

Behaviour of Lithium During Stripping – Isolated Lithium Formation

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Lithium metal (Li) batteries are promising candidates for next-generation high-energy density batteries, yet they suffer from low cycle life [1]. It is commonly observed that Li gets isolated from the current collector during cycling, diminishing the available capacity [2]. This isolated lithium is trapped in an insulating solid-electrolyte interphase (SEI) shell [2,3]. However, a fundamental understanding of how Li gets isolated is still lacking. Here, we present a combined experimental and theoretical study to uncover the origin of isolated Li formation [4]. We derive a thermodynamic consistent model for stripping of a single Li whisker, accounting for the interaction between Li and SEI. We find that this interaction leads to locally preferred stripping and isolated Li formation upon further stripping. Cryo transmission electron microscopy investigations of Li whiskers during stripping reveal that these local effects are pronounced at kinks and the tip of Li whiskers. Sources for heterogeneity can be locally varying geometry or heterogeneous SEI. Further, simulations reveal that higher stripping current densities lead to less isolated Li formation. This can be understood in terms of the auto-inhibitory behaviour of the stripping process, where the instability is suppressed above a critical current [5]. We conclude that isolated Li can only be fully avoided when Li-deposition is planar, and the SEI is homogeneous.

References

- [1] B. Horstmann, et al., "Strategies towards enabling lithium metal in batteries: interphases and electrodes", *Energy Environ. Sci.*, vol. 14, pp. 5289–5314, 2021.
- [2] C. Fang, et al., "Quantifying inactive lithium in lithium metal batteries", *Nature*, vol. 572, pp. 511–515, 2019.
- [3] J. Steiger, D. Kramer, & R. Mönig, "Microscopic observations of the formation, growth and shrinkage of lithium moss during electrodeposition and dissolution", *Electrochim. Acta*, vol. 136, pp. 529–536, 2014.
- [4] M. Werres et al. "Origin of heterogeneous stripping of lithium in liquid electrolytes", arXiv: 2301.04018v1, 2023.
- [5] M.Z. Bazant, "Thermodynamic stability of driven open systems and control of phase separation by electroautocatalysis", *Faraday Discuss.*, vol. 199, pp. 423–463, 2017.