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Title: Metabolic engineering for the optimization of hydrogen production in *Escherichia coli*: A review

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Abstract: Hydrogen is a potential sustainable energy source and it could become an alternative to fossil fuel combustion, thus helping to reduce greenhouse gas emissions. The biological production of hydrogen, instead of its chemical synthesis, is a promising possibility since this process requires less energy and is more sustainable and eco-friendly. Several microorganisms have been used for this purpose, but *Escherichia coli* is the most widely used in this field. The literature in this area has increased exponentially in the last 10 years and several strategies have been reported in an effort to improve hydrogen production. These strategies include heterologous gene expression, adaptive evolution and metabolic and protein engineering. On the other hand, culture conditions, including the use of carbon sources such as glycerol, a waste product from the biodiesel industry, have also been considered. This review includes a discussion of the hydrogenase complexes responsible for the hydrogen synthesis in *E. coli* and the central carbon metabolism pathways connected to this process. The different strategies applied to obtain engineered strains in which the carbon and electron fluxes are redirected towards hydrogen synthesis are also discussed. Yields and productivities of the most relevant engineered strains reported using several carbon sources are also compared.

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30th, October 2018

Dear Editor
Biotechnology advances

Please, find enclosed the manuscript entitled "**Metabolic engineering for the optimization of hydrogen production in *Escherichia coli*: A review**". The aim of this review is to study the mechanisms of hydrogen synthesis by *Escherichia coli* and the state of the art of the metabolic engineering strategies used to improve hydrogen yield and productivity in this microorganism. These strategies are summarized and illustrated in a metabolic diagram.

The authors of this work are: Antonio Valle, Domingo Cantero and Jorge Bolívar and all of them declare no financial or commercial conflict of interest and we proposed Jorge Bolívar Pérez and Antonio Valle Gallardo as corresponding authors for this paper. This is an original work and it has not been submitted previously to any journal.

Jorge Bolivar & Antonio Valle



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Jorge Bolivar & Antonio Valle

1 **Metabolic engineering for the optimization of hydrogen production in**
2 ***Escherichia coli*: A review**

3

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30 **Abstract**

31 Hydrogen is a potential sustainable energy source and it could become an alternative to
32 fossil fuel combustion, thus helping to reduce greenhouse gas emissions. The biological
33 production of hydrogen, instead of its chemical synthesis, is a promising possibility
34 since this process requires less energy and is more sustainable and eco-friendly. Several
35 microorganisms have been used for this purpose, but *Escherichia coli* is the most
36 widely used in this field. The literature in this area has increased exponentially in the
37 last 10 years and several strategies have been reported in an effort to improve hydrogen
38 production. These strategies include heterologous gene expression, adaptive evolution
39 and metabolic and protein engineering. On the other hand, culture conditions, including
40 the use of carbon sources such as glycerol, a waste product from the biodiesel industry,
41 have also been considered. This review includes a discussion of the hydrogenase
42 complexes responsible for the hydrogen synthesis in *E. coli* and the central carbon
43 metabolism pathways connected to this process. The different strategies applied to
44 obtain engineered strains in which the carbon and electron fluxes are redirected towards
45 hydrogen synthesis are also discussed. Yields and productivities of the most relevant
46 engineered strains reported using several carbon sources are also compared.

47

48 **Keywords:** hydrogen, *Escherichia coli*, glycerol, glucose, metabolic engineering,
49 hydrogenases.

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54 **1. Introduction**

55 Global warming is caused by greenhouse gas emissions from fossil fuel combustion
56 (Montzka et al., 2011) and this has been the primary source to satisfy the world's energy
57 demand (Davila-Vazquez et al., 2008). The scientific community agree that it is
58 necessary to mitigate the depletion of fossil fuel resources and to protect the
59 environment against their negative effects by exploring new renewable and
60 environmentally friendly energy sources that could replace fossil fuels (Dharmadi et al.,
61 2006).

62 Of all of the renewable energy sources, which include ethanol or algal diesel, hydrogen
63 (H_2) is an interesting alternative to be explored since it has several advantages when
64 compared to fossil fuels. These advantages include the following: (i) molecular
65 hydrogen has a higher energy bond by a factor of around 3 when compared to carbon
66 fuels (Hallenbeck and Ghosh, 2009; Momirlan and Veziroglu, 2005; Mudhoo et al.,
67 2011), (ii) nontoxic by-products are produced by breaking the H_2 covalent bond, (iii)
68 hydrogen is a renewable, efficient and clean energy source (Hansel and Lindblad, 1998)
69 and (iv) it is utilized for fuel cells in portable electronics, power plants and internal
70 combustion engines (Dunn, 2002).

71 However, H_2 is currently considered as a clean energy carrier rather than an energy
72 source because it does not come from natural sources. Indeed, H_2 is produced mainly by
73 the electrolysis of water (Armaroli and Balzani, 2011), by the water-gas shift reaction
74 (Yi and Harrison, 2005), and as a by-product from petroleum refining or chemical
75 production (Das and Veziroglu, 2001; Westermann et al., 2007). These methods for H_2
76 production are neither ecological nor environmentally friendly because they require
77 high energy input to generate H_2 gas (Das and Veziroglu, 2001). The commercial
78 production of H_2 by electrolysis of water achieves an efficiency of 75%; however, the

79 cost of this process is currently several times higher than that of hydrogen produced
80 from fossil fuels (Ewan and Allen, 2005). Moreover, most of the production depends on
81 fossil resources and this is clearly unsustainable (Edwards et al., 2007). It is therefore
82 necessary to improve this technology, not only by the development of H₂ storage and
83 distribution, but also through an efficient and sustainable production method. In this
84 respect, over the last few years the European Commission has developed several
85 activities through the technological initiative ‘Fuel Cells and Hydrogen Joint
86 Undertaking’ (FCH JU) under Horizon 2020 ((FCH)). This program is a unique public-
87 private partnership that supports research, technological development and
88 demonstration activities in fuel cell and H₂ energy technologies in Europe. Given this
89 background, biological H₂ or biohydrogen production is a feasible alternative when
90 compared to the commonly used technologies outlined above. The biological approach
91 has significant advantages over chemical methods since it could be performed at
92 relatively low temperatures (25–37 °C), at atmospheric pressure and at relatively high
93 rates (Trchounian and Trchounian, 2015). Efficient separation techniques from
94 bioreactor headspaces have also been developed (Fan, L.S. et al., 2009). The biological
95 production of H₂ can be classified into three major categories: (1) biophotolysis of water
96 using algae and cyanobacteria, (2) photofermentation of organic compounds by
97 photosynthetic bacteria and (3) dark fermentative production by heterotrophic
98 microorganisms (Hallenbeck, 2005). The theoretical yields of H₂ from dark
99 fermentations depend on the type of organism used in the fermentation (Mathews and
100 Wang, 2009) but in general it is more efficient than photosynthetic processes (Horizonte
101 2020 Comisión Europea).

102

103 Of all the H₂-producing microorganisms described in the literature, the bacterium
104 *Escherichia coli* is one of the most widely used for anaerobic fermentation due to its
105 well-known metabolism and also because it is relatively easy to rewire its metabolic
106 pathways through genetic engineering techniques (Atsumi and Liao, 2008; Blankschien
107 et al., 2010; Chen et al., 2013; Choi et al., 2012; Clomburg and Gonzalez, 2010;
108 Dharmadi et al., 2006; Ganesh et al., 2012; Joung et al., 2011; Maeda, Toshinari et al.,
109 2008; Nakamura and Whited, 2003; Trinh and Sreenc, 2009; Valle et al., 2012; Wang et
110 al., 2011; Yu et al., 2010). The interest in this field is currently increasing, as evidenced
111 by the exponential growth in publications since 2000. For instance, a search in
112 SCOPUS using the terms ‘hydrogen’ (physics, chemistry, environmental engineering,
113 biochemistry, technology, etc.), ‘hydrogen and *Escherichia coli*’ and ‘hydrogen and
114 *Escherichia coli* and engineering’ indicates an exponential growth in the last 18 years
115 (Fig. 1). It noteworthy that the growth rates of publications with the terms ‘hydrogen’
116 versus ‘hydrogen and *Escherichia coli*’ have been very similar in the last ten years,
117 although the number of publications related to engineering and *E. coli* to improve H₂
118 production has increased significantly since 2000, which reflects the growing number of
119 researchers interested in this topic and probably indicates a change in the paradigm of
120 biofuel production in the near future.

121 In anaerobic fermentative hydrogen production (FHP), numerous carbon sources such
122 as glycerol, glucose, formate, sucrose, lactose, galactose, arabinose, fructose, gluconate,
123 maltose, mannitol, sorbitol, trehalose, and xylose can be used (Goyal et al., 2013;
124 Rosales-Colunga and De León Rodríguez, 2015; Trchounian, 2015; Trchounian and
125 Trchounian, 2015). Nevertheless, in order to make H₂ production by *E. coli*
126 economically viable on a large-scale, organic substrates from agricultural, chemical,
127 food or by-product wastes should be used (Trchounian, 2015). In this regard, glycerol

128 has become an abundant and inexpensive carbon source because it is generated as a by-
129 product in the biodiesel fuel industry (Sarma et al., 2012). This compound has been
130 widely used in the generation of H₂ by *E. coli* and other bacteria (Cofré et al., 2016;
131 Dharmadi et al., 2006; Hu and Wood, 2010; Kim et al., 2010; Sarma et al., 2012; Tran
132 et al., 2015; Trchounian and Trchounian, 2015; Valle et al., 2015a). Several advances
133 have been made in studies that address the possibility of improving yields and
134 productivities by using glycerol (Akhtar and Jones, 2008; Cofré et al., 2016; Henstra et
135 al., 2007; Hu and Wood, 2010; Trchounian and Trchounian, 2015; Valle et al., 2015a;
136 Valle et al., 2017), glucose (Maeda et al., 2007a; Seol et al., 2014; Seol et al., 2016;
137 Trchounian and Trchounian, 2014b; Yoshida et al., 2006; Zhu et al., 2011) or even both
138 together in the same culture medium (Trchounian et al., 2011; Trchounian et al., 2014;
139 Trchounian and Trchounian, 2013) as the main carbon sources. Glycerol is a highly
140 reduced carbon source compared to sugars and it therefore offers an effective
141 opportunity to obtain H₂ and other reduced products by bacteria (Ganesh et al., 2012;
142 Khanna et al., 2012). The energy content of pure glycerol is 19.0 MJ/kg, although it is
143 only 25.30 MJ/kg for crude glycerol, probably due to the presence of methanol and
144 traces of biodiesel (Strathclyde). Such a high energy content in crude glycerol indicates
145 its great potential to be an effective substrate for H₂ production.

146

147 The margin for increasing the efficiency of H₂ production is potentially high since, from
148 a microbial evolutionary perspective, the low biological H₂ yields in the natural strains
149 can be attributed to the fact that microorganisms capable of FHP have developed their
150 metabolic pathways preferentially for cell growth rather than for H₂ synthesis.
151 Therefore, the yields and productivities can be improved significantly by metabolic
152 engineering, which eventually will depend on the carbon source used in the

153 fermentation. For this reason, some efforts have been focused on obtaining H₂ from
154 glycerol (Cai et al., 2011; Davila-Vazquez et al., 2008; Rosales-Colunga and De León
155 Rodríguez, 2015). In this review, we describe the state of the art for the strategies used
156 to enhance H₂ production in *E. coli* by using glycerol as the main carbon source and the
157 results are compared with those obtained on using glucose.

158

159 **2. Biosynthesis of H₂ in *E. coli***

160 *2.1 Fermentation of glycerol or glucose: from carbon source to formate*

161

162 H₂ is formed during the anaerobic mixed-acid fermentation in which pyruvate (PYR),
163 the final product of glycolysis, is converted into formate and acetyl-CoA, the precursors
164 of CO₂ + H₂ and ethanol, respectively (Dharmadi et al., 2006). Fermentation of glycerol
165 starts with its conversion to dihydroxyacetone phosphate (DHAP) in a process that is
166 mediated by a two-branch pathway: the oxidative branch by glycerol dehydrogenase
167 (GldA) and dihydroxyacetone kinase enzymes (DhaKLM), and the reductive branch by
168 glycerol kinase (GlpK) and glycerol 3-P dehydrogenase (GlpABC) enzymes. DHAP can
169 be metabolized in the glycolysis pathway to PYR, although a unique characteristic of
170 glycerol metabolism is that the conversion of phosphoenolpyruvate (PEP) into PYR is
171 coupled to DHA phosphorylation. In this process 1 NAD⁺ is reduced, in the assimilation
172 of glycerol and in the synthesis of 1,3-bisphosphoglycerate (1,3-BPG) (Cintolesi et al.,
173 2012), to yield 1 mol PYR per mol glycerol (Fig. 2). In the case of glucose
174 fermentation, PYR is generated from glycolysis and 2 NAD⁺ is also reduced in the same
175 step (conversion of glyceraldehyde 3-phosphate to 1,3-BPG), but the fermentation yield
176 is 2 mol PYR per mol glucose, i.e., double that obtained with glycerol (Fig. 2).
177 Independently of the substrate for FHP, the synthesis of formate from PYR is essential
178 for the production of H₂, which is formed concomitantly with Acetyl-CoA by pyruvate

179 formate lyase (PFL) in microaerobic and anaerobic conditions. Acetyl-CoA is then
180 reduced to ethanol to allow the NADH oxidation for glycolysis maintenance. PEP and
181 PYR are also the precursors for the synthesis of other fermentative end-products such as
182 succinate, acetate, acetolactate, lactate (Sawers and Clark, 2004), 1,2-propanediol
183 (PDO) (Murarka et al., 2008) and even butyric acid under certain conditions (Lugg et
184 al., 2008) (Fig. 2).

185 2.2 From formate to H₂: FHL systems and hydrogenases

186
187
188 One of the key elements in the strategy for improving H₂ production in *E. coli* is to
189 understand how the Formate Hydrogen Lyase (FHL) systems work. These multi-protein
190 complexes operate strictly under anaerobic conditions and they are not only able to
191 synthesize H₂ and CO₂ in a 1:1 molar ratio by decomposition of formate, but they are
192 also involved in H₂ recycling ($H_2 \rightarrow 2H^+ + e^-$). This is not the only way for the cell to
193 produce H⁺ as it can also be formed in the oxidation of NADH ($NADH \rightarrow NAD^+ H^+$),
194 which is produced during glycolysis (Das and Veziroğlu, 2001) and influences the way
195 in which FHL works.

196

197 H₂ synthesis is a process that is highly dependent on several factors such as pH_{out},
198 carbon source, temperature, redox potential (E_h), presence of exogenous electron
199 acceptors, ratios of end products and H₂ concentration (Trchounian, 2015; Trchounian,
200 2012). These factors probably determine which FHL is more active. FHL consists of
201 two enzyme complexes – formate dehydrogenase H (Fdh-H) and [Ni-Fe] hydrogenase
202 enzymes (Hyd) – in *E. coli* (Bagramyan and Trchounian, 2003; Sawers, 2005) and
203 other facultative anaerobe microorganisms (Sinha et al., 2015). The study of H₂
204 production has led to the discovery of multiple and reversible Hyd (Hyd-1, Hyd-2, Hyd-
205 3 and Hyd-4) enzymes. FHL-1 is composed of Hyd-3 complex encoded by the

206 *hycABCDEFGHI* operon, which is upregulated by an FhlA protein codified by the *fhl*
207 gene (Self and Shanmugam, 2000). This complex is well characterized and is the main
208 H₂-producing system during glycerol and glucose fermentation at acidic pH and it has
209 been widely described and reviewed by Trchounian's group (Trchounian, 2015;
210 Trchounian et al., 2012; Trchounian and Trchounian, 2015). The FHL-2 system is
211 composed of Hyd-4 and Fdh-H and it is associated with ATP_{ase}. This is the least known
212 of all the FHL and hydrogenase enzymes but it is also able to produce H₂ in both
213 glucose- and glycerol-based media (Bagramyan et al., 2002; Mnatsakanyan et al., 2004;
214 Trchounian and Trchounian, 2014b), although there is some controversy in this regard
215 (Noguchi et al., 2010; Self et al., 2003; Skibinski et al., 2002) and its physiological role
216 is not completely clear (Poladyan and Trchounian, 2009; Trchounian et al., 2012) (Fig.
217 2).

218
219 The other two hydrogenases, Hyd-1 and Hyd-2, play a minor role in H₂ synthesis and
220 they are probably more related to the recycling of H₂, which again depends on pH,
221 carbon source and H₂ concentration (Pinske et al., 2012), thus contributing to the
222 reduced quinone pool (Trchounian et al., 2012). Hyd-1 is encoded by the *hya* operon
223 and, despite the fact that it is not essential for anaerobic growth, it is needed to respond
224 to the [pH]_{out} shift from alkaline to acidic (Brøndsted and Atlung, 1994) and affects H⁺
225 translocation across the membrane. Since the proton motive force (Δp) changes during a
226 [pH]_{out} shift, it has been suggested that Hyd-1 is required to maintain Δp in an energy-
227 conserving manner (Trchounian, 2015). Hyd-2 is encoded by the *hyb* operon, the
228 expression is higher in alkaline media (Brøndsted and Atlung, 1994) and it is thought to
229 operate as an oxidizing H₂ enzyme under glycerol and glucose fermentation at acidic
230 pH. However, it is involved in H₂ production under glycerol fermentation at neutral and

231 slightly alkaline pH (Trchounian and Trchounian, 2009). Several pieces of evidence
232 suggest that either Hyd-1 or Hyd-2 operate preferentially under different conditions
233 (Trchounian, 2015).

234

235 The Hyd enzymes therefore appear to play important roles in increasing the fitness of
236 the bacteria to survive in different environments (Trchounian et al., 2012) (Fig. 2). The
237 physiological role of H₂ production mediated by Hyd enzymes seems to involve the
238 discharge of reducing power excess when other suitable electron acceptors, such as O₂,
239 are absent and for this reason H₂-producing Hyd enzymes are rapidly inhibited by O₂
240 (Laurinavichene et al., 2001; Lukey et al., 2010; Trchounian et al., 1998). It is
241 interesting to consider the relationship between H₂ and the H⁺ cycle, since Hyd enzymes
242 have H⁺ translocation activity and they interact with the F₀F₁-ATP_{ase}, which is also
243 involved in the H⁺ cycle (Rossmann et al., 1991). FHL-2 function requires catalytically
244 competent F₀F₁-ATP_{ase} and it is associated with K⁺ uptake via the ATP-dependent TrkA
245 transporter. In this hydrogenase the uptake of K⁺ is essential for H₂ synthesis, since the
246 mutation of the TrkA protein blocked the synthesis at [pH]_{out} 7.5 on using glucose as the
247 carbon source (Trchounian, 2015). This is a key mechanism in the generation of Δp and
248 the regulation of [pH]_{in}. In summary, Hyd-2 could be considered as a reversible enzyme
249 that can contribute to H₂ evolution more significantly in the absence of Hyd-3 and Hyd-
250 1. On the other hand, the oxidizing activity Hyd-1 and Hyd-2 depends on the presence
251 of an active F₀F₁-ATP synthase during growth at pH 7.5 and 5.5 with glucose or
252 glycerol fermentation. There is an inverse correlation between hydrogenase-1 and -2
253 activity at pH (5.5 and 7.5) during fermentative growth on glucose and a direct
254 correlation between hydrogenase activity at pH (5.5 and 7.5) during glycerol
255 fermentation. Since the H⁺ gradient is essential to ATP and H₂ synthesis, it seems that

256 both Hyd-1 and -2 play an important role in energy conservation in fermenting *E. coli*
257 cells (Trchounian et al., 2011). Nonetheless, further studies are required to elucidate the
258 physiological relationship between the different hydrogenases and their interaction with
259 these factors (i.e., pH and carbon source) within the bacterial membrane. This research
260 might help to provide a new perspective in the bioenergetics of fermentation.

261

262 **3. Several environmental parameters that affect H₂ production in *E. coli***

263

264 It can be deduced from the above discussion of FHL complexes that the environment
265 has a marked impact on H₂ production. The most influential factors are outlined below.

266 *3.1 Formate transport and pH*

267

268 Formate or formic acid is a monovalent anionic substance at neutral and slightly
269 alkaline pH and this is one of the major mixed-acid fermentation products in many
270 obligate and facultative anaerobes. This molecule can be used for *de novo* nucleotide
271 biosynthesis in *E. coli* (Jensen et al., 2008). Formate is produced by the action of PFL
272 and can be exported from the cytoplasm or, in the presence of terminal electron
273 acceptors, can be eliminated by the periplasmic formate dehydrogenase oxygen-
274 dependent (FDH-O) or nitrate-dependent (FDH-N). However, formate is re-imported to
275 the cytoplasm and used as a substrate for H₂ production by the formate dehydrogenase
276 H-dependent (FDH-H) as a component of FHL-1 when the cells enter the late
277 exponential growth phase or when terminal electron acceptors are absent (Sawers,
278 2005). As a weak acid, formate may act as an uncoupling factor that dissipates Δp and
279 therefore leads to a decrease of $[pH]_{out}$ potentiates. Formate transport in *E. coli* is
280 carried out by the two formate channels named FocA (member of formate-nitrite
281 transporter family FNT), which has an important role in both the export and import of

282 formate (Beyer et al., 2013), and the putative formate transporter FocB, which is less
283 well characterized than FocA (Wang et al., 2009) (Fig. 2). These channels appear to
284 control formate translocation in a pH-dependent manner. However, it should be noted
285 that during glucose fermentation at a pH value higher than 7 the FHL-2 complex is
286 probably responsible for H₂ production, whereas when the pH is below 7 FHL-1 it is
287 more active (Trchounian and Trchounian, 2014a). Formate can be transported across the
288 membrane under different conditions. The FocB transporter preferentially imports the
289 formate to produce H₂ at different [pH]_{out} with any carbon source (glucose or glycerol).
290 However, FocA exports formate during glucose fermentation at any pH and with
291 glycerol at pH 7.5, but it imports formate at pH values below 7 (Fan, Z. et al., 2009;
292 Trchounian and Trchounian, 2014a). From the results of these studies it can be deduced
293 that H₂ production during growth on different carbon sources depends on the *focA* and
294 *focB* genes. These channels seem to control formate translocation in a pH-dependent
295 manner and when both channels are absent it may lead to enhanced H₂ production
296 (Trchounian and Trchounian, 2014a). However, formate channels do not function in a
297 separate way and some interactions with other proteins have been identified. A study
298 published by Falke et al. (Falke et al., 2016) highlighted the glycyl-radical enzyme
299 (GRE) TdcE protein as a specific interaction partner of the formate-specific channel
300 FocA, which would play a similar role to that of PflB. Falke et al. (Falke et al., 2016)
301 ruled out the possibility that FocA-binding is a general property of GREs and only TdcE
302 and PflB interact with FocA.

303 *3.2 pH, hydrogenase activity and carbon source*

304

305 Trchounian's group has studied extensively the dependence of hydrogenase activities on
306 pH and their crosstalk functions (Trchounian, 2015; Trchounian et al., 2012;
307 Trchounian and Trchounian, 2015). H₂ production depends on [pH]_{out} and carbon

308 source (glycerol or glucose) in the culture medium and FHL systems are involved in the
309 regulation of $[pH]_{in}$ (Bock and Sawers, 2006); the results can be summarized as follows
310 and in Table 1:

311

312 (1) At $[pH]_{out}$ 5.5 hydrogenases Hyd-1 and Hyd-2 seem to have similar functions in
313 oxidizing hydrogen but Hyd-3 is involved in H_2 production, which has the
314 highest rate when H_2 uptake by Hyd-1 and Hyd-2 showed low activity.

315 (2) At pH 6.5 Hyd-3 is the main H_2 -producing enzyme during glycerol
316 fermentation, although it might operate in a reverse direction and have
317 significant H_2 uptake activity with glucose. Hyd-2 and Hyd-1 are involved in H_2
318 production in glycerol-containing media but operate in oxidizing mode in
319 culture media that contain both glucose and formate.

320 (3) At pH 7.5 Hyd-3 with glycerol shows H_2 uptake activity but Hyd-2 and Hyd-1
321 are involved in H_2 production in glycerol-containing media, as described for pH
322 6.5.

323 *3.3 H_2 and CO_2 partial pressure*

324

325 H_2 is produced under strictly anaerobic conditions and studies carried out in *E. coli* have
326 generally been performed in closed vials with headspace, although in several cases it
327 has been reported that headspace pressure can have a negative effect on the synthesis of
328 H_2 , theoretically by the recycling operating mode of H_2 for Hyd-1 and/or Hyd-2
329 (Maeda, Toshinari et al., 2008). Therefore, low partial pressure fermentation is often
330 applied to enhance H_2 production, as described in several studies (Dharmadi et al.,
331 2006; Kim et al., 2006; Maeda, Toshinari et al., 2008; Mizuno et al., 2000). The
332 concentration of CO_2 also has a negative effect on the H_2 production rate and yield
333 because the synthesis of succinate consumes CO_2 and PYR via OAA synthesis. This

334 latter pathway competes with the synthesis of formate and therefore H₂ production
335 decreases – as a consequence, rapid gas removal and separation of H₂ are essential to
336 increase the production.

337

338 **4. Metabolic engineering strategies for improving the synthesis of H₂**

339 Metabolically engineered microbial strains can give higher yields of target products for
340 biotechnological applications. Different strategies for metabolic engineering have been
341 carried out in the last 10 years with several microorganisms, although *E. coli* is the most
342 commonly used for improving H₂ production. Metabolic pathways in *E. coli* have been
343 widely studied (Blattner et al., 1997), but there are still some gaps in our knowledge
344 about H₂ production. For instance, very few studies have focused on the influence of
345 the carbon source in H₂ production. In this sense, several recent studies have shown
346 how the manipulation of certain genes can be beneficial for H₂ production using glucose
347 as the carbon source but they do not have the same effect on using glycerol and may
348 even have the opposite effect. For example, it has been reported that Hyd-1 and Hyd-2
349 are H₂-consuming enzymes during glucose metabolism, whereas they are H₂-producing
350 enzymes in the presence of glycerol at neutral and slightly alkaline pH (Trchounian et
351 al., 2011). In addition, anaerobic growth with glycerol is somewhat lower than with
352 glucose (Hu and Wood, 2010). Given the advances in the knowledge of metabolomics
353 and genetic manipulation techniques, the strategies for improving H₂ in *E. coli* have
354 evolved. In the past, genetic modifications were achieved through random processes
355 such as exposure to ultraviolet light, chemical mutagenesis etc., but nowadays genetic
356 engineering is widely used to obtain engineered strains with higher productivities
357 (Maeda et al., 2012). With the help of these new techniques, and based on the wide
358 availability of metabolic information on *E. coli*, several metabolic engineering strategies

359 to improve H₂ production can be designed *a priori* (Manish et al., 2007). It should be
360 noted, however, that genetic engineering is not always fruitful, as undesirable effects are
361 often encountered in the engineered organisms (Colletti et al., 2011). There is a need for
362 an extensive analysis and detailed understanding of metabolic fluxes and their
363 regulatory circuits that lead to H₂ formation. It is important to have a fundamental
364 knowledge of how the metabolic pathway flux responds to varying genetic and
365 environmental perturbations and whether the pathways can be deliberately redirected
366 towards H₂ synthesis rather than to the production of unwanted metabolites. The state of
367 the art of metabolic engineering in *E. coli* for H₂ production on using glycerol or
368 glucose as carbon sources is reviewed below and the most important strategies and
369 results reported to date are described and summarized in Table 2.

370

371 *4.1 Screening/High-throughput screening*

372

373 Screening and high-throughput screening methods can be useful to find new phenotypes
374 of H₂-producer knock out strains or potential genes to be overexpressed, as well as to
375 engineer further strains to improve yields and productivities. Mohd et al. (Mohd Yusoff
376 et al., 2013) reported four pseudogenes involved in H₂ metabolism with important roles
377 in cell physiology in a screening of 3985 *Escherichia coli* mutants from the Keio
378 Collection (Baba et al., 2006) by using H₂ chemochromic membranes. In another
379 screening of 150 single knock out mutants from the same collection, 12 novel strains
380 with enhanced yields for H₂ and/or ethanol production were detected in cells grown in a
381 glycerol-based medium by analyzing H₂ and ethanol production by gas chromatography
382 (Table 2B) (Baba et al., 2006; Valle et al., 2015a). In the same work, the knock out
383 genes that most significantly improved the parameters evaluated (including *gnd* and

384 *tdcE* genes) were combined and the resultant quintuple mutant significantly increased
385 the molar yields of H₂ production and glycerol consumption.

386

387 Barahona et al. (Barahona et al., 2016) suggested that there is a limitation in the
388 discovery or synthesis of better H₂-producing enzymes due to the absence of methods
389 for the high-throughput screening of H₂ production in biological systems. In an effort to
390 overcome this drawback, they engineered a natural H₂-sensing system of *Rhodobacter*
391 *capsulatus* to direct the emission of LacZ-dependent fluorescence in response to the
392 nitrogenase-produced H₂. This system was used in combination with fluorescence-
393 activated cell sorting flow cytometry to screen large libraries of nitrogenase Fe-protein
394 variants generated by random mutagenesis. This method was used to select an H₂-
395 overproducing Fe-protein variants lacking 40% of the wild type amino acid sequence,
396 and subsequently microbial H₂ production was improved by using directed evolution of
397 nitrogenases and hydrogenases in the selected protein variants (Barahona et al., 2016).
398 This strategy could be adapted to *E. coli* in order to improve H₂ production by
399 combining mutagenesis and high throughput screening.

400 *4.2 Mutagenesis of competitive pathways*

401

402 Several end-products synthesized in the mixed-acid fermentation, such as lactate,
403 acetate, ethanol and succinate, must be exported out from the cell in order to maintain
404 the NADH/NAD⁺ balance and intracellular pH.

405

406 Since these metabolites are reduced molecules, their syntheses are presumably
407 competitive pathways of H₂ production. Therefore, the removal of the synthesis of one
408 or more of these end-products should theoretically result in an increased flux towards

409 H₂ synthesis. This strategy has been applied to several reported strains (Sawers and
410 Clark, 2004).

411 • Lactate. Deletion of the lactate dehydrogenase gene (*ldh*) helps to drain PYR
412 flux towards H₂ (Table 2A and B), although this occurs only when the medium
413 is acidic, since this condition stimulates the formation of lactate (Hallenbeck and
414 Ghosh, 2012).

415 • Succinate is another end-product whose synthesis can be reduced by knocking
416 out the fumarate reductase (FRD) operon (*frdABCD*) responsible for the
417 conversion of fumarate to succinate. This would increase intracellular PYR and
418 therefore H₂ production. Several publications have concerned the use of single
419 and double *frd* knock out genes that improve H₂ production in glucose- and
420 glycerol-based media (Table 2A). However, the improvement surprisingly varies
421 significantly between the single knock outs (Valle et al., 2015b) and the multiple
422 mutant (Valle et al., 2017), thus indicating that each FRD subunit may have
423 different roles, with H₂ production being more favorable when FrdB and FrdC
424 subunits are deleted.

425 • Ethanol. Blocking the synthesis of ethanol by inactivating alcohol
426 dehydrogenase (*adhE*) severely impaired H₂ formation and cell growth when
427 glycerol was employed as the carbon source (Murarka et al., 2008; Tran et al.,
428 2014). This finding indicates that the synthesis of ethanol is essential to maintain
429 the redox balance and also to allow the synthesis of formate and Acetyl-CoA.

430 • Acetate production can be avoided by removing acetate-producing enzymes
431 such as AckA-PtA (acetate kinase, phosphate acetyl-transferase) or PoxB
432 (pyruvate oxidase) (Murarka et al., 2008), although deletion of the *poxB* gene

433 did not increase H₂ synthesis because this mutation reduced cell viability (Tran
434 et al., 2014).

435 • Butyrate is synthesized from acetyl-CoA via crotonyl-CoA. Despite the fact that
436 this metabolite has been detected in the culture under particular growth
437 conditions (Rosales-Colunga and De León Rodríguez, 2015), strategies for
438 removing butyrate have not been used to date, probably because its production is
439 relatively low.

440 • 1,2-Propanediol is synthesized by methylglyoxal synthase (MgsA) from DHAP
441 and it has been reported that the single knock out strain for the gene of this
442 enzyme increases slightly the production of H₂ in glycerol media (Tran et al.,
443 2014) (Table 2A).

444 • The conversion of PYR into acetyl-CoA is catalyzed by the PYR dehydrogenase
445 complex (PDH) under aerobic conditions, but in anaerobic conditions PYR is
446 transformed into acetyl-CoA and formate by PYR formate lyase (PFL), which is
447 essential for H₂ synthesis. However, under microaerobic conditions the two
448 enzymes can be functional (Sawers and Clark, 2004). Despite PDH operon
449 genes (*aceEF*, *lpd*) having low expression levels in anaerobic conditions, the
450 activity of this enzyme could be a competitive pathway in formate synthesis and
451 therefore deletion of the *aceE* subunit resulted in an increased H₂ yield with a
452 glucose-based medium (Maeda et al., 2007a). However, significant
453 improvements were not observed on using glycerol (Tran et al., 2014).

454 • Conversion of PEP into OAA by the anaplerotic enzyme PEP carboxylase (Ppc)
455 can be considered a competitive reaction since it diverts the C₃ compound
456 toward succinate. For this reason, the knock out of this gene increases H₂
457 productivity significantly (Tran et al., 2014) (Table 2A).

458 • Hydrogenases (Hyd-1, -2, -3, -4) are the key controlling enzymes in the
459 synthesis or oxidation of H₂ and their function depends on pH and carbon
460 sources, as described previously. Thus, removal of the H₂-consuming activity of
461 Hyd-1 and/or Hyd-2 by knocking out the large subunit of both enzymes
462 ($\DeltahyaB\DeltahybC$) in an acidic culture medium is a strategy that has been
463 successfully employed by several groups using both glycerol- (Maeda, T. et al.,
464 2008) and glucose-based media (Ghosh and Hallenbeck, 2010; Maeda et al.,
465 2007a; Maeda, Toshinari et al., 2008; Seol et al., 2014; Seol et al., 2016). In
466 addition, the overexpression of Hyd-3, the most important enzyme that forms
467 FHL-1, in FHP when glycerol is used at pH 7.5 increases H₂ production. Hence,
468 the FHL expression may be manipulated by overexpression of *fhlA* (Horizonte
469 2020 Comisión Europea) or deletion of the *hycA* gene that encoded a repressor
470 of FhlA expression (Penfold et al., 2003; Yoshida et al., 2005). Interestingly, a
471 combination of *fhlA* overexpression and disruption of *hycA* did lead to a 2.8-fold
472 increase in the H₂ formation rate (Yoshida et al., 2006) (Table 2A, B).

473

474 There are two additional formate dehydrogenases encoded by *fdnG* (a subunit of
475 nitrogen-dependent formate dehydrogenase-N) and *fdoG* genes (a subunit of oxygen-
476 dependent formate dehydrogenase-O) that consume formate but do not produce H₂.
477 However, the deletion of *fdoG* did not improve the H₂ yield in a glycerol-based medium
478 and the *fdnG* mutant showed only a moderate improvement in H₂ production (Maeda et
479 al., 2007a; Rossmann et al., 1991).

480

481 Other enzymes or regulator proteins involved in formate consumption that may compete
482 with the synthesis of H₂ are outlined below:

483

484 (1) Nitrate reductase A (whose α -subunit is encoded by the *narG* gene) consumes
485 formate in the conversion of nitrate into nitrite by using electrons produced from
486 formate by FDH-N. Deletion of the *narG* gene leads to a very modest H₂ productivity
487 but in combination with other mutations the H₂ productivity increases significantly
488 (Maeda et al., 2007a; Tran et al., 2014).

489 (2) FNR protein is a global DNA-binding transcriptional regulator that stimulates the
490 transcription of several genes required for fermentation and anaerobic respiration and
491 represses others like *hyf* operon, which encoded Hyd-4 (Salmon et al., 2003). As a
492 consequence, deletion of the *fnr* gene leads to a three-fold increase in HyfR expression
493 as the activator of *hyf* operon.

494 (3) The NarL protein is a global transcription repressor for the genes of FhlA,
495 PFL, and nickel transporter *nik* operon. Inactivation of the *narL* gene led to a two-fold
496 increase in the molar yield of H₂, with similar growth to that of the wild-type strain
497 (Fan, Z. et al., 2009).

498 (4) Formate channels (FocA and FocB) appear to control formate translocation
499 in a pH-dependent manner and the absence of both formate channels may lead to
500 enhanced FHP (Trchounian and Trchounian, 2014a). Several strategies for metabolic
501 engineering by manipulating these protein transporters with glycerol have been applied
502 to the $\Delta focA\Delta focB$ double mutant strain (Table 2A). However, the deletion of *focA* in
503 multiple mutant strains did not increase H₂ formation and, indeed, it was even lower
504 with respect to the wild type strain. These results indicate that the formate transporter is
505 necessary for an increase in hydrogen production in these mutant strains (Tran et al.,
506 2014) and this finding is consistent with those reported previously by Maeda et al.
507 (Maeda et al., 2007a) for the conversion of glucose into hydrogen.

508

509 *4.3 Adaptive evolution*

510

511 Adaptive evolution is a set of environmentally induced mutations that confer growth
512 advantages to the cell (Foster, 2007). In this strategy, an organism is subjected to serial
513 or continuous cultivation for many generations under non-optimal conditions in order to
514 select fitter genetic variants (Cooper et al., 2003). The specific growth rate in glycerol
515 medium of the wild type *E. coli* strain – and even in derivate mutant strains – is quite
516 low compared with that in glucose, which results in low H₂ productivity (Murarka et al.,
517 2008). In an effort to overcome this limitation, Hu & Wood (Hu and Wood, 2010)
518 obtained an efficient glycerol-utilizing strain by both adaptive evolution and chemical
519 mutagenesis using a selection method based on increased growth on glycerol. The
520 improved strain produced twenty times more H₂ in glycerol medium than the wild type
521 strain. It was also found that the activity of several enzymes related to glycerol
522 consumption were increased 4-fold with respect to the wild type strain, and a whole-
523 transcriptome study revealed that several competitive pathways were repressed in that
524 mutant.

525

526 *4.4 Homologous and heterologous expression*

527

528 Overexpression of *E. coli* native genes (homologous expression) or those from other
529 species (heterologous expression) has been used to improve H₂ production in *E. coli*.
530 The heterologous expression offers the advantages of obtaining more efficient enzyme
531 activities based on kinetic parameters (Valle et al., 2017). The main strategies reported
532 in the literature are described below:

533

534 (1) Homologous and heterologous hydrogenase-encoding genes. Several heterologous
535 hydrogenases and hydrogenase-related genes have been expressed in *E. coli*:

- 536 • The expression of HydA from *Clostridium butyricum* improved H₂ productivity
537 by around a factor of two (Subudhi and Lal, 2011).
- 538 • The expression of HydAEFG from *Clostridium* together with the expression of a
539 putative PYR flavodoxin/ferredoxin oxidoreductase (YdbK) from *E. coli*
540 increased H₂ production by 1.5 times (Akhtar and Jones, 2009).
- 541 • The expression of the hydrogenase HoxEFUYH from the cyanobacteria
542 *Synechocystis* in a Hyd-1 and -2 *E. coli* mutant gave rise to a 41-fold higher
543 production (Maeda et al., 2007b).
- 544 • Expression of the NADPH-dependent hydrogenases from *Clostridium sp.* and
545 *Bacillus subtilis* together with the increment of the NAD(P)H/NAD(P)⁺ ratio via
546 overexpression of *glpX* and *zwf* genes. This combination of heterologous
547 expression of NADPH-dependent hydrogenases and autologous gluconeogenesis
548 and PPP enzymes led to an increase in the H₂ yield by 3.5-fold (Kim et al.,
549 2011).

550 (2) Anaplerotic enzymes involved in carbon redirection toward H₂ production. The
551 human mitochondrial GTP-dependent PEP carboxykinase (hPEPCK-M) that catalyzes
552 more efficiently the conversion of OAA to PEP has been used due to its kinetic
553 parameters, the use of an energy donor (GTP) and non-allosteric regulation. That
554 heterologous expression was found to enhance significantly the H₂ yield and glycerol
555 consumption (Valle et al., 2017).

556 (3) Overexpression of transcription factors. The FHL complex is composed of FDH-
557 H and Hyd-3. The biosynthesis of FHL and PFL are up-regulated by the action of
558 several transcriptional regulators, including the global transcriptional factors Fnr,

559 amongst other proteins. The transcription of the *fhl* regulon is controlled by the primary
560 and secondary transcriptional activators FhlA and ModE. The biosynthesis of FDH-H
561 also requires the expression of the *selC* gene, which encodes tRNA for the incorporation
562 of selenocysteine to FHD-H. Therefore, the homologous overexpression of these
563 encoded protein genes, *fhlA*, *pfnr*, *pselC* and *pmodE*, has led to an improved H₂
564 production in a glucose-based medium (Fan, Z. et al., 2009).

565

566 4.5 Protein engineering

567

568 Protein engineering to improve H₂ production by increasing catalytic reactions is a
569 feasible approach. For instance, the *E. coli* Hyd-3 large subunit (HycE) has been
570 modified and this enhanced the FHL complex formation. The method used by Maeda et
571 al. (Maeda, T. et al., 2008) was based on mutagenesis of the HycE protein by an error-
572 prone polymerase chain reaction (epPCR) and employed a host that lacked hydrogenase
573 activity due to *hyaB* *hybC* *hycE* mutations. Seven enhanced HycE variants were
574 obtained and the best contained eight mutations and showed a 17-fold higher H₂-
575 producing activity than wild type HycE. Furthermore, shuffling of the DNA sequences
576 of the three most active HycE variants provided a variant that increased H₂ production
577 23-fold. This was the first engineered hydrogenase obtained by random mutation and it
578 seems to strengthen the interaction between the large and small subunits, thus enhancing
579 the flow of electrons for formate dehydrogenase-H (Maeda, T. et al., 2008). The same
580 method was also used to engineer the σ^{54} transcriptional factor FhlA, which activates
581 transcription of *fdhF*, and the *hyc*, *hyp* and *hydN-hypF* operons. The FhlA133 variant
582 isolated by Sanchez-Torres et al. (Sanchez-Torres et al., 2009) gave a four-fold increase
583 in H₂ production and the whole-transcriptome and promoter reporter analysis revealed
584 that FhlA133 improved H₂ production by increasing transcription of all of the genes

585 activated by FhlA (FHL complex). Surprisingly, *fhlA133* also induced four oxidative-
586 stress proteins and 12-stress related genes during anaerobic fermentation. This finding
587 suggests that an increased FHL expression affects the cell physiology and that an
588 increased expression of anti-stress proteins could probably facilitate H₂ synthesis
589 (Sanchez-Torres et al., 2009).

590

591 *4.6 Optimizing environmental conditions by Design of Experiment (DOE)*

592

593 The environmental conditions may be optimized for a given engendered strain. H₂
594 production has been used as a response variable in order to reduce the number of
595 experiments required to optimize the conditions for maximum yields. For instance, a 3^K
596 full factorial Box–Behnken design and response surface methodology (RSM) have been
597 employed for experimental design. Ghosh et al. (Ghosh and Hallenbeck, 2010) applied
598 this methodology and obtained a maximum molar H₂ yield on using glucose under the
599 optimal conditions at pH 6.5 in the *ΔhyaBΔhybCΔldhAΔfhlC* mutant strain (Rosales-
600 Colunga et al., 2010). Rosales-Colunga et al. (Rosales-Colunga et al., 2010) optimized
601 pH, temperature and substrate concentration of cheese, by using a strain deficient in a
602 protein repressor of Fhl activator factor (HycA) and lactose transporter (LacI). The
603 maximum yield obtained was comparable to the yield achieved in other H₂ production
604 processes with *Clostridium sp.* or mixed cultures (Table 2B). Another study in which
605 this method was used focused on optimizing culture media to improve the growth rate,
606 in order to obtain higher productivities of H₂ and ethanol and for scaling up the process
607 (Cofré et al., 2012).

608 *4.7 Metabolic Flux Analysis (MFA)*

609

610 The way in which anaerobic bacteria regulate flux distribution under different
611 environmental conditions remains to be elucidated. In this regard, the characterization

612 of metabolic fluxes is very important to design metabolic engineering strategies as the
613 flux is a fundamental parameter of cell physiology (Cai et al., 2011). Flux balance
614 analysis (FBA) was used to determine intracellular fluxes from substrate uptake to
615 product formation rates by applying metabolite balancing, which is based on the
616 stoichiometry of metabolic reactions (Varma and Palsson, 1994). The implementation
617 of this approach with proteomic, metabolomics and transcriptomic data led to the
618 concept of metabolic flux analysis (MFA). This approach has been applied to optimize
619 the production of lysine, acetate, and ethanol. However, MFA for H₂ production has
620 only attracted attention in recent years (Cai et al., 2011) and only Manish et al. (Manish
621 et al., 2007) have reported the theoretical capability of the microorganism to produce
622 H₂. They used existing experimental data to calculate fluxes in a batch culture and they
623 commented on the limitations in the yield of H₂. MFA was also used to determine the
624 feasible operating space for H₂ production against the varying yields of other
625 metabolites. This kind of study has helped to show that ethanol and acetate – but not
626 succinate or lactate – are necessary for H₂ production (Manish et al., 2007). It is
627 interesting to note that this deduction was corroborated by experimental results
628 (Murarka et al., 2008; Tran et al., 2014).

629

630 **5. Effect of carbon source and redox balance in H₂ synthesis**

631 Independently of the carbon source used, there are several genes that are involved in
632 fermentative pathways, H₂ synthesis, TCA-reductive and fermentative pathways whose
633 deletions have been widely reported and constitute the main targets for the metabolic
634 redirection in engineered strains. For instance, the deletions of the genes of
635 hydrogenases-1 and -2 (subunits \DeltahyaB and \DeltahybC , respectively), the repressor protein
636 of FhlA (\DeltahycA), the fumarate reductase (\DeltafrdABC), the lactate dehydrogenase (\DeltaldhA),

637 the \square -formate dehydrogenase-N ($\Delta fdnG$), and the phosphate acetyltransferase (Pta)
638 (underlined and bold in the gray square in Fig. 2), have been widely used to increase the
639 H₂ production rate using glucose or glycerol. However, not all of these strategies have
640 been tested using both carbon sources and there are some mutations that have only been
641 assayed with glycerol or glucose. In this sense, there are many mutation strategies, such
642 as deletion of acetate kinase ($\Delta ackA$), pyruvate oxidase (Δpox), pyruvate dehydrogenase
643 ($\Delta aceEF$), formate dehydrogenase-O ($\Delta fdoG$), phosphofructokinase ($\Delta pfkA$), that have
644 been tested in glucose-containing media but not in a glycerol C source that improves H₂
645 synthesis (marked as regular font in the gray square in Fig. 2)

646 On the other hand, some deletion strategies have worked in glycerol but not in glucose
647 media, for instance $\Delta rpiA$, $\Delta gldA$, $\Delta tdcB$, $\Delta tdcE$, $\Delta narG$, $\Delta mgsA$, $\Delta gatZ$. Nevertheless, it
648 has been reported that some mutations do not have the same final effect and even give
649 rise to an opposite result, as is the case of the gluconate dehydrogenase gene (Δgnd),
650 which has a positive effect when glycerol is used but its overexpression enhanced the
651 production of H₂ when glucose was used (marked in gray and also in the red square in
652 Fig. 2).

653 This differences in the engineering strategies on using glycerol or glucose are probably
654 a consequence of the numerous up- and down-regulation metabolic pathways caused by
655 the main regulator protein involved being a transcriptional dual regulator (CRP (cAMP
656 receptor protein) that regulates the expression of over 180 genes. Many of these genes
657 are involved the central carbon pathways and are also regulated by the *cra* protein,
658 which carries out a glycolytic flux-dependent process. This latter process is determined
659 by the concentration of fructose-1,6-bisphosphate synthesized in glucose fermentation,
660 which is the molecular effector of the *cra* gene (Kochanowski et al., 2013). The absence
661 of a rapidly metabolizable carbon source, such as glucose, results in an elevated level of

662 cAMP and subsequently cAMP-CRP increases, which in turn causes catabolite
663 repression. This protein also regulates the catabolism of secondary carbon sources
664 (Soberón-Chávez et al., 2017) and could be related to the enhancement in H₂ synthesis
665 reported for several mutant strains related to amino acid metabolism, antibiotic
666 resistance (Table 2A) or pseudogenes (Table 2B).

667 The strategies outlined above do not always work when either glucose or
668 glycerol are used as the C source. This could be due to the redox balance
669 NAD(P)H/NAD(P)⁺, which must be maintained at certain levels for an efficient
670 synthesis of formate and thus H₂. For instance, the deletion of alcohol dehydrogenase
671 ($\Delta adhE$), because it is a competitive pathway that consumes 2 NADH to produce
672 ethanol, did not improve H₂ production. Furthermore, in the case of malate
673 dehydrogenase (Mdh) and malic enzymes (MaeA, MaeB) that use NAD(P)H the
674 synthesis of H₂ was even impaired (Valle et al., 2015b). The function of these enzymes
675 and the regulation of the hydrogenase operon's gene expression are involved in
676 maintaining the redox and energy balances and are connected to the proton motive force
677 (PMF) in the intermembrane space. On the other hand, there are heterologous
678 NAD(P)H-dependent hydrogenases that have been overexpressed in *E. coli* (Table 2)
679 and the engineered strains required a higher NAD(P)H/NADP⁺ ratio to improve H₂
680 production. This has been achieved by overexpressing enzyme genes involved in
681 NADPH production. The carbon source and redox cofactors are two key factors to take
682 into account in the bacterial physiology for improving H₂ synthesis.

683

684 **6. Conclusions**

685 In this review, the *E. coli* metabolic pathways most commonly modified with the aim of
686 increasing carbon and electron flux towards H₂ biosynthesis are overviewed. The main

687 metabolic engineering strategies used for this purpose are also reviewed. This
688 information may be useful to design new strategies for the scale-up of H₂ production
689 with *E. coli*. The directed mutagenesis of \DeltahyaB , \DeltahybC , and \DeltahycA , which involve
690 hydrogenases-1, -2 and -3, together with \DeltafdnG , \DeltaldhA , \DeltafrdBC , and Δpta , which are
691 involved in competitive pathways of formate synthesis, are the most relevant deletions
692 for metabolic engineering to improve H₂ regardless of the use of glycerol or glucose as
693 carbon source. On the other hand, the overexpression of enzymes involved in glycerol
694 assimilation (GldA), transcriptional factor Hyd-3 operon (FhlA), heterologous enzyme
695 of human PEPCK, or hydrogenases have also been reported as potential strategies
696 together with directed mutagenesis to enhance H₂. There are numerous other strategies
697 (for instance, high-throughput screening, protein engineering, adaptive evolution or
698 random mutagenesis) that have been applied in order to design engineered strains. It is
699 also very useful to implement this information in modeling (MFA) or *in silico* studies in
700 order to predict the optimized pathway to enhance the production of the target product.
701 Although MFA is a suitable method for modeling, there is still a great deal of
702 information that can be obtained from multiomic platforms, which are very valuable
703 tools to design engineered strains that will be used as a microbial cell factory for H₂
704 production.

705

706

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712

713

714 **Declaration of interest**

715 The authors declare that they have no conflict of interest.

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1072 **Table and Figure caption**

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1074 Table 1. Activity (oxidizing or production mode) of hydrogenases 1, 2, 3 and 4 as a
1075 function of carbon source (glycerol or glucose) and pH_{out}. *In lacking Hyd-3 strains, in
1076 Minimal medium Hyd-2 activity is enhanced. **In MM and peptone medium. Hyd-3
1077 activity is increased when Hyd-2 is deleted. ‡Medium with glucose+ formate. †Low
1078 activity

1079

1080 Table 2. In an effort to facilitate the comprehension of the metabolic engineering
1081 strategies, the genotype of engineered mutant strains, the metabolic pathways redirected
1082 in every strain, strategies of metabolic engineering, environmental conditions such as
1083 pH, headspace, and parameters of hydrogen productivity, molar yields and the
1084 respective references, are described in for glycerol (A): *relativized molar yield respect
1085 to wild type strain values (mol H₂/ mol glycerol consumed). **Relativized productivity
1086 respect to wild type strain values (mol H₂/g DCW/h). ****Adding 10 mM formate to
1087 glycerol complex medium. †Complex-formate medium. ‡Minimal medium with glycerol
1088 1%. ¥Peptone glycerol medium 1%. For glucose fermentation (B): *Relativized molar
1089 yield respect to wild type strain values (mol H₂/ mol glucose consumed). **Relativized
1090 productivity respect to wild type strain values (mol H₂/g DCW/h //// mol H₂/mg protein
1091 /// mol H₂/ g DCW). *Cheese whey (20 g/L). **Lactose (5 g/L). ***Complex-formate
1092 medium. ****In vitro assay. *****Complex medium (contain fructose, galactose,
1093 maltose, lactose, glycerol citrate or succinate).

1094

1095 Figure 1. Curves representing the number of documents published from 1960 to 2017
1096 using different refined terms in the SCOPUS database: "hydrogen and *Escherichia coli*
1097 and engineering" (orange triangle); "hydrogen and *Escherichia coli*" (gray circles) and
1098 "hydrogen" (blue diamonds).

1099

1100 Figure 2. General metabolic pathways under anaerobic conditions, the different routes
1101 are represented by different colors. PPP (red), glycolysis (black), anaerobic respiration
1102 (TCA-cycle) (green); glycerol assimilation (brown); Entner–Doudoroff (orange),
1103 carbohydrates by using glycerol (*genes* in bold font) and for glucose fermentation

1104 (*genes* in regular font). The deleted genes are marked with a gray square and genes
1105 overexpressed are marked with a red square. Genes reported in both glycerol based
1106 medium and glucose based medium are underlined.

1107

1108

1109

Figure 1

SCOPUS

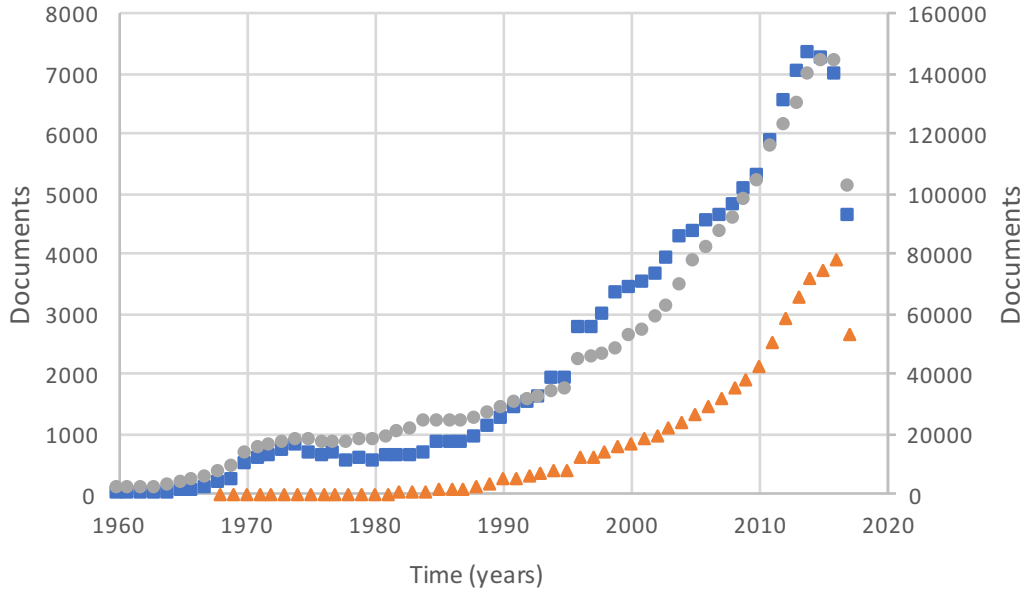


Table 1

		pH _{out}		
		5.5	6.5	7.5
Glycerol		Oxidizing mode ($H_2 \rightarrow 2H^+ + 2e^-$)		
		Hyd-1, Hyd-2 [†] , Hyd-4	Hyd-4	Hyd-3, Hyd-4
		Production mode ($H^+ + 2e^- \rightarrow H_2$)		
		Hyd-3 ^{**}	Hyd-3 ^{**} >Hyd-2, Hyd-1	Hyd-2 [*] > Hyd-1
Glucose		Oxidizing mode ($H_2 \rightarrow 2H^+ + 2e^-$)		
		Hyd-1, Hyd-2,	Hyd-3	Hyd-2 > Hyd-1
		Production mode ($H^+ + 2e^- \rightarrow H_2$)		
		Hyd-3, Hyd-4	Hyd-2, Hyd-1 [‡] , Hyd-4	Hyd-4, Hyd-3 [‡]

Table 2A. Carbon source: Glycerol

Strains, genotype (ID)	Phenotype	Metabolic pathways redirected	Metabolic Engineering strategies	pH _{out}	Headspace (partial pressure)	Yield*	Productivity**	References
BW25113 $\Delta hyaB \Delta hybC \Delta hycE$ /pBS(Kan)HycE variant satHycE12T366	Defective in Hyd-1, Hyd-2, Hyd-3 and overexpression of Hyd-3 catalytic subunit HycE with protein variation	Hydrogen synthesis by FHL-1 complex	Protein engineering	7	High	30	-	Maeda 2008 [†]
BW25113 $\Delta hyaB \Delta hybC \Delta hycE$ /pBS(Kan)HycE variant shufHycE1-9	Defective in Hyd-1, Hyd-2, Hyd-3 and overexpression of Hyd-3 catalytic subunit HycE with protein variation	Hydrogen synthesis by FHL-1 complex	Protein engineering	7	High	23	9	
MG1655 $\Delta frdA::FRT \Delta pta::FRT$ -tet-FRT (SY03)	Defective of FRQ subunit FrdA and phosphotransacetylase (PtA)	TCA- reductive branch Fermentative pathway	Mutagenesis of competitive pathways	6.3	Low	1.23	1.1	Yazdani 2008 [‡]
MG1655/pZSKLMgldA	Overexpression of dihydroxyacetone kinase (DhaKLM) and Glycerol dehydrogenase (GldA)	Glycerol assimilation	Homologous expression	6.3	Low	1.10	1.5	
BW25113 $\Delta frdC \Omega kan^R$ /pCA24N-gldA (HW2)	Defective of FRQ subunit FrdC and overexpression of GldA	Glycerol assimilation and TCA	Mutagenesis of competitive pathways Adaptive evolution Chemical mutagenesis Homologous expression	6.3	High	-	20	Hu H. 2010 [‡]
BW25113 $\Delta hyaB \Omega kan^R$	Defective of Hyd-1 large subunit (HyaB)	Reverse H ₂ oxidizing mode	Mutagenesis competitive pathways	6.5	High	-	1.76	Trchounian 2011 [‡]
BW25113 $\Delta hybC \Omega kan^R$	Defective of Hyd-2 large subunit (HybC)	Reverse H ₂ oxidizing mode	Mutagenesis competitive pathways	6.5	High	-	1.5	
<i>E. coli</i> B SS1		-	High-throughput screening from isolated strains	7		-	1.67	Suhaimi, 2012 [‡]
BW25113 $\Delta hyfG \Delta fhIA \Omega kan^R$	Defective in Hyd-4 large subunit (HyfG) and protein regulator of <i>hyc</i> , <i>hyb</i> and <i>hya</i> operons expression (FhIA)	Reverse H ₂ oxidizing mode	Mutagenesis competitive pathways	6.5	High	-	1.7	Trchounian 2012 [‡]
BW25113 $\Delta hyaB \Omega kan^R$	Defective in Hyd 1 large subunit (HyaB)	Reverse H ₂ oxidizing mode	Mutagenesis competitive pathways	7.5	High	-	1.34	Sanchesz Torres 2013 [‡]
BW25113 $\Delta frdC \Delta ldhA \Delta fdnG \Delta ppc \Delta narG \Delta mgsA \Omega kan^R$	Defective in FrdC, LdhA, FdnG, Ppc, NarG, methylglyoxal synthase (MgsA).	Fermentative pathway TCA reductive Branch Hydrogenase synthesis	Mutagenesis competitive pathways. Screening mutant strains Homologous expression	6.5	High	-	4.3	Tran 2014 [‡]
BW25113 $\Delta frdC \Delta ldhA \Delta fdnG \Delta ppc \Delta narG \Delta mgsA \Delta hycA \Omega kan^R$	Defective in FrdC, LdhA, FdnG, Ppc, NarG, MgsA and repressor of the FHL complex (HycA).					1.3	1.68	

Strains, genotype (ID)	Phenotype	Metabolic pathways redirected	Metabolic Engineering strategies	pH _{out}	Headspace (partial pressure)	Yield*	Productivity**	References
BW25113 Δ frdC Δ ldhA Δ fdnG Δ ppc Δ narG Δ mgsA Δ hycA Ω kan ^R /pCA24N-FdhF	Defective in FrdC, LdhA, FdnG, Ppc, NarG, MgsA and HycA. Overexpression of formate dehydrogenase large subunit (FdhF)	Fermentative pathway TCA reductive Branch Hydrogenase synthesis	Mutagenesis competitive pathways. Screening mutant strains Homologous expression	6.5	High	-	3.4	Tran 2014 [‡]
BW25113 Δ frdC Δ ldhA Δ fdnG Δ ppc Δ narG Δ mgsA Δ hycA Ω kan ^R /pCA24N-FhIA	Defective in FrdC, LdhA, FdnG, Ppc, NarG, MgsA and HycA. Overexpression of FhIA					-	3.1	
BW25113 Δ frdC Δ ldhA Δ fdnG Δ ppc Δ narG Δ mgsA Δ hycA Ω kan ^R	Defective in FrdC, LdhA, FdnG, Ppc, NarG, MgsA and HycA				Low	3.68	3.14	
BW25113 Δ focA Δ focB Ω kan ^R	Defective of formate transporter FocA and FocB	Hydrogenase synthesis	Mutagenesis competitive pathways	5.5	High	-	11	Trchounian, 2014
				6.5			3.2***	
BW25113 Δ aroM Ω kan ^R	Defective in shikimate kinase II	Aminoacids	Screening mutant strains Random mutagenesis	7.5	High	-	1.7	Tran 2015 [‡]
BW25113 Δ gatZ Ω kan ^R	Defective in tagatose-1,6-bisphosphate aldolase 2	Sugars					1.4	
BW25113 Δ ycgR Ω kan ^R	Defective in flagellar velocity braking protein, c-di-GMP-regulated	Motility					2.4	
BW25113 Δ yfgI Ω kan ^R	Defective in nalidixic acid resistance protein	Antibiotic resistance					2.1	
BW25113 Δ ldhA Δ gnd Δ frdBC Δ tdcE Ω kan ^R	Defective in LdhA, gluconate dehydrogenase (Gnd), FrdBC subunits and tdcE	Fermentative pathway TCA reductive Branch	Mutagenesis competitive pathways. Screening mutant strains	7.5	High	1.33	-	Valle 2015 [¥]
BW25113 Δ dcuD Ω kan ^R /pT-PEPCK	Defective in putative C4-dicarboxylate transporter (DcuD) and expression of human PEPCK mitochondrial	Putative C-4 transporter	Mutagenesis of competitive pathway Heterologous expression	6.3	Low	-	2.46	Valle 2017 [¥]
<i>E. coli</i> SS1/hycE	Overexpression of Hydrogenase 3-large subunit hycE (<i>E. coli</i>)	Synthesis of hydrogen	Screening mutant strains Homologous expression	5.8	High	1.41	<1.0	Soo 2017 [¥]
<i>E. coli</i> SS1/hydA	Overexpression of Hydrogenase from <i>Clostridium</i> (HydA)	Synthesis of hydrogen	Screening mutant strains Heterologous expression			1.26		

Table 2B. Carbon source: Glucose

Strains, genotype (ID)	Phenotype	Metabolic pathways redirected	Metabolic Engineering strategies	pH _{out}	Headspace (partial pressure)	Yield*	Productivity**	References
BW25113 $\DeltahyaB\Delta hybC\Delta hycA\Delta fdoG$ kan ^R /pCA24N-FhIA	Defective in Hyd-1 and Hyd-2, HycA and formate dehydrogenase-O (FdoG)	Hydrogenase synthesis	Mutagenesis of competitive pathway Homologous expression	6.8	High	1.5	26.3***	Maeda 2008
					Low		141***	
BW25113 $\DeltahyaB\Delta hybC\Delta hycA\Delta frdC\Delta ldhA\Delta aceE\Delta fdnG$ Ωkan ^R	Defective in Hyd-1 and Hyd-2, HycA, FrdC, LdhA, pyruvate dehydrogenase (AceE) and formate dehydrogenase-N (FdnG)	Hydrogenase synthesis fermentation, pyruvate metabolism and anaerobic respiration	Mutagenesis of competitive pathway Homologous expression	6.8	Low	2	4.3	Maeda 2007
BW25113 $\DeltahyaB\Delta hybC\Delta hycA\Delta fdoG\Delta frdC\Delta ldhA\Delta poxB$ Ωkan ^R	Defective in Hyd-1 and Hyd-2, HycA, FodG, FrdC, LdhA, and pyruvate oxidase (PoxB)					2	4.6	
BW25113 $\DeltahyaB\Delta hybC\Delta hycA\Delta fdoG\Delta frdC\Delta ldhA$ Ωkan ^R /pCA24N-FhIA	Defective in Hyd-1 and Hyd-2, HycA, FodG, FrdC, LdhA, and overexpression of FhIA					2.5	4.4	
W3110 $\Delta focA::FRT/pfnr$ (ZF13)	Defective in Formate transporter A and overexpression of global transcriptional regulator (Fnr) of anaerobic metabolism; activator of <i>fhIA</i> and <i>pflB</i> genes, and of the <i>hyp</i> , <i>nik</i> (Ni ⁺² ABC transporter) and <i>moa</i> (synthesis of Mb cofactor) operon genes, in pACYC177 vector with LacZ promoter and W3110 <i>fnr</i> gene.	Hydrogen synthesis, pyruvate metabolism, formate transporter	Mutagenesis of competitive pathway	7.6	High	-	9.65	Fan, 2009
W3110 $\Delta narL::FRT$ (ZF3)	Defective in global transcriptional regulator of anaerobic metabolism; represses the transcription of <i>nik</i> operon genes, <i>fdhF</i> and <i>pflB</i> genes; activates the transcription of formate dehydrogenase N, which couples formate oxidation to nitrate reduction.					1.77	-	
BW25113 $\Delta ydfW$ Ωkan ^R BW25113 $\Delta yclE$ Ωkan ^R BW25113 $\Delta ypdJ$ Ωkan ^R BW25113 $\Delta yqiG$ Ωkan ^R	Defective in pseudogenes	Involved as leader peptides, ribosomal proteins and toxic proteins	High-throughput screening	7	High	-	-	Mohd Yusoff 2013
BW25113 $\DeltahyaB\Delta hybC$ Ωkan ^R	Defective in Hyd-1 and Hyd-2	Reverse H ₂ oxidizing mode	Mutagenesis of competitive pathway	5.5	High	-	1.43	Trouchnian 2011
				6.5			2	
				7.5			1.4	

								2013
Strains, genotype (ID)	Phenotype	Metabolic pathways redirected	Metabolic Engineering strategies	pH _{out}	Headspace (partial pressure)	Yield*	Productivity**	References
DJT135 $\Delta hyaB\Delta hybC\Omega kan^R \Delta ldhA$, $\Delta fhlA\Omega Cm^R$	Defective in Hyd-1 and Hyd-2, in LdhA and FhlA	Reverse H ₂ oxidizing mode, fermentative pathways and synthesis of hydrogen.	Optimizing environmental conditions by Design of experiment (DOE)	7.5	Low	1.69	-	Ghosh, 2010
W3110 $\Delta hycA\Delta lac\Omega kan^R$	Defective in HycA and gene lac operon repressor (LacI)	Hydrogenase synthesis and lactose fermentation	Optimizing environmental conditions by Design of experiment (DOE)	7.5	Low	2.26**	1.78*	Rosales-Colunga 2010
BW25113 $\Delta hycA \Delta hyaAB \Delta hybBC \Delta ldhA \Delta frdAB \Omega kan^R$ (SH5)	Defective in HycA, Hyd-1, Hyd-2, LdhA, FrdAB	Synthesis of hydrogen, fermentative pathways and TCA-reductive branch	Mutagenesis of competitive pathway MFA	7.0	High	1.42	-	Seol, 2014
BW25113 $\Delta hycA \Delta hyaAB \Delta hybBC \Delta ldhA \Delta frdAB \Delta pfkA \Delta pta-ackA \Omega kan^R$ -adapted evolution- <i>pKD7-zwf-gnd</i> (SH8)	Defective in HycA, Hyd-1, Hyd-2, LdhA, FrdAB, phosphofructokinase (PfkA), Pta, acetate kinase (AckA) and overexpression of NADP ⁺ -dependent glucose-6-phosphate dehydrogenase (Zwf) and gluconate dehydrogenase (Gnd)	Pentose phosphate pathway (PPP) Embden-Meyerhof-Parnas (EMP) Synthesis of hydrogen, fermentative pathways and TCA-reductive branch.	Mutagenesis of competitive pathway Adaptive evolution MFA	7.0	High	1.32	-	Seol, 2016
				7.0	High	1.32	-	
$\Delta ldhA$	Defective in LdhA	Lactose fermentation	Metabolic flux analysis	-	-	1.35		Manish 2007
W3110 $\Delta hycA \Delta ldhA::Cm^I \Delta frdBC::Km^I$ / FhlA (SR14)	Defective in LdhA and FrdB and FrdC subunits.	Lactose fermentation, anaerobic respiration and hydrogen synthesis	Mutagenesis of competitive pathway	6.2	High	1.73	-	Yoshida, 2006
BL-21 (DE3)/pGEX-5X-hydA	pGEX-5X-3 vector carrying <i>Clostridium butyricum</i> hydA gene	Hydrogen synthesis	Heterologous expression	7.1	High	1.68	1.9	Subudhi, 2011
BL21(DE3) $\Delta iscR$ /pAF-pYdbK HydA, HydF, HydG, HydE and CpFdx	Defective in the transcriptional regulator of the <i>isc</i> operon involved in the assembly of Fe-S clusters into proteins.	Hydrogen synthesis	Mutagenesis of competitive pathway Heterologous expression	7.1	High	1.46	-	Akhtar, 2009****
BW25113 $\Delta hyaB \Delta hybC$ /pBS(Kan)Synhox	Defective in Hyd-1 and Hyd-2 large subunits and heterologous expression of hydrogenase (HoxEFUYH) from the cyanobacterium <i>Synechocystis</i> sp. PCC 6803	Hydrogen synthesis	Mutagenesis of competitive pathway Heterologous expression	6.8	High	41	-	Maeda 2007*****
BL21 (DE3)/ pETDuet-HydA-HydE (HFdY) and pCDFDuet-HydF-HydG and pRSFDuet-1-FdxA-YumC and pACYCD-Zwf-GlpX (Strain HFdYzg)	Heterologous expression of hydrogenases (HydAEFG) from <i>C. acetobutyricum</i> FdxA from <i>C. pasteurianum</i> ; YumC from <i>Bacillus subtilis</i> and Zwf from <i>E. coli</i> BL21 GlpX from <i>E. coli</i> BL21	Hydrogen synthesis	Heterologous expression	7.0	High	3.5	-	Kim, 2011

