- 1 Intrashelf basin record of redox and productivity changes along the Arabian margin of Neo-
- 2 Tethys during Oceanic Anoxic Event 1a
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16 ABSTRACT

17 The biotic, environmental, climatic, oceanic, and sea-level perturbations during the Early Aptian Oceanic Anoxic Event (OAE) 1a have been extensively documented from both deep- and shallow-18 marine deposits worldwide. However, there has been relatively little comparative assessment of 19 20 the simultaneous interplay among organic carbon burial, redox conditions, terrigenous output, and 21 productivity, leading to a lack of precise constraints on these relationships. Here, we use analyses of stable carbon isotopes ($\delta^{13}C_{org}$, $\delta^{13}C_{carb}$, and $\Delta^{13}C$), total organic carbon (TOC), detrital proxies 22 (Al, Si, Ti, K), redox-sensitive (RSTE: U, V, Mo) and productive-sensitive (PSTE: P, Cu, Ni) trace 23 elements from a continuous, predominantly carbonate succession of the Kazhdumi Intrashelf Basin 24 25 to evaluate the culprits for the OAE1a-associated changes in bottom-water oxygenation, organicrich layer formation, and biotic shifts along the Arabian margin of the Neo-Tethys. Concentrations 26 27 of Al-normalized RSTE and TOC values indicate that the bottom water conditions ranged from

oxic prior to and at the onset of the OAE 1a (carbon-isotope segments C2 to basalmost C4 sensu 28 29 Menegatti et al., 1998), to anoxic-suboxic but not euxinic (Mo<25 ppm) during the lower C4 through C5+C6 segments, and then returned to oxic-suboxic in the remaining C5+C6 segment. 30 The increase in Al-normalized PSTE coupled with TOC concentrations in the basal C4 is coeval 31 with a change from predominantly orbitolinid-ostreid to planktic foraminifera-radiolarian biota. 32 33 The periodically high productivity, driven both by the surface-water productivity as well as by phosphorus recycling from the sediments, continued through the C5+C6 segments as evidenced 34 35 by matching Si/Al and PSTE peaks (Cu/Al and Ni/Al). The study sheds new light on the causes of variations in bottom water deoxygenation, organic content, nutrient availability, and biotic shifts 36 in semi-restricted, relatively deep (>100 m), continental-margin basins during major oceanic 37 perturbations. 38

KEYWORDS: Kazhdumi Intrashelf Basin; High-resolution isotopic curve; Redox and productivity
sensitive trace elements; OAE 1a

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42 **1. Introduction**

The Early Aptian oceanic anoxic event (OAE1a, ~120 Ma) was a brief episode (1 to 1.3 Myr 43 duration; Li et al., 2008) of widespread oxygen deficiency of the ocean bottom-waters that resulted 44 45 in an extensive organic carbon sequestration in marine sediments (Schlanger and Jenkyns, 1976; Arthur et al., 1990). The event is globally recognized by its carbon-isotope signature that is 46 commonly subdivided into distinctive segments (Menegatti et al., 1998; Gröcke et al., 1999; 47 Gröcke, 2002; Castro et al., 2021; Percival et al., 2021), and is defined by a major negative carbon-48 49 isotope excursion (segment C3) followed by two pronounced positive excursions (C4 and C6), and post-OAE 1a continuation of high δ^{13} C values (C7) (Schlanger and Jenkyns, 1976; Arthur et al., 50 1990; Menegatti et al., 1998; Castro et al., 2021). The major OAE 1a culprit was the flood basalt 51 volcanism associated with the formation of the Greater Ontong Java Plateau, which resulted in a 52 massive release of isotopically light carbon into the atmosphere and the ocean and triggered global 53 warming (e.g., Tarduno et al., 1991; Tejada et al., 2009; Charbonnier and Föllmi, 2017; Matsumoto 54 et al., 2022). Other possible mechanisms include rapid methane release and High Arctic Large 55 Igneous Province (HALIP) sill intrusions into organic-rich marine sediments (e.g., Beerling et al., 56 57 2002; Adloff et al., 2020). The massive carbon emissions potentially increased the planetary pCO_2

by 3000 ppm (Beerling et al., 2002; Bauer et al., 2017) and arguably led to ocean acidification that 58 resulted in a significant decrease in pelagic biogenic calcite production (Erba, 2004; Erba et al., 59 2015; see also Gibbs et al., 2011). The Aptian (super)greenhouse conditions were interspersed with 60 cooling events (Jenkyns, 2018) that left a global record of significant sea-level fluctuations (e.g., 61 Matthews and Frohlich, 2002; Husinec and Jelaska, 2006; Graziano et al., 2016; Simmons et al., 62 2020; Bover-Arnal., 2022), likely driven by aquifer- and glacio-eustasy (Sames et al., 2020). Rapid 63 oceanic deoxygenation (<30 kyr interval) developed with the onset of OAE 1a (Bauer et al., 2021; 64 65 carbon-isotope segment C3) following the initial phases of large igneous province volcanism and biological crises, and well before intensified weathering that is recorded in oceanic chemistry 150-66 300 kyr later (Lechler et al., 2015). Large swaths of the ocean became ferruginous and anoxic for 67 >1 Myr, the latter at least transiently into the photic zone, which resulted in increased rates of 68 69 trace-metal transfer to the sediment (Algeo, 2004; Algeo and Rowe, 2012) and enhanced pyrite and sulfurized organic matter burial (Bauer et al., 2022). Given the relatively short duration of the 70 71 OAE 1a, the drawdown of trace elements was of limited duration, and as the event terminated their 72 concentrations in seawater likely recovered rapidly according to their seawater residence times 73 (~730 kyr for Mo, ~400 kyr for U, ~50 kyr for V, and shorter for other metals; Algeo, 2004).

74 While the environmental, climatic, and biotic impacts of the OAE la-associated 75 perturbations on deep- and shallow-water marine carbonate succession have been extensively 76 studied worldwide (e.g., Weissert et al., 1998; Huck et al., 2010; Husinec et al., 2012; Amodio and Weissert, 2017; Hueter et al., 2019; Steuber et al., 2022), the relationship between organic carbon 77 burial, redox conditions, terrigenous input, and productivity remains poorly constrained. 78 79 Carbonates precipitated in equilibrium with seawater incorporate trace elements and stable isotopes whose values reflect the composition of the ambient seawater (Brand and Veizer, 1980). 80 Following their deposition, however, they commonly underwent a complex series of marine, 81 82 meteoric, and burial diagenetic processes that altered their original trace element and stable isotope values and shifted them towards equilibrium with the ambient interstitial water (e.g., Brand and 83 84 Veizer, 1980; Swart and Kennedy, 2012). These processes, whether intrinsic (e.g., mineralogy, porosity, and permeability) or extrinsic (e.g., climate, bioturbation, and advecting fluid 85 composition and flow rate) are primarily influenced by the reactivity of the sediment, the duration 86 of reaction, and the degree of isotopic or elemental/chemical disequilibrium (Fantle et al., 2020). 87 88 Here, we reconstruct the palaeoredox and paleoproductivity changes across the OAE 1a by

utilizing a continuous carbonate sedimentary record from the Kazhdumi Intrashelf Basin exposed 89 in Fars Province (Iran). Given their predominantly mud-supported textures with limited 90 porosity/permeability, and lack of evidence for subaerial exposure in the study area (Jafarian et al., 91 2023), the potential for an extensive diagenetic modification of the Kazhdumi Intrashelf Basin 92 carbonate sediments was limited. We establish multiple geochemical records, including carbon 93 isotopes ($\delta^{13}C_{org}$, $\delta^{13}C_{carb}$, and $\Delta^{13}C$), total organic carbon (TOC), lithogenic elements (aluminum-94 Al, silica-Si, titanium-Ti, potassium-K), redox-sensitive trace elements (RSTE; uranium-U, 95 96 vanadium-V, molybdenum-Mo), and productive-sensitive trace elements (PSTE; copper-Cu, 97 nickel-Ni, phosphorus-P) to unravel the causes of variations in bottom-water deoxygenation, organic-rich layer formation, and biotic shifts taking place in semi-restricted, relatively deep (>100 98 99 m), continental-margin basins during major oceanic perturbations.

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2. Zagros Mountains and the Aptian Kazhdumi Intrashelf Basin

Spanning about 2000 km from the Turkish-Iranian border in the northwest to the Makran area in 102 103 southeastern Iran, the Zagros fold-thrust belt (ZFTB) occupies the northeastern margin of the Arabian Plate (Alavi, 2004). The Zagros Mountains are notable for their abundance of supergiant 104 hydrocarbon fields, making them the most productive fold-thrust belt on a global-scale (e.g., 105 Falcon, 1958; Murris, 1980; Bordenave and Burwood, 1989; Bordenave and Hegre, 2005). The 106 107 sedimentary evolution of the Zagros Mountains includes three phases (Heydari, 2008): the Persian Platform phase (latest Precambrian-Early Permian), the Arabian Platform phase (Permian-latest 108 109 Cretaceous), and the Zagros Foreland Basin phase (latest Cretaceous to present). During the Late 110 Pre-Cambrian to the Permian period, the Persian Platform was a comparatively stable continental platform that was submerged under an epeiric sea (Golonka, 2000; Heydari, 2008). In the Early 111 Permian, the platform experienced the separation of several microplates, which are commonly 112 113 known as the Cimmerian continent. This event resulted in the formation of the Neo-Tethys Ocean 114 (Golonka, 2000; Heydari, 2008). After the formation of the Neo-Tethys Ocean, the Zagros region became integrated with the Arabian Platform. Throughout the Jurassic and Cretaceous, the Arabian 115 Platform occupied the equatorial region and was connected to the northern margin of Africa (Smith 116 et al., 1994). In the Early Cretaceous, the southern margin of the Neo-Tethys bordered the NE 117 passive margin of the Arabian Plate, which was situated approximately 10-15° S of the equator 118

(Masse et al., 1993; Sharland et al., 2001). During that time, an extensive shallow-water carbonate 119 platform developed with two intrashelf basins, the Bab Basin to the south and southeast, and the 120 Kazhdumi Basin to the northwest (e.g., Piryaei, 2010; Razin et al., 2010; Mehrabi et al., 2015, 121 2018; Wohlwend et al., 2016; Bahrehvar et al., 2020; Fig. 1A). The development of the intrashelf 122 basins could have resulted from differential thermal subsidence, halokinetic movements linked to 123 124 the InfraCambrian Hormuz Salt Series, notably in the eastern Persian Gulf, or the reactivation of extensional basement faults like Kazerun and Hendijan faults in the NW Persian Gulf and the 125 126 onshore areas of the Zagros Basin (e.g., the Dezful Embayment, Izeh Zone, and Lorestan) (e.g., 127 Wrobel-Daveau et al., 2010; Pierson et al., 2010; Mehrabi and Rahimpour-Bonab, 2014). The Arabian Platform phase terminated in the latest Cretaceous as a result of the collision between the 128 Sanandaj-Sirjan microplate and the Arabian Platform (Golonka, 2000). In the course of the Zagros 129 130 Orogeny, the Arabian Plate eventually collided with the microcontinent of Iran, leading to the closure of the Neo-Tethys Ocean and the formation of the ZFTB (Berberian and King, 1981; 131 132 Golonka, 2000; Agard et al., 2005). However, the exact timing of the collision remains a subject of debate, with various estimates suggesting a range from the Late Cretaceous to Pliocene (e.g., 133 134 Stöcklin, 1968; Berberian and King, 1981; Koop and Stoneley, 1982; Fakhari, 2008; Horton et al., 2008). 135

The Kazhdumi Intrashelf Basin differentiated during the Early Aptian with a maximum water depth estimated at 135 m (van Buchem et al., 2010). The basin was connected with the Neo-Tehys Ocean, which enabled an efficient exchange of waters with the open ocean during the OAE 1a. The biostratigraphy, chemostratigraphy, lithostratigraphy, and facies of the Lower Aptian Lower Dariyan succession are summarized below.

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142 **2.1. Lithostratigraphy**

At its type locality measured at Kuh-e Gadvan (Fars Province, Iran), the Dariyan Formation is comprised of thick-bedded, massive, orbitolinid limestone (James and Wynd, 1965), which was formed in a shallow-marine environment. Except for the central and southern Lurestan, where it grades into shales and limestones of the Garau Formation, the Dariyan Formation is found throughout southwestern Iran (Fig. 1B). On the opposite, the Arabian side of the Persian Gulf, its stratigraphic equivalent is the Shu´aiba Formation. Towards the Kazhdumi Basin margin, the Dariyan Formation is subdivided into three informal units, including Lower Dariyan, Kazhdumi
Tongue (Middle Dariyan), and Upper Dariyan (e.g., van Buchem et al., 2010; Mehrabi et al., 2015,

151 2018; Jafarian et al., 2023).

In the study area, the 39.5-m-thick Lower Dariyan unit conformably overlies shales, marls, 152 153 and limestones of the Barremian-Aptian Gadvan Formation, and is conformably overlain by shales 154 and marls of the Kazhdumi Tongue (Middle Dariyan unit) (Fig. 2A–C). The lower part of the lower Dariyan unit is made up of thin- to medium-bedded, predominantly dark-gray, mud-supported 155 skeletal limestones with larger benthic foraminifera, subordinate planktonic foraminifera, and rare 156 157 ammonite casts, and interbedded with shell beds (ostreid limestone; Fig. 2D). In the middle part 158 of the lower Dariyan unit, which is regionally known as the radiolarian flood zone (RFZ) (James and Wynd, 1965; Vincent et al., 2010), dark gray radiolaria-rich limestones alternate with dark 159 160 gray shales and black chert bands (laterally discontinuous; up to 12 cm thick). The remaining upper part of the Lower Dariyan unit is made up of mud-supported, radiolarian and planktic foraminiferal 161 162 limestones. The contact between the Lower Dariyan unit and Kazhdumi Tongue is gradual and characterized by a color change from dark to greenish gray (Fig. 2B–C). 163

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165 **2.2. Lower Aptian facies**

The Lower Aptian facies and interpreted depositional settings of the Kazhdumi Intrashelf Basin were described in detail by van Buchem et al. (2010), Mehrabi et al. (2015, 2018), Bahrehvar et al. (2020), and Jafarian et al. (2023). In the study area, four distinct microfacies are identified, from deepest to shallowest: radiolarian wackestone and packstone (MF1; Fig. 3A), planktic foraminiferal wackestone and packstone (MF2; Fig. 3B), skeletal wackestone and floatstone (MF3; Fig. 3C), and ostreid packstone and rudstone (MF4; Fig. 3D). The main characteristics of each microfacies are summarized below after Jafarian et al. (2023).

Gray to dark gray radiolarian wackestone and packstone (MF1) is dominated by well-sorted, mostly spherical radiolarians (abundance ~20–40%) in a dark micrite matrix with organic matter. Sponge spicules (mainly monoaxon), echinoid spines, and planktonic foraminifers (mainly *Hedbergella* spp.) are subordinate, whereas coccoliths are rare. This microfacies is thinly bedded, locally laminated, and associated with chert bands. The majority of radiolarian tests are completely 178 filled with sparry calcite. Pyritization and chemical compaction are common. The MF1 179 microfacies formed in a deep water, distal part of the intrashelf basin. It comprises 45% of the 180 studied section.

Light gray to black, organic-rich planktic foraminiferal wackestone and packstone (MF2) contains predominantly well to moderately sorted planktonic foraminifers (abundance 10–20%; hedbergellids, globigerinelloids and leupoldinids), whereas sponge spicules, pelagic echinoids, bivalve fragments, and radiolarians are subordinate (5–10%). Most skeletal fragments and foraminiferal tests are completely replaced or filled with sparry calcite; pyritization is common. The MF2 microfacies deposited in a low-energy, open-marine setting likely with oxygen-depleted bottom conditions. It makes up ~25% of the studied section.

188 Skeletal wackestone and floatstone (MF3; corresponds to MF4 of Jafarian et al., 2023) is made up of medium- to fine-grained, poorly sorted allochems (abundance ~15%), including corals, 189 gastropods, echinoid spines, and planktonic foraminifers (rare) along with coarse-grained bivalve 190 fragments (e.g., *Exogyra* sp.) and larger benthic foraminifera (e.g., *Lenticulina* sp., *Choffatella* 191 decipiens, Palorbitolina lenticularis) in a micritic matrix. Hematite and pyrite have replaced some 192 193 of the shells and filled in the molds. The MF3 microfacies deposited in a mid-ramp, low- to moderate-energy setting likely close to the storm wave base. It constitutes 27.5% of the studied 194 section. 195

196 Ostreid packstone and rudstone (MF4; corresponds to MF6 of Jafarian et al., 2023) contains poorly sorted, medium- to coarse-sized (≤ 5 mm; abundance 15–20%) fragments of ostreid bivalves 197 198 (*Exogyra* sp.) that exhibit normal grading. Other skeletal components (abundance 5-10%) include 199 large benthic foraminifera (e.g., Lenticulina sp., C. decipiens), echinoderms, gastropods, and 200 serpulids, whereas planktonic foraminifers (Hedbergella spp.), echinoid spines, and sponge spicules are rare. Replacement (pyritization, hematitization), boring/burrowing (due to serpulid 201 202 worm activity), and fracture-filling calcite cementation are the most common diagenetic processes. 203 This microfacies formed in a mid-ramp setting with vigorous sediment reworking during storms. It makes up only 2.5% of the studied section. 204

205 **2.3. Biostratigraphy and chemostratigraphy**

The age of the studied Lower Dariyan unit of the Dariyan Formation has been established as Early 206 Aptian (Bedulian) based on benthic foraminiferal, planktic foraminiferal, and ammonite biozones, 207 208 and tied to carbon-isotope stratigraphy (e.g., Vincent et al., 2010; Schroeder et al., 2010; Yavari et 209 al., 2017; Mehrabi et al., 2018; Jafarian et al., 2023). The lower part of the Lower Dariyan unit contains abundant benthic foraminifera Palorbitolina lenticularis and Choffatella decipiens, which 210 211 are typical for the earliest Early Aptian (e.g., Peybernès, 1982; Husinec, 2001; Husinec and Sokač, 2006; Schroeder et al., 2010; Granier et al., 2017; Schlagintweit and Rashidi, 2022). The last 212 213 appearance of these larger benthic foraminifera is associated with the regional deepening and deposition of the RFZ. In the basin center, the RFZ spans the Deshayesites weissi to Deshayesites 214 deshayesi ammonite zones (Vincent et al., 2010) and includes the regional K80 maximum flooding 215 surface (Sharland et al., 2001; Mehrabi et al., 2018). The overlying Kazhdumi Tongue unit 216 contains the first appearance datum of *Mesorbitolina parva*, which is indicative of the late Aptian 217 Gargasian (Schroeder et al., 2010; Yavari et al., 2017). 218

219 The age of the Lower Dariyan interval at the Dareh Sefid section has further been constrained by chemostratigraphy and tied to carbon-isotope segments sensu Menegatti et al. 220 (1998). The C-isotope segments were identified based on the unsmoothed δ^{13} C curve following 221 Menegatti et al. (1998) original definition (Jafarian et al., 2023); the bulk carbonate δ^{13} C values 222 223 range from -0.8 to +4.6‰ VPDB (mean +1.9‰). The basal part of the unit corresponds to the 224 segment C2, which corresponds to the upper G. blowi zone. The remaining Lower Dariyan succession deposited during the OAE 1a (C3–C6) span from the uppermost *Globigerinelloides* 225 blowi into the upper Leupoldina cabri zone, with RFZ spanning from the upper C4 to the lower 226 C5 (Jafarian et al., 2023). Segment C2 is characterized by relatively uniform δ^{13} C values, and is 227 followed by a major negative shift (-1.4 to -1.6%; segment C3) marking the onset of OAE 1a. A 228 subsequent stepwise positive excursion to +2.8‰ VPDB corresponds to segment C4 at the Dareh 229 230 Sefid section. The remainder of OAE 1a exhibits relatively constant values (C5 + C6), with a pronounced positive shift at the top of this joint segment (Jafarian et al., 2023). 231

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233 **3.** Methodology

The samples were collected from the Dareh Sefid section (near a small village with the same name; it is also locally known as Banesh due to its proximity to a bigger homonymous village), which crops out NW of Shiraz City, Fars Province (Figs. 1–2). The 265-m-thick section has recently been
logged in detail and analyzed for microfacies, biostratigraphy, chemostratigraphy (stable carbon
and oxygen), and sequence stratigraphy (Jafarian et al., 2023). In this study, we focused on the
39.5-m-thick lowermost Aptian part of the section, and used additional geochemical analyses of
80 samples collected at 0.5–0.7 m intervals.

Sampling for $\delta^{18}O/\delta^{13}C_{carb}$ isotopes centered on the extraction of powders from the micritic 241 matrix using a microdrill under a reflected-light microscope. The $\delta^{18}O/\delta^{13}C_{carb}$ analysis is 242 described in detail in Jafarian et al. (2023, and their Supplementary Material Table). Given the 243 absence of subaerial exposure and non-depositional emergent episodes, the succession likely 244 contains a complete record of $\delta^{13}C_{carb}$ oscillations (cf., platform carbonates; Husinec and Read, 245 2018). For the $\delta^{13}C_{org}$ analysis (Table S1), the powdered rock samples were decalcified using 3M 246 HCl (at room temperature) in 50 ml centrifuge tubes for 16 h. Samples were subsequently 247 neutralized with de-ionized water and dried at 40°C for 36 h. The sample pellets were reground 248 249 and weighed into tin capsules for carbon isotope analysis. Stable isotope measurements were 250 performed at Durham University in the Stable Isotope Biogeochemistry Laboratory (SIBL) using 251 a Costech Elemental Analyzer (ESC4010) coupled to a Thermo Scientific Delta V Advantage isotope ratio mass spectrometer. Carbon isotope ratios are corrected for ¹⁷O and reported in 252 253 standard delta (δ) notation in per mil (∞) relative to Vienna Pee Dee Belemnite (VPDB). Isotopic 254 accuracy was monitored through routine analyses of in-house standards and international standards (IAEA-600, IAEA-CH-3, IAEA-CH-6, NBS 19, USGS24, USGS40). International and in-house 255 standards were run daily and provided a linear range of $\delta^{13}C_{org}$ between -46% and +3% for 256 257 isotopic correction. Analytical accuracy in sample $\delta^{13}C_{org}$ was better than 0.2‰ (2 σ) based on 258 standards and replicate sample analysis.

Total organic carbon (TOC) data (Table S1) were obtained as part of the isotopic analysis using the internal standard, glutamic acid (40.82 wt.% C). The Costech Elemental Analyzer was set to macro-oxygen for complete combustion to liberate all organic carbon. The isotope analyses were done in "no dilution mode" and all analyses produced a CO₂ signal > 800 mV, hence providing maximum confidence in the reported $\delta^{13}C_{org}$ values.

Bulk-rock major and trace elements (RSTE = redox-sensitive elements e.g., Mo, U, V and PSTE = productivity-sensitive elements e.g., Cu, Ni, P) were analyzed by X-ray fluorescence

(XRF) and inductively coupled plasma mass spectrometry (ICP-MS) on 40 samples at 266 ALSGLOBAL Laboratory, China (Table S2). To prepare the samples for the XRF analysis, the 267 268 whole-rock powders (200-mesh) were first dried in an oven for 4 h at 105°C. Then, approximately 0.6 g of each sample powder was mixed with 6 g of lithium borate flux mixture and 0.3 g of 269 oxidant, and a few drops of lithium bromide were added as a release agent. The mixture was then 270 271 fused in a furnace at 1050°C. The resulting flat glass discs were analyzed using the ZSX Primus II spectrometer. Samples were calibrated according to the Chinese National Reference Materials 272 (GBW07103, GBW07105, GBW07107, GBW07111). Using the theoretical α coefficient method, 273 the data were corrected and yielded a relative standard deviation of < 2 wt. %. 274

275 For ICP-MS, 200-mesh samples were dried in an oven at 105 °C for 12 h. Then, 50 mg of each dried sample was precisely weighed and transferred into a Teflon bomb. Following this, 1 ml each 276 277 of HNO₃ and HF were slowly introduced into the Teflon bomb. The Teflon bomb was placed in a stainless steel pressure jacket and heated to 190 °C for > 24 h in the oven. Once cooled, the Teflon 278 279 bomb was opened and placed on a hotplate at 140°C. The sample was evaporated to incipient dryness and 1 ml HNO₃ was added. The sample was then evaporated to dryness again. Next, a 280 281 solution containing 1 ml of HNO₃, 1 ml of MQ water, and 1 ml of 1ppm In internal standard solution was added. The Teflon bomb was released and placed in the oven at 190 °C for > 12 h. 282 283 Finally, the resulting solution was transferred into a polyethylene bottle, and the volume was adjusted to 100 g by adding 2 HNO₃. The Agilent 7700e ICP-MS was utilized to conduct trace 284 element analyses of whole rock samples. Lower detection limits range from 0.01 to 5 ppm. The 285 analytical reproducibility, as determined by performing replicate analyses on selected samples, 286 287 was found to be < 5%.

To minimize the effects of biogenic/authigenic carbonate enrichment and detrital dilution, 288 289 selected elements were normalized to aluminum (element/Al), which is the main component of 290 alumosilicate sediment fraction, does not move during diagenesis, and is therefore commonly used for comparison among different units (Tribovillard et al., 2006). Plots showing element/Al ratios 291 292 used in this study (V/Al, Mo/Al, U/Al, Ni/Al, Cu/Al, P/Al) against the stratigraphic interval, show the level of enrichment or depletion relative to their concentration in a reference, which in this case 293 294 is an average shale (AS) (Wedepohl, 1971, 1991; Table S3). Both the Al normalization and comparison to AS values have their drawbacks, and should not be used as the only method to 295

quantitatively identify elemental contributions (Van der Weijden, 2002). The triangular Al_2O_3 – SiO₂ – CaO diagram (Brumsack, 1989) was constructed to evaluate the variability of the three main sedimentary rock components recorded within the studied succession, i.e., clays, quartz and/or biogenic silica, and calcium carbonate. The strengths of association between selected variables were evaluated using the Pearson correlation.

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4. Results

303 4.1. Organic carbon isotopes ($\delta^{13}C_{org}$) and $\Delta^{13}C$

The $\delta^{13}C_{\text{org}}$ values range from -27.2‰ to -22.6‰ (mean= -25.4‰; Table S1; Fig. 4), with the 304 highest value recorded at the base of the RFZ. The basal part of the section (lower C2 segment) 305 shows an increase in values reaching -25.5% at -3 m, which is followed by a decrease to -27.2%306 at ~7 m (upper C2). The values increase across the C2–C3 boundary reaching -26.5‰ at the base 307 of C4. The values remain relatively constant through lower C4, which is followed by a pronounced 308 positive excursion to -22.6‰ at the base of the RFZ. From there, the values exhibit a stepwise 309 310 decrease to -25.6‰ at 22 m, followed by an increase to -23.8‰ at ~24 m; the latter is correlatable with a positive shift in TOC values in the upper RFZ. The remaining RFZ and Lower Dariyan 311 succession (segments C5+C6) exhibits relatively uniform values (average -24.5%) with three 312 minor positive excursions at \sim 32, \sim 34, and \sim 37 m. 313

The isotopic difference between paired isotopic measurements of bulk carbonate and 314 organic matter ($\Delta^{13}C = \delta^{13}C_{carb} - \delta^{13}C_{org}$) ranges from 23.8% to 28.4% (mean= 27.0%); Table S1; 315 Fig. 4). The basal part of the section (lower C2 segment) shows a stepwise decrease in values 316 reaching 25.6‰ at \sim 3 m, which is followed by an interval of relatively uniform values in the upper 317 C2 segment (27.2‰ to 27.5‰). The negative excursion begins with the onset of the OAE 1a (C2– 318 C3 boundary) and reaches the minimum Δ^{13} C value of 25‰ at the base of C4, which corresponds 319 to the lowest $\delta^{13}C_{carb}$ value (recorded in the section -0.8%; Fig. 4). This is immediately followed 320 by a positive excursion to 27.64‰ at 11 m and an interval of stepwise, gently increasing values. 321 A major negative excursion at the base of the RFZ (from 28.1% to 23.8%; 5.1% shift) is 322 immediately followed by a stepwise positive shift to 28‰ at 20 m in the lower RFZ (= upper C4 323 segment). The values remain relatively constant (mean= 27‰) towards the middle part of the 324

C5+C6 segment at 30 m, which is followed by a negative excursion to 25.7‰ at ~34 m, a positive
shift to 27% at 36 m, and finally a negative excursion to 25.5‰ on top of the studied interval.

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328 4.2. Total organic carbon

TOC values of the studied Lower Dariyan unit are relatively low (mean= 0.8 wt. %) and range 329 from 0.1 to 3.0 wt. % (Table S1, Fig. 4). The highest values are associated with black argillaceous 330 and thin-bedded limestones. Segments C2 and C3 have relatively constant, low TOC values 331 332 (mostly between 0.2 and 0.4 wt. %; max. 0.9 wt. %). The lower C4 segment (up to the base of 333 RFZ) is characterized by overall increasing values with two prominent positive excursions at ~ 13 m (2.2 wt. %) and ~17 m (3.0 wt. %). The latter value represents the highest TOC in the section, 334 which is recorded <1 m below the base of the RFZ (Fig. 4). The TOC values of the RFZ then 335 progressively decrease from 0.5 wt. % at 18 m (~middle C5) to 0.4 wt. % at 24 m (basal C5) with 336 a positive excursion to 2.4 wt. % at 23 m. Pronounced, stepwise increase in values characterizes 337 the remaining upper part of the RFZ, with the maximum value of 2.5 wt. % immediately above the 338 RFZ at 27.1 m (lower C5+C6 segment). The upper part of the section exhibits more heterogeneous 339 values with initially decreasing values to 0.2 wt. % at 30 m, which is followed by a stepwise 340 increasing trend to a maximum of 1.9 wt. % at 34 m. After a drop to 0.2 wt. % at 35 m, the 341 remaining part of the section exhibits a gentle stepwise increasing trend with maximum values 342 343 below 1.3 wt. %.

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345 **4.3. Major and trace elements**

The triangular $Al_2O_3 - SiO_2 - CaO$ diagram (Fig. 5A) shows that the majority of the samples plot at or below the 'carbonate dilution line', which connects the average shale (AS) point to the pure carbonate end-member and exhibits a tendency towards higher SiO₂ contents. The highest SiO₂ and the lowest CaO contents are recorded in segments C5+C6, especially in samples from the upper part of the RFZ (Table S2). Only three samples from the top of the studied interval (segment C6) show fair enrichment in Al content (Al₂O₃ (x5) content > 20 %; Fig. 5A).

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4.3.1. Detrital proxies 353

The titanium content of the analyzed samples varies from 0.01 to 0.78 wt. % (mean= 0.02 wt. %), 354 with the highest values recorded in the topmost part of the section (upper segment C5+C6); the 355 remaining section generally has much lower Ti values (mean=0.01 wt. %). The potassium content 356 357 ranges from 0.02 to 0.25 wt. % (mean= 0.09 wt. %). It mirrors the Ti trend, with the highest values 358 in the same interval (upper segment C5+C6), whereas the remainder of the section has mean K content of 0.07 wt. %. The Si content varies from 0.35 to 13.6 wt. % (mean= 4.8 wt. %), but the 359 values are ~5x greater in segment C5+C6 (7.52 wt. %) compared to segments C2 to C4 (mean= 360 1.56 wt. %). The above-selected lithogenic elements (Koinig et al., 2003) exhibit strong (Ti; R= 361 362 +0.98), moderate (K; R = +0.54) to weak (Si; R = +0.44) correlation with Al (wt.%) (Fig. 5B–D). Conversely, linear plots of K, Si, and Ti with Ca (%) indicate a moderate negative correlation 363 364 between the three elements (Fig. 5E–G).

365 The Al-normalized values of selected lithogenic elements (Si/Al, Ti/Al, and K/Al) are displayed in Figure 6 and Table S3. The K/Al and Ti/Al values are lower than AS values in the 366 same 2 intervals: 0-12.8 m (Segments C2 to basal C4), and 25.1-38.8 m (C5+C6, excluding its 367 basal part). Within the interval with most values > AS (lower C4 through basal C5+C6), three 368 peaks with the highest values are identified at 14.1 m (lower C4), 17.1 m (middle C4, right below 369 the RFZ), and 20.8 m (basal C5). The highest values below the RFZ base correspond to relatively 370 371 high TOC (2.58 %), although the highest TOC value (2.96 %) immediately below correlates with 372 low both K/Al and Ti/Al ratios. The Si/Al ratios are similar to slightly above the AS values up 373 until the mid-segment C4, i.e., right below the RFZ. Further up-section, with the exception of three values similar to AS, they are much higher with three peaks showing >10x greater values than AS 374 (at 20.1 m, 23.8 m, and 31.1 m). The remaining upper C5+C6 segment exhibits a gradual return to 375 376 AS values. In addition, the relatively low values of K, Si, and Ti at 16.8 m (mid-C4) coincide with 377 comparatively high TOC (2.58 %; Fig. 6; Table S3).

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4.3.2. Redox- and productive-sensitive trace elements

The selected RSTE (U, V, Mo) and PSTE (Cu, Ni, P) investigated in this study are shown in Table 380

S2, and normalized to Al in Figure 6 and Table S3. U/Al and V/Al values are relatively low from 381

the base of the section to 13.1 m (segments, C2, C3, and basal C4). This is followed by a gradual 382 increase in V/Al to a maximum value of 4130 at 23.1 m (top C4) and a subsequent decrease. The 383 384 corresponding U/Al values show a stepwise increase to a maximum of 83.8 in the mid-C4, 385 immediately below the base of RFZ; this is followed by a gradual decrease through upper C4 and lower C5+C6. Both U/Al and V/Al show relatively uniform values slightly greater than AS above 386 387 23.1 m (above RFZ). Contrary to U/Al and V/Al, the Mo/Al curve is characterized by more fluctuating values with several excursions to values significantly greater than AS (at 7.8 m in 388 389 segment C2, and 27.8 m and 31.8 m in segment C5+C6). Note that the entire upper part of segment C4 has high Mo/Al values, except for a single low point at 16.8 m (Fig. 6). All three high RSTE 390 values correspond to high TOC at 17.1 m (mid-C4, below the RFZ base) and 23.1 m (top C4). 391 Otherwise, the highest TOC values generally do not match all three (or two) RSTE highs. 392

393 Cu/Al and Ni/Al are characterized by low, relatively uniform values in segments C2 and C3 (0–10.8 m), and above 25.1 m (basal C5+C6) (Fig. 6). Both exhibit a smaller peak in lower 394 395 segment C4, and the highest value around the C4–C5 boundary (Ni/Al= 341.4, Cu/Al= 999.4); Ni/Al also shows a third peak with higher values in mid-C4 below the RFZ. From the base to ~ 11 396 397 m (C2 and C3 segments), there is a consistent decreasing trend observed in the P/Al ratios, with values from 0.2 to 0.05 (mean= 0.11). There are two distinct and prominent peaks in P/Al content, 398 399 one in basal C4 (0.52) and the other one at 20.8 m in the upper C4 (0.53). Subsequently, the P/Al 400 content in the C5+C6 segment is lower and fluctuates from 0.04 to 0.12, with four minor peaks of up to 29.1; the lowest of these positive excursions occurs at the RFZ top. 401

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403 **5. Discussion**

404 5.1. Fluctuations in TOC and δ^{13} C and assessment of primary values

The Lower Dariyan TOC values from the studied succession suggest that starting from segment C4, conditions in the Kazhdumi Basin were periodically favorable for the preservation of the organic matter (TOC>1 %), especially immediately preceding and during the RFZ deposition (RFZ values from 4.5 to 9 %; Bordenave and Hegre, 2010). This is within a range of values recorded from the Sub-Briançonnais basinal marls and limestones that deposited along the Iberian continental margin (TOC in segments C2–C6 ranges between 0 and 6 %; Charbonnier et al., 2018), the epicontinental Ionian Basin deposits (TOC also range between 0 and 6 %; Karakitsios, 1995), 412 or from other coeval intrashelf or deep-pelagic basinal sections worldwide (Castro et al., 2021, and 413 references therein). It is important to note, that the organic-rich deposits were geographically 414 restricted and characterized by a diachronous occurrence between different basins and carbonate 415 platforms (e.g., Huck et al., 2010; Föllmi, 2012; Husinec et al., 2012; Mutterlose et al., 2014; Blok 416 et al., 2022).

A moderate to strong relationship (R=+0.66) between $\delta^{13}C_{org}$ and $\delta^{13}C_{carb}$, similar scatter of values 417 (4.6‰ in $\delta^{13}C_{org}$, and 3.7‰ in $\delta^{13}C_{carb}$; Fig. 7A), and no to very week relationship (R= +0.08) 418 between $\delta^{13}C_{org}$ and TOC (Fig. 7B), suggests that the paired primary values are generally preserved 419 albeit likely affected to a certain degree by local environmental processes and diagenesis (cf., 420 Edwards and Saltzman, 2016). Although the Lower Dariyan $\delta^{13}C_{org}$ trends appear to be broadly 421 covariant with $\delta^{13}C_{carb}$, the decoupling in segment C3 and in the middle part of C5+C6 (Fig. 4) 422 suggests that the $\delta^{13}C_{org}$ signal may have periodically been modified by biological or geological 423 processes. This is likely because $\delta^{13}C_{org}$ is commonly subjected to a wide range of factors that can 424 425 affect its values, such as changing contributions of terrestrial and marine organic carbon, variations in the C-isotopic composition of the dissolved CO₂ metabolized by phytoplankton during carbon 426 427 fixation, changing fractionation of C-isotopes during photosynthesis, and changes in the surface ocean carbon reservoir (e.g., Lini et al., 1992; Sluijs and Dickens, 2012). 428

429 The positive shifts in $\delta^{13}C_{org}$ observed in the RFZ (mid-C4 to lower C5+C6), which pair well with the positive shifts in $\delta^{13}C_{carb}$ (Fig. 4), suggest a global increase in the burial rates of 430 431 organic carbon, potentially leading to falls in atmospheric pCO₂ (Kump and Arthur, 1999). The 432 shifts could have been driven by a range of factors, including increased productivity, decreased dissolved CO_2 in surface waters, as well as by changes in the organic material sources (Hatch et 433 al., 1987). A significant decrease in Δ^{13} C values associated with a paired increase in δ^{13} Corg and 434 $\delta^{13}C_{carb}$ in the lower RFZ, suggests that it was likely driven by a productivity increase in surface 435 waters (Patzkowsky et al., 1997), as evidenced by a contemporaneous increase in the abundance 436 of planktonic foraminifera and radiolarians (Jafarian et al., 2023). 437

The observed paired ($\delta^{13}C_{org}$ and $\delta^{13}C_{carb}$) negative excursions are not the result of a meteoric alteration of $\delta^{13}C_{carb}$ record (cf., Oehlert and Swart, 2014), as suggested by a lack of evidence for subaerial exposure within the studied succession (Jafarian et al., 2023). This is also supported by the lack of correlation on $\delta^{13}C_{org}$ vs. TOC and $\Delta^{13}C_{org}$ vs. TOC cross plots (Fig. 7B, C), as isotopically heavy bulk organic carbon values do not show consistently low TOC related to
increased respiration by either biological or geological processes, including thermal degradation
of organic matter during burial diagenesis (Popp et al., 1998; Hayes et al., 1999).

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5.2. Provenance of non-carbonate sediment fraction

447 The triangular phase diagram $(Al_2O_3 - SiO_2 - CaO; Fig. 5A)$ supports background sedimentation 448 dominated by benthic and pelagic carbonate skeletal fragments and an increased biogenous silica component due to the abundance of radiolarians and sponge spicules, especially within the RFZ. 449 This is also supported by a simultaneous rise in the abundance of productivity proxies (Cu/Al, 450 451 Ni/Al, and P/Al; discussed later). The very low Ti/Al, K/Al, and Si/Al values indicate pure 452 carbonate deposition during stages C2, C3, and basalmost C4. Since quartz and heavy minerals are commonly found together in coarser grain-size fractions and are indicative of high-energy 453 454 environments (e.g., Dellwig et al., 2000), the higher Ti/Al ratios that correspond with increased Si/Al ratios within the lower C4 segment are suggestive of the presence of detrital quartz (Hetzel 455 et al., 2011). The higher SiO₂ content within the interbedded shales of the uppermost Lower 456 457 Dariyan unit (segment C5+C6; Table S2) was due to the presence of biogenous silica, as also suggested by very low Ti/Al and K/Al, and high Si/Al values. 458

Strong Ti to moderate K positive correlation with Al suggests detrital input of these lithogenic 459 460 elements, whereas weak Si-Al correlation indicates that the SiO₂ source was both biogenous silica and detrital (Fig. 5B–D). Given that the Ti concentration is considered as a proxy for the input of 461 462 coarse- to medium-grained, and K as a proxy for fine-grained detrital fractions (Gambacorta et al., 463 2020 and references therein), the strong Ti to moderate K correlation with the fine-grained proxy 464 (Al) suggests an influx of both the coarse- to medium-grained and fine-grained detrital fractions into the basin. Siliciclastic input onto the markedly asymmetrical Lower Cretaceous Arabian 465 466 Platform was most pronounced during periods marked by concurrent tectonic uplift in the 467 hinterland, humid or seasonally wet climates, and significant eustatic falls (Yose et al., 2006; Davies et al., 2019). Accordingly, the uplift had exposed basement rock types and older 468 sedimentary rocks to weathering and erosion, influencing the slope profile and subsequently 469 470 affected sediment transport rates to the shelf, which is evidenced by the accumulation of thick sandstone-dominated units (e.g., Zubair and Burgan formations; Davies et al., 2019). In the 471

Kazhdumi Intrashelf Basin, tectonic control is suggested by the activation of the Kazerun Fault
system during the late Early Aptian (van Buchem et al., 2010). In the absence of an exposed
siliciclastic area in the vicinity of the basin (Fig. 1), it seems plausible that the very fine siliciclastic
detritus was blown in by the wind.

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5.3. Palaeoredox conditions in the Kazhdumi Intrashelf Basin

RSTE enrichment and their occurrence within the organic-rich sediments are commonly utilized 478 479 as a proxy for identifying the palaeoredox conditions and the changes thereof during deposition 480 and early diagenesis (Boning et al., 2004; Rimmer, 2004; Brumsack, 2006; Tribovillard et al., 2006). Thus, they reflect redox conditions of bottom waters and/or porewaters close to the 481 sediment water interface (Algeo and Li, 2020). Typically, RSTEs are more soluble under oxidizing 482 conditions and can be removed from the water column to the sediments either by biotic (mainly 483 484 phytoplankton) or abiotic processes, the latter particularly efficient under reducing conditions (Tribovillard et al., 2006). For instance, in anoxic environments, uranium is reduced and fixed in 485 marine carbonates as insoluble uranous (U^{4+}) , while in oxidizing conditions uranium ions keep the 486 higher oxidation state (U^{6+}) and form uranyl carbonate ($UO_2(CO_3)_3^{4-}$), which is soluble in ambient 487 water. Accordingly, anoxic deposits are normally more U-enriched than oxic sediments (Wignall 488 and Twitchett, 1996). Similar to uranium, Mo and V are also incorporated into the sediments under 489 490 the reducing conditions, either by precipitation as oxides and hydroxides, by the creation of organo-metal ligands in humic acids, or by surface adsorption (Algeo and Maynard, 2004; 491 492 Tribovillard et al., 2004). Furthermore, U and V can be reduced and accumulated in the sediments 493 under denitrifying settings (suboxic to anoxic), while Cu, Mo, Ni, and Zn are chiefly enriched 494 under sulfate-reducing environments (i.e., under a more advanced stage of deoxygenation; Tribovillard et al., 2006). Therefore, if U and V are enriched without the concurrent Mo 495 496 enrichment, then the bottom waters or porewaters were suboxic to anoxic without free hydrogen 497 sulfide (H₂S), whereas sediments that show simultaneous enrichments in U, V, and Mo are indicative of anoxic to euxinic conditions either at the sediment-water interface or in the water 498 column (Algeo and Maynard, 2004; Tribovillard et al., 2006). On the basis of our results, the 499 500 setting was non-euxinic as suggested by, except for two outliers, Mo concentrations <25 ppm (Scott and Lyons, 2012; Table S2). However, as noted by Algeo and Liu (2020), no single proxy 501

should be taken as a universally reliable redox indicator, albeit some are more dependable; inparticular, TOC and trace metal enrichment factors.

Lithologically, segments C2 through basalmost C4 are dominated by frequently 504 bioturbated, grayish-white to light gray, medium-to-thick bedded mollusk wackestone-floatstone 505 506 and packstone-rudstone with ostreids, rudists, benthic foraminifera, echinoids spines, and crinoids 507 (Jafarian et al., 2023). This, coupled with low TOC and low RSTE (Fig. 6; Table S3), suggests well-oxygenated conditions in the Kazhdumi Intrashelf Basin before and at the onset (C3 segment) 508 of the OAE 1a. Whether an increased value of Mo in the upper C2 indicates a short period of Mo 509 removal into sediments under reducing conditions is uncertain, as it is not matched with an increase 510 511 in other RSTE or TOC. The C4 segment is characterized by a predominance of TOC values averaging at 1.58 wt. %, indicating the periodic existence of a reducing environment that facilitated 512 513 the intermittent preservation of organic matter. The segment is characterized by the predominance of dark gray, thin-bedded and finely (millimeter-scale) laminated radiolarian mudstone, 514 515 wackestone and packstone with chert bands, organic-rich interbeds, and pyrite framboids (Jafarian et al., 2023). Within the C4 (excluding its basalmost part) to basalmost C5 segment there are three 516 517 intervals of significant enrichment in V, Mo, and U, which are relatively well correlated with some of the highest observed values of TOC in these sediments (Fig. 6). This co-occurrence of RSTE 518 519 enrichment within strata with higher TOC suggests anoxic but non-sulfidic conditions where the 520 uptake of RSTE was mainly limited by the availability of organic substrates (Algeo and Maynard, 2004; Stein et al., 2011). The latter is also suggested by the contemporaneous enrichment in Ni 521 522 and Cu, both of which occur in organic phases and may contribute to pyrite formation (Tribovillard 523 et al., 2006). Given that pyrite framboids precipitate relatively quickly under diffusion-limited 524 conditions and from waters at the supersaturation limit with respect to FeS₂ (Emmings et al., 2022), the local presence of pyrite framboids ($\sim 2-8 \mu m$; Jafarian et al., 2023) suggests that 525 526 euxinic conditions may have existed for limited periods of time within microenvironments, ambient bottom waters or porewaters. The majority of trace metals enriched in this succession can 527 528 form sulfides, leading to the assumption that sulfidic (euxinic) conditions could have been present at or near the sediment-water interface in some short periods, and that reductive mobilization 529 trapped these trace metals in the sediments (e.g., Hetzel et al., 2011; Sanchez-Hernandez and 530 Maurrasse, 2015). That the redox conditions during the lower C4-basal C5+C6 were periodically 531 532 suboxic is suggested by the intervals with decreased individual RSTE values and decreased TOC

(Fig. 6). The remaining studied succession (C5+C6) exhibits variations in TOC, where some of the higher TOC values correlate with increased Mo, albeit V and U remain low. Coupled with lithological characteristics (predominantly dark to light gray, mud-supported radiolarian and planktic foraminiferal limestone and marl), this suggests an oxic-suboxic syngenetic or early diagenetic setting for the C5+C6 segment.

538 Modern seawater authigenic Mo/U covariation in oxygen-depleted systems is influenced by variation in benthic redox conditions, the operation of particulate Mn-Fe-oxyhydroxide shuttle 539 within the water column, and trace-metal chemistry of the seawater (Algeo and Tribovillard, 540 2009). To evaluate trace-element enrichment in ancient and modern sediments, their 541 542 concentrations are expressed as enrichment factors (EF) (e.g., Calvert and Pedersen, 1993; Tribovillard et al., 2004, 2006; Brumsack, 2006; Algeo and Tribovillard, 2009). EF is simply 543 544 determined by normalizing an element to Al, which is considered as a proxy for detrital influx, and relative to the AS (Wedepohl, 1971, 1991). Although both Mo and U are enriched in sediments 545 546 under comparable water-column conditions, their geochemical behaviors differ significantly during the sedimentation process. The correlation between Mo-EF and U-EF can be used as a 547 548 practical proxy to investigate the level of oxygen depletion relative to the Mo/U ratios in modern seawater (Algeo and Tribovillard, 2009). EFs for Mo and U in the Lower Aptian Kazhdumi Basin 549 550 deposits were calculated (EF_{element} = (element/Al)_{sample}/(element/Al)_{AS}; Wedepohl, 1971; Taylor and McLennan, 1985; Table S3). The studied Lower Dariyan Fm. is characterized by a significant 551 552 enrichment in authigenic Mo (mean=155; samples from C2, C4 and C5+C6 segments with values over 300) and U (samples fall within the range of 3 to 200), with a weak (R = +0.48) positive 553 554 correlation for the entire dataset (Fig. 8). These values yield Mo/U ratios generally 0.3 to 3 times 555 the seawater ratio, albeit most samples are considerably more enriched in Mo relative to U.

In marine sediments, the absorption of authigenic U begins at the redox boundary of Fe(II)-Fe(III) under suboxic conditions (Zheng et al., 2002). Conversely, the absorption of authigenic Mo necessitates the existence of H_2S under anoxic to euxinic conditions (Zheng et al., 2000; Morford et al., 2001). Variations in sediment Mo/U ratios may occur as a result of differences in solubility attributable to redox conditions, with suboxic settings potentially yielding low Mo/U ratios and more anoxic settings potentially yielding high Mo/U ratios (Algeo and Tribovillard, 2009). The highest enrichment of U relative to Mo is recorded in the C4 sample immediately below the RFZ,

which suggests a strongly restricted basinal environment in that interval. A subset of samples from 563 the C4 segment with high TOC values exhibits the overall highest EFs with increasing Mo/U ratio 564 565 around the 1x seawater molar ratio (Fig. 8). A similar pattern has previously been recorded from the deep-water OAE 1a deposits from the Subbriançonnais Basin (Charbonnier et al., 2018). Such 566 covariation pattern is more similar to the present-day Peru Shelf, E. Tropical Pacific, than to the 567 568 euxinic setting of the Black Sea, and is suggestive of greater redox variation, i.e., from suboxic to sulfidic (Algeo and Tribovillard, 2009; see their Fig. 4). A similar trend is exhibited by a subset of 569 570 samples from segment C5+C6 with Mo/U ratios around the seawater value (1x SW). However, a second subset of samples from C5+C6, all samples from C3, and the majority of samples from C2 571 (Fig. 8; data points around and above the 3x SW line) exhibit a strong enrichment of Mo relative 572 to U. This is suggestive of the particulate Mn-Fe oxyhydroxide shuttle, which affects Mo but does 573 not affect the accumulation of U (e.g., Crusius et al., 1996; Algeo and Tribovillard, 2009; 574 Tribovillard et al., 2012). It is a mechanism by which the particulate Fe-Mn oxyhydroxides that 575 576 are formed at the chemocline absorb molybdate oxyanions as they settle through the water column, 577 and are then scavenged at the seafloor thus leading to Mo enrichment relative to other RSTE in 578 the sediment (Algeo and Tribovillard, 2009).

579 Finally, it is possible that, as a consequence of high-frequency changes in the redox 580 conditions, the primary Mo/U relationships were modified by the early post-depositional 581 remobilization of Mo and U (Algeo et al., 2004; Algeo and Tribovillard, 2009) but the extent of that is unknown. The overall shotgun-like pattern of Mo/U data clustering with only a weak trend 582 583 towards higher EFs with increasing Mo/U ratios suggests highly variable redox conditions in the 584 Kazhdumi Intrashelf Basin. Given the basin's moderately restricted setting along the NE Arabian 585 Plate margin, the connectivity and the watermass exchange between the basin and the Neo-Tethys likely controlled the variations in benthic redox diversity. 586

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588 5.4. Evaluation of palaeoproductivity based on trace element-TOC covariations

589 Copper and nickel are important micronutrients that are usually sourced from the settling organic 590 debris but can also be absorbed into Mn-Fe oxyhydroxides (e.g., Calvert and Pedersen, 1993; 591 Algeo and Maynard, 2004; Tribovillard et al., 2006). When sediment experiences reducing 592 conditions, these trace metals can get incorporated into Fe sulfides or form their own sulfide phases (Hetzel et al., 2011). Despite complete remineralization of organic matter, Ni and Cu can still be
retained in sediment through sulfides (typically as a solid solution within pyrite). As a result, Cu
and Ni are reliable indicators of the delivery of organic matter to the sediment (Algeo and
Maynard, 2004).

597 Following the low Cu and Ni values recorded in C2 through basalmost C4 segments, the 598 remaining C4 to basal C5+C6 segments are characterized by significant increases in Cu and Ni that match higher TOC values (Fig. 6), suggesting periodically increased organic matter influx. 599 600 The initial increase slightly trailed the increase in phosphorus content, which triggered the increase in primary productivity (discussed below), as evidenced by the abundance of radiolarians and 601 602 planktic foraminifera, particularly within the C4 to C5+C6 segments (Jafarian et al., 2023). The matching Si/Al and PSTE peaks (Cu/Al and NiAl; Fig. 6) indicate periodically high biogenic 603 604 productivity from mid-C4 to basalmost C5+C6 segments, especially immediately before and during deposition of the RFZ. It is important to note that not all of the biogenic silica generated by 605 606 the radiolarians was deposited because of the likely dissolution occurring in the overlying water 607 column (Wollast and Mackenzie, 1983).

608 Phosphorus serves as a crucial nutrient and energy carrier across various levels of the biosphere, and periods of extensive primary productivity and terrestrial biogeochemical 609 weathering are commonly characterized by significant increases in this element (e.g., Föllmi, 610 611 1996; Compton et al., 2000; Bodin et al., 2006). The effectiveness of phosphorus storage in a 612 sedimentary basin relies on the redox state, and when oxygen levels are depleted, a substantial 613 amount of phosphate can be recycled and returned into the bottom-waters (Van Cappellen and Ingall, 1994; Bodin et al., 2006; Mort et al., 2007; Beil et al., 2020). Phosphorous is transferred to 614 the sedimentary reservoir either through its assimilation into the organic matter or by its 615 616 aggregation with clay particles and Fe-Mn oxyhydroxides (Delaney, 1998).

The productivity played an important role in the development of anoxic conditions in the Kazhdumi Basin as further suggested by a strong increase in phosphorus content in the basal C4 segment that matches an increase in TOC, thus signaling an increase in nutrient availability. That this increase was not limited to this part of the Tethys is suggested by a coeval peak in TOC and phosphorus recorded in the Subbriançonnais Basin (Charbonnier et al., 2018; their Fig. 2). This correlates well with the previously suggested increase in nutrient availability that occurred in the

Kazhdumi Basin during the C3–C4 transition, as suggested by a lithological change from ostreid-623 624 dominated shallow-marine carbonates to deeper-water dark-gray limestones, chert bands, and 625 organic-rich shales with radiolarians, sponge spicules, and planktic foraminifera (Jafarian et al., 2023). Stein et al. (2011) argued that such increased input of phosphorous into the ocean acted as 626 a fertilizer for the southern Tethyan margin during the OAE 1a, which led to the growth of 627 628 radiolarian populations, particularly in the areas with significant upwelling. Although the latter was possible, in the absence of riverine inputs and lack of evidence for extensive upwelling, it is 629 likely that aeolian transport also played an important role in delivering terrestrial particulate 630 phosphorus (Mackey and Paytan, 2009) into the Kazhdumi Intrashelf Basin. The decrease in 631 phosphorus within the upper RFZ suggests that the high primary productivity was sustained by 632 phosphorus recycling from the sediments, which in turn led to enhanced carbon burial and 633 increased $\delta^{13}C_{carb}$ values recorded during that interval (cf., Beil et al., 2020). Several subsequent 634 peaks in phosphorus values also correlate with smaller increases in TOC. Given the overall lack 635 636 of association between lower phosphorus and higher RSTE values, it is unlikely that a significant part of phosphorus was remobilized in the bottom-waters by oxygen depletion (Charbonnier et al., 637 638 2018).

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640 5.5. Causes of variations in intrashelf basin bottom water conditions during OAE 1a

The OAE 1a represents a major episode of climatic, biotic, environmental, and climatic 641 642 perturbations related to a global warming event, increased marine productivity, oxygen-depleted 643 bottom water, and increased burial of organic matter in deep-water settings (e.g., Schlanger and 644 Jenkyns, 1976; Weissert and Bréheret, 1991; Jenkyns, 2010; Erba et al., 2015; Castro et al., 2021; Percival et al., 2021). However, its record is characterized by spatial variability in redox conditions 645 likely related to paleogeography, freshwater input, and depth, with deeper parts of the Tethys 646 characterized by anoxic-euxinic conditions and less reducing conditions in shallower 647 648 environments (Westermann et al., 2013; Nie et al., 2023). The OAE 1a onset was characterized by a massive input of isotopically light carbon into the ocean, as evidenced by a major negative $\delta^{13}C$ 649 excursion (Schlanger and Jenkyns, 1976; Arthur et al., 1990) in segment C3 sensu Menegatti et al. 650 651 (1998). Climatic instability characterized the period before (C2) and the onset (C3) of the OAE 652 1a, and the initial pulses of volcanic activity predated the main phase (Martínez-Rodríguez et al.,

2021). Also, the weathering regime was characterized by physical weathering and/or intensified 653 654 erosion under drier and/or markedly seasonal rainfall (Beil et al., 2020). Intensified greenhouse 655 conditions at the onset of the OAE 1a led to enhanced continental weathering, albeit the bottom 656 water oxygen levels were high enough to break down the organic matter through the water column (Westermann et al., 2013). The plate-marginal position of the Kazhdumi Intrashelf Basin restricted 657 it from significant detrital input during segments C2 through basalmost C4, as evidenced by 658 659 predominantly carbonate succession with low TOC. Abundant benthic biota, especially orbitolinids and ostreid bivalves, and extensive bioturbation, coupled with low RSTE suggest oxic 660 bottom water, and low PSTE are indicative of low productivity. The lag in nutrient increase, which 661 in deeper intrashelf basins with predominantly marly deposits is recorded by higher phosphorus 662 values already in C3 (e.g., Vocontian Basin; Giraud et al., 2018; Beil et al., 2020), was likely due 663 664 to paleogeographic setting of the basin along the NE Arabian Plate margin and its connectivity with the Neo-Tethys Ocean. 665

Segment C4 is associated with a global cooling episode with upper water column 666 temperatures dropping by $\sim 2^{\circ}$ C in the middle of C4, which is expressed as a positive shift in the 667 oxygen-isotope record, and a correlative major positive shift in δ^{13} C signaling extensive deposition 668 of organic matter (Jenkyns, 2018). The latter was enabled by the accelerated continental 669 weathering rates (Blätter et al., 2011; Bottini et al., 2012; Lechler et al., 2015) that resulted in 670 enhanced nutrient flux to the ocean and the associated increase in marine productivity. It is 671 important to note that the global record of organic carbon burial documented by the positive δ^{13} C 672 excursion is more representative for the duration of the OAE 1a than the stratigraphic extent of 673 674 local black shales (Beil et al., 2020). Also, an increase in the proportion of carbonate weathering, relative to organic carbon and silicate weathering, could have maintained prolonged positive 675 marine δ^{13} C excursions without substantially enhancing organic carbon burial (Kump and Arthur, 676 1999). On the Arabian Plate, the regional Early Aptian transgression (Aptian Supersequence with 677 maximum flooding surface K80; Sharland et al., 2001; van Buchem et al., 2010) promoted input 678 of nutrients derived from the weathered terrestrial sediments (e.g., Al-Fares et al., 1998; van 679 680 Buchem et al., 2002). An increase in continental weathering was promoted by a hot and humid or seasonally wet climate (Davies et al., 2019) with an enhanced hydrological cycle. Nutrient 681 682 availability boosted primary productivity and contributed to the development of the RFZ with

anoxic-suboxic bottom waters during the lower C4 to basalmost C5+C6 segments, as evidenced 683 by increased TOC and phosphorus in the Kazhdumi Basin and globally (e.g., Charbonnier et al., 684 685 2018). A significant deepening was in the basin also associated with a major biotic shift from shallow to deep open-marine biota dominated by radiolarians and planktic foraminifera. In the 686 shallower Bab Basin further south along the plate margin as well as on the Arabian carbonate 687 platform, the deepening was associated with the expansion of the microencrusters *Lithocodium*-688 Bacinella during segment C4 and continued through C6 (Immenhauser et al., 2005; Rameil et al., 689 2010; Yamamoto et al., 2013). The Lithocodium-Bacinella bloom has been associated with 690 platform top anoxia during the OAE 1a, apparently driven either by the upwelling of oxygen-691 depleted basinal waters or by the development of localized oxygen minimum zones (Hueter et al., 692 2019). The latter is supported by localized findings of dysaerobic platy lime mudstones on the 693 694 Adriatic Carbonate Platform that extend from the upper C4 to C7 (Husinec et al., 2012).

695 Following a significant global warming during the upper part of segment C4 and into lower C5, at least one cooling episode is globally recognized in segment C5+C6 (Jenkyns, 2018), 696 whereas fluctuations between drier and wetter climate are locally recorded into the lower C7 (Beil 697 et al., 2020). In the Kazhdumi Basin, this interval is characterized by a relative sea-level fall 698 (Jafarian et al., 2023), during which the major SiO_2 source was biogenous silica, as suggested by 699 low Si/Al and K/Al values (Fig. 6), and relatively high productivity was maintained by phosphorus 700 recycling. Oscillations in TOC, RSTE, and PSTE suggest periodic changes between oxic and 701 702 suboxic conditions, where the water column ventilation may have moderated TOC by organic matter destruction (cf., Nie et al., 2023). Overall, the unrelenting global productivity led to CO₂ 703 704 drawdown, more oxic conditions, and a post-OAE 1a climate overturn towards less humid 705 conditions (Westermann et al., 2013).

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707 Conclusions

A bio- and chemo-stratigraphically well-constrained succession of the Kazhdumi Intrashelf Basin exposed at Dareh Sefid, Fars Province, Iran was utilized to evaluate the trends in organic matter production, palaeoredox conditions, terrestrial input, and paleoproductivity during the OAE 1a (including carbon-isotope segments C2–C6 *sensu* Menegatti et al., 1998). The results suggest that the highly variable redox conditions in the Kazhdumi Basin during the OAE 1a were primarily controlled by the connectivity and watermass exchange with Neo-Tethys, and assisted by the nutrient input associated with continental weathering.

Carbon-isotope segments C2 through basalmost C4 are characterized by predominantly carbonate deposition with low TOC, low RSTE, and low PSTE. This, coupled with abundant orbitolinids, ostreid bivalves, and bioturbation, indicates oxic conditions prior to and at the onset (C3 segment) of the OAE 1a. The intermittently suboxic conditions are suggested by the intervals with increased RSTE or TOC. An increase in nutrient availability was marked by a significant increase in phosphorus content in the basalmost C4, which was matched by an increase in TOC.

721 The Kazhdumi Basin section spanning lower C4 through basalmost C5+C6 segments has 722 several intervals with co-occurrence of high RSTE, PSTE (Cu, Ni), and TOC, indicating anoxic, albeit non-sulfidic conditions with high biogenic productivity, primarily by radiolarians and 723 724 planktic foraminifera. That the surface-water productivity was driving the productivity increase is suggested by a significant decrease in Δ^{13} C values that correlates with a paired increase in δ^{13} Core 725 and $\delta^{13}C_{carb}$ in the lower RFZ. The decrease in the phosphorus content in the upper RFZ (mid-C4) 726 suggests that the high primary productivity was also periodically sustained by phosphorus 727 recycling from the sediments. 728

Segment C5+C6 (excluding its basalmost part, i.e., the top of RFZ) is part of a regressive sequence of predominantly dark gray, mud-supported radiolarian and planktic foraminiferal wackestone. The intermittently oxic-suboxic setting is indicated by variable TOC (0.2 to 2.5%), where some of the higher TOC values tie with increased molybdenum, albeit vanadium and uranium contents remain low. Both biogenous and detrital silica contributed to an increase in the SiO₂ content.

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736 Acknowledgements

This study presents the results of AJ's Ph.D. thesis and was financially supported by the National
Natural Science Foundation of China (Grant Nos. 41888101, 42072118). Funding for stable
isotope analysis of organic carbon was kindly covered by the Stable Isotope Biogeochemistry
Laboratory (SIBL) at Durham University. The senior author would like to thank Dr. G.C.

Gambacorta for valuable discussions. We thank editor Alex Dickson and three anonymousreviewers for their constructive comments.

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- 1162 Cosmochimica Acta 64, 4165-4178.
- 1163
- 1164 **Figure and Table captions**



Fig. 1. (A) Palaeogeographical map of the Arabian Plate during the Aptian (van Buchem et al.,
2010). (B) Generalized chronostratigraphy, biostratigraphy, and lithostratigraphy chart of the
Barremian–Lower Albian of the Zagros Mountains, Iran (van Buchem et al., 2010).



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Fig. 2. Outcrop photographs of the Aptian Dariyan Formation at the Dareh Sefid section. (A) 1170 1171 Panoramic view shows the location of the Dareh Sefid village relative to the hillside exposures of the Dariyan Fm. (B) Closer view of the conformable contacts between the Dariyan and Gadvan 1172 Fms. and between the Sarvak and the Kazhdumi Fms. on top of the Dariyan Formation. (C) Close-1173 up view of the Lower Dariyan contact with Kazhdumi Tongue. Note the color and lithology change 1174 1175 at the contact between Lower Dariyan unit (light gray thin-bedded limestones with frequent black chert bands (Bcb)) and Kazhdumi Tongue (middle unit; greenish gray shales). (D) Shelly 1176 limestone bed, Lower Dariyan unit. 1177



Fig. 3. Thin section photomicrographs of typical facies under plane-polarized light (PPL). (A)
radiolarian wackestone-packstone, MF1. (B) planktonic foraminiferal wackestone, MF2. (C)
Bioclast (coral) floatstone, MF3. (D) Bioclast packstone and rudstone, MF4.



Fig. 4. Stratigraphic column of the Lower Dariyan unit at the Dareh Sefid section showing lithology, carbon-isotope stages *sensu* Menegatti et al. (1988), chemostratigraphy, TOC content, major microfossils, and planktic foraminifera biozonation. The radiolarian flood zone (RFZ) is highlighted in yellow. Thick red lines represent 5-point moving averages. $\delta^{13}C_{carb}$ data and major microfossils from Jafarian et al. (2023). Tethyan planktic formainifera biozones after Castro et al. (2021). For location of the Dareh Sefid section see Figure 1.



Fig. 5. (A) Ternary diagram shows the relative proportions of Al₂O₃ (multiplied by 5; representing clay minerals), SiO₂ (representing quartz or biogenic silica), and CaO (representing carbonate) for samples from the Dareh Sefid section. The average shale (AS) composition (Wedepohl, 1971, 1991) and the carbonate dilution line are also included for reference. (B-G) Cross plots of selected lithogenic conservative elements (K, Ti, and Si) versus Al and Ca concentrations.



Fig. 6. The stratigraphic correlation of major and trace elements, TOC, and $\delta^{13}C_{org}$ which have been normalized to aluminum concentration, is paired with the subdivision of carbon isotope segments of Menegatti et al. (1988), in the Dareh Sefid section. The Al-normalized concentrations of each element in AS (Wedepohl, 1971, 1991) are shown by thin red lines.



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Fig. 7. Cross plots showing the relationships among organic ($\delta^{13}C_{org}$) and bulk carbonate ($\delta^{13}C_{carb}$) carbon isotopes, total organic carbon (TOC), and the isotopic difference between paired isotopic measurements of bulk carbonate and organic matter ($\Delta^{13}C = \delta^{13}C_{carb} - \delta^{13}C_{org}$).



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Fig. 8. Mo-EF versus U-EF for the Lower Aptian Lower Dariyan unit. Carbon isotope segments *sensu* Menegatti et al. (1988) after Jafarian et al. (2023). Sloping lines indicate Mo/U ratios equal
to the seawater value (1x SW) and to the multiples (3x SW) and fractions thereof (0.1x SW, 0.3x
SW).

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1210 Supplementary Material

1211**Table S1.** δ^{13} C and TOC data, Dareh Defid section, Lower Aptian Dariyan Formation, Fars1212Province, Iran.

Table S2. XRF and ICP-MS analysis data, Lower Aptian Dariyan Formation, Dareh Sefid
section, Fars Province, Iran.

1215 **Table S3.** Summary data of productive-sensitive (PSTE) and redox-sensitive trace elements

1216 (RSTE) normalized to Al and enrichment factors for U and Mo. Dareh Defid section, Lower

1217 Aptian Dariyan Formation, Fars Province, Iran.

Sample no.	Stratigraphic position (m)	C segment	δ ¹³ C _{carb} (‰VPDB)	$\frac{\delta^{13}C_{org}}{(\text{\%VPDB})}$	$\begin{array}{c} \Delta^{13}\mathrm{C} \\ (\text{\%VPDB}) \end{array}$	TOC (%)
DD 1	0.1	C2	0.93	-26.17	27.10	0.24
DD 2	0.9	C2	1.07	-25.12	26.19	0.35
DD 3	1.1	C2	1.10	-25.70	26.80	0.25
DD 4	1.8	C2	1.12	-25.84	26.96	0.15
DD 5	2.1	C2	0.91	-25.57	26.48	0.38
DD 6	2.8	C2	1.05	-25.50	26.55	0.56
DD 7	3.1	C2	1.01	-24.56	25.57	0.18
DD 8	3.8	C2	1.00	-26.42	27.42	0.20
DD 9	4.1	C2	0.95	-26.69	27.65	0.40
DD 10	4.8	C2	0.99	-26.56	27.55	0.43
DD 11	5.1	C2	0.65	-26.88	27.54	0.24
DD 12	5.8	C2	0.70	-26.50	27.20	0.22
DD 13	6.1	C2	0.25	-27.16	27.41	0.93
DD 14	6.8	C2	0.63	-27.24	27.87	0.41
DD 15	7.8	C2	0.69	-26.89	27.58	0.34
DD 16	8.1	C2	1.42	-26.44	27.86	0.42
DD 17	8.5	C3	0.80	-26.61	27.41	0.36
DD 18	9.1	C3	0.38	-26.43	26.82	0.65
DD 19	9.8	C3	0.38	-26.04	26.42	0.40
DD 20	10.1	C3	0.71	-25.74	26.45	0.71
DD 21	10.8	C3	-0.81	-25.81	25.00	0.35
DD 22	11.1	C4	1.35	-26.28	27.64	0.77
DD 23	11.8	C4	-0.17	-26.49	26.32	0.50
DD 24	12.1	C4	0.99	-26.39	27.38	0.34
DD 25	12.8	C4	1.29	-26.29	27.59	1.46
DD 26	13.1	C4	1.18	-27.22	28.40	2.18
DD 27	13.8	C4	1.28	-26.49	27.77	1.05
DD 28	14.1	C4	1.34	-26.44	27.79	0.62
DD 29	14.8	C4	0.36	-26.26	26.62	1.06
DD 30	15.1	C4	1.28	-26.19	27.47	1.22
DD 31	15.8	C4	1.36	-26.25	27.60	0.67
DD 32	16.1	C4	1.28	-26.87	28.15	0.86
DD 33	16.5	C4	1.42	-26.93	28.36	0.77
DD 34	16.8	C4	1.41	-26.80	28.20	2.96
DD 35	17.1	C4	1.06	-26.38	27.44	2.58
DD 36	17.8	C4	0.99	-27.16	28.15	0.55
DD 37	18.1	C4	1.16	-22.62	23.77	0.52
DD 38	18.8	C4	1.37	-25.39	26.77	0.48
DD 39	19.1	C4	1.30	-24.60	25.90	0.39
DD 40	19.7	C4	2.02	-23.78	25.80	0.38

Table S1. δ^{13} C and TOC data, Dareh Defid section, Lower Aptian Dariyan Formation, Fars Province, Iran.

DD 41	20	C4	2.08	-25.70	27.78	0.28
DD 42	20.1	C4	2.16	-25.79	27.95	0.49
DD 43	20.8	C4	2.22	-25.33	27.55	1.36
DD 44	21.1	C4	2.32	-24.71	27.03	0.41
DD 45	21.8	C4	1.83	-25.20	27.04	0.21
DD 46	22.1	C4	2.22	-25.60	27.82	0.32
DD 47	22.8	C4	2.48	-24.73	27.21	1.68
DD 48	23.1	C4	2.83	-24.03	26.86	2.38
DD 49	23.8	C5+C6	2.59	-23.75	26.34	0.37
DD 50	24.1	C5+C6	2.55	-25.14	27.69	0.11
DD 51	24.8	C5+C6	2.66	-25.14	27.79	0.31
DD 52	25.1	C5+C6	2.45	-24.99	27.44	0.82
DD 53	25.8	C5+C6	1.67	-24.65	26.32	2.35
DD 54	26.1	C5+C6	2.09	-24.93	27.01	1.57
DD 55	26.8	C5+C6	2.06	-25.13	27.18	0.72
DD 56	27.1	C5+C6	2.57	-24.31	26.88	2.46
DD 57	27.8	C5+C6	2.24	-24.86	27.11	0.81
DD 58	28.1	C5+C6	2.67	-24.79	27.46	0.58
DD 59	28.8	C5+C6	2.34	-24.78	27.11	0.58
DD 60	29.1	C5+C6	2.41	-24.82	27.23	1.17
DD 61	29.8	C5+C6	2.47	-24.75	27.22	0.66
DD 62	30.1	C5+C6	2.89	-24.83	27.72	0.21
DD 63	30.8	C5+C6	2.76	-24.67	27.43	0.30
DD 64	31.1	C5+C6	2.02	-24.37	26.40	1.04
DD 65	31.8	C5+C6	2.81	-24.00	26.82	1.53
DD 66	32.1	C5+C6	2.53	-23.88	26.40	1.42
DD 67	32.8	C5+C6	1.90	-24.11	26.01	0.60
DD 68	33.1	C5+C6	2.10	-24.02	26.12	1.04
DD 69	33.8	C5+C6	1.54	-24.13	25.67	0.80
DD 70	34.1	C5+C6	2.31	-23.85	26.15	1.87
DD 71	34.8	C5+C6	2.45	-24.00	26.45	0.22
DD 72	35.1	C5+C6	2.80	-24.48	27.28	0.28
DD 73	35.8	C5+C6	2.64	-24.70	27.34	0.37
DD 74	36.1	C5+C6	2.55	-24.84	27.39	0.89
DD 75	36.8	C5+C6	2.36	-24.82	27.18	0.52
DD 76	37.1	C5+C6	2.67	-24.38	27.05	1.37
DD 77	37.8	C5+C6	2.22	-24.25	26.47	0.52
DD 78	38.1	C5+C6	2.28	-24.77	27.05	0.50
DD 79	38.8	C5+C6	2.54	-24.64	27.18	1.32
DD 80	39.5	C5+C6	0.63	-24.83	25.46	0.89

Sample	Stratigraphic	С	XRF (%)							ICP-MS (ppm)				
no.	position (m)	segment	Al ₂ O ₃	CaO	K ₂ O	P ₂ O ₅	SiO ₂	TiO ₂	Cu	Ni	Mo	U	V	
DD 06	2.8	C2	0.57	53.20	0.07	0.100	2.58	0.01	3.5	11.9	4.84	2.53	23	
DD 10	4.8	C2	1.56	51.00	0.14	0.090	4.55	0.03	6.9	32.9	6.55	6.48	59	
DD 13	6.1	C2	0.79	52.40	0.09	0.050	2.77	0.02	4.6	21.3	1.18	2.78	27	
DD 14	6.8	C2	1.99	49.90	0.23	0.100	5.69	0.07	5.6	34.1	10.65	2.63	55	
DD 15	7.8	C2	1.21	51.60	0.14	0.070	3.67	0.04	6.4	107.5	27.3	2.68	98	
DD 17	8.5	C3	2.30	49.10	0.27	0.140	6.49	0.08	10.8	77.0	20.8	4.47	119	
DD 20	10.1	C3	1.23	51.70	0.12	0.130	4.06	0.03	23.0	28.9	11.05	5.34	107	
DD 21	10.8	C3	1.38	51.40	0.18	0.050	4.23	0.05	3.4	43.8	11.30	4.23	57	
DD 25	12.8	C4	0.45	53.00	0.07	0.110	2.32	0.01	12.4	32.2	2.67	3.04	60	
DD 26	13.1	C4	0.17	52.60	0.04	0.060	2.40	0.01	5.6	26.1	0.50	2.37	27	
DD 28	14.1	C4	0.16	54.20	0.04	0.020	1.39	0.03	12.1	38.0	5.76	6.59	69	
DD 32	16.1	C4	0.50	51.90	0.10	0.030	4.22	0.02	11.4	50.6	9.40	5.77	97	
DD 34	16.8	C4	0.25	52.70	0.06	0.030	1.84	0.01	10.6	41.9	0.45	7.23	93	
DD 35	17.1	C4	0.06	53.20	0.02	0.010	1.89	0.02	3.0	21.5	1.35	2.66	51	
DD 36	17.8	C4	0.30	53.60	0.07	0.050	2.59	0.01	13.5	21.6	7.14	5.33	104	
DD 42	20.1	C4	0.18	50.00	0.04	0.020	7.44	0.01	9.7	51.5	4.28	3.47	123	
DD 43	20.8	C4	0.11	52.40	0.03	0.040	3.50	0.02	6.1	29.1	2.90	2.68	95	
DD 47	22.8	C4	0.11	54.30	0.03	0.030	0.74	0.01	16.1	36.8	4.33	2.92	232	
DD 48	23.1	C4	0.14	54.00	0.04	0.030	0.89	0.01	25.3	23.9	2.08	2.73	306	
DD 49	23.8	C5+C6	0.10	52.80	0.03	0.020	4.17	0.01	6.1	52.9	2.32	1.94	164	
DD 52	25.1	C5+C6	0.82	43.50	0.09	0.070	18.05	0.01	25.0	45.3	2.15	3.15	380	
DD 53	25.8	C5+C6	0.63	43.30	0.08	0.030	17.07	0.01	16.0	39.5	7.57	2.72	420	
DD 54	26.1	C5+C6	0.67	41.70	0.08	0.030	20.88	0.02	17.2	46.9	3.04	4.42	447	
DD 55	26.8	C5+C6	0.52	46.10	0.07	0.040	14.08	0.01	10.2	31.3	3.56	3.18	294	
DD 56	27.1	C5+C6	0.97	44.00	0.10	0.150	15.60	0.02	26.2	67.3	1.84	5.24	395	
DD 57	27.8	C5+C6	0.64	45.90	0.07	0.040	15.00	0.01	14.0	39.8	15.95	2.85	239	
DD 58	28.1	C5+C6	0.77	44.20	0.08	0.060	17.34	0.01	13.8	23.1	2.03	2.49	219	
DD 60	29.1	C5+C6	0.58	43.80	0.07	0.130	17.43	0.01	13.2	36.0	5.25	3.54	172	

Table S2. XRF and ICP-MS analysis data, Lower Aptian Dariyan Formation, Dareh Sefid section, Fars Province, Iran.

DD 6431.1C5+C60.4045.500.040.06014.680.018.233.12.793.09169DD 6531.8C5+C60.8442.600.080.05019.240.0213.535.120.92.72216DD 6632.1C5+C60.8045.300.080.06014.940.0212.146.47.713.96349DD 6732.8C5+C62.3136.100.190.14029.070.0724.766.630.34.39295DD 6833.1C5+C61.2042.900.100.09018.220.0412.440.04.873.80266DD 6933.8C5+C60.5549.600.050.0708.040.019.129.35.192.61232DD 7034.1C5+C61.3545.200.110.28013.510.0426.360.95.334.69397DD 7436.1C5+C61.5747.400.140.07010.440.0412.231.67.332.23175DD 7637.1C5+C63.5241.000.210.09017.310.1228.945.913.854.33402DD 7838.1C5+C63.2541.600.270.11016.960.1229.843.211.353.06417	DD 61	29.8	C5+C6	0.59	44.80	0.07	0.050	16.68	0.01	6.7	25.8	6.18	2.05	124
DD 6531.8C5+C60.8442.600.080.05019.240.0213.535.120.92.72216DD 6632.1C5+C60.8045.300.080.06014.940.0212.146.47.713.96349DD 6732.8C5+C62.3136.100.190.14029.070.0724.766.630.34.39295DD 6833.1C5+C61.2042.900.100.09018.220.0412.440.04.873.80266DD 6933.8C5+C60.5549.600.050.0708.040.019.129.35.192.61232DD 7034.1C5+C61.3545.200.110.28013.510.0426.360.95.334.69397DD 7436.1C5+C61.5747.400.140.07010.440.0412.231.67.332.23175DD 7637.1C5+C63.5241.000.210.09017.310.1228.945.913.854.33402DD 7838.1C5+C63.2541.600.270.11016.960.1229.843.211.353.06417	DD 64	31.1	C5+C6	0.40	45.50	0.04	0.060	14.68	0.01	8.2	33.1	2.79	3.09	169
DD 6632.1C5+C60.8045.300.080.06014.940.0212.146.47.713.96349DD 6732.8C5+C62.3136.100.190.14029.070.0724.766.630.34.39295DD 6833.1C5+C61.2042.900.100.09018.220.0412.440.04.873.80266DD 6933.8C5+C60.5549.600.050.0708.040.019.129.35.192.61232DD 7034.1C5+C61.3545.200.110.28013.510.0426.360.95.334.69397DD 7436.1C5+C61.5747.400.140.07010.440.0412.231.67.332.23175DD 7637.1C5+C63.5241.000.210.09017.310.1228.945.913.854.33402DD 7838.1C5+C63.2541.600.270.11016.960.1229.843.211.353.06417	DD 65	31.8	C5+C6	0.84	42.60	0.08	0.050	19.24	0.02	13.5	35.1	20.9	2.72	216
DD 6732.8C5+C62.3136.100.190.14029.070.0724.766.630.34.39295DD 6833.1C5+C61.2042.900.100.09018.220.0412.440.04.873.80266DD 6933.8C5+C60.5549.600.050.0708.040.019.129.35.192.61232DD 7034.1C5+C61.3545.200.110.28013.510.0426.360.95.334.69397DD 7436.1C5+C61.5747.400.140.07010.440.0412.231.67.332.23175DD 7637.1C5+C63.5241.000.210.09017.310.1228.945.913.854.33402DD 7838.1C5+C63.2541.600.270.11016.960.1229.843.211.353.06417	DD 66	32.1	C5+C6	0.80	45.30	0.08	0.060	14.94	0.02	12.1	46.4	7.71	3.96	349
DD 6833.1C5+C61.2042.900.100.09018.220.0412.440.04.873.80266DD 6933.8C5+C60.5549.600.050.0708.040.019.129.35.192.61232DD 7034.1C5+C61.3545.200.110.28013.510.0426.360.95.334.69397DD 7436.1C5+C61.5747.400.140.07010.440.0412.231.67.332.23175DD 7637.1C5+C63.5241.000.210.09017.310.1228.945.913.854.33402DD 7838.1C5+C64.0039.900.300.14019.080.1330.954.16.753.52429DD 7938.8C5+C63.2541.600.270.11016.960.1229.843.211.353.06417	DD 67	32.8	C5+C6	2.31	36.10	0.19	0.140	29.07	0.07	24.7	66.6	30.3	4.39	295
DD 6933.8C5+C60.5549.600.050.0708.040.019.129.35.192.61232DD 7034.1C5+C61.3545.200.110.28013.510.0426.360.95.334.69397DD 7436.1C5+C61.5747.400.140.07010.440.0412.231.67.332.23175DD 7637.1C5+C63.5241.000.210.09017.310.1228.945.913.854.33402DD 7838.1C5+C64.0039.900.300.14019.080.1330.954.16.753.52429DD 7938.8C5+C63.2541.600.270.11016.960.1229.843.211.353.06417	DD 68	33.1	C5+C6	1.20	42.90	0.10	0.090	18.22	0.04	12.4	40.0	4.87	3.80	266
DD 7034.1C5+C61.3545.200.110.28013.510.0426.360.95.334.69397DD 7436.1C5+C61.5747.400.140.07010.440.0412.231.67.332.23175DD 7637.1C5+C63.5241.000.210.09017.310.1228.945.913.854.33402DD 7838.1C5+C64.0039.900.300.14019.080.1330.954.16.753.52429DD 7938.8C5+C63.2541.600.270.11016.960.1229.843.211.353.06417	DD 69	33.8	C5+C6	0.55	49.60	0.05	0.070	8.04	0.01	9.1	29.3	5.19	2.61	232
DD 7436.1C5+C61.5747.400.140.07010.440.0412.231.67.332.23175DD 7637.1C5+C63.5241.000.210.09017.310.1228.945.913.854.33402DD 7838.1C5+C64.0039.900.300.14019.080.1330.954.16.753.52429DD 7938.8C5+C63.2541.600.270.11016.960.1229.843.211.353.06417	DD 70	34.1	C5+C6	1.35	45.20	0.11	0.280	13.51	0.04	26.3	60.9	5.33	4.69	397
DD 76 37.1 C5+C6 3.52 41.00 0.21 0.090 17.31 0.12 28.9 45.9 13.85 4.33 402 DD 78 38.1 C5+C6 4.00 39.90 0.30 0.140 19.08 0.13 30.9 54.1 6.75 3.52 429 DD 79 38.8 C5+C6 3.25 41.60 0.27 0.110 16.96 0.12 29.8 43.2 11.35 3.06 417	DD 74	36.1	C5+C6	1.57	47.40	0.14	0.070	10.44	0.04	12.2	31.6	7.33	2.23	175
DD 78 38.1 C5+C6 4.00 39.90 0.30 0.140 19.08 0.13 30.9 54.1 6.75 3.52 429 DD 79 38.8 C5+C6 3.25 41.60 0.27 0.110 16.96 0.12 29.8 43.2 11.35 3.06 417	DD 76	37.1	C5+C6	3.52	41.00	0.21	0.090	17.31	0.12	28.9	45.9	13.85	4.33	402
DD 79 38.8 C5+C6 3.25 41.60 0.27 0.110 16.96 0.12 29.8 43.2 11.35 3.06 417	DD 78	38.1	C5+C6	4.00	39.90	0.30	0.140	19.08	0.13	30.9	54.1	6.75	3.52	429
	DD 79	38.8	C5+C6	3.25	41.60	0.27	0.110	16.96	0.12	29.8	43.2	11.35	3.06	417

Sample no.	Stratigraphic position (m)	C segment	K/Al	Si/Al	Ti/Al	P/Al	Mo/Al	U/Al	V/Al	Cu/Al	Ni/Al	EF U	EF Mo
DD 06	2.8	C2	0.1926	3.9978	0.0199	0.2567	16.0424	8.3858	76.2344	11.6009	39.4430	20.0337	109.0796
DD 10	4.8	C2	0.1408	2.5761	0.0218	0.0844	7.9326	7.8478	71.4538	8.3565	39.8446	18.7484	53.9374
DD 13	6.1	C2	0.1787	3.0969	0.0287	0.0926	2.8220	6.6484	64.5706	11.0009	50.9390	15.8830	19.1879
DD 14	6.8	C2	0.1813	2.5255	0.0398	0.0735	10.1110	2.4969	52.2165	5.3166	32.3742	5.9651	68.7495
DD 15	7.8	C2	0.1815	2.6789	0.0374	0.0847	42.6261	4.1845	153.0167	9.9929	167.8499	9.9969	289.8347
DD 17	8.5	C3	0.1841	2.4923	0.0394	0.0891	17.0857	3.6718	97.7501	8.8714	63.2501	8.7719	116.1739
DD 20	10.1	C3	0.1530	2.9154	0.0276	0.1547	16.9729	8.2023	164.3527	35.3281	44.3906	19.5953	115.4065
DD 21	10.8	C3	0.2046	2.7073	0.0410	0.0530	15.4703	5.7911	78.0358	4.6548	59.9643	13.8349	105.1895
DD 25	12.8	C4	0.2440	4.5536	0.0252	0.3577	11.2098	12.7632	251.9050	52.0604	135.1890	30.4913	76.2205
DD 26	13.1	C4	0.3691	12.4693	0.0666	0.5164	5.5567	26.3389	300.0633	62.2354	290.0612	62.9237	37.7828
DD 28	14.1	C4	0.3921	7.6732	0.2124	0.1829	68.0144	77.8150	814.7553	142.8774	448.7058	185.9003	462.4615
DD 32	16.1	C4	0.3137	7.4546	0.0453	0.0878	35.5186	21.8024	366.5218	43.0758	191.1959	52.0859	241.5077
DD 34	16.8	C4	0.3764	6.5007	0.0453	0.1756	3.4007	54.6382	702.8150	80.1058	316.6446	130.5308	23.1231
DD 35	17.1	C4	0.5228	27.8221	0.3775	0.2439	42.5090	83.7584	1605.8946	94.4644	676.9948	200.0991	289.0385
DD 36	17.8	C4	0.3660	7.6253	0.0378	0.2439	44.9650	33.5663	654.9531	85.0179	136.0287	80.1901	305.7385
DD 42	20.1	C4	0.3486	36.5073	0.0629	0.1626	44.9231	36.4213	1291.0133	101.8116	540.5462	87.0105	305.4530
DD 43	20.8	C4	0.4278	28.1032	0.2059	0.5321	49.8085	46.0299	1631.6576	104.7696	499.8025	109.9656	338.6713
DD 47	22.8	C4	0.4278	5.9418	0.1030	0.3991	74.3692	50.1520	3984.6796	276.5230	632.0526	119.8133	505.6713
DD 48	23.1	C4	0.4481	5.6149	0.0809	0.3136	28.0694	36.8411	4129.4432	341.4213	322.5284	88.0135	190.8571
DD 49	23.8	C5+C6	0.4705	36.8312	0.1133	0.2927	43.8315	36.6522	3098.4319	115.2466	999.4332	87.5622	298.0308
DD 52	25.1	C5+C6	0.1722	19.4421	0.0138	0.1249	4.9536	7.2576	875.5236	57.6002	104.3716	17.3385	33.6820
DD 53	25.8	C5+C6	0.1992	23.9317	0.0180	0.0697	22.7014	8.1569	1259.5252	47.9819	118.4553	19.4869	154.3578
DD 54	26.1	C5+C6	0.1873	27.5255	0.0338	0.0655	8.5723	12.4637	1260.4651	48.5011	132.2501	29.7757	58.2870
DD 55	26.8	C5+C6	0.2111	23.9155	0.0218	0.1126	12.9344	11.5537	1068.1742	37.0591	113.7206	27.6019	87.9467
DD 56	27.1	C5+C6	0.1617	14.2047	0.0234	0.2263	3.5838	10.2060	769.3491	51.0302	131.0815	24.3823	24.3680
DD 57	27.8	C5+C6	0.1716	20.7010	0.0177	0.0915	47.0846	8.4132	705.5309	41.3282	117.4901	20.0992	320.1502
DD 58	28.1	C5+C6	0.1630	19.8902	0.0147	0.1140	4.9808	6.1095	537.3429	33.8600	56.6786	14.5956	33.8671

Table S3. Summary data of productive-sensitive (PSTE) and redox-sensitive trace elements (RSTE) normalized to Al and enrichment factors for U and Mo. Dareh Defid section, Lower Aptian Dariyan Formation, Fars Province, Iran.

DD 60	29.1	C5+C6	0.1893	26.5429	0.0195	0.3280	17.1013	11.5312	560.2715	42.9976	117.2661	27.5480	116.2798
DD 61	29.8	C5+C6	0.1861	24.9703	0.0192	0.1240	19.7895	6.5645	397.0764	21.4546	82.6163	15.6825	134.5580
DD 64	31.1	C5+C6	0.1568	32.4150	0.0283	0.2195	13.1778	14.5947	798.2241	38.7304	156.3386	34.8669	89.6019
DD 65	31.8	C5+C6	0.1494	20.2304	0.0270	0.0871	47.0073	6.1177	485.8168	30.3636	78.9452	14.6152	319.6245
DD 66	32.1	C5+C6	0.1568	16.4945	0.0283	0.1097	18.2080	9.3520	824.2018	28.5755	109.5787	22.3419	123.8048
DD 67	32.8	C5+C6	0.1290	11.1151	0.0343	0.0887	24.7816	3.5905	241.2727	20.2015	54.4704	8.5776	168.5015
DD 68	33.1	C5+C6	0.1307	13.4106	0.0378	0.1097	7.6674	5.9827	418.7921	19.5226	62.9763	14.2928	52.1340
DD 69	33.8	C5+C6	0.1426	12.9114	0.0206	0.1862	17.8280	8.9655	796.9359	31.2591	100.6475	21.4187	121.2210
DD 70	34.1	C5+C6	0.1278	8.8390	0.0336	0.3035	7.4592	6.5635	555.5905	36.8061	85.2279	15.6803	50.7185
DD 74	36.1	C5+C6	0.1399	5.8733	0.0289	0.0652	8.8207	2.6835	210.5894	14.6811	38.0264	6.4109	59.9760
DD 76	37.1	C5+C6	0.0936	4.3434	0.0386	0.0374	7.4337	2.3240	215.7652	15.5115	24.6359	5.5521	50.5452
DD 78	38.1	C5+C6	0.1176	4.2131	0.0368	0.0512	3.1882	1.6626	202.6261	14.5947	25.5526	3.9719	21.6779
DD 79	38.8	C5+C6	0.1303	4.6092	0.0418	0.0495	6.5980	1.7788	242.4101	17.3233	25.1130	4.2496	44.8627



Citation on deposit: Jafarian, A., Husinec, A., Wang, C., Chen, X., Wang, M., Gröcke, D. R., ...Li, Y. (2024). Intrashelf basin record of redox and productivity changes along the Arabian margin of Neo-Tethys during Oceanic Anoxic Event 1a. Palaeogeography, Palaeoclimatology, Palaeoecology, 636, Article

111975. https://doi.org/10.1016/j.palaeo.2023.111975

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