–28] are shown in Figure

1.1.



**Figure 1.1** Experimental Hugoniot Data from References [5-28]

Analytical solutions to  $P-u_p$  and  $P-V$  Hugoniot curves can also be found by inserting the  $U_s - u_p$  relationship into the conservation equations.  $P-u_p$  and  $P-V$  solutions are shown in Equations 1.5 and 1.6

$$P = \rho_0 c_0 u_{p1} + \rho_0 s u_{p1}^2 \quad (1.5)$$

$$P = \frac{c_0^2 (V_0 - V)}{[V_0 - s_1 (V_0 - V)]^2} \quad (1.6)$$

Another important matrix in shock studies are rarefaction waves or release waves. Once a shock wave reaches a free surface, pressure returns to atmospheric. This sends a release wave back through the shocked material. A common experimental technique is to measure the velocity response of the free-surface, gaining insight to the release behavior. Also under sufficient loading, the material may release to a different material state. It is not uncommon to observe vaporization or melting. Experimental release data is shown in Figure 1.2 [21,29]. The lines in Figure 1.2 do not represent the release path but merely connect the shocked and released states.

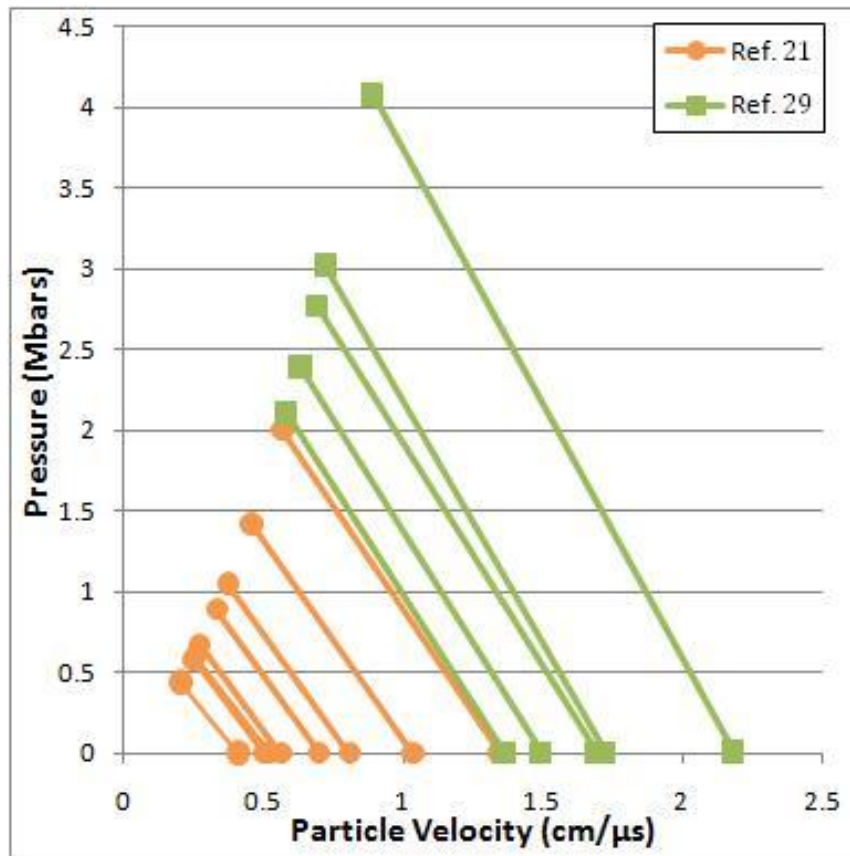


Figure 1.2 Experimental Release Data from References [21,29]

In addition to shock Hugoniot and release isentropes, reshock experimental data is compared. Experimental data from Nellis et. al [30] is shown in Figure 1.3. A reshock experiment is a common shock experiment where a second compression wave occurs after the material undergoes an initial shock. The reshock is commonly a result of impedance increase at a material interface.

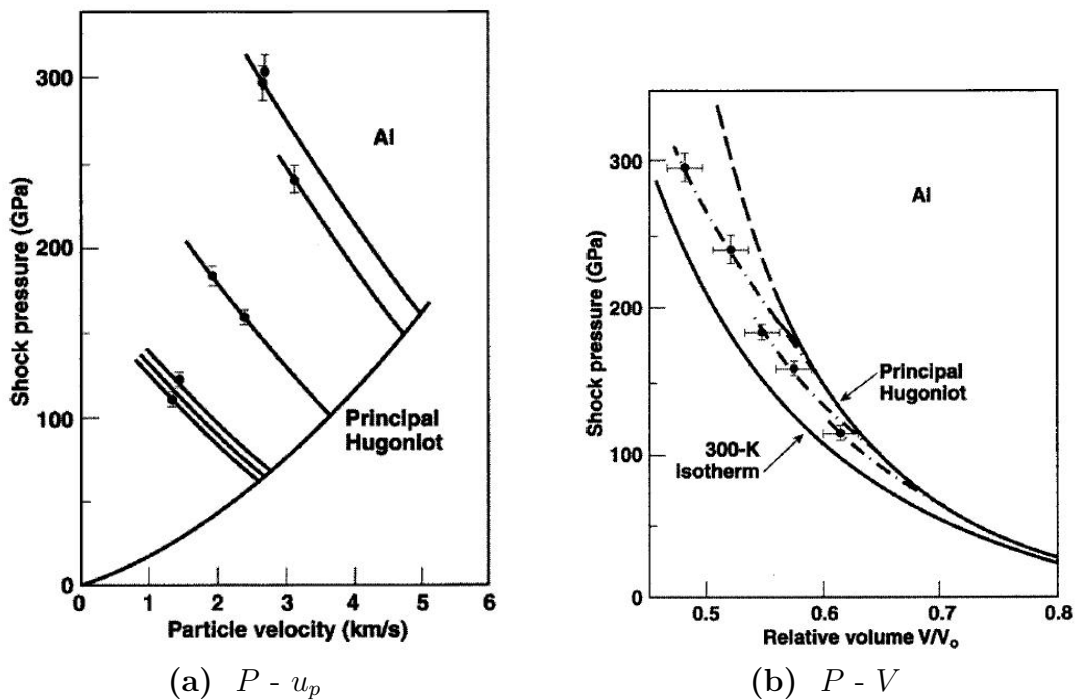


Figure 1.3 Nellis Mie-Grüneisen Reshock Results from Reference [30]

### 1.1.1 Hugoniot Calculations

In this work the Hugoniot relationships are calculated using the EOS and the conservation equations. The EOS results are then compared to the available experimental data. First it is assumed that the material is initially at normal conditions, i.e.  $P_0 = 0$ ,  $E_0 = 0$ , and  $u_p = 0$ , and these conditions correspond to the principle

Hugoniot. Through some manipulation the conservation of mass and momentum, Equations 1.1 and 1.2 take the form below:

$$U_s = \frac{V_0}{V_0 - V} u_p \quad (1.7)$$

$$P = \frac{u_p^2}{V_0 - V} \quad (1.8)$$

$$u_p^2 = P(V_0 - V) \quad (1.9)$$

Using Equation 1.9 and Equation 1.3, energy can be rewritten in terms of particle velocity shown in Equation 1.10.

$$E = \frac{u_p^2}{2} \quad (1.10)$$

Now pressure and energy are expressed as a function of particle velocity and specific volume. The Hugoniot states are found by substituting Equation 1.10 into the EOS and finding where  $P_{EOS}$  and  $P$  from the Hugoniot relationships are the same, shown in Equation 1.11.

$$\begin{aligned} P_{EOS} - P &= 0 \\ P_{EOS} - \frac{u_p^2}{V_0 - V} &= 0 \end{aligned} \quad (1.11)$$

### 1.1.2 Release Isentrope Calculations

The release process is commonly modeled isentropic. Unlike Hugoniots this process is path specific. The isentrope is calculated starting with the 2nd law of thermodynamics, Equation 1.12.

$$TdS = dE - PdV \quad (1.12)$$

By definition on an isentrope the change in entropy is zero; i.e.  $TdS = 0$ . Equation 1.12 is numerically differentiated and rearranged to Equation 1.13.

$$P = \frac{dE}{dV} = \frac{\Delta E}{\Delta V} \quad (1.13)$$

$$\frac{E_{i+1} - E_i}{V_{i+1} - V_i} = P_i \quad (1.14)$$

$$E_{i+1} = E_i + P_i (V_{i+1} - V_i) \quad (1.15)$$

In Equation 1.15 specific volume and pressure are assumed known and energy is calculated. If the path is reversed, this method can be used to calculate isentropic compression results. The free-surface velocity is calculated using Equation 1.16

$$u_{FS} = u_H - \int_{V_o}^V \left( -\frac{\partial P}{\partial V} \right)_S^{1/2} dV \quad (1.16)$$

### 1.1.3 Reshock Calculations

The reshock calculations are broken into two parts, the initial shock and the reshock. The initial shock is calculated using the same methods described in the principle Hugoniot calculation section. From the initial shock the reshock is calculated using the  $u_p$ ,  $P_1$ , and  $E_1$  from the initial shock as the starting state for the reshock. The reshock state is found using a method similar to the principle Hugoniot calculation but where the initial pressure, energy, and particle velocity are not equal to zero. Equations 1.17-1.20 are the full conservation equations used to calculate the reshock state.

$$(U_s - u_0) = (u_1 - u_0) \frac{\rho_1}{\rho_1 - \rho_0} \quad (1.17)$$

$$P_1 - P_0 = \rho (U - u_0) (u_1 - u_0) \quad (1.18)$$

$$P_1 = P_0 + \frac{(u_1 - u_0)^2}{(V_0 - V_1)} \quad (1.19)$$

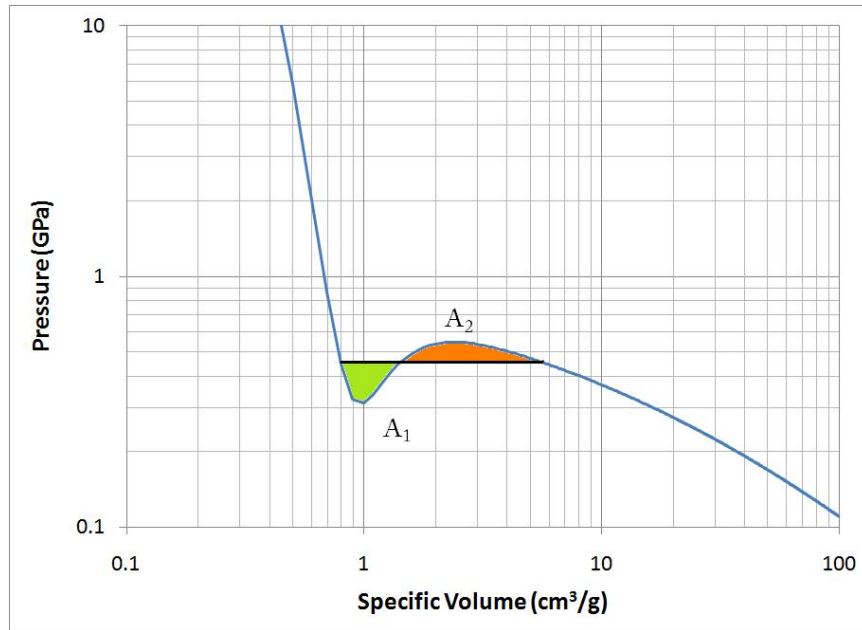
$$E_1 - E_0 = \frac{1}{2} (P_0 + P_1) (V_0 - V_1) \quad (1.20)$$

### 1.1.4 Multiphase Calculations

Some models include the necessary physics to model the phase change in the material. There are several different methods to accomplish this. First there are empirical relationships fitted to the multiphase boundaries. A common example is a Clausius-Clapeyron fit. This fit represents the saturation dome. The fit is commonly done independent of the EOS. The Linderman law is also an example of an empirical relationship. The Linderman law is a pressure-volume relationship used in calculating the melting transitions.

A more accurate approach to calculate the saturation region are Maxwell constructs [31]. Along a given isotherm the saturation pressure is found at the location where the area under and above the line are the same, shown in Figure 1.4 for the 5000 K isotherm. This method is used commonly in conjunction with a Van der Waals EOS.





**Figure 1.4** Maxwell Construct for 5000 K Isotherm

The final method for finding the stable state is using Gibb's free energy. For a detailed description on using Gibb's free energy for phase equilibrium see Wark [32]. In melting and polymorphic phase transitions, the state with the lowest Gibb's Free energy is the stable state. The saturation dome is found along an isotherm where the endpoints of the saturation pressure have the same Gibb's free energy.

## Chapter 2

# Equation of State

Equation of states have long been used to study material behaviors. A simple example of an EOS is the ideal gas equation, which relates pressure, specific volume and temperature. As the case with the ideal gas EOS, if volume and temperature are known, then pressure can be solved. If any two thermodynamic variables are known, the remaining thermodynamic variables can be found. It is important to note that most equations of state have limits. One would not want to apply the ideal gas equation in a region close to the saturation dome, where the ideal gas assumptions are not valid. So EOSs have evolved to problem-specific forms. EOS have been used to describe the material at the center of the earth and the behavior on stars that are galaxies away.

The primary focus of this work is on equations of state that describe materials in the shock regime. This is further broken down into two types, complete and incomplete EOS. A complete EOS,  $F = f(T, V)$ , is one that describes the thermodynamic state of a material. An incomplete EOS usually requires a second relationship to fully describe the thermodynamic state. Many commonly used shock physics EOSs are incomplete. These EOSs commonly take the form  $P = f(E, V)$  and an addi-

tional relationship is needed to describe energy, temperature, and entropy. Typically  $dE = Cp(dT)$  and the second law of thermodynamics,  $TdS = dE$ , are used to complete the EOS.

## 2.1 Mie-Grüneisen EOS

The Mie-Grüneisen EOS is a commonly used EOS in shock physics. It accurately represents the behavior of the Hugoniot relationships under modest compressions. Classically, the Mie-Grüneisen EOS considers only the effects of the lattice and zero-temperature contributions. Pressure and energy are determined from a given reference state; this could be room temperature, zero temperature, the Hugoniot, or other relationships. A detailed mathematical description of the Mie-Grüneisen EOS can be found in Gathers [3]. The Mie-Grüneisen EOS is commonly presented in the form shown in Equation 2.2. The Grüneisen parameter,  $\gamma_G$ , is assumed to be only a function of specific volume. The relationship for calculating  $\gamma_G$  is given in Equation 2.1.  $E_R$  and  $P_R$  are the given reference curves.

$$\gamma_G = V \left( \frac{\partial P}{\partial E} \right)_V \quad (2.1)$$

$$P - P_R = \frac{\gamma_G}{V} (E - E_R) \quad (2.2)$$

In many cases the Hugoniot is used as the reference curve. Under these circumstances Equation 2.2 can be rewritten as Equation 2.3. In Equation 2.3  $\gamma_g/V$  is rewritten using the assumption given in Equation 2.4

$$P = \frac{\gamma_0}{V_0} E - P_H \left( 1 - \frac{\gamma_0}{2} \frac{V_0 - V}{V} \right) \quad (2.3)$$

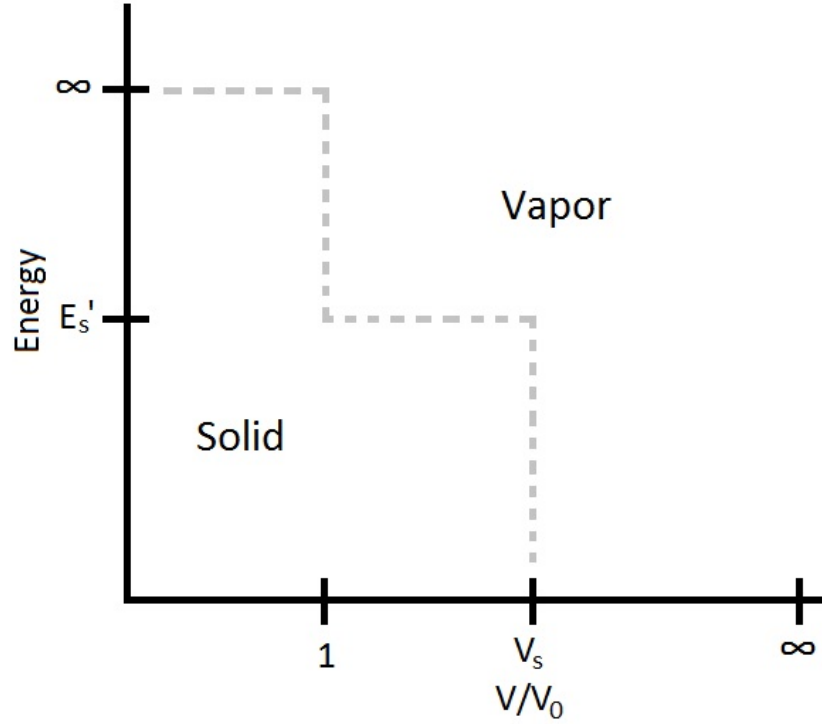
where

$$\frac{\gamma_G}{V} = \frac{\gamma_0}{V_0} = \text{Constant} \quad (2.4)$$

$$P_H = \frac{C_0^2 (V_0 - V)}{[V_0 - S (V_0 - V)]^2} \quad (2.5)$$

## 2.2 Tillotson EOS

The Tillotson EOS was developed to model hypervelocity impacts of metal materials [3, 33]. The Tillotson EOS divides the  $P$ - $V$  phase space into four regions. The first region is the small compression (Elastic Compression) region. This region was not considered in the Tillotson EOS but merely mentioned. Region II is strong compression (Shock waves) where  $V/V_0 < 1$  and  $E > 0$ . Region III is the release state from a strong compression but no phase change occurs, here  $V/V_0 < V_s$  and  $E < E'_s$ .  $E'_s$  is defined as the energy required to have "gas-like behavior". In Region IV the material is released from a strong enough shock state to create phase change in the material where  $1 < V/V_0 < V_s$  for  $E > E'_s$  and  $V/V_0$  for  $E > 0$ . The transition between the solid and the gas phase regions is shown in Figure 2.1



**Figure 2.1** Tillotson EOS phase regions

For regions II ( $V/V_0 < 1$  for  $E > 0$ ) and III ( $V/V_0 < V_s$  for  $E < E'_s$ )

$$P = \left( a + \frac{b}{\frac{E}{E_0 \eta^2} + 1} \right) \frac{E}{V} + A\mu + B\mu^2 \quad (2.6)$$

where:

$$\eta = \rho/\rho_0 = V_0/V$$

$$\mu = \eta - 1$$

$a, b, A, B$  are constants

The constants  $a, b,$  and  $A$  are fitted to a general EOS taking the form  $P = \gamma(E, E) + E/V + f(V)$ . The constant  $A$  is defined as  $A = C^2/V_0$  and the constants  $a$  and  $b$  follow the conditions  $a + b = \gamma_0$ . The remaining constant,  $B,$  is reserved to

adjust the EOS to best fit the thermodynamic  $P - V - E$  surface.

For region IV ( $1 < V/V_0 < V_s$  for  $E > E'_s$  and  $V/V_0 > V_s$  for  $E > 0$ )

$$P = aE\rho + \left\{ \frac{bE\rho}{\frac{E}{E_0\eta^2} + 1} + A\mu e^{-\beta[(V/V_0)-1]} \right\} e^{-\alpha[(V/V_0)-1]^2} \quad (2.7)$$

where:

$\alpha$  and  $\beta$  are constants for ideal gas convergence

The Tillotson EOS formulation allows for the treatment of vaporization and other physics-driven changes in the material response, without a large increase in mathematical complexities, but no considerations are made about the melting process or the liquid phase. The material is either all solid or all vapor in this approach.

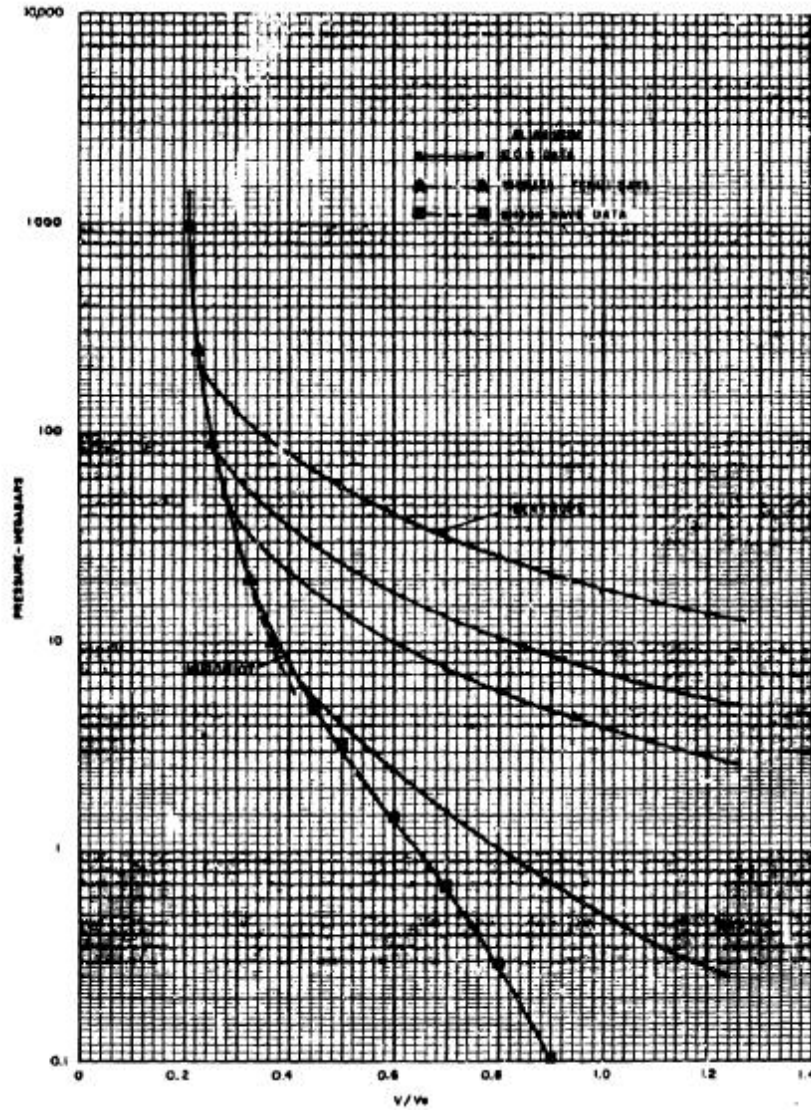


Figure 2.2 Hugoniot and Release Isentropes Results from Tillotson [33]

## 2.3 Multi-Branch Analytical EOS

The Multi-branch Analytic EOS (MBEOS) [3, 34] is a multi-phase EOS originally developed for Lithium. The EOS is broken into three regions,  $\rho \geq \rho_0$ ,  $\rho \leq \rho_0$ ,  $E > E_c$ , and  $\rho \leq \rho_0$ ,  $E < E_c$ . Where  $E_c$  is the critical point energy on the saturation dome.

Figure 2.3 shows how the regions are divided up over  $\rho$ - $E$  space.

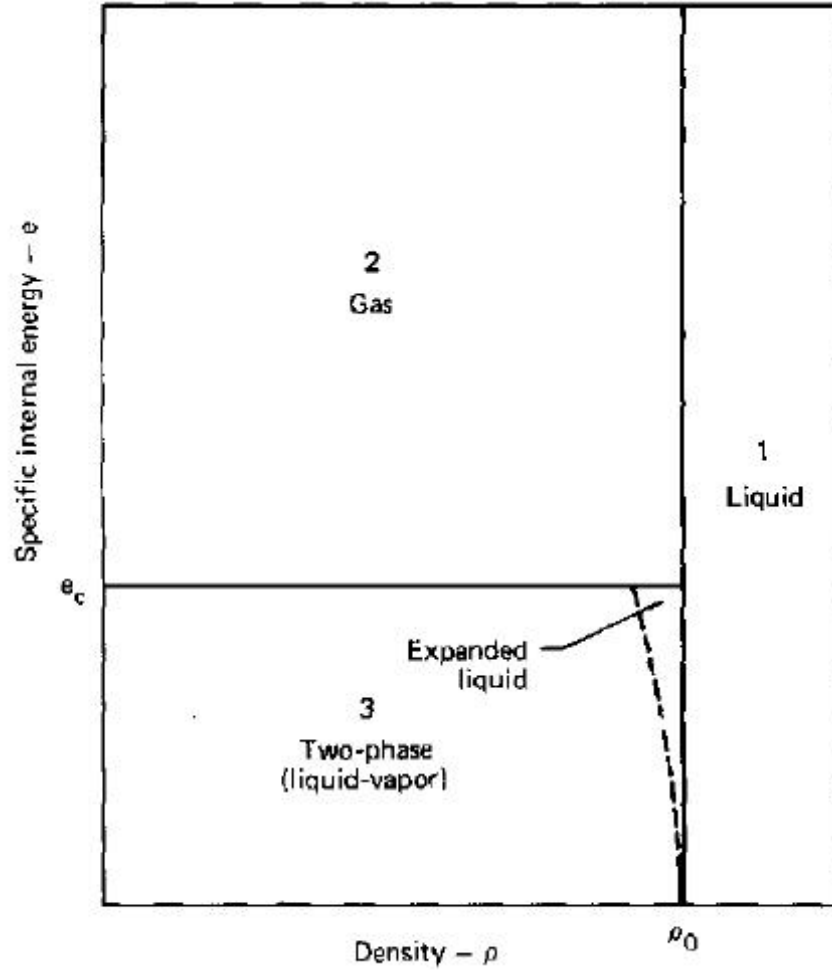


Figure 2.3 MBEOS Phase Space from Reference [3]

Region I:  $\rho \geq \rho_0$

This region is the solid phase response, it is represented by the Mie-Grüneisen EOS arranged in a slightly different form than previously shown.

$$P = P_H \left(1 - \frac{\gamma_0}{2} \eta\right) + \gamma_0 \rho_0 E \quad (2.8)$$



where:

$$P_H = \frac{\rho_0 c_0^2 \eta}{(1 - s\eta)^2} \quad \eta = 1 - \frac{\rho_0}{\rho} \quad (2.9)$$

Region II:  $\rho \leq \rho_0, E > E_c$

This region lies well above the critical point, such that an ideal gas assumption and ionization is considered. The ionization term,  $g(E)$ , models the dissociation of the electron shell surrounding the Lithium nucleus.

$$P(\rho, E) = \frac{2}{3}(E - E_c)\rho g(E) + \left[ \gamma_0 \rho_0 E - \frac{2}{3}(E - E_c)\rho_0 g(E) \right] (\rho/\rho_0)^k \quad (2.10)$$

$$g(E) = 1 - 0.7e^{[-2(y-1.233)^2]} - 0.7e^{[-0.5(y-4.468)^2]} \quad (2.11)$$

where :

$$y = \ln(E/E_c) \quad (2.12)$$

The constant  $k$  is fitted from tabular values in the expansion region. Region III:  $\rho \leq \rho_0, E < E_c$

Region III is the region that contains the saturation dome and expanded liquid state. In the expanded liquid region, the region between  $\rho_0$  and the saturation dome, is modeled using Equation 2.13. This allows for a smooth transition between the saturation dome and Region I. The saturation dome is determined through a Clausius-Clapeyron type fit, shown in Equation 2.13.  $P_c$  and  $E_c$  are the critical saturation pressure and energy. Constants  $m$  and  $n$  describe the saturation dome. For Lithium,  $m$  and  $n$  were fitted from tabulated data. On the vapor side of the saturation dome Region II and the saturation dome are joined using Equation 2.15.

$$P = \rho_0 C_0^2 (\rho/\rho_0 - 1) + \gamma_0 \rho_0 E \quad (2.13)$$

$$P = P_c \exp(m - nE_c/E) \quad (2.14)$$

$$P = \gamma_0 \rho_0 E (\rho/\rho_0)^k \quad (2.15)$$

## 2.4 Bushman-Lomonosov EOS

The Bushman-Lomonosov EOS originates from *Intense Dynamic Loading of Condensed Matter* by A.V. Bushman et. al. [1]. This model was later modified by Lomonosov [35]. The EOS is a complete EOS taking a free energy,  $F$ , approach. The EOS further separates free energy into three components; the cold compression curve ( $F_c$ ), atomic contributions ( $F_a$ ), and electronic contributions ( $F_e$ ). The cold and atomic contributions are further broken down into liquid and solid relationships.

$$F(V, T) = F_c(V) + F_a(V, T) + F_e(V, T) \quad (2.16)$$

The cold compression curve corresponds to the material behavior at the theoretical 0 K temperature state. Equation 2.17 behaves such that at  $E_c(V_{0c}) = 0$  and at  $E_c(V \rightarrow \infty) = E_s$ ,  $E_s$  is the sublimation energy. This assures that Equations 2.17 and 2.18 join smoothly. The value  $V_{0c}$  is the specific volume at zero pressure. The variable  $\sigma_c$  is the ratio of the specific volume at zero pressure over the current specific volume. The  $a_i$  variables are fitting constants for the solid phase cold curve.

$$F_c(V) = 3V_{0c} \sum_{i=1}^5 \frac{a_i}{i} (\sigma_c^{i/3} - 1) \quad (2.17)$$

The liquid region cold curve differs between Bushman and Lomonosov, shown in Equations 2.18 and 2.19. Both curves join with the solid curve at  $\sigma_c \geq 1$ . The parameters  $A, B, m, n, l$  in Equation 2.18 are fitting parameters. The parameters used

in Lomonosov's equation,  $A_c, B_c, C_c, m, n, l$ , are also fitting constants.

$$F_c(V) = V_{0c} \left[ A \left( \frac{\sigma_c^m}{m} - \frac{\sigma_c^l}{l} \right) + B \left( \frac{\sigma_c^n}{n} - \frac{\sigma_c^l}{l} \right) \right] + E_{sub} \quad (2.18)$$

$$F_c(V) = V_{0c} \left[ A_c \frac{\sigma_c^m}{m} + B_c \frac{\sigma_c^n}{n} + C_c \frac{\sigma_c^l}{l} \right] + E_{sub} \quad (2.19)$$

Atomic or lattice contributions are divided into two phases, solid and liquid. The solid phase represents the vibration of atoms in a crystal lattice. Both Bushman and Lomonosov use the same relationships to describe solid atomic contributions, shown in Equations 2.20 and 2.21. These equations are high temperature approximations of the Debye theory. In Equation 2.21,  $\Theta$  is an empirical relationship representing the characteristic temperature. The characteristic temperature represents vibrational frequency of the atoms in the lattice structure, for details see chapter 13 of Reference [36]. The variable  $B_s$  and  $D_s$  are constants fit to match the Grüneisen coefficient dependent on specific volume. Here  $\chi$  is defined by  $\chi = ln\sigma$ .

$$F_a(V, T) = 3RT \ln \{ \Theta(\sigma) / T \} \quad (2.20)$$

$$\Theta(\sigma) = \Theta_{0s} \sigma^{2/3} \exp \left\{ \frac{(\gamma_{0s} - \frac{2}{3})(B_s^2 + D_s^2)}{B_s} \tan^{-1} \left[ \frac{\chi B_s}{B_s^2 + D_s(\chi + D_s)} \right] \right\} \quad (2.21)$$

The atomic contributions in the liquid phases are shown in the below equations. Bushman and Lomonosov use the same relationship with the exception of the calculation of the characteristic temperature,  $\Theta$ . The liquid atomic contributions are broken into two parts,  $F_t$  and  $F_m$ . The anharmonicity of the atoms at high temperatures is represented in the  $F_t$  term.  $F_m$  models the melting curve and the liquid-state near the the melting curve.  $F_m$  also models the change in density that occurs during melting.

$$F_a(V, T) = F_t(V, T) + F_m(V, T) \quad (2.22)$$

The terms that describe the anharmonicity at high temperature are shown in the Equation 2.23. The leading terms,  $3R/2$ , ensure the liquid model asymptotes to the ideal gas assumptions. The variables  $T_{sa}$ ,  $T_{ca}$ ,  $B_l$ , and  $D_l$  are parameters determined from shock compression experiments. The variable  $\Theta_{0l}$  is found from  $\Theta_a(0) = T_{ca}$ . The difference between the Bushman, Equation 2.24, and Lomonosov, Equation 2.25, methods are the addition of the  $T_{ca}\Theta$  to  $\Theta$  in the Lomonosov method.

$$F_t(V, T) = \frac{3RT}{2} \left[ 1 + \frac{\sigma T_T}{(\sigma + \sigma_T)(T + T_T)} \right] \ln \left\{ \frac{\Theta(\sigma, T)}{T} \right\} \quad (2.23)$$

$$\Theta(\sigma, T) = T_{sa} \sigma^{2/3} \frac{\Theta_a(\sigma) + T}{T_{ca} + T} \quad (2.24)$$

$$\Theta(\sigma, T) = T_{sa} \sigma^{2/3} \frac{T_{ca} \Theta_a(\sigma) + T}{T_{ca} + T} \quad (2.25)$$

$$\Theta_a(\sigma) = \Theta_{0l} \exp \left\{ \frac{(\gamma_{0l} - \frac{2}{3})(B_l^2 + D_l^2)}{B_l} \arctan \left[ \frac{\chi B_l}{B_l^2 + D_l(\chi + D_l)} \right] \right\} \quad (2.26)$$

The remaining liquid atomic contributions are the melting effect. Both Bushman and Lomonosov use the same form, shown in Equation 2.27. The relative density of the liquid phase is defined as  $\sigma_l = \sigma/\sigma_m$ . The constants  $\sigma_m$  and  $T_m$  are the specific volume ratio and temperature of melting at standard pressure.

$$F_m(V, T) = 3R \left\{ \frac{2\sigma_l^2 T_m}{1 + \sigma_l^3} \left[ C_m + \frac{3A_m}{5} (\sigma_l^{5/3} - 1) \right] + (B_m - C_m) T \right\} \quad (2.27)$$

The last component to the Bushman and Lomonosov models is the electronic contribution. Bushman and Lomonosov use the same relationships, shown in Equations 2.28-2.31. The electronic contributions are the same for the liquid and solid phases. At high temperatures, the electronic heat capacity,  $\beta$ , asymptotes to  $3RZ/2$  which corresponds to an ideal electron gas with complete ionization.

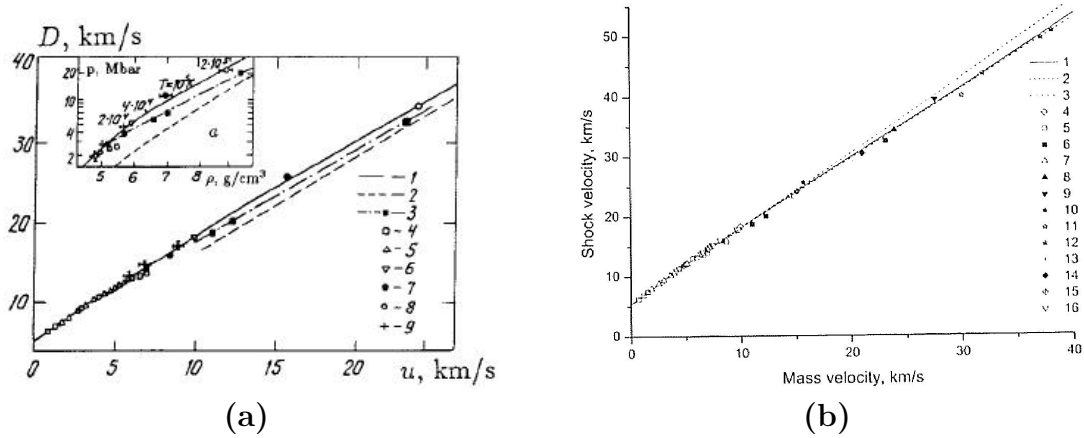
$$F_e(V, T) = -c_e(\sigma, T) T \ln \left\{ 1 + \frac{2\sigma^{-\gamma_e(\sigma, T)}}{3RTZ} \int^T \int^\tau \beta(\tau) d\tau dT \right\} \quad (2.28)$$

$$c_e(V, T) = \frac{3R}{2} \left[ Z + \frac{\sigma_Z T_Z^2 (1 - Z)}{(\sigma + \sigma_Z)(T^2 + T_Z^2)} \exp \left\{ -\frac{\tau_i(\sigma)}{T} \right\} \right] \quad (2.29)$$

$$\beta(T) = \beta_i + \left( \beta_0 - \beta_i + \beta_m \frac{T}{T_b} \right) \exp \left\{ -\frac{T}{T_b} \right\} \quad (2.30)$$

$$\gamma_e(\sigma, T) = \frac{2}{3} + \left( \gamma_0 - \frac{2}{3} + \gamma_m \frac{T}{T_g} \right) \exp \left\{ -\frac{T}{T_g} - \frac{(\sigma - \sigma_e)^2}{\sigma \sigma_d} \right\} \quad (2.31)$$

Both Bushman and Lomonosov generated a wide array of results. Figure 2.4 are  $U_s$ - $u_p$  Hugoniot results. Both models show good agreement with the presented experimental data. This includes the high pressure regions where melting may occur.



**Figure 2.4**  $U_s$ - $u_p$  Hugoniots from (a) Bushman [1] and (b) Lomonosov [35]

Lomonosov compared free surface velocity measurements in Figure 2.5. The model does a good job at lower free surface velocity but diverges significantly at higher values. The end result differs by approximately 10%.

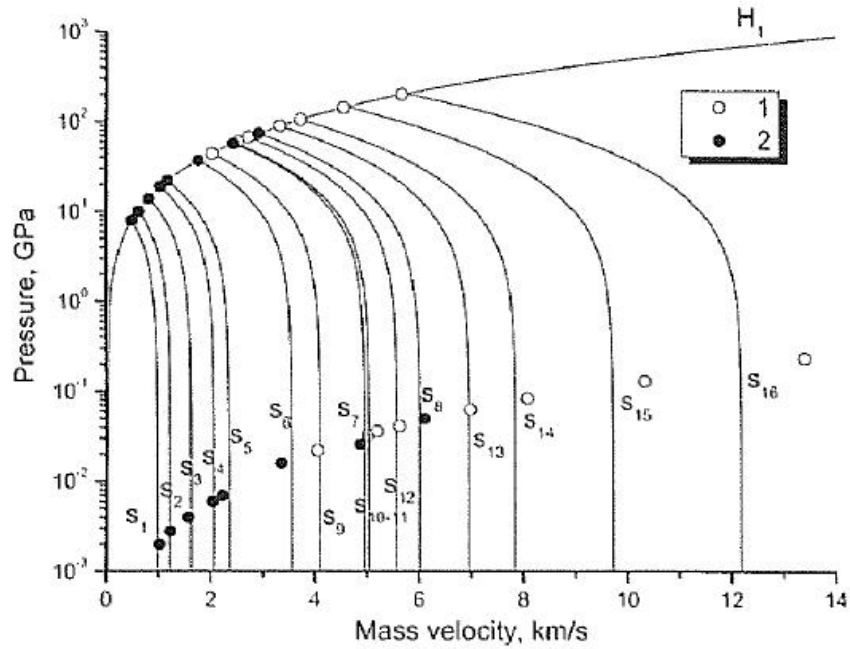


Figure 2.5 Release Isentropes Results from Lomonosov [35]

Both Bushman and Lomonosov made estimations of the critical properties. Bushman reported a critical pressure of  $0.571 \text{ GPa}$ , critical temperature  $7222 \text{ K}$ , and a critical specific volume of  $1.24 \text{ cm}^3/\text{g}$ . Lomonosov predicted a critical pressure of  $0.197 \text{ GPa}$ , critical temperature  $5520 \text{ K}$ , and a critical specific volume of  $1.423 \text{ cm}^3/\text{g}$ . The accompanying saturation domes are shown in Figure 2.6.

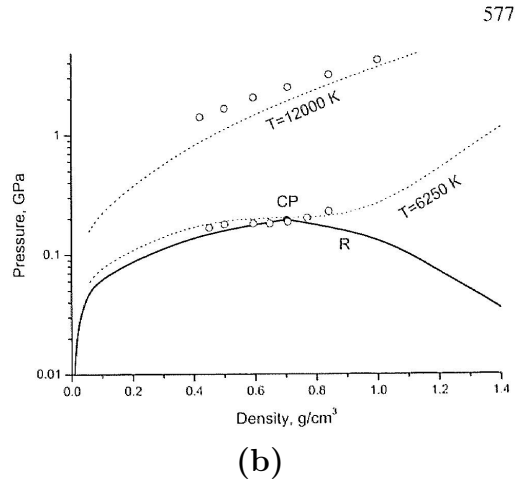
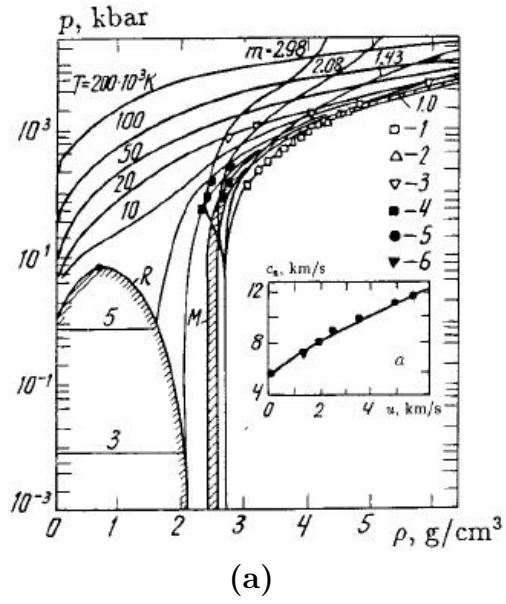


Figure 2.6 Saturation Domes from (a) Bushman [1] and (b) Lomonosov [35]

## Chapter 3

### Results

Several EOSs were simulated in this project, starting with the Mie-Grüneisen and Tillotson EOSs and moving to the more complex Multi-Branch and Bushman-Lomonosov EOSs. The Mie-Grüneisen, Tillotson, and Multi-Branch EOSs are all incomplete EOS while Bushman-Lomonosov EOS is a complete EOS.

The Mie-Grüneisen, Tillotson EOS, and the Multi-Branch EOS also used a shared program using the same routines to calculate Hugoniot and isentropes. Each EOS had its own module that included routines for initializing and calculating the thermodynamic state using the given inputs. The Bushman-Lomonosov EOS was considerably more complex. It was more straightforward to create an independent program to model the Bushman-Lomonosov EOS. The source code for both programs are given in the accompanying appendixes.



### 3.1 Mie-Grüneisen EOS

The first EOS modeled is the Mie-Grüneisen (MGR) equation of state. The Mie-Grüneisen was chosen as the starting point because it is a commonly used EOS in shock physics. As mentioned earlier the Mie-Grüneisen, given in Equation 3.1, uses a reference curve to model the pressure-volume space. This reference curve can take many forms but this work will look at a zero-Kelvin fit, room temperature fit, and the shock Hugoniot. One of the drawbacks to using the MGR EOS is that phase change is not modeled. Only the cold and vibrational contributions are included in the model.

$$P - P_R = \frac{\gamma_G}{V} (E - E_R) \quad (3.1)$$

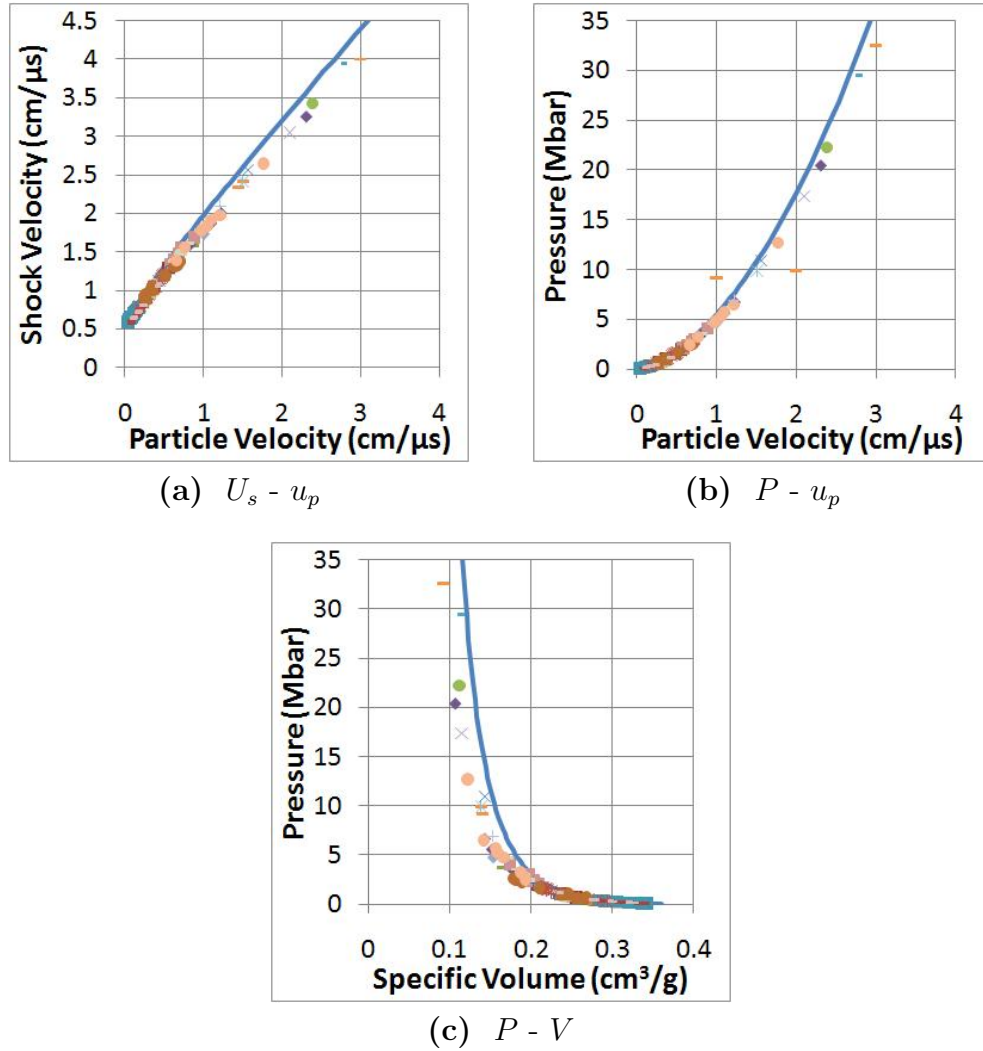
The zero-Kelvin fit used in this work is taken from the Bushman-Lomonosov EOS. The Mie-Grüneisen EOS was generated using the cold curve shown in Equation 3.2 and using the parameters listed in Table 3.1 taken from Reference [35]. The curve is constrained so that the energy and pressure are zero at  $V_{0c}$ . The pressure term is calculated by the  $P_c(V) = -\frac{dE_c(V)}{dV}$  relationship. The variable  $\sigma_c$  is the ratio of the specific volume at zero pressure over the current specific volume. This relationship is used as the reference curve in the Mie-Grüneisen EOS

$$E_c^s(V) = 3V_{0c} \sum_{i=1,5} \frac{a_i}{i} (\sigma_c^{i/3} - 1) \quad (3.2)$$

**Table 3.1** Aluminum Zero-Kelvin Mie-Grüneisen EOS parameters

$V_0$ $cm^3/g$	$\gamma_0$	$a_1$	$a_2$	$a_3$	$a_4$	$a_5$
0.3614	2.19	326.35	-1035.44	858.51	-160.59	11.17

Results of zero-Kelvin Mie-Grüneisen EOS Hugoniot calculations are shown in Figure 3.1. The EOS does a reasonable job predicting the Hugoniot behavior.



**Figure 3.1** Zero Kelvin Mie-Grüneisen Hugoniot Plots. Experimental data from References [5-28]

The next method was a room temperature fit. This fit was described by Wilkin in reference [37]. This method defines energy and pressure as zero at room temperature (300 K) and  $V/V_0 = 1$ . The reference curve for energy and pressure are shown in Equations 3.3 and 3.4. Parameters used in the room temperature fit are shown in Table 3.2. The parameters were calculated using Hugoniot fit and the equations given in Equation 3.5. Here  $x = 1 - V/V_0$ .

$$\epsilon_0 = \epsilon_{00} + \epsilon_{01}x + \epsilon_{02}x^2 + \epsilon_{03}x^3 + \epsilon_{04}x^4 \quad (3.3)$$

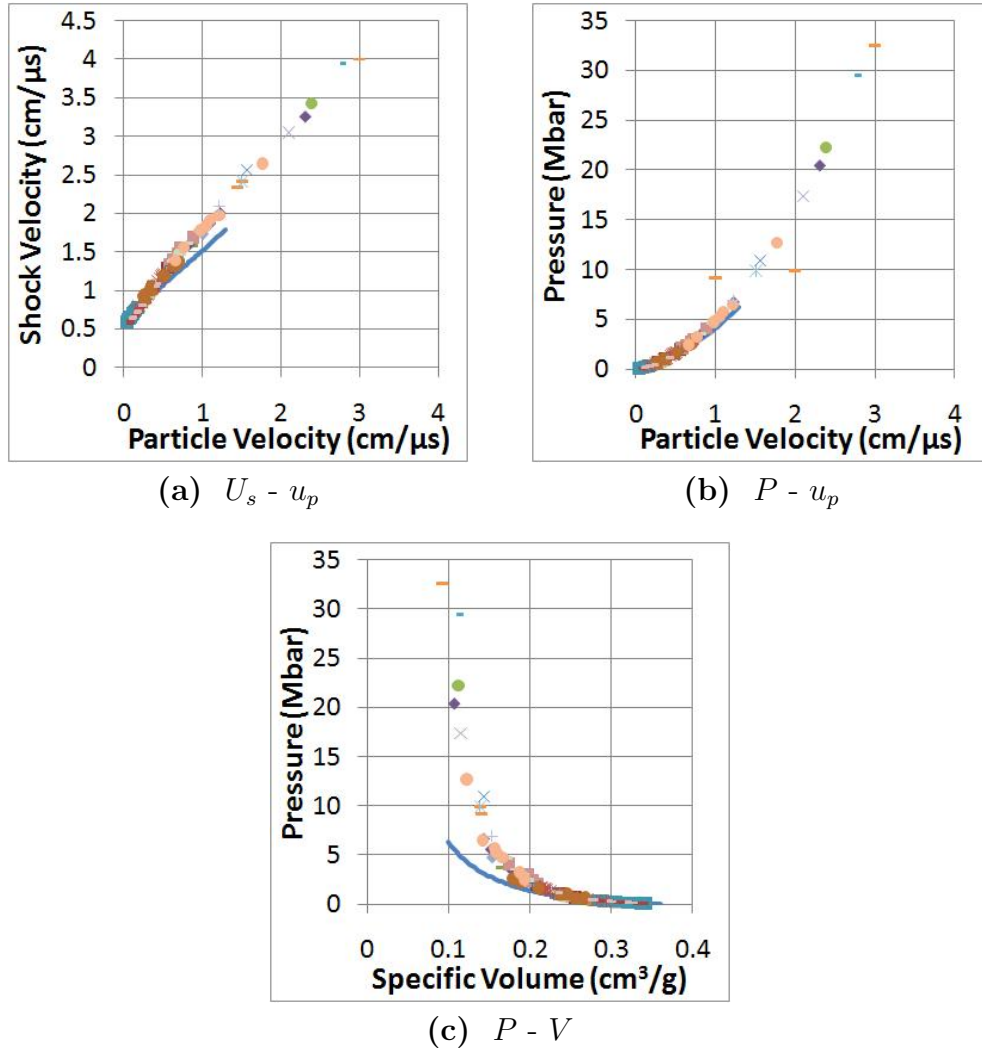
$$P_0 = -\frac{d\epsilon_0}{dV} = \frac{1}{V_0} (\epsilon_{01} + 2\epsilon_{02}x + 3\epsilon_{03}x^2 + 4\epsilon_{04}x^3) \quad (3.4)$$

$$\begin{aligned} e_0 &= e_{00} + e_{01}x + e_{02}x^2 + e_{03}x^3 + e_{04}x^4 \\ e_{00} &= -900R \\ e_{01} &= \gamma_0 e_{00} \\ e_{02} &= \frac{1}{2} (C + \gamma_0^2 e_{00}) \\ e_{03} &= \frac{1}{6} (4SC_0 + \gamma_0^3 e_{00}) \\ e_{04} &= \frac{1}{24} (-2\gamma_0 SC_0^2 + 18S^2 C_0^2 + \gamma_0^4 e_{00}) \end{aligned} \quad (3.5)$$

**Table 3.2** Aluminum Room Temperature Mie-Grüneisen EOS parameters

$\rho_0$ <i>cm</i> <sup>3</sup> / <i>g</i>	$\gamma_0$	$\epsilon_{00}$	$\epsilon_{01}$	$\epsilon_{02}$	$\epsilon_{03}$	$\epsilon_{04}$
2.712	2.14	-2.773e-3	-5.9342e-3	1.4226e-1	2.4498e-1	2.8429e-1

Results of room temperature Mie-Grüneisen EOS Hugoniot calculations are shown in Figure 3.2. The room temperature fit doesn't match the experimental data very well. It does give a reasonable match to low pressure region  $P < 1Mbar$ , but diverges from the experimental data at higher pressure.



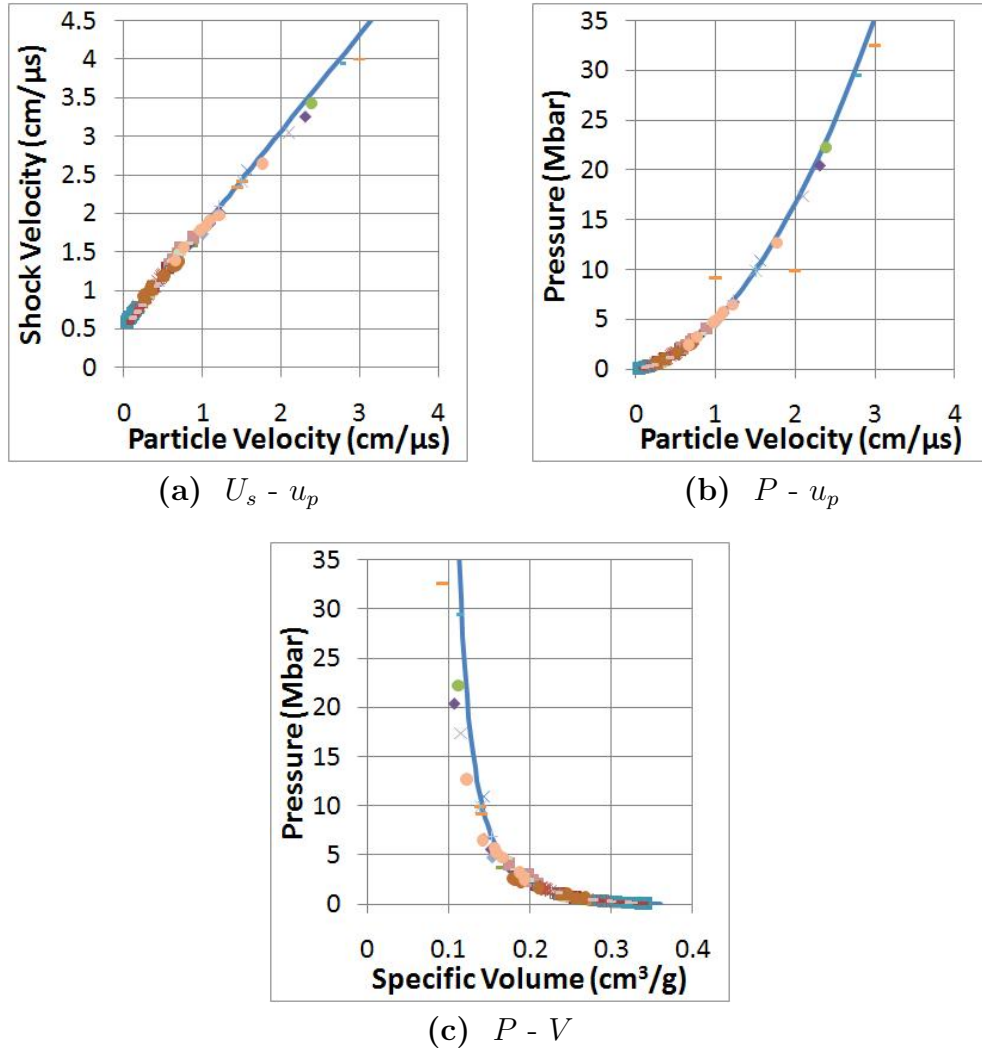
**Figure 3.2** Room Temperature Mie-Grüneisen Hugoniot Plots. Experimental data from References [5-28]

The final method is the Mie-Grüneisen EOS using the Hugoniot as a reference curve. This is a common method for calculating the Mie-Grüneisen EOS. A linear  $U_s - u_p$  was fitted to the experimental data and was used as the reference curve in the Mie-Grüneisen EOS. The parameters used in Mie-Grüneisen EOS are shown in Table 3.3.

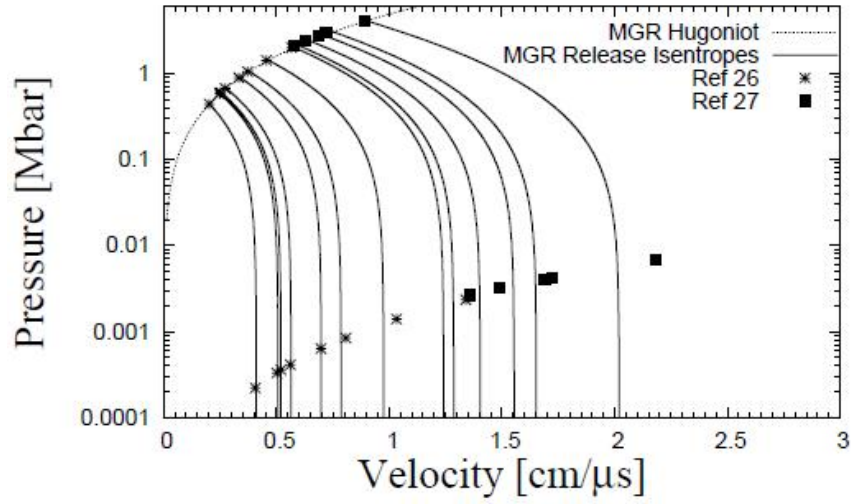
**Table 3.3** Aluminum Mie-Grüneisen EOS parameters

$\rho_0$ <i>g/cm<sup>3</sup></i>	$C_0$ <i>cm/<math>\mu</math>s</i>	$S$	$\gamma_0$
2.712	0.54518	1.2592	2.14

Results from the Hugoniot curve Mie-Grüneisen is shown in Figure 3.3. It gives good results matching the experimental principle Hugoniot. This is expected since the reference curve is the Hugoniot derived from the experimental data. The release isentropes, shown in Figure 3.4, match the low pressure states but don't do as well of a job at the higher states where phase change exists.



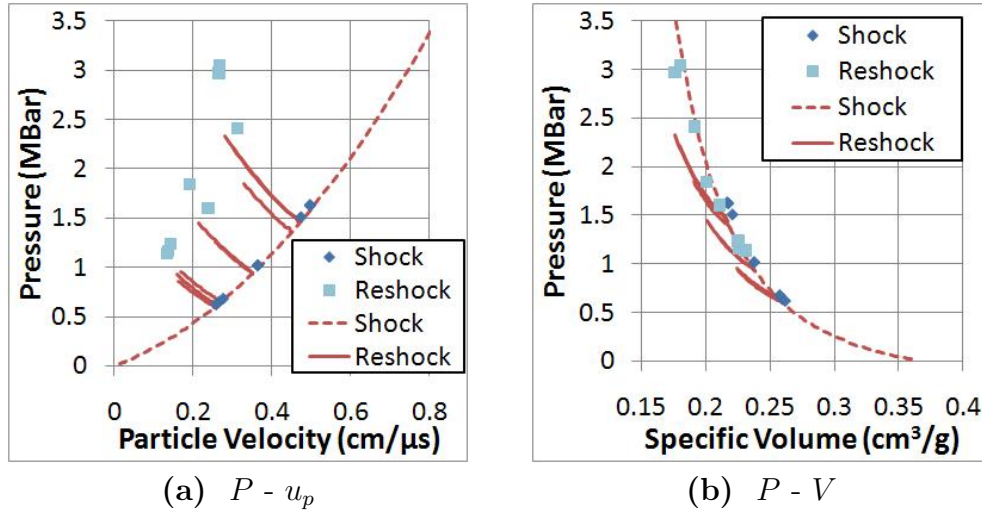
**Figure 3.3** Mie-Grüneisen Hugoniot Plots. Experimental data from References [5-28]



**Figure 3.4** Mie-Grüneisen Release Isentrope Results. Experimental data from References [21-29]

Reshock results were calculated using the method described in the reshock section, Section 1.1.3. In the Mie-Grüneisen type EOSs the Grüneisen,  $\gamma$ , strongly influences the off-Hugoniot behavior. Using the Hugoniot, parameters listed in Table 3.3 Hugoniot reshocks were calculated, shown in Figure 3.5.



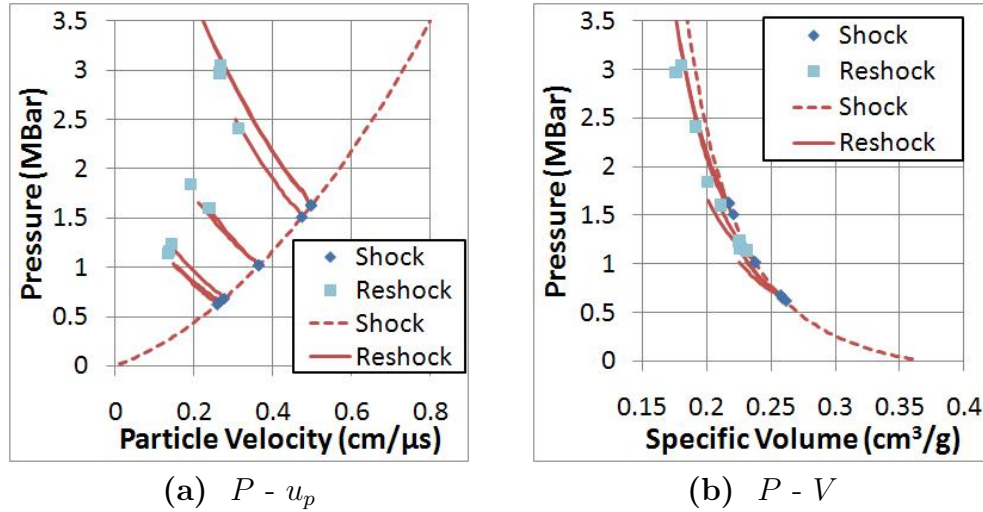


**Figure 3.5** Mie-Grüneisen Reshock Results. Experimental data from Reference [30].

In Reference [30], Nellis et. al., provided a  $U_s-u_p$  Hugoniot fit given in Table 3.4. Using these parameters the reshock experiments were recalculated. The resulting  $P-u_p$  and  $P-V$  curves are shown in Figure 3.6. The Nellis fit more accurately matches the experimental data. This is largely due to the difference in the Grüneisen parameter.

**Table 3.4** Aluminum Mie-Grüneisen EOS parameters from Reference [30]

$\rho_0$	$C_0$	$S$	$\gamma_0$
$g/cm^3$	$cm/\mu s$		
2.712	0.5386	1.339	1.35



**Figure 3.6** Nellis Mie-Grüneisen Reshock Results. Experimental data from Nellis [30].

## 3.2 Tillotson EOS

The Tillotson EOS was modeled using the parameters found in Table 3.5. These parameters were taken from Ref. [33]. Along with the EOS parameters, Tillotson also provided some additional numerical results for several metals. The numerical results consist of the Hugoniot and several release isentropes. First the results from Tillotson's [33] report were compared to the data calculated using the EOS code. This was done to verify that the Tillotson EOS is being modeled correctly.

**Table 3.5** Aluminum Tillotson EOS parameters

$a$	$b$	$A$ Mbar	$B$ Mbar	$E_0$ $cm^3/g$	$\alpha$	$\beta$	$E'_s$ $Mbarcm^3/g$	$V_s^a$ $cm^3/g$
0.5	1.63	0.752	0.65	0.05	5.0	5.0	0.03	1.1

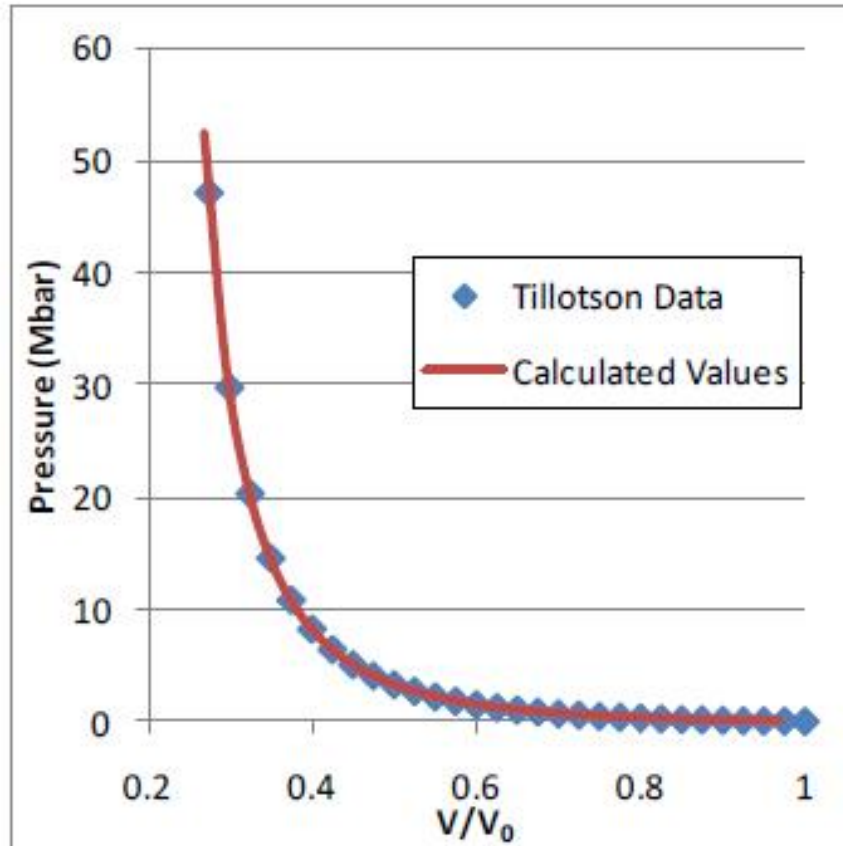
In Tillotson's report [33] a method for calculating the compression of materials and their release from high pressures was addressed. Tillotson reported on the principle Hugoniot and the release isentrope of several initial pressure/volume states. Calculation of the Hugoniot is straightforward and follows the approach demonstrated in Section 1.1.1.

To calculate the principle Hugoniot a Newton's solver was needed because particle velocity cannot be directly solved for in the compression portion of the Tillotson EOS shown in Equation 2.6. The Newton's solver, Equation 3.6, uses an initial guess,  $x_0$ , and calculates the desired function  $f$  and its derivative  $f'$ . A better guess of the correct answer is returned as  $x_1$ . This process is repeated until the function is close to the desired solution. In this particular problem the function we are solving for is  $f = P_{EOS} - P_{hugo}$ , where  $P_{EOS}$  is the pressure calculated by EOS using  $E = u_p^2/2$ .  $P_{hugo}$  is the pressure calculated using the Hugoniot relationships, i.e.  $P_{hugo} = u_p^2 / (V_0 - V_1)$ .

$$x_1 = x_0 - \frac{f(x_0)}{f'(x_0)} \quad (3.6)$$

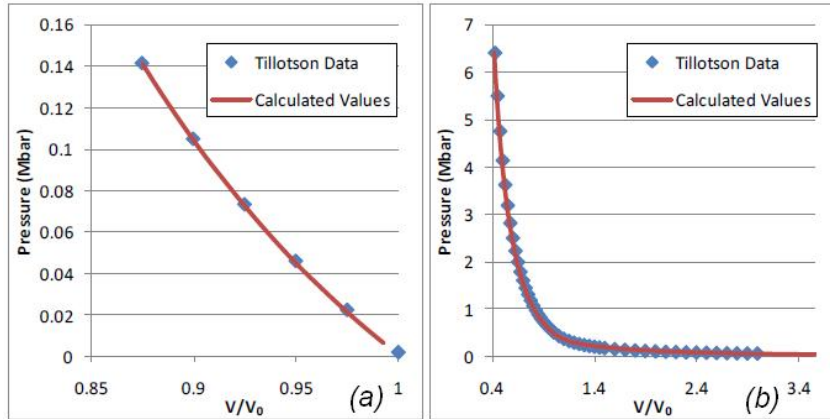
Figure 3.7 shows the comparison between the data calculated by Tillotson and what was generated during this work. The calculated values are in good agreement

with Tillotson's data, assuring that the model is being properly implemented.



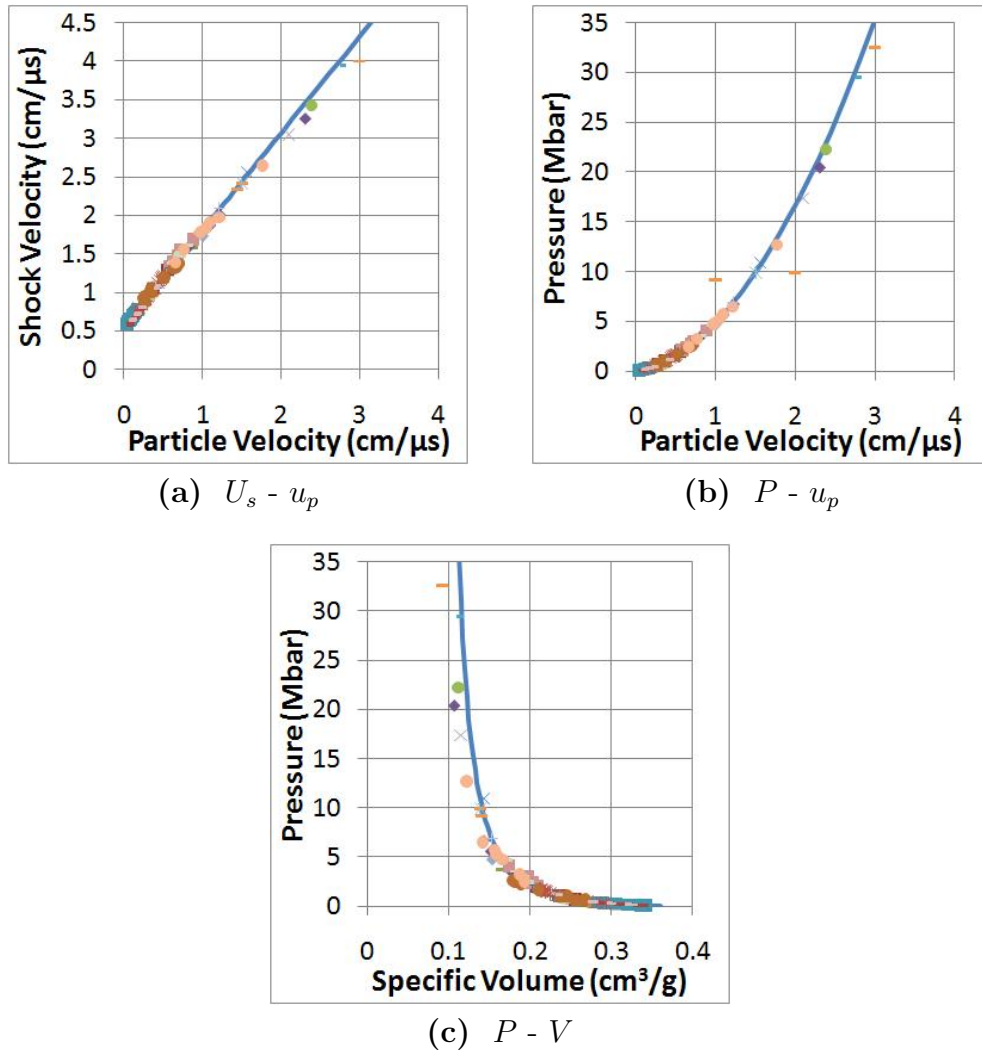
**Figure 3.7** Tillotson Hugoniot comparison with numerical data

In addition to the principle Hugoniot data, Tillotson provides seven release isentropes. The isentropes were found using the method described in Section 1.1.2. Two of the seven isentropes are shown in Figure 3.8. These two isentropes were chosen because one release exhibits no phase change, Figure 3.8a, while the other releases to the vapor phase, Figure 3.8b. The dramatic change in slope seen in Figure 3.8b is a result of the material transition from solid to vapor states.

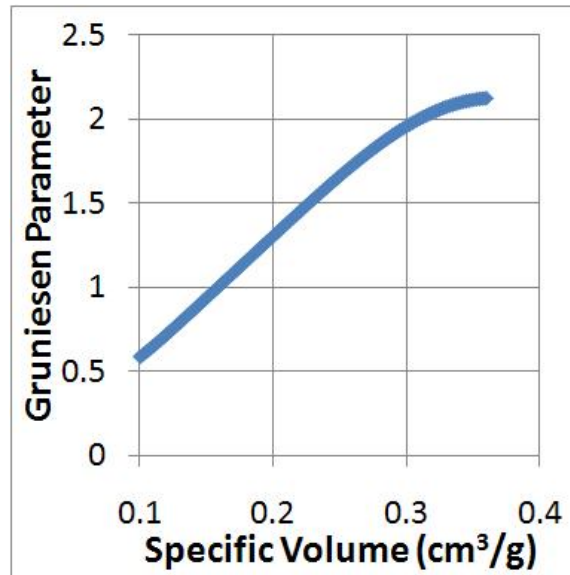


**Figure 3.8** Tillotson release isentrope comparison with numerical data

In addition to the numerical data provided by Tillotson, the EOS was compared to experimental Hugoniot and release isentrope data shown in Figure 3.9. In the Tillotson EOS the Grüneisen parameter isn't limited to the fixed ratio,  $\gamma/V = \gamma_0/V_0$ . The Grüneisen parameter in the Tillotson EOS is a function of energy and specific volume. The change of the Grüneisen parameter along the Hugoniot is shown in Figure 3.10

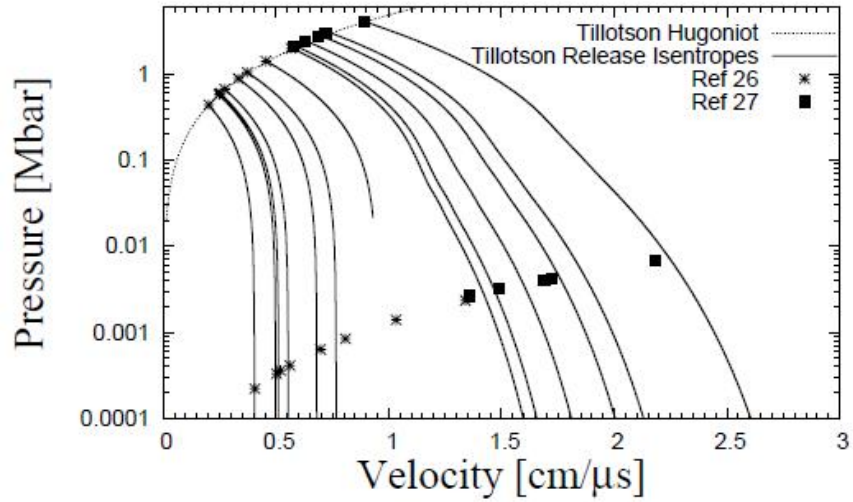


**Figure 3.9** Tillotson Hugoniot Plots. Experimental data from References [5-28]



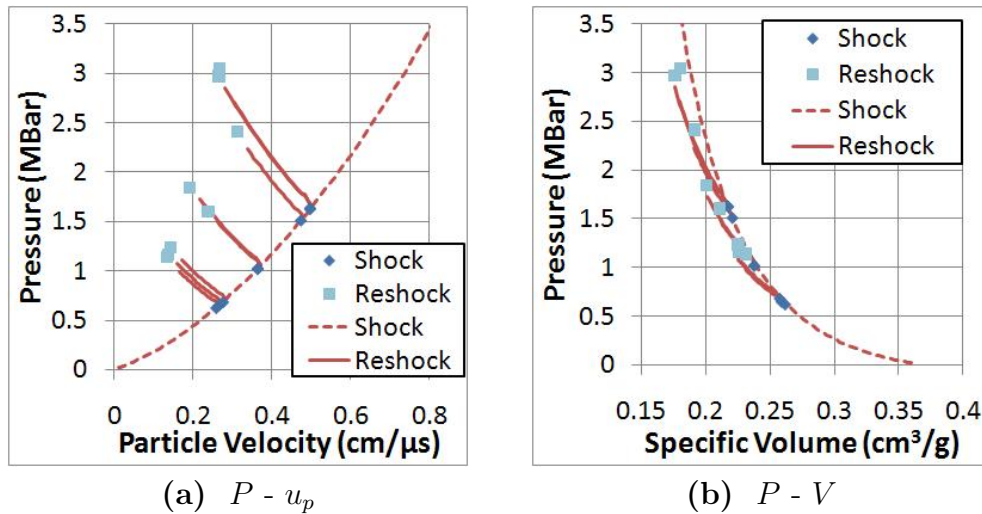
**Figure 3.10** Tillotson EOS Grüneisen parameter

The Tillotson EOS does a good job of matching the release isentrope free-surface velocity shown in Figure 3.11. In general Tillotson matches the release isentropes better than most EOS because it is a two-phase model. The material transitions from all solid to all vapor on release, which is sufficient for high-pressure release cases where melting and the liquid state aren't thought to play a large role in the release isentrope.



**Figure 3.11** Tillotson Release Isentrope Results. Experimental data from References [21-29]

Reshock states for the Tillotson EOS are shown in Figure 3.12. The Tillotson EOS does a better job predicting the reshock state because the Grüneisen parameter varies with specific volume and energy.



**Figure 3.12** Tillotson Reshock Results. Experimental data from Reference [30].



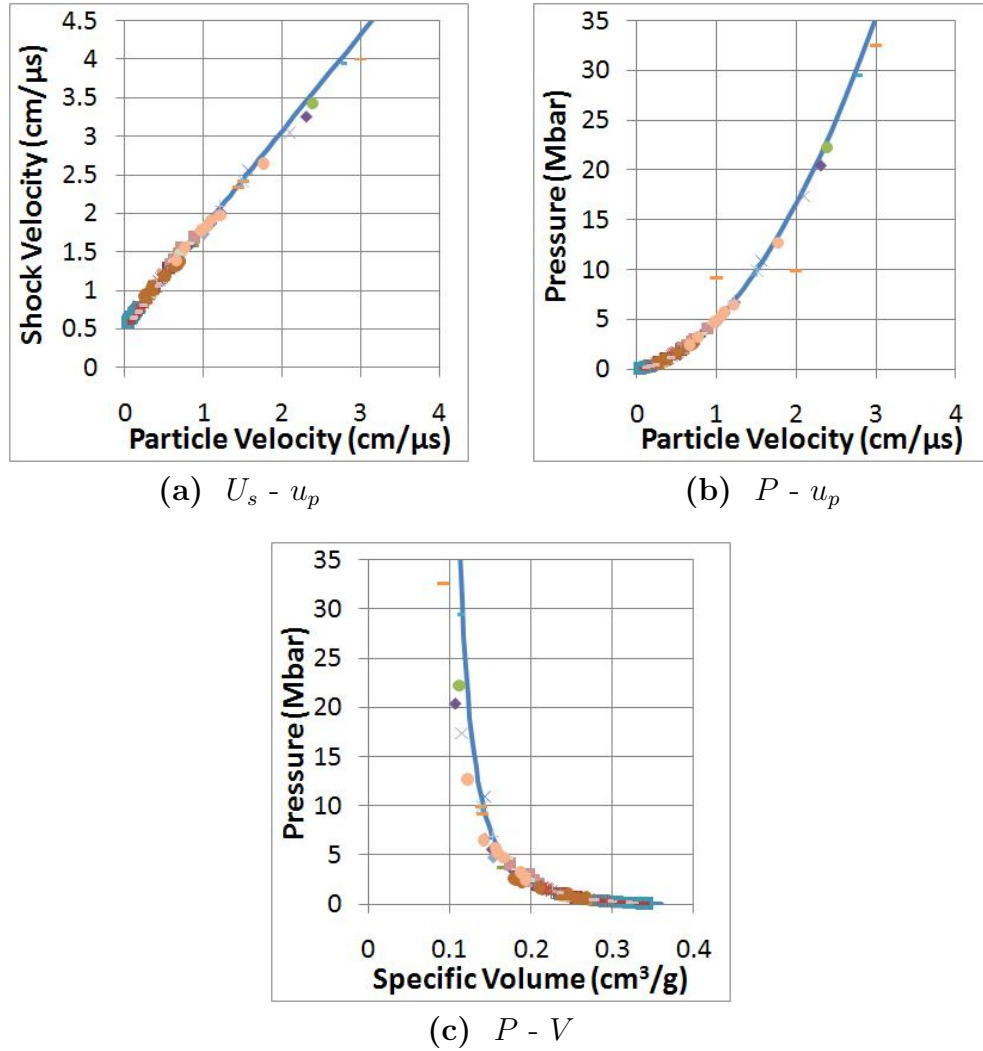
### 3.3 Multi-Branch Analytical EOS

The next EOS is the Multi-Branch Analytical EOS (MBEOS). The parameters used in the EOS are shown in Table 3.6. The effects of ionization shown in Equation 2.11 are neglected since it is particular to Lithium.

**Table 3.6** Aluminum MBEOS parameters

$\rho_0$ $g/cm^3$	$\gamma_0$	$T_0$ $K$	$P_c$ $Mbar$	$E_c$ $Mbar * cm^3/g$	$E_1$ $Mbar * cm^3/g$	$C_0$ $cm/\mu s$	$S$
2.712	2.14	298	0.0018202	0.122	0.01	0.54518	1.2592
$m$	$n$	$k$	$\xi$	$R$ $Mbar * cm^3/(gK)$			
1.0	1.1	1.1	2/3	3.08173e-6			

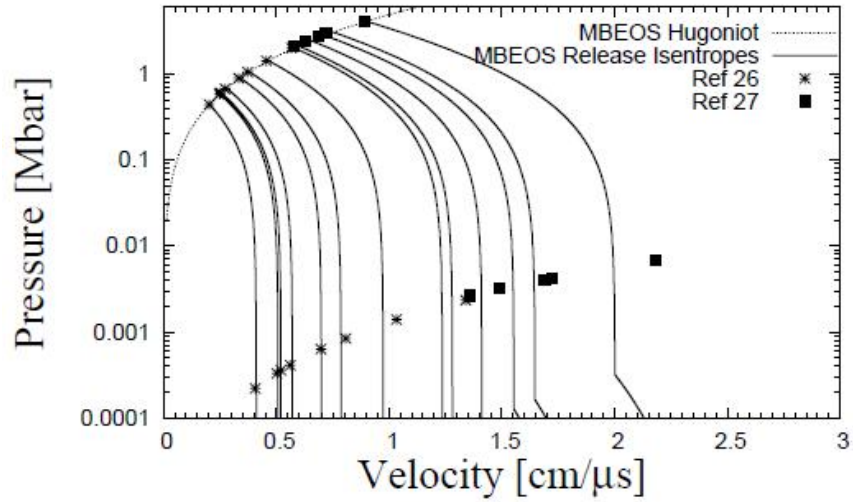
The Hugoniot, Figure 3.13, was compared to the experimental data. The EOS shows reasonable agreement with experimental data. This is expected since the MBEOS uses the Mie-Grüneisen model in the condensed phase region.



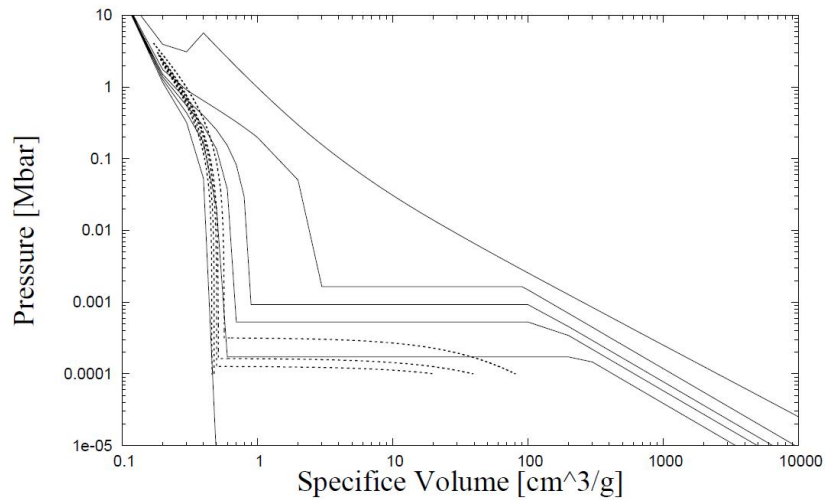
**Figure 3.13** MBEOS Hugoniot Plots. Experimental data from References [5-28]

The release isentropes were also generated for this EOS, Figure 3.14. The release isentropes show very little improvement from the Mie-Grüneisen results. Figure 3.15 is a series of constant energy lines to show the change in phase state. The dashed lines in the figure are a selection of high-pressure release isentropes. This shows that the material passes through the mixed phase region under a strong enough initial shock state. Recall that the phases are separated by the Clausius-Clapeyron given in

Equation 2.13.



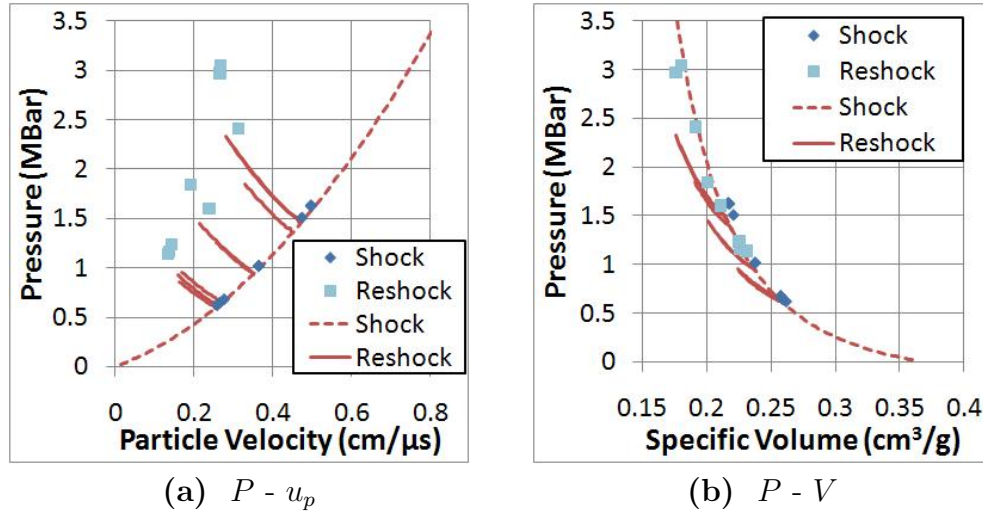
**Figure 3.14** MBEOS Release Isentropes Results. Experimental data from References [21-29]



**Figure 3.15** MBEOS Phase Diagram Results.

The results for the MBEOS reshock data are shown in Figure 3.16. The results are the same as the Mie-Grüneisen EOS. This is expected since the MBEOS uses the

Mie-Grüneisen EOS in the compression region.



**Figure 3.16** MBEOS Reshock Results. Experimental data from Reference [30].

### 3.4 Bushman-Lomonosov EOS

While reviewing Lomonosov [35] EOS approach, some mathematical discrepancies were found that didn't agree with the Bushman EOS method. This was particularly evident in the liquid atomic terms. So a hybrid approach was used. The cold contribution and the electronic terms in Table 3.7 are taken from the Lomonosov work while the atomic terms in Table 3.7 are taken from the Bushman model. A summary of the EOS parameters used are shown in Table 3.7.

This EOS is a multiphase EOS with three distinct phases; solid, liquid, and vapor. The liquid and vapor phases are modeling the liquid relationship described earlier. Instead of calculating the saturation points independently and interpolating between

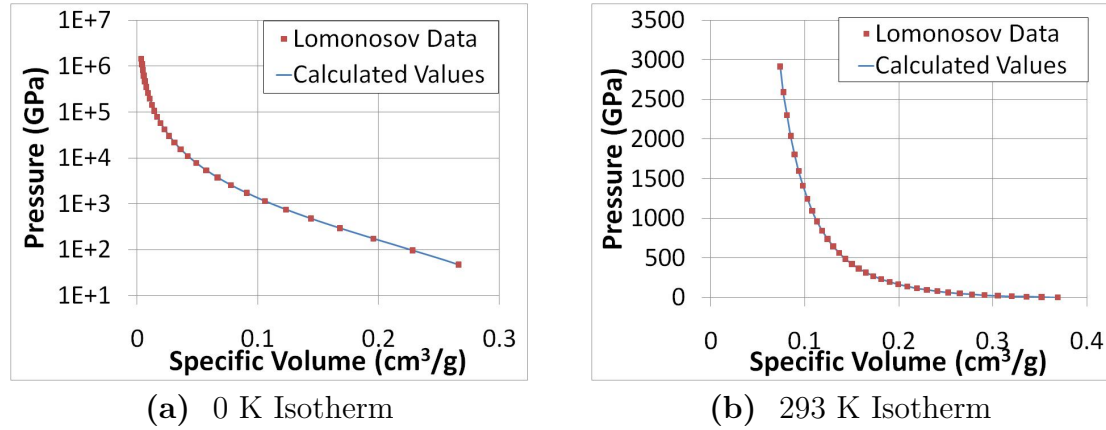
them, the saturation points are calculated during each set of the calculation. This is a computationally expensive method but removes errors associated with interpolation.

The saturation points were calculated in a multi-step process. First it is determined if the the thermodynamic state lies in the mixed-phase region. This is accomplished by tracking along a given isotherm for multiple specific volume with the same pressure. The next step is to determine saturation points for the given isotherm. The saturation points are calculated using Gibb's phase equilibrium using the method described in Section 1.1.4. The final step is to determine if the temperature-volume state lies in the mix-phase region. The transition between the solid and liquid region is also calculated using Gibb's phase equilibrium. The state with the lower Gibb's free energy is the stable state. No consideration of a mixed-phase region between the solid and liquid states were considered. It is assumed that the material is either all solid or all liquid.

**Table 3.7** Aluminum Bushman-Lomonosov EOS parameters

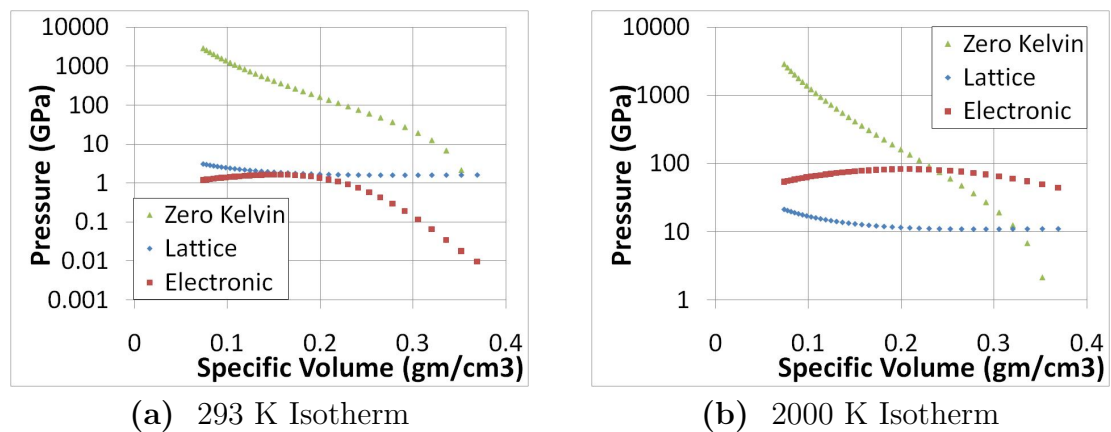
$V_0$ $g/cm^3$	$V_{0c}$ $g/cm^3$	$Z$	$R$	$a_1$	$a_2$	$a_3$	$a_4$	$a_5$
0.370	0.361	13	0.31	326.35	-1035.44	858.51	-160.59	11.17
$E_{sub}$ $g/cm^3$	$A_c$ $g/cm^3$	$B_c$	$C_c$	$m$	$n$	$l$	$T_m$ kK	$\sigma_m$
-12.1	-12.91	40.96	-28.05	8.0	4.99	0.70	0.933	0.923
$\Theta_{0s}$ kK	$\gamma_{0s}$	$B_s$	$D_s$	$T_T$ kK	$\sigma_T$	$T_{sa}$ kK	$T_{ca}$ kK	$\Theta_{0l}$ kK
0.1	2.19	0.7	0.7	30	0.14	6	25	157
$\gamma_{0l}$ kK	$B_l$	$D_l$	$A_m$	$B_m$	$C_m$	$T_Z$ kK	$\sigma_Z$	$T_i$ kK
1.78	1.05	0.0	2.24	-5.64	0.21	200	0.8	50
$\sigma_i$	$\gamma_0$	$\gamma_m$	$T_g$ kK	$\sigma_e$	$\sigma_d$	$T_b$	$\beta_i$	$\beta_0$
0.3	0.7	-0.5	300	1.0	9.99e9	8	0.0242	0.050
$\beta_m$								
0.0								

Lomonosov [35] included some numerical data including 0 K isotherm, 293 K isotherm, and the principal Hugoniot. Comparisons between the numerical data and the EOS are shown in Figure 3.17. The EOS is in good agreement with the numerical data from Lomonosov.



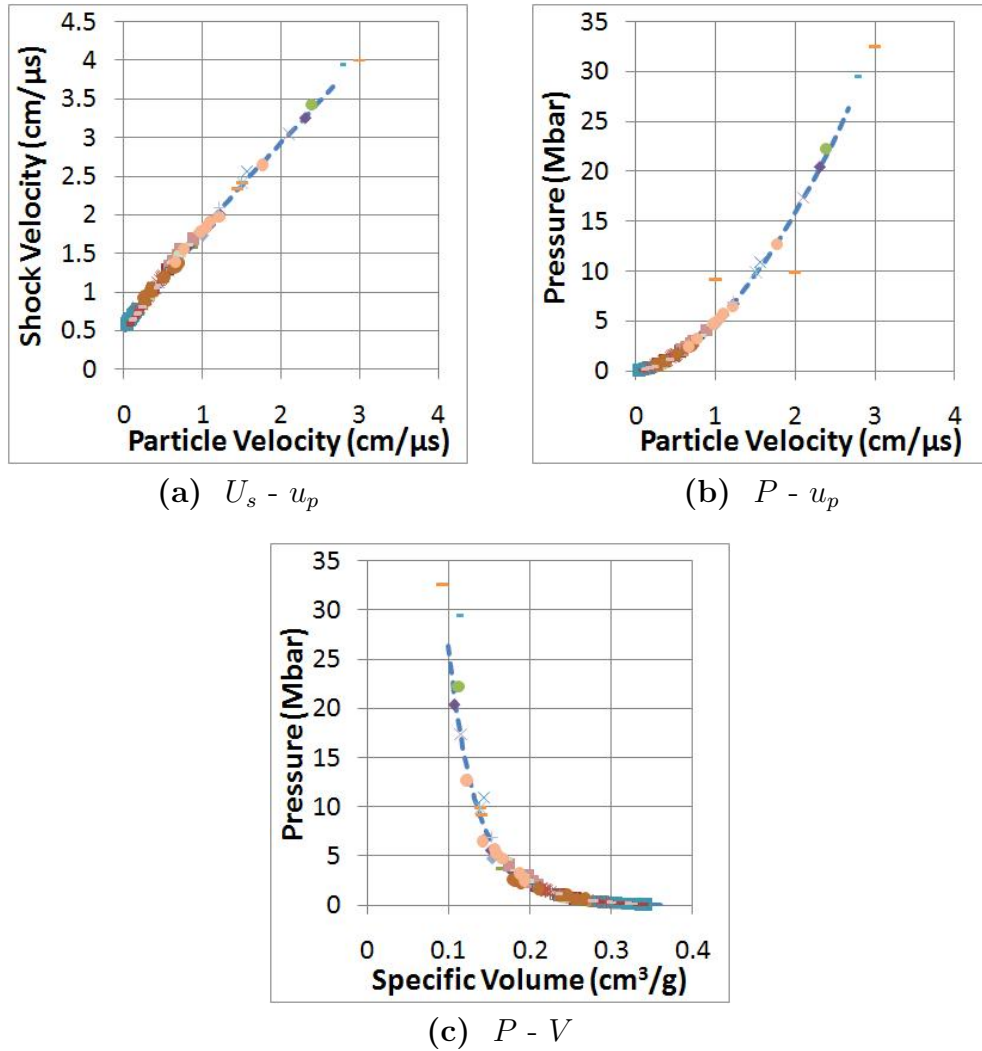
**Figure 3.17** Isotherms compared to Lomonosov Data

The three components that make up the Bushman-Lomonosov EOS; cold curve, lattice, and electronic contributions, can be evaluated independently as was done in Figure 3.18. The figure shows a low temperature and a higher temperature isotherm. In low temperature isotherms, the main contributor is the cold curve terms. As temperature increases, the lattice and the electronic terms play a more important role.



**Figure 3.18** Bushman-Lomonosov EOS Contributions

Figure 3.19 shows the Hugoniot plots using the Bushman-Lomonosov EOS. The EOS transitions from a solid to a liquid during compression, something not captured in other EOSs. This is shown by the transition from the solid line to the dashed line.

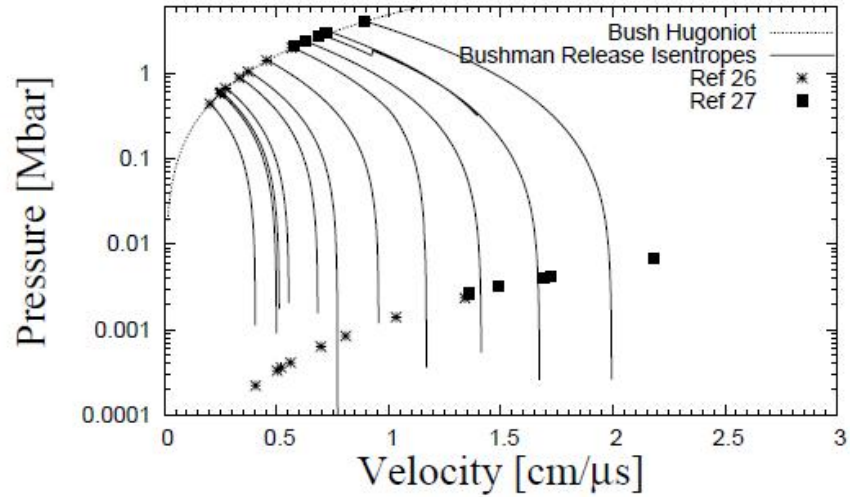


**Figure 3.19** Bushman-Lomonosov Hugoniot Plots. Experimental data from References [5-28]

The release isentropes, Figure 3.20, show good agreement at low initial pressure. As the pressure increases the EOS under-predicts the free surface velocity. A sharp

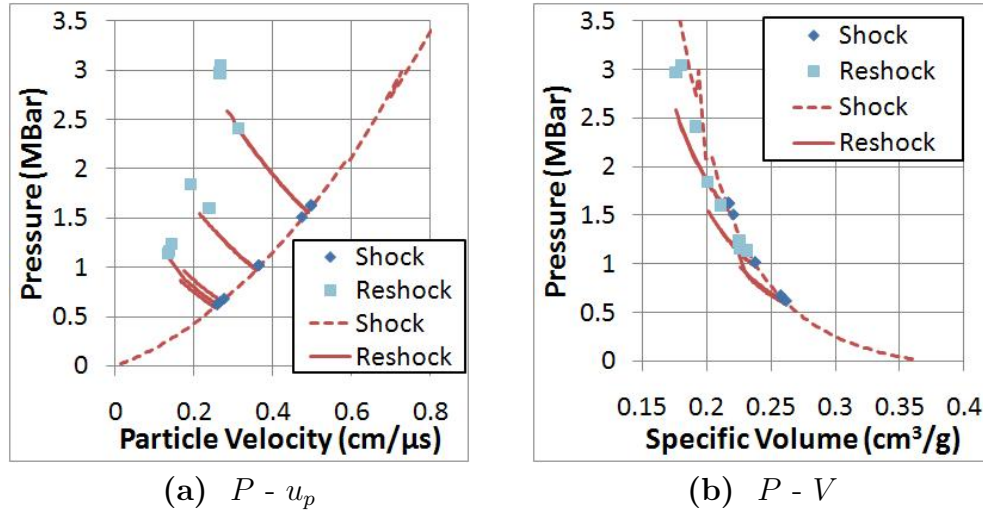


discontinuity seen in Figure 3.20 is a result of the solver not iterating to the correct solution. The Newton method was unable to converge on a solution because it was iterating between the solid and liquid phases.



**Figure 3.20** Bushman - Lomonosov Release Isentrope Results. Experimental data from References [21-29]

The calculated reshock states for the Bushman-Lomonosov EOS are shown in Figure 3.21. The Bushman-Lomonosov EOS does a reasonable job predicting the reshock state, though it under-predicts pressure.



**Figure 3.21** Bushman-Lomonosov Reshock Results. Experimental data from Reference [30].

The EOS predicts a critical temperature around 5500 K and a critical pressure of 1 GPa. These values differ from Bushman and Lomonosov results. Lomonosov predicted a critical temperature of 6250 K and a critical pressure of 0.197 GPa, while Bushman predicts a critical temperature of 7222 K and a critical pressure of 0.571 GPa.

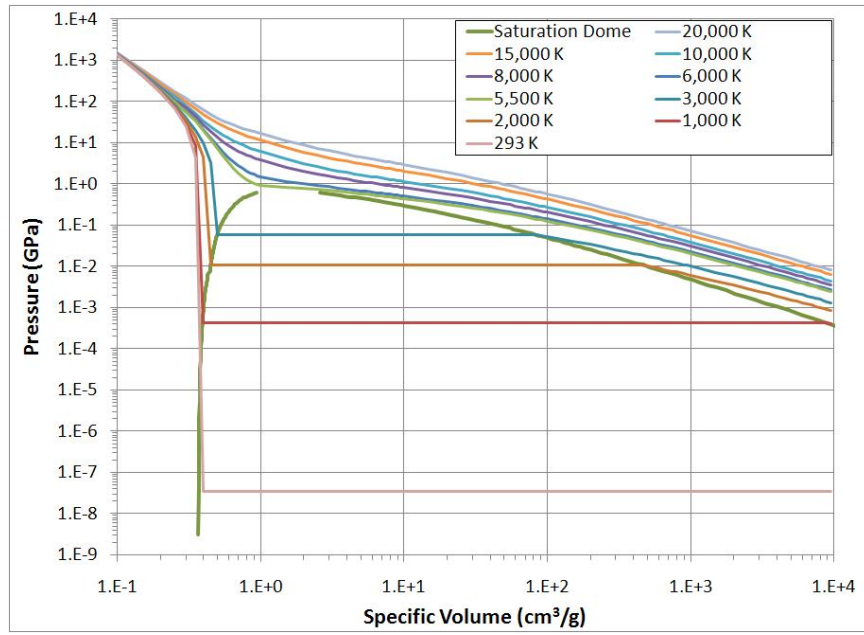


Figure 3.22 Bushman-Lomonosov Phase Diagram Results

## Chapter 4

### Conclusions

Several equations of state (EOS) for Aluminum were investigated in this thesis. Different types of EOS were also considered. An incomplete EOS formulation requires an additional relationship to fully define the thermodynamic state. This contradicts a complete EOS formulation which fully defines the thermodynamic state. Examples of incomplete EOSs modeled in this work are Mie-Grüneisen, Tillotson, and Multi-branch analytical EOS (MBEOS). The Bushman-Lomonosov EOS is the only complete EOS modeled here.

All EOS examined here do a good job of modeling the the Hugoniot at lower pressures shown in Figures 4.1 - 4.3. Mie-Grüneisen, Tillotson, and MBEOS have matching Hugoniot responses. This is expected since all three EOS use a Mie-Grüneisen form to represent the condensed phase. Bushman-Lomonosov EOS shows a different response. This is particularly evident at stronger shocks. Under stronger shocks, a solid-liquid phase change is predicted in the Bushman-Lomonosov EOS that is not captured in the other EOS. This is shown in dashed lines in Figures 4.1-4.3. The Bushman-Lomonosov best represents the shock Hugoniots over the range of states

examined here. Under lower pressure, all EOS give good results.

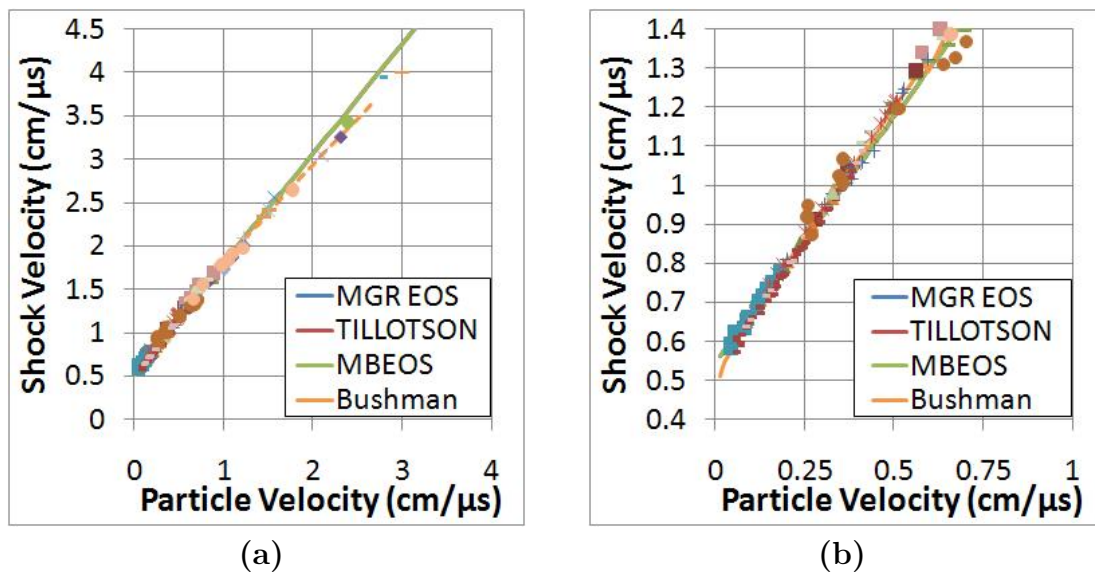


Figure 4.1  $U_s-u_p$  Results

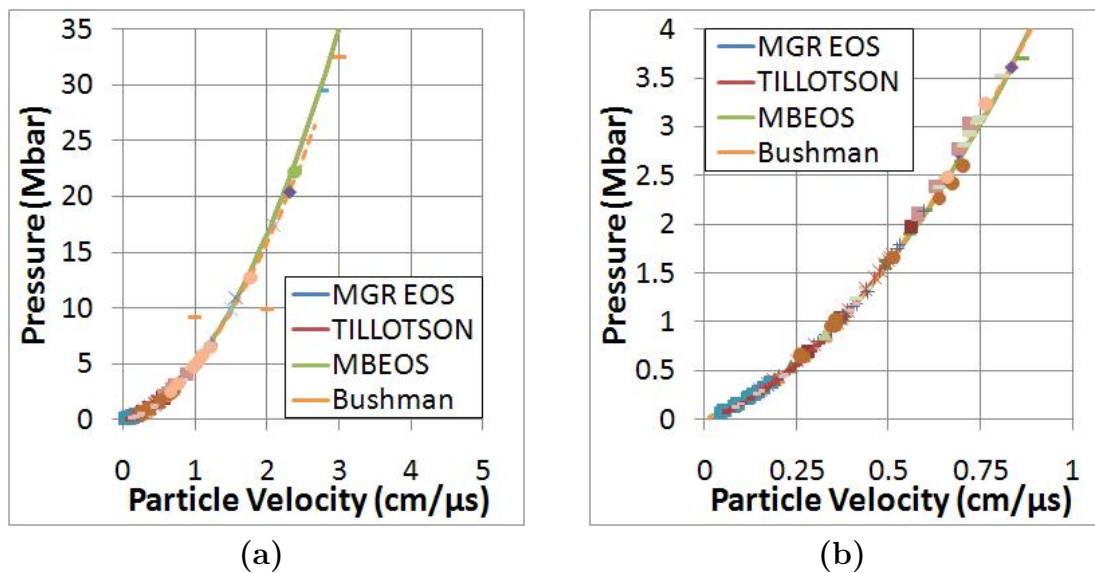


Figure 4.2  $P-u_p$  Results

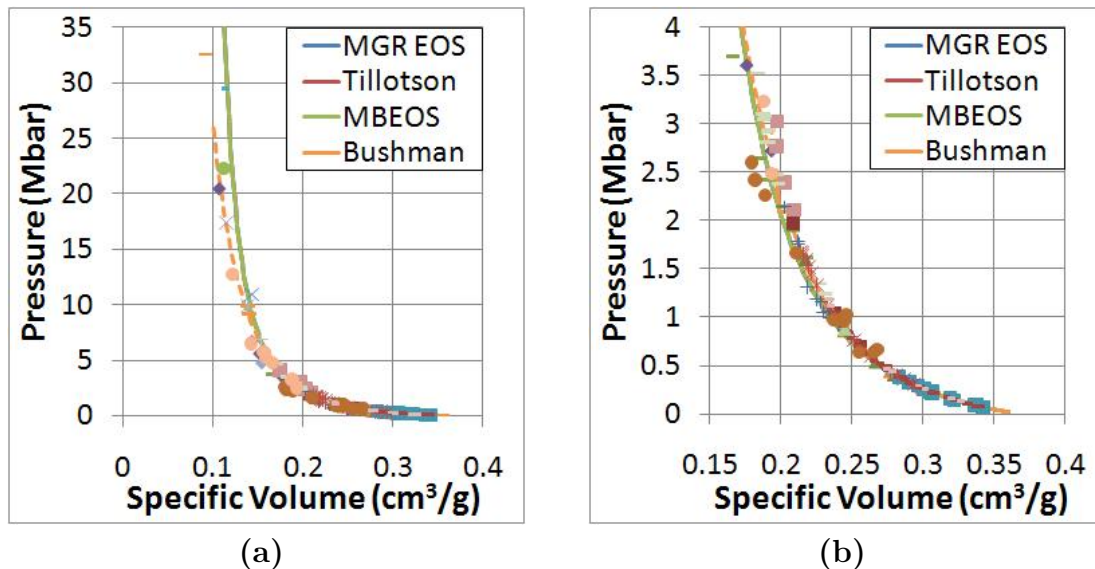
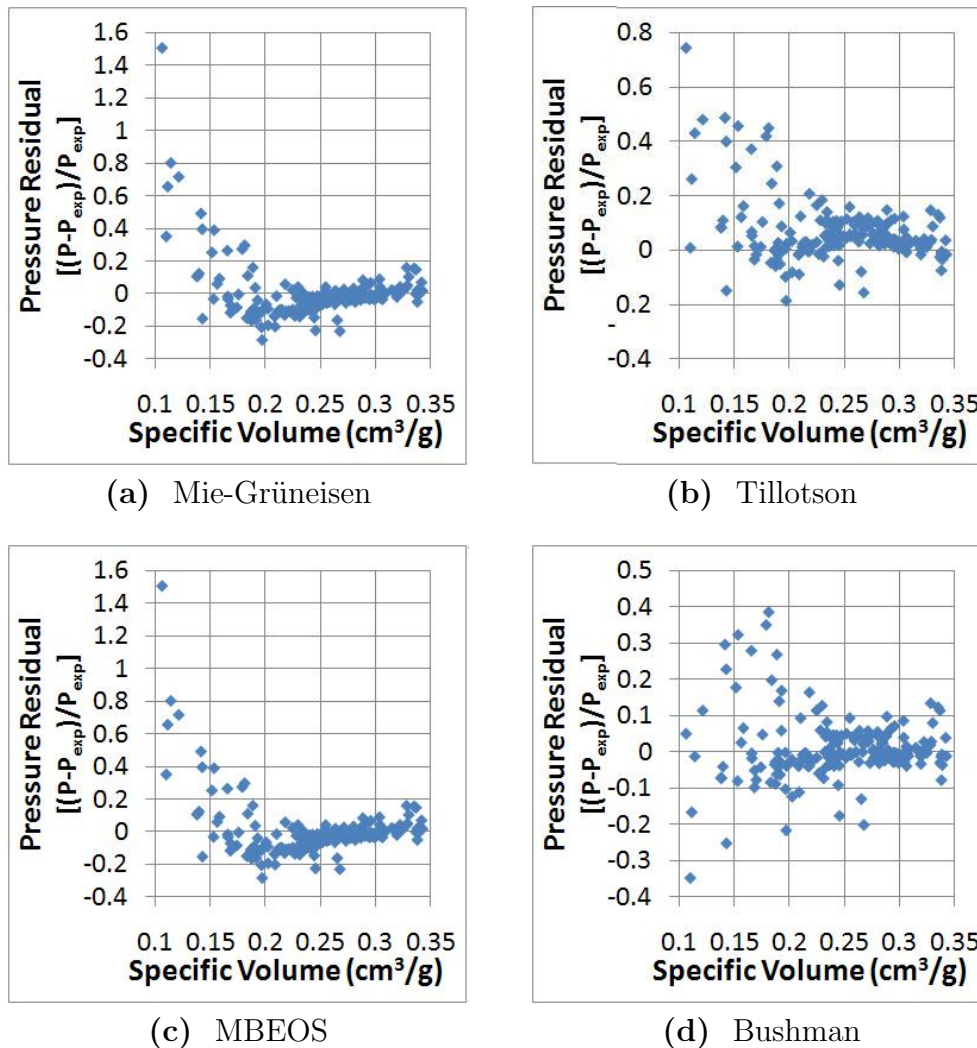


Figure 4.3  $P$ - $V$  Results

Pressure residuals plotted in Figure 4.4 show how the computation results deviate from the experimental results. All the models reasonably match the low compression region within 20 %. Under large compressions the Mie-Grüneisen and MBEOS EOS vary by 160 % while the Tillotson EOS only varies by a maximum of 80 %. The Bushman EOS performs the best with the largest deviation of 40 % over the entire experimental range.



**Figure 4.4** Residual pressure along the principle Hugoniot

The free surface velocity calculation shows that the Tillotson EOS is the most accurate for the experimental data. The remaining EOSs do an acceptable job under lower initial states. Once the shock pressure is greater than 1.5 MBar the free surface results diverge considerably.

The reshock calculations show how the EOSs perform off the principle Hugoniot. In the Mie-Grüneisen, Tillotson, and MBEOS EOSs the Grüneisen parameter has a

major effect on the calculated reshock state. The Tillotson does not use a constant Grüneisen ratio but calculated the Grüneisen parameter as a function of energy and specific volume. The Tillotson EOS performs the best of the three incomplete EOSs. The Bushman-Lomonosov EOS does a good job predicting the reshock state but underestimates pressure.



## Chapter 5

### Future Work

The work presented here could be extended in several different directions. First, the method created here could be extended to include additional EOS models and theory. The code could be improved to handle phase transitions more accurately and with less numerical noise. This would be accomplished by improving the numerical methods for determining the stable solution maybe via a mixing region between the solid-liquid transition. The code could be changed to include SESAME tabular output which would allow for new EOS table creation. The code is currently limited on which thermodynamic variables are needed to define a thermodynamic state. The code could be expanded so that for any two thermodynamic variables the remaining ones are found.

Another possible direction for future work is to incorporate models presented here into existing shock physics codes. This could be accomplished by using the individual modules that define each EOS. The shock physics code would need to provide thermodynamic variables and the remaining ones would be calculated and returned.

Finally, this work could be extended to develop new EOS models. The model theory could be created and tested using the framework developed here. A composite approach could be utilized to use a wide array of models. The composite model would require smoothing between each individual theory but may lead to more accurate behavior at extreme states.

## Bibliography

- [1] A. V. Bushman, G. I. Kanel', A. L. Ni, and V. E. Fortov, *Intense Dynamic Loading of Condensed Matter*. Washington, DC: Taylor & Francis, 1993.
- [2] M. A. Meyers, *Dynamic Behavior of Materials*. Wiley-Interscience, 1 ed., Sept. 1994.
- [3] G. R. Gathers, *Selected Topics in Shock Wave Physics and Equation of State Modeling*. River Edge, NJ: World Scientific, 1994.
- [4] Y. B. Zel'dovich and Y. P. Raizer, *Physics of Shock Waves and High-Temperature Hydrodynamic Phenomena*. Dover Publications, Mar. 2002.
- [5] L. V. Al'tshuler, S. B. Kormer, A. A. Bakanova, and R. F. Trunin, "Equations of state for aluminum, copper and lead in the high pressure region," *Journal of Experimental and Theoretical Physics*, vol. 11, pp. 573–579, 1960.
- [6] L. V. Al'tshuler, S. B. Kormer, M. I. Brazhnik, L. A. Vladimirov, M. P. Speranskaya, and A. I. Funtikov, "The isentropic compressibility of aluminum, copper, lead at high pressures," *Journal of Experimental and Theoretical Physics*, vol. 11, no. 4, pp. 766–775, 1960.

- 
- [7] L. V. Al'tshuler and A. P. Petrunin, "Rentgenographic investigation of compressibility of light substances under obstacle impact of shock waves," *Zhurnal Tekhnicheskoi Fiziki*, vol. 36, no. 6, pp. 717–725, 1961.
- [8] I. C. Skidmore and E. Morris, "Experimental equation-of-state data for uranium and its interpretation in the critical region," IAEA 173-216, Vienna, 1962.
- [9] W. H. Isbell, F. H. Shipman, and A. H. Jones, "Hugoniot equation of state measurements for eleven materials to five megabars," Materials Science Laboratory MSL-68-13, General Motors Corp., 1968.
- [10] R. G. McQueen, S. P. Marsh, J. W. Taylor, J. N. Fritz, and W. J. Carter, "The equation of state of solids from shock wave studies," in *High Velocity Impact Phenomena*, pp. 293–417, New York: Academic Press, 1970.
- [11] L. V. Al'tshuler and B. S. Chekin, "Metrology of high pulsed pressures," in *All-Union Pulsed Pressures Symposium*, vol. 1, (Moscow), pp. 5–22, 1974.
- [12] L. V. Al'tshuler, N. N. Kalitkin, L. V. Kuz'mina, and B. S. Chekin, "Shock adiabats for ultrahigh pressures," *Journal of Experimental and Theoretical Physics*, vol. 45, no. 1, pp. 167–171, 1977.
- [13] S. P. Marsh, *LASL Shock Hugoniot Data*. Berkeley: Univ. California Press, 1980.
- [14] L. V. Al'tshuler, A. A. Bakanova, I. P. Dudoladov, E. A. Dynin, R. F. Trunin, and B. S. Chekin, "Shock adiabats for metals. new data, statistical analysis and general regularities," *Journal of Applied Mechanics and Technical Physics*, vol. 22, p. 145, 1981.

- [15] L. P. Volkov, N. P. Voloshin, A. S. Vladimirov, V. N. Nogin, and V. A. Simonenko, "Shock compressibility of aluminum at pressure 10 mbar," *Journal of Experimental and Theoretical Physics Letters*, vol. 31, pp. 588–591, 1981.
- [16] A. C. Mitchell and W. J. Nellis, "Shock compression of aluminum, copper, and tantalum," *Journal of Applied Physics*, vol. 52, no. 5, pp. 3363–3374, 1981.
- [17] C. E. Ragan, "Shock compression measurements at 1 to 7 TPa," *Physical Review A*, vol. 25, p. 3360, June 1982. Copyright (C) 2009 The American Physical Society; Please report any problems to prola@aps.org.
- [18] C. E. Ragan, "Shock-wave experiments at threefold compression," *Physical Review A*, vol. 29, p. 1391, Mar. 1984. Copyright (C) 2009 The American Physical Society; Please report any problems to prola@aps.org.
- [19] V. A. Simonenko, N. P. Voloshin, A. S. Vladimirov, A. P. Nagibin, V. Nogin, V. A. Popov, V. A. Sal'nikov, and Y. A. Shoidin, "Absolute measurements of shock compressibility of aluminum at pressures  $p \geq 1$  TPa," *Journal of Experimental and Theoretical Physics*, vol. 61, no. 4, pp. 869–873, 1985.
- [20] R. F. Trunin, "Compressibility of various substances at high shock pressures," *Physics of the Solid Earth*, vol. 22, p. 103, 1986.
- [21] B. L. Glushak, A. P. Zharkov, M. V. Zhernokletov, V. Ternovoi, A. S. Filimonov, and V. E. Fortov, "Experimental investigation of the thermodynamics of dense plasmas formed from metals at high energy concentrations," *Journal of Experimental and Theoretical Physics*, vol. 69, no. 4, pp. 739–749, 1989.
- [22] M. A. Podurets, G. V. Simakov, and R. F. Trunin, "Transition of stishovite to a denser phase," *Fizika Zemli*, vol. 4, pp. 30–37, 1990.

- [23] M. A. Podurets, V. M. Ktitorov, R. F. Trunin, V. A. Popov, A. Y. Matveev, B. V. Pechenkin, and A. G. Sevast'yanov, "Shock wave compression of aluminum at pressures of 1,7 TPa," *Teplofizika Vysokikh Temperatur*, vol. 32, no. 6, pp. 952–955, 1994.
- [24] R. Trunin, M. Panov, and A. Medvedev, "Compressibility of iron, aluminum, molybdenum, titanium, and tantalum at shock-wave pressures of 1-2.5 TPa," *Journal of Experimental and Theoretical Physics Letter*, vol. 62, no. 7, pp. 591–594, 1995.
- [25] R. F. Trunin, M. A. Podurets, G. V. Simakov, V. A. Popov, and A. G. Sevast'yanov, "New data on aluminum, plexiglas and quartz compression obtained in strong shock wave of underground nuclear explosion," *Journal of Experimental and Theoretical Physics*, vol. 81, no. 3, p. 464, 1995.
- [26] R. F. Trunin, M. Panov, and A. Medvedev, "Shock compressibilities of iron, aluminum, and tantalum at terapascal pressures," *Khim. Fiz.*, vol. 14, no. 2-3, pp. 97–99, 1995.
- [27] R. F. Trunin, L. F. Gudarenko, M. V. Zhernokletov, and G. V. Simakov, "Experimental data on shock compressibility and adiabatic expansion of condensed substances," in *RFNC*, (Sarov), 2001.
- [28] M. D. Knudson, R. W. Lemke, D. B. Hayes, C. A. Hall, C. Deeney, and J. R. Asay, "Near-absolute hughoniot measurements in aluminum to 500 GPa using a magnetically accelerated flyer plate technique," *Journal of Applied Physics*, vol. 94, pp. 4420–4431, Oct. 2003.

- [29] A. A. Bakanova, I. P. Dudoladov, M. V. Zhernokletov, V. N. Zubarev, and G. V. Simakov, "Vaporization of shock-compressed metals on expansion," *Journal of Applied Mechanics and Technical Physics*, vol. 24, pp. 204–209, Mar. 1983.
- [30] W. J. Nellis, A. C. Mitchell, and D. A. Young, "Equation-of-state measurements for aluminum, copper, and tantalum in the pressure range 80–440 GPa (0.8–4.4 mbar)," *Journal of Applied Physics*, vol. 93, no. 1, pp. 304–310, 2003.
- [31] J. C. Maxwell, *The scientific paper of James Clerk Maxwell*. Mineola, NY: Dover Publications, 2003.
- [32] K. Wark, *Advanced thermodynamics for engineers*. McGraw-Hill, 1995.
- [33] J. H. Tilltson, "Metallic equations of state for hypervelocity impact," Technical Report GA-3216, General Atomic, San Diego, CA, July 1963.
- [34] L. Glenn and D. Young, "Dynamic loading of the structural wall in a lithium-fall fusion reactor," *Nuclear Engineering and Design*, vol. 54, pp. 1–16, Sept. 1979.
- [35] I. Lomonosov, "Multi-Phase equation of state for aluminum," *Laser and Particle Beams*, vol. 25, no. 04, pp. 567–584, 2007.
- [36] N. M. Laurendeau, *Statistical Thermodynamics*. New York, NY: Cambridge University Press, 2005.
- [37] M. L. Wilkins, *Computer Simulation of Dynamic Phenomena*. Scientific Computation, New York: Springer-Verlag, 1999.

# Appendix A

## Equation of State Source Code

The main EOS routine is two parts. A shared module where common variables are stored and the main driver routine. Below is the shared module.

```

MODULE EOS
  IMPLICIT NONE
  DOUBLE PRECISION rho0 , C0, s1 , v0 ,gamma0
  DOUBLE PRECISION Pr, Er
  DOUBLE PRECISION P,E,Up, Us, T, S,H,G
  DOUBLE PRECISION, dimension(10) :: a, ti, mb
  DOUBLE PRECISION, dimension(50) :: be
  INTEGER, dimension(5) :: nEOS
END MODULE EOS

```

This is the main driver routine for the EOS code.

```

PROGRAM eos_1_0
  USE EOS
  !-----
  !
  ! CODE NAME:          eos_1_0
  !
  ! VERSION NUMBER:    1.0
  !
  ! DATE CREATED:      28 JANUARY 2014
  !
  ! LANGUAGE:          FORTRAN 90
  !
  ! AUTHOR:            Aaron Ward
  !                   email: aaron.ward@corvidtec.com
  !
  ! DESCRIPTION:

```



---

!

! Calculate shock properties and relationships via different equation

! of state (EOS).

!

! The Mie-Gruneisen EOS states that  $P(V,E) = Pr + \gamma/V*(E-Er)$

! here the 'r' represents the references state. The simplist case is

! where the reference state is the known principle hugoniot.

!

! UPDATES:

DATE	WHO	Ver.	CHANGES
07/27/09	AJW	0.1	ADDED Mie-Gruneisen EOS to program
08/05/09	AJW	0.2	ADDED Tillotson EOS inputs
09/07/09	AJW	0.3	ADDED Release isentrope with calculating the free surface velocity
09/14/09	AJW	0.4	ADDED improved method for calculating the principle hugoniot
09/15/09	AJW	0.4	ADDED Tillotson principle hugoniot and single phase isentropic release
09/16/09	AJW	0.4	ADDED Tillotson 2 phase isentropic release
09/22/09	AJW	0.5	The EOSs have been removed form the main program and instead have been writen as Modules
09/22/09	AJW	0.5	The findV, isentrope sections have been rewritten with improvements. the initial guess of v in findV is still a consern an new method of determining should be found.
10/05/09	AJW	0.6	Tillotson EOS validated
10/18/09	AJW	0.7	EOS Modues rewritten, new aritecture setup for finding EOS variables. Given any two the remaining variables are found. This allows hugoniont, isentropic, etc logic to be moved back into the main program. Requiring a significate rewrite which concluded version 0.7
10/21/09	AJW	0.8	Multi-branch EOS modulus added
05/02/11	AJW	1.0	Added Compression Isentropes

!

---

**IMPLICIT NONE**

**INTEGER** sel

! DOUBLE PRECISION P1, v1

! Start of the Program

---

```

!   - Menu
!   ( I would like something like panda

      WRITE(*,100) 'EOS CODE', 'VERSION 0.9'

1000 CALL MENU(sel)

9000 WRITE(*,*) 'BYE'
!
100  format(25x,A/,23x,A/)
      END PROGRAM eos_1_0
!-----
!   Subroutine: MENU
!
!   DESCRIPTION:
!   Main Menu Subroutine tells options
!-----
      SUBROUTINE MENU(sel)
      IMPLICIT NONE
      integer sel
!
!   sel = 0
1000 WRITE(*,101) 'Select EOS to Model'
      WRITE(*,100) '-----MENU-----',
      WRITE(*,101) '(1) Mie-Gruneisen EOS'
      WRITE(*,101) '(2) Tillotson EOS'
      WRITE(*,101) '(3) Multi-branch EOS'
      WRITE(*,101) '(4) '
      WRITE(*,101) '(5) USER'
      WRITE(*,101) '(6) Exit System'
      read(*,*) sel
      if(sel.eq.1) then
          CALL mgrinput(sel)
      elseif(sel.eq.2) then
          CALL tilinput(sel)
      elseif(sel.eq.3) then
          CALL mbinput(sel)
!
!   elseif(sel.eq.4) then
!       CALL beinput(sel)
      elseif(sel.eq.5) then
          write(*,100) 'This feature will be added later to allow'
          write(*,100) 'user EOS to be added for testing pruposes'
          write(*,*) ''
          goto 1000
      elseif(sel.eq.6) then

```

```

        return
    else
        write(*,*) 'Not a valid option'
        goto 1000
    endif
    CALL OMENU()
    if (sel.ne.6) then
        goto 1000
    endif
    return
100 format(15x,A)
101 format(18x,A)

    END SUBROUTINE MENU
!-----
! Subroutine: OMENU
!
! DESCRIPTION:
! The OMENU or hugoniot menu allow you to select what is to be done next
! plotting, calculating based on the Mie-Gruneisen Hugoniot form.
!-----
SUBROUTINE OMENU()
    IMPLICIT NONE
    integer sel
1000 WRITE(*,100) '-----Options MENU-----'
    WRITE(*,101) '(1) Hugoniot'
    WRITE(*,101) '(2) Isotherms'
    WRITE(*,101) '(3) Reshock'
    WRITE(*,101) '(4) Isoenergy'
    WRITE(*,101) '(5) Release Isentrope'
    WRITE(*,101) '(6) Start Over'
    read(*,*) sel
    if(sel.eq.1) then
        CALL inpHugo()
    elseif(sel.eq.2) then
        CALL ISOTHERM()
    elseif(sel.eq.3) then
!       CALL inpRels()
        CALL reshock()
    elseif(sel.eq.4) then
        CALL ISOENERGY()
    elseif(sel.eq.5) then
        CALL HRISENTROPE()
    elseif(sel.eq.6) then
        RETURN

```

```

        else
            RETURN
        endif
        goto 1000
        return
100 format(15x,A)
101 format(18x,A)
    END SUBROUTINE OMENU

```

---

```

! Subroutine: inpHugo()
!
! DESCRIPTION:
! Values to be using in calculating princilpe hugo
!

```

---

```

SUBROUTINE inpHugo()
USE EOS
integer sel,step
DOUBLE PRECISION L,U

1000 WRITE(*,101) 'Select Independent Variable'
WRITE(*,101) '(1) v'
read(*,*) sel
if(sel.eq.1) then
    goto 2000
elseif(sel.eq.2) then
    goto 3000
elseif(sel.eq.2) then
    goto 4000
else
    write(*,101)'Not a valid input'
    goto 1000
endif
2000 WRITE(*,101) 'Enter Low, Upper V and increments'
READ(*,*) L,U,step
CALL Hugoniot(L,U,step)
GOTO 5000
3000 WRITE(*,*) 'Feature not added yet'
goto 2000
4000 WRITE(*,*) 'Feature not added yet'
goto 2000
5000 WRITE(*,*) 'Plot again yes (1)/No (2)'
read(*,*) sel
if(sel.eq.1) then
    goto 1000

```

```

    endif
    RETURN
100 format(10x, '———', A)
101 format(18x, A)
    END SUBROUTINE inpHugo

```

---

```

!
! Subroutine: inpRels()
!
! DESCRIPTION:
! Values to be using in calculating princilpe hugo
!

```

---

```

SUBROUTINE inpRels()
USE EOS
integer sel, step
DOUBLE PRECISION L,U
1000 WRITE(*,101) 'Select Independent Variable'
WRITE(*,101) '(1) V'
! WRITE(*,101) '(2) u'
! WRITE(*,101) '(3) P'
read(*,*) sel
if(sel.eq.1) then
    goto 2000
elseif(sel.eq.2) then
    goto 3000
elseif(sel.eq.2) then
    goto 4000
else
    write(*,101)'Not a valid input'
    goto 1000
endif
2000 WRITE(*,101) 'V and increments'
READ(*,*) L,step
U = v0
write(*,*) L,U,step
! CALL ISENTROPE(U,L,step)
call Comp_Isen(L,step)
GOTO 5000
3000 WRITE(*,*) 'Not added yet'
goto 2000
4000 WRITE(*,*) 'Not added yet'
goto 2000
5000 WRITE(*,*) 'Plot again yes (1)/No (2)'
read(*,*) sel
if(sel.eq.1) then

```

---

```
        goto 1000
    endif

RETURN
100 format(10x, '———', A)
101 format(18x, A)
END SUBROUTINE inpRels

SUBROUTINE mgrinput(sel)
USE MGR
INTEGER sel
call eosINP(sel)
RETURN
END SUBROUTINE

SUBROUTINE tilinput(sel)
USE TILLOTSON
INTEGER sel
call eosINP(sel)
RETURN
END SUBROUTINE

SUBROUTINE mbinput(sel)
USE MBEOS
INTEGER sel
call eosINP(sel)
RETURN
END SUBROUTINE

SUBROUTINE mgrFindV(P1, vo)
USE MGR
DOUBLE PRECISION P1, vo
call findV(P1, vo)
RETURN
END SUBROUTINE

SUBROUTINE tilFindV(P1, vo)
USE TILLOTSON
DOUBLE PRECISION P1, vo
call findV(P1, vo)
RETURN
END SUBROUTINE

SUBROUTINE mbFindV(P1, vo)
```

```

USE MBEOS
DOUBLE PRECISION P1,vo
call findV(P1,vo)
RETURN
END SUBROUTINE

```

```

SUBROUTINE mgrprop(P1,E1,V1,T1,Ss,temp)
USE MGR
DOUBLE PRECISION P1,E1,V1,T1,Ss
DOUBLE PRECISION,DIMENSION(3) :: temp
call mgreos(P1,E1,V1,T1,Ss,temp)
RETURN
END SUBROUTINE

```

```

SUBROUTINE tillprop(P1,E1,V1,T1,Ss,temp)
USE TILLOTSON
DOUBLE PRECISION P1,E1,V1,T1,Ss
DOUBLE PRECISION,DIMENSION(3) :: temp
call tilleos(P1,E1,V1,T1,Ss,temp)
RETURN
END SUBROUTINE

```

```

SUBROUTINE mbprop(P1,E1,V1,T1,Ss,temp)
USE MBEOS
DOUBLE PRECISION P1,E1,V1,T1,Ss
DOUBLE PRECISION,DIMENSION(3) :: temp
call multieos(P1,E1,V1,T1,Ss,temp)
RETURN
END SUBROUTINE

```

---

```

!  

! Subroutine: Hugoniot  

!  

! DESCRIPTION:  

! Plots/exports isotherms for over a given temperature range and number  

! of steps

```

---

```

SUBROUTINE Hugoniot(hig,low,incr)
use EOS
IMPLICIT NONE
DOUBLE PRECISION L,U,delta,Cu,low,hig,dump,E2,Up2,f2
DOUBLE PRECISION f,df,ierror,error,Peos,Ut,V2,P2,dsmall,E1
DOUBLE PRECISION,DIMENSION(3) :: temp
DOUBLE PRECISION,DIMENSION(10) :: junk

```

```

integer sel , step , i , incr

!      write(*,*) 'v0: ',v0
open(11,file='hugo.dat',access='append')
open(9)
L = low
U = hig
step = incr
delta = (U-L)/(step*1.)
WRITE(11,100) 'Principle Hugoniot Calculation'
if(nEOS(2).eq.1.or.nEOS(2).eq.2) then
  WRITE(11,104) rho0, C0, s1, gamma0
  if(nEOS(2).eq.2) then
    WRITE(11,105) a(1), a(2), a(3), a(4), a(5)
  endif
endif
WRITE(11,102) 'v[cm3/g]', 'P[Mbar]', 'Up[cm/micro-s]',
&           'Us[cm/micro-s]'
dsmall=0.99999
if(Up.eq.0.) then
  Up=0.1
endif
temp(1:3)=0.0
2100 do i=0,step
  Cu=L+delta*i
  dump=0
  error=0.01
  ierror=1.0
  do while (ierror.ge.error)
    Peos=0.0
    junk(1:10)=0.0
    E1=(Up**2./2.)
    write(9,*) (Up**2)/(v0-cu), E1
    if(nEOS(1).eq.1) then
      call mgrprop(Peos,E1,cu,junk(2),junk(3),temp)
    elseif(nEOS(1).eq.2) then
      call tillprop(Peos,E1,cu,junk(2),junk(3),temp)
    elseif(nEOS(1).eq.3) then
      call mbprop(Peos,E1,cu,junk(2),junk(3),temp)
    endif
    f=Peos-(Up**2)/(v0-cu)
    write(9,*) cu, Peos, E1, temp(2)
    Up2=dsmall*Up
    E2=(Up2**2./2.)
    junk(1:10)=0.0
  enddo
enddo

```



---

```

      write(9,*) (Up2**2)/(v0-cu), E1
      if(nEOS(1).eq.1) then
        call mgrprop(junk(1),E2,cu,junk(2),junk(3),temp)
      elseif(nEOS(1).eq.2) then
        call tillprop(junk(1),E2,cu,junk(2),junk(3),temp)
      elseif(nEOS(1).eq.3) then
        call mbprop(junk(1),E2,cu,junk(2),junk(3),temp)
      endif
      f2 = junk(1)-(Up2**2)/(v0-cu)
      df = (f2-f)/(Up2-Up)
      Ut=Up-f/df
      write(9,*) cu, Peos, E1, f2, df, Ut
      ierror = abs((Ut-Up)/Up)
      Up=abs(Ut)
    end do
    Up=Ut
    Us=(Up*v0) / (v0-cu)
    P = Up*Us/v0
    E = Up**2/2
    write(11,103) Cu,P, Up, Us
    write(*,103) Cu,P, Up, Us
  enddo
  write(11,*) ''
  write(11,*) ''
  close(11)
  close(9)
  return
100 format(10x,A)
101 format(18x,A)
102 format(16x,A,7x,A,3x,A,3x,A)
103 format(15x,e10.4,4x,e10.4,4x,e10.4,4x,e10.4,4x)
104 format(4x,'rho0: ',e10.4,4x,'c0: ',e10.4,4x,'s1: ',
& e10.4,4x,'g0: ',e10.4)
105 format(4x,'Wilkins Constants: ',e10.4,' : ',e10.4,
& ' : ', e10.4,' : ',e10.4)
  END SUBROUTINE Hugoniot
!-----
! Subroutine: Reshock
!
! DESCRIPTION:
! Calculated reshock state given principle shock strength, and reshock
! strength
!-----
SUBROUTINE reshock()

```

```

use EOS
IMPLICIT NONE
DOUBLE PRECISION up_1, P_1, E_1, V_1
DOUBLE PRECISION up_2, P_2, E_2, V_2
DOUBLE PRECISION L, U, delta, Ut, Cu, E1, E2, Up2
DOUBLE PRECISION f, f2, df, ierror, error, Peos, V2, dsmall
DOUBLE PRECISION, DIMENSION(3) :: temp
DOUBLE PRECISION, DIMENSION(10) :: junk
INTEGER step1, step2, i

open(11, file='reshock.data', access='append')
dsmall = 0.99999
WRITE(* ,101) 'Enter Hugoniot shock specific volume and increments'
READ(* ,*) V_1, step1
WRITE(* ,101) 'Enter reshock specific volume and increments'
READ(* ,*) V_2, step2
write(11,102) 'v[cm3/g]', 'P[Mbar]', 'Up[cm/micro-s]',
& 'Us[cm/micro-s]'
WRITE(* ,*) "Calculating principle hugoniot"
Up = 0.1
temp(1:3) = 0.0
L = V0
U = V_1
delta = (U-L)/(step1*1.0)
do i=1, step1
  Cu = L+delta*i
!   dump(1:10) = 0.0
  error = 0.01
  ierror = 1.0
  do while (ierror.ge.error)
    Peos = 0.0
    E1 = (Up**2./2.)
    if (nEOS(1).eq.1) then
      call mgrprop(Peos, E1, cu, junk(2), junk(3), temp)
    elseif(nEOS(1).eq.2) then
      call tillprop(Peos, E1, cu, junk(2), junk(3), temp)
    elseif(nEOS(1).eq.3) then
      call mbprop(Peos, E1, cu, junk(2), junk(3), temp)
    endif
    f=Peos-(Up**2)/(v0-cu)
!   write(9,*) cu, Peos, E1, temp(2)
    Up2=dsmall*Up
    E2=(Up2**2./2.)
    junk(1:10)=0.0
!   write(9,*) (Up2**2)/(v0-cu), E1

```

---

```

        if(nEOS(1).eq.1) then
            call mgrprop(junk(1),E2,cu,junk(2),junk(3),temp)
        elseif(nEOS(1).eq.2) then
            call tillprop(junk(1),E2,cu,junk(2),junk(3),temp)
        elseif(nEOS(1).eq.3) then
            call mbprop(junk(1),E2,cu,junk(2),junk(3),temp)
        endif
        f2 = junk(1)-(Up2**2)/(v0-cu)
        df = (f2-f)/(Up2-Up)
        Ut=Up-f/df
!       write(9,*) cu, Peos, E1, f2, df, Ut
        ierror = abs((Ut-Up)/Up)
        Up=abs(Ut)
    end do
    Up=Ut
    Us=(Up*v0) / (v0-cu)
    P = Up*Us/v0
    E = Up**2/2
    write(11,103) Cu,P, Up, Us
    write(*,103) Cu,P, Up, Us
enddo
write(*,*) "Principle Hugoniot calculation complete"
write(*,103) Cu,P, Up, Us
write(*,*) "Reshock calculation"
up_1 = Up
P_1 = P
E_1 = E
write(*,*) "Up_0      ,   P_0      ,   E_0"
write(*,*) up_1, P_1, E_1

temp(1:3) = 0.0
L = Cu
U = V_2
delta = (U-L)/(step2*1.0)
Up_2 = 0.01
do i=1, step2
    Cu = L+delta*i
!       dump(1:10) = 0.0
    error = 0.01
    ierror = 1.0
    do while (ierror.ge.error)
        Peos = 0.0
        P_2 = P_1+((Up_1-Up_2)**2.)/(V_1-cu)
!       E1 = (Up**2./2.)
        E1 = E_1 + 0.5*(P_2-P_1)*(V_1-cu)

```

```

    if (nEOS(1).eq.1) then
        call mgrprop(Peos ,E1 ,cu ,junk (2) ,junk (3) ,temp)
    elseif(nEOS(1).eq.2) then
        call tillprop (Peos ,E1 ,cu ,junk (2) ,junk (3) ,temp)
    elseif(nEOS(1).eq.3) then
        call mbprop (Peos ,E1 ,cu ,junk (2) ,junk (3) ,temp)
    endif
    f=Peos-P_2
!      write (9,*) cu , Peos ,P_2, E1
    Up2=dsmall*Up_2
!      E2=(Up2**2./2.)
    P_2 = P_1+((Up_1-Up2)**2.)/(V_1-cu)
    E2 = E_1 + 0.5*(P_2-P_1)*(V_1-cu)
    junk (1:10)=0.0
    write (9,*) P_2, E1
    if(nEOS(1).eq.1) then
        call mgrprop (junk (1) ,E2 ,cu ,junk (2) ,junk (3) ,temp)
    elseif(nEOS(1).eq.2) then
        call tillprop (junk (1) ,E2 ,cu ,junk (2) ,junk (3) ,temp)
    elseif(nEOS(1).eq.3) then
        call mbprop (junk (1) ,E2 ,cu ,junk (2) ,junk (3) ,temp)
    endif
    f2 = junk(1)-P_2
    df = (f2-f)/(Up2-Up_2)
    Ut=Up_2-f/df
    write (9,*) cu , Peos , E1, f2 , df , Ut
    ierror = abs((Ut-Up_2)/Up_2)
    write (9,*) ierror , error
    Up_2=abs(Ut)
end do
Up=Ut
Us=(Up-Up_1)*v0 / (v0-cu)
P = P_1+((Up_1-Up2)**2.)/(V_1-cu)
E = E_1 + 0.5*(P-P_1)*(V_1-cu)
write (11,103) Cu,P, Up, Us
write (*,103) Cu,P, Up, Us
enddo

return
100 format(10x,A)
101 format(18x,A)
102 format(16x,A,7x,A,3x,A,3x,A)
103 format(15x,e10.4,4x,e10.4,4x,e10.4,4x,e10.4,4x)
ENDSUBROUTINE Reshock

```

---

```

!  

!  Subroutine: ISOTHERMS  

!  

!  DESCRIPTION:  

!    Plots/exports isotherms for over a given temperature range and number  

!    of steps

```

---

```

SUBROUTINE ISOTHERM()  

  IMPLICIT NONE  

  DOUBLE PRECISION Ts    ! Start temperature  

  DOUBLE PRECISION Te    ! End temperature  

  integer num            ! number or increments  
  

  WRITE(*,*) 'This Feature has not been added yet'  

  return  
  

END SUBROUTINE ISOTHERM

```

---

```

!  

!  Subroutine: ISOENERGY  

!  

!  DESCRIPTION:  

!    Plots/exports Lines of constant energy

```

---

```

SUBROUTINE ISOENERGY()  

  USE EOS  

  IMPLICIT NONE  

  DOUBLE PRECISION E1      ! ENERGY  

  DOUBLE PRECISION vl,vu  ! start and engine specific energy  

  DOUBLE PRECISION delta,cu  

  DOUBLE PRECISION,DIMENSION(3) :: temp  

  DOUBLE PRECISION,DIMENSION(10):: junk  

  integer num,i ,m        ! number or increments  

  DOUBLE PRECISION level  

!  

!  open(12)  

  open(11, file='isoenergy.dat', access='append')  

  write(11,103) 'v[cm3/g]', 'P[Mbar]', 'E[Mbar-cm3/g]',  

&              'T[K]', 'S[Mbar-cm3/g-K]'  

  temp(1:3) = 0.  

  write(*,101) 'Enter energy state'  

  read(*,*) E1  

  write(*,101) 'Enter lower and upper specific volume'  

  read(*,*) vl,vu  

  write(*,101) 'Enter number of increments to be calculated'

```

---

```

      read(*,*) num
!      delta = (vu-vl)/(num*1.)
      delta = vl
      cu=vl
      do i=0,num
         junk(1:10) = 0.
         if(nEOS(1).eq.1) then
            call mgrprop(junk(1),E1,cu,junk(2),junk(3),temp)
         elseif(nEOS(1).eq.2) then
            call tillprop(junk(1),E1,cu,junk(2),junk(3),temp)
         elseif(nEOS(1).eq.3) then
            call mbprop(junk(1),E1,cu,junk(2),junk(3),temp)
         endif
         write(11,102) cu,junk(1),E1,junk(2),junk(3),nEOS(2)
         m=mod(i,9)
!         write(*,*) 'mode(i,10)',m
!         pause
!         write(12,*) i,m,cu,delta
         if(m.eq.0) then
            delta = cu
         endif
         cu=cu+delta
         if(cu.gt.vu) goto 1000
      enddo
1000 write(11,*) ''
      write(11,*) ''
      close(11)
      return
101 format(18x,A)
102 format(15x,e10.4,4x,e10.4,5x,e10.4,6x,e10.4,6x,e10.4,4x,i)
103 format(16x,A,7x,A,3x,A,3x,A,3x,A)
      END SUBROUTINE ISOENERGY

```

---

```

! Subroutine: Compression Isentrope
!
! Plots/exports compression isentrope starting at V0 to pressure P1
!

```

---

```

SUBROUTINE Comp_Isen(Upper,incr)
  use EOS
  IMPLICIT NONE
  DOUBLE PRECISION Pi, Ei, Po,E2,Eo
  DOUBLE PRECISION cu,l,Upper, V2
  DOUBLE PRECISION f, f2, df, delta ,dsmall
  DOUBLE PRECISION dP2,dP1

```

---

```

DOUBLE PRECISION, dimension(3) :: temp
DOUBLE PRECISION, dimension(10) :: junk

INTEGER          incr , step , i

!
open(11, file='isen_compress.dat', access='append')
l = V0
step=incr
delta = ((v0*1.1)-1)/step*1.
dsmall=0.99999
Ei = 0.00000001
Pi = 0.0
Up = 0.0

WRITE(11,102) 'v[cm3/g]', 'P[Mbar]', 'E[Mbar-cm3/g]',
&            'Ur[cm/micro-s]', 'Region'
write(11,103) l, Pi, Ei, Up, nEOS(2)
write(*,103) l, Pi, Ei, Up
junk(1:10) = 0.0
V2 = l*dsmall
E2 = Ei-(V2-L)*Pi
if(nEOS(1).eq.1) then
    call mgrprop(junk(1), E2, V2, junk(2), junk(3), temp)
elseif(nEOS(1).eq.2) then
    call tillprop(junk(1), E2, V2, junk(2), junk(3), temp)
elseif(nEOS(1).eq.3) then
    call mbprop(junk(1), E2, V2, junk(2), junk(3), temp)
endif
!   write(*,*) Pr, Er, E2, V2, v0, junk(1)
dP1 = (sqrt(-(junk(1)-Pi)/(V2-L)))
!   write(*,*) Pi, junk(1), V2, L, dP1
delta = ((v0*1.1)-1)/step*1.
i=0
Po = 1

cu = 1
do while(cu.gt.Upper)
    i=i+1
    Cu=l-delta*i ! Current location (v)
!!   Calculate Eo using foward difference method
!!   Shouldn't Pr+gamma0/v0(E-Er) be replaced with P ??
    Eo = Ei-delta*(Pi)
    junk(1:10)=0.0
    if(nEOS(1).eq.1) then
        call mgrprop(junk(1), Eo, Cu, junk(2), junk(3), temp)

```

```

elseif(nEOS(1).eq.2) then
  call tillprop(junk(1),Eo,Cu,junk(2),junk(3),temp)
elseif(nEOS(1).eq.3) then
  call mbprop(junk(1),Eo,Cu,junk(2),junk(3),temp)
endif
Po = junk(1)
V2=Cu*dsmall
E2 = Eo-(V2-Cu)*Po
junk(1:10)=0.0
!      Find P2
if(nEOS(1).eq.1) then
  call mgrprop(junk(1),E2,V2,junk(2),junk(3),temp)
elseif(nEOS(1).eq.2) then
  call tillprop(junk(1),E2,V2,junk(2),junk(3),temp)
elseif(nEOS(1).eq.3) then
  call mbprop(junk(1),E2,V2,junk(2),junk(3),temp)
endif
dP2 =(sqrt(-(junk(1)-Po)/(V2-Cu)))
!      write(*,*) Po, junk(1), V2,Cu, dP2
!      pause
Up=Up+0.5*(dP2+dP1)*(delta)
write(11,103) Cu, Po, Eo,Up,nEOS(2)
write(*,103) Cu, Po, Eo, Up
Ei=Eo
Pi=Po
dP1=dP2
enddo
!
close(11)
102 format(16x,A,6x,A,4x,A,3x,A,3x,A)
103 format(15x,e10.4,4x,e10.4,4x,e10.4,6x,e10.4,7x,i1)
END SUBROUTINE Comp_Isen

```

---

```

! Subroutine: ISENTROPE
!
! DESCRIPTION:
! Plots/exports ISENTROP

```

---

```

SUBROUTINE ISENTROPE(low , hig , incr )
  use EOS
  IMPLICIT NONE
  DOUBLE PRECISION low , hig
  DOUBLE PRECISION L, U, Cu, delta
  DOUBLE PRECISION Pi, Po, dump

```



---

```

DOUBLE PRECISION Ei , Eo
DOUBLE PRECISION Ur, dP1,dP2, E2,E1
DOUBLE PRECISION error , ierror , f ,df ,Up2,Peos ,v2 ,dsmall ,Ut ,f2
DOUBLE PRECISION,dimension(3) :: temp
DOUBLE PRECISION,dimension(10) :: junk

INTEGER          step , i , sel , incr

open(11 , file='isen.dat' , access='append')
L=low
U=hig
step=incr
dsmall=0.99999
!      dsmall=1.00001
!      L = v0 ! Lower bound
!      U = 0.203478 ! Upper Bound
!      step = 100
WRITE(11,100) 'Isentrope Calculation'
WRITE(11,*) 'Ploting from ',L,' to ',U
!      Up=0
!      Find initial Hugoniot Pressure given specific volume
error=0.0001
ierror=1.0
Up=0.1
do while (ierror.ge.error)
    Peos=0.0
    junk(1:10)=0.0
    E1=(Up**2./2.)
    if(nEOS(1).eq.1) then
        call mgrprop(Peos ,E1,L,junk(2) ,junk(3) ,temp)
    elseif(nEOS(1).eq.2) then
        call tillprop (Peos ,E1,L,junk(2) ,junk(3) ,temp)
    elseif(nEOS(1).eq.3) then
        call mbprop(Peos ,E1,L,junk(2) ,junk(3) ,temp)
    endif
    f=Peos-(Up**2)/(v0-L)
    Up2=dsmall*Up
    E2=(Up2**2./2.)
    junk(1:10)=0.0
    if(nEOS(1).eq.1) then
        call mgrprop(junk(1) ,E2,L,junk(2) ,junk(3) ,temp)
    elseif(nEOS(1).eq.2) then
        call tillprop (junk(1) ,E2,L,junk(2) ,junk(3) ,temp)
    elseif(nEOS(1).eq.3) then
        call mbprop(junk(1) ,E2,L,junk(2) ,junk(3) ,temp)

```

```

    endif
    f2 = junk(1)-(Up2**2)/(v0-L)
    df = (f2-f)/(Up2-Up)
    Ut=Up-f/df
    ierror = abs((Ut-Up)/Up)
    Up=abs(Ut)
    write(*,*) 'Up: ',Up
enddo
Ur=Up                ! units cm/micro-s
Ei= Ur**2./2.
if(nEOS(1).eq.1) then
    call mgrprop(junk(1),Ei,L,junk(2),junk(3),temp)
elseif(nEOS(1).eq.2) then
    call tillprop(junk(1),Ei,L,junk(2),junk(3),temp)
elseif(nEOS(1).eq.3) then
    call mbprop(junk(1),Ei,L,junk(2),junk(3),temp)
endif
!   Us=(Up*v0)/(v0-L)
!   Pi = Ur*Us/v0      ! units MBar
Pi=junk(1)
write(*,*) L, Pi, low
!
!   pause
!   Ei = Ur**2./2.    ! units MBar-cc/gm
WRITE(11,102) 'v[cm3/g] ', 'P[Mbar] ', 'E[Mbar-cm3/g] ',
&             'Ur[cm/micro-s] ', 'Region '
write(11,103) L,Pi,Ei,Ur,nEOS(2)
write(*,103) L,Pi,Ei,Ur
junk(1:10) = 0.0
V2 = L*dsmall
E2 = Ei-(V2-L)*Pi
if(nEOS(1).eq.1) then
    call mgrprop(junk(1),E2,V2,junk(2),junk(3),temp)
elseif(nEOS(1).eq.2) then
    call tillprop(junk(1),E2,V2,junk(2),junk(3),temp)
elseif(nEOS(1).eq.3) then
    call mbprop(junk(1),E2,V2,junk(2),junk(3),temp)
endif
write(*,*) Pr, Er, E2,V2,v0,junk(1)
dP1 = (sqrt(-(junk(1)-Pi)/(V2-L)))
write(*,*) Pi, junk(1), V2,L, dP1
delta = ((v0*1.1)-1)/step*1.
i=0
Po = 1
do while(Po.gt.0.0001)
    i=i+1

```

---

```

      Cu=L+delta*i ! Current location (v)
!!      Calculate Eo using foward difference method
!!      Shouldn't Pr+gamma0/v0(E-Er) be replaced with P ??
      Eo = Ei-delta*(Pi)
      junk(1:10)=0.0
      if(nEOS(1).eq.1) then
        call mgrprop(junk(1),Eo,Cu,junk(2),junk(3),temp)
      elseif(nEOS(1).eq.2) then
        call tillprop(junk(1),Eo,Cu,junk(2),junk(3),temp)
      elseif(nEOS(1).eq.3) then
        call mbprop(junk(1),Eo,Cu,junk(2),junk(3),temp)
      endif
      if(junk(1).lt.0) then
        goto 1000
      endif
      Po = junk(1)
      V2=Cu*dsmall
      E2 = Eo-(V2-Cu)*Po
      junk(1:10)=0.0
!      Find P2
      if(nEOS(1).eq.1) then
        call mgrprop(junk(1),E2,V2,junk(2),junk(3),temp)
      elseif(nEOS(1).eq.2) then
        call tillprop(junk(1),E2,V2,junk(2),junk(3),temp)
      elseif(nEOS(1).eq.3) then
        call mbprop(junk(1),E2,V2,junk(2),junk(3),temp)
      endif
      dP2=(sqrt(-(junk(1)-Po)/(V2-Cu)))
!      write(*,*) Po, junk(1), V2,Cu, dP2
!      pause
      Ur=Ur+0.5*(dP2+dP1)*(delta)
      write(11,103) Cu, Po, Eo,Ur,nEOS(2)
      write(*,103) Cu, Po, Eo, Ur
      Ei=Eo
      Pi=Po
      dP1=dP2
    enddo
1000 write(11,*) ''
      write(11,*) ''
      close(11)
      return
100 format(10x,A)
101 format(18x,A)
102 format(16x,A,6x,A,4x,A,3x,A,3x,A)
103 format(15x,e10.4,4x,e10.4,4x,e10.4,6x,e10.4,7x,i1)

```

```

104 format(4x,'rho0: ',e10.4,4x,'c0: ',e10.4,4x,'s1: ',
&      e10.4,4x,'g0: ',e10.4)
105 format(4x,'Wilkins Constants: ',e10.4,' : ',e10.4,
&      ' : ', e10.4,' : ',e10.4)

```

**END SUBROUTINE ISENTROPE**

---

```

!  

! Subroutine: HRISENTROPE  

!  

! DESCRIPTION:  

! Plots/exports The release isentrope from a given pressure

```

---

```

SUBROUTINE HRISENTROPE()
  use EOS
  IMPLICIT NONE
  DOUBLE PRECISION L, U, Cu, delta
  DOUBLE PRECISION Pg
  DOUBLE PRECISION v, P1, P2, E1, E2
  DOUBLE PRECISION df, Pi, Po, Ur,dP1,dP2, dump
  DOUBLE PRECISION Ei, Eo, ierror, error, P0
  DOUBLE PRECISION,DIMENSION(10) :: junk
  DOUBLE PRECISION,DIMENSION(3)  :: temp
  INTEGER          step, i, sel

1000 write(*,101) 'Pressure which released from'
     read(*,*) Pg
     if(nEOS(1).eq.1) then
       call mgrFindV(Pg,L)
     elseif(nEOS(1).eq.2) then
       call tilFindV(Pg,L)
     elseif(nEOS(1).eq.3) then
       call mbFindV(Pg,L)
     endif
     write(*,*) 'P:',Pg,'v:',L
     step = 10000
     U=L*10
     CALL ISENTROPE(L,U,step)
     goto 5000
5000 WRITE(*,*) 'Plot again yes (1)/No (2)'
     read(*,*) sel
     if(sel.eq.1) then
       goto 1000
     endif

```

**RETURN**

100 **format**(10x,A)

101 **format**(18x,A)

**END SUBROUTINE** HRISENTROPE

## Appendix B

# Mie-Grüneisen EOS Source Code

The Mie-Grüneisen Module used in the EOS program

```

MODULE MGR
  IMPLICIT NONE

  CONTAINS

  !-----
  ! SUBROUTINE: mgreos
  !
  ! DESCRIPTION:
  !   Calculates the thermodynamic variables using the
  !   Mie-Grüneisen EOS give two thermodynamic variables
  !-----
  SUBROUTINE mgreos(P1, E1, V1, T1, Ss, temp)
    USE EOS
    IMPLICIT NONE
    DOUBLE PRECISION P1, E1, V1, T1, Ss
    DOUBLE PRECISION, DIMENSION(3) :: temp
    DOUBLE PRECISION f, df, cu, error, ierror
    DOUBLE PRECISION P2, E2, V2, T2, S2
    !   temp has additional arguments that may not be necessary
    DOUBLE PRECISION, DIMENSION(5) :: junk
    INTEGER, DIMENSION(5) :: num
    !   DOUBLE PRECISION eta, mu

    !
    !   Checking for non zero variables
    !
    num(1:5) = 0
    if (P1.ne.0) then
      num(1)=1
    
```

```
endif
if(E1.ne.0) then
    num(2)=1
endif
if(V1.ne.0) then
    num(3)=1
endif
if(T1.ne.0) then
    num(4)=1
endif
if(Ss.ne.0) then
    num(5)=1
endif
!
! Finding the two give thermodynamic variables
!
if(num(1).eq.1) then
    if(num(2).eq.1) then
! Pressure and Energy
        goto 1000
    elseif(num(3).eq.1) then
! Pressure and Volume
        goto 1001
    elseif(num(4).eq.1) then
! Pressure and Temperature
        goto 1002
    elseif(num(5).eq.1) then
! Pressure and Entropy
        goto 1003
    else
        write(*,*) 'Two thermodynamic variables must be specified'
        return
    endif
elseif(num(2).eq.1) then
    if(num(3).eq.1) then
! Energy and Volume
        goto 1004
    elseif(num(4).eq.1) then
! Energy and Temperature
        goto 1005
    elseif(num(5).eq.1) then
! Energy and Entropy
        goto 1006
    else
        write(*,*) 'Two thermodynamic variables must be specified'
```

---

```

        return
    endif
elseif(num(3).eq.1) then
    if(num(4).eq.1) then
!       Volume and Temperature
        goto 1007
    elseif(num(5).eq.1) then
!       Volume and Entropy
        goto 1008
    else
        write(*,*) 'Two thermodynamic variables must be specified'
        return
    endif
elseif(num(4).eq.1) then
    if(num(5).eq.1) then
!       Temperature and Entropy
        goto 1009
    else
        write(*,*) 'Two thermodynamic variables must be specified'
        return
    endif
else
    write(*,*) 'Two thermodynamic variables must be specified'
    return
endif
!
!   Solving the MGR EOS using given Pressure and Energy
!   to find specific volume, temperature and entropy
1000 if(temp(1).ne.0) then
        cu=temp(1)    ! Initial guess
    else
        cu=v0*0.25
    endif

    error=0.01
    ierror=1.0

    do while (ierror.ge.error)
        call hugoRef(cu)
        P2=Pr+gamma0/v0*(E1-Er)
        f=P1-P2
        V2=cu*0.9999
        call hugoRef(V2)
        P2=Pr+gamma0/v0*(E1-Er)
        junk(1)=P1-P2
    
```



---

```

    df = (junk(1)-f)/(V2-cu)
    cu=cu-f/df
    call hugoRef(cu)
    P2 =Pr+gamma0/v0*(E1-Er)
    ierror=abs(P1-P2)/P1
enddo
!   Found the V that satisfies the given P and E
V1=cu
!
!   Need to figure out how to find entropy and Temperature
!
return
!
!   Solving the MGR EOS using given Pressure and Volume
!   to find entergy, temperature and entropy
!
1001 call hugoRef(V1)
if(temp(1).ne.0) then
    cu=temp(1)    ! Initial guess
else
    cu=Er    ! Reference Energy
endif

error=0.01
ierror=1.0

do while (ierror.ge.error)
!   hugoRef(V1)
    P2=Pr+gamma0/v0*(cu-Er)
    f=P1-P2
!   Need to replace with analytical expression
    E2=cu*0.9999
    P2 = Pr+gamma0/v0*(E2-Er)
    junk(1)=P1-P2
    df = (junk(1)-f)/(E2-cu)
    cu=cu-f/df
    P2 = Pr+gamma0/v0*(cu-Er)
    ierror=abs(P1-P2)/P1
enddo
!   Found the E that satisfies the given P and v
E1=cu
!
!   Need to figure out how to find entropy and Temperature
!
return

```

---

```
!  
!   Solving the MGR EOS using given Pressure and Temperature  
!   to find energy, specific volume, and entropy  
!  
1002 write(*,*) 'This Feature needs to be added'  
      return  
!  
!   Solving the MGR EOS using given Pressure and Entropy  
!   to find energy, specific volume, and temperature  
!  
1003 write(*,*) 'This Feature needs to be added'  
      return  
!  
!   Solving the MGR EOS using given Energy and Volume  
!   to find pressure, temperature and entropy  
!  
1004 call hugoRef(V1)  
      P1=Pr+gamma0/v0*(E1-Er)  
      return  
!  
!   Solving the MGR EOS using given Energy and Temperature  
!   to find pressure, specific volume, and entropy  
!  
1005 write(*,*) 'This Feature needs to be added'  
      return  
!  
!   Solving the MGR EOS using given Energy and Entropy  
!   to find pressure, specific volume, and temperature  
!  
1006 write(*,*) 'This Feature needs to be added'  
      return  
!  
!   Solving the MGR EOS using given Volume and Temperature  
!   to find pressure, energy, and entropy  
!  
1007 write(*,*) 'This Feature needs to be added'  
      return  
!  
!   Solving the MGR EOS using given Volume and Entropy  
!   to find pressure, energy, and temperature  
!  
1008 write(*,*) 'This Feature needs to be added'  
      return  
!  
!   Solving the MGR EOS using given Temperature and Entropy
```

---

```

!      to find pressure, energy, and specific volume
!
1009 write(*,*) 'This Feature needs to be added'
      return

      END SUBROUTINE

      SUBROUTINE hugoRef(vi)
      USE EOS
      IMPLICIT NONE
      DOUBLE PRECISION vi, x, x2, E2

      if(nEOS(2).eq.1) then
        goto 1000
      elseif(nEOS(2).eq.2) then
        goto 2000
      elseif(nEOS(2).eq.3) then
        goto 3000
      endif
1000 Pr = (C0**2*(v0-vi))/((v0-s1*(v0-vi))**2)
      Er = 0.5*Pr*(v0-vi)
      goto 5000
2000 x = 1.-(vi/v0)
      Er =(a(1)+a(2)*x+a(3)*x**2.+a(4)*x**3.+a(5)*x**4.)
      Pr =(a(2)+2*a(3)*x+3*a(4)*x**2.+4*a(5)*x**3.)*rho0
      goto 5000
3000 x = v0/vi
      Er = 3*v0*(a(1)*(x**(1./3.)-1)+a(2)/2.*(x**(2./3.)-1)
& + a(3)/3.*(x-1)+a(4)/4.*(x**(4./3.)-1)+a(5)/5.*(x**(5./3.)-1))
      Er = Er/100000
      Pr=(v0**(4./3.)*(a(1)*vi**(4./3.)+a(2)*vi*v0**(1./3.)+
& a(3)*(vi*v0)**(2./3)+a(4)*vi**(1./3.)*v0+
& a(5)*v0**(4./3.)))/vi**(8./3.)
      Pr = Pr/100
      goto 5000
5000 return
      END SUBROUTINE

      SUBROUTINE eosINP(sel)
      USE EOS
      IMPLICIT NONE
      integer sel
      nEOS(1) = sel ! MGR EOS Parameters

```

```

WRITE(*,100) '-----MENU-----',
WRITE(*,101) '(1) Enter Hugoniot Paramters',
WRITE(*,101) '(2) Enter Wilkin Energy Parameters',
WRITE(*,101) '(3) Elastic Cold Curve',
WRITE(*,101) '(5) Exit System'
read(*,*) sel
nEOS(2) = sel ! Reference Curve Definition
if(sel.eq.1.or.sel.eq.2) then
  write(*,100) 'Enter rho0,C0, s1, and gamma0'
  read(*,*) rho0, C0, s1, gamma0
  v0 = 1./rho0
  if(sel.eq.2) then
    write(*,100) 'Enter e00, e01, e02, e03, e04',
    write(*,100) 'see Wilkins pgs 60-63'
    read(*,*) a(1), a(2), a(3), a(4), a(5)
  endif
elseif(sel.eq.3) then
  write(*,100) 'Cold curve approximation',
  write(*,100) 'Enter v0, gamma0, a1,a2,a3,a4,a5',
  write(*,100) 'see Lomonosov',
  read(*,*) v0, gamma0,a(1),a(2),a(3),a(4),a(5)
  rho0 = 1./v0
endif
RETURN
100 format(15x, '-----',A)
101 format(18x,A)
END SUBROUTINE

```

```

! SUBROUTINE hugoProp(vi)
! USE EOS
! IMPLICIT NONE
! DOUBLE PRECISION vi
! CALL hugoRef(vi)
! Up = sqrt(((Pr-(gamma0/v0)*Er)/(1/(v0-vi)-gamma0/(2*v0))))
! Up = Up*10
! Us = (Up*v0) / (v0-vi)
! P = Up*Us/v0*0.01
! E = 0.5*P*(v0-vi)
! return
! END SUBROUTINE

```

```

SUBROUTINE findV(Pg,vout)
USE EOS

```

---

```

IMPLICIT NONE
DOUBLE PRECISION Pg, f, df, Peos, Peos2, f2, vout
DOUBLE PRECISION ierror, error, cu, P1, E1

error= 0.001
ierror = 1.0

cu=v0*0.5
do while(ierror.ge.error)
!   Up=sqrt(Pg*(v0-cu))
!   E1=(Up**2)/2.
CALL hugoRef(cu)
Up=sqrt(Pg*(v0-cu))
E1=(Up**2)/2.
Peos = Pr+(gamma0/v0)*(E1-Er)
f=Pg-Peos
Up=sqrt(Pg*(v0-cu*0.9999))
E1=(Up**2)/2.
CALL hugoRef(cu*0.9999)
Peos2 = Pr+(gamma0/v0)*(E1-Er)
f2 = Pg-Peos2
df= (f2-f)/(cu*0.9999-cu)
cu = cu -f/df
CALL hugoRef(cu)
Up=sqrt(Pg*(v0-cu))
E1=(Up**2)/2.
P1=Pr+(gamma0/v0)*(E1-Er)
ierror = abs((P1-Pg)/Pg)
end do
vout = cu
write(*,*) vout, Pg, P1
pause
return
END SUBROUTINE

END MODULE

```

# Appendix C

## Tillotson EOS Source Code

The Tillotson Module used in the EOS program.

```

MODULE TILLOTSON
  IMPLICIT NONE

  CONTAINS

!-----
!  SUBROUTINE:  tilleos
!
!  DESCRIPTION:
!      Caculaties the thermodynamic variables using the
!      Tillotson EOS give two thermodynamic variables
!-----

SUBROUTINE tilleos (P1, E1, V1, T1, Ss, temp)
  USE EOS
  IMPLICIT NONE
  DOUBLE PRECISION P1, E1, V1, T1, Ss
  DOUBLE PRECISION, DIMENSION(3)  :: temp
  DOUBLE PRECISION f, df, cu, error, ierror
  DOUBLE PRECISION P2, E2, V2, T2, S2
!   temp has additional arguments that may not be necessary
  DOUBLE PRECISION, DIMENSION(5)  :: junk
  INTEGER, DIMENSION(5)  :: num
  DOUBLE PRECISION eta, mu

!
!  Checking for non zero variables
!

  num(1:5) = 0
  if (P1.ne.0) then
    num(1)=1

```

```
endif
if(E1.ne.0) then
    num(2)=1
endif
if(V1.ne.0) then
    num(3)=1
endif
if(T1.ne.0) then
    num(4)=1
endif
if(Ss.ne.0) then
    num(5)=1
endif
!
! Finding the two give thermodynamic variables
!
if(num(1).eq.1) then
    if(num(2).eq.1) then
! Pressure and Energy
        goto 1000
    elseif(num(3).eq.1) then
! Pressure and Volume
        goto 1001
    elseif(num(4).eq.1) then
! Pressure and Temperature
        goto 1002
    elseif(num(5).eq.1) then
! Pressure and Entropy
        goto 1003
    else
        write(*,*) 'Two thermodynamic variables must be specified'
        return
    endif
elseif(num(2).eq.1) then
    if(num(3).eq.1) then
! Energy and Volume
        goto 1004
    elseif(num(4).eq.1) then
! Energy and Temperature
        goto 1005
    elseif(num(5).eq.1) then
! Energy and Entropy
        goto 1006
    else
        write(*,*) 'Two thermodynamic variables must be specified'
```

---

```

        return
    endif
elseif(num(3).eq.1) then
    if(num(4).eq.1) then
!       Volume and Temperature
        goto 1007
    elseif(num(5).eq.1) then
!       Volume and Entropy
        goto 1008
    else
        write(*,*) 'Two thermodynamic variables must be specified'
        return
    endif
elseif(num(4).eq.1) then
    if(num(5).eq.1) then
!       Temperature and Entropy
        goto 1009
    else
        write(*,*) 'Two thermodynamic variables must be specified'
        return
    endif
else
    write(*,*) 'Two thermodynamic variables must be specified'
    return
endif
1000 if(temp(1).ne.0) then
    cu=temp(1)    ! Initial guess
else
    cu=v0*0.25
endif

error=0.01
ierror=1.0

!
! Need to add logic for adding multi-phase check
!

eta = v0/cu
mu = eta-1
do while (ierror.ge.error)
    CALL findP (P2,E1,cu)
    f=P1-P2
    V2=cu*0.9999
    eta =v0/V2
    mu=eta-1
    CALL findP (P2,E1,V2)

```



---

```

    junk(1)=P1-P2
    df = (junk(1)-f)/(V2-cu)
    cu=cu-f/df
    eta = v0/cu
    mu = eta-1
    CALL findP (P2,E1,cu)
    ierror=abs(P1-P2)/P1
enddo
!   Found the V that satisfies the given P and E
V1=cu
!
!   Need to figure out how to find entropy and Temperature
!
return
!
!   Solving the Tillotson EOS using given Pressure and Volume
!   to find entergy, temperature and entropy
!
1001 if(temp(1).ne.0) then
        cu=temp(1)    ! Initial guess
    else
        cu=ti(1)*0.25    ! Eo
    endif

    error=0.01
    ierror=1.0

    eta = v0/V1
    mu = eta-1
do while (ierror.ge.error)
    CALL findP (P2,cu,V1)
    f=P1-P2
    E2=cu*0.9999
    CALL findP (P2,E2,V1)
    junk(1)=P1-P2
    df = (junk(1)-f)/(E2-cu)
    cu=cu-f/df
    CALL findP (P2,cu,V1)
    ierror=abs(P1-P2)/P1
enddo
!   Found the E that satisfies the given P and v
E1=cu
!
!   Need to figure out how to find entropy and Temperature
!
```

---

```

    return
!   Solving the Tillotson EOS using given Pressure and Temperature
!   to find energy, specific volume, and entropy
!
!
1002 write(*,*) 'This Feature needs to be added'
    return
!
!   Solving the Tillotson EOS using given Pressure and Entropy
!   to find energy, specific volume, and temperature
!
!
1003 write(*,*) 'This Feature needs to be added'
    return
!
!   Solving the Tillotson EOS using given Energy and Volume
!   to find pressure, temperature and entropy
!
!
1004 CALL findP(P1,E1,V1)

    return
!
!   Solving the Tillotson EOS using given Energy and Temperature
!   to find pressure, specific volume, and entropy
!
!
1005 write(*,*) 'This Feature needs to be added'
    return
!
!   Solving the Tillotson EOS using given Energy and Entropy
!   to find pressure, specific volume, and temperature
!
!
1006 write(*,*) 'This Feature needs to be added'
    return
!
!   Solving the Tillotson EOS using given Volume and Temperature
!   to find pressure, energy, and entropy
!
!
1007 write(*,*) 'This Feature needs to be added'
    return
!
!   Solving the Tillotson EOS using given Volume and Entropy
!   to find pressure, energy, and temperature
!
!
1008 write(*,*) 'This Feature needs to be added'
    return
!
```

---

```

!      Solving the Tillotson EOS using given Temperature and Entropy
!      to find pressure, energy, and specific volume
!
1009 write(*,*) 'This Feature needs to be added'
      return
END SUBROUTINE

SUBROUTINE findP(P1,E1,V1)
USE EOS
IMPLICIT NONE
DOUBLE PRECISION P1,E1,V1
DOUBLE PRECISION check, mu,eta

      check = V1/v0
      eta = v0/V1
      mu = eta -1.

      if(check.ge.ti(3)) then
          nEOS(2)=4
      elseif((check.gt.1).and.(E1.gt.ti(2))) then
          nEOS(2)=4
      else
          nEOS(2)=3
      endif

      if(nEOS(2).eq.3) then
          P1 = (ti(4)+ti(5)/(E1/(ti(1)*eta**2.)
&          +1.))*E1/V1+ti(6)*mu+ti(7)*mu**2.
      elseif(nEOS(2).eq.4) then
          P1 =ti(4)*E1/V1+(ti(5)*E1/(V1*(E1/(ti(1)*(eta**2.))+1.))+
&          ti(6)*mu*exp(-ti(9)*(V1/v0-1.))*exp(-ti(8)*((V1/v0)-1.))**2.)
      endif
      RETURN
END SUBROUTINE

SUBROUTINE eosINP(sel)
USE EOS
IMPLICIT NONE
integer sel
nEOS(1) = sel ! Tillotson EOS Parameters
write(* ,100) 'Tillotson EOS Parameters'
write(* ,100) 'Enter rho0 ,E0 ,Es ,Vs'
read(* ,*) rho0, ti(1), ti(2), ti(3)

write(* ,100) 'Enter a,b,A,B'

```

---

```

    read(*,*) ti(4),ti(5),ti(6),ti(7)
    write(*,100) 'Enter Alpha and Beta'
    read(*,*) ti(8),ti(9)
    v0 = 1./rho0
    RETURN
100 format(15x, '————', A)
    END SUBROUTINE

    SUBROUTINE findV(Pg, vout)
    USE EOS
    IMPLICIT NONE
    DOUBLE PRECISION Pg, f, df, Peos, Peos2, f2, vout
    DOUBLE PRECISION ierror, error, cu
    DOUBLE PRECISION eta, mu, eta2, mu2, vi2

    error=0.001
    ierror=1.0

    cu=v0*0.25

    do while(ierror.ge.error)
        eta=v0/cu
        mu=eta-1
        Up=sqrt(Pg*(v0-cu))
        E=(Up**2)/2.
        Peos = (ti(4)+ti(5)/(E/(ti(1)*eta**2)
&          +1))*(E/cu)+ti(6)*mu+ti(7)*mu**2
        f=Pg-Peos
        vi2=cu*0.9999
        eta2=v0/vi2
        mu2=eta2-1
        Up=sqrt(Pg*(v0-vi2))
        E=(Up**2)/2.
        Peos2 = (ti(4)+ti(5)/(E/(ti(1)*eta2**2)
&          +1))*(E/vi2)+ti(6)*mu2+ti(7)*mu2**2
        f2=Pg-Peos2
        df = (f2-f)/(vi2-cu)
        cu = cu - f/df
        P = (ti(4)+ti(5)/(E/(ti(1)*eta**2)
&          +1))*(E/cu)+ti(6)*mu+ti(7)*mu**2
!          write(*,*) cu, P, Pg
        ierror = abs((P-Pg)/Pg)
    enddo
    vout = cu

```

```
RETURN  
END SUBROUTINE
```

```
END MODULE
```

## Appendix D

# Multi-Branch Analytical EOS Source Code

The Multi-Branch Analytical Module used in the EOS program.

```
MODULE MBEOS
IMPLICIT NONE
```

```
CONTAINS
```

---

```
!
! CURRENT VALUES
! rho0 = 2.712 g/cm^3
! T0 = 298 K
! Pc = 0.0018202 Mbar
! Ec = 0.122 Mbar-cm^3/g
! E1 = 0.01 Mbar-cm^3/g! ~1 kJ/g This is esencietly the bottom of
! the sat dome
! gamma0 = 2.14
! c0 = 0.54518
! s = 1.2592
! m = 8 ! not correct
! n = 0.7 ! not correct
! k = 1 !assumed since i don't have gas data to compare
! !k is a fitting paramter and is only used in
! !the gas region
! xhi = 2/3 !No internal degree of freedome
! R = 3.08173e-6 (Mbar-cm^3)/(g-K)
!
```

---



---

```
!
```

---

```

! SUBROUTINE: multieos
!
! DESCRIPTION:
!   Caculaties the thermodynamic variables using the
!   Multi-branch analytical EOS give two thermodynamic
!   variables
!
!   The first attempt at this is using the models and
!   presentation shown in Gathers. This will probable
!   need to be tuned for Aluminum (the original work
!   was developed for Lithium)
!
!
SUBROUTINE multieos(P1,E1,V1,T1,Ss,temp)
USE EOS
IMPLICIT NONE
DOUBLE PRECISION P1, E1, V1, T1, Ss
DOUBLE PRECISION,DIMENSION(3) :: temp
DOUBLE PRECISION f,df,cu, error,ierror
DOUBLE PRECISION P2,E2,V2,T2,S2
!   temp has additional arguments that may not be necessary
DOUBLE PRECISION,DIMENSION(5) :: junk
INTEGER,DIMENSION(5) :: num
DOUBLE PRECISION eta,mu
!
! Checking for non zero variables
!
num(1:5) = 0
if(P1.ne.0) then
  num(1)=1
endif
if(E1.ne.0) then
  num(2)=1
endif
if(V1.ne.0) then
  num(3)=1
endif
if(T1.ne.0) then
  num(4)=1
endif
if(Ss.ne.0) then
  num(5)=1
endif
!

```

---

---

```
!      Finding the two give thermodynamic variables
!  
      if(num(1).eq.1) then
          if(num(2).eq.1) then
!<              Pressure and Energy
                goto 1000
          elseif(num(3).eq.1) then
!<              Pressure and Volume
                goto 1001
          elseif(num(4).eq.1) then
!<              Pressure and Temperature
                goto 1002
          elseif(num(5).eq.1) then
!<              Pressure and Entropy
                goto 1003
          else
              write(*,*) 'Two thermodynamic variables must be specified'
              return
          endif
      elseif(num(2).eq.1) then
          if(num(3).eq.1) then
!<              Energy and Volume
                goto 1004
          elseif(num(4).eq.1) then
!<              Energy and Temperature
                goto 1005
          elseif(num(5).eq.1) then
!<              Energy and Entropy
                goto 1006
          else
              write(*,*) 'Two thermodynamic variables must be specified'
              return
          endif
      elseif(num(3).eq.1) then
          if(num(4).eq.1) then
!<              Volume and Temperature
                goto 1007
          elseif(num(5).eq.1) then
!<              Volume and Entropy
                goto 1008
          else
              write(*,*) 'Two thermodynamic variables must be specified'
              return
          endif
      elseif(num(4).eq.1) then
```



```

        if(num(5).eq.1) then
!      Temperature and Entropy
          goto 1009
        else
          write(*,*) 'Two thermodynamic variabls must be specified'
          return
        endif
      else
        write(*,*) 'Two thermodynamic variabls must be specified'
        return
      endif
!
!      Solving the Tillotson EOS using given Pressure and Energy
!      to find specific volume, temperature and entropy
1000 if(temp(1).ne.0) then
        cu=temp(1)    ! Initial guess
      else
        cu=v0*0.25
      endif

      error=0.01
      ierror=1.0
!
!      Need to add logic for adding multi-phase check
!
      do while (ierror.ge.error)
        CALL findP (P2,E1,cu)
        f=P1-P2
        V2=cu*0.9999
        CALL findP (P2,E1,V2)
        junk(1)=P1-P2
        df = (junk(1)-f)/(V2-cu)
        cu=cu-f/df
        CALL findP (P2,E1,cu)
        ierror=abs(P1-P2)/P1
      enddo
!      Found the V that satisfies the given P and E
      V1=cu
      if(nEOS(2)==1) then ! Solid Phase
        junk(1) = -3*mb(9)*mb(1)
        junk(2) = 1-V1/v0
        junk(3) = junk(1)+gamma0*junk(1)*junk(2)+
&      0.5*(c0**2.+gamma0**2.*junk(1))*junk(2)**2.+
&      1./6.*(4*s1*C0**2.+gamma0**3.*junk(1))*junk(2)**3.
        T1=(E1-junk(3))/(3*mb(9))

```

---

```

    elseif(nEOS(2)==2) then ! Expanded phase, E greater than Ec
!       T1 = Tg+(v0/V1)**0.2*(Ts-Tg)
    elseif(nEOS(2)==3) then ! Interpolation region
!       ! E greater than E1 (what is E1???)

    else ! nEOS(2) =4 ! below dome

    endif

!
!   Need to figure out how to find entropy and Temperature
!
    junk(1:10) = 0.
    return

!
!   Solving the Tillotson EOS using given Pressure and Volume
!   to find entergy, temperature and entropy
!
1001 if(temp(1).ne.0) then
        cu=temp(1) ! Initial guess
    else
        cu=ti(1)*0.25 ! Eo
    endif

    error=0.01
    ierror=1.0

    do while (ierror.ge.error)
        CALL findP (P2,cu,V1)
        f=P1-P2
!       Need to replace with analytical expression
        E2=cu*0.9999
        CALL findP (P2,E2,V1)
        junk(1)=P1-P2
        df = (junk(1)-f)/(E2-cu)
        cu=cu-f/df
        CALL findP (P2,cu,V1)
        ierror=abs(P1-P2)/P1
    enddo

!   Found the E that satisfies the given P and v
    E1=cu

!
!   Need to figure out how to find entropy and Temperature
!
    return

```

---

```
!  
!   Solving the Tillotson EOS using given Pressure and Temperature  
!   to find energy, specific volume, and entropy  
!  
1002 write(*,*) 'This Feature needs to be added'  
      return  
!  
!   Solving the Tillotson EOS using given Pressure and Entropy  
!   to find energy, specific volume, and temperature  
!  
1003 write(*,*) 'This Feature needs to be added'  
      return  
!  
!   Solving the Tillotson EOS using given Energy and Volume  
!   to find pressure, temperature and entropy  
!  
1004 CALL findP(P1,E1,V1)  
  
      return  
!  
!   Solving the Tillotson EOS using given Energy and Temperature  
!   to find pressure, specific volume, and entropy  
!  
1005 write(*,*) 'This Feature needs to be added'  
      return  
!  
!   Solving the Tillotson EOS using given Energy and Entropy  
!   to find pressure, specific volume, and temperature  
!  
1006 write(*,*) 'This Feature needs to be added'  
      return  
!  
!   Solving the Tillotson EOS using given Volume and Temperature  
!   to find pressure, energy, and entropy  
!  
1007 write(*,*) 'This Feature needs to be added'  
      return  
!  
!   Solving the Tillotson EOS using given Volume and Entropy  
!   to find pressure, energy, and temperature  
!  
1008 write(*,*) 'This Feature needs to be added'  
      return  
!  
!   Solving the Tillotson EOS using given Temperature and Entropy
```

```

!      to find pressure, energy, and specific volume
!
1009 write(*,*) 'This Feature needs to be added'
      return
      END SUBROUTINE

SUBROUTINE findP (P1,E1,V1)
USE EOS
IMPLICIT NONE
DOUBLE PRECISION P1,E1,V1
DOUBLE PRECISION check , mu, eta
DOUBLE PRECISION ph, x
DOUBLE PRECISION,DIMENSION(10)      :: junk
!      open(12,access='append')
      if(V1.lt.v0) then
          nEOS(2) = 1      ! Solid Phase ~ using MGR
      elseif(E1.gt.mb(3)) then
          nEOS(2) = 2      ! Expanded phase, E greater than Ec
      elseif(E1.gt.mb(4)) then
          nEOS(2) = 3      ! Interpolation region
                          ! E greater than E1 (what is E1???)
      else
          nEOS(2) = 4      ! below dome
      endif

      if(nEOS(2).eq.1) then
!      There's probable better ways to do this but this is how it's presented
!      in gathers, optimization comes second
!      MGR eos for compression (solid) region (region 1)
          x = 1-V1/v0
          ph= rho0*c0**2.*x/(1.-S1*x)**2.
          P1=ph*(1-gamma0*x/2.)+gamma0*rho0*E1
      elseif(nEOS(2).eq.2) then
!      Gas Model (region 2)
!      write(12,*) V1, E1, mb(3), mb(8), mb(7),v0
          P1=mb(8)*(E1-mb(3))/V1+(gamma0/v0*E1-mb(8)*(E1-mb(3))/v0)
          &      *(v0/V1)**mb(7)/V1

      elseif(nEOS(2).eq.3) then
!      Interpolation Region 3
          junk(1) = rho0*c0*(v0/V1-1.)+gamma0*rho0*mb(4)      ! p41
          junk(2) = c0**2.                                     ! dpr41
          junk(3) = 0.0                                       ! dpe41
          junk(4) = mb(2)*exp(mb(5))*exp(-mb(6)*mb(3)/E1)    ! p511
          junk(5) = 0.0                                       ! dpr511

```

---

```

junk(6) = junk(4)*mb(6)*mb(3)/E1**2.          ! dpe511
junk(7) = gamma0*rho0*E1*(v0/V1)**mb(7)       ! p512
junk(8) = junk(7)*mb(7)*V1                    ! dpr512
junk(9) = junk(7)/E1                          ! dpe512
if (junk(4).gt.junk(7)) then                 ! p511>p512
    junk(4)=junk(7)                          ! now p51
    junk(5)=junk(8)                          ! dpr51
    junk(6)=junk(9)                          ! dpe51
endif
junk(7)=gamma0*rho0*mb(3)*(v0/V1)**mb(7)     !p2
junk(8)=junk(7)*mb(7)*V1                    !pdr2
junk(9)=0.0                                  !pde2
if(junk(1).gt.junk(4)) then                 !p41>p51
    P1=junk(1)+(E1-mb(4))*(junk(7)-junk(1))/(mb(3)-mb(4))
else
    P1=junk(4)+(E1-mb(4))*(junk(7)-junk(4))/(mb(3)-mb(4))
endif
elseif(nEOS(2).eq.4) then
junk(1)=mb(2)*exp(mb(5))*exp(-mb(6)*mb(3)/E1) !p51
junk(2)=gamma0*rho0*E1*(v0/v1)**mb(7)       !p52
if (junk(1).gt.junk(2)) then               !p51>p52
    junk(1) = junk(2)                       !p52=p5
endif
!
!
!
!
junk(2)=rho0*c0**2*(v0/V1-1)+gamma0*rho0*E1  !p4
if (junk(2).gt.junk(1)) then
    P1=junk(2)
else
    P1=junk(1)
endif
endif
junk(1:10)=0
RETURN
END SUBROUTINE
!
SUBROUTINE eosINP(sel)
USE EOS
IMPLICIT NONE
integer sel
nEOS(1) = sel ! Tillotson EOS Parameters
write(*,100) 'Multi=Branch EOS Parameters '
write(*,100) 'Enter rho0, gamma0, T0'
!
!
    read(*,*) rho0,gamma0, mb(1)
rho0 = 2.712
gamma0=2.14

```

```

mb(1) = 298
write(*,*) rho0,gamma0,mb(1)
write(*,100) 'Enter Pc,Ec,E1,C0,S'
!   read(*,*) mb(2),mb(3),mb(4),C0,S1
mb(2) = 0.0018202
mb(3) = 0.122
!   mb(4) = 0.01
mb(4) = mb(3) ! Still not sure what E1 does
C0 = 0.54518
s1 = 1.2592
write(*,*) mb(2),mb(3),mb(4),C0,S1
write(*,100) 'Enter m, n, k, xi, R'
!   read(*,*) mb(5),mb(6),mb(7),mb(8),mb(9)
mb(5) = 1.0      ! wrong
mb(6) = 1.1     ! wrong
mb(7) = 1.1
mb(8) = 2./3.
mb(9) = 3.08173e-6
write(*,*) mb(5),mb(6),mb(7),mb(8),mb(9)
v0 = 1./rho0
write(*,*) v0
RETURN
100 format(15x,'————',A)
END SUBROUTINE

```

```

SUBROUTINE findV(Pg,vout)
USE EOS
IMPLICIT NONE
DOUBLE PRECISION Pg,f,df,Peos,Peos2,f2,vout
DOUBLE PRECISION ierror,error,cu,E1
DOUBLE PRECISION eta,mu,eta2,mu2,vi2

error=0.001
ierror=1.0

cu=v0*0.25

!   Up=sqrt(Pg*(v0-cu))
!   E1=(Up**2)/2.
!   Peos = (ti(4)+ti(5))/(E/(ti(1)*eta**2)
!   &      +1))*(E/cu)+ti(6)*mu+ti(7)*mu**2
!   call findP(Peos,E1,cu)

do while(ierror.ge.error)
    Up=sqrt(Pg*(v0-cu))

```

---

```

      E1=(Up**2)/2.
      call findP (Peos ,E1 ,cu)
      f=Pg-Peos
      vi2=cu*0.9999
      Up=sqrt (Pg*(v0-vi2))
      E1=(Up**2)/2.
!      Peos2 = (ti(4)+ti(5))/(E/(ti(1)*eta**2)
!      &      +1))*(E/vi2)+ti(6)*mu2+ti(7)*mu2**2
      call findP (Peos2 ,E1 ,vi2)
      f2=Pg-Peos2
      df = (f2-f)/(vi2-cu)
      cu = cu -f/df
      Up=sqrt (Pg*(v0-cu))
      E1=(Up**2)/2.
!      write(*,*) 'Before hugoProp call '
!      P = (ti(4)+ti(5))/(E/(ti(1)*eta**2)
!      &      +1))*(E/cu)+ti(6)*mu+ti(7)*mu**2
      call findP (Peos ,E1 ,cu)
      write(*,*) cu ,Peos ,Pg
!      pause
      ierror = abs((Peos-Pg)/Pg)
enddo
      vout = cu

RETURN
END SUBROUTINE

END MODULE

```

## Appendix E

# Bushman-Lomonosov EOS Source Code

The Bushman-Lomonosov EOS was calculated using a separate program that the other EOSs. The complete program is listed below.

```

PROGRAM BushmanEOS
!-----
!  

! This program calculates shock hugoniot, release isentrop  

! end points using bushman EOS and  

! gibbs relationship  

!  

!-----
IMPLICIT NONE
DOUBLE PRECISION :: V1, vu, vl, vg, Ps
DOUBLE PRECISION :: T1, P1, E1, Ss, G
DOUBLE PRECISION :: vliq, vgas, Pg
DOUBLE PRECISION :: P2, gl, gg, dg1, dg2, dp, dg
DOUBLE PRECISION :: cu, Ptemp, f1, df, dv
DOUBLE PRECISION :: Tl, Tu
DOUBLE PRECISION :: delta
DOUBLE PRECISION :: error
DOUBLE PRECISION :: ierror, ierror2
DOUBLE PRECISION :: Plow, Pup
DOUBLE PRECISION :: F11, F12, Fs1, Fs2, V2, P1
DOUBLE PRECISION :: high, low
DOUBLE PRECISION :: v_1, v_2

DOUBLE PRECISION, DIMENSION(3) :: temp
DOUBLE PRECISION, DIMENSION(5) :: dump
DOUBLE PRECISION, DIMENSION(6) :: guess
DOUBLE PRECISION, DIMENSION(6, 1000) :: dome

```



---

```

DOUBLE PRECISION :: F1, Fs

INTEGER           :: step, tstep
INTEGER           :: step1, step2
INTEGER           :: i
INTEGER           :: counter, maxcount, check
INTEGER           :: flag

INTEGER           :: incr

maxcount = 100
delta = 1.001
error = 0.01
ierror = 1.0
ierror2 = 1.0
counter = 20
check = 0

!      E1 =

!      E1 = 27.48
E1 = 0.
P1 = 0.
Ss = 0.
!      Ss = 5.195
!      V1 = 0.2083
!      T1 = 0
!      temp(1) = 11000
V1= 0.37
!      T1 = 0.
T1 = .293
!      temp(1) = 1121.14833403170
!      temp(1) = 300
!      call busheos(P1,E1,V1,T1,Ss,temp)
call bush(P1,E1,V1,T1,Ss,temp)
write(*,101) V1,T1,P1,E1,Ss,temp(3),temp(2)
! Isentrope calculation
!      call HRISENTROPE()

! Reshock calculation
!      v_1 = 0.25885
!      v_1 = 0.237611

```

```

!      v_1 = 0.216409
!      v_1 = 0.260656
!      v_1 = 0.256342
!      v_1 = 0.216409
!      v_1 = 0.219653
v_1 = 0.237869
step1= 40
!      v_2 = 0.225701
!      v_2 = 0.201069
!      v_2 = 0.176217
!      v_2 = 0.230568
!      v_2 = 0.224189
!      v_2 = 0.179572
!      v_2 = 0.190892
v_2 = 0.210988
step2 = 40
call reshock(v_1 , step1 , v_2 , step2)

! Hugoniot calculation
!      high = 0.36
!      low = 0.1
!      incr = 105
!      call Hugoniot(low , high , incr)

!      open(11, file = 'Isotherm . dat ')

!      T1 = 20.0
!      vu = 10000
!      step = 1000
!      delta = 1./counter
!      V1 = 0.1
!      write(11,102) "v", "T", "P", "E", "S", "G", "State"
!      do while (check ==0)
!          do i=1, counter-2
!              P1 = 0.0
!              E1 = 0.0
!              Ss = 0.0
!              temp(1:3) = 0.0
!              call bush(P1, E1, V1, T1, Ss, temp)
!              G = E1+P1*V1-T1*Ss
!              write(11,101) V1, T1, P1, E1, Ss, G, temp(2)
!              V1 =V1+delta
!              if (V1. gt. vu) then
!                  check = 1

```

```

!           endif
!           enddo
!           delta = delta*10
!           enddo
!!
!           close(11)
!
101 format(3x,e10.4,4x,e10.4,4x,e10.4,
&         4x,e10.4,4x,e10.4,4x,e10.4,4x,e10.3)
102 format(3x,A,6x,A,6x,A,6x,A,6x,A,6x,A)
103 format(3x,e10.4,4x,e10.4,4x,e10.4,4x,e10.4)
104 format(3x,e10.4,4x,e10.4,4x,e10.4,4x,e10.4,4x,e10.4,4x,e10.4)
END PROGRAM

```

---

```

! Subroutine: Hugoniot
!
! DESCRIPTION:
! Plots/exports isotherms for over a given temperature range and
! number of steps

```

---

```

SUBROUTINE Hugoniot(hig,low,incr)
USE EOS
IMPLICIT NONE
DOUBLE PRECISION L, U, delta, Cu, low, hig,dump,E2,Up2,f2
DOUBLE PRECISION f,df,ierror,error,Peos,Ut,V2,P2,dsmall,E1
! DOUBLE PRECISION E,Up,Us,P,v0
DOUBLE PRECISION,DIMENSION(3) :: temp
DOUBLE PRECISION,DIMENSION(10) :: junk
integer sel, step, i, incr
integer counter, maxcount

! write(*,*) 'v0: ',v0
v0 = 0.370
counter = 0
maxcount = 100

open(11, file='hugo.dat', access='append')
open(9)
L = low
U = hig
step = incr

```

```

    delta = (U-L)/(step*1.)
    WRITE(11,100) 'Principle Hugoniot Calculation '
    WRITE(11,102) 'v[cm3/g] ', 'P[Mbar] ', 'Up[cm/micro-s] ',
&
    'Us[cm/micro-s] '
!
    dsmall=0.99999
    dsmall = 1.01
    if (Up.eq.0.) then
        Up=0.1
    endif
    temp(1:3)=0.0
!
    L = .300
2100 do i=0,step
    Cu=L+delta*i
    write(9,*) "v:", cu, "Up:", Up
    dump=0
    error=0.001
    ierror=1.0
!
    E1 = 5.0e-3
!
    call busheos (Peos, E1, cu, junk(2), junk(3), temp)

!
    return
do while (ierror.ge.error)
    Peos=0.0
    junk(1:10)=0.0
    E1=((Up**2.)/2.)
    write(9,107) cu, Up, (Up**2)/(v0-cu), E1
    call busheos (Peos, E1, cu, junk(2), junk(3), temp)
    f=Peos-(Up**2)/(v0-cu)
    write(9,107) cu, Peos, E1, temp(2)
    Up2=dsmall*Up
    E2=((Up2**2./2.))
    temp(1) =junk(2)
    junk(1:10)=0.0
    write(9,106) (Up2**2)/(v0-cu), E1, temp(1)
    call busheos (junk(1), E2, cu, junk(2), junk(3), temp)
    f2 = junk(1)-(Up2**2)/(v0-cu)
    df = (f2-f)/(Up2-Up)
    Ut=Up-f/df
    write(9,108) cu, Peos, E1, f2, df, Ut
    ierror = abs((Ut-Up)/Up)
    Up=abs(Ut)
    write(9,*) "ierror=", ierror
    write(9,*)
    if (counter.ge.maxcount) then
        write(9,*) "Cant converge moving to next point"

```

```

        ierror = 0.
    endif
    counter =counter +1
end do
write(9,*)
Up=Ut
Us=(Up*v0) / (v0-cu)
P = Up*Us/v0
E = Up**2/2
write(11,103) Cu,P, Up, Us, temp(2)
write(*,103) Cu,P, Up, Us, temp(2)
enddo
write(11,*) ''
write(11,*) ''
close(11)
close(9)
return
100 format(10x,A)
101 format(18x,A)
102 format(16x,A,7x,A,3x,A,3x,A)
103 format(15x,e10.4,4x,e10.4,4x,e10.4,4x,e10.4,4x,e10.4)
104 format(4x,'rho0: ',e10.4,4x,'c0: ',e10.4,4x,'s1: ',
& e10.4,4x,'g0: ',e10.4)
105 format(4x,'Wilkins Constants: ',e10.4,' : ',e10.4,
& ' : ', e10.4,' : ',e10.4)
106 format(3x,e10.4,4x,e10.4,4x,e10.4)
107 format(3x,e10.4,4x,e10.4,4x,e10.4,4x,e10.4)
108 format(3x,e10.4,4x,e10.4,4x,e10.4,4x,e10.4,4x,e10.4)
END SUBROUTINE Hugoniot

```

---

```

!
! Subroutine: Reshock
!
! DESCRIPTION:
! Calculated reshock state given principle shock strength, and reshock
! strength
!
!

```

---

```

SUBROUTINE reshock(V_1,step1,V_2,step2)
use EOS
IMPLICIT NONE
DOUBLE PRECISION up_1, P_1, E_1, V_1
DOUBLE PRECISION up_2, P_2, E_2, V_2
DOUBLE PRECISION L, U, delta, Ut,Cu,E1,E2,Up2
DOUBLE PRECISION f, f2,df, ierror, error, Peos, V2, dsmall

```

---

```

DOUBLE PRECISION,DIMENSION(3) :: temp
DOUBLE PRECISION,DIMENSION(10) :: junk
INTEGER step1, step2,i, counter, maxcount

counter = 0
maxcount = 20
v0 = 0.37

open(11, file='reshock.data', access='append')
dsmall = 0.99999
! WRITE(*,101) 'Enter Hugoniot shock specific volume and increments'
! READ(*,*) V_1, step1
! WRITE(*,101) 'Enter reshock specific volume and increments'
! READ(*,*) V_2, step2
write(*,*) " V1, step1, v2, step 2"
write(*,*) V_1, step1, V_2, step2
write(11,102) 'v[cm3/g]', 'P[Mbar]', 'Up[cm/micro-s]',
& 'Us[cm/micro-s]'
WRITE(*,*) "Calculating principle hugoniot"
Up = 0.1
temp(1:3) = 0.0
L = 0.37
U = V_1
delta = (U-L)/(step1*1.0)
! junk(2) = 0.325
do i=1, step1
Cu = L+delta*i
! dump(1:10) = 0.0
error = 0.01
ierror = 1.0
counter = 0
do while (ierror.ge.error)
Peos = 0.0
! temp(1) = junk(2)
junk(1:10) = 0.0
E1 = (Up**2./2.)
call busheos(Peos, E1, cu, junk(2), junk(3), temp)
f=Peos-(Up**2)/(v0-cu)
Up2=dsmall*Up
E2=(Up2**2./2.)
! temp(1) = junk(2)
junk(1:10)=0.0
call busheos(junk(1), E2, cu, junk(2), junk(3), temp)
f2 = junk(1)-(Up2**2)/(v0-cu)
df = (f2-f)/(Up2-Up)

```

```

        Ut=Up-f/df
        ierror = abs((Ut-Up)/Up)
!       ierror = abs(f/((Up**2)/(v0-cu)))
        Up=abs(Ut)
        if (counter.ge.maxcount) then
            write(9,*) "Cant converge moving to next point at"
            write(9,*) "V =", Cu
            ierror = 0.
        endif
        counter =counter +1
    end do
    if (counter.lt.maxcount) then
        Up=Ut
        Us=(Up*v0) / (v0-cu)
        P = Up*Us/v0
        E = Up**2/2
        write(11,103) Cu,P, Up, Us,temp(2)
        write(*,103) Cu,P, Up, Us,temp(2)
    else
        write(*,*) "Solution not found at V =",Cu
    endif

enddo
write(*,*) "Principle Hugoniot calculation complete"
write(*,103) Cu,P, Up, Us
write(*,*) "Reshock calculation"
up_1 = Up
P_1 = P
E_1 = E
write(*,*) "Up_0 , P_0 , E_0"
write(*,*) up_1 , P_1 , E_1

temp(1:3) = 0.0
L = Cu
U = V_2
counter = 0
delta = (U-L)/(step2*1.0)
Up_2 = 0.01
do i=1, step2
    Cu = L+delta*i
!    dump(1:10) = 0.0
    error = 0.01
    ierror = 1.0
    counter = 0
    do while (ierror.ge.error)

```

---

```

    Peos = 0.0
    P_2 = P_1+((Up_1-Up_2)**2.)/(V_1-cu)
!   E1 = (Up**2./2.)
    E1 = E_1 + 0.5*(P_2-P_1)*(V_1-cu)
    temp(1) = junk(2)
    junk(1:10)=0.0
    call busheos(Peos,E1,cu,junk(2),junk(3),temp)
    f=Peos-P_2
!   write(9,*) cu, Peos,P_2, E1
    Up2=dsmall*Up_2
!   E2=(Up2**2./2.)
    P_2 = P_1+((Up_1-Up2)**2.)/(V_1-cu)
    E2 = E_1 + 0.5*(P_2-P_1)*(V_1-cu)
    temp(1) = junk(2)
    junk(1:10)=0.0
!   write(9,*) P_2, E1
    call busheos(junk(1),E2,cu,junk(2),junk(3),temp)
    f2 = junk(1)-P_2
    df = (f2-f)/(Up2-Up_2)
    Ut=Up_2-f/df
!   write(9,*) cu, Peos, E1, f2, df, Ut
!   ierror = abs((Ut-Up_2)/Up_2)
    ierror = abs((P_2-Peos)/P_2)
!   write(9,*) ierror, error
    Up_2=abs(Ut)
    if (counter.ge.maxcount) then
        write(9,*) "Can't converge moving to next point at"
        write(9,*) "V =", Cu
        ierror = 0.
    endif
    counter =counter +1
end do
if (counter.lt.maxcount) then
    Up=Ut
    Us=(Up-Up_1)*v0 / (v0-cu)
    P = P_1+((Up_1-Up2)**2.)/(V_1-cu)
    E = E_1 + 0.5*(P-P_1)*(V_1-cu)
    write(11,103) Cu,P, Up, Us, temp(2)
    write(*,103) Cu,P, Up, Us, temp(2)
else
    write(*,*) "Solution not found at V =",Cu
endif
enddo

return

```



```

100 format(10x,A)
101 format(18x,A)
102 format(16x,A,7x,A,3x,A,3x,A)
! 103 format(15x,e10.4,4x,e10.4,4x,e10.4,4x,e10.4,4x)
103 format(15x,e10.4,4x,e10.4,4x,e10.4,4x,e10.4,4x,e10.4)

```

**ENDSUBROUTINE** Reshock

---

```

!
! Subroutine: ISENTROPE
!
! DESCRIPTION:
! Plots/exports ISENTROP
!
```

---

```

SUBROUTINE ISENTROPE(low , hig , incr , temp)
  use EOS
  IMPLICIT NONE
  DOUBLE PRECISION low , hig
  DOUBLE PRECISION L, U, Cu, delta
  DOUBLE PRECISION Pi, Po, dump
  DOUBLE PRECISION Ei, Eo, T0
  DOUBLE PRECISION Ur, dP1, dP2, E2, E1
  DOUBLE PRECISION error , ierror , f , df , Up2, Peos , v2 , dsmall , Ut , f2
  DOUBLE PRECISION S0, Ss
  DOUBLE PRECISION, dimension(3) :: temp
  DOUBLE PRECISION, dimension(10) :: junk

  INTEGER          step , i , sel , incr , counter

  open(11 , file='isen.dat' , access='append ')
  open(66)
  L=low
  U=hig
  step=incr
!   dsmall=0.99999
  dsmall = 0.999
  counter = 0
!   dsmall=1.00001
!   L = v0 ! Lower bound
!   U = 0.203478 ! Upper Bound
!   step = 100
  WRITE(11,100) 'Isentrope Calculation '
  WRITE(11,*) 'Ploting from ',L,' to ',U
!   Up=0
!   Find initial Hugoniot Pressure given specific volume

```

---

```

error=0.0001
ierror=1.0
!   Up=0.1
Up = temp(3)
junk(2) = temp(1)
write(* ,*) "temp 1:" , temp(1)
write(* ,*) "Up:" ,Up
do while ( ierror.ge.error)
    Peos=0.0
    temp(1) = junk(2)
    junk(1:10)=0.0
    E1=(Up**2.)/2.
    call busheos(Peos,E1,L,junk(2),junk(3),temp)
    f=Peos-(Up**2)/(v0-L)
    Up2=dsmall*Up
    E2=(Up2**2.)/2.
    temp(1) = junk(2)
    junk(1:10)=0.0
    call busheos(junk(1),E2,L,junk(2),junk(3),temp)
    f2 = junk(1)-(Up2**2)/(v0-L)
    df = (f2-f)/(Up2-Up)
    Ut=Up-f/df
    ierror = abs((Ut-Up)/Up)
    Up=abs(Ut)
    write(* ,*) 'Up: ',Up
end do
Ur=Ut           ! units cm/micro-s
Ei= Ur**2./2.
temp(1) = junk(2)
junk(1:10) = 0.0
call busheos(junk(1),Ei,L,junk(2),junk(3),temp)
Pi = Peos
write(* ,*) L, Pi, low
pause
WRITE(11,102) 'v[cm3/g] ', 'P[Mbar] ', 'E[Mbar-cm3/g] ',
&           'Ur[cm/micro-s] ', 'Region '
write(11,103) L, Pi, Ei, junk(2), junk(3), Ur, temp(2)
write(* ,*) L, Pi, Ei, junk(2), Ur, temp(2)
temp(1) = junk(2)
write(* ,*) "Tguess:" ,temp(1)
junk(1:10) = 0.0
V2 = L*dsmall
E2 = Ei-(V2-L)*Pi
call busheos(junk(1),E2,V2,junk(2),junk(3),temp)
S0 = junk(3)

```

```

write(*,*) Pi, Ei, E2,V2,v0,junk(1)
dP1 = (sqrt(-(junk(1)-Pi)/(V2-L)))
write(*,*) Pi, junk(1), V2,L, dP1, temp(2)
!   delta = ((v0*1.1)-l)/step*1.
delta = 0.001
i=0
Po = 1
Ss = S0
!   return
do while(Po.gt.0.0001)
!   do while(counter.lt.100000)
!       Ss = S0
           i=i+1
           Cu=L+delta*i ! Current location (v)
!!       Calculate Eo using foward difference method
!!       Shouldn't Pr+gamma0/v0(E-Er) be replaced with P ??
temp(1) = junk(2)
junk(1:10)= 0.0
call busheos(junk(1),junk(3), Cu, junk(2), Ss,temp)
write(66,*) Cu, Eo, junk(1)
if(junk(1).lt.0) goto 1000
Po = junk(1)
Eo = junk(3)
T0 = junk(2)
V2=Cu*dsmall
temp(1) = junk(2)
junk(1:10)=0.0
!       Find P2
call busheos(junk(1), junk(3), V2, junk(2), Ss, temp)
write(66,*) V2, E2, junk(1)
dP2 =(sqrt(-(junk(1)-Po)/(V2-Cu)))
!       write(*,*) Po, junk(1), V2,Cu, dP2
!       pause
Ur=Ur+0.5*(dP2+dP1)*(delta)
write(11,103) Cu, Po, Eo,T0,Ss,Ur, temp(2)
write(*,103) Cu, Po, Eo,T0, Ss,Ur,temp(2)
write(66,*) dP1, dp2, delta, Ur
write(66,*) ""
Ei=Eo
Pi=Po
dP1=dP2
!       counter = counter + 1
enddo
1000 write(11,*) ''
write(11,*) ''

```

```

        close(11)
        write(*,*) "End", junk(1)
        write(*,103) Cu, Po, Eo, U
        return
100 format(10x,A)
101 format(18x,A)
102 format(16x,A,6x,A,4x,A,3x,A,3x,A)
! 103 format(15x,e10.4,4x,e10.4,4x,e10.4,6x,e10.4,7x,i1)
103 format(15x,e10.4,4x,e10.4,4x,e10.4,4x,e10.4,4x,e10.4,
&      4x,e10.4,4x,e10.4)
104 format(4x,'rho0: ',e10.4,4x,'c0: ',e10.4,4x,'s1: ',
&      e10.4,4x,'g0: ',e10.4)
105 format(4x,'Wilkins Constants: ',e10.4,' : ',e10.4,
&      ' : ',e10.4,' : ',e10.4)

```

**END SUBROUTINE ISENTROPE**

---

```

!  

! Subroutine: HRISENTROPE  

!  

! DESCRIPTION:  

! Plots/exports The release isentrope from a given pressure

```

---

```

SUBROUTINE HRISENTROPE()
  use EOS
  IMPLICIT NONE
  DOUBLE PRECISION L, U, Cu, delta
  DOUBLE PRECISION Pg
  DOUBLE PRECISION v, P1, P2, E1, E2
  DOUBLE PRECISION df, Pi, Po, Ur,dP1,dP2, dump
  DOUBLE PRECISION Ei, Eo, ierror, error, P0
  DOUBLE PRECISION,DIMENSION(10) :: junk
  DOUBLE PRECISION,DIMENSION(3) :: temp
  INTEGER          step, i, sel

1000 write(*,101) 'Pressure which released from'
      read(*,*) Pg
      call findV(Pg,L,temp)
      write(*,*) "after findV"
      write(*,*) 'P: ',Pg, 'v: ',L, 'T: ',temp(1)
      step = 10000
      U=L*10
      CALL ISENTROPE(L,U,step,temp)
      goto 5000

```

```

5000 WRITE(*,*) 'Plot again yes (1)/No (2)'
      read(*,*) sel
      if(sel.eq.1) then
        goto 1000
      endif

```

```

RETURN

```

```

100 format(10x,A)

```

```

101 format(18x,A)

```

```

END SUBROUTINE HRISENTROPE

```

```

SUBROUTINE domeCheck(Pg,T1,vl,vu,step,flag,guess)

```

```

!
```

```

! This subroutine scans along the given isotherm checking the
! pressure state is below the given pressure (Pg)
! indicating the estimated location of specific volume with
! satisfies the P and T
!
```

```

!
```

```

! vl and vu are bounds of the search
! step is a legacy and can be removed
! flag is an indicator of how many solutions where found
! guess are the specific density which bound the given pressure
!
```

```

IMPLICIT NONE

```

```

DOUBLE PRECISION :: P2, P1,Pg,v2,v1, vl, vu, T1

```

```

DOUBLE PRECISION :: check, delta,delta2, vt

```

```

DOUBLE PRECISION :: fl1, fl2, fs1, fs2, Pl,Ps

```

```

DOUBLE PRECISION :: G1, Gs

```

```

DOUBLE PRECISION,DIMENSION(6) :: guess

```

```

DOUBLE PRECISION,DIMENSION(5) :: dump

```

```

DOUBLE PRECISION,DIMENSION(3) :: temp

```

```

INTEGER :: flag

```

```

INTEGER :: i

```

```

INTEGER :: step

```

```

INTEGER :: crap

```

```

INTEGER :: counter

```

```

crap = 0

```

```

open(14)

```

```

flag = 1

```

```

        guess(1:6) = 0.0
!       delta = 0.05
        delta2 = 1.001
!       check = 1.0
!       v1 = vl
!       v1 = 0.1
        if (vu.le.0.0) then
            vu = 10000000.
        endif
        if (vl.le.0.0) then
            vl = 0.1
        endif
!       v1 = 10000000.
!       vl = 0.1
        v1 = vu
        counter = 20

        if (Pg.le.0.0) then
! In the event that the given pressure is negative,
! then only two "real" solutions are possible
            flag = -1
            guess(1) = -9.99e9
            guess(5) = -9.99e9
            return
        endif

        delta = v1/counter
!       check = vl/10

!       P1 = 0.0
!       dump(1:5) = 0.
! LEGACY      call bush(P1,dump(1),v1,T1,dump(2),temp)
        CALL getF(v1,T1,fl1,fs1)
        vt = v1*delta2
        CALL getF(vt,T1,fl2,fs2)
        P1 = -(fl2-fl1)/(vt-v1)
        Ps = -(fs2-fs1)/(vt-v1)
        G1 = F11+P1*v1
        Gs = Fs1+Ps*v1
        if (v1.gt.0.5) then
            P1=P1
        else if (G1 > Gs) then
            P1=Ps
        else

```

```

        P1=P1
    endif

    write(14,103) P1,v1

    if (P1.gt.Pg) then
! If P at the maximum specific volume returns a pressure
! greater the the given pressure it will never pick the
! correct vgass since it is greater the the max
        flag = -2
        guess(1) = -9.99e9
        guess(5) = -9.99e9
        return
    endif

!     P1= -1000

!     do while (v1.lt.vu)
do while (v1.gt.vl)
    do i=1,(counter-2)
!         write(*,*) "i=",i,delta
        P2 = P1
        v2 = v1
        P1 = 0.0
        dump(1:5) = 0.
        v1 = v1-delta
! LEGACY         call bush(P1,dump(1),v1,T1,dump(2),temp)
CALL getF(v1,T1,fl1,fs1)
        vt = v1*delta2
CALL getF(vt,T1,fl2,fs2)
        P1 = -(fl2-fl1)/(vt-v1)
        Ps = -(fs2-fs1)/(vt-v1)
        G1 = F11+P1*v1
        Gs = Fs1+Ps*v1
        if (v1.gt.0.5) then
            P1=P1
        else if (G1 > Gs) then
            P1=Ps
        else
            P1=P1
        endif
    write(14,103) P1,v1,P2,v2
    if(((P1.gt.Pg).and.(P2.lt.Pg)).or.
&         ((P1.lt.Pg).and.(P2.gt.Pg))) then
!         pause

```

```

        write(14,*) '*****'
        guess(flag) = v2
        guess(flag+1) = v1
        flag=flag+2
    endif
    if(flag.eq.7) then
        v1=0.1*v1
        exit
    endif
enddo
    delta = delta/10
enddo
flag = (flag-1)/2
!   write(*,*) 'Number of possible solutions : ',flag
write(14,*) 'Number of possible solutions : ',flag
close(14)
return
103 format(3x,e10.4,4x,e10.4,4x,e10.4,4x,e10.4)
END SUBROUTINE

SUBROUTINE endPoints(P1,T1,guess ,v1 ,vg)
!
!   This subroutine finds the actual endpoints of the saturation dome
!   P1 is
!
    IMPLICIT NONE
    DOUBLE PRECISION :: v1 ,vg ,cu
    DOUBLE PRECISION :: T1 , P1 ,P2 ,Ptemp ,f1 , f2 , df
    DOUBLE PRECISION :: v1 , v2 ,dv ,delta , delta2
    DOUBLE PRECISION :: error , ierror
    DOUBLE PRECISION :: vt , fl1 , fl2 , fs1 , fs2 , G1 ,Gs ,P1 ,Ps
    DOUBLE PRECISION,DIMENSION(5) :: dump
    DOUBLE PRECISION,DIMENSION(3) :: temp
    DOUBLE PRECISION,DIMENSION(6) :: guess
!   guess is an initial guess for lower and upper
    INTEGER :: counter , maxcount
!
!   write(*,*) 'P1 :', P1 , ' T1: ',T1
open(17)
write(17,*) "EndPoints: ",P1, T1

    delta2 = 1.001
    delta = 1.001
    error = 0.01
    ierror= 1.0

```



```

dump(1:5) = 0.
temp(1:3) = 0.
Ptemp=0.
vl = .3 ! liquid state
vg = 100 ! gas state
counter = 0
maxcount = 100 ! limits number of iterations
!
!   cu =guess(5)
!   cu = 0.5*(guess(5)+guess(6))
cu = guess(5)
do while (ierror >= error)
!!   cu = vl
      dump(1:5) = 0.
      Ptemp = 0.
!     call bush(Ptemp,dump(1),cu,T1,dump(2),temp)
CALL getF(cu,T1,fl1,fs1)
      vt = cu*delta2
CALL getF(vt,T1,fl2,fs2)
      Pl = -(fl2-fl1)/(vt-cu)
      Ps = -(fs2-fs1)/(vt-cu)
      Gl = Fl1+Pl*cu
      Gs = Fs1+Ps*cu
      if (cu.gt.0.5) then
        Ptemp = Pl
      else if (Gl > Gs) then
        Ptemp=Ps
      else
        Ptemp=Pl
      endif
      f1 = Pl-Ptemp
      write(17,101) cu, T1, Ptemp, f1, -9.9e9, -9.9e9
      Ptemp = 0
      dump(1:5) = 0.
!     call bush(Ptemp,dump(1),cu*delta,T1,dump(2),temp)
CALL getF(cu*delta,T1,fl1,fs1)
      vt = cu*delta*delta2
CALL getF(vt,T1,fl2,fs2)
      Pl = -(fl2-fl1)/(vt-cu*delta)
      Ps = -(fs2-fs1)/(vt-cu*delta)
      Gl = Fl1+Pl*cu*delta
      Gs = Fs1+Ps*cu*delta
      if ((cu*delta).gt.0.5) then
        Ptemp = Pl
      else if (Gl > Gs) then

```

---

```

        Ptemp=Ps
    else
        Ptemp=P1
    endif
    df=(P1-Ptemp)-f1
    dv = delta*cu-cu
    df = df/dv
    write(17,101) delta*cu, T1, Ptemp, (P1-Ptemp), df, dv
    cu = cu - f1/df

    if ((cu.lt.guess(6)).or.(cu.gt.guess(5))) then
        if(cu.gt.guess(5)) then
            write(17,*) cu,">",guess(5)
            cu = cu+f1/df
            cu = 0.5*(cu+guess(5))
            write(17,*) "new cu =",cu
        else
            write(17,*) cu,">",guess(6)
            cu = cu+f1/df
            cu = 0.5*(cu+guess(6))
            write(17,*) "new cu =",cu
        endif
    endif
    Ptemp = 0
    dump(1:5) = 0.
!    call bush(Ptemp,dump(1),cu,T1,dump(2),temp)
    CALL getF(cu,T1,f11,fs1)
    vt = cu*delta2
    CALL getF(vt,T1,f12,fs2)
    P1 = -(f12-f11)/(vt-cu)
    Ps = -(fs2-fs1)/(vt-cu)
    G1 = F11+P1*cu
    Gs = Fs1+Ps*cu
    if (cu.gt.0.5) then
        Ptemp = P1
    else if (G1 > Gs) then
        Ptemp=Ps
    else
        Ptemp=P1
    endif
    ierror = ABS((P1-Ptemp)/P1)
!    ierror = ABS(P1-Ptemp)
    write(17,101) cu, T1, Ptemp,ierror, -9.9, -9.9
    counter= counter+1
    if(counter.ge.maxcount) then

```

```

        write(17,*) "Maximum iteration reached taking last value"
        ierror = 0.0
!       cu = vl + 1.0
!       counter = 0
    endif
enddo
vl = cu
write(17,*) "End of vl loop"
write(17,*) vl

!
ierror= 1.0

counter = 0
!   cu = guess(1)
cu = 0.5*(guess(1)+guess(2))
do while (ierror >= error)
!       cu = vg
dump(1:5) = 0.
Ptemp = 0.
!       call bush(Ptemp,dump(1),cu,T1,dump(2),temp)
CALL getF(cu,T1,fl1,fs1)
vt = cu*delta2
CALL getF(vt,T1,fl2,fs2)
P1 = -(fl2-fl1)/(vt-cu)
G1 = F11+P1*cu
!       Gs = Fs1+Ps*cu
if (cu.gt.0.5) then
    Ptemp = P1
!       else if (G1 > Gs) then
!           Ptemp=Ps
else
    Ptemp=P1
endif
f1 = P1-Ptemp
Ptemp = 0
dump(1:5) = 0.
!       call bush(Ptemp,dump(1),cu*delta,T1,dump(2),temp)
CALL getF(cu*delta,T1,fl1,fs1)
vt = cu*delta*delta2
CALL getF(vt,T1,fl2,fs2)
P1 = -(fl2-fl1)/(vt-cu*delta)
G1 = F11+P1*cu*delta
!       Gs = Fs1+Ps*cu*delta
if ((cu*delta).gt.0.5) then

```

```

        Ptemp = P1
!       else if (Gl > Gs) then
!           Ptemp=Ps
    else
        Ptemp=P1
    endif
    df=(P1-Ptemp)-f1
    dv = delta*cu-cu
    df = df/dv
    cu = cu - f1/df
    Ptemp = 0
    dump(1:5) = 0.
!       call bush(Ptemp,dump(1),cu,T1,dump(2),temp)
    write(17,101) cu,T1,Ptemp, f1,df,dv
    CALL getF(cu,T1,f11,fs1)
    vt = cu*delta2
    CALL getF(vt,T1,f12,fs2)
    P1 = -(f12-f11)/(vt-cu)
    G1 = F11+P1*cu
!       Gs = Fs1+Ps*cu
    if (cu.gt.0.5) then
        Ptemp = P1
!       elseif (Gl > Gs) then
!           Ptemp=Ps
    else
        Ptemp=P1
    endif
    ierror = ABS((P1-Ptemp)/P1)
    counter= counter+1
    if(counter.ge.maxcount) then
        write(17,*) "Maximum iteration reached taking last value"
        ierror = 0.0
!       cu = vg - 1.0
!       vg = cu
!       counter = 0
!       write(*,*) "cu reset to : ", cu
!       pause
    endif
enddo
vg = cu
write(17,*) vg

close(17)
!     write(*,*) 'Vl : ',vl, ' Vg : ', vf

```

---

```

    return
101 format(3x,e10.4,4x,e10.4,4x,e10.4,4x,e10.4,4x,e10.4,4x,e10.4,4x,e10.4)
    END SUBROUTINE

    SUBROUTINE getBounds(T1,Plow,Pup)
!      This subroutine returns the lower and upper pressure in the cubic
!      region of the isotherm
    IMPLICIT NONE
    DOUBLE PRECISION :: T1, Plow, Pup
    DOUBLE PRECISION :: P2, P1, V1,V2,vt
    DOUBLE PRECISION :: F11, Fs1, F12, Fs2, Pl, Ps, Gl, Gs
    DOUBLE PRECISION :: delta, delta2
    DOUBLE PRECISION :: ierror
    DOUBLE PRECISION :: error
    DOUBLE PRECISION :: Vtop

    INTEGER :: check, counter, i, inflec, state, crap

    error = 1.0
    ierror = 0.001
    V1 = 0.1
    Vtop = 100000.
    check = 0
    delta2 = 1.001

    inflec = 1
    state = 0

    open(77)

!      Sweep through the isotherm starting at v=0.1 to 1e5 on a log scale
!      check for inflection points between positive and negative slopes
    counter = 10
    delta = 1./counter
    crap = 0

    call getF(V1,T1,F11,Fs1)
    vt=V1*delta2
    call getF(vt,T1,F12,Fs2)
    Pl = -(F12-F11)/(vt-V1)
    Ps = -(Fs2-Fs1)/(vt-V1)
    Gl = F11+Pl*V1
    Gs = Fs1+Ps*V1
    if (V1.gt.0.5) then
        P1 = Pl

```

```

        state = 2
    else if (G1>Gs) then
        P1 = Ps
        state = 1
    else
        P1 = P1
        state = 2
    endif
    if (P1.lt.0.0) then
        crap = 1
    endif

write(77,103) T1, V1, P1, G1,Gs,state , inflec

do while (check == 0)
    do i=1,counter-1
        V2=V1+delta
        call getF(V2,T1,F11,Fs1)
        vt=V2*delta2
        call getF(vt,T1,F12,Fs2)
        P1 = -(F12-F11)/(vt-V2)
        Ps = -(Fs2-Fs1)/(vt-V2)
        G1 = F11+P1*V2
        Gs = Fs1+Ps*V2
        if (V2.gt.0.5) then
            P2 = P1
            state = 2
        else if (G1.gt.Gs) then
            P2 = Ps
            state = 1
        else
            P2 = P1
            state = 2
        endif
        write(77,103) T1, V2, P2,G1,Gs,state , inflec
        if (crap.eq.1) then
            if(P2.lt.0.0) then
                crap = 1
            else
                crap = 0
            endif
            else if ((P2.gt.P1).and.(inflec.eq.1)) then
! Found first inflection point
                Plow = P1
                inflec = 2

```

```

                elseif ((P2.lt.P1).and.(inflec.eq.2)) then
!   Found second inflection point
                Pup = P2
                inflec = 3
                check =1
                else if (V2.gt. Vtop) then
!   write(*,*) "V2 greater the Vtop"
                check = 1
                exit
                endif
                V1 = V2
                P1 = P2
!
                enddo
                delta = delta*10
            enddo

            close(77)
            return
103 format(3x,e10.4,4x,e10.4,4x,e10.4,4x,e10.4,4x,e10.4,4x,i1,4x,i1)
            END SUBROUTINE

SUBROUTINE satPoints(T1,Ps,vl,vg)
!   This subroutine returns the saturation endpoints of the isotherm T
!   using only the getF routine
!   Outputs include the saturation pressure and the specific volume,
!   energy, and entropy at the liquid and gas endpoints
IMPLICIT NONE
DOUBLE PRECISION T1, Ps, vl, vg, el, eg, sl, sg
DOUBLE PRECISION P1,P2, dg1, dg2, dp, dg
DOUBLE PRECISION f1, df,dv, delta, ierror
DOUBLE PRECISION P1, Pu, Pg
DOUBLE PRECISION vliq, vgas, Fl, Fs
DOUBLE PRECISION vbottom, vtop
DOUBLE PRECISION Gl, Gs, Glower, Ggas
DOUBLE PRECISION error

DOUBLE PRECISION,DIMENSION(3)      :: temp
DOUBLE PRECISION,DIMENSION(5)      :: dump
DOUBLE PRECISION,DIMENSION(6)      :: guess

INTEGER :: flag, step, i, check, counter, pcount, maxcount, count2
INTEGER :: check2

delta = 1.001

```

---

```

ierror = 1
error = 0.001
step = 1000 ! This may not be needed
vbottom = 0.1
vtop = 10000
maxcount = 100
counter = 0
pcount = 1
count2 = 0
check2 =1

open(15)
write(15,*) "Temperature=",T1
!
CALL getBounds(T1,P1,Pu)
write(15,*) "getBound returns"
write(15,101) T1, P1, Pu
! Initial pressure guess is the average of the P1 and Pu
! representing the max and min pressure seen in the cubic section
! P1 = 0.5*(P1+Pu)
P1 = 0.5*(Pu+P1)
if (P1.lt.0.0) then
    P1=0.5*Pu
endif
! write(*,*) P1

! P1=0.5*Pu
! domeCheck find rough location of specific volum
! endPoints routine finds the specific volume which corespond to the
! given Temperautre and guessed pressure
guess(1:6) = 0.0
call domeCheck(P1,T1,vbottom ,vtop ,step ,flag ,guess)
write(15,*) "domeCheck returns"
write(15,102) T1, P1, guess(1), guess(5)
if (flag.lt.3) then
    write(15,*) "Flag =",flag
    if(flag.eq.-2) then
        count2 = 1
        do while (check2==1)
            write(15,*) "Increasing vtop x10"
            vtop = vtop*10
            write(15,*) "vtop=",vtop
            call domeCheck(P1,T1,vbottom ,vtop ,step ,flag ,guess)

```



---

```

        if (flag.eq.3) then
            write(15,*) "Success"
            check2 = 0
        endif
        if (count2.eq.6) then
            write(15,*) "Failure"
            write(15,*) "Gas volume greater than max volume"
            vl = -9.99e-9
            vg = -9.99e-9
            Ps = -9.99e-9
        return
        endif
        count2=count2+1
    enddo
else
    write(15,*) "Outside saturation region"
    vl = -9.99e-9
    vg = -9.99e-9
    Ps = -9.99e-9
    return
endif
endif
check2 = 1

!         if (flag.lt.3) then
!         write(*,*) "less than three solutions found new guess"
!         check = 0
!         do while (check ==0)
!             counter = counter + 1
!             P1 = 0.5*(Pu+P1)
!             guess(1:6) = 0.0
!             flag = 0
!             CALL domeCheck(P1,T,vbottom , vtop , step , flag , guess)
!             if (flag.eq.3) then
!                 check=1
!             endif
!             if (counter==maxcount) then
!                 return
!             endif
!         enddo
!     endif
CALL endPoints(P1,T1,guess , vliq , vgas)

write(15,*) "endPoints returns"
write(15,102) T1, P1, vliq , vgas

```

---

```
CALL getF(vgas ,T1 ,F1 ,Fs)
G1 = F1 + P1*Vgas
Gs = Fs + P1*Vgas
Ggas = G1

CALL getF(vliq ,T1 ,F1 ,Fs)
G1 = F1 + P1*vliq
Gs = Fs + P1*vliq
if(vliq.gt.0.5) then
    Glower = G1
else if(G1>Gs) then
    Glower = Gs
else
    Glower = G1
endif

dg1 = Glower- Ggas
write(15,*) "First gibbs"
write(15,103) T1, P1, vliq ,vgas ,dg1

P2=P1*delta
CALL domeCheck(P2,T1,vbottom ,vtop ,step ,flag ,guess)
write(15,*) "domeCheck returns"
write(15,102) T1, P2, guess(1), guess(5)
CALL endPoints(P2,T1,guess ,vliq ,vgas)
write(15,*) "endPoints returns"
write(15,102) T1, P2, vliq , vgas

CALL getF(vgas ,T1 ,F1 ,Fs)
G1 = F1 + P2*vgas
Gs = Fs + P2*vgas
Ggas = G1

CALL getF(vliq ,T1 ,F1 ,Fs)
G1 = F1 + P2*vliq
Gs = Fs + P2*vliq

if(vliq.gt.0.5) then
    Glower = G1
else if(G1>Gs) then
    Glower = Gs
else
    Glower = G1
```

```

endif
dg2 = Glower - Ggas

write(15,*) "second gibbs"
write(15,103) T1, P2, vliq ,vgas ,dg2

ierror = abs(dg2/Glower)

!      write(15,*) Glower
!
!      P2 must never be negative because as  $v \rightarrow \text{inf}$   $P \rightarrow 0$ 
!      P will never be less than zero, hence the endpoints
!      routine finds the wrong endpoints
!

dP = P2-P1
dg = dg2-dg1
dg = dg/dp
P1 = P2
dg1 = dg2
P2 = P2-dg2/dg
if (P2.lt.0.0) then
    P2=P1*0.5
endif

!

do while (ierror >= error)
    pcount = pcount + 1
    CALL domeCheck(P2,T1,vbottom ,vtop ,step ,flag ,guess)
    write(15,*) "domeCheck returns"
    write(15,102) T1, P2, guess(1), guess(5)
!      write(15,*) "Flag=", flag
!      if (flag.eq.9) then
!          write(15,*) "Gas volume greater than max volume"
!          vl = -9.99e-9
!          vg = -9.99e-9
!          Ps = -9.99e-9
!          return
!      endif

    if (flag.lt.3) then
        write(15,*) "Flag=", flag
        if (flag.eq.-2) then
            count2 = 1
            do while (check2==1)
                write(15,*) "Increasing vtop x10"
                vtop = vtop*10

```

```
write(15,*) "vtop=", vtop
call domeCheck(P1, T1, vbottom, vtop, step, flag, guess)
if (flag.eq.3) then
  write(15,*) "Success"
  check2 = 0
endif
if (count2.eq.6) then
  write(15,*) "Failure"
  write(15,*) "Gas volume greater than max volume"
  vl = -9.99e-9
  vg = -9.99e-9
  Ps = -9.99e-9
  check = 0
  return
endif
count2=count2+1
enddo
else
  write(15,*) "Outside saturation region"
  return
  vl = -9.99e-9
  vg = -9.99e-9
  Ps = -9.99e-9
endif
endif
check2=1

CALL endPoint(P2, T1, guess, vliq, vgas)
write(15,*) "endPoints returns"
write(15,102) T1, P2, vliq, vgas

CALL getF(vgas, T1, F1, Fs)
G1 = F1 + P2*vgas
Gs = Fs + P2*vgas
Ggas = G1

CALL getF(vliq, T1, F1, Fs)
G1 = F1 + P2*vliq
Gs = Fs + P2*vliq

if (vliq.gt.0.5) then
  Glower = G1
else if (G1>Gs) then
  Glower = Gs
```

---

```

    else
      Glower = Gl
    endif
    dg2 = Glower-Ggas

    write(15,*) "loop gibbs"
    write(15,103) T1, P2, vliq ,vgas ,dg2

    ierror = abs(dg2/Glower)

    dP = P2-P1
    dg = dg2-dg1
    dg = dg/dp
    P1 = P2
    dg1 = dg2
    P2 = P2-dg2/dg
    if (P2.lt.0.0) then
      P2 = 0.5*P1
    endif
    if (P2>Pu) then
      P2 = Pu
    else if (P2<P1) then
      P2 = P1
    endif
    if (pcount.eq.maxcount) then
      write(15,*) "less than three solutions found"
      vl = -9.99e-9
      vg = -9.99e-9
      Ps = -9.99e-9
      return
    endif
  enddo
  Ps=P1
  vl = vliq
  vg = vgas
  write(15,*) "vl=", vl , "vg=", vg
  close(15)
!   vg , el , eg , sl , sg
  return !
101 format(3x,e10.4,4x,e10.4,4x,e10.4)
102 format(3x,e10.4,4x,e10.4,4x,e10.4,4x,e10.4)
103 format(3x,e10.4,4x,e10.4,4x,e10.4,4x,e10.4,4x,e10.4)
104 format(3x,e10.4,4x,e10.4,4x,e10.4,4x,e10.4,4x,e10.4,4x,e10.4)
  END SUBROUTINE

```

---

```

SUBROUTINE bush(P1,E1,V1,T1,Ss,temp)
  IMPLICIT NONE
  DOUBLE PRECISION P1, E1, V1, T1, Ss,F1
  DOUBLE PRECISION P2, E2, V2, T2, F2
  DOUBLE PRECISION F11,F12, Fs1,Fs2
  DOUBLE PRECISION P11,Ps1, S11, Ss1
  DOUBLE PRECISION P12, Ps2, S12,Ss2
  DOUBLE PRECISION E12, Es2
  DOUBLE PRECISION E11, Es1, G1,Gs
  DOUBLE PRECISION n1, n2,dn
  DOUBLE PRECISION deltaV, deltaT
  DOUBLE PRECISION cu
  DOUBLE PRECISION error, ierror
  DOUBLE PRECISION v0, ti
  DOUBLE PRECISION Ps, vl, vg,el,eg,sl,sg
  DOUBLE PRECISION Fs
  DOUBLE PRECISION x
  DOUBLE PRECISION,DIMENSION(3)      :: temp

  INTEGER,DIMENSION(5)      :: num
  v0 = 0.3690
  ti = .293
  deltaV = 1.0001
  deltaT = 1.001

!
!  open(18)
!
!  Checking for non zero variables
!

  num(1:5) = 0
  if(P1.ne.0) then
    num(1)=1
  endif
  if(E1.ne.0) then
    num(2)=1
  endif
  if(V1.ne.0) then
    num(3)=1
  endif
  if(T1.ne.0) then
    num(4)=1
  endif
  if(Ss.ne.0) then
    num(5)=1
  endif
endif

```

```

!
!   Finding the two give thermodynamic variables
!
      if(num(3).eq.1) then
          if(num(4).eq.1) then
!               Volume and Temperature
                  goto 1007
          endif
      endif
!   Should this call be it's own subroutine
!
1007 CALL getF(V1,T1,F11 ,Fs1)
      V2 = deltaV*V1
      CALL getF(V2,T1,F12 ,Fs2)
      P11 = -(F12-F11)/(V2-V1)
      Ps1 = -(Fs2-Fs1)/(V2-V1)
!   Caluclate S
      T2 = deltaT*T1
      CALL getF(V1,T2,F12 ,Fs2)
      S11 = -(F12-F11)/(T2-T1)
      Ss1 = -(Fs2-Fs1)/(T2-T1)
      E11 = F11 + T1*S11
      Es1 = Fs1 + T1*Ss1
!   Gl = F11 + P11*V1
      G1 = E11 + P11*V1 - T1*S11
!   Gs = Fs1 + Ps1*V1
      Gs = Es1 + Ps1*V1 - T1*Ss1
      write(18,*) T1,V1,P11 ,Ps1 ,E11 ,Es1 ,G1 ,Gs
!
!   temp(2) = 1 Solid
!   temp(2) = 2 liquid
!   temp(2) = 3 liquid-vapor
!   temp(2) = 4 solid-vapor
!   temp(2) = 5 vapor
!
!   if (V1 > 0.5) then
! Assume Vapor
!       P1 = P11
!       E1 = E11
!       Ss = S11
!       temp(2) = 2
!   else if(Gl>Gs) then

CALL satPoints(T1,Ps,vl ,vg)
write(18,103) T1, Ps, vl , vg, V1

```

---

```

!      write(*,*) "Finding Sat Points"
if ((V1.gt.vg).or.(V1.lt.vl).or.(Ps < 0.0)) then
!      write(*,*) "Not in a mixed region"
      write(18,*) "Not in a mixed region"
      if( V1 > 0.5) then
        P1 = P11
        E1 = E11
        Ss = S11
        temp(2) = 2
      else if(T1.ge.10.) then
        write(18,*) "Liquid"
        P1 = P11
        E1 = E11
        Ss = S11
! temp(2) = 2 — liquid is equilibrium state
        temp(2) = 2

      else if(G1>Gs) then
        write(18,*) "Solid"
        P1 = Ps1
        E1 = Es1
        Ss = Ss1
! temp(2) = 1 — Solid is equilibrium state
        temp(2) = 1
      else
        write(18,*) "Liquid"
        P1 = P11
        E1 = E11
        Ss = S11
! temp(2) = 2 — liquid is equilibrium state
        temp(2) = 2
      endif
!      else if ((V1<vg).and.(V1>vl)) then
else
      write(18,*) "In mix phase region"
!      V1 is between Vgas and Vliq is a mixed phase
!      temp(2) = 3
!      Find pressure, energy, entropy at gas endpoint
      CALL getF(vg,T1,F11,Fs)
      V2 = deltaV*vg
      CALL getF(V2,T1,F12,Fs2)
      P11 = -(F12-F11)/(V2-Vg)
      T2 = deltaT*T1
      CALL getF(Vg,T2,F12,Fs2)
      S11 = -(F12-F11)/(T2-T1)

```



---

```

      E11 = F11+T1*S11
!   Find pressure, energy, entropy at liquid/solid endpoint
      CALL getF (V1, T1, F11, Fs1)
      V2 = deltaV*v1
      CALL getF (V2, T1, F12, Fs2)
      P12 = -(F12-F11)/(V2-V1)
      Ps2 = -(Fs2-Fs1)/(V2-V1)
      CALL getF (V1, T2, F12, Fs2)
      S12 = -(F12-F11)/(T2-T1)
      Ss2 = -(Fs2-Fs1)/(T2-T1)
      E12 = F11+T1*S12
      Es2 = Fs1+T1*Ss2
      G1 = E12 + P12*V1 - T1*S12
      Gs = Es2 + Ps2*V1 - T1*Ss2
      x = (v1-V1)/(v1-vg)
      if (G1>Gs) then
        write (18,*) " Solid - Vapor"
        temp (2) = 4 ! Solid - Vapor
        temp (3) = x
!       P1 = 0.5*(Ps1+Ps2)
        P1 = Ps
        E1 = E11-x*(E11-Es2)
        Ss = S11-x*(S11-Ss2)
      else
        write (18,*) " Liquid - Vapor"
        temp (2) = 3 ! Liquid - Vapor
        temp (3) = x
!       P1 = 0.5*(Pl1+Pl2)
        P1 = Ps
        E1 = E11-x*(E11-E12)
        Ss = S11-x*(S11-S12)
      endif
    endif
!
  return
104 format (3x, e10.4, 4x, e10.4, 4x, e10.4, 4x, e10.4, 4x, e10.4)
103 format (3x, e10.4, 4x, e10.4, 4x, e10.4, 4x, e10.4, 4x, e10.4)
  END SUBROUTINE

SUBROUTINE getF (v, T, F1, Fs)
!
!
! The liquid part of the Bushman EOS
! returns the value of Free Energy
!

```

!

---

**IMPLICIT NONE**

**DOUBLE PRECISION** :: R  
**DOUBLE PRECISION** :: Z  
**DOUBLE PRECISION** :: V0c  
**DOUBLE PRECISION** :: V0  
**DOUBLE PRECISION** :: a1, a2, a3, a4, a5  
**DOUBLE PRECISION** :: Ac, Bc, Cc  
**DOUBLE PRECISION** :: m, n, l  
**DOUBLE PRECISION** :: Esub  
**DOUBLE PRECISION** :: Ta  
**DOUBLE PRECISION** :: sigmaa  
**DOUBLE PRECISION** :: Tca  
**DOUBLE PRECISION** :: Tsa  
**DOUBLE PRECISION** :: theta0l  
**DOUBLE PRECISION** :: theta0s  
**DOUBLE PRECISION** :: gamma0l  
**DOUBLE PRECISION** :: gamma0s  
**DOUBLE PRECISION** :: B1, D1  
**DOUBLE PRECISION** :: Bs, Ds  
**DOUBLE PRECISION** :: sigmam0  
**DOUBLE PRECISION** :: Tm0  
**DOUBLE PRECISION** :: Am, Bm, Cm  
**DOUBLE PRECISION** :: betai, betao, betam  
**DOUBLE PRECISION** :: Tb  
**DOUBLE PRECISION** :: sigmaz  
**DOUBLE PRECISION** :: Tz  
**DOUBLE PRECISION** :: sigmai  
**DOUBLE PRECISION** :: Ti  
**DOUBLE PRECISION** :: gammaei, gammaeoi, gammaeom  
**DOUBLE PRECISION** :: Tg  
**DOUBLE PRECISION** :: sigmae, sigmad  
**DOUBLE PRECISION** :: tth  
**DOUBLE PRECISION** :: v, T, P, s, F  
**DOUBLE PRECISION** :: sigmaC, sigma, sigmam, x  
**DOUBLE PRECISION** :: tau, Ce, Ce, gammaE, Be  
**DOUBLE PRECISION** :: Fcl, Fal, Fe, Fll, Fm, Ca, theta, thetacl  
**DOUBLE PRECISION** :: Fcs, Fas, Fl, Fs, thetacs

R = 0.31  
 Z = 13.  
 V0c = 0.361  
 V0 = 0.370  
 a1 = 326.35  
 a2 = -1035.44

---

a3 = 858.51  
a4 = -160.59  
a5 = 11.17  
Ac = -12.91  
Bc = 40 .96  
Cc = -28.05  
m = 8.0  
n = 4.99  
l = 0.7  
Esub = 12.1  
Ta = 30.0  
sigmaa = 0.14  
Tca = 25.0  
Tsa = 6.0  
theta0s = 0.1  
gamma0s = 2.19  
Bs = 0.6  
Ds = 0.36  
theta0l = 157.0  
gamma0l = 1.78  
Bl = 1.05  
Dl = 0.0  
sigmam0 = 0.923  
Tm0 = 0.933  
Am = 2.24  
Bm = -5.64  
Cm = 0.21  
betai = 0.0242  
betao = 0.05  
betam = 0.0  
Tb = 8.  
sigmaz = 0.8  
Tz = 200.  
sigmai = 0.3  
Ti = 50.  
gammaei = 0.4  
gammaeo = 0.7  
gammaem = -0.5  
Tg = 300.  
sigmae = 1.0  
sigmad = 9.99 e9  
tth = 2./3.

sigmaC = V0c/v

sigma = V0/v

```

    sigmam = sigma/sigmam0
    x       = log(sigma)
    tau_i   = Ti*exp(-sigma/sigma_i)
!       write(*,*) "Tau : ", tau_i

! Cold Contribution (Lomonosov)
! Solid
    Fcs = 3.*V0c*(a1/1*(sigmaC**(1./3.)-1)+a2/2*(sigmaC**(2./3.)-1)+
&        a3/3*(sigmaC**(3./3.)-1)+a4/4*(sigmaC**(4./3.)-1)+
&        a5/5*(sigmaC**(5./3.)-1))
! Liquid
    if (sigmaC.ge.1.) then
        Fcl = Fcs
    else
        Fcl =V0c*(Ac*(sigmaC**m)/m+Bc*(sigmaC**n)/n+Cc*(sigmaC**l)/l)
&        +Esub
    endif
! Atomic Contribution (Bushman)
! Solid
    thetacs=theta0s*sigma**(tth)*
&        exp((((gamma0s-(tth))*(Bs**2+Ds**2))/Bs)*
&        atan((x*Bs)/(Bs**2+Ds*(x+Ds))))
    Fas=3*R*T*log(thetacs/T)

! Liquid
    Ca = 3.*R/2.*(1+((sigma*Ta)/((sigma+sigmaa)*(T+Ta))))
    thetacl = theta0l*exp(((gamma0l-tth)*(Bl**2.+Dl**2.))/Bl*
&        atan((x*Bl)/(Bl**2.+Dl*(x+Dl))))
    theta = (sigma**tth)*Tsa*((thetacl+T)/(Tca+T))
    Fll = Ca*T*log(theta/T)
    Fm = 3.*R*((2.*(sigmam**2.)*Tm0)/(1.+(sigmam**3.)))*
&        (Cm+3*Am/5*((sigma**(5./3.))-1.)+(Bm-Cm)*T)
    Fal = Fll + Fm
! Electronic Contribution (Lomonosov)
!     Ce = 3.*R/2.*(Z+((sigmaz*sigma*(Tz**2.)*(1.-Z))/((sigma+sigmaz)*
!     &        (T**2.+Tz**2.)))*exp(-tau_i/T))
! Electronic Contribution (Bushman)
! Liquid and Solid
    Ce = 3.*R/2.*(Z+((sigmaz*(Tz**2.)*(1.-Z))/((sigma+sigmaz)*
&        (T**2.+Tz**2.)))*exp(-tau_i/T))
    Cei = 3.*R*Z/2.
    gammaE = gammae_i+(gammae_o-gammae_i+gammaem*T/Tg)*exp(-T/Tg-
&        ((sigma-sigmae)**2./(sigma*sigmad)))
    Be = (2./T**2.)*(Tb*(betam*T-(betai-betao-2*betam)*Tb)*exp(-T/Tb)
&        + betai*(T**2.)/2 - (betai-betao-betam)*Tb*T +

```

```

&      (betai-betao-2*betam)*Tb**2)
Fe = -Ce*T*log(1+((Be*T)/(2*Cei))*sigma**(-1.*gammaE))
!      write(*,*) 'Ce :',Ce," Cei :","Cei," gammaE:",gammaE," Be :","Be,""
!
!      write(*,*) 'Fc :',Fcs," Fa :","Fas," Fe:",Fe,""

F1 = Fcl + Fal + Fe
Fs = Fcs + Fas + Fe
!      F = F1
!      F=Fs
END SUBROUTINE

```

```

SUBROUTINE busheos(P1,E1,V1,T1,Ss,temp)
USE EOS

```

---

```

!  

!SUBROUTINE: busheos  

!  

!DESCRIPTION:  

!    Caculaties the thermodynamic variables using the  

!    Bushman EOS given two thermodynamic variables  

!  

!
```

---

```

IMPLICIT NONE
DOUBLE PRECISION P1, E1, V1, T1, Ss
DOUBLE PRECISION,DIMENSION(3)    :: temp

DOUBLE PRECISION f, df, cu, error, ierror
DOUBLE PRECISION P2, E2, V2, T2, S2,E0,P0,S0
DOUBLE PRECISION,DIMENSION(5)    :: junk

INTEGER, DIMENSION(5) :: num
INTEGER counter, maxcount, count2
v0 = 0.37

```

```

open(88)
write(88,*) "Input parameters"
write(88,101)V1 , T1 , P1 , E1 , Ss , temp(1)
!  

!Checking for non zero variables  

!  

num(1:5) = 0
if(P1.ne.0) then
    num(1)=1
endif

```

```

    if(E1.ne.0) then
        num(2)=1
    endif
    if(V1.ne.0) then
        num(3)=1
    endif
    if(T1.ne.0) then
        num(4)=1
    endif
    if(Ss.ne.0) then
        num(5)=1
    endif
!
!   Finding the two give thermodynamic variables
!
    if(num(1).eq.1) then
        if(num(2).eq.1) then
!           Pressure and Energy
            goto 1000
        elseif(num(3).eq.1) then
!           Pressure and Volume
            goto 1001
        elseif(num(4).eq.1) then
!           Pressure and Temperature
            goto 1002
        elseif(num(5).eq.1) then
!           Pressure and Entropy
            goto 1003
        else
!           write(*,*) 'Two thermodynamic variabls must be specified '
            return
        endif
    elseif(num(2).eq.1) then
        if(num(3).eq.1) then
!           Energy and Volume
            goto 1004
        elseif(num(4).eq.1) then
!           Energy and Temperature
            goto 1005
        elseif(num(5).eq.1) then
!           Energy and Entropy
            goto 1006
        else
!           write(*,*) 'Two thermodynamic variabls must be specified '
            return

```

---

```

        endif
    elseif(num(3).eq.1) then
        if(num(4).eq.1) then
!           Volume and Temperature
            goto 1007
        elseif(num(5).eq.1) then
!           Volume and Entropy
            goto 1008
        else
!           write(*,*) 'Two thermodynamic variables must be specified'
            return
        endif
    elseif(num(4).eq.1) then
        if(num(5).eq.1) then
!           Temperature and Entropy
            goto 1009
        else
!           write(*,*) 'Two thermodynamic variables must be specified'
            return
        endif
    else
!           write(*,*) 'Two thermodynamic variables must be specified'
        return
    endif
!
!   Solving the Bushman EOS using given Pressure and Energy
!   to find specific volume, temperature and entropy
1000 if(temp(1).ne.0) then
        cu=temp(1)    ! Initial guess
    else
        cu=v0*0.25
    endif

    error = 0.001
    ierror = 1.0

    E1 = E1*100
    P1 = P1/100

    return
!
!   Solving the Tillotson EOS using given Pressure and Volume
!   to find entergy, temperature and entropy
1001 if(temp(1).ne.0) then

```

---

```

        cu=temp(1)    ! Initial guess
    else
        cu=v0*0.25    ! Initial temperature guess
    endif

    return

!
! Solving the Bushman EOS using given Pressure and Temperature
! to find energy, specific volume, and entropy
!
1002 write(*,*) 'This Feature needs to be added'
    return

!
! Solving the Bushman EOS using given Pressure and Entropy
! to find energy, specific volume, and temperature
!
1003 write(*,*) 'This Feature needs to be added'
    return

!
! Solving the Bushman EOS using given Energy and Volume
! to find pressure, temperature and entropy
!
1004 error = 0.005
    ierror = 1.0
!    open(88)
    maxcount = 20
    counter = 0
    count2 = 0

    if (temp(1).ne.0) then
        cu=temp(1)/1000
    else
        cu=0.3    ! Initial guess 293 K
!        cu = 7
    endif
    E0 = 0.2748
    P0 = 1.839
!    S0 = 1.889
!    E0 = 0.1397e2
!    P0 = 0.3446e-7
    S0 = 0.5283e+2
!    S0 = 0.0
    E1 = E1*100+E0
    write(88,*) "E1 =",E1, "T1=",cu

```



---

```

E2 = 0.0
do while (ierror.ge.error)
  CALL bush(junk(2),E2,V1,cu,junk(3),temp)
  write(88,*) cu, V1, E2
  f=E1-E2
  T2 = cu*1.001
  junk(1:10) = 0.
  E2 = 0.0
  CALL bush(junk(2),E2,V1,T2,junk(3),temp)
  write(88,*) T2, V1, E2
  junk(1) = E1-E2
  df = (junk(1)-f)/(T2-cu)
  cu = cu - f/df
  if (cu.lt.0) then
    cu = cu+f/df + 1.
    write(88,*) "New T:" ,cu
  endif
  junk(1:10) = 0.
  E2 = 0.0
  write(88,*) cu, V1,E2
  CALL bush(junk(2),E2,V1,cu,junk(3),temp)
  ierror = abs(E1-E2)/E1
  write(88,*) cu, V1,E2,ierror
  if (counter.ge.maxcount) then
    write(88,*) "Maximum number of iteration reached"
    cu = 0.5*(cu+cu+f/df)
    write(88,*) "Reseting Cu to:" , cu
    count2 = count2+1
    counter = 0
  endif
  if (count2.eq.5) then
    write(88,*) "Could not converge on a answer"
    ierror =0.0
  endif
  counter = counter + 1
enddo
T1=cu*1000
P1=(junk(2)-P0)/100
E1= (E1-E0)/100
Ss=junk(3)-S0
!   close(88)

return
!
!   Solving the Bushman EOS using given Energy and Temperature

```

---

```

!      to find pressure, specific volume, and entropy
!
1005 write(*,*) 'This Feature needs to be added'
      return
!
!      Solving the Bushman EOS using given Energy and Entropy
!      to find pressure, specific volume, and temperature
!
1006 write(*,*) 'This Feature needs to be added'
      return
!
!      Solving the Bushman EOS using given Volume and Temperature
!      to find pressure, energy, and entropy
!
1007 CALL bush(P1,E1,V1,T1,Ss,temp)
      write(*,*) "Volume and Temperature"
      return
!
!      Solving the Bushman EOS using given Volume and Entropy
!      to find pressure, energy, and temperature
!
1008 error = 0.005
      ierror = 1.0
!      open(88)
      maxcount = 40
      counter = 0
      count2 = 0
      E0 = 0.2748
      P0 = 1.839
!      S0 = 1.889
!      E0 = 0.1397e2
!      P0 = 0.3446e-7
      S0 = 0.5283e2
!      S0 = 0.0
      Ss = Ss - S0

      if (temp(1).ne.0) then
          cu=temp(1)/1000
      else
          cu=0.3 ! Initial guess 293 K
!          cu = 7
      endif
      S2 = 0.0
      do while (ierror.ge.error)

```

---

```

CALL bush(junk(2),junk(3),V1,cu,S2,temp)
write(88,*) cu, V1, S2
f=Ss-S2
T2 = cu*1.001
junk(1:10) = 0.
E2 = 0.0
CALL bush(junk(2),junk(3),V1,T2,S2,temp)
write(88,*) T2, V1, S2
junk(1) = Ss-S2
df = (junk(1)-f)/(T2-cu)
cu = cu - f/df
if (cu.lt.0) then
    cu = cu+f/df + 1.
    write(88,*) "New T:" ,cu
endif
junk(1:10) = 0.
E2 = 0.0
write(88,*) cu, V1,E2
CALL bush(junk(2),junk(3),V1,cu,S2,temp)
ierror = abs(Ss-S2)/Ss
write(88,*) cu, V1,S2,ierror
if (counter.ge.maxcount) then
    write(88,*) "Maximum number of iteration reached"
    cu = 0.5*(cu+cu+f/df)
    write(88,*) "Reseting Cu to:" , cu
    count2 = count2+1
    counter = 0
endif
if (count2.eq.5) then
    write(88,*) "Could not converge on a answer"
    ierror =0.0
endif
    counter = counter + 1
enddo
T1=cu*1000
P1=(junk(2)-P0)/100
E1=(junk(3)-E0)/100
Ss=S2+S0
!   close(88)
return
!
!   Solving the Bushman EOS using given Temperature and Entropy
!   to find pressure, energy, and specific volume
!
1009 write(*,*) 'This Feature needs to be added'

```

```

    return

101 format(3x,e10.4,4x,e10.4,4x,e10.4,
&          4x,e10.4,4x,e10.4,4x,e10.4)

    END SUBROUTINE

!-----
!  

!  

!  

!  

!  

!  

!-----
SUBROUTINE FindV(Pg,vout,temp)
USE EOS
IMPLICIT NONE
DOUBLE PRECISION Pg, vout
DOUBLE PRECISION f, df, Peos, Peos2, f2
DOUBLE PRECISION ierror, error, cu, E1,v2

DOUBLE PRECISION,DIMENSION(3)  :: temp
DOUBLE PRECISION,DIMENSION(10) :: junk

v0 = 0.37

error = 0.001
ierror = 1.0
cu = v0*0.7
!  

!   cu = 0.276
temp(1:3) = 0.
junk(1:10) = 0.0
open(99)
write(*,*) "Start findV"

do while (ierror.ge.error)
    Up=sqrt(Pg*(v0-cu))
    E1=(Up**2)/2.
!  

!   E1 = 0.5*Pg*(v0-cu)
temp(1) = junk(2)
write(99,*) Up, E1, temp(1)
Peos = 0.0
junk(1:10) = 0.0
call busheos(Peos,E1,cu,junk(2),junk(3),temp)

```

---

```

    f = Pg - Peos
    write(99,*) Peos, Pg, cu

    v2 = cu*1.01
    Up=sqrt(Pg*(v0-v2))
    E1=(Up**2)/2.
!    E1 = 0.5*Pg*(v0-v2)
    temp(1) = junk(2)
    write(99,*) Up, E1, temp(1)
    Peos = 0.0
    Peos2 = 0.0
    junk(1:10) = 0.0
    call busheos(Peos2,E1,v2,junk(2),junk(3),temp)
    f2 = Pg- Peos2
    write(99,*) Peos2, Pg, v2

    df = (f2-f)/(v2-cu)
    cu = cu - f/df
    write(99,*) f, f2, df, cu

    Up=sqrt(Pg*(v0-cu))
    E1=(Up**2)/2
    temp(1) = junk(2)
    Peos2 = 0.0
    junk(1:10) = 0.0

    call busheos(Peos,E1,cu,junk(2),junk(3),temp)
    ierror = abs((Peos-Pg)/Pg)
    write(99,*) Peos, v2, ierror

enddo
temp(3) = Up
vout = cu
close(99)
write(* ,*) "Found V"
return

END SUBROUTINE

```