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1 **Effect of PEGylation on the biological properties of cationic carbosilane**
2 **dendronized gold nanoparticles**

3
4 Andrea Barrios-Gumiel,^{1,2,3} Javier Sánchez-Nieves,^{1,2,3} Elzbieta Pedziwiatr-
5 Werbicka,^{4,*} Viktor Abashkin,⁵ Natallia Shcharbina,⁶ Dzmitry Shcharbin,⁵ Sława
6 Glińska,⁷ Karol Ciepluch,⁸ Dorota Kuc-Ciepluch,⁸ Dominika Lach,⁴ Maria
7 Bryszewska,⁴ Rafael Gómez,^{1,2,3,*} F. Javier de la Mata^{1,2,3,*}

8
9 ¹ Dpto. de Química Orgánica y Química Inorgánica. Universidad de Alcalá (UAH).
10 Campus Universitario. E-28871 Alcalá de Henares (Madrid) Spain. Instituto de
11 Investigación Química "Andrés M. del Río" (IQAR). Universidad de Alcalá (UAH).

12 ² Networking Research Center for Bioengineering. Biomaterials and Nanomedicine
13 (CIBER-BBN), Spain.

14 ³ Instituto Ramón y Cajal de Investigación Sanitaria, IRYCIS, Spain.

15 ⁴ University of Lodz, Faculty of Biology and Environmental Protection, Department
16 of General Biophysics, 141/143 Pomorska Street, 90-236 Lodz, Poland;
17 elzbieta.pedziwiatr@biol.uni.lodz.pl.

18 ⁵ Institute of Biophysics and Cell Engineering of NASB, Minsk, Belarus;
19 shcharbin@gmail.com.

20 ⁶ Clinical Unit "Eleous" at Religious community «All Saints Parish in Minsk
21 Eparchy of Belarusian Orthodox Church», Minsk, Belarus; nata.shcharbina@mail.ru.

22 ⁷ University of Lodz, Faculty of Biology and Environmental Protection,
23 Laboratory of Microscopic Imaging and Specialized Biological Techniques,
24 Banacha 12/16, 90-237 Lodz, Poland; slawa.glinska@biol.uni.lodz.pl.

25 ⁸ Department of Biochemistry and Genetics, Jan Kochanowski University,
26 Świętokrzyska Street 15, 25-406 Kielce, Poland.

27

28 **ABSTRACT**

29 Heterofunctionalized gold nanoparticles (AuNP) were obtained in a one pot reaction
30 of gold precursor with cationic carbosilane dendrons (first to third generations, 1-3G)
31 and (polyethylene)glycol (PEG) ligands in the presence of a reducing agent. The final
32 dendron/PEG proportion on AuNPs depends on the initial dendron/PEG ratio (3/1, 1/1,
33 1/3) and dendron generation. AuNPs were characterized by transmission electron
34 microscopy (TEM), dynamic light scattering (DLS), ultraviolet spectroscopy (UV-VIS),
35 thermogravimetric analysis (TGA), nuclear magnetic resonance (¹H-NMR) and zeta
36 potential (ZP). Several assays have been carried out to determine the relevance of
37 PEG/dendron ratio and dendron generation in the biomedical properties of PEGylated
38 AuNPs and the results have been compared with those obtained for non-PEGylated
39 AuNPs. Finally, analyses of PEG recognition by anti-PEG antibodies were carried out.
40 In general, haemolysis, platelet aggregation and toxicity were reduced after PEGylation
41 of AuNPs, the effect being dependent on dendron generation and dendron/PEG ratio.
42 Dendron generation determines the exposure of PEG ligand and the interaction of this
43 ligand with AuNPs environment. On the other hand, increasing PEG proportion
44 diminishes toxicity but also favors interaction with antibodies.

45 **Keywords:** gold nanoparticles, carbosilane dendron, PEG, toxicity, haemolysis,
46 platelets aggregation, PEG-antibody.

47

48 INTRODUCTION

49 The design of carriers for biomedicine is an outstanding tool that pretends finding
50 versatile systems to deliver biomacromolecules or drugs.¹⁻⁵ The goals of carries are to
51 protect bioactive systems from degradation, to increase circulation time and solubility
52 and, if possible, to guide biomacromolecules or drugs to a specific target. The
53 transporting of these systems can be done through different approaches, being one of
54 them formation of electrostatic complexes. This option is widely used for nucleic acid
55 delivery, since its polyanionic nature favors formation of complexes with polycationic
56 systems.^{6, 7} Also, this approach can be used for proteins or peptides, which contain
57 charge domains of different nature, and drugs.⁷⁻¹⁴

58 One strategy to design carriers makes use of the enhance stability achieved with
59 polyfunctional compounds, due to establishment of multiple interactions that mimics the
60 behavior observed in biomacromolecules.¹⁵⁻¹⁷ To this end, several type of artificial
61 polyfunctional macromolecules have been developed: metal and metal oxide
62 nanoparticles (NP), soft nanoparticles, polymers, dendrimers, liposomes, etc.^{1, 7, 18, 19}

63 For example, the unique properties of metal NP, such as small size, high surface
64 area, ease of functionalization, associated to properties of gold, such as
65 biocompatibility, low toxicity and high stability, make gold nanoparticles (AuNP) ideal
66 for several biomedical applications, including as carriers.²⁰⁻²⁵

67 Other type of systems deeply studied as carriers of different active
68 biomacromolecules are dendritic systems (dendrimers and dendrons).^{18, 26, 27} These
69 multivalent molecules are built following iterative reactions leading to well-defined
70 hyperbranched structures. This fact allows for establishing structure/activity
71 relationships easily, in comparison with traditional polymers. Several types of
72 dendrimers have been developed, e.g. poly(amidoamine) (PAMAM), polypropilenimine

73 (PPI), phosphorus dendrimers, carbosilane dendrimers. They differ mainly on the
74 heteroatoms that compose their framework, which have huge influence on their
75 hydrophilic/hydrophobic character. In the particular case of carbosilane dendrimers,
76 their inner structure is clearly hydrophobic. However, if their surface is covered with
77 cationic groups they have shown good behavior as carriers,^{28, 29} being even able to cross
78 the blood brain barrier.^{30, 31}

79 Additionally, dendrimers and dendrons can stabilize AuNPs through terminal groups
80 of dendrimers or by anchoring dendrons on the surface of the NPs through the focal
81 point.³²⁻³⁵ If these dendronized NPs are decorated with the corresponding cationic
82 groups, they can also serve as carriers.³⁶⁻³⁸

83 Carriers injected in blood current will interact with a diversity of cells (e. g. red
84 blood cells, lymphocytes) and distributed through the body.³⁹ Therefore, the analyses of
85 their biological properties are priority.⁴⁰⁻⁴² It is important to note that the cationic
86 charges, which are responsible of stabilizing the complexes between cationic
87 macromolecules and nucleic acids (nanoplexes), are also responsible of the toxicity of
88 these systems.⁴³ Hence, several strategies to reduce toxicity have been developed:
89 synthesis of degradable systems, hiding of cationic charge, introducing biocompatible
90 ligands as polyethylenglycol (PEG), etc.^{13, 44}

91 In the particular case of PEG, it is widely used to camouflage carriers from non-
92 desired interactions in a biological environment, favoring also solubility and
93 biocompatibility (process known as PEGylation).⁴⁵⁻⁴⁸ The hydrophilic nature of PEG
94 favors formation of a hydration sphere that surrounds the nanocarriers. This steric shield
95 protects the charge surface from interaction with blood components and other carriers,
96 increasing also circulation time.^{49, 50} Particularly, PEGylation of NPs and dendrimers
97 reduces hemolysis⁵¹⁻⁵³ and interaction with several blood proteins.^{54, 55} Unfortunately,

98 this widespread strategy has also its own problems such as making difficult cellular
99 absorption or production of antibodies by the immune system that specifically recognize
100 PEG. These difficulties lead to loss of efficacy and appearance of adverse effects as
101 allergy reactions.⁵⁶⁻⁶⁰

102 Previously, we have described AuNPs covered with cationic dendrons that were
103 tested for gene delivery. However, their toxicity prevented the use of higher generation
104 systems and the transfection efficiency was low.³⁷ With the aim to improve their
105 biological properties, herein we report the synthesis of AuNPs covered with cationic
106 carbosilane dendrons of three different generations and PEG ligands. Several
107 dendron/PEG ratios have been employed with the aim to determine the influence of
108 dendron generation and dendron/PEG ratios on the biological properties of these
109 PEGylated AuNPs. These systems have been fully characterized through different
110 techniques. Biological assays to test their potential as vehicles for delivery of nucleic
111 acids or drugs have been carried out: haemolysis, toxicity, platelet aggregation, cellular
112 uptake, recognition by PEG antibodies.

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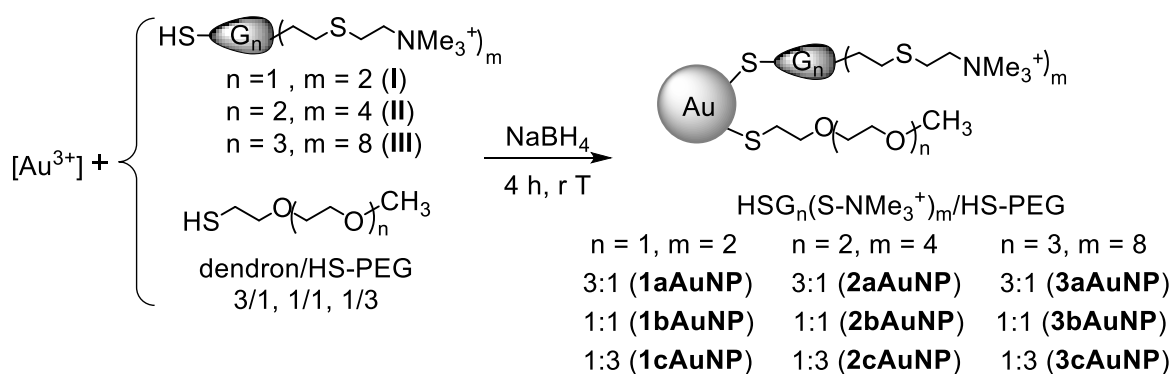
114 **RESULTS AND DISCUSSION**

115 **Synthesis and characterization of gold nanoparticles (AuNPs)**

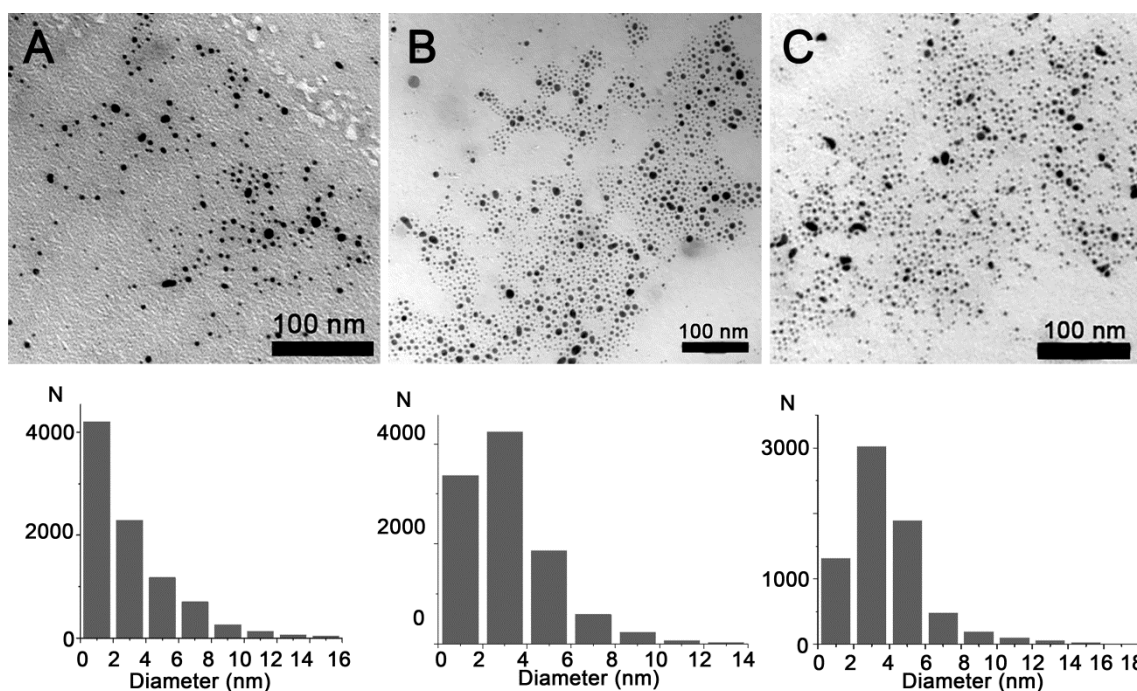
116 For simplicity, the following nomenclature will be used to name dendrons described
117 in this work: XG_nY_m , where 'X' represents the focal point, 'G_n', carbosilane dendron
118 generation, 'Y' denotes nature of peripheral groups and "m" the number of these
119 groups.

120 For preparation of bifunctionalized AuNPs, we have employed two ligands
121 containing a thiol moiety: the cationic dendrons $HSG_n(S-NMe_3^+)_m$ ($n = 1, m = 2$ (**I**); $n =$
122 $2, m = 4$ (**II**); $n = 3, m = 8$ (**III**), see Figure S1)³⁷ and a commercial

123 (polyethylene)glycol (PEG) ligand $(\text{CH}_3\text{O}(\text{CH}_2\text{CH}_2\text{O})_n\text{CH}_2\text{CH}_2\text{SH}$, HS-PEG, $M_n =$
 124 800). These new AuNPs were synthesized in water by the direct reaction of
 125 $\text{HAuCl}_4 \cdot 3\text{H}_2\text{O}$ with a mixture of both thiol derivatives and NaBH_4 as reducing agent.
 126 To evaluate the influence of PEG ligand on physical and biomedical properties of these
 127 new AuNPs, three different dendron/PEG ratios (3/1, 1/1, 1/3) were used (Scheme 1).
 128 The AuNPs thus obtained are named as $N_m\text{AuNP}$; where 'N' represents the carbosilane
 129 dendron generation (1-3) and 'm' represents the dendron/PEG ratio ($a = 3/1$, $b = 1/1$, $c =$
 130 $1/3$). Related homofunctionalized AuNPs follow the same nomenclature without letter
 131 'm': NAuNP . The heterofunctionalized AuNPs (**1a-cAuNP**, **2a-cAuNP**, **3a-cAuNP**)
 132 were produced in high yield as black solids soluble in water. This methodology is a
 133 straightforward procedure to prepare metal NPs with different ligands, reducing the
 134 amount of ligands with respect to the indirect or substitution procedure.³⁵
 135 Heterofunctionalized AuNPs **1-3,a-cAuNPs** were characterized by transmission
 136 electron microscopy (TEM), dynamic light scattering (DLS), ultraviolet spectroscopy
 137 (UV-VIS), thermogravimetric analysis (TGA), nuclear magnetic resonance ($^1\text{H-NMR}$)
 138 and zeta potential (ZP) (Table 1, Table S1).



140 **Scheme 1.** Synthesis of heterofunctionalized AuNPs ($n = 1, m = 2$ (**1a-cAuNP**); $n =$
 141 $2, m = 4$ (**2a-cAuNP**); $n = 3, m = 8$ (**3a-cAuNP**). Figure S1 depicts full structure of
 142 dendrons **I-III**.³⁷



143
144 **Figure 1.** TEM image and size distribution histogram associated to a) **1aAuNP**, b)
145 **1bAuNP** and c) **1cAuNP**.

146 Figure 1 shows TEM images and size distribution histograms of first generation
147 heterofunctionalized AuNPs **1a-cAuNP**. The corresponding images for higher
148 generation AuNPs are collected in Supporting Information (Figures S2-S7). The size of
149 these AuNPs (TEM) was independent of dendron generation or dendron/PEG ratio,
150 presenting diameters from 2 to 5 nm. These systems were stable under inert atmosphere,
151 maintaining size and shape for over 18 months (Figure S8). It should be noted that
152 stability of PEGylated AuNPs (**1-3,a-cAuNP**) was higher than non-PEGylated AuNPs
153 (**1-3AuNP**), according with the stabilizing capacity of this kind of polymer.⁶¹⁻⁶³

154 The hydrodynamic diameters of AuNPs in water solution were measured by DLS
155 (Figures S9-S17). The higher diameters obtained by DLS, compared with those
156 obtained by TEM, are related with the differences between both techniques. Whereas
157 TEM only measures the metallic core, the size obtained by DLS would include ligands
158 attached to the AuNPs and solvent layer at the interface. The polydispersity (PDI) and
159 the diameter (d_z) obtained by DLS can be used to calculate a theoretical d_n value (Cd_n

160 = $d_z/(1+Q)^5$; where Q correspond with PDI).^{37, 64, 65} These calculated diameters are
161 closer to those measured by TEM. UV spectra also confirmed AuNPs formation,
162 observing for all nanoparticles the band belonging to the surface plasmon resonance at
163 about 520 nm (Figures S18-S20).

164 Functionalization of AuNPs was detected by TGA and ¹H-NMR spectroscopy. TGA
165 gives information about the percentage of organic matter and inorganic matter in the
166 AuNPs (Table 1, Table S1, Figures S21-29). In ¹H-NMR spectroscopy, the resonances
167 of the different ligands on AuNP surface (**1-3,a-cAuNP**) were observed at similar
168 chemical shifts than those of the starting dendrons (**I-III**)³⁷ and HS-PEG ligand (Figures
169 S30-S38). The methylene groups closest to the gold surface CH₂SAu, corresponding
170 with the focal point of dendrons and PEG derivative, were not observed due to their
171 proximity to the gold surface. ¹H-NMR spectra also allowed calculation of final
172 dendron/PEG molar ratio of heterofunctionalized AuNPs, using integrals of external
173 methyl groups of trimethylammonium moieties of dendrons and methoxy group of PEG
174 ligand (Table 1, Table S1). This final dendron/PEG ratio is clear dependent on the
175 initial one and dendron generation, although PEG proportion on heterofunctionalized
176 AuNPs **1-3,a-cAuNP** is always higher than that expected. This result can be attributed
177 to the necessity to reduce charge repulsion between cationic dendrons.

178 It is important to note that introduction of PEG ligands entails a reduction of the
179 density in the number of functions on NP surface (ammonium groups of dendrons and
180 PEG ligands, Tables 1 and S1) with respect to non-PEGylated AuNP, except for **1a-**
181 **cAuNP**, which is covered with 1G dendrons. This fact can be related with relative sizes
182 of dendrons and PEG, but also with conformation of both ligands. While dendrons tend
183 to separate as much as possible to avoid charge repulsion, PEG can fold its structure

184 covering higher nanoparticle surface. This assessment is supported by the low density
185 value of PEG ligands on AuNP covered only with this type of ligand.

186 Comparison of Zeta potential values of heterofunctionalized AuNP **1-3,a-cAuNP**
187 with cationic homofunctionalized AuNP **1-3AuNP** showed an important reduction of
188 this value. Probably, this is consequence of the presence of lesser cationic dendrons on
189 AuNPs surface and because PEG ligand can mask dendron charge.

AuNPs	Molar ratio Dendron/PEG		%L ^b	d _n ^c	d _z ^d	PDI ^e	Cd _n ^f	P (F/nm ²) ^g	ZP ^h
	Theoretical	Obtained ^a							
1AuNP	-	-	45.2	1.8	10.3	0.391	2.0	11.6	50.0
1aAuNP	3/1	1/2.0	51.7	3.2	39.8	0.409	7.2	12.6	29.1
1bAuNP	1/1	1/2.9	59.9	3.2	24.5	0.471	3.6	16.1	23.8
1cAuNP	1/3	1/5.4	55.3	3.9	21.5	0.570	2.3	14.4	23.6.
2AuNP	-	-	67.7	2.2	15.8	0.444	2.5	71.8	63.7
2aAuNP	3/1	1/1.5	67.6	3.7	34.0	0.540	3.9	36.8	44.9
2bAuNP	1/1	1.1/1	66.6	2.8	23.3	0.298	6.3	30.2	41.1
2cAuNP	1/3	1/6.4	62.3	4.8	94.3	0.233	4.4	26.0	34.9
3AuNP	-	-	84.2	2.0	22.1	0.611	2.0	159.8	59.6
3aAuNP	3/1	1.5/1	77.4	3.9	26.0	0.473	3.8	84.8	48.0
3bAuNP	1/1	1/2.7	75.6	5.0	23.9	0.400	4.4	75.3	54.8
3cAuNP	1/3	1/6.0	68.0	3.3	35.5	0.381	7.1	27.5	46.0
PEGAuNP	-	-	64.6	3.1	256.1	0.475	36.8	13.7	21.5

Table 1. Selected data of AuNPs. a) Molar ratio obtained by NMR (integration of resonances of methyl groups of trimethylammonium moieties of dendrons and methoxy group of PEG ligand). b) % Organic matter obtained by TGA, corresponding with dendron and PEG coating. c) Diameter (d_n, nm) obtained by TEM. d) Diameter (d_z, nm) obtained by DLS. e) Polydispersity index (PDI) obtained by DLS. f) Diameter calculated (Cd_n, nm): Cd_n = d_z/(1+Q)⁵; Q corresponds with PDI.⁶⁵ g) Number of functions (F) per nm² on AuNP surface (dendrons: 2 F for 1G, 4 F for 2G, 8 F for 3G; PEG: 1 F), calculated from TEM diameter, NMR and TGA (see Table S1). h) Zeta potencial (mV). **1-3AuNP** were previously described.³⁷

Biological assays

Haemolysis and Cytotoxicity. The haemolysis data showed (Figure 2 and S39) that erythrocytes were very tolerant toward AuNPs covered with PEG and 1G (**1a-cAuNP**) and 2G (**2a-cAuNP**) dendrons in all the concentrations measured at 2 h and 24 h. Particularly, PEGylated NPs **2a-cAuNP** produced haemolysis below 5% even at 100 µg/ml, whilst non-PEGylated NPs **2AuNP** reached 70% haemolysis at 30 µg/ml. For **3a-cAuNP**, with the biggest 3G dendron, haemolysis over 20% was surpassed for AuNPs with lower PEG proportions (**3a-bAuNP**) at concentrations above 50 µg/ml after 24 h but not at 2 h. On the other hand, for **3cAuNP**, with the highest PEG proportion, haemolysis was kept below 20% at all concentrations studied. These data highlight the relevance of PEGylation, since non-PEGylated AuNPs covered with 2G cationic dendrons **2AuNP** were as hemolytic as **3AuNP**.³⁷

As eukaryotic model to study viability of non-PEGylated **1-3AuNP** and PEGylated **1-3,a-cAuNP**, HeLa cells were chosen (Figure 3). The general trend observed was an increase of viability for PEGylated AuNPs with respect to non-PEGylated AuNPs. This is mainly observed for AuNPs covered with 2G and 3G dendrons, since viability of AuNPs stabilized with 1G dendrons was similar for the non-PEGylated and PEGylated systems. Also, viability of AuNPs with higher generation dendrons improved with increasing amount of PEG ligands. Of all the systems studied, the data point out that PEGylation of AuNPs with second generation dendrons are the most benefited, being viability of **2bAuNP** at 70 ppm and of **2cAuNP** at 100 ppm clearly over 50%, whilst non-PEGylated **2-AuNP** at these concentrations were clearly toxic.

The low toxicity of gold core and the important influence of AuNPs substituents on toxicity has been previously reported.⁶⁶ On the other hand, it is known that positive charged systems destabilize negative charged cell membranes.⁴³ Hence, both haemolytic

and cytotoxicity properties of these dendronized AuNPs mainly arise from the presence of the cationic dendrons. Moreover, zeta potential analysis showed a diminishing of this value for PEGylated AuNPs, which can be related with the lower toxicity of PEGylated systems. For example, we have observed that viability of a fully PEGylated AuNPs was clearly over 80% at 100 ppm (**PEGAuNP**, Figure 3). However, PEGylation is not always an innocuous procedure and it has been described that this process affects the morphology and function of red blood cells in the long run, although do not have noticeable effects for short periods of time.^{67,68}

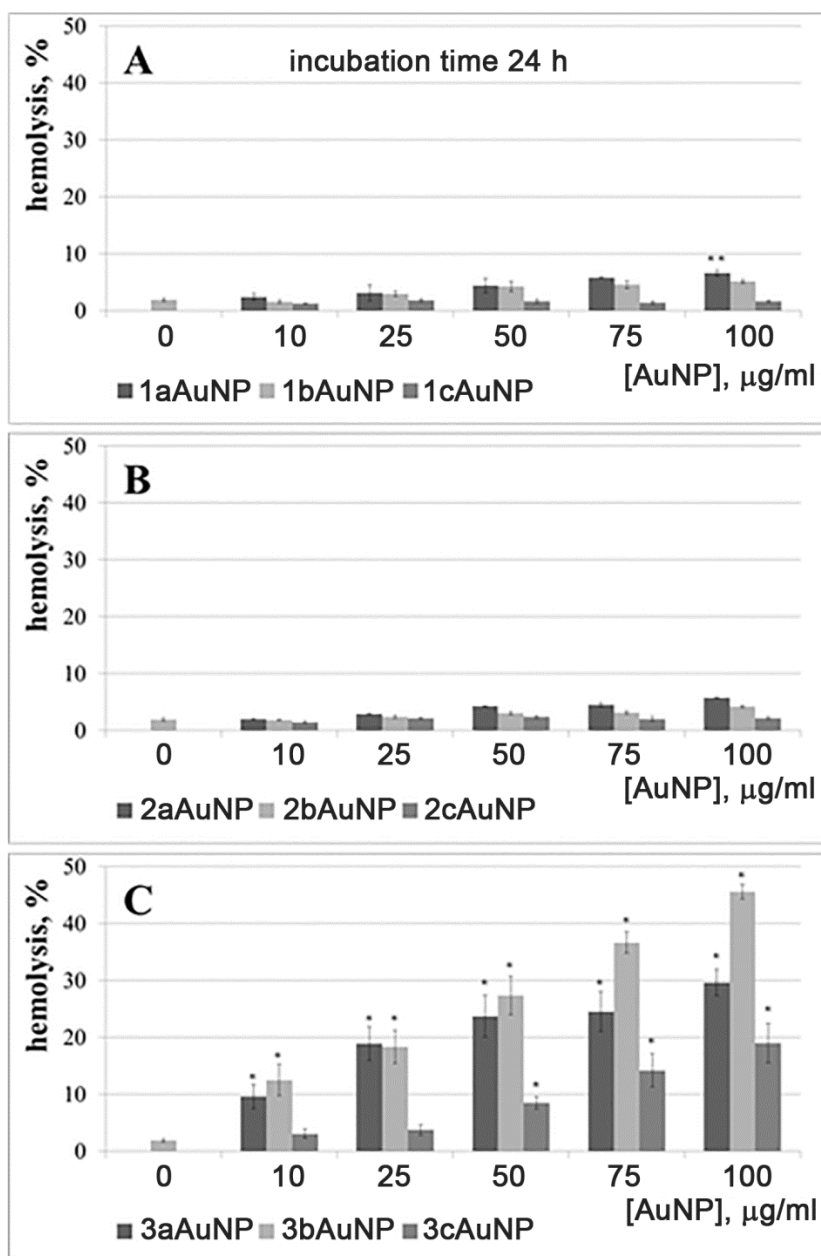


Figure 2. Extent of haemolysis after 24 h in the presence of PEGylated AuNPs decorated with dendrons: A) **1a-cAuNP**, B) **2a-cAuNP**, C) **3a-cAuNP**. Statistical significance of differences vs. control samples of blood erythrocytes alone (at * $p < 0.0002$, ** $p < 0.01$) was estimated by the post-hoc Newman-Keuls test. Statistical significance between haemolysis rates within the group with different degrees of PEGylation with equal concentrations occurs in case of **3a-cAuNP** with all samples inside groups between each other ($p < 0.005$) except **3a/3b** at concentration of 25 $\mu\text{g/mL}$.

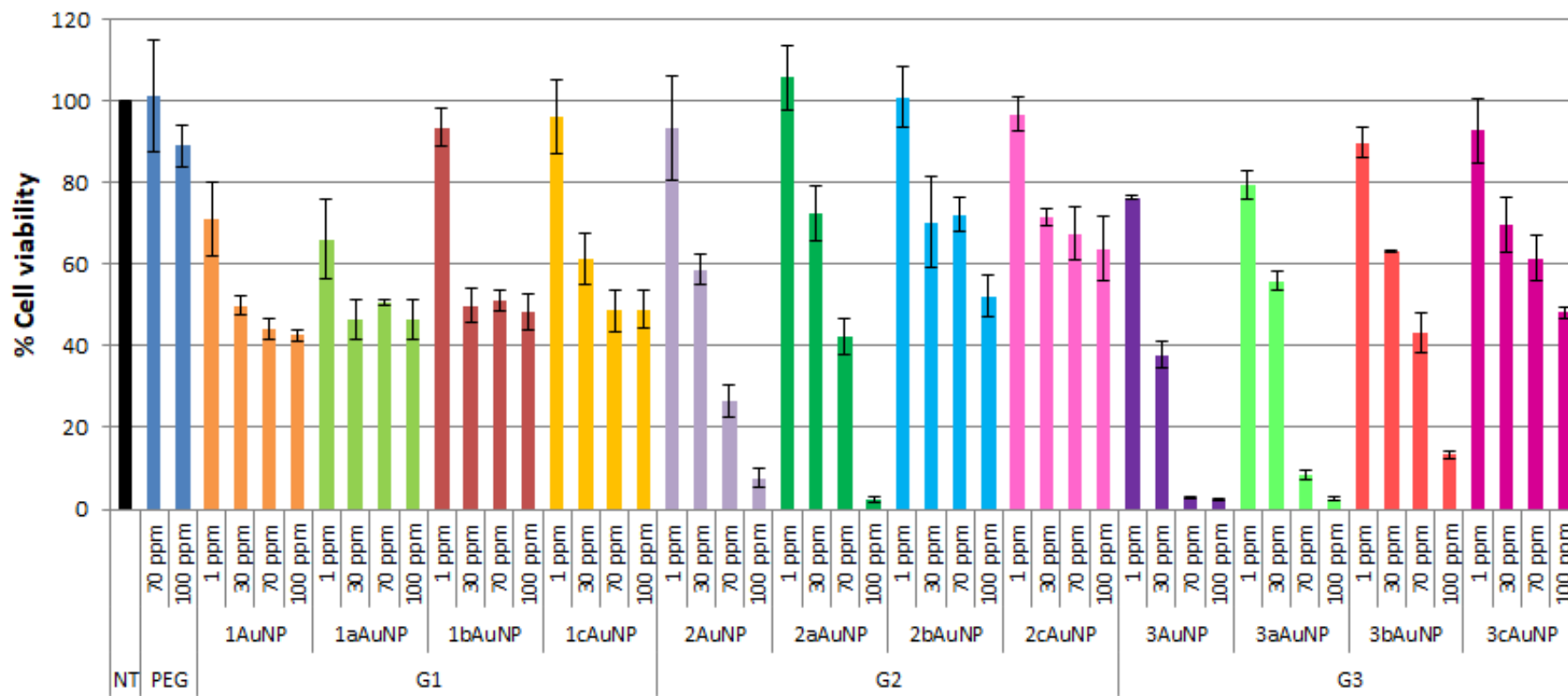


Figure 3. Cytotoxicity associated to homo and heterofunctionalized gold nanoparticles in HeLa cells 24 h post-loading using MTT assay.

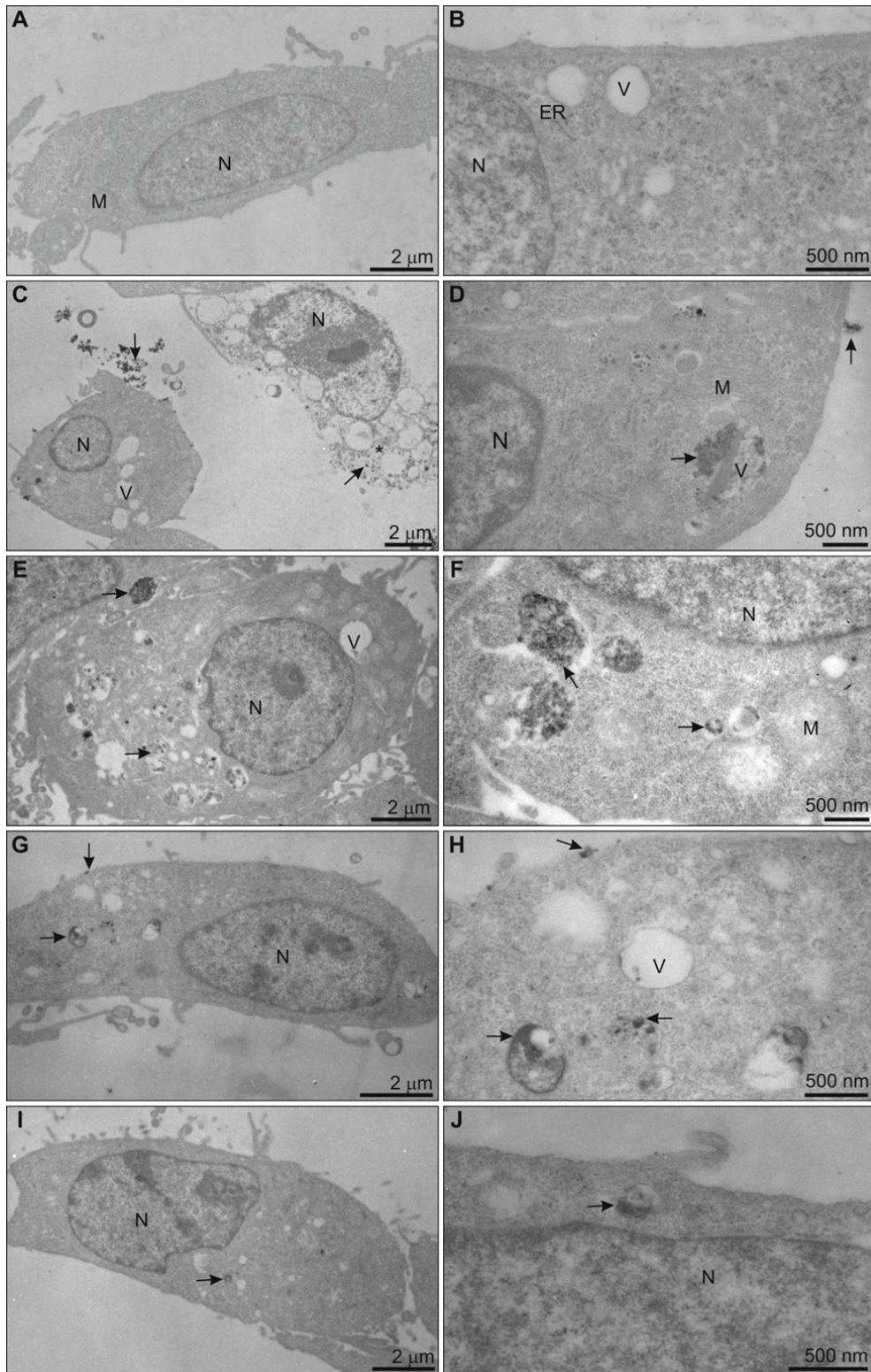


Figure 4. The ultrastructure of HeLa cells exposed to non-PEGylated 2AuNP and PEGylated 2a-cAuNP. A-B) control HeLa cell; C-D) 2AuNP (0.456 mg/ml); E-F) 2bAuNP; G-H) 2aAuNP; I-J) 2cAuNP. N – nucleus, M – mitochondrion, V – vesicle, ER – endoplasmic reticulum.

Ultrastructure of HeLa cells treated with PEGylated and non-PEGylated AuNPs.

The ultrastructure of HeLa cells treated with **2AuNP** (Figure 4C-D) as well as **2a-cAuNP** (Figure 4E-J) was examined in relation to the untreated cells (Figure 4A-B). The control HeLa cells were spindle shaped and had a big oval nucleus (Figure 4A-B). The endoplasmic reticulum cisternae and mitochondria were visible within quite dense area of cytoplasm filled with ribosomes. Only single vesicles were noticed in the cytoplasm and they did not contain any electron-dense cargo (Figure 4A-B). The cells treated with non-PEGylated **2AuNP** changed their shape to more round and they were more vesiculated than control ones. **The clusters of small electron dense granules were visible near cells (Figure 4C, arrows), adjusting their plasmalemma and in vesicles (Figure 4D, arrows). Such localization of AuNPs indicates endocytic pathway of uptake to cell.**⁶⁹ Some cells growing in AuNP solution showed necrosis features: loss of plasma membrane integrity, swelling of the cytoplasm and disruption of the organelles (Figure 2C, asterisk).

Regarding HeLa treatment with PEGylated AuNPs, those treated with **2bAuNP** showed only a slightly changed ultrastructure. They were rather oval in shape and had round nucleus. Moreover, many vesicles with electron-dense cargo (Figure 2E-F, arrows) were visible in the area of cytoplasm indicating intensive uptake of **2bAuNP**.

The cells exposed to **2aAuNP** and **2cAuNP** had the same shape as control ones (Figures 2G, 2I). The only noticeable effect of **2aAuNP** treatment was cell vesiculation and the presence of the electron-dense material adjacent plasma membrane and inside endocytic vesicles (Figures 2G-H, arrows). However the amount of vesicles and electron-dense deposits was significantly lower than in the **2bAuNP** treated cells.

HeLa cells exposed to **2cAuNP**, with the highest PEG content, were characterised by the lowest vesiculation and rarely found electron-dense material (Figures 2I-J,

arrows). This result suggests that this PEGylated AuNPs hardly entered tested cells, probably due to the lower content of cationic dendron 2G on their surface.

As commented above, localization of AuNPs indicates endocytic pathway of internalization. The metal core, due to its size, should not fundamentally change the mechanism and then it should be similar to that of dendrimers. However, there is no universal method for endocytosis and it is highly dependent on the type of cells. These AuNPs can penetrate the cell through receptor-mediated mechanisms, namely, clathrin- and caveola-dependent endocytosis, although the possibility of macropinocytosis is not excluded.^{70,71}

Lymphocyte proliferation. T lymphocytes (T-cells) are cells that act as regulatory agents of the immune system, defending the body from foreign antigens such as viruses and fungi. T cell proliferation initiates the immune response. The *in vitro* lymphocyte proliferation assay is used to determine whether cells can divide after exposure to a suitable stimulus or not, as well as to compare the ability of several cell populations to divide in response to the same stimulus.⁷² In order to determine if our systems constitute a non-specific antigen stimulus, the study of lymphocyte proliferation by **1-3,a-cAuNP** was carried out.

Peripheral blood mononuclear cells (PBMCs) were treated with PEGylated AuNPs (Figure 5). With the purpose of comparing these AuNPs with non-PEGylated (measured at 10 μ M and 20 μ M),³⁷ different concentrations were used for each generation: G1 (11 μ g/mL, 22 μ g/mL), G2 (15.2 μ g/mL, 30.4 μ g/mL), and G3 (25 μ g/mL, 50 μ g/mL). These concentrations in ppm for **1-3,a-cAuNP** corresponds with the molar concentration used for **1-3AuNP**. All considered AuNPs did not show antigenic effect (Figure 5) in given concentrations. Moreover, inhibition was observed in some cases.

Of all the NPs, **2aAuNP**, decorated with 2G dendrons and with the lowest PEG percentage, had clearly the strongest impact on lymphocyte proliferation (from 40% for 15.2 $\mu\text{g/mL}$ to 10% for 30.4 $\mu\text{g/mL}$). In the case of the **2bAuNP**, the level of proliferation fell from 88.8% (15.2 $\mu\text{g/mL}$) to 51.3% (30.4 $\mu\text{g/mL}$). Similar toxicity was shown by **2cAuNP** at concentration of 15.2 $\mu\text{g/mL}$ (85.5%) but slightly lower at a concentration of 30.4 $\mu\text{g/mL}$ (79.4%). For these AuNPs, the results showed that the degree of PEGylation seems to play a substantial role in the activity, probably by reducing interaction with the cationic groups of dendrons.

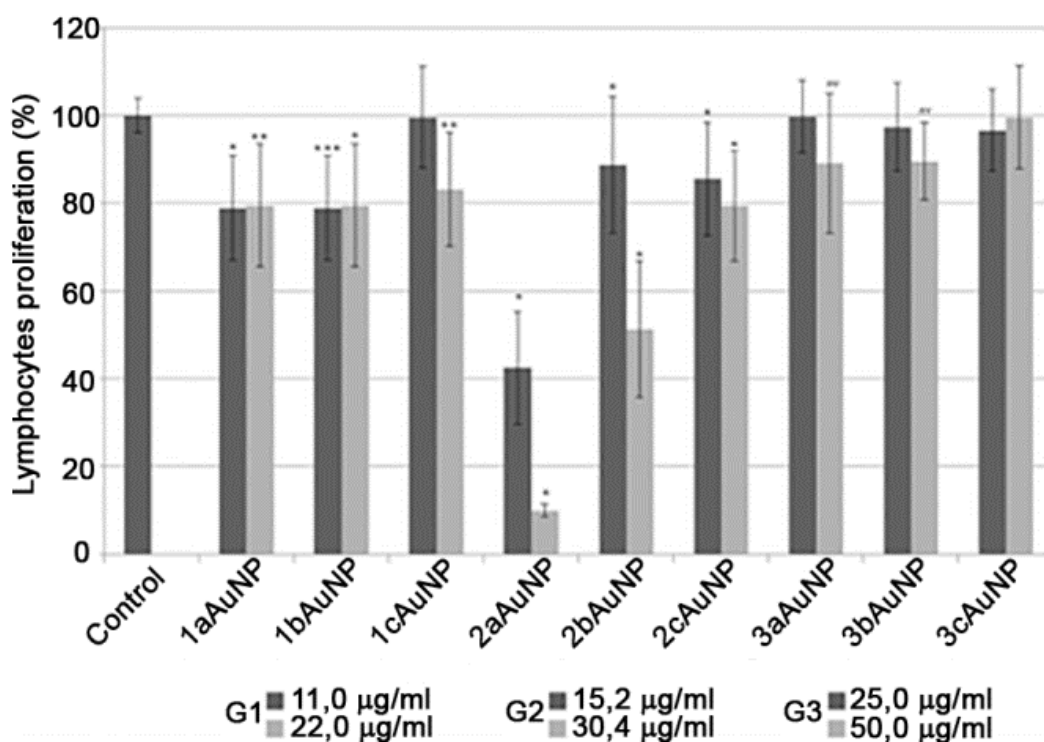


Figure 5. Extent of lymphocyte proliferation after 72 h in the presence of PEGylated gold nanoparticles **1-3,a-cAuNP**. Statistical significance of differences vs. control cells treated with PHA-M only (Phytohemagglutinin, M form) (at * $p < 0.00005$, ** $p < 0.0005$, *** $p < 0.003$, "" $p < 0.03$) was estimated by the post-hoc Newman-Keuls test.

Regarding AuNPs decorated with 1G dendrons **1aAuNP** and **1bAuNP**, they inhibited lymphocyte proliferation (about 20%) with no concentration dependence. In the case of the **1cAuNP**, with highest PEG percentage, at 11 $\mu\text{g/mL}$ they had no effect

at all, while at 22 $\mu\text{g/mL}$ proliferation was inhibited also about 20% (to 83.2%). Finally, **3a-cAuNPs** had the least impact on proliferation. The inhibition effect disappeared with increasing degree of PEGylation, as was observed at 50 $\mu\text{g/mL}$ for **3cAuNP**.

Platelets aggregation. The parameters of platelet aggregation induced by AuNPs in platelet rich plasma were measured (Figure 6). The concentrations of AuNPs to study platelets aggregation were chosen from the data of haemotoxicity of these NPs. Two concentrations were finally chosen: 10 and 100 $\mu\text{g/mL}$.

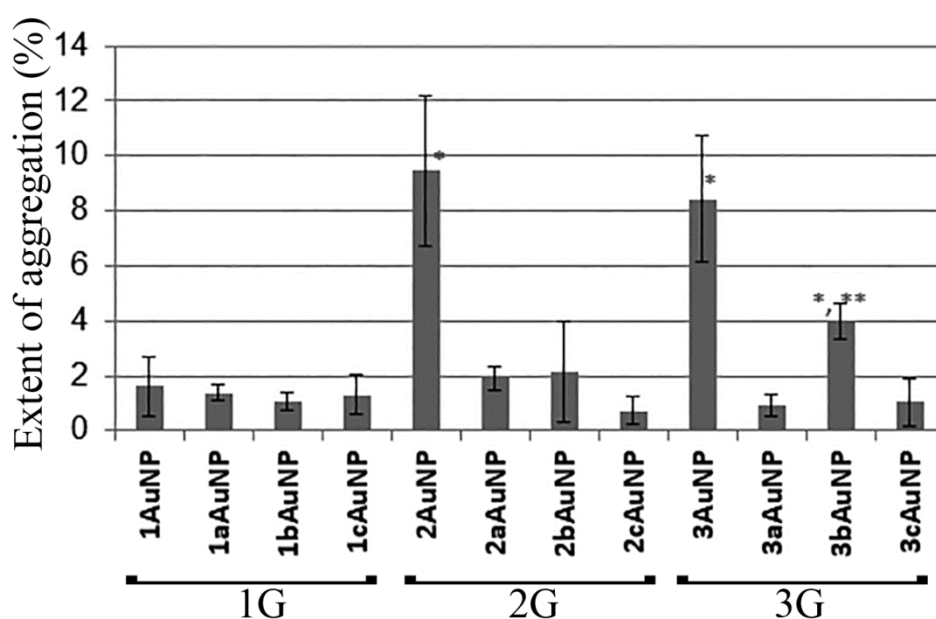


Figure 6. The extent in % of platelet aggregation induced by AuNPs in concentration of 100 $\mu\text{g/mL}$ in platelet rich plasma (225000-230000 cells per μL). $n = 6$, mean \pm S.D. * - $p < 0.05$ by ANOVA and post-hoc N-K test between (**1AuNP**, **2AuNP**, **3bAuNP**) and all other NP, ** - $p < 0.05$ by ANOVA and post-hoc N-K test between **3bAuNP** and (**1AuNP**, **2AuNP**).

All studied AuNPs did not induce aggregation of platelets below 10 $\mu\text{g/mL}$ (data not shown). At concentration of 100 $\mu\text{g/mL}$, PEGylated and non-PEGylated AuNPs differed in their effects, except for those covered with 1G dendrons. In these last cases, PEGylated **1a-cAuNP** and non-PEGylated **1AuNP** did not induce any effect on

platelets aggregation. In contrast, non-PEGylated AuNPs covered with 2G and 3G dendrons (**2AuNP**, **3AuNP**) significantly differed ($p < 0.05$) from PEGylated ones by their action on platelets. In general, PEGylated NPs **2a-cAuNP** and **3a-cAuNP** did not induce any effect on platelets aggregation, but **3bAuNP** (dendron/PEG = 1:1) that led to weak but statistically significant effect on aggregation of platelets. These data is in agreement with higher biocompatibility of PEGylated NPs.⁷³

Previously has been reported that colloidal AuNPs practically do not cause platelet aggregation.⁷⁴ The level of aggregation is also low for non-PEGylated nanoparticles here discussed. In this case, dendrons are probably not conducive to increasing aggregation, unlike typical dendrimers, because dendrons have less degrees of freedom while on the surface of these rigid nanoparticles.³⁷ Thus, dendrons on AuNPs cannot spread over the platelet membrane reducing the interaction with receptors that are involved in aggregation.^{75, 76}

Recognition of PEG with anti-PEG antibodies. PEGylation of biological and chemical compounds have many advantages such as good water solubility, lack of toxicity and low immunogenicity. However, long using PEGylated products lead to increase anti-PEG antibodies in blood. Thus, it is very important to check whether quantity of PEG chains and number and generation of dendrons attached to AuNPs have influence on PEG recognition by anti-PEG antibodies. To estimate the recognition of PEG attached to dendritic AuNPs, ELISA assays with anti-PEG antibodies were applied. These PEG-antibodies recognize the methyl group at the end of PEG chain.

On Figure 7, it can be observed that absorbance (optical density, OD) decreases in the presence of antibodies. The higher recognition by anti-PEG antibodies is observed for **2a-cAuNP**, with 2G dendrons. The lowest recognition is visible for dendritic **1a-cAuNP** and **3a-cAuNP**. Since anti-PEG antibodies recognize the methyl group at the

end of PEG chain, the more exposed PEG is, the higher recognition is observed. It seems that in **2a-cAuNP** PEG chains are more exposed and stretched due to the size of 2G dendrons, whilst in **3a-cAuNP** the bigger size of 3G dendrons hinder the interaction between antibodies and PEG. On the other hand, the PEG chain in **1a-cAuNP** containing the smallest 1G dendrons could be wrapped and therefore the end group of PEG chain would be less visible for antibodies.

On the other hand, we should expect that more PEG is attached to AuNPs surface, more antibodies can recognize these chains. However, the data point out that the higher recognition occurred for molar ratios 1:3 and 3:1 (dendron:PEG respectively). The lower recognition is for molar ratio 1:1. The explanation for the results is, probably, the nature of PEG chain alone. The PEG chain can be easily visible in presence of excess of dendrons (chain could be more stretched). With increasing PEG contain (molar ratio 1:1) the PEG is less visible because it could start to wrap around nanoparticles. In presence of excess of PEG, again PEG chains start to be more visible for antibodies. The above data suggest that recognition of PEGylated AuNPs depends on molar ratio between PEG and dendrons and on dendron generation.

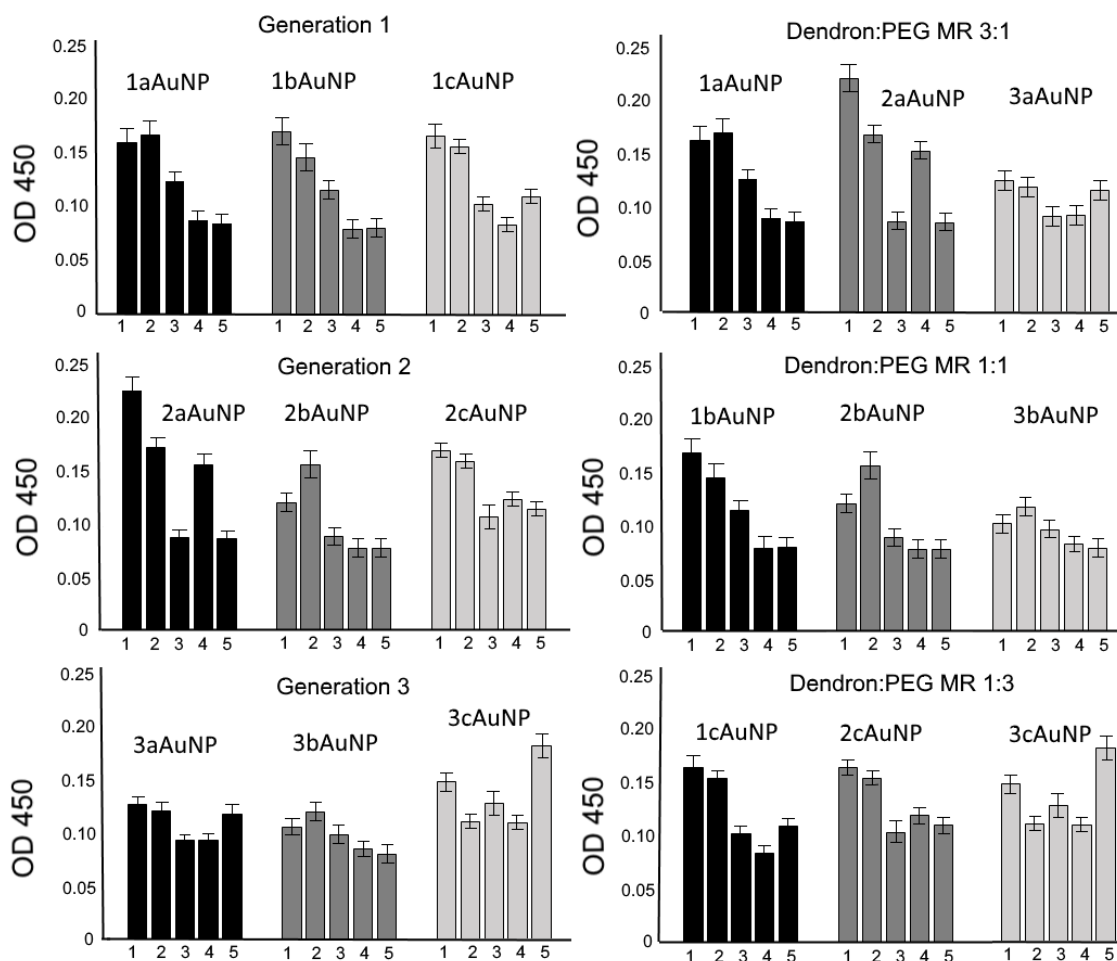


Figure 7. Direct ELISA antigen dose-response curve using anti-PEG antibodies.

Antibodies concentration series: 1- 0.125 mg/ml, 2- 0.0625 mg/ml, 3- 0.0312 mg/ml, 4- 0.0156 mg/ml, 5- 0.0078 mg/ml. A HRP-conjugated goat anti-rabbit IgM (1/1000) was used as the secondary antibody.

CONCLUSIONS

One pot reaction of gold precursor with cationic carbosilane dendrons and PEG is an adequate procedure for the synthesis of AuNPs covered with both type of ligands. The final dendron/PEG ratio is mainly dependent on initial one, observing higher presence of PEG ligands on AuNPs due to charge repulsion between cationic dendrons. However, the dendron/PEG ratio does not affect size of AuNPs. The presence of PEG chains on these heterofunctionalized AuNPs diminish the overall positive charge,

particularly on those AuNPs covered with lower generation dendrons. This behavior is due to the relative length of the PEG ligand with respect to dendron size, which overlaps positive charge of the dendrons.

The presence of PEG on dendronized AuNPs reduces haemolysis, platelet aggregation and modification of ultrastructure of a model of eukaryotic cells (HeLa). The uptake of AuNPs by cells was higher for non-PEGylated AuNPs (**2AuNP**) and PEGylated (**2bAuNP**). In the particular case of AuNPs with the higher proportion of PEG, minimum proportion of cationic dendron, their uptake was negligible, confirming the necessity of cationic groups to penetrate cell membrane.

Dendron generation and dendron/PEG ratio have an important influence on interaction of AuNPs with anti-PEG antibodies. The higher recognition by antibodies is for **2a-cAuNP**, with 2G dendrons. These data point out that in these AuNPs, the terminal group of PEG, which is recognized by antibodies, is more exposed due to the size of 2G dendrons. In **3a-cAuNP**, covered with 3G dendrons, these dendrons hide PEG chain, whilst in **1a-cAuNP**, covered with 1G dendrons, the smaller size of dendrons favours PEG folding making the methoxy group less available for interaction with antibodies.

Hence, the dendron/PEG ratio and dendron generation (size) clearly alter properties of AuNPs. This fact can be the key to choose the suitable system for transfection. Research on this idea is now being in progress.

MATERIALS AND METHODS

General considerations

All reactions were carried out under inert atmosphere and solvents were purified from appropriate drying agents when necessary. Unless otherwise stated, reagents were

obtained from commercial sources and used as received. Compounds HSG_n(S-NMe₃Cl)_m (where n = 1, m = 2; n = 2, m = 4; n = 3, m = 8) were synthesized as published³⁷. Thiol-ene reactions were carried out employing a HPK 125W Mercury Lamp from Heraeus Noblelight with maximum energy at 365 nm, in normal glassware under inert atmosphere. NMR spectra were recorded on a Varian Unity VXR-300 (300.13 (¹H)). Chemical shifts (δ) are given in ppm. ¹H resonances were measured relative to solvent peaks considering TMS = 0 ppm. UV-vis absorption was measured with a Perkin-Elmer Lambda 18 spectrophotometer. The spectra were recorded by measuring dilute samples in a quartz cell with a path length of 1 cm. **Zeta potential and DLS were measured in a Zetasizer Nano ZS (Malvern Instruments Ltd., UK) at 25 °C. Thermogravimetric analysis (TGA) were performed using a TGA Q500. Dry and pure samples (2-10 mg) were placed into platinum sample holder under nitrogen atmosphere. The measurements were recorded from 25 to 1000 °C, with heating rate of 10 °C/min.**

Synthesis of compounds

A description of the synthesis of first generation AuNPs with the equal ratio HSG₁(S-NMe₃Cl)/HS-PEG (**1bAuNP**) follows. The procedures and data of all compounds can be found in the Supporting information.

AuNP@PEG@SG₁(S-NMe₃Cl)₂ (1bAuNP). A mix of aqueous solution (53.2 mL, 12.5 mM) of compound HSG₁(S-NMe₃Cl)₂ (0.30 mmol, 149 mg) and HS-PEG (Mn = 800, 0.30 mmol, 240 mg) was added dropwise to an aqueous solution of HAuCl₄·3H₂O (20 mL, 30 mM, 0.6 mmol, 236 mg). NaBH₄ in water (15 mL, 200 mM, 3 mmol, 116 mg) was added dropwise before, and the mixture stirred 4 h. AuNP were purified by dialysis (MWCO 20 kDa) yielding **1bAuNP** (238 mg).

Data for: NMR (D₂O): ¹H NMR: δ 0.01 (SiCH₃), 0.56 (SCH₂CH₂CH₂CH₂Si), 0.87 (SiCH₂CH₂S), 1.45 (SCH₂CH₂CH₂CH₂Si), 1.73 (SCH₂CH₂CH₂CH₂Si), 2.62

(SiCH₂CH₂S), 2.89 (SCH₂CH₂N), 3.04 (NCH₃), 3.24 (OMe, PEG), 3.32 (SCH₂CH₂N), 3.56 (OCH₂CH₂O, PEG)). Reactant molar ratio HSG₁(S-NMe₃Cl)₂/HS-PEG molar ratio= 1/1. Calc. molar ratio (NMR) HSG₁(S-NMe₃Cl)₂/HS-PEG= 1/2.9. TGA (%): Au, 40.1; L, 59.9. UV-vis: 500 nm. Zeta Potential (mV): 23.8. DLS (nm): 24.47. Mean diameter of silver core (TEM) = 3.2 nm.

Hemotoxicity

Blood from healthy donors, obtained from Central Blood Bank in Lodz, was anticoagulated with 3% sodium citrate. Erythrocytes separated from blood plasma and leukocytes by centrifugation (4000 g, 10 min) at 4°C were washed 3 times with PBS (phosphate buffered saline; pH = 7.4). Erythrocytes were used immediately after isolation. PEGylated **1-3-a-cAuNP** were added at 10, 25, 50, 75 and 100 µg/ml at a 2% hematocrit value and incubated at 37°C. After 2 h and 24 h, these incubated suspensions were centrifuged (1000 g, 10 min) and haemolysis determined by measuring the free hemoglobin content in the supernatant at 540 nm. The percentage haemolysis was calculated from the formula:

$$\text{Haemolysis [\%]} = (A/Ac) \times 100\%$$

Where A is the absorbance of the sample, and Ac is the absorbance of the sample in water (100% haemolysis).

Transmission electron microscopy (TEM) of HeLa cells

Cells for TEM were fixed with 2.5% glutaraldehyde in 0.1 M PBS pH = 7.2 for 3 h at 4°C. Subsequently they were rinsed with the same buffer, scraped off and dispersed into 2 % agar. The material in the agar blocks was postfixed in 1% OsO₄ for 2 h at 4 C. Subsequently the material was dehydrated in a graded series of ethanol than propylene oxide and embedded in Epon-Spur's resin mixture. Sample resin blocks were sectioned on an Ultra Cut E (Reichert Jung, Germany) ultramicrotome with a diamond knife.

Ultrathin sections (60-70 nm) were placed on formvar coated nickel grids and stained with a uranyl acetate and lead citrate. The cell ultrastructure was examined in transmission electron microscope JEM 1010 (JEOL Ltd, Japan) at 80 kV. The developed films were scanned using Perfection V700 PHOTO scanner (Epson, Japan) at the resolution 1200 dpi.

Lymphocytes proliferation

Experiments were carried out on human lymphocytes, that isolated from whole blood of healthy donors. Lymphocytes were isolated using Histopaque 1077 in room temperature.

The cells were seeded at concentration 1×10^6 cells/mL in 100 μ L of full RPMI (10% FBS, 100 U/mL penicillin, 0.1 mg/ml streptomycin) on 96-well plates (100000 cells/well) and incubated at 37°C in humidified atmosphere with 5% CO₂.

Effects of dendronized and PEGylated AuNPs were observed by cell treatment with concentrations with respect to 10 and 20 μ M of non-PEGylated AuNPs.

The isolated cells were incubated 72 h in a humidified 37°C, 5% CO₂ incubator in the presence (test samples) or absence (control samples) of AuNPs, and in the presence or absence of phytohemagglutinin (PHA-M) to assess the inhibition or induction of proliferation, respectively. The final concentration of PHA-M was 10 μ g/ml.

After 72h incubation, cytotoxicity was evaluated by Alamar Blue assay (with resazurin, 10 μ g/ml). Fluorescence measurements were carried out on Fluoroscanner Ascent FL at $\lambda_{\text{ex}}=530$ nm and $\lambda_{\text{em}}=590$ nm. Viability of cells (%) was calculated relatively to the control (20 μ l of PBS added to the wells).

The degree of cell proliferation by AuNPs was calculated according to the formula:

$$Proliferation [\%] = \frac{A_{NP^+}}{A_{K^+}} \cdot 100\%$$

Where:

A_{den+} – absorbance of the samples treated with nanoparticle in the presence of PHA-M;

A_{K+} – absorbance of positive control.

The results are presented as mean \pm standard deviation; N = 6. Data were analyzed by conducting an ANOVA test with post-hoc Newman-Keulstests.

Platelets aggregation

Blood from healthy donors (men, 25-35 years) was taken. Aggregation of platelets was studied in platelet rich plasma.³⁷ For this, blood with addition of 3.8% of sodium citrate was centrifuged at 360 g for 5 min, and erythrocytes and leukocytes were removed to obtain platelet rich plasma.³⁷ Concentration of platelets in platelet rich plasma was at 225000 - 230000 cells per μL . Aggregation of platelets was studied using an automatic aggregometer AP2110 (SOLAR, Belarus). In the assay, 400 μL of platelets rich plasma was added to a thermostated (37°C) plastic tube at a final platelet concentration of 225000 - 230000 cells per μL .³⁷ AuNPs in two concentrations: 10 $\mu\text{g}/\text{mL}$ and 100 $\mu\text{g}/\text{mL}$ were then added to the platelet rich plasma to induce platelets aggregation. Data are expressed as mean \pm S.D. of 6 independent experiments. Significance was assessed using the one-way analysis of variance (ANOVA) with the post-hoc Newman-Keuls multiple comparisons test.

Recognition of PEG with anti-PEG antibodies

Wells were coated with 0.01 mg/ml (100 $\mu\text{L}/\text{well}$) of each dendritic gold nanoparticles. AuNPs were diluted in bicarbonate coating buffer and incubated on the plate overnight at 4°C. After that the plate was washed with PBS (containing 0.05% Tween20) 5 times and 150 μL of blocking buffer was added to each well. The plate was incubated for 1 h at 37°C and then washed 4 times in wash buffer. After blocking step the anti-PEG antibodies (Abcam) were added (in concentration range from 0.125 mg/ml

to 0.0078 mg/ml in volume 100 μ L) and samples were incubated for 1 h at 37°C. Again, after this step the plate was washed 3 times in wash buffer. Then, 100 μ L of horseradish peroxidase-conjugated secondary antibodies (diluted accordingly to company protocol) was added to each well and incubated for 1 h at 37°C. After washing the 100 μ L of OPD substrate solution (o-phenylenediamine dihydrochloride) was added to each well and incubated in room temperature for 30 min. After 30 min the stop solution (0.2 M H₂SO₄) was added and absorbance was measured at 450 nm.

SUPPORTING INFORMATION

Complete experimental section and methodology, synthesis of compounds, TEM images, NMR spectra.

CONFLICT OF INTEREST

The authors declare no conflict of interest.

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