Serbian Chemical Society Serbian Young Chemists' Club

# 9<sup>th</sup> Conference of the Young Chemists of Serbia **Book of Abstracts**



## 9th Conference of Young Chemists of Serbia Book of Abstracts

4th November 2023

**University of Novi Sad - Faculty of Sciences** 

9<sup>th</sup> Conference of Young Chemists of Serbia Novi Sad, 4th November 2023 Book of Abstracts

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#### Contents

Plenary Lecture	1
Invited Lectures	5
Oral presentations	11
Poster presentations	25
Chemistry and Society	27
Chemistry meets Biology	31
Developments in chemical synthesis	63
Environmental awareness	79
Physical and computational chemistry	97
Phytochemistry and Food Chemistry	117
Solution chemistry and Chemical equilibrium	149
Supramolecular Chemistry and Functional Materials	151
Author index	167

#### Preparation of NiO supported carbon paste electrode for sensitive and selective determination of Gallic acid in plant samples

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Gallic acid (GA), namely 3,4,5-trihydroxy benzoic acid, is a natural phenolic compound present in tea, wine, fruits, beverages, and various medicinal plants [1]. Due to their ability to scavenge free radicals and act as antioxidants, gallic acid and its derivatives, including lauryl-, propyl-, octyl-, tetradecyl-, hexadecyl-gallate, can prevent the oxidation and rancidity of oils and fats, and as a result, they are used as food additives [2]. There are numerous scientific reports on the biological and pharmacological activities of GA, with an emphasis on antioxidant, antimicrobial, anti-inflammatory, anticancer, cardioprotective, gastroprotective, and neuroprotective effects [1,2].

In this work, nickel oxide (NiO) nanoparticles were synthesized by the chemical coprecipitation method and used for modification of the carbon paste electrode for GA detection and determination. Morphological properties of prepared material were investigated using ICP-OES (inductively coupled plasma-optical emission spectrometry), XRD (X-ray diffraction), SEM and TEM (Scanning and Transmission Electron Microscopy). The electrochemical properties of the prepared electrode and the behavior of GA over the modified electrode were examined using CV (Cyclic Voltammetry), EIS (Electrochemical impedance spectroscopy), and SWV (Square Wave Voltammetry). The prepared electrode showed better electrocatalytic response than the bare carbon paste electrode. After square wave voltammetry (SWV) optimization, the electrode showed a wide linear working range from 0.2 to 100 µM at pH 3 of Britton-Robinson buffer solution (BRBS) as the supporting electrolyte. The excellent selectivity of the proposed method, with good repeatability and reproducibility, strongly suggests a potential application of the method for the determination of GA in plant samples.

#### References

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