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Pushpa Raghani
Boise State University

Pankaj Kumar
Boise State University

Balaji Ramanujam
Boise State University

Alex Punnoose
Boise State University

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Defect driven magnetism in doped SnO₂ nanoparticles: Surface effects

Raghani Pushpa, Pankaj Kumar, Balaji Ramanujam and Alex Punnoose

Department of Physics, Boise State University

Boise, ID, USA

Email: pankajsahota@gmail.com

Magnetism and energetics of intrinsic and extrinsic defects and defect clusters in bulk and surfaces of SnO₂ is investigated using first-principles to understand the role of surfaces in inducing magnetism in Zn doped nanoparticles. We find that Sn vacancies induce the largest magnetic moment in bulk and on surfaces. However, they have very large formation energies in bulk as well as on surfaces. Oxygen vacancies on the other hand are much easier to create than V_{Sn}, but neutral and V_O⁺² vacancies do not induce any magnetism in bulk as well as on surfaces. V_O⁺¹ induce small magnetism in bulk and on (001) surfaces. Isolated Zn_{Sn} defects are found to be much easier to create than isolated Sn vacancies and induce magnetism in bulk as well on surfaces. Due to charge compensation, Zn_{Sn}+V_O defect cluster is found to have the lowest formation energy amongst all the defects; it has a large magnetic moment on (001), a small magnetic moment on (110) surface and it is non-magnetic in bulk. Thus, we find that Zn_{Sn} and Zn_{Sn}+V_O defects on the surfaces of SnO₂ play an important role in inducing the magnetism in Zn-doped SnO₂ nanoparticles.

Keywords—density functional theory; defects; defect formation energies; magnetism

1. INTRODUCTION

Recently, ferromagnetism in dilute magnetic semiconductors (DMS) has received a considerable attention owing to potential applications in spintronic and optoelectronic devices [1, 2]. In DMS, magnetism can be induced in many ways. For instance, doping with magnetic transition metal (TM) is one of the well-accepted ways of inducing magnetism in DMS. However, doping with magnetic TMs can lead to segregation and precipitation, which can limit the applications of these compounds [4-9]. Therefore, an extensive research is undergoing to overcome these difficulties. The oxide based DMS are an ideal choice for induced magnetism in the presence of defects, as these systems have high Curie temperature (T_C). It has been found experimentally that some of the semiconductor oxides and nitrides show induced magnetism with nonmagnetic dopants such as Li, C, N, Zn and Cd. Examples include wide band gap semiconductors such as ZnO [10-12], TiO₂ [13] and SnO₂ [14-20].

Among the oxide semiconductors, SnO₂ has drawn much interest due to the fascinating properties such as wide band gap, high optical transparency, electrical conductivity and chemical sensitivity [21-23]. These properties make SnO₂ based structures an ideal candidate for many applications such as in solar cells and catalysis [24]. Recent reports on doping the bulk and nanoparticles of SnO₂ with magnetic TMs have shown huge ferromagnetism at room and above room temperatures. The possibility of room and high temperature ferromagnetism in magnetic TM doped SnO₂ systems has got a lot of attention in the field of DMS due to its applications in spintronic devices [25-33]. However, the exact mechanism of induced magnetism in the nanoparticles of SnO₂ with nonmagnetic dopants is still under debate. Nonmagnetic TM atom (Zn, Cu) doped nanoparticles of SnO₂ give rise to a large magnetic moment as compared to their bulk counterparts. To understand the onset of this magnetism, we study the intrinsic and extrinsic defects and defect clusters on various surfaces of SnO₂ in a nanoparticle [34-38].

A nanoparticle of a material can have a number of orientations of the surfaces. The areal predominance of these surfaces will depend upon the surface energy — energy per unit area of the surface, *i.e.*, the surface with highest energy will have lowest area and vice-versa. However, the exact ratio of the different surfaces in a nanoparticle will depend upon the experimental conditions such as temperature and pressure. Due to anisotropic nature of surfaces, they will exhibit anisotropic magnetic properties, *e.g.*, magnetism and magnetic anisotropy can be altered by

changing the chemical composition by creating the defects [39-41]. The overall magnetization of a nanoparticle will be a complex interplay among the magnetizations induced on different surfaces. To predict the overall magnetization of a nanoparticle, a much more complex modeling is required which is beyond the scope of this paper. In principle, one needs to calculate the formation energy (FE), magnetism and easy axis of magnetization for all the surfaces that would exist in a nanoparticle, which is a formidable task. **Instead, we investigate the magnetism and energetics of two surfaces having the highest and the lowest surface energies of 11.23 and 6.23 eV/Å² (1.82 and 1.01 J/m²) for (001) and (110) respectively, which are comparable to the values found in literature** [36, 42, 43]. Using density functional theory (DFT), we study the effect of intrinsic defects and defect clusters such as V_O, V_{Sn}, Zn_{Sn}+V_O and Zn_{Sn}+V_{Sn} on the electronic and magnetic properties of bulk and (001) and (110) surfaces of SnO₂. We find that Zn_{Sn}+V_O defect clusters have the lowest FE of all the defects considered in this paper and the FE further decreases on the surfaces and is minimum on (110) surface. The Zn_{Sn}+V_{Sn} defect on the other hand leads to a huge magnetic moment on (110) surface but it has higher FE as compared to the Zn_{Sn}+V_O defect. By calculating the magnetic moments of Sn vacancies on symmetric (S) and asymmetric (A) (001) and (110) surfaces of SnO₂, we find that easy axis of magnetization for V_{Sn} on (001) is out of plane whereas it is in-plane for (110).

2. COMPUTATIONAL DETAILS

We use DFT to study the electronic and magnetic properties of various point defects and defect clusters in bulk SnO₂ and its (001) & (110) surfaces. **Surfaces are modeled using slabs of atomic layers separated by a vacuum.** Within DFT, we employ the pseudopotential formalism as implemented in Quantum-ESPRESSO [44]. Interactions between the valence electrons and nuclei are treated using norm-conserving scalar relativistic pseudopotentials. The generalized gradient approximation (GGA) is employed for the exchange-correlation potential with Perdew-Burke-Ernzerhof (PBE) exchange correlation functional [45]. **The pseudopotentials of Sn, O and Zn are generated from their atomic configurations [Kr] 4d¹⁰ 5s² 5p², 1s² 2s² 2p⁴ and [Ar] 3d¹⁰ 4s². The corresponding valency of the atoms is +4,-2 and +2, respectively.** In general, SnO₂ crystallizes in the rutile structure. The optimized lattice parameters are $a = b = 4.843 \text{ \AA}$ and $c = 3.314 \text{ \AA}$ which are comparable to the earlier experimental and theoretical findings [5, 46]. To do the electronic

structure calculations, we use a Monkhorst-Pack scheme to generate the k -points [47]. For the $1 \times 1 \times 1$ unit cell that contains two Sn and four O atoms, a k -mesh of $6 \times 6 \times 9$ is used. For larger supercells, the k -mesh size is changed proportionally. A plane wave kinetic energy cutoff of 100 Ry is used for all calculations and forces are converged to 0.01 eV/\AA .

To determine the direction of magnetization axis, we use symmetric and asymmetric slabs. If the magnetization is perpendicular to the surface, then the total magnetization of the symmetric slab will be zero whereas it will be non-zero for an asymmetric slab. We realize that this is a crude way of determining the direction of magnetization axis and one has to perform anisotropy calculations including spin-orbit coupling. However, these calculations can be computationally demanding and time-consuming especially for large supercells used in this work. The symmetric slab contains seven atomic layers. The central three layers are fixed at their bulk-like sites, and rest of the layers on top and bottom of the slabs are allowed to relax. The in-plane size of the supercell is $2 \times 2 \times 3$ along both (001) and (110) directions with a vacuum of $\sim 13 \text{ \AA}$ along the z -direction. The asymmetric slab contains five atomic layers, having bottom three layers fixed at bulk-like sites. The schematic picture of a (110)-oriented slab with $\text{Zn}_{\text{Sn}} + \text{V}_{\text{Sn}}$ defect is shown in fig.1(a) and fig. 1(b). To do the supercell calculations, a k -mesh of size $4 \times 2 \times 1$ is used.

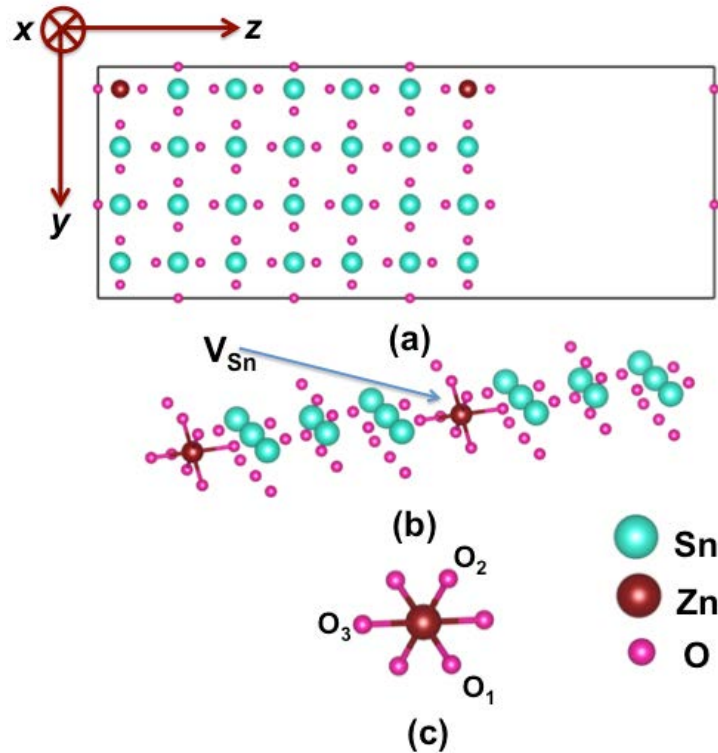


Fig. 1: (a) Schematic picture of the (110) slab with $Zn_{Sn}+V_{Sn}$ defect: (a) and (b) are the sideviews of the slab showing Sn atom substituted by Zn and Sn vacancy; (c) shows the O atoms surrounding the Zn atom in the doped region.

The formation energies of the defects are calculated using the equation:

$$E = E_{def} - E_{undef} + \sum_i \Delta n_i \mu_i$$

where E_{def} and E_{undef} represents the calculated total energies of systems in the presence and absence of defects, respectively. The quantity Δn_i is the difference in the number of atoms of i^{th} species between defect-free and defected supercells and μ_i is the chemical potential of i^{th} species. The chemical potentials of Sn and O atoms are calculated using the cohesive energies of bulk Sn and O_2 molecule respectively. Using these chemical potentials, the formation energies are calculated in Sn-rich and O-rich conditions. Detailed explanation on how to calculate the formation energies in Sn-rich and O-rich conditions is given in Ref.[48].

3. RESULTS AND DISCUSSION

3.1. Vacancies in bulk and at surfaces of SnO_2

Table 1 shows the calculated defect formation energies and magnetic moments per supercell in the presence of intrinsic defects such as V_{Sn} and V_O in bulk and surfaces of SnO_2 in both Sn- and O-rich conditions. Although, there is a large magnetic moment due to V_{Sn} in bulk, the FE for this vacancy is found to be very large~11.5 eV. These data agree very well with previous calculations [5, 49]. FE for V_O is much smaller than that of V_{Sn} but it does not induce any magnetism. We find that doubly ionized oxygen vacancies are also nonmagnetic. However, singly ionized oxygen vacancies do induce magnetism in SnO_2 in agreement with the previous work [4, 36, 38, 50].

Both the (001) and (110) surfaces are also found to show nonmagnetic behaviour in the presence of (neutral and doubly ionized) oxygen vacancies similar to that in bulk SnO_2 which is in agreement with the previous DFT results [15]. The singly ionized V_O on the other hand are

found to be magnetic on (001) surface and nonmagnetic on (110) surface. As expected, FEs of these defects on surfaces are lower than those in bulk. However, Sn vacancies are still much more difficult to create even on surfaces and cannot be created instantaneously in any environment. Oxygen vacancies on the other hand can be created instantaneously on surfaces in Sn-rich environment. Overall, creating both types of vacancies on (001) surface are easier than creating them on (110) surface, this is due to the higher surface energy of (001) than that of (110) surface.

The Sn and oxygen vacancies on the symmetric and asymmetric surfaces of (110) are found to have nearly same formation energies. However, Sn vacancy on the symmetric (001) slab has a higher FE of 9.01 eV than on the asymmetric (001) slab which is about 7.5 eV. We also find that the magnetic moments in both the cases are very different. The magnetic moment on the symmetric slab comes out to be zero whereas it is nonzero on the asymmetric slab. This suggests that the magnetic moment of V_{Sn} on (001) surface is perpendicular to the plane of the surface and the magnetic moments from the top and bottom of the surface cancel each other resulting in zero magnetic moment. On asymmetric slab, no such cancellation occurs and the magnetic moment of Sn vacancy on A-(001) surface is almost same as that on S & A-(110) surfaces. This suggests that the magnetic moment of Sn vacancy does not get affected by its environment and surface structure.

Table 1: Defect formation energies (in eV) of O and Sn vacancies in bulk and on surfaces within Sn- and O-rich conditions. M_{cell} is the magnetization of the supercell in μ_B per cell

System	Defect	Sn-rich	O-rich	M_{cell}
Bulk	V_{O}^0	1.08	3.27	0.00
	V_{Sn}^0	11.47	7.08	3.98
S-(001)	V_{O}^0	-1.68	0.51	0.00
	V_{Sn}^0	9.01	4.62	0.00
A-(001)	V_{O}^0	-1.67	0.52	-0.20
	V_{Sn}^0	7.46	3.07	4.18
S-(110)	V_{O}^0	-0.64	1.56	0.00

	V_{Sn}^0	10.10	5.71	4.00
A-(110)	V_O^0	-0.72	1.48	0.00
	V_{Sn}^0	10.09	5.71	3.83

On (110) surface, the magnetic moment of V_{Sn} is same for symmetric and asymmetric surfaces suggesting that magnetic moment is in the plane of the surface. From this data, we can say that easy axis of magnetization of V_{Sn} on (001) is perpendicular to surface, whereas it is in-plane for the (110) surface. As mentioned earlier, V_{Sn} are difficult to create as compared to V_O but V_{Sn} are responsible for induced magnetism in SnO_2 . Therefore, to induce the magnetism in SnO_2 , the FE of V_{Sn} needs to be reduced. One of the ways of doing this would be to introduce the dopants in the system, which is discussed in the next section.

3.2. Effect of Zn doping on intrinsic defects

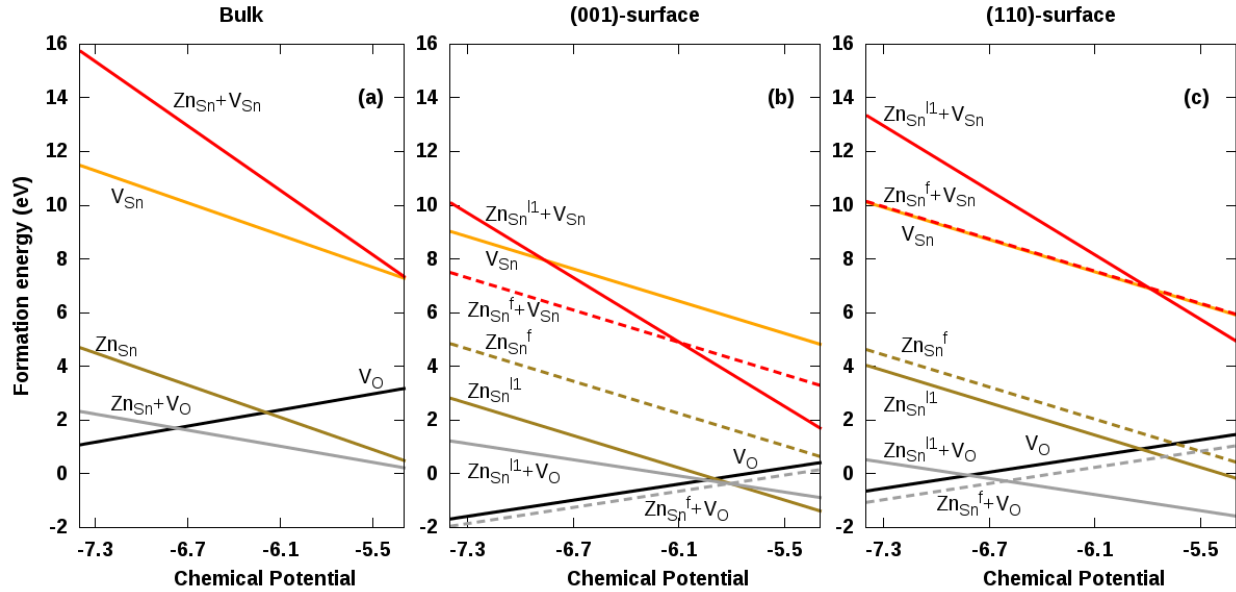


Fig. 2: Formation energies of various isolated defects and defect clusters calculated as a function of oxygen chemical potential: (a) in bulk SnO_2 ; (b) on (001) surface and (c) on (110) surface. The dashed lines are for doping Zn atom in the bulk region (Zn_{Sn}^f) and creating intrinsic defect in the surface layer

Here, we investigate the defect clusters involving Zn and intrinsic defects on the surfaces and their influence on the magnetism in SnO_2 . We consider two types of defect clusters: (i) defect

clusters entirely in the surface layer, and (ii) Zn_{Sn} in bulk and the intrinsic defect in the surface layer. Fig. 2 shows the variation in FE of different types of defects in bulk and on surfaces as a function of chemical potential of oxygen [48, 51]. From Fig. 2 (a), the FE of $Zn_{Sn} + V_O$ defect in bulk SnO_2 is 2.31 eV, which is much lower than that of the isolated Zn_{Sn} defect. It is also lower than the FE of isolated oxygen vacancy in O-rich conditions. The further decrease in FE of $Zn_{Sn} + V_O$ is due to the charge compensation between the low valence Zn atoms and negatively charged O atoms. As a result of this charge compensation, the oxygen vacancy associated with the Zn defect quenches the magnetic moment induced by Zn, resulting in zero total magnetization of the cell. FE of $Zn_{Sn} + V_{Sn}$ is 15.7 eV, which is much greater than those of isolated Zn_{Sn} and V_{Sn} defects. However, V_{Sn} with Zn defect increases the magnetic moment by about 40% as compared to the Zn_{Sn} defect alone. On the surfaces, the FE for the $Zn_{Sn}+V_O$ defect cluster further decreases and is lower than those of isolated Zn_{Sn} defects. Interestingly, the $Zn_{Sn} + V_O$ has the lowest FE on (110) surface than in bulk and (001) surface; whereas, all the other isolated defects are easier to form on (001) surface than on (110) surface. The lowest FE for $Zn_{Sn} + V_O$ on (110) surface can be inferred from the charge compensation caused by the reduction in surface co-ordination.

Doping Zn atom in the bulk region of the slab with O vacancy in the surface layer results in the decrease in FE of V_O from 0.51 to 0.25 eV in O-rich conditions on (001) surface. On (110) surface, it decreases from 1.48 to 1.14 eV. On the other hand, Zn_{Sn} defect in the bulk region of the slab does not affect the formation energy of V_{Sn} on (110) surface. However, FE of V_{Sn} decreases from 4.62 to 3.09 eV when Zn atom is substituted for Sn in the bulk region of (001) surfaces. Hence, we find that $Zn_{Sn} + V_O$ is much easier to create than $Zn_{Sn} + V_{Sn}$, both in bulk and on surfaces and $Zn_{Sn} + V_O$ has the lowest FE of all the intrinsic defects and defect clusters. Unlike the bulk case, this defect cluster is found to induce a finite magnetic moment on (001) surface due to lack of charge compensation caused by reduction in the surface co-ordination. FE of $Zn_{Sn} + V_{Sn}$ in bulk is much higher than those of isolated Zn_{Sn} and V_{Sn} defects. However, its FE on (001) surface greatly decreases to 10 eV on (001) surface whereas on (110) surface, it is 13.3 eV and induces a large magnetic moment of $\sim 6.00 \mu_B$ per cell on (110) surface, indicating a ferromagnetic coupling between the two defects. The $Zn_{Sn} + V_{Sn}$ defect is found to have a large increase in the magnetic moment than Zn_{Sn} alone on the (110) surface. The (001) surface does not retain the magnetic moment for $Zn_{Sn} + V_{Sn}$ defect cluster, but with antiferromagnetic coupling.

To get the atomic insight of the large magnetism in Zn doped (110) SnO₂ with intrinsic cation defects (V_{Sn}), we plot the orbital projected density of states (DOS) of O atoms surrounding the Zn dopant shown in Fig. 3. We considered only those O atomic sites, which are surrounding the dopant in the supercell. Note that, there are six O atoms surrounding the Zn atom, making a distorted octahedron. Due to the identical environment of three O atoms with other O atoms in the octahedron, we plotted the DOS of those O atoms that have different environment and are shown in Fig. 3(a-c). From the Fig. 3(a-c), it is clear that the magnetism is mainly caused by the 2*p* orbitals of O atoms. The O atoms facing the surface are found to produce higher DOS as compared to others as shown in Fig. 3(b). The large magnetism in the case of Zn_{Sn} + V_{Sn} defect originates from the ferromagnetic coupling of Zn_{Sn} and V_{Sn} defect and the hybridization of nearest neighbor 2*p* orbitals of oxygen atoms with the 3*d* orbitals of Zn dopant. In other words, when the higher valence atom such as Sn is substituted with lower valence dopant Zn, then it impinges holes into the 2*p* orbitals of the neighboring O atoms which leads to shift in the minority DOS of O atoms as shown in Fig. 3(a-c). Note that the dopant itself does not get magnetized extensively (Fig. 3(d)), but induces the magnetism in the neighboring O atoms. The partial DOS of nearest neighbor Sn atoms are not shown because they do not show any appreciable signatures of magnetism.

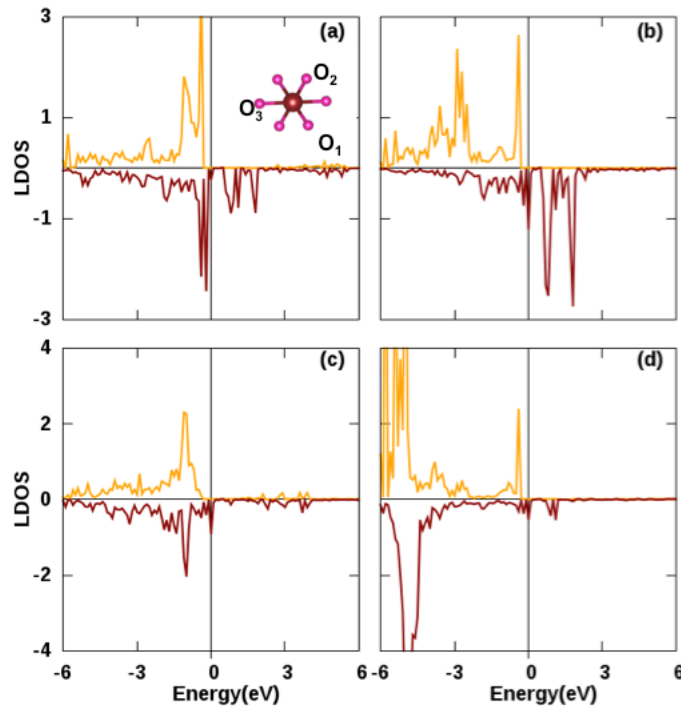


Fig. 2: Spin polarized projected density of states of (a-c) $2p$ orbitals of three types of O (O_1 , O_2 and O_3 as shown in the inset) atoms surrounding Zn (d) d orbitals of Zn dopant in a $Zn_{Sn}+V_{Sn}$ defect in (110) symmetric slab of SnO_2

4. CONCLUSIONS

In summary, magnetic behavior and energetics of different kinds of defects are studied in bulk, on (001) and (110) surfaces, to understand the role of surfaces in inducing the magnetism in SnO_2 nanoparticles. We find that isolated V_{Sn} gives large magnetic moment, both in bulk and on surfaces, but they are very difficult to create due to large formation energy. However, isolated Zn_{Sn} defects alone are found to have a larger magnetic moment and are also easier to create on both the (001) and (110) surfaces than in bulk. Sn vacancies along with Zn_{Sn} defects ($Zn_{Sn}+V_{Sn}$ defect cluster) become even harder to create even on surfaces (as expected), but they induce a large magnetic moment of $6 \mu_B/\text{cell}$ on (110) surface and a small magnetic moment of $2 \mu_B/\text{cell}$ on (001) surface. The Zn_{Sn} defects along with O defects are found to have the lowest formation energy of all the defects and induce small magnetism on surfaces. The magnetic moment induced by $Zn_{Sn}+V_O$ defects on (001) surface is larger than (110) surface, however, the FE of this defect is higher on (001) surface than on (110) surface. In fact, $Zn_{Sn}+V_O$ is the only defect that has lower formation energy on (110) surface than on (001) surface. Based on these observations we believe that $Zn_{Sn}+V_O$ defect clusters play an important role on surfaces in inducing the magnetism in SnO_2 nanoparticles.

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