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Various Methods of Reducing Calcium Oxalate Levels in Tubers: a Review

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1	Various Methods of Reducing Calcium Oxalate Levels in Tubers: a Review
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12	
13	Abstract
14	Tubers, as a highly promising agricultural commodity with distinctive flavors and nutritional
15	content, pose health challenges due to the presence of calcium oxalate. Excessive
16	consumption can lead to mechanical disturbances in the digestive and renal tubule systems.
17	This complicates processing techniques to ensure that tubers, before being used as raw
18	materials in the food and other industries, do not contain calcium oxalate. The research
19	method involves a literature review, in-depth exploration of theories and concepts, and
20	identification of variables related to the research method and context. From the literature
21	study, it is evident that focusing on understanding fermentation parameters, such as time,
22	temperature, pH, and the type and concentration of microorganisms, is crucial. This
23	information is expected to provide insights into potentially effective methods for reducing
24	calcium oxalate content in tubers, supporting the development of safer and healthier
25	agricultural and food processing practices.
26	
27	Keywords: Calcium oxalate; Various; Potential method; Reduction; Tubers.
28	r

29 Introduction

Tubers can thrive in Indonesia (Hatmi & Djaafar, 2014), but the level of consumption among the population remains low. According to the Food Security Agency (BKP) of Agriculture (2020), the tuber consumption value in 2019 was 15.9 kg/capita/year, which is still below the recommended 36.5 kg/capita/year. Tubers are known as a source of 34 carbohydrates in daily food. The utilization of tubers is closely related to the recognition of 35 their characteristics (Utami & Djaafar, 2014).

36 One type of tuber containing calcium oxalate crystals is the konjac tuber, which is almost evenly distributed throughout its central and peripheral parts (Chairiyah et al., 2011). 37 38 Consumption of calcium oxalate can cause itching, burning sensations, irritation in the mouth, 39 skin, and digestive tract (Dewi et al., 2017). Oxalic acid can bind calcium and is poorly 40 absorbed by the body. Excessive consumption of calcium oxalate and oxalic acid can lead to 41 kidney problems (Sari & Suhartati, 2015; Mitchell et al., 2019). Additionally, in terms of chemical structure, calcium oxalate crystals absorb calcium, which is essential for nerve 42 43 function and muscle fibers. In extreme cases, calcium absorption can lead to hypocalcemia 44 and fatal paralysis (Ardhian & Indrivani, 2013).

To eliminate calcium oxalate content in tubers, pre-treatment is necessary before they 45 46 are used as raw materials in the food and other industries. Therefore, in this paper, we discuss several potential methods to reduce calcium oxalate in tubers. 47

48

49 Materials and methods

The data is sourced from a search through the electronic databases Web of Science 50 51 (WoS) and Scopus, as leading scientific information platforms that access the most significant 52 scientific databases and publications across various fields of knowledge. To develop a profound understanding of the topic, the authors conducted a systematic literature review of 53 54 articles reviewed by peers over the last 10 years. Out of a total of 86 identified key studies, the authors categorized three method definitions for reducing calcium oxalate content in 55 56 tubers as the focus of the study: mechanical, chemical, and biological methods. After 57 reporting the findings of the systematic literature review, the authors discuss the best methods to reduce factors related to the reduction of calcium oxalate content in tubers. 58

59

Results and Discussion 60

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Calcium Oxalate Reduction Method in Tubers the results of a review of research 62 articles that reduce calcium oxalate in tubers can be seen in Figure 1.

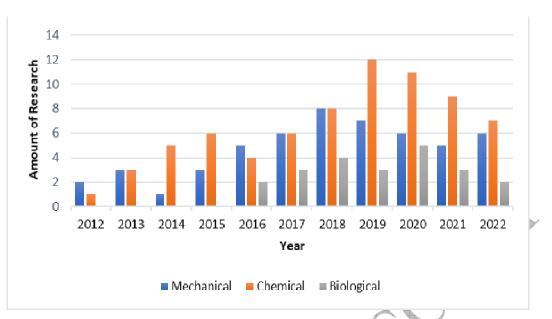


Figure 1. Research that has been carried out to reduce calcium oxalate in tubers

The figure 1 above explains three methods used to reduce the calcium oxalate content in tubers: mechanical, chemical, and biological. The processing processes of these three methods all begin with physical treatment in the form of stripping, washing, cutting, slicing, or crushing. This is adjusted to the advanced Method that will be used, whether mechanical, chemical, or biological.

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64 65

72 1. Chemical Method

Calcium oxalate can be eliminated through chemical treatment, such as dissolving it in strong acid to decompose it into oxalic acid (Lukitaningsih et al., 2012; Muttakin, 2015). Citric acid can penetrate the idioblast cell walls and dissolve calcium oxalate crystals (Purwaningsih & Kuswiyanto, 2016). Soaking in salt (NaCl) reduces calcium oxalate content through a reaction with Na+ and Cl- ions that attract calcium oxalate in the material (Muttakin, 2015). Table 1 summarizes various mechanical and chemical methods for reducing calcium oxalate in tubers from research conducted in the last five years.

80

81 Table 1. Several mechanical methods for reducing calcium oxalate in tubers

No	Method and Results	Refrences
Tub	ers Porang	
1	Boiling the chips using $6 - 8\%$ NaCl solution at 80° C for $5 - 30$ minute.	(Widari &

The higher the NaCl concentration and the longer the boiling time, the Rasmito, 2018) lower the calcium oxalate level in porang tuber flour. The best treatment was obtained by using 8% NaCl and boiling for 25 minutes with calcium oxalate levels 0.65%.

Soak 2 g of porang flour in 200 mL of 3 – 7% starfruit juice, 3 – 7% (Wardani *et al.*, lime juice, and 10 – 20% vinegar acid for 15 minutes. 2019)
The higher the concentration of the starfruit juice solution, the higher the percent reduction of calcium oxalate in porang tuber flour, namely 62.68%. The same thing happened when using acetic acid solution with a reduction percentage of calcium oxalate in porang tuber flour of 90.27%. Meanwhile, using lime juice solution did not show a higher percent reduction in calcium oxalate in porang tuber flour as the concentration of lime juice solution increased. Percent reduction of calcium oxalate in porang tuber flour as the concentration of lime juice flour was 65.94% when using a 5% lime juice solution.
3 Soaking 50 g of chips in 250 mL of 3 – 7% starfruit juice solution for 3 (Wardani &

soaking 50 g of entrys in 250 mL of 3 - 770 starting fuce solution for 3(wardam &x 15 minutes.Handrianto,The higher the concentration of starfruit juice solution, the higher the2019)percent reduction of calcium oxalate in porang tuber flour, namely37.38%.

4 Soak the chips in water for 0 – 180 minutes without peeling and with (Hadi & Kurniawan,

2020b)

The longer the soaking time, the levels of oxalate compounds in porang tuber flour decrease, both in the treatment without peeling and with peeling. In addition, in the treatment with peeling, the levels of oxalate compounds in porang tuber flour were smaller when compared to the treatment without peeling, where the levels of oxalate compounds were 197.57 ppm and 86.44 ppm respectively after soaking 180 minutes.

5 Soak 50 g of chips in 250 mL of 5% H2SO4 solution at room (Wardani & temperature, 40°C, 50°C and 60°C for 15 minutes. Arifiyana, The higher the soaking temperature, the percent reduction of calcium oxalate in porang tuber flour increased, namely 53.91% at a soaking

temperature of 60°C.

Boil the chips using 15% NaCl solution at 80°C for 25 minutes, then (Handayani *et* rinse with water and drain, and continue soaking in 0.02% sodium *al.,,* 2020) bisulfite solution for 10 minutes.

The method used can reduce 96.00% of calcium oxalate in porang tuber flour.

7 Soak 50 g chips in 250 mL of 5% lime solution at room temperature for (Wardani & 15, 30, 45, and 60 minutes.
The longer the soaking time in 250 mL of 5% lime solution at room 2021)

temperature, the percent reduction of calcium oxalate in porang tuber flour increased, namely 31.79% at a soaking time of 60 minutes.

8 Combination of soaking in 5% salt solution for 4 hours and blanching at 9 (Zainuri *et* for 15 minutes. *al.*, 2021)

The combination of the method of soaking in a 5% salt solution for 4 hours and blanching at a temperature of 90°C for 15 minutes can reduce the calcium oxalate content in porang tuber flour by 51.85%, which is greater than using just one method (soaking in the solution). 5% salt for 4 hours can reduce 14.81% calcium oxalate in porang tuber flour, while Blanching at 90°C for 15 minutes can reduce 18.52% calcium oxalate in porang tuber flour.

Purple Taro Tubers

Soak 100 g chips in 250 mL NaCl 5 – 15% at room temperature for 60 (Rofi'ana et al. minutes. 2018) Soaking 100 g of chips in 250 mL of 10% NaCl at room temperature for 60 minutes can reduce 22.89% of calcium oxalate in purple taro tuber flour and is the best treatment. Boiling or steaming the chips for 20 minutes (500 g chips in 1 L of water 2 (Sulaiman et at 100 °C) and soaking in a chemical solution (6% NaHCO3, 20% al., 2021) CH3COOH, and 10% NaCl).. Boiling and soaking in 10% NaCl for 60 minutes can reduce 75.03% of calcium oxalate in purple taro tubers and is the best treatment. 3 Soak 500 g chips in 1 L CH3COOH 20% for 30 minutes, 1 L NaCl 10% (Rozali et al.,

for 15 minutes, and 1 L NaHCO3 6% for 60 minutes, each at room 2021)

temperature.

Soaking 500 g chips in 1 L of 6% NaHCO3 for 60 minutes at room temperature can reduce 45.58% of calcium oxalate in purple taro tubers and is the best treatment.

Tubers Iles-iles

- Soak 1.5 kg chips in 0 15% NaCl solution at room temperature for 60 (Ulfa & minutes after soaking in warm water at 40°C for 3 hours. Nafi'ah, 2018) The higher the NaCl concentration, the percent reduction of calcium oxalate in iles-iles tuber flour increased, namely 91.60% at a NaCl concentration of 15%.
 Soaking chips at room temperature using 7% lime solution or 7% (Riyanto *et*
- CaCO3 CaCO3 Chips at room temperature using 7% lime solution or 7% (Riyanto *et al.*, 2020)

The method used can reduce 60 - 70% of calcium oxalate in iles-iles tuber flour.

calcium oxalate in kimpul tubers and is the best treatment.

Tubers Kimpul

Boiling or steaming the chips for 20 minutes (500 g chips in 1 L of water (Sulaiman et at 100°C) and soaking in a chemical solution (6% NaHCO3, 20% al, 2021) CH3COOH, and 10% NaCl).
Boiling 500 g of chips in 1 L of water at a temperature of 100 °C and soaking in 20% CH3COOH for 30 minutes can reduce 60.12% of

82

83 2. Mechanical Method

The mechanical method reduces calcium oxalate content in products by separating oxalate 84 85 crystals through size reduction using screening or filtration processes (Chairiyah et al., 2013). Although fast, this method requires significant energy, increases operational costs, and has the 86 87 potential to pollute the environment. Faridah et al. (2012) research showed that optimizing 88 mechanical milling of konjac flour resulted in a calcium oxalate content of 0.3%. Widjanarko et al.'s 89 (2015) study highlighted the impact of ball mill grinding time on yield, viscosity, and konjac flour 90 glucomannan content, with the best results obtained after 4 hours of grinding. The ball mill method 91 with chemical purification in 8 hours of grinding (L8) produced konjac flour with a calcium oxalate 92 content of 0.89%, flour size of 180-322.7 µm, and whiteness degree of 69.65 (Mawarni & 93 Widjanarko, 2015).

94

95 *3. Biological Methods*

96 Cultivation process is an alternative method for reducing oxalate content in food 97 materials by utilizing microbes that use oxalate as a source of nutrition. Several bacterial 98 strains, such as Lactobacillus sp., have proven to degrade calcium oxalate, with examples like 99 Lactobacillus plantarum and Lactobacillus acidophilus each having the ability to degrade 100 oxalate by 42% and 38%, respectively. Bacteria like Bacillus sp. also have the capability 101 through the enzyme oxalate decarboxylase (OxdC) (Akulenko et al., 2020; Conter et al., 102 2019). The fermentation process in kimchi, rich in oxalate, shows a reduction of up to 72.3% in swiss chard (Wadamori et al., 2014). Lactic acid bacteria, especially Lactobacillus sp., in 103 104 kimchi have the potential to degrade oxalate (Turroni et al., 2007). Table 2 presents biological 105 methods that, in the last five years, have assisted in reducing calcium oxalate in tubers with 106 the aid of initial physical treatments.

107

Table 2. Several biological methods used to reduce calcium oxalate in tubers with the help of physical treatment in the initial process

No	Method and Results	Reference	
Tubers Porang			
1	Ferment chips for 21 days at room temperature using Bacillus subtilis	(Koni et al.	
	which has been adapted to media containing 10% porang tubers for 4 days	2017)	
	$(pH = 5.5 and temperature = 37^{\circ}C)$		
	Fermentation with the addition of $10 - 30\%$ Bacillus subtilis (dbd)		
	can reduce calcium oxalate levels in porang tuber flour by around 58		
	- 65% without changing the nutritional composition.		
2	Wet milling and fermentation using Lactic Acid Bacteria (LAB) with a	(Ferdian &	
	LAB:water ratio = $1:1000$ for $6-24$ hours.	Perdana,	
		2021)	
	The longer the fermentation time, the calcium oxalate levels in porang		
	tuber flour decrease, except for the 24 hour fermentation time which is		
	higher compared to other fermentation times. In this method, the best		
	treatment was produced during fermentation for 18 hours with a calcium		
	oxalate content in tuber flour of 2.64%.		
3	Dry grinding and long fermentation time using Lactic Acid Bacteria (LAB)	(Ferdian &	

with a LAB:water ratio = 1:1000 for 6-24 hours. Pedana, 2021)

The longer the fermentation time, the calcium oxalate levels in porang tuber flour increase, except for the 12 hour fermentation time which is lower compared to other fermentation times. In this method, the best treatment was produced during fermentation for 12 hours with a calcium oxalate content in tuber flour of 4.08%.

- 4 Ferment chips (thickness ± 0.5 cm) at room temperature for 0 72 hours. (Sulastri *et al.*, The longer the fermentation time, the calcium oxalate levels in porang tuber flour increase, except for the 24 hour fermentation time which is lower compared to other fermentation times.
- 5 Ferment chips (thickness ± 0.5 cm) at room temperature using kimchi (Ridhoha & bacteria (Lactobacillus sp. and Bacillus sp.) at pH 5-7 and fermentation Suryadarma, time for 24-72 hours.
 2022)

The fermentation method at different pH conditions and fermentation times produces calcium oxalate levels in porang flour ranging from 0.050–0.069 mg/g, where fermentation for 24 hours at pH 6 is the best treatment.

Taro Tubers

1Ferment chips naturally (Chips:Air = 1:2) for 0–72 hours.(Indrastuti etThe longer the fermentation time, the calcium oxalate levels in taro tuberal., 2021)flour decrease, except for the fermentation time of 72 hours which is highercompared to the fermentation time of 48 hours.

Tubers Kimpul

A total of 500 g of chips was fermented in a closed bottle (bottle capacity (Sulaiman *et* 1.5 L) using Saccharomyces cerevisiae and Rhizopus inoculum for 48 and *al.*, 2020)
 72 hours.

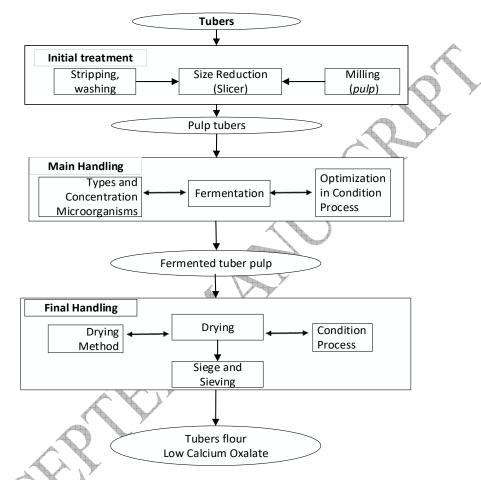
Fermentation for 72 hours produced calcium oxalate levels in kimpul flour of 0.334 mg/g, which was lower compared to fermentation for 48 hours which produced calcium oxalate levels in kimpul flour of 0.472 mg/g.

110

111 4. Initial Treatment of Tubers

The initial treatment for reducing calcium oxalate in tubers involves three subprocesses: peeling and washing, size reduction (slicing), and grinding (pulping) (Figure 2). Combining these subprocesses produces clean and small tuber pulp, impacting the calcium oxalate reduction process during fermentation (Ferdian & Perdana, 2021). Peeling and 116 washing at the beginning affect the resulting tuber flour's calcium oxalate levels, with the 117 peeling method showing a reduction (Hadi & Kurniawan, 2020b). Smaller pulp size facilitates 118 microbial reduction, and wet milling produces lower levels than dry milling (Ferdian & 119 Perdana, 2021). Future research need not optimize initial handling conditions.

120



121

Figure 2. Scheme of calcium oxalate reduction in tubers via biological or fermentation
 pathways

124

125 The primary treatment for reducing calcium oxalate in tubers involves one subprocess: 126 fermentation (Figure 2). Identifying process conditions (time, temperature, pH, 127 microorganism type, and concentration) is crucial for optimum conditions in reducing calcium 128 oxalate levels in tuber flour. Fermentation process optimization is a key agenda for future 129 research. The final treatment for reducing calcium oxalate in tubers includes two sub-130 processes: drying, flouring, and sieving (Figure 2). Combining these subprocesses is 131 applicable to producing low water and calcium oxalate content in tuber flour. Understanding

drying method and conditions is essential for maintaining flour quality post-fermentation. 132 133 Previous research indicates using an oven or cabinet dryer at 60°C for 6 - 24 hours for drying 134 tuber flour after the initial handling process. Sunlight drying has also been explored (Ferdian 135 & Perdana, 2021). Flouring and sieving depend on sieve type and size, with the hypothesis that using a sieve above 135 µm or a 100 mesh sieve allows calcium oxalate crystals to pass 136 137 through, resulting in low calcium oxalate content in tuber flour. Future research need not optimize conditions for drying, flouring, and sieving subprocesses in the final handling of 138 139 fermented tuber flour.

140

141 Conclusions

Reducing calcium oxalate levels in tubers can be achieved through three methods: 142 mechanical, chemical, and biological. Biological methods are considered potential because 143 144 they are safer, utilizing microbes beneficial to the body, and can reduce production costs 145 compared to other methods. Chemical methods involve high costs, limited availability, and have the potential to impact health if used over a prolonged period. Mechanical methods, 146 although fast, require high costs and energy and are challenging to apply on a small scale. In 147 biological methods, fermentation parameters such as time, temperature, pH, and the type of 148 149 microorganism need to be identified for optimal results.

150

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