

Evaluation of the production of second generation ethanol by coculture of *Saccharomyces cerevisiae* and *Pachysolen tannophilus* immobilized in sodium alginate

Avaliação da produção de etanol de segunda geração por co-cultura de Saccharomyces cerevisiae e Pachysolen tannophilus imobilizados em alginato de sódio

DOI:10.34117/bjdv8n9-220

Recebimento dos originais: 23/08/2022 Aceitação para publicação: 20/09/2022

Marcello Lima Bertuci

Mestre em Engenharia e Ciência de Alimentos Institution: Universidade Estadual Paulista "Júlio de Mesquita Filho" (UNESP) Address: Rua Cristóvão Colombo, 2265, Jardim Nazareth, CEP: 15054000, São José do Rio Preto – SP, Brasil E-mail: marcellobertuci@gmail.com

Mariane Daniella da Silva

Mestra em Engenharia e Ciência de Alimentos Institution: Universidade Estadual Paulista "Júlio de Mesquita Filho" (UNESP) Address: Rua Cristóvão Colombo, 2265, Jardim Nazareth, CEP: 15054000, São José do Rio Preto – SP, Brasil E-mail: marianedaniella@hotmail.com

João Pedro Cano

Engenheiro de Alimentos Institution: Universidade Estadual Paulista "Júlio de Mesquita Filho" (UNESP) Address: Rua Cristóvão Colombo, 2265, Jardim Nazareth, CEP: 15054000, São José do Rio Preto – SP, Brasil E-mail: joaopedro.cano@gmail.com

Crispin Humberto Garcia Cruz

Doutor em Ciência de Alimentos Institution: Universidade Estadual Paulista "Júlio de Mesquita Filho" (UNESP) Address: Rua Cristóvão Colombo, 2265, Jardim Nazareth, CEP: 15054000, São José do Rio Preto – SP, Brasil E-mail: crispinhumberto@gmail.com

ABSTRACT

The production of an alternative form of fuel that replaces fossil fuels has been increasingly studied due to the environmental impacts generated by its excessive use, as well as the depletion of these fossil energy sources. Ethanol obtained from the crushing of sugar cane has been used as a substitute for these fuels, mainly in the automotive area. However, alternative sources are being studied to produce the so called second generation bioethanol. This would avoid competition for food producing agricultural areas and agroindustrial waste is a great source for obtaining it. In general, these residues are not always completely reused and are disposed of inappropriately in the environment,



becoming contaminants. Therefore, the use of agroindustrial waste can become a renewable source of energy, in addition to reducing environmental impacts. The objective of this work is to produce second generation bioethanol as an alternative to the one currently used, using the rice husk hydrolyzate by the consortium formed by Saccharomyces cerevisiae and Pachysolen tannophilus. For this, an acid hydrolysis was performed with 2% sulfuric acid during 10 minutes of heating in an autoclave, after which the hydrolyzate was detoxified with the use of activated carbon. The crude and detoxified hydrolysates were used as a substrate for the fermentation medium with an initial concentration of 50 mg/mL of reducing sugars. The fermentation process with the use of both yeasts in the crude hydrolyzate medium, in the detoxified medium and in a synthetic medium composed of glucose, was carried out for 24 h, 30° C, 0 rpm and pH 6.5. The best results for the ethanol production of Saccharomyces cerevisiae was the synthetic medium with 20.6 mg/mL. For the yeast Pachysolen tannophilus, its highest production was in a synthetic medium with 11.67 mg/mL. The intercropping of the two yeasts proved to be efficient with a greater ethanol production reaching 21.5 mg/mL, the hydrolyzed and detoxified media showed great potential for ethanol production both in intercropping and in monoculture.

Keywords: agroindustrial waste, Bioethanol, consortium, rice husk, acid Hydrolysis.

RESUMO

A produção de uma forma alternativa de combustível que substitui os combustíveis fósseis tem sido cada vez mais estudada devido aos impactos ambientais gerados por seu uso excessivo, bem como pelo esgotamento dessas fontes de energia fóssil. O etanol obtido a partir da moagem da cana de acúcar tem sido utilizado como substituto desses combustíveis, principalmente na área automotiva. Entretanto, fontes alternativas estão sendo estudadas para produzir o chamado bioetanol de segunda geração. Isto evitaria a competição por áreas agrícolas produtoras de alimentos e os resíduos agroindustriais são uma grande fonte para sua obtenção. Em geral, estes resíduos nem sempre são completamente reutilizados e são descartados de forma inadequada no meio ambiente, tornando-se contaminantes. Portanto, o uso de resíduos agroindustriais pode se tornar uma fonte renovável de energia, além de reduzir os impactos ambientais. O objetivo deste trabalho é produzir bioetanol de segunda geração como alternativa ao atualmente utilizado, utilizando o hidrolisado de casca de arroz pelo consórcio formado pela Saccharomyces cerevisiae e Pachysolen tannophilus. Para isto, foi realizada uma hidrólise ácida com ácido sulfúrico a 2% durante 10 minutos de aquecimento em autoclave, após o que o hidrolisado foi desintoxicado com o uso de carvão ativado. Os hidrolisados brutos e desintoxicados foram utilizados como substrato para o meio de fermentação com uma concentração inicial de 50 mg/mL de açúcares redutores. O processo de fermentação com o uso de ambas as leveduras no meio hidrolisado bruto, no meio desintoxicado e em um meio sintético composto de glicose, foi realizado por 24 h, 30° C, 0 rpm e pH 6,5. Os melhores resultados para a produção de etanol de Saccharomyces cerevisiae foi o meio sintético com 20,6 mg/mL. Para o fermento Pachysolen tannophilus, sua maior produção foi em um meio sintético com 11,67 mg/mL. O intercruzamento das duas leveduras mostrou-se eficiente com uma maior produção de etanol atingindo 21,5 mg/mL, o meio hidrolisado e desintoxicado mostrou um grande potencial para a produção de etanol tanto no intercruzamento como na monocultura.

Palavras-chave: resíduos agroindustriais, Bioetanol, consórcio, casca de arroz, Hidrólise ácida.



1 INTRODUCTION

The concern to reduce the emission of gases related to the greenhouse effect, mainly derived from automobiles, has motivated the search for substitutes for fossil fuels that cause less environmental impact. Because of this, obtaining biofuels from alternative sources has become a major focus of study. Thus, they are an alternative to reduce the use of petroleum-derived fuels, whose supply and price depend on the moods of the producing countries, in addition to generating various environmental damages. (Dafnomilis et al. 2017; Küüt et al. 2017)

Biofuels are fuels produced from organic matter. One of the most intresting and outstanding is the production of second generation ethanol, which is obtained from agroindustrial waste. Several countries consider the use of biomass for ethanol production, making it an alternative with the capacity to increase economic development, in addition to reducing the emission of greenhouse gases, having a reduction when compared to emissions caused by gasoline. (Wei et al. 2014)

The exploration of alternatives to increase the production of biofuels has become the focus of biotechnological development, due to the large amount of resources required for production, requiring, in addition to technological advances, management models that guarantee technical, economic and environmental aspects. (Maldhure; Ehle 2013)

For the production of this bioethanol, which uses lignocellulosic biomass as a base, chemical (using acid and base), physical and biological (use of enzymes) processes can be used to break down the cellulose and hemicellulose present in the waste resulting in the release of sugars necessary to carry out the fermentative processes aimed at ethanol production (Kumari, Singh. 2018). However, for the consumption of these hexoses and pentoses, a microorganism co-culture process is necessary, due to the fact that there is no microorganism that can consume these two fermentable sugars. Therefore, co-culture is a viable option that improves fermentation yield, generating greater ethanol production (Xue et al. 2020).

Saccharomyces cerevisiae is the most widely used yeast for fermentation processes due to its ability to consume hexoses and is generally used for ethanol production (Behera et al. 2010).

Pachysolen tannophilus was the first yeast recognized for its ability to consume pentoses, due to this ability it can consume the sugars from biomass, which were not consumed by *S. cerevisiae* in fermentation, without generating competition for the



substrate. Thus, the relevance of co-culture benefits the process of obtaining ethanol. (Saleh et al. 2014)

Cell immobilization is a process used to increase productivity in fermentation processes, separating cells from microorganisms in continuous and long-term processes. The fermentative process using immobilized microorganisms is more efficient to take advantage of the potential of yeasts for consumption of substrates and tolerance to the medium. This technique improves production and yield for obtaining ethanol (Gandu et al. 2020).

Within this context, the objective of this work was to produce ethanol by individual cultures and co- culture of *Saccharomyces cerevisiae* and *Pachysolen tannophilus* immobilized in sodium alginate, using rice husk acid hydrolyzate.

2 MATERIALS AND METHODS

The experimental part of the work was developed at the Biopolymers Laboratory of the Department of Food Engineering and Technology, Institute of Biosciences, Letters and Exact Sciences - IBILCE / UNESP, in São José do Rio Preto / SP.

2.1 INDUSTRIAL RICE RESIDUE

The rice husks were acquired from establishments that commercialize rice based products, situated at Municipal Dam of São José do Rio Preto, São Paulo, Brazil (Latitude: 20°49'10.99" S and Longitude: 49°22'45.98" W).

2.2 PROCESSING OF RICE HUSKS

The industrial residue of rice consisting of husks was dried in an oven with air circulation, until it reached an approximate humidity of 12%. Subsequently, the shells were ground to increase the contact surface and stored in plastic containers. The particle size was homogenized at <0.64 mm using a Produtest brand atomizer.

2.3 RICE HUSK CHEMICAL HYDROLYSIS AND DETOXIFICATION

The hydrolysis of rice husks took place in Erlenmeyer bottles of 250ml, at a scale of 10/100 (w/v) of rice husk in sulfuric acid (H₂SO₄) with concentration (2 %). The bottles were submitted to heating in autoclave at 121 °C/1.1 atm. during 10 min. Afterwards, the pH of the hydrolyzate material was neutralized (pH 7) with NaOH (50%) and filtered using Whatman n°1 filter for the separation of residual solids, which were discarded. The



obtained supernatant was used as substract for the production of ethanol by two methods: raw hydrolyzate (RH), before being submitted into detoxification and detoxified raw hydrolyzate, that had already been through the detoxification stage with activated charcoal, for the removal of inhibitory agents from the microbial fermentation according to the methodology described by Mussatto; Roberto (2004).

Before carrying out the fermentation process with the RH and DRH the total sugar values were measured by the phenol-sulfuric method (Dubois et al. 1956) and reducing sugars total values by the copperarsenate method (Nelson 1944) and (Somogyi 1952). Also, the total phenolic compounds values were analyzed by the method described by Folin-Ciocalteau modified by Chaovanalikit; Wrolstad (2004).

2.4 MICROORGANISM, MAINTENANCE, INOCULUM PREPARATION AND FERMENTATION MEDIUM

The samples of *Saccharomyces cerevisiae* ATCC 26602 yeast were kindly given by the Department of Chemistry Engineering of University of Coimbra – Portugal and *Pachysolen tannophilus* CCT 1891, acquired from the Collection of Tropical Cultures (CCT) of the André Tosello Foundation - Research and Technology of Campinas, SP. The yeasts was stored YM Medium composed by glucose (10 g.L⁻¹), peptone (5 g.L⁻¹), yeast extract (3 g.L⁻¹), malt extract (3 g.L⁻¹) and agar (20 g.L⁻¹) at pH 5.0, after being cultivated for 24h a 30°C. The microorganism strains were stored under freezing conditions (4°C), while periodically being reactivated to maintain its viability.

The inoculum was prepared by the addition of microorganisms previously cultivated in YM broth in Erlenmeyer bottles of 250ml containing 100ml of YM broth (pH 5.0). The inoculum was standardized by spectrophotometry with an absorbance of 0.6 and wave length of 600nm.

The production of ethanol was performed in culture media (pH 7) composed by yeast extract (5 g.L⁻¹); KH₂PO₄ (1 g.L⁻¹); MgSO₄.7H₂O (1 g.L⁻¹), (NH4)₂SO₄ (1 g.L⁻¹) and a carbon supply (glucose, raw hydrolyzate or raw detoxified hydrolyzate).



2.5 CELL IMMOBILIZATION - CELL ENTRAPMENT IN ALGINATE GEL

The cells of *S. cerevisiae* ATCC 26602 and *P. tannophilus* CCT 1891 under the conditions described in item 2.4 were added in a solution of sodium alginate 2.0% (Dynamic), previously prepared and sterilized, under agitation for 15 minutes. Afterwards, it was dripped (using a peristaltic pump) in a 3.0% calcium chloride solution (Sigma) under agitation, obtaining calcium alginate spheres containing the microorganism approximately 2.5 mm in diameter and weight of 0.06 g; kept for 30 minutes under agitation, after this period, 3 g of these spheres were added in 125 ml Erlenmeyers containing 50 ml of the culture media used for the fermentation.

2.6 EFFECT OF CULTURE MEDIUM ON SUBSTRATE (CARBON SOURCE) CONSUMPTION AND ETHANOL BIOSYNTHESIS BY *S. CEREVISIAE* AND *P. TANNOPHILUS*

The culture media containing the crude hydrolyzate, the detoxified hydrolyzate and the synthetic medium with glucose was carried out in a D.B.O incubator without shaking (0 rpm) and constant temperature of 30 °C. The initial pH of the culture media was 6.5, with the initial concentration of reducing sugars of 50 mg / mL.

The fermentation was evaluated using the monoculture of S. cererevisiae and *P. tannophilus* and by the co-culture of both yeasts, immobilized in sodium alginate spheres.

The fermentation occurred during 24 h, where, samples were taken every 4 h at the phase-lag and at every 2 h at the phase-log for the purpose of evaluating the ethanol production, cellular biomass values, pH change, sugar values (reducing) and productivity calculation.

For analysis by the yeast consortium, it was performed with the inoculation of the yeast *S. cerevisiae* in the first 12 h of the fermentation, after, *P. tannophilus* was removed and inoculated.

2.7 ANALYTICAL METHODS

The final pH values were determined in the fermentation broth using the potentiometer Digmed pHmeter model DM20.

The cellular concentration was determined by turbidimetry in spectrophotometer Biochrom, model Libra S22.

The ethanol concentration values were determined by gas chromatography (GC) in the fermented broth free of cells, using a GC Thermo Scientific Model Series Focus



TR-WAX column HP-FFAP (25 m x 0.2 mm x 0.3 μ m) and flame ionization detector (FID). The oven temperature was kept at 70 °C (during the whole isothermal run), running time of 5min, the injector temperature was kept at 230°C; detector temperature at 270 °C and injection of 200 μ l of sample vapor. The samples were left at a water bath of 40°C until reaching its point of equilibrium.

The values of total ethanol productivity, expressed in mass of the product formed per unit of time and per unit of volume (g.L-1.h-1), will be obtained through Equation 1.

$$P = (Pf - Po)$$
(Equation 1)
(tf - to)

Where: P = productivity (g.L-1.h-1); Po = initial product mass (g.L-1); Pf = mass of final product (g.L-1); tf = final fermentation time (h); to = initial fermentation time (h).

3 RESULTS AND DISCUSSIONS

3.1 FERMENTATION WITH YEAST SACCHAROMYCES CEREVISIAE

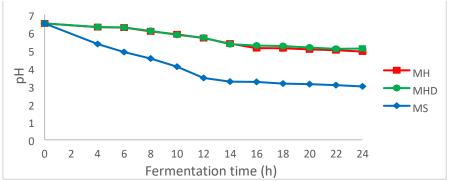
The fermentation was carried out in synthetic culture medium (MS), crude hydrolyzed medium (MH) and detoxified hydrolyzed medium (MHD), with a constant temperature of 30 °C; with initial pH of 6.5 and initial substrate (glucose) 5% (w / v); where aliquots were removed every 2 hours, after the first 4 hours of fermentation for 24 hours.

3.1.1 Change in pH in the culture medium

The initial pH of the culture media for fermentation containing, synthetic medium (MS), rice husk hydrolyzate (MH) and detoxified hydrolyzate (MHD) was 6.5. Figure 1 shows the performance in relation to pH during the alcoholic fermentation process during the 24 hour period. It becomes possible to see the reduction that occurred sharply during the first 12 hours, followed by a less mild reduction during the following hours. The final pH values after fermentation were 4.93 for hydrolyzed medium, 5.1 for detoxified and 2.93 for synthetic medium. It is possible to observe a small difference in relation to hydrolysates, however, considerable with the synthetic medium.



Fig 1. Kinetics of the fermentations pH by *Saccharomyces cerevisiae* during the 24 hours period determined in the 2-hour interval in the MS, MH and MHD media.

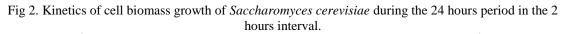


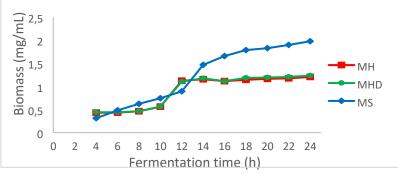
The performance of pH during the fermentation process acts directly on its yield, considering that these changes can influence the metabolism of the micro-organism as well as yield in the process. PH can play an important role in the face of fermentative inhibition, and consequently lead to changes in cellular metabolism. During ethanol production, the pH of the medium tends to decrease due to the consumption of substrates and byproduct formations, as it gradually decreases during this process (Yoshikawa et al. 2015).

3.1.1.1 Cell Growth (Biomass)

The inoculum of the yeast *S. cerevisiae* in the fermentation media was used to evaluate cell growth. Figure 2 shows the results obtained in the fermentations during the 24 hours period. This Figure shows that with the passage of time during the fermentation process, biomass increased for the 3 media used during this process (MH, MHD and MS). It would be possible to note that the synthetic medium (DM) has a higher biomass production compared to hydrolysates, such as an addition during the 12-hour period, it even produces 1.97 mg / mL at the end of the 24-hour process. For the hydrolyzed and detoxified media, they were very similar for the production of biomass during the fermentation process, with a greater production during the period of 10 hours and ending with a production of 1.2 mg / mL and 1.27 mg / mL respectively.







In a study on separation methods to distinguish the cell growth phases of *Saccharomyces cerevisiae*, researchers Lainiotiel al. (2010) determined the highest biomass production in the first 10 hours, following a reduction thereafter, in a culture medium with initial pH 5, testing temperatures of 25 and 30 °C. What differs from the present study, in which yeast showed the highest production after 12 h, with successive growth until the end of fermentation (24 h) at 30 °C.

3.1.1.2 Production of second generation bioethanol by the yeast *Saccharomyces* cerevisiae

Figure 3 shows the results of ethanol production during the 24-hour fermentation, with intervals of 2 hours from the 4-hour fermentation.

The best results obtained for ethanol production were 20.6 mg / mL for synthetic medium in 24 hours, 18.67 mg / mL for hydrolyzed medium in 22 hours and 19.17 mg / mL for detoxified in 22 hours. The synthetic medium showed stable production until the 24 hours period while the MH and MHD showed a decrease after the 22 hours period.

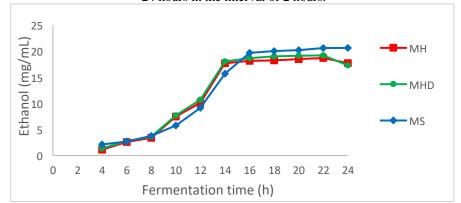


Fig 3. Kinetics of ethanol production in fermentation by *Saccharomyces cerevisiae* during the period of 24 hours in the interval of 2 hours.



As cell growth increased, ethanol production also increased, especially after the 12-hour fermentation period. As noted earlier in the 4 to 10 hours fermentation period, cell growth is still low, which corroborates the low ethanol production.

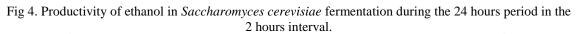
The authors Mishra et al. (2016) used strains of *Saccharomyces cerevisiae* immobilized with sodium alginate, in a concentrated bed reactor, during fermentations of 11 batch cycles, where they evaluated the consumption of the substrate (glucose) and the temperature behavior of 30 °C, 35 °C and 40 °C in relation to ethanol production in these media. These researchers obtained 90 mg / mL of ethanol at the end of the batches, where there was a total consumption of the substrate. They emphasized the efficiency of ethanol production at a temperature of 30 °C; the temperature with the lowest efficiency was 40 °C, the fermentation showed little consumption of glucose and ethanol production of 10 mg / mL. It is possible to verify that a high ethanol content was obtained in relation to the content obtained in this research, where 18.69 mg / mL of ethanol by MH and 19.25 mg / mL by MHD when using *S. cerevisiae*. Where the temperature used (30 °C) corroborates with the authors mentioned.

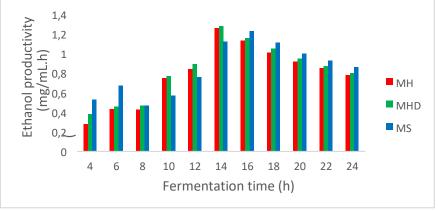
3.1.1.3 Productivity

The ethanol productivity for the media used in this study is shown in Figure 4, where it is possible to verify the highest response for productivity was 1.28 mg / mL.h for the detoxified medium in 14 hours, followed by the hydrolyzed medium presented 1, 26 mg / mL.h in 14 hours and 1.23 mg / mL.h for the synthetic medium in 16 hours. Such behavior was a reflection of the production of ethanol in hydrolyzed and detoxified media, which showed a higher production until the 14-hour fermentation period, followed by a stability in production. The behavior of the synthetic medium, in turn, showed less ethanol production in the first 16 hours of fermentation, however, after this period it was more efficient for production, following a constant increase until the end of the analysis in 24 hours.

As the fermentation proceeds, the yield decreases after 14 hours (MH, MHD) and in 16 hours (MS) this can occur due to the consumption of the substrate for growth in addition to the formation of other byproducts from the process.







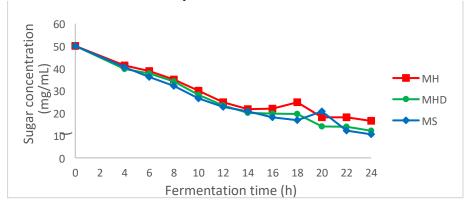
The researchers Sakomoto et al. (2012) used different combinations of yeast in 5 mg / mL of initial substrate in the fermentation to produce ethanol from rice husk, after 72 hours of fermentation obtained an amount of 24.4 mg / mL of ethanol consuming 93% of the sugars present in the medium thus presenting a productivity of 0.56 mg / mL.h for the yeast *Saccharomyces cerevisiae*. In this work, the highest yield content obtained was 1.26 mg / mL.h in 14 h of fermentation for the hydrolyzed medium, 1.28 mg / mL.h for the detoxified medium in 14 and 1.23 mg / mL.h in 16 h for the synthetic medium.

3.1.1.4 Reducing sugars in fermentation

The reducing sugars present in the fermentation media, which were consumed by the yeast to grow and produce ethanol. Where the initial concentration was 50 g / L for all media. Figure 5 shows during the 24 hours of the process, in which the synthetic medium obtained a higher consumption of sugars, leaving 10.5 g / L, for the hydrolyzate 16.5 g / L and for the detoxified 12.1 g / L.



Fig 5. Sugar consumption present in the fermentation media used by *Saccharomyces cerevisiae* during the 24 hours period in the 2 hours interval.



When using *Saccharomyces cerevisiae* in their study Boboescu et al. (2017) aimed to produce second generation ethanol using hemicellulose hydrolysates, obtaining a production of 74.27 mg / mL of ethanol with an initial concentration of 65.17 mg / mL of sugars. The authors emphasize the effectiveness of the strategy used in this process, due to the presence of glucose and xylose and the consumption of both during the process within 48 hours. Consumption took place in the first 24 hours, becoming stable after this period.

3.1.2 Fermentation with the yeast Pachysolen tannophilus

The fermentation was carried out in synthetic culture medium (MS), crude hydrolyzate (MH) and detoxified hydrolyzate (MHD). The established parameters were a temperature of 30 °C; initial pH of 6.5; and glucose substrates 5% (w / v); where rates were removed at times 4, 6, 8, 10, 12, 14, 16, 18, 20, 22 and 24 hours. With the best parameters determined, they were carried out for synthetic, hydrolyzed and detoxified hydrolyzed medium, during the 24 hours period.

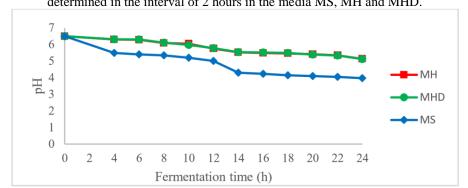
3.1.2.1 Change in pH in the fermentation medium

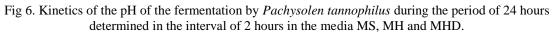
The media used for fermentation were synthetic media containing glucose, since yeast has the capacity to produce ethanol in media containing pentoses as a substrate, and these sugars are present in the rice husk residues; rice husk hydrolyzate and detoxified hydrolyzate all with an initial pH of 6.5, as well as the pH used for the media in the fermentation of *Saccharomyces cerevisiae*.

Figure 6 represents the values obtained in relation to the pH change of the media during the 24 hours fermentation process to obtain ethanol by the yeast *Pachysolen*



tannophilus. The pH values in the hydrolyzed and detoxified medium were very similar during the fermentation, showing a small reduction, reaching the values of 5.13 and 5.12 respectively, values higher than that of the synthetic medium 3.97 at the end of 24 hours of fermentation.

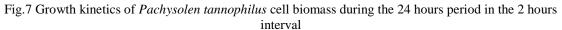


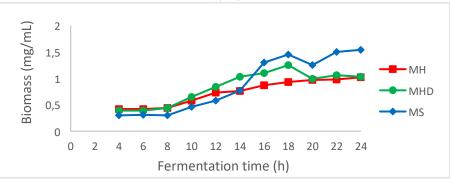


3.1.2.2 Cell growth (Biomass) of the yeast Pachysolen tannophilus

The yeast *P. tannophilus* immobilized in sodium alginate was evaluated in its growth during fermentation in Figure 7, which shows the results obtained during the 24 hours period. The first 4 hours were not evaluated because they are the adaptation and growth period of the yeast in the media, in this figure it is possible to observe the cellular growth of the synthetic, hydrolyzed and detoxified media. The growth becomes accentuated after 8 hours for all media, a period which was necessary for the yeast to adapt to the medium and to grow. The synthetic medium was less effective for the production of biomass during the first 14 hours of fermentation, when compared to the other two media in the study. The detoxified hydrolyzed medium was more effective in relation to the hydrolyzate until the period of 18 hours, after which there was a drop in which both presented similar results at the end of 24 hours, of 1.02 mg / mL and 10.3 mg / mL respectively the synthetic medium had a biomass production of 1.54 mg / mL for the same period.





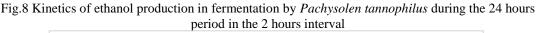


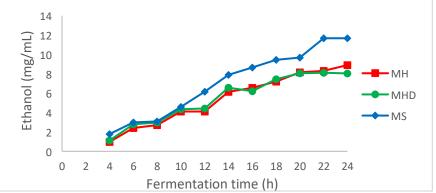
In their work, Romero et al. (2007) studied the growth of *P. tannophilus* over 24 hours, using acid hydrolysates from olive residues (branches / pruning), the authors found a marked growth during 24 hours in the hydrolyzed medium with 4 N sulfuric acid, in comparison with one of 1 N, reaching values of 1.62 mg / mL of cell growth and 0.75 mg / mL for 1 N for 20 hours. These results are similar to the results presented in this research, which were from 0.43 mg / mL to 1.20 mg / mL of cells, for the crude hydrolyzed medium and 0.44 mg / mL to 1.23 mg / mL for the detoxified hydrolyzed medium, in 24 hours of fermentation.

3.1.2.3 Bioethanol production

The production of ethanol by the yeast *Pachysolen tannophilus* during the 24 hours fermentation, with intervals of 2 hours from the 4 hour of fermentation are shown in Figure 8. The results achieved in the production of ethanol were 11.67 mg / mL for synthetic medium in 22 hours, remaining constant in 24 hours, 8.9 mg / mL for hydrolyzed medium in 24 hours and 8.11 mg / mL for detoxified in 22 hours with a small reduction in the period of 24 hours. The synthetic medium showed increasing production up to the 24 hours period while the MH and MHD showed a decrease after the 22 hours period.







In a study of the influence of concentrations of glucose, xylose for the production of ethanol and biomass by Pachysolen tannophilus during a period of 60 hours, where, Zhao et al. (2008) used 20g / L of xylose to 20 g / L of glucose (1: 1) were 8 mg / mL for ethanol production and 5.25 mg / mL of biomass, for the medium containing only glucose the results, it was possible observe a constant drop in the amount of glucose and biomass production after the 12-hour fermentation period, since those that used only xylose as a substrate in the medium had a production of 2.93 mg / mL of ethanol and 6.5 mg / mL of biomass, notice a drop in ethanol production after 50 hours of fermentation. Using the xylose medium, the yeast obtained greater cell growth, however this was not reflected in the production of ethanol, which was inferior to the medium with glucose. For the medium containing glucose and xylose the authors showed results of 4 mg / mL of ethanol during the 35 hours fermentation period, followed by a drop in production after this period for cell growth values were 6.5 mg / mL at end of the fermentation period. These results are inferior to those of this work, where in 8.1 h of fermentation 8.16 mg / mL of ethanol was obtained in the hydrolyzed medium and in 22 hours in 8.11 mg / mL was identified for the detoxified medium.

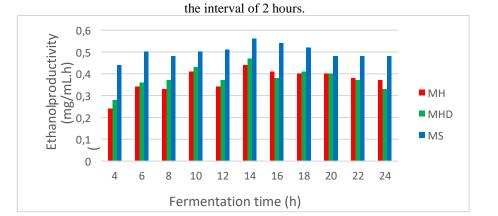
3.1.2.4 Productivity

The productivity of ethanol by the yeast *Pachysolen tannophilus* below compared to *Saccharomyces cerevisiae* the period of greatest productivity was in the period of 14 hours for all media. The results shown in figure 9, makes it possible to observe that the synthetic medium presented better yields when compared to the others reaching 0.56 g / Lh, other results obtained for that period were 0.44 and 0.47 mg / mL.h to the hydrolyzed and detoxified medium, respectively. Despite the fact that in the production of ethanol *P*.



tannophilus continued to produce more and more until the end of the process, in this 14 hours period the yeast adapted to the medium and grew in a way that s. u ethanol production was more efficient and in greater quantity compared to the previous period. The hydrolyzed medium showed lower productivity compared to the detoxified one, however, the period of 20 hours stabilized and later presented a higher productivity.

Fig 9. Productivity of ethanol in fermentation by Pachysolen tannophilus during the period of 24 hours in



The use of immobilized P. tannophilus was positive for ethanol production, Karagoz et al. (2019) reports the use of yeast immobilized in cryogel using glycerol as a substrate. An ethanol production of 8.2 mg / mL and a yield of 0.23 mg / mL were observed. Reporting the efficiency of immobilized yeast, as this technology increases productivity and improves control during the process. The productivity achieved in this research was 0.44 mg / mL.h for the hydrolyzed medium, 0.47 mg / mL .h for the detoxified hydrolyzate is 0.56 mg / mL.h all in 14 hours of fermentation being higher than the productivity achieved by the mentioned author.

3.1.2.5 Consumption of reducing sugars during fermentation

The sugars that were used by the yeast Pachyson tannophilus as a substrate for fermentation and ethanol production to occur. Where the initial concentration was 50 mg / mL for all media. Figure 10 shows during the 24 hours of the process, in which the synthetic medium obtained a higher consumption of sugars, with 15.97 mg / mL remaining, for the hydrolyzate 18.2 and 17.7 mg / mL for the detoxified one.

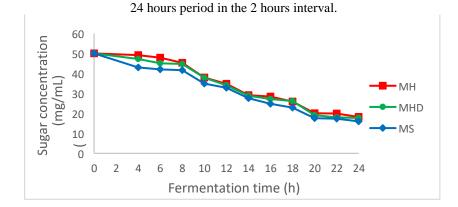
The consumption took place with greater efficiency after the period of 8 hours of fermentation, for all the studied means in this fermentative process. After this period, the





yeast adapted to the medium and obtained greater cell growth, thus leading to a greater consumption of sugars present until the end of the 24 hours fermentation process.

Fig 10. Sugar consumption present in the fermentation media used by Pachysolen tannophilus during the



Liu et al. (2012) used glycerol for the production of ethanol by *Pachysolen tannophilus* during the period of 90 hours of fermentation with an initial concentration of 25 mg / mL. Ethanol production was 6 mg / mL at the end of the evaluated process, in the same period all glycerol in the medium was consumed by the yeast. In the 25 hours period similar to the present study, glycerol consumption was low reaching a concentration of 20 mg / mL in the medium and a production of 2 mg / mL of ethanol for the same time.

3.1.3 Fermentation by intercropping of microorganisms

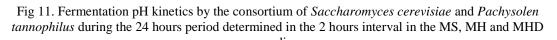
For the intercropping of *Saccharomyces cerevisiae* and *Pachysolen tannophilus* yeasts, the same conditions used for fermentation of hydrolysates and synthetic medium were tested. These conditions were determined by the best results in the tests previously tested for both microorganisms. Concentration of 5% of substrate was used for fermentation with an initial pH of 6.5, temperature of 30 °C and static agitation (0 rpm). Initially, *Saccharomyces cerevisiae* immobilized in sodium alginate during the first 12 hours was inoculated, later it was replaced in the media by *Pachysolen tannophilus* immobilized in sodium alginate. The parameters were analyzed in relation to pH, reducing sugars, ethanol production and cell growth from 4 hours of fermentation in the intervals of 2 hours until the end of 24 hours.

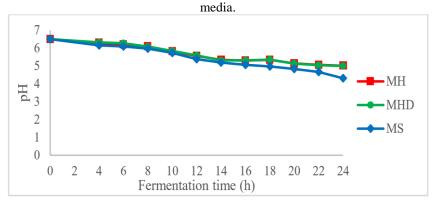
3.1.3.1 Change in pH in the culture medium

The synthetic medium showed a greater drop in pH according to the fermentation time with 4.31 after 24 hours, the hydrolyzed and detoxified hydrolyzed media had a very



similar pH ending the process period with 5.01 and 5 respectively according to figure 11. The replacement of the yeasts occurred after 12 hours, which led to a period of adaptation of *P. tannophilus* in the medium, in view of the drop in pH in these processes, it occurred in a more linear way, as the yeast needed time to develop and produce ethanol.





Ethanol production by kitchen waste (food waste), by the co-fermentation of *Saccharomyces cerevisiae* and *Pichia stipits* were a study by Ntaikou et al. (2018), who used different substrate concentrations of 20 mg / mL, 40 mg / mL, 60 mg / mL, with different concentrations of initial pH 5, 5.4, 5.8 respectively during the 50-hour fermentation period. The media showed a sharp drop during the first 10 hours of fermentation, reaching values of 2.5 pH for all media in 20 hours of fermentation, this value which in turn remained stable until the end in the period of 50 hours. The authors also verified the efficiency of the two yeast intercropping and its capacity to consume a greater amount of sugars present in the medium for ethanol production, as well as the capacity of these residues for the production of bioethanol.

3.1.3.2 Cell Growth (Biomass)

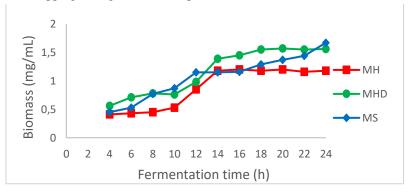
The yeasts *S. cerevisiae* and *P. tannophilus* were used in fermentation media to evaluate cell growth, with the first 12 hours of *S. cerevisiae* followed by *P. tannophilus* by intercropping until the end.

Figure 12 shows the results obtained in the fermentations during the time of 24 hours. This figure shows that over time during the fermentation process, biomass increased for the 3 media used during this process (MH, MHD and MS). Make it possible to note that the synthetic medium (MS) has a higher biomass production at the end of the



24 hour evaluated process compared to the hydrolysates, such as high cell growth in the detoxified hydrolyzed medium that stabilized in the 20 hours process period 1, 57 mg / mL, the hydrolyzed medium produced an amount of 1.18 mg / mL of biomass in 14 hours and stabilized at this value. For the hydrolyzed medium, it was less efficient for the production of biomass compared to the detoxified one using the intercropping of microorganisms.

Fig 12. Kinetics of biomass growth by the *Saccharomyces cerevisiae* and *Pachysolen tannophilus* intercropping during the 24 hours period determined in the 2 hours interval.



The production of biomass and ethanol by consortium of *Saccaromyces cerevisiae* and *Pachysolen tannophilus* in synthetic medium was used by Silva (2016). Parameters such as sugar consumption, ethanol production and biomass were analyzed during the 48 hours fermentation period with an initial substrate concentration of 20 mg / mL. The author determined greater biomass production in the 30 hours period, producing 3.91 mg / mL, 4.74 mg / mL of residual sugar after the 48 hours period, and his greatest ethanol production occurred in the 30 hours period. with 2.98 mg / mL using the intercrop. It was possible to verify in the study, that 100% of the glucose was consumed by the yeasts and that 10.5% of the xylose present was not consumed.

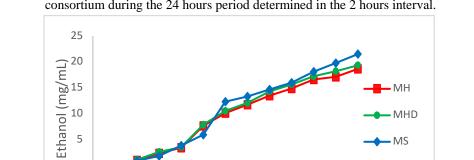
3.1.3.3 Production of second generation ethanol by the microorganism consortium

The maximum values for concentration in ethanol production obtained were 18.65 mg / mL for the hydrolyzate, 19.34 mg / mL for the detoxified hydrolyzate and 21.5 mg / mL for the synthetic medium. The synthetic medium showed better response to the consortium by the parameters used of 30 $^{\circ}$ C, static with an initial pH of 6.5, the efficiency of its production increased after the period of 10 hours and remained in growth until the end of 24 hours observed in the figure 13. The hydrolyzed and detoxified media showed



- MS

a very similar behavior for the production of ethanol during the 24 hours period, this behavior was also possible to be observed during cell growth and pH.



8

Δ 6

Fig 13. Kinetics of ethanol production by the Saccharomyces cerevisiae and Pachysolen tannophilus consortium during the 24 hours period determined in the 2 hours interval.

Sigh et al. (2014) used consortium of Saccharomyces cerevisiae and Scheffersomyces stipitis to obtain ethanol from rice husks, in this study concentrations of 10, 25, 50 and 70 mg / mL were used, obtaining a production of 2.5; 5.5; 12.56 and 20.8 respectively for 36 hours. The yeast intercropping proved to be 41% more effective than monoculture for ethanol production.

Fermentation time (h)

10 12 14 16 18 20 22 24

3.1.3.4 Productivity

0 0 2

The intercropping of the yeasts Pachysolen tannophilus and Saccharomyces *cerevisiae* in relation to their ethanol productivity is shown in Figure 14. It is possible to verify a higher productivity in 12 hours for synthetic medium with 1.03 mg / mL.hem sequence through the detoxified medium with 0, 90mg / mL.hem at 16 hours, the hydrolyzed medium showed a productivity of 0.84 g / L.h times 12, 14 and 16 hours later presenting a fall. The synthetic medium showed better responses after 12 hours followed by a reduction up to 24 hours, however its data were superior in relation to MH and MHD. The synthetic medium showed better response to the first fermentation periods of 4 to 10 hours, being superior to the synthetic and hydrolyzed medium, after presenting its peak in 16 hours, a small reduction occurred until the end of the process.



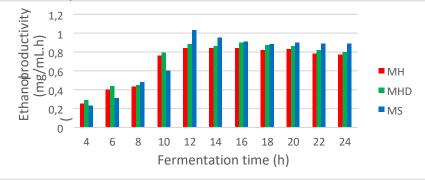


Fig 14. Ethanol productivity by the *Saccharomyces cerevisiae* and *Pachysolen tannophilus* intercropping during the 24 hours period determined in the 2 hours interval in the MS, MH and MHD media.

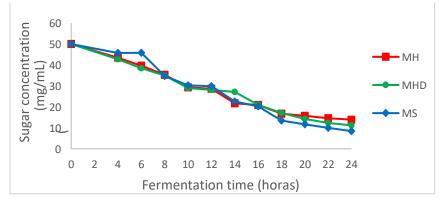
The effect of the immobilization of *Saccharomyces cerevisiae* and *Pachysolen tannophilus* studied in monoculture and in co-culture by Malik et al. (2021) comparing cotton stalk, chemical hydrolysis (alkaline) and enzymatic hydrolysis. The results obtained for the ethanol productivity of *S. cerevisiae* was 0.078 mg / mL.h-1 for the residue, 0.31 mg / mL.h⁻¹ for acid hydrolysis and 0.35 mg / mL.h-1 for the enzyme. For the yeast *P. tannophilus* they presented respective values of 0.058 mg / mL.h-1, 0.23 mg / mL.h⁻¹ and 0.21 mg / mL.h⁻¹. The author emphasizes the improvement of the efficiency in the production of ethanol for the intercropping of both yeasts obtaining values of 0.098 mg / mL.h⁻¹, 0.46 mg / mL.h⁻¹ and 0.44 mg / mL.h⁻¹ for the same parameters studied in monoculture, showing the effectiveness of the cell immobilization process and yeast coculture.

3.1.3.5 Consumption of Reducing Sugars in Fermentation

The reducing sugars present in the fermentation media, which were used by the yeast intercropping for the production of ethanol. The initial concentration was given by 50 mg / mL in the hydrolyzate and in the detoxified hydrolyzate of the sugars present and for the synthetic medium the same concentration using xylose and glucose (1: 1). Figure 15 shows during the 24 hours of the process, in which the synthetic medium obtained a higher consumption of sugars, with 8.46 g / L remaining, for the 13.91 hydrolyzate and 11.21 mg / mL for the detoxified one.



Fig 15. Sugar consumption present in the fermentation media used by the *Saccharomyces cerevisiae* and *Pachysolen tannophilus* intercropping during the 24 hours period in the 2 hours interval.



Malik et al. (2021) used in their study the yeasts *Saccharomyces cerevisiae* and *Pachysolen tannophilus* immobilized, individually and intercropped for the production of ethanol from cotton stalk. Three different substrate concentrations were used initially 90 mg / mL, 260 mg / mL and 280 mg / mL during the 96 hours fermentation period. The behavior of intercropped fermentation proved to be more effective than yeasts in separate process for the consumption of sugars present in the medium. Consortium of immobilized yeasts proved to be more effective in terms of consumption of substrates present in the medium, such as ethanol production, *S. cerevisiae* with chemical treatment had a production of 41.28 mg / mL while *P. tannhophilus* 29.88 mg / mL, when used in combination, showed a production of 56.47 mg / mL and about 92.26% of the use of sugars in the medium. Highlighting the efficiency of yeast immobilization for the ethanol production process.

4 CONCLUSIONS

The yeasts *Saccharomyces cerevisiae* and *Pachysolen tannophilus* showed potential for ethanol production, highlighting the efficiency of *S. cerevisiae* in monoculture for productivity.

The intercropping between the two yeasts proved to be efficient and promising, due to a higher production of ethanol and consumption of fermentable sugars in lignocellulosic material compared to monoculture. Cellular immobilization in sodium alginate was effective for all parameters of the study, emphasizing the importance of this technology to improve performance and facilitate the process.



REFERENCES

Dafnomilis I, Hoefnagels R, Pratama Y W, Schott D L, Lodewijks G, Junginger M (2017) Review of solid and liquid biofuel demand and supply in Northwest Europe towards 2030 – a comparison of national and regional projections. Renewable Sustainable Energy Review. https://doi.org/10.1016/j.rser.2017.04.108

Küüt A, Ilves R, Küüt K, Raide V, Ritslaid K J (2017) Influence of European Union Directives on the use of liquid biofuel in the transport sector. Procedia Engineering 187: 30–39. https://doi.org/10.1016/j.proeng.2017.04.346

Wei P, Cheng L-H, Zhang L, Xu X-H, Chen H-L, Gao C-J (2014) A review of membrane technology for bioethanol production. Renewable and Sustainable Energy Reviews 30: 388-400. https://doi.org/10.1016/j.rser.2013.10.017

Maldhure A V, Ehle J D (2013) Pyrolysis of purified kraft lignin in the presence of AlCl3 and ZnCl2. Journal of Environmental Chemical Engineering 1: 844-849. https://doi.org/10.1016/j.jece.2013.07.026

Furtado A T, Hekkert M P, Negro S O (2020) Of actors, functions, and fuels: Exploring a second generation ethanol transition from a technological innovation systems perspective in Brazil. Energy Research and Social Science 70. https://doi.org/10.1016/j.erss.2020.101706

Kumari D, Singh R (2018) Pretreatment of lignocellulosic wastes for biofuel production: A critical Review. Renewable and Sustainable Energy Reviews 90: 877–891. https://doi.org/10.1016/j.rser.2018.03.111

Xue J, Wu Y, Fu X, Li N, Sun J, Qiao Y (2020) Study on degradation characteristics and bacterial community structure changes of imobilized cells in straw-alginate beads in marine environment. Bioresource Technology Reports 10. https://doi.org/10.1016/j.biteb.2020.100402

Behera S, Kar S, Mohanty R C, Ray R C (2010) Comparative study of bio-ethanol
production from mahula (*Madhuca latifolia L.*) flowers by *Saccharomyces cerevisiae* and
Zymomonasmobilis. Applied Energy 87: 96-100.
https://doi.org/10.1016/j.apenergy.2009.11.018

Saleh M, Cuevas M, Garcia J F, Sanchez S (2014) Valorization of olive stones for xylitol and ethanol production from dilute acid pretreatment via enzymatic hydrolysis and fermentation by *Pachysolen tannophilus*. Biochemical Engineering Journal 90: 286–293. https://doi.org/10.1016/j.bej.2014.06.023

Gandu B, Rozenfeld S, Hirsch L O, Schechter A, Cahan R (2020) Immobilization of bacterial cells on carboncloth anode using alginate for hydrogen generation in a microbial electrolysis cell. Journal of Power Sources 455. https://doi.org/10.1016/j.jpowsour.2020.227986

Dubois M, Gilles K A, Hamilton J K, Rebers P A, Smith F (1956) Colorimetric method for determination of sugars and related substances. Analytical Chemistry St. Paul 28: 350-356. https://doi.org/10.1021/ac60111a017 Somogyi M (1952) Notes on sugar



determination. Journal of Biological Chemistry 195: 19-22. https://doi: 10.1016/S0021-9258(19)50870-5

Nelson N (1944) A photometric adaptation of Somogy method for determination of glucose. The Journal of Biological Chemistry 153: 375-380. https://doi.org/10.1016/S0021-9258(18)71980-7

Chaovanalikit A, Wrolstad R E (2004) Total anthocyanins and total phenolics of fresh and processed cherries and their antioxidant properties. Food Chemistry and Toxicology 69: 67-72. https://doi.org/10.1111/j.13652621.2004.tb17858.x

Yoshikawa K, Hirasawa ., Shimizu H (2015) Effect of malic enzyme on ethanol production by *Synechocystissp.* PCC 6803. Journal of Bioscience and Bioengineering 119: 82-84. <u>https://doi.org/10.1016/j.jbiosc.2014.06.001</u>

Lainioti G C, Kapolos J, Koliadima A, Karaiskakis G (2010) New separation methodologies for distinction of the growth phases of *Saccharomyces cerevisiae* cell cycle. Journal of chromatography A 1217: 1813-1820. https://doi.org/10.1016/j.chroma.2010.01.042

Mishra A, Sharma A. K, Sharma S, Bagai R, Mathur A S, Gupta R P, Tuli D K (2016) Lignocellulosic ethanol production employing immobilized *Saccharomyces cerevisiae* in packed bed reactor.Renewable Energy: 1-7. https://doi.org/10.1016/j.renene.2016.02.010

Sakomoto T, Hasunuma T, Hori Y, Yamada R, Kondo A (2012) Direct etanol production from hemicellulosic materials of rice straw by use of an engineered yeast strain codisplaying three types of hemicellulolytic enzymes on the surface of xylose-utilizing *Saccharomyces cerevisiae* cells. Journal of Biotechnology 158: 203-210. https://doi.org/10.1016/j.jbiotec.2011.06.025

Boboescu I-Z, Gelinas M, Beigbeder J-B, Lavoi J-M (2017) A two-step optimization strategy for 2nd generation ethanol production using softwood hemicelullosic hydrolysate as fermentation substrate. Bioresource technology. https://doi.org/10.1016/j.biortech.2017.07.107

Romero I, Sanchez S, Moya M, Castro E, Ruiz E, Bravo V (2007) Fermentation of olive tree prining acidhydrolysates by *Pachysolen tannophilus*. Biochemical Engineering Journal 36: 108-115. https://doi.org/10.1016/j.bej.2007.02.006

Zhao L, Zhang X, Tan T (2008) Influence of various glucose/xylose mixtures on ethanol production by *Pachysolen tannophilus*. Biomass and Bioenergy 32: 1156–1161. https://doi.org/10.1016/j.biombioe.2008.02.011

Karagoz P, Bill R M, Ozkan M (2019) Lignocellulosic ethanol production: Evaluation of new approaches, cell immobilization and reactor configurations. Renewable Energy. https://doi.org/10.1016/j.renene.2019.05.045 Liu X, Jensen P R, Workman M (2012) Bioconversion of crude glycerol feedstocks into ethanol by *Pachysolen tannophilus*. Bioresource Technology 104: 559-586. https://doi.org/10.1016/j.biortech.2011.10.065 Ntaikou I, Menis N, Alexandropoulou M, Antonopoulou G, Lyberatos G (2018) Valorization of kitchen biowaste for ethanol production via simultaneous saccharification



and fermentation using co-cultures of the yeasts *Saccharomyces cerevisiae* and *Pichia stipites*.Bioresource Technology. https://doi.org/10.1016/j.biortech.2018.04.109

Da Silva, D. C. M. PRE-TRATAMENTO E HIDROLISE DA CASCA DE UVA PARA LIBERACAO DE ACUCARES FERMENTESCIVEIS. 2016. 113 f. Tese (Doutorado em Engenharia e Ciência de alimentos) – Universidade Estadual Paulista, Instituto de Biociências, Letras e Ciências Exatas, São Jose do Rio Preto, 2016 Singh A, Bajar S, Bishnoi N R (2014) Enzymatic hydrolysis of microwave alkali pretreated rice husk for ethanol production by *Saccharomyces cerevisiae, Scheffersomyces stipites* and their coculture.Fuel 116: 699-702. https://doi.org/10.1016/j.fuel.2013.08.072

Malik K, Salama E-S, El-Dalatony M. M, Harraz F A, Al-Assari M. S, Zheng Y, Sharma P, Li X (2021) Cofermentation of immobilized yeasts boosted bioethanol production from pretreated cotton stalk lignocellulosic biomass: Long-term investigation. Industrial crops e Products 159 https://doi.org/10.1016/j.indcrop.2020.113122