



Título artículo / Títol article: Comparison of Simple and Rapid Extraction Procedures for the Determination of Pesticide Residues in Fruit Juices by Fast Gas Chromatography–Mass Spectrometry

Autores / Autors Laura Cherta, Joaquim Beltran, Elena Pitarch, Félix Hernández

Revista: Food Analytical Methods

Versión / Versió: Versió pre-print

Cita bibliográfica / Cita bibliogràfica (ISO 690): CHERTA, Laura, et al. Comparison of Simple and Rapid Extraction Procedures for the Determination of Pesticide Residues in Fruit Juices by Fast Gas Chromatography–Mass Spectrometry. Food Analytical Methods, 2013, vol. 6, no 6, p. 1671-1684.

url Repositori UJI: <http://hdl.handle.net/10234/87429>

Comparison of simple and rapid extraction procedures for the determination of pesticide residues in fruit juices by fast gas-chromatography-mass spectrometry

Laura Cherta, Joaquim Beltran, Elena Pitarch, Félix Hernández

Research Institute for Pesticides and Water, University Jaume I, Castellón, Spain.

Abstract

Three sample treatment methods, based on QuEChERS, solid-phase extraction (SPE) and solid-phase microextraction (SPME), were compared and evaluated in order to obtain the best conditions to determine pesticide residues in fruit juice by fast gas chromatography–mass spectrometry (single quadrupole GC-MS). Analysis were performed under selected ion monitoring, acquiring the three most abundant and/or specific ions for each analyte and using their relative intensity ratios as a confirmatory parameter. The 3 methodologies (QuEChERS, SPE and SPME) were validated taking 15 selected pesticides as model compounds, using commercial apple juice. QuEChERS procedure was based on the AOAC Official Method 2007.01, using acetonitrile (containing 1 % acetic acid) as extraction solvent and primary–secondary amine during the dispersive solid-phase extraction. Oasis hydrophilic–lipophilic balance cartridges were used for SPE, and polyacrylate fibers were used for direct immersion SPME procedure. Three isotopically labeled standards were added to the samples before extraction and used as surrogate standards. Validation parameters as recoveries, limits of detection, and limits of quantification (LOQ), as well as matrix effects and sample throughput, were obtained and compared for the three extraction procedures. QuEChERS was considered faster and led to the best quantitative results. In this way, validation was extended to up to 56 pesticides by applying QuEChERS in multi-fruit juice samples, obtaining LOQs ranging from 2 to 20 µg/L for most

compounds. Accuracy and precision were evaluated by means of recovery experiments at two concentration levels (10 and 100 µg/L), obtaining recoveries between 70 and 120 % in most cases and relative standard deviations below 15 %. Finally, the QuEChERS method was applied to the analysis of commercial juices, including mango–apple, pineapple, grapefruit and orange.

Keywords

QuEChERS; SPE; SPME; Pesticides; Fast gas chromatography-mass spectrometry; juices.

INTRODUCTION

Pesticide residues can remain in food after they are applied to crops, even after being washed, processed and prepared, and may result in adverse consequences to the human health. Their concentrations in processed food are usually lower than those observed in whole fruit due to their degradation through oxidative mechanisms or elimination during food processing, mainly after washing and peeling (Picó and Kozmutza 2007; Burchat et al. 1998; Patyal et al. 2004). The European Commission (2008) has set harmonized maximum residue levels (MRL) based on comprehensive assessment of the properties of the active substance and the residue behavior on treated crops. In most cases, no MRLs are set for processed food as juices. Then, the limit applied MRL applied for juice is the corresponding MRL for raw agricultural commodity, taking into account the concentration or dilution factor related to the manufacturing process (if available). Regulations and monitoring programs have to be adopted in order to strengthen food safety and control pesticide exposure to unacceptable levels in food. Analytical methodologies must be able to accurately determine the low concentration levels set up by the legislation. This is especially relevant for fruit and vegetable juices to have better knowledge of the pesticide levels actually present in this type of processed samples.

As it is already well-known, chromatographic techniques coupled to mass spectrometry (MS) are the most powerful tools for the identification and quantification of pesticides and other contaminants in food. Gas chromatography (GC) coupled to MS with single quadrupole analyzer operating in selected ion monitoring (SIM) has been widely applied for the multiresidue analysis of GC-amenable pesticides in different vegetable and fruit matrices (Mezcua et al. 2009; Mladenova and Shtereva 2009). The interest on reducing analysis time in multiresidue analysis has increased in the last years, looking for methods designed to determine as many compounds as possible in a short time. The use of fast GC allows rapid separations, satisfying current demands of higher sample throughput with not much sophisticated instrumentation (Dömötöröová and Matisová 2008; Kirchner et al. 2005). However, determination of pesticide residues in food typically requires multiple steps: extraction, cleanup and subsequent determination by GC, in some cases after derivatization; thus, faster sample treatment methods are also desirable to reach high sample throughput.

When dealing with liquid samples, like juices, a classical technique for sample preparation is solid-phase extraction (SPE). A wide variety of sorbents and elution solvents can be used depending on the characteristics of the compounds to be extracted. C₁₈ and Oasis hydrophilic–lipophilic balance (HLB) cartridges are among the most widely used in multiresidue methods (Marín et al. 2009; Xue et al. 2006; Piedra et al. 2000; Sabik et al. 2000; Picó et al. 2007; Cherta et al. 2012; Pitarch et al. 2007). Albero et al. (2005) developed a multiresidue method for the determination of pesticides using 10 ml juice samples C₁₈cartridge. Pang et al. (2006) used graphitized carbon black SPE cartridges in order to extract pesticides from 15 g of fruit juice. An immunoaffinity-based SPE procedure has been also applied for the determination of triazines in fruit juices (Dallüge et al. 1999).

A fast and simple alternative for sample treatment is the QuEChERS procedure (Anastassiades et al. 2003), which has been widely and successfully applied for the determination of pesticide residues in fruits and vegetables. It has been subjected to several modifications based on authors' preferences, but the AOAC Official Method

2007.01 (Lehotay et al. 2005) and the Standard Method EN 15662 (Payá et al. 2007) are the two official and most known versions. This methodology offers some advantages such as high sample throughput, high recoveries for a wide polarity and volatility range of pesticides and accurate results. QuEChERS combined with fast GC becomes a good choice to speed up multiresidue analysis. Although this procedure has been implemented for a wide range of commodities, especially fruits and vegetables in many routine laboratories with satisfactory results (Ciešlik et al. 2011; Dai et al. 2011; Jiang et al. 2009; Kolberg et al. 2011; Park et al. 2011; Cherta et al. 2012a), only few publications have been reported for the analysis of juice samples, especially using GC or fast GC. The original QuEChERS version has been applied for the determination of 118 pesticides in vegetable juice by GC-MS and liquid chromatography–tandem mass spectrometry (Nguyen et al. 2009). Furlani et al. (2011) reported the determination of pesticide residues in sugarcane juice by GC with electron capture detection, also applying the unbuffered original QuEChERS version.

Another interesting approach for juice samples is the use of solid-phase microextraction (SPME), which has been successfully applied in pesticide residue analysis in water, soil, food, and biological samples (Beltran et al. 2000, 2001, 2003; Kataoka et al. 2000; Cervera et al. 2011; Fuster et al. 2005). SPME has gained in popularity since it minimizes sample preparation and also allows performing extraction and preconcentration in a single step. The most common approach for nonvolatile pesticides is the application of SPME by direct immersion (DI-SPME) (Fidalgo-Used et al. 2006; Farajzadeh and Hatami 2004; Natangelo et al. 2002; Simplicio and Vilas Boas 1999), but its application to complex matrices is troublesome due to the absorption of interferences onto the fiber. This fact can be overcome if a previous solvent extraction is performed and the subsequent DI-SPME is applied over the separated aqueous extract (Kataoka et al. 2000; Zambonin et al. 2002) or by simply diluting the sample in order to simplify the matrix complexity (Sen et al. 1997). In the case of volatile compounds, the use of SPME in headspace mode (Hernández et al. 2002; López et al. 2001; Schurek et al. 2008; Serrano et al. 2009) allows minimizing the matrix interferences.

The aim of this work has been to critically compare three sample treatment methods based on QuEChERS, SPME and SPE in order to evaluate their applicability for pesticide residue analysis in fruit juice samples. All methods have been validated using apple juice samples. Determination has been performed by fast GC-MS with single quadrupole working under the SIM mode.

EXPERIMENTAL

Reagents and materials

Pesticide standards used for this work were purchased from Dr. Ehrenstorfer (Augsburg, Germany). Stock standard solutions (nominal concentration, 500 µg/ml) were prepared by dissolving reference standards in acetone and were stored in a freezer at -20 °C. Working standard mixtures for sample fortification were prepared by dilution of stock solutions in acetonitrile (for QuEChERS) and in acetone (for SPME and SPE).

Three isotopically labeled internal standards (ILIS) were used as surrogates: *p,p'*-DDE-D₈, terbuthylazine-D₅ (Dr. Ehrenstorfer) and hexachlorobenzene (HCB)-¹³C₆ (Cambridge Isotope Labs Inc., Andover, MA, USA). A working mixed solution of labeled standards was prepared by volume dilution of individual stock solutions with acetonitrile (MeCN) and acetone and stored at 4 °C.

Acetone, hexane, MeCN, dichloromethane (DCM), glacial acetic acid (HAc), anhydrous magnesium sulfate (MgSO₄), anhydrous sodium acetate (NaAc) and sodium chloride (NaCl) were purchased from Scharlab (Barcelona, Spain). All solvents were for pesticide residue analysis or high-performance liquid chromatography (HPLC) grade. Two types of 2 ml microcentrifuge tubes for dispersive solid-phase extraction (d-SPE; used for the cleanup step) containing 50 mg primary–secondary amine (PSA) and 150 mg anhydrous MgSO₄ or 50 mg PSA, 150 mg anhydrous MgSO₄ and 50 mg C₁₈ were obtained from Teknokroma (Barcelona, Spain).

Oasis HLB cartridges (200 and 60 mg) were purchased from Waters (Milford, MA, USA) and Bond Elut cartridges C₁₈ (500 mg) were obtained from Varian (Harbor City, CA, USA). SPME fibers of polydimethylsiloxane (PDMS, 100 μm), polyacrylate (PA, 85 μm), and divinylbenzene/carboxen/PDMS (DVB/CAR/PDMS, 50/30 μm) were purchased from Supelco (Madrid, Spain).

Sample material

Apple and multi-fruit juice used for the validation study were purchased from a local market in Castellón (Spain). Once the optimum method was validated, four different juices were analyzed to investigate the presence of pesticides and test the applicability of the method. Apple–mango, pineapple and grapefruit juices were purchased from a local market in Castellón. Natural orange juice was obtained from fresh oranges collected from local harvesters.

GC instrumentation

Chromatographic measurements were performed on a GC system (Shimadzu QP2010 Plus) equipped with an autosampler (Shimadzu AOC-5000) and coupled to a single quadrupole mass spectrometer (GCMS-QP2010 Plus). Compounds were separated on a SAPIENS-5MS (Teknokroma) capillary column (length 20 m × I.D. 0.10 mm × film 0.10 μm).

For the chromatographic analysis of QuEChERS extracts (in MeCN), injections (3 μl) were performed in programmable temperature vaporization (PTV) mode, which was programmed as follows: 40 °C (hold time, 0.5 min), maintaining the split valve open; once the valve was closed, a rate of 400 °C/min to 320 °C (hold time, 0.5 min) was applied, resulting in an injection total time of 1.70 min. During this time, initial oven temperature was maintained at 60 °C and then heated at a rate of 90 °C/min to 225 °C, then 15 °C/min to 270 °C, and finally 150 °C/min to 330 °C (2 min), resulting in a total analysis time of 8.93 min. Helium was used as carrier gas at a flow of 0.77 ml/min (corresponding to a linear velocity of 39.1 cm/s).

When SPME was performed, the injector was operated in splitless mode at 280 °C and the splitless time was 5 min. During this time, initial column oven temperature was maintained at 50 °C and then programmed as previously indicated. In this case, the total analysis time was 12.14 min.

The injector was also operated in splitless mode (1 µl) when SPE extracts were analyzed, although injection temperature was 320 °C, initial column temperature was 80 °C and splitless time was 1.2 min, so chromatographic run time was 8.01 min.

The mass spectrometer was operated in the electron ionization mode (70 eV). The source and the interface (transfer line) temperatures were adjusted to 225 and 300 °C, respectively. The scan time in SIM mode was set at 0.1 s. In SIM mode, the three most abundant and/or characteristic ions for each analyte were selected as target and reference ions. Solvent delay times of 3.5, 4 and 7 min for SPE, QuEChERS and SPME, respectively, were used to prevent damage to the filament of the ion source. Shimadzu software GCMSsolution was used to automatically process the data.

Analytical procedures

- ***QuEChERS extraction: AOAC Official Method 2007.01 (Lehotay et al. 2005)***

Fifteen milliliters of juice was poured in a 50-ml polypropylene centrifuge tube and 375 µl of surrogate standard solution mixture of 1 mg/L in MeCN was added and mixed on a vortex for 1 min. Extraction was carried out using 15 ml MeCN (with 1 % HAc) and shaking by hand during 30 s. Then, 6 g of anhydrous MgSO₄ and 1.5 g of anhydrous NaAc were added and immediately shaken vigorously by hand to prevent the formation of MgSO₄ agglomerates. Then, the tube was centrifuged at 3,000 rpm during 2 min.

For the cleanup step, 1 ml of the upper MeCN extract was poured into the d-SPE tubes containing 150 mg MgSO₄ and 50 mg PSA (in the case of orange juice samples, d-SPE tubes also contained 50 mg C₁₈). The tubes were shaken on a vortex for 30 s and centrifuged at 3,000 rpm for 2 min. The final MeCN supernatant extract

was directly injected into the GC system under the experimental conditions indicated before (PTV mode).

Matrix-matched calibration was used for each sample matrix in order to be able to adequately quantify analytes in real samples. In this way, 500 μ l of MeCN extract obtained from a blank sample were mixed with 50 μ l of the pesticide standard solution in MeCN of adequate concentration, also containing the three ILIS. Each analyte was quantified by using relative responses (areas) to the corresponding internal standard.

➤ ***SPE extraction***

Twenty-five microliters of surrogate standard mixture in acetone of 1 mg/L was added to 1 ml of juice sample and passed through the 200-mg (6 ml) Oasis cartridge, previously conditioned by passing 6 ml of methanol, 6 ml of ethyl acetate/DCM, 6 ml of methanol and 6 ml of deionized water. After loading the sample, cartridges were washed with 6 ml of deionized water and dried by passing air, using a vacuum for at least 30 min. The retained analytes were eluted with 5 ml ethyl acetate/DCM (50:50). The collected extract was evaporated, after the addition of 1 ml hexane, under a gentle nitrogen stream at 40 °C until 0.5 ml, adjusted to 1 ml with hexane and injected into the GC system under the experimental conditions indicated before. Quantification of analytes in samples was carried out from calibration curves prepared with standards in solvent, using relative responses of each compound to the corresponding ILIS.

➤ ***SPME extraction***

Extraction of juice samples was performed by direct immersion of a PA fiber into the sample, under magnetic stirring (600 rpm) for 30 min. Samples were prepared by adding 25 μ l of surrogate mixture (200 ng/ml) in acetone to 0.5 ml of juice and subsequent dilution with 1.5 ml of deionized water in a septum-capped 4-ml clear glass vial. Desorption of the fiber was carried out at 280 °C for 5 min in the splitless injector.

Quantification of analytes in samples was carried out using calibration curves prepared by spiking 2 ml of deionized water with 25 μl of pesticide standard solution of adequate concentration and 25 μl of surrogate mixture, both in acetone, and extracting these samples under the SPME procedure previously indicated. Relative responses of each compound to the corresponding internal standard were used.

Validation study

The three extraction methods used were validated using commercial apple juice samples in terms of linearity, accuracy, precision, limits of quantification (LOQ) and limits of detection (LOD). Confirmation capability of the method for positive samples was also evaluated.

Linearity was studied using calibration standards injected by triplicate. It was considered linear when regression coefficient was higher than 0.99 and the residuals lower than 30 % without any clear tendency.

Accuracy was estimated from recovery experiments at two concentration levels (10 and 100 $\mu\text{g/L}$ for QUEChERS and SPE; 1 and 10 $\mu\text{g/L}$ for SPME) ($n = 6$). Precision was expressed as repeatability (intraday precision) in terms of relative standard deviation (RSD, in percent) ($n = 6$) at each fortification level.

LOQ was estimated as the analyte concentration that produced a peak signal ten times that of the background noise and it was calculated using the chromatograms at the lowest fortification level tested with satisfactory recovery (70–120 %) and precision (RSD <20 %). LOD was estimated in the same way, but for a signal-to-noise ratio of 3.

In order to confirm peak identity in samples, the ratio between the quantification ion (target, Q) and the reference ions (q_i) was evaluated and compared with the theoretical value obtained from reference standard solutions. The confirmation criterion was based on the European Commission Decision 2002/657/EC (European Commission Decision 2002), which establishes the maximum tolerances for Q/q ratio deviation from theoretical values as a function of

relative intensities. Coincidence between the retention time in a sample and the corresponding standard was also required to confirm a positive finding (maximum deviation, ± 0.5 %).

RESULTS AND DISCUSSION

In a first step, for the optimization and comparison of extraction procedures, 15 selected pesticides (from a total of 56 pesticides studied in this work) were used (**Table 1**). Three sample treatments (QuEChERS, SPE and SPME) were studied in order to evaluate their advantages and disadvantages using apple juice sample as model matrix sample. The QuEChERS procedure was not optimized as it was based on the AOAC Official Method and applied in the same conditions as in our previous work (Cherta et al. 2013). SPME and SPE were subjected to an optimization study.

Table 1. List of compounds studied in method optimization.

Compound	QUECHERS		SPME		SPE		Monitored ions in SIM	
	t _R (min)	Time window (min) (SIM group)	t _R (min)	Time window (min) (SIM group)	t _R (min)	Time window (min) (SIM group)	Target ion	Reference ions
Trifluralin	4.542	4.30-4.70	7.931	7.80 – 8.05	3.772	3.60-3.86	264	290, 306
Atrazine	4.780	4.70-5.00	8.156	8.05 – 8.40	3.994	3.86-4.30	200	202, 215
Hexachlorobenzene- ¹³ C ₆ *	4.780		8.172		4.013		292	
Hexachlorobenzene	4.780		8.173		4.013		284	282, 286
Terbutylazine-D ₅ *	4.843		8.219		4.054		219	
Terbutylazine	4.853		8.229		4.064		214	173, 229
Chlorpyrifos methyl	5.214	5.00-5.40	8.586	8.40 – 8.80	4.423	4.30-4.60	286	125, 288
Alachlor	5.248		8.618		4.453		160	132, 188
Chlorpyrifos	5.508	5.40-5.65	8.868	8.80 – 9.10	4.703	4.60-4.90	314	197, 199
Aldrin	5.566		8.937		4.771		263	101, 261
Isodrin	5.783	5.65-5.90	9.148	9.10 – 9.35	4.982	4.90-5.10	193	195, 263
Endosulfan I	6.135	5.90-6.19	9.489	9.35 – 9.80	5.323	5.10-5.61	241	170, 239
<i>p,p'</i> -DDE-D ₈ *	6.240	6.19-6.45	9.575		5.413		254	
<i>p,p'</i> -DDE	6.256		9.591		5.428		246	248, 318
Dieldrin	6.348		9.693		5.528		263	265, 277
Endrin	6.548	6.45-7.00	9.889	9.80 – 10.20	5.723	5.61-6.10	263	261, 345
Endosulfan II ^(a)	6.626		9.960		5.796		241	243, 339
Bifenthrin	7.227	7.00-7.30	10.544	10.20 – 10.80	6.389	6.10-6.50	181	165, 166

* ILIS used in this work

^(a) Target ion modified to 243 in QuEChERS extraction.

SPME optimization

In order to establish the optimum conditions for the extraction of the selected pesticides in apple juice samples, several parameters of the SPME procedure (type of fiber, sample dilution, salt and solvent addition and extraction and desorption times) were considered separately. Extraction temperature was set at room temperature and magnetic stirring at 600 rpm.

Firstly, the selection of the fiber was carried out by testing three different fibers: PDMS, PA and DVB/CAR/PDMS, under the same SPME conditions and comparing the chromatographic responses obtained. Each fiber was immersed during 15 min into 0.5 ml of juice spiked at 50 µg/L (diluted with 2.5 ml of water) and desorbed into the GC at 280 °C during 5 min (DVB/CAR/PDMS was desorbed at 270 °C, according to the manufacturer's recommendation). The best results were obtained for the PA fiber, as shown in **Fig. 1**, so this fiber was used for further experiments.

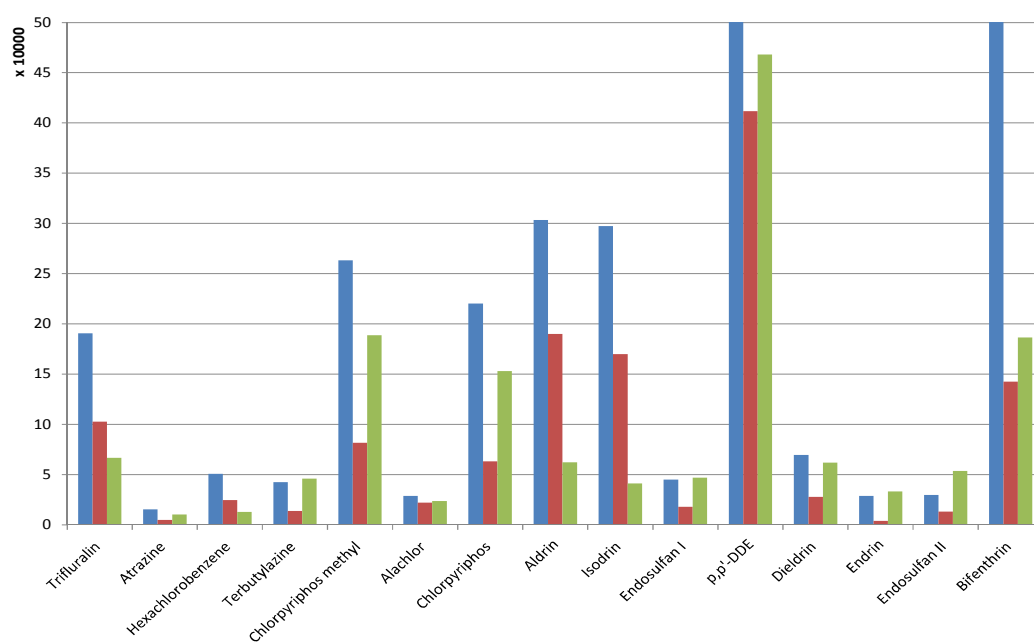


Fig. 1. Effect of SPME fiber type over extraction efficiency for selected pesticides (0.5 ml of 50 µg/L spiked apple juice, diluted with 2.5 ml of water; 15 min extraction).

A statistical optimization procedure based on a full factorial experiment design was applied. It allowed not only determining the optimum values for the selected variables but also detecting interactions between variables or identifying which ones did not affect the response. Optimization was carried out in a two-step scheme: first, a two-level full factorial design was applied to detect significant variables and, then, a surface response design was applied to determine the optimum values for those significant variables.

Three variables (addition of hexane/acetone (1:1), salting-out effect, and sample dilution) were studied at two levels (0 and 400 μ l for hexane, 0 and 20 % for NaCl, and 1.5 and 3.5 ml for H₂O). A 2³ factorial design was performed, including 3 central points, so a total number of 11 randomized experiments were done. The statistical software package Statgraphics Centurion XV was used to generate the table of experiments and to evaluate the results obtained. The area of each pesticide was used as response function. The main effects of each variable and all the interactions were studied by means of the resulting Pareto charts. **Fig. 2** illustrates an example of the corresponding Pareto chart obtained for HCB (as all compounds showed the same general trend).

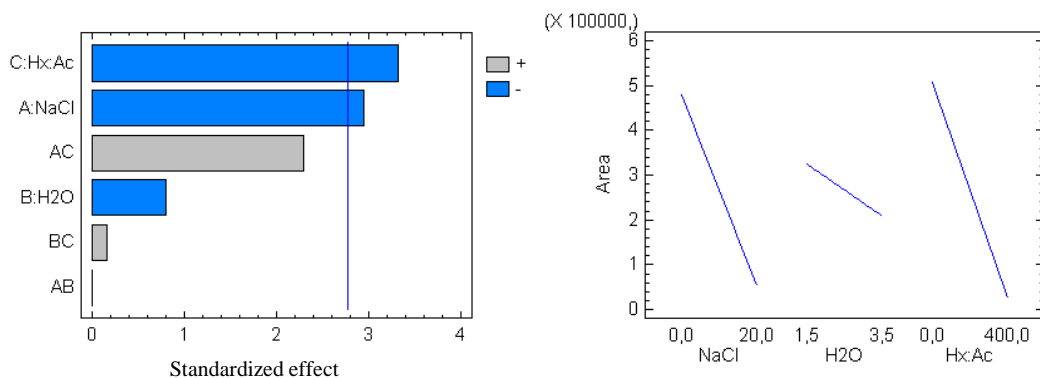


Fig. 2. Pareto chart of standardized effects of 2³ factorial design for HCB, using the peak area as the function response, and main effects plots.

The length of the horizontal bar in the chart is proportional to the absolute value of the estimated effect; the vertical line defines the 95 % confidence level. An effect is considered statistically significant if it exceeds this line. A general behavior was that solvent and salt addition presented a significant and negative effect, so both variables were selected for the next step in the optimization. As sample dilution was not significant but also had a negative effect, the lowest value tested, 1.5 ml, was selected. This minimum value was selected in order to have enough sample volume to cover the stationary phase of the SPME fiber and to have some matrix dilution effect that would improve quantification, as already described in the literature (Beltran et al. 2000).

Then, a 3^2 factorial design, including three central points, was performed in order to study salt and solvent addition at three levels. Values for these variables were set at the same levels as in the first design. This case required 12 randomized experiments and the response function used again was the peak area for each compound. The response surface for HCB (**Fig. 3**) obtained from the results of these experiments shows the negative effect of adding salt and solvent since extraction efficiency decreases proportionally to the addition of NaCl and hexane/acetone.

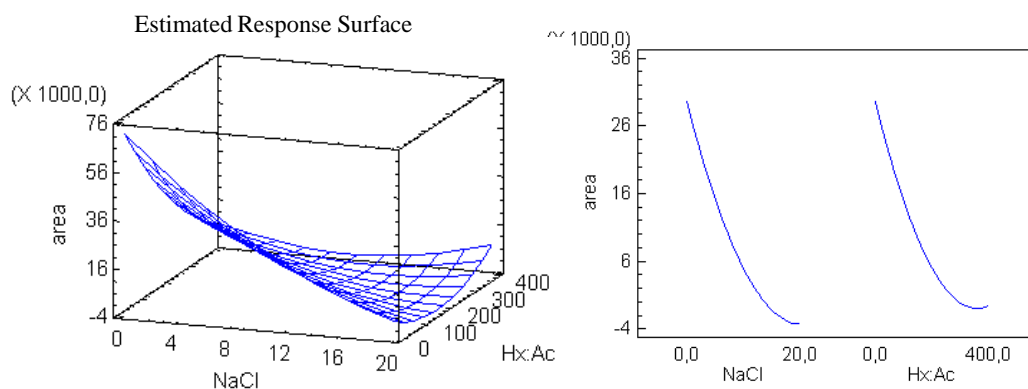


Fig. 3. Response surface obtained from a quadratic model for simultaneous optimization of salt and solvent addition, using the peak area as the function response, and main effects plots for HCB.

Therefore, the optimal conditions were found to be without modifying ionic strength or adding solvent. The effect of NaCl (or other ionic salts) has been widely discussed and different behaviors have been reported. In most cases, the salting out effect tends to increase extraction efficiency (Beltran et al. 1998; Boyd-Boland and Pawliszyn 1995), but this effect also depends on the solubility and polarity of the analytes and sometimes a decrease in sensitivity is noticed when larger amounts of salt are added (Cortés-Aguado et al. 2008; Magdic et al. 1996). Moreover, at higher concentration levels, NaCl crystals can occupy some of the active fiber sites and thus decrease extraction recoveries (Farajzadeh and Hatami 2004).

Optimization of the absorption equilibrium was performed by extracting replicate samples at different times (from 10 to 120 min). Analyte mass absorption, expressed as the peak area, was adjusted to a time-dependent equation given by Ai (1997):

$$n = n_o(1 - e^{-at})$$

where n and n_o are the amounts of analyte absorbed at a time t and at the equilibrium, respectively, and a is a parameter that measures how fast the absorption equilibrium can be reached in the SPME process. **Fig. 4** shows the results and the curves obtained after fitting the experimental data to the mentioned equation using the Statgraphics Centurion XV software for three of the studied compounds. Equilibrium time, estimated as the time necessary to extract 95 % of n_o , was calculated for all the pesticides, giving values higher than 120 min in all cases. The feasibility of working in nonequilibrium conditions was considered, and thus, using the fitted equations, it was stated that establishing an extraction time of 30 min would lead to an extraction of around 50 % with respect to the equilibrium situation for most compounds, and thus, analysis time would be considerably reduced.

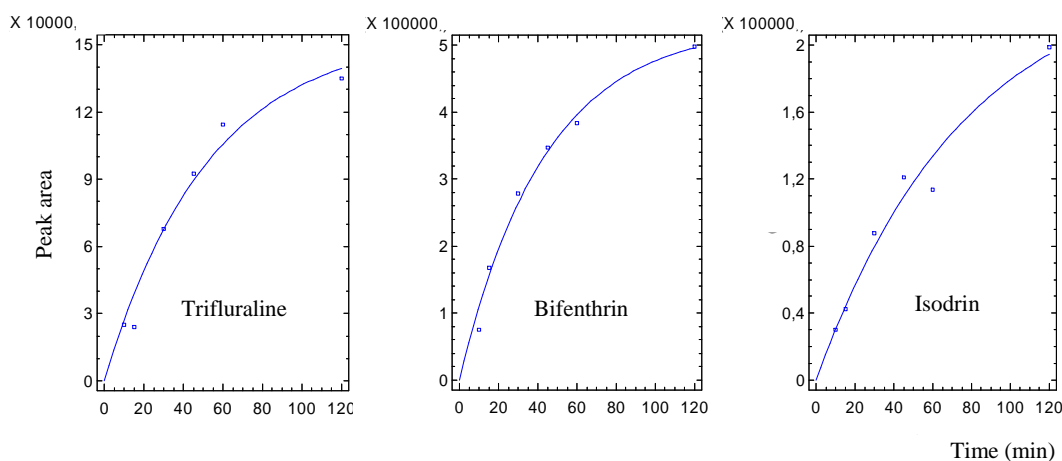


Fig. 4. SPME absorption time profiles for selected pesticides (0.5 ml of 10 $\mu\text{g/L}$ spiked apple juice, diluted with 2.5 ml of water).

Finally, desorption time was studied in the range of 1–9 min under the optimum extraction conditions previously indicated. Peak areas increased with longer desorption times until complete desorption was reached, at about 5 min for most compounds. During this long desorption time, in order to profit from the effect of cold trapping, the oven temperature was maintained isothermal at 50 °C.

SPE optimization

SPE parameters as sorbent type, elution volume, elution solvent, volume of sample and evaporation step were optimized in order to find the optimum conditions for the extraction of the selected pesticides in juice samples.

As a first step, 200 and 60 mg Oasis HLB cartridges and 500 mg Bond Elut cartridges C_{18} were tested using 1 ml of apple juice sample fortified with the 15 selected pesticides. Elution was carried out with 5 ml of ethyl acetate/DCM. More consistent recoveries with lower RSD were obtained when 200 mg Oasis HLB cartridges were used, so this sorbent was selected for further experiments.

Different solvents and mixtures of solvents were also studied to set the best elution conditions. Acetone, ethyl acetate, DCM and a mixture of ethyl acetate and

DCM (50:50) were used to elute analytes retained in the cartridges at different volumes (from 2 to 10 ml). The elution using ethyl acetate/DCM mixture led to higher recoveries (although not still completely satisfactory), and 5 ml was selected since poor recoveries were obtained using lower elution volumes and no significant differences were observed at higher volumes.

The evaporation step was carefully studied by evaluating the effect of evaporation until dryness. Poor recoveries were obtained when the SPE extract was evaporated until dryness and redissolved with hexane until 1 ml. Thus, the addition of 1 ml hexane before the evaporation was considered; in this case, the extract was evaporated until 0.5 ml and then adjusted to 1 ml with hexane, avoiding possible losses of analytes during the evaporation process.

Different volumes of sample were tested in order to evaluate the maximum amount of sample to be passed through the cartridge without affecting the retention of the analytes. One milliliter of apple juice fortified at 100 µg/L, 10 ml fortified at 10 µg/L and 100 ml fortified at 1 µg/L were loaded to the cartridge and the signal intensity was evaluated. No chromatographic peaks were observed when using 100 ml, since the cartridge was overloaded with matrix components, impeding the retention of the analytes. The use of 10 ml also had a negative effect on the chromatographic signal since a loss of 75 % with respect to the use of 1 ml was observed. Therefore, 1 ml of sample volume was selected, achieving satisfactory sensitivity and maintaining the high speed of sample preparation.

Comparison of analytical characteristics

In order to critically compare the three extraction procedures, accuracy, precision, LOD and LOQ were evaluated using apple juice blank samples spiked with the 15 pesticides selected as model. Three ILIS were used as surrogates in order to correct possible losses of the analytes during the extraction process and/or instrumental deviations. Terbutylazine-D₅ was used as internal standard for herbicides, organophosphorus (OP) insecticides and pyrethroids; DDE-D₈ was used

for organochlorine pesticides and trifluralin; and HCB-¹³C₆ was used for HCB. The specific internal standard used for each individual compound is indicated in **Table 2**.

Linearity was studied in the range 1–500 µg/L ($n = 3$) when QuEChERS and SPE were applied. In the case of SPME, the linearity was studied in the range 0.5–50 µg/L, and fitting the experimental data to quadratic curves; this concentration range could not be extended to higher values due to the large signal intensity of most compounds that saturated the detector (although a wider range could be achieved by the dilution of the sample before the SPME). The regression coefficients were higher than 0.99 for all compounds over the whole range tested in the three methodologies and the residuals lower than 30 %.

As regards the matrix effects, the corresponding study for QuEChERS procedure was performed in our previous work (Cherta et al. 2013), concluding that matrix-matched calibration curves were necessary to compensate for matrix effects in quantitative applications. On the contrary, no severe matrix effects were observed when SPE was applied, so calibration curves prepared in solvent could be used in this case, being this an important advantage. In the SPME procedure, matrix effects were evaluated by comparison of chromatographic responses of spiked water (2 ml) and spiked juice samples (0.5 ml juice and 1.5 ml water), both extracted by SPME. No significant differences or signal enhancements were observed, probably due to the matrix dilution already considered in the development of the procedure. Then, calibration curves prepared in water and juice were analyzed, obtaining similar calibration slopes for most compounds, so calibration prepared in HPLC water could be used instead of matrix-matched calibration curves in the SPME procedure.

Table 2. Average recovery (%) and R.S.D. (in parenthesis) after the application of three extraction techniques in apple juice (n=6) fortified at two concentration levels.

Compounds	QuEChERS			SPE			SPME		
	Fortification levels (µg/L)		LOD (µg/L)	Fortification levels (µg/L)		LOD (µg/L)	Fortification levels (µg/L)		LOD (µg/L)
	10	100		10	100		1	10	
Trifluralin ^(a)	98 (8)	101 (9)	0.05	0.2	0.2	0.2	0.6	0.001	0.003
Atrazine ^(b)	102(5)	84 (13)	0.7	3	3	0.5	2	0.05	0.2
Hexachlorobenzene ^(c)	106 (1)	116 (2)	0.2	0.7	0.1	0.1	0.3	0.007	0.03
Terbutylazine ^(b)	107 (6)	99 (8)	0.5	2	0.3	0.3	0.9	0.005	0.02
Chlorpyrifos methyl ^(b)	94 (5)	83 (15)	0.1	0.4	5	5	15	0.05	0.2
Alachlor ^(b)	104 (12)	94 (12)	0.3	1	0.2	0.2	0.6	0.1	0.3
Chlorpyrifos ^(b)	112 (6)	85 (14)	0.2	0.7	0.5	0.5	2	0.003	0.01
Aldrin ^(a)	102 (4)	109 (6)	0.2	0.7	3	3	10	0.02	0.06
Isodrin ^(a)	117 (5)	104 (8)	0.9	3	1	1	4	0.02	0.06
Endosulfan I ^(a)	-	103 (2)	6	18	-	9	27	0.04	0.2
p,p'-DDE ^(a)	100 (3)	103 (2)	0.08	0.3	0.2	0.2	0.6	0.008	0.03
Dieldrin ^(a)	68 (4)	108 (3)	0.9	3	0.7	0.7	3.0	0.02	0.06
Endrin ^(a)	111 (5)	105 (5)	1	4	5	5	15	0.05	0.2
Endosulfan II ^(a)	-	103 (3)	5	15	1	1	3	0.05	0.2
Bifenthrin ^(b)	118 (10)	93 (9)	0.2	0.7	0.3	0.3	0.9	0.03	0.1

(a), (b), (c) indicates the internal standard used for each analyte: (a) p,p'-DDE-D₈, (b) terbutylazine-D₈, (c) hexachlorobenzene-¹³C₆.

Underlined, not acceptable results. Detection (LOD) and quantification (LOQ) limits.

*Data not available due to heavy matrix interferences.

Accuracy and precision were evaluated by analyzing juice samples fortified at two concentrations ($n = 6$) of 10 and 100 $\mu\text{g/L}$ for QuEChERS and SPE and 1 and 10 $\mu\text{g/L}$ for SPME. Recoveries and RSD obtained for each analyte were calculated. As shown in **Table 2**, satisfactory recoveries (between 70 and 120 %) were obtained for all compounds at both spiking levels for the QuEChERS procedure, as well as adequate RSD values (lower than 15 %). Only two analytes, endosulfan I and II, could not be quantified at the lowest level due to poor sensitivity. LOQs ranged from 0.2 to 4 $\mu\text{g/L}$, except for endosulfan I and II (around 15 $\mu\text{g/L}$). Similar results were obtained for SPE, with the exceptions of chlorpyrifos methyl and aldrin, whose LOQs were slightly higher due to the inadequate recoveries presented at the lowest level. Endosulfan I and endrin could not be quantified at 10 $\mu\text{g/L}$, but satisfactory recoveries were obtained at the highest fortification level. The LOQs achieved with both methodologies are in agreement to those previously obtained by other authors for pesticides in fruit juices (Albero et al. 2005). On the other hand, the application of SPME led to an important gain in sensitivity, which can be appreciated in **Fig. 5**, reaching lower LOQs than those obtained in the other extraction methods (even 200 times lower for some compounds). This behavior was also reported in the literature for OP pesticides (Beltran et al. 1998), but in general terms, LODs obtained for pesticide residues ranges from 0.1 to 10 $\mu\text{g/L}$ (Hernández et al. 2002; López et al. 2001; Cortés-Aguado et al. 2008), so an important enhancement of sensitivity is achieved under conditions used in this work. However, five compounds presented inadequate recoveries at the lowest level, although they could be validated at 10 $\mu\text{g/L}$. LOQs obtained for QuEChERS and SPE could not reach the nanograms per liter level, but were low enough for regulations purposes, considering that MRLs are commonly set at 10 $\mu\text{g/L}$ in food commodities.

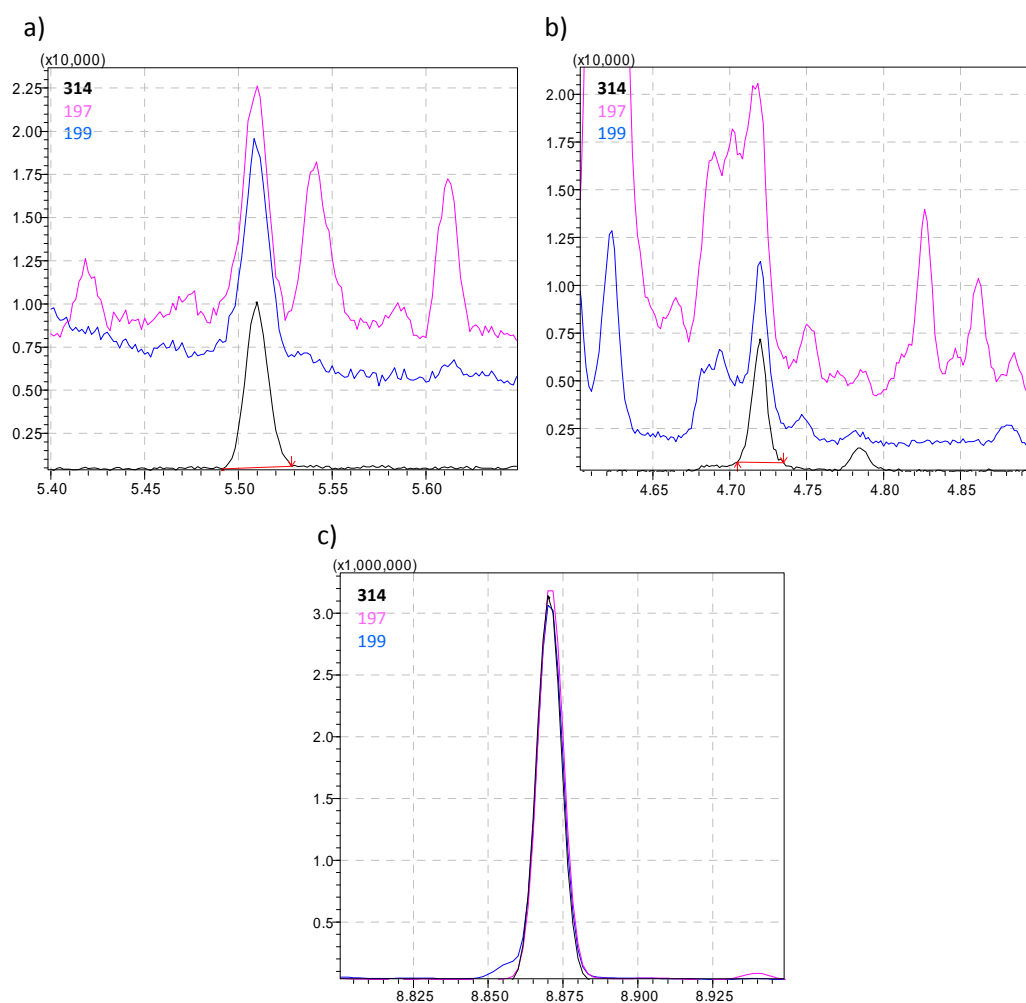


Fig 5. Comparison of chromatographic responses for chlorpyrifos in apple juice extract spiked at 0.01 mg/L after applying a) QuEChERS, b) SPE, and c) SPME. Target ion (*bold*) and two reference ions are shown.

As regards the confirmation of positive samples, the guidelines of the European Commission Decision (2002) establish that, after the acquisition of three ions (target (Q) and two reference ions (q_i)), the comparison of the two Q/q ratios measured in samples with those measured from reference standards shall lie within the maximum permitted tolerances. However, the expected Q/q ratios can be altered mainly due to matrix interferences. This is specially noticed at low concentration levels owing to the

lower abundance of the ions. In this work, we applied a more realistic criterion for the three methodologies: at the retention time of the analyte, three ions (target and reference) have to be observed in the sample and at least one Q/q ratio has to be accomplished.

It is noteworthy that differences on Q/q ratio accomplishment were observed depending on the extraction procedure applied. A higher number of compounds did not get ion ratios within the permitted tolerances after applying the SPE procedure. This seemed to be related to the fact that a higher number of reference ions were interfered by matrix coeluting components when SPE was applied, as illustrated in **Fig. 5** in the case of chlorpyrifos. Better results of Q/q ratio accomplishment were obtained for the QuEChERS procedure, surely due to the cleanup step included in the procedure. Thus, QuEChERS seemed to be a more adequate sample treatment for complex matrices than SPE. In the case of SPME, the higher sensitivity favored the compliance of Q/q ratios similar to QuEChERS, so it can be concluded that SPE is not a good enough extraction method for juice samples.

Extraction time was also evaluated, considering that the analysis method was based on fast GC. SPME involved a longer extraction time since samples were extracted one by one (30 min of extraction for sample), so it reduced dramatically the sample throughput. Decreasing the extraction time to a value similar to that of the chromatographic run (maximizing sample throughput) would lead to lower extraction efficiency (around 20 % with respect to the equilibrium situation for most compounds). Moreover, this technique requires an additional desorption time once injected into the GC, so it resulted in longer chromatographic time (10.8 min). QuEChERS is considered as a rapid method and less labor-consuming; around 10 samples can be extracted in approximately 2 h. Moreover, shorter chromatographic time was possible after applying QuEChERS extraction (chromatographic time was around 9 min), taking more benefit from the fast GC.

In summary, the main advantage of SPME was the null solvent consumption and the possibility of reaching very low LOQs. However, the poor reproducibility of SPME specially noticed at low levels complicated the performance of the calibration

curves and the subsequent quantification process. As regards QuEChERS, it was the faster extraction procedure, fitting well with fast GC, and led to more satisfactory quantification results. Therefore, QuEChERS was selected for further validation of a wider list of pesticides, included in **Table 3**.

Table 3. Fast GC-MS conditions for 56 pesticides studied in the QuEChERS extraction applied to multi-fruit juice.

Compound	t _R (min)	Window (min)	Scan time (s)	Monitored ions in SIM	
				Target ion	Reference ions
Dichlorvos	3.735	3.60-3.80	0.10	185	109, 187
Chlorpropham	4.520	3.80-4.61	0.10	127	154, 213
Trifluralin	4.542			264	290, 306
Phorate	4.652	4.61-4.81	0.13	260	121, 231
alpha-HCH	4.723			219	181, 217
Atrazine	4.780			200	202, 215
Hexachlorobenzene- ¹³ C ₆ *	4.780			292	
Hexachlorobenzene	4.780			284	282, 286
Terbutylazine-D ₅ *	4.843			219	
Terbutylazine	4.853	4.81-5.01	0.13	214	173, 229
beta-HCH	4.880			181	217, 219
Propyzamide	4.882			173	175, 255
Diazinon	4.885			137	152, 179
Lindane	4.890			181	183, 219
Pirimicarb	5.038	5.01-5.27	0.18	166	138, 238
Chlorothalonil	5.127			266	264, 268
Metribuzin	5.172			144	198, 199
Chlorpyrifos methyl	5.214			286	197, 288
Parathion methyl	5.235			263	216, 246
Alachlor	5.248			160	132, 188
Heptachlor	5.311	5.27-5.47	0.11	272	100, 102
Pirimiphos methyl	5.332			290	125, 244
Fenitrothion	5.388			109	260, 277
Malathion	5.400	5.47-5.67	0.15	127	125, 173
Fenthion	5.502			245	279, 280
Metholachlor	5.505			162	146, 238
Chlorpyrifos	5.508			314	197, 199
Parathion ethyl	5.566			291	139, 155
Aldrin	5.566	5.67-5.91	0.18	263	101, 261
Cyprodinil	5.822			224	210, 225
Pendimethalin	5.736			252	162, 192
Clofenvinphos	5.780			267	269, 323
Isodrin	5.783			193	195, 263

Table 3 (continued).

Compound	t _R (min)	Window (min)	Scan time (s)	Monitored ions in SIM	
				Target ion	Reference ions
Quinalphos	5.838			146	156, 157
Tolylfluanid	5.843	5.91-6.20	0.10	137	238, 240
Methidathion	5.985			145	93, 125
trans-Chlordane	6.011			375	371, 373
Endosulfan I	6.135			170	239, 241
<i>p,p'</i> -DDE-D ₈ *	6.240			254	
<i>p,p'</i> -DDE	6.256	6.20-6.50	0.10	246	248, 318
Buprofezin	6.309			105	104, 172
Dieldrin	6.348			263	265, 277
Endrin	6.548	6.50-6.88	0.15	263	261, 345
Endosulfan II	6.626			195	241, 339
<i>p,p'</i> -DDD	6.631			165	176, 199
Ethion	6.633			125	153, 384
Oxadixyl	6.713			132	120, 146
Propiconazole I	6.915	6.88-7.18	0.10	173	175, 259
Propiconazole II	6.640			173	175, 259
<i>p,p'</i> -DDT	6.650			165	199, 212
Endosulfan sulfate	6.952			272	227, 274
Bifenthrin	7.227	7.18-7.42	0.10	181	165, 166
Phosmet	7.337			160	104, 161
Methoxychlor	7.338			227	212, 228
Tetradifon	7.470	7.42 -7.65	0.10	159	227, 229
Pyriproxyfen	7.518			136	137, 186
Fenarimol	7.705	7.65-7.85	0.10	139	219, 251
Cypermethrin	8.168	7.85-8.90	0.10	163	127, 181
Fenvalerate	8.470			125	167, 169

* ILIS used in this work

Validation for QuEChERS procedure for 56 pesticides

A complete validation of QuEChERS was performed for 56 pesticides in multi-fruit juice. The three ILIS were again used as surrogates. The specific internal standard used for each compound is indicated in **Table 4**.

Table 4. Average recovery (in percent) and RSD (in parenthesis) for multi-fruit juice after QuEChERS extraction and fast GC-MS analysis.

Compounds	Fortification levels ($\mu\text{g/L}$)		LOD ($\mu\text{g/L}$)	LOQ ($\mu\text{g/L}$)
	10	100		
Dichlorvos ^(a)	96 (10)	86 (11)	0.6	2
Chlorpropham ^(a)	103 (12)	90 (10)	0.6	2
Trifluralin ^(b)	95 (13)	105 (7)	0.3	0.9
Phorate ^(a)	99 (8)	87 (9)	2	6
alpha-HCH ^(b)	99 (7)	104 (5)	2	6
Atrazine ^(a)	-	83 (12)	6	18
Hexachlorobenzene ^(c)	92 (5)	113 (3)	0.3	0.9
Terbutylazine ^(a)	109 (6)	95 (7)	2	6
beta-HCH ^(b)	88 (8)	87 (7)	2	6
Propyzamide ^(a)	-	100 (5)	7	21
Diazinon ^(a)	112 (6)	86 (8)	3	10
Lindane ^(b)	91 (9)	83 (8)	2	6
Pirimicarb ^(a)	104 (8)	82 (7)	2	6
Chlorothalonil ^(c)	-	-	-	-
Metribuzin ^(a)	-	86 (14)	12	36
Chlorpyrifos methyl ^(a)	91 (9)	81 (12)	0.3	0.9
Parathion methyl ^(a)	-	-	-	-
Alachlor ^(a)	96 (10)	88 (15)	1	3
Heptachlor ^(b)	68 (14)	98 (9)	2	6
Pirimiphos methyl ^(a)	120 (4)	111 (8)	2	6
Fenitrothion ^(a)	-	-	-	-
Malathion ^(a)	-	80 (11)	4	12
Fenthion ^(a)	-	109 (8)	10	30
Metholachlor ^(a)	105 (3)	88 (13)	0.7	2
Chlorpyrifos ^(a)	119 (7)	104 (7)	0.9	3
Parathion ethyl ^(a)	-	-	-	-
Aldrin ^(b)	110 (6)	85 (6)	2	6
Cyprodinil ^(c)	-	-	-	-
Pendimethalin ^(a)	-	95 (6)	8	24
Chlofenvinphos ^(a)	-	105 (10)	9	27
Isodrin ^(b)	76 (15)	78 (10)	2	6
Quinalphos ^(a)	-	92 (9)	6	18
Tolyfluanid ^(c)	-	-	-	-
Methidathion ^(a)	i.	i.	-	-
trans-Chlordane ^(b)	78 (12)	89 (6)	0.6	2
Endosulfan I ^(b)	-	89 (9)	8	24
<i>p,p'</i> -DDE ^(b)	87 (11)	82 (4)	0.6	2
Buprofezin ^(c)	-	120 (11)	15	45

Table 4 (continued).

Compounds	Fortification levels ($\mu\text{g/L}$)		LOD ($\mu\text{g/L}$)	LOQ ($\mu\text{g/L}$)
	10	100		
Dieldrin ^(b)	-	85 (10)	4	12
Endrin ^(b)	-	93 (9)	5	15
Endosulfan II ^(b)	-	107 (9)	10	30
<i>p,p'</i> -DDD ^(b)	119 (15)	78 (4)	2	6
Ethion ^(a)	-	88 (7)	5	15
Oxadixyl ^(c)	-	-	-	-
Propiconazole I ^(c)	-	120 (2)	20	60
Propiconazole II ^(c)	-	117 (8)	20	60
<i>p,p'</i> -DDT ^(b)	-	-	-	-
Endosulfan sulfate ^(b)	-	108 (12)	10	30
Bifenthrin ^(a)	83 (8)	72 (8)	1	3
Phosmet ^(a)	i.	i.	-	-
Methoxychlor ^(b)	-	-	-	-
Tetradifon ^(c)	-	<u>126 (5)</u>	-	-
Pyriproxyfen ^(c)	-	<u>127 (5)</u>	-	-
Fenarimol ^(c)	-	<u>129 (7)</u>	-	-
Cypermethrin ^(a)	-	-	-	-
Fenvalerate ^(a)	-	-	-	-

(a), (b), (c) indicates the internal standard used for each analyte: (a) *terbutylazine-D*₅, (b) *p,p'*-DDE-*D*₈, (c) hexachlorobenzene-¹³C₆.

Underlined, not acceptable results. Detection (LOD) and quantification (LOQ) limits.

i., analyte not detected due to matrix interferences on the three analyte ions.

Linearity using matrix-matched standards was studied in the range 5–500 $\mu\text{g/L}$ ($n = 3$). Residuals were lower than 30 % and correlation coefficients by linear curves were higher than 0.99.

Accuracy and precision were evaluated by analyzing juice samples fortified at two levels (0.01 and 0.1 mg/L, $n = 6$). Results of recoveries and RSD are shown in **Table 4**. Half of the compounds could not be validated at the lowest level due to insufficient sensitivity and/or matrix interferences, in agreement with previous works (Cherta et al. 2013), but most of them presented satisfactory recoveries (between 70 and 120 %) at 0.1 mg/L, as well as adequate RSD values (lower than 15 %). LOQs

ranged from 2 to 25 µg/L in most cases; exceptions were metribuzin, fenthion, buprofezin, endosulfan II, propiconazole and endosulfan sulfate, with LOQs between 30 and 60 µg/L. Results are in accordance with recent literature (Nguyen et al. 2009; Furlani et al. 2011).

In order to test the applicability of the GC-MS method developed for the 56 studied pesticides, it was applied to real commercially obtained samples. Representative samples of four matrices were selected and analyzed, including mango–apple, pineapple, grapefruit and natural orange juices. Multi-fruit juice was used to perform calibration curves. As it corresponds to healthy commercial juices, no positive findings were detected in any of the samples.

CONCLUSIONS

Three different sample treatments based on QuEChERS, SPE and SPME have been applied and evaluated for the determination of pesticides in juice samples by fast GC-MS. A comparative study in terms of validation results, extraction efficacy and extraction times has been carried out for 15 representative pesticides in order to establish the best extraction conditions.

Most compounds presented a similar behavior in terms of recoveries and RSD. However, SPME resulted in the most sensitive approach allowing to reach better LOQs (up to 200 times lower) in comparison with QuEChERS and SPE. On the other hand, more matrix interferences were observed after injecting the SPE extracts, leading to poorer *Q/q* ratio accomplishment that made identification of compounds in samples more problematic. Thus, SPE without additional cleanup seemed less adequate for complex matrices. Better results were obtained for the other two methodologies due to the cleanup step included in QuEChERS and the higher sensitivity achieved with SPME. As regards extraction times, SPME was the most time-consuming procedure and involved the longest chromatographic run time since an additional 5 min of desorption step in the injector was necessary. On the contrary,

QuEChERS led to the highest sample throughput, making feasible the analysis of around 30 samples in 1 day.

QuEChERS was considered the most appropriate sample treatment for juice samples, although SPME allowed reaching lower quantification limits. The QuEChERS procedure in combination with fast GC-MS was extended to the residue determination of 56 pesticides in multi-fruit samples, with acceptable results for the wide majority of compounds. Analysis of fruit juice samples of apple–mango, pineapple, grapefruit and orange revealed that any of the pesticides investigated were present at levels above the LOD, all well below the MRLs.

Acknowledgments

This work has been developed under the financial support of Bancaixa (P1-1B2009-25 and P1-1B2010-23). The authors are very grateful to Izasa S.A. for providing the chromatographic system Shimadzu QP2010 Plus and acknowledge the financial support of Generalitat Valenciana, as research group of excellence PROMETEO/2009/054. The authors wish to thank Teknokroma for providing the 20-m GC column SAPIENS 5-MS. L. Cherta is very grateful to University Jaume I for his pre-doctoral grant.

References

- Ai J (1997) *Anal Chem* 69:1230
- Anastassiades M, Lehotay SJ, Štajnbaher D, Schenck FJ (2003) *J AOAC Int* 86:412
- Albero B, Sánchez-Brunete C, Tadeo JL (2005) *Talanta* 66:917
- Beltran J, Lopez FJ, Cepria O, Hernandez F (1998) *J Chromatogr A* 808:257
- Beltran J, López FJ, Hernández F (2000) *J Chromatogr A* 885:389
- Beltran J, Peruga A, Pitarch E, López FJ, Hernández F (2003) *Anal Bioanal Chem* 376:502

Beltran J, Pitarch E, Egea S, López FJ, Hernández F (2001) *Chromatographia* 54:757

Boyd-Boland AA, Pawliszyn JB (1995) *J Chromatogr A* 704:163

Burchat CS, Ripley BD, Leishman PD, Ritcey GM, Kakuda Y, Stephenson GR (1998) *Food Addit Contam* 15:61

Cervera MI, Beltran J, Lopez FJ, Hernandez F (2011) *Anal Chim Acta* 704:87

Cieślik E, Sadowska-Rociek A, Ruiz JMM, Surma-Zadora M (2011) *Food Chem* 125:773

Cherta L, Beltran J, López FJ, Hernández F (2013) *Food Anal Meth* 6:1170

Cherta L, Beltran J, Portolés T, Hernández F (2012) *Anal Bioanal Chem* 402:2301

Cortés-Aguado S, Sánchez-Morito N, Arrebola FJ, Frenich AG, Vidal JLM (2008) *Food Chem* 107:1314

Dai R, Ren X, He X, Huo Y (2011) *Bull Environ Contam Toxicol* 86:559

Dallüge J, Hankemeier T, Vreuls RJJ, Brinkman UAT (1999) *J Chromatogr A* 830:377

Dömötörövá M, Matisová E (2008) *J Chromatogr A* 1207:1

European Commission (2008) Regulation (EC) No. 299/2008

European Commission Decision (2002) European Commission Decision 2002/657/EC. Official Journal of the European Community. 21 Aug 2002

Farajzadeh MA, Hatami M (2004) *Chromatographia* 59:259

Fidalgo-Used N, Montes-Bayón M, Blanco-González E, Sanz-Medel A (2006) *J Anal At Spectrom* 21:876

Furlani RPZ, Marcilio KM, Leme FM, Tfouni SAV (2011) *Food Chem* 126:1283

Fuster S, Beltran J, López FJ, Hernández F (2005) *J Sep Sci* 28:98

Hernández F, Pitarch E, Beltran J, López FJ (2002) *J Chromatogr B Biomed Anal Technol Biomed Life Sc* 769:65

Jiang Y, Li X, Xu J, Pan C, Zhang J, Niu W (2009) *Food Addit Contam Part A Chem Anal Control Expo Risk Assess* 26:859

Kataoka H, Lord HL, Pawliszyn J (2000) *J Chromatogr A* 880:35

Kirchner M, Matisová E, Hrouzková S, De Zeeuw J (2005) *J Chromatogr A* 1090:126

Kolberg DI, Prestes OD, Adaime MB, Zanella R (2011) *Food Chem* 125:1436

Lehotay SJ, Maštovská K, Lightfield AR (2005) *J AOAC Int* 88:615

López FJ, Pitarch E, Egea S, Beltran J, Hernández F (2001) *Anal Chim Acta* 433:217

Magdic S, Boyd-Boland A, Jinno K, Pawliszyn JB (1996) *J Chromatogr A* 736:219

Marín JM, Gracia-Lor E, Sancho JV, López FJ, Hernández F (2009) *J Chromatogr A* 1216:1410

Mezcua M, Martínez-Uroz MA, Wylie PL, Fernández-Alba AR (2009) *J AOAC Int* 92:1790

Mladenova R, Shtereva D (2009) *Food Addit Contam Part A Chem Anal Control Expo Risk Assess* 26:854

Natangelo M, Tavazzi S, Benfenati E (2002) *Anal Lett* 35:327

Nguyen TD, Yun MY, Lee GH (2009) *J Agric Food Chem* 57:10095

Pang G, Fan C, Liu Y, Cao Y, Zhang J, Fu B et al (2006) *Food Addit Contam* 23:777

Park J, Choi J, Abd El-Aty AM, Kim BM, Oh J, Do J et al (2011) *Food Chem* 128:241

Patyal SK, Lakhanpal AK, Nath A, Sharma PC (2004) *J Food Sci Technol* 41:316

Payá P, Anastassiades M, MacK D, Sigalova I, Tasdelen B, Oliva J et al (2007) *Anal Bioanal Chem* 389:1697

Picó Y, Fernández M, Ruiz MJ, Font G (2007) *J Biochem Biophys Methods* 70:117

Picó Y, Kozmutza C (2007) *Anal Bioanal Chem* 389:1805

Piedra L, Tejedor A, Hernando MD, Aguera A, Barcelo D, Fernández-Alba A (2000) *Chromatographia* 52:631

Pitarch E, Medina C, Portolés T, López FJ, Hernández F (2007) *Anal Chim Acta* 583:246

Sabik H, Jeannot R, Rondeau B (2000) *J Chromatogr A* 885:217

Schurek J, Portolés T, Hajslova J, Riddellova K, Hernández F (2008) *Anal Chim Acta* 611:163

Sen NP, Seaman SW, Page BD (1997) *J Chromatogr A* 788:131

Serrano E, Beltrán J, Hernández F (2009) J Chromatogr A 1216:127

Simplicio AL, Vilas Boas L (1999) J Chromatogr A 833:35

Xue N, Zhang D, Xu X (2006) Water Res 40:183

Zambonin CG, Cilenti A, Palmisano F (2002) J Chromatogr A 967:255