

Graphene-like Dispersion and Strong Optical Absorption in Two-Dimensional RP-type $\text{Sr}_3\text{Ti}_2\text{S}_7$ Perovskite

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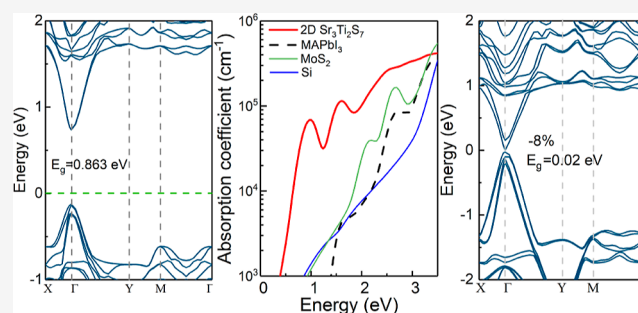
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ABSTRACT: Two-dimensional (2D) Ruddlesden–Popper (RP) perovskite alloys have recently become attractive due to many desired physical properties originating from distinct van der Waals-type layered structures. In this work, a novel 2D RP-type $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite material design is proposed by using first-principles calculations. Our results reveal that the 2D $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite possesses dynamically stable structures, direct band structures with a band gap value of 0.86 eV, and a smaller effective mass (0.15/0.25 m_0 for electron/hole) than MAPbI_3 and phosphorene. More importantly, 2D $\text{Sr}_3\text{Ti}_2\text{S}_7$ possesses wide optical spectra (from infrared-to ultraviolet-light region) and a higher absorption coefficient (10^5 cm^{-1}) than MAPbI_3 , silicon, and MoS_2 in the visible-light region. Interestingly, we also find that the ideal Dirac-like linear dispersion can appear near the Fermi level in the electronic band structures when compressive strain is applied. Especially, the Dirac-cone-like band structures can be realized when compressive strain is enhanced to -6% , indicating ultrahigh carrier mobility. These properties make the 2D $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite a promising candidate for future applications in solar cells and optoelectronic devices.



1. INTRODUCTION

Lead-based perovskite nanocrystals are considered as one of the most sustainable materials for the development of new-generation solar cells. Solar cells prepared incorporating perovskite material possess ideal optoelectronic properties that helped achieving an impressive power conversion efficiency of 25.5%.^{1–13} This performance of perovskite solar cells is higher than commercialized thin-film solar cells, such as silicon, GaAs, and $\text{Cu}(\text{In,Ga})\text{Se}_2$.^{14–16} However, despite their great potential, a persistent threat on future use and commercialization is the issue of toxic lead-content^{17–19} and longevity.^{20–24}

Thus, searching for an alternative new material with suitable direct band gap, long-term stability, nontoxicity, and high optical absorption has become a critical challenge in recent years. There is great concern among the scientific community for the toxicity of the lead content in these perovskite solar cells. In this direction, there are many attempts to find alternative Sn-based perovskite solar cells which show respectable performance over 13% but have detrimental long-term stability.^{25–28}

Recently, a new class of CaTiS_3 , BaZrS_3 , and CaHfSe_3 among 18 chalcogenide compounds have been reported to be ideal absorber materials because of their suitable band gaps (~ 0.9 – 1.7 eV).^{29–33} Leveille et al. predicted that SrSnS_3 and SrSnSe_3 can be good solar cell materials with a suitable direct

band gap (~ 0.9 – 1.6 eV), large optical absorption coefficient ($\sim 10^5 \text{ cm}^{-1}$), and small carrier effective mass (~ 0.12 – $0.65 m_0$).²⁶ Nie et al. have synthesized SrZrS_3 and BaZrS_3 and found they possess suitable direct band gap ~ 1.5 – 2.1 eV solar cell applications.^{34,35} Recently, two-dimensional (2D) Ruddlesden–Popper (RP) perovskite, as a representative of 2D-layered perovskites has captured a growing interest for it have also been widely exploited in other optoelectronic fields, such as solar cells, light-emitting diodes, lasers, and photodetectors.^{35–39} The recently synthesized $\text{Ba}_3\text{Zr}_2\text{S}_7$ RP perovskite possesses an optimal band gap (1.28–1.35 eV), a large absorption coefficient ($>10^4 \text{ cm}^{-1}$) near the band edge, and a long carrier lifetime (>60 ns). However, its large carrier effective mass directly led to its low carrier mobility. $\text{Cs}_2\text{PbI}_2\text{Cl}_2$ shows direct band structures but it does not show ideal optical absorption in the visible-light region due to the large band gap ($E_g \sim 3.04$ eV) and the existence of lead atoms makes it unfriendly to the environment.³⁵ Perovskite material can be divided into different structures as follows: (1)

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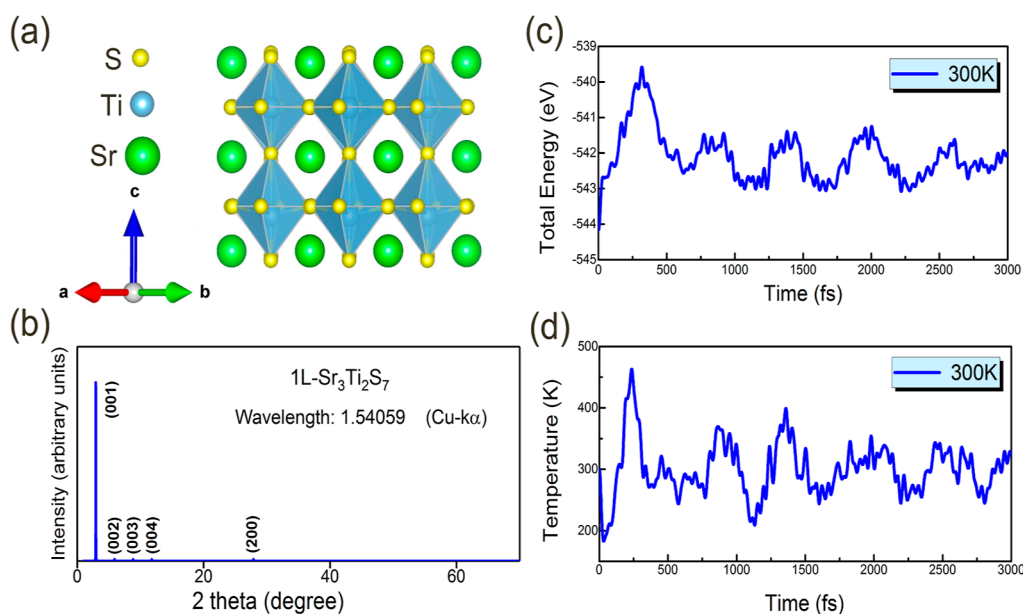


Figure 1. Lattice structure, X-ray diffraction (XRD), and structural stability of two-dimensional $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite. (a) Perspective view of the natural super lattice nanostructure in $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite. The green and yellow spheres represent the Sr and S atoms, respectively. The TiS_6 octahedra are in blue. (b) The XRD patterns for two-dimensional $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite with the wavelength of 1.54059 Å (Cu $K\alpha$) calculated by using the VESTA package (3D visualization program for structural models, volumetric data such as electron/nuclear densities, and crystal morphologies) and bulk $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite are shown in Figure S2 of the Supporting Information. (c,d) The ab initio molecular dynamics on total energy and temperature at 300 K.

ABX_3 3D structure with three different atomic sites; (2) $A'[A_{n-1}B_nX_{3n+1}]$ Dion–Jacobson (DJ) type structure; (3) $A'_2A_{n-1}B_nX_{3n+3}$ two-dimensional-111 type structure; (4) $A_2A_nB_nX_{3n+2}$ two-dimensional-110 type structure; and (5) $A'_2[A_{n-1}B_nX_{3n+1}]$ RP-type structure.

RP phases are a two-dimensional homologous series of the perovskite structure. Such layered structures can host interesting octahedral rotations and distortions that can lead to a noncentrosymmetric structure, which is a prerequisite for both polar nature and ferroelectric properties.^{39–43} 2D perovskite chalcogenides are formed by alternating a set number ($n = 2$) of perovskite layers with the chemical formula ABX_3 and a rock salt layer AX. Such a 2D perovskite has a general formula of $A_{n+1}B_nX_{3n+1}$ for the case of the same cations in perovskite and rock salt layer. In this direction, motivated by the recent findings in this study, we designed $\text{Sr}_3\text{Ti}_2\text{S}_7$ as a new type of RP perovskite and performed first-principles calculations to study its optoelectronic properties.

The results show that 2D $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite is dynamically stable with direct band structures with a band gap value of 0.86 eV and smaller effective mass (0.15/0.25 m_0 for electron/hole). Moreover, 2D $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite possesses wide optical spectra and higher absorption coefficient (10^5 cm^{-1}) than MAPbI_3 , silicon, and MoS_2 in the visible-light region. Interestingly, an ideal Dirac-like linear dispersion appears in the electronic band structures with compressive strain. Especially the Dirac-cone-like band structures can be realized when compressive strain is -6% . Our finding indicates that the suitable optoelectronic properties of 2D $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite make it ideal candidate for applications in solar cells and optoelectronic devices.

2. RESULTS AND DISCUSSION

2.1. Lattice Structures and Stability of Two-Dimensional $\text{Sr}_3\text{Ti}_2\text{S}_7$ Chalcogenide Perovskite.

RP-type perovskites have structural formula $A'_2A_{n-1}B_nX_{3n+1}$. 2D $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite possesses the $P4/nmm$ space group. It is an $n = 2$ RP-type perovskite with corner sharing TiS_6 octahedron surrounding a large A-site cation, as shown in Figure 1a. The optimized lattice parameters of 2D $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite are $a = b = 6.39 \text{ \AA}$. In order to clearly understand the electronic and crystal structure properties in the 2D $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite, we further calculated X-ray diffraction (XRD) of 2D $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite by using the VESTA package^{44,45} in the $0\text{--}70^\circ$ range of angles with expected Bragg reflections, which are shown Figure 1b. The XRD of the 2D $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite shows only one set of sharp 00l reflections. The most intense peak has a full-width-at-half-maximum (fwhm) of less than 0.01° . This confirmed that the crystal facets with layered-like features are along the (001) plane. Additional ab initio molecular dynamics simulations were conducted at 300 K for the monolayer supercell to study its room-temperature stability. As indicated by the AIMD results, illustrated in Figure 1c,d, the total energy shows a regular fluctuation from 0 to 3000 fs, which clarifies that the 2D $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite is stable at room temperature. During the AIMD (ab initio molecular dynamics on total energy and temperature at 300 K) simulations, all atoms in the cell vibrate around their equilibrium positions and no phase transition occurs at 300 K temperature. These results indicate the higher stability of the 2D $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite material. To further testify to the structural stability of 2D $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite, we calculated its formation energy. The formation energy is defined by following formula

$$E_f = \frac{(E_{\text{Sr}_3\text{Ti}_2\text{S}_7} - N_{\text{Sr}}E_{\text{Sr}} - N_{\text{Ti}}E_{\text{Ti}} - N_{\text{S}}E_{\text{S}})}{(N_{\text{Sr}} + N_{\text{Ti}} + N_{\text{S}})} \quad (1)$$

where $E_{\text{Sr}_3\text{Ti}_2\text{S}_7}$ is the total energy for the unit cell and $E_{\text{Sr}(\text{Ti},\text{S})}(N_{\text{Sr}(\text{Ti},\text{S})})$ is the single atom energy (number) of Sr, Ti,

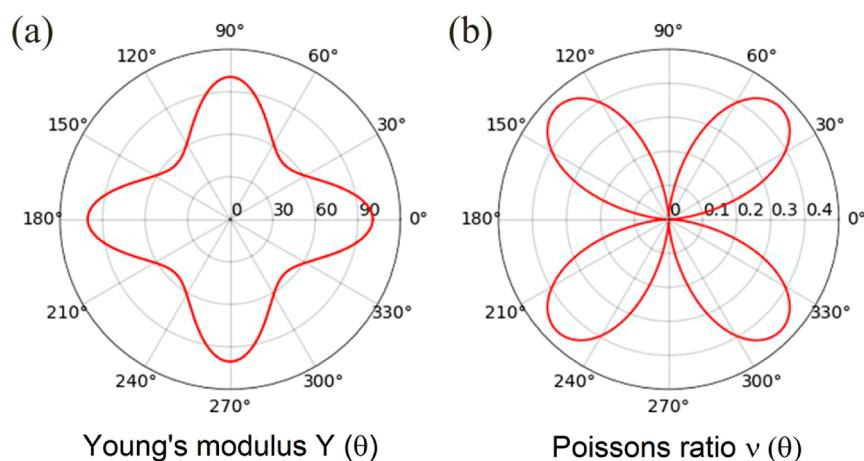


Figure 2. Mechanical properties of the two-dimensional $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite, the orientational-dependence of (a) Young's modulus $Y(\theta)$ and (b) Poisson's ratio $\nu(\theta)$ of the two-dimensional $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite. It is shown that the two-dimensional $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite exhibits anisotropic Young's modulus $Y(\theta)$ and Poisson's ratio $\nu(\theta)$ with a clear angle-dependent feature.

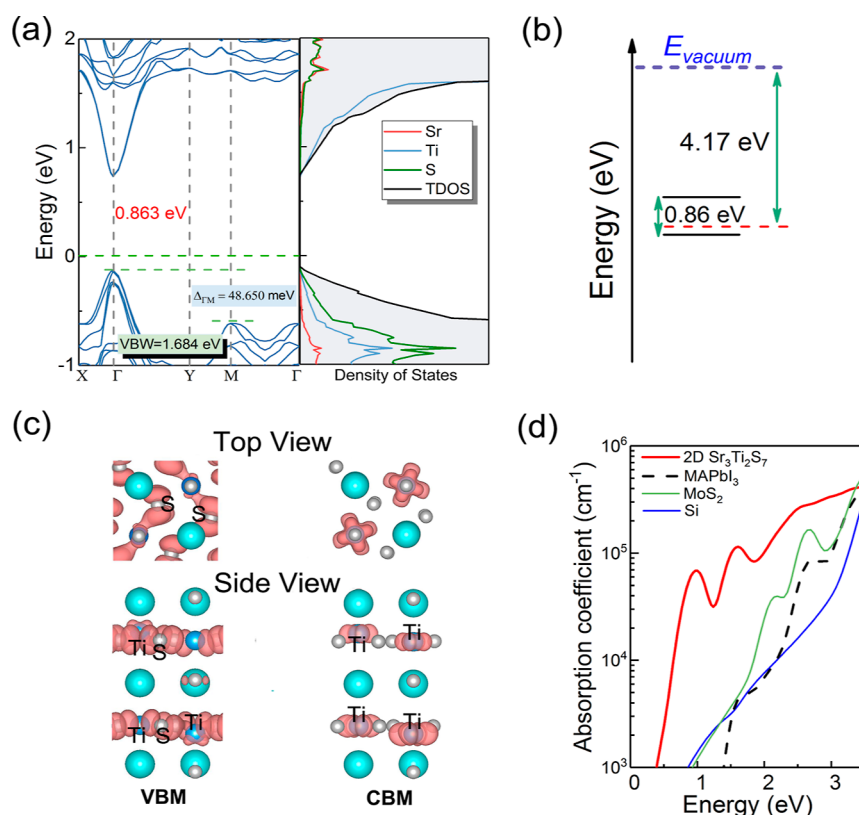


Figure 3. Electronic structure and optical absorption in two-dimensional $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite. (a) The HSE06 band structure and density of states, where the direct band gap and the top valence bandwidth (VBW) are 0.863 and 1.684 eV, respectively. (b) Schematic energy level diagrams of two-dimensional $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite. The top and bottom of the black solid lines represent the energy levels of two-dimensional $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite VBM and CBM, respectively. (c) The partial charge density for VBM and CBM of two-dimensional $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite with different view directions. (d) Optical absorption of two-dimensional $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite. For comparison, the optical absorption in important photovoltaic materials Si, MAPbI₃, and MoS₂.

and S atoms. As shown in Figure S3 (Supporting Information), it is clearly shows that the formation energy of 2D $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite is negative, indicating the higher structural stability. Moreover, its formation energy can be compared to MAPbI₃, phosphorene, graphene, MoS₂, and BN.^{40–45}

Moreover, It is also important to investigate the mechanical properties of perovskites material. Our calculated elastic constants are $C_{11}-C_{22} = 100.49$, $C_{12} = 0.79$, and $C_{66} =$

18.52 N/m, which satisfies the Born criteria $C_{11}C_{22}-C_{12}^2 > 0$ and $C_{66} > 0$ (for more details, see Table S2 in the Supporting Information).

All the above results show that the 2D $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite is mechanically stable.^{46–48} The calculated elastic constants of graphene, BN, Si, and MoS₂, are in agreement with previous results.^{49–52} Young's modulus Y^{2D} , Poisson's ratio ν^{2D} , shear modulus G^{2D} , and layer modulus γ of 2D $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite is

	Sr ₃ Ti ₂ S ₇	MAPbI ₃ (refs. 52-57)	Graphene (refs. 58-60)	Phosphorene (refs. 61-65)
Crystal structure				
Bandgap (eV)	0.86	1.49	0	1.55
Band structure				
Effective mass (m ₀)	m _e [*] ~ 0.15 m _h [*] ~ 0.25	m _e [*] ~ 0.36 m _h [*] ~ 0.32	Massless	m _e [*] ~ 0.3 m _h [*] ~ 0.3
Optical absorption coefficient α (cm ⁻¹)	~10 ⁵	~10 ⁵	~10 ⁵	~10 ⁵

Figure 4. Comprehensive comparison of crystal structures, band gap, band structure, efficient mass, and optical absorption in two-dimensional Sr₃Ti₂S₇ perovskite, MAPbI₃,^{52–57} graphene,^{58–60} and phosphorene.^{61–65} It can be clearly shown that the two-dimensional Sr₃Ti₂S₇ perovskite is an impressive two-dimensional semiconductor, which possesses remarkable electronic, optical, and transport properties.

also calculated, which is shown in Table S2. The derived Young's modulus (~100 N/m), shear modulus (~0.8 N/m), and layer modulus (~19 N/m) in 2D Sr₃Ti₂S₇ perovskite indicate that its stiffness is much smaller than that of monolayer graphene (340 N/m) and borophene (380 N/m).⁵² As a result, they have great potential as flexible devices. In addition, the orientation-dependent Young's modulus $Y^{2D}(\theta)$ and Poisson's ratio $\nu^{2D}(\theta)$ of 2D Sr₃Ti₂S₇ chalcogenide perovskite are displayed in Figure 2. It is shown that 2D Sr₃Ti₂S₇ perovskite exhibits highly anisotropic Young's modulus $Y^{2D}(\theta)$ and Poisson's ratio $\nu^{2D}(\theta)$ with clear angle-dependent mechanical properties. The minimum (maximum) Young's modulus (Poisson's ratio) is 54.24 (0.46) in the direction of 45, 135, 225, and 315°, and the maximum (minimum) Young's modulus (Poisson's ratio) is 100.48 (0.008) in the direction of 0, 90, 180, and 270°. In fact, the mechanical properties of isotropy or anisotropy can be clarified by analyzing the distribution of charge densities from the first-principle.⁵⁴ Additionally, calculated charge density distribution of 2D Sr₃Ti₂S₇ perovskite is presented in Figure S4. The charge density distribution of the 2D Sr₃Ti₂S₇ perovskite is not uniform, and it shows expanding the pattern along the *y* direction. In *y* directions, the electron orbitals are hybridized more strongly, leading to a larger overlap of wave functions and enhanced bonding power. Enhanced bonding power results into higher stiffness. The charge densities of 2D Sr₃Ti₂S₇ perovskite have the lowest value in the directions 45, 135, 225, and 315°, resulting in the reduced Young's modulus. Therefore, the 2D Sr₃Ti₂S₇ perovskites have strong anisotropic mechanical properties, which makes them play a vital role in the field of special flexible electronic devices.

To further investigate the electronic properties, we first calculated the band structures and density of states (PDOS) of 2D Sr₃Ti₂S₇ chalcogenide perovskite along the high-symmetry direction in the first Brillouin zone (BZ), as shown in Figure 3a. As both the conduction-band minimum (CBM) and valence-band maximum (VBM) are located at the Γ point in the BZ, this shows that 2D Sr₃Ti₂S₇ perovskite is a direct band

gap semiconductor. The calculated electronic band gap of the 2D Sr₃Ti₂S₇ perovskite is 0.86 eV, close to the band gap of bulk Ge,⁵⁴ which is of particular interest for optoelectronic applications. Furthermore, as seen from PDOS, the Ti_d orbital contributes to the CBM, while the VBM mainly consists of the S_p orbital, with small contributions from the Ti_d and p orbital. The PDOS of the Sr, Ti, and S with different orbital are presented in Figure S7, while Sr have almost no contribution to the band edge states. Moreover, we can find from Figure 3b that 2D Sr₃Ti₂S₇ chalcogenide perovskite have work function value of 4.17 eV. Considering the *k*-point effect on the band gaps of semiconductors, we also use the HSE06^{55,56} method to examine the band structures of 2D Sr₃Ti₂S₇ with the different *k*-points in Figure S5. It is shown that the band structures and band gaps remain nearly unchanged with different *k*-points, which indicates that the *k*-point has little effect on the electronic structure of the 2D Sr₃Ti₂S₇ perovskite. Moreover, we also consider the influence of spin-orbit coupling (SOC) effect on the 2D Sr₃Ti₂S₇ perovskite (Figure S6). The energy dispersion and band gap values have little change with respect to the original band structures due to relatively small atomic numbers. Thus, it has little effect on the electronic structure.

To further analyze the band-decomposed nature of the 2D Sr₃Ti₂S₇ chalcogenide perovskite, we compute the separate distributions of charge density for CBM and VBM in Figure 3c. The CBM is dominated by Ti atoms, while VBM is mainly contributed by S atoms, accompanied by Ti atoms, which is consistent with the results of DOS. The direct band structures and large DOS in the visible-light region (<1.5 eV) indicates the high optical absorption of the 2D Sr₃Ti₂S₇ perovskite. To explore its optical applications in Figure 3d, the optical absorption spectra of 2D Sr₃Ti₂S₇ chalcogenide perovskite materials is presented. It shows that 2D Sr₃Ti₂S₇ perovskite has a wide absorption spectrum from the far-infrared to the ultraviolet region and its absorption coefficient is up to 10⁴ and 10⁵ cm⁻¹ in the infrared- and visible-light regions because of its direct band gap and large DOS in this region. Moreover, we

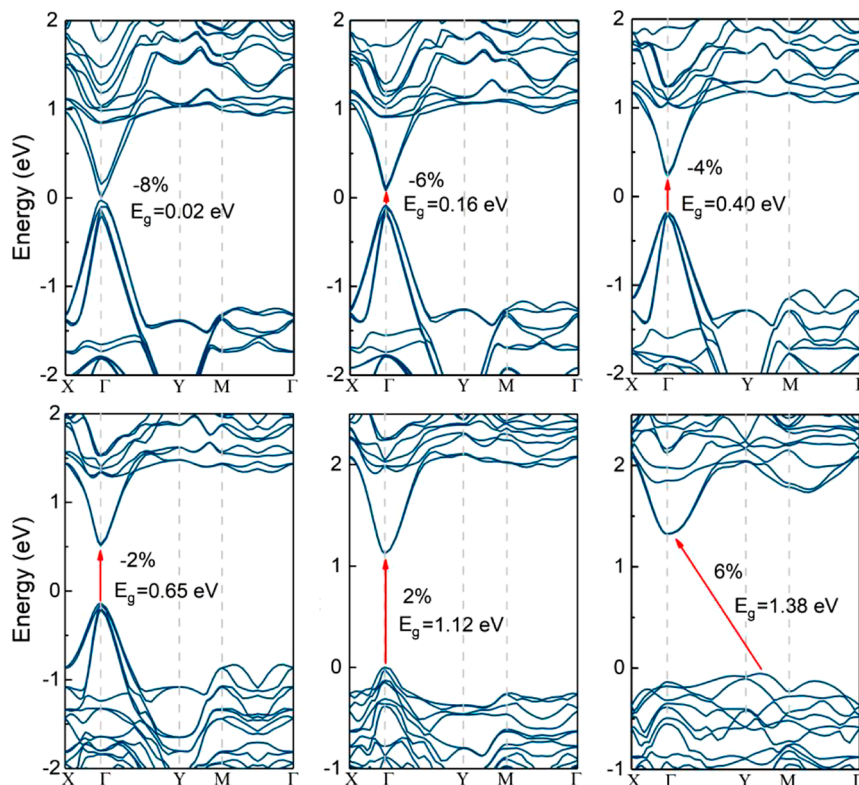


Figure 5. Band structures of 2D $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite with biaxial strains from -8 to 6% . The E_g is the band gap of two-dimensional $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite, and it is shown that biaxial strains have an effect on the band gap. The Fermi level is set to zero.

also compared the absorption spectra of 2D $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite with three highly efficient materials, MAPbI_3 , MoS_2 , and silicon. It is found that the optical spectra for 2D $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite covers the maximum wavelength. Particularly, the optical absorption is highest in the visible-light region. It can be physically attributed to strong interband optical transition in 2D $\text{Sr}_3\text{Ti}_2\text{S}_7$ and its smaller band gap than MAPbI_3 , MoS_2 , and silicon materials. To further understand the interband optical transition, we can analyze the CBM and VBM states from the orbital contribution. The CBM state is mainly contributed by the of S-p, Ti-s, and p orbitals energy level. The VBM state is dominated by the S-p, Ti-p, and d orbitals energy level. According to the selection rule for the electric dipole-allowed transitions, the allowed transitions are $|s\rangle \leftrightarrow |p\rangle$ and $|p\rangle \leftrightarrow |d\rangle$ transitions. Thus, $|p\rangle \rightarrow |s\rangle$ and $|d\rangle \rightarrow |p\rangle$ transitions for Ti atoms will dominate the optical absorption. These results demonstrates that 2D $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite can be used as light absorber material for solar cells.

To better study the materials properties of the 2D $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite, we make a comprehensive comparison between the 2D $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite and the well-known materials (MAPbI_3 , graphene, and phosphorene). The band gap, effective mass, and optical absorption of these materials are presented in Figure 4. The $\text{Sr}_3\text{Ti}_2\text{S}_7$ possesses a cubic structure and a square BZ, which implies its isotropic characteristic, such as isotropic optical properties. While phosphorene possesses a rectangle BZ, which results in its anisotropic optical properties. For effective mass, the 2D $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite possesses the lowest carrier effective mass except for graphene. The low effective mass for the 2D $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite originates from the almost linear energy dispersion near the VBM and CBM. The optical absorption coefficient of the 2D $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite is in the order of 10^5 cm^{-1} , which is similar to

other considered materials. These distinct graphene-like electronic structures, direct band gap of 0.86 eV , low carrier effective mass, and ideal optical absorption make 2D $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite potential candidates in the design of high-speed optoelectronic devices.

If the material is grown on the substrates, the external strain from the substrates may change its electronic structures.^{66–69} To consider the strain effect on the electronic properties of the 2D $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite, the electronic band structure with different biaxial strains is calculated and presented in Figure 5. When the tensile strain from 2 to 6% is applied, the band structures transfer directly to indirect. The position of the CBM is driven continuously to shift away from the Fermi level, leading to an increase of the band gap. On the contrary, when compressive strain is applied from -2 to -8% , the position of the CBM is driven continuously to shift close to the Fermi level, resulting in a narrowing of the direct band gap. Interestingly, when compressive strain is -8% , the Dirac-cone-like electronic structures appear with a “Dirac cone” lying at the Fermi level in the Γ points. The corresponding band gap is about 20 meV , which can be compared to silicene ($\sim 38 \text{ meV}$) and germanene ($\sim 0 \text{ meV}$) this indicates the ultrahigh carrier mobility.^{70,71} The result shows that compressive strain can make the band gap close and form the “Dirac-cone-like” electronic structures. From the experimental point of view, materials are generally grown on flexible substrates, such as graphene.^{70–73} In-plane compressive strain can be exerted by twisting the substrate, while tensile strain can be applied by stretching the substrate. Finally, we focus on the electron and hole effective masses at the CBM and VBM with the strain. The calculated results are shown in Table S3 and Figure S8. It is found that the electron/hole effective mass does not change significantly with compression strain applied from 0 to -8% .

The effective mass for electron and hole increasing from 0.15 (0.25) m_0 to 0.20 (0.50) m_0 , respectively. The hole effective mass is 1.23 m_0 with the compression strain of 6% due to their bigger energy dispersion slope at VBM. As a result, applying a strain can have a control over the effective mass which is the key characteristics for some optoelectronic designs.^{74–79}

3. CONCLUSIONS

In summary, we have theoretically predicted a new 2D $\text{Sr}_3\text{Ti}_2\text{S}_7$ RP chalcogenide perovskite and studied its mechanical, electronic, and optical properties by using first-principles calculations. Our results reveal that the 2D $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite possesses dynamically stable structures, direct band structures with a band gap value of 0.86 eV, and smaller effective mass (0.15/0.25 m_0 for electron/hole) than MAPbI_3 and phosphorene. Moreover, the 2D $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite possesses wide optical spectra (from the infrared-to ultraviolet-light region) and higher absorption coefficient (10^5 cm^{-1}) than MAPbI_3 , Si, and MoS_2 in the visible-light region. Interestingly, it is found that the ideal Dirac-like linear dispersion can appear near the Fermi level in the electronic band structures when compressive strain is applied. Especially, the Dirac-cone-like band structures can be realized when compressive strain is enhanced to -6% , indicating ultrahigh carrier mobility. All of these properties make the 2D $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite a promising candidate for future applications in solar cells and high-speed optoelectronic devices.

■ ASSOCIATED CONTENT

SI Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acs.cgd.3c00608>.

Computational methods in details, Vienna Ab initio Simulation Package (VASP) based on density functional theory; generalized gradient approximation (GGA) of Perdew–Burke–Ernzerhof (PBE); structures of the bulk phases of $\text{Sr}_3\text{Ti}_2\text{S}_7$ and two-dimensional $\text{Sr}_3\text{Ti}_2\text{S}_7$, respectively; support skeleton of corner-sharing TiS_6 octahedron; relaxed lattice constants and atomic positions for two-dimensional $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite material; XRD patterns of two-dimensional $\text{Sr}_3\text{Ti}_2\text{S}_7$ calculated by DFT in the $0\text{--}70^\circ$ range of angles with the radiation wavelength $\lambda = 1.54059 \text{ \AA}$ ($\text{Cu K}\alpha$), in which some of the significant peaks are labeled with the corresponding Miller indices; calculated formation energy of the two-dimensional $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite; mechanical properties for the two-dimensional $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite including elastic constants c_{ij} , shear modulus G^{2D} , layer modulus γ_{cal} , Young's modulus Y^{2D} in N/m, and Poisson's ratio ν (without dimension); distribution of the charge density in the ground states of the two-dimensional $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite (along the 001 direction); calculated band structures for the two-dimensional $\text{Sr}_3\text{Ti}_2\text{S}_7$ chalcogenide perovskite with the different k -points meshes; simulated band structure of two-dimensional $\text{Sr}_3\text{Ti}_2\text{S}_7$ both with and without the SOC effect using GGA-PBE functional; PDOS of the two-dimensional $\text{Sr}_3\text{Ti}_2\text{S}_7$ chalcogenide perovskite using the HSE06 functional; calculated effective mass and band gap for the two-dimensional $\text{Sr}_3\text{Ti}_2\text{S}_7$ perovskite with different biaxial strains using accurate HSE06

functional; and calculated effective mass of electrons/hole with different biaxial strains from -8 to 6% (PDF)

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