Viroporins

How to cite: Angew. Chem. Int. Ed. 2023, 62, e202309069 doi.org/10.1002/anie.202309069

In Situ Spectroscopic Detection of Large-Scale Reorientations of Transmembrane Helices During Influenza A M2 Channel Opening**

Ronja Rabea Paschke⁺, Swantje Mohr⁺, Sascha Lange, Adam Lange, * and Jacek Kozuch*

Abstract: Viroporins are small ion channels in membranes of enveloped viruses that play key roles during viral life cycles. To use viroporins as drug targets against viral infection requires in-depth mechanistic understanding and, with that, methods that enable investigations under in situ conditions. Here, we apply surfaceenhanced infrared absorption (SEIRA) spectroscopy to Influenza A M2 reconstituted within a solid-supported membrane, to shed light on the mechanics of its viroporin function. M2 is a paradigm of pH-activated proton channels and controls the proton flux into the viral interior during viral infection. We use SEIRA to track the large-scale reorientation of M2's transmembrane a-helices in situ during pH-activated channel opening. We quantify this event as a helical tilt from 26° to 40° by correlating the experimental results with solidstate nuclear magnetic resonance-informed computational spectroscopy. This mechanical motion is impeded upon addition of the inhibitor rimantadine, giving a direct spectroscopic marker to test antiviral activity. The presented approach provides a spectroscopic tool to quantify large-scale structural changes and to track the function and inhibition of the growing number of viroporins from pathogenic viruses in future studies.

Viroporins are a family of small, ion channel-forming viral membrane proteins that play essential roles at various steps of the viral life cycle and, thus, during infection.^[1] To date, many viroporins, such as the HIV-1 Vpu, Hepatitis p7, and SARS CoV-2 E protein, have been characterized regarding their structure, function, and virological significance.^[2] However, while new members are added steadily to this class of proteins,^[3] it was the M2 proton channel from

Influenza A virus (IAV)^[4] that garnered much interest for this family of proteins.^[5] During the 1980's, it was shown that anti-IAV adamantane drugs inhibit viral infection via interaction with M2, and thus by blocking its viroporin function.^[6] With the prevalence of diseases caused by pathogenic viruses but scarcity of potent treatments, this demonstrated that viroporins present a promising target for antiviral therapy. As such, a detailed understanding of their mechanisms of action and inhibition is required and, consequently, methods that enable their investigation under in situ conditions.

The IAV M2 protein is a tetrameric pH-dependent proton channel inside the viral envelope, which is activated at pH values <7 after endocytic entry of the virion, a step required for release of genetic material into the host cell.^[7] Extensive studies have revealed that four pore-lining His37 and Trp41 residues near the center of the transmembrane (TM) helices are crucial in the gating and proton conductance mechanism, respectively.^[8] According to the accepted model, protonation of the His37 tetrad from the viral exterior side opens the channel via electrostatic repulsion between the imidazolium groups to enable proton flow into the virion.^[9-11] The chemical step of His37 protonation has been investigated in detail using various methods, including x-ray crystallography,^[12,13] solid-state nuclear magnetic reso-(ssNMR),^[14–18] and nance molecular dynamics simulations.^[19,20] However, quantifying the large-scale reorganization of the tetrameric structure associated with the widening of the pore radius has been much more challenging, in particular from an experimental perspective. Whereas structures of M2 show different relative alignments of the TM helices, their comparison is complicated as they refer to constructs of various lengths. As such, optical spectroscopic investigations that enable monitoring channel function in situ have become important, as exemplified by 2D infrared

[*] R. R. Paschke,⁺ Dr. J. Kozuch Physics Department, Freie Universität Berlin, Experimental Molecular Biophysics Arnimallee 14, 14195 Berlin (Germany) E-mail: jacek.kozuch@fu-berlin.de
R. R. Paschke,⁺ Dr. J. Kozuch Research Building SupraFAB, Freie Universität Berlin Altensteinstr. 23a, 14195 Berlin (Germany)
S. Mohr,⁺ Dr. S. Lange, Prof. Dr. A. Lange Research Unit Molecular Biophysics, Leibniz-Forschungsinstitut für Molekulare Pharmakologie (FMP)

Robert-Rössle-Straße 10, 13125 Berlin (Germany)

E-mail: alange@fmp-berlin.de

Prof. Dr. A. Lange

Institut für Biologie, Humboldt-Universität zu Berlin Invalidenstraße 42, 10115 Berlin (Germany)

- $\left[^{+}\right] \,$ These authors contributed equally to this work.
- [**] A previous version of this manuscript has been deposited on a preprint server (https://doi.org/10.1101/2023.06.22.546036).
- C © 2023 The Authors. Angewandte Chemie International Edition published by Wiley-VCH GmbH. This is an open access article under the terms of the Creative Commons Attribution Non-Commercial License, which permits use, distribution and reproduction in any medium, provided the original work is properly cited and is not used for commercial purposes.

Angew. Chem. Int. Ed. 2023, 62, e202309069 (1 of 7)

© 2023 The Authors. Angewandte Chemie International Edition published by Wiley-VCH GmbH

spectroscopic work that proposed twisting motions of the TM helices after pH activation.^[21]

To shed more light on the mechanics of pH-induced channel opening of M2, we performed surface-enhanced infrared absorption (SEIRA) spectroscopic measurements on M2 reconstituted within a solid-supported bilayer lipid membrane (ssBLM; Figure 1), as a minimal model system for the viral membrane.^[22,23] SEIRA spectroscopy allows monitoring protonation events^[24,25] as well as quantifying large-scale helical reorientation^[26,27] in situ upon pH activation. This is achieved using a nano-structured Au film that provides plasmonic infrared (IR) signal enhancement of the reconstituted proteins in a single ssBLM by a factor of about $10^{2,\ensuremath{^{[28,29]}}}$ In turn, the bulk solution is spectrally invisible due to a sharp distance-dependent attenuation of the enhancement after just 10 nm from the surface. Furthermore, the surface selection rule enables detecting molecular (re)orientation events, as only IR modes with dipole moment changes perpendicular to the surface are enhanced. Here, we quantify the SEIRA-based results by comparing them to density functional theory (DFT)-derived computational spectroscopic predictions that are informed by structural data from ssNMR spectroscopy.

The IAV M2₁₈₋₆₀ construct used in this work is composed of a short segment of the unstructured N-terminal domain (res. 18-21), the TM helix (res. 22-46) and a segment of the short amphipathic helix (res. 47-60).^[30] This construct was expressed in E. coli and reconstituted into 1,2-diphytanoylsn-glycero-3-phosphatidylcholine (DPhPC) liposomes according to Fu et al.^[31] We used DPhPC to ensure comparability to previous studies,^[32,33] and thus facilitate the interpretation of SEIRA data and the judgment of ssNMR spectral quality (see below). Note that previous studies compared M2 in various lipids, showing no relevant structural changes induced by the lipid environment.[32,33] The M2-proteoliposomes were deposited on a 6-mercaptohexanol-functionalized SEIRA Au surface to create a M2ssBLM system revealing characteristic bands at 1659 cm⁻¹ and 1547 cm^{-1} that show an isotopic downshift by -45 to -25 cm^{-1} upon global ${}^{13}\text{C}/{}^{15}\text{N}$ -labelling of the M2 protein. These signals can, therefore, be assigned to the amide I and



Figure 1. Schematic representation of the Influenza A virus (IAV) M2containing solid-supported bilayer lipid membrane (ssBLM) on a nanostructured SEIRA Au film. SEIRA probes IR spectral changes within approx. 10 nm, i.e., of a single ssBLM layer, and provides information on reorientation events due to the polarization direction (surfaceselection rule). amide II modes of the α -helical protein backbone structure (Figure 2A, top).^[34] The peaks at 1736 and below 1500 cm^{-1} on the other hand do not shift and can, thus, be ascribed to the lipids of the ssBLM.^[26,35] To assure that these amide I and II signals reflect a well-folded protein, we recorded ssNMR spectra on the same proteoliposomes and demonstrate the native fold of M2 with most of the residues in the TM domain and the amphipathic helix being well resolved in a ¹⁵N-¹³C fingerprint spectrum (Figure 2B). Assignments (see Figure 2C for a sequential backbone walk) are in line with previous studies^[33] and suggest a tetrameric form, with peak doubling occurring for res. 28 to 34, pointing to structural plasticity in this region. Residues at the Nterminal part (res. 18-23) as well as at the C-terminal end (res. 53-60) could not be detected in dipolar coupling-based experiments, probably due to higher flexibility in these domains. As shown in Figure S1, the IR spectral features of M2 do not change upon ssBLM formation demonstrating that the protein structure is maintained when interfaced at the SEIRA Au surface.

To gain information on the pH-dependent mechanism of the channel, SEIRA difference spectra were recorded after decreasing the bulk solution's pH from 8 to 4 (Figure 2A, bottom). Difference spectroscopy is a common tool in IR spectroscopy as it allows to focus on changes upon a stimulus, while ignoring the portion of the signal that remained unchanged. As such, negative and positive peaks (denoted as "-" or "+", respectively, in the following) are ascribed to species that have been lost or gained during the pH change, respectively. Accordingly, we note a double peak at 1743 and 1710 cm⁻¹ (overlapping with a shifting band at 1714 cm⁻¹, see below) and a broader band at 1460 cm⁻¹, which are not affected by the ¹³C/¹⁵N-isotope labelling of M2 and, therefore, can be assigned to C=O stretching of the lipid's ester group^[36] and CH_x deformations of the lipid's acyl chains, respectively (see Supporting Information section 2.1 for bulk IR spectra with similar features).^[37] As discussed in Supporting Information section 2.3, these lipid absorption peaks only appear in the presence of M2 in the ssBLM, and are therefore a result of pH-induced structural changes of M2. However, we also recognize prominent signals that shift upon ¹³C/¹⁵N-labelling and can thus be directly assigned to changes of M2. First, signals at 1714(+), 1575(-), and 1400(-) cm⁻¹ were detected (approx. 1660, 1544, and 1367 cm⁻¹, respectively, in ${}^{13}C/{}^{15}N$ -M2), which are clear markers for the protonation of carboxylate groups.^[38] The M2₁₈₋₆₀ construct contains multiple carboxylate groups, namely at Asp21, Asp24 or Asp44; a potential protonation of one of these residues was suggested previously to contribute to the channel conductivity.^[39,40] It is worth noting that the -54 cm^{-1} shift of the 1714 cm⁻¹ is a rather large isotopic shift of ¹³C-labelling of C=O groups. However, similar peak shifts of -50 cm^{-1} have been observed previously;^[41,42] in addition, exact estimation of the corresponding peak center positions is aggravated by the complex band shape. Second, a prominent couple of a negative peak at 1660(-) and a positive signal at 1544-(+) cm⁻¹ is observed (1618 and 1518 cm⁻¹ in ${}^{13}C/{}^{15}N-M2)$, which can be assigned to the amide I and II vibrations of the

Angew. Chem. Int. Ed. 2023, 62, e202309069 (2 of 7)



Figure 2. A: SEIRA spectra of IAV M2 within ssBLM (top) and difference spectra after changing the pH from 8 to 4 (bottom) for unlabeled M2 (solid lines) and $^{13}C/^{15}$ N-labelled M2 (dashed lines); isotope shifts are shown by arrows. Amide I and II bands are indicated in red and blue, respectively; red and blue arrows highlight the direction of the corresponding difference peaks. Baseline correction of SEIRA difference spectra is shown in Figure S2. B: 2D N–Ca correlation spectrum of $^{13}C/^{15}$ N–M2 in DPhPC proteoliposomes at pH 7.8. Assigned resonances are indicated in the spectra. In the N–Ca spectrum, carbon and nitrogen resonances of the assigned peaks correspond to the same residue. Doubled peaks are indicated in italics, with the second resonance marked with an asterisk (*). W41* could not be assigned unambiguously from our data, thus being in brackets. C: Sequential walk for the amino acid stretch L46 to L43 shown in a strip plot of 3D experiments. Spectra are shown in black (NCaC β), red (NCaCO), green (CaNCO) and purple (NCOCa), strips sharing the same Ca- or N-chemical shift are marked with dashed lines and the respective residue number, steps of the backbone walk are indicated with solid lines.

 α -helical backbone of M2. Intriguingly, this specific feature of amide I and II peaks with intensity of opposite sign is a clear IR spectroscopic marker of reorienting a-helices, as observed in our previous work on antimicrobial peptides.^[26] This spectroscopic feature relies on the approximately perpendicular orientation of the transition dipole moments (TDMs) of the amide I and II vibrations (mostly C=O stretch, and C-N stretch and N-H bend of the peptide groups, respectively; see below and Scheme in Figure 3B). In combination with the surface selection rule in SEIRA, the orientation of the TDMs results in a preferential enhancement of one of the two vibrational modes. In the present case, this effect provides us with direct insight into the reorientation of the α-helices of M2 during pH-activated channel opening. We would like to note that we do not detect the protonation of the pH-sensing His37 residues, since the imidazole sidechain generally does not provide a clear IR-spectroscopic marker.^[43]

To illustrate this orientation-dependent spectroscopic effect more clearly and demonstrate its usefulness for a quantitative analysis of the orientation of α -helices, we

computed the theoretical spectra of the protein backbone with density functional theory (DFT) using the approach previously reported in Forbrig et al.^[26] and related work by Kubelka et al.^[44] Accordingly, in a simple model (Figure 3B), we performed a geometry optimization and normal mode analysis of the TM helix (res. 22-46) and the amphipathic helix (res. 48-60) individually. We used several starting structures based on the ssNMR-determined dihedrals obtained in this work (Table S1), a previously reported ssNMR structure (pdb 2l0j),^[45] and a fully unconstrained model giving overall similar results as summarized in Figure S6. Based on the TDMs of the normal modes contributing to the amide I and II bands (Figure 3B), we can model the polarized spectra. Figure 3A (top) shows a representative resulting computational absolute spectrum of the amide I and II signals (set to FWHM of 12 cm^{-1}) by considering the average orientation of the TM helices of 25° with respect to the symmetry axis of M2 as reported in literature for the closed state (Table S2). As in our previous work, the computed amide I/II-spectrum matches the experimental one very well in intensity and position. Only





Figure 3. A: DFT-computed spectrum (FWHM set to 12 cm^{-1}) of the M2 protein backbone (black, top) and difference spectrum for a change in helix tilt angle from 25° to 35° (black, bottom)—see inset. For comparison, grey lines indicate the related experimental spectra; amide I and II bands are highlighted in red and blue, respectively. Spectra were normalized to the amide I peak for clarity. B: The TDM vectors of amide I (red) and amide II (blue) of a TM α -helix in M2 are aligned preferentially along or perpendicular to the helix axis (PDB: 2l0j), respectively. The preferential orientation originates from coupling between the single amide units, whose amide I and II TDM vectors are approximately perpendicular to each other. C: Trend of the amide I/II ratio of the peak areas dependent on helix tilt angles from 0° to 90° in respect to the membrane surface normal. Confidence intervals are obtained from different starting structures in the DFT calculation (Figure S6).

the amide II shows a slight shift by approximately 10 cm^{-1} , which is acceptable considering the level of theory used in this work (bp86/6-31g*).^[26,44,46,47] Modelling a rigid body reorientation of the TM helix, we determined the computational difference spectra considering the polarization direction according to the surface selection rule in SEIRA (Figure 3A, bottom; experimental data are shown in light colors for comparison). We see a striking similarity in the spectral features of the amide I peak at 1659(-) cm⁻¹ and the amide II region at 1533(+) to $1492(+) \text{ cm}^{-1}$ between theoretical and experimental SEIRA difference spectra, providing the direct proof-of-concept for the experimental detection of a-helical reorientation upon pH-activated M2 channel opening. It is important to mention that a reorientation of the amphipathic helix (Figure S5D) shows a distinctly different spectral pattern, such that this approach suggests that the reorientation is localized to the TM region.

Having demonstrated the α -helical reorientation qualitatively, we now have a tool to monitor the orientation of the TM helices of M2 via changes in the amide I and II intensities. More specifically, we can use the amide I/II ratio as a measure for the quantification of the absolute orientation. We calculated the DFT-based amide I/II intensity ratio based on absolute spectra (Figure S5A, B) for TM helix tilt angles from 0° to 90° with respect to the surface normal (Figure 3C). To estimate the accuracy of the "amide I/II ratio vs. angle" calibration, we show the average and standard deviation over all structural models shown in Figure S6. We see that for an ideal vertical orientation of the TM helix in the M2 channel, a ratio of approx. 6 is expected, while a horizontal orientation would provide a ratio of around 0.7. Previously, we have shown that ideal vertical orientation of a helix can provide amide I/II ratios of > 15.^[22,26] In the present case, the amphipathic helix is oriented horizontally, where a amide I/II ratio of around 1 is expected,^[27] lowering this maximum achievable value.

Turning back to the experimental results, we now aim to quantify the pH-dependent reorientation of M2's TM helices during channel opening at pH <7. To monitor the opening of the M2 channels within the ssBLM, we performed a pH titration from 8 to 4 (Figure 4A, top), which shows a pHdependent increase of the spectral features observed above. As shown in Figure 4A (middle), this trend matches very well the DFT-based spectra following a reorientation in a range of 26 to 40°. Inspecting the changes of amide I and II, we notice that the absorption intensities respond to two pK_A values, one at 7 ± 0.5 and one at ≤5.5 based on a fit of two coupled sigmoidal curves (Figure 4B; Figure S5C includes fits of the markers for the carboxylate protonation). Previous work has suggested that in the closed state of M2 at least one His residue of the His37-tetrad is protonated, and further protonation occurs with pKA values of 7.1 and 5.4, respectively.^[18] Despite not detecting the direct IR fingerprint of the imidazole residues, we can observe the same protonation transitions as part of the amide I and II changes, and are, therefore, consistent with previous work. Our data further suggests that the more acidic pK_A does not originate (solely) from the protonation of the His37-tetrad but is coupled to aspartic acids (Figure S5C). The pHdependent amide I/II ratio (Figure 4C) obtained from the integrated absolute peak areas changes from 4.1 ± 0.2 to 2.5 ± 0.2 in the explored pH range. Using the calibration from computational spectra (Figure 3C), we are now able to translate this data to a pH-dependent TM helix tilt angle (Figure 4D). We obtain an angle of $26^{\circ} \pm 2^{\circ}$ at pH 8, which upon acidification results in a helix tilt of up to 40° and, therefore, an increase in angle by 14°. This result is consistent with literature where the average TM helix angles of the solved structures is 25° (see Table S2) and a $10-12^{\circ}$ difference for an opened and a closed state of the proton channel has been suggested previously.^[12,13] Interestingly, our data further suggests that this mechanical opening is guided by the more acidic pK_A value, as a plateau region is observed in the pH region of 8 to 7. It is important to note that the previously suggested pH-induced twist of the helices is not contradicted by our results and may be part of the mechanics of channel opening.^[21] However, our data suggests that such a helical twist would not exceed approx. 30° (Figure S6C).

Finally, having determined the IR spectral features of channel opening, we repeated the SEIRA pH titration in the presence of the known M2 inhibitor rimantadine (Figure 4A, bottom). Interestingly, while still observing the protonation of aspartates, the opposing changes in the amide I and II bands are notably absent. Plotting the amide I/II ratio and the associated helical tilt angles over pH (Figure 4C,D), we can determine similar angles in the closed

Communications



Figure 4. A: Experimental difference spectra of the pH titration of M2 in a ssBLM (top) and computational difference spectra for a stepwise TM helix reorientation from 26° to 40° (middle); experimental pH titration after adding the inhibitor rimantadine (bottom). Baseline correction of SEIRA difference spectra is shown in Figure S2. B: Experimental changes in the pH-dependent absorbance of amide I at 1659 cm⁻¹ (red) and the amide II at 1544 cm⁻¹ (blue). The fitted lines and pK_A values were obtained from global double sigmoidal fits considering pH-dependent spectral changes with two pK_A values; pK_{A2} was determined as 7.0 ± 0.5 ; pK_{A2} is indicated as ≤ 5.5 since the plateau at acidic conditions was not fully resolved. C: Experimental amide I/II ratio without inhibitor (black) and after adding rimantadine (grey triangles) over pH. The fitted lines result from intensity changes in B. D: Average TM α -helix tilt angles of M2 without inhibitor (black dots) and after adding rimantadine (grey triangles) depending on pH, calculated considering the computational calibration curve in Figure 3C. E: Schematic representation of the structure of M2 before (PDB: 6bkl) and after (PDB: 6boc) pH-dependent channel opening according to SEIRA spectra. Error bars in B and C were obtained from individual experiments; error bars in D result from error propagation (see Supporting Information 2.4).

state of around $26^{\circ}\pm 2^{\circ}$, but the large changes of the tilt angle of the TM helix at lower pH are absent. This provides direct evidence for a model, in which (part) of the inhibition mechanism involves a blockage of the dynamical changes of helix alignment.

To conclude, with the presented approach we have a tool to track in situ the large-scale structural changes associated with M2's pH-dependent channel activation, an essential process in IAV infection. Combining the surfaceselection rule inherent to SEIRA spectroscopy and ssNMRinformed computational spectroscopy, we quantify these changes as a reorientation of the channel's transmembrane helices by 14° (Figure 4E). However, this mechanical opening is inhibited upon addition of the antiviral drug rimantadine. As such, our data confirms the plethora of structural data that proposed that "fanned out" structures resolved an opened conformation of the channel, which we observed here under operating conditions. In the future, we aim to utilize this approach to enable a combined structural and functional analysis of viroporins of current relevance and contribute to the discovery of new inhibitors. In particular, combining results from SEIRA spectroscopy and ssNMR, which visualize complementary structural changes, can be of high benefit towards an extensive understanding of the action of viral proteins and finding of novel antiviral drugging strategies.

Supporting Information

Materials and methods (sample preparation, SEIRA spectroscopy, ssNMR spectroscopy, DFT calculations), supplementary discussions (unpolarized ATR-IR spectra, baselining, changes in lipid bands, error estimation) and supplementary figures and tables. The authors have cited additional references within the Supporting Information.^[48–58]

Acknowledgements

R.R.P. and J.K. thank Kenichi Ataka (Freie Universität Berlin, Germany) for helpful discussions and Joachim Heberle (Freie Universität Berlin, Germany) for the continuous support. We thank James Chou (Harvard Medical School, Boston, MA, USA) for generously providing the expression plasmid coding for the IAV M218-60 construct and the HPC Service of ZEDAT, Freie Universität Berlin, for computing time (10.17169/refubium-26754). The TOC Figure was created with the help of BioRender.com. This work was supported by DFG via the Sonderforschungsbereich 1078 "Protonation Dynamics in Protein Function" (project number 221545957, projects B09 and B10; J.K. and A.L., respectively). Open Access funding enabled and organized by Projekt DEAL.





Conflict of Interest

The authors declare no conflict of interest.

Data Availability Statement

The ssNMR chemical shifts are deposited in the BMRB under the ID 52099. The DFT results and analysis tool are deposited together with the main SEIRA spectroscopic dataset under github.com/jacekkozuch/Helix_Reoriention_M2.git. Further data reported in this work is available from the corresponding authors upon reasonable request.

Keywords: Bilayer Lipid Membrane • Influenza a Virus M2 • Proton Channels • SEIRA Spectroscopy • Viroporins

- [1] C. Scott, S. Griffin, J. Gen. Virol. 2015, 96, 2000-2027.
- [2] U. Breitinger, N. S. Farag, H. Sticht, H. G. Breitinger, Int. J. Biochem. Cell Biol. 2022, 145, 106185.
- [3] P. P. Singh Tomar, R. Oren, M. Krugliak, I. T. Arkin, *Viruses* 2019, 11, 632.
- [4] R. M. Pielak, J. J. Chou, Biochem. Biophys. Res. Commun. 2010, 401, 58–63.
- [5] L. H. Pinto, L. J. Holsinger, R. A. Lamb, Cell 1992, 69, 517– 528.
- [6] A. J. Hay, A. J. Wolstenholme, J. J. Skehel, M. H. Smith, EMBO J. 1985, 4, 3021–3024.
- [7] A. Helenius, Cell 1992, 69, 577–578.
- [8] E. Aledavood, B. Selmi, C. Estarellas, M. Masetti, F. J. Luque, Front. Mol. Biosci. 2022, 8, 796229.
- [9] M. Hong, W. F. DeGrado, Protein Sci. 2012, 21, 1620–1633.
- [10] R. M. Pielak, J. J. Chou, Biochim. Biophys. Acta Biomembr. 2011, 1808, 522–529.
- [11] J. Wang, J. X. Qiu, C. Soto, W. F. DeGrado, Curr. Opin. Struct. Biol. 2011, 21, 68–80.
- [12] J. L. Thomaston, N. F. Polizzi, A. Konstantinidi, J. Wang, A. Kolocouris, W. F. DeGrado, J. Am. Chem. Soc. 2018, 140, 15219–15226.
- [13] J. L. Thomaston, Y. Wu, N. Polizzi, L. Liu, J. Wang, W. F. DeGrado, J. Am. Chem. Soc. 2019, 141, 11481–11488.
- [14] J. Hu, R. Fu, K. Nishimura, L. Zhang, H.-X. Zhou, D. D. Busath, V. Vijayvergiya, T. A. Cross, *Proc. Natl. Acad. Sci.* USA 2006, 103, 6865–6870.
- [15] F. Hu, K. Schmidt-Rohr, M. Hong, J. Am. Chem. Soc. 2012, 134, 3703–3713.
- [16] M. Colvin, L. Andreas, J. Chou, R. Griffin, *Biochemistry* 2014, 53, 5987–5994.
- [17] Y. Miao, R. Fu, H.-X. Zhou, T. A. Cross, Structure 2015, 23, 2300–2308.
- [18] S. Y. Liao, Y. Yang, D. Tietze, M. Hong, J. Am. Chem. Soc. 2015, 137, 6067–6077.
- [19] H. Torabifard, A. Panahi, C. Brooks, Proc. Natl. Acad. Sci. USA 2020, 117, 3583–3591.
- [20] W. Chen, Y. Huang, J. Shen, J. Phys. Chem. Lett. 2016, 7, 3961–3966.
- [21] J. Manor, P. Mukherjee, Y. S. Lin, H. Leonov, J. L. Skinner, M. T. Zanni, I. T. Arkin, *Structure* **2009**, *17*, 247–254.
- [22] J. Kozuch, C. Steinem, P. Hildebrandt, D. Millo, Angew. Chem. Int. Ed. 2012, 51, 8114–8117.
- [23] J. Kozuch, K. Ataka, J. Heberle, Nat. Rev. Methods Primers 2023, 3, 70.

[24] K. Ataka, J. Drauschke, V. Stulberg, B. Koksch, J. Heberle, Biochim. Biophys. Acta Biomembr. 2022, 1864, 183873.

- [25] M. A. Fallah, C. Stanglmair, C. Pacholski, K. Hauser, *Lang-muir* 2016, 32, 7356–7364.
- [26] E. Forbrig, J. K. J. K. Staffa, J. Salewski, M. A. Mroginski, P. Hildebrandt, J. Kozuch, *Langmuir* 2018, 34, 2373–2385.
- [27] A. de Miguel Catalina, E. Forbrig, J. Kozuch, C. Nehls, L. Paulowski, T. Gutsmann, P. Hildebrandt, M. A. Mroginski, *Biochemistry* 2019, 58, 2447–2462.
- [28] M. Osawa, in *Handbook of Vibrational Spectroscopy, Vol. 1* (Eds.: J. M. Chalmers, P. R. Griffiths), John Wiley & Sons, Ltd, Chichester 2002, pp. 785–799.
- [29] K. Ataka, S. T. Stripp, J. Heberle, Biochim. Biophys. Acta 2013, 1828, 2283–2293.
- [30] J. R. Schnell, J. J. Chou, Nature 2008, 451, 591–595.
- [31] Q. Fu, A. Piai, W. Chen, K. Xia, J. J. Chou, Nat. Protoc. 2019, 14, 2483–2520.
- [32] L. B. Andreas, M. T. Eddy, R. M. Pielak, J. Chou, R. G. Griffin, J. Am. Chem. Soc. 2010, 132, 10958–10960.
- [33] L. B. Andreas, M. Reese, M. T. Eddy, V. Gelev, Q. Z. Ni, E. A. Miller, L. Emsley, G. Pintacuda, J. J. Chou, R. G. Griffin, *J. Am. Chem. Soc.* **2015**, *137*, 14877–14886.
- [34] P. I. Haris, G. T. Robillard, A. A. van Dijk, D. Chapman, G. T. Robillard, A. A. van Dijk, *Biochemistry* 1992, *31*, 6279–6284.
- [35] S. Wiebalck, J. Kozuch, E. Forbrig, C. C. Tzschucke, L. J. C. C. Jeuken, P. Hildebrandt, J. Phys. Chem. B 2016, 120, 2249–2256.
- [36] D. R. Gauger, H. Binder, A. Vogel, C. Selle, W. Pohle, J. Mol. Struct. 2002, 614, 211–220.
- [37] N. B. Colthup, L. H. Daly, S. E. Wiberley, in *Introduction to Infrared and Raman Spectroscopy*, 3rd Ed. (Eds.: N. B. Colthup, L. H. Daly, S. E. Wiberley), Academic Press, San Diego 1990, pp. 215–233.
- [38] A. Barth, C. Zscherp, Q. Rev. Biophys. 2002, 35, 369–430.
- [39] C. Ma, G. Fiorin, V. Carnevale, J. Wang, R. A. Lamb, M. L. Klein, Y. Wu, L. H. Pinto, W. F. DeGrado, *Structure* 2013, 21, 2033–2041.
- [40] B.-S. Jeong, R. B. Dyer, J. Am. Chem. Soc. 2017, 139, 6621– 6628.
- [41] J. Kozuch, S. H. Schneider, S. G. Boxer, ACS Chem. Biol. 2020, 15, 1148–1153.
- [42] S. H. Schneider, J. Kozuch, S. G. Boxer, ACS Cent. Sci. 2021, 7, 1996–2008.
- [43] A. Barth, Biochim. Biophys. Acta 2007, 1767, 1073-1101.
- [44] J. Kubelka, R. A. Gangani, D. Silva, T. A. Keiderling, J. Am. Chem. Soc. 2002, 124, 5325–5332.
- [45] M. Sharma, M. Yi, H. Dong, H. Qin, E. Peterson, D. D. Busath, H.-X. Zhou, T. A. Cross, *Science* **2010**, *330*, 509–512.
- [46] A. D. Becke, Phys. Rev. A 1988, 38, 3098–3100.
- [47] J. P. Perdew, Phys. Rev. B 1986, 33, 8822-8824.
- [48] H. Miyake, S. Ye, M. Osawa, *Electrochem. Commun.* 2002, 4, 973–977.
- [49] J. Hoffmann, J. Ruta, C. Shi, K. Hendriks, V. Chevelkov, W. T. Franks, H. Oschkinat, K. Giller, S. Becker, A. Lange, *Magn. Reson. Chem.* 2020, 58, 445–465.
- [50] H. Hu, K. Ataka, A. Menny, Z. Fourati, L. Sauguet, P. J. Corringer, P. Koehl, J. Heberle, M. Delarue, *Proc. Natl. Acad. Sci. USA* 2018, *115*, E12172–E12181.
- [51] R. H. Fogh, W. Boucher, W. F. Vranken, A. Pajon, T. J. Stevens, T. N. Bhat, J. Westbrook, J. M. C. Ionides, E. D. Laue, *Bioinformatics* 2005, 21, 1678–1684.
- [52] S. P. Skinner, R. H. Fogh, W. Boucher, T. J. Ragan, L. G. Mureddu, G. W. Vuister, J. Biomol. NMR 2016, 66, 111–124.
- [53] Y. Shen, F. Delaglio, G. Cornilescu, A. Bax, J. Biomol. NMR 2009, 44, 213–223.
- [54] M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, B. Mennucci, G. A. Petersson, H. Nakatsuji, M. Caricato, X. Li,

Angew. Chem. Int. Ed. 2023, 62, e202309069 (6 of 7)

© 2023 The Authors. Angewandte Chemie International Edition published by Wiley-VCH GmbH



H. P. Hratchian, A. F. Izmaylov, J. Bloino, G. Zheng, J. L. Sonnenberg, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, J. Montgomery, J. A. J. E. Peralta, F. Ogliaro, M. Bearpark, J. J. Heyd, E. Brothers, K. N. Kudin, V. N. Staroverov, R. Kobayashi, J. Normand, K. Raghavachari, A. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, N. Rega, J. M. Millam, M. Klene, J. E. Knox, J. B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, R. L. Martin, K. Morokuma, V. G. Zakrzewski, G. A. Voth, P. Salvador, J. J. Dannenberg, S. Dapprich, A. D. Daniels, Ö. Farkas, J. B. Foresman, J. V. Ortiz, J. Cioslowski, D. J. Fox, **2013**.

- [55] O. Gutiérrez-Sanz, E. Forbrig, A. P. Batista, M. M. Pereira, J. Salewski, M. A. Mroginski, R. Götz, A. L. De Lacey, J. Kozuch, I. Zebger, *Langmuir* 2018, *34*, 5703–5711.
- [56] B. M. Fung, A. K. Khitrin, K. Ermolaev, J. Magn. Reson. 2000, 142, 97–101.
- [57] A. Böckmann, C. Gardiennet, R. Verel, A. Hunkeler, A. Loquet, G. Pintacuda, L. Emsley, B. H. Meier, A. Lesage, J. Biomol. NMR 2009, 45, 319–327.
- [58] L. K. Tamm, S. A. Tatulian, *Q. Rev. Biophys.* **1997**, *30*, 365–429.

Manuscript received: June 27, 2023

Accepted manuscript online: September 21, 2023 Version of record online: October 17, 2023