

Radiationless decay spectrum of O 1s double core holes in liquid water

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ABSTRACT

We present a combined experimental and theoretical investigation of the radiationless decay spectrum of an O 1s double core hole in liquid water. Our experiments were carried out using liquid-jet electron spectroscopy from cylindrical microjets of normal and deuterated water. The signal of the double-core-hole spectral fingerprints (hypersatellites) of liquid water is clearly identified, with an intensity ratio to Auger decay of singly charged O 1s of 0.0014(5). We observe a significant isotope effect between liquid H₂O and D₂O. For theoretical modeling, the Auger electron spectrum of the central water molecule in a water pentamer was calculated using an electronic-structure toolkit combined with molecular-dynamics simulations to capture the influence of molecular rearrangement within the ultrashort lifetime of the double core hole. We obtained the static and dynamic Auger spectra for H₂O, (H₂O)₅, D₂O, and (D₂O)₅, instantaneous Auger spectra at selected times after core-level ionization, and the symmetrized oxygen-hydrogen distance as a function of time after double core ionization for all four prototypical systems. We consider this observation of liquid-water double core holes as a new tool to study ultrafast nuclear dynamics.

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I. INTRODUCTION

As nature's most important liquid, water remains of continuing interest to natural scientists because of its key role in chemistry and biology, since most chemical reactions and biological functions occur in aqueous environments, and because of its anomalous properties.¹ The latter is intimately linked to the hydrogen-bonding structure in liquid water. Various x-ray spectroscopy techniques sensitive to the chemical environment of a specific element have been

shown to be efficient probes of water's electronic and hydrogen-bond structures.² Photoemission spectra recorded upon ionization in the soft-x-ray region reflect the electronic and nuclear dynamics in aqueous solutions (see, e.g., Refs. 3–8). The simultaneous presence of electronic and nuclear dynamics in liquid water complicates the interpretation of the measurements. In a number of studies, altering the nuclear dynamics via isotopic substitution was used to aid in separating these two contributions.^{4,9,10} The dynamics in water induced by core-level ionization has also been studied by

x-ray emission spectroscopy of oxygen K -shell vacancies, exploring the excitation-energy dependence of the spectrum and the isotope effect.^{11–14} Another aspect of the interaction of water with ionizing radiation is the production of reactive decay products, such as OH radicals, ions, and free electrons, generated by the primary ionization followed by a cascade of reactions and secondary ionization processes.^{15–17} Therefore, knowledge of the dynamic response of water to ionizing x rays is an important aim of current water studies.

The time scale for the decay of a single oxygen core hole in liquid water can be estimated from the lifetime broadening in the O $1s$ main line of gaseous water as 4 fs.¹⁸ A window into even shorter time scales is provided by the decay of double core holes (DCHs).¹⁹ This refers to a state with two inner-shell holes created in sufficient temporal and spatial proximity to interact with each other. Whereas for single-core-level ionization, the liquid environment around a water molecule substantially influences the ultrafast core-hole-induced proton dynamics,^{4,10} the impact of the chemical environment, i.e., substances in the close neighborhood of the molecule, has so far not been addressed for DCHs. In an isolated water molecule, the DCH leads to an ultrafast dissociation of both hydrogen atoms as protons.^{19,20} It remains an open question how strongly the chemical environment influences this ultrafast dissociation. Moreover, for a single core hole, neighboring water molecules can to some extent directly participate in the core-hole decay via intermolecular Coulombic decay or via electron-transfer-mediated decay processes,^{4,10} which have not been reported for DCHs. Here, we will present experimental and theoretical results on the Auger decay of DCHs produced by simultaneously ionizing both oxygen core electrons of one water molecule in the liquid phase.

Double core holes were first discovered in the study of K -shell capture in radioactive isotopes, which were found to contain two K -shell holes in a single metal atom for some decay pathways.^{21,22} Analogous to single core holes, they may relax by either Auger decay or x-ray emission. The latter became the preferred probe to observe double K -shell holes in the decades following their first observation. Significant energy shifts in the x-ray emission of K -shell DCHs, compared to the characteristic x rays of the same element, were observed, which led to the term “hypersatellites” for these x-ray lines.²³ Studies of the Auger decay of DCHs greatly accelerated after it became apparent that these states are also created in ion-atom collisions; the term hypersatellites was soon also used to designate the energy-shifted Auger lines emerging from DCHs.²⁴

In recent years, strong experimental and theoretical activities have been devoted to the study of inner-shell double photoionization,²⁵ prompted both by the perspective to create such states using x-ray free-electron lasers (XFELs) with very high brightness^{26,27} as well as by the availability of sophisticated coincidence detection schemes that greatly enhance the amount of information on DCHs compared to using more conventional synchrotron light sources (see, e.g., Refs. 28–36). The high brightness of XFELs has promoted successive absorption of many x-ray photons by atoms at the atomic inner shells within a few tens of femtoseconds, leading to high spectral complexity.³⁷ The ensuing decay products could nevertheless be disentangled by suitable analysis methods.³⁸ We do not touch upon this type of experiment further here but restrict ourselves to DCHs produced by the absorption of a single photon

from a synchrotron radiation source. The DCH creation and an exemplary radiationless decay channel are sketched in Fig. 1. Mechanisms for single-photon double photoionization in general were delineated and compared, e.g., in Ref. 39. These authors distinguish between knockout, essentially an electron-electron collision, and shake-off, a quantum mechanical effect where the non-orthogonality of initial-state orbitals with a continuum wave function in the singly photoionized state fosters the transition of a second, bound electron into the continuum. For DCHs, a number of studies have targeted the latter of these mechanisms, but mostly in the context of singly ionized states that feature an additional core-to-valence *excitation* (as opposed to *ionization* in our experiment); see Refs. 33, 40, and 41 and, with a focus on water, Ref. 34. A complete picture of the creation of water DCHs is missing so far, however.

In our experiment, DCHs are created at a single molecular site of liquid water, i.e., we discuss states with configuration $(\text{H}_2\text{O}^{2+} 1s^{-2})(\text{H}_2\text{O})_N(\text{aq})$. This doubly ionized state is expected to relax via several radiationless decay steps. In gaseous water, the hypersatellite Auger spectrum of the first decay step was investigated experimentally and theoretically.^{19,20,42,43} The first radiationless decay step was found to take place within 1.5 fs (compared to the 4 fs lifetime of a singly ionized core hole in water; see earlier). In this sense, the liquid-water DCH can be considered a new tool for studying ultrafast nuclear dynamics on a yet shorter time scale. Here, we will present the electronic decay spectra of the same vacancy states measured in liquid water. In an attempt to isolate the effects of nuclear dynamics, measurements will be presented for normal and heavy water. Our experiment opens a window into liquid-water nuclear dynamics in the one-femtosecond range (i.e., faster compared to previous single-core-hole studies in solution), as we do not assume that the DCH lifetime is strongly influenced by the liquid environment.

For gaseous water, the experimental results showed that besides $1s^{-2}$ DCHs, the decay of satellite states,^{20,41,44} featuring a $1s^{-2}$ DCH and an additional, excited electron-hole pair, also plays a role. A greater importance of satellite states compared to single core ionization is common for DCHs; this leads not only to the creation of DCH states with an additional valence excitation but also to singly ionized states featuring core-hole ionization plus the excitation of a second core electron into the valence shell.^{40,45,46} The latter

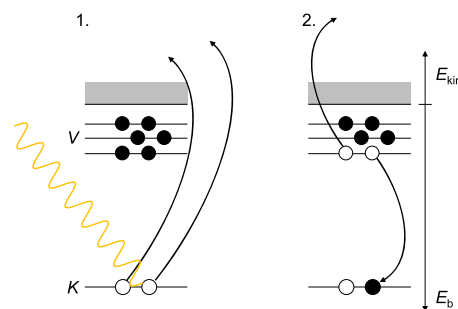


FIG. 1. Schematic showing the simultaneous ejection of two core electrons (left, label “K”) and the first step in the radiationless decay of the two core holes (right) by rearrangement of the valence shell (label “V”). The kinetic energy (E_{kin}) spectrum of the latter decay step in liquid water will be discussed here. E_b denotes the binding energy.

states constitute a series of resonances located in energy around the DCH production threshold. They are reached by the ejection of a single electron, while the creation of a non-resonant DCH state requires the ejection of an electron pair. The primary electrons pertaining to the resonant DCH satellites can, therefore, be detected by conventional electron spectroscopy. For experiments in gaseous water, see Refs. 34 and 47. For liquid water, we will return to this aspect below. The pair of primary electrons giving rise to non-resonant DCHs and doubly charged DCH satellite states can arbitrarily share the available excess energy and can only be identified by coincidence-detection methods that were not available for this study.^{28–32}

Several recent publications have discussed the hypersatellite Auger spectra of DCHs in other gaseous systems. Detailed results appeared on Ne, isoelectronic to water.^{48,49} For a molecular sample (CH_3I), the potential energy landscape of a DCH state could be extracted from high-resolution data.⁵⁰ The strongly repulsive nature of these states, which leads to ultrafast dissociation, was illustrated in Refs. 51 and 52 and will be discussed for water in this work.

II. METHODS

A. Experimental methods

The liquid-water experiments were performed using the EASI (Electronic structure from Aqueous Solutions and Interfaces) liquid-jet photoelectron spectroscopy instrument,⁵³ installed at the soft-x-ray beamline P04 (Ref. 54) of the synchrotron radiation source PETRA III (DESY, Hamburg, Germany). Data acquired in four different campaigns have been compiled for this work. EASI is based on a Scienta-Omicron HiPP-3 differentially pumped hemispherical electron analyzer. This analyzer uses a pre-lens system optimized for the detection of low-energy electrons but is also capable of detecting electrons with kinetic energies up to 1.5 keV. The main vacuum chamber contains efficient μ -metal shielding. All experiments were conducted with a pass energy of 200 eV and a nominal analyzer energy resolution of 2.0 eV (first campaign, entrance-slit setting 4.0 mm) or 0.4 eV (all other measurements, entrance-slit setting 0.8 mm, straight slit). For the liquid-water measurements, a small amount (≈ 50 mM) of NaCl salt was added to highly demineralized water (conductivity $\approx 0.2 \mu\text{S cm}^{-1}$) to maintain electrical conductivity and mitigate potentially deleterious sample-charging effects.^{55–57} This is common practice when measuring photoelectron spectra from liquid water.^{58,59} The liquid microjet was generated by injecting the sample solution into the interaction vacuum chamber through a 28 μm diameter glass capillary at a typical flow rate of 0.8 mL min^{-1} . A cooling jacket extending up to ~ 70 mm upstream of the nozzle and stabilized at 10°C was used to reduce the temperature of the liquid sample (see Ref. 53 for details). We then positioned the point of irradiation of the liquid jet at a distance of 0.5–0.8 mm from the 800 μm diameter skimmer orifice at the analyzer entrance. The pressure in the main chamber was kept at $\sim 5 \cdot 10^{-4}$ mbar using two turbo-molecular pumps (with a total pumping speed of $\approx 2600 \text{ L s}^{-1}$ for water) and three liquid-nitrogen cold traps (with a total pumping speed of $\approx 35000 \text{ L s}^{-1}$ for water). The solution was delivered using a Shimadzu LC-20AD HPLC pump equipped with an in-line degasser (Shimadzu DGU-20A5R). Measurements were carried out

for light and heavy water, respectively. Figure 2 shows a sketch of the experimental geometry.

The synchrotron-light propagation axis was orthogonal to the liquid jet, with both lying in the horizontal plane. We used the hemispherical electron analyzer positioned at a 130° angle with respect to the photon-beam propagation direction (backward-detection geometry⁵³), with its lens lying in the vertical plane, and with circular polarization of the photon beam. All photoemission measurements reported here were conducted using the 1200 l mm^{-1} laminar grating of the P04 beamline. For the first campaign, we used an exit-slit setting of 1000 μm and a resulting photon flux of $3.5 \cdot 10^{13}$ photons/s; for all other measurements, depending on the exact settings of the exit slit and the grating groove depth, the photon flux amounted to $4.1\text{--}7.7 \cdot 10^{12}$ photons/s. Due to the spatial dimensions of the EASI vacuum chamber, the interaction point is located ≈ 220 mm downstream of the nominal focus position when it is mounted at the P04 beamline. For our first experiments, this, in conjunction with the large exit-slit opening, lead to a vertical spot size of $\approx 400 \mu\text{m}$. In all subsequent campaigns, the matching of vertical spot size to the spatial extension of the liquid jet was improved by specific settings of the refocusing mirror unit of the beamline (a pair of Kirkpatrick–Baez mirrors), thus focusing the beam to 40 μm in the vertical direction. The horizontal spot size, extending along the liquid-jet flow direction, is 180 μm . The approximate photon-energy resolutions under these conditions are 3.0 eV (first campaign) and 0.31 eV (all other measurements). The photon energy was normalized by a procedure that optimizes the angle for specular reflection of the grating as a reference. As the exact photon energy is not decisive for the conclusions of this study, no further calibration was attempted. We chose to use a photon energy of 1400 eV to be well above the known gas-phase

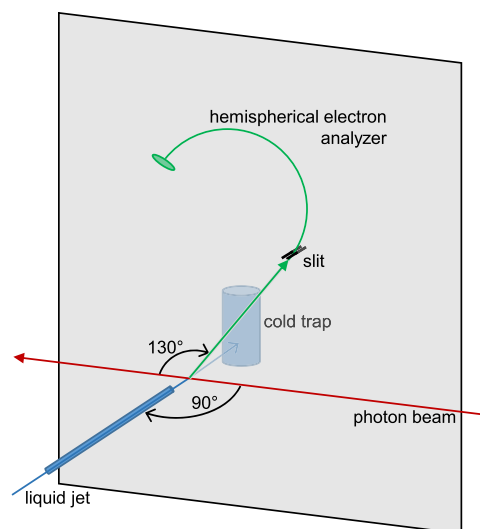


FIG. 2. Sketch of the experimental geometry. Circularly polarized synchrotron radiation (red arrow) is crossed by the liquid jet (blue arrow) in the horizontal plane under a right angle. Electrons are collected by a hemispherical analyzer arranged to have a vertical orbit plane aligned with the synchrotron radiation beam (green line: central electron trajectory). The entrance slit into the hemisphere is, therefore, parallel to the liquid jet. After crossing the interaction zone, the liquid jet is frozen out on a cold trap.

threshold of 1171 eV³² and the expected similar liquid-phase threshold for water double core holes. Measuring about 200 eV above the threshold represents a good compromise for photon flux and cross section.

The kinetic-energy scale of the hypersatellite Auger spectra was calibrated with respect to the maximum of the normal KVV Auger spectrum of liquid water, measured in conjunction with the DCH spectra and at the same photon energy. A value of (502.7 ± 0.3) eV was used for the normal Auger energy (see supplementary material).

B. Theoretical methods

To mimic the Auger electron spectrum of liquid water, we calculated the corresponding Auger spectrum for the central water molecule in a water pentamer in a tetrahedral arrangement using the electronic-structure toolkit XMOLECULE^{60,61} (version 0.2-145). With the aim of explaining the qualitative effects, we solely addressed pure K^{-2} configurations.

The Auger calculations were based on the one-center approximation.⁶² Specifically, a set of molecular orbitals was obtained for double-core-ionized configurations using the restricted Hartree-Fock method and employing the maximum-overlap method.⁶³

All calculations were performed using the 6-31G(d,p) basis set.^{64–66} Using the obtained orbitals, the final states of the Auger process were calculated using configuration interaction (CI), employing all two-valence-hole-one-core (2h-1c) and selected three-valence-hole-one-core-one-particle (3h-1c1p) configurations. The number of configurations that appear in such configuration-expansion calculations rapidly explodes with the size of the system. For the water pentamer [(H₂O)₅], the number of spin configurations exceeds 500 000, thus leading to a considerable computational challenge since numerous eigenvectors have to be calculated to cover the relevant final Auger states. To keep the calculations tractable, we had to constrain the employed configurations. To that end, valence orbitals were localized using the Foster-Boys localization procedure.⁶⁷ Only those configurations were considered, where not more than one valence hole is located outside the central water molecule. The idea behind this restriction is that electronic configurations where both valence holes are not covering the central water molecule, in which the core hole is located, have vanishing Auger yield. The described approximation is tested for a water dimer in the supplementary material. The restriction of the configuration space yielded a total of 62 084 doublet configurations. For each considered geometry, we computed the lowest 5000 roots to cover the relevant energy range of the Auger spectrum.

The Auger amplitudes involve the evaluation of two-electron integrals of the form

$$\langle ab|ck\rangle = \int d^3r_1 \int d^3r_2 \phi_a(\mathbf{r}_1)\phi_b(\mathbf{r}_2) \frac{1}{|\mathbf{r}_1 - \mathbf{r}_2|} \phi_c(\mathbf{r}_1)\phi_k(\mathbf{r}_2), \quad (1)$$

where $\phi_a(\mathbf{r})$ and $\phi_b(\mathbf{r})$ are valence orbitals, and $\phi_c(\mathbf{r})$ is a core orbital. The function $\phi_k(\mathbf{r})$ represents an energy-normalized continuum wave function with the Auger electron energy $\epsilon_k = k^2/2 = E_i - E_f$, where E_i and E_f are the initial and final bound-state energies, respectively.

The employed one-center approximation relies on the idea that the Auger effect is dominantly an inner-atomic process. The

molecular continuum wave function $\phi_k(\mathbf{r})$ is thus approximated by the atomic continuum wave function $\chi_k(\mathbf{r})$. Furthermore, the two-electron integrals are expanded using the linear combination of atomic orbitals employing a minimal basis set. Only coefficients on the atom on which the core hole is located are taken into consideration. The two-electron integral is thus approximated by

$$\langle ab|ck\rangle \simeq \sum_{\substack{\mu\nu\lambda \\ \text{on atom } \Lambda}} C_{\mu a} C_{\nu b} C_{\lambda c} \langle \mu\nu|\lambda\kappa\rangle. \quad (2)$$

Note that, for the atomic two-electron integrals $\langle \mu\nu|\lambda\kappa\rangle$, the atomic continuum energy $\epsilon_k = k^2/2$ for the corresponding atomic Auger transition is employed. The basis set used for the expansion of the molecular orbitals and the atomic two-electron integrals $\langle \mu\nu|\lambda\kappa\rangle$ were computed using the atomic electronic-structure program XATOM.⁶⁸

Even though the Auger matrix elements are evaluated based on a fully inner-atomic approximation, the present calculation still yields results where electronic holes are distributed over various atoms. This is because, in general, molecular orbitals spread over many atoms and, furthermore, configurational mixing spreads the created charge. This leads in practice to considerable two-center contributions, where neighboring water molecules also acquire charge.

To incorporate the effects of core-hole-induced nuclear dynamics during the lifetime of the double core hole, we propagated the molecular geometry in the double-core-hole state using molecular dynamics (MD) and employed a time step of 0.1 fs for up to 7 fs (~5 times the calculated DCH lifetime). Trajectories were initiated for isolated water (H₂O), the water pentamer [(H₂O)₅], and the deuterated water pentamer [(D₂O)₅]. The MD trajectories started with initial geometries and velocities generated by sampling the neutral-ground-state vibrational Wigner distribution. For (H₂O)₅, we obtained the equilibrium tetrahedral geometry and the corresponding Hessian matrix from geometry optimization using GAMESS (version 2012-R2)⁶⁹ employing a polarization-continuum model to mimic the surrounding water environment. In the MD calculations, we made sure that the respective core hole was always in the central water molecule. The effects of the nuclear dynamics on the Auger spectrum were described in Refs. 20 and 19. For both pentamer [(H₂O)₅] and isolated water (H₂O), the valence double-hole final states and the ensuing Auger spectrum were calculated for all 70 snapshots for a total of 50 sampled trajectories, respectively. For each time step, an instantaneous Auger spectrum was obtained by averaging over all trajectories and employing a Lorentzian line shape with a width determined by the calculated Auger decay rate Γ . The resulting Auger spectrum was then compiled from the instantaneous spectra, with the weights $\exp(-\Gamma t)$ accounting for the decay of the double-core-hole state.

III. RESULTS AND DISCUSSION

We start by discussing the experimental results of this study. For reference, the inner-shell photoemission spectrum of liquid water at a photon energy above the threshold for DCH creation (1171 eV for gas-phase water)³² is shown in Fig. 3(a). Although the P04 beamline is among the most brilliant synchrotron radiation

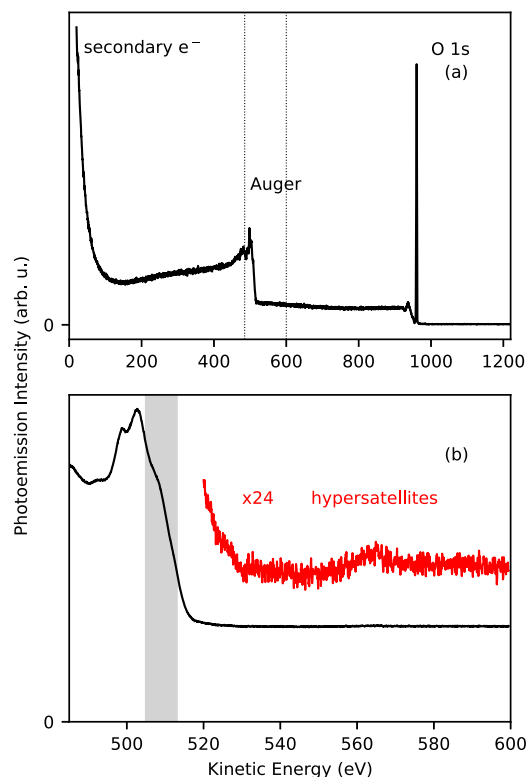


FIG. 3. (a) Full experimental photoemission spectrum of liquid water, excited with photons of 1500 eV. The kinetic-energy interval shown in the lower panel is marked by vertical dotted lines. (b) Photoemission spectrum of liquid water showing the kinetic-energy region of Auger decay marked in panel (a), here recorded at $h\nu = 1400$ eV. Normal KVV Auger decay can be seen between 480 and 515 eV. The gray-shaded region from 505 to 513 eV kinetic energy is associated with proton-transferred states, for which the nuclear dynamics in the K -shell-ionized state play a particular role.^{4,10} Auger decay of the DCH states, "hypersatellites," is much weaker due to the low cross section of DCH production by a single photon and is shifted to substantially higher kinetic energies. The upper trace in panel (b) shows a portion of the spectrum multiplied by 24 and vertically offset to improve its visibility. We have also performed experiments below the DCH threshold and at other photon energies. These data are presented in the supplementary material.

sources in the soft-x-ray range, we do not expect the production of doubly ionized states by the simultaneous absorption of two photons at the same or neighboring liquid-water sites. The spectrum therefore mostly exhibits signals from inner-shell single photoionization and the associated Auger decay. However, the largest signal intensity is observed at the low-energy tail due to inelastically scattered electrons trailing both the O 1s main line and the KVV Auger lines. We refer to Ref. 70 for a general discussion of inelastically scattered electrons from liquid-water photoionization. Auger decay of DCH states should be searched at the high-kinetic-energy side of the normal KVV Auger decay, shifted by some tens of eV with respect to the decay of singly ionized states.^{20,24} Such a spectral feature is indeed seen in Fig. 3(b), atop the background of inelastically scattered photoelectrons.

The result obtained for a much longer electron collection time and focusing on the energy region highlighted earlier is shown

in Fig. 4. Here, we present the average over 111 sweeps, after subtraction of a linear background from each individual sweep. The error bars are the standard deviation of the mean over all background-subtracted sweeps at the respective energy point. This type of analysis clearly confirms the existence of a feature between 558 and 572 eV kinetic energy. It was consistently observed at two different energies above the DCH creation threshold but not at a photon energy below it (see supplementary material). This feature identifies the first step in an Auger decay cascade initiated by single-photon double photoionization of liquid water. Its intensity, relative to the normal KVV Auger spectrum of singly ionized liquid water, was found to be 0.0014(5), $\sim 1/700$ (see supplementary material). This is on the same order of magnitude as found for the relative intensity of single- to double-core-hole creation in gaseous molecules.^{20,36}

Our analysis of the data in Fig. 4 reveals that the main source of remaining uncertainty is not Poissonian noise from the nature of the electron-counting process but fluctuations in the height and shape of the background. The background exceeds the intensity of the feature under discussion by more than an order of magnitude. Unfortunately, these fluctuations practically imposed an upper limit on the improvement in data quality we could obtain by extending the duration of the measurement (Fig. 4 corresponds to a wall-clock acquisition time of ~ 3 h). Moreover, between different campaigns, the intensity of the inelastic background and even its slope varied substantially. More details on the raw data and their analysis are given in the supplementary material.

As the main result of our experiments, we, therefore, present in Fig. 5(b) an average of the Auger hypersatellite spectra from several datasets. The error band now is derived from the standard deviation of the individual datasets (shown separately in the supplementary material), which were weighted equally to arrive at this result. Although, in particular, between 542 and 557 eV kinetic energy, strong variations in the experimental spectral shape as a function of kinetic energy can be seen, the presence of the hypersatellite signal is evident.

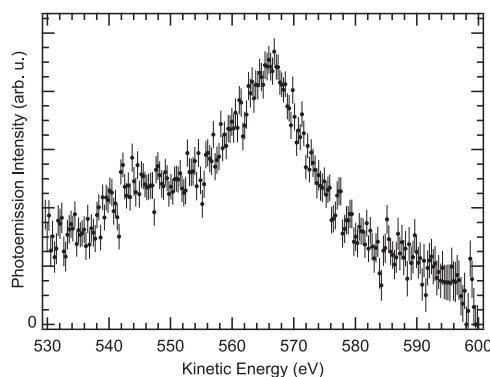


FIG. 4. Close-up on the photoemission intensity from liquid water on the high-kinetic-energy side of the normal Auger spectrum, recorded with a photon energy of 1400 eV. The feature between 558 and 572 eV of kinetic energy is identified with the first-step Auger decay of a DCH in liquid water. A linear background has been subtracted. See the text for details.

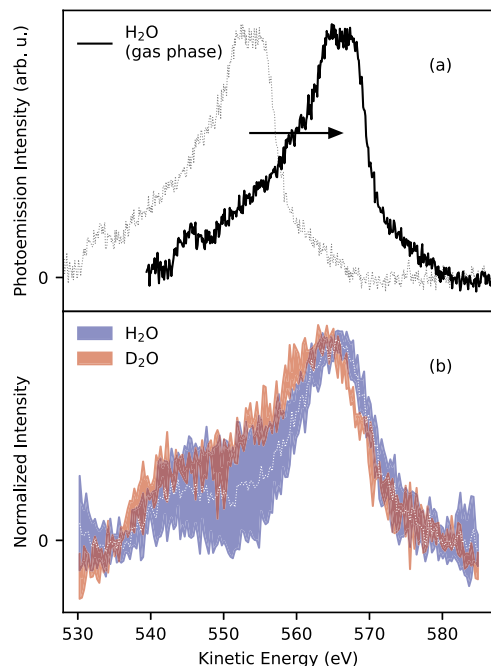


FIG. 5. Auger hypersatellite spectra of O 1s DCHs in gaseous (a) and liquid water (b). The data for gaseous water (dotted line) were shifted by 12.5 eV (horizontal arrow) to allow for an easier comparison and were published in Ref. 20. For the liquid, spectra were measured for normal and deuterated water. The colored error band depicts the standard deviation of several independently acquired datasets. For normal water, the average is indicated by a thin dotted line.

We can compare this signal to the one from gaseous water, shown for reference in Fig. 5(a). Qualitatively, we see a similar shape of the liquid-water hypersatellite Auger decay but shifted toward higher kinetic energy by about 12.5 eV. The shift was determined from the respective centers of gravity within the full width at half maximum of each peak. This significant shift of the Auger spectra of liquid water relative to those of the gas phase can be rationalized qualitatively by the Born solvation model.^{71,72} For liquid water, the Born model is well suited to reproduce the energy shift associated with the charge state of the species under consideration (termed “M” for the sake of discussion); this can be, e.g., M^+ upon direct ionization and M^{2+} in the case of a normal Auger process. Here, we extend it to M^{3+} in the present case of DCH formation. Accordingly, the creation of an additional charge in a molecule embedded in a solution induces polarization, and the resulting energy shift for a given charge state relative to the situation in vacuum amounts to

$$\Delta E = \frac{q^2}{2R} \left(1 - \frac{1}{\epsilon} \right), \quad (3)$$

where atomic units are employed, R is the assumed radius of the ionized molecule with charge q , and ϵ is the dielectric constant of the solution. Following Ref. 73, and since we have to consider mainly the electronic polarization, we employ the high-frequency dielectric constant $\epsilon = 1.8$ and an ion radius estimated from the crystal radius in ice of $R = 1.38$ Å. Those parameters have been found to

qualitatively describe K -shell binding energy shifts from vapor to liquid.⁷³ Using the same model, we obtain a relative shift of the doubly charged double-core-hole state to the triply charged state after the hypersatellite Auger decay of about 11.6 eV, in fair agreement with the observed shift of about 12.5 eV between the liquid and vapor K^{-2} Auger spectra in Fig. 5. Furthermore, when applying the same simple model to the single-core-hole case, we obtain a shift of 6.95 eV between gas-phase and liquid-phase Auger electrons. This agrees reasonably well with the experimentally observed value of ≈ 5 eV.^{4,74}

Comparing again the results for gaseous water in Fig. 5(a) to those for liquid water in Fig. 5(b), we observe that particularly large deviations between the various liquid-water datasets (seen as the width of the error band) occur in kinetic-energy regions where contributions of the original, unshifted gas-phase signal would be expected. To assure that the gas-phase signal makes a rather minor contribution, we have recorded reference spectra before or after measuring the hypersatellite Auger electron spectra. This reveals a gas-phase signal contribution of less than 10% (see supplementary material), far smaller than required to explain the deviations between different datasets contributing to Fig. 5(b). Whether the gas-phase contributions drifted toward a higher-intensity fraction during the lengthy hypersatellite data acquisition remains unsolved.

We now return to the original question of the dynamics of core-hole states in water on ultrashort time scales. For that, we have recorded the hypersatellite Auger spectra for normal and deuterated water [Fig. 5(b)]. A difference between the spectra can be clearly seen, most significantly in the position of the high-kinetic-energy flank of the hypersatellite Auger peak. Qualitatively, this is reminiscent of the differences between the normal KVV Auger spectra of liquid light and heavy water.⁴

Our results can be interpreted with assistance from the theoretically calculated K^{-2} Auger spectra. In order to capture the nuclear dynamics of the process, we have modeled the liquid sample by calculating Auger spectra for the central water molecule inside a water pentamer with tetrahedral arrangement, $(H_2O)_5$, and its deuterated variant $(D_2O)_5$. Figure 6(a) shows a comparison of the calculated Auger spectra obtained for $(H_2O)_5$ and $(D_2O)_5$. The spectra exhibit two dominant peaks at 565 eV and at 545 eV that can be associated with Auger transitions leaving two vacancies in the outer-valence levels ($3a_1$, $1b_2$, and $1b_1$) or one vacancy in the outer-valence levels and one in the inner-valence levels ($2a_1$). Despite the fact that only a single hypersatellite initial state (K^{-2}) was taken into account, whereas the experimental data may also involve other excited states based on a DCH (i.e., $K^{-2}V$, $K^{-2}L^{-1}$, or $K^{-2}L^{-1}V$),²⁰ the calculated Auger spectrum shows qualitatively similar features as the experimental data. The comparison of the two calculated spectra for the deuterated and non-deuterated pentamers also shows similar trends. As in the case of the experimental spectra, for $(H_2O)_5$, the calculated high-energy flank is shifted to somewhat higher energies compared to the spectrum for $(D_2O)_5$. In addition, for the calculated spectra, the extended tail at about 580 eV tends to be more pronounced for $(H_2O)_5$ than for $(D_2O)_5$. This difference in the Auger spectrum can be attributed to the ultrafast core-hole-induced proton dynamics, which is less pronounced for $(D_2O)_5$ due to the larger mass. The effect of the proton dynamics is highlighted by comparison to the

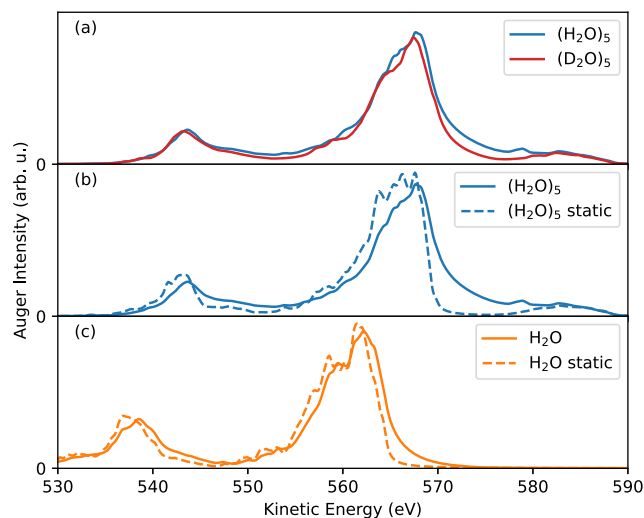


FIG. 6. Calculated Auger electron spectrum. (a) Comparison for the deuterated $[(D_2O)_5]$ and non-deuterated $[(H_2O)_5]$ water pentamers in a tetrahedral arrangement. (b) Auger spectrum for the water pentamer $[(H_2O)_5]$. (c) Auger spectrum for isolated water (H_2O). The dashed lines in panels (b) and (c) show instantaneous spectra at $t = 0$, i.e., without taking into account the effect of core-hole-lifetime nuclear dynamics.

calculated Auger spectrum for $(H_2O)_5$ employing a static geometry. As can be seen, the proton dynamics results in a pronounced tail and a slight shift of the high-energy flank to higher energies for the main Auger peak. The same trends appear for a water molecule in the gas phase, as shown in Fig. 6(c).^{19,20,44}

Effects specific to the liquid environment can be understood by comparing the Auger spectrum for $(H_2O)_5$ in Fig. 6(b) with the one calculated for isolated water that is shown in Fig. 6(c). The most striking difference is the significant shift of the entire spectrum to higher energies by about 5 eV. As described earlier, this shift can be attributed to the fact that the electronic structure of the surrounding water molecules contributes to the final double-valence-hole states, effectively lowering the pentamer's energy through polarization. The shift revealed from the experiment in Fig. 5 of about 12.5 eV is considerably larger than the theoretical one, which we mainly attribute to the fact that the calculation is for a water pentamer and not the full liquid environment. In addition, this discrepancy might also arise from the relatively small basis set that had to be chosen to make the computations feasible. Remember that the experimental spectra are in good agreement with the energy shift obtained from the simple Born model. We next comment on an additional spectral feature in Fig. 6, occurring near 580 eV kinetic energy for $(H_2O)_5$ and $(D_2O)_5$. This identifies final double-valence-hole states with charges delocalized over neighboring water molecules and thereby results in particularly high energy. These channels can be best characterized as core-level intermolecular Coulombic decay,⁷⁵ where electrons on the neighboring molecules participate in the radiationless decay. The feature connected with this decay process amounts to 6% of the overall decay. We expect that this pentamer feature will become more smeared out in liquid water due to the stronger fluctuations in the geometrical arrangements at room temperature

and effects from the second solvation layer. It is thus not surprising that this feature would be difficult to detect in the experimental data in Fig. 5.

Apart from the overall shift caused by polarization and the additional peak at 580 eV linked to decays directly involving the neighboring water molecules, the calculations show further effects of the local chemical environment. A comparison with the respective static spectra in Figs. 6(b) and 6(c) reveals that the effects of the nuclear dynamics in $(H_2O)_5$ are significantly stronger. Specifically, the marked tail on the high-energy side of the dominant Auger line located at 565 eV for $(H_2O)_5$ and at 560 eV for H_2O is more pronounced for $(H_2O)_5$ compared to H_2O .

The effect of the nuclear dynamics on the spectrum is further discussed in Fig. 7, where instantaneous Auger spectra at selected times are shown. The individual instantaneous spectra are weighted with their relative contribution according to the exponential decay in the core-hole lifetime of ≈ 1.5 fs in the calculations. In the double-core-hole state, the water molecule is highly dissociative, and both hydrogen atoms dissociate as protons.⁶¹ The dynamics is sketched for a typical trajectory in Fig. 7(a). The repulsion of the two protons reduces the charge around the oxygen atom [see Figs. 8(c) and 8(d) for details], ultimately leading to faster Auger electrons. As can be seen in Fig. 7(b), the spectral shifts result in marked tails on the high-energy side of the Auger peaks.

For the double-core-hole state, asymmetric stretching and bond-angle opening play a minor role during the lifetime of the Auger decay; instead, the dominant dynamics for the water pentamer as well as the isolated water molecule occurs along the symmetric stretching leading to the transfer of two protons to the neighboring molecules. This is different for the single-core-ionized water molecule, where, in the liquid phase, dynamics involves the transfer of a *single* proton^{4,10} and, in the gas phase, the molecule does not dissociate.¹⁹ Notably, as shown by Marchenko *et al.*,²⁰ we expect that for other double-core-hole satellite configurations, asymmetric stretching and bond-angle opening may also play a substantial role.²⁰ Moreover, we point out that proton dynamics for other double-core-hole satellites can be significantly faster compared to the pure DCH state, as was reported for the gas phase.²⁰ Figure 8 displays the geometrical variations of the trajectories used to compute the Auger spectra. It shows the symmetrized bond distance of the central water molecule of $(H_2O)_5$ as a function of time after core ionization. To make the difference in the dynamics more apparent, we plot the data for up to 5 fs, but one has to keep in mind that the decay lifetime of the double core hole is $1/\Gamma \approx 1.5$ fs. Figure 8(a) reveals that the fragmentation of the water molecule in the double-core-hole state is somewhat faster in the tetrahedral water structure of $(H_2O)_5$ than in H_2O . Even though the effect is rather small at early times, it points toward the fact that the liquid environment tends to accelerate the proton dynamics compared to the case of an isolated water molecule. This result can also be linked to the stronger effect of nuclear dynamics in $(H_2O)_5$ vs H_2O , as seen in Fig. 6(b). This result is plausible, considering that the presence of hydrogen bonds in $(H_2O)_5$ promotes proton dissociation relative to the situation in H_2O .

Figure 8(b) shows the dynamics for $(D_2O)_5$ and for D_2O . As one might expect, the dissociation for $(D_2O)_5$ is somewhat

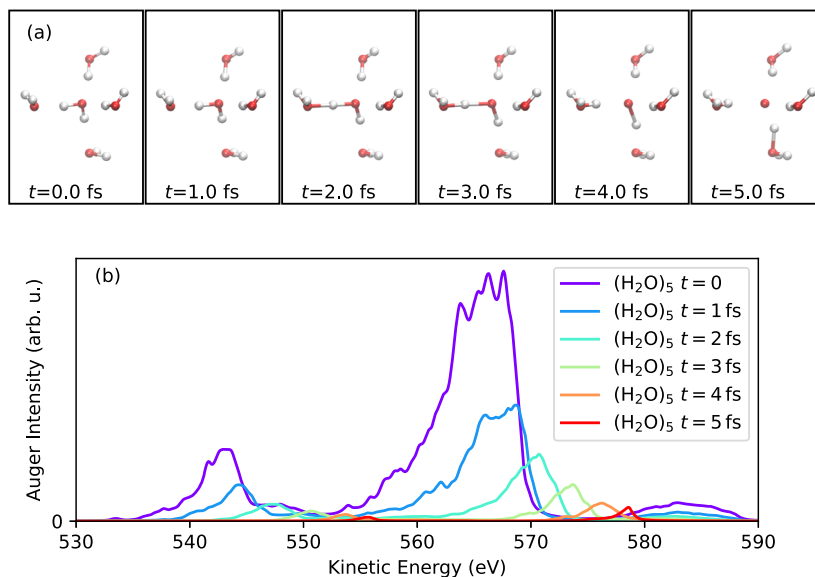


FIG. 7. (a) Snapshots of the molecule geometry for a typical $(\text{H}_2\text{O})_5$ trajectory. (b) Calculated instantaneous Auger spectra at selected times after core ionization.

slower than for $(\text{H}_2\text{O})_5$. When comparing the respective deuterated and non-deuterated species, one can see that deuteration leads to a similar deceleration of the proton dynamics as the transition from $(\text{H}_2\text{O})_5$ to H_2O . Our results show that the ultrafast core-hole-induced dynamics is promoted when the double-core-

ionized water molecule is in a tetrahedral hydrogen-bond environment. We speculate that this trend also holds for a liquid-water environment. Figures 8(c) and 8(d) show the partial charge (Mulliken charge) of the central oxygen atom and how it evolves as a function of time for $(\text{H}_2\text{O})_5$, H_2O [Fig. 8(c)], $(\text{D}_2\text{O})_5$, and D_2O

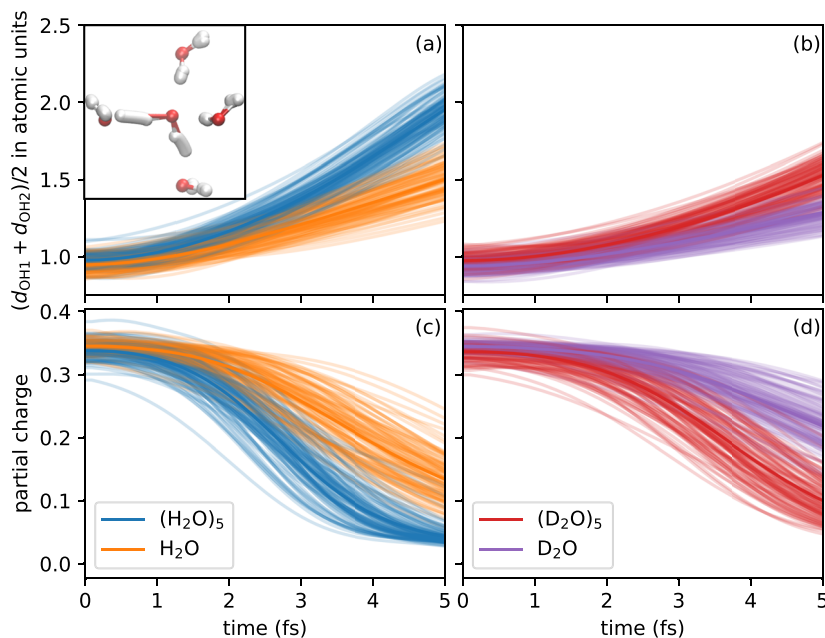


FIG. 8. Top panels: Symmetrized oxygen-hydrogen distance as function of time after double core ionization. (a) Comparison of water pentamer in a tetrahedral arrangement $[(\text{H}_2\text{O})_5]$ and isolated water (H_2O). The inset illustrates an example trajectory for the pentamer [see Fig. 7(a) for individual snapshots]. (b) Comparison of deuterated water pentamer $[(\text{D}_2\text{O})_5]$ and isolated deuterated water (D_2O). Bottom panels: Mulliken charge on the central oxygen as a function of time. Comparison for $(\text{H}_2\text{O})_5$ and H_2O (c) as well as $(\text{D}_2\text{O})_5$ and D_2O (d).

[Fig. 8(d)]. The partial charge of the central oxygen atom is initially about 0.35 due to the strong screening effects that involve electronic rearrangements from neighboring atoms toward the double-core-ionized oxygen atom. As the protons move away from the central oxygen atom, the partial charge assigned to the central oxygen atom decreases, illustrating the above-mentioned reduction of charge via proton repulsion.

We have tried to extract further experimental information on DCHs in liquid water by attempting to observe primary electrons corresponding to resonant $K^{-2}V$ satellite states at binding energies around the DCH threshold; no such signature could be identified, however (see supplementary material).

IV. SUMMARY AND OUTLOOK

We experimentally characterized the Auger electron hyper-satellite spectrum from the first step in the radiationless decay cascade of single-site double core holes in liquid water, produced by single-photon double photoionization. Compared to its gas-phase counterpart, the kinetic energy of the feature is blue-shifted by 12.5 eV, largely due to the polarization of the surrounding water molecules in the final state. Despite the ultrashort lifetime of an oxygen DCH being around 1.5 fs, nuclear dynamics within this time window is evidenced by a noticeable isotope effect when experiments are carried out with deuterated water. Calculations of the Auger spectrum of a DCH in a water pentamer allowed us to identify a symmetric stretch of the molecule as the main driver of nuclear dynamics and shed light on the temporal evolution of the Auger spectrum.

As an outlook, we would like to present some thoughts about future experiments on DCHs. Studies of the nuclear dynamics during normal Auger decay and the spectral features associated with it (see, e.g., Refs. 12, 13, and 76 and the review Ref. 2) had a considerable impact on the discussion about the structure of water. Extending these studies to DCHs will extend the range of parameters under which this dynamics can be probed.

One important aspect of the quality of DCH-related electron spectra we have identified is the ratio of the signal of interest to the background from the inelastic scattering of O 1s photoelectrons. This might be more favorable at higher photon energies for the following two reasons: (1) the background will be stretched out to a wider kinetic-energy range; and (2) the relative cross section of double- to single-core-hole creation depends on the excess energy. This is brought about by an interplay of two mechanisms for single-photon double photoionization that were discussed, e.g., in Ref. 39. If we assume this discussion to be universal, the maximum of the relative DCH production should be at an excess energy of roughly 1.4 times the binding energy of the DCH state, i.e., for our system at a photon energy of around 2800 eV. Experimentally, for Ne, isoelectronic to water, an increase in the double/single core-hole-production ratio was found up to photon energies of 5 keV.⁷⁷ We believe that these factors can lead to hypersatellite Auger spectra of better quality at higher photon energies, even though photon flux and absolute cross sections might be lower than in the present study. Moreover, we suggest liquid-water DCH experiments at XFELs, where the ratio between K^{-2} and K^{-1} should be larger and we would be less sensitive to an unknown and varying background.

As a further outlook, we want to point out that, at present, measuring time-resolved Auger spectra is within reach.⁷⁸ Therefore, as discussed earlier, our calculations reported here are interesting not only regarding the high-energy tail but also the energy shift (see Fig. 7). We envision time-resolved Auger spectra using intense x-ray free-electron lasers and, e.g., self-referenced attosecond streaking⁷⁸ of more complex systems than noble gases in the future. So far, e.g., the double-core-hole generation in O₂ molecules and the corresponding molecular-frame photoelectron angular distributions⁷⁹ and an alternative pathway to double-core-hole states in gas-phase water⁴³ have been studied at European XFEL.

Finally, we would like to mention that double-core-hole production has also been studied for anions, partly giving rise to significantly larger cross sections than in neutrals.⁸⁰ For example, the maximum cross section of (1s + 1s) ionization in C⁻ has been found to be 3 kb,⁸¹ compared to 0.07 kb when our experimental double/single core-ionization ratio is combined with a calculated value for the O 1s photoionization cross section.⁸² No explanation of this qualitative difference is known to the authors. Experimentally, the study of systems that, in solution, are present in anionic form is well possible since, in the aqueous phase, anions can be readily stabilized. Therefore, the search for a potential enhancement of the DCH cross section with charge state seems highly interesting.

SUPPLEMENTARY MATERIAL

In the supplementary material, we first describe details of the acquisition of the experimental data and their analysis. We then comment on various aspects of our results that may be of interest to a specialized reader.

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AUTHOR DECLARATIONS

Conflict of Interest

The authors have no conflicts to disclose.

Author Contributions

Florian Trinter: Investigation (equal); Methodology (equal); Writing – original draft (equal); Writing – review & editing (equal). **Ludger Inhester:** Methodology (equal); Visualization (equal); Writing – original draft (equal); Writing – review & editing (equal). **Ralph Püttner:** Conceptualization (equal); Writing – review & editing (equal). **Sebastian Malerz:** Investigation (equal); Writing – review & editing (equal). **Stephan Thürmer:** Investigation (equal); Writing – review & editing (equal). **Tatiana Marchenko:** Conceptualization (equal); Writing – review & editing (equal). **Maria Novella Piancastelli:** Conceptualization (equal); Writing – review & editing (equal). **Marc Simon:** Conceptualization (equal); Writing – review & editing (equal). **Bernd Winter:** Funding acquisition (equal); Investigation (equal); Project administration (equal); Supervision (equal); Writing – review & editing (equal). **Uwe Hergenhahn:** Data curation (equal); Formal analysis (equal); Funding acquisition (equal); Investigation (equal); Methodology (equal); Project administration (equal); Visualization (equal); Writing – original draft (equal); Writing – review & editing (equal).

DATA AVAILABILITY

The data that support the findings of this study are openly available in the Zenodo data repository at <https://doi.org/10.5281/zenodo.10523680>.⁸³

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