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Original Article

# Gas Chromatography-Mass Spectrometry (GC-MS) Analysis of Indole Alkaloids Isolated from *Catharanthus roseus* (L.) G. Don Cultivated Conventionally and Derived from *In vitro* Cultures

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#### **Abstract**

Catharanthus roseus (periwinkle) is a medicinal plant commonly known for its wide biological activity. In many countries different parts of this plant are used for the treatment of diabetes, hypertension and for menstrual regulation. Due to the ability of production of alkaloids, which can be applied in cancer therapy, is still extensively investigated. Two, the most valuable alkaloids (vincristine and vinblastine) are present in *C. roseus* in very low concentrations. Micropropagation is promising technique used to enhance the level of important secondary metabolites. The main objective of present study was alkaloids extraction from plants cultivated conventionally and derived from *in vitro* cultures. In this order the aerial parts of periwinkle were extracted with 96% ethanol at room temperature (method I) and heated with 96% ethanol at 55 °C for 90 minutes (method II). The obtained mixtures of different indole alkaloids were analyzed by gas chromatography – mass spectrometry (GC-MS). Analysis revealed the presence of 15 alkaloids, among which vindoline, vindorosine, isovindolinine and ajmalicine were the most abundant. The obtained results indicated that the propagation method had a significant effect on the percentage content of alkaloids in *C. roseus* herb. Plants derived from *in vitro* cultures were richer in vindorosine and vindoline, while conventionally cultivated – in tetrahydroalstonine and ajmalicine. Moreover, in case of isovindolinine, vindolinine and ajmalicine, extraction at 55 °C was more effective, while for pericyclivine – maceration at room temperature. Interestingly, the pericyclivine was not detected in the mixture of alkaloids obtained from periwinkle herb by the extraction at 55 °C.

Keywords: bioactivity, extraction method, Madagascar periwinkle, Vinca alkaloids, tissue cultures

#### Introduction

Catharanthus roseus (L.) G. Don, known as Vinca rosea or Madagascar periwinkle, can be found as an ornamental subshrub in gardens throughout the world (Larbie and Abboah-Offei, 2014). In addition to its pharmaceutical value, periwinkle has long been cultivated as a herbal medicine. It has been used for centuries to treat diabetes, high blood pressure, fever, malaria and chest complaints (Aslam et al., 2010; Gajalakshmi et al., 2013; Kumari and Gupta, 2013). In India, juice from the periwinkle flowers has been used to treat skin problems, dermatitis, eczema and acne, while juice from the leaves has been applied for wasp stings (El-Sayed and Cordell, 1981). The antibacterial, antifungal, antiviral and antioxidant

properties of this herb have also been well documented (Carev and Patterson, 1970; Garg, 2010).

C. roseus is also a rich source of monomeric and dimeric indole alkaloids, which possess antihypertensive and anticancer activities (Kulkarni et al., 1999; Mishra et al., 2001). Anticancer drugs, vincristine and vinblastine, are produced in vivo by the condensation of vindoline and catharanthine, two monomeric precursors (Noble, 1990; Laflamme et al., 2001). These two dimeric alkaloids may be used as single agents or in combination therapy for the treatment of acute leukemia and Hodgkin's disease as well as for a wide variety of neoplasms including breast, bladder and lung cancers (Noble, 1990; Ramirez et al., 1997; Tabakovic et al., 1997). Other alkaloids, such as ajmalicine and serpentine are used as antihypertensive and anti-

neuro-inflammatory agents, while vindoline, tetrahydroalstonine, catharanthine and vindolinine exhibit hypoglycemic activity (Svoboda, 1969; Marles and Farnsworth, 1995; Almagro *et al.*, 2015).

The alkaloid content varies considerably in the individual parts of *C. roseus*. In the leaves ranges from 0.32 to 1.16%, in the stems from 0.074 to 0.48%, in the flowers from 0.005 to 0.84%, while in roots – between 0.14 and 1.34% (Joy *et al.*, 1998).

Periwinkle is still intensively studied because of its valuable properties. Researchers are trying to increase the synthesis of alkaloids through the technique of tissue cultures. Also analytical methods for the rapid identification and quantitative extraction of alkaloids are continuously improved.

The aim of this research was to compare the alkaloid composition of *Catharanthus roseus* cultivated conventionally and derived from *in vitro* cultures using GC-MS method. Moreover, two different extraction methods were applied in order to show the influence of temperature on identified alkaloids.

#### Materials and Methods

Plant material

The research material consisted of herb (leafy flowered stems) of periwinkle (*Catharanthus roseus* (L.) G. Don 'Mediterranean Lilac' (Thompson & Morgan, England).

The field experiment was carried out at the experimental station which belongs to the Department of Horticulture of the West Pomeranian University of Technology Szczecin (North-Western Poland). The experiment was established in randomized blocks with four replications. The experimental plot area was 2.75 m² (40 plants per plot). The field was prepared according to agrotechnique proper for periwinkle cultivation (Thomas *et al.*, 2012). Mineral fertilization, in the form of NPK, was applied during the field preparation, in amounts: 20:30:30 kg·ha ¹, according to Khode *et al.* (2000).

There were two methods of plant propagation compared in this study: generative (conventional method of cultivation from seedlings) and *in vitro* propagation.

In the conventional method, the seedlings of periwinkle were produced in the greenhouse. Seeds were sown in trays with peat moss, on  $26^{th}$  February. The trays were kept in the greenhouse at the temperature  $22\pm2$  °C and relative humidity  $60\pm5\%$ . Seedlings were transferred to multiplates containing peat moss on  $24^{th}$  April and then transplanted into the open field on  $28^{th}$  June, at a density of  $25\times20$  cm.

In the *in vitro* propagation seeds of *Catharanthus roseus* were immersed in 70% ethanol for 30 seconds followed by surface disinfection in 7.5% NaClO for 15 minutes. Seeds were transferred onto initiation media containing macro- and microelements according to MS medium (Murashige and Skoog, 1962). Seedlings, initiated for growth, were proliferated three times on the MS media with 2.0 mg·dm<sup>-3</sup> KIN (kinetin) and 1.0 mg·dm<sup>-3</sup> IBA (indole-3-butyric acid). Propagated shoots of length 1cm from previous stage were transferred onto rooting media – ½ MS (half-strength concentrations of the major and minor salts of the MS medium) with 0.2 mg·dm<sup>-3</sup> NAA (α-naphtahleneacetic acid). At all stages of experiments the media were supplemented with 8 g·dm<sup>-3</sup> agar and 30 g·dm<sup>-3</sup> sucrose and 100 mg·dm<sup>-3</sup> inositol and their pH was adjusted to 5.7 using the solutions: 0.1 M NaOH and HCl. After adding growth regulators, the media were autoclaved for 20 minutes at the temperature 121 °C. Every stage

lasted 6 weeks. The cultures were placed in a growth chamber at the temperature  $24\pm1$  °C and then for month in a greenhouse at the temperature  $22\pm2$  °C.

After the acclimatization, plants were transplanted into the open field on  $28^{th}$  June, at a density of 25x20 cm.

During the growing season the crop management was carried out. It included mainly irrigation, weeding and soil cultivation.

The plant harvest was done at full-bloom stage, on  $29^{\text{th}}$  July. For the laboratory analyses 100 g samples (periwinkle herb), from each experimental plot  $(4 \times 100 \text{ g})$ , were taken. The obtained plant material was dried in a through-flow laboratory dryer set at 35 °C.

#### Chemicals

All solvents used in the experiments were of analytical grade. Dichloromethane, ethyl acetate, hexane, 96% ethanol, hydrochloric acid (35-35%), sulphuric acid (95%) and ammonia solution (25%) were purchased from Chempur (Piekary Śląskie, Poland) and used as received. Alkaloids standards, vindoline (purity 98%), ajmalicine (purity 98%) and tetrahydroalstonine (Aldrich CPR) were purchased from Sigma-Aldrich, Poznań, Poland. The n-alkane (C7-C40) standard solution (1000  $\mu g \ mL^{-1}$  each component in hexane) was purchased from Supelco, Poznań, Poland.

## Extraction of monoterpene indole alkaloids

Monoterpene indole alkaloids were extracted according to the methodology described by Misra *et al.* (2009) (method I) and Miranda-Ham *et al.* (2007) (method II), with small modifications.

*Method I:* dried, milled herb of periwinkle (2.5 g) was extracted with 96% ethanol (25 mL) through maceration (overnight) and filtered. The residue was again extracted with ethanol and pooled ethanolic extract was evaporated to the dryness. Green, dried residue was redissolved in ethanol (25 mL), diluted with distilled water (25 mL) and acidified with 3% hydrochloric acid to pH 1.

This mixture was extracted three times with hexane (3 x 25 mL). Hexane fraction was discarded. The remaining aqueous layer was cooled to 10 °C, basified with 25% ammonia solution to pH 10 and extracted with dichloromethane (3 × 25 mL). The combined dichloromethane extract was washed with 25 mL of brine, dried over anhydrous sodium sulphate and evaporated to dryness to give 0.04 g; 1.6% (conventional cultivation) and 0.05 g; 2% (*in vitro* propagation), brownish-yellow alkaloid mixture, respectively.

**Method II:** 2.5 g of dried, milled herb of periwinkle in 100 mL round bottomed flask containing 25 mL of 96% ethanol was heated at 55 °C for 90 minutes using magnetic stirrer. After filtration, the ethanolic extract was evaporated to the dryness and the residue was dissolved in 104 mL 2.5% sulphuric acid. The suspension was then extracted with ethyl acetate (2 x 25 mL), keeping the aqueous phase each time. After increasing pH of the aqueous phase to 10 with 25% ammonia solution, alkaloids were extracted with three equal volumes of ethyl acetate. The organic phase was washed with brine, dried over anhydrous sodium sulphate and evaporated to dryness to give 0.02g; 0.8% (conventional cultivation) and 0.03g; 1.2% (*in vitro* propagation), brownish-yellow alkaloid mixture, respectively.

The alkaloid mixtures obtained were dissolved in dichloromethane (1.5 mL) and subjected to GC/MS analysis.

Gas Chromatography-Mass Spectrometry (GC-MS) analysis of the alkaloid mixture

The qualitative GC-MS analysis of the obtained alkaloid mixture was carried out using an HP 6890 gas chromatograph, equipped with a HP-5MS capillary column (30 m  $\times$  0.25 mm; film thickness 0.25  $\mu m$ ) and coupled with HP 5973 Mass Selective Detector.

The flow rate of carrier gas (He) was  $1mL^{-1}$ . Samples of  $4\,\mu L$  were injected in the split mode at a ratio of 5:1. The injector and the transfer line were kept at 280 °C. The ion source temperature was 230 °C.

The initial temperature of the column was 40 °C for 5 min, then increased to 60 °C at a rate of 30 °C min<sup>-1</sup>, next to 230 °C at a rate of 6 °C min<sup>-1</sup> (kept constant for 10 min), and then increased to a final temperature of 280 °C at a rate of 30 °C min<sup>-1</sup>. The oven was held at this temperature for 30 min. Mass spectra were taken at 70 eV. Mass range was from 40 to 550 m/z. Solvent delay was 6 min. The total running time for a sample was about 76 min.

The identification of the alkaloids was confirmed by comparing the mass spectral data with those of authentic compounds and with data obtained from the literature (Djerassi *et al.*, 1962; Gilbert, 1965; Kohl *et al.*, 1981; Zocoler *et al.*, 2005; Zhao *et al.*, 2013; Akhgari *et al.*, 2015).

Further identification was made by comparison of their mass spectra with those stored in Wiley NBS75K, NIST 2005 and NIST 2008 mass spectral libraries (isovindolinine, vindolinine, ajmalicine, tetrahydroalstonine, pleiocarpamine).

Retention indices (RI) values were measured on HP-5MS column for all identified alkaloids. For RI calculation, a mixture of n-alkanes ( $C_7$ - $C_{40}$ ) was used, under the same chromatographic conditions which were used for the analysis of ethanolic extracts.

The relative percentage of each component was evaluated from the total peak area (TIC) by the apparatus software.

#### Statistical analysis

The results of the study were subjected to an analysis of variance which was performed with AWAR software, made by Department of Applied Informatics, Institute of Soil Science and Plant Cultivation in Puławy, Poland. The three experimental factors (I – alkaloid, II – propagation method, III – extraction method) were tested in a split-split-plot design and the means were separated using the Tukey's Studentized Range Test at P = 0.05

### **Results and Discussion**

In the current research the bioactive compounds of *C. roseus* extracts were analyzed by gas chromatography-mass spectrometry (GC-MS). A total of 15 different alkaloids have been identified (Table 1), among which vindoline (28.695-34.255%), vindorosine (10.665-18.890%) and ajmalicine (6.785-12.660%) were the most abundant. Some alkaloids remained unidentified due to the lack of reference substances as well as the lack of mass spectra libraries. According to Hisiger and Jolicoeur (2007), about eight indole alkaloids are commercially available and only few (about 11) of the large group of these valuable compounds produced by periwinkle are frequently analyzed. Moreover, GC is unsuitable to the analysis of bisindole alkaloids as they have high melting point (Chen *et al.*, 2013).

GC-MS chromatograms of ethanol extracts of *C. roseus* are presented in Fig. 1 and Fig. 2, while Fig. 3 shows the chemical structures of identified alkaloids.

Vindoline (30.865%) was the main alkaloid of periwinkle herb (Table 1). Vindorosine (14.746%) was the second one. High concentrations were also noted for ajmalicine (9.866%), isovindolinine (9.239%), vindolinine (8.209%) and tetrahydroalstonine (7.504%). The amounts of the other alkaloids ranged between 2.206 and 0.218%.

There was also a significant interaction recorded between alkaloid and propagation method. Significant differences were noted in case of the four of alkaloids. Higher amounts of vindorosine and vindoline were detected according to the *in vitro* plant propagation, while for tetrahydroalstonine and ajmalicine – conventional method of propagation. Moreover, an interaction between alkaloid and extraction method was also statistically significant for some of the compounds. In the case of isovindolinine, vindolinine and ajmalicine, extraction at 55 °C was more effective, while for pericyclivine – maceration at room temperature. When using extraction at 55 °C pericyclivine was not detected.

Mass spectrometry of *C. roseus* extracts indicated that isovindolinine, vindolinine, 19-epivindolinine, 20-epivindolinine and tabersonine have the same molecular ion at m/z 336 (Table 2).

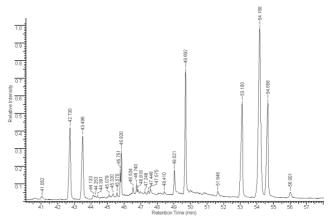


Fig. 1. GC-MS chromatogram obtained from alkaloids isolated from *Catharanthus roseus* L. (G.) Don by maceration at room temperature

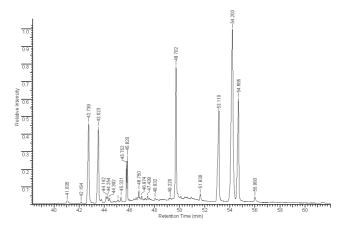


Fig. 2. GC-MS chromatogram obtained from alkaloids isolated from *Catharanthus roseus* L. (G.) Don by extraction at 55°C

Table 1. Alkaloids identified by GC/MS in ethanolic extracts of Catharanthus roseus (L.) G. Don. The data are given as the mean ±SD (n = 3)

No.	Name of alkaloid	R <sub>t</sub> [min]	RI -	Conventional propagation (area, %)			<i>In vitro</i> propagation (area, %)			3.6
				I	II	Mean	I	II	Mean	Mean
1.	Isovindolinine	42.73	2559	7.670±0.16	10.585±0.77	9.128	7.325±3.66	11.375±0.52	9.350	9.239
2.	Vindolinine	43.50	2581	7.255±0.28	9.200±0.66	8.227	6.480±3.70	9.900±0.55	8.190	8.209
3.	19-Epivindolinine	44.13	2600	0.650±0.28	0.695±0.06	0.673	0.570±0.16	$0.595 \pm 0.10$	0.582	0.627
4.	20-Epivindolinine	44.25	2606	0.590±0.32	0.625±0.08	0.608	0.565±0.21	$0.535 \pm 0.10$	0.550	0.579
5.	Tabersonine	44.39	2615	$0.270\pm0.01$	$0.385 \pm 0.02$	0.327	$0.115\pm0.08$	$0.305\pm0.02$	0.210	0.269
6.	Vincoline	45.08	2657	$0.320\pm0.03$	0.285±0.01	0.303	0.095±0.09	$0.170\pm0.00$	0.133	0.218
7.	Pleiocarpamine	45.75	2698	1.120±0.04	1.525±0.02	1.322	0.980±0.22	$1.450\pm0.05$	1.215	1.269
8.	Coronaridine	45.82	2704	2.240±0.03	2.120±0.07	2.180	2.405±0.35	2.060±0.09	2.233	2.206
9.	16-Epivindolinine-N <sub>b</sub> -oxide	46.74	2784	$0.885\pm0.09$	$0.700\pm0.04$	0.793	1.375±0.86	$0.490\pm0.10$	0.933	0.863
10.	Vindolidine	47.58	2852	0.565±0.12	0.175±0.01	0.370	0.655±0.50	0.295±0.05	0.475	0.423
11.	Pericyclivine	49.02	2960	2.545±0.49	_*	1.273	3.285±0.95	-	1.642	1.457
12.	Vindorosine	49.70	3007	10.665±0.39	11.640±0.12	11.153	17.790±3.18	18.890±0.05	18.340	14.746
13.	Tetrahydroalstonine	53.10	3193	$11.430 \pm 0.23$	11.660±0.18	11.545	$3.175\pm0.05$	$3.750\pm0.00$	3.463	7.504
14.	Vindoline	54.17	3238	28.695±1.19	29.225±1.13	28.960	34.255±5.45	31.285±1.19	32.770	30.865
15.	Ajmalicine	54.66	3259	10.865±0.68	12.660±0.08	11.762	6.785±0.57	9.155±0.38	7.970	9.866
Mean				5.718	6.099	5.908	5.724	6.017	5.870	5.889
LSD <sub>α=0.05</sub> for alkaloid (I)					4.698					
LSD <sub>0=0.05</sub> for propagation method (II)					n.s.					
$LSD_{\alpha=0.05}$ for extraction method (III)					n.s.					
$LSD_{\alpha=0.05}$ for interaction:										
$I \times II$		1.920								
$I \times III$		1.901								
$II \times III$	ns									

RI: retention indices relative to n-alkanes (C7-C40) on HP-5MS column

Table 2. Mass spectra data of alkaloids identified in ethanolic extracts of Catharanthus roseus L. cultivated conventionally and derived from in vitro cultures

Alkaloid	Molecular formula	$[M^{+}](\%)$	m/z (relative intensity),%
Isovindolinine	$C_{21}H_{24}N_2O_2$	336(100)	321 [M*-CH <sub>3</sub> ] (3), 305 [M*-OCH <sub>3</sub> ] (8), 277 [M*-CO <sub>2</sub> CH <sub>3</sub> ] (5), 230 (35), 216 (30), 202 (17), 183 (13), 170 (87), 156 (36), 135 (79), 134 (88), 120 (29), 108 (14), 91 (15)
Vindolinine	$C_{21}H_{24}N_2O_2$	336 (100)	321 (3), 305 (9), 277 (5), 230 (33), 216 (30), 202 (16), 183 (12), 170 (83), 156 (33), 135 (77), 134 (85), 120 (27), 115 (14), 107 (21), 93 (16)
19-Epivindolinine	$C_{21}H_{24}N_2O_2$	336 (100)	321 (6), 305 (8), 277 (8), 229 (21), 202 (12), 194 (25), 180 (19), 170 (40), 156 (20), 134 (51), 122 (44), 107 (10), 91 (11), 77 (14)
20-Epivindolinine	$C_{21}H_{24}N_2O_2$	336 (100)	321 (6), 305 (7), 277 (14), 229 (24), 202 (13), 194 (24), 180 (19), 170 (47), 156 (26), 134 (63), 122 (48), 107 (10), 91 (14), 77 (16)
Tabersonine	$C_{21}H_{24}N_2O_2$	336 (100)	305 (9), 277 (7), 229 (22), 216 (33), 202 (6), 194 (15), 180 (14), 168 (25), 156 (26), 135 (54), 122 (31), 107 (33), 92 (5), 93 (9), 77 (12)
Vincoline	$C_{21}H_{24}N_2O_4$	368 (2)	356 (7), 336 (100), 319 (6), 304 (16), 278 (12), 230 (18), 216 (21), 183 (22), 170 (62), 156 (36), 135 (50), 134 (64), 122 (22), 120 (24), 107 (25), 93 (13)
Pleiocarpamine	$C_{20}H_{22}N_2O_2$	322 (96)	307 (8), 293 (4), 292 (4), 263 [M+-CO <sub>2</sub> CH <sub>3</sub> ] (100), 247 (12), 232 (34), 218 (19), 194 (14), 180 (83), 167 (15), 131 (15), 121 (5), 108 (14)
Coronaridine	$C_{21}H_{26}N_2O_2$	338 (100)	323 (30), 309 (6), 253 (11), 214 (25), 136 (75), 135 (22), 124 (34), 122 (29), 108 (10), 96 (8), 77 (6), 41 (11)
16-Epivindolinine-N <sub>b</sub> -oxide	$C_{21}H_{24}N_2O_3$	352(3)	351 (7), 338 (2), 335 (23), 334 (100), 321 (1), 319 (3), 305 (3), 293 (21), 281 (14), 275 (11), 249 (22), 227 (58), 207 (31), 194 (60), 182 (23), 167 (90), 149 (74), 127 (19), 108 (71), 92 (14)
Vindolidine	$C_{24}H_{30}N_2O_5$	426(7)	367 (20), 282 (7), 281 (20), 267 (75), 266 (5), 207 (40), 158 (100), 144 (21), 143 (27), 135 (14), 107 (15), 93 (7), 92 (11), 77 915)
Pericyclivine	$C_{20}H_{22}N_2O_2$	322 (100)	321 (82), 307 (29), 291 [M+-OCH <sub>3</sub> ] (8), 263 [M+-CO <sub>2</sub> CH <sub>3</sub> ] (31), 249 (19), 235 (11), 207 (10), 181 (12), 180 (7), 169 (69), 168 (67), 154 (13)
Vindorosine	$C_{24}H_{30}N_2O_5$	426(6)	366 (7), 282 (13), 279 (16), 267 (13), 266 (26), 207 (2), 158 (94), 135 (100), 121 (26), 107 (18), 93 (9), 77 (6)
Tetrahydroalstonine	$C_{21}H_{24}N_2O_3$	352 (100)	337 (33), 293 [M <sup>+</sup> -CO <sub>2</sub> CH <sub>3</sub> ] (6), 251 (13), 223 (19), 197 (9), 184 (9), 170 (10), 169 (19), 156 (39), 129 (9), 115 (6), 77 (4), 55 (5)
Vindoline	$C_{25}H_{32}N_2O_6$	456 (15)	397 [M*-CO <sub>2</sub> CH <sub>3</sub> ] (9), 309 (13), 296 (25), 282 (29), 188 (84), 174 (37), 161 (22), 135 (100), 121 (42), 107 (20), 93 (12)
Ajmalicine	$C_{21}H_{24}N_2O_3$	352 (100)	$337 (6), 321 \left[M^+-OCH_3\right] (5), 293 \left[M^+-CO_2CH_3\right] (2), 265 (5), 225 (7), 209 (13), 184 (34), 170 (8), 169 (19), 156 (51), \\ 143 (7), 129 (7), 115 (5), 77 (4)$

Moreover, 19-epivindolinine, 20-epivindolinine and tabersonine showed similar fragmentation patterns and retention times (Table 1), very close to each other.

Dejerassi et al. (1962) reported that peaks at m/z 93, 107, 120, 121, 122 as well as at m/z 134, 135 are associated with the piperidine portion of the molecule and occurs in all mass spectra of the vindolinine series. This group is dominated by vindolinine, 19epivindolinine and their N<sub>b</sub>-oxides.

Vindolinine and tabersonine with [M<sup>+</sup>] at 336 m/z are considered as catharanthine isomers (Ferreres et al., 2010).

Interestingly, the base peak at m/z 336 (100%) in vincoline corresponds to the molecular ion of vindolinine (M<sup>+</sup>, 336). The presence of peaks at m/z 93, 107, 120, 122 and 135 shows that the both compounds belong to the same group. Svoboda et al. (1964) noted similarities in the case of IR and NMR spectra of these compounds.

I: maceration at room temperature; II: extraction at 55°C \* - not detected

n.s. - not significant

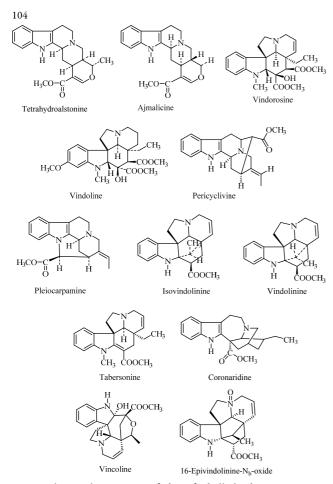


Fig. 3. Chemical structures of identified alkaloids

Pleiocarpamine shows prominent signals at m/z 322 (M<sup>+</sup>), m/z 263 (M<sup>+</sup>-CO<sub>2</sub>CH<sub>3</sub>) and m/z 180 (quinolium ion). Our results are in agreement with published data for pleiocarpamine (Hesse *et al.*, 1964; Akhgari *et al.*, 2015).

Comparing the mass spectrum of coronaridine with literature data (Zocoler *et al.*, 2005), we were observing only small differences in the relative ion intensities.

The mass spectrum of pericyclivine was characterized by the ions at m/z 322, 321, 291, 263 and 169. Similar fragmentation patterns was noted by Vieira *et al.* (2008).

The spectra of ajmalicine and tetrahydroalstonine exhibited fragments of the  $\beta$ -carboline skeleton at m/z 184, 170, 169 and 156, which is in accordance with the results published by Hesse (1974). However, both compounds were identified according to standards.

Vindorosine (molecular weight, 426) showed difference of 30 m/z in mass as compared to vindoline (molecular weight, 456). This may suggest that vindorosine is derivative of vindoline. Moreover, vindolidine and vindorosine have the same molecular ion at m/z 426. Slight differences were noted only in their intensity. Detailed fragmentation patterns of the mass spectra of these compounds can be found in Table 2. The major peaks observed in the mass spectrum of vindorosine were the same as those reported by Zhao *et al.* (2013).

Our results indicated that the GC-MS method can be successfully applied for the analysis of the mixture of periwinkle alkaloids without necessity of derivatization. Kreh *et al.* (1995), who applied for the first time GC-MS in the analysis of underivatized Amaryllidaceae alkaloids from *Narcissus* 

pseudonarcissus have demonstrated its advantages over the analysis of silylated samples. Tram et al. (2002) reported the results of GC-MS analysis of underivatized alkaloids from leaves of *Crinum latifolium*, while Tosun and Tamer (2004) used this method for determination of pyrrolizidine alkaloids in seeds of *Heliotropium europaeum*.

According to Subbaiyan *et al.* (2014), gas chromatography coupled to mass spectrometry is the best for identification of long chain hydrocarbons, alcohols, acids, esters, alkaloids, steroids and nitrogen compounds.

Catharanthus roseus is one of the main source of terpenoid indole alkaloids used for the preparation of diabetic drugs, cardiac drugs, in hypertension and in anticancer drugs formation by the pharmaceutical industries (Misra et al., 2009). Moreover, some of these therapeutic molecules are obtained by semi-synthesis using natural precursors extracted from *C. roseus* leaves.

The low content of dimeric alkaloids in the plant as well as their high market price encouraged the researchers to improve the production of these valuable metabolites through the technique of tissue cultures. The influence of concentration of different growth regulators on the production of callus and the amount of synthesized alkaloids have been investigated (Misawa, 1994; Junaid *et al.*, 2008; Kalidass *et al.*, 2010; Verma *et al.*, 2012; Soleimani *et al.*, 2013).

Soleimani *et al.* (2013) reported stimulatory effect of 2,4-D (2,4-dichlorophenoxyacetic acid) on callus formation as well as on the level of vinblastine and vincristine. They also found that increasing the concentration of kinetin resulted in a significant decrease in callogenesis and alkaloid production. Kalidass *et al.* (2010) demonstrated that the use of kinetin in a concentration above 1 mg·dm<sup>3</sup> had less effect of vinblastine production. Moreover, increase of its concentration was associated with decrease of vinblastine synthesis.

Verma *et al.* (2012) revealed that half strength basal MS medium supplemented with 2,4-D and BAP (0.5 mg/dm<sup>3</sup> and 1.0 mg/dm<sup>3</sup>, respectively), and with 6% sucrose was the best for callus biomass production along with high alkaloid content.

Generally, the auxins were approved as the best for callus proliferations and growth. However, the combination of auxins with cytokinins were found to be better for leaf callus growth and enhancement in alkaloids content (Misawa, 1994; Junaid *et al.*, 2008).

An antagonistic effect of gibberellins and cytokinins on the biosynthesis of monoterpenoid indole alkaloids in *C. roseus* was reported by Amini *et al.* (2009). According to them, the inhibitory effect of gibberellin was correlated with the silencing of two genes, encoding enzymes of the alkaloids biosynthetic pathway. Normally this process is regulated by cytokinins. In case of exogenously applied cytokinin they observed considerably increase of synthesis of ajmalicine and serpentine in untransformed cotyledon callus. The use of 2,4-D decreased the accumulation of these metabolites.

Garnier *et al.* (1996) found that BAP (6-benzylaminopurine) and in some cases NAA (naphtahleneacetic acid) stimulated the accumulation of vindoline and catharanthine in periwinkle.

In general, MS medium is most widely used in tissue cultures, but the number of inorganic and organic salts and their levels are usually different. The ammonium and nitrate ions inhibit alkaloids production (Van Gulik *et al.*, 1993). These salts are added to the media for growth promotion and differentiation.

The maximum production of alkaloids occurs at low levels of phosphate, nitrate and ammonia in the medium (Schlatmann et

al., 1992).

Vitamins and organic components had slight effect on the alkaloids production. Only in case of glucose positive effect on ajmalicine production in hairy-root culture was noted by Moreno et al. (1995). On the other hand, Zhao et al. (2001) showed that the addition of succinic acid, tryptamine and tryptophan to the culture medium caused significant growth of ajmalicine and catharanthine levels.

The present study indicated that the MS basal medium supplemented with kinetin (2.0 mg·dm³), IBA (1.0 mg·dm³), NAA (0.2 mg·dm³) and sucrose (30 mg·dm³) caused the increase of the production of vindorosine and vindoline in *C. roseus* tissue cultures. However, increasing the levels of important periwinkle alkaloids is extremely difficult and dependent on many factors. This explains why the role of different plant growth regulators in regulating monoterpene indole alkaloids (MIAs) biosynthetic pathway is still being extensively investigated.

#### Conclusion

The obtained results showed that the plants cultivated conventionally had higher amounts of ajmalicine and tetrahydroalstonine, while these from *in vitro* propagation – higher amounts of vindorosine and vindoline. Moreover, the temperature of extraction also affected the percentage content of isolated alkaloids. In case of isovindolinine, vindolinine and ajmalicine the more effective was extraction at 55 °C, while for pericyclivine – maceration at room temperature. Our study may find wide applications in the optimization of production and isolation of *Vinca* alkaloids.

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