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Interaction of Water with Atomic Layer Deposited Titanium Dioxide on p-Si Photocathode: Modeling of Photoelectrochemical Interfaces in Ultrahigh Vacuum with Cryo-Photoelectron Spectroscopy

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XPS Si2p

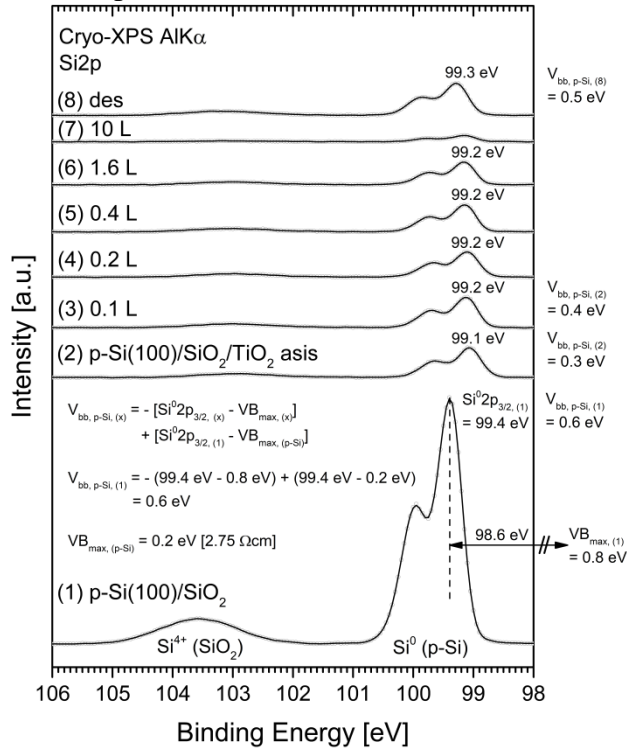


Figure S1. Cryo-XPS AlK α Si2p spectra of the p-Si(100)/SiO $_2$ interface prior to the atomic layer deposition of TiO $_2$ as well as of the subsequential water adsorption on p-Si/SiO $_2$ /TiO $_2$, followed by water desorption. Before TiO $_2$ deposition, the Si2p $_{3/2}$ peak of the p-Si is settled at 99.4, while the emission of Si $^{4+}$ (SiO $_2$) is found at 103.6 eV. The oxidation state Si $^{4+}$ states, that the SiO $_2$ layer is almost completely SiO $_2$. After the atomic layer deposition of TiO $_2$, a shift of 0.3 eV towards lower binding energies is observed. Upon water adsorption, in contrary to the TiO $_2$ core level in Fig. 5, a shift of 0.1 eV to higher binding energies is observed. With the equation given in the figure, the band bending $V_{bb, p-Si, (x)}$ in the p-Si at the p-Si/SiO $_2$ interface can be calculated for each experimental step x. $VB_{max, (x)}$ indicates the energy cut-off of the valence band maximum at the experimental step x, which was measured but is not shown here. For step 0 (pristine p-Si) $VB_{max, (0)} = 0.2$ eV was derived from the doping concentration and the resulting specific resistance, which is known.

UPS HeII

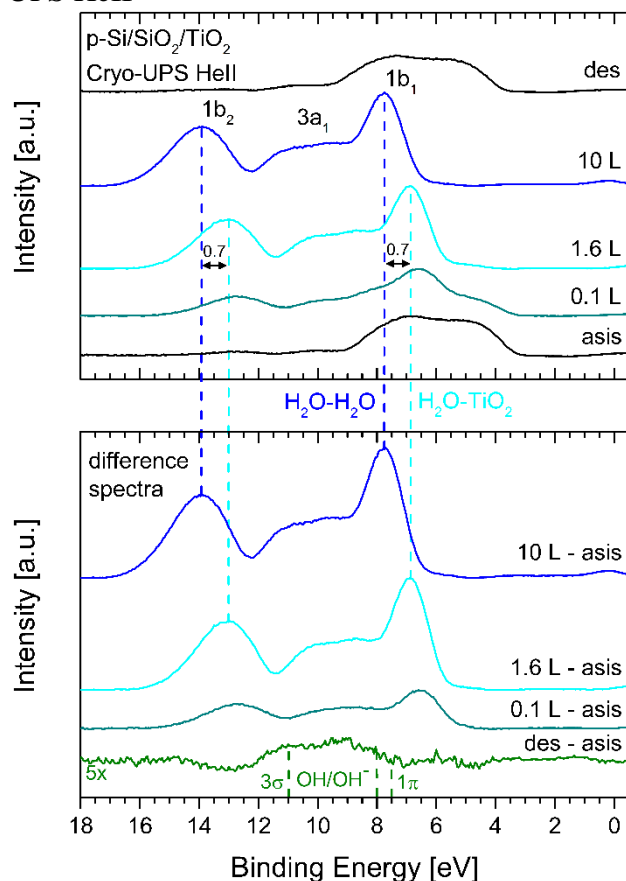


Figure S2. UPS HeII spectra and respective difference spectra for the water adsorption experiment on TiO₂. The binding energy difference between the H₂O-TiO₂ species and the H₂O-H₂O species is 0.7 eV, just as for the UPS HeI spectra in Figure 6. After desorption, chemisorbed OH and OH⁻ species can be detected on the surface to a very small extent. The presence of gap states in the HeII spectra could not be evaluated, since the HeII satellites directly interfere with the band gap region and satellite subtraction is more sophisticated than for HeI.