

# Air Quality Control in Mine Refuge Chamber with Ventilation through Pressure Air Pipeline

Zujing zhang<sup>a,b</sup>, Hongwei Wu<sup>c</sup>, Kequan Wang<sup>b</sup>, Rodney Day<sup>c</sup>, Yanping Yuan<sup>a\*</sup>

<sup>a</sup>School of Mechanical Engineering, Southwest Jiaotong University, Chengdu, 610031, China

<sup>b</sup>Chongqing Research Institute of China Coal Technology & Engineering Group, Chongqing, 400037, China

<sup>c</sup>School of Engineering and Computer Science, University of Hertfordshire, Hatfield, AL10 9AB, United Kingdom

\*Corresponding author. Email: [ypyuan@home.swjtu.edu.cn](mailto:ypyuan@home.swjtu.edu.cn)

**Abstract:** A combined experimental and numerical study was performed to improve the performance of the ventilation system in a mine refuge chamber (MRC). In the experiment, CO<sub>2</sub> cylinders and dispersion pipes were used to simulate the CO<sub>2</sub> release of 50 people, and 0.1 L/min per person of fresh air was provided by an air compressor. A new analytical model for a 50-person MRC was proposed and validated against the experimental data. Sensitivity analysis was carried out to investigate the effects of several control factors. The results indicated the following: (1) The ventilation system layout has a significant influence on the CO<sub>2</sub> concentration distribution in an MRC, while the uniformity of the CO<sub>2</sub> concentration distribution in the MRC may not be effective with increased number of air inlets. (2) Under a well-arranged ventilation system in the 50-person MRC, the average CO<sub>2</sub> concentration can be controlled at less than 0.5% with a ventilation rate of 0.1 m<sup>3</sup>/min per person, and less than 0.2% with a ventilation rate of 0.3 m<sup>3</sup>/min per person. (3) A quantitative correlation exists between the CO<sub>2</sub> concentration and ventilation volume rate, as well as the CO<sub>2</sub> release rate, for an MRC under a well-arranged ventilation system.

**Keywords:** Underground; Mine refuge chamber; Ventilation; CO<sub>2</sub> concentration; Air quality.

## 1 Introduction

Underground coal mining is a very dangerous operation owing to accidents involving gas explosion, water inrush, and roof collapse, among others (Tripathy and Ala, 2018). In China, it has been reported that over 90% of mine accidents with 10 or more casualties were mainly a result of explosion, water inrush, and fire, of which more than 60% were explosion accidents (Wang et al., 2014a; Zhu et al., 2019). Nearly 90% of the victims died of CO poisoning and asphyxia in coal mine explosion accidents (Vaught et al., 2000; National Research Council, 2013; Wang et al., 2014b) and 80% of the victims died indirectly from the flue gas in fire accidents (Charles and Inoka, 2012; Hansen and Haukur, 2013). Moreover, in China, 44% of coal mine explosion accidents had rescue times of more than two days (He et al., 2019). An effective method for solving the problem of personnel being harmed by fire and smoke is to establish life-saving facilities in the underground roadway (Zhang et al., 2012). A mine refuge chamber (MRC) is regarded as one such major life-saving facility that can provide a safe environment for miners in distress for no less than 96 h in coal mines (Bauer and Kohler, 2009; Margolis et al., 2011). The removal of the CO<sub>2</sub> accumulated in an MRC is one of the key requirements for breathable air (Mejías et al., 2014). However, this is a challenging issue owing the lack of electric power following an accident and the explosion-proof requirements of the electrical equipment in a coal mine.

It has been recognised that human metabolism is the main source of harmful gases in an MRC. The harmful gases produced by human metabolism are mainly CO<sub>2</sub>, accompanied by trace harmful gases such as CO, H<sub>2</sub>S, and NH<sub>3</sub>. Among these, CO<sub>2</sub> has been proven to be a direct contaminant

46 and used as an indicator of air quality for numerous years (Jia et al., 2018). Zhai et al. (2018)  
 47 indicated that, for an adult man, the CO<sub>2</sub> metabolism rate is 199 mL/min when lying down, 228 to  
 48 287 mL/min when sitting down, and 237 to 300 mL/min when standing up.  
 49

<b>Nomenclature</b>		<b>Subscripts</b>	
$a$	Constant in $Cst$ expression	exp	Experimental data
$c_0$	Initial volume concentration	in	Air inlet
$C_1, C_2,$	Turbulence model parameter	$i, j$	Elemental directions ( $i, j=1, 2$ and $3$ corresponding to the $x, y,$ and $z$ directions)
$C_{1e}, C_{3e}$	Turbulence model parameter	num	Numerical data
$c$	Volume concentration	out	Air outflow
$C_p$	Thermal capacity of air, J/(kg·K)	sta	Stable
$D_s$	Component diffusion coefficient	<b>Greek symbols</b>	
$g$	Gravitational acceleration, m <sup>2</sup> /s	$\sigma_k, \sigma_\epsilon$	Turbulence model parameter
$G$	Ventilation volume rate, m <sup>3</sup> /s	$\Theta$	Difference
$G_b$	Generation of turbulence kinetic energy due to buoyancy (J/s·m <sup>3</sup> )	$\rho$	Air density, kg/m <sup>3</sup>
$G_k$	Generation of turbulence kinetic energy due to the mean velocity gradients (J/s·m <sup>3</sup> )	$\lambda$	Air thermal conductivity, W/(m·K)
$k$	Turbulent kinetic energy (J/kg)	$\epsilon$	Turbulent energy dissipation (J/kg·s)
$N$	Number of people	$\beta$	Coefficient of thermal expansion, 1/K
$p$	Pressure, Pa	$\mu$	Dynamic viscosity, kg/m·s
$S$	Modulus of the mean rate-of-strain tensor	$\mu_t$	Turbulent eddy viscosity, (kg/m·s)
$S_s$	Component production rate	$\nu$	Kinematic viscosity (m <sup>2</sup> /s)
$T$	Temperature, °C	$\tau$	Time, s
$u$	Air velocity, m/s	<b>Acronyms</b>	
$v$	CO <sub>2</sub> release rate per person, m <sup>3</sup> /s	ATSI	Air temperature slow increase
$V$	Volume, m <sup>3</sup>	MRC	Mine refuge chamber
$x$	Coordinate direction vector	PCA	Personnel concentrated area
		RSS	Relatively stable state

50  
 51 The CO<sub>2</sub> concentration has a certain influence on the indoor environmental comfort (Cheung  
 52 et al., 2017). Li et al. (2018) stated that, when people are exposed to an environment with a CO<sub>2</sub>  
 53 concentration of 12,000 ppm and relative humidity of 85%, significant headache symptoms can  
 54 easily be observed. Zhang et al. (2016a) concluded that exposure to an environment with a CO<sub>2</sub>  
 55 concentration of 5,000 ppm for 2.5 h did not increase the intensity of health symptoms of healthy  
 56 college students when they performed simple or moderately difficult cognitive tests and certain  
 57 tasks resembling office work. Liu et al. (2017) indicated that exposure to an environment with a  
 58 CO<sub>2</sub> concentration of 3,000 ppm and a temperature of 35 °C did not cause any change in the  
 59 measurable responses of people. However, new evidence suggests that CO<sub>2</sub> concentrations below  
 60 the occupational level of 5,000 ppm may affect the ability to make decisions, although these levels  
 61 have not been demonstrated to cause negative effects on health or comfort, or result in measurable  
 62 physiological responses (Persily, 2015; Gall et al., 2016). Kajtar et al. (2012) concluded that  
 63 exposure to 3,000 or 4,000 ppm CO<sub>2</sub> for several hours results in decreased cognitive performance  
 64 via a proof of reading exercise. Du et al. (2018) recommended that the conditions for an MRC  
 65 should be controlled with an O<sub>2</sub> volume fraction of approximately 18% to 22.7% and a CO<sub>2</sub>

66 volume fraction of less than 1%.

67 An air curtain system was used to prevent harmful gases from pouring into the MRC and  
68 thereby to reduce the influence of harmful gas in the underground roadway on the MRC air quality  
69 (Zhang et al., 2016b; Wang et al., 2017). Three methods have been proposed to control the  
70 concentration of harmful gases in an MRC at an acceptable level: a hangable lithium hydroxide  
71 curtain, air purification devices, and ventilation. Jia et al. (2014) investigated the performance of  
72 three materials, namely  $\text{Ca}(\text{OH})_2$ ,  $\text{LiOH}$ , and  $\text{NaOH}$ , as  $\text{CO}_2$  absorbents, and the results indicated  
73 that the reaction rates of these three materials with  $\text{CO}_2$  gas from fast to slow were  $\text{NaOH}$ ,  $\text{LiOH}$ ,  
74 and  $\text{Ca}(\text{OH})_2$ . Gao et al. (2015) investigated the application of  $\text{KO}_2$  to supply oxygen and remove  
75  $\text{CO}_2$  gas. Their results demonstrated that, for a 15 g  $\text{KO}_2$  solid plate formed by pressure extrusion  
76 at 10 kN, the average oxygen production rate is  $11.88 \times 10^{-3}$  L/min, while the average  $\text{CO}_2$   
77 absorption rate is  $11.0 \times 10^{-3}$  L/min. Gai et al. (2016) developed a purification device accompanied  
78 by a novel modified soda lime for an MRC. It was found that the optimal combination of soda lime  
79 with different mass fractions was 6% additives, 12%  $\text{H}_2\text{O}$ , and 6%  $\text{NaOH}$ , while the most effective  
80 working mode for the purification device was 25 W of fan power. Du et al. (2018) developed a  
81 novel modified soda lime with higher adsorption properties. Their results demonstrated that the  
82 adsorption capacity was increased by 36.2% and the adsorption rate was increased by 39.5%  
83 compared to normal soda lime. The modified soda lime was added to an air-purifying device, in  
84 which power was provided by explosion-proof axial flow fan air circulation (Du, 2017). Zhang et  
85 al. (2017) studied the layout of the air purification devices in a 50-person MRC. Their results  
86 indicated that two air-purifying devices can meet the air quality control requirements in the MRC,  
87 and the appropriate distribution of the two devices is one at each end of the MRC.

88 Ventilation is an effective measure of the oxygen supply and  $\text{CO}_2$  removal in an MRC, for  
89 which a borehole from the surface directly to the MRC is the most advantageous and reliable option.  
90 For example, the 33 existing coal MRCs in the United States use ground boreholes to supply fresh  
91 air (Trackemas et al., 2015). The ventilation volume rate for MRCs should not be lower than 0.3  
92  $\text{m}^3/\text{min}$  per person, according to the *Policy on the construction and management of the coal mine*  
93 *underground emergency refuge system* in China (National Coal Mine Safety Administration, 2013).  
94 However, this is not realistic for certain coal mines, as the fresh air for the MRC is supplied by the  
95 existing mine air pressure system. According to the “Coal Mine Safety Regulations (2016)” of  
96 China, the acceptable minimum ventilation rate for a mine air pressure system is 0.1  $\text{m}^3/\text{min}$  per  
97 person. Through theoretical calculations, Gao et al. (2012) determined that an air supply volume of  
98 90 L/min per person can meet the indoor  $\text{CO}_2$  concentration control requirements in an MRC. Shao  
99 et al. (2016) numerically investigated the variations in the  $\text{CO}_2$  concentration in an MRC with one  
100 air inlet and one air outflow. It was found that the average  $\text{CO}_2$  concentration in the MRC was  
101 maintained at 1% when the  $\text{CO}_2$  release rate was 0.41 L/min per person and the ventilation rate was  
102 42 L/min per person. You et al. (2012) and Jin (2013) conducted an experiment to control the air  
103 quality in an 80-person MRC by means of ventilation. The MRC was located underground of the  
104 Changcun Coal Mine in Shanxi province, China, as illustrated in Fig. 1. Their results demonstrated  
105 that, when the ventilation rate was 100 L/min per person, the  $\text{CO}_2$  and  $\text{O}_2$  concentrations in the  
106 MRC could be maintained at approximately 0.3% to 0.34% and 19.6% to 19.8%, respectively. He  
107 (2017) conducted an experiment to control the air quality in a 50-person MRC using ventilation.  
108 The results indicated that the minimum ventilation rate for meeting the air quality control  
109 requirements was 84 L/min when the  $\text{CO}_2$  release rate was 0.5 L/min per person.

110 In general, previous research on air quality control in an MRC mainly focused on the  
111 development of air purification devices,  $\text{CO}_2$  adsorption, and the ventilation volume rate that can

112 meet the air quality control requirements in an MRC. It is well known that the ventilation system  
113 layout plays an important role in the air quality distribution in an MRC, but few, if any, studies  
114 have focused on this aspect. Thus, it is imperative to improve the ventilation system for air quality  
115 in the MRC. In this work, a combined experimental and numerical study was performed to  
116 investigate the effects of the air inlet and air outflow layout, the ventilation volume rate, and the  
117 CO<sub>2</sub> release rate by refugees on the CO<sub>2</sub> concentration distribution in an MRC under ventilation.



118  
119  
120

Fig. 1 Experimental scene of air quality control for 80-person MRC by ventilation. [36]

## 121 2 Experimental setup

### 122 2.1 Experimental environment and principles

123 The experiment was conducted in a comprehensive MRC laboratory with inner dimensions of  
124 20 m in length, 4 m in width, and 3 m in height. A screw air compressor (DFB-100A) with a  
125 volume flow rate of 11.3 m<sup>3</sup>/min was used to supply fresh air to the MRC, whereby the fresh air  
126 entered the MRC through buried ventilating pipes. The ventilation volume rate was controlled by a  
127 total valve. There were three muffler air inlets on each side of the long channel. The air inlets were  
128 arranged 1.8 m above the ground, and the distance between two adjacent air inlets was 3.5 m. A  
129 one-way automatic exhaust valve with a diameter of 110 mm was installed at each end wall of the  
130 MRC as the air outflow. The exhaust valve opened automatically to vent the contaminated air  
131 when the relative pressure in the MRC reached 180 Pa. Both air outlets were located 2.4 m above  
132 the ground. Figure 2 illustrates the experimental environment, equipment, and instruments.



Fig. 2 Experimental environment and equipment.

According to temporary provisions, the CO<sub>2</sub> concentration (in volume) in an MRC cannot be less than 1%, and the capacity for CO<sub>2</sub> removal cannot be less than 0.5 L/min per person. To save on time and costs, the initial CO<sub>2</sub> concentration in an MRC is approximately 1%. The “Coal Mine Safety Regulations (2016)” in China stipulates that the average air supply of the pressurised air self-rescue device should be no less than 0.1 m<sup>3</sup>/min per person. Therefore, in the experiment, the ventilation volume rate was set to 300 m<sup>3</sup>/h (0.1 L/min per person). To ensure that the CO<sub>2</sub> release rate was maintained at a constant value of 0.5 L/min per person, high-pressure CO<sub>2</sub> cylinders and dispersion gas supply pipes were used to replace the CO<sub>2</sub> release rate of people. The CO<sub>2</sub> gas was released from the high-pressure CO<sub>2</sub> cylinders and then entered the MRC through diffusion air-supply pipes that were installed on both sides of the room. The diffusion gas supply pipeline was a stainless steel pipe with a 15 mm diameter and 10 m length. The diameter of the gas supply hole was 1.5 mm, and the distance between two adjacent holes was 100 mm. The CO<sub>2</sub> release rate entering the MRC was 25 L/min to simulate the CO<sub>2</sub> released by 50 people.

It should be noted that the CO<sub>2</sub> release channel in this experiment differed from that in practical conditions for an MRC, which could result in several differences in the CO<sub>2</sub> concentrations at the various measuring points. However, under the same operating conditions of the CO<sub>2</sub> release rate, ventilation rate, and layout of the air inlets and outlets, the error of the indoor average CO<sub>2</sub> concentration produced by simulating human exhalation of CO<sub>2</sub> through a CO<sub>2</sub> cylinder will be relatively small. In this study, a comparative experiment was conducted to reduce the calculation error of the average CO<sub>2</sub> concentration caused by the different measuring points. Two fans were used to stir the indoor air to make the CO<sub>2</sub> concentration distribution more uniform, while the other experimental parameters were maintained the same.

## 2.2 Measurement and data acquisition

There were seven measurement points for measuring the CO<sub>2</sub> concentration in the MRC. The distribution of the measurement points is illustrated in Fig. 3.

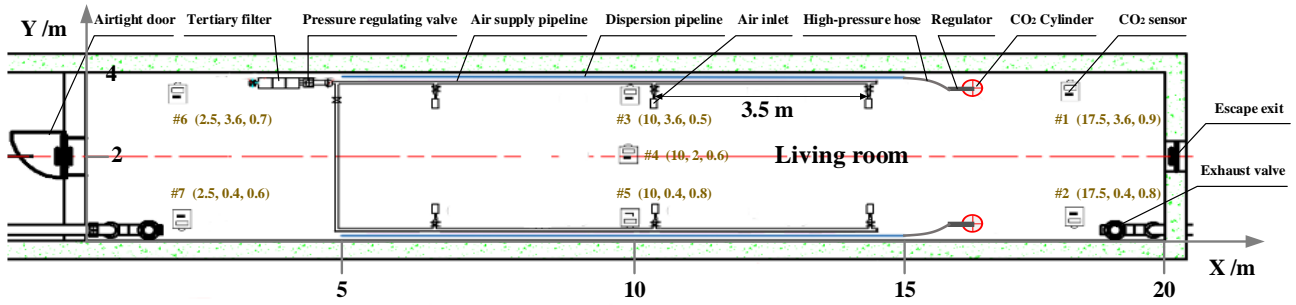


Fig. 3 Layout of air supply inlets and measurement points in living room.

The CO<sub>2</sub> concentration in the living room was monitored by infrared CO<sub>2</sub> sensors (GRG5H, Chongqing Research Institute of China Coal Technology and Engineering Group Co., Ltd.) in real time. The monitoring data were collected and saved automatically by the monitoring system platform. The ventilation volume rate for the living room was measured using a vortex flowmeter (SLDLUGB-DN50, Nanjing Senlod Measurement and Control Equipment Co., Ltd.) that was installed in the air-supply pipeline. The CO<sub>2</sub> release rate was measured by an electrically heated CO<sub>2</sub> decompression valve with a float flowmeter (YQT-731LR, Shanghai Regulator Co., Ltd.), which was linked to the high-pressure CO<sub>2</sub> cylinders. Figure 4 illustrates the interface between the measuring instruments and data acquisition system platform.



(a) Vortex flowmeter (b) CO<sub>2</sub> decompression valve (c) CO<sub>2</sub> sensor (d) Data acquisition platform

Fig. 4 Testing instruments and data acquisition system.

### 2.3 Experimental procedure

(1) The sensors were checked to ensure that they were working correctly and the data could be uploaded to the monitoring system platform.

(2) Operators entered the living room and closed the airtight door.

(3) The control valve of the CO<sub>2</sub> cylinder was opened and the flow rate of the CO<sub>2</sub> decompression valve was regulated at the maximum, so that the CO<sub>2</sub> concentration in the chamber increased to approximately 1%.

(4) The electric fans were turned on to stir the gas in the MRC until the CO<sub>2</sub> concentration approached 1%. (In the comparison experiment, the two fans were constantly running.)

(5) The CO<sub>2</sub> relief valves were adjusted to ensure that the CO<sub>2</sub> release rates on the two sides were 12 and 13 L/min, respectively.

(6) The air compressor was opened and the valve was adjusted to ensure that the ventilation volume rate for the MRC was 300 m<sup>3</sup>/h.

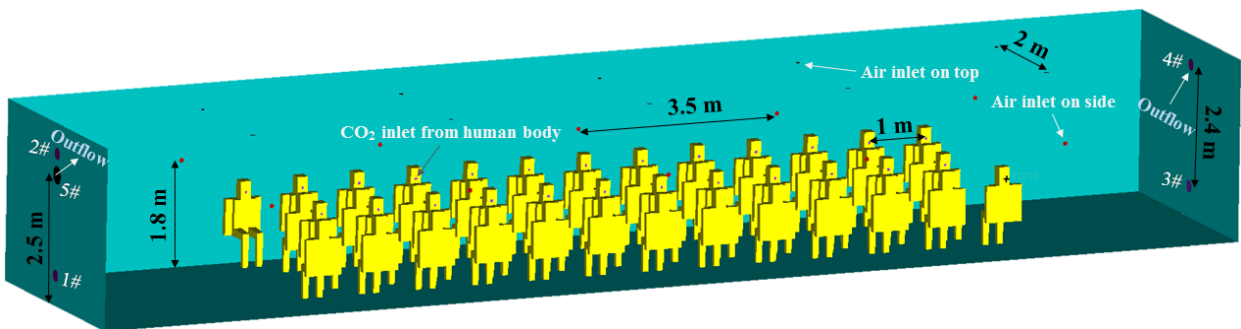
(7) The test was terminated once the CO<sub>2</sub> volume concentration was relatively stable in the living room.

191 **3. Computational details**

192 To optimise the layout of the air inlets and outlets of the ventilation system in the MRC, and  
193 to compute the optimum ventilation volume rate for an MRC, numerical analysis was also  
194 performed using ANSYS Fluent, which was consistent with the above experiment.

195 *3.1 Computational model*

196 For comparison with the above experimental results, a computational model of a 50-person  
197 MRC with an internal size of 20 m in length, 3 m in height, and 4 m in width was established.  
198 Considering the influence of the human body on the room space, a human model was incorporated  
199 into the computational model. The surface area and volume of the human model were 2 m<sup>2</sup> and  
200 0.067 m<sup>3</sup>, respectively. A square with an area of 0.08 m<sup>2</sup> was included on the head as the breathing  
201 gas outlet. A total of 50 people were divided into four rows in the room. Among these, there were  
202 13 people in each row near the two sides 0.3 m from the adjacent wall, and 12 people in each of  
203 the two middle rows, with a 0.4 m gap between two people's backs. The distance between two  
204 adjacent people in the same row was 1 m. The human bodies were placed 0.35 m above the ground  
205 to obtain high-quality boundary layer grids. To analyse the effects of the layout of both the air  
206 inlets and outflows on the CO<sub>2</sub> concentration distribution in the MRC, 20 air inlets with a diameter  
207 of 0.075 m, four air outflows with a diameter of 0.225 m, and one air outflow with a diameter of  
208 0.32 m were pre-positioned in the computational model. A total of 10 air inlets were located at the  
209 top of the MRC in two rows, with a row distance of 2 m, while the other 10 air inlets were located  
210 at either side of the MRC, at a distance of 1.8 m above the ground. The distance between two  
211 adjacent air inlets in the same row was 3.5 m. Two air outflows of the same area were located at  
212 the upper and lower parts of each wall at both ends of the MRC. The lower and upper air outflows  
213 were 0.3 and 2.7 m from the ground, respectively. The outlet with a diameter of 0.32 m was  
214 located at the upper part of the left end wall, 2.5 m from the ground. The geometric model of the  
215 50-person MRC is illustrated in Fig. 5.



216  
217 Fig. 5 Geometric model of 50-person MRC.  
218

219 The grids of the computational geometry were generated by ANSYS ICEM. Unstructured  
220 mesh was used owing to the complexity of the model structure. A grid independence study was  
221 performed to ensure that the numerical results were independent of the grid. Five grid models with  
222 different numbers of cells, namely  $1.12 \times 10^6$ ,  $1.68 \times 10^6$ ,  $2.31 \times 10^6$ ,  $3.34 \times 10^6$ , and  $4.51 \times 10^6$ ,  
223 were analysed under the same conditions. Figure 6 presents a comparison of the CO<sub>2</sub> concentration  
224 in the MRC at 1 and 2 h using the five different grids. It can be observed from Fig. 6 that, when  
225 the number of cells in the numerical model reached  $2.31 \times 10^6$ , the numerical results were not  
226 strongly affected by the number of grids. Therefore, considering the computational accuracy and  
227 resources, the grid model with  $2.31 \times 10^6$  cells was used for the following numerical analysis.

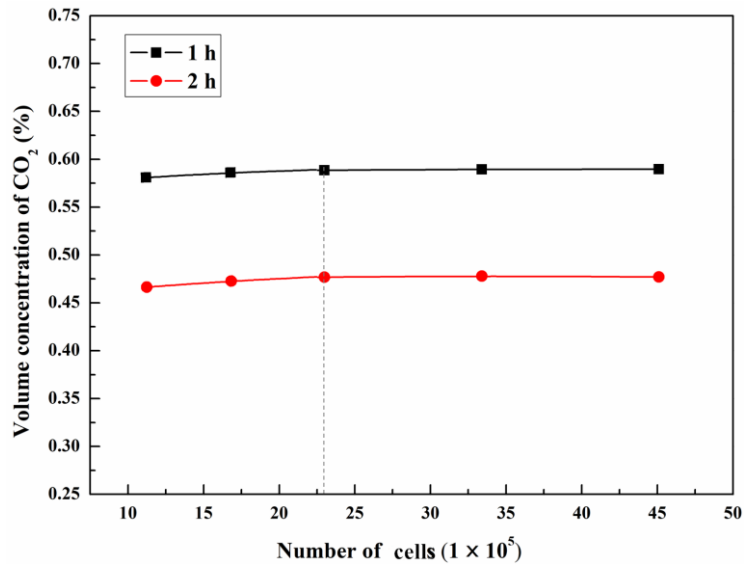


Fig. 6 Comparison of CO<sub>2</sub> concentrations in MRC at 1 and 2 h using five different grids.

### 3.2 Boundary conditions

In this work, trace gases such as CO, NH<sub>3</sub>, SO<sub>2</sub>, and NH<sub>3</sub> existing in the air environment and produced by human metabolism were neglected. The CO<sub>2</sub> gas was regarded as an indicator of the MRC air quality.

The human-exhaled gases were composed of O<sub>2</sub>, N<sub>2</sub>, CO<sub>2</sub>, and water vapour, the volume fractions of which were 16%, 78%, 4%, and 2%, respectively. The area for the human-exhaled gases was the inlet of the exhaled gas into the MRC. The velocity of the exhaled gas inlet was calculated according to the area of the exhaled gas inlets and CO<sub>2</sub> release rate. The velocity was 0.13 m/s when the CO<sub>2</sub> release rate was 0.5 L/min per person. The temperature of the exhaled gas was 35 °C, and the surface of the human body was defined as a constant temperature wall of 32 °C.

The volume fractions of the O<sub>2</sub>, N<sub>2</sub>, CO<sub>2</sub>, and water vapour in the fresh air for the air inlets were 21%, 78%, 0.03%, and 0.97%, respectively. In the model, six air inlets simultaneously delivered fresh air to the MRC at the same velocity. The velocity of the air inlets was calculated according to the area of the air inlets and ventilation volume rate of the MRC. The velocity was 3.204 m/s when the ventilation volume rate was 300 m<sup>3</sup>/h. The temperature of the air inlets was 32 °C.

The MRC exhaust vent was set as the air outflow. The number of air outflows was determined in terms of the working conditions. The walls of the MRC and other surfaces were set to a constant heat flux boundary with a heat flux rate of 0 W/m<sup>2</sup>.

In the initial air environment, regardless of the water vapour, the volume fractions of the O<sub>2</sub>, N<sub>2</sub>, and CO<sub>2</sub> were 20.5%, 78%, and 1.05%, respectively. The initial temperature was 25 °C.

### 3.3 Turbulence model

In this study, five operating conditions with different ventilation rates, namely 100, 150, 200, 250, and 300 L/min per person, were considered. The velocity of the air inlets ranged from 3.2 to 9.6 m/s. As the Reynolds number at the air inlet could be calculated as 9,091 to 29,032, the airflow in the MRC was turbulent.

The realizable  $k-\epsilon$  turbulence model was used because it exhibits strong performance with indoor airflows, temperature, and pressure in closed structures (Sørensen and Nielsen, 2003;



259 [Bacharoudis et al., 2007](#); [Piña-Ortiz et al., 2014](#); [Wu et al., 2015](#)). Enhanced wall treatment and a  
 260 full buoyancy effect were considered in the turbulence model. The pressure gradient and thermal  
 261 effect were neglected in the boundary function. Species transport was used for the component  
 262 analysis. The gaseous components, including O<sub>2</sub>, N<sub>2</sub>, CO<sub>2</sub>, and water vapour, were loaded into the  
 263 material item from the fluid material database.

264 The governing equations of the mass, momentum, and energy were given as follows ([Wu et](#)  
 265 [al., 2015](#)):

$$266 \quad \frac{\partial \rho}{\partial \tau} + \frac{\partial(\rho u_i)}{\partial x_i} = 0 \quad (1)$$

$$267 \quad \frac{\partial u_i}{\partial \tau} + \frac{\partial(u_i u_j)}{\partial x_j} = -\frac{1}{\rho} \frac{\partial P}{\partial x_i} + \frac{1}{\rho} \frac{\partial}{\partial x_j} \left[ \mu \left( \frac{\partial u_i}{\partial x_j} + \frac{\partial u_j}{\partial x_i} \right) - \overline{\rho u_i' u_j'} \right] - g_i \beta (T - T_0) \quad (2)$$

$$268 \quad \frac{\partial T}{\partial \tau} + \frac{\partial(u_j T)}{\partial x_j} = \frac{1}{\rho} \frac{\partial}{\partial x_j} \left( \frac{\lambda}{c_p} \frac{\partial T}{\partial x_j} \right) \quad (3)$$

269 The heat source was not considered in the energy equation, as the human surface was treated  
 270 as a constant temperature wall. The viscous dissipation could be ignored because almost no  
 271 mechanical energy was converted into heat.

272 The mass conservation equation of the component was ([Zhang et al., 2012](#)):

$$273 \quad \frac{\partial(\rho c)}{\partial \tau} + \frac{\partial(\rho u_j c)}{\partial x_j} = \frac{\partial}{\partial x_j} \left( D_s \frac{\partial(\rho c)}{\partial x_j} \right) + S \quad (4)$$

274 Two transport equations were formulated for the realizable  $k$ - $\varepsilon$  model, as follows ([Boulet et](#)  
 275 [al., 2010](#)):

$$276 \quad \frac{\partial k}{\partial \tau} + \frac{\partial(k u_j)}{\partial x_j} = \frac{1}{\rho} \frac{\partial}{\partial x_j} \left[ \left( \mu + \frac{\mu_\tau}{\sigma_k} \right) \frac{\partial k}{\partial x_j} \right] + \frac{G_k + G_b}{\rho} - \varepsilon \quad (5)$$

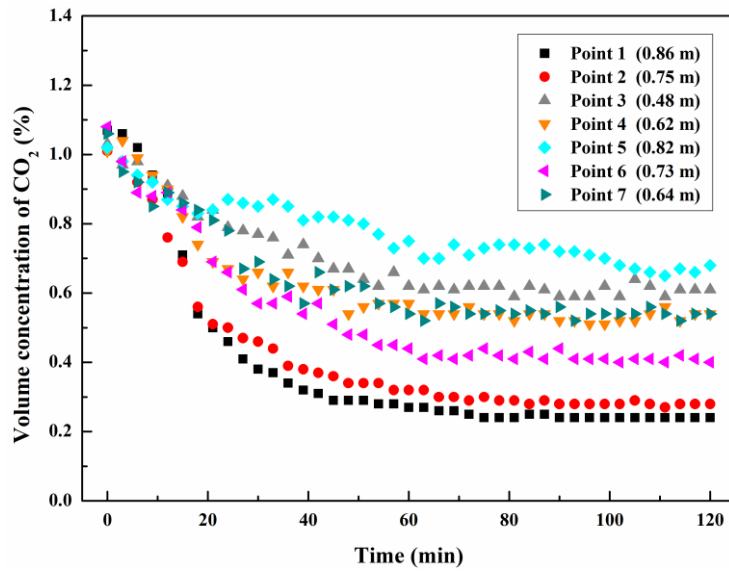
$$277 \quad \frac{\partial \varepsilon}{\partial \tau} + \frac{\partial(\varepsilon u_j)}{\partial x_j} = \frac{1}{\rho} \frac{\partial}{\partial x_j} \left[ \left( \mu + \frac{\mu_\tau}{\sigma_\varepsilon} \right) \frac{\partial \varepsilon}{\partial x_j} \right] + C_1 S \varepsilon + C_2 \frac{\varepsilon^2}{k + \sqrt{\nu \varepsilon}} - C_{1\varepsilon} \frac{\varepsilon}{k} C_{3\varepsilon} G_b \quad (6)$$

### 278 3.4 Calculation parameters

279 In this work, the Boussinesq assumption was used for the air operating density, taking the  
 280 gravity into account. The pressure-implicit with splitting of operators algorithm was applied. The  
 281 under-relaxation factors were 0.8 for the pressure, 0.2 for the momentum, and default values for the  
 282 other terms. The turbulence parameters of the air inlets, namely the turbulence intensity, turbulence  
 283 length scale, turbulence kinetic energy, and turbulence dissipation energy, were set as default values.  
 284 The pressure, power, turbulent kinetic energy, energy, time, and components were discretised by the  
 285 second-order upwind method. The convergence absolute criterion was 10<sup>-6</sup> for energy and 10<sup>-3</sup> for  
 286 the other items. The calculation was convergent when the time step ranged from approximately 0 to  
 287 5 s, and it was demonstrated that the numerical result was independent of the time step. In this work,  
 288 the time step was 5 s.

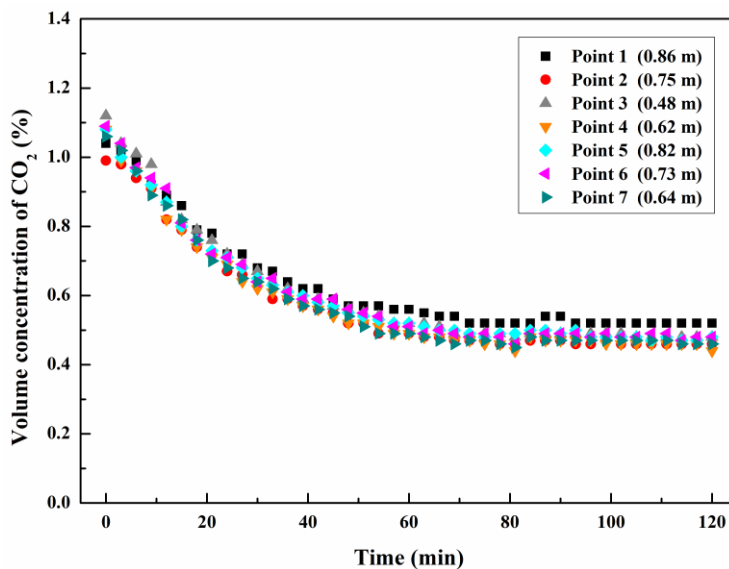
289 **4 Results and discussion**

290 *4.1 Variations in CO<sub>2</sub> concentration*



291  
292 Fig. 7 Variations in CO<sub>2</sub> concentration with time with ventilation volume rate of 300 m<sup>3</sup>/h.

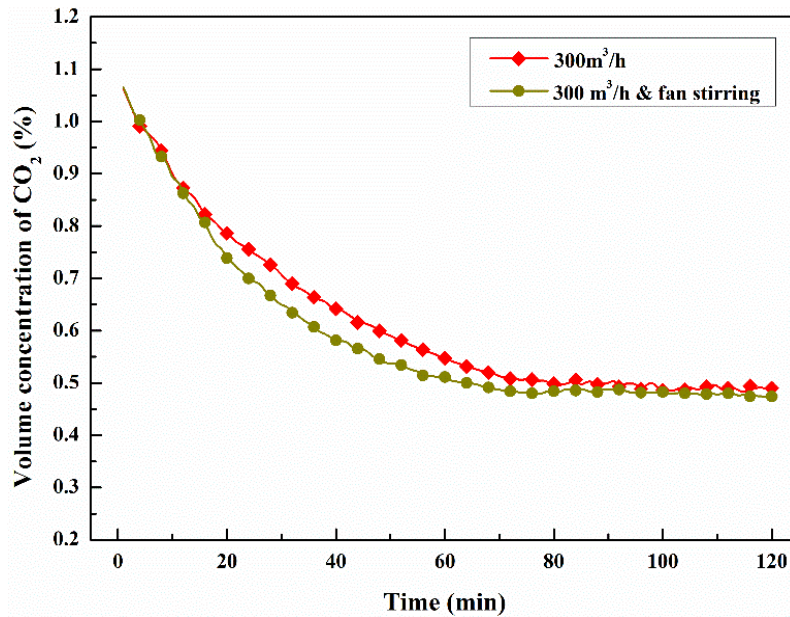
293 Figure 7 plots the variations in the CO<sub>2</sub> concentration at different measuring points when the  
294 ventilation volume rate was maintained at 300 m<sup>3</sup>/h. It can be observed from Fig. 7 that, in the  
295 MRC with an initial CO<sub>2</sub> concentration of 1.1%, the CO<sub>2</sub> concentration exhibited a decreasing  
296 trend and gradually tended towards a stable value when the ventilation volume rate was 300 m<sup>3</sup>/h.  
297 This indicates that, when the CO<sub>2</sub> release rate was 0.5 L/min and the ventilation volume rate was  
298 0.1 m<sup>3</sup>/min per person, the average CO<sub>2</sub> concentration in the MRC could be controlled below 1%.  
299 At the beginning of the ventilation, the CO<sub>2</sub> concentration decreased significantly, and it took  
300 approximately 1 h to reach a relatively stable state (RSS). It can be observed that, once the CO<sub>2</sub>  
301 concentration in the MRC reached the RSS, the CO<sub>2</sub> concentration value at each measurement  
302 point differed significantly, ranging from 0.25% to 0.75%, and the concentration difference  
303 between the highest and lowest points was approximately 0.5%. This implies that the uniformity of  
304 the CO<sub>2</sub> concentration distribution in the MRC was poor.



305  
306 Fig. 8 Variations in CO<sub>2</sub> concentration with time under combined action of air supply and fan  
307 stirring.

308

309 Figure 8 illustrates the variations in the CO<sub>2</sub> concentration at the measuring points when the  
310 ventilation volume rate for the MRC was 300 m<sup>3</sup>/h under the action of fan stirring. It can be  
311 observed from Fig. 8 that, when the ventilation volume rate was 300 m<sup>3</sup>/h and the fans were  
312 working to stir the indoor air, the CO<sub>2</sub> concentration in the MRC also exhibited an obvious  
313 decreasing trend within 1 h until reaching the RSS. The CO<sub>2</sub> concentration at the RSS stage ranged  
314 from 0.45% to 0.52%, and the concentration difference between the measuring points was less than  
315 0.1%. It should be noted that the disturbance of the fans caused the CO<sub>2</sub> concentration to be evenly  
316 mixed in the MRC.



317

318 Fig. 9 Comparison of average CO<sub>2</sub> concentrations for conditions with and without fan stirring.

319

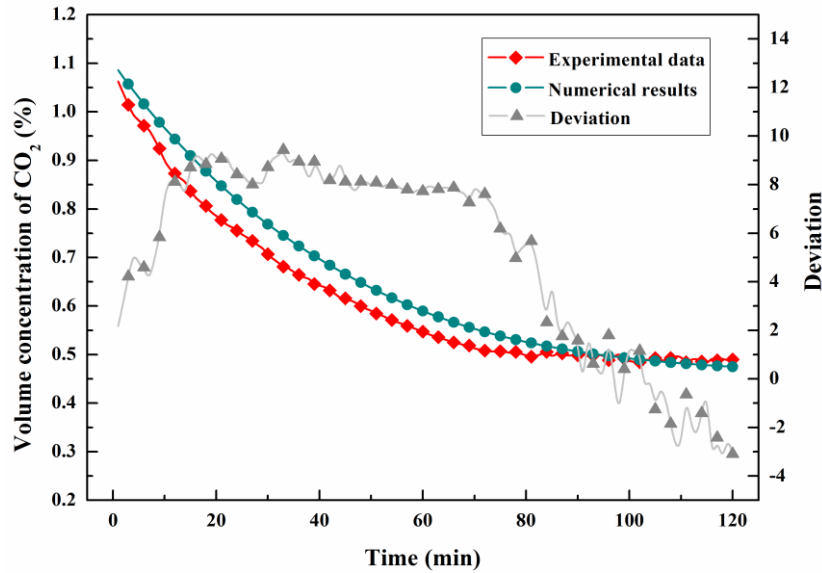
320 Figure 9 compares the average CO<sub>2</sub> concentrations of these measuring points with and  
321 without fan disturbance. It can be observed from Fig. 9 that, before the CO<sub>2</sub> concentration in the  
322 MRC tended to become stable, the average CO<sub>2</sub> concentrations of the measuring points with  
323 stirring were lower than those without, and the maximum difference between the two was  
324 approximately 0.06%. However, the variation trend of the average CO<sub>2</sub> concentration was not  
325 significantly affected by the fan stirring. The time of the CO<sub>2</sub> concentration entering the RSS was  
326 very close for these two cases, at approximately 70 to 80 min. During the RSS stage, the average  
327 CO<sub>2</sub> concentrations of these two cases were almost the same, with a concentration difference of  
328 less than 0.02%. Thus, it can be inferred that, even when the indoor airflow rate differs, the  
329 average CO<sub>2</sub> concentration in the MRC when reaching the steady state will be almost the same  
330 under the same layout of air inlets and air outlets, total CO<sub>2</sub> release rate, and ventilation volume  
331 rate.

332

333 The experimental results demonstrated that, when the CO<sub>2</sub> release rate was 0.5 L/min and the  
334 ventilation volume rate was 0.1 m<sup>3</sup>/min per person, the CO<sub>2</sub> concentration in the MRC could be  
335 effectively controlled below 0.8%, and the average CO<sub>2</sub> concentration was approximately 0.5%.  
336 However, the CO<sub>2</sub> concentration distribution in an MRC is not uniform under ventilation, and the  
337 air quality may be affected by poor ventilation systems owing to a higher CO<sub>2</sub> concentration in a  
338 local area. Thus, it is very important to improve the ventilation system layout to achieve acceptable  
air quality in the MRC.

339 4.2 Validation of numerical model

340 To verify the applicability of the numerical model, the numerical results were compared with  
341 the results of the above experiment without fan stirring. It should be noted that the distribution of  
342 the human CO<sub>2</sub> release outlets in the numerical model differed from that in the experiment, but the  
343 total CO<sub>2</sub> release rate was the same. Moreover, both the installation positions of the air inlets and  
344 the ventilation volume rate were the same in the numerical model and experiment, although there  
345 was a slight difference in the air inlet shape.



346 Fig. 10 Comparison of numerical results and experimental data.

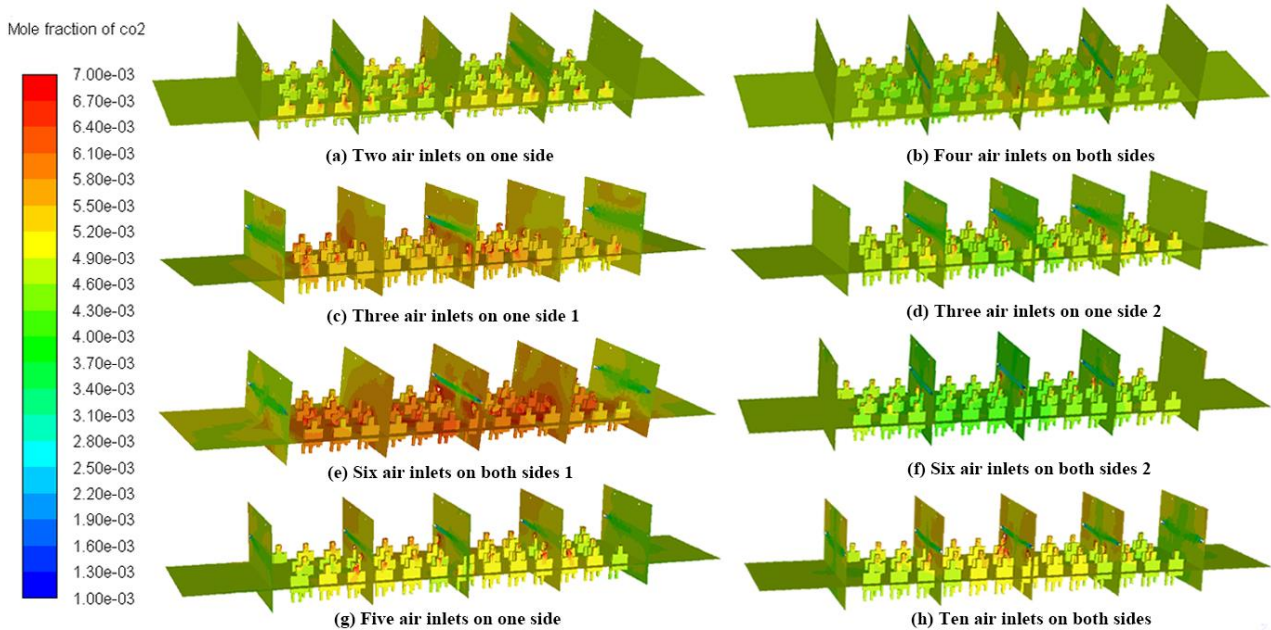
347  
348  
349 Figure 10 compares the average CO<sub>2</sub> concentrations of the measuring points varying with  
350 time in the MRC for the numerical analysis and experiment. The deviation between the numerical  
351 results and experimental data was calculated based on the average CO<sub>2</sub> concentration of the  
352 experiment, namely  $\Theta = (C_{Num} - C_{Exp}) / C_{Exp}$ . It can be observed from Fig. 10 that the predicted CO<sub>2</sub>  
353 concentration agreed well with the experimental data. The CO<sub>2</sub> concentration decreased rapidly  
354 from approximately 1.1% to 0.5% within approximately 70 to 80 min until it entered the RSS.  
355 From 0 to 1 h, the numerical results were slightly higher than the experimental data, the maximum  
356 difference in the CO<sub>2</sub> concentration was less than 0.7%, and the deviation was less than 10%. After  
357 reaching the stable state, the difference between the values decreased to 0.2% and the deviation  
358 was less than 4%, indicating that the numerical model is effective.

359 4.3 Sensitivity analysis

360 In practice, the effect of the air quality control in an MRC is influenced by several control  
361 parameters, such as the CO<sub>2</sub> release rate of people, ventilation rate, ventilation system layout, and  
362 personnel distribution. Among these, the personnel distribution exhibits a certain randomness. In  
363 terms of the overall space arrangement in an MRC, the space at both ends of the MRC is generally  
364 occupied by some equipment and lockers for storing household or escape items, and people are  
365 usually arranged in several rows in the remaining space, with a certain distance for them to walk.  
366 The personnel distribution implemented in the numerical 50-person MRC is a very common  
367 distribution pattern, which makes full use of the indoor space, and is also convenient for people to  
368 walk and talk to each other. Therefore, in this study, the influence of the personnel distribution on  
369 the air quality control was not considered.

370 4.3.1 Layout of air inlets

371 A series of numerical analyses with different air inlet layouts were performed to investigate the  
372 effect of the layout of the air inlets on the CO<sub>2</sub> concentration distribution in the MRC. Under these  
373 conditions, an outflow was located at the top of each end, the ventilation volume rate was 300 m<sup>3</sup>/h,  
374 and the CO<sub>2</sub> release rate was 0.5 L/min per person.



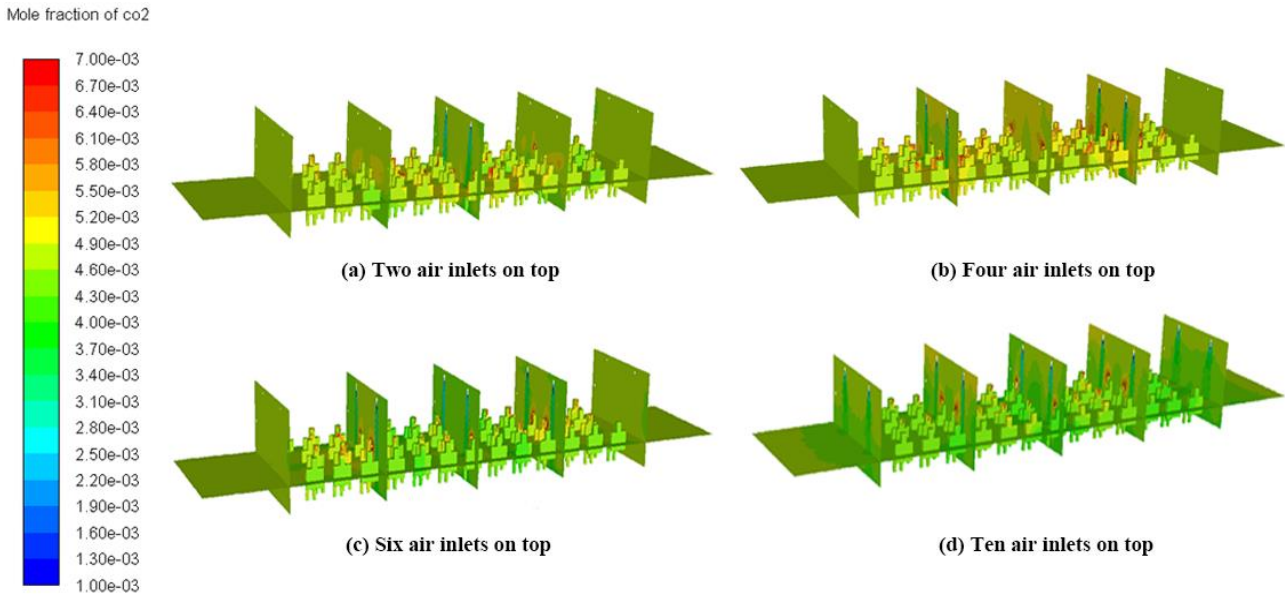
375  
376 Fig. 11 Distribution of CO<sub>2</sub> concentration under different air inlet layouts at both sides.  
377

378 Figure 11 illustrates the CO<sub>2</sub> concentration distribution when the ventilation volume rate was  
379 300 m<sup>3</sup>/h under different air inlet layouts at both sides of the MRC. It can be observed from Fig. 11  
380 that, when the ventilation volume rate was 300 m<sup>3</sup>/h, the CO<sub>2</sub> concentration ranged from 0.3% to  
381 0.7%. Therefore, it can be concluded that a ventilation volume rate of 0.1 m<sup>3</sup>/min per person can  
382 meet the requirements of controlling the CO<sub>2</sub> concentration in an MRC within 1%. However, the  
383 layout of the air inlets has an important influence on the CO<sub>2</sub> concentration distribution in the  
384 MRC.

385 According to Figs. 11(a), (c), (d), and (g), when the air inlets were located at one side of the  
386 tunnel, the CO<sub>2</sub> concentration distribution in Fig. 11(d) with three air inlets located in the  
387 personnel-concentrated area (PCA) was superior to that with the two air inlets (a). Moreover, when  
388 the number of air inlets increased to five and two of the inlets were located in the no-person area  
389 (g), the CO<sub>2</sub> concentration increased and the uniformity of the CO<sub>2</sub> concentration distribution in  
390 the MRC decreased obviously. It can be observed from Fig. 11(g) that the CO<sub>2</sub> concentration in the  
391 PCA was higher than that in the no-person area at both ends. Figures 11(c) and (d) clearly indicate  
392 that, for the cases with the same number of air inlets, the air inlets located in the PCA were  
393 obviously more conducive to air quality control than those distributed near both ends. Similarly, it  
394 can be observed from Figs. 11 (b) and (f) that, when the air inlets were arranged on both sides of  
395 the tunnel, the CO<sub>2</sub> concentration distribution under the condition of three air inlets located on  
396 each side of the PCA (f) was obviously superior to that of two air inlets located on each side (b).  
397 Comparing Fig. 11 (f) with (e) and (h), it can be observed that distributing or adding air inlets near  
398 both ends in the no-person area was not conducive to air quality control in the MRC. The reason is  
399 that the CO<sub>2</sub> concentration at both ends was relatively low when the air inlets were added into the  
400 no-person area, while the CO<sub>2</sub> concentration of the polluted air removed from the outflows was

401 relatively low, which reduced the efficiency of fresh air replacing the CO<sub>2</sub> contaminated gases.  
402 Therefore, when air outflows are located at both ends and air inlets are arranged on both sides in  
403 an MRC, the air inlets should be located in the PCA.

404 Moreover, it can be observed from Figs. 11(d) and (f) that, in these two cases with three air  
405 inlets located on only one side of the PCA (d) and on each side of the PCA (f), the difference in the  
406 CO<sub>2</sub> concentration distribution was relatively small, ranging from 0.3% to 0.5%. However, when  
407 three air inlets were located on each side of the PCA, the CO<sub>2</sub> concentration in the PCA was  
408 obviously better distributed than that with only three air inlets located on one side. Therefore,  
409 when air outflows are located at both ends, air inlets evenly distributed on both sides of the PCA  
410 will be more conducive to the removal of harmful gases in the MRC.



411

412 Fig. 12 Distribution of CO<sub>2</sub> concentrations under different air inlet layouts at top.

413

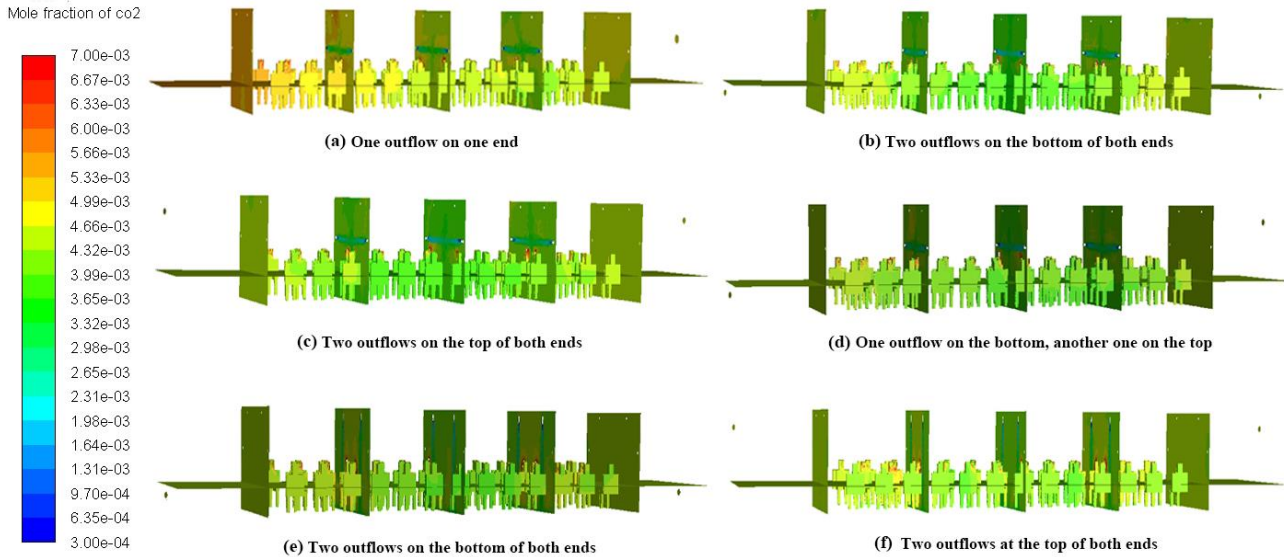
414 Figure 12 illustrates the CO<sub>2</sub> concentration distribution under different layouts of the air inlets  
415 located at the top of the MRC. It can be observed from Fig. 12 that, when the air inlets were  
416 located at the top of the MRC and the ventilation volume rate was 0.3 m<sup>3</sup>/h, the CO<sub>2</sub> concentration  
417 in the MRC ranged from 0.3% to 0.6% under the four different air inlet layouts. It can be  
418 concluded that the CO<sub>2</sub> concentration distribution in the MRC was relatively uniform when only  
419 two air inlets were located at the top of the middle of the MRC, as indicated in Fig. 12(a).  
420 However, the CO<sub>2</sub> concentration uniformity decreased when four air inlets were located at the top,  
421 as illustrated in Fig. 12(b). It is obvious that the CO<sub>2</sub> concentration in the PCA was higher than that  
422 in the no-person area. When there were six air inlets located at the top of the PCA, as illustrated in  
423 Fig. 12(c), the CO<sub>2</sub> concentration in the PCA was lower than that in the no-person area, which is  
424 more conducive to controlling the harmful gas concentration in the MRC for refugees. Moreover,  
425 it can be concluded that the CO<sub>2</sub> concentration in the MRC was generally well distributed when  
426 the number of air inlets located at the top increased to 10, as indicated in Fig. 12(d).

427

428 Comparing Figs. 10 and 11, it can be determined that, when the fresh air flowed into the  
429 MRC from the top air inlets, the CO<sub>2</sub> concentration in the lower space of the MRC was lower than  
430 that in the higher space, which is conducive to controlling the air quality in the breathing area.  
431 However, there was no significant difference in the CO<sub>2</sub> concentration distribution between the  
cases with six air inlets located at both sides and six air inlets located at the top in the PCA.

432 4.3.2 Layout of air outflows

433 A series of numerical analyses with different air outflow layouts were performed to investigate  
 434 the effect of the air outflow layout on the CO<sub>2</sub> concentration distribution in the MRC. In the  
 435 simulation, the ventilation volume rate was 300 m<sup>3</sup>/h and the CO<sub>2</sub> release rate was 0.5 L/min per  
 436 person.



437  
438 Fig. 13 Distribution of CO<sub>2</sub> concentration under different air outflow layouts.

439  
440 Figure 13 illustrates the distribution of the CO<sub>2</sub> concentration in the MRC under different air  
 441 outflow layouts. It can be observed from Fig. 13(a) that the uniformity of the CO<sub>2</sub> concentration  
 442 in the MRC was poor when only one outflow was installed at one end. Moreover, the CO<sub>2</sub>  
 443 concentration in the area near the end without outflow was higher than that in the area near the end  
 444 with outflow. Furthermore, Figs. 13(b), (c), and (d) indicate that, under the same air supply  
 445 conditions, when one air outflow was located at each end, there was no significant change in the  
 446 CO<sub>2</sub> concentration distribution in the MRC, regardless of whether both air outflows were located  
 447 at the upper part or lower part, or one was located at the upper part and the other at the lower part,  
 448 respectively. Comparing Figs. 13(e) and (f), it can be concluded that, when six air inlets were  
 449 distributed in two rows at the top of the MRC, the CO<sub>2</sub> concentration of the air outflows located at  
 450 the top of each end was slightly lower than that of the air outflows located at the bottom of each  
 451 end. This could be owing to the low CO<sub>2</sub> concentration in the lower space and high CO<sub>2</sub>  
 452 concentration in the upper space when the air inlets were arranged at the top.

453 According to the results, it is necessary to arrange one outflow at each end of the wall on both  
 454 ends of the MRC. The location of the outflow at the top or bottom of the wall has negligible effect  
 455 on the CO<sub>2</sub> concentration distribution in the MRC.

456 4.3.3 Ventilation volume rate

457 To investigate the effect of the enhanced ventilation volume rate on improving the air quality  
 458 in the MRC, five different ventilation volume rates, namely 0.1, 0.15, 0.2, 0.25, and 0.3 m<sup>3</sup>/min  
 459 per person, were analysed, with the other parameters remained the same. That is, three air inlets  
 460 were located at each side and one outflow was located at the top of each end, and the CO<sub>2</sub> release  
 461 rate was 0.5 L/min per person.

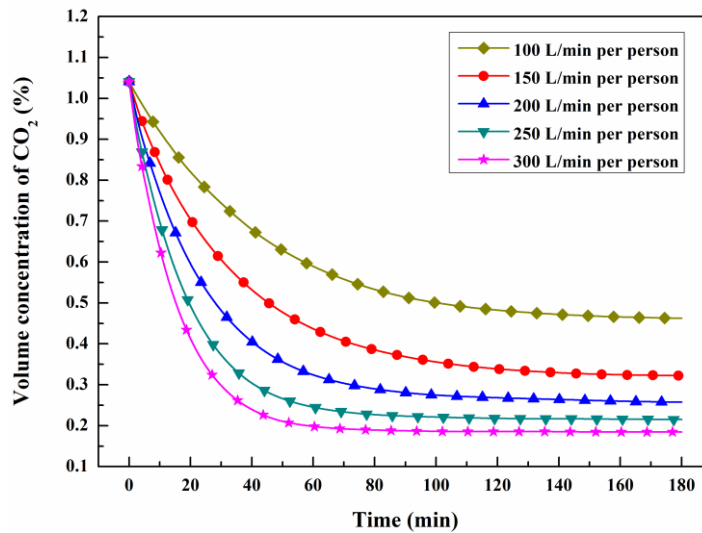


Fig. 14 Variations in CO<sub>2</sub> concentration with time under different ventilation volume rates.

462  
463  
464

465 Figure 14 plots the curves of the average CO<sub>2</sub> concentration in the MRC varying with time  
466 under different ventilation volume rates. It can be observed that, when the ventilation volume rate  
467 for the MRC was greater than or equal to 0.1 m<sup>3</sup>/min per person, the average CO<sub>2</sub> concentration  
468 decreased with time until it tended towards a relatively stable value of less than 0.5%. With the  
469 increase in the ventilation volume rate, both the time to reach relative stability and the stable CO<sub>2</sub>  
470 concentration decreased. The time to reach the RSS was approximately 2 h when the ventilation  
471 volume rate was 0.1 m<sup>3</sup>/min per person, and less than 1 h when the ventilation volume rate reached  
472 0.3 m<sup>3</sup>/min per person. The stable CO<sub>2</sub> concentrations for the five different ventilation rates of 0.1,  
473 0.15, 0.2, 0.25, and 0.3 m<sup>3</sup>/min per person were 0.483%, 0.32%, 0.257%, 0.215%, and 0.185%,  
474 respectively.

475 *4.3.4 CO<sub>2</sub> release rate*

476 To investigate the effect of the CO<sub>2</sub> release rate of human metabolism on the air quality control  
477 in the MRC, five different CO<sub>2</sub> release rates, namely 0.3, 0.35, 0.4, 0.45, and 0.5 L/min per person,  
478 were analysed, while the other parameters were maintained the same. That is, three air inlets were  
479 located at each side and one outflow was located at the top of each end, and the ventilation volume  
480 rate was 0.1 m<sup>3</sup>/min per person.

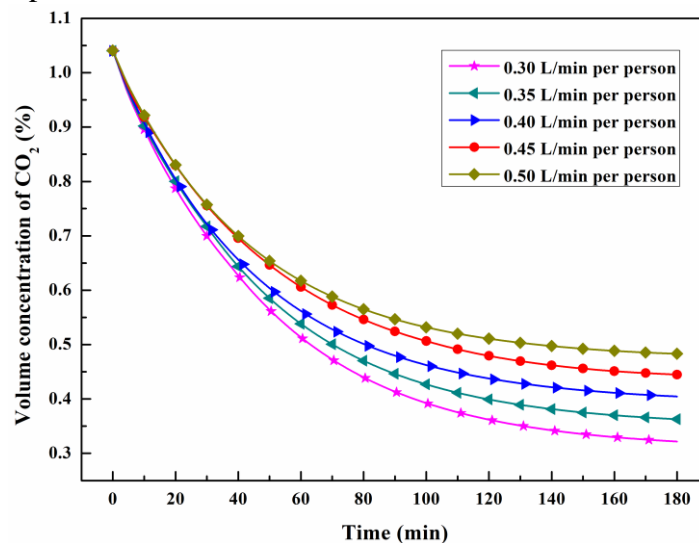


Fig. 15 Variations in CO<sub>2</sub> concentration with time under different CO<sub>2</sub> release rates.  
Figure 15 plots the curves of the average CO<sub>2</sub> concentration in the MRC varying with time

481  
482  
483



484 under different CO<sub>2</sub> release rates. It can be observed that, with the increase in the CO<sub>2</sub> release rate,  
 485 the time to reach the steady state decreased but the stable CO<sub>2</sub> concentration increased. The time to  
 486 reach the steady state was approximately 3 h when the CO<sub>2</sub> release rate was 0.3 L/min per person  
 487 and approximately 2 h when the CO<sub>2</sub> release rate reached 0.5 L/min per person. The stable CO<sub>2</sub>  
 488 concentrations for the five different CO<sub>2</sub> release rates of 0.3, 0.35, 0.4, 0.45, and 0.5 L/min per  
 489 person were 0.322%, 0.363%, 0.404%, 0.445%, and 0.483%, respectively.

#### 490 4.4 Discussion

##### 491 4.4.1 Prediction of CO<sub>2</sub> concentration in MRC

492 According to the mass conservation law, for a type of harmful gas in an MRC under ventilation,  
 493 the balanced equation can be expressed as

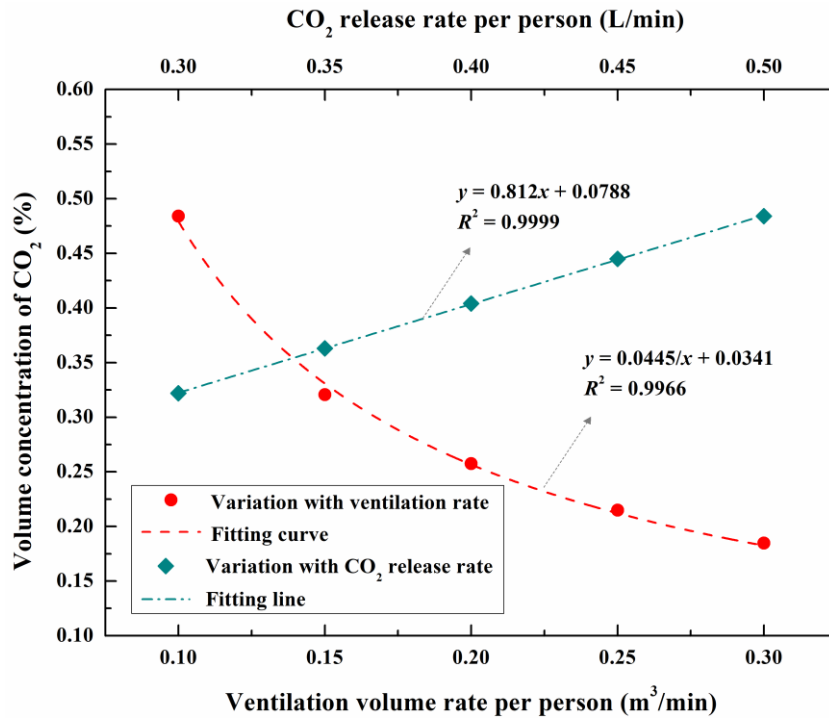
$$494 \quad V[C(\tau + \Delta\tau) - C(\tau)] = Nv\Delta\tau - G\Delta\tau[C_{\text{out}}(\tau) - C_{\text{in}}(\tau)]. \quad (7)$$

495 As the air supplied to the MRC is fresh air,  $C_{\text{in}}(\tau)$  is a constant. It can be assumed that the  
 496 concentration of harmful gases in an MRC is uniformly distributed, and the concentration of a type  
 497 of harmful gas at the outflows is equal to that in the indoor air; that is,  $C_{\text{out}}(\tau) = C(\tau)$ . By solving the  
 498 differential Eq. (7), the concentration of a harmful gas in the MRC can be expressed as:

$$499 \quad C(\tau) = \frac{Nv}{G} + C_{\text{in}} - \left( \frac{Nv}{G} + C_{\text{in}} - C_0 \right) e^{-\frac{G}{V}\tau}. \quad (8)$$

500 According to Eq. (8), it can be predicted that the harmful gas concentration in the MRC will  
 501 gradually tend towards a stable value as time increases, as follows:

$$502 \quad C_{\text{stable}} = \lim_{\tau \rightarrow +\infty} C(\tau) = \frac{Nv}{G} + C_{\text{in}}. \quad (9)$$



503  
 504 Fig. 16 Variations in CO<sub>2</sub> concentration with ventilation volume rate and CO<sub>2</sub> release rate.  
 505

506 Figure 16 plots the stable average CO<sub>2</sub> concentration varying with the ventilation volume rate

507 and CO<sub>2</sub> release rate, according to the above numerical results. It can be observed that the stable  
 508 average CO<sub>2</sub> concentration in the MRC decreased monotonically with the increase in the  
 509 ventilation volume rate when the CO<sub>2</sub> release rate was 0.5 L/min per person; it was inversely  
 510 proportional to the ventilation volume rate. The fitting formula was  $y = 0.0445/x + 0.0341$  with  $R^2$   
 511 = 0.9996. Moreover, it can easily be found that the stable average CO<sub>2</sub> concentration in the MRC  
 512 increased linearly with the increase in the CO<sub>2</sub> release rate when the ventilation volume rate was  
 513 0.1 m<sup>3</sup>/min per person. The fitting formula was  $y = 0.812x + 0.0788$  with  $R^2 = 0.9999$ .

514 According to the numerical results, the relationship between the CO<sub>2</sub> concentration in an  
 515 MRC and the ventilation volume rate as well as the CO<sub>2</sub> release rate is the same as that in Eq. (9).  
 516 The ventilation volume rate and CO<sub>2</sub> release rate are two relatively independent variables.  
 517 Therefore, the expression of the stable CO<sub>2</sub> concentration in the MRC can be assumed as

$$518 \quad C_{\text{stable}} = a \frac{Nv}{G} + C_{\text{in}} . \quad (10)$$

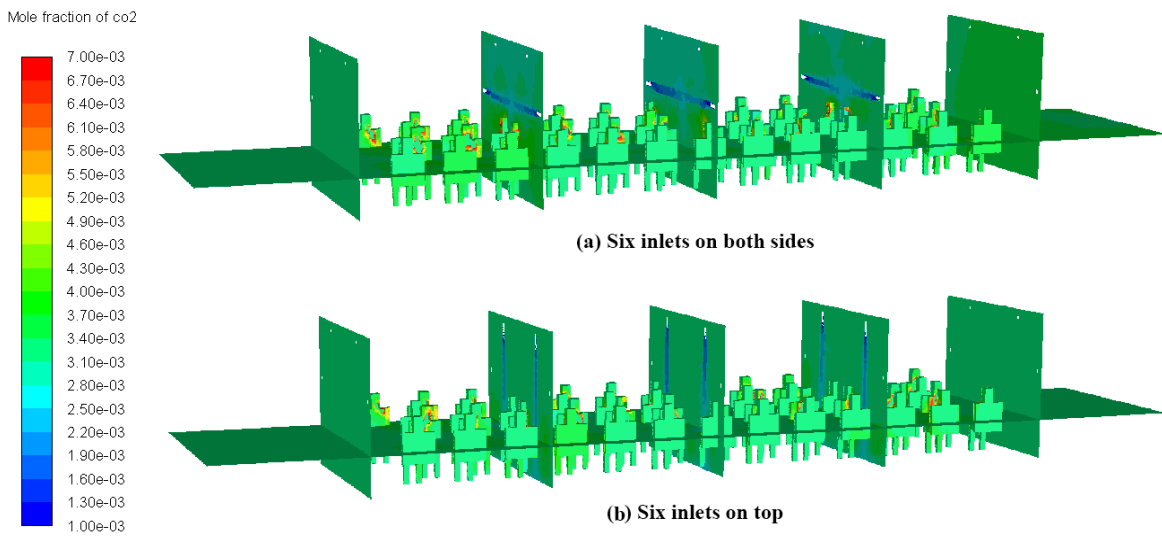
519 By introducing the ventilation volume rates and CO<sub>2</sub> release rates, as well as the corresponding  
 520 stable CO<sub>2</sub> concentration values, in the above analyses into Eq. (10), the optimal solution for  $a$   
 521 can be obtained as 0.92. Thus, the stable CO<sub>2</sub> concentration in the MRC can be calculated as

$$522 \quad C_{\text{stable}} = 0.92 \frac{Nv}{G} + C_{\text{in}} . \quad (11)$$

523 By comparing Eqs. (11) and (9), it can be concluded that, by optimising the ventilation system  
 524 in the MRC, the stable CO<sub>2</sub> concentration will be lower than the theoretical value, which indicates  
 525 that improving the ventilation system in an MRC is of significant value for controlling the air  
 526 quality.

#### 527 4.4.2 Variations in CO<sub>2</sub> concentration in practical MRC

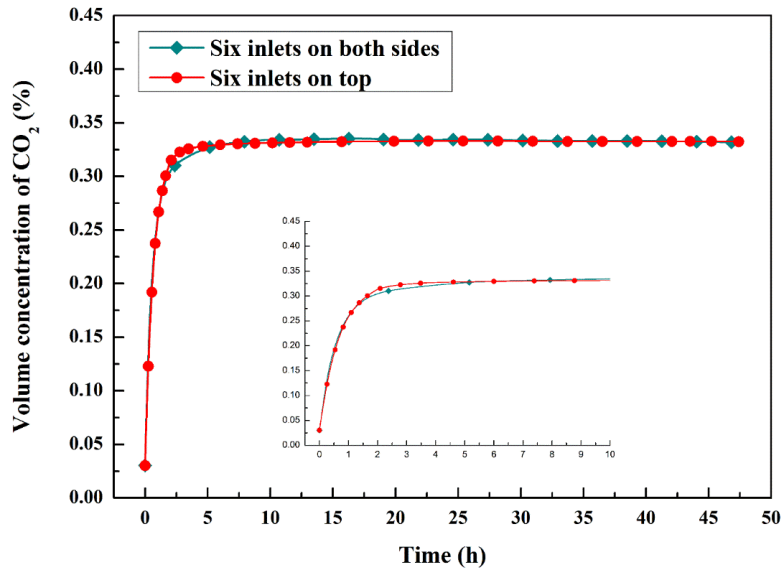
528 The average CO<sub>2</sub> release rate has been measured as approximately 0.34 L/min per person when  
 529 people sit in the MRC, according to an experiment of 50 men residing in an MRC for over 8 h  
 530 (Zhang, 2013). To simulate the variations in the CO<sub>2</sub> concentration in the MRC in a practical  
 531 refuge process, a typical case study was considered, in which the CO<sub>2</sub> release rate was 0.34 L/min,  
 532 the initial CO<sub>2</sub> concentration was 0.03%, and the ventilation volume rate was 0.1 m<sup>3</sup>/min per  
 533 person.



534 Fig. 17 Distribution of CO<sub>2</sub> concentration in MRC under two different ventilation modes.  
 535  
 536

537 Figure 17 illustrates the distribution of the CO<sub>2</sub> concentration under two different ventilation  
 538 modes: six inlets located on both sides and the top, and one outflow located on each end. It can be

539 observed from Fig. 17 that, in the affected zone of the air inlet jet, the CO<sub>2</sub> concentration was  
540 relatively low, at less than 0.3%. Outside the affected zone, the CO<sub>2</sub> concentration distribution was  
541 generally uniform, ranging from 0.3% to 0.45%. The CO<sub>2</sub> concentration distribution obtained in  
542 these two ventilation modes was fairly uniform.



543  
544 Fig. 18 Variations in average CO<sub>2</sub> concentration in MRC under two different ventilation modes.

545  
546 Figure 18 plots the variations in the CO<sub>2</sub> concentration with time under two different  
547 ventilation modes. It can be observed from Fig. 18 that, when the initial CO<sub>2</sub> concentration was  
548 0.03% and the ventilation rate was 0.1 m<sup>3</sup>/min per person, the average CO<sub>2</sub> concentration in the  
549 MRC increased rapidly within 2 h, and then tended towards a relatively stable value of 0.33%. In  
550 combination with the CO<sub>2</sub> concentration distribution in Fig. 17, it can be concluded that a  
551 ventilation volume rate of 0.1 m<sup>3</sup>/min per person can control the average CO<sub>2</sub> concentration below  
552 0.35% in a practical MRC, and the CO<sub>2</sub> concentration in the entire room will be less than 0.5%.  
553 Meanwhile, Figure 18 indicates that the average CO<sub>2</sub> concentration in the ventilation mode of air  
554 supply from the both sides during the initial stage was slightly lower than that in the ventilation  
555 mode of air supply from the top. However, the difference was not significant, and the value of the  
556 CO<sub>2</sub> concentration difference was less than 0.01%. After reaching the stable state, the average CO<sub>2</sub>  
557 concentration in the MRC under the two modes was almost the same.

#### 558 4.4.3 General ventilation design for common MRC

559 According to the above analysis, although people's safety may not be threatened when they  
560 are exposed to an environment with a CO<sub>2</sub> concentration of 1% or more, their cognitive ability  
561 may be affected, which in turn affects their judging abilities in a disaster area. Therefore, it is  
562 recommended that the CO<sub>2</sub> concentration of an MRC be limited to 0.5%. The layout of the air  
563 inlets and outflows in the MRC plays an important role in controlling the air quality. The air inlets  
564 should be arranged in the PCA of the MRC and evenly located on both sides or at the top. The  
565 outflows should be located at both ends of the MRC. When placing the inlets at the top, it is  
566 preferable to place the outlets at the upper parts of both side walls.

567 The CO<sub>2</sub> concentration in an MRC can be controlled below 0.5% when the ventilation volume  
568 rate reaches 0.1 m<sup>3</sup>/min per person, which can meet the oxygen supply and air-purification  
569 requirements of the MRC. According to Zhang et al (2018, 2019), for an MRC built in sandstone,  
570 when the initial rock temperature is less than 20 °C, the temperature requirement can be met  
571 without cooling measures, and when the initial rock temperature reaches 25 °C, the ventilation

572 requirement is  $0.1 \text{ m}^3/\text{min}$  per person. For an MRC with a low temperature in the initial rock, a  
573 ventilation volume rate of  $0.3 \text{ m}^3/\text{min}$  per person will increase the air supply difficulty for an MRC  
574 and incur unnecessary construction costs. Therefore, it is recommended that the ventilation rate for  
575 one person in an MRC is not subjected to a specific restriction, and the ventilation volume rate  
576 required for air quality control in an MRC can be calculated by Eq. (11).

## 577 **5 Conclusions**

578 This work has mainly focused on improving the ventilation system and determining the  
579 relationship between the harmful gas concentration and ventilation volume rate in an MRC. An  
580 experiment on ventilation to dilute the  $\text{CO}_2$  gas in a 50-person MRC was performed. A numerical  
581 model was developed and validated against the obtained experimental data. Several control factors,  
582 such as the layout of the air inlets and outflows, ventilation volume rate, and  $\text{CO}_2$  release rate in  
583 the MRC, were investigated in detail. According to the results of the experiment and numerical  
584 analyses, the following specific conclusions can be drawn:

585 (1) The  $\text{CO}_2$  concentration in an MRC can be controlled below 0.8% using a ventilation rate  
586 of  $0.1 \text{ m}^3/\text{min}$  per person, which can meet the air quality control requirements in the MRC.  
587 However, a poor ventilation system may result in excessive concentrations of harmful gases in a  
588 local space.

589 (2) The layout of the air inlets and outflows plays an important role in the distribution of the  
590  $\text{CO}_2$  concentration in the MRC. Increasing the number of the air inlets is not necessarily conducive  
591 to diluting the  $\text{CO}_2$  concentration. For an effective ventilation scheme, it is suggested that six air  
592 inlets be located evenly on both sides or at the top of the PCA in the MRC, with one outflow  
593 located at each end.

594 (3) The  $\text{CO}_2$  concentration in the MRC decreases inversely with an increase in the ventilation  
595 volume rate and increases linearly with an increase in the  $\text{CO}_2$  release rate. Under the condition  
596 whereby the air inlets and outlets are arranged according to the proposed ventilation system, the  
597 value of the average  $\text{CO}_2$  concentration in the MRC will be slightly lower than the theoretical  
598 result. The average  $\text{CO}_2$  concentration can be controlled below 0.5% with a ventilation rate of  $0.1$   
599  $\text{m}^3/\text{min}$  and less than 0.2% with a ventilation rate of  $0.3 \text{ m}^3/\text{min}$  per person.

600 (4) For a practical refuge process, when the ventilation system is arranged according to the  
601 proposed mode and the ventilation volume rate is  $0.1 \text{ m}^3/\text{min}$  per person, the  $\text{CO}_2$  concentration  
602 can reach the RSS within 2 h, the average  $\text{CO}_2$  concentration in the MRC is approximately 0.35%,  
603 and the  $\text{CO}_2$  concentration will be within 0.5% overall.

604

## 605 **Acknowledgments**

606 The authors would like to thank the financial support from the National Natural Science  
607 Foundation of China (NO. 51678488 and No. 51978575), and the Excellent Doctoral Thesis  
608 Cultivation Project of Southwest Jiaotong University (No. D-YB201703) for the financial support  
609 for this study.

## 610 **References**

611 Bacharoudis, E., Vrachopoulos, M.G., Koukou, M.K., Margaris, D., Filios, A.E., Mavrommatis,  
612 S.A., 2007. [Study of the natural convection phenomena inside a wall solar chimney with one wall  
613 adiabatic and one wall under a heat flux. Applied Thermal Engineering. 27, 2266-2275.](https://doi.org/10.1016/j.applthermaleng.2007.01.021)  
614 <https://doi.org/10.1016/j.applthermaleng.2007.01.021>

615 Bauer, E.R., Kohler, J.L., 2009. Update on refuge alternatives: research, recommendations and  
616 underground deployment. *Mining engineering*. 61, 51-57.  
617 <https://www.cdc.gov/niosh/mining/UserFiles/works/pdfs/uorarr.pdf>

618 Boulet, M., Marcos, B., Dostie, M., Moresoli, C., 2010. CFD modeling of heat transfer and flow  
619 field in a bakery pilot oven. *Journal of food engineering*. 97, 393-402.  
620 <https://doi.org/10.1016/j.jfoodeng.2009.10.034>

621 Charles, D.L., Inoka, E.P., 2012. Evaluation of criteria for the detection of fires in underground  
622 conveyor belt haulage ways. *Fire Safety Journal*. 51, 110-119.  
623 <https://www.ncbi.nlm.nih.gov/pmc/articles/PMC4641047/>

624 Cheung, T., Schiavon, S., Gall, E.T., Jin, M., Nazaroff, W.W., 2017. Longitudinal assessment of  
625 thermal and perceived air quality acceptability in relation to temperature, humidity, and CO<sub>2</sub>  
626 exposure in Singapore. *Building and Environment*. 115, 80-90.  
627 <https://doi.org/10.1016/j.buildenv.2017.01.014>

628 Du, Y., Wang, S., Jin, L.Z., Wang, S., Gai, W.M., 2018. Experimental investigation and theoretical  
629 analysis of the human comfort prediction model in a confined living space. *Applied Thermal  
630 Engineering*. 141, 61-69. <https://doi.org/10.1016/j.applthermaleng.2018.05.057>

631 Du, Y., Gai, W.M., Jin, L.Z., 2018. A novel and green CO<sub>2</sub> adsorbent developed with high  
632 adsorption properties in a coal mine refuge chamber. *Journal of Cleaner Production*. 126,  
633 216-229. <https://doi.org/10.1016/j.jclepro.2017.12.019>

634 Du, Y., Gai, W.M., Jin, L.Z., Sheng, W., 2017. Thermal comfort model analysis and optimization  
635 performance evaluation of a multifunctional ice storage air conditioning system in a confined  
636 mine refuge chamber. *Energy*. 141, 964-974. <https://doi.org/10.1016/j.energy.2017.09.123>

637 Gai, W.M., Deng, Y.F., Du, Y., 2016. Adsorption properties of modified soda lime for carbon  
638 dioxide removal within the closed environment of a coal mine refuge chamber. *Industrial &  
639 Engineering Chemistry Research*. 40, 10794-10802. <https://doi.org/10.1021/acs.iecr.6b02314>

640 Gall, E.T., Cheung, T., Luhung, I., Schiavon, S., Nazaroff, W.W., 2016. Real-time monitoring of  
641 personal exposures to carbon dioxide. *Building and Environment*. 104, 59-67.  
642 <https://doi.org/10.1016/j.buildenv.2016.04.021>

643 Gao, N., Jin, L.Z., Hu, H.H., Huang, X., Zhou, L., Fan, L.Y., 2015. Potassium superoxide oxygen  
644 generation rate and carbon dioxide absorption rate in coal mine refuge chambers. *International  
645 Journal of Mining Science and Technology*. 25, 151-155.  
646 <https://doi.org/10.1016/j.ijmst.2014.12.001>

647 Gao, N., Jin, L.Z., Wang, L., You, F., 2012. Research and application of oxygen supply system in  
648 Changcun coal mine refuge haven. *Journal of China Coal Society*. 37, 1021-1025. (In Chinese)

649 Hansen, R., Ingason, H., 2013. Heat release rate measurements of burning mining vehicles in an  
650 underground mine. *Fire Safety Journal*. 61, 12-25. <https://doi.org/10.1016/j.firesaf.2013.08.009>

651 He, T.M., 2015. Experimental study on purification by pressure ventilation of coal mine refuge  
652 chamber. *Industry and Mine Automation*. 8, 68-71. (In Chinese)

653 He, Z., Wu, Q., Wen, L.J., Fu, G., 2019. A process mining approach to improve emergency rescue  
654 processes of fatal gas explosion accidents in Chinese coal mines. *Safety Science*. 111, 154-166.  
655 <https://doi.org/10.1016/j.ssci.2018.07.006>

656 Jia, S.S., Lai, D.Y., Kang, J., Li, J.Y., Liu, J.J., 2018. Evaluation of relative weights for temperature,  
657 CO<sub>2</sub>, and noise in the aircraft cabin environment. *Building and Environment*. 131, 108-116.  
658 <https://doi.org/10.1016/j.buildenv.2018.01.009>

659 Jia, Y.X., Liu, Y.S., Liu, W.H., Li, Z.Y., 2014. Study on purification characteristic of CO<sub>2</sub> and CO  
660 within closed environment of coal mine refuge chamber. *Separation and Purification Technology*.

661 [130, 65-73. https://doi.org/10.1016/j.seppur.2014.04.014](https://doi.org/10.1016/j.seppur.2014.04.014)

662 Jin, L.Z., 2013. Research and experiment on air pressure volume of refuge chamber. Proceedings of  
663 the 2013 annual conference of the China Occupational Safety and Health Association, Fujian  
664 946-952. (In Chinese)

665 Kajtár, L., Herczeg, L., 2012. Influence of carbon-dioxide concentration on human well-being and  
666 intensity of mental work. *Idojaras (Budapest, 1905)*. 116, 145-169.  
667 [https://www.researchgate.net/publication/231558578\\_Influence\\_of\\_carbon-dioxide\\_concentration  
668 on\\_human\\_well-being\\_and\\_intensity\\_of\\_mental\\_work/link/0912f506c5e44909eb000000/dow  
669 nload](https://www.researchgate.net/publication/231558578_Influence_of_carbon-dioxide_concentration_on_human_well-being_and_intensity_of_mental_work/link/0912f506c5e44909eb000000/download)

670 Li, Y., Yuan, Y.P., Lia, C.F., Han, X., Zhang X.S., 2018. Human responses to high air temperature,  
671 relative humidity and carbon dioxide concentration in underground refuge chamber. *Building and  
672 Environment*. 131, 53-62. <https://doi.org/10.1016/j.buildenv.2017.12.038>

673 Liu, W.W., Zhong, W.D., Wargocki, P., 2017. Performance, acute health symptoms and  
674 physiological responses during exposure to high air temperature and carbon dioxide  
675 concentration. *Building and Environment*. 114, 96-105.  
676 <https://doi.org/10.1016/j.buildenv.2016.12.020>

677 Margolis, K.A., Westerman, C.Y.K., Kowalski-Trakofler, K.M., 2011. Underground mine Refuge  
678 Chamber Expectations Training: Program development and evaluation. *Safety Science*. 49,  
679 522-530. <https://doi.org/10.1016/j.ssci.2010.12.008>

680 Mejías, C., Jiménez, D., Munoz, A., Reyes-Bozo, L., 2014. Clinical response of 20 people in a  
681 mining refuge: study and analysis of functional parameters. *Safety Science*. 63, 204-210.  
682 <https://doi.org/10.1016/j.ssci.2013.11.011>

683 National Coal Mine Safety Administration, 2011. Temporary provisions on the construction and  
684 management of the coal mine underground emergency refuge system. (In Chinese)  
685 [http://www.chinacoal-safety.gov.cn/gk/tzgg/201101/t20110127\\_201854.shtml](http://www.chinacoal-safety.gov.cn/gk/tzgg/201101/t20110127_201854.shtml)

686 National Research Council, 2013. Improving Self-escape From Under-ground Coal Mines.  
687 Washington, DC: The National Academies Press. <https://doi.org/10.17226/18300>.

688 Persily, A., 2015. Challenges in developing ventilation and indoor air quality standards: The story  
689 of ASHRAE Standard 62. *Building and Environment*. 91, 61-69.  
690 <https://doi.org/10.1016/j.buildenv.2015.02.026>

691 Piña-Ortiz, A., Hinojosa, J.F., Maytorena, V.M., 2014. Test of turbulence models for natural  
692 convection in an open cubic tilted cavity. *International Communications in Heat and Mass  
693 Transfer*. 57, 264-273. <https://doi.org/10.1016/j.icheatmasstransfer.2014.08.011>

694 Shao, H., Jiang, S.G., Tao, W.Y., Wu, Z.Y., Zhang, W.Q, Wang, K., 2016. Theoretical and  
695 numerical simulation of critical gas supply of refuge chamber. *International journal of mining  
696 science and technology*. 26, 389-393. <https://doi.org/10.1016/j.ijmst.2016.02.004>

697 Sørensen, D.N., Nielsen, P.V., 2003. Quality control of computational fluid dynamics in indoor  
698 environments. *Indoor Air*. 13, 2-17. <https://doi.org/10.1111/j.1600-0668.2003.00170.x>

699 Trackemas, J.D., Thimons, E.D., Bauer, E.R., Sapko, M.J., Zipf, R. K., et al., 2015. Facilitating the  
700 Use of Built-in-place Refuge Alternatives in Mines. DHHS(NIOSH) Publication No. 2015-114,  
701 RI 9698, Pittsburgh, PA. <https://permanent.access.gpo.gov/gpo58847/2015-114.pdf>

702 Tripathy, D.P., Ala, C.K., 2018. Identification of safety hazards in Indian underground coal mines.  
703 *Journal of Sustainable Mining*. 17, 175-183. <https://doi.org/10.1016/j.jsm.2018.07.005>

704 Vaught, C., Brnich, M.J., Mallett, L.G., 2000. Behavioral and Organizational Dimensions of  
705 Underground Mine Fires. U.S. Department of Health and Human Services, Pittsburgh, PA.  
706 <https://www.cdc.gov/niosh/mining/UserFiles/works/pdfs/ic9450.pdf>

707 Wang, L., Cheng, Y.P., Liu, H.Y., 2014. An analysis of fatal gas accidents in Chinese coal mines.  
708 *Safety Science*. 62, 107-113. <https://doi.org/10.1016/j.ssci.2013.08.010>

709 Wang, L.J., Wang, Y.L., Cao, Q.Y., Li, X.D., Li, J.Q., et al., 2014. A framework for human error risk  
710 analysis of coal mine emergency evacuation in China. *Journal of Loss Prevention in the Process*  
711 *Industries*. 30, 113-123. <https://doi.org/10.1016/j.jlp.2014.05.007>

712 Wang, S., Jin, L.Z., Ou, S.N., Li, Y.G., 2017. Experimental air curtain solution for refuge  
713 alternatives in underground mines. *Tunnelling and Underground Space Technology*. 68, 74-81.  
714 <https://doi.org/10.1016/j.tust.2017.05.024>

715 Wu, T., Lei, C.W., 2015. On numerical modelling of conjugate turbulent natural convection and  
716 radiation in a differentially heated cavity. *International Journal of Heat and Mass Transfer*. 91,  
717 454-466. <https://doi.org/10.1016/j.ijheatmasstransfer.2015.07.113>

718 You, F., Jin, L.Z., Han, H.R., Gao, N., 2012. Research on air supply to refuge chamber. *China*  
719 *Safety Science Journal*. 22, 116-120. (In Chinese)

720 Zhai, Y.C., Li, M.H., Gao, S.R., Yang, L., Zhang, H., Arens, E., Gao, Y.F., 2018. Indirect  
721 calorimetry on the metabolic rate of sitting, standing and walking office activities. *Building and*  
722 *Environment*. 145, 77-84. <https://doi.org/10.1016/j.buildenv.2018.09.011>

723 Zhang, S.Z., Wu, Z.Z., Zhang, R., Kang, J.N., 2012. Dynamic numerical simulation of coal mine  
724 fire for escape capsule installation. *Safety Science*. 50, 600-606.  
725 <https://doi.org/10.1016/j.ssci.2011.09.007>

726 Zhang, X.J., Wargocki, P., Lian, Z.W., 2016. Human responses to carbon dioxide, a follow-up  
727 study at recommended exposure limits in non-industrial environments. *Building and environment*  
728 100, 162-171. <https://doi.org/10.1016/j.buildenv.2016.02.014>

729 Zhang, Z.J., 2013. Theoretical study on cooling technology of a coal mine refuge chamber. Master  
730 Degree Dissertation of Chongqing Research Institute of Coal Research Institute. (In Chinese)

731 Zhang, Z.J., Yuan, Y.P., Wang, K.Q., Gao, X.K., Cao, X.L., 2016. Experimental investigation on  
732 influencing factors of air curtain systems barrier efficiency for mine refuge chamber. *Process*  
733 *Safety and Environmental Protection*. 102, 534-546. <https://doi.org/10.1016/j.psep.2016.05.008>

734 Zhang, Z.J., Yuan, Y.P., Wang, K.Q., 2017. Effects of number and layout of air purification devices  
735 in mine refuge chamber. *Process Safety and Environmental Protection*. 105, 338-347.  
736 <https://doi.org/10.1016/j.psep.2016.11.023>

737 Zhang, Z.J., Day, R., Wang, K.Q., Wu, H.W., Yuan, Y.P., 2018. Thermal performance analysis of an  
738 underground closed chamber with human body heat sources under natural convection. *Applied*  
739 *Thermal Engineering*. 145, 453-463. <https://doi.org/10.1016/j.applthermaleng.2018.09.068>

740 Zhang, Z.J., Wu, H.W., Wang, K.Q., Day, R., Yuan, Y.P., 2019. Thermal performance of a mine  
741 refuge chamber with human body heat sources under ventilation, *Applied Thermal Engineering*  
742 162, 114243. <https://doi.org/10.1016/j.applthermaleng.2019.114243>

743 Zhu, Y.F., Wang, D.M., Shao, Z.L., Xu, C.H., Zhu, X.L., Qia, X.Y., Liu, F.M., 2019. A statistical  
744 analysis of coalmine fires and explosions in China. *Process Safety and Environmental Protection*.  
745 121, 357-366. <https://doi.org/10.1016/j.psep.2018.11.013>