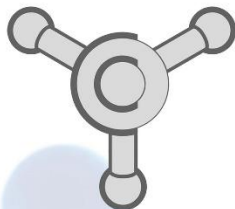


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Tamara TODOROVIĆ

Ljubodrag VUJISIĆ

Jelena RADIVOJEVIĆ

Vuk FILIPOVIĆ

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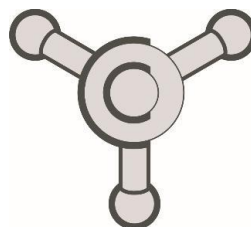
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SCIENTIFIC COMMITTEE

Dr Tamara TODORVIĆ

Dr Ljubodrag VUJISIĆ

Dr Jelena RADIVOJEVIĆ



ORGANIZING COMMITTEE

Dr Života SELAKOVIĆ

Vuk FILIPOVIĆ

Jelena LAZIĆ

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Materials science

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Removal of hexavalent chromium Cr(VI) from aqueous solutions using cellulose-magnetite membrane CelMag- M

Jovana M. Perendija¹, Dragana L. Milošević¹, Mladen D. Bugarčić², Aleksandar D. Marinković³

¹ University of Belgrade - SI Institute of Chemistry, Technology and Metallurgy National Institute, Center of Ecology and Technoeconomics, Belgrade, Serbia

² Institute for technology of nuclear and other raw materials, Belgrade, Serbia

³ University of Belgrade - Faculty of Technology and Metallurgy, Belgrade, Serbia

The industries of leather-tanning, mining and textile dyeing, generate large amounts of chromium-containing wastewater. Hexavalent Cr(VI) is highly poisonous and extremely mobile in the surface-water and groundwater in a broad pH range and therefore it has been identified as a potentially carcinogenic substance. The aim of the presented work was to develop cellulose-based membrane functionalized with magnetite, which could be used as an efficient adsorbent for the removal of hexavalent chromium Cr(VI) ions from aqueous solutions. Cellulose-based filter (CF) was functionalized with magnetite in three-step process. In the first and second step CF surface was modified using an ethanolic solution of (3-aminopropyl)triethoxysilane (APTES), and diethylenetriaminepentaacetic acid dianhydride, respectively. The introduction of amino and carboxylic groups provided successful precipitation of magnetite in the third step. The obtained cellulose-magnetite membrane (CelMag-M) was characterized by FTIR and SEM analysis. Adsorption of Cr(VI) onto CelMag-M was studied using batch-adsorption test. Under optimum pH conditions, the maximum experimental adsorption capacity of CelMag-M for Cr(VI) was found to be 111.2 mg g⁻¹. The adsorption process was endothermic, the equilibrium adsorption data could be best fitted to the Langmuir adsorption isotherm model and kinetics was in agreement with the pseudo-second-order rate equation.

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