Compatibility and performance of SOFCs based on lanthanum tungstates

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1 Abstract

Rare-earth tungstates with general composition "Ln₆WO₁₂" have attracted great attention in last few years due to their relatively high mixed protonelectron conductivity [1, 2]. One of the main advantages of these electrolytes, compared to the traditional perovskites based on BaCeO₃, is that they exhibit high tolerance towards CO₂ and H₂S environments. Therefore, this material is a potential electrolyte for proton conducting solid oxide fuel cells (PC-SOFC).

In this work, the lanthanum tungstate with composition $La_{27}W_4NbO_{55-\delta}$ (LWNO) has been tested as proton conductor electrolyte [3]. For this purpose, different electrodes and composite electrodes have been considered, including: $La_{0.8}Sr_{0.2}MnO_{3-\delta}$, $La_{0.6}Sr_{0.4}Co_{1-x}Fe_xO_{3-\delta}$, $La_{0.5}Sr_{0.5}Cr_{0.5}Mn_{0.5}O_{3-\delta}$, $SrFe_{0.75}Nb_{0.25}O_{3-\delta}$ and NiO.

Chemical compatibility between the cell components is investigated by X-ray powder diffraction (XRPD) and energy dispersive spectroscopy (EDS). Furthermore, area specific resistance (ASR) of the different electrodes is determined in symmetrical cells by impedance spectroscopy. XRPD and EDS analysis do not reveal significant bulk reactivity between most of these electrodes and LWNO electrolyte in the typical operating temperature range of a SOFC (600-900 °C).

However, minor interdiffusion of elements at the electrolyte/electrode interface affects both the ohmic losses and electrode polarization of the symmetric cells. ASR values are significantly improved by using a buffer layer of Ce_{0.8}Gd_{0.2}O_{1.9}, between the electrolyte and electrode materials, to prevent reactivity. A single cell with 350 µm thick electrolyte, NiO-Ce_{0.8}Gd_{0.2}O_{1.9} anode and La_{0.6}Sr_{0.4}Co_{0.8}Fe_{0.2}O_{3.8} cathode, generates maximum power densities of 140 and 18 mWcm⁻² at 900 and 650 °C, respectively. Hence, lanthanum tungstates could be competitive proton conductors for PC-SOFCs with similar performance to those based on BaZrO₃ if thin film electrolytes are used.

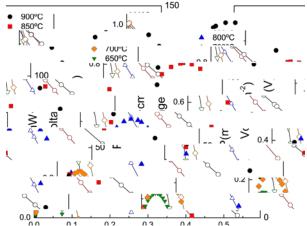


Fig. 1. Cell voltage and power density as a function of current density at different temperatures using air and 5% H₂–Ar as oxidant and fuel respectively.

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3 References

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